A Systematic Study of Actinide Production from the Interactions of Heavy Ions with ²⁴⁸Cm

by

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ABSTRACT

Production cross sections for heavy actinides produced from the interactions of ¹²C, 31P, 40Ar, and 44Ca ions with 248Cm were measured at energies ranging from 0.98 to 1.35 X Coulomb barrier. The recoiling reaction products were collected in copper or gold catcher foils located near the ²⁴⁸Cm target. Separate fractions of Bk, Cf, Es, Fm. and Md were obtained from a radiochemical separation procedure. For the ¹²C system, a He/KCl jet was used to transport the recoiling No activities of interest to a rotating wheel system. The isotopic distributions of the actinide products were found to be essentially symmetric about the maximum with full-widths-at-half-maximum of approximately 2.5 mass units. Isotopic distributions of the ¹²C, ³¹P, ⁴⁰Ar, and ⁴⁴Ca systems were found to be very similar to the 40,48Ca systems studied previously. The maxima of the isotopic distributions generally occurred for those reaction channels which involved the exchange of the fewest number of nucleons between the target and projectile for which the calculated excitation energy was a positive quantity. Additionally, the maxima of the excitation functions occurred at those projectile energies which were consistent with the calculated reaction barriers based upon a binary reaction mechanism. The experimental data from the four systems investigated were compared to several models of heavy ion interactions including a damped reaction mechanism, compound nucleus formation and subsequent particle evaporation, and classical partial wave calculations for binary systems.

DEDICATION

I would like to dedicate this thesis to the world's best teacher, Dr. John Hoffman. It is because of this man's enthusiasm and excellent teaching ability that I decided to go into chemistry. I thank him for giving me an excellent foundation of fundamental skills, without which, I could not have achieved my dreams. I hope that he will continue to be a role model for many other students in years to come. Keep up the great work!

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(x,y) = (x,y) + (x,y

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I. INTRODUCTION

A. Heavy Ion Reactions

1. General Features

Nuclear reactions can generally be described as those interactions between two nuclei, a nucleus and an elementary particle, or a nucleus and a photon in which a change in the nucleus takes place [FRI81]. The results of such interactions are numerous and are an active area of research in the field of nuclear science. However, nuclear reactions between heavy ions (HI) are especially important due to their unique characteristics. In particular, heavy ions have large nuclear radii, high nuclear charges and angular momenta, and small de Broglie wavelengths (defined below) [SCO77]. Since the reduced de Broglie wavelength is small compared to the internuclear distances involved in these types of reactions, heavy ion reactions can be treated in a classical manner [EIS75] [GLE83].

Presently, there are only 270 known stable nuclides. Euring the last seventy years, approximately 1300 radioisotopes have been discovered [GEN89] [SCO77]. However, it has been predicted that 6000 different nuclides could be formed from the interactions of 2 GeV U + U ions [SCO77]. Hence, heavy ion reactions have and will be an important tool in uncovering the secrets of nuclear structure and stability.

2. Classical Mechanics: An Introduction

A schematic diagram of a heavy ion projectile impinging on a target nucleus is shown in FIG. I.1. This figure illustrates, among other things, the impact parameter, b_m , and the scattering angle, θ .

The Coulomb barrier between two nuclei is the center-of-mass energy required to bring the two from infinite separation to a touching sphere configuration, and is given by

$$V_c = \frac{Z_1 Z_2 e^2}{R_1 + R_2},$$
 (1)

where Z_1 and Z_2 are the atomic numbers of the two nuclei, e is the elementary charge, and R_1 and R_2 are the respective nuclear radii. The distance of closest approach, d, for a head-on collision between these two nuclei is then

$$d = \frac{Z_1 Z_2 e^2}{E_{cm}},\tag{2}$$

where E_{Cm} is the center of mass kinetic energy. The reduced de Broglie wavelength, λ , for a nucleus is

$$\lambda = \underline{\qquad},$$
(3)

where π is $h/2\pi$, h is Planck's constant, m is the mass of the heavy ion, and v is the relative velocity of the ion. Hence, the Sommerfeld parameter, η , is used for these situations and is defined as

$$\eta = \frac{Z_1 Z_2 e^2}{\pi v} \tag{4}$$

This parameter is one half the ratio of the distance of closest approach to the reduced de Broglie wavelength. Once again, for the classical domain to be applicable, $\eta >> 1$. The Sommerfeld parameters have been calculated for ¹²C, ³¹P, ⁴⁰Ar, and ⁴⁴Ca with ²⁴⁸Cm and are shown in Table I.1. These calculations were done at the highest bombarding energy for each projectile.

For large η , the angular distributions for the projectile-like fragments have their maxima near the grazing angle [GAL76]. The grazing angle in the center-of-mass system, θ_{gr} , can be calculated by

$$\frac{\theta_{gr}}{2} = \frac{\varepsilon_C}{(2\varepsilon - \varepsilon_C)}, \qquad (5)$$

where $\varepsilon_C = V_C/\mu$, μ is the reduced mass of the system, and ε is the relative kinetic energy [MCF82] [SCO77]. Since the grazing angle can only be calculated at energies above the Coulomb barrier, the quarter-point angle from the Fresnel scattering model [FRA78] [SCO77] should be used for near- and below-Coulomb barrier interactions. The quarter-point angle, $\theta_{1/4}$, in the center-of-mass system is

$$\theta_{1/4} = 2\sin^{-1} \frac{\eta}{kR_{exp} - \eta}$$
(6)

where η is the Sommerfeld parameter, k is the asymptotic wave number, and R_{exp} is the experimentally determined interaction radius [WIL80]. The Fresnel model assumes there is a well-defined interaction radius, R_{exp} , for each target-projectile combination, that separates elastic scattering events and nuclear reactions in configuration space.

Using the sharp cutoff approximation, assuming a grazing collision with no mass transfer, and a loss of kinetic energy of 10 MeV, the scattering angles of the projectile-like and target-like products were calculated. The velocity of the center-of-mass of a dinuclear system in the laboratory frame of reference is

where E_p is the lab frame kinetic energy of the projectile, m_p is the mass of the projectile, and m_t is the mass of the target [TÜR89]. The center-of-mass velocity of the target-like fragment is given by

$$V_t = 1.389 - \frac{(TKEm_p)^{1/2}}{[m_t(m_t + m_p)]^{1/2}}$$
 (8)

where *TKE* is the total kinetic energy of the fragments and was assumed to be 10 MeV less than the center-of-mass energy of the projectile. The center-of-mass velocity of the projectile-like fragment is

$$V_p = 1.389 \frac{(TKEm_t)^{1/2}}{[m_p(m_t + m_p)]^{1/2}}$$
 cm/ns. (9)

The scattering angle of the projectile-like fragment in the laboratory system is then

$$\theta_{proj} = \arctan \frac{V_p \sin(\theta_{cm})}{V_p \cos(\theta_{cm}) + V_s}$$
(10)

where θ_{cm} is the angle between the nuclei in the entrance channel and exit channel. The scattering angle of the target-like fragment in the laboratory system is given by

$$\theta_{proj} = \arctan \frac{V_t \sin(\theta_{cm})}{V_t \cos(180^\circ - \theta_{cm}) + V_s}$$
(11)

Table I.1 lists θ_{proj} and θ_{targ} for the four systems investigated.

The particular type of reaction that results from an interaction between two heavy nuclei depends upon the impact parameter of the projectile, the kinetic energy of the projectile, and the masses of both the target and projectile [FRI81]. Since both nuclei are positively charged, the Coulomb barrier must be surmounted before the two will actually come in contact with each other. This can be achieved by giving the projectile sufficient kinetic energy.

3. Elastic and Inelastic Scattering

At large impact parameters where the separation distance R_{sep} is greater than the interaction radius R_{int} ($R_{int} = R_{proj} + R_{target}$), a process known as elastic scattering takes place. In this process the total kinetic energy of the system (projectile and target) is conserved. Heavy ions provide valuable information about interaction radii via elastic scattering. Another important type of process occurring at large impact parameters is termed inelastic scattering. In this process, the target nucleus is left in an excited state at the expense of the kinetic energy of the projectile. Extremely high-spin states can be induced in target nuclei by heavy ion projectiles through this type of process.

4. Transfer Reactions

When R_{sep} is approximately equal to R_{int} , the overlap of nuclear matter takes place, exchange of nucleons can occur, and the angular distributions of the projectile-like products will peak near the grazing angle. During this process, the projectile approaches the target, the two come into contact forming an intermediate complex with a slight overlap between the two nuclei. After exchanging energy, angular momentum, and nucleons, the intermediate then quickly separates into two new nuclei, a projectile-like fragment and a target-like fragment. Due to the binary nature and large impact

parameters of transfer reactions, the projectile-like fragment can carry off a large fraction of the kinetic energy and angular momentum, thus leaving the target-like product with little excitation energy. The contact time is on the order of 10^{-22} seconds [TÜR89]. Since the interaction times for these types of reactions are so short, little equilibration occurs; hence the neutron-to-proton ratio, angular momentum, and kinetic energy remain essentially unequilibrated.

5. Deeply Inelastic Reactions

If $R_{Sep} < R_{int}$, significant overlap of nuclear matter occurs, large numbers of nucleons, excitation energy, and angular momentum can be exchanged between the two nuclei, and the type of interaction is termed deeply inelastic. The intermediate complex formed between the target and projectile will rotate because of the amount of angular momentum introduced by the projectile. A large fraction of the projectile kinetic energy is transformed into internal excitation so that when the intermediate separates, two new products with a great deal of excitation energy are formed. The interaction times for deeply inelastic reactions are on the order of 10^{-21} s, sufficiently long to enable the neutron-to-proton ratio to equilibrate. There is heavy damping of the kinetic energy and some equilibration of orbital angular momentum [DOS85] [GRE85] [KRA77] [MOR81] [NÖR75] [RIE79] [SCH77].

6. Complete Fusion Reactions

At even smaller impact parameters, a projectile nucleus can totally amalgamate with a target nucleus resulting in a compound nucleus. During these reactions the identities of the target and projectile are lost and equilibration of the neutron-to-proton ratio, kinetic energy, and angular momentum is achieved. In some instances, however, p eequilibrium emission of particles can occur.

B. A Brief History of Transfer Reactions

Multinucleon transfer reactions were first investigated in detail by Kaufmann and Wolfgang in the late 50's and early 60's [KAU59] [KAU60]. Studying interactions of

¹⁶O projectiles with Al, Cu, and Sn targets, they found that the (p2n) and (2p3n) cross sections were larger than expected in comparison to single nucleon transfer reactions. These results contradicted those expected from transfer reactions involving a Coulomb barrier tunneling mechanism, leading them to propose the "contact transfer model." This model involved the interactions between a target and projectile at distances approximately equal to the radius of the projectile. Under these conditions a dumbbell-shaped intermediate could be formed, allowing nucleons to be exchanged. Because the Coulomb and centripetal forces were greater than the nuclear binding force, the intermediate would break up into a projectile-like fragment and target-like fragment before it had rotated one half of a revolution [KAU60].

Hahn et al. [HAH74] produced 245,244 Cf via the 239 Pu(12 C, αxn), 235 U(16 O, αxn), 237 Np(14 N, αxn), 242 Pu(12 C, $\alpha (x+3)n$) and 238 U(12 C, ^{5}n -6n) reactions. Based upon these data they made the following observations:

- 1. The probability of transferring a cluster of nucleons decreases as the size of the cluster increases. Viewing these reactions as the division of the projectile into an alpha particle and a residual cluster, they argued that it was logical that the alpha cluster nuclei such as ¹²C and ¹⁶O have larger yields than ¹⁴N.
- 2. The cross sections for the $(^{12}C,\alpha 2n)$ and $(^{12}C,\alpha 3n)$ channels were larger than the $(^{12}C,\alpha 5n)$ and $(^{12}C,\alpha 6n)$ channels. This signified to the authors that fission was effectively competing with the evaporation of more than two or three neutrons.

Focusing on the ²³⁸U(¹²C,5*n*-6*n*) and ²³⁹Pu(¹²C,α2*n*-α3*n*) channels, Hahn *et al.* performed a detailed comparison in which they measured excitation functions, recoil ranges, and angular distributions of the products. The data from the Cf isotopes produced by the ¹²C-²³⁸U system were consistent with a compound nucleus mechanism while the data from the same Cf isotopes produced in the ¹²C-²³⁹Pu system were consistent with a transfer mechanism. The compound nucleus products displayed forward-peaked angular distributions in the laboratory frame of reference while the

transfer reaction products had angular distributions that peaked at $\cong 17^{\circ}$ (at energies of 1.06 and 1.17 X Coulomb barrier). Based upon these results, Hahn *et al.* made use of a semi-classical, two step transfer reaction model involving the transfer of a cluster of nucleons in the first step and the subsequent Rutherford scattering of the projectile-like fragment in the second step.

Eskola *et al.* [ESK70] [ESK71] [ESK73] [ESK75] produced a variety of Es, Md, No, and Lr isotopes via (HI,xn), ($HI,\alpha xn$), and ($HI,2\alpha xn$) reactions on actinide targets. They found the production cross sections for the ($^{12}C,\alpha 2n$) and ($^{12}C,\alpha 3n$) channels were higher than the cross sections for the ($^{12}C,4n$) and ($^{12}C,5n$) channels. They viewed the ($^{12}C,\alpha 2n$) channel as a carbon projectile breaking up into a 8 Be nucleus and an α particle. In addition, the ($^{12}C,\alpha xn$) channels had higher production cross sections than the ($^{13}C,\alpha xn$) reaction channels. This artifact was attributed to α clustering in the ^{12}C nucleus. Eskola [ESK75] also found that cross sections for products with $Z_{target} < Z_{product} < Z_{target+projectile}$ were larger than the cross sections for compound nucleus products.

The fact that transfer reactions can produce target-like fragments with low excitation energy becomes advantageous in certain areas of the periodic table. Specifically, the production of "cold" target-like products becomes important in the actinide region where the fission barriers are on the order of 5-6 MeV [BRI79] [BRI80]. With low excitation energies, actinide products of transfer reactions have a greatly reduced probability of fission and/or particle emission during their deexcitation [HOF86]. The bombardment of actinide targets with heavy ion projectiles can effectively be used to produce a great variety of actinide products via transfer reactions. In particular, by choosing neutron-rich projectiles and/or targets, neutron-rich actinide products can be produced with easily detectable cross sections. By producing new, neutron-rich actinides, the so called "island of stability" or "superheavy island" (see FIG I.2) can possibly be reached. It is postulated that this island exists at the next closed proton and neutron shells of 114 and 184,

respectively [HER79].

C. Transfer Reaction Studies Using ²⁴⁸Cm as a Target

Because of the availability of relatively large quantities of 248 Cm, this nuclide has become a popular target material for many different transfer reaction studies. 248 Cm is an attractive target material for many reasons including its high neutron-to-proton ratio (N/Z = 1.583), long half-life ($t_{1/2} = 3.40 \times 10^5 \text{ y}$), and its availability in nearly isotopically pure milligram quantities [HOF85].

Numerous actinide production studies using ²⁴⁸Cm targets with various projectiles, ranging from ¹³C to ²³⁸U, have been performed in order to gain some insight into the reaction mechanism(s) involved [ESK75] [GÄG86] [HOF85A] [LEE82] [LEE83] [MOO86] [SCH82] [TÜR89] [WEL87].

In an attempt to study the influence of an additional pair of neutrons in the projectile on the actinide production cross sections, Lee *et al.* [LEE82] bombarded 248 Cm with the projectile pairs 16,18 O and 20,22 Ne. It was found that as the number of neutrons in the projectile was increased by two, the corresponding maxima of the mass-yield curves (plots of cross section vs. mass number for a given Z) for the above target elements ($Z_{product} > Z_{target}$) were increased by about two mass units. Hence, the neutron-rich projectile of each pair enhanced the neutron-rich actinide production.

Experiments using the projectiles ^{40,48}Ca on ²⁴⁸Cm targets were performed to see what the effect of the eight neutron excess of ⁴⁸Ca relative to ⁴⁰Ca would have on the final actinide production distribution [HOF85A]. Shifts to heavier masses of only 2-3 units in the mass-yield curves for the elements Bk, Cf, Es, and Fm were observed in the ⁴⁸Ca system relative to the ⁴⁰Ca system. The eight neutron excess of ⁴⁸Ca relative to ⁴⁰Ca did not increase the maxima of the isotopic distributions by eight mass units and hence had only a partial effect on the final isotopic distributions of the above target products.

Welch et al. [WEL85] [WEL87] observed in the ¹²⁹,132,136Xe-²⁴⁸Cm systems that

the peaks in the isotopic distributions for Fm, Bk, and Pu shifted to higher mass numbers as the projectile mass number increased. Shifts in the Cf and Es peaks could not be verified because production cross sections on the neutron-deficient side of the mass yield curves were not measured. However, production cross sections for the most neutron-rich above target nuclides were found to be independent of the projectile mass. Potential energy surfaces comparing the energies of the system of two touching spheres before and after the transfer of neutrons and protons were calculated and compared with the reaction cross sections. The potential energy surfaces were unable to successfully predict the maxima of the mass-yield curves, suggesting to the authors that deeply inelastic reactions may not be the dominant reaction mechanism.

Schädel et al. [SCH78] [SCH82] studied the ^{238}U - ^{248}Cm system in addition to the ^{238}U - ^{238}U system. Upon comparing various heavy ion- ^{248}Cm systems, they noted shifts in the centroids of the isotopic distributions of the actinide products. However, much to their surprise, they noted that the "nearby" ($Z_{product} \cong Z_{target}$) actinides produced from the ^{248}Cm targets interacting with $^{16,18}\text{O}$, $^{20,22}\text{Ne}$, ^{48}Ca , ^{136}Xe , and ^{238}U had peak yields that were all the same order of magnitude.

Since transfer reactions involving actinide targets provide a potential means of producing new neutron-rich actinides, a systematic study of these processes was continued as a primary goal of this thesis. Since ²⁴³Cm had been used as a target in many previous transfer reaction studies, it seemed logical to extend the systematic study of transfer reactions using this nuclide as the target material. With the addition of data from new target-projectile combinations, it was hoped that the present understanding of transfer reactions could be broadened and theoretical models developed which could accurately model the reaction mechanism involved, and provide predictions for unmeasured systems.

D. Rationale Behind Projectile Selection

The first projectile chosen for this study, ⁴⁴Ca, was used in order to further investigate

the effect of the neutron number of the projectile on the final isotopic distributions in transfer reactions involving ²⁴⁸Cm targets. Since previous studies had used the ⁴⁰Ca-²⁴⁸Cm and ⁴⁸Ca-²⁴⁸Cm systems, ⁴⁴Ca was an ideal candidate. ⁴⁴Ca has a neutron number of 24, which is between the neutron numbers of 20 and 24, respectively of the doubly magic nuclei ⁴⁶Ca and ⁴⁸Ca. Since both ⁴⁰Ca and ⁴⁸Ca are doubly magic, ⁴⁴Ca could be used to study the influence of neutron shell effects on the final isotopic distributions of the actinide products. This stable nuclide has a natural abundance of 2.086 atom % [WAL84] and 600 mg was obtained from Oak Ridge National Laboratory as a separated isotope (98.6 atom %) in the metallic form. In addition, reasonably intense beams ($\cong 100$ PnA) could be obtained from the 88-Inch Cyclotron at Lawrence Berkeley Laboratory.

The 40 Ar- 248 Cm system was studied next because the neutron-to-proton ratio of 40 Ar (1.22) is quite similar to the 1.20 ratio of 44 Ca. In addition, 40 Ar has the same mass number as the doubly magic 40 Ca. Since 40 Ar is a stable isotope of high natural abundance (99.600 atom %) [WAL84], it is readily available. It is also easily accelerated with high intensity by the 88-Inch Cyclotron.

31P was chosen as a projectile for two reasons. First of all it has an odd number of protons, and until this study, no detailed transfer reaction studies in the actinide region had been performed using projectiles with odd Z. The odd Z projectile was used in order to see if any projectile pairing effects could be observed in the final isotopic distributions. Secondly, ³¹P has a neutron-to-proton ratio of 1.07, similar to the ratios of other projectiles used in previous studies. Since ³¹P has a natural abundance of 100 atom % [WAL84], it was rather easily obtainable. Red phosphorus was used in the ion source at the 88-Inch Cyclotron.

The final projectile used in this study was 12 C. This projectile was chosen because it has a neutron-to-proton ratio of 1.00, exactly the same as 40 Ca. Since the lightest projectile used in a detailed transfer reaction study on 248 Cm was 16 O, the extension of

this study down to an even lighter element was desirable to see if any mechanistic changes occur. ^{12}C can be thought of as simply a cluster of three α particles. Hence, the $(^{12}\text{C},\alpha xn)$ and $(^{12}\text{C},2\alpha xn)$ reaction channels might be enhanced. Additionally, ^{12}C is easily obtainable in gaseous forms such as CH₄ or CO₂ which make ideal sources for the 88-Inch Cyclotron ECR ion source.

Table I.1. Calculated Sommerfeld parameter, η , for ^{12}C , ^{31}P , ^{40}Ar , and ^{44}Ca with ^{248}Cm .

Projectile	V _C (MeV)	Bombardir Energy (MeV)	ng η	θ_{proj} (degrees)	θ _{targ} (degrees)
12 _C	68.6	93	32	66	53
31 _P	156.1	21	282	64	54
40 _{Ar}	181.9	246	102	64	53
⁴⁴ Ca	199.8	271	112	63	53

Figure Captions

Figure I.1. Schematic diagram of a heavy ion projectile (A_1, Z_1) impinging upon a stationary target nucleus (A_2, Z_2) . b_m is the impact parameter, θ is the scattering angle, R_1 is the projectile nucleus radius, and R_2 is the target nucleus radius.

Figure I.2. Schematic representation of the chart of the nuclides taken from [GRE89].

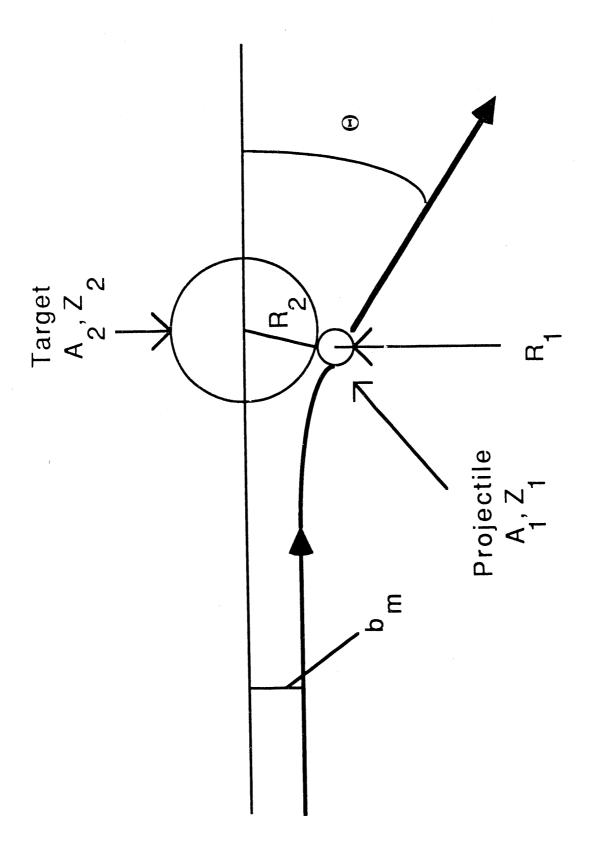


Fig. I.1

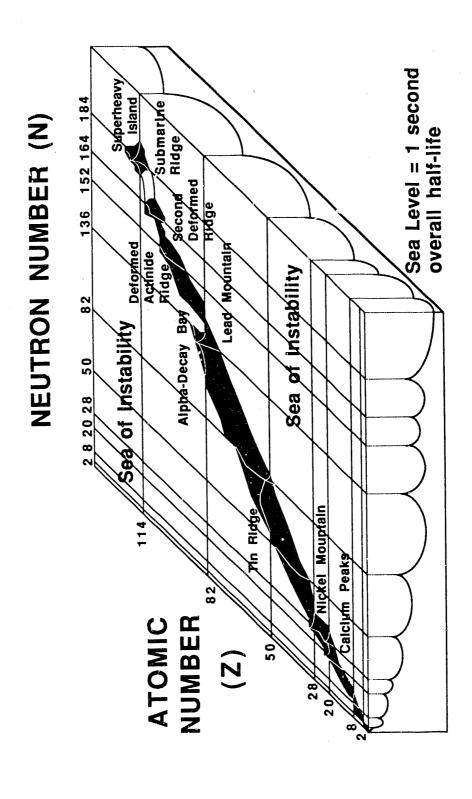


Fig. I.2

II. EXPERIMENTAL

A. Target Requirements

Very strict guidelines are required of targets used in heavy ion bombardments at particle accelerators [HOF88]. When measuring reaction cross sections, the target must be thin and uniform so the energy of the bombarding particles is well defined upon passage through the target material. In addition, the target material must also be thin enough so that the recoiling target-like fragments can escape from the target material if a catcher foil technique is used.

The passage of a heavy ion beam through an actinide target produces a large amount of heat and extreme ionizing conditions. The target and its backing must therefore be able to withstand such conditions. For these reasons Cm_2O_3 (m.p. 2260 \pm 25 °C) [BAY73] was chosen as the target material and Be metal (m.p. 1278 \pm 5 °C) [CRC75] as the target backing.

The 248Cm targets (97.44% 248Cm, 2.53% 246Cm, 0.025% 245Cm, 0.010% 247Cm, and 1 X 10-4% 244Cm) were prepared by the electrodeposition of Cm in a 0.6 cm diameter on a Be foil in stepwise fashion from an isopropanol solution [GRE85]. After depositing approximately 70 μg/cm² of ²⁴⁸Cm, the Be was placed in an oven and heated to 500 °C to convert the Cm to Cm₂O₃. To determine its thickness, the target was assayed at low geometry with an α-spectrometer system utilizing a Si(Au) solid state surface barrier detector. The electrodeposition, heating, and counting were repeated until the desired target thickness was reached. Table II.1 lists the thicknesses of the targets used in the various experiments of this study.

B. Target System

All of the irradiations of the ²⁴⁸Cm targets were performed at the Lawrence Berkeley Laboratory 88-Inch Cyclotron. Because a radioactive target was used, precautions had to be taken to insure that the 88-Inch Cyclotron would not become contaminated in the

event of a vacuum failure. For this reason the "slammer valve" was used. Should a vacuum problem arise, the "slammer valve" would simply seal the beam line between the target system and the cyclotron, effectively isolating the cyclotron from possible contamination with ²⁴⁸Cm. The positive ion beams, after exiting the cyclotron passed through a "beam wobbler". This device is simply the stator of an induction motor used to move the beam in circles over the target area in order to spread the beam and prevent any local overheating in the target [MOO83]. From the wobbler, the positive ion beams passed through a water-cooled graphite collimator (inner diameter 6 mm), a 1.8 mg/cm² HAVAR isolation foil, 0.2 mg/cm² nitrogen cooling gas, the 2.75 mg/cm² Be target backing, and finally the Cm₂O₃ target material. The reaction products recoiling out of the target were stopped and collected in 1.6 cm diameter copper or gold catcher foils located approximately 0.4 cm from the target. The compositions and thicknesses of the catcher foils used in these experiments are listed in Table II.1. The accelerated ions that did not interact with the target material were stopped by a water-cooled graphite beam The total electrical current deposited in the target system by the beam was measured by an integrating electrometer. The integrated beam current was recorded at regular intervals throughout all of the bombardments so that accurate beam histories could be used in the calculation of production cross sections. A schematic diagram of the target system at the 88- Inch Cyclotron is shown in Fig. II.1.

C. Catcher Foil Collection Efficiency

Since both the target and catcher foil have finite dimensions, the probability of a recoiling reaction product hitting the catcher foil must be considered. This problem has been treated in detail by Moody and only a brief summary will be presented here [MOO83]. If one has a circular catcher foil of radius a at distance b from a circular target of radius R, the probability of a recoiling reaction product missing the catcher foil may easily be calculated assuming the target and catcher foil are parallel (see Fig. II.2). At any point r from the axis connecting the centers of the target and catcher foil, reaction

products are emitted at a variety of angles θ with respect to the beam axis. However, if the reaction products are emitted at angles smaller than certain angle θ_{min} , all such products are collected regardless of their orientation about the axis. Likewise, if the recoiling reaction products are emitted at angles larger than angle θ_{max} , all such products will miss the catcher foil, regardless of their orientation. If, however, the recoiling reaction products are emitted at angles between θ_{min} and θ_{max} , a certain fraction of the products will hit the catcher foil, depending on the azimuthal angle at which the products are emitted. The probability of a recoiling reaction product missing the catcher foil when it is emitted at a distance r from the axis at an angle θ (where $\theta_{min} < \theta < \theta_{max}$) is given by

$$P_{\theta,r} = -\sin^{-1} \frac{b^2 \tan^2 \theta - a^2 + r^2}{\pi} + -\frac{\pi}{2rb \tan \theta}$$
 (1)

By evaluating this expression over the entire area of the target and over all azimuthal angles, the total collection efficiency of the catcher foil can be obtained. A computer code was written (see Appendix A) to calculate the probability of a recoiling reaction product hitting the catcher foil as a function of the angle at which the recoiling nucleus was emitted (in the lab frame) [LEY90B]. The results of this calculation for the target system used in this study are plotted in Fig. II.3. From Fig. II.3 it can be seen that the probability of a recoiling product nucleus hitting the catcher foil is 1 if it is emitted between 0° and 51° and then decreases monotonically to 0 between 51° and 71°.

The laboratory system scattering angles for the target-like fragments, θ_{targ} , which are listed in Table II.1, indicate that the largest angles at which the target-like fragments recoil is 53-54°. At 54°, the probability of a recoiling nucleus hitting the catcher foil is 0.95. Hence, even at the highest bombarding energies, losses due to the recoiling atoms missing the catcher foil are small.

The maximum recoil energy possible for a binary-type interaction, assuming that the reaction Q value and the amount of mass transferred is small, is for a completely elastic collision at an impact parameter of zero. For 319-Mev ⁴⁴Ca ions, the recoiling energy of a target-like product is about 140 MeV. The range of 140 MeV ²⁴⁸Bk ions in Au is 12 mg/cm² [NOR70], much thinner than the 19.6 mg/cm² Au catcher foils used. Maximum recoil energies for target-like products have been calculated for the other systems in the present study and are given in Table II.2 along with their associated ranges in the catcher foil material and the thicknesses of the catcher foils used. In all cases the catcher foils were sufficiently thick to stop the recoiling reaction products, assuming the reactions products hit the catcher foil, especially since most products will recoil with significantly less than the maximum possible recoil energy.

In many instances, however, the Q value and/or the mass transfer for a particular reaction channel are significant and cannot be ignored, especially in the case of the ¹²C system. Fortunately recoil range data exist on the actinide products from the ¹⁸O-245Cm and 18O-249Cf systems [MCF82], 246Cf and 252Fm produced from the 18O-²⁴⁵Cm reaction at 1.18 X Coulomb barrier, were both found to have an average range of approximately 0.25 mg/cm² in Al. Based on range and stopping-power tables [NOR70], this corresponds to a range of roughly 0.65 mg/cm² in Au, much thinner than the 2.47 mg/cm² Au catcher foils actually used in the ¹²C-²⁴⁸Cm system. However, some tailing of the ²⁴⁶Cf and ²⁵²Fm occurred up to 0.7 mg/cm² in Al, corresponding to about 2 mg/cm² in Au. Once again though, the Au catcher foils should be sufficiently thick to stop these recoiling target-like fragments. From the range of ²⁵²Fm nuclei in the Al, the average recoil energy is about 5 MeV, and the maximum recoil energy is approximately 15 MeV. As Moody [MOO83] points out, there is measurable activity in the first foils from the stack experiments of McFarland [MCF82]; hence, there is a probability that a few percent of the recoiling reactions products were stopped in the target material. In fact, Moody attributes discrepancies between Bk cross sections reported for two identical experiments he performed (except for target thickness) as "probably partly due recoil losses in the thicker target" [MOO83].

Considering the experiments where ²⁴⁸Cm was bombarded with 93 MeV ¹²C ions, if compound nucleus formation takes place, then the recoiling ²⁶⁰No compound nucleus products should have about 4.3 MeV of kinetic energy assuming no kinetic energy is converted to excitation energy. The range of 4.3 MeV ²⁶⁰No ions in the target material is about 0.6 mg/cm² [NOR70], much greater than the 0.182 mg/cm² thickness of the ²⁴⁸Cm target. Likewise, the range of 4.3 MeV ²⁶⁰No ions in Au is about 0.5 mg/cm²; hence, the ^{2.47} mg/cm² Au catcher foils used in this set of experiments are more than adequate to stop the recoiling compound nucleus products.

At the other extreme, consider the case of compound nucleus formation from 71 MeV ¹²C ions interacting with ²⁴⁸Cm. The recoiling ²⁶⁰No ions should have about 3.3 MeV of kinetic energy, again assuming no excitation energy. The range of 3.3 MeV ²⁶⁰No ions in the target is about 0.4 mg/cm², once again greater than the 0.182 mg/cm² target thickness. The range of 3.3 MeV ²⁶⁰No ions in the Au catcher foil material is approximately 0.4 mg/cm², compared to the 2.47 mg/cm² thickness of the Au catcher foil. Similar arguments can be made for the ³¹P, ⁴⁰Ar, and ⁴⁴Ca systems.

Türler considered the scenario at higher bombarding energies where the target-like reaction products are emitted at angles near the limits of the target arrangement [TÜR89]. The center-of-mass scattering angle θ_{cm} is smaller than the calculated grazing angle for the projectile-like fragment if the projectile and target stick together and rotate for a finite length of time. Similarly for the target-like reaction products, recoils at angles larger than the grazing angle may be possible. Considering the $^{44}\text{Ca-}^{248}\text{Cm}$ system at projectile energies of 230.8-236.5 MeV, Türler found losses of projectile-like products due to absorption in the target material at scattering angles of 180° - $\theta_{cm} \ge 140^{\circ}$ (This corresponds to a scattering angle in the laboratory frame of reference of $\Omega_{lab} = 68^{\circ}$).

From the above arguments, it can be seen that even for the most extreme case studied here (^{31}P at 7 MeV/u), the probability of a recoiling target-like product hitting the catcher foil is 0.95. Therefore these losses can be ignored especially at the lower bombarding energies. Additionally, it has been shown that the catcher foils were sufficiently thick to catch those products with the maximum possible recoil energy. It will be assumed in all of the following calculations that 100% of the recoiling reaction products are caught by the catcher foils.

D. Experimental Conditions

The Experimental conditions for each irradiation are shown in Table II.3. All of the beam energies have been corrected for the appropriate energy loss in the target system and represent the average energy of the ions in the ²⁴⁸Cm target material in the laboratory frame of reference. The energy loss of the beam in the target was calculated using range and stopping power tables for heavy ions [HUB80] [NOR70]. Several values had to be estimated due to insufficient data in the range and stopping power tables. The stopping powers of oxygen and nitrogen were interpolated from the stopping powers of adjacent elements, while uranium was used as an approximation for curium. The stopping power for the complex alloy HAVAR was approximated with Ni. The composition of HAVAR is shown in Table II.4 [GRE85].

E. Manual Chemical Separations

Heavy actinide transfer reaction products constitute only a small fraction of the nuclides produced from the interactions of heavy ions with actinide targets. Typical actinide production cross sections are on the order of microbarns whereas the total reaction cross section is approximately one barn. Therefore, chemical separation procedures must be employed in order to isolate the actinide products of interest. The chemical separations, however, must meet several criteria. First of all they must effectively and cleanly separate the actinides from other interfering activities such as the lanthanides. Secondly, the actinides must be separated from each other. Separate

actinide fractions allow Z identification and aid in the determination of independent production yields of the actinides of interest. Separate elemental fractions are also vital when two nuclides with similar $t_{1/2}$'s emit gamma rays with the same energy (i.e. $^{250}\text{Es}m$ $t_{1/2}$ = 2.2 h and ^{250}Bk $t_{1/2}$ = 3.22 h) [KEG85]. In addition, the chemical separation procedures should be efficient and relatively fast in order to isolate nuclides with half lives on the order of 30 to 45 minutes.

Ion exchange chromatography has become one of the most important tools for the separation of the lanthanides and actinides [STE61]. The resins used in ion exchange chromatography are simply three-dimensional structures with numerous ionizable groups affixed. The most common resins are polystyrenes cross-linked with divinylbenzene [DYE61]. The degree of cross-linking is directly proportional to the porosity of the resin. High porosity is desired since this feature will determine the number of available sites for ion exchange to take place and will therefore increase the selectivity of the resin [STE61]. In addition, greater cross-linking decreases the elasticity of the three dimensional matrix and hence the swelling that takes place is reduced [MAS71].

1. Anion Exchange Chromatography

Anion exchange resins exchange anionic species between the stationary resin phase and the mobile liquid phase. The active sites in the resin are typically quaternary amines. The elution position, or chromatographic retention of an element is dependent upon the competition for the element between the stationary resin phase and the complex in the mobile liquid phase. Ions with different charge densities form complexes of different stabilities. Hence the elution positions of different species vary and an effective separation can be performed [COO52].

2. Cation Exchange Chromatography

Cation exchange resins exchange cationic species between the resin phase and the mobile liquid phase according to

$$A^{+} + R^{-}B^{+} = B^{+} + R^{-}A^{+},$$
 (2)

where A^+ is the cationic species of interest, B^+ is mobile phase ion and, R^- is the ionic exchange site [BAU78]. Typical cation exchange resins consist of cross-linked polystyrene with sulfonic acid groups (-SO₃H) [FRI81]. Separations usually involve the adsorption of ions from a solution on the resin followed by the selective desorption of the elements of interest from the column with different solutions [STE61].

α-Hydroxyisobutyrate Separations

The elements Am through Lr, like the analogous lanthanides (Eu through Lu), all exist as +3 ions in aqueous solutions except for No which exists as a +2 ion. This similarity presents a problem in the separation of the heavier actinides. They can, however, be separated on the basis of their trivalent ionic radii. It is well known that the ionic radii of the actinides decrease in going from Ac+3 (0.118 nm) to Lr+3 (0.088 nm). The ionic radii of the +3 actinides are listed in Table II.5 [BRA86].

A very common method for the separation of the lanthanides and transcurium elements from each other, based upon ionic radii, is the elution of the trivalent ions from a cation exchange resin with a complexing agent. Most commonly, organic acid anions such as citrate, glycolate, lactate, and α -hydroxyisobutyrate (α -HIB) are used [HIG60]. However, ammonium α -hydroxyisobutyrate typically gives better separation factors and has become one of the most widely used complexing agents [CHO56] [MAS71] [SMI56].

It has been shown that when α -HIB is present in concentrations of 0.1 M or greater, it will form -1 charged species with +3 ions of the form,

$$4L^{-} + M^{+3} = [ML_4]^{-},$$
 (3)

where L represents α -hydroxyisobutyrate and M is the metal ion of interest [CHO61]. Complexing agents such as α -hydroxyisobutyrate stabilize the cation in the mobile phase. Cations with higher charge densities will therefore spend a greater amount of time in the mobile phase when compared to cations with smaller charge densities. Hence, the cations with the highest charge densities will be eluted first. Since the charge

density of the actinides increases as the atomic number increases, the actinide elements are eluted from the cation exchange resin in order of decreasing atomic number.

3. Reverse Phase HDEHP Column Chromatography

a. Separation of the Actinides from the Lanthanides

The lanthanides, produced primarily by fission, are major products of heavy ion reactions with actinide targets. Because they are produced in large quantities, if they are present in the actinide fractions, their activities would mask the actinide activities. In order to collect radiochemically pure samples so that clean gamma spectra could be obtained, it was necessary to separate the actinides from the lanthanides. The separation of actinides from lanthanides is not trivial due to the similarities in chemical properties and ionic radii. The actinides, however, form stronger chloride and thiocyanate complexes than the corresponding lanthanides [HIG60]. Based on this property, an effective separation of actinides from lanthanides can be achieved on a cation exchange resin column using 13 M HCl as the eluent. Since the actinides form stronger neutral or anionic chloride complexes, they are more stable in the mobile phase and elute from the column more quickly than the lanthanides.

b. Elution of Actinide Fractions Containing Short-Lived Nuclides Via Reverse

Phase HDEHP Column Chromatography

The α -HIB chemical separation procedure can be too time consuming for the quick elution of elemental fractions containing short-lived nuclides such as 251 Bk ($t_{1/2}$ = 57 m). If one waits for the isolation of the Bk fraction on the α -hydroxyisobutyrate column, which is generally the last step in the chemical separation scheme, 3 or 4 hours could elapse, resulting in very little remaining 251 Bk activity for gamma-ray counting. An alternative separation technique, utilizing reverse phase column chromatography can be used to produce a chemical fraction containing Bk quickly. Reverse phase column chromatography utilizes a polar mobile phase and a nonpolar stationary phase such as silica particles with hydrocarbon chains attached to the surface [DOR89]. Hence, the

transplutonium elements can be effectively and quickly separated by a column of bis(2-ethylhexyl) orthophosphoric acid (HDEHP) adsorbed on hydrophobic diatomaceous earth (Celite 545) as the stationary phase with mineral acids as the mobile phase [HOR69] [HOR69A] [HOR72] [WIN63]. With this technique, the transplutonium elements are eluted in order of increasing Z, resulting in a separation of the Bk fraction in about one hour.

4. Recoil Atom Collection Foils

a. Chemical Procedure for Copper Catcher Foils

After each irradiation, the copper catcher foil was removed and taken to a chemistry laboratory. The foil was dissolved in 1 ml of a solution of 15.8 M HNO₃ with 50 µl of 70% HClO₄. Appropriate yield tracers such as ²⁴⁹Cf were added in order to monitor the chemical yield of the actinides throughout the separation procedure. After dissolution, the solution was evaporated to dryness until NO2 fumes were evolved. The resulting residue was dissolved in 1 ml of 10 M HNO3 and cooled to room temperature. When the solution reached room temperature it was sorbed onto a 3-mm diameter X 60-mm long column of Bio Rad AG-1, X-8 (200-400 mesh) anion exchange resin which had been pretreated with 10 M HNO₃. The trivalent actinides, Ra, and Cu were eluted with approximately 0.5 ml of 10 M HNO3. Th, Pa, U, Np, and Pu remained on the column and were subsequently eluted with various solutions. These activities, however, were not used in this study and will not be discussed in this text. A full treatment of the Th-Pu activities produced in the interactions of ⁴⁴Ca with ²⁴⁸Cm is given in [TÜR89]. The effluent containing the trivalent actinides, Ra, and Cu was evaporated to dryness. The remaining activity was dissolved in about 1 ml of 9 M HCl and passed through a 10-mm diameter X 50 mm long column of Bio Rad AG-1, X-8 (200-400 mesh) anion exchange resin which had been pretreated with 9 M HCl. The trivalent actinides and Ra were eluted from the column with the 9 M HCl while Cu was adsorbed on the resin.

The effluent from the second anion exchange column was evaporated to dryness and

the remaining activity was dissolved in 0.1 M HCl and passed through a 2-mm diameter X 50-mm long column of Bio Rad AG, MP-50 (200-400 mesh) cation exchange resin. This column had previously been treated with 0.1 M HCl. Ac and Am through Md were eluted with freshly saturated 13 M HCl in the first four free column volumes, ahead of Ra and the lanthanides. A known amount of ¹⁷⁰Tm tracer was added to the resulting fraction containing Ac and Am through Md. This fraction was dried, dissolved in 0.1 ml H₂O and dried again to ensure the removal of any residual acid. The residue was dissolved in about 0.2 ml of 0.02 M HNO₃ and passed through a 2-mm diameter X 50-mm long column of HDEHP on hydrophobic Celite 535. Actinium was eluted in that third free column volume with 0.1 M HNO₃. Upon adding 0.04 M HNO₃, Am and Cm were eluted in the second free column volume with Bk coming off in the fourth free column volume. 0.08 M HCl was then used to elute Cf through Md from the column.

To both the Am and Bk fractions, 1 mg each of La carrier and Zr/Ba hold-back carriers were added. 1 ml of 1 M HF was then added resulting in the coprecipitation of AmF3, and BkF3 along with the LaF3 in each fraction. Millipore type GS (22 µm) nitrocellulose filters were used to filter the precipitates. The dried precipitates were then mounted for gamma-ray spectroscopy.

The fraction containing the Cf, Es, Fm, and Md was heated to dryness to remove any residual acid and subsequently dissolved in 0.5 M HCl and sorbed on a 2-mm diameter by 50-mm long column of Dowex-50, X-12 cation exchange resin (7-10 μ m) kept at 80 °C. The HCl was removed from the column with 100 μ l of H₂O. The trivalent actinides were eluted from the column in separate fractions using 0.5 M ammonium alphahydroxyisobutyrate (α -HIB) at a pH of 3.80. The purified chemical fractions were collected on Pt discs in 2-drop increments. Each disc was monitored for alpha and spontaneous fission (SF) activities with a windowless 2π gas flow detector and for β -activity with a proportional counter. Those discs containing the same chemical element were combined on a single piece of Pt, dried, flamed, and mounted on aluminum

counting cards. The chemical procedure used on the copper catcher foils is represented schematically in Fig. II.4 and Fig. II.5.

b. Chemical Procedure for Gold Catcher Foils

After the first three experiments, it was realized that the interferences from other activities were small, therefore the chemical separations could be greatly simplified while still allowing accurate detection of the decay of the actinide nuclides of interest. Specifically, the cation exchange column utilizing AG MP-50 and the HDEHP reverse phase chromatographic column were excluded. Gold recoil catcher foils were used rather than copper, further simplifying the procedure. These modifications reduced the chemical separation times from 3.5 hours to about 2 hours. In addition, before each long bombardment a short bombardment of approximately 1 hour was performed. The short bombardments allowed the normalization of the +3 actinide chemical yields of the long bombardments and the determination of short-lived 250 Fm ($t_{1/2}$ = 30 m) cross sections.

The recoil products from the short bombardments were caught on a gold catcher foil which was subsequently removed and dissolved in approximately 0.25 ml of hot aqua regia containing a known amount of ²⁴¹Am tracer. An ether extraction was performed on this solution to remove about 95% of the gold [MYL11]. The aqueous solution was heated in a water bath to remove the dissolved ether and subsequently passed through a 3-mm diameter X 10-mm long column of Bio Rad AG-1, X-8 anion exchange resin in 12 M HCl. The remaining gold was adsorbed on the column while the trivalent actinides passed through the column. The effluent was heated to dryness, redissolved in 3 M HCl and aken to dryness again to remove any residual nitric acid. The remaining activity was dissolved in 3 M HCl and dried on a small piece of platinum, which was then mounted on an aluminum counting card for alpha-particle pulse height analysis. A flowchart of the chemical procedure for the gold foils from the short irradiations is given in Fig. II.6.

The gold catcher foil from each long bombardment was removed and dissolved in hot

aqua regia containing a known amount of 241 Am tracer. An ether extraction was performed, the aqueous solution was heated in a water bath to remove the dissolved ether, and then passed through a 3-mm diameter X 50-mm long column of Bio Rad AG-1, X-8 anion exchange resin in 12 M HCl. The eluate was evaporated to dryness and furned twice with concentrated HClO₄ to remove any residual organic material. The remaining activity was dissolved in 0.5 M HCl and sorbed on a 2-mm diameter by 50-mm long column of cation exchange resin (Dowex- 50, X-12, 7-10 μ m) maintained at 80 °C. The column was washed with approximately 100 μ l of H₂O to remove the HCl. Separate fractions of the trivalent actinides were eluted with 0.5 M ammonium alphahydroxyisobutyrate of pH 3.80. The eluate was collected on Pt discs in two-drop increments and dried. The discs were monitored for alpha and spontaneous fission activities with a windowless 2π gas flow detector, and for β - activity with a proportional counter. The activity on those discs containing the same chemical element were combined on a single piece of Pt for pulse height analysis. The chemical procedure used on the Au catcher foils from the long bombardments is presented in Fig. II.7.

F. Automated Chromatographic Chemical Element Separator System

It was desired to measure the cross sections of the various Md isotopes produced in the interactions of 12 C with 248 Cm. Due to the low production cross sections and short half-lives, a procedure was needed which incorporated reproducible chemical separations of the Md from the other reaction products on a time scale comparable to the half-lives of the Md isotopes of interest (254 Md $t_{1/2}$ = 10 m, 255 Md $t_{1/2}$ = 27 m, and 256 Md $t_{1/2}$ = 27 m). The Automated Chromatographic Chemical Element Separator System (ACCESS) was designed exactly for this type of separation [PAL88]. ACCESS is a computer-controlled separation system composed of high-performance liquid chromatography (HPLC) pumps and pneumatically actuated valves which guide a variety of solutions through a sample injection loop, and a chromatography column. The computer also controls an automated fraction collector. The modular design of ACCESS allows for great flexibility in the

type of separations that can be performed. For example, anion exchange, cation exchange, and reverse-phase extraction chromatography can be used with ACCESS to perform rapid, chemical separations. Two schematic diagrams of ACCESS are illustrated in Figs. II.8 and II.9.

Instead of using Au or Cu catcher foils to stop and collect the reaction products, a He/KCl-jet system was employed. The recoiling reaction products were thermalized in He (1.2 bar) which was seeded with KCl aerosol. After the thermalized reaction products became attached to the aerosol, the KCl aerosol was pumped through a polyvinyl chloride capillary tube to a collection site located above the cave shielding. The activityladen KCl aerosol was collected for approximately 20 minutes on 2 cm X 2 cm pieces of Teflon. At the end of the collection period, the KCl was dissolved in 95 µl of 0.05 M ammonium alpha-hydroxyisobutyrate (pH \cong 2, not adjusted), and introduced into ACCESS via the sample injection loop. This solution was sorbed on a 2-mm diameter X 70-mm long column containing cation exchange resin (Hamilton X-12, 10-15 µm) kept at 80 °C. The trivalent actinides were eluted in order of decreasing Z with 0.5 M ammonium alpha-hydroxyisobutyrate (pH = 3.36). The isolated Md and Fm fractions were collected in polyethylene centrifuges cones via an automated fraction collector, evaporated to dryness on Pt discs, and flamed for alpha pulse height analysis. Table II.6 lists the specific collection times and beam integrals for the experiments performed with ACCESS. It is estimated that the efficiency of the He/KCl-jet transport system is 75%. This efficiency was determined by comparing the yields of Fm isotopes from the transport system to those from a separate 1-h irradiation utilizing a Au catcher foil to collect the recoiling product atoms. The chemical procedure used on the Au foils from these 1-h irradiations is identical to that described in II.E.4.b.

G. Data Acquisition

1. Merry-Go-Round Realtime Data Acquisition and Graphics System

A rotating wheel system was used for the on-line measurements of the No cross

sections from the $(^{12}\text{C},xn)$ reactions with ^{248}Cm . The He/KCl-jet system was again utilized to transport the recoiling reaction products from the target system to the "Merry-Go-Round" (MG) horizontal, rotating wheel system (see Fig II.10) [HOF80]. The MG wheels rotated through a series of six passivated, ion-implanted, planar silicon detector pairs (100 mm² active area, 300-micron depletion depth) mounted above and below the plane of the wheel (see Fig. II.11). Thin polypropylene foils ($40 \pm 15 \,\mu\text{g/cm}^2$), mounted on 12.67 mm steel washers (6.3 mm i.d. hole), were placed at 80 equally spaced positions on the periphery of each MG wheel.

The activity-laden KCl aerosol was deposited on the first polypropylene foil for a preset time, after which the wheel was rotated 4.5°, moving the first collection foil from the collection site to a position located between the first pair of silicon detectors. After the second collection time, the wheel was rotated again, moving the first collection foil between the second pair silicon detectors, and moving the second collection foil between the first pair of detectors. This process continued until all 80 collection positions had been used. The MG wheel was replaced with a new one after about five complete revolutions in order to minimize the accumulation of any long-lived activities.

The α particle detection efficiency was determined to be 30%. The efficiency of the He/KCl-jet was determined with a separate 1-h irradiation utilizing an Au catcher foil as described earlier. The output signals from the ion implanted silicon detectors were digitized by Ortec AD-811 ADC's which were controlled by a Standard Engineering CAMAC crate controller. All events were stored on magnetic tape, in list mode, by our Realtime Data Acquisition and Graphics System (RAGS). Each event was tagged with a time, channel number (proportional to alpha particle energy), and detector number.

2. Counting Procedure

High purity Ge or Ge(Li) gamma-ray spectrometer systems were used to assay the Bk fractions. The Fm fractions were assayed with a fission-alpha spectrometer system utilizing Si(Au) surface barrier detectors with active areas of 100 mm². In the ⁴⁴Ca

experiments, the Cf fractions were first assayed on a gamma-ray spectrometer system to observe the K x-rays from the electron capture decay of ²⁴⁷Cf, and then on a fission-alpha spectrometer system. However, for the ⁴⁰Ar, ³¹P, and ¹²C experiments, the Cf fractions were assayed only on the fission-alpha spectrometer system. The Es samples were simultaneously assayed for alpha, gamma, and spontaneous fission activities by mounting the purified samples in a vacuum chamber containing a 100 mm² Si(Au) surface barrier detector. This chamber was then positioned next to a Ge(Li) gamma-ray detector. A schematic diagram of the alpha-gamma counting chamber is displayed in Fig. II.12. All of the separated samples were continuously counted for four weeks after the end of bombardment and at appropriate intervals so that unambiguous nuclide identification could be established by measuring both the decay energies and the half-lives.

H. Data Processing and Calculations

1. Processing of Alpha Data

Each alpha-spectrum consisted of a 1024-channel histogram with all fission events (events above a 30 MeV threshold) stored in channel zero. A typical Cf fraction alpha-spectrum is shown in Fig. II.13. All of the alpha spectra from each elemental fraction were summed so that every peak present could be located and tentatively identified based upon characteristic alpha-decay energies. This summing procedure aided in the identification of low intensity peaks which might not have been easily seen in any individual spectrum. Once all of the peaks were identified, regions of interest were set around the appropriate peaks. A region of interest was also set on channel zero to include any fissioning activities that might be present. The regions of interest in the individual spectra were integrated using INTEG2, a simple computer code [HAL86]. INTEG2 summed the counts in each channel within the regions of interest and constructed a decay curve for each region based on these summed counts and the times of measurement for each spectrum. The decay curves were subsequently fit by an error

weighted non-linear least squares program, EXFIT [GRE85]. Activities at the end of bombardment and the associated errors were calculated from the decay curve fitting procedure of EXFIT. Once the initial activities were calculated, the absolute count rates were determined by correcting for the abundances of specific decay modes, detector efficiencies, chemical yields, and the fraction of the alpha peaks contained within the regions of interest. Table II.7 lists the α emitters used in this study along with the associated half-lives, decay energies, and α decay branches. The nuclides measured which decay by spontaneous fission are shown in Table II.8.

In order to determine the efficiencies of each 100 mm² Si(Au) surface barrier detector, a standard alpha source containing a known amount of ²⁴¹Am was prepared before each experiment. The efficiencies typically ranged from 16 to 21% with the efficiency for fission fragments assumed to be twice that of alpha particles.

2. Processing of Gamma Data

Each gamma spectrum consisted of a 4096-channel histogram. Fig. II.14 shows the first 2000 channels of a representative gamma-spectrum from a Bk fraction eluted from the α-HIB column. The gamma spectra from each elemental fraction were analyzed using the SAMPO computer code [ROU69]. This code contains algorithms for line-shape, energy, and efficiency calibrations. In addition it has automatic peak-search and peak-fitting routines, as well as the ability to deconvolute multiplet peaks. Decay curves were constructed from the peak areas found by SAMPO. Initial activities were calculated with CLSQ [CUM63], a non-linear, least squares, multi-component decay curve fitting program. Abundances of the specific gamma-decay energies were used to calculate the absolute disintegration rates at the end of bombardment of the nuclides of interest. Table II.9 lists some decay information for the nuclides of interest which decay by electron capture (EC) or beta emission (β-).

3. Calculation of Cross Sections

The number of product nuclei produced per unit time, dN_p/dt , for a constant beam of

particles striking a thin target is

$$\frac{dN_p}{dt} = IN_t \sigma, \tag{4}$$

where I is the number of incident particles striking the target per unit time, N_t is the number of target atoms per cm², and σ is the cross section for the production of this product. However, if the nuclide being produced is unstable and decays during the bombardment, then the net rate of production of this nuclide is

$$\frac{dN_p}{dt} = IN_t \sigma - \lambda_p N_p, \tag{5}$$

where λ_p is the decay constant of the product and N_p is the number of product nuclei present at time t. Assuming that I is constant, Eq. 2 can be solved by rearrangement, introduction of an integrating factor, and finally integration to give [STE66]

$$N_{t}I\sigma\{1-\exp[-\lambda_{p}(t-t_{o})]\}$$

$$N_{p} = N_{po}\exp[-\lambda_{p}(t-t_{o})] + \frac{\lambda_{p}}{\lambda_{p}}, \qquad (6)$$

where N_p is the number of target nuclei present at the end of bombardment, t is the end of bombardment time, t_o is the start of bombardment, and N_{po} is the number of product nuclei present at the beginning of bombardment. If at the start of bombardment there are no product nuclei present, then Eq. 3 reduces to

$$N_{t}I\sigma\{1-\exp[-\lambda_{p}(t-t_{o})]\}$$

$$N_{p} = \frac{\lambda_{p}}{\lambda_{p}}.$$
(7)

Rearranging Eq. 4 and solving for σ , one obtains

$$\sigma = \frac{\lambda_p N_p}{N_t I\{1-\exp[-\lambda_p(t-t_o)]\}} = \frac{A_p}{N_t I\{1-\exp[-\lambda_p(t-t_o)]\}},$$
(8)

where A_p is the activity of the product at the end of bombardment. In reality, the beam intensity is not constant, so the beam history must be divided into j intervals giving

$$\sigma = \frac{A_p}{N_t \sum_{j} [1 - \exp{-\lambda_p(t_{j+1} - t_j)}] \exp[-\lambda_p(t - t_{j+1})]},$$
(9)

where I_j is the average beam intensity during the jth interval, t_j is the time at the start of the jth interval, and t is the time at the end of bombardment. Therefore, Eq. (9) was used to calculate the cross sections of the nuclides of interest.

Table II.1. Projectile, ²⁴⁸Cm target thicknesses, catcher foil material, and catcher foil thicknesses.

Exp#	Projectile	Target Thickness (mg/cm²)	Catcher Foil Material	Catcher Foil Thickness (mg/cm ²)
1-3	44Ca	0.644	Cu	5.7-7.0
4,5	44Ca	0.644	Au	19.6
6-8	40_{Ar}	0.644	Au	19.6
9,10	40_{Ar}	0.491	Au	12.3
11-15	31p	0.491	Au	12.3
16-20	12 _C	0.182	Au	2.47

Table II.2. Maximum recoil energy, range in catcher foil, thickness of catcher foil, and composition of catcher foil for experimental systems used in this study.

Proj	Max. Recoil Energy (MeV)	Range in Catcher Foil (mg/cm ²)	Thickness of Catcher Foil (mg/cm²)	Catcher Foil Material
^{2}C	15	2.1	2.47	Au
1 _P	81	8.7	12.3	Au
0_{Ar}	124	11.2	12.3	Au
⁴ Ca ⁴ Ca	141	12.0	19.6	Au
¹⁴ Ca	141	5.7	7.0	Cu

Table II.3. Experimental conditions for the long irradiations of 248 Cm with 44 Ca, 40 Ar, 31 P, and 12 C ions.

Exp#	Proj.	Charge State	Ave. Energy in the Target (MeV)	Length of Bombardment (min)	Ave. Beam Flux (PnA)
1	44Ca	10+	231.7	650	90.0
2	44Ca	10+	251.9	474	28.0
3	44Ca	12+	274.8	763	18.3
4	44Ca	12+	297.0	326	11.7
5	44Ca	12+	319.1	450	16.7
6	40_{Ar}	9+	207.0	412	77.8
7	40 _{Ar}	11+	224.9	276	59.1
8	40Ar	11+	245.0	480	31.8
9	40_{Ar}	11+	265.5	641	36.4
10	40Ar	11+	285.5	415	29.1
11	31 _P	8+	173.8	675	62.5
12	31 _P	8+	190.0	654	100
13	31 _P	8+	207.4	743	93.8
14	31 _P	8+	223.1	538	52.5
15	31 _P	8+	238.9	647	50.0
16	12 _C	4+	70.7	355	675
17	^{12}C	4+	76.4	595	800
18	12 _C	4+	83.6	484	800
19	12 _C	4+	90.1	406	500
20	12C	4+	97.3	489	675

Table II.4. Composition of the alloy HAVAR. Taken from [GRE85].

Element	% By Weight	
Co	42.5	
Cr	20.0	
Fe	17.9	
Ni	13.0	
W	2.8	
Mo	2.0	
Mn	1.6	
C	0.2	
Be	0.04	

Table II.5. Ionic radii of +3 actinides [BRA86].

Z	Symbol	Radius (nm)
89	Ac	1.120
90	Th	1.088
91	Pa	1.060
92	U	1.035
93	Np	1.013
94	Pu	0.994
95	Am	0.976
96	Cm	0.960
97	Bk	0.945
98	Cf	0.932
99	Es	0.920
100	Fm	0.909
101	Md	0.898
102	No	0.889
103	Lr	0.880

Table II.6 Experimental conditions for the irradiations of ²⁴⁸Cm with 71-MeV ¹²C+4 to produce Md isotopes.

Run #	Length of Irradiation (min)	Beam Integral (µC)	
1	20	2740	
2	30	3991	
3	36	4623	
4	34	4136	
5	32	3723	
6	31	3917	

Table II.7. Decay properties of α emitters used in the present study [BRO78].

elide	Half Life	α Decay Energy* (MeV)	Total α Decay Branch (%)
246	35.7 h	6.758	100
248	333.5 d	6.26	100
250	13.08 y	6.031	100
252	2.65 y	6.118	96.908
251	33.0 h	6.492	0.49
252	471.7 d	6.632	78
253	20.47 d	6.633	100
250	30.0 m	7.44	100
252	25.39 h	7.04	100
253	3.00 d	6.943	12
254	3.24 h	7.187	99.94
255	20.07 h	7.022	100
_	248 250 252 251 252 253 250 252 253 254	248 333.5 d 250 13.08 y 252 2.65 y 251 33.0 h 252 471.7 d 253 20.47 d 250 30.0 m 252 25.39 h 253 3.00 d 254 3.24 h	246 35.7 h 6.758 248 333.5 d 6.26 250 13.08 y 6.031 252 2.65 y 6.118 251 33.0 h 6.492 252 471.7 d 6.632 253 20.47 d 6.633 250 30.0 m 7.44 252 25.39 h 7.04 253 3.00 d 6.943 254 3.24 h 7.187

^{*}Most abundant decay energy.

Table II.8. Decay properties of spontaneously fissioning nuclides [BRO78].

Nuclide	Total Half Life	SF Branch
Cf 252 254	2.64 y 60.5 d	3.092 99.69
Fm 256	2.627 h	91.9

Table II.9. Half lives, decay modes, photopeak energies, and absolute abundances of photopeaks for those nuclides decaying via electron capture (EC) or beta emission (β⁻) [BRO78].

Nuc	clide	Half Life	Decay Mode	Photopeak Energy (keV)	Absolute Abundance of Photopeak (%)
Bk	244 245 246	4.35 h 4.90 d 1.80 d	EC EC EC	217.6 252.9 798.7	100* 31.3 61.4
	248m	23.5 h	ß-	550.7	4.59
	250	3.22 h	ß-	989.0	45.23
	251	55.6 m	ß-	152.8	100*
Cf	247	3.15 h	EC	112.1+	33.7
Es	249	1.70 h	EC	379.5	40.4
	250m	2.1 h	EC	989.0	16.8
	250g	8.6 h	EC	829.0	73.6
	251	33.0 h	EC	177.6	≅2.0

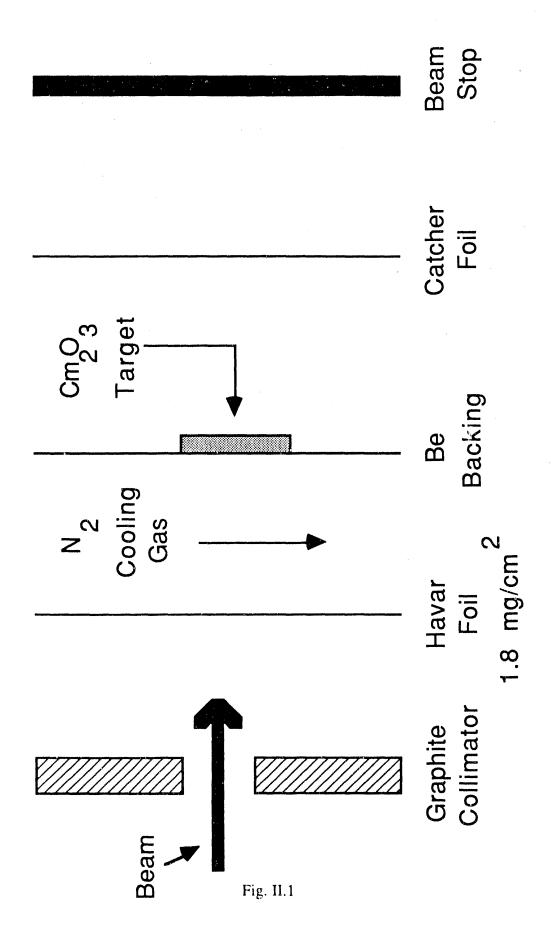
^{*}The absolute abundances are unknown and are therefore assumed to be 100%.

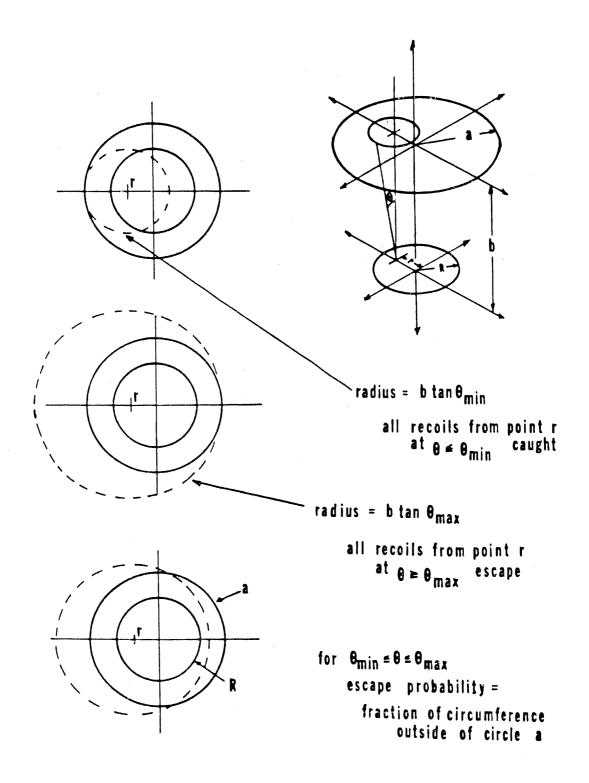
⁺This is actually the photopeak energy of the $K_{\alpha 1} \ x\text{-ray}$ from the Bk daughter.

Figure Captions

- Figure II.1. Target system schematic diagram. The distance between the graphite collimator and the HAVAR foil is about 40 mm, the target is 3 mm from the HAVAR foil, the catcher foil is approximately 4 mm from the target, and the beam stop is about 120 mm from the catcher foil.
- Figure II.2. Illustrative diagram of the calculations used to determine the probability of a recoiling reaction product missing the Cu and Au catcher foils as a function of laboratory angle. Taken from [MOO83].
- Figure II.3. Recoil collection efficiency as a function of laboratory angle for the present target system in Cave 0 at the LBL 88-Inch Cycletron.
 - Figure II.4. Flowchart of the chemical procedure used on the Cu catcher foils (Part I).
- Figure II.5. Flowchart of the chemical procedure used on the Cu catcher foils (Part II).
- Figure II.6. Flowchart of the chemistry used on the Au foils from the short irradiations.
 - Figure II.7. Chemistry flowchart for the Au foils from the long irradiations.
- Figure II.8. Automated control system diagram for ACCESS. The Gilson FC 203 fraction collector is controlled by the personal computer (PC) through an RS-232C serial interface. The conditioning of the analytical column is controlled by software throughout each experiment. Schematic taken from [PAL88].
- Figure II.9. Schematic diagram of ACCESS for Md experiments. Taken from [KAD90].
- Figure II.10. Schematic diagram of the Merry-Go-Round Realtime Data Acquisition and Graphics System "MG-RAGS" at the LBL 88-Inch Cyclotron.
 - Figure II.11. Schematic diagram of "MG" wheel and detector arrangement.
 - Figure II.12. Alpha-gamma counting chamber used to assay isolated Es samples.
 - Figure II.13. Typical alpha spectrum from an isolated Cf fraction.

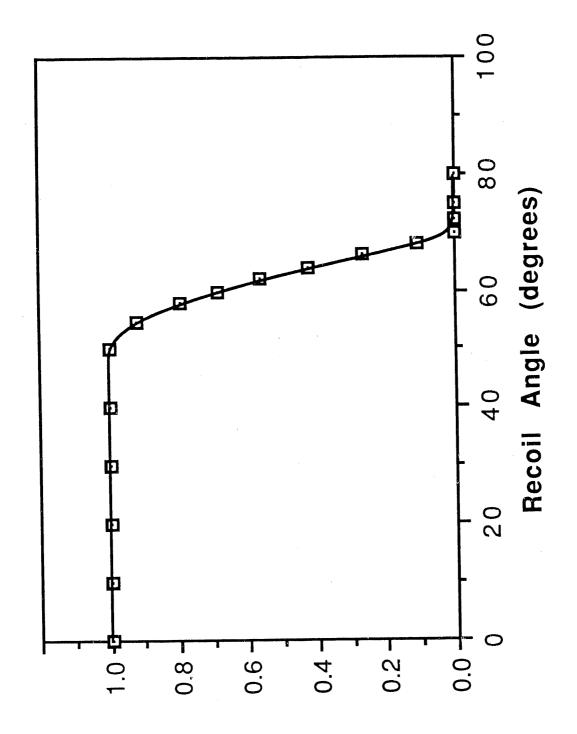
Figure II.14. Typical gamma spectrum from an isolated Bk fraction.





XBL 836-10249

Fig. II.2



Prob. of Catching Recoil

Fig. II.3

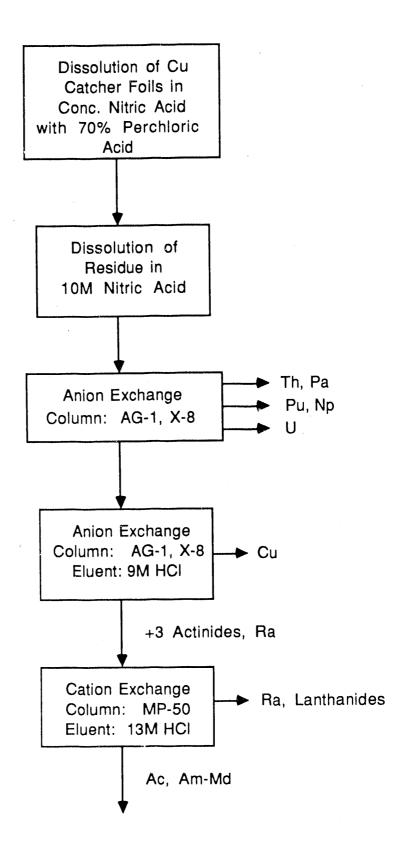


Fig. II.4

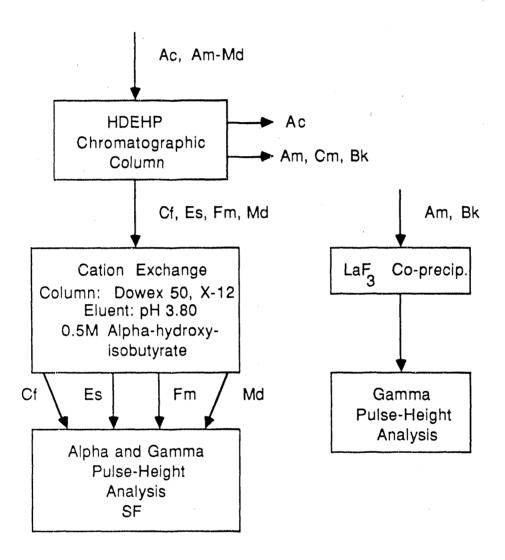


Fig. II.5

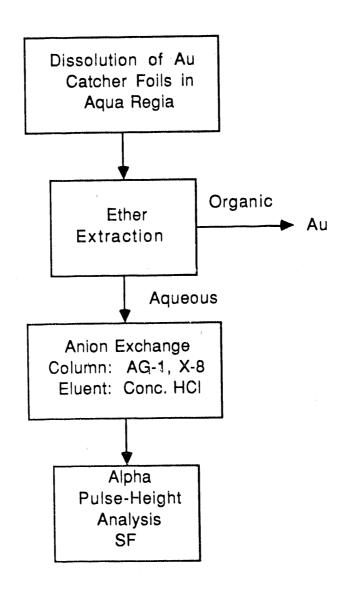


Fig. II.6

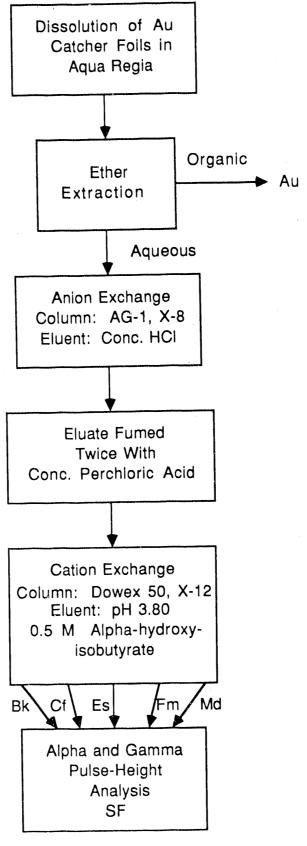


Fig. II.7

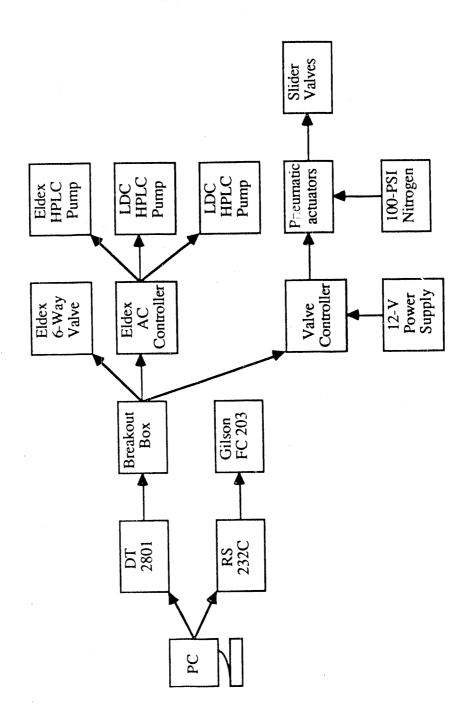


Fig. II.8

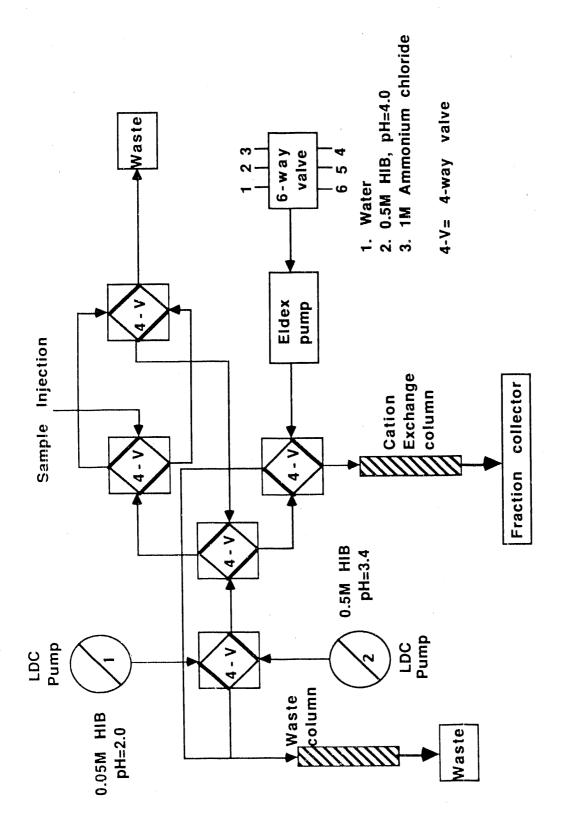


Fig. II.9

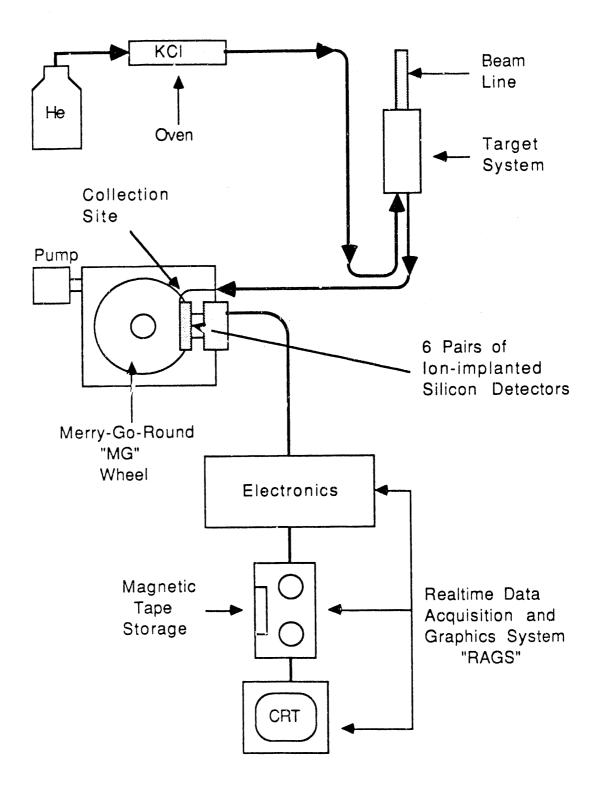
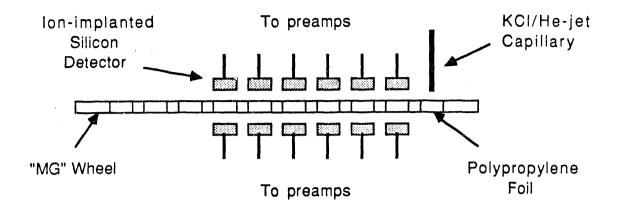
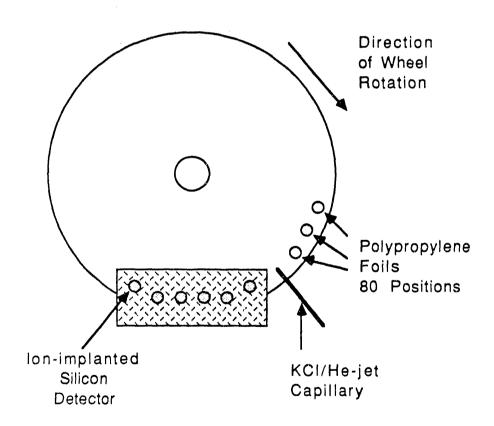


Fig. II.10



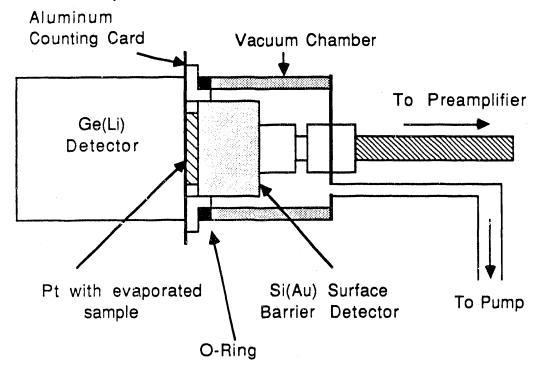
Side View



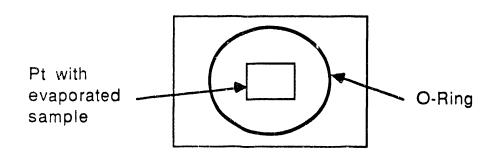
Top View

Fig. II.11

Alpha-Gamma Counting Chamber



Side View



Front View of Counting Card

Fig. II.12

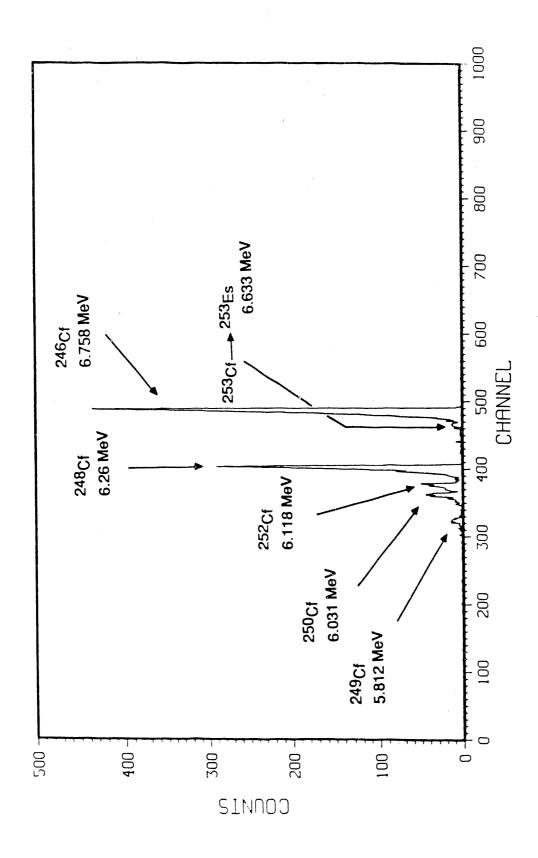


Fig. II.13

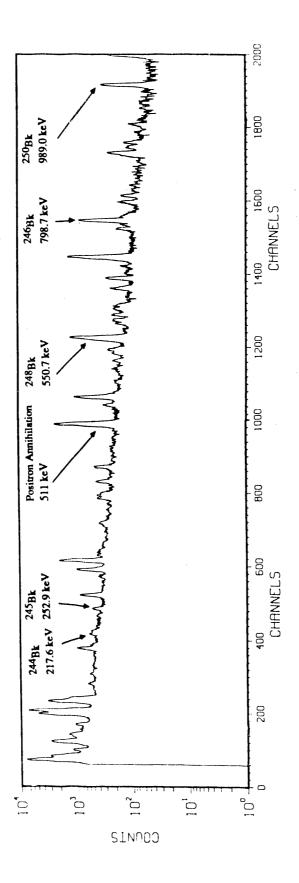


Fig. II.14

III. RESULTS AND DISCUSSION

A. Cross Sections and Excitation Functions

Theoretically, for the capture of charged particles, the cross section should increase with increasing bombarding energy of the projectile, since the maximum impact parameter, b_m , for such processes goes as

$$b_m = R\left(1 - \frac{V_C}{\varepsilon}\right). \tag{1}$$

R is the interaction radius, V_C is the Coulomb barrier between the target and the projectile, and ε is the kinetic energy of the projectile. In addition, the cross section for the capture of charged particles can be estimated based upon the geometric cross section of a disk with radius equal to the interaction radius, R:

$$\sigma_{geom} = \pi R^2. \tag{2}$$

However for two nuclei to classically interact, $b_m < R$. Therefore, an upper limit for the capture of charged particles can be written as

$$\sigma_{ul} = \pi R^2 \left(1 - \frac{V_C}{\varepsilon} \right) = \sigma_{geom} \left(1 - \frac{V_C}{\varepsilon} \right). \tag{3}$$

From Eq. 3 it can be seen that the cross section asymptotically approaches the geometric cross section at high ε . However, as more energy is deposited into the system, the probability for particle emission and/or fission increases. Hence, at some higher energy, the decrease in yield due to particle emission and fission should be greater than the increase resulting from the higher bombarding energy. As a result the net production

cross section for a given nuclide decreases.

Cross sections for nuclides produced from the interactions of 12 C, 31 P, 40 Ar, and 44 Ca with 248 Cm are listed in Tables III.1-III.5. The energies of the projectiles are given in the lab frame of reference. All cross sections are given in microbarns (μ b). The statistical standard deviation, s, for a given cross section is listed as a percentage of the production cross section. A less than sign (<) in front of the cross section indicates that no activity was detected, therefore an upper limit on the activity was set at the 1σ (68%) confidence level. In particular, for nonexistent alpha peaks, it was assumed that <2.3 counts were seen in the specific counting interval where no activity was detected. For those nonexistent gamma peaks, a formalism developed by Gregorich [GRE85] was used to determine upper limits. This formalism is based upon the definition of chi squared. Hence, the upper limit on the activity of a nuclide at the end of bombardment is given by

$$\begin{array}{ccc}
N & \exp(-2\lambda t_i) \\
A_O = \sum & & \\
i=1 & \sigma_i 2
\end{array} , \tag{4}$$

where λ is the decay constant, σ_i is the error in the background count rate of the *i*th interval, and t_i is the time between the end of bombardment and the *i*th interval.

A given nuclide can be produced directly via a transfer reaction or indirectly by the decay of a parent nuclide produced in the bombardment. Hence, in order to obtain the cross section for the transfer reaction, the amount of the nuclide produced via the indirect route(s) must be subtracted from the amount produced directly. Therefore, corrections have been made, where appropriate, to the cross sections of the daughter ruclides from the decay of radioactive parent nuclides during the bombardment and the time interval between the end of bombardment and chemical separation. For example, the decay of 248 Bk m ($t_{1/2} = 23.7$ h) to 248 Cf ($t_{1/2} = 334$ d) represents a significant

correction ($\cong 25\%$) to the amount of 248 Cf produced directly via a transfer reaction.

Only the relative intensities are known for the gamma rays from 244 Bk. However, by assuming the electron capture branch to be 100%, and the fluorescence yield to be 0.978, and knowing the relative intensities of the Cm x-rays, the absolute intensity of the 217.6 keV gamma ray was estimated to be $37.4 \pm 12\%$ [BRO78]. The decay scheme of 251 Bk has not been measured, therefore the cross sections for this nuclide are based upon the assumption that the absolute intensity of the most abundant gamma ray is 100%; therefore, these values are lower limits only.

The cross sections plotted for ²⁴⁸Bk and ²⁵⁴Es are in reality only lower limits due to the fact that there were unobserved isomers. The effects of these unobserved isomers on the mass-yield curves are discussed in section III.G.c.

B. Excitation Energies

Hoffman and Hoffman [HOF82] have developed a formalism to calculate the energy available for product excitation in transfer reactions. This model is based upon a binary-type reaction in which the target and projectile exchange nucleons but do not amalgamate into a compound nucleus. The center of mass bombarding energy, E_{cm} , of the projectile is given by

$$E_{cm} = E_b(i) + KE(i) = E_b(f) + KE(f) + KE(R) - Q_{gg} + E_X,$$
 (5)

where $E_b(i)$ and $E_b(f)$ are the entrance channel and exit channel Coulomb barriers respectively, KE(i) is the residual kinetic energy of the projectile (the amount of energy over the entrance channel Coulomb barrier), KE(R) is the rotational kinetic energy of the target-like fragment, Q_{gg} is the ground state Q value of the reaction channel (mass of the reactants - mass of the products), and E_X is the excess energy available for product excitation (excitation energy).

KE(f) was calculated assuming that the kinetic energy per nucleon of the projectile-

like fragment is the same as the residual kinetic energy per nucleon of the projectile at the point of contact with the target, assuming two touching spheres.

Hence

$$KE(f) = [KE(i)] \frac{A(f)}{A(i)} = [E_{cm} - E_b(i)] \frac{A(f)}{A(i)}, \qquad (6)$$

where A(i) is the mass of the projectile and A(f) is the mass of the projectile-like fragment. Since this model was developed for reactions at energies near the entrance channel Coulomb barrier $(E_{cm} \subseteq E_b(i))$, KE(f) and KE(R) are always small in comparison to E_{χ} . KE(R) was calculated using

$$KE(R) = \frac{\hbar^2 [J(J+1)]}{2J}, \qquad (7)$$

where J is the rotational angular momentum, and I is the nuclear moment of inertia. For these calculations I was set equal to one half the moment of inertia of a rigid sphere.

If E_X is negative, an energy barrier must be overcome for the reaction to proceed with a high probability. Reactions with negative excitation energies should therefore display excitation functions with maxima at some energy above the Coulomb barrier. The cross sections should then decrease at higher projectile kinetic energy as fission or particle emission becomes energetically possible. For reactions with positive excitation energies (of the order of neutron binding energies or fission barriers) the excitation function should peak at or near the Coulomb barrier and decrease with increasing projectile kinetic energy. E_X 's have been calculated for several target-like and projectile-like fragments from the 12 C, 31 P, 40 Ar, 40 Ca, 44 Ca, and 48 Ca systems and are given in Tables III.6-III.10.

C. Division of Energy

Lee et al. [LEE83] estimated the fraction, \boldsymbol{F}_t , of the projectile kinetic energy transformed into target-like fragment excitation energy using the equation

$$F_t = \frac{E_{f,n} - E_X}{E_M - V_C}, \tag{8}$$

where $E_{f,n}$ is the height of the fission barrier [BRI79] [BRI80] or the neutron binding energy [SEE67] in MeV whichever is lower, E_X is the excitation energy described earlier, E_M is the maximum in the experimental excitation function, and V_C is the calculated Coulomb barrier. This formula assumes that the maximum in the excitation function occurs when the product excitation energy is comparable to the fission barrier or neutron binding energy. This calculation in not applicable when the reaction energy is greater than the neutron binding energy or the fission barrier, because enough energy is already available to cause depletion of the cross section by fission or neutron emission. The calculated F_t values for the ^{12}C , ^{31}P , ^{40}Ar , and ^{44}Ca systems are given in Tables III.11-III.14. E_t , the fraction of energy transferred assuming energy transfer is proportional to mass transfer, is also shown in Tables III.11-III.14. E_t , which can appear as excitation energy of the heavy product, was calculated using the formula

$$E_t = \frac{M}{A} \tag{9}$$

where M is the number of nucleons transferred and A is the mass number of the projectile. F_t and E_t values for the actinides produced in the ¹²C, ³¹P, ⁴⁰Ar, and ⁴⁴Ca systems are discussed with each individual system in the following sections.

D. 12C-248Cm System

Production cross sections for the measured isotopes of Bk, Cf, Es, and Fm from the interactions of ¹²C with ²⁴⁸Cm along with the statistical standard deviation, s, are shown in Table III.1. The measured production cross sections for some Md and No isotopes are given in Table III.2. The calculated Coulomb barrier is 72 MeV in the lab system. Ghiorso et al. [GHI67] measured production cross sections for ²⁵⁵No, ²⁵⁶No, and ²⁵⁷No at 71 MeV from the ¹²C-²⁴⁸Cm system. They obtained values of 380, 740, and 80 nb respectively for ²⁵⁵No, ²⁵⁶No, and ²⁵⁷No, versus 21.5, 203, and 33.2 nb obtained in the present study. Hoffman et al. [HOF90] have also measured the ²⁴⁸Cm(¹²C,4n)²⁵⁶No cross section at 71 MeV and obtained a value of 250 nb, in agreement with the value of this work. The discrepancy between Ghiorso's values and those in this thesis are not understood.

1. Isotopic Distributions

The isotopic distributions for Bk, Cf, Es, Fm, and No from the interactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm are shown in Fig. III.1. This energy was chosen so that this system could be compared to other systems at the same energy relative to the entrance channel Coulomb barrier. All of the data points for each element are connected with lines. These lines are included only to aid the eye in viewing the plot. The mass yield curves have maxima at ²⁴⁸Bk, ²⁵⁰Cf, ²⁵¹Es, ²⁵³Fm, and ²⁵⁵No. The FWHM's are 2.1, 1.7, <1.5, 1.8, and 1.8 mass units for the Bk, Cf, Es, Fm, and No curves, respectively. Since neither ²⁵⁰Es nor ²⁴⁹Es were seen at this energy, only an upper limit can be set on the width of the Es curve.

Isotopic distributions for Bk, Cf, Es, Fm, Md, and No from the interactions of 71-MeV (0.98 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm are plotted in Fig. III.2. The curves have FWHM's of 1.7, 1.6, <1.0, 1.3, 1.0, and 1.0 mass units for Bk, Cf, Es, Fm, Md, and No respectively. The widths of these isotopic distributions are narrower than the corresponding widths at 1.16 X Coulomb barrier.

2. Excitation Functions

a. Excitation Functions for Bk Isotopes

The excitation functions for the Bk isotopes 245, 246, 248, and 250 are displayed in Fig. III.3. The lines connecting the experimental cross sections for a given isotope are included only to aid the eye. Assuming a binary-type transfer mechanism, the production of all of the Bk isotopes requires the transfer of one proton from the projectile to the target, along with the transfer of various numbers of neutrons to or from the target. For example $^{248}Bk^m$ requires the transfer of only one neutron whereas the production of 244Bk requires a five neutron transfer from the target to the projectile. Therefore, the isotope requiring the transfer of the fewest number of nucleons for which E_X is positive (but not large enough to cause fission or neutron emission) should have the highest cross section. ²⁴⁹Bk meets these requirements. This nuclide requires the transfer of just one proton from the projectile to the target and has an E_X of -0.7 MeV. However, the $^{249}\mathrm{Bk}$ cross sections were not included for two reasons. First of all ²⁴⁹Bk is used as both a tracer and a target in our laboratory and secondly its measurement is extremely difficult. Since ²⁴⁹Bk is used in our laboratory, contamination is a possibility that has to be considered. The difficulty in measurement stems from the fact that the absolute intensities of the most abundant gamma ray and alpha branch from ²⁴⁹Bk are on the order of 1 X 10-5. Since ²⁴⁹Bk has a 99+% B branch, its activity can be determined from the growth of the 249 Cf daughter. However, the long half-life of 249 Cf ($t_{1/2}$ = 351 y) makes the detection of this nuclide difficult also. From the experimental data it can be seen that ²⁴⁸Bk^m has the highest cross sections over the energy range studied followed by 250Bk, 246Bk, and 245Bk. The production of both 248Bkm and 250Bk requires the transfer of two nucleons between the target and projectile. However, the E_X value of 248Bkm is -3.5 MeV whereas 250Bk has an E_X of -7.6 MeV; hence, the higher production cross sections for ²⁴⁸Bk^m relative to ²⁵⁰Bk. ²⁴⁵Bk, which requires the transfer of five nucleons has lower production cross sections than ²⁴⁶Bk (requiring the transfer of four nucleons) over the energy range studied. Since the excitation functions are increasing with increasing bombarding energy, they are consistent with their calculated negative E_X values.

b. Excitation Functions for Cf Isotopes

Fig. III.4 displays the excitation functions for 246 Cf, 248 Cf, 250 Cf, 252 Cf, and 253 Cf. As can be seen from this figure, 250 Cf, the isotope requiring the transfer of the fewest nucleons between the target and projectile, has the highest production cross sections. Two of the isotopes, 250 Cf and 252 Cf, have positive E_x 's and display excitation functions which have maxima near the Coulomb barrier, consistent with their calculated E_x 's of 4.9 and 6.9 MeV. 246 Cf, 248 Cf, and 253 Cf each have negative E_x 's with excitation functions consistent with this fact.

c. Excitation Functions for Es Isotopes

The excitation functions measured for the Es isotopes 251, 252, 253, 254m, and 255 are plotted in Fig. III.5. The isotope requiring the transfer of the fewest nucleons, 251, has the highest cross sections at two energies, with 252 having the highest cross sections at the other three energies. 255 Es, the isotope requiring the transfer of the most nucleons (seven), has the lowest cross sections. The shapes of the excitation functions of these Es isotopes are consistent with their calculated positive E_X , values with the possible exception of 251 Es. No 251 Es was found at 71 MeV; therefore, only an upper limit could be set on the cross section.

d. Excitation Functions for Fm Isotopes

Fig. III.6 shows the excitation functions for the Fm isotopes 252, 253, 254, 255, and 256. ²⁵³Fm, which requires the transfer of five nucleons has the highest cross sections at three energies, followed by ²⁵²Fm, ²⁵⁴Fm, ²⁵⁵Fm, and finally ²⁵⁶Fm. It should be noted that even though ²⁵²Fm requires the transfer of only four nucleons, it has lower production cross sections than ²⁵³Fm at the three lower energies. The higher cross

sections for 253 Fm could be due to the greater E_X value of this nuclide (5.7 MeV) as compared to 252 Fm ($E_X = 3.0$ MeV). This would explain the lower cross sections of 253 Fm at the two higher energies. Since more energy would be available for fission and neutron evaporation, the cross section for 253 Fm should diminish faster than the cross section for 252 Fm. 252 Fm has an excitation function shape that is not consistent with its calculated positive E_X value. Likewise, the shape of 256 Fm's excitation function is consistent with a negative E_X value, not its very positive value of 22.4 MeV.

e. Ft and Et Values

 F_t and E_t values for the ¹²C system are shown in Table III.11. The F_t 's range in value from 0.08 to 0.72. No trends can be seen in the F_t values and no correlations between F_t and E_t are apparent. In general, however, E_t is less than F_t , indicating that the fraction of energy transferred is greater than the fraction of mass transferred.

E. 31p.248Cm System

The production cross sections along with the statistical standard deviation, s, for the measured isotopes of Bk, Cf, Es, and Fm from the interactions of ³¹P with ²⁴⁸Cm are listed in Table III.3.

1. Isotopic Distributions

Isotopic distributions for Bk, Cf, Es, and Fm produced from the interactions of 207-MeV (1.17 X Coulomb barrier) ³¹P ions with ²⁴⁸Cm are plotted in Fig. III.7. The isotopic distributions have maxima at ²⁴⁸Bk^m, ²⁵⁰Cf, ²⁵¹Es, and ²⁵²Fm. All of these nuclides correspond to the isotope requiring the transfer of the fewest number of nucleons for a given element, except for ²⁴⁸Bk^m. All of the curves appear to be symmetric with full-widths-at-half-maximum (FWHM) of 2.5, 2.5, <1.5, and 2.3 mass units for Bk, Cf, Es, and Fm respectively. Since only upper limits were set on the ²⁴⁹Es, and ²⁵⁰Es cross sections, only an upper limit can be set on the FWHM of the Es curve. Additionally, the maxima and the widths of the mass-yield curves stay constant over the energy range investigated.

2. Excitation Functions

a. Excitation Functions for Bk Isotopes

Excitation functions for the Bk isotopes 244, 245, 246, 248m, and 250 are shown in Fig. III.8. 248 Bk m has the highest cross sections at all energies, with 246 Bk next, followed by 250 Bk, 245 Bk, and finally 244 Bk. All of the Bk excitation functions have maxima at about 207 MeV. The shapes of the excitation functions of 244 Bk and 248 Bk m are consistent with their positive excitation energies. 245 Bk, 246 Bk, and 250 Bk have excitation functions which are very similar in shape. They increase by about a factor of three between 174 and 207 MeV and then show a decrease by more than a factor of two over the next 32 MeV. The shapes of the excitation functions of 245 Bk and 246 Bk, however, do not appear to be consistent with their calculated positive excitation energies. Since their E_x 's are positive, one would not expect an increase in the cross sections at the lower bombarding energies

b. Excitation functions for Cf isotopes

Excitation functions for the various Cf isotopes measured in this study are shown in Fig. III.9. 250 Cf has the highest cross sections, followed by 248 Cf, 246 Cf, 252 Cf, and finally 253 Cf. 250 Cf requires the transfer of the fewest number of nucleons, two protons, and additionally it has an E_X of 7.2 MeV. 248 Cf has much higher cross sections than 252 Cf even though both require the transfer of four nucleons. In fact, 252 Cf can be produced by the transfer of an alpha particle from the projectile to the target. However, 248 Cf has an E_X of 8.8 MeV while 252 Cf has an E_X of only 0.6 MeV. This could explain the lower yields of 252 Cf relative to 248 Cf. The 248 Cf, 250 Cf, and 252 Cf excitation functions have similar shapes, consistent with their positive E_X 's. However, for 253 Cf, the excitation function at the Coulomb barrier is near its maximum and decreases with increasing projectile energy, inconsistent with its E_X of -8.1 MeV. 246 Cf, with an E_X of 6.9 MeV, has an excitation function which is consistent with a negative E_X value.

c. Excitation functions for Es isotopes

Fig. III.10 shows excitation functions for the Es isotopes 251, 252, 253, 254, and 255. The highest cross sections occur for ²⁵¹Es, the isotope requiring the transfer of the fewest number of nucleons (three protons). As the number of nucleons transferred increases, the cross sections decrease. In addition, the shapes of the excitation functions are consistent with their calculated excitation energies.

d. Excitation functions for Fm isotopes

Excitation functions for several isotopes of Fm are shown in Fig. III.11. The shapes of the curves are all similar, and again they are consistent with the calculated positive excitation energies. 252 Fm, the Fm isotope requiring the transfer of the fewest number of nucleons (four protons) has the highest cross sections with 256 Fm, which requires the transfer of eight nucleons, having the lowest cross sections. 255 Fm seems to be an exception in this case. Even though it requires the transfer of seven nucleons, it has higher cross sections at three energies than both 254 Fm and 250 Fm which each require the transfer of six nucleons. The higher cross sections of 255 Fm relative to 254 Fm and 250 Fm cannot be explained on the basis of excitation energies. 255 Fm has an 250 Fm respectively. However, it should be noted that the 255 Fm cross sections have relatively large errors.

e. Ft and Et Values

Table III.12 lists the calculated F_t and E_t values for the $^{31}\mathrm{P}$ system. For 12 of the 21 cases, the value of E_x was gre. In than $E_{f,n}$ resulting in a negative value for F_t . The F_t 's from the $^{31}\mathrm{P}$ system range in value from 0.01 to 0.69, similar to the values from the $^{40}\mathrm{Ar}$ and $^{44}\mathrm{Ca}$ systems (discussed below). No obvious trends in the F_t 's can be seen. There is good agreement between F_t and E_t for $^{252}\mathrm{Cf}$, $^{255}\mathrm{Es}$, $^{255}\mathrm{Fm}$, and $^{256}\mathrm{Fm}$. However, there are significant deviations for $^{250}\mathrm{Bk}$, $^{253}\mathrm{Cf}$, $^{254}\mathrm{Es}^m$, $^{250}\mathrm{Fm}$, and $^{253}\mathrm{Fm}$.

F. 40Ar-248Cm System

The cross sections obtained from the 40 Ar system are summarized in Table III.4 with the statistical standard deviations, s.

1. Isotopic distributions

Isotopic distributions for Bk, Cf, Es, and Fm from the interactions of 245-MeV (1.16 X Coulomb barrier) 40 Ar ions with 248 Cm are plotted in Fig. III.12. The maxima of the mass-yield curves for the 40 Ar system occur at 248 Bkm, 250 Cf, 251 Es, and 252 Fm. Since the cross sections for the neutron-deficient Fm isotopes were not measured, the peaks of the mass distributions for the Fm isotopes for this system cannot be located with certainty. However it can be said that the peaks occur at a mass number of 252 or less. The measured isotopic distributions are essentially symmetric about the maximum with FWHM's of 3.3, 2.8, and 1.5 mass units for Bk, Cf, and Es, respectively. In addition, the maxima and the widths of the mass-yield curves for the 40 Ar system stay essentially constant over the energy range in which cross sections were measured.

2. Excitation Functions

a. Excitation functions for Bk isotopes

Excitation functions for the Bk isotopes 244, 245, 246, 248m, and 250 produced from the interactions of ⁴⁰Ar with ²⁴⁸Cm are shown in Fig. III.13. The shapes of the functions are consistent with the calculated excitation energies. ²⁴⁸Bk^m has the highest production cross section, with ²⁵⁰Bk having the next highest, followed by ²⁴⁶Bk, ²⁴⁵Bk, and finally ²⁴⁴Bk.

b. Excitation functions for Cf isotopes

The excitation functions measured for the Cf isotopes 246, 248, 250, 252, 253, and 254 are shown in Fig. III.14. These are all similar in shape showing an increase by a factor of 2.5 when the bombarding energy is increased by 18 MeV, then they level off for the next 41 MeV and finally decrease by a factor of 2.5 over the next 20 MeV. The highest cross sections occur for 250 Cf which results from the transfer of only two

nucleons (two protons), the fewest possible, and the lowest cross sections occur for 254Cf which results from the transfer of six nucleons (two protons and four neutrons).

c. Excitation functions for Es isotopes

Excitation functions for the Es isotopes 250-255 are plotted in Fig. III.15. Again the shapes of the functions are consistent with the calculated excitation energies, and again the highest cross section occurs for ²⁵¹Es resulting from the transfer of the fewest number of nucleons, in this case three protons. A decrease in the production cross sections is seen as the number of nucleons transferred increases.

The fraction of the yield going to $^{250}\text{Es}g$ ($\sigma(6^+)/[\sigma(6^+) + \sigma(1^-)]$) is given in Table III.15 as a function of projectile energy for the ^{40}Ar system. These data are plotted in Fig. III.16 also. The fraction decreases with increasing projectile bombarding energy from 207 to 245 MeV, and levels off at about 0.25 over the next 41 MeV. Hence, the majority of the yield (except at the lowest bombarding energy) for this mass number is going into the low spin $^{250}\text{Es}m$ (1-). Perhaps the fraction of the yield going to $^{250}\text{Es}g$ is decreasing with increasing bombarding energy due to increased fission in $^{250}\text{Es}g$ relative to $^{250}\text{Es}m$.

Chasteler *et al.* [CHA87], for the ¹⁶O-²⁴⁹Cf system, found the fraction of the yield going to ²⁵⁰Esg decreases linearly with increasing bombarding energy. The values went from 0.41 to 0.34 when the projectile energy was increased from 1.0 to 1.4 X Coulomb barrier. For the ¹⁸O-²⁴⁹Cf system, Lee *et al.* [LEE83] reported the fraction going to ²⁵⁰Esg to be 0.28, 0.39, 0.36, and 0.27 at 1.0, 1.1, 1.3, and 1.5 X Coulomb barrier, respectively. The fraction of the yield going to ²⁵⁰Esg in the ⁴⁸Ca-²⁴⁸Cm system ¹HOF85] was reported to be 0.35 and 0.31 at 1.1 and 1.3 X Coulomb barrier, respectively. Hence, it appears that in these heavy ion-actinide systems near the Coulomb barrier, the low spin isomer receives a significant fraction of the yield.

The $^{254}\text{Es}^m$ cross section seems to fit the systematics of the other Es isotopes in the mass yield curve from the ^{40}Ar system at 1.16 X Coulomb barrier. If a significant

fraction of the yield was going to $^{254}\text{Es}g$ (7+), then the cross section for $^{254}\text{Es}m$ (2+) would appear low on the mass yield curve. Finally, since the cross sections for $^{248}\text{Bk}m$ (1-, $t_{1/2} = 23.4$ h) were measured and those of $^{248}\text{Bk}g$ (6+, $t_{1/2} = >9$ y) were not, then one must consider the fraction of yield going to the 6+ state. If one assumes that the partitioning of the yield is similar to the case of $^{250}\text{Es}m$ and $^{250}\text{Es}g$, then the fraction of the yield going to the 6+ state would be approximately 0.57, 0.47, 0.28, 0.25, and 0.26 at 207, 225, 245, 266, and 286 MeV, respectively.

d. Excitation functions for Fm isotopes

Excitation functions for isotopes of Fm are shown in Fig. III.17. The curves are all similar in shape and are consistent with their calculated positive excitation energies. The fewest number of nucleons which can be transferred to make a Fm isotope is four, and indeed, ²⁵²Fm has the highest cross section. The cross sections decrease as the number of nucleons transferred increases, with ²⁵²Fm having the largest cross section and ²⁵⁶Fm having the smallest.

e. F, and E, Values

 F_t and E_t values for the 40 Ar system are shown in Table III.13. It should be noted that the Cf excitation functions are relatively flat; therefore, the E_M values for these isotopes were difficult to locate, thus affecting the F_t values. The F_t 's range in value from 0.02 to 0.29, with average values of 0.16, 0.14, 0.18, and 0.19 for the Bk, Cf, Es, and Fm isotopes respectively. As can be seen, the average F_t 's generally increase with increasing atomic number. However, Cf does not follow this trend. Once again, no correlation was found between F_t and E_t for a given nuclide; however, in general, E_t is less than F_t indicating that the fraction of energy transferred is greater than the fraction of mass transferred.

G. 44Ca-248Cm System

Production cross sections for the measured isotopes of Bk, Cf, Es, and 14m from the interactions of 44 Ca with 248 Cm along with the statistical standard deviation, s, are

shown in Table III.5.

1. Isotopic Distributions

Isotopic distributions for Bk, Cf, Es, and Fm from the interactions of 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions with ²⁴⁸Cm are shown in Fig. III.18. The maxima of the isotopic distributions occur at ²⁴⁸Bkm, ²⁵⁰Cf, ²⁵¹Es, and less than or equal to ²⁵²Fm. As mentioned before, these are the nuclides which correspond to the isotopes requiring the transfer of the fewest number of nucleons for a given element (with the exception of ²⁴⁸Bkm). Since the cross sections for the neutron-deficient Fm isotopes were not measured for this system, the peak of the mass-yield curve for Fm cannot be located with certainty.

The isotopic distributions are symmetric and display full-widths-at-half-maximum of 3.0, 3.5, and 2.0 for Bk, Cf, and Es respectively. The width of the Fm peak could not be determined due to insufficient data on the neutron-deficient Fm isotopes.

2. Excitation Functions

a. Excitation Functions for Bk Isotopes

Excitation functions for the Bk isotopes 244, 245, 246, 248m, and 250 are plotted in Fig. III.19. 248 Bk m has the highest production cross sections, followed by 246 Bk, 250 Bk, 245 Bk and finally 244 Bk. The production of 248 Bk requires the transfer of one proton and one neutron as does the production of 250 Bk. However, the 1.4 MeV E_X value of 248 Bk makes the production of this nuclide more energetically favorable than the production of 250 Bk which has an E_X value of -4.7 MeV. 246 Bk, which requires the transfer of 4 nucleons, has cross sections which are greater at four energies than the cross sections of 250 Bk. This again can be attributed to the positive E_X value (5.5 MeV) for this reaction channel as opposed to the negative value for the 250 Bk channel. The shape of 250 Bk's excitation function is consistent with its negative E_X value of -4.7 MeV. However, the excitation function shapes of the other Bk isotopes are inconsistent with their calculated positive E_X values.

b. Excitation Functions for Cf Isotopes

Excitation functions for the Cf isotopes 246, 247, 248, 250, 252, and 253 are plotted in Fig. III.20. The isotope requiring the transfer of the fewest nucleons, 250 Cf, (two protons) has the highest production cross sections at all of the measured energies. 248 Cf has the next highest cross sections followed by 252 Cf, 247 Cf, 246 Cf, and finally 253 Cf. Both 248 Cf and 252 Cf require the transfer of four nucleons; however, 248 Cf has a positive E_X value (7.2 MeV) whereas 252 Cf has a negative value (-0.2 MeV). Five nucleons have to be transferred between the target and projectile in order to produce both 247 Cf and 253 Cf. 247 Cf has much higher cross sections than 253 Cf, as expected from their calculated E_X values of 5.9 MeV and -5.7 MeV respectively. 246 Cf, 247 Cf, and 250 Cf have excitation functions which are consistent with their calculated positive E_X values. Additionally the shapes of the excitation functions from 252 Cf and 253 Cf are consistent with their calculated negative E_X values.

c. Excitation Functions for Es Isotopes

Excitation functions of the Es isotopes 250, 251, 252, 253, and 254m are shown in Fig. III.21. The solid lined labeled 250 is the sum of the cross sections from 250m and 250g. $^{251}{\rm Es}$, the isotope requiring the transfer of the fewest nucleons, has the highest peak cross section and the highest cross sections at three energies. $^{250}{\rm Es}$ has the next highest cross sections, followed by $^{252}{\rm Es}$, $^{253}{\rm Es}$, and $^{254}{\rm Es}^m$. Hence, as the number of nucleons transferred increases, the cross sections decrease. $^{252}{\rm Es}$, $^{253}{\rm Es}$, and $^{254}{\rm Es}^m$ have excitation functions which are consistent with their negative E_X values. Both $^{251}{\rm Es}$ and $^{250}{\rm Es}$, however, have excitation functions which are inconsistent with their calculated positive excitation energies. But if one looks at the excitation functions of $^{250}{\rm Es}$ and $^{250}{\rm Es}$ separately, it appears that $^{250}{\rm Es}$ has an excitation function that is consistent with a positive E_X while $^{250}{\rm Es}$ has one that is consistent with a negative E_X . $^{250}{\rm Es}$ has its peak cross section at 275 MeV while the peak cross section of $^{250}{\rm Es}$ 8 is at 252 MeV. Table III.15 lists the fraction of the yield going to $^{250}{\rm Es}$ 8 (6+) as a

function of projectile bombarding energy for the 44 Ca system. These data are also plotted in Fig. III.22. Like the 40 Ar system, the fraction decreases with increasing projectile energy up to 275 MeV, and then stays relatively constant over the next 44 MeV. Based upon these results, the fraction of the yield going to 248 Bk8 (6⁺) can be estimated to be 0.38, 0.29, 0.15, 0.19, and >0.16 at 232, 252, 275, 297, and 319 MeV, respectively. The 254 Es m cross section does not appear to be unusually low on the mass yield curve when compared to the other Es isotopes from the 44 Ca system. Hence, it appears only a small fraction of the yield for mass number 254 is going to 254 Es 8 (7⁺).

d. Excitation Functions for Fm Isotopes

Fig. III.23 displays excitation functions for the Fm isotopes 252, 253, 254, and 256. The isotope requiring the transfer of the fewest number of nucleons, 252 Fm (four nucleons) has the highest cross sections, followed by 253 Fm (five nucleons), 254 Fm (six nucleons), and 256 Fm (eight nucleons). The shapes of these excitation functions are all consistent with their calculated positive E_X values.

e. F_1 and E_1 values

 F_t and E_t values for the various actinide products from the interactions of ^{44}Ca with ^{248}Cm are given in Table III.14. F_t values were found to be approximately 0.01-0.63 with average values of \leq 0.03, 0.22, 0.24, and 0.14 for the Bk, Cf, Es, and Fm isotopes respectively. The average F_t values increase with increasing Z going from Bk to Fm with the exception of Fm. There is no obvious trend between F_t and E_t for a given nuclide. As with the ^{12}C and ^{40}Ar systems, E_t is less than F_t indicating that the fraction of energy transferred is greater than the fraction of mass transferred. Lee et al. [LEE83] also found this to be the case for the $^{18}\text{O-}^{248}\text{Cm}$ and $^{18}\text{O-}^{249}\text{Cf}$ systems, where the average F_t value was approximately 0.5. The calculated F_t values for the ^{44}Ca system would be smaller than those from the $^{18}\text{O-}^{248}\text{Cm}$ and $^{18}\text{O-}^{249}\text{Cf}$ systems if the energy transferred is proportional to the fraction of mass transferred.

H. Comparison of the Various Heavy Ion-248Cm Systems

The maxima of the isotopic distributions for the ¹²C, ¹⁸O, ³¹P, ⁴⁰Ar, ⁴⁰Ca, ⁴⁴Ca, ⁴⁸Ca, and ¹³²Xe systems are listed in Table III.16 [HOF82] [LEY90] [LEY90A] [TÜR90], along with the neutron-to-proton ratios of the projectiles. All of the systems show maximum cross sections for Bk at mass ²⁴⁸ except for the most neutron-deficient projectile, ⁴⁰Ca, for which the maximum occurs at mass ²⁴⁷, and the neutron-rich projectiles ¹⁸O, ⁴⁸Ca, and ¹³²Xe which have maxima at about mass number ²⁴⁹. The maximum cross sections for Cf occur at mass ²⁵⁰ in all cases except for the ⁴⁰Ca system again, for which the maximum is at ²⁴⁸. The maxima of the Es curves are at mass ²⁵¹ for ¹²C, ³¹P, ⁴⁴Ca, ⁴⁰Ar, and ¹³²Xe, at ²⁵⁰ for ⁴⁰Ca, ^{≤252} for ¹⁸O, and at ²⁵² for ⁴⁸Ca. The Fm curves have their maxima around ²⁵² for the ⁴⁰Ca, ³¹P, ⁴⁴Ca, and ⁴⁰Ar systems, at ²⁵³ for the ¹²C and ¹³²Xe systems, and around ²⁵⁴ for the ¹⁸O and ⁴⁸Ca systems. Once again, it should be noted that the maximum cross sections for the ⁴⁰Ar and ⁴⁴Ca systems could not be located with certainty due to insufficient data on the neutron-deficient Fm isotopes.

The values of the FWHM's for the isotopic distributions from the ¹²C, ⁴⁰Ca, ³¹P, ⁴⁴Ca, ⁴⁰Ar, ¹⁸O, ⁴⁸Ca, and ¹³²Xe systems are shown in Table III.17. The ¹²C system has the smallest FWHM's and ¹³²Xe has the largest. However, there doesn't seem to be any pattern in the FWHM's from the other systems. Additionally, the FWHM values for these system are similar to the value of 3.0 reported previously for the ²³⁸U system [SCH82].

The final isotopic distributions for Bk from the interactions of a number of heavy ions (HI) with 248 Cm are shown in Fig. III.24. The ΔN on the x axis represents the total number of neutrons transferred. A (+) sign indicates neutrons are transferred from the projectile to the target, while a (-) sign indicates the opposite. The 31 P system has the highest production cross section for 248 Bk m by about a factor of two. This high peak cross section for 248 Bk m might be due to the odd proton in 31 P. Since an odd proton

would be unpaired in the nucleus, the probability of transferring one unpaired proton from an odd Z nucleus to the target should be greater than the probability of transferring one proton from an even Z nucleus. This argument is supported by the E_X values for Bk isotopes from these systems (Table III.6). The E_{χ} values from the $^{31}{\rm P}$ system are greater than the corresponding values from ¹²C, ⁴⁰Ar, ⁴⁴Ca, and ⁴⁸Ca systems. However, for some neutron-deficient Bk isotopes ($^{244-247}$ Bk), the 40 Ca system has E_X values that are greater than those of the ³¹P system. This should be expected since the neutron-toproton ratio of ⁴⁰Ca is only 1.00. With such a low neutron-to-proton ratio, ⁴⁰Ca should preferentially transfer protons to the target nucleus. 12C has the lowest cross sections for all of the Bk isotopes measured. This is to be expected, since the E_X values for this system are generally the most negative of the six systems listed. The neutron-deficient Bk isotope yields are essentially independent of the projectile (with the exception of the neutron-rich 48Ca curve and the energetically unfavorable ¹²C curve), whereas the neutron-rich Bk isotopes are projectile dependent. ⁴⁸Ca, the most neutron-rich projectile gives the highest yields while ¹²C and ⁴⁰Ca, the most neutron-deficient projectiles, give the lowest yields of the neutron-rich Bk isotopes.

Isotopic distributions for Cf from the reactions of several heavy ions with ²⁴⁸Cm are shown in Fig. III.25. The energetically unfavorable ¹²C and ⁴⁸Ca systems (see Table III.7) give the lowest cross sections for the neutron-deficient Cf isotopes. The cross sections for the neutron-rich Cf isotopes are all quite similar for ³¹P, ⁴⁰Ar, ⁴⁴Ca, and ⁴⁸Ca. The most neutron-deficient projectiles, ¹²C and ⁴⁰Ca, are the exceptions, with the lowest cross sections for the neutron-rich Cf isotopes.

Isotopic distributions for Es produced from the interactions of some heavy ions with ²⁴⁸Cm are shown in Fig III.26. ⁴⁰Ar consistently has the highest yields for the isotopes ²⁵⁰⁻²⁵⁴Es. The curves for the ¹²C and ³¹P systems seem to have the narrowest FWHM's.

Fig. III.27 illustrates the final isotopic distributions for Fm produced from the

interactions of several heavy ions with ²⁴⁸Cm. ⁴⁰Ar has the highest cross sections for the neutron-rich Fm isotopes, and ⁴⁸Ca has the lowest cross sections for the neutron-deficient Fm isotopes.

Based upon the detailed comparison of the interactions of numerous heavy ions with 248Cm, the following observations can be made:

- 1. In general the cross sections decrease as the number of nucleons transferred increases.
- 2. With few exceptions, the shapes of the excitation functions are consistent with the calculated excitation energies. Specifically, for reactions with negative excitation energies, the cross sections increase with increasing projectile bombarding energy until the reaction barrier is exceeded. Once the maximum is reached at some energy above the Coulomb barrier, the cross sections decrease due to fission and/or particle emission. Reactions with positive excitation energies display excitation functions that peak at or near the Coulomb barrier and decrease with increasing projectile kinetic energy.
- 3. The maxima of the isotopic distributions occur for those reaction channels which involve the apparent transfer of the fewest number of nucleons for which E_X is positive. This lends support to a binary-type reaction mechanism in which the target-like fragment has little excitation energy. However, deviations from this behavior, such as, the 40 Ca system indicate that factors other than E_X might also be influencing the final production cross section distributions.
- 4. The shapes of the isotopic distributions are all fairly symmetric about the maximum, with a FWHM of ≥ 2.5 mass units. The symmetric isotopic distributions should follow for a given Z if the production cross section for transferring + n neutrons (transferring neutrons from the projectile to the target) is equal to the production cross section for transferring n neutrons provided the E_X values for the two reaction channels are similar.
 - 5. The neutron-rich projectiles such as ⁴⁸Ca enhance the production of neutron-rich

products, while neutron-deficient projectiles such as \$^{12}{\rm C}\$ and \$^{40}{\rm Ca}\$ enhance the production of neutron-deficient products.

TABLE III.1. Cross sections from the bombardment of ^{248}Cm with ^{12}C (6 = 71.9 MeV, lab system).

,	8		7.4	 3	1.2	7.6	1 1	1.0	28	1.5	7.3	3.3	:	1	1 1	1	9.7	13	5.8	2.7	29	38	6.8	17	4.4	99	9.4
97 MeV	Cross Section		3.05	97.5	1610	295	<4.32	2.85	339	1540	24.0	0.62	<0.11	<1.23	<2.14	<0.42	37.1	25.3	2.77	0.32	0.016	0.11	4.19	2.10	0.56	0.39	0.074
>	s (%)	(o/)	1	2.1	1.2	1.4	1 1	1.0	2.0	1.5	0.9	3.5	1] [1	i i	8.5	56	8.2	2.8	27	49	4.8	10	4.1	50	8.9
90 MeV	Cross Section		<4.83	59.0	1380	328	<1.15	2.19	366	2250	38.9	0.62	<0.13	<2.71	<4.95	<0.72	64.8	93.2	4.61	0.45	0.036	0.031	5.85	4.69	0.75	0.62	0.11
>	5	(e/	0.9	1.2	1.2	1.2	1 1	1.0	2.0	£.3	3.7	3.1	1	1 1 1	1	1 1	7.9	-	25	20	21	30	3.0	3.4	2.0	46	6.3
84 MeV	Cross		3.08	41.8	1170	225	<0.76	1.97	418	4310	81.8	0.55	<0.11	<2.48	<5.04	<0.76	204	154	11.0	0.71	0.058	0.066	12.3	14.3	1.71	0.74	0.20
>	S (%)	(<u>%</u>	1	2.1	1.7	2.3	!	2.9	14	1.5	5.2	9.9	i i	1 1	!	!	5.7	8.7	11	15	25	62	4.2	8.0	2.9	30	4.3
76 MeV	Cross Section	<u>9</u>	<2.14	16.0	507	71.3	<0.69	0.91	238	5990	143	0.45	<0.42	<1.36	<2.99	<0.45	135	141	13.0	1.01	690.0	0.018	11.7	29.1	4.26	1.76	0.20
>	S	(%)	!	18	12	14	1	8.6	2.0	1.5	1.3	42	1	1	1	1	1 1	28	3.8	4.9	;	1	17	9	1.0	53	29
71 MeV	Cross Section						<1.01						٧		_	_					<0.30		4.57	28.1	6.03	1.20	0.058
	1	Nucide	BK 245	246	248 <i>m</i>	250	251	Cf 246	248	250	252	253	254	Es 249	250m	2500	251	252	253	254 <i>m</i>	255	Fm 250	252	253	254	255	256

Table III.2. Cross sections for Md and No isotopes produced from the bombardment of 248 Cm with 12 C ions.

	71 N	1eV	76 N	ИeV	84 N	ИeV
	Cross		Cross		Cross	
	section	S	section	S	section	S
Nuclide	(µb)	(%)	(µb)	(%)	(µb)	(%)
Md 254	0.053	26	ad to ma ma			
255	. 0.33	11				
256	0.049	13				
257	< 0.048	***		-		
No 254					0.050	14
255	0.022	17	0.177	4	0.077	8
256	0.203	10	0.075	15	0.024	56
257	0.033	11	0.015	19	0.010	45

TABLE III.3. Cross sections from the bombardment of 248 Cm with 31 P (4 Cm = 175.6 MeV, lab system).

leV	s (%)	2.5 1.1 50 1.3	0.3 0.9 12 14	10 8.8 6.0 7.1	64 10 12 12 46 59
239 MeV	Cross Section (µb)	<2.96 189 1090 3590 552 <7.88	44.8 824 2270 29.4 0.81	<27.8 <33.7 <4.28 59.4 21.5 2.10 0.078 0.005	0.13 1.37 0.69 0.16 0.59
> 0	s (%)	3.4 2.2 47 1.9	0.3 0.8 0.5 7.8 25	5.8 6.5 7.4 7.4	10 11 2.7 7.7 50 56
223 MeV	Cross Section (µb)	7.20 436 1830 3880 1050 <34.4	75.9 1640 3790 75.9 1.10	<18.1 <35.4 <2.85 83.9 37.8 3.11 0.097	0.17 4.68 2.65 0.56 0.43
AeV	s s	3.1 4.1 1.2 1.2 1.1	0.5 1.1 0.5 5.9 30	4.1 9.8 7.6 4.1	15 2.7 5.0 4.4 36
207 MeV	Cross Section (µb)	37.9 726 2590 8680 1330 <5.23	125 2340 5410 105 3.15	<8.73 <43.9 <2.34 179 68.9 5.23 0.17	1.79 14.9 9.97 1.59 0.79
leV	s (%)	2.2 1.1 3.5 1.0	0.5 1.2 4.2 1.5	2.8 7.3 5.0 3.8	8.1 21 0.8 2.4 51
190 MeV	Cross · Section (µb)	23.7 452 1910 7100 991 <2.57	71.9 1450 3460 73.6 3.45	<2.75 <5.84 <48.4 258 89.6 7.26 0.01	1.13 12.1 9.71 2.09 4.11 0.082
ev V	s (%)	7.1 7.1 1.6 1.6	0.5 0.6 8.5	4.3 58 11 11	15 58 2.2 5.3 58
174 MeV	Cross Section (µb)	20.3 224 829 4270 486 <6.25	31.4 892 2850 50.1 2.98	<pre><6.71 <6.71 <11.6 +) <35.1 113 27.5 4.22 0.074 <0.01</pre>	
	Nuclide	Bk 244 245 246 246 248 <i>m</i> 250	Cf 246 248 250 252 252 253	Es 249 250m (1-) 250g (6+) 251 252 253 254m 255	Fm 250 252 253 254 255 255

TABLE III.4. Cross sections from the bombardment of 248 Cm with 40 Ar (6 c = 211.2 MeV, lab system).

_		S	(%)	5.2	2:7	5.6	22	1.5	11	4.9	0.3	90	2 0	. o. 1	٠.'	4.5	23	20	32	20	- 2	د ا بر	;	0 4	1.9	5.7	5.6	, u	5 6	17
286 MeV	Cross	Section	(qrl)						100								18.5	6.48	71.3	19.5	7.5	95.0	0.00	0.03	2.28	2.29	0.73		0.42	0.03
>		S	(%)	=	1.9	21	37	<u>t.</u>	15	0.3	7 0			رن د ن	.2	3.2	24	20	. 4	000	י זע	, ,	ر. د د	9	7.2	0.9	4.2	, c	ۍ د د	12
266 MeV	Cross	Section	(qrl)	8.53	293	1115	3900	1660	240	0.69	1560	7600	0007	852	14.9	1.16	8.09	20.2	119	61.2	2.4.5		0.5	0.16	7.71	5.20	2 55	6.3	0.89	G.19
leV		S	(%)	4.7	23	ر د د	<u> </u>	1.7	25	ر در		† r	٠.٠	0.8	1.5	3.6	6.2	3 -	9.5	, ,	- 6	٥ ٥٠	7.7	18	5.1	9 6		£.3	1,5	9.9
245 MeV	Cross	Section	(qrl)	19.5	414		100	1670	350	77.9	1.7.	1630	2850	955	17.1	1.42	166	001	0.00	4 - 0	187	41.0	4.04	0.23	16.4	02.6	,	5.18	3.41	0.59
e^		S	(%)	27) c	? •		7.9	•	† 0	9.0	0.5	0.7	1.2	2.9	ď	2 •	0 0	0 2 0	7.9	1.	1.3	25	5.0	ο α) ·	4.	3.5	3.3
225 MeV	Cross	Section	(qrl)	115	0.00	5.63	0000	3000	27.5	3	0.4.0	1320	5160	755	15.7	1.22	0 99	6.00	59.R	250	179	37.7	3.63	0.11	18.4	7 0 0	0.01	6.58	4.43	0.46
eV		Ú,	(%)	Č	, r	C. 4		0.0	3.9 9.0	(٥.٥	6.0	0.7	6.0	8	3.8	C	3 .	5) ·	15	7.9	6.1	6.	4	ď	9 6	y.	. 8.	23	4.9
207 MeV	Cross	Section	(qrl)	•	4.20	84.8	342	907	311 64.8		19.2	449	2100	301	4.58	0.46		1.) 23.1	(6+) 30.9	112	76.3	17.5	1.19	0.07	7 40	D	6.58	6.14	2.40	0.34
			Nuclide		BK 244	245	246	248m	250 251		C! 246	248	250	252	253	254		Es 250m (1-)	250g (251	252	253	254m	255	מינים בי	7C7 III.	253	254	255	256

TABLE III.5. Cross sections from the bombardment of 248 Cm with 44 Ca (6 = 235.3 MeV, lab system).

319 MeV			%	16	14	25	25	7.0	1	10	1	4	8.0	12	28	:	•	1	4	27	34	د .	15	!	25	14	20		40
319	Cross	Section	<u>£</u>	21.2	322	972	2710	1380	<209	19.3	<142	647	2200	231		<0.14	1	<48.4	9.10	41.6	15.3	2.06	0.15	<5.90	3.60	1.17	0.26	<0.16	0.039
ev		s	(%)	15	8.4	12	27	10	1	10	4.9	8.5	5.0	12	20	17	1	56	9.0	18	24	15	15	1	6.0	5.8	14	1	8.3
297 MeV	Cross	Section	(qrl)	29.0	503	1360	3350	1050	<76.7	126	226	1808	5340	1410	7.56	0.79	<6.92	122	28.5	160	6.69	12.0	0.74	<1.95	7.90	6.24	1.38	<0.28	0.11
275 MeV		S	(%)	6.0	=	4.4	14	7.0	1 6 5	7.0	9.1	5.0	4.0	10	=	:	4.9	25	17	12	21	4	15	;	9.1	8.7	3.4	12	10
275	Cross	Section	(qrl)	27.2	391	1220	2990	828	<400	131	266	2380	5490	1710	8.46	<0.60	38.1	176	32.0	250	118	9.19	0.86	<1.28	13.9	13.2	3.26	3.09	0.18
leV		S	(%)	17	6.4	10	17	0.9	32	8.0	18	7.0	9.0	10	=	84	15	22	7.5	22	13	10	17		8.0	12	9.0	30	30
252 MeV	Cross	Section	(qnl)	18.7	315	870	2080	513	395	113	254	1630	7040	856	5.12	0.20	49.1	87.3	35.2	127	45.3	9.91	99.0	<3.90	20.9	13.0	4.53	0.39	0.34
le V		S	(%)	6.8	7.9	8.1	6.8	5.5	34	5.2	9.6	7.8	28	24	4	62	10	25	4.9	56	15	-	20	•	8.0	16	=	1	15
232 MeV	Cross	Section	(Qrl)							41.5	168	580	1690	368	1.17	0.042	34.4	1	+ 0		18.4	10.3	0.27	<0.46	8.13	3.32	1.53	<0.019	0.19
232 MeV 252 MeV			Nuclide	RIV 244	245	246	248m	250	251	C1 246		248	250	252	253	254	Fs 249	250m	2500 (6	251	252	253	254m	Fm 250	252	253	254	255	256

Table III.6. Excitation energies, E_{χ} , for Bk isotopes produced from the interactions of various heavy ions with ²⁴⁸Cm.

Product Nuclide	12 _C	31p	40 _{Ar}	40 _{Ca}	⁴⁴ Ca	48Ca
Bk 244	-19.2	7.6	0.1	17.2	6.8	-16.4
245	-12.2	11.3	3.0	16.5	8.8	-10.9
246	- 9.3	8.6	1.3	11.9	5.5	- 9.6
247	- 3.5	10.8	2.9	11.1	5.4	- 4.4
248	- 3.5	6.4	- 0.2	5.8	1.4	- 5.0
249	- 0.7	6.0	0.2	4.2	0.3	- 3.4
250	- 7.6	0.0	- 3.2	- 4.3	- 4.7	- 7.1
251	- 9.9	- 2.3	- 3.2	-10.2	- 6.0	- 7.7
	······································					

Table III.7. Excitation energies, E_x , for Cf isotopes produced from the interactions of various heavy ions with 248 Cm.

Product Nuclide	12 _C	31p	40 _{Ar}	40 _{Ca}	⁴⁴ Ca	48 _{Ca}
Cf 246	-15.3	6.9	1.7	19.9	8.4	-14.3
247	-11.6	6.3	1.4	16.1	5.9	-13.3
248	- 3.0	8.8	4.2	16.7	7.2	- 8.0
249	- 0.9	6.7	2.1	12.1	4.1	- 6.8
250	4.9	7.2	4.2	11.8	4.7	- 2.8
251	2.8	2.5	0.9	4.7	0.1	- 6.1
252	6.9	0.6	2.4	1.6	- 0.2	- 5.8
253	- 7.6	- 8.1	- 3.1	- 9.2	- 5.7	- 9.6
254	-12.8	-13.9	- 4.5	-16.4	- 6.6	- 9.6

Table III.8. Excitation energies, E_x , for Es isotopes produced from the interactions of various heavy ions with 248 Cm.

Product Nuclide	12 _C	31p	40 _{Ar}	40 _{Ca}	⁴⁴ Ca	48 _{Ca}
Es 249	-11.7	8.6	- 1.3	15.8	2.3	-16.8
250	- 6.9	7.1	- 1.4	13.4	1.2	-14.8
251	0.4	9.7	1.4	13.7	2.5	-10.6
252	1.3	6.1	0.0	8.3	- 0.5	-12.2
253	5.3	5.6	2.1	5.6	- 0.3	- 9.8
254	2.7	- 0.8	- 1.7	- 2.3	- 3.7	-11.8
255	2.7	- 2.5	- 2.3	- 8.2	- 4.1	-11.2
256	-14.4	-14.5	- 7.8	-19.4	- 9.8	-14.5

Table III.9. Excitation energies, E_x , for Fm isotopes produced from the interactions of various heavy ions with 248 Cm.

Product Nuclide	12 _C	31 _P	40 _{Ar}	40 _{Ca}	44Ca	48 _{Ca}
Fm 251	- 5.2	4.0	- 3.1	16.1	0.4	-18.7
252	3.0	7.2	1.4	18.7	3.4	-13.1
253	5.7	5.6	0.9	13.9	1.3	-14.4
254	12.5	6.2	4.2	13.1	3.2	10.5
255	15.5	1.9	1.2	6.5	0.0	-11.6
256	22.4	0.9	2.7	3.8	1.7	- 9.5

Table III.10. Excitation energies, E_{χ} , for Md isotopes produced from the interactions of various heavy ions with ²⁴⁸Cm.

Product Nuclide	12 _C	31p	40 _{Ar}	⁴⁰ Ca	44 _{Ca}	⁴⁸ Ca
Md 253		6.9	- 6.0	17.2	- 2,4	-23.1
254		6.3	- 5.3	14.2	- 2.8	-22.8
255	4.5	8.3	- 1.8	14.2	0.0	-18.3
256	9.6	4.6	- 2.8	9.2	- 1.2	-18.9
257	18.8	5.5	- 1.0	7.4	1.1	-14.4
258	17.6	0.0	- 3.3	- 0.1	- 2.5	-15.3
259	***	- 0.9	- 3.1	- 5.5	- 2.8	-12.9

Table III.11. Calculated fraction of energy transferred, F_t , from the ¹²C projectile to the ²⁴⁸Cm target. E_M is the maximum in the experimental excitation functions and E_t is the fraction of energy transferred assuming energy transfer is proportional to mass transfer. $V_c \cong 72$ MeV in the lab frame. An * indicates that $(E_{f,n} - E_x) < 0$. See explanation in text. Data taken from [LEY90A].

Nuclide	E_{M} (MeV)	F_t	E_{t}	Nuclide	E _M (Me)		E_{t}
Bk 245	>97	<0.75	0.42	Es 251	81	0.55	0.25
246	>97	< 0.62	0.33	252	72		0.33
248m	>97	< 0.38	0.17	253	72		0.42
250	90	0.72	0.17	254m	76	0.66	0.50
•				255	76	0.67	0.58
Cf 246	>97	< 0.82	0.50	Fm 252	80	0.34	0.33
248	84	0.68	0.33	253	76	*	0.42
250	76	0.08	0.17	254	72	*	0.50
252	72	*	0.33	255	76	*	0.58
253	>90	< 0.71	0.42	256	80	*	0.6

Table III.12. Calculated fraction of energy transferred, F_t , from the $^{31}\mathrm{P}$ projectile to the $^{248}\mathrm{Cm}$ target. E_M is the maximum in the experimental excitation functions and E_t is the fraction of energy transferred assuming energy transfer is proportional to mass transfer. $V_c \cong 176~\mathrm{MeV}$ in the lab frame. An * indicates that $(E_{f,n} - E_x) < 0$. See explanation in text. Taken from [LEY90A].

Nuclide	E_{M}	F_{t}	E_t	Nuclide	E_{M}	F_{t}	E_t
1	(MeV)				(MeV)		•
Bk 244	200	*	0.19	Es 251	195	*	0.10
245	207	*	0.16	252	195	*	0.13
246	207	*	0.13	253	190	*	0.16
248m	200	*	0.06	254m	195	0.32	0.19
250	207	0.17	0.06	255	220	0.18	0.23
Cf 246	207	*	0.19	Fm 250	202	0.13	0.19
248	207	*	0.13	252	201	*	0.13
250	207	*	0.06	253	200	0.01	0.16
252	207	0.15	0.13	254	196	*	0.19
253	195	0.69	0.16	255	190	0.26	0.23
				256	195	0.25	0.26

TABLE III.13. Calculated fraction of energy transferred, F_t , from 40 Ar projectile to 248 Cm target. E_M is the maximum in the experimental excitation functions, and E_t is the fraction of energy transferred assuming it is apportioned according to the mass fraction transferred. $V_c \cong 211$ MeV in the lab frame. Data taken from [LEY90].

Nuclide	E_{M}	F_t	E_t	Nuclide	E_{M}	F_t	E_t
A	(MeV)		•	*	(MeV)	•	•
Bk 244	240	0.21	0.15	Es 250	243	0.21	0.10
245	245	0.10	0.13	251	245	0.12	0.08
246	242	0.15	0.10	252	240	0.18	0.10
248m	250	0.15	0.05	253	237	0.13	0.13
250	255	0.19	0.05	254m	237	0.26	0.15
				255	250	0.20	0.18
Cf 246	245	0.10	0.15	Fm 252	235	0.18	0.10
248	255	0.02	0.10	253	228	0.28	0.13
250	266	0.02	0.05	254	220	0.17	0.15
252	245	0.08	0.10	255	230	0.21	0.18
253	245	0.23	0.13	256	245	0.09	0.20
254	245	0.29	0.15				

TABLE III.14. Calculated fraction of energy transferred, F_t , from ⁴⁴Ca projectile to ²⁴⁸Cm target. E_M is the maximum in the experimental excitation functions, and E_t is the fraction of energy transferred assuming it is apportioned according to the mass fraction transferred. $V_c \cong 237$ MeV in the lab frame. An * indicates that $(E_{f,n} - E_x) < 0$. See explanation in text. Data taken from [LEY90].

Nuclide	E_{M}	F_t	E_t	Nuclide	E_{M}	F_t	E_t		
	(MeV)			(MeV)					
Bk 244	285	*	0.14	Es 250	275	0.11	0.09		
245	285	*	0.11	251	275	0.08	0.07		
246	282	0.01	0.09	252	273	0.16	0.09		
248m	286	0.08	0.05	253	266	0.20	0.11		
250	≥319	≤0.01	0.05	254m	273	0.24	0.14		
				255	252	0.63	0.16		
Cf 246	275	*	0.14	Fm 252	252	0.16	0.09		
247	275	*	0.11	253	264	0.16	0.11		
248	275	*	0.09	254	256	0.13	0.14		
250	260	0.21	0.05	256	252	0.26	0.18		
252	280	0.13	0.09						
253	270	0.32	0.11						
254	292	0.21	0.14		4				

Table III.15. Fraction of yield going to ²⁵⁰Esg (6⁺) from the ⁴⁰Ar and ⁴⁴Ca systems.

			40_{Ar}						
	Energy (MeV)								
	207	225	245 	266	286				
σ(6 ⁺)	0.57	0.45	0.00	0.0"	0.00				
$\sigma(6^+) + \sigma(1^-)$	0.57	0.47	0.28	0.25	0.26				
			⁴⁴ Ca						
	Energy (MeV)								
	232	252	275	297	319				
σ(6 ⁺)	0.38	0.29	0.15	0.19	>0.16				
$\sigma(6^+) + \sigma(1^-)$	0.50	0.27	0.15	0.17	, 0,10				

Table III.16. Neutron-to-proton ratios, N/Z, and peaks of isotopic distributions for the interactions of 84-MeV (1.16 X Coulomb barrier ¹²C ions, 253-MeV (1.08 X Coulomb barrier) ⁴⁰Ca ions, 207-MeV (1.17 X Coulomb barrier) ³¹P ions, 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions, 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions, 113-MeV (1.20 X Coulomb barrier) ¹⁸O ions, 280-MeV (1.19 X Coulomb barrier) ⁴⁸Ca ions, and 805-MeV (1.13 X Coulomb barrier) ¹³²Xe ions with ²⁴⁸Cm targets. Data for ^{40,48}Ca taken from [HOF82], ³¹P taken from [LEY90A], ⁴⁴Ca taken from [TUR90], ⁴⁰Ar taken from [LEY90], ¹⁸O taken from [LEE83], and ¹³²Xe taken from [WEL87].

Projectile	N/Z	Peak of Isotopic Distributions				
		Bk	Cf	Es	Fm	
		(Mass Number)				
12 _C	1.00	248	250	251	253	
40Ca	1.00	247	248	250	251-252	
31p	1.07	248	250	251	252	
44Ca	1.20	248	250	251	≤ 252	
40_{Ar}	1.22	248	250	251	≤ 252	
$18_{\mathbf{O}}$	1.25	249-250	250	≤252	254	
⁴⁸ Ca	1.40	249	250	252	254-255	
132Xe	1.44	249-250	250	251	253	

Table III.17. Neutron-to-proton ratios, N/Z, and full-widths-at-half-maximum of isotopic distributions from the interactions of 84-MeV (1.16 X Coulomb barrier ¹²C ions, 253-MeV (1.08 X Coulomb barrier) ⁴⁰Ca ions, 207-MeV (1.17 X Coulomb barrier) ³¹P ions, 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions, 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions, 113-MeV (1.20 X Coulomb barrier) ¹⁸O ions, 280-MeV (1.19 X Coulomb barrier) ⁴⁸Ca ions, and 805-MeV (1.13 X Coulomb barrier) ¹³²Xe ions with ²⁴⁸Cm targets. Data for ^{40,48}Ca taken from [HOF82], ³¹P taken from [LEY90A], ⁴⁴Ca taken from [TUR90], ⁴⁰Ar taken from [LEY90], ¹⁸O taken from [LEE83], and ¹³²Xe taken from [WEL87].

Projectile	N/Z	Full-Width-At-Half-Maximum				
		Bk	Cf	Es	Fm	
		(Mass Number)				
12 _C	1.00	2.1	1.8	<1.5	1.8	
40Ca	1.00	2.3	2.5	2.3	2.6	
31p	1.07	2.5	2.5	<1.5	2.3	
44Ca	1.20	3.0	3.5	2.0		
40_{Ar}	1.22	3.3	2.8	1.5	~	
18 _O	1.25	2.2	2.8		3.0	
⁴⁸ Ca	1.40	≅3.8	2.0	**=	2.0	
132Xe	1.44	≅3.8	2.0	2.2	≅ 2.7	

Figure Captions

- Figure III.1. Isotopic distributions for Bk, Cf, Es, and Fm from the reactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm.
- Figure III.2. Isotopic distributions for Bk, Cf, Es, and Fm from the reactions of 71-MeV (0.98 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm.
- Figure III.3. Excitation functions for isotopes of Bk from the interactions of ¹²C with ²⁴⁸Cm.
- Figure III.4. Excitation functions for isotopes of Cf from the interactions of ¹²C with ²⁴⁸Cm.
- Figure III.5. Excitation functions for isotopes of Es from the interactions of 12_{C} with 248_{Cm}
- Figure III.6. Excitation functions for isotopes of Fm from the interactions of ¹²C with ²⁴⁸Cm.
- Figure III.7. Isotopic distributions for Bk, Cf, Es, and Fm from the reactions of 207-MeV (1.17 X Coulomb barrier) ³¹P ions with ²⁴⁸Cm. Taken from [LEY90A].
- Figure III.8. Excitation functions for isotopes of Bk from the interactions of ³¹P with ²⁴⁸Cm. Taken from [LEY90A].
- Figure III.9. Excitation functions for isotopes of Cf from the interactions of ³¹P with ²⁴⁸Cm. Taken from [LEY90A].
- Figure III.10. Excitation functions for isotopes of Es from the interactions of ³¹P with ²⁴⁸Cm. Taken from [LEY90A].
- Figure III.11. Excitation functions for isotopes of Fm from the interactions of ³¹P with ²⁴⁸Cm. Taken from [LEY90A].
- Figure III.12. Isotopic distributions for Bk, Cf, Es, and Fm from the interactions of 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.13. Excitation functions for isotopes of Bk from the interactions of ⁴⁰Ar with ²⁴⁸Cm. Taken from [LEY90].

- Figure III.14. Excitation functions for isotopes of Cf from the interactions of ⁴⁰Ar with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.15. Excitation functions for isotopes of Es from the interactions of ⁴⁰Ar with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.16. Fraction of yield going to $^{250}\text{Es}g$ ($\sigma(6^+)/[\sigma(6^+) + \sigma(1^-)]$) as a function of energy for the ^{40}Ar - ^{248}Cm system.
- Figure III.17. Excitation functions for isotopes of Fm from the interactions of 40 Ar with 248 Cm. Taken from [LEY90].
- Figure III.18. Isotopic distributions for Bk, Cf, Es, and Fm from the interactions of 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.19. Excitation functions for isotopes of Bk from the interactions of ⁴⁴Ca with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.20. Excitation functions for isotopes of Cf from the interactions of ⁴⁴Ca with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.21. Excitation functions for isotopes of Es from the interactions of ⁴⁴Ca with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.22. Fraction of yield going to $^{250}\text{Es}g$ ($\sigma(6^+)/[\sigma(6^+) + \sigma(1^-)]$) as a function of energy for the $^{44}\text{Ca}-^{248}\text{Cm}$ system.
- Figure III.23. Excitation functions for isotopes of Fm from the interactions of ⁴⁴Ca with ²⁴⁸Cm. Taken from [LEY90].
- Figure III.24. Isotopic distributions for Bk produced in the interactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions, 207-MeV (1.17 X Coulomb barrier) ³¹P ions, 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions, 253-MeV (1.08 X Coulomb barrier) ⁴⁰Ca ions, 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions, and 280-MeV (1.19 X Coulomb barrier) ⁴⁸Ca ions with ²⁴⁸Cm. ΔN is the total number of neutrons transferred. A (+) sign indicates that neutrons are transferred from the projectile to the target, while a (-) sign indicates neutron transfer in the opposite direction.

Figure III.25. Isotopic distributions for Cf produced in the interactions of 84-MeV (1.16 X Coulomb barrier) 12 C ions, 207-MeV (1.17 X Coulomb barrier) 31 P ions, 245-MeV (1.16 X Coulomb barrier) 40 Ar ions, 253-MeV (1.08 X Coulomb barrier) 40 Ca ions, 275-MeV (1.16 X Coulomb barrier) 44 Ca ions, and 280-MeV (1.19 X Coulomb barrier) 48 Ca ions with 248 Cm. ΔN is the total number of neutrons transferred. A (+) sign indicates that neutrons are transferred from the projectile to the target, while a (-) sign indicates neutron transfer in the opposite direction.

Figure III.26. Isotopic distributions for Es produced in the interactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions, 207-MeV (1.17 X Coulomb barrier) ³¹P ions, 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions, 253-MeV (1.08 X Coulomb barrier) ⁴⁰Ca ions, 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions, and 280-MeV (1.19 X Coulomb barrier) ⁴⁸Ca ions with ²⁴⁸Cm. ΔN is the total number of neutrons transferred. A (+) sign indicates that neutrons are transferred from the projectile to the target, while a (-) sign indicates neutron transfer in the opposite direction.

Figure III.27. Isotopic distributions for Fm produced in the interactions of 84-MeV (1.16 X Coulomb barrier) 12 C ions, 207-MeV (1.17 X Coulomb barrier) 31 P ions, 245-MeV (1.16 X Coulomb barrier) 40 Ar ions, 253-MeV (1.08 X Coulomb barrier) 40 Ca ions, 275-MeV (1.16 X Coulomb barrier) 44 Ca ions, and 280-MeV (1.19 X Coulomb barrier) 48 Ca ions with 248 Cm. ΔN is the total number of neutrons transferred. A (+) sign indicates that neutrons are transferred from the projectile to the target, while a (-) sign indicates neutron transfer in the opposite direction.

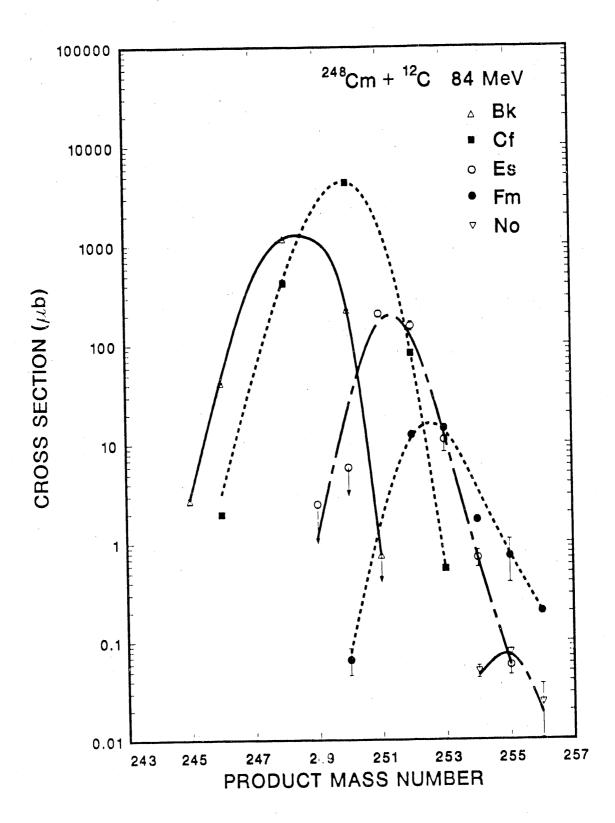


Fig. III.1

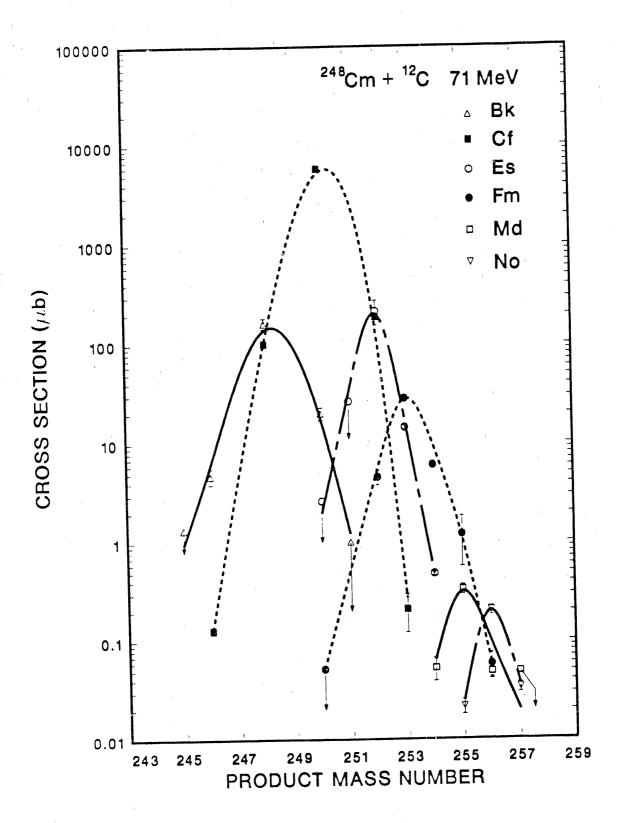


Fig. III.2

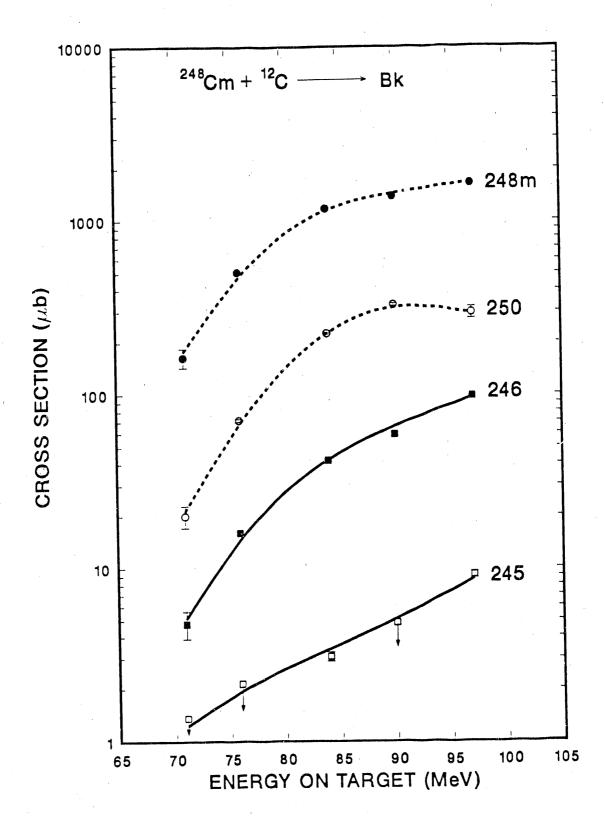


Fig. III.3

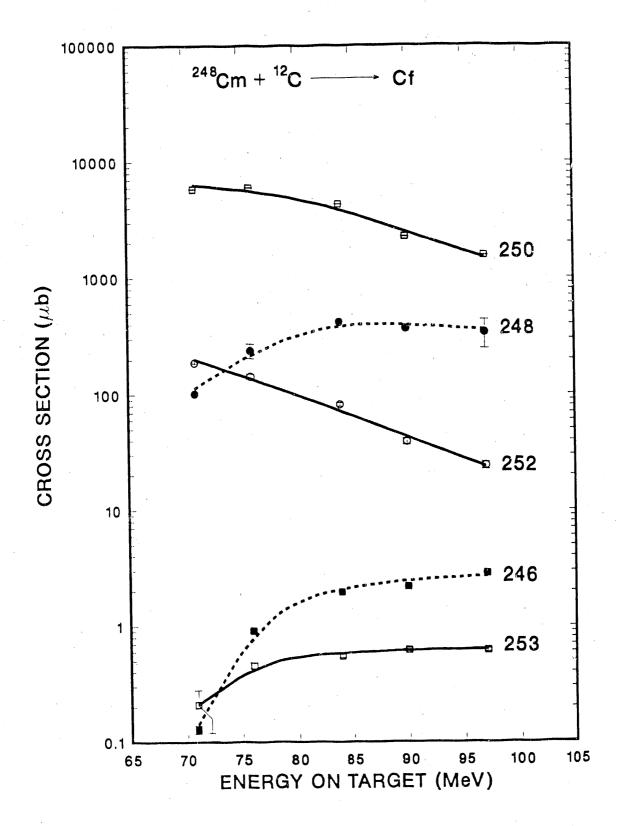


Fig. III.4

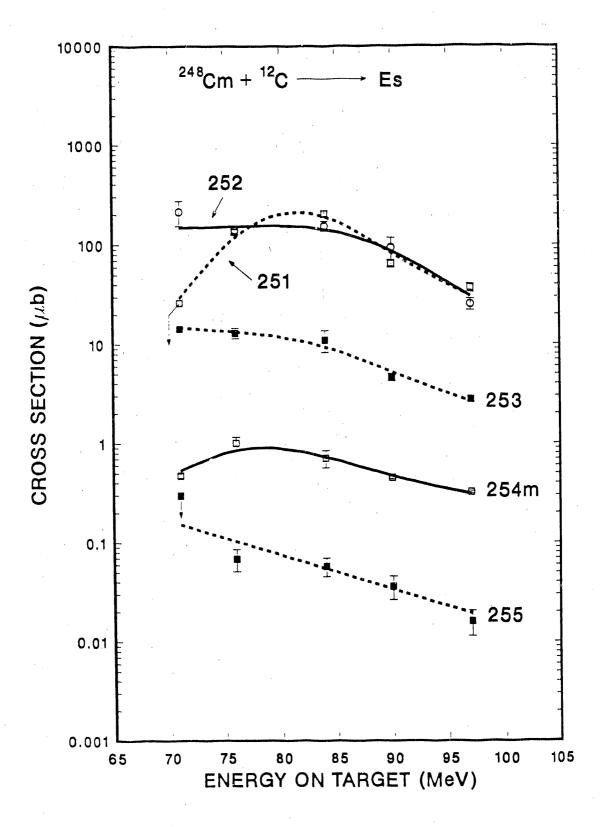


Fig. III.5

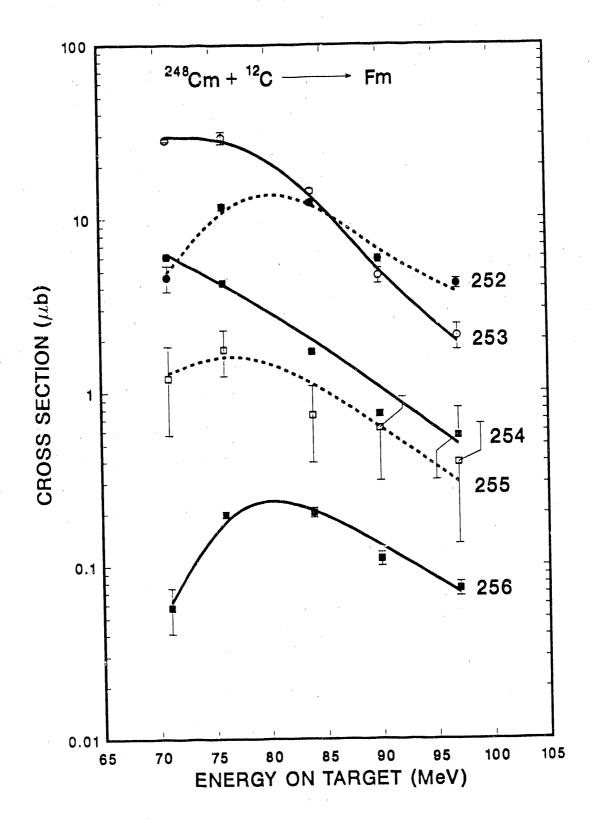


Fig. III.6

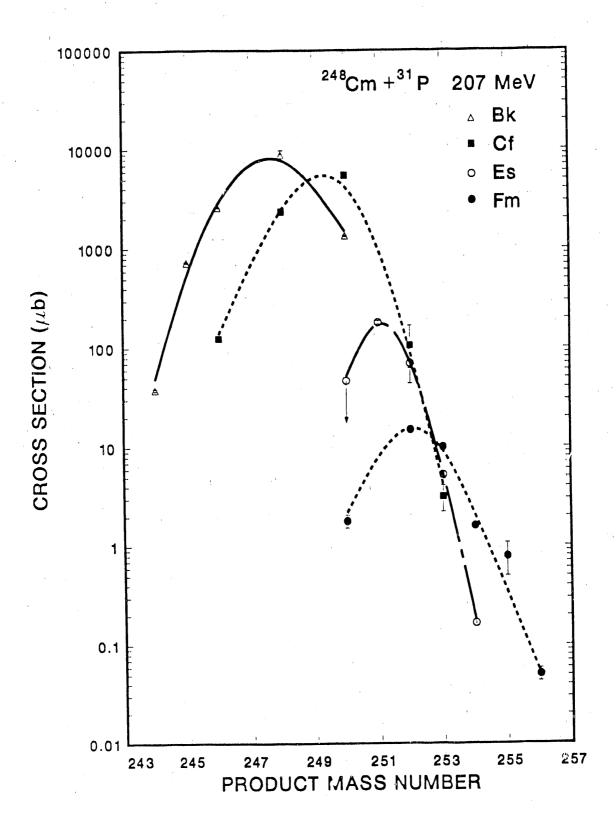


Fig. III.7

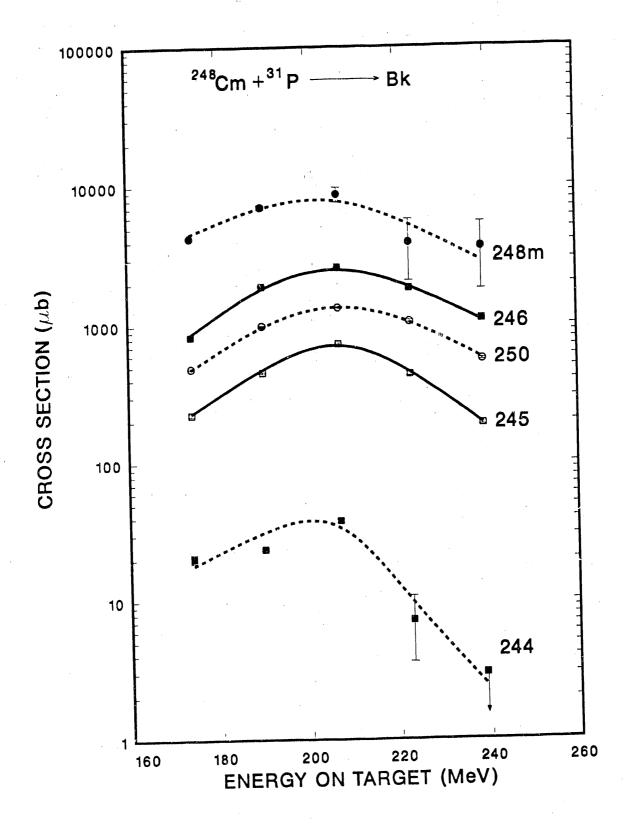


Fig. III.8

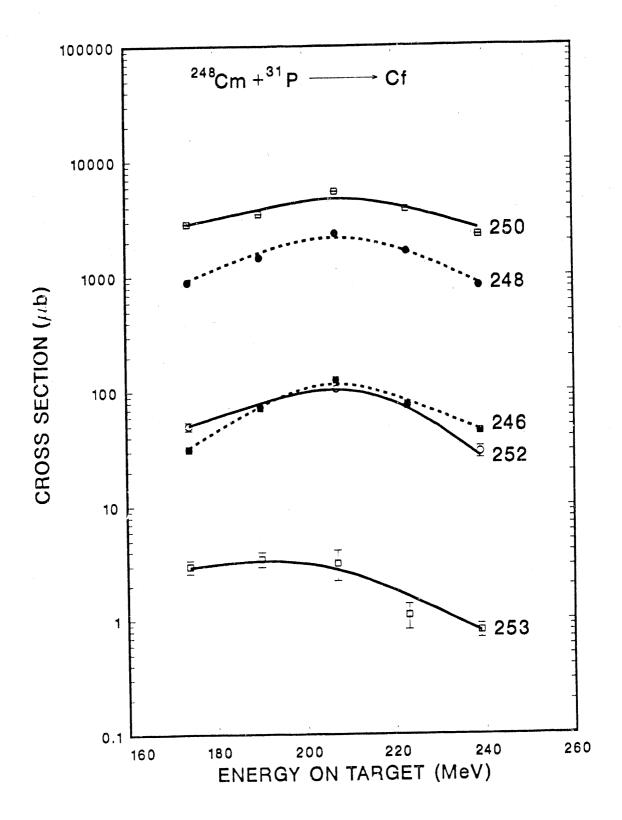


Fig. III.9

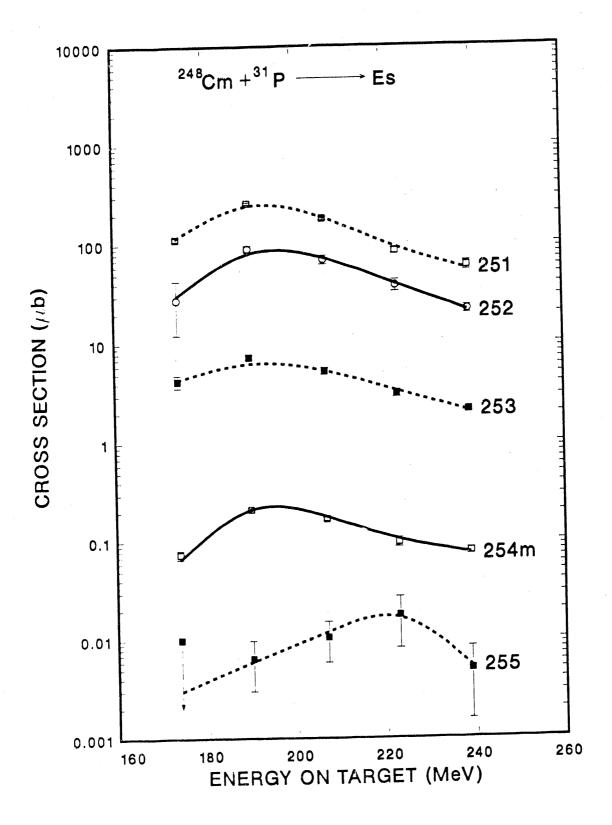


Fig. III.10

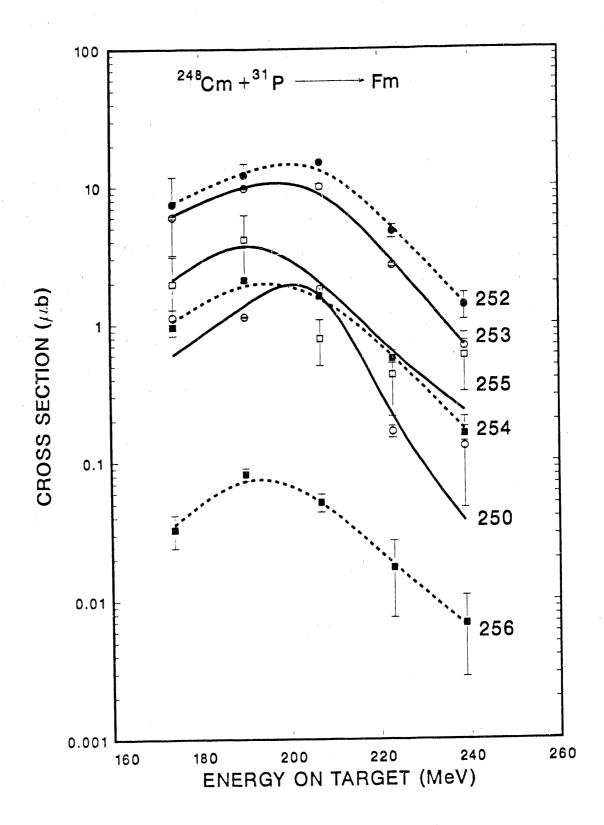


Fig. III.11

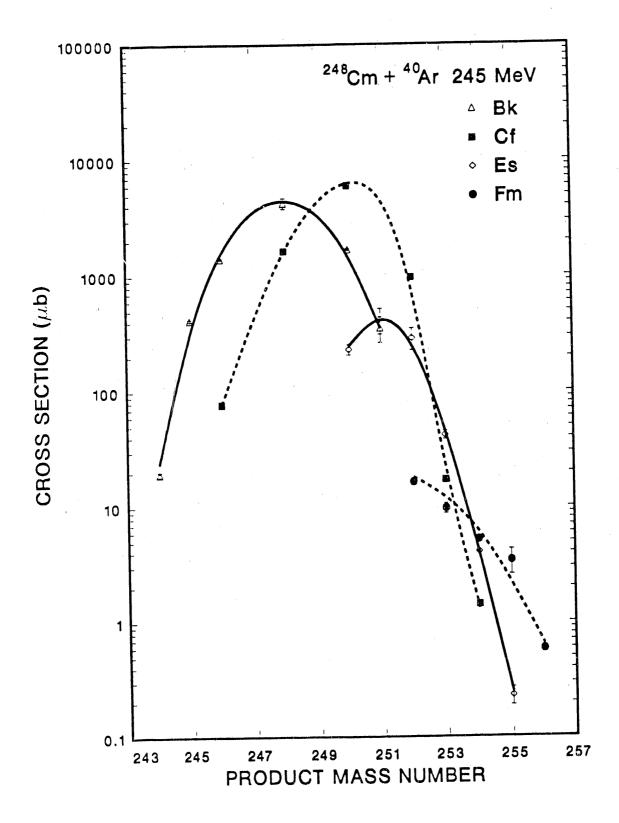


Fig. III.12

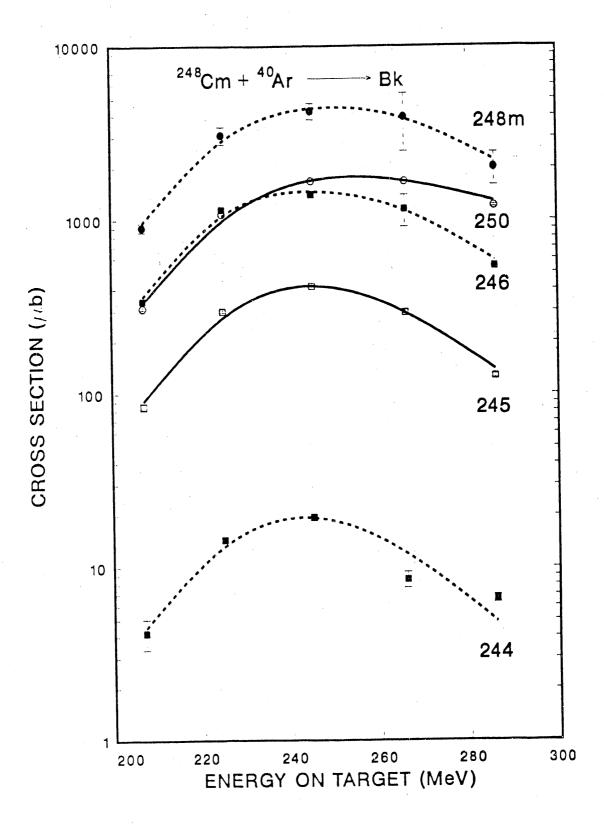


Fig. III.13

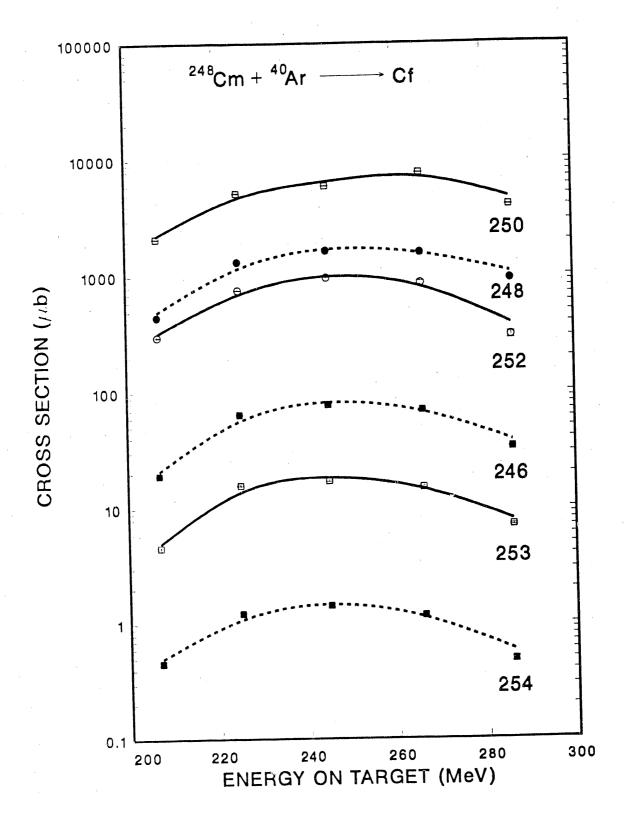


Fig. III.14

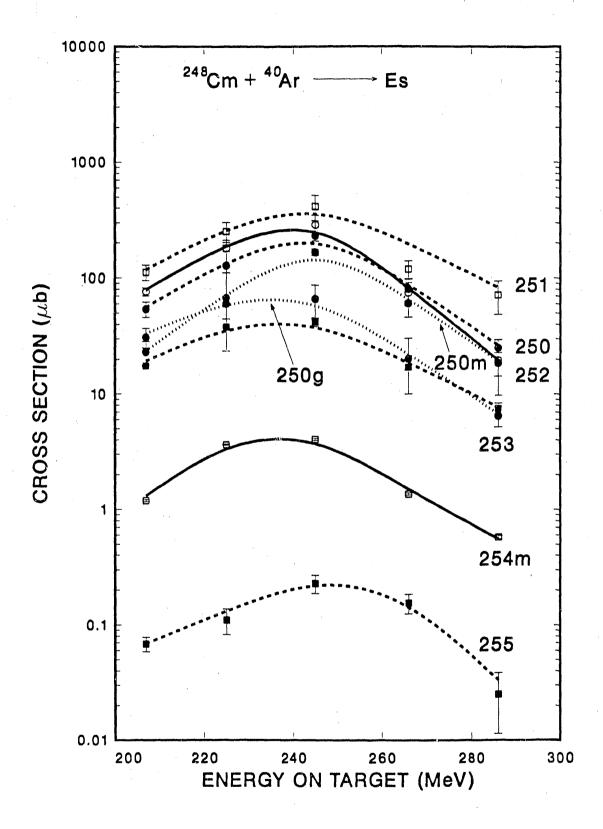


Fig. III.15

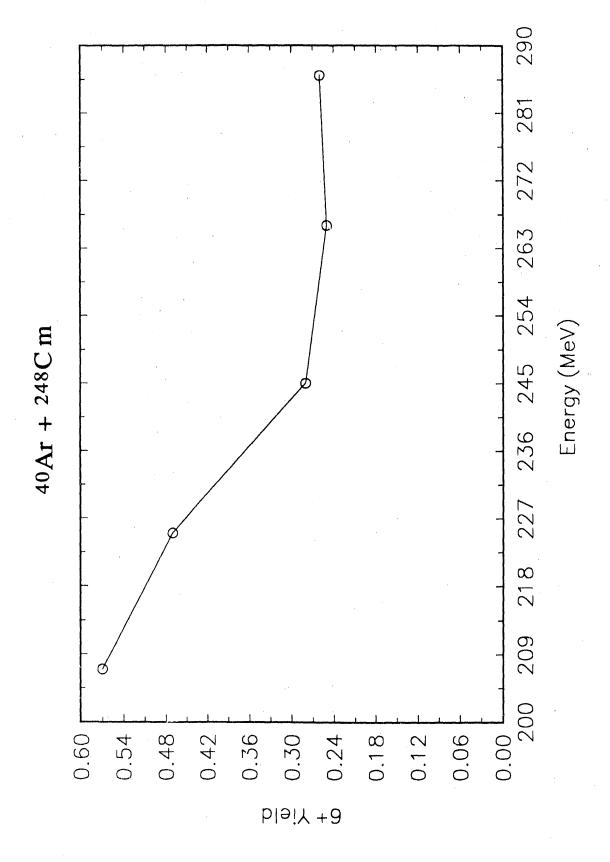


Fig. III.16

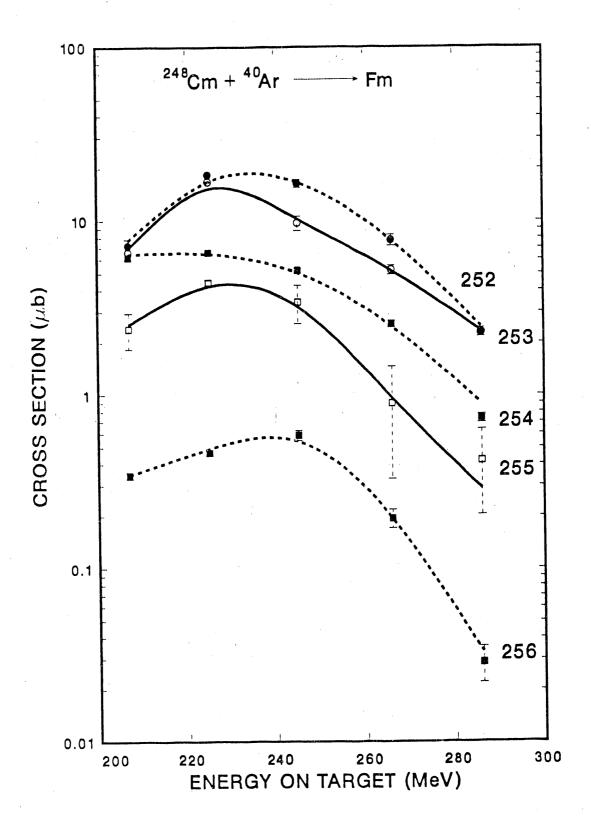


Fig. III.17

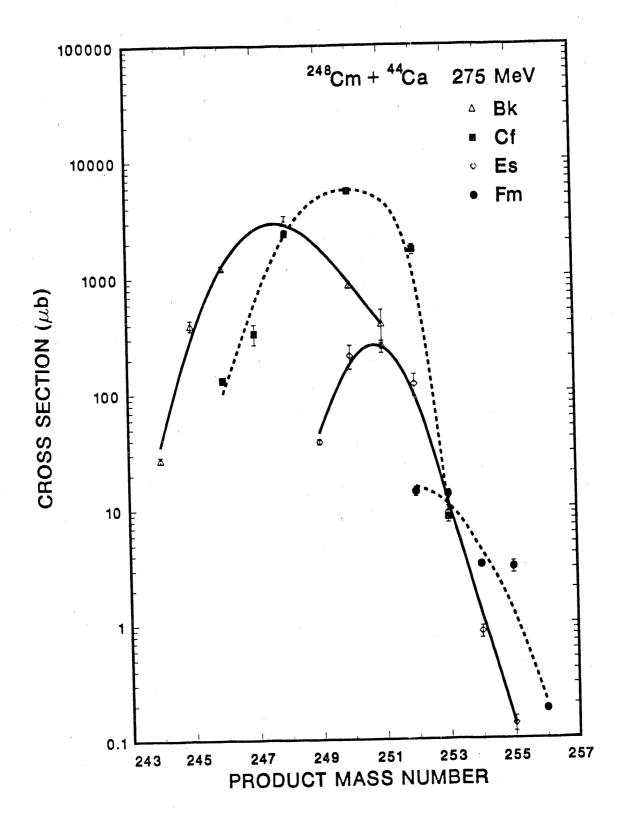


Fig. III.18

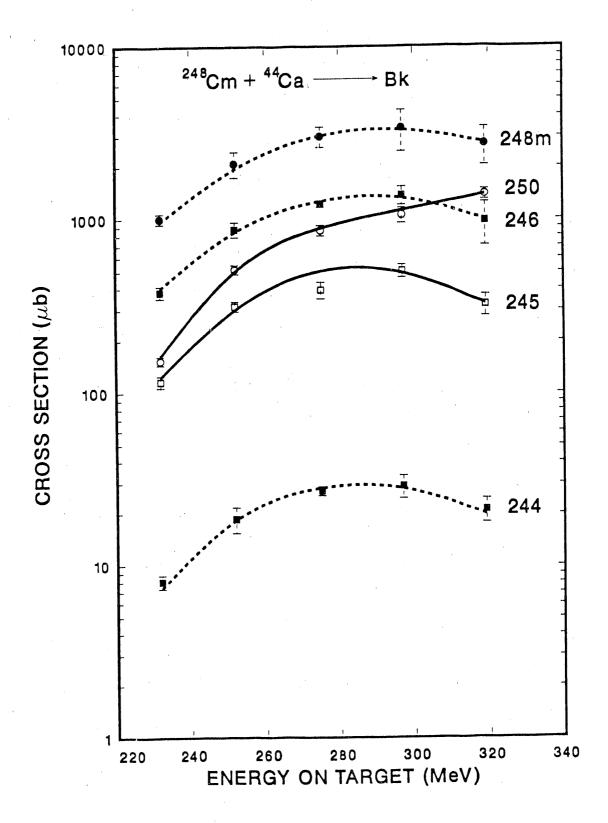


Fig. III.19

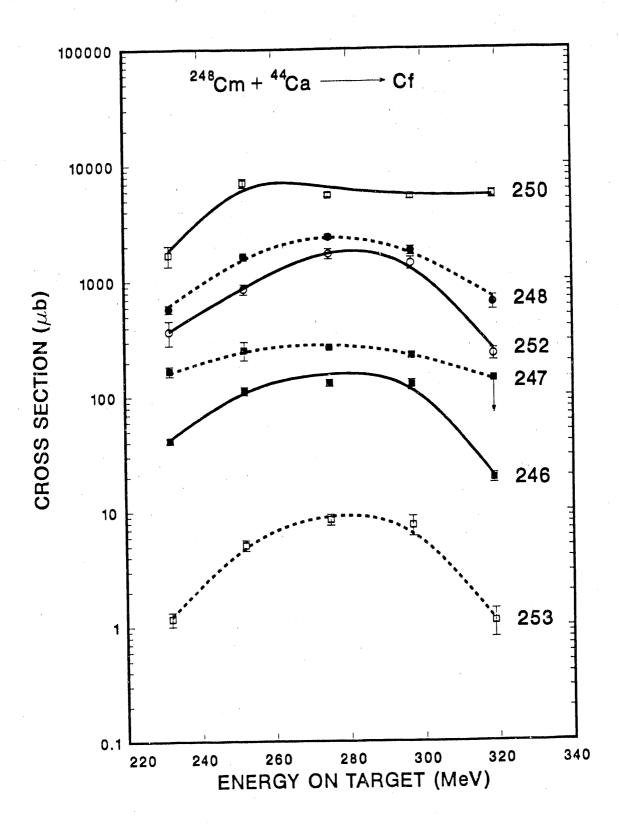


Fig. III.20

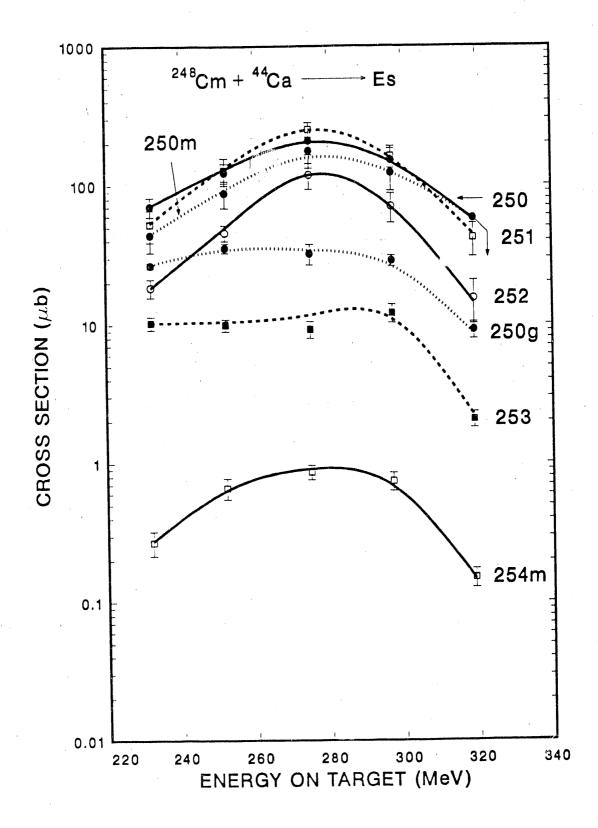


Fig. III.21

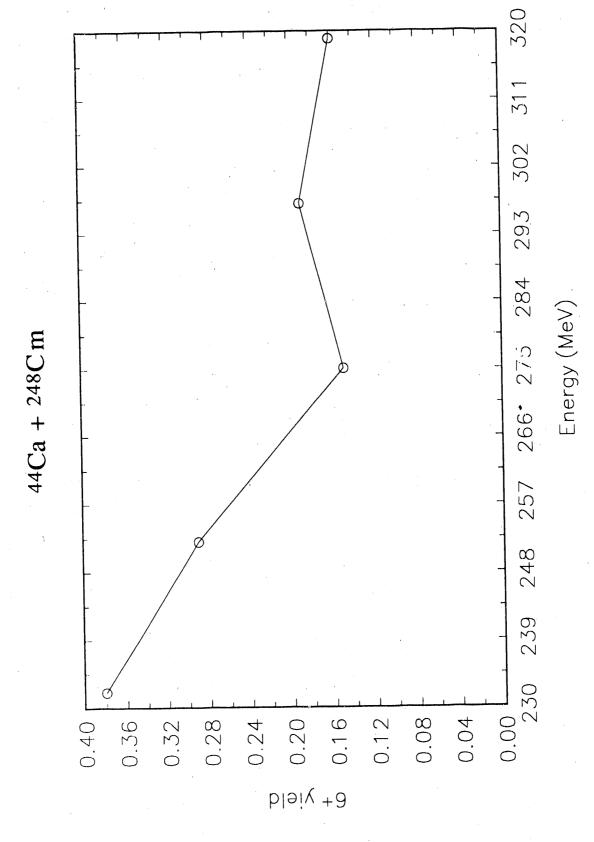


Fig. III.22

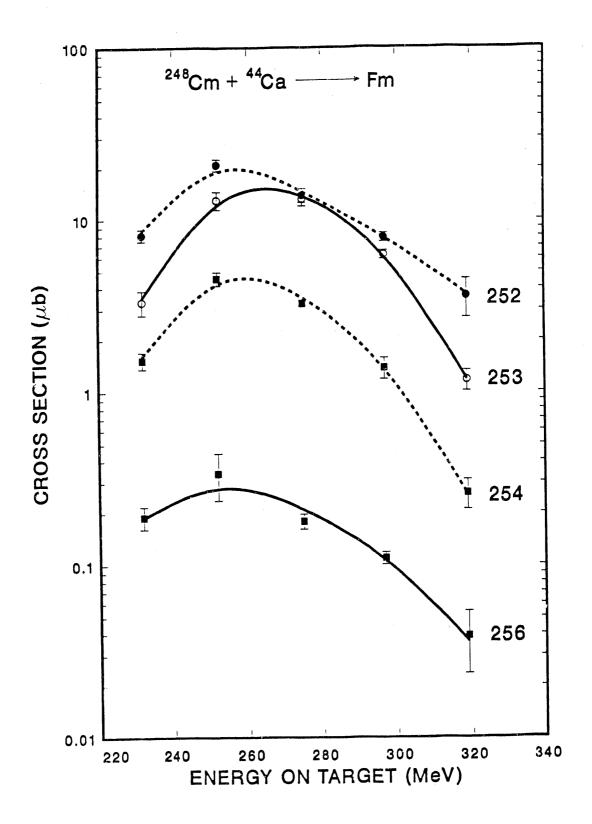


Fig. III.23

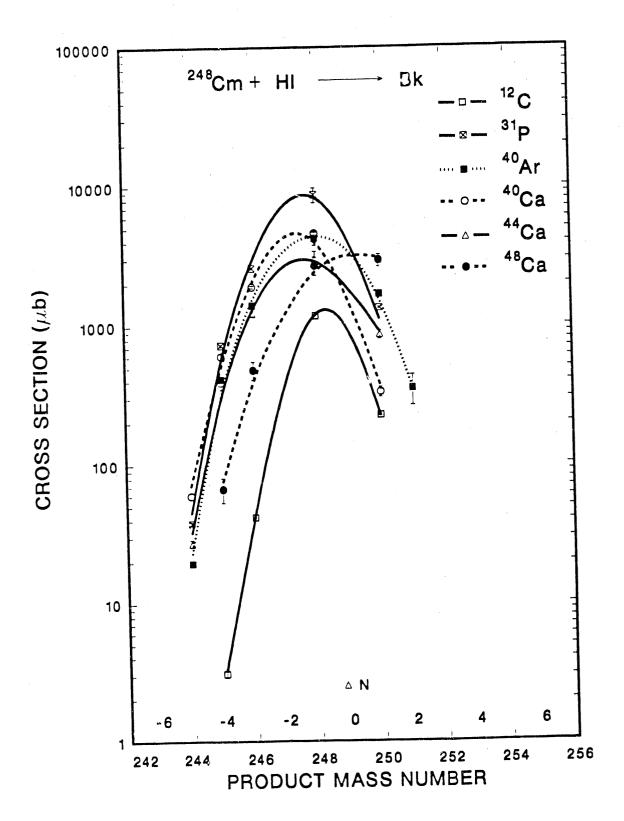


Fig. III.24

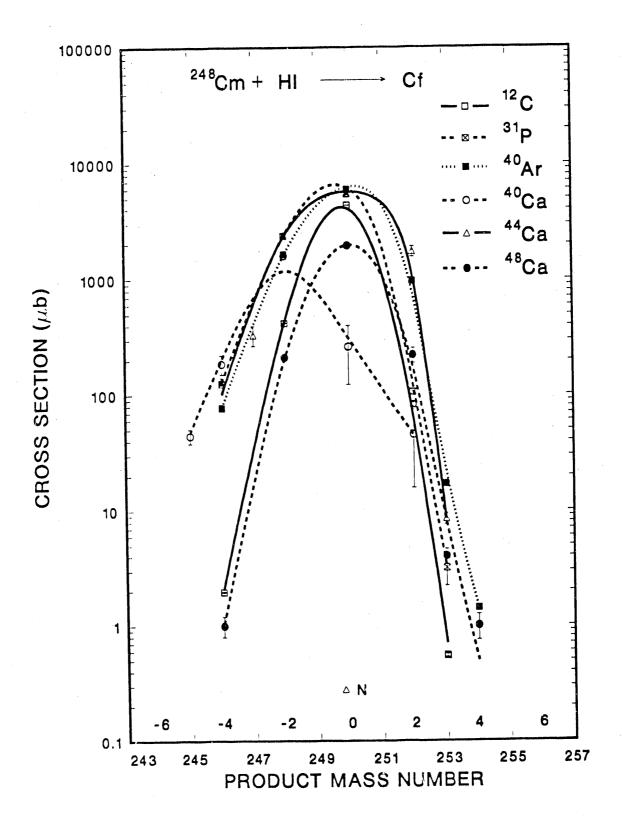


Fig. III.25

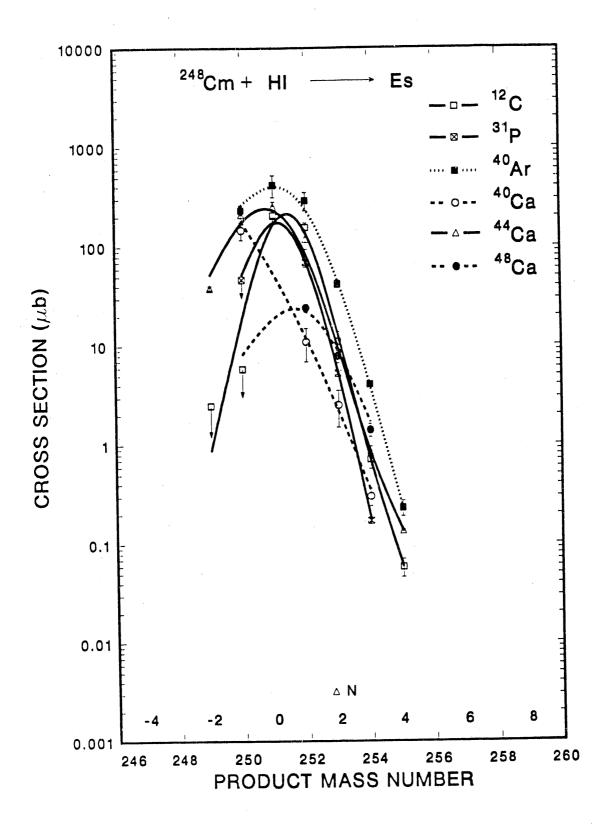


Fig. III.26

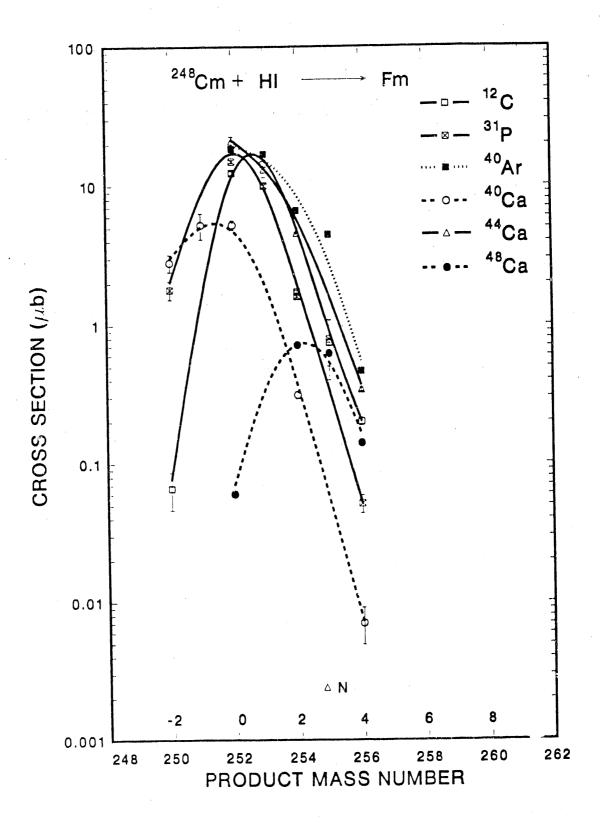


Fig. III.27

IV. THEORY

The theoretical treatment of transfer reactions is no trivial undertaking due to the extremely complex situations and the nonequilibrium conditions involved. However, several theories have been developed to model these processes [ABU80] [BIM72] [BRI72] [FRI83] [GAL69] [GAL70] [GRE85] [HAH74] [HAR83] [HOF88A] [KAU60] [NÖR75] [RAN86] [SIE71] [SIK72] [UDA82] [WIL80A]. These theoretical approaches are varied and range from simple semiclassical calculations to very intricate computations utilizing the distorted-wave Born approximation (DWBA). This list, however, is by no means complete and only a few of these theories will be discussed.

A. Potential Energy Surfaces

Potential energy surfaces (PES) have proven to be useful, especially in deeply inelastic reactions, for predicting the primary mass distributions of reaction products [TÜR89] [WEL85]. The most probable mass distribution should coincide with the highest level density in the dinuclear complex (composed of the projectile-like fragment and the target-like fragment), corresponding to a minimum in the potential energy of the system [GRO81]. V_{prod} , the potential energy of a specific product pair at an interaction radius, R_{int} is

$$V_{prod} - M_{tl} - M_{pl} = V_{Coul} + V_{nucl} + V_{cent}, \tag{1}$$

where M_{tl} and M_{pl} are the ground-state masses of the target-like and projectile-like fragments respectively, V_{Coul} is the Coulomb potential, V_{nucl} is the nuclear proximity potential [BLO77], and V_{cent} is the centrifugal potential. The potential energy, PE, of a specific point on the PES is then defined as

$$PE = V_{prod} - V_{react} - Q_{gg}, \tag{2}$$

where V_{react} is the potential energy of the reactants and Q_{gg} is the ground state Q value. Thus, the PES is constructed from these PE points.

Türler has calculated *PES* for the 40,44,48Ca-248Cm systems at $\equiv 1.1$ X Coulomb barrier, with a corresponding grazing angular momentum of about 70 % [TÜR89]. Contour maps of these three systems are shown in Fig. IV.1. The equipotential energy contour lines are 4 MeV apart, with the dashed lines representing 0 MeV. The dotted lines connect the calculated minimum potential energy points for each Z; hence, defining the valley of N/Z equilibration. The position of 248Cm (the injection point) is depicted by a cross (+). Experimental maxima of the isotopic distributions are represented by filled circles and open circles are estimates of the maxima of unmeasured distributions.

In the 40 Ca system, the transfer of about four protons from the projectile to the target is required to reach N/Z equilibration for the above target products. The position of 248 Cm in this system lies on a very "steep" wall of the PES on the neutron-rich side. Additionally, the experimentally determined maxima of the isotopic distributions all lie on the neutron-rich side of the N/Z equilibration valley.

Like the 40 Ca system, the 44 Ca system has its injection point also displaced to the neutron-rich side of the valley of N/Z equilibration by about four mass units. The experimentally determined maxima of the isotopic distributions lie on the neutron-rich side of the N/Z equilibration valley, except for those products resulting from reactions involving the exchange of large numbers of nucleons such as Th and Fm. It should be noted that the maxima of the isotopic distributions for the above target products lie on or near the N=152 neutron subshell. Hence, this closed shell probably has a measurable effect on the final product distribution.

The 48 Ca system differs significantly from the 40 Ca and 44 Ca systems since the injection point is located on the line of N/Z equilibration. All of the experimental maxima of the isotopic distributions for Z=92-100 lie on or very near the N/Z

equilibration line, indicating low initial excitation energy. Gäggeler [GÄG86] attributed the displacement of the maxima of the isotopic distributions of those products with $Z \le 90$ to neutron evaporation from the excited primary fragments. The fact that the experimental maxima of the isotopic distributions for the 40 Ca and 44 Ca systems lie on the neutron-rich side of the valley of N/Z equilibration indicates that the reaction products, particularly the above target products with $Z_{prod} \equiv Z_{target}$, have essentially no excitation energy [TÜR89].

The *PES* have been calculated for the 12 C, 31 P, and 40 Ar systems and contour maps are shown in Figs. IV.2-IV.4. The masses were calculated using the Myers and Swiatecki liquid droplet model [MYE77] [MYE80]. The odd-even term in the mass formula was neglected. Shell effects were included in these calculations while angular momentum effects were ignored. The equipotential lines have a spacing of 5 MeV. The minimum potential of each Z is represented by an open circle. These circles have been connected by a dotted line to aid in viewing the contour plot. The solid circles represent the experimental maxima of the isotopic distributions, and the solid square represents the injection point, 248 Cm.

The injection point for the 12 C system (see Fig. IV-2) lies on the neutron-rich side of the N/Z equilibration line. However, the experimental maxima of the isotopic distributions lie on the neutron-deficient side of the N/Z equilibration line, possibly due to neutron evaporation from excited primary fragments. Türler found the 18 O system to behave similarly, with the experimental maxima of the isotopic distributions also being located on the neutron-deficient side of the N/Z equilibration line [TÜR89].

Fig. IV.3 shows the *PES* contour plot for the ^{31}P system. The experimental maxima of the Fm and Es isotopic distributions lie on the N/Z equilibration line, whereas the maxima of the Bk and Cf isotopic distributions lie on the neutron-rich side. Like the ^{44}Ca system, the maxima of the isotopic distributions for Z=97-100 from the ^{31}P system lie on or near the N=152 neutron subshell.

The *PES* contour plot of the 40 Ar system is shown in Fig. IV.4. As with the 31 P system, the experimental maxima of the Bk and Cf isotopic distributions lie on the neutron-rich side of the N/Z equilibration line. However, the maxima of the Fm and Es curves lie on the neutron-deficient side of the N/Z equilibration line, possibly indicating the formation of hot primary fragments and subsequent neutron evaporation.

As can be seen from Figs IV.1-IV-4, the *PES* do a relatively poor job of calculating the primary mass distributions of the above target reaction products ($Z_{product} \cong Z_{target}$) for the systems studied. N/Z equilibration is not reached in many instances, especially in reactions involving the transfer of 1-2 protons and various numbers of neutrons. However, N/Z equilibration is eventually reached after larger numbers of nucleons are exchanged between the target and projectile.

Additionally, since the peaks of the isotopic distributions, in many instances, lie on the neutron-rich side of the N/Z equilibration line, the primary reaction products are most likely being formed with a small amount of excitation energy. This, however, does not hold for the ^{12}C or ^{18}O systems which both display maxima of isotopic distributions that are located on the neutron-deficient side of the N/Z equilibration line.

B. Two-Step Transfer Model

As mentioned in Section I.B, Hahn et al. [HAH74] performed a detailed comparison between the experimental results of the 239 Pu(12 C, α 2- α 3n) reaction channels and the calculated values from a simple two-step kinematic model. Hahn et al. used the general expression obtained by Siemens et al. [SIE71] to treat only those cases where nucleons are transferred from the projectile to the target. In the first step of this semiclassical model, a cluster of nucleons is transferred from the projectile to the target. The transfer is assumed to occur at the nuclear surface and the velocity of the cluster is taken to be the same as the velocity of the projectile. Step two involves the Rutherford scattering of the light and heavy fragments formed in the transfer reaction. The model, however, does not treat the details of the separation of the projectile into the transferred aggregate and the

residual product, nor does it consider the probability of a specific transfer process occurring. Additionally, the reaction Q value is not treated in the calculation.

Specifically then, the final velocity, V_{Rf} , at angle θ_R of the target-like fragment resulting from a transfer reaction is from the following two processes (see Fig. IV.5):

- 1. A group of nucleons is transferred from the projectile to the target, resulting in a recoiling nucleus R with velocity V_{RI} .
- 2. erford scattering between R and the remaining light nucleus results in velocity V_{R2} at angle ϕ .

For a specific reaction channel

$$P + T - (T + A) + L = R + L,$$
 (3)

where P is the projectile, T is the target, A is the transferred aggregate, L is the light residue, and R is the recoiling nucleus. Hence, for step 1

$$V_{RI} = \frac{M_A}{M_R} V_P, \tag{4}$$

where M_A and M_R are the masses of the transferred aggregate and recoiling heavy product nucleus respectively, and V_P is the velocity of the projectile.

Then, for step 2

$$V_{R2} = \frac{2M_L}{M_R + M_L} V_{P} \cos \phi, \tag{5}$$

where M_L is the mass of the light residue, and

$$\cos\phi = \frac{V_C}{2E_O - V_C},\tag{6}$$

where V_C is the Coulomb barrier between R and L at the distance of closest approach, and E_O is the energy in the center-of-mass system given by

$$E_O = E_P - \frac{M_L M_R}{M_P (M_R + M_L)}, \qquad (7)$$

where E_P is the kinetic energy of the projectile.

The calculated kinetic energy of the target-like fragment, E_R , and the emission angle of the target-like fragment, θ_R , are compared with the experimental results obtained by Hahn *et al.* in Fig. IV.6A. The curves calculated for various reaction channels are labeled with the lighter fragment of that specific reaction channel (example: $^{239}\text{Pu}(^{12}\text{C},^{3}\text{He})^{248}\text{Cf}$). As can be seen from the figure, the model does a fair job of reproducing the experimental results. Specifically, it reproduces the rapid decrease in θ_R with decreasing projectile bombarding energy.

Fig. IV.6B illustrates the results of a model developed by Galin *et al.* [GAL69] [GAL70] in which the incoming and outgoing trajectories are expressed only in terms of Rutherford scattering. This model requires the transfer to occur at the point of closest approach, R_{min} , where the incoming (i) and outgoing (f) trajectories meet. Therefore

$$Z_{P}Z_{T}e^{2}$$

$$R_{min} = \frac{}{2E_{i}} \left[1 + \csc(\frac{1}{2}\theta_{i})\right], \tag{8}$$

and

$$R_{\min} = \frac{Z_L Z_R e^2}{2E_f} \left[1 + \csc(\frac{1}{2}\theta_f) \right], \tag{9}$$

where E is the energy available in the center-of-mass system, and

$$E_f = E_i - Q. \tag{10}$$

The sum $\frac{1}{2}\theta_i + \frac{1}{2}\theta_f$ is then taken to be the effective angle at which the light product is observed.

For Hahn's calculations, a distance of 14 fm corresponding to $r_0 = 1.65$ fm was used to match the incoming and outgoing trajectories. The Q values were chosen such that in each reaction sufficient excitation energy was available to evaporate neutrons from the Cf transfer reaction products to reach 245 Cf or 244 Cf. Once again, the calculated curves reproduce the experimental data fairly well, consistent with the transfer of a "Be" cluster from 12 C to 239 Pu. Hahn *et al.* point out the similarity of the calculated curves from the two different theories and caution against trying to interpret recoil data such as these in light of a specific transfer reaction model. However, Hahn *et al.* also note that these two theories do emphasize the importance of Coulombic scattering in transfer reactions.

C. Gregorich's Model of Heavy Ion Damped Reactions

Gregorich [GRE85] has developed a simple model for the theoretical determination of reaction cross sections of target-like products resulting from heavy ion bombardments of actinide targets. The model assumes a binary reaction mechanism in which the projectile collides with the target forming a dinuclear complex, which then exists for a time comparable to the nuclear transit time. During the lifetime of this dinuclear complex, nucleons, energy and angular momentum are exchanged. The model, however, ignores the exchange of angular momentum. Additionally, the charge distribution is assumed to be unequilibrated. An empirical description of the charge distribution as described by Wollersheim [WOL84] is used to calculate $\delta^2\sigma/\delta Z\delta E$. All of the kinetic energy loss is assumed to appear as excitation energy of the resulting dinuclear complex,

and the equilibration of N/Z is achieved by the transfer of neutrons only. The neutrons and excitation energy are statistically distributed between the two products according to a simple Fermi gas level density, yielding $\delta^3\sigma/\delta Z\delta N\delta E^*$, the triple differential cross section of the target-like products, where E^* is the excitation energy. This distribution is then allowed to deexcite by the emission of neutrons in competition with fission to obtain the final cross section distribution. Charged particle emission was not considered.

Reaction cross sections based upon Gregorich's model were calculated using the computer code VMOD [GRE85] for target-like fragments for the ^{31}P , ^{40}Ar , and ^{44}Ca systems at ≈ 1.16 X Coulomb barrier and are shown in Figs. IV.7-IV.9. The experimental data points are connected with dashed lines while solid lines connect the calculated points.

For the ³¹P system (Fig. IV.7), centroids of the calculated isotopic distributions of Bk, Cf, Es, and Fm are too low when compared to the experimental centroids. The widths of the calculated Cf and Fm curves match the experimental widths fairly well. However, the width of the calculated Es distribution is much too wide. The calculated cross sections are typically too low, indicating that a significant fraction of the target-like reaction products are surviving neutron evaporation and fission. Hence, the reaction products are being formed with very little excitation energy. The model reproduces the Fm distribution better than the other distributions, although the calculated cross sections are lower than the experimental cross sections.

The calculated isotopic distributions for the ⁴⁰Ar system are shown in Fig. IV.8. The centroids for the Bk, Cf, and Es isotopic distributions are once again too low; however, the widths of the calculated distributions are fairly close to the widths of the experimental distributions. The calculated Es distributions reproduce the experimental distribution quite well, whereas the calculated Fm distribution is a bit high. The model does a very poor job of reproducing the Bk and Cf experimental distributions.

Fig. IV.9 illustrates the calculated and experimental isotopic distributions for the

44Ca system. As with the other two systems, the centroids of the calculated distributions are too low, with the possible exception of the Fm distribution. The experimental cross sections are mostly higher than the calculated cross sections. The model does an excellent job of reproducing the Fm experimental isotopic distribution.

Cross sections, using Gregorich's model, were also calculated for the ⁴⁴Ca system at the Coulomb barrier. These results are plotted in Fig. IV.10. Like the ⁴⁴Ca system at 1.16 X Coulomb barrier, the model typically estimates cross sections for Bk and Cf isotopes which are too low compared to the experimental cross sections. However, the calculated Fm cross sections are too large. The centroids of the calculated isotopic distributions are also low, with the exception of the Fm distribution.

Gregorich's model of damped reactions involving heavy ion projectiles and actinide targets inadequately reproduces the Bk, Cf, and Es experimental isotopic distributions from the ³¹P, ⁴⁰Ar, and ⁴⁴Ca systems. In contrast, the model satisfactorily duplicates the Fm experimental isotopic distributions from the three systems. In most of the cases studied, the experimental peak centroids of the isotopic distributions are on the neutron-rich side of the calculated peak centroids, indicating that *N/Z* equilibration is not being reached in these reactions. Also, cross sections calculated using Gregorich's model are consistently low for many of the above target actinide products, especially isotopes of Bk, Cf, and Es. This disagreement between Gregorich's model and the data indicate that these reaction products are being produced "cold"; therefore, surviving fission and/or neutron emission. Hence, Bk, Cf, Es, and perhaps some of the Fm isotopes produced in these reactions do not come from a damped reaction mechanism, but rather from a quasielastic type of reaction.

D. Compound Nucleus Calculations: ALICE Code

The computer code ALICE [BLA82] was used to calculate cross sections of above target reaction products from the ¹²C-²⁴⁸Cm system based upon compound nucleus formation and subsequent particle evaporation. The code utilizes a standard Weisskopf-

Ewing evaporation calculation [WEI40] with multiple particle emission including neutrons, protons, deuterons, and alphas. The code also includes the Bohr-Wheeler transition state model for fission [BOH39], and the hybrid/geometry dependent hybrid models for precompound decay [BLA71] [BLA72].

The results of the ALICE code calculation at 71 and 84 MeV are plotted in Figs. IV.11 and IV.12 respectively, along with the experimental data at these energies. The Md cross sections at 71 MeV are grossly overpredicted by ALICE which also incorrectly calculates the peak of the isotopic distribution to be at mass number 257. At both energies, ALICE overpredicts the Fm cross sections by three orders of magnitude in some cases. The calculated peak Es cross section at 71 MeV is at mass number 255 and has a value of 0.5 μb. All other calculated Es cross sections at this energy are less than 0.01 μb. Hence, ALICE is underpredicting these Es cross section by several orders of magnitude. The calculated peak of the Es isotopic distribution at 84 MeV is at mass number 253 versus the experimental maximum at 251. At 84 MeV, ALICE does a better job of reproducing the Es experimental cross sections. However, at this energy, the calculated cross sections for the neutron-deficient Es isotopes are too low. Looking at the Cf isotopic distributions, ALICE predicts the peaks to be at mass numbers 252 and 251 for 71 and 84 MeV respectively, while the experimental maxima are at mass number 250 at both energies. In addition, the calculated Cf cross sections are too low at both energies. Finally, the calculated Bk cross sections are less than 0.01 µb for all Bk isotopes at both energies, which is not in agreement with the experimental data. From these results one can conclude that ALICE does not do an adequate job of calculating the cross sections for target-like fragments produced in the ¹²C-²⁴⁸Cm system, thus indicating that a compound nucleus-type mechanism is not involved in the formation of these products.

E. Partial Wave Calculations

When a heavy ion projectile interacts with a target nucleus, the angular momentum, L, of the system is given by

$$L = pb, (11)$$

where p is the linear momentum and b is the impact parameter. Since the reduced de Broglie wavelength, λ , is

$$\chi = \frac{\pi}{p} \,, \tag{12}$$

then

$$p = \frac{\pi}{\gamma}. (13)$$

Substituting in for p in Eq. (11), one gets

$$L = \frac{b \hslash}{\chi} \tag{14}$$

The quantum mechanical restraints on the system require

$$L = l\hbar, \tag{15}$$

where $l = 0, 1, 2, \dots$

Hence,

$$b = \mathcal{H}. \tag{16}$$

Therefore, the cross section, σ_l , which corresponds to a collision having angular momentum H is given by

$$\sigma_l = \pi \lambda^2 [(l+1)^2 - l^2] T_l \tag{17}$$

$$=\pi \lambda^2 [l^2 + 2l + 1 - l^2] T_l \tag{18}$$

$$=\pi \mathcal{K}^2[2l+1]T_l,\tag{19}$$

where T_l is the transmission coefficient. The total reaction cross section is then

$$\sigma_r = \sum_{l=0}^{\infty} \sigma_l \tag{20}$$

$$= \pi \lambda^2 \sum_{l=0}^{\infty} (2l+1)T_l. \tag{21}$$

1. Sikkeland's Theory of Compound Nucleus Formation

Sikkeland *et al.* [SIK68] have calculated heavy-ion induced compound nucleus reaction cross sections using

$$\sigma_{\mathcal{X}}(E_i) = G^{\mathcal{X}} \sigma_{Cn} P_{\mathcal{X}}, \tag{22}$$

where

$$G = \begin{bmatrix} \prod & \Gamma_n & 1/x \\ \prod & (-----)_i \end{bmatrix}, \qquad (23)$$

$$i=1 \qquad \Gamma_n + \Gamma_f$$

and

$$\sigma_{cn}P_X = \sum_{l=0}^{l} \sigma_l(E_i)P_{X,l}(E) . \qquad (24)$$

G is the mean value of $\Gamma_n/(\Gamma_n + \Gamma_f)$, σ_{cn} is the cross section for the formation of the compound nucleus, P_X is the probability of emitting x neutrons, l_{cn} is the cutoff value where the reaction mechanism changes from compound nucleus reactions to grazing reactions, $\sigma_l(E_i)$ is the cross section for the *l*th partial wave at bombarding energy E_i , and

 $P_{X,l}(E)$ is the probability of emitting exactly x neutrons from a compound nucleus of angular momentum l and effective excitation energy E, based upon the Jackson model [JAK56]. Charged particle evaporation was ignored in these calculations.

2. Sikkeland's Theory of Transfer Reactions

Eq. (22) was modified by Sikkeland e. al. [SIK72] to calculate cross sections for transfer reactions such as (12 C, αxn) in which a 8 Be is effectively transferred from the projectile to the target. Thus, the cross sections for transfer reactions are calculated from angular momentum values above the sharp cutoff value using

$$\sigma_{j\chi}(E_i) = P_j G^{\chi} \sum_{l=l_{CR}+1}^{\infty} \sigma_{l}(E_i) P_{\chi,l}(E), \qquad (25)$$

where P_j is the mean probability of transferring exactly j nucleons in an incomplete fusion process. Based upon existing experimental data, Sikkeland proposed an empirical method of calculating P_j ,

$$\log P_j = -\frac{1}{(0.1 + Z_j)},$$
(26)

where Z_j is the number of protons transferred to the target nucleus.

When comparing $(^{12}\text{C},\alpha xn)$ and $(^{13}\text{C},\alpha xn)$ reactions on ^{249}Cf (see Table IV.1), it was noted that the same value of P_j did not fit both reaction channels well. The predicted energies for x=1 and 2 reactions agreed with the experimental data. However, for the x=3 and 4 reaction channels, the maximum predicted cross sections occurred at energies which were much too high. Additionally, for the $(^{12}\text{C}, 2\alpha 3n)$ reaction, the predicted cross section value agreed fairly well with the experimental data. However, the predicted energy at which the maximum occurred was too large.

3. Wilczynski's Sum-Rule Model

Wilczynski has developed a "Sum-Rule Model" to calculate absolute cross sections for complete and incomplete fusion reactions [WIL80A]. The absolute cross sections for each reaction channel can be determined from

$$l_{max} \qquad Q_{gg}(i) - Q_{c}(i)$$

$$\sigma(i) = \pi \lambda^{2} \sum_{l=0}^{\infty} (2l+1)N_{l}T_{l}(i)\exp\left(-\frac{1}{T}\right), \qquad (27)$$

where N_l is the normalization factor, T_l is the transmission coefficient, Q_{gg} is the ground state Q value, Q_c is the change in the Coulomb interaction energy due to charge transfer, and l_{max} is the largest l value for which the total nucleus-nucleus potential is attractive and/or the distance of closest approach is smaller than the sum of the half-density radii.

The computer code SUMRUL, obtained from Oak Ridge [BEE87], was used to calculate the absolute cross sections for various exit channels for the \$^{12}\text{C}\$-\$^{248}\text{Cm}\$ and \$^{40}\text{Ar}\$-\$^{248}\text{Cm}\$ systems. This code is based upon Wilczynski's "Sum-Rule Model." The results of the calculation along with some experimental results are plotted in Figs IV.13-IV.16. The experimental cross sections are connected with dashed lines while the calculated cross sections are connected with solid lines. For the \$^{12}\text{C}\$ system, the model calculates peak cross sections for Bk and Cf which are too low by one or two orders of magnitude (see Fig IV.13). In addition, the model predicts that the peaks of the mass yield curves will be at mass numbers 249 and 252 for Bk and Cf, respectively, whereas the experimental peaks are at 248 and 250, respectively. Fig IV.14 illustrates the calculated Es and Fm cross sections for the \$^{12}\text{C}\$ system. The model calculates cross sections for the heavier Es isotopes which are too high and misses the peak by two mass units. For the Fm cross sections, the model calculates the peak to be at mass number 256 while the experimental maximum is between 252 and 253. The model also overestimates the neutron-rich Fm isotope cross sections by several orders of magnitude.

The cross sections calculated using this model are fairly consistent with the experimental cross sections for the Bk and Cf isotopes produced in the interactions of 40 Ar with 248 Cm (Fig. IV.15). The calculated cross sections for the neutron-rich Cf isotopes are too high. The model fails badly on the neutron-rich Es and Fm isotopes as can be seen in Fig. IV.16. Not only are the calculated cross sections orders of magnitude too high, but the calculated peaks of the isotopic distributions are also wrong.

This model, however, does not include fission. If one assumes that the heavy fragment is emitted completely cold then the distributions in Figs. IV.13-IV.16 would effectively be the final isotopic distributions. If, however, the heavy fragment receives an amount of excitation energy which is comparable to its fission barrier and/or neutron binding energy, then the production cross section will be reduced. $\Gamma_n/(\Gamma_n + \Gamma_f)$ values for isotopes of Bk, Cf, Es and Fm, based on the empirical equations of Sikkeland *et al.* [SIK68] have been calculated and are given in Table IV.2. Including fission with the Bk and Cf cross section calculations for the ¹²C system (Fig. IV.13) would cause greater disagreement with the experimental data. The calculated Es and Fm cross sections (Fig. IV.14) for the neutron-rich isotopes would have better agreement with the experimental cross sections. For the ⁴⁰Ar system, inclusion of fission would lower the cross sections of the neutron-rich Cf isotopes resulting in closer agreement between experimental and calculated cross sections. The calculated Es and Fm cross sections would be more consistent with the experimental data with the inclusion of fission.

4. Presentation of New Model: PWAVE Calculations

A simple method involving the calculation of geometric cross sections for binary reactions was developed by Hoffman and Hoffman [HOF88A]. This classical treatment involves the calculation of Rutherford trajectories for a given system. Specifically, given the A, Z, and energy of the projectile and the A and Z of the target, two parameters RD and CD are calculated. RD is the difference between the radius of the projectile plus the target and the distance of closest approach of projectile and target for each Rutherford

trajectory. CD is the difference between the critical radius and the distance of closest approach for each Rutherford trajectory. If RD>0, an overlap between the two nuclei occurs and a binary reaction can take place. For CD>0, complete fusion between the target and projectile takes place. Hence, the impact parameters corresponding to RD=0 and CD=0 are calculated. The impact parameter area between these limits is then integrated to give the geometric cross section. This model assumes that the most probable number of nucleons exchanged between the target and projectile is related to the volume of the nuclear overlap region (i.e., the number of nucleons in the overlap region) and the exchange of nucleons is represented by a Gaussian distribution about this most probable value. The total binary reaction cross section for a given impact parameter is then distributed among twelve different mass numbers (one to twelve nucleons transferred) according to the Gaussian distribution. The impact parameter is incremented and the process is repeated over the interval CD<0< RD.

Cross sections based upon the model developed by Hoffman and Hoffman for the 44Ca system at the Coulomb barrier are plotted in Figs. IV.17-IV.18. The calculated cross sections are connected by dotted lines while the experimental cross sections are connected by dashed lines. It should be noted that the calculated cross section for transferring x nucleons is the total cross section of all such transfers involving x nucleons. For example, there are sixteen possible ways to transfer four nucleons between two nuclei; therefore, these multiplicities must be taken into account when comparing the calculated cross sections with the experimental data for a particular reaction channel. The calculated isotopic distributions have widths that are approximately the same as the experimental distributions; however, the calculated cross sections for the neutron-deficient isotopes are typically too low.

Hoffman and Hoffman [HOF82] found that the reaction cross sections for above target nuclides produced in the $^{16,18}\text{O}_{-}^{248}\text{Cm}$ systems were roughly proportional to $\exp(cE^*)$ where c varies from 0.35 for Fm to 0.6 for Bk. Artukh $et\ al.$ [ART71] found a

similar relationship for the $^{16}\text{O}\text{-}232\text{Th}$ and $^{16}\text{O}\text{-}197\text{Au}$ systems. Based on these results, the cross sections calculated with the Hoffman model were multiplied by the proportionality constant $\exp(cE^*)$, where c was set equal to 0.4. These cross sections are connected with solid lines in Figs. IV.17 and IV.18. As can be seen from these figures, the calculated Bk and Es cross sections match the experimental values very well. The model does a reasonable job for the neutron-rich and neutron-deficient. Cf isotopes, but the calculated cross sections for $^{249\text{-}252}\text{Cf}$ are slightly smaller than the experimental cross sections. For $^{255\text{-}256}\text{Fm}$, the model does an excellent job. However, it overestimates the cross sections of $^{252\text{-}254}\text{Fm}$. The calculated peaks of the isotopic distributions are at ^{247}Bk , ^{248}Cf , ^{251}Es , and ^{252}Fm , versus the experimental maxima at ^{248}Bk , ^{250}Cf , ^{250}Es , and ^{252}Fm .

Figs. IV.19-IV.21 illustrate the calculated cross sections from the Hoffman model using the proportionality $\exp(0.4E^*)$ for the 40 Ar, 31 P, and 12 C systems. The calculated cross sections are connected by solid lines while the experimental points are connected with dashed lines. The calculated Bk and Fm isotopic distributions from the 40 Ar system (see Fig. IV.19) match the experimental distributions very well. Like the 44 Ca system, the calculated cross sections for the neutron-deficient and neutron-rich Cf isotopes fit the experimental values very well, however, the model once again calculates cross sections for $^{249-252}$ Cf which are too low. The calculated cross sections for $^{253-254}$ Es accurately depict the experimental values but the calculated cross sections for the other measured Es isotopes are too low.

The calculated Bk and Cf isotopic distributions from the ³¹P system (see Fig. IV.20) fit the experimental distributions reasonably well. However, the model overestimates the cross sections of all the measured Es and Fm isotopes. Additionally, the model correctly predicts the peaks of the Cf, Es, and Fm isotopic distributions to be at mass numbers 250, 251 and 252 respectively.

Discrepancies arise between the calculated cross sections and the experimental cross

sections from the 12 C system (Fig. IV.21). The model underestimates the yields of the neutron-deficient Cf and Es isotopes. Since the E_X values for the Fm and Md isotopes from the 12 C system increase drastically with A (see Tables III.9 and III.10), the calculated cross sections also increase; hence, the model does a very poor job on the neutron-rich Fm and Md isotopes. However, the model does a fairly good job of calculating the Bk isotopic distribution even though the calculated cross sections are a little low.

Assuming that the excitation energy is partitioned according to the mass fraction of the projectile transferred to the target, the effective excitation energy, E_{eff} , can be calculated using

$$E_{eff} = \frac{N}{A} E_{\chi}, \tag{28}$$

where N is the total number of nucleons transferred between the target and projectile, A is the mass number of the projectile, and E_X is the excitation energy described earlier.

The results of this calculation at 236, 275 and 319 MeV for the ⁴⁴Ca system are given in Table IV.III. As can be seen, even at the highest bombarding energy studied, the effective excitation energy is less than 2.5 MeV for the above target products; hence, these products are being formed essentially cold. The experimental data corroborates this conclusion since the cross sections for the various Bk, Cf, and Es isotopes measured do not appreciably decrease even at energies of 1.37 X Coulomb barrier. Similar results were obtained for the ³¹P and ⁴⁰Ar systems.

The effective excitation energies for the ^{12}C system are given in Table IV.IV. Since the mass number of ^{12}C is much smaller than ^{44}Ca , ^{40}Ar , and ^{31}P , the fraction of mass transferred from the projectile to the target is much greater; therefore, the effective excitation of the above target products is greater. The E_{eff} 's of ^{12}C system for the Bk

and Cf isotopes of interest are \leq 4.7 MeV. However, for the Es and Fm isotopes, the E_{eff} values begin to exceed the neutron binding energies and/or fission barriers. For these isotopes, one would thus expect depletion of the cross sections due to fission and/or neutron emission.

Table IV.1. Experimental and calculated peak cross sections for No and Fm isotopes produced from the interactions of ^{12,13}C with ²⁴⁹Cf. Taken from [ESK75].

	$(HI,\alpha xn)$									
		· x :	= 1	•	x = 2					
Projectile	σ _{exp} (μb)			E (MeV)	-					
12 _C	0.16	67	0.02	71	6.0	71	1.98	75		
- 13 _C					0.46	70	0.35	72		
***************************************		X	= 3		x = 4					
Projectile	σ_{exp}	E	σ_{calc}	E	σ_{exp}	\mathbf{E}	σ_{calc}	E		
	(µb)	(MeV)	(µb)	(MeV)	(μb)	(MeV)	(µb)	(MeV)		
12 _C	2.60	77	1.01	81	0.50	87	0.07	101		
13 _C	1.70	75	1.07	78	0.25	83	0.13	90		
Processing and Prof. Sales are consistent and accounting group of				(HI,2	20xn)					
	x = 3				x = 4					
Projectile	σ_{exp}	E	σ_{calc}	E	σ_{exp}	E	σ_{calc}	E		
	(µb)	(MeV)	(µb)	(MeV)	(µb)	(MeV)	(µb)	(MeV)		
12 _C	300	82	175	136		≅ 95	≥6.68	≥156		

Table IV.2. Calculated $\Gamma_n/(\Gamma_n + \Gamma_f)$ values for various actinides based upon the empirical equations of Sikkeland *et al.* [SIK68].

	Γ_n		Γ_n	
Nuclide	$\Gamma_n + \Gamma_f$	Nuclide	$\Gamma_n + \Gamma_f$	
Bk 244	0.20	Es 249	0.15	
245	0.25	250	0.20	
246	0.32	251	0.26	
247	0.39	252	0.32	
248	0.47	253	0.35	
249	0.55	254	0.37	
250	0.63	255	0.40	
251	0.66	256	0.43	
Cf 246	0.12	Fm 250	0.07	
247	0.16	251	0.09	
248	0.21	252	0.12	
249	0.26	253	0.16	
250	0.33	254	0.18	
251	0.41	255	0.19	
252	0.43	256	0.21	
253	0.46	254	0.49	

Table IV.3. Excitation energies, E_X , and effective excitation energies, E_{eff} , for above target actinide products of the 44 Ca- 248 Cm system at 236 MeV (1.0 X Coulomb barrier), 275 Mev (1.16 X Coulomb barrier) and 319 MeV (1.35 X Coulomb barrier). E_{eff} was calculated assuming E_X is apportioned according to the fraction of projectile mass transferred between the target and projectile.

MeV	319 MeV		275 MeV		236 1	i.
$E_{\it eff}$	E_{χ}	E_{eff}	E_{χ}	$E_{\it eff}$	E_{χ}	Nuclide
(MeV)	(MeV)	(MeV)	(MeV)	(MeV)	(MeV)	
 ~~~	-0.1	0.5	3.6	0.9	6.8	Bk 244
0.4	3.8	0.7	6.4	1.0	8.8	245
0.2	2.2	0.4	4.0	0.5	5.5	246
0.3	3.8	0.3	4.6	0.4	5.4	247
0.1	1.4	0.1	1.4	0.1	1.4	248
0.0	1.9	0.0	1.1	0.0	0.3	249
	-1.6		-3.3		-4.7	250
	-1.5		-3.9		-6.0	251
0.7	5.1	1.0	6.8	1.1	8.4	Cf 246
0.5	4.3	0.6	5.2	0.7	5.9	247
0.7	7.2	0.7	7.2	0.7	7.2	248
0.4	5.6	0.3	4.8	0.3	4.1	249
0.4	7.8	0.3	6.2	0.2	4.7	250
0.3	4.6	0.2	2.2	0.0	0.1	251
0.5	5.7	0.2	2.6		-0.2	252
0.2	1.6		-2.3		-5.7	253
0.3	2.0		-2.6		-6.6	254
0.4	3.8	0.3	3.0	0.3	2,3	Es 249
0.4	4.3	0.2	2.6	0.1	1.2	250
0.5	7.0	0.3	4.6	0.2	2.5	251
0.5	5.5	0.2	2.3		-0.5	252
0.8	7.0	0.4	3.1		-0.3	253
0.7	4.9	0.0	0.3		-3.6	254
0.9	5.7	0.1	0.5		-4.1	255
0.2	1.1		-4.7		-9.9	256
0.6	4.9	0.3	2.5	0.0	0.4	Fm 251
0.8	9.3	0.6	6.1	0.3	3.4	252
1.0	8.5	0.5	4.7	0.1	1.3	253
1.6	11.8	1.0	7.2	0.4	3.2	254
1.6	9.8	0.7	4.6	0.0	0.0	255
2.3	12.7	1.3	6.9	0.3	1.7	256

Table IV.4. Excitation energies,  $E_X$ , and effective excitation energies,  $E_{eff}$ , for above target actinide products of the  $^{12}\text{C}$ - $^{248}\text{Cm}$  system at 72 MeV (1.0 X Coulomb barrier), 84 MeV (1.16 X Coulomb barrier) and 97 MeV (1.35 X Coulomb barrier).  $E_{eff}$  was calculated assuming  $E_X$  is apportioned according to the fraction of projectile mass transferred between the target and projectile.

	72 N	72 MeV		84 MeV		eV
Nuclide	$E_{\chi}$ (MeV)	E _{eff} (MeV)	$E_{\mathcal{X}}$ (MeV)	E _{eff} (MeV)	$E_{x}$ (MeV)	$\frac{E_{\it{eff}}}{({ m MeV})}$
Bk 244	-19.2	m ₁ = -	-23.3		-27.7	as of 19
245	-12.2		-15.2		-18.4	
246	-9.3		-11.2		-13.4	
247	-3.5		-4.4		-5.5	-
248	-3.5		-3.5		-3.5	
249	-0.7		0.3	0.0	1.3	0.1
250	-7.6		-5.8		-3,8	
251	-9.9		-7.2	***	-4.3	***
Cf 246	-15.3		-17.2	***	-19.4	
247	-11.6		-12.6	w et w	-13.6	
248	-3.0		-3.0		3,0	
249	-0.9		0.0	0.0	1.0	0.3
250	4.8	0.8	6.7	1.1	8.7	1.4
251	2.8	0.7	5,5	1.4	8.4	2.1
252	6.9	2.3	10.4	3.5	14.2	4.7
253	-7.6		-3.4		1.3	0.5
254	-12.8		-7.8		-2.3	
Es 249	-11.7		-10.8		-9.8	
250	-6.9		-5.1		-3.1	
251	0.4	0.1	3.1	0.8	6.0	1.5
252	1.3	0.5	4.8	1.6	8.6	2.9
253	5.3	2.2	9.6	4.0	14.2	5.9
254	2.7	1.4	7.7	3.9	13.2	6.6
255	2.7	1.6	. 8.4	4.9	14.6	8.5
256	-14.4		-8.0	***	-1.1	
Fm 252	3.0	1.0	6.5	2.2	10.3	3.4
253	5.7	2.4	10.0	4.2	14.7	6.1
254	12.5	6.2	17.5	8.7	22.9	11.5
255	15.5	9.0	21.2	12.4	27.4	16.0
256	22.4	14.9	28.8	19.2	35.7	23.8
Md 255	4.5	2.6	10.2	6.0	16.4	9.6
256	9.6	6.4	16.0	10.7	23.0	15.3
257	18.8	14.1	25.9	19.4	33.5	25.1
258	17.6	14.6	25.2	21.0	33.5	27.9

#### **Figure Captions**

Figure IV.1. Potential energy surface contour plots for the 40,44,48Ca-248Cm systems at a center-of-mass energy of 1.1 X Coulomb barrier and l = 50 M. The equipotential energy contour lines have a spacing of 4 MeV. The dashed lines represent 0 MeV. The position of the 248Cm target (called the injection point) is signified with a cross (+). The dotted lines connect the calculated minimum potential energy points of each Z. The filled circles indicate the experimentally determined maxima of the isotopic distributions and the open circles are the estimated maxima of isotopic distributions based upon experimental data where maxima where not seen. Figure taken from [TÜR89].

Figure IV.2. Potential energy surface contour plot of the  12 C system at 0  $\pi$ . The open circles represent the minimum calculated potential energy of each Z and are connected by a dotted line. The solid circles represent the experimentally determined maxima of the isotopic distributions. The  248 Cm injection point is represented by a filled square.

Figure IV.3. Potential energy surface contour plot of the  $^{31}P$  system at  $0 \,\text{H}$ . The open circles represent the minimum calculated potential energy of each Z and are connected by a dotted line. The solid circles represent the experimentally determined maxima of the isotopic distributions. The  $^{248}Cm$  injection point is represented by a filled square.

Figure IV.4. Potential energy surface contour plot of the  40 Ar system at  $0 \,\text{H}$ . The open circles represent the minimum calculated potential energy of each Z and are connected by a dotted line. The solid circles represent the experimentally determined maxima of the isotopic distributions. The  248 Cm injection point is represented by a filled square.

Figure IV.5. Laboratory system vector diagram for Two-Step Transfer Model. Recoiling target-like fragment R with mass  $M_R$  has a velocity  $V_{Rf}$  at angle  $\theta_R$  which is

the resultant of velocity  $V_{RI}$  at 0°, due to the transfer of an aggregate of nucleons from the projectile to the target, and of velocity  $V_{R2}$  at angle  $\phi$ , due to Rutherford scattering with the remaining projectile-like product L, with mass  $M_L$ . Figure taken from [HAH74].

Figure IV.6. Experimentally determined recoil energies,  $E_R$ , and scattering angle of the heavy recoiling reaction products using the Two-Step Transfer Model of Hahn [HAH74]. These results were derived from range and angular distributions measured for  $^{245}\text{Cf}$  and  $^{244}\text{Cf}$ . The circles represent the centroids of the distributions while the triangles represent the maxima of the angular distributions. The curved lines in part (a) are the results of a calculation involving the Two-Step Transfer Model, while those lines in part (b) are the results of a calculation which has considered only Rutherford scattering. Each curve is identified by the He isotope remaining after the transfer reaction. The solid lines represent those cases in which sufficient excitation energy is available in the intermediate nucleus to evaporate neutrons to reach  $^{244}\text{Cf}$ , while the dashed curves represent those cases with sufficient excitation energy to reach  $^{245}\text{Cf}$ . Figure taken from [HAH74].

Figure IV.7. Calculated isotopic distributions (symbols connected by solid lines) for Bk, Cf, Es, and Fm from the interactions of 207-MeV (1.17 X Coulomb barrier) ³¹P ions with ²⁴⁸Cm. These calculations are based upon Gregorich's model of heavy ion damped reactions. The experimentally determined isotopic distributions are represented by the symbols connected by dashed lines.

Figure IV.8. Calculated isotopic distributions (symbols connected by solid lines) for Bk, Cf, Es, and Fm from the interactions of 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions with ²⁴⁸Cm. These calculations are based upon Gregorich's model of heavy ion damped reactions. The experimentally determined isotopic distributions are represented by the symbols connected by dashed lines.

Figure IV.9. Calculated isotopic distributions (symbols connected by solid lines) for

Bk, Cf, Es, and Fm from the interactions of 275-MeV (1.16 X Coulomb barrier) ⁴⁴Ca ions with ²⁴⁸Cm. These calculations are based upon Gregorich's model of heavy ion damped reactions. The experimentally determined isotopic distributions are represented by the symbols connected by dashed lines.

Figure IV.10. Calculated isotopic distributions (symbols connected by solid lines) for Bk, Cf, Es, and Fm from the interactions of 236-MeV (1.0 X Coulomb barrier) ⁴⁴Ca ions with ²⁴⁸Cm. These calculations are based upon Gregorich's model of heavy ion damped reactions. The experimentally determined isotopic distributions are represented by the symbols connected by dashed lines.

Figure IV.11. Calculated isotopic distributions (symbols connected by solid lines) for Es, Fm, and Md from the interactions of 71-MeV (0.98 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm. The computer code ALICE was used for the calculations, which assumes compound nucleus formation followed by particle evaporation. The experimentally determined isotopic distributions are represented by the symbols connected by dashed lines.

Figure IV.12. Calculated isotopic distributions (symbols connected by solid lines) for Cf, Es, and Fm from the interactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm. The computer code ALICE was used for the calculations, which assumes compound nucleus formation followed by particle evaporation. The experimentally determined isotopic distributions are represented by the symbols connected by dashed lines.

Figure IV.13. Calculated and experimental isotopic distributions for Bk and Cf produced from the interactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm. Wilczynski's Sum-Rule Model was used to determine the theoretical cross sections. The experimental data are connected with dashed lines, while the calculated points are connected with solid lines.

Figure IV.14. Calculated and experimental isotopic distributions for Es and Fm

produced from the interactions of 84-MeV (1.16 X Coulomb barrier) ¹²C ions with ²⁴⁸Cm. Wilczynski's Sum-Rule Model was used to determine the theoretical cross sections. The experimental data are connected with dashed lines, while the calculated points are connected with solid lines.

Figure IV.15. Calculated and experimental isotopic distributions for Bk and Cf produced from the interactions of 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions with ²⁴⁸Cm. Wilczynski's Sum-Rule Model was used to determine the theoretical cross sections. The experimental data are connected with dashed lines, while the calculated points are connected with solid lines.

Figure IV.16. Calculated and experimental isotopic distributions for Es and Fm produced from the interactions of 245-MeV (1.16 X Coulomb barrier) ⁴⁰Ar ions with ²⁴⁸Cm. Wilczynski's Sum-Rule Model was used to determine the theoretical cross sections. The experimental data are connected with dashed lines, while the calculated points are connected with solid lines.

Figure IV.17. Calculated and experimental isotopic distributions for Bk and Cf produced from the interactions of 275-MeV (1.16 X Coulomb barrier)  44 Ca ions with  248 Cm. The model developed by Hoffman and Hoffman was used to calculate the theoretical cross sections. The calculated data points based upon the geometric cross section are connected by dotted lines, the calculated cross sections multiplied by  $\exp(0.4E^*)$  are connected by solid lines, and the experimental cross sections are connected by dashed lines.

Figure IV.18. Calculated and experimental isotopic distributions for Es and Fm produced from the interactions of 275-MeV (1.16 X Coulomb barrier)  44 Ca ions with  248 Cm. The model developed by Hoffman and Hoffman was used to determine the theoretical cross sections. The calculated data points based upon the geometric cross section are connected by dotted lines, the calculated cross sections multiplied by  $\exp(0.4E^*)$  are connected by solid lines, and the experimental cross sections are

connected by dashed lines.

Figure IV.19. Calculated and experimental isotopic distributions for Bk, Cf, Es, and Fm produced from the interactions of 245-MeV (1.16 X Coulomb barrier)  40 Ar ions with  248 Cm. The model developed by Hoffman and Hoffman was used to determine the theoretical cross sections. The calculated cross sections multiplied by  $\exp(0.4E^*)$  are connected by solid lines and the experimental cross sections are connected by dashed lines.

Figure IV.20. Calculated and experimental isotopic distributions for Bk, Cf, Es, and Fm produced from the interactions of 207-MeV (1.17 X Coulomb barrier)  31 P ions with  248 Cm. The model developed by Hoffman and Hoffman was used to determine the theoretical cross sections. The calculated cross sections multiplied by  $\exp(0.4E^*)$  are connected by solid lines and the experimental cross sections are connected by dashed lines.

Figure IV.21. Calculated and experimental isotopic distributions for Bk, Cf, Es, Fm, and Md produced from the interactions of 71-MeV (1.0 X Coulomb barrier)  12 C ions with  248 Cm. The model developed by Hoffman and Hoffman was used to determine the theoretical cross sections. The calculated cross sections multiplied by  $\exp(0.4E^*)$  are connected by solid lines and the experimental cross sections are connected by dashed lines.

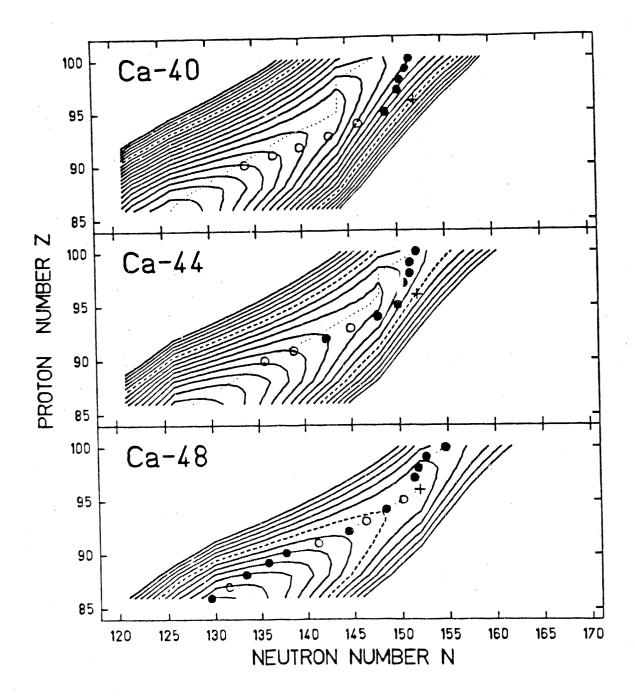


Fig. IV.1

## C-12 + CM-248

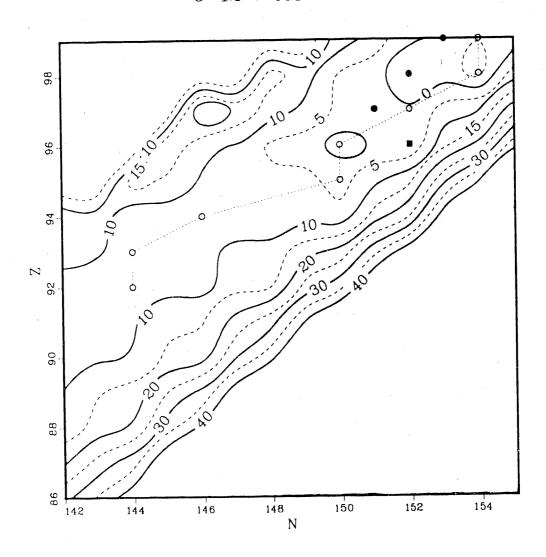
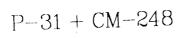


Fig. IV.2



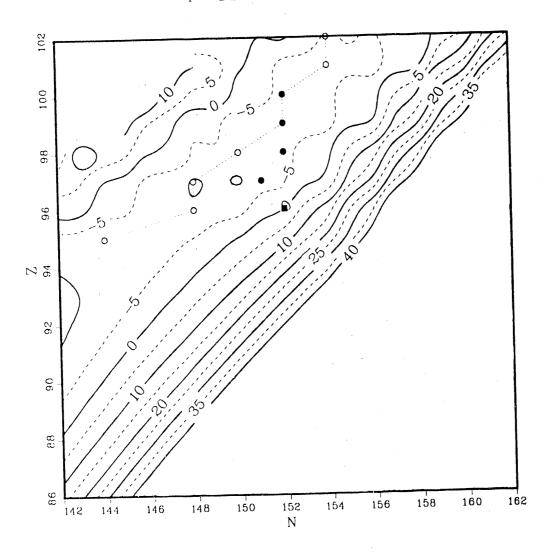


Fig. IV.3

# AR-40 + CM-248

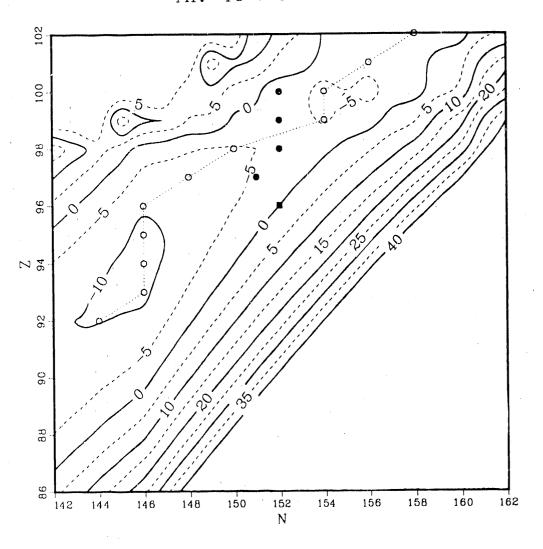
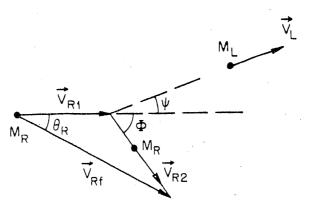


Fig. IV.4



TWO-STEP TRANSFER MODEL

Fig. IV.5

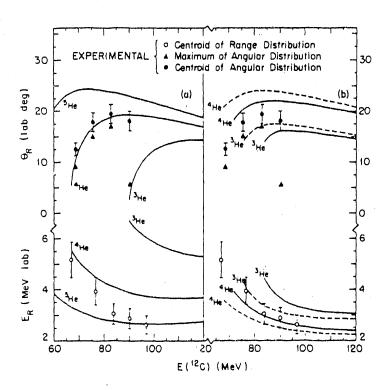


Fig. IV.6

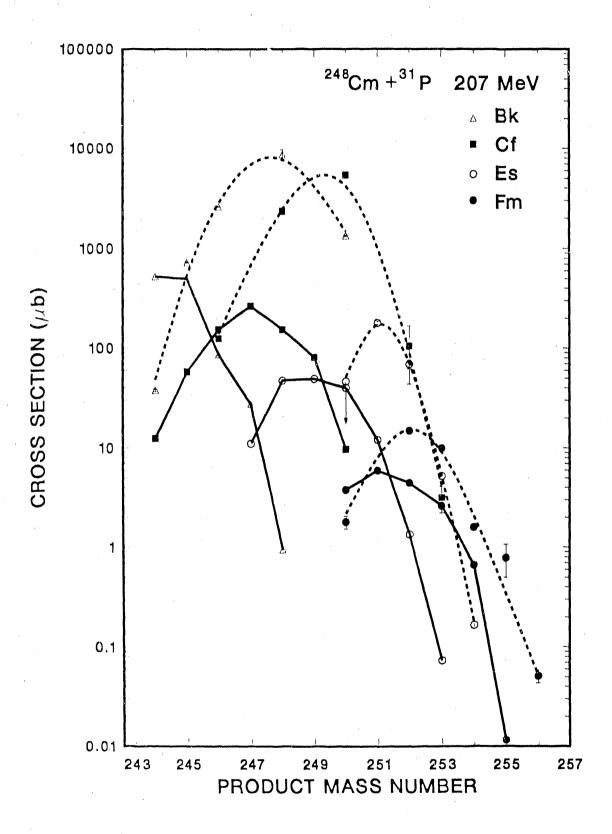


Fig. IV.7

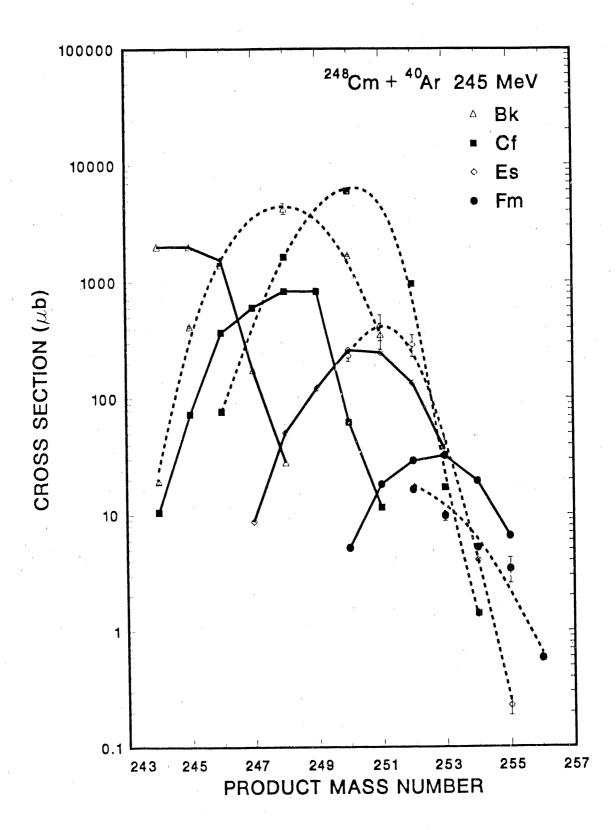


Fig. IV.8

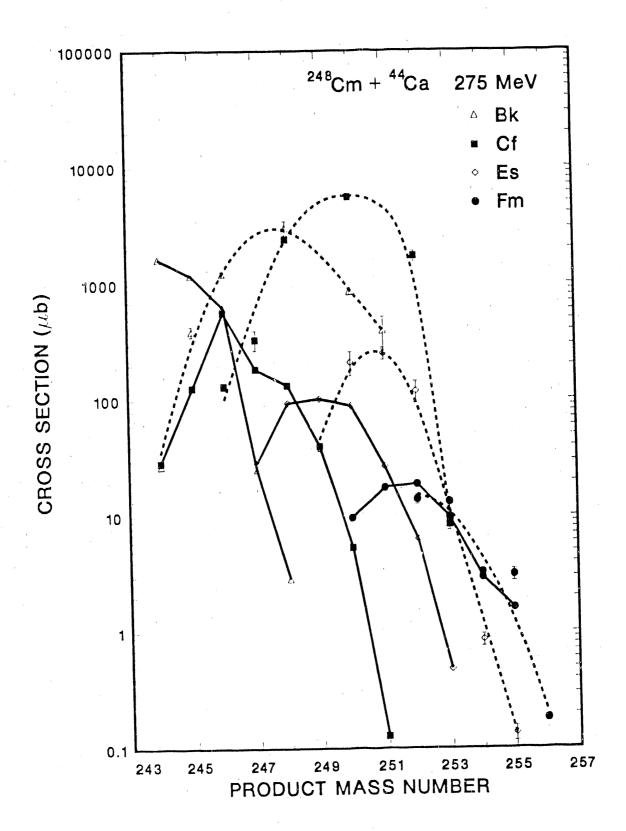


Fig. IV.9

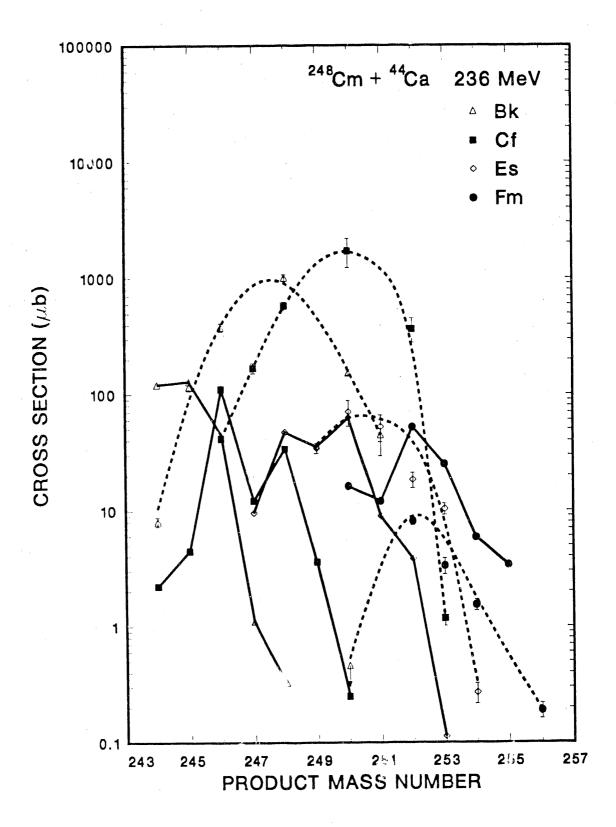


Fig. IV.10

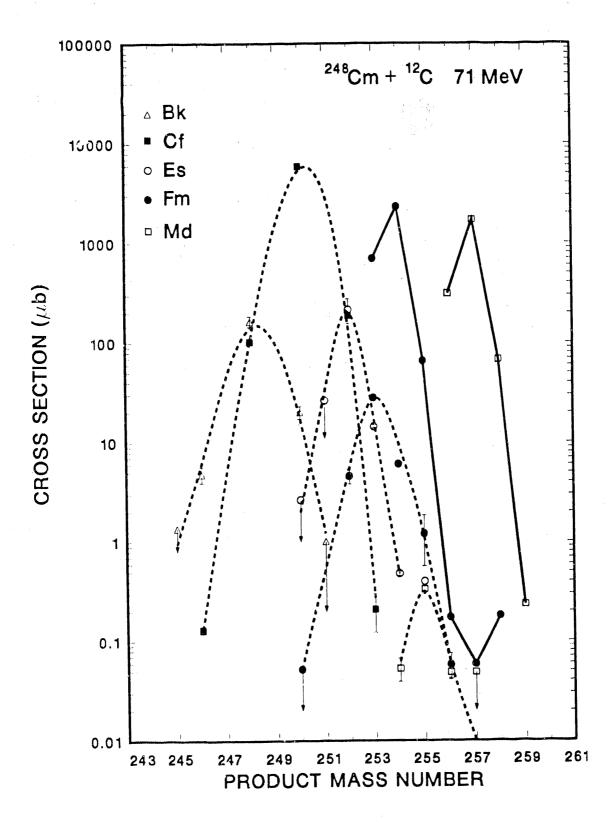


Fig. IV.11

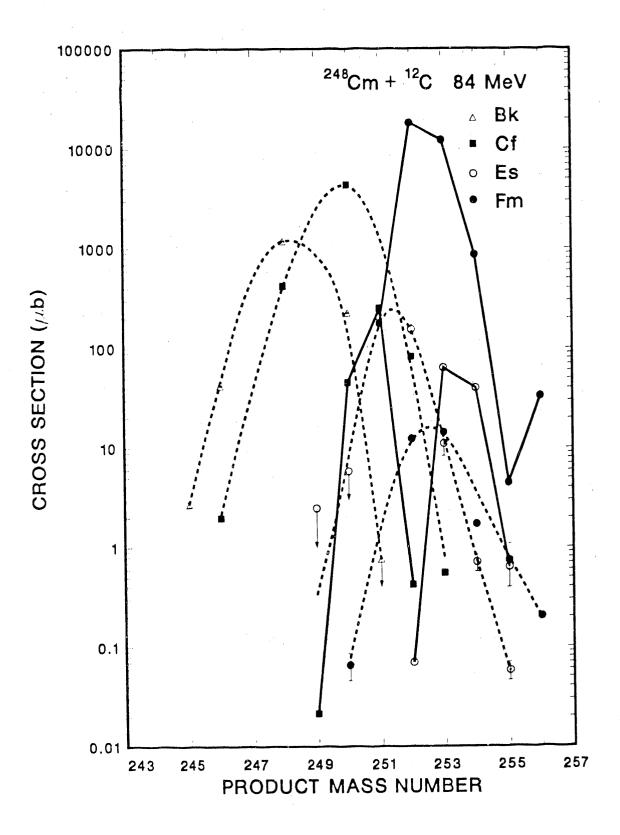


Fig. IV.12

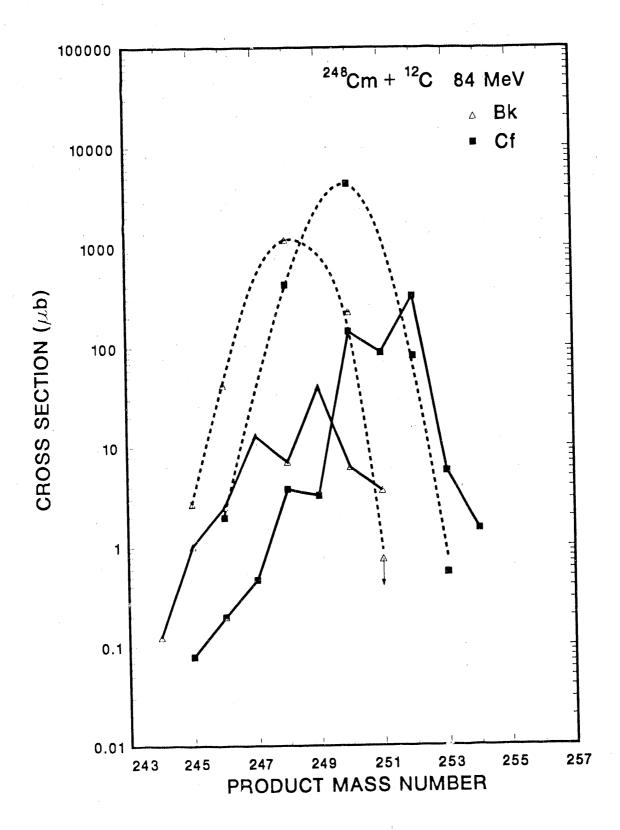


Fig. IV.13

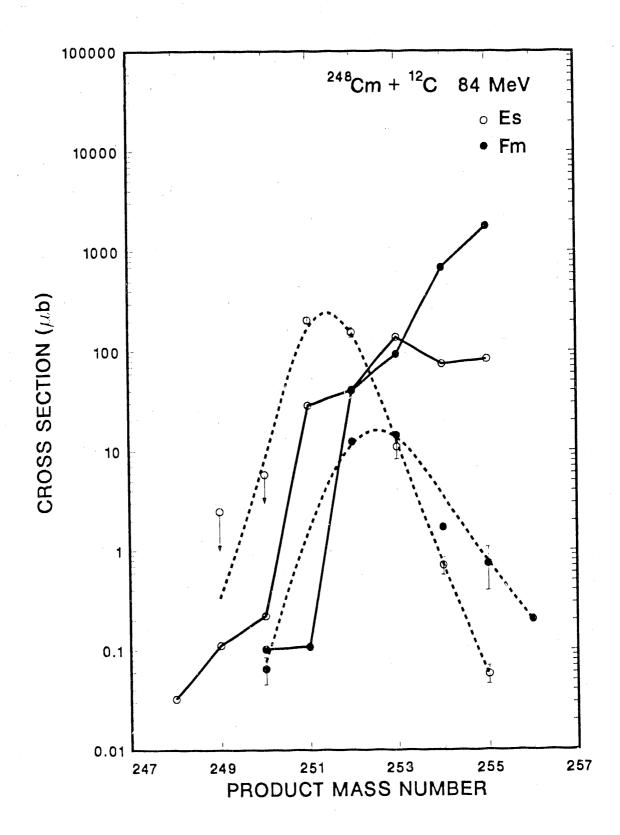


Fig. IV.14

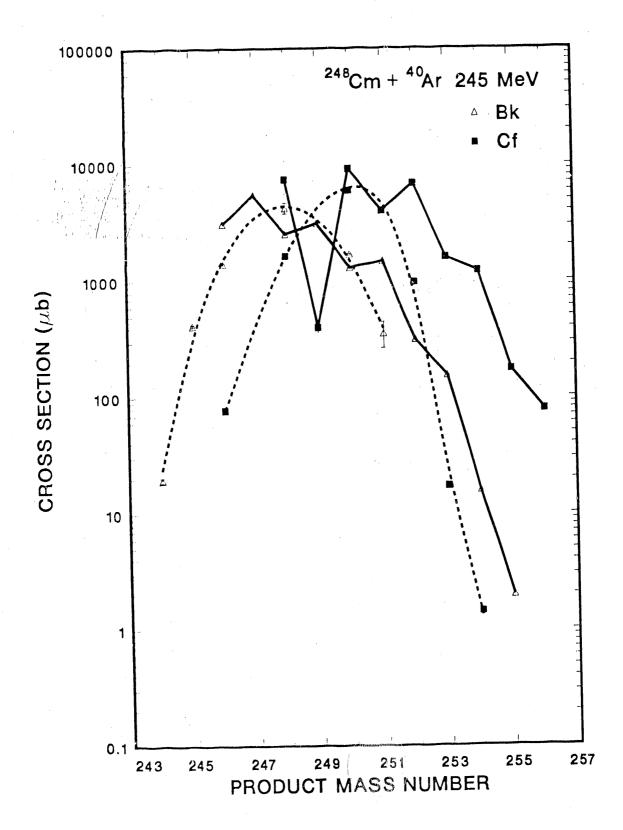


Fig. IV.15

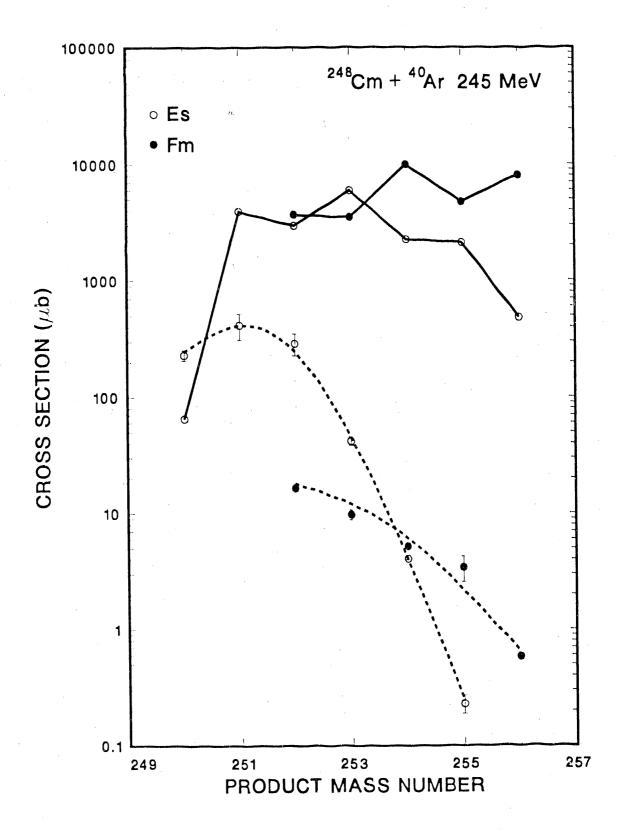


Fig. IV.16

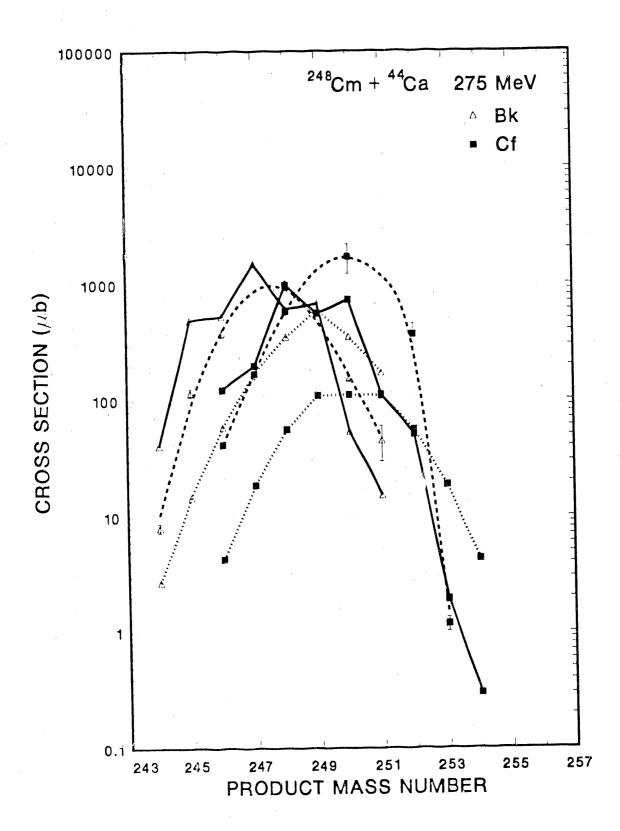


Fig. IV.17

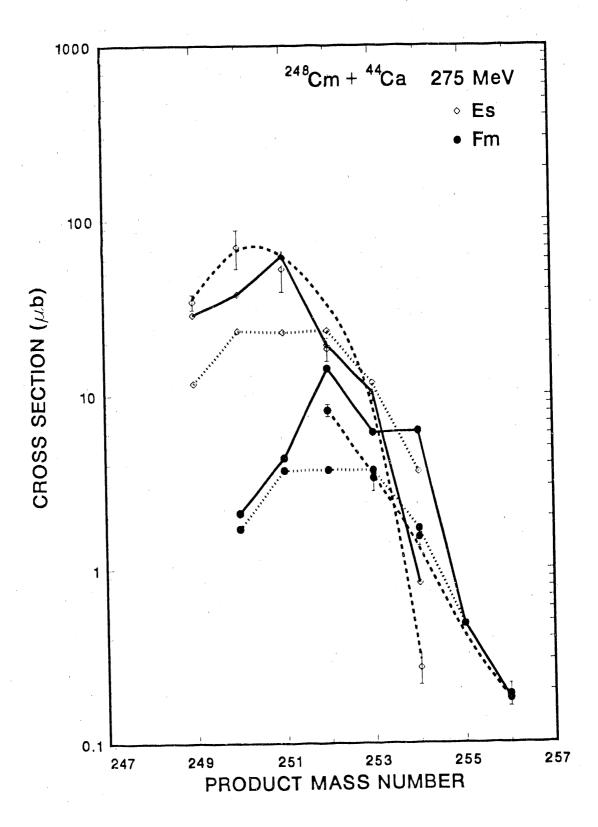


Fig. IV.18

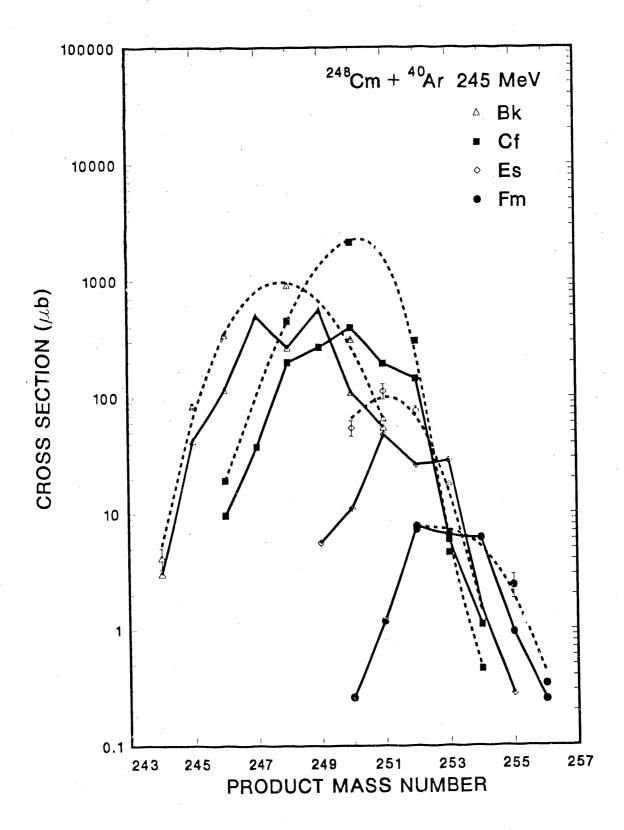


Fig. IV.19

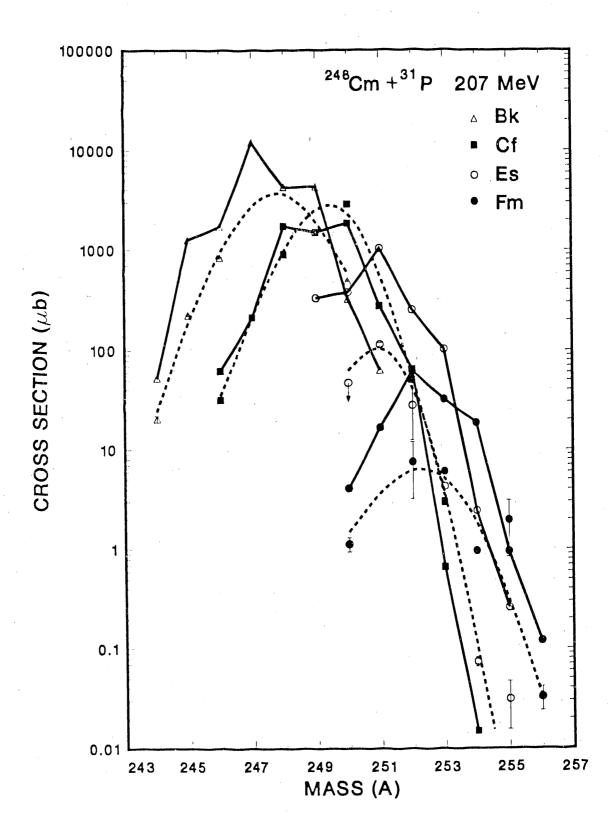


Fig. IV.20

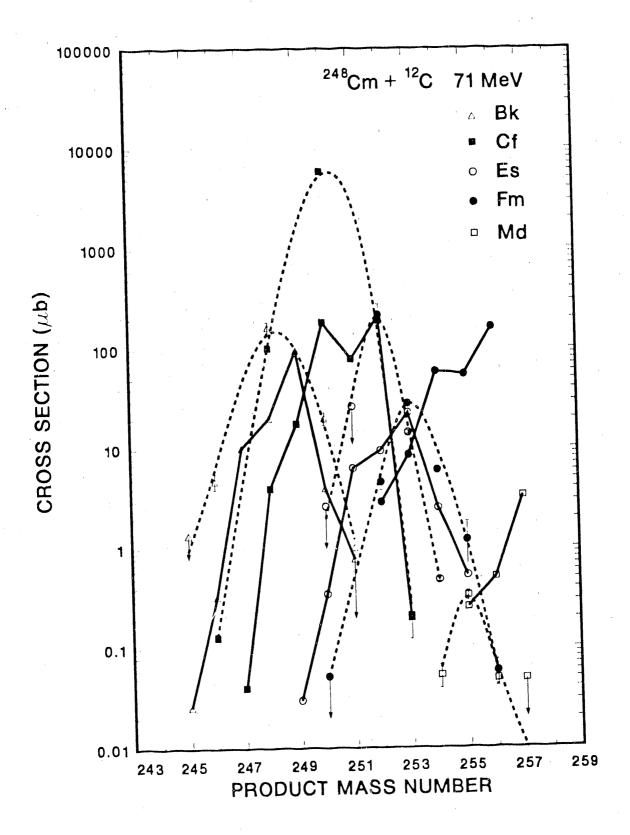


Fig. IV.21

# IV. CONCLUSIONS

248Cm was bombarded with ¹²C, ³¹P, ⁴⁰Ar, and ⁴⁴Ca ions at energies of 0.98, 1.06, 1.16, 1.25, and 1.35 X Coulomb barrier. The recoiling reaction products were collected by copper or gold catcher foils. A radiochemical procedure was used to obtain separate fractions of Bk, Cf, Es, Fm, and Md. A He/KCl jet was used to transport the recoiling No nuclides produced in the ¹²C system to a rotating wheel system.

The isotopic distributions of the above target reaction products ( $Z_{product} > Z_{target}$ ) were found to be essentially symmetric about the maximum with full-widths-at-half-maximum (FWHM) of <1.5 to 3.3 mass units. These results are similar to FWHM values found previously for other heavy ion- 248 Cm systems. The peaks of the isotopic distributions generally occur for those reaction channels involving the apparent transfer of the fewest number of nucleons for which the calculated excitation energy is a positive quantity. The maxima of the excitation functions occur at those projectile energies which are consistent with the calculated reaction barriers based upon a binary reaction mechanism.

Potential Energy Surfaces (*PES*) do a relatively poor job of calculating the primary mass distributions of the above target actinide reaction products for the systems studied. A model based upon heavy ion damped reactions was used to calculate the final isotopic distributions of the above target actinide products. This model does an inadequate job of reproducing the Bk, Cf, and Es experimental isotopic distributions of the ³¹P, ⁴⁰Ar, and ⁴⁴Ca systems. The model, however does a good job reproducing the Fm experimental isotopic distributions. In most cases, the experimental peak centroids of the isotopic distributions lie on the neutron-rich side of the calculated peak centroids, indicating that *N/Z* equilibration is not being achieved in these reactions. The damped reaction model consistently gives cross sections which are too low for many of the above target actinide reaction products, especially for isotopes of Bk, Cf, and Es. Since the cross sections for

isotopes of Bk, Cf, and Es from the ¹²C, ³¹P, ⁴⁰Ar, and ⁴⁴Ca systems do not appreciably decrease over the energy range investigated, one can conclude that these products are being formed with little or no excitation energy. A quasi-elastic mechanism is thus probably responsible for the production of isotopes of Bk, Cf, Es, and perhaps some Fm.

The computer code ALICE, based upon compound nucleus formation and subsequent particle evaporation, was used to calculate actinide production cross sections for the ¹²C system. ALICE does a very poor job of reproducing the isotopic distributions of the above target products, thus indicating that a compound nucleus mechanism is not involved in the production of these nuclides.

Hoffman and Hoffman developed a model utilizing a classical geometric cross section calculation for binary reactions. The calculated isotopic distributions based upon this model agree quite well with the experimental isotopic distributions. This model, however, does show some discrepancies with the experimental isotopic distributions from the ¹²C system.

Transfer reactions provide a useful tool for the production of "cold", neutron-rich actinides. It appears that the  $^{40}\text{Ca-}^{248}\text{Cm}$  system might be promising for the production of  $^{274}111$ . The  $E_X$  value for this reaction channel is 0.5 MeV. Another possibility is the production of  $^{291}114$  ( $E_X = 0.8$  MeV) employing the  $^{48}\text{Ca-}^{248}\text{Cm}$  system. Perhaps the  $^{40}\text{Ca-}^{249}\text{Bk}$  system could be used to produce  $^{272}111$  or  $^{278}111$  which have  $E_X$  values of 0.9 and 1.4 MeV respectively. With the choice of such heavy projectiles at energies near the Coulomb barrier, the fraction of excitation energy transferred to the target-like fragment should be low, thus producing "cold" neutron-rich actinides. Hence, by choosing the appropriate target-projectile-energy combinations, transfer reactions can be used to produce new, neutron-rich actinides and possibly reach the "island of stability."

### APPENDIX A

### Catcher foil collection efficiency code

```
PROGRAM CATFOIL
      WRITE (*,*)'This program calculates the probablity of
      WRITE (*,*)'hitting the catcher foil for reaction'
      WRITE (*,*)'products emitted at angles between one'
      WRITE (*,*)'and eighty-nine degrees inclusive with'
      WRITE (*,*)'respect to the beam axis as a function'
      WRITE (*,*)'of angle.'
      WRITE (*,*)'Equations taken from K. Moody Ph.D. Thesis,'
      WRITE (*,*)'LBL-16249, PAGE 84.'
      WRITE (*,*)
      WRITE (*,*)'Enter the radius of the catcher foil (RFOIL),'
      WRITE (*,*)'the radius of the target (RTARGET), and'
      WRITE (*,*)'the distance between the target and catcher'
      WRITE (*,*)'foil (D), in that order.'
      WRITE (*,*)'Use mm as the unit of choice.'
      WRITE (*,*)
      READ (*,*)RFOIL,RTARGET,D
      WRITE (*,*)
      PI=3.14159
      PROBMISS=0.0
      PROBHIT=0.0
      WRITE (*,100)
 100 FORMAT ('What do you want to call the output file?')
      OPEN (2,File=' ',Status='New')
      WRITE (2,105)
 105 FORMAT (' ANGLE PROB. MISS PROB. HIT')
C
      Convert degrees into radians (Multiply by PI/180)
      DO 1, J=1,89
      TRAD=FLOAT(J)*PI/180.
      PROBHIT=0.
      PROBMISS=0.
      DO 2, K=1,1000
      TR=FLOAT(K)*RTARGET/1000.
\mathbf{C}
      IF RECOILING PRODUCTS MISS ALL OF THE TIME
      IF ((RFOIL+TR-D*TAN(TRAD)).LT.0)THEN
      PROBM=1.
      PROBH=0.
C
      IF RECOILING PRODUCTS HIT ALL THE TIME
      ELSE IF ((RFOIL-TR-D*TAN(TRAD)),GE.0)THEN
      PROBM=0.
      PROBH=1.
C
      THE OTHER CASES (SOMETIMES HITS AND MISSES)
```

**ELSE** 

 $A1=((D^{**2})^{*}(TAN(TRAD)^{*}TAN(TRAD)))-(RFOIL^{**2})+(TR^{**2})$ 

B1=(2.*TR*D*TAN(TRAD))

C1=A1/B1

PROBM=(1./PI)*((ASIN(C1)+PI/2.))

PROBH=1.-PROBM

**ENDIF** 

PROBHIT=PROBHIT+PROBH*TR

PROBMISS=PROBMISS+PROBM*TR

PROBT=PROBHIT+PROBMISS

2 CONTINUE

WRITE (2,200)J,PROBMISS/PROBT,PROBHIT/PROBT

- 200 FORMAT (2X,I5,7X,F6.4,8X,F6.4)
  - 1 CONTINUE

**STOP** 

**END** 

## REFERENCES

- [ABU80] A.B. Abul-Magd, Z. Physik <u>A298</u>, 143 (1980).
- [ART71] A.G. Artukh, V.V. Avdeichikov, J. Erö, G.F. Gridnev, V.L. Mikheev, V.V. Volkov, and J. Wilczynski, Nucl. Phys. <u>A160</u>, 511 (1971).
- [BAU78] H.R. Bauer, G.D. Christian, and J.E. O'Reilly, <u>Instrumental Analysis</u>
  (Allyn and Bacon Inc. Boston 1978).
- [BAY73] R.D. Baybarz, J. Inorg. Nucl. Chem., <u>35</u>, 4149 (1973).
- [BEE87] J. Beene, private communication, (1987).
- [BIM72] R. Bimbot, D. Gardes, and M.F. Rivet, Nucl. Phys. <u>A189</u>, 193 (1972).
- [BLA71] M. Blann, Phys. Rev. Lett., <u>27</u>, 337 (1971).
- [BLA72] M. Blann, Phys. Rev. Lett., <u>28</u>, 757 (1972).
- [BLA82] M. Blann and J. Bisplinghoff, Lawrence Livermore National Laboratory Report UCID-19614 (1982).
- [BLO77] J. Blocki, J. Randrup, W.J. Swiatecki, and C.F. Tsang, Ann. Phys. <u>105</u>, 427 (1977).
- [BOH39] N. Bohr and J.A. Wheeler, Phys. Rev. <u>56</u>, 426 (1939).
- [BRA86] S.G. Bratsch and J.J. Lagowski, J. Phys. Chem. <u>90</u>, 307 (1986).
- [BRI72] D.M. Brink, Phys. Lett. <u>40B</u>, 37 (1972).
- [BRI79] H.C. Britt, in <u>Proceedings of the Fourth International Atomic Energy</u>

  <u>Symposium on Physics and Chemistry of Fission</u>, Jülich, 1979 (IAEA, Vienna, 1980), Vol. I, p. 3.
- [PRI80] H.C. Britt, E. Cheifetz, D.C. Hoffman, J.B. Wilhelmy, R.J. Dupzyk, and R.W. Lougheed, Phys. Rev. C 21, 761 (1980).
- [BRO78] E. Browne, J.M. Dairiki, and R.E. Doebler, <u>Table of the Isotopes</u>, Second Edition (J. Wiley and Sons, Inc., New York 1978).
- [CHA87] R.M. Chasteler, R.A. Henderson, D. Lee, K.E. Gregorich, M.J. Nurmia, R.B. Welch, and D.C. Hoffman, Phys. Rev. C <u>36</u>, 1820 (1987).

- [CHO56] G.R. Choppin, B.G. Harvey, and S.G. Thompson, J. Inorg. Nucl. Chem. 2, 66 (1956).
- [CHO56A] G.R. Choppin and R.J. Silva, J. Inorg. Nucl. Chem. 3, 153 (1956).
- [CHO61] G.R. Choppin abd J.A. Chopoorian, J. Inorg. Nucl. Chem. 22, 97 (1961).
- [COO52] G.B. Cook, and J.F. Duncan, <u>Modern Radiochemical Practice</u> (Oxford at the Clarendon Press, Oxford 1952).
- [CRC75] <u>Handbook of Chemistry and Physics</u>, 56th Edition (CRC Press, Inc., Cleveland 1975).
- [CUM63] J.B. Cumming, Brookhaven National Laboratory Report BNL-6470 (1963).
- [DOR89] J.G. Dorsey and K.A. Dill, Chem. Rev. <u>89</u>, 331 (1989).
- [DOS85] T. Dossing and J. Randrup, Nucl. Phys. <u>A433</u>, 215 (1985).
- [DYE61] F.F. Dyer and G.W. Liddicotte, <u>The Radiochemistry of Copper NAS-NS-3027 (1961)</u>.
- [EIS85] J.M. Eisenberg and W. Greiner, <u>Nuclear Models</u>, <u>Collective and Single Particle Phenomenon</u>, Second Edition (North Holland Publishing Co., Amsterdam 1975).
- [ESK70] P. Eskola, K. Eskola, M. Nurmia, and A. Ghiorso, Phys. Rev. C 2, 1058 (1970).
- [ESK71] K. Eskola, P. Eskola, M. Nurmia, and A. Ghiorso, Phys. Rev. C <u>4</u>, 632 (1971).
- [ESK73] P. Eskola, Phys. Rev. C 7, 280 (1973).
- [ESK75] P. Eskola, Ph.D. thesis, University of Helsinki, 1975.
- [FRA78] W.E. Frahn, Nucl. Phys. <u>A302</u>, 267 (1978).
- [FRI81] G. Friedlander, J.W. Kennedy, E.S. Macias, and J.M. Miller, <u>Nuclear and Radiochemistry</u>, Third Edition (John Wiley and Sons, New York 1981).
- [FRI83] W.A. Friedman, Phys. Rev. C <u>27</u>, 569 (1983).

- [GÄG86] H. Gäggeler, W. Brüchle, M. Brügger, M. Schädel, K. Sümmerer,
  G. Wirth, J.V. Kratz, M. Lerch, Th. Blaich, G. Herrmann, N. Hildebrand,
  N. Trautmann, D. Lee, K.J. Moody, K.E. Gregorich, R.B. Welch,
  G.T. Seaborg, D.C. Hoffman, W.R. Daniels, M.M. Fowler, and
  H.R. von Gunten, Phys. Rev. C 33, 1983 (1986).
- [GAL69] J. Galin, B. Gatty, M. Lefort, J. Peter, X. Tarrago, and R. Basile, Phys. Rev. 182, 1267 (1969).
- [GAL70] J. Galin, D. Guerreau, M. Lefort, J. Peter, X. Tarrago, and R. Basile,Nucl. Phys. <u>A159</u>, 461 (1970).
- [GAL76] J. Galin, J. Phys. (Paris) Colloque C 5, 83 (1976).
- [GEN89] General Electric Company, <u>Nuclides and Isotopes</u>, Fourteenth Edition (General Electric, San Jose 1989).
- [GHI67] A. Ghiorso, T. Sikkeland, and M.J. Nurmia, Phys. Rev, Lett. <u>18</u>, 401 (1967).
- [GLE83] N.K. Glendenning, <u>Direct Nuclear Reactions</u> (Academic Press, New York 1983).
- [GRE89] K.E. Gregorich and G.T. Seaborg, Lawrence Berkeley Laboratory
  Preprint LBL-27947
- [GRE85] K.E. Gregorich, Ph.D. thesis, University of California, 1985.
- [GRO81] D.H.E. Gross and K.M. Hartmann, Phys. Rev. C <u>24</u>, 2526 (1981).
- [HAH74] R.L. Hahn, P.F. Dittner, K.S. Toth, and O.L. Keller, Phys. Rev. C <u>10</u>, 1889 (1974).
- [HAL86] H.L. Hall (unpublished).
- [HAR83] B.G. Harvey and M.J. Murphy, Phys. Lett. <u>130B</u>, 373 (1983).
- [HER79] G. Herrmann, Nature <u>280</u>, 543 (1979).
- [HIG60] G.H. Higgens, <u>The Radiochemistry of the Transcurium Elements</u>, NAS-NS-3031 (1960).

- [HOF80] D.C. Hoffman, D. Lee, A. Ghiorso, M. Nurmia, and K. Aleklett, Phys. Rev. C <u>22</u>, 1581 (1980).
- [HOF85] D.C. Hoffman, <u>Americium and Curium Chemistry and Technology</u>
  (D. Reidel Publishing Company, Dordecht, 1985), p. 241-250.
- [HOF85A] D.C. Hoffman, M.M. Fowler, W.R. Daniels, H.R. von Gunten, D. Lee,
  K.J. Moody, K. Gregorich, R. Welch, G.T. Seaborg, W. Brüchle,
  M. Brügger, H. Gäggeler, M. Schädel, K. Sümmerer, G. Wirth,
  Th. Blaich, G. Herrmann, N. Hildebrand, J.V. Kratz, M. Lerch, and
  N. Trautmann, Phys. Rev. C 31, 1763 (1985).
- [HOF86] D.C. Hoffman, Lawrence Berkeley Laboratory Preprint LBL-21384 (1986).
- [HOF88] D.C. Hoffman, in <u>Proceedings of the Second International Conference on Nuclear and Radiochemistry</u>, Brighton, 1988, also Lawrence Berkeley Laboratory Preprint LBL-26277 (1987).
- [HOF88A] D.C. Hoffman, "Transfer Reactions with Transuranium Elements,"

  3rd Chemical Congress of North America, Award Symposium for
  Günter Herrmann, June 6-10, 1988, Toronto, Canada, also Lawrence
  Berkeley Laboratory Preprint LBL-29502 (1990).
- [HOF90] D.C. Hoffman, D.M. Lee, K.E. Gregorich, M.J. Nurmia, R.B. Chadwick,
  K.B. Chen, K.R. Czerwinski, C.M. Gannett, H.L. Hall, R.A. Henderson,
  B. Kadkhodayan, S.A. Kreek, and J.D. Leyba, Phys. Rev. C 41, 631 (1990).
- [HOR69] E.P. Horwitz, C.A. Bloomquist, and D.J. Henderson, J. Inorg. Nucl. Chem. 31, 1149 (1969).
- [HOR69A] E.P. Horwitz, C.A. Bloomquist, D.J. Henderson, and D.E. Nelson,J. Inorg. Nucl. Chem. <u>31</u>, 3255 (1969).
- [HOR72] E.P. Horwitz and C.A. Bloomquist, J. Inorg. Nucl. Chem. <u>34</u>, 3851

(1972).

- [HUB80] F. Hubert, A. Fleury, R. Bimbot, and D. Gardes, Ann. Phys. 5, 1 (1980).
- [JAC56] J.D. Jackson, Can. J. Phys. <u>31</u>, 767 (1956).
- [KAD90] B. Kadkhodayan, unpublished (1990).
- [KAU59] R. Kaufmann and R. Wolfgang, Phys. Rev. Lett. 3, 232 (1959).
- [KAU60] R. Kaufmann and R. Wolfgang, Phys. Rev. <u>121</u>, 192 (1960).
- [KRA77] J.V. Kratz, H. Ahrens, W. Bögl, W. Brüchle, G. Franz, M. Schädel,
  I. Warnecke, G. Wirth, G. Klein, and M. Weis, Phys. Rev. Lett. <u>39</u>, 984 (1977).
- [LEE82] D. Lee, H.R. von Gunten, B. Jacak, M. Nurmia, Y. Liu, C. Luo, G.T. Seaborg, H.R. von Gunten, and D.C. Hoffman, Phys. Rev. C 25, 286 (1982).
- [LEE83] D. Lee, K.J. Moody, M.J. Nurmia, G.T. Seaborg, H.R. von Gunten, and D.C. Hoffman, Phys. Rev. C 27, 2656 (1983).
- [LER87] R.G. Leres, Lawrence Berkeley Laboratory Preprint LBL-24808, (1987).
- [LEY90] J.D. Leyba, R.A. Henderson, H.L. Hall, C.M. Gannett, R.B. Chadwick
   K.R. Czerwinski, S.A. Kreek, G.R. Haynes, K.E. Gregorich, D.M. Lee,
   M.J. Nurmia, and D.C. Hoffman, Phys. Rev. C 41, 2092, (1990).
- [LEY90A] J.D. Leyba, R.A. Henderson, H.L. Hall, K.R. Czerwinski,
   B.A. Kadkhodayan, S.A. Kreek, E.K. Brady, K.E. Gregorich, D.M. Lee,
   M.J. Nurmia, and D.C. Hoffman, submitted to Phys. Rev. C.
- [LEY90B] J.D. Leyba (unpublished).
- [MAS71] D.L. Massart, <u>Cation-Exchange Techniques in Radiochemistry</u>, NAS-NS-3113 (1971).
- [MCF82] R.M. McFarland, Ph.D. thesis, University of California, 1982.
- [MCF83] R.M. McFarland, A. Ghiorso, and G.T. Seaborg, Lawrence Berkeley Laboratory Preprint LBL-15529 1983.

- [MOR81] L.G. Moretto, Rep. Prog. Phys. <u>44</u>, 533 (1981).
- [MOO83] K.J. Moody, Ph.D. thesis, University of California, 1983.
- [MOO86] K. Moody, D. Lee, R.B. Welch, K.E. Gregorich, G.T. Seaborg, R.W. Lougheed, and E.K. Hulet, Phys. Rev. C <u>33</u>, 1315 (1986).
- [MYE74] W.D. Myers and W.J. Swiatecki, Ann. Phys. <u>84</u>, 186 (1974).
- [MYE80] W.D. Myers, <u>Droplet Model of Atomic Nuclei</u>, IFI/Plenum, New York (1980).
- [MYL11] F. Mylius and C. Hüttner, Bec. Deut. Chem. Ges. <u>44</u>, 1315 (1911).
- [NOR70] R.C. Northcliffe and R.F. Schilling, Nucl. Data Tables <u>A7</u>, 233 (1970).
- [NÖR75] W. Nörenberg, Z. Phys. <u>A274</u>, 241 (1975).
- [PAL88] C.E.A. Palmer, H.L. Hall, P.A. Baisden, D.C. Hoffman, and D.M. Lee,
  LLNL Nuclear Chemistry Division Annual Report FY 1988, Report No.
  UCAR 10062-88 (1988).
- [PEP57] P.F. Peppard, G.W. Mason, J.L. Maier, and W.J. Driscoll, J. Inorg. Nucl. Chem. 4, 334 (1957).
- [RAN79] J. Randrup, Nucl. Phys. <u>A327</u>, 490 (1979).
- [RAN86] J. Randrup, Nucl. Phys. <u>A452</u>, 105 (1986).
- [RIE79] C. Riedel and W. Nörenberg, Z. Phys <u>A290</u>, 385 (1979).
- [ROU69] J.T. Routti and S.G. Prussin, Nucl. Inst. Meth. <u>72</u>, 125 (1969).
- [SCH77] W.U. Schröder and K.A. Dill, Ann. Rev. Nucl. Sci. <u>27</u>, 465 (1977).
- [SCH78] M Schädel, J.V. Kratz, H. Ahrens, W. Brüchle, G. Franz, H. Gäggeler,
  I. Warnecke, G. Wirth, G. Herrmann, N. Trautmann, and W. Weis,
  Phys. Rev. Lett. 41, 469 (1978).
- [SCH81] W.U. Schröder, J.R. Huizenga, and J. Randrup, Phys. Lett. <u>98B</u>, 355 (1981).
- [SCH82] M. Schädel, W. Brüchle, H. Gäggeler, J.V. Kratz, K. Sümmerer,G. Wirth, G. Herrmann, R. Stakemann, G. Tittel, N. Trautmann,

- J.M. Nitschke, E.K. Hulet, R.W. Lougheed, R.L. Hahn, and R.L. Ferguson, Phys. Rev. Lett. 48, 852 (1982).
- [SCO77] D.K. Scott, Lawrence Berkeley Laboratory Preprint LBL-7111 (1977).
- [SEE67] P.A. Seeger and R.C. Perisho, Los Alamos Scientific Laboratory Report LA-3751, September 1967.
- [SIE71] P.J. Siemens, J.P. Bondorf, D.H.E. Gross, and F. Dickmann, Phys. Lett. <u>36B</u>, 24 (1971).
- [SIK68] T. Sikkeland, A. Ghiorso, and M.J. Nurmia, Phys. Rev. <u>172</u>, 1232 (1968).
- [SIK72] T. Sikkeland, N.H. Shafrir, and N. Trautmann, Phys. Lett <u>42B</u>, 201 (1972).
- [SMI56] H.L. Smith and D.C. Hoffman, J. Inorg. Nucl. Chem. 3, 243 (1956).
- [STE61] P.C. Stevenson and W.E. Nervik, <u>The Radiochemistry of the Rare Earth's</u>, <u>Scandium</u>, <u>Yttrium</u>, and <u>Actinium</u>, NAS-NS-3020 (1961).
- [STE66] P.C. Stevenson, <u>Processing of Counting Data</u>, NAS-NS-3109 (1966).
- [TÜR89] A Türler, Ph.D. thesis, University of Bern, 1989.
- [UDA82] T. Udagawa, D. Price, and T. Tamura, Phys. Lett. <u>118B</u>, 45 (1982).
- [VAN84] R. Vandenbosch, A. Lazzarini, D. Leach, P.K. Lock, A. Ray, and A. Seamster, Phys. Rev. Lett. <u>52</u>, 1964 (1984).
- [WAL84] F.W. Walker, D.G. Miller, and F. Feiner, <u>Chart of the Nuclides</u>,
  Thirteenth Edition, General Electric, San Jose, 1984.
- [WEI40] V.F. Weisskopf and D.H. Ewing, Phys. Rev. <u>57</u>, 472 (1940).
- [WEL85] R.B. Welch, Ph.D. thesis, University of California, 1985.
- [WEL87] R. Welch, K.J. Moody, K.E. Gregorich, D.M. Lee, and G.T. Seaborg, Phys. Rev. C 35, 204 (1987).
- [WIL80] W.W. Wilcke, J.R. Birkelund, H.J. Wollersheim, A.D. Hoover,J.R. Huizenga, W.U. Schröder, and L.E. Tubbs, At. Data Nucl. Data

Tables, 25, 389 (1980).

[WIL80A] J. Wilczynski, K. Siwek-Wilczynska, J. van Driel, S. Gonggrijp,
 D.C.J.M. Hageman, R.V.F. Janssens, J. Lukasiak, and R.H. Siemssen,
 Phys. Rev. Lett. 45, 606 (1980).

[WIN63] J.W. Winchester, J. Chromatog. <u>10</u>, 502 (1963).

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