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PHYSICOCHEMICAL CHARACTERISTICS OF FERTILIZER INDUSTRY EFFLUENT AND ITS TOXICOLOGICAL IMPACT ON THE ACTIVITY OF ACETYLCHOLINESTERASE (AChE) IN FRESHWATER TELEOSTS Heteropneustes fossilis AND Labeo rohita

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ABSTRACT

Industrial revolution is a good indicator of economic development of a country; however, it can be a threat to the flora and fauna if the untreated effluent of an industry is discharged. The present study is aimed to assess the comparative toxicological impacts of treated and untreated industrial effluents on acetylcholinesterase (AChE) activity in Heteropneustes fossilis and Labeo rohita, the most common edible fishes having diverse characters which include differences in morphology, habitat, food and feeding, etc. The physico-chemical parameters such as pH, electrical conductivity (EC), alkalinity, hardness, dissolved oxygen (DO), phosphate, sulphate, nitrate, free ammonia, chloride, zinc, iron, chromium and potassium of both untreated and treated effluent from the fertilizer industry were also analyzed as these parameters were not in range as per ISO guidelines. The LC_{so} value for untreated effluent was 2.34% (v/v) and 0.80% (v/v) for 96 h in H. fossilis and L. rohita, respectively, while no mortality was recorded in the treated effluent. The AChE activity in both fish species was found to decline in metabolically responsive organs like brain, muscle and gills through exposure to sub-lethal concentrations (1/15th, 1/10th and 1/5th of LC₅₀ value) of the untreated effluent for 96 h. Further studies on biochemical and molecular aspects may reveal the mechanism of their action.

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INTRODUCTION

Fertilizer industries use a variety of chemical compounds due to which the effluents are complex in nature. The effluents are rich in various organic compounds, heavy metals, dissolved and suspended solids. The effluents adversely affect the metabolism of fish inhabiting the aquatic bodies causing mortality and impairment in physiological activities (Bobmanuel et al., 2006; Kaur et al., 2010; Roopadevi and Somasekhar, 2012). On the basis of pollution load, fertilizer producing industries have been categorized under red list industries in India by the Ministry of Environment, Forest and Climate Change (2016).

In the aquatic ecosystem, fish play crucial role in the monitoring of water contamination as they respond in most sensitivity to variations in aquatic pollution (Naigaga et al., 2011). Stinging catfish H. fossilis (name "stinging" due to its venomous nature) is a native fish species of India and is economically important due to its nutritive and medicinal value (Talwar and Jhingran, 1991; Tiwari et al., 2017). Its body is elongated and depressed towards head region in order to survive in drastic conditions (e.g. drought, enemy and carnivorous activity) which make them hardy in nature (Jawad, 2015). It is carnivorous in nature, found at benthic region (in mud) during day time and exhibits nocturnal behaviour, i.e. it is much active during night as it frequently comes up for air gulping (Pati et al., 1998). The other species, Labeo rohita, is one of the major and most widespread carp species in India. It is suitable for experimental purposes due to its easy capturing and maintenance in the laboratory and is also economically important. L. rohita is found throughout the year in lakes and rivers of India, has spindle shaped body for floatation purpose and is basically planktivorous (Das and Moitra, 1955).

LC₅₀ is the most reliable and preliminary test for toxicity evaluation in fish (Shenkani and Das, 2014; Shreelekshmy et al., 2016). Acetylcholinesterase (AChE, EC: 3.1.1.7), an acetylcholine (ACh) hydrolyzing enzyme (Gaitonde et al., 2006), plays an important role in neurotransmission in both the vertebrates and invertebrates (Valbonesi et al., 2003). Inhibition of AChE by pesticides, heavy metals or industrial effluents results in ACh accumulation in synaptic cleft, causing neuromuscular paralysis and asphyxiation, resulting finally in mortality (Xuereb et al., 2009; Rakhi et al., 2013). Recent studies suggest that AChE is a very sensitive indicator of stress and hence can be useful for biomonitoring and assessment of the level of contamination (Richetti et al., 2011).

In an era of industrialization, developing countries face a big problem in proper industrial waste treatment prior to its disposal due to technical limitations and heavy economic expenses behind the treatment process (Majumdar et al., 2007). Therefore, an effective management to prevent or at least minimize waste generation by developing

technologies for waste recycling and reuse are the need of the hour (Jadhav and Hocheng, 2012). Many ecological incidents occurred in the past across the planet, and there are still high probabilities that untreated industrial effluent may reach the nearby aquatic bodies and adversely affect the organisms living therein. Hence, in the present study an attempt has been made to determine the physicochemical characteristics of both treated and untreated effluents, their acute toxicity and their impact on AChE activity in fishes.

MATERIALS AND METHODS

Animal maintenance and chemicals

Fishes, Heteropneustes fossilis and Labeo rohita, were procured from the local fish market (Allahabad, Uttar-Pradesh, India) and treated with potassium permanganate solution (KMnO₄, 0.5% w/v) for one min to remove any kind of subcutaneous adherent(s). The fishes were acclimatized in glass aquarium (capacity 50 L) with dechlorinated aerated tap water (pH 6.8 ± 0.2, dissolved oxygen 6.8 ± 0.5 mg L⁻¹ and total hardness 111.4 ± 4.0 mg L⁻¹). Before exposure, quality of water was assessed and tested according to the American Public Health Association Guidelines (APHA, 1985). For fifteen days, fishes were acclimatized at room temperature (26.0±1.2°C) and were fed commercially available food pellets (Tokyu, India) ad libitum. All of the chemicals used in the present study were of analytical grade and standard quality.

Experimental design

Physicochemical analysis

The untreated and treated samples were collected from a fertilizer production unit situated in Indian Farmers Fertiliser Cooperative Ltd. (IFFCO; Phulpur, Allahabad, India) and brought to the laboratory for physicochemical analysis followed by experimentation. Standard methods were used for the analysis of collected samples. pH and electrical conductivity (EC) were determined electrometrically. The physicochemical parameters such as alkalinity, hardness, dissolved oxygen, free ammonia and chloride were determined by titration method. Determinations of phosphate, sulphate and nitrate concentrations were done by spectrophotometric methods. Zinc, iron, chromium and potassium concentrations were determined by using atomic absorption spectrophotometer (Perkin Elmer Analyst 200, USA).

Determination of acute toxicity (LC₅₀)

Feeding was stopped twenty four hours before experimentation. Twelve individuals of H. fossilis of average length of 13.0 \pm 2.0 cm and average weight of 21.0 \pm 2.25 g were placed in each of the aquaria containing 12 L of water (n = 12/glass aquarium). Untreated and treated effluent samples were added as

per following concentrations - control, 1.95, 2.10, 2.25, 2.40, 2.55, 2.70% (v/v) and control, 25, 50, 75, 100% (v/v), respectively. Similarly, twelve individuals of L. rohita of average length of 12.0 ± 1.69 cm and average weight of 43.0 ± 1.43 g were placed in each of the aquaria containing 12 L of water (n = 12/glass aquarium). Untreated and treated effluent samples were added as per following concentrations - control, 0.75, 0.85, 0.90, 0.95, 1.00, 1.05% (v/v) and control, 20, 40, 60, 80 and 100% (v/v), respectively. At an interval of twenty four hours, the water of aquaria was changed and different concentrations of fresh effluents were added. Fish kept in effluent-free medium served as the control. The test was performed for 96 h treatment period and dead fish were removed as the test proceeded. The number of dead fish per group was recorded against the time of their mortality in a tabular form, as specified by Sprague (1969). The 96 h LC₅₀ value of untreated and treated effluents was calculated using arithmetic method of Kärber (1931), modified by Dede and Kaglo (2001).

Subacute toxicity assays

Preparation of tissue homogenate

The fishes were sacrificed by using mild anaesthesia (Trichloromethane) after exposure to three sublethal concentrations of effluents (1/15th of LC₅₀, 1/10th of LC₅₀ and 1/10th of LC₅₀) for 96 h. Key organs (brain, muscle and gills) of exposed fishes along with the control were surgically removed and thoroughly rinsed in 0.69% cold saline, at 4-6°C and blotted dry. The tissues were weighed and homogenized (10%, w/v) in 50 mM sodium phosphate buffer (pH 8.0), containing 0.1% Triton X-100, using Potter-Elvehjam homogenizer fitted with a Tefloncoated pestle under ice cold conditions. For 30 min, the homogenates were kept in cold with intermittent stirring and centrifuged at 4°C for 30 min at 10,000 g in a refrigerated centrifuge (Model-3K30 Sigma, St. Louis, USA). The corresponding supernatants were used afresh for determination of protein contents by AChE assay.

Assay of Acetylcholinesterase (AChE)

AChE activity was assayed according to Ellman's method (Ellman et al., 1961). The reaction mixture (3 mL) contained 1.5 mL of 100 mM sodium phosphate buffer (pH 8.0), 0.3 mL of 5 mM DTNB [5,5'-dithiobis-(2-nitrobenzoic acid), prepared in 10 mM sodium phosphate buffer, pH 7.5 containing 15 mg sodium bicarbonate added per 10 mL of solution], 0.3 mL of 5 mM Acetylthiocholine iodide (ATI), 0.1 mL of supernatant and 0.8 mL of distilled water. In a UV-visible double beam spectrophotometer (Shimadzu, Model: UV 1800, SL - 02480, Japan), the increase in absorbance was monitored at 412 nm and 28°C for 3 min with quartz cuvettes against distilled water as blank. For each tissue, measurements were made in triplicate. Simultaneously, two blanks were also used. One (enzyme blank) contained phosphate buffer, DTNB and ATI but not

the enzyme to determine the spontaneous hydrolysis of ATI, and the second (substrate blank) contained phosphate buffer, DTNB and enzyme protein but no substrate to correct any non-AChE-dependent formation of thionitrobenzoic acid (TNB). One unit of AChE activity was expressed as nanomoles of hydrolyzed substrate/min/mg protein under experimental conditions.

Estimation of protein

The protein content in cell-free extract of different fish tissues was determined by the method described by Lowry et al. (1951) using bovine serum albumin (BSA) as the standard.

Statistical analysis

The data of physicochemical parameters are represented as mean ± SEM. The lethal concentration for 96 h was calculated by Kärber method (1931) as adopted by Dede and Kaglo (2001), and confidence interval calculated by SPSS 16.0. The data for AChE activity and protein content were represented as mean ± SEM and were analyzed by the one-way analysis of variance (one-way ANOVA) using SPSS software (16.0, 2008). To note the variations between control versus experimental groups and among the experimental groups (1/15th, 1/10th and 1/5th exposed groups), Duncan's multiple range post hoc test was applied. The results were considered to be significant at p < 0.05 and p < 0.01 levels (confidence levels 95% and 99%, respectively). All experiments were carried out in triplicate and data were statistically analyzed according to Brunning and Knitz (1977).

RESULTS AND DISCUSSION

Physico-chemical characteristics

physicochemical characteristics Various like pH, conductivity, alkalinity, hardness, dissolved oxygen, concentration of phosphate, sulphate, nitrate, chloride, zinc, iron, potassium, chromium and free ammonia of untreated and treated fertilizer industry effluents were studied (Table 1). Some of the parameters like pH, conductivity, chloride content, free ammonia and chromium have been found quite high, while dissolved oxygen, concentration of nitrate, zinc and iron were even below the permissible limit of the Bureau of Indian Standards (BIS 10500:1991, 2003). However, others like alkalinity, hardness and the concentration of sulphate, phosphate and potassium were in moderate range.

The pH of untreated and treated effluent samples was 9.99 ± 0.17 and 7.2 ± 0.048 , respectively, and similarly alkalinity was 627.0 ± 0.54 and 192.0 ± 0.49 mg L⁻¹, respectively. The higher pH (showing alkalinity) of the untreated sample is probably due to the presence of ammonia which may result in bio-concentration of metals in aquatic organisms and cause adverse effect on their health (Campbell and Stokes, 1985; Karthikeyan et al., 2004; Durrani et al., 2007). Conductivity of water

Table 1. Comparative account of physico-chemical parameters of the untreated and treated effluent sample (Mean ± SEM) of the fertilizer industry and their permissible range (BIS 10500: 1991, 2003)

S. No	Parameters	Permissible Range	Untreated effluent	Treated effluent	Method employed
1.	рН	6.5 - 8.2	9.99 ±0.17	7.2 ± 0.048	Electrometric
2.	Conductivity	200 (μScm ⁻¹)	9463 ±1.82	217 ± 1.08	Electrometric
3.	Alkalinity	200 (mg L-1)	627.0 ± 0.54	192 ±0.49	Titrimetric
4.	Hardness	300-600 (mg L ⁻¹)	672.0 ± 0.09	302 ± 0.84	Titrimetric
5.	Dissolved Oxygen	5 (mg L ⁻¹)	3.33 ± 0.09	4.98 ± 0.23	Titrimetric
6.	Phosphate	5 (mg L ⁻¹)	7.32 ± 0.50	5.02 ± 0.11	Spectroscopic
7.	Sulphate	200 - 400 (mg L ⁻¹)	865.0 ± 1.58	331.8 ± 0.28	Spectroscopic
8.	Nitrate	45 – 100 (mg L ⁻¹)	10.89 ± 0.57	63.22 ± 0.88	Spectroscopic
9.	Free NH ₃	0-5 (mg L ⁻¹)	403.6 ± 0.47	4.7 ± 0.29	Titrimetric
10.	Chloride	250-1000 (mg L ⁻¹)	1667.2 ± 0.56	857.2 ±0.96	Titrimetric
11.	Zinc	<2 (mg L ⁻¹)	0.132 ± 0.01	1.97 ± 0.05	Atomic Absorption Spectrophotometric
12.	Iron	1.0 (mg L-1)	0.079 ± 0.01	0.92 ± 0.03	Atomic Absorption Spectrophotometric
13.	Potassium	<20 (mg L ⁻¹)	26.67 ± 0.72	19.20 ± 0.82	Atomic Absorption Spectrophotometric
14.	Chromium	0.05 (mg L ⁻¹)	1.274 ± 0.09	0.06 ± 0.01	Atomic Absorption Spectrophotometric

shows the ability of solvent to conduct electricity which depends on the presence of ions and their concentrations (Abdullah and Musta, 1999). Conductivity is significantly correlated to the pH, alkalinity, hardness, concentrations of phosphate, sulphate, free ammonia, chloride, ion mobility and temperature, and is an indicator of salinity or total salt content of discharged effluent (Kumar and Sinha, 2010). A quite higher level of conductivity in the untreated effluent than in the treated one is probably due to the presence of a higher level of nitrate, free ammonia, chromium content, etc. (Table 1). Hardness is a measure of capacity of water to precipitate soap, contributed by presence of calcium, magnesium and presence of minerals. The hardness of both the untreated and treated samples was almost close to the permissible range. Dissolved oxygen in the untreated effluent sample was 3.33 ± 0.09 mg L⁻¹, which is much lower than in the treated one (as per ISO-guidelines). A dissolved oxygen level > 5 mg L⁻¹ is necessary to support adequately the metabolism and health of fishes (Bhatnagar and Devi, 2013; Nirgude et al., 2013). Thus the observed lower dissolved oxygen recorded in the untreated effluent seems inadequate for the maintenance of good health of fish. Effluent from the fertilizer industry contains a higher level of phosphate and sulphate, therefore higher concentrations of these in the untreated effluent sample were recorded. Increased phosphate results in eutrophication of aquatic ecosystem, which might be a cause for mortality of fish in aquatic system (Grubb et al., 2000; Horrigan et al., 2002). A higher level of chloride content observed in the untreated sample may be due to the usage of hydrochloric acid, hypochloric acid, chlorine gas, etc., as raw materials during various processes. A higher free ammonia and a lower nitrate content in the untreated sample indicates the presence of nitrogen in bound form (NH,OH) which decomposes into free ammonia and water, resulting in increase in ammonia concentration in comparison to the treated effluent (Table 1, Fig. 1). The zinc and iron levels in the untreated effluent sample are close to the permissible range. However, level of potassium is higher than in the treated sample, which may cause physiological imbalance in the fish (Sah et al., 2002). Chromium level is much higher in the untreated effluent sample than in the treated sample. It is non-biodegradable and therefore has potential to cause toxicity and biomagnifications in aquatic organisms (Wicklund-Glynn and Olsson, 1991; Praveena et al., 2013).

Toxicity assessment

Acute toxicity assessment provides a comparative lethal value of tested effluents in both catfish and carps. The obtained values show susceptibility of fish to particular contaminants and reflect their survival potential. Accordingly, catfish were more tolerant than carps due to their physiological and ecological condition (Abedi et al., 2012). LC_{50} for 96 h for the untreated effluent in *H. fossilis* and *L. rohita* was determined as 2.35 and 0.85 % (v/v) (confidence interval: 1.94 - 2.47 and 0.786 – 0.958),

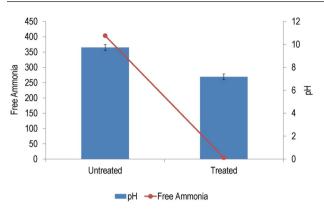


Fig 1. pH value and concentration of free ammonia in the untreated and treated effluent sample of the fertilizer industry. Data represent mean \pm S. E. M.

respectively (Table 2&3). A similar finding was reported by Yadav et al. (2007) and El-Samara et al. (1986) in *Channa striatus* (70% v/v) and *Mugil cephalus* (0.74 and 0.76% v/v exposed to north and south plant fertilizer effluents, respectively). However, fishes exposed to dechlorinated tap-water and the treated effluent sample were observed to be healthy and normal.

However, reports regarding lethal toxicity in fishes exposed to fertilizer industry effluent are very scanty. Only a few are currently available but for varying exposure duration (shorter than 96 h) and for different fish species. The LC₅₀ values at 24, 48 and 72 h were 72.05, 30.81 and 15.26% (v/v), respectively, in *Oreochromis niloticus*, and 26.18, 10.32 and 9.84% (v/v), respectively, for hybrid catfish (*Heterobranchus bidorsalis* $\[\bigcirc \]$ X *Clarias gariepinus* $\[\bigcirc \]$) (Bobmanuel et al., 2006). The LC₅₀ values of different industrial effluents for 96 h exposure of different fish species have been reported. These values were 20, 6 and 22% (v/v) in *L. rohita* for tannery, electroplating

and textile effluents, respectively (Muley et al., 2007); 4.21 and 2.5% (v/v) in *H. fossilis* and *Lebistus reticulates*, respectively, exposed to galvanizing and raw bulk drug industrial effluents (Majumdar et al., 2007; Deshpande and Satyanarayan, 2011); 51.20 mL L⁻¹ in *Oreochromis niloticus* exposed to pharmaceutical effluent and 1.67 mL L⁻¹ in *Poecilia reticulate* exposed to chrome plating industrial effluents (Obasi et al., 2014; Ahila Angelin et al., 2015); 0.259% (v/v) in *Lebistus reticulate* exposed to the untreated pesticide producing industrial effluent (Chavan et al., 2016). These results suggest that the level of toxicity was dependent on nature of effluents discharged by the industries threatening the inhabiting species.

Acetylcholinesterase activity

Biological markers are sensitive and cost-effective tools for identifying hazards of environmental contamination. As AChE activity has been shown to be a potential biochemical indicator of toxic stress in different fish species; an attempt has been made to study the effect of fertilizer industrial effluent on its activity in different tissues of *H. fossilis* and *L. rohita*. After exposure to sublethal concentrations (1/15th, 1/10th and 1/5th of LC_{50}) of the untreated effluent for 96 h, AChE activity was found to decrease in a concentration-dependent manner in both fish species. The order of decrease in AChE activity was brain > gills > muscles for all sublethal concentration of the untreated effluent studied, however, no significant change was observed in the treated effluent exposed fishes (Fig 2 and 3).

AChE is a transmembrane protein present in cell membranes of vertebrates and invertebrates (Jebali et al., 2013) which hydrolyzes the neurotransmitter acetylcholine into choline and acetic acid in synapses as well as neuromuscular junctions. However, only scanty information is available related to the impact of

Table 2. LC_{50} determination value of the untreated fertilizer industry effluent in *H. fossilis* for 96 h based on arithmetic method of Kärber (1931) as adopted by Dede and Kaglo (2001)

Concentration %(v/v)	Concentration difference	Number of alive fish	Number of dead fish	Mean mortality	Mean mortality x concentration difference
Control	0	12	0	0	0
1.95	0.15	12	0	0	0
2.10	0.15	10	2	1	0.15
2.25	0.15	08	4	3.0	0.45
2.40	0.15	05	7	5.5	0.825
2.55	0.15	03	9	08	1.20
2.70	0.15	0	12	10.5	1.575
					∑ = 4.2

Summation indicates sum (Mean mortality x Concentration difference) LC_{50} for LC_{100} - Σ (Mean mortality x Concentration difference)/ number of organisms per group untreated effluent 2.70 - 4.2/12 = 2.70 - 0.35 = 2.35 % (v/v)

Table 3. LC_{50} determination value of the untreated fertilizer industry effluent in *L. rohita* for 96 h based on arithmetic method of Kärber (1931) as adopted by Dede and Kaglo (2001)

Concentration %(v/v)	Concentration difference	Number of alive fish	Number of dead fish	Mean mortality	Mean mortality x concentration difference
Control	0	12	0	0	0
0.75	0.75	10	2	1	0.75
0.85	0.10	08	4	3	0.30
0.90	0.10	05	7	5.50	0.55
0.95	0.05	04	8	7.50	0.375
1.00	0.05	2	10	9	0.45
1.05	0.05	0	12	11	0.55
					∑ = 2.975

Summation indicates sum (Mean mortality x Concentration difference) LC_{50} for LC_{100} - Σ (Mean mortality x Concentration difference)/ number of organisms per group untreated effluent 1.05 - 2.975/12 = 1.05 - 0.248 = 0.80% (v/v)

fertilizer industry effluents on this enzyme. The waste water released from fertilizer industry contains various contaminants which may act synergistically and if accumulated by inhibiting AChE activity may pose serious threat to the inhabiting species (Matthiessen et al., 1993; Bocquene et al., 1995; Forget et al., 1999; Larry et al., 2006). Hence, measurement of fish AChE activity has been considered as a suitable indicator for biomonitoring aquatic/industrial pollution (Jebali et al., 2013). The results further indicate that this effluent may act on AChE as a key target for its action. The negative impact caused by the untreated effluent, such as decrease in AChE activity and protein content in fish tissues (Fig 2 and 3), could be attributed to the presence of certain heavy metals and ammonia in it which may bind irreversibly on active site (Ravindra, 1999). The physical factors of fertilizer industry effluent such as dissolved oxygen, pH, temperature and salinity may also influence certain biochemical indices in both fishes by altering multiple pathways (Bobmanuel et al., 2006; Kumari et al., 2010). In aquatic ecosystems, fertilizer industry effluent containing low concentrations (Sultana and Rao, 1998) of zinc may cause tissue damage by reacting with proteins and could also affect the respiratory efficiency, ion exchange and osmoregulatory function of the organs (Lloyd, 1992). The heavy metals present in the effluent in association with other chemicals may distort the cell organelles and alter the activity of several enzymes (Jan et al., 2015). This may disturb the physiological status of the exposed fishes, H. fossilis and L. rohita. In conclusion, the untreated fertilizer industry effluent has a deleterious impact in both fishes at high sublethal concentrations by inducing significant alterations in the activity of AChE in various fish tissues. The results indicated that the contents present in this effluent specifically targeted neurotransmission system individually and/or synergistically. Even though

the fertilizer industry effluent at high concentrations markedly affected the fish organs tested, the magnitude of response varied from the one organ to the other. The decline in AChE activity could be considered as a useful tool for the assessment of industrial effluent toxicity, as it has been the case for other xenobiotic compounds such as organophosphate pesticides and heavy metals (Yadav et al., 2009; Rajkumar and Milton, 2011; Al-Ghais, 2013; Ramesh and Nagarajan, 2016; Tiwari et al., 2017).

CONCLUSION

The physico-chemical characteristics of the untreated fertilizer industry effluent sample exhibited high pH, conductivity, free ammonia, chloride and chromium while low dissolved oxygen, iron and nitrate might be the reason for its high toxicity. The treated effluent sample had no toxic effect even when undiluted. This may be due to the fact that the physico-chemical parameters were in defined range of ISO guidelines (BIS 10500, 1991, 2003). The sublethal exposure of the untreated effluent sample clearly resulted in an inhibitory effect on the AChE activity in a concentration-dependent manner. These results indicate impairment of neuronal signalling on exposure to fertilizer industry effluent. However, further biochemical and molecular studies may help in better understanding of the mechanism of action of fertilizer industrial pollutants. The Government of India has made it mandatory for all industries to treat their effluents properly before their discharge. However, accidental discharge or inadequate management of untreated effluents may have disastrous effect on aquatic organisms.

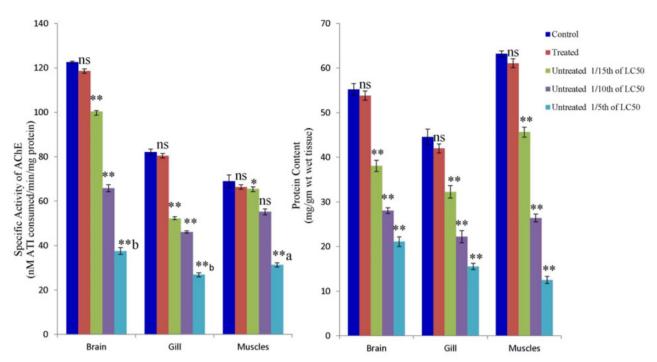


Fig 2. Specific activity of AChE (nM ATI/min/mg protein) and protein content (mg/gm wet wt. of tissue) in brain, muscle and gills of *H. fossilis* exposed to the untreated and treated fertilizer industry effluent. Data represent mean \pm S. E. M. (**p < 0.01; Control versus 1/15th, 1/10th and 1/5th of LC₅₀ and a p < 0.05; Control versus 1/15th, 1/10th and 1/5th of LC₅₀; ns- not significant).

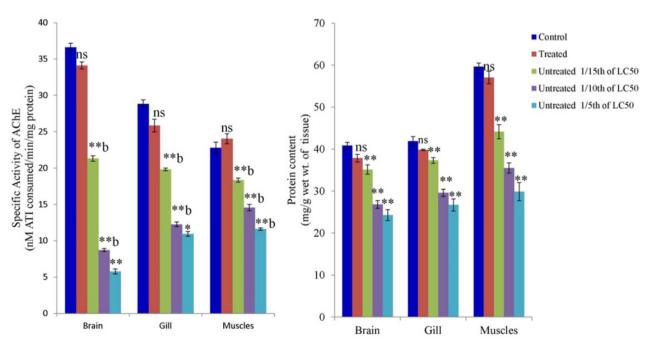


Fig 3. Specific activity of AChE (nM ATI/min/mg protein) and protein content (mg/gm wet wt. of tissue) in brain, muscle and gills of *L. rohita* exposed to the untreated and treated fertilizer industry effluent. Data represent mean \pm S. E. M. (**p < 0.01; Control versus 1/15th, 1/10th and 1/5th of LC₅₀. b p < 0.01; Control versus 1/15th, 1/10th and 1/5th of LC₅₀; ns- not significant).

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SAŽETAK

FIZIKALNO-KEMIJSKE ZNAČAJKE OTPADNIH VODA INDUSTRIJE GNOJIVA I NJIHOV TOKSI-KOLOŠKI UTJECAJ NA AKTIVNOST ACETILKOLIN-ESTERAZE (AChE) KOD SLATKOVODNIH TELEOS-TA Heteropneustes fossilis I Labeo rohita

Industrijska revolucija je dobar pokazatelj gospodarskog razvoja države, međutim, može biti i prijetnja za floru i faunu ako se u prirodu ispuštaju nepročišćene otpadne vode industrije. Cilj ovog istraživanja procijeniti toksikološke učinke pročišćenih nepročišćenih industrijskih otpadnih voda na aktivnost acetilkolinesteraze (AChE) kod Heteropneustes fossilis i Labeo rohita, jestivih riba različitih karakteristika poput razlika u morfologiji, staništu, ishrani, itd. Analizirani su fizikalno-kemijski parametri otpadnih voda kao što su: pH, električna vodljivost (EC), alkalitet, tvrdoća, otopljeni kisik (DO), fosfati, sulfati, nitrati, slobodni amonijak, kloridi, cink, željezo, krom i kalij jer ti parametri nisu bili u rasponu prema ISO smjernicama. Vrijednost LC_{so} za nepročišćene vode iznosila je 2,34% (v/v) i 0,80% (v/v) tijekom 96 h kod H. fossilis i L. rohita. Smrtnost riba u obrađenim otpadnim vodama nije zabilježena. Tijekom izlaganja subletalnim koncentracijama (1/15, 1/(10), 1/5 LC_{so}) kod obje vrste riba aktivnost AChE u metabolički osjetljivim organima (mozak, mišići, škrge) je bila smanjena. Daljnja istraživanja biokemijskih i molekularnih pokazatelja mogla bi približiti mehanizam njihovog djelovanja.

Ključne riječi: Aktivnost AChE, akutna toksičnost, *Heteropneustes fossilis, Labeo rohita,* otpadne vode industrije gnojiva, fizikalno-kemijske karakteristike

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