Supporting Information:

- 1. Experimental data with stabilized enolates and cyclohexenyl substrates
- 2. Non-linear experiments
- 3. Kinetic data
- 4. Calculated Data
- 5. Crystal data

1. Experimental data with stabilized enolates:

General Procedure for Preparation of Allyl Enol Carbonates:



To a solution of potassium *t*-butoxide (5.88 g, 52.5 mmol, 1.05 equiv) in DMF (100 mL) was added 2-methylcyclohexanone (6.13 mL, 50 mmol, 1.0 equiv). After 12 h, the reaction mixture was cooled in an ice bath and allyl chloroformate (6.4 mL, 60 mmol, 1.2 equiv) was added in a dropwise fashion. After an additional 30 min in the ice bath and 15 min at 25 °C, the reaction mixture was quenched into water (250 mL), extracted with DCM/hexanes 2/1 (4 x 125 mL), dried (MgSO₄), and evaporated. Chromatography (2.5 to 4 % Et₂O in Hexanes on SiO₂) afforded the allyl enol carbonate¹ (4.49 g, 46% yield) as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 5.94 (ddt, *J* = 17.4, 10.5, 5.6 Hz, 1H), 5.36 (dq, *J* = 17.1, 1.5 Hz, 1H), 5.26 (dq, *J* = 10.2, 1.2 Hz, 1H), 4.63 (dt, *J* = 5.7, 1.4 Hz, 2H), 2.13 (m, 2H), 2.02 (m, 2H), 1.70 (m, 2H), 1.59 (m, 2H), 1.55 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.1, 142.2, 131.5, 120.8, 118.8, 68.5, 30.0, 26.6, 23.1, 22.3, 15.7; IR (Neat Film NaCl) 3936, 1755, 1275, 1239, 1037 cm⁻¹; HRMS *m/z* calc'd for C₁₁H₁₆O₃ [M]⁺: 196.1100, found 196.1092.

General Procedure for 1.0 mmol Preparative Reactions of Allyl Enol Carbonates:



A 50 mL rb flask equipped with a magnetic stir bar was flame dried under vacuum. After cooling under dry argon, Pd₂(dba)₃ (22.9 mg, 0.025 mmol, 0.025 equiv) and (*S*)-*t*-Bu-PHOX (24.2 mg, 0.0625 mmol, 0.0625 equiv) were added. After the flask was flushed with argon, THF (30 mL) was added and the contents were stirred at 25 °C for 30 min, at which time the allyl enol carbonate (196.2 mg, 1.0 mmol, 1.0 equiv) was added by syringe in one portion. When the reaction was complete by TLC, the reaction mixture was evaporated under reduced pressure and the residue chromatographed (2 to 3 % Et₂O in pentane on SiO₂) to afford (*S*)-2-allyl-2-methylcyclohexanone (129.6 mg, 85.1% yield) as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 5.75-5.61 (m, 1H), 5.05 (s, 1H), 5.01 (m, 1H), 2.40-2.31 (m, 3H), 2.21 (dd, *J* = 13.8, 7.5 Hz, 1H), 1.78 (m, 5H), 1.56 (m, 1H), 1.06 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 215.4, 133.7, 117.9, 48.4, 41.9, 38.8, 38.5,

27.4, 22.6, 21.0; IR (Neat Film NaCl) 2934, 2865, 1707, 1451, 912 cm⁻¹; HRMS m/z calc'd for C₁₀H₁₆O [M]⁺: 152.1201, found 152.1204; [α]D²⁸ -22.90° (c 2.09, hexane, 98% ee).

Asymmetric Allylation with Stabilized Enolates

Table SI1 summarizes results obtained with stabilized enolates with low pK_a values relative to the typical ketone enolates employed in these reactions. Despite the low levels of enantioselectivity, chemical yield is very high. This may imply that an alternate mechanism is accessible with this low pK_a substrates (e.g., an outer sphere mechanism).

Table SI1. Asymmetric Allylation via Stabilized Allyl Enol Carbonates.



^{*a*} Reactions were performed using 1.0 mmol of substrate in THF (0.033 M in substrate) at 25 °C with 2.5 mol% $Pd_2(dba)_3$ and 6.25 mol% (*S*)-*t*-BuPHOX. ^{*b*} Absolute stereochemistry of products assigned by analogy. ^{*c*} Isolated yields. ^{*d*} Measured by chiral GC or HPLC.

Characterization data for compounds in Table SI1:

Products shown in entries 1, 2, and 3 have been reported previously.²



Table SI1, Entry 1: Prepared by the general procedure in 43% yield as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 7.35-7.18 (comp. m, 5H), 5.80 (ddt, *J* = 17.4, 10.5, 5.4 Hz, 1H), 5.20 (ddt, *J* = 17.4, 1.8, 1.2 Hz, 1H), 5.18 (ddt, *J* = 10.5, 1.5, 1.2 Hz, 1H), 5.02 (ddd, *J* = 5.7, 1.5, 1.5 Hz, 2H), 2.46-2.38 (m, 2H), 2.37-2.30 (m, 2H), 1.90-1.72 (comp. m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 152.8, 143.4, 138.8, 131.3, 128.1, 127.6, 126.8, 125.9, 118.4, 68.4, 30.1, 27.1, 22.8, 22.5; IR (Neat Film NaCl) 3081, 3057, 3024, 2938, 2862, 1753, 1687, 1601, 1492, 1444, 1367, 1238, 1178, 1091, 1036, 941, 784, 760, 700 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₆H₁₈O₃ [M]⁺: 258.1256, found 258.1256.



Table SI1, Entry 2: Prepared by the general procedure in 59% yield as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 7.29-7.11 (comp. m, 4H), 6.00 (ddt, *J* = 17.1, 10.2, 6.0 Hz, 1H), 5.43 (ddt, *J* = 17.1, 1.8, 1.2 Hz, 1H), 5.33 (ddt, *J* = 10.2, 1.5, 1.2 Hz, 1H), 4.72 (ddd, *J* = 6.0, 1.5, 1.2 Hz, 2H), 2.97 (t, *J* = 7.8 Hz, 2H), 2.55 (tq, *J* = 8.1, 1.5 Hz, 2H), 2.00 (t, *J* = 1.5 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 152.7, 145.9, 135.0, 134.1, 131.2, 127.1, 126.6, 126.5, 123.4, 119.7, 119.2, 68.9, 28.7, 26.0, 10.9; IR (Neat Film NaCl) 3021, 2993, 2944, 2891, 2836, 1757, 1674, 1488, 1451, 1365, 1304, 1279, 1246, 1217, 1181, 1157, 1031, 1018, 986, 943, 782, 760 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₁₅H₁₆O₃ [M]⁺: 244.1100, found 244.1095.



Table SI1, Entry 3: Prepared by the general procedure in 78% yield as a light yellow oil; ¹H NMR (300 MHz, CDCl₃) δ 5.96 (ddt, J = 17.1, 10.5, 5.7 Hz, 1H), 5.39 (ddt, J = 17.1, 1.5, 1.5 Hz, 1H), 5.28 (ddt, J = 10.5, 1.5, 1.2 Hz, 1H), 4.67 (ddd, J = 5.7, 1.2, 1.2 Hz, 2H), 4.15 (q, J = 7.2 Hz, 2H), 2.44-2.34 (m, 2H), 2.32-2.24 (m, 2H), 1.80-1.58 (comp. m, 4H), 1.24 (t, J = 7.2 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.6, 154.8, 152.2, 131.2, 119.2, 118.3, 69.0, 60.5, 28.6, 25.1, 21.9, 21.5, 14.0; IR (Neat Film NaCl) 3087, 1983, 2942, 2866, 1760, 1715, 1666, 1449, 1368, 1233, 1189, 1081, 1056, 1035, 994, 946, 767 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₃H₁₈O₅ [M]⁺: 254.1154, found 254.1153.

Table SI1, Entry 4: Prepared by a modification of the general procedure using TEA as the base and THF as solvent in 79% yield as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 5.95 (ddt, J = 17.1, 10.2, 6.0 Hz, 1H), 4.93 (ddt, J = 17.1, 2.7, 1.2 Hz, 1H), 5.37 (ddt, J = 10.2, 2.1, 0.9 Hz, 1H), 5.07 (q, J = 1.8 Hz, 2H), 4.74 (ddd, J = 6.0, 1.2, 0.9 Hz, 2H), 1.81 (t, J = 1.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 172.6, 163.4, 150.2, 130.0, 120.8, 109.4, 70.3, 67.5, 6.9; IR (Neat Film NaCl) 3089, 2958, 2931, 1774, 1702, 1446, 1392, 1360, 1330, 1240, 1132, 1079, 1025, 945, 889, 775, 754 cm⁻¹; HRMS (EI+) m/z calc'd for C₉H₁₁O₅ [M + H]⁺: 199.0606, found 199.0600.



Table SI1, Entry 4: Prepared by the general procedure. Purified by flash chromatography (SiO₂, 2 \rightarrow 12% EtOAc in hexanes). 87% yield, 2% ee. $R_f = 0.20$ (10%)

EtOAc in hexanes); ¹H NMR (300 MHz, CDCl₃) δ 5.62 (dddd, J = 17.7, 9.6, 7.5, 7.2 Hz, 1H), 5.13 (app. ddd, J = 9.6, 1.8, 0.9 Hz, 1H), 5.12 (app. ddd, J = 17.1, 1.5, 0.9 Hz, 1H), 4.59 (d, J = 17.1 Hz, 1H), 4.44 (d, J = 17.4 Hz, 1H), 2.53-2.37 (m, 2H), 1.28 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 209.9, 176.6, 130.1, 121.1, 72.5, 45.6, 40.2, 19.0; IR (Neat Film NaCl) 3543, 3083, 2983, 2939, 2877, 1803, 1758, 1642, 1454, 1436, 1378, 1341, 1231, 1122, 1065, 1043, 998, 912, 664 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₈H₁₀O₃ [M]⁺: 154.0630, found 154.0626.



Table SI1, Entry 5: Prepared by a modification of Leplawy's procedure³ in 96% yield as a colorless oil that solidifies on standing; Mp 37.5-39 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.97-7.90 (m, 2H), 7.45-7.38 (comp. m, 3H), 5.99 (ddt, *J* = 17.4, 10.5, 5.7 Hz, 1H), 5.45 (ddt, *J* = 17.4, 1.5, 1.2 Hz, 1H), 5.36 (ddt, *J* = 10.5, 1.2, 1.2 Hz, 1H), 4.78 (ddd, *J* = 6.0, 1.2, 1.2 Hz, 2H), 2.14 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 155.3, 155.0, 151.6, 146.2, 130.5 (2C), 128.9, 127.3, 126.1, 120.6, 70.7, 10.5; IR (Neat Film NaCl) 3066, 2930, 1786, 1669, 1554, 1490, 1450, 1367, 1213, 1082, 1069, 1026, 992, 939, 774, 711, 692 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₁₄H₁₃O₄N [M]⁺: 259.0845, found 259.0855.



Table SI1, Entry 5: Prepared by the general procedure. Purified by flash chromatography (SiO₂, 4 \rightarrow 7% Et₂O in hexanes). 89% yield, 2% ee. $R_f = 0.39$ (25% Et₂O in hexanes); ¹H NMR (300 MHz, CDCl₃) δ 7.99 (ddd, J = 7.2, 1.5, 1.2 Hz, 2H), 7.57 (tt, J = 7.8, 1.2 Hz, 1H), 7.48 (ddd, J = 7.8, 6.9, 1.5 Hz, 2H), 5.67 (dddd, J = 17.1, 9.9, 7.5, 6.9 Hz, 1H), 5.18 (dddd, J = 17.1, 1.5, 1.5 Hz, 1H), 5.11 (dddd, J = 10.2, 1.5, 0.9, 0.9 Hz, 1H), 2.64 (dddd, J = 13.8, 6.9, 0.9, 0.9 Hz, 1H), 2.57 (dddd, J = 13.8, 7.5, 1.2, 1.2 Hz, 1H), 1.53 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 180.2, 159.8, 132.7, 130.8, 128.7, 127.9, 125.9, 120.4, 69.7, 42.3, 23.2; IR (Neat Film NaCl) 3078, 2982, 2934, 1819, 1655, 1581, 1493, 1451, 1321, 1293, 1177, 1094, 1071, 1005, 930, 889, 780, 700 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₃H₁₃O₂N [M]⁺: 215.0946, found 215.0938.

Table SI2. Methods for the Determination of Enantiomeric Excess.

entry	product	compound assayed	assay conditions	retention time of major isomer (min)	retention time of minor isomer (min)	% ee
1			GC, G-TA 100 °C isotherm	11.13	12.74	88
2			HPLC Chiracel OJ 2% EtOH in hexane isocratic, 1.0 mL/mi	19.81 n	13.82	85
3	•	0 0 0	GC, G-TA 100 °C isotherm	19.67	21.64	2
4	N O Ph	N O Ph	HPLC Chiracel OD-H 2% IPA in hexane isocratic, 1.0 mL/mi	6.61 n	5.40	2

Tsuji Allylation with cyclohexenyl substrates



Sample Procedure

In a flame dried 1-dram vial, $Pd_2(dba)_3$ (4.6 mg, 0.005 mmol), (*S*)-*t*-BuPHOX (4.7 mg, 0.0625 mmol), and TBAT (18.9 mg, 0.035 mmol) were combined. The vial was evacuated for 10 minutes prior to addition of THF (3 mL). The mixture was allowed to stir at 25 °C for 30 min prior to addition of tridecane (10 µL, 0.4 mmol), silyl enol ether (18.4 mg, 0.1 mmol), and bromocyclohexene (16.1 mg, 0.105 mmol) via syringe. GC yield was determined by a GC assay with tridecane as the internal standard. (Isothermal at 80 °C for 5 min, then ramp from 80 °C to 115 °C at 10 °C/min, then isothermal at 115°C for 75 min. Silyl enol ether: 5.759 min, tridecane: 7.329 min, bromocyclohexene: 8.426 min, minor product diastereomer: 72.223 min, major product diastereomer: 73.434 min). Enantiomeric excess was determined by an Agilent 6850 GC utilizing a G-TA column (30 mm x 0.25 cm) with 1.0 mL/min carrier gas flow. The method utilized for enantiomeric excess determination was isothermal at 110°C for 60 min (major product diastereomer: 48.282 min (major enantiomer) and 55.842 min). Isolation of products as a

mixture of diastereomers was accomplished by flash chromatography (1 cm x 20 cm SiO_2 , 2% ether in pentane).



¹H NMR (300 MHz) δ 5.74 (complex multiplet, 1H), 5.49 (dddd, 0.7H, J = 11.0, 1.9, 1.9, 1.9 Hz), 5.20 (dddd, 0.3H, J = 10.1, 2.1, 2.1, 2.1 Hz), 2.84-2.29 (complex multiplet, 3H), 2.01-1.22 (complex multiplet, 12H), 0.90 (s, 2.1 H), 0.89 (s, 0.9 H); ¹³C NMR (75 MHz, CDCl₃) δ 216.7, 216.1, 129.6, 129.5, 128.0, 127.3, 52.0, 51.7, 39.5, 39.0, 38.6, 37.0, 36.0, 28.0, 27.7, 25.4, 24.0, 23.1, 23.0, 22.9, 21.1, 21.0, 19.4, 18.6; IR (Neat Film, NaCl) 3023, 2934, 2862, 1705, 1452, 1313, 1121 cm⁻¹; HRMS *m*/*z* calc'd for C₁₃H₂₄O [M⁺]: 192.1514, found 192.1519.

(1) Tsuji, J.; Minami, I.; Shimizu, I. Tetrahedron Lett. 1983, 24, 1793-1796.

(3) Slomczynska, U.; Kaminski, Z. J.; Leplawy, M. T. Bull. Pol. Acad. Sci. Chem. 1991, 39, 85-88.

^{(2) (}a) Trost, B. M.; Schroeder, G. M.; Kristensen, J. Angew. Chem., Int. Ed. 2002, 41, 3492-3495. (b) Trost, B. M.; Radinov, R.; Grenzer, E. M. J. Am. Chem. Soc. 1997, 119, 7879-7880.

2. Non-linear experiments:

Material and Methods. Unless otherwise stated, reactions were performed in flamedried glassware under argon atmosphere using dry, deoxygenated solvents. Solvents were dried by passage through an activated alumina column under argon. Tris(dibenzylideneacetone)dipalladium(0) (Pd₂(dba)₃) was purchased from Strem and stored in a dessicator under argon atmosphere prior to use. (S)-t-Bu-PHOX, (S)-i-Pr-PHOX, (R)-i-Pr-PHOX, and all substrates were prepared by our previously reported methods.4, Reaction temperatures were controlled by an IKAmag temperature modulator. Thin-layer chromatography (TLC) was performed by using E. Merck silica gel 60 F254 precoated plates (0.25 mm) and visualized by UV fluorescence quenching or ICN silica gel (particle size 0.032-0.063 mm) was used for flash anisaldehyde. chromatography. Analytical chiral HPLC was performed with an Agilent 1100 Series HPLC, utilizing a Chiracel OJ column (4.6 mm x 25 cm) obtained from Daicel Chemical Industries, Ltd with visualization at 254 nm. Analytical achiral GC was performed with an Agilent 6850 GC utilizing a DB-WAX column (30 mm x 0.24 mm with 1.0 mL/min carrier gas flow). Temperature controlled ¹H NMR kinetic experiments were performed on a Varian Inova 500 MHz.

General Procedures for Nonlinear Experiments



THF stock solutions with the desired enantiomeric excess of *i*-Pr-PHOX were freshly prepared prior to each experiment. The enantiomeric excess of the *i*-Pr-PHOX delivered was confirmed by subsequent chiral HPLC with a Chiracel OJ column using 1% ethanol in hexanes (1.0 mL/min) as an eluent on the remaining stock solution ((*S*)-*i*-Pr-PHOX: 13.16 min and (*R*)-*i*-Pr-PHOX: 7.60 min).

A 1-dram vial equipped with a stirbar was flame dried twice under vacuum. After cooling under nitrogen, $Pd_2(dba)_3$ (4.6 mg, 0.005 mmol) was added. The vial was evacuated for 5 minutes. THF (3 mL total) was added and then *i*-Pr-PHOX (4.8 mg, 0.0125 mmol) in THF was added via syringe. Contents were allowed to stir for 30 minutes at 25 °C prior to addition of benzyl β -ketoester **SI1** (27.2 mg, 0.1 mmol) via syringe. The reaction progress was monitored by TLC. Upon completion, the reaction, was concentrated and purified via column chromatography (1 cm x 11.5 cm SiO₂, 20% ether in pentane). Subsequently, the enantiomeric excess of product was determined by chiral HPLC with a Chiracel OJ column using 1% ethanol in hexanes (1.0 mL/min) as an eluent (**SI2**: 15.942 min and 24.345 min).

Comparison of the enantiomeric excess of the product versus the enantiomeric excess of the *i*-Pr-PHOX revealed a linear relationship (Figure 1). The absence of a nonlinear effect suggests that the active catalyst in our Tsuji allylation system involves one molecule of *i*-Pr-PHOX, thus one palladium metal center. Furthermore, the absence

⁴ Behenna, D. C.; Stoltz, B. M. J. Am. Chem. Soc. 2004, 126, 15044-15045.

⁵ Mohr, J. T.; Behenna, D. C.; Harned. A. M.; Stoltz, B. M. *Angew. Chem., Int. Ed.* **2005**, *44*, 6924-6927.

of a nonlinear effect suggests that the rate determining step does not involve a bimetallic system, such as a palladium-enolate and a palladium π -allyl complex.



Figure 1. Plot of Enantiomeric Excess of β-Ketoester SI2 versus Enantiomeric Excess of *i*-Pr-PHOX

3. Kinetic Data:

General Procedures for Kinetic Experiments

Determination of Substrate Order



Solid Pd₂(dba)₃ (2.3 mg, 0.0025 mmol) and (*S*)-*t*-Bu-PHOX (2.4 mg, 0.00625 mmol) were placed in a NMR tube equipped with a screw cap and a Teflon septum. The NMR tube was then placed under vacuum and backfilled with argon (3 x 1 min). THF-*d*₈ (0.2 mL, dried over sodium benzophenone ketyl) was added to the NMR tube via syringe under a positive pressure of argon. The mixture was heated at 40 °C for 30 min. The mixture was then cooled to -78 °C using a CO₂/acetone bath. A THF-*d*₈ solution (0.3 mL, 0.1 M in substrate total) of allyl enol carbonate **SI3** (12.2 mg, 0.05 mmol) and 1,4-dimethoxybenzene (2.4 mg, 0.0175 mmol) were added to the reaction mixture under argon. Before recording the ¹H NMR spectrum, the sample was allowed to warm for 5 - 10 seconds and mixed. Reaction progress was monitored by ¹H NMR spectroscopy at 0 °C, where integral areas of the allylic protons of **SI3** (dt, 2H, $\delta = 4.677$ ppm, J = 5.5, 1.0 Hz) relative to the phenyl protons of the dimethoxybenzene internal standard (s, 4H, $\delta = 6.795$ ppm) were obtained at 5 minute intervals. The experiment was concluded upon complete conversion of **SI3**, which was determined by the disappearance of the allylic protons of **SI3**.

Analysis of consumption of **SI3** over time is consistent with a zero-order dependence in allyl enol carbonate (Figure 2).



Figure 2. Plot of Consumption of Allyl Enol Carbonate SI3 versus Time as Observed by ¹H NMR Spectroscopy

Determination of Palladium(PHOX) Order



Asymmetric Tsuji allylation of enol carbonate **SI5** was carried out in an identical manner to those previously reported at different concentrations of the in situ generated Pd(PHOX) complex.¹ Reaction progress was monitored by an achiral GC equipped with a DB-WAX column with tridecane as the internal standard. GC yield was determined by using an acquisition method that ramped the temperature from 70 °C to 175 °C at a rate of 5 °C/min (tridecane: 6.915 min, cycloalkanone **SI6**: 12.185 min, and enol carbonate **SI5**: 17.697 min).

Rate constants (k_{obs}) were determined at 25 °C by GC analysis. The dependence of reaction rate on the concentration of the in situ generated Pd(PHOX) was measured at a constant concentration of **SI15** (0.03 M) and a constant concentration of tridecane (0.015M). Figure 3 shows that the reaction is first-order in Pd(PHOX) complex.





Based on these preliminary kinetic experiments, we believe that our asymmetric Tsuji allylation is zero-order in substrate and first-order in *in situ* palladium(PHOX) complex.

4. Calculated Data:

Calculation details: All calculations were run using Jaguar 6.5

Geometry optimization was done with a mixed basis set (MIDI! on all atoms except N, P, Pd, the six atoms from the allyl fragment and enolate component which were treated with LACVP or $6-31G^*$).

"Gas Phase Energy" was calculated as single point energy calculations from the above geometries with the LACV3P**++ basis set with both B3LYP and PBE.

"Solvent Phase Energy" was calculated with B3LYP and the LACV3P** basis set. Diffuse functions were omitted, as they appear to have adverse effects on Jaguar's solvation model. Solvent = THF (probe radius = 2.527Å, $\varepsilon = 7.52$)

"Zero Point Energy" was calculated from analytic vibrational frequencies.

All transition states were reported were fully optimized and yielded one imaginary frequency.

Contact John Keith for other calculation details (johnk@wag.caltech.edu)

	Complex 1		
	(B3LYP)		
G	as Phase Energy	= -654.64331885	491 = Eh
Se	olvent Phase Energy	= -0.010957458	54 = Eh
Ze	ero Point Energy	= 159.985	= kcal/mol
		(PBE)	
G	as Phase Energy	= -653.82932310	282 = Eh
	Geome	try Coordinates	
C1	0.0113083798	0.0124257236	-0.0061783496
C2	-0.0013818048	0.0086824472	1.3267454073
C3	1.2339260089	0.0153211569	2.1717260330
04	1.1429594509	-1.0908202308	3.1072486300
C5	2.1186404294	-1.1171196145	4.0385647212
06	3.0180904838	-0.3191945640	4.1324261749
07	2.0035448532	-2.1647199625	4.8809113190
C8	0.8991992312	-3.0435401244	4.8363028339
C9	1.0771331487	-4.2973061924	4.4017656946
H10	-0.9133147377	0.0566512194	-0.5891741503
HII	0.9476871255	-0.0189251237	-0.5744052870
H12	-0.9534029285	0.0384538672	1.8713083079
HI3	2.1474818658	-0.0980871386	1.5652816886
H14	1.3285573872	0.9488691482	2.7535568736
C15	-1.5746866876	-3.4044541922	5.1922740477
C16	-0.3551976806	-2.4877409835	5.4538668734
CI7	-0.0529567564	-5.3043896563	4.4860652339
C18	-1.1917306988	-4.8771957822	5.4433840026
HI9	-1.9051008295	-3.2856823628	4.1486577507
H20	-2.4102397826	-3.1039696803	5.8392585659
H21	-0.2035478549	-2.3634255284	6.5406674169
H22	-0.5522573151	-1.4841068032	5.0483581460
H23	-0.4659570753	-5.4740200848	3.4/481518/2
H24	0.3551672995	-6.2753897821	4.8132244080
H25	-0.8589056652	-4.9939727985	6.4867831713
H26	-2.0626868838	-5.532218/705	5.302115/953
C27	2.3661026241	-4./9664/9686	3.8064196230
H28	2.7491463200	-5.65/5604840	4.3802683709
H29	2.2066227425	-5.148241/926	2.7722771430
H30	3.1442248494	-4.0244807837	5.1952011238

	-
Complex 2	

......

	(B3LYP)		
Gas Phase Energy	= -466.037385550)19	= Eh
Solvent Phase Energy	= -0.0076306673	33	= Eh
Zero Point Energy	= 151.022	= kca	al/mol
	(PBE)		
Gas Phase Energy	= -465.412944018	344	= Eh

	Geomet	ry Coordinates		
C1	-0.0109108095	-0.0090913358	-0.0039725405	
C2	-0.0019377287	0.0036042017	1.3311872844	
C3	1.2380737348	-0.0031750635	2.1857876305	
C4	1.4287766106	-1.2691160917	3.0695595585	
C5	2.7189398847	-1.0964231308	3.9031287517	
O6	3.5264703528	-0.2162058183	3.6688334943	
C7	0.2197252134	-1.4655974661	4.0267158875	
C8	0.4425853342	-2.5256322103	5.1342095225	
С9	1.6991627758	-2.1908086263	5.9645485923	
C10	2.9434642135	-2.0867930224	5.0391924948	
H11	-0.0056320434	-0.5013459258	4.5157686164	
H12	-0.6675204448	-1.7358556422	3.4339196855	
H13	0.5474941665	-3.5277876176	4.6914167333	
H14	-0.4422202193	-2.5569111685	5.7863780637	
H15	1.5517832876	-1.2318804997	6.4865076425	
H16	1.8737722150	-2.9553599127	6.7343017634	
H17	3.8338193270	-1.7655138907	5.5946626836	
H18	3.1533849642	-3.0833841603	4.6143920855	
C19	1.6086725483	-2.5137696692	2.1667783431	
H20	2.4990070331	-2.4061609466	1.5279721473	
H21	1.7168858220	-3.4408988269	2.7458276242	
H22	0.7345201419	-2.6209798584	1.5084046329	
H23	1.2232556197	0.8767106710	2.8575081124	
H24	-0.9639896574	0.0470791753	1.8593952008	
H25	0.9176493118	-0.0351836527	-0.5858169872	
H26	-0.9443975068	0.0188140773	-0.5750236071	
H27	2.1318023368	0.1153535642	1.5506968142	
		••••••••••••••••	•••••••••••••	

Complex CO2

Gas Phase Energy Zero Point Energy	(B3LYP) = -188.64660127352 = Eh = 7.282 = kcal/mol
Gas Phase Energy	(PBE) = -188.45726211825 = Eh
Geom	netry Coordinates

O3	-1.7834484094	-0.0452538213	3.7234453188
O2	0.0327656439	1.4194127886	3.5572223432
C1	-0.8753363139	0.6870813764	3.6403431163

Com	plex Pd(PHOX)	

((B3LYP)		
Gas Phase Energy	= -1566.716.	36231574	= Eh
Solvent Phase Energy	= -0.01365	948285	= Eh
Zero Point Energy	= 282.286	= kca	al/mol
	(PBE)		

Gas Phase Energy = -1565.11360608546 = Eh

	Geomet	ry Coordinates	
Pd1	0.1771925079	0.0303972044	-0.4792913529
N2	0.3026055938	-0.1410796128	2.7832384005
P3	2.2198705967	0.1131084513	0.4386177259
C4	-1.0038182842	-0.4993172805	3.3625614198
C5	-1.3695267186	0.7361541683	4.2307918808
06	-0.3882895714	1.7308207778	3.8622197991
C7	0 5365601358	1 0775218072	3 0867538636
C8	1 7119479275	1.8758559013	2 6970712247
C9	2 5690975521	1 5236726441	1 6174755080
H10	-1 2808860521	0 5535120633	5 3074731832
H11	-2 3613727745	1 1431853758	4 0214715692
нн 1111 1112	1 7172300412	0.5028340006	2 5308864505
C13	0.08/1078360	1 8701715/35	4 0045000747
C13	-0.3641378303	2 1070062027	4.0943009747
U15	-2.3048332201	-2.1070002927	4.7373300320
Ш1 Ц16	2 16/0/58600	-1.3712200232	2 0897277476
П10 Ц17	-3.1049438090	2.0076202088	5 2020555245
П1/ С19	-2.4031370939	-3.0970292088	5 1607106212
U10	0.11//1343/0	-1.9123691036	5.109/100515
П19 1120	-0.0260202493	-1.1333903903	5 6667759700
H20	0.1230704120	-2.0000419941	3.0007738799
П21 С22	1.1022625775	-1./330811//1	4.7207160910
C22	-0.7141053043	-2.9/21399416	2.5212475709
П23 1124	-0.0923445310	-5.9577556702	3.3312473700
H24 1125	-1.4991023109	-2.9905245508	2.2804003029
П23 С26	0.2420939313	-2.0103202074	2.3303066020
C_{20}	3.0400411160	-3.7363110336	2.3046716042
C_{28}	4.1074920047	-2.4955000171	2 1012167370
C_{20}	2 8405525532	-1.3200440390	1 3280108613
C29	2.8403323332	2 6405304324	0.7628757074
C30	2.3080003340	2.0403304324	1 3/01708173
U31 U32	1 2263028740	-5.8005787580	2 0601381037
П32 Ц33	4.2203020740	2 / 31 / 105 37 /	2.9001301937
П33 Ц34	3 8388033638	-2.4314403374	2 0480214760
1134 1125	1.0555010704	-0.3033148073	0.1222012057
ПЭЭ Ц26	2 9521177860	-2.09/3239437	-0.1333013037
П30 С27	2.0321177000	-4.7093333780	0.0011931033
C_{28}	3.40011/3313	1 2026606001	-2.03/03/093/
C30	4.2002346073	1.2030000991	-3.0062311302
C39	3.2307099374	1.024/192022	-2.00/965//5/
C40	3.3013402203	0.3211378322	-0.03302/0/32
C41 C42	4.8525299578	-0.2141929008	-0.0939888/28
C42	5.8049571905	-0.0441991918	-1./000423110
H45	0.2283709031	0.7952972902	-3.0409822040
H44	3.94/69439/8	1.7523924171	-3.9099742196
H45	2.2440352795	1.41/54818/5	-2.1322906313
H46	5.1159062955	-0.7705053528	0.1999946198
H4/	6.7974292943	-0.4/053090//	-1.5/84051954
C48	3.6692839902	2.3615463383	1.368/018/2/
H49	4.5411402151	2.128/39/114	0.5515/94144
C30	1.9/63230183	3.0296087707	5.4539024330
H)1	1.3009806790	3.2/90098495	4.2042189791
C52	3.928/604198	3.494/888886	2.1392257713
H33	4.1913122591	4.10/3/88890	1.912343/951
C54	3.0/0/496909	3.8370912632	3.18016096//
HDD	0.2040320738	4.7200000803	3.7890262267

Complex Enolate

	Gas Phase Energy Solvent Phase Energy Zero Point Energy	(B3LYP) = -348.71065236 = -0.084252591 = 102.902	365 = Eh 24 = Eh = kcal/mol		
(PBE) Gas Phase Energy = -348.25500040654 = Eh					
	Geome	try Coordinates			
01	-0.0003410345	0.0018492883	-0.0018906891		
C2	-0.0012930859	-0.0008335598	1.2724903838		
C3	1.1061593054	-0.0010872893	2.1166904574		

C_{2}	-0.0012/5005/	-0.00003333770	1.272+705050
C3	1.1061593054	-0.0010872893	2.1166904574
C4	-1.4104526170	-0.0319872575	1.9233471036
C5	2.4818715361	0.0739164076	1.5196859652
C6	-1.4515514456	0.3591684973	3.4170181569
C7	-0.3527287552	-0.4105932402	4.1753994277
C8	1.0376837131	-0.0101397774	3.6211509483
H9	-1.2650581487	1.4419209104	3.5240176167
H10	-2.4459585505	0.1554018823	3.8553292856
H11	-0.5098798273	-1.4922663929	4.0247324928
H12	-0.4093491028	-0.2110971796	5.2604028648
H13	1.3059029375	0.9803472051	4.0738041013
H14	1.7916277693	-0.7143291969	4.0427607809
H15	-1.8367868734	-1.0464439177	1.7975540590
H16	-2.0494777870	0.6403990799	1.3261341092
H17	2.3821126706	0.0632008815	0.4232794062
H18	3.0422779241	0.9968736269	1.8062482500
H19	3.1410928032	-0.7749511975	1.8204725621

Complex 3

	(B3LYP)		
Gas Phase Energy	= -1683.907	95339188	= Eh
Solvent Phase Energy	= -0.05367	430515	= Eh
Zero Point Energy	= 330.264	= kc	al/mol

(PBE)

Ga	s Phase Energy	= -1682.15272185	5669	= Eh
	Geome	try Coordinates		
Pd1	0.0179829299	-0.0676708739	-0.0	340612533
C2	0.5050195355	-1.1692203769	1.84	489587378
C3	-0.8985553714	-1.0206510966	1.6	545794544
C4	1.3163391626	-0.0389784858	1.89	956242167
H5	-1.5090774567	-1.9019537550	1.43	854221069

1.8535746508

2.1262357489

2.2504672281

2.3957602772 -0.1371207332

-1.4192523527 -0.1889278914

0.9276249187 0.9130875980

H6

H7

H8

H9	0.9636611927	-2.1435452811	1.6917623608
N10	1.1511428117	1.0692029511	-1.4617307199
P11	-1.7174002497	-0.1717162380	-1.5812161704
C12	2.6446626690	1.1016133800	-1.4946703082
C13	2.9128829823	2.4582786137	-2.1982793762
014	1 6458038779	2 7356867692	-2 8905046568
C15	0 7042142647	1 9499153489	-2 3012413409
H16	3 6070106065	2 4401837701	2 9562777418
C17	1 0801338212	2.4401037791	4 6701348073
C18	0.0015160272	3 7020162060	3 31/201/301
C10	1 0864813043	2 7504205402	2 3080023311
C19 C20	-1.0804813043	-2.7394303492	2.3980923311
C20	-1.4020904090	-1.4/06033070	-2.8390104300
C21 C22	-1.0300193694	-1.24/9210333	-4.2131439/44
C22	-1.4431481568	-2.2800855522	-5.12531849/1
H23	-0.9335313877	-4.36343/6384	-5.3934074944
H24	-0.6224696672	-4.7828988999	-2.9639494457
H25	-0.9402425784	-2.9505439510	-1.336/592315
H26	-1.9276546004	-0.2655093506	-4.5751408796
H27	-1.5807296881	-2.1020685290	-6.1880005070
C28	-6.0108222495	-0.6827395226	0.0847630174
C29	-5.5822308034	-1.5014328706	-0.9587416830
C30	-4.2929564870	-1.3649872842	-1.4782699849
C31	-3.4198230169	-0.4035262947	-0.9472616242
C32	-3.8571442232	0.4141129064	0.1112558776
C33	-5.1470706008	0.2775924727	0.6174870591
H34	-7.0156694884	-0.7918028681	0.4852892341
H35	-6.2527440211	-2.2501124139	-1.3737276006
H36	-3.9715041448	-2.0086007331	-2.2926869648
H37	-3.1922296286	1.1650736918	0.5348615338
H38	-5.4801222065	0.9178646759	1.4308987877
C39	-2.0917909438	3.8288111183	-3.9397626248
C40	-3.1955760773	3.0087813541	-3.7309847663
C41	-3 0479854883	1 8169764747	-3 0216946696
C42	-1 8029857361	1 4161325566	-2 5147376653
C43	-0 6774653169	2 2487612543	-2 7363715505
C44	-0.8484695501	3 4481733444	-3 4474833487
U44 U45	2 10/33503/6	<i>A</i> 7645713061	A A827813227
П 4 5 Ц46	-2.1945550540 4 1734051120	3 2020850556	4 1118200136
1140 1147	3 0204600432	1 1032225546	-4.1118290130
1147	-3.9204099432	1.1932223340	2 6005204852
H40	0.01499/1010	4.0650106902	-3.0093204833
П49 С50	2 2672106692	0.1222000060	-1.4921003989
C50	3.20/3190063	-0.1522099009	-2.2290481400
(51	4.8097960218	0.0355752320	-2.2321092500
H52	5.1372215469	0.8949872402	-2.8338522574
H53	5.19865///41	0.15/4569696	-1.2094740686
H54	5.2753864147	-0.8611778049	-2.6646233451
C55	2.7466860750	-0.2452939683	-3.6/969400/4
H56	2.9375390805	0.6699273619	-4.2584065984
H57	3.2584862794	-1.0737003009	-4.1902023908
H58	1.6679877242	-0.4548105081	-3.7011281686
C59	2.9295243491	-1.4292044742	-1.4578823501
H60	3.3997035260	-2.2895861753	-1.9559421718
H61	3.3142422646	-1.3859212143	-0.4274744391
H62	1.8451801419	-1.6056103459	-1.4248356794
H63	3.0184804616	1.1146049470	-0.4645288410
*****	*****	****	******

	Complex 4						
	(B3LYP)						
	Gas Phase Energy $= -2032.73911500101 = Eh$						
	Solvent Phase Energy	= -0.017845631	76 = Eh				
	Zero Point Energy	= 434.031	= kcal/mol				
	(PBE)						
	Gas Phase Energy	= -2030.53944632	2755 = Eh				
_	Geome	try Coordinates					
Pd	0.1522421755	-0.1116392671	-0.0014489177				
C2	0.3201673009	-0.1796904055	2.2760129712				
03	2.7683828369	-0.1292058406	-0.0509452580				
C4	-0.3089759399	1.0020591816	1.8725967702				
C5	0.3572462693	1.8198794116	0.9189059555				
C6	3.6231836400	-1.0320514691	-0.4454565077				
C/	4.5645342779	-0.6017/01382	-1.5790101993				
C8	3./319944213	-2.2986969229	0.0963151434				
C9	2.9319355116	-2.6638/38931	1.31/6139803				
	4.7627721980	-3.3113686164	-0.3585935803				
HI	-0.1635/96499	2.6681043577	0.4843450687				
HI.	2 -0.2222383097	-0.9166488314	2.8606049645				
HI.	3 1.4435092171 1.4025140507	1.8/9345599/	0.9323271725				
H14	+ 1.4033149307	-0.2011482144	2.2437892374				
HI.	-1.3002392380	1.1559400514	2.0790270703				
D1'	0.0904120280	-2.1721915229	-0.02480/1/39				
ГI. U19	7 0.1740600923	0.4320736241	-2.2431717911				
ц10 Ц10	$3 \qquad 5.5555450025$	-0.3790031703	1 3704720165				
111 ЦЭ	7 - 4.0020042027	3 75668/1867	1 4007000411				
112) Ц2	1 3 3736753121	2 1850001038	2 2/327/6779				
H2	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-2.1050771050	1 2329248766				
$H2^{\prime}$	$\frac{1}{3}$ $\frac{1}$	-4 3222079238	-0.4201911164				
H2	4 5 5694359641	-3 4207278375	0.4055290059				
$C2^{i}$	5 -0.7350831020	-3 2196094825	0.2145874496				
C2	5 0.0253384366	-4 4963013059	-0 2322589823				
02'	7 0.5646050655	-4.1275435673	-1.5433923246				
C28	3 0.5490572057	-2.7557291926	-1.5872860900				
H29	-0.5979310895	-5.3814233610	-0.3767989277				
C30) 1.0751352456	-0.8356551795	-3.2059721504				
C3	1.1800742899	-2.1879310773	-2.7905657485				
H3	2 0.8671385414	-4.7299403361	0.4300132078				
C33	3 -2.2861720057	-3.2868486983	0.0078957052				
C34	4 -2.8503553237	-4.4322244334	0.8875736219				
H3:	5 -2.4968156257	-5.4213090582	0.5626847308				
H3(-2.5704486745	-4.2925639033	1.9434975890				
H3′	-3.9486254196	-4.4383230060	0.8269120766				
C38	-2.6429574335	-3.5374779809	-1.4747726814				
H39	-2.1569173063	-4.4433254395	-1.8650556956				
H4(-3.7310120313	-3.6646211075	-1.5783460509				
H4	-2.3418826533	-2.6849977453	-2.0990721550				
C42	2 -2.9326790960	-1.9607299977	0.4721737948				
H4.	-4.0245459550	-2.0136723051	0.3421425130				
H44	4 -2.7256056999	-1.7760957874	1.5377230994				

H45	-2.5505778966	-1.1091203781	-0.1060660246
H46	-0.5403211568	-2.9996832719	1.2703192523
C47	2.3860184274	4.3928572056	-3.2837949551
C48	1.1849262514	4.1040696560	-3.9307794581
C49	0.4990063076	2.9223334720	-3.6431855483
C50	1.0156992509	2.0222375547	-2.6982728271
C51	2.2242796977	2.3187630128	-2.0406796476
C52	2.9016472517	3.4996854342	-2.3422000820
H53	2.9179077889	5.3148280951	-3.5105141300
H54	0.7783647615	4.7996636351	-4.6623478042
H55	-0.4393082329	2.7098002870	-4.1494234414
H56	2.6195443006	1.6088146650	-1.3055510017
H57	3.8379878854	3.7243131230	-1.8353142604
C58	-4.0140849884	0.8486431113	-4.2294907723
C59	-3.0426757843	0.0654841238	-4.8504161246
C60	-1.7758545288	-0.0722575192	-4.2781553383
C61	-1.4704573260	0.5697076485	-3.0683288784
C62	-2.4649169194	1.3419170502	-2.4406972350
C63	-3.7217207787	1.4881266566	-3.0229061547
H64	-4.9979071726	0.9582920145	-4.6804358217
H65	-3.2670363641	-0.4380494793	-5.7884578786
H66	-1.0232965470	-0.6749349737	-4.7796770291
H67	-2.2502441947	1.8272918150	-1.4903614712
H68	-4.4779273956	2.0969750059	-2.5318640115
C69	1.6776467046	-0.4539878202	-4.4182651942
H70	1.6202426167	0.5839995041	-4.7339037256
C71	1.8801960366	-3.0948204178	-3.6107465389
H72	1.9634346085	-4.1239615571	-3.2816818655
C73	2.3570314890	-1.3677921209	-5.2172415639
H74	2.8120335257	-1.0408065248	-6.1494141375
C75	2.4586284413	-2.6978214199	-4.8070394514
H76	2.9941504587	-3.4227225064	-5.4152182616
C77	5.8108451629	-1.4802030793	-1.7353239197
H78	6.5162215574	-1.2846717852	-0.9091958086
H79	6.3415542060	-1.2405257562	-2.6715555184
C80	5.4125045433	-2.9724497525	-1.7051510232
H81	6.2936522790	-3.6113461306	-1.8783039381
H82	4.7016915621	-3.1642056614	-2.5260393040

Complex 5

	, ompren e	
	(B3LYP)	
Gas Phase Energy	= -2032.73488137	7273 = Eh
Solvent Phase Energy	= -0.018730375	75 = Eh
Zero Point Energy	= 434.098	= kcal/mol
Frequency	= -34.91	

(PBE)

Gas Phase Energy = -2030.53248097952 = Eh

Pd1	0.0000000000	0.00000000000	0.0000000000
O2	0.0000000000	0.0000000000	2.4803720451
C3	2.3476054165	0.0000000000	0.7367316679
P4	-1.8800020861	-0.5618249390	-1.1806220877

N5	-0.7195656370	2.0597627224	0.0137173621	
C6	0.9064332782	-1.8840233523	0.2142974169	
C7	2.0445691690	-1.0617588596	-0.0915735806	
C8	-0.7922314100	0.6676829591	3.2821701313	
C9	-2.2509241518	0 1857639555	3 2879022041	
C10	-0.4063556970	1 7062882460	4 1018820247	
C10	1 0/2556070/	2 0837572704	4.1010020247	
C12	1 2702764862	2.0037372704	4.0842006402	
C12	-1.3/93/04003	2.4034673294	4.9845000402	
	-3.1383402101	0.7770323138	4.3901100130	
U14	-2.839//18/13	2.284/2/4280	4.3370714443	
HIS	-2.9358046298	0.2743380753	5.3511223962	
HI6	-4.2047722555	0.6135394931	4.1628775917	
HI7	-3.521/14/132	2.7350407823	5.2969811729	
H18	-3.0092349046	2.7942325503	3.5925930708	
H19	-2.7000559021	0.3900064897	2.2954888901	
H20	-2.2191406357	-0.9177934332	3.3776356079	
H21	1.2619118657	3.1338681065	3.9709727183	
H22	1.3566809869	2.0156057803	5.3358840075	
H23	1.6974826371	1.4175034753	3.6882268715	
H24	-1.1277811425	3.5473387431	4.9780721229	
H25	-1.2778659160	2.1626772194	6.0543878341	
H26	0.6299523797	-2.6735728266	-0.4795275726	
H27	3.0728822401	0.7458542597	0.4238115325	
H28	0.6660856868	-2.0624677761	1.2615306105	
H29	1.9397681570	0.0474465421	1.7466374505	
H30	2.5138443074	-1.1399840470	-1.0713262590	
C31	0.1812309612	3.1604316961	0.4629063979	
C32	-0.7452405288	3.9527408768	1.4154757460	
033	-2.0914874689	3.6008132210	0.9327958474	
C34	-1 9379362622	2,4149678241	0 2662991991	
H35	-0.6577173350	5 0398267895	1 3699702015	
C36	-3 3331200492	0 4700475481	-0 6838786444	
C37	-3 2113039348	1 7446025095	-0.0738916922	
H38	-0.6625795277	3 5742926563	2 4425828016	
C39	0 7450457047	3 9829183995	-0 7415342329	
C40	1 6788928218	5.0867728182	-0.1831908470	
U40	1 1333003055	5.8178336242	0.4308846595	
11+1 1142	0 1010220100	1 6407621426	0.4200802510	
П42 Ц42	2.4010330420	4.0497021420	1.0152001057	
П43 С44	2.1433961093	1.6002404156	-1.0132001937	
C44	-0.3900383902	4.0292424130	-1.30/0040334	
П4Ј Ц46	-1.0344233910	5.2700320102	-0.9488040423	
H40	0.0403812048	3.2309039384	-2.3003418098	
H4/	-1.01/0/44416	3.8597922931	-2.0401356505	
C48	1.5760122526	3.05/95038/5	-1.6615697933	
H49	1.9939656734	3.6413559812	-2.4962698698	
H50	2.4117096252	2.6038245279	-1.1088845448	
H51	0.9573327549	2.2498820850	-2.0736224508	
H52	1.0163878771	2.7227230956	1.0184537406	
C53	-3.4086102107	-4.9030843740	-0.5674705694	
C54	-3.3279831636	-4.3939449216	-1.8624661290	
C55	-2.8845741818	-3.0871164255	-2.0789992967	
C56	-2.5132856862	-2.2791085131	-0.9942049296	
C57	-2.5818701933	-2.8056336904	0.3091561554	
C58	-3.0355866321	-4.1061042413	0.5169557627	
H59	-3.7541665409	-5.9212640420	-0.4019386526	
H60	-3.6090571493	-5.0145894062	-2.7102397005	

H61	-2.8244554032	-2.7014474119	-3.0935496837
H62	-2.2712339676	-2.1974572829	1.1576893747
H63	-3.0889145554	-4.5027041241	1.5287555954
C64	-1.4395461626	0.0044342230	-5.7793216043
C65	-2.6908340590	0.1962592220	-5.1932725625
C66	-2.8553650403	0.0399216749	-3.8163992255
C67	-1.7630296496	-0.3100851497	-3.0062840665
C68	-0.5044470040	-0.4853757023	-3.6055647271
C69	-0.3463340711	-0.3368857458	-4.9827212123
H70	-1.3153088822	0.1284774518	-6.8528320453
H71	-3.5436197597	0.4722026535	-5.8099053953
H72	-3.8341496035	0.2030161266	-3.3744114483
H73	0.3526202106	-0.7246791946	-2.9792211061
H74	0.6334966583	-0.4780263819	-5.4328751065
C75	-4.6206080175	-0.0498147446	-0.9005027928
H76	-4.7270762954	-1.0440822164	-1.3251947583
C77	-4.3817107142	2.4461824658	0.2675069522
H78	-4.2784088137	3.4157065455	0.7403735308
C79	-5.7665363634	0.6634457911	-0.5578048907
H80	-6.7482084736	0.2304573169	-0.7366032364
C81	-5.6440672192	1.9221918765	0.0236956770
H82	-6.5282403983	2.4903939387	0.3019617665
******	*****	***********	*****

Complex 6

	(B3LYP)			
Gas Phase Energy	= -2032.72926	163396	= Eh	
Solvent Phase Energy	= -0.0218296	9499	= Eh	
Zero Point Energy	= 434.068	= kc	al/mol	
Frequency $= -30.63$				

Gas Phase Energy	= -2030.52840672257	= Eh

		-)	
Pd1	0.0000000000	0.0000000000	0.0000000000
C2	0.0000000000	0.0000000000	2.4380822781
C3	0.6432168078	0.0000000000	5.2288898795
C4	-0.9577996270	0.8242757005	1.8292527455
C5	-1.9333911447	0.3010441896	0.9393466568
C6	1.3920387959	-1.0411775719	4.6521739538
07	2.4038882128	-0.8626009024	3.8926831343
C8	0.9582318422	-2.4964169892	4.8908529369
C9	1.1490531325	1.4175118987	5.1694290282
C10	-0.5291132928	-0.2378446570	6.1606933875
H11	-2.6383125028	0.9742306191	0.4606679979
H12	-2.2888799627	-0.7201752021	1.0704217078
H13	-0.8301103775	1.9037677609	1.8923316849
H14	0.9290103527	0.3785585874	2.8291907710
H15	-0.1499539666	-1.0723479593	2.4991565056
H16	1.0755042597	-3.0454280781	3.9377318532
H17	1.6900573488	-2.9549284256	5.5873659314
H18	0.3709363831	2.1387011460	4.8407575547
H19	1.4785716180	1.7684365765	6.1722558610
H20	2.0158725557	1.4938857868	4.4933295049

H21	-0 4306748395	0.4183825616	7 0521528150
1121 ЦЭЭ	1 4820600508	0.930087822	5 6810176663
1122 NO2	-1.4620090306	0.0039907022	0.0701657119
N23	2.2012/55411	-0.1770214513	-0.2/0165/118
P24	-0.0/4683002/	-0.1/06195/45	-2.349/321554
C25	3.1417415294	-0.7281884090	0.7519737679
C26	4.3768076860	0.1876453930	0.5678148220
O27	4.1972964798	0.7472717266	-0.7857698346
C28	2.8740324098	0.5651857183	-1.0812655715
C29	2.4219869655	1.2718017843	-2.3027869054
C30	1,1917621682	1.0385633280	-2.9727103546
H31	5 3405861920	-0 3246122924	0 5811828515
Ц32	1 3875007254	1.02/0633226	1 2786452064
H32 H22	7.3073777234	0.6102644112	1.2700452004
П33 С24	2.7140343000	-0.0192044112	1.7303104000
C34	3.43/5695046	-2.2484693420	0.5347891135
C35	4.3907486126	-2.6996420717	1.6/3/633360
H36	5.3864747323	-2.2393179696	1.5824731914
H37	3.9597180661	-2.4271499633	2.6491859319
H38	4.5280525539	-3.7907324593	1.6299833012
C39	4.0695216000	-2.5273875995	-0.8464098140
H40	4.9965819577	-1.9542667694	-0.9971420737
H41	4.3206008154	-3.5959200446	-0.9286627601
H42	3.3723668218	-2.2834327564	-1.6610162573
C43	2 1196406902	-3 0445101136	0 6744536143
H44	2 3253943365	-4 1224262846	0.5863410312
H45	1 6726112849	-2 8563674501	1 6605266346
H46	1 4006668427	-2.0505074501	-0 1079690935
C47	3 016/307725	1 0237571877	4 683/0501/6
C47	3 1101506713	0.0008040066	5 1816108066
C40	-3.1101300713	0.0008040000	-5.1810198000
C49	-1.9460347004	-0.3737047013	-4.3042330337
C50	-1.3839113122	0.2/12/30800	-3.31100/8/13
C51	-2.4103528689	1.2928404124	-2.81183/6405
C52	-3.5634403651	1.6698/1351/	-3.49/3505416
H53	-4.8211297859	1.31268888881	-5.2138114521
H54	-3.3846515179	-0.5092779083	-6.1026847695
H55	-1.3291652504	-1.1724067821	-4.9036477189
H56	-2.1491921337	1.7870732624	-1.8782323947
H57	-4.1931451028	2.4637717511	-3.1015363532
C58	1.0991936356	-4.2519871459	-4.2399010222
C59	1.7062582256	-3.0882490547	-4.7107716932
C60	1.3797715802	-1.8494949096	-4.1550951219
C61	0 4425787059	-1 7645983729	-3 1129760199
C62	-0 1530088040	-2 9457673083	-2 6361437772
C63	0.1677128116	-4 1780335204	-3 2029527657
С05 Н64	1 35/2028781	5 2150802/30	-5.2027527057 A 6763301844
110 4 1165	2 4254528428	2 1416000041	5 5165006608
	2.4334330420	-3.1410090041	-5.5105090000
	1.6337370034	-0.9493703263	-4.3363949304
H6/	-0.8/12/56638	-2.8985923454	-1.81966/086/
H68	-0.3040158359	-5.083/941180	-2.8285858033
C69	3.0096058630	2.9563150416	-3.9655288892
C70	3.3081916232	2.2303795607	-2.8181216893
C71	0.9155798386	1.7771799082	-4.1331398154
C72	1.8088181324	2.7271029623	-4.6278451081
H73	3.7131767993	3.6979080775	-4.3358575125
H74	4.2418481011	2.3973850102	-2.2929354265
H75	-0.0206353150	1.6132140828	-4.6575701994
H76	1.5595325253	3.2848774661	-5.5278170572

C77	-0.6739835636	-1.6916908541	6.6302127243
H78	-1.6635797548	-1.8512185521	7.0878784722
H79	0.0799410992	-1.9102276847	7.4059672475
C80	-0.4583710845	-2.6651725868	5.4495242864
H81	-1.2066190791	-2.4485879632	4.6667337224
H82	-0.6270041583	-3.7050178951	5.7739204697
*****	*****	*****	*****

Compl	lex	7
-------	-----	---

Eh
Ξh
ıol
]

(PBE) Gas Phase Energy = -2030.52549546915 = Eh

	Geomet	ry Coordinates	
Pd1	0.0000000000	0.0000000000	0.0000000000
C2	0.0000000000	0.0000000000	2.3211968669
C3	0.8424999870	0.0000000000	5.2512512366
C4	-0.6604120923	-1.9348412135	0.9595233626
C5	2.0348324454	-0.2310662807	4.5376139376
06	2.6348456752	0.6466958488	3.8354158531
C7	2.6662417406	-1.6330421228	4.5701704904
C8	0.2961210408	1.3961934591	5.3680717406
C9	0.1728499396	-1.0508920472	6.1098325762
H10	0.7948244678	0.5765149642	2.7720940497
H11	-1.0279369486	0.3229104883	2.4686154440
H12	1.2858408351	-1.6542438342	1.8480005605
H13	-0.3944955452	-2.8749897192	0.4852676950
H14	-1.7244022252	-1.7326377282	1.0634760359
H15	3.5597004710	-1.5853182383	5.2270714149
H16	3.0593542203	-1.8489673429	3.5582365913
H17	-0.7921611092	1.4502644686	5.1542871484
H18	0.4156091570	1.7923032950	6.4009815514
H19	0.8276757331	2.0791754177	4.6873643011
H20	-0.1086039186	-0.6061760141	7.0889316396
H21	-0.7975838993	-1.3648843601	5.6603378580
N22	-0.2745441670	-0.2905545527	-2.1920555215
P23	0.8648308863	2.1113477013	-0.6693151461
C24	-1.2509872732	-1.2571523956	-2.7626386475
C25	-0.4420654562	-1.9002773930	-3.9231160958
O26	0.6345547654	-0.9384740923	-4.1632467737
C27	0.6946263401	-0.1464163899	-3.0415768310
C28	1.8728726616	0.7334422716	-2.9853464257
C29	2.0761494729	1.7610679965	-2.0218505857
H30	-0.9906321073	-2.0177097100	-4.8602362182
H31	0.0126772887	-2.8586026226	-3.6381708761
H32	-1.4950380278	-2.0010319604	-1.9954775416
C33	2.4241038645	2.7233170959	1.5561167438
C34	1.7400677103	3.2648340468	0.4560900697
C35	1.7305566879	4.6604610113	0.2713410300
H36	2.4342950681	1.6572310425	1.7657888990

C37	-0.1744459942	3.8005104158	-2.7353763444
C38	-0.3842402897	3.1941069380	-1.4871547730
C39	-1.5873254814	3.4462197271	-0.8046222505
H40	0.7497590375	3.6229391528	-3.2795306915
C41	-2.5829048012	-0.5723812117	-3.2115703309
C42	-3.5226457050	-1.6611799856	-3.7906283054
H43	-3.1206050633	-2.1126142666	-4.7095296194
H44	-3.6984259948	-2.4631779990	-3.0568533627
H45	-4.4954173395	-1.2140115013	-4.0424949399
C46	-2.3236751391	0.5166456810	-4.2766675801
H47	-1.7994590868	0.1136099881	-5.1557923258
H48	-3.2821669514	0.9339498639	-4.6198067842
H49	-1.7273284056	1.3400238101	-3.8615835900
C50	-3.2711894530	0.0620790727	-1.9813446505
H51	-4.2359934795	0.5002592874	-2.2786198113
H52	-3.4654380926	-0.6960811123	-1.2072950761
H53	-2.6515770794	0.8562591986	-1.5472490587
C54	3.0863060738	3.5602725378	2.4558857286
C55	2.4027650897	5.4884113118	1.1684566451
H56	-1.7673404030	2.9824138821	0.1636368688
C57	3.0798840591	4.9403968382	2.2615620909
H58	3.5686606868	3.1019120375	3.3147293051
H59	3.5894325554	5.5940185150	2.9665952082
H60	2.3898130592	6.5665337320	1.0198503902
H61	1.1923191041	5.0998890573	-0.5650727955
C62	3.2639293977	2.5054802833	-2.0831420359
H63	3.4376797169	3.2834967942	-1.3462185188
C64	2.8582121102	0.5133783170	-3.9652365103
H65	2.6930188108	-0.2734663361	-4.6926562066
C66	4.2268004642	2.2702537433	-3.0626286700
H67	5.1353855670	2.8677815670	-3.0804999502
C68	4.0213668206	1.2695081458	-4.0094654907
H69	4.7664559798	1.0728666145	-4.7766134510
C70	-1.1435577183	4.6448497973	-3.2830470551
H71	-0.9667010913	5.1117833019	-4.2499459543
C72	-2.5469733724	4.2973968449	-1.3506503759
H73	-3.4690537648	4.4911595023	-0.8068178154
C74	-2.3280888886	4.8971398964	-2.5920592894
H75	-3.0784118973	5.5595794610	-3.0183503598
C76	1.7313693390	-2.7447299319	5.0553938736
H77	2.2909677741	-3.6804210814	5.2200821173
H78	0.9637938530	-2.9596843707	4.2909866494
C79	1.0289212136	-2.3009106400	6.3589482287
H80	0.4080627879	-3.1172395038	6.7609951908
H81	1.8016646973	-2.0804973041	7.1152871298
C82	0.2694098235	-1.2734770317	1.7771592089
***********	*******	************	*********************

Complex 8

(B3LYP)					
Gas Phase Energy	= -2032.75792237	478	= Eh		
Solvent Phase Energy	= -0.0132273134	14	= Eh		
Zero Point Energy	= 434.122	= kca	al/mol		

Gas Phase Energy	= -203055097086362	= Eh
Oas r hase Energy	2030.33097080302	$-L\Pi$

	Geomet	ry Coordinates		
Pd1	0.1915648604	-0.0578528428	-0.0045695048	
C2	0.7380779138	0.2487995492	1.9995464524	
C3	0.9177841207	-1.0073451258	2.7347292550	
C4	2.1009685961	-1.5430077692	3.0837744176	
C5	2.4878983867	-3.6194572433	0.6346249865	
06	1.0583871202	-1.9540125174	-0.2257736421	
C7	2.2726742622	-2.3890021033	0.0776056574	
C8	3.4297966073	-1.4751329042	-0.3067694587	
C9	1.3453136245	-4.5030156999	1.0554435136	
C10	3.8735081920	-4.2179363732	0.7748355573	
H11	1.6597711792	0.8380857605	1.9368195007	
H12	-0.0661278683	0.8647676841	2.4088914371	
H13	0.0075837368	-1.5471634442	2.9999230666	
H14	2.1666842094	-2.4793286188	3.6288899558	
H15	3.0398243445	-1.0476341725	2.8491588813	
N16	-0.3151411952	-0.2210320715	-2.2673198189	
P17	-1.0099909164	1.8893277359	-0.2353169597	
H18	1.2026386203	-5.3564874568	0.3580888722	
H19	1.5340795723	-4.9528330440	2.0514005731	
H20	0.4021506133	-3.9378683602	1.0961707804	
H21	3.4558833885	-1.3843806675	-1.4135071228	
H22	3.2005713018	-0.4583710791	0.0668447451	
H23	4.1612766875	-4.2676650211	1.8487845849	
H24	3.8539533942	-5.2762285658	0.4391254461	
C25	-0.1707710001	-1.5388970127	-2.9376804206	
C26	0.5848833580	-1.1678773667	-4.2428486570	
O27	0.3965718734	0.2867028847	-4.3570544158	
C28	-0.0234016542	0.6990763044	-3.1184031584	
C29	-0.0901801771	2.1649562186	-2.9507016808	
C30	-0.5336065898	2.8158246961	-1.7713310450	
H31	0.1849787453	-1.6192735173	-5.1539130856	
H32	1.6635538895	-1.3603616856	-4.1744656024	
H33	0.4335428166	-2.1672215716	-2.2734206758	
C34	-1.5507008039	-2.2342628594	-3.1622908909	
C35	-1.3000056189	-3.5938690621	-3.8631928080	
H36	-0.8746059321	-3.4698997206	-4.8704068576	
H37	-0.6150941371	-4.2177932248	-3.2693802954	
H38	-2.2505255673	-4.1371046791	-3.9707983782	
C39	-2.4955577412	-1.3642139207	-4.0210720513	
H40	-2.0550758561	-1.1204001307	-4.9998245432	
H41	-3.4376912446	-1.9035547547	-4.2023374903	
H42	-2.7378972057	-0.4251752261	-3.5037520189	
C43	-2.2033919921	-2.5068433753	-1.7866159330	
H44	-3.1236858616	-3.0968506834	-1.9199341474	
H45	-1.5148537869	-3.0636939639	-1.1349875996	
H46	-2.4641750329	-1.5673286597	-1.2833208377	
C47	-0.7351988819	5.1428542117	3.0777342034	
C48	-1.9902940587	4.7276393083	2.6361133318	
C49	-2.1006118621	3.7627045552	1.6327710106	
C50	-0.9499837117	3.1987892116	1.0607850552	
C51	0.3118483935	3.6159099203	1.5230068995	
C52	0.4159497924	4.5860115161	2.5170725798	
H53	-0.6524707600	5.8936163832	3.8605807615	

H54	-2.8904676073	5.1553437529	3.0726229093
H55	-3.0854016586	3.4503657429	1.2964933333
H56	1.2147396958	3.1780202374	1.1024854807
H57	1.3977544977	4.9010793435	2.8635438565
C58	-5.5721822676	1.0976674868	-0.6537203254
C59	-4.9793523413	2.0817837082	-1.4447955083
C60	-3.6099851672	2.3339098202	-1.3456845603
C61	-2.8160674451	1.5952320583	-0.4544319317
C62	-3.4200745858	0.5953567211	0.3276738616
C63	-4.7910605004	0.3570153269	0.2342728769
H64	-6.6399550725	0.9052587831	-0.7322854924
H65	-5.5846466717	2.6582142658	-2.1414020266
H66	-3.1622611951	3.1085298952	-1.9630976861
H67	-2.8092648528	0.0016576511	1.0050616425
H68	-5.2473410856	-0.4162861624	0.8481598024
C69	-0.5766164268	4.2191986417	-1.7623943464
H70	-0.9095791948	4.7354282688	-0.8675032058
C71	0.3066134829	2.9389340228	-4.0524075135
H72	0.6546991590	2.4236847708	-4.9406427424
C73	-0.1888534396	4.9709030312	-2.8705946740
H74	-0.2348239360	6.0568682123	-2.8287902764
C75	0.2606249955	4.3274412581	-4.0192348967
H76	0.5763238190	4.9016115004	-4.8869892838
C77	4.9562887125	-3.4610120967	-0.0033264612
H78	4.8661386095	-3.6803287843	-1.0812050486
H79	5.9587135159	-3.7919330880	0.3110557758
C80	4.7963226149	-1.9385504327	0.2112779548
H81	5.6071565314	-1.3895053766	-0.2934912805
H82	4.8835268496	-1.7225425270	1.2893223041

Complex 9

	(B3LYP)		
Gas Phase Energy	= -2032.746151	86694	= Eh
Solvent Phase Energy	= -0.01150958	455	= Eh
Zero Point Energy	= 434.278	= kc	al/mol
Frequency	= -310.	74	

(PBE)

Gas Phase Energy = -2030.54481985386 = Eh

		2 -	
Pd1	-0.0043666097	0.0634387862	0.0584997996
C2	0.0070301491	0.1093603823	2.2104800572
C3	1.3809836889	0.0923046314	2.5751574707
C4	2.1337719483	1.2231012929	2.9104985345
C5	3.2615043957	2.1293264681	1.3058756248
O6	1.6508404679	1.4960421691	-0.2819578274
C7	2.1093206112	2.3759665974	0.5197675089
C8	1.2867406393	3.6483812887	0.7117120749
C9	4.1528897960	0.9682068027	0.9203252161
C10	3.9828945429	3.2963699640	1.9819053361
H11	-0.5402218720	1.0258575014	2.4494500813
H12	-0.5664594077	-0.7887870614	2.4359080111
H13	1.8911886327	-0.8702890690	2.5576346448

H14	3.0695012544	1.0739744176	3.4420746077
H15	1.5994052343	2.1238049488	3.2026376188
N16	0.0598027013	-0.1507606747	-2.3652811753
P17	-1 8988267118	-1 1875206227	-0 3477454845
H18	4 8116316350	1 2485030675	0.0735001053
н10 H10	1.0110510550	0.6703548080	1 7587025662
LI19	3 563/780877	0.0793348089	0.6127374514
1120	1 200 4 22 7 6 1 1	1 1 2 6 4 6 4 4 1 5 6	0.012/3/4314
П21 1122	1.2064257011	4.1304044130	-0.2/9098004/
H22	0.2541/49651	3.3496393037	0.9756055892
H23	3.98/3328/09	3.1615/144/5	3.0835539236
H24	5.0493474817	3.2771056818	1.6831225435
C25	1.3309108023	0.1810614157	-3.0550807095
C26	0.8910225442	1.2580287268	-4.0897827309
O27	-0.5680682715	1.1190895635	-4.1437226055
C28	-0.9058379715	0.3663336690	-3.0417826594
C29	-2.3556196621	0.2503022358	-2.8003711737
C30	-2.9294723790	-0.4275592628	-1.6914924094
H31	1.2716436199	1.1104281738	-5.1038867802
H32	1.1212299331	2.2758671052	-3.7497625271
H33	2.0131888195	0.6092412024	-2.3126094410
C34	1.9944710527	-1.0877589318	-3.6779034594
C35	3.2954968468	-0.6618860827	-4.4036098376
H36	3.0945550658	-0.0098999411	-5.2665946508
H37	3 9712282665	-0 1305765168	-3 7158460289
H38	3 8216624034	-1 5526772970	-4 7781852931
C39	1 0427151133	-1.7914018068	-4 6709814172
U37 Н40	0.7070600166	1 1106221/18	5 /682127000
11 4 0 1141	1 5575815220	-1.1100221418	5 1462602127999
1141 1142	0.1565272175	2.0401033097	-3.1402092117
П42 С42	0.1303372173	-2.1/90052194	-4.149/964029
C45	2.3003733904	-2.0030/30/41	-2.3393839493
H44	2.8776872764	-2.94/8889243	-2.9550/2/816
H45	3.0424607685	-1.5817340672	-1.8185428345
H46	1.4711677153	-2.4015219368	-2.0023770477
C47	-4.9857758571	-1.8215422525	3.0822534802
C48	-4.6702699354	-2.8969135612	2.2525319378
C49	-3.7572341243	-2.7377091227	1.2083770539
C50	-3.1437451996	-1.4952556675	0.9828513043
C51	-3.4608629400	-0.4212893720	1.8344622973
C52	-4.3797126106	-0.5822574108	2.8696882541
H53	-5.6966301602	-1.9495905712	3.8958736944
H54	-5.1374082326	-3.8659337610	2.4167936563
H55	-3.5227954964	-3.5838104971	0.5681301190
H56	-2.9803705094	0.5434447887	1.6830382378
H57	-4.6170652447	0.2585608854	3.5180215106
C58	-1 0423446840	-5 5146784482	-1 8369275927
C59	-2.0652162923	-4 7811237959	-2 4378130859
C60	-2.3378093279	-3 4776209438	-2.4570150055
C61	1 5817506032	2 8888154214	0.001670/820
C61	-1.3017390932	2 6217485021	-0.9910794029
C62	-0.3433110004	-3.031/403021	-0.4041133909
	-0.20309/1403	-4.7302010273	-0.01/1342313
H04	-0.8343014004	-0.3311394994	-2.1040302183
HOD	-2.05/25109/1	-5.2253558197	-5.2355813982
H66	-3.1436/69655	-2.92040/6/42	-2.4882240108
H67	0.0643119259	-3.17/0472487	0.3757621197
H68	0.5194360829	-5.5035453497	-0.3491395012
C69	-4.3309567383	-0.4629720977	-1.5913714220

H70	-4.7900339729	-0.9696964429	-0.7484630146
C71	-3.1991766733	0.8608780164	-3.7429400764
H72	-2.7407847681	1.3829060717	-4.5756876027
C73	-5.1507369355	0.1403645374	-2.5425295950
H74	-6.2320589425	0.0880192955	-2.4339354764
C75	-4.5818586235	0.8083042804	-3.6235973487
H76	-5.2103802919	1.2895048275	-4.3693208260
C77	3.3935843916	4.6746888398	1.6482914839
H78	3.6828770373	4.9697891388	0.6253733622
H79	3.8019437690	5.4363281057	2.3301100420
C80	1.8507165035	4.6395234511	1.7395643891
H81	1.4323039335	5.6436565142	1.5706238144
H82	1.5613964410	4.3467043411	2.7621993770
********	******************	******	*****

_____ Complex 10

_____ (B3LYP)

Gas Phase Energy	= -2032.79265164	845 = Eh
Solvent Phase Energy	= -0.0132024782	= Eh
Zero Point Energy	= 436.003	= kcal/mol

(PBE)

Gas Phase Energy = -2030.58452882899 = Eh

Geometry	Coordinates
----------	-------------

	Geoinet	ry Coordinates	
Pd1	0.0720538868	0.0428463229	0.0099922639
C2	0.1823952393	0.0807276368	3.1441744177
C3	0.9868244414	-0.0413482100	4.4706878791
C4	1.0117446381	0.4512350308	1.9246127186
C5	0.8479732080	1.6552857713	1.2186102544
C6	1.9459401444	-1.2505665304	4.3933555328
O 7	1.7797394260	-2.1590009219	3.5946976409
C8	3.0834756682	-1.3205780504	5.4113248468
C9	-0.0108204601	-0.3537065252	5.6196570138
C10	1.7656846439	1.2674364822	4.7784009153
H11	1.6760690159	2.0855625627	0.6590397637
H12	0.0702612855	2.3606437920	1.5081520885
H13	1.9744286139	-0.0554575430	1.8517660042
H14	-0.3256968273	-0.8772717915	2.9927053433
H15	-0.6035676672	0.8337681560	3.2886683122
N16	0.0484820169	-2.2853000136	-0.3976119107
P17	-0.7102341132	0.0214395134	-2.2148560190
H18	3.7706645397	-2.1220425249	5.0937819529
H19	2.6427740306	-1.6388325815	6.3782108942
H20	0.4849699728	-0.4715208688	6.5985465661
H21	-0.7381239401	0.4726328396	5.7119503953
H22	-0.5731314347	-1.2806682257	5.4109840625
H23	2.2926459144	1.5902870483	3.8618999508
H24	1.0323602254	2.0635735350	5.0074200204
C25	0.0609439448	-3.2580504675	0.7255767213
C26	1.2631248171	-4.1793755031	0.3705211187
O27	1.5050457660	-3.9039475631	-1.0474204185
C28	0.8256728704	-2.7365105650	-1.3214621184
C29	1.1008358857	-2.1782701010	-2.6615948386
C30	0.4890408738	-1.0109593988	-3.1967527048

H31	1.0716161971	-5.2510148996	0.4680682596
H32	2.1650133576	-3.9099959040	0.9350579742
H33	0.2646088251	-2.7222304050	1.6595664200
C34	-1.3072661327	-3.9997669336	0.8635278086
C35	-1.2098812150	-4.9996551446	2.0435941864
H36	-0 4866097167	-5 8042555496	1 8447936330
H37	-0.9092094092	-4 4848574196	2 9689330704
H38	-2 1899178524	-5 4701321508	2 2140042606
C39	-1 6726280358	-4 7520967625	-0 4362427854
H40	-0.8854136775	-5 4613605143	-0 7329157546
H41	-2 6021536753	-5 3227565150	-0 2879394314
H42	-1 8358339521	-4 0468267949	-1 2624684923
C43	-2.4121125123	-2.9666683809	1 1845480490
С 4 5 Н44	3 37//282653	3 /81/8731/5	1 331/55808/
1144 1145	2 1726515138	2 4156700105	2 1068300847
114J 114J	2 5240247405	-2.4130799193	0.3673154852
C47	4 8363112080	2.2441552582	2 0723084437
C47	-4.0303112900	-2.0144393340	-2.9123904431
C48	-5.7515922214	-2.5705750274	-5.7720246755
C49	-2.3013840374	-1.7803920203	-3.3080041311
C50	-2.5200197965	-0.8280844018	-2.3322219347
C51 C52	-3.4183013847	-0.48/3280143	-1./44232/129
C52	-4.66/424000/	-1.0694183139	-1.9590789706
H53	-5.8093280350	-2.4/311/33/4	-3.1356856461
H54	-3.8/6561992/	-3.10//399192	-4.5635379229
H55	-1.6666922446	-2.0622871319	-4.204/590342
H56	-3.288/940948	0.23/626/5/2	-0.9428108291
H5/	-5.5094217762	-0./901086644	-1.3290662510
C58	-1.0241618164	3.935/099438	-4.7255940687
C59	-0.1324344483	3.823/060081	-3.6579522014
C60	-0.0645545666	2.6401959335	-2.923/900966
C61	-0.8/53930000	1.5430660785	-3.258/2006/5
C62	-1.//45918263	1.0092103034	-4.3300014785
C63	-1.84/922/101	2.8591134108	-5.0561857332
H64	-1.0842297636	4.8625131638	-5.2926029702
H65	0.5033051520	4.6649184764	-3.3890527821
H66	0.60/8/59339	2.5615382272	-2.0710561244
H67	-2.4238350807	0.8383667700	-4.5961507454
H68	-2.5518526350	2.9460398511	-5.8818049236
C69	0.8634098162	-0.6036993693	-4.48/4960931
H70	0.4144613913	0.2885664182	-4.9115175531
C/1	2.0446382301	-2.8727927175	-3.43/06466/6
H72	2.5050639468	-3.7562853405	-3.0096107953
C/3	1.8006810884	-1.30/92/9906	-5.2410127888
H74	2.0638814782	-0.9595474772	-6.2377039883
C/5	2.3948084026	-2.449/335188	-4.7128787974
H76	3.1313153560	-3.0074737718	-5.2868418177
C//	3.8063600601	0.0213914446	5.6105285093
H7/8	4.5404854526	-0.0664440798	6.4252479976
H79	4.3696318528	0.2801472431	4.6984207991
C80	2.7902258113	1.1494348280	5.913950/169
H81	2.2935087270	0.9442125103	6.8766427542
Hð2	3.3219/81402	2.1039402224	0.0322300343

CALIFORNIA INSTITUTE OF TECHNOLOGY BECKMAN INSTITUTE X-RAY CRYSTALLOGRAPHY LABORATORY

Date 20 July 2004

Crystal Structure Analysis of:

DCB24

(shown below)

For	Investigator: Doug Behe	nna	ext. 2116
	Advisor: B. M. Stoltz		ext. 6064
	Account Number:	BMS.JandJ-2.11-GR	ANT.000006
By	Michael W. Day	116 Beckman e-mail: mikeday	ext. 2734 @caltech.edu

Contents

- Table 1. Crystal data
- Figures Figures
- Table 2. Atomic Coordinates
- Table 3. Selected bond distances and angles
- Table 4. Full bond distances and angles
- Table 5. Anisotropic displacement parameters
- Table 6. Hydrogen bond distance and angle
- Table 7. Observed and calculated structure factors (available upon request)



DCB24

Note: The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 245187. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 245187."

Empirical formula	$[C_{28}H_{31}NOPPd]^+ PF_6^- \cdot \frac{1}{2}C_2H_5OH$
Formula weight	702.91
Crystallization Solvent	Ethanol
Crystal Habit	Fragment
Crystal size	0.35 x 0.34 x 0.23 mm ³
Crystal color	Colorless
Data Colle	ection
Type of diffractometer	Bruker SMART 1000
Wavelength	0.71073 Å MoKα
Data Collection Temperature	100(2) K
θ range for 15322 reflections used in lattice determination	2.31 to 41.00°
Unit cell dimensions	
Volume	3004.98(18) Å ³
Z	4
Crystal system	Monoclinic
Space group	C2
Density (calculated)	1.554 Mg/m ³
F(000)	1428
θ range for data collection	1.77 to 42.31°
Completeness to $\theta = 42.31^{\circ}$	85.0 %
Index ranges	$-32 \leq h \leq 32,-28 \leq k \leq 29,-20 \leq l \leq 15$
Data collection scan type	ω scans at 3 ϕ settings of 20=-28° and 2 at 20=-59°
Reflections collected	28501
Independent reflections	15572 [R _{int} = 0.0351]
Absorption coefficient	0.787 mm ⁻¹
Absorption correction	SADABS
Max. and min. transmission	0.8397 and 0.7702

Table 1. Crystal data and structure refinement for DCB24 (CCDC 245187).

Table 1 (cont.)

Structure solution and Refinement

Structure solution program	SHELXS-97 (Sheldrick, 1990)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 1997)
Refinement method	Full matrix least-squares on F ²
Data / restraints / parameters	15572 / 1 / 408
Treatment of hydrogen atoms	Riding
Goodness-of-fit on F ²	1.343
Final R indices [I>2 σ (I), 13582 reflections]	R1 = 0.0373, wR2 = 0.0725
R indices (all data)	R1 = 0.0459, wR2 = 0.0748
Type of weighting scheme used	Sigma
Weighting scheme used	$w=1/\sigma^2(\text{Fo}^2)$
Max shift/error	0.004
Average shift/error	0.000
Absolute structure parameter	-0.019(13)
Largest diff. peak and hole	1.422 and -0.710 e.Å ⁻³

Special Refinement Details

The propyl ligand, C26-C27-C28, is disordered in two alternate orientations, differing by "up-down" positions for C27. Additional disorder is observed in one PF_6 counterion and an included solvent molecule, modeled as ethanol hydrogen bonded to a fluorine of one counterion.

Refinement of F^2 against ALL reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 , conventional R-factors (R) are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma$ (F^2) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.







Table 2. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters (Å²x 10³) for DCB24 (CCDC 245187). U(eq) is defined as the trace of the orthogonalized U^{ij} tensor.

	X	У	Z	U _{eq}	Occ
Pd(1)	2838(1)	4922(1)	2994(1)	18(1)	1
P(1)	2327(1)	5673(1)	4282(1)	19(1)	1
O(1)	3784(1)	7349(1)	2545(1)	22(1)	1
N(1)	3187(1)	6087(1)	2419(2)	17(1)	1
C(1)	3099(1)	6431(1)	5056(2)	18(1)	1
C(2)	3275(1)	6569(1)	6320(2)	23(1)	1
C(3)	3845(1)	7170(1)	6920(2)	24(1)	1
C(4)	4208(1)	7669(1)	6242(2)	25(1)	1
C(5)	4040(1)	7544(1)	4982(2)	21(1)	1
C(6)	3512(1)	6903(1)	4381(2)	17(1)	1
C(7)	3458(1)	6748(1)	3074(2)	17(1)	1
C(8)	3836(1)	6994(1)	1381(2)	25(1)	1
C(9)	3264(1)	6242(1)	1155(2)	20(1)	1
C(10)	2441(1)	6383(1)	190(2)	23(1)	1
C(11)	2594(2)	6514(2)	-1064(2)	33(1)	1
C(12)	1931(1)	5586(2)	100(2)	29(1)	1
C(13)	2015(1)	7156(1)	510(2)	28(1)	1
C(14)	2024(1)	5153(1)	5500(2)	26(1)	1
C(15)	2543(2)	4578(2)	6254(2)	33(1)	1
C(16)	2337(2)	4162(2)	7193(2)	41(1)	1
C(17)	1591(2)	4315(2)	7372(3)	45(1)	1
C(18)	1070(2)	4884(2)	6615(2)	43(1)	1
C(19)	1277(2)	5303(2)	5676(2)	35(1)	1
C(20)	1492(1)	6337(1)	3474(2)	22(1)	1
C(21)	888(1)	5982(1)	2497(2)	24(1)	1
C(22)	256(1)	6481(2)	1823(2)	33(1)	1
C(23)	218(2)	7332(2)	2102(3)	50(1)	1
C(24)	812(2)	7686(2)	3061(4)	63(1)	1
C(25)	1439(1)	7192(2)	3747(3)	44(1)	1
C(26)	3314(2)	3965(2)	1935(3)	41(1)	1
C(27A)	3201(3)	3620(2)	2979(5)	30(1)	0 563(11)
C(27B)	2776(4)	3655(3)	2334(7)	32(2)	0.333(11) 0.437(11)
C(28)	2576(3)	3683(2)	3388(3)	56(1)	1
P(2)	0	3840(1)	0	18(1)	1
F(1)	0	2828(1)	0	36(1)	1
F(2)	0	4853(2)	0	32(1)	1
F(3)	955(3)	3812(3)	530(5)	49(1)	0.77(3)
F(4)	-75(4)	3827(4)	1359(6)	51(1)	0.77(3)
F(3B)	900(13)	3927(11)	140(50)	100(9)	0.23(3)
F(4B)	170(30)	3897(16)	1430(20)	94(10)	0.23(3)
P(3)	5000	4813(1)	0	36(1)	1
F(5)	5000	3850(3)	0	210(4)	1
F(6)	5000	5758(3)	0	223(5)	1
F(7)	4919(1)	4802(3)	1329(2)	120(2)	1
F(8)	4076(1)	4798(4)	-481(2)	120(1)	1

C(30)	4962(4)	5145(5)	5617(6)	55(4)	0.50
C(31)	5035(8)	5160(5)	4359(7)	106(8)	0.50
O(30)	5453(3)	4644(3)	3892(5)	68(2)	0.50

171.47(9)
37.19(19)
134.59(17)
36.07(16)
135.98(13)
23.7(2)
67.92(10)
103.55(9)
33.40(17)
36.35(15)
101.37(8)
87.15(5)
136.98(16)
132.75(14)
168.99(7)

Table 3. Selected bond lengths [Å] and angles [°] for DCB24 (CCDC 245187).

Pd(1)-C(28)	2.087(2)	P(3)-F(6)	1.492(5)
Pd(1)-N(1)	2.1020(15)	P(3)-F(5)	1.520(5)
Pd(1)-C(27B)	2.127(4)	P(3)-F(8)	1.549(2)
Pd(1)-C(27A)	2.153(4)	P(3)-F(8)#2	1.549(2)
Pd(1)-C(26)	2.239(2)	P(3)-F(7)#2	1.5592(19)
Pd(1)-P(1)	2.2639(5)	P(3)-F(7)	1.5592(19)
P(1)-C(20)	1.814(2)	C(30)-C(31)	1.4740
P(1)-C(14)	1.8179(19)	C(31)-O(30)	1.3080
P(1)-C(1)	1.8262(18)		
O(1)-C(7)	1.337(2)	C(28)-Pd(1)-N(1)	171.47(9)
O(1)-C(8)	1.465(2)	C(28)-Pd(1)-C(27B)	37.19(19)
N(1)-C(7)	1.287(2)	N(1)-Pd(1)-C(27B)	134.59(17)
N(1)-C(9)	1.502(2)	C(28)-Pd(1)-C(27A)	36.07(16)
C(1)-C(2)	1.397(3)	N(1)-Pd(1)-C(27A)	135.98(13)
C(1)-C(6)	1.412(2)	C(27B)-Pd(1)-C(27A)	23.7(2)
C(2)-C(3)	1.399(3)	C(28)-Pd(1)-C(26)	67.92(10)
C(3)-C(4)	1.382(3)	N(1)-Pd(1)-C(26)	103.55(9)
C(4)-C(5)	1.390(3)	C(27B)-Pd(1)-C(26)	33.40(17)
C(5)-C(6)	1.406(2)	C(27A)-Pd(1)-C(26)	36.35(15)
C(6)-C(7)	1.482(3)	C(28)-Pd(1)-P(1)	101.37(8)
C(8)-C(9)	1.526(3)	N(1)-Pd(1)-P(1)	87.15(5)
C(9)-C(10)	1.549(3)	C(27B)-Pd(1)-P(1)	136.98(16)
C(10)-C(13)	1.529(3)	C(27A)-Pd(1)-P(1)	132.75(14)
C(10)-C(12)	1.529(3)	C(26)-Pd(1)-P(1)	168.99(7)
C(10)-C(11)	1.541(3)	C(20)-P(1)-C(14)	105.92(10)
C(14)-C(15)	1.388(3)	C(20)-P(1)-C(1)	103.80(8)
C(14)-C(19)	1.401(3)	C(14)-P(1)-C(1)	105.63(9)
C(15)-C(16)	1.388(3)	C(20)-P(1)-Pd(1)	112.83(6)
C(16)-C(17)	1.401(4)	C(14)-P(1)-Pd(1)	121.10(7)
C(17)-C(18)	1.386(5)	C(1)-P(1)-Pd(1)	106.03(6)
C(18)-C(19)	1.391(3)	C(7)-O(1)-C(8)	106.45(14)
C(20)-C(25)	1.394(3)	C(7)-N(1)-C(9)	107.51(15)
C(20)-C(21)	1.407(3)	C(7)-N(1)-Pd(1)	128.28(14)
C(21)-C(22)	1.391(3)	C(9)-N(1)-Pd(1)	123.75(12)
C(22)-C(23)	1.386(4)	C(2)-C(1)-C(6)	118.94(16)
C(23)-C(24)	1.385(4)	C(2)-C(1)-P(1)	120.43(14)
C(24)-C(25)	1.385(4)	C(6)-C(1)-P(1)	120.61(14)
C(26)-C(27B)	1.259(6)	C(1)-C(2)-C(3)	121.19(18)
C(26)-C(27A)	1.372(6)	C(4)-C(3)-C(2)	119.57(19)
C(27A)-C(28)	1.314(6)	C(3)-C(4)-C(5)	120.13(18)
C(27B)-C(28)	1.345(7)	C(4)-C(5)-C(6)	120.84(18)
P(2)-F(3B)	1.543(18)	C(5)-C(6)-C(1)	119.03(17)
P(2)-F(3B)#1	1.543(18)	C(5)-C(6)-C(7)	116.43(15)
P(2)-F(4B)	1.57(2)	C(1)-C(6)-C(7)	124.49(15)
P(2)-F(4B)#1	1.57(2)	N(1)-C(7)-O(1)	116.61(16)
P(2)-F(4)	1.590(5)	N(1)-C(7)-C(6)	128.16(16)
P(2)-F(4)#1	1.590(5)	O(1)-C(7)-C(6)	115.06(14)
P(2)-F(2)	1.598(2)	O(1)-C(8)-C(9)	103.71(14)
P(2)-F(1)	1.5982(18)	N(1)-C(9)-C(8)	101.78(15)
P(2)-F(3)	1.603(4)	N(1)-C(9)-C(10)	112.07(15)
P(2)-F(3)#1	1.603(4)	C(8)-C(9)-C(10)	115.76(17)

Table 4. Bond lengths [Å] and angles $[\circ]$ for DCB24 (CCDC 245187).

C(13)-C(10)-C(12)	110.76(17)	F(4)-P(2)-F(2)	90.8(3)
C(13)-C(10)-C(11)	110.13(18)	F(4)#1-P(2)-F(2)	90.8(3)
C(12)-C(10)-C(11)	107.80(17)	F(3B)-P(2)-F(1)	95.1(7)
C(13)-C(10)-C(9)	111.34(16)	F(3B)#1-P(2)-F(1)	95.1(7)
C(12)-C(10)-C(9)	109.42(17)	F(4B)-P(2)-F(1)	93.2(9)
C(11)-C(10)-C(9)	107.27(17)	F(4B)#1-P(2)-F(1)	93.2(9)
C(15)-C(14)-C(19)	119.5(2)	F(4)-P(2)-F(1)	89.2(3)
C(15)-C(14)-P(1)	118.74(16)	F(4)#1-P(2)-F(1)	89.2(3)
C(19)-C(14)-P(1)	121.78(19)	F(2)-P(2)-F(1)	180.0
C(14)-C(15)-C(16)	120.8(2)	F(3B)#1-P(2)-F(3)	164.4(18)
C(15)-C(16)-C(17)	119.7(3)	F(4B)-P(2)-F(3)	75.9(19)
C(18)-C(17)-C(16)	119.5(2)	F(4B)#1-P(2)-F(3)	104.3(19)
C(17)-C(18)-C(19)	120.8(2)	F(4)-P(2)-F(3)	90.6(3)
C(18)-C(19)-C(14)	119.7(3)	F(4)#1-P(2)-F(3)	89.4(3)
C(25)-C(20)-C(21)	118.5(2)	F(2)-P(2)-F(3)	91.6(2)
C(25)-C(20)-P(1)	122.91(16)	F(1)-P(2)-F(3)	88.4(2)
C(21)-C(20)-P(1)	118.53(15)	F(3B)-P(2)-F(3)#1	164.4(18)
C(22)-C(21)-C(20)	120.12(19)	F(4B)-P(2)-F(3)#1	104.3(19)
C(23)-C(22)-C(21)	120.4(2)	F(4B)#1-P(2)-F(3)#1	75.9(19)
C(24)-C(23)-C(22)	119.8(2)	F(4)-P(2)-F(3)#1	89.4(4)
C(25)-C(24)-C(23)	120.2(3)	F(4)#1-P(2)-F(3)#1	90.6(3)
C(24)-C(25)-C(20)	120.9(2)	F(2)-P(2)-F(3)#1	91.6(2)
C(28)-C(27A)-C(26)	128.5(5)	F(1)-P(2)-F(3)#1	88.4(2)
C(26)-C(27B)-C(28)	136.7(6)	F(3)-P(2)-F(3)#1	176.8(4)
F(3B)-P(2)-F(3B)#1	169.8(13)	F(6)-P(3)-F(5)	180.000(1)
F(3B)-P(2)-F(4B)	90.7(14)	F(6)-P(3)-F(8)	90.9(2)
F(3B)#1-P(2)-F(4B)	88.7(14)	F(5)-P(3)-F(8)	89.1(2)
F(3B)-P(2)-F(4B)#1	88.7(14)	F(6)-P(3)-F(8)#2	90.9(2)
F(3B)#1-P(2)-F(4B)#1	90.7(14)	F(5)-P(3)-F(8)#2	89.1(2)
F(4B)-P(2)-F(4B)#1	173.5(18)	F(8)-P(3)-F(8)#2	178.3(4)
F(3B)-P(2)-F(4)	105.8(18)	F(6)-P(3)-F(7)#2	90.64(19)
F(3B)#1-P(2)-F(4)	74.3(18)	F(5)-P(3)-F(7)#2	89.36(19)
F(4B)#1-P(2)-F(4)	165(2)	F(8)-P(3)-F(7)#2	92.36(12)
F(3B)-P(2)-F(4)#1	74.3(18)	F(8)#2-P(3)-F(7)#2	87.63(12)
F(3B)#1-P(2)-F(4)#1	105.8(18)	F(6)-P(3)-F(7)	90.64(19)
F(4)-P(2)-F(4)#1	178.4(5)	F(5)-P(3)-F(7)	89.36(19)
F(3B)-P(2)-F(2)	84.9(7)	F(8)-P(3)-F(7)	87.63(12)
F(3B)#1-P(2)-F(2)	84.9(7)	F(8)#2-P(3)-F(7)	92.36(12)
F(4B)-P(2)-F(2)	86.8(9)	F(7)#2-P(3)-F(7)	178.7(4)
F(4B)#1-P(2)-F(2)	86.8(9)	O(30)-C(31)-C(30)	127.1

Symmetry transformations used to generate equivalent atoms: #1 -x,y,-z #2 -x+1,y,-z

Table 5. Anisotropic displacement parameters (Å²x 10⁴) for DCB24 (CCDC 245187). The anisotropic displacement factor exponent takes the form: $-2\pi^2$ [h² a^{*2}U ¹¹ + ... + 2 h k a^{*} b^{*} U¹²]

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
$\underline{\mathbf{Pd}(1)}$	249(1)	152(1)	169(1)	-17(1)	102(1)	-31(1)
P(1)	249(1) 209(2)	217(2)	157(2)	-38(2)	86(2)	-57(2)
O(1)	270(6)	206(6)	201(7)	-11(5)	116(5)	-71(5)
N(1)	270(0) 204(7)	184(6)	156(8)	-27(5)	87(6)	-26(5)
C(1)	165(7)	215(7)	168(9)	-20(6)	54(6)	-17(5)
C(2)	219(8)	303(9)	177(9)	-41(7)	90(6)	-23(6)
C(2)	205(8)	354(10)	166(9)	-79(7)	51(7)	-6(7)
C(4)	184(8)	309(9)	235(10)	-102(7)	43(6)	-28(6)
C(5)	187(0) 183(7)	219(7)	217(9)	-44(6)	61(6)	-25(5)
C(6)	152(6)	186(6)	157(8)	-25(5)	45(5)	-7(5)
C(0)	152(0)	171(6)	187(8)	0(5)	69(5)	-16(5)
C(8)	339(10)	257(9)	205(10)	-22(7)	156(8)	-94(7)
C(0)	261(8)	209(7)	169(9)	-16(6)	122(6)	-43(6)
C(10)	297(9)	265(8)	146(9)	-36(6)	90(7)	-33(7)
C(10)	459(13)	381(12)	140(9) 182(11)	8(8)	141(9)	-24(9)
C(12)	310(10)	332(10)	208(10)	-58(8)	62(7)	-86(8)
C(12)	301(10)	319(10)	200(10) 203(10)	15(8)	62(7)	20(7)
C(14)	351(10)	292(9)	196(10)	-70(6)	156(8)	-132(7)
C(15)	479(13)	311(10)	261(12)	-10(8)	201(10)	-86(9)
C(16)	698(18)	313(11)	271(12)	-1(9)	233(12)	-136(11)
C(10)	830(20)	342(12)	308(13)	-148(10)	375(12)	-305(13)
C(18)	540(13)	487(13)	403(12)	-178(16)	373(14) 351(11)	-278(15)
C(10)	403(12)	419(12)	300(12)	-104(9)	225(10)	-149(9)
C(20)	172(7)	261(8)	228(10)	-94(7)	67(6)	-52(6)
C(20)	232(8)	267(8)	226(10)	-87(7)	70(7)	-72(6)
C(21)	232(0) 221(9)	419(12)	325(13)	-131(9)	26(8)	-40(8)
C(22)	221(5) 261(11)	445(14)	680(20)	-170(14)	-48(11)	86(10)
C(24)	311(13)	421(15)	980(30)	-389(17)	-102(15)	122(11)
C(24)	221(10)	410(13)	591(18)	-309(17)	-44(10)	35(8)
C(26)	668(17)	230(10)	422(16)	-52(9)	325(14)	39(10)
C(27A)	400(30)	150(14)	370(30)	$\frac{32(9)}{4(14)}$	120(20)	-16(13)
C(27R)	400(30) 440(40)	154(18)	410(40)	-103(18)	190(30)	-37(18)
C(28)	1130(30)	165(9)	595(19)	100(10)	600(20)	-74(12)
0(20)	1150(50)	105())	555(15)	10(10)	000(20)	/ 1(12)
P(2)	208(3)	164(2)	159(3)	0	52(2)	0
F(1)	453(11)	158(7)	586(14)	0	334(10)	0
F(2)	458(9)	159(8)	316(9)	0	61(7)	Ő
F(3)	201(11)	356(18)	770(30)	35(16)	-63(14)	-7(11)
F(4)	890(30)	440(20)	310(20)	47(14)	357(19)	125(16)
F(3B)	390(70)	340(50)	2400(300)	500(110)	640(110)	170(50)
F(4B)	2200(300)	320(60)	120(60)	130(40)	150(120)	130(130)
P(3)	413(4)	349(6)	402(5)	0	235(3)	0
F(5)	2790(100)	370(20)	3230(110)	0	1010(80)	0
F(6)	4250(130)	290(20)	3080(100)	0	2540(100)	0
F(7)	772(14)	2450(40)	439(11)	130(20)	293(10)	-630(20)
F(8)	562(12)	2360(50)	671(14)	-350(30)	185(11)	-300(20)

C(30)	250(30)	870(110)	530(50)	330(50)	120(30)	250(40)
C(31)	890(100)	540(90)	1800(200)	-350(90)	380(130)	-80(80)
O(30)	770(30)	560(30)	640(30)	-10(20)	130(30)	380(20)

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
O(30)-H(30)F(7)	0.85	1.99	2.799(6)	157.8

Table 6. Hydrogen bonds for DCB24 (CCDC 245187) [Å and °].

Symmetry transformations used to generate equivalent atoms: #1 -x,y,-z #2 -x+1,y,-z