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Effect of Phytosanitary Irradiation on the Postharvest Quality of Seedless Kishu Mandarins (*Citrus kinokuni mukakukishu*)

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Running head: Postharvest quality of irradiated mandarins

24 Abstract

Transnational trade of 'Seedless kishu' mandarins (Citrus kinokuni mukakukishu) would 25 require them to be subjected to a suitable phytosanitary treatment. Irradiation is used as 26 an effective treatment for many fruit, but the effect on quality of kishu mandarins is 27 unknown. 'Seedless kishu' mandarins were treated with gamma irradiation (150, 400, and 28 1000 Gy) and stored for three weeks at 6 °C and then for one week at 20 °C. Irradiation 29 at 400 and 1000 Gy promoted browning of the calyx end and fungal infection. Irradiation 30 caused immediate reductions in pulp firmness, vitamin E, individual sugars and 31 carotenoids but increased the content of organic acids, except ascorbic acid, and phenolic 32 compounds. The volatile profile of tested fruit was also differentially altered by irradiation. 33 Most of these initial changes were dose dependent. 'Seedless Kishu' mandarins are 34 significantly sensitive to irradiation and are not suitable for treatment at the studied doses. 35

36

Keywords: Phytochemicals; Ionizing energy; Postharvest storage; Citrus; Bioactive
 compounds; Mandarin

39

40 **1. Introduction**

The cultivation and consumption of mandarins has increased steadily in recent years in the U.S.A. (Baldwin & Jones, 2012). However, the domestic production of mandarins is insufficient and seasonal imports from Spain, Chile, Morocco, and Peru are required to satisfy the domestic demand of this fruit. These seasonal imports provide 30% of the mandarins consumed in the U.S.A (Baldwin et al., 2012). Currently, the importation of

new mandarin varieties from China is under consideration by the United States Animal
and Plant Health Inspection Service (APHIS, 2014). The seedless kishu mandarin
(*Citrus kinokuni mukakukishu*) is one of the varieties under consideration for import.
This variety is also available in California. It is a small, sweet, aromatic mandarin with
an easy to peel, thin, tight rind (UCR, 2016), but very little is known about the chemical
properties of this variety of mandarin.

International trade of fruit involves the risk of introducing pests if adequate phytosanitary 52 treatments are not applied before fruit shipment. Irradiation represents an alternative 53 54 method to control insects and possesses advantages over chemical and thermal methods, especially in terms of human safety, fruit quality and environmental impacts 55 (McDonald et al., 2013). The success and advantages of generic phytosanitary 56 irradiation doses for the postharvest control of pests in fruits have clearly been 57 demonstrated (Hallman, 2012). Currently, two generic irradiation doses (150 and 400 58 Gy) are approved by the APHIS for fruits to be exported to the continental U.S.A., and 59 the maximum irradiation dose allowed in foods by the FDA is 1000 Gy (FDA, 2008; 60 Hallman, 2012). These two generic doses allow the control of the most important 61 62 quarantine pests of citrus fruits (Hallman, 2012; Zhang, Deng, Fu and Weng, 2014a). However, the success of irradiation as a phytosanitary treatment depends not only on 63 its capacity to kill or neutralize target insects but also on the tolerance of fruits to 64 65 ionizing energy. Depending upon the fruit, irradiation may result in an increase in ethylene biosynthesis and respiration rate, physiological disorders (rind disorder, loss of 66 glossiness, pitting, and other skin injuries), softening, retardation of color development, 67 68 deterioration of pulp flavor, accumulation of fermentative metabolites, and the alteration

of levels of some bioactive compounds (Miller, McDonald & Chaparro, 2000; Oufedjikh, 69 Mahrouz, Amiot & Lacroix, 2000; Ladaniya, Singh & Wadhavan, 2003; Alonso, Palou, 70 del Rio & Jacas, 2007: Palou, Marcilla, Rojas, Argudo, Alonso & Jacas, 2007). 71 However, most of these responses in mandarins have mainly been observed at doses 72 that differ from the generic doses, and especially at doses that exceed 1000 Gy. The 73 response of mandarins at irradiation doses of 150 and 400 Gy is virtually not known nor 74 the effect of irradiation on certain chemical attributes of mandarins. Based on literature, 75 it can be hypothesized that a dose of 150 Gy is considerably low for phyto-toxic effects 76 77 on mandarins, and that at 1000 Gy, some negative impacts may be manifested. Thus, the objectives of this work were to characterize the physical and chemical properties of 78 'Seedless kishu' mandarins and also to determine the effect of gamma irradiation at 79 generic doses on the physical and chemical attributes of mandarins during simulated 80 sea shipment and subsequent retail distribution. 81

82

83 **2. Materials and Methods**

2.1 Chemicals and solvents. All reagents and solvents were of analytical or HPLC grade and were purchased from Fisher Scientific (Fair Lawn, NJ, U.S.A). The standard compounds for sugars (D-(+)-glucose, D-(-)-fructose and sucrose), phenolic compounds (gallic acid, *p*-coumaric acid, ferulic acid, chlorogenic acid, hesperidin, narirutin, rutin and (-)-epicatechin), organic acids (citric, succinic, L-(+)-tartaric, DL-malic, L-ascorbic, oxalic, and fumaric), *all*-rac-α-tocopherol, volatile compounds, and some carotenoids (all-*trans*β-cryptoxanthin, all-*trans*-lutein, all-*trans*-α and all-*trans*-β-carotene from carrots) were purchased from Sigma-Aldrich (St. Louis, MO, U.S.A.). The other carotenoids were
obtained from CaroteNature GmbH (Lupsingen, Switzerland).

93 2.2 Fruit procurement, treatment, and storage

'Seedless kishu' mandarins (Citrus kinokuni mukakukishu) were harvested in Exeter, CA, 94 U.S.A. The average weight, diameter, and height of the fruits were 37.7 g, 42.1 mm and 95 27.9 mm, respectively. They were hand cleaned and packed in 6.8 kg cartons (29.2 cm 96 wide, 43.2 cm long, and 14 cm high) by California Citrus Specialties. Most commercial 97 packing houses do not apply wax or fungicides to these mandarins and none was applied 98 to these fruit. After cleaning, the fruit was ground transported to Chapman University and 99 kept at 20 °C. The next day, the fruit was taken to Sterigenics, Inc. (Tustin, CA, U.S.A.), 100 101 for treatment. Ten cases of mandarins were located two rows high and five across in front of a ⁶⁰Co source (~37PBg). Dose mapping was conducted by placing 24 alanine pellet 102 dosimeters (FarWest Technology, In., Goleta, CA, U.S.A.) at various locations in the 103 104 cases. The dose rate was 0.637 Gy/s. Ten cases of mandarins were placed exactly in the same configuration as the dummy cases to receive treatment at a target dose of 150, 105 400, and 1000 Gy (4.6-5.5% uncertainty) and Dmax/Dmin ratio of 1.33. Midway through 106 treatment, the boxes were rotated 180° to ensure uniform treatment. After treatment, the 107 mandarins were transported to Chapman University and stored at 6 °C (relative humidity 108 (RH) = 85-90%) for 21 days. Then, the cases were opened and kept at 20 °C (RH = 85-109 90%) for 7 days. After 2, 21 and 28 days, three cartons of each experimental group were 110 removed from storage. Two cartons were used for physical and chemical analyses. One 111 112 carton of each treatment was intermittently removed from storage to evaluate the development of disorders and fungal infections. For chemical analyses, at least 150 fruits 113

were juiced using an Elite Gourmet MaxiMatic Juice Extractor and 20 subsamples of 40 mL each were centrifuged (20000 g/ 20 °C/10 min) to separate the solids and liquid of the juice, according to Stinco et al. (2013). The obtained solids and liquids were distributed into five samples. The solids were evaluated for carotenoid and α -tocopherol content and tristimulus color while the liquid was evaluated for total soluble solids content (TSS), titratable acidity (TA), individual sugars, individual organic acids, individual and total phenolic compounds, and volatile compounds.

121

122 2.3 Peel disorders and fungal infections

The incidence of oleocelosis, pitting, rotting and browning was evaluated in one case of fruit per treatment. The cases used for this evaluation were briefly removed from storage and returned to it after evaluation of fruit by five trained judges. The results were expressed as % of total fruit (number of fruit manifesting the disorder/total number of fruit in the case) showing noticeable symptoms of each disorder.

128

129 2.4 Evaluation of TSS, TA, tristimulus color, and firmness

TSS was measured in the centrifuged juice using a hand-held refractometer (ATAGO Co. Ltd.; Tokyo, Japan). For TA, the centrifuged juice was diluted with water (1:10, v/v), titrated with 0.1 N NaOH to an end point of pH 8.2 and expressed as citric acid (%). The tristimulus color (L*, a* and b*) of mandarin pulp obtained following centrifugation was evaluated using a CM-2500d Minolta Spectrophotometer (Ramsey, NJ, U.S.A). In

135 preparation for firmness measurements, ten fruit were scored longitudinally and carefully peeled and segmented. Peel firmness was measured using a TA.XT2 Texture Analyzer 136 (Texture Technology Corp., Scarsdale, N.Y., U.S.A. and Stable Microsystems, 137 Godalming, Surrey, U.K.) at the equatorial axis of the fruit peels with a cylindrical puncture 138 probe (i.d. 3 mm) to penetrate through a distance of 10 mm at a speed of 2 mm/s. The 139 maximum force (N) was recorded. To measure firmness of the flesh, 150 g of mandarin 140 segments were placed in a Kramer Shear Cell (TA-91) and pressed with the five flat-blade 141 attachment at a speed of 4.0 mm/s. The maximum force (N) and area under the curve 142 143 were recorded.

144

145 2.5 Analysis of sugars

The content of individual sugars was determined according to Ornelas-Paz et al. (2013), 146 with some modifications. Aliquots of juice (100 µL) were diluted with HPLC grade water 147 (2 mL), filtered using a nylon membrane with a pore size of 0.45 µm (Pall Corp., New 148 York, U.S.A.), and automatically injected (20 µL) into a 1100 series HPLC system (Agilent 149 150 Inc., CA, U.S.A.) equipped with a refractive index detector. The sugars (sucrose, glucose 151 and fructose) were separated in a SUGAR SC 1821 (300 x 8.0 mm I.D., 6 µm particle size) ion-exchange column (Showa Denko K.K.; Tokyo, Japan) at 80 °C. The mobile 152 phase was HPLC-grade water at a flow rate of 0.8 mL/min. The sugars were identified 153 154 by comparing their chromatographic behavior with that of reference compounds. Quantitative data were obtained by calibration curves constructed with three independent 155 sets of dilutions of reference compounds (six concentration points for each set). 156

157

158 2.6. Analysis of organic acids

This analysis was based on the methodology described by Ornelas-Paz et al. (2013). One 159 milliliter of centrifuged juice was diluted with 3 mL of 5 mM H₂SO₄. The mixture was filtered 160 and injected into the HPLC system described above but connected to a diode array 161 detector. The separation of the organic acids was performed in an Aminex HPX-87H (300 162 x 7.8 mm I.D., 9 µm particle size) ion-exchange column (Bio-Rad Laboratories., CA, 163 U.S.A.) at 60 °C. The mobile phase was composed of 5mM H₂SO₄ and acetonitrile (90:10, 164 v/v) at a flow rate of 0.4 mL/min. The ascorbic acid was monitored at λ = 260 nm while the 165 other organic acids were monitored at λ =210 nm. The identification and quantification of 166 the acids were performed using standard compounds. 167

168

169 2.7. Analysis of phenolic compounds

170 The analysis of individual and total phenols was performed simultaneously. The juice was filtered with a membrane of 0.45 µm pore size and directly injected (100 µL) into the HPLC 171 described previously. The separation of phenolic compounds was performed in a Kinetex 172 C18 (100 x 4.6 mm I.D., 5 µm particle size) (Phenomenex; Torrance, CA, U.S.A.) at 30 173 °C. The phenolic compounds were monitored at λ = 280, 320, 350 and 520 nm. The mobile 174 phase consisted of 2% acetic acid (A), and acetonitrile (B), according to the following 175 gradient: 100% A at 0 min, 93% A at 12 min, 89% A at 20 min, 86% A at 35 min, 84% A 176 at 36 min, 82% A at 41 min, 79% A at 44, 0% A from min 55 to 60. The flow rate was 1 177 mL/min. The phenolic compounds were identified and guantified by using reference 178

compounds. The UV-Vis spectrum of individual phenols was also used for identificationpurposes.

For total phenols content, 100 µL of filtered juice were mixed with 100 µL of Folin-181 Ciocalteu reagent, 3 mL of deionized water and 100 µL of 20% Na₂CO₃. The mixture was 182 vigorously shaken for 1 min and incubated for 1h in the darkness. The absorbance was 183 evaluated five times at 765 nm using a FLUOstar Omega microplate reader (BMG 184 LABTECH Inc.; Cary, NC, U.S.A.). The absorbance values were corrected with those 185 generated with blank reactions. The quantification was performed using a calibration 186 curve constructed with several sets of dilutions of gallic acid. The results were expressed 187 as mg gallic acid equivalents (GAE) per liter of juice. 188

189

190 2.8 Determination of carotenoid and tocopherol content

191 These compounds were analyzed simultaneously according to Ornelas-Paz, Yahia and Gardea-Bejar (2007). Briefly, 4 g of mandarin pulp was mixed with CaCO₃ (0.2 g) and 20 192 mL of methanol. The mixture was filtered through a Whatman paper No. 3, recovering the 193 methanolic extract. The retained solids were sequentially depigmented with 100 mL of 194 methanol and 75 mL of a mixture of hexane: acetone (1:1, v/v). The extract was placed 195 into a separatory funnel and 30 mL of hexane were added. The mixture was shaken 196 vigorously and 40 mL of 10% Na₂SO₄ were added. After phase separation, the upper 197 phase was recovered and the solvent evaporated at 40 °C under reduced pressure. 198 These residues were analyzed without and with saponification in order to identify free and 199 esterified xanthophylls. For saponification, the residues were dissolved in 30 mL of diethyl 200 ether and 0.6 mL of 40% KOH in methanol were added. The mixture was vigorously 201

202 shaken and kept in the darkness for 16 h. Then, the sample was washed with water and the organic solvent evaporated at reduced pressure. Crude and saponified residues were 203 dissolved in methanol (4 mL), filtered and injected in the HPLC system (20 µL) described 204 above. The separation was performed in a YMC C30 column (150 x 4.6 mm I.D., 3 µm 205 particle size) (YMC America Inc., Allentown, PA, U.S.A.) at 15 °C. The UV-Vis spectra of 206 carotenoids was recorded from 300 to 750 nm in steps of 1 nm. The α-tocopherol was 207 monitored with a fluorescence detector ($\lambda ex = 294 \text{ nm}$, $\lambda em = 326 \text{ nm}$). The mobile phase 208 was composed of water (solvent A), methanol (solvent (B) and methyl tert-butyl ether 209 (MTBE, solvent C) according to the following gradient: 4%A/ 94.5%B/ 1.5%C at 0 min, 210 4%A/ 68%B/ 28%C at 31 min, 4%A/ 30%B/ 66%C at 83 min. The identification of 211 carotenoid esters was performed by comparing the chromatographic behavior of crude 212 213 and saponified extracts. The chromatographic behavior and UV-Vis of peaks of the sample were also compared with those of reference compounds. The quantification was 214 performed using calibration curves constructed with reference compounds. The cis 215 isomers were quantified as all-*E* carotenoids. 216

217 2.9 Volatile analysis

Five samples per treatment (5 mL each) were placed into headspace vials (12 x 32 mm) and 5 mL of saturated sodium chloride was added. 1-pentanol (1.1 mL, final concentration of 490 µg/L) was added to each sample as internal standard. The headspace vials were capped with a Teflon-coated septum. The analysis of volatile components was completed using solid phase microextraction (SPME) with a 75-µm carboxen/polydimethylsiloxane fiber (Supelco, St. Louis, MO, USA). A Gerstel MPS-2 system (Gerstel, Linthicum, MD, USA) was utilized to automate the analytical procedure. 225 Initially, the sample was equilibrated at 40 °C for 15 min followed by deployment of the SPME fiber into the headspace vial and the trapping of headspace volatiles at 40 °C for 226 30 min, all processes with agitation at 4.2 s⁻¹. This was followed by desorption of the 227 volatile compounds from the SPME fiber at 280 °C in the splitless inlet of an Agilent 7980 228 GC (Agilent, Palo Alto, CA) equipped with an FID detector at 280 °C. The flow of hydrogen 229 and air were 0.67 mL/s and 7.5 mL/s, respectively. The separation was performed with 230 an Agilent HP-5MS ultra-inert column (30m x 0.25 mm I.D., 0.25 µm film thickness) using 231 He as carrier gas (0.02 mL/s). Oven temperature was held at 32 °C for 3 min and then 232 ramped up to 200 °C at a rate of 0.1 °C/s. Identification of compounds was performed by 233 using standards, retention indices for *n*-alkanes, and comparing the MS spectra of volatile 234 compounds with those of Wiley/NBS library. The MS spectra were obtained by switching 235 the GC from the FID detector to the mass spectrometer (Agilent 5975) and analyzing 236 identical samples. Semi-quantification of volatile concentrations from the FID data was 237 obtained by utilizing calibration curves generated from standards placed into deodorized 238 mandarin juice. The quantification of ethanol, esters, alcohols aldehydes, terpenes, 239 terpene alcohols and ketones was performed using calibration curves made with ethanol, 240 ethyl acetate, 3-methyl butanol, E-2-hexenal, α -pinene, 4-terpineol, and carvone, 241 respectively. 242

243

244 2.10 Statistics

The data were analyzed using a completely randomized design. The statistical significance of the differences between treatments was determined using ANOVA

followed by the Tukey–Kramer post hoc test; 0.05 was the significance limit. Data analysis
was performed using JMP statistical software (SAS Institute Inc., Cary, NC, U.S.A.).

249

250 **3. Results and discussion**

3.1 Damage and decay

252 Fungal infection and browning of the calyx end were the main causes of damage and decay in irradiated fruit (Table 1). Control fruit did not develop fungal infections or 253 browning during the experiment. The incidence of fungal infections was minimal (0.6-254 255 1.3%) and independent of irradiation dose during cold storage. Similar rates of spoilage were observed in mandarins of several genotypes irradiated up to 300 Gy during 256 refrigerated storage (Miller et al., 2000). In the fourth week of storage at 20 °C, fruit 257 treated at 150 Gy maintained low incidence of rotting (1.3%). However, 400 and 1000 Gy 258 treated samples were heavily infected and incidence of fungal infection in these fruit in 259 their cases was so high, that it was difficult to evaluate the fruit, thus no tests could be 260 performed at the last test day. Similarly, Zhang et al. (2014a) demonstrated that the 261 postharvest decay of Shatang mandarins increased at irradiation dose at and above 400 262 263 Gy. Ladaniya et al. (2003) also demonstrated that mandarins irradiated with high doses (1000-1500 Gy) were more susceptible to fungal infections than those treated with low 264 irradiation doses (250-500 Gy). Rojas-Argudo et al. (2012) showed that high levels of 265 phytoalexins provide resistance to fungal rot in mandarins. However, they observed that 266 physical damage in irradiated mandarins, especially when stored at ambient 267 temperatures, can lead to a decrease in phytoalexins and the resultant development of 268

fungal infection. This is in line with our observation of fungal rot developing in irradiatedmandarins but only when the fruit was stored at room temperature.

271 Browning of the calyx (Fig.1), was observed exclusively in irradiated fruit, with 150 Gy fruit showing the lower incidence of this disorder (76.6 %) after the third week of storage, 272 as compared with fruit treated at the higher doses (91.9-99.3%). The severity of browning 273 seemed to increase with irradiation dose, especially for fruit located on the top layer of 274 the cases but browning severity was not objectively evaluated in this study and further 275 studies are needed in this regard. Irradiation-induced peel injury has been observed in 276 oranges (Macfarlane and Roberts, 1968, Guerrero et al., 1967) and grapefruit and 277 mandarins (Ladaniya, 2008) and attributed to higher respiratory rates, and increased 278 279 activity of enzymes such as peroxidase and phenyl alananine lyase leading to an increase in phenolic compounds. Pitting, a disorder observed in irradiated citrus fruit (McDonald 280 et al., 2013) was not manifested in the irradiated mandarins in this study. 281

Irradiation at 400 and 1000 Gy decreased pulp firmness (Supplementary Table 1). Similar results have been reported for mandarins of several genotypes (Miller et al., 2000; Ladaniya et al., 2003). This effect might be a consequence of irradiation-mediated modification of cell wall components. Bustos and Mendieta (1988) demonstrated that irradiation doses below 1000 Gy were able to modify protopectin and pectin, two polysaccharides involved in fruit firmness, of Valencia oranges. On the other hand, there was no significant changes in peel firmness due to irradiation or storage (data not shown).

Irradiation did not cause an immediate effect on tristimulus color of pulp; however, yellow
and red components (a* and b* values) were reduced in the 400 and 1000 Gy irradiated

291 samples after 21 days of cold storage (Supplementary Table 1). This decrease in a* and b* values of the pulp coincided with darkening of the peel, but did not correlate with 292 carotenoid values. The color values for control and 150 Gy fruit were similar during 293 storage. Similarly, Mitchell, McLauchlan, Isaacs, Williams, and Nottingham (1992) 294 observed moderate effects of irradiation (75 and 300 Gy) on tristimulus color of 'Ellendale' 295 and 'Imperial' mandarins, especially of a* and b* values, after storage. The tristimulus 296 color of clementines was minimally affected by irradiation (300 Gy) during cold storage 297 (Mahrouz et al., 2002). 298

299

300 3.2 TSS and sugars

The contents of TSS and individual sugars are shown in Table 2. Our values for TSS are within the range typically reported (8.3-15.5%) for mandarins (Mahrouz et al., 2002; Mahmoud, Mohamed, Botros & Sabri, 2011). Sucrose was the most abundant of the individual sugars, followed by fructose and glucose, within the concentration ranges previously reported for this fruit and particularly similar to those reported for 'Garbí', 'Fortune' and 'Kara' mandarins (Matsumoto & Ikoma, 2012; Sdiri, Bermejo, Aleza, Navarro & Salvador, 2012).

The effect of irradiation on TSS has extensively been studied, finding that this variable is slightly or not affected by wide range of irradiation doses (75-2400 Gy) in several mandarin genotypes (Mitchell et al., 1992; Mahmoud et al., 2011). In our study, we did not observe a consistent effect of irradiation on TSS or glucose and fructose. Sucrose, however, was reduced by irradiation at 400 and 1000 Gy. After three weeks, the 1000 Gy

fruit showed higher glucose and fructose concentrations suggesting that these changes 313 were a consequence of a sugar interconversion process whereby the higher dose 314 affected the activity or the biosynthesis of enzymes such as invertases, sucrose 315 synthases, and sucrose phosphate synthases (Yativ, Harary & Wolf, 2010). The novo 316 biosynthesis of these enzymes could also be involved in these changes since some 317 studies in other nonclimateric fruits such as Chinese bayberry fruit have demonstrated 318 that irradiation promoted the expression of genes coding these enzymes (Shi, Cao, Shao, 319 Chen, Yang & Zheng, 2016). An increase in sugar content immediately after irradiation 320 has been observed in mangoes (Naresh, Varakumar, Variyar, Sharma & Reddy, 2015) 321 and Chinese bayberry fruit (Shi et al., 2016). In our study, the effect of storage on sugar 322 content was not apparent. 323

324

325 3.3 Organic acids

The TA and concentration of organic acids in tested fruit are shown in Table 2. Our values 326 327 of TA are within the range typically reported (0.2-1.9%) for irradiated and non-irradiated mandarins from several cultivars (Miller et al., 2000; Palou et al., 2007). The organic acids 328 evaluated in this study have also been reported previously for mandarins and other citrus 329 fruits, although the content of oxalic, fumaric, and tartaric acids are often not quantified in 330 mandarins (Matsumoto et al., 2012; Sdiri et al., 2012). As expected, citric acid was the 331 most abundant among tested acids. Our values for this acid were similar to those reported 332 (4.9-16.9 g/L) for several mandarin genotypes (Matsumoto et al., 2012; Sdiri et al., 2012). 333 Citrus fruits are regarded as an important source of ascorbic acid. The initial content of 334

this acid in tested mandarins was in the range typically reported (21-600 mg/L) previously
for this fruit (Mitchell et al., 1992; Mahrouz et al., 2002).

Irradiation did not affect TA (Table 2). Miller et al. (2000) observed that the TA values 337 increased the irradiation dose in Murcott mandarins; however, in other mandarin 338 339 genotypes, irradiation either did not alter or decreased TA. Interestingly, the content of individual organic acids increased immediately after irradiation application (Table 2) as 340 compared to the control. The increase in organic acids suggests alterations of the normal 341 function of the tricarboxylic acid cycle. Similarly, Surendranathan and Nair (1980) 342 observed the accumulation of organic acids in irradiated preclimateric bananas as 343 compared with non-irradiated fruit. They demonstrated that gamma irradiation shifted the 344 glycolytic pathway to the pentose phosphate pathway, causing a reduction in the 345 production of energy and an increased usage of proteins to enhance the gluconeogenic 346 347 flux. The increase of organic acids and reducing sugars by irradiation and the probable involvement of amino acids on such increases of sugars and acids was also hypothesized 348 in carrots (Ismail, Afifi & Fahmy, 1977). Few studies are available in this regard, with 349 350 limitations in the range of irradiation doses, storage conditions, and fruit type. Thus, our data suggest that irradiation modified the activity of several enzymes (i.e. 351 phosphoenolpyruvate carboxykinase, isocitratelyase, fructose diphosphatase, Asp- α -KG, 352 Ala- α -KG, and those of glyoxylate cycle) that initially caused an increase in organic acid 353 content at the beginning of the experiment. After storage for three weeks however, the 354 impact of irradiation was diminished. In regards to ascorbic acid, there was an initial 355 increase in this acid but after three weeks of storage, ascorbic acid showed a clear dose 356 dependent decrease (Table 2). Similar dose dependent reductions in ascorbic acid have 357

been observed in Nagpur mandarins (Ladaniya et al., 2003) and Imperial mandarins (Mitchell et al., 1992). The extent of decrease in mandarins has been shown to depend on genotype (Mitchell et al., 1992). The irradiation-mediated loss of vitamin C in vegetable foods has been attributed to the direct oxidation of ascorbic through the action of free radicals generated by the water radiolysis as well as by the involvement of ascorbic acid in the protection of other compounds against the oxidative damage (Wong & Kitts, 2001).

364

365 3.4 Phenolic compounds

The phenolic content in tested fruit is shown in Table 3. Our values of total phenols (TPC) 366 are within the range reported (263.1-557.3 mg/L) for eleven mandarin cultivars cultivated 367 in Spain (Simón-Grao et al., 2014). Similar values of TPC were also reported for Chinese 368 369 mandarins (Zhang et al., 2014b). Three phenolic acids and four flavonoids were identified in the juice of tested mandarins (Table 3). Ferulic acid was the most abundant phenolic 370 acid in tested juice, as observed in fruit from other genotypes (Shen, Sun, Qiao, Chen, 371 372 Liu & Ye, 2013). Hesperidin and narirutin were the most abundant flavonoids in tested juice, as reported for many other mandarin genotypes (Zhang et al., 2014b; Shen et al., 373 2013; Oufedjikh, Mahrouz, Lacroix, Amiot & Taccini, 1998). The levels of hesperidin in 374 tested fruit were similar to those reported (229-287 mg/L) for irradiated and non-irradiated 375 mandarins (Rojas-Argudo et al., 2012). Interestingly, other flavonoids characteristic of 376 377 mandarins, like naringin, naringenin and neohesperedin, were not detected. Recently, Zhang et al. (2014b) demonstrated that the flavonoid composition in mandarins strongly 378 depended on genotype, with some genotypes showing quite similar qualitative 379

composition of phenolic compounds (phenolic acids and flavonoids) to that observed inthe present study.

Irradiation immediately increased the content of total and individual phenols, except 382 hesperidin (Table 3). Similar increases of phenolic compounds, including that of 383 384 hesperidin, after application of low (510 and 875 Gy) and high irradiation (37900 Gy) doses has been observed in Clemenules and mandarin pomaces (Kim, Lee, Lee, Nam & 385 Lee, 2008; Rojas-Argudo et al., 2012). In contrast, Oufedjikh et al. (1998) observed small 386 decreases in hesperidin and other phenolic compounds immediately after irradiation 387 application (300 Gy) in Clementines. The causes for this immediate increase in phenols 388 389 by irradiation in mandarins and other fruits has been attributed to the stimulation of phenylalanine ammonia-lyase (PAL) activity as a response of the fruit to the stress 390 caused by ionizing energy, as occurs with other stressing postharvest treatments (Rojas-391 392 Argudo et al., 2012; Shen et al., 2013). Undoubtedly, the irradiation-mediated generation of free radicals from water splitting could also reduce the levels of phenolic compounds; 393 however, our data suggest that this negative effect was low or masked by the positive 394 effect of PAL stimulation by irradiation. 395

The TPC and tested individual phenols significantly increased in control fruit during storage (Table 3) although the effect of storage was not clear for irradiated fruit. After cold storage, the total phenolics was highest in the 1000 Gy mandarins but individual phenol content was mostly lower in irradiated than in control fruit, demonstrating that irradiation exert a negative effect on the content of some phenolic during long storage of mandarins. Oufedjikh et al. (1998) also observed that the phenolic content was consistently lower in irradiated than in control mandarins during cold storage for 49 days.

404 3.5 Carotenoids and α-tocopherol

405 The crude extract contained 20 different carotenoid species according to their UV-Vis spectra, including Z and all-E isomers of free and esterified of carotenoids. Excepting 406 data of β -cryptoxanthin, all-E and Z isomers of free xanthophylls were observed in small 407 amounts. The content of the different isomers of each carotenoid were grouped in their 408 free and esterified forms and shown in Table 4. β-cryptoxanthin and violaxanthin, in free 409 and esterified form, were the most abundant xanthophylls in tested fruit. Similar 410 concentrations and relative abundances for these xanthophylls have been reported in 411 Ponkan and Satsuma mandarins (Lin & Chen, 1995; Matsumoto, Ikoma, Kato, Nakajima 412 413 & Hasegawa, 2009). Luteoxanthin, mutatoxanthin and zeinoxanthin were not observed in tested fruit. β -carotene was more abundant than α -carotene; both of which were 414 observed at very small amounts (Table 6), agreeing with previous studies on mandarin 415 416 carotenoids (Lin et al., 1995; Matsumoto et al., 2009). The carotene content was significantly lower than that of total xanthophylls, as observed in fruit from several 417 mandarin genotypes (Matsumoto et al., 2009). 418

The effect of irradiation on carotenoid composition of mandarins has not been previously documented. In our study, irradiation did not promote all-*E* to *Z* isomerization, as evidenced by the absence of increases in the content of *Z* isomers as irradiation dose increased. As expected, irradiation immediately reduced the content of tested carotenoids in a dose dependent trend (Table 4). This detrimental effect of irradiation on carotenoid content might be attributed to the irradiation-mediated generation of hydroxyl radicals

425 from water splitting and the subsequent production of other highly reactive free radicals, like peroxyl and alkoxyl radicals, which show strong reactivity with carotenoids and 426 vitamin E and concomitant depletion of these antioxidants (Bramley et al., 2000; van den 427 Berg et al., 2000). Such radicals are characteristic of membrane lipid oxidation (Bramley 428 et al., 2000), allowing the hypothesis that irradiation caused lipid oxidation in tested fruit. 429 This inference is supported by the immediate and sustained reduction of α -tocopherol 430 levels in tested fruit as the irradiation dose increased (Table 4) since tocopherols are the 431 most potent antioxidants of lipids in the cellular membranes (Bramley et al., 2000). The 432 initial reduction of the carotenoid content by irradiation was statistically similar for free 433 and esterified versions of the same xanthophyll. Similarly, Pérez-Gálvez and Mínguez-434 Mosquera (2002) demonstrated that antioxidant capacity and degradation susceptibility 435 of some carotenoids by free radicals were not dependent on esterification. 436

437 In general, cold storage reduced the levels of carotenoids and α -tocopherol (Table 4), with this effect being more evident for control than for irradiated fruit. However, the levels 438 of tested compounds increased in irradiated and control fruit upon removal from cold 439 440 storage, suggesting that carotenoids are being generated. The effect of temperature storage on carotenoid biosynthesis has previously been studied. Matsumoto et al. (2009) 441 demonstrated that the content of many carotenoids was higher in fruit stored at 20 °C 442 than 5 °C, explaining the increased levels of some carotenoids observed in the present 443 study by the end of the experiment. They observed that cold storage (5 °C) reduced the 444 carotenoid content, as occurred in our study. 445

Fifty volatile compounds were identified in control and irradiated mandarin samples 448 consisting of aldehydes, alcohols, esters, sesquiterpenes, monoterpenes and ketones 449 (data not shown). These components have been commonly found in other varieties of 450 mandarins (Tietel, Plotto, Fallik, Lewinsohn, Porat, 2010; Ummarat, Arpaia, Obenland, 451 2015). Thirty-nine of these volatiles were significantly changed due to irradiation 452 treatment at one or more of the three time points (Table 5). At the beginning of the 453 experiment, irradiated fruit had higher volatile concentrations following both 150 and 400 454 Gy treatments but had reduced concentrations in 13 volatiles, relative to the control, after 455 1000 Gy treatment. This loss in concentration was particularly prevalent in the aldehydes 456 (4), ketones (2) and terpene alcohols (4). Alcohols were increased by irradiation treatment 457 the greatest degree and, with the exception of octanol, concentrations were greater than 458 459 those in untreated fruit, even at 1000 Gy. Storage for 3 weeks at 6 °C after irradiation treatment magnified the effect of the 400 Gy and 1000 Gy treatments and resulted in 460 greater increases in most of the volatile compounds relative to that seen in fruit at the 461 beginning of the experiment. In this case, 1000 Gy caused significant increases in nearly 462 every volatile compound. The large impact of cold storage was most dramatically 463 illustrated in the response of ethanol to irradiation at 6 and 20 °C. In contrast, for many 464 of the volatiles the effect of 150 Gy treatment was lessened and only six of the 39 465 compounds remained significantly different from the untreated control following cold 466 storage. Holding the mandarins an additional week at 20 °C following cold storage also 467 had a large impact on the response to irradiation at 150 Gy. In this case, ethanol and 2-468 methyl-3-buten-2-ol and most of the esters were further increased in amount while many 469

470 of the other volatiles lost in concentration relative to that measured following cold storage. Ethanol has often been noted to be a marker of stress in fruit tissues and has been 471 previously reported to accumulate in enhanced amounts in citrus as a result of irradiation 472 (McDonald et al., 2013), indicating an impact of the irradiation treatment on the 473 metabolism of the fruit. In comparison to the prior research, however, ethanol 474 concentrations were considerably lower in this study, even following irradiation. This was 475 likely in a large part due to the fact that the 'Seedless Kishu' were unwaxed and probably 476 had higher internal oxygen levels. Ethyl esters, compounds that can increase in response 477 to high ethanol levels and contribute to off-flavor (Tietel, Fallik, Lewinsohn & Porat, 2011), 478 were also present in much smaller amounts than observed in the prior study. This may 479 indicate that it might be safer to irradiate unwaxed fruit, as was done in this test, in terms 480 of considering the potential effects on flavor. 481

482

483 **4. Conclusions**

484 Seedless Kishu mandarins have physical and chemical attributes similar to other mandarins, although there were some differences such as the lack of phenolic 485 compounds- naringin, naringenin and neohesperedin. The kishu mandarins were highly 486 sensitive to gamma irradiation even at 150 Gy, the lowest generic doses approved by 487 USDA-APHIS for postharvest phytosanitary treatment. Irradiation negatively affected the 488 489 appearance, firmness, and promoted fungal infections during storage. Irradiation also significantly modified the composition of many compounds involved in the sensory, 490 nutrient and health-promoting attributes of mandarins. These effects were in many cases 491

dose-dependent. We conclude that the Kishu mandarin is not a good candidate to betreated with irradiation for phytosanitary purposes.

494

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- 651

652 **Figure Captions**

Fig. 1. Damage evident in irradiated kishu mandarins after three weeks of storage.