

CCA-369

548.7:549.752:546.33-841-791

Preliminary Note

## Synthesis and Crystallographic Data of Sodium Thorium Triphosphate, $\text{NaTh}_2(\text{PO}_4)_3$ , and Sodium Uranium(IV) Triphosphate, $\text{NaU}_2(\text{PO}_4)_3$

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Received February 2, 1965

Sodium thorium triphosphate was first prepared by K. A. Wallroth<sup>1</sup> and sodium uranium(IV) triphosphate by M. A. Colani<sup>2</sup>. Both compounds are very stable and practically insoluble in acids. Single crystals of these compounds were prepared by methods which slightly differ from the above mentioned.

To obtain the single crystals of  $\text{NaTh}_2(\text{PO}_4)_3$ , a mixture of 2.5 g.  $\text{Na}_4\text{P}_2\text{O}_7 \cdot 10\text{H}_2\text{O}$ , 1.3 g.  $\text{ThO}_2$  and 2.5 g.  $\text{NaCl}$  was heated to 350°C for 24 hours in a platinum crucible. The insoluble residuum, left after boiling in water, was heated again with 2.5 g.  $\text{Na}_4\text{P}_2\text{O}_7 \cdot 10\text{H}_2\text{O}$  and 3.0 g.  $\text{B}_2\text{O}_3$  to 1200°C for 24 hours. This time the fused mixture was cooled very slowly and then washed with acidulated water. The crystals obtained were colourless, transparent monoclinic prisms which were insoluble in all acids; their chemical composition, determined by analysis ( $\text{ThO}_2 = 68.35\%$ ,  $\text{P}_2\text{O}_5 = 27.72\%$ ), corresponds to the formula  $\text{NaTh}_2(\text{PO}_4)_3$ .

To obtain single crystals of  $\text{NaU}_2(\text{PO}_4)_3$  a mixture of 4.0 g.  $\text{NaPO}_3$ , 2.7 g.  $\text{UO}_2$  and 3.0 g.  $\text{B}_2\text{O}_3$  was kept in nitrogen atmosphere for 24 hours at 1200°C in a graphite crucible and then very slowly cooled. The crystals, separated from the solid mass by washing with acidulated water, were dark green monoclinic prisms; they are soluble in a mixture of aqua regia and hydrofluoric acid. The results of the complete chemical analysis ( $\text{Na}_2\text{O} = 4.15\%$ ,  $\text{P}_2\text{O}_5 = 27.53\%$  and  $\text{UO}_2 = 67.25\%$ ) are in agreement with the formula  $\text{NaU}_2(\text{PO}_4)_3$ .

### Crystallographic data

The crystallographic data of both compounds are as follows:

Compound	a(Å)	b(Å)	c(Å)	$\beta$	$D_m$	$D_c$	Z	Space group
$\text{NaTh}_2(\text{PO}_4)_3$	17.39	6.80	8.13	$101^\circ 8'$	5.41 g. cm <sup>-3</sup>	5.44 g. cm <sup>-3</sup>	4	$C_{2h}^4$ (Cc)
$\text{NaU}_2(\text{PO}_4)_3$	17.23	6.70	8.03	$101^\circ 22'$	5.63 g. cm <sup>-3</sup>	5.71 g. cm <sup>-3</sup>	4	$C_{2h}^4$ (Cc)

The lattice constants were determined from oscillation and Weissenberg photographs using  $\text{CuK}\alpha$  radiation. The a : b : c ratios and the angle  $\beta$  for  $\text{NaU}_2(\text{PO}_4)_3$  found in this and in previous work are in close agreement:

$$\begin{aligned} a : b : c &= 2.5605 : 1 : 1.1918, \quad \beta = 101^\circ 3' \quad (\text{Colani } ^2) \\ a : b : c &= 2.571 : 1 : 1.194, \quad \beta = 101^\circ 22' \quad (\text{this work}). \end{aligned}$$

The densities ( $D_m$ ) were determined pycnometrically using decalin as liquid, and ( $D_c$ ) were calculated for four formula units in the elementary cell. The

space group can be  $C_s^4$  (Cc) or  $C_{2h}^6$  (C2/c) because hkl reflections are absent with  $h+k$  odd and h0l reflections are absent with  $l$  odd. Because the piezoelectric effect was found on both compounds, the space group is  $C_s^4$  (Cc). The test was made by the Bergman method as modified by Iitaka<sup>3</sup>. The complete crystal structure determination on the thorium compound is now in progress.

The authors wish to express their sincere thanks to Dr. Š. Mesarić for her help in analytical determinations and to Professor D. Grdenić and Dr. S. Ščavničar for helpful suggestions.

## REFERENCES

1. K. A. Wallroth, *Bull. Soc. Chim.* **39** (1883) 316.
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## IZVOD

**Preparacija i kristalografski podaci o natrijevom torijevom fosfatu i natrijevom uran(IV) fosfatu**

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Opisane su metode za dobivanje monokristala natrijevog torijevog fosfata i natrijevog uran(IV) fosfata. Oba spoja kristaliziraju u monoklinskom sustavu. Metodom rendgenske difrakcije određeni su njihovi kristalografski podaci:

Spoj	a(Å)	b(Å)	c(Å)	$\beta$	Z	Prostorna grupa
$NaTh_2(PO_4)_3$	17.39	6.80	8.13	$101^\circ 8'$	4	$C_s^4$ (Cc)
$NaU_2(PO_4)_3$	17.23	6.17	8.03	$101^\circ 22'$	4	$C_s^4$ (Cc)

Na kristalima oba spoja ustanovljen je piezoelektrični efekt.

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PRIRODNO MATEMATIČKI FAKULTET  
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Primljeno 2. veljače 1965.