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Part I. The First Enantiospecific, Stereospecific Total Synthesis of the Indole Alkaloid Ervincidine. Part II. The Synthesis of Alpha 5 Subtype Selective Ligands for Gaba(a) /Benzodiazepine Receptors

Sundari K. Rallapalli University of Wisconsin-Milwaukee

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# PART I. THE FIRST ENANTIOSPECIFIC, STEREOSPECIFIC TOTAL SYNTHESIS OF THE INDOLE ALKALOID ERVINCIDINE PART II. THE SYNTHESIS OF ALPHA 5 SUBTYPE SELECTIVE LIGANDS FOR GABA(A)/BENZODIAZEPINE RECEPTORS

by

Sundari K. Rallapalli

A Dissertation Submitted in

Partial Fulfillment of the

Requirements for the Degree of

Doctor of Philosophy

in Chemistry

at

The University of Wisconsin-Milwaukee

December 2012

#### ABSTRACT

# PART I. THE FIRST ENANTIOSPECIFIC, STEREOSPECIFIC TOTAL SYNTHESIS OF THE INDOLE ALKALOID ERVINCIDINE PART II. THE SYNTHESIS OF ALPHA 5 SUBTYPE SELECTIVE LIGANDS FOR GABA (A)/BENZODIAZEPINE RECEPTORS

by

Sundari K. Rallapalli

The University of Wisconsin-Milwaukee, 2012 Under the Supervision of Professor James M. Cook

.

Part I. The first enantiospecific, stereospecific total synthesis of ervincidine **89** has been accomplished from commercially available D-(+)-tryptophan **37** which has served both as the chiral auxiliary and the starting material. Moreover, this is the first synthesis which unequivocally sets the stereochemistry of the hydroxyl group at C-6 in sterospecific fashion, as well as the C-16 hydroxy methyl group. The stereospecific conversion of D-(+)-tryptophan **37** into the key template (–)- $N_a$ -H,  $N_b$ -benzyl tetracyclic ketone **49** via the asymmetric Pictet-Spengler reaction (600 gram scale) and Dieckmann cyclization on multi-hundred gram scale was reduced to only two reaction vessels. The optically active tetracyclic ketone **49** was converted into the core pentacyclic framework **56** using the intramolecular palladium-mediated enolate cross coupling reaction which was developed here in Milwaukee to afford the core pentacylic framework 56. This robust reaction could be scaled up to multigram scale in this series. Important to success here were the sequence of chemical reactions which included a Wittig reaction, a regioselective hydroboration and protection/deprotection steps in order to provide regiospecific oxidation at C-6. The IBX mediated oxidation and the Luche reduction using CeCl<sub>3</sub>.  $7H_{20}$  in the presence of NaBH<sub>4</sub> afforded the first enantiospecific, stereospecific total synthesis of ervincidine **89**. The indole alkaloid ervincidine **89** could be prepared from D-(+)-tryptophan 37 in 13 reaction vessels in 19.2% overall yield. Another important experiment was the epimerization of the C-6 alcohol with 0.2N HCl which indicated that care must be employed in isolation of these alkaloids which contain a benzylic hydroxyl group. This research process developed here also provides a general entry to the C-6 hydroxy substituted indole alkaloids of either alpha or beta stereochemistry. Two other diasteromers were made to rule them out as potential structures. This research corrects the errors in Glasby's book and Lousnamma's review and clarifies the work of Yunusov et al. as well as providing the correct absolute configuration of the C-6 hydroxyl function in ervincidine 89.

Part II. GABA<sub>A</sub>/BzR chloride ion channels comprise the major inhibitory neurotransmitter system in the CNS. This central role carries with it a direct influence on many diseases of the CNS. Inverse agonists acting at  $\alpha$ 5 subunits containing GABA<sub>A</sub> receptors are thought to act as cognitive enhancers while eliminating unwanted side effects associated with non-selective compounds. From the recent work of Rowlett, Cook et.al. it was demonstrated that novel  $\alpha$ 5 selective inverse agonist PWZ-029 was evaluated as a cognitive enhancer in rhesus monkeys in the CANTAB paradigm. This ligand had the ability to reverse cholinergic deficits in performance induced by the antimuscarinic scopolamine under mixed trial conditions. In the ORD task, PWZ-029 showed only a modest trend for enhancement of performance, but when task difficulty was increased by testing with difficult trials only, PWZ-029 robustly increased performance. This enhancement was reversed by administration of the  $\alpha$ 5 GABA (A) subtype selective antagonist XLi-093 and this antagonism in turn was reversed by increasing the dose of PWZ-029. In addition, PWZ-029 enhanced performance in the DNMS task using the 10 minute delay with distracters. This ligand also exhibited anxiolytic activity in some primates and was an orally active anticonvulsant in rats. These findings are consistent with a key role for  $\alpha$ 5 GABA<sub>A</sub> receptors in the treatment of age-associated memory impairment and Alzheimer's disease.

То

My Parents Sugunavathi and Satyanarayana Rallapalli

and

My Husband Srihari Vedartham

and

My Son Anirudh Vedartham and Daughter Amulya Vedartham

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# I. THE FIRST ENANTIOSPECIFIC, STEREOSPECIFIC TOTAL SYNTHESIS OF THE INDOLE ALKALOID ERVINCIDINE

#### **1. Introduction**

# **1.1. Classification and biological activity of the sarpagine/macroline related indole** alkaloids

Alkaloids are of importance due to their biological activity. They are distributed diversely and have been isolated from plant and animal sources worldwide.<sup>1-5</sup> Indole alkaloids are of prominence because of the similarity with tryptophan which is an essential amino acid. The medicinal properties<sup>6</sup> of these indole alkaloids remain of great interest as well as the nature of their structure and stereochemistry. From the synthetic point of view, these complex molecules remain of paramount significance. The enantiospecific and regiospecific synthesis of alkaloids or their enantiomers has always been a challenge to the synthetic chemist.<sup>7-9</sup> Synthetic routes are designed and revised continually in order to provide material for testing and for commercial uses. For all of these reasons, indole alkaloids play an important role in organic chemistry. This chapter includes the various classes of ajmaline and sarpagine-related indole alkaloids and also provides the previous synthetic methods to construct these molecules.<sup>10</sup>

#### Biological activity of sarpagine/macroline related indole alkaloids

Indole alkaloids of the sarpagine/macroline/ajmaline type comprise one of the largest groups of structurally related indole natural products.<sup>10</sup> These indole alkaloids are primarily isolated from three plant families: *Apocynaceae, Rubiaceae* and *Loganiaceae*.<sup>11-13</sup> Interest in the macroline/sarpagine alkaloids originated as a result of folk tales that described the medicinal properties of the plants from which these alkaloids were isolated.<sup>9,14</sup> Various species of *Rauwolfia* which are broadly distributed among Asia

and Africa <sup>15-20</sup>are widely used in traditional Chinese medicine for the treatment of neuralgia, migraine<sup>21</sup> and hypertension.<sup>18,20,22</sup>

As early as 1958,<sup>23</sup> and later in 1967<sup>24</sup> the bisindole macralstonine **1a** (Figure 1), which had been isolated from *Alstonia macrophylla* Wall and *Alstonia muelleriana* Domin, was shown to exhibit potent hypotensive activity.

Nine alkaloids from *Alstonia angustifolia* were tested by Wright et al.<sup>25</sup> for antiprotozoal activity against *Entamoeba histolytica* and *Plasmodium falciparum* in vitro (Table 1). Three dimeric alkaloids, macralstonine O-acetate **1b**,<sup>26</sup> (-)-macrocarpamine **2a**,<sup>27</sup> and (+)-villalstonine **3a**<sup>28, 29</sup> (Figure 1) were found to possess significant activity against both protozoa. Macrocarpamine **2a** was found to be the most active antiamoebic compound with an antiamoebic activity one fourth that of the standard drug emetine. Villalstonine **3a** was found to be the most potent alkaloid against *P. falciparum*, and was about fifteen

#### Figure 1. Bisindoles







**2a** R=H, macrocarpamine **2b** R=OMe,10-methoxy macrocarpamine

**3a** R = H, villalstonine **3b** R = OMe,10-methoxy villalstonine

times less active than the antimalarial drug chloroquine. The acetate of macralstonine **1b** was much more active against both protozoa when compared with the parent macralstonine **1a**. The monomeric alkaloids were found to be considerably less active than the bisindoles. These results were in agreement with the studies carried out later by Houghton et al. (Tables 2 and 3).<sup>30</sup> The results of these *in vitro* studies provide some basis for the traditional use of the plant extract from *Alstonia angustifolia* for the treatment of amoebic dysentery and malaria by the people of the Malay peninsula.<sup>31,32</sup>

R

Me

Me

Alkaloids	<i>E. histolytica</i> ED <sub>50</sub> (95% C. I.) μM	<i>P. falciparum</i> ED <sub>50</sub> (95% С. І.) µМ
alstonerine	75.3(65.0-85.6)	46.3(27.7-77.3)
alstophylline	67.7(57.2-78.2)	82.3(65.9-102)
macralstonine	inactive at 70	inactive at 178
O-acetyl macralstonine	15.51(14.78-16.24)	3.43(1.86-6.34)
villalstonine	11.8(11.7-12.0)	2.92(1.11-3.14)
pleiocarpamine	47.4(46.8-52.90)	20.5(12.6-33.17)
macrocarpamine	8.21(7.76-8.48)	9.36(7.20-12.1)
11-methoxyakuammicine	70.5(65.3-75.6)	41.3(26.5-64.3)
norfurocurarine	84.1(82.9-89.6)	129(70.2-239)
vincamajine	inactive at 70	138(79.3-238)

Table 1. Antiamoebic and Antimalarial Activity of Selected Alstonia Alkaloids<sup>25</sup>

ED<sub>50</sub>: Effective dose at 50% inhibition. See Table 2 for source of P.falciparum

The bisindole alkaloid accedinisine **4a** was first reported to be isolated from the root of epithelium *T. accedens* in 1976.<sup>33</sup> Later in 1994, accedinisine **4a** and *N*<sup>2</sup> demethylaccedinisine **4b** were isolated from *Peschiera van heurkii*. These two bisindole alkaloids exhibited potent antileishmanial and antibacterial activities.<sup>34</sup> Dimeric alkaloids are often more potent in vivo than their monomeric progenitors. Much of the early isolation and structural work on the *Alstonia* alkaloids was performed in the laboratories of Elderfield<sup>35</sup> and Schmid<sup>36,37</sup> and was followed by the biomimetic interconversions of LeQuesne et al.<sup>26,28,29,38-40</sup>

Alkaloids	K1 (μM, n = 3)	T9-96 (μM, n = 3)	K1/T9-96 ratio
O-acetyl macralstonine	$0.53\pm0.09$	$12.4 \pm 1.6$	0.04
villalstonine	$0.27\pm0.06$	$0.94\pm0.07$	0.29
alstomacroline	$1.12\pm0.35$	$10.2\pm0.4$	0.11
macrocarpamine	$0.36\pm0.06$	>39*	<0.009
chloroquine diphosphate	$0.20\pm0.07$	$0.019\pm0.002$	10.53

**Table 2.**  $IC_{50}$  Values of Alkaloids Tested Against *Plasmodia falciparum* (K1 and T9-96 strains)<sup>30</sup>

K1: the multi-drug resistant strain, which is highly resistant to chloroquine; T9-96: the chloroquine sensitive strain; n: number of independent experiments; \*: % inhibition at 39  $\mu$ M (7.5  $\mu$ g/mL), the highest concentration tested was 32.6 ± 4.1  $\mu$ M. The experiment was done with infected human red blood cells.

During the structure determination of *Alstonia* bisindole alkaloids by Schmid et al.<sup>36,37</sup> macroline **6** (Figure 3) was obtained as a degradation product from villalstonine **3a**. To date, macroline **6** has not been isolated as a natural product but it is believed to be a biomimetic precursor to many *Alstonia* alkaloids.<sup>26,28,29,38,39</sup>

Studies on individual sarpagine/ajmaline indole alkaloids have shown normacusine B exhibited sedative and ganglion blocking activity,<sup>41</sup> lochnerine elicited hypoglycemic activity,<sup>42</sup> pericyclivine has been shown to exhibit weak cytotoxic activity against P-388 leukemia,<sup>43</sup> while gardnutine, gardnerine, and hydroxygardnutine have been shown to elicit ganglion blocking effects as well as anticancer activity.<sup>44-46</sup> The mixture of constituents (normacusine B, affinisine,  $N_a$ -methylpericyclivine, and voachalotine) present in *Peschiera van Heurckii* from the tropical rain forest in Bolivia is known for its alleged leishmanicidal and bactericidal activity.<sup>34</sup> Like many quaternary alklaoids, the

major quaternary sarpagine alkaloid from *Strychnos angolensis*, 11-methylmacusine A, demonstrated muscle relaxant

ALKALOID	n	$IC_{50}(\mu g/mL)$
talcarpine	3	$40.3 \pm 2.9$
pleiocarpamine	2	$6.44\pm0.98$
alstoumerine	2	$13.1 \pm 2.7$
20-epi-antirhine	3	$7.51 \pm 1.17$
alstonerine	2	$9.67 \pm 3.57$
alstophylline	2	$12.7\pm1.7$
macralstonine	3	$8.92\pm2.95$
O-methylmacralstonine	2	$0.85\pm0.20$
O-acetylmacralstonine	3	$0.53\pm0.09$
alstomacrophylline	2	$1.10\pm0.30$
villalstonine	3	$\boldsymbol{0.27\pm0.06}$
villalstonine N <sub>b</sub> -oxide	2	$10.7\pm1.9$
alstomacroline	3	$1.12\pm0.35$
macrocarpamine	3	$0.36\pm0.06$
chloroquine diphosphate	3	$0.20\pm0.07$

**Table 3.** IC<sub>50</sub> Values of Alkaloids Tested Against Plasmodia falciparum (K1 strain)<sup>47</sup>

K1 strain: multidrug resistant strain; n: number of independent experiments. The test was done with infected human red blood cells.

activity.<sup>48</sup> The crude mixture of nine alkaloids present in *Ervatamia yunnanesis* is used in Chinese folk medicine for the treatment of hypertension.<sup>22</sup> Scientists have begun to

evaluate these alkaloids for activity against cancer<sup>49,50</sup> and HIV,<sup>51,52</sup> however, the paucity of isolable material from these plants has retarded in depth evaluation.

Figure 2. Dimeric Indole Alkaloids accedinisine and N'-demethylaccedinisine



Figure 3. Macroline, Sarpagine and Ajmaline



The number of macroline-related alkaloids which have been isolated has increased rapidly. At this time, the group contains more than 150 alkaloids, of which over a 100 are monomeric indoles. The remainder belongs to the bisindole class of alkaloids.<sup>40,53-55</sup> The

sarpagine group of alkaloids (represented by **7**, Figure 3) is the largest class of natural products related to the macroline bases and both series originate from common biogenetic intermediates. Illustrated in Figure 3 are the basic macroline and sarpagine skeletons. The biogenetic numbering of LeMen and Taylor<sup>56</sup> is used throughout this chapter. Note the four stereocenters at C-3, C-5, C-15 and C-16 in macroline. The  $\beta$ -hydrogen atom at C-15 of this group, as well as the chiral centers at C-3, C-5 and C-16 are the same for the sarpagine series. The two classes can be related in a synthetic sense by a Michael addition of the nitrogen atom of N-4 of macroline **6** to the  $\alpha$ ,  $\beta$ -unsaturated carbonyl system at C-21, or by direct 1, 2 addition of N-4 to the ketone at C-19, as illustrated in Figure 4.<sup>54,55</sup> Conversely, sarpagine can be converted into macroline **6** by a retro-Michael reaction or Hoffmann elimination of the *N*<sub>b</sub>-methyl intermediate, as demonstrated earlier by LeQuesne et al.<sup>38</sup> (see Figure 4).

Figure 4. Biosynthetic Relationship Between Macroline and Sarpagine



The macroline/sarpagine alkaloids bear important structural similarities to the ajmaline alkaloids, the latter of which are well known for their biological activity.<sup>57</sup> Ajmaline **8**, is a clinically important indole alkaloid with historical significance.<sup>58,59</sup> Both these bases are structurally related by the presence of the quinuclidine ring and the C-5–C-16 bond linkage. The absolute configurations of the stereogenic centers at C-3, C-5 and C-15 of both series are identical, except at C-16 which is antipodal to the sarpagine series. The biogenetic connection between the sarpagine and ajmaline alkaloids proposed earlier<sup>60-62</sup> was confirmed by Stöckigt et al.<sup>63-66</sup> by conversion of 16-epi-vellosimine **14** into deacetylvinorine **16** in the presence of the enzyme, acetyl-CoA dependent *vinorine synthase* (see Figure 5).

Figure 5. Biosynthetic Relationship Between Sarpagine and Ajmaline Type Alkaloids



Structure	Alkaloid	Refs.
R'H	R=H, R'=H, vellosimine	(40)
	R=Me, R'=H, N <sub>a</sub> -methylvellosimine	(40)
	R=H, R'=OMe, 10-methoxyvellosimine	(40)
	R=Me, R'=OMe, majvinine	(40)
	R=H, R'=H, normacusine B	(40)
	R=H, R'=Ac, O-acetylnormacusine B	(40)
N N	R=H, R'=COPh, O-benzoyl normacusine B	(40)
RHH	R=H, R'=Me, O-methylnormacusine B	(40)
/	R=Me, R'=H, affinisine	(40)
R <sub>1</sub> O H	$R=R_1=R_2=H$ , sarpagine	(40)
	R=H, R <sub>1</sub> =Me, R <sub>2</sub> =H, lochnerine	(40)
	R=R <sub>1</sub> =H, R <sub>2</sub> =Ac, O-acetylsarpagine	(40)
RHHH	R=Me, R <sub>1</sub> =R <sub>2</sub> =H, $N_a$ -methylsarpagine	(40)
	$R_1 = R_2 = H$ , pericyclivine	(40)
$R_2$ $H$ $T$	$R_1$ =CH <sub>3</sub> , $R_2$ =H, $N_a$ -methylpericyclivine	(40)
	R <sub>1</sub> =H, R <sub>2</sub> =OH, 10-hydroxy pericyclivine	(40)
	$R_1=H, R_2=OMe, 10$ -methoxy pericyclivine	(40)
1	$R_1$ =Me, $R_2$ =OH, 10-hydroxy <sup>-</sup> $N_a$ -methylpericyclivine	(40)
	P -Me P -OMe 10 mothery	(AO)

**Table 4.** Sarpagine Indole Alkaloids  $(1)^{40}$ 

**Table 4.** Sarpagine Indole Alkaloids (2)

 <sup>67-69</sup>

Structure	Alkaloid	Refs.
$R_2$ $H$	$R_1=R_3=H$ , $R_2=OH$ , 16-episarpagine $R_1=R_3=H$ , $R_2=OMe$ , lochvinerine $R_1=R_2=H$ , $R_3=OCH_3$ , gardnerine	(40) (40) (40)
	R'=OMe, R= H, 18-hydroxy lochnerine R'=H, R=Me, 18-hydroxy affinisine	(40) (40)
H <sub>3</sub> CO H H H H H H H H H H	18-hydroxygardnerine	(40)
H <sub>3</sub> CO H N Me H H	11-methoxy- <i>N<sub>a</sub></i> -methyldihydro- pericyclivine	(68)
	R=H, R'=CH=CHCOCH <sub>3</sub> , difforine R=CH <sub>3</sub> , R'=CO <sub>2</sub> CH <sub>3</sub> <i>N<sub>a</sub></i> -methyl-16-epipericyclivine	(40, 69) (40, 70)

Structure	Alkaloid	Refs.
$R_1$ $H_1$ $H_2$ $H_1$	R <sub>1</sub> =OH, R <sub>2</sub> =CH <sub>3</sub> verticillatine (Cl salt) R <sub>1</sub> =H, peraksine (vomifoline)	(40) (40)
H H Me H Br H H H H	macrosalhine bromide	(40)
H H H CHO	O-acetylpreperakine	(40)
R <sub>1</sub>	$R_1$ =H, $R_2$ =CH <sub>2</sub> OH, 19( <i>S</i> ),20( <i>R</i> )-dihydroperaksine	(40)
	$R_1$ =OH, $R_2$ =CH <sub>2</sub> OH, 10-hydroxy-19( <i>S</i> ),20( <i>R</i> )- dihydroperaksine	(40)
H H H T	$R_1$ =H, $R_2$ =CHO, 19( <i>S</i> ),20( <i>R</i> )-dihydroperaksine-17- al	(40)
	R=R"=H, R'=CH <sub>2</sub> OH, 16-epinormacusine B	(40)
	R=Me, R'=CH <sub>2</sub> OH, R"=H, 16-epiaffinisine	(40)
	R=Me, R'=CH <sub>2</sub> OAc, R"=H, O-acetyl-16-epiaffinisine	(40)

### **Table 4.** Sarpagine Indole Alkaloids (3)

Structure	Alkaloids	Refs.
Me H H Me	voacoline	(40)
H H H H Me	trinervine	(40)
H H H H H H Me	venecurine	(40)
	R=H, R'=Me, talpinine R=OMe, R'=H, 21-hydroxy cyclolochnerine	(40) (40)
H H H H OH	eburnaphylline	(40)

### **Table 4.** Sarpagine Indole Alkaloids (4)

Structure	Alkaloid	Refs.
	neo-sarpagine	(71)
R H H H H H H H H H H	R=H, N <sub>a</sub> -demethylaccedine (amerovolfine) R=OH, 3-hydroxysarpagine	(72) (73)
	ervincidine	(40, 138,139)
$R_2$ $H_1$ $H_1$ $H_2$ $H_1$	R <sub>1</sub> =CH <sub>3</sub> , R <sub>2</sub> =H, R <sub>3</sub> =CH <sub>3</sub> , dehydro-16-epiaffinisine R <sub>1</sub> =H, R <sub>2</sub> =OCH <sub>3</sub> , R <sub>3</sub> =CH <sub>3</sub> , gardnutine R <sub>1</sub> =H, R <sub>2</sub> =OCH <sub>3</sub> , R <sub>3</sub> =CH <sub>2</sub> OH, 18-hydroxygardnutine	(40) (40) (40, 74)
R R H/,, H H Me H H H H	R=H, dehydrovoachalotine R=OH, 17-hydroxy- dehydrovoachalotine	(40) (40)

**Table 4.** Sarpagine Indole Alkaloids  $(5)^{70-73}$ 

Structure	Alkaloid	Refs.
	alstoumerine	(40)
OH H H H H H H	koumidine	(40)
$ \begin{array}{c} H \\ H \\$	R <sub>1</sub> =H, R <sub>2</sub> =CHO, polyneuridine aldehyde R <sub>1</sub> =H, R <sub>2</sub> =CH <sub>2</sub> OH, polyneuridine R <sub>1</sub> =CH <sub>3</sub> , R <sub>2</sub> =CHO, voachalotinal R <sub>1</sub> =CH <sub>3</sub> , R <sub>2</sub> =CH <sub>2</sub> OH, voachalotine	<ul> <li>(40)</li> <li>(40)</li> <li>(40)</li> <li>(40)</li> </ul>
$R_2$ $H$ $T$ $R_1$ $H$ $H$ $H$ $H$ $H$ $H$ $H$	$R_1=R_2=H, R_3=CH_2OH,$ E-akuammidine $R_1=R_2=H, R_3=CH_2OAc,$ acetylakuammidine $R_1=CH_3, R_2=OCH_3, R_3=CH_2OH,$ 10-methoxy- $N_a$ -methylakuammidine	(40) (40) (40)

# Table 4. Sarpagine Indole Alkaloids (6)

Structure	Alkaloid	Refs.
H H H H	panarine	(40)
	R <sub>1</sub> =CH <sub>2</sub> OH, R <sub>2</sub> =H, macusine B	(40)
	R <sub>1</sub> =CH <sub>2</sub> OCH <sub>3</sub> , R <sub>2</sub> =H, O-methylmacusine B	(40)
$H^{R_2}$	R <sub>1</sub> =CO <sub>2</sub> CH <sub>3</sub> , R <sub>2</sub> =H, alkaloid Q3	(40)
	R <sub>1</sub> =H, R <sub>2</sub> =CH <sub>2</sub> OCH <sub>3</sub> , O-methyl-16-epimacusine B	(40)
	$R_1=H, R_2=CO_2^-, 16$ -epipanarine	(40)
	R <sub>1</sub> =CO <sub>2</sub> CH <sub>3</sub> , R <sub>2</sub> =CH <sub>2</sub> OH macusine A	(40)
	$R_1$ =CH <sub>2</sub> OH, $R_2$ =CO <sub>2</sub> CH <sub>3</sub> , $N_b$ -methylakuammidine	(40)
R'O	R=H, R'=H, spegatrine	(40)
H OH	R=H, R'=Me, lochneram	(40)
	R=Me, R'=H, <i>N<sub>a</sub></i> -methylsarpagine metho salt	(40)
$R_2O$	$R_1$ =CH <sub>3</sub> , $R_2$ =H, $N_a$ -methyl-11-hydroxymacusine A	(40)
	N <sub>1</sub> -11, N <sub>2</sub> -CH <sub>3</sub> , 11-methoxymacusine A	(40)

# **Table 4.** Sarpgine Indole Alkaloids (7)

Structure	Alkaloids	Refs.
$H_{3}CO N H H H$	R <sub>1</sub> =H, R <sub>2</sub> =CO <sub>2</sub> Et, fuchsiaefoline R <sub>1</sub> =CH <sub>2</sub> OH, R <sub>2</sub> =CO <sub>2</sub> Me, 12-methoxy- $N_b$ -methylvoachalotine R <sub>1</sub> =CH <sub>2</sub> OH, R <sub>2</sub> =CO <sub>2</sub> Et 12-methoxy- $N_b$ -methylvoachalotine ethyl ester	(40) (40) (40)
H H OMe N H H H H H H H H	19,20-dihydro-O-methylmacusine B	(40)
$ \begin{array}{c} H \\ H \\ T \\ T \\ H \\ H$	$R_1=H, R_2=CO_2CH_3, R_3=CH_2OH, \\19,20-dihydroakuammidine \\R_1=H, R_2=CH_2OH, R_3=CO_2CH_3, \\19,20-dihydropolyneuridine \\R_1=CH_3, R_2=CH_2OAc, R_3=CO_2CH_3, \\17-O-acetyl-19,20-dihydrovoachalotine \\R_1=CH_3, R_2=CH_2O-dihydrovoachalotine \\R_1=CH_3, R_2=CH_2O-dihydrovoachalotine \\R_1=CH_3, R_3=CO_2CH_3, \\R_1=CH_3, R_3=CH_2O-dihydrovoachalotine \\R_1=CH_3, R_3=CH_2O-dihydrovoachalotine \\R_1=CH_3, R_3=CH_3O-dihydrovoachalotine \\R_1=CH_3, R_3=CO_2CH_3, \\R_1=CH_3, R_3=CH_3O-dihydrovoachalotine \\R_1=CH_3, R_3=CO_2CH_3, \\R_1=CH_3, R_3=CH_3O-dihydrovoachalotine \\R_1=CH_3, R_3=CO_3CH_3, \\R_1=CH_3, R_3=CH_3O-dihydrovoachalotine \\R_1=CH_3, R_3=CH_3O-dihydrovoachalotine \\R_1=CH_3, R_3=CH_3O-dihydrovoachalotine \\R_1=CH_3, R_3=CO_3CH_3, \\R_1=CH_3O-dihydrovoachalotine \\R_2=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_2=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_2=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_2=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_2=CH_3O-dihydrovoachalotine \\R_1=CH_3O-dihydrovoachalotine \\R_2=CH_3O-dihydrovoachalotine \\R_3=CH_3O-dihydrovoachalotine \\R_3=CH_3$	(40) (40) (40)
$H = H OR_2$	$R_1$ =CH <sub>3</sub> , $R_2$ =H, affinisine $N_b$ -oxide $R_1$ =H, $R_2$ =CH <sub>3</sub> , O-methylnormacusine B $N_b$ -oxide	(40) (40)
H H (''CO <sub>2</sub> CH <sub>3</sub> N Me H H H	21-hydroxyvoachalotine	(40)

Structure	Alkaloid	Refs.
	R =R <sub>1</sub> =H, 16-epiaffinine R =Me, R <sub>1</sub> =H, hystrixnine R =H, R <sub>1</sub> =OMe, pelirine	(75) (76) (77)
	affinine	(78)
$ \begin{array}{c}                                     $	R=H, 19 <i>E</i> , gelsempervine-A R=Ac, 19 <i>E</i> , gelsempervine-B R=H, 19 <i>Z</i> , gelsempervine-C R=Ac, 19 <i>Z</i> , gelsempervine-D	(79) (79) (79) (79)
H HO H 19	19Z, 19Z-16-epivoacarpine 19ZE, 19 <i>E</i> -16-epivoacarpine	(79) (79)

**Table 4.** Sarpagine Indole Alkaloids (9)<sup>74-78</sup>
Structure	Alkaloid	Refs.
H O H H OH	erichsonine	(80)
H CO <sub>2</sub> CH <sub>3</sub> H O H	pagicerine	(81)
	R =H, amerovolficine R =OMe, 10-methoxy-16-de(methoxycarbonyl) pagicerine	(72) (82)
H H H H N H N H <sub>2</sub> CO <sub>2</sub> CH <sub>3</sub> '''H H N H	pagisulfine	(83)

# **Table 4.** Sarpagine Indole Alkaloids (10)<sup>71,79-82</sup>

Structure	Alkaloid	Refs.
R <sub>1</sub> N Me H H H H H H H H H H H H H H H H H H	$R_1 = R_2 = H_2$ , alstonerine $R_1 = OMe$ , $R_2 = H_2$ , alstophylline $R_1 = OMe$ , $R_2 = O$ , 6-oxoalstophylline	(84) (85) (85)
$R_{2}$ $R_{2}$ $R_{2}$ $R_{2}$ $R_{2}$ $R_{3}$ $H$ $H$ $R_{3}$ $H$	$R_1$ =OMe, $R_2$ =H, $R_3$ =H <sub>2</sub> , 19,20-dehydro,10-methoxytalcarpine $R_1$ =H, $R_2$ =OMe, $R_3$ =H <sub>2</sub> , alstophyllal	(27)
	$R_1$ =H, $R_2$ =OMe, $R_3$ =O, 6-oxo-alstoph $R_1$ =H, $R_2$ =H, $R_3$ =H <sub>2</sub> , alstonerinal	yllal (85) (85, 88)
$Me H H R_2$	R <sub>1</sub> =H, R <sub>2</sub> =CHO, talcarpine R <sub>1</sub> =H, R <sub>2</sub> =CH <sub>2</sub> OH, macrocarpine A R <sub>1</sub> =OH, R <sub>2</sub> =CH <sub>2</sub> OH, alstohentine	(85, 86) (85, 86) (85)
	R=CHO, secotalcarpine R=CH <sub>2</sub> OH, macrocarpine B R=CH <sub>2</sub> OAc, macrocarpine C	(86) (85, 86) (86)
H N Me H H H O H	alstomacroine	(87)

 Table 5. Macroline Related Indole Alkaloids (1)
 83-87

Structure	Alkaloid	Refs.
R H NH H Me O	R=H, alstonisine R=OCH <sub>3</sub> , <i>N<sub>b</sub></i> -demethylalstophylline oxindole	(89)
R H N Me O H H H H H H H H H H	R <sub>1</sub> =H, alstonal R <sub>1</sub> =OCH <sub>3</sub> , <i>N<sub>b</sub></i> -demethylalstophyllal oxindole	(89)
R H <sub>3</sub> C H <sub>1</sub> C O H <sub>3</sub> C H <sub>1</sub> C O H <sub>1</sub> C O H <sub>1</sub> C O H Me O	R=H, 16-hydroxyalstonisine R=OCH <sub>3</sub> , 16-hydroxy- <i>N<sub>b</sub></i> -demethyl alstophylline oxindole	(89) (89)
R H N Me O H H H H H H H H H H H H H H H H H H	$R_1$ =H, 16-hydroxyalstonal $R_1$ =OCH <sub>3</sub> , 16-hydroxy- $N_b$ -demethyl alstophyllal oxindole	(89) (89)
R H N Me O	macroxine	(89)

# Table 5. Macroline Related Indole Alkaloids (2)<sup>88</sup>

Structure	Alkaloids	Refs.
H H Me H H H H H H H C H <sub>2</sub> OH	macrodasine A	(86, 90)
H H Me H H H H H H H H H H H H H H H H H	macrodasine B	(86)
	alstomicine	(85)

# **Table 5.** Macroline Related Indole Alkaloids (3)

Table 6. Bisindole Alkaloids (1)



(40) R = $CH_2OH$ , undulatine



MeO

Ĥ

`N´ `` ⊢ H Me

ÇO<sub>2</sub>CH<sub>3</sub>

Ĥ.

macralstonidine

(40)

**Table 6.** Bisindole alkaloids  $(2)^{90,91}$ 



**Table 6.** Bisindole Alkaloids  $(3)^{92-94}$ 



 Table 6. Bisindole Alkaloids (4)



**Table 6.** Bisindole Alkaloids  $(5)^{95}$ 



## Table 6. Bisindole Alkaloids (6)

Structure	Alkaloid	Refs.
R		
Me-N	R =H, villalstonine	(26, 27)
H, Me	R =H, $N_b$ -O, villalstonine N <sub>b</sub> -oxide	(28)
	R =OMe, 10-methoxy villalstonine	(26, 27)
Me ,0	R =OMe, $N_b$ -O, 10-methoxy villalstonine $N_b$ -oxide	(26, 27)
MeO <sub>2</sub> C		



## Table 6. Bisindole Alkaloids (7)



# **Table 6.** Bisindole Alkaloids (8)<sup>96,97</sup>





R =H, 11-methoxy-10-(11'-vincorinyl)-vincamajine	(98)	
R =Ac, 11-methoxy-10-(11'-vincorinyl)-vincamedine	(98)	



# **Table 6.** Bisindole Alkaloids (9)<sup>98,99</sup>



Structure	Alkaloid	Refs.
Me-N H,,, Me N HH,, Me N HH,, Me OR MeO <sub>2</sub> C H H	R =H, villamine R =Ac, O-acetylvillamine	(36, 37) (36, 37)
H N Me H H H O R	R =H, macroline R =Ac, O-acetylmacroline	(36, 37) (36, 37)
H N Me H H	anhydromacrosalhine-methine	(101)
	$N_a$ -desmethylalstophylline	(101)

# **Table 7.** Degradation Products of Alstonia Alkaloids<sup>100</sup>

# 1.2. Previous approaches to the tetracyclic azabicyclo [3.3.1] nonane core framework5 of the sarpagine/macroline related indole alkaloids

As illustrated in Figure 3, as well as Tables 4, 5, and 7, a core tetracyclic system (rings A, B, C and D) appears to be a common structural feature of the macroline related sarpagine and ajmaline alkaloids. A general approach to the synthesis of these macroline-related alkaloids should involve enantiospecific synthesis of this core structure with the correct configurations at stereocenters C-3, and C-5 and the appropriate functional groups at C-15/C-16 for further transformations. The racemic azabicyclo [3.3.1] nonane has been prepared in the late 1970's by Yoneda,<sup>101</sup> Mashimo,<sup>102</sup> Kluge<sup>103</sup> and Hobson,<sup>104</sup> and was later improved by Soerens<sup>105</sup> on a kilogram scale. In 1988, Zhang et al.<sup>106</sup> achieved the synthesis of the optically active tetracyclic ketone 25 in a stereospecific fashion by employing the asymmetric Pictet-Spengler reaction. Many improvements have been made to prepare both the  $N_a$ -H and the  $N_a$ -CH<sub>3</sub> tetracyclic ketones. Other recent routes toward the formation of the tetracyclic core, which deserve mention, include the racemic synthesis by Rassat,<sup>107,108</sup> the aza Diels-Alder approach by Kuethe,<sup>109</sup> the olefin metathesis process of Martin<sup>110</sup> and the *cis*-Pictet-Spengler approach by Bailey<sup>111</sup> and Magnus.<sup>112</sup>

## 1.2.1. A racemic synthesis by Rassat et al. <sup>59, 60</sup>

In 2000, Rassat et al.<sup>107, 108</sup>designed the synthesis of 2,6-dihyroxy-9-azabicyclo[3.31]nonane **25** from the bis epoxide **18** (Scheme 1), which could be carried out on large scale. The bis epoxide **18** was easily converted into a 1:1 mixture of regioisomeric diols **19** and **20** via transannular cyclization with benzylamine in refluxing water. The undesired diol **20** was then converted into **19** by reacting it with trifluroacetic anhydride in refluxing dichloromethane, followed by hydrolysis with sodium hydroxide to furnish **19** in 95% yield (from **18**). This bicyclic nonane **19**, was then converted into **25**, as illustrated in Scheme 1.



Monoprotection of the diol **19** with *tert*-butyldimethylsilyl chloride, followed by replacement of the *N*-benzyl group with the *N*-benzoyl group gave the desired monol **22**. Swern oxidation of the alcohol **22**, followed by removed of the silyl group gave **23** in 95% yield (2 steps). Fischer indole cyclization of **23** with *N*-methyl, *N*-phenylhydrazine

in refluxing methanolic HCl led to the indole, which on reduction with  $LiAlH_4$  provided the  $N_b$ -benzyl derivative 24. The racemic tetracyclic ketone 25 could then be obtained by Swern oxidation of 24 in 73% yield.

**1.2.2.** Approach to the core tetracyclic structure via the aza-Diels Alder/intramolecular Heck cyclization by Kuethe et al.<sup>61</sup>



Scheme 2

Keuthe and co-workers adopted a [4+2] annulation strategy for construction of the tetracyclic cores **30a,b**. As illustrated in Scheme 2, indoles **26a,b** on treatment with n-BuLi or LDA in refluxing MTBE, followed by addition of iodine furnished the iodo compounds **27a,b** individually. Dess-Martin oxidation of the primary alcohol provided the aldehydes **28a,b**. The aza-Diels Alder reaction of **28a,b** with the Danishefsky diene provided **29a,b**, which was subjected to an intramolecular Heck cyclization to furnish the desired cyclized indole scaffolds in both series **30a,b**. Although this route requires stoichiometric amounts of palladium in the Heck coupling reaction, it does provide the tetracyclic framework in five steps, albeit the carbonyl group is not in an ideal position for regiospecific transformations.

# **1.2.3.** Approach to the core structure 36 via a ring-closing olefin metathesis by Martin et al.<sup>110</sup>

During their extensive study on the use of olefin metathesis reactions for the synthesis of various azabicyclo structures with the nitrogen in the one-atom bridge, Martin et al.<sup>110</sup> were able to construct the tetracyclic core (ABCD rings), necessary for the synthesis of the macroline/sarpagine alkaloids. As illustrated in Scheme 3, the dihydro- $\beta$ -carboline **31** (obtained from L-tryptophan) was N-protected with Cbz-Cl, followed by esterification to provide a diastereomeric mixture of aminals **32**. This diastereomeric mixture of aminals **32** was then treated with allyltrimethylsilane and boron-trifluoride etherate to afford the *cis*-tetrahydro- $\beta$ -carboline **33** as a mixture of two compounds in a 5.5:1 diastereomeric ratio which must be separated. Selective reduction of the ester, followed by a Wittig reaction yielded the alkyne **34** in a one-pot procedure. Treatment of the alkyne **34** with

### Scheme 3



Grubb's first generation catalyst promoted the enyne metathesis to furnish the tetracyclic diene **35** in essentially quantitative yield. Selective oxidation of the exocylic olefin provided the aldehyde **36** in 54% yield (two steps).

# **1.2.4.** Synthesis of the tetracyclic core 42 via the *cis*-selective Pictet-Spengler reaction by Bailev et al.<sup>111</sup>

Bailey employed the *cis*-selective Pictet-Spengler reaction to synthesize the tetracyclic intermediate **42**. As Illustrated in Scheme 4, the synthesis began with L-(-)-tryptophan **37'** which was converted into the homologated nitrile **38** via the reduction/

activation/replacement sequence in 50% overall yield. The homologated nitrile **38** was then treated with methyl propynoate in refluxing chloroform for 100 hours to afford the enamines **39a** and **39b**, after which the mixture was directly added to excess trifluoroacetic acid at 0 °C to provide a mixture of *cis* and *trans* diastereomers (**40a** and **40b**) in the ratio of 77:23. The diastereomeric mixture was then converted into the  $N_b$ -benzyl compound, which was further methylated to provide **41** in 39% yield (2 steps). The ring closure was mediated by the action of lithium diethylamide at -78 °C on ester **41** to afford the tetracyclic core in **42** in 90% yield. Although the tetracyclic ketone **42** can be synthesized by this method, this route suffers due to the number of steps involved in comparison to other routes and also from the poor diastereoselectivity obtained in the Pictet-Spengler reaction.



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### 1.2.5. Synthesis of the tetracyclic core 48 by Magnus et al.<sup>64</sup>

In 1989, Magnus et al.<sup>64</sup> reported the synthesis of the  $N_a$ -H tetracyclic ketone **49** from L-(-)-tryptophan **37'** which he employed for the total synthesis of (+)-koumine, (+)taberpsychine and (+)-koumidine. Illustrated in Scheme 5 is the *cis*-selective Pictet-Spengler reaction, which began with L-(-)-tryptophan**37'**.

Scheme 5



The L-(-)-tryptophan **37'** was converted into the  $N_a$ -benzyl, tryptophan methyl ester **43** by first benzylation with sodium amide/PhCH<sub>2</sub>Cl in 92% yield, followed by Fischer esterification of the alkylated L-(+)-tryptophan in 68% yield. This ester **43** was then

reacted with benzaldehyde at 22 °C and the imine which resulted was reduced with sodium borohydride in methanol to provide  $N_a$ -benzyl,  $N_b$ -benzyl tryptophan methyl ester **44** in 95% yield. This intermediate was then converted into a mixture of diastereomeric acids **45a/45b**, by employing the conditions of the Pictet-Spengler reaction as developed by Zhang et al.<sup>106</sup> Esterification of this mixture with MeOH/CISiMe<sub>3</sub> provided the esters **46a/46b** (2:1) in 80% yield. The major *cis*-diastereomer **46a** was isolated in 58% yield by crystallization from methanol and carried through to the tetracyclic ketone **48** via a Dieckmann cyclization and acid-mediated decarboxylation, as illustrated in Scheme 5 following the route developed in Milwaukee and reported in the racemic series by Yoneda et al.<sup>101</sup>

#### 1.3. Previous syntheses of sarpagine/macroline related indole alkaloids

### **1.3.1.** Synthesis of sarpagine/macroline related indole alkaloids in Milwaukee<sup>113</sup>

Vellosimine **14a** and normacusine B are the simplest representatives of the family of sarpagine indole alkaloids. Until 2000, no enantiospecific total synthesis of members of the sarpagine series was reported. The partial total synthesis of (-)-koumidine [which is similar in structure to **14a**, except it has a more stable *Z*-ethylidene double bond and the *S* configuration at C-16], was first reported by Sakai.<sup>114</sup> Magnus later synthesized the enantiomer of (-)-koumidine via a Pictet- Spengler reaction. However, establishment of the double bond (*Z*:*E*=1:5.7) was not stereospecific<sup>112</sup> and the reaction was hampered by poor diasteroselectivity in the Pictet-Spengler reaction. Entry into the pentacyclic sarpagine skeleton in Milwaukee for the total synthesis of ajmaline previously, was through a macroline framework which did not have the potential for incorporation of the

*E*-ethylidene double bond sterospecifically and also involved a large number of steps. Martin reported the total synthesis of geissoschizine with stereoselective establishment of the *E*-ethylidene double bond by an elimination process.<sup>115</sup> Furthermore Rawal<sup>116</sup> and Bosch<sup>117</sup> reported the total synthesis of Strychnos alkaloids with stereocontrolled establishment of the double bond by a Heck coupling reaction.

## **1.3.1.1.** The enantiospecific total synthesis of (+)-vellosimine by Wang<sup>10,118,119</sup>

In 2000 Wang reported an efficient, enantiospecific total synthesis of (+)-vellosimine **14a** with the first stereospecific establishment of the ethylidene double bond by a key palladium (enolate-mediated) carbon-carbon bond forming process. This key reaction provided a very efficient approach and allowed a direct entry to the basic pentacylic skeleton of the sarpagine alkaloid system, and permitted introduction of the C-19—C-20 olefinic bond with the *E* configuration, stereospecifically. The synthesis began with the tetracyclic ketone **49**, which was subjected to the conditions of catalytic hydrogenation to provide the  $N_a$ -H tetracyclic ketone **50** in 95% yield. Alkylation of the secondary amine in **50** with *Z*-1-bromo-2-iodo-2-butene **51**, a unit prepared by Ensley et al.<sup>120</sup> and employed by Rawal,<sup>116</sup> Bosch<sup>117</sup> and Kuehne<sup>121</sup> provided the  $N_b$ -*Z*-2'-iodo-2'-butenyl tetracyclic ketone **52** in 87% yield. This *Z*-olefin **51** can now be prepared on a 500 gram scale in a regiospecific fashion in three steps from crotonaldehyde.<sup>122</sup> Ketone **52** was then converted into the  $\alpha$ ,  $\beta$ -unsaturated aldehyde **53** in high yield (see Scheme 6).

Numerous attempts to effect the intramolecular Michael addition in  $\alpha$ , $\beta$ -unsaturated aldehyde **53** by generating a vinyl anion, followed by cyclization failed to give the desired product.<sup>10,119</sup> Radical-mediated cyclization<sup>121</sup> was then attempted on **53** by

Scheme 6



treating it with  $Bu_3SnH/AIBN$  in refluxing benzene. A mixture of (+)-vellosimine **14a**, accompanied by the undesired Z-olefinic isomer **54** in a 1:3 (*E/Z*) ratio was obtained in 40% overall yield. The olefinic bond had isomerized under the radical conditions to provide the thermodynamically more stable olefin as the major isomer.<sup>10</sup> Attempts to

carry out the Heck reaction under the conditions developed by Rawal,<sup>123</sup> as well as under modified conditions did not provide the desired *E*-olefin.

Muratake and Natsume<sup>124</sup> had reported a direct palladium-mediated arylation at the  $\gamma$ position of a simple  $\alpha$ ,  $\beta$ -unsaturated aldehyde under basic conditions in high yield. When the  $\alpha$ ,  $\beta$ -unsaturated aldehyde 53 was stirred under similar conditions, the C-14 insertion product (aldehyde 55) was isolated in 71% yield. With the formation of this product it was felt a similar mechanism could be envisaged to occur with iodo olefin 52. The relative acidity inherent at the  $\alpha$ -position of the ketone in 52 would facilitate the enolate mediated Pd(0)-cyclization into the desired sarpagine system in 14a. When iodo olefin 52 was subjected to the optimized conditions of the intramolecular palladium-mediated cross coupling reaction (3 mol% Pd(OAc)<sub>2</sub>, 30 mol% PPh<sub>3</sub>),<sup>10</sup> it furnished the cyclized product 56 stereospecifically in 80% yield. Wittig reaction of this ketone was followed by hydrolysis to provide the thermodynamically more stable  $\alpha$ -aldehyde in **14a**. Consequently, the first total synthesis of the  $N_a$ -H substituted sarpagine indole alkaloid (+)-vellosimine **14a** was accomplished from commercially available D-(+)-tryptophan methyl ester 107 (Scheme 16) in seven reaction vessels in 27% overall yield. With the success achieved in stereospecific construction of the sarpagine skeleton via the enolate mediated palladium cross coupling reaction, a similar strategy was employed for the synthesis of the related sarpagine alkaloids in the  $N_a$ -H as well as  $N_a$ -methyl series.<sup>113</sup>

# 1.3.1.2. Total synthesis of the enantiomer of macroline, the $N_a$ -H analog of macroline and trinervine via a regiospecific hydroboration process by Liu et al.<sup>125-127</sup>

From the earlier work of LeQuesne et al.,<sup>38</sup> the biogenetic origin of macroline **6** could be envisaged to arise from affinisine (enantiomer of **57'**, see Scheme 7) via a retro Michael

reaction (Figure 4). Consequently, the total synthesis of the enantiomer of macroline **6'** was designed based on this approach. Shortly after the synthesis of (+)-vellosimine **14a**, Liu<sup>125-127</sup> followed the same strategy (from L-tryptophan **37'**) to provide the enantiomer of affinisine **57'**.



The hydroxyl group of the enantiomer of affinisine **57'** was protected as the *t*butyldimethylsilyl ether to afford olefin **58'** (Scheme 7). Hydroboration of olefin **58'** with 1 equivalent of  $BH_3$ . THF at 0 °C resulted in the formation of the  $N_b$ -BH<sub>3</sub> complex **59'**. Further addition of 5 equivalents of  $BH_3$ . THF at 0 °C afforded a mixture of the secondary alcohols **60a',b'** and undesired tertiary alcohol **61'** in a ratio of 7:3.<sup>125</sup> The poor yield (49% of the desired secondary alcohols **60a',b'**), as well as the formation of the undesired

tertiary alcohol **61'** prompted a more detailed study of this process (see Scheme 8). After executing various reaction conditions,<sup>127</sup> which involved the use of hindered boranes and



a change of the reaction temperature as well as the use of a bulkier TIPS protecting group, the desired secondary alcohols could be prepared (the TIPS group was employed because it was found to be more stable and bulkier than the TBDMS). As illustrated in

Scheme 8, the TIPS group was used to protect the alcohol **57'** to provide the ether **62'** in 90% yield. Hydroboration of olefin **62'** under the modified conditions with 9 equivalents of BH<sub>3</sub>/DMS at room temperature provided the desired secondary alcohols **63a'** and **63b'** in 80% yield after oxidative workup.<sup>126,127</sup>



Swern oxidation of this mixture of secondary alcohols furnished the ketones **64a',b'** in 85% yield. This mixture of ketones was heated in THF in the presence of 1.5 equivalents of HCl (1N aqueous) for 3 hours, after which ketone **65'**, which was devoid of the boron species, was obtained in 90% yield. The methylation of **65'** with methyl iodide gave the

 $N_b$ -methiodide salt which was subjected to the modified conditions for ring cleavage (*t*-BuOK in refluxing THF/EtOH [6:1]) to provide the enantiomer **66'** of the macroline equivalent. Deprotection of the TIPS group in **66'** with TBAF in THF afforded the enantiomer **6'** of macroline in 86% yield (see Scheme 8).

Normacusine B 67 was converted into the  $N_a$ -H analog of macroline 71 (Scheme 9) following the same sequence of steps as outlined for 6' in Scheme 8. Trinervine 69, which retained the basic sarpagine skeleton, and contained a unique hemiketal ring formed between the C-17 (OH group) and the ketone at C-19 was synthesized from intermediate 68 on stirring it with 10 equivalents of 1 N aqueous HCl in refluxing THF (80% yield). The acidic conditions at reflux released the free amine, cleaved the TIPS group, as well as catalyzed the hemiketal formation. The total synthesis of trinervine 69 was, therefore, completed in enantiospecific fashion in an overall yield of 20% (from the ketone 50, see Scheme 6) in 10 reaction vessels (see Scheme 9).

## 1.3.1.3. Improved total synthesis of macroline by Liao et al.<sup>128</sup>

In an effort to decrease the amount of phosphine ligand employed for the cross coupling process, the Buchwald-Hartwig  $\alpha$ -arylation<sup>129</sup> conditions were studied. The original Buchwald conditions<sup>129</sup> were modified by introducing DPEphos, a bidentate ligand with a large bite angle, a stronger base such as NaO-*t*-Bu and Pd(dba)<sub>2</sub> as the catalyst. The improved route toward the synthesis of macroline is illustrated in Scheme 10.

Previous to this, the same sequence of steps was carried out as depicted in Scheme 8, albeit under modified reaction conditions. The  $N_a$ -methylvellosimine **74** was converted into affinisine **58** (Scheme 7) by NaBH<sub>4</sub>/EtOH reduction in 90% yield. Protection of the

OH group was carried out using TIPSOTf/2,6-lutidine in  $CH_2Cl_2$  to provide the ether **62**. Hydroboration followed by oxidation provided the secondary alcohols **75a** and **75b**. The



 $N_{b}$ -BH<sub>3</sub> complex was cleaved by refluxing with 2 equivalents of acetic acid in THF to provide the free amine **75a,b**. Because of the inconsistent results obtained in the Swern-oxidation of the secondary alcohols **75a,b**<sup>130</sup>, Dess Martin oxidation was employed to furnish the desired ketone **65** in 82% yield. Treatment with methyl iodide in THF

provided the quaternary salt, which underwent a retro-Michael reaction to provide the macroline equivalent **66** in 20.4% overall yield from tryptophan methyl ester (not shown in Scheme). Removal of the TIPS group with TBAF monohydrate, provided **6** in 86% yield.

### **1.3.2.** A biomimetic total synthesis of (+)- $N_a$ -methyl vellosimine 74 by Martin et al.



Scheme 11

van Tamelen<sup>62,131</sup> reported a biogenetic-type synthesis of ajmaline **8** in which the key step was the conversion of acid **76** into a mixture of epimeric aldehydes represented by **74** (Scheme 11). A proposition supported by a subsequent report of a biogenetic type synthesis of ajmaline was made later by van Tamelen and Olivier.<sup>132,133</sup> This seminal cyclization to furnish the sarpagine skeleton proceeded via the proposed iminium ion **77**, which was generated by decarbonylation of an activated form of the carboxyl group in

**76**. Aldehyde **74** was then converted into a relay compound which eventually was transformed into ajmaline **8**. Twenty-five years after van Tamelen's original work, Lounasmaa and coworkers reinvestigated this biogenetic-type cyclization.<sup>134</sup> They did not observe the expected "biogenetic-type" cyclization of a related iminium ion. Consequently, Lounasmaa et al.<sup>134</sup> reported they could not execute this type of cyclization and cast doubt on the biogenetic hypothesis of van Tamelen. It must be pointed out; however, Lounasmaa et al.<sup>134</sup> did not employ the exact substrate of van Tamelen. The group in Milwaukee as well as Martin and coworkers realized that Lounasmaa's work was not executed with the same substrate as van Tamelen. In order to solve this controversy, Martin et al.<sup>135</sup> developed a biogenetic entry into  $N_a$ -methyl vellosimine **74** which employed an iminium ion mediated cyclization similar to van Tamelen's original proposal.

As shown in Scheme 12, dihydrocarboline **31**, which was readily obtained from commercially available D-tryptophan **37**, was allowed to react with the vinyl ketene acetal to afford a single product and then directly converted into the *t*-butyl ester **78**. The  $N_b$ -acylation of amine **78** with diketene furnished an intermediate  $\beta$ -keto amide that underwent facile cyclization via an intramolecular Michael reaction upon addition of potassium *tert*-butoxide to give keto amide **79**.

The *E*-ethylidene side chain in amide **80** was obtained in good yield by following the reduction/activation/elimination sequence to give a single geometric isomer, as depicted in Scheme 12. The lactam **80** was then methylated at the  $N_a$ -indole position, after which the  $N_a$ -methyl amide was reduced to the amine **81** in 90% yield. This was followd by selective acid catalyzed ester hydrolysis in the presence of methyl thioanisole which

proceeded smoothly to give the acid **82**. The acid moiety in **82** was then transformed into the amide with NH<sub>4</sub>OH in the presence of EDCl in 86% yield followed by dehydration of the amide with trifluoroacetic anhydride to provide the nitrile **83** in 90% yield. The ester group in **83** was selectively reduced in the presence of the cyano group by reaction with lithium borohydride in THF and the subsequent oxidation of the hydroxyl group with Dess-Martin periodionane gave the aldehyde **84** in 83% yield. The aldehyde **84** was then converted into the silyl enol ether **85** with TBDMSCl in the presence of NaH. The  $\alpha$ aminonitrile **85** was then treated with BF<sub>3</sub>·OEt<sub>2</sub> in benzene and this was followed by exposure to aqueous potassium hydroxide in MeOH. The  $N_a$ -methylvellosimine **74** was obtained in 54% yield. This biomimetic total synthesis of  $N_a$ -methylvellosimine **74** provided experimental support for the cyclization step originally proposed by van Tamelen as one of the key steps in the biosynthesis of the sarpagine and ajmaline alkaloids. It also illustrated the Lounasmaa's criticism of the cyclization process<sup>134</sup> may be unfounded since he did not use the exact same substrate as van Tamelen.<sup>131</sup>

### Scheme 12



#### 2. Objectives and synthetic approach

The unique and complex architecture of the indole alkaloids coupled with their largely unexplored potential in medicine or as tools for biological studies make these compounds attractive targets for total synthesis. The sarpagine alkaloids contain the common structural element of the parent pentacyclic sarpagan ring system.<sup>57</sup> These indole alkaloids contain a double bond at C19-C20 and lack functionality at the C-21 position. In all known sarpagine indole alkaloids the stereochemistry at C-3 (S), C-5 (R) and C -15 (R) has been established. These three stereochemical centers have a fixed configuration because of the rigid framework of the molecule. The double bond at C19-C20 has an E-configuration. The alkaloid ervincidine belongs to the sarpagine class and has been isolated from the epigeal part of *Vinca erecta* Rgl. et Schmalh. <sup>136</sup> The UV spectrum is characteristic for indole bases. The IR spectrum has a broad band at 3000-3330 cm<sup>-1</sup> due to the presence of OH and NH groups connected by a hydrogen bond. This alkaloid belongs to the akuammidine subgroup and is a hydroxyl derivative of tombosine. <sup>136</sup>

#### Figure 6. Sarpagine Type Indole Alkaloids



According to Glasby's book of alkaloids  $^{137}$  and the review by Lounasmaa,  $^{57}$  the structure of ervincidine has four stereocenters C-3 (S), C-5 (R) , C -15 (R) and C -16(R). The stereochemistry of the alcohol at the C-6 position was not assigned. The paper published

by Yunusov etal.,<sup>136</sup> has reported the structure of ervincidine similar to that of the polyneuridine subclass with four stereocenters C-3(S), C-5(R), C-15(R) and C-16(S). But again the stereochemistry of the C-6 alcohol was not assigned. As illustrated in Figure 7, the proposed structure of ervincidine by Glasby<sup>5</sup> and Lousnamaa<sup>57</sup> has the functional group at C-16 (R) with the alpha stereochemistry and that of Yunusov<sup>136</sup> has the functional group at C-16 (S) with the less stable beta stereochemistry which stimulated the interest in this indole alkaloid. The synthesis of this natural product was also challenging because the only information from the literature<sup>136</sup> was the optical rotation which was  $+29.5^{\circ}(c \ 0.6, \text{ MeOH})$ . <sup>138</sup> No unequivocal spectral information (NMR, CHN analysis) was given. Another aspect which made the synthesis of this molecule important was the establishment of the stereochemistry of the C-6 alcohol function. Examination of both the reports from the literature<sup>40,136</sup> did not assign the stereochemistry to the C-6 alcohol. The mass spectrum and the molecular ion peak of ervincidine <sup>136</sup> was similar to the spectra of the sarpagine and tombosine alkaloids. <sup>139</sup> The difference of 16 m/e units indicated the presence of a hydroxyl group in ervincidine. The presence of an m/e ion peak at 169 in the mass spectrum suggested the presence of the hydroxyl group in the aliphatic part of the molecule possibly in ring C. On the basis of the comparison of the mass spectrum of ervincidine with that of tombosine, akuammidine and gardenine and with the proposed biogenentic interrelationship with gardnutine and hydroxygardnutine,  $^{140}$  the position of the hydroxyl group in ervincidine was probably at C-6. $^{136}$
# Figure 7. The Indole Alkaloid Ervincidine



<sup>a</sup> Ervincidine as proposed by Glasby <sup>137</sup> and Lousnamaa <sup>40</sup>



<sup>b</sup> Ervincidine as proposed by Yunusov *etal*.<sup>136</sup>

The objective of this research was to complete the first enantiospecific , stereospecific total synthesis of the indole alkaloid ervincidine and correct the stereochemical ambiguity reported in the literature. This ambiguity as pointed out was because the stereochemistry of the CH<sub>2</sub>OH group at C-16 was not assigned unequivocally and neither was the hydroxyl group at C-6 position. Moreover, the design of a synthetic entry into the C-6 alcohol stereospecifically, as well as an assessment (thermodynamic stability) of the stability of the C-6 hydroxyl group was of significant importance in indole alkaloid chemistry. As shown in Scheme 13, in a retrosynthetic sense, both the diastereomers **88** and **89** might be available via a common intermediate, pentacyclic ketone **56**. The chirality of **56** was to be employed to introduce the correct stereocenters in the target alkaloids **88** or **89**.

The ketone **56** could presumably be converted into vellosomine **14a**, which could be employed as an intermediate in the total synthesis of **88**. The synthesis of the axial alcohol E-16-epinormacusine B **112** could be achieved by a hydroboration/oxidation process. This intermediate once obtained, might further provide 6-hydroxy-16 demethoxycarbonyl polyneuridine **89** *via* protection/deprotection of hydroxyl group,

selective oxidation by IBX at C-6 and a reduction. The chirality of **56** was to be employed to introduce the correct stereocenters in the target alkaloids **88** or **89**.



Scheme 13. Retrosynthetic Analysis of ervincidine 88 and 89

The important initial stereogenic centers would be established from the 1, 3 *trans* transfer of chirality in the asymmetric Pictet-Spengler reaction.<sup>141-146</sup> The retrosynthetic strategy rests on the general approach which has been used to synthesize the macroline-sarpagine type alkaloids in Milwaukee. The thermodynamically stable  $\alpha$ -aldehyde **14a** would arise via Wittig reaction on the pentacylic ketone **56**. The *N*<sub>a</sub>-H, *N*<sub>b</sub>-H tetracyclic ketone **50** should be available from D-(+)-tryptophan **37** on a large scale via the asymmetric Pictet-Spengler reaction and the Dieckmann reaction as key steps.<sup>147-149</sup> This synthetic strategy was an efficient way to provide **56** in gram quantities. The less stable  $\beta$ -axial alcohol could be synthesized using the pentacyclic ketone **56** via a Wittig reaction coupled with a hydroboration/oxidation following the published procedure.<sup>122,150,151</sup>

# **3. Results and Discussion**

**3.1.** The First enantiospecific, stereospecific total synthesis of the sarpagine alkaloids related to structures of the proposed ervincidine (88 and 89) including ervindicine

#### **3.1.1.** Synthesis of the (-)- $N_a$ -H, $N_b$ -H tetracyclic ketone 50 on large scale

As planned, the synthesis of isomers of ervincidine **88** and **89** began with the preparation of the (–)- $N_a$ -H,  $N_b$ -benzyl tetracyclic ketone **49** as the key building block on large scale following the procedure developed in Milwaukee. As illustrated in Scheme 14, Fischer esterification of (+)-**37** with methanolic hydrogen chloride gave the methyl ester **90** on 400 gram scale. This was then converted into  $N_b$ -benzyltryptophan methyl ester **91** on stirring with benzaldehyde (1.1 equivalents) at room temperature (6 hours), followed by reduction of the imine which resulted *in situ* with sodium borohydride [-10 °C to -8 °C (2-5 hours, according to the scale)]. The optical purity of the  $N_b$ -benzyltryptophan methyl ester **91** prepared in this manner was found to be greater than 98% ee as determined by comparison to data from authentic samples.<sup>149</sup>

To introduce the desired stereocenter at C-1 the asymmetric Pictet-Spengler reaction<sup>141-146</sup> was employed with methyl 4,4-dimethoxy-butyrate **95**,<sup>146</sup> (which can be prepared on a kilogram scale) and  $N_b$ -benzyl tryptophan **91** to provide the *trans* diester **92b** as a single diastereomer in 92% yield. This process was scaled up to the 400 gram level and purification was achieved by recrystallization from ethanol. As shown in Scheme 15, the lone pair of electrons of the  $N_b$ -nitrogen atom of **91** (Scheme 15) presumably attacked the oxonium ion and this was followed by loss of the elements of methanol to form the iminium ion intermediate **96**. Attack of the iminium ion at the indole C-3 position could

form either the anti spiroindolenine **98b** or the syn spiroindolenine **98a** intermediate. The anti intermediate **98b** could rearrange directly into the desired *trans* diester **91b**; however, in the case of the disfavored syn spiroindolenine **98a**,<sup>152</sup> it was felt the rearrangement to the *cis* diester **92a** would occur followed by the acid-catalyzed isomerization at C-1 to afford the more stable *trans* diester **92b**.<sup>141</sup> Consequently, during the process of this Pictet-Spengler reaction, the *cis* diester **92a** derived from **97a** was indeed detected by TLC but was converted entirely into the *trans* diester **92b**, presumably via the carbocationic intermediate<sup>143</sup> **III** as planned. After work up, the *trans* diester **92b** was obtained as the only diastereomer in greater than 98% ee and high yield on 100 gram scale.

When the *trans* diester **92b** was treated with three equivalents of sodium hydride in the presence of excess methanol in refluxing toluene,<sup>106</sup> the *trans* diester **92b** was converted entirely into *trans*  $\delta$ -lactam **100**. No further change was observed (Scheme 16) on prolonged heating. However, when diester **92b** was treated with nine equivalents of sodium hydride in the presence of excess methanol in refluxing toluene under these same reaction conditions for an extended period of time (72 hours), the desired Dieckmann  $\beta$ -ketoester **102** was obtained in 88% yield. Presumably, the  $\delta$ -lactam **100** formed and then underwent ring-opening<sup>10,101</sup> to provide the *trans* diester **92b** under these conditions. This diester **101a** was in equilibrium with the *cis* intermediate **101b** and the latter material was eventually converted into  $\beta$ -ketoester **102** via the Dieckmann process.







It was felt the  $\delta$ -lactam **100** was favored in the equilibrium over intermediates **101a** and **101b**; however, the irreversible formation of the enolate (see **102**) in the Dieckmann cyclization gradually promoted completion of the process to provide the  $\beta$ -ketoester **102**, exclusively (Scheme 16). Hydrolysis of the  $\beta$ -ketoester **102** and decarboxylation under acidic conditions was achieved in one step to provide the template (-)- $N_a$ -H,  $N_b$ -benzyl tetracyclic ketone **49** in 90% yield. The enantiomeric purity of ketone **49** was determined to be greater than 98% ee by conversion of it into (-)- $N_a$ -methyl,  $N_b$ -benzyl tetracyclic ketone **25** whose enantiomeric purity had been determined previously by Zhang.<sup>10,106,153</sup>

D-(+)-Tryptophan methyl ester 90 can also be diastereospecifically converted (via 92b) into azabicyclononone 49 in greater than 98% ee in a two pot process on multihundred gram scale, as depicted in Scheme 17. More specifically, after  $N_{\rm b}$ -benzylation of 90 with benzaldehyde and sodium borohydride (1 eq) in methanol was completed, trifluoroacetic acid (TFA) was added to the reaction vessel at 0 °C to neutralize the alkaline mixture. After removal of the solvent under reduced pressure, CH<sub>2</sub>Cl<sub>2</sub>, TFA and acetal 95 were added to the vessel at 0 °C and the modified Pictet-Spengler reaction (600 g scale) was carried out to provide the *trans* diester **92b** in 83% crystalline yield. In the second sequence (150 g scale), the Dieckmann cyclization was completed and the reaction solution was then cooled to 0 °C after which it was carefully quenched with glacial acetic acid. After removal of the solvent under reduced pressure, concentrated glacial acetic acid, aqueous hydrochloric acid and water were added to the residue in the same vessel at 0 °C. The acidic hydrolysis and decarboxylation of **102** was then executed at reflux to provide (-)-49 in 80% crystalline yield in greater than 98% ee.<sup>10</sup> The benzyl group in tetracyclic ketone 49 was removed by converting 49 first into its hydrochloride salt. The HCl salt of 49 was then subjected to the conditions of catalytic debenzylation [Pd/C, H<sub>2</sub> (1 atm), bench top] to furnish the  $N_a$ -H,  $N_b$ -H tetracyclic ketone 50 in 95% yield (Scheme 17).



In conclusion, the basic tetracyclic core of the  $N_a$ -H,  $N_b$ -H tetracyclic ketone **50**, which contained rings ABCD required for the synthesis of the C-19 methyl substituted sarpagine indole alkaloids was constructed enantiospecifically and stereospecifically via the asymmetric Pictet-Spengler/Dieckmann protocol on large scale.<sup>119,147,148</sup>

# 3.1.2. Synthesis of the -N<sub>a</sub>-H pentacyclic ketone 56

 $N_{b}$ -alkylation of the secondary amine **50** with (*Z*)-1-bromo-2-iodo-2-butene **51** under the conditions developed in Milwaukee (dry THF/K<sub>2</sub>CO<sub>3</sub>/ reflux)<sup>122</sup> <sup>118,154</sup> led to the formation of the desired product **52**, but was very sluggish, and led to considerable formation of baseline impurities. Since substitution by amines on bromides was clearly an S<sub>N</sub>2 type process, it was then decided to increase the nucleophilicity of the amine by employing a more polar aprotic solvent. It was well known that the nucleophilic strength was dependent on the solvent employed in the S<sub>N</sub>2 reaction. Dry acetonitrile on the other

hand proved to be the most suitable solvent for this process for the reaction went to completion at room temperature to give **52** in 76% yield. In order to increase the yield, vinyl iodide **51** was used in excess (3-4 equivalents). But this reaction condition did not improve the yield but increased the base line impurities.

Initially, the vinyl iodide **51** required for this route was synthesized according to a literature procedure.<sup>120</sup> However, the free-radical hydrostannation of the propargyl alcohol was difficult and the desired iodide was extremely hard to purify. Importantly, as illustrated in Scheme 18, an improved method was recently developed<sup>122</sup> to prepare **51** *via*  $\alpha$ -iodination of an  $\alpha$ , $\beta$ -unsaturated aldehyde **103** under the conditions of Kraft<sup>82,155</sup> which exclusively formed only the Z-isomer **104**. This material without further purification underwent the subsequent reduction<sup>83</sup> to provide the desired alcohol **105** in 97% overall yield. This alcohol **105** was treated with PBr<sub>3</sub> in dry ether at 0 °C for 12 hours to provide the desired bromide **51** in 93% yield . The sequence was scaled up to the 200 g level scale starting from crotonaldehyde **103** to obtain the desired allylic bromide **51** on a large scale as well.

The ketone **52** was initially converted into the desired ketone **56** in 80% yield *via* an intramolecular palladium (enolate mediated) cross-coupling reaction analogous to the process developed by Wang *et al.*<sup>118,156,157</sup> Further development for this palladium-mediated cross coupling reaction was achieved when ketone **52** was subjected in combination with 5.0 mol%  $Pd_2(dba)_3$ , 7.0 mol% DPEphos as a ligand and 1.5 equivalent of NaO*t*Bu as a base in THF at 80 °C for 5 hours. The desired ketone **56** was obtained in 83% yield which was superior to the previously reported results (**Scheme 5**).



A plausible Hartwig-Buchwald catalytic cycle for the enolate-driven palladium catalyzed  $\alpha$ -vinylation reaction is illustrated in Scheme 19. It was felt the active Pd(0)L<sub>2</sub> species was generated primarily by reduction of Pd(OAc)<sub>2</sub> with triphenylphosphine in this phosphine-assisted catalytic cycle.<sup>158,159</sup> The active Pd(0)L<sub>2</sub> species could then undergo oxidative addition on the vinyl iodide to form a vinylpalladium(II) iodide complex. Ligation to the oxygen and rearrangement to the carbon atom at C(15) can then occur. Finally, reductive elimination of the carbon bound Pd(II) intermediate would furnish the desired E-olefin **56** while regenerating the active Pd(0)L<sub>2</sub> species. A key to the success of this Pd-mediated cross coupling reaction rested on the enolate concentration. This must be high enough to permit ligand substitution to occur.<sup>125</sup> It was well known that vinyl iodides would be quite reactive in the oxidative-addition step. If nucleophilic substitution

of the iodide by the nucleophilic enolate in the coupling process was comparably slow, the rate of byproduct formation would be significant. Eventually, a highly reproducible yield could be obtained by employing an excess of the phosphine ligand.<sup>160</sup> In the case of different substrates, sometimes it was necessary to carry out the reaction with a 10:1 ratio of PPh<sub>3</sub> (30 mol%) to Pd(OAc)<sub>2</sub> (3 mol%). The process then took three days to go to completion in comparison to that of five hours required under the original conditions, but the amount of byproducts (Scheme 19) was decreased. The yield of the sequence in this series was 81%. Furthermore, the best solvent system to date was  $DMF/H_2O$  (9/1), as compared to CH<sub>3</sub>CN, pure DMF, or DMF/H<sub>2</sub>O (4/1) which resulted in no reaction or low vields.<sup>127</sup> Because the conditions employed for the enolate driven palladium coupling process were different from the recent Buchwald-Hartwig arylation reactions,<sup>160</sup> efforts were directed toward intramolecular  $\alpha$ -vinylation under the conditions of the arylation process.<sup>160</sup> This might permit lower catalyst loading<sup>160</sup> and be effective for both E and Z ethylidenes. The homogenous reaction conditions might permit a faster rate and thereby facilitate scaleup. In fact, Liao first reported a Pd(0) catalyzed  $\alpha$ -vinylation process in which the cheaper DPEphos was employed in the case of a Z-ethylidene<sup>130</sup> for the synthesis of (-)-koumidine.<sup>150</sup> Moreover, utilization of this ligand for the synthesis of the E-ethylidene 56 was also executed by Liao et al. in 80% yield.<sup>150</sup> However, a related approach has also been effective based on a palladium-catalyzed coupling of aminotethered vinyl halides with ketones reported by Solé, Bonjoch et al.<sup>161,162</sup> In addition, another process was also carried out with ketone 52 in the presence of 6.0 mol%  $Pd(PPh_3)_4$  and a base PhOK (which was generated previously in the reaction vessel from 2.0 equivalent of PhOH and 1.5 equivalent of KOtBu). The most notable aspect of the



Different trace byproducts **52**, **50** or **106** are ocassionally observed depending on what catalyst was employed and how much base was used.

latter attempt was that it did not require an additional ligand. The intramolecular cyclization took place to afford ketone **56** in 80% yield again in stereospecific fashion.

The use of the weaker potassium phenolate base limited the presence of any acetylene byproduct formed from loss of HI from the halide **52**.

Ketone 56 was then subjected to the Wittig reaction with methoxymethyl triphenylphosphonium chloride and anhydrous potassium tert-butoxide to provide a mixture of two stereoisomeric enol ethers (not shown). After a short wash column, the mixture of enol ethers was hydrolyzed under acidic conditions to provide vellosimine 14a in 90% yield (overall yield for two steps). Since the aldehyde at C(16) was in the more stable alpha configuration because of a *syn* pentane interaction with the indole methylene bridge, the mixture was simply stirred until all of the  $\beta$ -epimer was converted into the more stable, natural epimer present in (+)-vellosimine **14a**. The aldehyde function of **14a** was then reduced with sodium borohydride to provide the alcohol normacusine B 106, the spectral data of which are in complete agreement with the natural product. The C (17)functionalized alcohol was then protected as the trisopropylsilyl ether employing 2,6 lutidene as the base to provide the ether 107. Repeated attempts to oxidize the C-6 position by DDQ on 108 were unsuccessful. The conditions employed with the first equivalent of DDQ deprotected the TIPS functional group forming alcohol 107 and the second equivalent oxidized the alcohol 107 to aldehyde 14a. Finally, the oxidation at the C (6) position was successfully accomplished by oxidation using IBX at 80°C (radical process as shown in Scheme 21) to form the desired ketone 108.<sup>130,163</sup>



The silyl analogue was then stirred in THF with tetrabutyl-ammonium fluoride monohydrate to provide the C-16 substituted monol **110**. Illustrated in Figure 8 is the ORTEP view of the crystal structure of ketone **110**.







Figure 8. ORTEP View of the Crystal Structure of the monol 110

**3.1.3.** Completion of the regiospecific, stereospecific total synthesis of ervincidine proposed by Glasby and Lousnamaa

The selective reduction of the ketone **110** was done by a Luche reduction<sup>164</sup> using sodium borohydride in combination with cerium chloride heptahydrate to afford the final product **88** with C-3 (S), C-5 (R), C-6 (R), C -15 (R) and C -16(R) stereochemistry as a single diastereomer. Luche reduction is a selective reduction of 1,2 enones using NaBH<sub>4</sub> in combination with CeCl<sub>3</sub>. For the methanolysis of NaBH<sub>4</sub>, CeCl<sub>3</sub> which is a lewis acid catalyst is selected. The harder reducing agent (see Scheme 22) favor 1, 2 reduction thus leading to the formation of a single diastereomer regiospecifically. Illustrated in **Figure 9** is the ORTEP view of the crystal structure of **88.** However, the optical rotation of this diastereomer **88** was  $[a]^{26}_{D} +79^{\circ}$  (c 0.6, in methanol) which is not in agreement with the value reported by Yunusov in the literature.<sup>40,137</sup> The diastereomer **88** was stirred in 0.2N HCl at 0°C. Examination by TLC (silica gel; CH<sub>2</sub>Cl<sub>2</sub>: MeOH 9:1; Rf: 0.73) showed a new spot at lower Rf value (0.73) which indicated the complete epimerization of **111** at the C-6 position. Unfortunately the new spot could not be isolated due to shortage of the material at this point. It is felt the C-6 alpha hydroxyl group is presumably the thermodynamically more stable isomer based on this preliminary experiment.



Figure 9. ORTEP View of the Crystal Structure of the Diol 88 which shows the alpha



hydrogen at C-6 with C-5

Since the optical rotation of diastereomer **88** obtained was not in agreement with the literature which was  $+29.5^{40,136,137}$  the focus of this research turned towards the synthesis of the N<sub>a</sub>-H axial alcohol at the C-16 position proposed by Yunusov<sup>136</sup> in order to construct the diastereomer with the polyneuridine-related stereochemistry in an enantiospecific fashion.

# **3.1.4.** Completion of the regiospecific, stereospecfic total synthesis of ervincidine proposed by Yunusov<sup>136</sup>

The synthesis of E-16-epinormacusine B **112** began with the pentacyclic ketone **56**. The E-16-epinormacusine B **112** was obtained from **56** via a Wittig reaction which was carried out with triphenylphosphonium bromide in benzene, in the presence of potassium *t*-butoxide to afford the diene **111** in 90% yield. When the two double bonds in the diene

**111** are compared it was felt the less hindered double bond at C(16)-C(17), relative to the ethylidene at C(19)-C(20) would be more easily oxidized with a hindered hydroborating reagent. This was in agreement with the earlier work of Magnus and Cook et al. Treatment of diene **111** with the 9-BBN reagent was chosen as the hydroborating agent to facilitate attack from the less hindered face of the C(16)-C(17) double bond relative to the C(19)-C(20) site. This also hinders the formation of the N<sub>b</sub>-9-BBN complex which might hinder attack from the alpha face of the molecule. If it became important to obtain the alcohol with the alpha stereochemistry at C-6, simple epimerization (H<sup>+</sup>) should be effective based on the previous results shown here for the epimerization of **111**.

Scheme 23



The synthesis of ketone **114** from alcohol **112** was carried out analogous to the previous preparation of the TIPS derivative **108** from alcohol **107** in 90% yield. The oxidation of the C-6 position was achieved by radical oxidation using IBX<sup>165,166</sup> in 85% yield.



As shown in Scheme 25, the silvl group from ketone **114** was removed by treatment with wet TBAF in THF to give **115** in 90% yield. The selective reduction of the ketone **115** was done by a Luche reduction to achieve the stereospecific synthesis of the natural product ervincidine **89** as a single diastereomer in 90% yield. The optical rotation of **89**  $[\alpha]^{26}{}_{\rm p}$  +29.00° (c 0.6, in methanol) was in agreement with that of the literature<sup>136</sup>  $[\alpha]^{26}{}_{\rm p}$  +29.5° (c 0.6, in methanol) which completed the first enantiospecific, stereospecific total synthesis of the natural product ervincidine **89** (Scheme 25). The analysis of the spectral data of the alkaloid **89** follows here.







Figure 10: Proton spectrum of natural product Ervincidine 89



Figure 11: Carbon spectrum of natural product ervincidine 89

To assign the stereochemistry of the C-6 alcohol a detailed study of the 1D and 2D NOESY experiments were carried out on the synthetic diol **89.** The methyl group (3H-18) detected as a three proton doublet showed strong connectivity only to H-19. The proton of the aromatic ring at H-9 and H-6, showed a weak NOE. No NOEs were found between H-15 and H-6. The protons of H-15 exhibited a strong NOE with H-17, H-16, H-14 and H-18. The proton of H-6 showed strong NOE with H-5. A weak NOE was observed between the protons labeled H-6 and H-3.

Figure 12. Selected NOEs of ervincidine 89





Figure 13. 1D NOEs Observed after Irradiation of Protons at H-3

Analysis of the 2D NOESY experiments (Figure 12) further confirmed, the result obtained by the 1D NOESY experiments. Good NOEs were found between H-3 and H-5 (strong) and H-14 (strong). The irradiation of the proton at C-3 resulted in the enhancement of the signals at H-5 and H-14, indicating that the protons at H-3 and H-5 (see **Figure 13**) were located in the pseudo axial position. These results were supported by examination of the 2D NOE spectrum for H-6 which exhibited cross peaks at H-5 (strong).

Figure 14. 1D NOEs Observed after Irradiation of Protons at H-15



The 1D NOE signals were observed between H-15 and H-18 (strong), H-17 (strong), H-16 (strong) and H-14 (strong). The strong NOE between H-15 and with one of the H-17 protons and not an NOE with the other proton at H-17 indicated that one proton at C-17 was located in the  $\alpha$  position while the other proton located at C-17 was found to be  $\beta$ . The NOE signals between H-15 and H-18 (strong), H-19 (weak) and H-5 (weak) were in agreement with the assignment.

Figure 15. 1D NOEs Observed after Irradiation of Protons at H-17



The 2D NOE signals observed between the two proton at C-17 were very strong. The signals between H-17 and H-15 (strong), H-17 and H16 (strong) and H-17 and H-14 (strong) were also strong.

Figure 16. 1D NOEs Observed after Irradiation of Protons at H-6



The 1D NOE signal observed between H-6 and H-5 was very strong and the NOE between H-6 and H-3 was weak. No NOE is seen between H-6 and H-15. Weak NOE signals are found between H-6 and H-9 as well as H-6 and H-18. The NOE signal between H-5 and H-16 was strong with a coupling constant of 9.8 Hz, consequently the dihedral angle between the protons at H-5 and H-16 was close to 0° (eclipsed conformation). A weak NOE was observed between H-3 and H-6. The coupling constant between H-5 and H-6 was 5.1 Hz which indicated that both the protons were cis to each other. The angle between both the H-5 and H-6 protons was about 60° (staggered conformation) from models. Both the alcohol at C (17) and the proton at the C-6 position are trans to each other. If the OH were axial and the proton at H-6 were equatorial, a strong NOE would have been observed with H-15 which in this case was not seen. In summary, the 2D COSY, NOESY and 1D NOE experiments support the structure of ervincidine **89**, as indicated in Scheme 25.



Figure 17. 1D NOE data H-15 irradiated











# **3.1.5.** Completion of the regiospecific, stereospecfic total synthesis of the diasteromer (118) of Ervincidine

To synthesize the diastereomer with the opposite stereochemistry to that of ervincidine **89** at the C-6 position, diene **111** was treated with IBX to give the ketone **117** at the C-6 position in 80% yield. In order to shorten the synthetic route and also synthesize the other isomer, 9-BBN was chosen which reduces the C-6 keto group to the alcohol and also acts as a hydroborating agent. <sup>167,168</sup> Since 9-BBN is a bulky group the attack might take place from the top face of the molecule. The boron might coordinate with the C-6 carbonyl oxygen as well as the N<sub>b</sub> nitrogen atom thus leading to the formation of the alpha diastereomer **118**. This process was executed as shown in scheme 26 in 70% yield The optical rotation of **118**  $[\alpha]^{26}_{p}$ +17.00° (c 0.6, in methanol) was not in agreement with the literature. <sup>40,137</sup> It is entirely possible that 9-BBN complexed<sup>127,169</sup> with the N<sub>b</sub>-nitrogen function and blocked the attack from the bottom face of the ketone so that the carbonyl reduction occurred from the top face to give **118**.



#### 4. Conclusion

In conclusion, the first enantiospecific, stereospecific total synthesis of ervincidine 89 has been accomplished from commercially available D(+)-tryptophan 37 which has served both as the chiral auxiliary and the starting material. Moreover, this is the first synthesis which unequivocally sets the stereochemistry of the hydroxyl group at C-6 in sterospecific fashion, as well as the C-16 hydroxy methyl group. The stereospecific conversion of D-(+)-tryptophan 37 into the key template (–)- $N_a$ -H,  $N_b$ -benzyl tetracyclic ketone **49** via the asymmetric Pictet-Spengler reaction (600 gram scale) and Dieckmann cyclization on multi-hundred gram scale was reduced to only two reaction vessels. The optically active tetracyclic ketone **49** was converted into the core pentacyclic framework 56 using the intramolecular palladium-mediated enolate cross coupling reaction which was developed here in Milwaukee<sup>130</sup> to afford the core pentacylic framework **56**. This robust reaction could be scaled up to multigram scale in this series. Important to success here were the sequence of chemical reactions which included a Wittig reaction, a regiospecific hydroboration and protection/deprotection steps in order to provide regiospecific oxidation at C-6. The IBX mediated oxidation and the Luche reduction using CeCl<sub>3</sub>. 7H<sub>2</sub>0 in the presence of NaBH<sub>4</sub> afforded the first enantiospecific, stereospecific total synthesis of ervincidine 89. The indole alkaloid ervincidine 89 could be prepared from D-(+)-tryptophan 37 in 13 reaction vessels in 19.2% overall yield. Another important experiment was the epimerization of the C-6 alcohol with 0.2N HCl which indicated the care that must be employed in isolation of these alkaloids which contain a benzylic hydroxyl group. This research process developed here also provides a
general entry into C-6 hydroxy substituted indole alkaloids either alpha or beta stereochemistry.

Two other diasteromers were made to rule them out as potential structures. This research corrects the errors in Glasby's book<sup>137</sup> and Lousnamma's review <sup>40</sup> and clarifies the work of Yunusov et al as well as providing the correct absolute configuration of the C-6 hydroxyl function in ervincidine **89**.<sup>136</sup>

### **5. Experimental Details**

#### **General Experimental Considerations:**

All reactions were carried out under an argon atmosphere with dry solvents using anhydrous conditions unless otherwise stated. Tetrahydrofuran (THF) and diethyl ether were freshly distilled from Na/benzophenone ketyl prior to use. Dichloromethane was distilled from calcium hydride prior to use. Methanol was distilled over magnesium sulfate prior to use. Benzene and toluene were distilled over Na prior to use. Acetonitrile was distilled over CaH<sub>2</sub> prior to use. Reagents were purchased of the highest commercial quality and used without further purification unless otherwise stated. Thin layer chromatography (TLC) was performed using Dynamic Adsorbents Inc. UV active silica gel, 200  $\mu$ m, plastic backed plates or; Dynamic Adsorbents Inc. UV active alumina N, 200  $\mu$ m, F-254 plastic backed plates. Flash and gravity chromatography were performed using silica gel P60A, 40-63  $\mu$ m purchased from Silicycle. Basic alumina (Act I, 50-200  $\mu$ m) for chromatography was purchased from Dynamic Adsorbents. Neutral alumina (Brockman I, ~150 mesh) for chromatography was purchased from Sigma-Aldrich. TLC plates were visualized by exposure to short wavelength UV light (254 nm). Indoles were visualized with a saturated solution of ceric ammonium sulfate in 50% sulfuric acid.<sup>170</sup> Elemental analyses were performed on a Carlo Erba model EA-1110 carbon, hydrogen, and nitrogen analyzer. All samples submitted for CHN analyses were first dried under high vacuum for a minimum of six hours using a drying pistol with isopropyl alcohol or benzene as the solvent with potassium hydroxide pellets in the drying bulb. Melting points were taken on a Stuart melting point apparatus SMP3 manufactured by Barloworld Scientific US Ltd. Proton (<sup>1</sup>H NMR) and carbon high resolution nuclear magnetic resonance spectra (<sup>13</sup>C NMR) were obtained on a Bruker 300-MHz or a GE 500-MHz NMR spectrometer. <sup>1</sup>H NMR data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublets, dt = doublet of triplets, ddd = doublet of doublet of doublet, td = triplet of doublets, qd = quartet of doublets, sex = sextet, m = multiplet), integration, and coupling constants (Hz),  ${}^{13}C$  NMR data are reported in parts per million (ppm) on the  $\delta$  scale. The low resolution mass spectra (LRMS) were obtained as electron impact (EI, 70eV), which were recorded on a Hewlett-Packard 5985B gas chromatography-mass spectrometer, while high resolution mass spectra (HRMS) were recorded on a VG Autospec (Manchester, England) mass spectrometer. HRMS recorded by electrospray ionization (ESI) methods were performed at the Laboratory for Biological Mass Spectrometery at Texas A&M University on a API QStar Pulsar model, manufactured by MDS Sciex. Optical rotations were measured on a JASCO Model DIP-370 digital polarimeter. Infra-red spectra were recorded on a Thermo Nicolet Nexus 870 FT-IR or a Perkin Elmer 1600 series FT-IR spectrometer.



#### **D-(-)-Tryptophan methyl ester 90**

To a round bottom flask (5000 mL) which contained a freshly saturated solution of anhydrous methanolic hydrogen chloride (2500 mL), D-(+)-tryptophan 37 (400 g, 1.96 mol) was added. The mixture which resulted was heated to reflux for 8 h and then allowed to cool to rt. The crystalline product which formed upon cooling was collected by filtration and washed with cold ether to provide D-(-)-tryptophan methyl ester•HCl. The free base 90 was prepared by treatment of the corresponding HCl salt with cold aq NH<sub>4</sub>OH (10%), followed by extraction with CH<sub>2</sub>Cl<sub>2</sub> in methanol (4:1, 3 x 1000 mL). The solvent was evaporated under reduced pressure, after which the solid was redissolved in CH<sub>2</sub>Cl<sub>2</sub> and dried (K<sub>2</sub>CO<sub>3</sub>). The DCM layer was removed under reduced pressure to provide D-(-)-tryptophan methyl ester 90 (392 g, 91%). mp 92-93 °C (free base);  $[\alpha]_D^{27}$ = -37.2 (c 1.0 in MeOH); **IR** (KBr) 1730 cm<sup>-1</sup>; <sup>1</sup>**H** NMR (250 MHz, CDCl<sub>3</sub>, free base)  $\delta$ 1.62 (2H, s), 3.05 (1H, dd, J = 14.5, 7.5 Hz), 3.29 (1H, dd, J = 14.6, 5.8 Hz), 3.74 (3H, s), 3.86 (1H, dd, J = 7.5, 4.8 Hz), 7.04-7.23 (3H, m), 7.35 (1H, d, J = 7.5 Hz), 7.61 (1H, d, J = 7.5 Hz), 8.24 (1 H, s); <sup>13</sup>C NMR (62.8 MHz, CDCl<sub>3</sub>, free base)  $\delta$  30.83, 51.86, 55.06, 111.19, 118.73, 119.49, 122.11, 122.87, 127.57, 136.37, 175.71; EIMS (m/e, relative

intensity) 218 (M<sup>+</sup>, 78), 159 (57), 130 (100), 117 (33). **Anal. Calcd. for C<sub>12</sub>H<sub>14</sub>O<sub>2</sub>N<sub>2</sub>:** C, 66.06; H, 6.42, N, 12.84. Found: C, 66.14; H, 6.33; N, 12.77.



#### **D**-(+)- $N_{\rm b}$ -Benzyltryptophan methyl ester 91

To a round bottom flask (3000 mL) which contained a solution of D-(-)-tryptophan methyl ester 90 (408 g, 2.0 mol) in dry CH<sub>3</sub>OH (1500 mL), benzaldehyde (260.0 g, 2.3 mol) was added. The solution which resulted was stirred with an overhead stir for 6 h at 22 °C, until examination by TLC (silica gel) indicated the disappearance of 90. The mixture was then cooled in an ice-salt bath to -5 °C [It was much easier to maintain the inside temperature between  $-10^{\circ}$ C to  $-5^{\circ}$ C by the employment of a dry ice bath (without solvent)]. If the temperature falls below -15°C, a large amount of solid will appear and it is too difficult to stir the reaction mixture. Sodium borohydride (42 g, 1.1 mol) was then added portionwise at -5 °C over a period of 2.5 h [The slow addition and lower temperature of this process are critical to avoid racemization of the chiral center]. The solution was allowed to stir for 3 h and then followed by the addition of ice water (50 mL). The solvent was removed under reduced pressure. The residue was dissolved in CHCl<sub>3</sub> (2000 mL) and washed with brine (2 x 500 mL). The organic layer was dried  $(K_2CO_3)$  and the solvent was removed under reduced pressure to give the free base 3 (525 g, 90%) which could be further purified by crystallization from EtOH. **mp** 109-110 °C;  $[\alpha]_{D}^{22} = +9.07^{\circ} (c \ 1.0 \text{ in CH}_{3}\text{OH}), [\text{lit.}^{152} [\alpha]_{D}^{22} = +8.65^{\circ} (c \ 1.0 \text{ in CH}_{3}\text{OH})]; IR (film)$ 1745 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  2.10 (1H, s), 3,15 (2H, m), 3.65 (3H, s), 3.75 (3 H, s), 3.80 (3H, m), 6.90 (1H, s), 7.10 (1H, t, *J* = 8.2 Hz), 7.25 (5H, m), 7.55 (1H, d, *J* = 8.2 Hz); <sup>13</sup>C NMR (62.8 MHz, CDCl<sub>3</sub>)  $\delta$  29.35, 51.68, 52.20, 61.38, 111.18, 111.38, 118.84, 119.45, 122.07, 122.88, 127.03, 127.62, 128.19, 128.32, 136.30, 139.79, 175.36; EIMS (*m/e*, relative intensity) 308 (M<sup>+</sup>, 28), 249 (22), 178 (57), 130 (100).

**Anal. Calcd. for** C<sub>19</sub>H<sub>20</sub>O<sub>2</sub>N<sub>2</sub>: C, 74.03; H, 6.49; N, 9.08. Found: C, 73.99; H, 6.56; N, 8.99.



## Preparation of methyl 4,4-dimethoxybutyrate 95<sup>171</sup>

To a round bottom flask (12 L) equipped with an overhead stir which contained a solution of methyl acrylate **93** (1075 g, 12.5 mol) and nitromethane (762.5 g, 12.5 mol) in  $CH_2Cl_2$  (2500 mL) at 0 °C, was added a solution of NaOH (60 g) in water (2500 mL) at 0 °C. The mixture which resulted was stirred at rt for 48 h. The organic layer was separated and washed with water (2 x 500 mL) and dried (MgSO<sub>4</sub>). The solvent was then removed under reduced pressure and the residue was distilled under reduced pressure (95-100 °C/8 mm of Hg) to give the pure methyl  $\gamma$ -nitro butyrate **94** (540 g, 30%). This was dissolved in methanolic sodium methoxide [5400 mL, 0.8 N, freshly prepared from 173 g of NaH (60% dispersion in mineral oil)] and the solution which resulted was added dropwise with

an overhead sitr at a rate of 1 drop per second to a round bottom flask (12 L) which contained a solution of conc sulfuric acid (1350 mL) in methanol (4050 mL) at -45 to -50 °C. After the addition was completed, the reaction mixture was poured into  $CH_2Cl_2$  (8 L). The organic layer was separated and washed with ice water (2 x 1000 mL), aq 4 N NaOH (2 x 600 mL) and dried ( $K_2CO_3$ ). The solvent was removed under reduced pressure and the residue was distilled (110-115 °C/8 mm of Hg) to provide methyl 4,4-dimethoxybutyrate **95** (540 g, 90%). <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  1.92 (2H, q, *J* = 7.5 Hz), 2.37 (2H, t, *J* = 7.5 Hz), 3.31 (6H, s), 3.66 (3H, s), 4.38 (1H, t, *J* = 15.8 Hz); <sup>13</sup>C NMR (62.8 MHz, CDCl<sub>3</sub>)  $\delta$  27.92, 29.07, 51.45, 53.14, 103.14; CIMS (*m/e*, relative intensity) 375 (M<sup>+</sup> + 1, 100). The spectral data for **95** were identical to the published values. <sup>171</sup>



Diastereospecific preparation of the *trans*-(1S,3R)-(-)-2-benzyl-3-methoxycarbonyl-1-methoxycarbonylethyl-1,2,3,4-tetrahydro-9*H*-pyrido-[3,4-b]indole (92b) via the newly modified Pictet-Spengler reaction.

To a round bottom flask (5000 mL) which contained a solution of optically active  $N_{\rm b}$ benzyl-D-tryptophan methyl ester **90** (400 g, 1.3 mol) in CH<sub>2</sub>Cl<sub>2</sub> (2500 mL) was added methyl 4,4-dimethoxybutyrate acetal **95** (315 g, 1.9 mol) and TFA (180 g, 2.4 eq) at 0 °C

with stirring. The reaction mixture which resulted was allowed to stir at rt for 10 d and then cooled in an ice bath and brought to pH = 8 with a cold aq solution of NaOH (3 N). The aq layer was separated and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 1500 mL). The combined organic layers were washed with brine (2 x 600 mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The volume of the solution was reduced to 300 mL under reduced pressure and EtOAc (100 mL) and hexanes (300 mL) were added to the above solution. The solution was cooled to -20 °C. The trans diester 109b (402 g, 76%) precipitated out as white crystals and the mother liquor was concentrated. The residue which resulted was purified by flash chromatography (silica gel, EtOAc/hexane, 1/4) to provide additional **109b** (83 g, 16%). The combined yield of **92b** (485 g) was 92%. **mp** 152-153 °C;  $[\alpha]_D^{27} = -35.7^\circ$  (*c* 1.4, in CHCl<sub>3</sub>), [lit.<sup>114</sup>[ $\alpha$ ]<sub>D</sub><sup>27</sup>=-38.0° (*c* 1.0 in CHCl<sub>3</sub>)]; **IR** (film) 3310, 1731, 1707 cm<sup>-1</sup>; <sup>1</sup>**H NMR** (250 MHz, CDCl<sub>3</sub>)  $\delta$  1.85-2.15 (2H, m), 2.20-2.50 (2H, m), 3.03 (1H, dd, J = 15.8, 5.3 Hz), 3.12 (1H, dd, J = 15.8, 8.8 Hz), 3.48 (3H, s), 3.59 (1H, d, J = 13.6 Hz), 3.75 (3H, s), 3.84 (1H, d, J = 13.6 Hz), 3.87-3.93 (1H, m), 3.98 (1H, dd, J = 8.8, 5.3 Hz), 7.07-7.35 (8H, m), 7.43 (1H, d, J = 7.2 Hz), 7.98 (1H, s); <sup>13</sup>C NMR (62.8 MHz, CDCl<sub>3</sub>)  $\delta$  21.39, 28.94, 29.88, 51.45, 51.85, 53.51, 54.81, 56.79, 107.51, 110.99, 118.16, 119.51, 121.77, 127.08, 127.13, 128.28, 129.13, 134.26, 136.50, 139.41, 173.42, 174.18; EIMS (m/e, relative intensity) 406 (M<sup>+</sup>, 60), 347 (45), 319 (100), 169 (50).

**Anal. Calcd. for** C<sub>24</sub>H<sub>26</sub>O<sub>4</sub>N<sub>2</sub>**:** C, 70.91; H, 6.45; N, 6.89. Found: C, 70.88; H, 6.47; N, 6.91.



Dieckmann cyclization of the N<sub>a</sub>-H diester 92b to provide (6*S*,10*S*)-(-)-methyl-9-oxo-12-benzyl-6,7,8,9,10,11-hexahydro-6,10-imino-5*H*-cyclooct[*b*]indole-8-carboxy-late 102

To a round bottom flask (5000 mL) which contained a solution of  $N_a$ -H, trans diester **92b** (100 g, 0.25 mol) in toluene (3000 mL), which had been predried by azeotropic removal of H<sub>2</sub>O by a DST (refluxed 6 h), was added sodium hydride (88.7 g, 2.2 mol, 60% dispersion in mineral oil) at 0 °C. Anhydrous CH<sub>3</sub>OH (174 mL, 4.3 mol) was then added into the above mixture dropwise under Ar at 0 °C. The solution which resulted was held at reflux for 72 h (the flask was covered with aluminum foil on the top to keep the temperature at reflux without carbonizing any compound on the sides of the flask). The reaction was quenched with ice. The aq layer was extracted with  $CH_2Cl_2$  (3 x 1 L). The combined organic extracts were washed with brine and dried (K<sub>2</sub>CO). The solvent was removed under reduced pressure and the mineral oil was separated by decantation. The residue which resulted was purified by flash chromatography (silica gel, EtOAc/hexane, 1/4) to provide the  $N_a$ -H,  $\beta$ -ketoester **102** (82 g, 88%). mp 149-150 °C:  $[\alpha]_{D}^{27} = -177.4^{\circ}$  (c, 1.0 in CHCl<sub>3</sub>); **IR** (KBr) 1670, 1630cm<sup>-1</sup>; <sup>1</sup>H **NMR** (250M Hz, CDCl<sub>2</sub>)  $\delta$  2.30 (1H, d, J = 15.6 Hz), 2.82 (1H, dd, J = 15.5, 5.6 Hz), 2.90 (1H, d, J = 15.3 Hz), 3.18 (1H, dd, J = 15.9, 5.9 Hz), 3.66 (3H, s), 3.71 (1H, d, J = 13.4 Hz), 3.77 (1H, d,

J = 5.3 Hz), 3.82 (1H, d, J = 13.4 Hz), 3.98 (1H, d, J = 5.4 Hz), 7.11 (1H, t, J = 7.1 Hz), 7.16 (1H, t, J = 7.0 Hz), 7.24-7.39 (6H, m), 7.50 (1H, d, J = 7.1 Hz), 7.63 (1H, s), 11.98 (1H, s); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  22.2, 28.6, 49.7, 51.3, 55.2, 55.9, 94.3, 106.2, 110.8, 118.1, 119.5, 121.6, 127.0, 127.2, 128.4, 128.7, 133.4, 135.7, 138.3, 171.6, 172.5; CIMS (CH<sub>4</sub>) (*m/e*, relative intensity) 389 (M<sup>+</sup> + 1, 100).

**Anal. Calcd. for** C<sub>23</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>**:** C, 73.78; H, 5.92; N, 7.48. Found: C, 74.19; H, 6.23; N, 7.35.



Preparation of (6S,10S)-(-)-9-oxo-12-benzyl-6,7,8,9,10,11-hexahydro-6,10-imino-5*H*cyclo-oct[*b*]indole 49 via acid mediated hydrolysis of the  $N_a$ -H,  $\beta$ -ketoester 102

To a round bottom flask (500 mL) which contained the  $N_a$ -H,  $\beta$ -ketoester **102** (45.0 g, 0.12 mol) was added glacial acetic acid (167 mL), aq hydrochloric acid (245 mL, conc.) and water (65 mL) with stirring (magnetic stir). The solution which resulted was heated at reflux for 8 h. After removal of the solvent under reduced pressure, the residue was brought to pH = 9 with a cold aq solution of NaOH (3 N). The mixture which resulted was extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 x 250 mL) and the combined organic extracts were washed with a saturated aq solution of NH<sub>4</sub>Cl (100 mL), brine (2 x 100 mL) and dried (K<sub>2</sub>CO<sub>3</sub>). Removal of the solvent under reduced pressure afforded an oil. After a short wash column on silica gel, the  $N_a$ -H,  $N_b$ -benzyltetracyclic ketone **49** (24.5 g, 64%) was

crystallized from EtOAc/ hexane (1:4, 30 mL). The mother liquor was concentrated under reduced pressure and the residue was chromatographed on silica gel with EtOAc/hexane (1:4) to provide additional ketone **49** (10.5 g, 27%). The combined yield of **49** (35.0 g) was 91%.  $[\alpha]_D^{27} = -246.7^\circ$  (*c* 1.05 in CHCl<sub>3</sub>); **IR** (KBr) 2933, 1715 cm<sup>-1</sup>; <sup>1</sup>**H NMR** (250 MHz, CDCl<sub>3</sub>)  $\delta$  2.02(1H, m), 2.15 (1H, m), 2.49 (2H, m), 2.71 (1H, d, *J* = 16.9 Hz), 3.27 (1H, dd, *J* = 16.9, 6.8 Hz), 3.78 (2H, s), 3.80 (1H, s), 4.02 (1H, s), 7.17 (1H, dt, *J* = 7.3, 1.0 Hz), 7.23 (1H, dt, *J* = 7.6, 1.0 Hz), 7.30-7.38 (6H, m), 7.54 (1H, d, *J* = 7.6 Hz), 7.81 (1 H, s); <sup>13</sup>C NMR (62.8 MHz, CDCl<sub>3</sub>)  $\delta$  20.40, 30.37, 34.51, 49.40, 56.13, 65.22, 106.73, 110.94, 118.21, 119.72, 122.01, 126.86, 127.42, 128.43, 128.60, 131.98, 135.85, 138.25, 210.40; **CIMS** (*m/e*, relative intensity) 317 (M<sup>+</sup> + 1).

**Anal. Calcd. for C<sub>21</sub>H<sub>20</sub>N<sub>2</sub>O:** C, 79.72; H, 6.37; N, 8.85. Found: C, 79.51; H, 6.37; N, 8.85.



To a round bottom flask (12 L) which contained a solution of D-(-)-tryptophan methyl ester 90 (400 g, 1.83 mol) in CH<sub>3</sub>OH (5000 mL) was added benzaldehyde (230 g, 2.2 mol) with stirring (overhead stir). The solution which resulted was stirred for 6 h at rt. The mixture was then cooled to -10 °C and sodium borohydride (55 g, 1.3 mol) was added portionwise over a period of 3 h. The internal pot temperature was kept between -10 °C and -5 °C with a dry ice bath (solid dry ice). After analysis by TLC (silica gel, EtOAc/hexane, 1/1) indicated the disappearance of imine, the mixture was allowed to stir for an additional 0.5 h followed by the slow addition of CF<sub>3</sub>COOH (260 mL) at 0 °C. The solvent was removed under reduced pressure and the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (3000 mL). After the solution which resulted was cooled to 0 °C, methyl 4,4dimethoxybutyrate acetal 95 (324 g, 2.0 mol) and TFA (500 g, 2.4 eq) were added to this vessel at 0 °C. The reaction mixture was stirred at rt for 10 d, after which it was cooled in an ice bath and brought to pH = 8 with a cold aq solution of NaOH (3 N). The aq layer was separated and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 5000 mL). The combined organic layers were washed with brine (2 x 200 mL) and dried ( $K_2CO_3$ ). The volume of the solution was reduced to 300 mL under reduced pressure and EtOAc (100 mL) and hexanes (300 mL) were added to the above solution. The solution was cooled to -20 °C. The *trans* diester **92b** (510 g, 69%) precipitated as white crystals and the mother liquor was concentrated and the residue which resulted was purified by flash chromatography (silica gel, EtOAc/hexane, 1/4) to provide additional 92b (114 g, 14%). The combined yield of 92b

(624 g) was 83% and the solid was identical to the *trans* diester **92b** obtained from  $N_{\rm b}$ -benzyl-D-tryptophan methyl ester **90** in an earlier experiment.

One pot process for converting *trans*-(1*S*,3*R*)-(-)-2-benzyl-3-methoxy carbonyl-1methoxycarbonylethyl-1,2,3,4-tetrahydro-9*H*-pyrido[3,4-*b*]indole 92b into (6*S*,10*S*) -

(-)-9-oxo-12-benzyl-6,7,8,9,10,11-hexahydro-6,10-imino-5*H*-cyclooct[*b*] indole 49<sup>10</sup>

The *trans* diester **92b** (150 g, 246 mmol) was dissolved in dry toluene (3500 mL) in a round bottom flask (5000 mL) equipped with an overhead stirrer, Dean Stark trap (DST) and a reflux condenser. The solution which resulted was heated to reflux for 6 h with stirring. The above predried solution was cooled to 0 °C and sodium hydride (150 g of 60% dispersion NaH in mineral oil) was added to the above vessel under an atmosphere of Ar. Dry CH<sub>3</sub>OH (170 mL) was added carefully (a large amount of  $H_2$  was evolved at this point). The mixture was stirred at rt for 0.5 h, and then heated to reflux for an additional 72 h. Note: The top of the flask was covered with aluminium foil to keep the reflux and prevent carbonization of the lactam intermediate during the **Dieckmann process.** After analysis by TLC (silica gel, EtOAc/hexane, 1/4) indicated the disappearance of 92b, the reaction mixture was cooled to 0 °C and then quenched carefully with glacial acetic acid (300 mL). The solvent was removed under reduced pressure and glacial acetic acid (480 mL), aq hydrochloric acid (750 mL, conc.) and water (200 mL) were added to the above vessel. The mixture which resulted was heated at reflux for 8 h. After removal of the solvent under reduced pressure, the residue was brought to pH = 9 by addition of an aq solution of cold NaOH (3 N). The mixture which resulted was extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 x 1000 mL) and the combined organic extracts were washed with brine (2 x 500 mL) and dried (K<sub>2</sub>CO<sub>3</sub>). Removal of the solvent under

reduced pressure afforded an oil which was chromatographed on silica gel with EtOAc/hexane (3:7) to provide the tetracyclic ketone **50** (90 g, 80%), which was identical to the ketone obtained from ketoester **102** (see above). Note: A lactam intermediate is formed first in the Dieckmann reaction which must open up to isomerize and undergo the Dieckmann reaction. Do not mistake this first product (lactam) for the  $\beta$ -keto ester intermediate by TLC.



# Catalytic debenzylation of 49 to provide (6*S*,10*S*)-(-)-9-oxo-12-*H*-6,7,8,9,10,11hexahydro-6,10-imino-5*H*-cyclooct[*b*]indole 50 over Pd/C/H<sub>2</sub>

Tetracyclic ketone **49** (40 g, 127 mmol) was dissolved in anhydrous ethanol (1300 mL). A saturated solution of EtOH/HCl (g) was then added dropwise into the above mixture until the solid completely dissolved. The solvent was then removed under reduced pressure to furnish an HCl salt. The residue was then dissolved in dry ethanol (1000 mL) and the solvent was removed under reduced pressure to remove the excess HCl. This process was repeated 3 times to make sure there was no excess HCl. The HCl salt was degassed under reduced pressure at rt and back filled with argon (2 times). Dry Pd/C (10% by wt, 8.1 g, 60 mmol) was added to the above HCl salt, followed by addition (slow addition, initially) of dry ethanol (1300 mL). The mixture was degassed under reduced pressure at rt and back filled with H<sub>2</sub> (3 times). The mixture which resulted was allowed to stir on bench top at rt under an atmosphere of H<sub>2</sub> (balloon) for 10 h. Analysis

by TLC (silica gel plate was exposed to NH<sub>3</sub> vapours) indicated the absence of starting ketone **49**. The catalyst was removed by filtration (Celite) and washed with ethanol (3 x 100 mL). The solvent was removed under reduced pressure. The residue was dissolved in a mixture of CHCl<sub>3</sub> and brought to pH = 8 with 10% aq NH<sub>4</sub>OH. The aq layer was extracted with CHCl<sub>3</sub> (3 x 400 mL). The combined organic layers were washed with brine (2 x 200 mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure to afford the crude amine which was chromatographed (flash) on silica gel (CHCl<sub>3</sub>/EtOH, 9/1) to provide pure  $N_a$ -H,  $N_b$ -H tetracyclic ketone **50** (27.2 g, 95%). **FTIR** (NaCl) 3393, 3382, 1705 cm<sup>-1</sup>, <sup>1</sup>**H NMR** (300 MHz, CDCl<sub>3</sub>)  $\delta$  2.08-2.15 (2H, m), 2.36-2.50 (3H, m), 2.80 (1H, d, J = 16.4 Hz), 3.09 (1H, dd, J = 16.5, 6.8 Hz), 3.92 (1H, d, J = 6.7 Hz), 4.27 (1H, d, J = 3.9 Hz), 7.10 (1H, d, J = 7.4 Hz), 7.16 (1H, d, J = 7.4 Hz), 7.29 (1H, d, J = 7.9 Hz), 7.44 (1H, d, J = 7.62 Hz), 8.06 (1H, bs), <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>) 211.0, 135.7, 134.0, 126.9, 122.1, 119.7, 118.1, 110.9, 107.5, 59.8, 46.1, 35.0, 32, 25.8; CIMS (*m/e*, relative intensity) 227 (M<sup>+</sup>+1, 100%).

**Anal. Calcd. For C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>O:** C, 74.31%; H, 6.24%; N, 12.38%. Found: C, 74.39%; H, 6.35%; N, 12.41%.



#### (Z)-1-bromo-2-iodobut-2-ene 51<sup>155</sup>

A solution of crotonaldehyde **103** (300mL. 3.63 mol) in THF-H<sub>2</sub>O (1:1, 1500mL) was cooled to -15 to -20 °C (dry ice). To the cold solution was added K<sub>2</sub>CO<sub>3</sub> (600g, 4.45 mol, 1.2 equiv), I<sub>2</sub> (1383g, 5.45 mol, 1.5 equiv) and DMAP (88.6g, 0.726 mol, 0.2 equiv) successively (the temperature was increased, controlled by the rate of addition of I<sub>2</sub>). The reaction was allowed to stir for 12 h until the disappearance of starting material (TLC, silica gel, I<sub>2</sub>) during which the temperature was allowed to increase to rt. Then solid Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (600g) and a saturated solution of Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (about 5L) was added until the reaction mixture became colorless. The mixture was extracted with EtOAc (3x8L) and the organic layers were combined. The majority of the solvent was removed under reduced pressure. The residue **104** was dissolved in THF-H<sub>2</sub>O (9:1, 8L) and cooled to -5 to -10 °C (dry ice) after which NaBH<sub>4</sub> (70g, 1.8 mol, 0.5 equiv) was added in portions

and the temperature maintained at 0°C. The mixture was extracted with EtOAc (3x8L) and dried (MgSO<sub>4</sub>). The solvent was removed under reduced pressure to provide a brown oil. After a simple wash column to remove some baseline material, iodide **105** was obtained as light yellow oil **105** (650g, 90%). <sup>1</sup>**H NMR** (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.02 (q, 1H, J = 3.9Hz), 4.30 (s, 2H), 1.85 (t, 3H, J=6.3Hz). The spectral properties of this iodide were identical to the published values.<sup>120</sup> (*Z*)-2-Bromo-2-buten-1-ol **105** (2 g, 10 mmol) was dissolved in anhydrous ethyl ether (20 mL). Phosphorus tribromide (0.380 mL, 4 mmol) was added dropwise to this solution at 0°C. This reaction mixture which resulted was stirred for 18 h at rt. The reaction was quenched with a cold aq solution of K<sub>2</sub>CO<sub>3</sub> and extracted with ethyl ether after which it was washed with brine. The organic layer was directly used in the next step.<sup>172</sup> <sup>1</sup>H NMR: 6.08 (q, J = 6.4 Hz, 1H); 4.36 (s, 2H); 1.81 (d, J = 6.4 Hz, 3H). The spectral properties of this bromide were identical to the published values.<sup>121</sup> This was directly used in the following alkylation.



Alkylation of (6S,10S)-(-)-9-oxo-12-H-6,7,8,9,10,11-hexahydro-6,10-imino-5Hcyclooct[b] indole (50) to provide (6S,10S)-(-)-9-oxo-12-(Z-2' -iodo-2' -butenyl)-6,7,8,9,10,11-hexahydro-6,10-imino-5H-cyclooct[b] indole 52

A solution of the N<sub>a</sub>-H, N<sub>b</sub>-H tetracyclic ketone **50** (1.5g, 6.637mmol) was dissolved in dry CH<sub>3</sub>CN (20mL). To this mixture, K<sub>2</sub>CO<sub>3</sub> (3.66g, 26.54 mmol) and Z-1-bromo-2iodo-2-butene **51** (2.6g, 9.95mmol) were added. This reaction mixture was stirred for 14h at rt. Analysis by TLC indicated the absence of tetracyclic ketone. The  $K_2CO_3$  was removed by passing the mixture through a bed of celite with EtOAC. After removal of the solvent under reduced pressure, the crude product was purified by flash chromatography (silica gel, hexane:EtOAC = 9:1) to provide  $N_b$ -Z-2'-iodo-2'-butenyl tetracyclic ketone **52** (2.04g, 76%). FTIR (NaCl) 3393, 1705, 1450cm-1; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 1.73 (3H, d, J=6.27Hz), 2.19-2.05 (2H, m), 2.37-2.47 (2H,m), 2.65(1H, d, J=16.87Hz), 3.05 (1H, dd, J=16.94, 6.72 Hz), 3.19-3.37 (2H,m), 3.64 (1H, d, J=6.37Hz), 3.96 (1H, bs), 5.76 (1H, q, J=6.17Hz), 6.97-7.16 (2H, m), 7.25 (1H, d, J=7.72Hz), 7.40 (1H, d, J=7.52Hz), 8.05 (1H, bs);  $^{13}$ C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  20.60, 21.71, 30.35, 34.49, 49.73, 63.36, 64.07, 106.69, 108.47, 110.94, 118.09, 119.62, 121.97, 125.72, 132.11, 132.76, 135.79, 210.40; CIMS (m/e, relative intensity) 407 (M<sup>+</sup>+1, 100%). Anal. Calcd. For C<sub>18</sub>H<sub>19</sub>N<sub>2</sub>O: C, 53.22%; H, 4.71%; N, 6.90%. Found: C, 53.15%; H, 4.80%; N, 6.64%.



# Palladium catalyzed cyclization of (6S, 10S)-(-)-9-oxo-12-(Z-2'-iodo-2'butenyl)-6,7,8,9,10,11-hexahydro-6,10-imino-5H-cyclooct [b] indole (52) to provide pentacyclic ketone (56)

The  $N_b$ -Z-2' -iodo-butenyl tetracyclic ketone 52 (1 g, 2.46 mmol) was dissolved in anhydrous THF (20mL). To this mixture NaO<sup>t</sup>Bu (0.354 g, 3.69 mmol), DPEphos (0.0927 g, 7.0 mol %) and Pd<sub>2</sub> (dba)<sub>3</sub> (0.112g, 5.0 mol %) was added. This reaction mixture was degassed 3-4 times under vacuum and back filled with argon and placed in a preheated oil bath at 80 °C for 5hrs. The reaction was guenched with water and ethyl acetate. The organic layer was separated and dried ( $Na_2SO_4$ ). The EtOAc was then removed under reduced pressure and the residue was flash chromatographed with  $CH_2Cl_2/MeOH$  (4.5:0.5) to provide the coupling product **56** (0.569g, 83% yield). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 1.63 (3H, J=6.86 Hz, d), 2.15-2.20 (1H, m), 2.38 (1H, t, J=9.9 Hz), 2.97 (1H, dd, J=15.58, 6.20 Hz), 3.27 (1H, d, J=14.43 Hz), 3.37 (1H, bs), 3.59 (1H, d, J=5.70 Hz), 3.78 (2H, bs), 4.26 (1H, d, J=59.11 Hz), 5.49 (1H, q, J=6.85 *Hz*), 7.05-7.15 (2H, m), 7.25 (1H, d, *J*=7.25*Hz*), 7.47 (1H, d, *J*=4.7*Hz*), 7.93 (s, 1H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>) δ 12.68, 22.39, 36.40, 44.55, 50.83, 55.24, 64.17, 105.62, 110. 89, 118.54, 119.69, 121. 26, 122.00, 126.88, 131.88, 135.87, 136. 34, 217.00; CIMS (m/e, relative intensity) 291 (M<sup>+</sup> +1, 100%); EIMS (m/e, relative intensity) 278 (M<sup>+</sup>, 10%), 250, 75%), 249 (85%), 182 (6%), 169 (100%), 168 (5%). **HRMS** C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O: calcd. 278.1419; found 278.1437. This material was used directly in the next step.



Conversion of the pentacyclic ketone (56) into (+)-3-ethylidene-1,3,4,7,12,12bhexahydro-2H, 6H- 2,6-methano-indole[2,3,- $\alpha$ ]-quinolizine-13-carboxaldehyde[(+)vellosimine, 14a] *via* the Wittig reaction followed by acid mediated hydrolysis.

A mixture of anhydrous potassium t-butoxide (12.8g, 0.114 mol) and methoxylmethyl triphenylphosphonium chloride (36g, 0.105 mol) in dry benzene (500mL) was allowed to stir at rt for 1h. The pentacyclic ketone 56 (4g, 14.4mmol) in THF (160mL) was then added into the above orange colored solution dropwise at rt. The mixture which resulted was stirred at rt for 24h. The mixture was diluted with EtOAc (3x700mL), washed with H<sub>2</sub>O (3x50mL), brine (50mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure to afford an oil. The solvent was removed under reduced pressure and the residue was dissolved (without further purification) in a solution of aq HCl (2N) in H<sub>2</sub>O-THF (1:1, 400mL). The solution which resulted was refluxed 55 °C (oil bath temperature) under an atmosphere of argon for 6h. The reaction mixture was brought to rt and concentrated under reduced pressure. The reaction mixture was diluted with water and extracted with ether to remove the triphenyl phosphine oxide. The aq layer was brought to pH 8 with an ice cold aq solution of NaOH (1N). The aq layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> and the combined organic layers were washed with H<sub>2</sub>O (3x100mL), brine (100 mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure to afford an oil which was crystallized to provide the aldehyde **14a** (3.78g, 90%). The spectral data for (+)-vellosimine **14a** were in good agreement with the published values.<sup>19,173</sup>



Sodium borohydride mediated reduction of (+)-3-ethylidene-1,3,4,7,12,12bhexahydro-2H,6H-2,6-methano-indole[2,3- $\alpha$ ]-quinolizine-13-carboxaldehyde[(+)vellosimine, 14a] to provide the (+)-3-ethylidene-1,3,4,7,12,12b-hexahydro-13hydroxymethyl-2H,6H-2,6-methano-indole [2,3- $\alpha$ ]-quinolizine [(+)-normacusine B, 107].

The (+)-vellosimine **14a** (100mg, 0.34mmol) was dissolved in EtOH (5mL). The NaBH<sub>4</sub> (12mg, 0.33 mmol) was added to the above solution at 0 °C in one portion. The mixture was then stirred for 8 h. The reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> (50mL) and poured into cold water (10mL). The aq layer was extracted with additional CH<sub>2</sub>Cl<sub>2</sub> (3x20mL) and the combined organic layers were washed with brine (10mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure to afford the crude product which was chromatographed to provide normacusine B **107** (90mg, 90%). [ $\alpha$ ]<sub>D</sub><sup>28</sup> = +40.5 ° (c=0.75, C<sub>2</sub>H<sub>5</sub>OH); lit [ $\alpha$ ]<sub>D</sub> =+42 ° (c=1.0, C<sub>2</sub>H<sub>5</sub>OH)<sup>174</sup>: FTIR (NaCl) 3198 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.57 (3H, d, *J*=6.76 *Hz*), 1.66 (1H, dt, *J* =12.54, 3.69 *Hz*), 1.78 (1H, q, *J*=7.49 *Hz*), 1.93 (1H, t, *J*=10.00Hz), 2.61(1H, d, *J*=15.37Hz), 2.72 (1H, s), 2.74 (1H, d, *J*=6.51 *Hz*), 3.03 (1H, dd, J=15.39, 5.10 Hz), 3.43-3.49 (4H, m), 4.07 (1H, d,

J=10.29Hz), 5.26 (1H, q, J=6.68Hz), 7.03-7.14 (1H, m), 7.28 (1H, d, J=6.00Hz), 7.42 (1H, d, J=7.00Hz), 8.22 (1H, s); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  12.75, 26.95, 27.58, 33.37, 44.14, 050/48, 54.47, 55.85, 64.90, 0104.52, 110.99, 116.84, 118.07, 119.35,121.44, 127.59, 135.33, 136.33, 137.80; **EIMS** (*m/e*, relative intensity) 294 (M<sup>+</sup>, 84%), 293 (85%), 279 (10%), 263 (30%), 169 (100%), 168 (79%). **HRMS** C<sub>19</sub>H<sub>22</sub>N<sub>2</sub>O: calcd. 294.1732; found 294.1705. The spectral data were identical to that reported in the literature.<sup>174</sup>



Synthesis of 3-ethylidiene-13-(triisopropylsilanyloxymethyl)-1,3,4,7,12,12bhexahydro-2H,6H-2,6-methano-indolo [2,3-α] quinolizine 108 from normacusine B 107 with 2,6-lutidine and triisopropylsilyl chloride.

A solution of normacusine B **107** (1.0g, 3.4mmol) was stirred in dry  $CH_2Cl_2$  (30mL) in a 125mL round bottom flask. At 0 °C, 2, 6–lutidine (1.58mL, 13.6mmol), followed by triisopropylsilyl chloride (1.82mL, 6.8mmol) were added. The reaction mixture was allowed to stir for 2h until analysis of the mixture by TLC indicated the absence of starting material. Water (50.0mL) was added to the solution to quench the reaction mixture and it was extracted with  $CH_2Cl_2$  (3x200mL). The organic layer was separated and dried (Na<sub>2</sub>SO<sub>4</sub>). The organic layer was removed under reduced pressure and the residue was purified by flash chromatography (silica gel,  $CH_2Cl_2/MeOH$  (95:5) to

provide **108** (1.38g, 90%).  $[\alpha]_{D} = +45.14 \circ (c=0.14 \text{ in CH}_{2}Cl_{2}); {}^{1}H NMR (300 MHz, CDCl_{3}) \delta 1.05 (m, 21H), 1.62 (d, 3H, <math>J=6.7Hz$ ), 1.74 (m, 2H), 2.05 (t, 1H, J=10.1Hz), 2.70 (d, 1H, J=15.5Hz), 2.79 (m, 2H), 3.08 (dd, 1H, J=15.3 and 5.3Hz), 3.60 (m, 4H), 4.17 (d, 1H, J=7.5Hz), 5.39 (q, 1H, J=6.8Hz), 7.13 (m, 2H), 7.34 (d, 1H, J=7.5Hz), 7.48 (d, 1H, J=7.4Hz), 8.03(br, 1H);  ${}^{13}C$  NMR (75.5 MHz, CDCl\_{3})  $\delta$  11.83, 12.73, 17.96, 27.10, 27.29, 33.56, 44.26, 50.55, 54.93, 56.05, 65.56, 104.72, 110.76, 116.44, 118.03, 119.20, 121.23, 127.74, 135.48, 136.18, 137.87; MS (EI) *m/e* (relative intensity) 450 (M<sup>+</sup>, 16), 263 (45), 182 (31), 169 (89), 168 (100). Anal.Calcd. for C<sub>28</sub>H<sub>42</sub>N<sub>2</sub>OSi: C, 74.61; H, 9.39; N, 6.22. Found: C, 74.46; H, 9.31, N, 6.06



IBX mediated oxidation to provide (6S,11S,11aR,E)-9-ethylidene-11-((triisopropylsilyl)oxy)methyl)-6,8,9,10,11,11a-hexahydro-6,10-methanoindolo[3,2b]quinolizin-12(5H)-one (109)<sup>130</sup>

To a solution of triisopropylsilyl ether **108** (100mg, 0.22mmol) in EtOAc/DMSO (10mL/5mL) was added IBX (0.552mg, 0.88mmol) in one portion at rt. The mixture which resulted was stirred at 80°C overnight and the reaction progress was monitored by TLC (silica gel, EtOAc). The reaction mixture was cooled at 0 °C and quenched with a

saturated solution of aq NaHCO<sub>3</sub> (4mL), followed by treatment with a saturated solution of aq Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (5mL). After this, the mixture was stirred for an additional 10min at 0°C. The aq layer was extracted with additional amounts of EtOAc (3x10mL) and the combined organic layers was washed with brine (10mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure to provide the crude oil which was purified by flash chromatography [silica gel, hexane: EtOAc (1:1)] to provide the benzylic ketone **109** (87mg, 85%). <sup>1</sup>**H NMR** (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.05 (m, 21H), 1.66 (d, 3H, *J*=6.3*Hz*), 1.82 (d, 1H, *J*=6.3*Hz*), 2.13 (t, 1H, J=9.9*Hz*), 2.86 (d, 1H, *J*=7*Hz*), 3.17 (s, 1H), 3.5 (m, 3H), 3.8 (dd, 1H, *J*<sub>1</sub>=9.6*Hz*, *J*<sub>2</sub>=4.2*Hz*), 4.3 (d, 1H, *J*=8.4*Hz*), 5.47 (q, 1H, *J*=7*Hz*), 7.39 (t, 2H, *J*=7*Hz*), 7.49 (t, 1H, *J*=7*Hz*), 7.96 (d, 1H, *J*=7*Hz*), 8.14 (d, 1H, *J*=7.2*Hz*), 9.6 (br, 1H); <sup>13</sup>C **NMR** (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  11.91, 12.97, 18.03, 29.72, 32.50, 42.63, 50.48, 54.85, 63.92, 64.60, 106.49, 111.64, 118.52, 121.64, 122.74, 123.61, 124.47, 132.30, 136.14, 154.67, 192.01; **HRMS (ESI)** calculated for C<sub>28</sub>H<sub>41</sub>N<sub>2</sub>O<sub>2</sub>Si : 465.2937; found : 465.2950.



Synthesis of (6S,11R,11aR,E)-9-ethylidene-11-(hydroxymethyl)-6,8,9,10,11,11ahexahydro-6,10-methanoindolo[3,2-b]quinolizin-12(5H)-one (110)

A solution of benzylic ketone **109** (20mg, 0.043mmol) was stirred in THF (1 mL) in a 5mL round bottom flask. At 0°C, excess TBAF hydrate was then added to the mixture and it

was allowed to warm to rt. The reaction mixture was stirred for 2h until analysis of the mixture by TLC indicated the absence of starting material. The reaction was quenched with water (10mL) and extracted with EtOAc (3x10mL), washed with brine and dried (Na<sub>2</sub>SO<sub>4</sub>). After removal of the solvent under reduced pressure, the residue was purified by flash chromatography [EtOAc : hexane (4:1)] to provide the target monol **110** (0.011g, 85%) . <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.68 (d, 3H, *J*=6*Hz*), 1.87 (d, 1H, *J*=12*Hz*), 2.12 (br, 1H), 2.28 (t, 1H, *J*=12*Hz*), 2.84 (d, 1H, *J*=6 *Hz*), 3.25 (s, 1H), 3.66 (m, 5H), 4.35 (dd, 1H, *J*<sub>1</sub>=9*Hz*, *J*<sub>2</sub>=3*Hz*), 5.55 (q, 1H, *J*=7.5*Hz*), 7.25 (m, 2H), 7.43 (d, 1H, *J*=7*Hz*), 7.98 (d, 1H, *J*=8.4*Hz*); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  11.60, 12.52, 19.30, 29.17, 42.58, 50.19, 54.38, 63.17, 64.32, 105.38, 111.63, 117.97, 120.54, 122.24, 123.19, 124.29, 133.29, 136.80, 156.43, 193.70; **HRMS (ESI)** calculated for C<sub>19</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub> : 309.1603; found : 309.1588. This material was used in a later step. The structure of ketone **110** was confirmed by X-ray analysis (see Figure 8).



Preparation of (6S,11R,11aR,12R,E)-9-ethylidene-11-(hydroxymethyl)-5,6,8,9,10,11,11a,12-octahydro-6,10-methanoindolo[3,2-b]quinolizin-12-ol (88)

A solution of alcohol 110 (15mg, 0.049mmol) was stirred in MeOH (1 mL) in a 5mL flask. At -78°C, CeCl<sub>3</sub>.7H<sub>2</sub>O (19mg, 0.054mmol) and NaBH<sub>4</sub> (2mg, 0.049mmol) was added to the mixture and it was allowed to warm to rt. The reaction mixture was stirred for 3h until analysis of the mixture by TLC indicated the absence of starting material. The reaction was quenched with aq NH<sub>4</sub>Cl (5mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3x10mL), washed with brine and dried (Na<sub>2</sub>SO<sub>4</sub>). After removal of the solvent under reduced pressure, the residue was purified by flash chromatography  $[CH_2Cl_2 : MeOH (90:10)]$  to provide the target diol 88 (13mg, 90%);  $\mathbf{R}_f \ 0.75$  (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 9 : 1);  $[\alpha]^{25}$ <sub>D</sub>  $+79^{\circ}$  (c 0.6 MeOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.67 (d, 3H, J=6.6Hz), 1.80 (td, 1H,  $J_1=12.6Hz$ ,  $J_2=6Hz$ ), 2.13 (m, 1H), 2.2 (m, 1H), 2.7 (t, 1H, J=6.6 Hz), 3.01 (s, 1H), 3.5 (m, 3H), 3.6 (m, 2H), 4.10 (d, 1H, J=8Hz), 5.23 (d, 1H, J=6Hz), 5.47 (q, 1H, J=6.6 Hz), 7.04 (m, 2H), 7.3 (d, 1H, J=7.8Hz), 7.78 (d, 1H, J=7.8Hz); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>) § 11.60, 27.20, 32.46, 39.18, 50.16, 55.58, 59.71, 63.85, 65.98, 102.27, 110.61, 116.86, 118.64, 119.18, 120.78, 126.51, 135.19, 136.91, 138.54; **HRMS (ESI)** calculated for  $C_{19}H_{23}N_2O_2$ : 311.1760; found : 311.1758. The structure of diol **88** was confirmed by X-ray analysis (see Figure 9).



Synthesis of (+)-3-ethylidene-12-methyl-1,3,4,7,12,12b-hexahydro-2H,6H-2,6-

#### methano-13-methylene-indole[2,3-a]quinolizine (111) via the Wittig reaction.

A mixture of anhydrous potassium *tert*-butoxide (3.24g, 2.75mmol) and methyltriphenylphosphonium bromide (8.93 g, 2.5 mmol) in dry benzene (150 mL) was allowed to stir at rt for 1 h. The pentacyclic ketone 56 (1.46 g, 0.50 mmol) in THF (50 mL) was then added into the above orange-colored solution dropwise at rt. The mixture which resulted was stirred at rt for 4 h. The mixture was diluted with EtOAc (500 mL), washed with H<sub>2</sub>O (3 x 50 mL), brine (100 mL), and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure and the oil which resulted was flash chromatographed (silica gel, CHCl<sub>3</sub>/MeOH) 40:1) to provide the olefin **111** (1.30 g, 90%): FTIR (CHCl<sub>3</sub>) 3439 cm-1; <sup>1</sup>**H NMR (300 MHz, CDCl<sub>3</sub>)**  $\delta$  1.67 (dt, 3H, J = 6.8, 1.8 Hz), 1.87 (dt, 1H, J= 9.2, 2.8 Hz, 2.25 (ddd, 1 H, J = 11.8, 10.0, 1.7 Hz), 3.02 (dd, 1 H, J = 15.4, 1.3 Hz), 3.24 (dd, 1 H, J = 15.5, 5.2 Hz), 3.35 (m, 1 H), 3.60 (s, 3 H), 3.84 (m, 2 H), 3.93 (m, 1 H), 4.35 (dd, 1 H, J = 9.9, 2.3 Hz), 4.86 (t, 2 H, J = 2.4 Hz, 5.34 (q, 1 H, J = 7.0 Hz), 7.08 (td, 1 H, J = 6.9, 1.2 Hz), 7.20 (td, 1 H, J = 6.9, 1.1 Hz), 7.27 (d, 1 H, J = 7.6 Hz), 7.50 (td, 1 H, J = 7.6 Hz); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  12.4, 26.2, 29.2, 35.3, 36.3, 49.6, 56.1, 56.8, 103.8, 105.4, 108.7, 115.4, 118.1, 118.8, 120.9, 127.0, 136.4, 137.3, 138.2, 151.6; EIMS (*m*/*z*, relative intensity) 290 (M+, 100), 275 (13), 182 (76). Anal. **Calcd** for C<sub>20</sub>H<sub>22</sub>N<sub>2</sub>: C, 82.72; H, 7.04; N, 9.65. Found: C, 82.84; H, 7.29; N, 9.46.



# 1,3,4,7,12,12b-hexahydro-2*H*,6*H*-2,6-methano-13-methylene-indole[2,3-*a*]quinolizine (111)

To a solution of olefin 111 (200 mg, 0.72 mmol) in THF (20 mL) was added 9-BBN (0.5 M in THF, 7.23 mL, 3.6 mmol) dropwise at 0 °C. The solution was allowed to warm to rt and stirred for 1.5 h. The reaction mixture was then cooled to 0 °C and NaBO<sub>3</sub>·  $4H_2O$ (1.67g, 10.8mmol) was added (Kabalka process) and the reaction temperature allowed to warm to rt. The mixture which resulted was stirred for 2 h at rt. diluted with  $CH_2Cl_2$  (200) mL), washed with  $H_2O$  (3 x 50 mL) as well as brine (100 mL), and dried ( $K_2CO_3$ ). The solvent was removed under reduced pressure and the residue was chromatographed (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH; 9:1) to provide alcohol **112** (149 mg, 70%). FTIR (CHCl<sub>3</sub>) 3200, 2918 cm-1;  $[\alpha]_{D} = +6.92^{\circ}$  (c 0.25, MeOH) [lit.1  $[\alpha]^{D} = +3^{\circ}$  (c 0.25, MeOH)]; <sup>1</sup>H **NMR (300 MHz, CDCl<sub>3</sub>)**  $\delta$  1.64 (dt, 3 H, J = 6.8, 1.8 Hz), 1.84 (m, 2 H), 2.24 (m, 1 H), 2.89 (q, 1 H, J = 2.8 Hz), 3.01 (m, 2 H), 3.27 (dd, 1 H, J = 10.5, 8.5 Hz), 3.48-3.76 (m, 4 H), 4.22 (dd, 1 H, J = 9.0, 4.2 Hz), 5.23 (q, 1 H, J = 6.8 Hz), 7.08 (td, 1 H, J = 7.1, 1.3 *Hz*), 7.16 (td, 1 H, J = 7.1, 1.3 Hz), 7.36 (d, 1 H, J = 7.3 Hz), 7.49 (d, 1 H, J = 7.3 Hz), 8.60 (s, 1 H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>) δ 12.7, 22.3, 26.1, 27.0, 41.8, 50.1, 52.8, 55.9, 60.9, 105.6, 111.2, 114.9, 118.1, 119.3, 121.7, 126.0, 135.6, 136.4, 137.4; Anal. Calcd for C<sub>19</sub>H<sub>22</sub>N<sub>2</sub>O: C, 77.52; H, 7.53; N, 9.52. Found: C, 77.80; H, 7.79; N, 9.34



Preparation of (6S,11S,11aS,E)-9-ethylidene-11-(((triisopropylsilyl)oxy)methyl)-5, 6, 8, 9, 10, 11, 11a,12-octahydro-6,10-methanoindolo[3,2-b]quinolizine (113)

The synthesis of 113 (TIPS derivative) from monol 112 was carried out analogous to the preparation of 108 from 107 in 90% yield. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.03 (d, 21H, *J*=3*Hz*), 1.57 (d, 3H, *J*=6*Hz*), 2.01 (m, 1H), 2.08 (m, 1H), 2.26 (t, 1H, *J*=12*Hz*), 2.81 (s, 1H), 2.97 (d, 1H, *J*=15*Hz*), 3.30 (m, 3H), 3.57 (m, 3H), 4.76 (d, 1H, *J*=9*Hz*), 5.21 (q, 1H, *J*=9*Hz*), 7.28 (m, 2H), 7.45 (d, 1H, *J*=9*Hz*), 9.88 (br, 1H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  12.69, 22.80, 26.53, 27.44, 42.99, 50.12, 52.25, 56.65, 61.76, 106.78, 110.55, 112.93, 118.21, 121.23, 126.73, 136.38, 137.26, 141.34; **HRMS** (ESI) calculated for C<sub>28</sub>H<sub>43</sub>N<sub>2</sub>OSi : 451.3145; found : 451.3172. This material was employed directly in the next step.



Preparation of (6S,11S,11aR,E)-9-ethylidene-11-((triisopropylsilyl)oxy)methyl)-

6,8,9,10,11,11a-hexahydro-6,10-methanoindolo[3,2-b]quinolizin-12(5H)-one (114)

The synthesis of ketone 114 from TIPS derivative 113 was carried out analogous to the preparation of 109 from 108 in 85% yield. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  0.96 (d, 21H, *J*=2.7*Hz*), 1.66 (d, 3H, *J*=7*Hz*), 1.81 (t, 1H, J=11Hz), 2.06 (dd, 1H, *J*<sub>1</sub>=9.9*Hz*, *J*<sub>2</sub>=3.3*Hz*), 2.25 (m, 1H), 3.35 (d, 1H, *J*=2), 3.47 (t, 1H, *J*=11), 3.74 (d, 2H, *J*=2.4*Hz*), 3.78 (s, 1H), 3.84 (dd, 1H, *J*<sub>1</sub>=10.2*Hz*, *J*<sub>2</sub>=4.8), 4.14 (dd, 1H, *J*<sub>1</sub>=10.5*Hz*, *J*<sub>2</sub>=4.5), 5.32 (q, 1H, *J*=6.9*Hz*), 7.27 (m, 3H), 8.13 (d, 1H, J=6.9Hz), 8.9 (br, 1H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  11.85, 12.72, 17.94, 25.99, 27.81, 38.52, 50.07, 55.52, 59.87, 63.54, 108.57, 111.27, 114.29, 121.77, 122.82, 123.50, 123.88, 135.75, 139.17, 155.29, 191.47; HRMS (ESI) calculated for C<sub>28</sub>H<sub>41</sub>N<sub>2</sub>OSi : 465.2937; found : 465.2914. This material was employed directly in the next step.



Preparation of (6S,11S,11aR,E)-9-ethylidene-11-(hydroxymethyl)-6,8,9,10,11,11ahexahydro-6,10-methanoindolo[3,2-b]quinolizin-12(5H)-one (115)

The synthesis of monol 115 from TIPS derivative 114 was carried out analogous to the preparation of 110 from 109 in 85% yield <sup>13</sup>H NMR  $\delta$  1.68 (td, 3H,  $J_1$ =6.6 $H_z$ ,

 $J_2=3.9Hz$ ), 2.06 (m, 2H), 2.26 (m, 1H), 3.3 (m, 1H), 3.67 (dd, 1H,  $J_1=11Hz$ ,  $J_2=5Hz$ ), 3.77 (m, 3H), 4.28 (dd, 1H,  $J_1=11Hz$ ,  $J_2=5Hz$ ), 5.4 (q, 1H, J=6Hz), 7.27 (m, 2H), 7.47 (m, 1H), 8.01 (m, 1H); <sup>13</sup>C NMR (75.5 MHz, CD<sub>3</sub>OD)  $\delta$  11.47, 24.85, 27.60, 38.07, 49.51, 54.62, 58.32, 63.23, 107.63, 111.63, 114.36, 120.63, 122.40, 123.30, 123.48, 136.70, 138.16, 156.29, 191.83 ; **HRMS** (ESI) calculated for C<sub>19</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub> : 309.1603; found : 309.1588. This material was employed directly in the next step.



Preparation of (6S,11S,11aR,12R,E)-9-ethylidene-11-(hydroxymethyl)-

5,6,8,9,10,11,11a,12-octahydro-6,10-methanoindolo[3,2-b]quinolizin-12-ol (89)

The synthesis of 89 from 115 was carried out analogous to the preparation of 88 from 110 in 85% yield.  $\mathbf{R}_f \ 0.71$  (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 9 : 1);  $[\alpha]^{25}_{\mathbf{D}} + 29.0^{\circ}$  (*c* 0.6 in MeOH); [lit.<sup>136</sup>  $[\alpha]_{\mathbf{D}}^{22} = +29.5^{\circ}$  (*c* 0.6 in CH<sub>3</sub>OH)]; <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD)  $\delta$  1.70 (d, 3H, *J*=6.9*Hz*), 1.94 (m, 2H), 2.26 (m, 1H), 2.9 (q, 1H, *J*=2.4 *Hz*), 3.14 (dd, 1H, *J*<sub>1</sub>=11.4*Hz*, *J*<sub>2</sub>=6*Hz*), 3.61 (dd, 1H, *J*<sub>1</sub>=10*Hz*, *J*<sub>2</sub>=5.4*Hz*), 3.6 (s, 3H), 3.8 (m, 1H), 4.2 (dd, 1H, *J*<sub>1</sub>=8.4*Hz*, *J*<sub>2</sub>=4*Hz*), 5.4 (m, 2H), 7.0 (t, 1H, *J*=7*Hz*), 7.08 (t, 1H, *J*=7*Hz*), 7.32 (d, 1H, *J*=8.1*Hz*), 7.76 (d, 1H, *J*=7.8*Hz*); <sup>13</sup>C NMR (300MHz, CD<sub>3</sub>OD)  $\delta$  11.48, 26.31, 28.15, 42.53, 50.10, 55.66, 58.72, 61.00, 67.33, 110.22, 110.58, 114.22, 118.64, 119.68, 120.85, 125.04, 136.17, 136.99, 138.87; **HRMS** (ESI) calculated for C<sub>19</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>:

311.1760; found : 311.1773. The optical rotation and mass spectrum were in excellent agreement with the natural product.  $^{136}$ 



Synthesis of (6S,11aR,E)-9-ethylidene-11-methylene-6,8,9,10,11,11a-hexahydro-6,10methanoindolo[3,2-b]quinolizin-12(5H)-one (116)

To a solution of diene **111** (100mg, 0.36mmol) in EtOAc/DMSO (5mL/2.5mL) was added IBX (0.225mg, 1.44mmol) in one portion at rt. The mixture was stirred at 80°C overnight and the reaction progress was monitored by TLC (silica gel, EtOAc). The reaction mixture was cooled at 0 °C and quenched with a saturated solution of aq NaHCO<sub>3</sub> (4mL), followed by treatment with a saturated solution of aq Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (5mL). After this, the mixture was stirred for an additional 10min at 0°C. The aq layer was extracted with additional amounts of EtOAc (3x10mL) and the combined organic layers was washed with brine (10mL) and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure to provide the crude oil which was purified by chromatography [silica gel, hexane: EtOAc (3:1)] to provide the benzylic ketone **116** (84mg, 80%). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) 1.66 (d, 3H, *J*=6*Hz*), 1.9 (m, 1H), 2.36 (m, 1H), 3.6 (m, 3H), 4.0 (m, 1H), 4.9 (d, 1H, *J*=2*Hz*), 5.0 (d, 1H, *J*=2*Hz*), 5.39 (q, 1H, *J*=6.8*Hz*), 7.27 (m, 2H), 7.4 (m, 1H), 7.6 (m, 2H), 8.0 (m, 1H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  11.13, 34.44, 38.50, 50.13, 54.41, 67.44, 105.89, 106.23, 124.30, 128.49, 128. 65, 131.60, 131.74, 132.39,

135.75, 136.72, 145.10, 155.95, 190.17; **HRMS** (ESI) calculated for  $C_{19}H_{19}N_2O$  : 291.1497; found : 291.1513. This material was employed directly in the next step.



Preparation of (6S,11S,11aR,12S,E)-9-ethylidene-11-(hydroxymethyl)-

5,6,8,9,10,11,11a,12-octahydro-6,10-methanoindolo[3,2-b]quinolizin-12-ol (117)

To a solution of olefin **116** (100 mg, 0.344 mmol) in THF (10 mL) was added 9-BBN (0.5 M in THF, 1.63 mL, 1.72 mmol) dropwise, at 0 °C. The solution was allowed to warm to rt and stirred for 1.5 h. The reaction mixture was then cooled to 0 °C and NaBO<sub>3</sub>· 4H<sub>2</sub>O (0.795g, 5.16mmol) was added and the reaction temperature allowed to warm to rt. The mixture which resulted was stirred for 2 h at rt, diluted with CH<sub>2</sub>Cl<sub>2</sub> (50 mL), washed with H<sub>2</sub>O (3 x 50 mL) as well as brine (50 mL), and dried (K<sub>2</sub>CO<sub>3</sub>). The solvent was removed under reduced pressure and the residue was chromatographed (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH; 9:1) to provide alcohol **117** (74mg, 70% yield). **R**<sub>f</sub> 0.73 (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH, 9 : 1);  $[\alpha]^{25}_{D} = +17.0^{\circ}$  (*c* 0.6 in CH<sub>3</sub>OH); [lit.<sup>136</sup>  $[\alpha]_D^{22} = +29.5^{\circ}$  (*c* 0.6 in CH<sub>3</sub>OH)]; <sup>1</sup>**H NMR** (300 MHz, CD<sub>3</sub>OD)  $\delta$  1.70 (d, 3H, *J=6Hz*), 1.92 (m, 2H), 2.27 (m, 1H), 2.9 (s, 1H), 3.16 (dd, 1H, *J*<sub>1</sub>=*11.4Hz*, *J*<sub>2</sub>=*6Hz*), 3.61 (m, 3H), 3.8 (m, 1H), 4.2 (m, 1H), 5.3 (q, 1H, *J=6.6Hz*), 5.36 (d, 1H, *J=4.8Hz*), 7.0 (m, 2H), 7.35 (s, 1H), 7.82 (d, 1H, *J=7.5Hz*); <sup>13</sup>C NMR (300MHz, CD<sub>3</sub>OD)  $\delta$  11.48, 26.35, 28.22, 42.59, 50.06.

55.69, 58.72, 59.17, 61.07, 67.42, 110.23, 110.57, 114.09, 118.62, 119.67, 120.81, 125.07, 136.35, 136.98, 139.12; **HRMS** (ESI) calculated for  $C_{19}H_{22}N_2O_2$  : 311.1760; found : 311.1748

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# II. THE SYNTHESIS OF ALPHA 5 SUBTYPE SELECTIVE LIGANDS FOR

### GABA (A) /BENZODIAZEPINE RECEPTORS

#### **1. Introduction**

The GABA<sub>A</sub> /benzodiazepine receptor (BzR) is the major neurotransmitter in the central nervous system which is inhibitory in action.<sup>1</sup> This receptor is involved in many functions of the CNS as well as in many diseases.<sup>2</sup> Many of these diseases occur due to the improper balance of neurotransmitters which can be restored using therapeutics. The benzodiazepines continue to be the most widely used drug to treat various disorders such as anxiety disorders<sup>3</sup> and epilepsy<sup>4</sup> but they have side effects including muscle relaxant/ataxic, sedative-hypnotic effects, hypersensitive behavior, phobias, along with effects which lead to or complicate drug abuse<sup>5</sup>.

The  $\gamma$ -aminobutyric acid (GABA<sub>A</sub>) receptors are heteroligomeric membrane-bound protein complexes that are composed of several subunits and are termed type A, type B and type C. In the case of type A receptors, the inhibitory effects of GABA mediated by these receptors can be modulated by a number of pharmacological agents that selectively bind to allosteric sites on these ion channels.<sup>6-9</sup> GABA<sub>A</sub> receptors are pentameric assemblies of proteins derived from a family of subunits ( $6\alpha 3\beta 3\gamma 1\delta 1\pi 1\theta 1\epsilon$  and  $3\rho$ ) which form the chloride ion channel. The most common form of native GABA<sub>A</sub> receptors contains  $\alpha$ ,  $\beta$ , and  $\gamma$  subunits in a 2:2:1 stoichiometry.<sup>10-12</sup> It has been shown that recombinant receptors which contain these subunits most closely mimic the biological, electrophysiological and pharmacological properties of native GABA<sub>A</sub> receptors which contain a benzodiazepine recognition site (GABA<sub>A</sub>/Bz receptors).<sup>13</sup>The benzodiazepine binding site is located at the interface between <sup>14,15</sup> $\alpha$  and  $\gamma$  subunits.<sup>16</sup> While it is clear the  $\gamma$  subunit is also required for benzodiazepine binding,<sup>11,13</sup> the fact that most native GABA<sub>A</sub> receptors contain a  $\gamma$ 2 subunit is in agreement with the  $\alpha$  subunit as the key determinant of benzodiazepine binding and efficacy. However, it is clear the gamma subunit is also required for benzodiazepine binding (see Figure 1) was mentioned. Recent advances in molecular biology have identified 21 subunits which comprise a series of GABA<sub>A</sub>/BzR subtypes (ion channels).<sup>17-21</sup> Importantly, it has now been found that certain disease states, as well as tolerance and dependence are related to the up or down regulation of specific subunits.



**Figure 1**. Absolute subunit arrangement of the  $\alpha 1\beta 2\gamma 2$  GABA<sub>A</sub> receptor when viewed from the synaptic cleft. The GABA binding sites are located at the  $\beta^+\alpha^-$  subunit interfaces and the modulatory Bz BS (Bz) is located at the  $\alpha^+\gamma^{\Box}$  subunit interface. <sup>22,23</sup>

#### 1.2. Benzodiazepine pharmacology

Ligands that bind to the BZ site can influence the binding of GABA to its receptor and thereby alter the flux of chloride ions through the ion channel.<sup>24</sup> Ligands at the BZ site are categorized as agonists, inverse agonists, or antagonists. Agonists enhance the effects of GABA by increasing the frequency of channel opening to provide a net

hyperpolarization of the neuron and a decreased excitability. BZ inverse agonists have the opposite effect and decrease the flow of chloride ions (negative modulators, NAM) which results in a depolarization and an increased neuronal excitability. Between the two efficacy extremes, there is a continuum of partial agonists and partial inverse agonists as well as antagonists, the latter of which do not alter chloride flow and are functionally silent. These different efficacies are reflected in different behavioral effects of BzR site ligands in mammalian species. The schematic diagram of subunits and ligands of the benzodiazepine-GABA receptor-chloride channel complex can be seen in Figure 2.



Subunits other than  $\alpha$ ,  $\beta$ , and  $\gamma$  exist.

**Figure 2**. The schematic diagram of subunits and ligands of the benzodiazepine-GABA receptor-chloride channel complex. Reprinted with permission of the author.

The interaction with benzodiazepines (BZ, Figure 3) has been a major influence in studies on GABA receptors because of the long history of the therapeutic application of BZs as anxiolytics, anticonvulsants, sedative-hypnotics, and muscle relaxants.<sup>9,25</sup> Although the BZs were first introduced into clinical practice in the early 1960s, it was not until 1975 that these drugs were shown to act by potentiating the inhibitory actions of GABA in the brain.<sup>1,25</sup> The presence of high-affinity specific binding sites for BZs in the mammalian brain was then demonstrated by Squires, Haefely et al. <sup>6,26,27</sup> Converging lines of evidence established that these sites were contained in the same macromolecule as the GABA sites and the chloride channel, moreover, that all three elements were coupled allosterically.<sup>28-30</sup> The term "GABA/BZ receptor" came into use for this complex (and is still encountered). Until recently, progress in this field was driven by the synthesis of a vast range of BZs and BZ-like drugs, which acted at the allosteric BZ sites in the brain. GABAA receptors exhibited clinical anxiolytic or sedative properties which correlated to their binding potencies in the CNS.<sup>31</sup> The structures of some ligands which act at the BZ site are shown in Figure 3.



Figure 3. GABA, diazepam, triazolam and midazolam.

BDZs, such as diazepam(2), are the most potent anticonvulsent, muscle relaxant, sedative-hypnotic and anxiolytic compounds in clinical use.<sup>32</sup> They do not substitute for GABA, but instead enhance the inhibitory effects of GABA. BDZ's allosterically bind to the receptor at a different location than GABA does and enhance the chloride channels conductance by increasing the frequency of gated channel opening.<sup>33-37</sup> BDZ and compounds interacting with the benzodiazepine site of GABA<sub>A</sub> receptors can only modulate ongoing GABAergic activity (see Figure 4). These compounds cannot elicit chloride ion flux in the absence of GABA<sup>33,37</sup> and thus can exhibit an extremely low degree of toxicity. Many different classes of compounds have been shown to interact with the BDZ binding site of GABA<sub>A</sub> receptors.<sup>38-42</sup> In each of these classes, compounds

could be identified that enhanced or reduced the action of GABA on GABA<sub>A</sub> receptors, as mentioned above.



Figure 4: Mode of action of benzodiazepines. Reprinted with permission of the author

## **1.3. Molecular Modeling**

An extensive study of the BzR pharmacophore has been carried out in Milwaukee.<sup>42-47</sup> Based on data from at least 166 ligands at BzR, a unified BzR pharmacophore receptor model of the BDZ receptor has been developed. Outlined in Figure 5 are the proposed descriptors required for high affinity binding to the inclusive pharmacophore.<sup>43,48</sup> The important points of the model include three anchor points which were termed L<sub>1</sub>, H<sub>1</sub>, and H<sub>2</sub>; region L<sub>1</sub> represents a center of lipophilic interaction with the ligands, while H<sub>1</sub>, H<sub>2</sub> and  $A_2$  represent two hydrogen bond donor sites and one hydrogen bond acceptor site on the receptor protein, respectively, in the receptor binding domain.<sup>49</sup> Three additional lipophilic regions were also proposed, termed  $L_2$ ,  $L_3$  and  $L_{Di}$ , which are important for binding affinity and selectivity as well as efficacy of ligands. It is noteworthy to mention that lipophilic interactions between a ligand and the receptor include vander Waals interactions, as well as potential  $\pi$ - $\pi$  and p- $\pi$  stacking between the aromatic moieties of the ligand and groups on the receptor protein.



**Figure 5.** Diazepam in the Pharmacophore/Receptor Model for the BzR Site.

Pyrazolo[3,4-c]quinolin-3-one CGS-9896 (dotted line), a diazadiindole (thin line), and diazepam (thick line) fitted to the inclusive pharmacophore model for the BzR. Sites H1 and H2 represent hydrogen bond donor sites on the receptor protein complex, while A2 represents a hydrogen bond acceptor site necessary for potent inverse activity in vivo. L1, L2, L3 and LDi are four lipophilic regions in the binding pharmacophore. Descriptors S1, S2, and S3 are regions of negative steric repulsion.

Areas of negative steric repulsion with the protein are described as  $S_1$ ,  $S_2$ , and  $S_3$ . Examination of the included volumes indicated that the shapes of binding pockets for  $\alpha 1$ ,  $\alpha 2$  and  $\alpha 3$  subtypes are very similar to each other. Region L2 for the  $\alpha 5$  containing subtype appeared to be larger in size than the analogous region of the other receptor subtypes. The LDi region in contrast appeared to be larger in the  $\alpha 1$  subtype than in the other subtypes. <sup>50</sup> The receptor subtype descriptor termed the DI site is related to  $\alpha 4$  and  $\alpha 6$  BzR, which is devoid of or has a very small lipophilic pocket L<sub>3</sub> in comparison to the other four ( $\alpha 1$ ,  $\alpha 2$ ,  $\alpha 3$ , and  $\alpha 5$ ) subtypes The DI region appears to be near the outside of the receptor binding cleft and may be important as well for interactions at the  $\alpha$ 1 subtype which results sometimes in increased  $\alpha 1$  subtype selective affinity. With the aid of these models, several series of ligands have been synthesized and evaluated pharmacologically to determine the interaction with the GABA/benzodiazepine receptor subtype selective ligands.

#### 2. Objective of this research

Benzodiazepines and related drugs have been documented to disrupt memory in human and non-human subjects. <sup>10,13,47,51</sup>Broadly stated, administration of Bz-type drugs can result in a loss of the ability to form new memories. The memory deficits induced by Bz-type drugs are strikingly similar to those associated with lesions of the hippocampus.<sup>52</sup> Neurons in the hippocampus express structurally diverse GABA<sub>A</sub> receptors.<sup>53,54</sup> The highest density of  $\alpha$ 5GABA<sub>A</sub> receptors is found in the hippocampus proper (see Figure 6) and these receptors are mostly extra-synaptic.<sup>53</sup> The distribution of  $\alpha$ 5GABA<sub>A</sub> receptors in the hippocampus raises the possibility that this subtype plays a distinct role in memory deficits induced by Bz-type drugs.<sup>55</sup> While memory deficits induced by Bz-type drugs represent a clinical burden, understanding the mechanisms underlying these deficits might represent a therapeutic boon.



**Figure 6**: Distribution of  $\alpha$ 5 receptor in the Hippocampus. Reprinted with permission of author.

In this regard, there is increasing evidence that inverse agonists with selectivity for the  $\alpha$ 5GABA<sub>A</sub> receptor subtype may enhance cognition.<sup>56</sup> Since  $\alpha$ 5 GABA<sub>A</sub> subtypes modulate memory processes, the specific aim of this research is to synthesize  $\alpha$ 5 subtype selective ligands which can be used to treat disorders such as cognitive dysfunction which includes Alzheimer's disease devoid of side effects (sedative, addictive, hypnotic, muscle relaxant and convulsant effects). A key rationale for this

approach is that GABA neurons in the hippocampus and temporal cortical areas are largely preserved in patients  $^{57,58}$  with Alzheimer's disease (AD), and these preserved neurons contain  $\alpha$ 5GABA<sub>A</sub> receptor subtypes that are only moderately reduced in number in AD.<sup>58</sup>

Another critical therapeutic goal of this research is to synthesize ligands which would also reduce anxiety. In fact, BZ-type drugs are commonly given to dementia patients exhibiting "agitation" (usually defined as "inappropriate verbal, vocal, or motor activity not explainable by apparent needs or confusion").<sup>59</sup> Thus, in addition to enhancing cognition, it would be advantageous to develop analogs which may provide a unique approach to reduce agitation/anxiety in patients with cognitive disorders.

#### **3. Results and Discussion**

#### **3.1.** Synthesis of $\alpha$ 5 selective 1, 4 benzodiazepines

The need for  $\alpha$ 5 selective ligands to study the hippocampus, as well as results based on previous structure-activity relationships<sup>60-64</sup> prompted a study of agonist selectivity by modification of 7-substituted benzodiazepines.<sup>60-63</sup> Earlier in this laboratory,<sup>49,62-66</sup> a series of acetylene based  $\alpha$ 5 selective inverse agonists were prepared.<sup>67</sup> These agents were among the most  $\alpha$ 5 selective ligands ever reported.<sup>67</sup> Because the acetylene function (interaction in lipophilic pocket L<sub>2</sub>) enhanced selectivity at  $\alpha$ 5 BzR, it was decided to effect a similar change with orally active benzodiazepines to extend the SAR. In the past two decades computer modeling has evolved into a full-fledged discipline independent from traditional structure-activity-relationships (SAR) and molecular

biology and proven to be a valuable tool in drug discovery.<sup>68-76</sup> This technique was important in the development of selective ligands for GABA/BzR subtypes due to the diversity of structural classes of active ligands which bound to these subtypes.<sup>31,74,76</sup> For this the unified pharmacophore/receptor model (see Figure 5) which was described previously was used.

In order to develop orally active high affinity selective ligands for  $\alpha 5$  containing isoforms at the BzR, diazepam was chosen as the starting point. According to classical structure-activity-relationships,<sup>77</sup> diazepam does not bind to  $\alpha 4$  and  $\alpha 6$  containing isoforms (these are termed "diazepam insensitive sites"),<sup>77,78</sup> consequently, interaction at  $\alpha 4$  and  $\alpha 6$  sites was retarded from the beginning. It was felt diazepam bound to all diazepam sensitive (DS) receptor subtypes at the three anchor points:  $H_1$ ,  $H_2$  and  $L_1$ .<sup>65,71,79</sup> The imine nitrogen atom was believed to be hydrogen bonded to a hydrogen bond donor site on the receptor protein $(H_2)$ , the carbonyl oxygen was felt to be hydrogen bonded to another hydrogen bonding descriptor( $H_1$ ), while the benzene A-ring interacted with the receptor protein at a lipophilic site  $(L_1)$ . In addition, it was felt diazepam also interacted in lipophilic pockets  $L_2$  and  $L_3$ , as shown in Figure 7. Lipophilic pocket  $L_3$ was believed to be very small in  $\alpha 4$  and  $\alpha 6$  containing isoforms which retarded affinity of 1,4-benzodiazepines at these subtypes.<sup>49,79,80</sup> When the agonist QH-II-066 11 was aligned in the included volume of pharmacophore/receptor for  $\alpha 2\beta 3\gamma 2$  and  $\alpha 5\beta 3\gamma 2$ subtypes, the fit was excellent (Figure 9 and Figure 10) respectively.



Figure 7. Diazepam (2) aligned in the pharmacophore model.<sup>75,81</sup>



**Figure 8:** The ligand QH-II-66, (**11**, a 7-acetylenyl-1, 4-benzodiazepine, in black), was designed to offer better occupation of this region and exhibit  $\alpha$ 5 selectivity. Note that diazepam shown in the Figure was intentionally offset from its original position, which had overlapped completely with QH-II-066, solely for the sake of visualization.<sup>65</sup>,<sup>66</sup>



**Figure 9**. The ligand, QH-II-66 **11**, aligned in the included volume of the pharmacophore/receptor model for the  $\alpha 2\beta 3\gamma 2$  subtype.



Figure 10. The ligand, QH-II-66 11, aligned in the included volume of the

pharmacophore/receptor model for the  $\alpha 5\beta 3\gamma 2$  subtype.

Based on the findings at  $\alpha$ 5 containing isoforms<sup>64,74,76,81</sup> and the results in the imidazobenzodiazepine series described previously,<sup>61-63</sup> a diazepam analog with an acetylenic function at position-7(QH-II-066, **11**) was prepared, and better selectivity *in vitro* for  $\alpha$ 5 containing isoforms was immediately achieved in the agonist series.<sup>74-76,82</sup> It was felt that lipophilic pocket L<sub>2</sub> was larger in  $\alpha$ 5 containing subtypes as compared to  $\alpha$ 1 containing isoforms,<sup>47,74-76</sup> hence the  $\alpha$ 5 subtype tolerated a larger lipophilic group in region L<sub>2</sub>. To develop ligands more selective for  $\alpha$ 5 containing BzR subtypes, interactions at H<sub>1</sub>, H<sub>2</sub> or L<sub>3</sub> must be modified since not many changes can be done at the L<sub>2</sub> pocket.

## **3.1.2.** Synthesis of gram quantities of the 1, 4 benzodiazepine (QH-II-066)

The large scale synthesis here of this  $\alpha 5\beta 3\gamma 2$  selective ligand **11** (Scheme 1), which began from 2-aminobenzophenone **5** and followed the classic work of Fryer, Sternbach et al.<sup>83,84</sup> as modified by Huang and Liu, was carried out.<sup>65,85</sup> Benzophenone **5** was acylated on 100 gram scale by bromoacetylbromide. The bromoacetamide **6** which resulted was then treated with a saturated solution of ammonia (gas) in methanol at reflux to afford benzodiazepine **7** in 99% yield. Regioselective bromination<sup>84</sup> of this ligand at position-7 was realized after treatment with 2 equivalents of bromine in acetic acid in the presence of 2 equivalents of concentrated sulfuric acid at room temperature. After the reaction was allowed to stir for seven days, ninety percent of the monobrominated benzodiazepine **8** was obtained after alkaline workup. It was necessary to stir the reaction with only 2 equivalents of bromine for several days to prevent dibromination. Attempts to use additional bromine or to heat the reaction mixture resulted in polybromination. It was difficult to remove the dibrominated side product from the desired 1, 4-benzodiazepine **8**, consequently, the reaction was run over several days. The bromide **8** then was reacted with trimethylsilyacetylene in the presence of a palladium catalyst to provide trimethylsilyl analog **9**.<sup>62,86,87</sup> This TMS-protected material **9** was then methylated with methyl iodide/sodium hydride to give the N-methyl benzodiazepine **10**. In this manner the acetylenic function was not methylated. This material was subjected to fluoride-mediated disilation on 20 g scale to furnish gram quantities of **11** (QH-II-066) for studies in rodents and primates.

In agreement with the molecular modeling, QH-II-066 **11** indeed bound more potently to  $\alpha 5\beta 3\gamma 2$  subtypes than to  $\alpha 1\beta 3\gamma 2$  GABA/BzR receptor sites. This important result indicated that perhaps QH-II-066 might exhibit anxiolytic activity but would exert less sedative-hypnotic, muscle relaxant or ataxic effects when compared to the control (diazepam **2**).



Scheme 1: Synthesis of QH-II-066

Compound	α1	α2	α3	α4	α5	α6
Diazepam (2)	14	7.8	13.9	$ND^{a}$	13.4	$ND^{a}$
QH-II-066 (11)	76.3	42.1	47.4	$ND^{a}$	6.8	>3000

**Table 1.** *In Vitro* Receptor Binding Data of Diazepam (2) and **QH-II-066** (11). Binding Affinity at  $\alpha x \beta 3 \gamma 2$  GABA<sub>A</sub>/BzR Receptor Subtypes (Values are Reported in nM)

ND: not determined yet. <sup>*a*</sup>Binding at  $\alpha$ 4 and  $\alpha$ 6 subtypes has not been determined, but since the 6-phenyl group is present, the ligand will not bind to  $\alpha$ 4 and  $\alpha$ 6 subtypes.

Studies by Lelas and Cook et al.<sup>88</sup> reflect that QH-II-066 **11** has functional selectivity *in vivo* with diazepam-like efficacy at  $\alpha$ 5 and partial efficacy at the  $\alpha$ 1 subtype. This study showed that QH-II-066 had anxiolytic activity with reduced sedative effects compared to diazepam **2**. This study also indicated that the 7-acetyleno substituted diazepam analog exhibited less potency against ECS induced seizures relative to diazepam than against PTZ-induced seizures. Hence  $\alpha$ 1 subtype may play an important role in ECS-induced seizures than in PTZ-induced seizures (see Appendix II).

Recent studies by Weiss and Cook et al.<sup>89</sup> show that activation of GABA (A) receptors induces apoptosis in neuroblastoma. The mechanism by which GABA-A activation inhibited the growth of neuroblastoma cells was assessed. To enhance the signal readout of this pathway,  $\alpha$ 5 selective GABA-A receptor agonist QH-II-066 was used. Apoptosis was induced in both LAN-5 (high GABA-A expression levels) and Kelly cells (low GABA-A expression levels) (**Figure 11 A and B**), as assessed by annexin V flow cytometry. Immunoblotting revealed that upon addition of GABA and QH-II-066, cleaved PARP, an indicator of apoptosis, was increased. Notably, immunoblotting revealed a decrease in activity of the PI3K/AKT and MAPK pro-growth and survival pathways, indicated by decreased phosphorylation levels of Akt and Erk, respectively (**Figure 11 C**). Together these data suggest that specific activation of the GABA-A receptor decreases cell viability, induces apoptosis, and suppresses pro-growth and survival signaling pathways in neuroblastoma cell lines (see Appendix III).





Figure 11. GABA activation decreases viability and induces apoptosis in neuroblastoma cells (A) Lan5 and Kelly cells were treated with GABA 100  $\mu$ M or the selective ligand for GABA (QH-II-066, 11) 1  $\mu$ M or both and analyzed by annexin V-FITC flow cytometry for apoptosis after 72h. (B) Percentages of positive cells for apoptotic marker annexin V are mean  $\pm$  s.e. for triplicate samples with treatments indicated. (C) Cells analyzed by immunoblot using antisera indicated show decreased phosphorylation of Akt and Erk.

#### **3.1.3. Synthesis of PS-1-37**

Since the  $\alpha$ 5 *in vitro* selectivity and efficacy of QH-II-066 **11** was increased in regard to the potent controls, it was decided to maintain the acetylenic function at C (7) and determine this potential effect at L<sub>2</sub> of other orally active BZ agonists.

PS-I-37 is a stable analog of XHe-II-053 related to an agonist developed by Upjohn, and was prepared for study. A mixture of the bromo benzodiazepine 3 was taken up in pyridine and phosphorous pentasulfide to form the thione 12. This was further reacted with glycine,  $Na_2CO_3$  in ethanol and a water mixture to form acid 13 as a yellow solid. This was suspended in dry  $CH_2Cl_2$  and DCC was added. The suspension which resulted was stirred at 40°C for 2hrs. The solvent was removed under vacuum to give the ketone 14 as brown oil. The cyclized product was dissolved in benzene, after which dimethylformamide diethyl acetal and triethyl amine were added. The residue was then crystallized from ethyl acetate to give amine 15. A solution of 15 in dry toluene was treated with 1-methyl piperazine and refluxed for 5hrs and then the solvent was removed under vacuum to give a gum which was crystallized to furnish 16. To the suspension of compound 16 in acetonitrile and triethyl amine was added bis(triphenylphosphine) palladium(II) acetate to furnish the trimethylsilyl analogue 17. The trimethyl silyl was removed using TBAF in THF to provide the acetylenic product 18 (PS-I-37) and this material was anxiolytic in monkeys (Dr.Rowlett, Harvard).



**Table 2.** *In Vitro* Receptor Binding Data on Diazepam (2) and **PS-1-37** (18). Binding Affinity at  $\alpha x \beta 3\gamma 2$  GABA<sub>A</sub>/BzR Receptor Subtypes (Values are Reported in nM)

Compound	α1	α2	α3	α4	α5	ab
Diazepam (2)	14	7.8	13.9	$ND^{a}$	13.4	$ND^{a}$
PS-1-37 (18)	193	35	435	$ND^{a}$	22	5000

ND: not determined yet. <sup>*a*</sup>Binding at α4 subtypes has not been determined, but since the 6-phenyl

group is present, the ligand will not bind to  $\alpha 4$  and  $\alpha 6$  subtypes.



N=2 for each receptor type

Figure 12. The Oocyte Efficacy Data on PS-1-37(18).

**Table 3.** EC<sub>50</sub> and Hill Slope for Modulation of Control GABA EC<sub>3</sub> by **PS-1-37** (18) in Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Results Represent EC<sub>50</sub> (95% Confidence Interval for EC<sub>50</sub>) and Hill Slope (Nh)  $\pm$  SD from at Least Four Cells from Two Batches.

		α1β3γ2	α2β3γ2	α3β3γ2	α4β3γ2	α5β3γ2	α6β3γ2
	EC <sub>50</sub>	>1,1µM	>228nM	>2,3µM		87nM	
PS-I-37	95%CI					48-158	
	Nh						

**Table 4.** Modulation of Control GABA EC3 by Diazepam and **PS-1-37** (18) on XenopusOocytes Expressing Different GABAA Receptor Subtypes. Values are Presented asMean  $\pm$  SEM of at Least Four Oocytes from at Least Two Batches

		α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
	10nM	135±4%	193±5%	167±8%	160±2%
Diazepam(2)	100nM	246±16%	400±22%	461±34%	322±7%
	1µM	345±27%	508±29%	776±44%	420±12%
	10µM	366±27%	520±14%	749±29%	377±14%
	10nM	114±2%	119%	ns	136±1%
<b>PS-1-37</b> (18)	100nM	144±1 %	172±3%	ns	259±16%
	1µM	272±3 %	271±15%	324±9%	438±12%
	10µM	402±12%	311±1%	402±29%	345±4%

All the above values were significantly different from the respective control currents (p<0.01, Student's*t*-test) except where indicated (n.s.). NS means not significant.

#### 3.1.4. Synthesis of 8-substituted imidazobenzodiazepines

As mentioned above, the  $\alpha$ 5 subtypes are primarily found in the hippocampus<sup>90</sup> which is important for memory and learning<sup>91</sup>. Studies done by Mohler et al. <sup>92</sup>on  $\alpha$ 5 knockin mice provided strong evidence that hippocampal extrasynaptic  $\alpha$ 5 GABA<sub>A</sub> receptors play a critical role in associative learning.<sup>92</sup> The 8-substituted imidazobenzodiazepines such as RY-23 and RY-24 exhibited a greater affinity and selectivity at the  $\alpha$ 5 subtype. Previously, the subtype selective inverse agonist activity<sup>60-64,85</sup> of a series of  $\alpha$ 5 subtype selective ligands [(**20**, RY-023), (**21**, RY-024), (**22**, RY-079) and (**23**, RY-080)] was reported. The design was based on the structure of **19** (Ro 15-4513) and, in addition, McKernan, Atack et al. (see **24**) reported several related ligands as well.<sup>93,94</sup> These ligands are BzR inverse agonists *in vivo* and have been shown to enhance cognition.<sup>17-21</sup> One of these ligands was shown by Bailey et al.<sup>91</sup> to be important in the acquisition of fear conditioning and provided further evidence for the involvement of hippocampal GABA<sub>A</sub>/benzodiazepine receptors in learning and anxiety. This was supported by the work of DeLorey et al.<sup>95</sup> in a memory model with a ligand closely related to  $\alpha$ 5 subtype selective inverse agonists **20** and **21**.



**Figure 13**. Ro 15-4513 (**19**), RY-023 (**20**), RY-024 (**21**), RY-079 (**22**), RY-80 (**23**) and the Merck compound (**24**).

The ligands **20-23** and **24** depicted in Figure 13 were 40-70 fold more selective for  $\alpha$ 5 subtypes in comparison to  $\alpha$ 1 subtypes; however, better selectivity remains a goal of paramount importance. In collaboration with Sieghart *et al.* the efficacy of inverse

agonist RY 24 (21) was determined (Figure 46). It was demonstrated that RY 24 (21) was a potent inverse agonist at  $\alpha$ 5 subtypes in oocytes with a much weaker efficacy at the other subtypes.



Figure 14. Subtype selective efficacy of RY-24 (21); dose response curves for RY-24 in oocytes expressing different subunit combinations of GABA<sub>A</sub> receptors. Subtype combinations are indicated in legends. cRNA–injected Xenopus oocytes were held at – 60 mV under two-electrode voltage clamp. Increasing concentrations of RY-24 were superfused together with a GABA concentration eliciting ~ 20% of the maximal current amplitude. RY-24 was pre-applied for 30 sec before the addition of GABA, which was co-applied with the drugs until a peak response was observed. Data were normalized for each curve assuming 100% for the response in the absence of RY-24. RY-24 was made up and diluted as stock solution in DMSO. Final DMSO concentrations perfusing the oocyte were 0.1%. Values are presented as mean  $\pm$  SD of at least 4 oocytes from at least 2 batches.

Inspection of the data in Table 4 reveals some observations worth noting. While the lipophilic substituent at  $R_8$  of RY-24 **21** and RY-80 **23** decreased the affinity for  $\alpha 1$ ,  $\alpha 2$  and  $\alpha 3$  subtypes, it retained affinity for  $\alpha 5$  and actually increased affinity for the  $\alpha 6$  subtype. This data again supports the importance of the occupation of the lipophilic pocket  $L_2$  for potent selectivity at the  $\alpha 5$  subtype.

**Table 5.** *In Vitro* Receptor Binding Data of  $\alpha$ 5 ligands. Binding Affinity at  $\alpha x\beta 3\gamma 2$ GABA<sub>A</sub>/BzR Receptor Subtypes (Values are Reported in nM)



				$\mathbf{K}_{i}\left(\mathbf{nM}\right)^{a}$				
Ligand	<b>R</b> <sub>8</sub>		R <sub>3</sub>	α1	α2	α3	α5	α6
RY-24 (	(21) ———	-H	<i>t</i> -Bu	26.9	26.3	18.7	0.4	5.1
RY-23 (	(20)	—тмѕ	<i>t</i> -Bu	197	143	255	2.61	58.6
RY-80 (	(23) ———	-H	Et	28.4	21.4	25.8	0.49	28.8
RY-79 (	(22)	—тмѕ	Et	121	142	198	5.0	114

<sup>a</sup> "Synthesis and Pharmacological Properties of Novel 8-Substituted Imidazobenzo-diazepines:
High Affinity, Selective Probes for α5 Containing GABA<sub>A</sub> Receptors, "Liu, R.; Hu, R.; Zhang,
P.; Skolnick, P.; Cook, J. M., *J. Med. Chem.*, **39**, 1928-1934 (**1996**).
In agreement with previous studies, <sup>61</sup> these  $\alpha$ 5 selective ligands were highly selective in suppressing ethanol-maintained responding (Figure 25).<sup>96</sup> As previously stated, the hippocampus contains the greatest concentration of  $\alpha$ 5-containing receptors in the CNS.<sup>90,97</sup> and it is possible that these hippocampal  $\alpha$ 5 receptors may regulate alcoholmotivated responding following systemic administration of an  $\alpha$ 5-selective agent. Furthermore, RY-24 also antagonized the motor-impairing and sedative effects of ethanol in Long-Evans rats. Combined with additional studies within the ventral pallidum (VP), it has been proposed that the GABAergic systems within the VP and hippocampal pathways may represent new extensions of the mesolimbic ethanol reward circuitry. Although these data do not strongly support a direct role for the modulatory influences of intrinsic efficacy in the behaviors examined, the synthesis of  $\alpha 5$  subtype selective ligands provides researchers a unique opportunity to explore the role of this subtype in the neurobehavioral effects of alcohol.<sup>61,98</sup> However, the efficacy (inverse agonist, negative modulation) of RY-23 and RY-24 at  $\alpha 1$  and  $\alpha 2$  subtypes precludes potential clinical uses of such agents for both agents are proconvulsant and RY-24 is also convulsant.

Additional behavioral studies of RY-24 were performed by Helmstetter *et al.* and provided further support for the role of the hippocampus in anxiety and learning.<sup>91</sup> Moreover, the data suggested that Bz BSs within the hippocampus are important for the acquisition of fear conditioning. Although this subtype selective ligand has been shown to be an inverse agonist at the  $\alpha$ 5 subtype,<sup>62,99</sup> this study suggested that RY-24 may act as an agonist at other alpha subtypes because larger doses of RY-24 were not as anxiogenic as the smaller doses and resulted in decreased learning. Consistent with the

studies of Stephens *et al.* using  $\alpha$ 5 knock-out mice<sup>100</sup> and the efficacy studies of Lüddens, June and Cook *et al.*,<sup>101</sup> these findings support the concept that the pharmacology observed depends upon the dose, behavioral paradigm employed and subunit composition activated. Ligands such as RY-24 have proven to be valuable in the study of the biochemical and pharmacological properties of GABA<sub>A</sub> receptors and have permitted insight into the role this protein plays in anxiety and learning.

In brief, isatoic anhydride **25** and sarcosine were heated in DMSO to form benzodiazepine **26** in 84% yield. Benzodiazepine **26** was treated with bromine in the presence of CH<sub>3</sub>COOH and H<sub>2</sub>SO<sub>4</sub> to provide the bromide **27** in 80% yield as illustrated in Scheme 3. Deprotonation of **27** with KOtBu in THF followed by treatment with diethyl chlorophosphate provided the intermediate enol phosphate.<sup>102</sup> The enol phosphate was stirred with a solution of *t*-butyl isocyanoacetate and KOtBu to yield the imidazobenzodiazepine **28** in 70% yield. This bromide was converted into **20** by a Heck-type coupling reaction<sup>60,62</sup>using bis(acetate) bis(triphenylphosphine) palladium(II) to provide TMS analog **RY 23 (20)** and the silyl group was removed in high yield on treatment with TBAF/H<sub>2</sub>O/THF to provide **RY 24 (21)**.<sup>60</sup>



#### Scheme 3: Synthesis of RY-23 and RY-24

# 4. Synthesis of $\alpha 5$ selective BzR bivalent ligands

Bivalent ligands have been termed as compounds that contain two pharmacophores joined through a connecting unit. The general structure for bivalent ligands is P-X-P (P: pharmacophore; X: spanner).<sup>103-105</sup> They may exhibit enhanced selectivity and potency relative to their monovalent analogues (P-X) when a suitable spanner X is employed. The study by Portoghese et al. on bivalent ligands that bind to different opioid receptor subtypes was among one of the most successful reported to date.<sup>103-105</sup>

A linker of 3 CH<sub>2</sub>- groups is often called the Leonard linker from its founder Nelson J. Leonard.

To enhance subtype selectivity, the bivalent form<sup>103,104,106</sup> of RY024 (**17**),<sup>60,62</sup> that is 1,3-bis(8-acetyleno-5,6-dihydro-5-methyl-6-oxo-4H-imidazo-[1,5a][1,4]-benzodiazepine-3-carboxy) propyl diester, (**33**, XLi093) was designed and prepared. The

synthesis,  $\alpha 5$  subtype selectivity and antagonist properties of this ligand are described below.

Deprotonation of **27** with KOtBu in THF followed by treatment with diethyl chlorophosphate provided the intermediate enol phosphate. The enol phosphate was stirred with a solution of ethyl isocyanoacetate and KOtBu to yield the imidazobenzodiazepine **29** in 70% yield.<sup>102</sup> This bromide was converted into **22** by a Heck-type coupling reaction<sup>60,62</sup> and the silyl group was removed to provide acetylene **23** in high yield on treatment with TBAF/H<sub>2</sub>O/THF.<sup>60</sup> Hydrolysis of the ester function of **23** provided the acid **32** in excellent yield and this material was subjected to a standard CDI-mediated coupling reaction with a three carbon linker to furnish bivalent ligand XLi093 **33** in 73.4% yield on gram scale.





The binding affinity of XLi093 (**33**) *in vitro* was determined on  $\alpha 1$ -6 $\beta 3\gamma 2$  Ltk<sup>-</sup> cells and is illustrated in Table 6. This ligand XLi093 **33** bound to  $\alpha 5\beta 3\gamma 2$  subtypes with a K<sub>i</sub> of 15 nM, but exhibited little or no affinity at other BzR/GABA<sub>A</sub> subtypes. Although the efficacy at all receptor subtypes was decreased by incorporation of the pharmacophore of **23** into the bivalent structure, the loss of affinity at  $\alpha 1$ ,  $\alpha 2$ ,  $\alpha 3$ ,  $\alpha 4$  and  $\alpha 6$  subtypes was profound. Presumably, since the second unit of XLi093 **33** lies in the extracellular domain of the BzR, the enriched selectivity is entropic. This has been the most selective  $\alpha 5$  subtype selective ligand reported, to date. Depicted in Figure 15 is the ORTEP drawing of the crystal structure of XLi093 (see Appendix VI for coordinates). The J values (J = 5.38) calculated from the dihedral angles were in excellent agreement with those from the solution determined NMR spectrum (J=6.39), hence support for the conformation as shown (Figure 16) derives from the NMR data. This conformation of XLi093 was then placed in the BzR/GABA<sub>A</sub>  $\alpha 5$  pharmacophore receptor model previously reported<sup>42,72,76</sup> from this laboratory; the fit was excellent.

Based on the binding affinity of XLi093 (**33**) and the fit at the  $\alpha$ 5 subtype, a study of the efficacy of Xli093 was then pursued in oocytes by Sieghart, Furtmueller, Li, Cook et al.<sup>107</sup> The effects of XLi093 (**33**) on chloride currents in GABA<sub>A</sub> receptors was determined.<sup>108</sup> The activity of XLi093 on GABA<sub>A</sub> receptors was characterized using *Xenopus* oocytes expressing the GABA<sub>A</sub> receptor subunits alpha 1 to alpha 6 in combination with beta 3 and gamma 2 subunits. Using the two electrode voltage clamp method, currents in the  $\mu$ A range were measured for all subunits.



Figure 15. Crystal Structure of Bivalent Ligand XLI093 (33).



Figure 16. XLi093(33) Aligned in the Included Volume of the Pharmacophore/Receptor Model for the  $\alpha$ 5 $\beta$ 3 $\gamma$ 2 Subtype.

 Table 6. In Vitro Binding Affinities at GABA<sub>A</sub>/BzR Subtypes. <sup>a</sup>Values Reported are in nM.

Ligand	α1	α2	α3	α4	α5	α6
RY079 ( <b>22</b> )	121	142	198	159	5.0	114
RY080 (23)	28.4	21.4	25.8	53	0.49	28.8
XLi093 (33)	>1000	>1000	858	1550	15	>2000

<sup>a</sup>K<sub>i</sub> values represent the mean of two determinations which differed by less than 10%. Data were generated using Ltk<sup>-</sup> cell membranes expressing human  $\alpha x\beta 3\gamma 2$  receptors combinations in response to application of a saturating concentration of GABA (10 mM). Two electrode voltage clamp experiments were performed to test whether XLi093 triggered chloride currents,<sup>108,109</sup> modulated GABA-induced currents or antagonized the effects of benzodiazepines in oocytes that express GABA<sub>A</sub> receptors.

Bivalent ligand XLi093 **33** at concentrations up to 1  $\mu$ M did not trigger chloride currents in any of the tested subtypes of the GABA<sub>A</sub> receptor. This dimer, however, at micromolar concentrations modulated GABA-induced currents in an alpha subtype specific manner. To test for agonistic or inverse agonistic effects of XLi093, the compound was coapplied with a concentration of GABA that induced app. 3% of the maximum current amplitude. At concentrations of XLi093 **33** up to 100 nM, this ligand caused no significant modulation of GABA-induced currents in any subtype of GABA<sub>A</sub> receptors tested. At a concentration of 1  $\mu$ M, XLi093 **33** caused a significant increase in GABA-induced currents in the GABA<sub>A</sub> receptor subtypes  $\alpha 4\beta 3\gamma 2$  [+37.6 ± 2.6 % (SEM), p < 0.0001 in four oocytes] and  $\alpha 6\beta 3\gamma 2$  [+ 17.0 ± 2.5 % (SEM), p = 0.0014 in four oocytes]. Furthermore, 1 µM XLi093 caused a reduction in GABA induced currents in the subtype  $\alpha 5\beta 3\gamma 2$  [-4.0 ± 1.5 % (SEM), p = 0.0075 in six oocytes]. However, 1 µM of XLi093 caused no significant modulation of GABA-induced currents in the GABA<sub>A</sub> receptor subtypes  $\alpha 1\beta 3\gamma 2$ ,  $\alpha 2\beta 3\gamma 2$  and  $\alpha 3\beta 3\gamma 2$ .

Ligand XLi093 (**33**) shifted the dose response curve for diazepam in a subtype specific manner. To test for benzodiazepine antagonistic effects of XLi093, dose response experiments for the stimulation of GABA-induced currents by diazepam in the absence or presence of 1  $\mu$ M XLi093 **33** were performed. Similar conditions for different subtypes were provided by using a concentration of GABA that induced app. 3% of the maximum current amplitude in the respective cell. In oocytes expressing GABA<sub>A</sub> receptors of subunit combination  $\alpha 5\beta 3\gamma 2$ , the presence of 1  $\mu$ M XLi093 (**33**), shifted the dose response curves for the stimulation of GABA induced currents by diazepam to the right (Figure 17). Even high concentrations of diazepam could not fully overcome this inhibition since maximum current stimulations in the presence of XLi093 (**33**) reached only app. 85% of current stimulations that were seen in the absence of XLi093 (**33**).

The dose response curves for diazepam with and without 1  $\mu$ M XLi093 (**33**) in oocytes expressing different subunit combinations of GABA<sub>A</sub> receptors are depicted (Figure 17). cRNA–Injected *Xenopus* oocytes were held at –60 mV under two-electrode voltage clamp. Increasing concentrations of diazepam with or without 1  $\mu$ M of XLi093 (**33**) were superfused together with a GABA concentration eliciting app. 3% of the maximal current amplitude. Diazepam and XLi093 were preapplied for 30 sec before the addition of GABA, which was coapplied with the drugs until a peak response was observed. Data were normalized for each curve assuming 100% for the response in the absence of diazepam and XLi093. Drugs were made up and diluted as stock solutions in DMSO. Final concentrations of DMSO perfusing the oocyte were 0.1%. Values are presented as mean  $\pm$  SD of at least four oocytes from at least two batches.



Figure 17. Effects of XLi093 (33) on GABA<sub>A</sub> receptors were tested by two-electrode voltage clamp experiments in cRNA injected *Xenopus* oocytes that functionally expressed several subtype combinations of GABA<sub>A</sub> receptors. In oocytes expressing GABA<sub>A</sub> receptors of the subtypes  $\alpha 1\beta 3\gamma 2$ ,  $\alpha 2\beta 3\gamma 2$ , or  $\alpha 3\beta 3\gamma 2$ , 1 µM XLi093 (33)

caused only marginal shifts of dose response curves for the stimulation of GABAinduced currents by diazepam.

Receptor binding studies indicated that the bivalent ligand XLi093 33 bound almost exclusively to the  $\alpha$ 5 subtype of GABA<sub>A</sub> receptors. In summary, the effect of this compound on various GABA<sub>A</sub> receptors expressed in Xenopus oocytes was fully investigated by electrophysiological studies. Analysis of the data indicated that bivalent ligand XLi093 up to a concentration of 1  $\mu$ M did not trigger chloride currents in any one of the tested GABA<sub>A</sub> receptor subtypes. At 1µM, XLi093 did not modulate GABA induced chloride flux in  $\alpha 1\beta 3\gamma 2$ ,  $\alpha 2\beta 3\gamma 2$ ,  $\alpha 3\beta 3\gamma 2$  but stimulated GABA-induced currents in  $\alpha 4\beta 3\gamma 2$  and  $\alpha 6\beta 3\gamma 2$  and slightly inhibited currents in  $\alpha 5\beta 3\gamma 2$ . At 1  $\mu$ M, XLi093 only marginally influenced diazepam stimulation of GABA-induced current in  $\alpha 1\beta 3\gamma 2$ ,  $\alpha 2\beta 3\gamma 2$  and  $\alpha 3\beta 3\gamma 2$ , but shifted the diazepam dose response curve to the right in  $\alpha 5\beta 3\gamma 2$  receptors. The slight reduction in the maximal diazepam stimulation in the presence of 1µM XLi093 might have been due to the previously described reduction in current stimulation at high diazepam concentration<sup>109</sup> or might indicate interaction of XLi093 via two different binding sites. Importantly, bivalent ligand XLi093 was able to dose dependently and completely inhibit diazepam stimulated currents in  $\alpha 5\beta 3\gamma 2$ receptors. It was a quite selective benzodiazepine receptor site antagonist at  $\alpha 5$ receptors. As noted, ligands such as XLi093 have been important in the determination of which physiological function(s) are subserved by this GABA<sub>A</sub>  $\alpha 5\beta 3\gamma 2$  subtype. There are many scientists who have reversed the amnesic effects of diazepam in vivo with XLi-093 in rodents and primates by Savic, Rowlett and Platt et al (see below).

# 4.1. Binding Studies of XLi-093 (33) in vivo by Rowlett and Cook et al.<sup>55</sup>

Recent studies by Rowlett and Cook et al. <sup>55</sup> have shown that diazepam significantly impaired performance in a hole board memory task, in which mice learned the location of 4 baited holes (out of 16 possible) over the course of 4 trials.



**Figure 18**: *In vivo* binding of Xli-093. The memory-impairing effects of diazepam were blocked (**Figure 18**) by a dose of (**33**; 3.0 mg/kg) that occupied ~66% of  $\alpha$ 5 GABA(A) binding sites but less than 25% (non-significant) of the  $\alpha$ 1 GABA(A) and  $\alpha$ 2/ $\alpha$ 3GABA(A) binding sites. Memory impairment measured via the holeboard task occurred at a dose of diazepam that resulted in a relatively high degree of  $\alpha$ 5 GABA(A) binding site occupancy (64%). Interestingly, even at a saturating dose of diazepam (10 mg/kg), XLi-093 (**33**) was still able to block the memory impairing effects of diazepam.



**Figure 19**: XLi-093 (3.0 mg/kg) blocked the DZ-induced suppression of performance ratio on the holeboard task. Note that XLi-093 showed no effect when administered with vehicle.

This performance impairment suggested that diazepam-induced a working memory deficit. The novel antagonist XLi-093 (**33**) was able to antagonize this deficit and showed appreciable subtype selectivity for the  $\alpha$ 5 GABAA receptor *in vivo* consistent with previous findings. *In vivo* binding studies with XLi-093 showed negligible occupancy of  $\alpha$ 1GABA(A) and  $\alpha$ 2/3GABA(A) binding sites; however, occupancy of  $\alpha$ 5GABA(A) binding sites was significant at 1.0 mg/kg and above (see Figure 18).

The data suggests  $\alpha$ 5 GABA(A) receptors play an important role in diazepam-induced memory impairments. Identification of compounds with reduced  $\alpha$ 5 GABA(A) activity may result in sedative-anxiolytics that lack memory-impairing effects of classical benzodiazepines.

*In vivo* binding studies (see Table 7) with XLi-093 (**33**) showed negligible occupancy of  $\alpha$ 1 GABA<sub>A</sub> and  $\alpha$ 2/3GABA<sub>A</sub> binding sites; however, occupancy of  $\alpha$ 5 GABA(A) binding sites was significant at 1.0 mg/kg and above. The ability of a novel  $\alpha$ 5 GABA(A) subtype selective antagonist, XLi-093 (**33**; 0.3-10.0mg/kg), to occupy  $\alpha$ 5 GABA receptors *in vivo* (*in vivo* [3H]L-655,708 binding) had been evaluated. Based on percent binding occupancy, the ability of a range of XLi-093 doses to reverse impairments induced by the minimal effective dose of diazepam (DZ (3.0 mg/kg)) was assessed. Ligand XLi-093 (**33**) reversed DZ-induced impairments at doses of 1.0 and 3.0 mg/kg, which occupied ~50-65% of [3H]L-655,708 binding. In contrast, these doses resulted in only ~20% occupancy of  $\alpha$ 1 GABA and  $\alpha$ 2/3 GABA(A) receptors in cerebellum and spinal cord, respectively, (determined by *in vivo* [3H]flumazenil binding). These findings are consistent with a key role for  $\alpha$ 5GABA(A) receptors in BZ agonist induced impairments in learning and memory.

**Table 7**: Percent of  $\alpha$ 5GABA (A) binding sites occupied by XLi-093 and DZ at doses used in the behavioral studies.

Ligand (Dose, i.p.)	% α5GABAA Occupancy
XLi (3.0)	65.8
DZ (1.0)	63.8
DZ(3.0)	83.5
DZ (10.0)	99

It had been demonstrated that bivalent ligand XLi-093 bound tightly to  $\alpha 5\beta 3\gamma 2$  receptor subtypes illustrated that bivalent ligands will bind to BzR and since the second unit of this bivalent ligand presumably lies in the extracellular domain of the BzR, this ligand had very poor or no affinity at the other subtypes. It is now obvious that this is the first  $\alpha 5$  bivalent ligand which was not only orally active, but also passed through the blood brain barrier. This was an important result for these bivalent ligands in regard to use in science.



Scheme 5: Synthesis of Bivalent Ligands XLi-356 and XLi-210

The bivalent ligand XLi356 (**34**) was obtained from XLi093 (**33**) in 99% yield *via* catalytic hydrogenation (Pd/C, H<sub>2</sub>). While XLi093 was found to be an antagonist, ligand XLi356 is an inverse agonist in  $\alpha$ 5 $\beta$ 3 $\gamma$ 2 oocytes and a weak agonist at  $\alpha$ 1 $\beta$ 3 $\gamma$ 2 sites.



**Figure 20**: Subtype selective efficacy of XLi356 (34), Dose response curves for XLi356 in oocytes expressing different subunit combinations of GABA<sub>A</sub> receptors. Subtype combinations are indicated in legends. cRNA–injected Xenopus oocytes were held at - 60 mV under two-electrode voltage clamp by Sieghart et al.

**Table 8**. In Vitro Binding affinities of bivalent ligands at GABA<sub>A</sub>/BzR subtypes.<sup>a</sup>Values Reported are in nM.

Ligand	α1	α2	α3	α4	α5	α6
XLi356 (34)	1851.5	4202.5	8545	ND	100.5	5000
XLi210 (35)	231	661	2666	ND	5.4	54.22
XLi093 (33)	>1000	>1000	858	1550	15	>2000

<sup>a</sup>K<sub>i</sub> values represent the mean of two determinations which differed by less than 10%. Data were generated using Ltk<sup>-</sup> cell membranes expressing human  $\alpha x \beta 3\gamma 2$  receptors.

DeLorey has recently shown that XLi-356 (**34**) does potently reverse scoploamine induced memory deficits in rodents. XLi356 was found to reverse scopolamine induced memory deficits in mice. When XLi356 was looked at in audio cued fear conditioning, the results show no activity. This suggests that the effect of XLi356 is selective through  $\alpha$ 5 receptors which are abundant in the hippocampus which is highly associated with contextual memory. Audio cued memory instead is amygdale-based, and is not affected by an  $\alpha$ 5 subtype selective compound.<sup>110</sup>

**Contextual Memory** 





Figure 21: Visual and audio que data for XLi356

As illustrated in Figure 21, scopolamine (1 mg/kg) reduced freezing (ie. impairs memory) typically caused by pairing the context (the cage) with a shock. The drug XLi356 when given at 10mg/kg attenuates the impairment of memory returning the freezing to the level that one typically sees the mouse freeze at (i.e. veh). In the audio cued paradigm, memory was triggered by sound not the context. XLi356 was not able to reverse this type of memory effect which is amygdale driven.

The biology of a bivalent ligand with a 3-carbon linker XLi-093 (**33**) led us to synthesize the analogous bivalent ligand with a 5-carbon linker. The synthesis of XLi-210 (**35**) was done analogous to the conditions employed for XLi-093. The acid **32** was subjected to a standard CDI-mediated coupling reaction using a 5-carbon linker to furnish bivalent ligand XLi210 **35** in 60% yield in dry DMF (Scheme 5).



**Figure 22**: Subtype selective efficacy of XLi210 (**35**): dose response curves for XLi210 in oocytes expressing different subunit combinations of GABA<sub>A</sub> receptors. Subtype combinations are indicated in legends. cRNA–injected Xenopus oocytes were held at – 60 mV under two-electrode voltage clamp.

This bivalent ligand (**35**) showed 50 fold selectivity at  $\alpha$ 5 over  $\alpha$ 1 (see Table 8). The efficacy data shows that the ligand is an agonist. This ligand should have properties exactly opposite that of PWZ-029 which is an inverse agonist (NAM). Positive controls and an  $\alpha$ 5 antagonist render this a very strong approach to hippocampal-mediated drug design in regard to the development of agents to treat cognition deficits in Alzheimer's and age-related memory deficits.

**Table 9.** EC<sub>50</sub> and Hill Slope for Modulation of Control GABA EC<sub>3</sub> by **XLi 210 (35)** in Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Results Represent EC<sub>50</sub> (95% Confidence Interval for EC<sub>50</sub>) and Hill Slope (Nh)  $\pm$  SD from at Least Four Cells from Two Batches.

		α1β3γ2	α2β3γ2	$\alpha 3\beta 3\gamma 2$	α5β3γ2
	EC <sub>50</sub>	262nM	>324nM	175nM	
XLi-210	95%CI	(15-4543)		(7-4210)	
	Nh	0.89		0.94	

**Table 10.** Modulation of Control GABA  $EC_3$  by XLi 210 (**35**) on Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Values are Presented as Mean  $\pm$  SEM of at Least Four Oocytes from at Least Two Batches.

		α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
	10nM	ns	ns	ns	ns
XLi210 (35)	100nM	115±6 %	117±7%	121	110±3%
	1µM	143±18 %	133±9%	150±25%	111±2%
	10µM	153±20%	144±4%	156±19%	123±6%

All the above values were significantly different from the respective control currents (p<0.01, Student's*t*-test) except where indicated (n.s.). n.s. means not significant.

## 5. Synthesis of Bivalent Ligands with Oxygen Linkers

In order to enhance water solubility and potency, bivalent ligands with a oxygen atom in the linker were synthesized. The insertion of an oxygen atom into the linker might increase the water solubility and hence enhance the molecular hydrophilicity which should play an important role in the pharmacokinetic properties of the ligand.

Scheme 6: Synthesis of Bivalent Ligands with Oxygen Linkers XLi-347, XLi-374 and SR-II-35



The oxygen linked analog XLi 347 (**36**) and SR-II-35 (**42**) were synthesized analogous to the conditions employed for the preparation of XLi-093 (**33**), the synthesis of the 5-carbon atom linked dimer **35** (XLi210) had been was carried out. The acid (**32**) available

from Scheme 4, was subjected to a standard CDI-mediated coupling reaction to furnish bivalent ligands XLi 347 (**36**) and SR-II-35 (**39**) in 60% yield in dry DMF (see Scheme 6).

For the synthesis of XLi 374 (39), the preparation of the required oxygen-substituted linker (a dibromodimethylether) was based on the method of Tischtschenki and Rabcevitsch-Zubkovski<sup>111,112</sup>. Bromine was added slowly to water over a four hour period, to which a mixture of paraformaldehyde (37) and red phosphorus was also added at a comparable rate. The reaction mixture was ice-water-cooled during this process. The temperature was maintained at 0°C. The final reaction mixture was allowed to stand for 20 hours. The heavy crude dibromo ether layer was then separated, saturated with hydrogen bromide, and dried over P<sub>2</sub>O<sub>5</sub> for one day. Distillation at 158°C at 1 atm provided the ether from one fraction as pure dibromodimethyl ether (a colorless liquid). With the dibromodimethyl ether in hand, acid 32 was dissolved in dry dimethylformamide (DMF). To this solution the dibromodimethyl ether (38) was added, and this was followed by addition of TEA. After the mixture stirred for 26 hours at ambient temperature, the solvent was removed under reduced pressure and the residue was distributed between CH<sub>2</sub>Cl<sub>2</sub> and water. The organic layer was washed multiple times with water and dried over Na<sub>2</sub>SO<sub>4</sub>, and the solvent was removed under reduced pressure almost to dryness. Subsequent flash chromatography of the crude oil provided the bivalent oxygen (C-O-C) dimer XLi374 (**39**) as a white solid.

The nature of the functional groups in a ligand plays an important role in receptor binding, of course, as well as the conformation in solution. The more the information about the stable conformation(s) of molecules, the better is the understanding of the structure-activity relationships. It was essential from the beginning of the present study to determine the conformation of these bivalent ligands, which contained 3 to 5 atom linkers, since the steric requirements for affinity to the Bz receptor must be satisfied. In traditional medicinal chemistry, computer assisted molecular modeling programs and X-ray analysis contribute greatly in the search for stable conformations. However, some problems with these methods should not be neglected. Using computer modeling to determine the stable conformation of molecules containing many freely rotating bonds, such as those contained in bivalent ligands is difficult. Although X-ray crystallography is the ultimate arbiter of chemical structure, it has many limitations beyond the obvious need for crystals: it often does not reflect accurately the conformation is crucial in drug design. However, NMR spectroscopy is a powerful technique in drug discovery and its role in conformational analysis cannot be surpassed by other spectroscopic methods.<sup>42,113</sup>

#### **5.1.** Low Temperature NMR studies and X-ray Analysis

Herein is described a method utilizing low temperature NMR for the determination of the solution stable conformation of a series of GABA<sub>A</sub>-benzodiazepine bivalent ligands with different monomeric units and linkers. The conformations in solution were determined by NMR spectroscopy and compared with those in the crystal structure. The combination of low temperature NMR and X-ray analysis provided accurate structural information required for understanding structure-activity relationships and drug design. The influence of the molecular structure of the linker on the conformation is also discussed.

# 5.2. Results

Recently it was shown the active dimers XLi093 (**33**) and DMH-D-053 (**31**) existed in a linear conformation in the solid state while dimer dm-III-96 (**43**) with a oxygen containing linker folded back upon itself, as illustrated in Figure 23.<sup>114</sup>

Since the bioactive conformation in solution may or may not parallel that in the crystal structure, the lowest energy solution structure must be established in order to correlate conformation with biological activity. Thus, NMR experiments at variable temperatures were performed and data were collected in different solvents. In methylene chloride or chloroform at room temperature, only a single set of signals was detected for both bivalent ligands **33** and **42**.

**Table 11**. In Vitro binding affinities of bivalent ligands at GABA<sub>A</sub>/BzR subtypes.<sup>a</sup>Values Reported are in nM.

Ligand	α1	α2	α3	α4	α5	α6
RY-080 (23)	28.4	21.4	25.8	5.3	0.49	28.8
XLi-093 ( <b>33</b> )	>1000	>1000	858	1550	15	>2000
XLi-374 ( <b>39</b> )	3795	2694	1864	ND	76.14	ND
XLi-347 ( <b>36</b> )	828.05	690.2	ND	ND	13.87	ND
SR-II-35 ( <b>42</b> )	200.7	47.03	94.66	ND	38.38	ND
XHe-II-053 ( <b>30</b> )	287	45	96	1504	13.8	1000
dm-III-96 ( <b>43</b> )	460	5000	ND	ND	5000	5000
DMH-D-053 ( <b>31</b> )	236	7.4	272	>5000	194.2	>5000



Figure 23. Crystal structures of XLi-093, DMH-D-053, and DM-III-96.

At low temperature, it was found that the linear dimer **33** exhibited only a small splitting of about 3 Hz for some of the aromatic protons in the <sup>1</sup>HNMR spectra( see Figure 24),<sup>114</sup> while two clearly separated sets of signals were observed for the folded dimer **43** (Figure 25). For example, as seen in Figure 25 for **43**, the signal of <sup>1</sup>H (7.92 ppm at 298 K) was split into two peaks at  $\delta$ 7.91 and 7.88 ppm, respectively, at 273° K. The doubling of the signals is consistent with disruption of the symmetry between the two domains of the molecule as expected if **43** adopted a static folded structure similar to the crystalline state.



Ligand	Monounit 1	Monounit 2	Linker	Conformation in solution	Crystal structure
<b>32,</b> RY-080	А	-	-	-	-
<b>33</b> , XLi-093	А	А	(CH <sub>2</sub> ) <sub>3</sub>	linear	linear
<b>35</b> , XLi -210	А	А	(CH <sub>2</sub> ) <sub>5</sub>	linear	ND
<b>36</b> , XLi -347	А	А	$(CH_2)_2O(CH_2)_2$	folded	ND
<b>39</b> , XLi-374	А	А	CH <sub>2</sub> OCH <sub>2</sub>	folded	ND
<b>30,</b> XHe-II-053	В	-	-	-	-
<b>31,</b> DMH-D-053	В	В	(CH <sub>2</sub> ) <sub>3</sub>	linear	linear
<b>43</b> , dm-III-96	В	В	$(CH_2)_2O(CH_2)_2$	folded	folded
<b>42,</b> SR-II-35	С	С	CH <sub>2</sub> OCH <sub>2</sub>	folded	ND



**Figure 24.** Aromatic region of <sup>1</sup>HNMR spectra of XLi-093 (**33**) in CD<sub>2</sub>Cl<sub>2</sub> at variable temperatures.



**Figure 25.** Aromatic region of <sup>1</sup>HNMR spectra of SR-II-35 (**42**) in  $CD_2Cl_2$  at variable temperatures.

However, the possibility could not be ruled out that the split in the signals was caused by slowing a dynamic process within each domain, such as conformational interconversion of the seven-membered ring.



**Figure 26.** Aliphatic region of <sup>1</sup>HNMR spectra of SR-II-35 (42) in  $CD_2Cl_2$  at variable temperatures.

The analysis of these data indicated that the line broadening at the lowest temperature was due to one of the conformational processes mentioned above, whereas the doubling of the peaks in **42** was caused by the presence of two domains. Certainly an interdomain

interaction existed between the two heterocyclic units of **42**. It was therefore concluded the internal mobility of the molecule decreased as expected, when the temperature was lowered which permitted observation of the two sets of signals of **42** on the NMR time scale. It was thus suggested that only when the molecule preferred the folded conformation in solution were two sets of signals observed. The preferred conformations of the molecules in CDCl<sub>3</sub> and CD<sub>2</sub>Cl<sub>2</sub> correlated quite well with those observed in the crystal structures (Table 12).

The study was then expanded by varying the nature of the linker and monomer. It was found that **31**, **33**, **34**, and **35** exhibited only one set of NMR signals at low temperature, whereas the NMR signals of ligands **43** and **42** split into two sets at low temperature. Low temperature NMR studies were performed in  $CD_2Cl_2$ . It was concluded that **43** and **42** preferred a folded conformation, while **32**, **33**, **34**, and **35** assumed a linear conformation. These conclusions are supported by a crystal structure obtained for bivalent ligand **33** which indicated **33** was present in a linear conformation in the solid state. These results are illustrated in Table 12.

Attempts to run the NMR experiments in water failed since the ligands were not sufficiently soluble in  $D_2O$ . However, the spectra of both the linear and folded dimer **33** and **42** respectively could be carried out in MeOH- $d_4$ . The solvating properties of methanol, of course, more closely resemble those of water and this more closely mimics aqueous physiological conditions as well. The conformations of dimers in MeOH- $d_4$  were consistent with conformations in hydrophobic solvents.

### **5.3.** Discussion

From the results described above, it is clear that dimers which contain an oxygen atom in the linker tend to adopt a folded conformation. Analysis of these data indicated in the hydrophilic solvent, **43** also had a higher tendency to fold back upon itself than **33**, as it did in the hydrophobic media (Figure 23). In fact, the tendency of **43** to assume a folded structure appeared to be higher in methanol than in  $CD_2Cl_2$  or  $CDCl_3$ , as the free rotation of the molecule was limited (Figure 25). On the other hand, for dimer **33**, which preferred a linear structure in the solid state and in lipophilic solvents, only one set of signals was observed at room temperature in MeOH- $d_4$  (Figure 27).

It is interesting to note that in the preparation of the samples, dimer 33 was less soluble in MeOH- $d_4$  than was 42 and 43. It has been established that ligands are easier to dissolve in a solvent when the ligands surface energy can be minimized. Bivalent ligands 42 and 43 were more polar than dimer 33 and had a higher tendency to fold back. Consequently, 42 was presumably, easier to dissolve in methanol because its surface energy was minimized. On the other hand, when ligand 33 was dissolved in a more polar solvent such as methanol, the surface energy may be forced to be minimized.



**Figure 27.** Aromatic region of <sup>1</sup>H NMR spectra of XLi-093 (**33**) in MeOH-*d4* at different temperatures.

Since the conformation of molecules XLi-093 (**33**) and SR-II-35 (**42**) in methanol agree with those in  $CD_2Cl_2$  or  $CDCl_3$ , the behavior of these ligands in  $CD_2Cl_2$  or  $CDCl_3$  should reflect those in aqueous solution. The stable conformation of the compounds determined in  $CD_2Cl_2$  or  $CDCl_3$  were correlated with the newly generated receptor binding data (Table 11). In the pharmacophore/receptor model, the bivalent ligands in the linear conformation align well (Figure 28).



**Figure 28.** Bivalent ligand XLi093 (**33**) aligned in the included volume of the pharmacophore/receptor model for the  $\alpha 5\beta 3\gamma 2$  subtype.

Importantly, bivalent ligands **31**, **33**, **34**, and **35** with carbon only linkers preferred the linear conformation as the stable conformation, independent of the number of linker atoms. In contrast, replacing the middle carbon of either linker (CH<sub>2</sub>)<sub>3</sub> or (CH<sub>2</sub>)<sub>5</sub> with an oxygen atom altered the stable conformation of the molecules **42** and **43** from linear to folded. The only difference between bivalent pairs **33** and **39** with linear and folded conformations as the stable ones, respectively, was the center atom. In compound **33** (XLi-093) the middle atom was carbon, while in **39** (XLi-374) oxygen was present. Consequently, it was decided to focus attention on the conformational difference between carbon and oxygen containing linkers.

It was well-known<sup>115,116</sup> that the carbon chain in both small molecules and polymers favored the *anti*-conformation which results in a linear arrangement of atoms. From examination of the linkers (Figure 29), it was easily seen that linkers **C** and **D** can be regarded as oligomers of oxymethylene (OCH<sub>2</sub>)<sub>2</sub> and oxyethylene (OCH<sub>2</sub>CH<sub>2</sub>)<sub>2</sub>.



**Figure 29.** The linkers of the bivalent ligands.

It was well documented that the preference for the *gauche-gauche* conformation<sup>117-119</sup> of a simple open-chain acetal such as dimethoxymethane (CH<sub>3</sub>OCH<sub>2</sub>OCH<sub>3</sub>) could be predicted on the basis of the anomeric effect and related stereoelectronic effects. In this conformation the polar X–O bonds are favorably oriented such that a lone pair orbital of the oxygen atom was almost antiperiplanar to the X–O bond (Figure 30, lone pair orbital and X–O bond in same color). This permitted maximum overlap of the n orbital of the oxygen atom with the  $\sigma^*$  orbital of the X–O bond. This was not possible in the *anti-anti* or *gauche-trans* conformations and the two rabbit-ear interactions<sup>117-121</sup> (Figure 30) engendered by each pair of adjacent oxygen atoms in the *anti-anti* conformer are avoided.



Figure 30. The conformations for dimethoxymethane.

Furthermore, in the related polymer of dimethoxymethane with the two-bond repeating sequence, poly(oxymethylene) (**POM**), the *gauche* conformation was, in fact, markedly preferred over trans and the polymer existed in a helical (all *gauche*) conformation<sup>116,122-</sup><sup>124</sup> rather than in the all-*anti* one.

Similarly, much effort has been spent on the investigation of the conformational characteristics of 1,2-dimethoxyethane (glyme, CH<sub>3</sub>OCH<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub>) as a model molecule for understanding the conformations of poly(oxyethylene), **POE**. It had long been established that POE chains have a large fraction of bonds in *gauche* conformation and assumed a helical conformation overall.<sup>115,116,122-124</sup> It has been proposed that the oxygen gauche effect<sup>116,125,126</sup> and 1, 5-CHO interaction<sup>127-129</sup> within the molecule were responsible for the *gauche*-rich conformations.

Based on this pioneering work, the correlation was made that the conformation of the linkers in the bivalents **33-39** and **42-43** adopt *anti* (**B**), *gauche-gauche* (**C**), *trans*-

*gauche-trans* (**D**) conformations, respectively, regardless of the monomeric units which comprise them. The arrangement in space (disregarding the direction) of every unit in the linkers is depicted in Figure 31. The end-to-end distance of each unit (**C** and **D**) which adopted the *gauche* conformation was shorter than the one in the *anti*-conformer. The more units in the linker, the shorter the *gauche* linker becomes. Moreover, it was recognized that the X–O bond length (1.43Å) is often appreciably shorter than the C-C bond (1.54 Å).<sup>130</sup> Therefore, it was believed that the linkers with the oxygen atom in the middle favored the helical conformation and rendered the two monomeric units in each dimer sufficiently close to each other with suitable dihedral angles to facilitate the intramolecular lipophilic-lipophilic (aromatic-aromatic) interaction. This was regarded as one of the most important factors to stabilize the folded structure as the preferred conformation.<sup>131-135</sup> For these same reasons, bivalent ligands with the linker B adopted the linear conformation.



Figure 31. Newman projection for linkers B, C and D.

For the higher analog of **POM** and **POE**, namely, poly(trimethylene oxide)  $[(CH_2)_3O]_x$ (**POM**<sub>3</sub>), *trans-gauche-gauche-trans* was slightly preferred over all-*trans* and *trans-* *trans-gauche-trans* comformers in the crystalline state.<sup>116</sup> The preference for the *gauche* state in this case was only 0.2 kcal/mol were as in the **POM** and **POE** examples was 1.5 and 0.4 kcal/mol, respectively.<sup>122</sup> Since the energy difference between *gauche* and *trans* was low, the linker **A** had more flexibility than the other linkers (**C** and **D**) to rotate freely and less tendency to occur as gauche. This could lead to improper end-to-end distances or dihedral angles for the interaction between the aromatic monounits. Hence dimers connected with linker **A** could not be stabilized in the folded conformation even though the same aromatic monomeric units were contained in the molecules.

Bivalent ligands were evaluated in competition binding assays for specific GABA<sub>A</sub> membrane proteins using [<sup>3</sup>H] flunitrazepam as the radiolabel. This assay measured the ability of the ligand to displace flunitrazepam. Data is reported as K<sub>i</sub> according to the Cheng-Prusoff equation.<sup>136</sup> Biological data are presented in Table 11 for dimers synthesized using different spanners (or linkers). The linkers in **42** (SR-II-35) and **43** (dm-III-96) contained an oxygen atom. Compounds **31**, **33**, **34**, and **35** contained all carbon linkers. Binding data indicated decreased affinity when bivalents contained an oxygen atom in the linker. Bivalent Ligand **33** (XLi-093) and **35** (XLi-210) showed increased selectivity for the  $\alpha$ 5 subtype as compared to parent monomer **23**. Bivalent ligand **39** (XLi-374) which was analogous to **33** (XLi-093) with the exception of the oxygen atom present in the linker, bound with less affinity at the  $\alpha$ 5 subtype. Likewise bivalent ligand **31** (DMH-D-053) showed increased selectivity versus monomer **30** (XHe-II-53). However, the bivalent **43** (dm-III-96) containing an oxygen atom in the linker did not bind. The data suggests that dimers which contain a single oxygen atom

in the linker bind less (decreased affinity) to the Bz/GABA (A) receptor. This is due to their propensity to adopt a folded conformation. One strategy to increase hydrophilicity and avoid a folded conformation is to extend the linker length and insert two opposing oxygen atoms. This approach is currently underway.

Comparison of the results of low temperature NMR studies to crystal structures has provided enough information to demonstrate that low temperature NMR can be used as a quick method to identify dimeric ligands with a tendency to fold back upon themselves, as compared with those preferring a linear conformation. A correlation with binding data shows that the suitability of a ligand in the  $\alpha$ 5 BzR/Gabaergic subtype is heavily influenced by its conformation in solution. Variable temperature NMR thus can be used as a tool for screening bivalent ligands for their *in vivo* suitability. It was also clear the presence of one oxygen atom in the linker was the principle cause for the dimer to fold back onto itself.

# 6. Synthesis of $\alpha$ 5 subtype selective inverse agonist PWZ-029 on multigram scale

PWZ-029 (**47**) was effective in improving memory and cognition, especially in models with baseline cognitive deficits. The high specificity of PWZ-029 (**47**) for  $\alpha$ 5-containing GABA<sub>A</sub> receptors has successfully mitigated adverse side effects of seizures and sedation, which are commonly observed with typical benzodiazepines (BZ) and nonselective inverse agonists of the GABA<sub>A</sub> complex. The novel ligand PWZ-029, which was synthesized on large scale and characterized electrophysiologically, possesses *in vitro* binding selectivity and moderate inverse agonist functional selectivity
at  $\alpha$ 5-containing GABA<sub>A</sub> receptors. This ligand has also been examined in rats in the passive and active avoidance, spontaneous locomotor activity, elevated plus maze and grip strength tests, primarily predictive of the effects on the memory acquisition, basal locomotor activity, anxiety level and muscle tone, respectively. The improvement of task learning was detected at the dose of 5 mg/kg in the passive, but not the active avoidance test. The inverse agonist PWZ-029 had no effect on anxiety nor muscle tone, whereas at higher doses (10 and 20 mg/kg) it decreased locomotor activity in this one paradigm. This effect was antagonized by flumazenil and also by the lower (but not the higher) dose of an agonist (SH-053-2'F-R-CH<sub>3</sub>) selective for GABA<sub>A</sub> receptors containing the  $\alpha_5$  subunit. The hypolocomotor effect of PWZ-029 was not antagonized by the antagonist  $\beta$ -CCt exhibiting a preferential affinity for  $\alpha_1$ -subunit containing receptors. These data suggest that moderate negative modulation at GABA<sub>A</sub> receptors containing the  $\alpha 5$  subunit is a sufficient condition for eliciting enhanced encoding/consolidation of declarative memory, while the influence of higher doses of modulators at these receptors on motor activity shows an intricate pattern whose relevance and mechanism wait to be defined.

One desirable property in this regard is the pro-amnesic activity of BZ site inverse agonists, repeatedly reported in animal models,<sup>137,138</sup> as well as in human volunteers.<sup>139,140</sup> However, this desirable effect is confounded by different concomitant psychomotor effects (increased vigilance, anxiogenic and/or proconvulsant state), some of which have been described in memory studies with non-selective inverse agonists in humans, urging their early termination.<sup>139</sup> Point mutated mice could not be used to identify the receptor subtypes mediating the promnesic activity of inverse agonists

because an unexplained switch to the agonist mode of action occurs when an inverse agonist at wild type diazepam-sensitive recombinant GABA<sub>A</sub> receptors is tested at the respective point-mutated receptors.<sup>141,142</sup> Thus, inverse agonists exerted agonistic-like sedative and anticonvulsant effects in mice with the point-mutated  $\alpha 1$  subunits,<sup>142</sup> while the corresponding experiments in models of learning and memory were not performed.<sup>142</sup> Nevertheless, behavioral examination of genetically modified animals conducted to date has indicated the  $\alpha_1$  and  $\alpha_5$  subunit-containing GABA<sub>A</sub> receptors comprise the 'memory-modulating' population of these receptors.<sup>143,144</sup> It is notable that GABA<sub>A</sub> receptors containing the  $\alpha 5$  subunit are abundantly expressed in the hippocampus,<sup>11,145,146</sup> the structure substantially involved in memory formation.<sup>147</sup> Recent evidence from animal studies with affinity-selective<sup>148</sup> or efficacy-selective ligands<sup>149-151</sup> has confirmed that the  $\alpha_5$  subunit was significantly involved in cognition enhancement mediated by the negative modulation of GABA<sub>A</sub> receptor functions. Moreover, it was shown in humans that pre-treatment with an  $\alpha 5$  efficacy-selective inverse agonist significantly reduces the amnesic effect of alcohol on learning a word However, the affinity- or efficacy-selectivity of the ligands, as well as the list.<sup>152</sup> diversity of the behavioral tasks used in their characterization to date, were of limited extent, which necessitated screening of newer BZ site negative modulatory ligands, to determine the putative therapeutic role of such compounds in various disorders with diminished cognitive capabilities in humans.<sup>56</sup>

In this regard, the BZ site ligand PWZ-029 was synthesized (Scheme 7) and prepared on large scale. A mixture of 5-chloroisatoic anhydride **44** and sarcosine in DMSO was heated to reflux to furnish benzodiazepine **45** in 84% yield. Potassium *t*-butoxide was

added to a solution of **45** in THF at 0 °C and stirred for 20 min. The reaction was cooled to  $-35^{\circ}$ C and diethylchlorophosphate was added slowly. After stirring at 0°C for 30 min, the reaction mixture was cooled to  $-78^{\circ}$ C and ethyl isocyanoacetate was added followed by potassium *t*-butoxide to form **40** in 89% yield. A solution of imidazobenzodiazepine **40** can be reduced with LiBH<sub>4</sub> to furnish alcohol **46** in 67% yield as colorless crystals. To a slurry of KOH in DMSO at room temperature were added alcohol **46** and CH<sub>3</sub>I to give PWZ-029 (**47**), as an off-white powder in 95% yield on 1-15 gram scale. The alcohol was dissolved in a solution of KOH in DMSO and treated with CH<sub>3</sub>CH<sub>2</sub>I at room temperature to give SR-II-81 (**48**).

### Scheme 7: Synthesis of PWZ-029



Electrophysiology experiments in *Xenopus* oocytes demonstrated that PWZ-029, at a concentration of 1  $\mu$ M, significantly reduced GABA initiated control currents, thereby indicating a partial inverse agonist effect of PWZ-029 on GABA<sub>A</sub> receptors possessing the  $\alpha$ 5 isoforms (Figure 32). Illustrated in Figure 32, are the concentration–effects curve

for modulation of GABA<sub>A</sub> elicited currents by PWZ-029 on *Xenopus* oocytes expressing GABA<sub>A</sub> receptor subtypes  $\alpha 1\beta 3\gamma 2$ ,  $\alpha 2\beta 3\gamma 2$ ,  $\alpha 3\beta 3\gamma 2$ , and  $\alpha 5\beta 3\gamma 2$ . Concentrations of GABA<sub>A</sub> that elicit 3% of the maximum GABA<sub>A</sub>-triggered current of the respective cells were applied alone and with various concentrations of PWZ-029. Control currents represent responses in the absence of PWZ-029. Data points represent means±SEM from 4 oocytes from  $\geq 2$  batches [1 µMPWZ-029 resulted in 114±4%, 105±8%, 118±5% and 80±4% of control current (at GABA<sub>A</sub> EC3) in  $\alpha 1\beta 3\gamma 2$ ,  $\alpha 2\beta 3\gamma 2$ ,  $\alpha 3\beta 3\gamma 2$ , and  $\alpha 5\beta 3\gamma 2$  receptors, respectively]. All these values except the one for  $\alpha 2\beta 3\gamma 2$  receptors were significantly different from that of the respective control currents (p<0.01, Student's t-test). Note that 100 nM is the pharmacologically relevant concentration.

Table	<b>13</b> .	In	Vitro	Binding	Affinities	of	bivalent	ligands	at	GABA <sub>A</sub> /BzR	subtypes.
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Code	MW	α1	α2	α3	α4	α5	α6
Merck	291.73	>300	>300	>300	ND	38.8	>300
Moltech	291.73	920	ND	ND	ND	30	ND
UNC-Roth	291.73	362.4	180.330	328.2	ND	6.185	ND

<sup>a</sup>Values Reported are in nM.



Figure 32. Oocyte data for PWZ-029 by Sieghart et al.<sup>153</sup>

This ligand as mentioned, has also been examined in rats in the passive and active avoidance, spontaneous locomotor activity, elevated plus maze and grip strength tests, primarily predictive of the effects on the memory acquisition, basal locomotor activity, anxiety level and muscle tone, respectively. The improvement of task learning was detected at the dose of 5 mg/kg in the passive avoidance task (Figure 33). The passive avoidance task is a one trial fear-motivated avoidance task in which the mouse learns to refrain from stepping through a door to an apparently safer but previously punished dark compartment. The latency to refrain from crossing into the punished compartment serves as an index of the ability to avoid, and allows memory to be assessed.



Figure 33. Fear conditioned contextual memory of PWZ-029

The PWZ-029 at 5 mg/kg, administered before the acquisition session, significantly increased retention session latency relative to the control group.<sup>154</sup>

The inverse agonist PWZ-029 (47) had no effect on anxiety or muscle tone, whereas at higher doses (10 and 20 mg/kg) it decreased locomotor activity. This effect was antagonized by flumazenil and also by the lower (but not the higher) dose of an agonist (SH-053-R-CH3-2'F) selective for GABA<sub>A</sub> receptors containing the  $\alpha$ 5 subunit. The hypolocomotor effect of PWZ- 029 (47) was not antagonized by the antagonist  $\beta$ CCt, which exhibits a preferential affinity for  $\alpha$ 1- subunit-containing receptors. These data suggest that moderate negative modulation at GABA<sub>A</sub> receptors containing the  $\alpha$ 5 subunit is a sufficient condition for eliciting enhanced encoding/consolidation of declarative memory, while the influence of higher doses of modulators at these receptors on motor activity shows an intricate pattern whose relevance and mechanism await to be defined.<sup>154</sup> In fact, in the conflict paradigm in Rowlett's lab in rhesus monkeys, PWZ-029 was a weak anxiolytic with no sedation observed in either the suppressed or nonsuppressed portion of the test. Recently, Stables et al. at NINDS has shown PWZ-029 to be a weak anticonvulsant. The PWZ-029 (**47**) an  $\alpha$ 5 inverse agonist cannot be proconvulsant or convulsant (see Appendix IV) and consequently will not exhibit the side effects of other inverse agonists (NAM).



**Figure 34.** Passive avoidance task<sup>154</sup>

The effects of DMCM (0.2 mg/kg) and PWZ-029 (2, 5 and 10 mg/kg) on retention performance in a passive avoidance task (\*p<0.05 compared to solvent (SOL) group). Number of animals per treatment group: 10.

Recent studies by Rowlett, Rallapalli and Cook et al. on PWZ-029 showed that this ligand acted as an inverse agonist at  $\alpha$ 5GABA<sub>A</sub> receptor subtypes which enhanced memory. Rowlett's used several primate models, including the Cambridge Neuropsychological Test Automated Battery, or CANTAB. A clear strength of the CANTAB approach is that its tasks are based on corresponding procedures developed for human patients, which greatly facilitate the ultimate goal of translating findings from monkeys to humans.

Because PWZ-029 has been shown to be safe and to have cognitive-enhancing effects in rodents<sup>154</sup> experiments were initiated with this compound in rhesus monkeys as mentioned. Initial results, shown in Figure 35, are quite provocative and exciting: PWZ-029 enhanced performance in the DNMS (Delayed non-matching to sample) to task using the 10-min delay with distracters (Figure 35A) and although it had no effect alone, PWZ-029 completely reversed the scopolamine-induced deficit in the ORD (Object Retrieval with Detours ) task (Figure 35B). Interestingly, PWZ-029 induced anti-conflict (anxiolytic) effects in monkeys without the concomitant response rate-suppressing effects characteristic of BZ-type drugs (Figure 35C). It is felt that these latter effects are due to PWZ-029's partial agonist effects at  $\alpha 2/3GABA_A$  receptor subtypes, whereas the cognition-enhancing effects of this compound are due to inverse agonist (NAM) action at  $\alpha 5GABA_A$  receptors.

PWZ-029 (47) provides an important lead compound based not only on potential effectiveness against cognitive impairment, but also as a potential treatment for anxiety and agitation in patients with dementias, similar to the anxiolytic effects of classical BZs

(Meehan et al. 2002). Studies will continue with PWZ-029 and expand them to include comparative studies with non-selective agonists and  $\alpha$ 5GABA<sub>A</sub>-preferring agonists; as well as with pre-treatments with  $\alpha$ 5GABA<sub>A</sub>-preferring antagonists such as XLi-093 (**33**). In addition, in our initial screens compounds will be sought with similar, and hopefully improved, efficacy profiles. For example, a compound that is an  $\alpha$ 5GABA<sub>A</sub> partial inverse agonist, an  $\alpha$ 2/3GABA<sub>A</sub>-preferring partial agonist, but ineffective at  $\alpha$ 1GABA<sub>A</sub> receptors is both desirable and feasible in our synthesis/screening program.



**Figure 35:** Preliminary profile of cognitive enhancing and anxiolytic like effects of PWZ-029 in monkeys. Data are from n=2 monkeys (A &C) and n=4 monkeys (B). Red arrow indicates the dose (0.01mg/kg, i.v.) of PWZ-029 tested in A & B

Importantly, in the ORD task, PWZ-029 showed only a modest trend for enhancement of performance (Fig. 36A), but when task difficulty was increased by testing with difficult trials only, PWZ-029 robustly increased performance (Fig. 36B). This enhancement was reversed by administration of the  $\alpha$ 5GABA<sub>A</sub>-preferring antagonist XLi-093, and this antagonism, in turn, was reversed by increasing the dose of PWZ-029 (Fig. 36C). Finally, PWZ-029 completely reversed deficits in performance induced by the memory-impairing anti-cholinergic scopolamine (Fig. 36D).



**Figure 36.** Cognitive-enhancing effects of PWZ-029 in the rhesus monkey Object Retrieval with Detours (ORD) task (N=5 monkeys). **A.** Effects of PWZ-029 on ORD tests consisting of both easy and difficult trials .No effect of the compound was observed

(mean % correct first reaches, p>0.05, repeated measures ANOVA). **B.** PWZ-029 enhanced performance in the ORD task when tested with difficult trials only (\*p<0.05 vs. vehicle, V, Bonferroni t-tests). **C.** Enhancement of ORD performance by 0.018 mg/kg of PWZ-029 was attenuated by the  $\alpha$ 5GABA<sub>A</sub>-preferring antagonist XLi-093, and this antagonism was surmountable by increasing the PWZ-029 dose (\*p<0.05 vs. vehicle, \*\*p<0.05 vs. 0.018 mg/kg PWZ-029, Bonferroni t-tests). **D.** PWZ-029 reversed performance impaired by 0.01 mg/kg of scopolamine. (\*p<0.05 vs. vehicle; \*\*p<0.05 vs. scopolamine alone, Bonferroni t-tests).

## 6.2. Synthesis of PWZ-029 analogs

The chemistry used to design fluoro analogs of PWZ-029 (**47**) can be adapted to design ligands of XHE-II-053 (**30**) with less potential for metabolization by esterases. Fluorine atoms in place of the carbonyl group satisfy current included volumes and electronegativity requirements of the Milwaukee-based pharmacophore receptor model. The covalent radius of a hydrogen is 37 pm whereas a fluorine is 71 pm.

This approach is in agreement by a recent compounds patented by Roche.<sup>156</sup> Several new analogs have been proposed and docked in the Milwaukee based pharmacophore by Clayton et al.<sup>157</sup>



Affinity for  $\alpha x \beta 3 \gamma 2$  (x = 1-6) benzodiazepine receptor isoforms

	Alpha 1	Alpha 2	Alpha 3	Alpha 4	Alpha 5	Alpha 6
Roche	174.3	185.4	79.6	ND	4.6	ND



Affinity of PWZ-029 for  $\alpha x\beta 3\gamma 2$  (x = 1-6) benzodiazepine receptor

	Alpha 1	Alpha 2	Alpha 3	Alpha 4	Alpha 5	Alpha 6
Merck	>300	>300	>300	ND	38.5	>300
Moltech	920	ND	ND	ND	30	ND
UNC-Roth	362.4	180.330	328.2	ND	6.185	ND

Figure 37: Binding affinity data of PWZ-029 (47) and Roche difluoro ligand 49 The new analogs of PWZ-029 are of strong interest due to the recent contextual memory impairment data wherein doses of scopolamine decreased memory while the  $\alpha$ 5 ligand reversed this (see Scheme 8).



Scheme 8: Synthesis of various analogs of PWZ029

The synthesis of amine analog of PWZ-029 can be seen in Scheme 9. A mixture of imidazobenzodiazepine was refluxed with methyl amine in anhydrous  $CH_2Cl_2$  to give SR-II-097 (57) in 78% yield. The efficacy data evaluated by Sieghart et al. (shown in

Figure 38) indicated that the ligand SR-II-097 (57) was an inverse agonist at  $\alpha$ 5 and was an agonist at  $\alpha$ 1 when compared to PWZ-029 (see Figure 38 and Table 15).



Scheme 9: Synthesis of SR-II-097

Figure 38: Oocyte data of SR-II-097

**Table 14.** EC<sub>50</sub> and Hill Slope for Modulation of Control GABA EC<sub>3</sub> by **SR-II-097 (57)** in Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Results represent EC<sub>50</sub> (95% confidence interval for EC<sub>50</sub>) and Hill Slope (Nh)  $\pm$  SD from at least four cells from two batches.

		α1β3γ2	α2 <i>β</i> 3γ2	α3 <i>β</i> 3γ2	α5 <i>β</i> 3γ2
	EC <sub>50</sub>	378nM	>6µM	>18µM	95nM
SR-II-097 ( <b>57</b> )	95%CI	(58-2426)			(0.01-6000)
	Nh	0.80			-3.5

**Table 15.** Modulation of Control GABA EC<sub>3</sub> by SR-II-097 (**57**) on Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Values are presented as mean  $\pm$  SEM of at least four oocytes from at least two batches.

		α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
	10nM	ns	ns	ns	ns
SR-II-097( <b>57</b> )	100nM	ns	ns	ns	ns
	$1 \mu M$	121±2 %	ns	ns	88±5%
	10µM	123±5%	119±1%	125±13%	88±4%

All the above values were significantly different from the respective control currents (p<0.01, Student's*t*-test) except where indicated (n.s.). n.s. means not significant. The fluoro analog SR-III-38 was prepared using DAST (diethylaminosulfur trifluoride) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> at room temperature in 70 % yield.

Scheme 10: Synthesis of SR-III-38



Examination of the oocyte data (shown in Figure 39) indicated no significant subtype selective efficacy.



Figure 39: Oocyte data on SR-III-38 by Sieghart et al.

**Table 16.** EC<sub>50</sub> and Hill Slope for Modulation of Control GABA EC<sub>3</sub> by **SR-III-38 (58)** in Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Results represent EC<sub>50</sub> (95% confidence interval for EC<sub>50</sub>) and Hill Slope (Nh)  $\pm$  SD from at least four cells from two batches.

		α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
	EC <sub>50</sub>	>0.6µM		>43µM	
SR-III-38 <b>(58)</b>	95%Cl				
	Nh	0.4		1.5	

**Table 17.** Modulation of Control GABA EC<sub>3</sub> by **SR-III-38 (58)** on Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Values are presented as mean  $\pm$  SEM of at least four oocytes from at least two batches.

	α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
10nM	ns	ns	ns	ns
100nM	105±1 %	ns	ns	ns
1µM	112±10 %	ns	105±3%	ns
10µM	115±7%	ns	115±17%	ns
	10nM 100nM 1μM 10μM	α1β3γ2   10nM ns   100nM 105±1 %   1µM 112±10 %   10µM 115±7%	α1β3γ2 α2β3γ2   10nM ns ns   100nM 105±1 % ns   1µM 112±10 % ns   10µM 115±7% ns	α1β3γ2 α2β3γ2 α3β3γ2   10nM ns ns ns   100nM 105±1 % ns ns   1µM 112±10 % ns 105±3%   10µM 115±7% ns 115±17%

All the above values were significantly different from the respective control currents (p<0.01, Student's*t*-test) except where indicated (n.s.). n.s. means not significant. To improve the selectivity, the difluoro analog SR-III-61 was synthesized from an aldehyde (**59**). The alcohol **46** was treated with Dess Martin periodinane in dry  $CH_2Cl_2$  to afford aldehyde **59** in 96% yield. Treatment of aldehyde in dry  $CH_2Cl_2$  with DAST resulted in the difluro analog SR-III-61 (**49**) in 65% yield.



Scheme 11: Synthesis of SR-III-61

Figure 40: Oocyte data of SR-III-61 by Sieghart et al.

**Table 18.** EC<sub>50</sub> and Hill Slope for Modulation of Control GABA EC<sub>3</sub> by **SR-III-61 (49)** in Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Results represent EC<sub>50</sub> (95% Confidence interval for EC<sub>50</sub>) and Hill Slope (Nh)  $\pm$  SD from at least four cells from two batches.

		α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
	EC <sub>50</sub>	>10µM	>3nM		>0.03µM
SR-III-61 <b>(49)</b>	95%CI		(1-27nM)		
	Nb		0.8		
			0.0		

**Table 19.** Modulation of Control GABA EC<sub>3</sub> by **SR-III-61** (**49**) on Xenopus Oocytes Expressing Different GABA<sub>A</sub> Receptor Subtypes. Values are presented as mean  $\pm$  SEM of at least four oocytes from at least two batches.

	α1β3γ2	α2β3γ2	α3β3γ2	α5β3γ2
10nM	ns	109±4 %	ns	102±7 %
100nM	ns	119±6%	112±2 %	107±5%
1µM	110±4 %	116±1%	116±2%	109±8%
10µM	113±4%	113±1%	144±16%	110±9%
	10nM 100nM 1μM 10μM	α1β3γ2   10nM ns   100nM ns   1μM 110±4 %   10μM 113±4%	α1β3γ2 α2β3γ2   10nM ns 109±4 %   100nM ns 119±6%   1µM 110±4 % 116±1%   10µM 113±4% 113±1%	α1β3γ2 α2β3γ2 α3β3γ2   10nM ns 109±4 % ns   100nM ns 119±6% 112±2 %   1µM 110±4 % 116±1% 116±2%   10µM 113±4% 113±1% 144±16%

All the above values were significantly different from the respective control currents (p<0.01, Student's*t*-test) except where indicated (n.s.). n.s. means not significant. Data of SR-III-61 (**49**) from the electrophysiology indicated (in oocytes) was an agonist at  $\alpha$ 3 subtypes at high concentrations. However at a pharmacologically relevant concentration (100nM, -7) the ligand demonstrated efficacy at all subtypes. Unfortunately, the efficacy data showed no significant subtype selectivity. The synthesis of the other PWZ analogs is currently underway.

#### 7. Conclusions

In this study, it was clear that QH-II-066 (11) which is  $\alpha$ 5 subtype selective ligand is an agonist. It was shown<sup>88</sup> that QH-II-066 (11) exhibited functional selectivity *in vivo* with potent diazepam-like efficacy at  $\alpha$ 5 subtypes and partial efficacy at the  $\alpha$ 1 subtype. Recent studies by Weiss and Cook et al.<sup>89</sup> have shown that activation of GABA (A) by QH-II-066 (11) induces apoptosis in neuroblastoma. This is very important finding. The mechanism by which GABA-A activation inhibited the growth of neuroblastoma cells was assessed by Cho, Cook et al. QH-II-066 (11), an  $\alpha$ 5 agonist passes through the blood brain barrier which opens up the potential to apply in neurology and psychology as chemotherapeutic agent. These results also highlight a possible connection between the role of neurotransmitters in nervous system development and the regulation of neuroblastoma growth as shown by Weiss, Rallapalli and Cook et al. (unpublished, see Appendix III).

Bivalent ligand XLi-093 (**33**) has clearly been shown to be the first subtype selective antagonist at  $\alpha$ 5 subtypes which is known (Rowlett, Savic et al.) to cross the blood brain barrier. Rowlett showed that the memory impairing effects of diazepam were blocked by

a dose of XLi-093 (3.0mg/kg) that occupied about 66% of the  $\alpha$ 5 GABA<sub>A</sub> binding sites. A series of bivalent ligands with oxygen linker were synthesized based on  $\alpha 5$  selectivity of XLi-093 (33). Comparison of the results of low temperature NMR studies to crystal structures has provided enough information to demonstrate that low temperature NMR can be used as a quick method to identify dimeric ligands with a tendency to fold back upon themselves, as compared with those preferring a linear conformation. А correlation with binding data shows that the suitability of a ligand in the  $\alpha$ 5 BzR/Gabaergic subtype is heavily influenced by its conformation in solution. Variable temperature NMR thus can be used as a tool for screening bivalent ligands for their in vivo suitability. It is also clear the presence of one oxygen atom in the linker was the principle cause for the dimer to fold back onto itself, consequently resulted in diminished activity. From the recent work of Rowlett, it was demonstrated that novel  $\alpha 5$ selective inverse agonist PWZ-029 (47) enhanced cognition in rhesus monkeys in the CANTAB paradigm. This ligand had the ability to reverse cholinergic deficits in performance induced by the antimuscarinic scopolamine under mixed trial conditions. In the ORD task, PWZ-029 showed only a modest trend for enhancement of performance. But when task difficulty was increased by testing with difficult trials only, PWZ-029 robustly increased performance. This enhancement was reversed by administration of the  $\alpha$ 5 GABA (A) subtype selective antagonist XLi-093 and this antagonism in turn was reversed by increasing the dose of PWZ-029. In addition, PWZ-029 enhanced performance in the DNMS task using the 10 minute delay with distracters. This ligand also exhibited anxiolytic activity in some primates and was a weakly active anticonvulsant in rats (NINDS, Jim Stables et al. see Appendix IV). These findings are

consistent with a role for PWZ-029 (47)  $\alpha$ 5 GABA<sub>A</sub> receptors in the treatment of ageassociated memory impairment and Alzheimer's disease. It is the subject of an SBIR grant submitted December 5<sup>th</sup>, 2012 by Physiogenix (see Appendix VII).

# 8. General Experimental Details

All reactions were carried out under an argon atmosphere with dry solvents using anhydrous conditions unless otherwise stated. Tetrahydrofuran (THF) and diethyl ether were freshly distilled from Na/benzophenone ketyl prior to use. Dichloromethane was distilled from calcium hydride prior to use. Methanol was distilled over magnesium sulfate prior to use. Benzene and toluene were distilled over sodium and acetonitrile was distilled over CaH<sub>2</sub> prior to use. Reagents were purchased of the highest commercial quality and used without further purification unless otherwise stated. Thin layer chromatography (TLC) was performed using Dynamic Adsorbents Inc. UV active silica gel, 200 µm, plastic backed; Dynamic Adsorbents Inc. UV active alumina N, 200 µm, F-254 plastic backed plates. Flash and gravity chromatography were performed using silica gel P60A, 40-63 µm purchased from Silicycle. Elemental analyses were performed on a Carlo Erba model EA-1110 carbon, hydrogen, and nitrogen analyzer. All samples submitted for CHN analyses were first dried under high vacuum for a minimum of six hours using a drying pistol with isopropyl alcohol or benzene as the solvent with potassium hydroxide pellets in the drying bulb. Melting points were taken on a Stuart melting point apparatus SMP3 manufactured by Barloworld Scientific US Ltd. Proton (<sup>1</sup>H NMR) and carbon high resolution nuclear magnetic resonance spectra (<sup>13</sup>C NMR) were obtained on a Bruker 300-MHz or a GE 500-MHz NMR spectrometer. <sup>1</sup>H NMR data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublets, dt = doublet of triplets, ddd = doublet of doublet of doublet, td = triplet of doublets, qd = quartet of doublets, sex = sextet, m = multiplet), integration, and coupling constants (Hz). <sup>13</sup>C NMR data are reported in parts per million (ppm) on the  $\delta$  scale. The low resolution mass spectra (LRMS) were obtained as electron impact (EI, 70eV), which were recorded on a Hewlett-Packard 5985B gas chromatography-mass spectrometer, while high resolution mass spectra (HRMS) were recorded on a VG Autospec (Manchester, England) mass spectrometer. HRMS recorded by electrospray ionization (ESI) methods were performed at the Laboratory for Biological Mass Spectrometery at Texas A&M University on an API QStar Pulsar model, manufactured by MDS Sciex.

**5-Phenyl -1,4 benzodiazepine-2-one** (**7**) : The 2-amino-benzophenone **5** (100.6g, 0.51mol) and NaHCO<sub>3</sub> (128.6g, 1.5mol) were suspended in dry CH<sub>2</sub>Cl<sub>2</sub> (1.3L) at rt. A solution of bromoacetyl bromide (113.2g, 0.56mol, 48.9mL) in dry CH<sub>2</sub>Cl<sub>2</sub> (200mL) was added dropwise over 30 min and the mixture was allowed to stir for another 3h. The reaction was allowed to stir until analysis by TLC (silica gel) indicated the absence of starting material, after which the above reaction mixture was concentrated under vacuum to provide amide intermediate **6** which was added directly to a saturated solution of anhydrous ammonia in CH<sub>3</sub>OH (1.5L). The mixture was heated to reflux for 6h until analysis by TLC (silica gel) indicated the absence of the 2-bromoacetyl intermediate **6** prepared above. After the solvent was removed under reduced pressure, the solid which remained was washed with hexane/EtOH (1:4) to provide pure benzodiazepine **7** (107g, 95%) as a white powder. **3**: mp: 135-145 °C; **IR** (KBr) 3454, 3185, 3063, 2891, 1681,

1614, 1467 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  9.68 (s, 1H), 7.09-7.59 (m, 9H), 4.31 (s, 2H) MS (EI) m/e 236 (M<sup>+</sup>, 93), 208 (100), 180 (28), 152 (19); Anal. Cald for C<sub>15</sub>H<sub>12</sub>N<sub>2</sub>O: C, 76.25; H, 5.12; N, 11.86; Found C, 76.21; H, 5.19; N, 11.94. The spectral data were in agreement with the literature.<sup>65</sup>

**7-Bromo-5-phenyl-1,4-benzodiazepine-2-one 8**: Benzodiazepine **7** (61.4g, 0.274mol) was dissolved in acetic acid (1L) and was treated with H<sub>2</sub>SO<sub>4</sub> (15.2mL, 0.55mol, 36N) and bromine (87.1 g, 0.55mol). After stirring at rt for 10d, the reaction mixture was poured into ice water and brought to pH 5-6 with an aq solution of 6N NaOH. The crude bromide **4** precipitated and was collected by vacuum filtration and recrystallized from EtOH to provide the bromide **8** as a white solid in 90% yield. **8**: mp: 188-194 °C; IR (KBr) 3466, 3173, 3075, 2953, 1687, 1602, 1480 cm<sup>-1</sup>: <sup>1</sup>H NMR (DMSO-d, 300MHz)  $\delta$  10.95(s, 1H), 7.11-7.89 (m, 8H), 4.25 (s, 2H). MS (CI) m/e 316 (M<sup>+</sup> +1, 78), 265 (15), 237 (100), Anal. Cald for C<sub>15</sub>H<sub>12</sub>N<sub>2</sub>O: C, 57.16; H, 3.52; N, 8.89; Found C, 57.56; H, 3.49; N, 8.82. This material was identical to that reported in the literature. <sup>65</sup>

**7-Trimethylsilylacetyleno-5-phenyl-1,3-dihydrobenzo[e]-1,4-diazepine-2-one 9:**The 6- Bromo-benzodiazepine **8** (5g, 15.86mmol) was dissolved in dry triethyl amine (108mL) and acetonitrile (68mL). This mixture was degassed 3-4 times under vacuum and back filled with argon. To this degassed mixture, bis (triphenylphosphine) palladium (II) acetate (0.713g, 1.1mmol) and trimethylsilylacetylene (4.48mL, 31.72 mmol) was added and the mixture was heated to reflux under argon. After 12 h the reaction mixture was cooled to rt and filtered through a bed of celite with EtOAc. The filtrate was concentrated in vacuuo and the residue was purified via flash chromatography [silica gel, EtOAc:hexane (1:1)] to furnish the TMS analog **9** as a yellow powder (4.5g, 85%): **mp**: 190.0-191.5 °C; **IR** (KBr) 3011, 2281, 1686, 1610, 1486 cm<sup>-1</sup>: <sup>1</sup>**H NMR** (CDCl<sub>3</sub>, 300 MHz)  $\delta$ 10.17 (s, 1H), 7.21-7.61 (br, 7H), 7.09 (d, 1H, J=8.25Hz), 4.31 (s, 2H), 0.21 (s, 9H) ; **MS** (**CI**) m/e (relative intensity) 333 (M<sup>+</sup> +1, 100). **Anal. Cald for C**<sub>20</sub>**H**<sub>20</sub>**N**<sub>2</sub>**OSi**: C, 72.25; H, 6.06; N, 8.43; Found C, 72.20; H, 6.11; N, 8.52

**1-Methyl-** 7 - trimethylsilylacetyleno-5- phenyl - 1,3-dihydrobenzo[e]-1,4-diazepine-**2-one 10**: The acetyleno amine 9 (1g, 3mmol) was dissolved in dry THF (15mL) at -15°C and NaH (60% mineral oil, 144mg, 3.6mmol) was added to the solution in one portion. The slurry was then stirred for 30 min and then CH<sub>3</sub>I (0.28mL, 4.5mmol) was added to the mixture and this mixture was allowed to warm to rt. After the mixture was allowed to stir overnight, the reaction was quenched with water (200mL) and EtOAc (100mL). The mixture was extracted with EtOAc and then washed with brine and dried (Na<sub>2</sub>SO<sub>4</sub>). After the removal of the solvent under reduced pressure, the residue was purified by flash chromatography [hexane:EtOAc (75:25)] to provide the TMS-analog 10 (0.9g, 85%) as a white solid : mp: 177-178 °C; IR (KBr) 2954, 2147, 1687, 1612, 1491 cm<sup>-1</sup>: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz,)  $\delta$  7.20-7.27 (m, 3H), 7.07 (d, 1H, *J*=8.58Hz), 7.05 (s,1H), 4.60 (d, 1H, *J*=10.8Hz), 3.54 (d, 1H, *J*=10.8 Hz), 3.18 (s, 3H); MS (EI) m/e 346 (M<sup>+.</sup> 90), 318 (100), 303 (19), 165 (22), 151 (20). Anal. Cald for C<sub>21</sub>H<sub>22</sub>N<sub>2</sub>OSi: C, 72.79; H, 6.40; N, 8.08; Found C, 72.50; H, 6.68; N, 8.04

**1-Methyl-7-acetyleno-5- phenyl- 1,3-dihydrobenzo[e]-1,4-diazepine-2-one 11**: A solution of trimethylsilyl acetyleno benzodiazepine **10** (0.87g, 2.5 mmol) was dissolved in THF (200mL) and the solution was cooled to -30°C. A solution of TBAF (1.0M in THF; 3.01mL, 3mmol) was then added to the mixture and it was allowed to warm to rt. After stirring at rt for 20min the reaction mixture was quenched with water (200mL) and

extracted with EtOAc (200mL), washed with brine and dried (Na<sub>2</sub>SO<sub>4</sub>). After removal of the solvent under reduced pressure the residue was purified by flash chromatography [EtOAc : hexane (1:1)] to provide the target compound **11** (**QH-II-066**) (0.62g, 90%) as light yellow crystals. mp: 163-165 °C; IR (KBr) 2965, 1680, 1605 cm<sup>-1</sup>: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.62 (dd, 1H, J=8.5Hz, 2.0Hz), 7.35-7.45 (m, 4H), 7.27-7.42 (d, 1H, *J*=8.5Hz), 4.80 (d, 1H, *J*=10.9Hz), 3.85 (d, 1H, *J* = 10.8Hz), 3.46 (s, 3H), 3.11 (s, 3H); **MS (EI)** m/e 274 (M<sup>+</sup>, 100), 259(12), 246 (10), 189 (12), 122 (19), 105 (42). **Anal. Cald for C<sub>18</sub>H<sub>14</sub>N<sub>2</sub>O.2/3H<sub>2</sub>O:** C, 75.51; H, 4.89; N, 9.78; Found C, 75.59; H, 5.17; N, 9.62. This material was identical to that reported in the literature. <sup>47,65</sup>

### (Z)-8-Bromo-2-((4-methylpiperidin-1-yl)methylene)-6-phenyl-2,4-dihydro-1H-

benzo[f]imidazo[1,2-a][1,4]diazepin-1-one 16: A suspension of 7-bromo-1,3-dihydro-5-phenyl-2H-1,4-benzodiazepin-2-thione 12 (1.6g, 4.83mmol), glycine (1.81g, 24.2mmol) and Na<sub>2</sub>CO<sub>3</sub> (1.84g, 17.4mmol) in EtOH (38mL)-H<sub>2</sub>O (16mL) was stirred at reflux for 1 h and then poured into water (100mL), after which the mixture was filtered to remove a small amount of 7-bromo-1,3-dihydro-5-phenyl-2H-1,4-benzodiazepin-2one which remained. The filtrate was extracted with CHCl<sub>3</sub>. The CHCl<sub>3</sub> extract was discarded and then the aq layer was adjusted to pH 4 with a solution of aq 2N HCl after which it was extracted with CHCl<sub>3</sub> (3x25mL). Evaporation of the CHCl<sub>3</sub> solution gave pure acid 13 (1.2g, 67%) as a yellow solid. Acid 13 (350mg, 0.941mmol) was suspended in dry CH<sub>2</sub>Cl<sub>2</sub> (10mL) and DCC (223mg, 1.08mmol) was added. The suspension which resulted was stirred at 40°C for 2h and then cooled to 0°C. The mixture which resulted was filtered and the solvent was removed under vacuum to give 8-bromo-2,4-dihydro-6phenyl-1H-imidazo[1,2-a][1,4] benzodiazepine-1-one 14 as a brown oil. The cyclized amide **14** (250mg) was dissolved in dry benzene (6mL), after which dimethylformamide diethylacetal (130mg, 0.883 mmol) and triethyl amine (89mg, 0.883 mmol) were added. The solution which resulted was stirred at rt for 1 h and the solvent was removed under vacuum. The residue was crystallized from EtOAc-MeOH to give **15** (200mg, 70%). A solution of **15** (180mg, 0.440 mmol) in dry toluene (5mL) was treated with 1-methyl piperazine (1mL) and heated to reflux for 5h. The solvent was removed under vacuum to give a gum which crystallized from CH<sub>2</sub>Cl<sub>2</sub>-Et<sub>2</sub>O to furnish **16** (146mg, 72%) as a yellow solid. **mp** >250 °C; IR (KBr) 3324, 2932, 2787, 1692, 1624, 1475, 1402, 1297, 1137, 933 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  7.95 (d, 1H, *J*=8.8*Hz*), 7.72 (dd, 1H, *J*=12*Hz*), 4.50-4.60 (m, 1H), 4.20-4.30 (m, 1H), 4.16 (d, 1H, *J*=12Hz), 3.50-3.58 (m, 2H), 2.40-2.60 (m, 4H), 2.34 (s, 3H); **MS (m/z)** 465 (100). Anal. Calcd. for C<sub>23</sub>H<sub>22</sub>N<sub>5</sub>OBr. 1/3H<sub>2</sub>O: C, 58.79; H, 4.95; N, 14.89. Found: C, 58.73; H, 4.86; N, 14.83. The spectral data of **16** were identical to the published results.<sup>158</sup>

(Z)-2-((4-Methylpiperidin-1-yl)methylene)-6-phenyl-8-((trimethylsilyl)ethynyl)-2,4dihydro-1H-benzo[f]imidazo[1,2-a][1,4]diazepin-1-one 17: To a suspension of amide 16 (140mg, 0.302mmol) in dry acetonitrile (4mL) and triethylamine (3mL) was added bis (triphenylphosphine)-palladium (II) acetate (22.6mg, 0.03mmol). The solution which resulted was degassed and trimethylsilyl acetylene (0.1mL, 0.7mmol) was added. The mixture was heated to reflux and stirred overnight. After removal of the solvent under vacuum, the residue was dissolved in  $CH_2Cl_2$  and washed with a saturated aq solution of NaHCO<sub>3</sub> and brine. The organic layer was dried (Na<sub>2</sub>CO<sub>3</sub>), filtered and concentrated under vacuum. The residue was purified by flash column chromatography (EtOAc: MeOH 9:1) to furnish the trimethylsilyl analog **17** (100mg, 69%) as a pale yellow solid. **mp** >250 °C; **IR** (KBr) 3436, 2936, 2794, 2154, 1682, 1625, 1489, 1136, 847cm<sup>-1</sup>; <sup>1</sup>H **NMR** (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.80 (d, 1H, J=8.5Hz), 7.68 (dd, 1H, J=1.9Hz, 8.5Hz), 7.55-7.59 (m, 2H), 7.37-7.49 (m, 4H), 7.16 (s, 1H), 4.99 (d, 1H, J=12Hz), 4.50-4.60 (m, 1H), 4.20-4.30(m, 1H), 4.13 (d, 1H, J=12.4Hz), 3.48-3.58(m, 2H), 2.4-2.6 (m, 4H), 2.35 (s, 3H), 0.23 (s, 9H); **MS** (**m**/**z**) 482 (100). Anal. Calcd. for C<sub>28</sub>H31N<sub>5</sub>OSi. H<sub>2</sub>O: C, 67.60; H, 6.54; N, 13.59. Found: C, 67.30; H, 6.66; N, 14.01.

### (Z)-8-Ethynyl-2-((4-methylpiperidin-1-yl)methylene)-6-phenyl-2,4-dihydro-1H-

**benzo** [**f**]**imidazo**[**1**,**2**-**a**][**1**,**4**]**diazepin-1-one 18** (**PS-1-37**): A solution of the trimethylsilyl analog **17** (65mg, 0.135mmol) in THF (15mL) was stirred with TBAF hydrate (45mg, 0.175mmol) at -5°C for 5min. After this, H<sub>2</sub>O (5mL) was added to the solution to quench the reaction and stirring continued at -5°C for one half h. The solution was then extracted with EtOAc (3x40mL) and the organic layer was washed with water. After removal of the solvent under reduced pressure, diethyl ether was added to the residue to precipitate a solid. The solid was filtered and washed with CHCl<sub>3</sub>-Et<sub>2</sub>O (1:15) to provide the acetyl target **18** (40mg, 73%). **mp** 223-224 °C; **IR** (KBr) 3298, 2935, 2786, 1695, 1628, 1364, 1136, 1002, 778cm; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.04 (d, 1H, *J*=8.5Hz), 7.71 (dd, 1H, *J*=1.9Hz, 8.5Hz), 7.55-7.58 (m, 2H), 7.36-7.48 (m, 4H), 7.17 (s, 1H), 5.0 (d, 1H, *J*=12.1Hz), 4.5-4.6 (m, 1H), 4.2-4.3 (m, 1H), 4.16 (d, 1H, *J*=12.1Hz), 3.5-3.6 (m, 2H), 3.08 (s, 1H), 2.4-2.6 (m, 4H), 2.35 (s, 3H); **MS (m/z)** (100). The spectral data of **14** were identical to the published results.<sup>159</sup>

**4-Methyl-3,4-dihydro-1H- benzo[e][1,4]diazepine-2,5-dione(26):** A mixture of isatoic anhydride **25** (20g, 101 mmol) and sarcosine (9.02g, 101 mmol) in DMSO (160mL) was heated at 150°C for 5 hr, after which the solution was cooled to rt and poured into ice water (750mL) to furnish a light brown solid. This solid was collected by filtration, washed with water (3x200mL) and dried. The benzodiazepine **26** was obtained as a light brown solid (19g, 84%). This material was used directly in the next experiment.

### 7-Bromo-4-methyl-3,4-dihydro-1H-benzo[e][1,4]diazepine-2,5-dione(27):

Benzodiazepine **26** (32.4g, 0.17mol) was added to a solution of acetic acid (400mL) in a 3-neck reaction flask. Sodium acetate (27.9g, 0.34mol) was added to this solution. Bromine (13.1mL, 0.255mol) was added using an addition funnel and the solution which resulted was allowed to stir for 18h and then poured into ice water (200mL) to provide a white solid. This solid was collected by filtration and then washed (3x200mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). The bromobenzodiazepine **27** (36.6g, 80%) was used directly in the next experiment. **mp** 240-242 °C; <sup>1</sup>**H NMR** (DMSO-d<sub>6</sub>, 300MHz)  $\delta$ 10.54 (br s, 1H), 7.81 (d, 1 H, J = 2.2 Hz), 7.67 (dd, 1 H, J = 2.4, 8.6 Hz), 7.04 (d, 1 H, J = 8.7 Hz), 3.87 (s, 2 H), 3.09 (s, 3 H); **MS** (**CI**) m/z 269 (M + H).

## tert-Butyl-8-bromo-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5-

a][1,4]benzodiazepine-3-carboxylate (28): The *t*-BuOK (7.50 g, 66.91 mmol) was added to a solution of benzodiazepine 27 (15 g, 55.76 mmol) in anhydrous THF (1500 mL) at 0 °C and the mixture which resulted was stirred for 20 min. The reaction mixture was cooled to -35 °C and diethyl chlorophosphate (10.47 mL, 72.48 mmol) was added slowly. After stirring at 0°C for 30 min the mixture was cooled to -78°C and t-butyl

isocyanoacetate (8.926 mL, 61.33 mmol) was added, and this was followed by the addition of *t*-BuOK (6.88g, 61.33 mmol) at -78°C. After stirring at rt for 4 h the reaction was quenched with a saturated aq NaHCO<sub>3</sub> solution (500 mL) and extracted with EtOAc ( $3 \times 1000$  mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to give a solid residue. This solid residue was triturated with Et<sub>2</sub>O (250 mL) and the ester **28** precipitated as an off-white solid. The mother liquor was further purified by flash chromatography on silica gel (gradient elution 40– 60% EtOAc in hexane) to afford additional ester **28** in a combined yield of (15.3g, 70%). **mp** 180-183 °C; <sup>1</sup>**H NMR** (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.33 (s, 1H), 7.90 (s, 1H), 7.62 (d, 1H, *J* =10.8 Hz), 7.03 (d, 1H, *J* = 8.6 Hz), 4.71 (br, s, 1H), 4.13 (br, s, 1H), 3.10 (s, 3H), 1.56 (s, 9H); **MS (EI)** *m/e* 394 (M<sup>+</sup>, 100). The spectral data for **28** were identical to the published values.

#### tert-Butyl-8-(trimethylsilyl)-acetylene-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5-

a][1,4]benzodiazepine3-carboxylate (20): A mixture of imidazobenzodiazepine 28 (7.5g, 19.12mmol) was dissolved in dry triethyl amine (223mL) and acetonitrile (149mL; 3:2 ratio). This mixture was degassed under argon three times. To this degassed mixture, bis (triphenylphosphine)-palladium (II) acetate (1g, 7 mol %) was added. The mixture was again degassed (with vacuum) three times under argon. Trimethylsilyl acetylene (5.52mL, 38.24mmol) was added and this reaction mixture was heated to reflux for 8h under argon. The reaction mixture was cooled to rt and the solvents were concentrated under reduced pressure. This concentrated mixture was dissolved in ethyl acetate and passed through a bed of celite. After the removal of solvent under reduced pressure the residue was purified by flash chromatography (silica gel, EtOAc) to afford a yellow solid of **RY-23 (20)** (6.5g, 83%). **mp** 199-200 °C. **IR** 

(KBr) 2971, 2156, 1723, 1662, 1500 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.13 (s, 1H), 7.84 (s, 1H), 7.64 (d, 1H, J = 8.2 Hz), 7.32 (d, 1H, J = 8.2 Hz), 5.11 (br, 1H), 4.31 (br, 1H), 1.63 (s, 9H), 0.25 (s, 9H); MS (EI) m/e 409 (M<sup>+</sup>, 4), 354 (19), 353 (82), 335 (45), 307 (100), 279 (23), 251 (16), 143 (13), 160 (21), 107 (14); Anal. Calcd for C<sub>22</sub>H<sub>27</sub>N<sub>3</sub>O<sub>3</sub>Si: C, 64.52; H, 6.64; N, 10.26; Found C, 64.57; H, 6.70; N, 10.46. The spectral data for **20** were in excellent agreement with the literature. <sup>85</sup>

### tert-Butyl-8-acetylene-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5-

**a**][1,4]benzodiazepine3-carboxylate (21): A solution of 20 (1.2g, 2.93 mmol) in THF (360mL) was treated with a solution of TBAF solution (1.0M in THF; 3.52mL,3.51mmol) at -30°C. The mixture which resulted was allowed to stir for 30 min at rt, after which the mixture was added to H<sub>2</sub>O (50mL) and extracted with EtOAc (3x100mL). The combined organic extracts were washed with brine (50mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). After removal of the solvent under reduced pressure, the residue was purified by a wash column (silica gel, EtOAc) to give 21 (RY-24) as a yellow solid (0.89g, 90%): mp 213-214 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.16 (s, 1H), 7.85 (s, 1H), 7.69 (d, 1H, *J* = 8.3 Hz), 7.35 (d, 1H, *J* = 8.3 Hz), 5.15 (br, 1H), 4.32 (br, 1H), 3.23 (s, 3H), 3.20 (s, 1H), 1.63 (s, 9H); MS (EI) m/e 337 (M<sup>+</sup>, 2), 282(15), 281 (78), 264 (31), 263(56), 235 (100), 229(22), 207 (36), 138 (16), 101(19); Anal. Calcd for C<sub>19</sub>H<sub>19</sub>N<sub>3</sub>O<sub>3</sub>: C, 67.64; H, 5.68; N, 12.45; Found C, 67.50; H, 5.78; N, 12.41. The spectral data for 21 were in excellent agreement with the reported values.

8-Ethynyl-5,6-dihydro-5-methyl-6-oxo-4H-imidazo [1,5-a][1,4] benzodiazepine-3carboxylic acid 32 : Ethyl-8-ethynyl-5,6-dihydro-5-methyl-6-oxo-4H-imidazo [1,5a][1,4] benzodiazepine-3-carboxylate 23 (1.5g, 5.3mmol),available from the literature, was dissolved in EtOH (100mL), after which an aq solution of 2N NaOH (15mL) was added to the solution. The mixture which resulted was heated to 70°C and allowed to stir for 1hr. The mixture was then allowed to cool to rt and the EtOH was removed under reduced pressure. Aq 2N HCl was added to the residue to bring the pH to approximately 3-4. A precipitate formed which was removed by filtration, washed with water (3x50mL) and dried. The 3-carboxylic acid **32** was obtained as a white solid in 90% yield; **mp** 315-316 °C. <sup>1</sup>H NMR (CD<sub>3</sub>OD, 300MHz)  $\delta$  8.20 (s, 1H), 7.80 (dd, 1H, *J* =2.5Hz, 10 Hz), 7.69 (s, 1H), 5.22 (br, 1H), 4.50 (br, 1H), 3.74 (s, 1H), 3.28 (s, 3H), ; **MS (EI)** m/e 281 (M<sup>+</sup>, 80), 277(100), 263 (60), 235 (84), 207(55) **Anal. Calcd for C**<sub>15</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>: C, 64.04; H, 3.94; N, 14.95; Found C, 64.40; H, 4.12; N, 14.76. The spectral data for **20** were in excellent agreement with the published reports. <sup>160</sup>

#### 1,3-Bis(8-acetyleno-5,6-dihydro-5-methyl-6-oxo-4H-

imidazo[1,5a][1,4]benzodiazepine-3-carboxy) propyl diester 33 (XLi-093) : To a solution of carbonyl diimidazole (230.3mg, 0.57mmol) in anhydrous DMF (5mL) was added 8-ethyl-5,6-dihydro-5-methyl-6-oxo-4H-imidazo [1,5-a][1,4] benzodiazepine-3-carboxylic acid 24 (200mg, 0.71mmol). The solution which resulted was allowed to stir for 2h at rt. Analysis of the material by TLC (silica gel) indicated the absence of starting material. To this solution, 1, 3 propanediol (27.1mg, 0.36mmol) in dry DMF (0.5mL) and DBU (114.2mg, 0.75mmol) in dry DMF (0.5mL) were added to the solution at rt. The mixture was allowed to stir for 12h until analysis by TLC (silica gel) indicated the absence of starting material. The reaction mixture was then poured into ice water (30mL) and extracted with EtOAc (3x50mL). The combined organic layers were washed with H<sub>2</sub>O (5x50mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). The solvent was removed under reduced

pressure and the residue was purified by flash chromatography (silica gel, EtOAc/MeOH: 4:1) to provide the dimer **33** (**XLi-093**; 157mg, 74%) as white solid. **mp** >230 °C (dec); **IR** (NaCl) 3247, 1725, 1641, 1359, 1253, 1061 cm<sup>-1</sup>; <sup>1</sup>H NMR (CD<sub>3</sub>OD, 300MHz)  $\delta$  8.20 (d, 2H, *J*=1.76Hz), 7.89 (s, 2H), 7.72 (dd, 2H, *J* =6.43Hz, 1.86 Hz), 7.41 (d, 2H, *J*=8.3Hz), 5.31 (br, 2H), 4.57 (t, 4H, *J*=6.2Hz), 4.40 (s, 2H), 3.26 (s, 6H), 3.24 (s, 2H), 2.37 (m, 2H); <sup>13</sup>C NMR (CD<sub>3</sub>OD, 300MHz)  $\delta$  26.2, 34.4, 40.7, 60.2, 78.7, 79.7, 120.4, 121.6, 127.1, 127.7, 130.1, 133.4, 134.1, 134.9, 161.3, 164.1; **MS (FAB)** m/e 603 (M<sup>+</sup> +1, 100) **Anal. Calcd for C<sub>33</sub>H<sub>26</sub>N<sub>6</sub>O<sub>6</sub>.2/3 CH<sub>3</sub>OH: C**, 64.81; H, 4.63; N, 13.47; Found C, 64.56; H, 4.72; N, 13.76. The spectral data for XLi-093 were identical to the published values. <sup>160</sup>

**1,3-Bis(8-ethyl-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5a][1,4] benzodiazepine-3-carboxy) propyl diester 34:** Bis (8-acetyleno-5,6-dihydro-5-methyl-6-oxo-4Himidazo[1,5a][1,4]-benzodiazepine-3-carboxy) propyl diester **33** (500mg, 0.83mmol) was dissolved in EtOH (150mL) and degassed 3-4 times with a balloon of hydrogen after which Pd/C (176mg) was added to the solution at rt. The slurry was allowed to stir for 12h under H<sub>2</sub>. The catalyst was removed by filtration and washed with EtOH. The EtOH was removed under reduced pressure to furnish a residue. This material was purified by flash chromatography (silica gel, EtOAc:MeOH, 9:1) to provide (**34)XLi-356** (504mg, 99%) as white crystals: **mp** 125-133°C; **IR** (NaCl) 3407, 2964, 2358, 1725, 1640, 1499 cm<sup>-</sup>1; <sup>1</sup>**H NMR** (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.15 (s, 1H), 7.89 (t, 4H, *J*=3.2Hz, 5.3Hz), 7.48 (d, 2H, *J*=1.8Hz), 7.33 (d, 2H, *J*=8.2Hz), 5.16 (br, 2H), 4.56 (t, 4H, *J*=6.1, 12.2 Hz), 4.48 (br, 2H), 3.26 (s, 6H), 2.78 (dd, 4H, *J*=7.5, 15.1Hz), 2.39 (m, 2H), 1.29 (m, 6H) ; **MS (EI)** m/e (relative intensity) 611 (M<sup>+</sup>+1, 100); **Anal Calcd for** 

#### 1,5-Bis(8-acetyleno-5,6-dihydro-5-methyl-6-oxo-4H-

imidazo[1,5a][1,4]benzodiazepine-3-carboxy) pentyl diester 35: To a solution of carbonyl diimidazole (173mg, 1.06mmol) in anhydrous DMF (5mL) was added 8-ethyl-5,6-dihydro-5-methyl-6-oxo-4H-imidazo [1,5-a][1,4] benzodiazepine-3-carboxylic acid **32** (150mg, 0.534mmol). The solution which resulted was stirred for 2h at rt. Analysis of the mixture by TLC (silica gel) indicated the absence of starting material. To this solution, 1,5 pentanediol (27.8mg, 0.267mmol) in dry DMF (0.5mL) and DBU (114.2mg, 0.75mmol) in dry DMF (0.5mL) were added at rt. The mixture was allowed to stir for 12h until analysis by TLC (silica gel) indicated absence of starting material. The reaction mixture was then poured into ice water (30mL) and extracted with EtOAc (3x50mL). The combined organic layers were washed with H<sub>2</sub>O (5x50mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). The solvent was removed under reduced pressure and the residue was purified by flash chromatography (silica gel, EtOAc/MeOH: 4:1) to provide the dimer 35 XLi-**210** (123mg, 89.2%) as a white solid. **mp** 132-138 °C ; **IR** (NaCl) 3422, 3280, 2931, 1714, 1635, 1487, 1249, 1064 cm<sup>-1</sup>; <sup>1</sup>**H NMR** (CD<sub>3</sub>OD, 300MHz) δ 8.18 (s, 2H), 7.86 (s, 2H), 7.70 (m, 2H), 7.36 (dd, 2H, J=8.1, 16Hz), 5.29 (s, 2H), 4.39 (s, 8H), 3.52 (s, 2H), 3.24 (s, 6H), 1.90 (m, 4H); **MS** (**FAB**) m/e 631 ( $M^+$  +1, 100) Anal. Calcd for C<sub>35</sub>H<sub>30</sub>N<sub>6</sub>O<sub>6</sub>,5/3 H<sub>2</sub>O: C, 63.61; H, 4.83; N, 12.72; Found C, 63.16; H, 4.72; N, 13.06.

Bis(8-acetyleno-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5a][1,4]benzodiazepine-3-carboxy) diethylene glycol diester 36 (XLi-347). To a solution of carbonyl diimidazole (243 mg, 1.50 mmol) in anhydrous DMF (8 mL) was added 8-ethynyl-5, 6dihydro-5-methyl-6-oxo-4H-imidazo[1,5-a][1,4]-benzodiazepine-3-carboxylic acid **32** (351.4 mg, 1.25 mmol). This solution was stirred for 2 h at rt until analysis by TLC (silica gel) indicated the absence of starting material. To the solution which resulted was then added diethylene glycol (66.3 mg, 0.62 mmol) in dry DMF (0.5 mL) and also DBU (190 mg, 1.25 mmol) in dry DMF (0.10 mL) at rt. The mixture was stirred at rt for 8 h until analysis by TLC (silica gel) indicated the reaction was complete. The reaction mixture was then poured into ice water (30 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). The combined organic layer was washed with  $H_2O$  (5 x 50 mL), brine and dried (Na<sub>2</sub>SO<sub>4</sub>). The solvent was removed under reduced pressure and the residue was purified by flash chromatography (silica gel, EtOAc/CH<sub>3</sub>OH, 4:1) to provide **36** (150 mg) as a white solid in 60% yield. 28: mp 118-121°C (dec.); IR (NaCl) 3247, 2880, 1725, 1639, 1496, 1253, cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.19 (d, 2H, *J*=1.76Hz), 7.90(s, 2H), 7.72(dd, 2H, J=1.8 Hz and J=8.3 Hz), 7.39 (d, 2H, J=6.1 Hz), 5.20(br, 2H), 4.37(br, 2H), 3.67 (t, 4H, J=5.1Hz), 3.23(s,6H), 3.21(s, 2H); <sup>13</sup>C NMR (53.4 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 35.4, 42.3, 63.5, 68.9, 79.6, 81.3, 122.2, 122.4, 128.4, 129.4, 131.9, 135.1, 135.7, 135.8, 136.0, 162.5, 165.2; MS (FAB,NBA) m/e (relative intensity) 633(M<sup>+</sup>+1, 11). Anal. **Calcd** for C<sub>34</sub>H<sub>28</sub>N<sub>6</sub>O<sub>7</sub> •1/3 C<sub>2</sub>H<sub>5</sub>OH: C, 61.38; H, 4.45; N, 12.39. Found: C, 60.99; H, 4.62; N, 12.16.

**Dibromodimethyl ether 38^{111}**. Bromine (120 g) was added slowly via an addition funnel to a capped three-neck flask containing water (22.4 ml) in equal portion intervals over a 2 h period. Paraformaldehyde (120 g) and red phosphorus (27.2 g) were added to the bromine/water solution at a comparable rate in equal portions after each bromine

addition. The reaction vessel and vials containing the starting material were kept capped and only opened briefly to allow sample removal and addition. The reaction mixture was ice water-cooled during the addition period of the process. After completing the addition of all the starting material the final reaction mixture was set aside for 20 h. The heavy crude ether layer, dibromodimethyl ether, was separated, saturated with hydrogen bromide, and dried with phosphorus pentaoxide (P<sub>2</sub>O<sub>5</sub>) for 24 h. Simple distillation at atmospheric pressure afforded the colorless dibromo-dimethyl ether **38** as a fraction with b. p. = 146-159 °C. The data for **38** were identical to the literature values.<sup>112</sup>

Bis(8-acetyleno-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5a][1,4]benzodiazepine-3-carboxy) dimethyl glycol diester 39 (XLi374). 8-Ethynyl-5, 6-dihydro-5-methyl-6oxo-4H-imidazo[1,5-a][1,4]-benzodiazepine-3-carboxylic acid 32 (100 mg, 0.356 mmol) was dissolved in DMF (10 mL) and to this solution dibromodimethyl ether (36.3 mg, 0.178 mmol) was added, followed by TEA(2.0 mL). After 26 h of stirring at ambient temperature the solvent was removed under reduced pressure and the residue was distributed between CHCl<sub>3</sub> and water. The organic phase was washed multiple times with water and dried over Na<sub>2</sub>SO<sub>4</sub>. The dried crude ether was applied on a column of silica gel to afford dimer 39 (86 mg) in 80% yield. 39: mp >220 °C (dec.); IR (KBr) 3419, 3237, 2910, 1714, 1635, 1561, 1498 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$  8.15 (s, 2H), 7.90 (s, 2H), 7.74(dd, 2H, *J=1.6Hz*, *J=8.2Hz*), 7.40(d, 2H, *J=8.1* Hz), 5.74 (s, 4H), 5.29(br, 2H), 4.37 (br, 2H), 3.31(s,2H), 3.18(s, 6H); MS(FAB,NBA) *m/e* (relative intensity) 605(M<sup>+</sup>+1, 100). Anal. Calcd for C<sub>32</sub>H<sub>24</sub>N<sub>6</sub>O<sub>7</sub> •3/2 CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub>: C, 61.99; H, 4.93; N, 11.41. Found: C, 61.36; H, 4.52; N, 11.96.
#### 8-Chloro-5-methyl-6-Oxo-5,6-dihydro-4H-benzo-[f]-imidazo-[1,5-a]-[1,4]-

**dizaepine-3-carboxylic acid (41).** The ester **40** (2 g, 6 mmol) was dissolved in EtOH (150 mL) to form a solution and then 10% aq sodium hydroxide solution (40 mL) was added to the mixture. The mixture was heated to reflux for 0.5 h and the EtOH was removed under reduced pressure, after which the solution was allowed to cool. The pH was then adjusted to 4 by adding 10% aq HCl dropwise. The solid which precipitated was filtered and the solid acid was washed with water and ether. The solid was dried to provide acid **41**: mp >260oC; IR (neat) 3100, 2424, 1661, 1642, 1562 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, DMSO)  $\delta$  8.35 (s, 1H), 7.77-7.88 (m, 3H), 5.28 (br s, 1H), 4.55 (br s, 1H), 3.09 (s, 3H); MS (EI) m/e (relative intensity) 291 (M+, 44), 273 (61), 245 (100), 230 (47), 217 (62), 75 (37). **Anal. Calcd.** for C<sub>13</sub>H<sub>10</sub>ClN<sub>3</sub>O<sub>3</sub>• 0.3CH<sub>2</sub>Cl<sub>2</sub>: C, 50.57; H, 3.37; N, 13.32. Found: C, 50.64; H, 3.25; N, 13.12. This material was used directly in a later step.

#### Bis(8-chloro-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5a][1,4]benzodiazepine-3-

**carboxy**) **diethylene glycol diester 42(SR-II-35).** To a solution of carbonyl diimidazole (0.197g, 1.22 mmol) in anhydrous DMF (8 mL) was added 8-ethynyl-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5-a][1,4]-benzodiazepine-3-carboxylic acid **41** (0.297g, 1.01 mmol). The solution which resulted was stirred for 2 h at rt until analysis by TLC (silica gel) indicated the absence of starting material. To the solution which resulted was then added diethylene glycol (48 mg, 0.46 mmol) in dry DMF (0.5 mL) and also DBU (186mg, 1.22 mmol) in dry DMF (0.10 mL) at rt. The mixture was stirred at rt for 8 h until analysis by TLC (silica gel) indicated the reaction was complete. The reaction

mixture was then poured into ice water (30 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). The combined organic layer was washed with H<sub>2</sub>O (5 x 50 mL), brine and dried (Na<sub>2</sub>SO<sub>4</sub>). The solvent was removed under reduced pressure and the residue was purified by flash chromatography (silica gel, EtOAc/CH<sub>3</sub>OH, 4: 1) to provide **41** (177 mg) as a white solid in 60% yield. **41**: <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.03 (d, 2H, *J*=2.4*Hz*), 7.83(s, 2H), 7.72(dd, 2H, *J*=3.3 *Hz and J*=9.3 *Hz*), 7.40 (s,1H), 7.35(s, 1H), 5.18 (br, 2H), 4.55 (s,4H), 4.36 (br, 2H), 3.93 (t, 4H, *J*=4.8 *Hz*), 3.26 (s,6H); **MS (EI)** *m/e* 652 (M<sup>+</sup>).

**7-Chloro-4-methyl-3,4-dihydro-1H-benzo[e][1,4]diazepine-2,5-dione (45):** A mixture of 5-chloroisatoic anhydride **44** (20g, 101 mmol) and sarcosine (9.02g, 101 mmol) in DMSO (160mL) was heated at 150°C for 5 hr, after which it was cooled to rt and poured into ice water (750mL) to furnish a light brown solid. This solid was collected by filtration, washed with water (3x200mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). The benzodiazepine **45** was obtained as a light brown solid (19g, 84%). This material was used directly in the next experiment.

Ethyl-8-chloro-5, 6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5-][1,4]benzodiazepine3carboxylate (40): *t*-BuOK (8.99 g, 80.14 mmol) was added to a solution of amide 45 (15 g, 66.77 mmol) in anhydrous THF (1500 mL) at 0 °C and the solution allowed to stir for 20 min. The reaction mixture which resulted was cooled to -35 °C and diethyl chlorophosphate (12.48 mL, 86.80 mmol) was added slowly. After stirring at 0 °C for 30 min, the mixture was cooled to -78 °C and ethyl isocyanoacetate (8.027 mL, 73.44 mmol) was added and this was followed by the addition of *t*-BuOK (8.24g, 73.44 mmol). After stirring at rt for 4 h, the reaction mixture was quenched with a saturated aq solution of NaHCO<sub>3</sub> (500 mL) and extracted with EtOAc ( $3 \times 1000$  mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to give a solid residue. This solid residue was treated with Et<sub>2</sub>O (250 mL) and the ester **40** was precipitated as an off-white solid. The mother liquor was further purified by flash chromatography on silica gel (gradient elution, 40– 60%, EtOAc in hexane) to afford additional ester **40** with an overall yield of (19g, 89%). **mp** 192-193 °C; <sup>1</sup>**H NMR** (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.1 (d, 1H, *J* = 2.4 Hz), 7.90 (s, 1H), 7.62 (dd, 1H, *J* = 8.7, 2.5 Hz), 7.40 (d, 1H, *J* = 8.6 Hz), 5.23 (br, s, 1H), 4.46 (q, 2H, J=7.12Hz), 4.13 (br, s, 1H), 3.27 (s, 3H), 1.47 (t, 3H, *J*=7.12Hz); **MS (EI)** *m*/*e* 319 (M<sup>+</sup>, 100). The data for this material were identical to the published values.<sup>161</sup>

8-Chloro-5, 6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5- $\alpha$ ][1,4]benzodiazepine-3methyl alcohol (46). A solution of imidazobenzodiazepine 40 (5 g, 15.6 mmol) in a mixture of ethyl ether (50 mL), and THF (50 mL) was stirred with LiBH<sub>4</sub> (2.0 M in THF, 9 mL, 18 mmol). The mixture which resulted was then heated to reflux for 30 min, after which it was cooled to rt and treated with a saturated aq solution of NaHCO<sub>3</sub> (5 mL). The solvent was concentrated and the residue was taken up in EtOAc (100 mL). The organic layer was washed with water (2 × 20 mL), brine (20 mL) and dried (MgSO<sub>4</sub>). After removal of the solvent under reduced pressure the residue was purified by flash chromatography (silica gel, EtOAc) to afford alcohol 46 as colorless crystals (2.9 g, 67%): mp 252-253 °C; IR (KBr) 3500 (br, OH), 3100, 1667, 1612, 823 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.00 (d, 1H, *J* = 2.4 Hz), 7.80 (s, 1H), 7.55 (dd, 1H, *J* = 8.7, 2.4 Hz), 7.30 (d, 1H, *J* = 8.6 Hz), 4.70 (d, 2H, *J* = 4.2 Hz), 4.40 (s, 2H), 3.20 (s, 3H); **MS** (**EI**) m/e 279 (M<sup>+</sup>, 41), 277 (M<sup>+</sup>, 100), 259 (84), 246 (55), 231(41). The data for this material were identical to the published values.<sup>19</sup>

#### Methyl(8-chloro-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5- $\alpha$ ][1,4]

**benzodiazepin-3-yl)methyl ether (47).** To a slurry of KOH (2.41g, 44 mmol) in DMSO (56 mL) at rt were added alcohol **46** (3g, 11 mmol) and excess CH<sub>3</sub>I. The mixture which resulted was stirred for 5-10 min and then poured into ice water (100 mL) after which it was extracted with EtOAc ( $3 \times 100$  mL). The combined organic extracts were washed with brine (50mL) and dried (MgSO<sub>4</sub>). After removal of solvent under reduced pressure, the residue was purified by a wash column on silica gel (EtOAc) to give ether **47** (**PWZ-029**) as an off-white powder (2.83g, 90%): **mp** 193-194 °C; **IR** (KBr) 3122, 2973, 1632, 1611, 811 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.00 (d, 1H, *J* = 2.4 Hz), 7.80 (s, 1H), 7.55 (dd, 1H, *J* = 8.7, 2.5 Hz), 7.30 (d, 1H, *J* = 8.6 Hz), 4.55 (br, 2H), 4.38 (s, 2H), 3.42 (s, 3H), 3.18 (s, 3H). The data for this material were identical to the published values.

Ethyl (8-chloro-5,6-dihydro-5-methyl-6-oxo-4H-imidazo[1,5- $\alpha$ ][1,4] benzodiazepin-3-yl)methyl ether (48). To slurry of KOH (2.41g, 44 mmol) in DMSO (56 mL) at rt were added alcohol 46 (3g, 11 mmol) and excess C<sub>2</sub>H<sub>5</sub>I. The mixture which resulted was allowed to stir for 5-10 min, poured into ice water (100 mL), and extracted with EtOAc (3 × 100 mL). The combined organic extracts were washed with brine (50mL) and dried (MgSO<sub>4</sub>). After removal of the solvent under reduced pressure the residue was purified by a wash column on silica gel (EtOAc) to give ether 48 (SR-III-81) as an off-white powder (2.98g, 90%): mp 137-138 °C; IR (KBr) 31282, 2973, 1634, 1601, 810 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  7.92 (d, 1H, *J*=2.4Hz), 7.86 (s, 1H), 7.50 (dd, 1H, *J*=8.5, 2.4 Hz), 7.30 (d, 1H, *J*=8.6Hz), 4.55 (s, 2H), 3.60 (q, 2H, *J*=7.0Hz), 3.12 (s, 3H), 1.22 (t, 3H, *J*=7.0Hz). Anal. Calcd. For C<sub>15</sub>H<sub>16</sub>ClN<sub>3</sub>O2. 0.0235 H<sub>2</sub>O, C 58.84, H 5.28, N 13.72. Found 58.94, H 5.40, N 13.32

#### 8-Chloro-N,5-dimethyl-6-oxo-5,6-dihydro-4H-benzo[f]imidazo-[1,5-

a][1,4]diazepine-3-carboxamide(57): The ester 40 (100mg, 0.313 mmol) was dissolved in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (5mL). To this mixture, a solution of CH<sub>3</sub>NH<sub>2</sub> (0.147mL, 1.56 mmol) was added. This reaction mixture (in a sealed tube) was placed in a preheated oil bath at 45°C overnight. The solvent was then removed under reduced pressure and the residue was purified by flash chromatography with CH<sub>2</sub>Cl<sub>2</sub>/MeOH (4.5:0.5) to provide 3-N-methyl amide 57 (74mg, 78% yield). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.1 (d, 1H, *J* = 2.4Hz), 7.90 (s, 1H), 7.62 (dd, 1H, *J* = 8.6, 2.5 Hz), 7.40 (d, 1H, *J* = 8.6Hz), 5.23 (br, 1H), 5.23 (br, 1H), 3.28 (s, 3H), 3.02 (d, 3H, *J*=3.9Hz) ; MS (EI) *m/e* 304 (M<sup>+</sup>, 100). Anal. Calcd. For C<sub>14</sub>H<sub>13</sub>ClN<sub>4</sub>O<sub>2</sub>. 0.28 H<sub>2</sub>O, C 54.25, H 4.41, N 18.07. Found 54.39, H 4.52, N 17.69

**8-Chloro-3-(fluoromethyl)-5-methyl-4H-benzo[f]imidazo[1,5-a][1,4]diazepin-6(5H)one (58) :** The alcohol **46** (100mg, 0.360mmol) was dissolved in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (5 mL). At 0°C, DAST (diethylaminosulfur trifluoride, 0.07mL, 0.720mmol) <sup>162</sup> was added very slowly and the reaction was allowed to stir for 2h. The reaction mixture was then quenched with aq solution of NaHCO<sub>3</sub> and allowed to stir for 1h at 0°C and then extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with brine and dried (MgSO4). After removal of the solvent under reduced pressure the residue was purified by a wash column on silica gel (gradient elution 4:1 EtOAc: hexane) to afford the fluoro analog **58** (**SR-III-38**; 0.070mg, 70%). <sup>1</sup>**H NMR** (CDCl<sub>3</sub>, 300MHz)  $\delta$  8.05 (dd, 1H, *J*=2.3Hz, 1.4Hz), 7.89 (d, 1H, *J*=1.4Hz), 7.62 (m, 1H), 7.37 (dd, 1H, *J* = 8.4, 1.2 Hz), 5.57 (s, 1H), 5.41 (s, 1H), 4.40 (br, 2H), 3.25 (d, 3H, J=1.3Hz); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 282MHz) δ -205.21 (t, 1F, *J*=52Hz); MS (EI) *m/e* 279 (M<sup>+</sup>, 100). Anal. Calcd. For C<sub>13</sub>H<sub>11</sub>ClFN<sub>3</sub>O.0.13 (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>O, C 56.11, H 4.28, N 14.54. Found 55.86, H 3.97, N 14.17

#### 8-Chloro-3-(difluoromethyl)-5-methyl-4H-benzo[f]imidazo[1,5-a][1,4]diazepin-

6(5H)-one 49: The alcohol 46 (1g, 3.6mmol) was dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (100mL) and Dess-Martin periodinane<sup>163</sup> (2.29g, 5.41mmol) was added in one portion at 0°C. This solution which resulted was stirred at rt for 2 h. The reaction mixture which resulted was then quenched with a saturated aq solution of  $Na_2S_2O_3$ , washed with NaHCO<sub>3</sub> and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). After removal of the solvent under reduced pressure, the residue was purified by a wash column on silica gel (gradient elution 1:9 EtOAc: hexane) to afford the aldehyde **59** (0.953g, 96%). <sup>1</sup>**H NMR** (CDCl<sub>3</sub> 300MHz)  $\delta$  10.35 (s, 1H), 8.09 (d, 1H) J=2.43Hz), 7.94 s, 1H), 7.64 (dd, 1H, J=8.55, 2.41Hz), 7.41 (d, 1H, J=8.6Hz), 5.23 (br, s, 1H), 4.6 (br, s, 1H), 3.25 (s, 3H); **HRMS (ESI)** calculated for  $C_{13}H_{11}ClN_3O_2$ : 276.0540; found : 276.0536 The aldehyde 59 (100mg, 0.363mmol) was dissolved in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (5 mL). At 0°C DAST (diethylaminosulfur trifluoride, 0.07mL, 0.725mmol) was added very slowly and the reaction was allowed to stir for 2h at 0°C. The reaction mixture was then quenched with an aq solution of  $NaHCO_3$  and allowed to stir for 1hr and extracted  $(CH_2Cl_2)$ . The combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). After removal of the solvent under reduced pressure, the residue was purified by a wash column on silica gel (gradient elution 4:1 EtOAc: hexane) to afford the difluoro analog 49 (0.070mg, 65%). <sup>1</sup>H NMR (CDCl<sub>3</sub> 300MHz)  $\delta$ 8.05 (dd, 1H, J=2.4Hz), 7.88 (s, 1H), 7.62 (dd, 1H, J = 8.6 Hz, 2.4 Hz), 7.37 (d, 1H, J =

8.5Hz), 6.83 (t, 1H, *J*=55Hz), 4.55 (d, 1H, *J*= 14Hz), 4.45 (d, 1H, *J*=14Hz), 3.2 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 300MHz)  $\delta$  35.41, 42.01, 112.52 (t, *J* = 234Hz), 123.2, 128.11, 130.38 (d, *J*=31.8Hz), 132.43, 132.67, 134.50, 134.80, 165.14; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 282MHz)  $\delta$  -111.76 (dd, *J*<sub>(*F*-*F*)</sub>=309Hz, *J*<sub>(H-F)</sub> =55Hz), -109.78 (dd, *J*<sub>(*F*-*F*)</sub>=309Hz, *J*<sub>(H-F)</sub> =55Hz); **HRMS (ESI)** Calculated for C<sub>13</sub>H<sub>11</sub>ClF<sub>2</sub>N<sub>3</sub>O : 298.0559; found: 298.0561.

#### 8-Chloro-3-(difluoromethyl)-5-methyl-4H-benzo[f]imidazo[1,5-a][1,4]diazepin-

6(5H)-one 60: The acid 41 (300mg, 1.02 mmol) was dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (5mL) and at  $0^{\circ}$ C, excess SOCl<sub>2</sub> was added in one portion. The reaction mixture was heated to reflux for 2h until analysis of the mixture by TLC indicated the absence of starting material. The reaction mixture was concentrated under reduced pressure on a rotavapor a few times with CH<sub>2</sub>Cl<sub>2</sub> to remove the excess SOCl<sub>2</sub> and dried under vacuum. The solid which resulted was stirred in dry THF at -60°C. In a separate flask, bis[2-(N,Ndimethylamino)ethyl] ether<sup>37</sup> (0.239mL, 1.25mmol) was stirred in dry THF (5mL) and at  $0^{\circ}$ C, EtMgBr (0.419mL, 1.25mmol) was slowly added over 15min. After stirring the reaction mixture for 5min, the mixture which resulted was quenched with an aq solution of  $NH_4Cl$  and extracted with EtOAc (3x10mL). The combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). After removal of the solvent under reduced pressure the residue was purified by a wash column on silica gel (gradient elution 4:1 EtOAc: hexane) to afford the ketone 60 (0.167g, 53 %). HRMS (ESI) Calculated for  $C_{15}H_{14}ClN_3O_2$ : 304.0853; found:304.0861. The ketone **60** (100mg, 0.360mmol) was dissolved in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (5 mL). At 0°C, DAST (diethylaminosulfur trifluoride, 0.07mL, 0.720mmol)<sup>162</sup> was added very slowly and the reaction was allowed to stir for 2h. No product formation was observed. Addition of excess DAST (diethylaminosulfur

trifluoride) did not work either. When various fluorinating agents such as Xtal-fluor or Deoxo-fluor were employed they did not yield the product. Excess addition of these fluorinating agents also was futile.

### 9. Computer Modeling

The core structures of the ligands were taken from available X-ray crystallographic coordinates or generated using the SYBYL fragment library.<sup>164</sup> The structures which resulted were energy minimized using MM2 (molecular mechanics program 2) or MMFF (Merck molecular force field), and the subsequent Monte Carlo conformational searches were carried out on MacroModel 6.0 on a Silicon Graphics Personal Iris 4D/35 workstation or a Silicon Graphics Octane SI 2P 175 R10000 workstation, respectively. The low energy conformations were then fully optimized *via* molecular orbital calculations at the 3-21G basis set with torsional angles fixed. The structures which resulted were further calibrated with 6-31G\* single point calculations at an "SCF=TIGHT" convergence criteria *via* Gaussian 92<sup>165</sup> on a Silicon Graphics Indigo R4400 workstation, or Gaussian 94<sup>166</sup> on a Silicon Graphics Octane SI2P175R10000 workstation.<sup>43</sup>

#### 9.1. Competition Binding Assays (With Dr. Majumder and Dr. Roth)

Competition binding assays were performed in a total volume of 0.5 mL at 4 °C for 1 h using [<sup>3</sup>H] flunitrazepam as the radiolabel. A total of 6  $\mu$ g of cloned human GABA<sub>A</sub> receptor DNA containing the desired  $\alpha$  subtype along with the  $\beta$ 2 and  $\gamma$ 2 subunits were used for transfecting the HEK 293T cell line using Fugene 6 (Roche Diagnostic) transfecting reagent. Cells were harvested 48 h after transfection, washed with Tris-HCl

buffer (pH 7.0) and Tris Acetate buffer (pH 7.4) and the resulting pellets were stored at -80 °C until assayed. On the day of the assay, pellets containing 20-50  $\mu$ g of GABA<sub>A</sub> receptor protein were resuspended in (50 mM Tris-acetate pH 7.4 at 4 °C) and incubated with the radiolabel as previously described.<sup>167</sup> Nonspecific binding was defined as radioactivity bound in the presence of 100  $\mu$ M diazepam and represented less than 20% of total binding. Membranes were harvested with a Brandel cell harvester followed by three ice-cold washes onto polyethyleneimine-pretreated (0.3%) Whatman GF/C filters. Filters were dried overnight and then soaked in Ecoscint A liquid scintillation cocktail (National Diagnostics; Atlanta, GA). Bound radioactivity was quantified by liquid scintillation counting. Membrane protein concentrations were determined using an assay kit from Bio-Rad (Hercules, CA) with bovine serum albumin as the standard.

## 10. Experimental procedures for the electrophysiology using the twoelectrode voltage clamp (with Dr. R. Furtmüller and Dr. W. Sieghart)<sup>168-171</sup>

Ligand effects on GABA<sub>A</sub> receptors were tested by two-electrode voltage clamp experiments in cRNA injected Xenopus oocytes that functionally expressed several subtype combinations of GABA<sub>A</sub> receptors.<sup>168</sup>

#### **10.1.1 Preparation of cloned mRNA**

Cloning of GABA<sub>A</sub> receptor subunits  $\alpha 1$ ,  $\beta 3$  and  $\gamma 2$  into pCDM8 expression vectors (Invitrogen, CA) has been described elsewhere.<sup>172</sup> GABA<sub>A</sub> receptor subunit  $\alpha 4$  was cloned in an analogous way. cDNAs for subunits  $\alpha 2$ ,  $\alpha 3$  and  $\alpha 5$  were gifts from P. Malherbe and were subcloned into a pCI-vector. cDNA for subunit  $\alpha 6$  was a gift from P.

Seeburg and was subcloned into the vector pGEM-3Z (Promega). After linearizing the cDNA vectors with appropriate restriction endonucleases, capped transcripts were produced using the mMessage mMachine T7 transcription kit (Ambion, TX). The capped transcripts were polyadenylated using yeast poly(A) polymerase (USB, OH) and were diluted and stored in diethylpyrocarbonate-treated water at -70 °C.

#### **10.1.2 Functional expression of GABA**<sub>A</sub> receptors

The methods used for isolating, culturing, injecting and defolliculating of the oocytes were identical to those as described previously.<sup>109,168,173</sup> Briefly, mature female Xenopus laevis (Nasco, WI) were anaesthetized in a bath of ice-cold 0.17 % Tricain (Ethyl-maminobenzoate, Sigma, MO) before decapitation and removal of the frog ovary. Stage 5 to 6 oocytes with the follicle cell layer around them were singled out of the ovary using a platinum wire loop. Oocytes were stored and incubated at 18 °C in modified Barths medium (MB, containing 88 mM NaCl, 10 mM HEPES-NaOH (pH 7.4), 2.4 mM NaHCO<sub>3</sub>, 1 mM KCl, 0.82 mM MgSO4, 0.41 mM CaCl<sub>2</sub>, 0.34 mM Ca(NO<sub>3</sub>)<sub>2</sub>) that was supplemented with 100 U/mL penicillin and 100 µg/mL streptomycin. Oocytes with follicle cell layers still around them were injected with a total of 2.25 ng of cRNA. This solution contained the transcripts for the different  $\alpha$  subunits and the  $\beta$  3 subunit at a concentration of 0.0065 ng/nL as well as the transcript for the  $\gamma$ 2 subunit at 0.032 ng/nL. After injection of the cRNA, oocytes were incubated for at least 36 h before the enveloping follicle cell layers were removed. To this end, oocytes were incubated for 20 min at 37 °C in MB that contained 1 mg/mL collagenase type IA and 0.1 mg/mL trypsin inhibitor I-S (both Sigma). This was followed by osmotic shrinkage of the oocytes in doubly concentrated MB medium supplied with 4 mM Na-EGTA. Finally, the oocytes were transferred to a culture dish containing MB and were gently pushed away from the follicle cell layer which stuck to the surface of the dish. After removal of the follicle cell layer, oocytes were allowed to recover for at least 4 h before being used in electrophysiological experiments.

## **10.1.3. Electrophysiological experiments**<sup>171,174</sup>

For electrophysiological recordings, oocytes were placed on a nylon-grid in a bath of Xenopus Ringer solution (XR, containing 90 mM NaCl, 5 mM HEPES NaOH (pH 7.4), 1 mM MgCl<sub>2</sub>, 1 mM KCl and 1 mM CaCl<sub>2</sub>). The oocytes were constantly washed by a flow of 6 mL/min XR which could be switched to XR containing GABA and/or drugs. Drugs were diluted into XR from DMSO solutions resulting in a final concentration of 0.1 % DMSO perfusing the oocytes. Drugs were preapplied for 30 sec before the addition of GABA, which was coapplied with the drugs until a peak response was observed. Between the two applications, oocytes were washed in XR for up to 15 min to ensure full recovery from desensitization. For current measurements the oocytes were impaled with two microelectrodes  $(2-3 \text{ m}\Omega)$  which were filled with 2 mM KCl. All recordings were performed at rt at a holding potential of -60 mV using a Warner OC-725C two-electrode voltage clamp (Warner Instruments, Hamden, CT). Data were digitized, recorded and measured using a Digidata 1322A data acquisition system (Axon Instruments, Union City, CA). Results of concentration response experiments were fitted using GraphPad Prism 3.00 (GraphPad Software, San Diego, CA). The equation used for fitting the Y=Bottom concentration response curves was (Top-+Bottom)/(1+10^((LogEC50-X)\*HillSlope)); X represents the logarithm of concentration, Y represents the response; Y starts at Bottom and goes to Top with a sigmoid shape.

### **10.2.** Anticonvulsant screen (with Dr. James Stables)<sup>174</sup>

The standard models incorporated into anticonvulsant screening include the maximal electroshock test (MES), the subcutaneous Metrazol test (scMET), and evaluations of toxicity (TOX)-which is sedation in this study. The data for each condition is presented as a ratio of either the number of animals protected or toxicity (loss of locomotor activity) over the number of animals tested at a given time point and dose. Experiments were performed in male rodents (albino Carworth Farms No. 1 mice (intraperitoneal route, ip) and albino Sprague-Dawley rats (oral route and ip). The mice weighed between 18 and 25 g, while rats were between 100 and 150 g. All animals had free access to food and water except during the actual testing period. Housing, handling, and feeding were all in accordance with recommendations contained in the "Guide for the Care and Use of Laboratory Animals". All of the test ligands were administrated in suspensions of 0.5% (w/v) of methyl cellulose in water. The volumes administered were 0.01 mL/g of body weight for mice and 0.04 mL/10 g for rats. Anticonvulsant activity was established using the maximal electroshock seizure (MES) and subcutaneous metrazole seizure threshold (scMET) tests.<sup>175,176</sup>

## 10.3. In Vivo Experiments (with S. Lelas)<sup>88</sup>

#### 10.3.1. Subjects

All animals were maintained in accordance with the guidelines of the Committee on Animals of the Bristol-Myers Squibb Company and the "Guide for Care and Use of Laboratory Animals" (Institute of Animal Laboratory Resources, 1996). Research protocols were approved by the Bristol-Myers Squibb Company Institutional Animal Care and Use Committee.

Male CDF rats weighing approximately 350 g at the time of the experiment (conflict), male Sprague-Dawley rats weighing 180-300 g (locomotor activity, rotorod studies, and drug discrimination studies) and male CF1 mice weighing 20-22 g at the time of the experiment (seizure studies) were purchased from Charles River Laboratories (Wilmington, Mass., USA). For the conditioned fear studies, male Sprague-Dawley rats weighing approximately 225 g at the time of the experiment were purchased from Harlan (Indianapolis, Ind., USA). Animals were housed in pairs or fours in clear plastic shoebox cages with Beta Chip bedding in rooms maintained at constant temperature (21-23°C) and humidity (50  $\pm$  10%) with a 12 h light/dark cycle (lights on at 0600 h, except for mice lights were on at 0500 h), except for the drug discrimination study for which animals were housed individually in suspended wire cages.

Animals had *ad libitum* access to food and water throughout the studies, except for the conflict and drug discrimination studies in which animals were restricted to 9 g (conflict) or 12 g (drug discrimination) of laboratory rodent chow (Bio-Serv, Frenchtown, NJ) per day in addition to the food earned in operant box, with unlimited access to water. All training and testing was done between 0600 and 1300 h Monday through Friday of each week.

#### **10.3.2.** Procedures

#### **10.3.2.1. GABA**<sub>A</sub> Binding Assay

HEK293 cells were trypsinized and counted. Cells were transfected using the Amaxa electroporator and 3 million cells per sample. Each transfection was performed in

Amaxa solution "V" using electroporation program Q-01 (Amaxa Inc., Gaithersburg, Md., USA). 1  $\mu$ g of each GABA subunit ( $\alpha$ ,  $\beta$ , and  $\gamma$ ) was used per transfection. Following electroporation, cells were plated onto poly-lysin coated glass coverslips and used for the experiments.

#### **10.3.2.2. Electrophysiological Studies**

Whole-cell patch-clamp experiments were performed using HEK cell lines transiently expressing two different configurations of GABA<sub>A</sub> constructs, on the first or second day after the transfection. Cells were plated onto 25 mm Petri dishes and viewed using Nikon TE200 inverted microscope, equipped with the mercury lamp to monitor GFP fluorescence. Cells were continuously perfused with extracellular physiological solution containing (in mM): NaCl 140, KCl 2.5, CaCl<sub>2</sub> 2, MgSO<sub>4</sub> 2, Na<sub>2</sub>HPO<sub>4</sub> 1.25, HEPES 10, Glucose 11, pH 7.4 and 340 mOsm (modified from Brown et al., 2002). Patch pipettes were pulled using Narishige vertical puller from fire-polished borosillicate glass capillaries. Pipette resistance was between 3.5 and 6 MOm. The intracellular solution contained (in mM): CsCl 140, HEPES 10, EGTA 5, MgATP 2, MgCl<sub>2</sub> 3, pH 7.3 and 340 mOsm.

Cells were voltage-clamped at -20 mV via an EPC10 amplifier (HEKA Electronics, Southboro, Mass.). Preliminary experiments have shown that amplitude of the GABA-induced current in these cells was not significantly different at -60, -40 or -20 mV holding potentials. At the same time, -20 mV holding potential facilitated longer experimental time which enabled testing of several concentrations of the test compounds. Drug solutions were applied to the cells via a multi-channel local perfusion

system (AutoMate Scientific, Inc., San Francisco, Calif.) with the estimated solution change time of 20 ms.

GABA was applied to the cell for 20 s with a 60 s washout period between applications, which was found to be a sufficient time period to reverse any desensitization that may occur. In addition, to ensure that there is no desensitization that may affect GABA responses at any given concentration of a GABA modulator, GABA was applied twice during control (no drug) and once at each concentration of the test compound. Test compounds were applied in increasing concentrations (1, 10, 100, or 1000 nM) 60 s prior to GABA (3  $\mu$ M) applications.

#### 10.3.3. Geller Conflict with Incremental Shock in Rats

Experiments were conducted in 8 model E10-10 Coulbourn operant chambers (Coulbourn Instruments, Allentown, Penn., USA; 28 x 26 x 31 cm) with an electrified grid floor which were housed in light-proof, sound-attenuated, and fan-ventilated chambers. On one wall of each operant chamber is a house light and a centrally placed food hopper 3 cm from the floor that delivered one 45-mg food pellet (Dustless Precision Pellets, Bio-Serv, Frenchtown, N.J., USA). To the left was mounted a lever response module with cue lights above. Foot-shock was delivered using an H13-16 Universal Shocker. Experiments were controlled with a Compaq DeskPro computer running Graphic State Notation software through Habitest Universal Linc interface equipment (Coulbourn Instruments).

Rats were trained to stability on an alternating multiple Random Interval 30 s/Fixed Ratio 1 Food +Shock paradigm (RI15/FR1 F+S). The paradigm ran through four cycles beginning with the Random Interval (RI) schedule which ran for 12 minutes then switched to the Fixed Ratio (FR) schedule which was cued and ran for 3 minutes. The RI had a range of 15 to 45 s with the food pellet being delivered with the first lever press after the expiration of the interval. The FR schedule delivered a food pellet after each lever press along with a foot-shock that increased with each lever press beginning at zero on the first lever press then increasing to 0.05 mA on the second lever press and incrementing by 0.05 mA with each subsequent lever press. The shock was reset to zero at the beginning of each RI/FR cycle.

#### 10.3.4. Protection against Pentylenetetrazole- and Shock-Induced Seizures in Mice

Pentylenetetrazole (PTZ) was administered at 125 mg/kg s.c. The number of animals surviving was recorded at 30 minutes and 60 minutes after administration of PTZ. Electroshock is administered using an Ugo Basile ECT, Unit 7801 seizure apparatus (Ugo Basile, Italy) and corneal electrodes soaked in 0.9% saline. Mice received a shock of 30 mA for 0.3 s. The number of animals that exhibited the hind-limb extensor component of the seizure was recorded.

#### **10.3.5.** Conditioned Fear in Rats

Experiments were conducted in standard modular test chambers (ENV-008, Med Associates, St. Albans, Vt., USA, 30.5 x 24.1 x 21 cm) with an electrified grid floor which were housed in light-proof, sound-attenuated, and fan-ventilated chambers. Foot-shock was delivered using a shocker/scrambler module (ENV-414). Experiments were controlled with a desktop computer using Med-PC software through Smart Ctrl<sup>™</sup> interface equipment (Med-Associates).

On day 1, rats were placed in the test chambers for a 7-minute conditioning session consisting of 2.5 minutes of habituation, a 2-s foot-shock at 0.7 mA, followed by 2.5

minutes of no foot-shock, another 2-s, 0.7 mA foot-shock, and another 2 minutes with no foot-shock. Twenty-four hours after the conditioning session, rats were placed in the test chamber again for 7 minutes with no foot-shocks and percent time freezing was quantified by video image analysis (Freezeview software using filter 25 and bout length of 0.75 s; Actimetrics and Coulbourn Instruments, Allentown, Penn., USA).

#### **10.3.6.** Spontaneous Locomotor Activity in Rats

The testing apparatus consisted of Plexiglas chambers ( $42 \times 42 \times 30 \text{ cm}$ ) equipped with Digiscan activity monitors (Omnitech Electronics, Columbus, Ohio, USA) that detect interruptions of eight photobeams spaced 5 cm apart and 2.5 cm above the floor. Horizontal activity was recorded in 5-minute bins for a total of 60 minutes and expressed as total distance covered (in cm).

#### **10.3.7. Rotorod Performance in Rats**

The degree of motor coordination or balance was determined using a standard accelerating rotorod treadmill (Ugo Basile, Comerio-Varese, Italy) that was 6 cm in diameter and 24 cm above the base. The speed was increased gradually from 2 rpm to a maximum speed of 20 rpm. The time each animal remained on the rotating rod was automatically recorded, up to a maximum of 5 minutes. Three acclimation trials were conducted for each animal prior to administration of drugs. The time on rotorod from the third trial was used to counterbalance animals for subsequent drug testing.

#### **10.3.8.** Chlordiazepoxide Discrimination in Rats

Twelve model E10-10 Coulbourn operant conditioning chambers (Coulbourn Instruments, Allentown, Penn., USA;  $28 \times 26 \times 31 \text{ cm}$ ) were housed in light-proof, sound-attenuated, and fan-ventilated chambers. Each operant conditioning chamber was

equipped with two non-retractable levers, requiring a downward force equivalent to 15 g (0.15 N), that were mounted 3 cm from the side wall, 3 cm above the metal grid floor, and 5 cm from a centrally placed food tray that delivered one 45-mg food pellet (Dustless Precision Pellets, Bio-Serv, Frenchtown, N.J.). The experimental chambers were connected to a Micro PDP11/73 computer using a LAB LINC interface. A SKED-11 operating system (State System, Kalamazoo, Mich.) was used to record and control behavior.

After habituation to the operant conditioning chamber, rats were trained to alternate daily between response levers on an FR 1 schedule of food reinforcement. Once lever pressing was well established, the reinforcement contingency was increased incrementally to an FR 10 schedule of food reinforcement, while maintaining the lever alternation. Rats were then trained to discriminate between an intraperitoneal (IP) injection of drug (5.0 mg/kg CDP) and vehicle (0.25% methylcellulose) administered 30 minutes prior to the start of the session. Ten consecutive responses on the injection-appropriate lever resulted in food delivery. Responses on the incorrect lever did not result in pellet delivery. Following an injection of CDP, responding on the right lever was reinforced for half the rats and responding on the left lever was reinforced for the other rats. In each two-week period there were five drug days and five saline days, with the constraint that there not be more than three consecutive drug or vehicle days.

Drug test sessions were conducted once a week (Fridays) with training sessions scheduled on intervening days (Monday – Thursday). The criterion for testing was achieved if no more than three incorrect responses occurred prior to the first reinforcer for at least 9 out of 10 consecutive sessions. During test sessions, the lever on which the

rat first made 10 consecutive responses resulting in delivery of the first food pellet was designated the "selected " lever and subsequent pellet delivery was made contingent on responding on this selected lever on an FR10 schedule. The selected lever, the number of responses prior to the first food pellet delivery, and the total number of responses in the session were recorded for each rat. The duration of test sessions was ten minutes. Dose-response functions were determined for CDP, diazepam, and QH-ii-066.

## 11. Experiments of GABA signaling in neuroblastoma (Weiss and Cook et al.)<sup>89</sup>

#### **11.1. Methods Summary:**

Backcross mice were generated by crossing *TH-MYCN* transgenic FVB/N mice to wild-type 129/SvJ, and subsequently crossing F1 offspring to wild-type 129/SvJ. Mice were observed for at least one year to be considered tumor-free. DNA was isolated from spleen and genotyped using a combination of microsatellite and SNP markers. RNA from SCG was analyzed using Affymetrix Mouse Exon arrays. Susceptibility linkage was analyzed using R-QTL. Differential gene expression was assessed using Significance Analysis of Microarrays, and eQTL were calculated using custom software as described previously.

#### **11.2. Supplementary Methods:**

**11.2.1. Mice** All mice were obtained from the Jackson Labs (Bar Harbor, ME), and were housed and treated following UCSF IACUC guidelines. Tumor-negative backcross mice were followed until one year of age (the latest tumor was detected at 342 days). Superior cervical ganglia (SCG) were surgically isolated and snap-frozen in liquid nitrogen. SCG were isolated from the parental control groups at 21 days.

**11.2.2. Taqman analysis of transgene expression:** Taqman expression analysis was performed on 6 mice (3 female, 3 male) from each strain. Proprietary assays for human MYCN and controls L18 and mGUS were obtained from Applied Biosystems (Carlsbad, CA).

**11.2.3. Genotyping** DNA was isolated from spleen tissue using a proteinase K lysis followed by phenol chloroform extraction. Microsatellite marker genotyping was carried out by the Marshfield Clinic (Marshfield, WI), and CIDR (Baltimore, MD). SNP genotyping was performed using template-directed primer extension with fluorescence polarization detection (FP-TDI, Acycloprime II, Perkin Elmer, Waltham, MA) and SNPStream<sup>2</sup> 48-plex (Beckman Coulter, Brea, CA). The marker set had an average spacing of 8 MB genome-wide (excluding the high density of markers on chromosome 10).

**11.2.4. Linkage analysis** Interval mapping was performed using the R-QTL<sup>3</sup> package in the R statistical language. Genotypes flagged as probable errors by R-QTL were discarded. The genetic map positions were determined using the physical map positions (NCBI 37/mm9), followed by re-estimation of the map using R-QTL, and likely mis-

mapped markers were discarded. Linkage analysis was performed on a 1-cM grid. Genome-wide significance thresholds were determined by running 1000 permutations for each dataset. Interval analysis was performed using the binary mode of the "EM" model. All results reported as significant were significant at a 5% genome-wide error rate. 95% confidence intervals (CI) were determined using the lodint function in R-QTL. Genes within the confidence intervals were determined by counting all genes in the UCSC genome assembly mapping between markers flanking the 95% CI.

**11.2.5. Expression Arrays** RNA from superior cervical ganglia was isolated using the RNEasy kit (QIAGEN, Valencia, CA), as we found these buffers were more effective at disrupting the ganglia than Trizol. 1µg of RNA was used as a starting template for the RiboMinus rRNA subtraction (Invitrogen, Carlsbad, CA) followed by the ST labeling protocol (Affymetrix, Santa Clara, CA). Labeled samples were hybridized to Affymetrix Mouse Exon 1.0 arrays. Array quality control was performed using the Affymetrix Expression Console. 2-way comparisons between homogenous groups (e.g. male vs. females or 129/SvJ vs FVB/N males) were performed using the Significance Analysis of Microarrays (SAM) package<sup>4</sup> using a 5% false discovery rate.

**11.2.6. eQTL analysis** Arrays were normalized using RMA in the XPS package (http://www.bioconductor.org/packages/2.6/bioc/html/xps.html). eQTL were calculated as described in <sup>5</sup>; briefly, linkage between gene expression and loci was assessed by linear regression with genome-wide significance assessed using a FDR-based method.

**11.2.7. Cell Culture** Neuroblastoma cell lines were grown in RPMI media with 10% serum and antibiotics with the exception of SK-N-BE(2) (DMEM/F12, 10% serum) and

IMR-32 (DMEM, 10% serum plus non-essential amino acids). Nor-NOHA was obtained from Bachem (Torrance, CA) and dissolved in DMSO. GABA was obtained from Sigma (St. Louis, MO) and dissolved in H<sub>2</sub>O. Picrotoxin and bicuculline were obtained from Sigma and dissolved in DMSO.

**11.2.8. Western blotting** Equal amounts of total protein were loaded for 4%–12% SDS-polyacrylamide gel electrophoresis and transferred to nitrocellulose membranes. After blocking, membranes were blotted with Arg1 (R&D Systems, Minneapolis MN), GABA B R1, Erk (Santa Cruz Biotech, Santa Cruz, CA) , GABA A R $\beta$ 2/3, GABA B R2,  $\beta$ -tubulin (Millipore, Billerica, MA), cleaved-PARP (Asp214), p-Akt (Ser473), Akt, pErk (Thr202/Tyr204) (Cell Signaling Technology, Danvers, MA), and GAPDH (Upstate Biotechnology, Lake Placid NY). Antibodies were detected with HRP-linked anti-mouse, anti-rabbit (Amersham/GE, Piscataway, NJ), or anti-sheep IgG (Calbiochem/EMD, Gibbstown, NJ), followed by enhanced chemiluminescence (Amersham/GE, Piscataway, NJ).

**11.2.9. WST-1 assay** Cells were grown in 12-well plates and treated for 72 hours prior to the addition of WST-1 reagent ( $100\mu$ l per 1ml media). Absorbance was read after 60 minutes. Results were plotted using Graphpad Prism software, and p-values were calculated using a two-tailed Student's t-test.

**11.2.10. FACS** Cells were plated on 6-well dishes and treated as indicated for 24 hours. They were then harvested, fixed in 70% ethanol for 30 minutes, then stained with a propidium iodide (PI) dye and RNase solution. All samples were analyzed on a FACSCaliber flow cytometer (Becton Dickenson, Franklin Lakes, NJ) and analyzed

using ModFit (Verity Software, Topsham ME). Apoptosis was detected by flow cytometry for annexin V-FITC per manufacturer's protocol (Annexin V-FITC detection kit, BioVision, Milpitas, CA) using FlowJo software (Tree Star, Inc, Ashland, OR).

## 12. Behavioral experiments (with Savic et al.)<sup>154</sup>

Experiments were carried out on male Wistar rats (Military Farm, Belgrade, Serbia), weighing 220-250 g. All procedures in the study conformed to EEC Directive 86/609 and were approved by the Ethical Committee on Animal Experimentation of the Medical Faculty in Belgrade. The rats were housed in transparent plastic cages, six animals per cage, and had free access to pelleted food and tap water. The temperature of the animal room was  $22\pm1^{\circ}$ C, the relative humidity 40-70%, the illumination 120 lux, and the 12/12h light/dark period (light on at 6:00 h). All handling and testing took place during the light phase of the diurnal cycle. Throughout the study the animals were used only once, with the exception of the grip strength measurement, which was done immediately after the tracking of behavior on the elevated plus maze. Spontaneous locomotor activity and elevated plus maze behavior were analyzed by the ANY-maze Video Tracking System software (Stoelting Co., Wood Dale, IL, USA). All drugs were dissolved/suspended with the aid of sonication in a solvent containing 85% distilled water, 14% propylene glycol, and 1% Tween 80, and were administered intraperitoneally in a volume of 1 ml/kg, 15 min before behavioral testing (for active and passive avoidance test, before the acquisition session). Time of administration, as well as, the doses of DMCM in various tests were chosen based on our previous studies (Savić et al., 2005a; 2005b; 2006). For interaction studies in the locomotor activity assay, the first treatment indicated in combination was administered into the lower left quadrant of the peritoneum 20 minutes before testing, and the second treatment 5 minutes later into the lower right quadrant of the peritoneum.

#### **12.1.** Passive Avoidance (PA) Paradigm

The experiments were performed in an adapted Automatic reflex conditioner (Ugo Basile, Milan, Italy, Model 7051), as described earlier.<sup>177</sup> In short, the apparatus consisted of a shuttle-box, equipped with a grid floor and divided with a sliding door into a lit and a dark compartment, and a programming unit. The animals were submitted to two, 24 hour – separated sessions. The acquisition session started by placing individual subjects in the illuminated compartment. After 30 seconds, the entrance to the dark compartment was opened, and as soon as the rat had entered it with all four paws, the footshock (2 s, 0.3 mA) was delivered. Immediately afterwards, the rat was returned to its home cage. The same procedure was repeated 24 hours later (retention session), without footshock. A cut-off time of 180 seconds was used on the training day, whereas, on the retention trial, a ceiling of 300 seconds was imposed.

# 13. Two-Way Active Avoidance (AA) Paradigm (with Savic et al.)<sup>154</sup>

The AA test was performed in automated two-way shuttle-boxes (Campden Instruments, Sileby, UK), as described earlier.<sup>177</sup> In short, the animals were submitted to two, 24-hour separated sessions. Training and test sessions were procedurally identical. Animals were placed singly into the shuttle box and left to freely explore for 15 minutes, and

habituation crossings were automatically counted. Afterward, they received 30 tone foot shock trials. During the first 5 seconds of each trial, a sound signal was presented (broadband noise of 69 dB), allowing the animal to avoid shocks by moving to the other compartment (avoidance response). If the animal did not respond within this period, a foot shock of 0.5 mA (7-s duration) was applied. Crossing to the adjacent compartment during the shock discontinued its delivery. The animal could move freely in the apparatus between trials (18-s intertrial intervals), and the intertrial crossings were automatically counted.

## 14. Measurement of Locomotor Activity (with Savic et al.)

Activity of single rats in a clear Plexiglas chamber (40 x 25 x 35 cm) under dim red light (20 lux) was recorded for a total of 30 minutes, without any habituation period, using ANY-maze software (as described above). For purposes of improving data analysis, the central 20% of the chamber ( $200 \text{ cm}^2$ ) was virtually set as a central zone. The minimum percentage of animal that must have been in the zone for an entry to occur was set at 70%, and 50% of the animal must have remained in the zone for an exit not to occur.

# 15. Behavior on the Elevated Plus Maze (EPM) (with Savic et al.)

The apparatus consisted of two open (50 x 10 cm, with a ledge of 0.3 cm) and two enclosed arms (50 x 10 x 40 cm), connected by a junction area (10 x 10 cm). The illumination in the experimental room was 10 lux on the surface of the arms. At the beginning of the experiment, single rats were placed in the center of the maze, facing

one of the enclosed arms, and their behavior was recorded for 5 min. An entry into an open or closed arm was scored when 90% of the animal crossed the virtual line separating the central square of the maze from the arm, whereas an exit occurred when more than 90 % of the animal left the respective arm. After each trial, the maze was cleaned with dry and wet towels.

## 16. Grip Strength Test (with Savic et al.)

Muscle strength was assessed by the grip strength meter (Ugo Basile, Milan, Italy, model 47105). When pulled by the tail, the rat grasps the trapeze connected to a force transducer, and the apparatus measures the peak force of experimenter's pull (in g) necessary to overcome the strength of the animal's forelimbs grip. Each animal was given three consecutive trials, and the median value was used for further statistics.

#### **16.1. Statistical Analysis**

All numerical data presented in the figures were given as the mean  $\pm$  SEM, except for results from the PA test (median latency with 25th, 75th interquartile range; data were non-parametric because the procedure involved a cut-off). For electrophysiological data Student's t-test was used for statistical analysis. Data from PA test were assessed by a Kruskal-Wallis non-parametric ANOVA, with post-hoc comparison relative to solvent control by a Dunn's test ( $\alpha$ =0.05). Data from the AA, EPM, grip strength and activity assay were assessed by a one-way ANOVA. If the ANOVA was significant, each treatment condition was compared with control by a Dunnett's test ( $\alpha$ =0.05). Where appropriate, the assessment of the antagonist influence on the inverse agonist effect was

conducted by a student's t-test. Statistical analyses were performed with ANY-maze Video Tracking System software (Stoelting Co., Wood Dale, IL, USA) and SigmaStat 2.0 (SPSS, Inc., Chicago, IL, USA).

## 17. *In vivo* studies on XLi-093 in mice (Rowlett and Cook et al)<sup>55</sup>

**Subjects** 

C57BL/6 Male Mice, 6 weeks old, n=8-10 per group

#### **17.1. Drugs**

1. Diazepam, Xli-093, flumazenil, bretazenil, and flunitrazepam were injected at doses between 03. and 30.0 mg/kg, i.p. and dissolved in a 50% PG and 50% distilled water vehicle

 [<sup>3</sup>H]Flumazenil (1.5μCi/30g mouse, i.v), [<sup>3</sup>H]L655-708 (1.5μCi/30g mouse, i.v.), both were dissolved in a saline vehicle

#### **17.2. HOLEBOARD MEMORY TASK**

**Apparatus**: Holeboard insert containing 16 holes in a sound attenuated locomotor activity chamber (Med Associates, Inc.)

Same 4 holes were baited for 4 consecutive 5-min sessions. Baited holes are changed daily. Mice were trained for 14 days or until they reached criteria. After training, testing

with DZ, XLi-093(XLi), DZ+ flumazenil (FLZ), or DZ+XLi-093 began. Performance was measured as a ratio of the number of novel correct holes (i.e., correct hole visited only once) /Total holes visited. Perfect performance was indicated by a ratio of 1.0.

### **17.3.** *IN VIVO* BINDING ASSAY

1. To determine binding site occupancy at  $\alpha 1$  and  $\alpha 2/3$ GABAA receptors, mice were injected with [<sup>3</sup>H] flumazenil ([<sup>3</sup>H]FLZ) (1.5  $\mu$ Ci/mouse, i.v.). Pretreatments (30 min): DZ (0.3-10.0 mg/kg, i.p.), XLi-093 (0.03-10.0 mg/kg, i.p.), flunitrazepam (30.0mg/kg, i.p., used to determine non-specific binding). Mice were sacrificed 3 min after i.v. injection of tritiated ligand. Cerebellum and spinal cord were quickly removed, homogenized, and stored in ScintiSafe. All samples were processed with a Packard 1900CA liquid scintillation analyzer.

2. For binding site occupancy at  $\alpha$ 5GABA(A)receptors, methods followed those above using [<sup>3</sup>H]L655,708 (1.5µCi/mouse, i.v.). Non-specific binding was determined using bretazenil (5.6 mg/kg, i.p.). Mice were sacrificed 1 min after injection with tritiated ligand and the forebrain was quickly removed and prepared as described above.

18. CANTAB work (Rowlett and Cook et al.)<sup>155</sup>: Preferential inverse agonist action at α5GABA<sub>A</sub> receptors resulted in enhancement of performance on memory tasks.

**18.1. Subjects:** Experiments were conducted with adult male rhesus monkeys (*Macaca mulatta*), aged 10-15 years (corresponding roughly to "middle aged" monkeys), born and reared at the New England Primate Research Center breeding facility. The monkeys were experimentally naïve at the beginning of the study. Monkeys were prepared with a non-allergic nylon vest (Lomir) and seated in primate chairs (Crisp Instruments) enclosed in ventilated, sound-attenuating chambers (Med Associates); or prepared with the non-allergic nylon vest attached to a flexible, stainless steel tether (Lomir) connected to the top of the monkey's home cage. The restraint chair system was used in CANTAB studies, and the tether system was used in ORD, observational and MAP studies. All monkeys were housed in a temperature- and humidity-controlled housing room and maintained on a 12-hr on/12-hr off light/dark schedule (lights on at 0600 hr). Water was available continuously in the home cage.

**18.2. Food availability:** The protocol for food restriction developed by Taffe (2004) was used. Briefly, the monkeys in all studies were weighed weekly, and these weights were used to calculate food (LabDiet 5038) allotments based on metabolic energy requirements for rhesus monkeys. The calculation was based on the equation:

 $y=15.504x^{-0.8219}$ , in which y=food biscuits/kg/day and x=body weight in kg (Taffe 2004). Monkeys were maintained at 100% of this allotment for at least 3 weeks or until weights are stable (no upward or downward trend over three weight determinations). Once stable, the monkeys' food allotments were gradually reduced to 70 to 85% (note that this was not a reduction in *weight*, rather a reduction in *food allotment*), an amount shown to result in stable and accurate performance in CANTAB tasks. Weekly weighing occured throughout the studies to adjust feeding (within the 70-85% range) based on a combination of weight, performance, and overall health (as determined by routine physical exams conducted by the Clinical Veterinary Staff).

**18.3.** Surgery and drug delivery: All compounds were administered via the intravenous (i.v.) route. We used this route in order to minimize pharmacokinetic variables and to allow direct comparisons with other procedures at Harvard Medical School (NERPC) designed to study BZ effects in rhesus monkeys (e.g., i.v. self-administration models of abuse potential). Monkeys were implanted with venous catheters using the surgical procedures described by Platt et al. (2005). Surgery was conducted under aseptic conditions using isoflurane/oxygen anesthesia. Catheters were passed by way of a jugular, femoral or brachial vein to the level of the right atrium. The distal end of the catheter was passed subcutaneously to a mid-scapular exit point. Catheters were flushed routinely with saline containing Heparin (100-150 U/ml), and physical exams were conducted that occasionally included contrast-dye infusion into the catheter, followed by X-ray to verify catheter patency and proper placement. Using these procedures, catheters can remain patent for up to 3 years.

**18.4. General design and data analyses:** Separate groups of 6-8 monkeys were used in the CANTAB vs. the ORD/MAP/observation studies, with each monkey in a group receiving all treatments whenever possible.

Dependent measures for the studies depended on the individual tasks, as described below. The studies employed within-subjects design, in which each subject serves as its own control. This design permits scientifically meaningful results to be obtained with fewer animals than would be required using most other approaches. For individual subject analysis, data during test sessions were compared to data from control sessions and significance of effects were determined on the basis of 95% confidence intervals established under baseline conditions. For comparison of doses within individual tasks, repeated-measures ANOVAs were used to evaluate statistical reliability when warranted. For all studies, multiple comparisons were assessed for statistical reliability using *a priori* Bonferroni t-tests, when applicable. Comparisons of drug potencies were conducted by computing  $ED_{50}$  values (dose engendering 50% of the maximum effect). The  $ED_{50}$ s were computed using non-linear regression analysis techniques.

### **18.5. CANTAB Tasks in Monkeys.**

**18.5.1** *Apparatus:* As described above, monkeys were trained to sit in primate chairs that will be placed in a sound-attenuating chamber. Inside this chamber were a computer-controlled touch-screen monitor (Campden or Lafayette Instruments, Inc,), equipped with a dispenser to deliver 190-mg flavored food pellets (BioServe, Inc.).

Experimental events were controlled by the CANTAB software system on a computer in an adjacent room.

18.5.2. Delayed non-matching to sample (DNMS). The DNMS task was a short-term recognition memory task involving sets of visual discriminations. A sample stimulus is presented in the center of the screen, and the animal must touch this stimulus within 30 sec. After a touch, the screen was blanked and following a variable retention interval, two stimuli were presented on the lower left and right of the screen. The retention interval consisted of 0, 1, 3, or 10 min; if performance was highly variable, the delays were adjusted for individual monkeys based on performance such that short, medium, and long delays were tested. Of the two "matching" stimuli, one stimulus was identical to the sample stimulus ("matching" stimulus) and the other was novel ("nonmatching" stimulus). A touch directed to the non-matching stimulus was followed by reinforcer delivery. In addition to the four retention interval conditions, a simultaneous condition was included in which the sample stimulus remained present while the matching and non-matching stimuli were present. A session consisted of 10 trials of each retention interval, presented in a randomly intermixed fashion (total trials=50). The CANTAB software used 469 shapes and 7 colors to ensure that discriminations were unique for approximately 120,000 trials. The primary dependent measured for the DNMS task are the percent correct responses at each delay and the latency to respond at each delay. The number of trials completed also was recorded as a measure of motivation and/or ability to perform the task.

18.5.3. Paired associated learning (PAL). Large colored abstract stimuli were displayed in one of 4 positions (top center, bottom center, left middle, right middle).

The subject was required to touch the sample stimulus, which then disappeared. After a 1-sec delay, the same pattern reappears (choice phase) in two or more locations on the screen (the original location plus one or more novel locations). The subject was required to touch the stimulus that was presented in the same location as the sample item to obtain a food pellet. Task difficulty was increased by increasing the number of stimulus-location associations required on each trial. Training was initiated with sessions that present 25 1-stimulus trials and 25 2-stimulus trials until performance averaged >50% correct trials on the 2-stimulus trials. Next, monkeys received sessions of 25 2-stimulus trials and 25 3-stimulus trials until performance averaged >25% correct trials on the 3-stimulus trials. This part of the PAL task measured memory for stimulus-location associations.

Next, a learning component is introduced in which a trial repeats if a mistake was made in attempting to complete this trial (i.e. the monkey gets 0-2 of 3 stimulus-locations correct in a 3-stimulus trial). Up to 6 attempts at a given trial were allowed. Finally, 4stimulus trials were added when performance of 3-stimulus trials exceeded 50% correct (thus, the PAL task measured both memory and learning of stimulus-location associations). The dependent measures were the percent correct at each difficulty level and the latencies to respond.

## 18.5.4. Intradimensional/extradimensional set shifting (ID/ED). The

ID/ED shift task evaluated the ability of a subject to attend to specific attributes of a stimulus, as well as the ability to shift attention to other attributes when required. This task consisted of a series of eight discrimination learning stages wherein touching only one of two stimuli presented on the screen resulted in food pellet delivery. Within any

given stage of the task, a pair of stimuli was presented and the same stimulus was associated with reinforcement (S+ stimulus) until the performance criteria were met (18 of 20 consecutive trials correct). Correct choices must be made within 30 sec, and following a correct choice the screen was blanked for 5 sec while an incorrect choice (S- stimulus) resulted in a 0.2 sec tone and a 9-sec period of blank screen.

The task consisted of 4 levels of stimulus sets, each progressively more difficult. In the first stimulus set (Level 1), two distinctly shaped stimuli were presented, with touches on one shape resulting in food pellet delivery. This level (and all subsequent levels) included a stimulus reversal, in which the same two shapes are retained but pressing the S+ stimulus now does not result in reinforcement, whereas touching the former Sstimulus did result in food pellet delivery. In Level 2, a compound discrimination was presented. For this discrimination, the two shapes from Level 1 are present, but additional stimuli, consisted of lines, were superimposed onto the existing shapes. Because the shape discrimination from the previous stage did not change, the lines are irrelevant to this discrimination. Level 3 consisted of the intra-dimensional (ID) shift stage. For this discrimination, two new shapes with new lines were presented. This was considered an ID shift due to the fact that despite new shapes and lines being introduced, the shape remained the relevant dimension for the discrimination. In the final level, the extra-dimensional (ED) shift will be introduced. Thus, in Level 4, new shapes and lines were presented; however, the lines—not the shape—were the relevant discriminative stimulus. Performance was determined as the number of errors at each stage, and the data were subjected to a square root transformation to achieve normal distributions.

18.5.5. Progressive-Ratio (PR). PR procedures consisted of response requirements that progressively increased across a session until responding ceases. The last response requirement completed, termed "break point", provided a quantitative measure of the reinforcing effectiveness of a stimulus. For these studies, a session was initiated by a colored box appearing in the middle of the screen. Touching the box once resulted in food pellet delivery, and the response requirement were progressively increased following each reinforcer by an incremental value beginning at 1 and doubling after 8 response requirements were completed successfully. Thus, the response requirement sequences consisted of: Increment= 1, response requirements= 1, 2, 3, 4, 5, 6, 7, 8; Increment = 2, response requirements = 10, 12, 14, 16, 18, 20, 22, 24; Increment = 4, response requirements = 28, 32, 36, 40, 44, 48, 52, 56; and so on. Sessions were a maximum of 30 min, and were terminated if 3 min elapses without a response. Performance was measured as the break point and number of food pellets delivered per session. This procedure provided a measure of the ability of the animal to respond, as well as an assessment of "motivation" to perform.

**18.5.6.** *Training*. Initial training began with shaping the basic response of touching the screen. A large colored square was displayed that occupied most of the screen. Touching the screen inside the box was a "correct" response and results in food reinforcement. With each correct response, the size of the square decreased until the monkey readily pressed a 1-in box in any location on the screen. This initial training typically required up to 5 sessions.

Currently 4 monkeys were trained at various stages of the DNMS and ID/ED task. These monkeys initiated training on PAL (a task that requires a lengthy training period, Taffe et al. 2004) with "reminder" trials for DNMS and ID/ED scheduled one day per week. We anticipated approximately 5-6 more months of training for PAL performance to reach acceptable performance.

For new monkeys, CANTAB training was initiated by gradually introducing different tasks, and also incorporate training on the ORD task. In the first phase of training, the PAL task was introduced 5 days/week. After approximately 6 months, the DNMS task was introduced on days 1-4 of the week, with PAL training continuing on day 5. In our experience, DNMS training typically took up to 3 months, once criterion performance was met, DNMS training was initiated on days 1-3, with PAL training scheduled for days 4 and 5, respectively. PAL training may take up to and over 12 months, therefore, after 9 months the ID/ED task was introduced into the sequence once or twice per week on randomly chosen days (this task did not require special training, Weed et al., 1999). In the final phase of training, two tasks per day were introduced and the PR schedule introduced according to the schedule: day 1: PR/PAL; day 2: DNMS/PR; day 3: PAL; day 4: DNMS/PR; day 5: ID/ED/PR.

	<u>Day 1</u>	<u>Day 2</u>	<u>Day 3</u>	<u>Day 4</u>	<u>Day 5</u>	<u>Day 6</u>	<u>Day 7</u>	<u>Day 8</u>
<u>Cγcle 1:</u>	PAL	ID/ED	ORD	DNMS/PR	DNMS/PR	ORD	ID/ED	PAL
	Dose A	vehicle						
<u>Cycle 2:</u>	DNMS/PR	ORD	IDÆD	PAL	PAL	ID/ED	ORD	DNMS/PR
	Dose A	vehicle						

**Table 20**: Schedule of CANTAB tests for a single dose of compound plus vehicle
**18.5.7. Testing.** Once training criteria were reached on all tasks, the 8-day cycles shown in **Table 20** were initiated. The two cycles shown in the Table allowed a dose of test compound and vehicle to be determined twice in all tasks. For each compound, vehicle plus at least 4 doses were evaluated. All doses of a compound (or dose of compound plus antagonist) were tested in irregular order, with a compound finished prior to moving to the next compound. A test session consisted of i.v. injections of compound(s) at an appropriate pretreatment time prior to the session. Based on the scheduling shown in Table 20, for CANTAB a dose-response function was required approximately 10 weeks to complete.

## 19. Object Retrieval with Detours (ORD) Task. To augment

CANTAB-based testing, parallel assessments were conducted using the ORD Task, which was a manual test sensitive to frontostriatal deficits (Taylor et al. 1990; Jentsch et al. 1999) and has been validated in pharmacological studies at NERPC. The task required the monkey to retrieve a small piece of palatable food from a transparent box that was open on one side only. The box was attached to a tray that can be positioned within easy reach of the monkey. Different levels of task complexity were achieved by varying the orientation of the open side of the box, the position of the box on the tray, and the position of the food relative to the opening. The level of cognitive complexity and motor difficulty of each task was rated *a priori* using the criteria of Taylor et al. (1990). The 12 task configurations varied from those characterized by high cognitive complexity/low motor difficulty to low cognitive complexity/high motor difficulty.

Training on the ORD task typically required 14-21 sessions, and tests were scheduled as described above in Table 20. The primary dependent measure for the ORD task was "percent success", which was the total number of reaches minus "incorrect" (i.e., a reach in which the food item was not obtained) and "barrier" reaches (i.e., a reach in which the monkey contacted a closed side of the box), divided by total reaches and then multiplied by 100. Barrier reaches were also analyzed separately as a measure of perseverative behavior, and reach latencies (i.e., time to obtain food or maximum trial length of 20 seconds) were recorded as a measure of motor impairment.

## 20. Assessment of Motor Coordination and Behavioral Effects in Monkeys.

**20.1. MAP** apparatus and procedure. To evaluate fine motor coordination, monkeys were trained using the movement assessment panel (MAP) system developed by Gash and colleagues (Gash et al. 1999). The MAP (Quanteon, Inc.) consists of a Lexan® box attached to a clear Lexan® panel that was mounted on to home cage (**Figure 41**). Portals in the front panel and on the sides of the food receptacle allowed monkeys to retrieve small food objects (e.g., miniature marshmallows or raisins). Arrays of photocells are attached to the portals on the front panel (Portal A) and food receptacle (Portal B). The amount of time that the photobeams for Portal A were broken was considered a measure of gross motor coordination, whereas the amount of time that the photobeams for Portal B were broken was considered a measure of fine motor

coordination (Gash et al. 1999). MAP studies occurred in the home cage of monkeys prepared with the tether system. The MAP apparatus fit the cage door, and monkeys received initial training by placing food treats in different parts of the apparatus until they consistently retrieved treats from the food receptacle.



Training occurred 5 days per week until latencies retrieved the treat are stable (no upward or downward trends over 3 consecutive sessions) and testing occurred on day 2 and day 5 of this 5-day cycle.

**20.1.1.** Observation studies. Observers were trained in the use of the behavioral scoring instrument developed in previous studies at NERPC. Each observer will undergo at least 20 hrs of observational training until they reached a > 90% reliability criterion based on percent agreement scores. The scoring system included 19 categories

encompassing a wide range of species-typical and drug-induced behaviors including locomotion, environmental manipulation, foraging, self-grooming, scratching, stereotypic behavior such as self-grasping, back flipping, and hair pulling, ataxia, sleep posture, moderate and deep sedation. These behavioral categories have proven to be useful for characterization of the sedative and motoric effects of BZ-type compounds in previous research at NERPC.

Observation studies occurred in the same monkeys that were trained on the MAP task. Observation studies were conducted in these monkeys in alternate weeks, on the 2<sup>nd</sup> and 3<sup>rd</sup> day of the 5-day cycle (i.e., MAP testing occurred one week, observation tests occurred the following week, and so on). Video cameras were used to record the monkeys' behavior during observation sessions. This provided an archival record of experimental sessions, permitted subsequent scoring of videotapes by independent observers, and eliminated the possibility that an observer's presence disturbed the monkeys' behavior. All behaviors were scored during 5-min sampling periods as described above. Each 5-min period was subdivided into twenty 15-s intervals, and frequency scores were calculated as the proportion of 15-s intervals in which a particular behavior occurred. Data was analyzed separately in individual subjects and for the group when appropriate used the statistical approach described earlier at NERPC.

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## III. Appendix I: Crystal Data for 110 and 111

X-ray Crystal Structure Determination for 110 and 111

Single-crystal X-ray diffraction data on **110** and **111** were collected at 296 and 100 °K respectively. All data were collected using MoK $\alpha$  radiation ( $\lambda = 0.71073$  Å) and a Bruker APEX 2 CCD area detector. Samples were prepared for data collection by coating with high viscosity microscope oil (Paratone-N, Hampton Research). The oil-coated crystal was mounted on a MicroMesh mount (MiTeGen, Inc.) and transferred immediately to the diffractometer. Corrections were applied for Lorentz, polarization, and absorption effects. All structures were solved by direct methods and refined by full-matrix least squares on  $F^2$  values using the programs found in the SHELXTL suite (Bruker, SHELXTL v6.10, 2000, Bruker AXS Inc., Madison, WI). Parameters refined included atomic coordinates and anisotropic thermal parameters for all non-hydrogen atoms.

Full information on data collection, refinement, and results of the x-ray studies are given in Tables 1 to 14

 Table 1. Crystal data and structure refinement for 110.

Empirical formula	$C_{19}H_{20}N_2O_2$	
Formula weight	308.37	
Temperature	296(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	
Unit cell dimensions	a = 9.6247(4)  Å	$\alpha = 90^{\circ}$
	b = 11.6485(5) Å	$\beta = 90^{\circ}$
	c = 13.7072(5)  Å	$\dot{\gamma} = 90^{\circ}$
Volume	1536.76(11) Å <sup>3</sup>	·
Z	4	
Density (calculated)	$1.333 \text{ Mg/m}^3$	
Absorption coefficient	$0.087 \text{ mm}^{-1}$	
F(000)	656	
Crystal size	0.549 x 0.141 x 0.104 m	m <sup>3</sup>
Theta range for data collection	2.29 to 29.23°	
Index ranges	-12<=h<=13, -15<=k<=2	15, -18<=l<=18
Reflections collected	17885	
Independent reflections	4117 [R(int) = 0.0313]	
Completeness to theta = $29.23^{\circ}$	99.4 %	
Absorption correction	Semi-empirical from equ	ivalents
Max. and min. transmission	0.9910 and 0.9537	
Refinement method	Full-matrix least-squares	s on F <sup>2</sup>
Data / restraints / parameters	4117 / 0 / 208	
Goodness-of-fit on F <sup>2</sup>	1.051	
Final R indices [I>2sigma(I)]	R1 = 0.0441, wR2 = 0.1	120
R indices (all data)	R1 = 0.0560, wR2 = 0.1	197
Absolute structure parameter	0.9(14)	
Largest diff. peak and hole	0.275 and -0.226 e.Å <sup>-3</sup>	
Absolute structure parameter Largest diff. peak and hole	0.9(14) 0.275 and -0.226 e.Å <sup>-3</sup>	

	Х	У	Z	U(eq)
N(1)	1218(2)	4427(1)	8185(1)	37(1)
C(2)	2099(2)	5027(2)	8752(1)	33(1)
C(3)	3328(2)	5685(2)	8403(1)	34(1)
N(4)	3535(1)	6617(1)	9129(1)	34(1)
C(5)	3857(2)	6092(1)	10089(1)	34(1)
C(6)	2664(2)	5348(2)	10454(1)	35(1)
O(6)	2569(2)	5140(1)	11332(1)	47(1)
C(7)	1757(2)	4884(2)	9725(1)	33(1)
C(8)	551(2)	4152(2)	9753(1)	33(1)
C(9)	-305(2)	3714(2)	10488(1)	42(1)
C(10)	-1425(2)	3030(2)	10227(2)	47(1)
C(11)	-1703(2)	2774(2)	9258(2)	49(1)
C(12)	-880(2)	3187(2)	8508(2)	43(1)
C(13)	242(2)	3876(2)	8775(1)	35(1)
C(14)	4643(2)	4903(2)	8333(1)	39(1)
C(15)	5780(2)	5425(2)	8987(1)	37(1)
C(16)	5253(2)	5408(2)	10054(1)	37(1)
C(17)	6311(2)	5866(2)	10792(2)	53(1)
O(17)	6185(2)	7054(1)	10952(1)	60(1)
C(18)	8403(2)	6398(2)	7934(2)	60(1)
C(19)	7098(2)	7033(2)	8158(1)	44(1)
C(20)	6001(2)	6639(2)	8636(1)	37(1)
C(21)	4712(2)	7351(2)	8821(2)	41(1)

**Table 2.** Atomic coordinates (  $x \ 10^4$ ) and equivalent isotropic displacement parameters (Å<sup>2</sup>  $x \ 10^3$ ) for **110.** U(eq) is defined as one third of the trace of the orthogonalized U<sup>ij</sup> tensor.

N(1)-C(2)	1.345(2)	N(1)-C(13)	1.395(2)
N(1)-H(1A)	0.8600	C(2)-C(7)	1.384(2)
C(2)-C(3)	1.489(2)	C(3)-N(4)	1.485(2)
C(3)-C(14)	1.562(3)	C(3)-H(3A)	0.9800
N(4)-C(21)	1.480(2)	N(4)-C(5)	1.484(2)
C(5)-C(6)	1.523(2)	C(5)-C(16)	1.563(2)
C(5)-H(5A)	0.9800	C(6)-O(6)	1.231(2)
C(6)-C(7)	1.432(2)	C(7)-C(8)	1.441(2)
C(8)-C(9)	1.398(2)	C(8)-C(13)	1.411(2)
C(9)-C(10)	1.387(3)	C(9)-H(9A)	0.9300
C(10)-C(11)	1.387(3)	C(10)-H(10A)	0.9300
C(11)-C(12)	1.384(3)	C(11)-H(11A)	0.9300
C(12)-C(13)	1.395(3)	C(12)-H(12A)	0.9300
C(14)-C(15)	1.541(3)	C(14)-H(14A)	0.9700
C(14)-H(14B)	0.9700	C(15)-C(20)	1.509(3)
C(15)-C(16)	1.548(2)	C(15)-H(15A)	0.9800
C(16)-C(17)	1.531(3)	C(16)-H(16A)	0.9800
C(17)-O(17)	1.406(3)	C(17)-H(17A)	0.9700
C(17)-H(17B)	0.9700	O(17)-H(17C)	0.8200
C(18)-C(19)	1.490(3)	C(18)-H(18A)	0.9600
C(18)-H(18B)	0.9600	C(18)-H(18C)	0.9600
C(19)- $C(20)$	1.325(3)	C(19)-H(19A)	0.9300
C(20)- $C(21)$	1.514(3)	C(21)-H(21A)	0.9700
C(21)-H(21B)	0.9700		0.7700
C(2)-N(1)-C(13)	109.18(14)	C(2)-N(1)-H(1A)	125.4
C(13)-N(1)-H(1A)	125.4	N(1)-C(2)-C(7)	110.16(15)
N(1)-C(2)-C(3)	125.62(14)	C(7)-C(2)-C(3)	124 10(15)
N(4)-C(3)-C(2)	105.51(13)	N(4)-C(3)-C(14)	111.05(14)
C(2)-C(3)-C(14)	111.26(14)	N(4)-C(3)-H(3A)	109.6
C(2)- $C(3)$ - $H(3A)$	109.6	C(14)-C(3)-H(3A)	109.6
C(21) - N(4) - C(5)	109.31(14)	C(21)-N(4)-C(3)	109.49(14)
C(5)-N(4)-C(3)	108.68(13)	N(4)-C(5)-C(6)	109.19(11) 111.59(14)
N(4)-C(5)-C(16)	111 26(13)	C(6)-C(5)-C(16)	111.64(13)
N(4)-C(5)-H(5A)	107.4	C(6)-C(5)-H(5A)	107.4
C(16)-C(5)-H(5A)	107.4	O(6)-C(6)-C(7)	124 21(16)
O(6)-C(6)-C(5)	119 25(16)	C(7)- $C(6)$ - $C(5)$	121.21(10) 11646(14)
C(2) - C(7) - C(6)	119.23(10)	C(2)-C(7)-C(8)	10674(14)
C(6)-C(7)-C(8)	134 12(15)	C(9)-C(8)-C(13)	100.74(14) 118 55(16)
C(9)-C(8)-C(7)	134.12(15) 135.20(16)	C(13)-C(8)-C(7)	10624(14)
C(10)-C(9)-C(8)	133.20(10) 118 74(18)	C(10)-C(0)-U(0)	120.6
C(8) C(9) H(9A)	120.6	$C(10)$ - $C(9)$ - $\Pi(9R)$	120.0 121.41(18)
C(11) C(10) H(10A)	120.0	C(11)- $C(10)$ - $C(3)$	121.41(10)
C(12) C(11) C(10)	117.3 121.72(18)	C(12) C(11) H(11A)	119.5
C(12)- $C(11)$ - $C(10)$	121.72(16)	$C(12)$ - $C(11)$ - $\Pi(11A)$ C(11) $C(12)$ $C(12)$	117.1 116.67(19)
$C(10)-C(11)-\Pi(11A)$ $C(11)-C(12)-\Pi(12A)$	117.1	C(11)-C(12)-C(13) C(13) C(12) U(12A)	121.07(18)
$C(12) - C(12) - \Pi(12A)$ $C(12) - C(12) - \Pi(12A)$	121.7 120 41(16)	$C(12) - C(12) - \Pi(12A)$ C(12) - C(12) - C(2)	121.7 122.01(16)
V(12)-V(13)-IN(1) N(1) C(12) C(9)	127.41(10) 107.67(15)	C(12)-C(13)-C(0) C(15)-C(14)-C(2)	122.91(10) 109.01(14)
$N(1)-U(13)-U(\delta)$	107.07(15)	C(13)-C(14)-C(3)	108.01(14)
C(15)-C(14)-H(14A)	110.1	C(3)-C(14)-H(14A)	110.1
U(13)-U(14)-H(14B)	110.1	C(3)-C(14)-H(14B)	110.1
H(14A)-U(14)-H(14B)	108.4	C(20)-C(15)-C(14)	100.53(15)
C(20)- $C(15)$ - $C(16)$	111.10(15)	C(14)-C(15)-C(16)	108.14(14)
C(20)-C(15)-H(15A)	110.3	C(14)-C(15)-H(15A)	110.3

 Table 3. Bond lengths [Å] and angles [°] for 110.

Table 3. (continued)

С(16)-С(15)-Н(15А)	110.3	C(17)-C(16)-C(15)	113.68(15)
C(17)-C(16)-C(5)	111.96(15)	C(15)-C(16)-C(5)	107.71(13)
C(17)-C(16)-H(16A)	107.8	C(15)-C(16)-H(16A)	107.8
C(5)-C(16)-H(16A)	107.8	O(17)-C(17)-C(16)	112.84(17)
O(17)-C(17)-H(17A)	109.0	C(16)-C(17)-H(17A)	109.0
O(17)-C(17)-H(17B)	109.0	C(16)-C(17)-H(17B)	109.0
H(17A)-C(17)-H(17B)	107.8	C(17)-O(17)-H(17C)	109.5
C(19)-C(18)-H(18A)	109.5	C(19)-C(18)-H(18B)	109.5
H(18A)-C(18)-H(18B)	109.5	C(19)-C(18)-H(18C)	109.5
H(18A)-C(18)-H(18C)	109.5	H(18B)-C(18)-H(18C)	109.5
C(20)-C(19)-C(18)	127.06(19)	C(20)-C(19)-H(19A)	116.5
C(18)-C(19)-H(19A)	116.5	C(19)-C(20)-C(15)	126.46(17)
C(19)-C(20)-C(21)	123.16(16)	C(15)-C(20)-C(21)	110.16(14)
N(4)-C(21)-C(20)	111.02(14)	N(4)-C(21)-H(21A)	109.4
C(20)-C(21)-H(21A)	109.4	N(4)-C(21)-H(21B)	109.4
C(20)-C(21)-H(21B)	109.4	H(21A)-C(21)-H(21B)	108.0

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
N(1)	39(1)	46(1)	27(1)	0(1)	1(1)	-5(1)
C(2)	33(1)	36(1)	30(1)	1(1)	2(1)	1(1)
C(3)	34(1)	38(1)	30(1)	2(1)	2(1)	-4(1)
N(4)	33(1)	33(1)	38(1)	2(1)	2(1)	1(1)
C(5)	36(1)	34(1)	32(1)	-3(1)	0(1)	-4(1)
C(6)	37(1)	36(1)	30(1)	-2(1)	3(1)	-1(1)
O(6)	54(1)	58(1)	27(1)	-1(1)	2(1)	-13(1)
C(7)	33(1)	37(1)	30(1)	1(1)	3(1)	-3(1)
C(8)	33(1)	35(1)	31(1)	2(1)	1(1)	2(1)
C(9)	42(1)	48(1)	36(1)	4(1)	5(1)	-1(1)
C(10)	41(1)	47(1)	53(1)	8(1)	13(1)	-8(1)
C(11)	40(1)	47(1)	59(1)	3(1)	0(1)	-11(1)
C(12)	41(1)	43(1)	45(1)	-2(1)	-4(1)	-6(1)
C(13)	34(1)	37(1)	34(1)	0(1)	1(1)	1(1)
C(14)	42(1)	36(1)	39(1)	-4(1)	10(1)	-2(1)
C(15)	32(1)	32(1)	47(1)	2(1)	5(1)	3(1)
C(16)	37(1)	33(1)	41(1)	5(1)	-1(1)	-2(1)
C(17)	50(1)	54(1)	54(1)	7(1)	-13(1)	-5(1)
O(17)	42(1)	63(1)	76(1)	-22(1)	1(1)	-12(1)
C(18)	48(1)	61(1)	72(2)	-5(1)	20(1)	-5(1)
C(19)	44(1)	42(1)	47(1)	1(1)	9(1)	-6(1)
C(20)	35(1)	35(1)	42(1)	2(1)	0(1)	-3(1)
C(21)	39(1)	33(1)	50(1)	8(1)	5(1)	-2(1)

**Table 4.** Anisotropic displacement parameters (Å<sup>2</sup> x 10<sup>3</sup>) for **110**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2$ [ h<sup>2</sup> a<sup>\*2</sup>U<sup>11</sup> + ... + 2 h k a<sup>\*</sup> b<sup>\*</sup> U<sup>12</sup> ]

	Х	У	Z	U(eq)
H(1A)	1251	4390	7559	44
H(3A)	3128	6019	7762	41
H(5A)	3978	6719	10557	40
H(9A)	-127	3878	11140	50
H(10A)	-2001	2738	10711	56
H(11A)	-2462	2312	9109	58
H(12A)	-1066	3013	7859	52
H(14A)	4421	4131	8550	47
H(14B)	4963	4863	7663	47
H(15A)	6643	4984	8929	44
H(16A)	5050	4609	10225	44
H(17A)	6188	5466	11407	63
H(17B)	7241	5701	10558	63
H(17C)	6767	7260	11352	91
H(18A)	9023	6889	7577	90
H(18B)	8190	5733	7548	90
H(18C)	8838	6162	8531	90
H(19A)	7048	7787	7939	53
H(21A)	4904	7913	9325	49
H(21B)	4462	7761	8231	49

**Table 5.** Hydrogen coordinates (  $x \ 10^4$ ) and isotropic displacement parameters (Å<sup>2</sup> x 10<sup>3</sup>) for **110**.

C(13)-N(1)-C(2)-C(7)	0.8(2)	C(13)-N(1)-C(2)-C(3)	176.81(16)
N(1)-C(2)-C(3)-N(4)	152.43(16)	C(7)-C(2)-C(3)-N(4)	-32.1(2)
N(1)-C(2)-C(3)-C(14)	-87.0(2)	C(7)-C(2)-C(3)-C(14)	88.4(2)
C(2)-C(3)-N(4)-C(21)	-177.83(13)	C(14)-C(3)-N(4)-C(21)	61.50(17)
C(2)-C(3)-N(4)-C(5)	62.83(16)	C(14)-C(3)-N(4)-C(5)	-57.84(17)
C(21)-N(4)-C(5)-C(6)	178.61(14)	C(3)-N(4)-C(5)-C(6)	-61.93(16)
C(21)-N(4)-C(5)-C(16)	-55.97(17)	C(3)-N(4)-C(5)-C(16)	63.48(16)
N(4)-C(5)-C(6)-O(6)	-158.11(16)	C(16)-C(5)-C(6)-O(6)	76.7(2)
N(4)-C(5)-C(6)-C(7)	25.1(2)	C(16)-C(5)-C(6)-C(7)	-100.12(18)
N(1)-C(2)-C(7)-C(6)	173.05(16)	C(3)-C(2)-C(7)-C(6)	-3.0(3)
N(1)-C(2)-C(7)-C(8)	-1.0(2)	C(3)-C(2)-C(7)-C(8)	-177.07(16)
O(6)-C(6)-C(7)-C(2)	-169.54(18)	C(5)-C(6)-C(7)-C(2)	7.1(2)
O(6)-C(6)-C(7)-C(8)	2.5(3)	C(5)-C(6)-C(7)-C(8)	179.14(18)
C(2)-C(7)-C(8)-C(9)	-178.1(2)	C(6)-C(7)-C(8)-C(9)	9.1(4)
C(2)-C(7)-C(8)-C(13)	0.80(19)	C(6)-C(7)-C(8)-C(13)	-171.94(19)
C(13)-C(8)-C(9)-C(10)	-0.2(3)	C(7)-C(8)-C(9)-C(10)	178.7(2)
C(8)-C(9)-C(10)-C(11)	0.2(3)	C(9)-C(10)-C(11)-C(12)	-0.1(3)
C(10)-C(11)-C(12)-C(13)	-0.1(3)	C(11)-C(12)-C(13)-N(1)	-178.39(19)
C(11)-C(12)-C(13)-C(8)	0.2(3)	C(2)-N(1)-C(13)-C(12)	178.43(18)
C(2)-N(1)-C(13)-C(8)	-0.3(2)	C(9)-C(8)-C(13)-C(12)	0.0(3)
C(7)-C(8)-C(13)-C(12)	-179.15(17)	C(9)-C(8)-C(13)-N(1)	178.81(16)
C(7)-C(8)-C(13)-N(1)	-0.32(19)	N(4)-C(3)-C(14)-C(15)	-4.76(18)
C(2)-C(3)-C(14)-C(15)	-121.97(15)	C(3)-C(14)-C(15)-C(20)	-56.62(17)
C(3)-C(14)-C(15)-C(16)	62.88(17)	C(20)-C(15)-C(16)-C(17)	-65.57(19)
C(14)-C(15)-C(16)-C(17)	177.85(16)	C(20)-C(15)-C(16)-C(5)	59.09(18)
C(14)-C(15)-C(16)-C(5)	-57.49(17)	N(4)-C(5)-C(16)-C(17)	121.05(16)
C(6)-C(5)-C(16)-C(17)	-113.56(17)	N(4)-C(5)-C(16)-C(15)	-4.63(18)
C(6)-C(5)-C(16)-C(15)	120.75(15)	C(15)-C(16)-C(17)-O(17)	90.4(2)
C(5)-C(16)-C(17)-O(17)	-31.9(2)	C(18)-C(19)-C(20)-C(15)	-5.5(4)
C(18)-C(19)-C(20)-C(21)	-179.5(2)	C(14)-C(15)-C(20)-C(19)	-108.6(2)
C(16)-C(15)-C(20)-C(19)	133.8(2)	C(14)-C(15)-C(20)-C(21)	65.96(18)
C(16)-C(15)-C(20)-C(21)	-51.6(2)	C(5)-N(4)-C(21)-C(20)	65.37(19)
C(3)-N(4)-C(21)-C(20)	-53.6(2)	C(19)-C(20)-C(21)-N(4)	164.91(18)
C(15)-C(20)-C(21)-N(4)	-9.9(2)		

 Table 6. Torsion angles [°] for cook106.

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
N(1)-H(1A)O(6)#1	0.86	2.10	2.8409(19)	143.7
O(17)-H(17C)N(4)#2	0.82	2.25	2.743(2)	119.5

**Table 7.** Hydrogen bonds for **110** [Å and  $^{\circ}$ ].

Symmetry transformations used to generate equivalent atoms: #1 -x+1/2,-y+1,z-1/2 #2 x+1/2,-y+3/2,-z+2 
 Table 8. Crystal data and structure refinement for 111.

Empirical formula	$C_{21}H_{30}N_2O_4$	
Formula weight	374.47	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P 21 21 21	
Unit cell dimensions	a = 8.3186(7)  Å	$\alpha = 90^{\circ}$
	b = 12.4731(10) Å	$\beta = 90^{\circ}$
	c = 19.1532(16) Å	$\gamma = 90^{\circ}$
Volume	1987.3(3) Å <sup>3</sup>	
Z	4	
Density (calculated)	$1.252 \text{ Mg/m}^3$	
Absorption coefficient	0.086 mm <sup>-1</sup>	
F(000)	808	
Crystal size	0.463 x 0.429 x 0.091 mm <sup>3</sup>	
Theta range for data collection	1.95 to 29.14°	
Index ranges	-11<=h<=11, -17<=k<=16, -24	<=l<=26
Reflections collected	22977	
Independent reflections	5326 [R(int) = 0.0221]	
Completeness to theta = $29.14^{\circ}$	99.6 %	
Absorption correction	Semi-empirical from equivalen	ts
Max. and min. transmission	0.9931 and 0.9658	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	5326 / 2 / 252	
Goodness-of-fit on F <sup>2</sup>	1.103	
Final R indices [I>2sigma(I)]	R1 = 0.0356, $wR2 = 0.1041$	
R indices (all data)	R1 = 0.0367, wR2 = 0.1055	
Absolute structure parameter	0.3(7)	
Largest diff. peak and hole	0.329 and -0.273 e.Å <sup>-3</sup>	

	Х	У	Z	U(eq)
 N(1)	8034(1)	7267(1)	1454(1)	19(1)
C(2)	9226(1)	6894(1)	1016(1)	16(1)
C(3)	9215(1)	5825(1)	663(1)	16(1)
N(4)	10222(1)	5969(1)	23(1)	14(1)
C(5)	11870(1)	6313(1)	231(1)	14(1)
C(6)	11787(1)	7506(1)	449(1)	14(1)
O(6)	13218(1)	7827(1)	813(1)	18(1)
C(7)	10356(1)	7674(1)	908(1)	15(1)
C(8)	9838(1)	8606(1)	1287(1)	17(1)
C(9)	10416(1)	9662(1)	1352(1)	21(1)
C(10)	9558(2)	10381(1)	1759(1)	25(1)
C(11)	8144(2)	10067(1)	2108(1)	26(1)
C(12)	7541(2)	9036(1)	2053(1)	23(1)
C(13)	8393(1)	8315(1)	1633(1)	19(1)
C(14)	9850(1)	4886(1)	1124(1)	19(1)
C(15)	11491(1)	4539(1)	832(1)	18(1)
C(16)	12574(1)	5545(1)	793(1)	17(1)
C(17)	14318(1)	5261(1)	620(1)	22(1)
O(18)	15055(1)	4601(1)	1132(1)	29(1)
C(19)	12250(2)	2212(1)	268(1)	40(1)
C(20)	11531(2)	3126(1)	-134(1)	27(1)
C(21)	11204(1)	4106(1)	107(1)	19(1)
C(22)	10360(1)	4933(1)	-348(1)	18(1)
O(1S)	5487(1)	6196(1)	2188(1)	26(1)
C(1S)	6066(2)	6072(2)	2882(1)	39(1)
O(2S)	13630(1)	7920(1)	2198(1)	34(1)
C(2S)	12416(2)	8163(1)	2687(1)	34(1)

**Table 9.** Atomic coordinates (  $x \ 10^4$ ) and equivalent isotropic displacement parameters (Å<sup>2</sup>  $x \ 10^3$ ) for **111**. U(eq) is defined as one third of the trace of the orthogonalized U<sup>ij</sup> tensor.

N(1)-C(2)	1.3796(14)	N(1)-C(13)	1.3849(16)
N(1)-H(1A)	0.8800	C(2)-C(7)	1.3681(15)
C(2)-C(3)	1.4956(15)	C(3)-N(4)	1.4957(14)
C(3)-C(14)	1.5580(15)	C(3)-H(3A)	1.0000
N(4)-C(22)	1.4790(13)	N(4)-C(5)	1.4898(14)
C(5)-C(6)	1.5469(14)	C(5)-C(16)	1.5569(15)
C(5)-H(5A)	1.0000	C(6)-O(6)	1.4362(13)
C(6)-C(7)	1.4942(14)	C(6)-H(6A)	1.0000
O(6)-H(6B)	0.8400	C(7)-C(8)	1.4366(15)
C(8)-C(9)	1.4073(16)	C(8)-C(13)	1.4191(16)
C(9)-C(10)	1.3856(17)	C(9)-H(9A)	0.9500
C(10)-C(11)	1.4089(19)	C(10)-H(10A)	0.9500
C(11)-C(12)	1.384(2)	C(11)-H(11A)	0.9500
C(12)-C(13)	1.3988(16)	C(12)-H(12A)	0.9500
C(14)-C(15)	1.5377(16)	C(14)-H(14A)	0.9900
C(14)-H(14B)	0.9900	C(15)-C(21)	1.5081(17)
C(15)-C(16)	1.5462(16)	C(15)-H(15A)	1.0000
C(16)-C(17)	1.5302(16)	C(16)-H(16A)	1.0000
C(17)-O(18)	1.4209(15)	C(17)-H(17A)	0.9900
C(17)-H(17B)	0.9900	O(18)-H(18A)	0.8400
C(19)-C(20)	1.4994(19)	C(19)-H(19A)	0.9800
C(19)-H(19B)	0.9800	C(19)-H(19C)	0.9800
C(20)-C(21)	1.3352(17)	C(20)-H(20A)	0.9500
C(21)-C(22)	1.5219(16)	C(22)-H(22D)	0.9900
C(22)-H(22E)	0.9900	O(1S)-C(1S)	1.4221(17)
O(1S)-H(1S)	0.8400	C(1S)-H(1SC)	0.9800
C(1S)-H(1SB)	0.9800	C(1S)-H(1SA)	0.9800
O(2S)-C(2S)	1.4099(17)	O(2S)-H(2S)	0.8402(10)
C(2S)-H(2SA)	0.9800	C(2S)-H(2SB)	0.9800
C(2S)-H(2SC)	0.9800		
C(2)-N(1)-C(13)	108.26(10)	C(2)-N(1)-H(1A)	125.9
C(13)-N(1)-H(1A)	125.9	C(7)-C(2)-N(1)	110.26(10)
C(7)-C(2)-C(3)	124.79(10)	N(1)-C(2)-C(3)	124.78(10)
C(2)-C(3)-N(4)	105.01(8)	C(2)-C(3)-C(14)	114.34(9)
N(4)-C(3)-C(14)	111.40(9)	C(2)-C(3)-H(3A)	108.6
N(4)-C(3)-H(3A)	108.6	C(14)-C(3)-H(3A)	108.6
C(22)-N(4)-C(5)	107.98(8)	C(22)-N(4)-C(3)	109.36(8)
C(5)-N(4)-C(3)	109.37(8)	N(4)-C(5)-C(6)	107.96(8)
N(4)-C(5)-C(16)	110.68(8)	C(6)-C(5)-C(16)	114.89(9)
N(4)-C(5)-H(5A)	107.7	C(6)-C(5)-H(5A)	107.7
C(16)-C(5)-H(5A)	107.7	O(6)-C(6)-C(7)	109.63(8)
O(6)-C(6)-C(5)	111.23(8)	C(7)-C(6)-C(5)	109.26(9)
O(6)-C(6)-H(6A)	108.9	C(7)-C(6)-H(6A)	108.9
C(5)-C(6)-H(6A)	108.9	C(6)-O(6)-H(6B)	109.5
C(2)-C(7)-C(8)	107.02(10)	C(2)-C(7)-C(6)	122.41(10)
C(8)-C(7)-C(6)	130.54(10)	C(9)-C(8)-C(13)	119.14(10)
C(9)-C(8)-C(7)	134.35(11)	C(13)-C(8)-C(7)	106.46(10)
C(10)-C(9)-C(8)	118.66(11)	C(10)-C(9)-H(9A)	120.7
C(8)-C(9)-H(9A)	120.7	C(9)-C(10)-C(11)	121.11(12)
C(9)-C(10)-H(10A)	119.4	C(11)-C(10)-H(10A)	119.4
C(12)- $C(11)$ - $C(10)$	121.62(11)	C(12)-C(11)-H(11A)	119.2
C(10)-C(11)-H(11A)	119.2	C(11)-C(12)-C(13)	117.21(12)

 Table 10.
 Bond lengths [Å] and angles [°] for 110.

Table 10. (continued)

C(11)-C(12)-H(12A)	121.4	C(13)-C(12)-H(12A)	121.4
N(1)-C(13)-C(12)	129.79(11)	N(1)-C(13)-C(8)	107.97(10)
C(12)-C(13)-C(8)	122.22(11)	C(15)-C(14)-C(3)	107.85(9)
C(15)-C(14)-H(14A)	110.1	C(3)-C(14)-H(14A)	110.1
C(15)-C(14)-H(14B)	110.1	C(3)-C(14)-H(14B)	110.1
H(14A)-C(14)-H(14B)	108.5	C(21)-C(15)-C(14)	107.15(9)
C(21)-C(15)-C(16)	109.83(9)	C(14)-C(15)-C(16)	107.80(9)
C(21)-C(15)-H(15A)	110.7	C(14)-C(15)-H(15A)	110.7
C(16)-C(15)-H(15A)	110.7	C(17)-C(16)-C(15)	111.98(9)
C(17)-C(16)-C(5)	110.40(9)	C(15)-C(16)-C(5)	108.26(9)
C(17)-C(16)-H(16A)	108.7	C(15)-C(16)-H(16A)	108.7
C(5)-C(16)-H(16A)	108.7	O(18)-C(17)-C(16)	113.16(10)
O(18)-C(17)-H(17A)	108.9	C(16)-C(17)-H(17A)	108.9
O(18)-C(17)-H(17B)	108.9	C(16)-C(17)-H(17B)	108.9
H(17A)-C(17)-H(17B)	107.8	C(17)-O(18)-H(18A)	109.5
C(20)-C(19)-H(19A)	109.5	C(20)-C(19)-H(19B)	109.5
H(19A)-C(19)-H(19B)	109.5	C(20)-C(19)-H(19C)	109.5
H(19A)-C(19)-H(19C)	109.5	H(19B)-C(19)-H(19C)	109.5
C(21)-C(20)-C(19)	126.91(14)	C(21)-C(20)-H(20A)	116.5
C(19)-C(20)-H(20A)	116.5	C(20)-C(21)-C(15)	127.85(11)
C(20)-C(21)-C(22)	121.13(11)	C(15)-C(21)-C(22)	110.94(9)
N(4)-C(22)-C(21)	110.67(9)	N(4)-C(22)-H(22D)	109.5
C(21)-C(22)-H(22D)	109.5	N(4)-C(22)-H(22E)	109.5
C(21)-C(22)-H(22E)	109.5	H(22D)-C(22)-H(22E)	108.1
C(1S)-O(1S)-H(1S)	109.5	O(1S)-C(1S)-H(1SC)	109.5
O(1S)-C(1S)-H(1SB)	109.5	H(1SC)-C(1S)-H(1SB)	109.5
O(1S)-C(1S)-H(1SA)	109.5	H(1SC)-C(1S)-H(1SA)	109.5
H(1SB)-C(1S)-H(1SA)	109.5	C(2S)-O(2S)-H(2S)	110.8(3)
O(2S)-C(2S)-H(2SA)	109.5	O(2S)-C(2S)-H(2SB)	109.5
H(2SA)-C(2S)-H(2SB)	109.5	O(2S)-C(2S)-H(2SC)	109.5
H(2SA)-C(2S)-H(2SC)	109.5	H(2SB)-C(2S)-H(2SC)	109.5

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
N(1)	13(1)	25(1)	18(1)	-1(1)	4(1)	-1(1)
C(2)	13(1)	21(1)	13(1)	1(1)	1(1)	1(1)
C(3)	12(1)	19(1)	15(1)	3(1)	1(1)	-2(1)
N(4)	13(1)	16(1)	14(1)	0(1)	0(1)	-1(1)
C(5)	12(1)	17(1)	13(1)	1(1)	0(1)	0(1)
C(6)	11(1)	19(1)	12(1)	0(1)	1(1)	-1(1)
O(6)	14(1)	26(1)	14(1)	0(1)	0(1)	-5(1)
C(7)	13(1)	19(1)	14(1)	0(1)	1(1)	1(1)
C(8)	14(1)	22(1)	14(1)	-2(1)	-1(1)	2(1)
C(9)	19(1)	24(1)	21(1)	-3(1)	-1(1)	-1(1)
C(10)	24(1)	25(1)	28(1)	-8(1)	-3(1)	2(1)
C(11)	23(1)	33(1)	23(1)	-10(1)	-1(1)	8(1)
C(12)	18(1)	34(1)	18(1)	-5(1)	2(1)	4(1)
C(13)	14(1)	26(1)	15(1)	-2(1)	0(1)	3(1)
C(14)	18(1)	21(1)	18(1)	6(1)	2(1)	-1(1)
C(15)	17(1)	19(1)	20(1)	5(1)	0(1)	1(1)
C(16)	13(1)	20(1)	16(1)	5(1)	-2(1)	2(1)
C(17)	14(1)	27(1)	25(1)	5(1)	-1(1)	3(1)
O(18)	24(1)	32(1)	32(1)	5(1)	-8(1)	9(1)
C(19)	41(1)	25(1)	54(1)	3(1)	-6(1)	13(1)
C(20)	27(1)	24(1)	32(1)	-1(1)	-3(1)	6(1)
C(21)	17(1)	18(1)	23(1)	3(1)	0(1)	0(1)
C(22)	20(1)	16(1)	17(1)	-1(1)	-2(1)	-1(1)
O(1S)	19(1)	37(1)	21(1)	6(1)	0(1)	-1(1)
C(1S)	33(1)	57(1)	25(1)	9(1)	-5(1)	0(1)
O(2S)	29(1)	56(1)	18(1)	-3(1)	-2(1)	11(1)
C(2S)	38(1)	41(1)	23(1)	5(1)	7(1)	13(1)

**Table 11.** Anisotropic displacement parameters (Å<sup>2</sup> x 10<sup>3</sup>) for **110**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2$ [ h<sup>2</sup> a<sup>\*2</sup>U<sup>11</sup> + ... + 2 h k a<sup>\*</sup> b<sup>\*</sup> U<sup>12</sup> ]

	X	У	Z	U(eq)
H(1A)	7191	6900	1596	22
H(3A)	8089	5657	516	19
H(5A)	12571	6265	-192	16
H(6A)	11666	7957	21	17
H(6B)	13840	8142	535	26
H(9A)	11375	9876	1123	25
H(10A)	9928	11098	1802	30
H(11A)	7592	10575	2390	31
H(12A)	6589	8826	2290	28
H(14A)	9965	5125	1614	23
H(14B)	9087	4278	1111	23
H(15A)	11990	3979	1136	22
H(16A)	12547	5916	1256	20
H(17A)	14347	4886	164	26
H(17B)	14946	5931	572	26
H(18A)	15198	4957	1500	44
H(19A)	12363	1588	-40	60
H(19B)	11547	2028	661	60
H(19C)	13309	2421	445	60
H(20A)	11281	2993	-610	33
H(22D)	10974	5035	-785	21
H(22E)	9274	4670	-472	21
H(1S)	4845	6716	2172	39
H(1SC)	6968	6564	2959	58
H(1SB)	6429	5333	2951	58
H(1SA)	5202	6236	3213	58
H(2S)	13284(9)	7984(18)	1788(1)	51
H(2SA)	12095	8915	2635	51
H(2SB)	12826	8045	3161	51
H(2SC)	11485	7699	2606	51

**Table 12.** Hydrogen coordinates (  $x \ 10^4$ ) and isotropic displacement parameters (Å<sup>2</sup> x 10<sup>3</sup>) for **110**.
C(13)-N(1)-C(2)-C(7)	0.34(13)	C(13)-N(1)-C(2)-C(3)	-175.08(10)
C(7)-C(2)-C(3)-N(4)	-20.77(14)	N(1)-C(2)-C(3)-N(4)	153.99(10)
C(7)-C(2)-C(3)-C(14)	101.63(12)	N(1)-C(2)-C(3)-C(14)	-83.61(13)
C(2)-C(3)-N(4)-C(22)	177.07(9)	C(14)-C(3)-N(4)-C(22)	52.78(11)
C(2)-C(3)-N(4)-C(5)	59.02(10)	C(14)-C(3)-N(4)-C(5)	-65.27(11)
C(22)-N(4)-C(5)-C(6)	166.16(8)	C(3)-N(4)-C(5)-C(6)	-74.93(10)
C(22)-N(4)-C(5)-C(16)	-67.31(10)	C(3)-N(4)-C(5)-C(16)	51.60(11)
N(4)-C(5)-C(6)-O(6)	165.99(8)	C(16)-C(5)-C(6)-O(6)	41.96(12)
N(4)-C(5)-C(6)-C(7)	44.82(10)	C(16)-C(5)-C(6)-C(7)	-79.21(11)
N(1)-C(2)-C(7)-C(8)	-1.32(13)	C(3)-C(2)-C(7)-C(8)	174.09(10)
N(1)-C(2)-C(7)-C(6)	-179.39(10)	C(3)-C(2)-C(7)-C(6)	-3.98(17)
O(6)-C(6)-C(7)-C(2)	-130.32(11)	C(5)-C(6)-C(7)-C(2)	-8.19(14)
O(6)-C(6)-C(7)-C(8)	52.11(15)	C(5)-C(6)-C(7)-C(8)	174.24(11)
C(2)-C(7)-C(8)-C(9)	-175.80(13)	C(6)-C(7)-C(8)-C(9)	2.1(2)
C(2)-C(7)-C(8)-C(13)	1.77(12)	C(6)-C(7)-C(8)-C(13)	179.63(11)
C(13)-C(8)-C(9)-C(10)	0.58(17)	C(7)-C(8)-C(9)-C(10)	177.92(12)
C(8)-C(9)-C(10)-C(11)	0.79(19)	C(9)-C(10)-C(11)-C(12)	-1.1(2)
C(10)-C(11)-C(12)-C(13)	-0.09(19)	C(2)-N(1)-C(13)-C(12)	178.83(12)
C(2)-N(1)-C(13)-C(8)	0.81(12)	C(11)-C(12)-C(13)-N(1)	-176.27(12)
C(11)-C(12)-C(13)-C(8)	1.51(18)	C(9)-C(8)-C(13)-N(1)	176.43(10)
C(7)-C(8)-C(13)-N(1)	-1.59(12)	C(9)-C(8)-C(13)-C(12)	-1.78(17)
C(7)-C(8)-C(13)-C(12)	-179.79(11)	C(2)-C(3)-C(14)-C(15)	-109.10(11)
N(4)-C(3)-C(14)-C(15)	9.75(12)	C(3)-C(14)-C(15)-C(21)	-64.15(11)
C(3)-C(14)-C(15)-C(16)	54.00(11)	C(21)-C(15)-C(16)-C(17)	-72.68(12)
C(14)-C(15)-C(16)-C(17)	170.90(9)	C(21)-C(15)-C(16)-C(5)	49.27(12)
C(14)-C(15)-C(16)-C(5)	-67.15(11)	N(4)-C(5)-C(16)-C(17)	135.38(9)
C(6)-C(5)-C(16)-C(17)	-102.04(11)	N(4)-C(5)-C(16)-C(15)	12.47(12)
C(6)-C(5)-C(16)-C(15)	135.05(10)	C(15)-C(16)-C(17)-O(18)	-62.10(13)
C(5)-C(16)-C(17)-O(18)	177.19(10)	C(19)-C(20)-C(21)-C(15)	0.8(2)
C(19)-C(20)-C(21)-C(22)	-175.55(14)	C(14)-C(15)-C(21)-C(20)	-120.57(14)
C(16)-C(15)-C(21)-C(20)	122.61(13)	C(14)-C(15)-C(21)-C(22)	56.12(12)
C(16)-C(15)-C(21)-C(22)	-60.71(12)	C(5)-N(4)-C(22)-C(21)	56.56(11)
C(3)-N(4)-C(22)-C(21)	-62.36(11)	C(20)-C(21)-C(22)-N(4)	-176.72(11)
C(15)-C(21)-C(22)-N(4)	6.34(13)		

Table 13.Torsion angles [°] for 110.

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
N(1)-H(1A)O(1S)	0.88	2.02	2.8717(13)	163.7
O(6)-H(6B)N(4)#1	0.84	1.92	2.7569(12)	172.2
O(18)-H(18A)O(1S)#2	0.84	2.05	2.8593(15)	162.8
O(1S)-H(1S)O(2S)#3	0.84	1.81	2.6472(15)	173.4
O(2S)-H(2S)O(6)	0.8402(10)	1.880(3)	2.6783(13)	158.3(7)

Table 14. Hydrogen bonds for 110 [Å and °].

Symmetry transformations used to generate equivalent atoms: #1 x+1/2,-y+3/2,-z #2 x+1,y,z #3 x-1,y,z

## IV. Appendix II: Data on QH-II-066

In vivo and in vitro profiles of the  $GABA_A/\alpha 5$ -preferring agonist QH-ii-066 and the nonselective agonist diazepam

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In order to investigate the possible role of  $GABA_A/\alpha 5$  receptors in behavioral effects of benzodiazepines (BZ), the effects of QH-ii-066, the GABA<sub>A</sub>/ $\alpha$ 5-preferring agonist, were compared with diazepam in binding and electrophysiology assays and behavioral models of anxiety, epilepsy, and side effects (activity, motor coordination, and subjective effects). QH-ii-066, while having similar affinity for BZ-sensitive GABAA receptor subtypes, showed functional selectivity for  $GABA_A/\alpha 5$  receptors over GABA<sub>A</sub>/ $\alpha$ 1 receptors. QH-ii-066 increased punished responding in the Geller conflict test in rats and protected against pentylenetetrazole (PTZ)-induced seizures in mice at doses similar to those of diazepam, but was less potent than diazepam in protecting against electroconvulsive shock (ECS)-induced seizures, suggesting that  $GABA_A/\alpha 1$ receptors may play a bigger role in ECS- than PTZ-induced seizures. OH-ii-066 showed a larger separation between doses effective in the conflict model and those effective in the conditioned fear, locomotor activity, rotorod, and chlordiazepoxide drug discrimination. In the case of locomotor activity and chlordiazepoxide drug discrimination, the decreased potency of QH-ii-066 compared with diazepam indicate that GABA<sub>A</sub>/a1 receptors play a more prominent role in sedative and chlordiazepoxide-like subjective effects of BZs. These data suggest that compounds like QH-ii-066 with reduced efficacy at  $GABA_A/\alpha 1$ receptors may be effective anxiolytics with a reduced potential for side effects.

Although benzodiazepines (BZ) are the most frequently prescribed anxiolytics, their use is accompanied by significant side effects, most notably, sedation, ataxia, memory impairment, potentiation of their effects by ethanol, development of tolerance and dependence, and abuse liability. Efforts to discover anxiolytics with reduced side effect profiles have centered on two types of compounds, those acting by a different mechanism (e.g., 5-HT<sub>1A</sub> receptor agonists and CRF<sub>1</sub> receptor antagonists) and BZ-like compounds showing functional selectivity, i.e., acting with high efficacy at some GABA<sub>A</sub> receptors and with low or no efficacy on other GABA<sub>A</sub> receptors. The latter approach assumes that the therapeutic and adverse effects of classical BZs are mediated by different GABA<sub>A</sub> receptors and that reducing binding affinity and/or efficacy at receptors mediating adverse effects will result in a superior anxiolytic compared with classical BZs.

Zolpidem and zaleplon, two compounds with selectivity for GABA<sub>A</sub>/ $\alpha$ 1 receptors, decrease sleep latency and increase sleep duration (e.g., Fry et al., 2000) and are marketed for treatment of insomnia. In addition, recent studies with transgenic mice and GABA<sub>A</sub> receptor-selective compounds have indicated that GABA<sub>A</sub>/ $\alpha$ 1 receptors mediate sedative/ataxic effects of classical BZs (Rudolph et al., 1999; McKernan et al., 2000; Kralic et al., 2002).

The role of other BZ-sensitive GABA<sub>A</sub> receptors, containing  $\alpha 2$ ,  $\alpha 3$ , or  $\alpha 5$  subunits, in the behavioral effects of classical BZs is less clear. Mice with a point mutation rendering GABA<sub>A</sub>/ $\alpha 2$  receptors insensitive to diazepam (GABA<sub>A</sub>/ $\alpha 2$  receptor knock-in mice) do not exhibit anxiolytic or myorelaxant effects when treated with diazepam, whereas GABA<sub>A</sub>/ $\alpha 3$  receptor knock-in mice showed a normal anxiolytic response to diazepam, with reduced myorelaxant effects (Löw et al., 2000; Crestani et al., 2001). However, selective GABA<sub>A</sub>/ $\alpha 3$  receptor inverse agonists appear to be anxiogenic (Collins et al., 2002), indicating that GABA<sub>A</sub>/ $\alpha 3$  receptors may be involved in anxiolytic effects of BZs. In addition, compounds with higher intrinsic efficacy at GABA<sub>A</sub>/ $\alpha 5$  receptors, and lower or no efficacy at GABA<sub>A</sub>/ $\alpha 1$  and GABA<sub>A</sub>/ $\alpha 5$  receptors,

show anxiolytic and anticonvulsant effects in rodents (Griebel et al., 2001). These studies suggest that both  $GABA_A/\alpha 2$  and  $GABA_A/\alpha 3$  receptors may play a role in the anxiolytic effects of classical BZs. Thus, an ideal compound, an anxiolytic with reduced motoric side effect profile, would have no affinity or, at least, no efficacy at the  $GABA_A/\alpha 1$  receptors and high affinity and high efficacy at  $GABA_A/\alpha 2$  and  $GABA_A/\alpha 3$  receptors.

The role of GABA<sub>A</sub>/ $\alpha$ 5 receptors remains to be elucidated. Thus far, only one study used GABA<sub>A</sub>/ $\alpha$ 5 receptor knock-in mice to investigate the role of GABA<sub>A</sub>/ $\alpha$ 5 receptors. Collinson et al. (2002) showed that these mice show no overt phenotype: their baseline anxiety-related response, as measured by elevated plus maze, did not differ from those of wild-type mice, they showed no spontaneous seizures, no impairment in rotorod performance, and normal anxiolytic-like responses to chlordiazepoxide (CDP). These animals did show faster learning of a Morris water maze task, suggesting that GABA<sub>A</sub>/ $\alpha$ 5 receptors may play a role in the memory-impairing effects of classical BZs. This result is consistent with the finding that most GABA<sub>A</sub>/ $\alpha$ 5 receptors are located in the hippocampus, a brain region shown to be involved in learning and memory. Furthermore, inverse agonists at this site show memory-enhancing effects in animals (Chambers et al., 2003), and some of these compounds also show anxiogenic and proconvulsant effects (Navarro et al., 2002).

As is the case with GABA<sub>A</sub>/ $\alpha$ 2- and GABA<sub>A</sub>/ $\alpha$ 3 receptor-selective compounds, studies with agonists selective for GABA<sub>A</sub>/ $\alpha$ 5 receptors are scarce. QH-ii-066 (7-acetyleno-1,3-dihydro-1-methyl-5-phenyl-2H-1,4-benzodiazepin-2-one), a compound with 6-fold selectivity for GABA<sub>A</sub>/ $\alpha$ 5 receptors compared with GABA<sub>A</sub>/ $\alpha$ 2 and GABA<sub>A</sub>/ $\alpha$ 3 receptors and 11-fold selectivity compared with GABA<sub>A</sub>/ $\alpha$ 1 receptors (Huang et al., 2000), differs from classical BZs in that it did not share subjective effects with triazolam in squirrel monkeys (Lelas et al., 2002). Like classical BZs, QH-ii-066 protected against seizures induced by BZ inverse agonists DMCM and RY-24 (Skolnick et al., 1997).

The present studies were designed to compare the effects of QH-ii-066 and diazepam in an array of *in vitro* and *in vivo* assays. QH-ii-066 is a 7-acetyleno congener of diazepam (Figure 1) and thus, diazepam was chosen as the classical BZ for comparison. The two compounds were characterized in binding and electrophysiology functional assays (HEK whole-cell patch clamp) and in models of anxiety (conflict), pentylenetetrazole (PTZ)- and electroconvulsive shock (ECS)- induced seizures, learning and memory (conditioned fear), sedation (locomotor activity), ataxia (rotorod performance), and BZ-like subjective effects (CDP drug discrimination) in rodents. Differences observed between QH-ii-066 and diazepam in these assays would provide an indication of the potential contribution of  $GABA_A/\alpha5$  receptors to these effects of classical BZs.

## **METHODS**

#### Subjects

All animals were maintained in accordance with the guidelines of the Committee on Animals of the Bristol-Myers Squibb Company and the "Guide for Care and Use of Laboratory Animals" (Institute of Animal Laboratory Resources, 1996). Research protocols were approved by the Bristol-Myers Squibb Company Institutional Animal Care and Use Committee.

Male CDF rats weighing approximately 350 g at the time of the experiment (conflict), male Sprague-Dawley rats weighing 180-300 g (locomotor activity, rotorod studies, and drug discrimination studies) and male CF1 mice weighing 20-22 g at the time of the experiment (seizure studies) were purchased from Charles River Laboratories (Wilmington, Mass., USA). For the conditioned fear studies, male Sprague-Dawley rats weighing approximately 225 g at the time of the experiment were purchased from Harlan (Indianapolis, Ind., USA). Animals were housed in pairs or fours in clear plastic shoebox cages with Beta Chip bedding in rooms maintained at constant temperature (21-23°C) and humidity (50  $\pm$  10%) with a 12 h light/dark cycle (lights on at 0600 h, except for mice lights were on at 0500 h), except for the drug discrimination study for which animals were housed individually in suspended wire cages.

Animals had *ad libitum* access to food and water throughout the studies, except for the conflict and drug discrimination studies in which animals were restricted to 9 g (conflict) or 12 g (drug discrimination) of laboratory rodent chow (Bio-Serv, Frenchtown, NJ) per day in addition to the food earned in operant box, with unlimited access to water. All training and testing was done between 0600 and 1300 h Monday through Friday of each week.

## **Procedures**

#### **GABA**<sub>A</sub> Binding Assay

HEK293 cells were trypsinized and counted. Cells were transfected using the Amaxa electroporator and 3 million cells per sample. Each transfection was performed in Amaxa solution "V" using electroporation program Q-01 (Amaxa Inc., Gaithersburg, Md., USA). 1  $\mu$ g of each GABA subunit ( $\alpha$ ,  $\beta$ , and  $\gamma$ ) was used per transfection. Following electroporation, cells were plated onto poly-lysin coated glass coverslips and used for the experiments.

## **Electrophysiological Studies**

Whole-cell patch-clamp experiments were performed using HEK cell lines transiently expressing two different configurations of GABA<sub>A</sub> constructs, on the first or second day after the transfection. Cells were plated onto 25 mm Petri dishes and viewed using Nikon TE200 inverted microscope, equipped with the mercury lamp to monitor GFP fluorescence. Cells were continuously perfused with extracellular physiological solution containing (in mM): NaCl 140, KCl 2.5, CaCl<sub>2</sub> 2, MgSO<sub>4</sub> 2, Na<sub>2</sub>HPO<sub>4</sub> 1.25, HEPES 10, Glucose 11, pH 7.4 and 340 mOsm (modified from Brown et al., 2002). Patch pipettes were pulled using Narishige vertical puller from fire-polished borosillicate glass capillaries. Pipette resistance was between 3.5 and 6 MOm. The intracellular solution contained (in mM): CsCl 140, HEPES 10, EGTA 5, MgATP 2, MgCl<sub>2</sub> 3, pH 7.3 and 340 mOsm.

Cells were voltage-clamped at -20 mV via an EPC10 amplifier (HEKA Electronics, Southboro, Mass.). Preliminary experiments have shown that amplitude of the GABA-induced current in these cells was not significantly different at -60, -40 or -20 mV holding potentials. At the same time, -20 mV holding potential facilitated longer experimental time which enabled testing of several concentrations of the test compounds. Drug solutions were applied to the cells via a multi-channel local perfusion system

(AutoMate Scientific, Inc., San Francisco, Calif.) with the estimated solution change time of 20 ms.

GABA was applied to the cell for 20 s with a 60 s washout period between applications, which was found to be a sufficient time period to reverse any desensitization that may occur. In addition, to ensure that there is no desensitization that may affect GABA responses at any given concentration of a GABA modulator, GABA was applied twice during control (no drug) and once at each concentration of the test compound. Test compounds were applied in increasing concentrations (1, 10, 100, or 1000 nM) 60 s prior to GABA (3  $\mu$ M) applications.

#### **Geller Conflict with Incremental Shock in Rats**

Experiments were conducted in 8 model E10-10 Coulbourn operant chambers (Coulbourn Instruments, Allentown, Penn., USA; 28 x 26 x 31 cm) with an electrified grid floor which were housed in light-proof, sound-attenuated, and fan-ventilated chambers. On one wall of each operant chamber was a houselight and a centrally placed food hopper 3 cm from the floor that delivered one 45-mg food pellet (Dustless Precision Pellets, Bio-Serv, Frenchtown, N.J., USA). To the left was mounted a lever response module with cue lights above. Foot-shock was delivered using an H13-16 Universal Shocker. Experiments were controlled with a Compaq DeskPro computer running Graphic State Notation software through Habitest Universal Linc interface equipment (Coulbourn Instruments).

Rats were trained to stability on an alternating multiple Random Interval 30 s/Fixed Ratio 1 Food +Shock paradigm (RI15/FR1 F+S). The paradigm ran through four cycles beginning with the Random Interval (RI) schedule which ran for 12 minutes then switched to the Fixed Ratio (FR) schedule which was cued and ran for 3 minutes. The RI had a range of 15 to 45 s with the food pellet being delivered with the first lever press after the expiration of the interval. The FR schedule delivered a food pellet after each lever press along with a foot-shock that increased with each lever press beginning at zero on the first lever press then increasing to 0.05 mA on the second lever press and incrementing by

0.05 mA with each subsequent lever press. The shock was reset to zero at the beginning of each RI/FR cycle.

#### Protection against Pentylenetetrazole- and Shock-Induced Seizures in Mice

Pentylenetetrazole (PTZ) was administered at 125 mg/kg s.c. The number of animals surviving was recorded at 30 minutes and 60 minutes after administration of PTZ. Electroshock is administered using a Ugo Basile ECT, Unit 7801 seizure apparatus (Ugo Basile, Italy) and corneal electrodes soaked in 0.9% saline. Mice received a shock of 30 mA for 0.3 s. The number of animals that exhibited the hind-limb extensor component of the seizure was recorded.

## **Conditioned Fear in Rats**

Experiments were conducted in standard modular test chambers (ENV-008, Med Associates, St. Albans, Vt., USA, 30.5 x 24.1 x 21 cm) with an electrified grid floor which were housed in light-proof, sound-attenuated, and fan-ventilated chambers. Foot-shock was delivered using a shocker/scrambler module (ENV-414). Experiments were controlled with a desktop computer using Med-PC software through Smart Ctrl<sup>™</sup> interface equipment (Med-Associates).

On day 1, rats were placed in the test chambers for a 7-minute conditioning session consisting of 2.5 minutes of habituation, a 2-s foot-shock at 0.7 mA, followed by 2.5 minutes of no foot-shock, another 2-s, 0.7 mA foot-shock, and another 2 minutes with no foot-shock. Twenty-four hours after the conditioning session, rats were placed in the test chamber again for 7 minutes with no foot-shocks and percent time freezing was quantified by video image analysis (Freezeview software using filter 25 and bout length of 0.75 s; Actimetrics and Coulbourn Instruments, Allentown, Penn., USA).

#### **Spontaneous Locomotor Activity in Rats**

The testing apparatus consisted of Plexiglas chambers (42 x 42 x 30 cm) equipped with Digiscan activity monitors (Omnitech Electronics, Columbus, Ohio, USA) that detect interruptions of eight photobeams spaced 5 cm apart and 2.5 cm above the floor. Horizontal activity was recorded in 5-minute bins for a total of 60 minutes and expressed as total distance covered (in cm).

## **Rotorod Performance in Rats**

The degree of motor coordination or balance was determined using a standard accelerating rotorod treadmill (Ugo Basile, Comerio-Varese, Italy) that was 6 cm in diameter and 24 cm above the base. The speed was increased gradually from 2 rpm to a maximum speed of 20 rpm. The time each animal remained on the rotating rod was automatically recorded, up to a maximum of 5 minutes. Three acclimation trials were conducted for each animal prior to administration of drugs. The time on rotorod from the third trial was used to counterbalance animals for subsequent drug testing.

## **Chlordiazepoxide Discrimination in Rats**

Twelve model E10-10 Coulbourn operant conditioning chambers (Coulbourn Instruments, Allentown, Penn., USA; 28 x 26 x 31 cm) were housed in light-proof, soundattenuated, and fan-ventilated chambers. Each operant conditioning chamber was equipped with two non-retractable levers, requiring a downward force equivalent to 15 g (0.15 N), that were mounted 3 cm from the side wall, 3 cm above the metal grid floor, and 5 cm from a centrally placed food tray that delivered one 45-mg food pellet (Dustless Precision Pellets, Bio-Serv, Frenchtown, N.J.). The experimental chambers were connected to a Micro PDP11/73 computer using a LAB LINC interface. A SKED-11 operating system (State System, Kalamazoo, Mich.) was used to record and control behavior.

After habituation to the operant conditioning chamber, rats were trained to alternate daily between response levers on an FR 1 schedule of food reinforcement. Once lever pressing was well established, the reinforcement contingency was increased incrementally to an FR 10 schedule of food reinforcement, while maintaining the lever alternation. Rats were then trained to discriminate between an intraperitoneal (IP) injection of drug (5.0 mg/kg CDP) and vehicle (0.25% methylcellulose) administered 30

minutes prior to the start of the session. Ten consecutive responses on the injectionappropriate lever resulted in food delivery. Responses on the incorrect lever did not result in pellet delivery. Following an injection of CDP, responding on the right lever was reinforced for half the rats and responding on the left lever was reinforced for the other rats. In each two-week period there were five drug days and five saline days, with the constraint that there not be more than three consecutive drug or vehicle days.

Drug test sessions were conducted once a week (Fridays) with training sessions scheduled on intervening days (Monday – Thursday). The criterion for testing was achieved if no more than three incorrect responses occurred prior to the first reinforcer for at least 9 out of 10 consecutive sessions. During test sessions, the lever on which the rat first made 10 consecutive responses resulting in delivery of the first food pellet was designated the "selected " lever and subsequent pellet delivery was made contingent on responding on this selected lever on an FR10 schedule. The selected lever, the number of responses prior to the first food pellet delivery, and the total number of responses in the session were recorded for each rat. The duration of test sessions was ten minutes. Doseresponse functions were determined for CDP, diazepam, and QH-ii-066.

#### Drugs

GABA, diazepam, PTZ, and CDP HCl were purchased from Sigma (St. Louis, Mo., USA). QH-ii-066 was synthesized by Xiaoyan Li, Wenyuan Yin, and Dongmei Han at the University of Wisconsin. Compounds were prepared as suspensions in an aqueous vehicle of 0.25% methylcellulose (Type A15c, Dow Chemicals), except for diazepam in the conditioned fear study which was prepared in 40% propylene glycol / 10% ethanol / 50% benzoic acid solution. Stock suspensions were bead-milled overnight using three layers of 4-mm glass beads. Dilutions were made using the 0.25% methylcellulose vehicle. Compounds were administered orally by gavage (PO) 60 minutes before

behavioral testing in a volume of 2 ml/kg body weight. Doses of all drugs were calculated and are expressed in terms of the free base weight.

#### **Data Analysis**

In the electrophysiology experiments, the amplitude of the  $I_{GABA}$  in each cell was normalized to  $I_{GABA}$  produced by 3  $\mu$ M GABA in the absence of drug in the same cell. Data (% increase of  $I_{GABA}$ ) from 3-6 cells for each concentration were averaged and concentration-response curves for each test compound were constructed using ExcelFit. All data are presented as mean  $\pm$  1 SEM.

Results for conflict, locomotor activity, and rotorod performance are expressed as the mean  $\pm 1$  SEM. Results for the PTZ- and ECS-induced seizures are presented as the percent of animals protected from death and percent of animals protected from the hindlimb extensor component of the seizure, respectively. Results for the drug discrimination study are expressed as the percent of animals selecting the CDP-appropriate lever. A drug was considered to substitute fully for CDP if 75% of the animals selected the CDPappropriate lever. Response rates were expressed as percentage of control rate (average of the vehicle training session immediately before and the vehicle session immediately after the test) for individual animals, averaged across animals, and plotted as a function of dose (mean  $\pm 1$  SEM). All data were subjected to analysis of variance (ANOVA) followed by individual mean comparisons using Fisher's Least Significant Difference Test (Kirk, 1968) where appropriate. The significance level was set at *P*<0.05. ED<sub>50</sub> values were calculated by linear regression.

#### **GABA**<sub>A</sub> Binding Assay

Binding affinities of QH-ii-066 and diazepam at different rat GABA<sub>A</sub> receptor subtypes are shown in Table 1. QH-ii-066 showed similar affinity for the six rat GABA<sub>A</sub> receptors examined (range of  $K_i = 40-72$  nM). Similarly, diazepam showed comparable binding affinity at all GABA<sub>A</sub> receptor subtypes studied (range of  $K_i$  values was 20-28 nM). QH-ii-066 had a 2- to 3-fold lower affinity than diazepam at the six receptors studied.

#### **Electrophysiological Studies**

Application of GABA resulted in a dose-dependent increase in Cl<sup>-</sup> currents in GABA<sub>A</sub>/ $\alpha$ 1 $\beta$ 2 $\gamma$ 2 and GABA<sub>A</sub>/ $\alpha$ 5 $\beta$ 2 $\gamma$ 2 receptors transiently expressed in HEK cells (data not shown). The EC<sub>50</sub> value for GABA was 7.55±0.58. Diazepam and QH-ii-066 both potentiated GABA-induced Cl- currents at both GABA<sub>A</sub>/ $\alpha$ 1 $\beta$ 2 $\gamma$ 2 and GABA<sub>A</sub>/ $\alpha$ 5 $\beta$ 2 $\gamma$ 2 receptors in a concentration-dependent manner (Figure 2). At the GABA<sub>A</sub>/ $\alpha$ 1 $\beta$ 2 $\gamma$ 2 receptors, maximum potentiation of I<sub>GABA</sub> by diazepam occurred at 1  $\mu$ M and reached 389% of control, whereas maximum potentiation of I<sub>GABA</sub> by QH-ii-066 occurred at 100 nM and reached only 200% (at 1  $\mu$ M, the potentation of I<sub>GABA</sub> by QH-ii-066 reached 180% of control). In contrast, at the GABA<sub>A</sub>/ $\alpha$ 5 $\beta$ 2 $\gamma$ 2 receptors, maximum potentiation of I<sub>GABA</sub> by diazepam and QH-ii-066 occurred at the same concentration (1  $\mu$ M) and reached similar levels (282% and 289% of control, respectively).

#### **Geller Conflict with Incremental Shock in Rats**

Diazepam and QH-ii-066 increased punished responding without altering the rate of responding in the unpunished component (Figure 3). Diazepam significantly increased responding in the punished component to 178% of control at the dose of 3.0 mg/kg PO [F(3,21)=3.38, P=0.04]. QH-ii-066 significantly increased punished responding to 149% of control at the dose of 10 mg/kg PO [F(3,15)=3.7, P=0.04]. Thus, QH-ii-066 is 3-fold less potent than diazepam in increasing punished responding in rats. The doses of

diazepam and QH-ii-066 that produced a significant increase in punished responding did not alter rates of responding in the punished component [F(3,21)=1.72, P=0.19 for diazepam; F(3,15)=3.14, P=0.06 for QH-ii-066].

## Protection from Pentylenetetrazole- and Shock-Induced Seizures in Mice

Diazepam and QH-ii-066 both fully protected against both PTZ- (at 1.0 and 10 mg/kg PO, respectively) and ECS- (10 and 100 mg/kg PO, respectively) induced seizures in mice (Figure 4). The ED<sub>50</sub> values for the protective effects of diazepam and QH-ii-066 are 0.27 mg/kg and <3 mg/kg against PTZ-induced seizures, respectively. The ED<sub>50</sub> values for the protective effects of diazepam and QH-ii-066 are 0.88 mg/kg and 18.57 mg/kg against ECS-induced seizures, respectively. Thus, QH-ii-066 is approximately 11-fold and 21-fold less potent in protecting against PTZ- and ECS-induced seizures, respectively, than diazepam. Both compounds are more potent in protecting against PTZ-than ECS-induced seizures, 3-fold more potent in the case of diazepam and more than 6-fold more potent in the case of QH-ii-066.

#### **Conditioned Fear in Rats**

For both diazepam and QH-ii-066 experiments, vehicle-treated rats showed an increase in freezing observed on test day (day 2) compared with animals who did not receive shock (diazepam study: F(1,95)=19.88, P<0.0001; QH-ii-066 study: F(1,95)=13.27, P<0.0004). Diazepam significantly decreased freezing on test day relative to the vehicle-treated group (Figure 5, left panel). A dose of diazepam comparable to that which was effective in the conflict model (2.5 mg/kg) decreased percent freezing by 44% (F(1,95)=6.46, P=0.01) and a higher dose (5.0 mg/kg) also significantly decreased percent freezing by 39% (F(1,95)=5.11, P=0.03). Unlike diazepam, QH-ii-066 did not alter freezing on test day relative to the vehicle-treated group at doses up to and including 10 mg/kg (Figure 5, right panel; F(1,95)=0.5, P=0.48), the lowest effective dose in the conflict model.

#### **Spontaneous Locomotor Activity in Rats**

Diazepam and QH-ii-066 dose-dependently decreased spontaneous locomotor activity recorded 60-120 minutes after oral administration (corresponding to the 60-min pretreatment time and 60-min test duration in the conflict test) [diazepam F(5,42)=16.66, P<0.0001, Figure 6, left panel; QH-ii-066 F(6,49)=4.17, P=0.002, Figure 6, right panel]. The lowest effective dose of diazepam in the conflict model (3.0 mg/kg) did not alter locomotor activity; however, a 3-fold higher dose (10 mg/kg) resulted in a significant reduction of locomotor activity (by 34%, P=0.001), as did the doses of 30 and 100 mg/kg (38%, P=0.0004 and 75%, P<0.0001, respectively). In comparison, the lowest effective dose of QH-ii-066 in the conflict model (10 mg/kg) also did not alter locomotor activity; however, a 10-fold higher dose (100 mg/kg) resulted in a significant decrease in locomotor activity (by 49%; P=0.04).

## **Rotorod Performance in Rats**

Diazepam dose-dependently decreased motor coordination in rats as measured by time on rotorod [F(3,28)=5.43, P<0.005, Figure 7, left panel]. The dose of 100 mg/kg of diazepam resulted in a significant decrease in time on rotorod to 72% of control (*P*=0.001). The dose of 3.0 mg/kg of diazepam, the lowest effective dose of diazepam in the conflict model, did not alter the time on rotorod (*P*=0.84); thus, the difference between the lowest effective dose of diazepam in the conflict model of diazepam in the conflict model compared with rotorod performance is 33-fold.

Unlike diazepam, QH-ii-066 did not alter time on rotorod in rats up to 100 mg/kg [F(3,27)=1.0, P=0.41, Figure 7, right panel]. The lowest effective dose of QH-ii-066 in the conflict model (10 mg/kg) did not alter rotorod performance (P=0.96) and neither did the highest dose of QH-ii-066 tested (100 mg/kg, P=0.44). Thus, the difference between the lowest effective dose of QH-ii-066 in the conflict model compared with rotorod performance is greater than 10-fold.

#### **Chlordiazepoxide Discrimination in Rats**

All rats acquired stimulus control with the training dose of CDP of 5.0 mg/kg (IP, administered 30 minutes before session). These data were reported by Lelas et al. (2003). Briefly, the mean response rates (responses/min  $\pm$  SEM) for vehicle and CDP were 67.1  $\pm$  8.5 and 91.7  $\pm$  9.8, respectively. The ED<sub>50</sub> value for CDP was 1.7 mg/kg. All three doses of CDP significantly increased response rates relative to vehicle control rates.

Both diazepam and QH-ii-066 dose-dependently increased the percent of rats selecting the CDP lever (Figure 8). At the lowest effective dose of diazepam in the conflict model (3.0 mg/kg) 86% of animals selected the CDP lever, and at 10 mg/kg 100% of animals selected the CDP lever. The ED<sub>50</sub> value for diazepam is 2.82 mg/kg. In comparison, the lowest effective dose of QH-ii-066 in the conflict model (10 mg/kg) resulted in 63% of animals selecting the CDP lever, and at 30 mg/kg 100% of animals selected the CDP lever. The ED<sub>50</sub> value for QH-ii-066 is 8.56 mg/kg. Thus, in the case of diazepam, there is no separation between the doses effective in the conflict model and those resulting in CDP-lever selection, whereas for QH-ii-066 this separation is 3-fold.

Diazepam increased mean response rates, expressed as percent of control, at 5.0 mg/kg to 158% of control [t(6)=2.69, P=0.04)], but did not alter rates of responding at the lower (1.0 mg/kg) or the higher (10 mg/kg) dose tested. The highest dose of QH-ii-066 tested (30 mg/kg) decreased response rates to 77% of control [t(8)=2.38, P=0.04].

#### DISCUSSION

The purpose of the present studies was to compare the *in vitro* and *in vivo* profiles of diazepam, a non-selective classical BZ agonist, and QH-ii-066, a 7-acetyleno congener of diazepam. In contrast to the previously published data (Huang et al., 2000), the present study showed that QH-ii-066 did not bind selectively to any GABA<sub>A</sub> receptor subtype. Affinity of QH-ii-066 at GABA<sub>A</sub>/ $\alpha$ 1, GABA<sub>A</sub>/ $\alpha$ 2, GABA<sub>A</sub>/ $\alpha$ 3, and GABA<sub>A</sub>/ $\alpha$ 5 receptors was similar, ranging between 40 and 72 nM. Binding affinity of QH-ii-066 and diazepam at the GABA<sub>A</sub>/ $\alpha$ 1 and GABA<sub>A</sub>/ $\alpha$ 5 receptors was similar in the presence of either  $\beta$ 2 or  $\beta$ 3 subunits. In the previous publication (Huang et al., 2000), QH-ii-066 showed a 6-fold selectivity for  $GABA_A/\alpha 5$  over  $GABA_A/\alpha 2$  and  $GABA_A/\alpha 3$  receptors and 11-fold selectivity over  $GABA_A/\alpha 1$  receptors. Thus, the difference in binding affinity for the GABA $_{A}/\alpha$ 5 receptors between the two studies is 6-fold. It is possible that this discrepancy is due to the fact that the Huang et al. (2000) study used cells expressing human GABA receptors, whereas the present study used rat GABA receptors. When compared with diazepam, binding data from the present study show that the 7-acetyleno congener of diazepam exhibits approximately 2-fold lower affinity for all BZ-sensitive GABAA receptor subtypes.

In contrast to the binding data, electrophysiology studies showed a functional selectivity of QH-ii-066 compared with diazepam. The intrinsic efficacy of QH-ii-066 at the GABA<sub>A</sub>/ $\alpha$ 5 receptor was similar to the efficacy of diazepam, as measured by potentiation of GABA-induced Cl<sup>-</sup> currents in HEK cells transiently expressing GABA<sub>A</sub>/ $\alpha$ 5 receptors. However, QH-ii-066 was approximately half as efficacious as diazepam in HEK cells transiently expressing GABA<sub>A</sub>/ $\alpha$ 1 receptors. Thus, while QH-ii-066 did not show binding selectivity, it did exhibit functional selectivity, with diazepam-like efficacy at GABA<sub>A</sub>/ $\alpha$ 5 receptors and partial efficacy at GABA<sub>A</sub>/ $\alpha$ 1 receptors.

*In vivo*, the effects of QH-ii-066 and diazepam were compared directly in behavioral models of anxiety, epilepsy, and side effects (sedation, motor incoordination,

and subjective effects). QH-ii-066 and diazepam administered orally significantly increased punished responding in the Geller conflict test with incremental shock, an index of anxiolytic-like activity in rats. QH-ii-066 was 3-fold less potent in this model compared with diazepam. This loss of potency could be accounted for by lower affinity of QH-ii-066 at all BZ-sensitive GABA<sub>A</sub> receptors. The present data suggest that QH-ii-066 retained anxiolytic-like activity despite loss of efficacy at the GABA<sub>A</sub>/ $\alpha$ 1 receptors. These data support the conclusion from studies using the GABA<sub>A</sub>/ $\alpha$ 1 receptor knock-out or knock-in mice showing normal anxiolytic-like response to diazepam in these animals (Kralic et al., 2002; Rudolph et al., 1999; McKernan et al., 2000), suggesting that GABA<sub>A</sub>/ $\alpha$ 1 receptors do not play an important role in the anxiolytic effects of BZs. While efficacy of QH-ii-066 at the GABA<sub>A</sub>/ $\alpha$ 2 or GABA<sub>A</sub>/ $\alpha$ 3 receptors was not investigated in the present study, the similar effects of QH-ii-066 and diazepam in these receptors.

In addition to the anxiolytic-like effects, QH-ii-066 and diazepam also showed protective effects against PTZ- and ECS-induced seizures. Diazepam was 3-fold less potent in protecting against ECS-induced seizures compared with PTZ-induced seizures; this difference was increased to 6-fold in the case of QH-ii-066. When the two compounds are compared in potency in protecting against each type of seizures, diazepam was 11-fold more potent than QH-ii-066 in protecting against PTZ-induced seizures and 21-fold more potent against ECS-induced seizures. Thus, it appears that the 7-acetyleno substituted diazepam analog QH-ii-066 showed a greater loss of potency in protecting against ECS-induced seizures relative to diazepam than against PTZ-induced seizures. These data may suggest that GABA<sub>A</sub>/ $\alpha$ 1 receptors play a more prominent role in the ECS-induced seizures than in PTZ-induced seizures. SL651498, a compound with lower efficacy at GABA<sub>A</sub>/ $\alpha$ 1 receptors (Griebel et al., 2001), and bretazenil, a low-efficacy agonist at many GABA<sub>A</sub> subtypes including GABA<sub>A</sub>/ $\alpha$ 1 receptors (Martin et al., 1988), were also more potent in reversing PTZ- than ECS-induced seizures. In addition, Griebel

et al. (1999) showed that  $\beta$ -CCT, an GABA<sub>A</sub>/ $\alpha$ 1 receptor-preferring antagonist, did not alter protective effects of diazepam and zolpidem against PTZ-induced seizures, whereas it blocked the effects of the agonists against isoniazid-induced seizures. Taken together, these data suggest that PTZ-induced seizures are less dependent on the activity at the GABA<sub>A</sub>/ $\alpha$ 1 receptors than ECS-induced seizures.

The conditioned fear model has been used to study "cognitive-affective" processes in anxiety disorders (e.g., Uys et al., 2003). In this model, animals are exposed to pairings of a neutral stimulus (conditioned stimulus (CS); e.g., tone or context) with an aversive stimulus (unconditioned stimulus (UCS); e.g., shock). Following acquisition, behaviors typically seen after the presentation of UCS (e.g., startle, freezing, or suppression of waterdrinking) are now seen when the CS is presented alone. BZs reduced expression of UCSappropriate behaviors in the presence of the CS (e.g., Davis, 1979; Sanger and Joly, 1985). The same effects are obtained following administration of scopolamine (e.g., Rudy, 1996; Lindner et al., 2003), a memory-impairing agent without any anxiolytic-like activity, suggesting that the conditioned fear assay is more sensitive to memory-impairing rather than anxiolytic-like effects of BZs. In the present study, diazepam at the doses of 2.5 and 5.0 mg/kg PO resulted in a decrease in freezing, whereas QH-ii-066 did not affect freezing at the doses up to and including 10 mg/kg, PO. Thus, QH-ii-066 appears to be 2-fold less potent in producing memory deficits in the conditioned fear test. It has been suggested that GABA<sub>A</sub>/ $\alpha$ 5 receptors mediate memory-impairing effects of BZs (e.g., Collinson et al., 2002). Our electrophysiology studies showed comparable efficacy of QH-ii-066 and diazepam in potentiating the GABA currents through the GABA<sub>A</sub>/ $\alpha$ 5 receptors. The observed difference in potency in the conditioned fear assay could be due to lower affinity of QH-ii-066 at all GABA<sub>A</sub> receptors studied. Therefore, higher doses of QH-ii-066 would need to be tested to determine if lower efficacy of QH-ii-066 at GABA<sub>A</sub>/ $\alpha$ 1 receptors compared with diazepam would result in lower potential to produce memory deficits in animals. However, the present study indicates that QH-ii-066 may have a better therapeutic index than diazepam with respect to memory impairment since the doses of diazepam effective in the conflict and conditioned fear models effects were, whereas QHii-066 showed at least a 3-fold separation between doses effective in these tests.

Effects of QH-ii-066 and diazepam on motoric behaviors were assessed using the locomotor activity assay as an index of sedative effects and the rotorod performance as an index of motor incoordination. QH-ii-066 was 10-fold less potent than diazepam in reducing locomotor activity and at least 3-fold less potent in reducing time on rotorod. These differences in potency to produce sedation and motor incoordination lead to a better therapeutic index for QH-ii-066 compared with diazepam, despite a 3-fold loss of potency in producing anxiolytic-like effects (Table 2). Thus, QH-ii-066 shows a 10-fold or greater separation between its effects on motoric behavior and its anxiolytic-like effects in rats. By comparison, diazepam showed only a 3-fold separation in dose between the sedative and anxiolytic-like effects.

Since QH-ii-066 has only 2-fold lower affinity for the six GABA<sub>A</sub> receptors compared with diazepam, the 10-fold loss of potency of QH-ii-066 in reducing locomotor activity suggests that diminished intrinsic efficacy of QH-ii-066 at the GABA<sub>A</sub>/ $\alpha$ 1 receptors also contributed to its reduced potency in producing sedative effects. These data support studies with transgenic mice and GABA<sub>A</sub>/ $\alpha$ 1-selective agonists in indicating that GABA<sub>A</sub>/ $\alpha$ 1 receptors mediate sedative/ataxic effects of classical BZs (e.g., Rudolph et al., 1999; McKernan et al., 2000; Kralic et al., 2002; Fry et al., 2000).

Finally, effects of QH-ii-066 and diazepam were compared in animals trained to discriminate CDP. Drug discrimination may be used as an index of abuse liability when compounds of the same pharmacological class are compared. In the present studies, animals were trained to discriminate CDP, a classical BZ with no apparent receptor selectivity. Diazepam at 3.0 mg/kg and QH-ii-066 at 30 mg/kg fully substituted for CDP, suggesting that at some dose, both compounds produce CDP-like subjective effects. As

was the case for locomotor activity, lower affinity of QH-ii-066 at GABA<sub>A</sub> receptors alone does not account entirely for the lower potency of QH-ii-066 compared with diazepam in substituting for CDP, suggesting that GABA<sub>A</sub>/ $\alpha$ 1 receptors may be an important part of the CDP stimulus in rat drug discrimination. In addition, these data also suggest that compounds like QH-ii-066 with lower efficacy at GABA<sub>A</sub>/ $\alpha$ 1 receptors may have reduced abuse liability compared with classical BZs.

In summary, QH-ii-066, the 7-acetyleno congener of diazepam, showed a 2-fold loss of affinity at all GABA<sub>A</sub> receptors studied relative to diazepam. However, QH-ii-066 also showed a loss of efficacy at the GABA<sub>A</sub>/ $\alpha$ 1 receptors compared with diazepam, while retaining diazepam-like efficacy at the GABA<sub>A</sub>/ $\alpha$ 5 receptors. *In vivo*, QH-ii-066 showed anxiolytic-like and protective effects against PTZ-induced seizures in rodents at doses similar to those of diazepam. QH-ii-066 was less potent than diazepam in protecting against ECS-induced seizures, indicating that ECS-induced seizures are mediated by GABA<sub>A</sub>/ $\alpha$ 1 receptors to a greater extent than PTZ-induced seizures. QH-ii-066 was also less potent than diazepam in reducing activity and in producing CDP-like subjective effects, suggesting that QH-ii-066 and other compounds with lower intrinsic efficacy at the GABA<sub>A</sub>/ $\alpha$ 1 receptors may be effective anxiolytics in humans with a reduced potential for side effects.

# Table 1.

Binding affinity of QH-ii-066 and diazepam for rat GABA<sub>A</sub> receptors.

	GABA <sub>A</sub> Receptor K <sub>i</sub> (nM)					
Compound	α1β2γ2	α1β3γ2	α2β3γ2	α3β3γ2	α5β2γ2	α5β3γ2
QH-ii-066	52±3*	56±1***	51±2**	72±8**	44±1**	40±8***
Diazepam	18	20	20	28	22	20
Ratio <sup>a</sup>	2.9	2.8	2.6	2.6	2.0	2.0

\* n=5 determinations

\*\* n=3 determinations

\*\*\* n=2 determinations

<sup>a</sup> Ratio of binding affinity, QH-ii-066:diazepam

Table 2.

Lowest effective doses (mg/kg) and therapeutic indices for QH-ii-066 and diazepam in behavioral models.

Assay	QH-ii-066	Diazepam	Ratio
Conflict	10	3.0	3-fold
Conditioned Fear	>10	2.5	>4-fold
Locomotor Activity (LMA)	100	10	10-fold
Rotorod	>100	100	≥3-fold
Chlordiazepoxide	30	3.0	10-fold
Drug Discrimination (DD)			
Conflict:LMA Ratio	10-fold	3-fold	
Conflict:Rotorod Ratio	>10	33-fold	
Conflict:DD Ratio	3-fold	0	

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#### **LEGENDS FOR FIGURES**

Figure 1. Chemical structures of QH-ii-066 and diazepam.

Figure 2. Effects of diazepam and QH-ii-066 on GABA-induced Cl<sup>-</sup> currents in two GABA<sub>A</sub> constructs ( $\alpha 1\beta 2\gamma 2$  and  $\alpha 5\beta 2\gamma 2$ ). Abscissae: concentration (log, nM). Ordinates: Percent control (3  $\mu$ M GABA without drug) I<sub>GABA</sub> for n=3-6 cells per concentration.

Figure 3. Anxiolytic-like effects of diazepam and QH-ii-066 in the Geller conflict with incremental shock test in rats. Abscissae: dose (mg/kg). Ordinates: mean ( $\pm$ SEM) response rates expressed as percent of control rates (vehicle training days) for n=6-8 animals per dose. Diazepam and QH-ii-066 were administered in 0.25% methylcellulose PO, 60 minutes prior to testing. \* *P*<0.05.

Figure 4. Effect of diazepam and QH-ii-066 on pentylenetetrazole- and electroconvulsive shock-induced seizures in mice. Abscissae: dose (mg/kg). Ordinates: percent animals protected for n=12 animals per dose. Diazepam and QH-ii-066 were administered in 0.25% methylcellulose PO, 60 minutes prior to testing.

Figure 5. Effect of diazepam and QH-ii-066 in conditioned fear test in rats. Abscissae: dose (mg/kg). Ordinates: mean ( $\pm$ SEM) percent time spent freezing for n=16-24 animals per dose. Diazepam and QH-ii-066 were administered in 0.25% methylcellulose PO, 60 minutes prior to behavioral testing. \* *P*<0.05. Figure 6. Effect of diazepam and QH-ii-066 on spontaneous locomotor activity in rats. Abscissae: dose (mg/kg). Ordinates: mean ( $\pm$ SEM) distance traveled (in cm) for n=8 animals per dose. Diazepam and QH-ii-066 were administered in 0.25% methylcellulose PO, 60 minutes prior to behavioral testing. \* *P*<0.05.

Figure 7. Effect of diazepam and QH-ii-066 on motor coordination in the rotorod test in rats. Abscissae: dose (mg/kg). Ordinates: mean ( $\pm$ SEM) time on rotorod (in seconds) for n=7-8 animals per dose. Diazepam and QH-ii-066 were administered in 0.25% methylcellulose PO, 60 minutes prior to behavioral testing. \* *P*<0.05.

Figure 8. Discriminative stimulus and rate-decreasing effects of chlordiazepoxide, diazepam, and QH-ii-066 in rats (n=7-12) discriminating between chlordiazepoxide and vehicle. Abscissae: dose (mg/kg). Ordinates: percent of rats selecting the chlordiazepoxide lever (upper panel) and mean response rates ( $\pm$  SEM) expressed as percent control rates (vehicle training days) (lower panel). Chlordiazepoxide was administered in 0.25% methylcellulose IP, 30 minutes before testing. Diazepam and QH-ii-066 were administered in 0.25% methylcellulose PO, 60 minutes prior to behavioral testing. \* *P*<0.05. The dashed line in the upper panel refers to 75% of animals selecting the drug lever (criterion for substitution).

Figure 1



Figure 2

EΡ







Dose (mg/kg)












## V. Appendix III: Data on QH-II-066

### eQTL and receptor pharmacology implicate GABA signaling in neuroblastoma

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#### **Summary**

The signaling pathways driving neuroblastoma, a tumor derived from neural progenitors, remain largely unidentified. In an engineered mouse model for neuroblastoma, susceptibility to tumors is strain-specific and genetically complex. To identify genes involved in tumor susceptibility, we analyzed genetic control of gene expression (eQTL) in the peripheral sympathetic nervous system. Here we show that genes controlling GABA neurotransmitter signaling, including arginase (*Arg1*), co-localized with susceptibility loci. Activation of GABA-A receptors in human neuroblastoma cells induced apoptosis, while inhibition of arginase slowed proliferation. These results demonstrate the ability of a neurotransmitter to regulate cell growth, and present novel neuroblastoma therapeutic targets.

#### Introduction

Neuroblastoma, a pediatric tumor of the peripheral nervous system, arises from sympathetic neural progenitors that normally differentiate into post-mitotic neurons shortly after birth. Mutations that drive myeloid, epithelial, and glial tumors are largely not present in neuroblastoma, suggesting this tumor arises from disruption of growth pathways specific to the neural lineage. In a transgenic model driven by the MYCN proto-oncogene, amplification of which is a hallmark of high-risk neuroblastoma<sup>1,2</sup>, mice develop tumors with histological and genomic characteristics of human neuroblastoma<sup>3,4</sup>. In this model, tumor incidence varies among mouse strains, and we exploited this genetic variance to identify novel molecular pathways driving neuroblastoma development.

In this study, genetic linkage analysis revealed a locus on chromosome 10 that interacted with numerous secondary loci to influence tumor susceptibility. To identify candidate genes at these loci, we superimposed expression quantitative trail loci (eQTL) in sympathetic superior cervical ganglia (SCG) from backcross mice. Liver arginase (*Arg1*), a component of both the urea cycle and the gamma-amino butyric acid (GABA) biosynthetic pathway, was the strongest eQTL at the primary chromosome 10 locus. Multiple genes and eQTL involved in GABA neurotransmitter signaling were located at the secondary susceptibility loci, providing an intriguing genetic link between this pathway and neuroblastoma biology. We tested this link using compounds targeting the *Arg1*-GABA pathway in human neuroblastoma cell lines. Both nor-NOHA, an *Arg1* inhibitor, and, more strikingly, nanomolar concentrations of the neurotransmitter GABA, inhibited cell growth and viability in vitro. The GABA response was dependent on GABA-A receptor expression and was rescued with GABA-A-specific antagonists. Activation of GABA-A signaling induced apoptosis and was associated with decreased activity of PI3K/AKT and MAPK signaling, two central pathways in cell growth and survival. The implication of arginine metabolism and GABA signaling in the pathogenesis of neuroblastoma presents a combination of therapeutic targets. Additionally, the demonstration that the neurotransmitter GABA inhibits growth of tumor cells derived from a neural lineage highlights an emerging role for neurotransmitters in regulating development of the peripheral nervous system.

#### Results

#### Neuroblastoma penetrance in TH-MYCN mice is strain-dependent

Transgenic mice were generated on a Balb/c x C57B6/J background and showed ~10% incidence of tumors. As mice were crossed into strain FVB/NJ, incidence decreased to zero by 2 generations (**Supplementary Figure 1A**). Conversely, penetrance of tumors steadily increased with each successive backcross into strain 129/SvJ, leveling out at ~60%<sup>5</sup>. Levels of transgene expression were similar between genetic backgrounds (**Supplementary Figure 1B**).

We next crossed resistant transgenic FVB/NJ mice to susceptible wild-type 129/SvJ. The resulting F1 mice showed 4% incidence of tumors, suggesting that resistance was genetically dominant. To generate a genetically diverse population for linkage mapping, we backcrossed transgenic F1 animals to wild-type 129/SvJ mice. Incidence of tumors in the resulting N1 backcross generation was 38% (N=203), with an average survival (109 days) identical to that of mice carrying the transgene in a pure 129/SvJ background <sup>5</sup>

#### Linkage analysis identifies a tumor susceptibility modifier on chromosome 10

To identify genomic loci associated with tumor susceptibility, we genotyped 203 mice using a combination of microsatellite and SNP markers (see **Supplementary Methods**). Interval mapping for linkage to tumor susceptibility produced a maximum LOD (log of odds) score on chromosome 10 at marker RS36323433 (LOD=4.1, **Figure** 

**1A, B,** significant at a 5% genome-wide error rate, see **Supplementary Methods**). However, at this locus, heterozygous mice were tumor prone (55%), while mice homozygous for the 129/SvJ allele were resistant (25% incidence) (**Figure 1C**). Additionally, tumor susceptibility was strongly influenced by gender. Performing the analysis using sex as an interacting covariate increased the LOD score to 4.9. When mice were segregated by sex and analyzed independently, this locus was not significant in female mice but was in males (LOD=4.3, N=82). It may be relevant in this regard that neuroblastoma is slightly more prevalent in boys (1.3:1) than in girls<sup>6</sup>.

Since the segregation of genotypes with tumor susceptibility at the chromosome 10 locus did not match the patterns of tumor susceptibility in the parent strains, we next considered a more complex genetic model to explain susceptibility. A 2-QTL test identified several loci that interacted with the chromosome 10 locus, with similar LOD scores (**Figure 1D, Supplementary Table 1**). Interestingly, while all of these loci interacted with the chromosome 10 locus, they did not interact with each other.

#### Expression QTL analysis identifies Arg1 as a candidate modifier

The 95% confidence interval of the peak on chromosome 10 spanned 47Mb and contained over 281 genes (see **Supplementary Methods**), complicating identification of candidates. Hypothesizing that susceptibility to tumors may be governed by differential expression of genes within this locus, we narrowed our field of candidates by comparing mRNA expression levels in neural crest-derived sympathetic superior cervical ganglia (SCG) from transgenic male and female 129/SvJ and FVB/NJ mice using Affymetrix

Mouse Exon arrays. Surprisingly, only 8 genes were differentially expressed between sexes, and with the exception of a RIKEN clone on chromosome 2, all mapped to either the X or Y chromosomes (**Supplementary Figure 1C, Supplementary Table 3**), excluding gender-specific gene expression as a mechanism for the gender effect at the chromosome 10 locus. When males of the two strains were compared, however, 9,820 genes were differentially expressed genome-wide (**Supplementary Figure 1D, Supplementary Table 3**), including 137 within the 95% confidence interval for the chromosome 10 susceptibility locus (representing almost half of all genes within the locus).

Gene expression levels can be influenced by complex interactions among cis- and trans- acting factors. One method for distinguishing these factors involves generating a genetically heterogeneous population (such as our backcross population), measuring gene expression levels, and treating the expression level of each gene as a quantitative trait (expression QTL or eQTL<sup>7,8</sup>). Using this method, the cis- and trans- acting alleles influencing gene expression are decoupled from each other, and genes whose differential expression is due to cis-acting factors at a locus can be distinguished from genes under the control of trans-acting factors at other loci. Genes with eQTL overlapping with QTL for physiological phenotypes have been shown to control these phenotypes<sup>8,9</sup>. We thus utilized this technique to identify eQTL within our chromosome 10 locus, as well as at the numerous other secondary loci identified in the 2-QTL tumor susceptibility analysis.

We measured mRNA expression levels in 116 SCG from the backcross population and tested for linkage between gene-level expression calls for each gene and germline variation, to identify 342 eQTL acting both locally and on another chromosomes (**Supplementary Table 4**). Four eQTL mapped to the susceptibility locus on chromosome 10, with Arg1 (liver arginase) showing the strongest eQTL. When this measurement was refined by interval mapping, Arg1 expression was linked to a QTL at chromosome 10 with a LOD score of 18.2 (**Figure 2A**). Arg1 overlapped directly with our tumor susceptibility locus, making Arg1 our top candidate gene (**Figure 2C, Table 1**). Mice heterozygous at that locus had almost 2-fold higher expression of Arg1compared to mice homozygous for the 129/SvJ allele (**Figure 2B**). Notably, among 137 genes differentially expressed at the chromosome 10 locus in parental strains, only four showed significant eQTL mapping to the locus, with Arg1 showing the strongest eQTL in backcross animals. These data suggest that differential expression of the other genes in the parents was due in part to *trans*-acting factors, illustrating the power of eQTL analysis in dissecting control of gene expression at a genetic locus and filtering candidate genes.

#### eQTL for GABA-related genes map to secondary susceptibility loci

We next looked for eQTL that mapped to the secondary tumor susceptibility loci that interacted with the chromosome 10 locus, and noted eQTL for genes related to GABA neurotransmitter signaling at two of the loci (**Table 1**). Notably, a trans-eQTL on chromosome 2 controlled expression of the *Gabra3* receptor subunit on the X chromosome (**Figure 3A-C**). This eQTL overlapped directly with the secondary susceptibility locus on chromosome 2 (**Figure 3G, Supplementary Figure 2A, Table 1**). Mice harboring alleles resulting in high *Arg1* expression and low *Gabra3* expression showed the strongest susceptibility to tumors (**Figure 3C**). Similarly, an eQTL for the GABA transporter *Slc6a1* (**Figure 3D-F**) mapped to a secondary susceptibility locus on chromosome 4 (**Figure 3G, Supplementary Figure 2B, Table 1**), though the gene was on chromosome 6. Importantly, the chromosome 2 and 4 secondary susceptibility loci showed the highest LOD scores in male mice (**Supplementary Table 2**).

We next investigated secondary susceptibility loci that lacked GABA-related eQTL candidates and found several GABA-related genes mapping within 10Mb of susceptibility peaks. The locus on chromosome 1 (LOD 7.8) flanked by markers RS5056599 (116MB) and D1MIT1001 (131Mb) , LOD 7.8) is in close proximity to *Dbi* (diazepam binding inhibitor, 122Mb), a gene that modulates GABA receptor activity<sup>10</sup>. Similarly, the locus on chromosome 9 centered near 117Mb (D9MIT201, LOD 7.9) is 4 Mb from the *Trak1* gene, which encodes a trafficking factor that modulates GABA receptor homeostasis<sup>11</sup>. The locus on chromosome 7 centered at 144Mb (RS13479509, LOD 6.4) is 4 MB from ornithine aminotransferase (*Oat*), which converts ornithine to glutamate (the substrate for GABA synthesis). Finally, the locus on chromosome 17 centered near 35Mb (D17MIT231, LOD 6.7) is 2 Mb from the GABA-B receptor 1. Together, at least 6 secondary susceptibility loci co-localized with genes in the GABA pathway, and/or eQTL controlling these genes (**Table 1**).

#### Inhibition of Arg1 decreases viability of human neuroblastoma cells

The higher-expressing Arginase allele conferring tumor susceptibility is nested within an overall resistant genetic background in purebred mice (either FVB/NJ or FVB/NJ x 129/SvJ F1). The allelic variation in the secondary loci (harboring components of the GABA pathway) only showed an effect in combination with alleles at other loci. This genetic complexity precluded validation *in vivo*. The association of a gene with susceptibility to tumors does not necessarily imply an ongoing role in tumor maintenance. Nevertheless, since our ultimate goal was to identify clinically-relevant biochemical pathways, we next analyzed ArgI and GABA in human neuroblastoma cell lines. As there is little precedent for these pathways in regulation of cell growth and thus little basis for the specific downstream pathways involved, we assessed the effect of compounds targeting this pathway on overall growth and viability.

We detected *Arg1* by western blot across a panel of human neuroblastoma cell lines (**Figure 4A**). We treated this panel with the reversible arginase inhibitor nor-NOHA (N-Omega-hydroxy-nor-arginine)<sup>12</sup> to test whether inhibition of arginase could impair tumor growth. Viability was decreased in CHP-126, SY5Y, and Kelly cells at  $100\mu$ M (**Supplementary Figure 3A**). Both Kelly and SY5Y cultures accumulated in G1 phase with decreased numbers of cells in S phase, without a significant apoptotic fraction assessed by flow cytometry (**Supplementary Figure 3B**). In SY5Y cells this effect was nearly equivalent to the potent PI3K/mTOR inhibitor PI-103. We conclude that Arg1 expression modulates viability and proliferation in human neuroblastoma cells.

#### GABA-A activation induces apoptosis in neuroblastoma

We next investigated the role of GABA signaling in the control of viability. While all neuroblastoma cell lines tested expressed the two human GABA-B receptors, expression of the GABA-A receptor subunits was variable (Figure 4A). In response to GABA, viability was unchanged in cell lines CHP-126, Kelly and SY5Y, all of which expressed low levels of the GABA-A receptor (Supplementary Figure 3C). In contrast, the cell line LAN-5, which expressed high levels of the GABA-A receptor, was particularly sensitive to GABA, showing a 40% decrease in viability at doses as low as 100nM. To demonstrate that this effect was specific to the GABA-A receptor, we co-treated cells with GABA and either picrotoxin or bicuculline. Picrotoxin, a non-competitive GABA-A channel blocker, was able to fully rescue the effects of both 100nM and 1mM GABA. In contrast, the competitive inhibitor bicuculline was able to reverse the effects of GABA in a dose-dependent manner; 10µM bicuculline reversed the effects of 100nM but not 1mM GABA treatment (Figure 4B). The dose-dependent and independent effects of bicuculline and picrotoxin, respectively, are consistent with the pharmacological mechanisms of each molecule on the GABA-A receptor, and demonstrate the specificity of the GABA-A receptor's role in cell viability.

We next assessed the mechanism by which GABA-A activation inhibited the growth of neuroblastoma cells. To enhance the signal readout of this pathway, we utilized the potent, selective GABA-A receptor agonist QHii066, a benzodiazepine derivative<sup>13,14</sup>. Apoptosis was induced in both LAN-5 (high GABA-A expression levels) and Kelly cells (low GABA-A expression levels) (**Figure 4 C and D**), as assessed by annexin V flow cytometry. Immunoblotting revealed that upon addition of GABA and QHii066, cleaved PARP, an indicator of apoptosis, was increased. Notably, immunoblotting revealed a decrease in activity of the PI3K/AKT and MAPK pro-growth and survival pathways, indicated by decreased phosphorylation levels of Akt and Erk,

respectively (**Figure 4E**). Together these data suggest that specific activation of the GABA-A receptor decreases cell viability, induces apoptosis, and suppresses pro-growth and survival signaling pathways in neuroblastoma cell lines.

#### Discussion

The pathways regulating development of the peripheral nervous system are distinct from those controlling epithelial and other cell types. Perhaps as a consequence, common genomic aberrations driving tumorigenesis in epithelial and glial tumors rarely show abnormalities in neuroblastoma, a tumor of the sympathetic peripheral nervous system.

Mouse model systems have provided a tremendous resource for the identification and validation of genes involved in cancer. Though the influence of strain background on tumor penetrance is frequently observed in mouse models of cancer, only a handful of genes underlying this susceptibility have been identified<sup>15-20</sup>, mostly due to the limited resolution of quantitative trait linkage mapping. To decipher biological mechanisms underlying genetic linkage to tumor susceptibility in our mouse neuroblastoma model, we investigated patterns of gene expression in peripheral sympathetic nervous tissue. Analysis of gene expression as a function of genotype (in other words, treating gene expression levels as heritable traits and performing linkage analysis) has facilitated identification of candidate genes for quantitative trait loci underlying physiological differences<sup>7-9</sup>. Genetic control of gene expression is more direct than genetic influence over complex physiological phenotypes. Thus, analysis of expression quantitative trait loci or eQTL provides a relatively strong signal in linkage analysis, enabling the direct identification of specific candidate genes among hundreds of genes at an overlapping QTL for a physiological phenotype. eQTL analysis has identified several genes modifying tumor susceptibility in mouse models for cancer<sup>21-23</sup>. In the current study, eQTL suggested candidate genes from a common biological pathway at multiple loci simultaneously, including genes located on chromosomes that were controlled in *trans* by unidentified elements at the QTL loci, a phenomenon that would have been missed by conventional analysis of candidate genes at the susceptibility loci.

eQTL analysis identified *Arg1* as a candidate neuroblastoma modifier gene at chromosome 10. One of the strongest secondary susceptibility loci, a locus on chromosome 2, overlapped with a particularly prominent eQTL governing expression of a GABA-A neurotransmitter receptor subunit. Numerous other susceptibility loci overlapped with other GABA-related genes, suggesting a mechanism for the pattern of genetic linkage to tumor susceptibility and implicating an interaction between upregulation of arginase activity and down-regulation of GABA receptors as cooperating mechanisms promoting susceptibility to tumors.

While Arg1 is associated with the urea cycle in liver, it is also expressed in a number of tissues that do not carry out the urea cycle, including sympathetic nerve ganglia<sup>24</sup>, suggesting a role in other biochemical pathways. In neurons, Arg1 is part of the GABA synthesis pathway, producing ornithine, a precursor of glutamate and of GABA. The biochemical link between Arg1 and the GABA pathway potentially explains our unique genetic pattern of several secondary loci interacting with a single gene/locus. Only one cytoplasmic arginase gene exists (Arg1). However, the GABA signaling pathway may be perturbed at several genomic loci, including several components of the GABA-A receptor, genes for which are dispersed throughout the genome. Thus, in a genetic model involving perturbation of arginase and GABA signaling, many GABA-related genes could interact with the single Arg1 gene on chromosome 10 with similar

effects (disrupting two connected pathways), while GABA genes may not interact with each other (as multiple disruptions would have redundant effects on the same pathway).

While emerging evidence supports the role for GABA signaling in the control of neural cell growth (see below), a role for Arg1 is less obvious. Several downstream outputs could account for increased expression of Arg1 predisposing mice to tumors. Arginase may act as an immunosuppressant<sup>25,26</sup>, a competitor for the substrate arginine with the growth-inhibitory nitric oxide synthases (NOSs)<sup>27-30</sup>, or a producer of ornithine, the substrate for polyamine synthesis linked to tumorigenesis<sup>31</sup> and neural proliferation<sup>32</sup>. The rate-limiting polyamine synthetic enzyme, ornithine decarboxylase (ODC), is a wellestablished target for c-myc and MYCN<sup>33</sup>. ODC inhibitors inhibit neuroblastoma development in vitro and in TH-MYCN mice<sup>34,35</sup>, and are currently in clinical trials for neuroblastoma (http://www.nmtrc.org/phase-i-dfmo/). Ornithine is also a substrate for synthesis of glutamate and GABA. Notably, ornithine aminotransferase (Oat), which converts ornithine to glutamate, maps to a susceptibility locus. GABA inhibits neuronal growth and promotes differentiation<sup>36</sup>. Thus, at least one ultimate output of Arg1 activity (GABA) could inhibit both induction and further growth of tumors, driving selection for secondary genetic lesions that disable this pathway.

While *Arg1* activity has not been studied extensively in cancer, inhibition of arginase has been shown to disrupt growth of breast cancer cells<sup>37</sup>. Arginase expression has also been linked to neuronal viability. Clinically, *Arg1* mutations cause arginemia, characterized by neurodegeneration (OMIM 207800). While hepatic production of neurotoxic metabolites are speculated to cause this neurodegeneration<sup>38,39</sup>, our data, coupled with detection of *Arg1* expression in neural tissues<sup>24,40,41</sup>, suggest that *Arg1* may

be involved in growth and survival signaling in neuroblasts, and inhibition of Arg1 may have an intrinsic cytotoxic effect in neurons and neuroblastoma cells.

The observation that mice with lower expression of GABA-A receptor subunits are tumor prone is consistent with GABA's known role in neuronal cell growth and differentiation<sup>36,42</sup>, and suggest that the association of down-regulation of GABA-A receptors with more aggressive human neuroblastomas has biological significance<sup>43</sup>. However, since the receptor is formed as a pentameric combination of 19 possible subunits, with changing expression patterns and biological roles<sup>42</sup>, testing this hypothesis is not straightforward. GABA signaling has been shown to negatively regulate growth of neural crest stem cells<sup>36</sup>. In our hands, neuroblastoma cells expressing the GABA-A receptor showed decreased viability, increased apoptosis, and diminished activity of mitogenic signaling pathways in response to GABA-A activation, supporting a role of GABA signaling in tumor cell growth and survival. These results parallel the neurotrophin model, based on the observation that dysregulation of TRK neurotrophin receptors correlates with clinical outcome in neuroblastoma<sup>44</sup>. Our results are consistent with this paradigm, suggesting that decreased signaling through the GABA pathway (rather than TRKs) may promote neuroblastoma development.

Neuroblastoma is a common pediatric tumor with a unique biology, making the development of novel targeted therapeutics problematic. As a result, improvements in clinical outcomes have been modest over the last several decades. Arginase inhibitors are under investigation for highly prevalent diseases such as hypertension<sup>27</sup>. Additionally, the implication of GABA and other neurotransmitters in regulating growth of neuroblastoma opens up the potential to apply numerous clinically-approved drugs used

in neurology and psychology as chemotherapeutic agents. These results also highlight a possible connection between the role of neurotransmitters in nervous system development and the regulation of neuroblastoma growth.

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#### **Author Contributions:**

CSH performed the majority of the experiments and data analysis in the manuscript. DQ performed the eQTL and statistical analysis and helped prepare the manuscript. QF, CC and YB performed biochemical assays. DDG and KN assisted with mouse work. LP, JC, JM, and PK provided technical support for genotyping data generation and analysis. YKS, JC, SK, TVD, and JK provided technical support for expression array data generation and analysis. AB, JK, YJC, JMC and TVD helped design experiments and prepare the manuscript. WAW supervised the project. CSH and WAW designed experiments and wrote the paper.

#### **Methods Summary:**

Backcross mice were generated by crossing *TH-MYCN* transgenic FVB/N mice to wild-type 129/SvJ, and subsequently crossing F1 offspring to wild-type 129/SvJ. Mice were observed for at least one year to be considered tumor-free. DNA was isolated from spleen and genotyped using a combination of microsatellite and SNP markers. RNA

from SCG was analyzed using Affymetrix Mouse Exon arrays. Susceptibility linkage was analyzed using R-QTL. Differential gene expression was assessed using Significance Analysis of Microarrays, and eQTL were calculated using custom software as described previously<sup>21</sup>. Cell viability and cell-cycle status were assessed using a WST-1 assay and flow cytometry. For details see **Supplementary Methods**.

# Table 1: Candidate genes at susceptibility loci

susceptibility			eQTL locus					
locus (Mb)	LOD	eQTL gene	(Mb)	eQTL p-val	perm. p-val			
28	5.0	Arg1	25	2.92E-20	<0.001			
166	7.0	Gabra3	178	5.57E-34	< 0.001			
115	7.4	Slc6a1	129	2.81E-05	0.001			
	susceptibility locus (Mb) 28 166 115	susceptibility locus (Mb) LOD 28 5.0 166 7.0 115 7.4	susceptibility locus (Mb) LOD eQTL gene 28 5.0 Arg1 166 7.0 Gabra3 115 7.4 Slc6a1	susceptibilityeQTL locussusceptibilitycQTL gene(Mb)285.0Arg1251667.0Gabra31781157.4Slc6a1129	susceptibility         eQTL locus           locus (Mb)         LOD         eQTL gene         (Mb)         eQTL p-val           28         5.0         Arg1         25         2.92E-20           166         7.0         Gabra3         178         5.57E-34           115         7.4         Slc6a1         129         2.81E-05			

# A. Candidate eQTL at susceptibility loci

# B. GABA genes at susceptibility loci

	susceptibility			
chromosome	locus (Mb)	LOD	gene	gene location (Mb)
1	131	7.8	Dbi	122
9	117	7.8	Trak1	121
17	35	6.9	Gabbr1	37
7	144	6.3	Oat	140

### C. Gene descriptions

	location	
gene	(chromosome)	description
Arg1	10	liver arginase
Gabra3	X*	GABA-A receptor, subunit alpha 3
Slc6a1	6*	GABA transporter, removes GABA from synaptic cleft
Dbi	1	diazepam binding inhibitor; modulates the action of the GABA receptor
Trak1	9	trafficking protein, kinesin binding 1, regulates GABA receptor
Gabbr1	17	GABA-B receptor, 1
Oat	7	ornithine aminotransferase (produces glutamate, a GABA precursor)

\*controlled by trans eQTL

Figure 1. A locus on chromosome 10 and multiple secondary loci are linked to tumor susceptibility. (A) LOD plot for tumor susceptibility shows a single significant locus on chromosome 10. Dotted line indicates 5% genome-wide significance threshold (LOD=2.81, 1000 permutations). (B) LOD plot of chromosome 10 only. Hash marks on the horizontal axis indicate marker positions. Dotted line indicates 5% genome-wide significance threshold (LOD=2.81, 1000 permutations). (C) Effect plot for the marker closest to the maximum LOD score (RS36323433) showing incidence of tumors as a function of genotype. (**D**) Two-QTL analysis using sex as an interacting covariate reveals that multiple secondary loci interact with the chromosome 10 locus. The top left half shows the results of the additive model (no epistatic interactions), and the bottom right half shows results of the "full" model, which accounts for epistatic interactions. The bar on the right displays the correspondence between color on the chart and LOD scores (the left side of the bar corresponds to the additive analysis on the top left half of the plot, while the right side of the bar, with a different scale, corresponds to the full model analysis on the bottom right). Locations of the maximum LOD scores are shown in Supplementary Table 1.

Figure 2. An eQTL for Arg1 co-localizes with the tumor susceptibility locus on chromosome 10. (A) Interval mapping for Arg1 expression, the most significant eQTL in the chromosome 10 region, showing a LOD score of 18.2 on chromosome 10, centered at the physical location of the Arg1 gene. Line indicates 5% genome-wide significance threshold. (B)  $Log_2 Arg1$  expression level as a function of genotype. (C) A plot of LOD

scores on chromosome 10 for susceptibility (black line) and the Arg1 eQTL (blue line) showing co-localization of the peaks.

Figure 3. Secondary susceptibility loci co-localize with eQTL for GABA-related genes.

(A) A trans-eQTL on chromosome 2 controls expression of the GABA-A receptor subunit 3 (*Gabra3*) on the X chromosome (arrow) (LOD=30.9). (B) *Gabra3* expression as a function of D2MIT148 genotype. (C) Plot of tumor incidence as a function of genotypes at the chromosome 2 and 10 loci. (D) Expression QTL for the *Slc6a1* GABA transporter on chromosome 4 (LOD=4.1). Arrow indicates the physical location of *Slc6a1* on chromosome 6. (E) *Slc6a1* expression as a function of D4MIT203 genotype.
(F) Tumor incidence as a function of genotypes at loci on chromosomes 4 and 10. (G) Genome-wide plot of LOD scores for the *Gabra3* (black), *Slc6a1* (green), *Arg1* (blue) eQTL and tumor susceptibility for a single-locus model (purple) and the combined effect of the peak of the chromosome 10 locus plus points across the rest of the genome (red), showing correspondence between eQTL peaks and primary and secondary susceptibility loci. Two-dimensional plots for *Gabra3* and *Slc6a1* are shown in **Supplementary Figure 2**.

Figure 4. GABA activation decreases viability and induces apoptosis inneuroblastoma cells (A) Western blot for Arg1, GABA-A, and GABA-B receptors inneuroblastomacelllines.

(**B**) GABA treatment of LAN-5, a GABA-A expressing cell line, results in decreased viability that is rescued by the GABA competitive inhibitor bicuculline (bic) at 10 $\mu$ M, in a GABA dose-dependent manner; and by the non-competitive channel blocker picrotoxin (picro) at 100 $\mu$ M in a GABA dose-independent manner. (**C**) Lan5 and Kelly cells were treated with GABA 100  $\mu$ M or the selective ligand for GABA QHii066 1  $\mu$ M or both and analyzed by annexin V-FITC flow cytometry for apoptosis after 72h. (**D**) Percentages of positive cells for apoptotic marker annexin V are mean  $\pm$  s.e. for triplicate samples with treatments indicated. (**E**) Cells analyzed by immunoblot using antisera indicated show decreased phosphorylation of Akt and Erk.









Supplementary Figure 1













0.0

CHP126

Kelly

SY5Y

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#### **Supplementary Methods:**

**Mice:** All mice were obtained from the Jackson Labs (Bar Harbor, ME), and were housed and treated following UCSF IACUC guidelines. Tumor-negative backcross mice were followed until one year of age (the latest tumor was detected at 342 days). Superior cervical ganglia (SCG) were surgically isolated and snap-frozen in liquid nitrogen. SCG were isolated from the parental control groups at 21 days.

**Taqman analysis of transgene expression:** Taqman expression analysis was performed on 6 mice (3 female, 3 male) from each strain. Proprietary assays for human MYCN and controls L18 and mGUS were obtained from Applied Biosystems (Carlsbad, CA). MYCN relative to mGUS is shown in **Supplementary Figure 1B**.

**Genotyping:** DNA was isolated from spleen tissue using a proteinase K lysis followed by phenol chloroform extraction. Microsatellite marker genotyping was carried out by the Marshfield Clinic (Marshfield, WI), and CIDR (Baltimore, MD). SNP genotyping was performed using template-directed primer extension with fluorescence polarization detection (FP-TDI<sup>1</sup>, Acycloprime II, Perkin Elmer, Waltham, MA) and SNPStream<sup>2</sup> 48plex (Beckman Coulter, Brea, CA). Markers and map positions are shown in **Supplementary Table 5**. The marker set had an average spacing of 8 MB genome-wide (excluding the high density of markers on chromosome 10).

**Linkage analysis:** Interval mapping was performed using the R-QTL<sup>3</sup> package in the R statistical language. Genotypes flagged as probable errors by R-QTL were discarded. The genetic map positions were determined using the physical map positions (NCBI

37/mm9), followed by re-estimation of the map using R-QTL, and likely mis-mapped markers were discarded. Linkage analysis was performed on a 1-cM grid. Genome-wide significance thresholds were determined by running 1000 permutations for each dataset. Interval analysis was performed using the binary mode of the "EM" model. All results reported as significant were significant at a 5% genome-wide error rate. 95% confidence intervals (CI) were determined using the lodint function in R-QTL. Genes within the confidence intervals were determined by counting all genes in the UCSC genome assembly mapping between markers flanking the 95% CI.

**Expression Arrays:** RNA from superior cervical ganglia was isolated using the RNEasy kit (QIAGEN, Valencia, CA), as we found these buffers were more effective at disrupting the ganglia than Trizol. 1µg of RNA was used as a starting template for the RiboMinus rRNA subtraction (Invitrogen, Carlsbad, CA) followed by the ST labeling protocol (Affymetrix, Santa Clara, CA). Labeled samples were hybridized to Affymetrix Mouse Exon 1.0 arrays. Array quality control was performed using the Affymetrix Expression Console. 2-way comparisons between homogenous groups (e.g. male vs. females or 129/SvJ vs FVB/N males) were performed using the Significance Analysis of Microarrays (SAM) package<sup>4</sup> using a 5% false discovery rate. Results are presented in **Supplementary Table 3** and plotted in **Supplementary Figures 1C** and **1D**.

**eQTL analysis:** Arrays were normalized using RMA in the XPS package (http://www.bioconductor.org/packages/2.6/bioc/html/xps.html). eQTL were calculated as described in <sup>5</sup>; briefly, linkage between gene expression and loci was assessed by linear regression with genome-wide significance assessed using a FDR-based method.

**Cell Culture:** Neuroblastoma cell lines were grown in RPMI media with 10% serum and antibiotics with the exception of SK-N-BE(2) (DMEM/F12, 10% serum) and IMR-32 (DMEM, 10% serum plus non-essential amino acids). Nor-NOHA was obtained from Bachem (Torrance, CA) and dissolved in DMSO. GABA was obtained from Sigma (St. Louis, MO) and dissolved in H<sub>2</sub>O. Picrotoxin and bicuculline were obtained from Sigma and dissolved in DMSO.

Western blotting: Equal amounts of total protein were loaded for 4%-12% SDSpolyacrylamide gel electrophoresis and transferred to nitrocellulose membranes. After blocking, membranes were blotted with Arg1 (R&D Systems, Minneapolis MN), GABA B R1, Erk (Santa Cruz Biotech, Santa Cruz, CA), GABA A Rβ2/3, GABA B R2, βtubulin (Millipore, Billerica, MA), cleaved-PARP (Asp214), p-Akt (Ser473), Akt, pErk (Thr202/Tyr204) (Cell Signaling Technology, Danvers, MA), and GAPDH (Upstate Biotechnology, Lake Placid NY). Antibodies were detected with HRP-linked antianti-rabbit (Amersham/GE, Piscataway, NJ), mouse. or anti-sheep IgG (Calbiochem/EMD, Gibbstown, NJ), followed by enhanced chemiluminescence (Amersham/GE, Piscataway, NJ).

**WST-1 assay:** Cells were grown in 12-well plates and treated for 72 hours prior to the addition of WST-1 reagent (100µl per 1ml media). Absorbance was read after 60 minutes. Results were plotted using Graphpad Prism software, and p-values were calculated using a two-tailed Student's t-test.

**FACS:** Cells were plated on 6-well dishes and treated as indicated for 24 hours. They were then harvested, fixed in 70% ethanol for 30 minutes, then stained with a propidium

iodide (PI) dye and RNase solution. All samples were analyzed on a FACSCaliber flow cytometer (Becton Dickenson, Franklin Lakes, NJ) and analyzed using ModFit (Verity Software, Topsham ME). Apoptosis was detected by flow cytometry for annexin V-FITC per manufacturer's protocol (Annexin V-FITC detection kit, BioVision, Milpitas, CA) using FlowJo software (Tree Star, Inc, Ashland, OR).
### **Supplementary Figure Legends**

**Supplementary Figure 1.** Incidence of neuroblastoma and SCG gene expression, but not MYCN transgene expression, are dependent on strain. (A) Incidence of tumors in FVB/NJ and 129/SvJ as a function of backcross number. (B) Taqman expression showing equivalent levels of *TH-MYCN* transgene expression in all tissues tested. Analysis of brain, adrenal gland, and superior cervical ganglia from each strain showed equivalent levels of expression (C) SAM analysis showing 3 genes differentially regulated between male and female FVB/NJ mice (N=5 in each group) (D) SAM analysis showing 9,820 genes differentially expressed between strains in male ganglia

**Supplementary Figure 2.** Correspondence between *Gabra3* and *Slc6a1* eQTL and secondary susceptibility loci. (A) 2-qtl lod scores are plotted for the intersection of chromosomes 10 (vertical) and 2 (horizontal) in the top panel, with the scale at right indicating LOD scores for the full (additive plus interaction) model. The contour line indicates a 1.5 LOD drop from the maximal value for the chromosome (See Table 1). The bottom panel shows the eQTL LOD plot for *Gabra3* expression, indicating that the peak LOD score falls within a 1.5-LOD (95%) confidence interval for the susceptibility locus. (**B**) plot similar to A, showing chromosome 4 on the horizontal axis and the eQTL lod score for *Slc6a1* in the bottom panel.

Supplementary Figure 3. Treatment of neuroblastoma cell lines with arginase inhibitors inhibits proliferation, while the effect of GABA on viability is GABRA-A expression dependent. (A) WST-1 assay showing a dose-dependent decrease in viability following 72h treatment of neuroblastoma cells with varying doses of the Arg1 inhibitor nor-NOHA. (C) FACS plot showing increased G1 phase in Kelly and SY5Y cells treated with nor-NOHA for 24 hours. The dual PI3K/mTOR inhibitor PI103 is shown for comparison. (C) Three neuroblastoma cell lines with low GABA-A expression show no change in viability in response to GABA. Viability was assessed by WST-1 assay at 72h.

### **Supplementary Tables:**

Supplementary Table 1: Tumor susceptibility interaction LOD scores, all mice, sex as an interacting covariate.

Supplementary Table 2: Tumor susceptibility interaction LOD scores, male mice only.

Supplementary Table 3: Genes differentially expressed between 129/SvJ and FVB/N

Supplementary Table 4: eQTL significant at a 5% genome-wide error rate

**Supplementary Table 5: Markers and map positions** 

#### **Supplementary References:**

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## VI. Appendix IV: Anticonvulsant Screening Data

	Anticonvulsant Screening Program																				
	Test 8 Results - Anticonvulsant Identification (Rats I.P.)																				
Add ID:	41200	07 I	D	Screen ID:				1													
Solvent Code: Animal Weight:		MC t: 105	5.0 - 14	0.0 g		Solve	ent	Prep:		M	SB									_	
Date Started:		20-	20-Nov-2012				Date Completed:				20-Nov-2012										
Reference:		480	): 216																		
Respon	se																				
Time (Hou		Hours)		0.25		0.5		1.0		2.0		4.0		6.0		8.0		24		3.0	
Test	Dose		Dths	N / F	с	N / F	с	N/F	с	N / F	с	N / F	с	N / F	с	N/F	с	N / F	с	N / F	с
MES	30.0			1/4		0 / 4		0 / 4		0 / 4		0 / 4		Ι		1		1		1	
тох	30.0			0 / 4		0 / 4		0 / 4		0 / 4		0 / 4		1		1		1		/	

Note: N/F = number of animals active or toxic over the number tested.

C= Comment code

Comments to Supplier:

## Anticonvulsant Screening Program Test 8 Results - Anticonvulsant Identification (Rats I.P.)

Add ID:	41200	7	D Screen ID: 1																		
Solvent Code:		MC	0 14	0.0 ~		Solve	ent	Prep:		м	SB										
Date Started:		20-	Nov-20	0.0 g 12		Date Completed:				20-Nov-2012											
Reference:		480	): 216																		
Respon	se																				
	Time (H	lours)	rs) 0.2			0.5		1.0		2.0		4.0		6.0		8.0		24		3.0	
Test	Dose		Dths	N/F	с	N / F	с	N/F	с	N / F	с	N/F	с	N / F	с	N / F	с	N / F	с	N / F	с
MES	30.0			1/4		0 / 4		0 / 4		0 / 4		0 / 4		1		1		Ι		1	
тох	30.0			0/4		0/4		0/4		0/4		0/4		1		1		1		1	

Note: N/F = number of animals active or toxic over the number tested.

C= Comment code

Comments to Supplier:

## VII. Appendix V: Data on QH-II-066

# Modulation of $\alpha$ 5 subunit-containing GABA<sub>A</sub> receptors alters alcohol drinking by rhesus monkeys

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#### Abstract

*Background:* Alcohol's ability to potentiate the activity of  $\gamma$ -aminobutyric acid (GABA) at GABA<sub>A</sub> receptors has been implicated as a key mechanism underlying the behavioral effects of alcohol. The complex molecular biology of these receptors raises the possibility that particular receptor subtypes may play unique roles in alcohol's abuserelated effects and that subtype-selective ligands with therapeutic specificity against alcohol might be developed. This study evaluated the capacity of  $\alpha$ 5GABA<sub>A</sub> receptor ligands to alter selectively the reinforcing effects of alcohol. Methods: Two groups of rhesus monkeys were trained to orally self-administer alcohol or sucrose under fixedratio schedules and limited daily access conditions. Additionally, following daily selfadministration sessions, the behavior of each monkey was scored for both speciestypical and drug-induced behaviors. *Results:* Concentrations of 1-6% alcohol maintained self-administration above water levels, engendered pharmacologicallyrelevant blood alcohol levels ranging from 90-160 mg/dl, and produced changes in behavior typical of alcohol intoxication. Concentrations of 0.3-3% sucrose also reliably maintained self-administration. The  $\alpha$ 5GABA<sub>A</sub> receptor agonist QH-ii-066 enhanced and the  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonist L-655,708 inhibited alcohol but not sucrose drinking. The changes in alcohol drinking could be reversed with the  $\alpha$ 5GABA<sub>A</sub> receptor antagonist XLi-093. However, L-655,708 increased yawning in both alcohol- and sucrose-drinkers, possibly indicative of an anxiogenic effect. Conclusions: These findings suggest a prominent and specific role for  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms in the reinforcing effects of alcohol. Moreover, these results suggest that  $\alpha$ 5GABA<sub>A</sub> receptors may represent a novel pharmacological target for the development of medications to reduce drinking. Of ligands modulating this receptor,  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonists may hold the most promise as alcohol pharmacotherapies.

**Keywords:** GABA<sub>A</sub> receptors, alcohol self-administration, monkey, pharmacotherapy, QH-ii-066, L-655,708, XLi-093, naltrexone

#### Introduction

Alcohol's ability to potentiate the activity of  $\gamma$ -aminobutyric acid (GABA) at GABA<sub>A</sub> receptors has been implicated as a key mechanism underlying the abuse-related effects of alcohol in both human and non-human subjects (e.g., Hyytiä and Koob, 1995; June et al., 1995; 1998; Korpi, 1994; Söderpalm and Hansen, 1998). Molecular biological studies have shown that the mammalian GABA<sub>A</sub> receptor is a pentamer consisting of subunits from at least five different families, including the  $\alpha$ ,  $\beta$ , and  $\gamma$  subunits (McKernan and Whiting, 1996; Pritchett et al., 1989; Rudolph et al., 2001). In addition, most of these subunits have a number of different isoforms (e.g.,  $\alpha 1$ - $\alpha 6$ ,  $\beta 1$ - $\beta 3$ ,  $\gamma 1$ - $\gamma 3$ ; Olsen and Sieghart, 2008). The complex molecular biology of GABA<sub>A</sub> receptors raises the possibility that subtype-selective agents with therapeutic specificity against alcohol might be developed.

GABA<sub>A</sub> receptors expressing the  $\alpha$ 5 subunit ( $\alpha$ 5GABA<sub>A</sub> receptors) comprise only a small proportion of native GABA<sub>A</sub> receptors and are found primarily in the hippocampus (Sur et al., 1999; Uusi-Oukari and Korpi, 2010). Projections from the CA1 and CA3 hippocampal fields have been shown to innervate brain regions implicated in the reinforcing effects of alcohol (e.g., Sesack and Grace, 2010). Several lines of evidence support a potential role for  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms in the abuserelated effects of alcohol. In this regard, using data from the Collaborative Study on the Genetics of Alcoholism, Song and colleagues (2003) reported a significant association between *GABRA5* (the gene encoding the  $\alpha$ 5 subunit of the GABA<sub>A</sub> receptor) and alcohol dependence. In alcohol discrimination studies in squirrel monkeys, we have shown that the  $\alpha$ 5GABA<sub>A</sub> receptor-selective agonists QH-ii-066 and panadiplon mimicked the discriminative stimulus effects of alcohol (Platt et al., 2005). Moreover, the discriminative stimulus effects of alcohol and the alcohol-like discriminative stimulus effects of QH-ii-066 were dose-dependently blocked by L-655,708, an  $\alpha$ 5GABA<sub>A</sub> receptor-selective inverse agonist (Platt et al., 2005). With respect to the reinforcing effects of alcohol, intrahippocampal or i.p. administration of other  $\alpha$ 5GABA<sub>A</sub> receptor-selective inverse agonists decreased alcohol self-administration in rats at doses that did not disrupt self-administration of saccharin or water (June et al., 2001; McKay et al., 2004). Finally, mice lacking the  $\alpha$ 5 subunit of the GABA<sub>A</sub> receptor show both reduced preference for and reduced consumption of alcohol (Boehm et al., 2004).

The purpose of the present study was to evaluate the contribution of  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms to the reinforcing effects of alcohol in rhesus monkeys. We determined the ability of  $\alpha$ 5GABA<sub>A</sub> receptor-selective ligands to augment and/or attenuate alcohol self-administration. The selective ligands included the agonist QH-ii-066, the antagonist XLi-093 and the inverse agonist L-655,708. Results with the  $\alpha$ 5GABA<sub>A</sub> receptor ligands were compared with results obtained with naltrexone, an FDA approved pharmacotherapy to treat alcohol dependence. Drugs that reduce alcohol self-administration may have potential as candidate anti-alcohol medications. An important consideration in the development of an alcohol pharmacotherapy is the pharmacological specificity of the candidate medication's effects. One technique used to evaluate the specificity of a medication's effect is to compare the doses of a drug to reduce behaviour maintained by alcohol relative to behaviour maintained by a non-drug reinforce. To that end, we investigated the extent to which L-655,708 modulated the reinforcing effects of alcohol at doses that did not alter sucrose self-administration. Finally, compliance with any medication regimen is tied closely with the extent to which the medication produces untoward side effects. We evaluated the observable behavioral effects of  $\alpha$ 5GABA<sub>A</sub> receptor ligands in both alcohol- and sucrose-drinking

monkeys to assess the degree to which the drugs elicited behavioral effects that might limit their clinical utility (e.g., sedation, anxiety, convulsions).

### Materials and Methods

*Subjects.* Nine adult male rhesus monkeys (*Macaca mulatta*), weighing 9.5 to 12.1 kg, were studied in daily experimental sessions. Three of the monkeys (two assigned to the alcohol group; one assigned to the sucrose group) had prior experience in operant i.v. self-administration (either cocaine or midazolam) procedures. The remaining monkeys were experimentally naive. Monkeys were housed in a colony room with a 12-h light/dark cycle. Monkeys had unrestricted access to water via the standard cage-associated automatic sipper line and were fed a diet of monkey chow (Harlan Teklad Monkey Diet, Harlan Teklad, Madison, WI), supplemented with fresh fruit. All animals were maintained in accordance with the guidelines of the Committee on Animals of the Harvard Medical School and the "Guide for Care and Use of Laboratory Animals" of the Institute of Laboratory Animal Resources, National Research Council, Department of Health, Education and Welfare Publication No. (NIH) 85-23, revised 1996. Research protocols were approved by the Harvard Medical School Institutional Animal Care and Use Committee.

*Apparatus.* Monkeys were housed individually in stainless steel primate cages that also served as the experimental chambers. One side of the cage was customized to form an operant drinking panel. The panel consisted of two retractable sippers (model #: ENV-652AM; Med Associates, Inc., Georgia, VT) equipped with solenoids to minimize dripping. Each sipper was connected with Tygon tubing to a stainless steel liquid reservoir mounted on the outside of the drinking panel. A response lever (Med Associates, Inc., Georgia, VT) was positioned below each sipper. Each press of a lever

produced an audible click and was recorded as a response. Colored lights mounted above the sippers could be illuminated to serve as visual stimuli. In these experiments, only one side of the drinking panel was active.

**Alcohol self-administration.** Five monkeys were trained to respond under a fixed-ratio (FR) schedule of oral alcohol reinforcement. In the presence of a white light, completion of every 10<sup>th</sup> response (FR 10) resulted in a change in illumination from white to red light and extension of the drinking spout. Displacement of the drinking spout up, down, left or right during extension resulted in alcohol delivery. Fluid flow continued for as long as the monkey displaced the spout or for 30-s, whichever was shorter. Within the 30-s sipper extension time, the monkey could stop drinking (i.e., release the spout) and resume drinking (i.e., displace the spout) as many times as it wanted. The actual duration and volume of intake, within the constraints of the schedule, were under the control of the subject. Between extensions of the drinking spout, all lights were off briefly, and responses had no scheduled consequences. Daily self-administration sessions lasted 3-h. The reinforcing effects of a range of concentrations of alcohol (0.5 -16% w/v), as well as water, were evaluated initially in each monkey. Each alcohol concentration (or water) was made available for a minimum of five consecutive selfadministration sessions and until stable intake was observed (i.e., no upward or downward trend in mls and/or g/kg consumed over a period of at least three consecutive days).

The effects of pretreatment with varying doses of the  $\alpha$ 5GABA<sub>A</sub> agonist QH-ii-066 (0.1 – 5.6 mg/kg), the  $\alpha$ 5GABA<sub>A</sub> antagonist XLi-093 (0.03 – 0.3 mg/kg) or the  $\alpha$ 5GABA<sub>A</sub> inverse agonist L-655,708 (0.1 – 1.8 mg/kg) and their vehicles were determined on self-administration of a 2% alcohol solution. This alcohol concentration was chosen because it maintained intakes significantly higher than that of water. In addition, it was on the ascending limb of the alcohol concentration-response function, allowing for detection of both increases and decreases in intake as a consequence of drug pretreatment. Each dose of QH-ii-066, XLi-093 and L-655,708 was studied for a minimum of five consecutive sessions and until intake was stable. Between evaluation of different doses of the  $\alpha$ 5GABA<sub>A</sub> ligands, monkeys were returned to baseline self-administration (i.e., a minimum of five days of self-administration of a 2% alcohol solution in the absence of any pretreatment). This step allowed assessment of baseline levels of intake over the course of the study. The  $\alpha$ 5GABA<sub>A</sub> ligands were studied in random order in individual animals, and all doses of a given ligand were studied before moving on to another drug. For comparison, the effects of the opioid antagonist naltrexone (0.01 – 0.3 mg/kg) also were determined.

Subsequently, maximally effective doses of QH-ii-066 (0.3 mg/kg) and L-655,708 (1.8 mg/kg) and their vehicles were given as pretreatments before sessions in which a wider range of alcohol concentrations were made available for self-administration. These studies determined the degree to which the ligands altered the shape and/or position of the alcohol concentration-response function. Finally, the capacity of XLi-093 to reverse the effects of QH-ii-066 and L-655,708 was determined. In all instances, the effects of the  $\alpha$ 5GABA<sub>A</sub> ligands at each alcohol concentration were studied for a minimum of five consecutive sessions and until intake was stable.

**Sucrose self-administration.** A separate group of four monkeys was trained to respond under a FR 10 schedule of oral sucrose reinforcement. This schedule and solution delivery was identical to that for alcohol self-administration, except that a sucrose solution instead of alcohol was available for self-administration. The reinforcing

effects of a range of concentrations of sucrose (0.1 - 3% w/v), as well as water, were evaluated initially in each monkey. Each sucrose concentration (or water) was made available for a minimum of five consecutive self-administration sessions and until stable intake was observed.

The effects of pretreatment with varying doses of the QH-ii-066 or L-655,708 and their vehicles were determined on self-administration of a 0.3% sucrose solution. This sucrose concentration was chosen because it maintained intakes that were significantly higher than that of water and roughly equivalent to intakes maintained by 2% alcohol in the alcohol self-administration group. Each dose of QH-ii-066 and L-655,708 was studied for a minimum of five consecutive sessions and until intake was stable. Between evaluation of different doses of the  $\alpha$ 5GABA<sub>A</sub> ligands, monkeys were returned to baseline self-administration. The  $\alpha$ 5GABA<sub>A</sub> ligands were studied in irregular order in individual animals, and all doses of a given ligand were studied before moving on to another test drug.

*Observable behavior.* Following daily self-administration sessions, the behavior of each monkey (regardless of whether the animal self-administered alcohol or sucrose) was scored during a five-min sampling period using a focal animal approach as described in Platt et al. (2000; 2002) and as modified for the rhesus monkey. That is, after the self-administration session, an observer who was not informed about the drugs under investigation watched a specific animal for five min. During this sampling period, the observer recorded each instance a particular behavior (Table 1) occurred in 15-s intervals during the period. Scores were calculated from these data as the number of 15-s intervals in which a particular behavior occurred (maximum possible score for each behavior was 20). A randomization schedule was employed so that order effects

between animals were minimized. Eight observers performed the scoring across the duration of the study. Typically, two observers were used daily – one observing the alcohol drinkers, one observing the sucrose drinkers. To ensure reliability across observers, each individual underwent at least 20 h of training until they reached an interobserver reliability criterion of  $\geq$  90% based on percent agreement scores calculated between all possible pairs.

**Blood alcohol levels (BAL).** At select time points across the study (e.g., once stable intake was obtained at a particular alcohol concentration or after a particular drug treatment), monkeys self-administering alcohol were anesthetized with ketamine immediately following the session and three – five mls of blood were collected from the femoral vein. Samples were immediately centrifuged at 3200 rpm for 8-12 minutes. The plasma was drawn off, transferred to polypropylene tubes, and then frozen until analysis. Triple determinations of BALs were conducted using a rapid high performance plasma alcohol analysis using alcohol oxidase with an AM1 series analyzer and Analox Kit GMRD-113 (Analox Instruments USA, Lunenberg, MA). This process detects BAL ranges from 0 to 350 mg/dl using an internal standard of 100 mg/dl.

**Data analysis.** After each alcohol self-administration session, intake for individual subjects was determined in two ways: volume (mls) and alcohol dose (g/kg) consumed. Dose was calculated as: [volume consumed (mls) X alcohol concentration (g/ml)]/weight (kg). After each sucrose self-administration session, intake for individual subjects was the volume consumed (mls). Data are expressed as mean intake over the last three sessions of water or alcohol/sucrose availability. Because the intake measurements for alcohol self-administration are correlated, with the exception of Fig. 1, we present only the data representing alcohol dose. A one-way repeated measures

ANOVA was used to evaluate the effect of alcohol/sucrose concentration on total volume consumed.

To compare the effects of the α5GABA<sub>A</sub> ligands and naltrexone on alcohol selfadministration, intakes following pretreatment with QH-ii-066, XLi-093; L-655,708 or naltrexone were converted to % of baseline intake. Baseline intake was considered to be the mean dose of alcohol consumed across the three days of no pretreatment immediately prior to tests with a given dose of a compound. Separate one-way repeated measures ANOVAs followed by Bonferroni t-tests compared the effects of the pretreatment drugs to the effect of vehicle on intake. In curve-shifting experiments, separate two-way repeated measures ANOVAs compared the effects of either QH-ii-066 or L-655,708, alone or combined with XLi-093, to the effect of vehicle on intake.

To compare the effects of the  $\alpha$ 5GABA<sub>A</sub> ligands on sucrose self-administration, intakes following pretreatment with QH-ii-066 or L-655,708 were converted to % of baseline intake. Baseline intake was considered to be the mean mls of sucrose consumed across the three days of no pretreatment immediately prior to tests with a given dose of a compound. Separate one-way repeated measures ANOVAs compared the effects of the pretreatment drugs to the effect of vehicle on intake.

Observational data were treated in two ways. To evaluate the effect of increasing intake of alcohol on observable behavior, frequency scores for each behavior were averaged across the alcohol self-administration subjects to obtain group means. Mean scores (± SEM) for each behavior then were plotted as a function of intake. For these analyses, intake (g/kg) was converted to "number of drinks". We use an operational definition of a "drink" as 0.25 g/kg alcohol or 17 g alcohol in a 70 kg human, representing the upper range of worldwide definitions of a drink (cf. Grant and Bennett, 2003; Kalant and Poikolainen, 1999). To determine statistical reliability of treatment

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effects on each behavior, the effect of drink number was determined by separate oneway ANOVAs. Only the behavioral data generated during determination of the alcoholconcentration response function were included in these analyses. To evaluate the effect of  $\alpha$ 5GABA<sub>A</sub> ligands on observable behavior, frequency scores for each behavior were averaged separately across the alcohol self-administration subjects and the sucrose self-administration subjects to obtain means. For each group, mean scores (± SEM) for each behavior were plotted as a function of dose. For normally distributed data (as assessed using the Shapiro-Wilk test), separate two-way mixed factor ANOVAs determined the effects of the between-group factor (alcohol vs. sucrose) and withingroup factor (dose) and their interaction on each behavior. In all cases, significant main effects were examined post-hoc with Bonferroni t-tests. The alpha level for all tests was  $P \le 0.05$ .

**Drugs.** Ethanol (95%; Pharmco Products. Brookfield, CT) and diluted to the appropriate concentrations with tap water. QH-ii-066 and XLi-093 were synthesized as described in Huang et al. (2000) and Li et al. (2003), respectively. These compounds were dissolved in 50-80% propylene glycol/20-50% 0.9% saline solution. L-655,708 (Tocris Bioscience, Ellsville, MO) was dissolved in 50% propylene glycol/50% 0.9% saline solution. Naltrexone HCI (Sigma/RBI, St. Louis, MO) was dissolved in 0.9% saline solution. Intramuscular injection volumes were 0.02 - 0.05 ml/kg body weight. Pretreatment times were 10 min for the  $\alpha$ 5GABA<sub>A</sub> ligands and 30 min for naltrexone.

#### Results

Alcohol self-administration. Oral presentation of alcohol maintained selfadministration in all monkeys under the FR schedule of alcohol delivery and limited 396

access conditions (Fig. 1). As the concentration of alcohol was increased, intake first increased and then decreased, resulting in an inverted u-shaped concentration-response function. For the group, alcohol concentrations between 1 and 6% (w/v) maintained intakes significantly higher than that during water availability (Fig. 1A; F(7,28)=17.6, P<0.001, Bonferroni t-tests, P<0.05). At these concentrations, monkeys drank approximately 0.8 – 1.9 g/kg of alcohol (Fig. 1B). Analysis of blood samples verified that the volumes consumed when concentrations  $\geq 2\%$  (w/v) were available produced BALs in excess of 80 mg/dl in all monkeys (Fig. 1C).

In addition to producing orderly effects on drinking behavior, alcohol altered several observable behaviors (Fig. 2). All five self-administration monkeys had selfadministration episodes resulting in intake of 1 - 8 drinks; while three of the five monkeys had self-administration episodes resulting in intake of nine or more drinks. These higher intakes were observed after 13% of the self-administration sessions. While intakes below the equivalent of 7 drinks failed to alter any observable behavior, intakes of at least 7 drinks produced reliable changes in measures of passive visual, locomotion, tactile/oral exploration, stereotypy, scratching, and ataxia compared to levels of these behaviors observed during water availability. Locomotion, across a range of 7 -9 drinks, was significantly increased compared to levels observed during water availability (F(10,716)=6.2, P<0.001, Bonferroni t-tests, P<0.05). At the highest intake (≥ 10 drinks), though, locomotion returned to water levels, likely coinciding with a significant increase in passive visual behavior (F(10,716)=2.0, P<0.05, Bonferroni t-tests, P<0.05). During water availability, ataxia was not observed. However, a significant drinkdependent increase in ataxia occurred at intakes of  $\geq 8$  drinks (F(10,716)=4.0, P<0.001, Bonferroni t-tests, P<0.05). Significant drink-dependent decreases in behaviors also were observed. At intakes ≥7 drinks, scratching was significantly reduced compared to water levels (F(10,716)=8.0, P<0.001, Bonferroni t-tests, P<0.05). Alcohol also

produced an overall drink-dependent decrease in levels of tactile/oral exploratory behavior (F(10,716)=2.2, P<0.05, Bonferroni t-tests, n.s.) and stereotypic behavior (F(10,716)=2.4, P<0.01, Bonferroni t-tests, n.s.). Alcohol had no other significant effects on observable behavior over the range of intakes observed during determination of the alcohol concentration-response function.

 $\alpha$ **5GABA**<sub>A</sub> ligands and alcohol self-administration. Daily pretreatment with the  $\alpha$ 5GABA<sub>A</sub> ligands or naltrexone produced qualitatively different effects on alcohol (2% w/v) intake depending on the compound (Fig. 3). For example, the two lowest doses (0.1 and 0.3 mg/kg) of QH-ii-066 induced small but significant increases in alcohol consumption (125 - 135%) of the baseline intake of  $1.4\pm0.2$  g/kg) compared to vehicle (F(5,20)=18.2, P<0.001, Bonferroni t-tests, P<0.05). At higher doses of QH-ii-066, intake returned to baseline levels and did not differ from that following vehicle administration. Problems with solubility prohibited testing higher doses of QH-ii-066. In the case of XLi-093, none of the doses evaluated altered self-administration of alcohol compared to vehicle, and intake was maintained at baseline levels (1.8±0.02 g/kg). In contrast, daily pretreatment with L-655,708 resulted in dose-dependent reductions in alcohol self-administration compared to vehicle (F(4,16)=37.6, P<0.001), with the two highest doses (1 and 1.8 mg/kg) producing significant decreases in intake (Bonferroni ttests, P<0.05). At the highest dose of L-655,708, intake was suppressed maximally to approximately 40% of the baseline intake of 1.6±0.2 g/kg. The suppression of alcohol self-administration by L-655,708 was similar to that induced by naltrexone. As expected, naltrexone engendered dose-dependent decreases in alcohol self-administration that reliably differed from vehicle at doses of 0.1 and 0.3 mg/kg (F(4,16)=7.7, P<0.01, Bonferroni t-tests, P<0.05). The highest dose of naltrexone maximally suppressed

alcohol self-administration to 38% of the baseline intake of  $1.7\pm0.2$  g/kg. In general, for all drugs, the effect observed on the first day of a given pretreatment was similar to that observed on the final day of administration (5 – 7 days), providing no evidence of the development of tolerance to the effects of QH-ii-066, L-655,708 or naltrexone (data not shown).

Based on the results of the study described above, doses of the  $\alpha$ 5GABA<sub>A</sub> ligands that maximally altered alcohol self-administration were studied in combination with a wider range of alcohol concentrations (Fig. 4). In general, pretreatment with QHii-066 (0.3 mg/kg) enhanced self-administration of 0.5 – 4% (w/v) alcohol solutions and resulted in a leftward shift in the ascending limb of the alcohol concentration-response function (Fig. 4, left). By and large, this leftward shift could be reversed if the animals also received a pretreatment of the highest dose of XLi-093 (0.3 mg/kg). These findings are confirmed by a significant alcohol concentration X treatment interaction (F(6,24)=5.2, P<0.01, Bonferroni t-tests, P<0.05). In contrast, pretreatment with L-655,708 (1.8 mg/kg) attenuated self-administration of 1 – 6% (w/v) alcohol solutions and resulted in an overall rightward shift in the ascending limb of the alcohol concentration-response function (Fig. 4, right). As with QH-ii-066, pretreatment with the highest dose of XLi-093 completely reversed the L-655,708-induced rightward shift in the alcohol concentration-response function. These findings are confirmed by a significant alcohol concentration by a significant alcohol concentration-response function. These findings are confirmed by a significant alcohol concentration-response function. These findings are confirmed by a significant alcohol concentration-response function. (F(6,24)=3.9, P<0.01, Bonferroni t-tests, P<0.05).

 $\alpha$ 5GABA<sub>A</sub> ligands and sucrose self-administration. Oral presentation of sucrose maintained self-administration in all monkeys under the FR schedule of sucrose delivery and limited access conditions (Fig. 5A). As the concentration of sucrose was increased, intake also increased. Patterns of responding were comparable to those

observed during alcohol self-administration (data not shown). For the group, sucrose concentrations between 0.3 and 3% (w/v) maintained intakes significantly higher than that during water availability (F(4,12)=9.5, P<0.01, Bonferroni t-tests, P<0.05). At 0.3% (w/v), individual subjects displayed intakes ranging from ~ 450 to 1100 mls. The average intake (± SEM) for the group at this concentration was 839±157 mls and did not differ significantly from the intake of the alcohol self-administration group at 2% (w/v) (906 ± 92 mls; t(7)=0.4, P=n.s.). Daily pretreatment with either QH-ii-066 or L-655,708 did not markedly alter sucrose self-administration at any dose compared to their respective vehicles (Fig. 5B).

α5GABA<sub>A</sub> ligands and observable behavior. In the absence of drug pretreatment and under water availability, the alcohol- and sucrose-drinking monkeys did not differ from each other for the majority of behaviors (Supplemental Figure 1). However, the sucrose drinkers did exhibit significantly higher levels of tactile/oral exploration (t(7)=-4.3, P<0.005) and locomotion (t(7)=-2.4, P<0.05) and tended to present more (t(7)=-2.5, P<0.05). Neither QH-ii-066 nor L-655,708, at any dose evaluated, markedly altered the majority of observable behaviors. Likewise, naltrexone failed to alter any of the behaviors. There were, however, two exceptions (Fig. 6A and 6B). QH-ii-066 dose-dependently decreased tactile/oral exploration in both the alcoholdrinking and sucrose-drinking monkeys, with a dose of 3 mg/kg significantly reducing this behavior compared to vehicle levels. This effect was confirmed by a significant main effect of QH-ii-066 dose on this behavior (Shapiro-Wilk Normality test – passed; F(4,28)=4.6, P<0.01, Bonferroni t-tests, P<0.05). On the other hand, L-655,708 dosedependently increased yawning in both self-administration groups, with the highest dose (1.8 mg/kg) engendering significantly more yawning compared to vehicle (Shapiro-Wilk Normality test – passed; F(4,28)=2.8, P<0.05, Bonferroni t-tests, P<0.05).

#### Discussion

In the present study, alcohol served as a reinforcer in all monkeys, maintaining intake levels above those of water. At concentrations of  $\geq 2\%$  (w/v), monkeys selfadministered pharmacologically relevant doses of alcohol and produced BALs ranging from 90 – 160 mg/dl. The overall intakes (i.e., g/kg) observed in this study are generally typical of intakes observed for macaques self-administering alcohol under limited access conditions (Grant and Johanson, 1988; Henningfield and Meisch, 1978; 1979; 1981; Katner et al., 2004; Vivian et al., 1999). Differences in intake between the present study and those cited could be due to a number of methodological differences including length of self-administration session, addition of flavorants, induction techniques, concentrations of alcohol made available for self-administration, and the selfadministration apparatus itself. In fact, the latter difference could be especially important in the cases where the intakes observed in the present study are higher than those of the cited studies. In our procedure, the monkey must press a lever to gain access to a drinking spout connected to an alcohol reservoir. Depression of the drinking spout during extension results in alcohol delivery and fluid flow continues for as long as the monkey displaces the spout or for 30-s, whichever is shorter. The actual duration and volume of intake, within the constraints of the schedule, is under the control of the subject. Depending upon the available alcohol concentration, we observed average volumes of 12.5 mls of alcohol delivered per spout extension (range: ~3.5 mls/extension at 16% w/v; ~21 mls/extension at 2% w/v; data not shown). This average volume/delivery is much larger than the volumes/delivery in the cited studies in which the drinking spouts are designed to deliver a set amount of alcohol (typically 0.5 - 1 ml) each delivery. Evidence from rats would suggest that alcohol intake should increases as the volume per reinforcement increases (Henningfield and Meisch, 1975).

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When given the opportunity to consume a wide range of alcohol concentrations, monkeys drank sufficient quantities of alcohol to alter observable behavior. In many instances, these behavioral changes mirrored the effects of alcohol in intoxicated humans or those exposed to an acute alcohol challenge. For example, it is wellestablished that alcohol has a biphasic effect on activity (e.g., Martin et al., 1993; Morean and Corbin, 2010) such that, in human subjects, alcohol-induced stimulant effects (e.g., feelings of energized, stimulated, vigorous) are generally observed at lower BALs on the ascending limb of the blood alcohol curve, and alcohol-induced sedative effects (e.g., feelings of heavy head, inactive, sluggish) are reported at higher BALs on the descending limb of the blood alcohol curve. In monkeys, we observed a reliable increase in locomotor activity across a lower range of drinks (and presumably lower BALs) compared to the range where we observed increases in passive visual behavior and decreases in tactile/oral exploration, possible indices of sedative-like effects. Alcohol also is known to reduce the response to stressors and can produce anxiolyticlike effects in humans (e.g., Eckardt et al., 1998; Thomas et al., 2003). Similarly, alcohol produced a drink-dependent decrease in stereotypy and scratching, two behaviors that have been associated with anxiety in nonhuman primates and which are amenable to reduction by typical anxiolytics (e.g., Birkett et al., 2005; Castles et al., 1999; Schino et al., 1996; Tiefenbacher et al., 2005a; 2005b; Troisi, 2002). Finally, alcohol has been shown to dose-dependently increase the amount of body sway or static ataxia in humans (e.g., Schuckit, 1985); likewise, in monkeys, observable ataxia increased in a drink-dependent manner. Together, these findings illustrate and reinforce the translational nature of our non-human primate drinking model.

Using this model and GABA<sub>A</sub> receptor subtype-selective ligands, we assessed the contribution of  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms to the reinforcing effects of alcohol. We found that the agonist QH-ii-066 modestly increased, and the inverse agonist L- 655,708 decreased, intake of alcohol (2% w/v). Subsequently, the maximally effective dose of QH-ii-066 was found to enhance intake of a wider range of alcohol concentrations; whereas, the maximally effective dose of L-655,708 was found to inhibit intake of a wider range of alcohol concentrations. Both leftward (QH-ii-066) and rightward (L-655,708) shifts in the alcohol concentration-response function were abolished by XLi-093, despite it not altering drinking on its own. These observations are in agreement with results from rodent studies showing that other  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonists can reduce drinking (June et al., 2001; McKay et al., 2004). In conjunction with observations that  $\alpha$ 5GABA<sub>A</sub> receptor-subunit knockout mice exhibit reduced preference for and consumption of alcohol (Boehm et al., 2004), these findings suggest a prominent role for  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms in the reinforcing effects of alcohol. Moreover, the lack of effect on drinking behavior with XLi-093 alone raises the possibility that neutral efficacy (i.e., antagonism) at this specific GABA<sub>A</sub> receptor subtype is not sufficient to inhibit drinking, rather inverse efficacy is required.

In our hands, L-655,708 was as effective as the prototypical alcohol pharmacotherapy naltrexone in reducing drinking in the monkeys. Coupled with the observation that L-655,708 also can attenuate the subjective/interoceptive effects of alcohol (Platt et al., 2005), a growing body of research suggests that  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonists may have potential as candidate anti-alcohol medications. An important consideration in the development of an alcohol pharmacotherapy is the pharmacological specificity of the candidate medication's effect. One technique used to evaluate the specificity of a medication's effect is to compare the doses of a drug needed to reduce behavior maintained by alcohol relative to behavior maintained by a non-drug reinforcer. For example, if a drug reduces alcohol-maintained behavior to a greater extent than it reduces behavior maintained by the non-drug reinforcer, then the

drug may have selective effects on alcohol self-administration (c.f. Mello and Negus, 1996). In the present study, L-655,708, across the same dose range studied in the alcohol drinkers, did not alter drinking in the sucrose drinkers. This result suggests that the inhibitory effects of L-655,708 are selective for alcohol. Moreover, that sucrose drinking also was not modified by QH-ii-066 provides additional support for a key role for  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms in modulating the abuse-related effects of alcohol in particular, rather than drinking behavior in general.

Because adverse side effects can limit the utility of a pharmacotherapy, and because non-selective inverse agonists at the GABA<sub>A</sub>/benzodiazepine site often can have convulsant/proconvulsant effects (e.g., Braestrup et al., 1983), it was of interest to determine the behavioral profile of L-655,708. In both alcohol- and sucrose-drinkers, L-655,708 was relatively benign and there was no indication of any interaction between L-655,708 and alcohol. Notably, even with repeated administration, L-655,708 was devoid of convulsant activity. This agrees with studies in rats showing that L-655,708 does not induce seizures on its own, nor does it alter the dose of pentylenetetrazole required to induce seizures (Atack et al., 2006). The sole observable effect of L-655,708 was to increase vawning in both groups of monkeys. In nonhuman primates, increases in yawning often have been interpreted as indicating increased stress and anxiety (Castles et al., 1999; Lutz et al., 2004), and yawning can be induced in monkeys with administration of an anxiogenic drug (Major et al., 2009). Likewise, L-655,708 exhibited anxiogenic-like activity in mice in an elevated plus-maze procedure (Navarro et al., 2002). While the effects of L-655,708 have not been determined in alcoholics, there is evidence for transient anxiogenic effects of flumazenil, a benzodiazepine receptor antagonist, in alcoholics in early withdrawal (Nutt et al., 1993). Taken together, these findings certainly would suggest that an  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonist is likely to

have anxiogenic effects in humans. While a relatively mild side effect, anxiogenesis would likely limit how an L-655,708-type drug might be used effectively in a treatment setting. For example, given that stress-induced negative affective states and withdrawal-induced anxiety are important facilitators of relapse to excessive alcohol use (e.g., Breese et al., 2005; Heilig et al., 2010), a drug that induces anxiety would not be a practical anti-relapse medication. Rather, our results would suggest that an  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonist may be most useful as a pharmacotherapy to assist individuals with achieving abstinence.

In summary, our findings suggest a prominent and selective role for  $\alpha$ 5GABA<sub>A</sub> receptor mechanisms in the reinforcing effects of alcohol. The results implicate  $\alpha$ 5GABA<sub>A</sub> receptors as potential pharmacological targets for medications to reduce drinking and assist individuals in achieving abstinence. In particular,  $\alpha$ 5GABA<sub>A</sub> receptor inverse agonists may hold the most promise as alcohol pharmacotherapies.

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#### SPECIES-TYPICAL: Passive visual (VIS) Animal is standing or sitting motionless with eves open Rest/Sleep posture (RSP) Idiosyncratic posture adopted by monkeys during rest or sleep; eyes closed, easily roused by external stimulation (e.g., tapping on cage) Locomotion At least two directed steps in the horizontal and/or vertical plane (LOC) Tactile/oral exploration (TAC) Any tactile or oral manipulation of the cage or environment Sweeping and/or picking through wood chip substrate Forage (FOR) Picking, scraping, spreading or licking of an animal's own hair Self-groom (GRM) (SCR) Vigorous strokes of the hair with the finger or toenails Scratch Vocalization (VOC) Species-typical sounds emitted by monkey (not differentiated into different types) To open mouth wide and expose teeth Yawn (YWN) Posture involving presentation of rump, belly, flank, and/or neck) to observer or other Present (PRE) monkev Threat/aggress (THR) Multifaceted display involving one or more of the following: open mouth stare with teeth partially exposed, evebrows lifted, ears flattened or flapping, rigid body posture, piloerection, attack (e.g., biting, slapping) of inanimate object or other monkey Grin-like facial expression involving the retraction of the lips exposing clenched teeth; (FGR) Fear grimace may be accompanied by flattened ears, stiff, huddled body posture, screech/chattering vocalizations (BSP) An involuntary twitch or shudder of the entire body; also "wet dog" shake Body spasm Lip smack (LIP) Pursing the lips and moving them together to produce a smacking sound: often accompanied by moaning Any vigorous shaking of the cage that may or may not make noise Cage shake (CSH) Stereotypy Any repetitive, ritualized pattern of behavior that serves no obvious function (STY) Any notable behavior not indicated above (e.g., masturbation, nose rub, lip droop, (OTH) Other vomit/retch) **PROCEDURE-RELATED:** Lever press (LVR) Depression of lever manipulanda on the drinking panel Drink (DRI) Mouth contact to fluid delivery sippers DRUG-INDUCED: Any slip, trip, fall or loss of balance Ataxia (ATX) Procumbent posture Loose-limbed posture (sitting or lying on cage bottom) accompanied by eye closure; not (PRO) easily aroused by external stimulation (e.g., tapping on cage)

### Table 1. Behavioral categories, abbreviations, and definitions<sup>a</sup>.

<sup>a</sup> Adapted from Novak et al., 1992; Weerts et al., 1998; Platt et al., 2000, 2002.

#### **Figure Legends**

- Figure 1 Oral self-administration of varying alcohol concentrations and water under a fixed-ratio schedule and limited access conditions in monkeys. For each monkey, mean intakes for the last three sessions at each concentration were calculated. Data points represent the mean intake  $\pm$  SEM (A. mls; B. g/kg) for the group (N = 5). \* P<0.05 vs. water (gray bar = SEM for water availability). C. Mean  $\pm$  SEM blood alcohol levels (mg/dl) determined once stable intake was obtained at a particular alcohol concentration for the group (N = 5).
- Figure 2 Effects of increasing number of drinks on observable behavior (one drink = 0.25 g/kg alcohol). Data points represent the mean  $\pm$  SEM at each drink (N = 34 115). \* P<0.05 vs. water (gray bar = SEM for water availability).
- Figure 3 Effects of daily pretreatment with QH-ii-066, XLi-093, L-655,708 or naltrexone on intake maintained by a 2% (w/v) alcohol solution. For each monkey, mean intakes for the last three sessions at each dose were calculated. Data points represent the mean intake  $\pm$  SEM presented as a % of baseline g/kg for the group (N = 5). \* P<0.05 vs. vehicle. Baseline intake for QH-ii-066 = 1.4 $\pm$ 0.2 g/kg; XLi-093 = 1.8 $\pm$ 0.02 g/kg; L-655,708 = 1.6 $\pm$ 0.2 g/kg; naltrexone = 1.7 $\pm$ 0.2 g/kg (one-way ANOVA; F(3,16)=0.97, n.s.).
- Figure 4 Modulation by QH-ii-066 and L-655,708, administered alone or in conjunction with XLi-093, of alcohol self-administration in monkeys. Points are group means ± SEM based on the last three sessions at each alcohol concentration in five monkeys. \* P<0.05 vs. vehicle.</li>
- Figure 5 A. Oral self-administration of varying sucrose concentrations and water under a fixed-ratio schedule and limited access conditions in monkeys. For each monkey, mean intakes for the last three sessions at each concentration were calculated. Data points represent the mean intake ± SEM (mls) for the group (N

= 4). \* P<0.05 vs. water (gray bar = SEM for water availability). B. Effects of daily pretreatment with QH-ii-066 or L-655,708 on intake maintained by a 0.3% (w/v) sucrose solution. For each monkey, mean intakes for the last three sessions at each dose were calculated. Data points represent the mean intake  $\pm$  SEM presented as a % of baseline mls for the group (N = 4). Baseline intake for QH-ii-066 = 739±178 mls; L-655,708 = 953±142 mls (t-test; t(6)=-0.94, n.s.).

- Figure 6 Observable effects of QH-ii-066 on tactile/oral exploration (A) and L-655,708 on yawning (B). Bars/points are group means ± SEM in the sucrose self-administration group (N = 4) and the alcohol self-administration group (N = 5). \* P<0.05 vs. vehicle.</li>
- Supplemental Figure 1 Behavioral profile of alcohol and sucrose self-administration monkeys under water availability (see Table 1 for abbreviations). \* P<0.05








Figure 3:



Dose (mg/kg, i.m.)



Figure 5:





Identification code	(33)	
Empirical formula	$C_{33}H_{26}N_6O_6$	
Formula weight	602.60	
Temperature	295(2) K	
Wavelength	1.54178 Å	
Crystal system	Monoclinic	
Space group	P2/n	
Unit cell dimensions	a = 6.9314(1) Å	α= 90°.
	b = 9.2889(1) Å	$\beta = 95.316(1)^{\circ}.$
	c = 22.5403(4)  Å	$\gamma = 90^{\circ}$ .
Volume	1445.02(4) Å <sup>3</sup>	
Z	2	
Density (calculated)	1.385 Mg/m <sup>3</sup>	
Absorption coefficient	0.809 mm <sup>-1</sup>	
F(000)	628	
Crystal size	0.24 x 0.10 x 0.02 mm <sup>3</sup>	
Theta range for data collection	3.94 to 66.97°.	
Index ranges	-7<=h<=7, -10<=k<=10, -	-22<=l<=26
Reflections collected	7178	
Independent reflections	2450 [R(int) = 0.0286]	
Completeness to theta = $66.97^{\circ}$	95.3 %	
Absorption correction	Semi-empirical from equi	valents
Max. and min. transmission	0.984 and 0.823	
Refinement method	Full-matrix least-squares	on F <sup>2</sup>
Data / restraints / parameters	2450 / 0 / 211	
Goodness-of-fit on F <sup>2</sup>	1.045	
Final R indices [I>2sigma(I)]	$R1 = 0.0451, wR2 = 0.12^{\circ}$	76
R indices (all data)	$R1 = 0.0574, wR2 = 0.13^{\circ}$	78
Extinction coefficient	0.0020(6)	
Largest diff. peak and hole	0.369 and -0.181 e.Å <sup>-3</sup>	

VIII. Appendix VI (a). Crystal data and structure refinement for XLi093 (33).

**Appendix III(b)**. Atomic coordinates (x  $10^4$ ) and equivalent isotropic displacement parameters (Å<sup>2</sup>x  $10^3$ ) for XLi093(**33**).

	Х	У	Z	U(eq)
C(1)	583(3)	4957(2)	3838(1)	60(1)
N(2)	-713(2)	5543(2)	4205(1)	66(1)
C(3)	302(3)	6423(2)	4553(1)	63(1)
N(4)	2206(2)	6468(2)	4437(1)	58(1)
C(5)	3729(3)	7252(2)	4757(1)	56(1)
C(6)	3678(3)	7399(2)	5366(1)	62(1)
C(7)	5123(3)	8119(2)	5706(1)	64(1)
C(8)	6661(3)	8721(2)	5433(1)	60(1)
C(9)	6658(3)	8620(2)	4822(1)	60(1)
C(10)	5230(3)	7871(2)	4466(1)	57(1)
C(11)	5409(3)	7917(2)	3806(1)	63(1)
O(11)	6076(2)	9000(2)	3588(1)	79(1)
N(12)	4863(2)	6755(2)	3477(1)	68(1)
C(13)	4333(3)	5396(2)	3747(1)	67(1)
C(14)	2394(3)	5524(2)	3978(1)	57(1)
C(15)	-123(3)	3891(2)	3389(1)	62(1)
O(15)	-1791(2)	3537(2)	3294(1)	88(1)
O(16)	1313(2)	3338(2)	3099(1)	66(1)
C(17)	735(3)	2297(3)	2632(1)	68(1)
C(18)	2500	1438(4)	2500	70(1)
C(19)	5152(4)	6746(4)	2849(1)	103(1)
C(20)	8236(3)	9416(2)	5781(1)	70(1)
C(21)	9564(4)	9941(3)	6065(1)	92(1)

U(eq) is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

C(1)-C(14)	1.371(3)
C(1)-N(2)	1.386(3)
C(1)-C(15)	1.468(3)
N(2)-C(3)	1.295(3)
C(3)-N(4)	1.369(2)
N(4)-C(14)	1.370(3)
N(4)-C(5)	1.423(2)
C(5)-C(6)	1.384(3)
C(5)-C(10)	1.403(3)
C(6)-C(7)	1.377(3)
C(7)-C(8)	1.397(3)
C(8)-C(9)	1.379(3)
C(8)-C(20)	1.437(3)
C(9)-C(10)	1.401(3)
C(10)-C(11)	1.504(3)
C(11)-O(11)	1.229(2)
C(11)-N(12)	1.344(3)
N(12)-C(19)	1.448(3)
N(12)-C(13)	1.464(3)
C(13)-C(14)	1.491(3)
C(15)-O(15)	1.203(2)
C(15)-O(16)	1.342(2)
O(16)-C(17)	1.457(3)
C(17)-C(18)	1.513(3)
C(18)-C(17)#1	1.513(3)
C(20)-C(21)	1.177(3)
C(14)-C(1)-N(2)	109.99(18)
C(14)-C(1)-C(15)	131.11(18)
N(2)-C(1)-C(15)	118.90(17)
C(3)-N(2)-C(1)	105.25(16)
N(2)-C(3)-N(4)	112.49(18)
C(3)-N(4)-C(14)	106.56(16)
C(3)-N(4)-C(5)	126.97(17)

**Appendix III**(c). Bond lengths [Å] and angles [°] for XLi093(**33**).

C(14)-N(4)-C(5)	126.31(15)
C(6)-C(5)-C(10)	120.76(18)
C(6)-C(5)-N(4)	117.91(17)
C(10)-C(5)-N(4)	121.33(18)
C(7)-C(6)-C(5)	121.08(18)
C(6)-C(7)-C(8)	119.6(2)
C(9)-C(8)-C(7)	118.85(19)
C(9)-C(8)-C(20)	120.46(18)
C(7)-C(8)-C(20)	120.7(2)
C(8)-C(9)-C(10)	122.81(18)
C(9)-C(10)-C(5)	116.79(19)
C(9)-C(10)-C(11)	115.74(17)
C(5)-C(10)-C(11)	127.30(17)
O(11)-C(11)-N(12)	122.2(2)
O(11)-C(11)-C(10)	119.17(19)
N(12)-C(11)-C(10)	118.64(18)
C(11)-N(12)-C(19)	119.2(2)
C(11)-N(12)-C(13)	122.06(19)
C(19)-N(12)-C(13)	117.7(2)
N(12)-C(13)-C(14)	110.26(17)
N(4)-C(14)-C(1)	105.71(16)
N(4)-C(14)-C(13)	117.83(16)
C(1)-C(14)-C(13)	136.41(19)
O(15)-C(15)-O(16)	123.6(2)
O(15)-C(15)-C(1)	124.09(19)
O(16)-C(15)-C(1)	112.30(17)
C(15)-O(16)-C(17)	115.99(16)
O(16)-C(17)-C(18)	108.60(16)
C(17)#1-C(18)-C(17)	116.3(3)
C(21)-C(20)-C(8)	177.6(3)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1/2,y,-z+1/2

**Appendix III(d)**. Anisotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>) for XLi093 (**33**). The anisotropic displacement factor exponent takes the form:  $-2\pi^2$ [ h<sup>2</sup> a<sup>\*2</sup>U<sup>11</sup> + ... + 2 h k a<sup>\*</sup> b<sup>\*</sup> U<sup>12</sup> ]

	U <sup>11</sup>	U <sup>22</sup>	U33	U <sup>23</sup>	U13	U <sup>12</sup>	
C(1)	45(1)	62(1)	74(1)	0(1)	14(1)	-3(1)	
N(2)	42(1)	77(1)	81(1)	-6(1)	17(1)	-4(1)	
C(3)	43(1)	66(1)	82(1)	-6(1)	18(1)	-2(1)	
N(4)	40(1)	56(1)	79(1)	-6(1)	16(1)	-4(1)	
C(5)	43(1)	49(1)	77(1)	-1(1)	13(1)	0(1)	
C(6)	51(1)	60(1)	78(1)	1(1)	19(1)	-6(1)	
C(7)	59(1)	65(1)	70(1)	2(1)	13(1)	-6(1)	
C(8)	50(1)	57(1)	74(1)	1(1)	8(1)	-3(1)	
C(9)	45(1)	54(1)	81(1)	3(1)	14(1)	-4(1)	
C(10)	43(1)	52(1)	77(1)	0(1)	13(1)	-1(1)	
C(11)	45(1)	68(1)	79(1)	-2(1)	17(1)	-8(1)	
O(11)	76(1)	81(1)	81(1)	3(1)	21(1)	-24(1)	
N(12)	52(1)	80(1)	77(1)	-13(1)	23(1)	-12(1)	
C(13)	48(1)	62(1)	94(2)	-16(1)	23(1)	-5(1)	
C(14)	45(1)	53(1)	75(1)	-4(1)	15(1)	-1(1)	
C(15)	50(1)	70(1)	68(1)	3(1)	13(1)	-5(1)	
O(15)	49(1)	116(1)	100(1)	-28(1)	14(1)	-17(1)	
O(16)	53(1)	72(1)	73(1)	-10(1)	15(1)	-8(1)	
C(17)	63(1)	77(1)	65(1)	-4(1)	9(1)	-9(1)	
C(18)	78(2)	69(2)	66(2)	0	10(1)	0	
C(19)	94(2)	135(2)	87(2)	-28(2)	37(2)	-31(2)	
C(20)	61(1)	73(1)	77(1)	8(1)	10(1)	-12(1)	
C(21)	77(2)	114(2)	82(2)	12(2)	1(1)	-40(2)	

	Х	У	Z	U(eq)	
H(3)	-205	6965	4848	75	
H(6)	2650	7006	5549	75	
H(7)	5076	8205	6116	77	
H(9)	7646	9068	4639	72	
H(13B)	5296	5148	4070	80	
H(13A)	4303	4633	3452	80	
H(17B)	199	2793	2276	82	
H(17A)	-250	1659	2763	82	
H(18)	3010(40)	810(30)	2867(10)	84	
H(19A)	5480	7698	2726	155	
H(19B)	3983	6437	2623	155	
H(19C)	6186	6096	2780	155	
H(21)	10530(50)	10430(40)	6280(14)	119	

**Appendix III** (e). Hydrogen coordinates (  $x \ 10^4$ ) and isotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>) for XLi093 (**33**).

C(14)-C(1)-N(2)-C(3)	-0.3(2)
C(15)-C(1)-N(2)-C(3)	179.53(19)
C(1)-N(2)-C(3)-N(4)	0.5(2)
N(2)-C(3)-N(4)-C(14)	-0.4(2)
N(2)-C(3)-N(4)-C(5)	-176.01(19)
C(3)-N(4)-C(5)-C(6)	34.5(3)
C(14)-N(4)-C(5)-C(6)	-140.2(2)
C(3)-N(4)-C(5)-C(10)	-144.9(2)
C(14)-N(4)-C(5)-C(10)	40.4(3)
C(10)-C(5)-C(6)-C(7)	-1.9(3)
N(4)-C(5)-C(6)-C(7)	178.69(18)
C(5)-C(6)-C(7)-C(8)	0.5(3)
C(6)-C(7)-C(8)-C(9)	2.2(3)
C(6)-C(7)-C(8)-C(20)	-176.9(2)
C(7)-C(8)-C(9)-C(10)	-3.5(3)
C(20)-C(8)-C(9)-C(10)	175.54(19)
C(8)-C(9)-C(10)-C(5)	2.1(3)
C(8)-C(9)-C(10)-C(11)	177.80(18)
C(6)-C(5)-C(10)-C(9)	0.6(3)
N(4)-C(5)-C(10)-C(9)	-179.99(16)
C(6)-C(5)-C(10)-C(11)	-174.48(19)
N(4)-C(5)-C(10)-C(11)	4.9(3)
C(9)-C(10)-C(11)-O(11)	-32.8(3)
C(5)-C(10)-C(11)-O(11)	142.3(2)
C(9)-C(10)-C(11)-N(12)	145.98(19)
C(5)-C(10)-C(11)-N(12)	-38.9(3)
O(11)-C(11)-N(12)-C(19)	3.0(3)
C(10)-C(11)-N(12)-C(19)	-175.8(2)
O(11)-C(11)-N(12)-C(13)	171.04(19)
C(10)-C(11)-N(12)-C(13)	-7.7(3)
C(11)-N(12)-C(13)-C(14)	73.5(3)
C(19)-N(12)-C(13)-C(14)	-118.2(2)
C(3)-N(4)-C(14)-C(1)	0.2(2)
C(5)-N(4)-C(14)-C(1)	175.81(18)

**Appendix III** (**f**). Torsion angles [°] for XLi093(**33**).

C(3)-N(4)-C(14)-C(13)	178.00(19)
C(5)-N(4)-C(14)-C(13)	-6.4(3)
N(2)-C(1)-C(14)-N(4)	0.1(2)
C(15)-C(1)-C(14)-N(4)	-179.8(2)
N(2)-C(1)-C(14)-C(13)	-177.1(2)
C(15)-C(1)-C(14)-C(13)	3.0(4)
N(12)-C(13)-C(14)-N(4)	-64.7(3)
N(12)-C(13)-C(14)-C(1)	112.3(3)
C(14)-C(1)-C(15)-O(15)	-176.2(2)
N(2)-C(1)-C(15)-O(15)	3.9(3)
C(14)-C(1)-C(15)-O(16)	4.6(3)
N(2)-C(1)-C(15)-O(16)	-175.23(18)
O(15)-C(15)-O(16)-C(17)	2.4(3)
C(1)-C(15)-O(16)-C(17)	-178.41(17)
C(15)-O(16)-C(17)-C(18)	-161.41(18)
O(16)-C(17)-C(18)-C(17)#1	-52.04(12)
C(9)-C(8)-C(20)-C(21)	-78(6)
C(7)-C(8)-C(20)-C(21)	101(6)

Symmetry transformations used to generate equivalent atoms: #1 -x+1/2,y,-z+1/2

## **IX. Appendix VII**: SBIR grant by Physiogenix

# **Research Strategy**

## A. Significance

Alzheimer's disease (AD) is a complex and progressive neurodegenerative pathology that currently affects 5.4 million Americans at an estimated total cost of \$200 billion [Alzheimer's Association, 2012]. Although the monetary costs greatly impact our economy, the emotional and social burdens the disease imparts on individuals and their families are immeasurable. The complexity of the disease is due in part to the multiple stages of AD, ranging from pre-symptomatic to severe dementia. As the disease progresses into the symptomatic phase of pathogenesis, individuals begin losing basic cognitive and learning abilities, which eventually can lead to the complete loss of memory, cognition, and executive functions [1, 2]; a point at which many AD sufferers lose the freedom of independent living. Along with the complexity of defining the exact causes of AD, there is a multitude of controversies and gaps in our understanding of the disease progression, which has made the treatment and prevention of AD especially difficult. Furthermore, current treatments for AD have variable successes [2-5] and demonstrate efficacy in approximately half of the treated individuals [National Institute of Aging, Alzheimer's Association, 2012], thus substantiating the need for new approaches to AD therapeutics [1, 2]. Here, we propose our ligand PWZ-029, an inverse agonist with functional selectivity at gamma-aminobutyric acid class A (GABA<sub>A</sub>) receptors containing  $\alpha$ 5 subunits, for the mitigation of cognitive deficits associated with AD and age-related dementias.

*Current treatments of Alzheimer's disease are largely ineffective and recent approaches to treating and preventing the disease have lost momentum in clinical trials.* Currently, there are five drug therapies that are FDA-approved for the treatment of

cognitive symptoms associated with AD. Cholinesterase inhibition is the mechanism of action utilized by four of the five AD therapies (e.g., donepezil) while N-methyl-Daspartate (NMDA) antagonism constitutes the action of memantine, which is prescribed to treat moderate to severe AD [2]. Unfortunately, the efficacy of these drugs is variable, with positive results reported in only half of the users [3-5]. Thus, the necessity to further characterize the mechanism of disease progression, which will theoretically identify novel targets for drug therapies, is increasingly evident. Since the mid-1990s, the bulk of AD drug development efforts have surrounded the amyloid cascade hypothesis, which postulates that the neurotoxicity associated with AD pathogenesis is largely the result of alterations in the processing of amyloid precursor proteins in the brain. The result of these alterations can lead to the production and eventual aggregations of plague-forming amyloid-beta (A $\beta$ ) peptides [2, 6-9]. Although the presence of extracellular A $\beta$  plaques and intracellular neurofibrillary tangles are generally recognized as hallmarks of AD, recent attempts to commercialize compounds that reduce the accumulation of or increase the clearance of A $\beta$  have been met by failure during clinical trials [2]. Thus, new approaches to treating or preventing AD are of immediate need. Recent approaches to characterizing the neuronal processes and alterations that result in or from hallmark anatomical changes of AD pathology are gaining momentum [10-12], thereby leading to novel targets for treating and preventing AD, such as the GABA<sub>A</sub> receptor complex [2, 7, 8, 13-15]. To this end, we aim to develop PWZ-029, a selective inverse agonist of  $\alpha$ 5GABA<sub>A</sub> receptors, to mitigate cognitive deficits associated with early (i.e., mild and moderate) stages of AD, as well as age-associated dementias.

The GABA<sub>A</sub> receptor complex is the primary modulator of neuronal inhibition and possesses diverse functionalities. The pathogenesis of AD targets neuronal networks that are especially vulnerable to the toxic effects of disease progression. Cholinergic neurons of the forebrain and the glutamatergic receptors and neurons in the

hippocampus are prime examples of networks affected by AD pathology. These excitatory networks of the brain are predominantly modulated by gamma-aminobutyric (GABA) neurotransmissions. GABA is the chief inhibitory neurotransmitter of the mammalian central nervous system (CNS) [7, 8, 14, 16, 17] and is regulated by three classes of GABA receptors, GABA<sub>A</sub>, GABA<sub>B</sub>, and GABA<sub>C</sub>, with the GABA<sub>A</sub> receptor complex being the primary regulator of GABA-induced CNS depression [7, 8, 16] and the primary target for benzodiazepine (BZ) pharmaceuticals [7, 8, 14, 17].

The GABA<sub>A</sub> receptor complex is a pentameric, chloride-ion-gated channel through which GABA regulates fast synaptic inhibitory neurotransmissions [7, 16]. The heterogeneity of this ionotropic receptor is due to the diversity of subunit compositions. Characterizations of the GABA<sub>A</sub> receptor complex define 8 subunits with a total of 20 different subtypes:  $\alpha$ 1-6,  $\beta$ 1-4,  $\gamma$ 1-3,  $\rho$ 1-3,  $\delta$ ,  $\varepsilon$ ,  $\pi$ , and  $\theta$  [13, 16-24]. Although the exact characterization of subtype and subunit composition in relation to GABA<sub>A</sub> receptor functionality is yet unclear, research supports that the majority of these complexes are composed of at least two  $\alpha$  and  $\beta$  subunits [14, 16, 25-27] and one  $\gamma$  subunit [8, 14, 16, 27]. These findings suggest that GABA<sub>A</sub> receptors have immense structural and functional diversity, and the implications for this diverse functioning potential have profound implications for the specificity and resulting pharmacological effects of ligands targeting the  $GABA_A$ complex [14]. For instance, many established BZ drugs (e.g., diazepam) have nonspecific binding behaviors, which have been shown to elicit side effects like sedation, hypnotic behavior, and affected motor coordination [14]; side effects that are not attractive for AD therapeutics aiming to improve memory and cognition. The  $\alpha$ 1 subtype, for example, is ubiquitously distributed in mammalian CNS [13, 14, 16] and has been shown to play a paramount role in the sedative side effects associated with ligands of the GABA<sub>A</sub> receptor [13, 14, 28]. Conversely,  $\alpha$ 5 subtypes, which are targeted by PWZ-

029, are predominantly concentrated in the hippocampus, specifically in the Ammon's horn of the dorsal CA1 region of rodent brains [13, 14, 16], and are characterized as modulators of memory and cognition [13, 14, 16, 24, 29-31]. These findings support the hypothesis that highly specific ligands that selectively bind α5 subtypes are promising targets for potent mitigators of AD-associated cognitive decline [15]. Our preliminary data for PWZ-029 substantiates this hypothesis and shows practical evidence that our compound significantly improves cognition without the sedative side effects common to other BZ ligands.

Down-regulating compensatory mechanisms related to AD pathology may reduce the cognitive impairments associated with mild and moderate stages of the disease. Neuronal death that is commonly observed in the brains of AD patients has been shown to be caused partly by overstimulation of NMDA and 2-amino-3-(3-hydroxy-5-methyl-isoxazol-4-yl) propanoic acid (AMPA) receptors via toxic amounts of the excitatory neurotransmitter, glutamate [32, 33]. Researchers speculate that these overstimulated pathways lead to compensatory mechanisms that aim to maintain the excitatory/inhibitory (E/I) balance of the brain's regulatory processes. In other words, a pathologically overactive excitatory pathway can lead to the compensatory up-regulation of inhibitory processes, specifically GABA-mediated CNS inhibition [7, 8]. These findings, in combination with research showing the relative conservation of the GABAergic hippocampal networks in AD pathology [7, 29, 34-37], suggest that inverse agonist ligands that selectively act on GABA<sub>A</sub> receptors to inhibit the action of GABA in the brain are excellent candidates for memory and cognition therapeutics [15], such as PWZ-029.

### B. Innovation

Preliminary preclinical data strongly suggests that PWZ-029 is effective in improving memory and cognition, especially in models with baseline cognitive deficits. The high specificity of PWZ-029 for α5-containing GABA<sub>A</sub> receptors has successfully mitigated adverse side effects of seizures and sedation, which are commonly observed with benzodiazepines (BZ). Furthermore, the unique functional selectivity of PWZ-029 has successfully overcome functional interference and side effects that have stopped previous attempts to commercialize nonselective inverse agonists of the GABA<sub>A</sub> complex. Furthermore, because the core chemical structure of PWZ-029 classifies it as a benzodiazepine (Figure 1A) we are confident that the safety and toxicology experiments proposed here would be consistent with the vast body of research demonstrating the established safety and toxicological profiles of the BZ class of compounds.

Our collaborator, Dr. James Cook, synthesized PWZ-029 as part of an effort to explore the difference between the pharmacophores for the diazepam-sensitive versus the diazepam-insensitive GABA<sub>A</sub> receptors [38]. One of the products of his laboratory's research was PWZ-029, which was designed to be a partial inverse agonist with functional selectivity for the  $\alpha$ 5 subunit-containing GABA<sub>A</sub> receptor [39]. Our approach to mitigating the cognitive impairments of AD is innovative because PWZ-029 acts with a novel GABA-reducing mechanism rather than targeting cholinergic, glutamatergic, or A $\beta$ pathways. Success in the proposed experiments will drive our team towards commercializing a benzodiazepine drug that improves memory and cognition in AD patients without eliciting unwanted sedative and convulsive side effects. These strides in our research will not only lead to a marketable product with strong intellectual property, but we will also publish our sensitive findings so that scientific understanding for treatment and mitigation of AD will advance.

## C. Approach

## C1. PHASE I FINAL REPORT/PRELIMINARY STUDIES

Our successful Phase I milestone data combined with significant findings from our company-sponsored preliminary studies encouraged us to pursue a Phase II SBIR grant for our cognition-enhancing compound, PWZ-029. In summary, our significant findings demonstrate: 1) a selective binding affinity of PWZ-029 to α5 subunits on GABA<sub>A</sub> receptors, which was substantiated by an auspicious in-vitro efficacy profile, 2) minimal behavioral side effects of PWZ-029 compared to benzodiazepines and other inverse agonists with non-specific binding affinities, and 3) the ability of PWZ-029 to improve cognition and overcome impaired baseline performance in a myriad of memory tests that were carried out in multiple species.

**In-vitro characterization of binding affinity and efficacy.** Extensive in vitro binding and efficacy experiments that showed a selective affinity of PWZ-029 to  $\alpha$ 5GABA<sub>A</sub> receptors (Figure 1.); in vitro data that were exemplified by enhanced performance in behavioral assays [39], specifically the contextual fear conditioning paradigm [30] (Figure 3). These data, in addition to binding and electrophysiological assays carried out by our collaborator, Dr. James Cook (University of Wisconsin-Milwaukee), demonstrate that PWZ-029 binds selectively (i.e.,  $\approx$ 20- to 30-fold greater selectivity) to  $\alpha$ 5 subunitcontaining receptors compared to the other isoforms of the GABA<sub>A</sub> receptor complex, specifically the  $\alpha$ 1 subunit-containing receptors, which are associated with sedative side effects of non-selective BZ ligands (Figure 1B and 1C). Electrophysiology experiments in *Xenopus* oocytes demonstrated that PWZ-029, at a concentration of 1  $\mu$ M, significantly reduced GABA initiated control currents, thereby indicating a partial inverse agonist effect of PWZ-029 on GABA<sub>A</sub> receptors possessing the  $\alpha$ 5 isoforms (Figure 1B). Additional experiments carried out by the NIMH Psychoactive Drug Screening Program (PDSP) at the University of North Carolina Chapel Hill Medical School showed that PWZ-029 is a highly selective ligand of the BZ-binding site, confidently indicating that the observed cognitive enhancing effects elicited by this ligand are likely the result of binding at the GABA<sub>A</sub> receptor complex and not other receptor classes, such as cannabinoids, histamines, prostaglandins, glutamate, Ca<sup>+</sup> channel, transporters, adrenergic, acetylcholine, dopamine, serotonin, and opiates (data not shown).



C.	Ligand	MW	α1	α2	α3	α4	α5	α6
	PWZ-029	291.7	362.4	180.3	328.2	ND	6.2	ND

Figure 1. Structural, electrophysiological, and binding characterizations of PWZ-029 show the ligand is a partial inverse agonist with functional selectivity and in vitro efficacy at  $\alpha$ 5 subunit-containing GABA<sub>A</sub> receptors. <u>Panel A</u>. Chemical structure of PWZ-029 classifies the compound as a benzodiazepine, a class of compounds with a well-characterized safety profile. <u>Panel B</u>. Oocyte electrophysiology data demonstrating the selectivity of GABA attenuation by inverse activity with  $\alpha$ 5 isoforms. <u>Panel C</u>. Binding affinity (Ki, nM) and selectivity of PWZ-029 against the various  $\alpha$ -containing GABA<sub>A</sub> receptors.

Preliminary profiling of PWZ-029 pharmacokinetics, maximum tolerated dose, and oral bioavailability (data not shown). In our Phase I rat experiments, we showed that peak plasma concentrations were detected by LC/MS at 0.25 hours (h) post-dose when PWZ-029 was administered by intravenous (1 mg/kg, i.v.) and oral gavage (5 mg/kg, p.o.) routes. These preliminary data indicate a rapid absorption into the blood compartment via the oral route. The concentration-time curves for both i.v. and p.o. administrations indicated at least two separate phases of redistribution with significant amounts of compound in the plasma compartment at 4 h and remaining detectable for 24 h post-dose. The bioavailability of PWZ-029 was found to be roughly 11% when calculated by the area under the curve method and when accounting for the larger concentration used for the p.o. dose. In addition, brain samples were taken from rats 2 h after dosing to determine the potential of PWZ-029 to traverse the blood brain barrier. PWZ-029 was detected in brain tissue at 2 h after oral dosing and represented 2.1% of peak plasma concentration. A brain/blood ratio of 0.7 was determined for PWZ-029, with a brain/blood ratio of greater than 0.3-0.5 serving as our reference for sufficient access to the target organ. This was the first report showing both the oral bioavailability and target organ accessibility of PWZ-029.

Additionally, a dose escalation experiment investigated oral doses of PWZ-029 at 30, 100, 300, and 1000 mg/kg. Doses were given in single administrations, animals were observed for toxicological effects for 48 h post-dose, and then the next escalating dose was administered with the same experimental design followed for all phases of the escalation paradigm. All experimental animals survived each arm of dosing, and body weight loss was not observed for the duration of the experiment. Although mild tremors and facial twitches were noted, specifically at 300 and 1000 mg/kg, there were no overt signs of toxicity observed.

**Minimal side effects observed with PWZ-029 administrations in mice, rats, and non-human primates.** Single-dose proconvulsant studies utilizing male Swiss mice (CrI:CFW(SW)) treated with pentylenetrazole (PTZ) were conducted at NINDS and at PsychoGenics (Tarrytown, NY). Results demonstrated that PWZ-029 is not proconvulsant at any dose (10, 20, or 40 mg/kg) in the PTZ mouse model (data not shown). Furthermore, Harris et al. [30] found that systemic dose administrations in mice of PWZ-029 at a dose of 30 mg/kg did not result in convulsions.

Our Phase I experiments utilized a standard battery of behavioral and motor coordination tests to elucidate side effects that are common to the BZ class of compounds. The rotarod test was used to assess motor coordination effects of single oral doses of PWZ-029 at 5, 10, and 20 mg/kg versus saline- and diazepam-treated rats in the Dahl/SS rat strain. No signs of impaired motor coordination or sedation were observed at any of the three PWZ-029 doses (data not shown). The Dahl/SS rat strain was also utilized for the elevated plus maze and the open field test. Oral doses of PWZ-029 at 5, 10, and 20 mg/kg were tested against saline-treated and DMCM-treated rats in both assays. The combined results of these experiments suggest that PWZ-029 activity does not elicit anxiogenic side effects relative to the DMCM- or diazepam-treated control rats (data not shown). We suspect that the overall data suggest a mild sedative effect of PWZ-029; however, a sedative effect was not observed in the rotarod studies.

Targeted studies in monkeys were conducted in order to assess possible anxiolytic-like effects, abuse potential, and sedative-motor effects. In a rhesus monkey conflict model of the anxiolytic effects of drugs [40], PWZ-029 was ineffective over a 1000-fold dose range (up to 1.0 mg/kg, i.v.), suggesting that although this compound is a weak agonist at  $\alpha$ 1,  $\alpha$ 2, and  $\alpha$ 3 subunit-containing receptors, the efficacy at these subtypes is insufficient to induce anxiolysis-like behavior or alter operant responding. Similarly,

PWZ-029 lacked reinforcing effects over a 300-fold dose range in monkeys selfadministering the BZ, midazolam (i.v.), under a progressive-ratio procedure, suggesting that this compound has no abuse potential (data not shown).

Figure 2 represents the results of sedation studies in rhesus monkeys treated with 3.0 mg/kg PWZ-029 (i.v., panel A) or alprazolam (i.v., panel B) and then monitored for 120 minutes (min) post-dose. PWZ-029 did not induce significant signs of sedation at doses 200-300 times higher than the lowest dose shown to enhance performance in cognitive tasks (see below). Conversely, the non-selective BZ, alprazolam, induced significant rest/sleep posture and deep sedation, thus providing a positive comparator.



Figure 2. Sedation studies carried out with PWZ-029- (A) and alprazolam- (B) treated rhesus monkeys (n=5). Data were obtained as modified frequency scores by trained observers unaware of the treatments or purpose of the studies (interrater reliability scores > 90%). "Rest/sleep posture" was defined as the monkey assuming a posture associated with rest or assumed sleep, with eyes closed for at least 3 sec, but responsive to an auditory cue. "Deep sedation" was recorded if the monkey assumed an atypical posture, with eyes closed for at least 3 sec, and non-responsive to an auditory cue. These measures are adapted from the American Association for Anesthesiology standards for levels of sedation. All data presented as mean  $\pm$  SEM.

<u>PWZ-029 demonstrates efficacy in the attenuation of cognitive deficits in mice,</u> <u>rats, and NHP.</u> Assessments of the contextual fear response (i.e., % freezing) to stimulus (Figure 3) in scopolamine-treated, male C57BL/6 mice demonstrated that PWZ-029 attenuated the scopolamine-induced impairment of contextual memory (Harris, et al., 2008).



Figure 3. Pavlovian fear conditioned contextual memory test in scopolamine-treated C57BL/6 mice. Each mouse was injected via intraperitoneal injection with vehicle (white bar), 1.5 mg/kg scopolamine (black bar), or 1.5 mg/kg scopolamine plus 10 mg/kg PWZ-029 (gray bar). N=12 mice/group. \*P<0.05 versus scopolamine-treated mice (Harris et al., 2008)

In a company-sponsored Morris Water Maze (MWM) experiment, we examined the effects of PWZ-029 in a novel rat model of Alzheimer's disease (AD) - the FAB-Samaritan rat (Taconic Farms, Inc., Hudson, NY). FAB-Samaritan rats were carried through the MWM task (1 day of visible acquistion followed by 2 days of hidden platform tests) in subsequent weeks following brain infusions of an AD-inducing solution [41]. Infusion pumps were installed in the animals, and rats received brain infusions of saline (positive model control) or a proprietary FAB solution (negative control and test groups). Animals (n=8/group) were dosed for ≈28 days via p.o. gavage with vehicle (positive model and negative controls), donepezil (positive drug control, 3 mg/kg), or PWZ-029

(10 mg/kg) at ≈30 minutes prior to entering the maze. In the second arm of testing for the animals, PWZ-029-treated rats performed significantly better on Day 2 of the hidden phase in both latency to platform and distance travelled when compared to the FAB-Samaritan negative control animals (P<0.05 versus negative control, data not shown). As an additional part of our SBIR Phase I project, we conducted preliminary experiments to evaluate the extent to which PWZ-029 could alter memory processes in a translationally-relevant primate species. Figure 3 shows very promising results with PWZ-029 in the Object Retrieval with Detours (ORD) task in rhesus monkeys. The ORD task in monkeys is often described as providing a measure of executive functioning [42, 43]. Impairments in executive functioning (i.e., decision making, planning, response inhibition, cognitive flexibility) often emerge after mild cognitive impairment in the progression to AD [44]. Importantly, Ballard et al. reported that an inverse agonist with selectivity for α5-containing receptors enhanced performance in the ORD task, providing initial proof-of-concept support for our studies [42].

The ORD task consists of a food-retrieval procedure in which a small food treat (e.g., banana slice) is placed in a transparent box with one open side, and the animal's ability to retrieve the treat is measured (see Figure 4). Performance in the ORD task was not enhanced (mean percentage of successfully-completed trials in a daily session) by i.v. injections of PWZ-029 when the task included easy trials (i.e., open side directly in front of monkey, Figure 4A); however, performance was robustly enhanced when trials were difficult (i.e., open side on right or left, treat deep within the box, Figure 4B). When performance on the ORD task was attenuated by the cholinergic agent, scopolamine, PWZ-029 treatment effectively reversed this deficit (Figure 4C). Finally, the enhancement of ORD performance induced by PWZ-029 could be blocked by an antagonist selective for  $\alpha$ 5-containing GABA<sub>A</sub> receptors (Figure 4D). Together, these

findings provide strong support for PWZ-029 as a cognitive enhancer of executive functioning in a primate species with close genetic similarity to humans.



Figure 4. Effects of PWZ-029 in the **Object Retrieval with Detour (ORD)** task. Monkeys were trained to retrieve food from an opaque box with one open end (above right). Panel A. PWZ-029 was when tested with а combination of Easy and Difficult Trials. PWZ-029 В. enhanced performance when only Difficult Trials were tested (\*p<0.05 vs. vehicle, V). Panel C. Scopolamine (S, 0.01 mg/kg) inhibited ORD performance, which was reversed by PWZ-029 (\*\*p<0.05 vs. S). Panel D. Enhanced performance with PWZ-029 in Difficult Trials was blocked by the antagonist XLi-093 (\*p<0.05 vs. V; \*\*p<0.05 vs. PWZ-029 alone, P).

#### **C2. EXPERIMENTAL APPROACH**

Aim 1. Define the exposure/safety relationship of PWZ-029 in rats to determine whether this compound exhibits a sufficient level of safety to support Phase I clinical trials. The safety and potential target-organ toxicity of a drug must be established to demonstrate a sufficient efficacy/safety margin prior to entering a compound into clinical trials [45]. Here, escalating-dose and 7-day repeat dosing studies will be carried out in rats with the goal of establishing the no observable adverse effects level (NOAEL), dose-limiting toxicities (DLT), maximum tolerated dose (MTD), and the safety/exposure relationship of PWZ-029. We will use these data to support the definitive, IND-enabling GLP safety studies needed prior to initiating a Phase I clinical trial.

**Rationale:** Determining the toxicology and safety of a compound prior to testing in humans is crucial to the process of drug development [45]. As multiple benzodiazepines are currently on the market, their general safety profile is well documented. To establish the specific safety profile of PWZ-029, a series of safety/toxicological experiments will be carried out to assess the MTD and NOAEL in acute and multiple dosing paradigms. The toxicology assessments will be carried out in Sprague Dawley (SD) rats. We will determine the general safety profile of PWZ-029 and incorporate pharmacokinetic measurements to establish its safety/exposure relationship. Potential differences in toxicological responses between genders will also be evaluated. In parallel to the in vivo safety studies, standard in vitro assays for selectivity, mutagenicity, and clastogenicity/aneugenicity will be conducted to assess the potential for 'off-target' toxicity.

**Experimental Design**: Information gained from the escalating-dose MTD studies will drive the doses of PWZ-029 to be used in subsequent 7-day toxicological studies, and the outcomes of these studies will then determine the ultimate design of the GLP-safety

studies. A preliminary dose-ranging study in rats (using doses up to 1000 mg/kg) has already been conducted as part of an initial safety evaluation of PWZ-029. Although no overt findings of toxicological significance were observed, increased facial movements (twitching) were noted, especially at 300 and 1000 mg/kg. Therefore, a second MTD study will be conducted in rats using doses up to 2000 mg/kg, paying particular attention to potential abnormal muscular activity (i.e., twitching/tremors).

In addition to the Phase I single dose pharmacokinetic (PK) analysis that has already been conducted with PWZ-029 in rats, PK analysis will be carried out after a single dose and after 4 daily doses to assess the potential for drug accumulation or the potential for induction of liver metabolism that may affect drug exposure upon chronic dosing. Results from these rat safety studies will be used to seek funding from other sources for further study and to design PK, single-dose MTD, and sub-chronic safety studies in beagle dogs. Additionally, in vitro safety assessment will be conducted by an established contract research organization (Cerep Inc., Redmond, WA). The desired outcome of these studies is to further define the safety and selectivity profile of PWZ-029.

<u>PK analysis to establish safety/exposure ratio in rats</u>. Three male SD rats (8-weeks old, 250 g) will be dosed p.o. with 10 mg/kg of PWZ-029. Blood will be sampled after 0.08, 0.25, 0.5, 1, 2, 4, 8, 18 and 24 h to assess PWZ-029 exposure and to determine single dose PK parameters. Subsequently, rats will be dosed at 24 h intervals, and after the fourth daily dose, blood will be sampled following the schedule utilized on Day 1 to determine whether multiple-day dosing affects any of the PK parameters.

<u>Escalating-dose MTD studies</u>. Male SD rats (8-weeks old, 250 g) will be dosed via p.o. gavage with either vehicle (n = 5) or with escalating doses of PWZ-029 (100, 500 and 2000 mg/kg; n = 5/group). On day 1 of the study, rats will be dosed with vehicle or the lowest dose of PWZ-029 (100 mg/kg) and then observed for behavioral effects. For the first 120 minutes, animals will be intensely monitored for signs of acute toxicity (i.e.

morbidity and mortality) with special attention being paid toward unusual muscular activity (i.e., twitching, tremors and/or convulsions). Animals will be subsequently monitored for an additional 46 h for overt signs of toxicity such as ataxia, greater than 10% decrement in body weight, tremors, loss of righting response, morbidity, and mortality. Following the initial dose, rats will be dosed at 48 h intervals with either vehicle or the escalating doses of PWZ-029 and monitored as described for the 100-mg/kg dose. Forty-eight hours after the final vehicle or 2000 mg/kg dose of PWZ-029, animals will be terminated and gross pathological observations documented. Blood will be collected to assess clinical chemistries, and the adrenals, brain, heart, kidneys, liver, spleen, thyroid with parathyroid, skeletal muscle and thymus will be weighed, collected, and stored in formalin for histological analysis.

<u>In vivo safety assessment in rats (7-day repeat dose study)</u>. A low, mid, and high dose of PWZ-029 will be established based upon the findings from the dose-ranging experiments. A total of 32 SD rats will be distributed among 8 experimental groups.

- 16 female SD rats will be distributed into 4 experimental groups (n=4/group): Control group, low dose group, mid dose group and high dose group.
- 16 male SD rats will be distributed among 4 dose groups as with the females.

All experimental animals will be dosed via p.o. gavage once a day for 7 consecutive days. Detailed clinical observations including unusual muscular activity (i.e., twitching, tremors and/or convulsions), body weights, and food consumption will be recorded daily. On day 7, blood will be collected 1 h after administering the final dose for drug exposure and clinical chemistries. Animals will be terminated, and the adrenals, brain, heart, kidneys, liver, spleen, thyroid with parathyroid, skeletal muscle, and thymus will be collected, weighed, and stored in formalin for histology. The 1 h post-dose timing was chosen based on the PK data obtained from Phase I experiments.

<u>In vitro selectivity and gene safety assessment</u>. The potential for off-target pharmacology/toxicology will be assessed using a broad panel of ~150 human receptors and enzymes The potential for genetic toxicity will be assessed using the Ames fluctuation assay and the in vitro micronucleus assays run in the presence and absence of S9 liver fractions (Cerep Inc., Redmond, WA).

<u>Benchmark for Success</u>. Results from the PK study will refine the PK parameters estimated in the preliminary studies that indicated rapid oral absorption of PWZ-029 into the blood compartment with a bioavailability of about 11% and biphasic elimination kinetics. Success will be determined from the escalating-dose MTD studies if we observe of a sufficiently high MTD to proceed to a multiple-dose safety assessment in rats. Successful outcomes of the 7-day repeat dosing study will establish a relationship between chronic dosing and DLT, define a therapeutic index (TI) sufficient for continued development of PWZ-029, and demonstrate that PWZ-029 does not induce its own metabolism nor show drug accumulation upon multiple-day dosing. Successful outcomes will Finally, in vitro analyses will be successful will demonstrate that PWZ-029 does not induce its own metabolism nor show drug accumulation upon multiple-day dosing. Finally, successful in vitro analyses will be defined as demonstration of an in vitro selectivity profile ≥50-fold and a lack of gene toxicity.

Anticipated Problems and Solutions. The increased facial movements (i.e., twitching) that were observed in the preliminary dose-ranging study may be confirmed or another unanticipated DLT may be discovered. If increased facial twitching is confirmed, we will determine whether it decreases or stops in a multiple-day dosing study (i.e., toleration). Assessment of any toxicological observations will require their assessment in multiple-day dosing studies and protocols will be designed to incorporate the necessary measurements. Identification of off-target activity in the selectivity panel would first require confirmation in secondary enzyme and/or cell-based assays. Confirmed off-

target activity would require the evaluation of potential side effects and further toxicology studies would need to incorporate assessment of these in study designs. Positive results in the genetic toxicity assays would require the identification of a back-up compound devoid of this activity.

Aim 2. Examine the potential of PWZ-029, an inverse agonist of  $\alpha$ 5GABA<sub>A</sub> receptors, to mitigate cognitive deficits in a transgenic mouse model of AD. Despite an ever-growing population of aged individuals, there are few, effective therapeutic options for individuals suffering from cognitive deficits associated with AD. To expand on our preliminary evidence that suggests inverse agonist action at  $\alpha$ 5-subunit containing GABA<sub>A</sub> receptors can mitigate AD-associated cognitive decline, we will test the efficacy of PWZ-029 to improve performance in the contextual-fear conditioning assay with the APPswe Tg2576 mouse model of AD. All behavioral experiments will be controlled with age-matched, non-transgenic littermates, as well as positive and negative drug controls (n=10 mice/group). From these experiments, we expect to gain proof-of-concept data that supports inverse agonist action at  $\alpha$ 5GABA<sub>A</sub> receptors as a novel therapeutic option for improving cognition in those suffering from AD.

**<u>Rationale</u>**: Research carried out to characterize the distribution of the various GABA<sub>A</sub> receptor subtypes in rodent brains shows that  $\alpha$ 5GABA<sub>A</sub> subunits are highly concentrated in the Ammon's horn of the hippocampus, specifically in the CA1 dorsal region [13, 14, 16]; a region of the hippocampus that is specifically targeted in context-dependent fear conditioning assays [46-50]. Furthermore, studies have shown that the  $\alpha$ 5GABA<sub>A</sub> subunits are spared in the hippocampus, even with the pathophysiological brain shrinkage and neuronal death that is commonly associated with AD [7, 29, 34-37]. To this end, we hypothesize that PWZ-029, an inverse agonist that has repeatedly demonstrated high selectivity for  $\alpha$ 5GABA<sub>A</sub> receptors, will improve performance in

context-dependent fear conditioning experiments, thereby establishing proof-of-concept for a novel therapeutic mechanism to mitigate AD-associated cognitive decline.

**Experimental Design**: The following experiments will have oversight by Dr. Catherine Kaczorowski, a leading AD expert from the Medical College of Wisconsin. The evidence gained from these experiments will guide the direction in which impactful and successful commercialization of PWZ-029 is most likely.

Formulation. Formulation of PWZ-029 will be carried out identically for all Aims. All compounds will be administered via p.o. gavage 30-60 minutes prior to entering the fear conditioning experiments. PWZ-029 will be synthesized by Cambridge Major Laboratories (Germantown, WI). PWZ-029 will be administered to the animals only on the day of training (see conditioning protocol below) at 10 mg/kg, a dose shown to improve performance on memory tasks in adult mice (Figure 3) and rats (refer to MWM data in the preliminary data section of this proposal), suggesting that a 10 mg/kg dose of PWZ-029 will enhance cognition in the APPswe model [30, 39]. Premeasured amounts of PWZ-029 will be dissolved in a widely used solvent that contains 85% distilled water, 14% propylene glycol, and 1% Tween 80 [30, 38, 39]. Donepezil (catalog # D6821), which will serve as the positive drug control, will be purchased from Sigma-Aldrich (St. Louis, MO), formulated in vehicle, and administered daily at 0.3 mg/kg for 6 weeks prior to the fear-conditioning assay, including the training day of the behavioral assay [51, 52]. Triazolam, a non-selective BZ agonist that has been shown to inhibit performance in fear conditioning assays [53] will serve as the negative drug control, and will also be purchased from Sigma Aldrich (catalog # T9772), formulated in vehicle, and administered at 0.03 mg/kg prior to the training phase of the fear conditioning protocol. To acclimate all mice to the same handling/dosing exposure and to maintain blinded treatments, PWZ-029 and triazolam groups will be dosed with vehicle for the 6-week period that donepezil will be dosed prior to the conditioning assay. There will be no

doses administered on the testing day when freezing behaviors are scored.

<u>Research Model</u>. Taconic Farms, Inc. (Hudson, NY) will provide male APPswe Tg2576 mice (001349-T-RD1) along with the age-matched, non-transgenic, wild type littermates (001349-W-RD1). Prior to being shipped to the PhysioGenix Animal Facility, the mice will be screened for the RD1 mutation, which causes blindness. Any animal carrying the mutation will be excluded and replaced. Animals will be shipped from Taconic at  $\approx$ 6 months of age and housed at PhysioGenix until the animals are aged to 9-12 months, which is the age where A $\beta$  deposition notably affects performance in memory tasks [54]. Body weights will be collected weekly for the entire duration that the animals are housed at PhysioGenix as a means of monitoring animal health and to calculate dosing volumes for each study animal.

Contextual fear conditioning assay to assess associative learning and memory. Behavioral assays will be carried out in an isolated laboratory where traffic and sound inference is negligible. Experiments will be carried out during the light phase of a 12/12hr light/dark cycle, and all technicians will be blinded to the treatments of the animals. Fear chambers will be procured from Med Associates (St. Albans, VT). Study animals will be acclimated to the experimental room for 60-90 minutes before entering into the fearconditioning assay. After the period of acclimation, an animal will be placed into the fear chamber, which will serve as the novel context, i.e., conditioned stimulus. The animal will be allowed to explore the novel context for 148 s, at the end of which a foot shock (duration of 2 s at 0.4 mA) will be introduced [55, 56]. The animal will remain in the chamber for 30 s following the foot shock before returning to the home cage. Twentyfour hours later, the animal will be returned to the experimental room, allowed to acclimate in the room, and placed back into the chamber. Freezing, which will be defined as the complete absence of movement, except for respiratory actions, will be scored every 10 s for 4 min.

<u>Benchmark for Success</u>. Success will be defined as a significant increase (P<0.05) in percent freezing at the 24-hour time point for APPswe transgenic and/or non-transgenic mice treated with PWZ-029 relative to the negative controls (vehicle and triazolam).

<u>Anticipated Problems and Solutions</u>. If differences are not observed between the treatment groups, we will consult Dr. Kaczorowski for procedural amendments that may enhance the differentiating potential of the fear conditioning assay (e.g., condition mice to the context prior to administering unconditioned stimulus training). We may still find that PWZ-029 does not enhance performance in this mouse model and in this assay. If these results are encountered, in our next round of funding, we will test the ability of PWZ-029 to mitigate cognitive deficits in mouse models of 'normal' aging.

Aim 3. Examine the potential of PWZ-029 to enhance executive function and memory in rhesus monkeys. Our research during the SBIR Phase I project showed that PWZ-029 enhanced performance on the ORD task of executive function in a manner consistent with a role for  $\alpha$ 5-containing GABA<sub>A</sub> receptors. We first will expand these findings to assess specific aspects of executive function, including set-shifting and response inhibition. In addition, we will introduce studies on working memory—a cognitive domain regulated by and influencing executive function [57]. Notably, deficits of this domain are considered a hallmark of AD [2, 44]. Finally, we will evaluate fundamental processes that may influence results from executive function and memory tasks, specifically attention and motivation.

**<u>Rationale</u>**: In order to evaluate the extent to which PWZ-029 enhances executive function and memory, our proposal will utilize the Cambridge Neuropsychological Test Automated Battery (CANTAB). A clear strength of the CANTAB approach is that its tasks

are based on corresponding procedures developed for human patients, which will greatly facilitate our ultimate goal of translating findings from monkeys to humans. For executive function, we will use two tasks, including the intradimensional/extradimensional (ID/ED) set-shifting task, which evaluates the ability of a subject to acquire discriminations, attend to specific stimulus attributes, and shift attention to other attributes, thus providing an assessment of cognitive flexibility. To evaluate "impulsivity," a behavior shown to be impaired in AD [58], we will use the delayed reinforcement choice (DRC) task, in which monkeys must choose between a delayed, large reward or an immediate, small reward. To assess memory, we will use the self-ordered spatial search (SOSS) task, which is a short-term spatial working memory task similar to the radial-arm maze procedure used in rodents [59]. Attention will be assessed with the serial-reaction time (SRT) task [60], and motivational processes will be assessed with a progressive-ratio (PR) procedure. Altogether, these tasks will provide a comprehensive evaluation of the ability of PWZ-029 to enhance executive function and memory, as well as to assess the relative roles of attentional and motivational variables that potentially contribute to PWZ-029's ability to enhance cognition.

**Experimental Design**: Our overall approach will be similar to that described for our ORD results in the preliminary data section of this proposal (see Figure 4): (1) examine a range of doses of PWZ-029 (0.001-1.0 mg/kg) compared to a positive control (donepezil, 0.01 or 0.03 mg/kg [61]); (2) assess the ability of PWZ-029 to reverse the performance-attenuating effects of scopolamine (0.01 mg/kg, which will serve as a negative control; and (3) assess the ability of the preferential  $\alpha$ 5-containing receptor antagonist, XLi-093 (0.1-1.0 mg/kg), to block enhanced performance, if present, of a maximally effective dose of PWZ-029.

<u>Route of Administration</u>. In order to compare our results with the results from the Phase I studies, we will initially administer all drugs via the i.v. route using indwelling chronic venous catheters (implanted using the methods described in detail by Platt et al., 2011) [62]. The use of the i.v. route will also help to minimize limitations of interpretation due to differences in pharmacokinetic variables. Oral dosing is a key consideration, however, and in the event that a positive effect of PWZ-029 emerges for a cognitive task, we will evaluate targeted doses using the intragastric (i.g.) route via a nasogastric tube in awake monkeys. Our co-PI, Dr. Rowlett, has extensive experience with this methodology [63-69]. We will choose the maximally effective dose, assuming a decrease in potency when switching from i.v. to i.g. administration, and increase (or decrease) the dose by ½ log unit in order to adjust for potency differences across the two routes.

<u>Subjects</u>. Experiments will be conducted with adult male and female rhesus monkeys (*Macaca mulatta*), born and reared at the New England Primate Research Center breeding facility and/or obtained from outside sources. Based on our research during the Phase I grant, as well as power analyses derived from the literature, we are proposing a sample size of at least N=6 monkeys. For cognitive testing, monkeys will be prepared with a non-allergenic nylon vest (Lomir Biomedical, Inc., Malone, NY) and seated in primate chairs (Crist Instruments, Hagerstown, MD) enclosed in ventilated, sound-attenuating chambers (Med Associates). All animals will be monitored for food intake, and will undergo relatively mild food restriction, based on limiting caloric intake (for detailed methods, see Taffe, 2004 [70]). The monkeys will be weighed and assessed for body condition once per week. Weekly weighing of the conflict monkeys will occur throughout the studies to adjust feeding based on a combination of weight, performance, and overall health (assessed in consultation with the clinical veterinary staff).

<u>CANTAB tasks for assessing executive function, memory, attention, and motivation</u>. All monkeys to be used in this application will receive training to sit in the chair, with different stages of training for the individual tasks (most monkeys will be trained and ready by the project start date). During a given experimental session, the monkey will be placed into the chair and brought to a sound-attenuated chamber (Med Associates) with ventilation, white noise, and a CANTAB touch screen unit. Sessions typically last no more than 2 hours, and 1 or 2 tasks are conducted per session (usually a single session/day). During tests with PWZ-029 (administered 5-10 min before the session), only a single task will be scheduled.



Intradimensional/Extradimensional Set Shifting (ID/ED). This task consists of a series of four discrimination learning stages, plus reversals, wherein touching only one of two stimuli presented on the screen results in food pellet delivery (for summary, see Figure 5). Within any given stage of the task, a pair of stimuli is presented and the same stimulus is associated with reinforcement (S+ stimulus) until the performance criteria are met (18 of 20 consecutive trials correct). Correct choices must be made within 30 s, and following a correct choice, the screen is blanked for 5 sec while an incorrect choice (Sstimulus) results in a 0.2 s tone and a 9-s period of blank screen. Four distinct stimulus sets will be used in 4 stages. In the first stimulus set (stage 1A), two distinctly shaped stimuli will be presented, with touches on one shape resulting in food pellet delivery. Stage 1B will consist of a stimulus reversal, in which the same two shapes are retained, but pressing the S+ stimulus now does not result in reinforcement, whereas touching the former S- stimulus will result in food pellet delivery. In Stage 2A, a compound discrimination is presented. For this discrimination, the two shapes from Stage 1B are present, but additional stimuli, consisting of lines, will be superimposed onto the existing shapes. Because the shape discrimination from the previous stage does not change, the lines are irrelevant to this discrimination. Stage 2B will consist of a shape reversal, similar to Stage 1B.

Stage 3A will consist of the ID shift stage. For this discrimination, two new shapes with new lines will be presented. This is considered an ID shift due to the fact that despite new shapes and lines being introduced, the shape remains the relevant dimension for the discrimination. Stage 3B will consist of a reversal of the shape S+ and S- from Stage 3A. In the final two stages, the ED shift will be introduced. Thus, in Stage 4A, new shapes and lines will be presented; however, the lines—not the shape—will be the relevant discriminative stimulus. Stage 4B will consist of a reversal of the line S+ and S-
contingencies. Performance is determined as the number of errors at each stage, and the data will be subjected to a square root transformation to achieve normal distributions [59]. Data will be analyzed with repeated measures ANOVA and Bonferroni t-tests for multiple comparisons.

<u>Delayed Reinforcement Choice (DRC)</u>. Our preliminary data with the ORD task suggested the possibility that PWZ-029 enhances the ability of the monkey to withhold or inhibit prepotent responding. The ID/ED task will allow recording of persistent incorrect responding, similar to barrier reaches in the ORD task. A prominent interpretation of this type of behavior is dysfunctional inhibitory control, or "impulsivity." A more complex form of inhibitory control involves choice behavior, which can be investigated using the DRC task, also commonly referred to as "delayed discounting" task.

The CANTAB "Impulsive Choice" task offers subjects a choice between two reinforcers, which can differ in magnitude, probability, and delay. We are focusing on delay of reinforcement in the DRC task, based on pharmacological studies [71, 72]. In this task, a



#### Figure 6. Delayed reinforcement choice task from the monkey CANTAB system. A. Schematic of the task.

**B.** Prediction of the effects of a cognitive enhancer on data analyzed by discount functions. The k rate is the slope of discounting for the hyperbolic function—in this example, k is reduced by the cognitive enhancer, indicating a higher degree of choice for the larger, but delayed, reward.

choice is offered between a relatively large reward (4 or more food pellets) and a small reward (1 food pellet). A trial will initiate by a touch of a single stimulus, which is replaced by two stimuli, referred to as the "delay" and "immediate" stimulus. Touching the delay stimulus results in a blank screen for a specified duration (e.g., 0-80 s), followed by a signaled delivery of the large reward. Touching the immediate stimulus will result in signaled delivery of the small reward with no delay. After reward, the screen will be blank for 10 s. Data will be analyzed initially by comparing the percentage of delay stimulus responding as a function of delay (typical delay sequence= 0, 10, 20, 40, 80 s, randomly presented). As the delay is increased, animals characteristically choose the large reward less [71] (for summary, see Figure 6A). An improvement in performance by PWZ-029 would be manifested as a significant increase in percentage of choices for the

delayed, yet larger reward, relative to control conditions. Data also may be analyzed using a hyperbolic discounting function [73], which provides a rate (k) of discounting value (Figure 6B). Enhancement of performance by PWZ-029 would be reflected as a significant decrease in the k value.

<u>Self-Ordered Spatial Search (SOSS)</u>. The SOSS task is a short-term spatial working memory task similar to the radial-arm maze procedure used in rodents (see Figure 7, [59]).



In each trial, two, three or four small, colored, rectangular boxes are displayed on the screen in positions randomly allocated from 16 possible locations. The monkey must touch a box within 30 s of stimulus onset. After each successful touch, the color of the touched box is briefly (100 ms) changed and the screen is then blanked and a reinforcer delivered. After a 2-s delay, the boxes are re-displayed and the monkey must touch a box that was not previously touched in the trial in order to receive a food pellet. The trial is completed when the animal has either touched all boxes without a repetition (correct), touched a box that had previously been selected in that trial (error), or failed to touch a box within 30 s of stimulus presentation (omission). A tone and a 4-s screen blank will follow errors and omissions. After 5 s, another trial is presented with stimuli in new (randomly allocated) positions. A session will consist of 40 trials grouped into 6 blocks that differ according to the number of boxes presented. The different blocks, with

number of trials in parentheses, will be: 2 boxes (5), 3 boxes (7), 4 boxes (7), 3 boxes (8), 4 boxes (8), and 2 boxes (5). Accuracy scores will be calculated for each trial type by dividing the number of correctly completed trials by the number of trials in which there was at least one response (omissions will be excluded from the calculation). Accuracy scores will be analyzed by repeated measures ANOVA and Bonferroni t-tests, with trial type and PWZ-029 dose as factors.

<u>Serial Reaction Time (SRT)</u>. This task is based on methods described in detail by Weed and Gold [60]. Briefly, the task consists of a discrete-trials procedure in which each trial begins with the presentation of a grid with 5 open circles (3 cm diameter) on the touch screen. The monkeys will be trained to press and hold a lever for a variable duration (between 0.75 to 2.5 s, randomly determined) and then to release the lever at the onset of a 2 cm yellow stimulus presented in one of the 5 circles. If the monkey touches this circle within 2 s, the screen will go blank and 2 food pellets will be delivered, and a 3 s inter-trial interval will begin. The yellow stimulus duration will consist of brief flashes of 0.02, 0.1, and 1.0 s durations. If the monkey fails to press the lever or fails to respond to the screen within the allotted time, or touches an incorrect circle, then a tone will sound and the screen will go blank for 4 s until the next trial, with no food reward given (for summary, see Figure 8).



Each session will consist of 45 trials with the 3 flash durations presented randomly 3 times in each of the 5 choice circles.

Data that will be analyzed include latency to release the lever following stimulus onset and the time taken to move the hand from the lever to the screen. Failure to touch the screen after pressing the lever will be considered an omission error, which will be analyzed as the average number of errors per trial. All data will be analyzed with repeated measures ANOVA with dose and stimulus duration as factors, as well as with Bonferroni t-tests for multiple comparisons.

<u>Progressive-Ratio (PR) schedule of reinforcement</u>. PR procedures consist of response requirements that progressively increase across a session until responding ceases (Figure 9). The last response requirement completed, termed "break point," provides a quantitative measure of the reinforcing effectiveness of a stimulus. For these studies, a colored box appearing in the middle of the screen will initiate a session. Touching the box once will result in food pellet delivery, and the response requirement will progressively increase following each reinforcer by an incremental value beginning at 1 and doubling after 8 response requirements are completed successfully. Thus, the

response requirement sequences will consist of: Increment=1, response requirements=1, 2, 3, 4, 5, 6, 7, 8; Increment=2, response requirements=10, 12, 14, 16, 18, 20, 22, 24; Increment=4, response requirements=28, 32, 36, 40, 44, 48, 52, 56; and so on.



Sessions will be a maximum of 20 min and will be terminated if 3 min elapses without a response. Performance will be measured as the break point and number of food pellets delivered per session. This procedure provides a measure of the ability of the animal to respond, as well as an assessment of "motivation" to perform. We do not anticipate that PWZ-029 will significantly alter the mean break points or number of food pellets delivered, which would rule out changes in motivational levels as an interpretation for changes in cognitive performance.

<u>Benchmark for Success</u>. Success will be defined as  $\alpha$ 5-containing receptor mediated enhancements in ID/ED, DRC, and SOSS performance, in the absence of robust changes in SRT and PR.

<u>Anticipated Problems and Solutions</u>. SRT enhancements might suggest that broader cognitive processes are augmented simply due to an enhancement in attention, which in and of itself might be therapeutically useful. Change in PR performance, however, is a more significant issue, given that increases in motivation might explain increases in task performance by a non-cognitive mechanism (e.g., enhanced food palatability). We are unaware of any precedent in the literature for this type of effect.

Recent evidence has suggested an age-dependency for cognitive-enhancing effects of inverse agonists at α5 subunit-containing GABA<sub>A</sub> receptors [15]. For our ORD studies, the monkeys generally were in the "middle-aged" category (≈15-20 years old), a time period documented to show mostly moderate cognitive impairments [57, 74] and a similar group of monkeys will be available for the present project. If warranted, future plans will involve targeted cognitive tests with PWZ-029 in aged monkeys (20+ years old) depending on availability of monkeys in this age range.

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#### **BINDING AFFINITY**

Experiment	Type of Test	Model				Fig	gure				Summary of Results
Functional and subtype selectivity of PWZ-029	Used to determine the subtype selectivity of PWZ-029 to the GABA <sub>A</sub> receptors, especially the α5 subtype.	Xenpus oocytes electrophysiology			a 1301 1301 1301 1302 1302 1302 1302 1302		α 1β 3γ2 α 2β 3γ2 α 3β 3γ2 α 5β 3γ2 π Μ 100 n Μ [P W Z-0	т т т т т т т т т т т т т т т т т т т			PWZ-029 concentrations of up to 1 μM demonstrated a significant 20% reduction of control GABA elicited current indicating a partial inverse agonist effect at GABA <sub>A</sub> receptors possessing the α5 subtype. PWZ-029 also exhibited agonistic selectivity for the α1 and α3 bearing GABA <sub>A</sub> receptors. P<0.01, Student's t-test.
	Used to determine the	Mouse fibroblast	Code	Affinity of MW	PWZ-029 for a1	αχβ3γ2 (x = α2	1-6) benzod a3	iazepine rece ¤4	ptor isoforms α5	α6	PWZ-029 binding affinity
Binding Affinity of PWZ-029	affinity of PWZ-029 to the GABA <sub>A</sub> receptor subtype, $\alpha$ 5.	$\begin{array}{c} \text{L(tk) cells} \\ \text{ffinity of PWZ-029} \\ \text{to the GABA}_{\text{A}} \\ \text{eceptor subtype, } \alpha 5. \end{array} \\ \begin{array}{c} \text{L(tk) cells} \\ \text{expressing} \\ \text{recombinant} \\ \text{human GABA}_{\text{A}} \\ \text{receptor subunits} \end{array}$	Merck	291.73	>300	>300 ND	>300 ND	ND ND	38.8	>300 ND	for α5 sutypes are near 60- fold times that of α1 subunits.
			UNC-Roth	291.73	362.4	180.330	328.2	ND	6.185	ND	
		receptor subunits	UNC-Roth	291.73	362.4	180.330	328.2	ND	6.185	ND	

## SIDE-EFFECTS

Experiment	Type of Test	Model	Figure	Summary of Results
Convulsant/ Proconvulsant Effects	Used to determine the convulsant/proconvulsant drug side-effects.	Male Swiss mice (Crl:CFW(SW)) treated with pentylenetetrazole (PTZ)	N/A	PWZ-029 did not cause convulsions at any concentration (10, 20, or 40 mg/kg) in the PTZ mouse model.
Rotarod	Rotarod tests are used to assess motor coordination. The longer an animal stays on the rotating rod, the more coordination their motor skills. The test further assesses drug side-effects, such as dizziness and muscle relaxation. If these effects are present the subject is expected to spend less time on the rotating rod.	Dahl Salt-Sensitive (SS) rats (n=12/group) were treated per os with either vehicle, diazepam, or one of three concentrations of PWZ-029. After a training period, the animals were subjected to the rotating rod test for a maximum of 300 seconds.	<figure><section-header></section-header></figure>	No significant decrease in motor coordination was observed at any of the three PWZ-029 concentrations. However, the success rate of rats increased with PWZ- 029 treatment.

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## **SIDE-EFFECTS Continued**

Experiment	Type of Test	Model			Figure			Summary of Results
Operant Responding	Very basically assesses the individual's response that is required to carry out cognitive tasks. This experiment is highly sensitve to sedative- motor effects of conventional benzodiazepines (e.g. alprazolam).	Rhesus monkeys treated with vehicle and then increasing concentrations of alprazolam or PWZ- 029. This Operant Responding task required the animal to press the lever 20 times to receive a food pellet.	Rate (Responses/s)	Operant Respondin 	ng 029 	- alpra	azolam	PWZ-029 showed no evidence of inhibition in the baseline levels of operant responding maintained by food.
Observational Studies	Based on defined behaviors observed in animals under baseline conditions, post-dose observations are indicative of drug side- effects.	Rhesus monkeys were treated with vehicle, alprozolam, zolpidem, or PWZ-029. Post- dose observations were documented as no change, increase, decrease or increase and decrease in respective behavior.	Table 2. And at least one of Compounds Alprazolam Zolpidem PWZ-029 *Loc was ass (Bonferroni t-	alysis of all behavioral categories following lose significantly different from vehicle (N=4-6 No Change Loc, Grm, Scr, Ste, For, Voc, Thr, Yaw, BSm, Pre, Dri, NRu, LSm, CSh, LDr Loc*, Grm, Scr, Ste, For, Voc, Thr, Yaw, BSm, Pre, Dri, NRu, LSm, RSP, MSe, CSh, LDr Pas, Loc, Grm, Tac, Scr, Ste, For, Voc, Thr, Yaw, BSm, Pre, Dri, NRu, Atx, LSm, RSP, CSh, MSe, DSe, LDr lociated with a significant overall main effect o tests, P>0.05).	g multiple doses of the three monkeys). See Table 1 for of Increase RSP, MSe, DSe, Atx DSe, Atx	e test compounds. Sig fefinitions of abbreviation Decrease Pas, Tac Pas	nificant effects were identified as ns. Increase & Decrease Tac Tac	No increase or decrease in behavior were observed for PWZ-029.

#### **SIDE-EFFECTS Continued**

Experiment	Type of Test	Model		Figure		Summary of Results
Sedation Measures	Based on defined sedative postures for primates, sedative side-effects can be measured with these experiments.	Rhesus monkeys were treated with 3.0 mg/kg PWZ-029 (i.v.) or one of two concentrations of alprazolam and then monitored for 120- minutes post-dose.	Sleep Posture 3.0 mg/kg 15- 10- 5- 0- 0- 7.5-15-30-60-120 Sleep Posture 0.1 mg/kg 15- 10- 5- 0- 0- 7.5-15-30-60-120 120	Moderate Sedation 3.0 mg/kg	Deep Sedation 3.0 mg/kg	PWZ-029 did not alter either species-specific behavior or induce significant signs of sedation at doses 200- 300 times higher than the lowest dose that enhanced ORD performance. Conversely, the non- selective benzodiazepine alprazolam suppressed some aspects of normal, species-typical behavior and engendered rest/sleep posture, moderate and deep sedation; thus providing a positive control for the PWZ-029 data.

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## IMPAIRED BASELINE RECOVERY AND COMPLEX MEMORY

Experiment	Type of Test	Model	Figure	Summary of Results
Pavlovian Fear Conditioned Contexual Memory	Assesses the contexual fear response (i.e. freezing) to stimulus.	Scopolamine-treated C57BL/6 mice (n=12/group) - Each mouse received i.p. injections with either vehicle (0.9% saline containing 2.5% encapsin), 1.5 mg/kg scopolamine, or 1.5 mg/kg scopolamine plus 10 mg/kg PWZ- 029.	<sup>100</sup> p>0.05 50 min point of the scopolamine pwz-oze + scop	PWZ-029 attenuated the scopolamine-induced impairment of contextual memory [3]. *** <i>p</i> <0.05.

## PWZ-029 Preliminary Data Compilation IMPAIRED BASELINE RECOVERY AND COMPLEX MEMORY Continued

Experiment	Type of Test	Model	Figure	Summary of Results
Object Retrieval with Detours (ORD)	ORD is a manual test sensitive to frontostriatal deficits that investigates executive function.	Rhesus monkeys were treated with vehicle or 1 of 3 concentrations of PWZ-029 and then subjected to 15 mixed or difficult task configurations over 14- 21 experimental sessions. The percent success rate of object retrieval was measured.		PWZ-029 enhanced cognition under conditions in which baseline performance was impaired (i.e. difficult trials). *P<0.05 vs. vehicle (V)

## IMPAIRED BASELINE RECOVERY AND COMPLEX MEMORY Continued

Experiment	Type of Test	Model	Figure	Summary of Results
Object Retrieval with Detours (ORD)	ORD is a manual test sensitive to frontostriatal deficits that investigates executive function.	Rhesus monkeys were treated with vehicle, scopolamine, or 1 of 3 concentrations of PWZ- 029 and then subjected to 15 mixed or difficult task configurations over 14-21 experimental sessions. The percent success rate of object retrieval was measured.	100 75 50 25 ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓	Scopolamine-induced deficits in ORD performance were reduced with PWZ-029. *P<0.05 vs. Vehicle (V); **P<0.05 vs. scopolamine (S)
Delayed Non- Match to Sample Task (DNMS)	Short-term recognition memory task	Rhesus monkeys treated with vehicle or increasing concentrations of PWZ- 029 (n=2-3), and subjected to DNMS tasks with and without delay.	Delayed Non-Match to Sample Task	Performance was not affected by PWZ-029 without delay. A 0.1 mg/kg dose of PWZ- 029 significantly enhanced performance when a 1-minute delay was introduced. *P<0.05

## IMPAIRED BASELINE RECOVERY AND COMPLEX MEMORY Continued

Experiment	Type of Test	Model	Figure	Summary of Results
DNMS	Short-term recognition memory task	Rhesus monkeys treated with vehicle or 0.01 mg/kg of PWZ- 029 (n=4), and subjected to DNMS tasks without delay, as well as with a 10- minute delay and distractors.	A. Delayed Non-Match to Sample 10-Min Delay w/Distractors * 40 20 0 No Delay Veh PWZ	At 0.01 mg/kg, PWZ-029 significantly enhances recognition memory in the DNMS task when a 10-minute delay and distractors are introduced. *P<0.05
Object Retrieval with Detours	Theorized to assess Executive Function	Rhesus Monkeys treated with and without scopolamine and/or PWZ-029 (n=4).	B. Object Retrieval with Detours 100 80 60 40 20 0 Scopolamine + + PWZ-029 + +	PWZ-029 appeared to recover the scopolamine-induced deficit of retrieving the objects. *P<0.05 vs. no treatment (first bar)

PWZ-029 Preliminary Data Compilation IMPAIRED BASELINE RECOVERY AND COMPLEX MEMORY Continued

Experiment	Type of Test	Model	Figure	Summary of Results
Acquisition Phase of MWM	Assess spatial learning and working memory	Male, Dahl SS, n=10, red bars = 10 mg/kg PWZ-029	(i) 9.0 8.0 7.0 6.0 5.0 4.0 3.0 2.0 1.0 Vis 1 2 NW RW Day P<.05 vs. Vehicle	PWZ-029 significantly improves the ability of SS rats to acquire the location of the hidden platform after four days of acquisition training.
MWM	Assess spatial learning and working memory	Male, FAB-Samaritan rats; blue bars = positive control, red bars = negative control, green bars = donepezil, 3 mg/kg; turquoise bars = PWZ-029, 10 mg/kg	5.0 3.0 2.0 1.0 0.0 Day 1 Visible Day 2 Hidden Platform Phase	On the second week of the MWM protocol, PWZ-029 significantly improved performance in the experiment compared to the Group 2 (negative control).
MWM	Assess spatial learning and working memory	Male, FAB-Samaritan rats; blue bars = positive control, red bars = negative control, green bars = donepezil, 3 mg/kg; turquoise bars = PWZ-029, 10 mg/kg	Tay 1 Visible Day 2 Hidden Day 3 Hidden Platform Phase	On the second week of the MWM protocol, PWZ-029 significantly improved performance in the experiment compared to the Group 2 (negative control).

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- II. Synthesis of Alpha 5 Subtype Selective Ligands for GABA(A)/Benzodiazepine Receptors

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Minor: Physical Chemistry