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Evaluation of The Relationship Between Bulk Organic Precursors and Disinfection Byproduct Formation for Advanced Oxidation Processes

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Abstract: Advanced oxidation processes (AOPs) are gaining traction as they offer mineralization potential rather than transferring contaminants between media. However, AOPs operated with limited energy and/or chemical inputs *Chemosphere*, Vol 121 (February 2015): pg. 39-46. DOI. This article is © Elsevier and permission has been granted for this version to appear in <u>e-Publications@Marquette</u>. Elsevier does not grant permission for this article to be further copied/distributed or hosted elsewhere without the express permission from Elsevier.

can exacerbate disinfection byproduct (DBP) formation, even as precursors such as dissolved organic carbon, UV_{254} , and specific UV absorbance (SUVA) decrease. This study examined the relationship between DBP precursors and formation using TiO₂ photocatalysis experiments, external AOP and non-AOP data, and predictive DBP models. The top-performing indicator, SUVA, generally correlated positively with trihalomethanes and haloacetic acids, but limited-energy photocatalysis yielded contrasting negative correlations. The accuracy of predicted DBP values from models based on bulk parameters was generally poor, regardless of use and extent of AOP treatment and type of source water. Though performance improved for scenarios bounded by conditions used in model development, only 0.5% of the model/dataset pairings satisfied all measured parameter boundary conditions, thereby introducing skepticism toward model usefulness. Study findings suggest that caution should be employed when using bulk indicators and/or models as a metric for AOP mitigation of DBP formation potential, particularly for limitedenergy/chemical inputs.



Graphical abstract

Keywords: Disinfection byproduct; Advanced oxidation; Titanium dioxide photocatalysis; Trihalomethane; Haloacetic acid; Model

1. Introduction

Increasing implementation of advanced oxidation processes (AOPs) makes it imperative to improve understanding of their disinfection byproduct (DBP) mitigation potential, including the extent to which traditional DBP indicators and the predictive DBP formation models based on them reliably predict AOP performance. This study evaluated the relationship between traditional bulk DBP precursors and DBP formation following AOP treatment using pilot-scale TiO₂ photocatalysis experiments, external DBP data (both AOP and non-

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AOP), and published predictive models for total trihalomethanes (TTHMs), haloacetic acids (HAA5s), and subspecies thereof.

Since their discovery in 1974 (Bellar et al., 1974 and Rook, 1974), the production of over 600 potentially harmful DBPs has attracted considerable attention (Krasner et al., 2006). Drinking water DBPs are widely regulated, as described in the Supplementary Information (SI), and increasing interest in water reclamation and reuse has sparked recognition of the occurrence and impacts of DBPs in wastewater (Wei et al., 2009). Chlorine disinfection alternatives such as chloramines, chlorine dioxide, ozone, or ultraviolet (UV) may reduce the formation of trihalomethanes (THMs) and haloacetic acids (HAAs); however, they may still form DBPs, and can shift DBP production toward species of greater health concern (Richardson, 2005). Additionally, alternative disinfectants may be ineffective against certain pathogens, increase nitrification and corrosion in the distribution system, and/or lack disinfectant residual (USEPA, 2006 and Zhang et al., 2008). Thus, the most common DBP reduction strategy is removal of precursor natural organic matter (NOM) prior to disinfection (Kulkarni and Chellam, 2010). The best available techniques – enhanced coagulation or softening, granular activated carbon (GAC), or membrane filtration (Liu et al., 2008b and USEPA, 2006) – are often sufficient to control DBP formation; however, alternatives such as AOPs are also being investigated (Mayer et al., 2014 and WHO, 2000).

The highly reactive, nonspecific radical species, such as HO, produced by AOPs are capable of mineralizing NOM, oxidizing trace emerging contaminants, and inactivating microorganisms (Gerrity et al., 2009). Given their potential to mineralize NOM rather than capturing and transferring it to another phase, implementation of AOPs is becoming increasingly attractive (Gerrity et al., 2009) and Mayer et al., 2014), making it critically important to examine their effect on DBP formation (Dotson et al., 2010). Photocatalysis provides one pathway by which HO radicals are produced using ultraviolet (UV) light ($\lambda < 387$ nm) to irradiate semiconductors such as TiO₂. However, AOPs such as TiO₂ photocatalysis are unlikely to be operated at sufficiently high energy and/or chemical inputs to mineralize NOM. Instead, under more practical operating conditions, AOPs are likely to incompletely

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oxidize organics, resulting in a shift toward smaller, less aromatic, and more hydrophilic moieties (Liu et al., 2008a, Mayer et al., 2014 and Sarathy and Mohseni, 2010), which can exacerbate DBP formation beyond initial levels (Dotson et al., 2010, Gerrity et al., 2009, Liu et al., 2008a and Mayer et al., 2014). For example, 5 kW h m⁻³ TiO₂/UV AOP treatment increased TTHMs up to 119% and HAA5s up to 299% when applied to waters of varying quality, ranging from untreated to finished drinking water (Gerrity et al., 2009 and Mayer et al., 2014). When higher AOP energy/chemical inputs are used, more complete NOM mineralization follows, resulting in decreased DBP formation, e.g., reductions of up to 95% TTHMs and 96% HAA5s (Gerrity et al., 2009 and Mayer et al., 2014). The initial trend of increasing DBP production using limited-input AOP treatments contrasts with decreases in traditional bulk DBP precursors such as dissolved organic carbon (DOC), ultraviolet absorbance at a wavelength of 254 nm (UV_{254}), and specific UV absorbance (SUVA). This suggests that traditional bulk indicators of DBPs may not provide suitable performance metrics for AOPs, which fundamentally augment NOM characteristics rather than physically removing it.

2. Materials and methods

2.1. TiO₂ photocatalysis

Samples were collected from several locations in Central Arizona drinking water treatment trains using a mixture of surface and groundwater, including following sedimentation (SW), following GAC filtration (GAC), and from the distribution system (DS). Details of the treatment trains are provided in the <u>SI</u>. The samples were used as the influent water for pilot-scale TiO₂ photocatalysis experiments.

Photocatalysis was performed using the Purifics (London, ON) Photo-Cat® Lab reactor at an optimized concentration of 1 g L⁻¹ (<u>Gerrity et al., 2009</u> and <u>Liu et al., 2008b</u>) suspended reagent-grade Degussa P25 TiO₂ (Dusseldorf, Germany). The reactor was operated in batch configuration, as described previously (<u>Gerrity et al.,</u> <u>2009</u> and <u>Mayer et al., 2014</u>). For each source water, dark adsorption (no UV) samples were collected to provide a measure of physical removal due to adsorption and retention on the Photo-Cat's®

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submicron-pore-size ceramic membrane filter. Additionally, three levels of treatment were tested for each water: limited (5 kW h m⁻³), medium (80 kW h m⁻³), and extended photocatalysis (160 kW h m⁻³), providing a spectrum of partial, incomplete, and near-complete NOM oxidation (<u>Mayer et al., 2014</u>).

2.2. Water quality parameters

Following photocatalysis, turbidity was measured using a Hach (Loveland, CO) model 2100P turbidimeter, and a Mettler meter (Columbus, OH) was used for pH. Traditional DBP indicators quantified included DOC, UV₂₅₄, SUVA, and bromide (Br⁻). Particulate matter was removed from untreated samples using 0.45 μ m Pall (Port Washington, NY) Acrodisc® GHP membrane filters. DOC samples were acidified with 1 M HCl, and analyzed using a Shimadzu (Kyoto, Japan) 5050A Total Organic Carbon Analyzer. A Hach DR 5000 spectrophotometer was used to measure UV₂₅₄, and SUVA was calculated by normalizing UV₂₅₄ values with respect to DOC. A Dionex (Sunnyvale, CA) DX-120 ion chromatograph was used to measure Br⁻.

2.3. Disinfection byproduct formation

Treated samples were chlorinated in accordance with a modified simulated distribution system test (SDS-THM), which targeted a chlorine residual of 1 mg L⁻¹ after 24 h at 28 °C (<u>APHA</u>, 2005 and <u>Mayer et al.</u>, 2014). A Hach D4/4000U spectrophotometer and DPD free chlorine reagent powder pillows were used to analyze free chlorine. The dominant weight-based classes of chlorine-DBPs, TTHMs and HAA5s (<u>Krasner et al.</u>, 2006), were measured as described by <u>Mayer et al.</u> (2014). To gauge the evolution of DBP formation over time, samples were collected at 10–15 min, 1 h, 2 h, and 6 h post-chlorination. 24 h DBP formation was reported by <u>Mayer et al.</u> (2014).

2.4. External DBP datasets

In addition to internal DBP data from this study, external AOP and non-AOP datasets of TTHMs, HAA5s, and/or subspecies were collected from published literature to provide additional data to investigate the relationship between precursors and DBPs and to

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evaluate DBP model performance. This external data also functioned as a control to ensure that study findings were not artifacts of the laboratory protocol used in the current study. Twenty external datasets encompassing a range of experimental parameters were selected, as detailed in the <u>SI</u>.

2.5. DBP models

Internal and external DBP data were compared with DBP predictions using 149 published chlorine-DBP models, as described in the <u>SI</u>. These models predicted mass concentrations of TTHMs, HAA5s, and/or subspecies using traditional DBP indicators: DOC, UV₂₅₄, SUVA, chlorine dose (D), pH, temperature (T), Br⁻, and disinfection contact time (t). Models incorporating other parameters were not included as this information was not readily available for the datasets analyzed here. Additionally, this approach to model selection supported the study objective of evaluating the relationship between common bulk indicators of DBPs and measured DBP formation. This dictates that the predictive results reported here are purely empirically-derived due to the exclusion of semi-mechanistic models.

2.6. Statistics

GraphPad Prism was used to compute all statistics at a significance level of a = 0.05. Normality was tested using the D'Agostino-Pearson omnibus test. The relationship between DBP indicators and DBP formation was assessed using the Pearson product-moment coefficient of correlation for normally distributed datasets, while the nonparametric Spearman rank correlation coefficient was used for non-normal distributions. To assess similarities between the means of groups, multiplicity-adjusted *P* values were calculated using one-way ANOVA for normally distributed groups, while the nonparametric Kruskal–Wallis one-way analysis of variance was used for non-normal distributions. For groups with statistical differences, post-hoc tests were used to identify dissimilar datasets by comparing each mean to all other means using Tukey's Honestly Significant Difference (HSD) test following ANOVA or the nonparametric Dunn's test following Kruskal–Wallis tests.

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3. Results and discussion

Initial source water quality differed among samples, reflecting the range of treatments applied prior to photocatalysis, as shown in <u>Table 1</u>.

Table 1. Water quality parameters for TiO₂ photocatalysis experiments.

Water source	Treatment level	pН	Turbidity (NTU)	DOC (mg L ⁻¹)	UV ₂₅₄ (cm ⁻¹)	SUVA (L(mg m) ⁻¹)	Br- (µg L ⁻¹)	24-h TTHMs (µg L ⁻¹)	24-h HAA5s (µg L ⁻¹)	24-h Specific TTHMs (µg L ⁻¹)	24-h Specific HAA5s (µg L ⁻¹)
CAPª	Untreated	8.50	2.20	5.70	0.046	0.81	109	-	-	-	-
SRª	Untreated	8.02	2.86	4.85	0.086	1.78	25	-	-	-	-
SW	Control	8.08	0.90	3.13	0.039	1.26	10	52.5	19.2	16.8	6.12
	Dark adsorption	7.75	0.20	2.70	0.029	1.06	0	62.4	28.9	23.1	10.72
	5 kWh m ⁻³	7.73	0.13	2.51	0.019	0.74	10	109	44.4	43.4	17.71
	80 kWh m ⁻³	8.43	0.23	0.54	0.003	0.61	10	11.0	4.04	20.4	7.48
	$160 \text{ kWh} \text{ m}^{-3}$	7.49	0.07	0.79	0.003	0.34	10	3.43	3.00	4.3	3.80
GAC	Control	8.31	0.39	2.09	0.032	1.51	32	43	18.7	20.6	8.93
	Dark adsorption	8.02	0.21	1.56	0.014	0.90	0	33.8	15.9	21.6	10.21
	5 kWh m ⁻³	8.06	0.21	1.56	0.009	0.55	2	60.3	29.0	38.7	18.60
	80 kWh m ⁻³	8.31	0.20	0.22	0.000	0.00	0	7.20	2.71	32.7	12.32
	160 kWh m ⁻³	8.44	0.16	0.45	0.000	0.00	0	1.92	0.00	4.3	0.00
DS	Control	8.42	0.09	3.02	0.013	0.44	0	29.6	8.97	9.8	2.97
	Dark adsorption	8.32	0.10	1.59	0.015	0.97	0	57.4	28.3	36.1	17.80
	5 kWh m ⁻³	8.28	0.13	1.94	0.014	0.73	0	64.8	35.8	33.9	18.74
	80 kWh m ⁻³	8.51	0.11	0.60	0.003	0.50	0	36.8	18.8	61.3	31.35
	160 kWh m ⁻³	8.38	0.09	0.44	0.002	0.45	0	17.6	8.76	40.0	19.91
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^aUntreated water from the Central Arizona Project Canal (CAP) and the Salt River (SR) were not tested, but provide a basis of comparison for water quality (and level of treatment) of the SW, GAC, and DS samples.

3.1. DBP precursors

Increasing removal of the bulk organic DBP precursors DOC, UV_{254} , and SUVA was observed following photocatalytic oxidation, as shown in <u>Table 1</u>. In comparison to DOC, faster and greater overall removal of UV_{254} , and hence SUVA, occurred. This signifies a progressive shift in NOM toward smaller, less aromatic species, whereas DOC reduction is only achieved following physical removal (dark adsorption), extended treatment due to more complete mineralization, or not at all due to refractory compounds, in agreement with previous reports (<u>Gerrity et al., 2009</u>, <u>Huang et al.</u>,

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<u>2008</u>, <u>Liu et al., 2008a</u> and <u>Mayer et al., 2014</u>). Changes in the NOM size distribution (quantified using high-performance size exclusion chromatography) and spectra (quantified using regional integration of excitation–emission matrix fluorescence spectroscopy) in response to limited energy photocatalytic treatment demonstrated shifts toward smaller, less aromatic, and less humic moieties, as reported by <u>Mayer et al. (2014)</u>.

Concentrations in excess of 100 μ g L⁻¹ Br⁻ trigger concern about the formation of brominated DBPs (<u>Zhang et al., 2011</u>), which exhibit higher toxicity compared to their chlorinated counterparts (<u>Plewa et</u> <u>al., 2002</u>). However, chlorinated species tend to dominate in low Br⁻ waters (<u>Richardson et al., 1999</u>). In this study, a maximum of 32 μ g L⁻¹ Br⁻ was observed, effectively limiting the formation of brominated DBPs.

3.2. DBP formation and relation to precursors

As illustrated for the SW in Fig. 1, DBP evolution is characterized by rapid initial formation followed by a slower rate of generation several hours post chlorination (Gallard and von Guten, 2002). Fig. 1 also illustrates that dark adsorption and limited photocatalysis exacerbate DBPs beyond untreated control samples. Adsorption of higher molecular weight, hydrophobic NOM to TiO₂ surfaces can shift organic profiles toward more reactive species (Mayer et al., 2014), as shown by the increase in the specific DBP values shown in Table 1. Similar findings using industry-relevant energy and chemical inputs have been observed for TiO₂ photocatalysis (Gerrity et al., 2009, Liu et al., 2008b and Mayer et al., 2014) and other AOPs (Dotson et al., 2010, Kleiser and Frimmel, 2000 and Metz et al., 2011). This appears to stem from HO, radicals, and is not observed in non-AOP systems, e.g., UV alone (Kleiser and Frimmel, 2000). Incomplete NOM oxidation resulting from limited-input AOPs opens the rings of aromatic structures and cleaves double-bonded carbon structures, thereby yielding more reactive sites, which can increase DBP formation (Metz et al., 2011). Given sufficient energy/chemical inputs, more complete oxidation of NOM significantly reduces DBP concentrations.

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Fig. 1. Time evolution of (a) TTHM and (b) HAA5 species following chlorine disinfection in the SW as a function of time and extent of photocatalytic treatment. As shown in the <u>SI</u>, the GAC and DS data followed similar patterns, albeit with lower overall DBP formation.

The contrasting trend of decreasing bulk organic DBP precursors and increasing DBP formation using limited photocatalysis is apparent in <u>Table 1</u>. Limited-energy inputs correspond to spikes in specific TTHMs and HAA5s, which are normalized to DOC. This indicates that although the concentration of DOC in the system did not decrease substantially using 5 kW h m⁻³, its DBP formation potential increased, signifying fundamental shifts toward more reactive moieties. Although contrasting DBP precursor/formation trends were observed for limited photocatalysis, linear correlation coefficients indicated that DOC, UV₂₅₄, and SUVA generally exhibited positive correlations with DBP formation, *Chemosphere*, Vol 121 (February 2015): pg. 39-46. DOI. This article is © Elsevier and permission has been granted for this version to appear in <u>e-Publications@Marquette</u>. Elsevier does not grant permission for this article to be further copied/distributed or hosted elsewhere without the express permission from Elsevier. as shown in Fig. 2. Of these NOM-related parameters, SUVA correlated positively with DBPs most consistently, with the major exception of the 5 kW h m⁻³ dataset. For this dataset, the correlations between precursors and DBPs varied widely, even within individual parameter groupings. Notably, the majority of the DBP species were negatively correlated with SUVA at 5 kW h m⁻³ (though CF, DBCM, and DBA demonstrated strong positive correlations), whereas nearly all SUVA correlations were positive for the other treatment conditions. This supports the hypothesis that the incomplete oxidation of organics caused by limited-input AOPs alters NOM character such that traditional bulk measures such as SUVA are less reliable indicators of DBP formation compared to their effectiveness in physical removal scenarios such as enhanced coagulation or GAC filtration. The altered NOM composition resulting from limited-input AOPs exhibits different chlorine reactivity, which is not adequately conveyed by bulk properties such as SUVA, suggesting that caution is warranted when using bulk NOM precursors as indicators of AOP performance. With additional energy input, e.g., 80 and 160 kW h m⁻³, organics are more completely oxidized, DOC and UV₂₅₄ decrease, and SUVA demonstrates a strong positive correlation to DBP formation.



Fig. 2. Linear correlation coefficients (Pearson for normally distributed datasets, otherwise Spearman) between DBP formation and precursor measurements. Values closer to +1 indicate positive correlations, while values closer to -1 indicate negative correlations. Panels (a) through (e) are data from the current study, grouped by extent of photocatalytic treatment, while panel (f) illustrates correlations for all datasets combined, both internal and external.

Fig. 2 shows that DBP formation was also generally positively correlated with D and *t*, while Br⁻ and pH demonstrated mixed positive and negative associations with DBPs. A positive correlation between increasing pH and THMs is commonly reported, while HAAs are often *Chemosphere*, Vol 121 (February 2015): pg. 39-46. DOI. This article is © Elsevier and permission has been granted for this version to appear in <u>e-Publications@Marquette</u>. Elsevier does not grant permission for this article to be further copied/distributed or hosted elsewhere without the express permission from Elsevier.

negatively correlated (<u>WHO, 2000</u>). Full statistical results are provided in the <u>SI</u>.

3.3. DBP modeling assessment

3.3.1. Internal data

Comparisons of internal data and predicted TTHMs (combined results of 47 models) and HAA5s (combined results of 8 models) are illustrated in Fig. 3. Linear regression equations were used to gauge the overall accuracy of model predictions. The low R^2 values of 0.17 and 0.16 for TTHMs and HAA5s, respectively, signify poor overall model accuracy. Notably, these regressions are the combined measure of model fit, whereas the average R^2 values for individual TTHM models applied to this dataset was 0.58, with a range of 0.00–0.87. The average R^2 from individual HAA5 models was 0.15, with a range of 0.01–0.27. The R^2 values for all assessments are provided in the <u>SI</u>.

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Fig. 3. Predicted versus measured DBP concentrations for all photocatalytic data from the current study. The predicted values are the results of (a) 47 independent published TTHM models and (b) 8 independent published HAA5 models. The insets show an enlarged view of data points up to 20 μ g L⁻¹ measured DBPs.

In addition to $R^2 = 1.0$, a perfect fit would produce slope = 1.0 and intercept = 0. Fig. 3 indicates that the models tend to over-predict at low DBP concentrations and under-predict at high concentrations (slopes <1 and intercepts >0). This suggests that empirical models commonly over-predict for conditions least conducive to DBP formation and under-predict for those most conducive, consistent with individual modeler reports (<u>Amy et al., 1987</u>, <u>Amy et al., 1998</u> and <u>Sohn et al.</u>,

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<u>2004</u>). One possible explanation for this over/under-estimate trend is that these "extreme" formation conditions are more likely to extend beyond model boundary conditions, producing inherently less accurate predictions due to extrapolation, a practice which is typically discouraged (<u>Amy et al., 1987</u>).

The coefficients of determination (R^2) from comparisons of predicted DBP model values versus measured DBP formation for the internal data, grouped by level of photocatalytic treatment, are shown in Fig. 4a. ANOVA and post-hoc tests determined that model accuracy differed across all levels of treatment with the exception of the 80 and 160 kW h m⁻³ pairing. This suggests that model accuracy (a gauge of the predictive relationship between bulk precursors and DBP formation) differs as a function of photocatalytic treatment, i.e., the degree of NOM oxidation. Contrary to the hypothesis, however, incomplete oxidation at the 5 kW h m⁻³ treatment level, which is characterized by contradictory trends between bulk DBP indicators and DBP formation, exhibited the highest degree of overall model accuracy. The same trend was apparent with the HAA dataset, although statistics indicated that only the Dark Adsorption and 5 kW h m⁻³ pair differed significantly in model accuracy. Full statistical results are provided in the SI.

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Fig. 4. Box and whisker plots showing the coefficients of determination (R^2) for predicted versus measured DBP values (including TTHM, HAA5, and subspecies thereof) for (a) photocatalysis-treated samples from the current study and (b) all internal and external datasets grouped by source water type. The whiskers denote the maximum and minimum values in the dataset exclusive of outliers, which were identified as data points located more than 1.5 times the interquartile range (IQR) beyond the IQR.

3.3.2. Source water

The internal and external DBP datasets analyzed here included water from varying sources, including untreated, partially treated, fully *Chemosphere*, Vol 121 (February 2015): pg. 39-46. DOI. This article is © Elsevier and permission has been granted for this version to appear in <u>e-Publications@Marguette</u>. Elsevier does not grant permission for this article to be further

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treated, and synthetic waters. These differences may impact model performance as the majority of DBP models were developed using untreated water, and as such may be less accurate for partially to fully-treated source waters, which vary in quantity and character of NOM.

The *R*² values from predicted DBP model values versus internal and external DBP measurements, grouped by type of source water, are shown in Fig. 4b. Statistical analyses showed that the accuracy of the predictive THM and HAA models differed as a function of source water. Post-hoc tests determined that THM model accuracy for the untreated and coagulated waters was significantly different from all other source waters. For HAAs, model accuracy differed across all water types with the exception of the settled and finished pair and filtered and finished pair. These findings suggest greater model accuracy is achieved when using untreated or coagulated waters, which are also the water sources used to develop the majority of the models assessed in this study. Additionally, modeling capabilities using synthetic samples were guite poor, possibly the result of more homogenous chemical compositions than those typically encountered in natural waters, for which the models were developed. Full statistical analyses are provided in the <u>SI</u>.

3.3.3. AOP versus non-AOP

The consolidated results of the 33 348 THM and 4773 HAA pairs of predicted DBPs versus internal and external data are shown in Fig. <u>5</u>. Each box depicts the distribution of R^2 values from the predicted versus measured DBPs for a given dataset. Not all models yielded meaningful predictions for all datasets as a result of missing input data, inability to calculate R^2 due to model returning identical values for all scenarios, etc. Analysis of variance and post hoc tests identified differences in HAA model performance between the TiO₂ photocatalysis and non-TiO₂ AOP datasets. No statistical difference was found in the accuracy of model predictions among the THM datasets. Full statistical analyses are provided in the <u>SI</u>.

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Fig. 5. Box and whisker plots of the coefficients of determination (R^2) for predicted versus measured THM and HAA values (including TTHM, HAA5, and subspecies thereof). Values are calculated from a maximum of 105 THM models (48 HAA models) for 31 different AOP and non-AOP THM datasets (16 HAA data sets). Data points located more than 1.5 times the interquartile range (IQR) outside the IQR are identified as outliers. The whiskers denote the maximum and minimum values in the dataset exclusive of outliers.

Fig. 3, Fig. 4 and Fig. 5 demonstrate generally poor performance by the predictive DBP models, regardless of the source water and/or treatment applied. This alludes to deeper challenges in the application of DBP models to external datasets beyond simply suggesting that caution be used when employing bulk indicators of DBP formation as a metric of AOP performance. Limitations of some existing models include calibration using a limited database, applicability to a specific water source or group thereof, disregard of important parameters such as reaction time, inadequate validation using independent external data, and/or inclusion of terms not

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typically measured such as site-specific parameters (<u>Amy et al.,</u> <u>1987</u>).

One of the key implications of the empirical nature of most models is the introduction of specific boundaries related to source water, water quality parameters, and treatment conditions. Most models were developed using untreated, coagulated, or finished conventionally-treated water. As such, satisfactory model performance may be limited to a narrow range of treatment scenarios. Accordingly, the overall poor performance of the models tested here may be a function of applying them to datasets that did not satisfy all boundary conditions. Unfortunately, strict agreement with all model boundary conditions is infrequently encountered (Amy et al., 1987). Of the 2085 THM and 557 HAA model/dataset combinations evaluated, only 12 (0.5%) combinations satisfied all constraints of model parameter boundary conditions. For these 12 datasets, the average R^2 value was 0.59 (ranging from 0.06 to 0.87), which supports the expectation that improved model accuracy is achieved for data within the boundary conditions. Notably, the majority of the datasets satisfying the boundary conditions were for models that did not report all conditions used to develop specific parameters, particularly Br⁻.

Based on this analysis, it appears that many models are so specific in their application, based on the boundary conditions implemented during model development, that extrapolation is necessary for the vast majority of water systems. The inaccuracy this introduces brings into question the usefulness of many of these models outside of the specific scenario for which they were derived. Furthermore, this study suggests that caution should be employed when using models and/or bulk organic indicators such as DOC, UV₂₅₄, and SUVA to assess AOP performance, even in cases of qualitative comparison, and specifically when operating under limited-input energy and/or chemical conditions. More accurate predictions of AOP performance can be expected through development of site-specific mechanistic (or at least semi-mechanistic) models which directly account for hydroxyl radical interactions with organic matter.

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Appendix A. Supplementary material

1	
23	Supplementary Information
4	Evaluation of the relationship between bulk organic precursors and disinfection
5	byproduct formation for advanced oxidation processes
6	
7 8	Brooke K. Mayer ^{1*} , Erin Daugherty ² , and Morteza Abbaszadegan ²
9	
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17	The supplementary information consists of 67 pages, including 3 figures and 18 tables.
18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33	 Supplementary Information The Supplementary Information provides additional relevant details of this study, including: List of several significant disinfection byproduct (DBP) regulations Description of treatment trains for source waters used in photocatalysis experiments Identification of all external DBP datasets mined from the literature and any assumptions employed when using them List of all DBP models and the conditions under which they were developed Illustrations of the evolutions of DBPs as a function of time and extent of photocatalytic treatment Linear correlation coefficients for datasets grouped by type of source water Table of all coefficient of determinations (R²) calculated for model/dataset pairings Full numerical statistics for tests of normal distributions, analysis of variance, and post hoc analyses.
34 35 36 37 38 39	1. Regulated Disinfection Byproducts The 20 th century heralded substantial public health improvements through the advent of widespread drinking water disinfection. Yet, in the last several decades, concern over the production of disinfection byproducts (DBPs) formed through the interactions of oxidizing disinfectants and natural organic matter (NOM) has stimulated increasing regulations on levels of residual disinfectants in distribution systems as well as maximum

regulations on levels of residual disinfectants in distribution systems as well as maximum contaminant levels for DBPs. Table S1 lists several current DBP regulations established by

41 regulatory agencies.

Species	Chemical	Shorthand	Main	U.S.	Health	European	Australia ^d	World
	Formula	Notation	Causative	Environmental	Canada	Union		Health
			Disinfectant	Protection	(HC) [♭]	(EU) ^c		Organization
				Agency				(WHO) ^e
				(USEPA) ^a				
Total Trihalomethanes		TTHMs	Chlorine	80	100	100	250	f
Chloroform	CHCl₃	CF	Chlorine					300
Bromodichloromethane	CHBrCl ₂	BDCM	Chlorine		16			60
Dibromochloromethane	CHBr ₂ CI	DBCM	Chlorine					100
Bromoform	CHBr ₃	BF	Chlorine					100
Haloacetic Acids		HAA5s	Chlorine	60	80			
Monochloroacetic Acid	$C_2H_3CIO_2$	CAA	Chlorine				150	20
Dichloroacetic Acid	$C_2H_2CI_2O_2$	DCA	Chlorine				100	50
Trichloroacetic Acid	$C_2HCI_3O_2$	TCA	Chlorine				100	200
Monobromoacetic Acid	$C_2H_3BrO_2$	MBA	Chlorine					
Dibromoacetic Acid	$C_2H_2Br_2O_2$	DBA	Chlorine					
Bromate	BrO₃		Ozone	10	10	10	20	10
Chlorite	CIO ₂ -		Chlorine	1,000	1,000		800	700
			dioxide					
Chlorate	CIO3-		Chlorine		1,000			700
			dioxide					
Chloral hydrate	$C_2H_3CI_3O_2$	СН	Chlorine				20	
(Trichloroacetaldehyde)								
Chlorophenols			Chlorine				0.1	
2-Chlorophenol	C ₆ H ₅ ClO		Chlorine				300	
2,4-Dichlorophenol	$C_6H_4Cl_2O$		Chlorine				200	
2,4,6-Trichlorophenol	C ₆ H ₃ Cl ₃ O		Chlorine		5		20	200
Cyanogen Chloride (as	CNCI		Chloramines				80	70
cyanogen)								
Formaldehyde	CH ₂ O		Chlorine,				500	
			Ozone					
Haloacetonitriles		HANs	Chlorine					
Dichloroacetonitrile	C ₂ HCl ₂ N	DCAN	Chlorine					20
Dibromoacetonitrile	C ₂ HBr ₂ N	DBAN	Chlorine					70
N-Nitrosodimethylamine	$C_2H_6N_2O$	NDMA	Chloramines		0.04			0.1
^a USEPA, 2009		^d NHMRC	C and NRMMC, 20	011	f The s	sum of the ra	atio of the co	ncentration of
^b Health Canada, 2012		e WHO, 2	2008		each	to its respec	ctive guidelin	e value
^c EU, 1998					shou	ld not excee	d 1 _	

42 Table S1 – Regulated disinfection byproducts (all concentrations in µg L⁻¹)

43

44 2. Source Water for Pilot-scale TiO₂ Photocatalysis Experiments

45 As described by Mayer et al. (2014), the settled water (SW) was collected following 46 coagulation using approximately 12 mg L^{-1} alum (Al₂(SO₄)₃), flocculation, and sedimentation 47 with a hydraulic residence time of 97 min. The granular activated carbon-filtered water 48 (GAC) was collected from a direct filtration system consisting of microfiltration followed by 49 GAC filtration operated at a loading rate of 2.4 qpm ft⁻² and an empty bed contact time of 50 17 min. The distribution system water (DS) was collected from a point in the direct 51 filtration plant's distribution system with a water age of approximately 45 hr and a residual 52 chlorine concentration of 0.3 mg L^{-1} .

53 **3. DBP Datasets**

54 The sources for the 20 external DBP datasets mined from the literature are described 55 in Tables S2 and S3. The data included TiO_2 photocatalysis, non- TiO_2 advanced oxidation 56 processes (AOPs), and non-AOP treatments. The source water and water quality 57 parameters characterizing the datasets encompass a wide range of values in an effort to 58 evaluate effectiveness of model predictions over a wide range of scenarios. As noted by 59 Amy et al. (1987), the ideal literature source in which all relevant modeling parameters, 60 operating conditions, and resulting DBP concentrations were clearly identified and 61 quantified, was rarely encountered. Thus, rational assumptions were made in an attempt to 62 faithfully represent the data. Notations are made in the table regarding all data-related 63 assumptions. These assumptions introduced inherent inaccuracies through estimates of 64 graphical presentations, assumptions for parameters and operating conditions not clearly 65 identified and/or quantified, etc. Notably, bromide concentrations were often unreported. 66 When possible, external literature searches were used to quantify typical bromide content of 67 the source water. In several cases, the bromide concentration was reportedly negligible (or 68 assumed to be based on limited influence of saline water). In these scenarios, the bromide 69 ion concentration was recorded as 1 μ g L⁻¹, which is negligible, but enables model 70 computations that would otherwise return a null output using bromide ion concentrations of 71 zero. Similar approaches are reported by Amy et al. (1998) and Chen and Westerhoff

72 (2010).

Source	Process Description	Source Water	DBP	Comments
Current Study	TiO ₂ photocatalysis using	Settled, GAC-filtered, and	TTHM,	For UV_{254} below the detection limit, a value
	low-pressure UV, 1 g L ⁻¹	distribution system water	HAA5,	of 0.00001 cm ⁻¹ was used to avoid null
	TiO ₂	from central AZ, U.S.	subspecies	model outputs. ^b
(Chin and	O ₃ , UV, UV/ O ₃	Seymour Reservoir, BC,	CF, HAA5	Based on reported 8-d chlorine residual of 10
Bérubé, 2005)	(low-pressure UV lamp)	Canada		- 20 mg L ⁻¹ , estimated excess D = 25 mg L ⁻¹ . a, b, c
(Dotson et al., 2010)	Low-pressure and medium-	Sand and GAC-filtered water from Cincinnati, OH, U.S.	TTHM	Br ⁻ and pH were assumed to stay constant through AOP treatment. ^{a,c}
(Gerrity et al	TiO_2 photocatalysis using	Raw surface water from	ТТНМ.	For UV ₂₅₄ below the detection limit, a value
2009)	low-pressure UV, 1 g L^{-1}	central Arizona, U.S.	subspecies	of 0.00001 cm ⁻¹ was used to avoid null
	TiO ₂			model outputs. ^b
(Kent et al.,	TiO ₂ photocatalysis using	French River Water, NS,	ттнм,	Residual chlorine after 24 hr target of 1 mg
2011)	low-pressure UV, 1 g L ⁻¹	Canada	HAA5,	L ⁻¹ , so D was estimated as being equal to
	sputtered or suspended TiO ₂		subspecies	DOC. Temperature not given; assumed to
				be 20°C. Brookman et al. (2011) reported no
				Br ⁻ in source water. ^{a, b}
(Kleiser and	O_3/H_2O_2 , UV/H_2O_2	River Ruhr, Germany	TTHM	Temperature not given; assumed to be
Frimmel, 2000	(low-pressure UV lamp)			20°C. ^{a, b}
(Lamsal et al.,	$UV, O_3/H_2O_2, UV/H_2O_2,$	French River, Nova Scotia,	TTHM,	Residual 24 hr chlorine target of 1 mg L ⁻¹ , so
2011)	O ₃ /UV using low-pressure	Canada	HAA5	D was estimated as being equal to DOC.
	UV			I wo non-reported DOC values were
				estimated using nearby data. Brookman et
(Live at al	TiQ shata aatalusia usia a	Cumthatia humaia a sid		al. (2011) reported no Br In source water.
(Liu et al.,	Holdight blue fuereseet	Synthetic – numic acid	CF	Followed Method 5/10D. Reported a target
20088)	blacklight blue hubrescent			residual of 3 – 5 mg L \pm after 7 d. Liu et al.
	$ampiped UV(TiO_2, and a combined UV(TiO_2, d))$			$(2000D, 2007)$ used $D \approx 25$ Hig/L, so this
				value was assumed. ^{3/3}
(Liu ot al		Myponga Posonyoir in South	ттнм	D reported was $20 - 30$ mg l^{-1} so used
(Liu et al., 2008b)	using blacklight blue		наа5	D reported was $20 - 50$ mg L ⁻⁷ , so used
20000)	fluoroscont Jamps 0,1, g L ⁻¹	Australia	TIAAS	used value reported by Chow et al. (2004)
	TiO_{2} and combined LIV			for the Myponga Peservoir ^a
	TiO_2/H_2O_2 treatment			
(Liu et al		Surface water from the	ттнм	D reported was $20 - 30$ mg l ⁻¹ so used
2010)	blacklight blue fluorescent	Myponga Reservoir and Coni		average of 25 mg L^{-1} Br ⁻ not reported so
2010)	lamps $0.1 \text{ g} \text{ l}^{-1} \text{ Ti} \Omega_2$	Hollow Australia		used value reported by Chow et al. (2004)
		Honow, Australia		for the Myponga Reservoir.
(Mayer et al	TiO ₂ photocatalysis using	Settled, GAC-filtered, and	TTHM.	For UV ₂₅₄ below the detection limit, a value
2014)	low-pressure UV, 1 a L^{-1}	distribution system water	HAA5,	of 0.00001 cm ⁻¹ was used to avoid null
,	TiO ₂	from central AZ, U.S.	subspecies	model outputs. ^b

73 **Table S2 – Description of advanced oxidation process (AOP) DBP datasets used in this comparative study**

Source	Process Description	Source Water	DBP	Comments
(Philippe et al.,	TiO ₂ photocatalysis using	Raw water from the Severn	TTHM	D was equal to 5xs DOC, which was read
2010a)	medium-pressure UV,	Trent region of the UK		from plots. No Br was reported, so value
	1 g L ⁻¹ TiO ₂			reported by Brown (2009) was used. ^a
(Philippe et al.,	TiO_2 photocatalysis using	Synthetic – amino acids	CF	D was equal to 5xs DOC, which was read
2010b)	medium-pressure UV,			from plots. ^{a, b}
	1 g L ⁻¹ TiO ₂			
(Liu et al.,	UV/H_2O_2 , UV/Fe^{+3} ,	Synthetic – humic acid	TTHM	Reported excess chlorine dose. Liu et al.
2007)	$UV/H_2O_2/Fe^{+3}$ using			(2008b, 2007) used D = 25 mg L ⁻¹ , so this
	blacklight blue fluorescent			value was assumed. ^{a, b}
	lamps			
(Sarathy and	Low-pressure, high output	Surface water from the	TTHM,	a, b, c
Mohseni, 2010)	lamp for UV/H_2O_2 with	Capilano Reservoir, BC,	HAA5	
	15 mg L ⁻¹ H ₂ O ₂	Canada		
^a At least some va	alues estimated from graphical	data presentations.		
^b Bromide reporte	ed as zero or assumed to be neg	gligible, so concentration of 1 µg	g L ⁻¹ used to	avoid null model outputs.
^c TOC concentrati	ons reported rather than DOC.			

Table S3 – Description of non-AOP DBP datasets used in this comparative study

Source	Process Description	Source Water	DBP	Comments					
(Chowdhury et al., 2010)	Untreated water	Synthetic water samples of differing characteristics	TTHM, subspecies	а					
(Elshorbagy, 2000)	Finished (desalinated) water	Field samples from Abu-Dhabi, UAE	ттнм	a, c					
(Fabbricino and Korshin, 2009)	Untreated water	Seawater from Naples, Italy	HAA5, subspecies	Reported in µmol L-1 and converted to µg L-1.a					
(Gang et al., 2003)	Raw and treated water	Surface water from the Mississippi River, Garden City, Maysville, and Lake Vandalia, U.S.	ТТНМ	a,b					
(Hua and Reckhow, 2008)	Raw water	Surface water from Cambridge, MA, U.S.	ттнм	а					
(Zhang et al., 2011)	Raw water	Surface water from China	ттнм						
a At least some values estimated from gra	a At least some values estimated from graphical data presentations.								
b Bromide reported as zero or assumed to	b Bromide reported as zero or assumed to be negligible, so concentration of 1 µg L-1 used to avoid null model outputs.								
c TOC concentrations reported rather than	1 DOC								

78 **4. DBP Models**

79 In recent decades, hundreds of predictive DBP formation models have been 80 published (refer to reviews by Chowdhury et al. (2009) and Sadig and Rodriguez (2004)). 81 They have been developed to qualify the relative significance of water quality and 82 operational disinfection parameters, investigate kinetics, provide alternatives to field 83 monitoring of DBPs, aid in assessments of exposure and health risk, and support 84 comparative economic/performance decisions (Amy et al., 1987; Hong et al., 2007; Sadig 85 and Rodriguez, 2004). The sheer number of models exemplifies the challenge of attempting 86 to develop a universally applicable model (Golfinopoulos and Arhonditsis, 2002). 87 Mechanistic DBP models are exceedingly difficult to derive due to seasonal, locational, and 88 temporal variations in water quality, as well as the complexity of aquatic chemistry – in 89 terms of both disinfection kinetics and interactions in natural water matrices arising from 90 heterogeneous NOM (Kulkarni and Chellam, 2010). Accordingly, most modelers have 91 attempted to establish statistically-based empirical and/or semi-mechanistic linear or 92 nonlinear regression models using commonly measured water guality parameters and 93 operating conditions (Sadig and Rodriguez, 2004).

94 The 149 published chlorine-DBP models identified for use in this study are listed in 95 Table S4. The models all used variations of common DBP indicators such as dissolved 96 organic carbon (DOC), ultraviolet absorbance at a wavelength of 254 nm (UV₂₅₄), specific 97 UV absorbance (SUVA), chlorine dose (D), pH, temperature (T), bromide ion concentration 98 (Br⁻), and disinfection contact time (t) to predict mass-based concentrations of total 99 trihalomethanes (TTHMs), haloacetic acids (HAA5s), and subspecies thereof. Models 100 incorporating other parameters, such as chlorophyll (Golfinopoulos and Arhonditsis, 2002; 101 Golfinopoulos et al., 1998); fulvic acid (Rodrigues et al., 2007); specific rate constants 102 (Chang et al., 2006; Fabbricino and Korshin, 2009); nitrogen, ammonia, or combined 103 chlorine (Espigares et al., 2003); or site-specific parameters such as DBP yield coefficients 104 or THM formation potential (Amy et al., 1998; Gang et al., 2002; Li and Zhao, 2006) were 105 not included. This dictates that the predictive results reported here are purely empirically-106 derived due to the exclusion of semi-mechanistic models incorporating kinetic rate 107 constants, species-specific reactivity parameters, artificial neural network modeling, etc. 108 (Kulkarni and Chellam, 2010; Milot et al., 2002; Westerhoff et al., 2000). Exclusion of 109 these more advanced models was obligatory since site-specific rates were not readily 110 available for the DBP datasets and artificial neural networks are not generalizable models 111 that can be expressed in mathematical form, but rather are designed to adapt to specific 112 datasets through model training.

113 As part of the modeling effort, total organic carbon (TOC) and DOC were treated 114 synonymously as they typically differ by less than 5%, and can be used interchangeably with little error (Owen et al., 1993; Rathbun, 1996a). Table S4 lists the NOM parameter 115 116 used to develop the model (e.g., TOC or DOC), but when data was substituted into the 117 model's NOM placeholder, DOC (as reported in the data sets) was treated as equivalent to 118 TOC and vice versa. The conditions for model development are listed in Table S5. Empirical 119 models such as these are typically recommended for application within the boundary 120 conditions defined by the parameters from each specific study.

121 **5.** Time Evolution of DBPs

122 The 24-hr evolution of TTHMs and HAA5s is illustrated in Figure S1 as a function of 123 the level of photocatalytic treatment and type of source water. Figure S2 shows the time 124 evolution of the 4 subspecies of TTHMs and 5 subspecies of HAA5s for the SW, GAC, and 125 DS, as a function of the level of photocatalytic treatment.

#	Source	Model
	Tot	tal Trihalomethanes = TTHM = CF + BDCM + DBCM + BF
1	(Al-Omari et al., 2004)	TTHM = $4.527(t/60)^{0.127}(D)^{0.595}(TOC)^{0.596}(Br^{-}/1000)^{0.103}(pH)^{0.66}$
2	(Amy et al., 1998)	$TTHM = 10^{-1.385} (DOC)^{1.098} (D)^{0.152} (Br^{-})^{0.068} (T)^{0.609} (pH)^{1.601} (t)^{0.263}$
3	(Amy et al., 1998)	$TTHM = 10^{0.651} (DOC)^{0.752} (D)^{0.246} (Br^{-})^{0.185} (t)^{0.258}$
4	(Amy et al., 1998)	$TTHM = 10^{0.387} (DOC)^{0.839} (D)^{0.287} (Br^{-})^{0.259} (t)^{0.270}$
5	(Amy et al., 1998)	$TTHM = 10^{0.518} (DOC)^{0.801} (D)^{0.261} (Br^{-})^{0.223} (t)^{0.264}$
6	(Boyalla, 2004)	TTHM = $0.0001(D)^{3.14}(pH)^{1.56}(TOC)^{0.69}(t)^{0.175}$
7	(Chang et al., 1996)	$TTHM = 108.80(TOC)^{0.2466}(t)^{0.2956}(UV_{254})^{0.9919}(D)^{0.126}$
8	(Chang et al., 1996)	$TTHM = 12.7254(TOC)^{0.2910}(t)^{0.2706}(D)^{-0.0719}$
9	(Chang et al., 1996)	$TTHM = 131.7492(t)^{0.2931}(UV_{254})^{1.075}(D)^{0.1064}$
10	(Chen and Westerhoff, 2010)	$TTHM = 1147(DOC)^{0.00}(UV_{254})^{0.83}(Br^-+1)^{0.27}$
11	(Harrington et al., 1992)	$TTHM = [0.00309(DOC^*UV_{254})^{0.440}(D)^{0.409}(t)^{0.265}(T)^{1.06}(pH-2.6)^{0.715}(Br^-/1000+1)^{0.036}] * [106(Br^-/1000+1)^{0.036}] * [106(Br^-/1000+1)^$
		$/1000+1)^{0.48}(UV_{254})^{-0.089}]$
12	(Hong et al., 2007)	$TTHM = 10^{-1.375}(t)^{0.258}(D/DOC)^{0.194}(pH)^{1.695}(T)^{0.507}(Br^{-})^{0.218}$
13	(McBean et al., 2008)	TTHM = $10^{0.715} (DOC_{raw})^{0.322} (DOC_{treated})^{0.761} (D_{pre})^{0.206} (D_{post})^{0.184} (T)^{0.204}$
14	(Milot et al., 2002)	TTHM = $0.032(DOC)^{1.053}(t)^{0.263}(pH)^{1.168}(D)^{0.275}(T)^{1.040}$
15	(Rathbun, 1996a)	TTHM = $14.6(pH-3.8)^{1.01}(D)^{0.206}(UV_{254}/2)^{0.849}(t)^{0.306}$
16	(Rodriguez et al., 2000)	TTHM = $0.044(DOC)^{1.030}(t)^{0.262}(pH)^{1.149}(D)^{0.277}(T)^{0.968}$
17	(Rodriguez et al., 2000)	TTHM = $1.392(DOC)^{1.092}(pH)^{0.531}(T)^{0.255}$
18	(Semerjian et al., 2009)	TTHM = $-9.22+360.97(UV_{254})+0.54(T)+0.12(t)$
19	(Semerjian et al., 2009)	TTHM = $40.44(UV_{254})^{0.41}(Br^{-}/1000)^{0.44}$
20	(Semerjian et al., 2009)	TTHM = $sqrt[17.31+10.52(D)^2+259728.60(SUVA)^2]$
21	(Semerjian et al., 2009)	TTHM = $exp[-0.21+62.43(UV_{254})+0.06(T)]$
22	(Semerjian et al., 2009)	TTHM = $4.01 + 319.88(UV_{254}) + 2.42(D)$
23	(Semerjian et al., 2009)	$TTHM = 8.08(D)^{0.41}$
24	(Semerjian et al., 2009)	TTHM = $sqrt[42.10+29.23(D)^2+353375.0(UV_{254})^2]$
25	(Semerjian et al., 2009)	TTHM = $exp[1.53+42.21(UV_{254})]$
26	(Semerjian et al., 2009)	TTHM = $-12.66+0.48(t)+35.06(Br^{-}/1000)+10.26(D)$
27	(Semerjian et al., 2009)	TTHM = $450.3(t)^{0.21}(Br^{-}/1000)^{1.64}(D)^{0.34}(T)^{0.801}$
28	(Semerjian et al., 2009)	TTHM = $sqrt[-471.11+0.48(t)^2+1856.07(Br^{-}/1000)^2+404.38(D)^2]$
29	(Semerjian et al., 2009)	TTHM = $exp[1.43+0.02(t)-0.04(T)+2.78(Br^{-}/1000)+0.48(D)]$
30	(Sérodes et al., 2003)	TTHM = 16.9 + 16.0(TOC) + 3.319(D) - 1.135(T) + 1.139(t)
31	(Sérodes et al., 2003)	$TTHM = -0.101(TOC)^{3.914}(t)^{0.117}$
32	(Sérodes et al., 2003)	TTHM = $21.2+2.447(D)+0.449(t)$
33	(Sohn et al., 2004)	TTHM = $75.7(UV_{254})^{0.593}(D)^{0.332}(Br^{-})^{0.060}(t)^{0.264}$
34	(Sohn et al., 2004)	TTHM = $23.9(DOC*UV_{254})^{0.403}(D)^{0.225}(Br^{-})^{0.141}(t)^{0.264}$
35	(Sohn et al., 2004)	TTHM = $[10^{0.518}(\text{DOC})^{0.801}(\text{D})^{0.261}(\text{Br})^{0.223}(t)^{0.264}]*(1.156)^{(\text{pH}-7.5)}(1.0263)^{(\text{T}-20)}$

126 **Table S4 – Published DBP models considered as part of this study**

#	Source	Model
36	(Sohn et al., 2004)	TTHM = $0.00253(DOC)^{1.22}(D)^{0.442}(T)^{1.34}(pH)^{1.75}(t)^{0.34}$
37	(Sohn et al., 2004)	TTHM = $0.012(DOC^*UV_{254})^{0.47}(D)^{0.48}(T)^{1.10}(pH)^{2.38}(t)^{0.35}$
38	(Toroz and Uyak, 2005)	$TTHM = 11.967(TOC)^{0.398}(T)^{0.158}(D)^{0.702}$
39	(Urano et al., 1983)	TTHM = 8.2×10^{-4} (pH-2.8)(TOC)(D) ^{0.25} (t) ^{0.36}
40	(Uyak et al., 2005)	TTHM = $0.0707(TOC+3.2)^{1.314}(pH-4.0)^{1.496}(D-2.5)^{-0.197}(T+10)^{0.724}$
41	(USEPA, 2001) ^b	$TTHM = [23.9(DOC^*UV_{254})^{0.403}(D)^{0.225}(Br^{-})^{0.141}(t)^{0.264}]^*(1.1560)^{(pH-7.5)}(1.0263)^{(T-20)}$
42	(USEPA, 2001)	$TTHM = 17.7(DOC^*UV_{254})^{0.475}(D)^{0.173}(Br^{-})^{0.246}(1.316)^{(pH-8.0)}(1.036)^{(T-20)}(t)^{0.366}$
43	(Zhu, 1995)ª	TTHM = $-480.51+36.67(DOC)+4.92(D)+0.16(Br)+5.70(T)+33.45(pH)+0.92(t)$
44	(Zhu, 1995)	TTHM = $-826.40+36.76(DOC)+4.96(D)+0.15(Br^-)+261.1*log(T)+33.27(pH)+66.75(t)^{0.267}$
45	(Zhu, 1995)	$TTHM = 10^{-1.385} (DOC)^{1.098} (D)^{0.152} (Br^{-})^{0.068} (T)^{0.609} (pH)^{1.601} (t)^{0.263}$
46	(Zhu, 1995)	$TTHM = 10^{-0.377} (UV_{254})^{0.482} (D)^{0.339} (Br^{-})^{0.023} (T)^{0.617} (pH)^{1.609} (t)^{0.261}$
47	(Zhu, 1995)	TTHM = $10^{-0.549}$ (DOC*UV ₂₅₄) ^{0.421} (D) ^{0.145} (Br ⁻) ^{0.041} (T) ^{0.614} (pH) ^{1.606} (t) ^{0.261}
		Individual Trihalomethane Species
48	(Amy et al., 1998)	$CF = 10^{-1.205} (DOC)^{1.617} (D)^{-0.094} (Br^{-})^{-0.175} (T)^{0.607} (pH)^{1.403} (t)^{0.306}$
49	(Amy et al., 1998)	$BDCM = 10^{-2.874} (DOC)^{0.901} (D)^{0.017} (Br^{-})^{0.733} (T)^{0.498} (pH)^{1.511} (t)^{0.199}$
50	(Amy et al., 1998)	$DBCM = 10^{-5.649} (DOC)^{-0.226} (D)^{0.108} (Br^{-})^{1.81} (T)^{0.512} (pH)^{2.212} (t)^{0.146}$
51	(Amy et al., 1998)	$BF = 10^{-7.83} (DOC)^{-0.983} (D)^{0.804} (Br^{-})^{1.765} (T)^{0.754} (pH)^{2.139} (t)^{0.566}$
52	(Amy et al., 1998)	$CF = 10^{1.331} (DOC)^{1.11} (D)^{0.324} (Br^{-})^{-0.532} (t)^{0.341}$
53	(Amy et al., 1998)	$BDCM = 10^{-0.203} (DOC)^{0.504} (D)^{0.126} (Br^{-})^{0.474} (t)^{0.187}$
54	(Amy et al., 1998)	$DBCM = 10^{-5.398} (DOC)^{0.943} (D)^{-0.228} (Br^{-})^{2.678} (t)^{0.175}$
55	(Amy et al., 1998)	$BF = 10^{-9.6} (DOC)^{0.203} (D)^{-0.639} (Br^{-})^{4.487} (t)^{0.319}$
56	(Amy et al., 1998)	$CF = 10^{1.092} (DOC)^{1.179} (D)^{0.378} (Br^{-})^{-0.454} (t)^{0.326}$
57	(Amy et al., 1998)	$BDCM = 10^{-0.416} (DOC)^{0.599} (D)^{0.125} (Br^{-})^{0.533} (t)^{0.205}$
58	(Amy et al., 1998)	$DBCM = 10^{-5.127} (DOC)^{0.194} (D)^{0.433} (Br^{-})^{2.427} (t)^{0.294}$
59	(Amy et al., 1998)	$BF = 10^{-9.427} (DOC)^{-0.329} (D)^{-0.035} (Br^{-})^{4.335} (t)^{0.307}$
60	(Amy et al., 1998)	$CF = 10^{1.211} (DOC)^{1.149} (D)^{0.345} (Br^{-})^{-0.492} (t)^{0.333}$
61	(Amy et al., 1998)	$BDCM = 10^{-0.311} (DOC)^{0.556} (D)^{0.121} (Br^{-})^{0.505} (t)^{0.196}$
62	(Amy et al., 1998)	$DBCM = 10^{-5.248} (DOC)^{0.55} (D)^{0.105} (Br^{-})^{2.549} (t)^{0.234}$
63	(Amy et al., 1998)	$BF = 10^{-9.5} (DOC)^{-0.075} (D)^{-0.34} (Br^{-})^{4.409} (t)^{0.313}$
64	(Chen and Westerhoff, 2010)	$CF = 1805(DOC)^{0.11}(UV_{254})^{1.22}(Br^{-}+1)^{-2.19}$
65	(Chen and Westerhoff, 2010)	$BDCM = 137(DOC)^{0.16}(UV_{254})^{0.94}(Br^{-}+1)^{3.66}$
66	(Hong et al., 2007)	$BDCM = 10^{-3.201}(t)^{0.297}(pH)^{2.878}(t)^{0.414}(Br^{-})^{0.371}$
67	(Hong et al., 2007)	$CF = 10^{-0.748}(t)^{0.210}(D/DOC)^{0.221}(pH)^{1.374}(T)^{0.532}(Br^{-})^{-0.184}$
68	(Malcom Pirnie Inc, 1992)	$CF = 0.078(TOC^*UV_{254})^{0.616}(D)^{0.391}(t)^{0.265}(T)^{1.15}(pH-2.6)^{0.8}$
69	(Malcom Pirnie Inc, 1992)	$BDCM = 0.863(TOC^*UV_{254})^{0.177}(D)^{0.309}(t)^{0.271}(T)^{0.72}(pH-2.6)^{0.925}(Br^-+1)^{0.0722}$
70	(Malcom Pirnie Inc, 1992)	$DBCM = 2.57(UV_{254}/TOC)^{-0.184}(D)^{-0.0746}(t)^{0.57}(pH-2.6)^{1.35}(Br^{-}+1)^{2.08}$
71	(Malcom Pirnie Inc, 1993)	$CF = 0.997(TOC)^{0.580}(UV_{254})^{0.580}(D)^{0.814}(t)^{0.278}(Br^{-}+1)^{-4.27}(T)^{0.569}(pH-2.6)^{0.759}$
72	(Montgomery Watson, 1993)	$CF = 0.064(TOC)^{0.329}(UV_{254})^{0.874}(Br + 0.01)^{0.404}(pH)^{1.161}(D)^{0.561}(t)^{0.269}(T)^{1.018}$
73	(Montgomery Watson, 1993)	BDCM = $0.0098(Br^{-})^{0.181}(pH)^{2.55}(D)^{0.497}(t)^{0.256}(T)^{0.519}$

#	Source	Model
74	(Montgomery Watson, 1993)	BDCM = $1.325(TOC)^{-0.725}(Br^{-})^{0.794}(D)^{0.632}(t)^{0.204}(T)^{1.441}$
75	(Montgomery Watson, 1993)	$DBCM = 14.998(TOC)^{-1.665}(Br^{-})^{1.241}(D)0.729(t)^{0.261}(T)^{0.989}$
76	(Montgomery Watson, 1993)	$DBCM = 0.028(UV_{254})^{-1.175}(TOC)^{-1.078}(Br^{-})^{1.573}(pH)^{1.956}(D)^{1.072}(t)^{0.2}(T)^{0.596}$
77	(Montgomery Watson, 1993)	$BF = 6.533(TOC)^{-2.031}(Br^{-})^{1.388}(pH)^{1.603}(D)^{1.057}(t)^{0.136}$
78	(Rathbun, 1996b)	$CF = 0.442(pH)^2(D)^{0.229}(DOC)^{0.912}(Br^{-}/1000)^{-0.116}$
79	(Rathbun, 1996b)	$BDCM = 17.5(pH)^{1.01}(D)^{0.0367}(DOC)^{0.228}(Br^{-}/1000)^{0.513}$
80	(Rathbun, 1996b)	$DBCM = 26.6(pH)^{1.80}(D)^{-0.0928}(DOC)^{-0.758}(Br^{-}/1000)^{1.20}$
81	(Rathbun, 1996b)	$BF = 0.290(pH)^{3.51}(D)^{-0.347}(DOC)^{-0.330}(Br^-/1000)^{1.84}$
82	(USEPA, 2001) ^b	$CF = [266(DOC^*UV_{254})^{0.403}(D)^{0.424}(Br^{-})^{-0.679}(t)^{0.333}]^*(1.1322)^{(pH-7.5)}(1.0179)^{(T-20)}$
83	(USEPA, 2001) ^b	$BDCM = [1.68(DOC^*UV_{254})^{0.260}(D)^{0.114}(Br^{-})^{0.462}(t)^{0.196}]^*(1.0977)^{(pH-7.5)}(1.0260)^{(T-20)}$
84	(USEPA, 2001) ^b	$DBCM = [8*10^{-3}(DOC*UV_{254})^{-0.056}(D)^{-0.157}(Br^{-})^{1.425}(t)^{0.148}]*(1.1271)^{(pH-7.5)}(1.0212)^{(T-20)}$
85	(USEPA, 2001) ^b	$BF = [4.4*10^{-5} (DOC*UV_{254})^{-0.300} (D)^{-0.221} (Br^{-})^{2.134} (t)^{0.143}] * (1.3907)^{(pH-7.5)} (1.0374)^{(T-20)}$
86	(USEPA, 2001)	$CF = [101.0(DOC^*UV_{254})^{0.615}(D)^{0.699}(Br^{-})^{-0.468}(t)^{0.336}]^*(1.099)^{(pH-7.5)}(1.035)^{(T-20)}$
87	(USEPA, 2001)	$BDCM = [7.57(DOC^*UV_{254})^{0.443}(D)^{0.563}(Br^{-})^{0.0739}(t)^{0.281}]^*(1.355)^{(pH-7.5)}(1.030)^{(T-20)}$
88	(USEPA, 2001)	$DBCM = [3.99(DOC^*UV_{254})^{0.535}(D)^{0.125}(Br^{-})^{365}(t)^{0.322}]^*(1.436)^{(pH^{-}7.5)}(1.037)^{(T^{-}20)}$
89	(USEPA, 2001)	$BF = [1.47*10^{-1} (DOC*UV_{254})^{0.408} (D)^{-0.115} (Br^{-})^{0.961} (t)^{0.324}] * (1.438)^{(PH-7.5)} (1.048)^{(T-20)}$
90	(Zhu, 1995)	$CF = -366.42 + 31.03(DOC) + 5.27(D) - 0.23(Br^{-}) + 4.50(T) + 25.99(pH) + 0.72(t)$
91	(Zhu, 1995)	$BDCM = -71.05 + 6.66(DOC) - 0.65(D) + 0.19(Br^{-}) + 0.80(T) + 4.15(pH) + 0.12(t)$
92	(Zhu, 1995)	$DBCM = -33.88-0.41(DOC)+0.16(D)0.16(Br^{-})+0.33(T)+2.55(pH)+0.06(t)$
93	(Zhu, 1995)	$BF = -8.67 - 0.63(DOC) + 0.14(D) + 0.05(Br^{-}) + 0.07(T) + 0.75(pH) + 0.02(t)$
94	(Zhu, 1995)	CF = -762.77+31.20(DOC)+5.12(D)+143.53exp(-0.0029(Br ⁻))
		+210.09*log(T)+25.87(pH)+41.05(t) ^{0.300}
95	(Zhu, 1995)	$BDCM = -135.54 + 6.41(DOC) - 3.93(D)^{0.310} + 1.04^{*}(3.20 + 0.35(Br^{-}) - 0.00039(Br^{-})^{2})$
		+34.16*log(T)+4.12(pH)+16.06(t) ^{0.194}
96	(Zhu, 1995)	$DBCM = -54.52 - 0.39(DOC) + 0.11(D) + 1.01^{*}(-2.99 + 0.10(Br^{-}) - 0.00011(Br^{-})^{2})$
		+13.16*log(T)+2.53(pH)+11.73(t) ^{0.154}
97	(Zhu, 1995)	$BF = -7.51 + 0.62(DOC) + 0.069(D) + 0.99(1.34 - 0.23(Br) - 0.00015(Br)^{2})$
		$+2.76*\log(T)+0.75(pH)+0.062(t)^{0.764}$
98	(Zhu, 1995)	$CF = 10^{0.96} (UV_{254})^{0.604} (D)^{0.286} (Br^{-})^{-0.242} (T)^{0.619} (pH)^{1.416} (t)^{0.304}$
99	(Zhu, 1995)	$BDCM = 10^{-1.708} (UV_{254})^{0.590} (D)^{-0.022} (Br^{-})^{0.697} (T)^{0.505} (pH)^{1.519} (t)^{0.196}$
100	(Zhu, 1995)	$DBCM = 10^{-5.694} (UV_{254})^{-0.005} (D)^{-0.024} (Br^{1.820}(T)^{0.510}(pH)^{2.211}(t)^{0.146}$
101	(Zhu, 1995)	$BF = 10^{-9.26} (UV_{254})^{-0.722} (D)^{0.926} (Br^{-})^{1.804} (T)^{0.741} (pH)^{2.158} (t)^{0.571}$
102	(Zhu, 1995)	$CF = 10^{-0.040} (DOC^* UV_{254})^{0.578} (D)^{-0.038} (Br^{-})^{-0.217} (T)^{0.615} (pH)^{1.412} (t)^{0.304}$
103	(Zhu, 1995)	$BDCM = 10^{-2.06} (DOC^*UV_{254})^{0.422} (D)^{-0.111} (Br^{-})^{0.714} (T)^{0.502} (pH)^{1.515} (t)^{0.196}$
104	(Zhu, 1995)	$DBCM = 10^{-5.763} (DOC^*UV_{254})^{-0.049} (D)^{0.051} (Br^{-})^{1.817} (T)^{0.511} (pH)^{2.211} (t)^{0.146}$
105	(Zhu, 1995)	$BF = 10^{-8.7/6} (DOC^* UV_{254})^{-0.496} (D)^{0.996} (Br^{-})^{1.784} (T)^{0.744} (pH)^{2.146} (t)^{0.569}$
	Hal	oacetic Acids = HAA5 = MCA + DCA + TCA + MBA + DBA
106	(McBean et al., 2008)	HAAs = $10^{3.333}(\text{DOC}_{raw})^{0.380}(\text{DOC}_{treated})^{0.7/4}(\text{D}_{pre})^{0.102}(\text{pH}_{raw})^{-2.599}$
107	(Sérodes et al., 2003)	$HAAs = 2.720 + (TOC)^{0.553} + (D)^{0.458} + (t)^{0.295}$
108	(Sérodes et al., 2003)	$ HAAs = 1.33 + (TOC)^{2.612} + (D)^{0.102} + (T)^{0.255} + (t)^{0.102}$

#	Source	Model
109	(Sérodes et al., 2003)	HAAs = -8.202 + 4.869(TOC) + 1.053(D) + 0.364(t)
110	(USEPA, 2001)	HAA5 = $30.0(TOC)^{0.997}(D)^{0.278}(Br^{-})^{-0.138}(T)^{0.341}(pH)^{-0.799}(t)^{0.169}$
111	(USEPA, 2001)	$HAA5 = 30.7(DOC^*UV_{254})^{0.302}(D)^{0.541}(Br^{-})^{-0.012}(0.932)^{(pH-7.5)}(1.021)^{(T-20)}(t)^{0.161}$
112	(USEPA, 2001)	$HAA5 = 41.2(DOC^*UV_{254})^{0.498}(D)^{0.388}(Br^{-})^{-0.156}(0.867)^{(pH-7.5)}(1.021)^{(T-20)}(t)^{0.263}$
		Individual Haloacetic Acid Species
113	(Amy et al., 1998)	$MCA = 0.45(t)^{-0.009}(T)^{0.573}(pH)^{-0.279}(D)^{0.397}(DOC)^{0.173}(Br^{-})^{0.029}$
114	(Amy et al., 1998)	MBA = 6.21*10 ⁻⁵ (t) ^{0.090} (T) ^{0.707} (pH) ^{0.604} (D) ^{0.754} (DOC) ^{-0.584} (Br ⁻) ^{1.100}
115	(Amy et al., 1998)	$DCA = 0.30(t)^{0.218}(T)^{0.456}(pH)^{0.200}(D)^{0.379}(DOC)^{1.396}(Br^{-})^{-0.149}$
116	(Amy et al., 1998)	$TCA = 92.68(t)^{0.180}(T)^{0.299}(pH)^{-1.627}(D)^{0.331}(DOC)^{1.152}(Br)^{-0.229}$
117	(Amy et al., 1998)	$DBA = 3.59*10^{-5}(t)^{0.095}(T)^{0.380}(pH)^{-0.001}(D)^{0.673}(DOC)^{-1.086}(Br^{-})^{2.052}$
118	(Amy et al., 1998)	$MCA = 12.82(t)^{-0.066}(DOC)^{-0.377}(Br^{-})^{-0.303}(D)^{0.671}$
119	(Amy et al., 1998)	$MBA = 3.97 \times 10^{-3} (t)^{0.132} (DOC)^{0.409} (Br^{-})^{0.834} (D)^{0.095}$
120	(Amy et al., 1998)	$DCA = 10.96(t)^{0.230}(DOC)^{0.704}(Br^{-})^{-0.514}(D)^{0.751}$
121	(Amy et al., 1998)	$TCA = 6.22(t)^{0.164}(DOC)^{0.900}(Br^{-})^{-0.267}(D)^{0.697}$
122	(Amy et al., 1998)	$DBA = 4.84*10^{-5}(t)^{0.077}(DOC)^{-0.424}(Br^{-})^{2.222}(D)^{0.379}$
123	(Amy et al., 1998)	$MCA = 11.49(t)^{-0.021}(DOC)^{-0.668}(Br^{-})^{-0.415}(D)^{1.061}$
124	(Amy et al., 1998)	$MBA = 3.33*10^{-3}(t)^{-0.020}(DOC)^{0.211}(Br^{-})^{0.925}(D)^{0.313}$
125	(Amy et al., 1998)	$DCA = 6.31(t)^{0.213}(DOC)^{0.846}(Br^{-})^{-9.416}(D)^{0.742}$
126	(Amy et al., 1998)	$TCA = 3.97(t)^{0.163}(DOC)^{1.063}(Br^{-})^{-0.215}(D)^{0.680}$
127	(Amy et al., 1998)	$DBA = 3.92*10^{-5}(t)^{0.068}(DOC)^{-0.318}(Br^{-})^{2.256}(D)^{0.397}$
128	(Amy et al., 1998)	$MCA = 12.30(t)^{-0.043}(DOC)^{-0.522}(Br^{-})^{-0.355}(D)^{0.859}$
129	(Amy et al., 1998)	$MBA = 3.68 \times 10^{-3} (t)^{0.077} (DOC)^{0.299} (Br^{-})^{0.877} (D)^{0.206}$
130	(Amy et al., 1998)	$DCA = 8.38(t)^{0.222}(DOC)^{0.777}(Br^{-})^{-0.466}(D)^{0.744}$
131	(Amy et al., 1998)	$TCA = 4.98(t)^{0.163}(DOC)^{0.988}(Br^{-})^{-0.240}(D)^{0.683}$
132	(Amy et al., 1998)	$DBA = 4.41*10^{-5}(t)^{0.072}(DOC)^{-0.374}(Br^{-})^{2.237}(D)^{0.387}$
133	(Chen and Westerhoff, 2010)	$DCA = 188.5(DOC)^{0.57}(UV_{254})^{0.73}(Br^{-}+1)^{-2.42}$
134	(Chen and Westerhoff, 2010)	$TCA = 1282(DOC)^{0.19}(UV_{254})^{1.44}(Br^{-}+1)^{-2.85}$
135	(Montgomery Watson, 1993)	$MCA = 1.634(TOC)^{0.753}(Br^{-}+0.01)^{-0.085}(pH)^{-1.124}(D)^{0.509}(t)^{0.300}$
136	(Montgomery Watson, 1993)	$DCA = 0.606(TOC)^{0.291}(UV_{254})^{0.726}(Br^{-}+0.01)^{-0.568}(D)^{0.48}(t)^{0.239}(T)^{0.665}$
137	(Montgomery Watson, 1993)	$TCA = 87.182(TOC)^{0.355}(UV_{254})^{0.901}(Br^{-}+0.01)^{0.679}(pH)^{1.732}(D)^{0.881}(t)^{0.264}$
138	(Montgomery Watson, 1993)	$MBA = 0.176(TOC)^{1.664}(UV_{254})^{-0.624}(Br^{-})^{-0.795}(pH)^{-0.927}(t)^{0.145}(T)^{0.45}$
139	(Montgomery Watson, 1993)	$DBA = 84.945(TOC)^{-0.62}(UV_{254})^{0.651}(Br^{-})^{1.073}(D)^{-0.2}(t)^{0.12}(T)^{0.657}$
140	(USEPA, 2001)	$MCA = 4.58(DOC^*UV_{254})^{-0.090}(D)^{0.662}(Br^{-})^{-0.224}(1.042)^{(pH-7.5)}(1.024)^{(T-20)}(t)^{0.043}$
141	(USEPA, 2001)	$DCA = 60.4(DOC^*UV_{254})^{0.397}(D)^{0.665}(Br^{-})^{-0.558}(1.034)^{(pH^{-}7.5)}(1.017)^{(T^{-}20)}(t)^{0.222}$
142	(USEPA, 2001)	$TCA = 52.6(DOC^*UV_{254})^{0.403}(D)^{0.749}(Br^{-})^{-0.416}(0.8739)^{(pH-7.5)}(1.014)^{(T-20)}(t)^{0.163}$
143	(USEPA, 2001)	$MBA = 2.06*10^{-2} (DOC*UV_{254})^{0.358} (D)^{-0.101} (Br^{-})^{0.812} (0.6526)^{(pH-7.5)} (1.162)^{(T-20)} (t)^{0.043}$
144	(USEPA, 2001)	$DBA = 9.42*10^{-5} (DOC*UV_{254})^{0.0590} (D)^{0.182} (Br^{-})^{2.109} (1.210)^{(pH^{-7.5})} (1.007)^{(T^{-20})} (t)^{0.070}$
145	(USEPA, 2001)	$MCA = 1.31*10^{-1}(DOC*UV_{254})^{0.202}(D)^{0.275}(Br^{-})^{-0.958}(0.124)^{(pH-7.5)}(1.036)^{(T-20)}(t)^{0.923}$
146	(USEPA, 2001)	$DCA = 38.4(DOC^*UV_{254})^{0.503}(D)^{0.421}(Br^{-})^{-0.393}(0.867)^{(pH-7.5)}(1.019)^{(T-20)}(t)^{0.293}$

#	Source	Model				
147	(USEPA, 2001)	$TCA = 47.8(DOC^*UV_{254})^{0.627}(D)^{0.729}(Br^{-})^{-0.425}(0.602)^{(pH-7.5)}(1.011)^{(T-20)}(t)^{0.174}$				
148	(USEPA, 2001)	$MBA = 3.0*10^{-1} (DOC*UV_{254})^{0.093} (D)^{0.964} (Br^{-})^{-0.408} (0.134)^{(pH-7.5)} (1.054)^{(T-20)} (t)^{0.554}$				
149	(USEPA, 2001)	$DBA = 3.96*10^{-1} (DOC*UV_{254})^{0.509} (D)^{-0.251} (Br^{-})^{0.689} (1.302)^{(pH-7.5)} (1.019)^{(T-20)} (t)^{0.310}$				
TTHM=	=total trihalomethanes (\Box g L ⁻¹); Cl	F =chloroform (\Box g L ⁻¹); BDCM=bromodichloromethane (\Box g L ⁻¹); DBCM=dibromochloromethane				
(□g L ⁻	\Box g L ⁻¹); BF=bromoform (\Box g L ⁻¹); HAA=haloacetic acid (\Box g L ⁻¹); MCA=monochloroacetic acid (\Box g L ⁻¹); DCA=dichloroacetic acid (\Box g L ⁻¹);					
TCA=t	richloroacetic acid (\Box g L ⁻¹); BAA=I	promoacetic acid (\Box g L ⁻¹); MBA=monobromoacetic acid (\Box g L ⁻¹); TOC=total organic carbon (mg L ⁻				
¹); DO	C=dissolved organic carbon (mg L	⁻¹); UV ₂₅₄ =ultraviolet absorbance at a wavelength of 254 nm (cm ⁻¹); D=chlorine dose (mg L ⁻¹);				
T=tem	perature (°C); t=reaction time (hr); Br ⁻ =bromide ion concentration (\Box g L ⁻¹); SUVA=specific ultraviolet absorbance at a wavelength				
of 254	nm (L (mg-m) ⁻¹)					
^a Zhu	Zhu (1995) published additional models, including predictive power functions, which gave the best fit, and are reported in Amy et al.					
(1998	3).					
^b Repo	rted in USEPA (2001) based on So	hn et al. (2004) and Amy et al. (1998).				

Table S5 – Conditions for model development

Source	Model # ^k	Source Water	Organic Content (mg L ⁻¹)	Temperatu re (°C)	Time (hr)	Chlorine Dose (mg L ⁻¹)	рН	Bromide (µg L ⁻¹)
(Al-Omari et al., 2004)	1	Raw and settled waters from Amman, Jordan	0.7≤TOC≤3.1	20	0.5≤t≤24	1≤D≤5	7.65≤pH≤8.3 5	920≤Br⁻ ≤2,140
(Amy et al., 1998)	2, 48- 51	Raw surface and groundwaters from the U.S.	1.2≤DOC≤10.6	15≤T≤25	2≤t≤168	1.51≤D≤33.55	6.5≤pH≤8.5	7≤Br⁻ ≤600
(Amy et al., 1998)	3, 4, 5, 52–63	Alum or iron- coagulated waters from the U.S.	1.00≤DOC≤7.7 7	20	2≤t≤168	1.11≤D≤24.75	7.5	36≤Br⁻ ≤308
(Amy et al., 1998)	113- 117	Raw surface and groundwaters from the U.S.	1.2≤DOC≤10.7	15≤T≤25	2≤t≤168	2.11≤D≤26.4, 0.5≤D:DOC≤3 mg mg ⁻¹	6.5≤pH≤8.5	7≤Br⁻ ≤560
(Amy et al., 1998)	118- 122	Alum-coagulated waters from the U.S.	1≤DOC≤4.6	20	2≤t≤168	1.11≤D≤14.19, 1≤D:DOC≤3 mg mg ⁻¹	7.5	36≤Br⁻ ≤308
(Amy et al., 1998)	123- 127	Iron-coagulated waters from the U.S.	1.03≤DOC≤4.2 0	20	2≤t≤168	5≤D≤24, 1≤D:DOC≤3 mg mg ⁻¹	7.5	37≤Br⁻ ≤308

Source	Model # ^k	Source Water	Organic Content (mg L ⁻¹)	Temperatu re (°C)	Time (hr)	Chlorine Dose (mg L ⁻¹)	рН	Bromide (µg L ⁻¹)
(Amy et al., 1998)	128- 132	Alum or iron- coagulated waters from the U.S.	1.0≤DOC≤4.6	20	2≤t≤168	1.11≤D≤24, 1≤D:DOC≤3 mg mg ⁻¹	7.5	36≤Br⁻ ≤308
(Boyalla, 2004)	6	Raw water from Newfoundland, Canada	5.58≤TOC≤27 4.01≤DOC≤10. 88	23	1≤t≤120	22.36≤D ≤31.25	3.95≤pH≤9.0 8	
(Chang et al., 1996)	7, 8, 9	Raw water from Taiwan	0.61≤TOC≤1.2 4	21≤T≤27	0.5≤t≤16 8	3.5≤D≤5.0	7.8≤pH≤8.0	<10≤Br⁻ ≤20
(Chen and Westerhoff, 2010)	10, 64, 65, 133, 134	Wastewater, groundwater, surface water, and coagulated waters from the U.S.	0.6≤DOC≤23	25	24≤t≤72	D=3(DOC)+ 8(NH ₃ -N)+10	~8.2	0≤Br ⁻ ≤1,000
(Harrington et al., 1992)	11	Raw waters from the U.S.	3≤TOC≤13.8	10≤T≤30	0.1≤t≤16 8	1.5≤D≤69	4.6≤pH≤9.8	10≤Br⁻ ≤245
(Hong et al., 2007)	12, 66, 67	Raw water from Dongjiang River, Hong Kong		15≤T≤25	6≤t≤168	0.5≤D/DOC≤3 mg mg ⁻¹	6.5≤pH≤8.5	10≤Br⁻ ≤310
(Malcom Pirnie Inc, 1992)	68, 69, 70	Raw water	2.8≤TOC≤11 ⁹	13≤T≤20 ^g	0.1≤t≤ 105g	3.0≤D≤25.0 ^g	5.6≤pH≤9.0º	5≤Br⁻ ≤43 ^g
(Malcom Pirnie Inc, 1992)	71							
(McBean et al., 2008)	13, 106	Treated waters from Ontario, Canada	1.3≤DOC≤13.2 ^f 0.8≤DOC≤5.6	1.0≤T≤25.4		$0.42 \le D \le 5.5$ $0.15 \le D_{pre} \le 4.7$ $0.03 \le D_{post} \le 2.89$	7.14≤pH≤8.5 7 ^f 7.02≤pH≤8.3 4	
(Milot et al., 2002)	14	Raw and treated waters from the U.S.	2.2≤DOC or TOC≤13.8	10≤T≤30	0.1≤t≤16 8	1.5≤D≤69.0	4.6≤pH≤10	
(Montgomer y Watson, 1993)	72, 77, 135- 139		2.8≤TOC≤11 ^b	10≤T≤20 ^b	0.1≤t≤96 ♭	3.0≤D≤25.3 ^b	5.6≤pH≤9 ^b	

Source	Model # ^k	Source Water	Organic Content (mg L ⁻¹)	Temperatu re (°C)	Time (hr)	Chlorine Dose (mg L ⁻¹)	рН	Bromide (µg L ⁻¹)
(Montgomer y Watson, 1993)	73		2.8≤TOC≤11 ^b	10≤T≤20 ^b	0.1≤t≤96 ♭	3.0≤D≤25.3 ^b	5.6≤pH≤9 ^b	For D/Br⁻ <75 ^j
(Montgomer y Watson, 1993)	74		2.8≤TOC≤11 ^b	10≤T≤20 ^b	0.1≤t≤96 ^b	3.0≤D≤25.3 ^b	5.6≤pH≤9 ^b	For D/Br⁻ >75 ^j
(Montgomer y Watson, 1993)	75		2.8≤TOC≤11 ^b	10≤T≤20 ^b	0.1≤t≤96 ♭	3.0≤D≤25.3 ^b	5.6≤pH≤9 ^b	For D/Br⁻ <50 ^j
(Montgomer y Watson, 1993)	76		2.8≤TOC≤11 ^b	10≤T≤20 ^b	0.1≤t≤96 ♭	3.0≤D≤25.3 ^b	5.6≤pH≤9 ^b	For D/Br⁻ >50 ^j
(Rathbun, 1996b)	78-81	Untreated Mississippi, Missouri, and Ohio River water from the U.S.	2.2≤DOC≤12	25	168	15≤D≤50	6.85≤pH≤9.7 5°	10≤Br⁻ ≤85°
(Rathbun, 1996b)	15	Untreated Mississippi, Missouri, and Ohio River water from the U.S.	2.2≤DOC≤13 ^c	25	2≤t≤168	15≤D≤50	5.5≤pH≤10	
(Rodriguez et al., 2000)	16	Untreated waters	2.2≤DOC≤7.7	10≤T≤30	0≤t≤168	1.5≤D≤50	4.6≤pH≤10	10 ≤Br⁻ ≤420
(Rodriguez et al., 2000)	17	Distribution system in Quebec, Canada	1.2≤DOC≤7.7 ⁱ	6≤T≤20 ⁱ			6.4≤pH≤8.0 ⁱ	
(Semerjian et al., 2009)	18-29	Treated waters from Lebanon		15≤T≤28	0.33≤t≤4 8		6.5≤pH≤8.5	700 ≤Br⁻ ≤830
(Sérodes et al., 2003)	30, 107	Raw water from Charlesbourg, Canada	1.5≤TOC≤1.8	8≤T≤20	1≤t≤48	2≤D≤4	7.5	
(Sérodes et al., 2003)	31, 108	Treated water from Quebec City, Canada	1.6≤TOC≤2.2	8≤T≤20	1≤t <u>≤</u> 48	1≤D≤3	7.5	

Source	Model # ^k	Source Water	Organic Content (mg L ⁻¹)	Temperatu re (°C)	Time (hr)	Chlorine Dose (mg L ⁻¹)	рН	Bromide (µg L ⁻¹)
(Sérodes et al., 2003)	32, 109	Treated water from Sainte Foy, Canada	2.1≤TOC≤2.6	8≤T≤20	1≤t≤48	1≤D≤3	7.5	
(Sohn et al., 2004)	33, 34	Raw and coagulated waters from the U.S.	1≤DOC≤10.6 ^d	15≤T≤25ª	2≤t≤168 ^d	1.11≤D≤33.5 ^d	6.5≤pH≤8.5 ^d	7≤Br⁻ ≤600 ^d
(Sohn et al., 2004)	35	Temp. and pH correction for Amy et al., 1987	1≤DOC≤10.6 ^d	15≤T≤25 ^d	2≤t≤168 ^d	1.11≤D≤33.5 ^d	6.5≤pH≤8.5 ^d	7≤Br⁻ ≤600 ^d
(Sohn et al., 2004)	36, 37	Raw waters from the U.S. (short-term models)	3.0≤DOC≤11.7	10≤T≤20	0.1≤t≤2	1.5≤D≤1.7	4.6≤pH≤8.3	10≤Br⁻ ≤152
(Toroz and Uyak, 2005)	38	Distribution system in Istanbul, Turkey	2.98≤TOC≤3.4 1	11.07≤T≤2 6.74		3.2≤D≤6.2	7.34≤pH≤8.5 0	145≤Br⁻ ≤255
(Urano et al., 1983)	39	Humic acid	2.1≤TOC≤9 ^c	20	1≤t≤48°	50	4≤pH≤10 ^c	
(Uyak et al., 2005)	40	Raw water from Istanbul, Turkey	4.20≤TOC≤6.2 0	7.2≤T≤22.7		2.82≤D≤6.75	7.10≤pH≤7.9 0	
(USEPA, 2001)	110	Raw surface and groundwaters from the U.S.	1.2≤DOC≤10.6	15≤T≤25	2≤t≤168	1.51≤D≤33.55	6.5≤pH≤8.5	7≤Br⁻ ≤600
(USEPA, 2001)	41, 82- 85, 111, 140- 144	Coagulated waters from the U.S.	1.00≤DOC≤7.7 7	20	2≤t≤168	1.11≤D≤24.75	7.5	23≤Br ⁻ ≤308
(USEPA, 2001)	42, 86- 89, 112, 145- 149	GAC-filtered waters from the U.S.	0.14≤D0C≤2.0	3≤T≤33	2≤t≤168	0.5≤D≤3.0	6.7≤pH≤10	10≤Br ⁻ ≤570
(Zhu, 1995)	43-47, 90-105	Raw waters from the U.S.	1.2≤DOC≤10.6	15≤T≤25	2≤t≤168	1.51≤D≤33.55	6.5≤pH≤8.5	7≤Br⁻ ≤600
TOC=total organic carbon (mg L ⁻¹); DOC=dissolved organic carbon (mg L ⁻¹); UV ₂₅₄ =ultraviolet absorbance at a wavelength of 254 nm (cm ⁻¹); D=chlorine dose (mg L ⁻¹); T=temperature (°C); t=reaction time (hr); Br ⁻ =bromide ion concentration (\Box g L ⁻¹); SUVA=specific ultraviolet absorbance at a wavelength of 254 nm (L (mg-m) ⁻¹)								
Source	Model # ^k	Source Water	Organic Content (mg L ⁻¹)	Temperatu re (°C)	Time (hr)	Chlorine Dose (mg L ⁻¹)	рН	Bromide (µg L ⁻¹)
--	--	--	--	---	--	---	------------------	----------------------------------
 ^a Boundary cd ^b As reported ^c Estimated fr ^d Boundary cd ^e Reported in ^f The model d water DOC ^g As reported ^h Bench-scale ⁱ Reported bo ^j As reported ^k Model # ref 	onditions for in Milot et com data pro- onditions for USEPA (20 differentiate and pH in Gang et data colle undaries an in Sadiq ar ers to num	r model provided in al. (2002) resented in graphical or model provided in 001) based on Amy e es between raw and cal. (2002) cted from Amy et al. re 10 th and 90 th perce nd Rodriguez (2004) ber assigned in Table	Amy et al. (1987) form Amy et al. (1998) t al. (1998) treated water. The (1987), Montgome entile values from t	e first entry is f ery Watson (19 the entire field-	or raw water 91), and Rat scale databa	r DOC and pH, while t thbun (1996a) ase	he second is for	treated



134 Figure S1 – Time evolution of TTHMs and HAA5s following chlorine disinfection. Panels (a) and (d) illustrate the formation of 135 TTHMs and HAA5s, respectively, for the SW as a function of time and extent of photocatalytic treatment; (b) and (e) illustrate 136 the same for the GAC-filtered water; (c) and (f) illustrate the same for the DS water.



139 Figure S2 – Time evolution of TTHM and HAA5 species following chlorine disinfection. Panels (a) and (d) illustrate the

140 formation of THMs and HAAs, respectively, for the SW as a function of time and extent of photocatalytic treatment; (b) and (e) 141 illustrate the same for the GAC-filtered water; (c) and (f) illustrate the same for the DS water.

142 6. Correlation Analyses of Empirical Data Parameters and DBP Formation

143 6.1 Normal Distribution Testing

144 Prior to testing correlations, each data grouping (e.g., extent of photocatalytic

145 treatment, type of source water) was tested for normal distribution using the D'Agostino-

146 Pearson omnibus test (a = 0.05). The results of these tests are reported in Tables S6 – S8, with normally distributed datasets shaded in gray. 147

148 Table S6 - P values from the D'Agostino-Pearson omnibus test for normality of internal

149 DBP data grouped by extent of photocatalytic treatment

Data Group	All AOPs	Non-AOPs	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻³
D	0.0004	0.4086	0.0369	0.0256	0.0329
t	0.0029	0.0115	0.0416	0.0416	0.0416
рН	<0.0001	0.0002	0.0551	0.0551	0.0551
DOC	0.0271	< 0.0001	0.0678	0.0344	0.0292
Br⁻	<0.0001	0.0027	0.0316	0.0291	0.0291
UV ₂₅₄	0.0216	0.0002	0.0762	0.0291	0.0568
SUVA	0.0606	0.4687	0.0396	0.0311	0.0338
BDCM	0.0115	0.0579	0.0266	< 0.0001	< 0.0001
CF	<0.0001	0.0031	0.0386	0.0002	< 0.0001
BF	<0.0001	< 0.0001	0.0002	0.3090	0.0218
DBCM	0.0335	0.1987	0.8709	< 0.0001	< 0.0001
TTHM	0.0033	0.7868	0.0007	< 0.0001	< 0.0001
MCA	0.0527	0.0010	0.1757	0.1270	0.0461
DCA	0.1050	0.2076	0.0418	0.0066	0.0034
TCA	0.1705	0.1330	0.2957	0.3600	0.1770
MBA	<0.0001	0.0051	0.0309	а	а
DBA	0.0003	0.0780	0.1271	0.0058	< 0.0001
HAA5	0.2060	0.0764	0.9224	0.0002	0.3540
n =	45	25 - 30	15	15	15
^a Values did not	change, so no P va	alue could be com	puted		

150

151 Table S7 – P values from the D'Agostino-Pearson omnibus test for normality of all THM

152 data (internal and external) grouped by type of source water

		1	0			1	
Data Group	Untreated	Coagulated	Settled	Filtered	Finished	Synthetic	All Data
D	< 0.0001	<0.0001	< 0.0001	0.0082	0.2394	< 0.0001	< 0.0001
t	< 0.0001	а	0.0177	< 0.0001	0.0177	< 0.0001	< 0.0001
Т	< 0.0001	< 0.0001	а	< 0.0001	а	<0.0001	< 0.0001
pН	< 0.0001	0.0021	а	< 0.0001	а	< 0.0001	< 0.0001
DOC	< 0.0001	0.0311	< 0.0001	0.0184	0.0002	0.0721	< 0.0001
Br⁻	< 0.0001	< 0.0001	0.0065	< 0.0001	0.0066	<0.0001	< 0.0001
UV ₂₅₄	< 0.0001	0.0153	0.0018	0.0345	0.1553	0.0794	< 0.0001
SUVA	< 0.0001	0.1111	0.0041	0.0618	0.0886	< 0.0001	< 0.0001
BDCM	0.0059	0.0187	0.1268	0.1913	0.1913	b	< 0.0001
CF	< 0.0001	0.0175	0.5105	0.0168	0.0168	<0.0001	< 0.0001
BF	< 0.0001	0.0085	0.0002	0.0250	0.0250	b	< 0.0001
DBCM	< 0.0001	0.2021	0.2468	0.0003	0.0003	b	0.1998
TTHM	< 0.0001	0.1932	0.0165	0.4758	0.2382	0.1256	< 0.0001
n =	127 - 211	24	25	25 - 39	25	15 – 79	198 - 369
^a Values di	d not change,	so no P value co	ould be com	puted			
^b No values	s reported in d	atasets analyze	d	-			

153

Table S8 – P values from the D'Agostino-Pearson omnibus test for normality of all HAA data
 154

155 (internal and external) grouped by type of source water

Data Group	Untreated	Settled	Filtered	Finished	All Data
D	<0.0001	<0.0001	0.2394	0.1088	<0.0001
t	0.0123	0.0179	0.0179	0.0179	<0.0001
Т	0.0089	b	b	b	0.0389
pН	0.0217	b	b	b	<0.0001
DOC	<0.0001	<0.0001	0.0002	0.1076	<0.0001
Br⁻	0.0001	0.0065	0.0066	b	<0.0001
UV ₂₅₄	<0.0001	0.0041	0.1531	<0.0001	<0.0001
SUVA	0.1123	0.0513	0.0771	0.0906	<0.0001
MCA	а	<0.0001	0.0030	0.9891	0.0071
DCA	а	0.0217	0.1618	0.2585	<0.0001
TCA	а	0.2757	0.3805	0.0293	<0.0001
MBA	а	<0.0001	< 0.0001	0.0090	<0.0001
DBA	а	<0.0001	0.0067	0.0037	0.0013
HAA5	0.0609	0.9440	0.3489	0.0653	<0.0001
n =	4 - 31	25	25	25	75 - 11
^a n too small to	compute P value				

^b Values did not change, so no P value could be computed

156 157

6.2 Analyses of Linear Correlation Grouped by Type of Source Water

158 Figure S3 shows the correlation analyses between parameters commonly associated 159 with DBP formation and the DBPs themselves, grouped by type of source water. Linear 160 correlation coefficients (either the Pearson product-moment coefficient of correlation for 161 normally distributed data or the Spearman rank correlation coefficient for non-normal 162 distributions) were computed for datasets. All data, both internal and external, were 163 considered in the statistical analysis. Figure S3 includes varying numbers of data pairs per 164 group, dependent upon reports from individual literature sources, datasets in which the 165 parameters were varied, etc. (for example, if contact time was held constant for a particular 166 dataset, no correlation can be computed).

167 Although the data matrices are sparse for some groups, it appears that D, T, t, DOC, 168 UV_{254} , and SUVA are generally positively correlated with DBP formation. The relationship of 169 THMs and HAAs with pH and Br⁻ is less straightforward as both positive and negative 170 correlations are observed. The treated samples show less variation within parameter 171 groups, while the untreated samples show the greatest variation. This may stem from untreated samples containing more complex NOM, whereas following different stages of 172 drinking water treatment (e.g., coagulation, sedimentation, and filtration), less NOM is 173 174 present. Moreover, the NOM composition is likely more homogenized as physicochemical 175 treatments such as coagulation tend to target certain NOM moieties, i.e., the aromatic 176 fraction, which is most closely linked to DBP formation (Weishaar et al., 2003).



177

179 Figure S3 – Linear correlation coefficients (Pearson for normally distributed datasets,

180 otherwise Spearman) between DBP formation and precursor measurements for different

181 types of source water. Values closer to +1 indicate positive correlations, while values closer

182 to -1 indicate negative correlations. Data includes all applicable internal and external 183 datasets.

184 7. DBP Model Performance

185 Coefficients of determination (R^2) were computed for each of the 2,642 datasets of 186 THM and HAA model/data pairings analyzed in this study. All R² values, representing the 187 model's ability to accurately predict DBP formation using bulk precursors such as DOC, 188 UV_{254} , SUVA, D, t, T, and Br⁻, are provided in Tables S9 and S10 for THMs and HAAs, 189 respectively. The values ranged from 0.00 - 0.99, with an overall average of 0.39. The R² 190 values shown in bold, underlined font are for the datasets which satisfied all boundary 191 conditions under which the model was developed. A total of 12 (0.5%) of the THM and HAA 192 model/data datasets satisfied all model boundary conditions. Notably, the majority of the 193 datasets satisfying the boundary conditions were for models that did not report all 194 conditions used to develop specific parameters, particularly Br⁻. Moreover, many of the DBP 195 models included in this study were bounded by time periods ≥ 2 hr, which may adversely 196 impact model performance for the rapid reactions characteristic of short-term DBP 197 formation for internal DBP measurements at 10 – 15 min and 1 hr.

While it is recommended that models be applied to datasets satisfying the boundary conditions, based on this analysis, it appears that many models are so specific in their application, due to the boundary conditions implemented during model development, that extrapolation is necessary for the vast majority of water systems. The inaccuracy this introduces brings into question the usefulness of many of these models outside of the specific scenario from which they were derived, even in cases of qualitative comparisons.

_	Source	Specie	Source	Model ⁱ				Cu	rrent Stu	udy					(Gerrity	et al., 2009))
Mode # ^j		S	# of Sampl es	Validat ion	All Dat a	Non - AOP	TiO ₂	SW	GAC	DS	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻³	All Dat a	Raw Wate r	Coagulate d Water	TiO ₂
1	(Al-Omari et al., 2004)ª	ТТНМ		0.94 (NR ^k)	0.2 9	0.0 0	0.5 9	0.6 0	0.57	0.1 1	0.7 0	0.0 8	0.0 0	0.2 8	0.78	0.78	0.5 5
2	(Amy et al., 1998)	TTHM	786 (80)	0.90 (0.84 ^d)	0.3 3	0.0 1	0.8 4	0.6 0	0.61	0.0 4	0.9 1	0.4 9	0.0 4	0.3 4	0.72	0.65	0.7 9
48	(Amy et al., 1998)	CF	786 (80)	0.87	0.0 2	0.0 4	0.2 5	0.6 4	0.70	0.0 3	0.0 4	0.8 1	0.0 7	0.2 7	0.62	0.64	0.8 4
49	(Amy et al., 1998)	BDCM	786 (80)	0.90	0.1 7	0.0 7	0.4 7	0.3 3	0.17	0.4 4	0.6 5	0.0 2	0.0 4	0.1 4	0.57	0.50	0.4 5
50	(Amy et al., 1998)	DBCM	786 (80)	0.89	0.0 1	0.0 2	0.0 6	0.0 8	0.22	0.0 4	0.2 5	0.1 0	0.0 6	0.0 1	0.17	0.15	0.0 0
51	(Amy et al., 1998)	BF	786 (80)	0.61	0.0 0	0.0 0	0.0 0	0.0 7	0.03	0.1 0	0.9 5	0.5 8	0.7 4	0.0 7	0.05	0.02	0.0 8
3	(Amy et al., 1998)	TTHM	143 ^e	0.87	0.3 1	0.0 2	0.6 7	0.5 7	0.47	0.0 8	0.8 8	0.1 7	0.0 1	0.1 1	0.73	0.67	0.6 1
52	(Amy et al., 1998)	CF	143 ^e	0.90	0.1 9	0.4 5	0.2 1	0.2 2	0.41	0.4 8	0.0 0	0.1 6	0.1 4	0.2 6	0.85	0.71	0.6 1
53	(Amy et al., 1998)	BDCM	143 ^e	0.84	0.2 2	0.0 9	0.3 9	0.3 1	0.24	0.4 8	0.7 3	0.0 4	0.0 6	0.0 6	0.51	0.47	0.4 3
54	(Amy et al., 1998)	DBCM	143 ^e	0.31	0.0 4	0.0 2	0.0 1	0.4 2	0.22	0.2 4	0.2 2	0.0 9	0.0 6	0.0 1	0.49	0.32	0.0 0
55	(Amy et al., 1998)	BF	143 ^e	0.73	0.0 0	0.0 1	0.3 4	0.0 7	0.02	0.0 7	0.8 9	0.5 0	0.7 4	0.0 0	0.13	0.00	0.0 0
4	(Amy et al., 1998)	TTHM	143 ^e	0.88	0.2 8	0.0 2	0.6 3	0.5 4	0.39	0.0 7	0.3 0	0.9 6	0.0 0	0.1 0	0.73	0.66	0.5 9
56	(Amy et al., 1998)	CF	143 ^e	0.92	0.2 3	0.4 9	0.2 9	0.2 5	0.48	0.4 9	0.0 1	0.2 1	0.3 3	0.2 4	0.85	0.74	0.6 4
57	(Amy et al., 1998)	BDCM	143 ^e	0.84	0.2 1	0.0 9	0.4 1	0.3 3	0.22	0.4 6	0.7 3	0.0 4	0.0 5	0.0 5	0.49	0.48	0.4 2
58	(Amy et al., 1998)	DBCM	143 ^e	0.34	0.0 3	0.0 3	0.0 1	0.0 5	0.21	0.2 3	0.1 3	0.0 9	0.0 4	0.0 0	0.59	0.31	0.0 0
59	(Amy et al., 1998)	BF	143 ^e	0.72	0.0 0	0.0 1	0.0 4	0.0 1	0.02	0.0 0	0.8 9	0.5 0	0.7 4	0.0 1	0.02	0.00	0.0 2
5	(Amy et al., 1998)	TTHM	286 ^e	0.87 (0.98 ^d)	0.2 9	0.0 2	0.6 4	0.5 5	0.42	0.0 7	0.8 6	0.1 2	0.0 1	0.1 0	0.73	0.66	0.6 0

205 **Table S9** – Average correlation coefficients (R² values) for pairs of predictive THM models versus measured THM datasets

_	Source	Specie	Source	Model ⁱ				Cu	rent Stu	Jdy					(Gerrity	et al., 2009))
Mode # ^j		S	# of Sampl es	Validat ion	All Dat a	Non - AOP	TiO ₂	SW	GAC	DS	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻³	All Dat a	Raw Wate r	Coagulate d Water	TiO ₂
60	(Amy et al.,	CF	286 ^e	0.91	0.2	0.4	0.2	0.2	0.45	0.4	0.0	0.1	0.2	0.2	0.85	0.73	0.6
61	(Amy et al.,	BDCM	286 ^e	0.83	0.2	0.0	0.4	0.3	0.23	0.4	0.7	0.0	0.0	0.0	0.49	0.47	0.4
62	(Amy et al.,	DBCM	286 ^e	0.32	0.0	9 0.0	0.0	0.2	0.22	0.2	0.1	4 0.0	0.0	0.0	0.66	0.32	0.0
63	(Amy et al.,	BF	286 ^e	0.72	4 0.0	0.0	0.1	0.0	0.02	0.0	0.8	8 0.5	0.7	0.0	0.04	0.00	0.0
6	(Boyalla,	TTHM	51	0.77	0.0	0.0	0.1	0.1	0.52	0.2	9 0.5	0.0	4	0.2	0.50	0.78	0.2
7	(Chang et	TTHM	120	0.97	0.2	0.0	0.8	0.3	0.36	0.2	0.8	0.4	0.1	4 0.0 7	0.76	0.55	0.2
8	(Chang et	TTHM	120	0.94	0.2 9	9 0.0 8	0.4 4	0.4	0.41	0.0	0.6 8	0.5	0.3 1	0.0	0.14	0.52	9 0.5 3
9	(Chang et al 1996)	TTHM	120	0.95	0.2	0.1	0.8	, 0.3 5	0.36	0.3	0.8	0.3 9	0.2	0.0	0.76	0.53	0.2
10	(Chen and Westerhoff, 2010)	TTHM	210	0.87	0.1 4	0.0 0	0.6 8	0.1 9	0.22	0.3 8	0.2	0.0 2	0.0 1	0.0 8	0.75	0.57	0.3 2
64	(Chen and Westerhoff, 2010)	CF	210	0.88	0.0 8	0.3 7	0.0 2	0.0 4	0.09	0.8 2	0.3 3	0.0 1	0.0 9	0.1 2	0.85	0.52	0.1 1
65	(Chen and Westerhoff, 2010)	BDCM	210	0.69	0.0 1	0.0 1	0.3 2	0.0 8	0.08	0.7 1	0.2 6	0.0 2	0.0 0	0.0 1	0.56	0.54	0.0 1
11	(Harrington et al., 1992)	TTHM		NR (NR ^{b,c,k})	0.3 8	0.0 9	0.8 1	0.5 1	0.60	0.1 9	0.8 6	0.3 7	0.1 2	0.4 1	0.72	0.68	0.6 7
12	(Hong et al., 2007)	TTHM	72	0.90	0.0 1	0.1 1	0.0 0	0.0 0	0.07	0.0 2	0.8 5	0.0 6	0.1 4	0.0 2	0.15	0.18	0.0 2
66	(Hong et al., 2007)	BDCM	72	0.90	0.1 0	0.2 0	0.0 9	0.0 4	0.20	0.0 8	0.9 6	0.1 5	0.2 5	0.0 6	0.04	0.11	0.0 0
67	(Hong et al., 2007)	CF	72	0.85	0.0	0.2	0.0 6	0.0	0.05	0.0	0.0 0	0.1	0.0 3	0.0	0.06	0.10	0.0
68	(Malcom Pirnie Inc, 1992)	CF			0.0 9	0.4 9	0.0 2	0.0 7	0.16	0.6 9	0.1 8	0.0 3	0.0 3	0.3 0	0.69	0.56	0.3 5

_	Source	Specie	Source	Model ⁱ				Cu	rrent Sti	udy					(Gerrity	et al., 2009))
Mode # ^j		S	# of Sampl es	Validat ion	All Dat a	Non - AOP	TiO ₂	SW	GAC	DS	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻³	All Dat a	Raw Wate r	Coagulate d Water	TiO ₂
69	(Malcom Pirnie Inc, 1992)	BDCM			0.1 6	0.1 4	0.2 9	0.2 3	0.20	0.4 3	0.8 1	0.0 6	0.1 0	0.3 0	0.70	0.57	0.4 5
70	(Malcom Pirnie Inc, 1992)	DBCM			0.0 2	0.0 2	0.0 3	0.0 0	0.22	0.7 0	0.1 5	0.0 8	0.0 5	0.0 0	0.29	0.21	0.0 1
71	(Malcom Pirnie Inc, 1993)	CF			0.0 9	0.5 4	0.0 1	0.0 7	0.07	0.7 0	0.1 7	0.0 3	0.0 4	0.0 9	0.71	0.11	0.1 7
13	(McBean et al., 2008)	TTHM	180	0.73	0.2 4	0.1 2	0.7 2	0.4 0	0.67	0.0 3	0.2 0	0.1 5	0.0 2	0.2 6	0.95	0.82	0.8 0
14	(Milot et al., 2002)	TTHM	1483	0.87	0.3 3	0.0 1	0.8 4	0.5 9	0.68	0.0 5	0.8 8	0.5 5	0.0 5	0.3 5	<u>0.63</u>	<u>0.54</u>	0.6 7
72	(Montgomer y Watson, 1993)	CF			0.2 1	0.1 0	0.8 9	0.2 1	0.06	0.7 3	0.8 6	0.6 6	0.7 4	0.2 9	0.79	0.84	0.6 5
73	(Montgomer y Watson, 1993)	BDCM			0.1 1	0.4 4	0.0 7	0.0 3	0.22	0.2 0	0.9 2	0.2 1	0.2 4	0.1 1	0.50	0.32	0.0 1
74	(Montgomer y Watson, 1993)	BDCM			0.0 3	0.0 8	0.0 4	0.2 6	0.06	0.1 8	0.6 9	0.0 0	0.0 5	0.0 7	0.37	0.09	0.2 4
75	(Montgomer y Watson, 1993)	DBCM			0.1 0	0.1 7	0.0 7	0.0 5	0.36	0.2 9	0.8 0	0.0 0	0.4 8	0.1 0	0.25	0.47	0.2 1
76	(Montgomer y Watson, 1993)	DBCM			0.1 4	0.0 2	0.1 3	0.5 3	0.50	0.4 5	0.2 0	0.0 9	0.0 9	0.1 0	0.72	0.56	0.1 9
77	(Montgomer y Watson, 1993)	BF			0.1 0	0.0 1	0.1 0	0.0 6	0.28	0.7 6	0.3 0	0.1 5	0.2 3	0.0 0	0.01	0.45	0.0 0
78	(Rathbun, 1996b)	CF		0.97	0.3 3	0.3 2	0.4 9	0.2 5	0.55	0.8 3	0.0 0	0.1 2	0.3 8	0.3 3	0.61	0.45	0.6 9
79	(Rathbun, 1996b)	BDCM		0.86	0.0 7	0.0 0	0.1 4	0.0 2	0.16	0.7 1	0.2 6	0.0 1	0.0 0	0.0 7	0.02	0.23	0.1 8
80	(Rathbun, 1996b)	DBCM		0.94	0.0 2	0.0	0.1 8	0.5 0	0.16	0.6 8	0.4 8	0.1 7	0.0 9	0.0 2	0.68	0.38	0.0 9

_	Source	Specie	Source	Model ⁱ				Cu	rrent Sti	udy					(Gerrity	et al., 2009)
Mode # ^j		S	# of Sampl es	Validat ion	All Dat a	Non - AOP	TiO ₂	SW	GAC	DS	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻³	All Dat a	Raw Wate r	Coagulate d Water	TiO ₂
81	(Rathbun, 1996b)	BF		0.78	0.0	0.0 6	0.0 2	0.1 6	0.00	0.8 5	0.5 2	0.1 5	0.3 9	0.0	0.71	0.18	0.0 7
15	(Rathbun, 1996a)	TTHM	685	0.99	0.3 4	0.1 4	0.8 4	0.4 1	0.43	0.3 2	0.8 5	0.4 2	0.2 9	0.2 2	0.75	0.74	0.3 1
16	(Rodriguez et al., 2000)	TTHM	1483	0.90	0.3 3	0.0 1	0.8 4	0.5 9	0.68	0.0 5	0.8 8	0.5 5	0.0 5	0.3 5	0.64	0.56	0.6 9
17	(Rodriguez et al., 2000)	ТТНМ	113	0.34	0.2 2	0.2 8	0.7 9	0.3 9	0.66	0.0 1	0.2 1	0.2 5	0.0 4	0.1 9	<u>0.82</u>	0.79	0.8 1
18	(Semerjian et al., 2009)	ТТНМ	53	0.38	0.2 6	0.0 2	0.8 3	0.2 6	0.37	0.4 5	0.6 2	0.4 3	0.4 3	0.1 4	0.96	0.61	0.4 5
19	(Semerjian et al., 2009)	ТТНМ	53	0.25	0.0 8	0.0 0	0.3 6	0.1 4	0.19	0.3 7	0.1 7	0.0 1	0.0 0	0.1 2	0.73	0.61	0.4 2
20	(Semerjian et al., 2009)	ТТНМ	53	0.39	0.2 6	0.1 3	0.4 5	0.1 4	0.41	0.8 1	0.1 6	0.0 9	0.2 1	0.1 3	0.79	0.46	0.1 6
21	(Semerjian et al., 2009)	ТТНМ	53	0.29	0.0 6	0.0 0	0.7 5	0.0 5	0.17	0.4 2	0.2 1	0.1 3	0.0 5	0.0 0	0.86	0.10	0.1 9
22	(Semerjian et al., 2009)	ТТНМ	60	0.23	0.1 7	0.0 1	0.3 5	0.1 3	0.33	0.4 3	0.1 9	0.0 0	0.0 1	0.1 0	0.79	0.56	0.3 0
23	(Semerjian et al., 2009)	ТТНМ	60	0.12	0.0 2	0.0 3	0.0 1	0.2 6	0.35	0.4 1	0.1 7	0.0 0	0.0 1	0.1 3	0.56	0.76	0.0 3
24	(Semerjian et al., 2009)	ТТНМ	60	0.33	0.0 7	0.0 1	0.0 8	0.0 1	0.20	0.4 6	0.1 9	0.0 1	0.0 2	0.0 7	0.78	0.54	0.1 9
25	(Semerjian et al., 2009)	TTHM	60	0.11	0.1 0	0.0 0	0.7 7	0.0 8	0.21	0.4 1	0.2 1	0.1 3	0.0 5	0.0 0	0.74	0.12	0.1 4
26	(Semerjian et al., 2009)	TTHM	160	0.54	0.0 5	0.1	0.0 3	0.0 1	0.18	0.3 5	0.5 9	0.0 3	0.0 1	0.1 0	0.54	0.76	0.0 2

_	Source	Specie	Source	Model ⁱ				Cu	rent Stu	ypr					(Gerrity	et al., 2009))
Mode # ^j		S	# of Sampl es	Validat ion	All Dat a	Non - AOP	TiO ₂	SW	GAC	DS	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻³	All Dat a	Raw Wate r	Coagulate d Water	TiO ₂
27	(Semerjian	TTHM	160	0.70	0.0	0.0	0.0	0.0	0.11	0.1	0.4	0.0	0.0	0.0	0.47	0.17	0.0
	et al., 2009)				1	0	1	0		1	8	3	1	1			0
28	(Semerjian et al.,	TTHM	160	0.31	0.0 2	0.0 4	0.0 1	0.2 1	0.34	0.4 8	0.2 4	0.0 0	0.0 1	0.1 2	0.53	0.76	0.0 2
	2009)																
29	(Semerjian et al., 2009)	TTHM	160	0.61	0.0 1	0.0 0	0.0 0	0.1 0	0.10	0.3 4	0.3 4	0.0 6	0.0 2	NA	0.18	0.14	NA
30	(Sérodes et al., 2003)ª	TTHM	53	0.78	0.3 0	0.0 0	0.6 7	0.5 3	0.59	0.0 6	0.7 1	0.3 6	0.2 0	0.0 3	0.20	0.39	0.2 5
31	(Sérodes et al., 2003) ^a	TTHM	55	0.89	0.0 2	0.2 7	0.6 2	0.1 8	0.35	0.1 1	0.2 1	0.5 0	0.1 1	0.0 1	0.53	0.62	0.4 2
32	(Sérodes et al., 2003) ^a	TTHM	51	0.56	0.0 9	0.1 7	0.0 7	0.0 7	0.08	0.1 2	0.6 4	0.2 3	0.2 2	0.1 2	0.54	0.76	0.0 2
33	(Sohn et	TTHM		0.90	0.3	0.1	0.7	0.4	0.48	0.3	0.8	0.2	0.1	0.1	0.78	0.62	0.4
	al., 2004)				6	3	6	6		3	9	4	5	7			1
34	(Sohn et	TTHM		0.92	0.3	0.1	0.8	0.4	0.69	0.1	0.6	0.6	0.3	0.1	0.77	0.66	0.4
25	dl., 2004) (Sobp.ot	ттим	288 d	0.02	03		3	5 05	0 43	9	9	0.1	4	03	0.80	0.74	3
55	al., 2004)	1 1 1 11*1	2005	0.92	0.5	2	6	5	0.45	7	8	5	1	3	0.80	0.74	7
36	(Sohn et	TTHM		0.91	0.3	0.0	0.7	0.5	0.61	0.0	0.8	0.5	0.0	0.3	0.63	0.54	0.6
~	al., 2004)				0	2	9	7		4	9	2	4	5			4
37	(Sohn et al., 2004)	IIHM		0.91	0.3	0.1	0.7	0.4 7	0.51	0.1	0.8	0.4	0.1 4	0.4	0.76	0.73	0.6 3
38	(Toroz and	TTHM	13	0.91	0.1	0.0	0.4	0.3	0.64	0.1	0.1	0.0	0.0	0.3	0.77	0.86	0.5
	Uyak, 2005)ª				8	0	0	0		3	9	1	2	7			8
39	(Urano et al., 1983)	TTHM			0.3 3	0.0 4	0.7 7	0.5 7	0.60	0.0 6	0.8 6	0.5 9	0.1 0	0.2 3	0.59	0.59	0.5 9
40	(Uyak et	TTHM		0.96	0.4	0.1	0.8	0.4	NA ^f	NA ^f	NA ^f	NA ^f	NA ^f	0.3	0.63	0.47	0.6
	al., 2005) ^a			(0.86)	5	8	0	5						9			7
41	(USEPA, 2001)	ГТНМ	288	0.92	0.3	0.0	0.7	0.4 9	0.44	0.1	0.9	0.2	0.0 8	0.3	0.93	0.88	0.7
82	(USEPA, 2001)	CF	288	0.90	0.2	0.0	0.4 3	0.1 7	0.33	0.5 8	0.0	0.9 5	0.8 9	0.4 1	0.96	0.83	0.6 0

_	Source	Specie	Source	Model ⁱ				Cur	rent Stu	Jdy					(Gerrity	et al., 2009))
#j		S	# of	Validat	All	Non	TiO ₂	SW	GAC	DS	5	80	160	All	Raw	Coagulate	TiO ₂
Σ			Sampl	ion	Dat	-					kWh	kWh	kWh	Dat	Wate	d Water	
			es		а	AOP					m ⁻³	m⁻3	m⁻³	а	r		
83	(USEPA,	BDCM	288	0.85	0.1	0.0	0.4	0.3	0.25	0.5	0.6	0.0	0.0	0.3	0.83	0.81	0.5
	2001)		0.45		4	1	6	4		6	5	1	0	1	0.01		5
84	(USEPA,	DBCM	245	0.94	0.0	0.0	0.0	0.0	0.11	0.0	0.3	0.0	0.0	0.0	0.01	0.00	0.0
85	(LISEPA	BE	171	0.93	0.0	00	0	<u> </u>	0.10	4	9	0.0	2	<u> </u>	0.07	0.05	0
05	2001)	ы	1/1	0.55	0.0	0.0	4	5	0.10	0.0	7	6	3	3	0.07	0.05	8
42	(USEPA,	TTHM	2946	0.83	0.2	0.0	0.7	0.4	0.31	0.1	0.9	0.1	0.0	0.3	0.88	0.86	0.6
	`2001) [′]				6	6	3	4		6	2	9	7	5			7
86	(USEPA,	CF	2960	0.73	0.2	0.0	0.6	0.2	0.49	0.6	0.2	0.8	0.4	0.3	0.90	0.80	0.6
	2001)				1	9	1	3		3	7	7	9	1			8
87	(USEPA,	BDCM	2959	0.58	0.3	0.0	0.7	0.4	0.47	0.5	0.8	0.2	0.0	0.4	0.92	0.89	0.4
	2001)				4	8	8	8		3	8	5	6	2			6
88	(USEPA,	DBCM	2948	0.73	0.2	0.0	0.7	0.3	0.24	0.2	0.8	0.1	0.0	0.3	0.88	0.86	0.0
	2001)				0	3	0	9		6	5	0	3	1			0
89	(USEPA,	BF	2948	0.77	0.0	0.0	0.4	0.2	0.14	0.5	0.6	0.0	0.0	0.2	0.61	0.72	0.0
42	2001)		706	0.01	6	1	3	9	0.00	5	8	0	0	4	0.76	0.70	6
43	(Zhu, 1995)	TTHM	786	0.81	0.2 q	0.1	0.8	0.4 8	0.68	0.0 R	0.5	0.6	0.1 8	0.2	0.76	0.70	0.5 8
90	(7hu 1995)	CF	786	0.79	0.2	01	0.8	04	0.76	0.8	0.5	0.6	01	03	0.80	0.72	0.6
50	(2110, 1999)	Ci	700	0.75	8	0	5	6	0.70	0.0	2	2	7	2	0.00	0.72	4
91	(Zhu, 1995)	BDCM	786	0.67	0.2	0.1	0.8	0.4	0.53	0.6	0.4	0.6	0.1	0.1	0.72	0.71	0.4
					6	8	6	7		0	9	6	9	5			0
92	(Zhu, 1995)	DBCM	786	0.79	0.0	0.0	0.0	0.0	0.07	0.0	0.6	0.1	0.2	0.0	0.11	0.07	0.0
					0	3	1	5		1	9	1	0	2			3
93	(Zhu, 1995)	BF	786	0.68	0.1	0.1	0.5	0.2	0.06	0.6	0.2	0.0	0.2	0.0	0.02	0.05	0.2
					0	6	1	7		0	1	1	1	0			0
44	(Zhu, 1995)	TTHM	786	0.83	0.3	0.0	0.5	0.5	0.47	0.0	0.7	0.4	0.3	0.2	0.72	0.67	0.5
0.4	(7) (005)	05	706	0.01	0	0	6	2	0.50	/	/	/	4	8	0.75		3
94	(Zhu, 1995)	CF	786	0.81	0.3	0.0	0.6	0.5	0.52	0.4	0.7	0.5	0.3	0.3	0.75	0.68	0.5
05	(7hu 1005)	BDCM	786	0.71	0.2	0	2 05	05	0.44	03	07	<u> </u>	03	01	0.71	0.67	0.2
55	(ZIIU, 1993)	DUCIT	700	0.71	7	0.0	9	3	0.77	8	9	9	0.5	1	0.71	0.07	9
96	(Zhu, 1995)	DBCM	786	0.80	, 0.0	0.1	0.0	0.0	0.09	0.0	0.5	0.3	0.3	0.0	0.10	0.02	0.0
	(,	550.1	,	0.00	6	5	4	8	0.00	2	9	4	7	1	0.10	0.02	1
97	(Zhu, 1995)	BF	786	0.75	0.0	0.2	0.3	0.2	0.28	0.5	0.0	0.2	0.4	0.0	0.09	0.03	0.0
					9	9	7	0		1	0	9	2	1			2

_	Source	Specie	Source	Model ⁱ				Cu	rent Stu	Jdy					(Gerrity	et al., 2009))
#jde		S	# of	Validat	All	Non	TiO ₂	SW	GAC	DS	5	80	160	All	Raw	Coagulate	TiO ₂
ž			Sampl	ion	Dat	-					kWh	kWh	kWh	Dat	Wate	d Water	
			es		а	AOP					m ⁻³	m ⁻³	m⁻³	а	r		
45	(Zhu, 1995)	TTHM	786		0.3	0.0	0.8	0.6	0.61	0.0	0.9	0.4	0.0	0.3	0.72	0.65	0.7
					3	1	4	0		4	1	9	4	4			9
46	(Zhu, 1995)	ТТНМ	786	0.70	0.4 0	0.1 6	0.7 3	0.4 7	0.57	0.3 1	0.8 9	0.3	0.2 3	0.4 9	0.82	0.83	0.6 7
98	(Zhu, 1995)	CF	786	0.61	0.3	0.2	0.7	0.3	0.60	0.6	0.4	0.7	0.6	0.4	0.87	0.82	0.6
					8	0	0	5		0	2	5	2	5			4
99	(Zhu, 1995)	BDCM	786	0.88	0.0	0.0	0.4	0.2	0.16	0.6	0.5	0.0	0.0	0.2	0.88	0.87	0.3
	(=)				7	1	4	5		2	5	0	0	7			7
10	(Zhu, 1995)	DBCM	786	0.89	0.0	0.0	0.0	0.0	0.11	0.0	0.3	0.0	0.0	0.0	0.11	0.12	0.0
0	(71	DE	706	0.61	1	0	1	0	0.00	2	8	5	2	0	0.02	0.00	1
10	(Znu, 1995)	BF	786	0.61	0.0	0.0	0.0	0.1	0.00	0.0	0.7	0.0	0.0	0.1	0.02	0.02	0.1
17	(7hu 100E)	ттым	706	0.01	4	2	0		0 55	0 1	0	1	0	4	0.96	0.02	1
47	(ZIIU, 1995)	IIII	/00	0.01	0.5 Q	0.0 Q	0.0	0.5	0.55	0.1	0.0	0.4	0.1 Q	0.4	0.00	0.85	0.0 Q
10	(7hu 1995)	CE	786	0.73	03	0.0	0.8	04	0.61	0.6	05	0.8	05	03	0.91	0.84	0.6
2	(2110, 1999)	Ci	700	0.75	4	9	2	0.4	0.01	2	5	2	6	2	0.51	0.04	5
10	(Zhu, 1995)	BDCM	786	0.90	0.0	0.0	0.5	0.2	0.16	0.5	0.5	0.0	0.0	0.2	0.82	0.86	0.4
3	(2.10) 2555)	00011	,	0150	8	0	0	8	0110	7	4	0	0	2	0.02	0100	1
10	(Zhu, 1995)	DBCM	786	0.89	0.0	0.0	0.1	0.3	0.01	0.4	0.3	0.1	0.0	0.1	0.17	0.14	0.1
4					5	0	8	6		7	8	6	7	6			3
10	(Zhu, 1995)	BF	786	0.61	0.0	0.0	0.0	0.1	0.03	0.1	0.7	0.0	0.0	0.1	0.06	0.00	0.0
5					2	2	5	2		3	6	1	0	6			7
	Number of	Samples	5		75	30	45	25	25	25	15	15	15	124	6	24	76
	Minin	num		0.11	0.0	0.0	0.0	0.0	0.00	0.0	0.0	0.0	0.0	0.0	0.01	0.00	0.0
				(0.84)	0	0	0	0		0	0	0	0	0			0
	Maxir	num		0.99	0.4	0.5	0.8	0.6	0.76	0.8	0.9	0.9	0.8	0.4	0.96	0.89	0.8
				(0.98)	5	4	9	4		5	6	6	9	9			4
	Aver	age		0.75	0.1	0.0	0.4	0.2	0.32	0.3	0.5	0.2	0.1	0.1	0.57	0.50	0.3
				(0.89)	7	9	2	8		3	3	5	7	7			4
	Standard I	Deviatior	1	0.22	0.1	0.1	0.3	0.2	0.21	0.2	0.3	0.2	0.2	0.1	0.29	0.28	0.2
- • •				(0.08)	3	2	2	0		5	1	5	0	4			7
	del developed i	using TOC	rather th	an DOC													
^o K ² I	not reported to	r validatio	n data se	t rmod by (Craina		(1002)										

^c Independent external validation performed by Greiner et al. (1992)
 ^d External validation reported by Zhu (1995) using data from Montgomery Watson (1991)
 ^e Reported by Zhu (1995)

_	Source	Specie	Source	Model ⁱ				Cu	rrent Stu	udy					(Gerrity	et al., 2009))
#jde		S	# of	Validat	All	Non	TiO ₂	SW	GAC	DS	5	80	160	All	Raw	Coagulate	TiO ₂
Σ			Sampl	ion	Dat	-					kWh	kWh	kWh	Dat	Wate	d Water	
			es		а	AOP					m ⁻³	m⁻³	m⁻³	а	r		
f NA	- Not applicabl	le; model	does not r	eturn mea	aningfu	I result	ts. Pos	sible re	easons i	nclude	all pred	dictions	are id	entical	(so no F	R ² value), inp	out
da	ta missing (e.g	nissing (e.g., no UV ₂₅₄), etc.															
^g Re	ported by USE	nissing (e.g., no UV ₂₅₄), etc. ed by USEPA (2001)															
^h So	me models cou	Id not be	calculated	for some	data s	ets due	e to lacl	< of UV	254 data	. For t	his sub	set of a	data/m	odels,	n = 181	for the AOP	data
se	t, n = 201 for i	non-AOP d	lata, n = 1	L43 for Ti	D ₂ data	set, ai	nd n =	38 for	all othei	r types	of AOF	' data					
i Val m	ues reported an odel	re internal	modeling	values fo	r # of s	sources	and va	alidatio	n, while	values	s show	n in pa	renthe	ses are	for exte	ernal validati	on of
^j Mo	del # refers to	number a	ssigned in	Table S4													
^k NR	- Not reported	d; althoug	h validatio	on was pei	formed	d, R ² w	as not i	provide	ed								

208 Table S9 (Continued) – Average correlation coefficients (R² values) for the predicted THMs versus measured THMs

	Source	Specie						Externa	I AOP Data				
Model # ^j		S	(Lamsa l et al., 2011)	(Liu et al., 2010)	(Liu et al., 2008a)	(Liu et al., 2008b)	(Liu et al., 2007)	(Philipp e et al., 2010b)	(Philipp e et al., 2010a)	(Dotso n et al., 2010)	(Sarathy and Mohseni, 2010)	(Ken t et al., 2011)	(Kleiser and Frimmel, 2000)
1	(Al-Omari et al., 2004)ª	TTHM	0.56	0.73		0.95	0.08		0.73	0.29	0.29	0.87	0.32
2	(Amy et al., 1998)	TTHM	0.56	0.81		0.97	0.08		0.73	0.04	0.29	0.87	0.34
48	(Amy et al., 1998)	CF			0.52			0.06				0.84	
49	(Amy et al., 1998)	BDCM										0.74	
50	(Amy et al., 1998)	DBCM										NA ^f	
51	(Amy et al., 1998)	BF										NA ^f	
3	(Amy et al., 1998)	TTHM	0.56	0.76		0.96	0.08		0.74	0.09	0.29	0.86	0.33
52	(Amy et al., 1998)	CF			0.72			0.06				0.84	

	Source	Specie						External	AOP Data				
Model # ^j		S	(Lamsa l et al., 2011)	(Liu et al., 2010)	(Liu et al., 2008a)	(Liu et al., 2008b)	(Liu et al., 2007)	(Philipp e et al., 2010b)	(Philipp e et al., 2010a)	(Dotso n et al., 2010)	(Sarathy and Mohseni, 2010)	(Ken t et al., 2011	(Kleiser and Frimmel, 2000)
53	(Amy et al., 1998)	BDCM		, , , , , , , , , , , , , , , , , , ,			/					0.74	
54	(Amy et al., 1998)	DBCM										NA ^f	
55	(Amy et al., 1998)	BF										NA ^f	
4	(Amy et al., 1998)	TTHM	0.56	0.78		0.96	0.08		0.73	0.10	0.29	0.86	0.33
56	(Amy et al., 1998)	CF			0.73			0.07				0.84	
57	(Amy et al., 1998)	BDCM										0.74	
58	(Amy et al., 1998)	DBCM										NA ^f	
59	(Amy et al., 1998)	BF										NA ^f	
5	(Amy et al., 1998)	TTHM	0.56	0.77		0.96	0.08		0.73	0.09	0.29	0.86	0.33
60	(Amy et al., 1998)	CF			0.72			0.06				0.84	
61	(Amy et al., 1998)	BDCM										0.74	
62	(Amy et al., 1998)	DBCM										NA ^f	
63	(Amy et al., 1998)	BF										NA ^f	
6	(Boyalla, 2004)	TTHM	0.57	0.75		0.96	0.08		0.57	0.55	0.29	0.88	0.33
7	(Chang et al., 1996)	TTHM	0.91	0.88		0.86	NA ^f		0.42	0.07	0.70	NA ^f	0.97
8	(Chang et al., 1996)	TTHM	0.56	0.65		0.93	0.07		0.76	0.01	0.29	0.86	0.31
9	(Chang et al., 1996)	TTHM	0.90	0.88		0.86	NA ^f		0.39	0.10	0.70	NA ^f	0.97

	Source	Specie						Externa	I AOP Data				
Model # ^j		S	(Lamsa l et al., 2011)	(Liu et al., 2010)	(Liu et al., 2008a)	(Liu et al., 2008b)	(Liu et al., 2007)	(Philipp e et al., 2010b)	(Philipp e et al., 2010a)	(Dotso n et al., 2010)	(Sarathy and Mohseni, 2010)	(Ken t et al., 2011)	(Kleiser and Frimmel, 2000)
10	(Chen and Westerhoff, 2010)	TTHM	0.85	0.93		0.89	NA ^f		0.41	0.12	0.71	NA ^f	0.98
64	(Chen and Westerhoff, 2010)	CF			NA ^f			NA ^f				NA ^f	
65	(Chen and Westerhoff, 2010)	BDCM										NA ^f	
11	(Harrington et al., 1992)	TTHM	0.81	0.94		0.95	NA ^f		0.58	0.01	0.73	NA ^f	0.99
12	(Hong et al., 2007)	TTHM	NA ^f	0.50		0.89	0.07		NA ^f	0.31	0.29	NA ^f	0.30
66	(Hong et al., 2007)	BDCM										NA ^f	
67	(Hong et al., 2007)	CF			0.27			0.00				NA ^f	
68	(Malcom Pirnie Inc, 1992)	CF			NA ^f							NA ^f	
69	(Malcom Pirnie Inc, 1992)	BDCM										NA ^f	
70	2(Malcom Pirnie Inc, 1992)	DBCM										NA ^f	
71	(Malcom Pirnie Inc, 1993)	CF			NA ^f							NA ^f	
13	(McBean et al., 2008)	TTHM	<u>0.56</u>	0.68		0.96	0.08		0.74	<u>0.06</u>	0.29	<u>0.86</u>	0.33
14	(Milot et al., 2002)	TTHM	0.56	0.81		0.97	0.08		<u>0.72</u>	0.08	<u>0.29</u>	<u>0.87</u>	0.34
72	(Montgomery Watson, 1993)	CF			NA ^f							NA ^f	
73	(Montgomery Watson, 1993)	BDCM										0.74	

	Source	Specie						External	AOP Data				
Model # ^j		S	(Lamsa l et al., 2011)	(Liu et al., 2010)	(Liu et al., 2008a)	(Liu et al., 2008b)	(Liu et al., 2007)	(Philipp e et al., 2010b)	(Philipp e et al., 2010a)	(Dotso n et al., 2010)	(Sarathy and Mohseni, 2010)	(Ken t et al., 2011)	(Kleiser and Frimmel, 2000)
74	(Montgomery Watson, 1993)	BDCM										0.74	
75	(Montgomery Watson, 1993)	DBCM										NA ^f	
76	(Montgomery Watson, 1993)	DBCM										NA ^f	
77	(Montgomery Watson, 1993)	BF										NA ^f	
78	(Rathbun, 1996b)	CF			0.48			0.06				0.84	
79	(Rathbun, 1996b)	BDCM										0.74	
80	(Rathbun, 1996b)	DBCM										NA ^f	
81	(Rathbun, 1996b)	BF										NA ^f	
15	(Rathbun, 1996a)	TTHM	0.87	0.92		0.89	NA ^f		0.43	0.08	0.71	NA ^f	0.98
16	(Rodriguez et al., 2000)	TTHM	0.56	0.80		0.97	0.08		0.72	0.08	0.29	0.87	0.34
17	(Rodriguez et al., 2000)	TTHM	<u>0.56</u>	0.81		0.97	0.08		0.73	0.01	0.29	<u>0.86</u>	<u>0.34</u>
18	(Semerjian et al., 2009)	TTHM	0.88	0.90		0.87	NA ^f		0.39	0.12	0.70	NA ^f	0.97
19	(Semerjian et al., 2009)	TTHM	0.74	0.95		0.94	NA ^f		0.46	0.12	0.71	NA ^f	0.99
20	(Semerjian et al., 2009)	TTHM	0.73	0.81		0.88	NA ^f		0.18	0.48	0.68	NA ^f	0.97
21	(Semerjian et al., 2009)	TTHM	0.94	0.65		0.72	NA ^f		0.23	0.11	0.42	NA ^f	0.89
22	(Semerjian et al., 2009)	TTHM	0.89	0.90		0.87	NA ^f		0.58	0.01	0.70	NA ^f	0.97
23	(Semerjian et al., 2009)	TTHM	0.56	0.00		0.00	NA ^f		0.76	0.70	NA ^f	0.86	NA ^f

	Source	Specie						Externa	AOP Data				
Model # ^j		S	(Lamsa l et al., 2011)	(Liu et al., 2010)	(Liu et al., 2008a)	(Liu et al., 2008b)	(Liu et al., 2007)	(Philipp e et al., 2010b)	(Philipp e et al., 2010a)	(Dotso n et al., 2010)	(Sarathy and Mohseni, 2010)	(Ken t et al., 2011)	(Kleiser and Frimmel, 2000)
24	(Semerjian et al., 2009)	TTHM	0.92	0.78		0.80	NA ^f		0.66	0.06	0.67	NA ^f	0.92
25	(Semerjian et al., 2009)	TTHM	0.97	0.65		0.72	NA ^f		0.23	0.11	0.52	NA ^f	0.92
26	(Semerjian et al., 2009)	TTHM	0.56	NA ^f		NA ^f	NA ^f		0.74	0.70	NA ^f	0.86	0.00
27	(Semerjian et al., 2009)	ТТНМ	0.56	0.00		0.00	NA ^f		0.76	0.57	0.00	0.86	NA ^f
28	(Semerjian et al., 2009)	TTHM	0.56	0.00		NA ^f	NA ^f		0.74	0.71	NA ^f	0.86	NA ^f
29	(Semerjian et al., 2009)	TTHM	0.56	NA ^f		NA ^f	0.00		0.26	0.12	NA ^f	0.88	0.00
30	(Sérodes et al., 2003)ª	TTHM	0.56	0.80		0.97	0.08		0.74	0.14	0.29	0.86	0.34
31	(Sérodes et al., 2003)ª	TTHM	0.57	0.76		0.88	0.13		0.56	0.01	0.28	0.88	0.43
32	(Sérodes et al., 2003)ª	TTHM	0.56	NA ^f		NA ^f	NA ^f		0.74	0.70	NA ^f	0.86	0.00
33	(Sohn et al., 2004)	TTHM	0.84	0.95		0.92	NA ^f		0.48	0.02	0.71	NA ^f	0.98
34	(Sohn et al., 2004)	TTHM	0.82	0.94		0.94	NA ^f		0.56	0.00	0.73	NA ^f	0.99
35	(Sohn et al., 2004)	TTHM	0.56	0.77		0.96	0.08		0.73	0.09	0.29	0.86	0.33
36	(Sohn et al., 2004)	TTHM	0.56	0.82		0.97	0.09		0.71	0.11	0.29	0.87	0.34
37	(Sohn et al., 2004)	TTHM	0.84	0.93		0.93	NA ^f		0.54	0.00	0.73	NA ^f	0.99
38	(Toroz and Uyak, 2005) ^a	TTHM	0.56	0.68		0.94	0.07		0.73	0.42	0.29	0.86	0.32
39	(Urano et al., 1983)	TTHM	0.56	0.80		0.97	0.08		0.73	0.07	0.29	0.87	0.34
40	(Uyak et al., 2005) ^a	TTHM	NA ^f	0.82		0.97	NA ^f		0.73	0.62	0.29	0.87	0.34

	Source	Specie						External	AOP Data				
Model # ^j		S	(Lamsa l et al., 2011)	(Liu et al., 2010)	(Liu et al., 2008a)	(Liu et al., 2008b)	(Liu et al., 2007)	(Philipp e et al., 2010b)	(Philipp e et al., 2010a)	(Dotso n et al., 2010)	(Sarathy and Mohseni, 2010)	(Ken t et al., 2011)	(Kleiser and Frimmel, 2000)
41	(USEPA, 2001)	TTHM	0.82	0.94		0.94	NA ^f		0.56	0.00	0.73	NA ^f	0.99
82	(USEPA, 2001)	CF			NA ^f			NA ^f				NA ^f	
83	(USEPA, 2001)	BDCM										NA ^f	
84	(USEPA, 2001)	DBCM										NA ^f	
85	(USEPA, 2001)	BF										NA ^f	
42	(USEPA, 2001)	TTHM	0.83	0.93		0.93	NA ^f		0.53	0.01	0.73	NA ^f	0.99
86	(USEPA, 2001)	CF			NA ^f			NA ^f				NA ^f	
87	(USEPA, 2001)	BDCM										NA ^f	
88	(USEPA, 2001)	DBCM										NA ^f	
89	(USEPA, 2001)	BF										NA ^f	
43	(Zhu, 1995)	TTHM	0.56	0.80		0.97	0.08		0.74	0.10	0.29	0.86	0.34
90	(Zhu, 1995)	CF			0.59			0.05				0.84	
91	(Zhu, 1995)	BDCM										0.74	
92	(Zhu, 1995)	DBCM										NA ^f	
93	(Zhu, 1995)	BF										NA ^f	
44	(Zhu, 1995)	TTHM	0.56	0.80		0.97	0.08		0.74	0.10	0.29	0.86	0.34
94	(Zhu, 1995)	CF			0.59			0.05				0.84	
95	(Zhu, 1995)	BDCM										0.74	
96	(Zhu, 1995)	DBCM										NA ^f	
97	(Zhu, 1995)	BF										NA ^f	
45	(Zhu, 1995)	TTHM	0.56	0.81		0.97	0.08		0.73	0.04	0.29	0.87	0.34
46	(Zhu, 1995)	TTHM	0.81	0.96		0.93	NA ^f		0.51	0.01	0.71	NA ^f	0.99
98	(Zhu, 1995)	CF			NA ^f			NA ^f				NA ^f	
99	(Zhu, 1995)	BDCM										NA ^f	

	Source	Specie						External	AOP Data				
ť#		S	(Lamsa	(Liu	(Liu	(Liu	(Liu	(Philipp	(Philipp	(Dotso	(Sarathy	(Ken	(Kleiser
del			l et al.,	et	et al.,	et al.,	et	e et al.,	e et al.,	n et	and	t et	and
10(2011)	al.,	2008a	2008b	al.,	2010b)	2010a)	al.,	Mohseni,	al.,	Frimmel,
2				2010))	2007			2010)	2010)	2011	2000)
10	(Zhu, 1995)	DBCM										NA ^f	
0													
10 1	(Zhu, 1995)	BF										NA ^f	
47	(Zhu, 1995)	TTHM	0.82	0.94		0.94			0.55	0.01	0.73	NA ^f	0.99
10	(Zhu, 1995)	CF			NA ^f		NA ^f	NA ^f				NA ^f	
2													
10	(Zhu, 1995)	BDCM										NA ^f	
3	(=) (0.0.7)												
10 4	(Zhu, 1995)	DBCM										NA'	
10	(Zhu, 1995)	BF										NA ^f	
5													
	Number of Sam	ples	5	12	10	10	5	54	6	14	9	4	14
	Minimum		0.56	0.00	0.27	0.00	0.00	0.00	0.18	0.00	0.00	0.74	0.00
	Maximum		0.97	0.96	0.73	0.97	0.13	0.07	0.76	0.71	0.73	0.88	0.99
	Average		0.68	0.76	0.58	0.88	0.08	0.05	0.60	0.18	0.46	0.84	0.59
	Standard Deviat	tion	0.15	0.23	0.16	0.21	0.02	0.02	0.17	0.22	0.22	0.05	0.35

^b R² not reported for validation data set

^c Independent external validation performed by Greiner et al. (1992)

^d External validation reported by Zhu (1995) using data from Montgomery Watson (1991)

^e Reported by Zhu (1995)

^f NA - Not applicable; model does not return meaningful results. Possible reasons include all predictions are identical (so no R² value), input data missing (e.g., no UV₂₅₄), etc.

⁹ Reported by USEPA (2001)

^h Some models could not be calculated for some data sets due to lack of UV_{254} data. For this subset of data/models, n = 181 for the AOP data set, n = 201 for non-AOP data, n = 143 for TiO₂ data set, and n = 38 for all other types of AOP data

ⁱ Values reported are internal modeling values for # of sources and validation, while values shown in parentheses are for external validation of model

^j Model # refers to number assigned in Table S4

^k NR – Not reported; although validation was performed, R² was not provided

210	Table S9 (Continued) – Average correlation coefficients (R ² values) for pairs of predictive THM models versus measured THM
211	datasets

Łj	Source	Species			External Non	-AOP Data				Groupe	d Data	
Model #			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
1	(Al-Omari et al., 2004) ^a	TTHM		0.26	0.47	0.70	0.87	0.76	0.25	0.59	0.2 9	0.36
2	(Amy et al., 1998)	TTHM		0.17	0.62	0.70	0.79	0.88	0.48	0.41	0.6 4	0.56
48	(Amy et al., 1998)	CF	0.63	0.43					0.08	0.00	0.0 8	0.63
49	(Amy et al., 1998)	BDCM		0.12					0.70	0.61	0.7 0	
50	(Amy et al., 1998)	DBCM		0.19					0.09	0.07	0.0 9	
51	(Amy et al., 1998)	BF							0.01	0.17	0.0 1	
3	(Amy et al., 1998)	TTHM		0.56	0.80	0.75	0.41	0.83	0.38	0.38	0.4 9	0.37
52	(Amy et al., 1998)	CF	0.60	0.59					0.07	0.07	0.0 6	0.59
53	(Amy et al., 1998)	BDCM		0.62					0.75	0.55	0.7 5	
54	(Amy et al., 1998)	DBCM		0.72					0.04	0.01	0.0 4	
55	(Amy et al., 1998)	BF							0.00	0.23	0.0 0	
4	(Amy et al., 1998)	TTHM		0.51	0.80	0.75	0.43	0.78	0.23	0.38	0.4 2	0.32
56	(Amy et al., 1998)	CF	0.60	0.60					0.07	0.08	0.0 5	0.60
57	(Amy et al., 1998)	BDCM		0.62					0.74	0.55	0.7 4	
58	(Amy et al., 1998)	DBCM		0.70					0.07	0.00	0.0 7	
59	(Amy et al., 1998)	BF							0.00	0.25	0.0 0	

Ŀ.	Source	Species			External Non	-AOP Data				Groupe	d Data	
Model #			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
5	(Amy et al., 1998)	TTHM		0.72	0.80	0.75	0.42	0.80	0.32	0.37	0.4 5	0.34
60	(Amy et al., 1998)	CF	0.60	0.60					0.07	0.08	0.0 6	0.59
61	(Amy et al., 1998)	BDCM		0.62					0.75	0.55	0.7 5	
62	(Amy et al., 1998)	DBCM		0.72					0.06	0.00	0.0 6	
63	(Amy et al., 1998)	BF							0.00	0.24	0.0 0	
6	(Boyalla, 2004)	TTHM		0.11	0.28	0.57	0.76	0.35	0.10	0.44	0.0 7	0.22
7	(Chang et al., 1996)	TTHM		0.60	NA ^f	0.92	0.27	0.74	0.72	0.58	0.6 7	0.86
8	(Chang et al., 1996)	TTHM		0.61	0.18	0.47	0.06	0.58	0.55	0.16	0.4 4	0.78
9	(Chang et al., 1996)	TTHM		0.60	NA ^f	0.92	0.25	0.71	0.72	0.60	0.6 5	0.84
10	(Chen and Westerhoff, 2010)	ТТНМ		0.02	NA ^f	0.60	0.00	0.81	0.34	0.53	0.6 6	0.07
64	(Chen and Westerhoff, 2010)	CF	0.88	0.04					0.41	0.47	0.0 7	0.87
65	(Chen and Westerhoff, 2010)	BDCM		0.51					0.04	0.50	0.0 4	
11	(Harrington et al., 1992)	ТТНМ		0.28	NA ^f	0.87	0.92	0.87	0.57	0.69	0.6 6	0.90
12	(Hong et al., 2007)	TTHM		0.16	0.83	0.30	0.83	0.11	0.00	0.12	0.0 2	0.15
66	(Hong et al., 2007)	BDCM		0.01					0.49	0.49	0.4 9	
67	(Hong et al., 2007)	CF	0.49	0.37					0.02	0.01	0.0 1	0.49

Łj	Source	Species			External Non	-AOP Data				Groupe	d Data	
Model #			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
68	(Malcom Pirnie Inc, 1992)	CF		0.24					0.45	0.42	0.0 6	0.95
69	(Malcom Pirnie Inc, 1992)	BDCM		0.13					0.69	0.70	0.6 9	
70	(Malcom Pirnie Inc, 1992)	DBCM		0.23					0.02	0.06	0.0 2	
71	(Malcom Pirnie Inc, 1993)	CF		0.18					0.46	0.38	0.0 5	0.95
13	(McBean et al., 2008)	ТТНМ		0.00	0.30	0.36	0.19	0.85	0.39	0.57	0.5 2	0.32
14	(Milot et al., 2002)	ТТНМ		0.26	0.44	0.71	0.86	0.84	0.53	0.60	0.5 8	0.46
72	(Montgomery Watson, 1993)	CF	0.91	0.24					0.55	0.19	0.7 5	0.91
73	(Montgomery Watson, 1993)	BDCM		0.02					0.39	0.62	0.3 9	
74	(Montgomery Watson, 1993)	BDCM		0.69					0.00	0.45	0.0 0	
75	(Montgomery Watson, 1993)	DBCM		0.34					0.14	0.04	0.1 4	
76	(Montgomery Watson, 1993)	DBCM		0.17					0.03	0.26	0.0 3	
77	(Montgomery Watson, 1993)	BF							0.01	0.25	0.0 1	
78	(Rathbun, 1996b)	CF	0.58	0.07					0.03	0.01	0.0 3	0.58
79	(Rathbun, 1996b)	BDCM		0.11					0.67	0.50	0.6 7	
80	(Rathbun, 1996b)	DBCM		0.15					0.01	0.08	0.0 1	
81	(Rathbun, 1996b)	BF							0.03	0.22	0.0 3	
15	(Rathbun, 1996a)	TTHM		0.12	NA ^f	0.92	0.81	0.77	0.71	0.74	0.6 7	0.85
16	(Rodriguez et al., 2000)	TTHM		0.26	0.46	0.72	0.86	0.84	0.54	0.60	0.6 0	0.46

Łj	Source	Species			External Non	-AOP Data				Groupe	d Data	
Model #			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
17	(Rodriguez et al., 2000)	TTHM		0.00	0.09	0.41	0.32	0.85	0.48	0.37	0.7 0	0.38
18	(Semerjian et al., 2009)	ТТНМ		0.56	NA ^f	0.80	0.11	0.69	0.76	0.64	0.6 8	0.93
19	(Semerjian et al., 2009)	ТТНМ		0.02	NA ^f	0.57	0.00	0.44	0.21	0.44	0.6 1	0.28
20	(Semerjian et al., 2009)	ТТНМ		0.00	NA ^f	0.18	0.13	0.06	0.47	0.57	0.4 9	0.36
21	(Semerjian et al., 2009)	ТТНМ		0.00	NA ^f	0.46	0.00	0.08	0.05	0.22	0.0 9	0.42
22	(Semerjian et al., 2009)	ТТНМ		0.00	NA ^f	0.61	0.16	0.84	0.23	0.66	0.2 7	0.70
23	(Semerjian et al., 2009)	TTHM		NA ^f	0.21	0.54	0.18	0.71	0.11	0.60	0.0 7	0.23
24	(Semerjian et al., 2009)	TTHM		0.00	NA ^f	0.61	0.13	0.85	0.10	0.64	0.1 2	0.53
25	(Semerjian et al., 2009)	TTHM		0.00	NA ^f	0.53	0.00	0.11	0.08	0.24	0.1 3	0.47
26	(Semerjian et al., 2009)	TTHM		0.58	0.55	0.81	0.37	0.64	0.14	0.62	0.0 4	0.47
27	(Semerjian et al., 2009)	TTHM		0.17	0.55	0.71	0.51	0.09	0.00	0.29	0.0 0	0.21
28	(Semerjian et al., 2009)	TTHM		0.49	0.46	0.65	0.17	0.64	0.09	0.56	0.0 3	0.32
29	(Semerjian et al., 2009)	TTHM		0.10	NA ^f	0.55	0.34	NA ^f	NA ^f	0.00	NA ^f	0.08
30	(Sérodes et al., 2003)ª	TTHM		0.56	0.01	0.68	0.24	0.83	0.61	0.56	0.5 3	0.73
31	(Sérodes et al., 2003)ª	TTHM		0.61	0.31	0.36	0.14	0.72	0.20	0.50	0.4 8	0.17
32	(Sérodes et al., 2003)ª	TTHM		0.56	0.37	0.50	0.33	0.64	0.36	0.62	0.1 0	0.74
33	(Sohn et al., 2004)	TTHM		0.62	NA ^f	0.92	0.47	0.88	0.76	0.64	0.7 8	0.87
34	(Sohn et al., 2004)	TTHM		0.34	NA ^f	0.88	0.39	0.70	0.78	0.61	0.7 3	0.91

¢j	Source	Species			External Non			Groupe	d Data			
Model #			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
35	(Sohn et al., 2004)	TTHM		0.28	0.75	0.75	0.85	0.81	0.23	0.42	0.3 9	0.39
36	(Sohn et al., 2004)	ТТНМ		0.19	0.47	0.71	0.92	0.80	0.47	0.64	0.5 0	0.44
37	(Sohn et al., 2004)	ТТНМ		0.12	NA ^f	0.87	0.91	0.83	0.66	0.72	0.7 2	0.92
38	(Toroz and Uyak, 2005)ª	ТТНМ		0.00	0.25	0.52	0.17	0.73	0.22	0.67	0.0 2	0.30
39	(Urano et al., 1983)	ТТНМ		0.22	0.51	0.74	0.80	0.85	0.68	0.39	0.7 1	0.74
40	(Uyak et al., 2005)ª	TTHM		0.00	NA ^f	0.28	0.38	0.48	0.00	0.05	0.1 0	0.09
41	(USEPA, 2001)	TTHM		0.29	NA ^f	0.88	0.83	0.93	0.53	0.61	0.8 1	0.81
82	(USEPA, 2001)	CF	0.97	0.53					0.45	0.40	0.0 1	0.97
83	(USEPA, 2001)	BDCM		0.26					0.78	0.76	0.7 8	
84	(USEPA, 2001)	DBCM		0.46					0.10	0.04	0.1 0	
85	(USEPA, 2001)	BF							0.00	0.19	0.0 0	
42	(USEPA, 2001)	TTHM		0.17	NA ^f	0.88	0.80	0.91	0.43	0.57	0.7 7	0.60
86	(USEPA, 2001)	CF	0.95	0.58					0.52	0.46	0.6 0	0.95
87	(USEPA, 2001)	BDCM		0.01					0.70	0.84	0.7 0	
88	(USEPA, 2001)	DBCM		0.04					0.18	0.00	0.1 8	
89	(USEPA, 2001)	BF							0.05	0.24	0.0 5	
43	(Zhu, 1995)	TTHM		0.07	0.39	0.65	0.56	0.85	0.35	0.59	0.2 8	0.65
90	(Zhu, 1995)	CF	0.59	0.23					0.06	0.05	0.0 4	0.58

¢j	Source	Species			External Non			Groupe	d Data			
Model #			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
91	(Zhu, 1995)	BDCM		0.14					0.78	0.62	0.7 8	
92	(Zhu, 1995)	DBCM		0.17					0.17	0.11	0.1 7	
93	(Zhu, 1995)	BF							0.03	0.04	0.0 3	
44	(Zhu, 1995)	TTHM	0.59	0.25	0.56	0.69	0.59	0.85	0.44	0.57	0.4 1	0.59
94	(Zhu, 1995)	CF		0.40					0.07	0.06	0.0 6	0.58
95	(Zhu, 1995)	BDCM		0.32					0.76	0.62	0.7 6	
96	(Zhu, 1995)	DBCM		0.02					0.27	0.09	0.2 7	
97	(Zhu, 1995)	BF							0.04	0.42	0.0 4	
45	(Zhu, 1995)	TTHM		0.17	0.62	0.70	0.79	0.88	0.48	0.41	0.6 4	0.56
46	(Zhu, 1995)	TTHM		0.17	NA ^f	0.91	0.91	0.87	0.77	0.75	0.8 0	0.90
98	(Zhu, 1995)	CF	0.91	0.43					0.63	0.54	0.6 1	0.91
99	(Zhu, 1995)	BDCM		0.11					0.66	0.78	0.6 6	
10 0	(Zhu, 1995)	DBCM		0.19					0.11	0.05	0.1 1	
10 1	(Zhu, 1995)	BF							0.00	0.18	0.0 0	
47	(Zhu, 1995)	TTHM		0.17	NA ^f	0.88	0.79	0.90	0.67	0.64	0.8 2	0.90
10 2	(Zhu, 1995)	CF	0.95	0.43					0.84	0.71	0.7 2	0.95
10 3	(Zhu, 1995)	BDCM		0.11					0.68	0.73	0.6 8	
10 4	(Zhu, 1995)	DBCM		0.19					0.01	0.15	0.0 1	

¢j	Source	Species			External Non	-AOP Data				Groupe	d Data	
Model ∉			(Chin and Bérubé, 2005)	(Chowdhur y et al., 2010)	(Elshorbag y, 2000)	(Gang et al., 2002)	(Hua and Reckhow, 2008)	(Zhang et al., 2011)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
10 5	(Zhu, 1995)	BF							0.00	0.17	0.0 0	
	Number of Sam	ples	13	32	40	56	20	13	243 ^h	261 ^h	191 h	52 ^h
Minimum		0.49	0.00	0.01	0.18	0.00	0.06	0.00	0.00	0.0 0	0.07	
	Maximum		0.97	0.72	0.83	0.92	0.92	0.93	0.84	0.84	0.8 2	0.97
	Average		0.72	0.29	0.46	0.67	0.47	0.69	0.33	0.39	0.3 4	0.58
	Standard Devia	tion	0.18	0.23	0.23	0.19	0.32	0.25	0.28	0.24	0.3 0	0.27

^b R² not reported for validation data set

^c Independent external validation performed by Greiner et al. (1992)

^d External validation reported by Zhu (1995) using Montgomery Watson (1991)

^e Reported by Zhu (1995)

^f NA - Not applicable; model does not return meaningful results. Possible reasons include all predictions are identical (so no R² value), input data missing (e.g., no UV₂₅₄), etc.

^g Reported by USEPA (2001)

^h Some models could not be calculated for some data sets due to lack of UV_{254} data. For this subset of data/models, n = 181 for the AOP data set, n = 201 for non-AOP data, n = 143 for TiO₂ data set, and n = 38 for all other types of AOP data

ⁱ Values reported are internal modeling values for # of sources and validation, while values shown in parentheses are for external validation of model

^j Model # refers to number assigned in Table S4

^k NR – Not reported; although validation was performed, R² was not provided

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213 **Table S10** – Average correlation coefficients (R² values) for pairs of predictive HAA models versus measured HAA datasets

#j	Source	Specie	Source	e Model ⁱ	Current Study									External Non-
del		5	# of	Validatio	All	Non-	TiO ₂	SW	GAC	DS	5 kWh	80	160	(Fabbricino
ω			Sample	n	data	AOP					m⁻³	kWh m⁻³	kWh m⁻ ₃	and Korshin, 2009)
	(Amy et al.,	HAA5 ^d	738		0.0	0.13	0.1	0.0	0.34	0.0	0.84	0.43	0.06	0.12
-	1998)		(60)		6		4	0		2				
11	(Amy et al.,	MCA	738	0.14	0.0	0.09	0.2	0.0	0.09	0.0	0.13	0.20	0.17	0.07
3	1998) (Amy ot al	MBA	(60)	0.43	8	0.03	1	03	0.27	0	0.32	0.00	0.04	0.00
4	(Any et al., 1998)	MDA	(60)	0.45	0.0	0.05	0.0	6	0.27	7	0.52	0.00	0.04	0.00
11	(Amy et al.,	DCA	738	0.83	0.2	0.04	0.8	0.2	0.64	0.1	0.22	0.02	0.00	0.69
5	1998)		(60)	(0.86)	2		1	2		0				
11	(Amy et al.,	TCA	738	0.87	0.3	0.41	0.5	0.2	0.64	0.1	0.09	0.18	0.14	0.16
6	1998)		(60)	(0.91)	3		6	4		9				
11	(Amy et al.,	DBA	738	0.77	0.0	0.00	0.0	0.0	0.42	0.8	0.48	NA ^f	NA ^f	0.19
/	1998)		(60)		0	0.00	3	/	0.10	1	0.22	0.46	0.00	0.24
	(Amy et al., 1998)	HAA5°	288		0.0	0.06	0.0 5	0.0	0.13	0.0	0.22	0.46	0.02	0.24
11	(Amy et al.,	MCA	288	0.3	0.0	0.01	0.1	0.0	0.13	0.0	0.03	0.22	0.24	0.23
8	1998)				2		4	1		1				
11	(Amy et al.,	MBA	288	0.33	0.0	0.01	0.3	0.0	0.33	0.2	0.30	0.01	0.03	0.15
9	1998)				3		3	2		4	0.05	0.10		0.50
12	(Amy et al.,	DCA	288	0.84	0.2	0.30	0.4	0.1	0.39	0.1	0.25	0.10	0.00	0.59
12	(Amy et al	тсл	200	0.02	1	0 55	0 1	2	0.64	9	0.01	0.22	0.11	0.14
1	(Any et al., 1998)	ICA	200	0.95	0.5	0.55	0.4	3	0.04	0.2	0.01	0.52	0.11	0.14
12	(Amy et al.,	DBA	288	0.85	0.0	0.00	0.0	0.0	0.44	0.4	0.50	NA ^f	NA ^f	0.29
2	1998)				0		3	5	••••	6				
	(Amy et al.,	HAA5 ^d	288		0.0	0.08	0.0	0.0	0.00	0.0	0.34	0.50	0.05	0.25
	1998)				3		1	0		0				
12	(Amy et al.,	MCA	288	0.44	0.0	0.02	0.1	0.0	0.14	0.0	0.13	0.25	0.28	0.74
3	1998)				2		3	0		4				
12	(Amy et al.,	MBA	288	0.39	0.0	0.02	0.2	0.0	0.15	0.2	0.30	0.01	0.03	0.03
4	(Amy et al		288	0.9	∠ 0.2	0.25	0.6	01	0 54	01	0.24	0.00	0.00	0.61
5	1998)	DCA	200	0.5	3	0.25	0	4	0.54	9	0.27	0.00	0.00	0.01
12	(Amy et al.,	TCA	288	0.94	0.3	0.49	0.5	0.2	0.66	0.2	0.03	0.26	0.13	0.16
6	`				5		6	3		4				
12	(Amy et al.,	DBA	288	0.92	0.0	0.00	0.0	0.0	0.44	0.3	0.51	NAf	NA ^f	0.30
7	1998)				1		3	6		3				

#j	Source	Specie s	Source	e Model ⁱ	Current Study									External Non- AOP
Model			# of Sample s	Validatio n	All data	Non- AOP	TiO ₂	SW	GAC	DS	5 kWh m ⁻³	80 kWh m ⁻³	160 kWh m ⁻ 3	(Fabbricino and Korshin, 2009)
	(Amy et al., 1998)	HAA5 ^d	288		0.0 3	0.07	0.0 4	0.0 0	0.05	0.0 0	0.27	0.48	0.03	0.25
12 8	(Amy et al., 1998)	MCA	288	0.36	0.0 2	0.02	0.1 4	0.0 1	0.14	0.0 3	0.07	0.24	0.27	0.40
12 9	(Amy et al., 1998)	MBA	288	0.35	0.0 3	0.01	0.2 7	0.0 0	0.26	0.2 8	0.30	0.01	0.03	0.15
13 0	(Amy et al., 1998)	DCA	288	0.88 (0.95)	0.2 2	0.28	0.5 4	0.1 3	0.47	0.1 9	0.00	0.71	0.23	0.60
13 1	(Amy et al., 1998)	TCA	288	0.93 (0.97)	0.3 5	0.52	0.5 2	0.2 3	0.65	0.2 6	0.33	0.08	0.00	0.15
13 2	(Amy et al., 1998)	DBA	288	0.84	0.0 0	0.00	0.0 3	0.0 6	0.44	0.4 1	0.06	0.41	0.48	0.29
13 3	(Chen and Westerhoff, 2010)	DCA	207	0.84	0.1 5	0.36	0.3 0	0.0 4	0.15	0.4 8	0.86	0.10	0.15	0.44
13 4	(Chen and Westerhoff, 2010)	TCA	207	0.81	0.1 8	0.75	0.0 5	0.1 2	0.09	0.6 5	0.23	0.61	0.40	0.62
10 6	(McBean et al., 2008)	HAA	180	0.72	0.1 2	0.32	0.1 6	0.0 0	0.13	0.0 0	0.87	0.10	0.28	0.36
13 5	(Montgomer y Watson, 1993)	MCA			0.0 7	0.10	0.1 7	0.0 0	0.33	0.0 0	0.19	NA ^f	NA ^f	0.18
13 6	(Montgomer y Watson, 1993)	DCA			0.2 0	0.34	0.5 9	0.1 0	0.42	0.3 2	0.88	0.25	0.30	0.71
13 7	(Montgomer y Watson, 1993)	ТСА			0.0 3	0.05	0.5 7	0.0 0	0.02	0.3 6	0.03	0.17	0.09	0.16
13 8	(Montgomer y Watson, 1993)	MBA			0.0 2	0.00	0.0 2	0.1 2	0.02	0.0 3	0.07	0.28	0.01	0.23
13 9	(Montgomer y Watson, 1993)	DBA			0.2 9	0.07	0.4 2	0.0 5	0.16	0.5 1	0.01	0.10	0.14	0.83

#j	Source	Specie	Source	e Model ⁱ					Curren	t Study	/			External Non-
ē		5	# of	Validatio	٨	Non-	TiO	SW/	GAC	DS	5 kWb	80	160	(Eabbricino
ро			‴ 01 Samnle	n	data		1102	500	UAC	05	m ⁻³	kWh	$kWh m^{-}$	and Korshin
Σ			Sample		uutu	701						m ⁻³	3	2009)
10	(Sérodes et	HAA5	53	0.89	0.1	0.01	0.2	0.1	0.52	0.0	0.74	0.22	0.24	0.63
7	al., 2003)ª				3		1	2		1				
10	(Sérodes et	HAA5	55	0.8	0.0	0.29	0.1	0.1	0.91	0.0	0.65	0.30	0.43	0.92
8	al., 2003)ª				3		8	0		7				
10	(Sérodes et	HAA5	51	0.92	0.1	0.08	0.2	0.1	0.67	0.0	0.05	0.34	0.24	0.68
9	al., 2003)ª				2		4	0		0				
11	(USEPA,	HAA5	738	0.87	0.0	0.15	0.1	0.0	0.33	0.0	0.20	0.61	0.41	0.77
0	2001)		200	0.04	/	0.27	5	0	0.15	1	0.01	0.25	0.25	0.75
11	(USEPA, 2001)	паар	288	0.94	2	0.27	0.2	0.0	0.15	0.0	0.01	0.25	0.35	0.75
14	(USEDA	МСА	288	0.27	0.0	0.05	0.2	00	0.12	03	0.52	NΔf	NΔf	0.02
0	2001)	INCA	200	0.27	1	0.05	0.2	0.0	0.12	1	0.52	IN/A		0.02
14	(USEPA,	DCA	288	0.84	0.2	0.35	0.5	0.1	0.49	0.3	0.30	0.01	0.03	0.65
1	2001)	_			3		6	0		0				
14	(USEPA,	TCA	288	0.9	0.3	0.73	0.3	0.1	0.55	0.4	0.76	0.23	0.28	0.17
2	2001)				2		8	8		3				
14	(USEPA,	MBA	288	0.4	0.0	0.01	0.5	0.1	0.21	0.4	0.38	0.01	0.02	0.70
3	2001)				5		3	4		0				
14	(USEPA,	DBA	288	0.82	0.0	0.00	0.0	0.0	0.44	0.5	0.46	0.48	0.33	0.56
4			2007	0.50	1	0.17	2	1	0.10	1	0.02	0.27	0.40	0.76
2	(USEPA, 2001)	пааз	2007	0.59	0.0	0.17	0.1	0.0	0.10	0.0	0.02	0.27	0.40	0.76
14	(USEPA	МСА	2841	0.29	00	0.00	00	0.0	0.11	0.0	0.01	NΔf	NΔ ^f	0.15
5	2001)	110/1	2011	0.25	0	0.00	3	2	0.11	0	0.01	1473		0.15
14	(USEPA,	DCA	2881	0.54	0.2	0.26	0.6	0.1	0.52	0.2	0.16	0.12	0.00	0.65
6	2001)				2		8	4		2				
14	(USEPA,	TCA	2891	0.61	0.2	0.70	0.4	0.1	0.49	0.3	0.84	0.43	0.06	0.24
7	2001)				9		7	6		9				
14	(USEPA,	MBA	2892	0.19	0.0	0.05	0.0	0.0	0.01	0.1	0.13	0.20	0.17	0.11
8	2001)		2006	0.75	5	0.00	2	0	0.46	4	0.00	0.00	0.04	0.70
14	(USEPA,	DRA	2886	0.75	0.0	0.00	0.0	0.0	0.46	0.3	0.32	0.00	0.04	0.79
9	9 2001)			∠ 75	30	<u> </u>	25	25	25	15	15	15	14	
	Number of Samples			75	50		25	25	25	15	15	15	14	
	Minimum		0.14	0.0	0.00	0.0	0.0	0.00	0.0	0.00	0.00	0.00	0.00	
				U		U	U		U					

#j	Source	Specie	Source	e Model ⁱ					Curren	t Study	/			External Non-
-		S									[AUP
d€			# of	Validatio	All	Non-	TiO ₂	SW	GAC	DS	5 kWh	80	160	(Fabbricino
9			Sample	n	data	AOP					m ⁻³	kWh	kWh m⁻	and Korshin,
~			S									m⁻³	3	2009)
	Maximum			0.94	0.3	0.75	0.8	0.3	0.91	0.8	0.88	0.71	0.48	0.92
					5		1	6		1				
	Aver	age		0.67	0.1	0.18	0.2	0.0	0.33	0.2	0.30	0.24	0.17	0.39
					1		7	8		1				
	Standard	Deviatior	1	0.26	0.1	0.21	0.2	0.0	0.22	0.2	0.27	0.19	0.15	0.27
					1		2	9		0				

 $^{\rm b}$ R² not reported for validation data set

^c Independent external validation performed by Greiner et al. (1992)

^d No model number provided as HAA5 was calculated as the sum of 5 HAA

^e Reported by Zhu (1995)

^f NA - Not applicable; model does not return meaningful results. Possible reasons include all predictions are identical (so no R² value), input data missing (e.g., no UV₂₅₄), etc.

^g Reported by USEPA (2001)

^h Some models could not be calculated for some data sets due to lack of UV_{254} data. For this subset of data/models, n = 181 for the AOP data set, n = 201 for non-AOP data, n = 143 for TiO₂ data set, and n = 38 for all other types of AOP data

ⁱ Values reported are internal modeling values for # of sources and validation, while values shown in parentheses are for external validation of model

^j Model # refers to number assigned in Table S4

214

215 **Table S10 (Continued)** – Average correlation coefficients (R² values) for pairs of predictive HAA models versus measured HAA 216 datasets

_	Source	Specie			External AOP D	ata			Grouped	Data	
bde #j		S	(Lamsal	(Liu et	(Sarathy and	(Kent et	(Chin and	All	All Non-	All	Othe
Σ			et al.,	al.,	Mohseni,	al., 2011)	Bérubé,	AOPs	AOP	TiO ₂	r
			2011)	2008b)	2010)		2005)				AOPs
	(Amy et al., 1998)	HAA5 ^d	0.84	0.57	0.02	0.68	0.66	0.70	0.80	0.79	0.21
113	(Amy et al., 1998)	MCA				0.01		0.21	0.09	0.21	
114	(Amy et al., 1998)	MBA				NA ^f		0.00	0.33	0.00	
115	(Amy et al., 1998)	DCA				0.69		0.84	0.72	0.84	
116	(Amy et al., 1998)	TCA				0.68		0.84	0.61	0.84	

_	Source	Specie			External AOP D	Data			Grouped	Data	
Mode # ^j		S	(Lamsal et al., 2011)	(Liu et al., 2008b)	(Sarathy and Mohseni, 2010)	(Kent et al., 2011)	(Chin and Bérubé, 2005)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
117	(Amy et al., 1998)	DBA				NA ^f		0.02	0.52	0.02	
	(Amy et al., 1998)	HAA5 ^d	0.84	0.85	0.02	0.68	0.63	0.58	0.56	0.65	0.44
118	(Amy et al., 1998)	MCA				0.01		0.12	0.00	0.12	
119	(Amy et al., 1998)	MBA				NA ^f		0.33	0.35	0.33	
120	(Amy et al., 1998)	DCA				0.69		0.82	0.73	0.82	
121	(Amy et al., 1998)	TCA				0.68		0.85	0.67	0.85	
122	(Amy et al., 1998)	DBA				NA ^f		0.02	0.63	0.02	
	(Amy et al., 1998)	HAA5 ^d	0.84	0.87	0.02	0.68	0.75	0.58	0.66	0.77	0.38
123	(Amy et al., 1998)	MCA				0.01		0.12	0.01	0.12	
124	(Amy et al., 1998)	MBA				NA ^f		0.21	0.34	0.21	
125	(Amy et al., 1998)	DCA				0.69		0.83	0.76	0.83	
126	(Amy et al., 1998)	TCA				0.68		0.86	0.70	0.86	
127	(Amy et al., 1998)	DBA				NA ^f		0.02	0.62	0.02	
	(Amy et al., 1998)	HAA5 ^d	0.84	0.86	0.02	0.68	0.66	0.59	0.62	0.72	0.42
128	(Amy et al., 1998)	MCA				0.01		0.12	0.00	0.12	
129	(Amy et al., 1998)	MBA				NA ^f		0.27	0.35	0.27	
130	(Amy et al., 1998)	DCA				0.69		0.82	0.74	0.82	
131	(Amy et al., 1998)	TCA				0.68		0.85	0.69	0.85	
132	(Amy et al., 1998)	DBA				NA ^f		0.02	0.62	0.02	
133	(Chen and Westerhoff, 2010)	DCA				NA ^f		0.30	0.77	0.30	
134	(Chen and Westerhoff, 2010)	TCA				NA ^f		0.05	0.90	0.05	
106	(McBean et al., 2008)	HAA	0.84	0.95	0.02	0.69	0.63	0.51	0.50	0.81	0.00

_	Source	Specie			External AOP D			Grouped	Data		
Mode # ^j		S	(Lamsal et al., 2011)	(Liu et al., 2008b)	(Sarathy and Mohseni, 2010)	(Kent et al., 2011)	(Chin and Bérubé, 2005)	All AOPs	All Non- AOP	All TiO ₂	Othe r AOPs
135	(Montgomery Watson, 1993)	MCA				0.01		0.09	0.00	0.09	
136	(Montgomery Watson, 1993)	DCA				NA ^f		0.59	0.85	0.59	
137	(Montgomery Watson, 1993)	TCA				NA ^f		0.57	0.03	0.57	
138	(Montgomery Watson, 1993)	MBA				NA ^f		0.02	0.11	0.02	
139	(Montgomery Watson, 1993)	DBA				NA ^f		0.42	0.20	0.42	
107	(Sérodes et al., 2003)ª	HAA5	0.84	0.95	0.02	0.69	0.62	0.64	0.56	0.69	0.21
108	(Sérodes et al., 2003)ª	HAA5	0.84	0.78	0.01	0.68	0.74	0.32	0.42	0.90	0.09
109	(Sérodes et al., 2003)ª	HAA5	0.84	0.94	0.02	0.69	0.65	0.54	0.57	0.77	0.16
110	(USEPA, 2001)	HAA5	0.84	0.94	0.02	0.68	0.65	0.68	0.64	0.80	0.20
111	(USEPA, 2001)	HAA5	0.79	0.96	0.40	NA ^f	0.93	0.87	0.70	0.99	0.60
140	(USEPA, 2001)	MCA				NA ^f		0.20	0.01	0.20	
141	(USEPA, 2001)	DCA				NA ^f		0.56	0.79	0.56	
142	(USEPA, 2001)	TCA				NA ^f		0.38	0.73	0.38	
143	(USEPA, 2001)	MBA				NA ^f		0.53	0.36	0.53	
144	(USEPA, 2001)	DBA				NA ^f		0.02	0.64	0.02	
112	(USEPA, 2001)	HAA5	0.75	0.89	0.41	NA ^f	0.91	0.70	0.74	0.90	0.49
145	(USEPA, 2001)	MCA				NA ^f		0.03	0.01	0.03	
146	(USEPA, 2001)	DCA				NA ^f		0.68	0.84	0.68	
147	(USEPA, 2001)	TCA				NA ^f		0.47	0.89	0.47	
148	(USEPA, 2001)	MBA				NA ^f		0.02	0.03	0.02	
149	(USEPA, 2001)	DBA				NA ^f		0.02	0.81	0.02	
	Number of Sample	S	6	5	9	4	13	77 ^h	49	52 ^h	25

_	Source	Specie			External AOP D	Data			Grouped	Data	
#jde		S	(Lamsal	(Liu et	(Sarathy and	(Kent et	(Chin and	All	All Non-	All	Othe
Σ			et al.,	al.,	Mohseni,	al., 2011)	Bérubé,	AOPs	AOP	TiO ₂	r
			2011)	2008b)	2010)		2005)				AOPs
	Minimum		0.62	0.57	0.01	0.01	0.62	0.00	0.00	0.00	0.00
	Maximum		0.93	0.96	0.41	0.69	0.93	0.87	0.90	0.99	0.60
	Average		0.71	0.87	0.09	0.53	0.71	0.41	0.51	0.46	0.29
	Standard Deviation			0.11	0.16	0.29	0.11	0.31	0.29	0.34	0.19

^b R² not reported for validation data set

^c Independent external validation performed by Greiner et al. (1992)

^d No model number provided as HAA5 was calculated as the sum of 5 HAA

^e Reported by Zhu (1995)

^f NA - Not applicable; model does not return meaningful results. Possible reasons include all predictions are identical (so no R² value), input data missing (e.g., no UV₂₅₄), etc.

^g Reported by USEPA (2001)

^h Some models could not be calculated for some data sets due to lack of UV_{254} data. For this subset of data/models, n = 181 for the AOP data set, n = 201 for non-AOP data, n = 143 for TiO₂ data set, and n = 38 for all other types of AOP data

ⁱ Values reported are internal modeling values for # of sources and validation, while values shown in parentheses are for external validation of model

^j Model # refers to number assigned in Table S4

219 8. Statistical Analyses on Grouped Modeling Results

220 Coefficients of determination (R^2) were calculated for each model/dataset pairing, 221 although not all models yielded meaningful predictions for all datasets as a result of missing 222 input data, inability to calculate R^2 due to model returning identical values for all scenarios, 223 etc. The R^2 values were grouped in accordance with level of photocatalytic/AOP treatment, 224 type of source water, etc. in order to further explore model accuracy (i.e., the relationship 225 between precursor measures and DBP formation) using statistical analyses, all of which 226 were performed at a significance level of a = 0.05.

227 8.1 Predictive Model Accuracy as a Function of Extent of Photocatalytic 228 Treatment

The D'Agostino-Pearson omnibus test revealed that at least one THM and HAA dataset did not follow a normal distribution, as shown in Table S11, with normally distributed datasets shaded in gray.

Table S11 – P values from the D'Agostino-Pearson omnibus test for normality for R² values

233 of DBP model predictions versus internal data grouped by extent of photocatalytic treatment

Data Group	THMs	HAAs
Dark Adsorption	<0.0001	0.0016
5 kWh m ⁻³	<0.0001	0.0700
80 kWh m ⁻³	0.0004	0.2516
160 kWh m ⁻³	<0.0001	0.0138
n =	108 - 109	41 - 48

234 235

The Kruskal-Wallis nonparametric test was used to test for differences in group

means. The tests revealed significant differences in the accuracy (R^2) of DBP model predictions among the levels of photocatalytic treatment tested in this study (P < 0.0001 for THMs and P = 0.0256 for HAAs). Dunn's post hoc test was used to ascertain which groups were significantly different. The results are shown in Tables S12 and S13, with significantly different pairings shaded in gray.

Table S12 – Dunn's post-hoc analysis for THM model accuracy using internal data grouped
 by extent of photocatalytic treatment

Data Group	Dark Adsorption	5 kWh m⁻³	80 kWh m ⁻³	160 kWh m ⁻³
Dark Adsorption				
5 kWh m⁻³	<0.0001			
80 kWh m ⁻³	< 0.0001	< 0.0001		
160 kWh m ⁻³	0.0267	< 0.0001	0.1716	
n per aroup = 108 - 10)9			

243

244 **Table S13** – Dunn's post-hoc analysis for HAA model accuracy using internal data grouped

245 by extent of photocatalytic treatment

Data Group	Dark Adsorption	5 kWh m⁻³	80 kWh m ⁻³	160 kWh m ⁻³
Dark Adsorption				
5 kWh m ⁻³	0.0411			
80 kWh m ⁻³	0.2455	>0.9999		
160 kWh m ⁻³	>0.9999	0.2632	0.9509	
n per group = $41 - 48$				

246

247 **8.2** *Predictive Model Accuracy as a Function of Type of Source Water*

248 The D'Agostino-Pearson omnibus test revealed that at least one THM and HAA

249 dataset did not follow a normal distribution, as shown in Table S14, with normally

250 distributed datasets shaded in gray.
251 **Table S14** – P values from the D'Agostino-Pearson omnibus test for normality for R^2 values 252 of THM and HAA model predictions versus internal data grouped by type of source water

Data Group	THMs	HAAs	
Untreated	< 0.0001	< 0.0001	
Coagulated	<0.0001	а	
Settled	<0.0001	0.0034	
Filtered	< 0.0001	0.1910	
Finished	<0.0001	0.0728	
Synthetic	< 0.0001	а	
n =	104 - 591	48 - 114	
$^{\circ}$ No HAA measurements for this type of source water reported in the datasets analyzed			

253

254 The Kruskal-Wallis nonparametric test was used to test for differences in group 255 means. The tests revealed significant differences in the accuracy (R^2) of DBP model predictions among the levels of photocatalytic treatment tested in this study (P < 0.0001 for 256 257 THMs and HAAs). Dunn's post hoc test was used to ascertain which groups were 258 significantly different. The results are shown in Tables S15 and S16, with significantly 259 different pairings shaded in gray. 260

- **Table S15** Dunn's post-hoc analysis for THM model accuracy for aggregated internal and
- 261 external data grouped by source water type

Data Group	Untreated	Coagulated	Settled	Filtered	Finished	Synthetic
Untreated						
Coagulated	0.0001					
Settled	< 0.0001	<0.0001				
Filtered	< 0.0001	<0.0001	>0.9999			
Finished	< 0.0001	0.0006	>0.9999	>0.9999		
Synthetic	< 0.0001	<0.0001	>0.9999	>0.9999	>0.9999	
n = 104 - 591						

262

263 **Table S16** – Dunn's post-hoc analysis for HAA model predictions versus aggregated internal 264 and external data grouped by source water type

Data Group	Untreated	Settled	Filtered	Finished
Untreated				
Settled	< 0.0001			
Filtered	0.0317	< 0.0001		
Finished	< 0.0001	0.0512	0.1842	
n = 48 - 114				

265

266 8.3 DBP Model Accuracy for TiO₂ Photocatalysis, Other AOP, and Non-AOP 267 Treatment

268 The D'Agostino-Pearson omnibus test revealed that at least one THM and HAA 269 dataset did not follow a normal distribution, as shown in Table S17, with normally 270 distributed datasets shaded in gray.

- 271
- 272 **Table S17** – P values from the D'Agostino-Pearson omnibus test for normality for R² values 273 of THM and HAA model predictions versus internal and external data grouped by type of
- 274 AOP treatment

Data Group	THMs	HAAs
TiO ₂ Photocatalysis	<0.0001	<0.0001

Non-TiO ₂ AOP	<0.0001	<0.0001
Non-AOP	<0.0001	<0.0001
n =	380 - 1130	92 - 403

275

276 The Kruskal-Wallis nonparametric test was used to test for differences in group 277 means. The tests revealed significant differences in the accuracy (R^2) of DBP model 278 predictions among the levels of photocatalytic treatment tested in this study (P < 0.0001 for 279 HAAs). However, no significant difference was observed for THMs (P = 0.0568). Dunn's 280 post hoc test was used to ascertain which HAA groups were significantly different. The results are shown in Table S18, with significantly different pairings shaded in gray. 281 282 **Table S18** – Dunn's post-hoc analysis for HAA model accuracy of data grouped by type of 283 AOP treatment

Data Group	TiO ₂ Photocatalysis	Non-TiO ₂ AOPs	Non-AOPs
TiO ₂ Photocatalysis			
Non-TiO ₂ AOPs	0.0001		
Non-AOPs	0.09204	0.1424	
n = 92 - 403			

284

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