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Photoanodic behavior of vapor-liquid-solid-grown, lightly doped, crystalline Si microwire arrays[†]

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Arrays of n-Si microwires have to date exhibited low efficiencies when measured as photoanodes in contact with a 1-1'-dimethylferrocene (Me₂Fc^{+/0})–CH₃OH solution. Using high-purity Au or Cu catalysts, arrays of crystalline Si microwires were grown by a vapor-liquid-solid process without dopants, which produced wires with electronically active dopant concentrations of 1×10^{13} cm⁻³. When measured as photoanodes in contact with a Me₂Fc^{+/0}– CH₃OH solution, the lightly doped Si microwire arrays exhibited greatly increased fill factors and efficiencies as compared to n-Si microwires grown previously with a lower purity Au catalyst. In particular, the Cu-catalyzed Si microwire array photoanodes exhibited open-circuit voltages of ~0.44 V, carrier-collection

efficiencies exceeding ~0.75, and an energy-conversion efficiency of 1.4% under simulated air mass 1.5 G illumination. Lightly doped Cu-catalyzed Si microwire array photoanodes have thus demonstrated performance that is comparable to that of optimally doped p-type Si microwire array photocathodes in photoelectrochemical cells.

The structuring of semiconductors on the nano- and micro-scale is a promising approach for the fabrication of scalable and efficient devices for the production of electricity and fuels from sunlight.^{1–3} In contrast to a traditional geometry that is characterized by planar light absorbers and planar electrical junctions, wire-based architectures orthogonalize the directions of light absorption and carrier collection.⁴ Such a structure provides both a long optical path length for efficient light absorption and a short distance for minority-carrier collection, therefore allowing the incorporation of defective materials with short minority-carrier diffusion lengths into devices that can produce high energy-conversion efficiencies. Specifically, devicephysics modeling of radial junction p-n Si microwires has predicted that crystalline Si microwire arrays can yield efficiencies comparable to those achieved by wafer-based planar Si solar cells, even for Si microwires that have a 1 μ m minority-carrier diffusion length.⁴

Arrays of crystalline p-Si microwires grown by the vapor–liquid– solid (VLS) process have demonstrated promising performance in regenerative^{5,6} and fuel–forming⁷ photoelectrochemical cells, as well as in photovoltaic devices.^{8,9} For example, p-type Si microwire array

Broader context

To make solar energy cost-competitive with fossil fuels and thus drive the large-scale deployment of photovoltaics, significant technological advances must be made to reduce the installed cost of photovoltaics to < \$1/Watt_{peak}. Recently, Si wire arrays grown by the vapor–liquid–solid technique have emerged as a promising technology for the fabrication of efficient and inexpensive photovoltaics, as well as for artificial photosynthetic devices. The thin-film Si wire array photovoltaic technology utilizes an atmospheric pressure, chemical vapor deposition growth process, inexpensive Si precursors, and earth-abundant catalysts, and produces flexible, crystalline Si solar cells that have the potential to approach the efficiencies of wafer-based crystalline Si solar cells. Herein we demonstrate improved performance of Si microwire array photoanodes in contact with a regenerative non-aqueous electrolyte that provides a conformal, high barrier-height contact to structured semiconductor materials. Hence, lightly doped Cu–catalyzed Si microwire array photoanodes can achieve performance comparable to that of optimally doped p-type Si microwire array photocathodes.

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photoelectrodes in contact with an aqueous methyl viologen ($MV^{2^{+/+}}$) redox system have yielded open-circuit voltages (V_{oc}) approaching 0.45 V under 100 mW cm⁻² of simulated air mass (AM) 1.5 G illumination, with near-unity internal quantum yields, demonstrating the efficient radial collection of carriers in the wire-array geometry. Arrays of radial junction n⁺p-Si microwires have demonstrated thermodynamically based photoelectrode energy-conversion efficiencies of >5% for the production of H₂ from H₂O. Analogous arrays in solid-state photovoltaic devices have achieved an efficiency of 7.8%, with V_{oc} exceeding 0.5 V, under simulated AM 1.5 G illumination.

In contrast, n-Si microwire array photoanodes in contact with a 1-1'-dimethylferrocene (Me₂Fc^{+/0})-CH₃OH solution under simulated AM 1.5 conditions have only exhibited V_{oc} values of 0.39 V, in conjunction with low fill factors and low short-circuit photocurrent densities (J_{sc}), resulting in photoelectrode efficiencies, η , of ~0.1%.¹⁰ Given that the Me₂Fc^{+/0}–CH₃OH electrolyte in contact with planar, crystalline n-Si photoanodes produces Voc values that are only limited by bulk recombination/generation,¹¹ the comparatively low performance of n-Si microwire array photoanodes is presumably indicative of the inferior material quality of the n-Si microwires. Two factors may have contributed to the poor electronic quality of the VLS-grown n-Si microwires: the purity of the catalyst used and the choice of metal catalyst. The n-Si microwire arrays were grown with a 99.999% (5N) Au VLS catalyst,^{10,12-14} as compared to the higher purity 99.9999% (6N) Cu catalyst that has been used to grow p-Si microwires.^{5,7,9,15-18} Uncontrolled impurities in the 5N Au catalyst produced microwires that had high variability in the observed electronically active dopant concentrations, $N_{\rm d}$, with values ranging between 1×10^{14} and 1×10^{20} cm⁻³.¹⁶ Additionally, the use of Au as the VLS catalyst, as opposed to Cu, may have limited the efficiency of the Si microwire arrays. Although both Au and Cu form mid-gap traps in Si, in planar Si solar cells Cu has a less detrimental effect than Au, with a minority-carrier lifetime degradation threshold concentration of 4 \times 10¹⁷ cm⁻³ for Cu as compared to 3×10^{13} cm⁻³ for Au.^{19,20} The focus of this work was to determine whether the low photoelectrode efficiencies observed for n-Si microwire arrays in contact with Me₂Fc^{+/0}-CH₃OH are an inherent, fundamental property of the system or whether improved performance could be obtained through control over the electronic properties of the bulk and surface chemistry of Si wire array photoelectrodes.

We report herein the photoelectrochemical behavior of Si microwire arrays that have been fabricated using a 6N VLS catalyst, for both Au and Cu. The device performance of the Si microwire arrays was probed using the Me₂Fc^{+/0}–CH₃OH junction, because this system provides a conformal contact to the microwires, and obviates the need to fabricate a diffused metallurgical junction. Additionally, the Me₂Fc^{+/0}–CH₃OH system generates an inversion layer in contact with n-Si, and the resulting semiconductor/liquid interface has a low effective surface recombination velocity.²¹

To produce the desired photoanodes, arrays of square–packed Si microwires were grown on a planar n⁺-Si(111) substrate using the VLS process, without dopants, but with 6N Au as the growth catalyst (see ESI†). The resulting Si wires were oriented in the (111) direction, with diameters of 2.25–3.0 μ m and heights of 65–75 μ m, with an average areal packing fraction (η_f) of 11.0%. Four-point resistance and gate-dependent conductance measurements indicated that the undoped silicon microwires were nominally p-type, with resistivities

of 800 \pm 700 Ω -cm, corresponding to an acceptor concentration, $N_{\rm a}$, of $\sim 1 \times 10^{13}$ cm⁻³ (Figure S1).

Current density vs. potential (*J*–*E*) measurements of Au-catalyzed Si microwire array photoanodes, and of control photoanodes in which the wires had been physically removed after growth on the n⁺-Si substrate, were measured in contact with 200 mM Me₂Fc-0.4 mM Me₂FcBF₄ in CH₃OH under 100 mW cm⁻² of simulated AM 1.5 G illumination (Fig. 1A). The Si wire array photoanodes exhibited $V_{\rm oc} = 334 \pm 21$ mV, $J_{\rm sc} = 10.0 \pm 1.3$ mA cm⁻², and fill-factors, ff = 0.34 ± 0.05 , with a photoelectrode energy-conversion efficiency $\eta =$ $1.1 \pm 0.3\%$. The n⁺-Si(111) control substrates for which the wires had been physically removed after growth produced $V_{\rm oc} = 233 \pm 38$ mV, $J_{\rm sc} = 1.0 \pm 0.2$ mA cm⁻², and $ff = 0.20 \pm 0.04$. The observed properties of the Si microwire array photoelectrodes can therefore predominately be attributed to the behavior of the VLS–grown crystalline Si wires in contact with the Me₂Fc⁺⁰–CH₃OH electrolyte.

The behavior of the same Au-catalyzed Si microwire photoanodes was also measured in contact with a higher concentration of the oxidized form of the redox couple, 25 mM Me₂FcBF₄, to reduce the concentration overpotential losses at the photoelectrode. The photoelectrodes were illuminated using an 808 nm laser diode, such that the J_{sc} value matched the value of J_{sc} that was obtained at low Me₂Fc⁺ concentrations under 100 mW cm⁻² of simulated AM 1.5 G illumination. Fig. 1B shows the performance of the arrays in the presence of either 0.4 or 25 mM Me₂FcBF₄, with the latter cell exhibiting a fill factor of $ff_{808} = 0.47 \pm 0.04$ and an efficiency $\eta_{808} =$ $2.7 \pm 0.7\%$. After correcting both the 0.4 and 25 mM Me₂FcBF₄ J–E data for concentration overpotential and uncompensated resistance losses, the corrected fill factor and photoelectrode efficiency values were $ff_{\rm corr} = 0.57 \pm 0.05$ and $\eta_{\rm corr} = 2.0 \pm 0.5\%$, respectively (calculations in ESI \dagger). The corrected J-E data are indicative of the inherent performance of the Si microwire arrays, without experimental artifacts arising from measurement in an unoptimized electrochemical cell configuration.

The photoelectrode efficiency of Si microwire arrays that were grown using the 6N-purity Au VLS catalyst represents a significant improvement relative to initial measurements of the photoanodic performance of n-Si microwire arrays in contact with the Me₂Fc^{+/0}-CH₃OH system. However, the V_{oc} of the Si microwire arrays grown with Au was still substantially less than the V_{oc} values produced by either p-type or n⁺p-Si microwire array photocathodes that were grown with Cu and tested in contact with the MV^{2+/+}(aq) redox system. For Si microwires grown with a Au VLS catalyst, bulk Au concentrations up to $1.7 \times 10^{16} \,\mathrm{cm^{-3}}$ have been previously measured, corresponding to the thermodynamic equilibrium concentration of Au in Si at the growth temperature.14 The concentration of Au within the wires thus greatly exceeded the degradation threshold concentration of Au in planar Si solar cells, and could have contributed to the lower $V_{\rm oc}$ values that were measured for wires that were grown with a Au catalyst. Indeed, Si microwire arrays that were grown by a Au-catalyzed VLS process have previously shown $V_{\rm oc}$ values up to 500 mV in photovoltaic device structures, but only after repeated thermal oxidation and etching steps that should getter Au at the surfaces of the Si wires.8

Given the low tolerance for Au in Si solar cells, 6N Cu was subsequently used for VLS-catalyzed Si wire growth. Cu catalyzed, hexagonally packed Si microwire arrays were fabricated without dopants, with the resulting wires being 2.0–2.5 μ m in diameter and 70–80 μ m in height, and providing an average areal packing fraction



Fig. 1 Current density *vs.* potential (J-E) data for Au–catalyzed Si microwire array photoelectrodes in contact with the Me₂Fc^{+/0}–CH₃OH redox system (a) under 100 mW cm⁻² of AM 1.5 G illumination and (b) with varying amounts of varying amounts of Me₂FcBF₄, to demonstrate and correct for the electrochemical concentration overpotential resistance and series resistance losses within the cell.

 $(\eta_{\rm f})$ of 9.4% (Fig. 2). Similar to the Au-catalyzed wires, the Cu catalyzed wires were slightly p-type, with resistivities of 1000 ± 600 Ω -cm, as measured by four-point resistance measurements, corresponding to $N_{\rm a}$ of $\sim 1 \times 10^{13}$ cm⁻³.

J–*E* measurements of the Cu-catalyzed Si microwire arrays in contact with the Me₂Fc^{+/0}–CH₃OH electrolyte under 100 mW cm⁻² of simulated AM 1.5 G illumination (Fig. 3A) showed $V_{oc} = 437 \pm 8$ mV, $J_{sc} = 7.9 \pm 0.5$ mA cm⁻², and $ff = 0.40 \pm 0.02$, with a photoelectrode efficiency of $\eta = 1.4 \pm 0.1\%$. The Cu-catalyzed Si microwire array photoanodes measured herein exhibited a slightly smaller J_{sc} than the Au-catalyzed Si microwire array photoanodes, consistent with the Cu-catalyzed wire arrays possessing a smaller areal packing fraction than the Au-catalyzed wire arrays. The n⁺-Si(111) control substrates with the wires physically removed produced $V_{oc} = 262 \pm 17$ mV, $J_{sc} = 1.3 \pm 0.3$ mA cm⁻², and $ff = 0.21 \pm 0.01$, again demonstrating that the wafer



Fig. 2 (a) Side view scanning electron microscope image of a cleaved array of Si microwires, scale bar = 40 μ m. (b) Top view of a Si microwire hexagonal array, and the noted axis of rotation θ_{ν} , scale bar = 20 μ m.

substrate did not contribute substantially to the photoresponse of Si microwire photoelectrodes. The Cu-catalyzed Si microwire array photoanodes measured under 808 nm illumination in contact with 25 mM Me₂FcBF₄, to reduce concentration overpotential losses, exhibited a fill factor of $ff_{808} = 0.60 \pm 0.02$ and an efficiency of $\eta_{808} = 3.4 \pm 0.2\%$ (Fig. 3B). After correcting for concentration overpotential and uncompensated resistance losses, the corrected fill factor and efficiency were $ff_{corr} = 0.61 \pm 0.04$ and $\eta_{corr} = 2.1 \pm 0.1\%$, respectively, for Si microwire photoanodes measured under AM 1.5 G illumination. Thus, the undoped Cu-catalyzed Si wire array photoanodes in contact with Me₂Fc^{+/0}–CH₃OH not only exhibited improved performance relative to the initial n-Si wire array measurements, but also yielded efficiencies that were very similar to those observed for optimally doped p-Si wire array photocathodes in contact with MV^{2+/+}(aq).^{5,18}

To further investigate the carrier-collection efficiency of the Cucatalyzed Si microwire arrays, the external quantum yield, Γ_{ext} , of Si microwire photoanodes in contact with Me₂Fc^{+/0}–CH₃OH was recorded as a function of the incident angle of illumination (θ_y). The photoelectrodes were rotated about the θ_y axis of the wire array, as shown in Fig. 2B, from $\theta_y = 0$ –60°. At normal incidence of illumination ($\theta_y = 0^\circ$), the arrays showed the lowest external quantum yield, with $\Gamma_{ext} \sim 0.28$ under visible illumination (Fig. 4A), corresponding to high transmission through an array that had an average $\eta_f = 9.4\%$. The prominent resonant peaks in the external quantum yield can be attributed to whispering-gallery modes in the hexagonal wires, in which light can circularly propagate at the periphery due to multiple total internal reflections.^{22,23} Despite the low packing fraction of wires, the arrays effectively collected 28% of the incident photons,



Fig. 3 J-E data for Cu–catalyzed Si microwire array photoelectrodes in contact with the Me₂Fc⁺⁰–CH₃OH redox system (a) under 100 mW cm⁻² of AM 1.5 G illumination and (b) with varying amounts of Me₂FcBF₄.





Fig. 4 Angle-resolved (a) external quantum yield and (b) absorption, and (c) calculated internal quantum yield at normal incidence, for Si microwire arrays.

demonstrating optical concentration within the array. The spectral response of the wire arrays strongly depended on the angle of incident illumination, with a peak of $\Gamma_{ext} = 0.86$ at $\theta_y > 52^\circ$. In contrast, photoanodes for which the Si wires had been physically removed from the substrate exhibited negligible photocurrent, with $\Gamma_{ext} < .03$. The current of such electrodes also did not vary with angle, further indicating that the degenerately doped Si substrate did not substantially contribute to the response of the Si wire array photoanodes.

The significant increase in Γ_{ext} as θ_y increased indicates that the J_{sc} of the wire arrays under AM 1.5 G illumination was primarily limited by light absorption in the array, and not by carrier collection. Convolution of the spectral response at $\theta_y > 52^\circ$ with the AM 1.5 G spectra resulted in a predicted J_{sc} value of 26 mA cm⁻². As has been demonstrated recently, this calculated J_{sc} can be attained by incorporating light-trapping elements such as a back-reflector, anti-reflective coatings, and/or scattering particles into the device structure.^{9,24}

To calculate the internal quantum yield, Γ_{int} , of the Cu-catalyzed Si microwire array photoanodes, optical absorption measurements as a function of wavelength and angle were performed on the same wire arrays that were used for collection of the external quantum yield data. An integrating sphere was used to perform optical transmission and reflection measurements on peeled films of wires that had been embedded in polydimethylsiloxane (PDMS), as described previously.²⁴ As expected, the measured absorption was lowest at normal incidence, corresponding to large transmission through the sparsely packed, highly oriented array (Fig. 4B). The absorption rapidly increased with increasing angle of incident illumination, reaching a plateau value of 0.86. The value of Γ_{int} for the Si microwire array photoelectrodes was subsequently calculated by dividing the Γ_{ext} at normal incidence by the absorption of the array at normal incidence. Fig. 4C compares the values of Γ_{ext} and absorption at normal incidence, resulting in a peak Γ_{int} value of 0.73 \pm 0.05.

From the values of Γ_{int} and the optical absorption coefficient $\alpha(\lambda)$, the minority-carrier diffusion length L_{n} , for a planar device can be approximated by use of eqn (1).^{11,25}

$$\Gamma_{\rm int} = \frac{1}{1 + \frac{1}{\alpha(\lambda)L_{\rm n}}} \tag{1}$$

Analysis of Γ_{int} for the Cu-catalyzed Si microwire array photoanodes in the near-infrared region (800 nm $\leq \lambda \leq$ 950 nm), in which the optical penetration depth α^{-1} did not exceed the length of the wires, yielded an effective hole diffusion length, $L_{\rm n,eff}$, of 75–85 µm. This value is not a true diffusion length, given that the assumptions of eqn (1) do not apply in a radial geometry, but rather a comparison to the diffusion length that would be needed to produce similar near-IR carrier collection efficiencies in a planar Si device structure (Fig. 5). The calculated $L_{\rm n eff}$ value is significantly larger than the 30 μ m minoritycarrier diffusion length that has been measured previously for moderately doped, Cu-catalyzed, VLS-grown Si microwires.9,17 This observation further demonstrates the advantages of using a radial junction, which produces a longer effective diffusion length than the actual minority-carrier diffusion length, by extending the device response further into the near-IR region of the spectrum as compared to planar Si-based devices with comparable bulk electronic properties.

With V_{oc} values of 0.44 V, a photoelectrode energy-conversion efficiency of $\eta = 1.4\%$, and high carrier-collection efficiencies, the undoped Cu-catalyzed Si microwire array photoanodes have demonstrated performance that is comparable to that of optimally doped p-type Si microwire array photocathodes. However, the measured Γ_{int} deviated from the value of unity previously measured for p-type Si microwire photocathodes, and from the unity value that is predicted by radial junction theory for a wire having a radius smaller than the minority-carrier diffusion length.⁴ The Γ_{int} of the wire array photoanodes at normal incidence was also lower than the



1.0

Fig. 5 A comparison of the measured Γ_{int} of Cu-catalyzed Si microwire array photoanodes and several calculated Γ_{int} responses for planar Si photoelectrodes, with L_n , ranging from 5–200 µm.

 Γ_{ext} value that was measured at high angles, implying a change in Γ_{int} with a change in the angle of incident illumination. The Si microwires, however, are not expected to have a minority-carrier diffusion length smaller than their 1.25 µm radius, given that 30 µm minority carrier diffusion lengths have been reported previously for Cu-catalyzed VLS-grown Si microwires. These apparent discrepancies can be explained by the presence of Cu silicide located at the tops of the wires, as visualized by a high-contrast region in scanning electron microscopy images (Figure S2). This region persisted after chemical etching to remove the metallic Cu VLS catalyst, and varied in thickness from 200-900 nm, depending on the cooling conditions after VLS growth. This silicide most likely acted as a region of low lifetime for carriers that were generated at the tops of the wires, where a significant fraction of the light is absorbed,²⁶ thus decreasing the internal quantum vield, particularly at normal incidence. The quantum yield would therefore not only deviate from unity due to this electronically defective region, but would also vary as the excitation profile changed with incident angle of illumination, consistent with the observations reported herein.

Due to their very low dopant concentration, the Si wires are expected to be operating under high-level injection conditions, in which the concentration of photogenerated carriers exceeds the equilibrium concentration of majority carriers in the wire.27 The initial device physics model only encompassed wires under low-level injection that were not fully depleted, making the model potentially not applicable to undoped microwires operating under field-free conditions. Modeling of radial junction nanowire devices under highlevel injection conditions predicts that these devices should have poor carrier collection efficiencies, due to the full depletion within the nanowire resulting in large majority-carrier recombination losses. 16,28 Work is currently underway to understand in more detail the properties of both nano- and microscale radial geometry devices with low dopant densities that are operating under high-level injection conditions.

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References

- 1 B. Tian, T. J. Kempa and C. M. Lieber, Chem. Soc. Rev., 2009, 38, 16 - 24.
- 2 A. I. Hochbaum and P. D. Yang, Chem. Rev., 2010, 110, 527-546.
- 3 M. G. Walter, E. L. Warren, J. R. McKone, S. W. Boettcher, Q. X. Mi, E. A. Santori and N. S. Lewis, *Chem. Rev.*, 2010, 110, 6446-6473.
- 4 B. M. Kayes, H. A. Atwater and N. S. Lewis, J. Appl. Phys., 2005, 97, 114302
- 5 S. W. Boettcher, J. M. Spurgeon, M. C. Putnam, E. L. Warren, D. B. Turner-Evans, M. D. Kelzenberg, J. R. Maiolo, H. A. Atwater and N. S. Lewis, Science, 2010, 327, 185-187.
- 6 J. M. Spurgeon, S. W. Boettcher, M. D. Kelzenberg, B. S. Brunschwig, H. A. Atwater and N. S. Lewis, Adv. Mater., 2010, 22, 3277-3281.
- 7 S. W. Boettcher, E. L. Warren, M. C. Putnam, E. A. Santori, D. Turner-Evans, M. D. Kelzenberg, M. G. Walter, J. R. McKone, B. S. Brunschwig, H. A. Atwater and N. S. Lewis, J. Am. Chem. Soc., 2011, 133, 1216-1219.
- 8 C. E. Kendrick, H. P. Yoon, Y. A. Yuwen, G. D. Barber, H. T. Shen, T. E. Mallouk, E. C. Dickey, T. S. Mayer and J. M. Redwing, Appl. Phys. Lett., 2010, 97, 143108.
- 9 M. C. Putnam, S. W. Boettcher, M. D. Kelzenberg, D. B. Turner-Evans, J. M. Spurgeon, E. L. Warren, R. M. Briggs, N. S. Lewis and H. A. Atwater, Energy Environ. Sci., 2010, 3, 1037-1041.
- 10 J. R. Maiolo, B. M. Kayes, M. A. Filler, M. C. Putnam, M. D. Kelzenberg, H. A. Atwater and N. S. Lewis, J. Am. Chem. Soc., 2007, 129, 12346-12347.
- 11 M. L. Rosenbluth and N. S. Lewis, J. Am. Chem. Soc., 1986, 108, 4689-4695.
- 12 M. D. Kelzenberg, D. B. Turner-Evans, B. M. Kayes, M. A. Filler, M. C. Putnam, N. S. Lewis and H. A. Atwater, Nano Lett., 2008, 8, 710-714.
- 13 B. M. Kayes, M. A. Filler, M. C. Putnam, M. D. Kelzenberg, N. S. Lewis and H. A. Atwater, Appl. Phys. Lett., 2007, 91, 103110.
- 14 M. C. Putnam, M. A. Filler, B. M. Kayes, M. D. Kelzenberg, Y. B. Guan, N. S. Lewis, J. M. Eiler and H. A. Atwater, Nano Lett., 2008, 8, 3109-3113.
- 15 M. C. Putnam, D. B. Turner-Evans, M. D. Kelzenberg, S. W. Boettcher, N. S. Lewis and H. A. Atwater, Appl. Phys. Lett., 2009, 95, 163116.
- 16 M. D. Kelzenberg, Ph.D., California Institute of Technology, 2010.
- 17 M. D. Kelzenberg, D. B. Turner-Evans, M. C. Putnam, S. W. Boettcher, R. M. Briggs, J. Y. Baek, N. S. Lewis and H. A. Atwater, Energy Environ. Sci., 2011, 4, 866-871.
- 18 E. L. Warren, S. W. Boettcher, M. G. Walter, H. A. Atwater and N. S. Lewis, J. Phys. Chem. C, 2011, 115, 594-598.
- 19 R. H. Hopkins and A. Rohatgi, J. Cryst. Growth, 1986, 75, 67-79.
- 20 J. R. Davis, A. Rohatgi, R. H. Hopkins, P. D. Blais, P. Raichoudhury, J. R. Mccormick and H. C. Mollenkopf, IEEE Trans. Electron Devices, 1980, 27, 677-687.
- 21 F. Gstrein, D. J. Michalak, W. J. Royea and N. S. Lewis, J. Phys. Chem. B, 2002, 106, 2950-2961.
- T. Nobis, E. M. Kaidashev, A. Rahm, M. Lorenz and 22 M. Grundmann, Phys. Rev. Lett., 2004, 93.
- 23 L. Y. Cao, J. S. White, J. S. Park, J. A. Schuller, B. M. Clemens and M. L. Brongersma, Nat. Mater., 2009, 8, 643-647.
- 24 M. D. Kelzenberg, S. W. Boettcher, J. A. Petykiewicz, D. B. Turner-Evans, M. C. Putnam, E. L. Warren, J. M. Spurgeon, R. M. Briggs, N. S. Lewis and H. A. Atwater, Nat. Mater., 2010, 9, 239-244.
- 25 W. W. Gartner, Phys. Rev., 1959, 116, 84-87.
- 26 M. D. Kelzenberg, M. C. Putnam, D. B. Turner-Evans, N. S. Lewis and H. A. Atwater, 2009 34th IEEE Photovoltaic Specialists Conference, Vols. 1-3, 2009, 001948-001953.
- 27 A. Luque and S. Hegedus, ed., Handbook of Photovoltaic Science and Engineering, Wiley, Chichester, England; Hoboken, NJ 2003.
- 28 J. M. Foley, M. J. Price, J. I. Feldblyum and S. Maldonado, Energy Environ. Sci., 2012, 5, 5203-5220.