



# THE EFFECT OF HYDROGEN ON THE PROPERTIES OF DUPLEX STAINLESS STEEL

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## Introduction

Duplex stainless steels with a two phase austenite/ferrite microstructure have excellent mechanical properties and a high corrosion resistance. However, they are susceptible to hydrogen induced degradation. In this work, the interaction between hydrogen (H) and a 50/50 duplex stainless steel was investigated.

### Material characterisation Duplex stainless steel type UNS S32205 Mo Mn Other Cu 0.24, P 0.02, S 0.005 22.45 2.63 1.81 0.38 5.31 Wt% 0.013 ND Ferrite 150 μm

Austenite fraction = 50.0 ± 1.5 %

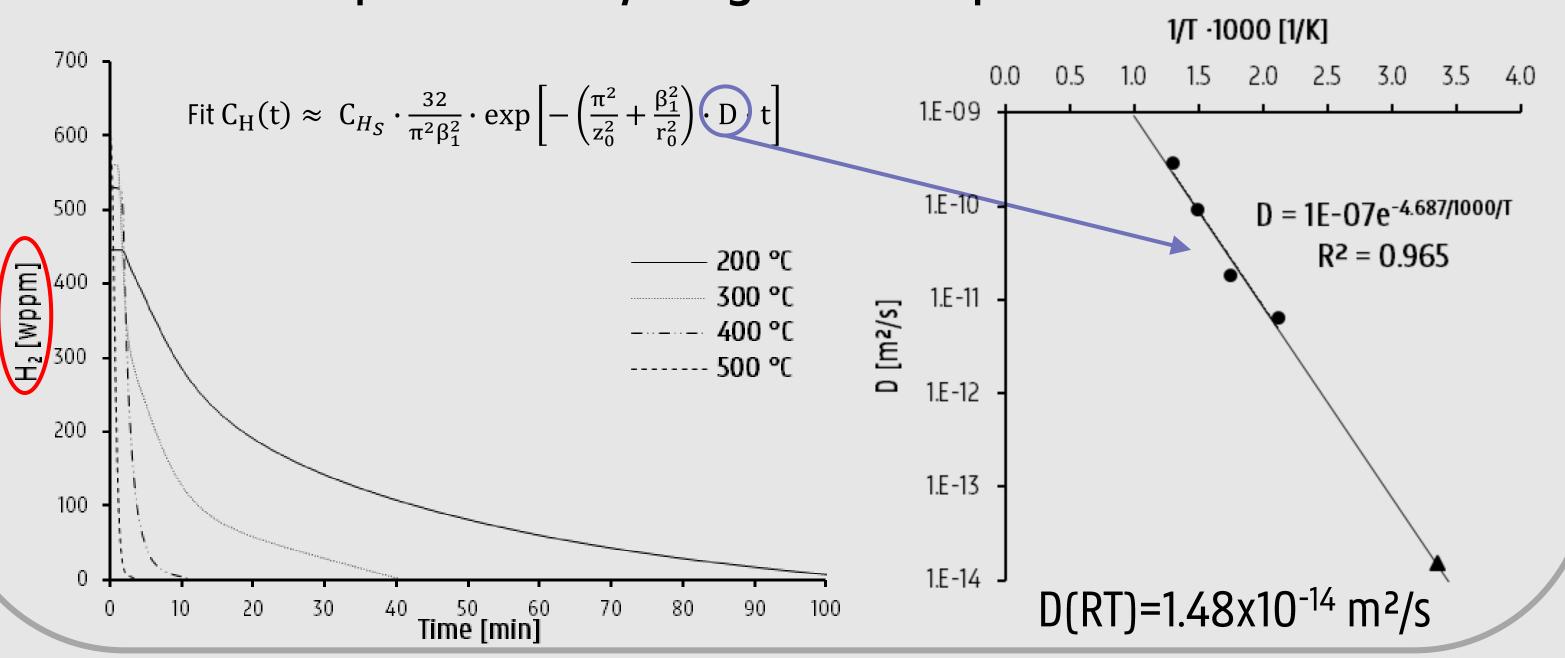
Austenite

# Hydrogen/metal interaction

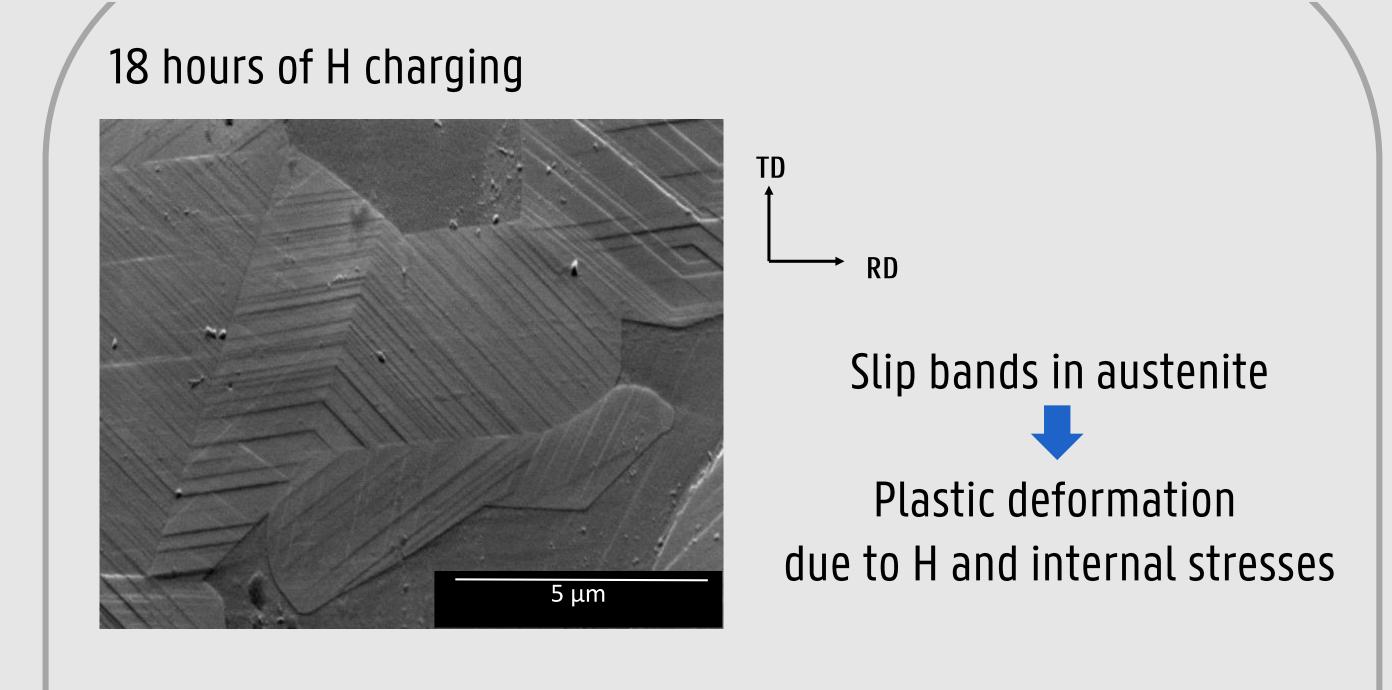
0.8 mA/cm² in 0.5M H<sub>2</sub>SO<sub>4</sub> and 1 g/l thiourea

# Total hydrogen content (melt extraction) 0.3 mm 1.5 mm $C_{H_S} = 680 \text{ wppm}$ $D_{app} = 1.64 \times 10^{-14} \text{ m}^2/\text{s}$ $C_{harging time [days]}$ Charging time [days]

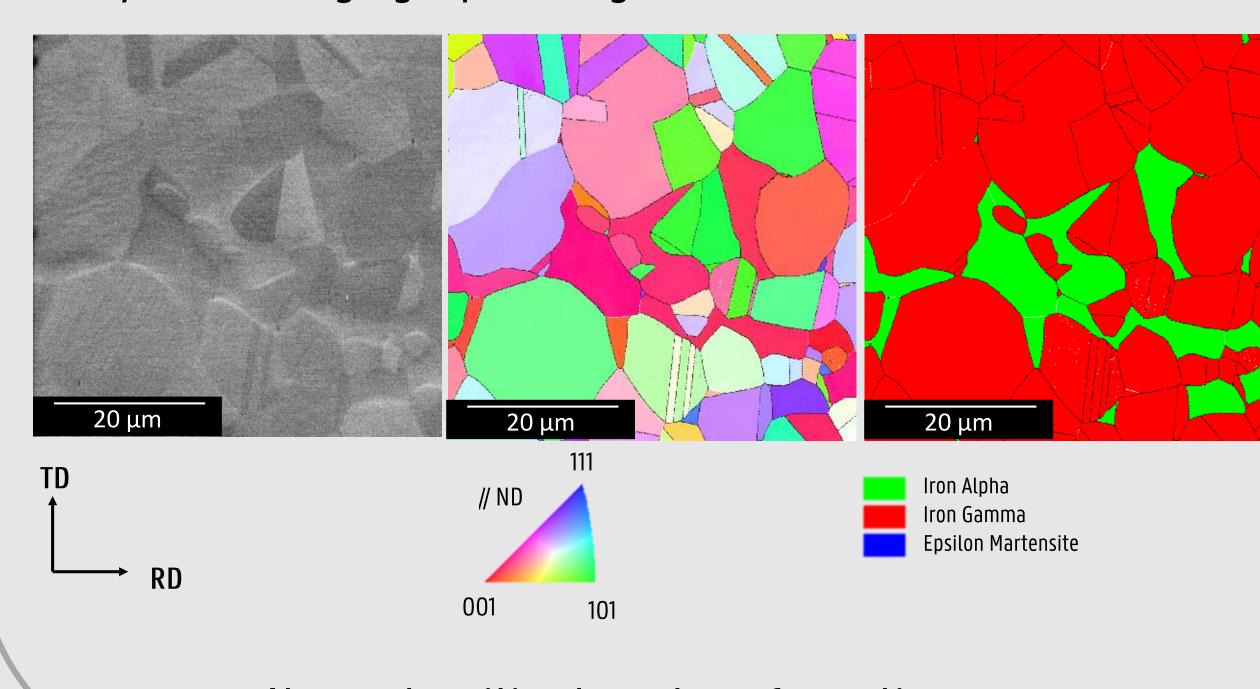
## Constant temperature hydrogen desorption tests



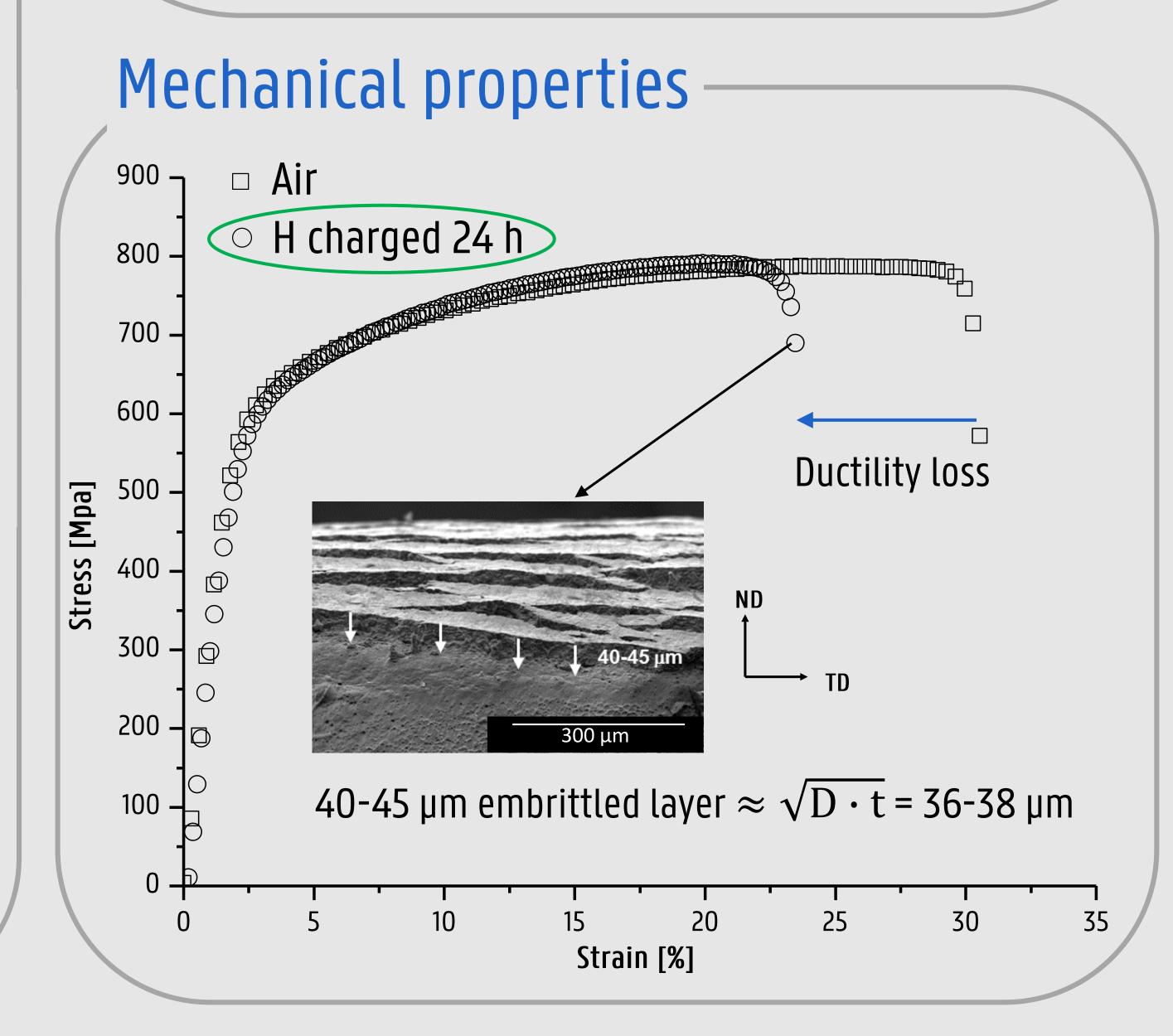
# µ-structural changes during H charging



7 days of H charging + polishing



No martensitic phase transformations



## Conclusions

- A saturation level of 680 wppm was obtained after approximately 10 days of charging
- A diffusion coefficient of 1.64x10<sup>-14</sup> m²/s was found by melt extraction; 1.48x10<sup>-14</sup> m²/s was found from hydrogen desorption tests
- Slip bands were generated during hydrogen charging
- Correlation between H affected zone and H diffusion distance was found

