DESIGN OF A PULSED X-RAY SYSTEM FOR FLUORESCENT LIFETIME **MEASUREMENTS WITH A TIMING ACCURACY OF 109 PS***

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Beryllium Exit Window

ABSTRACT

We describe the design of a table-top pulsed x-ray system for measuring fluorescent lifetime and wavelength spectra of samples in both crystal and powdered form. The novel element of the system is a light-excited x-ray tube with a tungsten anode at +30 kV potential. The S-20 photocathode is excited by a laser diode with a maximum rate of 10 MHz, each pulse having <100 ps fwhm (full-width at halfmaximum) and >10⁷ photons. In a collimated 2 mm \times 2 mm beam spot 40 mm from the anode we expect >1 x-ray per pulse. A sample is exposed to these x-rays and fluorescent photons are detected by a microchannel PMT with a photoelectron transit time spread of 60 ps fwhm, a sapphire window, and a bialkali photocathode (wavelength range 180-600 nm). The combined time spread of a laser diode, the x-ray tube, and a microchannel tube has been measured to be 109 ps fwhm. To measure wavelength spectra, a reflection grating monochromator is placed between the sample and the PMT.

1 BACKGROUND AND INTRODUCTION

In previous work, an electron synchrotron was used in single-bunch mode to measure the x-ray excited fluorescence of over 400 compounds. Interesting fluorescent emissions were detected from several compounds by this method, including CeF3, PbCO3, PbSO4, Yb2O3, CeCl3, BaCl2, and CuI [1, 2]. At the same time, a light-excited x-ray tube was developed for producing very brief (= 100 ps) pulses of x-rays (Figure 1) [3].

In this work, we describe the design of a table-top pulsed x-ray fluorescence measurement system using a laser diode and a light-excited x-ray tube. This system has a number of advantages over the electron synchrotron technique. Among these are (i) measuring samples without having to transport personnel and equipment to a remote site, (ii) measuring samples immediately rather than waiting for the next single-bunch beam run, (iii) a narrower x-ray pulse width (<100 ps), and (iv) the ability to vary the excitation repetition period from 100 ns to well beyond 10 μ s.

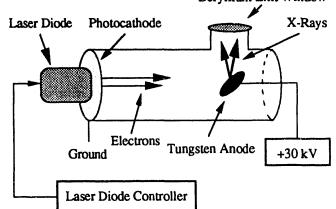


Figure 1. Schematic diagram of the pulsed x-ray source. A laser diode excites the photocathode with 60 ps pulses at ≤10 MHz and the photoelectrons are accelerated toward the tungsten anode at 30 kV potential to produce x-rays.

The method of Bollinger and Thomas [4] (as modified by Moszynski and Bengtson [5] for coincident 511 keV annihilation photons) is also used to measure fluorescent decay spectra. However, the timing accuracy is limited to about 200 ps fwhm by jitter in the trigger scintillator, and it is difficult to determine the fraction of slow components.

2 PULSED X-RAY SOURCE

2.1 Laser Diode

The PLP-01 Picosecond Light Pulser consists of a controller and a laser diode head. The pulse width and number of photons per pulse depend only weakly on the pulse rate, which can be varied from 1 Hz to 10 MHz. Each light pulse is preceded by a timing pulse with a time jitter of ± 10 ps. In addition, the laser diode can be externally triggered or operated in dc mode. See Table 1 for additional details.

Table 1. Characteristi	cs of Hamamatsu	PLP-01 Light Pulser
with C3551-01 Control	ller and LDH065	Laser Diode Head

Emission wavelength	650 nm
Peak power	>100 mW
Pulse width	50- 100 ps fwhm
Pulse repetition rate	dc to 10 MHz
Photons per pulse	> 10 ⁷
Timing pulse jitter	±10 ps

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2.2 Light-excited X-ray Tube

The light-excited x-ray tube (Figure 1), is essentially a single-stage photomultiplier tube. It has an S-20 photocathode at ground potential and a tungsten anode at +30 kV. Light liberates photoelectrons at the photocathode, which are accelerated to 30 keV to strike the tungsten target, producing bremsstrahlung x-rays. The photocathode to anode transit time is 0.8 ns and the anode surface is oriented at 45° with respect to the incident electron beam. See Table 2 for additional details.

The maximum anode current of 50 μ A corresponds to a maximum photon rate of 3 × 10¹⁵/s at 10 % quantum efficiency. At 10 MHz and 2 × 10⁷ photons per pulse, the laser diode will provide 2 × 10¹⁴ photons/s.

Table 2. Characteristics of Hamamatsu N5084 Light-excited X-Ray Tube

Overall tube length	152 mm
Tube Diameter	52 mm
Photocathode	S-20
Quantum efficiency @ 650 nm	> 10%
Photocathode diameter	12 mm
Target material (anode)	Tungsten (45°)
Output window material	Beryllium
Output window diameter	20 mm
Output window thickness	0.5 mm
Cooling	Natural air
Tube voltage (max)	30 kV
Tube current (max)	50 µA
Photocathode photon rate (max)	3×10^{15} /s @ 10% QE

3 RESULTS

3.1 Measurement of Timing Response

The light-excited x-ray tube was tested by illuminating its photocathode with laser diode pulses and measuring the x-rays with a Be-window microchannel electron multiplier (Figure 2). The laser diode was coupled to the x-ray tube with an optical fiber. The measured time distribution of detected x-rays is shown in Figure 3. The 109 ps fwhm includes three components: the width of the laser diode pulses (55 ps), time spread in the light-excited x-ray tube, and the timing resolution of the microchannel tube (57 ps). The contribution of the lightexcited x-ray tube is inferred to be 75 ps.

3.2 System for Fluorescence Measurements

Figure 4 shows the system for measuring fluorescence decay timing spectra from powder or crystal samples. The delayed coincidence method is used to record the time distribution between the x-ray pulses and the individual fluorescent photons.

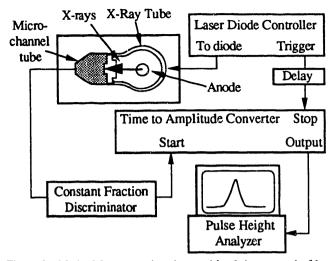


Figure 2. Method for measuring the combined time spread of laser diode, light-excited x-ray tube, and microchannel x-ray detector. The x-rays pass through Be windows to interact directly with the microchannel multiplier.

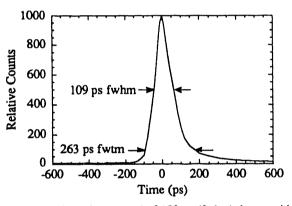


Figure 3. Combined time spread of 109 ps (fwhm) due to width of laser diode pulse (55 ps), and time spreads in the light-excited x-ray tube (75 ps) and microchannel x-ray detector (57 ps). The spectrum was measured with the setup shown in Figure 2, using an x-ray tube voltage of 10 kV.

The laser diode head is proximity coupled to the x-ray tube photocathode for maximum photon rate. The microchannel PMT has a bialkali photocathode, a sapphire window, a wavelength range from 180 to 600 nm, and a 60 ps transit time spread. The PMT is thermoelectrically cooled to -20° C to reduce the dark rate to a few photoelectrons/s.

Emission spectra can be measured by inserting a monochromator between the sample and the microchannel phototube. To extend the upper limit to 900 nm, a phototube with a quartz window, a GaAs photocathode, and 4 ns transit time spread can be used.

The sample chamber is evacuated to a pressure below 10^{-3} mm Hg to avoid fluorescence from air. A current meter at the photocathode is used to monitor the photoelectron current and a current meter at the power supply is used to monitor the sum of the anode current and any leakage currents to ground.

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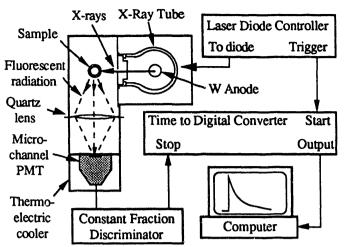


Figure 4. System for exciting samples with <100 ps pulses of xrays and measuring the fluorescent photons with a microchannel phototube. See text for details.

For measuring the system time response for the fluorescence time measurement system shown in Figure 4, we plan to position the microchannel PMT in the x-ray beam at an angle to produce photoelectrons at the photocathode. This measurement includes (i) the laser diode pulse shape, (ii) time spread in the x-ray tube, (iii) photoelectron time spread in the PMT between the photocathode and the microchannel plate and (iv) electron time spread in the microchannel multiplier.

Safety interlocks allow the high voltage power supply to be turned on only after the high voltage cables and sample chamber are connected and the sample chamber is evacuated. The high voltage supply trips off whenever the current exceeds 50 μ A or an ionization chamber monitor exceeds a set limit.

The system shown in Figure 4 has a parts cost of about 50,000 (U.S.), with the light-excited x-ray tube, the microchannel PMT, and the laser diode being the most expensive components.

3.3 Data Acquisition and Analysis

Time differences between the x-ray excitation and the arrival of fluorescent photons are digitized by a time to digital converter (TDC) with a bin width of 50 ps and a maximum dynamic range of 800 μ s, corresponding to the maximum 24-bit CAMAC data number. The TDC can detect multiple photon detections in the conversion time range, and these events are rejected to reduce bias and permit substantially higher data rates [6]. A compression algorithm preserves the 50 ps bin width near the time of the excitation but combines data into progressively wider bins at later times. In a typical case, 2×10^5 bins spanning a 10 μ s conversion range are compressed into 2×10^3 bins for fitting. The fitting program determines fractions, decay times, and rise times of fluorescent components by fitting a sum of exponentials (convolved with the system timing response) to the compressed data [7].

4 CONCLUSIONS

By combining a state-of-the-art laser diode and a recently developed light-excited x-ray tube, we have designed a tabletop system for measuring fluorescent intensities and decay times with a timing accuracy of about 100 ps. The data acquisition uses photon counting and signal averaging, so that accurate measurements are possible for even weakly fluorescing samples. The system is more accurate than methods using coincident nuclear radiation and more accurate and convenient than methods using synchrotron x-radiation.

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Reference to a company or product name does not imply approval or recommendation by the University of California or the U.S. Department of Energy to the exclusion of others that may be suitable.

REFERENCES

- S. E. Derenzo, W. W. Moses, R. Perera, et al., "Prospects for new inorganic scintillators," *IEEE Trans Nucl Sci*, vol. NS-37, pp. 203-208, 1990.
- [2] S. E. Derenzo, W. W. Moses, J. L. Cahoon, et al., "X-ray fluorescence measurements of 412 inorganic compounds," *IEEE Nuclear Science Symposium Conference Record* 91CH3100-5, vol. 1, pp. 143- 147, 1992.
- [3] K. Oba, "Instrument to measure fluorescence which has occurred in a sample stimulated by x-rays," U.S. patent #4,724,536, issued to Hamamatsu Photonics, Japan, Feb 9, 1988.
- [4] L. M. Bollinger and G. E. Thomas, "Measurement of the time dependence of scintillation intensity by a delayedcoincidence method," *Rev Sci Instr*, vol. 32, pp. 1044-1050, 1961.
- [5] M. Moszynski and B. Bengtson, "Light pulse shapes from plastic scintillators," Nucl Instr Meth, vol. 142, pp. 417-434, 1977.
- [6] W. W. Moses, "A method to increase scintillation lifetime measurement rates using a multi-hit TDC," Nucl Instr Meth (in preparation), 1993.
- [7] S. Derenzo, "TAUFIT- a fitting program to determine fluorescent decay components as sum of exponentials convolved with an impulse response," Lawrence Berkeley Laboratory Report No. LBL-33050, 1993.



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