

Crystal structure of hexapraseodymium(III) trinitride tetrasulfide chloride, Pr₆N₃S₄Cl

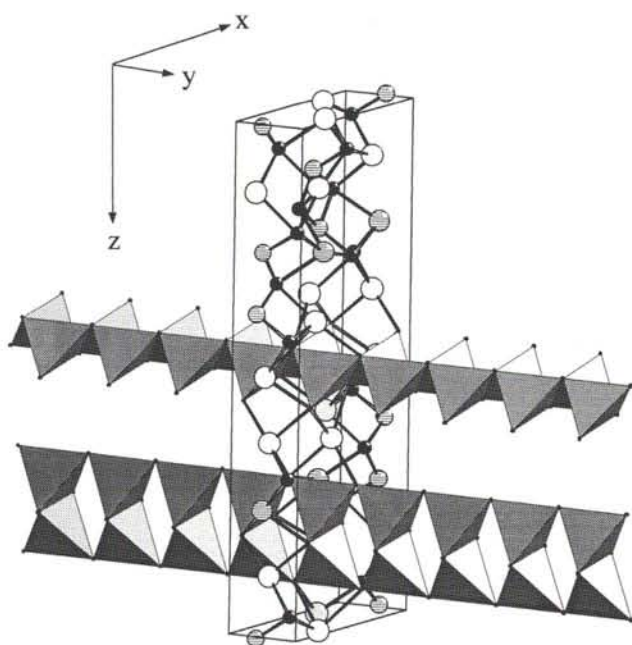
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Abstract

CIN₃Pr₆S₄, orthorhombic, *Pnma* (No. 62), $a = 11.3987(8)$ Å, $b = 4.0235(3)$ Å, $c = 26.999(1)$ Å, $V = 1238.3$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.037$, $R_w(F^2) = 0.079$, $T = 293$ K.

Source of material

Pr₆N₃S₄Cl was prepared from PrCl₃, praseodymium powder, sulfur, and sodium azide (NaN₃) as nitrogen source (molar ratio 2:16:12:3) with small amounts of NaCl or a slight excess of PrCl₃ as fluxing agents [1, 2]. The mixture was heated in a sealed evacuated silica vessel at 1123 K (7 d) and subsequently cooled down (25 K/d) at ambient temperature.

Discussion

Pr₆N₃S₄Cl belongs to the Nd₆N₃S₄Cl-type of structure [1]. The orthorhombic unit cell contains two different types of chains built up by edge- and vertex-sharing N³⁻-centred (Pr³⁺)₄ tetrahedra $\frac{1}{\infty}[(N1)(Pr1)_{3/3}(Pr2)_{1/1}]^{3+}$ and $\frac{1}{\infty}[(Pr5)_{2/2}(N2)(Pr3)_{(1+1)/(1+1)}(Pr4)_{(1+1)/(1+1)}(N3)(Pr6)_{2/2}]^{6+}$, both commensurate in translation along [010], which are depicted in the figure (Pr: black, N: small, S and Cl: large circles) by closed polyhedra [3–7].

Table 1. Data collection and handling.

Crystal:	reddish brown, transparent, lath-shaped needles, size 0.03 x 0.08 x 0.15 mm
Wavelength:	Mo K α radiation (0.7107 Å)
μ :	239.9 cm ⁻¹
Diffractometer, scan mode:	Siemens-Stoe, profile fitted ω -2 θ -scan [11]
2 θ_{max} :	64.4°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	4648, 2350
Criterion for F_{obs} , $N(hkl)_{\text{gt}}$:	$F_{\text{obs}} > 3 \sigma(F_{\text{obs}})$, 1665
$N(\text{param})_{\text{refined}}$:	86
Programs:	SHELXS-86 [8], SHELXL-93 [9], DIAMOND [10]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Pr(1)	4c	0.40104(5)	1/4	0.46236(2)	0.0116(2)	0.0132(3)	0.0138(2)	0	-0.0012(2)	0
Pr(2)	4c	0.24374(6)	1/4	0.06443(2)	0.0138(3)	0.0140(3)	0.0158(2)	0	-0.0023(2)	0
Pr(3)	4c	0.43801(5)	1/4	0.19285(2)	0.0124(3)	0.0146(3)	0.0151(2)	0	-0.0008(2)	0
Pr(4)	4c	0.24533(6)	1/4	0.32661(2)	0.0130(3)	0.0134(3)	0.0138(2)	0	-0.0004(2)	0
Pr(5)	4c	0.44107(6)	1/4	0.92262(2)	0.0126(3)	0.0181(4)	0.0158(2)	0	-0.0010(2)	0
Pr(6)	4c	0.38142(6)	1/4	0.71331(2)	0.0135(3)	0.0142(3)	0.0149(2)	0	-0.0012(2)	0
N(1)	4c	0.0980(8)	1/4	0.0133(3)	0.009(4)	0.017(5)	0.012(3)	0	0.000(3)	0
N(2)	4c	0.0621(9)	1/4	0.3747(3)	0.020(5)	0.004(4)	0.017(4)	0	-0.004(3)	0
N(3)	4c	0.1111(9)	1/4	0.2606(3)	0.018(5)	0.016(5)	0.016(4)	0	-0.004(3)	0
S(1)	4c	0.2041(3)	1/4	0.89640(9)	0.014(1)	0.013(1)	0.016(1)	0	0.0017(9)	0

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
S(2)	4c	0.0431(3)	1/4	0.13630(9)	0.015(1)	0.022(2)	0.016(1)	0	-0.0018(9)	0
S(3)	4c	0.1449(3)	1/4	0.50411(9)	0.013(1)	0.018(2)	0.019(1)	0	0.0001(9)	0
S(4)	4c	0.1900(3)	1/4	0.63676(9)	0.016(1)	0.017(2)	0.014(1)	0	0.0035(9)	0
Cl	4c	0.1374(3)	1/4	0.76448(9)	0.017(1)	0.018(1)	0.017(1)	0	-0.0007(9)	0

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