# Structure-Based Design and Synthesis of Benzothiazole Phosphonate Analogues with Inhibitors of Human ABAD-A $\beta$ for Treatment of Alzheimer's disease 

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#### Abstract

Amyloid binding alcohol dehydrogenase (ABAD), a mitochondrial protein, is a cofactor facilitating amyloid- $\beta$ peptide (A $\beta$ ) induced cell stress. Antagonizing $A \beta-A B A D$ interaction protects against aberrant mitochondrial and neuronal function and improves learning memory in the Alzheimer's disease (AD) mouse model. Therefore, it offers a potential target for Alzheimer's drug design, by identifying potential inhibitors of A $\beta$-ABAD interaction. 2D QSAR methods were applied to novel compounds with known $\mathrm{IC}_{50}$ values, which formed a training set. A correlation analysis was carried out comparing the statistics of the measured $\mathrm{IC}_{50}$ with predicted values. These selectivity-determining descriptors were interpreted graphically in terms of principle component analyses, which are highly informative for the lead optimization process with respect to activity enhancement. A 3D pharmacophore model also was created. The 2D QSAR and 3D pharmacophore models will assist in hi-throughput screening. In addition, ADME descriptors were also determined to study their pharmacokinetic properties. Finally, ABAD molecular docking study of these novel molecules was undertaken to determine whether these compounds exhibit significant binding affinity with the binding site. We have synthesized only the compounds that have shown the best drug like properties as candidates for further studies.


## Keywords

Molecular Docking; Quantitative Structure Activity Relationship; ADME Prediction; ABAD Inhibitors; Benzothiazole Aminophosphonates

## Introduction

Alzheimer's disease (AD) is one of the most common dementias showing slowly progressive cognitive decline. Alzheimer's brain is characterized by accumulation of amyloid beta peptide $(A \beta)$ and the formation of neurofibrillary tangles. $A \beta$ plays a central role in the development of AD pathology and contributes to neuronal, synaptic, and cognitive malfunction. Mitochondrial and synaptic dysfunction is an early pathological feature of $\mathrm{AD}(1-6)$. Recent studies have highlighted the significance of mitochondrial $\mathrm{A} \beta$ accumulation and synaptic mitochondrial dysfunction. A $\beta$ progressively accumulates in synaptic mitochondria and impairs mitochondrial structure and function including

[^0]membrane potential, membrane permeability transition pore, respiration, energy metabolism, oxidative stress, mitochondrial dynamics, and calcium homeostasis (4, 7-19). Thus, strategies that suppress/attenuate $A \beta$-induced mitochondrial toxicity in addition to $A \beta$ levels in the brain and improve cognitive function are critical for preventing and/or halting at an early stage of AD.

Amyloid binding alcohol dehydrogenase (ABAD), a mitochondrial enzyme, plays a key role in mitochondrial dysfunction and in the pathogenesis of AD. This enzyme has attracted considerable interest because of its ability to interact with $A \beta$. Importantly, the interaction of $A B A D$ with $A \beta$ mediates mitochondrial and synaptic dysfunction (9, 20). Antagonizing $A \beta$ $A B A D$ interaction with the $A B A D$ decoy peptide that encompasses the amino residues responsible for $\mathrm{A} \beta$ binding to ABAD protects against aberrant mitochondrial and neuronal function and improves learning memory in AD transgenic mice (9, 17, 20). Furthermore, interception of $\mathrm{A} \beta-\mathrm{ABAD}$ interaction also significantly reduces mitochondrial and cerebral $A \beta$ accumulation (17). These data support that $A \beta-A B A D$ interaction is a potential target of the drug development for treatment of AD.

In this way, the search for inhibitors of $A \beta-A B A D$ interaction has started and using an ELISA-based screening assay, frentizole, an FDA-approved immunosuppressive drug, was identified as a novel inhibitor of $A \beta-A B A D$ interaction. Analyzing the frentizole SAR studies, we have developed novel benzothiazole ureas with a 30 -fold improvement in potency(21). Recently, AG18051 (1-azepan-1-yl-2-phenyl-2-(4-thioxo-1,4-dihydropyrazolo[3,4-d]pyrimidin-5-yl)-ethanone) was also identified as a potent inhibitor of ABAD.(22) However, currently available inhibitors of $A \beta-A B A D$ interaction have the disadvantages of low solubility, poorly crossing the blood brain barrier ( BBB ), high toxicity, and low cell permeability. Current efforts to design $A \beta-A B A D$ inhibitors have been unsuccessful due largely to poor ADME (Absorption, Distribution, Metabolism and Excretion) properties. To overcome the limitation of currently available ABAD inhibitors, we have designed a new class of small molecular inhibitors of $A \beta-A B A D$ interaction via phosphonate derivatives. The goal of this study is to identify the potential blockers of $A \beta$ $A B A D$ interaction as therapeutic targets of $A D$.

The quantitative structure-activity relationship (QSAR) analysis helps to derive highly applicable models that allow designing novel and reactive molecules. Any statistical tool and/or graphical model can predict physicochemical properties to increase the selectivity and affinity of new compounds.(23) These properties may be measured by a variety of calculated descriptors, i.e., potential hydrogen bond donors or acceptors, molar refractivity, hydrophobicity, $\operatorname{Slog} P$. We used descriptors to evaluate the drug-like properties of our compounds. In this present study, we also applied 2D QSAR and 3D ligand based pharmacophore methods to predict the inhibitory activity of a test set of compounds by constructing a model from a training set. For the QSAR model, linearity of the correlation and root mean square error (RSME) and correlation factor $\left(\mathrm{R}^{2}\right)$ indicate the robustness of the model. The accuracy of the pharmacophore model was evaluated by selecting hits from a test set. Docking studies to ABAD were carried out to determine the binding affinity differences of our compounds. Reliable models were obtained that provided interesting information with respect to binding affinity and compound selectivity discrimination.

In our previous work, we have described the synthesis and evaluation of a novel class of benzothiazole urea derivatives as potent $A \beta-A B A D$ inhibitors.(24) Based on benzothiazole urea and frentizole structure activities studies, we have designed benzothiazole amino and frentizole phosphonate derivatives. Molecular docking, QSAR studies and pharmacokinetics/absorption, distribution, metabolism, and excretion (ADME) prediction
for any given scaffold of interest are the most popular methods of computer aided drug design.

With respect to ADME prediction, we have predicted the cell permeability of Caco2 (human colon adenocarcinoma) cell permeability, MDCK (Madin-Darby canine kidney), human intestinal absorption, BBB penetration and plasma protein binding. Finally, the compounds with satisfactory properties were chosen for molecular docking against ABAD protein to find out whether they are able to bind to the protein and inhibit. The results from the molecular docking can be used to interpret the efficacy of the novel molecules to inhibit ABAD and in turn they may be useful as drugs for the treatment and management of AD . Based on the results of these studies, we have synthesized the compounds that have shown the best drug-like properties.

## Materials and Methods

## General procedure for the synthesis of small molecular weight benzothiazole phosphonate derivatives (Scheme 1)

To a stirred solution of aromatic/heterocyclic aldehydes (2) in 10 mL of toluene, $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$ ( $5 \mathrm{~mol} \%$ ) was added. The mixture was stirred magnetically for $10-15$ minutes, later substituted 2-amino benzothiazoles (1) and dimethyl/diethyl phosphite (3) in anhydrous toluene ( 5 mL ) were added drop-wise with stirring at room temperature for 30 minutes. The reaction mixture was refluxed with stirring for 5-12 hours. After completion of the reaction, as determined by TLC (ethyl acetate:hexane), the solvent was removed in a rotaryevaporator. The reaction mixture was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ). The combined EtOAc extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure to afford products $\mathbf{4 a - j}$, which were passed through a column of silica gel and eluted with EtOAchexane (1:1).

## General procedure for the synthesis of urea/thiourea phosphonate derivatives (Scheme 2)

A mixture of urea/thiourea ( 1 mmol ) and methyl 5-formyl-2-hydroxybenzoate ( 2 mmol ) in ethanol ( 15 mL ) was refluxed for the appropriate reaction (TLC) time. The solvent was evaporated in vacuo and the resulting crude material was purified by chromatography on a short column of silica gel (EtOAc: petroleum ether, 1:3) and then recrystallized from ethanol/dichloromethane (4:1) to afford the urea derivatives.

A mixture of urea derivatives ( 1 mmol ), toluene ( 20 mL ), and dimethyl phosphite (DMP)/ diethyl phosphate (DEP) $(2 \mathrm{mmol})$ was heated at $110^{\circ} \mathrm{C}$ for 15 hours. The solvent was evaporated in vacuo and the resulting crude material was purified by chromatography on a short column of silica gel (EtOAc: dichloromethane, 1:3) and then recrystallized from ethanol (4:1) to afford the target molecules.

## Construction of models and Molecular Dynamics of the novel leads (25)

The 3D structure of all ligands were constructed in MOE (Molecular Operating Environment) working environment and subjected to energy minimization.(26) The Merck Molecular Force Field (MMFF94x) and parameter set was employed and the related potential energy terms were enabled for all bonded interactions, Van der Waals interactions, electrostatic interactions and restraints. The non-bonded cut off value was enabled between 8-10 Ang. The Generalized Born implicit salvation model was enabled. The gradient was set to 0.05 and force filed partial charges were enabled to calculate during minimization process. The dynamics simulations were carried out with the initial temperature set to 30 K and increased to a 300 K run time temperature. Heat time and cool time were set to 0 pico seconds. The final stabilized conformations were used for the construction of a local data
base of present novel leads and chosen to determine 2D and 3D descriptors, including Lipinski rules that define drug-like properties.

## Construction of a 2D QSAR model (25)

The QSAR suite of applications in MOE was used to calculate and analyze the data and build numerical models of the data for prediction and interpretation purposes. Any QSAR model for a given set of molecules correlates the activities with properties inherent to each molecule in the set itself. A database of 45 compounds was used to generate independent training and test data sets. Initially, the QSAR descriptors SlogP, Density, Molar refractivity, Molecular weight, atomic polarizability, $\log \mathrm{P}(\mathrm{o} / \mathrm{w}), \operatorname{logS}$, Polar surface area, Van der Waals volume and radius of gyration were calculated for a limited 50 compound training set of moleculesto determine and select compounds from our larger set with drug-like properties. A comparison of these descriptors suggested important properties for drug-like compound selection.

Fitting the experimental data-\$PRED_was chosen as dependent variable and the remaining descriptors (SlogP_VSA) and xsurf_CW) as independent variables of the database. A QSAR model was constructed choosing \$PRED as activity filed and the remaining descriptors as model fields. Regression analysis was performed for the training data set and RMSE and $\boldsymbol{r}^{2}$ values of the fit were reported. This fit model was saved as the QSAR model and used for the prediction of activities of compounds of test data set.

Cross-Validating the Model—The above QSAR fit was used for both model validation and cross validation. This validation procedure will evaluate the predicted activities and the residuals for the training set molecules. The predicted, residual and Z-score values were calculated for both model and cross validations.

Graphical Analysis-The predictive ability of the model was assessed using a correlation plot by plotting the predicted (\$PRED) values (X-axis) versus the predicted IC50 activities (Y-axis). This correlation plot was used to find out the outliers that have a Z-score beyond the range of 1.5 .

Estimation and validation of predicted activities of test set-The QSAR model fit obtained above was used to evaluate the predicted (\$PRED) values of 21 test set compounds. Regression analysis was performed for the test data set and RMSE and $\boldsymbol{r}^{2}$ values of the fit were reported.

Pruning the Descriptors-Pruning the descriptors is necessary to select the optimum set of molecules under consideration. 'QuaSAR-Contingency,' a statistical application in MOE was used to describe the best molecules in the data set. The results were analyzed using Principle Component Analysis (PCA) and the purpose of which is to reduce the dimensionality of set of molecular descriptors by linearly transforming the data or defying a property that would be important to drug design. A Three-dimensional scatter graphical plot was generated using the first three Principal Components (PCA1, PCA2 and PCA3).

## ADMET prediction $(27,28)$

Absorption, Distribution, Metabolism, Excretion, \& Toxicity (ADMET) properties of the 20 novel compounds were calculated using the preADMET online server (http:// preadmet.bmdrc.org/). The ADMET properties, human intestinal absorption, in vitro Caco-2 cell permeability, in vitro Maden Darby Canine Kidney (MDCK) cell permeability, in vitro plasma protein binding and in vivo BBB penetration were predicted using this program.

## Molecular Docking (29, 30)

Preparation of ABAD protein-The three-dimensional structure of ABAD was retrieved from Protein Data Bank (PDB:http://www.rcsb.org/pdb, PDB ID: 1SO8). To relieve any close contacts, the protein structure was loaded into MOE molecular modeling software and all the water molecules and hetero atoms were removed and polar hydrogen's were added. Protonation of 3D structure was done for all the atoms in implicit solvated environment (Born salvation model) at a specified temperature of $300 \mathrm{~K}, \mathrm{pH}$ of 7 and with a salt concentration value of 0.1 . A non-bonded cut off value of $8-10 \AA$ was applied to the Leonard-Jones terms. After the protonation, the complete structure was Energy minimized in MMFF94x force field at a gradient cut off value of 0.05 . Molecular dynamics simulations were carried out at a constant temperature of 300 deg K for a heat time of 10 pico seconds. The total simulations were carried out for a total period of 10 nano seconds. The time step was considered as 0.001 and the temperature relaxation time was set to 0.2 pico seconds. The position, velocity and acceleration were saved per every 0.5 pico seconds.

Prediction of Binding site for Ligands-The binding site for ABAD was predicted through PDBSum (http://www.ebi.ac.uk/pdbsum/). The protein structural information was analyzed at PDBSum and its link to Catalytic Site Atlas (CAS) was followed (http:// www.ebi.ac.uk/thornton-srv/databases/cgi-bin/CSA/CSA_Site-_Wrapper.pl?pdb=1so8). The catalytic site residues were Asn121, Ser155, Tyr168 and Lys172, which were based on the structure of Trihydroxynaphthalene Reductase (1YBV).

## Molecular Docking(31)

The ligand data base generated from the list of all novel ligand molecules was docked into the specified binding domain of the ABAD receptor. A total of 30 conformations were generated for each Ligand-Receptor complex and among them, the conformation with least docking score was considered for further analysis. The interaction of all ligand molecules in the binding domain cavity was analyzed from ligand interaction study of MOE.(32) The ligand-receptor complexes were analyzed by both London $\Delta \mathrm{G}$ free energy approximations and interaction energies, $\Delta \mathrm{E}$.

## Results and Discussion

ABAD inhibitor design
A total of 20 compounds were designed, and their capacity to inhibit $A \beta-A B A D$ interaction was predicted by using, quantitative structure activity relationship studies, preADME properties, docking studies. The preliminary SAR study indicated that the benzothiazole amine moiety is required for the inhibition of $A \beta-A B A D$ interaction.

Based on the frentizole SAR study and the previous report, we have developed novel benzothiazole ureas with a 30 -fold improvement in potency;(21) benzothiazole urea and frentizole analogs provide remarkable enhancements of permeation across biological membranes and of oral bioavailability. Based on frentizole and benzothiazole urea derivatives SAR studies, we have designed and synthesized novel small drug molecules as urea and frentizole phosphonate derivatives (Schemes $1 \&$ Scheme 2), which might have the capacity to cross the BBB and inhibit $\mathrm{A} \beta-\mathrm{ABAD}$ interaction. While for frentizole a wide range of analogues have been prepared and investigated, up to now there are no reports dealing with derivatives of urea and frentizole phosphonates or their related analogues. Compounds 1-12 and 16-20 are substituted benzothiazole amines bearing a methoxy- or fluoro and dimethyl/diethyl phosphonates, substituted aromatic/heterocyclic aldehydes, respectively. Compounds $\mathbf{1 3 - 1 5}$ are the urea/thiourea phosphonates moieties. The rational design of 1-20 had also been strongly supported by known phosphonate prodrugs and
docking studies showing that Tyr652 and Phe656 play a pivotal role in the ABAD drug binding, by promoting cation $-\pi, \pi-\pi$, and hydrophobic interactions with the basic nitrogen and aromatic rings of drugs. This might confer the capacity to cross the BBB and to inhibit $A \beta-A B A D$ interaction.

The major novelty of the present approach is the use of a benthiazole phosphonate moiety, which readily penetrates biological membranes such as the blood-brain barrier (BBB) and enters the target organ.(33-35) According, phosphate esters are frequently used as a prodrug strategy, especially for water insoluble compounds, since the phosphate group confers the following characteristic features to the xenobiotic: 1) decreases the adverse effects of the drugs, 2) help in readily crossing the blood-brain barrier (BBB) and enters the target organ, 3 ) increases water solubility and thereby enables delivery of the drug parenterally, 4) cleavage of the phosphonate carrier/drug entity in vivo provides a hydrophilic, negatively charged intermediate, which is "locked" in the brain or other organ and which provides significant and sustained delivery of the active drug species to the target organ, 5) as phosphonate moiety induces polar nature to the derivatives, bio transformation ( by Phase I enzymes) is not necessary for drug disposition, thereby reducing drug-drug interaction.(33, $34,36,37$ )

## Chemistry

To decrease adverse effects, facilitate passage across the BBB, and increase water solubility, phosphonate derivatives were synthesized for use as potential inhibitors of $A \beta-A B A D$ interactions. A series of analogs were synthesized with variations at the aromatic rings and their linking group. As shown in Scheme 1, various aromatic/heterocyclic aldehydes (2) were individually mixed with $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}(5 \mathrm{~mol} \%)$ in anhydrous toluene $(5 \mathrm{~mL})$. The mixture was stirred magnetically for $10-15$ minutes, after which time substituted benzothiazole amines (1) and dimethyl/diethyl phosphite (3) in anhydrous toluene ( 5 mL ) were added dropwise with stirring at room temperature for 30 minutes. The reaction mixture was refluxed with stirring for 5-12 hours. Herein, we report that $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$ is an extremely efficient catalyst for the formation of benthiazole aminophosphonates by a one pot, threecomponent reaction of aromatic/heterocyclic aldehydes, substituted benzothiazole amines, and a dimethyl/diethyl phosphite in dry toluene (Scheme 1). The synthetic route utilized to make our target "ABAD inhibitors" compounds $\mathbf{4 a - 4 n}$ is shown in Scheme 1.

To determine the best experimental conditions, the reaction of 6-fluorobenthiazole amine, methyl 5-formyl-2-hydroxybenzoate, and DMP was considered a model (spectral data has been provided in the supporting information section). The progress of the reaction was monitored by TLC (ethyl acetate:hexane $=1: 4$ ) analysis. After completion of the reaction, the solvent was removed under reduced pressure. The reaction mixture was extracted with EtOAc $(3 \times 10 \mathrm{~mL})$. The combined EtOAc extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure to afford a white solid, which on passing through a column of silica gel and on elution with EtOAc-hexane (80:20) afforded compound (4a). The best results were obtained in the presence of $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$ at reflux for 6 h , affording the desired methyl 5-((dimethoxyphosphoryl) ((6-fluorobenzo[d]thiazol-2-yl) amino)methyl)-2hydroxybenzoate in $81 \%$ yield (4a). The remaining reactions were carried out following this general procedure. On each occasion, the spectral data (IR, NMR, and MS) of prepared known compounds were identical with those reported in the literature. We have synthesized few representative examples using DMP, DEP, methoxy, fluoro substituted amines, and different aromatic/heterocyclic aldehydes, respectively.

The alternative synthetic route for the preparation of, urea/thiourea phosphonates derivatives uses a mixture of urea/thiourea and methyl 5-formyl-2-hydroxybenzoate in ethanol under reflux to form an imine intermediate, which is further reacted with dialkyl/aryl phosphates to
generate the phosphonate derivatives (Scheme 2).(38) The remaining title compounds will be synthesized for further study $\mathrm{A} \beta-\mathrm{ABAD}$ interaction to evaluate the inhibitory activity.

A total of 20 compounds have been designed and tested as of date, and their capacity to inhibit $A \beta-A B A D$ interaction was predicted by using, QSAR and 3D pharmacophore methods, preADME properties, docking studies and drug-like analysis. A preliminary SAR through our docking studies of these compounds indicated that the benzothiazole amine moiety is required for inhibition of $\mathrm{A} \beta-\mathrm{ABAD}$ interaction. Urea/thiourea compounds 11, 12, $\mathbf{1 3}, 14,15$ and 16 are not quite as effective because their molecular weight is too big. Substitutions on the benzothiazole ring and phenyl rings dramatically affected potency. Small electron-withdrawing groups were preferred at the benzothiazole ring with F particularly favored. Further, compounds with a hydroxyl group at the para position of the phenyl ring are predicted to more potent.

## Molecular dynamics and 2D QSAR study of ligand molecules

The stabilized conformations obtained at the end of the molecular dynamics simulations were used to construct the molecular compound database. For the entire database, molecular QSAR descriptors were calculated and graphs were plotted to analyze whether they are in the optimum range or not. The compounds $\mathbf{1 1}, 12,13,14,15$ and 16 are violating the drug like properties especially with respect to molecular weight (Figures. 1 to Fig. 13). The remaining compounds all show the satisfactory values, which make them behave as good drugs.

## 2D QSAR model descriptors

| SlogP_VSA(i,p) | 2D Subdivided surface areas. Each descriptor in a series is defined to be the sum of the $v_{i}$ over all <br> atoms $i$ such that $p_{i}$ is in a specified range $(a, b)$ |
| :--- | :--- |
| vsurf_CW(i=1,8) | 3D volume capacity descriptors |
| SlogP | Log of the octanol/water partition coefficient |
| Density | Molecular mass density |
| MW/Vol (VDW) | Weight divided by vdw_vol (amu/Å3) |
| SMR | Molecular refractivity, (including implicit hydrogens) |
| MW | Molecular weight (including implicit hydrogens) in atomic mass units |
| TPSA | Total polar surface area |
| $\operatorname{logs}$ | Log of the aqueous solubility (mol/L) |

The correlation plot generated form the regression analysis showed a linear relationship among the training set of 19 compounds (Fig. 14). The model exhibited excellent linearity $\left[\$ P R E S=0.9\left(\mathrm{IC}_{50}\right)+9.0\right]$, with a $\mathrm{RMSE}=0.9, \mathrm{R}^{2}=0.97$ and a cross-validated $\mathrm{R}^{2}=0.61$. The reliability of the QSAR model was further established by applying this model on an independent set of 21 test compounds and its predictive ability was evaluated. The correlation plot of the test compounds was reasonable and expected based on a limit set of compounds in the training set. The resultant correlation plot of regression analysis showed a linear relationship for the final test data set $\left[\$ P R E S ~=0.6\left(\mathrm{IC}_{50}\right)+54.0\right]$, with a $\mathrm{RMSE}=0.8$, $\mathrm{R}^{2}=0.6$. (Fig 15). As new compound data is added to the training set, we would expect the accuracy for prediction of our QSAR model to improve (Figures $14 \& 15$ ).

## Principal Component Analysis

A principal component analysis using the QSAR descriptors showed that the first three PCA eigenvectors included $63 \%$ of the variance. All the data values found to lie in the range of
-3 to +3 , each spot in the plot represents a molecule and color coded by $\mathrm{IC}_{50}$ activity. Most interestingly the top most active compounds in our data set, as shown in magenta are isolated at PCA1 $=2$ and PCA $=2$ (Figure 16). This could provide an addition criterion for compound selection.

## 3D Pharmacophore model of ligand molecules (Figure 17)

ADME predictions $(39,40)$ —ADME properties are important conditions and major parts of pharmacokinetics. Whatever the compound that is going to be a drug should have the perfect ADME properties, and then only it will be approved as a drug in clinical tests. The ADME predictions of present 20 compounds are showing satisfactory results. Among the 20 , the compounds $\mathbf{1}, \mathbf{2}, \mathbf{3}, \mathbf{4}, \mathbf{5}, \mathbf{6}, \mathbf{7}, \mathbf{8}, \mathbf{9}, \mathbf{1 0}, \mathbf{1 1}, \mathbf{1 4}, \mathbf{1 7}, \mathbf{1 8}, 19$ and $\mathbf{2 0}$ are showing well intestinal absorption and the compounds 12, 13, 15 and 16 are showing moderate absorption. All of them are showing middle permeability for in vitro Caco- 2 cells and low permeability for in vitro MDCK cells. in vivo blood-brain barrier penetration capacity was predicted to be having middle absorption to CNS (Central Nervous System) for the compounds 1, 2, 3, 4, 5, $\mathbf{6}, \mathbf{7}, \mathbf{8}, \mathbf{9}, 10,17,18,19$ and $\mathbf{2 0}$ where as low absorption to CNS was observed for the compounds $11,12,13,14,15$ and 16 . BBB penetration is a crucial pharmacokinetic property because CNS-active compounds must pass across it and CNS-inactive compounds must not pass across it to avoid CNS side effects. Generally the degree of plasma protein binding of any drug influences not only the drug action but also its disposition and efficacy. Usually, the drug that is unbound to plasma proteins will be available for diffusion or transport across cell membranes and finally interacting with the target. Here in with respect to ADME, the percent of drug bound with plasma proteins was predicted and the compounds $\mathbf{2 , 5 , 6 , 7 , 8}$, $\mathbf{9}, \mathbf{1 0}, \mathbf{1 4}, \mathbf{1 7}, \mathbf{1 8}, 19$ and 20 are predicted to bound strongly and the compounds $\mathbf{1}, \mathbf{3}, \mathbf{4}, \mathbf{1 1}$, $12,13,15$ and 16 are predicted to bound weakly to plasma proteins. The predicted ADME properties and their values are presented in the Table 1.

## Molecular Docking

Molecular docking of the 20 novel compounds against ABAD active site revealed that all of them with better $\mathrm{IC}_{50}$ 's showed good interaction with good docking scores dominated by hydrogen bonding and phosphonate salt-bridge formation with the binding domain of ABAD. Hydrophobic interactions also were observed to play a contributing role. The least and highest docking scores were found with the compounds 14 and 16, respectively, but both of them are not showing satisfactory QSAR descriptions and violating drug like properties. All of the compounds finally had better interactions (supporting information document). The residues Ser 155 and Val 156 of ABAD active sites are found to be playing a predominant role in interaction with all the compounds. Lys 172 is showing arene interactions, which is because of hydrophobicity of the ring structures with the compounds 4, 6, 11 and 19, which in turn explains strong interaction with the ABAD (Figure 18, 19 \& Table 3 in supplemental file). The docking scores and the bonding information with ABAD residues are tabulated and shown in Table 2.

## Conclusions and Future Directions

From the docking, 2D QSAR and pharmacophore studies, it is concluded that most of these compounds are reasonable inhibitors of ABAD. The exceptions are compounds 11, 12, 13, 14, 15 and 16 as they are not showing proper drug-like QSAR and satisfactory ADMET properties. The ligand - complexes generated from molecular docking process suggest that the molecules are good ABAD inhibitors as they are showing good binding affinity with the ABAD receptor. This molecular docking study will not be used to screen the molecules that are having inhibitory activity against $A B A D$, since it is an inefficient. However, our demonstration of 2D QSAR and 3D pharmacophore models to predict activity should be
very useful to screen a huge number of new compounds. We hope to combine these models to give us greater accuracy in the prediction of active candidates. We would like to point out that only the most active compounds that we have synthesized are going to be used in upcoming studies to discover even more potent analogues for the treatment of the Alzheimer disease.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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## Abbreviations

| ABAD | Amyloid binding alcohol dehydrogenase |
| :--- | :--- |
| A $\boldsymbol{\beta}$ | amyloid beta |
| AD | Alzheimer's disease |
| QSAR | quantitative structure-activity relationship |
| CoMFA | comparative molecular field analysis |
| ADMET | absorption, distribution, metabolism and excretion and toxicity |
| FDA | Food and Drug Administration |
| BBB | blood brain barrier |
| Caco2 | human colon adenocarcinoma |
| MDCK | Madin-Darby canine kidney |
| SAR | structure-activity relationship |
| DMP | dimethyl phosphite |
| DEP | diethyl phosphite |
| SlogP | Log of the octanol/water partition coefficient |
| CNS | central nervous system |
| MOE | molecular operating environment |
| RMSE | Root mean square error |
| PCA | Principle Component Analysis |
| PDB | Protein Data Bank |
| MMFF94x | The Merck Molecular Force Field |

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Figure 1.
Plot showing the Log of octanol/water partition coefficient of test compounds including implicit hydrogens $(\mathbf{S l o g} \mathbf{P})$. This property is an atomic contribution model that calculates $\log P$ from the given structure; i.e., the correct protonation state.


Figure 2.
Plot showing the Molecular mass density of test compounds (Weight divided by vdw_vol (amu/Å3)).


Figure 3.
Plot showing the Molecular refractivity of test compounds (including implicit hydrogens). This property is an atomic contribution model that assumes the correct protonation state.


Figure 4.
Plot showing the Molecular weight (including implicit hydrogens) of test compounds in atomic mass units.


Figure 5.
Plot showing the Sum of the atomic polarizabilities of test compounds including implicit hydrogens.


Figure 6.
Plot showing the $\log$ of the octanol/water partition coefficient $(\log \mathbf{P}(\mathbf{o} / \mathbf{w}))$ of test compounds including implicit hydrogens. This property is calculated from a linear atom type model with $\mathrm{r} 2=0.931$ and RMSE $=0.393$.


Figure 7.
Plot showing the log of the aqueous solubility (logs) (mol/L) of test compounds. This property is calculated from an atom contribution linear atom type model.


Figure 8.
Plot showing the polar surface area ( $\AA 2$ ) of test compounds calculated using group contributions to approximate the polar surface area.


Figure 9.
Plot showing the van der Waals volume ( $\AA$ 3) of test compounds.


Figure 10.
Plot showing the Lipinski rule drug like (blue) and violation (black) of test compounds. The blue colored line with a value of 1 shows the drug likeliness and value of beyond 0 indicates the Lipinski rule violation. The compounds $11,12,13,14,15$ and 16 are showing violations of 2 each.


Figure 11.
Plot showing the radius of gyration of test compounds.


Figure 12.
Plot showing the count of hydrogen bonds donors in each test compound. All of them showing a count of less than 5 , which is one of the desired properties of Lipinski rule.


Figure 13.
Plot showing the count of hydrogen bonds acceptors in each test compound. All of them showing a count of less than 10 , except the compounds 12,13 and 16 , which means that they are violating the desired properties of Lipinski rule.


Figure 14.
Linear correlation graph comparing a independent test set of measured IC50's with predicted values based on the calculated 2D QSAR model. The linearity of the test model is shown with the values of the error (RMSE) and correlation factor $\left(\mathrm{R}^{2}\right)$.


Figure 15.
Linear correlation graph comparing a independent test set of measured IC50's with predicted values based on the calculated 2D QSAR model. The linearity of the test model is shown with the values of the error (RMSE) and correlation factor $\left(\mathrm{R}^{2}\right)$.


Figure 16.
PCA plot of the complete set of ABAD compounds. The first three eigenvectors are shown (PCA1, PCA2 and PCA3), which constituted $63 \%$ of the variance. The spheres indicate the position of each of the 45 compounds. Based on SlogP a clear separation is observed. Those compounds with calculated $\operatorname{Slog} P(<2)$, are shown as magenta spheres) are clustered about $\mathrm{PCA} 1=1, \mathrm{PCA} 2=2, \mathrm{PCA} 3=2$ ). Those spheres in red are calculated $>4$, and green designate between these extremes.


Figure 17.
Showing the resulting pharmacophore model (shown as volume sheres) based on the activities (IC50) of 20 compounds in the training set. The various pharmacophore sites are labeled as to the features or properties (H-bond donor, H-bond acceptor, hydrophilic/ aromatic and carboxyl acceptor/anion donor). The 6 top most active ABAD inhibitors from the test set are overlayed with the pharmacophore model.


Figure 18.
Binding site for active $\mathrm{IC}_{50}$ compound $1 \& 2$. Compound 1 (left) binding affinity is contributed to 4 H -bonds to Ser 155 Ala 156 and Arg 116, as shown by the dotted yellow interactions. Compound 2 (right) binding affinity is contributed to $\Pi$-stacking interactions between Lys 172, Arg 116 and a H-bond to Gln 115.


Figure 19.
2D Ligand interaction maps of compound 1 (left) compound 2 (right) binding to ABAD .

|  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Compound Entry | $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ - CHO | $\mathrm{R}_{3}$ | Product |
| 1 | OMe | $5-\mathrm{COOCH}_{3}, 4-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{3}$ | Methyl | 4 a |
| 2 | F | $5-\mathrm{COOCH}_{3}, 4-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{3}$ | Methyl | 4b |
| 3 | OMe | $5-\mathrm{COOCH}_{3}, 4-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{3}$ | Ethyl | 4 c |
| 4 | OMe | $5-\mathrm{COOCH}_{3}, 4-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{3}$ | Ethyl | 4d |
| 5 | OMe | $4-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Methyl | 4 e |
| 6 | OMe | $2-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Methyl | 4f |
| 7 | OMe | $2-\mathrm{S} . \mathrm{C}_{4} \mathrm{H}_{3}$ | Methyl | 4 g |
| 8 | OMe | $2-\mathrm{NH} . \mathrm{C}_{4} \mathrm{H}_{3}$ | Methyl | 4h |
| 9 | OMe | $4-\mathrm{N} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Ethyl | 4i |
| 10 | OMe | $2-\mathrm{O} . \mathrm{C}_{4} \mathrm{H}_{3}$ | Methyl | 4 |
| 17 | OMe | $4-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Ethyl | 4k |
| 18 | OMe | $4-\mathrm{F} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Methyl | 41 |
| 19 | OMe | $4-\mathrm{F} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Ethyl | 4 m |
| 20 | OMe | 5-Br, $2-\mathrm{OH} . \mathrm{C}_{6} \mathrm{H}_{4}$ | Ethyl | 4 n |

Scheme 1.
Representative classes of compounds containing the N-C-P scaffold and selected molecular targets of ABAD
(

## Scheme 2.

Alternative synthetic route of urea/thiourea phosphonates derivatives

## Table 1

ADME properties predicted for 20 novel compounds.

| Compound | ${ }^{\boldsymbol{a}}$ Human intestinal absorption (\%) | $b_{\text {in }}$ vitro Caco-2 cell permeability ( $\mathrm{nm} / \mathrm{sec}$ ) | $c_{i n}$ vitro MDCK cell permeability ( $\mathrm{nm} / \mathrm{sec}$ ) | $d_{\text {in }}$ vitro plasma protein binding (\%) | $e_{\text {in }}$ vivo blood-brain barrier penetration (C.brain/C.blood) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 90.635 | 21.557 | 0.2484 | 83.69 | 0.2324 |
| 2 | 92.359 | 21.480 | 0.4673 | 91.89 | 0.4803 |
| 3 | 92.583 | 21.655 | 0.0732 | 88.53 | 0.4080 |
| 4 | 92.583 | 21.655 | 0.0732 | 88.12 | 0.4082 |
| 5 | 94.664 | 21.654 | 4.4672 | 97.01 | 0.2629 |
| 6 | 94.663 | 21.654 | 4.2637 | 100.00 | 0.2434 |
| 7 | 95.914 | 21.714 | 14.2191 | 100.00 | 0.3311 |
| 8 | 89.279 | 21.694 | 25.7699 | 100.00 | 0.1876 |
| 9 | 96.488 | 21.710 | 6.4507 | 100.00 | 0.1989 |
| 10 | 94.924 | 21.713 | 25.2219 | 100.00 | 0.3009 |
| 11 | 79.779 | 11.035 | 0.0436 | $87.56$ | 0.0110 |
| 12 | 49.368 | 17.055 | 0.0434 | 78.52 | 0.0258 |
| 13 | 21.008 | 19.226 | 0.0434 | 60.27 | 0.0306 |
| 14 | 83.536 | 16.043 | 0.0436 | 90.71 | 0.0126 |
| 15 | 62.784 | 19.787 | 0.0447 | 81.16 | 0.0139 |
| 16 | 26.739 | 20.563 | 0.0434 | 66.55 | 0.0325 |
| 17 | 95.453 | 21.697 | 0.6218 | 96.40 | 0.3688 |
| 18 | 97.777 | 21.719 | 4.7661 | 100.00 | 0.6795 |
| 19 | 97.876 | 21.723 | 0.8251 | 98.20 | 0.9963 |
| 20 | 95.990 | 19.217 | 0.0218 | 100.00 | 0.3767 |

${ }^{a}$ Human intestinal absorption is the sum of bioavailability and absorption evaluated from ratio of excretion or cumulative excretion in urine, bile and feces. The value between $0 \%-20 \%$ indicates poor absorption, $20 \%-70 \%$ moderate absorption and $70 \%-100 \%$ well absorption.
${ }^{b}$ Caco- 2 cells are derived from human colon adenocarcinoma and possess multiple drug transport pathways through the intestinal epithelium. The value $<4$ indicates low permeability, $4-70$ middle permeability and $>70$ high permeability.
${ }^{c}$ MDCK cell system may use as good tool for rapid permeability screening. The value $<25$ indicates low permeability, 25-500 middle permeability and $>500$ high permeability
${ }^{d}$ The percent of drug binds to plasma protein. The value $<90 \%$ indicates weak binding and $>90 \%$ indicates strong binding to plasma proteins.

Table 2
Molecualr docking interaction of 20 novel compounds against ABAD active site.

| $a^{\text {Compound }}$ | $b$ Docking Score | ${ }^{c}$ No. Hydrogen Bonds | $d^{\text {Interacting Residues of ABAD }}$ |
| :---: | :---: | :---: | :---: |
| 1 | -13.203 | 5 | Gln 115, Asn 121, Ser 155, Ala 158, Lys 172 |
| 2 | -11.741 | 2 | Gln 115, Asp 119 |
| 3 | -10.757 | 3 | Phe 114, Gln 115, Ser 155 |
| 4 | -11.160 | Arene interaction | Lys 172 |
| 5 | -10.927 | 2 | Ser 155, Val 156 |
| 6 | -12.043 | 4 | Asn 121, Ser 155, Ser 155, Val 156 |
|  |  | Arene interaction | Lys 172 |
| 7 | -11.906 | 5 | Asn 121, Ser 155, Ser 155, Val 156, Lys 172 |
| 8 | -11.423 | 4 | Ser 155, Val 156, Lys 172, Lys 172 |
| 9 | -10.727 | 4 | Asn 121, Ser 155, Ser 155, Val 156 |
| 10 | -11.557 | 2 | Ser 155, Val 156 |
| 11 | -11.616 | 2 | Gly 95, Ser 155 |
|  |  | Arene interaction | Lys 172 |
| 12 | -10.358 | 1 | Glu 160 |
| 13 | -14.813 | 2 | Lys 172, Glu 160 |
| 14 | -11.626 | -- | -- |
| 15 | -11.785 | 5 | Gln 115, Asp 119, Asn 121, Ser 155, Lys 172 |
| 16 | -10.190 | 2 | Gln 115, Ser 155 |
| 17 | -10.623 | 1 | Gln 115 |
| 18 | -10.695 | 1 | Lys 172 |
| 19 | -10.990 | 2 | Ser 155, Val 156 |
|  |  | Arene interaction | Lys 172 |
| 20 | -12.774 | 3 | Asp 119, Ser 155, Lys 172 |

[^1]
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    Supporting Information Available: Experimental details and spectroscopic data of synthesized compounds and Interaction of remaining 18 novel inhibitors with ABAD active site. Two-dimensional linear representations of ligand-receptor complexes and threedimensional graphical representations of ligand-receptor complexes were presented.

[^1]:    ${ }^{a}$ The novel ABAD inhibitors.
    ${ }^{b}$ Docking scores generated during MOE docking between the novel leads and ABAD binding domain.
    ${ }^{c}$ Number of hydrogen bonds formed between the ABAD binding domain and the novel leads.
    $d_{\text {The interacting active site residues of } A B A D ~ p r o t e i n ~ w i t h ~ t h e ~ n o v e l ~ i n h i b i t o r s ~ i n ~ t h e ~ l i g a n d-r e c e p t o r ~ c o m p l e x ~}^{\text {a }}$

