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# Studies in the Sphingolipids Series. XII.\* Structure of the Cerebrin Anhydro Base of Yeast (C<sub>20</sub>-Phytosphingosine Anhydro Base)

M. Proštenik, B. Majhofer-Oreščanin, M. Munk-Weinert, and B. Ries-Lešić

Department of Biochemistry, Institute »Ruđer Bošković« and Department of Chemistry, Medical Faculty, University of Zagreb, Zagreb, Croatia, Yugoslavia

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It was confirmed that the cerebrin anhydro base from yeast (IIa) isolated according to the procedure established by Reindel et al.<sup>2,3</sup> has a C<sub>20</sub>-structure rather than C<sub>18</sub>-structure as one might conclude erroneously by simple comparison of physical constants of both bases. The tetrahydrofuran structure was also demonstrated by means of the IR spectroscopy. For this purpose following new compounds derived from C<sub>20</sub>-anhydro base were prepared: N,O-diacetyl (IIb), N,O-ditosyl (IIc), O-acetyl-N-benzoyl (IId) and N-phthaloyl derivative (III); in addition, 2-hexadecyl-3,4-dihydroxy-tetrahydrofuran (IV) — a deamination product of the base was described.

Carter, Celmer, Lands, Mueller, and Tomizawa<sup>1</sup> have discovered that phytosphingosine — a base of plant sphingolipides, obtained from corn, soybean and other phosphatides — has a structure of an  $C_{18}$  amino alcohol (Ia).\*\* On the other hand, Reindel et al.<sup>2,3</sup> alotted a C<sub>20</sub>-structure to the base isolated from yeast (cerebrin base, Ib). The existence and preparation of the corresponding anhydro bases (IIa and C18-base) was also reported by both schools. Recently, the  $C_{20}$ -structure of cerebrin base — isolated exactly according to the Reindel's procedure - was confirmed in our laboratories and the correct distribution of four contiguous functional groups established as well (Ib)<sup>4</sup>. It was announced at the same time that the base might be a mixture of the  $C_{18}$  and  $C_{20}$ -compounds. This was confirmed experimentally by Oda and Kamiya<sup>5</sup> by means of the reverse phase column chromatography of fatty acids obtained by the oxidation of the yeast base, and by Sweeley6 who analyzed the oxidation products by virtue of gas chromatography. More recently, O'Connell and Tsien<sup>7</sup> have described in detail the isolation and identification of phytosphingosine anhydro base of corn phosphatides. The base - described for the first time in a free condition — has a  $C_{18}$ -chain and the physical constants were in good agreement with those reported for the  $C_{20}$ -base (cerebrin anhydro base, IIa).

\* Paper XI: D. Keglević, A. Kisić, N. Krvavica-Feretić, and M. Proštenik, Croat. Chem. Acta 31 (1959) 41.

<sup>\*\*</sup> The name *phytosphingosine* was proposed by the same authors preferentially for the  $C_{18}$ -base. Since that time the  $C_{20}$ -base has been found with certainty in nature and there are indications for the occurrence of other homologues. From now on we shall term all homologous bases with the same distribution of four functional groups as  $C_{18}$ -phytosphingosine,  $C_{20}$ -phytosphingosine *etc.* 

In this paper we wish to contribute additional data in order to support the  $C_{20}$ -formulation of the yeast anhydro base as isolated previously<sup>2,3</sup>. The only product which could be isolated by Reindel's procedure was the lower melting isomer, m. p. 88—89<sup>6</sup>, of the  $C_{20}$ -base. It is interesting to mention on this occasion that the  $C_{18}$  and  $C_{20}$ -anhydro bases may exist each in two isomeric forms — as a lower and a higher melting one. In paper chromatographic assay on Whatman No. 1 paper impregnated with silicic acid IIa appeared to be homogeneous showing a single spot<sup>8</sup>.

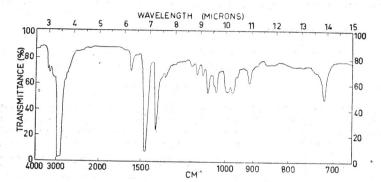
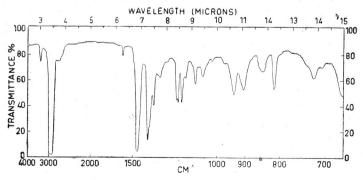
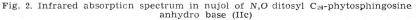


Fig. 1. Infrared absorption spectrum in nujol of C20-phytosphingosine anhydro base (IIa)

The base IIa was converted into several hitherto either unknown or unsufficiently characterized derivatives such as N,O-diacetyl (IIb), N,O-ditosyl (IIc), O-acetyl-N-benzoyl (IId) and N-phthaloyl (III) compound. The N,O-ditosyl derivative prepared from the lower melting base (m. p. 88—89°) was reduced with lithium aluminium hydride in order to remove either the  $C_3$ -hydroxy group, or — what was more probable — only the O-tosyl group. Reactions like this usually proceed with retention of the N-tosyl groups. Unexpectingly, both tosyl groups were split off simultaneously, and the higher melting isomer (m. p. 105°) of IIa was obtained. The mixed melting point with the lower melting base was 97°. Deamination of IIa by means of nitrous acid proceeded smoothly yielding the nitrogen-free 2-hexadecyl-3,4-dihydroxytetrahydrofuran

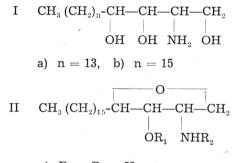




(IV). All these compounds gave good elementary analyses for the  $C_{20}$ -formulation. The analytical data exclude the admixture of noticeable quantities of the  $C_{18}$ -base.

The presence of the tetrahydrofuran ring in IIa was detected by means of the infrared spectroscopy. The absorption characteristic for the tetrahydrofuran structure occurs at 912 and 1075  $\rm cm^{-1}$ .

In conclusion, the bases isolated from corn and soybean phosphatides and — on the other hand — from yeast using Reindel's procedure are not identical. The former has a  $C_{18}$  the latter a  $C_{20}$ -chain, although the physical properties such as melting points and specific rotations are nearly identical. Consequently, the comparison of properties of both bases cannot be a reliable means of identification.



a) 
$$R_1 = R_2 = H$$

b)  $R_1 = R_2 = CH_3CO$ 

- c)  $R_1 = R_2 = p CH_3C_6H_4SO_2$
- d)  $R_1 = CH_3CO$ ,  $R_2 = C_6H_5CO$

 $\cap$ 

III 
$$CH_{3} (CH_{2})_{15}$$
- $CH$ - $CH$ - $CH$ - $CH$ - $CH_{2}$   
 $|$   $|$   $OH$   $N(CO)_{2}C_{6}H_{4}$ 

IV 
$$CH_3 (CH_2)_{15}$$
- $CH$ - $CH$ - $CH$ - $CH$ - $CH_2$ 

#### EXPERIMENTAL

The melting points are uncorrected. The infrared absorption spectra were measured on a Perkin-Elmer Model 134 spectrophotometer.

### Starting Materials

The preparation of  $C_{20}$ -phytosphingosine anhydro base (IIa) employed in this investigation has been achieved starting with crude yeast cerebrin following exactly the two-stage procedure given by Reindel *et al.*<sup>2,3</sup>

# N,O-Ditosyl $C_{20}$ -Phytosphingosine Anhydro Base (IIc)

 $C_{20}$ -Phytosphingosine anhydro base (IIa, 200 mg., m. p. 88–89<sup>0</sup>), toluene sulphochloride (512 mg.) and pyridine (1.5 ml.) were mixed and left to stand at room temperature for 24 hrs. The mixture was then poured onto 10 g. of crushed ice and cooled in the refrigerator for 2—3 hrs. During this time the oily phase crystallized completely. The substance was collected and washed with cold water. One crystallization from methanol of the crude product gave 350 mg. of colourless powder, m. p. 64—65°. One more crystallization from the same solvent raised the m. p. to 75—76°. The analytical sample melted — after four additional crystallizations — at 79—80°.  $[\alpha]_{12}^{18}$ —18.8° (c, 0.8 in methanol).

Anal. 7.935 mg. subst.: 18.74 mg. CO<sub>2</sub>, 5.92 mg. H<sub>2</sub>O 7.870 mg. subst.: 0.147 ml. N<sub>2</sub> (20.5°, 763 mm)  $C_{34}H_{55}NO_6S_2$  (635.93) calc'd.: C 64.22; H 8.40; N 2.20°/° found: C 64.45; H 8.35; N 2.18°/°

# C<sub>20</sub>-Phytosphingosine Anhydro Base, m. p. 105<sup>o</sup>

A 400 mg. sample of the ditosyl derivative IIc (m. p. 79–80°) was suspended in 40 ml. of dry ether and added to a solution of 400 mg. of lithium aluminium hydride in 40 ml. of ether. After refluxing for 5 hrs. the reaction mixture was treated carefully with water, the ether phase separated by decantation and the solid residue washed with ether. The combined ether extracts were evaporated to dryness yielding 200 mg. of a waxy solid, m. p. 82–84°. The crude substance was recrystallized four times from methanol. The crystals melted at 105° indicating the identity with the higher melting form. The mixed melting point with the lower melting isomer (m. p. 88–89°) was 97°.

> Anal. 8.055 mg. subst.: 21.59 mg.  $CO_2$ , 9.00 mg.  $H_2O$ 6.490 mg. subst.: 0.253 ml.  $N_2$  (25%, 736 mm)  $C_{20}H_{41}NO_2$  (327.54) calc'd.: C 73.33; H 12.62; N 4.28% found: C 73.15; H 12.50; N 4.33%

# N,O-Diacetyl $C_{20}$ -Phytosphingosine Anhydro Base (IIb)

A mixture of IIa (200 mg., m. p. 88–89<sup>0</sup>), acetic anhydride (1 ml.) and pyridine (2 ml.) was heated at 100<sup>o</sup> for 2 hrs. The reaction mixture was then poured into cold water (15 ml.), the separated colourless, crystalline solid filtered by suction and washed thoroughly with water. The crude product (190 mg.,  $75.7^{0/0}$ , m. p. 73–75<sup>o</sup>) was recrystallized from petroleum ether and melted at 76–77<sup>o</sup>.

Anal. 6.071 mg. subst.: 15.62 mg. CO<sub>2</sub>, 6.02 mg. H<sub>2</sub>O C<sub>24</sub>H<sub>45</sub>NO<sub>4</sub> (411.61) calc'd.: C 70.03; H 11.20% found: C 70.21; H 11.10%

# O-Acetyl-N-benzoyl $C_{20}$ -Phytosphingosine Anhydro Base (IId)

The acetylation of the *N*-benzoyl derivative (100 mg., m. p.  $105^{0}$ )<sup>2,3</sup> with acetic anhydride (0.5 ml.) and pyridine (1 ml.) was performed in a known way. The crude product was crystallized three times from ethanol and melted at  $89-90^{0}$ .

Anal. 5.365 mg. subst.: 14.51 mg. CO<sub>2</sub>, 4.97 mg. H<sub>2</sub>O C<sub>29</sub>H<sub>47</sub>NO<sub>4</sub> (473.67) calc'd.: C 73.53; H 10.00% found: C 73.81; H 10.37%

# N-Phthaloyl $C_{20}$ -Phytosphingosine Anhydro Base (III)

The mixture of IIa (1.31 g., 4 m*M*, m. p. 88–89°), phthalic anhydride (592 mg.) and toluene (150 ml.) was refluxed under continuous separation of water formed during the reaction. The reaction was completed in 2 hrs. Evaporation of the solvent at  $50-60^\circ$  in vacuo to dryness gave 1.9 g. of the crude product melting at  $102-107^\circ$ . Three crystallizations from ethanol yielded the analytically pure compound melting at  $102-103.5^\circ$ .

Anal. 8.060 mg. subst.: 21.73 mg. CO<sub>2</sub>, 6.80 mg. H<sub>2</sub>O 8.010 mg. subst.: 0.216 ml. N<sub>2</sub> (24<sup>0</sup>, 765 mm) C<sub>28</sub>H<sub>43</sub>NO<sub>4</sub> (457.63) calc'd.: C 73.48; H 9.47; N 3.06<sup>0</sup>/<sub>0</sub> found: C 73.57; H 9.44; N 3.12<sup>0</sup>/<sub>9</sub>

# 2-Hexadecul-3.4-dihudroxutetrahudrofuran (IV)

To a solution of IIa (600 mg., m. p. 88-89%) in glacial acetic acid (15 ml.) solid sodium nitrite (210 mg.) was added in small portions. A strong evolution of nitrogen took place. After standing at room temperature overnight the solution was diluted with water (50 ml.), the resulted precipitate extracted with ether and the extracts dried with sodium sulphate. Evaporation of the solvent gave a somewhat sticky residue which was recrystallized from methanol. The yield of the colourless substance -m. p. 51-520 - amounted to 400 mg. (67%). It was crystallized again from methanol to give a product melting at 57-59°.

> Anal. 8.275 mg. subst.: 22.83 mg. CO<sub>2</sub>, 8.96 mg. H<sub>2</sub>O C20H40O3 (328.54) calc'd.: C 73.12; H 12.27% found: C 73.58; H 12.12%

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### IZVOD

#### Studije u redu sfingolipoida. XII. Struktura anhidro-cerebrinske baze iz kvasca (C<sub>20</sub>-fitosfingozin anhidro baze)

# M. Proštenik, B. Majhofer-Oreščanin, M. Munk-Weinert i B. Ries-Lešić

Na temelju novih analitičkih podataka potvrđeno je, da anhidro-cerebrinska baza (IIa) pripremljena iz kvaščeva cerebrina točno prema podacima Reindel-a i sur.<sup>2,3</sup> ima lanac  $C_{20}$ , a ne  $C_{18}$ . Uspoređivanjem fizikalnih svojstava (tališta i specifičnih skretanja), što su objavili neki autori<sup>1,7</sup>, moglo bi se pogrešno zaključiti da su identične. Tališta i specifična skretanja ne mogu biti kriterij za identifikaciju bliskih homolognih baza cerebrinskoga reda. Dokazana je također tetrahidrofuranska struktura anhidrocerebrinske baze s pomoću IR spektroskopije.

BIOKEMIJSKI ODJEL INSTITUT »RUDER BOŠKOVIĆ«

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