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Mineralogy and Phase Chemistry of Mount St. Helens Tephra Sets W and Y as Keys to Their Identification

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Voluminous and widespread tephra sets were produced frequently during the last 36,000 yr of volcanic activity at Mount St. Helens. Numerous tephra sets have been defined by D. R. Mullineaux, J. H. Hyde, and M. Rubin (1975, *U.S. Geological Survey Journal of Research*, 3, 329-335) on the basis of field relations, Fe-Mg phenocryst assemblage, and ^{14}C chronology and are valuable marker beds for regional stratigraphic studies. In this study modal abundances and mineral compositions were determined (via petrographic and electron microprobe techniques) for numerous samples of individual layers within tephra sets W and Y to evaluate the degree of compositional variability within and between tephra layers and criteria by which to distinguish among Mount St. Helens and other Pacific Northwest tephra sets. Although individual layers within a set (e.g., W_e , W_n) cannot be distinguished from each other on the basis of mineralogic characteristics examined, mineral compositions allow distinction among layers W and Y and other Pacific Northwest tephra sets (e.g., Mazama, Glacier Peak). Fe-Ti oxide compositions and $T-f_{\text{O}_2}$ estimates derived using coexisting magnetite-ilmenite are especially useful due to the compositional homogeneity of these minerals both within and between samples of a given unit over a wide geographic area. The silicates show more compositional variability than the oxides, but $\text{SiO}_2/\text{Al}_2\text{O}_3$ contents in hornblende and Fe/Mg ratios in hypersthene aid in distinguishing among Pacific Northwest tephra sets.

INTRODUCTION

The present-day cone and amphitheater of Mount St. Helens volcano are spatially associated with an ancestral volcanic center which came into existence at least 36,000 yr ago (Hyde, 1973; Mullineaux *et al.*, 1975). Since that time, eruptions of pumiceous tephra sets, pyroclastic flows, lavas, and domes were produced intermittently from the volcano. Detailed investigations of the eruptive history conducted previous to the 1980 eruptions have provided a time-stratigraphic framework for many of the eruptive products (Hopson, 1971; Hyde, 1973, 1975; Crandell and Mullineaux, 1973, 1978; Crandell *et al.*, 1975; Mullineaux *et al.*, 1975; Hoblitt *et al.*, 1980). A sequence of numerous tephra sets has been defined by Mullineaux *et al.* (1975) on the basis of field relations, ferromagnesian phenocryst

assemblages, and ^{14}C geochronology (Fig. 1). Within each set there may be more than one layer of tephra, representing eruptions so closely spaced in time that no significant soil horizon developed between eruptions.

Many of the tephra layers are quite voluminous and widespread. Some layers (e.g., We, Wn) have thicknesses of 20 cm or more at distances of tens of kilometers from the volcano (Crandell and Mullineaux, 1978) and several layers have been traced as far as 900 km from the source (e.g., Westgate and Fulton, 1975; Smith *et al.*, 1977; Westgate, 1977; Brewster and Barnett, 1979).

Due to their widespread distribution and their well-documented chronology, Mount St. Helens (MSH) tephras, when properly identified, are valuable marker beds for regional stratigraphic correlation and can be employed in a variety of geologic, archaeological, and limnologic studies. MSH tephras are intercalated with tephras from other volcanic sources in the Pacific Northwest (e.g., Mount Mazama and Glacier Peak). Whereas tephras derived from different centers are usually distinguishable from one another, tephras erupted from the same center but at significantly different times may differ only slightly (Izett *et al.*, 1970, p. 121). As Westgate and Fulton noted (1975, p. 489), positive identification of distal portions of widespread airfall eruptive products may not be possible simply on the basis of field criteria such as coloration, degree of weathering, granulometry, thickness, and stratigraphic position. Detailed petrographic and compositional studies are necessary in order to distinguish and utilize individual MSH units in stratigraphic investigations.

It is necessary to evaluate the degree of homogeneity of individual MSH layers before mineralogic criteria can be used confidently in stratigraphic correlations of tephras. In this paper we report new mineralogical data, including modal abundances and mineral compositions, for two of the most voluminous silicic tephra sets, W and Y. These data are used to estimate the inherent variability in mineralogical and compositional characteristics to be expected for these and other tephra units. As published mineralogic data for the MSH tephras are sparse, the data collected in this study will also provide a basis for petrogenetic studies and evaluation of eruptive mechanisms for the explosive silicic tephras.

PREVIOUS WORK

Work toward characterization and identification of MSH tephtras has involved identification of ferromagnesian silicate assemblages (Hyde, 1973; Mullineaux *et al.*, 1975). The difficulty with this approach lies in the facts that the assemblages are repeated in successive MSH units (Fig. 1) and some are similar to those of tephtra from other Cascade sources (Randle *et al.*, 1971; Porter, 1978).

Other workers have characterized some of the tephtras on the basis of elemental ratios in the glass as determined by electron microprobe (e.g., Smith and Westgate, 1969; Okazaki *et al.*, 1972; Smith *et al.*, 1977). Figure 2 illustrates Ca-Fe-K ratios for Glacier Peak tephtra, the climactic eruption of Mount Mazama, Bridge River tephtra, and numerous MSH tephtras. On this basis one can readily distinguish among Glacier Peak, Mazama, Bridge River, and several MSH tephtras; however, differences between some MSH tephtras are not very great and individual eruptions within a set cannot always be distinguished (e.g., Ye and Yn). The effectiveness of this method may be lessened when applied to older units due to alteration or leaching of the glasses which is commonly accompanied by depletion of alkalies (Lipman, 1965).

Small differences in trace element abundances in glass and mineral separates have allowed successful discrimination of certain units (Dudas *et al.*, 1973; Randle *et al.*, 1971; Sarna-Wojcicki *et al.*, 1980). However, this requires preparation of pure fractions (not always feasible), and possible compositional zoning within individual tephtra layers has not been evaluated.

Scanning electron microprobe studies of glass shards have been made for tephtra layers We and Wn to define morphologic characteristics (Smith *et al.*, 1977). No distinguishing morphology for pumice lumps of lapilli size were observed, but these units could be differentiated on the basis of morphological characteristics of ash-size glass shards from more than 200 km from the volcano.

Westgate and his colleagues (Westgate and Fulton, 1975; Westgate *et al.*, 1977; Westgate and Evans, 1978; Brewster and Barnett, 1979; Mathewes and Westgate, 1980) have recognized the effectiveness of iron-titanium oxide composition as a tool in distinguishing tephtras in the plains of southwestern Canada. Their investigations include units which they believe to be the distal portions of tephtras from MSH on the basis of several criteria. However, no systematic detailed characterization has been performed on individual layers near the volcano where stratigraphic relations can be discerned unambiguously from field criteria.

METHODS OF STUDY

Sampling

Tephra sets W and Y are each composed of several layers of pumice lapilli which represent separate eruptions and are distinguishable near the volcano. The initial sampling was carried out in the fall of 1977 with the assistance of D. R. Mullineaux. Sampling was completed in the summer of 1978. For this study, samples of layers We, Wn, Ye, and Yn were taken from numerous localities within each layer (Fig. 3 and Table 1). At a few localities samples were collected at different horizons within a given tephra layer. Only crystal-rich pumice clasts were sampled because these are clearly magmatic products that record preeruptive conditions in the chamber(s) and because the effects of crystal winnowing and/or contamination are minimized.

Analytical Methods

Heavy-mineral separates were examined by standard petrographic techniques to determine the mineral assemblage and modal abundances. Chemical analyses of the heavy minerals, as well as plagioclases, were determined by use of an ETEC electron microprobe with both natural and synthetic mineral standards. Either 15 or 20 kV emission current, 0.025 current, 20-sec counting periods, and 1- μ m electron beam-diameter were routine operating conditions. Rim and core analyses were performed to check for zoning and overall homogeneity. Raw data were corrected for dead time of detectors, instrument current drift, spectrometer background and matrix (ZAF) effects using the computer program PROBEG. Accuracy is estimated to be no worse than 2% for major elements and 5-10% for minor elements, on the basis of replicate microprobe analyses compared with wet chemical analyses of the standards.

RESULTS

Modal Abundances

Results of modal analyses of the heavy-mineral separates are listed in Table 2. Figure 4 illustrates hypersthene-hornblende-Fe-Ti oxide proportions for units We and Wn, and cummingtonite-hornblende-Fe-Ti oxide proportions for units Ye and Yn. Unit Y shows a large range in modes, mostly due to variations in contents of the oxide minerals, whereas W is more uniform. Layers examined within W are indistinguishable, and layers Ye and Yn are only marginally distinct.

Electron Microprobe Studies

Plagioclases in layers of W and Y exhibit a large compositional range within each layer (Fig. 5), the total range being An₃₅₋₈₅. Oscillatory, normal, and reverse zoning were observed. Neither core compositions nor zoning patterns were distinctive for sets W and Y and layers within W and Y. No correlation of composition with stratigraphic position was evident.

Table 3 presents representative analyses of orthopyroxene from samples of We and Wn (orthopyroxene was not observed in set Y) and compositional ranges for the two layers are illustrated in Figure 6. The range of composition is not large (~En₅₀₋₆₀) and there is no relation between stratigraphic position and composition of the hypersthene. Rims of some of the analyzed grains are slightly more magnesian (En₅₇) relative to the cores (En₅₄). In comparison, orthopyroxene in Glacier Peak tephra has a higher enstatite component (En₆₅₋₆₆ (Westgate and Evans, 1978); Fig. 6).

Cummingtonite is present in layers Ye and Yn as discrete prismatic phenocrysts and as cores of grains rimmed by hornblende. Discrete grains and cores of zoned grains were analyzed, and Table 4 lists representative analyses; compositional ranges are shown in Figure 6. Most grains analyzed are compositionally homogeneous, but a few of the discrete grains within Ye were found to have more magnesian rims (see analyses for sample SH-17, Table 4). The range in composition is rather limited (~10 mole%) and is overlapping between the two layers. No correlation was observed between composition and mode of textural occurrence.

Hornblende occurs in both sets W and Y and was analyzed in each layer examined (see Table 5 for representative analyses). Within a given layer, the amphibole exhibits a large range in composition. In fact, the entire range of composition was observed not only in samples collected from different localities of a given unit, but even within single pumice clasts. Ca-Fe-Mg diagrams show the overlapping but small range (~10 mole%) of these components in different tephra units (Fig. 6). On the basis of SiO₂ and Al₂O₃ contents in hornblendes one can distinguish set W from set Y, but only marginally Glacier Peak tephra (Fig. 7). Layers examined within a given tephra set cannot be distinguished from each other (e.g., We and Wn). Hornblende in MSH layer Y is indistinguishable from hornblende in Glacier Peak tephra, but the presence of cummingtonite in Y and its absence in Glacier Peak tephra aid in proper identification.

Iron-titanium oxides occur as micro-phenocrysts and as inclusions within silicate phenocrysts in all the tephra studied. The oxides were not observed to exhibit exsolution or oxidation characteristics. Analyses were recalculated according to the method of Carmichael (1967) to estimate stoichiometric contents of FeO and Fe₂O₃. The ilvospinel content of the titanomagnetite and hematite content of the ilmenite solid solutions were calculated from these adjusted analyses. The results are listed in Table 6 (see the Appendix for a complete listing of analyses for individual MSH samples). The analyses of Fe-Ti oxides in Ye and Yn of this study are similar to reported analyses of these minerals in distal MSH "Y" (Table 7). The exceptions are small differences in MgO content in ilmenite and calculated FeO and Fe₂O₃ (assuming stoichiometry), which could be due to use of different standards and techniques.

In contrast to the silicates, titanomagnetite and ilmenite grains in MSH tephra layers are compositionally homogeneous whether they occur as phenocrysts or as inclusions within other minerals. Furthermore, these phases are compositionally distinctive between different tephra sets. Differences in MgO, Al₂O₃, and TiO₂ contents in titanomagnetite allow distinction of tephra sets Y and W from each other and from other Pacific Northwest tephra, but not layer Ye from Yn nor layer We from Wn (Figs. 8a and b). Likewise, TiO₂ content in ilmenite is a useful discriminator.

Temperature-Oxygen Fugacity Estimates

Differences in preemptive physical conditions may have existed for different tephra. The compositional data were utilized in estimating temperatures and oxygen fugacities of equilibration of coexisting oxides using the method of Buddington and Lindsley (1964) to ascertain whether the values would allow distinction of tephra. The results for average compositions of Fe-Ti oxides are included in Table 6.

Samples of tephra set W collected from different localities give a small spread of Fe-Ti oxide composition which in turn implies a temperature range of 810-833°C and a $-\log f_{\text{O}_2}$ range of 12.7-13.3 for layer Wn and 821-848°C and 12.9-13.0 for layer We. Considering the precision of the geothermometer and geobarometer, there is no correlation between T, f_{O_2} , and stratigraphic level within W tephra. Data for tephra set Y also yield a small range of temperature and $-\log f_{\text{O}_2}$ of 860-884°C and 10.6-11.1, respectively.

Difficulties arise in comparing the temperatures and oxygen fugacities calculated in this study to temperatures and oxygen fugacities calculated for other Pacific Northwest tephra using published Fe-Ti oxides analyses, due to differences in FeO and Fe₂O₃ calculations as mentioned previously. Thus, published FeO and Fe₂O₃ contents were recalculated to total FeO (where total FeO = (0.9 x Fe₂O₃) + FeO). FeO and Fe₂O₃ contents were then determined in the same manner as for analyses of this study. The resulting values and calculated ilmenite and hematite contents (mole percentage, determined in the same manner as for analyses of this study) are included in Table 7. Temperatures and oxygen fugacities calculated using the data are listed in Table 7 and plotted in Figure 9 along with average temperatures and oxygen fugacities calculated for MSH tephra. Published analyses of MSH "Y" oxides give $T-f_{O_2}$ values very close to the range found using analyses of this study. With the exception of Glacier Peak and Mazama tephra, it appears possible to distinguish several Pacific Northwest tephra, including sets from Mount St. Helens, on this basis. It was not possible to distinguish between individual layers within MSH sets W and Y.

DISCUSSION AND CONCLUSIONS

Near-source samples of Mount St. Helens tephra set W are readily distinguished from set Y on the basis of their mineralogic characteristics, but within these sets, individual layers cannot be distinguished from each other on the basis of mineralogy and phase chemistry. No single characteristic is in itself the key to distinguishing among individual Pacific Northwest tephra. Within the constraints of field relations and available ¹⁴C ages, the most distinctive parameters are phenocryst assemblage, ferromagnesian phenocryst composition, composition of iron-titanium oxides, and calculated temperature and oxygen fugacity estimates based on compositions of coexisting magnetite and ilmenite. Table 8 summarizes some of the characteristics useful in tephra identification.

Possibly the most effective technique in tephra distinction is identification of ferromagnesian silicate assemblages augmented, where necessary, by analyses of iron-titanium oxides. Glass-encased magnetite and ilmenite grains are commonly found in downwind tephra deposits and are considerably easier to analyze than glass shards. Also, we have found the Fe-Ti oxide minerals to be remarkably homogeneous both within and between samples of a given unit over a wide geographic area. Analysis of glasses may be complicated by volatilization of alkalis (especially sodium) and very fine-grained, vesicular shards. These problems and possible post-eruption alteration may hinder correct identification of tephra on the basis of glass chemistry, especially where only small chemical differences exist for tephra erupted from the same volcanic center.

Distinction of tephra on the basis of iron-titanium oxide composition is dependent on precise determination of the minor elements. Differences in analyses made in different laboratories may result from the use of different standards, operating conditions, and data reduction methods. Thus, in order to apply the techniques used here effectively and to facilitate interlaboratory comparison, it is advised that samples of proximal tephra known to be from a specific unit (e.g., W, Y) be analyzed as "internal standards".

Work in progress includes characterization of other proximal and distal Mount St. Helens tephra (sets T, S, P, J, M, K, C), including mineral compositions, glass chemistry, and geothermometry and geobarometry.

APPENDIX

The following is a list of iron-titanium oxide compositions for samples of tephra sets W and Y. The mean for n (number of) analyses is given, with the standard deviation in parentheses.

TITANOMAONNETITE COMPOSITIONS

Unit: Sample:	Wn DS-11	Wn DS-40	Wn DS-45	Wn DS-50	Wn DS-51
SiO ₂	0.22 (0.03)	0.23 (0.08)	0.17 (0.07)	0.23 (0.08)	0.24 (0.04)
TiO ₂	9.63 (0.16)	9.66 (0.17)	9.66 (0.13)	9.82 (0.77)	9.78 (0.12)
Al ₂ O ₃	2.39 (0.09)	2.48 (0.03)	2.42 (0.04)	2.43 (0.09)	2.31 (0.14)
FeO	38.1 (0.10)	37.7 (0.03)	37.9 (0.20)	38.3 (0.06)	38.2 (0.15)
Fe ₂ O ₃	46.5 (0.12)	46.5 (0.04)	46.1 (0.24)	45.6 (0.10)	45.7 (0.18)
MnO	0.32 (0.03)	0.36 (0.03)	0.33 (0.02)	0.38 (0.02)	0.36 (0.03)
MgO	1.16 (0.05)	1.42 (0.48)	1.13 (0.12)	1.01 (0.17)	1.08 (0.11)
Sum	98.32	98.35	97.71	97.79	98.87
n	5	5	5	4	5

Unit: Sample:	Wn DS-57	Wn DS-60	We DS-19	We DS-30	We DS-47
SiO ₂	0.26 (0.08)	0.26 (0.03)	0.26 (0.04)	0.25 (0.09)	0.25 (0.04)
TiO ₂	9.63 (0.07)	9.78 (0.24)	9.54 (0.18)	9.78 (0.15)	9.66 (0.11)
Al ₂ O ₃	2.36 (0.19)	2.41 (0.07)	2.09 (0.09)	2.09 (0.12)	2.12 (0.09)
FeO	38.0 (0.17)	39.5 (0.07)	37.7 (0.03)	37.6 (0.16)	37.6 (0.09)
Fe ₂ O ₃	45.7 (0.21)	43.7 (0.08)	46.9 (0.04)	46.3 (0.19)	46.3 (0.12)
MnO	0.34 (0.02)	0.38 (0.01)	0.34 (0.04)	0.38 (0.01)	0.33 (0.04)
MgO	1.01 (0.10)	1.01 (0.19)	1.30 (0.07)	1.40 (0.16)	1.31 (0.03)
Sum	97.31	97.04	98.13	97.80	97.59
n	5	4	4	5	5

Unit: Sample:	We DS-55	We SH-18	Yn DS-58	Yn DS-13	Yn DS-54
SiO ₂	0.21 (0.03)	0.20	0.11 (0.04)	0.12 (0.03)	0.15 (0.04)
TiO ₂	9.66 (0.24)	9.78	5.62 (0.31)	5.68 (0.25)	5.48 (0.18)
Al ₂ O ₃	2.07 (0.40)	1.84	2.55 (0.08)	2.68 (0.10)	2.76 (0.06)
FeO	37.8 (0.13)	38.4	34.1 (0.24)	34.4 (0.10)	33.8 (0.11)
Fe ₂ O ₃	46.2 (0.16)	47.2	54.3 (0.39)	54.2 (0.15)	54.2 (0.18)
MnO	0.37 (0.03)	0.35	0.30 (0.02)	0.33 (0.02)	0.36 (0.04)
MgO	1.10 (0.12)	1.02	1.24 (0.24)	1.17 (0.13)	1.33 (0.14)
Sum	97.41	98.79	98.22	98.58	98.08
n	5	1	5	5	5

Unit: Sample:	Yn DS-44	Ye DS-56	Ye DS-48	Ye DS-31	Ye DS-36
SiO ₂	0.11 (0.02)	0.11 (0.02)	0.13 (0.05)	0.12 (0.03)	0.14 (0.04)
TiO ₂	5.52 (0.18)	5.84 (0.13)	5.80 (0.26)	5.74 (0.11)	5.81 (0.32)
Al ₂ O ₃	2.83 (0.12)	2.67 (0.15)	2.80 (0.10)	2.72 (0.11)	2.71 (0.11)
FeO	34.3 (0.27)	34.0 (0.16)	33.9 (0.26)	34.3 (0.13)	34.5 (0.27)
Fe ₂ O ₃	55.2 (0.44)	53.6 (0.25)	53.5 (0.41)	54.1 (0.21)	54.0 (0.42)
MnO	0.36 (0.03)	0.33 (0.02)	0.35 (0.02)	0.36 (0.01)	0.36 (0.03)
MgO	1.33 (0.12)	1.38 (0.18)	1.41 (0.06)	1.24 (0.14)	1.20 (0.04)
Sum	99.65	97.93	97.89	98.58	98.72
n	4	4	5	5	5

Unit: Sample:	Ye DS-52	Ye DS-20	Ye DS-38
SiO ₂	0.11 (0.04)	0.10 (0.03)	0.12 (0.04)
TiO ₂	5.80 (0.40)	5.85 (0.19)	5.69 (0.11)
Al ₂ O ₃	2.95 (0.37)	2.74 (0.12)	2.72 (0.14)
FeO	34.4 (0.24)	34.6 (0.17)	34.3 (0.19)
Fe ₂ O ₃	53.9 (0.38)	54.5 (0.26)	54.1 (0.29)
MnO	0.36 (0.02)	0.36 (0.02)	0.36 (0.02)
MgO	1.26 (0.05)	1.24 (0.11)	1.20 (0.12)
Sum	98.78	99.39	98.49
n	5	5	1

ILMENITE COMPOSITIONS

Unit: Sample:	Wn DS-11	Wn DS-40	Wn DS-45	Wn DS-51
SiO ₂	0.24 (0.06)	0.23 (0.06)	0.18 (0.03)	0.14 (0.06)
TiO ₂	44.6 (0.60)	44.7 (0.65)	45.2 (0.31)	45.1 (0.24)
Al ₂ O ₃	0.02 (0.02)	tr ^a	tr	tr
FeO	37.5 (0.12)	37.1 (0.16)	37.7 (0.26)	37.2 (0.24)
Fe ₂ O ₃	13.3 (0.04)	13.9 (0.06)	13.6 (0.09)	14.0 (0.09)
MnO	0.48 (0.01)	0.48 (0.03)	0.51 (0.05)	0.48 (0.04)
MgO	1.39 (0.24)	1.62 (0.17)	1.49 (0.17)	1.71 (0.18)
Sum	97.53	98.03	98.68	98.63
n	5	5	5	5
Unit: Sample:	Wn DS-60	We DS-19	We DS-30	We DS-47
SiO ₂	0.23 (0.09)	0.18 (0.09)	0.12 (0.04)	0.27 (0.08)
TiO ₂	44.8 (0.33)	44.8 (0.38)	44.8 (1.03)	45.2 (0.43)
Al ₂ O ₃	0.08 (0.06)	tr	tr	tr
FeO	37.1 (0.27)	36.9 (0.17)	36.5 (0.17)	37.1 (0.20)
Fe ₂ O ₃	13.7 (0.10)	13.8 (0.06)	13.9 (0.07)	13.9 (0.08)
MnO	0.48 (0.02)	0.50 (0.03)	0.50 (0.02)	0.51 (0.03)
MgO	1.70 (0.20)	1.78 (0.15)	1.64 (0.09)	1.88 (0.17)
Sum	98.09	97.96	97.46	98.86
n	4	5	4	5
Unit: Sample:	We DS-55	We SH-18	Yn ^b	Ye ^b
SiO ₂	0.17 (0.10)	0.19	0.17 (0.06)	0.21 (0.11)
TiO ₂	44.7 (0.26)	44.7	39.9 (0.7)	39.3 (1.2)
Al ₂ O ₃	tr	0.06	0.25 (0.11)	0.33 (0.14)
FeO	36.4 (0.21)	36.3	32.1 (0.4)	31.4 (0.5)
Fe ₂ O ₃	14.4 (0.03)	15.9	24.2 (0.3)	24.9 (0.4)
MnO	0.52 (0.03)	0.53	0.48 (0.03)	0.46 (0.04)
MgO	1.94 (0.07)	2.00	1.98 (0.25)	2.09 (0.28)
Sum	98.13	99.68	99.08	98.69
n	4	1	6	8

^a tr = trace amount

^b Represents an average of n analyses performed on several random samples of the tephra layer.

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REFERENCES

- Brewster, G. R., and Barnett, R. L. (1979). Magnetites from a new unidentified tephra source, Banff National Park, Alberta. *Canadian Journal of Earth Sciences* **16**, 1294-1297.
- Buddington, A. F., and Lindsley, D. H. (1964). Iron-titanium oxide minerals and synthetic equivalents. *Journal of Petrology* **5**, 310-357.
- Carmichael, I. S. E. (1967). The iron-titanium oxides of salic volcanic rocks and their associated ferromagnesian silicates. *Contributions to Mineralogy and Petrology* **14**, 36-64.
- Crandell, D. R. (1971). "Postglacial Lahars from Mount Rainier Volcano, Washington." U.S. Geological Survey Professional Paper 677.
- Crandell, D. R., and Mullineaux, D. R. (1973). "Pine Creek Volcanic Assemblage at Mount St. Helens, Washington." U.S. Geological Survey Bulletin 1383-A.
- Crandell, D. R., and Mullineaux, D. R. (1978). "Potential Hazards from Future Eruptions of Mount St. Helens Volcano, Washington." U.S. Geological Survey Bulletin 1383-C.
- Crandell, D. R., Mullineaux, D. R., and Rubin, M. (1975). Mount St. Helens: Recent and future behavior. *Science* **187**, 438-441.
- Dudas, M. J., Harward, M. D., and Schmitt, R. A. (1973). Identification of dacitic tephra by activation analysis of their primary mineral phenocrysts. *Quaternary Research* **3**, 307-315.
- Fryxell, R. (1965). Mazama and Glacier Peak volcanic ash layers: Relative ages. *Science* **147**, 1288-1290.
- Hoblitt, R. P., Crandell, D. R., and Mullineaux, D. R. (1980). Mount St. Helens eruptive behavior during the past 1,500 yr. *Geology* **8**, 555-559.
- Hopson, C. A. (1971). Eruptive sequence at Mount St. Helens, Washington. *Geological Society of America Abstracts with Programs* **3**, 138.

Hyde, J. H. (1973). "Late Quaternary Volcanic Stratigraphy, South Flank of Mount St. Helens, Washington." Ph.D. dissertation, University of Washington, Seattle.

Hyde, J. H. (1975). "Upper Pleistocene pyroclastic flow deposits and lahars south of Mount St. Helens volcano. Washington." U.S. Geological Survey Bulletin 1383-B.

Izett, G. A., Wilcox, R. E., Powers, H. A., and Desborough, G. A. (1970). The Bishop ash bed, a Pleistocene marker bed in the western United States. *Quaternary Research* **1**, 121-132.

Lipman, P. W. (1965). "Chemical Composition of Glassy and Crystalline Volcanic Rocks." U.S. Geological Survey Bulletin 1201-D, D1-D24.

Mathewes, R. W., and Westgate, J. A. (1980). Bridge River tephra: Revised distribution and significance for detecting old carbon errors in radiocarbon dates of limnic sediments in southern British Columbia. *Canadian Journal of Earth Sciences* **17**, 1454-1461.

Mullineaux, D. R., Hyde, J. H., and Rubin, M. (1975). Widespread late glacial and postglacial tephra deposits from Mount St. Helens volcano, Washington. *U.S. Geological Survey Journal of Research* **3**, 329-335.

Mullineaux, D. R., Wilcox, R. E., Ebaugh, W. F., Fryxell, R., and Rubin, M. (1978). Age of the last major scabland flood of the Columbia Plateau in Eastern Washington. *Quaternary Research* **10**, 171-180.

Okazaki, R., Smith, H. W., Gilkeson, R. A., and Franklin, J. (1972). Correlation of West Blacktail ash with pyroclastic layer T from the 1800 A.D. eruption of Mount St. Helens. *Northwest Science* **46**, 77-89.

Porter, S. C. (1978). Glacier Peak tephra in the North Cascade Range, Washington: Stratigraphy, distribution, and relationship to late-glacial events. *Quaternary Research* **10**, 30-41.

Powers, H. A., and Wilcox, R. E. (1964). Volcanic ash from Mount Mazama (Crater Lake) and from Glacier Peak. *Science* **144**, 1334-1336.

Randle, K., Goles, G. G., and Kittleman, L. R. (1971). Geochemical and petrological characterization of ash samples from Cascade Range volcanoes. *Quaternary Research* **1**, 261-282.

Sarna-Wojcicki, A. M., Meyer, C. E., Russell, P. C., and Woodward, M. (1980). Chemical correlation and fission-track age dating of late Cenozoic tephra units in northern California, Oregon, Washington, and western Nevada. A presentation at the Cascade Conference, Feb. 19-20, 1980, Menlo Park, Calif.

Smith, D. G. W., and Westgate, J. A. (1969). Electron probe technique for characterizing pyroclastic deposits. *Earth and Planetary Science Letters* **5**, 313-319.

Smith, H. W., Okazaki, R., and Knowles, C. R. (1977). Electron microprobe analysis of glass shards from tephra assigned to set W, Mount St. Helens, Washington. *Quaternary Research* **7**, 202-217.

Westgate, J. A. (1977). Identification and significance of late Holocene tephra from Otter Creek, southern British Columbia and localities in west-central Alberta. *Canadian Journal of Earth Sciences* **14**, 2593-2600.

Westgate, J. A., and Fulton, R. J. (1975). Tephro-stratigraphy of Olympia interglacial sediments in south-central British Columbia, Canada. *Canadian Journal of Earth Sciences* **12**, 489-502.

Westgate, J. A., and Evans, M. E. (1978). Compositional variability of Glacier Peak tephra and its stratigraphic significance. *Canadian Journal of Earth Sciences* **15**, 1554-1567.

Westgate, J. A., Christiansen, E. A., and Boellstorff, J. D. (1977). Wascana Creek ash (Middle Pleistocene) in southern Saskatchewan: Characterization, source, fission track age, paleomagnetism and stratigraphic significance. *Canadian Journal of Earth Sciences* **14**, 357-374.

Figure 1:

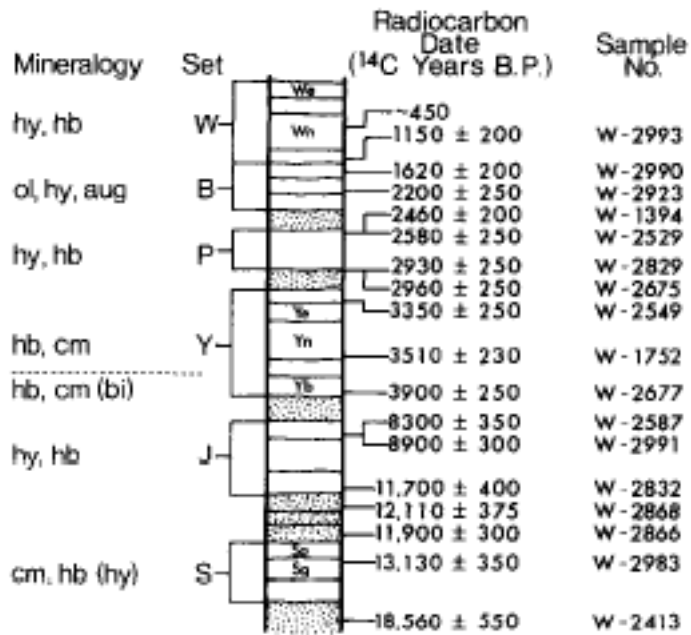


FIG. 1. Diagrammatic section showing stratigraphic units exposed on the northern and eastern slopes of Mount St. Helens (previous to the 1980 eruptions). Layer Wn is dated at ~450 yr B.P. by tree-ring counts (Crandell, 1971). Data taken from Crandell and Mullineaux (1973), Mullineaux *et al.* (1975, 1978), and Hyde (1975). Key: ol = olivine, hy = hypersthene, cm = cummingtonite, hb = hornblende, bi = biotite, aug = augite, () = in minor abundance.

Figure 2:

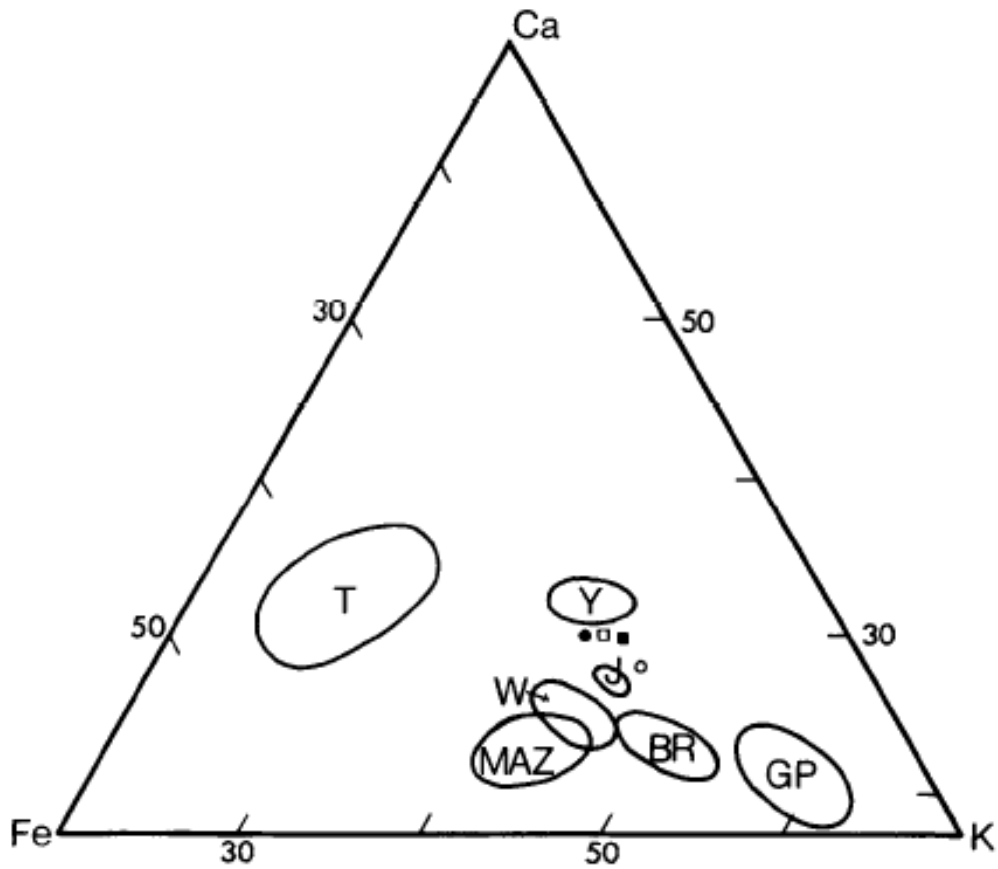


FIG. 2. Ca-Fe-K ratios in glasses from various Cascade tephras: Glacier Peak (GP), the climactic eruption of Mount Mazama (MAZ), Bridge River tephra (BR), and Mount St. Helens tephra units T, W, Y, J, So (open circle), Sg (solid circle), Mm (open square), Mp (solid square). Data from Smith and Westgate (1969), Smith *et al.* (1977), Mullinaux *et al.* (1978), and Westgate and Evans (1978).

Figure 3:

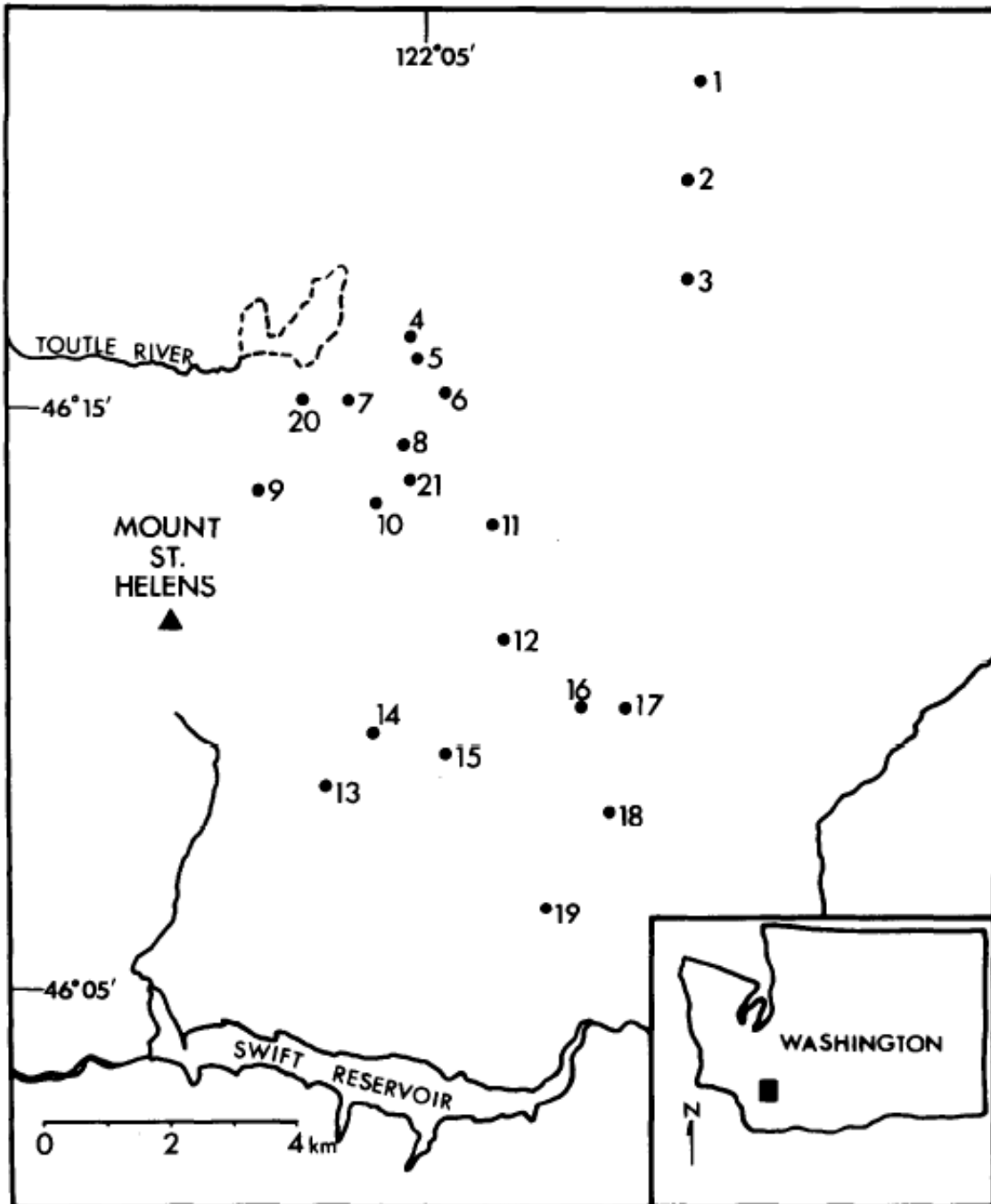


FIG. 3. Sample location map (Table 1). Dashed line indicates Spirit Lake's boundary before the May 18, 1980, eruption.

Table 1:

TABLE 1. SAMPLE LOCATION DESCRIPTIONS

Location No.	Description	Sample No.	Unit
1	NW $\frac{1}{4}$ SW $\frac{1}{4}$, Sec. 17, T. 10 N., R. 7 E.; roadcut, west side of Road 125	DS-53	Wn
		DS-54	Yn
2	NE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 30, T. 10 N., R. 7 E.; roadcut, east side of Road 125	DS-51	Wn
		DS-52	Yn
3	NE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 6, T. 9 N., R. 7 E.; roadcut, east side of Road 125	DS-50	Wn
4	NE $\frac{1}{4}$ SW $\frac{1}{4}$ SW $\frac{1}{4}$, Sec. 8, T. 9 N., R. 6 E.; roadcut, east side of Road 100	DS-45	Wn
5	SW $\frac{1}{4}$ NE $\frac{1}{4}$ NW $\frac{1}{4}$, Sec. 17, T. 9 N., R. 6 E.; roadcut, north side of Road N92	DS-60	Wn
6	SE $\frac{1}{4}$ SE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 17, T. 9 N., R. 6 E.; roadcut, Road N930	DS-57	Wn
		DS-59	Ye
		DS-58	Yn
7	NW $\frac{1}{4}$ NE $\frac{1}{4}$ NE $\frac{1}{4}$, Sec. 24, T. 9 N., R. 5 E.; roadcut, north side of Road 100	SH-4	Wn
		SH-5	Wn
		SH-6	Wn
8	SE $\frac{1}{4}$ SE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 19, T. 9 N., R. 6 E.; roadcut, west side of Road N92	DS-11	Wn
		DS-13	Yn
9	SE $\frac{1}{4}$ SE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 27, T. 9 N., R. 5 E.; timberline viewpoint	DS-40	Wn
10	SE $\frac{1}{4}$ NE $\frac{1}{4}$ NW $\frac{1}{4}$, Sec. 31, T. 9 N., R. 6 E.	DS-15	Yn
11	SE $\frac{1}{4}$ NE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 33, T. 9 N., R. 6 E.; roadcut, east side of Road N930	DS-55	We
		DS-56	Ye
12	NE $\frac{1}{4}$ SE $\frac{1}{4}$ NW $\frac{1}{4}$, Sec. 15, T. 8 N., R. 6 E.; roadcut, east side of Road N92	DS-19	We
		DS-20	Ye
13	SE $\frac{1}{4}$ NE $\frac{1}{4}$ NE $\frac{1}{4}$, Sec. 36, T. 8 N., R. 5 E.; roadcut, north side of Road N83	DS-30	We
		DS-31	Ye
14	NE $\frac{1}{4}$ NW $\frac{1}{4}$ NE $\frac{1}{4}$, Sec. 30, T. 8 N., R. 6 E.; Streamcut exposure on east side of Pine Creek	DS-34	We
15	NE $\frac{1}{4}$ NE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 29, T. 8 N., R. 6 E.; roadcut, east side of Road N833	DS-36	Ye
16	SW $\frac{1}{4}$ SE $\frac{1}{4}$ NE $\frac{1}{4}$, Sec. 23, T. 8 N., R. 6 E.; roadcut, east side of Road N917	DS-26	Ye
17	SW $\frac{1}{4}$ NE $\frac{1}{4}$, Sec. 25, T. 8 N., R. 6 E.; roadcut, east side of Road 125	DS-48	Ye
18	SW $\frac{1}{4}$ NW $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 36, T. 8 N., R. 6 E.; roadcut, east side of Road 125	DS-47	We
		DS-46	Ye
19	NW $\frac{1}{4}$ SW $\frac{1}{4}$ NW $\frac{1}{4}$, Sec. 11, T. 7 N., R. 6 E.; roadcut, north side of N833	DS-38	Ye
20	SW $\frac{1}{4}$ NW $\frac{1}{4}$ NW $\frac{1}{4}$, Sec. 24, T. 9 N., R. 5 E.; roadcut, west side of Road 100	DS-44	Yn
21	NW $\frac{1}{4}$ SE $\frac{1}{4}$ SE $\frac{1}{4}$, Sec. 13, T. 9 N., R. 5 E.; roadcut, west side of Road 100	SH-17	Ye

Table 2:

TABLE 2. MODAL PROPORTIONS FOR HEAVY MINERAL SEPARATES^a

Unit	Sample	Hypersthene	Hornblende	Cummingtonite	Opagues
We	DS-30	71 ± 5	15 ± 2	—	14 ± 2
We	DS-55	75 ± 5	8 ± 2	—	17 ± 2
We	DS-47	82 ± 5	5 ± 1	—	13 ± 2
We	DS-19	72 ± 5	8 ± 2	—	20 ± 3
We	DS-47	82 ± 5	2 ± 1	—	9 ± 2
Wn	DS-50	69 ± 5	14 ± 2	—	17 ± 2
Wn	DS-51	72 ± 5	9 ± 2	—	19 ± 2
Wn	DS-60	61 ± 4	19 ± 3	—	20 ± 3
Wn	DS-40	72 ± 4	12 ± 2	—	16 ± 2
Wn	DS-45	73 ± 4	11 ± 2	—	16 ± 2
Wn	DS-57	66 ± 5	15 ± 2	—	19 ± 2
Wn	SH-4	75 ± 5	15 ± 2	—	10 ± 2
Wn	SH-5	72 ± 5	11 ± 2	—	17 ± 2
Wn	SH-6	78 ± 5	8 ± 2	—	14 ± 2
Wn	DS-11	75 ± 5	12 ± 1	—	13 ± 1
Ye	SH 17	—	18 ± 2	37 ± 3	45 ± 4
Ye	DS-56	—	23 ± 3	35 ± 4	42 ± 4
Ye	DS-48	—	22 ± 3	45 ± 4	33 ± 3
Ye	DS-46	—	29 ± 3	55 ± 4	16 ± 2
Ye	DS-59	—	33 ± 3	33 ± 3	34 ± 3
Ye	DS-38	—	29 ± 3	58 ± 4	13 ± 2
Ye	DS-20	—	18 ± 2	33 ± 3	49 ± 4
Ye	DS-31	—	23 ± 3	29 ± 3	48 ± 4
Ye	DS-36	—	19 ± 2	34 ± 3	47 ± 4
Yn	DS-13	—	27 ± 3	26 ± 3	47 ± 4
Yn	DS-54	—	37 ± 3	18 ± 2	45 ± 4
Yn	DS-58	—	33 ± 3	36 ± 3	32 ± 3
Yn	DS-44	—	18 ± 2	16 ± 2	66 ± 5

^a Specific gravity >2.9672, the fraction of 100–150 mesh size was analyzed (2.75–3.25 ϕ units). More than 300 grains were counted in each sample. Uncertainties given are one standard deviation estimated from counting statistics.

Figure 4:

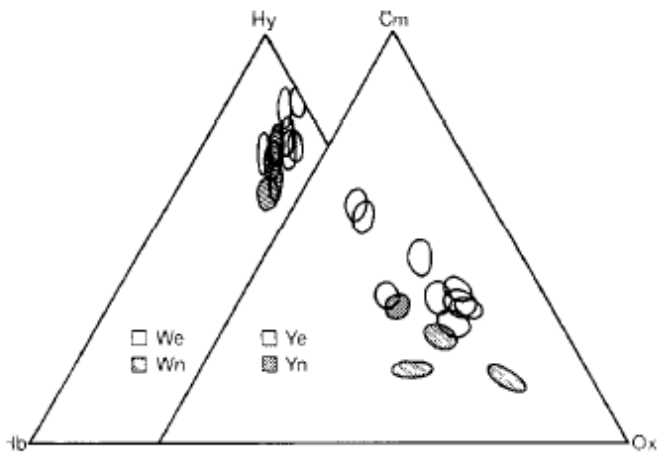


FIG. 4. Hypersthene-hornblende-Fe-Ti oxide and cummingtonite-hornblende-Fe-Ti oxide abundances for tephra sets W and Y, respectively. For each sample, the modal abundance and the uncertainty due to counting statistics are plotted.

Figure 5:

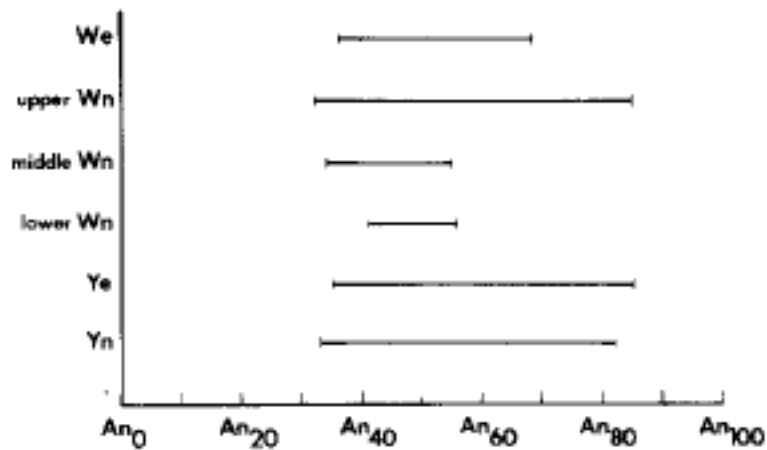


FIG. 5. Anorthite content in plagioclases in units Ye, Yn, We, and Wn.

Table 3:

TABLE 3. REPRESENTATIVE ANALYSES OF ORTHOPYROXENE					
Sample:	We DS-55	We DS-45	Wn SH-6	Wn SH-5	Glacier peak ^a [mean (σ)] ^b
SiO ₂	50.9	50.8	51.2	51.9	54.23 (0.36)
TiO ₂	0.09	0.09	0.11	0.20	0.14 (0.04)
Al ₂ O ₃	1.36	0.66	0.72	1.38	0.85 (0.32)
∑ FeO	25.5	27.6	27.9	25.3	20.30 (0.36)
MnO	0.64	0.75	0.80	0.66	1.05 (0.10)
MgO	21.4	18.2	18.6	19.9	22.73 (0.33)
CaO	0.78	0.86	0.85	1.13	0.86 (0.10)
Na ₂ O	n.d. ^c	n.d.	n.d.	n.d.	0.01 (0.01)
K ₂ O	n.d.	n.d.	n.d.	n.d.	0.04 (0.01)
Sum	100.67	98.96	100.18	100.47	100.21
Number of cations on the basis of six oxygens					
Si	1.923	1.971	1.964	1.958	1.998
Ti	0.003	0.003	0.003	0.006	0.004
Al ^{IV}	0.060	0.029	0.032	0.042	0.002
Al ^{VI}	0.000	0.001	0.000	0.019	0.035
Fe	0.805	0.897	0.896	0.798	0.625
Mn	0.020	0.024	0.026	0.021	0.033
Mg	0.201	1.050	1.060	1.116	1.248
Ca	0.032	0.036	0.035	0.046	0.034
Na	—	—	—	—	0.001
K	—	—	—	—	0.002
Mole percentage					
Mg	58.9	53.0	53.2	56.9	65.4
Fe	39.5	45.2	45.0	40.7	32.8
Ca	1.6	1.8	1.8	2.4	1.8

^a Taken from Westgate and Evans (1978).

^b σ = standard deviation.

^c Not detected.

Figure 6:

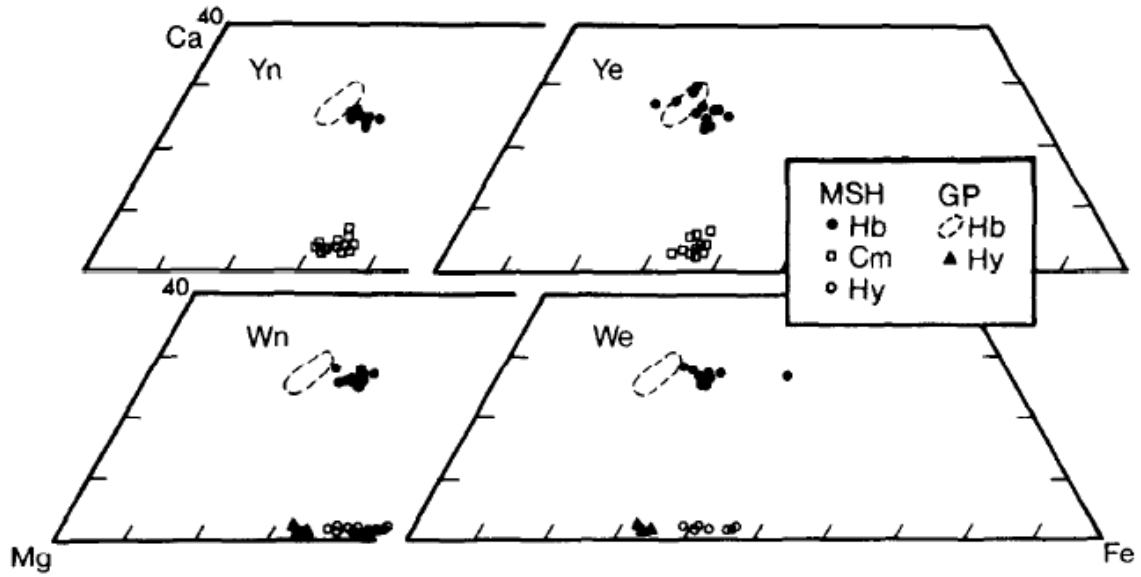


FIG. 6. Compositional ranges of ferromagnesian silicates in MSH Yn, Ye, Wn, and We, and Glacier Peak tephtras. Abbreviations as in Figure 1.

Table 4:

TABLE 4. REPRESENTATIVE ANALYSES
OF CUMMINGTONITE

Sample:	Yn	Yn	Ye ^a	Ye ^b
	DS-13	DS-13	SH-17	SH-17
SiO ₂	54.7	52.3	54.5	53.8
TiO ₂	0.24	0.33	0.21	0.18
Al ₂ O ₃	2.59	3.64	2.13	3.13
∑FeO	19.8	17.4	20.5	19.0
MnO	0.68	0.56	0.74	0.67
MgO	18.5	20.2	17.7	19.4
CaO	1.49	2.23	1.94	1.48
Na ₂ O	0.29	0.65	0.17	0.32
K ₂ O	n.d.	n.d.	n.d.	n.d.
Sum	98.29	97.31	97.89	97.98
Number of cations on basis of 23 oxygens				
Si	7.818	7.532	7.864	7.701
Ti	0.026	0.036	0.023	0.019
Al	0.436	0.618	0.362	0.528
Fe	2.367	2.096	2.462	2.275
Mn	0.082	0.068	0.090	0.081
Mg	3.940	4.355	3.806	4.139
Ca	0.228	0.344	0.300	0.227
Na	0.080	0.181	0.048	0.089
Mole percentage				
Mg	60.3	64.1	57.9	62.3
Fe	36.2	30.8	37.9	34.3
Ca	3.5	5.1	4.6	3.4

^a Core of discrete cummingtonite phenocryst.^b Rim of grain of previous analysis.

Table 5:

TABLE 5. REPRESENTATIVE ANALYSES OF HORNBLENDE												
	We	We	We	Wn	Wn	Wn	Ye	Ye	Ye	Yn	Yn	Yn
SiO ₂	40.6	42.5	45.2	41.2	42.5	44.8	42.7	44.9	47.1	49.3	44.0	48.2
TiO ₂	2.76	3.00	1.62	2.45	2.81	1.86	2.03	1.20	0.85	0.67	1.81	0.87
Al ₂ O ₃	14.5	10.3	7.97	13.8	10.4	8.82	13.8	10.2	9.27	7.68	12.2	9.46
Σ FeO	12.6	16.1	15.9	11.9	15.7	15.2	10.5	14.5	14.2	15.7	12.2	13.5
MnO	0.11	0.22	0.11	0.16	0.24	0.25	0.11	0.34	0.40	0.39	0.22	0.26
MgO	12.8	11.6	12.8	14.1	12.2	13.2	14.4	14.8	14.4	14.1	14.0	14.7
CaO	10.8	10.6	10.8	10.9	10.8	10.6	10.7	9.25	9.31	8.57	10.3	9.70
Na ₂ O	2.22	2.18	2.22	2.44	2.13	1.82	2.05	1.72	1.30	1.17	1.75	1.20
K ₂ O	0.26	0.30	0.26	0.25	0.30	0.25	0.27	0.19	0.15	0.20	0.23	0.15
Sum	96.65	96.80	96.88	97.20	97.08	96.80	96.56	97.10	96.98	97.78	96.71	98.04
Number of cations on basis of 23 oxygens												
Si	6.040	6.417	6.810	6.082	6.390	6.696	6.263	6.628	6.905	7.172	6.475	6.950
Al ^{IV}	1.960	1.583	1.190	1.918	1.610	1.304	1.737	1.372	1.095	0.828	1.525	1.050
Al ^{VI}	0.582	0.250	0.226	0.484	0.234	0.250	0.649	0.403	0.507	0.489	0.591	0.558
Ti	0.309	0.341	0.184	0.272	0.318	0.209	0.224	0.133	0.094	0.073	0.200	0.094
Fe	1.568	2.033	2.003	1.496	1.974	1.900	1.288	1.790	1.741	1.910	1.501	1.628
Mn	0.014	0.028	0.031	0.020	0.031	0.032	0.014	0.043	0.050	0.048	0.027	0.032
Mg	2.838	2.610	2.874	3.102	2.734	2.941	3.148	3.256	3.146	3.057	3.048	3.159
Ca	1.722	1.715	1.711	1.724	1.740	1.698	1.682	1.463	1.463	1.336	1.627	1.499
Na	0.640	0.638	0.494	0.698	0.621	0.527	0.583	0.492	0.370	0.330	0.499	0.335
K	0.049	0.058	0.048	0.047	0.058	0.048	0.051	0.036	0.028	0.037	0.043	0.028
Mole percentage												
Mg	46.3	41.1	43.6	49.1	42.4	45.0	51.5	50.0	49.5	48.5	49.4	50.3
Fe	25.6	32.0	30.4	23.6	30.6	29.0	21.0	27.5	27.4	30.3	24.3	25.9
Ca	28.1	26.9	26.0	27.3	27.0	26.0	27.5	22.5	23.1	21.2	26.3	23.8

Figure 7:

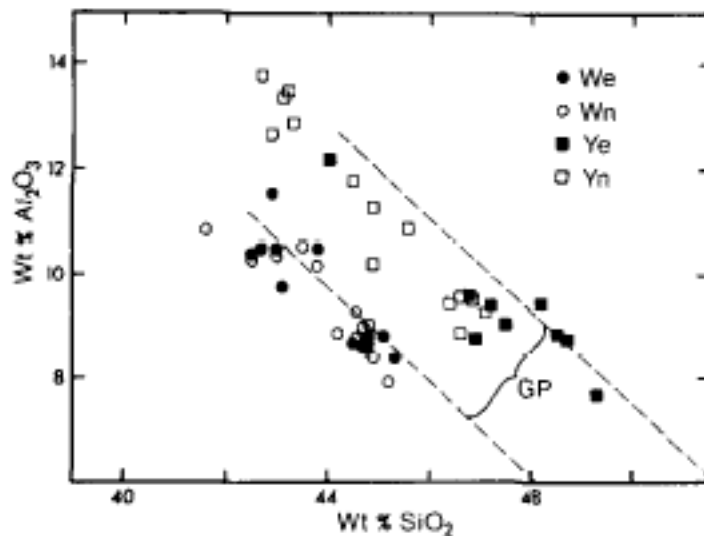


FIG. 7. Variation of Al₂O₃ (wt%) with SiO₂ (wt%) in hornblende in MSH tephtras. The compositional range of hornblende in Glacier Peak tephra (Westgate and Evans, 1978) is indicated by the dashed boundaries.

Table 6:

TABLE 6. AVERAGE COMPOSITIONS OF IRON-TITANIUM OXIDES IN MOUNT ST. HELENS TEPHRAS

	We	Wn	Ye	Yn
			Magnetite	
SiO ₂	0.24 (0.05) ^a	0.23 (0.06)	0.12 (0.04)	0.12 (0.04)
TiO ₂	9.62 (0.27)	9.70 (0.15)	5.79 (0.23)	5.58 (0.23)
Al ₂ O ₃	2.09 (0.10)	2.42 (0.16)	2.76 (0.24)	2.70 (0.13)
FeO	37.6 (0.2)	38.0 (0.2)	34.3 (0.3)	34.1 (0.3)
Fe ₂ O ₃	46.4 (0.2)	45.9 (0.3)	53.9 (0.4)	54.4 (0.4)
MnO	0.36 (0.03)	0.35 (0.03)	0.36 (0.02)	0.34 (0.03)
MgO	1.28 (0.15)	1.13 (0.24)	1.24 (0.19)	1.26 (0.17)
Sum	97.59	97.73	98.47	98.50
n ^b	20	34	33	19
Mole% Üsp	29.3	29.7	17.7	17.0
			Ilmenite	
SiO ₂	0.20 (0.09)	0.20 (0.07)	0.21 (0.11)	0.17 (0.06)
TiO ₂	44.8 (0.6)	44.9 (0.5)	39.3 (1.2)	39.9 (0.7)
Al ₂ O ₃	tr	tr	0.33 (0.14)	0.25 (0.11)
FeO	36.8 (0.2)	37.3 (0.3)	31.4 (0.5)	32.1 (0.4)
Fe ₂ O ₃	14.0 (0.1)	13.7 (0.1)	24.9 (0.4)	24.2 (0.3)
MnO	0.50 (0.03)	0.49 (0.03)	0.46 (0.04)	0.48 (0.03)
MgO	1.81 (0.16)	1.58 (0.22)	2.09 (0.28)	1.98 (0.25)
Sum	98.11	98.17	98.69	99.08
n	18	24	8	6
Mole% Hem	13.5	13.2	24.1	23.2
T (°C)	821	819	884	860
-log f _{O₂}	13.0	13.1	10.6	11.1

^a Standard deviation given in parentheses.

^b n = number of analyses.

Table 7:

TABLE 7. AVERAGE COMPOSITIONS OF IRON-TITANIUM OXIDES IN PACIFIC NORTHWEST TEPHRAS				
	St. Helens "Y"	Glacier Peak	Bridge River	Mazama
			Magnetite ^a	
SiO ₂	n.g. ^b	0.05 (0.03)	n.g.	n.g.
TiO ₂	5.85 (0.29)	6.15 (0.06)	7.56 (0.26)	8.66 (0.36)
Al ₂ O ₃	2.61 (0.09)	2.23 (0.06)	1.87 (0.08)	2.09 (0.17)
FeO	31.33 (0.30)	34.78 (0.12)	30.39 (0.37)	30.03 (0.32)
Fe ₂ O ₃	58.54 (0.85)	54.98 (0.03)	57.38 (0.77)	55.68 (0.64)
MnO	0.34 (0.04)	0.36 (0.04)	0.45 (0.05)	0.44 (0.09)
MgO	1.18 (0.10)	1.32 (0.04)	1.99 (0.18)	2.40 (0.23)
Sum	99.85	99.87	99.64	99.30
<i>n</i>	45	3	36	19
FeO ^c	34.6	34.8	34.6	34.9
Fe ₂ O ₃ ^c	54.9	55.0	52.7	50.2
Mole% Üsp ^c	17.6	18.3	22.3	25.6
			Ilmenite ^d	
SiO ₂	0.10 (0.04)	0.05 (0.05)	0.26 (0.04)	0.16 (0.09)
TiO ₂	40.30 (0.75)	38.33 (0.54)	37.54 (0.75)	39.65 (0.82)
Al ₂ O ₃	0.36 (0.02)	0.31 (0.02)	0.33 (0.03)	0.36 (0.02)
FeO	32.94 (0.66)	30.65 (0.55)	29.37 (0.85)	30.36 (0.52)
Fe ₂ O ₃	23.94 (1.41)	28.51 (1.27)	29.92 (0.90)	25.14 (1.25)
MnO	0.38 (0.03)	0.38 (0.05)	0.36 (0.11)	0.38 (0.03)
MgO	1.70 (0.07)	1.96 (0.08)	2.36 (0.11)	2.86 (0.13)
Sum	99.72	100.19	100.14	98.91
<i>n</i>	8 ^e	4	15 ^e	21 ^e
FeO ^c	32.9	30.7	29.5	30.5
Fe ₂ O ₃ ^c	23.9	28.5	29.8	25.0
Mole% Hem ^c	22.9	27.1	28.4	24.0
<i>T</i> (°C)	857	954	1019	966
-log <i>f</i> _{O₂}	11.2	9.3	8.4	9.6

^a Data for St. Helens Y, Bridge River, and Mazama tephtras taken from Brewster and Barnett (1979). Data for Glacier Peak taken from Westgate and Evans (1978).

^b n.g. = not given.

^c Recalculated values; see text for discussion.

^d Taken from Mathewes and Westgate (1980).

^e Each analysis is based on an average of ten grains.

Figure 8:

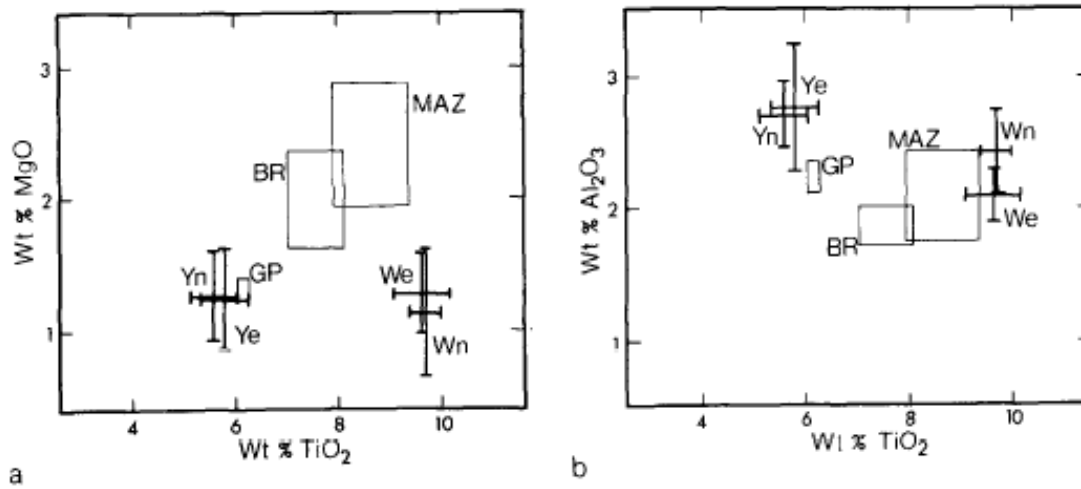


FIG. 8. MgO and TiO₂ and Al₂O₃ and TiO₂ in titanomagnetites from MSH We, Wn, Ye, Yn (this study), Glacier Peak tephra (Westgate and Evans, 1978), Mazama ash and Bridge River tephra (Brewster and Barnett, 1979). Mean value $\pm 2\sigma$ (σ = standard deviation) is plotted. Abbreviations as in Figure 1.

Figure 9:

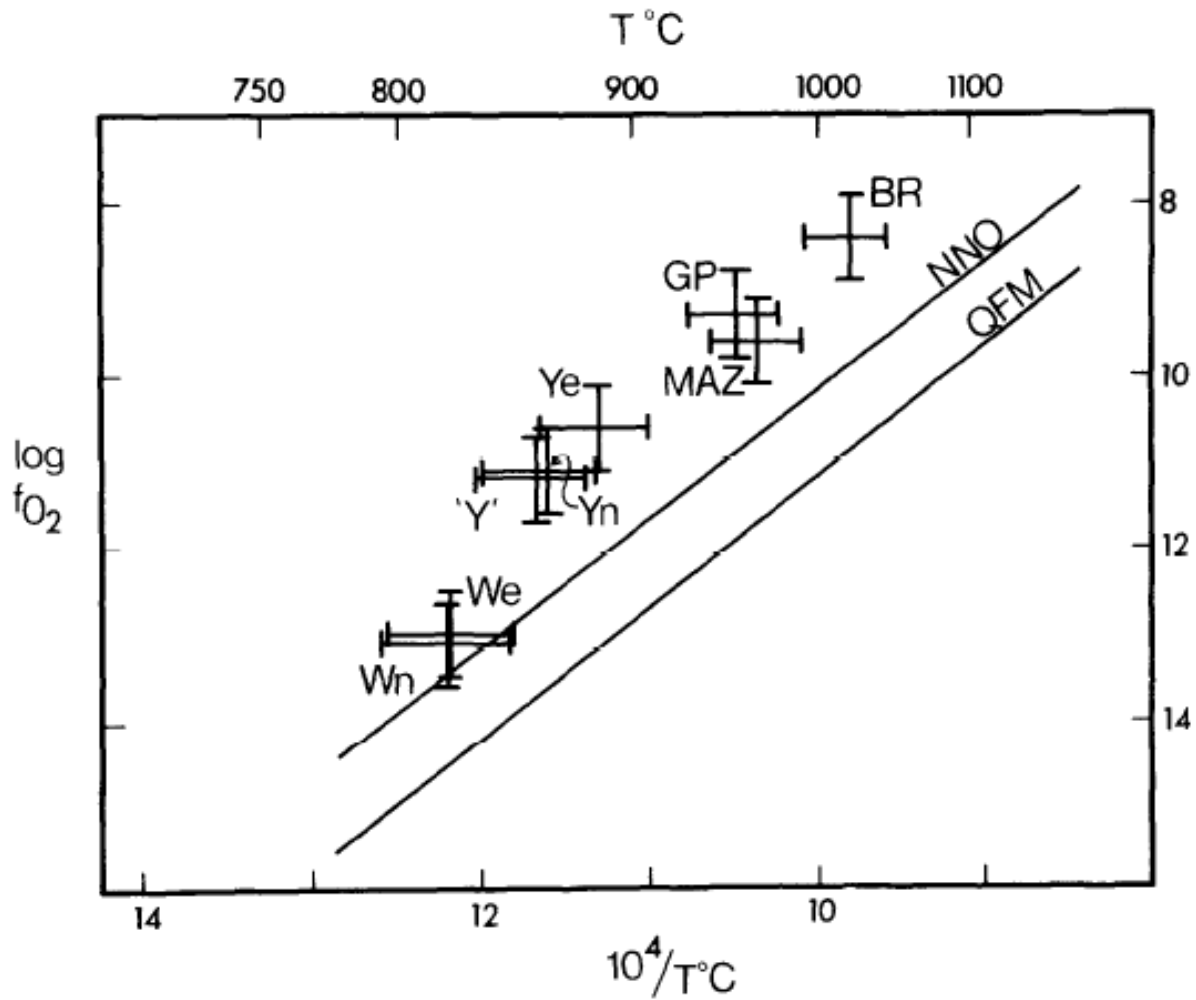


FIG. 9. Temperature vs $-\log f_{\text{O}_2}$ for various Pacific Northwest tephras. The average values are plotted with error bars indicating a range of $\pm 25^\circ\text{C}$ and ± 0.5 log unit. "Y" represents $T-f_{\text{O}_2}$ values derived using previously published analyses of Fe-Ti oxides attributed to tephra set Y; other abbreviations as in Figure 1.

Table 8:

TABLE 8. CHARACTERISTICS OF SELECTED PACIFIC NORTHWEST TEPHRAS

Unit	Age (¹⁴ C yr B.P.)	Ferromagnesian silicate assemblage ^a	Mg # ^{b,c}	Ilmenite Composition ^c	
				TiO ₂ (wt%) [mean (σ)]	Hematite (mole%)
Wc	>450 ^d	hy, hb	hy: 52–59	44.8 (0.6)	13.5
Wn	<1,150	hy, hb	hy: 53–64	44.9 (0.5)	13.2
Ye	>3,000 ^d	hb, cm (hy)	cm: 57–64	39.3 (1.2)	24.1
Yn	<4,000	hb, cm (hy)	cm: 60–65	39.9 (0.7)	23.2
Bridge River	~2500 ^e	hy, aug, hb (bi)	—	37.54 (0.75)	28.4
Glacier Peak	~12,000 ^f	hy, hb	hy: 65–67	38.33 (0.54)	27.1
Mazama	~6,600 ^f	aug, hy, hb	—	39.65 (0.82)	24.0

^a Abbreviations for minerals as in Figure 1. Data from Powers and Wilcox (9164), Mullineaux *et al.* (1975), and Mathewes and Westgate (1980).

^b Mg # = (Mg/(Mg + Fe)) × 100 (molar ratio).

^c Compositional data for MSH tephra from this study; other data are from sources cited elsewhere. Mole percentage hematite and ulvospinel, temperature, and $-\log f_{O_2}$ have been calculated in a similar manner for each unit as discussed in text.

^d Mullineaux *et al.* (1975).

^e Mathewes and Westgate (1980).

^f Fryxell (1965).

TABLE 8—Continued

Magnetite composition ^a					
TiO ₂ (wt%) [mean (σ)]	Al ₂ O ₃ (wt%) [mean (σ)]	MgO (wt%) [mean (σ)]	Ülvospinel (mole%)	T (°C)	$-\log f_{O_2}$
9.62 (0.27)	2.09 (0.10)	1.28 (0.15)	29.3	~ 835	~13.0
9.70 (0.15)	2.42 (0.16)	1.13 (0.24)	29.7	~ 820	~13.1
5.79 (0.23)	2.76 (0.24)	1.24 (0.19)	17.7	~ 880	~10.6
5.58 (0.23)	2.70 (0.13)	1.26 (0.17)	17.0	~ 860	~11.1
7.56 (0.26)	1.87 (0.08)	1.99 (0.18)	22.3	~1020	~ 8.4
6.15 (0.06)	2.23 (0.06)	1.32 (0.04)	18.3	~ 955	~ 9.3
8.66 (0.36)	2.09 (0.17)	2.40 (0.23)	25.6	~ 965	~ 9.6