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**A Mathematical Model for
Multicomponent Separations on the
Continuous Annular Chromatograph**

Robert L. Bratzler
John M. Begovich

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CHEMICAL TECHNOLOGY DIVISION

A MATHEMATICAL MODEL FOR MULTICOMPONENT SEPARATIONS
ON THE CONTINUOUS ANNULAR CHROMATOGRAPH

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A MATHEMATICAL MODEL FOR MULTICOMPONENT SEPARATIONS
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HIGHLIGHTS

A model for multicomponent separations on ion exchange columns has been adapted for use in studying the performance of the continuous annular chromatograph. The model accurately predicts solute peak positions in the column effluent and qualitatively predicts trends in solute effluent resolution as a function of increasing bandwidth of the solute feed pulse. The major virtues of the model are its simplicity in terms of the calculations involved and the fact that it incorporates the nonlinear solute-resin binding isotherms common in many ion exchange separations. Because dispersion effects are not accounted for in the model, discrepancies exist between the shapes of the effluent peaks predicted by the model and those determined experimentally.

1. INTRODUCTION

Conventional chromatographic separations have been modeled using both numerical and analytical calculations. Only the numerical methods have been successful in incorporating all of the effects which occur in chromatography. Unfortunately, the calculations are usually long and involved and do not yield much insight into the physics of the process. On the other hand, the analytical methods usually involve assumptions which yield only approximate results of the real experiment. However, they do provide a theoretical basis for predicting the results of a chromatographic separation with fewer calculations than with the numerical methods.

If the discussion is restricted to analytical models developed by ion exchange chromatographic separations, it must include the following effects in the most general model: (1) solute and solvent convection axially through the bed (column); (2) solute dispersion and diffusion, both parallel and normal to the direction of flow; and (3) nonlinear sorption and desorption of the solute by the bed packing. To date no analytical model has been developed which includes all of these effects. A model does exist for unidimensional convection and dispersion with linear binding isotherms.¹ This model is adequate for those instances in which the concentrations of solutes are much smaller than the concentrations at which the bed packing would be saturated (overloaded). For those cases in which the solute loading is relatively high, an analytical model has been developed by Rhee² which treats nonlinear solute binding isotherms, but it does not include diffusion and dispersion effects. Others³ have developed models based on the idea that the column may be approximated as a number of ideal mixing cells, all connected in series. Although this approach is based on relatively crude geometrical assumptions, the results predicted have been shown to agree fairly well with experimental findings, provided the bed is sufficiently long.

Both cell models⁴ and numerical models⁵ have been employed in describing the continuous annular chromatograph (CAC) developed at the Oak Ridge National Laboratory. A schematic diagram of the apparatus is shown in Fig. 1. A regulated gas overpressure provides the driving force for a constant flow rate of eluent through the column. The feed inlet is held stationary while the column is rotated, subjecting each portion of the annulus to feed for a short period of time followed by

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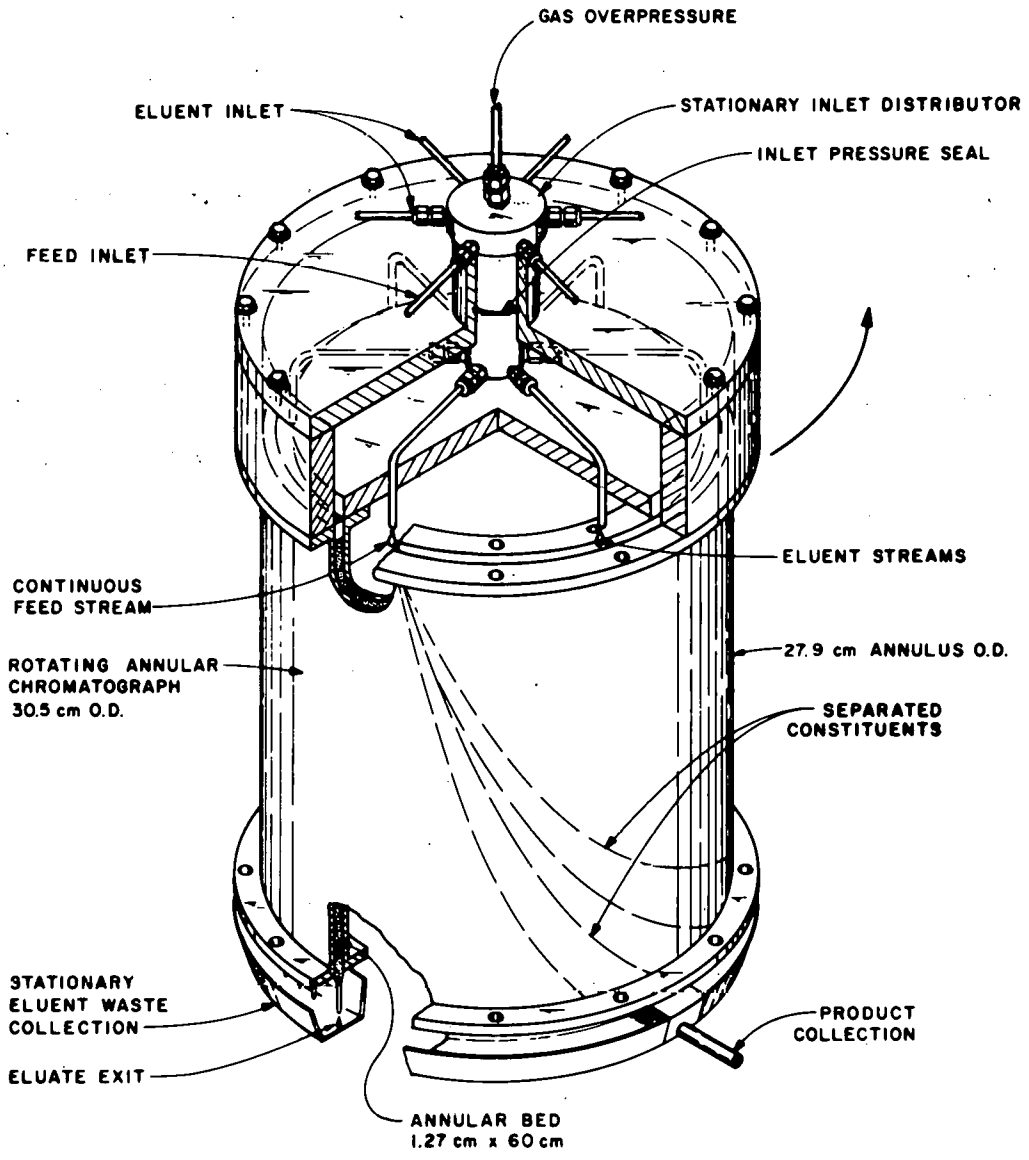


Fig. 1. Schematic of the pressurized continuous annular chromatograph.

elution by the eluent stream. As elution proceeds, the eluted substances separate as they progress vertically down the annulus, giving the appearance of helices as the annulus rotates. Component species with a high affinity for the column packing exit from the rotating annulus at a greater distance from the feed point compared to species with lower distribution coefficients. Thus, multicomponent separations can be made on a continuous basis depending on the sorption characteristics of the component species.

The purpose of the work reported here is to assess the applicability of the Rhee analytical model to the performance of the CAC. The following sections describe the model, and the results predicted by the model are compared with experimental results.

2. SUMMARY OF THE ANALYTICAL MODEL

Consider a conventional chromatographic column to which a liquid solute mixture is fed for some short-time interval, t_p , and is followed by a stream containing just solvent (the eluent stream). If there is sufficient variation in the affinities of the individual solutes for the column packing material, the solutes will appear separated in the effluent stream. The governing partial differential material balance on the column is:

$$D_i \frac{\partial^2 C_i}{\partial x^2} - \frac{v}{\epsilon} \frac{\partial C_i}{\partial x} = \frac{\partial C_i}{\partial t} + \frac{(1-\epsilon)}{\epsilon} \frac{\partial B_i}{\partial t} \quad (1)$$

$$\frac{\partial B_i}{\partial t} = k_{fi} C_i \left(N - \sum_{j=1}^n B_j \right) - k_{ri} B_i, \quad (2)$$

where the nomenclature used is given in Table 1. Equation (1) results from a material balance performed on the mobile (liquid) phase, whereas Eq. (2) applies to the immobile (solid) phase. These equations are general and cannot be solved without simplifying the assumptions and specifying the initial and boundary conditions. For the case considered here, these latter conditions are:

$$t = 0, \text{ all } x: C_i = B_i = 0 \quad (3)$$

$$0 < t \leq t_p, x = 0: C_i = C_{i0}$$

$$t > t_p, x = 0: C_i = 0$$

$$\lim_{x \rightarrow \infty} C_i = 0$$

Thus, the column is initially free of solute; up to time, t_p , the column feed contains solute of some constant concentration; thereafter, solute-free eluent is fed to the column. Neglecting column end effects, it is assumed that the concentration vanishes as the axial position approaches infinity.

The model is simplified by neglecting the second-order dispersion terms in Eq. (1). A second simplification is to assume that the binding reaction [Eq. (2)] is sufficiently fast so that local chemical equilibrium exists throughout the column. This second assumption is valid in most cases, because the flow rates through ion exchange columns are relatively slow and the binding reactions are quite fast. The first assumption is made, not from physical insight, but rather to facilitate an analytical solution to the problem. Indeed, dispersion effects play a large role in determining the performance of chromatographic separations.

Table 1. Nomenclature

Symbol	Definition and units
B_i	Concentration of solute i bound to resin, meq/ml resin
C_i	Concentration of solute i in liquid phase, meq/ml solution
C_{i0}	Concentration of solute i in liquid feed, meq/ml solution
D_i	Dispersion coefficient for solute i , cm^2/sec
k_{fi}	Forward binding-rate constant for solute i , (ml solution)/(sec-meq)
k_{ri}	Reverse binding-rate constant for solute i , sec^{-1}
K_i	k_{fi}/k_{ri} = binding constant, ml solution/meq
L	Overall bed length, cm
N	Resin capacity, meq/ml resin
n	Number of solutes in system
t_p	Duration of solute feed pulse, sec
t	Time, h
v	Eluent superficial velocity, cm/min
x	Axial coordinate, cm
ϵ	Bed void fraction, ml voids/ml bed
θ	Angular displacement, deg arc
ϕ_i	$K_i C_i$, dimensionless transformed solute i concentration
ω	Annular bed rotation rate, deg arc/h

Even with these simplifying assumptions, Eqs. (1) and (2) are difficult to solve because of the nonlinearity in the concentration terms in Eq. (2). For instantaneous binding, Eq. (2) may be rearranged into the following form:

$$B_i = \frac{K_i N C_i}{1 + \sum_{j=1}^n K_j C_j} \quad (4)$$

This equation is often called the Langmuir binding isotherm. A plot of this function is shown in Fig. 2. It adequately describes the "saturability" of the ion exchange resin as the concentration of solute in the mobile phase is increased. This equation is based on the concept that the resin has only a finite number of binding sites. As sites become filled, the tendency of any particular solute to bind decreases. This is an important consideration in the description of multicomponent chromatography, because the presence of a solute with a higher affinity for the resin will effectively displace solutes with lower resin affinities. The linear binding model described in ref. 1 does not include this effect.

The problem as posed has been solved analytically.² To apply the solution to the CAC, in which solute bands are distributed both axially and circumferentially, only a simple transformation from time to circumferential displacement is needed:

$$\theta = \omega t \quad (5)$$

Thus, these two coordinates may be viewed as interchangeable.

The solution to the problem is best understood from plots in the x-t (or x- θ) plane. Consider the separation of a three-component mixture (three solutes plus solvent). For clarity, the solutes are numbered in

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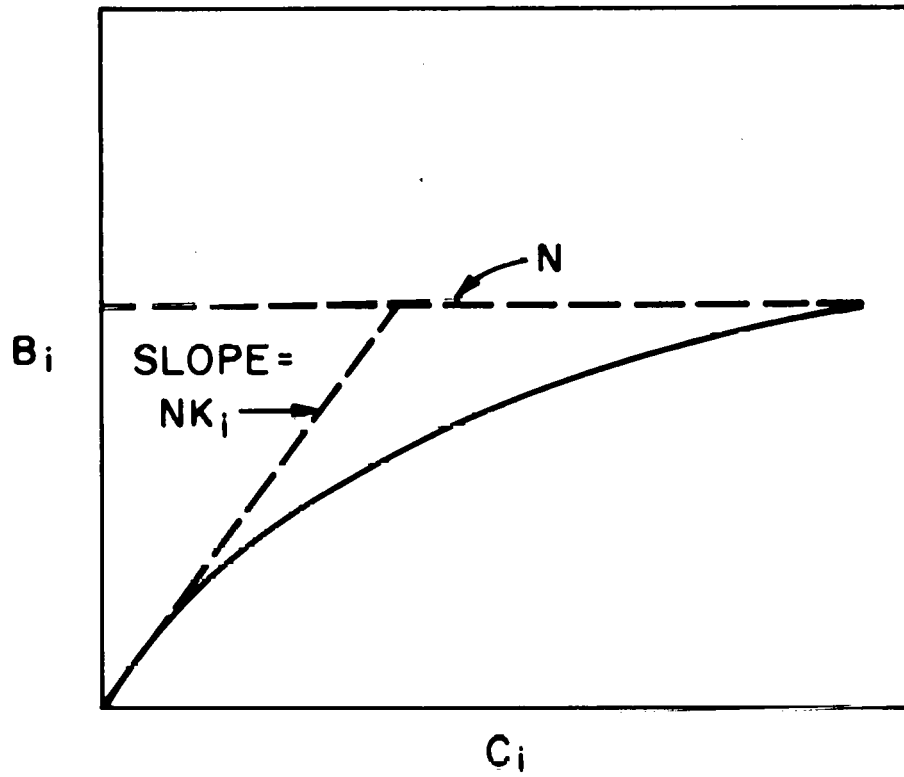


Fig. 2. Langmuir isotherm.

the order of increasing affinity for the resin, solute 1 having the lowest affinity and solute 3 having the highest. Because solute 3 will spend a greater proportion of time on the immobile phase, it will advance through the column at a slower rate than solutes 1 and 2. Thus if the column is initially solute-free when the mixture is applied ($t = 0$), a series of shock waves, one for each solute, will be established which demarcate the position of the solute fronts in the column. This is shown in Fig. 3. The solute fronts are sharp (i.e., the shock fronts) because dispersion has been neglected. Solute 1 front advances the fastest. The numbers of Fig. 3 indicate the solute content in the front and trailing regions near each of the three shock waves. To the left of shock 3, all three solutes coexist at concentrations equal to the respective feed concentrations. To the right of shock 3, only solutes 1 and 2 are present, and to the right of shock 2, only solute 1 exists.

At some later time, t_p , the column feed is switched to pure solvent, and elution of the column begins. Because the solvent does not interact with the resin, it will advance through the column faster than any of the solutes. Solute that resides on the resin will be removed progressively, starting with solute 1 (the solute with the lowest affinity for the resin). Thus, the mobile phase concentration of solute 1 will gradually decrease from its maximum value (= the inlet feed concentration) to zero. Graphically, this is represented by a fan of isoconcentration lines originating at $(0, t_p)$ as shown in Fig. 4. This family of lines may be regarded as the solvent front permeating the column.

There will be as many fans as there are solutes in the column, in this case, three. The fan for solute 1 will have smallest slopes (because

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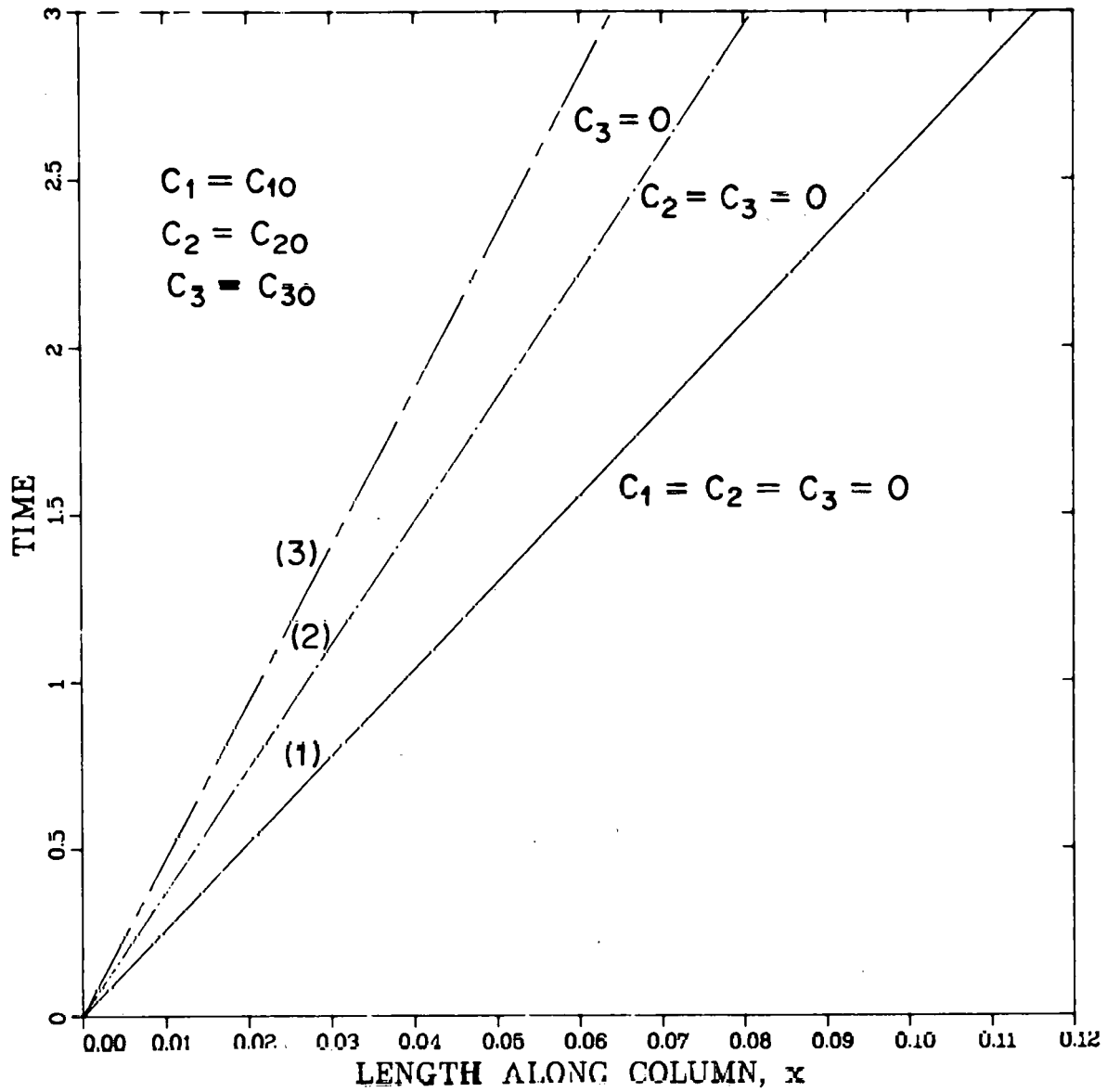


Fig. 3. Solute fronts advancing through a column in the loading of a three component mixture.

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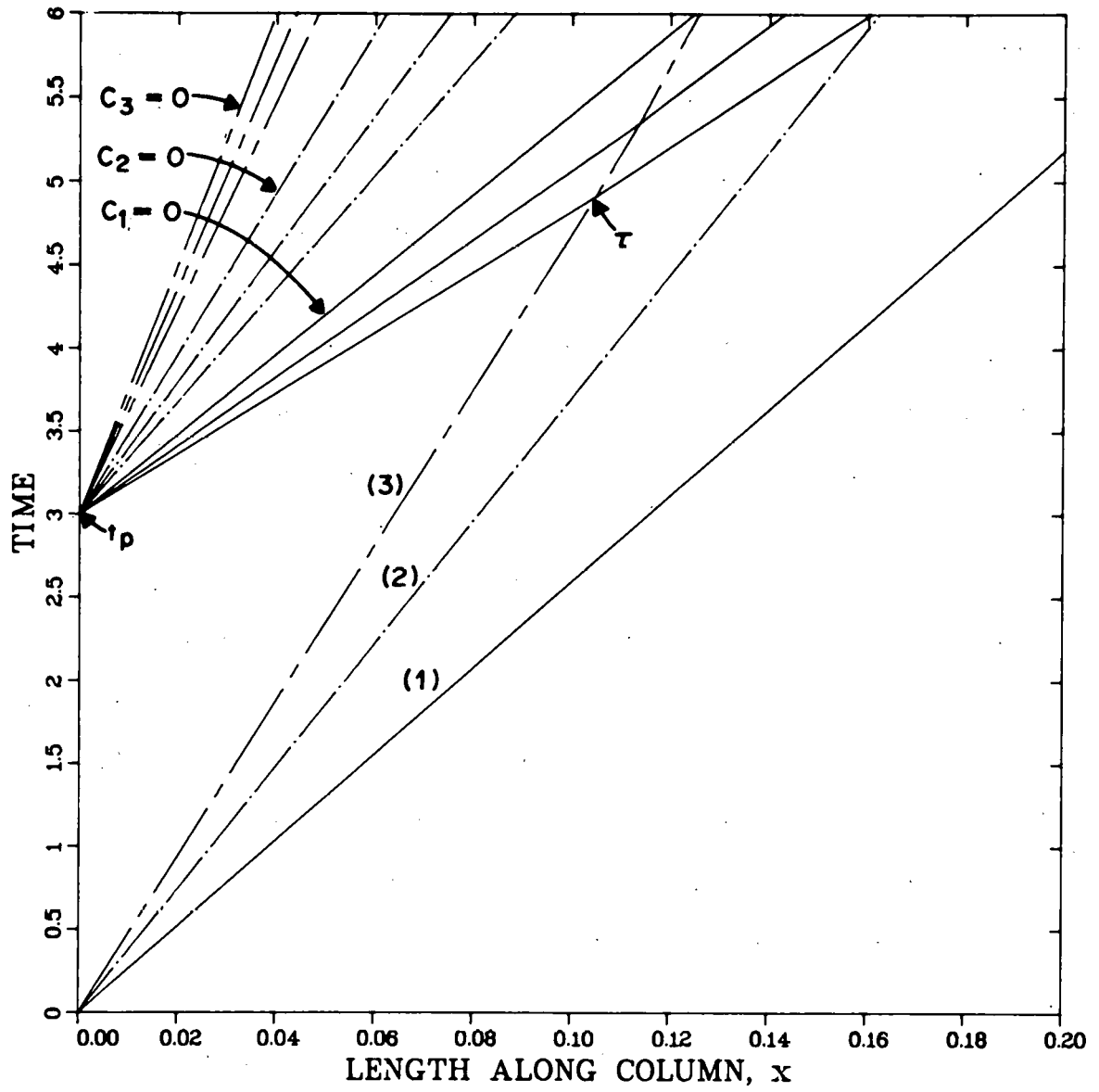


Fig. 4. Interaction of the isoconcentration lines in fan 1 with shock 3.

solute 1 is removed from the resin first). All three solutes are present along each of the isoconcentration lines in fan 1. At time, τ , fan 1 starts to overtake shock 3. This results in an interaction between the two waves. The net effect of the interaction is that shock 3 will move at a slower rate through the column (steeper slope) because there will no longer be any solute 1 competing for sites on the resin. The solvent fan 1 will also be slowed because there will be an absence of solute 3 to the right of shock 3. That is, solutes 1 and 2 will spend a greater proportion of time on the resin phase because competition from solute 3 is no longer a factor. Fan 1 will proceed down the column until it overtakes shock 2 and a similar transmissive interaction will ensue. In this case, solute 2 is depleted, leaving only solute 1 in fan 1. Again the speed at which the isoconcentration waves move through the column will be retarded due to the absence of solute 2 to the right of shock 2. This is shown in Fig. 5. When the last of fan 1 ($C_1 = 0$) has interacted with shock 2, solutes 1 and 2 will have been separated. This occurs at point A in Fig. 5. At times and bed lengths greater than those of point A, the separation distance between solutes 1 and 2 increases. Point A thus represents the minimum time and distance required to separate solutes 1 and 2.

After interaction with shock 2, fan 1 starts to interact with shock 1. Unlike the previous fan-shock interactions, the fan wave is not transmitted in this case. Instead, it is absorbed, thereby slowing the advance of shock 1. What ultimately results at long times and distances is shown on Fig. 6. The solvent front (fan 1) minimum velocity isoconcentration line (corresponding to $C_1 = 0$) moves at the same rate as the shock wave.

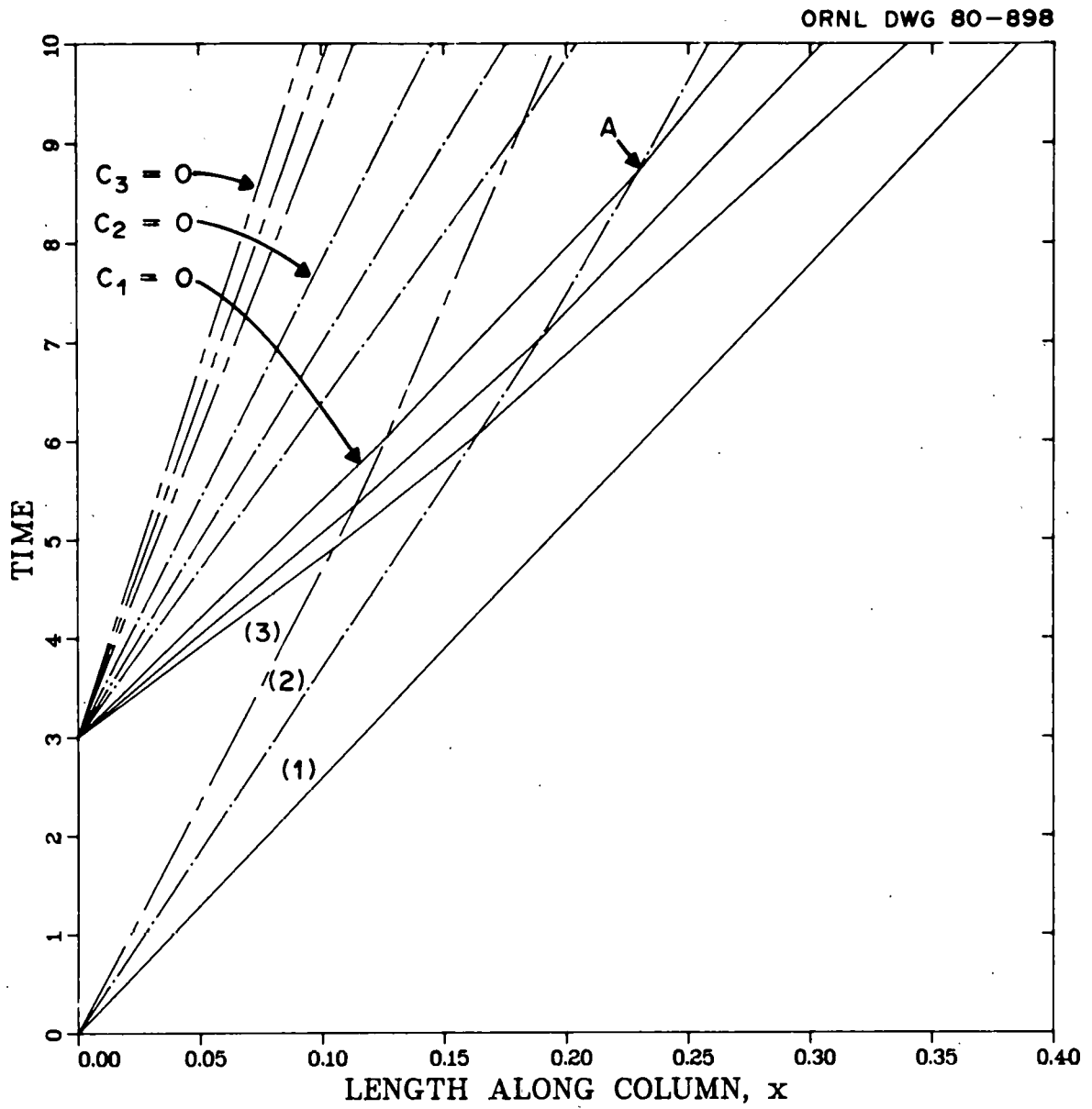


Fig. 5. Interaction of the isoconcentration lines in fan 1 with shock 2.

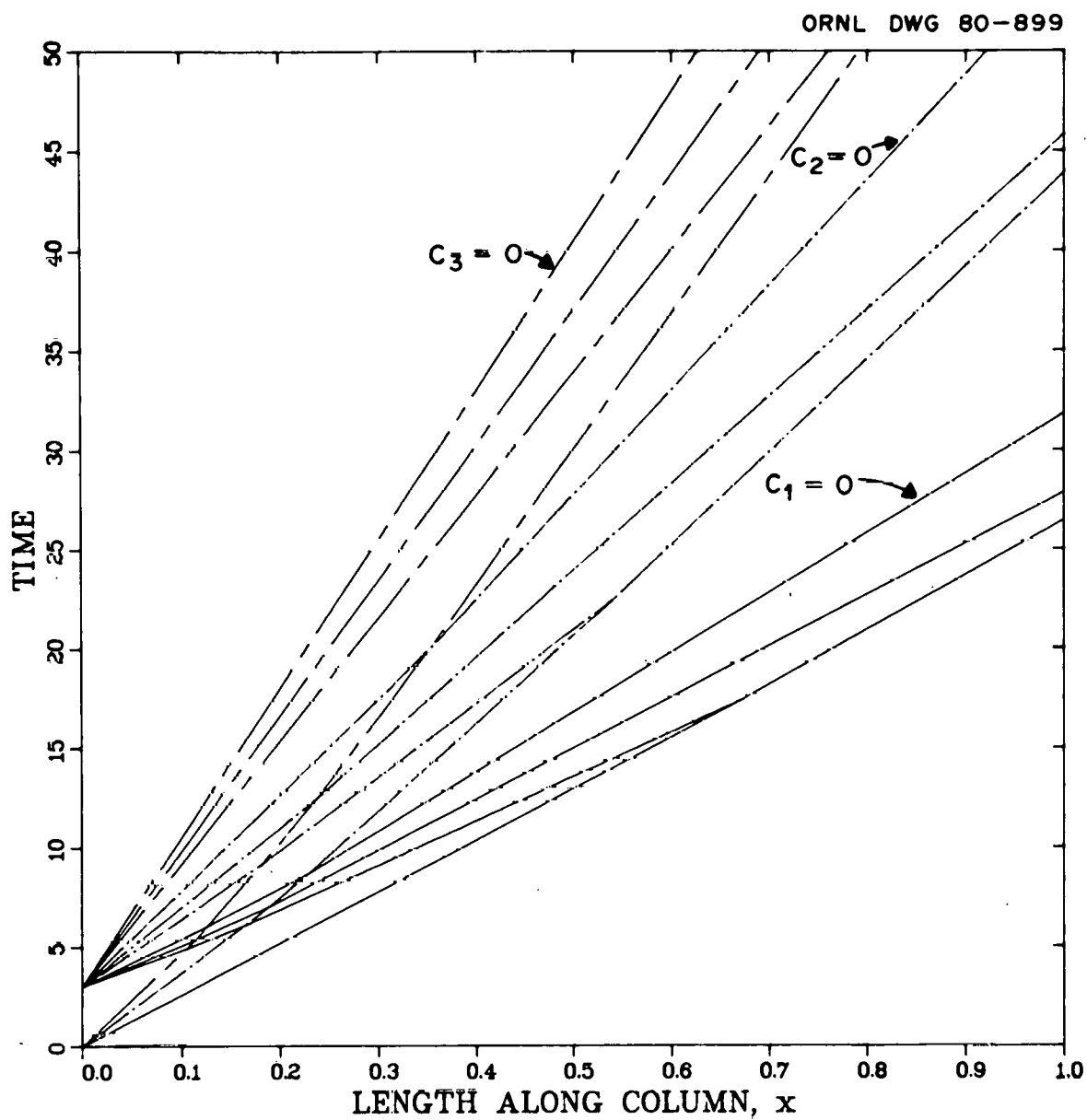


Fig. 6. Absorption of the isoconcentration lines in fan 1 with shock 1.

Thus, the two lines are parallel and separated by a constant distance at all bed lengths. This distance represents the solute 1 bandwidth.

Similar fan-shock interactions occur with the other waves. Typical final results are shown in Fig. 7 (only one fan line is shown for each fan wave). Point B depicts the time and position at which solute 2 is completely separated from solute 3. The effluent concentration profiles at any particular bed length may be determined by simply following along a vertical line through each of the shock and fan waves. Figures 8 and 9 show typical profiles for $x = 0.2$ and $x = 1.0$ respectively. At $x = 0.2$, there is incomplete separation between solutes 2 and 3. At $x = 1.0$, all three solutes have been separated.

To summarize, if one knows the inlet concentrations for each solute; the binding constants, K_i ; the void volume of the column, ϵ ; the eluent superficial velocity, v ; and the total capacity of the resin, N , the effluent concentration as a function of time at any arbitrary bed length, x , can be predicted. A computer program has been written to perform the calculations and is described in detail in Appendix A.

3. CASES STUDIED

The model was tested by performing calculations based on experimental data already generated on the CAC. The experiment separated copper, nickel, and cobalt (Co-1 and Co-2) amine complexes using Dowex 50W-X8 50- to 60- μm ion exchange resin (NH_4^+ form) with 1 M $(\text{NH}_4)_2\text{CO}_3$ at pH 7.88 as the eluent.⁶ The purpose of the experiments was to study the effect of variation in the feed bandwidth (t_p) on product resolution.

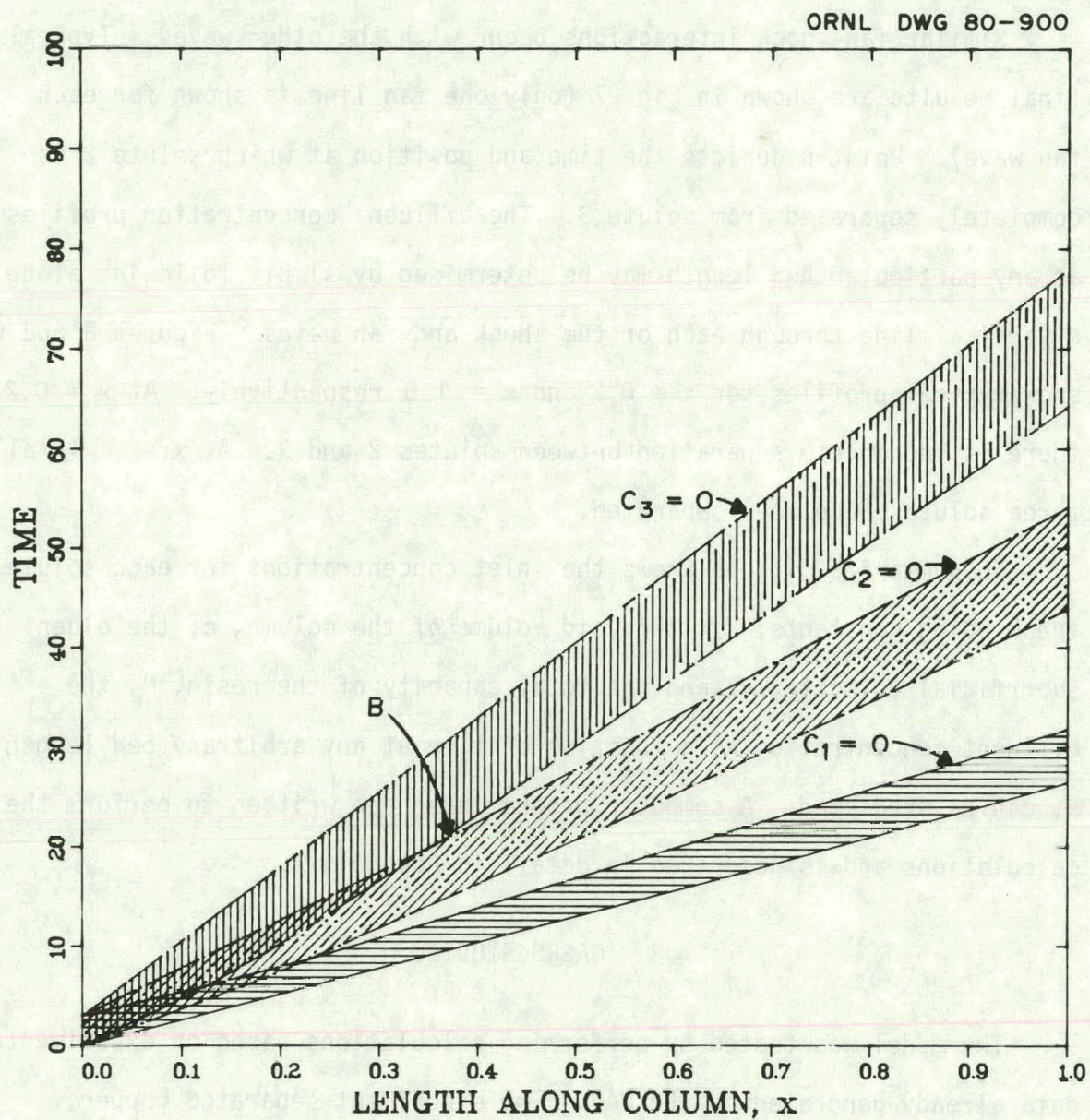


Fig. 7. Development of the solute bands along the column length.

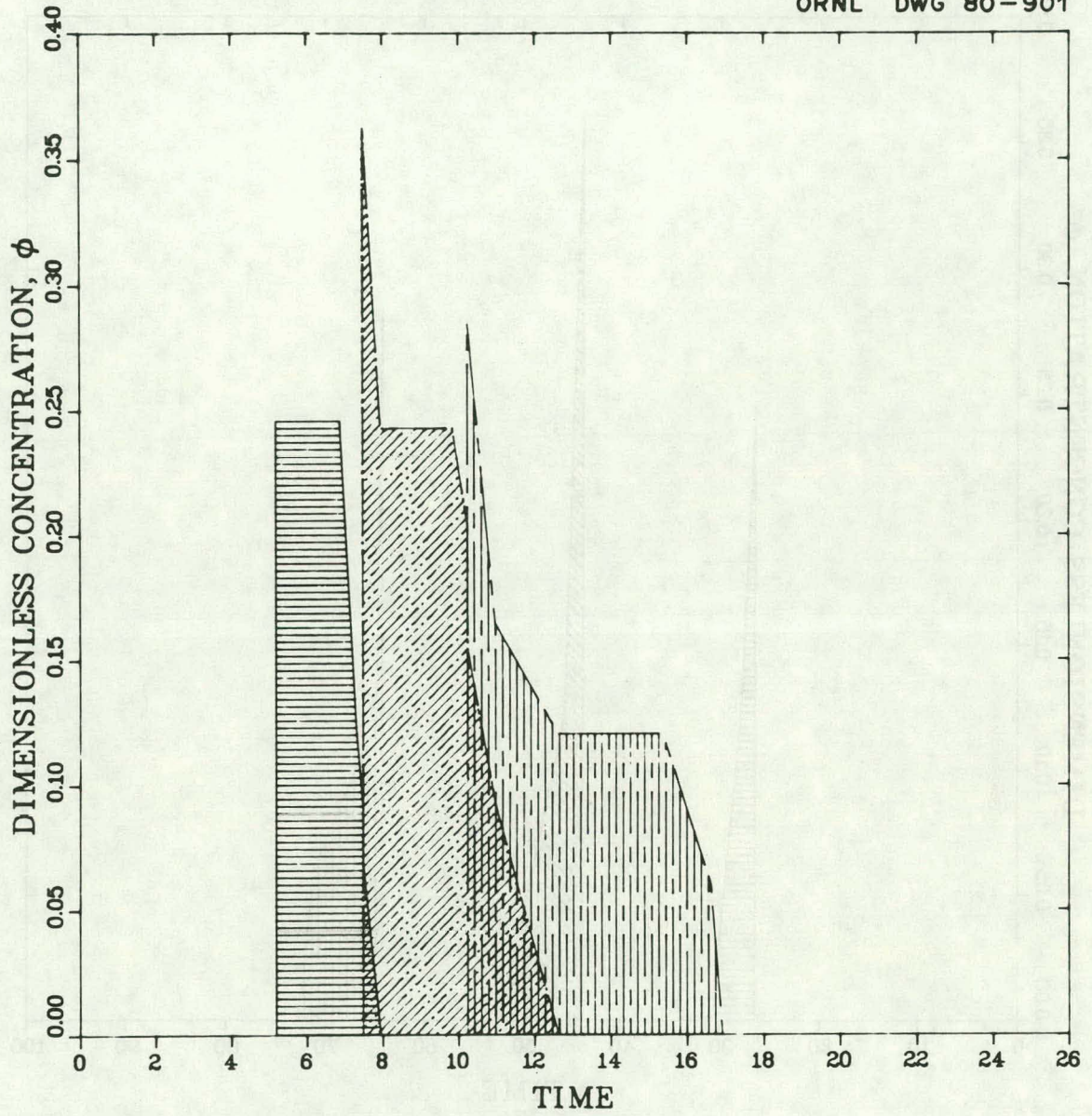


Fig. 8. Typical concentration profile at $x = 0.2$.

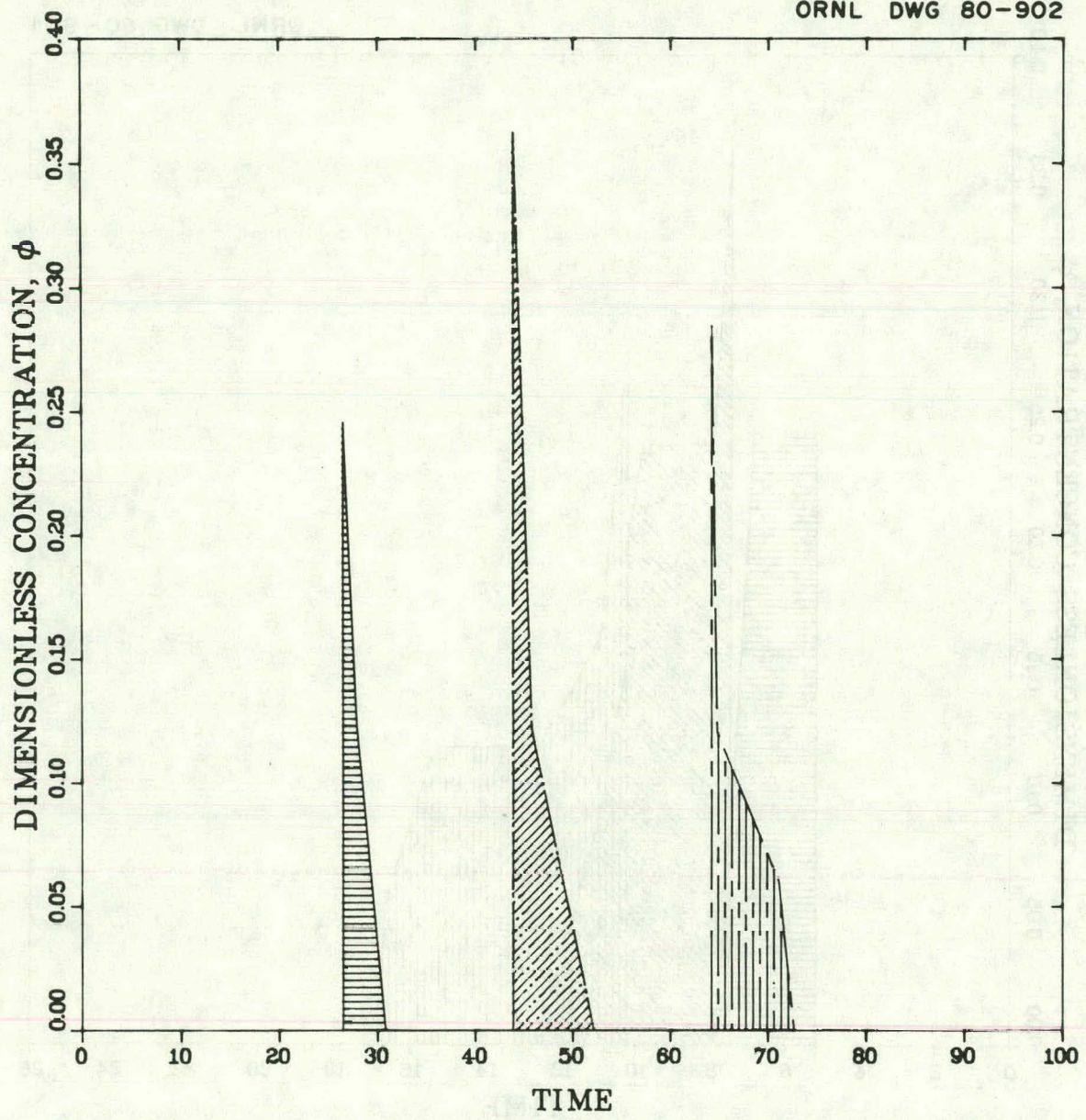


Fig. 9. Typical concentration profile at $x = 1.0$.

The binding constants for each of the solutes and the resin capacity used in the calculation are given in Table 2. The resin capacity used was the nominal capacity quoted by the resin manufacturer. The binding constants were determined by a trial-and-error method. Initial estimates were modified so that the results of the calculation matched those for one particular set of experimental conditions. These constants were not varied in any of the subsequent calculations. (Attempts to measure the binding constants in separate equilibrium binding experiments yielded inconsistent results, presumably due to the large errors involved in the experiments.) The algorithm used to determine the binding constants is fairly straightforward and is described in Appendix A.

Table 3 summarizes the experimental conditions used in the calculations. The feed bandwidth was varied from 5 to 38 mm. Because the bed was rotating at $62.2^\circ/\text{h}$ and the diameter of the annular bed is 28.2 cm, these bandwidths correspond to pulse widths ranging from 2 to 15 min. The feed bandwidth is varied experimentally by changing the solute mixture feed rate. When the feed rate exceeds the eluent rate, the feed band spreads. Conversely, at feed rates lower than the eluent rate, the feed pulse is confined to a narrower region and undoubtedly undergoes some dilution in the process. For the cases studied here, the eluent rate was 2.53 cm/min. For feed rates less than this eluent rate, the inlet solute concentrations were assumed to be diluted in direct proportion to the ratio of feed to eluent flow rates.

Table 2. Binding isotherm data used in the model calculations

Resin capacity: $N = 1.78$ meq/ml resin.

Solute complex	$\frac{K_i}{n}$ (ml solution/meq)	$\frac{K_{app}^a}{n}$ (ml solution/ml resin)
Cu	1.18	2.10
Ni	2.18	3.88
Co-1	2.92	5.20
Co-2	7.73	13.76

$^a K_{app}$ is the distribution coefficient at very low concentrations:

$$B_i = \frac{NK_i C_i}{1 + \sum_{j=1}^n K_j C_j} \approx K_{app} C_i \quad \text{if} \quad \sum_{j=1}^n K_j C_j \ll 1 .$$

Table 3. Experimental conditions used in the analytical model

$$v = 2.53 \text{ cm/min}$$

$$\epsilon = 0.38$$

$$L = 50 \text{ cm}$$

$$\text{Feed bandwidth} = 5 \text{ to } 38 \text{ mm}$$

$$\omega = 62.2^\circ/\text{h}$$

Solute complex	C_{io}^a [(meq/ml solution) $\times 10^2$]	C_{io}^a (g/liter)
Cu^{2+}	1.57	0.5
Ni^{2+}	13.63	4
Co-1^+	0.848	0.5
Co-2^{2+}	1.697	0.5

^aAssuming the feed rate \geq eluent rate. For the 5- and 7-mm bandwidth cases, these values were assumed to be 1.0/2.53 and 2.0/2.53 of the values shown here.

4. RESULTS

Bed concentration profiles were computed for the range of bandwidths listed in Table 3. The results predicted for a 7-mm feed bandwidth are shown in Fig. 10 on the $x/L-\theta$ plane. Solute bands for each solute in the system are represented by a pair of broken lines. The lowest band is for the copper complex, followed in order by the nickel, Co-1, and Co-2 complexes. This figure corresponds to the trajectory of the bands that would be observed during the operation of the CAC. In the calculations, L was assumed to be 50 cm, equal approximately to the overall bed length of the CAC. The effluent concentration can thus be determined by a cut at $x/L = 1$. This eluent profile is shown in Fig. 11 where $\phi_i (= K_i C_i)$ as a function of θ is plotted. In the case shown, the copper complex exits at 34° , the nickel complex at 55° , the Co-1 complex at 73° , and the Co-2 complex at 177° . Note that the shape of the peaks is not Gaussian. This, of course, is because dispersion effects were neglected in the calculation. Results for each of the other cases tested are shown in Figs. B.1-B.18 found in Appendix B.

Referring to the figures in Appendix B, the model predicts complete separation of all four solutes for bandwidths up to 19 mm (7.7°). With a feed bandwidth of 19 mm, the model predicts that the nickel complex will emerge slightly contaminated with Co-1. At even larger feed bandwidths, the degree of overlap between the two bands increases. For the range of bandwidths used, the model predicts complete separation of the copper and nickel complexes.

Experimentally, the nickel and Co-1 bands overlapped for feed bandwidths greater than 22 mm, in agreement with the model. However, contrary

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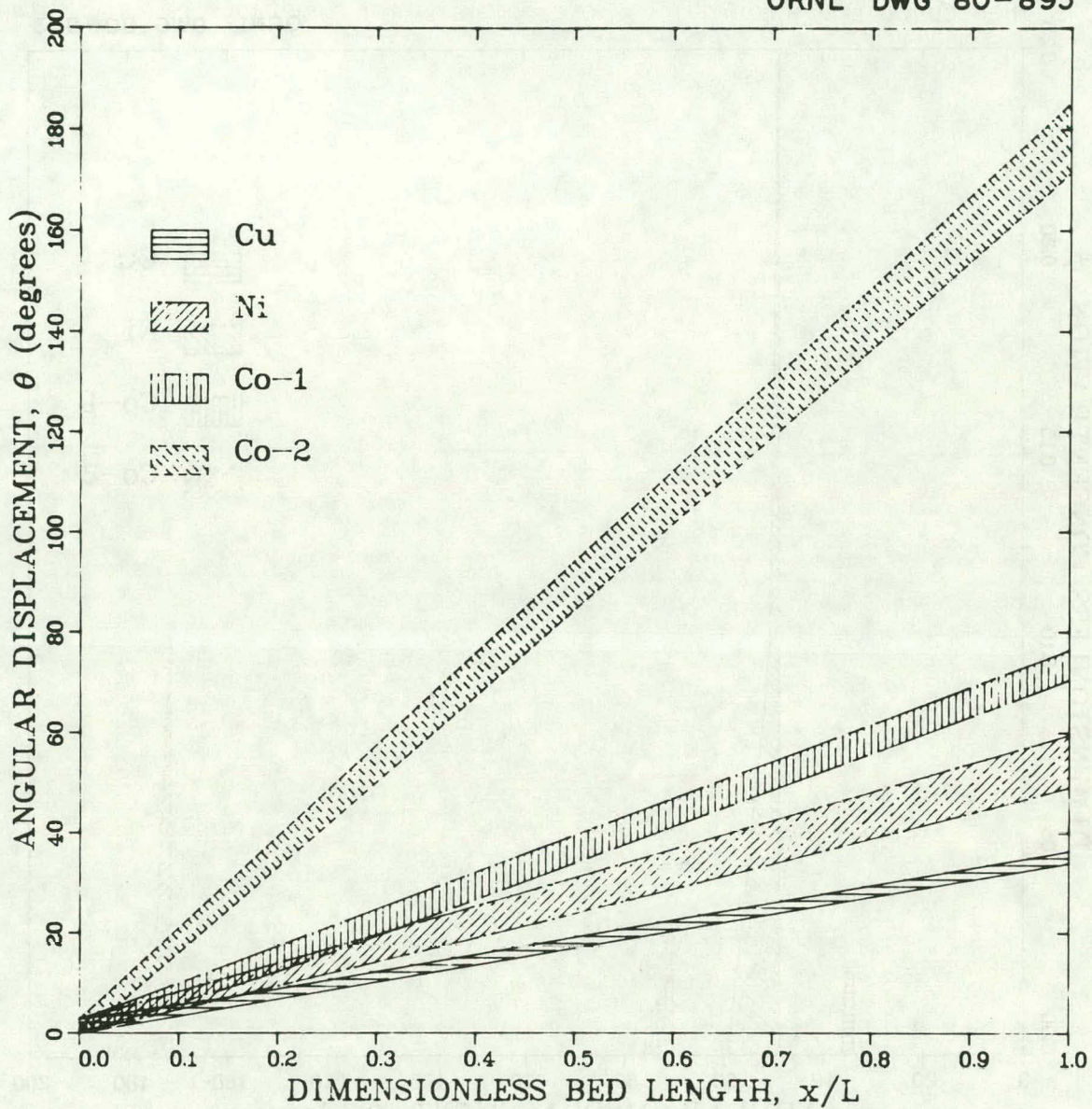


Fig. 10. Solute bands along the column length for a 7-mm feed bandwidth.

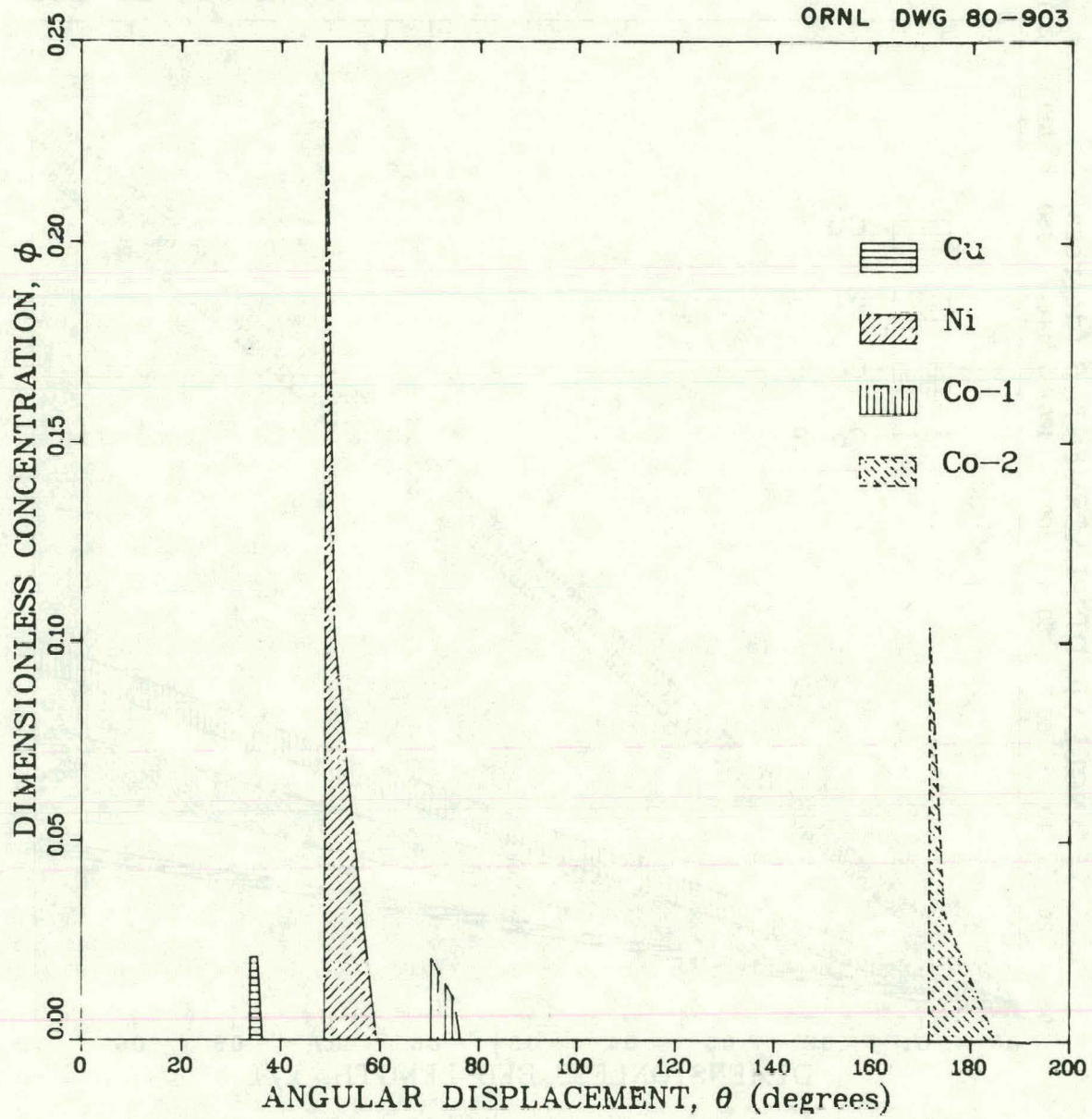


Fig. 11. Concentration profile at the column exit for a 7-mm feed bandwidth.

to predictions, the copper and nickel bands also overlapped for feed bandwidths of 19 mm and greater. This discrepancy between the model predictions and the experimental results may be explained in part by the omission of dispersion effects in the model. Also, the binding constants used in the calculation may not have been sufficiently accurate.

Although the model may not have predicted the band overlaps properly, it did a reasonably good job of estimating the position of the effluent concentration peaks. The comparison between experimentally measured and model-predicted peak positions is shown in Fig. 12. Most of the data points are scattered near the 45° line of identity. The major exception to this observation is the data for the position of the Co-2 peaks. The model generally overpredicted these peak locations, again probably due to the choice of the binding coefficient for Co-2.

The column effluent solute bandwidths predicted by the model were, in general, understandably smaller than those determined experimentally. The comparison of theory and experiment is shown in Figs. 13 and 14. In the case of copper, the model and experimental results show qualitative, but not quantitative, agreement (Fig. 13). The results obtained with nickel are somewhat surprising. Over the range of column feed bandwidths tested, the model agrees quite well with the experimental findings. Results obtained with Co-1 and Co-2 (Fig. 14) are similar to those of copper in that the model predicts the correct trend but not the correct quantitative values.

The comparison between predicted and measured peak resolution is shown in Fig. 15; peak resolution is defined in ref. 7. A value of unity means that the adjacent peaks are just touching. Values above unity infer

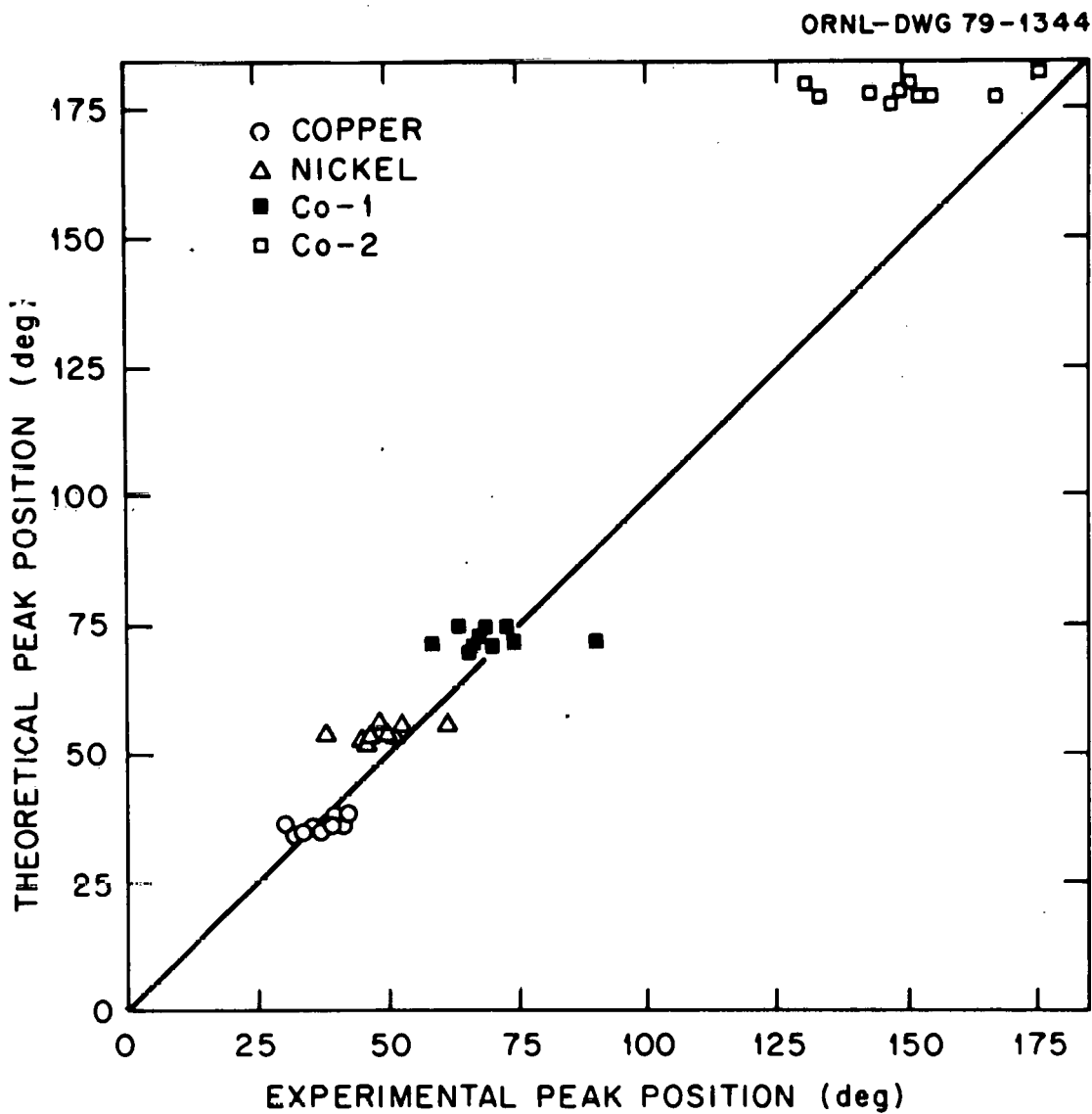


Fig. 12. Comparison of predicted and measured peak positions.

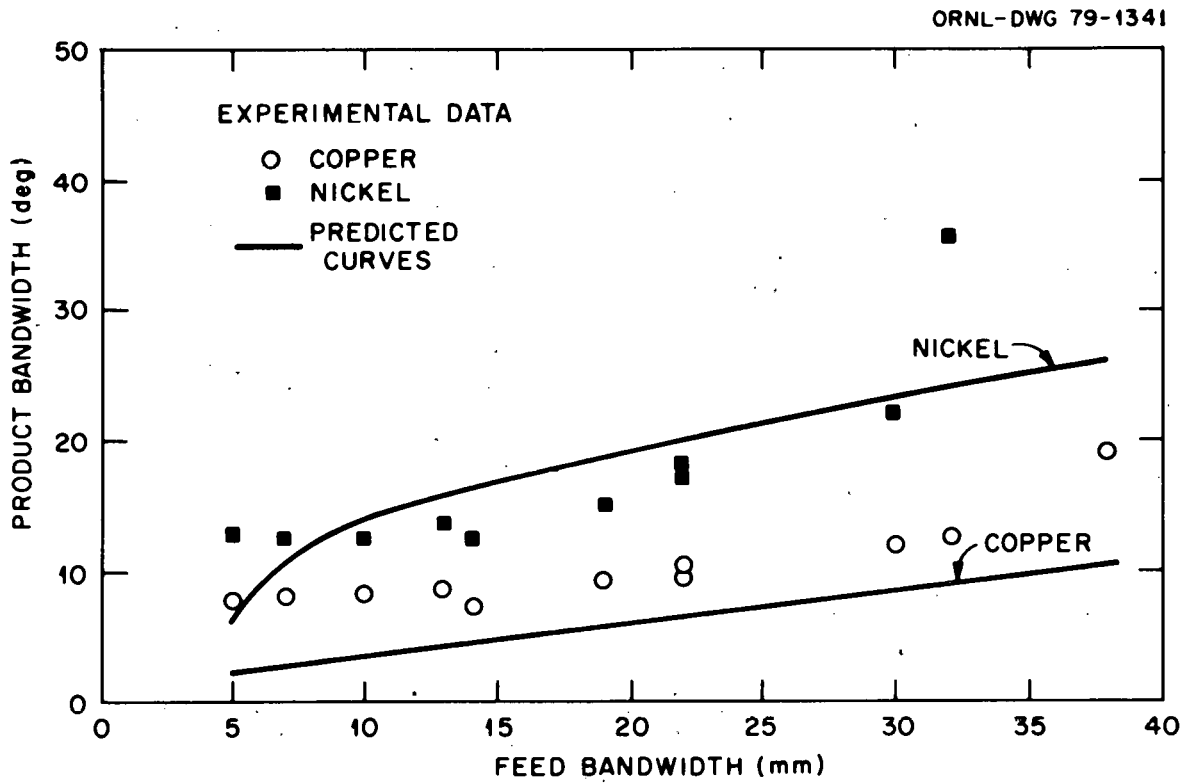


Fig. 13. Experimental and predicted product bandwidths for nickel and copper.

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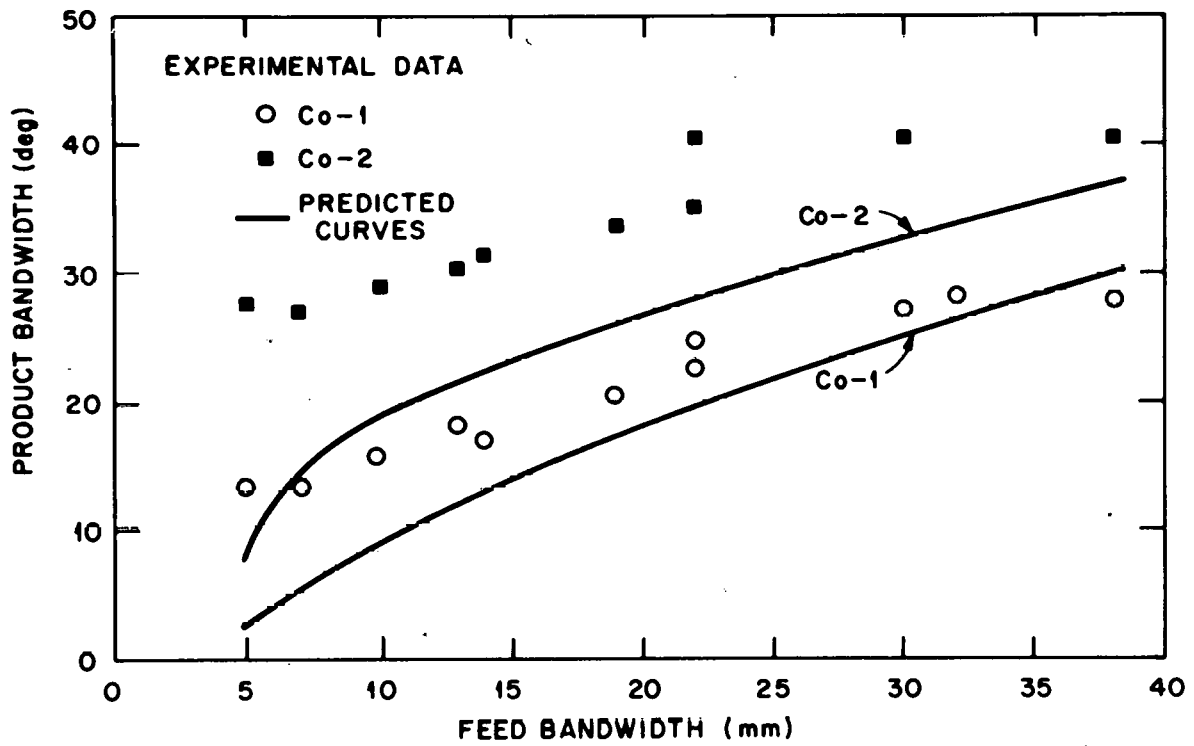


Fig. 14. Experimental and predicted product bandwidths for Co-1 and Co-2.

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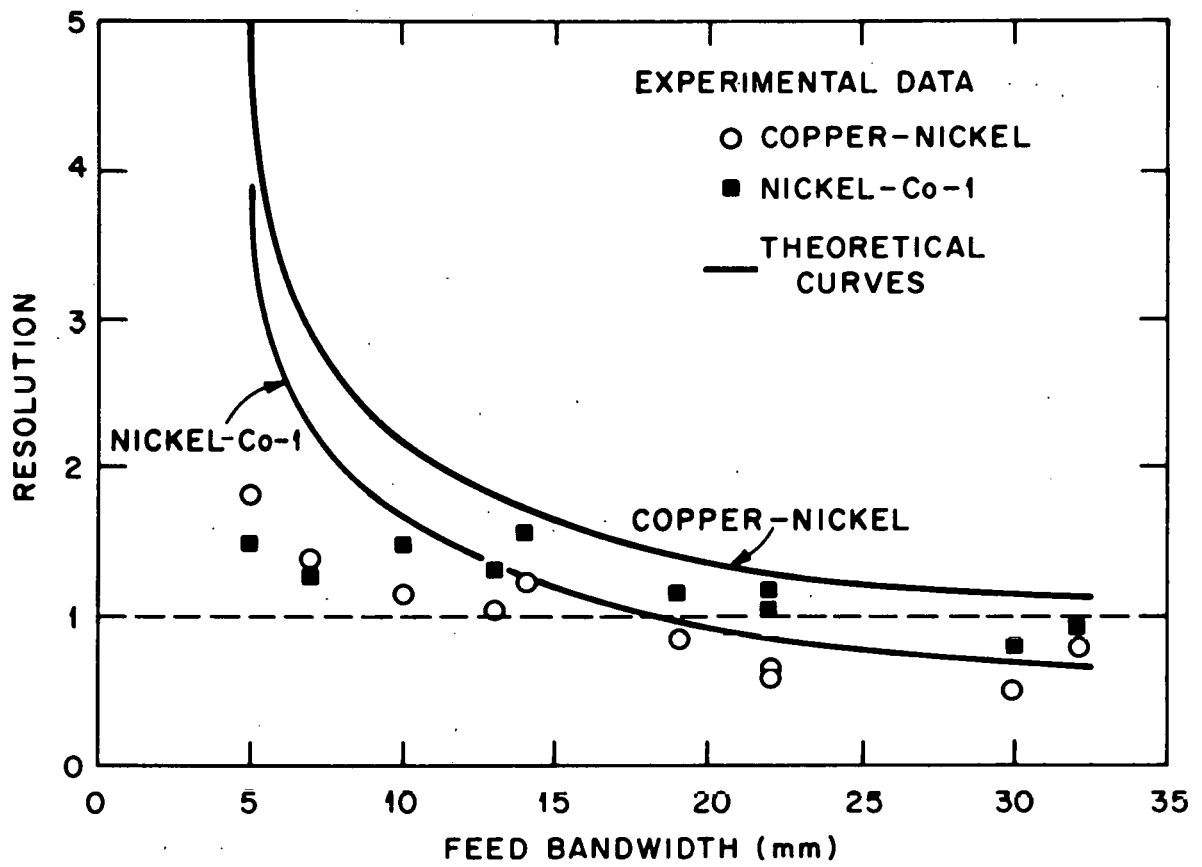


Fig. 15. Product resolution.

separation of the two solutes, and values below unity infer contamination. The model predicted separation of the copper and nickel complexes over the entire range of feed bandwidths. Experimental results showed an overlap of the two solute bands at feed bandwidths in excess of 19 mm. The agreement between the model and experimental results was better in the case of the nickel--Co-1 separation. The onset of contamination at about 22 mm was confirmed experimentally.

5. DISCUSSION

If a mathematical model is to predict the performance of the CAC accurately, it must incorporate all the effects of convection, dispersion, and nonlinear binding isotherms. The shortcoming of the model tested here is the omission of the dispersion effects. These dispersion effects are particularly important if one wishes to determine solute resolution in the column effluent. However, if only a prediction of gross behavior at different operating conditions is desired, this model may be sufficient. In models which include dispersion, it is necessary to estimate the dispersion coefficient for each solute. Since there is no reliable way to make this estimate better than within an order of magnitude, the success of these more complete models is also doubtful.

The virtues of the Rhee model are its simplicity in terms of the calculations involved and the fact that it provides a convenient framework for insight into the physical processes which occur in chromatographic separations. It also allows consideration of the interactions between solutes and the resin in multicomponent separations. For example, one could expect to predict the effect of a doubling of the inlet feed

composition of all the solutes being separated with reasonable accuracy. Another desirable feature of this approach is that the theory could be modified to account for gradient elution operations. This area might warrant future exploration in light of the successful gradient elution runs on the CAC.⁶

6. CONCLUSIONS

The multicomponent chromatographic model² has been compared to experimental results obtained on the CAC. The model predicts the position of the effluent solute peaks quite well under conditions of varying input feed bandwidths. It also predicts the correct trends in effluent solute resolution as the feed bandwidth is increased. Given a method for superimposing dispersion effects on the predicted output concentration profiles, one would be better able to make a quantitative prediction of band overlap and contamination. One possible use of this model would be to predict CAC performance as the solute concentrations in the feed stream were varied. More experimental data need to be generated with wide ranges in feed composition before comparisons with the model can be made. The model may be adapted for gradient elution operations, and this may prove useful in assessing the gradient elution mode in the industrial setting.

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APPENDIX A. COMPUTER PROGRAM

All calculations were performed on a PDP-11 computer. A basic description of how to use the program follows. No attempt will be made to describe the program algorithm in detail.

1. Running the program MCHAR

Input data. Input data are read from a data file, MCHAR.DAT. The data should be in the following format:

Line 1: K_i (6F8.4 format)
 Line 2: C_j^i (6F8.4 format)
 Lines 3 and 4: C_{i0} (4E10.4 format)
 Line 5: N, ϵ (2F8.4 format)
 Line 6: n, nic, pw, v, ω (2I2, 3F8.4 format)

where the symbols are defined in Table A.1. An example of input data is also shown in Table A.1.

In the example shown, $K_1 = 1.18$, $K_2 = 2.18$, $K_3 = 2.92$, and $K_4 = 7.73$. Note that the K_i 's must be entered in ascending numerical order. Since the bcd was assumed to be initially solute free, $C_j^i = 0$. The feed (entry) solute concentrations are: $C_{10} = 0.0354$, $C_{20} = 0.1533$, $C_{30} = 0.01909$, and $C_{40} = 0.03818$ meq/ml ($C_{50} = C_{60} = 0$). The resin binding capacity, N , = 1.78 meq/ml resin; $\epsilon = 0.38$; the total number of solutes, n , = 4 (the program may handle up to six component systems); the number of isoconcentration lines in each "fan wave," nic , = 8 (8 is the maximum number possible); the feed bandwidth, pw , = 12.0 mm; the superficial eluent velocity, v , = 2.53 cm/min; and, finally, the bed rotation rate, ω , = 62.2 °/hr.

Table A.1. Program input data

 Sample data file:

1.18, 2.18, 2.92, 7.73, 0., 0.
 0., 0., 0., 0., 0., 0
 3.54E-02, 1.533E-01, 1.909E-02, 3.818E-02
 0.0E-00, 0.0E-00, 0.0E-00, 0.0E-00
 1.78, 0.38
 4, 8, 12., 2.53, 62.2

Notations used:

K_i - solute binding constants, ml solution/meq (line 1)
 C_j^i - initial solute concentrations in ion exchange bed, meq/ml solution (line 2)
 C_{i0} - feed solute concentrations, meq/ml (lines 3 and 4)
 N - resin binding capacity, meq/ml resin (line 5 - #1)
 ϵ - column void fraction (line 5 - #2)
 n - number of solutes (line 6 - #1)
 nic - number of isoconcentration line in each "fan wave" (line 6 - #2)
 pw - feed bandwidth, mm (line 6 - #3)
 v - superficial eluent velocity, cm/min (line 6 - #4)
 ω - column rotation rate, "/hr (line 6 - #5)

Executing the program. If MCHAR.DAT has the proper input data, only "RUN MCHAR" needs to be entered. After results are printed, the line terminal will ask if a plot is wanted (in the X- θ plane as shown in Fig. B.1). If the response is yes, the maximum values for X and θ are requested. In this case, X is the dimensionless bed length, x/L. After entering the requested value, press "return" and the plot will be made on the cathode ray terminal (CRT). A hard copy of the plot may be obtained using the "copy" key on the CRT keyboard. The line terminal will ask if another plot with different scaling is wanted; if so, it will ask for maximum values for X and θ .

Once all the desired X- θ plots have been obtained, concentration profiles may be drawn (similar to those shown on Fig. B.2) at any desired dimensionless bed length. Enter only the appropriate responses to the questions posed by the line terminal. The coordinates of the curves to be plotted are shown ahead of time on the line printer. This should aid in determining the best scaling parameters to enter.

2. Estimation of binding constants, K_i

Suppose one wishes to model a particular experimental run in which the θ_{i0} (position of the peak maxima) are known. To a first approximation, the binding constants, K_i , may be estimated from:

$$\theta_v \left(1 + \left(\frac{1 - \epsilon}{\epsilon} \right) NK_i \right) = \theta_{i0} , \quad (\text{A.1})$$

where θ_v is the angular position (in degrees) at which the pure eluent would exit (analogous to the void volume time in gel permeation chromatography).

$$\theta_v = \frac{\epsilon \omega X}{V} . \quad (\text{A.2})$$

These relationships are exact if the competitive binding effects among the solutes are minimal. The program-predicted θ_{i0} will differ from the experimental values to the extent that competitive binding is important.

A second guess of the K_i may be made based on the following:

$$\frac{K_{i\text{-program}}}{\theta_{i\text{-program}} - \theta_v} = \frac{K_{i\text{-new}}}{\theta_{i\text{-expt}} - \theta_v} . \quad (\text{A.3})$$

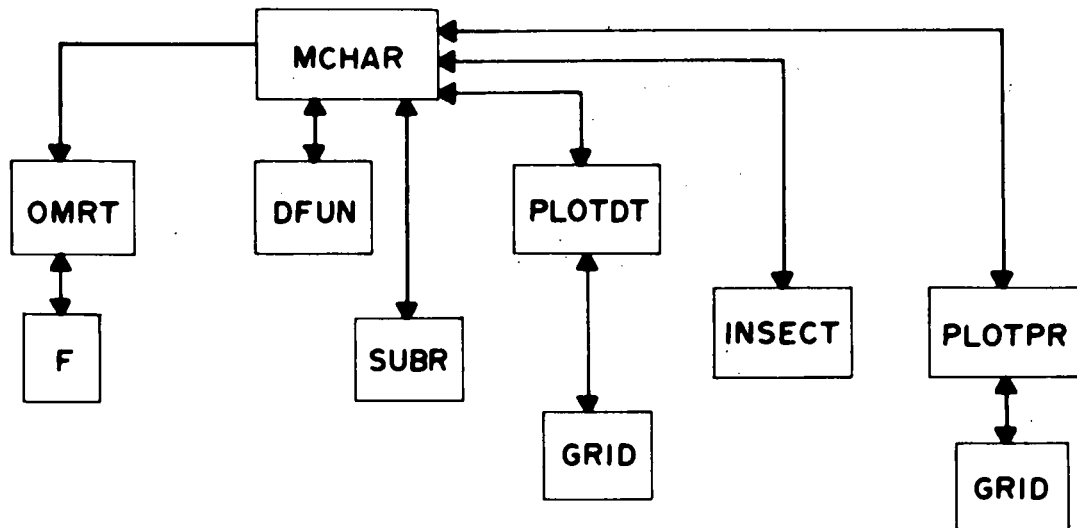
The $K_{i\text{-new}}$ which is calculated will usually result in $\theta_{i\text{-program}}$, which matches the experimental values quite closely.

3. Program components

The program FORTRAN code consists of a MAIN section (called MCHAR for method of characteristics) and subroutines. MCHAR calls all the necessary subroutines and function subprograms. A listing of the various subroutines and program organization is shown in Fig. A.1.

4. Program listing

The detailed FORTRAN code is listed on the following pages.



Functions of subprograms

1. MCHAR - main driving program; calculates initial slopes of shock waves and fan waves; calculates initial isoconcentrations associated with all waves.
2. OMRT,F - finds ω_i 's as defined in ref. 2.
3. DFUN - calculates $1 + \sum_{i=1}^n K_i C_i$.
4. SUBR - determines intersection coordinates and new isoconcentrations associated with shock and fan waves.
5. PLOTDT - plots X- θ figures on CRT.
6. GRID - supplies grid for plots.
7. INSECT - determines the coordinates of all shock and fan waves at a particular bed length.
8. PLOTPR - plots concentration profiles.

Fig. A.1. Organization of program.

```

REAL*4 K(6),N,NM(6,6),LA(6),NE(6),NI(6)
LOGICAL*1 XDATE(9),XTIME(8)
DIMENSION CI(6),CE(6),PHII(6),PHIE(6),D(6),PHIC(6,6),SSH(6),
1DD(6,8),XJ(6,6),PHI(6,6,6,8),XS(6,6,8),TAU(6,6,8),WI(6),
1 WE(6), SCW(6,8),DS(6),PHICS(6,6),NMS(6,6),ETA(8),TAUA(6),
2 DEG(6,6,8), DEGA(6),TI(72),PHICON(72,6),DEE(72)
COMMON TY,SSH,SCW,TAU,DD,D,WE,WI,EPS,PHI,N,K,DI,DE,TAUA
1 ,V,ROT
COMMON /BLK2/ XDATE,XTIME
COMMON /BLK3/ XISO,PHICS,DEE,PHICON,ICNT
COMMON /BLK4/ DEG,XS,M,NIC,TYD,DEGA
CALL ASSIGN(2,'MCHAR.DAT',9)
CALL ASSIGN(1,'TT7:')
READ(2,1000) K
READ(2,1000) CI
READ(2,1050) CE
READ(2,1000) N,EPS
READ(2,1100) M,NIC,PW,V,ROT
CALL DATE(XDATE)
CALL TIME(XTIME)
WRITE(6,1110) XDATE,XTIME,K,CI,CE
WRITE(6,1120) N,PW,V,ROT
TY=.470763*PW*V/ROT
DO 100 J=1,M
PHII(J)=K(J)*CI(J)
PHIE(J)=K(J)*CE(J)
100 CONTINUE
DE=DFUN(PHIE)
DI=DFUN(PHII)
DO 200 J=1,M
NE(J)=N*PHIE(J)/DE
NI(J)=N*PHII(J)/DI
200 CONTINUE
CALL OMRT(N,K,NI,M,WI,IER)
GO TO (210,2000), IER
210 WRITE(6,1130) WI
CALL OMRT(N,K,NE,M,WE,IER)
GO TO (220,2100), IER
220 WRITE(6,1140) WE
DO 500 IM=1,M
DO 400 KK=1,M
PROD=PHIC(IM)
DK=DE
DO 300 J=1,KK
PROD=PROD*(1.-N*K(IM)/WI(J))/(1.-N*K(IM)/WE(J))
IF (IM .GT. 1) GO TO 300
DK=DK*WE(J)/WI(J)
300 CONTINUE
IF (IM .GT. 1) GO TO 320
D(KK)=DK
320 PHIC(IM,KK)=PROD
NM(IM,KK)=N*PHIC(IM,KK)/D(KK)
400 CONTINUE
500 CONTINUE

```

```

WRITE(6,1150)
DO 600 KK=1,M
WRITE(6,1200) KK,(PHIC(IM,KK),IM=1,M)
600 CONTINUE
DO 645 IM=1,M
DO 640 KK=1,M
PROD=PHIE(IM)
DK=DE
DO 630 J=KK+1,M
IF (J .GT. M) GO TO 638
PROD=PROD*(1.-N*K(IM)/WI(J))/(1.-N*K(IM)/WE(J))
DK=DK*WE(J)/WI(J)
630 CONTINUE
IF (IM .GT. 1) GO TO 635
DS(KK+1)=DK
635 PHICS(IM,KK)=PROD
NMS(IM,KK)=N*PHICS(IM,KK)/DS(KK+1)
GO TO 640
638 DS(1)=DI
PHICS(IM,KK)=PHIE(IM)
NMS(IM,KK)=NE(IM)
640 CONTINUE
645 CONTINUE
DO 650 IM=1,M
SSH(IM)=EPS+(1.-EPS)*WE(IM)/DS(IM)
650 CONTINUE
DELD=D(1)-DE
DO 700 KK=1,M
LA(KK)=WI(KK)*D(KK)
DO 670 L=1,NIC
XL=NIC-L
DD(KK,L)=D(KK)-XL*DELD/(NIC-1)
SCW(KK,L)=EPS+(1.-EPS)*LA(KK)/DD(KK,L)**2
DO 660 IM=1,M
IF (IM .NE. KK) GO TO 655
IF ( KK .EQ. 1) GO TO 652
DOLD=D(KK-1)
PHIOLD=PHIC(IM,KK-1)
651 XJ(IM,KK)=PHIOLD/(DOLD-D(KK))
GO TO 657
652 DOLD=DE
PHIOLD=PHIE(IM)
GO TO 651
655 XJ(IM,KK)=K(IM)*NM(IM,KK)/(N*K(IM)-WI(KK))
657 PHI(M,IM,KK,L)=PHIC(IM,KK)+XJ(IM,KK)*(DD(KK,L)-D(KK))
660 CONTINUE
670 CONTINUE
DELD=D(KK+1)-D(KK)
700 CONTINUE
WRITE(6,1240)
WRITE(6,1250) SSH
WRITE(6,1290)
DO 710 KK=1,M
WRITE(6,1300) KK,(SCW(KK,LL),LL=1,NIC)

```



```

710  CONTINUE
      WRITE(6,1340)
      DO 720 KK=1,M
      WRITE(6,1300) KK,(DD(KK,L),L=1,NIC)
720  CONTINUE
      WRITE(6,1390)
      DO 750 KK=1,M
      DO 740 L=1,NIC
      WRITE(6,1400) KK,L,(PHI(M,IM,KK,L),IM=1,M)
740  CONTINUE
750  CONTINUE
      WRITE(6,1430)
      DO 770 KK=1,M
      WRITE(6,1440) KK,(PHICS(IM,KK),IM=1,M)
770  CONTINUE
      CALL SUBR
      TYD=ROT*TY*50./(V*60.)
      CALL PLOTDT
780  TYPE 2400
      ACCEPT 2500,IA
      IF (IA .EQ. 0) CALL FINITT(0,700)

C
C      FIND INTERSECTIONS WITH SHOCK WAVES
C

      CALL INSECT
      CALL PLOTPR(M)
      TYPE 2600
      ACCEPT 2700,IA
      IF (IA .NE. 0) GO TO 780
      CALL FINITT(0,700)
1000  FORMAT(6F8.4)
1050  FORMAT(4E16.4)
1100  FORMAT(2I2,3F8.4)
1110  FORMAT('1',//T10,9A1,3X,8A1,//T10,'INPUT DATA',/T10,10('*'),///
1T20,'K'S:',/T20,3('*'),//T20,6F10.4,//T20,'CI'S:',/T20,
24('*'),//T20,6F10.4,///T20,'CE'S:',/T20,4('*'),//T20,6F10.4)
1120  FORMAT(///T20,'N=',F5.2,' MEQUIV/ML',/T20,'PULSE WIDTH=',
1 F4.1,' MM',/T20,'SUPERFICIAL VELOCITY=',F5.2,' CM/MIN',/T20,
2 'ROTATION RATE=',F4.0,' DEGREES/HR')
1130  FORMAT(///T20,'OMEGA-ENTRY'S:',/T20,13('*'),//T20,6F10.4)
1140  FORMAT(///T20,'OMEGA-INITIAL'S:',/T20,16('*'),//T20,6F10.4)
1150  FORMAT('1',//T10,'CONSTANT CONCENTRATIONS',/T10,23('*'),//T20,
1 'WAVE NO.',T40,'PHI',/T30,80('*'),/)
1200  FORMAT(T23,I2,T30,6E13.3)
1240  FORMAT(////,T10,'SLOPES OF SHOCK WAVES=',/)
1250  FORMAT(T20,6E13.3)
1290  FORMAT(////,T10,'INITIAL SLOPES OF C-WAVES=',//T20,'WAVE NO.',
1 T30,'L=1',T43,'L=2',T56,'L=3',T69,'L=4',T82,'L=5',/)
1300  FORMAT(T23,I2,T28,8E11.3)
1340  FORMAT(////,T10,'VALUES FOR D=',/)
1390  FORMAT(////T10,'ISO-CONCENTRATIONS (INITIAL)=',//T20,'WAVE NO.',
1 T30,'L',T40,'M',/)
1400  FORMAT(T23,I2,T30,I1,T38,6E13.3)
1430  FORMAT(////T10,'ISOCONCENTRATIONS ASSOCIATED WITH SHOCK WAVES',

```

```
1 /T20,'WAVE NO.',T33,'M',/)  
1440 FORMAT(T23,I2,T29,6E13.3)  
2000 WRITE(6,2200)  
      STOP  
2100 WRITE(6,2300)  
      STOP  
2200 FORMAT('1',//T10,'***ERROR***-NO ROOT COULD BE FOUND FOR ''WI'',  
1 CHECK INPUT DATA.')2300 FORMAT('1',/T10,'***ERROR***NO ROOT COULD BE FOUND FOR ''WE'',  
2 CHECK INPUT DATA.')2400 FORMAT(' DO YOU WANT ANY PLOTS OF CONCENTRATION PROFILES?  
1(0=NO; 1=YES):',$(  
2500 FORMAT(I2)  
2600 FORMAT(' DO YOU WANT ANOTHER PLOT OF CONCENTRATION PROFILES  
1 AT A LARGER BED LENGTH? (0=NO; 1=YES):',$(  
2700 FORMAT(I2)  
      END
```

```

SUBROUTINE OMRT(N,K,NI,M,WI,IER)
REAL*4 N,K(6),NI(6)
DIMENSION WI(6)
DO 700 J=1,M
IF (NI(J) .NE. 0) GO TO 50
WI(J)=N*K(J)
GO TO 700
50  IF (J .EQ. 1) XL=.0001
    IF (J .NE. 1) XL=N*K(J-1)+.0001
    XR=N*K(J)-.0001
    FL=F(XL,K,NI,N,M)
    FR=F(XR,K,NI,N,M)
    IF (FL/FR .GT. 0.) GO TO 800
    IF (FL .LT. 0.) GO TO 100
    XPOS=XL
    XNEG=XR
    GO TO 300
100  XNEG=XL
    XPOS=XR
300  X=.5*(XPOS+XNEG)
    DO 400 I=1,50
    FX=F(X,K,NI,N,M)
    IF (ABS(FX) .LT. .001) GO TO 600
    IF (FX .LT. 0.) GO TO 320
    XPOS=X
    GO TO 340
320  XNEG=X
340  X=.5*(XPOS+XNEG)
400  CONTINUE
    WRITE(6,1000) J
    IER=2
    RETURN
600  WI(J)=X
700  CONTINUE
    IER=1
    RETURN
800  WRITE(6,1100)
    IER=2
    RETURN
1000 FORMAT('1',/T10,'***ERROR***-NO ROOT COULD BE FOUND FOR J=',I2,
1' IN OMRT.')
1100 FORMAT('1',/T10,'***ERROR***-INITIAL ESTIMATES DID NOT BRACKET
1 THE SOLUTION IN OMRT.')
END

```

```
FUNCTION F(X,K,NI,N,M)
REAL*4 K(6),NI(6),N
SUM=0.
DO 100 I=1,M
SUM=SUM+K(I)*NI(I)/(N*K(I)-X)
100 CONTINUE
F=1.-SUM
RETURN
END
```

```
FUNCTION DFUN(PHI)
DIMENSION PHI(6)
DFUN=0.
DO 100 I=1,6
DFUN=DFUN+PHI(I)
100 CONTINUE
DFUN=DFUN+1.0
RETURN
END
```

```

SUBROUTINE SUBR
REAL*4 K(6),N,NM(6,6),LA(6),NE(6),NI(6)
LOGICAL*1 XDATE(9),XTIME(8)
DIMENSION CI(6),CE(6),PHII(6),PHIE(6),D(6),PHIC(6,6),SSH(6),
1DD(6,8),XJ(6,6),PHI(6,6,6,8),XS(6,6,8),TAU(6,6,8),WI(6),
1 WE(6), SCW(6,8),DS(6),PHICS(6,6),NMX(6,6),ETA(8),TAUA(6),
2 DEG(6,6,8), DEGA(6)
COMMON TY,SSH,SCW,TAU,DD,D,WE,WI,EPS,PHI,N,K,DI,DE,TAUA
1 ,V,ROT
COMMON /BLK2/ XDATE,XTIME
COMMON /BLK4/ DEG,XS,M,NIC,TYD,DEGA
TI=0.
XI=0.
DO 950 KK=1,M
IF (KK .NE. 1) TI=TAU(M,KK-1,NIC)
IF (KK .NE. 1) XI=XS(M,KK-1,NIC)
XS(M,KK,1)=(TY-TI+SSH(M)*XI)/(SSH(M)-SCW(KK,1))
TAU(M,KK,1)=SCW(KK,1)*XS(M,KK,1)+TY
DIN=DD(KK,1)
DO 900 IM=M,1,-1
OMNEW=WI(IM)
OMOLD=WE(IM)
IF(KK .LT. IM) WRITE(6,1450) IM
DO 850 L=1,NIC
IF (IM .NE. M) GO TO 833
IF (KK .EQ. M) GO TO 833
IF (L .EQ. NIC) GO TO 820
XS(IM,KK,L+1)=XS(IM,KK,L)*(DD(KK,L+1)*(DIN-DD(KK,NIC)*WI(KK)/
1OMNEW)/(DIN*(DD(KK,L+1)-DD(KK,NIC)*WI(KK)/OMNEW)))*2
TAU(IM,KK,L+1)=SCW(KK,L+1)*XS(IM,KK,L+1)+TY
820 IF (IM .GT. KK) SCW(KK,L)=EPS+(1.-EPS)*(OMNEW/OMOLD)*WI(KK)*
1 DD(KK,NIC)/(DD(KK,L))*2
SSH(IM)=EPS+(1.-EPS)*OMNEW/DD(KK,NIC)
SUM=0.
DO 830 I=1,M
IF (IM .EQ. 1) GO TO 850
IF (I .EQ. IM) GO TO 822
PHI(IM-1,I,KK,L)=PHI(IM,I,KK,L)*(1.-N*K(I)/OMNEW)/
1 (1.-N*K(I)/OMOLD)
GO TO 824
822 PHI(IM-1,I,KK,L)=0.
824 SUM=SUM+PHI(IM-1,I,KK,L)
830 CONTINUE
IF (KK .LT. IM) WRITE(6,1500) KK,L,(PHI(IM-1,I,KK,L),I=1,M)
DD(KK,L)=SUM+1.
GO TO 850
833 IF (IM .NE. KK) GO TO 846
IF (IM .NE. M) GO TO 834
IF (L .GT. 1) GO TO 8331
XS(IM,KK,1)=(TY-TAU(IM,KK-1,NIC)+SSH(IM)*XS(IM,KK-1,NIC))/
1 (SSH(IM)-SCW(KK,1))
TAU(IM,KK,1)=SCW(KK,1)*XS(IM,KK,1)+TY
GO TO 820
8331 DOLD=DD(KK-1,NIC)

```

```

DNEW=D(KK-1)
XS(IM, KK, L)=XS(IM, KK, 1)*((1.-DOLD/DNEW)/(1.-DOLD/DD(KK, L)))**2
TAU(IM, KK, L)=SCW(KK, L)*XS(IM, KK, L)+TY
GO TO 820
834 IF ((KK .EQ. 1) .AND. (L .EQ. 1)) GO TO 836
    IF ( L .NE. 1) GO TO 840
    XS(IM, KK, 1)=(TAU(IM, KK-1, NIC)-TAU(IM+1, KK, 1)-XS(IM, KK-1, NIC)*
1 SSH(IM)+ XS(IM+1, KK, 1)*SCW(KK, 1))/(SCW(KK, 1)-SSH(IM))
    TAU(IM, KK, 1)=SSH(IM)*(XS(IM, KK, 1)-XS(IM, KK-1, NIC))+
1 TAU(IM, KK-1, NIC)
    GO TO 820
836 XS(IM, KK, 1)=(-TAU(IM+1, KK, 1)+XS(IM+1, KK, 1)*SCW(KK, 1))/(SCW(KK, 1)
1 -SSH(IM))
    TAU(IM, KK, 1)=SSH(IM)*XS(IM, KK, 1)
    GO TO 820
840 IF (L .GT. 2) GO TO 843
    DELD=DD(KK, 3)-DD(KK, 2)
    IF(KK .EQ. 1) GO TO 841
    DOLD=DD(KK-1, NIC)
    GO TO 8411
841 DOLD=DI
8411 DSTAR=DD(KK, 1)
    DO 842 LL=1, NIC
    ETA(LL)=FETA(TAU(IM+1, KK, LL), TAU(IM, KK, 1), SCW(KK, LL),
1 XS(IM+1, KK, LL), XS(IM, KK, 1))
842 CONTINUE
843 SUM=0.
    DO 845 LL=1, L
    SUM=SUM+ETA(LL)
845 CONTINUE
    XIETA=(SUM-.5*(ETA(L)+ETA(1)))*DELD
    IF (L .EQ. NIC) GO TO 820
    XS(IM, KK, L)=((DOLD*DD(KK, L)**2)/((1.-EPS)*(DD(KK, L)-DOLD)
1 *WI(KK)*DD(KK, 1)))*(ETA(L)-(1./(DD(KK, L)-DOLD))*XIETA)
2 +XS(IM, KK, 1)
    TAU(IM, KK, L)=SCW(KK, L)*(XS(IM, KK, L)-XS(IM+1, KK, L))+
1 TAU(IM+1, KK, L)
    GO TO 820
846 IF (IM .LT. KK) GO TO 900
    IF (L .NE. 1) GO TO 847
    IF (KK .EQ. 1) GO TO 8461
    XS(IM, KK, 1)=(TAU(IM, KK-1, NIC)-TAU(IM+1, KK, 1)-XS(IM, KK-1, NIC)*
1 SSH(IM) +XS(IM+1, KK, 1)*SCW(KK, 1))/(SCW(KK, 1)-SSH(IM))
    TAU(IM, KK, 1)=SSH(IM)*(XS(IM, KK, 1)-XS(IM, KK-1, NIC))+
1 TAU(IM, KK-1, NIC)
    GO TO 820
8461 XS(IM, KK, 1)=(-TAU(IM+1, KK, 1)+XS(IM+1, KK, 1)*SCW(KK, 1))/
1 (SCW(KK, 1)-SSH(IM))
    TAU(IM, KK, 1)=SSH(IM)*XS(IM, KK, 1)
    GO TO 820
847 IF (L .GT. 2) GO TO 8481
    DELD=DD(KK, 3)-DD(KK, 2)
    DO 848 LL=1, NIC
    ETA(LL)=FETA(TAU(IM+1, KK, LL), TAU(IM, KK, 1), SCW(KK, LL),

```

```

1 XS(IM+1, KK, LL), XS(IM, KK, 1))
848 CONTINUE
8481 SUM=0.
DO 849 LL=1, L
SUM=SUM+ETA(LL)
849 CONTINUE
XIETA=(SUM-.5*(ETA(L)+ETA(1)))*DELD
DNEW=DD(KK, L)-DD(KK, NIC)*WI(KK)/OMNEW
XS(IM, KK, L)=XS(IM, KK, 1)+(DD(KK, L)**2/((1.-EPS)*OMNEW*DNEW))*
1 (ETA(L)-(1./DNEW)*XIETA)
TAU(IM, KK, L)=SCW(KK, L)*(XS(IM, KK, L)-XS(IM+1, KK, L))+
1 TAU(IM+1, KK, L)
GO TO 820
850 CONTINUE
XMAX=XS(1, 1, NIC-1)
DO 860 L=1, M-1
IF (XS(L+1, L+1, NIC-1) .GT. XMAX) XMAX=XS(L+1, L+1, NIC-1)
860 CONTINUE
XMAX=XMAX+1.
DO 890 L=1, M
TAU(L, L, NIC)=TAU(L, L, NIC-1)+SSH(L)*(XMAX-XS(L, L, NIC-1))
TAUA(L)=TAU(L-1, L, NIC)+SCW(L, NIC)*(XMAX-XS(L-1, L, NIC))
DEGA(L)=ROT*TAUA(L)*50./(V*60.)
XS(L, L, NIC)=XMAX
890 CONTINUE
IF (IM .EQ. KK) GO TO 895
WRITE(6, 1550) IM, KK
WRITE(6, 1600) (DD(KK, L), L=1, NIC)
WRITE(6, 1620) KK, IM
WRITE(6, 1625) (SCW(KK, L), L=1, NIC)
895 WRITE(6, 1630) SSH(IM)
900 CONTINUE
950 CONTINUE
D WRITE(6, 1650) XDATE, XTIME
WRITE(6, 1660)
DO 980 I=M, 1, -1
DO 970 KK=1, M
IF (KK .GT. I) GO TO 980
DO 960 L=1, NIC
DEG(I, KK, L)=ROT*TAU(I, KK, L)*50./(V*60.)
WRITE(6, 1700) I, KK, L, XS(I, KK, L), TAU(I, KK, L), DEG(I, KK, L)
960 CONTINUE
970 CONTINUE
980 CONTINUE
1450 FORMAT(///T10, 'ISOCONCENTRATIONS AFTER INTERACTION WITH SHOCK
1 WAVE NO.', I2, //T20, 'WAVE NO.', T30, 'L', T35, 'M:', //)
1500 FORMAT(T23, I2, T30, I1, T33, 6E13.3)
1550 FORMAT(///T10, 'NEW D''S AFTER SHOCK NO.', I2, ' WAVE NO.',
1 I2, //)
1600 FORMAT(T30, 5E13.3, /T30, 5E13.3)
1620 FORMAT(///T10, 'NEW C-WAVE SLOPES:', I2, ' WAVE AFTER', I2,
1 ' SHOCK', //)
1625 FORMAT(T30, 5E13.3, /T30, 5E13.3)
1630 FORMAT(///T10, 'NEW SHOCK WAVE SLOPE', /T30, E13.3)

```

```
1650 FORMAT('1',/T10,9A1,3X,8A1)
1660 FORMAT(///T10,'COORDINATES OF INTERSECTIONS
1 OF C-WAVES AND SHOCK WAVES',/T10,57('*'),/T20,'SHOCK NO.',T30,
2 'WAVE NO.',T40,'L',T55,'X',T65,'TAU',T75,'DEGREES ROT.',/)
1700 FORMAT(T23,I1,T33,I1,T40,I1,T50,F10.4,T60,F10.4,T70,F10.2)
RETURN
END
```

```
FUNCTION FETA(U,V,W,X,Y)
FETA=U-V-W*(X-Y)
RETURN
END
```



```

SUBROUTINE PLOTDT
LOGICAL*1 XTIME(8),XDATE(9)
DIMENSION Y(6,6,8),X(6,6,8),DEGA(6)
COMMON /BLK1/MINX,MAXX,MINY,MAXY
COMMON /BLK2/XDATE,XTIME
COMMON /BLK4/ Y,X,NSOL,NIC,YI,DEGA
ICNT=0
TYPE 100
100 FORMAT(' DO YOU WANT A PLOT? (0=NO; 1=YES)',*)
MINX=150
MAXX=900
MINY=150
MAXY=700
ACCEPT 150,IA
150 FORMAT(I2)
IF (IA .EQ. 0) RETURN
50 CALL INITT(480)
ICNT=ICNT+1
TYPE 60
60 FORMAT(' ENTER XMAX AND YMAX IN F8.4 FORMAT. ')
ACCEPT 70, XMAX,YMAX
70 FORMAT(2F8.4)
CALL TWINDO(MINX,MAXX,MINY,MAXY)
CALL DWINDO(0.,XMAX,0.,YMAX)

C
C     DRAW SHOCK WAVES
C
CALL MOVEA(0.,0.)
DO 300 I=NSOL,1,-1
DO 200 J=1,NSOL
IF (J .GT. 1) GO TO 260
DO 180 K=1,NIC
CALL DASHA(X(I,J,K),Y(I,J,K),I)
180 CONTINUE
200 CONTINUE
260 CALL MOVEA(0.,0.)
300 CONTINUE

C
C     DRAW C-WAVES
C
CALL MOVEA(0.,YI)
DO 800 J=1,NSOL
K=NIC
DO 600 I=NSOL,1,-1
IF (J .GT. 1) GO TO 600
IF ((I .EQ. J) .AND. (K .EQ. NIC)) GO TO 550
CALL DASHA(X(I,J,K),Y(I,J,K),J)
GO TO 600
550 CALL DASHA(X(NSOL,NSOL,NIC),DEGA(J),J)
600 CONTINUE
CALL MOVEA(0.,YI)
800 CONTINUE
CALL GRID(0.,XMAX,0.,YMAX)
IXO=MINX

```

```
IY0=730
CALL MOVABS(IX0,IY0)
CALL ANMODE
WRITE(1,900) XDATE,XTIME,ICNT
900  FORMAT(1H+,$,9A1,3X,8A1,5X,'GRAPH NO.',I2)
     TYPE 1000
1000 FORMAT(' DO YOU WANT ANOTHER PLOT WITH DIFFERENT SCALING? (0=NO;
1 1=YES);',%)
     ACCEPT 1100, IA
1100 FORMAT(I2)
     IF (IA .NE. 0) GO TO 50
     RETURN
     END
```

```

SUBROUTINE GRID(XMIN,XMAX,YMIN,YMAX)
COMMON /BLK1/MINX,MAXX,MINY,MAXY
IDELY=(MAXY-MINY)/5
IDELX=(MAXX-MINX)/5
CALL MOVABS(MINX,MINY)
IX=MINX
DO 100 I=1,6
CALL DRWABS(IX,MAXY)
IX=IX+IDELX
100 CALL MOVABS(IX,MINY)
CONTINUE
IY=MINY
CALL MOVABS(MINX,MINY)
DO 200 I=1,6
CALL DRWABS(MAXX,IY)
IY=IY+IDELY
200 CALL MOVABS(MINX,IY)
CONTINUE
CALL CSIZE(IHORZ,IVERT)
IXEXP=ALOG10(XMAX)
IYEXP=ALOG10(YMAX)
XMAXN=XMAX/(10**IXEXP)
XMINN=XMIN/(10**IXEXP)
YMINN=YMIN/(10**IYEXP)
YMAXN=YMAX/(10**IYEXP)
DELX=(XMAXN-XMINN)/5.
DELY=(YMAXN-YMINN)/5.
IXO=MINX-2*IHORZ
IYO=MINY-1.5*IVERT
CALL MOVABS(IXO,IYO)
X=XMINN
DO 300 I=1,6
CALL ANMODE
WRITE(1,1000) X
X=X+DELX
IXO=IXO+IDELX
CALL MOVABS(IXO,IYO)
1000 FORMAT(1H+,$,F4.2)
300 CONTINUE
IXO=MAXX-7*IHORZ
IYO=IYO-2*IVERT
IXEXP=-IXEXP
IYEXP=-IYEXP
CALL MOVABS(IXO,IYO)
CALL ANMODE
WRITE(1,1100) IXEXP
1100 FORMAT(1H+,$, 'X10**', I2)
IXO=MINX-4*IHORZ
IYO=MINY-.5*IVERT
CALL MOVABS(IXO,IYO)
Y=YMINN
DO 400 I=1,6
CALL ANMODE
WRITE(1,1200) Y

```

```
1200  FORMAT(1H+,$,F4.2)
      Y=Y+DELY
      IYO=IYO+IDELY
      CALL MOVABS(IXO,IYO)
400   CONTINUE
      IYO=MAXY-2*IVERT
      IXO=MINX-7*IHORZ
      CALL MOVABS(IXO,IYO)
      CALL ANMODE
      WRITE(1,1300) IYEXP
1300  FORMAT(1H+,$, 'X10**',I2)
      RETURN
      END
```

```

SUBROUTINE INSECT
REAL*4 K(6),N
LOGICAL*1 XDATE(9),XTIME(8)
DIMENSION D(6),SSH(6),DD(6,8),PHI(6,6,6,8),XS(6,6,8),
1 TAU(6,6,8),WI(6),WE(6), SCW(6,8),PHICS(6,6),TAUA(6),
1 DEG(6,6,8), DEGA(6),TI(72),PHICON(72,6),DEE(72),PHICSN(6,6)
COMMON TY,SSH,SCW,TAU,DD,D,WE,WI,EPS,PHI,N,K,DI,DE,TAUA
1 ,V,ROT
COMMON /BLK2/ XDATE,XTIME
COMMON /BLK3/ XISO,PHICS,DEE,PHICON,ICNT
COMMON /BLK4/ DEG,XS,M,NIC,TYD,DEGA
C
C      FIND INTERSECTIONS WITH SHOCK WAVES
C
TYPE 2600
ACCEPT 2700, XISO
ICNT=0
DO 200 KK=1,M
DO 100 I=1,M
PHICSN(I,KK)=PHICS(I,KK)
100 CONTINUE
200 CONTINUE
DO 900 I=1,M
DO 850 KK=1,M
IF (KK .GT. I) GO TO 850
DO 800 L=1,NIC
IF (XISO .GT. XS(I,KK,L)) GO TO 800
ICNT=ICNT+1
XRT=XS(I,KK,L)
TRT=TAU(I,KK,L)
IF ((KK .EQ. 1) .AND. (L .EQ. 1)) GO TO 790
IF ( L .EQ. 1) GO TO 785
XLT=XS(I,KK,L-1)
TLT=TAU(I,KK,L-1)
GO TO 795
785 XLT=XS(I,KK-1,NIC)
TLT=TAU(I,KK-1,NIC)
GO TO 795
790 XLT=0,
TLT=0,
795 TI(ICNT)=(XISO-XLT)*(TLT-TRT)/(XLT-XRT)+TLT
IF (I .EQ. 1) GO TO 7958
DO 7955 KK=1,I-1
IF (XISO .LT. XS(I,KK,NIC)) GO TO 7956
PHICSN(KK,I-1)=0.
PHICSN(KK,I)=0.
GO TO 7955
7956 IF (XISO .LT. XS(I,KK,1)) GO TO 7958
C
C      XISO INTERSECTS BETWEEN L==1 AND L=NIC
C
DO 7957 KKK=KK,I-1
PHICSN(KKK,I-1)=(XISO-XLT)*(PHI(I-1,KKK,KKK,L)-
1 PHI(I-1,KKK,KKK,L-1))/ (XRT-XLT) + PHI(I-1,KKK,KKK,L-1)

```

```

      PHICSN(KKK,I)=(XISO-XLT)*(PHI(I,KKK,KKK,L)-PHI(I,KKK,KKK,L-1))/
1 (XRT-XLT) + PHI(I,KKK,KKK,L-1)
7957 CONTINUE
7955 CONTINUE
7958 DO 796 MM=1,M
      IF (I .EQ. 1) GO TO 7995
      PHICON(ICNT,MM)=PHICSN(MM,I-1)
      GO TO 796
7995 PHICON(ICNT,MM)=0.
796 CONTINUE
      ICNT=ICNT+1
      TI(ICNT)=TI(ICNT-1)
      DO 799 MM=1,M
      PHICON(ICNT,MM)=PHICSN(MM,I)
799 CONTINUE
      GO TO 900
800 CONTINUE
850 CONTINUE
900 CONTINUE
C
C      FIND INTERSECTIONS WITH C-WAVES
C
      DO 990 KK=1,M
      DO 980 L=1,NIC
      DO 970 I=M,1,-1
      IF (I .LT. KK) GO TO 970
      IF (XISO .GT. XS(I,KK,L)) GO TO 970
      ICNT=ICNT+1
      XRT=XS(I,KK,L)
      TRT=TAU(I,KK,L)
      IF (I .EQ. M) GO TO 950
      XLT=XS(I+1,KK,L)
      TLT=TAU(I+1,KK,L)
      GO TO 960
950 XLT=0.
      TLT=TY
960 TI(ICNT)=(XISO-XLT)*(TLT-TRT)/(XLT-XRT)+TLT
      DO 966 MM=1,M
      PHICON(ICNT,MM)=PHI(I,MM,KK,L)
966 CONTINUE
      GO TO 980
970 CONTINUE
980 CONTINUE
990 CONTINUE
C
C      PUT TI'S IN ASCENDING ORDER
C
      IEND=ICNT-1
      DO 999 I=1,IEND
      DO 997 J=1,I
      II=I-J+1
      IF (TI(II+1) .GE. TI(II)) GO TO 997
      TSAV=TI(II)
      TI(II)=TI(II+1)

```

```
TI(II+1)=TSAV
DO 995 MM=1,M
PHISAV=PHICON(II,MM)
PHICON(II,MM)=PHICON(II+1,MM)
PHICON(II+1,MM)=PHISAV
995 CONTINUE
997 CONTINUE
999 CONTINUE
WRITE(6,2750) XDATE,XTIME,XISO
DO 998 I=1,ICNT
DEE(I)=ROT*TI(I)*50./(V*60.)
WRITE(6,2800) TI(I),DEE(I),(PHICON(I,MM),MM=1,M)
998 CONTINUE
RETURN
2600 FORMAT(' ENTER THE DIMFNSIONLESS BED LENGTH DESIRED IN F8.4
1FORMAT. ')
2700 FORMAT(F8.4)
2750 FORMAT('1',/T10,9A1,3X,8A1, //T10, 'PROFILE DATA', /T10, 11('*'), //T20,
1 'TAU', T30, 'DEGREES', T40, 'PHI''S:', T80,
2 'X=', F5.2, //)
2800 FORMAT(T15, 8F10.4)
END
```

```

SUBROUTINE PLOTPR(NSOL)
LOGICAL*1 XTIME(8),XDATE(9)
DIMENSION X(72),Y(72,6),DUM(6,6)
COMMON /BLK2/XDATE,XTIME
COMMON /BLK1/MINX,MAXX,MINY,MAXY
COMMON /BLK3/ XISO,DUM,X,Y,NPTS
ICNT=0
50  CALL INITT(480)
    ICNT=ICNT+1
    TYPE 60
60  FORMAT(' ENTER RANGE (MINIMUM AND MAXIMUM) VALUES FOR ANGULAR RO
ITATION IN 2F8.4 FORMAT.')
```

ACCEPT 70, XMIN,XMAX

```

70  FORMAT(2F8.4)
    TYPE 80
80  FORMAT(' ENTER RANGE (MINIMUM AND MAXIMUM) VALUES OF
1CONCENTRATIONS IN 2F8.4 FORMAT.')
```

ACCEPT 70, YMIN,YMAX

```

MINX=150
MAXX=900
MINY=150
MAXY=700
CALL TWINDO(MINX,MAXX,MINY,MAXY)
CALL DWINDO(XMIN,XMAX,YMIN,YMAX)
```

C
C DRAW CURVES FOREACH SOLUTE
C

```

DO 200 I=1,NSOL
CALL MOVEA(X(1),0.)
DO 100 J=1,NPTS
CALL DASHA(X(J),Y(J,I),I)
100  CONTINUE
200  CONTINUE
CALL GRID(XMIN,XMAX,YMIN,YMAX)
IXO=MINX
IYO=730
CALL MOVABS(IXO,IYO)
CALL ANMODE
WRITE(1,300) XDATE,XTIME,XISO,ICNT
300  FORMAT(1H+,$,9A1,3X,8A1,5X,'X=',F4.1,3X,'GRAPH NO.',I2)
    TYPE 1000
1000 FORMAT(' DO YOU WANT ANOTHER PLOT WITH DIFFERENT SCALING?
1 (0=NO; 1=YES):',6)
ACCEPT 1100, IA
1100 FORMAT(I2)
IF (IA .NE. 0) GO TO 50
RETURN
END
```


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APPENDIX B. GRAPHICAL RESULTS

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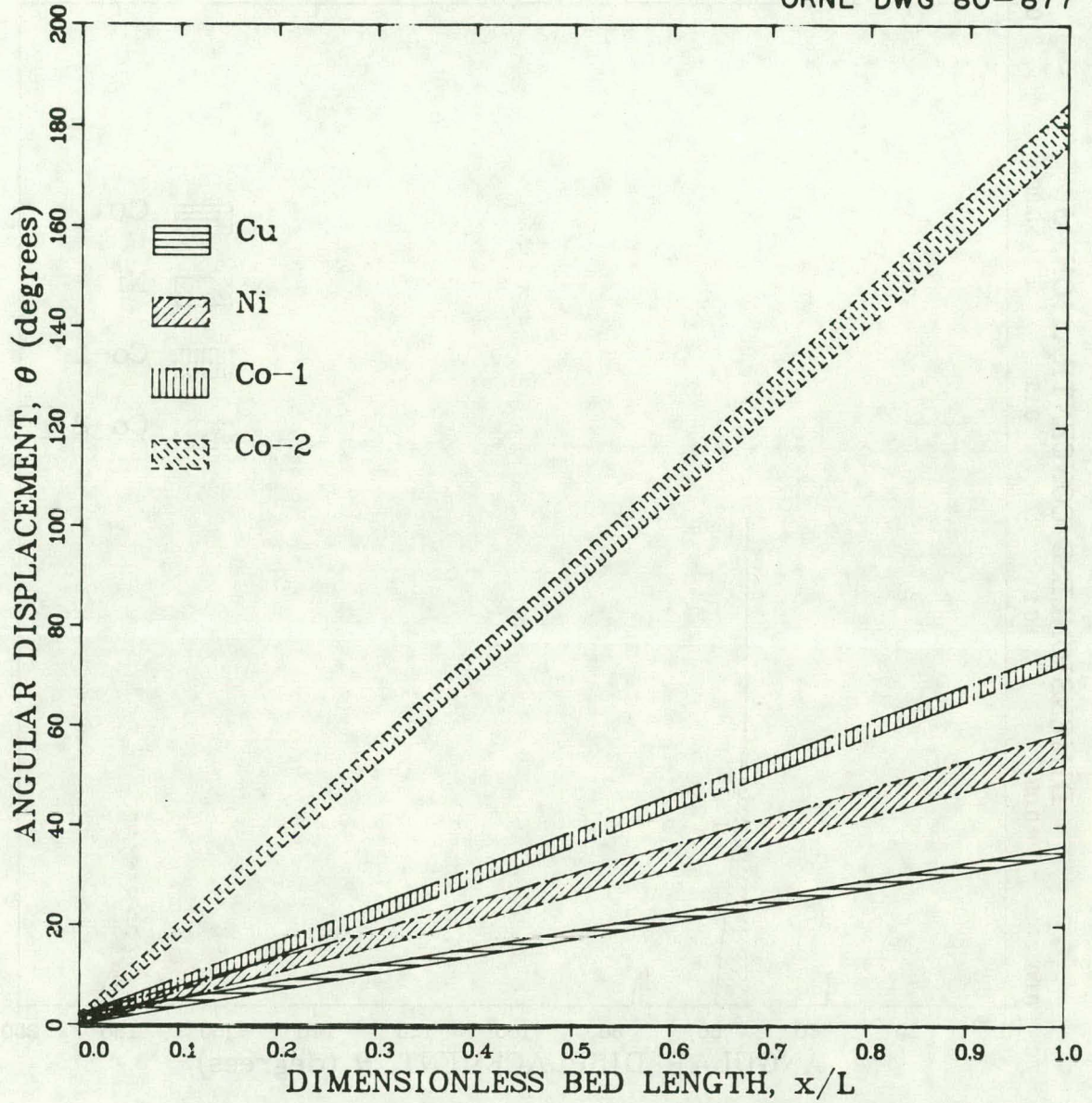


Fig. B.1. Solute bands along the column length for a 5-mm feed bandwidth.

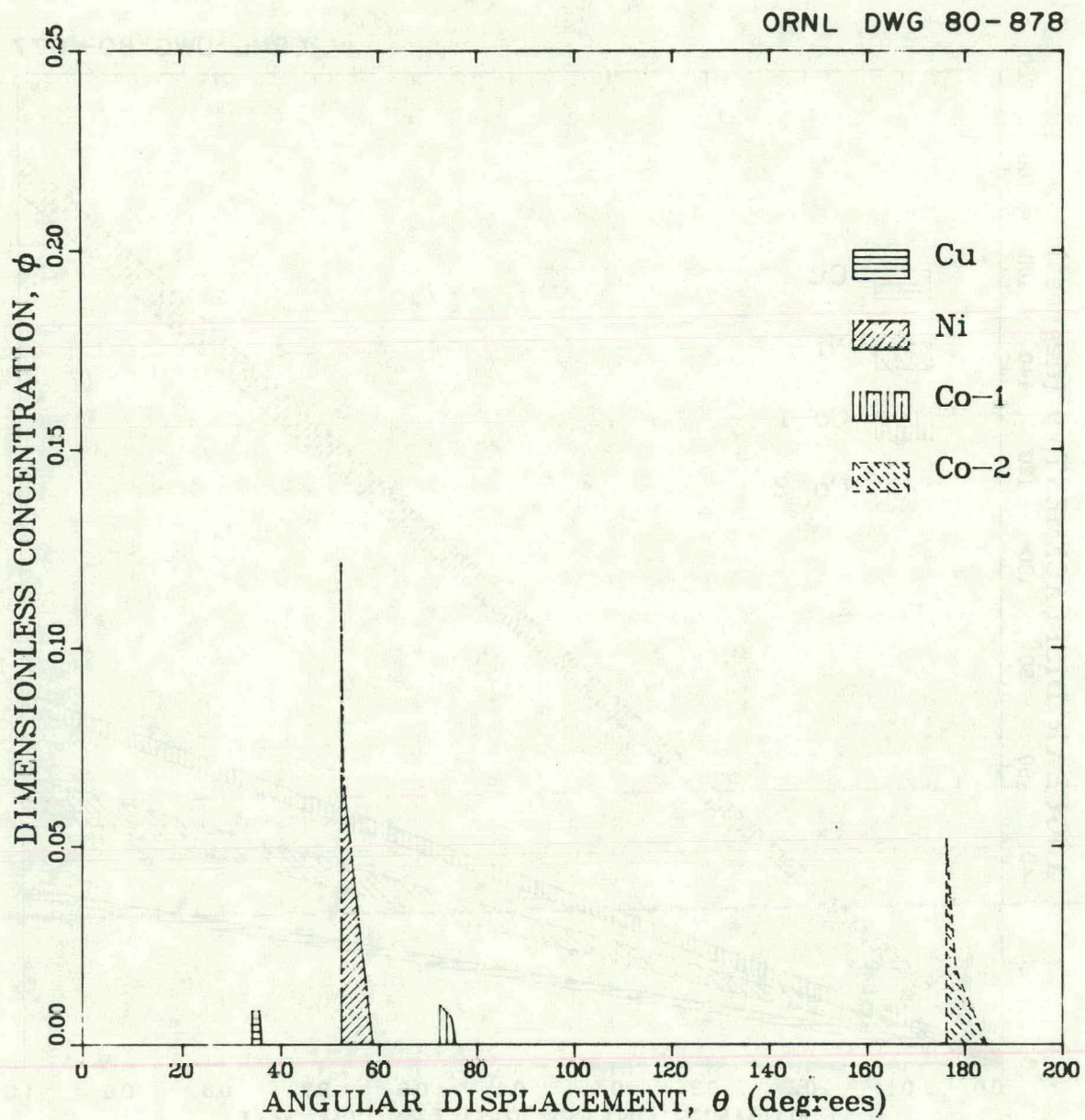


Fig. B.2. Concentration profile at the column exit for a 5-mm feed bandwidth.

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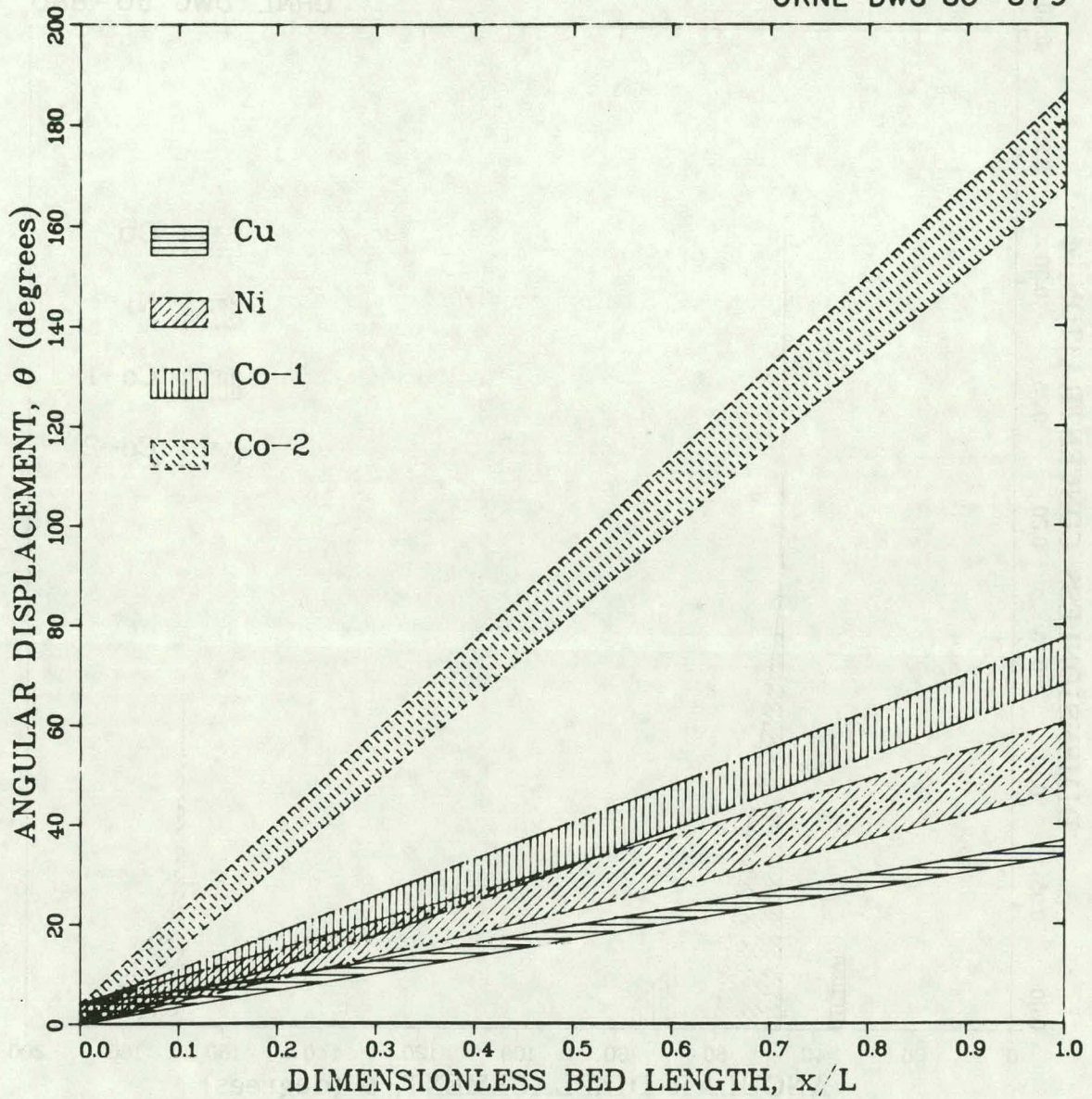


Fig. B.3. Solute bands along the column length for a 10-mm feed bandwidth.

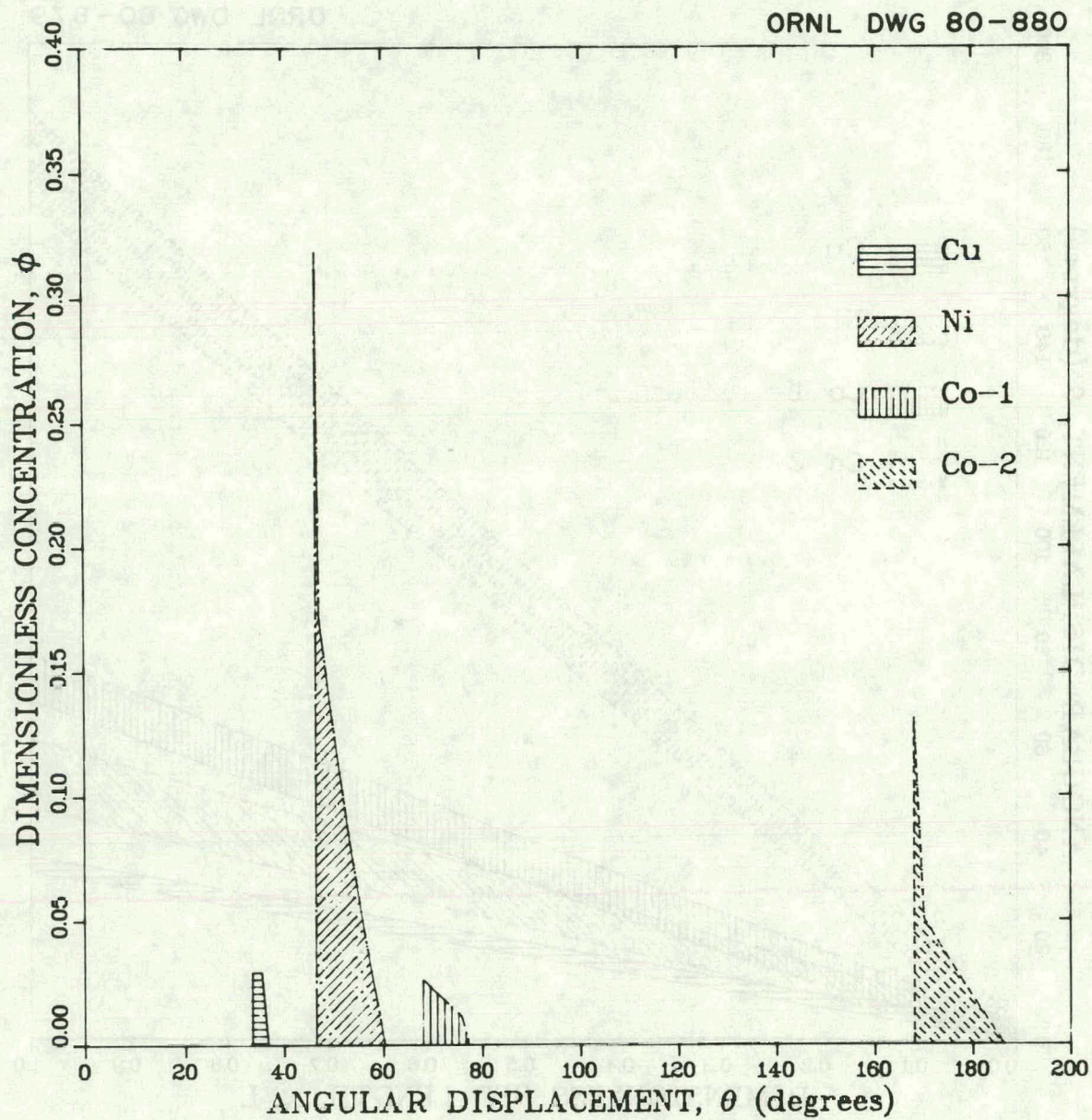


Fig. B.4. Concentration profile at the column exit for a 10-mm feed bandwidth.

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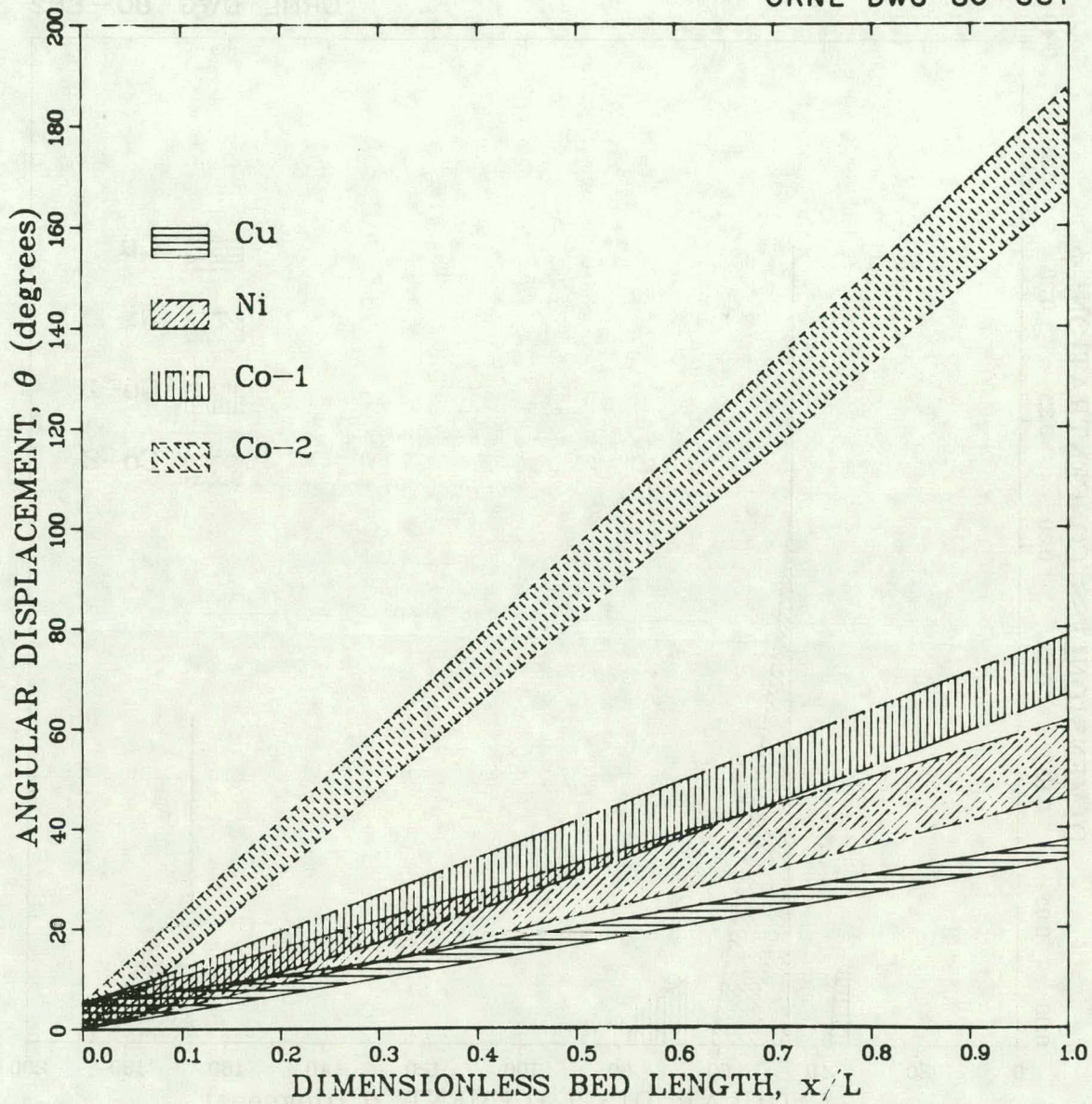


Fig. B.5. Solute bands along the column length for a 13-mm feed bandwidth.

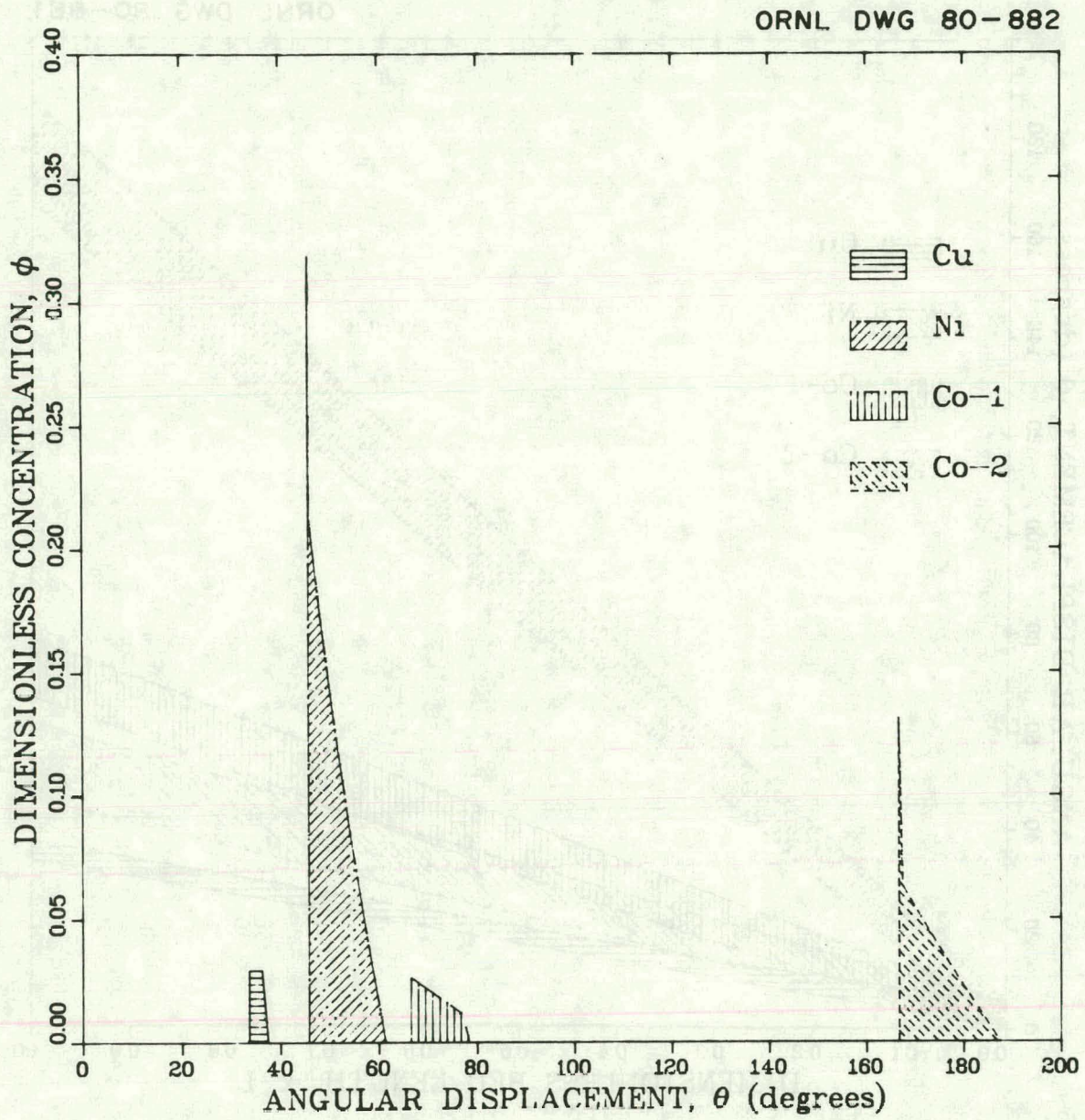


Fig. B.6. Concentration profile at the column exit for a 13-mm feed bandwidth.

ORNL DWG 80-883

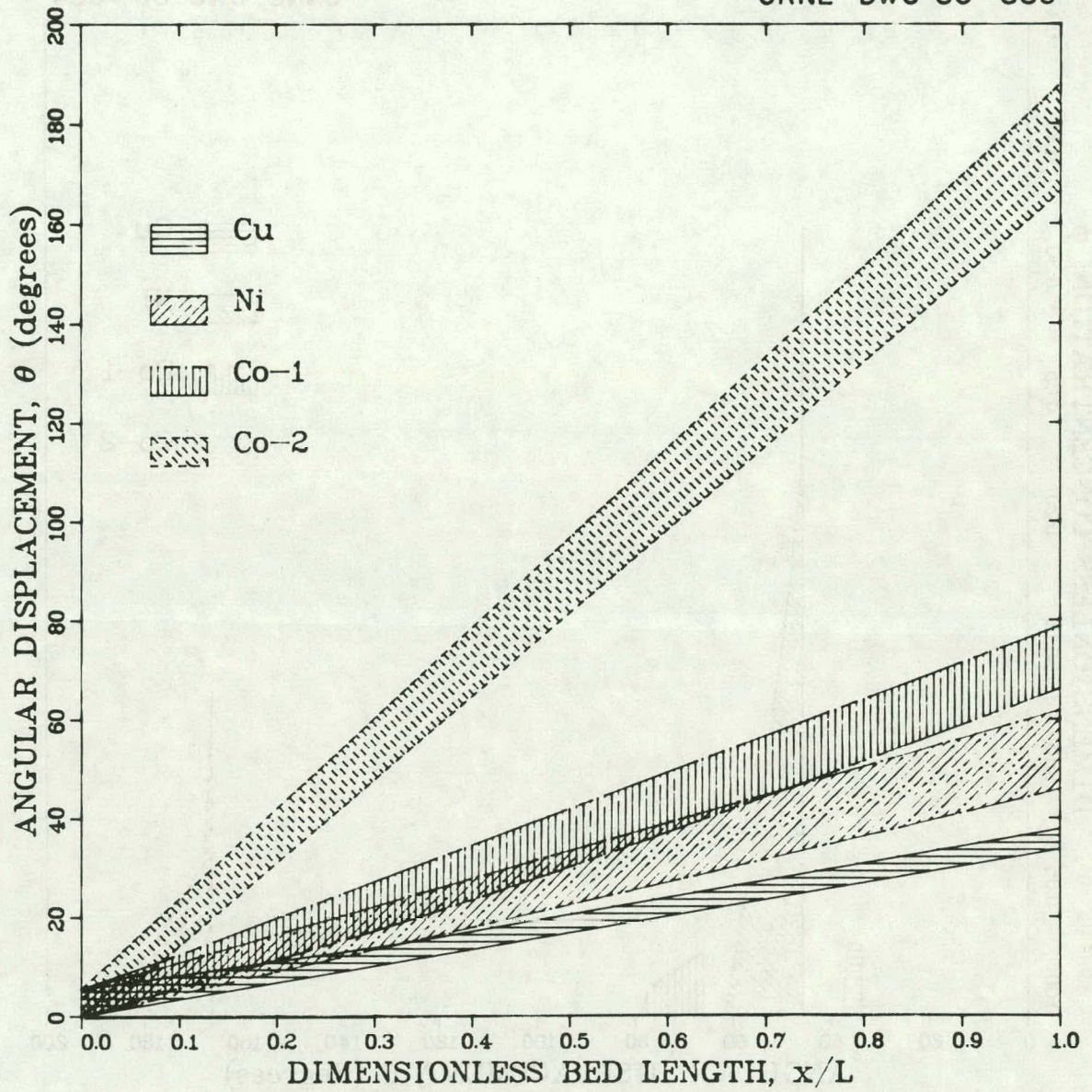


Fig. B.7. Solute bands along the column length for a 14-mm feed bandwidth.

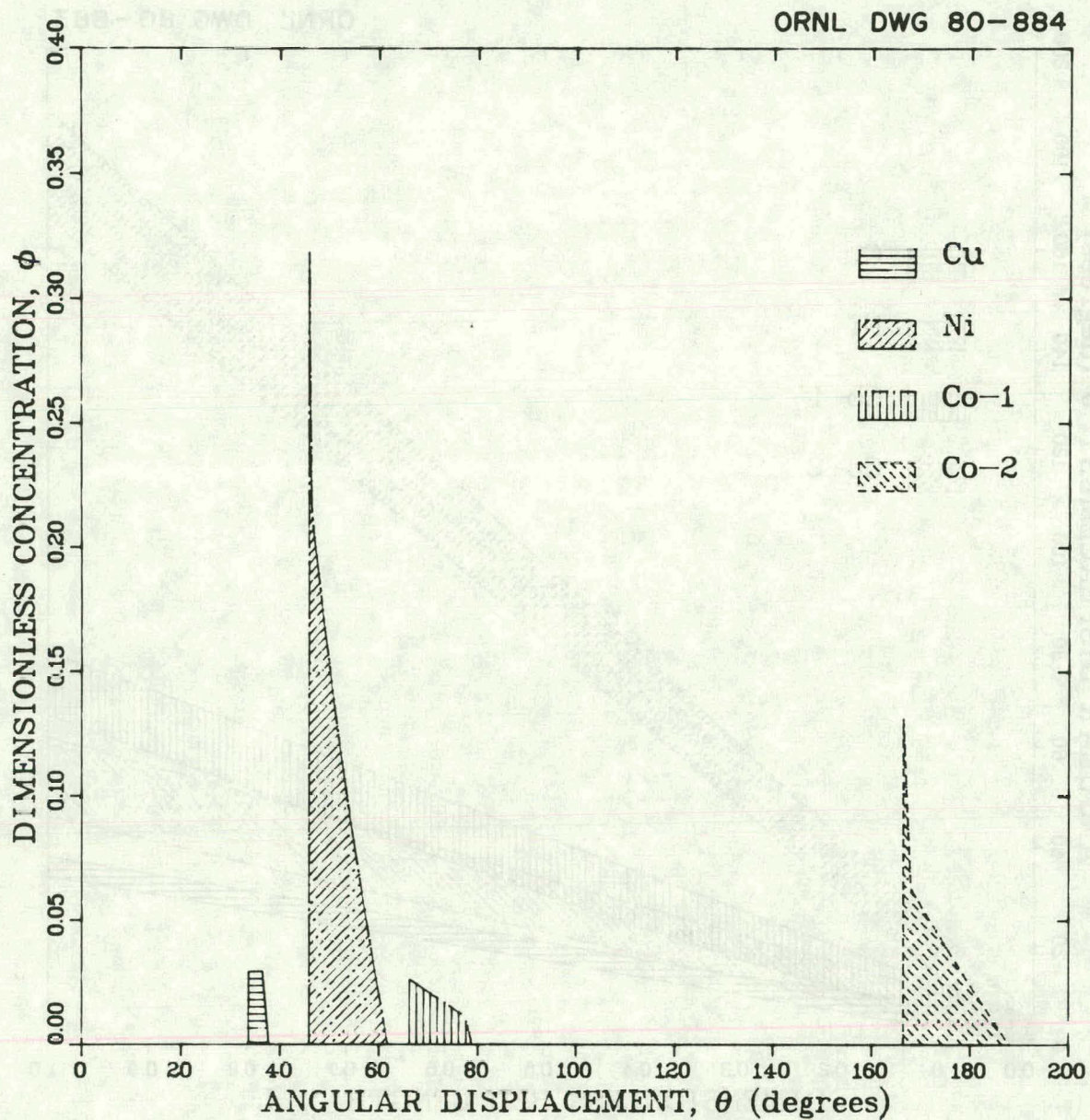


Fig. B.8. Concentration profile at the column exit for a 14-mm feed bandwidth.

ORNL DWG 80-885

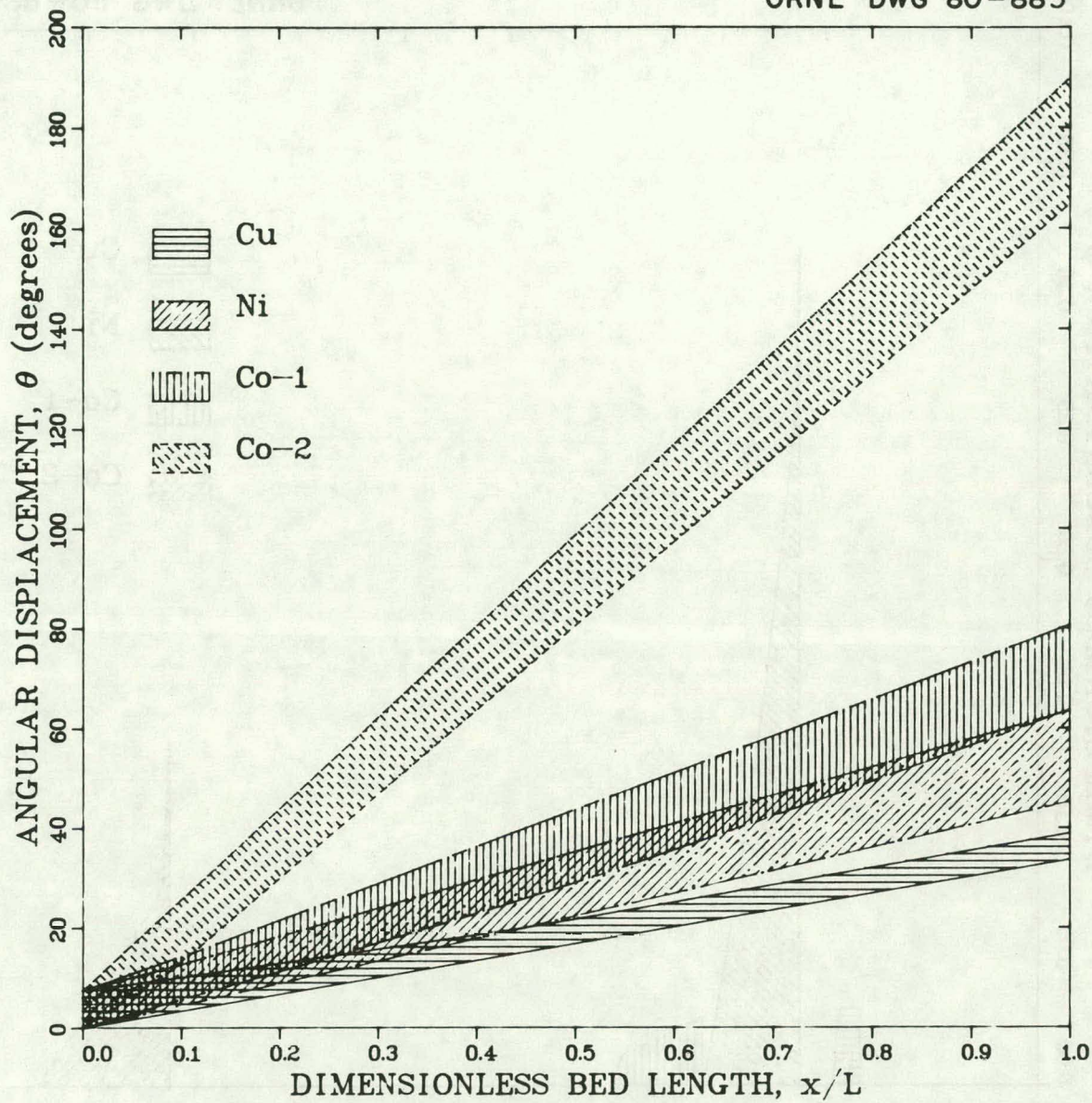


Fig. B.9. Solute bands along the column length for a 19-mm feed bandwidth.

ORNL DWG 80-886

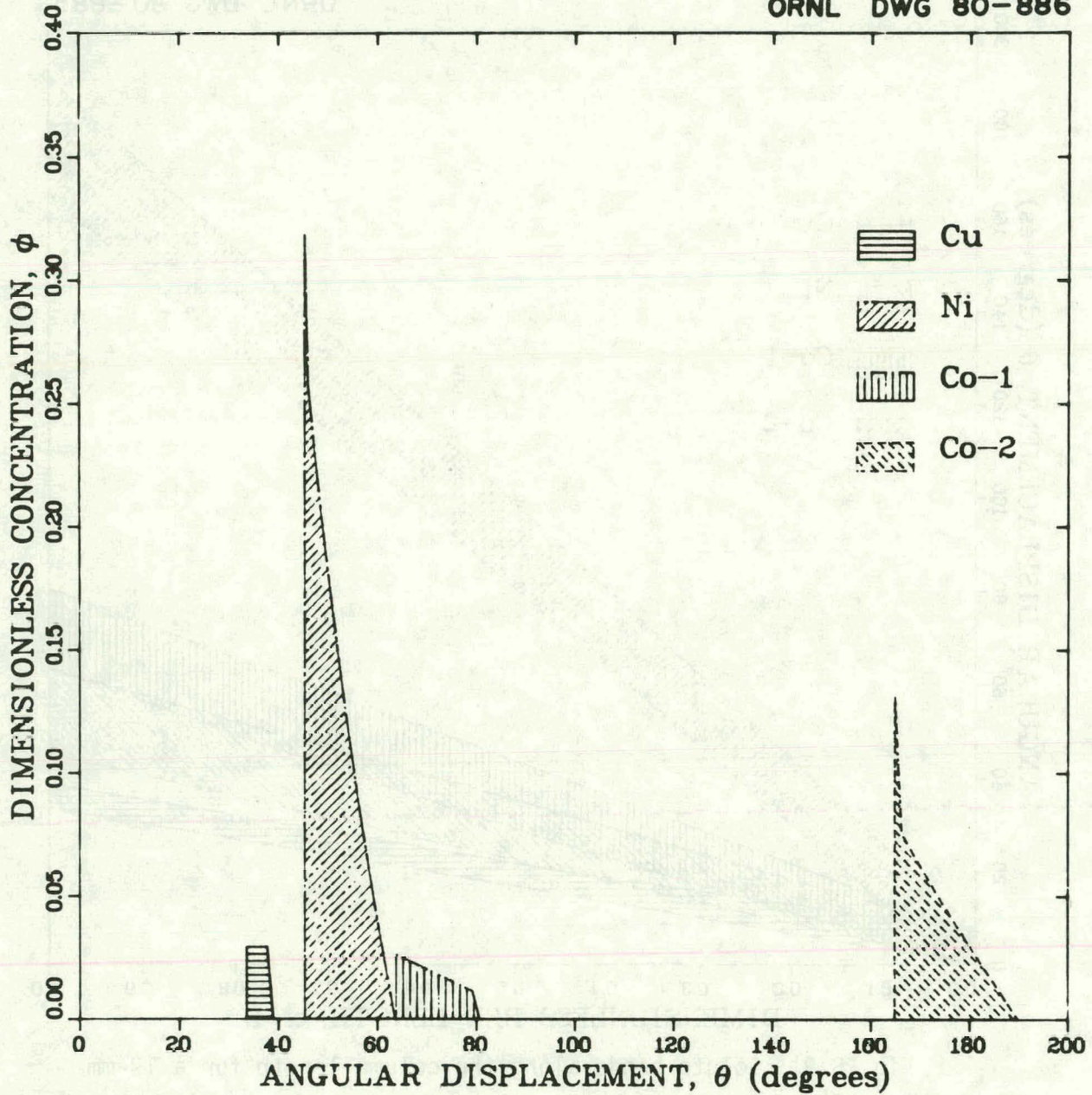


Fig. B.10. Concentration profile at the column exit for a 19-mm feed bandwidth.

ORNL DWG 80-887

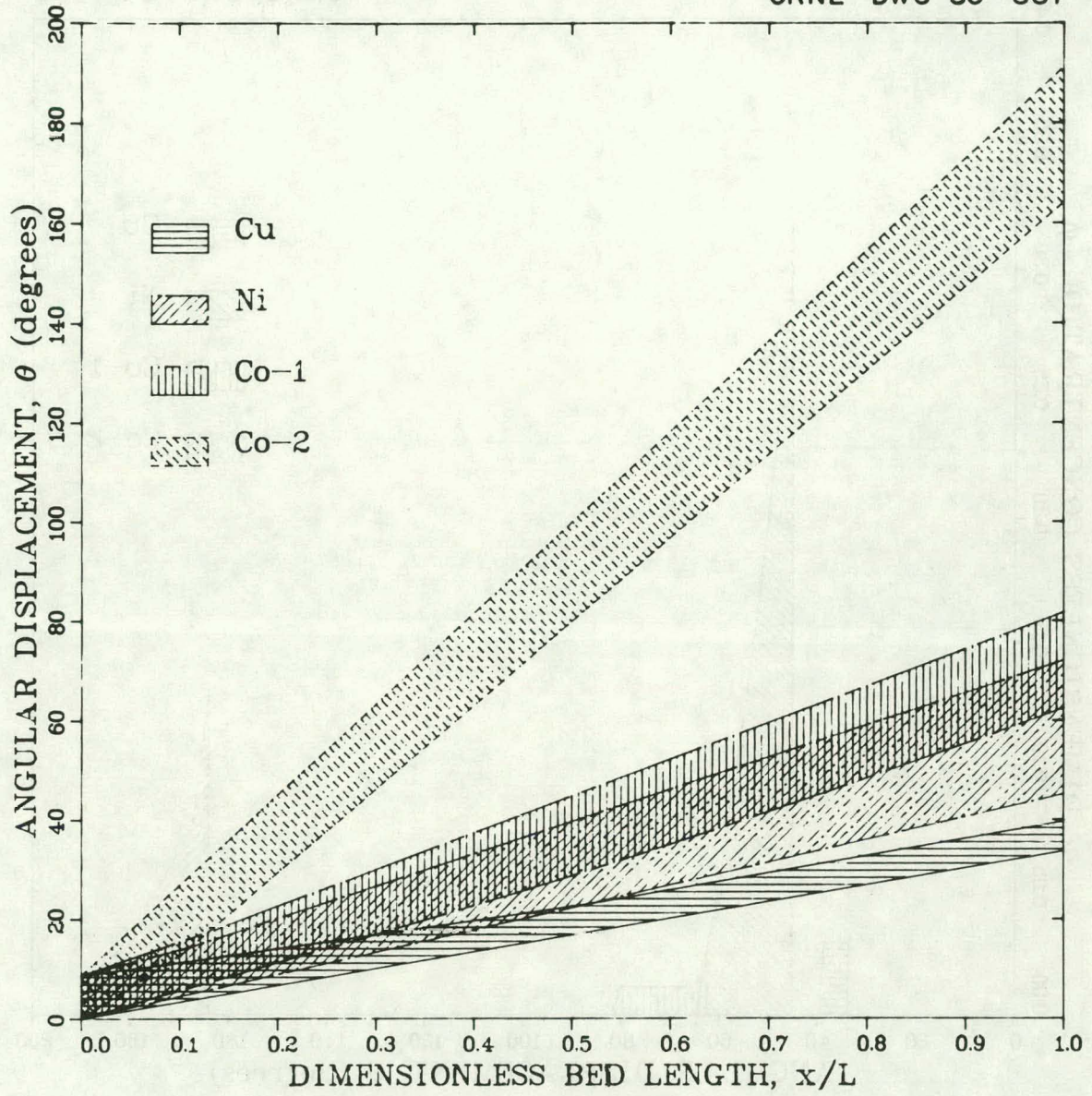


Fig. B.11. Solute bands along the column length for a 22-mm feed bandwidth.

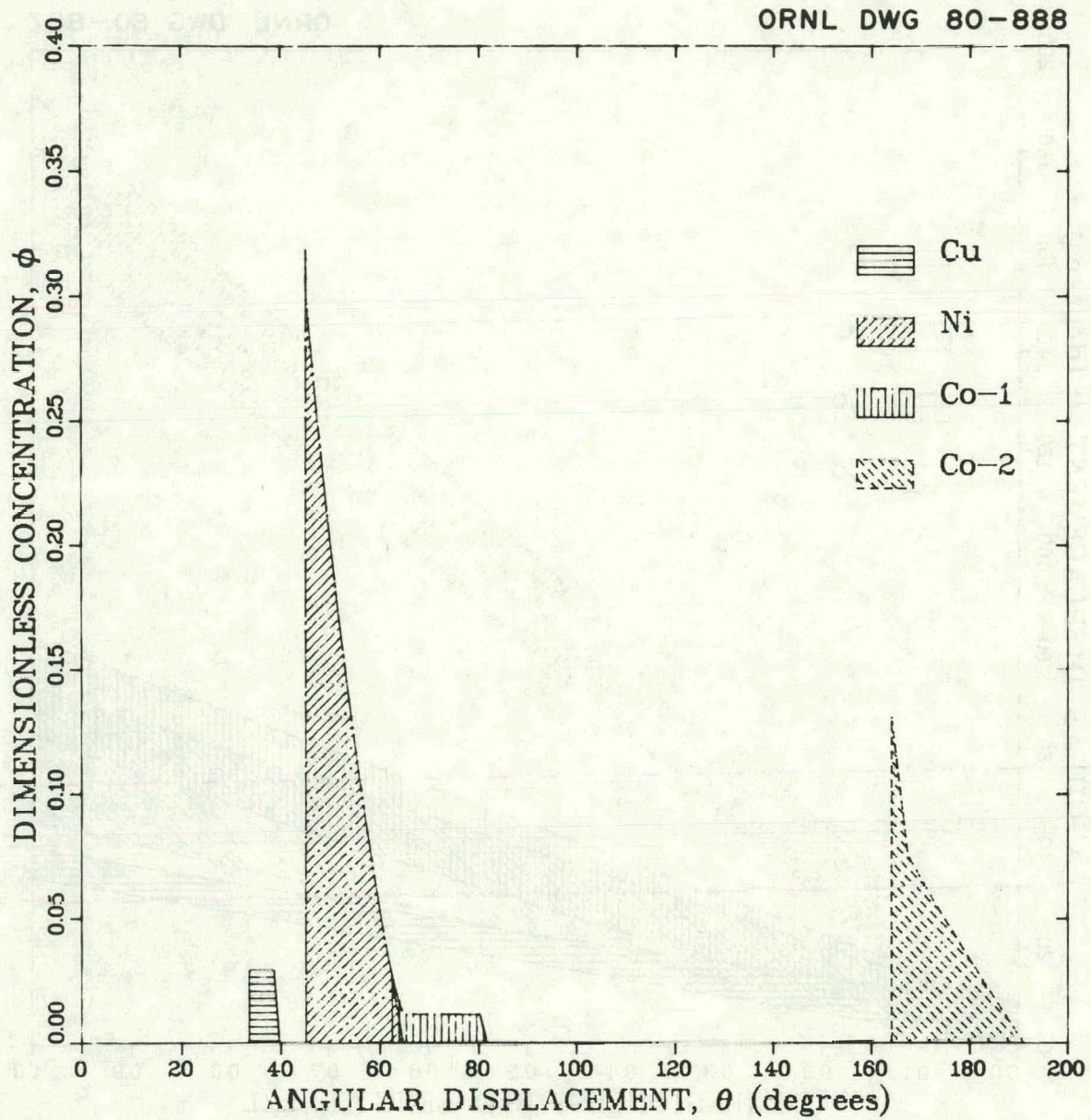


Fig. B.12. Concentration profile at the column exit for a 22-mm feed bandwidth.

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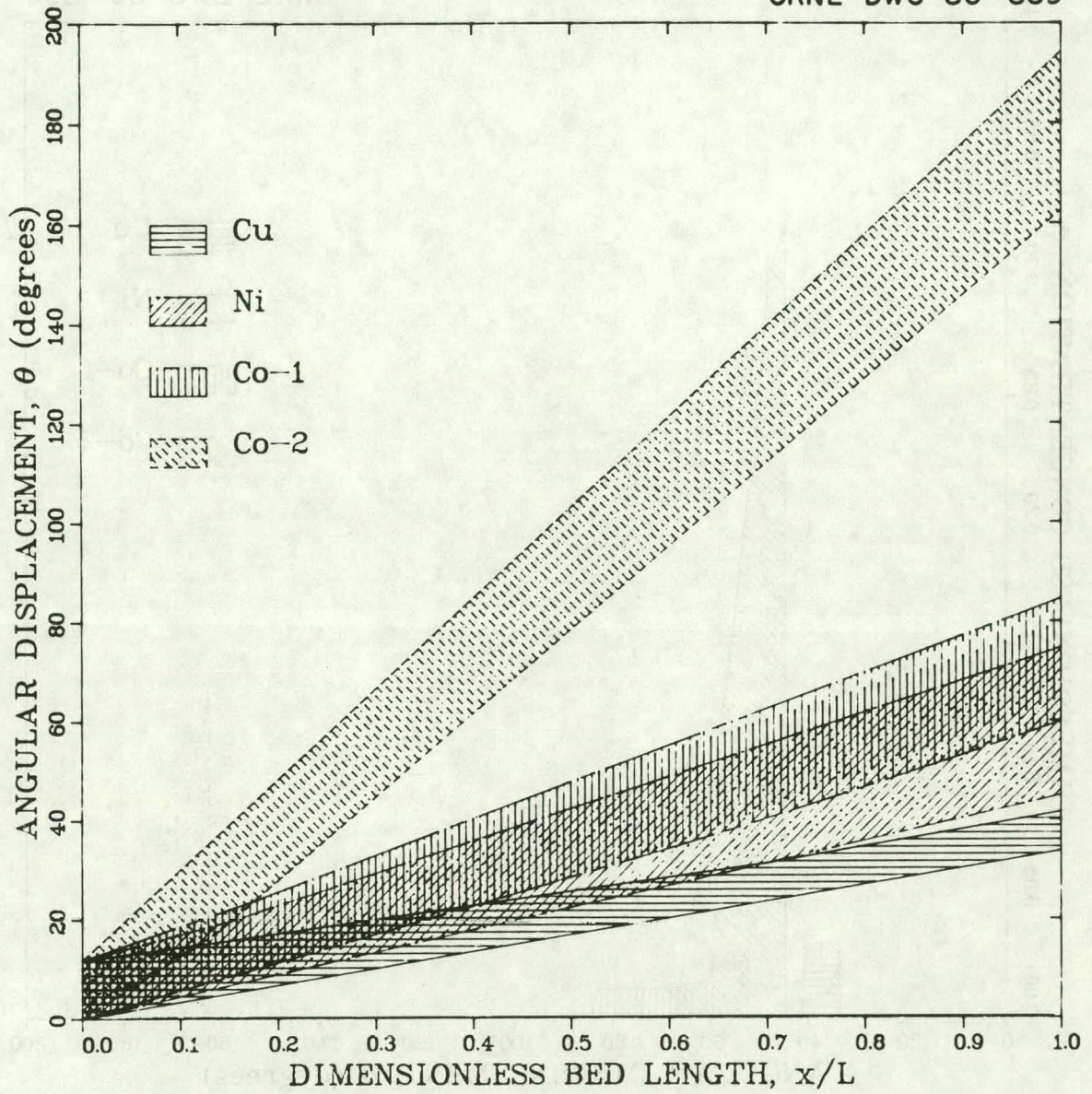


Fig. B.13. Solute bands along the column length for a 30-mm feed bandwidth.

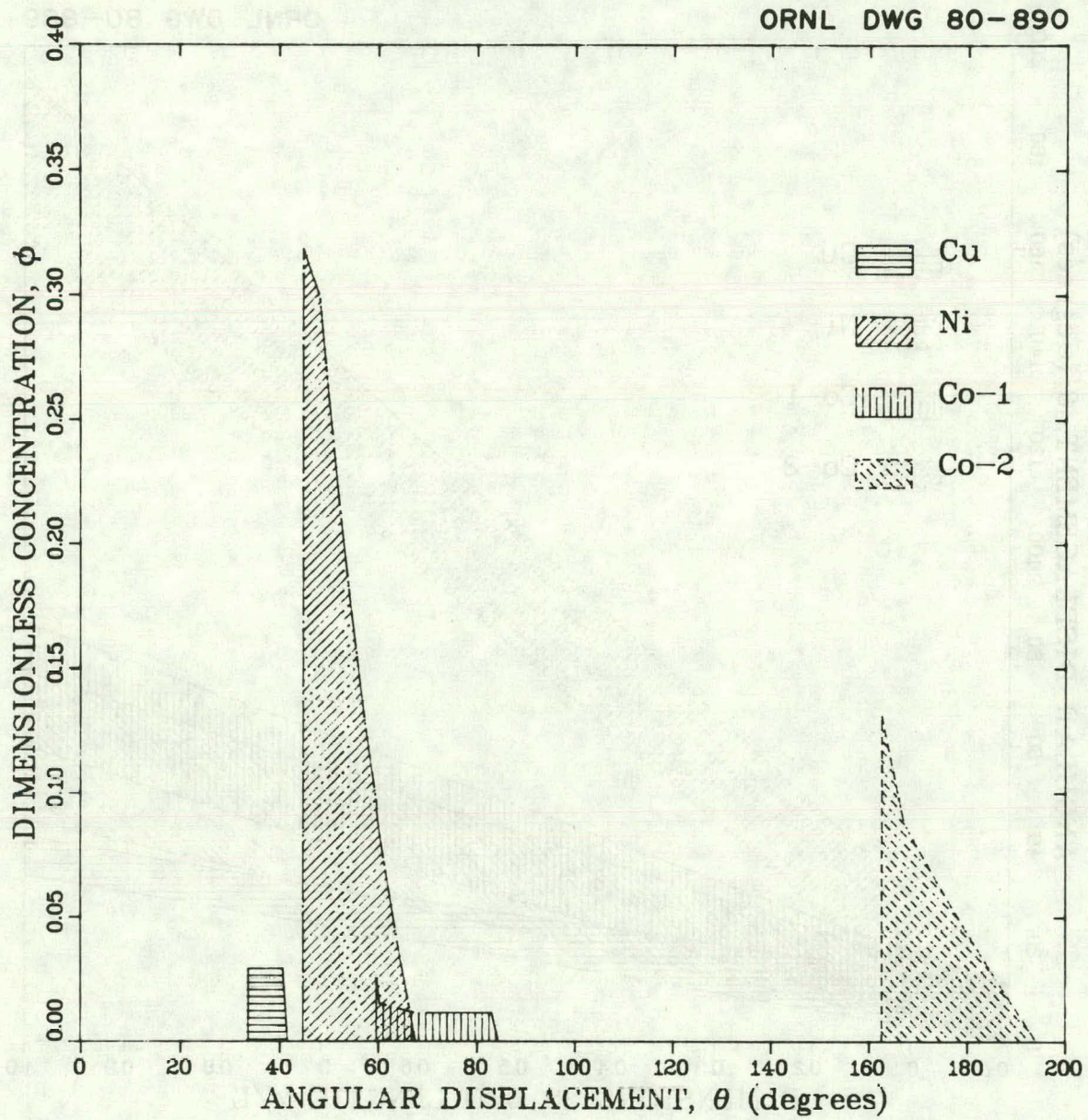


Fig. B.14. Concentration profile at the column exit for a 30-mm feed bandwidth.

ORNL DWG 80-891

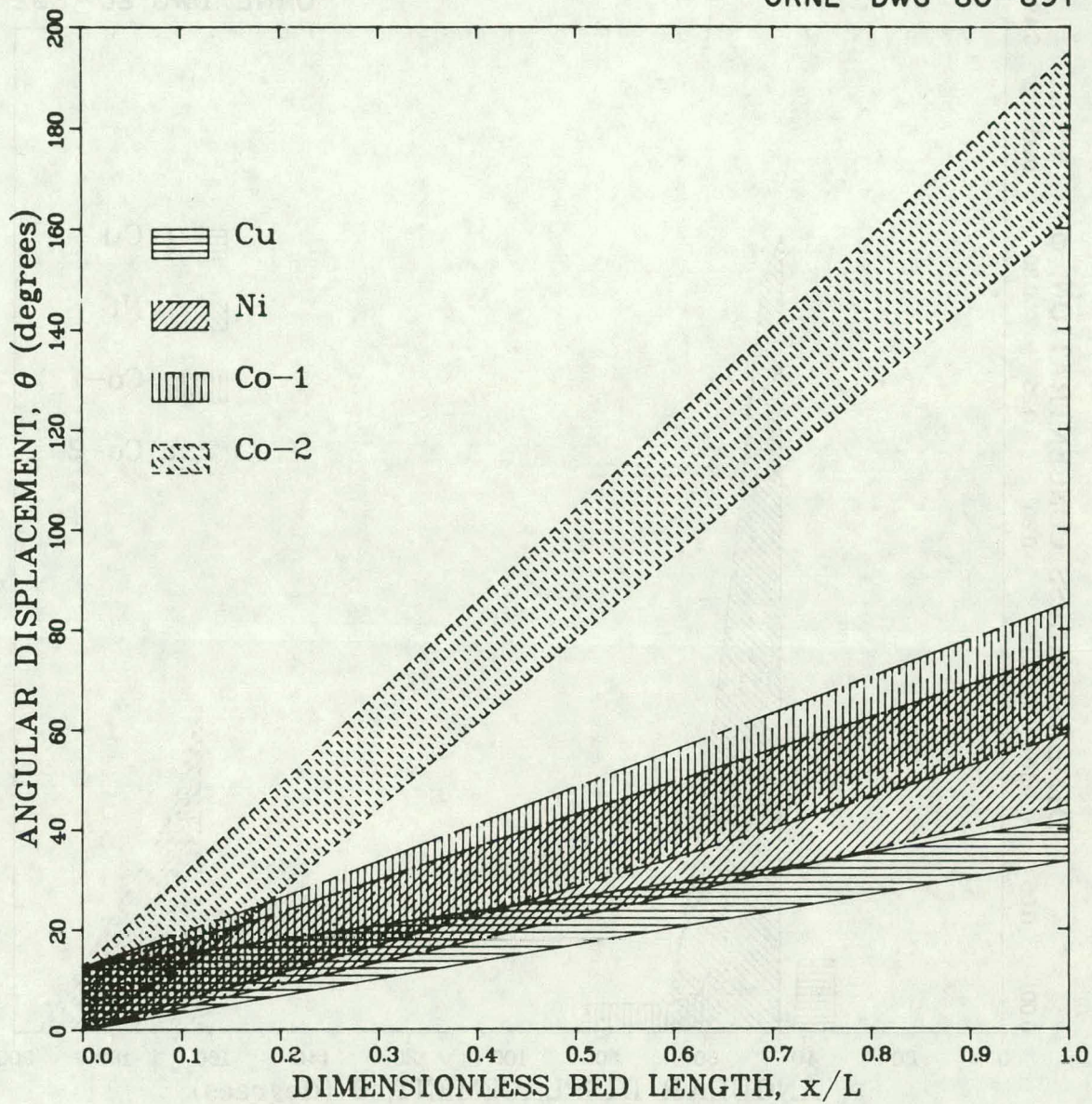


Fig. B.15. Solute bands along the column length for a 32-mm feed bandwidth.

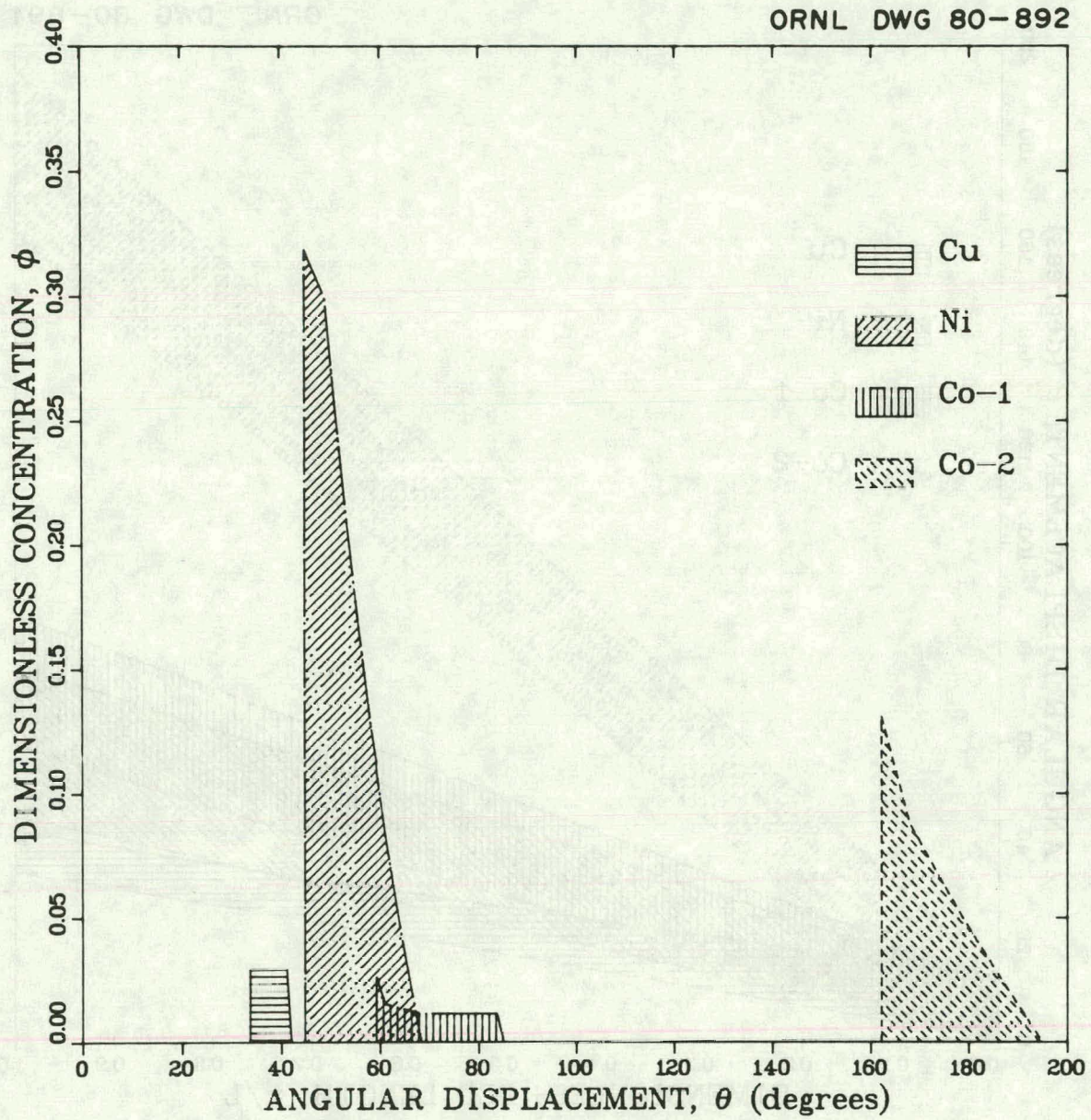


Fig. B.16. Concentration profile at the column exit for a 32-mm feed bandwidth.

ORNL DWG 80-893

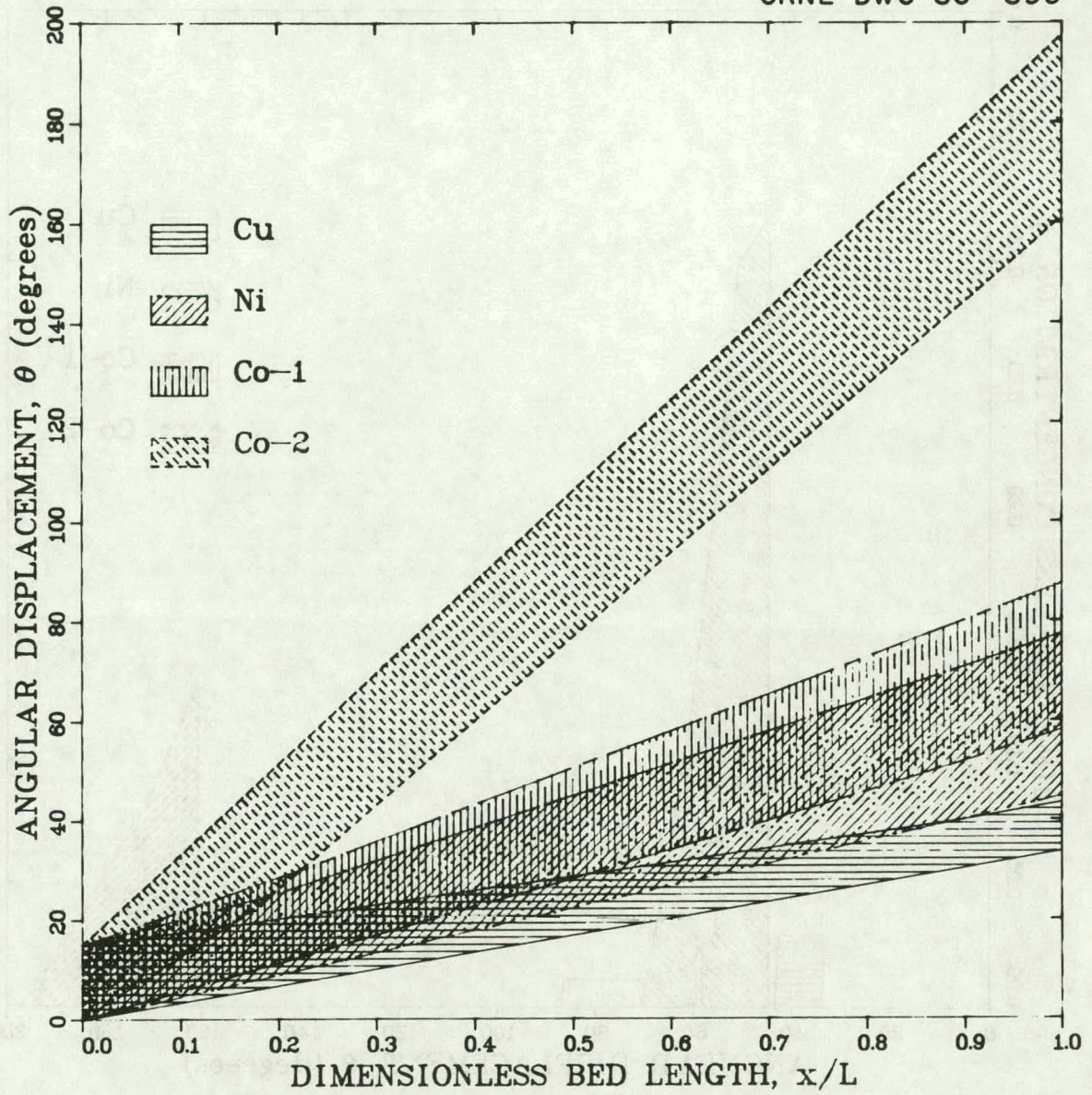


Fig. B.17. Solute bands along the column length for a 38-mm feed bandwidth.

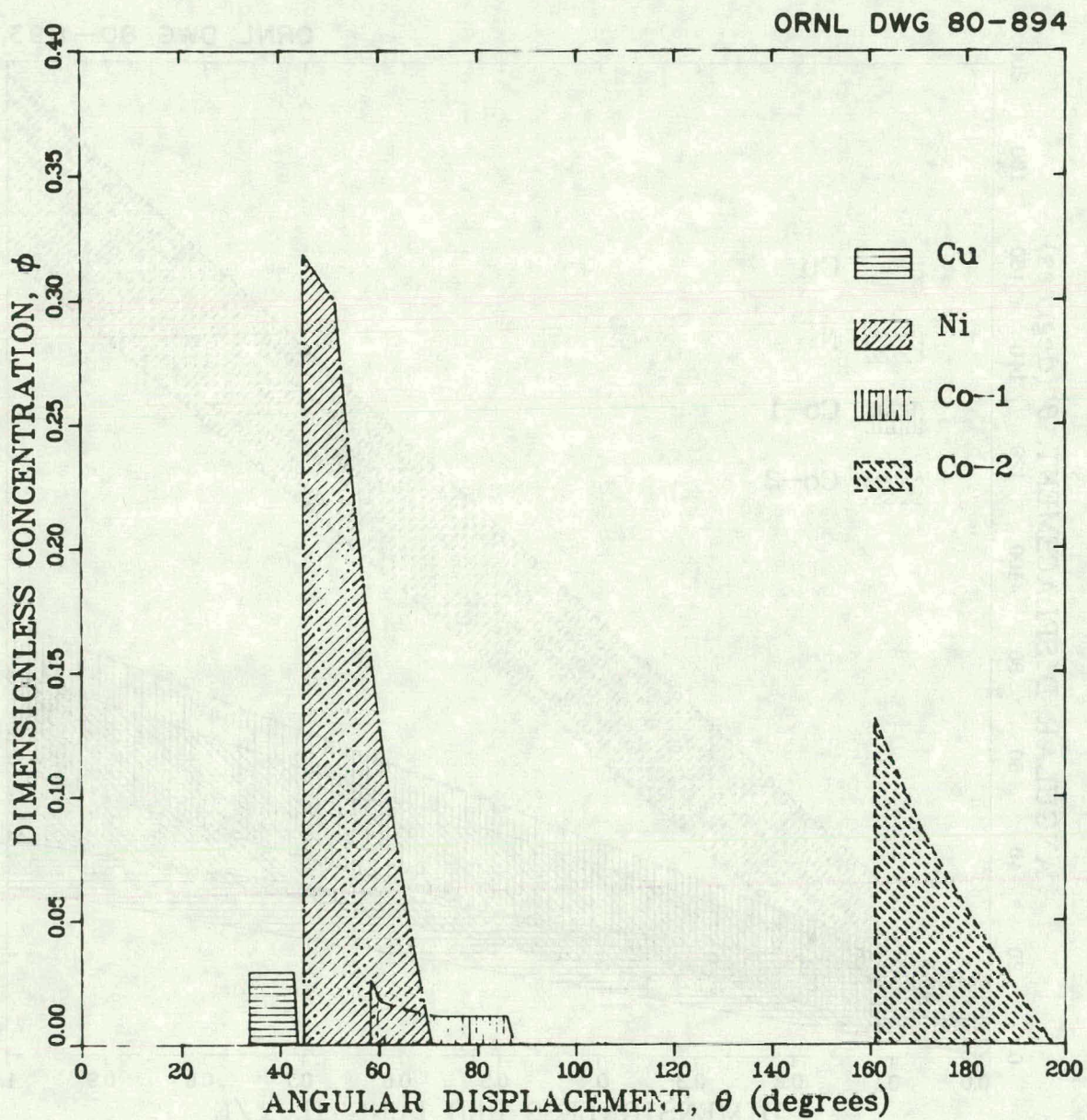


Fig. B.18. Concentration profile at the column exit for a 38-mm feed bandwidth.

8. REFERENCES

1. L. Lapidus and N. Amundson, "Mathematics of Adsorption in Beds. VI. The Effect of Longitudinal Diffusion in Ion Exchange and Chromatographic Columns," *J. Phys. Chem.* 56, 984 (1952).
2. H. K. Rhee, R. Aris, and N. R. Amundson, "On the Theory of Multi-component Chromatography," *Trans. R. Soc.* A267, 419 (1970).
3. R. Aris and N. R. Amundson, *AIChE J.* 3, 230 (1957).
4. C. D. Scott, R. D. Spence, and W. G. Sisson, "Pressurized Annular Chromatograph for Continuous Separations," *J. Chromatogr.* 126, 381 (1976).
5. R. Y. Shah, J. H. Austin, and L. M. Burelle, Comprehensive Model for an Annular Continuous Chromatograph, ORNL/MIT-205 (December 1974).
6. R. M. Canon and W. G. Sisson, "Operation of an Improved, Continuous Annular Chromatograph," *J. Liq. Chromatogr.* 1(4), 427 (1978).
7. R. M. Canon, Section 2.1 in Advanced Technology Section Semiannual Progress Report for the Period Oct. 1, 1977, to Mar. 31, 1978. Volume 2: Engineering Science Programs, ORNL/TM-6377/V2 (July 1980).

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