1	Identification of recalcitrant compounds in a pilot-scale AB system: an Adsorption (A) stage
2	followed by a Biological (B) stage to treat municipal wastewater.
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22 Abstract

35	1. Introduction
34	
33	Microbial Products (SMP), sewage sludge, GC-MS
32	Keywords AB process, anaerobic biodegradability, dissolved organic compounds, Soluble
31	
30	proportion of acids (10% and 10%) and alcohols (16% and 10%) was observed.
29	digestion of these sludges, a greater proportion of aromatics (42% and 58%) and a lower
28	was found to be 349 \pm 1 mL CH ₄ /g VS and 238 \pm 12 mL CH ₄ /g VS, respectively. After anaerobic
27	(27.9% and 21%), alcohols (25.6% and 15%) and acids (30.2% and 15%). The methane potential
26	supernatants identified respectively 43 and 19 organic compounds consisting mainly of aromatics
25	Chromatography-Mass Spectrometry (GC-MS). The GC-MS analysis of A-stage and B-stage
24	biodegradability and low molecular weight compounds present in the supernatant using Gas
23	This manuscript presents a comparison of the A-stage and B-stage sludges in terms of anaerobic

36 In the recent years, research efforts aiming to improve energy efficiency of wastewater treatment processes in large centralized wastewater treatment plants (WWTPs) have increased. Concerns 37 over global warming impacts, energy sustainability, and biosolids generation are among several 38 39 key drivers towards the establishment of more energy-efficient WWTPs (Chai et al., 2015). The 40 biosolids management system is cost-intensive as it typically accounts for 25-60% of the total operational costs of conventional activated sludge (CAS)-based WWTPs (Canales et al., 1994; 41 Verstraete & Vlaeminck, 2011). Innovative design and treatment strategies, therefore, are required 42 43 to achieve more cost-effective and energy self-sufficient WWTPs by minimizing energy consumption while increasing its recovery. 44

46	An approach towards an energy-neutral, if not -positive, wastewater treatment process is to
47	recover the potential energy available in raw municipal wastewaters (Shizas & Bagley, 2004). A
48	well-structured strategy deploying a two-stage process, the so-called AB process, has been
49	suggested for the recovery of caloric energy content from sewage organics (Böhnke, 1977;
50	Meerburg et al., 2015; Versprille et al., 1984). The first stage is an extremely high loaded
51	biosorption stage (A-stage), which is subsequently followed by a low loaded biological stage (B-
52	stage) to ensure the removal of dissolved organics and ammonia. The A-stage treatment at the
53	entry of WWTP allows biological concentration of sewage with minimum oxidation of organics to
54	CO ₂ , and consequently producing a concentrated sludge stream to be channeled to the anaerobic
55	digester. The entrapped organics (chemical energy) can then be recovered through an efficient
56	conversion to biogas without significant energy losses (Verstraete et al., 2009). A characteristic
57	feature of the A-stage reactor is operation with high food to microorganisms (F/M) ratios, short
58	hydraulic retention times (HRTs), and short solid retention times (SRTs), to achieve high reduction
59	rate of sewage organics (Boehnke et al., 1997). Indeed, the treatment with short SRT has been
60	demonstrated to significantly improve the biodegradability of sludge in the downstream anaerobic
61	digester (Ge et al., 2013). The separation of excess sludge in the A-stage can be achieved through
62	an intermediate clarifier (henceforth referred to as 'A-stage clarifier') or dynamic membrane
63	filtration unit (Ersahin et al., 2012; Roest et al., 2012).
64	

During the biosorption process, the A-sludge retains particulate and colloidal organic substances
within the biomass matrix, and therefore leaving mainly dissolved organics in the effluents. This
would mean reduced aeration energy requirement and lower sludge production in the following B-

68	stage (Versprille et al., 1984), and therefore may lead to considerable energy savings and overall
69	reduction in biosolids generation. There is currently little information available regarding the
70	biodegradability of the excess sludge and the types of dissolved organics leaving the A and B
71	stages. Effluents from biological processes contain a wide range of complex organic compounds,
72	including soluble microbial products (SMP) and extracellular polymeric substances (EPS),
73	released during bacterial metabolism in mixed culture in bioreactors. Generally, in order to
74	evaluate the performance of biological wastewater treatment processes, only the common generic
75	parameters are measured. These include measures such as chemical oxygen demand (COD),
76	biochemical oxygen demand (BOD), mixed liquor volatile suspended solids, and total organic
77	carbon (TOC), which are done according to Standard Methods from the American Public Health
78	Association (APHA) (Eaton and Franson, 2005). It is important to clearly identify the primary
79	components of SMPs and ECPs in order to understand the fundamental mechanisms of biological
80	activity that create these compounds, and how to reduce these compounds in the effluent.
81	Preliminary results from Aquino (2004) on the identification of SMPs using GC-MS surprisingly
82	revealed long chain alkenes and alkanes, as well as some aromatic compounds such as phthalates
83	in significant concentration (low mg/L). Shen et al. (2012) showed that the concentration of SMPs
84	in wastewater treatment plants ranged roughly from 5 to 25 mg TOC/L, with the major component
85	being polysaccharides (ca. 3–18 mg/L) followed by humic substances (ca. 2–10 mg/L); while the
86	protein concentration was relatively low (<5 mg/L). The SMPs presented a broad molecular weight
87	distribution from smaller than 1 kDa to over 100 kDa. In addition, these compounds constitute the
88	main foulants in membrane bioreactors which are being used more widely around the world (Mei
89	et al., 2014).

90 Thus so far, there is virtually no report on the A-sludge's biodegradability and its comparison with
91 the B-sludge, a more conventional type of sludge, and the type of organics and their concentration
92 in each stage.

93

In this study, gas chromatography coupled with mass spectrometry (GC-MS) was used to identify 94 95 recalcitrant low molecular weight (MW) organics (<580 Da) that were not adsorbed in the A-stage 96 and appeared in the influent to the B-stage. Moreover, the recalcitrant compounds and soluble 97 microbial products (SMPs) produced in the B-stage were also identified and compared with those in the A-stage. These are the compounds that are most likely to foul the membrane when MBRs 98 99 are used in the B-stage, and that could also appear in the final effluent. There is therefore interest 100 to shed more light on these compounds, in particular from an AB process treating combined 101 industrial municipal wastewaters.

102

103 2. Material and methods

104 2.1 Reactors configuration and operating conditions

105 A pilot unit was operated with an AB process to treat real municipal wastewater from households 106 and small businesses in Singapore. The pilot plant was run in a continuous flow mode with an average wastewater flow of 1000 m³/d. It consisted of an equalization tank, 2 coarse (5 mm) rotary 107 108 drum screen units, a high-rate A-stage contact tank, a primary/A-stage clarifier, 2 fine (2 mm) 109 rotary drum screen units, and an ultrafiltration membrane bioreactor (MBR) system which 110 comprised 5 biological tanks (2 anoxic tanks and 3 aerobic tanks), 1 membrane tank and 1 111 deoxygenation tank. A simplified schematic diagram of the pilot plant is shown in Figure 1. The 112 raw influent consisted of a mixture of incoming municipal wastewaters and dewatered digested

113	sludge and was drawn through submersible pumps operating in constant flowrate mode. Initial
114	screening was subsequently performed through 5 mm perforated screen units followed by a screw
115	conveyor type grit removal system. The A-stage was designed with an SRT of 0.5 d (calculated
116	over the entire contact tank and clarifier) and a total HRT of 2 h, consisting of 0.5 h and 1.5 h for
117	the contact tank and clarifier, respectively. To protect the downstream MBR process, 2 mm fine
118	screens were provided for the removal of smaller solid particles. The following B-stage was
119	operated with a 5-h HRT in the Modified Ludzack – Ettinger (MLE) configuration with a step-feed
120	of 50% influent to the first anoxic zone and the other 50% to the second anoxic zone. A target SRT
121	of 5 d was set in order to maintain the slow-growing nitrifying organisms for N removal. Dissolved
122	oxygen (DO) concentrations were maintained at 0.3 and 1 mg O_2/L in the corresponding contact
123	tank and aerobic tanks.

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127 2.2 Physicochemical analyses

Sludge samples were taken from the pilot plant on 26th March 2015. Physico-chemical parameters 128 129 such as Total Solids (TS), Volatile Solids (VS), Total Suspended Solids (TSS), Volatile Suspended Solids (VSS) and COD concentrations were immediately analyzed in accordance with Standard 130 Methods for the Examination of Water and Wastewater (APHA, 1995). Calorific value was 131 132 determined using an oxygen bomb calorimeter (IKA, Malaysia) to measure the energy content in the sludge. The calorimeter unit consisted of a stainless steel bomb, a water jacket, an ignition unit, 133 134 a thermometer, and a mechanical stirrer. Internal volume of the stainless steel bomb was approximately 350 mL and the volume of water jacket surrounding the bomb was 2 L. The 135

136	mechanical stirrer was used to keep the water jacket uniformly mixed. After centrifugation, the
137	biomass pellet was frozen at -20°C and subsequently freeze-dried at 0.01 mbar vacuum and -45°C
138	overnight. Next, the dried samples were crushed into powder, weighed and combusted using high
139	pressure oxygen (30 bar) in bomb calorimeter. The temperature rise in the water jacket during
140	combustion was used to calculate the energy content of sludge samples. The heat capacity of the
141	bomb was determined using benzoic acid as a standard (Shizas & Bagley, 2004).
142	
143	
144	2.3 Liquid-Liquid extraction
145	Liquid-liquid extraction was performed on 100 mL of filtered supernatant (<0.45 m) using 70 mL
146	Dichloromethane (GC-MS grade, Merck). This solvent was chosen because it has been used by
146 147	Dichloromethane (GC-MS grade, Merck). This solvent was chosen because it has been used by other researchers for SMP analysis on GC-MS (Wu & Zhou, 2010). All glassware was washed
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147 148	other researchers for SMP analysis on GC-MS (Wu & Zhou, 2010). All glassware was washed with acetone prior to the procedure. A blank containing only distilled water was run along as
147 148 149	other researchers for SMP analysis on GC-MS (Wu & Zhou, 2010). All glassware was washed with acetone prior to the procedure. A blank containing only distilled water was run along as control. Mixing was provided for 3 minutes by manually inverting the extraction funnel and
147 148 149 150	other researchers for SMP analysis on GC-MS (Wu & Zhou, 2010). All glassware was washed with acetone prior to the procedure. A blank containing only distilled water was run along as control. Mixing was provided for 3 minutes by manually inverting the extraction funnel and separation of the 2 phases was then allowed for 5 minutes. Traces of water were removed by
147 148 149 150 151	other researchers for SMP analysis on GC-MS (Wu & Zhou, 2010). All glassware was washed with acetone prior to the procedure. A blank containing only distilled water was run along as control. Mixing was provided for 3 minutes by manually inverting the extraction funnel and separation of the 2 phases was then allowed for 5 minutes. Traces of water were removed by mixing the solvent phase with 2 spoons of Na ₂ SO ₄ . Solvent evaporation was then carried out at

155 2.4 Gas Chromatography – Mass Spectrometry

156 The samples (injection volume: $1 \ \mu L$) were then analyzed using a Shimadzu gas chromatograph

equipped with an autosampler and a QP2010Ultra mass spectrometry detector (Shimadzu, Japan).

158 The analytes were separated using an Rtx® -5MS column of 30m x 0.25 mm with a film thickness

159	of 0.25 μ m. The temperature program of the GC-MS oven was: 50°C, hold 7 min, rate 7°C min ⁻¹ to
160	325°C, hold 14 min. Helium was used as a carrier gas at a column flowrate of 1 mL/min. The
161	injector temperature was set at 280°C (splitless injection mode), and the MS was operated in the
162	electron impact ionisation mode (70 eV). The transfer line and ion source temperatures were 280
163	and 230°C, respectively. Scan runs were made with a range from m/z 30 to 580. The
164	chromatographic peaks were identified either by direct analysis of the mass spectrum or/and
165	comparison with the NIST11 library (National Institute of Standards and Technology,
166	Gaithersburg, MD, USA, http://www.nist.gov/srd/mslist.htm). The retention indexes were
167	calculated by the library according to alkanes standards retention times (Trzcinski & Stuckey,
168	2010). Quantification was done separately for each unknown compound using the alkane with the
169	closest retention time.

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172 2.4 Biochemical Methane Potential

Biochemical methane potential (BMP) of the A-stage and B-stage sludges was determined in batch 173 assays using an Automatic Methane Potential Test System (AMPTS II, Bioprocess Control, 174 175 Sweden). The assay was performed to examine the biodegradability of substrate subjected to the 176 anaerobic incubation through the measurement of its cumulative methane production. The AMPTS 177 reactor was seeded with anaerobic sludge which was collected from a mesophilic digester at Ulu 178 Pandan Water Reclamation Plant in Singapore. The assay was conducted at 35°C for 179 approximately 28 days. Prior to the assay, the inoculum was degassed at 35°C for one week to 180 remove the residual carbon source. Biomedium containing nutrients and vitamin was prepared in

181	accordance with Owen et al. (1979). 200 mL of inoculum, 100 mL of substrate, and 50 mL of
182	biomedium were added to each reactor which was subsequently flushed with nitrogen gas at 5 psi
183	for approximately 5 min. Batch reactor without substrate addition was used as negative control and
184	its methane production was subtracted from the methane production in the test bottles. All assays
185	were performed in duplicate. The composition of biogas was analyzed with gas chromatography as
186	previously reported (Tian et al., 2014). The percentage of biodegradability was calculated through
187	stoichiometric conversion of CH ₄ production from organic degradation as described in Speece
188	(1996). Sample preparation of the anaerobically digested sludge prior to GC-MS analysis was done
189	as described above. SMPs from the anaerobic inoculum used in the AMPTS were also analyzed
190	following the same procedure and is referred to as "AMPTS control" in results and discussion.
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192	3. Results and discussion
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3.2 GC-MS analysis of recalcitrant compounds and SMPs in AB process

205	144 peaks appeared on the chromatograph from the A-stage sludge supernatant (Supplementary
206	material), but only 43 (30%) could be identified with a match percentage greater than 80% (Figure
207	2 top left). Their concentration was not higher than 5 μ g/L, except for a few acid compounds
208	detected at a higher concentration such as dodecanoic (11.2 μ g/L), hexadecanoic (28.5 μ g/L), oleic
209	(21.1 μ g/L) and octadecanoic acids (20.5 μ g/L) (Table 2). Long chain fatty acids (LCFA) originate
210	from the degradation of fats, oils and grease present in raw sewage. LCFAs could have been taken
211	up by Poly-phosphate accumulating microorganisms (PAO) in the B stage.
212	It has been recently reported that LCFA can be used as sole carbon source for EBPR and were
213	found to enhance PAO activity (Tayà et al., 2015). It is also possible that some of the compounds
214	detected in this study by GC-MS were inhibitory or toxic to PAOs which can explain why the Bio-
215	P removal was not stable according to Qing (2015).
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216 217 218 219	Aromatic compounds were found in the low MW range (<150 Da) as well as in the high MW range (>300 Da) and bis(2-ethylhexyl) isophthalate was the largest aromatic compound in this sample with a MW of 390 Da. Overall, it was found that the compounds were mainly aromatic
216 217 218 219 220	Aromatic compounds were found in the low MW range (<150 Da) as well as in the high MW range (>300 Da) and bis(2-ethylhexyl) isophthalate was the largest aromatic compound in this sample with a MW of 390 Da. Overall, it was found that the compounds were mainly aromatic (27.9%), alcohols (25.6%) or acids (30.2%) (Figure 2 top right). The other compounds were
216 217 218 219 220 221	Aromatic compounds were found in the low MW range (<150 Da) as well as in the high MW range (>300 Da) and bis(2-ethylhexyl) isophthalate was the largest aromatic compound in this sample with a MW of 390 Da. Overall, it was found that the compounds were mainly aromatic (27.9%), alcohols (25.6%) or acids (30.2%) (Figure 2 top right). The other compounds were

- chromatograph that shows more peaks compared to the B-stage (Supplementary material). The B-
- stage chromatogram also displayed a flatter baseline which is an indication that it had fewer peaks.

Similarly, the number of identified peaks with a match percentage greater than 80% was higher in
the A-stage with 43 peaks versus 19 peaks in the B-stage supernatant. However, the B-stage
supernatant was less characterized than the A-stage supernatant with 23% of the peaks being
identified versus 30% for the A-stage supernatant.

231 The A-stage supernatant contained high molecular weight (MW) compounds with Retention Index 232 (RI) greater than 3000 and the greatest molecular weight was 534 Da for 9-Octadecenoic acid (Z)-, 233 octadecyl ester. In contrast, the B-stage supernatant did not contain any compounds with RI greater 234 than 3000 indicating that high MW compounds from the A-stage were hydrolyzed. This is relevant 235 since membrane modules (ultrafiltration) are submerged in the B-stage membrane tank and the 236 type of organics, their concentration and molecular weight will affect the fouling because they are 237 the same size as the pore diameter (Mei et al., 2014). From this study, there were clear differences 238 between the A-stage and B stage in terms of number of compounds, the type of organics, their 239 concentration and molecular weight. The A-stage is a rapid physical separation step and the 240 compounds detected in the A-stage supernatant are therefore very likely to be recalcitrant from raw 241 sewage. In contrast, the B-stage is a biological step and soluble microbial products are more likely 242 to be dominant in that sample.

Zhou et al. (2009b) investigated SMPs in the effluent of a bench scale aerobic sequencing batch
reactor treating distillery wastewater and found only 13 components by GC-MS whereas in this
study 19 were found in the B-stage supernatant; They found that alkanes and esters such as
heneicosane (19.8%), hexadecanoic acid, butyl ester (18.4%) and tetratetracontane (10.4%) were in
significant percentage of the total compounds. Alkanes such as octacosane (3.3%), hentriacontane
(2.4%), dotriacontane (2.4%) and acids such hexadecanoic acid, trimethylsilyl ester (1.2%) and
acetic acid, octadecyl ester (3.8%) were also found but in lower proportions. Alkanes were the

250	most common compounds which were found in the effluent of a SAMBR-treated solid waste
251	leachate (Trzcinski & Stuckey, 2010) and UASB effluent (Zhou et al., 2009a). These long chain
252	carbohydrates (or alkanes) and esters are frequently found in the biological treatment effluent and
253	are known to be the main components of SMP in aerobic reactors (Janga et al., 2007; Liang et al.,
254	2007). In this study, aromatic, alcohols and acids were more dominant presumably due to the more
255	complex raw wastewater and also because of the short SRT applied in the pilot plant. It is known
256	that the accumulation of SMPs becomes more pronounced at short SRTs (Liang et al., 2007).
257	
258	Overall, there was a radical shift of compounds between the A-stage and B-stage. In fact, the B-
259	stage supernatant consisted of completely different compounds, except three: flutolanil (a common
260	pesticide), triacetine and n-Nonadecanol-1, and their concentrations decreased compared to the A-
261	stage supernatant, showing that indeed some compounds could be biodegraded in the process or
262	removed through adsorption to the B-stage sludge. The new compounds in B-stage were either
263	SMPs or biodegradation end-products of residual COD in the soluble phase.
264	The B-stage supernatant contained very diverse compounds such as aromatics (21%), alcohols
265	(15%), acids (15%), alkanes (10%), alkenes (15%), aldehydes (10%), amide (5%) and ester (5%)
266	as shown in Figure 2 (bottom, right).
267	
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269	3.3 Anaerobic Biodegradability
270	The cumulative methane production is shown in Figure 3 where it can be seen that 349 ± 1 mL
271	CH ₄ /g VS and 238 \pm 12 mL CH ₄ /g VS were produced from the A-stage and B-stage sludges,

respectively, showing the greater biodegradability (+47%) of the A-stage sludge. From the COD

273	mass balance and considering the theoretical COD equivalence of 395 mL CH ₄ per gram COD
274	(Speece, 1996), it was derived that 53% and 42% of the COD in A-stage and B-stage sludges were
275	converted to methane gas, respectively.
276	
277	Moreover, the respective methane content in the biogas were 64% and 54% showing the higher
278	energy content of the biogas obtained from the A-stage sludge. This is consistent with the calorific
279	value given in Table 1 which confirms that the A-stage yielded sludge with a greater carbon
280	content and biodegradability potential compared to the more conventional aerated waste activated
281	sludge. This indicates the capacity of the AB system to rapidly capture the carbon from raw
282	sewage and channel it to the existing anaerobic digester to increase energy production.
283	
284	3.4 GC-MS analysis of recalcitrant compounds and SMPs after anaerobic digestion (AD)
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286	After anaerobic digestion (AD) tests, SMPs and recalcitrant compounds in the supernatant of the
287	digested sludges were also analyzed using GC-MS. A few peaks (identified by ** in
288	Supplementary Table S2) were also found in the inoculum used in the anaerobic biodegradability
289	test, for instance p-cresol which was detected in relatively high concentration. It was found that the
290	number of peaks decreased from 144 to 124 in the digested A-stage sludge (Figure 4 top left). This
291	shows that some compounds were anaerobically degraded to methane, CO ₂ or converted to new
292	biomass while new molecules appeared as end-product of the anaerobic process or SMPs produced
293	by anaerobic metabolism. Among these 124 peaks, only 31 (or 25%) were identified and only 6
294	were in common before and after the anaerobic biodegradability test (identified by *** in
295	

from the anaerobic metabolism. One of them was oleic acid and its concentration had decreased from 21.14 μ g/L before AD to 1.4 μ g/L after AD. However, the concentration of some of these increased through the anaerobic digestion test which could be the result of biological degradation of colloids and large molecules in the sludge sample.

300

The number of low molecular weight compounds (with RI lower than 1200) was 3 before AD
(Table 2), and this increased to 10 after AD (Supplementary Table S2) showing that high
molecular weight compounds were hydrolyzed to low molecular weights compounds during
anaerobic digestion tests. The number of compounds with RI>3000 (chain with more than 30
Carbons) was 5 before AD and 4 after AD. In both B-stage supernatants (before AD in Table S1
and after AD in Table S3) no compounds with RI>3000 was found showing a different molecular
weight distribution than in A-stage.

308 It was observed that the distribution of compounds also changed with a significantly greater 309 proportion of aromatic compounds: 42% after AD versus 28% before AD. This is because 310 aromatic compounds are generally more recalcitrant and therefore represent a major fraction of 311 residual compounds after AD. All the aromatic compounds detected after AD were smaller than 312 206 Da which is different than before AD where they were found in the low (<150 Da) and high ranges of MW (>300 Da). From the results of the A-stage sludge, it can be added that aromatic 313 314 biodegradation end-products and SMPs were all smaller then about 200 Da (aromatics are shown 315 with † in Supplementary S2). Alcohols and acids were secondary compounds with 16% and 10% 316 of the total number of compounds, respectively (Figure 4 top right). These proportions were 25.6% 317 and 30% in the sample before AD (Figure 2 top right).

In conclusion, there were fewer compounds after AD with a higher proportion of aromatics and alower proportion of acids and alcohols.

320

321 In the B-stage supernatant the total number of compounds decreased from 84 to 76 before and after 322 AD, respectively, while the number of identified peaks remained 19 (Figure 4 bottom left). When 323 comparing before and after AD, only 2 compounds (2,4,7,9-Tetramethyl-5-decyn-4,7-diol and 324 propanoic acid, 2-methyl-, 1-(1,1-dimethylethyl)-2-methyl-1,3-propanediyl ester) were common in 325 both samples indicating that there was a radical shift of compounds during anaerobic digestion of 326 the B-stage sludge. The number of compounds with RI lower than 1200 was 2 before AD and 327 became 10 after AD showing that hydrolysis of larger molecular weight compounds was taking 328 place during the BMP tests (Supplementary materials Tables S1 and S3). 329 The proportion of various compounds significantly changed during the AD process. The 330 percentage of aromatic compounds increased to 58 % while the percentage of alcohols and acids 331 decreased to 10% each (Figure 4 bottom right). The further stabilization in the B stage due to the 332 process configuration was confirmed with a lower number of compounds compared to the A-stage 333 supernatant (19 versus 31) and also by a higher degree of aromaticity: 58% versus 42%. This was 334 expected since the SRT is longer in the B-stage (5 days) than in the A-stage (0.5 days) and 335 retention of bacteria by the membrane in the B-stage can also contribute to a better biodegradation 336 of SMPs. The role of the A-stage is also to provide protection to the B-stage and buffer any 337 organic shock that may occur. The higher number of compounds in the A-stage compared to the B-338 stage showed that indeed the process configuration allowed for fewer contaminants ending up in 339 the B-stage. This provides protection for the biological process in the B-stage as fewer toxic or 340 inhibitory compounds were detected.

342	In this study aromatics were detected in both the aerobic sludges (from A-stage and B-stage) and
343	the anaerobically digested sludges, but the degree of aromaticity was greater in the anaerobically
344	digested sludge.
345	The concentrations were typically less than 5 μ g/L which is too low to explain the residual SCOD
346	given in Table 1: 153 mg/L and 38 mg/L in the A-stage and B-stage sludge, respectively.
347	This is because the use of GC-MS is limited to the identification of non-polar, volatile and
348	thermostable compounds and many peaks in the chromatograms could not be identified.
349	Techniques such as LC-MS or Matrix Assisted Laser Desorption Ionization-Time of Flight-Mass
350	Spectrometry (MALDI-ToF-MS) would certainly shed more light on the nature of the high MW
351	compounds that were not detected and could explain the residual COD in the effluent.
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356	4. Conclusions
357	This study showed that the supernatant of both A-stage and B-stages sludges contained aromatics
358	(27.9% and 21.1% of identified compounds), long chain alkanes (7% and 10.5%), alcohols (25.6%
359	and 15.8%), acids (30.2% and 15.8%) and esters (2.3% and 5.3%). More methane could be
360	produced from the A-stage sludge (349 \pm 1 mL CH ₄ /g VS) compared to the B-stage sludge (238 \pm 12
361	mL CH ₄ /g VS). After anaerobic digestion of these sludges, the total number of compounds
362	detected by GC-MS was lower, and there was a greater proportion of aromatic compounds (42%
363	and 58%).

365	5. Acknowledgments
366	The authors would like to thank the Public Utilities Board (PUB) of Singapore for financial
367	support of this research.
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369	6. References
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