

# OPTIMIZATION OF ENERGY SAVINGS IN SHOE SOLE PRODUCTION

Ivana Špelić<sup>1</sup>, Alka Mihelić-Bogdanić<sup>1</sup>, Rajka Budin<sup>2</sup>

<sup>1</sup>University of Zagreb, Faculty of Textile Technology, Prilaz baruna Filipovića 28a, 10000 Zagreb, Croatia

E-mail: [ispelic@tff.hr](mailto:ispelic@tff.hr); [amihelic@tff.hr](mailto:amihelic@tff.hr)

<sup>2</sup>University of Zagreb, Faculty of Chemical Engineering and Technology, Marulićev trg 19, 10000 Zagreb, Croatia

E-mail: [rbudin@fkit.hr](mailto:rbudin@fkit.hr)

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## Abstract

Since 1970's, the most popular shoe soles are (EVA) soles, made from Ethylene Vinyl Acetate, copolymer consisting of ethylene and vinyl acetate. The thermosets and elastomers (rubbers), among them accounting EVA, encompasses around 30% of the tonnage of all synthetic polymers produced, with the synthetic rubbers exceeding the tonnage of natural rubber. With that in mind, the energy savings in rubber processing plants are of great importance and the paper analyses the potential of energy savings in shoe soles production process made of Ethylene Vinyl Acetate (EVA). The shoe soles made from EVA are lightweight, easy to mould, water and moisture resistant, highly elastic, shock absorbent, great thermal insulators, highly durable, etc.

The energy savings using the process return condensate in shoe sole production process are presented. Using the return condensate results in lower make up water consumption, substantial fuel savings needed to produce steam and lower chemical consumption. Returning hot process condensate to the boiler results in oil savings of 14,9%. Also, the thermal pollution is reduced by 95,3%, while the volume of the flue gases is lowered from 17,11 m<sup>3</sup><sub>FG</sub>/kg<sub>NEC</sub> to 14,57 m<sup>3</sup><sub>FG</sub>/kg<sub>NEC</sub> or by 14,8%. Such a system enables both the oil savings and reduces the thermal pollution.

The comparison of process with and without flue gases heat recovery shows fuel savings of 18,76%, while the temperature reduces from 221°C to 137,39°C and while the volume of the flue gases is lowered to 13,90 m<sup>3</sup><sub>FG</sub>/kg<sub>P</sub>.

## Ključne riječi

energy savings; return condensate; shoe sole production, Ethylene Vinyl Acetate (EVA), thermal pollution reduction.

## 1. INTRODUCTION

The last century was marked by appearance of new plastic foams, among others new thermosetting foams were introduced to the global market. Thermosetting foams, among which there are also natural and synthetic rubbers, are defined for not having thermoplastic properties and are prepared by simultaneous occurrence of polymer formation and gas generation. As a general rule, any polymer-forming reaction can be used for foam preparation. First rubber production began in World War I as natural or raw rubber products were introduced (Landrock, 1995). Engineering materials are classified as metals, ceramics, polymers and composites. Polymers are further classified as thermoplastics, thermosets and elastomers. Thermoplastic polymers can be subjected to multiple heating and cooling cycles without substantially altering the molecular structure of the polymer. Common thermoplastics include polyethylene, polystyrene, polyvinylchloride, and nylon (Groover, 2010).

There are several industries involved in the production and processing of rubber, but today most of the rubber

used is classified as synthetic rubber, which is produced by the petrochemical industry in processor (fabricator) plants. Many foamed rubber parts, such as shoe soles, are produced by molding. Products formulated from thermoplastic polymers, which include ethylene vinyl acetate, polyethylene, styrene block copolymer, butyl rubber, polyamide, polyurethane, and polyester, can be prepared as hot melt. Single-component thermoplastic hardens from molten state after cooling from elevated temperatures (Groover, 2010).

The shoe soles or outsoles are the bottom part of the shoes in direct contact with the floor. Nowadays they are usually made from synthetic polymers such as Polyisoprene (IR), Thermoplastic Polyurethane (TPU), Polyurethane foams, Ethylene Vinyl Acetate (EVA) or Polyvinylchloride (PVC). Synthetic polymers are made from various petroleum-based monomers and the polymers have become the main specialized materials for footwear industry (Karkalića et al., 2017).

Since the creation in 1970's, the most popular soles are Ethylene Vinyl Acetate (EVA) soles (Wang et al., 2012; Fuss et al., 2014). A large range of articles using EVA are being produced for the footwear industry (Lopes et al., 2015). Ethylene Vinyl Acetate is polymer based copolymer consisting of ethylene and vinyl acetate (Shi et al., 2008; Brito e Dias et al., 2018), also called expanded rubber or foam rubber, Figure 1. It is a polyolefin with the ethylene weight percentage varying from 60 - 90% (Mike Chung, 2002), while the vinyl acetate weight percentage usually varies from 10 to 40% (Nautiyal, 2012).

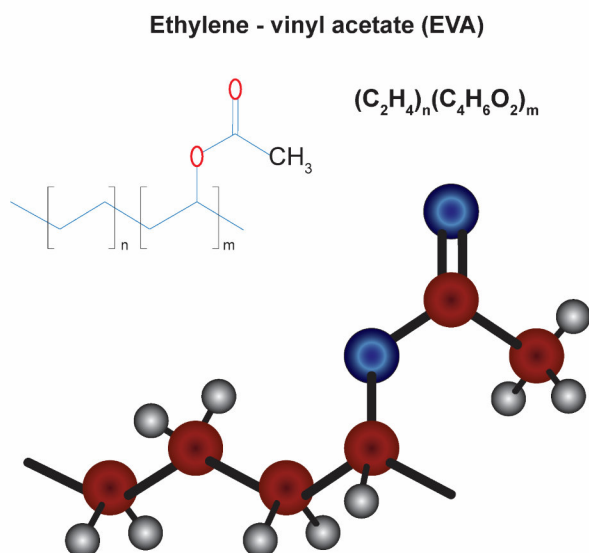


Figure 1. The structural formula of Ethylene Vinyl Acetate (EVA)

The Ethylene Vinyl Acetate can be injected like other polyolefin elastomers (new class of polymers, mostly copolymers of ethylene-butene or ethylene-octen, produced using metallocen catalyst) (Mike Chung, 2002). This EVA material has good clarity, low temperature toughness, stress-crack resistance, hot-melt adhesive water proof properties, and resistance to UV radiation, similar softness, extreme elasticity and flexibility of elastomeric material, but it can also be processed like other thermoplastic materials (Mike Chung, 2002; Wang et al., 2012; Nautiyal, 2012). The main characteristics of EVA is considerable elasticity characterized by the Young Modulus ranging from 15 to 80 MPa (Brito e Dias et al., 2018).

EVA is also known as expanded rubber or foam rubber. The material is known to be lightweight and easy to mould, but also owns its' popularity to cheaper price when compared to natural rubber. Other popular properties involve good water and moisture resistance, high elasticity, great shock absorption, great thermal insulation properties, durability, low-temperature toughness, stress-crack resistance, hot-melt adhesive water proof properties and resistance to UV radiation (Jeng et al., 2012). Soles made from Ethylene vinyl

acetate (EVA) are known for their softness and flexibility. Figure 2. They are processed like other thermoplastics materials. EVA has little or no odour and is competitive with rubber and vinyl products in many electrical applications (Nautiyal, 2012).



Figure 2. The EVA shoe soles for sneakers

### 1.1 SHOE SOLES INJECTION MOLDING PROCESS

Principal molding processes for rubber are (Groover, 2010):

1. compression molding,
2. transfer molding, and
3. injection molding.

There are two basic ways to mold the EVA soles. First one is the compression molded EVA (CMEVA) is made by filling a mold with EVA pellets to achieve desired shape or by compressing a block of EVA foam inside a metal mold. The heat and pressure are applied to the mold, causing the pellets to melt or causing the EVA foam to expand and fill the mold cavity. The second one is the injection molding process. Direct injection is the process of forcing a heated thermoplastics polymer of a highly plastic state under high pressure into the cavity of a mold to achieve desired shape. The process produces discrete components that are almost always net shape. The production cycle time is typically in the range of 10 to 30 sec (Groover, 2010). When the material solidifies in the mold it takes the desired shape and is removed from the mold. The injection molding is quite popular due to zero wastage as the exact amount of EVA is required. A single cavity or multiple cavities moulds are generally made of tools steal, aluminium or stainless steel (López, 2014). The injection molding is economical only for large production quantities. With injection molding of rubber, there are risks of premature curing. Advantages of injection molding include better dimensional control, less scrap and shorter cycle times. Because of high mold costs, large production quantities

are required to justify injection molding (Groover, 2010).

An injection molding machine consists of, Figure 3:

1. the plastic injection unit (similar to the extruder) and
2. the mold clamping unit (power press).

The plastic injection unit consists of a barrel that is fed by a hopper containing a supply of plastic pellets. Inside the barrel is a reciprocating unit, which turns and heats the polymer, and also injects the molten plastic into the mold. The mold clamping unit holds the two platens, affixed and a moveable platen, in proper alignment with each other, and keeps the mold closed during injection by applying a clamping force sufficient to resist the injection force and opens and close the mold (Groover, 2010).

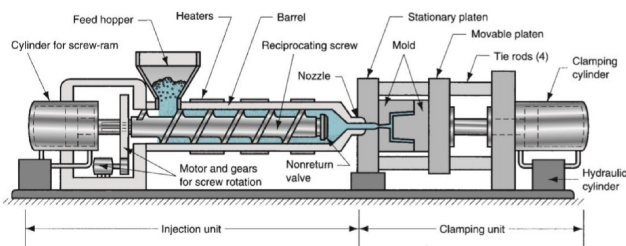


Figure 3. The injection molding machine

The EVA mixed compound is taken to an injector that can be rotating or multi-station since the average injection cycle of each mold is around 7 minutes, so with a multi-station machine you can achieve a faster and more efficient production (Ferreir et al., 2018).

The steps of the EVA injection process (Ferreir et al., 2018):

- (a) The mixture is added to the injector storage tank.
- (b) The injection volume is calculated and adjusted according to the mold.
- (c) The material is heated to 95 °C and injected into the mold at a pressure of ~1200bar.
- (d) The mold is heated to 170 °C to activate the expander and also the crosslinking agents
- (e) The mold is kept closed at 170 °C until the crosslinks are fully completed, about 1.5min / mm thick
- (f) The mold is opened, and the EVA sole jumps out of the cavity rapidly and begins to expand.

The technical description for the selected Automatic Eva Foam Injection Molding Machine and the yearly energy analysis is shown in Table 1.

Table 1. The Technical Description for Automatic Eva Foam Injection Molding Machine

Technical Description	
Work Stations	6
Clamping Force (kg)	170000-230000
Mould-Opening Stroke (mm)	350
Standard Thickness of Mould (mm)	100-210-230
Mould Plate Size (mm)	2(290*550)
Height of Injection Nozzle (mm)	60
Method of Heating	electric
Heating Power (kw)	8.04
Injection System	
Injectors (set)	2
Diameter of Screw (mm)	60/70
L/D Ratio of Screw	23
Max Injection Capacity (cc)	920-1100
Injection Pressure	1100
Injection Speed (cm/sec)	10
Rotating Speed of Screw (r.p.m)	0-190
Temperature Control (section)	4
Electricity Consumption	
Heater Barrel (kw)	12.1
Motor for Injector Moving (kw)	1.5
Heating Board of Mould (kw)	76.8
Hydraulic System (kw)(50hz)	40.5
Vacuum Pump (kw)	2.2
Cooling Fan of Barrel (HP)	1
Total Electricity (kw)	50/70/150
Oil Tank Size (L)	1000
Machine Weight (kg)	23800
Machine size (m)	6.5*4.5*2.8
Output (Pairs/day)	5000
Working times (h/day)	8
Monthly working times (days/month)	25

## 1.2 INPUT DATA AND ENERGY ANALYSIS OF TECHNOLOGICAL PROCESS

The energy consumption in shoe sole process manufacturing at rate  $n=2500$  per shift (or 5000 pairs per day or 625 pairs per hour), is carried out. This technology (Fig. 2) includes processes in mixer, warming mill, calender, cutter, extruder, cooler etc. The plant works 8 hours in day, 25 days in month or  $\tau=2400$  hours yearly, i.e. so the plant use factor becomes  $\beta=27$ , 4%. Such a basic process needs electrical energy in amount of  $e_{el}=3,44 \text{ kWh}_e/\text{kg}$  product and dry saturated steam (13 bar) in mass of  $d_s=14,870 \text{ kg}_s/\text{kg}$  product. The process uses also water in mass of  $d_w=4,57 \text{ kg}_w/\text{kg}$  product (Mihelić-Bogdanić i Budin, 2008). Dry saturated steam is produced in an oil fuelled boiler with efficiency  $\eta_b=80\%$ .

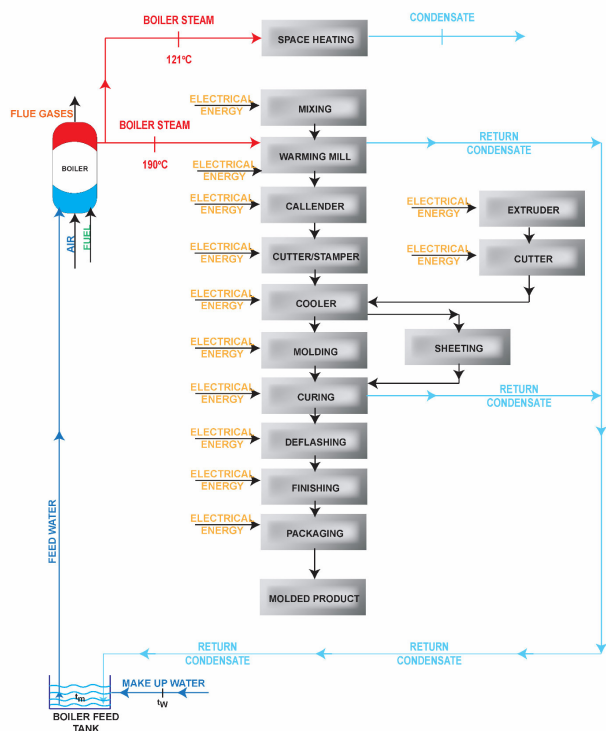


Figure 4. Shoe sole process flow diagram

The percentage composition of oil by mass is: C:H:N:S = 85,3%:11,6%:0,6%:2,5% and is burned with excess air coefficient  $\alpha=1,25$ . A part of saturated steam  $d_{SP}=14,170 \text{ kg}_S/\text{kg}_P$  is used for industrial process and the rest  $d_{SH}=0,7 \text{ kg}_S/\text{kg}_H$  for space heating. The lower heating value of oil is  $H_L = 41268 \text{ kJ/kg}$  (Mihelić-Bogdanić i Budin, 2008). Feed water enters the boiler with temperature  $t_B=25^\circ\text{C}$  and the whole condensate with temperature  $t_C=121^\circ\text{C}$  is withdrawn to the surrounding. The air required for oil combustion passes into firebox with temperature  $t_a = 25^\circ\text{C}$ , while the temperature of exhaust stacks is  $t_{FG}=221^\circ\text{C}$ .

The heat transferred to the steam generator is:  $q_s=d_s(h_s-h_b)=14,870(2785-104,68) = 39856,36 \text{ kJ/kg}_P$ , where  $h_s$  and  $h_b$  are the steam and water enthalpies taken from thermodynamically tables (Mihelić – Bogdanić and Budin, 2008).

From this data the unit mass of oil requirement using heat balance is:

$$d_F = q_s / H_L \times \eta_B = 39856,36 / 41268 \times 0,80 = 1,21 \text{ kg}_F/\text{kg}_P$$

The specific steam consumption is:

$$d_{SP} = d_s / d_F = 14,870 / 1,21 = 12,29 \text{ kg}_S/\text{kg}_F$$

or the oil consumption becomes:

$$d_{SF} = d_F / d_s = 1,21 / 14,870 = 0,0814 \text{ kg}_F/\text{kg}_S$$

Taking into consideration yearly operating time  $\tau=2400$  hours and shoe sole rate  $n=625$  pairs, oil consumption is:

$$D_{FY} = d_F \times \tau \times n = 1,21 \times 2400 \times 625 = 1,815 \times 10^6 \text{ kg}_F$$

The total condensate  $d_C$  from the process  $d_{CP}=14,170 \text{ kg}_C/\text{kg}_P$  and from the space heating  $d_{CH}=0,7 \text{ kg}_C/\text{kg}_H$  with temperature  $t_C=121^\circ\text{C}$  is withdrawn to the surrounding (Fig. 3).

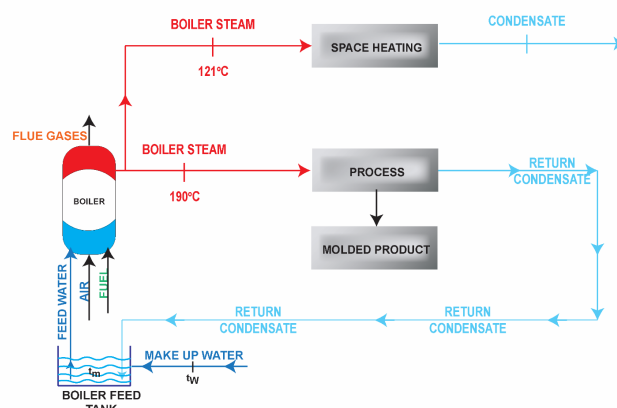


Figure 5. Returning hot process condensate

The heat of the condensate is:

$q_C = (d_{CP} + d_{CH})h_{121} = (14,170 + 0,7) 507 = 7539,1 \text{ kJ/kg}_P$ . To improve process efficiency [9], the waste condensate could be applicable for feed water preheating.

## 2. ANALYSES OF COMBUSTION PRODUCTS

The products of fuel combustion are mostly gaseous (Eastop and McConkey, 1994). For complete oil combustion, 25% air is supplied. The minimum oxygen volume  $V_{O_2 \text{ min}}$  and stehiometric air volume  $V_a$  required for combustion is expressed as:

$$V_{O_2 \text{ min}} = (22,4/12) \{C + 3[H - (o-s)/8]\} = 2,26 \text{ m}^3_{O_2}/\text{kg}_F$$

$$V_a = V_{O_2 \text{ min}} / 0,21 = 10,8 \text{ m}^3_a/\text{kg}_F$$

To make safe the complete combustion of oil the excess coefficient of  $\alpha=1,25$  was taken.

The actual volume of air is:

$$V_{a\alpha} = V_a \times \alpha = 10,8 \times 1,25 = 13,5 \text{ m}^3_a/\text{kg}_F$$

The total volume of wet flue gases  $V_{FG}$  consists of carbon dioxide, sulphur dioxide, nitrogen, excess oxygen and water vapour:

$$V_{FG} = V_{CO_2} + V_{SO_2} + V_{N_2} + V_{O_2} + V_{H_2O}$$

Volume of each gas component is calculated using following expressions:

$$V_{CO_2} = (22,4/12)C = (22,4/12)0,853 = 1,592 \text{ m}^3_{CO_2}/\text{kg}_F$$

$$V_{SO_2} = (22,4/32)S = (22,4/32)0,025 = 0,0175 \text{ m}^3_{SO_2}/\text{kg}_F$$

$$V_{N_2} = (22,4/28)S + 0,79 \times V_{a\alpha} = (22,4/28)0,006 + 0,79 \times 13,5 = 10,67 \text{ m}^3_{N_2}/\text{kg}_F$$

$$V_{O_2} = 0,21(\alpha-1)V_a = 0,21(1,25-1)10,8 = 0,567 \text{ m}^3_{O_2}/\text{kg}_F$$

$$V_{H_2O} = (22,4/2)H + (22,4/18)W = (22,4/2)0,116 + (22,4/18)0 = 1,299 \text{ m}^3_{H_2O}/\text{kg}_F$$

and the whole volume of exhaust gases with excess air is:

$$V_{FG} = 14,1455 \text{ m}^3_{FG}/\text{kg}_F$$

The proportional of each constituent expressed as a percentage of the total wet product is:

$$\text{CO}_2 : 11,25\%; \text{SO}_2 : 0,12\%; \text{N}_2 : 75,43\%; \text{O}_2 : 4,02\%; \\ \text{H}_2\text{O} : 9,18\%.$$

The specific heat of exhaust gases with temperature  $t_{FG} = 221^\circ\text{C}$  and the percentage by volume of products as well as the specific heat of each gas, should be obtained (Bošnjaković, 1971):

$$c_{pFG} = c_{pCO_2} \cdot c\%_{CO_2} \cdot V_{CO_2} + c_{pSO_2} \cdot x\%_{SO_2} \cdot V_{SO_2} + c_{pN_2} \cdot x\%_{N_2} \cdot V_{N_2} + c_{pO_2} \cdot x\%_{O_2} \cdot V_{O_2} + c_{pH_2O} \cdot x\%_{H_2O} \cdot V_{H_2O} = 1,392 \text{ kJ/m}^3 \text{ grad}.$$

The energy released on the complete combustion per unit mass of the fuel or per unit of products depends on total flue gases volume, specific heat and flue gases temperature:

$$q_{FG} = V_{FG} \cdot c_{pFG} \cdot t_{FG} = 14,1455 \times 1,392 \times 221 = 4351,61 \text{ kJ}_{FG}/\text{kg}_F \\ q_{FGP} = q_{FG} \cdot d_F = 4351,61 \times 1,21 = 5265 \text{ kJ}_{FG}/\text{kg}_P.$$

The volume of exhaust gases per unit of product rejected to the surrounding with temperature  $t_{FG} = 221^\circ\text{C}$  is:

$$V_{FGP} = V_{FG} \cdot d_F = 14,1455 \times 1,21 = 17,11 \text{ m}^3_{FG}/\text{kg}_P.$$

### 3. OPTIMIZATION OF ENERGY SAVINGS IN SHOE SOLE PRODUCTION

#### 3.1. PROCESS CONDENSATE HEAT RECOVERY

The returning hot condensate to the boiler has many several reasons. Using return process condensate to heat feed water less amount of make-up water is required, less oil is needed to produce steam from hot water rather than cold water. Return of high purity condensate i.e. distilled water reduces also chemicals and treatment cost as well as energy losses due to the boiler blowdown [8]. In analysed procedure the process condensate in amount of  $d_{CR} = 14,170 \text{ kg}_C/\text{kg}_P$  with temperature  $t_C = 121^\circ\text{C}$  is returned and mixed with the make-up water in mass  $d_W = 0,700 \text{ kg}_W/\text{kg}_P$  with temperature  $t_W = 25^\circ\text{C}$ .

Considering the adiabatic mixing process, the feed water will enter the boiler with temperature:

$$t_m = (d_{CR} \cdot t_C + d_W \cdot t_W) / d_s = (14,170 \times 121 + 0,700 \times 25) / 14,870 = 116,48^\circ\text{C}.$$

The oil consumption is after heat balance:

$$d_{FC} = d_s \cdot (h_s - h_m) / H_L \cdot \eta_B = 14,870 (2785 - 492) / 41268 \times 0,8 = 1,03 \text{ kg}_F/\text{kg}_P.$$

The specific steam consumption is:

$$d_{SPC} = d_s / d_{FC} = 14,870 / 1,03 = 14,44 \text{ kg}_S/\text{kg}_P.$$

Yearly consumption is:

$$D_{FCY} = d_F \cdot x \cdot \eta = 1,03 \times 2400 \times 625 = 1,545 \times 10^6 \text{ kg}_F.$$

Heat condensate for feed water preheating is:

$$Q_C = d_C \cdot h_{121} = 14,170 \times 507 = 7184,19 \text{ kJ}_C/\text{kg}_P.$$

The application of return condensate with process without condensate heat recovery shows oil savings of:

$$S = (d_F - d_{FC}) / d_F = (1,21 - 1,03) / 1,21 = 0,1487 \text{ i.e. } 14,9\%.$$

The condensate heat, which is discharged to the surrounding becomes:

$$q_{CH} = d_{CH} \cdot h_{121} = 0,7 \times 507 = 354,9 \text{ kJ}_C/\text{kg}_P.$$

So, the thermal pollution is decreased from 7539,1  $\text{kJ}_C/\text{kg}_P$  to 354,9  $\text{kJ}_C/\text{kg}_P$  or for 95,3%.

In this analyses case the flue gases volume per unit of product that is rejected to the atmosphere with temperature  $t_{FG} = 221^\circ\text{C}$  is:

$$V_{FGC} = V_{FG} \cdot d_{FC} = 14,1455 \times 1,03 = 14,57 \text{ m}^3_{FG}/\text{kg}_P.$$

The volume of exhaust gases is lowered from 17,11  $\text{m}^3_{FG}/\text{kg}_P$  to 14,57  $\text{m}^3_{FG}/\text{kg}_P$  or for 14,8%.

#### 3.2. FEED WATER PREHEATING WITH FLUE GASES

Input data summary:

The economizer efficiency  $\eta_E = 80\%$

The economizer water inlet temperate  $t_{WEI} = 116,78^\circ\text{C}$

The volume of exhaust gases  $V_{FG} = 14,1455 \text{ m}^3_{FG}/\text{kg}_F$

The economizer flue gases inlet temperate  $t_{FGEI} = 221^\circ\text{C}$

The specific steam consumption  $d_{SP} = 14,44 \text{ kg}_S/\text{kg}_F$

The specific heat of water  $c_{pH_2O} = 4,187 \text{ kJ/kg K}$

The specific heat of flue gases  $c_{pFG} = 1,392 \text{ kJ}_{FG}/\text{m}^3 \text{ deg}$

The mass of dry saturated steam  $d_s = 14,870 \text{ kg}_S/\text{kg}_P$

The enthalpy of steam  $h_s = 2785 \text{ kJ/kg}$

The fuel lower heating value  $H_L = 41268 \text{ kJ/kg}_F$

Cold water with temperature  $t_{WEI} = 116,78^\circ\text{C}$  enters the economizer where is heated with flue gases in amount

$V_{FG} = 14,1455 \text{ m}^3_{FG}/\text{kg}_F$  and inlet temperature

$t_{FGEI} = 221^\circ\text{C}$ . The specific steam consumption is

$d_{SP} = 14,44 \text{ kg}_S/\text{kg}_F$ . With specific heat of water  $c_{pH_2O} = 4,187 \text{ kJ/kg K}$  and flue gases  $c_{pFG} = 1,392 \text{ kJ}_{FG}/\text{m}^3 \text{ deg}$  calculated previously.

The economizer water outlet temperate ( $t_{WEO}$ ) can be

calculated from the energy balance equation:

$V_{FG} \cdot c_{pFG} \cdot (t_{FGEI} - t_{WEO}) \cdot \eta_E = d_{SP} \cdot c_{pH_2O} \cdot (t_{WEO} - t_{WEI})$

$t_{WEO} = [V_{FG} \cdot c_{pFG} \cdot (t_{FGEI} - t_{WEI}) \cdot \eta_E + d_{SP} \cdot c_{pH_2O} \cdot t_{WEI}] /$

$d_{SP} \cdot c_{pH_2O}$

$t_{WEO} = 143,71^\circ\text{C}$

The economizer flue gases outlet temperate ( $t_{FGEo}$ ) can

be calculated from the relation:

$V_{FG} \cdot c_{pFG} \cdot (t_{FGEI} - t_{FGEo}) = d_{SP} \cdot c_{pH_2O} \cdot (t_{WEO} - t_{WEI})$

$t_{FGEo} = [d_{SP} \cdot c_{pH_2O} \cdot (t_{WEO} - t_{WEI}) - V_{FG} \cdot c_{pFG} \cdot t_{FGEI}] / V_{FG} \cdot$

$c_{pFG}$

$t_{FGEo} = 137,39^\circ\text{C}$

The fuel consumption is then calculated as:

$$d_{FE} = d_s \cdot (h_s - h_{w_{E0}}) / \eta_B \cdot H_L = 14,870 \cdot (2785 - 601,7) / 0,8 \cdot 41268 = 0,983 \text{ kg}_F / \text{kg}_P.$$

where  $h_s$  and  $h_{w_{E0}}$  are enthalpies of steam and water.

The comparison of process with and without flue gases heat recovery shows fuel savings of:

$$S_1 = (d_F - d_{FE}) / d_F = (1,21 - 0,983) / 1,21 = 0,1876 \text{ i.e. } 18,76\%.$$

The flue gases rejected to the atmosphere is now:

$$V_{FG E} = V_{FG} \cdot d_{FE} = 14,155 \cdot 0,983 = 13,90 \text{ m}^3_{FG} / \text{kg}_P.$$

## CONCLUSIONS

Calculating the energy savings in rubber processing plants is of great importance since the synthetic rubbers accounts for the majority of rubber production. Most of the shoe soles today are made from Ethylene Vinyl Acetate (EVA). This paper analyses the potential of energy savings using the process return condensate and feed water preheating using the flue gases heat recovery in EVA shoe soles production process. The energy savings are seen through lower make up water consumption, substantial fuel savings and lower chemical consumption. Returning hot process condensate to the boiler results in oil savings of 14,9%, the thermal pollution is reduced by 95,3%, while the volume of the flue gases is lowered from 17,11  $\text{m}^3_{FG} / \text{kg}_{NEC}$  to 14,57  $\text{m}^3_{FG} / \text{kg}_{NEC}$  or by 14,8%. Also, The implementation of flue gases heat recovery for feed water preheating can improve the technological process. By applying the presented method, the flue gases are cooled which causes increasing of feed water temperature.

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