## Highly Active Platinum Catalysts for Nitrile and Cyanohydrin Hydration: Catalyst Design and Ligand Screening via High-Throughput Techniques

Xiangyou Xing<sup>§†,‡</sup>, Chen Xu<sup>§†,‡</sup>, Bo Chen<sup>†</sup>, Chengcheng Li<sup>†</sup>, Scott C. Virgil<sup>‡</sup>\* and Robert H. Grubbs<sup>‡</sup>\*

<sup>†</sup>Shenzhen Grubbs Institute, Southern University of Science and Technology (SUSTech), Shenzhen,

518055, China

<sup>‡</sup>The Warren and Katharine Schlinger Laboratory for Chemistry and PChemical Engineering, Division

of Chemistry and Chemical Engineering, California Institute of Technology, Pasadena, California 91125,

United States

rhg@caltech.edu and svirgil@caltech.edu

## **Supplementary Information**

#### **General Procedures**

Unless otherwise stated, reactions were performed in brand-new Fisherbrand scintillation vials in a nitrogen filled glove box using dry, degassed. Organic solvents were dried by passage through an activated alumina column under argon and water was distilled under the protection of nitrogen. Commercial reagents (Sigma Aldrich or Alfa Aesar) were used as received with the exception of cyanohydrins which are purified by distillation before use. Dimethyl phosphine oxide was synthesized by following the known procedure<sup>1</sup> and it was kept in the nitrogen filled glovebox. Ligand screening was performed by using Freeslate Core Module 2 system which was enclosed in a nitrogen filled glovebox. Reaction progress was monitored by thin-layer chromatography (TLC) or Agilent 1290 UHPLC-LCMS analyses. TLC was performed using E. Merck silica gel 60 F254 precoated glass plates (0.25 mm) and visualized by UV fluorescence quenching or KMnO<sub>4</sub> staining. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian Inova 500 spectrometer (500 MHz and 126 MHz, respectively) or a Bruker AV III HD spectrometer equipped with a Prodigy liquid nitrogen temperature cryoprobe (400 MHz and 101 MHz, respectively), and are reported in terms of chemical shift relative to residual CHCl<sub>3</sub> (δ 7.26 and  $\delta$  77.16 ppm, respectively). Data for <sup>1</sup>H NMR spectra are reported as follows: chemical shift ( $\delta$  ppm) (multiplicity, coupling constant (Hz), integration). Abbreviations are used as follows: s = singlet, bs =broad singlet, d = doublet, t = triplet, q = quartet, m = complex multiplet. High-resolution mass spectra (HRMS) were obtained from the Caltech Mass Spectral Facility using a JEOL JMS-600H High Resolution Mass Spectrometer with fast atom bombardment (FAB+) ionization mode or were acquired using an Agilent 6200 Series TOF with an Agilent G1978A Multimode source in electrospray ionization (ESI+) mode.

Reagents were purchased from commercial vendors as follows: Parkins catalyst was purchased from Strem Chemicals and stored in a nitrogen-filled glovebox. Dichloro(1,5-cyclooctadiene)Platinum, silver salts (silver trifluoromethanesulfonate *et al*) were purchased from Sigma-Aldrich and stored in a nitrogen-filled glovebox. Nitriles (*p*-tolunitrile *et al*) and cyanohydrins (mandelonitrile *et al*) were also purchased from Sigma-Aldrich.

## 1. Ligand screening by using high-throughput techniques



Hydrocinnamonitrile (7) was selected as our module substrate and the screening was performed by Freeslate core module 2 high-throughput screening system, which was enclosed in a nitrogen filled glovebox. 42 bidentate ligands were investigated at a time. All the materials were prepared as stock solution: (COD)PtCl<sub>2</sub>, 0.02 M in CH<sub>2</sub>Cl<sub>2</sub>; PMe<sub>2</sub>OH, 0.02 M in THF; AgOTf, 0.04 M in THF; bidentate ligands, 0.02 M in THF; hydrocinnamonitrile (7), 0.2 M in THF with 1 mmol% 4,4'-ditert-butylbiphenyl as internal standard. 2 mL vials (42) were arranged in a freeslate plate (7 rows × 6 column), and the screening was conducted in 20 µmol scale with the following procedure: 1) ligands (50 µL) were dispensed into each vial, and then (COD)PtCl<sub>2</sub> (50 µL) was dispensed. The reaction mixture was then stirred for 2 hours at room temperature; 2) PMe<sub>2</sub>OH (50 µL) was dispensed into the above reaction mixture, and then AgOTf (50 µL) was dispensed; 3) water (150 µL) was dispensed to the above reaction mixture; 4) hydrocinnamonitrile (100 µL) was dispensed to the reaction mixture and the reaction was warmed up to 40 °C. After stirred for 10 minutes, the plate was taken outside of the glovebox and cooled to room temperature. Then, the reaction was monitor by UHPLC-LCMS, and the conversion of each reaction was calculated based on the internal standard.

Ligand	Conversion	Ligand	Conversion	Ligand	Conversion
DPEnhos	25%	DTBPF	30%	(S)-C <sub>3</sub> -TunePhos	57%
Xantohos	0%	Ph₂P <sup>t</sup> Bu₂PFerroce	0%	(R)-SYNPHOS	57%
<sup>t</sup> BuXantnhos	0%	(2S.4S)-Et-FerroTANE	64%	BiPhePhos	0%
rac-BINAP	54%	(2S.5S)-Me-Ferrocelane	64%	Josiphos SL-J001-1	33%
	0%	(2S.5S)-iPr-Ferrocelane	5%	Josiphos SL-J008-2	0%
	14%	(S)-CF <sub>2</sub> -PHOX	31%	Josiphos SL-J009-1	4%
	66%	(S)- <sup>t</sup> Bu-PHOX	3%	Josiphos SL-J013-1	0%
( <i>R</i> )-Phanenhos	32%	(4S)- <sup>t</sup> Bu-Me₀box	0%	Josiphos SL-J005-1	49%
	72%	(4S)-Ph-H <sub>2</sub> box	0%	Josiphos SL-J015-1	65%
( <i>P P</i> )-Chiranhos	21%	2.2'-Bis[(4S)-4-benzyl-2-oxazoline]	22%	Josiphos SL-J505-1	25%
(R,R)-Olinaphos	42%	(S.S)-Me-BPE	31%	Walphos SL-W001-1	3%
	69%	(28 58)-Et-DUPHOS	20%	Walphos SL-W009-1	10%
	69%	( <i>R</i> )-QUINAP	22%	Walphos SL-W009-1	20%
DIPF	6%	(R)-SEGPHOS	63%	Walphos SL-W009-1	5%

Table 1. Ligand screen and their corresponding conversions



Figure 1. Ligand screen and their corresponding conversions

## Note:

DPEphos: Bis[(2-diphenylphosphino)phenyl]ether;

Xantphos: 4,5-Bis(diphenylphosphino)-9,9-dimethylxanthene;

<sup>t</sup>BuXantphos: 9,9-Dimethyl-4,5-bis(di-*tert*-butylphosphino)xanthene;

rac-BINAP: (±)-2,2'-Bis(diphenylphosphino)-1,1'-binaphthalene;

(*S*)-BINAPINE: (3S,3'S,4S,4'S,11bS,11'bS)-(+)-4,4'-Di-t-butyl-4,4',5,5'-tetrahydro-3,3'-bi-3H-dinaphtho[2,1-c:1',2'-e]phosphepin;

(S)-(3,5-*t*-Bu-MeOBIPHEP): (S)-2,2'-Bis[bis(3,5-di-*tert*-butyl)phosphino]-6,6'-dimethoxy-1,1'-biphenyl;

*R*,*R*)-DIOP: (–)-2,3-*O*-Isopropylidene-2,3-dihydroxy-1,4-bis(diphenylphosphino)butane;

(*R*)-PhanePhos: (*R*)-(–)-4,12-Bis(diphenylphosphino)-[2.2]-paracyclophane;

(*R*)-SDP: Spirobiindane-bis-PPh<sub>2</sub>;

(R,R)-Chiraphos: Ph<sub>2</sub>P-CH(Me)CH(Me)-PPh<sub>2</sub>;

(*R*,*R*)-Norphos: Ph<sub>2</sub>P-norbornene-PPh<sub>2</sub>;

DPPF: 1,1'-Ferrocenediyl-bis(diphenylphosphine);

DIPF: 1,1'-Ferrocenediyl-bis(di-*iso*-propylphosphine);

(*S*,*S*)-BDPP: Ph<sub>2</sub>P-CH(Me)CH<sub>2</sub>CH(Me)-PPh<sub>2</sub>;

DTBPF: 1,1'-Ferrocenediyl-bis(di-tert-butyllphosphine);

Ph<sub>2</sub>P<sup>t</sup>Bu<sub>2</sub>PFerroce: 1-Diphenylphosphino-1'-di-tert-butylphosphinoferrocene;

(2S,4S)-Et-FerroTANE: (-)-1,1'-Bis[(2S,4S)-2,4-Diethylphosphotano]Ferrocene;

(2*S*,5*S*)-Me-Ferrocelane: 1,1'-Bis[(2*S*,5*S*)-2,5-dimethylphospholano]Ferrocene;

(2*S*,5*S*)-<sup>*i*</sup>Pr-Ferrocelane: 1,1'-Bis[(2*S*,5*S*)-2,5-diisopropylphospholano]Ferrocene;

(*S*)-CF<sub>3</sub>-PHOX: (*S*)-4-tri-fluoro-2-[2-(diphenylphosphino)phenyl]-2-oxazoline;

(S)- <sup>t</sup>Bu-PHOX: (S)-4-*tert*-Butyl-2-[2-(diphenylphosphino)phenyl]-2-oxazoline;

(4*S*)- <sup>*t*</sup>Bu-Me<sub>2</sub>box: 2,2'-Isopropylidenebis[(4*S*)-4-*tert*-butyl-2-oxazoline];

(4*S*)-Ph-H<sub>2</sub>box: 2,2'-Methylenebis[(4*S*)-4-phenyl-2-oxazoline];

(2*S*,5*S*)-Me-BPE: (–)-1,2-Bis[(2*S*,5*S*)-2,5-dimethylphospholano]ethane;

(2R,5R)-Et-DUPHOS: (-)-1,2-Bis[(2R,5R)-2,5-diethylphospholano]benzene

(*R*)-QUINAP: (*R*)-(+)-1-(2-Diphenylphosphino-1-naphthyl)isoquinoline;

(*R*)-SEGPHOS: (*R*)-(+)-5,5'-Bis(diphenylphosphino)-4,4'-bi-1,3-benzodioxole, [4(*R*)-(4,4'-bi-1,3-benzodioxole)-5,5'-diyl]bis[diphenylphosphine;

(*S*)-C<sub>3</sub>-TunePhos: (*R*)-1,13-Bis(diphenylphosphino)-7,8-dihydro-6*H*-dibenzo[f,h][1,5]dioxonin; (*R*)-SYNPHOS: R-(+)-6,6'-Bis(diphenylphosphino)-2,2',3,3'-tetrahydro-5,5'-bi-1,4-benzodioxin;

SL-J001-1 : (*R*)-1-[(*S*<sub>P</sub>)-2-(Diphenylphosphino)ferrocenyl]ethyldicyclohexylphosphine;

SL-J008-2: (S)-1-{( $R_P$ )-2-[Bis[3,5-bis(trifluoromethyl)phenyl]phosphino]ferrocenyl}ethyldi(3,5-xylyl)phosphine

- SL-J009-1: (*R*)-1-[(*S*<sub>P</sub>)-2-(Dicyclohexylphosphino)ferrocenyl]ethyldi-*tert*-butylphosphine
- SL-J013-1: (*R*)-1-{(*S*<sub>P</sub>)-2-[Bis(4-methoxy-3,5-dimethylphenyl)phosphino]ferrocenyl}ethyldi-*tert*-butylphosphine
- SL-J005-1: (*R*)-1-[(*S*<sub>P</sub>)-2-(Diphenylphosphino)ferrocenyl]ethyldi(3,5-xylyl)phosphine
- SL-J015-1: (*R*)-1-{(*S*<sub>P</sub>)-2-[Di(2-furyl)phosphino]ferrocenyl}ethyldi(3,5-xylyl)phosphine
- SL-J505-1: (R)-1-[(S<sub>P</sub>)-2-(Di-*tert*-butylphosphino)ferrocenyl]ethylbis(2-methylphenyl)phosphine

 $SL-W001-1: (R)-1-\{(R_P)-2-[2-(Diphenylphosphino)phenyl] ferrocenyl\} ethylbis [3,5-bis-(trifluoromethyl)phenyl] phosphine \\$ 

 $SL-W009-1: (R)-1-\{(R_P)-2-[2-[Di(3,5-xylyl)phosphino]phenyl]ferrocenyl\}ethyldi(3,5-xylyl)phosphine \\ SL-M001-2: (2S,2'S)-1,1'-Bis[(S)-(dimethylamino)phenylmethyl]-2,2'-bis(diphenylphosphino)ferrocene \\ SL-M003-2: (R)-1-\{(R_P)-2-[2-[Di(3,5-xylyl)phosphino]phenyl]ferrocenyl\}ethyldi(3,5-xylyl)phosphine \\ \label{eq:slower}$ 

## 2. General procedure for synthesis of the catalysts 2a and 2b



**Step 1**: In a nitrogen filled glovebox, to a 20mL scintillation vial with a magnetic stir bar were added DPPF [1,1'-ferrocenediyl-bis(diphenylphosphine)] (832 mg, 1.5 mmol), (COD)PtCl<sub>2</sub> (561 mg, 1.5 mmol) and 4 mL CH<sub>2</sub>Cl<sub>2</sub>, then the vial was capped and taken outside of the dry box and stirred at room temperature for 4 hours. The yellow solution was filtered and concentrated to a yellow solid. The solid was recrystallized from CH<sub>2</sub>Cl<sub>2</sub> and Hexane to give (DPPF)PtCl<sub>2</sub> (1.03 g, 85% yield).

**Step 2**: In a nitrogen filled glovebox, to a 10 mL vial with a magnetic stir bar were added (DPPF)PtCl<sub>2</sub> (205 mg, 0.25 mmol), silver salt (for AgOTf: 64 mg, 0.25 mmol; for AgBF<sub>4</sub>: 49 mg, 0.25 mmol), dimethylphosphine oxide (20 mg, 0.26 mmol) and 2 mL CH<sub>2</sub>Cl<sub>2</sub>. The vial was taken outside of the dry box and stirred at room temperature for 3 hours. The yellow solution was filtered and concentrated to provide a yellow solid. The solid was recrystallized from  $CH_2Cl_2$  and Hexane to give **2a** (185 mg, 73% yield) or **2b** (178 mg, 76% yield).

## Catalyst 2a:

<sup>1</sup>**H** NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  7.96 – 7.89 (m, 4H), 7.81 (ddd, *J* = 11.9, 8.3, 1.3 Hz, 4H), 7.69 – 7.64 (m, 2H), 7.63 – 7.56 (m, 6H), 7.51 (td, *J* = 7.7, 2.6 Hz, 4H), 4.51 (d, *J* = 1.8 Hz, 2H), 4.40 (d, *J* = 1.8 Hz, 2H), 4.39 (q, *J* = 1.9 Hz, 2H), 4.13 (q, *J* = 1.9 Hz, 2H), 1.72 (d, *J* = 2.5 Hz, 3H), 1.70 (d, *J* = 2.5 Hz, 3H) ppm

<sup>13</sup>**C** NMR (126 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  134.9 (dd, J = 11.1, 1.6 Hz), 134.2 (d, J = 11.9 Hz), 132.3 (d, J = 2.8 Hz), 131.5 (d, J = 2.7 Hz), 130.7, 130.2, 130.1, 129.5, 128.7 (d, J = 11.6 Hz), 128.3 (d, J = 11.0 Hz), 75.7 (d, J = 12.3 Hz), 75.5 (d, J = 10.8 Hz), 74.6 (d, J = 7.5 Hz), 73.9 (d, J = 8.2 Hz), 18.9 (d, J = 5.0 Hz), 18.6 (d, J = 3.8 Hz) ppm

<sup>31</sup>**P** NMR (121 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  105.0 (d, J = 24.4 Hz), 101.2 (d, J = 22.9 Hz), 93.5 (d, J = 20.0 Hz), 89.6 (d, J = 20.0 Hz), 82.0 (d, J = 19.2 Hz), 78.1 (d, J = 20.4 Hz), 34.6 (d, J = 15.5 Hz), 30.7 (d, J = 18.3 Hz), 25.5 (d, J = 15.8 Hz), 21.7 (d, J = 15.8 Hz), 16.43 (d, J = 15.3 Hz), 14.9 (dd, J = 20.0, 15.7 Hz), - 0.97 ppm

<sup>19</sup>**F NMR (282 MHz, CD<sub>2</sub>Cl<sub>2</sub>):** δ -79.1 ppm

**HRMS (ESI+):** [C36H34P3OFePt]<sup>+</sup>: 826.0814, found 826.0816.

#### Catalyst 2b:

<sup>1</sup>**H** NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  8.01 (ddt, J = 12.8, 6.8, 1.5 Hz, 4H), 7.89-7.77 (m, 4H), 7.64-7.55 (m, 7H), 7.54-7.41 (m, 4H), 4.46 (dd, J = 2.1, 1.1 Hz, 2H), 4.44 (q, J = 1.9 Hz, 2H), 4.32 (q, J = 1.6 Hz, 2H), 4.05 (q, J = 1.9 Hz, 2H), 1.64 (d, J = 2.6 Hz, 3H), 1.61 (d, J = 2.6 Hz, 3H) ppm

<sup>13</sup>**C** NMR (101 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  135.0 (dd, J = 11.1, 1.6 Hz), 134.6 (d, J = 11.8 Hz), 131.6 (d, J = 2.9 Hz), 131.4 (d, J = 2.3 Hz), 131.2 (d, J = 2.5 Hz), 130.9 (d, J = 2.5 Hz), 130.0, 129.3, 128.3 (d, J = 12.0 Hz), 128.2 (d, J = 12 Hz), 75.4 (dd, J = 10.5, 1.9 Hz), 75.2 (d, J = 10.3 Hz), 74.3 (d, J = 7.1 Hz), 73.5 (d, J = 7.7 Hz), 18.4 (d, J = 5.1 Hz), 18.0 (d, J = 5.3 Hz) ppm

<sup>31</sup>**P NMR (121 MHz, CD<sub>2</sub>Cl<sub>2</sub>):** δ 95.8 (d, *J* = 20.0 Hz), 92.0 (d, *J* = 20.2 Hz), 25.6 (d, *J* = 15.4 Hz), 21.8 (d, *J* = 15.7 Hz), 14.5 (dd, *J* = 20.1, 15.5 Hz) ppm

<sup>19</sup>F NMR (282 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ -150.8 ppm

**HRMS** (ESI+): [C36H34P3OFePt]<sup>+</sup>: 826.0814, found 826.0814.

## 3. Synthesis of catalyst 2c



**Step 1:** In a nitrogen filled glovebox, to a 20 mL scintillation vial with a magnetic stir bar were added 1,1'-Bis[bis(5-methyl-2-furanyl)phosphino]ferrocene (182 mg, 0.32 mmol), (COD)PtCl<sub>2</sub> (120 mg, 0.32 mmol) and 3 mL CH<sub>2</sub>Cl<sub>2</sub>, then the vial was taken outside of the dry box and stirred at room temperature for 10 hours. The yellow solution was filtered and then evaporated to provide yellow solid, which was recrystallized through 1 mL CH<sub>2</sub>Cl<sub>2</sub> and 1 mL hexane to give yellow precipitate. The solid was collected and dried in *vacuo* to yield (dmfpf)PtCl<sub>2</sub> (220 mg, 82% yield).

**Step 2:** In a nitrogen filled glovebox, to a 10 mL scintillation vial with a magnetic stir bar were added (dmfpf)PtCl<sub>2</sub> (209 mg, 0.25 mmol), silver trifluoromethanesulfonate (64 mg, 0.25 mmol), dimethylphosphine oxide (20 mg, 0.26 mmol) and 2 mL CH<sub>2</sub>Cl<sub>2</sub>. Then the vial was taken outside of the glovebox and was stirred at room temperature for 3 hours. The orange solution was filtered and CH<sub>2</sub>Cl<sub>2</sub> was evaporated to provide orange solid, which was recrystallized through CH<sub>2</sub>Cl<sub>2</sub> and Hexane to give yellow precipitate. The solid was collected and dried in vacuo to yield catalyst 2c (198 mg, 76% yield).

<sup>1</sup>**H** NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  7.01 (dt, J = 15.5, 2.9 Hz, 4H), 6.31 – 6.18 (m, 4H), 4.49 (dq, J = 5.8, 1.6 Hz, 4H), 4.41 (q, J = 2.1 Hz, 2H), 4.33 – 4.26 (m, 2H), 2.43 (dd, J = 15.3, 0.8 Hz, 12H), 2.08 – 1.74 (m, 6H) ppm

<sup>13</sup>C NMR (101 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ 159.6 (d, J = 5.5 Hz), 159.5 (d, J = 6.6 Hz), 140.9, 140.7, 134.0, 139.6, 126.2 (d, J = 19.0 Hz), 125.0 (d, J = 19.3 Hz), 108.4 (d, J = 8.0 Hz), 107.8 (d, J = 8.0 Hz), 75.5–75.00 (m), 74.34(d, J = 8.4 Hz), 74.0 (d, J = 9.4 Hz), 19.2 (d, J = 5.1 Hz), 18.8 (d, J = 5.1 Hz), 13.8, 13.8 ppm <sup>31</sup>P NMR (162 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ 102.40 (d, J = 17.8 Hz), 99.41 (d, J = 18.9 Hz), 93.75 (d, J = 18.6 Hz), 90.76 (d, J = 18.6 Hz), 85.10 (d, J = 18.8 Hz), 82.11 (d, J = 18.8 Hz), -2.95 (d, J = 19.5 Hz), -5.93 (d, J = 19.7 Hz), -9.63 (d, J = 19.6 Hz), -12.02 (t, J = 18.9 Hz), -12.62 (d, J = 19.5 Hz), -16.31 (d, J = 20.5 Hz), -19.30 (d, J = 19.4 Hz), -23.89 (t, J = 19.1 Hz), -35.76 (t, J = 19.1 Hz) ppm <sup>19</sup>F NMR (282 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ -79.0 (d, J = 3.2 Hz) ppm HRMS (ESI+): [C32H34P305FePt]<sup>+</sup>: 842.0611, found 842.0607.

## 4. Syntheses of catalysts 2d and 2e



**Step 1:** In a nitrogen filled glovebox, to a 10 mL scintillation vial with a magnetic stir bar were added 1,1'-Bis[bis(4-(trifluoromethyl)phenyl)phosphino]ferrocene <sup>2</sup> (265 mg, 0.32 mmol), (COD)PtCl<sub>2</sub> (100 mg, 0.27 mmol) and 3 mL CH<sub>2</sub>Cl<sub>2</sub>, then the vial was taken outside of the dry box and stirred at room temperature for 3 hours. The yellow solution was filtered and then evaporated to provide yellow solid, which was recrystallized through 1 mL CH<sub>2</sub>Cl<sub>2</sub> and 1 mL hexane to give yellow precipitate. The solid was collected and dried in *vacuo* to yield [dp(4-CF<sub>3</sub>)pf]PtCl<sub>2</sub> (280 mg, 95% yield).

**Step 2:** In a nitrogen filled glovebox, to a 10 mL scintillation vial with a magnetic stir bar were added  $[dp(4-CF_3)pf]PtCl_2$  (280 mg, 0.26 mmol), silver trifluoromethanesulfonate (65 mg, 0.26 mmol), dimethylphosphine oxide (21 mg, 0.27 mmol) and 2 mL CH<sub>2</sub>Cl<sub>2</sub>. Then the vial was taken outside of the glovebox and was stirred at room temperature for 3 hours. The orange solution was filtered and CH<sub>2</sub>Cl<sub>2</sub> was evaporated to provide orange solid, which was recrystallized through CH<sub>2</sub>Cl<sub>2</sub> and Hexane to give yellow precipitate. The solid was collected and dried in vacuo to yield catalyst 2c (270 mg, 82% yield).

<sup>1</sup>**H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>):**  $\delta$  7.01 (dt, *J* = 15.5, 2.9 Hz, 4H), 6.31 – 6.18 (m, 4H), 4.49 (dq, *J* = 5.8, 1.6 Hz, 4H), 4.41 (q, *J* = 2.1 Hz, 2H), 4.33 – 4.26 (m, 2H), 2.43 (dd, *J* = 15.3, 0.8 Hz, 12H), 2.08 – 1.74 (m, 6H) ppm

<sup>13</sup>C NMR (101 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  159.6 (d, *J* = 5.5 Hz), 159.5 (d, *J* = 6.6 Hz), 140.9, 140.7, 134.0, 139.6, 126.2 (d, *J* = 19.0 Hz), 125.0 (d, *J* = 19.3 Hz), 108.4 (d, *J* = 8.0 Hz), 107.8 (d, *J* = 8.0 Hz), 75.5–75.00 (m), 74.34 (d, *J* = 8.4 Hz), 74.0 (d, *J* = 9.4 Hz), 19.2 (d, *J* = 5.1 Hz), 18.8 (d, *J* = 5.1 Hz), 13.8, 13.8 ppm <sup>31</sup>P NMR (162 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  102.40 (d, *J* = 17.8 Hz), 99.41 (d, *J* = 18.9 Hz), 93.75 (d, *J* = 18.6 Hz), 90.76 (d, *J* = 18.6 Hz), 85.10 (d, *J* = 18.8 Hz), 82.11 (d, *J* = 18.8 Hz), -2.95 (d, *J* = 19.5 Hz), -5.93 (d, *J* 

= 19.7 Hz), -9.63 (d, J = 19.6 Hz), -12.02 (t, J = 18.9 Hz), -12.62 (d, J = 19.5 Hz), -16.31 (d, J = 20.5Hz), -19.30 (d, J = 19.4 Hz), -23.89 (t, J = 19.1 Hz), -35.76 (t, J = 19.1 Hz) ppm <sup>19</sup>**F NMR (282 MHz, CD<sub>2</sub>Cl<sub>2</sub>):**  $\delta$  -79.0 (d, J = 3.2 Hz) ppm

**HRMS (ESI+):** [C40H31ClF12FeOP3Pt]<sup>+</sup>: 1134.0082, found 1134.0061.



Step 1: In a nitrogen filled glovebox, to a 10 mL scintillation vial with a magnetic stir bar were added 1,1'-Bis[bis(3,5-(trifluoromethyl)phenyl)phosphino]ferrocene <sup>2</sup>(308 mg, 0.28 mmol), (COD)PtCl<sub>2</sub> (100 mg, 0.27 mmol) and 3 mL CH<sub>2</sub>Cl<sub>2</sub>, then the vial was taken outside of the dry box and stirred at room temperature for 3 hours. The yellow solution was filtered and then evaporated to provide yellow solid, which was recrystallized through 1 mL CH<sub>2</sub>Cl<sub>2</sub> and 1 mL hexane to give yellow precipitate. The solid was collected and dried in *vacuo* to yield [dp(3,5-CF<sub>3</sub>)pf]PtCl<sub>2</sub> (330 mg, 91% yield).

Step 2: In a nitrogen filled glovebox, to a 10 mL scintillation vial with a magnetic stir bar were added [dp(4-CF<sub>3</sub>)pf]PtCl<sub>2</sub> (200 mg, 0.15 mmol), silver trifluoromethanesulfonate (39 mg, 0.15 mmol), dimethylphosphine oxide (12 mg, 0.15 mmol) and 2 mL CH<sub>2</sub>Cl<sub>2</sub>. Then the vial was taken outside of the glovebox and was stirred at room temperature for 3 hours. The orange solution was filtered and  $CH_2Cl_2$ was evaporated to provide orange solid, which was recrystallized through CH<sub>2</sub>Cl<sub>2</sub> and Hexane to give yellow precipitate. The solid was collected and dried in vacuo to yield catalyst **2e** (171 mg, 73% yield).

<sup>1</sup>**H NMR (400 MHz, CDCl<sub>3</sub>):**  $\delta$  8.20 (s, 4H), 8.16 (d, J = 13.5 Hz, 4H), 8.05 (d, J = 11.3 Hz, 4H), 4.74 (s, 2H), 4.61 (s, 2H), 4.35 (d, J = 1.7 Hz, 2H), 3.91 (d, J = 1.7 Hz, 2H), 1.86 (d, J = 9.6 Hz, 6H).

<sup>13</sup>C NMR (101 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  159.6 (d, J = 5.5 Hz), 159.5 (d, J = 6.6 Hz), 140.9, 140.7, 134.0, 139.6, 126.2 (d, J = 19.0 Hz), 125.0 (d, J = 19.3 Hz), 108.4 (d, J = 8.0 Hz), 107.8 (d, J = 8.0 Hz), 75.5–75.00 (m), 74.34(d, J = 8.4 Hz), 74.0 (d, J = 9.4 Hz), 19.2 (d, J = 5.1 Hz), 18.8 (d, J = 5.1 Hz), 13.8, 13.8 ppm<sup>31</sup>**P NMR** (**162 MHz, CD<sub>2</sub>Cl<sub>2</sub>**):  $\delta$  102.40 (d, J = 17.8 Hz), 99.41 (d, J = 18.9 Hz), 93.75 (d, J = 18.6 Hz), 90.76 (d, J = 18.6 Hz), 85.10 (d, J = 18.8 Hz), 82.11 (d, J = 18.8 Hz), -2.95 (d, J = 19.5 Hz), -5.93 (d, J = 19.7 Hz), -9.63 (d, J = 19.6 Hz), -12.02 (t, J = 18.9 Hz), -12.62 (d, J = 19.5 Hz), -16.31 (d, J = 20.5Hz), -19.30 (d, J = 19.4 Hz), -23.89 (t, J = 19.1 Hz), -35.76 (t, J = 19.1 Hz) ppm <sup>19</sup>F NMR (282 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  -79.0 (d, J = 3.2 Hz) ppm **HRMS** (**ESI**+): [C44H27ClF24FeOP3Pt]<sup>+</sup>: 1405.9577, found 1405.9560.

## 5. Hydration of nitriles and cyanohydrins by catalyst 2a and 2c

#### Hydration of acetonitrile (9) with catalyst 2a and AgOTf at 40 °C 5.1.

Catalyst **2a**  

$$AgOTf$$
  $H_2O$   $H_2O$   $H_2O$   $H_3CONH_2$   
**9**  $12 h \text{ or } 40 \text{ h}$  **10**

To a 20 mL scintillation vial with magnetic stir bar was added catalyst **2a** (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), acetonitrile (86.16 mmol, 4.50 mL) and water (2.1 mL). The reaction mixture was then warmed up to 40  $\mathbb{C}$  and was stirred for 12 hours. After then, the solvent was evaporated by rotary evaporator under reduced pressure. The solid collected was washed by diethyl ether twice (5 mL for each) and amide **10** (59.30 mmol, 3.503 g, 69% yield, 5930 TON) was obtained. To a 20 mL scintillation vial with magnetic stir bar was added catalyst **2a** (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), acetonitrile (162.75 mmol, 8.5 mL) and water (4 mL). The reaction mixture was then warmed up to 40  $\mathbb{C}$  and was stirred for 40 hours. After then, the solvent was evaporated by rotary evaporator under reduced pressure. The solid collected was washed by diethyl ether twice (10 mL for each) and amide **10** (127.12 mmol, 7.508g, 78% yield, 12712 TON) was obtained. <sup>1</sup>H NMR (**500 MHz, Deuterium Oxide):**  $\delta$  6.01 (d, *J* = 39.7 Hz, 2H), 1.99 (d, *J* = 0.5 Hz, 3H).

<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>): δ 173.3, 22.7 ppm

**HRMS** (**ESI**+): calc'd for C9H11NO [M+H]<sup>+</sup> : 60.0474, found 60.0449.

## 5.2. General procedure I for hydration of nitriles with catalyst 2a and AgOTf at 40 °C

$$R^{\text{CN}} \xrightarrow{\text{CN}} 40 \, {}^{\circ}\text{C}, \, 12 \text{ h} \xrightarrow{\text{CN}} R^{\text{CN}}$$

In a nitrogen filled glovebox, to an 8 mL vial with a magnetic stir bar were added catalyst 2a (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), THF (2 mL) and water (2 mL). The mixture was stirred for 2 minutes, then the vial was taken outside of the glovebox and nitriles (10 mmol or 4 mmol) were added. The reaction mixture was then warmed up to 40 °C and was stirred for 12 hours. After then, it was diluted with water (5 mL) and extracted by ethyl acetate twice (10 mL for each). The combined organic layer was washed with brine (10 mL), and then solvent was removed by rotary evaporator to produce the crude products, which were purified by recrystallization. The products were obtained in 52%-98% yields with TONs ranging from 207 to 981.



The general procedure I was followed. The desired product **8** (1.46 g, 98% yield, 980 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/3 mL).

<sup>1</sup>**H NMR (500 MHz, CDCl<sub>3</sub>):**  $\delta$  9.31-9.21 (m, 2H), 9.22-9.09 (m, 3H), 7.92 (s, 1H), 7.62 (s, 1H), 4.93-4.89 (t, *J* = 7.6 Hz, 2H), 4.47 (t, *J* = 7.6 Hz, 2H) ppm <sup>13</sup>**C NMR (126 MHz, CDCl<sub>3</sub>):**  $\delta$  176.5, 143.0, 130.3, 130.2, 128.0, 39.2, 33.2 ppm

**HRMS (ESI+):** calc'd for C9H11NO [M+H]<sup>+</sup> : 150.0913, found 150.0911



The general procedure I was followed. The desired product **11p** (1.44 g, 98% yield, 981 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/4 mL).

<sup>1</sup>**H NMR (500 MHz, DMSO-***d*<sub>6</sub>): δ 7.57-7.51 (m, 3H), 7.43-7.33 (m, 4H), 7.11 (s, 1H), 6.60 (d, *J* = 15.9 Hz, 1H) **ppm** 

<sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 167.1, 139.6, 135.3, 129.9, 129.4, 128.0, 122.8 ppm HRMS (ESI+): calc'd for C9H9NO [M+H]<sup>+</sup>: 148.0757, found 148.0760.



The general procedure I was followed. The desired product **12p** (1.25 g, 93% yield, 926 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/2 mL).

<sup>1</sup>**H** NMR (500 MHz, DMSO- $d_6$ ):  $\delta$  7.91 (s, 1H), 7.79 (d, J = 8.2 Hz, 2H), 7.29 (s, 1H), 7.27-7.23 (m, 2H), 2.36 (s, 3H) ppm

<sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 168.2, 141.5, 131.9, 129.2, 127.93, 21.4 ppm HRMS (ESI+): calc'd for C8H9NO [M+H]<sup>+</sup>: 136.0757, found 136.0757.



The general procedure I was followed. The desired product **13p** (1.64 g, 96% yield, 960 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/4 mL).

<sup>1</sup>**H NMR (500 MHz, DMSO-***d***<sub>6</sub>):** δ 8.19 (s, 1H), 8.06-8.03 (m, 2H), 7.84-7.78 (m, 2H), 7.62 (s, 1H) ppm

<sup>13</sup>C NMR (126 MHz, DMSO- $d_6$ ):  $\delta$  167.1, 138.5, 131.6 (q, J = 32.0 Hz), 128.75, 125.7 (q, J = 3.8 Hz), 124.4 (q, J = 273.4 Hz) ppm



The general procedure I was followed. The desired product **14p** (0.82 g, 92% yield, 921 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/4 mL).

<sup>1</sup>**H NMR (500 MHz, Deuterium Oxide):**  $\delta$  3.72 (t, *J* = 6.1 Hz, 2H), 2.38 (t, *J* = 6.1 Hz, 2H) ppm

<sup>13</sup>C NMR (126 MHz, Deuterium Oxide): δ 177.3, 57.7, 37.7 ppm

**HRMS** (**ESI**+): calc'd for C3H7NO2 [M+H]<sup>+</sup>: 90.0555, found 90.0555.



The general procedure I was followed. The desired product **15p** (1.21 g, 96% yield, 960 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/3 mL).

<sup>1</sup>**H NMR (500 MHz, DMSO-***d*<sub>6</sub>):  $\delta$  7.81 (dd, *J* = 1.8, 0.8 Hz, 1H), 7.78 (bs, 1H), 7.38 (bs, 1H), 7.10 (dd, *J* = 3.4, 0.8 Hz, 1H), 6.60 (dd, *J* = 3.4, 1.7 Hz, 1H) ppm

<sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 159.8, 148.5, 145.4, 114.0, 112.2 ppm

**HRMS (ESI+):** calc'd for C5H5NO2 [M+H]<sup>+</sup>: 112.0399, found 112.0411



The general procedure I was followed. The desired product **16p** (1.21 g, 96% yield, 960 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/3 mL).

<sup>1</sup>**H NMR (500 MHz, DMSO-***d*<sub>6</sub>): δ 8.14 (dd, *J* = 3.0, 1.3 Hz, 1H), 7.80 (s, 1H), 7.56 (dd, *J* = 5.0, 2.9 Hz, 1H), 7.49 (dd, *J* = 5.0, 1.3 Hz, 1H), 7.25 (s, 1H) ppm.

<sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 164.1, 138.4, 129.5, 127.6, 127.0 ppm HRMS (ESI+): calc'd for C5H5NOS [M+H]<sup>+</sup>: 128.0165, found 128.0168



The general procedure I was followed. The reaction was performed with catalyst **2a** (1 mg, 0.001 mmol), silver trifluoromethanesulfonate (0.26 mg, 0.001 mmol), (*E*)-1-cyano-3-phenylallyl acetate **17** (201 mg, 1 mmol). The desired product **17p** (188 mg, 86% yield, 860 TON) was obtained <sup>1</sup>H NMR (**400 MHz, CDCl**<sub>3</sub>):  $\delta$  7.42 – 7.27 (m, 5H), 6.78 (d, *J* = 15.9 Hz, 1H), 6.27 (dd, *J* = 15.9, 7.1 Hz, 1H), 6.10 (s, 1H), 5.76 (dd, *J* = 7.1, 1.2 Hz, 2H), 2.21 (s, 3H) ppm <sup>13</sup>C NMR (**101 MHz, CDCl**<sub>3</sub>):  $\delta$  170.6, 169.3, 135.4, 128.7, 128.6, 126.9, 122.0, 74.2, 21.0 ppm. **HRMS (ESI+):** calc'd for C12H13NNaO3 [M+Na]<sup>+</sup>: 242.0788, found 242.0788



Preparation of substrate **18**: Phenylpropiolaldehyde (1.0 g, 7.69mmol), TMSCN (0.84g, 8.84mmol) were mixed neat, and LiCl (1 mg, cat.) was added. The resulting mixture was stirred at rt under Ar. After 1h, the reaction mixture was evaporated under vacuum, and the residue was dissolved in Ac<sub>2</sub>O (1 mL). Scandium(III) trifluoromethanesulfonate (5 % mol) was added. After stirring for 1 h, the reaction mixture was purified by flash chromatography (EA/Hex = 1/10) to give **18** (1.22g, 82%). **1H NMR (400 MHz, CDCl3)**  $\delta$  7.55 – 7.33 (m, 5H), 6.28 (s, 1H), 2.22 (s, 3H) ppm **13C NMR (101 MHz, CDCl3)**  $\delta$  168.4, 132.2, 130.1, 128.6, 120.2, 113.5, 88.7, 51.0, 20.3 ppm **HRMS (ESI+):** calc'd for C12H10NO2 [M+H]<sup>+</sup>: 200.0706, found 200.0706.

Hydration of **18**: the general procedure I was followed. The reaction was performed with catalyst **2a** (1 mg, 0.001 mmol), silver trifluoromethanesulfonate (0.26 mg, 0.001 mmol) and **18** (199 mg, 1 mmol). The desired product **18p** (200 mg, 92% yield, 920 TON) was obtained.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.52 – 7.29 (m, 5H), 6.28 (d, J = 45.1 Hz, 2H), 2.22 (s, 3H) ppm <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>): δ 169.1, 167.5, 132.1, 129.4, 128.4, 121.2, 87.5, 81.5, 63.8, 20.8 ppm HRMS (ESI+): calc'd for C12H11NNaO3 [M+Na]<sup>+</sup>: 240.0631, found 240.0628



The general procedure I was followed. The reaction was performed with catalyst **2a** (0.3 mg, 0.0003 mmol), silver trifluoromethanesulfonate (0.07 mg, 0.0003 mmol), N-(1-cyano-2-phenylethyl)acetamide **19** (56 mg, 0.3 mmol). The desired product **19p** (38 mg, 61% yield, 610 TON) was obtained.

<sup>1</sup>**H NMR (400 MHz, DMSO)**:  $\delta$  8.03 (d, J = 8.5 Hz, 1H), 7.44 (s, 1H), 7.33 – 7.14 (m, 5H), 7.03 (s, 1H), 4.41 (td, J = 9.6, 4.7 Hz, 1H), 2.98 (dd, J = 13.7, 4.7 Hz, 1H), 2.71 (dt, J = 17.4, 8.7 Hz, 1H), 1.75 (s, 3H) ppm

ppm <sup>13</sup>C NMR (101 MHz, DMSO) δ 173.6, 169.5, 138.7, 129.6, 128.5, 126.6, 54.3, 38.1, 23.0 ppm HRMS (ESI+): calc'd for C11H15N2O2 [M+H]<sup>+</sup>: 207.1128, found 207.1128.



The general procedure I was followed. The desired product **20p** (520 mg, 96% yield, 385 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (3 mL/3 mL).

<sup>1</sup>**H NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>):** δ 7.48 (dd, *J* = 7.5, 1.4 Hz, 1H), 7.38 (td, *J* = 7.5, 1.5 Hz, 1H), 7.30-7.23 (m, 2H), 6.17-5.19 (m, 2H), 2.51 (s, 3H) ppm

<sup>13</sup>C NMR (126 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ 171.5, 136.3, 135.4, 131.1, 130.1, 126.8, 125.6, 19.7 ppm HRMS (ESI+): calc'd for C8H9NO [M+H]<sup>+</sup>: 136.0757, found 136.0753.



The general procedure I was followed. The desired product **21p** (420 mg, 52% yield, 207 TON) was obtained after recrystallization by  $CH_2Cl_2/Hexane$  (2 mL/4 mL).

<sup>1</sup>H NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ 7.32-7.28 (m, 2H), 7.22-7.18 (m, 2H), 5.52 (s, 1H), 5.28 (s, 1H), 2.52-2.42 (m, 2H), 2.37 (d, J = 0.6 Hz, 3H), 2.08-1.99 (m, 2H), 1.88-1.77 (m, 2H), 1.76-1.66 (m, 2H) ppm <sup>13</sup>C NMR (126 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ 178.7, 141.2, 136.5, 129.2, 126.5, 58.7, 36.6, 23.8, 20.6 ppm HRMS (ESI+): calc'd for C13H17NO [M+H]<sup>+</sup>: 204.1383, found 204.1386



The general procedure I was followed. The desired product **22p** (658 mg, 92% yield, 368 TON) was obtained after washing with  $Et_2O$  (5 mL) and recrystallization by  $CH_2Cl_2/Hexane$  (3 mL/4 mL).

<sup>1</sup>**H NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>):** δ 5.67 (s, 2H), 2.07 (p, *J* = 3.3 Hz, 3H), 1.89 (d, *J* = 2.9 Hz, 6H), 1.87-1.71 (m, 6H) ppm

<sup>13</sup>C NMR (126 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ 180.4, 40.5, 39.3, 36.4, 28.3 ppm HRMS (ESI+): calc'd for C11H17NO [M+H]<sup>+</sup>: 180.1383, found 180.1385



The general procedure I was followed. The desired product **23p** (310 mg, 64% yield, 254 TON) was obtained after washing with Et<sub>2</sub>O (5 mL) and recrystallization by CH<sub>2</sub>Cl<sub>2</sub>/Hexane (3 mL/3 mL). <sup>1</sup>H NMR (**500 MHz, Deuterium Oxide**):  $\delta$  8.72 (dd, *J* = 2.3, 0.9 Hz, 1H), 8.52 (dd, *J* = 5.0, 1.6 Hz, 1H), 8.04 (ddd, *J* = 8.0, 2.3, 1.6 Hz, 1H), 7.40 (ddd, *J* = 8.0, 5.0, 0.9 Hz, 1H) ppm <sup>13</sup>C NMR (**126 MHz, Deuterium Oxide**):  $\delta$  170.4, 151.7, 147.5, 136.3, 129.0, 124.1 ppm **HRMS (ESI+):** calc'd for C6H6N2O [M+H]<sup>+</sup>: 123.0553, found 123.0554.



Chiral nitrile **24** (96% *ee*) was prepared following Stahl and Liu's procedure.<sup>3</sup> HPLC (OD-H, 0.46\*25 cm, 5 $\mu$ m, hexane / ethanol = 98/2, flow 1 mL/min, detection at 254 nm) retention time = 13.824 min (minor) and 29.712 min (major).



Peak#	Ret. Time	Area	Height	Conc.	Unit	Mark	Name
1	13.561	23044349	1144191	49.117		М	
2	29.079	23873244	323152	50.883		M	
Total		46917593	1467343				



Hydration of **24**: the general procedure I was followed. The reaction was performed with catalyst 2a (1.7 mg, 0.0017mmol), silver trifluoromethanesulfonate (0.5 mg, 0.0017 mmol), (R)-2-(naphthalen-1-yl) propanenitrile **24** (120 mg, 0.662 mmol). The desired product **24p** (115 mg, 87% yield, 348 TON) was obtained with 97% *ee*.

<sup>1</sup>**H** NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.06 (d, J = 8.2 Hz, 1H), 7.89 (d, J = 8.3 Hz, 1H), 7.81 (d, J = 7.9 Hz, 1H), 7.64 – 7.42 (m, 4H), 5.29 (d, J = 44.6 Hz, 2H), 4.34 (q, J = 7.2 Hz, 1H), 1.72 (d, J = 7.2 Hz, 3H) ppm.

<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>): δ 177.0, 137.0, 134.1, 131.5, 129.1, 128.3, 126.7, 126.02, 125.7, 124.9, 123.3, 43.4, 17.8 ppm.

HRMS (ESI+): calc'd for C13H13NO [M+H] <sup>+</sup>200.1075, found 200.1066

HPLC (OD-H, 0.46\*25 cm, 5 $\mu$ m, hexane / ethanol = 98/2, flow 1 mL/min, detection at 254 nm) retention time = 12.405 min (minor) and 17.702 min (major).

<Chromatogram>



#### <Peak Table>

Detecto	or A 254nm						
Peak#	Ret. Time	Area	Height	Conc.	Unit	Mark	Name
1	12.256	5484153	269608	49.957		M	
2	17.792	5493507	183224	50.043		M	
Total		10977660	452832				



# 5.3. General procedure II for hydration of cyanohydrins by catalyst 2a and AgOTf at 40 °C:

HO CN AgOTf, THF/H<sub>2</sub>O  
R<sub>1</sub> R<sub>2</sub> 40 °C, 12 h 
$$R_1$$
  $R_2$   $R_1$   $R_2$   $R_1$   $R_2$   $R_1$   $R_2$   $R_1$   $R_2$   $R_1$   $R_2$   $R_1$   $R_2$   $R_2$   $R_1$   $R_2$   $R_2$   $R_1$   $R_2$   $R_2$   $R_1$   $R_2$   $R_2$   $R_2$   $R_2$   $R_1$   $R_2$   $R_2$ 

In a nitrogen filled glovebox, to a 8 mL vial with a magnetic stir bar were added **2a** (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), THF (2 mL) and water (2 mL), and the mixture was stirred for 2 minutes, then the vial was taken outside of the glovebox and cyanohydrin (2 mmol to 10 mmol) was added. The reaction mixture was then warm up to 40  $\,^{\circ}$ C and was stirred for 12 hours. After then, it was diluted with water (5 mL) and was extracted with Et<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub> (2:1) for twice (5 mL for each). Water was evaporated to produce crude product as white solid, which was washed with Et<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub> (1:1, 5 mL), and then dried in vacuo. The products were obtained in 22%-93% yield with TON ranging from 43 to 880.



The general procedure II was followed. The reaction was performed with catalyst 2a (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), acetone cyanohydrin 1 (360 *u*L, 3 mmol). The desired product 1p (115 mg, 37% yield, 112 TON) was obtained.

<sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>): δ 7.06 (s, 1H), 6.96 (m, 1H), 1.21 (s, 6H) ppm <sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 179.5, 72.1, 28.1 ppm

**HRMS (ESI+):** calc'd for C4H10NO2 [M+H]<sup>+</sup>: 104.0706, found 104.0705.



The general procedure II was followed. The reaction was performed with catalyst 2a (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), mandelonitrile (360 uL, 3 mmol). The desired product mandelamide 2p (218 mg, 48% yield, 144 TON) was obtained.

<sup>1</sup>**H NMR (500 MHz, DMSO-***d*<sub>6</sub>):  $\delta$  7.46-7.38 (m, 3H), 7.35-7.31 (m, 2H), 7.30-7.23 (m, 1H), 7.19 (s, 1H), 6.02 (d, *J* = 3.8 Hz, 1H), 4.85 (s, 1H) ppm

<sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 175.0, 141.8, 128.3, 127.7, 126.9, 73.9 ppm HRMS (ESI+): calc'd for C8H10NO2 [M+H]<sup>+</sup>: 152.0706, found 152.0695.



The general procedure II was followed. The reaction was performed with catalyst 2a (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), cyclohexanone cyanohydrin 3 (250 mg, 2 mmol). The desired product 3p (61 mg, 22% yield, 43 TON) was obtained.

<sup>1</sup>**H NMR (500 MHz, Deuterium Oxide):**  $\delta$  1.61 (td, J = 13.7, 4.6 Hz, 2H), 1.56-1.29 (m, 6H), 1.10 (qt, J = 12.7, 3.9 Hz, 1H) ppm

<sup>13</sup>C NMR (126 MHz, Deuterium Oxide): δ 183.4, 75.0, 32.9, 24.5, 20.3 ppm HRMS (ESI+): calc'd for C7H13NO2 [M+H]<sup>+</sup>: 144.1019, found 144.1019.



The general procedure II was followed. The reaction was performed with catalyst **2a** (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), and glycolic acid nitrile **4** (760 uL, 70% in water, 10 mmol). The desired product **4p** (660 mg, 88% yield, 880 TON) was obtained.

<sup>1</sup>**H NMR (500 MHz, Deuterium Oxide):** δ 3.96 (s, 2H) ppm

<sup>13</sup>C NMR (126 MHz, Deuterium Oxide): δ 178.0, 60.5 ppm

**HRMS** (**ESI**+): calc'd for C2H5NO2 [M+H]<sup>+</sup>: 76.0399, found 76.0417.



The general procedure II was followed. The reaction was performed with catalyst **2a** (10 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), and lactonitrile **5** (350 uL, 5 mmol). The desired product **5p** (412 mg, 93% yield, 463 TON) was obtained.

<sup>1</sup>**H NMR (500 MHz, DMSO-***d*<sub>6</sub>): δ 7.12 (s, 1H), 7.06 (s, 1H), 5.32 (d, *J* = 4.9 Hz, 1H), 3.90 (qd, *J* = 6.8, 4.9 Hz, 1H), 1.21 (d, *J* = 6.8 Hz, 3H) ppm

<sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>): δ 177.5, 67.5, 21.4 ppm

HRMS (ESI+): calc'd for C3H7NO2 [M+H]<sup>+</sup>: 90.0555, found 90.0549.

## 5.4. General procedure III for hydration of nitriles and cyanohydrins by using catalyst 2c and AgOTf at room temperature:



Hydration of nitriles and cyanohydrins with catalyst 2c (catalyst loading 0.1%-0.5%) was following general procedure I and II respectively with exception of performing the reactions at room

temperature. For nitriles, the products were obtained in 45%-95% yields with TONs ranging from 156 to 950; for cyanohydrins, the products were obtained in 37%-98% yields with TONs ranging from 74 to 395.



The reaction was performed with catalyst **2c** (10.3 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), acetone cyanohydrin **1** (370 uL, 4 mmol). The desired product **1p** (223 mg, 54% yield, 216 TON) was obtained.



The reaction was performed with catalyst 2c (10.3 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), cyclohexanone cyanohydrin **3** (250 mg, 2 mmol). The desired product **3p** (106 mg, 37% yield, 74 TON) was obtained.

HO CN 
$$AgOTf, THF/H_2O$$
  
4 HO CONH<sub>2</sub>  
4 HO CONH<sub>2</sub>  
4 HO CONH<sub>2</sub>

The reaction was performed with catalyst 2c (10.3 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), glycolic acid nitrile **4** (300 uL, 4 mmol, 70% in water). The desired product **4p** (295 mg, 98% yield, 395 TON) was obtained.



The reaction was performed with catalyst 2c (10.3 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), 3-thiophenecarbontrile **16** (900 *u*L, 10 mmol). The desired product **16p** (1.21 g, 95% yield, 950 TON) was obtained.



S20

The reaction was performed with catalyst 2c (10.3 mg, 0.01 mmol), silver trifluoromethanesulfonate (2.6 mg, 0.01 mmol), *o*-tolunitrile **17** (480 uL, 4 mmol). The desired product **17p** (210 mg, 45% yield, 156 TON) was obtained.

## 6. Rate comparison experiments with different catalysts



In order to investigate the relative reaction rates with different hydration catalysts, hydrocinnamonitrile (7) was selected as our module substrate and it was prepared as 1 M solution in THF with 1% mmol 4,4'-di-*tert*-butylbiphenyl as internal standard. Six reactions were performed in parallel (generally in 0.5 mL THF and 0.5 mL H<sub>2</sub>O, 0.2 mmol scale).

1) Parkins catalyst (1a) at 40 °C (0.5 mL EtOH, 0.5 mL H<sub>2</sub>O);

**2**) catalyst **2a** without additives at 40  $\mathbb{C}$ ;

3) catalyst 2b and AgBF<sub>4</sub> at room temperature;

4) catalyst 2a and AgOTf at 40 °C;

**5**) catalyst **2b** and AgBF<sub>4</sub> at 40 °C;

6) catalyst **2c** and AgOTf at room temperature.

Samples were taken at indicated times and they were monitored by UHPLC-LCMS, and the conversion was obtained by calculating relative peak area based on internal standard. The data was described below:

Entry	Catalyst	Temperature	additive	Time	Conversion
1	Parkins Catalyst	40 °C	-	10 min 30 min 50 min 90 min 120 min 180 min 240 min 300 min	3.3% 8.3% 15.2% 34.0% 50.2% 72.3% 90.9% 100%
2	Catalyst 2a	40 °C	-	10 min 20 min 30 min 40 min 50 min 60 min 90 min 120 min 180 min	10.2% 18.2% 28.4% 42.2% 53.7% 63.0% 84.2% 95.4% 100%
3	Catalyst 2a	40 °C	AgOTf	5 min 10 min	75.5% 100%
4	Catalyst 2b	23 °C	AgBF <sub>4</sub>	10 min 20 min 30 min 40 min 50 min 60 min 70 min	16.4% 32.3% 49.3% 64.8% 75.1% 91.9% 100%
5	Catalyst 2b	40 °C	AgBF <sub>4</sub>	5 min 10 min	78.3% 100%
6	Catalyst 2c	23 °C	AgOTf	5 min 10 min	85.5% 100%

 Table 2. Reaction conversions of the six hydration reactions of 7



Figure 2. Comparison of the conversions of catalyst 1a and 2a-c in hydration of 7 within 5 h



Figure 3. Comparison of the conversions of catalyst 1a and 2a-c in hydration of 7 within 50 min S23

#### 7. Rate comparison on hydration of acetonitrile (9) at room temperature



Those reactions were performed in parallel. In a nitrogen filled glovebox, Parkins catalyst (4.3 mg, 0.01 mmol), catalyst 2a and AgOTf (10 mg, 0.01 mmol; 2.6 mg, 0.01 mmol), catalyst 2c and AgOTf (10.3 mg, 0.01 mmol), catalyst 2a mmol), catalyst 2c (10.3 mg, 0.01 mmol), catalyst 2c and AgOTf (10.3 mg, 0.01 mmol), catalyst 2c (10.3 mg, 0.01 mmol), were dispensed into 20 mL scintillation vials, respectively. Acetone nitrile (3.2 mL, 60 mmol) and water (1.1 mL, 60 mmol) were added to each of them. The reaction solution was taken outside of the glovebox and performed at room temperature for 12 hours. The solvent was removed by rotary evaporator to provide white solid, which was washed with  $CH_2Cl_2/Et_2O$  (1:1, 5 mL) and dried in *vacuo*. Accordingly, 48 mg, 1056 mg, 2630 mg and 410 mg of acetamides were obtained, respectively.

Entry	Catalysts	Additives	Product (mg)	ΤΟΝ
1	Parkins	-	48	81
2	2a	-	1056	1790
3	2c	AgOTf	2630	4457
4	2c	-	410	695

Tabl	le	3
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Figure 4 Comparison of TON of catalyst 1a, 2a and 2c in hydration of acetonitrile at room temperature

## 8. Rate comparison experiments in hydration of nitrile 7 with catalysts 2a, 2c, 2d

and 2e.



In order to investigate the relative reaction rates with different hydration catalysts, hydrocinnamonitrile (7) was selected as our module substrate and it was prepared as 1 M solution in THF with 1% mmol 4,4'-di-*tert*-butylbiphenyl as internal standard. Four reactions were performed in parallel (generally in 0.5 mL THF and 0.5 mL H<sub>2</sub>O, 0.2 mmol scale).

- 1) catalyst 2a and AgOTf at 40 °C;
- 2) catalyst 2c and AgOTf at room temperature;

3) catalyst 2d and AgOTf at 40 °C;

4) catalyst 2e and AgOTf at 40 °C.

Samples were taken at indicated times and they were monitored by UHPLC-LCMS, and the conversion was obtained by calculating relative peak area based on internal standard. The data was described below:

Entry	Catalyst	Temperature	additive	Time	Conversion
1	Catalyst 2a	40 °C	AgOTf	5 min 10 min	75.5% 100%
2	Catalyst 2c	23 °C	AgOTf	5 min 10 min	85.5% 100%
3	Catalyst 2d	40 °C	AgOTf	10 min 20 min 30 min 40 min 50 min 60 min	37.4% 54.2% 70.4% 81.3% 91.8% 100%
4	Catalyst 2e	40 °C	AgOTf	10 min 20 min 30 min 40 min 50 min 60 min 90 min 120 min 180 min 240 min 300 min 360min 15 h	$\begin{array}{c} 11.3\%\\ 23.5\%\\ 33.0\%\\ 39.8\%\\ 45.4\%\\ 50.5\%\\ 57.1\%\\ 61.5\%\\ 65.5\%\\ 67.3\%\\ 68.2\%\\ 69.7\%\\ 72.6\%^{*}\end{array}$

**Table 4.** Reaction conversions of the four hydration reactions of 7

\*Note: after 15 hours, further conversion with catalyst 2e is not observed.

## 9. X-Ray Structure Determination

Crystals were mounted on polyimide MiTeGen loops with STP Oil Treatment and placed under a nitrogen stream. Low temperature (100K) diffraction data for 2a was collected on a Bruker AXS KAPPA APEX II diffractometer (50kV and 30mA) coupled to a APEX II CCD detector (equipped with a TRIUMPH graphite monochromator) with Mo-K<sub> $\alpha$ </sub> radiation ( $\lambda = 0.71073$  Å). High temperature (221K and 293K) diffraction data for 2a was collected on a Bruker AXS D8 VENTURE KAPPA diffractometer (50 kV and 1mA) coupled to a PHOTON 100 CMOS detector (equipped with Helios focusing multilayer mirror optics) with Mo-K<sub>a</sub> radiation ( $\lambda = 0.71073$  Å). Low-temperature diffraction data (q- and w-scans) for 2b and 2c were collected on a Bruker AXS D8 VENTURE KAPPA diffractometer coupled to a PHOTON 100 CMOS detector with Mo-K<sub>a</sub> radiation ( $\lambda = 0.71073$  Å) from an I<sub>u</sub>S micro-source. All diffractometer manipulations, including data collection, integration, and scaling were carried out using the Bruker APEXII software.<sup>4</sup> Absorption corrections were applied using SADABS.<sup>5</sup> Space groups were determined on the basis of systematic absences and intensity statistics. The structures were solved by direct methods using SHELXS or by intrinsic phasing using SHELXT<sup>6</sup>, and were refined against  $F^5$  on all data by full-matrix least squares with SHELXL-2014<sup>6</sup> using established refinement techniques.<sup>7</sup> All non-hydrogen atoms were refined anisotropically. Unless otherwise noted, all hydrogen atoms were included into the model at geometrically calculated positions and refined using a riding model. The isotropic displacement paramet67ers of all hydrogen atoms were fixed to 1.2 times the U value of the atoms they are linked to (1.5 times for methyl groups). Crystallographic data for 2a, 2b and 2c can be obtained free of charge from The Cambridge Crystallographic Data Centre (CCDC) via www.ccdc.cam.ac.uk/data\_request/cif under CCDC deposition numbers 1837963-1837974 Graphical representation of the structure with 50% probability thermal ellipsoids was generated using Mercury visualization software.<sup>8</sup>

	2a	2b	2c
	1837963,		
CCDC Number	1837964,	1837973	1837974
	1837965		
Empirical formula	$C_{37}H_{35}ClF_3FeO_4P_3PtS$	C <sub>36</sub> H <sub>35</sub> BClF <sub>4</sub> FeOP <sub>3</sub> Pt	$C_{33}H_{35}ClF_3FeO_8P_3PtS$
Formula weight	1012.01	949.75	1027.97
T (K)	100	100	100
Crystal system	Triclinic	Triclinic	Triclinic
Space group	P-1	P-1	P-1
a, Å	12.020(4)	10.6888(12)	9.1811(8)
b, Å	14.966(4)	11.6609(14)	12.1939(10)
c, Å	21.674(7)	14.5386(17)	16.6464(14)
α, °	74.877(11)	93.888(4)	87.574(4)
β, °	79.712(18)	102.803(4)	77.784(3)
γ, °	89.614(12)	97.548(4)	86.681(3)
Volume, Å <sup>3</sup>	3700(2)	1742.9(4)	1817.5(3)
Z	4	2	2
$d_{\text{calc}}, \text{g/cm}^3$	1.817	1.810	1.878
Abs. coeff. $(mm^{-1})$	4.482	4.691	4.572
$\theta$ range, $^{\circ}$	0.990 to 43.791	2.170 to 36.372	2.354 to 36.388
Abs. correction	Semi-empirical	Semi-empirical	Semi-empirical
GOF	1.007	1.181	1.020
$R_{I}^{a}, wR_{2}^{b}, [I > 2\sigma(I)]$	0.0262, 0.0493	0.0222, 0.0510	0.0339, 0.0591
${}^{a}\mathbf{R}_{1} = \Sigma \parallel$	$F_o - F_c  /\Sigma F_o $ . <sup>b</sup> wR <sub>2</sub> =	$= [\Sigma[w(F_o^2 - F_c^2)^2] / \Sigma[v]$	$w(F_0^2)^2]^{1/2}$ .

**Table 5.** Crystal and refinement data for compounds 2a, 2b and 2c



Figure 5. Structure of 2a with 50% probability anisotropic displacement ellipsoids. The second molecule of 2a and second triflate anion are not shown for clarity.

#### **Special Refinement Details for 2a**

Compound **2a** crystallizes in the triclinic space group P-1 with two complexes in the asymmetric unit. One triflate is disordered with a 78:22 ratio. The structure suggested a possible phase transition; after a variable temperature unit cell determination showed a doubling of the c-axis as the temperature decreases, additional data sets were collected at 221 and 293K on another crystal to see if the triflates would be ordered. The structure is still P-1 but with a Z' now of 1 instead of 2 as at 100K. However the sole triflate is still disordered, with a 56:44 ratio at 221K and a 54:46 ratio at 293K.



Figure 6. Structure of 2b with 50% probability anisotropic displacement ellipsoids.

## **Special Refinement Details for 2b**

Compound **2b** crystallizes in the triclinic space group P-1 with one molecule in the asymmetric unit. The hydrogen atom bound to O1 could not be located in the difference Fourier synthesis and was included into the model at geometrically calculated positions, and refined using a riding model.



Figure 7. Structure of 2c with 50% probability anisotropic displacement ellipsoids.

## **Special Refinement Details for 2c**

Compound **2c** crystallizes in the triclinic space group P-1 with one molecule in the asymmetric unit. The coordinates for the hydrogen atom bound to O5 was located in the difference Fourier synthesis and refined semi-freely with the help of a restraint on the O-H distance (0.84(4) Å).

## **10. References:**

<sup>1</sup>Collingwood, S. P.; Taylor, R. J. Synlett, **1998**, *3*, 283.

<sup>2</sup> These two ligands were made according to: Clark, J. S. K.; Voth, C. N.; Ferguson, M. J.; Stradiotto, M. *Organometallics* **2017**, *36*, 679.

<sup>3</sup>Zhang, W.; Wang, McCann, F. S. D.; Wang, D.; Chen, P.; Stahl, S. S.; Liu, G. *Science*, **2016**, *353*, 1014-1018.

<sup>4</sup> APEX2, Version 2 User Manual, M86-E01078, Bruker Analytical X-ray Systems, Madison, WI, **June 2006**.

<sup>5</sup> Sheldrick, G.M. "SADABS (version 2008/1): Program for Absorption Correction for Data from Area Detector Frames", University of Göttingen, **2008**.

<sup>6</sup> Sheldrick, G. Acta Crystallogr., Sect. A: Found. Crystallogr. 2008, 64, 112.

<sup>7</sup> Müller, P. Crystallogr. Rev. **2009**, 15, 57.

<sup>8</sup> Macrae, C. F.; Edgington, P. R.; McCabe, P.; Pidcock, E.; Shields, G. P.; Taylor, R.; Towler M.; Van de Streek, J. *J. Appl. Cryst.* **2006**, *39*, 453.



## CARBON01

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### PHOSPHORUS01





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Nucleus	19F
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Spectrometer	spect
Author	
Solvent	CDC13
Temperature	296.1
Pulse Sequence	zg30
Experiment	1D
Number of Scans	16
Receiver Gain	63
Relaxation Delay	1.0000
Pulse Width	10.7100
Acquisition Time	3.2768
Acquisition Date	2018-09-15T22:20:00
Modification Date	2018-09-18T06:48:09
Spectrometer Frequency	500.13
Spectral Width	10000.0
Lowest Frequency	-1911.5
Nucleus	1H
Acquired Size	32768
Spectral Size	65536











Parameter Title Comment Origin Owner Site	Value chenbo-X-cat 13-0915 chenbo-X-cat 13-0915 Bruker BioSpin GmbH nmr	135. 36 135. 36 134. 52 134. 52 134. 27 133. 64 133. 64 125. 62 125. 56 125. 56 125. 55 125. 55 125. 55 122. 07 122. 07	75.63 $75.54$ $75.54$ $75.44$ $75.01$ $74.95$	19.30 18.95 18.93
Spectrometer Author	spect			
Solvent Temperature Pulse Sequence Experiment Number of Scans Receiver Gain Relaxation Delay Pulse Width Acquisition Time Acquisition Date Modification Date Spectrometer Frequency Spectral Width Lowest Frequency Nucleus Acquired Size Spectral Size	CDC13 296.2 zgpg30 1D 4096 193 2.0000 9.6000 1.1010 2018-09-15T22:21:52 2018-09-18T06:48:09 y 125.76 29761.9 -2305.8 13C 32768 65536			$F_{3}C \rightarrow CF_{3} \rightarrow Me \rightarrow Me \rightarrow He \rightarrow He \rightarrow He \rightarrow He \rightarrow He \rightarrow He$
				U
210 200 190		150 140 130 120 110 100 f1 (ppm)	90 80 70	60 50 40 30 20 10 0 -10

Parameter Title Comment Origin Owner Site Spectrometer	R 5 සි පි 5 5 සේ ස් ස් ස් සේ සේ \\ch enbo-X-cat 13-0915 chenbo-X-cat 13-0915 Bruker BioSpin GmbH nmr	28. 90 28. 59 24. 74 24. 74 25. 67 22. 45 22. 45 22. 45 22. 45 22. 45 19. 50 19. 50 13. 26 6. 37	
Author	-		
Solvent	CDC13		F <sub>3</sub> C
Temperature	296.2		)CF₃
Pulse Sequence	zgpg30		
Experiment	1D		
Number of Scans	3096		Mo Mo
Receiver Gain	193		
Relaxation Delay	2.0000		+ P-We
Pulse Width	12.0000		E Pt I
Acquisition Time	0.4020		Fe / CI OH
Acquisition Date	2018-09-16T04:06:27		C'D-B
Modification Date	2018-09-18T06:48:09		
Spectrometer Frequ	iency 202.45		/=\ // `) OTf <sup>-</sup>
Spectral Width	81521.7		
Lowest Frequency	-50883.7		CE.
Nucleus	31P 22769		
Acquired Size	52706 65536		F <sub>3</sub> C
Spectral Size	00000		
			24
			Zu

<b>T</b>			• T	·	·	·	• T	· _	- T		·				<u> </u>	$\overline{}$			<u> </u>	<u>, , , , , , , , , , , , , , , , , , , </u>					
50	130	110	90	80	70	60	50	40	30	20	10	0	-20	-	-40	-60	-80	-100	-120	-140	-160	-180	-200	-220	-240
															f1	(ppm)								S46	

Demomenter	
Parameter	
litle	chenbo-X-2-cat 13#-0911
Comment	chenbo-X-2-cat 13#-0911
Origin	Bruker BioSpin GmbH
Owner	nmr
Site	
Spectrometer	AVANCE NEO 400 MHZ DIGITAL NMR SPECTROMETER
Author	
Solvent	CDC13
Temperature	296.6
Pulse Sequence	Zgig
Experiment	1D
Number of Scans	256
Receiver Gain	101
Relaxation Delay	1.0000
Pulse Width	18.0000
Acquisition Time	0.7209
Acquisition Date	2018-09-12T08:17:56
Modification Date	2018-09-12T11:25:26
Spectrometer Frequency	y 376. 50
Spectral Width	90909.1
Lowest Frequency	-83104.4
Nucleus	19F
Acquired Size	65536
Spectral Size	65536



2d

Т Т -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 f1 (ppm) 20 10 -10 -20 -30 -40 -50 -60 -200 -210 -22 0 -70 -80 -190S47

D	2604 2604 2604 2604 2604
Parameter	
Title	chenbo-X-2-cat 12#-K
Comment	chenbo-X-2-cat 12#
Origin	Bruker BioSpin GmbH
Owner	nmr
Site	
Spectrometer	AVANCE NEO 400 MHZ DIGITAL NMR SPECTROMETER
Author	
Solvent	CDC13
Temperature	295. 3
Pulse Sequence	zg30
Experiment	1D
Number of Scans	16
Receiver Gain	101
Relaxation Delay	1.0000
Pulse Width	10.0000
Acquisition Time	3. 9977
Acquisition Date	2018-09-06T12:54:19
Modification Date	2018-09-06T13:01:07
Spectrometer Frequency	400.13
Spectral Width	8196. 7
Lowest Frequency	-1636.9
Nucleus	1H
Acquired Size	32768
Spectral Size	65536



 ${\textstyle \nwarrow}^{1.\,88}_{1.\,85}$ 

 $\overbrace{}^{-4.1}_{-4.1}$ 





Parameter Title Comment Origin Owner Site	Value chenbo-X-cat 12#-0915 chenbo-X-cat 12-0915 Bruker BioSpin GmbH nmr	134.41         133.58         133.58         133.58         133.58         133.58         133.58         133.58         133.58         133.58         133.58         133.51         133.51         132.65         132.66         132.66         132.73         132.66         132.73         132.66         132.73         132.66         132.73         132.73         132.66         133.73         132.66         133.73         132.74         132.75         133.75         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41         123.41	76. 93 75. 87 75. 53 75. 23	₹19. 55 19. 25
Spectrometer Author	spect			
Solvent Temperature Pulse Sequence Experiment Number of Scans Receiver Gain Relaxation Delay Pulse Width Acquisition Time Acquisition Date Modification Date Spectrometer Frequency Spectral Width Lowest Frequency Nucleus Acquired Size Spectral Size	CDC13 296.1 zgpg30 1D 4096 193 2.0000 9.6000 1.1010 2018-09-16T07:48:19 2018-09-18T06:48:10 cy 125.77 29761.9 -2305.8 13C 32768 65536			$F_{3}C \rightarrow F_{3}CF_{3} \rightarrow F_{4} \rightarrow F_{6} \rightarrow F_{6} \rightarrow F_{6} \rightarrow F_{7} \rightarrow F_{7}$

fl (ppm)

Parameter Title Comment Origin Owner Site	窓 二 芝 冬 ら ð		
Spectrometer Author Solvent Temperature Pulse Sequence Experiment Number of Scans Receiver Gain Relaxation Delay Pulse Width Acquisition Time Acquisition Date Modification Date Spectrometer Freque	spect CDC13 296.2 zgpg30 1D 3096 193 2.0000 12.0000 0.4020 2018-09-16T09:57:42 2018-09-18T06:48:10 ency 202.45	F₃C →	CF <sub>3</sub> CF <sub>3</sub> CF <sub>3</sub> Me Pt CF <sub>3</sub> Me Pt OH CI OH OH
Spectral Width Lowest Frequency Nucleus Acquired Size Spectral Size	81521. 7 -50883. 7 31P 32768 65536	F <sub>3</sub> C	CF <sub>3</sub> CF <sub>3</sub> 2e

-40 -60 f1 (ppm) Т 90 80 70 60 50 40 30 20 10 0 -200 -220 -120 -140 -160 50 130 110 -20 -80 -100 -180 -240 S50

Paramatar	13.52 Value aufor
Titlo	chenho= $Y=2-cat$ 12#=0011
Commont	chenho-Y-2-cat 12#-0011
Origin	Reukon RioSpin (mbl
Origin	
Owner Site	
Site	
Spectrometer	AVANCE NEO 400 MHZ DIGITAL NMR SPECTROMETER
Author	
Solvent	CDC13
Temperature	296.2
Pulse Sequence	zgig
Experiment	1D
Number of Scans	256
Receiver Gain	101
Relaxation Delay	1.0000
Pulse Width	18.0000
Acquisition Time	0.7209
Acquisition Date	2018-09-12T02:38:25
Modification Date	2018-09-12T11:25:26
Spectrometer Frequency	376.50
Spectral Width	90909.1
Lowest Frequency	-83104.4
Nucleus	19F
Acquired Size	65536
Spectral Size	65536



<b>T</b>												· .											· · · · ·	T T
20	10	0	-10	-20	-30	-40	-50	-60	-70	-80	-90	-100	-110	-120	-130	-140	-150	-160	-170	-180	-190	-200	-210	-22
												fl (ppm	1)									S51		







CARBON01	76.47	42.97	30. 32 30. 18 28. 01	9.16	
xxy-vi-164	ī	Ī	Ϋ́Ζ		
Parameter	Value				
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-164/ CARBON01.fid/ fid				
Title	CARBONO1				
Origin	Varian				
Solvent	cdc13				
Temperature	25.0				
Pulse Sequence	s2pul				
Experiment	1D				
Probe	autox7991			NH-	
Number of Scans	1200				
Receiver Gain	30				
Relaxation Delay	1.0000			8	
Pulse Width	4. 6125			له	
Presaturation Frequency					
Acquisition Time	1.0420				
Acquisition Date	2017-05-21T20:56:13				
Spectrometer Frequency	125.65				
Spectral Width	31446. 5				
Lowest Frequency	-1903. 3				
Nucleus	13C		I		
Acquired Size	32768				
Spectral Size	65536				
	L				<u> </u>
· · · · · · · · · ·			· · · · ·		<del>г – – ,</del>
230 220 210	200 190 180 170 160	0 150 140	130 120	110 100 90 80 70 60 50 40 30 20 10 0 -1	10

120 110 100 f1 (ppm)



 $\sum_{5.97}^{6.05}$ 

1.99 1.99



#### PROTON01

## xy-vi-214

Parameter	Value
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/
	data/ xy-v1-214/ PROTONO1.fid/ fid
Title	PROTON01
Origin	Varian
Solvent	dmso
Temperature	25.0
Pulse Sequence	s2pu1
Experiment	1D
Probe	autox7991
Number of Scans	32
Receiver Gain	46
Relaxation Delay	1.0000
Pulse Width	5.8000
Presaturation Frequency	
Acquisition Time	3.0000
Acquisition Date	2017-07-04T11:32:55
Spectrometer Frequency	499.65
Spectral Width	8000.0
Lowest Frequency	-1002.1
Nucleus	1H
Acquired Size	24000
Spectral Size	65536





CARBON01	60	58	31	88 36 77
Parameter	Value 😸	39.	35.	29. 27. 22.
xy <del>Dáta</del> 2141e Name	/ Volumes/ nmrdata-¼/ xiangyou/ vnmrsys/ data/ xy- vi-214/ CARBON01.fid/ fid	ī	ī	VZ I
Title	CARBON01			
Origin	Varian			
Solvent	dmso			
Temperature	25.0			
Pulse Sequence	s2pul			
Experiment	1D			
Probe	autox7991			
Number of Scans	1500			
Receiver Gain	30			
Relaxation Delay	1.0000			
Pulse Width	4. 6125			
Presaturation Frequency				
Acquisition Time	1.0420			
Acquisition Date	2017-07-04T11:35:19			
Spectrometer Frequency	125.65			
Spectral Width	31446.5			
Lowest Frequency	-1903. 2			
Nucleus	13C			
Acquired Size	32768			
Spectral Size	65536			







Me

CONH2

4

\_1 S58

0

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12p

Parameter	Value								
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-177/ PROTONO1.fid/ fid								
Title	PROTON01								
Origin	Varian								
Solvent	dmso								
Temperature	25.0								
Pulse Sequence	s2pul								
Experiment	1D								
Probe	autox7991								
Number of Scans	32								
Receiver Gain	40								
Relaxation Delay	1.0000								
Pulse Width	5.8000								
Presaturation Frequency	7								
Acquisition Time	3. 0000							1	
Acquisition Date	2017-06-04T10:47:15								
Spectrometer Frequency	499.65								
Spectral Width	8000.0								
Lowest Frequency	-1002.1								
Nucleus	1H								
Acquired Size	24000								
Spectral Size	65536								
- <u>r</u>									
							J		
		i <del>, t</del> ri	ЬH					н	
		04 > 12 <	$^{91}_{-14}$					- 00	
		i? i	0 N					r. N	
13 19		8	- I 7	6	י ו ה	1	о Г	I	1 · 1 9 1
10 12	11 10 9	0	(	fl (nnm)	) )	4	ა		ے 1 د

CARBONO1 Parameter xxyDati-F77e Name	Value / Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data xxy-vi-177/ CARBON01.fid/ fid	 	<ul> <li>131.90</li> <li>129.16</li> <li>127.93</li> </ul>				
Title	CARBON01						
Origin	Varian						
Solvent	dmso						
Temperature	25.0						
Pulse Sequence	s2pul						
Experiment	1D						
Probe	autox7991						
Number of Scans	1200						
Receiver Gain	30						
Relaxation Delay	1.0000						
Pulse Width	4.6125				Me		
Presaturation					1		
Acquisition Time	1 0420			ĺ			
Acquisition Date	$2017 - 06 - 04T10 \cdot 49 \cdot 39$			ų			
Spectrometer	125.65				I		
Frequency	1				CONH <sub>2</sub>		
Spectral Width	31446.5		1.		12n		
Lowest Frequency	-1903.2				120		
Nucleus	13C						
Acquired Size	32768						
Spectral Size	65536						

120 110 100 f1 (ppm) 210 200 180 170 160 150 140 130 -10 S59

xxy-vi-199









RBON01	01.10	88. 88. 51 55. 56 57 55. 56 57 53. 31 53. 31 53. 31 54. 57 55. 56 57 57 57 57 57 57 57 57 57 57 57 57 57	
v-vi-199			
Parameter	Value		
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-199/ CARBON01.fid/ fid		
Title	CARBON01		
Origin	Varian		
Solvent	dmso		
Temperature	25.0		
Pulse Sequence	s2pul		
Experiment	1D		
Probe	autox7991		
Number of Scans	1200		
Receiver Gain	30		
Relaxation Delay	1.0000		
Pulse Width	4. 6125		
Presaturation Frequency			
Acquisition Time	1.0420		
Acquisition Date	2017-07-21T15:38:58		
Spectrometer Frequency	125.65		
Spectral Width	31446.5		
Lowest Frequency	-1903.2		
Nucleus	13C		
Acquired Size	32768		
Spectral Size	65536		

CONH<sub>2</sub> 13p

CF<sub>3</sub>



PROTON01

xxy-vi-170-1





CARBON01
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Parameter	Value
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-170-1/ CARBON01.fid/ fid
Title	CARBON01
Origin	Varian
Solvent	d2o
Temperature	25.0
Pulse Sequence	s2pul
Experiment	1D
Probe	autox7991
Number of Scans	1000
Receiver Gain	30
Relaxation Delay	1.0000
Pulse Width	4. 6125
Presaturation Frequency	
Acquisition Time	1.0420
Acquisition Date	2017-05-25T10:33:01
Spectrometer Frequency	125.65
Spectral Width	31446.5
Lowest Frequency	-1903.2
Nucleus	13C
Acquired Size	32768
Spectral Size	65536



14p

-	1		1 1		· · ·				'	·			'				·				· 1			1	- 1
23	0 22	20 2	210	200	190	180	170	160	150	140	130	120	110	100	90	80	70	60	50	40	30	20	10	0	-10
												1	l (ppm	1)										S63	

xxy-vi-162



Parameter	Value										
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy- vi-162/ PROTON01.fid/ fid										
Title	PROTON01										
Origin	Varian										
Solvent	dmso										
Temperature	25.0										
Pulse Sequence	s2pul										
Experiment	1D										
Probe	autox7991										
Number of Scans	16							K L CC			
Receiver Gain	40							0. 00			
Relaxation Delay	1.0000							15p			
Pulse Width	5.8000										
Presaturation Frequency	7										
Acquisition Time	3.0000										
Acquisition Date	2017-05-20T11:03:28										
Spectrometer Frequency	499.65										
Spectral Width	8000.0										
Lowest Frequency	-1002.1										
Nucleus	1H										
Acquired Size	24000										
Spectral Size	65536										
13 12	11 10 9	8 7	6 f1 (ppm)	5	4	3	2	1	0	-1 <b>S64</b>	-

CARBON01	84	45	$^{04}$
xxy-vi-162	— 159.		/ / / 114. / 112.

Data File Name       / Volumes/ num/star-1/ xxy-vri-162/ CAMBOOL fid/ File         Title       CABDOOL         Origin       Varian         Solvent       daso         Solvent       daso         Temperature       25.0         Pulse Sequence       Spul         Experiment       1D         Probe       mutox2991         Numbor of Scans       1500         Receiver diration       30         Relaxation Dulay       1.0000         Putse Width       4.0125         Presentation       1.020         Acquisition Thme       1.020         Acquisition Thme       1.0420         Acquisition Thme       1.020         Acquired Size	Parameter	Value	
Title       CARBOND         Origin       Varian         Solvent       dmso         Tomorature       25.0         Pulse Sequence       Sapel         Excoratent       10         Probe       mitox7991         Number of Seans       1500         Receiver Gain       30         Relaxation belay       1.0000         Presseturation       10         Programer       2017-06-2011:01:18         Spectral Tidth       31446.5         Lowest Frequency       1003.2         Nucleus       13C         Acquisition Time       1903.2         Nucleus       13C         Acquired Size       32768         Spectral Size       65306	Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-162/ CARBON01.fid/ fid	
Origin         Varian           Solvent         dmso           Temperature         25.0           Pulse Sequence         sepuil           Experiment         ID           Probe         autox7991           Number of Scans         1500           Reaceiver Game         30           Reaceiver Game         10000           Priseation Delay         1.0000           Presaturation         Frequency           Acquisition Time         1.020           Acquisition Date         2017-05-20711:04:48           Spectranetre         125.65           Frequency	Title	CARBON01	
Solvent       dmso         Temperature       25.0         Pulse Sequence       s2pul         Experiment       10         Probe       autox7991         Nubber of Scans       1500         Receiver Gain       30         Relaxation Delay       1.0000         Pulse Width       4.6125         Presentration	Origin	Varian	
Temperature       25.0         Pulse Sequence       s2pul         Experiment       10         Probe       autox7991         Number of Scans       1500         Receiver Gain       30         Relaxation Delay       1.0000         Pulse Width       4.6125         Presenturation       Presenturation         Prequency       -         Acquisition Date       2017-05-20111:04:48         Spectrometer       125.65         Prequency       -         Acquisition Date       31446.5         Lowest Frequency       -         Nucleus       13C         Acquired Size       2536	Solvent	dmso	
Pulse Sequence       s2pul         Experiment       10         Probe       autox7991         Number of Scans       1500         Receiver Gain       30         Relaxation Delay       1.0000         Pulse Width       4.6125         Presaturation	Temperature	25.0	
Experiment       ID         Probe       autox7991         Number of Scans       1500         Receiver Gain       30         Relaxation Delay       1.0000         Pulse Widh       4.6125         Presaturation       1         Frequency       1         Acquisition Date       2017-05-20T11:04:48         Spectrometer       125.65         Frequency       125.65         Spectral Width       31446.5         Lowest Frequency       13C         Acquisition Zate       22768         Spectral Size       65536	Pulse Sequence	s2pul	
Probe     autox7991       Number of Scans     1500       Receiver Gain     30       Relaxation Delay     1.0000       Pulse Width     4.6125       Presenturation     Trequency       Acquisition Time     1.0420       Acquisition Date     2017-05-20711:04:48       Spectrometer     125.65       Prequency     -       Spectral Width     31446.5       Lowest Frequency     -       Acquired Size     32768       Spectral Size     65336	Experiment	1D	
Number of Scans       150         Recavin Delay       1.0000         Pulse Width       4.6125         Presaturation	Probe	autox7991	
Receiver Gain     30       Relaxation Delay     1.0000       Pulse Width     4.6125       Presaturation     15p       Frequency     15p       Acquisition Time     1.0420       Acquisition Date     2017-05-20T11:04:48       Spectrometer     125.65       Frequency     1903.2       Spectral Width     31446.5       Lowest Frequency     13C       Acquired Size     32768       Spectral Size     65536	Number of Scans	1500	
Relaxation Delay 1.0000   Pulse Width 4.6125   Presaturation Image: Stress of the	Receiver Gain	30	
Pulse Width     4.6125       Presaturation     Frequency       Acquisition Time     1.0420       Acquisition Date     2017-05-20T11:04:48       Spectrometer     125.65       Frequency       Spectral Width     31446.5       Lowest Frequency     -1903.2       Nucleus     13C       Acquired Size     32768       Spectral Size     65536	Relaxation Delay	1.0000	CONH2
Presaturation 15p   Prequency 10420   Acquisition Date 2017-05-20T11:04:48   Spectrometer 125.65   Prequency -   Spectral Width 31446.5   Lowest Prequency -1903.2   Nucleus 13C   Acquired Size 32768   Spectral Size 65536	Pulse Width	4. 6125	0 2
Acquisition Time1. 0420Acquisition Date2017-05-20T11:04:48Spectrometer125. 65Frequency-Spectral Width31446. 5Lowest Frequency-1903. 2Nucleus13CAcquired Size32768Spectral Size65536	Presaturation Frequency		15p
Acquisition Date2017-05-20T11:04:48Spectrometer125.65Frequency125.65Spectral Width31446.5Lowest Frequency-1903.2Nucleus13CAcquired Size32768Spectral Size65536	Acquisition Time	1.0420	
Spectrometer Frequency125.65Spectral Width31446.5Lowest Frequency-1903.2Nucleus13CAcquired Size32768Spectral Size65536	Acquisition Date	2017-05-20T11:04:48	
Spectral Width31446.5Lowest Frequency-1903.2Nucleus13CAcquired Size32768Spectral Size65536	Spectrometer Frequency	125.65	
Lowest Frequency -1903.2 Nucleus 13C Acquired Size 32768 Spectral Size 65536	Spectral Width	31446. 5	
Nucleus 13C Acquired Size 32768 Spectral Size 65536	Lowest Frequency	-1903. 2	
Acquired Size 32768 Spectral Size 65536	Nucleus	13C	
Spectral Size 65536	Acquired Size	32768	
	Spectral Size	65536	
	a ya a sa kata na kata na kata na kata na kata na kata kat	$\mathcal{M}$	Λ

120 110 100 f1 (ppm) 10 0 180 170 -10 S65

xxy-vi-161



Parameter	Value								
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy- vi-161/ PROTONO1.fid/ fid								
Title	PROTONO1								
Origin	Varian								
Solvent	dmso								
Temperature	25.0								
Pulse Sequence	s2pul								
Experiment	1D								
Probe	autox7991								
Number of Scans	16								
Receiver Gain	32							CONIL	
Relaxation Delay	1.0000							CONF	2
Pulse Width	5.8000								
Presaturation Frequency	y							s	
Acquisition Time	3.0000								
Acquisition Date	2017-05-20T10:58:55							16p	
Spectrometer Frequency	499.65								
Spectral Width	8000.0								
Lowest Frequency	-1002.1								
Nucleus	1H								
Acquired Size	24000								
Spectral Size	65536								
		d .							
		日 1 1 1 1 1 1 1 1 1 1 1 1 1							
4 13 12	11 10 9	8 7	6	5	4	3	2	1 0	-]
			fl (ppm)						S66

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Parameter	Value			
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ x. vi-161/ CARBON01.fid/ fid	xy-		
Title	CARBON01			
Origin	Varian			
Solvent	dmso			
Temperature	25.0			
Pulse Sequence	s2pu1			
Experiment	1D			
Probe	autox7991			
Number of Scans	1500			
Receiver Gain	30		,CONH₂	
Relaxation Delay	1.0000		IT IS	
Pulse Width	4.6125			
Presaturation Frequency	Į.		S	
Acquisition Time	1.0420	I.	16p	
Acquisition Date	2017-05-20T10:07:06			
Spectrometer Frequency	125.65			
Spectral Width	31446.5			
Lowest Frequency	-1903.2			
Nucleus	13C			
Acquired Size	32768			
Spectral Size	65536			

120 110 100 f1 (ppm) 180 170 140 130 -10

Parameter	Value	7 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	2.21
Title	chenbo-X-2-106		
Comment	chenbo-X-2-106		
Origin	Bruker BioSpin GmbH		
Owner	nmr		
Site			
Spectrometer	AVANCE NEO 400 MHZ DIGITAL	NMR SPECTROMETER	
Author			
Solvent	CDC13		
Temperature	294. 9		
Pulse Sequence	zg30		
Experiment	1D		$\wedge \wedge$
Number of Scans	16		ſΎ
Receiver Gain	101		
Relaxation Delay	1.0000		$\checkmark$
Pulse Width	10.0000		
Acquisition Time	3. 9977		17n
Acquisition Date	2018-08-30T08:08:02		11 P
Modification Date	2018-08-30T08:22:20		
Spectrometer Freque	ncy 400.13		
Spectral Width	8196.7		
Lowest Frequency	-1636.4		
Nucleus	1H		
Acquired Size	32768		
Spectral Size	65536		



OAc

n

NH<sub>2</sub>



Parameter	Value 0 0 0 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	22 22	
Title	1cc-080913-2		
Comment			
Origin	Bruker BioSpin GmbH		
Owner	nmr		
Site			
Spectrometer	AVANCE NEO 400 MHZ DIGITAL NMR SPECTROMETER		
Author			
Solvent	CDC13	OAc	
Temperature	295.0		
Pulse Sequence	zg30		
Experiment	1D		
Number of Scans	8		
Receiver Gain	80	ίŇ	
Relaxation Delay	1.0000	[ ]	
Pulse Width	10.0000	$\checkmark$	
Acquisition Time	3. 9977		
Acquisition Date	2018-09-13T20:42:27	S-18	
Modification Date	2018-09-13T20:45:49		
Spectrometer Freque	ncy 400.13		
Spectral Width	8196.7		
Lowest Frequency	-1638.1		
Nucleus	1H		
Acquired Size	32768		
Spectral Size	65536		
•			
	/// L /		
		н	
		Т 00 ю	
			,
16 15 14	13 12 11 10 9 8 7 6 5 f1 (npm)	4 3 2 1 0 -1 -2 -3	5 -
	II (hhm)	S70	

Parameter Title Comment	≋ Value 1cc-080913-2	→ 132. 24 → 130. 07 → 128. 55	 	 
Origin Owner Site	Bruker BioSpin GmbH nmr			
Spectrometer Author	AVANCE NEO 400 MHZ DIGITAL NMR SPEC	TROMETER		OAc
Solvent Temperature Pulse Sequence Experiment Number of Scans Receiver Gain Relaxation Delay Pulse Width Acquisition Time Acquisition Date Modification Date Spectrometer Frequency Spectral Width Lowest Frequency Nucleus Acquired Size Spectral Size	CDC13 295.7 zgpg30 1D 84 32 2.0000 10.0000 1.3763 2018-09-13T20:48:13 2018-09-13T20:45:49 Cy 100.61 23809.5 -1843.5 13C 32768 32768			S-18
			1	

⊤ 150 110 100 90 f1 (ppm) 170 160 -10






Parameter Title Comment Origin Owner Site Spectrometer Author Solvent Temperature Pulse Sequence Experiment Number of Scans Receiver Gain Relaxation Delay Pulse Width Acquisition Time Acquisition Date Spectrometer Frequence Spectral Width Lowest Frequency Nucleus Acquired Size Spectral Size	<sup>1</sup> / <sub>2</sub> <sup>2</sup> / <sub>3</sub> <sup>1</sup> / <sub>3</sub>	$   \begin{array}{l}                                     $
nen beste bester dit vijnen han bester bled de statutet bester stierte nen plate bester og vijnen per per per per son de stierte		t de zoektij kronnen bie bied te new de biet en gewennen die de ster te de ster de ster de ster de ster de ste Program te program te vergen en boure geget op de ster de biet en over te die ster begen oorde bie en program e
210 200 190	180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 f1 (ppm)	) 30 20 10 0 -10 <b>\$75</b>

xxy-vi-175

Parameter	Value										
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-175/ PROTON01.fid/ fid										
Title	PROTONO1										
Origin	Varian										
Solvent	cd2c12										
Temperature	25.0										
Pulse Sequence	s2pu1										
Experiment	1D										
Probe	autox7991										
Number of Scans	32										
Receiver Gain	44							( )	—Ме		
Relaxation Delay	1.0000							$\searrow$			
Pulse Width	5. 8000							ć	CONH <sub>2</sub>		
Presaturation Frequency								20p	4		
Acquisition Time	3.0000										
Acquisition Date	2017-05-28T20:18:59										
Spectrometer Freque	ncy 499.65										
Spectral Width	8000.0										
Lowest Frequency	-1002.1	1									
Nucleus	1H										
Acquired Size	24000										
Spectral Size	65536		1								
			M				u				
		0.91 0.92 1.89	2.33			Ц С	- <b>1</b> }				
13 12	11 10 9	8 7	6 f1 (ppm)	5	4	3	2	1	0	-1 <b>S76</b>	- <u>`</u>

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)1	1. 53	5. 29 5. 38 1. 06 5. 84 5. 61
-175	— 17	
Parameter	Value	
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy- vi-175/ CARBON01.fid/ fid	
Title	CARBON01	
Origin	Varian	
Solvent	cd2c12	
Temperature	25.0	
Pulse Sequence	s2pul	
Experiment	1D	
Probe	autox7991	
Number of Scans	1200	
Receiver Gain	30	
Relaxation Delay	1.0000	
Pulse Width	4. 6125	
Presaturation Frequency		
Acquisition Time	1.0420	i n
Acquisition Date	2017-05-28T20:21:24	
Spectrometer Frequency	125.65	
Spectral Width	31446.5	
Lowest Frequency	-1903.3	
Nucleus	13C	
Acquired Size	32768	
Spectral Size	65536	

Me CONH<sub>2</sub> 20p

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— 19.67

110 100 f1 (ppm) -10 

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# P\$33332 xxy-vi-165-1



CARBON01	. 65	. 51	. 51	0 0 0 2 2 2 2 0 4 4 4 4 4 4 4 4 4 4 4 4
xxy-vi-165-1			— 129 — 126	
Parameter Data File Name Title Origin Solvent Temperature Pulse Sequence Experiment	Value / Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-165-1/ CARBON01.fid/ fid CARBON01 Varian cd2c12 25.0 s2pu1 1D			
Probe Number of Scans Receiver Gain Relaxation Delay Pulse Width Presaturation Frequency Acquisition Time Acquisition Date Spectrometer Frequency Spectral Width Lowest Frequency Nucleus Acquired Size Spectral Size	autox7991 1200 30 1.0000 4.6125 1.0420 2017-05-22T21:55:49 125.65 31446.5 -1903.3 13C 32768 65536			Me CONH2 21p

120 110 100 f1 (ppm) 210 200 180 170 160 150 140 130 -10

PROTON01

### xxy-vi-170

$\begin{array}{c} 0.8\\ 0.8\\ 0.6\\ 0.6\\ 0.6\\ 0.6\\ 0.6\\ 0.6\\ 0.6\\ 0.6$	74 74 73 73 73

Parameter	Value					
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-170/ PROTONO1.fid/ fid					
Title	PROTONO1					
Origin	Varian					
Solvent	cd2c12					
Temperature	25.0				1	
Pulse Sequence	s2pu1					
Experiment	1D					
Probe	autox7991					
Number of Scans	16					
Receiver Gain	46					
Relaxation Delay	1.0000					
Pulse Width	5.8000					
Presaturation Frequenc	y					
Acquisition Time	3, 0000					CONH
Acquisition Date	2017-05-24T23:20:16					1
Spectrometer Frequency	499.65					
Spectral Width	8000.0					
Lowest Frequency	-1002.1					
Nucleus	1H					22p
Acquired Size	24000					
Spectral Size	65536					
-						
			1			
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			21-		23 ( ) (000)	
			~i			
4 13 12	11 10 9	8 7	6 5	4 3	2 1	0 -1 -
_ 10 18		- ·	fl (ppm)		- 1	S80
						000

). 37				46 25 41	25
				- 40. - 39.	
Value / Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xy-vi-170/ CARBON01.fid/ fid CARBON01 Varian cd2c12 25.0 32pu1 10 autox7991 1000 30 1.0000 30 1.0000 4.6125 1.0420 2017-05-24T23:21:36 125.65 31446.5 -1903.3 13C 32768 65536					CONH2 22p
	Value / Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-170/ CARBONO1. fid/ fid CARBONO1 Varian cd2c12 25.0 s2pu1 1D autox7991 1000 30 1.0000 4.6125 1.0420 2017-05-24T23:21:36 125.65 31446.5 -1903.3 13C 32768 65536	Value / Volumes/ nmrdata-1/ xiangyou/vnmrsys/ data/ xxy-ri-170/ CARBONO1.fid/ fid CARBONO1 Varian cd2c12 25.0 s2pu1 10 autox7991 1000 30 1.0000 4.6125 1.0420 2017-05-24T23:21:36 125.65 31446.5 -1903.3 132 32768 65536	Value / Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-1200 (ARBONOL fid/ fid CARBONOL Varian ed2cl2 25.0 32pal 10 autor7991 100 30 1.0000 4.6125 1.0420 2017-05-24T23:21:36 125.6 31446.5 -1903.3 132 32768 65536	F           Value           / Voluess/ murdata.l/ xiamgou/vurmssd/data/ xxy/vill0/CARDONOL.rid/ fd           CARDONI           Value           Value           Sapul           Jb           autox7991           1000           30           1.000           30           1.000           31           2017-05-24123:21:36           125           31446.5           -1903.3           32768           65536	Fig         Value         Valuesyood/vmm;sts/data/ xxx=vi=170/CCABBOND.fid/ fid         CARROND.1fid/ fid         Values         varian         cd2e12         25.0         s2pul         10         autox7591         1000         30         1.020         2017 05 24723:21:36         125.65         31446.5         -1909.3         302         22768         66366

180 170 160 150 140 130 120 110 100 90 f1 (ppm) -10

S81

xxy-vi-173



Parameter	Value
Data File Name	/ Volumes/ nmrdata-1/
	xiangyou/ vnmrsys/
	PROTONO1.fid/ fid
Title	PROTON01
Origin	Varian
Solvent	d2o
Temperature	25. 0
Pulse Sequence	s2pu1
Experiment	1D
Probe	autox7991
Number of Scans	32
Receiver Gain	52
Relaxation Delay	1.0000
Pulse Width	5.8000
Presaturation	
Frequency	
Acquisition Time	3.0000
Acquisition Date	2017-05-27T22:27:28
Spectrometer	499.65
Frequency	
Spectral Width	8000.0
Lowest Frequency	-1002.1
Nucleus	1H
Acquired Size	24000
Spectral Size	65536





CARBON01		0. 43	1. 65 7. 46	5. 26	9. 03	6 5 +
xxy-vi-173		- 17(	- 151	- 13(	- 129	
Parameter	Value	I	1 1	I	I	
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-173/ CARBON01.fid/ fid					
Title	CARBON01					
Origin	Varian					
Solvent	d2o					
Temperature	25.0					
Pulse Sequence	s2pu1					
Experiment	1D					
Probe	autox7991					
Number of Scans	1200					
Receiver Gain	30					
Relaxation Delay	1.0000					
Pulse Width Presaturation Frequency	4. 6125					
Acquisition Time	1.0420					CONH <sub>2</sub>
Acquisition Date	2017-05-27T22:29:52					
Spectrometer Frequency	125.65			1		23p
Spectral Width	31446.5					
Lowest Frequency	-1903. 2					
Nucleus	13C					
Acquired Size	32768					
Spectral Size	65536				ł	
230 220 210	200 190 180	170 160	150 14	0	130	120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm) \$83

S83









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CARBON01		60	0
xxy-vi-153		2	— 28.
Parameter	Value		
Data File Name / Y xi xx fi	Volumes/ nmrdata-1/ iangyou/ vnmrsys/ data/ xy-vi-153/ CARBON01.fid/ id		
Title CA	ARBON01		
Origin Va	arian		
Solvent dm	NSO		
Temperature 25	5. 0		
Pulse Sequence s2	2pul		
Experiment 1D	D		
Probe au	utox7991		
Number of Scans 15	500		
Receiver Gain 30	C		
Relaxation Delay 1.	0000		
Pulse Width 4.	6125		
Presaturation Frequency			
Acquisition Time 1.	0420		
Acquisition Date 20	D17-07-18T17:44:52		
Spectrometer 12 Frequency	25.65		
Spectral Width 31	1446. 5		0
Lowest Frequency -1	1903. 2		но
Nucleus 13	3C		Me NH2
Acquired Size 32	2768		Me
Spectral Size 65	5536		10
		in a superior of the second	

120 110 100 f1 (ppm) 210 200 180 170 160 150 140 130 -10

### PROTON01

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xxy-vi-mandeloamide

Parameter	Value								
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-mandeloamide/ PROTONO1.fid/ fid								
Title	PROTONO1								
Origin	Varian								
Solvent	dmso								
Temperature	25.0								
Pulse Sequence	s2pul								
Experiment	1D								
Probe	autox7991								
Number of Scans	16								
Receiver Gain	50								
Relaxation Delay	1.0000								/
Pulse Width	5.8000								
Presaturation Frequency									
Acquisition Time	3.0000								
Acquisition Date	2017-05-08T18:49:21								
Spectrometer Frequency	499.65								
Spectral Width	8000.0								
Lowest Frequency	-1002.1								
Nucleus	1H								
Acquired Size	24000								
Spectral Size	65536								
							1		
			2. 70 1. 92 0. 94	0. 89 <b>म</b>	1. 05				
13 1	2 11 10	9	8 7	6 f1 (ppm)	5	4	3	2	1





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### xxy-vi-197

Parameter	Value			
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy- vi-197/ PROTONO1.fid/ fid			
Title	PROTON01			
Origin	Varian			
Solvent	d2o			
Temperature	25.0			
Pulse Sequence	s2pul			
Experiment	1D			
Probe	autox7991			
Number of Scans	32			
Receiver Gain	48			
Relaxation Delay	1.0000			C→ OH
Pulse Width	5.8000			$\langle \chi^{\circ}$
Presaturation Frequency	7		· · · · · · · · · · · · · · · · · · ·	CONH <sub>2</sub>
Acquisition Time	3.0000			
Acquisition Date	2017-07-04T10:36:03			Зр
Spectrometer Frequency	499.65			
Spectral Width	8000.0			
Lowest Frequency	-1002.1			
Nucleus	1H			
Acquired Size	24000			
Spectral Size	65536			
F F F F F F F F F F F F F F F F F F F				
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			J	/////M
			۲ بر	
			~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	9:5 1. C
13 12	11 10 9 8	7 6 5	4 3 2	1 0 -1
		f1 (ppm)		S90

CARBON01	.48	8	92	43 30
xxy-vi-197	— 183	74.	32.	<u> </u>
Parameter	Value			
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-197/ CARBONO1 fid/ fid			

	011100101.110/ 110
Title	CARBON01
Origin	Varian
Solvent	d2o
Temperature	25.0
Pulse Sequence	s2pu1
Experiment	1D
Probe	autox7991
Number of Scans	1500
Receiver Gain	30
Relaxation Delay	1.0000
Pulse Width	4.6125
Presaturation Frequency	
Acquisition Time	1.0420
Acquisition Date	2017-07-04T10:38:28
Spectrometer Frequency	125.65
Spectral Width	31446.5
Lowest Frequency	-1903.2
Nucleus	13C
Acquired Size	32768
Spectral Size	65536



120 110 100 f1 (ppm) -10 

S91

PROTON01

### xxy-vi-178

Parameter	Value	
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-178/ PROTON01.fid/ fid	
Title	PROTONO1	
Origin	Varian	
Solvent	d2o	
Temperature	25.0	
Pulse Sequence	s2pul	
Experiment	1D	
Probe	autox7991	
Number of Scans	32	
Receiver Gain	36	
Relaxation Delay	1.0000	
Pulse Width	5. 8000	
Presaturation Frequency		
Acquisition Time	3.0000	
Acquisition Date	2017-06-08T13:22:02	HO CONH2
Spectrometer Frequency	499. 65	40
Spectral Width	8000. 0	46
Lowest Frequency	-1002. 1	
Nucleus	1H	
Acquired Size	24000	
Spectral Size	65536	
		t-l
		L-4
		2 00 
13 12	11   10   9   8   7   6   5	4 3 2 1 0 -1

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L . A	KB	1 1 1	
UI1			

Parameter	Value
Data File Name	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-178/ CARBON01.fid/ fid
Title	CARBON01
Origin	Varian
Solvent	d2o
Temperature	25.0
Pulse Sequence	s2pul
Experiment	1D
Probe	autox7991
Number of Scans	1200
Receiver Gain	30
Relaxation Delay	1.0000
Pulse Width	4. 6125
Presaturation Frequency	
Acquisition Time	1.0420
Acquisition Date	2017-06-08T13:24:27
Spectrometer Frequency	125.65
Spectral Width	31446.5
Lowest Frequency	-1903.2
Nucleus	13C
Acquired Size	32768
Spectral Size	65536

HO-CONH2

-60.51

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230	220	210	200	190	180	170	160	150	140	130	120	110	100	90	80	70	60	50	40	30	20	10	0	-10
											t	fl (ppm	)									SOS	2	



 $\sim$  7.12  $\sim$  7.06  $<_{5.32}^{5.33}$ 

 $<^{1.\,22}_{1.\,20}$ 

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1/1	11	1 3/ 7	1 8 9		

### xxy-vi-160

— 177. 48

S95

-10

Data File Name Title Origin	/ Volumes/ nmrdata-1/ xiangyou/ vnmrsys/ data/ xxy-vi-160/ CARBON01.fid/ fid CARBON01 Varian dmso		
Title Origin	CARBON01 Varian dmso		
Origin	Varian dmso		
~ + + 0 + + +	dmso		
Solvent			
Temperature	25.0		
Pulse Sequence	s2pul		
Experiment	1D		
Probe	autox7991		
Number of Scans	1500		
Receiver Gain	30		
Relaxation Delay	1.0000		
Pulse Width	4. 6125		
Presaturation Frequency			Me-
Acquisition Time	1.0420		3
Acquisition Date	2017-05-20T09:12:52		ं
Spectrometer Frequency	125.65		
Spectral Width	31446.5		
Lowest Frequency	-1903.2		
Nucleus	13C		
Acquired Size	32768		
Spectral Size	65536		
			L I

120 110 100 f1 (ppm)