

Soil O₂ controls denitrification rates and N₂O yield in a riparian wetland

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[1] Wetland soil oxygen (O₂) is rarely measured, which limits our understanding of a key regulator of nitrogen loss through denitrification. We asked: (1) How does soil [O₂] vary in riparian wetlands? (2) How does this [O₂] variation affect denitrification rates and end products? and (3) How does [O₂] variation and previous exposure to O₂ affect trace gas fluxes? We collected a continuous seven-month record of [O₂] dynamics in a “wet” and “dry” riparian zone. In April 2009, soil [O₂] ranged from 0 to 13% and consistently increased with increasing distance from the stream. [O₂] gradually declined in all sensors until all sensors went anoxic in early September 2009. In mid-fall, a dropping water table increased soil [O₂] to 15–20% within a 2–3 day period. We measured denitrification using the Nitrogen-Free Air Recirculation Method (N-FARM), a direct measurement of N₂ production against a helium background. Denitrification rates were significantly higher in the wetter areas, which correlated to lower O₂ conditions. Denitrification rates in the drier areas correlated with [O₂] in the early spring and summer, but significantly decreased in late summer despite decreasing O₂ concentrations. Increasing [O₂] significantly increased core N₂O production, and therefore may be an important control on nitrous oxide yield. Field N₂O fluxes, however, were highly variable, ranging from 0 to 800 ug N m⁻² hr⁻¹ with no differences between the wet and dry sites. Future research should focus on understanding the biotic and abiotic controls on O₂ dynamics, and O₂ dynamics should be included in models of soil N cycling and trace gas fluxes.

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1. Introduction

[2] Denitrification is the microbial metabolism that oxidizes organic carbon while reducing nitrate (NO₃⁻) under hypoxic or anaerobic conditions. Since the end product of the reaction is the dominant atmospheric gas, N₂, the process is notoriously difficult to measure despite decades of methodological development [Groffman *et al.*, 2006]. Therefore, denitrification remains one of the greatest uncertainties in nitrogen mass balance studies at ecosystem [Steinheimer *et al.*, 1998], regional [van Breeman *et al.*, 2002] and global [Galloway *et al.*, 2008; Gruber and Galloway, 2008] scales.

[3] Point-scale measurements of denitrification that have been made in aquatic and terrestrial ecosystems indicate that a variety of ecosystems denitrify at highly variable rates. However, constrained mass balance studies suggest that approximately 50% of the 270 Tg N added annually to terrestrial ecosystems crosses the threshold into aquatic ecosystems [Seitzinger *et al.*, 2006]. This points to the importance of aquatic-terrestrial interfaces for removing N;

however, the controls and function of these dynamic transitional zones remains poorly characterized. Riparian zone wetlands are one class of these transitional areas, and are often considered to be “hotspots” for biogeochemical processes, including denitrification [Vidon, 2010; Mayer *et al.*, 2007]. As such, they are important zones for intercepting NO₃⁻ before it moves into aquatic ecosystems where it can cause downstream eutrophication and harmful algal blooms [Howarth *et al.*, 2011; Paerl *et al.*, 2002]. Riparian zones are defined in part by their high degree of spatial and temporal variation in denitrification [Harms and Grimm, 2008] and greenhouse gas (GHG) fluxes [DeSimone *et al.*, 2010]. Understanding how this temporal and spatial heterogeneity in biogeochemical functions translates into patterns of nutrient removal remains a key challenge to modeling and predicting biogeochemical dynamics linked aquatic and terrestrial ecosystems [Grimm *et al.*, 2003; McClain *et al.*, 2003].

[4] During denitrification, a fraction of the NO₃⁻ is not fully reduced and escapes as nitrous oxide, N₂O. N₂O is the most potent of the three biogenic greenhouse gases (including methane, CH₄, and carbon dioxide, CO₂) with a radiative forcing ~300× that of CO₂. Atmospheric N₂O has been increasing due to increased reactive N (e.g., NO₃⁻) from agricultural activities [Galloway *et al.*, 2004; Davidson, 2009]. Higher NO₃⁻ concentrations, low pH and the presence of O₂ tend to promote increased N₂O fluxes [Firestone *et al.*, 1980; Bollmann and Conrad, 1998; Robertson and Groffman,

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2007]. The ability to reduce N₂O is encoded by the *nosZ* gene, which is more sensitive to O₂ than other denitrification genes [Richardson *et al.*, 2009]. The N₂O mole fraction (N₂O / [N₂O + N₂]), also called the N₂O yield, is used to express the proportion of gaseous loss to N₂O via denitrification. N₂O yields are generally low in wetlands (<0.1), but can be considerably higher in upland soils (≥0.4) [Beaulieu *et al.*, 2011; Schlesinger, 2009]. There is great interest in and uncertainty about N₂O yields in transitional zones such as riparian buffers [Groffman *et al.*, 1992; Reay *et al.*, 2009]. N₂O is also produced by nitrification, a microbial metabolism that converts ammonium (NH₄⁺) to NO₃⁻. The complex interactions between producing and consuming processes make field-level N₂O fluxes notoriously difficult to interpret. Field-level N₂O fluxes are tied to water-filled pore space (WFPS) [Pathak and Nedwell, 2001], which is also linked to O₂ dynamics.

[5] In addition to N₂O, microbes are responsible for the production and consumption of methane (CH₄) and carbon dioxide (CO₂). CH₄ is produced through methanogenesis, another form of anaerobic microbial metabolism. Methanogenesis tends to occur once all other electron donors (e.g., O₂, nitrate) are exhausted. Methanogenesis rates can be enhanced by labile substrates [Glatzel *et al.*, 2004; Yavitt and Lang, 1990] and warmer seasons [Gleason *et al.*, 2009; Yavitt *et al.*, 2000]. Ecosystem-level controls on fluxes include water table fluctuations [Altor and Mitsch, 2006; Gleason *et al.*, 2009] and plant productivity and vascular transport [Whiting and Chanton, 1993]. Stable water levels promote larger CH₄ fluxes over variable or “pulsing” hydrology [Altor and Mitsch, 2006], suggesting a potential connection to O₂ and more oxidizing conditions under pulsing hydrology. However, work in tropical rain forest soils has suggested that CH₄ fluxes can remain high even at 9–19% soil O₂ [Teh *et al.*, 2005].

[6] While O₂ is generally considered the dominant proximal control of the microbial metabolisms that perform denitrification and produce and consume the three biogenic GHGs, O₂ is seldom measured in terrestrial ecosystems [Silver *et al.*, 1999]. Consequently, our understanding of how soil [O₂] varies is limited. This restricts our understanding of how soil O₂ variation affects ecosystem processes including GHG production [Whalen, 2005], nutrient dynamics [Crawford, 1992], and redox [Pett-Ridge and Firestone, 2005]. The direct effects of O₂ on denitrification have been tested in laboratory settings, often on cultures isolated from soils or from soil “slurries” [Firestone *et al.*, 1980; Fazzolari *et al.*, 1998; Parkin and Tiedje, 1984; Sexstone *et al.*, 1985]; however, there have been few studies that have evaluated O₂ controls on intact core denitrification (but see Parkin and Tiedje [1984]) or in situ GHG fluxes.

[7] In this study we asked: 1) How does soil [O₂] vary in riparian wetlands? 2) How does this [O₂] variation affect denitrification rates and end products? and 3) How does seasonal variation in [O₂] affect greenhouse gas fluxes? We hypothesized that soil [O₂] would be spatially and temporally dynamic, with concentrations depending on proximity to the stream, which influences water table and also depend on season, which controls plant and microbial consumption of O₂ and soil moisture content. We further hypothesized that areas and periods of decreased soil [O₂] would have

increased denitrification rates and CH₄ production, but expected that that N₂O fluxes would be highest in areas with the most variable O₂ levels. To answer these questions, we tested five hypotheses: 1) Soil O₂ was dynamic, with [O₂] dependent on proximity to the stream and water table; 2) Soil [O₂] would vary seasonally; 3) Areas of decreased soil [O₂] would have increased denitrification rates; 4) Decreasing soil [O₂] would increase N₂O fluxes; and 5) Biogenic greenhouse gases, including CH₄ and N₂O, would increase under decreased [O₂] conditions. Though we already understand how O₂ affects denitrification from microcosm studies, this study expands our understanding to how field-level fluctuations in soil [O₂] and subsequent differential exposure of the in situ microbial community to soil O₂ can control denitrification rates and end products. More practically, previous studies have shown that O₂ effects on denitrification are strong and immediate, suggesting that continuous measurements of in situ [O₂] levels may be a useful tool temporal scaling of point measurements.

2. Materials and Methods

2.1. Study Sites

[8] Our study site was located on the property of the Cary Institute of Ecosystem Studies in Millbrook, NY, (USA) adjacent to Gifford Stream, a first order, groundwater-fed tributary of the East Branch of Wappingers Creek. Vegetation in the area is predominantly mature sugar maples (*Acer saccharum*) mixed with younger ash (*Fraxinus sp.*) and birch (*Betula lenta*). The nearby fields were in agricultural production through the middle of the 20th century, but have been managed as mowed fields in more recent decades. The areas draining into our research site are gravelly loam soils (Hoosic series) derived from schist [Soil Survey Staff, 2006].

[9] Our sampling site was a 2 m × 4 m plot of riparian wetland immediately adjacent to the stream (Figure 1a). The site was characterized by wetland plants including skunk cabbage (*Symplocarpus foetidus*) and jewelweed (*Impatiens capensis*). Soils were characterized by a 10-cm organic horizon, underlain by outwash from the adjacent stream (Fluvaquentic Humaquept, Soil Survey Staff [2006]). These soils were water-saturated from early winter through spring, but dried up considerably during the growing season. The site was divided into the “wet” zone, closer to the stream, which was more fully inundated though it did not have standing water, and the “dry” zone, which had lower soil moisture, but was only 1-m from the “wet” area.

[10] Five intact soil cores (2.4 cm diameter × 8 cm length) were collected from both areas monthly and immediately taken to the lab for analysis. Samples were taken by driving a piece of sharpened PVC pipe, split down the side and then taped together, into the soil. After removing the pipe from the soil, the tape was removed and intact cores could be accessed.

2.2. Soil O₂ and Moisture Measurements

[11] Four Apogee diffusion-head soil O₂ sensors (SO-100 series, Apogee Instruments, Logan UT; accuracy <0.02%/day, repeatability ±0.001% O₂) per plot were buried with the diffusion heads at 6-cm. Sensors were spaced approximately 0.5 m apart from the upper portion of the dry zone (sensor #1)

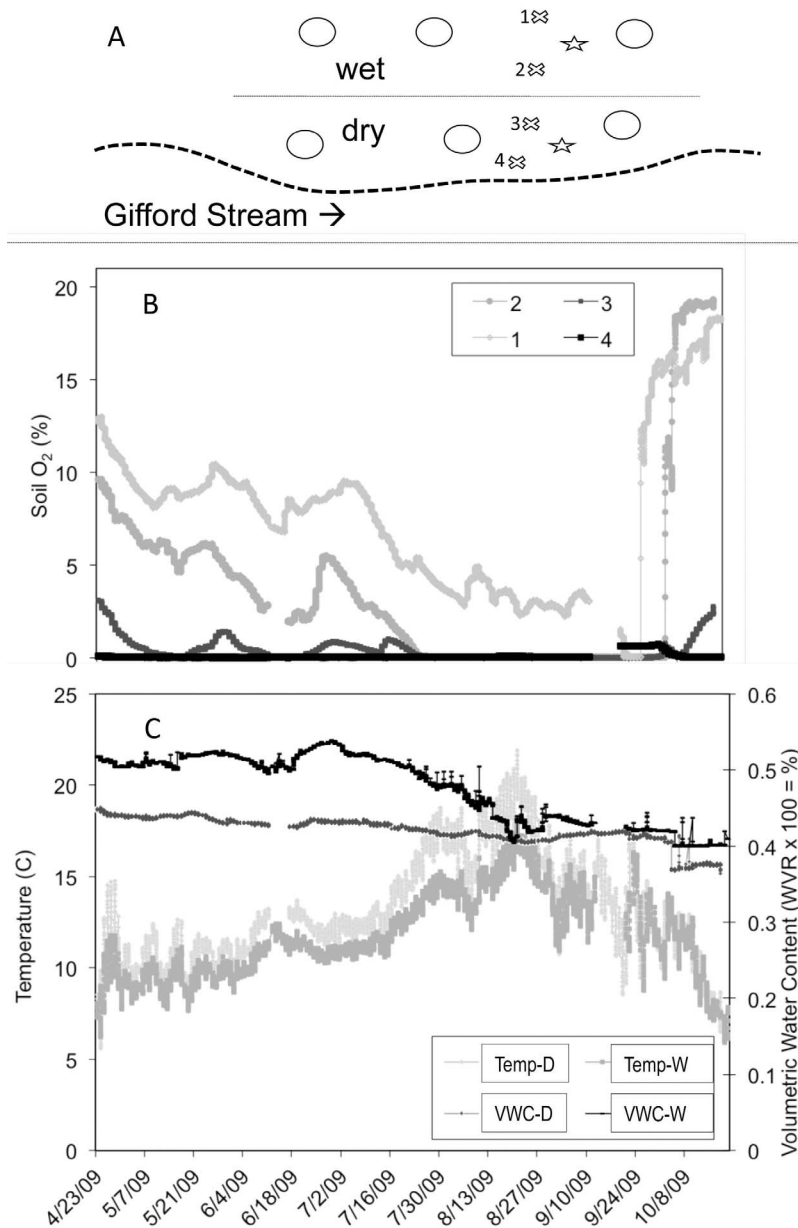


Figure 1. (a) Plot layout for the Gifford Riparian Wetland, which contained sensors for (b) soil oxygen, soil moisture and (c) temperature. Grey circles represent the location of static gas flux chambers. Numbered “x” symbols represent the location of O₂ sensors, with seasonal O₂ concentrations for the sensors in Figure 1b. Stars represent the soil moisture and temperature sensors with corresponding data in Figure 1c. “Temp-D” and “VWC-D” refer to the temperature and volumetric water content sensors from the dry zone; the same abbreviations followed by “-W” applies to the wet zone sensors.

to the lower portion of the wet zone (sensor #4) (Figure 1a). O₂ sensors #1 and 4 also recorded soil temperature. The sensors were controlled by a Campbell Scientific Data logger (CR-800; Campbell Scientific, Logan UT), which also controlled two Campbell CS616 water content reflectometers to measure soil moisture in the wet and dry areas (Figure 1c). Sensors were calibrated prior to deployment, and were set to collect soil [O₂] and volumetric water content (VWC) every hour.

2.3. Measuring Denitrification With the Nitrogen-Free Air Recirculation Method (N-FARM)

[12] Our N-FARM flow-through core measurement system is described by *Burgin et al.* [2010] and is based on those built and described by others [*Butterbach-Bahl et al.*, 2002; *Swerts et al.*, 1995]. Cores were encased in glass bottles with gas-tight lids connected to a gas-tight flow injection system built from Swagelok connections (Swagelok, Crawford Fitting Co., Solon, OH) inline with a Shimadzu GC8A gas

Table 1. Average N₂ and N₂O Fluxes Measured From Cores in Different Months at Different O₂ Concentrations

Month in 2009	Percent O ₂	N ₂ Flux $\mu\text{g N m}^{-2} \text{hr}^{-1}$		N ₂ O Flux $\mu\text{g N m}^{-2} \text{hr}^{-1}$	
		Dry	Wet	Dry	Wet
May	0	1657 ± 201	2053 ± 216^a	6.3 ± 5.6	6.0 ± 0.6
	5	907 ± 131	1076 ± 132	28.5 ± 14.1	11.6 ± 2.9
June	0	2331 ± 168	2644 ± 146	56.8 ± 17.8	-30.6 ± 23.6
	5	1972 ± 455	1486 ± 205	31.0 ± 6.8	35.5 ± 6.9
July	10	1454 ± 437	1233 ± 272	31.3 ± 13.1	48.5 ± 16.1
	0	2725 ± 195	2813 ± 289	5.5 ± 3.2	3.5 ± 2.1
	5	1410 ± 245	1662 ± 160	45.0 ± 15.1	20.5 ± 1.7
August	0	1305 ± 204	2489 ± 344	0.0 ± 0.0	32.5 ± 29.5
	5	926 ± 134	1577 ± 177	51.3 ± 38.3	67.3 ± 31.0
September	0	926 ± 207	2377 ± 149	0.0 ± 0.0	0.0 ± 0.0
	5	1002 ± 194	1268 ± 102	137.5 ± 24.4	13.8 ± 11.8
October	10	470 ± 88	857 ± 145	58.0 ± 10.1	10.0 ± 2.0
	0	525 ± 131	2560 ± 300	0.0 ± 0.0	0.0 ± 0.0
	20	353 ± 97	997 ± 259	82.5 ± 15.1	49.6 ± 32.6

^aThe O₂ concentration based on in situ sensor information at the time of sample collection is denoted by boldface text.

chromatograph (GC; Kyoto, Japan) with a thermal conductivity detector (TCD) to measure N₂, CO₂, and O₂ and an electron capture detector (ECD) to measure N₂O. The glass jars were housed in a Plexiglas box, which was filled with water over the top of the lids to aid in preventing air leaks into the jars. The system has a total of 10 jars; for each experiment, four cores from each the wet and dry areas were analyzed, and two jars remained empty as blanks. The advantages and disadvantages of the N-FARM have been addressed previously [Burgin *et al.*, 2010], but briefly, the major advantage of our system is the ability to precisely measure very small changes in [N₂] and [N₂O] (detection limits described below) on relatively large soil samples (whole, intact cores). The major limitation of this method is the very low throughput of the analysis; it generally takes 7–10 days to analyze eight cores (including “killed” cores, see below for detail), depending on the number of [O₂] treatments and rates of denitrification.

[13] Once cores were loaded into jars, a gas mixture of helium and oxygen (HelOx), with O₂ concentrations at 0% (ultra high purity He), 5, 10 and 20% was used to replace the existing N₂ containing atmosphere. As in other systems, the incubation gas was repeatedly injected into the cores and then removed by very low vacuum (500 torr; K. Butterbach-Bahl, personal communication, 2009), which switched with a slight over-pressurization (860 torr) at 90-s intervals. Methods development tests showed that a 14-h vacuum/flush cycle on 90-s intervals resulted in 560 switches, creating a long and effective serial dilution and evacuation of the headspace, which removed all traces of atmospheric N₂ thereby negating the large background interference that would dilute the signal of denitrification we were trying to measure. After the 14-h flush/vacuum cycle, the HelOx gas was flushed through the cores (with no vacuum) for an additional 2 h to ensure the cores were at equilibrium.

[14] Once cores were done flushing, the system was set to “incubate” mode wherein gases that were produced by the cores accumulated in the jars. Gases were sampled at time intervals that were dependent on the rates of production, but were generally 5–6 h for low [O₂] incubation conditions and 12–24 h for high [O₂] incubations. For each month, both the wet and dry site cores were run at the lowest recorded [O₂] from the site data (Figure 1, generally 0% O₂; Table 1)

and highest recorded [O₂] (Figure 1, O₂ varied by season; Table 1). In June and September, the same cores were run at multiple O₂ concentrations to examine the effects of different O₂ concentrations on denitrification activity.

[15] To sample the headspace of the cores, 50-mL (approximately 1/10th the volume of one core) was released from a pressurized chamber into the core, slightly over-pressurizing the headspace. After mixing, this over-pressurization was released and allowed to flush the system lines and loops, which contained a total volume of approximately 10 mL. The sample from the loop was then transferred onto the GC columns. Flux rates were calculated by regression of N₂-N versus time of incubation corrected for the blanks and any dilution that occurred via sampling. These were expressed in areal terms by dividing by the area of the core cylinder.

[16] To ensure that measured N₂ production was coming from denitrification and not from leakage or degassing from small soil pores, we used the rate of N₂ production in “killed” cores to correct for any background N₂ that may not have been flushed from the soil pore space. After being run as “live” for ambient denitrification rates, the same cores were killed by autoclaving three times over two days at 134°C for 60 min. Killed cores were only incubated at 0% O₂ (anoxic conditions) which would maximize any potential denitrification. The rates of N₂ flux from killed cores were very low (50.8 ± 8.1) in comparison to the live cores (~500–3000 $\mu\text{g N m}^{-2} \text{h}^{-1}$), ensuring we were measuring a strong signal of denitrification activity.

2.4. Soil-Atmosphere Trace Gas Flux Methods

[17] Our field trace gas flux method was similar to that described previously [Bowden *et al.*, 1991]. We used 287 mm-dia, 5 cm-tall PVC cylinders for chambers with gas sampling ports in the center of the chamber lid. These lids were placed on PVC base rings of the same size, which were permanently installed at the site. Ten mL gas samples were collected using a syringe at 0, 10, 20, and 30 min after the placement of the chamber lids onto the bases. Samples were transferred to evacuated glass vials which were stored at room temperature before analysis by gas chromatography with ECD (N₂O), TCD (CO₂) and flame ionization detection (FID, CH₄). Fluxes were calculated from the linear rate of

Table 2. Monthly Soil NO₃⁻ and NH₄⁺ Concentrations at Dry and Wet Sampling Sites

	Dry Zone		Wet Zone	
	NO ₃ ⁻	NH ₄ ⁺	NO ₃ ⁻	NH ₄ ⁺
	<i>mg N kg Dry Soil⁻¹</i>			
May	0.3 ± 0.1	3.2 ± 0.1	BDL ^a	3.2 ± 0.6
June	0.4 ± 0.1	2.2 ± 0.6	N/A ^b	N/A
July	BDL	6.0 ± 0.3	BDL	4.0 ± 0.3
August	BDL	3.1 ± 0.2	BDL	1.8 ± 0.3
September	0.3 ± 0.1	3.1 ± 0.1	BDL	1.1 ± 0.1
October	N/A	N/A	2.6 ± 0.4	0.8 ± 0.1

^aBelow detection limit.^bNot available.

change in gas concentration, corrected for the chamber volume, outside temperature during the collection and surface area of the underlying soil. We collected samples monthly from three chambers in each the wet and dry zones of the Gifford Riparian Wetland.

2.5. Soil Characteristics

[18] Composite soil samples were taken every month from the wet and dry zones (Table 2). Soil organic matter content was determined in August 2009 by loss on ignition at 450°C for 4 h. The wet zone soils were 7.2 ± 0.7% carbon, whereas the dry zone soil was 7.6 ± 0.2% carbon. Concentrations of inorganic N (NO₃⁻ and NH₄⁺; Table 2) were determined by extraction with 2M KCl and colorimetric analysis with a Lachat Flow Injection Analyzer.

2.6. Statistical Analysis

[19] Mean values of the response variables (gas fluxes) were calculated from four replicate cores or three replicate field gas flux chambers. When appropriate, fluxes were log transformed to normalize data. A one-way Analysis of Variance (ANOVA) was conducted to test for differences in N₂ or N₂O flux between the wet versus dry zones. The effects of site (e.g., wet versus dry zones) and month of measurement on greenhouse gas dynamics were assessed with a two-way repeated measures ANOVA (with interactions) of plot means from three replicate gas flux chambers from each sampling date. Effects with a p < 0.05 were determined to be significant. ANOVA was performed using JMP 9.0.

3. Results

[20] Soil [O₂] ranged from 0 to 20% depending on the proximity to the stream and the season (Figure 1). Soil O₂ was consistently low in the area closest to the stream (sensor 4) and increased further from the stream (sensor 1, Figure 1b). Soil [O₂] was also higher in the spring, and decreased over the course of the late spring and summer. All four sensors were briefly anoxic for a week in late September, after which occurred a very fast (<3 days) switch to completely oxic conditions in the two sensors furthest from the stream (sensors 1 and 2, Figure 1b), with a more muted increase in [O₂] in sensor 3. The volumetric water content (VWC) of the wet area was consistently ~5% higher than that of the dry area throughout much of the spring and summer (Figure 1c). The VWC of the two areas converged in late summer (September) and remained similar until early

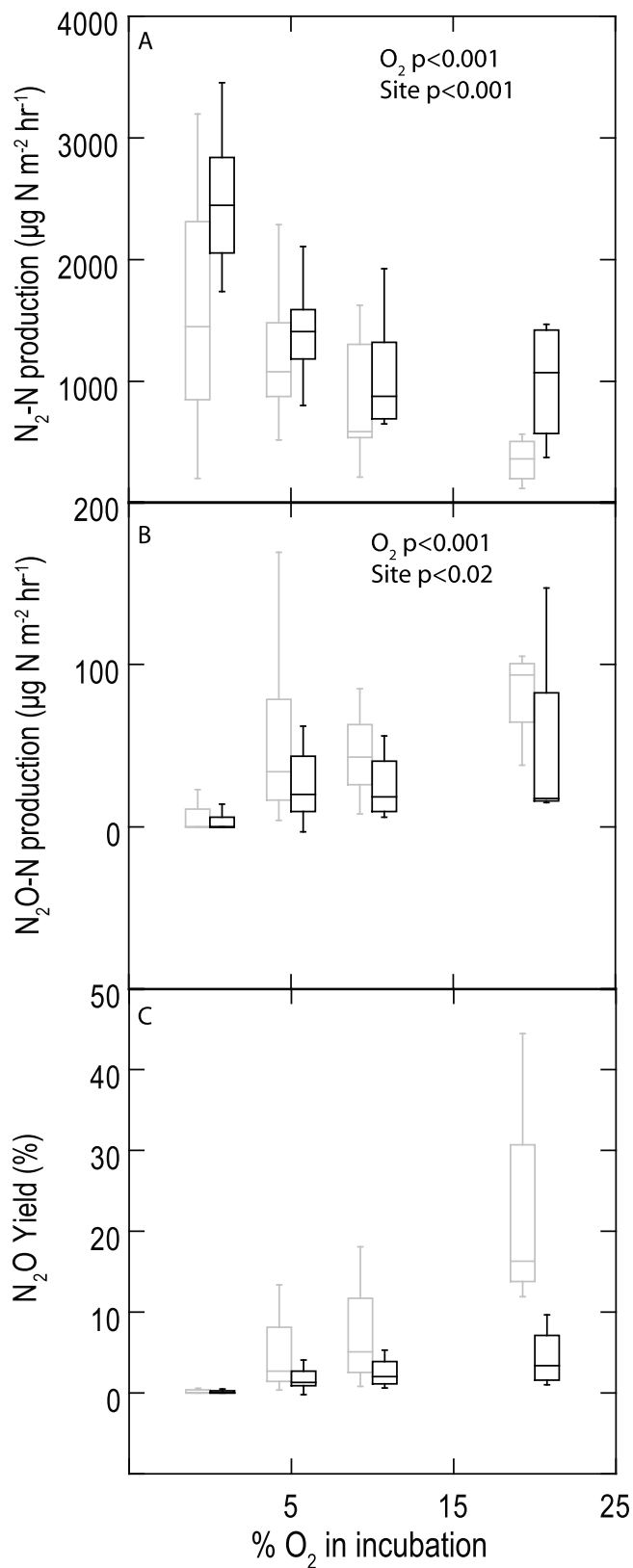


Figure 2. (a) N₂ production, (b) N₂O production, and (c) N₂O yield (expressed as a percentage) from wet (black) and dry (gray) zone cores at different O₂ concentrations. Whiskers denote the minimum and maximum measured values and the box's middle line denotes the median value.

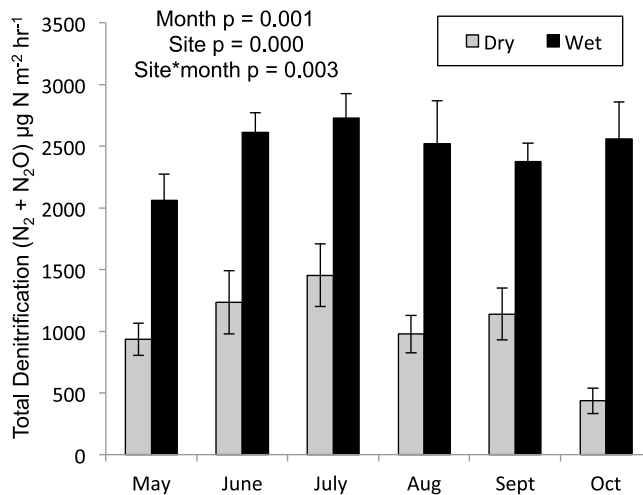


Figure 3. Total denitrification over a six-month period in wet and dry zone soils. Bars represent the mean denitrification rate ($N_2 + N_2O$) measured in four cores (± 1 standard error of the mean) per zone at field appropriate O_2 concentrations in each month. Rates for N_2 and N_2O production are in Table 2.

October when the VWC of the dry area quickly dropped, coincident with the rapid increase in soil $[O_2]$ (Figures 1b and 1c). The wet area was also consistently 1–2°C cooler than the dry area for most of the spring and summer before converging in early fall (Figure 1c). In the period before the rapid $[O_2]$ switch in late September, soil $[O_2]$ was significantly correlated with VWC in the dry site (dry sensor 1 $r = 0.92$, sensor 2 = 0.93), and had a much weaker correlation in the wet site (sensor 3 $r = 0.51$, sensor 4 $r = -0.51$). Soil $[O_2]$ was also significantly correlated with soil temperature in the dry site (dry sensor 1 $r = -0.83$, sensor 2 = -0.82), and again had a much weaker correlation in the wet site (sensor 3 $r = -0.54$, sensor 4 $r = 0.57$). In both sites, however, soil temperature and VWC were significantly correlated (dry $r = -0.88$; wet $r = -0.80$), indicating that the effects of VWC (soil moisture) and temperature on $[O_2]$ dynamics cannot be separated.

[21] N_2 production was generally 5–10× greater than N_2O production from cores collected in the “wet” and “dry” zones (Figures 2b and 2c). Increasing $[O_2]$ significantly decreased N_2 production ($F_{3,104} = 21.5$; $p < 0.001$) and significantly increased N_2O production ($F_{3,108} = 20.8$; $p < 0.001$) in both the wet and dry sites (Figure 2). The wet site produced significantly more N_2 than the dry site (Figure 2a; $F_{1,104} = 9.4$; $p = 0.003$). In general, the dry site cores produced more N_2O than the wet site cores, though the difference was not statistically significant (Figure 2b; $F_{1,108} = 2.1$; $p < 0.15$). N_2O yield, which is a composite of both N_2 and N_2O production, ranged from ~0–36% in the dry site and 0–6.5% in the wet site. N_2O yield was consistently higher in the dry zone compared to the wet zone and increased with increasing O_2 (Figure 2c).

[22] Denitrification rates, calculated as the sum of N_2 and N_2O fluxes and corrected for field-appropriate O_2 concentrations (Table 1), were generally 2× higher in the wet zone compared to the dry zone (Figure 3; $F_{3,111} = 11.4$; site $p = 0.000$; month $p = 0.001$). The presence of a significant

interaction term between month and site ($p = 0.003$) indicates that the difference in denitrification rates between the wet and dry areas is not consistent across time; this is apparent by comparing the difference in the sites in July and October. The major factor driving this difference in denitrification rates is the soil $[O_2]$ (Figure 1b). Dry zone N_2 fluxes ranged from a low of 353 $\mu g N m^{-2} hr^{-1}$ in October to a high of 1972 $\mu g N m^{-2} hr^{-1}$ in June (Table 2). Wet zone N_2 fluxes were relatively uniform, ranging from a low of 2053 $\mu g N m^{-2} hr^{-1}$ in May to a high of 2813 $\mu g N m^{-2} hr^{-1}$ in July (Table 2). The maximum denitrification for the wet zone was 2813 $\mu g N m^{-2} hr^{-1}$ (July) and 2725 $\mu g N m^{-2} hr^{-1}$ (July) for the dry zone, indicating that both zones could reach similar denitrification potentials under the optimal conditions (Table 2).

[23] CO_2 fluxes were significantly higher in the wet site compared to the dry site ($p = 0.02$ for effect of site; $p = 0.49$ for sample month; $p = 0.76$ for month*site) and showed a clear seasonal pattern in the wet zone with maximal fluxes of 37 $mg C m^{-2} hr^{-1}$ in July and much lower fluxes in the spring and fall (Figure 4a). This pattern was absent, however, in the dry zone where CO_2 fluxes were much more consistent between months, often around 10 $mg C m^{-2} hr^{-1}$ (Figure 4a). N_2O fluxes ranged from 0–0.8 $ng N m^{-2} hr^{-1}$ and did not correlate with season or site, and were often near zero in both the dry and wet zones (Figure 4b). CH_4 fluxes ranged from 0–258 $\mu g C m^{-2} hr^{-1}$ and were similarly variable with no consistent seasonal pattern (Figure 4c). In general, the wet and dry zone had similar CH_4 fluxes with the exception of June and July in which the dry zone had a significantly higher flux than the wet zone (Figure 4c).

4. Discussion

4.1. Improved Estimates of Denitrification in Forests

[24] Our results demonstrate that the combination of our N-FARM technique [Burgin *et al.*, 2010; M. V. Kulkarni *et al.*, A comparison of denitrification rates as measured using direct flux and $15N$ tracer methods in northeastern forest soils, submitted to *Biogeochemistry*, 2011] and an understanding of field-level variation in soil O_2 dynamics (Figure 1b) yields better constrained estimates of in situ denitrification rates (Figure 3). Direct measurement of N_2 production without any inhibitors or supplemental NO_3^- stimulation is rare, especially in soils [Groffman *et al.*, 2006]. The high atmospheric background of N_2 necessitates that these measurements be made in enclosure-type systems, which introduce some of their own limitations, including sampling disturbance effects and lag time between sampling and measurement. In spite of these limitations, we found significant differences in the denitrification rates between the wet and dry zone soils across O_2 levels (Figure 2).

[25] The range of denitrification rates (353–2813 $\mu g N m^{-2} hr^{-1}$) that we measured under $[O_2]$ matched to field conditions (Figure 1b) corresponds to 22.4–67.5 $mg N m^{-2} day^{-1}$. This is comparable to the range of rates we measured at these same sites in 2008 [Burgin *et al.*, 2010], as well as those documented for other riparian ecosystems, including 19 $mg N m^{-2} day^{-1}$ in a Georgia hardwood riparian zone [Lowrance *et al.*, 1995] and 3–82 $mg N m^{-2} day^{-1}$ in a Pennsylvania hardwood riparian zone [Watts and Seitzinger, 2000]. Extrapolating monthly rates (Figure 3) yields a range of 45–109 $kg N ha^{-1} season^{-1}$ (assuming a 184 d season),

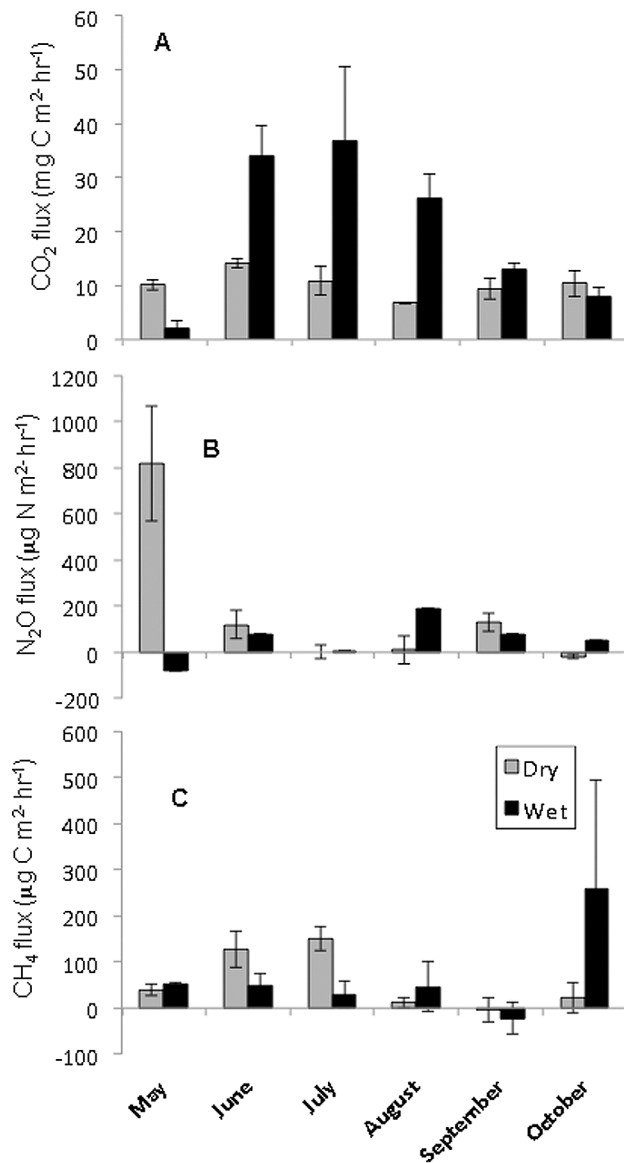


Figure 4. Field fluxes of (a) carbon dioxide, (b) nitrous oxide, and (c) methane for dry (gray bars) and wet (black bars) zone soils collected once per month. Bars are an average of three replicate static chambers ± 1 standard error of the mean.

which is comparable to the range measured in Tuttingen (Germany) forests ($14\text{--}94 \text{ kg N ha}^{-1} \text{ year}^{-1}$) using a similar system to directly measure N_2 and N_2O [Dannenmann *et al.*, 2008].

[26] While comparable to estimated denitrification ranges from other studies, our scaled range of $45\text{--}109 \text{ kg N ha}^{-1} \text{ season}^{-1}$ is high compared to the amount of denitrification thought to occur in Northeastern deciduous forests. Although mass balance studies frequently show large amounts of missing N, denitrification is thought to be unimportant to northeastern forests because early studies suggested that N gas fluxes were low in the region [Bowden and Bormann, 1986; Davidson and Swank, 1990]. These studies, however, only measured N_2O production across a limited number of sites. More recent studies using improved methodology for

measuring N_2 fluxes, including the data we present herein, indicate that N_2 is the dominant end product of denitrification [Dannenmann *et al.*, 2008]. Our results agree with this finding and strengthen the idea that N_2O fluxes may not be a good predictor of overall denitrification activity.

4.2. Understanding O₂ Dynamics in Riparian Wetlands

[27] The basic question asked herein is: How does soil O₂ vary in riparian wetlands? Though O₂ has long been recognized as an important driver for determining microbial metabolism, relatively little is known about field-level O₂ dynamics in soils [Burgin *et al.*, 2010; Liptzin *et al.*, 2011; Silver *et al.*, 1999]. We found support for our hypotheses that: 1) soil O₂ was dynamic, with O₂ concentrations dependent on proximity to the stream and water table, and 2) soil O₂ would vary seasonally (Figure 1). While few studies have measured soil O₂, those that have find [O₂] is dynamic under varying hydrologic conditions resulting from connection to a water table [Burgin *et al.*, 2010], seasonal water table dynamics [Faulkner and Patrick, 1992; Megonigal *et al.*, 1993] or precipitation [Liptzin *et al.*, 2011; Silver *et al.*, 1999]. Despite small differences in water content between the dry and wet zones (Figure 1b), water dynamics are also clearly linked to soil [O₂] variation in the Gifford Riparian Wetland (Figure 1a).

[28] The most striking observations from the soil O₂ record were the low values at high temperature and relatively low water content in mid summer, and the rapid shift from complete anoxia to oxic conditions, which occurred in late September over a period of less than three days (Figure 1). The mid-summer pattern of decreasing [O₂] was likely driven by plants, from both direct oxygen consumption by root respiration and stimulation of microbial respiration by root exudation and turnover [Woldendorp, 1962]. A similar rapid shift in fall was also documented in the Gifford Riparian Wetland in 2008 [Burgin *et al.*, 2010], though the shift occurred much earlier in the season, likely because 2008 was drier than 2009. This shift did not correspond to any changes in precipitation, but was correlated with a slight decrease in soil VWC (Figure 1). We therefore hypothesize that the rapid switch occurs because the soil has dried to the point that the macropores no longer have water blocking the diffusion of atmospheric air into the soils. Plant senescence may also play a role. Rapid transitions such as these are known to occur in soils and are thought to be important in controlling microbial processes and gas fluxes [Metivier *et al.*, 2009]. These rapid increases in [O₂] have also been documented in the generally low-O₂ rain forest soils under extended dry periods [Liptzin *et al.*, 2011] and floodplain wetlands of the Savannah River [Megonigal *et al.*, 1993]. This rapid switch is a stark contrast to the gradual decline in [O₂] for most of the season, which took months to gradually become anoxic (Figure 1a). A similar pattern to that documented in Gifford Wetland was also seen in forested wetlands, wherein a late spring rise in the water table corresponded to decreasing soil [O₂] and mid-summer drying led to increased soil [O₂] [Megonigal *et al.*, 1993]. Clearly both physical (e.g., water dynamics) and biological (e.g., plant and microbial community) factors control soil [O₂]; however, the relative contribution of these to the overall pattern of soil O₂ concentration remains unknown. More research in both the areas of soil O₂ dynamics across various ecosystems, coupled with

bench-scale manipulations of biotic and abiotic factors would enhance our nascent understanding of soil O₂ dynamics and drivers.

4.3. Controls on Denitrification Rates and End Products

[29] The second question we sought to address was: How does O₂ variation affect denitrification rates and end products? O₂ controls denitrification through inhibition of denitrification enzymes; however, not all denitrifying bacteria respond similarly to the presence of oxygen [Abou Seada and Ottow, 1985; Morley et al., 2008]. Our results indicate that increasing soil O₂ concentration clearly decreases denitrification rates and increases the fraction of N₂O in the final end products (Figure 2), lending support to our 3rd and 4th hypotheses. The connection between O₂ concentration, denitrification rates and N₂O flux has been known for over thirty years [Firestone et al., 1980]; however, our findings extend this understanding by providing the additional context of an increased understanding of field-level variation in O₂ and how this can feedback to affect denitrification and N₂O yield.

[30] The relationship between soil O₂ and denitrification differed between the wet and dry sites, with higher denitrification rates in the wet site at any given level of O₂ (Figure 2). One likely explanation is that soil macropore [O₂] (measured by the soil O₂ probes (Figure 1)), and microsite [O₂] (where denitrification occurs) differs between the sites due to differences in moisture content and soil properties (texture, organic matter content). These results suggest that establishing relationships between denitrification and [O₂] in different soil types may be a useful approach for scaling results from field to landscape and regional scales [Groffman and Tiedje, 1989].

[31] The different O₂:denitrification relationships that we observed in the wet versus dry zones are likely due to differences in microbial community acclimation to the particular environmental conditions. That is, the wet zone had significantly more denitrification than the dry zone soils at each soil [O₂] because a microbial community more acclimated to anoxic conditions was present. It has recently been suggested that knowledge of denitrifier abundance may not lend additional predictive power to understanding the controls on denitrification [Attard et al., 2011]. However, our new method appears sensitive enough to discern differences in denitrification under nearly identical conditions (O₂, Figure 2; available N, Table 1; organic carbon, section 2.5). This suggests that other methods to measure denitrification (e.g., the commonly used acetylene block technique) are not sensitive enough to distinguish differences in the denitrification activity of microbial communities acclimated to even slight variation in environmental conditions.

[32] While our new method appears sensitive enough to distinguish differences in the denitrification capacity of the wet and dry zones, it cannot discern whether the N₂O generated is from nitrification or denitrification. O₂ is known to affect both processes by increasing the proportion of transformation going to N₂O [Bollmann and Conrad, 1998]. Bollmann and Conrad [1998] found that nitrification was the main source of N₂O at lower soil moistures, whereas denitrification predominated N₂O fluxes at higher soil moistures. Thus, in more aerated soils, the majority of N₂O fluxes are

thought to stem from nitrification [Bremner and Blackmer, 1981]. Given that we cannot distinguish the two processes using our N-FARM method, we warn that the estimates of N₂O yield may be an overestimation of N₂O flux due to denitrification (Figure 2c). However, given the high soil moisture contents of these soils (Figure 1c), it is likely that much of the N cycling activity is due to denitrification.

[33] The controls on N₂O production are of particular interest because N₂O currently accounts for 6% of radiative forcing and destroys stratospheric ozone [Intergovernmental Panel on Climate Change, 2008]. N₂O yield increased with increasing exposure to O₂ and ranged from 0 to 36%, similar to the range reported in two literature reviews of N₂O yield [Beaulieu et al., 2011; Schlesinger, 2009]. The balance between electron donors (organic C) and acceptors (NO₃⁻) is often invoked to predict N₂O yield [Firestone and Davidson, 1989]. Other studies have cited a connection between N₂O yield and soil moisture or water filled pore space [Bergsma et al., 2002; Ciarlo et al., 2007; Rudaz et al., 1999; Ruser et al., 2006; Scheer et al., 2008]. However, our data suggests that soil O₂ is perhaps the dominant control on N₂O yield. [O₂] is highly correlated with water filled pore space, consistent with the often-observed connection between soil moisture and N₂O yield as well. The distinctive connection between soil [O₂] and N₂O yield in different soils suggest that measuring these relationships may be a useful approach for sorting out the extreme variation in yield that has been observed in previous studies [Beaulieu et al., 2011; Schlesinger, 2009].

4.4. O₂ Effects on Greenhouse Gas Fluxes

[34] In addition to understanding how soil O₂ varied, and how the variation affected denitrification rates, we also asked: How does seasonal variation in O₂ affect in situ greenhouse gas fluxes? We measured fluxes of N₂O, CO₂ and CH₄ from the wet and dry zones to test the hypothesis that gases indicative of increased anaerobic conditions (e.g., CH₄ and N₂O) would increase under decreasing field O₂. Greenhouse gas flux rates measured from the Gifford Riparian Wetland fall within the published ranges of fluxes from other riparian sites, as compiled by Soosaar et al. [2011]. N₂O fluxes at the Gifford site ranged from -20–820 μg N m⁻² hr⁻¹, which is comparable to range of -25–104 μg N m⁻² hr⁻¹ from a mixed forest riparian zone [Dhondt et al., 2004] and 108–566 μg N m⁻² hr⁻¹ in an Alder riparian zone [Hefting et al., 2006]. CO₂ fluxes in our site ranged from 2.2–37 mg CO₂ m⁻² hr⁻¹, within the range of 0–365 mg CO₂ m⁻² hr⁻¹ measured in a gray alder riparian zone over an eight-year period [Soosaar et al., 2011]. CH₄ flux at our site ranged from -22–258 μg CH₄ m⁻² hr⁻¹, well within the range of -38–561 μg CH₄ m⁻² hr⁻¹ in the study reported by Soosaar et al. [2011] and generally higher than the range of 0.06–0.15 μg CH₄ m⁻² hr⁻¹ in riparian zones of northern hardwood forests [Hopfensperger et al., 2009]. Of the three gases, only CO₂ fluxes showed a clear seasonal signal with increased CO₂ in the summer compared to spring or fall probably due to the influence of increased plant respiration during the growing season. However, this was only apparent in the wet zone, whereas the CO₂ flux from the dry zone was relatively consistent over time. Neither N₂O nor CH₄ fluxes (Figure 4) correlated were correlated to O₂ patterns (Figure 1) in the wet or dry zone.

[35] Field-level N₂O fluxes often exhibit micro-scale spatial variability, which makes understanding and predicting the controls on N₂O fluxes difficult [DeSimone et al., 2010; Groffman and Gold, 1998; van den Heuvel et al., 2009]. Temporal patterns are also difficult to discern, though many studies have documented relatively large fluxes in early spring compared to other seasons [DeSimone et al., 2010; Dhondt et al., 2004]. The field fluxes of N₂O measured in the Gifford Riparian Wetland also fit the pattern of high spatial and temporal variation, with the highest flux occurring in May. While we hypothesized that field-level O₂ concentrations would correlate with N₂O fluxes, we did not see a corresponding pattern in the data. Therefore, while O₂ exhibits tight control on N₂O flux at some spatial scales (e.g., cores), it does not seem to translate to predicting patterns at larger spatial scales (e.g., gas chambers). However, this is not altogether surprising since data sets with much more temporal and spatial resolution than ours have also failed to link N₂O fluxes with key drivers [van den Heuvel et al., 2009]. Temporal scaling may be an issue here—we measured trace gases once a month, whereas O₂ was measured hourly. Spatial factors may also be important, as macropore [O₂] may not be reflective of conditions within the much larger area of the GHG chamber and may not account for the balance of N₂O producing and consuming sites within the chamber area. Understanding the spatial and temporal complexity of these controlling factors will be necessary to build models capable of predicting N loss through denitrification and N₂O fluxes associated with this activity.

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References

- Abou Seada, M. N. I., and J. C. G. Ottow (1985), Effect of increasing oxygen concentrations on total denitrification and nitrous oxide release from soil by different bacteria, *Biol. Fertil. Soils*, *1*, 31–38, doi:10.1007/BF00710968.
- Altor, A. E., and W. J. Mitsch (2006), Methane flux from created riparian marshes: Relationship to intermittent versus continuous inundation and emergent macrophytes, *Ecol. Eng.*, *28*(3), 224–234, doi:10.1016/j.ecoleng.2006.06.006.
- Attard, E., S. Recous, A. Chabbi, C. De Berranger, N. Guillaumaud, J. Labreche, L. Philippot, B. Schmid, and X. Le Roux (2011), Soil environmental conditions rather than denitrifier abundance and diversity drive potential denitrification after changes in land uses, *Global Change Biol.*, *17*, 1975–1989, doi:10.1111/j.1365-2486.2010.02340.x.
- Beaulieu, J. J., et al. (2011), Nitrous oxide emission from denitrification in stream and river networks, *Proc. Natl. Acad. Sci. U. S. A.*, *108*(1), 214–219, doi:10.1073/pnas.1011464108.
- Bergsma, T. T., G. P. Robertson, and N. E. Ostrom (2002), Influence of soil moisture and land use history on denitrification end-products, *J. Environ. Qual.*, *31*, 711–717, doi:10.2134/jeq2002.0711.
- Bollmann, A., and R. Conrad (1998), Influence of O₂ availability on NO and N₂O release by nitrification and denitrification in soils, *Global Change Biol.*, *4*, 387–396, doi:10.1046/j.1365-2486.1998.00161.x.
- Bowden, R. D., J. M. Melillo, P. A. Stuedler, and J. D. Aber (1991), Effects of nitrogen additions on annual nitrous oxide fluxes from temperate forest soils in the Northeastern United States, *J. Geophys. Res.*, *96*(D5), 9321–9328, doi:10.1029/91JD00151.
- Bowden, W. B., and F. H. Bormann (1986), Transport and loss of nitrous oxide in soil-water after forest clear-cutting, *Science*, *233*(4766), 867–869, doi:10.1126/science.233.4766.867.
- Bremner, J. M., and A. M. Blackmer (1981), Terrestrial nitrification as a source of atmospheric nitrous oxide, in *Denitrification, Nitrification and Atmospheric Nitrous Oxide*, edited by C. C. Delwiche, pp. 151–170, Wiley, New York.
- Burgin, A. J., P. Groffman, and D. Lewis (2010), Factors regulating denitrification in a riparian wetland, *Soil Sci. Soc. Am. J.*, *74*(5), 1826–1833, doi:10.2136/sssaj2009.0463.
- Butterbach-Bahl, K., G. Willibald, and H. Papen (2002), Soil core method for direct simultaneous determination of N₂ and N₂O emissions from forest soils, *Plant Soil*, *240*(1), 105–116, doi:10.1023/A:1015870518723.
- Ciarlo, E., M. Conti, N. Bartoloni, and G. Rubio (2007), The effect of moisture on nitrous oxide emissions from soil and the N₂O/(N₂O + N₂) ratio under laboratory conditions, *Biol. Fertil. Soils*, *43*, 675–681, doi:10.1007/s00374-006-0147-9.
- Crawford, R. M. M. (1992), Oxygen availability as an ecological limit to plant distribution, *Adv. Ecol. Res.*, *23*, 93–185, doi:10.1016/S0065-2504(08)60147-6.
- Dannenmann, M., K. Butterbach-Bahl, R. Gasche, G. Willibald, and H. Papen (2008), Dinitrogen emissions and the N₂:N₂O emission ratio of a Rendzic Leptosol as influenced by pH and forest thinning, *Soil Biol. Biochem.*, *40*(9), 2317–2323, doi:10.1016/j.soilbio.2008.05.009.
- Davidson, E. A. (2009), The contribution of manure and fertilizer nitrogen to atmospheric nitrous oxide since 1860, *Nat. Geosci.*, *2*(9), 659–662, doi:10.1038/ngeo608.
- Davidson, E. A., and W. T. Swank (1990), Nitrous oxide dissolved in soil solution: An insignificant pathway of nitrogen loss from a southeastern hardwood forest, *Water Resour. Res.*, *26*(7), 1687–1690.
- DeSimone, J., M. L. Macrae, and R. A. Bourbonniere (2010), Spatial variability in surface N₂O fluxes across a riparian zone and relationships with soil environmental conditions and nutrient supply, *Agric. Ecosyst. Environ.*, *138*(1–2), 1–9, doi:10.1016/j.agee.2010.03.007.
- Dhondt, K., P. Boeckx, G. Hofman, and O. Cleemput (2004), Temporal and spatial patterns of denitrification enzyme activity and nitrous oxide fluxes in three adjacent vegetated riparian buffer zones, *Biol. Fertil. Soils*, *40*(4), 243–251, doi:10.1007/s00374-004-0773-z.
- Faulkner, S. P., and W. H. Patrick (1992), Redox processes and diagnostic wetland soil indicators in bottomland hardwood forests, *Soil Sci. Soc. Am. J.*, *56*(3), 856–865, doi:10.2136/sssaj1992.03615995005600030030x.
- Fazzolari, E., B. Nicolardot, and J. C. Germon (1998), Simultaneous effects of increasing levels of glucose and oxygen partial pressures on denitrification and dissimilatory nitrate reduction to ammonium in repacked soil cores, *Eur. J. Soil Biol.*, *34*(1), 47–52, doi:10.1016/S1164-5563(99)80006-5.
- Firestone, M. K., and E. A. Davidson (1989), Microbiological basis of NO and N₂O production and consumption in soils, in *Exchange of Trace Gases Between Terrestrial Ecosystems and the Atmosphere*, edited by M. O. Andreae and D. S. Schimel, pp. 7–21, Wiley, New York.
- Firestone, M. K., R. B. Firestone, and J. M. Tiedje (1980), Nitrous oxide from soil denitrification: Factors controlling its biological production, *Science*, *208*, 749–751, doi:10.1126/science.208.4445.749.
- Galloway, J. N., et al. (2004), Nitrogen cycles: Past, present, and future, *Biogeochemistry*, *70*(2), 153–226, doi:10.1007/s10533-004-0370-0.
- Galloway, J. N., A. R. Townsend, J. W. Erisman, M. Bekunda, Z. Cai, J. R. Freney, L. A. Martinelli, S. P. Seitzinger, and M. A. Sutton (2008), Transformation of the nitrogen cycle: Recent trends, questions, and potential solutions, *Science*, *320*(5878), 889–892, doi:10.1126/science.1136674.
- Glatzel, S., N. Basiliko, and T. Moore (2004), Carbon dioxide and methane production potentials of peats from natural, harvested, and restored sites, eastern Quebec, Canada, *Wetlands*, *24*(2), 261–267, doi:10.1672/0277-5212(2004)024[0261:CDAMPP]2.0.CO;2.
- Gleason, R. A., B. A. Tangen, B. A. Browne, and N. H. Euliss Jr. (2009), Greenhouse gas flux from cropland and restored wetlands in the Prairie Pothole Region, *Soil Biol. Biochem.*, *41*(12), 2501–2507, doi:10.1016/j.soilbio.2009.09.008.
- Grimm, N. B., et al. (2003), Merging aquatic and terrestrial perspectives of nutrient biogeochemistry, *Oecologia*, *137*(4), 485–501, doi:10.1007/s00442-003-1382-5.
- Groffman, P. M., and A. J. Gold (1998), Nitrous oxide production in riparian zones and groundwater, *Nutr. Cycl. Agroecosyst.*, *52*, 179–186, doi:10.1023/A:1009719923861.
- Groffman, P. M., and J. M. Tiedje (1989), Denitrification in north temperate forest soils: Relationships between denitrification and environmental factors at the landscape scale, *Soil Biol. Biochem.*, *21*(5), 621–626.
- Groffman, P. M., J. M. Tiedje, D. L. Mokma, and S. Simkins (1992), Regional scale analysis of denitrification in north temperate forest soils, *Landscape Ecol.*, *7*, 45–53, doi:10.1007/BF02573956.
- Groffman, P. M., M. A. Altabet, J. K. Böhlke, K. Butterbach-Bahl, M. B. David, M. K. Firestone, A. E. Giblin, T. M. Kana, L. P. Nielsen, and M. A. Voytek (2006), Methods for measuring denitrification: Diverse approaches to a difficult problem, *Ecol. Appl.*, *16*(6), 2091–2122, doi:10.1890/1051-0761(2006)016[2091:MFMDDA]2.0.CO;2.

- Gruber, N., and J. N. Galloway (2008), An Earth-system perspective of the global nitrogen cycle, *Nature*, 451, 293–296, doi:10.1038/nature06592.
- Harms, T. K., and N. B. Grimm (2008), Hot spots and hot moments of carbon and nitrogen dynamics in a semiarid riparian zone, *J. Geophys. Res.*, 113, G01020, doi:10.1029/2007JG000588.
- Hefting, M. M., R. Bobbink, and M. P. Janssens (2006), Spatial variation in denitrification and N₂O emission in relation to nitrate removal efficiency in a N-stressed riparian buffer zone, *Ecosystems*, 9(4), 550–563, doi:10.1007/s10021-006-0160-8.
- Hopfensperger, K. N., C. M. Gault, and P. M. Groffman (2009), Influence of plant communities and soil properties on trace gas fluxes in riparian northern hardwood forests, *For. Ecol. Manage.*, 258(9), 2076–2082, doi:10.1016/j.foreco.2009.08.004.
- Howarth, R., F. Chan, D. J. Conley, J. Garnier, S. C. Doney, R. Marino, and G. Billen (2011), Coupled biogeochemical cycles: Eutrophication and hypoxia in temperate estuaries and coastal marine ecosystems, *Front. Ecol. Environ.*, 9(1), 18–26, doi:10.1890/100008.
- Intergovernmental Panel on Climate Change (2008), *Climate Change 2007: Synthesis Report. Contribution of Working Groups I, II and III to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*, edited by Core Writing Team et al., 104 pp., Geneva, Switzerland.
- Liptzin, D., W. L. Silver, and M. Detto (2011), Temporal dynamics in soil oxygen and greenhouse gases in two humid tropical forests, *Ecosystems*, 14(2), 171–182, doi:10.1007/s10021-010-9402-x.
- Lowrance, R., G. Vellidis, and R. K. Hubbard (1995), Denitrification in a restored riparian forest wetland, *J. Environ. Qual.*, 24(5), 808–815, doi:10.2134/jeq1995.00472425002400050003x.
- Mayer, P. M., S. K. Reynolds, M. D. McCutchen, and T. J. Canfield (2007), Meta-analysis of nitrogen removal in riparian buffers, *J. Environ. Qual.*, 36, 1172–1180, doi:10.2134/jeq2006.0462.
- McClain, M. E., et al. (2003), Biogeochemical hot spots and hot moments at the interface of terrestrial and aquatic ecosystems, *Ecosystems*, 6(4), 301–312, doi:10.1007/s10021-003-0161-9.
- Megonigal, J. P., W. H. Patrick, and S. P. Faulkner (1993), Wetland identification in seasonally flooded forest soils: Soil morphology and redox dynamics, *Soil Sci. Soc. Am. J.*, 57, 140–149, doi:10.2136/sssaj1993.03615995005700010027x.
- Metivier, K. A., E. Pattey, and R. F. Grant (2009), Using the *ecosys* mathematical model to simulate temporal variability of nitrous oxide emissions from a fertilized agricultural soil, *Soil Biol. Biochem.*, 41, 2370–2386, doi:10.1016/j.soilbio.2009.03.007.
- Morley, N., E. M. Baggs, P. Dörsch, and L. Bakken (2008), Production of NO, N₂O and N₂ by extracted soil bacteria, regulation by NO₂⁻ and O₂ concentrations, *FEMS Microbiol. Ecol.*, 65(1), 102–112, doi:10.1111/j.1574-6941.2008.00495.x.
- Paerl, H. W., R. L. Dennis, and D. R. Whittall (2002), Atmospheric deposition of nitrogen: Implications for nutrient over-enrichment of coastal waters, *Estuaries*, 25(4), 677–693, doi:10.1007/BF02804899.
- Parkin, T. B., and J. M. Tiedje (1984), Application of a soil core method to investigate the effect of oxygen concentration on denitrification, *Soil Biol. Biochem.*, 16(4), 331–334, doi:10.1016/0038-0717(84)90027-0.
- Pathak, H., and D. B. Nedwell (2001), Nitrous oxide emission from soil with different fertilizers, water levels and nitrification inhibitors, *Water Air Soil Pollut.*, 129, 217–228, doi:10.1023/A:1010316215596.
- Pett-Ridge, J., and M. K. Firestone (2005), Redox fluctuation structures microbial communities in a wet tropical soil, *Appl. Environ. Microbiol.*, 71(11), 6998–7007, doi:10.1128/AEM.71.11.6998-7007.2005.
- Reay, D. S., A. C. Edwards, and K. A. Smith (2009), Importance of indirect nitrous oxide emissions at the field, farm and catchment scale, *Agric. Ecosyst. Environ.*, 133, 163–169, doi:10.1016/j.agee.2009.04.019.
- Richardson, D., H. Felgate, N. Watmough, A. Thomson, and E. Baggs (2009), Mitigating release of the potent greenhouse gas N₂O from the nitrogen cycle—Could enzymic regulation hold the key?, *Trends Biotechnol.*, 27(7), 388–397, doi:10.1016/j.tibtech.2009.03.009.
- Robertson, G. P., and P. Groffman (2007), Nitrogen Transformations, in *Soil Microbiology, Ecology and Biochemistry*, edited by E. A. Paul, pp. 341–362, Academic, San Diego, Calif.
- Rudaz, A. O., E. Walti, G. Kyburz, P. Lehmann, and J. Fuhrer (1999), Temporal variation in N₂O and N₂ fluxes from a permanent pasture in Switzerland in relation to management, soil water content and soil temperature, *Agric. Ecosyst. Environ.*, 73(1), 83–91, doi:10.1016/S0167-8809(99)00005-5.
- Ruser, R., H. Flessa, R. Russow, G. Schmidt, F. Buegger, and J. C. Munch (2006), Emission of N₂O, N₂ and CO₂ from soil fertilized with nitrate: Effect of compaction, soil moisture and rewetting, *Soil Biol. Biochem.*, 38(2), 263–274, doi:10.1016/j.soilbio.2005.05.005.
- Scheer, C., R. Wassmann, K. Butterbach-Bahl, J. P. A. Lamers, and C. Martius (2008), The relationship between N₂O, NO, and N₂ fluxes from fertilized and irrigated dryland soils of the Aral Sea Basin, Uzbekistan, *Plant Soil*, 314(1–2), 273–283, doi:10.1007/s11104-008-9728-8.
- Schlesinger, W. H. (2009), On the fate of anthropogenic nitrogen, *Proc. Natl. Acad. Sci. U. S. A.*, 106(1), 203–208, doi:10.1073/pnas.0810193105.
- Seitzinger, S., J. A. Harrison, J. K. Böhlke, A. F. Bouwman, R. Lowrance, B. Peterson, C. Tobias, and G. Van Drecht (2006), Denitrification across landscapes and waterscapes: A synthesis, *Ecol. Appl.*, 16(6), 2064–2090, doi:10.1890/1051-0761(2006)016[2064:DALAWA]2.0.CO;2.
- Sexstone, A. J., N. P. Revsbech, T. B. Parkin, and J. M. Tiedje (1985), Direct measurement of oxygen profiles and denitrification rates in soil aggregates, *Soil Sci. Soc. Am. J.*, 49, 645–651, doi:10.2136/sssaj1985.03615995004900030024x.
- Silver, W. L., A. E. Lugo, and M. Keller (1999), Soil oxygen availability and biogeochemistry along rainfall and topographic gradients in upland wet tropical forest soils, *Biogeochemistry*, 44(3), 301–328, doi:10.1007/BF00996995.
- Soil Survey Staff (2006), *Keys to Soil Taxonomy*, 10th ed., U.S. Dep. of Agric. Nat. Resour. Conserv. Serv., Washington, D. C.
- Soosaar, K., U. Mander, M. Maddison, A. Kanal, K. Lohmus, J. Truu, and J. Augustin (2011), Dynamics of gaseous nitrogen and carbon fluxes in riparian alder forests, *Ecol. Eng.*, 37, 40–53, doi:10.1016/j.ecoleng.2010.07.025.
- Steinheimer, T. R., K. D. Scoggin, and L. A. Kramer (1998), Agricultural chemical movement through a field-size watershed in Iowa: Subsurface hydrology and distribution of nitrate in groundwater, *Environ. Sci. Technol.*, 32(8), 1039–1047, doi:10.1021/es970598j.
- Swerts, M., G. Uytterhoeven, R. Merckx, and K. Vlassak (1995), Semi-continuous measurement of soil atmosphere gases with gas flow soil core method, *Soil Sci. Soc. Am. J.*, 59(5), 1336–1342, doi:10.2136/sssaj1995.03615995005900050020x.
- Teh, Y. A., W. L. Silver, and M. E. Conrad (2005), Oxygen effects on methane production and oxidation in humid tropical forest soils, *Global Change Biol.*, 11(8), 1283–1297, doi:10.1111/j.1365-2486.2005.00983.x.
- van Breeman, N., et al. (2002), Where did all the nitrogen go? Fate of nitrogen inputs to large watersheds in the northeastern U.S.A., *Biogeochemistry*, 57, 267–293, doi:10.1023/A:1015775225913.
- van den Heuvel, R. N., M. M. Hefting, N. C. G. Tan, M. S. M. Jetten, and J. T. A. Verhoeven (2009), N₂O emission hotspots at different spatial scales and governing factors for small scale hotspots, *Sci. Total Environ.*, 407(7), 2325–2332, doi:10.1016/j.scitotenv.2008.11.010.
- Vidon, P. (2010), Riparian zone management and environmental quality: A multi-contaminant challenge, *Hydrol. Processes*, 24, 1532–1535, doi:10.1002/hyp.7740.
- Watts, S., and S. P. Seitzinger (2000), Denitrification rates in organic and mineral soils from riparian sites: A comparison of N₂ flux and acetylene inhibition methods, *Soil Biol. Biochem.*, 32(10), 1383–1392, doi:10.1016/S0038-0717(00)00056-0.
- Whalen, S. C. (2005), Biogeochemistry of methane exchange between natural wetlands and the atmosphere, *Environ. Eng. Sci.*, 22(1), 73–94, doi:10.1089/ees.2005.22.73.
- Whiting, G. J., and J. P. Chanton (1993), Primary production control of methane emission from wetlands, *Nature*, 364(6440), 794–795, doi:10.1038/364794a0.
- Woldendorp, J. W. (1962), The quantitative influence of the rhizosphere on denitrification, *Plant Soil*, 17, 267–270, doi:10.1007/BF01376229.
- Yavitt, J. B., and G. E. Lang (1990), Methane production in contrasting wetland sites: Response to organic chemical components of peat and to sulfate reduction, *Geomicrobiol. J.*, 8(1), 27–46, doi:10.1080/01490459009377876.
- Yavitt, J. B., C. J. Williams, and R. K. Wieder (2000), Controls on microbial production of methane and carbon dioxide in three *Sphagnum*-dominated peatland ecosystems as revealed by a reciprocal field peat transplant experiment, *Geomicrobiol. J.*, 17(1), 61–88, doi:10.1080/014904500270503.

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