

# Exploring the Ring-Closing Metathesis for the Construction of the Solomonamide Macrocylic Core: Identification of Bioactive Precursors

Iván Cheng-Sánchez,<sup>[a]</sup> Paloma Carrillo,<sup>[b]</sup> Antonio Sánchez-Ruiz,<sup>[c]</sup> Beatriz Martínez-Poveda,<sup>[b]</sup> Ana R. Quesada,<sup>[b]</sup> Miguel A. Medina,<sup>[b]</sup> Juan M. López-Romero<sup>[a]</sup> and Francisco Sarabia<sup>[a]\*</sup>

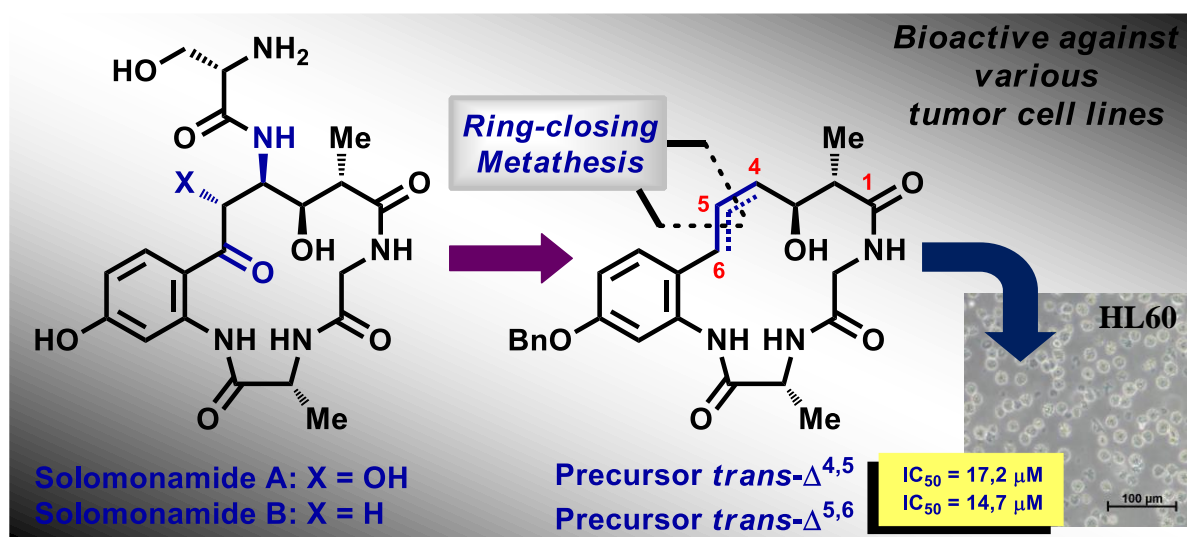
<sup>[a]</sup> Department of Organic Chemistry; Faculty of Sciences. University of Malaga. Campus de Teatinos s/n. 29071. Malaga (SPAIN)

<sup>[b]</sup> Department of Biochemistry and Molecular Biology; Faculty of Sciences. University of Malaga. Campus de Teatinos s/n. 29071. Malaga (SPAIN)

<sup>[c]</sup> Organic Chemistry Section; Faculty of Pharmacy. University of Castilla-La Mancha. Avda Dr. José María Sánchez Ibáñez s/n. 02008. Albacete (SPAIN)

Telephone: 34-952 134258. Fax: 34-952 131941

Email: [frsabria@uma.es](mailto:frsabria@uma.es)



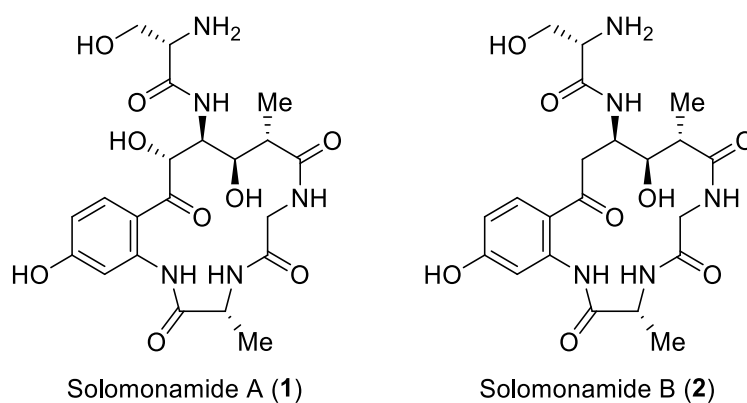
This is the preprint version of our manuscript, corresponding to the article that has been published in final form at JOURNAL OF ORGANIC CHEMISTRY with DOI: 10.1021/acs.joc.7b02988

1  
2  
3  
4  
5  
6  
7 **Abstract:** New synthetic strategies directed towards the novel cyclopeptides solomonamides  
8  
9 have been explored utilizing an olefin metathesis as the key reaction. In the various strategies  
10  
11 investigated, we worked on minimally oxidized systems and the olefin metathesis reaction  
12  
13 demonstrated efficiency and validity for the construction of the macrocyclic core. The  
14  
15 described synthetic strategies towards the solomonamides are well suited for the subsequent  
16  
17 access to the natural products and represent flexible and diversity-oriented routes that allow  
18  
19 for the generation of a variety of analogues via oxidative transformations. In addition,  
20  
21 preliminary biological evaluations of the generated solomonamide precursors revealed anti-  
22  
23 tumor activity against various tumor cell lines.  
24  
25  
26  
27  
28

---

## 1. INTRODUCTION

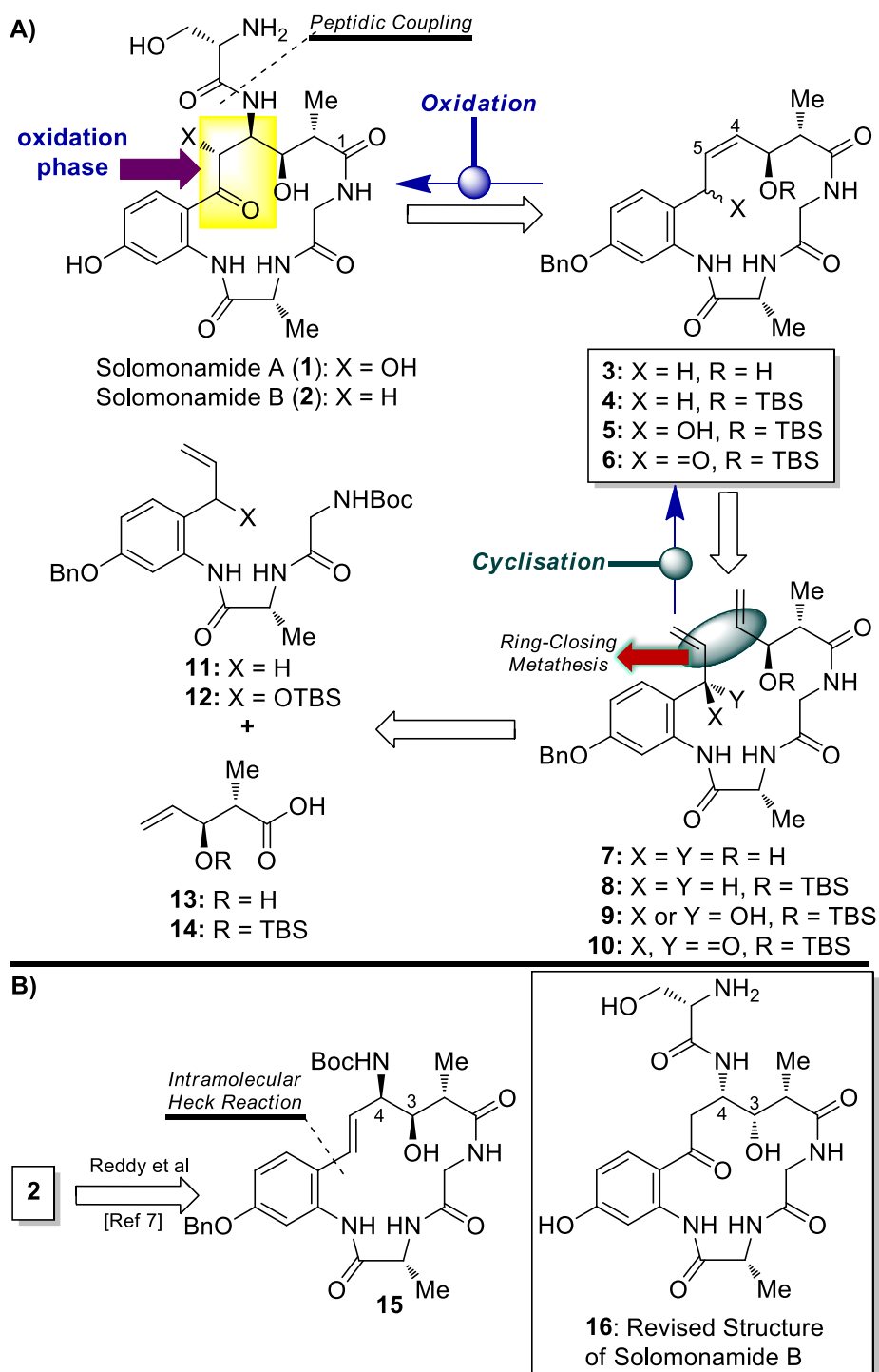
Recently isolated from the marine sponge *Theonella swinhoei*, the solomonamides A (**1**) and B (**2**) (Figure 1) are natural products with interesting and promising biological properties.<sup>1</sup> Structurally characterized by unprecedented cyclopeptidic-type frameworks, these natural products have become potential leads for drug discovery and join a broad and impressive family of other bioactive natural products provided by this marine sponge.<sup>2</sup> An exhaustive spectroscopic analysis of both compounds has resulted in the elucidation of their intricate cyclic structures, revealing the presence of three conventional amino acids (D-Ala, Gly and L-Ser) and an unprecedented 4-amino(2'-amino-4'-hydroxyphenyl)-3,5-dihydroxy-2-methyl-6-oxohexanoic acid (ADMOA) and its corresponding 5-deoxy derivative (AHMOA) for solomonamides A and B, respectively. Assignment of the absolute configurations of ADMOA and AHMOA required additional studies, involving a combination of spectroscopic and theoretical studies (QM *J* based analysis and DFT *J*/<sup>13</sup>C calculations), which resulted in the proposal of the depicted absolute configurations as the most likely. Biologically, solomonamide A (**1**) displayed potent anti-inflammatory activity, causing a significant 60% reduction of inflammation in an animal model of edema at 100 µg/Kg. Unfortunately, the extreme scarcity of the solomonamides has precluded a thorough biological evaluation. In fact, the anti-inflammatory activity of solomonamide B (**2**) was not evaluated due to limited amounts. The unique and unprecedented structures of the solomonamides, together with their intriguing biological properties, have generated intense synthetic activity.<sup>3</sup> For example, the Reddy group<sup>4</sup> has recently reported a total synthesis of a deoxy analogue of solomonamide B<sup>5</sup> together with an array of simple unfunctionalized analogues,<sup>6</sup> culminating with a total synthesis of the natural solomonamide B,<sup>7</sup> which has led to a revision of the initially proposed structure as will be detailed later.



**Figure 1.** Originally Assigned Structures of the Solomonamides

Our ongoing interest in the discovery and development of new potential leads based on cyclopeptidic- and cyclodepsipeptidic-type compounds,<sup>8</sup> prompted us to initiate a research program directed toward the total synthesis of this novel and unexplored class of cyclopeptides. With the aim of establishing a flexible and divergent synthetic strategy capable of providing not only the natural products, but also provide an entry into a plethora of analogues for biological studies, we sought to explore the ring-closing metathesis (RCM) reaction as the key step for construction of the macrocycle.<sup>9</sup> This cyclisation step would be followed by an oxidation phase, which would incorporate the functional groups needed to reach the final oxidation stage found in the natural products. From a strategic perspective, we considered that the construction of the macrocyclic core at the 4,5-bond would be capable of providing rapid access not only to the final products, but also to analogues from late stage intermediates, allowing for the facile entry into numerous scaffolds. Furthermore, it is worthy to note that this synthetic strategy utilizes simple starting materials, avoiding the construction of the complex ADMOA residue, which can be constructed in the later stages of the synthesis through an epoxidation of the olefins **3-6**, followed by an oxirane-ring opening process to introduce the amine group. Accordingly, as detailed in Scheme 1, our delineated strategy in retrosynthetic terms begins with the straightforward amide disconnection of the L-serine

1  
2 residue, followed by the removal of the functional groups, which would be introduced by  
3  
4 means of oxidative manipulations (oxidation phase) of the resulting metathesis products. In  
5  
6 this way, the synthetic strategy would render the corresponding macrocyclic alkenes  
7  
8 represented by the *cis*- $\Delta^{4,5}$  derivatives **3-6**, which would possess or not various functionalities  
9  
10 at the benzylic position. All these macrocyclic compounds, in turn, could be obtained from the  
11  
12 corresponding acyclic precursors **7-10** via a ring-closing metathesis process. Finally, the  
13  
14 preparation of such precursors would be achieved by simple peptidic-like assembly between  
15  
16 the amine derivatives of the corresponding Boc derivatives **11-12** and the olefinic acids **13** or  
17  
18 **14** (Scheme 1, part A). Preliminary results in this synthetic direction have been recently  
19  
20 published<sup>10</sup> and support the viability of this approach. In this manuscript we wish to report a  
21  
22 full account of all the synthetic studies carried out in our laboratories which have been  
23  
24 initiated upon the basis of the previous retrosynthetic scheme. In addition, during the  
25  
26 execution of this synthetic work, Reddy et al published the total synthesis of solomonamide B,  
27  
28 based on an elegant intramolecular Heck reaction, which provided the advanced  
29  
30 solomonamide precursor **15**. More importantly, the total synthesis of solomonamide B led to  
31  
32 the revision of the initially proposed structure for **2**, with the correction of the configurations  
33  
34 at C-3 and at C-4 positions to the (3*S*, 4*S*)-isomer (compound **16**) instead of the proposed (3*R*,  
35  
36 4*R*) for solomonamide B (**2**) (Scheme 1, part B).<sup>7</sup> As the present synthetic studies were  
37  
38 initiated prior to the Reddy publication, we targeted the initially proposed structures for the  
39  
40 solomonamides.  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60



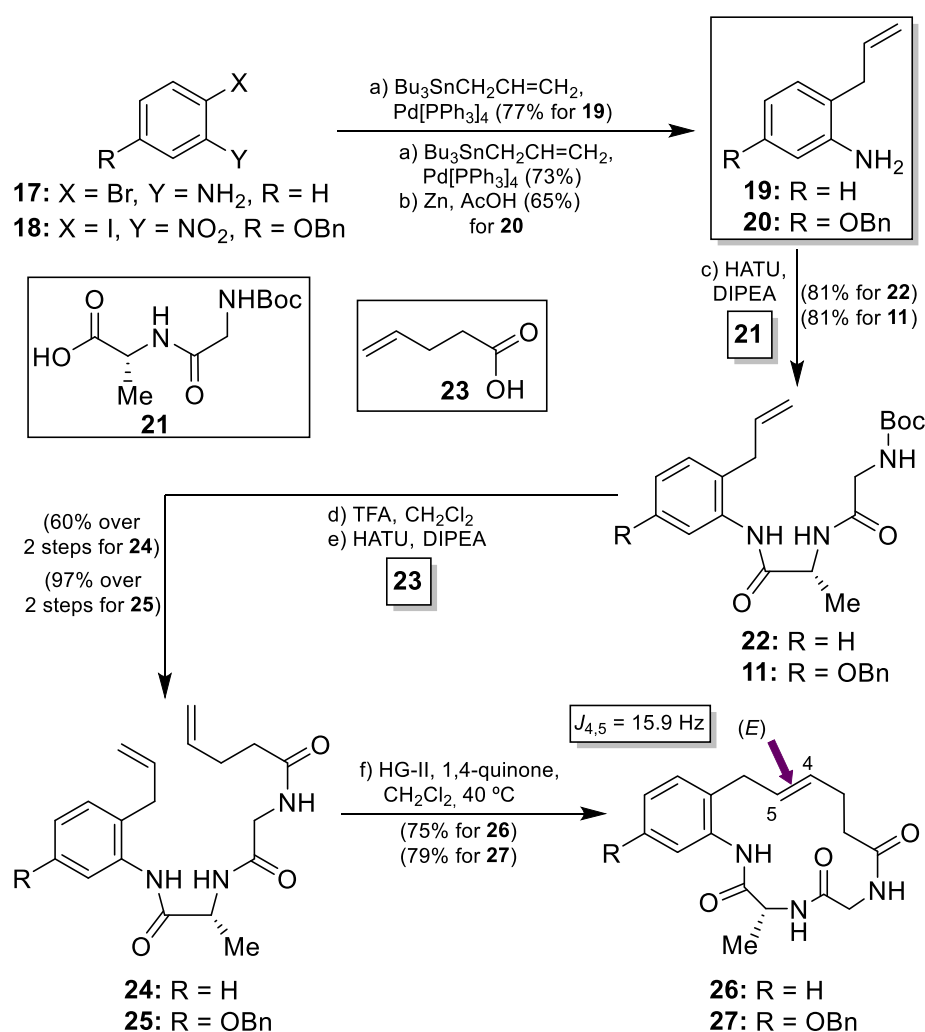
**Scheme 1.** Retrosynthetic Analysis for the Solomonamides (A) and Synthetic Work by Reddy (B)

## 2. RESULTS AND DISCUSSION

**2.1. Ring-Closing Metathesis at the C4-C5 Bond.** Encouraged by the appealing features of this synthetic strategy, as mentioned above, we initiated the synthetic route. Our initial forays toward the solomonamide structures were conducted to demonstrate the viability of the olefin metathesis approach, based on the disconnection of the 4,5-positions bond, using the model compounds **24** and **25** (Scheme 2). To this end, the readily accessible dipeptides **24** and **25** were prepared from 2-bromo aniline **17** and the iodinitrobenzene derivative **18**,<sup>11</sup> respectively, according to the synthetic sequence depicted in Scheme 2, entailing a Stille reaction for compound **19**, and a sequential Stille reaction/reduction, for compound **20**. Couplings of the resulting anilines **19** and **20** with dipeptide **21**<sup>12</sup> furnished the corresponding dipeptides **11** and **22**, which, after Boc deprotection, were coupled with commercial acid **23** to yield the targeted ring-closing metathesis precursors **24** and **25**. Thus, **24** and **25** were treated with 10 mol% of Hoveyda-Grubbs 2<sup>nd</sup> generation (HG-II) catalyst in refluxing dichloromethane in the presence of *p*-benzoquinone<sup>13</sup> to obtain the expected macrocycles **26** and **27** in excellent 75 and 79% yields, respectively, as the sole products of the reaction. The newly formed double bonds ( $\Delta^{4,5}$ ) of **26** and **27** were determined in both cases to be exclusively in the *E*-configuration, as evidenced by a coupling constant *J* of 15.9 Hz, therefore revising our previous assignment, which was incorrect<sup>10</sup> (Scheme 2).

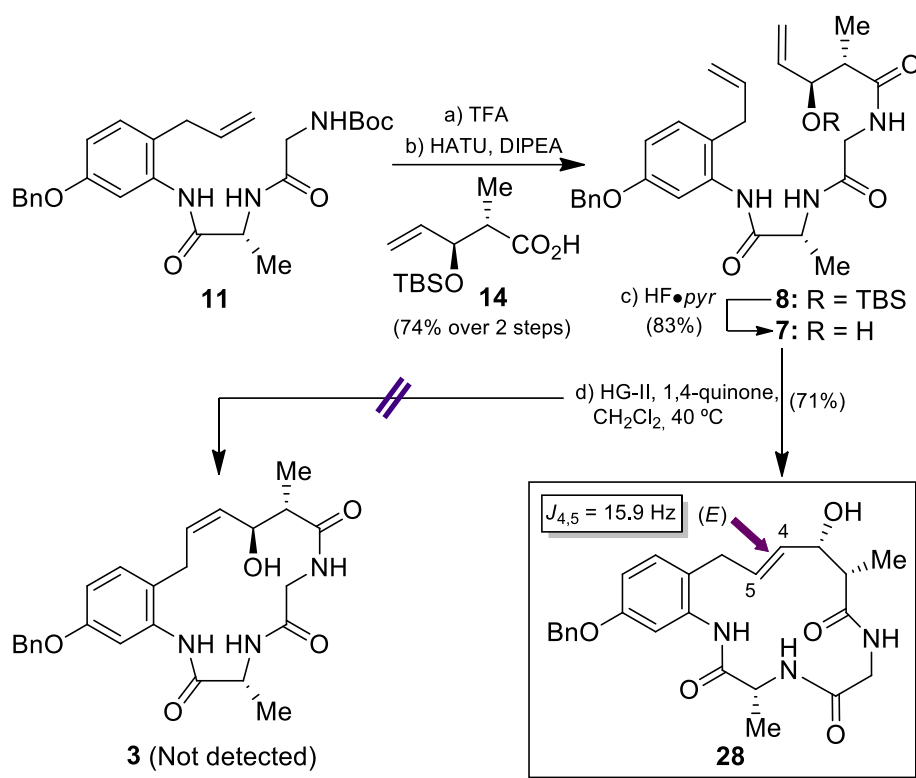
Despite this stereochemical outcome, and with the possibility in mind that the structural pattern of the acyclic precursor could influence the double bond geometry and switch in favour to the desired *Z*-isomer,<sup>14</sup> we continued with the assembly of compounds **11** and **14**<sup>10</sup> in a manner similar to as described before for **24**, providing **8** in a 74% overall yield. Upon exposure of **8** to HG-II catalyst, the expected macrocycle **4** was not obtained, instead recovering starting material, together with a significant degree of decomposition. Attributing

steric factors to this failed cyclisation, the silyl protecting group was removed by treatment of **8** with HF•pyr to give allylic alcohol **7**. Various reports in the literature, describing metathesis reactions involving allylic alcohols, indicate that these structural systems may favour the closing process,<sup>15</sup> although other studies point out that these systems result in detrimental effects for the metathesis reaction.<sup>16</sup> Nonetheless, when allylic alcohol **7** was subjected to the action of HG-II catalyst in dichloromethane at 40 °C in the presence of *p*-benzoquinone, the *E*-olefin **28** was obtained exclusively in a gratifying 71% yield, with no formation of the required *Z*-isomer **3**, as revealed by the <sup>1</sup>H NMR spectra of the crude reaction mixture (Scheme 3).



**Scheme 2.** Towards the Total Synthesis of Solomonamides: RCM of Model Compounds **24** and **25**





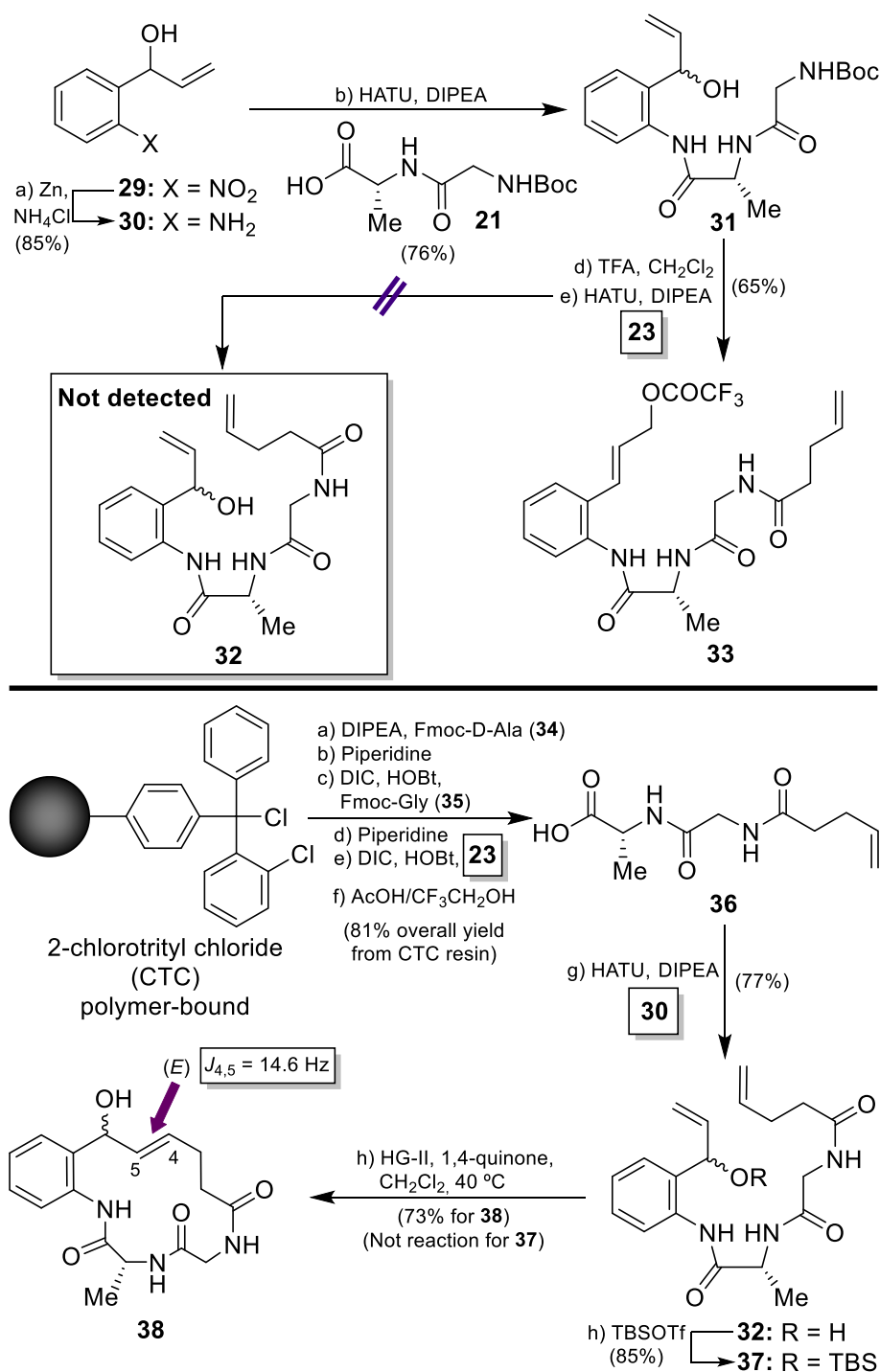
**Scheme 3.** Towards the Total Synthesis of Solomonamides: Synthesis of Macrocycle **28**

In parallel to these preliminary works, we accomplished related synthetic studies extended to more functionalised precursors that would provide for a shortened the path towards the completion of the synthesis of these natural substances. In this direction, we pursued the preparation of the macrocycles **5** and **6** (See scheme 1) as potential advanced precursors. In order to rapidly inspect the validity of this approach, we initially worked with a model system represented by the deoxy aromatic derivatives. Thus, the readily accessible allylic alcohol **29**<sup>17</sup> was transformed into the aniline **30** by treatment with Zn/NH<sub>4</sub>Cl. Following the delineated synthesis for previous compounds, **30** was coupled with dipeptide **21** to yield the coupled product **31** as a 1:1 mixture of diastereoisomers. The assembly of **31** with acid **23** was preceded by Boc deprotection under conventional acidic conditions, followed by amide coupling assisted by HATU. Disappointingly, the expected coupling product **32** was not detected, instead the derivative **33** was obtained as a result of a cationic rearrangement of the

1  
2 labile allylic alcohol **31**, which probably occurred during the Boc deprotection step under the  
3  
4 acidic conditions used (Scheme 4). Different attempts for removal the Boc group under mild  
5  
6 acidic (TMSCl, TMSOTf, Sn(OTf)<sub>2</sub>, TiCl<sub>4</sub>, SnCl<sub>4</sub>), neutral (I<sub>2</sub>, TBAF, H<sub>2</sub>O at reflux) or basic  
7  
8 conditions (Na<sub>2</sub>CO<sub>3</sub>, K<sub>3</sub>PO<sub>4</sub>, NaO<sup>t</sup>Bu)<sup>18</sup> resulted similarly unfruitful with the formation of  
9  
10 rearranged byproducts, recovery of starting material or degradation, respectively. As a  
11  
12 consequence of these results, we considered a direct coupling between aniline **30** and the  
13  
14 peptidic fragment **36**, which was efficiently prepared by utilising solid phase peptide synthesis  
15  
16 (SPPS) as described in Scheme 4. Thus, the coupling of aniline **30** and acid **36** provided the  
17  
18 desired diolefinic precursor **32** in a 77% yield. Finally, the ring-closing metathesis of **32** under  
19  
20 the same conditions as in previous cases, afforded the macrocyclic derivative **38** in a 73%  
21  
22 yield as the *E*-isomer, supported by the *J*<sub>4,5</sub> coupling constant (14.6 Hz), and as a 1:1 mixture  
23  
24 of diastereoisomers (Scheme 4). Interestingly, the protected precursor **37**, prepared by  
25  
26 silylation of **32** by the action of TBSOTf, did not provide the corresponding macrocyclic  
27  
28 derivative when it was subjected to the action of the HG-II catalyst under the same conditions  
29  
30 used for **32**, indicating that steric factors may be responsible for the failed ring closure.  
31  
32  
33  
34  
35  
36  
37  
38  
39

40 The extension of this synthetic scheme to the aromatic system contained in the natural  
41  
42 products was initiated with the aldehyde **40** prepared from commercially available **39**.<sup>19</sup>  
43  
44 Subsequent treatment of aldehyde **40** with magnesium vinylbromide provided allylic alcohol  
45  
46 **41**, which required the reduction of the nitro group to the corresponding amine to give **42**.  
47  
48 This seemingly simple operation however proved more problematic than expected. For  
49  
50 example, treatment of **41** with Zn/NH<sub>4</sub>Cl did not provide the expected aniline **42**, instead a  
51  
52 complex mixture of unidentifiable degradation products were obtained. We surmised that the  
53  
54 benzyloxy group installed in the aromatic ring was playing a crucial role in the reactivity of  
55  
56  
57  
58  
59  
60

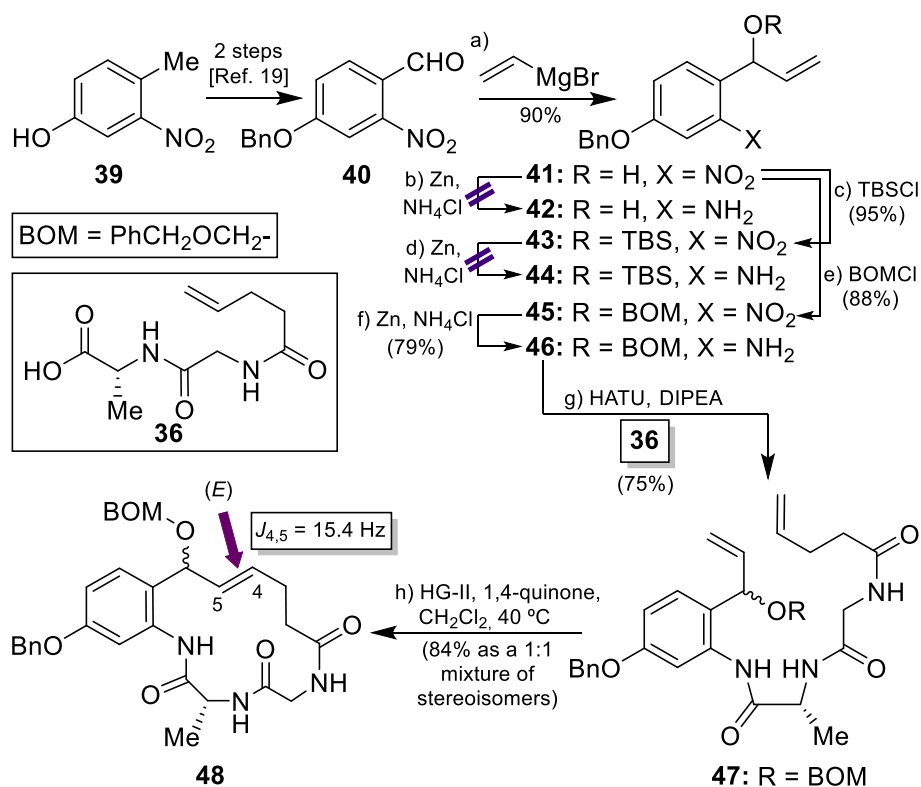
the benzylic alcohol, making it especially sensitive even under the weak acidic conditions of the reduction reaction.



**Scheme 4.** Towards the Total Synthesis of the Solomonamides via Diolefin **32**

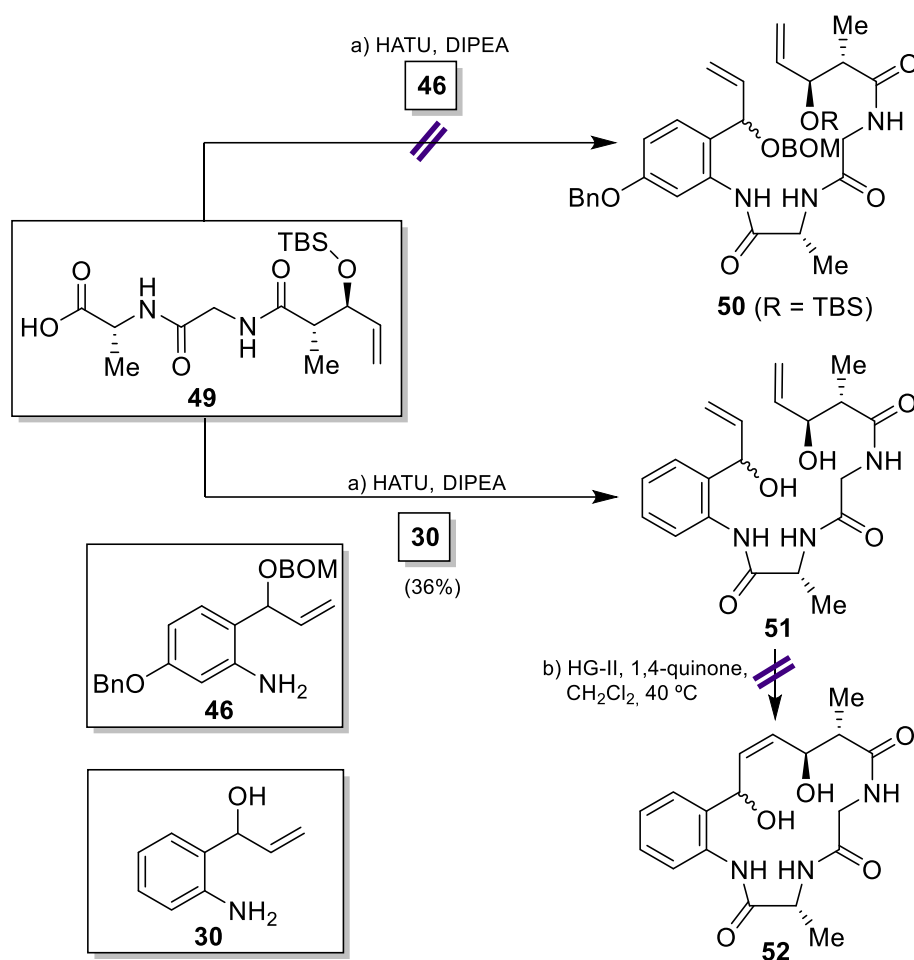
To circumvent this rather disappointing outcome, we decided to protect the hydroxyl group as the silyl ether **43**. However, the Zn-mediated reduction of the nitro group did not provide the

1  
2 expected aniline **44**, instead the result again was the formation of a complex mixture of  
3 degradation products. Other reduction methods, including  $\text{LiAlH}_4$  or Ni-Raney, were  
4 degradation products. Other reduction methods, including  $\text{LiAlH}_4$  or Ni-Raney, were  
5 attempted but they also were unsuccessful. Finally, the more robust BOM ( $\text{PhCH}_2\text{OCH}_2-$ )  
6 protecting group proved to be the solution for this problematic reduction, as the BOM  
7 derivative **45** provided the expected aniline **46** when treated with  $\text{Zn}/\text{NH}_4\text{Cl}$  in a reasonable  
8 and reproducible 79% yield. The linkage of aniline **46** and dipeptide **36** was then performed as  
9 described above for **32** to obtain compound **47**, which was subjected to the ring-closing  
10 metathesis to provide macrocycle **48** in an excellent 84% yield as a 1:1 mixture of  
11 diastereoisomers and the *E*-olefin as the only detectable double bond isomer. It is intriguing  
12 that this ring-closing metathesis process proceeded in such good yield, despite the presence of  
13 a protecting group at the hydroxyl group of the diolefinic precursor (Scheme 5).



Scheme 5. Towards the Total Synthesis of the Solomonamides via Diolefin **47**

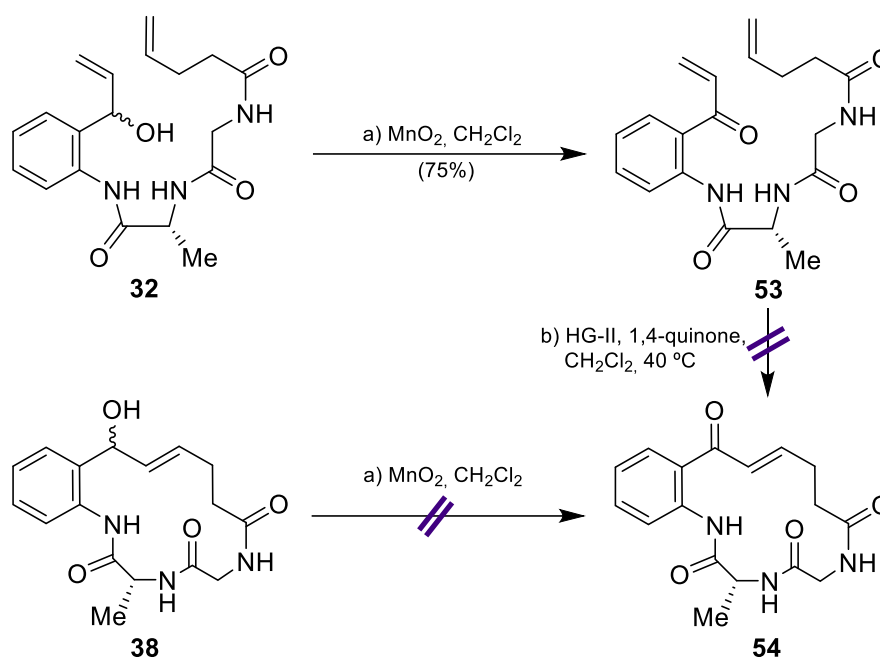
1  
2 The implementation of this synthetic scheme onto the more functionalised system found in the  
3  
4 natural products was then confronted by the assembly of acid **49**, prepared by SPPS (See  
5  
6 Experimental Part) and **46** under conventional conditions employed in this work  
7  
8 (HATU/DIPEA). However, no desired product **50** was obtained despite attempts with an array  
9  
10 of coupling reagents such as BOP, PyBOP, DIC/HOBt or DIC/HOAt. Reasoning that steric  
11  
12 hindrances around the acid **49** could explain this failure, we attempted the synthesis of the  
13  
14 advanced precursor **50** by sequential couplings of aniline **46** with the amino acid derivatives  
15  
16 **34** and **35** and the less sterically encumbered acid **14**. Unfortunately, the last coupling with  
17  
18 acid **14** did not provide the expected product **50**, instead a complex mixture of HATU  
19  
20 derivatives was obtained from the starting acid. Intrigued by these failed reactions, we turned  
21  
22 our attention to the simple aniline **30**, in which the hydroxyl group at C6 position was free, in  
23  
24 order to determine the reasons of the serious hurdles found during this synthetic course. To  
25  
26 this end, aniline **30** and acid **49** were exposed to the action of HATU/DIPEA and the  
27  
28 corresponding coupling product **51** was obtained, albeit in a low 36% yield, with the  
29  
30 unexpected cleavage of the silyl protecting group. Several attempts of this reaction revealed  
31  
32 its lack of reproducibility, confirming the difficulties of such a coupling. Although obtained in  
33  
34 low yield, compound **51** was in hand and we were in position to investigate the ring-closing  
35  
36 metathesis reaction for this more complex system. The reaction, undertaken under similar  
37  
38 conditions as in previous cases, did not give the expected macrocycle **52**, thus demonstrating  
39  
40 the unsuitability of these substrates in the construction of the macrocyclic core of the  
41  
42 solomonamides via RCM. The reason for failure is most likely due to the sensitivity of the  
43  
44 reaction to steric factors arising from the presence of substituents at both allylic and benzylic  
45  
46 positions (scheme 6).  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60



**Scheme 6.** Exploration of the RCM in More Complex Systems

In yet another attempt to access the macrocyclic solomonamide precursors functionalised at the benzylic position, we considered the incorporation of a ketone at this position, since this is the functional group present in the natural products. In addition, the generation of the  $\alpha,\beta$ -unsaturated ketone system, after the macrocyclisation process, would allow for the rapid and facile access to an epoxide via oxidation of the double bond. Notably, we were skeptical about the success of the metathesis reaction in such a system, represented by compound **53**, given that the olefin is unactivated as it is conjugated ( $\alpha,\beta$ -unsaturated phenylketone). Despite this concern, we opted to experimentally verify the chemical reactivity of **53** under the action of the Hoveyda-Grubbs and related catalysts.

To this aim, we prepared ketone **53** by oxidation of the alcohol **32** with  $\text{MnO}_2$  and then, subjected to the catalytic action of the Hoveyda-Grubbs 2<sup>nd</sup> generation, keeping in mind the difficulty of this reaction due to the inactivated nature of the double bond of the  $\alpha,\beta$ -unsaturated ketone system. Indeed, this reaction did not provide any desired product, even when the reaction was forced using drastic conditions (ex. toluene at 60 °C and 100 °C), instead providing recovered starting material in all cases. In another attempt to obtain cyclic ketone **54**, cyclic alcohol **38** was treated with  $\text{MnO}_2$  but the result was similarly unsuccessful, with only starting material recovered and no detection of the coveted ketone **54** (Scheme 7).

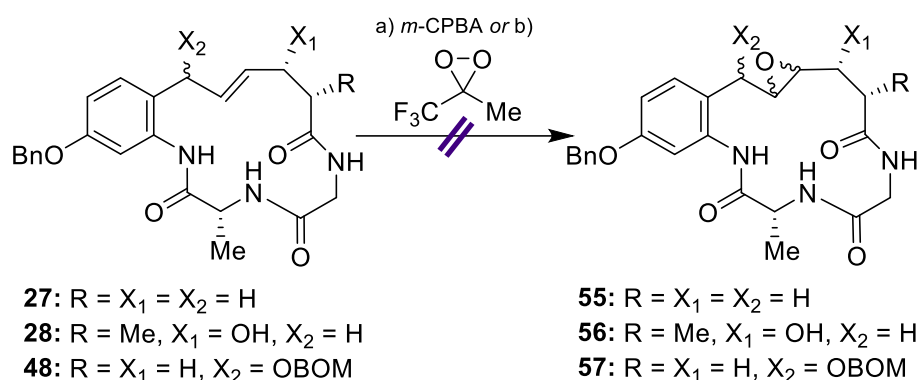


**Scheme 7.** Exploration of the RCM of the Diolefinic ketone **53**

Although a *Z*-geometry for the  $\Delta^{4,5}$  double bond was proposed to provide access to the *syn*-1,2-difunctionalized system present at these positions in the natural products, the fact that we obtained instead the *E*-olefinic macrocycles in all cases represented an additional difficulty in this study to obtain the final targets. Nonetheless, this synthetic strategy was not discarded at this point, as synthetic methods are available to provide the required *syn* isomer from a *trans* double bond.<sup>20</sup> Consequently, we decided to continue with the present synthetic approach and

1  
2 the next step was to evaluate the feasibility of such compounds to provide more oxidized  
3  
4 derivatives through an oxidation phase that would give access to the final products. In this  
5  
6 sense, we studied the oxidation of compounds **27**, **28** and **48** as representative macrocyclic  
7  
8 precursors of the solomonamides. Thus, when **27** or **28** were subjected to the oxidative action  
9  
10 of *m*-CPBA, we found, to our dismay, that these reactions did not yield the expected epoxides  
11  
12 **55** or **56**. Instead, starting materials were recovered in both cases. In light of these  
13  
14 discouraging results, we attempted the epoxidation utilizing the dioxirane derived from  
15  
16 trifluoromethylketone,<sup>21</sup> however the result was similarly frustrating, with no formation of any  
17  
18 desired oxidation products. Other oxidative reactions were screened, such as a  
19  
20 dihydroxylation reaction mediated by OsO<sub>4</sub>, and electrophilic additions, mediated by the  
21  
22 actions of bromine or NBS, but these did not provide favorable results, not detecting  
23  
24 formation of any of the possible oxidation products. The poor solubility observed for these  
25  
26 cyclic compounds in common organic solvents could explain the lack of reactivity found for  
27  
28 them towards the oxidative reagents. However, the more soluble derivative **48** also proved to  
29  
30 be unreactive when it was subjected to the epoxidation reagents (*m*-CPBA and dioxirane  
31  
32 species), resulting in the recovery of starting material and no detection of epoxide **57** (Scheme  
33  
34 8). Theoretical studies are currently in progress in order to justify the lack of reactivity exerted  
35  
36 by these macrocyclic compounds. Thus, initial results from DFT calculations carried out in  
37  
38 solution showed that the HOMO orbital of the molecules is centered around the intracyclic  
39  
40 phenyl ring with a low electronic density around the alkene moiety.<sup>22</sup> This result suggests that  
41  
42 the main reactive point for the oxidant should not be the alkene, as it could be expected, but  
43  
44 the aromatic ring. At this stage, no further work was required to recognize that oxidation of  
45  
46 the previous solomonamide precursors was extremely hard to accomplish. Therefore, a new  
47  
48 approach was considered.  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60





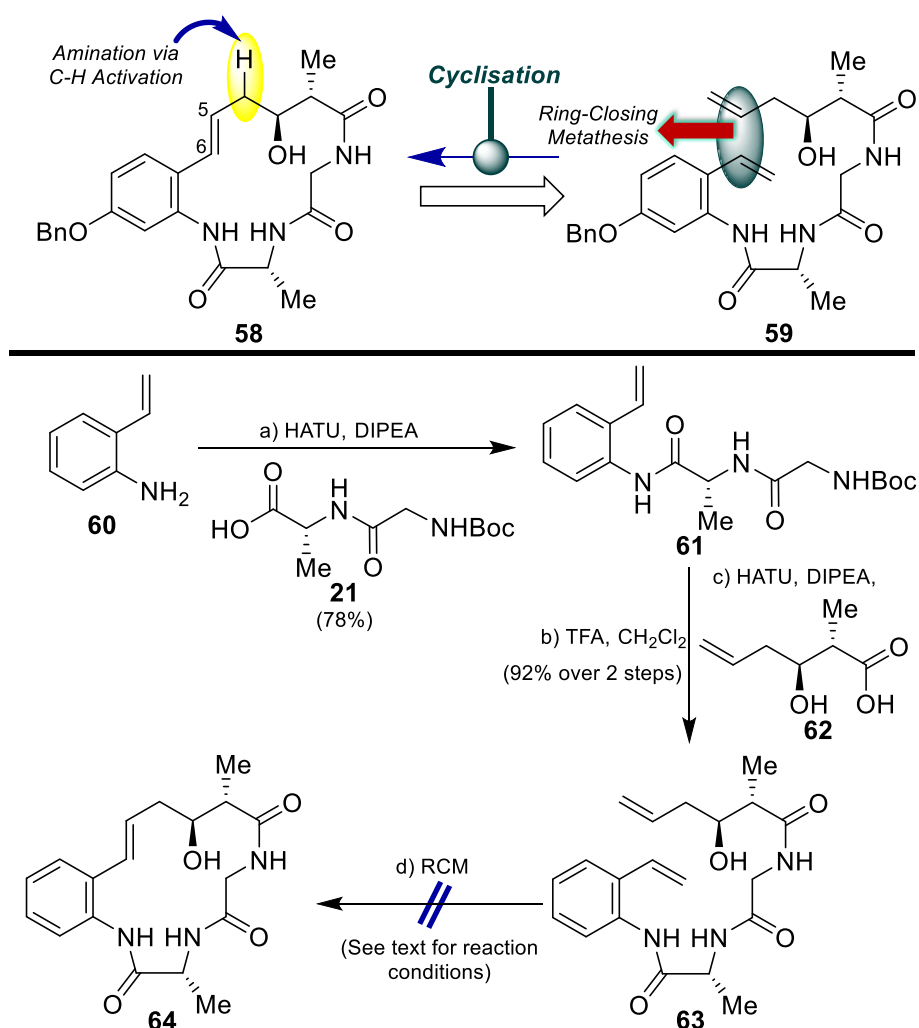
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

**Scheme 8.** Attempts of Oxidation of the RCM Products

**2.2. Ring-Closing Metathesis at the C5-C6 Bond.** Given the results obtained during the synthetic studies towards the solomonamides utilizing the macrocyclic construction at the C4-C5 bond, we turned our attention to the construction of the macrocyclic core of the solomonamides at the C5-C6 bond. In relation to the synthetic strategy explored in the previous section, the removal of the functional groups, which should be accessible during an oxidation phase, would lead to the relatively simple precursor **58**. Whereas the styryl double bond could be transformed into a ketone or an  $\alpha$ -hydroxy ketone via Wacker or dihydroxylation oxidations for solomonamides A and B, respectively, the introduction of the required amine group at the C4 position was initially planned via a C-H activation. The syntheses of the key precursor **58** would be performed by a RCM of the acyclic derivative **59**, whose synthesis would be achieved by simple peptidic-like assemblies between the corresponding amine and olefinic acid. As in the previous synthetic exploration, we preferred to initiate this study with the model compound **63** to test the viability of the new synthetic proposal. The preparation of this RCM precursor was successfully achieved from the simple aniline **60**<sup>23</sup> by sequential couplings with dipeptide **21** and hydroxy acid **62**<sup>10,24</sup> with HATU. With compound **63** in hand, we proceeded with the olefin metathesis reaction by use of the HG-II catalyst in refluxing dichloromethane. However, it was with much disappointment, that

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

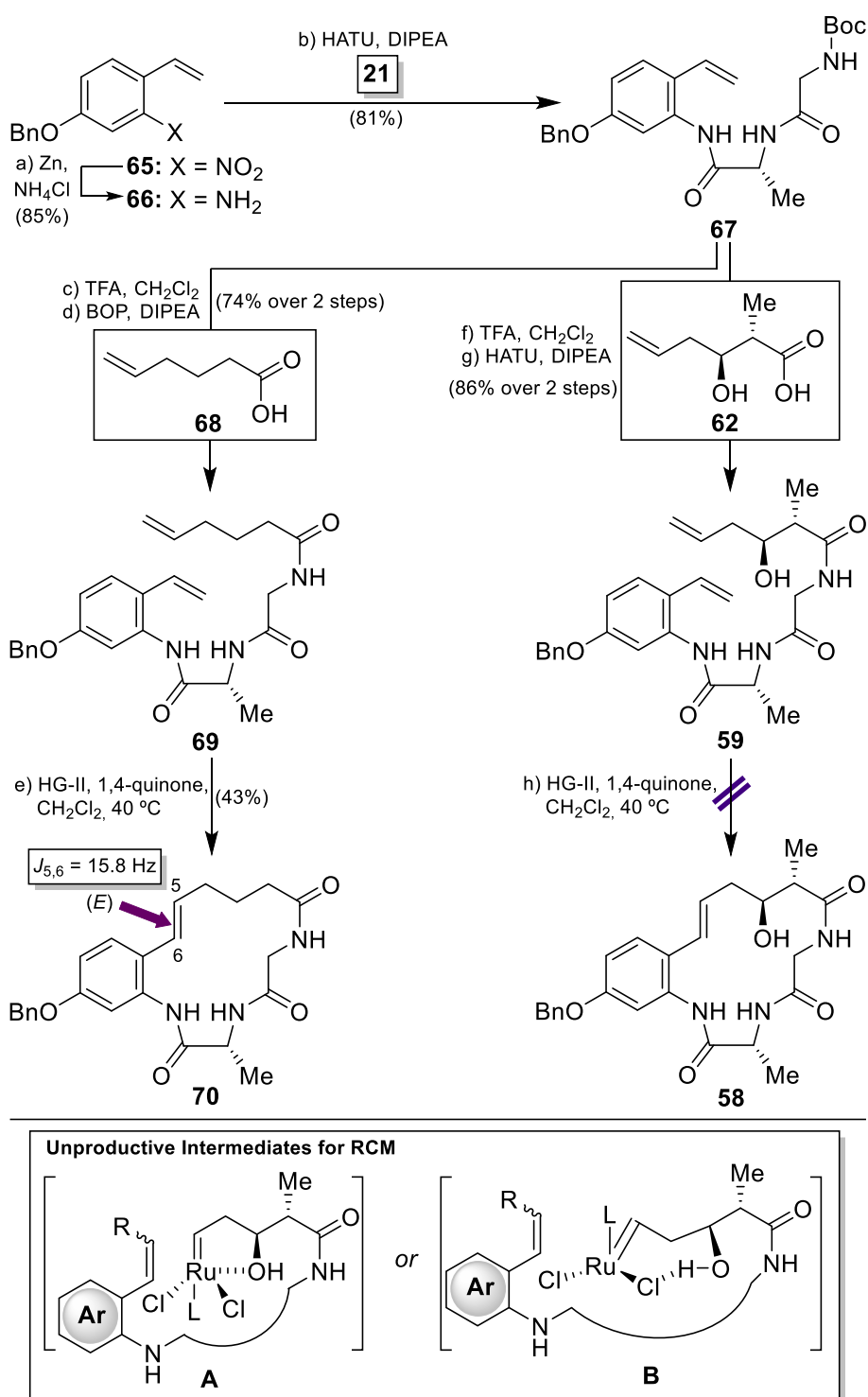
this reaction failed to afford any macrocyclic product, leading instead to decomposition and/or polymerization, together with the recovery of some starting material (~12%). Additional attempts that included more forcing conditions (toluene at 65°C or 100 °C) and other catalysts (Grubbs 1<sup>st</sup> and 2<sup>nd</sup> generations, Hoveyda-Grubbs 1<sup>st</sup> generation) were thwarted, with no detection of the desired macrocyclic product **64** (Scheme 9).



**Scheme 9.** Second Approach to the Solomonamides via RCM: The C5-C6 Disconnection

Despite these discouraging results, we decided to press forward with the synthetic strategy by exploring the ring closing metathesis in the real system represented by the product **59**. We reasoned that in the case of the metathesis precursor **59**, the electronic effect of the benzyloxy

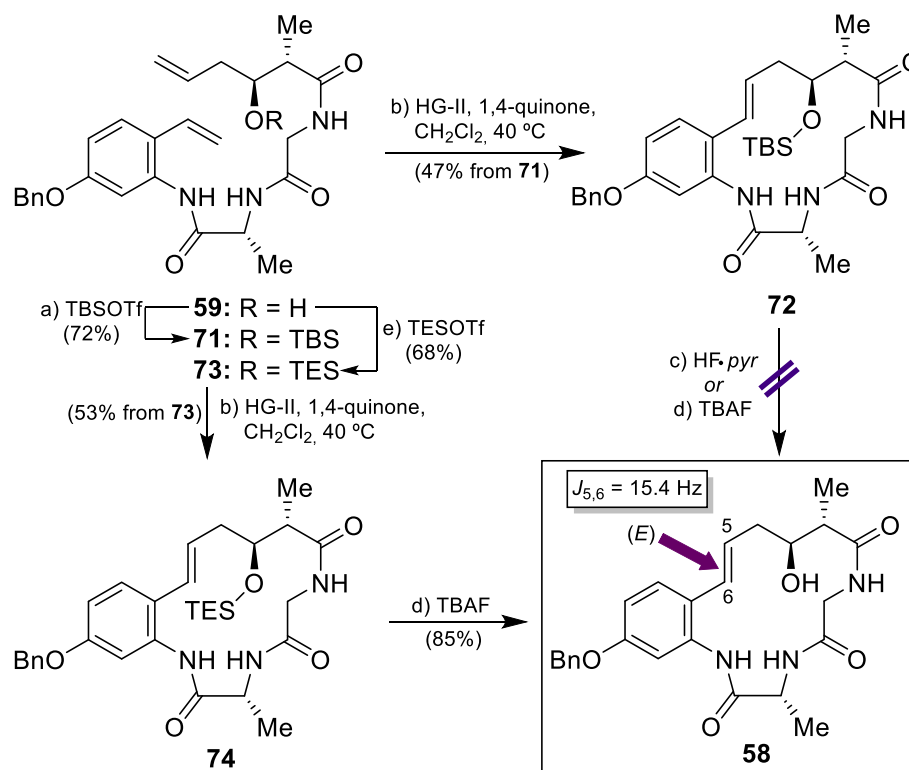
1  
2 group at the *para* position with respect to the styryl double bond could exert a favourable  
3  
4 effect in terms of reactivity for the unreactive olefin. In order to confirm this hypothesis, we  
5  
6 prepared the styryl derivative **67** from compound **65** in a very good overall yield. We then  
7  
8 proceeded to test the construction of the macrocyclic ring of the solomonamide model system  
9  
10  
11 **70**. To this end, compound **67** was coupled with commercial acid **68** to obtain the acyclic  
12  
13 precursor **69** in a 74% yield. Having prepared compound **69**, we were primed to test our  
14  
15 hypothesis regarding the favourable electronic donating effect of the benzyloxy group on the  
16  
17 ring-closing metathesis reaction. Indeed, we were gratified to discover that cyclic compound  
18  
19 **70** was obtained in a 43% yield and, exclusively, as the *E*-isomer, when the acyclic precursor  
20  
21 **69** was subjected to the action of HG-II catalyst in refluxing dichloromethane in the presence  
22  
23 of 1,4-benzoquinone (Scheme 10). In light of this encouraging result, the ring-closing  
24  
25 metathesis was extended to the advanced intermediate **59**, which was efficiently prepared  
26  
27 from the Boc derivative **67** and acid **62** in 86% overall yield, according to the previously  
28  
29 described fragment coupling protocol. To our dismay, treatment of **59** with the HG-II catalyst,  
30  
31 under the same conditions previously employed in earlier cases, did not provide the coveted  
32  
33 solomonamide precursor **58**, providing instead degradation products and recovered starting  
34  
35 material. A possible explanation for the failure of the metathesis reaction is the formation of  
36  
37 either the five-membered ring intermediate **A**,<sup>25</sup> by chelation of the ruthenium carbenoid with  
38  
39 the hydroxyl group, or the seven-membered ring chelate intermediate **B**,<sup>26</sup> in which  
40  
41 sequestration of the ruthenium carbenoid species occurred through a hydrogen bond of the  
42  
43 hydroxyl group with the chlorine atom, which could explain the inactivation of the catalyst for  
44  
45 the RCM (Scheme 10).  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60



**Scheme 10.** RCM of the Acyclic Precursors **59** and **69**

As support for this mechanistic hypothesis,<sup>27</sup> we decided to protect the hydroxyl group to avoid the formation of the proposed unproductive intermediates. Therefore, the TBS

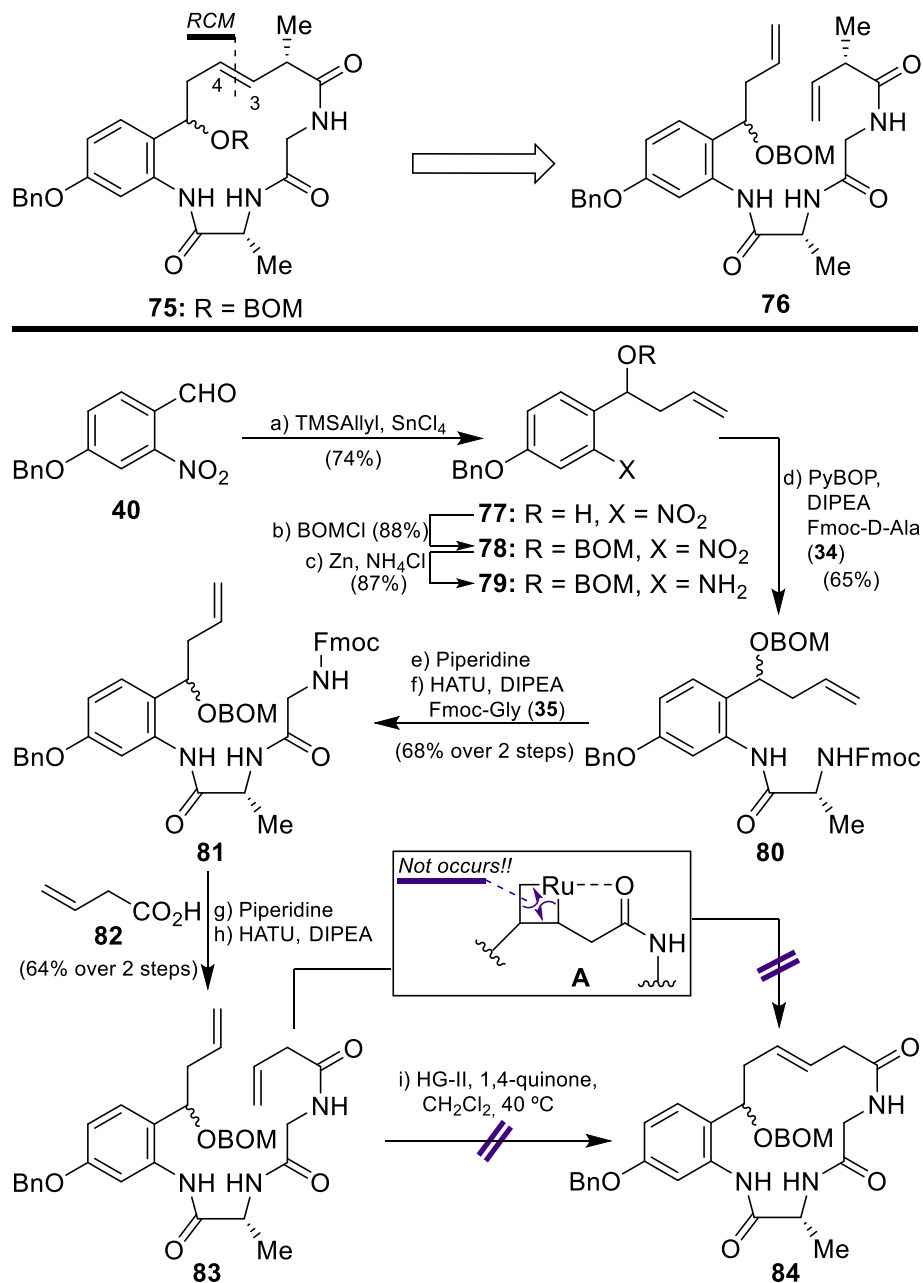
1  
2 derivative **71** was prepared in a good yield by treatment of **59** with TBSOTf in the presence of  
3  
4 2,6-lutidine. The resulting silyl ether **71** was then subjected to the action of the HG-II catalyst  
5  
6 and, to our delight, the macrocyclic product **72** was obtained in a reasonable 47% yield. The  
7  
8 completion of the synthesis of the natural products should require an eventual removal of the  
9  
10 TBS protecting group. With this future objective in mind, we proceeded with the desilylation  
11  
12 reaction by treatment of **72** with HF•pyr. Surprisingly, the expected hydroxyl derivative **58**  
13  
14 was not obtained, instead recovered starting material. Other fluoride-based reagents (TBAF,  
15  
16 TBAF-AcOH) afforded the same frustrating result. Therefore, we decided to replace the TBS  
17  
18 group with a more labile protecting group, choosing TES as a suitable alternative. Then,  
19  
20 compound **73** was prepared by reaction of **59** with TESOTf/2,6-lutidine and its suitability as a  
21  
22 viable substrate for the preparation of **58** was evaluated. Gratifyingly, the RCM reaction of **73**  
23  
24 provided **74** in a similar yield as for **72** (53%) and its desilylation, by treatment with TBAF,  
25  
26 yielded the desired macrocyclic **58** in 85% yield (Scheme 11).



**Scheme 11.** RCM of the Acyclic Precursors **71** and **73**: Synthesis of **58**

1  
2 **2.3. Ring-Closing Metathesis at the C3-C4 Bond.** Prior to the completion of the synthetic  
3  
4 study towards the natural solomonamides, for which meaningful quantities of **58** or precursor  
5  
6 **74** were required, we opted to complete the exploration of the RCM reaction by exploring the  
7  
8 C6-C7 bond as another disconnection point to construct the macrocyclic system of the  
9  
10 solomonamides. This new disconnection strategy would be based on the retrosynthetic  
11  
12 analysis depicted in Scheme 12. Having established the viability of the RCM strategy for the  
13  
14 construction of the macrocyclic core of the solomonamides at two different sites, the C5-C6  
15  
16 and the C4-C5 bonds, we decided to extend the RCM-based strategy to the C3-C4 position,  
17  
18 scanning all the options along the six-carbons chain contained in the ADMOA fragment. In  
19  
20 this new scenario, the removal of the functional groups at the C3 and C4 positions revealed  
21  
22 the compound **75** as a potential precursor for solomonamide B, which could be obtained from  
23  
24 the acyclic derivative **76** via RCM. In the synthetic direction, aldehyde **40** was transformed  
25  
26 into the amine **79** without problems according to the synthetic sequence described in Scheme  
27  
28 12. Our experience gathered during the synthetic campaign led us to choose BOM as the most  
29  
30 suitable protecting group for the benzylic alcohol and to introduce sequentially the two amino  
31  
32 acids (D-Ala and Gly) through the Fmoc derivatives **34** and **35** to obtain compound **81**. Prior  
33  
34 to the preparation of the subtarget RCM precursor **76**, we considered the use of the  
35  
36 commercially available acid **82** to test the possibilities of this new strategy. Having prepared  
37  
38 model compound **83**, treatment with HG-II catalyst under similar reaction conditions as  
39  
40 previously used, did not provide the desired result with the no formation of the macrocyclic  
41  
42 compound **84**. In fact, a rational explanation for this result could be the formation of a stable  
43  
44 5-membered ring chelate between the ruthenium carbene and the carbonyl group of the amide  
45  
46 (intermediate **A**), which renders the metallacyclobutane intermediate unreactive toward the  
47  
48 retro [2 + 2] cycloaddition<sup>28</sup> (Scheme 12). At this stage, we decided to terminate this synthetic  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

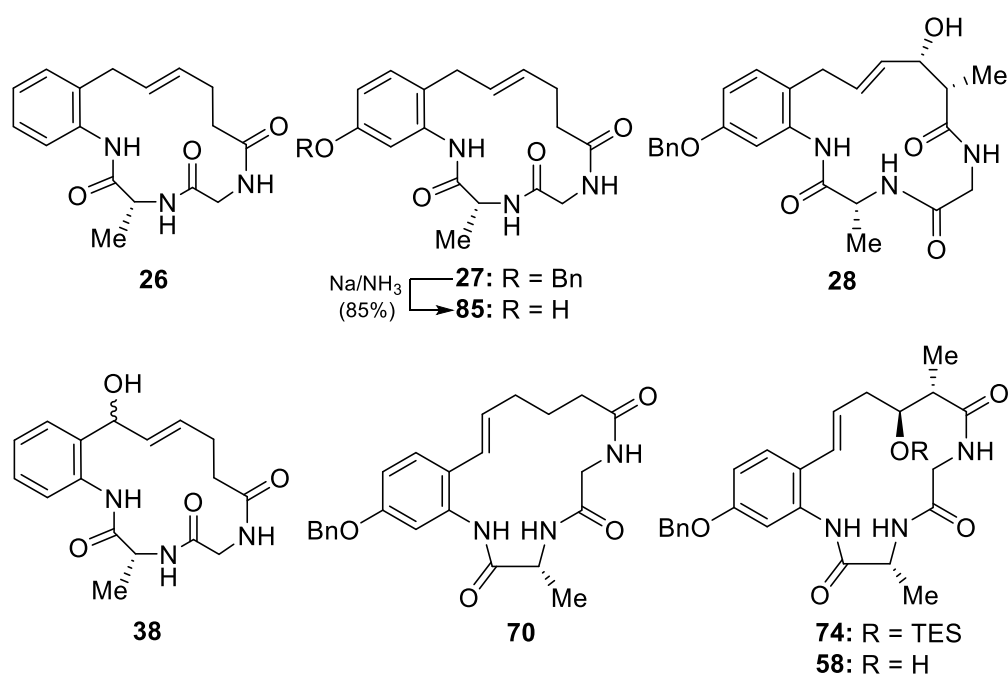
approach, taking into account that this strategy could be reconsidered in the future by use of Lewis acids,<sup>29</sup> which could be investigated as useful additives to promote the ring closure as has been demonstrated in cross-metathesis reactions bearing similar structural motifs.



**Scheme 12.** Third Approach to the Solomonamides via RCM: The C3-C4 Disconnection

## 2.4. Biological Activity of Solomonamide Precursors.

At this stage of the synthetic work, with an efficient route defined to obtain the final natural products, we were intrigued with the biological properties of the synthesized products, recognizing that the macrocyclic peptides represented unprecedented and novel molecular architectures of biological interest. For this reason, we decided to explore and investigate their biological activities. To this aim, we performed preliminary biological evaluations, which consisted of the measurement of the antitumor properties of selected compounds against a panel of various tumor cell lines. The chosen solomonamide derivatives for this study were **26**, **27**, **28**, **38**, **70**, **74** and **58** as representative compounds of the different scaffolds generated during the synthetic work. In addition, compound **85**, obtained from **27** by treatment with Na/NH<sub>3</sub> (See Figure 2), was included for this study to evaluate the effect of the protecting group upon biological activity.



**Figure 2.** Synthetic Solomonamide Precursors Submitted to Biological Evaluations



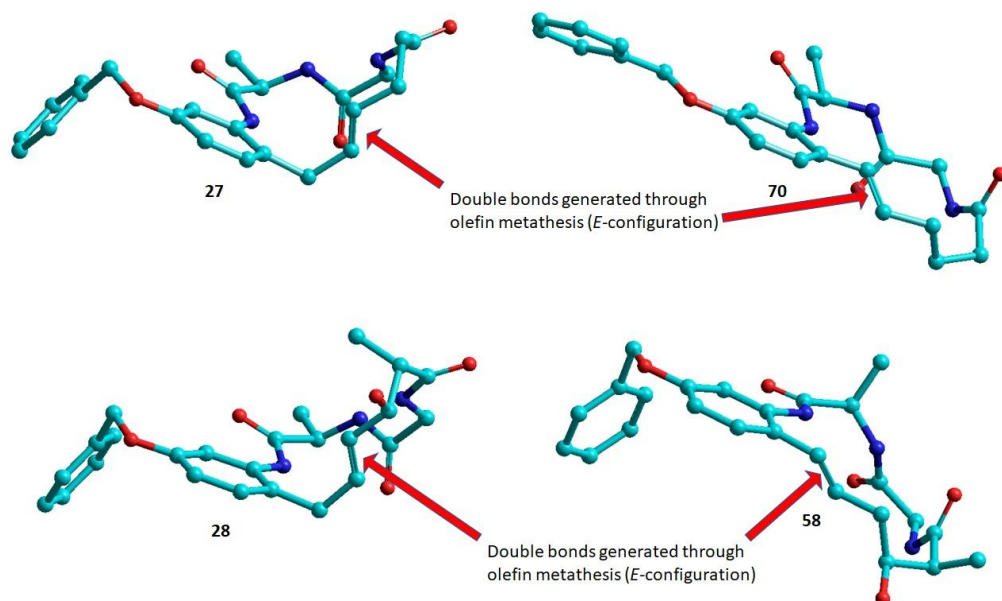
1  
2 As a first attempt to characterize the biological activities of the synthetic solomonamide  
3  
4 analogues, the cytotoxicity profile of the aforementioned compounds was examined using  
5  
6 nine different cancer cell lines (see Table 1), as well as a primary culture of non transformed  
7  
8 bovine aorta endothelial (BAEC) cells.<sup>30</sup> The results of these biological evaluations,  
9  
10 summarized in Table 1, clearly revealed a relevant cytotoxic activity for only one compound,  
11  
12 **58**, which displayed the best values of inhibition of the series in the low  $\mu\text{M}$  range, against all  
13  
14 the tumor cell lines. In addition, **58** was also cytotoxic against the endothelial cells line  
15  
16 (BAEC), which may indicate a putative antiangiogenic effect of this compound.<sup>30</sup> In contrast,  
17  
18 the other compounds were practically devoid of antitumor activity except compounds **28** and  
19  
20 **70**, which displayed cytotoxicity, particularly against the HL60 tumor cell line albeit  
21  
22 moderately. From these results, a highlight is the notable enhancement of activity observed  
23  
24 upon introduction of functional groups in the polyketide fragment contained in these  
25  
26 compounds, as concluded when biological activities of **70** and **58** are compared, or **27** and **28**  
27  
28 for the case of the HL60 cell line. In addition, the isomeric derivative of **58**, compound **28**,  
29  
30 displayed similar antiproliferative activity as **58** against HL60, but, in contrast, was inactive  
31  
32 against the other cell lines, including BAEC, revealing the importance of the position of the  
33  
34 double bond of the molecule upon biological activities. This structural effect is also reflected  
35  
36 when the biological activities of the isomeric compounds **27** and **70** are compared. In an  
37  
38 attempt to make sense of this result, molecular modelling studies were conducted for  
39  
40 compounds **27**, **28**, **70** and **58** to obtain the corresponding minimized structures (Figure 3).<sup>31</sup>  
41  
42 The resulting models revealed a certain structural distortion when **27** and **70**, on one side, and  
43  
44 **28** and **58**, on the other, are compared, respectively. Thus, whereas **70** and **58** are almost flat,  
45  
46 **27** and **28** displayed considerable twisting. These differences in the resulting minimized  
47  
48 molecular structures may explain the differences observed for the biological activities found  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

for these compounds. Additionally, the overall shape of **27** is very similar to **28**, and the same is evident also for **70** and **58**. In this case, despite these conformational similarities, it is clear that the functional groups incorporated in **28** and **58**, in particular the hydroxyl group, may exert a key biological interaction through a hydrogen bonding interaction with an acceptor-type residue located at the active site of the biological targets. These interactions may justify the greater activities displayed by **28** and **58** compared with **27** and **70**, respectively.

**Table 1.** In vitro Antitumor Activities of Solomonamide Precursors against Various Tumor Cell Lines and BAEC (IC<sub>50</sub>, μM)<sup>a</sup>

Compound	HL60 <sup>b</sup>	KU812F <sup>c</sup>	U937 <sup>d</sup>	HT-1080 <sup>e</sup>	MDA-MB-231 <sup>f</sup>	U87MG <sup>g</sup>	HepG2 <sup>h</sup>	HT-29 <sup>i</sup>	U2OS <sup>j</sup>	BAEC <sup>k</sup>
<b>26</b>	> 50	> 50	> 50	> 100	> 100	> 100	> 100	> 100	> 100	> 100
<b>27</b>	> 50	n.d.	n.d.	> 100	> 100	> 100	> 100	> 100	> 100	43,8 ± 1,2
<b>85</b>	> 50	> 50	> 50	> 100	> 100	> 100	> 100	> 100	> 100	> 100
<b>28</b>	17,2 ± 6,6	n.d.	n.d.	> 50	> 50	n.d.	n.d.	n.d.	n.d.	> 100
<b>38</b>	> 50	> 50	> 50	> 100	> 100	> 100	> 100	> 100	> 100	> 100
<b>70</b>	20,5 ± 5,2	> 50	> 50	31,1 ± 2,8	> 100	> 100	> 100	> 100	> 100	69,6 ± 12,5
<b>74</b>	> 50	> 50	> 50	> 100	> 100	> 100	> 100	> 100	> 100	> 100
<b>58</b>	14,7 ± 4,4	14,0 ± 3,05	7,02 ± 2,32	16,3 ± 2,9	16,5 ± 1,01	34,8 ± 5,1	18,9 ± 1,5	13,3 ± 0,6	12,9 ± 3,1	18,1 ± 2,2

[a] In vitro cytotoxicities were determined according to the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide dye reduction assay as detailed in experimental part. The IC<sub>50</sub> values were obtained from semilogarithmic dose-response plots as the concentration of compound yielding a 50% of cell survival. [b] HL60: Human promyelocytic leukemia. [c] KU812F: Chronic Myelogenous leukemia. [d] U937: Histiocytic lymphoma. [e] HT1080: Human fibrosarcoma. [f] MDA-MB-231: Human breast adenocarcinoma. [g] U87MG: Glioblastoma. [h] HepG2: Hepatocellular carcinoma. [i] HT-29: Colorectal adenocarcinoma. [j] U2OS: Osteosarcoma. [k] BAEC: Non transformed bovine aortic endothelial cells.



**Figure 3.** Computer-generated Minimum-Energy Conformations for Compounds **27**, **28**, **58** and **70**

### 3. CONCLUSIONS

In conclusion, an extensive synthetic exploration directed towards the solomonamides was conducted based on a ring-closing metathesis as the key reaction for the rapid and efficient access to their macrocyclic cores. The result of this synthetic study was the establishment of an efficient ring closure process which proceeded in high yields and complete stereoselectivity. During the course of these efforts, unexpected hurdles arose along the way, mainly with: 1) the reactivity of the hydroxyl group at the benzylic position (compounds **31** and **41**); 2) the reactivity of diolefins containing allylic and homoallylic alcohols (compounds **7**, **8**, **32**, **47**, **59**, **63**, **71** and **73**) or containing a  $\beta,\gamma$ -unsaturated carbonyl system (case of compound **83**) toward the ruthenium catalysts, and 3) the reactivity of macrocyclic olefins (compounds **27**, **28** and **48**) toward oxidative reagents. Although many of these synthetic problems were overcome, others remained elusive and represent synthetic challenges for future works. In relation to the

1 ring-closing metathesis reactions, many of the described findings support the  
2 observations and results reported by other authors for this reaction on this class of  
3  
4 observations and results reported by other authors for this reaction on this class of  
5  
6 structural systems. More importantly, not only did we explore the scope and limitations  
7  
8 of the ring-closing metathesis in the synthesis of the macrocyclic core of the  
9  
10 solomonamides, but we also identified several structurally related solomonamide  
11  
12 precursors possessing significant cytotoxicities against various tumor cell lines,  
13  
14 including endothelial cells, in the low  $\mu\text{M}$  range. These preliminary biological  
15  
16 evaluations of relatively simple compounds, devoid of the functional groups present in  
17  
18 the natural counterparts, portend promising antitumor properties for the natural  
19  
20 products and qualify them as new scaffolds of biological and medicinal interest.  
21  
22 Therefore, the described chemistry highlights the benefits of the olefin metathesis  
23  
24 reaction in the field of the total synthesis of natural products, featuring convergency  
25  
26 and flexibility for structural diversity and has allowed the identification of bioactive  
27  
28 compounds with interesting antitumor properties. The completion of the synthesis of  
29  
30 the natural products, the design of new analogues based upon compound **58**, and  
31  
32 further biological studies to elucidate the mechanism of its antitumoral and  
33  
34 antiangiogenic activities are currently in progress and they will be reported in due  
35  
36 course.  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46

## 47 **EXPERIMENTAL SECTION**

48  
49  
50 **General Techniques.** All reactions were carried out under an argon atmosphere with dry,  
51  
52 freshly distilled solvents under anhydrous conditions, unless using aqueous reagents or  
53  
54 otherwise noted. All solvents used in reactions were dried and distilled using standard  
55  
56 procedures. Tetrahydrofuran (THF) was distilled from sodium benzophenone, and methylene  
57  
58  
59  
60

1  
2 chloride (CH<sub>2</sub>Cl<sub>2</sub>) from calcium hydride. Yields refer to chromatographically and  
3  
4 spectroscopically (<sup>1</sup>H NMR) homogeneous materials, unless otherwise stated. All solutions  
5  
6 used in workup procedures were saturated unless otherwise noted. All reagents were  
7  
8 purchased at highest commercial quality and used without further purification unless  
9  
10 otherwise stated. All reactions were monitored by thin-layer chromatography (TLC) using  
11  
12 0.25 mm silica gel plates (60F-254) using UV light (254 nm) as visualizing agent and acidic  
13  
14 ceric ammonium molybdate/ phosphomolybdic acid or potassium permanganate solutions and  
15  
16 heat as developing agents. Flash column chromatography (FCC) was performed using silica  
17  
18 gel (60 Å, particle size 230-400 mesh) under air pressure. All solvents used for  
19  
20 chromatographic purifications were distilled prior to use. <sup>1</sup>H and <sup>13</sup>C NMR spectra were  
21  
22 recorded on a Bruker DPX-400 MHz instrument and calibrated using residual undeuterated  
23  
24 solvent as an internal reference. Chemical shifts are reported in ppm with the resonance  
25  
26 resulting from incomplete deuteration of the solvent as the internal standard (<sup>13</sup>CDCl<sub>3</sub>: 7.26  
27  
28 ppm, s and 77.0 ppm, t; <sup>13</sup>CD<sub>3</sub>OD: 4.87 ppm, s, 3.31 ppm, quin and 49.1 ppm, sep; <sup>13</sup>C<sub>2</sub>D<sub>6</sub>OS:  
29  
30 2.49 ppm, quin and 39.52 ppm, sep). Data are reported as follows: chemical shift δ/ppm  
31  
32 (multiplicity, coupling constants *J* (Hz) and integration (<sup>1</sup>H only)). The following  
33  
34 abbreviations were used to explain the multiplicities: s = singlet; d = doublet; t = triplet; q =  
35  
36 quartet; quin = quintet; b = broad; m = multiplet or combination thereof. <sup>13</sup>C signals are  
37  
38 singles, unless otherwise stated. High resolution mass spectrometry (HRMS) was performed  
39  
40 on a H-ESI and APCI mass spectrometer in positive mode and using an ion trap (Orbitrap) as  
41  
42 the mass analyzer type. HRMS signals are reported to 4 decimal places and are within ± 5  
43  
44 ppm of theoretical values. Specific optical rotations were recorded on a Perkin-Elmer 241  
45  
46 polarimeter with a sodium halogen lamp (λ = 589 nm) and a cell path length of 100 mm (*c*

1  
2 given in g/100 mL). Melting points were collected using a Gallenkamp or a Griffin melting  
3  
4 point system using a gradient of 0.5 °C per min.  
5

6 **Biological Material and Methods.** Bovine aortic endothelial cells (BAEC) were isolated by  
7  
8 collagenase digestion, as previously described,<sup>30</sup> and maintained in Dulbecco's modified  
9  
10 Eagle's medium (DMEM) containing glucose (1 g/L) supplemented with 10% FBS  
11  
12 (DMEM/10% FBS). All the cancer cell lines used in this study were obtained from the  
13  
14 American Type Culture Collection (ATCC). Human fibrosarcoma HT-1080, hepatocellular  
15  
16 carcinoma HepG2 and glioblastoma U87MG cells were maintained in Eagle's Minimum  
17  
18 Essential Medium (EMEM) supplemented with 10% FBS. Human colon adenocarcinoma HT-  
19  
20 29 cells and human osteosarcoma U2OS cells were maintained in DMEM containing glucose  
21  
22 (4,5 g/L) supplemented with 10% FBS. Human breast cancer carcinoma MDA-MB-231,  
23  
24 chronic myelogenous leukemia KU812F and histiocytic lymphoma U937 cells were  
25  
26 maintained in RPMI1640 medium supplemented with 10% FBS. Acute promyelocytic  
27  
28 leukemia HL-60 cells were maintained in RPMI1640 medium supplemented with 20% FBS.  
29  
30 All culture medium contained glutamine (2 mM), penicillin (50 IU/mL), streptomycin (50  
31  
32 µg/mL) and amphotericin (1.25 µg/mL) and all cell lines were grown at 37 °C and humidified  
33  
34 5% CO<sub>2</sub> atmosphere.  
35  
36  
37  
38  
39  
40  
41  
42  
43

44 **Aniline 19.** To a solution of bromoaniline **17** (1.0 g, 5.81 mmol, 1.0 equiv) and Pd[PPh<sub>3</sub>]<sub>4</sub>  
45  
46 (336 mg, 0.29 mmol, 0.05 equiv) in DMF (15 mL) was added dropwise allyltri-*n*-butyltin (2.2  
47  
48 mL, 6.97 mmol, 1.2 equiv). The solution was then heated at 80 °C for 12 h. After this time,  
49  
50 the mixture was diluted with diethyl ether and washed with water. The organic layer was  
51  
52 separated and washed with water four times, dried over anhydrous MgSO<sub>4</sub>, filtered and the  
53  
54 solvent evaporated under reduced pressure. The crude product was purified by flash column  
55  
56  
57  
58  
59  
60

1 chromatography (silica gel, 5% EtOAc in hexanes) to obtain aniline **19** (596 mg, 77%) as a  
2 yellow oil:  $R_f = 0.60$  (silica gel, 20% EtOAc in hexanes);  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$   
3  
4 (ppm) 7.13 – 7.05 (m, 2 H), 6.78 (td,  $J = 7.4, 1.1$  Hz, 1 H), 6.72 – 6.69 (m, 1 H), 6.04 – 5.92  
5  
6 (m, 1 H), 5.18 – 5.09 (m, 2 H), 3.64 (bs, 2 H), 3.33 (d,  $J = 6.1$  Hz, 2 H);  $^{13}\text{C NMR}$  (100 MHz,  
7  
8  $\text{CDCl}_3$ )  $\delta$  (ppm) 144.8, 136.0, 130.2, 127.6, 124.0, 118.9, 116.1, 115.9, 36.5; HRMS (H-ESI)  
9  
10  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_9\text{H}_{12}\text{N}$  134.0970; found 134.0967.

11  
12  
13  
14  
15  
16  
17  
18 **Aniline 20.** To a solution of iodonitrobenzene **18**<sup>11</sup> (6.0 g, 16.90 mmol, 1.0 equiv) and  
19  
20  $\text{Pd}[\text{PPh}_3]_4$  (3.0 g, 3.40 mmol, 0.15 equiv) in DMF (30 mL) was added dropwise allyltri-*n*-  
21  
22 butyltin (6.4 mL, 20.30 mmol, 1.2 equiv). The solution was then heated at 60 °C for 15 h.  
23  
24 After this time, the mixture was diluted with  $\text{Et}_2\text{O}$  and water. The organic layer was separated  
25  
26 and washed with water four times, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent  
27  
28 evaporated under reduced pressure. The crude product was purified by flash column  
29  
30 chromatography (silica gel, 100% Hexanes) to obtain the corresponding allyl derivative (3.3 g,  
31  
32 73%) as a yellow oil:  $R_f = 0.56$  (silica gel, 20% EtOAc in hexanes);  $^1\text{H NMR}$  (400 MHz,  
33  
34  $\text{CDCl}_3$ )  $\delta$  (ppm) 7.54 (d,  $J = 2.7$  Hz, 1 H), 7.46 – 7.33 (m, 5 H), 7.28 – 7.25 (m, 1 H), 7.16  
35  
36 (dd,  $J = 8.5, 2.7$  Hz, 1 H), 5.96 (ddt,  $J = 16.6, 10.1, 6.4$  Hz, 1 H), 5.11 (s, 2 H), 5.09 – 5.07 (m,  
37  
38 1 H), 5.05 – 5.03 (m, 1 H), 3.62 (dt,  $J = 6.4, 1.4$  Hz, 2 H);  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ )  $\delta$   
39  
40 (ppm) 157.4, 135.8, 135.4, 132.8, 128.7, 128.3, 127.5, 127.1, 120.4, 116.7, 110.3, 70.6, 36.3;  
41  
42 HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{16}\text{H}_{16}\text{NO}_3$  270.1130; found 270.1129. To a solution  
43  
44 of the allyl derivative obtained above (1.5 g, 5.60 mmol, 1.0 equiv) in acetic acid (15 mL) was  
45  
46 added Zn dust (1.0 g, 16.20 mmol, 3.0 equiv) in ten portions of 100 mg each over 30 min at  
47  
48 25 °C. The reaction mixture was stirred at this temperature until completion monitoring by  
49  
50 TLC (1 h) and then, quenched by addition of 10% NaOH solution at 0 °C. The mixture was  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2 extracted with CH<sub>2</sub>Cl<sub>2</sub>, and the organic phase washed with water, dried over anhydrous  
3  
4 MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The resulting residue was  
5  
6 purified by flash column chromatography (silica gel, 5% EtOAc in hexanes) to obtain **20** (773  
7  
8 mg, 65%) as an orange oil: *R<sub>f</sub>* = 0.38 (silica gel, 20% EtOAc in hexanes); <sup>1</sup>H NMR (400 MHz,  
9  
10 CDCl<sub>3</sub>)  $\delta$  (ppm) 7.46 – 7.37 (m, 4 H), 7.35 – 7.30 (m, 1 H), 6.95 (d, *J* = 8.3 Hz, 1 H), 6.41  
11  
12 (dd, *J* = 8.2, 2.5 Hz, 1 H), 6.35 (d, *J* = 2.5 Hz, 1 H), 5.95 (ddt, *J* = 16.6, 10.5, 6.1 Hz, 1 H),  
13  
14 5.14 – 5.11 (m, 1 H), 5.11 – 5.08 (m, 1 H), 5.03 (s, 2 H), 3.67 (bs, 2 H), 3.26 (dt, *J* = 6.1, 1.5  
15  
16 Hz, 2 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 158.6, 145.8, 137.3, 136.4, 130.9, 128.5,  
17  
18 127.8, 127.4, 116.9, 115.8, 104.9, 102.6, 69.9, 35.8; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for  
19  
20 C<sub>16</sub>H<sub>18</sub>NO 240.1388; found 240.1392.  
21  
22  
23  
24  
25  
26  
27

28 **Dipeptide 22.** To a solution of aniline **19** (700 mg, 5.25 mmol, 1.0 equiv) and Boc-Gly-D-  
29  
30 Ala-OH (**21**)<sup>12</sup> (1.3 g, 5.25 mmol, 1.0 equiv) in DMF (15 mL) was added HATU (3.0 g, 7.88  
31  
32 mmol, 1.5 equiv) and DIPEA (1.0 mL, 5.25 mmol, 1.0 equiv) at 0 °C and the mixture was  
33  
34 stirred for 12 h at 25 °C. After this time, a saturated aqueous NH<sub>4</sub>Cl solution was added and  
35  
36 the organic layer was separated. The aqueous phase was extracted with EtOAc, and the  
37  
38 combined organic layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and  
39  
40 the solvent evaporated under reduced pressure. The crude product was purified by flash  
41  
42 column chromatography (silica gel, 20% EtOAc in hexanes → 50% EtOAc in hexanes) to  
43  
44 obtain dipeptide **22** (1.5 g, 81%) as a white foam: *R<sub>f</sub>* = 0.60 (silica gel, 100% EtOAc); [ $\alpha$ ]<sub>D</sub><sup>25</sup>  
45  
46 = – 6.23 (*c* 0.45, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.96 (bs, 1 H), 7.79 (d, *J* =  
47  
48 8.0 Hz, 1 H), 7.23 (dd, *J* = 8.1, 1.7 Hz, 1 H), 7.18 (dd, *J* = 7.5, 1.5 Hz, 1 H), 7.15 – 7.10 (m, 1  
49  
50 H), 6.87 (s, 1 H), 5.97 (ddt, *J* = 16.2, 10.2, 6.0 Hz, 1 H), 5.15 (dd, *J* = 10.2, 1.6 Hz, 1 H), 5.06  
51  
52 (dd, *J* = 17.2, 1.6 Hz, 1 H), 4.64 – 4.55 (m, 1 H), 3.88 – 3.80 (m, 2 H), 3.41 – 3.33 (m, 2 H),  
53  
54  
55  
56  
57  
58  
59  
60



1  
2 1.47 (d,  $J = 7.0$  Hz, 3 H), 1.44 (s, 9 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 170.4, 170.0,  
3  
4 156.2, 135.9, 135.4, 131.1, 130.2, 127.1, 125.8, 123.9, 116.5, 49.5, 38.7, 36.1, 31.3, 28.3,  
5  
6 17.8; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{19}\text{H}_{28}\text{N}_3\text{O}_4$  362.2080; found 362.2079.  
7  
8  
9

10  
11 **Dipeptide 11.** To a solution of aniline **20** (730 mg, 3.10 mmol, 1.0 equiv) and Boc-Gly-D-  
12  
13 Ala-OH (**21**) (752 mg, 3.10 mmol, 1.0 equiv) in  $\text{CH}_2\text{Cl}_2$  (15 mL) was added HATU (1.7 g,  
14  
15 4.60 mmol, 1.5 equiv) and DIPEA (0.5 mL, 3.10 mmol, 1.0 equiv) and the mixture was stirred  
16  
17 for 12 h at 25 °C. After this time, the reaction mixture was diluted with  $\text{CH}_2\text{Cl}_2$  and washed  
18  
19 sequentially with 1 N HCl and a saturated aqueous  $\text{NaHCO}_3$  solution. The organic layer was  
20  
21 separated, washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent  
22  
23 evaporated under reduced pressure. The crude product was purified by flash column  
24  
25 chromatography (silica gel, 20% EtOAc in hexanes) to obtain dipeptide **11** (1.15 g, 81%) as a  
26  
27 pale brown solid:  $R_f = 0.69$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -3.41$  ( $c$  0.10,  $\text{CH}_2\text{Cl}_2$ ); mp =  
28  
29 83-84 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.07 (bs, 1 H), 7.62 (d,  $J = 2.3$  Hz, 1 H), 7.44  
30  
31 – 7.27 (m, 6 H), 7.05 (d,  $J = 8.5$  Hz, 1 H), 6.94 (d,  $J = 7.3$  Hz, 1 H), 6.73 (dd,  $J = 8.4, 2.6$  Hz,  
32  
33 1 H), 5.94 (ddt,  $J = 16.2, 10.2, 5.9$  Hz, 1 H), 5.14 – 5.04 (m, 2 H), 5.02 (s, 2 H), 4.59 (p,  $J =$   
34  
35 7.0 Hz, 1 H), 3.91 – 3.77 (m, 2 H), 3.37 – 3.23 (m, 2 H), 1.46 (d,  $J = 7.0$  Hz, 3 H), 1.44 (s, 9  
36  
37 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 170.1, 169.9, 157.9, 156.0, 136.9, 136.3, 130.8,  
38  
39 128.5, 127.9, 127.7, 127.5, 122.3, 116.2, 112.2, 109.4, 80.6, 70.0, 49.6, 39.2, 36.5, 28.2, 17.6;  
40  
41 HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{26}\text{H}_{34}\text{N}_3\text{O}_5$  468.2498; found 468.2487.  
42  
43  
44  
45  
46  
47  
48  
49  
50

51 **Diolefin 24.** TFA (5.0 mL) was added to a solution of dipeptide **22** (360 mg, 0.99 mmol, 1.0  
52  
53 equiv) in  $\text{CH}_2\text{Cl}_2$  (60 mL) at 0 °C and the reaction mixture was stirred at 25 °C until depletion  
54  
55 of starting material as judged by TLC (3 h). Then, the TFA excess was removed under  
56  
57  
58  
59  
60

1  
2 reduced pressure to obtain the corresponding ammonium trifluoroacetate salt as a brown solid  
3  
4 which was dissolved in DMF (10 mL). To this solution, Kosher acid **23** (0.1 mL, 0.99 mmol,  
5  
6 1.0 equiv), HATU (379 mg, 0.99 mmol, 1.0 equiv) and DIPEA (0.52 mL, 2.99 mmol, 3.0  
7  
8 equiv) were added and the resulting solution was stirred at 25 °C for 12 h. After this time, a  
9  
10 saturated aqueous NH<sub>4</sub>Cl solution was added and the organic layer was separated. The  
11  
12 aqueous phase was extracted with EtOAc, and the combined organic layers were washed with  
13  
14 brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced  
15  
16 pressure. The resulting residue was purified by flash column chromatography (silica gel, 30%  
17  
18 EtOAc in hexanes) to obtain diolefin **24** (205 mg, 60% over two steps) as a white solid:  $R_f =$   
19  
20 0.23 (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -8.50$  ( $c$  0.51, CH<sub>2</sub>Cl<sub>2</sub>); mp = 80-81 °C ; <sup>1</sup>H NMR  
21  
22 (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.82 (bs, 1 H), 7.29 – 7.27 (m, 1 H), 7.25 – 7.23 (m, 1 H), 7.21 –  
23  
24 7.17 (m, 1 H), 7.16 – 7.11 (m, 1 H), 6.64 (d,  $J = 6.4$  Hz, 1 H), 6.17 (bs, 1 H), 5.98 (dt,  $J =$   
25  
26 16.4, 6.3 Hz, 1 H), 5.83 (dt,  $J = 16.9, 5.8$  Hz, 1 H), 5.19 – 5.14 (m, 1 H), 5.12 – 5.09 (m, 1 H),  
27  
28 5.08 – 5.01 (m, 2 H), 4.60 – 4.51 (m, 1 H), 3.98 (d,  $J = 5.4$  Hz, 2 H), 3.41 – 3.35 (m, 2 H),  
29  
30 2.44 – 2.33 (m, 4 H), 1.48 (d,  $J = 7.0$  Hz, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 173.1,  
31  
32 169.9, 169.1, 136.7, 136.0, 135.3, 130.4, 130.3, 127.4, 125.7, 123.6, 116.6, 115.9, 49.7, 43.1,  
33  
34 36.4, 35.4, 29.4, 17.9; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for C<sub>19</sub>H<sub>26</sub>N<sub>3</sub>O<sub>3</sub> 344.1974; found  
35  
36 344.1965.

37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47 **Diolefin 25.** Dipeptide **11** (150 mg, 0.32 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was treated  
48  
49 with TFA (1.6 mL) in exactly the same manner as previously described for synthesis of **24**. A  
50  
51 solution of the ammonium salt and Kosher acid **23** (30  $\mu$ L, 0.32 mmol, 1.0 equiv), in CH<sub>2</sub>Cl<sub>2</sub>  
52  
53 (10 mL) was treated with HATU (122 mg, 0.32 mmol, 1.0 equiv) and DIPEA (0.17 mL, 0.96  
54  
55 mmol, 3.0 equiv) and the resulting solution was stirred at 25 °C for 12 h. After this time, a  
56  
57  
58  
59  
60

1 saturated aqueous  $\text{NH}_4\text{Cl}$  solution was added and the organic layer was separated. The  
2  
3 aqueous phase was extracted with EtOAc, and the combined organic layers were washed with  
4  
5 brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent evaporated under reduced  
6  
7 pressure. The resulting residue was purified by flash column chromatography (silica gel, 20%  
8  
9 EtOAc in hexanes  $\rightarrow$  80% EtOAc in hexanes) to obtain diolefin **25** (140 mg, 97% over two  
10  
11 steps) as a white solid:  $R_f = 0.24$  (silica gel, 100%);  $[\alpha]_D^{25} = -6.37$  ( $c$  0.06,  $\text{CH}_2\text{Cl}_2$ ); mp =  
12  
13 87-88 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.02 (bs, 1 H), 7.63 (bs, 1 H), 7.45 – 7.28 (m,  
14  
15 6 H), 7.05 (d,  $J = 8.5$  Hz, 1 H), 6.74 (dd,  $J = 8.4, 2.6$  Hz, 1 H), 6.43 (bs, 1 H), 5.94 (ddd,  $J =$   
16  
17 16.5, 11.1, 5.8 Hz, 1 H), 5.81 (ddd,  $J = 17.0, 11.2, 6.2$  Hz, 1 H), 5.15 – 5.04 (m, 4 H), 5.03 (s,  
18  
19 2 H), 4.60 – 4.52 (m, 1 H), 4.04 – 3.93 (m, 2 H), 3.37 – 3.24 (m, 2 H), 2.43 – 2.31 (m, 4 H),  
20  
21 1.46 (d,  $J = 7.0$  Hz, 3 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 173.5, 170.8, 169.5, 157.7,  
22  
23 136.9, 136.7, 136.3, 136.0, 130.7, 128.5, 127.9, 127.5, 123.6, 116.1, 115.6, 112.2, 110.4, 70.0,  
24  
25 49.8, 43.1, 38.6, 36.2, 36.1, 29.3, 17.7; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{26}\text{H}_{32}\text{N}_3\text{O}_4$   
26  
27 450.2393; found 450.2390.

28  
29  
30  
31  
32  
33  
34  
35  
36  
37 **Macrocycle 26.** Diolefin **24** (45 mg, 0.13 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation  
38  
39 catalyst (8 mg, 0.01 mmol, 0.10 equiv) and *p*-benzoquinone (1 mg, 0.01 mmol, 0.10 equiv)  
40  
41 were dissolved in degassed  $\text{CH}_2\text{Cl}_2$  (7 mL, 0.02 M) and the reaction mixture was heated at 40  
42  
43 °C for 12 h. After this time, the solvent was removed under reduced pressure and the resulting  
44  
45 crude product was purified by flash column chromatography (silica gel, 20% EtOAc in  
46  
47 hexanes  $\rightarrow$  2% MeOH in  $\text{CH}_2\text{Cl}_2$ ) to obtain macrocycle **26** (31 mg, 75%) as a white solid:  $R_f$   
48  
49 = 0.56 (silica gel, 10% MeOH in  $\text{CH}_2\text{Cl}_2$ );  $[\alpha]_D^{25} = -6.11$  ( $c$  0.08, MeOH); mp = 188-189 °C;  
50  
51  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta$  (ppm) 8.50 (bs, 1 H), 8.45 (d,  $J = 7.3$  Hz, 1 H), 8.37 (t,  $J =$   
52  
53 5.9 Hz, 1 H), 7.46 (dd,  $J = 7.8, 0.9$  Hz, 1 H), 7.23 – 7.09 (m, 3 H), 5.44 (dt,  $J = 15.9, 5.4$  Hz, 1  
54  
55  
56  
57  
58  
59  
60

1  
2 H), 5.08 (dt,  $J = 15.9, 6.4$  Hz, 1 H), 4.35 (p,  $J = 7.1$  Hz, 1 H), 3.85 (dd,  $J = 14.7, 6.4$  Hz, 1 H),  
3  
4 3.44 (dd,  $J = 14.7, 5.7$  Hz, 1 H), 3.26 (dd,  $J = 15.8, 3.6$  Hz, 1 H), 3.08 (dd,  $J = 15.5, 5.1$  Hz, 1  
5  
6 H), 2.28 – 2.08 (m, 4 H), 1.31 (d,  $J = 7.2$  Hz, 3 H);  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta$  (ppm)  
7  
8 173.1, 171.3, 170.8, 136.5, 134.4, 130.9, 130.3, 128.5, 127.1, 126.0, 125.9, 49.7, 43.7, 34.8,  
9  
10 31.1, 27.9, 17.1; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{17}\text{H}_{22}\text{N}_3\text{O}_3$  316.1661; found  
11  
12 316.1655.  
13  
14  
15  
16  
17

18 **Macrocycle 27.** Diolefin **25** (135 mg, 0.30 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation  
19  
20 catalyst (20 mg, 0.03 mmol, 0.10 equiv) and *p*-benzoquinone (4.0 mg, 0.03 mmol, 0.10 equiv)  
21  
22 were dissolved in degassed  $\text{CH}_2\text{Cl}_2$  (16 mL, 0.02 M) and the reaction mixture was heated at 40  
23  
24  $^\circ\text{C}$  for 12 h. After this time, the solvent was removed under reduced pressure and the resulting  
25  
26 crude product was purified by flash column chromatography (silica gel, 30% EtOAc in  
27  
28 hexanes  $\rightarrow$  3% MeOH in  $\text{CH}_2\text{Cl}_2$ ) to obtain macrocycle **27** (100 mg, 79%) as a white solid:  $R_f$   
29  
30 = 0.5 (silica gel, 10% MeOH in  $\text{CH}_2\text{Cl}_2$ );  $[\alpha]_D^{25} = -4.97$  ( $c$  0.07, MeOH); mp = 198–199  $^\circ\text{C}$ ;  
31  
32  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  (ppm) 8.54 (bs, 1 H), 8.52 (d,  $J = 7.3$  Hz, 1 H), 8.37 (t,  $J =$   
33  
34 6.0 Hz, 1 H), 7.46 – 7.31 (m, 5 H), 7.29 (d,  $J = 2.6$  Hz, 1 H), 7.06 (d,  $J = 8.4$  Hz, 1 H), 6.77  
35  
36 (dd,  $J = 8.4, 2.7$  Hz, 1 H), 5.41 (dt,  $J = 15.9, 5.4$  Hz, 1 H), 5.13 – 5.05 (m, 1 H), 5.04 (s, 2 H),  
37  
38 4.36 (p,  $J = 7.1$  Hz, 1 H), 3.86 (dd,  $J = 14.8, 6.5$  Hz, 1 H), 3.43 (dd,  $J = 14.7, 5.6$  Hz, 1 H),  
39  
40 3.20 (dd,  $J = 16.4, 3.9$  Hz, 1 H), 3.03 (dd,  $J = 16.2, 5.0$  Hz, 1 H), 2.28 – 2.03 (m, 4 H), 1.30  
41  
42 (d,  $J = 7.2$  Hz, 3 H);  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta$  (ppm) 173.1, 171.2, 171.0, 157.4,  
43  
44 137.6, 137.3, 131.5, 130.7, 128.8, 128.2, 128.1, 128.1, 128.0, 125.8, 111.8, 69.7, 49.7, 43.4,  
45  
46 34.7, 33.7, 27.9, 16.8; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{24}\text{H}_{28}\text{N}_3\text{O}_4$  422.2080; found  
47  
48 422.2074.  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

**Diolefin 8.** Dipeptide **11** (328 mg, 0.70 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was treated with TFA (3.5 mL) in exactly the same manner as previously described for synthesis of **24**. A solution of the ammonium salt and acid **14**<sup>10</sup> (171 mg, 0.70 mmol, 1.0 equiv), in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was treated with HATU (400 mg, 1.05 mmol, 1.0 equiv) and DIPEA (0.35 mL, 2.10 mmol, 3.0 equiv) and the resulting solution was stirred at 25 °C for 12 h. After this time, a saturated aqueous NH<sub>4</sub>Cl solution was added and the organic layer was separated. The aqueous phase was extracted with EtOAc, and the combined organic layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 20% EtOAc in hexanes) to obtain diolefin **8** (250 mg, 74% over two steps) as a white solid:  $R_f = 0.73$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -6.06$  (*c* 0.05, CH<sub>2</sub>Cl<sub>2</sub>); mp = 98-99 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 8.12 (bs, 1 H), 7.68 (d, *J* = 7.3 Hz, 1 H), 7.54 (d, *J* = 2.6 Hz, 1 H), 7.44 – 7.27 (m, 5 H), 7.06 (m, 1 H), 6.78 – 6.71 (m, 1 H), 6.70 – 6.63 (m, 1 H), 5.92 (ddt, *J* = 16.5, 10.2, 6.1 Hz, 1 H), 5.79 (ddd, *J* = 17.3, 10.3, 7.2 Hz, 1 H), 5.27 (m, 1 H), 5.10 (m, 1 H), 5.01 (s, 2 H), 4.60 – 4.51 (m, 1 H), 4.33 (dd, *J* = 17.1, 7.4 Hz, 1 H), 4.15 – 4.04 (m, 2 H), 3.62 (dd, *J* = 17.1, 4.6 Hz, 1 H), 3.36 – 3.22 (m, 1 H), 2.32 (dq, *J* = 13.7, 6.8 Hz, 1 H), 1.42 (d, *J* = 7.0 Hz, 3 H), 1.10 (d, *J* = 6.9 Hz, 3 H), 0.94 (s, 9 H), 0.19 (s, 3 H), 0.17 (s, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 175.5, 170.8, 169.8, 157.8, 138.2, 136.8, 136.3, 136.1, 130.9, 128.5, 127.9, 127.5, 122.7, 117.9, 116.3, 112.2, 109.9, 70.1, 49.8, 46.6, 43.1, 38.9, 38.5, 25.2, 18.1, 17.4, 16.2, 13.6, 2.9, -4.5, -4.7; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>33</sub>H<sub>48</sub>N<sub>3</sub>O<sub>5</sub>Si 594.3363; found 594.3365.

**Diolefin 7.** To a solution of diolefin **8** (90 mg, 0.15 mmol, 1.0 equiv) in THF (10 mL) was added HF•pyr (0.6 mL) at 0 °C and the mixture was stirred for 12 h. The reaction mixture was

1  
2 quenched with a saturated aqueous NaHCO<sub>3</sub> solution and diluted with EtOAc. The organic  
3  
4 layer was separated and the aqueous phase was extracted with EtOAc. The combined organic  
5  
6 layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent  
7  
8 evaporated under reduced pressure. The residue was purified by flash column chromatography  
9  
10 (silica gel, 2% MeOH in CH<sub>2</sub>Cl<sub>2</sub>) to obtain diolefin **7** (60 mg, 83%) as a white solid: R<sub>f</sub> = 0.66  
11  
12 (silica gel, 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub>); [ $\alpha$ ]<sup>25</sup><sub>D</sub> = -4.91 (*c* 0.08, CH<sub>2</sub>Cl<sub>2</sub>); mp = 90-91 °C; <sup>1</sup>H NMR  
13  
14 (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 8.16 (bs, 1 H), 7.69 (d, *J* = 7.1 Hz, 1 H), 7.56 (d, *J* = 2.6 Hz, 1 H),  
15  
16 7.43 – 7.28 (m, 5 H), 7.05 (d, *J* = 8.4 Hz, 1 H), 6.88 – 6.81 (m, 1 H), 6.73 (dd, *J* = 8.4, 2.6 Hz,  
17  
18 1 H), 5.93 (ddd, *J* = 16.3, 11.1, 6.0 Hz, 1 H), 5.80 (ddd, *J* = 17.3, 10.3, 7.2 Hz, 1 H), 5.28 (dd,  
19  
20 *J* = 16.0, 2.2 Hz, 1 H), 5.20 (d, *J* = 10.3 Hz, 1 H); 5.14 – 5.04 (m, 2 H), 5.01 (s, 2 H), 4.59 –  
21  
22 4.48 (m, 1 H), 4.31 (dd, *J* = 17.2, 7.5 Hz, 1 H), 3.62 (dd, *J* = 17.1, 4.8 Hz, 1 H), 3.33 – 3.26  
23  
24 (m, 2 H), 2.35 (dq, *J* = 13.6, 6.8 Hz, 1 H), 1.41 (d, *J* = 7.1 Hz, 3 H), 1.09 (d, *J* = 6.8 Hz, 3 H);  
25  
26 <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 175.5, 170.7, 169.9, 157.8, 138.4, 136.9, 136.3, 136.2,  
27  
28 130.8, 128.5, 127.9, 127.5, 122.8, 117.7, 116.3, 111.9, 109.9, 70.1, 49.8, 43.2, 36.4, 23.8,  
29  
30 19.6, 17.3, 13.5; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>27</sub>H<sub>34</sub>N<sub>3</sub>O<sub>5</sub> 480.2498; found  
31  
32 480.2502.  
33  
34  
35  
36  
37  
38  
39  
40  
41

42 **Macrocycle 28.** Diolefin **7** (30 mg, 0.06 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation  
43  
44 catalyst (4.0 mg, 0.01 mmol, 0.10 equiv) and *p*-benzoquinone (1.0 mg, 0.01 mmol, 0.10  
45  
46 equiv) were dissolved in degassed CH<sub>2</sub>Cl<sub>2</sub> (3 mL, 0.02 M) and the reaction mixture was  
47  
48 heated at 40 °C for 12 h. After this time, the solvent was removed under reduced pressure and  
49  
50 the resulting crude product was purified by flash column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>  
51  
52 → 5% MeOH in CH<sub>2</sub>Cl<sub>2</sub>) to obtain **28** (20 mg, 71%) as a white solid: R<sub>f</sub> = 0.44 (silica gel,  
53  
54 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub>); [ $\alpha$ ]<sup>25</sup><sub>D</sub> = -8.4 (*c* 0.05, MeOH); mp = 203-204 °C; <sup>1</sup>H NMR (400  
55  
56  
57  
58  
59  
60

1  
2 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  (ppm) 8.67 (bs, 1 H), 8.49 (t, *J* = 5.5 Hz, 1 H), 7.46 – 7.29 (m, 6 H), 7.07  
3  
4 (d, *J* = 2.7 Hz, 1 H), 6.82 (dd, *J* = 8.3, 2.7 Hz, 1 H), 5.62 (dt, *J* = 15.9, 5.11 Hz, 1 H), 5.05 (s,  
5  
6 2 H), 4.98 (dd, *J* = 15.9, 7.0 Hz, 1 H), 4.32 – 4.23 (m, 1 H), 4.13 (dd, *J* = 6.6, 4.7 Hz, 1 H),  
7  
8 3.85 (dd, *J* = 14.3, 5.8 Hz, 1 H), 3.19 – 2.99 (m, 3 H), 2.68 – 2.64 (m, 1 H), 1.30 (d, *J* = 7.3  
9  
10 Hz, 3 H), 0.85 (d, *J* = 6.9 Hz, 3 H); <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  (ppm) 174.9, 171.5,  
11  
12 170.6, 157.5, 137.6, 137.5, 131.6, 131.0, 130.2, 128.8, 128.2, 128.1, 127.6, 113.3, 112.3, 72.7,  
13  
14 69.7, 49.6, 44.9, 43.6, 33.8, 16.9, 11.0; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>25</sub>H<sub>30</sub>N<sub>3</sub>O<sub>5</sub>  
15  
16 452.2186; found 452.2188.  
17  
18  
19  
20  
21  
22

23 **Aniline 30.** To a solution of allylic alcohol **29**<sup>17</sup> (1.9 g, 10.60 mmol, 1.0 equiv) in EtOH (60  
24 mL) was added a solution of NH<sub>4</sub>Cl (2.8 g, 53.02 mmol, 5.0 equiv) in water (40 mL) followed  
25  
26 by Zn dust (10 g, 159.06 mmol, 15.0 equiv) in ten portions of 1 g each over 30 min at 25 °C.  
27  
28 The mixture was stirred at this temperature for 12 h and then the reaction mixture was diluted  
29  
30 with CH<sub>2</sub>Cl<sub>2</sub> and water, then filtered and rinsed with CH<sub>2</sub>Cl<sub>2</sub>. The filtrate was diluted with  
31  
32 water and extracted with CH<sub>2</sub>Cl<sub>2</sub>, and the organic phase washed with brine, dried over  
33  
34 anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The resulting  
35  
36 residue was purified by flash column chromatography (silica gel, 10% EtOAc in hexanes) to  
37  
38 obtain aniline **30** (1.4 g, 85%) as a yellow oil: *R*<sub>f</sub> = 0.48 (silica gel, 40% EtOAc in hexanes);  
39  
40 <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.13 – 7.06 (m, 2 H), 6.75 (td, *J* = 7.5, 1.2 Hz, 1 H),  
41  
42 6.65 (dd, *J* = 7.9, 0.9 Hz, 1 H), 6.15 (ddd, *J* = 17.0, 10.4, 5.3 Hz, 1 H), 5.34 (dt, *J* = 17.2, 1.4  
43  
44 Hz, 1 H), 5.25 (dt, *J* = 10.4, 1.5 Hz, 1 H), 5.15 (d, *J* = 5.2 Hz, 1 H), 3.75 (bs, 2 H); <sup>13</sup>C NMR  
45  
46 (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 144.8, 138.3, 128.9, 128.0, 126.4, 118.6, 117.1, 115.4, 74.2;  
47  
48  
49 HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>9</sub>H<sub>12</sub>NO 150.0919; found 150.0907.  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

**Dipeptide 31.** To a solution of aniline **30** (1.3 g, 8.71 mmol, 1.0 equiv) and Boc-Gly-D-Ala-OH (**21**) (2.1 g, 8.71 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was added HATU (3.6 g, 13.07 mmol, 1.5 equiv) and DIPEA (1.5 mL, 8.71 mmol, 1.0 equiv) and the mixture was stirred for 12 h at 25 °C. After this time, a saturated aqueous NH<sub>4</sub>Cl solution was added and the organic layer was separated. The aqueous phase was extracted with EtOAc, and the combined organic layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 20% EtOAc in hexanes → 40% EtOAc in hexanes) to obtain dipeptide **31** (2.5 g, 76%, 1:1 mixture of diastereoisomers) as a white foam. Data assigned for the mixture of diastereoisomers:  $R_f = 0.56$  (silica gel, 100% EtOAc); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 9.75 (d,  $J = 8.6$  Hz, 1 H), 7.96 (t,  $J = 8.2$  Hz, 1 H), 7.39 – 7.28 (m, 1 H), 7.25 – 7.15 (m, 1 H), 7.10 (m, 1 H), 7.06 – 6.98 (m, 1 H), 6.08 – 5.92 (m, 1 H), 5.85 – 5.64 (m, 1 H), 5.20 (d,  $J = 5.7$  Hz, 1 H), 5.16 – 5.07 (m, 1 H), 4.54 – 4.35 (m, 1 H), 3.89 – 3.62 (m, 2 H), 1.37 (s, 9 H), 1.33 (d,  $J = 7.2$  Hz, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 170.5, 170.4, 170.4, 170.1, 170.1, 138.1, 138.1, 136.3, 128.6, 128.5, 128.1, 127.9, 124.4, 122.4, 122.3, 116.0, 115.9, 80.9, 60.4, 50.3, 50.1, 44.1, 44.0, 28.3, 28.3, 17.8, 17.8; HRMS (H-ESI)  $m/z$ : [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>28</sub>N<sub>3</sub>O<sub>5</sub> 378.2029; found 378.2034.

**Trifluoroacetate Derivative 33.** Dipeptide **31** (1.0 g, 2.65 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was treated with TFA (13 mL) in exactly the same manner as previously described for synthesis of **24**. A solution of the ammonium salt and Kosher acid **23** (0.27 mL, 2.65 mmol, 1.0 equiv), in CH<sub>2</sub>Cl<sub>2</sub> (35 mL) was treated with HATU (1.0 g, 2.65 mmol, 1.0 equiv) and DIPEA (1.4 mL, 7.95 mmol, 3.0 equiv) and the resulting solution was stirred at 25 °C for 12 h. After this time, a saturated aqueous NH<sub>4</sub>Cl solution was added and the organic layer was



1  
2 separated. The aqueous phase was extracted with EtOAc, and the combined organic layers  
3  
4 were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated  
5  
6 under reduced pressure. The resulting residue was purified by flash column chromatography  
7  
8 (silica gel, 30% EtOAc in hexanes) to obtain trifluoroacetate derivative **33** (784 mg, 65%) as a  
9  
10 white solid:  $R_f = 0.18$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -9.14$  ( $c$  0.12, CH<sub>2</sub>Cl<sub>2</sub>); mp = 109-  
11  
12 110 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 8.62 (bs, 1 H), 7.55 – 7.35 (m, 2 H), 7.25 – 7.20  
13  
14 (m, 1 H), 7.18 – 7.12 (m, 1 H), 6.85 (d,  $J = 15.4$  Hz, 1 H), 6.15 (dt,  $J = 15.4, 6.6$  Hz, 1 H),  
15  
16 5.77 – 5.66 (m, 1 H), 5.01 – 4.90 (m, 4 H), 4.69 – 4.58 (m, 1 H), 4.01 – 3.90 (m, 2 H), 2.37 –  
17  
18 2.22 (m, 4 H), 1.42 (d,  $J = 5.4$  Hz, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 173.8, 171.2,  
19  
20 169.9, 157.7, 136.7, 134.3, 132.0, 129.7, 129.1, 126.7, 126.3, 125.1, 123.1, 115.7, 113.1, 68.5,  
21  
22 49.8, 43.3, 35.1, 29.3, 17.4; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for C<sub>21</sub>H<sub>25</sub>F<sub>3</sub>N<sub>3</sub>O<sub>5</sub> 456.1746;  
23  
24 found 456.1752.  
25  
26  
27  
28  
29  
30  
31  
32

33 **Acid 36.** A 5 mL polypropylene syringe fitted with polyethylene porous disk charged with 2-  
34 chlorotriyl chloride (CTC) resin (300 mg, L=1.3 mmol/g, 0.39 mmol, 1.0 equiv), was loaded  
35 with a solution of Fmoc-D-Ala-OH (**34**) (364 mg, 1.17 mmol, 3.0 equiv) and DIPEA (0.23  
36 mL, 1.36 mmol, 3.5 equiv) in dry DMF (3 mL). The resulting suspension was shaken at 280  
37 rpm for 30 h, then the solution was unloaded and the resin was washed by shaking with dry  
38 DMF (5 x 3 mL). The resulting yellow resin was used in the next step.  
39

40 The polypropylene syringe loaded with the yellow resin was treated with a solution of 20%  
41 piperidine in DMF (3 x 3 mL x 10 min at 280 rpm). After the last run, the resin was washed  
42 with dry DMF (5 x 3 mL) and loaded with a solution of Fmoc-Gly-OH (**35**) (232 mg, 0.78  
43 mmol, 2.0 equiv), HOBt (105 mg, 0.78 mmol, 2.0 equiv) and DIC (0.15 mL, 0.97 mmol, 2.5  
44 equiv) in dry DMF (3 mL). The resulting suspension was shaken at 280 rpm for 24 h, and then  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2 the solution was unloaded and the resin washed with dry DMF (5 x 3 mL). The resulting  
3  
4 yellow resin was used in the next step.  
5

6  
7 To a 5 mL polypropylene syringe fitted with polyethylene porous disk and loaded with the  
8  
9 resulting Fmoc protected dipeptide was added a solution of 20% piperidine in DMF (3 x 3 mL  
10  
11 x 10 min at 280 rpm). After the last run, the resin was washed with dry DMF (5 x 3 mL) and  
12  
13 treated with a solution of Koser acid **23** (80  $\mu$ L, 0.78 mmol, 2.0 equiv) HOBt (105 mg, 0.78  
14  
15 mmol, 2.0 equiv) and DIC (0.2 mL, 0.97 mmol, 2.5 equiv) in dry DMF (3 mL). The resulting  
16  
17 suspension was shaken at 280 rpm for 24 h, and then the solution was unloaded and the  
18  
19 resin washed with dry DMF (5 x 3 mL). The resulting yellow resin was treated with a solution  
20  
21 of CH<sub>2</sub>Cl<sub>2</sub>/AcOH/TFE (7:2:1, 3 mL) for 30 min. After that, the solution was collected and the  
22  
23 resin washed with CH<sub>2</sub>Cl<sub>2</sub> (2 x 3 mL). All the collected organic solvents were evaporated  
24  
25 under reduced pressure and the resulting acid **36** (71 mg, 81 % overall yield from CTC resin)  
26  
27 was obtained as a white solid which not required further purification:  $R_f = 0.21$  (silica gel,  
28  
29 100% EtOAc);  $[\alpha]_D^{25} = -5.21$  ( $c$  0.07, CH<sub>2</sub>Cl<sub>2</sub>); mp = 89-90 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  
30  
31  $\delta$  (ppm) 6.91 (bs, 1 H), 6.62 (bs, 1 H), 5.84 (ddd,  $J = 16.2, 11.1, 5.9$  Hz, 1 H), 5.10 (dd,  $J =$   
32  
33 17.1, 1.5 Hz, 1 H), 5.04 (dd,  $J = 10.3, 1.0$  Hz, 1 H), 4.63 – 4.53 (m, 1 H), 4.10 (dd,  $J = 16.4,$   
34  
35 5.5 Hz, 1 H), 3.94 (dd,  $J = 17.0, 4.9$  Hz, 1 H), 2.46 – 2.35 (m, 4 H), 1.48 (d,  $J = 6.9$  Hz, 3 H);  
36  
37 <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 175.4, 175.3, 162.9, 136.7, 115.8, 42.5, 36.7, 31.6,  
38  
39 29.4, 23.3; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for C<sub>10</sub>H<sub>17</sub>N<sub>2</sub>O<sub>4</sub> 229.1188; found 229.1194.  
40  
41  
42  
43  
44  
45  
46  
47  
48

49 **Diolefin 32.** A solution of the aniline **30** (31 mg, 0.20 mmol, 1.0 equiv) and acid **36** (46 mg,  
50  
51 0.20 mmol, 1.0 equiv) in dry DMF (8 mL) was treated with HATU (76 mg, 0.20 mmol, 1.0  
52  
53 equiv) and DIPEA (40  $\mu$ L, 0.20 mmol, 1.0 equiv) and the resulting solution was stirred at 25  
54  
55 °C for 12 h. After this time, a saturated aqueous NH<sub>4</sub>Cl solution was added and the organic  
56  
57  
58  
59  
60

1  
2 layer was separated. The aqueous phase was extracted with EtOAc, and the combined organic  
3  
4 layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent  
5  
6 evaporated under reduced pressure. The crude product was purified by flash column  
7  
8 chromatography (silica gel, 20% EtOAc in hexanes → 80% EtOAc in hexanes) to obtain  
9  
10 diolefin **32** (55 mg, 77%, 1:1 mixture of diastereoisomers) as a white foam. Data assigned for  
11  
12 the mixture of diastereoisomers:  $R_f = 0.37$  (silica gel, 100% EtOAc); <sup>1</sup>H NMR (400 MHz,  
13  
14 CDCl<sub>3</sub>)  $\delta$  (ppm) 9.83 (bs, 1 H), 9.66 (bs, 1 H), 8.16 (d,  $J = 8.1$  Hz, 1 H), 8.11 (d,  $J = 8.1$  Hz, 1  
15  
16 H), 7.32 – 7.23 (m, 4 H), 7.16 – 7.10 (m, 2 H), 7.06 (td,  $J = 7.5, 0.7$  Hz, 2 H), 6.80 – 6.66 (m,  
17  
18 2 H), 6.14 – 5.99 (m, 2 H), 5.82 (dd,  $J = 11.6, 5.2$  Hz, 1 H), 5.76 (dd,  $J = 10.3, 6.2$  Hz, 1 H),  
19  
20 5.35 – 5.25 (m, 2 H), 5.24 – 5.22 (m, 2 H), 5.20 – 5.12 (m, 2 H), 5.07 (dd,  $J = 18.3, 2.9$  Hz, 2  
21  
22 H), 5.04 – 4.99 (m, 2 H), 4.84 (bs, 2 H), 4.56 (dd,  $J = 14.7, 7.4$  Hz, 1 H), 4.48 (dd,  $J = 14.1,$   
23  
24 6.9 Hz, 1 H), 4.14 (dd,  $J = 6.6, 2.7$  Hz, 1 H), 4.10 (dd,  $J = 6.2, 3.2$  Hz, 1 H), 3.82 (d,  $J = 4.8$   
25  
26 Hz, 1 H), 3.79 – 3.75 (m, 1 H), 2.40 – 2.29 (m, 8 H), 1.46 (d,  $J = 7.8$  Hz, 3 H), 1.44 (d,  $J = 7.8$   
27  
28 Hz, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 174.0, 173.9, 170.5, 170.4, 169.4, 169.3,  
29  
30 138.4, 138.2, 136.8, 136.5, 131.0, 130.8, 128.5, 128.4, 128.4, 128.3, 128.1, 127.8, 124.3,  
31  
32 124.2, 122.2, 122.1, 115.8, 115.7, 115.7, 115.6, 50.6, 50.2, 42.9, 42.8, 35.3, 35.2, 31.9, 31.4,  
33  
34 29.6, 29.3, 17.7, 17.6; HRMS (H-ESI)  $m/z$ : [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>26</sub>N<sub>3</sub>O<sub>4</sub> 360.1923; found  
35  
36 360.1917.  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46

47 **Silyl Ether 37.** To a solution of **32** (45 mg, 0.14 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was  
48  
49 added 2,6-lutidine (30  $\mu$ L, 0.25 mmol, 2.0 equiv) at 0 °C and the mixture was stirred 10 min at  
50  
51 this temperature. After this time TBSOTf (60  $\mu$ L, 0.25 mmol, 2.0 equiv) was added at 0 °C  
52  
53 and the mixture was stirred 12 h at 25 °C. Then, the reaction was quenched by addition of  
54  
55 H<sub>2</sub>O. After decantation of the organic layer, the aqueous phase was extracted with EtOAc, and  
56  
57  
58  
59  
60

1  
2 the combined organic layers were washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered  
3  
4 and the solvent evaporated under reduced pressure. The crude product was purified by flash  
5  
6 column chromatography (silica gel, 20% EtOAc in hexanes  $\rightarrow$  100% EtOAc) to obtain silyl  
7  
8 ether **37** (50 mg, 85%, 1:1 mixture of diastereoisomers) as a white solid. Data assigned for the  
9  
10 mixture of diastereoisomers:  $R_f = 0.80$  (silica gel, 100% EtOAc); mp = 121-122 °C;  $^1\text{H NMR}$   
11  
12 (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.16 (dd,  $J = 8.1, 7.9$  Hz, 1 H), 7.31 – 7.27 (m, 1 H), 7.12 – 7.03  
13  
14 (m, 2 H), 7.01 – 6.95 (m, 1 H), 6.42 (d,  $J = 5.2$  Hz, 1 H), 6.03 – 5.91 (m, 1 H), 5.88 – 5.76 (m,  
15  
16 1 H), 5.33 – 5.10 (m, 3 H), 5.10 – 4.98 (m, 2 H), 4.47 (p,  $J = 6.9$  Hz, 1 H), 4.02 – 3.96 (m, 2  
17  
18 H), 2.44 – 2.36 (m, 2 H), 2.37 – 2.31 (m, 2 H), 1.50 – 1.42 (m, 3 H), 0.91 (s, 9 H), 0.14 (s, 3  
19  
20 H), 0.00 (s, 3 H);  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 172.7, 169.7, 169.7, 168.3, 168.2,  
21  
22 138.9, 138.7, 136.9, 136.4, 136.3, 130.5, 130.2, 128.7, 128.6, 127.9, 127.9, 124.4, 124.3,  
23  
24 122.3, 122.3, 115.7, 114.5, 114.4, 50.1, 49.9, 42.9, 38.6, 35.5, 29.4, 25.7, 19.1, 19.1, 18.3,  
25  
26 18.3, -4.9, -5.0, -5.1, -5.1; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{25}\text{H}_{40}\text{N}_3\text{O}_4\text{Si}$  474.2788;  
27  
28 found 474.2775.  
29  
30  
31  
32  
33  
34  
35  
36

37 **Macrocycle 38.** Diolefin **32** (30 mg, 0.08 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation  
38  
39 catalyst (5.0 mg, 0.01 mmol, 0.10 equiv) and *p*-benzoquinone (1.0 mg, 0.01 mmol, 0.10  
40  
41 equiv) were dissolved in degassed  $\text{CH}_2\text{Cl}_2$  (4 mL, 0.02 M) and the reaction mixture was  
42  
43 heated at 40 °C for 15 h. After this time, the solvent was removed under reduced pressure and  
44  
45 the resulting crude product was purified by flash column chromatography (silica gel,  $\text{CH}_2\text{Cl}_2$   
46  
47  $\rightarrow$  5% MeOH in  $\text{CH}_2\text{Cl}_2$ ) to obtain macrocycle **38** (21 mg, 73%, 1:1 mixture of  
48  
49 diastereoisomers) as a white solid. Data assigned for the mixture of diastereoisomers:  $R_f =$   
50  
51 0.46 (silica gel, 10% MeOH in  $\text{CH}_2\text{Cl}_2$ ); mp = 153-154 °C;  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$   
52  
53 (ppm) 7.58 (dd,  $J = 14.1, 8.0$  Hz, 2 H), 7.45 – 7.38 (m, 2 H), 7.30 – 7.20 (m, 4 H), 5.78 – 5.68  
54  
55  
56  
57  
58  
59  
60

(m, 1 H), 5.58 – 5.51 (m, 1 H), 5.35 (dt,  $J = 14.6, 6.9$  Hz, 1 H), 5.15 (d,  $J = 6.5$  Hz, 1 H), 5.10 (d,  $J = 6.9$  Hz, 1 H), 4.53 (q,  $J = 7.1$  Hz, 1 H), 4.31 (q,  $J = 7.3$  Hz, 1 H), 4.15 (d,  $J = 14.8$  Hz, 1 H), 3.89 (d,  $J = 15.6$  Hz, 1 H), 3.75 (d,  $J = 15.4$  Hz, 1 H), 3.52 (d,  $J = 14.8$  Hz, 1 H), 2.53 – 2.44 (m, 4 H), 2.41 – 2.26 (m, 4 H), 1.48 (d,  $J = 7.4$  Hz, 3 H), 1.46 (d,  $J = 7.2$  Hz, 3 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 175.1, 174.9, 172.7, 171.7, 171.2, 171.2, 136.9, 136.7, 134.3, 133.9, 133.8, 133.5, 130.9, 129.9, 127.4, 127.3, 126.9, 126.4, 125.9, 125.8, 125.8, 125.3, 71.9, 71.6, 50.9, 49.6, 43.6, 42.7, 34.9, 33.5, 28.5, 26.8, 15.8, 15.5; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{17}\text{H}_{22}\text{N}_3\text{O}_4$  332.1610; found 332.1588.

**Allylic Alcohol 41.** To a solution of nitrobenzaldehyde derivative **40**<sup>19</sup> (1.8 g, 6.99 mmol, 1.0 equiv) in THF (25 mL) was added vinylmagnesium bromide (10 mL, 1.0 M in THF, 9.79 mmol, 1.4 equiv) at  $-78$  °C. After being stirred for 3.5 h, the mixture was quenched with 25 mL of 0.01 N HCl, diluted and extracted with EtOAc three times. The combined organic layers were washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 5% EtOAc in hexanes) to obtain allylic alcohol **41** (1.8 g, 90%) as a yellow oil:  $R_f = 0.79$  (silica gel, 40% EtOAc in hexanes);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 7.66 (d,  $J = 8.7$  Hz, 1 H), 7.53 (d,  $J = 2.7$  Hz, 1 H), 7.47 – 7.36 (m, 6 H), 7.25 (dd,  $J = 8.7, 2.7$  Hz, 1 H), 6.08 (ddd,  $J = 17.2, 10.5, 5.1$  Hz, 1 H), 5.75 – 5.70 (m, 1 H), 5.43 (dt,  $J = 17.2, 1.4$  Hz, 1 H), 5.27 (dt,  $J = 10.5, 1.4$  Hz, 1 H), 5.14 (s, 2 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 158.3, 148.9, 138.1, 135.7, 130.2, 129.8, 128.8, 128.5, 127.6, 120.7, 115.9, 110.3, 70.67, 69.7; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{16}\text{H}_{16}\text{NO}_4$  286.1079; found 286.1080.

**Silyl Ether 43.** A solution of allylic alcohol **41** (800 mg, 4.47 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (40 mL) was treated with imidazole (395 mg, 5.80 mmol, 1.3 equiv) and TBSCl (875 mg, 5.80 mmol, 1.3 equiv) at 0 °C. The resulting solution was stirred for 12 h at 25 °C and then diluted with CH<sub>2</sub>Cl<sub>2</sub>, washed with a saturated aqueous NH<sub>4</sub>Cl solution and brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The residue was purified by flash column chromatography (silica gel, 5% EtOAc in hexanes) to obtain silyl ether **43** (1.2 g, 95%) as a colourless oil: *R<sub>f</sub>* = 0.88 (silica gel, 40% EtOAc in hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 7.74 (d, *J* = 8.8 Hz, 1 H), 7.50 (d, *J* = 2.6 Hz, 1 H), 7.47 – 7.38 (m, 5 H), 7.25 (dd, *J* = 8.8, 2.7 Hz, 1 H), 5.98 (ddd, *J* = 17.0, 10.3, 5.0 Hz, 1 H), 5.84 (dt, *J* = 5.0, 1.4 Hz, 1 H), 5.35 (dt, *J* = 17.0, 1.6 Hz, 1 H), 5.13 (s, 2 H), 5.09 (dt, *J* = 10.3, 1.6 Hz, 1 H), 0.92 (s, 9 H), 0.13 (s, 3 H), 0.09 (s, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 157.8, 147.7, 139.7, 135.8, 131.6, 129.8, 128.8, 128.4, 127.6, 120.9, 113.9, 109.3, 70.6, 69.9, 25.8, 18.3, -3.5; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>22</sub>H<sub>30</sub>NO<sub>4</sub>Si 400.1944; found 400.1958.

**Benzyloxymethyl Acetal 45.** To a solution of allylic alcohol **41** (460 mg, 1.61 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was added DIPEA (1.12 mL, 6.45 mmol, 4.0 equiv) and BOMCl (1.2 mL, 6.45 mmol, 4.0 equiv) at 0 °C and the mixture was stirred at 25 °C for 15 h. After this time, the reaction was quenched by addition of a saturated aqueous Na<sub>2</sub>CO<sub>3</sub> solution and the crude mixture was stirred for 30 min. Then, the aqueous phase was extracted with CH<sub>2</sub>Cl<sub>2</sub> and the organic phase washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The resulting residue was purified by flash column chromatography (silica gel, 5% EtOAc in hexanes) to obtain benzyloxymethyl acetal **45** (528 mg, 88%) as a pale yellow oil: *R<sub>f</sub>* = 0.56 (silica gel, 20% EtOAc in hexanes); <sup>1</sup>H NMR (400

1  
2 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.65 (d,  $J$  = 8.8 Hz, 1 H), 7.49 (d,  $J$  = 2.7 Hz, 1 H), 7.43 – 7.28 (m, 10  
3  
4 H), 7.22 (dd,  $J$  = 8.8, 2.7 Hz, 1 H), 5.95 (ddd,  $J$  = 17.1, 10.3, 6.0 Hz, 1 H), 5.79 (d,  $J$  = 6.1 Hz,  
5  
6 1 H), 5.35 (dt,  $J$  = 17.2, 1.4 Hz, 1 H), 5.23 (dt,  $J$  = 10.4, 1.3 Hz, 1 H), 5.11 (s, 2 H), 4.91 –  
7  
8 4.77 (m, 2 H), 4.71 (d,  $J$  = 6.9 Hz, 1 H), 4.62 – 4.51 (m, 2 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$   
9  
10 (ppm) 158.2, 148.9, 137.6, 136.8, 135.8, 129.9, 128.8, 128.5, 128.4, 128.4, 127.9, 127.7,  
11  
12 127.5, 120.6, 117.0, 109.8, 92.6, 73.4, 70.7, 69.8; HRMS (H-ESI)  $m/z$ : [M + H]<sup>+</sup> calcd for  
13  
14 C<sub>24</sub>H<sub>24</sub>NO<sub>5</sub> 406.1655; found 406.1651.  
15  
16  
17  
18  
19  
20

21 **Aniline 46.** To a solution of benzyloxymethyl acetal **45** (480 mg, 1.18 mmol, 1.0 equiv) in  
22  
23 EtOH (10 mL) was added a solution of NH<sub>4</sub>Cl (317 mg, 5.92 mmol, 5.0 equiv) in water (7.5  
24  
25 mL) followed by Zn dust (1.2 g, 17.76 mmol, 15.0 equiv) in twelve portions of ~100 mg each  
26  
27 over 30 min at 25 °C. The mixture was stirred at this temperature for 12 h and then the  
28  
29 reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> and water, filtered and rinsed with CH<sub>2</sub>Cl<sub>2</sub>. The  
30  
31 filtrate was diluted with water and extracted with CH<sub>2</sub>Cl<sub>2</sub>, and the organic phase washed with  
32  
33 brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced  
34  
35 pressure. The resulting residue was purified by flash column chromatography (silica gel, 15%  
36  
37 EtOAc in hexanes) to obtain aniline **46** (351 mg, 79%) as a pale yellow oil:  $R_f$  = 0.35 (silica  
38  
39 gel, 30% EtOAc in hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.42 – 7.29 (m, 10 H),  
40  
41 7.01 (d,  $J$  = 8.4 Hz, 1 H), 6.35 (dd,  $J$  = 8.3, 2.5 Hz, 1 H), 6.29 (d,  $J$  = 2.5 Hz, 1 H), 6.12 (ddd,  
42  
43  $J$  = 17.2, 10.4, 5.7 Hz, 1 H), 5.33 (dt,  $J$  = 17.3, 1.6 Hz, 1 H), 5.25 (dt,  $J$  = 10.4, 1.6 Hz, 1 H),  
44  
45 5.16 (dt,  $J$  = 5.7, 1.5 Hz, 1 H), 5.02 (s, 2 H), 4.84 – 4.76 (m, 2 H), 4.67 – 4.62 (m, 2 H), 4.16  
46  
47 (bs, 2 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 159.8, 146.7, 137.8, 137.2, 136.5, 130.6,  
48  
49 128.5, 128.4, 127.9, 127.9, 127.7, 127.4, 116.3, 104.2, 102.9, 91.9, 78.1, 77.2, 69.9, 69.8;  
50  
51 HRMS (H-ESI)  $m/z$ : [M + H]<sup>+</sup> calcd for C<sub>24</sub>H<sub>26</sub>NO<sub>3</sub> 376.1913; found 376.1901.  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3  
4 **Diolefin 47.** A solution of the aniline **46** (50 mg, 0.13 mmol, 1.0 equiv) and acid **36** (30 mg,  
5 0.13 mmol, 1.0 equiv) in dry DMF (7 mL) was treated with HATU (50 mg, 0.13 mmol, 1.0  
6 equiv) and DIPEA (22  $\mu$ L, 0.13 mmol, 1.0 equiv) and the resulting solution was stirred at 25  
7  $^{\circ}$ C for 12 h. After this time, a saturated aqueous  $\text{NH}_4\text{Cl}$  solution was added and the organic  
8 layer was separated. The aqueous phase was extracted with EtOAc, and the combined organic  
9 layers were washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent  
10 evaporated under reduced pressure. The crude product was purified by flash column  
11 chromatography (silica gel, 30% EtOAc in hexanes) to obtain diolefin **47** (57 mg, 75%, 1:1  
12 mixture of diastereoisomers) as a white solid. Data assigned for the mixture of  
13 diastereoisomers:  $R_f = 0.81$  (silica gel, 80% EtOAc in hexanes); mp = 90-91  $^{\circ}$ C;  $^1\text{H}$  NMR  
14 (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.89 (d,  $J = 16.4$  Hz, 1 H), 7.98 (d,  $J = 2.5$  Hz, 1 H), 7.47 – 7.27  
15 (m, 11 H), 7.08 (dd,  $J = 8.5, 1.8$  Hz, 1 H), 6.70 (ddd,  $J = 8.5, 2.6, 1.2$  Hz, 1 H), 6.00 (ddd,  $J =$   
16 18.5, 9.3, 4.1 Hz, 1 H), 5.83 (dddd,  $J = 12.6, 8.8, 6.1, 4.7$  Hz, 1 H), 5.30 – 5.20 (m, 3 H), 5.12  
17 – 4.99 (m, 5 H), 4.80 (ddd,  $J = 15.5, 11.0, 6.7$  Hz, 2 H), 4.67 (dd,  $J = 14.2, 12.2$  Hz, 1 H), 4.59  
18 – 4.54 (m, 2 H), 2.44 – 2.37 (m, 2 H), 2.36 – 2.29 (m, 2 H), 1.45 – 1.38 (m, 3 H);  $^{13}\text{C}$  NMR  
19 (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 176.3, 171.9, 171.9, 170.5, 170.4, 159.4, 137.5, 137.4, 137.2,  
20 136.9, 136.8, 136.4, 136.4, 130.4, 128.6, 128.6, 128.5, 128.1, 128.1, 128.0, 127.6, 120.2,  
21 120.2, 116.8, 116.8, 115.8, 115.7, 115.6, 111.1, 111.0, 108.6, 108.5, 92.2, 92.0, 78.2, 78.2,  
22 70.2, 70.1, 49.8, 49.7, 38.6, 35.7, 33.0, 29.4, 28.7, 19.1, 18.9; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$   
23 calcd for  $\text{C}_{34}\text{H}_{40}\text{N}_3\text{O}_6$  586.2917; found 586.2921.  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53

54 **Macrocycle 48.** Diolefin **47** (15 mg, 0.03 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation  
55 catalyst (1.6 mg, 0.003 mmol, 0.10 equiv) and *p*-benzoquinone (1.0 mg, 0.003 mmol, 0.10  
56  
57  
58  
59  
60



equiv) were dissolved in degassed  $\text{CH}_2\text{Cl}_2$  (2 mL, 0.02 M) and the reaction mixture was heated at 40 °C for 15 h. After this time, the solvent was removed under reduced pressure and the resulting crude product was purified by flash column chromatography (silica gel, 20% EtOAc in hexanes  $\rightarrow$  60% EtOAc in hexanes) to obtain macrocycle **48** (12 mg, 84%, 1:1 mixture of diastereoisomers) as a white solid. Data assigned for the mixture of diastereoisomers:  $R_f = 0.21$  (silica gel, 80% EtOAc in hexanes); mp = 207-208 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.81 (bs, 1 H), 8.53 (bs, 1 H), 8.51 (d,  $J = 2.6$  Hz, 1 H), 8.35 (d,  $J = 2.6$  Hz, 1 H), 7.46 – 7.28 (m, 23 H), 7.04 (d,  $J = 8.4$  Hz, 1 H), 6.70 (dd,  $J = 8.5, 2.6$  Hz, 1 H), 6.61 (dd,  $J = 8.3, 2.6$  Hz, 1 H), 6.20 – 6.11 (m, 1 H), 6.08 – 5.98 (m, 2 H), 5.82 – 5.72 (m, 2 H), 5.60 (dd,  $J = 15.4, 5.7$  Hz, 1 H), 5.20 – 5.16 (m, 1 H), 5.15 – 5.10 (m, 1 H), 5.07 (s, 4 H), 4.91 – 4.74 (m, 4 H), 4.71 – 4.63 (m, 4 H), 4.60 (d,  $J = 7.5$  Hz, 2 H), 4.52 – 4.43 (m, 2 H), 2.51 (dd,  $J = 9.1, 4.1$  Hz, 4 H), 2.39 – 2.20 (m, 4 H), 1.54 (d,  $J = 7.2$  Hz, 3 H), 1.49 (d,  $J = 7.4$  Hz, 3 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 172.7, 172.7, 172.2, 170.4, 170.3, 169.5, 159.8, 159.2, 138.7, 138.1, 137.7, 136.9, 136.9, 132.4, 132.3, 132.0, 131.6, 128.7, 128.6, 128.5, 128.5, 128.5, 128.1, 127.9, 127.9, 127.9, 127.9, 127.8, 127.7, 127.6, 127.6, 127.5, 119.8, 118.3, 110.7, 109.9, 106.9, 106.7, 91.5, 90.9, 75.9, 74.5, 69.9, 69.9, 69.7, 69.5, 51.8, 51.5, 50.9, 36.9, 36.4, 30.2, 29.7, 29.6, 17.9, 16.5; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{32}\text{H}_{36}\text{N}_3\text{O}_6$  558.2604; found 558.2615.

**Acid 49.** A 5 mL polypropylene syringe fitted with polyethylene porous disk charged with 2-chlorotrityl chloride (CTC) resin (363 mg, L=1.3 mmol/g, 0.47 mmol, 1.0 equiv), was loaded with a solution of Fmoc-D-Ala-OH (**34**) (441 mg, 1.42 mmol, 3.0 equiv) and DIPEA (0.29 mL, 1.65 mmol, 3.5 equiv) in dry DMF (3 mL). The resulting suspension was shaken at 280 rpm for 30 h, then the solution was unloaded and the resin was washed by shaking with dry

1  
2 DMF (5 x 3 mL). The resulting yellow resin was used in the subsequent step. The  
3  
4 polypropylene syringe loaded with the yellow resin was treated with a solution of 20%  
5  
6 piperidine in DMF (3 x 3 mL x 10 min at 280 rpm). After the last run, the resin was washed  
7  
8 with dry DMF (5 x 3 mL) and loaded with a solution of Fmoc-Gly-OH (**35**) (281 mg, 0.94  
9  
10 mmol, 2.0 equiv), HOBt (128 mg, 0.94 mmol, 2.0 equiv) and DIC (0.18 mL, 1.18 mmol, 2.5  
11  
12 equiv) in dry DMF (3 mL). The resulting suspension was shaken at 280 rpm for 24 h, and then  
13  
14 the solution was unloaded and the resin washed with dry DMF (5 x 3 mL). The resulting  
15  
16 yellow resin was used in the next step. To a 5 mL polypropylene syringe fitted with  
17  
18 polyethylene porous disk and loaded with the Fmoc protected dipeptide resin was added a  
19  
20 solution of 20% piperidine in DMF (3 x 3 mL x 10 min at 280 rpm). After the last run, the  
21  
22 resin was washed with dry DMF (5 x 3 mL) and treated with a solution of the acid **14** (231  
23  
24 mg, 0.94 mmol, 2.0 equiv), HOBt (127 mg, 0.94 mmol, 2.0 equiv) and DIC (0.18 mL, 1.18  
25  
26 mmol, 2.5 equiv) in dry DMF (3 mL). The resulting suspension was shaken at 280 rpm for 24  
27  
28 h, and then the solution was unloaded and the resin washed with dry DMF (5 x 3 mL). The  
29  
30 resulting yellow resin was treated with a solution of CH<sub>2</sub>Cl<sub>2</sub>/AcOH/TFE (7:2:1, 3 mL) for 30  
31  
32 min. After that, the solution was collected and the resin washed with CH<sub>2</sub>Cl<sub>2</sub> (2 x 3 mL). All  
33  
34 the collected organic solvents were evaporated under reduced pressure and the resulting acid  
35  
36 **49** (90 mg, 84% overall yield from CTC resin) was obtained as a colorless solid which not  
37  
38 required further purification:  $R_f = 0.30$  (silica gel, 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub>);  $[\alpha]_D^{25} = -5.44$  ( $c$   
39  
40 0.08, CH<sub>2</sub>Cl<sub>2</sub>); mp = 92-93 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.20 (d,  $J = 5.7$  Hz, 1 H),  
41  
42 5.76 (ddd,  $J = 17.2, 10.4, 6.8$  Hz, 1 H), 5.20 (d,  $J = 17.1$  Hz, 1 H), 5.15 (d,  $J = 10.5$  Hz, 1 H),  
43  
44 4.61 – 4.47 (m, 1 H), 4.17 (d,  $J = 6.5$  Hz, 1 H), 4.08 (dd,  $J = 16.6, 5.7$  Hz, 1 H), 3.86 (dd,  $J =$   
45  
46 16.6, 4.8 Hz, 1 H), 2.41 (p,  $J = 6.9$  Hz, 1 H), 1.42 (d,  $J = 7.0$  Hz, 3 H), 1.11 (d,  $J = 7.1$  Hz, 3  
47  
48 H), 0.86 (s, 9 H), 0.03 (s, 3 H), 0.02 (s, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 175.9,  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2 168.8, 163.2, 138.8, 116.6, 76.0, 47.8, 42.9, 36.8, 25.8, 18.1, 18.0, 14.7, -4.3, -5.1; HRMS (H-  
3  
4 ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{17}H_{33}N_2O_5Si$  373.2159; found 373.2164.  
5  
6  
7  
8

9 **Diolefin 51.** To a solution of aniline **30** (64 mg, 0.43 mmol, 1.0 equiv) and acid **49** (160 mg,  
10 0.43 mmol, 1.3 equiv) in DMF (10 mL) was added HATU (163 mg, 0.43 mmol, 1.0 equiv) and  
11 DIPEA (0.10 mL, 0.43 mmol, 1.0 equiv) at 0 °C and the mixture was stirred for 12 h at 25 °C.  
12  
13 After this time, a saturated aqueous  $NH_4Cl$  solution was added and the organic layer was  
14 separated. The aqueous phase was extracted with EtOAc, and the combined organic layers  
15 were washed with brine, dried over anhydrous  $MgSO_4$ , filtered and the solvent evaporated  
16 under reduced pressure. The crude product was purified by flash column chromatography  
17 (silica gel, 25% EtOAc in hexanes  $\rightarrow$  100% EtOAc) to obtain diolefin **51** (60 mg, 36%, 1:1  
18 mixture of diastereoisomers) as a white foam. Data assigned for the mixture of  
19 diastereoisomers:  $R_f$  = 0.23 (silica gel, 100% EtOAc);  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm)  
20 7.28 – 7.19 (m, 2 H), 7.17 – 7.10 (m, 1 H), 7.05 (t,  $J$  = 7.3 Hz, 1 H), 6.14 – 5.96 (m, 1 H),  
21 5.78 – 5.61 (m, 1 H), 5.32 – 5.06 (m, 5 H), 4.43 – 4.25 (m, 1 H), 4.18 – 4.07 (m, 1 H), 3.99 (s,  
22 1 H), 3.69 – 3.55 (m, 1 H), 2.44 (s, 1 H), 2.35 – 2.22 (m, 1 H), 1.36 (d,  $J$  = 5.7 Hz, 3 H), 1.00  
23 – 0.94 (m, 3 H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 176.9, 176.5, 171.6, 171.2, 170.7,  
24 170.5, 138.1, 137.9, 137.8, 136.1, 131.4, 131.1, 128.6, 128.5, 128.2, 128.1, 124.8, 124.8,  
25 122.6, 122.3, 118.4, 118.0, 116.7, 116.2, 74.9, 74.3, 60.5, 50.5, 49.5, 46.8, 46.8, 30.7, 29.7,  
26 17.4, 17.3, 13.8, 13.5; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{20}H_{28}N_3O_5$  390.2029; found  
27 390.2037.  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51

52  
53 **Ketone 53.** To a solution of diolefin **32** (27 mg, 0.08 mmol, 1.0 equiv) in  $CH_2Cl_2$  (15 mL)  
54 was added  $MnO_2$  (131 mg, 1.50 mmol, 20.0 equiv) and the dark solution was stirred at 25 °C  
55  
56  
57  
58  
59  
60

1  
2 for 48 h. The mixture was filtered through a pad of celite and rinsed with CH<sub>2</sub>Cl<sub>2</sub>. The solvent  
3  
4 was evaporated under reduced pressure to obtain ketone **53** (20 mg, 75%) as a white solid  
5  
6 which did not require purification:  $R_f = 0.43$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -6.98$  ( $c$   
7 0.09, CH<sub>2</sub>Cl<sub>2</sub>); mp = 111-112 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 11.91 (bs, 1 H), 8.72  
8  
9 (dd,  $J = 8.5, 1.0$  Hz, 1 H), 7.90 (dd,  $J = 8.0, 1.6$  Hz, 1 H), 7.59 (ddd,  $J = 8.6, 7.5, 1.4$  Hz, 1 H),  
10  
11 7.26 – 7.15 (m, 2 H), 7.04 (d,  $J = 7.2$  Hz, 1 H), 6.73 (t,  $J = 4.9$  Hz, 1 H), 6.42 (dd,  $J = 16.9,$   
12  
13 1.6 Hz, 1 H), 5.98 (dd,  $J = 10.6, 1.6$  Hz, 1 H), 5.91 – 5.77 (m, 1 H), 5.08 (ddd,  $J = 17.1, 3.2,$   
14  
15 1.6 Hz, 1 H), 5.01 (ddd,  $J = 2.9, 2.4, 1.2$  Hz, 1 H), 4.67 (p,  $J = 7.2$  Hz, 1 H), 4.32 (dd,  $J =$   
16  
17 16.8, 5.8 Hz, 1 H), 4.06 (dd,  $J = 16.8, 4.8$  Hz, 1 H), 2.45 – 2.35 (m, 4 H), 1.54 (d,  $J = 7.2$  Hz,  
18  
19 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 194.2, 172.9, 171.3, 169.2, 140.7, 136.9, 135.3,  
20  
21 133.0, 131.3, 131.0, 122.9, 122.5, 121.0, 115.8, 50.4, 43.3, 35.5, 29.4, 18.2; HRMS (H-ESI)  
22  
23  $m/z$ :  $[M + H]^+$  calcd for C<sub>19</sub>H<sub>24</sub>N<sub>3</sub>O<sub>4</sub> 358.1767; found 358.1789.  
24  
25  
26  
27  
28  
29  
30  
31  
32

33 **Dipeptide 61.** To a solution of aniline **60**<sup>23</sup> (200 mg, 1.68 mmol, 1.0 equiv) and Boc-Gly-D-  
34  
35 Ala-OH (**21**) (413 mg, 1.68 mmol, 1.0 equiv) in DMF (10 mL) was added HATU (960 mg,  
36  
37 2.52 mmol, 1.5 equiv) and DIPEA (0.30 mL, 1.68 mmol, 1.0 equiv) and the mixture was  
38  
39 stirred for 12 h at 25 °C. After this time, the reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> and  
40  
41 washed sequentially with 1 N HCl and saturated aqueous NaHCO<sub>3</sub> solution. The organic layer  
42  
43 was separated, washed with brine and dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent  
44  
45 evaporated under reduced pressure. The crude product was purified by flash column  
46  
47 chromatography (silica gel, 30% EtOAc in hexanes) to obtain dipeptide **61** (455 mg, 78%) as  
48  
49 a white solid:  $R_f = 0.52$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -6.36$  ( $c$  0.06, CH<sub>2</sub>Cl<sub>2</sub>); mp = 84-  
50  
51 85 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 8.13 (bs, 1 H), 7.78 (d,  $J = 8.1$  Hz, 1 H), 7.45 (d,  
52  
53  $J = 7.8$  Hz, 1 H), 7.15 (t,  $J = 7.5$  Hz, 1 H), 6.85 – 6.79 (m, 1 H), 6.76 (d,  $J = 9.1$  Hz, 1 H), 5.68  
54  
55  
56  
57  
58  
59  
60

(dd,  $J = 17.4, 1.2$  Hz, 1 H), 5.43 (dd,  $J = 11.0, 1.2$  Hz, 1 H), 5.12 (bs, 1 H), 4.66 (q,  $J = 7.1$  Hz, 1 H), 3.92 – 3.77 (m, 2 H), 1.50 (d,  $J = 7.0$  Hz, 3 H), 1.44 (s, 9 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 170.4, 170.0, 156.1, 136.9, 136.2, 131.1, 130.1, 127.2, 125.7, 123.9, 116.5, 49.5, 38.6, 36.1, 28.2, 17.7; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{18}\text{H}_{26}\text{N}_3\text{O}_4$  348.1923; found 348.1921.

**Diolefin 63.** Dipeptide **61** (152 mg, 0.44 mmol, 1.0 equiv) in  $\text{CH}_2\text{Cl}_2$  (25 mL) was treated with TFA (2.2 mL) in exactly the same manner as previously described for synthesis of **24**. A solution of the ammonium salt and acid **62**<sup>10,24</sup> (50 mg, 0.44 mmol, 1.0 equiv), in DMF (15 mL) was treated with HATU (167 mg, 0.44 mmol, 1.0 equiv) and DIPEA (0.23 mL, 1.31 mmol, 3.0 equiv) and the resulting reaction mixture was stirred at 25 °C for 12 h. After this time, a saturated aqueous  $\text{NH}_4\text{Cl}$  solution was added and the organic layer was separated. The aqueous phase was extracted with EtOAc, and the combined organic layers were washed with brine and dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica gel,  $\text{CH}_2\text{Cl}_2 \rightarrow 2\%$  MeOH in  $\text{CH}_2\text{Cl}_2$ ) to obtain diolefin **63** (150 mg, 92% over two steps) as a white solid:  $R_f = 0.60$  (silica gel, 10% MeOH in  $\text{CH}_2\text{Cl}_2$ );  $[\alpha]_D^{25} = -5.43$  ( $c$  0.07,  $\text{CH}_2\text{Cl}_2$ ); mp = 104-105 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.34 (bs, 1 H), 7.67 – 7.57 (m, 1 H), 7.46 (d,  $J = 7.6$  Hz, 1 H), 7.26 – 7.13 (m, 2 H), 6.82 – 6.71 (m, 1 H), 6.68 (t,  $J = 7.6$  Hz, 1 H), 6.46 (bs, 1 H), 5.81 (td,  $J = 16.9, 7.7$  Hz, 1 H), 5.67 (dd,  $J = 17.4, 1.1$  Hz, 1 H), 5.42 – 5.34 (m, 2 H), 5.18 – 5.12 (m, 1 H), 4.67 – 4.51 (m, 1 H), 4.25 (dd,  $J = 16.7, 7.3$  Hz, 1 H), 3.79 – 3.62 (m, 2 H), 3.14 (qd,  $J = 7.3, 4.4$  Hz, 2 H), 2.17 (dd,  $J = 13.7, 8.2$  Hz, 1 H), 1.42 (d,  $J = 6.7$  Hz, 3 H), 1.39 (d,  $J = 6.6$  Hz, 3 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 176.1, 171.2, 170.2, 133.5, 131.8,

1  
2 128.3, 126.4, 126.1, 126.0, 124.5, 124.1, 119.0, 117.5, 73.3, 55.6, 43.5, 39.2, 18.6, 17.2, 12.4.

3  
4 HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{20}H_{28}N_3O_4$  374.2080; found 374.2072.

5  
6  
7  
8  
9 **Styryl 65.** To a solution of iodonitrobenzene **18** (2.7 g, 7.60 mmol, 1.0 equiv) and  $Pd[PPh_3]_4$   
10 (1.3 g, 1.14 mmol, 0.15 equiv) in DMF (10 mL) was added dropwise tri-*n*-butyl(vinyl)tin (2.7  
11 mL, 9.12 mmol, 1.2 equiv). The solution was then heated at 60 °C for 48 h. After this time,  
12  
13 the mixture was diluted with  $Et_2O$  and water. The organic layer was separated and washed  
14  
15 with water four times, dried over anhydrous  $MgSO_4$ , filtered and the solvent evaporated under  
16  
17 reduced pressure. The crude product was purified by flash column chromatography (silica gel,  
18  
19 2% EtOAc in Hexanes) to obtain styryl **65** (1.4 g, 74%) as a yellow oil:  $R_f = 0.51$  (silica gel,  
20  
21 20% EtOAc in hexanes);  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 7.54 (dd,  $J = 10.3, 5.7$  Hz, 2  
22  
23 H), 7.46 – 7.34 (m, 5 H), 7.20 (ddd,  $J = 8.7, 2.7, 0.6$  Hz, 1 H), 7.11 (ddd,  $J = 11.5, 10.9, 5.5$   
24  
25 Hz, 1 H), 5.65 (dd,  $J = 17.3, 1.0$  Hz, 1 H), 5.40 (dd,  $J = 11.0, 1.0$  Hz, 1 H), 5.13 (s, 2 H);  $^{13}C$   
26  
27 NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 158.4, 135.7, 132.0, 129.5, 128.8, 128.7, 128.5, 127.6,  
28  
29 126.1, 120.8, 117.4, 109.8, 70.7; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{15}H_{14}NO_3$   
30  
31 256.0974; found 256.0981.  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41

42 **Aniline 66.** To a solution of styryl **65** (550 mg, 2.16 mmol, 1.0 equiv) in EtOH (20 mL) was  
43  
44 added a solution of  $NH_4Cl$  (576 mg, 10.77 mmol, 5.0 equiv) in water (14 mL) followed by Zn  
45  
46 dust (2.1 g, 32.32 mmol, 15.0 equiv) in ten portions of ~200 mg each over 30 min at 25 °C.  
47  
48 The mixture was stirred at this temperature for 12 h and then the reaction mixture was diluted  
49  
50 with  $CH_2Cl_2$  and water, filtered and rinsed with  $CH_2Cl_2$ . The filtrate was diluted with water  
51  
52 and extracted with  $CH_2Cl_2$ , and the organic phase washed with brine, dried over anhydrous  
53  
54  $MgSO_4$ , filtered and the solvent evaporated under reduced pressure. The resulting residue was  
55  
56  
57  
58  
59  
60

1 purified by flash column chromatography (silica gel, 10% EtOAc in hexanes) to obtain aniline  
2  
3  
4 **66** (413 mg, 85%) as a yellow oil:  $R_f = 0.47$  (silica gel, 30% EtOAc in hexanes);  $^1\text{H}$  NMR  
5  
6 (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 7.45 – 7.31 (m, 5 H), 7.27 – 7.21 (m, 1 H), 6.71 (dd,  $J = 17.4$ ,  
7  
8 11.1 Hz, 1 H), 6.44 (dd,  $J = 8.5, 2.4$  Hz, 1 H), 6.32 (d,  $J = 2.5$  Hz, 1 H), 5.54 (dd,  $J = 17.4, 1.5$   
9 Hz, 1 H), 5.22 (dd,  $J = 11.0, 1.5$  Hz, 1 H), 5.04 (s, 2 H), 3.78 (bs, 2 H);  $^{13}\text{C}$  NMR (100 MHz,  
10  
11  $\text{CDCl}_3$ )  $\delta$  (ppm) 159.6, 144.9, 137.2, 132.2, 128.6, 128.5, 127.9, 127.4, 117.5, 113.8, 105.7,  
12  
13 102.3, 69.9; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{15}\text{H}_{16}\text{NO}$  226.1232; found 226.1229.  
14  
15  
16  
17  
18  
19  
20

21 **Dipeptide 67.** To a solution of aniline **66** (340 mg, 1.51 mmol, 1.0 equiv) and Boc-Gly-D-  
22  
23 Ala-OH (**21**) (483 mg, 1.962 mmol, 1.0 equiv) in DMF (5 mL) was added HATU (861 mg,  
24  
25 2.26 mmol, 1.5 equiv) and DIPEA (0.5 mL, 3.02 mmol, 1.0 equiv) and the mixture was stirred  
26  
27 for 12 h at 25 °C. After this time, a saturated aqueous  $\text{NH}_4\text{Cl}$  solution was added and the  
28  
29 organic layer was separated. The aqueous phase was extracted with EtOAc, and the combined  
30  
31 organic layers were washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent  
32  
33 evaporated under reduced pressure. The crude product was purified by flash column  
34  
35 chromatography (silica gel, 15% EtOAc in hexanes  $\rightarrow$  35% EtOAc in hexanes) to obtain  
36  
37 dipeptide **67** (554 mg, 81%) as a yellow foam:  $R_f = 0.64$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -$   
38  
39 3.96 ( $c$  0.15,  $\text{CH}_2\text{Cl}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 8.30 (d,  $J = 13.1$  Hz, 1 H), 7.60  
40  
41 (bs, 1 H), 7.46 – 7.30 (m, 7 H), 6.78 (d,  $J = 2.7$  Hz, 1 H), 6.76 – 6.69 (m, 1 H), 5.56 (d,  $J =$   
42  
43 17.3 Hz, 1 H), 5.31 (d,  $J = 11.6$  Hz, 1 H), 5.18 (bs, 1 H), 5.04 (s, 2 H), 4.68 (p,  $J = 7.1$  Hz, 1  
44  
45 H), 3.88 – 3.79 (m, 2 H), 1.47 (d,  $J = 6.3$  Hz, 3 H), 1.43 (s, 9 H);  $^{13}\text{C}$  NMR (100 MHz,  
46  
47  $\text{CDCl}_3$ )  $\delta$  (ppm) 170.2, 170.1, 158.8, 156.2, 136.8, 135.1, 131.4, 128.5, 127.9, 127.6, 123.2,  
48  
49 120.4, 116.0, 112.8, 109.1, 80.7, 70.1, 49.6, 44.4, 28.3, 17.5; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$   
50  
51 calcd for  $\text{C}_{25}\text{H}_{32}\text{N}_3\text{O}_5$  454.2342; found 454.2340.  
52  
53  
54  
55  
56  
57  
58  
59  
60

**Diolefin 69.** Dipeptide **67** (70 mg, 0.15 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was treated with TFA (2.2 mL) in exactly the same manner as previously described for synthesis of **24**. A solution of the resulting ammonium salt and hexenoic acid **68** (18 μL, 0.15 mmol, 1.0 equiv), in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was treated with BOP (66 mg, 0.15 mmol, 1.0 equiv) and DIPEA (77 μL, 0.45 mmol, 3.0 equiv) and the resulting solution was stirred at 25 °C for 12 h. After this time, a saturated aqueous NH<sub>4</sub>Cl solution was added and the organic layer was separated. The aqueous phase was extracted with EtOAc, and the combined organic layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The resulting residue was purified by flash column chromatography (silica gel, 30% EtOAc in hexanes) to obtain diolefin **69** (51 mg, 74% over two steps) as a white solid:  $R_f = 0.45$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -9.02$  ( $c$  0.11, CH<sub>2</sub>Cl<sub>2</sub>); mp = 82-83 °C ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 8.24 (bs, 1 H), 7.58 (bs, 1 H), 7.44 – 7.30 (m, 6 H), 7.06 (s, 1 H), 6.80 – 6.75 (m, 1 H), 6.72 (dd,  $J = 16.1, 9.7$  Hz, 1 H), 6.34 (s, 1 H), 5.74 (ddt,  $J = 17.0, 10.0, 6.8$  Hz, 1 H), 5.56 (d,  $J = 17.3$  Hz, 1 H), 5.29 (d,  $J = 11.0$  Hz, 1 H), 5.05 – 5.00 (m, 3 H), 5.00 – 4.93 (m, 1 H), 4.71 – 4.61 (m, 1 H), 4.06 – 3.93 (m, 2 H), 2.23 (td,  $J = 7.8, 1.7$  Hz, 2 H), 2.10 – 2.02 (m, 2 H), 1.77 – 1.68 (m, 2 H), 1.48 (d,  $J = 7.0$  Hz, 3 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 173.8, 170.1, 169.4, 158.9, 137.7, 136.7, 135.1, 131.4, 128.6, 128.0, 127.6, 127.6, 123.0, 116.1, 115.5, 112.7, 109.0, 70.1, 49.8, 43.3, 35.4, 33.1, 24.5, 17.6; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for C<sub>26</sub>H<sub>32</sub>N<sub>3</sub>O<sub>4</sub> 450.2393; found 450.2391.

**Macrocycle 70.** Diolefin **69** (19 mg, 0.08 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation catalyst (3 mg, 0.004 mmol, 0.10 equiv) and *p*-benzoquinone (1.0 mg, 0.004 mmol, 0.10 equiv) were dissolved in degassed CH<sub>2</sub>Cl<sub>2</sub> (2 mL, 0.02 M) and the reaction mixture was



1  
2 heated at 40 °C for 12 h. After this time, the solvent was removed under reduced pressure and  
3  
4 the resulting crude product was purified by flash column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>  
5  
6 → 3% MeOH in CH<sub>2</sub>Cl<sub>2</sub>) to obtain macrocycle **70** (6.1 mg, 43%) as a white solid:  $R_f = 0.76$   
7  
8 (silica gel, 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub>);  $[\alpha]^{25}_D = -5.32$  (*c* 0.09, MeOH); mp = 196-197 ; <sup>1</sup>H NMR  
9  
10 (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  (ppm) 8.54 (bs, 1 H), 8.40 (t, *J* = 5.9 Hz, 1 H), 7.46 – 7.35 (m, 5 H),  
11  
12 7.32 (d, *J* = 7.0 Hz, 1 H), 7.29 (d, *J* = 2.5 Hz, 1 H), 7.06 (d, *J* = 8.5 Hz, 1 H), 6.77 (dd, *J* = 8.4,  
13  
14 2.6 Hz, 1 H), 5.42 (dt, *J* = 15.8, 5.2 Hz, 1 H), 5.04 (s, 2 H), 4.40 – 4.31 (m, 1 H), 3.85 (dd, *J* =  
15  
16 14.8, 6.3 Hz, 1 H), 3.45 (dd, *J* = 14.8, 5.6 Hz, 1 H), 3.20 (dd, *J* = 15.5, 5.1 Hz, 1 H), 3.03 (dd,  
17  
18 *J* = 15.5, 5.2 Hz, 1 H), 2.34 – 2.04 (m, 4 H), 1.30 (d, *J* = 7.1 Hz, 3 H); <sup>13</sup>C NMR (100 MHz,  
19  
20 DMSO-*d*<sub>6</sub>)  $\delta$  (ppm) 173.1, 171.4, 171.1, 157.4, 137.6, 137.3, 131.6, 130.7, 128.9, 128.2,  
21  
22 128.1, 128.09, 128.0, 125.8, 111.8, 69.7, 49.7, 43.4, 34.7, 33.7, 27.9, 16.8; HRMS (H-ESI)  
23  
24 *m/z*: [M + H]<sup>+</sup> calcd for C<sub>24</sub>H<sub>28</sub>N<sub>3</sub>O<sub>4</sub> 422.2079; found 422.2073.  
25  
26  
27  
28  
29  
30  
31  
32

33 **Diolefin 59.** Dipeptide **67** (420 mg, 0.93 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was treated  
34  
35 with TFA (4.6 mL) in exactly the same manner as previously described for synthesis of **24**. A  
36  
37 solution of the corresponding ammonium salt and acid **62** (174 mg, 1.20 mmol, 1.3 equiv), in  
38  
39 CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was treated with HATU (528 mg, 1.39 mmol, 1.0 equiv) and DIPEA (0.5  
40  
41 mL, 2.78 mmol, 3.0 equiv) and the resulting solution was stirred at 25 °C for 12 h. After this  
42  
43 time, a saturated aqueous NH<sub>4</sub>Cl solution was added and the organic layer was separated. The  
44  
45 aqueous phase was extracted with EtOAc, and the combined organic layers were washed with  
46  
47 brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced  
48  
49 pressure. The resulting residue was purified by flash column chromatography (silica gel, 50%  
50  
51 EtOAc in hexanes → 100% EtOAc) to obtain diolefin **59** (67 mg, 86% over two steps) as a  
52  
53 white solid:  $R_f = 0.46$  (silica gel, 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub>);  $[\alpha]^{25}_D = -3.98$  (*c* 0.07, MeOH);  
54  
55  
56  
57  
58  
59  
60

1  
2 mp = 88-89 °C; <sup>1</sup>H NMR (400 MHz, MeOD) δ (ppm) 7.53 (d, *J* = 8.7 Hz, 1 H), 7.44 – 7.27  
3  
4 (m, 6 H), 7.04 (d, *J* = 2.6 Hz, 1 H), 6.89 (dd, *J* = 8.7, 2.6 Hz, 1 H), 6.80 (dd, *J* = 17.5, 11.1 Hz,  
5  
6 1 H), 5.98 – 5.82 (m, 1 H), 5.62 (dd, *J* = 17.4, 1.2 Hz, 1 H), 5.20 (dd, *J* = 11.0, 1.3 Hz, 1 H),  
7  
8 5.11 (dd, *J* = 18.9, 1.7 Hz, 1 H), 5.07 (s, 2 H), 5.07 – 5.03 (m, 1 H), 4.55 – 4.47 (m, 1 H), 4.02  
9  
10 (d, *J* = 16.9 Hz, 1 H), 3.79 (d, *J* = 16.9 Hz, 1 H), 3.71 (td, *J* = 7.3, 4.0 Hz, 1 H), 2.48 – 2.38  
11  
12 (m, 2 H), 2.27 – 2.18 (m, 1 H), 1.46 (d, *J* = 7.2 Hz, 3 H), 1.11 (d, *J* = 6.9 Hz, 3 H); <sup>13</sup>C NMR  
13  
14 (100 MHz, MeOD) δ (ppm) 177.3, 172.7, 170.5, 163.5, 158.7, 137.1, 134.9, 134.2, 131.6,  
15  
16 128.1, 127.6, 127.2, 126.4, 116.5, 113.5, 112.9, 112.1, 73.1, 69.7, 42.3, 38.6, 35.6, 30.3, 16.5,  
17  
18 12.9; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>27</sub>H<sub>34</sub>N<sub>3</sub>O<sub>5</sub> 480.2499; found 480.2501.  
19  
20  
21  
22  
23  
24  
25

26 **Diolefin 71.** To a solution of diolefin **59** (35 mg, 0.07 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was  
27  
28 added 2,6-lutidine (20 μL, 0.15 mmol, 2.0 equiv) at 0 °C and the mixture was stirred 10 min at  
29  
30 this temperature. After this time TBSOTf (0.03 mL, 0.15 mmol, 2.0 equiv) was added at 0 °C  
31  
32 and the mixture was stirred for 12 h at 25 °C. Then, the reaction was quenched by addition of  
33  
34 H<sub>2</sub>O. After decantation of the organic layer, the aqueous phase was extracted with EtOAc, and  
35  
36 the combined organic layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered  
37  
38 and the solvent evaporated under reduced pressure. The crude product was purified by flash  
39  
40 column chromatography (silica gel, 20% EtOAc in hexanes → 100% EtOAc) to obtain  
41  
42 diolefin **71** (31 mg, 72%) as a white solid: *R*<sub>f</sub> = 0.80 (silica gel, 100% EtOAc); [α]<sub>D</sub><sup>25</sup> = – 6.98  
43  
44 (*c* 0.06, MeOH); mp = 96-97 °C; <sup>1</sup>H NMR (400 MHz, MeOD) δ (ppm) 7.52 (d, *J* = 8.7 Hz, 1  
45  
46 H), 7.44 – 7.40 (m, 2 H), 7.39 – 7.33 (m, 2 H), 7.33 – 7.27 (m, 1 H), 7.06 (d, *J* = 2.6 Hz, 1 H),  
47  
48 6.89 (dd, *J* = 8.7, 2.7 Hz, 1 H), 6.79 (dd, *J* = 17.5, 11.1 Hz, 1 H), 5.89 (dddd, *J* = 16.5, 10.4,  
49  
50 8.2, 6.0 Hz, 1 H), 5.62 (dd, *J* = 17.5, 1.3 Hz, 1 H), 5.19 (dd, *J* = 11.0, 1.3 Hz, 1 H), 5.11 –  
51  
52 5.07 (m, 1 H), 5.07 (s, 2 H), 5.06 – 5.03 (m, 1 H), 4.56 – 4.49 (m, 1 H), 4.07 (d, *J* = 16.5 Hz, 1  
53  
54  
55  
56  
57  
58  
59  
60

1  
2 H), 3.98 – 3.89 (m, 1 H), 3.70 (d,  $J = 16.6$  Hz, 1 H), 2.58 – 2.46 (m, 1 H), 2.40 – 2.22 (m, 2  
3  
4 H), 1.47 (d,  $J = 7.1$  Hz, 3 H), 1.08 (d,  $J = 7.0$  Hz, 3 H), 0.87 (s, 9 H), 0.08 (s, 3 H), 0.03 (s, 3  
5  
6 H);  $^{13}\text{C}$  NMR (100 MHz, MeOD)  $\delta$  (ppm) 176.9, 172.6, 169.9, 158.7, 137.1, 134.9, 133.7,  
7  
8 131.6, 128.1, 127.5, 127.2, 126.4, 125.9, 116.6, 113.4, 112.9, 112.0, 73.3, 69.7, 49.4, 45.9,  
9  
10 42.0, 38.1, 24.9, 17.5, 16.8, 12.7, -5.6, -6.1; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  
11  
12  $\text{C}_{33}\text{H}_{48}\text{N}_3\text{O}_5\text{Si}$  594.3363; found 594.3367.  
13  
14  
15  
16  
17

18 **Macrocycle 72.** Diolefin **71** (20 mg, 0.03 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation  
19  
20 catalyst (2 mg, 0.003 mmol, 0.10 equiv) and *p*-benzoquinone (1.0 mg, 0.003 mmol, 0.10  
21  
22 equiv) were dissolved in degassed  $\text{CH}_2\text{Cl}_2$  (2 mL, 0.02 M) and the reaction mixture was  
23  
24 heated at 40 °C for 12 h. After this time, the solvent was removed under reduced pressure and  
25  
26 the resulting crude product was purified by flash column chromatography (silica gel, 20%  
27  
28 EtOAc in hexanes  $\rightarrow$  60% EtOAc in hexanes) to obtain macrocycle **72** (9.0 mg, 47%) as a  
29  
30 white solid:  $R_f = 0.51$  (silica gel, 70% EtOAc in hexanes);  $[\alpha]_D^{25} = -9.21$  ( $c$  0.08, MeOH);  
31  
32 mp = 205-206 °C;  $^1\text{H}$  NMR (400 MHz, MeOD)  $\delta$  (ppm) 7.44 – 7.27 (m, 6 H), 7.08 (d,  $J = 2.6$   
33  
34 Hz, 1 H), 6.85 (dd,  $J = 8.7, 2.8$  Hz, 1 H), 6.45 (d,  $J = 15.8$  Hz, 1 H), 5.98 (ddd,  $J = 15.4, 7.5,$   
35  
36 3.1 Hz, 1 H), 5.07 (s, 2 H), 4.25 (q,  $J = 7.2$  Hz, 1 H), 4.13 (td,  $J = 6.6, 3.7$  Hz, 1 H), 3.96 (d,  $J$   
37  
38 = 14.6 Hz, 1 H), 3.74 (d,  $J = 14.6$  Hz, 1 H), 2.59 – 2.49 (m, 3 H), 1.54 (d,  $J = 7.2$  Hz, 3 H),  
39  
40 1.15 (d,  $J = 7.2$  Hz, 3 H), 0.95 (s, 9 H), 0.15 (overlap two singlets, 6 H);  $^{13}\text{C}$  NMR (100 MHz,  
41  
42 MeOD)  $\delta$  (ppm) 176.8, 171.6, 171.3, 158.2, 137.2, 134.5, 128.6, 128.1, 127.5, 127.2, 126.3,  
43  
44 126.2, 125.4, 113.3, 112.3, 73.4, 69.7, 50.9, 46.3, 43.3, 38.9, 24.9, 17.5, 14.6, 12.0, -5.7, -6.1;  
45  
46  
47  
48  
49  
50  
51 HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{Na}]^+$  calcd for  $\text{C}_{31}\text{H}_{43}\text{N}_3\text{O}_5\text{SiNa}$  588.2870; found 588.2868.  
52  
53  
54  
55  
56  
57  
58  
59  
60

**Diolefin 73.** To a solution of diolefin **59** (53 mg, 0.11 mmol, 1.0 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added 2,6-lutidine (30 μL, 0.28 mmol, 2.5 equiv) at 0 °C and the mixture was stirred 10 min at this temperature. After this time TESOTf (0.06 mL, 0.30 mmol, 2.5 equiv) was added at 0 °C and the mixture was stirred for 12 h at 25 °C. Then, the reaction was quenched by addition of H<sub>2</sub>O. After decantation of the organic layer, the aqueous phase was extracted with EtOAc, and the combined organic layers were washed with brine, dried over anhydrous MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 15% EtOAc in hexanes → 60% EtOAc) to obtain diolefin **73** (45 mg, 68%) as a white solid:  $R_f = 0.80$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -9.41$  ( $c$  0.08, MeOH); mp = 92-93 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.53 (d,  $J = 8.7$  Hz, 1 H), 7.45 – 7.40 (m, 2 H), 7.39 – 7.33 (m, 2 H), 7.31 (dt,  $J = 5.3, 2.1$  Hz, 1 H), 7.06 (d,  $J = 2.6$  Hz, 1 H), 6.89 (dd,  $J = 8.5, 2.4$  Hz, 1 H), 6.80 (dd,  $J = 17.5, 11.0$  Hz, 1 H), 5.89 (dddd,  $J = 16.5, 10.3, 8.1, 6.1$  Hz, 1 H), 5.62 (dd,  $J = 17.5, 1.3$  Hz, 1 H), 5.20 (dd,  $J = 11.0, 1.3$  Hz, 1 H), 5.13 – 5.03 (m, 4 H), 4.53 (q,  $J = 7.0$  Hz, 1 H), 4.06 (d,  $J = 16.6$  Hz, 1 H), 3.97 – 3.91 (m, 1 H), 3.74 (d,  $J = 16.5$  Hz, 1 H), 2.54 – 2.44 (m, 1 H), 2.40 – 2.22 (m, 2 H), 1.47 (d,  $J = 7.1$  Hz, 3 H), 1.09 (d,  $J = 7.0$  Hz, 3 H), 0.95 (t,  $J = 7.9$  Hz, 9 H), 0.61 (q,  $J = 7.6$  Hz, 6 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 176.9, 172.7, 169.9, 158.7, 137.1, 134.9, 133.9, 131.6, 128.1, 127.5, 127.2, 126.4, 125.9, 116.6, 113.4, 112.9, 112.0, 73.4, 69.7, 49.4, 46.1, 41.9, 38.5, 16.7, 12.9, 5.9, 4.5; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for C<sub>33</sub>H<sub>48</sub>N<sub>3</sub>O<sub>5</sub>Si 594.3363; found 594.3359.

**Macrocycle 74.** Diolefin **73** (22 mg, 0.04 mmol, 1.0 equiv), Hoveyda-Grubbs 2<sup>nd</sup> generation catalyst (2 mg, 0.004 mmol, 0.10 equiv) and *p*-benzoquinone (1.0 mg, 0.004 mmol, 0.10 equiv) were dissolved in degassed CH<sub>2</sub>Cl<sub>2</sub> (2 mL, 0.02 M) and the reaction mixture was

1  
2 heated at 40 °C for 12 h. After this time, the solvent was removed under reduced pressure and  
3  
4 the resulting crude product was purified by flash column chromatography (silica gel, 30%  
5  
6 EtOAc in hexanes → 60% EtOAc in hexanes) to obtain macrocycle **74** (11 mg, 53%) as a  
7  
8 white solid:  $R_f = 0.28$  (silica gel, 70% EtOAc in hexanes);  $[\alpha]_D^{25} = -11.08$  ( $c$  0.09, MeOH);  
9  
10 mp = 198-199 °C;  $^1\text{H NMR}$  (400 MHz, MeOD)  $\delta$  (ppm) 7.44 – 7.40 (m, 2 H), 7.39 – 7.33 (m,  
11  
12 3 H), 7.33 – 7.27 (m, 2 H), 7.08 (d,  $J = 2.6$  Hz, 1 H), 6.85 (dd,  $J = 8.6, 2.5$  Hz, 1 H), 6.46 (d,  $J$   
13  
14 = 15.7 Hz, 1 H), 6.02 – 5.93 (m, 1 H), 5.07 (s, 2 H), 4.24 (q,  $J = 7.1$  Hz, 1 H), 4.14 (dd,  $J =$   
15  
16 10.2, 6.6 Hz, 1 H), 3.95 (d,  $J = 14.6$  Hz, 1 H), 3.74 (d,  $J = 14.4$  Hz, 1 H), 2.65 – 2.46 (m, 3 H),  
17  
18 1.54 (d,  $J = 7.2$  Hz, 3 H), 1.17 (d,  $J = 7.2$  Hz, 3 H), 1.02 (t,  $J = 7.9$  Hz, 9 H), 0.70 (q,  $J = 7.5$   
19  
20 Hz, 6 H);  $^{13}\text{C NMR}$  (100 MHz, MeOD)  $\delta$  (ppm) 176.8, 171.6, 171.3, 158.1, 137.2, 134.5,  
21  
22 128.7, 128.1, 127.5, 127.2, 126.3, 126.0, 125.4, 113.3, 112.3, 73.2, 69.9, 50.9, 46.3, 39.1,  
23  
24 29.3, 14.6, 12.2, 5.9, 4.4; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{31}\text{H}_{44}\text{N}_3\text{O}_5\text{Si}$  566.3050;  
25  
26 found 566.3067.  
27  
28  
29  
30  
31  
32  
33  
34

35 **Macrocycle 58.** To a solution of macrocycle **74** (5.0 mg, 0.01 mmol, 1.0 equiv) in THF (3  
36  
37 mL) was added at 0 °C TBAF (18  $\mu\text{L}$ , 1.0 M in THF, 0.018 mmol, 2.0 equiv). After 95 min  
38  
39 the solvent was removed under reduced pressure and the crude mixture was purified by  
40  
41 preparative TLC (silica gel 60 F<sub>254</sub>, 1 mm, 100% EtOAc) to obtain macrocycle **58** (3.4 mg,  
42  
43 85%) as a white solid:  $R_f = 0.27$  (silica gel, 100% EtOAc);  $[\alpha]_D^{25} = -7.72$  ( $c$  0.04, MeOH);  
44  
45 mp = 201-202 °C;  $^1\text{H NMR}$  (400 MHz, MeOD)  $\delta$  (ppm) 7.44 – 7.40 (m, 2 H), 7.38 – 7.32 (m,  
46  
47 3 H), 7.32 – 7.27 (m, 1 H), 7.19 (d,  $J = 2.6$  Hz, 1 H), 6.85 (dd,  $J = 8.6, 2.6$  Hz, 1 H), 6.41 (d,  $J$   
48  
49 = 15.4 Hz, 1 H), 6.00 (dt,  $J = 15.4, 7.4$  Hz, 1 H), 5.08 (s, 2 H), 4.55 (s, 1 H), 4.45 (q,  $J = 7.1$   
50  
51 Hz, 1 H), 4.06 (d,  $J = 14.9$  Hz, 1 H), 3.87 (td,  $J = 6.8, 2.9$  Hz, 1 H), 3.72 (d,  $J = 14.9$  Hz, 1 H),  
52  
53 2.56 – 2.50 (m, 3 H), 1.47 (d,  $J = 7.2$  Hz, 3 H), 1.25 (d,  $J = 7.2$  Hz, 3 H);  $^{13}\text{C NMR}$  (100  
54  
55  
56  
57  
58  
59  
60

MHz, MeOD)  $\delta$  (ppm) 176.7, 171.6, 170.7, 158.3, 137.1, 134.5, 128.9, 128.1, 127.6, 127.5, 127.2, 126.7, 125.1, 113.0, 111.3, 72.3, 69.8, 50.1, 44.9, 42.9, 39.2, 15.3, 13.5; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{25}H_{30}N_3O_5$  452.2186; found 452.2170.

**Homoallylic Alcohol 77.** A solution of aldehyde **40** (395 mg, 1.54 mmol, 1.0 equiv) in  $CH_2Cl_2$  (15 mL) was cooled to 0 °C and  $SnCl_4$  (0.8 mL, 0.77 mmol, 0.5 equiv, 1.0 M in  $CH_2Cl_2$ ) was added slowly over 10 min at this temperature and then stirred for 10 min at 25 °C. Allyl trimethylsilane (0.35 mL, 2.30 mmol, 1.5 equiv) was added quickly and the reaction was stirred for 15 min, poured into  $Et_2O$  and after decantation the organic phase was washed with brine, dried over anhydrous  $MgSO_4$ , filtered and the solvent evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 5%  $EtOAc$  in Hexanes) to obtain the homoallylic alcohol **77** (340 mg, 74%) as a yellow solid:  $R_f$  = 0.67 (silica gel, 40%  $EtOAc$  in hexanes); mp = 74-75 °C;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 7.71 (d,  $J$  = 8.8 Hz, 1 H), 7.52 (d,  $J$  = 2.7 Hz, 1 H), 7.46 – 7.34 (m, 5 H), 7.26 – 7.23 (m, 1 H), 5.88 (dddd,  $J$  = 16.9, 10.5, 7.8, 6.4 Hz, 1 H), 5.26 – 5.19 (m, 2 H), 5.17 (t,  $J$  = 1.1 Hz, 1 H), 5.12 (s, 2 H), 2.67 (dddt,  $J$  = 14.0, 6.4, 3.8, 1.3 Hz, 1 H), 2.47 – 2.34 (m, 2 H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 158.0, 148.4, 135.8, 134.2, 131.5, 129.3, 128.8, 128.4, 127.6, 120.8, 118.9, 109.9, 70.7, 68.2, 42.8; HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{17}H_{18}NO_4$  300.1236; found 300.1238.

**Benzyloxymethyl Acetal 78.** To a solution of the homoallylic alcohol **77** (226 mg, 0.76 mmol, 1.0 equiv) in  $CH_2Cl_2$  (10 mL) was added DIPEA (0.52 mL, 3.02 mmol, 4.0 equiv) and BOMCl (0.56 mL, 3.02 mmol, 4.0 equiv) at 0 °C and the mixture was stirred at 25 °C for 15 h. After this time, the reaction was quenched by addition of saturated aqueous  $Na_2CO_3$  solution

1  
2 and the crude mixture was stirred for 30 min. Then, the aqueous phase was extracted with  
3  
4  $\text{CH}_2\text{Cl}_2$  and the organic phase washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and  
5  
6 the solvent evaporated under reduced pressure. The resulting residue was purified by flash  
7  
8 column chromatography (silica gel, 3% EtOAc in hexanes) to obtain benzyloxymethyl acetal  
9  
10 **78** (280 mg, 88%) as a pale yellow oil:  $R_f = 0.50$  (silica gel, 20% EtOAc in hexanes);  $^1\text{H}$   
11  
12 NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 7.67 (d,  $J = 8.7$  Hz, 1 H), 7.53 (d,  $J = 2.6$  Hz, 1 H), 7.46 –  
13  
14 7.22 (m, 11 H), 5.94 (ddt,  $J = 17.1, 10.1, 7.0$  Hz, 1 H), 5.35 (dd,  $J = 8.0, 4.2$  Hz, 1 H), 5.15  
15  
16 (dd,  $J = 17.0, 1.7$  Hz, 2 H), 5.11 (s, 2 H), 4.91 (d,  $J = 9.0$  Hz, 1 H), 4.71 (dd,  $J = 21.3, 5.5$  Hz,  
17  
18 2 H), 4.66 (s, 1 H), 4.59 (d,  $J = 6.9$  Hz, 1 H), 4.47 (d,  $J = 11.8$  Hz, 1 H), 2.67 – 2.59 (m, 1 H),  
19  
20 2.57 – 2.48 (m, 1 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 158.1, 148.9, 137.6, 135.8, 134.3,  
21  
22 130.3, 129.7, 128.8, 128.4, 128.4, 127.8, 127.7, 127.6, 120.8, 117.8, 109.7, 93.1, 73.3, 70.7,  
23  
24 69.8, 42.0; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{25}\text{H}_{26}\text{NO}_5$  420.1811; found 420.1815.  
25  
26  
27  
28  
29  
30  
31  
32

33 **Aniline 79.** To a solution of benzyloxymethyl acetal **78** (280 mg, 0.67 mmol, 1.0 equiv) in  
34  
35 EtOH (6 mL) was added a solution of  $\text{NH}_4\text{Cl}$  (179 mg, 3.34 mmol, 5.0 equiv) in water (4 mL)  
36  
37 followed by Zn dust (655 mg, 10.01 mmol, 15.0 equiv) in seven portions of ~100 mg each  
38  
39 over 30 min at 25 °C. The mixture was stirred at this temperature for 15 h and then the  
40  
41 reaction mixture was diluted with  $\text{CH}_2\text{Cl}_2$  and water, filtered and rinsed with  $\text{CH}_2\text{Cl}_2$ . The  
42  
43 filtrate was diluted with water and extracted with  $\text{CH}_2\text{Cl}_2$ , and the organic phase washed with  
44  
45 brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and the solvent evaporated under reduced  
46  
47 pressure. The resulting residue was purified by flash column chromatography (silica gel, 10%  
48  
49 EtOAc in hexanes) to obtain aniline **79** (217 mg, 87%) as a yellow oil:  $R_f = 0.60$  (silica gel,  
50  
51 30% EtOAc in hexanes);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 7.41 – 7.30 (m, 10 H), 6.94 (d,  
52  
53  $J = 8.4$  Hz, 1 H), 6.34 (dd,  $J = 8.3, 2.5$  Hz, 1 H), 6.28 (d,  $J = 2.5$  Hz, 1 H), 5.81 (ddt,  $J = 17.2,$   
54  
55  
56  
57  
58  
59  
60

1  
2 10.1, 7.0 Hz, 1 H), 5.12 (ddd,  $J = 17.1, 3.4, 1.4$  Hz, 1 H), 5.05 (ddt,  $J = 10.2, 2.1, 1.0$  Hz, 1  
3  
4 H), 5.01 (s, 2 H), 4.90 (d,  $J = 9.1$  Hz, 1 H), 4.73 – 4.71 (m, 2 H), 4.67 – 4.63 (m, 1 H), 4.52 (d,  
5  
6  $J = 11.7$  Hz, 1 H), 4.20 (bs, 2 H), 2.83 – 2.74 (m, 1 H), 2.62 – 2.53 (m, 1 H);  $^{13}\text{C}$  NMR (100  
7  
8 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 159.5, 146.5, 137.8, 137.2, 135.3, 130.6, 128.6, 128.4, 128.0, 127.9,  
9  
10 127.7, 127.5, 117.1, 116.9, 104.1, 102.9, 91.9, 77.9, 69.8, 69.7, 38.7; HRMS (H-ESI)  $m/z$ : [M  
11  
12 + H] $^+$  calcd for  $\text{C}_{25}\text{H}_{28}\text{NO}_3$  390.2069; found 390.2071.  
13  
14  
15  
16  
17

18 **Peptide 80.** To a solution of aniline **79** (217 mg, 0.56 mmol, 1.0 equiv) and Fmoc-D-Ala-OH  
19  
20 (**34**) (225 mg, 0.72 mmol, 1.3 equiv) in  $\text{CH}_2\text{Cl}_2$  (10 mL) was added PyBOP (435 mg, 0.84  
21  
22 mmol, 1.5 equiv) and DIPEA (0.2 mL, 1.11 mmol, 2.0 equiv) at 0 °C and the mixture was  
23  
24 stirred for 15 h at 25 °C. After this time, a saturated aqueous  $\text{NH}_4\text{Cl}$  solution was added and  
25  
26 the organic layer was separated. The aqueous phase was extracted with EtOAc, and the  
27  
28 combined organic layers were washed with brine, dried over anhydrous  $\text{MgSO}_4$ , filtered and  
29  
30 the solvent evaporated under reduced pressure. The crude product was purified by flash  
31  
32 column chromatography (silica gel, 30% EtOAc in hexanes) to obtain peptide **80** (247 mg,  
33  
34 65%, 1:1 mixture of diastereoisomers) as a white foam. Data assigned for the mixture of  
35  
36 diastereoisomers:  $R_f = 0.20$  (silica gel, 30% EtOAc in hexanes);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  
37  
38  $\delta$  (ppm) 9.32 (d,  $J = 11.0$  Hz, 1 H), 8.03 (d,  $J = 8.7$  Hz, 1 H), 7.77 (d,  $J = 7.4$  Hz, 2 H), 7.61  
39  
40 (d,  $J = 6.0$  Hz, 2 H), 7.47 – 7.27 (m, 13 H), 7.21 (dd,  $J = 7.2, 5.5$  Hz, 2 H), 7.02 (d,  $J = 8.4$  Hz,  
41  
42 1 H), 6.69 (dd,  $J = 8.4, 2.5$  Hz, 1 H), 5.72 (td,  $J = 16.8, 6.9$  Hz, 1 H), 5.40 (d,  $J = 6.5$  Hz, 1 H),  
43  
44 5.13 – 5.01 (m, 4 H), 4.77 – 4.64 (m, 3 H), 4.64 – 4.56 (m,  $J = 11.4, 8.2$  Hz, 1 H), 4.54 – 4.45  
45  
46 (m, 1 H), 4.43 – 4.32 (m, 2 H), 4.23 (dd,  $J = 14.3, 7.2$  Hz, 1 H), 2.73 – 2.61 (m, 1 H), 2.53 –  
47  
48 2.40 (m, 1 H), 1.51 – 1.42 (m, 3 H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 170.3, 159.1,  
49  
50 159.0, 155.8, 143.8, 141.3, 137.3, 137.2, 136.8, 134.2, 134.1, 129.9, 129.9, 128.5, 128.5,  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60



1  
2 128.4, 127.9, 127.9, 127.9, 127.9, 127.9, 127.8, 127.6, 127.1, 125.2, 125.0, 120.0, 117.9,  
3  
4 117.8, 111.1, 108.3, 108.1, 92.4, 78.6, 70.1, 70.0, 67.1, 51.6, 47.1, 39.9, 39.9, 18.9, 18.9;  
5  
6 HRMS (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for  $C_{43}H_{43}N_2O_6$  683.3121; found 683.3115.  
7  
8  
9

10  
11 **Dipeptide 81.** To a solution of peptide **80** (176 mg, 0.26 mmol, 1.0 equiv) in  $CH_2Cl_2$  (5 mL)  
12  
13 was added piperidine (0.13 mL, 1.29 mmol, 5.0 equiv) and the reaction mixture was stirred at  
14  
15 25 °C for 5 h. After this time, the organic solvent was removed under reduced pressure and the  
16  
17 resulting crude product was purified by flash column chromatography (silica gel, 30% EtOAc  
18  
19 in hexanes  $\rightarrow$  60% EtOAc in hexanes) to obtain the corresponding amine (110 mg, 93%, 1:1  
20  
21 mixture of diastereoisomers) as a white solid. Data assigned for the mixture of  
22  
23 diastereoisomers:  $R_f$  = 0.18 (silica gel, 80% EtOAc in hexanes); mp = 82-83 °C;  $^1H$  NMR  
24  
25 (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 8.15 – 8.01 (m, 1 H), 7.48 – 7.21 (m, 12 H), 7.12 – 6.99 (m, 1 H),  
26  
27 6.74 – 6.62 (m, 1 H), 5.84 – 5.67 (m, 1 H), 5.13 – 4.99 (m, 5 H), 4.83 – 4.64 (m, 4 H), 4.58 –  
28  
29 4.45 (m, 1 H), 3.76 (bs, 1 H), 2.71 (dt,  $J$  = 14.8, 8.0 Hz, 1 H), 2.50 (dt,  $J$  = 13.6, 6.6 Hz, 1 H),  
30  
31 1.53 – 1.25 (m, 3 H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 172.3, 171.9, 159.1, 159.0, 137.6,  
32  
33 137.5, 137.5, 137.5, 137.4, 137.3, 137.3, 137.3, 136.9, 136.9, 134.5, 134.4, 128.5, 128.5,  
34  
35 128.5, 128.5, 128.1, 128.0, 127.9, 127.9, 127.6, 121.4, 121.2, 117.8, 117.7, 110.8, 110.6,  
36  
37 108.5, 108.2, 92.3, 92.3, 70.1, 70.0, 69.9, 61.4, 61.2, 51.4, 51.3, 40.1, 39.9, 16.4, 15.8. To a  
38  
39 solution of the amine obtained above (105 mg, 0.23 mmol, 1.0 equiv) and Fmoc-Gly-OH (**35**)  
40  
41 (88 mg, 0.30 mmol, 1.3 equiv) in  $CH_2Cl_2$  (5 mL) was added HATU (130 mg, 0.34 mmol, 1.5  
42  
43 equiv) and DIPEA (0.1 mL, 0.46 mmol, 2.0 equiv) at 0 °C and the mixture was stirred for 15 h  
44  
45 at 25 °C. After this time, a saturated aqueous  $NH_4Cl$  solution was added and the organic layer  
46  
47 was separated. The aqueous phase was extracted with EtOAc, and the combined organic  
48  
49 layers were washed with brine, dried over anhydrous  $MgSO_4$ , filtered and the solvent  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2 evaporated under reduced pressure. The crude product was purified by flash column  
3  
4 chromatography (silica gel, 15% EtOAc in hexanes → 40% EtOAc in hexanes) to obtain  
5  
6 peptide **81** (123 mg, 73%, 1:1 mixture of diastereoisomers) as a white foam. Data assigned for  
7  
8 the mixture of diastereoisomers:  $R_f = 0.27$  (silica gel, 50% EtOAc in hexanes);  $^1\text{H NMR}$  (400  
9  
10 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 7.99 – 7.96 (m, 1 H), 7.75 (dd,  $J = 7.5, 3.4$  Hz, 2 H), 7.59 (d,  $J = 7.3$   
11  
12 Hz, 2 H), 7.45 – 7.23 (m, 16 H), 7.03 (t,  $J = 8.9$  Hz, 1 H), 6.71 – 6.66 (m, 1 H), 5.72 (ttd,  $J =$   
13  
14 13.7, 7.0, 3.3 Hz, 1 H), 5.53 (bs, 1 H), 5.13 – 5.06 (m, 2 H), 5.04 (s, 2 H), 4.77 – 4.55 (m, 6  
15  
16 H), 4.44 (d,  $J = 6.8$  Hz, 2 H), 4.22 (t,  $J = 6.9$  Hz, 1 H), 3.97 – 3.78 (m, 2 H), 2.64 (dt,  $J = 16.3,$   
17  
18 7.0 Hz, 1 H), 2.44 (ddd,  $J = 20.6, 13.8, 6.5$  Hz, 1 H), 1.49 – 1.44 (m, 3 H);  $^{13}\text{C NMR}$  (100  
19  
20 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 174.6, 170.1, 170.0, 165.6, 159.1, 159.1, 156.8, 156.7, 143.7, 143.7,  
21  
22 141.3, 141.3, 137.2, 136.8, 134.2, 134.2, 130.0, 129.9, 128.6, 128.1, 128.1, 128.0, 127.8,  
23  
24 127.7, 127.6, 127.1, 127.0, 125.1, 125.0, 121.3, 121.2, 120.0, 119.9, 117.9, 117.9, 111.2,  
25  
26 111.2, 108.5, 108.3, 92.6, 92.5, 78.3, 77.8, 70.3, 70.2, 70.1, 65.4, 49.9, 47.1, 44.5, 38.7, 18.4,  
27  
28 18.3; HRMS (H-ESI)  $m/z$ :  $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{45}\text{H}_{46}\text{N}_3\text{O}_7$  740.3336; found 740.3327.  
29  
30  
31  
32  
33  
34  
35  
36

37 **Diolefin 83.** To a solution of peptide **81** (63 mg, 0.09 mmol, 1.0 equiv) in  $\text{CH}_2\text{Cl}_2$  (5 mL) was  
38  
39 added piperidine (0.04 mL, 0.43 mmol, 5.0 equiv) and the reaction mixture was stirred at 25  
40  
41  $^\circ\text{C}$  5 h. After this time, the organic solvent was removed under reduced pressure and the  
42  
43 resulting crude amine was used in the next step without purification. To a solution of the  
44  
45 crude amine (~0.09 mmol) and acid **82** (10  $\mu\text{L}$ , 0.11 mmol, 1.3 equiv) in  $\text{CH}_2\text{Cl}_2$  (5 mL) was  
46  
47 added HATU (49 mg, 0.13 mmol, 1.5 equiv) and DIPEA (30  $\mu\text{L}$ , 0.17 mmol, 2.0 equiv) at 0  
48  
49  $^\circ\text{C}$  and the mixture was stirred for 12 h at 25  $^\circ\text{C}$ . After this time, a saturated aqueous  $\text{NH}_4\text{Cl}$   
50  
51 solution was added and the organic layer was separated. The aqueous phase was extracted  
52  
53 with EtOAc, and the combined organic layers were washed with brine, dried over anhydrous  
54  
55  
56  
57  
58  
59  
60

1  
2 MgSO<sub>4</sub>, filtered and the solvent evaporated under reduced pressure. The crude product was  
3  
4 purified by flash column chromatography (silica gel, 30% EtOAc in hexanes → 100% EtOAc)  
5  
6 to obtain diolefin **83** (32 mg, 64% over 2 steps, 1:1 mixture of diastereoisomers) as a white  
7  
8 solid. Data assigned for the mixture of diastereoisomers:  $R_f = 0.62$  (silica gel, 100% EtOAc);  
9  
10 mp = 95-96 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.95 (t,  $J = 2.6$  Hz, 1 H), 7.46 – 7.27 (m,  
11  
12 12 H), 7.03 (t,  $J = 8.6$  Hz, 1 H), 6.85 (s, 1 H), 6.69 (dt,  $J = 8.4, 2.8$  Hz, 1 H), 5.99 – 5.84 (m, 1  
13  
14 H), 5.80 – 5.65 (m, 1 H), 5.27 – 5.19 (m, 2 H), 5.16 – 5.06 (m, 2 H), 5.05 (s, 2 H), 4.76 – 4.63  
15  
16 H), 4.56 – 4.46 (m, 2 H), 4.03 – 3.95 (m, 2 H), 3.07 – 3.00 (m, 2 H), 2.65 (ddd,  $J = 15.3,$   
17  
18 13.9, 7.7 Hz, 1 H), 2.44 (ddd,  $J = 20.9, 13.1, 6.4$  Hz, 1 H), 1.49 – 1.43 (m, 3 H); <sup>13</sup>C NMR  
19  
20 (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 174.5, 171.8, 171.4, 170.1, 168.8, 165.7, 160.2, 159.1, 140.9,  
21  
22 137.2, 137.1, 136.8, 136.8, 134.3, 134.2, 130.7, 129.9, 129.9, 128.8, 128.6, 128.1, 128.1,  
23  
24 128.0, 127.7, 127.7, 127.6, 127.0, 120.2, 120.1, 117.9, 117.8, 111.2, 111.1, 108.6, 108.4, 92.7,  
25  
26 92.5, 78.8, 70.3, 70.3, 70.2, 70.1, 50.1, 50.1, 43.1, 43.0, 41.2, 41.2, 38.7, 18.3, 18.2; HRMS  
27  
28 (H-ESI)  $m/z$ :  $[M + H]^+$  calcd for C<sub>34</sub>H<sub>40</sub>N<sub>3</sub>O<sub>6</sub> 586.2917; found 586.2905.  
29  
30  
31  
32  
33  
34  
35  
36

37 **Macrocycle 85.** To a stirred solution of macrocycle **27** (28 mg, 0.07 mmol, 1.0 equiv) in THF  
38  
39 (10 mL) at -78 °C was added condensed liquid NH<sub>3</sub> (~8 mL) *via* cannula. Then, small pieces  
40  
41 of Na (130 mg, 5.61 mmol, 85 equiv) were added to the mixture until the formation of a deep  
42  
43 blue solution. The reaction was stirred at -78 °C for 2 h and after this time it was quenched by  
44  
45 slowly addition of MeOH at the same temperature. The mixture was allowed to reach room  
46  
47 temperature and concentrated under reduced pressure to a volume of ~10 mL. The reaction  
48  
49 was neutralized with Dowex-H<sup>+</sup>, washed with MeOH and concentrated under reduced  
50  
51 pressure. The crude product was purified by flash column chromatography (silica gel, 7%  
52  
53 MeOH in CH<sub>2</sub>Cl<sub>2</sub>) to obtain macrocycle **85** (18 mg, 85%) as a white solid:  $R_f = 0.34$  (silica  
54  
55  
56  
57  
58  
59  
60

1  
2 gel, 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub>); [ $\alpha$ ]<sub>D</sub><sup>25</sup> = - 6.77 (c 0.09, MeOH); mp = 125-126 °C; <sup>1</sup>H NMR  
3  
4 (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  (ppm) 9.30 (bs, 1 H), 8.51 (d, *J* = 7.2 Hz, 1 H), 8.38 (bs, 1 H), 7.03  
5  
6 (d, *J* = 2.4 Hz, 1 H), 6.91 (d, *J* = 8.2 Hz, 1 H), 6.49 (dd, *J* = 8.2, 2.5 Hz, 1 H), 5.40 (dt, *J* =  
7  
8 15.1, 5.0 Hz, 1 H), 5.07 (dt, *J* = 15.1, 5.7 Hz, 1 H), 4.32 (p, *J* = 7.0 Hz, 1 H), 4.11 (bs, 1 H),  
9  
10 3.84 (dd, *J* = 14.8, 6.4 Hz, 1 H), 3.43 (dd, *J* = 14.8, 5.6 Hz, 1 H), 3.16 – 3.09 (m, 1 H), 2.96  
11  
12 (dd, *J* = 16.4, 5.2 Hz, 1 H), 2.28 – 2.08 (m, 4 H), 1.29 (d, *J* = 7.1 Hz, 3 H); <sup>13</sup>C NMR (100  
13  
14 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  173.1, 171.2, 170.9, 156.3, 136.9, 131.4, 131.0, 127.8, 123.7, 112.7,  
15  
16 112.4, 49.7, 49.1, 43.4, 33.8, 27.9, 16.9; HRMS (H-ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>17</sub>H<sub>22</sub>N<sub>3</sub>O<sub>4</sub>  
17  
18 332.1610; found 332.1618.  
19  
20  
21  
22  
23  
24

25 **Cell growth assay.** The 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide or  
26  
27 MTT dye reduction assay in 96-well microplates was used.<sup>32</sup> This assay is dependent on the  
28  
29 reduction of MTT by mitochondrial dehydrogenases of a viable cell to a blue formazan  
30  
31 product, which can be measured spectrophotometrically. 2.5 × 10<sup>3</sup> BAEC, 3 × 10<sup>3</sup> HL-60 and  
32  
33 KU812F and 2 × 10<sup>3</sup> HepG2, HT-1080, HT-29, MDA-MB-231, U2OS, U937 and U87MG  
34  
35 cells in a total volume of 100  $\mu$ L of their respective growth medium were incubated with  
36  
37 serial dilutions 1:1 of the tested compounds. After 3 days of incubation (37 °C and 5% CO<sub>2</sub> in  
38  
39 a humid atmosphere), 10  $\mu$ L of MTT (5 mg/mL in phosphate-buffered saline) were added to  
40  
41 each well, and the plate was incubated for a further 4 h at 37 °C. The resulting formazan was  
42  
43 dissolved in 150  $\mu$ L of 0.04 N HCl/2-propanol and read at 550 nm. IC<sub>50</sub> values were  
44  
45 calculated from semi-logarithmic dose-response plots as those concentrations of compound  
46  
47 yielding 50% cell survival, taking the values obtained for the control to be 100%. IC<sub>50</sub> results  
48  
49 are expressed as means  $\pm$  S.D. of at least three independent experiments.  
50  
51  
52  
53  
54  
55  
56

## 57 ASSOCIATED CONTENT

58  
59  
60

## Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at

DOI:

<sup>1</sup>H and <sup>13</sup>C NMR spectra for all new compounds, dose-response plots of the biological studies and theoretical calculations data.

## AUTHOR INFORMATION

### Corresponding Author

\* E-mail: frsarabia@uma.es

### Author Contributions

The manuscript was written through contributions of all authors. / All authors have given approval to the final version of the manuscript. / ‡I. C.-S. performed all the synthetic work and designed the experiments. A. S.-R. and J. M. L.-R. performed the theoretical calculations. P. C., B. M.-P., A. R. Q. and M. A. M. performed the biological evaluations. F. S. and I. C.-S. conceived and directed the study.

### Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENT

This work was financially supported by the Ministerio de Ciencias e Innovación (MICINN) (ref. CTQ2014-60223-R and CTQ2016-76311-R) and Junta de Andalucía and “Fondo Europeo de Desarrollo Regional-FEDER” (P12 CTS-1507). I. C.-S. thanks Ministerio de Educación, Cultura y Deporte for a predoctoral fellowship (FPU programme). The authors thank Dr. J. I. Trujillo from Pfizer (Groton, CT) for assistance in the preparation of this

manuscript. The authors thank the Unidad de Espectroscopía de Masas and the NMR facility of the University of Málaga for exact mass and NMR spectroscopic assistance.

## REFERENCES

1. Festa, C.; De Marino, S.; Sepe, V.; D'Auria, M. V.; Bifulco, G.; Débitus, C.; Bucci, M.; Vellecco, V.; Zampella, A. *Org. Lett.* **2011**, *13*, 1532–1535.
2. (a) Wegerski, C. J.; Hammond, J.; Tenney, K.; Matainaho, T.; Crews, P. *J. Nat. Prod.* **2007**, *70*, 89–94. (b) Festa, C.; De Marino, S.; Sepe, V.; Monti, M. C.; Luciano, P.; D'Auria, M. V.; Débitus, C.; Bucci, M.; Vellecco, V.; Zampella, A. *Tetrahedron* **2009**, *65*, 10424–10429 and references therein.
3. (a) Kavitha, N.; Kumar, V. P.; Chandrasekhar, S. *Tetrahedron Lett.* **2013**, *54*, 2128–2130. (b) Kavitha, N.; Chandrasekhar, S. *Org. Biomol. Chem.* **2015**, *13*, 6242–6248.
4. (a) Kashinath, K.; Vasudevan, N.; Reddy, D. S. *Org. Lett.* **2012**, *14*, 6222–6225. (b) Kashinath, K.; Dhara, S.; Reddy, D. S. *Org. Lett.* **2015**, *17*, 2090–2093.
5. Vasudevan, N.; Kashinath, K.; Reddy, D. S. *Org. Lett.* **2014**, *16*, 6148–6151.
6. Reddy, D. S.; Kormirishetty, K.; Natrajan, V. WO Patent 2014083578 A1, Nov 27, 2013; *Chem. Abstr.* **2014**, *161*, 70679.
7. Kashinath, K.; Jachak, G. R.; Athawale, P. R.; Marelli, U. K.; Gonnade, R. G.; Reddy, D. S. *Org. Lett.* **2016**, *18*, 3178–3181.
8. (a) Sarabia, F.; Chammaa, S.; Sánchez-Ruiz, A.; Martín-Ortiz, L.; López-Herrera, F. J. *Curr. Med. Chem.* **2004**, *11*, 1309–1332. (b) Sarabia, F.; Chammaa, S. *J. Org. Chem.* **2005**, *70*, 7846–7857. (c) Sarabia, F.; Chammaa, S.; García-Castro, M. *J. Org. Chem.* **2005**, *70*, 7858–7865. (d) Sarabia, F.; Chammaa, S.; García-Ruiz, C. *J. Org. Chem.* **2011**,

- 1  
2 76, 2132–2144. (e) Goh, S.; Hohmeier, A.; Stone, T. C.; Offord, V.; Sarabia, F.; García-  
3  
4 Ruiz, C.; Good, L. *Appl. Environ. Microbiol.* **2015**, *81*, 5650–5659.
- 5  
6  
7 9. Nicolaou, K. C.; Bulger, P. G.; Sarlah, D. *Angew. Chem. Int. Ed.* **2005**, *44*, 4490–4527.
- 8  
9  
10 10. Cheng-Sánchez, I.; García-Ruiz, C.; Sarabia, F. *Tetrahedron Lett.* **2016**, *57*, 3392–3395.
- 11  
12 11. Jin-Gim, H.; Li, H.; Lee, E.; Ryu, J.-H.; Jeon, R. *Bioorg. Med. Chem. Lett.* **2013**, *23*,  
13  
14 513–517.
- 15  
16 12. Asif, K.; Himaja, M.; Ramana, M. V.; Sikarwar, M. S. *Asian J. Chem.* **2012**, *24*,  
17  
18 2739–2743.
- 19  
20  
21 13. Hong, S. H.; Sanders, D. P.; Lee, C. W.; Grubbs, R. H. *J. Am. Chem. Soc.* **2005**, *127*,  
22  
23 17160–17161.
- 24  
25  
26 14. For representative examples of the influence of the structural pattern in the  
27  
28 stereochemistry of the ring-closing metathesis, see: (a) Nicolaou, K. C.; He, Y.;  
29  
30 Vourloumis, D.; Vallberg, H.; Roschangar, F.; Sarabia, F.; Ninkovic, S.; Yang, Z.;  
31  
32 Trujillo, J. I. *J. Am. Chem. Soc.* **1997**, *119*, 7960–7973. (b) Vassilikogiannakis, G.;  
33  
34 Margaros, I.; Tofi, M. *Org. Lett.* **2004**, *6*, 205–208. (c) Nicolaou, K. C.; Montagnon, T.;  
35  
36 Vassilikogiannakis, G.; Mathison, C. J. N. *J. Am. Chem. Soc.* **2005**, *127*, 8872–8888.
- 37  
38  
39 15. (a) Hoye, T. R.; Zhao, H. *Org. Lett.* **1999**, *1*, 1123–1125. (b) Maishal, T. K.; Sinha-  
40  
41 Mahapatra, D. K.; Paranjape, K.; Sarkar, A. *Tetrahedron Lett.* **2002**, *43*, 2263–2267. (c)  
42  
43 Imahori, T.; Ojima, H.; Tateyama, H.; Mihara, Y.; Takahata, H. *Tetrahedron Lett.* **2008**,  
44  
45 49, 265–268. (d) Imahori, T.; Ojima, H.; Yoshimura, Y.; Takahata, H. *Chem. Eur. J.*  
46  
47 **2008**, *14*, 10762–10771.
- 48  
49  
50  
51  
52  
53 16. (a) Cheng-Sánchez, I.; García-Ruiz, C.; Guerrero-Vásquez, G. A.; Sarabia, F. *J. Org.*  
54  
55 *Chem.* **2017**, *82*, 4744–4757. (b) Paquette, L. A.; Efremov, I. *J. Am. Chem. Soc.* **2001**,  
56  
57  
58  
59  
60

- 1  
2 123, 4492–4501. (c) Gurjar, M. K.; Yakambram, P. *Tetrahedron Lett.* **2001**, *42*,  
3  
4 3633–3636.  
5  
6  
7 17. Carrión, M. D.; Chayah, M.; Entrena, A.; López, A.; Gallo, M. A.; Acuña-Castroviejo, D.;  
8  
9 Camacho, M. E. *Bioorg. Med. Chem.* **2013**, *21*, 4132–4142.  
10  
11 18. For a comprehensive revision of different methods of Boc cleavage see: (a) Dandepally,  
12  
13 S. R.; Williams, A. L. *Tetrahedron Lett.* **2009**, *50*, 1071–1074. (b) Kumar, G. P.;  
14  
15 Rambabu, D.; Rao, M. V. B.; Pal, M. *J. Chem.* **2013**, *2013*, 916960; *Chem. Abstr.* **2013**,  
16  
17 *160*, 723475.  
18  
19 19. Hengartner, U.; Batcho, A. D.; Blount, J. F.; Leimgruber, W.; Larscheid, M. E.; Scott, J.  
20  
21 *W. J. Org. Chem.* **1979**, *44*, 3748–3752.  
22  
23 20. (a) Martín-Ortiz, L.; Chammaa, S.; Pino-González, M. S.; Sánchez-Ruiz, A.; García-  
24  
25 Castro, M.; Assiego, C.; Sarabia, F. *Tetrahedron Lett.* **2004**, *45*, 9069–9072. (b)  
26  
27 Miyashita, M.; Mizutani, T.; Tadano, G.; Iwata, Y.; Miyazawa, M.; Tanino, K. *Angew.*  
28  
29 *Chem. Int. Ed.* **2005**, *44*, 5094–5097. (c) Yu, X.-Q.; Yoshimura, F.; Ito, F.; Sasaki, M.;  
30  
31 Hirai, A.; Tanino, K.; Miyashita, M. *Angew. Chem. Int. Ed.* **2008**, *47*, 750–754.  
32  
33 21. Yang, D.; Wong, M.-K.; Yip, Y.-C. *J. Org. Chem.* **1995**, *60*, 3887–3889.  
34  
35 22. DFT calculations were carried out using Gaussian 09 at the B3LYP/6-31G' level of  
36  
37 theory using the SMD solvation methodology, choosing acetonitrile as solvent.  
38  
39 23. Aoyama, A.; Endo-Umeda, K.; Kishida, K.; Ohgane, K.; Noguchi-Yachide, T.; Aoyama,  
40  
41 H.; Ishikawa, M.; Miyachi, H.; Makishima, M.; Hashimoto, Y. *J. Med. Chem.* **2012**, *55*,  
42  
43 7360–7377.  
44  
45 24. Wagner, H.; Harms, K.; Koert, U.; Meder, S.; Boheim, G. *Angew. Chem. Int. Ed.* **1996**,  
46  
47 *35*, 2643–2646.  
48  
49 25. Engelhardt, F. C.; Schmitt, M. J.; Taylor, R. E. *Org. Lett.* **2001**, *3*, 2209–2212.  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60



- 1  
2 26. Hoveyda, A. H.; Lombardi, P. J.; O'Brien, R. V.; Zhugralin, A. R. *J. Am. Chem. Soc.*  
3  
4 **2009**, *131*, 8378–8379.  
5  
6  
7 27. (a) Lin, Y. A.; Davis, B. G. *Beilstein J. Org. Chem.* **2010**, *6*, 1219–1228. (b) Fuwa, H.;  
8  
9 Saito, A.; Sasaki, M. *Angew. Chem. Int. Ed.* **2010**, *49*, 3041–3044. (c) Schmidt, B.;  
10  
11 Staude, L. *J. Org. Chem.* **2009**, *74*, 9237–9240. (d) Schmidt, B.; Nave, S. *Chem.*  
12  
13 *Commun.* **2006**, 2489–2491.  
14  
15  
16 28. (a) Fürstner, A.; Thiel, O. R.; Lehmann, C. W. *Organometallics* **2002**, *21*, 331–335. (b)  
17  
18 Choi, T.-L.; Chatterjee, A. K.; Grubbs, R. H. *Angew. Chem. Int. Ed.* **2001**, *40*,  
19  
20 1277–1279. (c) Chatterjee, A. K.; Choi, T.-L.; Sanders, D. P.; Grubbs, R. H. *J. Am.*  
21  
22 *Chem. Soc.* **2003**, *125*, 11360–11370.  
23  
24  
25  
26 29. For representative examples see: (a) Lübbe, C.; Dumrath, A.; Neumann, H.; Beller, M.;  
27  
28 Kadyrov, R. *ChemCatChem* **2014**, *6*, 105–108. (b) Nagarapu, L.; Gaikwad, H. K.; Bantu,  
29  
30 R.; Manikonda, S. R.; Kumar, C. G.; Pombala, S. *Tetrahedron Lett.* **2012**, *53*,  
31  
32 1287–1291. (c) Pentzer, E. B.; Gadzikwa, T.; Nguyen, S. T. *Org. Lett.* **2008**, *10*,  
33  
34 5613–5615. (d) Vedrenne, E.; Dupont, H.; Oualef, S.; Elkaïm, L.; Grimaud, L. *Synlett*  
35  
36 **2005**, 670–672.  
37  
38  
39  
40  
41 30. Cárdenas, C; Quesada, A. R.; Medina, M. A. *Cell. Mol. Life Sci.* **2006**, *63*, 3083–3089.  
42  
43  
44 31. Minimum energy conformations were calculated using the PM3 method found in  
45  
46 HyperChem 5.0 software. Optimization was performed using Polak-Ribiere algorithm  
47  
48 until the RMS gradient reached a value below 0.001 kcal/(Å·mol).  
49  
50  
51 32. Mosmann, T. *J. Immunol. Methods* **1983**, *65*, 55–63.  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60