

Carbon Metabolism of Methylotrophic Methanogens and Asgard Archaea in Marine Sediments

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Summary	1
Zusammenfassung	3
Chapter 1	5
Introduction	5
1.1. Carbon utilization in marine sediments	5
1.2. Carbon metabolisms in methanogenesis	9
1.2.1. Hydrogenotrophic methanogens	9
1.2.2. Methylotrophic methanogens	10
1.2.3. Acetoclastic methanogens	11
1.3. Carbon metabolisms in other uncultured archaea	
1.3.1. Archaeal diversity in the Helgoland Mud sediments	13
1.3.2. Carbon fixation	
1.3.3. Alkane and fatty acids degradation	14
1.3.4. Fermentation and organic polymer degradation	
1.4. Identification of active microorganisms by stable isotope probing	
1.4.1. Nucleic acid-SIP	
1.4.2. Lipid-SIP	
1.5. Objectives and structure of the thesis	
1.6. Reference	
Chapter 2	
DNA and RNA stable isotope probing of methylotrophic methanogenic archaea	
Abstract	
2.1. Introduction	
2.2. Materials	
2.2.1. Sediment slurry incubations	
2.2.2. Media	
2.2.3. Nucleic acid extraction	
2.2.4. DNA- and RNA-SIP	
2.3. Methods	
2.3.1. SIP incubations	
2.3.2. Nucleic acids extraction	
2.3.3. Density separation of isotopically labeled nucleic acids	
2.3.4. Anticipated results	
2.3.5. Experimental run time	
2.4. Notes	
2.5. Figures	
2.6. Table	
2.7. References	

Contents

Chapter 3	55
CO ₂ conversion to methane and biomass in obligate methylotrophic methanogens in ma	
sediments	
Abstract	
3.1. Introduction	58
3.2. Materials and Methods	59
3.2.1. Sediment incubation setup for SIP	59
3.2.2. Pure culture setup	60
3.2.3. Slurry incubations inoculated with M. methylutens	
3.2.4. Gas analysis	61
3.2.5. Nucleic acids extraction, quantification and DNase treatment	61
3.2.6. Isopycnic centrifugation, gradient fractionation and reverse transcription	61
3.2.7. Quantitative PCR (qPCR)	62
3.2.8. Sequencing and bioinformatics analysis	62
3.2.9. Lipid analysis	62
3.2.10. δ^{13} C calculation	62
3.3. Results	63
3.3.1. Methylotrophic methanogenesis and increase in methanogenic archaea	63
3.3.2. Carbon assimilation into RNA and identification of metabolically active archaea.	64
3.3.3. Methane formation from DIC during methylotrophic methanogenesis	65
3.4. Discussion	66
3.4.1. Carbon assimilation by methylotrophic methanogens in sediment incubations	66
3.4.2. CO ₂ reduction to methane by obligate methylotrophic methanogens	68
3.5. Acknowledgments	70
3.6. Conflict of interest	70
3.7. Data availability	70
3.8. Table	70
3.9. Figures	72
3.10. Supplementary figures	78
3.11. Reference	87
Chapter 4	93
Asgard archaea are active in marine sediment carbon cycling	93
Abstract	95
4.1. Introduction	96
4.2. Results and discussion	97
4.2.1. Asgard archaea are diverse and ubiquitous	97
4.2.2. Asgard archaea fix CO_2 and degrade organic polymers based on stable isotope pro	obing
4.2.3. Potential metabolic capabilities of Asgard archaea revealed by metagenomics and	
metatranscriptomics	98

4.2.4. Alkane metabolism	.100
4.3. Methods	.102
4.3.1. Asgard archaea 16S rRNA gene dataset construction	. 102
4.3.2. Phylogenetic position and distribution of Asgard archaea 16S rRNA gene sequences	103
4.3.3. SIP incubations	. 103
4.3.4. Isopycnic centrifugation, gradient fractionation and sequencing	. 104
4.3.5. Clone library construction	. 105
4.3.6. Quantitative PCR (qPCR)	. 105
4.3.7. Sediment sample collection and processing	. 106
4.3.8. Metagenomic assembly, genome binning and gene annotation	. 106
4.3.9. Phylogenetic analyses of Asgard archaea and functional genes	. 107
4.3.10. Metabolic pathway construction	. 107
4.3.11. Molecular modelling and dynamics simulation	. 108
4.3.12. Evolutionary analysis	.108
4.3.13. Data availability	. 109
4.4. Acknowledgements	. 109
4.5. Author contributions	. 109
4.6. Author Information	. 109
4.7. Figures	.110
4.8. Supplementary tables	.114
4.9. Supplementary figures	. 123
4.10. References	.134
Chapter 5	.141
General Discussion	. 141
5.1. Carbon metabolism in methylotrophic methanogens and Asgard archaea	.144
5.1.1. Methyl metabolism	.144
5.1.2. Inorganic carbon assimilation	. 145
5.2. Carbon incorporation in other archaea	. 146
5.2.1. Carbon metabolism in Bathyarchaeota	. 146
5.2.2. Carbon metabolism in anaerobic methanotrophs (ANMEs)	. 147
5.2.3. Carbon metabolism in Thermoplasmata and potential acetate metabolism in archaea	. 148
5.3. Nucleic acid-SIP for detecting archaea-high sensitivity meets low activity	. 149
5.4. Carbon fixation in bacteria in anoxic marine sediments	. 150
5.5. Reference	.152
Acknowledgements	. 159

Abbreviation

AAI: amino acid identity alkB: alkane hydroxylase gene ANMEs: anaerobic methanotrophs AQDS: anthraquinone-2,6-disulfonic acid AR: archaeol; AR-OH: hydroxyarchaeol ASW: artificial sea water CBB cycle: Calvin-Benson-Bassham cycle CH₃-H₄MPT: methyl-tetrahydromethanopterin CsCl: cesium chloride CsTFA: cesium trifluoroacetate DIC: dissolved inorganic carbon EMP: Embden-Meyerhof-Parnas pathway EtOH: ethanol GB: gradient buffer GC-c-IRMS: gas chromatography combustion isotope ratio mass spectrometry LPA: linear polyacrylamide MAGs: metagenome-assembled genomes MBGD: marine benthic group D mcrA: methyl-coenzyme M reductase subunit alpha MeOH: methanol MZ: methanogenic zone PCI: phenol:chloroform:isoamylalcohol PEG: polyethylene glycol RMSD: root mean square deviation RubisCO: ribulose-1,5-bisphosphate carboxylase/oxygenase SIP: stable isotope probing SMTZ: sulfate-methane transition zone SRZ: sulfate reduction zone TCA cycle: tricarboxylic acid cycle THF-WL: tetrahydrofolate dependent Wood-Ljungdahl THMPT-WL: tetrahydromethanopterin dependent Wood-Ljungdahl TIC: total inorganic carbon TNS: tris-NaCl-sodium dodecyl sulfate TOC: sulfate-methane transition zone

WL: Wood-Ljungdahl pathway

Summary

Carbon is the central element of life, as it is involved in building up of biological constituents and energy metabolisms in the cell. Archaea, - the most recently recognized domain of life - hold a crucial phylogenetic position in the evolution of life, but for most archaeal phyla, little is known about their role and activity in carbon metabolism. Archaea inhabit a variety of environments such as soils, sediments, sea water, and the guts of animals. Specifically in marine sediments, Thaumarchaeota, Euryarchaeota, Bathyarchaeota, Woesearchaeota and Asgard archaea are commonly found in archaeal communities.

Methanogens affiliated to Euryarchaeota are well-known players in carbon metabolism, i.e., acetoclastic, hydrogenotrophic, and methylotrophic methanogenesis. Based on pure culture studies and genomic evidence, significant amounts of the biomass of methylotrophic methanogens growing on methyl substrates is derived from inorganic carbon. However, the *in situ* activity of these methanogens in carbon assimilation is unclear as the large inorganic carbon pool in marine sediment potentially affects carbon utilization patterns. To address this hypothesis, we initially applied nucleic acid stable isotope probing (SIP) to detect methylotrophic methanogens in marine sediment incubations. SIP results showed that ¹³C-labeled dissolved inorganic carbon (DIC) is necessary to identify methylotrophic methanogens, as illustrated by the nucleic acid synthesis pathway in these methanogens that 70-80% of carbon stems from DIC rather than methanol.

In parallel, lipid-SIP suggested that DIC contributed to more than 60% from incubations with sediment from the sulfate reduction zone (SRZ), i.e., 20% higher than expected from lipid synthesis pathway. We further unexpectedly found that up to 12% methane was formed from DIC in autoclaved slurry incubations inoculated with the marine methylotrophic methanogen *Methanococcoides methylutens*. Similarly, methane formation from CO_2 during methylotrophic methanogenesis was also observed with SRZ sediment incubations. In the same sediment incubations a higher amount of inorganic carbon was incorporated into lipids than expected, indicating that more DIC was assimilated into biomass than expected. Thus, the CO_2 conversion to methane and biomass may play an important role in marine sediments.

In the most recently discovered super phylum of the Archaea, the Asgard archaea might hold the key to understand the evolutionary origin of eukaryotes. Unlike methanogens, however, the diversity, carbon metabolism and the activity of Asgard archaea in marine sediments are still unknown. In this study, five new groups of Asgard archaea namely Kariarchaeota, Balderarchaeota, Hodarchaeota, Lagarchaeota and Gerdarchaeota are reported. In experiments with ¹³C-DIC, potential electron donors and electron acceptors, subgroup of Asgard archaea i.e., Lokiarchaeota was detected in the heavy SIP fractions from the incubations amended with organic polymers or sulfur, suggesting their activities of carbon fixation, organic polymers (cellulose, lignin and humic acid) degradation and sulfur

metabolism. Furthermore, metagenomes were sequenced from heavy fractions of DNA-SIP samples obtained in the aforementioned experiments and from DNA extracted from mangrove sediments in the southeast coast of China. These metagenomes indicate that Asgard archaea harbor pathways of inorganic carbon fixation and degradation of cellulose, protein, short-chain and medium-chain alkane as well as assimilatory sulfate reduction. Crucially, the methyl coenzyme M reductase genes found in Helarchaeota have extended the potential of short-chain hydrocarbon oxidation to the Asgard archaea in this study. Overall, these findings illustrate that Asgard archaea actively utilize organic and inorganic carbon at the same time in mixotrophic fashion, which might play critical roles in carbon cycling of marine sediments.

In particular, the successful detection of methylotrophic methanogens and Asgard archaea in marine sediments by nucleic acid-SIP with ¹³C-DIC suggested a crucial role of inorganic carbon in carbon metabolisms of these archaea. Given that many archaea harbor the acetyl-CoA associated carbon fixation pathway, my findings indicate that inorganic carbon assimilation might be ubiquitous in archaea when supply or availability of organic carbon are not sufficient in marine sediments.

Zusammenfassung

Kohlenstoff ist für alles Leben auf der Erde von zentraler Bedeutung: Er bildet das Grundgerüst aller zellulärer Komponenten und ist am Energiestoffwechsel beteiligt. Archaeen – die zuletzt entdeckte Domäne irdischen Lebens – besetzen eine Schlüsselposition innerhalb der organismischen Evolution, aber über die physiologische Rolle der meisten Stämme der Archaeen im Kohlenstoffkreislauf ist noch sehr wenig bekannt. Archaeen kommen in einer Vielzahl von Habitaten vor, beispielsweise in Böden, Sedimenten, aquatischen Lebensräumen oder im gastrointestinalen Trakt vieler Metazoa.

Innerhalb dieser wichtigen Gruppe von Organismen sind methanogene Euryarchaeota als Teilnehmer des Kohlenstoffkreislaufes gut bekannt, zum Beispiel in der acetoklastischen, hydrogenotrophen und methylotrophen Methanbildung. Durch Studien an Reinkulturen und Genomanalysen weiß man, dass methylotrophe Methanogene einen nicht unerheblichen Anteil ihrer Biomasse aus anorganischen Kohlenstoffquellen bestreiten, wenn sie auf methylierten Verbindungen wachsen. Jedoch ist nicht bekannt, ob diese Stoffwechselwege auch *in situ* beschritten werden; allerdings ist zu vermuten, dass die großen Mengen anorganisch gebundenen Kohlenstoffs in der Umwelt auch Auswirkungen auf die Strategie der Kohlenstoffaufnahme dieser Mikroorganismen haben. Um diese Hypothese zu prüfen, benutzten wir eine Technik zur stabilen Isotopenbeprobung (SIP). Zunächst wurden marine Sedimente mit ¹³C-markierten Substraten inkubiert, um methylotrophe Methanogene zu detektieren. Dabei zeigte sich, dass die Markierung mit gelöstem anorganischen Kohlenstoff (engl. DIC) notwendig ist, um dieses Ziel zu erreichen, da zwischen 70 – 80% des in den Nukleinsäuren dieser Organismen gebundenen Kohlenstoffs aus anorganischen Quellen stammt – nicht aus Methanol.

Parallel dazu konnte in Inkubationen mit Sedimenten aus der Sulfatreduktionszone (SRZ) mit Hilfe der Isotopenanalyse der zellulären Fettsäuren (Lipid-SIP) gezeigt werden, dass der gelöste anorganische Kohlenstoff zu mehr als 60% des in den Fettsäuren gebundenen Kohlenstoffs beiträgt, also etwa 20% mehr als durch den Fettsäuresyntheseweg vorhergesagt. Wir konnten auch die unerwartete Entdeckung machen, dass in Inkubationen mit inokuliertem *Methanococcoides methylutens*, einem methylotrophen Methanbildners, mit autoklaviertem Sediment 12% des hergestellten Methans aus dem DIC-Reservoir stammt. In ähnlicher Weise konnte die Methansynthese aus CO₂ auch während der eigentlich methylotrophen Methanogenese in SRZ-Sedimenten beobachtet werden. Hierbei wurde ebenfalls mehr anorganischer Kohlenstoff in die zellulären Fettsäuren, also Biomasse, eingebaut als erwartet. Daraus folgend lässt sich eine wichtige Rolle der CO₂-Fixierung und -Konversion zu Methan und Biomasse in marinen Sedimenten vermuten.

Der erst kürzlich entdeckte archaeale Superstamm (engl. super phylum) der Asgard-Archaeen könnte eine Schlüsselrolle beim evolutionären Übergang von den Pro- zu den Eukaryonten gespielt haben. Im Gegensatz zu den Methanogenen jedoch ist sehr wenig über Diversität, Physiologie und Aktivitäten dieser Gruppe bekannt. In der vorliegenden Arbeit werden 5 neue Gruppen innerhalb dieses Superstammes eingeführt, namentlich Kariarchaeota, Balderarchaeota, Hodarchaeota, Lagarchaeota und Gerdarchaeota. In Experimenten mit ¹³C-markiertem DIC und verschiedenen potentiellen Elektronendonoren und -akzeptoren wurden Asgard-Archaeen, zum Beispiel Lokiarchaeota, in den schweren Nukleinsäurefraktionen derjenigen Inkubationen gefunden, die unter Zugabe von organischen Polymeren oder Schwefel abliefen. Dies zeigt eine ökophysiologische Funktion dieser Organismen im Abbau von organischen Polymeren (z.B. Lignin, Cellulose und Huminsäuren), in der Kohlenstofffixierung und im Schwefelstoffwechsel an. Weiterhin wurden Metagenome aus den schweren Fraktionen dieser Experimente und aus DNA aus Mangrovensedimenten von der chinesischen Südostküste gebildet. Die Analyse dieser Metagenome zeigte, dass Asgard-Archaeen über das genetische Potential verfügen, um Cellulose, Proteine, kurz- und mittelkettige Alkane zu verstoffwechseln, anorganischen Kohlenstoff zu fixieren, und assimilatorische Sulfatreduktion zu betreiben. Besonders wichtig ist die Detektion von Genen für die Methyl-Coenzym-M-Reduktase in Helarchaeota, was das funktionelle Spektrum der Asgard-Archaeen um die Fähigkeit der Oxidation kurzkettiger Kohlenwasserstoffe erweitert. Insgesamt konnte gezeigt werden, dass Asgard-Archaeen mixotroph sowohl organische als auch anorganische Kohlenstoffverbindungen gleichzeitig aufnehmen können, was für den Kohlenstoffkreislauf in marinen Sedimenten eine wichtige Rolle spielen könnte.

Besonders die Entdeckung von methylotrophen und Asgard-Archaeen in marinen Sedimenten mit Hilfe der Beprobung mit ¹³C-DIC deutet auf eine wesentliche Funktion von anorganischen Kohlenstoffverbindungen im Stoffwechsel dieser Archaeen hin. Zieht man in Betracht, dass viele Archaeen über den Acetyl-CoA-assoziierten Kohlenstofffixierungsweg verfügen, zeigen meine Befunde, dass anorganischer Kohlenstoff ubiquitär von allen Archaeen assimiliert wird, wenn der Kohlenstoffbedarf nicht aus organischen Quellen gedeckt werden kann.

Chapter 1

Introduction

1.1. Carbon utilization in marine sediments

Marine sediments are the sink of particles in suspension settling down from the water column. These particles contain aggregates of organic compounds, which mainly originate from terrestrial or aquatic primary production as dominant autotrophic process (Burdige 2005; Schlünz and Schneider 2000; Schubert and Calvert 2001). For our study site at Helgoland Mud Area in the German Bight of North Sea (Fig. 1), the input of fresh water provided relatively rich organic carbon and minerals with high sedimentation rates of above 13 mm/year prior to 1250 A.D and 7.7 mm/year during the last century (Dominik et al., 1978; Hebbeln et al., 2003). In contrast, sedimentation in the deep sea sediments occurred over millions of years (Roy et al., 2012; Zhou and Kyte 1992). However, these sedimentations are responsible for organic carbon input and feed microorganisms in marine sediments.

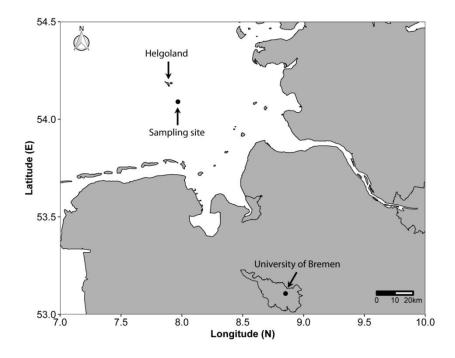


Fig. 1 Sampling site in the Helgoland Mud Area

(The map data was obtained from the Database of Global Administrative Areas (GADM) (http://www.gadm.org/country) and the figure was made in R3.5.2)

Global deposition of organic carbon in the ocean reaches 169×10^{12} g C yr⁻¹ of which ~75% organic carbon originates from continental shelf, (Smith et al., 2015), providing a large amount of organic carbon as C source for microbial metabolism. The composition of organic compounds in marine sediments is complex, mainly comprising of carbohydrates, aromatics (humic substances and lignin), aliphatic compounds, lipids, protein and amino acids (Arndt et al., 2013; Carter and Mitterer 1978;

Colombo et al., 1996; Goni et al., 1997; Guerzoni and Rovatti 1987; Mayer et al., 1986; Oni et al., 2015b; Prahl et al., 1994; Smith et al., 2015; Volkman et al., 1987). These organic compounds harbor versatile functional groups, indicating a variety of degradability by microorganisms (De Leeuw and Largeau 1993). Thus, this organic carbon can serve as electron donor and carbon source through the sediment core and likely shapes microbial communities and activities. In the upper layers, microorganisms preferentially take up easily degradable organic compounds (Wakeham et al., 1997), which is coupled with the reduction of oxygen, nitrate, manganese oxides, iron oxides and sulfate. As general rule, oxygen and nitrate is quickly depleted, followed by metal (manganese and iron oxides) and sulfate reduction, according to the availabilities and redox potentials of electron acceptors from upper to deeper sediment layers (Jørgensen and Kasten 2006). For example in the Helgoland Mud sediments, the concentration of dissolved manganese quickly increases from the top of sediment core (~2 cm), followed by Fe^{2+} (~4 cm) (Fig. 2). Sulfate depletion is observed from approximately 10 - 86cm below subsurface (Fig. 2 and Fig. 3). The dynamics of the geochemical profile in marine sediments reflects that these electron acceptors fuel microbial metabolisms by organic carbon degradation. In the deeper sediment below 100 cm in the Helgoland Mud sediments, the occurrence of methane indicates organic carbon that can be consumed and that fermentation intermediates are used for methanogenesis (Fig. 3). Furthermore, diffusion of methane from the lower sediment to the upper layers feed methane oxidizers when electron acceptor (sulfate) is available. Therefore, in the Helgoland Mud area, although the content of total organic carbon fluctuates through sediment core (Fig. 3), recalcitrance such as aromatic compounds and unsaturated compounds with low hydrogen to carbon ratios accumulates in deeper sediments (Oni et al., 2015b), resulting in a low degradability in deeper sediments.

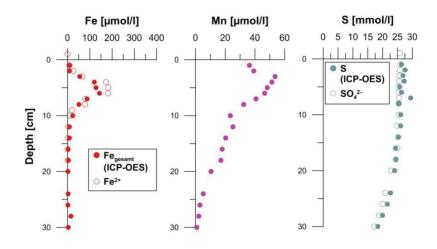


Fig. 2 Geochemical profiles of dissolved iron, manganese and sulfur in the Helgoland Mud sediment in the upper sediment layers (down to 30 cm depths, data were obtained from multicorer sampling). Sediment cores were collected in 2015 during the RV HEINCKE cruise HE443. This data is unpublished and kindly provided by Dr. Susann Henkel (AWI, Bremerhaven).

Anaerobic degradation of organic matter in marine sediments involves fermentation, sulfate reduction, iron reduction and manganese reduction (Canfield et al., 1993). Organic compounds such as carbohydrates can be fermented to short-chain fatty acids or H_2 , and the terminal degraders (methanogens) will utilize these carbon sources or electron donors for methane formation when electron acceptors are unavailable (Schink 1997; Tromp et al., 1995). In the presence of electron acceptors, long-chain fatty acids and recalcitrant compounds can be used for microbial growth anaerobically (Annweiler et al., 2000; Coates et al., 1995; Egland et al., 1997). Consequently, turning over of organic carbon in deep marine sediments leads to the accumulation of methane and CO_2 .

Besides organic carbon, CO_2 is an important carbon source as suggested by presence of hydrogenotrophic methanogens in deeper layers (Lazar et al., 2011; Zeleke et al., 2013), since CO_2 fixation by hydrogenotrophic methanogens represents an input of inorganic carbon into biosphere. Inorganic carbon is a large pool (up to 40 mM (Zhuang et al., 2018)) in deeper sediment (Fig. 3) including CO_2 , mineral inorganic carbon and dissolved inorganic carbon (DIC). Except for methanogens, DIC is also an important carbon source for autotrophs and mixotrophs in sediments such as sulfur-oxidizing *Gammaproteobacteria*, anaerobic methanotrophs (ANMEs) and Bathyarchaeota via the ribulose-1,5-bisphosphate carboxylase/oxygenase (RubisCO) pathway or Wood-Ljungdahl pathway (Dyksma et al., 2016; Kellermann et al., 2012; Yu et al., 2018).

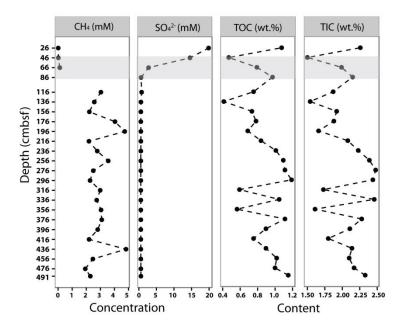


Fig. 3 Geochemical profiles of methane, sulfate, total organic carbon (TOC) and total inorganic carbon (TIC) in the Helgoland Mud sediment from the gravity corer. Sediment was collected during the RV HEINCKE cruise HE443. Gray bar denotes the sulfate-methane transition zone (SMTZ). This data is unpublished and is kindly provided by Dr. Susann Henkel (AWI, Bremerhaven).

Additionally, CO_2 fixation during pyruvate formation from acetyl-CoA for gluconeogenesis and anaplerotic reactions for replenishing citric acid cycle intermediates elevate the contribution of inorganic carbon to biomass substantially in both archaea and bacteria (Perez and Matin 1982; Deppenmeier et al., 2002).

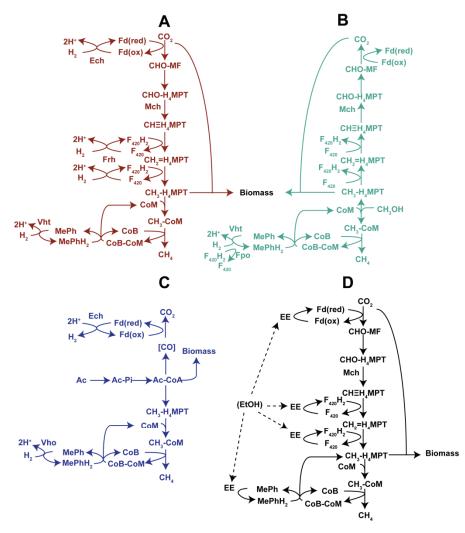


Fig. 4 Methanogenic pathways in *Methanosarcina barkeri* including hydrogenotrophic (H₂/CO₂; A), methylotrophic (e.g., methanol; B), acetoclastic (acetate; C) methanogenesis and syntrophic methane production via extracellular electron transfer (D).

The pathways were constructed according to previous studies (Guss et al., 2005; Rotaru et al., 2014; Thauer 1998). Abbreviations: Ech, ferredoxin-dependent hydrogenase; Frh, F_{420} -dependent hydrogenase; Vho, methanophenazine-dependent hydrogenase; Fpo, F_{420} dehydrogenase; CHO-MF, formyl-methanofuran; CHO-H₄MPT, formyl-tetrahydromethanopterin; CH=H₄MPT, methenyl-tetrahydromethanopterin; CH₂=H₄MPT, methyl-ne-tetrahydromethanopterin; CH₃-H₄MPT, methyl-tetrahydromethanopterin; CH₃-H₄MPT, methyl-tetrahydromethanopterin; CH₃-GoM, methyl-coenzyme M; CoM, coenzyme M; CoB, coenzyme B; CoM-CoB, mixed disulfide of CoM and CoB; Mph/MphH₂, oxidized and reduced methanophenazine; $F_{420}/F_{420}H_2$, oxidized and reduced Factor 420; Fd(ox)/Fd(red), oxidized and reduced ferredoxin; Ac, acetate; Ac-Pi, acetyl-phosphate; Ac-CoA, acetyl-Coenzyme A; EE, extracellular electron; EtOH, ethanol. Dashed lines indicate extracellular electron transfer. The figure was made by Weichao Wu (Marum) and Xiuran Yin.

1.2. Carbon metabolisms in methanogenesis

Methanogenesis, as an ancient pathway diverged before 3.51 billion years (Wolfe and Fournier 2018), is the terminal step of organic carbon degradation when the simple compounds or electron donors such as CO_2/H_2 , methanol and acetate are accumulated (Ferry and Lessner 2008). As one of greenhouse gases, annual methane emission can reach 500 – 600 Tg globally, in which 5 – 6% methane is originated from ocean (Reay et al., 2018). These large amounts of methane are generated via three main pathways: hydrogenotrophic, methylotrophic and acetoclastic methanogenesis.

1.2.1. Hydrogenotrophic methanogens

Most orders of methanogens are capable of methane formation from H_2 and CO_2 , including *Methanopyrales, Methanococcales, Methanobacteriales, Methanomicrobiales, Methanocellales* and *Methanosarcinales* (Liu and Whitman 2008; Thauer et al., 2008). Hydrogenotrophic methanogenesis is CO_2 -dependent. During methanogenesis, CO_2 is reduced to CH_4 with H_2 as electron donor according to the following reaction:

$$4 H_2 + CO_2 \rightarrow CH_4 + 2 H_2O$$
 $\Delta G^{\circ \prime} = -131 \text{ kJ/mol CH}_4 \text{ (Thauer et al., 2008) (1)}$

In this pathway (Fig. 4A), a range of hydrogenases in *Methanosarcinales* (Ech, Frh, Vht) or obligate hydrogenotrophic methanogens (Eha, Frh, Mvh) (Thauer et al., 2010) are used to catalyze reduction reactions with H_2 as electron donor and obtain electron carriers such as reduced ferredoxin and $F_{420}H_2$. These electron carriers further reduce CO₂ to formyl groups (CHO-MF and CHO-H₄MPT), methenyl group (CH=H₄MPT), methylene group (CH₂=H₄MPT), methyl groups (CH₃-H₄MPT and CH₃-CoM) and CH₄, subsequently. In versatile methanogens (*Methanosarcinales*), methanophenazine-dependent hydrogenase (Vht) catalyzes the reduction of methanophenazine, with which reduced coenzyme B (CoB) is formed, required for catalyzing methyl coenzyme M (CH₃-CoM) reduction to CH₄. The last step of methane formation (reduction of CH₃-CoM to methane) is catalyzed by another hydrogenase, i.e., F_{420} -non-reducing hydrogenase (Mvh) in obligate hydrogenotrophic methanogens as methanophenazine-dependent hydrogenase is restricted to the order of *Methanosarcinales* containing cytochromes (Thauer et al., 2010; Thauer et al., 2008).

Under standard conditions, hydrogenotrophic methanogenesis produces sufficient energy for cell growth (reaction 1). In fact, obligate hydrogenotrophic methanogenesis without cytochromes are more efficient in obtaining ATP than *Methanosarcinales* (Thauer et al., 2008), promoting the adaption of these methanogenes to utilization of low H₂ partial pressures (\geq 5 Pa) (Thauer et al., 2010). As hydrogenotrophic methanogenes are observed in a variety of anoxic environments, methane formation from H₂ and CO₂ is regarded as main methanogenic pathway in marine sediments (Katayama et al., 2015; Lazar et al., 2011).

Carbon assimilation by hydrogenotrophic methanogens involves CO_2 reduction, overlapping partly with the dissimilatory pathway, i.e., all the way to the formation of methyl-tetrahydromethanopterin (CH₃-H₄MPT) (Fig. 4). As the key precursor for biomass synthesis, acetyl-CoA is synthesized from CH₃-H₄MPT as methyl group donor and CO₂ as carboxyl group donor (Fig. 4). Both obligate and facultative hydrogenotrophic methanogens harbor the same pathway for autotrophic carbon assimilation.

1.2.2. Methylotrophic methanogens

There are two types of methyl-dependent methanogenesis pathways: disproportionation of C-1 compounds into methane and CO₂, and reduction of methyl group by H_2 to methane (Borrel et al., 2013; Lang et al., 2015) (Fig. 4B). The reactions are as follows:

 $4 \text{ CH}_{3}\text{OH} \rightarrow 3 \text{ CH}_{4} + \text{CO}_{2} + 2 \text{ H}_{2}\text{O} \qquad \Delta G^{\circ'} = -105 \text{ kJ/mol CH}_{4} \text{ (Liu and Whitman 2008) (3)}$ $CH_{3}\text{OH} + H_{2} \rightarrow CH_{4} + H_{2}\text{O} \qquad \Delta G^{\circ'} = -113 \text{ kJ/mol CH}_{4} \text{ (Liu and Whitman 2008) (4)}$

Disproportionation of methyl substrates was found in the family of *Methanosarcinaceae* and *Methermicoccaceae*. In this pathway, the methyl group is delivered to CH_3 -H₄MPT or methyl-tetrahydrosarcinapterin (CH_3 -H₄SPT) and subsequently to CO_2 via the reverse CO_2 reduction pathway. On the other hand, reducing equivalents ($F_{420}H_2$) stemming from methyl group oxidation are used to reduce the methyl group and generate CH_4 . For H_2 -dependent methylotrophic methanogenesis, however, reduction of methyl substrates by H_2 to methane is not involved in oxidation of methyl group to CO_2 . The methyl group can be reduced with electrons from H_2 to CH_4 directly. For this pathway, the versatile methanogenes in the family of *Methanosarcinaceae*, the genus of *Methanosphaera* and the order of *Methanomassiliicoccales* have the capability of methanol reduction to methane using H_2 as electron donor; in fact, in the latter, hydrogen is obligatory for methylotrophic methanogenesis (Borrel et al., 2014; Lambie et al., 2015; Poehlein et al., 2018).

Methanogens having the ability to use methyl substrates for methanogenesis are identified in different orders. For example, in the order of *Methanosarcinales*, the obligate methylotrophic methanogens only using C1 compounds for methane formation such as the genus of *Methanosarcina (Methanolobus, Methanomethylovorans* and *Methermicoccus*, as well as the versatile *Methanosarcina* (Cheng et al., 2007; Liu and Whitman 2008). Methylotrophic methanogens placed into other groups including *Methanomassiliicoccales, Candidatus* Methanofastidiosa and the newly proposed phylum Verstraetearchaeota, in which H₂ is required to reduce methyl group to methane because of the lack of methyl group oxidation capability to CO₂ (Borrel et al., 2014; Nobu et al., 2016; Vanwonterghem et al., 2016).

Methylotrophic methanogens are able to utilize a variety of methyl substrates including methanol, methyl amines, methionine, methyl sulfides, glycine betaine, choline and dimethylethanolamine for methanogenesis (Cha et al., 2013; Mochimaru et al., 2009; Watkins et al., 2014; Watkins et al., 2012). Specifically in marine sediments, methyl compounds are detectable. For instance, methanol originates from degradation of pectin and lignin (Fall and Benson 1996; Schink and Zeikus 1980). Methylated amines are derived from glycine betaine and choline, which are organic osmoprotectants in marine organisms (King 1984). Dimethylsulfide is formed by decomposition of dimethylsulfoniopropionate, an osmoprotectant in algae (Dacey and Wakeham 1986). These methyl substrates even harbor a high concentration, i.e., up to 69 µM of methanol, 3 µmol/kg trimethylamine and 15 µmol/kg of dimethylsulfoniopropionate (Yanagawa et al., 2016; Zhuang et al., 2016; Zhuang et al., 2018; Zhuang et al., 2014). These observations highlight the important role of methylotrophic methanogenesis in marine sediments, especially in SRZ and hypersaline sediments (Zhuang et al., 2016). Notably, methyl compounds are considered as non-competitive substrates for methanogenesis, since sulfate reducing microorganisms apparently do not compete with methanogenesis for these compounds (Oremland and Polcin 1982).

In methylotrophic methanogens, acetyl-CoA is formed from CH_3 - H_4SPT and CO_2 , indicating both CO_2 and methyl group are the main carbon sources. Mixotrophic growth, i.e. utilizing both methyl group carbon and CO_2 for assimilation, was observed in pure cultures of *Methanosarcina barkeri*, with equivalent proportions of methanol and CO_2 contributing to total cell carbon (Weimer and Zeikus 1978).

1.2.3. Acetoclastic methanogens

Acetoclastic methanogens utilize acetate for energy and carbon source. For methanogenesis, acetate is disproportionated to CO_2 and CH_4 , which can be performed by the order of *Methanosarcinales* (*Methanosarcinaceae* and *Methanosaetaceae*). The reaction is as follows:

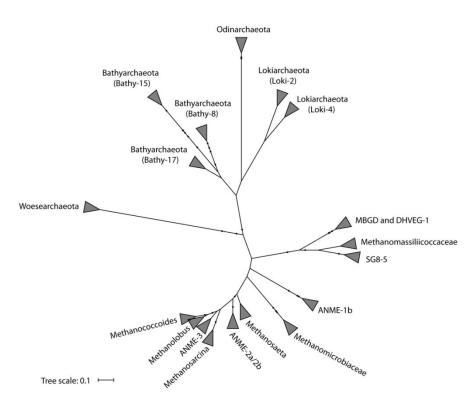
$$CH_3COOH \rightarrow CH_4 + CO_2$$
 $\Delta G^{\circ} = -35.6 \text{ kJ/mol } CH_4 \text{ (Conrad 1999) } (2)$

In acetoclastic methanogens, acetate is converted to acetyl-phosphate and further to acetyl-CoA via acetate kinase and phosphate acetyltransferase, respectively. As an important intermediate, acetyl-CoA can be disproportionated to CO_2 and CH_4 under catalysis of a series of enzymes. Regarding carbon assimilation, acetyl-CoA is the key precursor for pyruvate synthesis and further for macromolecule biosynthesis, i.e., protein, nucleic acids and lipids (Fig. 4C).

Compared with hydrogenotrophic and methylotrophic methanogenesis, acetoclastic methanogens generate less energy under standard conditions. However, acetate is the one of the major intermediates of organic polymer degradation formed by fermenting and syntrophic bacteria, and subsequently used

for methane formation in marine sediments (Beulig et al., 2018; Carr et al., 2018; Sorensen et al., 1981; Yoshioka et al., 2015). Acetate concentrations range from $2 - 23 \mu$ M in marine sediments (Iniri et al., 2010; Zhuang et al., 2018), indicating the availability as energy and carbon source for these methanogens.

In addition, methanogenesis involved in ethanol metabolisms also occurs in a syntrophic pathway by *Methanosaeta* spp. (Fig. 4D) (Lovley 2017; Rotaru et al., 2014). In this pathway, extracellular electrons generated from degradation of ethanol by *Geobacter* species are transferred via conductive pili to methanogens (Rotaru et al., 2014). In the presence of these extracellular electrons, CO_2 will be reduced to methane via the hydrogenotrophic methanogenesis pathway. On the other hand, the generated byproduct of ethanol degradation, i.e., acetate, is also used by *Methanosaeta* for methane formation via the acetoclastic methanogenesis pathway.



1.3. Carbon metabolisms in other uncultured archaea

Fig. 5 The unrooted 16S rRNA tree of archaea found in Helgoland Mud sediment and sediment incubations based on clone sequences. The raw tree contained about 300 clones was built by using maximum likelihood algorithm and bootstrapping (n=1000). 2 clones from each branch were picked from the original tree to generate this pruned tree. The tree was built by Ajinkya Kulkarni and Xiuran Yin.

1.3.1. Archaeal diversity in the Helgoland Mud sediments

Archaea affiliated to marine benthic group D (MBGD), Bathyarchaeota and Lokiarchaeota (phylum of Asgard archaea) tend to be found in organic-rich sediments (Durbin and Teske 2012). With a high sedimentation rate and organic carbon input, these archaea are also detected in Helgoland Mud sediments and incubations, including Thaumarchaeota, Euryarchaeota, Bathyarchaeota and Asgard archaea (Lokiarchaeota and Odinarchaeota). Among Euryarchaeota, methanogens with the three methanogenic pathways are found from archaea clones and next-generation sequencing (Fig. 5) (Oni et al., 2015a; Oni et al., 2015b). These methanogens comprised the family of *Methanosarcinaceae* (*Methanococcoides, Methanolobus* and *Methanosarcina*), *Methanosaetaceae*, *Methanomicrobiaceae* and *Methanomassiliicoccales*-like group. Except for methanogens, anaerobic methanotrophs (ANMEs) including three subgroups are also found (ANME-1b, ANME-2a/2b and ANME-3), indicating the activities of anaerobic methane oxidation in Helgoland Mud sediments. The other Euryarchaeota such as MBGD, SG8-5 are also detectable in original sediment and incubations (Fig. 5). High relative abundance of Thaumarchaeota is detected from the surface sediment, suggesting oxygen-dependent ammonia oxidation by members of this phylum.

Bathyarchaeota and Asgard archaea are important archaeal subgroups in marine sediments. The in situ activities of Bathyarchaeota and Lokiarchaeota are not well known, but evidence exists that Bathyarchaeota is the most dominant archaeal phylum in deep sediments of Helgoland Mud Area (> 50%) (Oni et al., 2015b) and perform CO_2 fixation and degradation of aromatic compounds such as lignin by the subgroup of Bathy-8 (Meng et al., 2014; Yu et al., 2018). By contrast, activity of Asgard archaea is unknown in marine sediments. Asgard archaea is the newly described archaeal super phylum, which include Lokiarchaeota, Thorarchaeota, Odinarchaeota, Heimdallarchaeota and Helarchaeota. These archaea harbor eukaryotic signature proteins and have a crucial phylogenetic position close to eukaryotes, suggesting a role as close relatives of eukaryotes (Spang et al., 2015; Zaremba-Niedzwiedzka et al., 2017). The metagenome-assembled genomes detected from marine sediments suggest Lokiarchaeota and Thorarchaeota might fix CO₂ associating with organic carbon degradation; Heimdallarchaeota potentially utilize nitrate or oxygen to breakdown organic compounds; Odinarchaeota can ferment carbohydrates and Helarchaeota have the capability of hydrocarbon oxidation (Bulzu et al., 2019; Seitz et al., 2019; Seitz et al., 2016; Spang et al., 2019). However, these predicted metagenomic evidences directly obtained from original marine sediments showed all the abilities of Asgard archaea, while the activity in marine sediments is still unknown.

1.3.2. Carbon fixation

Carbon fixation is ubiquitous in methanogens, e.g., autotrophic growth of hydrogenotrophic methanogens and mixotrophic growth of methylotrophic methanogens. In these methanogens, CO_2 is

incorporated into biomass by the Wood-Ljungdahl pathway (reductive acetyl-CoA pathway) and reduced to acetyl-CoA under catalysis of carbon monoxide dehydrogenase/acetyl-CoA synthase (CODH/ACS). Specifically, the complete Wood-Ljungdahl pathway is widespread in most archaea in anoxic Helgoland Mud sediment (Table 1), indicating the potential capability of these archaea in carbon fixation. Methanogens, ANMEs, MBGD, Bathyarchaeota, Woesearchaeota and Lokiarchaeota have the genes encoding the type III ribulose-1,5-bisphosphate carboxylase/oxygenase (RubisCO), the key enzyme catalyzing the CO₂ fixation in the Calvin-Benson-Bassham (CBB) cycle (Table 1). However, the CBB cycle in these archaea is incomplete (Table 1) since the gene encoding Phosphoribulokinase is not present. Although the RubisCO gene is detected and expressed in methanogens (Table 1) (Allen et al., 2009; Goodchild et al., 2004), its function is still enigmatic; potentially RubisCO is involved in metabolism of adenosine 5-monophosphate (Sato et al., 2007) or a new carbon fixation pathway (reductive hexulose-phosphate pathway) (Kono et al., 2017). Similarly, the tricarboxylic acid (TCA) cycle is also incomplete in most archaea with the exception of the MBGD (Table 1) (Lazar et al., 2017; Zhou et al., 2018). In addition, some archaea are "organoautotrophs" such as ANME-1 and Bathyarchaeota since they specifically utilize CO₂ for carbon source and methane and lignin for energy, respectively (Kellermann et al., 2012; Yu et al., 2018), although the details of their carbon-fixing pathways are unclear.

1.3.3. Alkane and fatty acids degradation

The *mcr* genes, encoding methyl coenzyme M reductase, have been found to be central in both methanogenesis and alkane oxidation. In marine sediments, methane and short-chain alkanes will be formed via the mediation of biological process (Hinrichs et al., 2006) and subsequently oxidized by other archaea in anoxic sediments when electron acceptors are available (Ettwig et al., 2016; Hinrichs et al., 1999; Jørgensen et al., 2004; Laso-Perez et al., 2016). These alkane oxidizing archaea are mainly placed within the group of ANMEs. For example, ANME-1 and ANME-2 archaea perform methane oxidation associated with sulfate reduction by the *Desulfosarcina/Desulfococcus* group (Blumenberg et al., 2004; Michaelis et al., 2002; Schreiber et al., 2010). Besides, short-chain hydrocarbons such as butane potentially can be degraded by novel ANME-1, Bathyarchaeota and Asgard archaea (Table 1), since the phylogenetic position of detected *mcr* genes are close to the putative butane oxidizer (Syntrophoarchaea) (Dombrowski et al., 2018; Evans et al., 2015; Laso-Perez et al., 2016; Seitz et al., 2019). The expansive and divergent *mcr* genes found in recent years reflect the wide distribution and complex metabolisms encoded by in *mcr* genes in the kingdom of archaea (Evans et al., 2019).

Interestingly, beta-oxidation is detected in some archaea (novel ANME-1, Bathyarchaeota and Asgardarchaeota) (Table 1) (Dombrowski et al., 2018; He et al., 2016; Seitz et al., 2019). Specifically, these archaea harbor the capability of alkane oxidization, potentially involving alkane oxidation to

fatty acids and further conversion to acetyl-CoA via beta-oxidation. Additionally, beta-oxidation in these archaea may be also associated with the degradation of fatty acids, which are the important class of compounds in marine sediments (Kniemeyer et al., 2007; Pearson et al., 2001; Sinninghe Damsté et al., 2003).

1.3.4. Fermentation and organic polymer degradation

In many archaea, most steps of Embden-Meyerhof-Parnas (EMP) pathway are present (Table 1). However, except for Woesearchaeota (Lazar et al., 2017; Liu et al., 2018), the lack of evidence of sugar degradation and fermentation to acetate indicates that these archaea might use the reverse pathway to obtain sugar-phosphate, which is further used for nucleotide synthesis. Therefore, although acetate metabolism is present in these archaea, it is unclear whether acetate is produced or incorporated acetate as the relevant enzymes (acetyl-CoA synthetase) catalyze acetate metabolism are reversible (Schäfer et al., 1993).

Functional genes encoding protein breaking down and incorporating into cells were detected in MBGD and DHVEG-1, Bathyarchaeota and Asgard archaea (Table 1) (He et al., 2016; Lazar et al., 2016; Lazar et al., 2017; MacLeod et al., 2019; Zhou et al., 2018). Protein degradation might be an important survival strategy for these archaea, since protein can provide both carbon and nitrogen source for growth. Nevertheless, the direct evidence about the activities of protein degradation by these archaea is unclear to date. More promisingly, Bathyarchaeota is active in lignin degradation as subgroup of Bathy-8 is enriched in a long-term incubation amended with lignin (Yu et al., 2018). Considering divergent carbon metabolic pathways in archaea found from anoxic Helgoland Mud sediment, archaea might have versatile strategies in carbon utilization, potentially including carbon fixation, alkane degradation, aromatic compound degradation and the intermediates utilization.

	WL	CBB	TCA	Alkane metabol.	β-oxid.	EMP	Acetate metabol.	Protein degrad.	Aromatic degrad.	Ethanol metabol.
Methanosarcinaceae	+++	++	++	+++ ^a	-	++	+++ ^e	-	-	+++ ^g
Methanosaetaceae	+++	++	++	$+++^{a}$	-	++	$+++^{e}$	-	-	+++ ^g
Methanomicrobiaceae	+++	++	++	$+++^{a}$	-	++	$++^{f}$	-	-	-
Methanomassiliicoccaceae	+	++	++	++ ^a	-	++	$++^{f}$	-	-	-
ANME-1	+++	-	++	+++ ^{a,b,c}	+++	++	$++^{f}$	+	-	-
ANME-2a/2b	+++	++	++	$+++^{b}$	-	++	$++^{f}$	-	-	-
MBGD and DHVEG-1	+++	-	+++	-	-	++	$+++^{f}$	+++	-	$+++^{h}$
Woesearchaeota	++	-	+	-	-	++	$+++^{f}$	-	-	$+++^{h}$
Bathyarchaeota	+++	-	++	+++ ^{a,c}	$+++^{d}$	++	$+++^{f}$	+++	+++	$+++^{h}$
Asgard archaea	+++	++	++	+++	$+++^{d}$	++	$+++^{f}$	+++	-	$+++^{h}$

Table 1 Summary of the main carbon metabolism pathways in archaea found in anoxic sediment of Helgoland Mud Area

TCA, tricarboxylic acid cycle; CBB, Calvin-Benson-Bassham cycle; 3-HP/4-HB, 3-hydroxypropionate/4-hydroxybutyrate cycle; WL, Wood-Ljungdahl pathway; Alkane metabol., Alkane metabolisms; EMP, Embden–Meyerhof–Parnas pathway; Protein degrad., Protein degradation; β-oxid., β-oxidation; Aromatic degrad., Aromatic compounds degradation; Acetate metabolisms; Ethanol metabol., Ethanol metabolisms.

+++, pathway complete; ++, main gene cluster detected; +, few gene cluster detected; -, no relevant gene detected; na, not applicable.

a, methanogenesis; b, methane oxidation; c, short-chain alkane degradation; d, long-chain alkane degradation; e, acetoclastic methanogenesis; f, acetogenesis or acetate utilization; g, extracellular electron transfer mediated methanogenesis; h, ethanol generation.

The summary is based on the KEGG metabolic pathways and the previous studies (Allen et al., 2009; Beulig et al., 2019; Deppenmeier et al., 2002; Dombrowski et al., 2018; Evans et al., 2015; He et al., 2016; Lazar et al., 2016; Lazar et al., 2017; Liu et al., 2018; MacLeod et al., 2019; Meng et al., 2014; Meyerdierks et al., 2010; Rotaru et al., 2014; Seitz et al., 2019; Spring et al., 2010; Wang et al., 2014; Zhou et al., 2018).

1.4. Identification of active microorganisms by stable isotope probing

Stable isotope probing (SIP) is an incubation-dependent technique that uses substrates labelled with stable isotopes such as ¹³C-carbon, ¹⁵N-nitrogen or ¹⁸O-oxygen for detecting microbial activities by tracking the incorporated labeled substrates into macromolecules such as nucleic acids and lipids, i.e., DNA-SIP, RNA-SIP and lipid-SIP.

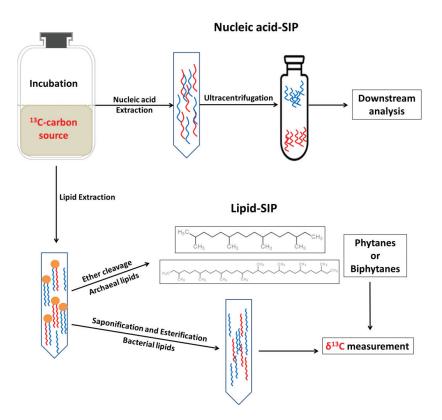


Fig. 6 Workflow of nucleic acid- and lipid-SIP based on ¹³C-carbon incorporation

1.4.1. Nucleic acid-SIP

For detecting active microbial members in environmental samples by nucleic acid-SIP, ¹³C-carbon, ¹⁵Nnitrogen or ¹⁸O-oxygen substrates can be assimilated into nucleic acids, thereby increasing the density of DNA and RNA (Cadisch et al., 2005; Lueders et al., 2004; Schwartz et al., 2016). This is the basis for separating labeled nucleic acids with higher density from unlabeled ones (lower density) by isopycnic centrifugation (Fig. 6) (Radajewski et al., 2000). Considering the difference of DNA and RNA densities, nucleic acid-SIP is conducted in cesium chloride and cesium trifluoroacetate gradients, respectively (Lueders et al., 2003; Manefield et al., 2002; Radajewski et al., 2000). In combination with next generation sequencing of the labeled and unlabeled nucleic acids, nucleic acid-SIP harbors a high resolution and sensitivity in identification of taxonomic levels (Aoyagi et al., 2015; Aoyagi et al., 2018; Singer et al., 2017). Furthermore, combining nucleic acid-SIP with metagenomics and metatranscriptomics provides the direct evidence for construction and characterization of carbon metabolic pathways (Bradford et al., 2018; Chemerys et al., 2014; Coyotzi et al., 2016; Fortunato and Huber 2016; Kalyuhznaya et al., 2009). Except for ¹³C-carbon substrates, ¹⁵N-nucleic acid-SIP is conducted to track nitrogen metabolisms in environmental samples (Buckley et al., 2007; Roh et al., 2009), but with a higher threshold level (atom% >50%) than ¹³C-nucleic acid-SIP (>10%) (Cadisch et al., 2005; Manefield et al., 2002). In addition, by amending with $H_2^{18}O$ into soils or sediments, the *in situ* microbial growth and activity have been detected by using ¹⁸O-nucleic acid-SIP (Rettedal and Brozel 2015; Schwartz et al., 2016). However, most of these studies detected bacteria activities based on carbon and nitrogen assimilation. For identifying archaeal activity, it is necessary and worthy to try the nucleic acid-SIP method because of its sensitivity for detecting uncultured archaea with low activity.

1.4.2. Lipid-SIP

Unlike the taxonomic identification of active microorganisms by nucleic acid-SIP, lipid-SIP harbors a lower phylogenetic resolution but measurement of δ^{13} C or δ D-values by gas chromatography combustion isotope ratio mass spectrometry (GC-c-IRMS) in lipids allows us to track carbon and hydrogen assimilation into biomass quantitatively and high-sensitively (Berry et al., 2015; Boschker et al., 1998; Kopf et al., 2016; Wegener et al., 2016). Before δ^{13} C and δ D lipid measurement from environmental samples, fatty acids or isoprenoid chains are obtained by saponification/esterification and ether cleavage, respectively (Fig. 6) (Elvert et al., 2003; Liu et al., 2012). Lipid-SIP has been used for assessing carbon assimilation patterns, carbon fixation and organic carbon degradation in a variety of environments (Blaser et al., 2015; Boschker and Middelburg 2002; Hinrichs et al., 1999; Wegener et al., 2012).

1.5. Objectives and structure of the thesis

A number of studies about methanogenesis in marine sediments in recent years have indicated the importance of methylotrophic methanogenesis. Methylotrophic methanogens can utilize methyl substrates for methane formation in presence of sulfate and sulfate reducers successfully because it is regarded as non-competitive substrate. However, it is unclear, which methylotrophic methanogens are active in the sulfate reduction zone and the methanogenic zone of marine sediments, although methyl substrates and methanogenic methylotrophic activity are detected. Marine sediment harbor a large pool of inorganic carbon, and thus, this ambient CO_2 may affect carbon utilization by methylotrophic methanogens. Hence, identifying and characterizing methylotrophic methanogens in marine sediments allows us to know how carbon can be used in marine sediments by these archaea.

Apart from methylotrophic methanogens, uncultured archaea such as Bathyarchaeota, Asgard archaea and MBGD and DHVEG-1 are found in the Helgoland sediments while their activities are not well studied. Especially for Asgard archaea, studying their activity will help us understand the survival strategy of these archaea. Because Asgard archaea are the closest relatives of Eukarya, even their carbon metabolism can provide insight into evolutionary relationships between Archaea and Eukarya. Nevertheless, some studies suggest that these uncultured archaea might utilize a decent amount of CO_2 for carbon assimilation.

In this study, I used SIP methods including nucleic acid- and lipid-SIP to further understand the carbon utilization patterns in methylotrophic methanogens and uncultured archaea. Specifically, I worked on the following objectives:

- 1) Identify methylotrophic methanogens in different depths of marine sediment by RNA and DNA-SIP (sulfate reduction zone and methanogenic zone).
- 2) Quantify carbon utilization patterns in methylotrophic methanogens by nucleic acid-SIP and lipid-SIP.
- Identify uncultured archaea in deep marine sediment by nucleic acid-SIP with providing potential electron donor, electron acceptor and ¹³C-labeled dissolved inorganic carbon.

In brief, this thesis contains three manuscript chapters (chapter 2 to 5):

Chapter 2: **DNA and RNA stable isotope probing of methylotrophic methanogenic archaea**. In this chapter, we developed a labeling strategy to identify methylotrophic methanogens by amendment with ¹³C-DIC using nucleic acid-SIP. In combination with Illumina sequencing of nucleic acid samples from the "heavy" and "light" fractions, we showed that ¹³C-DIC is necessary to use in order to obtain sufficient labeled nucleic acids in heavy gradient fractions after density separation of nucleic acids.

Chapter 3: CO₂ conversion to methane and biomass in obligate methylotrophic methanogens in marine sediments. In order to identify methylotrophic methanogens in the Helgoland Mud sediments, we applied RNA-SIP on the incubations amended with ¹³C-DIC and unlabeled methanol and successfully detected methylotrophic methanogens. By analyzing carbon assimilation into biomass using lipid-SIP, we found that the contribution of inorganic carbon was higher than that of methanol. The further study of the autoclaved slurry incubations with inoculated *Methanococcoides methylutens* suggested that obligate methylotrophic methanogens have the unexpected ability of using CO₂ for methane formation during methylotrophic methanogenesis. The CO₂-dependent methanogenesis in methylotrophic methanogens may play an important role in environment.

Chapter 4: Asgard archaea are key participants in marine sediment carbon cycling. In this study, we showed that amendment of ¹³C-DIC and electron donors into the SIP incubations was an effective strategy on identifying uncultured archaea (Asgard archaea). In the presence of organic polymers (lignin, cellulose and humic acid) or sulfur, we found that Asgard archaea were incorporating ¹³C-DIC in these SIP incubations. Combining metagenomes and metatranscriptomes of the Asgard archaea obtained from SIP samples and mangrove sediment in the southeast coast of China, we demonstrated that Asgard archaea harbor genes for alkane, protein, fatty acids and cellulose degradation, and are actively involved in organic polymer degradation as well as CO_2 fixation.

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Chapter 2

DNA and RNA stable isotope probing of methylotrophic methanogenic archaea

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Experimental concept and design	80%
Acquisition of experimental data	80%
Data analysis and interpretation	80%
Preparation of figures and tables	70%
Drafting of manuscript	50%

Abstract

Methylotrophic methanogenic archaea are an integral part of the carbon cycle in various anaerobic environments. Different from methylotrophic bacteria, methylotrophic methanogens assimilate both, the methyl compound and dissolved inorganic carbon. Here, we present DNA- and RNA-stable isotope probing (SIP) methods involving an effective labeling strategy using ¹³C-labeled dissolved inorganic carbon (DIC) as carbon source along with methanol as dissimilatory substrate.

Keywords Methylotrophic methanogens, Stable isotope probing (SIP), Labeling strategy, Dissolved inorganic carbon, Methanol

Running head: SIP of methylotrophic methanogens

2.1. Introduction

A large part of the globally emitted climate relevant trace gas methane is produced by methanogenic archaea, which use three main groups of substrate: hydrogen (formate) and CO₂, acetate, and methyl compounds (e.g., methanol, methyl amine) [1, 2]. Using ¹³C-labeled acetate and CO₂, acetoclastic and hydrogenotrophic methanogens have been identified using stable isotope-probing (SIP) of nucleic acids [3, 4, 5]. Conversely, methylotrophic methanogens have not been studied using SIP of nucleic acids.

SIP of nucleic acids capitalizes on assimilation of isotopically labeled compounds (e.g., ¹³C, ¹⁵N) into DNA and RNA, density separation of differently labeled nucleic acid, and identification of actively assimilating microorganisms in isotopically "heavy" DNA or RNA [6, 7, 8]. Originally, the method was described for identifying methylotrophic bacteria with ¹³C-labeled methanol in soil microcosms [6], but has been extended to a wide range of labeled compounds used and physiological guilds traced. Methylotrophic bacteria use C-1 compounds as carbon and energy source [9], and¹³C-labeled methyl substrates have been demonstrated to be highly effective in labeling nucleic acids of methylotrophs using SIP of nucleic acids [10, 11]. Contrastingly, methylotrophic methanogens assimilate carbon mixotrophically, for example, using both, organic and inorganic carbon sources at the same time. In methanogenic archaea, de novo nucleotide biosynthesis is based on acetyl-CoA synthesis from methyltetrahydrosarcinopterin and CO₂ via the Wood-Ljungdahl pathway [12, 13], with 50% of acetyl carbon originating from inorganic carbon in methylotrophic methanogens [14]. Pyruvate, a precursor for nucleic acid synthesis in methanogens, is formed by further incorporation of CO_2 into acetyl-CoA [15], elevating the inorganic carbon contribution to 67% in nucleic acids. Thus in case of methylotrophic methanogens, the typical assumption in SIP labeling that the dissimilated substrate (here, ¹³C-methyl compounds) will also be dominantly present in the biomass formed, will result in low levels of labeling of DNA and RNA. As a remedy, we have developed a labeling strategy for targeting methylotrophic methanogens involving a mixture of ¹³C-CO₂ and a C-1 compound such as methanol to ensure sufficient labeling of these methanogens in DNA- and RNA-SIP [16]. Here, we present the workflow for labeling of methlyotrophic methanogenic archaea by DNA- and RNA-SIP using a combination of ¹³C-labeled and unlabeled methanol and DIC with marine sediment incubations as an example. Major steps of the method encompass (i) labeling of methylotrophic methanogens in sediment incubations, (ii) nucleic acid extraction, and (iii) density separation of differently labeled DNA or RNA. The chapter is concluded by discussing anticipated results of the method.

2.2. Materials

Chemicals used for sediment slurry incubations are prepared using Milli-Q water and are 'pro analysi (p. a.)' (*see* **Note 1**). Chemicals and reagents used for DNA and RNA based processing are ultrapure, molecular biology grade and devoid of DNA and RNA nucleases. They are prepared in diethyl pyrocarbonate (DEPC) treated, autoclaved water (*see* **Note 2**). All solutions are sterilized by autoclaving (120 °C for 20 min, 103 kPa) and filtered through 0.22 μ m polyethersulfone filters into sterilized glass bottles (*see* **Note 3**). All glassware, spoons, spatulas, and magnetic stir bars are sterilized and made nuclease-free by baking at 180 °C, 3 hours.

2.2.1. Sediment slurry incubations

Incubations for SIP of methylotrophic methanogens are performed in slurries prepared with anoxic marine sediment (*see* **Note 4**).

- Artificial Sea Water (ASW): Dissolve 26.4 g NaCl, 11.2 g MgCl₂·6H₂O, 1.5 g CaCl₂·2H₂O, and 0.7 g KCl in 1 L Milli-Q water (*see* Note 5). Transfer into 1-L serum bottle and crimp-seal with butyl rubber stopper. Solution is sterilized and made anoxic (*see* Note 6).
- 500 mM Methanol stock solution: Add 98.7 mL Milli-Q water into a sterile serum bottle, crimpseal with a butyl rubber stopper, and make it anoxic. Weigh out solutions of ¹³C-methanol (1.652 g) or unlabeled methanol (1.602 g) and transfer to anoxic water using a N₂ flushed, sterile needle (0.8 x 40 mm, 21G) and syringe and autoclave (*see* Note 7).
- 3. 500 mM DIC solution: Transfer 4.25 g sodium ¹³C-bicarbonate or 4.20 g unlabeled sodium bicarbonate into 120-mL serum bottle and crimp-seal with butyl rubber stopper. The bottle is made anoxic by flushing with N₂ for 10 min (*see* Note 8). Prepare anoxic water by flushing 100 mL Milli-Q water with N₂. Using a N₂ flushed sterile needle and syringe, add 99 mL anoxic water to dissolve the bicarbonate and autoclave.
- 4. Environmental samples: Marine sediment sample (see Note 9).

2.2.2. Media

- Lysogeny broth (LB) medium (100 mL): Dissolve 1 g tryptone, 0.5 g yeast extract and 0.5 g NaCl in 90 mL Milli-Q water. Adjust pH to 7 and volume to 100 mL. Transfer the medium to 150-mL glass flasks and seal with cotton plugs. Sterilize by autoclaving.
- 2. *E. coli* OD 2, ¹³C-labeled medium (*E. Coli*-OD2 C, Silantes, Germany): Ready to use *E. coli* medium. Transfer the medium to 150-mL Erlenmeyer flasks and seal with cotton plugs. Sterilize by autoclaving.

2.2.3. Nucleic acid extraction

- 1. 0.1 mm zirconium beads (Carl Roth, Karlsruhe): Sterilized by baking at 180°C for 3 hours in a glass bottle.
- 2. 120 mM sodium phosphate buffer (pH 8): 112.87 mM Na₂HPO₄ and 7.12 mM NaH₂PO₄.
- Tris-NaCl-sodium dodecyl sulfate (TNS): 500 mM Tris-HCl pH 8.0, 100 mM NaCl, 10% sodium dodecyl sulfate (SDS) (w/v).
- 4. Phenol:chloroform:isoamylalcohol pH 8 (PCI; 25:24:1; v/v/v)
- 5. Chloroform:isoamylalcohol (CI; 24:1; v/v) (see Note 10).
- 30% Polyethylene glycol (PEG) 6000 (w/v): Slowly dissolve 60 g of PEG 6000 in 100 mL 1.6 M NaCl and make up volume to 200 mL. Sterilize by autoclaving (*see* Note 11).
- 7. 70% ethanol (v/v): Prepare using DEPC treated water and store at 4 $^{\circ}$ C.
- 8. Quant-iT PicoGreen dsDNA assay kit (Invitrogen-Thermo Fischer Scientific) (see Note 12).
- 9. Quant-iT RiboGreen kit (Invitrogen-Thermo Fischer Scientific).
- 10. RNase-free DNase.
- 11. Bead beater (Fast-Prep 24, MP Biomedicals).
- 12. Refrigerated centrifuge.
- 13. NanoDrop spectrophotometer (ND-1000, Peqlab Biotechnologie).
- 14. Microplate fluorometer (Fluoroskan Ascent FC, Thermo labsystems)

2.2.4. DNA- and RNA-SIP

- 1. Gradient Buffer (GB): 0.1 M Tris-HCl (pH 8), 0.1 M KCl, 1 mM ethylenediaminetetraacetic acid. Prepare using DEPC-treated water, autoclave in baked glassware.
- 7.163 M Cesium chloride solution (CsCl, 1.847 g/mL): 217.07 g Cesium chloride dissolved in 180 mL Milli-Q water. Filter sterilize into serum bottles crimp-sealed with butyl rubber stoppers (*see* Note 13).
- 3. Cesium trifluoroacetate (2 ± 0.05 g/mL): Cesium trifluoroacetate solution is directly obtained from GE Healthcare, United Kingdom, in 120-mL rubber stopper sealed serum bottles.
- 4. Formamide (\geq 99.5%, deionized): Stored at -20 °C until use. (*see* Note 14)
- 5. Digital refractometer (AR200, Reichert Technologies).
- 6. Vertical ultracentrifuge rotor (VTi 65.1, Beckman Coulter) with tube adapters and seals.
- 7. Ultracentrifuge (Optima XE-90, Beckman Coulter).
- 8. Quick seal 16 x 45 mm tubes (Beckman Coulter).
- 9. Microbalance.
- 10. Beckman cordless tube topper.

- 11. Syringe pump (Aladdin syringe pump, AL-1000, WPI).
- 12. Stop watch.
- 13. 7.5 M ammonium acetate.
- 14. 100% isopropanol.
- 15. GenElute-Linear polyacrylamide (LPA, 25 µg/mL; Merck).
- 16. RNaseZap (Thermo Fischer Scientific).

2.3. Methods

SIP experiments start with the incubation of the microbial community, ideally, under close to natural conditions and in triplicate biological replicates (3.1), followed by the nucleic acid extraction (3.2), density separation of nucleic acids by ultracentrifugation (3.3.1–3.3.6), and precipitation, quantification and analysis of DNA and RNA from "light" and "heavy" density fractions (3.3.7–3.3.8, 3.5) (*see* **Note 15**).

2.3.1. SIP incubations

- Transfer 200 g sediment into glass bottle under constant flushing of N₂ (or use an anaerobic glove box when available). Transfer 800 mL of ASW to prepare a 1:4 (w/v) sediment slurry dilution. Flush the headspace of the bottle with N₂ for 3 min and seal with a butyl stopper. Mix sediment slurry until homogeneity is achieved (*see* Note 16).
- To set up anoxic sediment slurry incubations (n=3), flush 120-mL serum bottles with N₂ for 2 min and crimp-seal using a butyl rubber stopper. Under continuous N₂ flushing and stirring, transfer 50 mL of sediment slurry to each 120-mL serum bottle using a sterile needle (0.8 x 40 mm, 21G) and syringe.
- 3. To remove residual oxygen and ambient CO₂ in serum bottles, sediment slurries are made anoxic as stated in Note 6.
- 4. Pre-incubate sediment slurries for 10 days at 30 °C in the dark (see Note 17).
- 5. Prepare a working methanol solution (50 mM) by transferring stock solution (500 mM) to a serum bottle containing sterile anoxic water. Add 1 mL of ¹³C-labeled or unlabeled methanol solution (50 mM) to the respective sediment slurries using a sterile needle (0.8 x 40 mm, 21G) and syringe (Table 1; Fig. 1). Similarly, add 1 mL of ¹³C-DIC or unlabeled DIC to sediment slurries (*see* Note 18).
- 6. Incubate the sediment slurries at 30 °C (see Note 19).
- 7. Track methane concentration in headspace by gas chromatography (GC) [17], and stop the incubation when methane concentrations in the headspace stop increasing (Fig. 2A) (*see* **Note 20**).

8. Up to 8 mL of slurry is transferred in 2-mL aliquots into sterile Eppendorf tubes, per biological replicate incubation and stored at -80 °C (*see* Note 21).

2.3.2. Nucleic acids extraction

- 1. 2 mL sediment slurries (see step 8, Section 3.1) are centrifuged for 5 min to obtain sediment pellets after discarding the supernatant.
- Add 750 μL of 120 mM sodium phosphate buffer (pH 8) to the sediment pellet and vortex to resuspend the pellet.
- 3. Transfer the re-suspended sediment to a 2-mL sterile screw cap vial containing 0.5 g of sterile zirconium beads.
- 4. Add 250 μL TNS solution and gently mix the tubes without vortexing (see Note 22).
- 5. Disrupt cells by bead beating for 1 min at 6.5 m/s and centrifuge for 10 min (see Note 23).
- 6. Carefully transfer the supernatant to a 2-mL tube.
- 7. Add 1 volume PCI and gently vortex for 10 seconds (see Note 24).
- 8. Incubate for 10 min at room temperature and centrifuge for 5 min.
- 9. Transfer upper aqueous phase to a 2-mL tube (see Note 25).
- 10. Add 1 volume CI to the aqueous phase and gently vortex for 10 seconds and centrifuge for 5 min.
- 11. Transfer the aqueous phase to a 2-mL tube and fill the tube up to the 2 mL mark with 30% PEG solution. Gently mix by turning the tube up and down and incubate for 30 min at room temperature to precipitate the nucleic acids.
- 12. Pellet the precipitated nucleic acids by centrifuging at 45 min.
- 13. Discard supernatant and add 1 mL cold 70% ethanol to wash pellet (see Note 26).
- 14. Centrifuge for 5 min and repeat step 13.
- 15. Discard supernatant and air dry pellet under a fume hood for 5 min (see Note 27).
- 16. Dissolve DNA or RNA pellet in 50 µL DEPC treated water (see Note 28).
- 17. Quantify DNA with Quant-iT PicoGreen dsDNA assay kit and assess its quality using NanoDrop spectrophotometer.

Steps 18–21 for RNA extraction only:

- 18. To remove DNA from extracts for RNA-SIP, extracts are treated with DNase using RQ1 DNase.
- 19. For removal of DNases, add 500 μ L of DEPC treated water and repeat steps 7–16.
- 20. Quantify RNA with Quant-iT RiboGreen kit and assess its quality using NanoDrop spectrophotometer.

Conduct PCR with archaeal primers to confirm absence of DNA after DNase treatment (*see* Note 29).

2.3.3. Density separation of isotopically labeled nucleic acids

2.3.3.1 Preparation of Escherichia coli DNA and RNA standards

- 1. Inoculate a culture of *E. coli* DSM 498 strain into 50 mL of liquid LB medium and 50 mL of ¹³C-labeled *E. coli* medium under a sterile clean bench.
- 2. Incubate cultures at 37 °C overnight at 150 rpm in a shaker incubator. Collect cells by centrifugation.
- 3. Repeat steps from section 3.2 to obtain ¹³C-labeled and unlabeled *E. coli* DNA and RNA.

2.3.3.2 Preparation of samples for DNA-SIP

- Gently mix 1.5 μg of DNA from SIP incubations with 1.5 mL GB. Similarly, mix 1.5 μg of ¹³Cand unlabeled *E. coli* DNA (a total of 3 μg) with 1.5 mL GB (*see* Note 30).
- 2. Add 6.16 mL CsCl to the DNA-GB mixture.
- Vortex gently and measure refractive index (RI) of each mixture (75 μL from each fraction) using the refractometer in "nD-TC" mode after calibrating with Milli-Q water. Adjust the RI of the mixtures to 1.4022 by adding 30 μL of GB or CsCl (*see* Note 31).
- 2.3.3.3 Preparation of samples for RNA-SIP
 - Mix 750 ng of RNA from sediment slurries with 1.3 mL GB. Similarly, mix 375 ng of ¹³C- and unlabeled *E. coli* RNA (a total of 750 ng) with 1.3 mL GB.
 - 2. Add 240 µL formamide and 6 mL of CsTFA to the RNA-GB mixture (see Note 32).
 - 3. Gently mix, measure RI of each mixture as stated above (3.3.2) and adjust the RI to 1.7328 by adding 30 μL aliquots of GB or CsTFA.

2.3.3.4 Preparation of ultracentrifugation

- 1. Transfer sample mixtures to 6.9 mL Quick Seal tubes using a 10-mL sterile syringe and needle (0.8 x 40 mm, 21G). Fill the Quick Seal tubes up to their shoulder mark (*see* **Note 33**).
- 2. Balance (with tube adapters and rotor seals for the rotor) the weight of two tubes against each other on a microbalance so that the weight difference is ± 0.01 g. Use leftover sample mixtures from before. (*see* Note 34).
- 3. Seal tubes with Beckman cordless tube topper (see Note 35).

- 4. Place the tubes into the vertical rotor (VTi 65.1) and fit the tube adapters and rotor seals.
- 5. Tighten the rotor seals using a wrench and a pressure not greater than 120 psi.
- 6. Centrifuge for 40 hours at 20 °C and 192,600 g for DNA-SIP and for 65 hours at 20 °C and 124,000 g for RNA-SIP (see Note 36).

2.3.3.5 Fractionation of nucleic acids

- After centrifugation, carry rotor from centrifuge to bench with minimum disturbance. Remove the tubes from rotor using a wrench and attach one of the tubes to a clamp stand such that the tube is stable but not under high pressure (*see* Note 37). Prepare 16 sterile, 1.5-mL Eppendorf tubes below the fixed sample tube to collect gradient fractions.
- 2. Fill a 20-mL sterile syringe, attached to a long tube, with DEPC treated water and attach it to the pre-calibrated syringe pump (pump rate of 1 mL/min).
- 3. Attach a sterile needle (0.5 x 16 mm, 25G) to the end of the tube. Pump water to remove any air bubbles from the setup. Refill syringe if necessary.
- 4. Carefully pierce the needle into the shoulder of Quick Seal tube such that the needle opening faces the solution in the tube (*see* **Note 38**).
- 5. Pierce the bottom of the tube with another sterile needle (0.5 x 16 mm, 25G) and remove it.
- Pump the Milli-Q water from the needle at the top by starting the syringe pump and collect 16 fractions (420 μL) in 1.5-mL Eppendorf tubes placed below the tube by manual shifting each tube every 25 sec (using preset stopwatches) (*see* Note 39).

2.3.3.6 Density measurement of gradient fractions

- 1. Measure RI of fractions (30 μ L from each fraction) starting with the lightest fraction (*see* Note 40).
- 2. For fractions where only GB was added, density of each fraction was estimated by weighing out $100 \ \mu L$ of each fraction medium three times on a microbalance (*see* Note 41).
- 3. Plot the RI and density values against each other and obtain a linear curve equation. Determine the densities of sample fractions using the RI values of each fraction in this equation.
- 2.3.3.7 DNA precipitation and quantification from fractions
 - 1. Add 2 volumes PEG and 1 μ L (25 μ g) linear polyacrylamide (LPA) to each fraction and mix by inverting the tubes (*see* Note 42).
 - 2. Incubate for 2 hours at room temperature to precipitate the DNA and centrifuge for 30 min at 20 °C. Discard supernatant.

- 3. Wash the DNA pellet with 500 µL cold 70% ethanol and centrifuge for 10 min at 20 °C.
- 4. Air-dry the pellet under a fume hood and elute DNA in 30 µL DEPC treated water.
- 5. Quantify DNA using Quant-iT PicoGreen dsDNA kit.
- 6. Plot DNA quantities against the fraction densities to analyze the outcome of SIP (Fig. 2B).
- 2.3.3.8 RNA precipitation and quantification from fractions
 - 1. Precipitate RNA from fractions by adding 1 volume of isopropanol and 1/5 volume of 7.5 M ammonium acetate each.
 - 2. Mix by inverting the tubes and incubate at room temperature for 20 min.
 - 3. Centrifuge the tubes for 1 hour at 4 °C. Discard supernatant.
 - 7. Wash the RNA pellet with 500 μ L cold 70% ethanol and centrifuge for 5 min at 4 °C.
 - 4. Air-dry the pellet under a fume hood and elute RNA in 15 μ L DEPC water.
 - 5. Quantify RNA using Quant-iT RiboGreen kit.
 - 6. Plot RNA quantities against the fraction densities to analyze the outcome of SIP (Fig. 2C).

2.3.4. Anticipated results

In our laboratory, we have successfully labeled both, DNA and RNA, of methylotrophic methanogens in marine sediment incubations using ¹³C-DIC as substrates. Labeling of RNA was strongest, when ¹³C-DIC was used as substrate as indicated by the larger amount of heavy RNA compared to ¹³C-methanol (Fig. 2C). Illumina sequencing of 16S rRNA showed that members of the *Methanococcoides*, a genus known to harbor methylotrophs, were predominantly present in heavy RNA from both ¹³C-DIC, and ¹³C-methanol; the occurrence of these *Methanococcoides* spp. in the light fractions of ¹³C-methanol incubations might be explained with their strong enrichment, however, the lower degree of labeling from ¹³C-methanol.

Labeling of DNA was much less pronounced regardless of labeled substrate used, as expected because of longer incubation times required for DNA replication and incorporation of label. Still, the power of PCR and Illumina sequencing of16S rRNA genes showed that *Methanococcoides* were the active methylotrophic methanogens in DNA-SIP (Fig. 3). Similar trends were observed in control incubations where no methanol was amended confirming the necessity to use ¹³C-DIC as a carbon source to track methylotrophic methanogens from environments.

In general, the total recovery of nucleic acids after density separation can be expected to be much lower than what was loaded into the gradient (between 10-30%); especially the labeled nucleic acids in the heavy fractions have a low yield as often only small amounts of labeled nucleic acids are formed, which affects recovery by nucleic acid precipitation substantially (*see* Note 42). To identify potentially active

methylotrophic methanogens, nucleic acids from one or more fractions can be pooled (to counteract low yields) and used as templates for downstream processes such as metagenomics and next generation sequencing.

2.3.5. Experimental run time

DNA- and RNA-SIP experiments (sections 3.2-3.3.8) for 6 samples can be performed ideally over the course of 5–6 days respectively. The breakdown is as follows: (i) nucleic acid extraction and quantification - 1 day; (ii) preparation of SIP samples and ultracentrifugation - 2 days for DNA-SIP, 3 days for RNA-SIP; (iii) gradient fractionation, RI and density measurements - 1 day; and (iv) nucleic acid quantification from fractions - 1 day. We recommend starting the ultracentrifuge runs at 16:00 h to ensure that it ends in the morning giving ample time to process the fractions on the same day. It takes 2–3 hours for the ultracentrifuge to stop as the deceleration is set to coast. Incubation times for your samples may vary based on activity rates of methylotrophic methanogens. Before starting the experiment, running test incubations using DIC and methanol and measuring CH₄ formation gives a good estimate of incubation time.

2.4. Notes

- 1. Chemicals and reagents can also be prepared using double distilled water instead of Milli-Q water if necessary.
- 2. Treatment with DEPC is useful for inactivating RNA nucleases. To prepare DEPC treated nuclease-free water, add 1 mL to 1000 mL Milli-Q water. Shake vigorously and incubate at 37 °C for 12 hours. Autoclave for 15 minutes to remove traces of DEPC and inactivate DNases. Filter sterilize through a 0.22 μm filter in a sterile bottle.
- 3. Avoid using cellulose or acetate based filter for sterilizing solutions as they can contaminate the solutions with unwanted carbon traces.
- 4. The protocol described here uses marine sediment samples but the labeling strategy can also be used with other samples, e.g., lake sediment, anoxic soil etc. The solution for preparing anoxic slurries has to be adjusted accordingly for the habitat type, e.g. marine, brackish, or freshwater.
- 5. Since sulfate reduction is more thermodynamically favorable over methylotrophic methanogenesis, ASW is prepared without the addition of sulfate to prevent the preferential enrichment of sulfate reducing microorganisms which are found in many environments.
- 6. Solutions and sediment slurry incubations in serum bottles are made anoxic by replacing the headspace gas with N₂ (99.999%) and removing dissolved O₂ from the solution by three repetitive cycles of gassing and vacuum. N₂ gas is introduced into the bottle by penetrating the butyl rubber

stopper with a sterile needle (0.8 x 40 mm, 21G) attached to a sterile filter (0.2 μ m, cellulose acetate) up to a pressure of max. 0.5 bar. Fitted to the same line via a 3-way stop cock with Luer lock adapters, vacuum is applied to remove gas from the headspace and the solution until no gas bubbles appear from within the solution (~ 5–10 minutes). Safety note: wear safety goggles at all time when working with compressed gases and glassware.

- 7. No vacuum is used while making methanol and DIC stock solutions anoxic as suggested in Note 6 due to their volatile nature.
- 8. The bottle is made anoxic by flushing with N₂, as mentioned in Note 6, but without the use of vacuum. Instead, another needle (0.55 x 25 mm, 24G) is introduced into the bottle through the rubber septa to allow the headspace air to be exchanged ("chimney method").
- 9. We collected marine sediment samples in the form of a 5 m long gravity core from the Helgoland Mud Area, North Sea, on board of the RV HEINCKE (Cruise 443; core HE443-077-1). This core was cut immediately on board into 1 m long sections, sealed and stored at 4 °C and brought to the lab. Finally, cores were sectioned into 25 cm sections which were homogenized and stored anoxically (see Note 6.) in 2.5 L jars at 4 °C until use.
- 10. Both PCI and CI are commercially available. Due to their hazardous nature, all steps involving these chemicals should be done under a fume hood along with the recommended protective gear (gloves, eye protection).
- Since PEG solution is highly viscous, it forms two phases after autoclaving: the lower dense PEG 6000 phase and the upper aqueous NaCl phase. Upon cooling, gently shake the bottle to mix these phases completely.
- 12. Kits are used as per manufacturer's instructions.
- 13. Dissolution of cesium chloride into water can be enhanced by continuous stirring on a magnetic stirring device set at a temperature between 50–100 °C. The solution is filtered in 30–50 mL stocks using a sterile needle (0.8 x 40 mm, 21G), 0.22 μm filter, and a 50-mL syringe. The solution is stored in butyl rubber stopper sealed serum bottles to prevent cesium chloride crust formation due to evaporation [18].
- 14. Storage of formamide at -20 °C prevents it from decomposing and lengthens its storage time. Storing in 2 mL aliquots allows for faster thawing times before use.
- 15. Disinfect all working benches and micropipettes with 70% ethanol before beginning any lab work. Treat working benches with RNaseZap before conducting RNA extraction procedures and working with RNA samples. Cool table top centrifuges to 4 °C before using them for DNA and RNA related work. All centrifugation steps in this section are carried out between 15,294–20,817 g at 4 °C unless specified.

- 16. Mixing of sediment samples with ASW helps to mimic *in situ* conditions and ensures homogenization of the sediment sample and the microbial community within. As the sediment samples are dense in nature, dilution (1:4 with ASW) also helps easier sub-sampling of the incubations for DNA and RNA extractions.
- 17. Pre-incubation of sediment slurries is helpful in activating the microbial community, which was surviving on minimum metabolic activity at 4 °C. Storing in dark prevents the growth of phototrophic microorganisms.
- 18. Having an estimate of indigenous DIC in the sample can be helpful in determining the amount of ¹³C-DIC that should be added to sediment slurry incubations. Since indigenous DIC is preferably assimilated by microbes over ¹³C-DIC, the latter should be added in excess in order to successfully label the active populations. In this study, supplementing 10 mM of ¹³C-DIC elevated the ¹³C-DIC level to 70% 85% based on GC coupled combustion isotope ratio mass spectrometer (GC-c-IRMS) measurements. Samples from different environments may have variable DIC concentrations. Thus, quantifying ¹³C-DIC in the slurry incubations after amending ¹³C-DIC is highly recommended.
- 19. We incubated the sediment slurries at 30 °C as the process of methanogenesis is thermodynamically favorable at mesophilic temperatures [19].
- 20. Preferably, stop the incubations immediately once the methanogenesis is complete (i.e. no increase of methane over 1–2 time points of measurement; Fig. 2A) to avoid cross-feeding of label and the degradation of primarily labeled microorganisms. Track changes in methane concentration in the headspace on a daily basis.
- 21. It can be difficult to obtain high amounts of labeled nucleic acids, especially RNA, from certain environmental samples due to many factors: low microbial activity rates, low starting biomass and loss of nucleic acids during extraction due to physical or enzyme based denaturation. We collected large amounts of slurry (up to 8 mL for DNA and RNA, respectively, per replicate treatment) from each biological replicate incubation to ensure that sufficient amounts of nucleic acids could be extracted. Here, nucleic acid extracts from all replicates were pooled together to obtain DNA and RNA in one tube in order to have sufficient amounts of nucleic acids. Otherwise, nucleic acids from biological replicates can be treated separately to check for community differences in parallel treatments.
- 22. Avoid vortexing after addition of TNS solution to prevent foaming from SDS. Make sure that there is at least 500 μ L headspace left after addition of TNS for efficient cell disruption during bead beating.

- 23. After bead beating, samples must always be placed on ice or in cold blocks (+4 °C) to prevent denaturation of DNA and RNA by exo- and endonucleases present within the samples.
- 24. Avoid strong vortexing of samples during the extraction of total nucleic acids as shearing forces can fragment DNA and RNA.
- 25. It is important to avoid disturbing the aqueous and organic phases formed after centrifugation on PCI treatment. Since phenol is partially miscible in the aqueous phase, traces of phenol originating from the organic phase can inhibit downstream processes such as PCR. Transferring small volumes of the aqueous phase using low-volume micropipette is useful to avoid uptake of phenol from the organic phase. If the aqueous phase after PCI treatment looks brownish due to high amount of organic content coming from the sediment samples, repeat the PCI treatment to obtain higher quality of nucleic acids.
- 26. Use of "cold" 70% ethanol improves DNA precipitation efficiency and ensures that the precipitated DNA is not denatured.
- 27. Air drying is done to remove residual ethanol. Preferably, remove traces of ethanol using a 100 μL micropipette to expedite drying. However, avoid "over-drying" (i.e. extended periods of drying), which renders resuspension of the nucleic acid pellet more difficult.
- 28. Dissolution of nucleic acids can be enhanced by slow and steady mixing of 50 μL nuclease-free water for 1–2 minutes using the micropipette. This prevents shearing of nucleic acids and ensures complete dissolution of the nucleic acids into the water. 50 μL of dissolved nucleic acids are transferred from one replicate tube to another and the dissolution procedure is repeated in order to pool all nucleic acids together.
- 29. PCR using general archaeal primer pairs such as Arc109f [20] and Arc912rt [21] must be conducted on DNase treated samples to confirm the absence of DNA in the samples. This is done to prevent DNA contamination into the RNA fractions after SIP. If the archaeal PCR yields a product, then repeat the DNase treatment to remove any trace of DNA within the sample. Generally, 2 μL of DNase treated sample can be used to detect the presence of DNA. Appropriate controls must be tested along with the samples: (i) negative control, (ii) positive control (e.g., genomic DNA of *Methanosarcina barkeri*), (iii) DNase kit control where DNase treatment was carried out without the addition of sample to detect post process contaminants and (iv) spiking of some DNase treated samples with genomic DNA of *Methanosarcina barkeri* to detect inhibition of PCR.
- 30. For every first ultracentrifuge run using a fresh stock of either CsCl or CsTFA, it is necessary to centrifuge a GB sample without any nucleic acid as a standard. RI and densities of fractions obtained from the GB standard are plotted against each other to obtain a standard linear curve

equation to quantify densities of samples. Accuracy of density measurements rely on calibrated micropipettes. To obtain a "gold standard", *E. coli* standards should be always used with every sample run to check the consistency of density separation. At a time, we recommend to process only 4–6 samples (including *E. coli* standard) due to the high number of fractions obtained after fractionation (i.e. at least 16 fractions/ sample). Up to 5 μ g DNA and 1.5 μ g RNA can be used for ultracentrifugation. Higher quantities of DNA and RNA than recommended can lead to improper separation of labeled and unlabeled DNA and RNA post ultracentrifugation.

- 31. Estimates of RI and densities of mixtures can be done by mixing different volumes of GB to fixed volumes of CsCl or CsTFA [22]. This information is useful for determining the starting RI and density, where the density is ~1.705 g/mL for DNA-SIP and ~1.8 g/mL for RNA-SIP. This ensures the complete retrieval of labeled and unlabeled DNA and RNA after isopycnic separation (Fig 2B, 2C).
- 32. Formamide is specifically added to RNA-SIP samples to break secondary structures formed within the RNA molecules which can affect their migration of RNA along the CsTFA gradient. Mix the RNA-GB mixture with formamide before mixing the RNA with CsTFA to prevent RNA precipitation.
- 33. Avoid filling the Quick Seal tubes completely as the tubes have to be heat sealed later. Leaving a headspace ensures easier removal of air bubbles which can be removed by gently tapping the tubes or turning the tubes up and down after sealing. Always place the tubes on a suitable stand that can hold the ultracentrifuge tubes, their respective adapters, and centrifuge lids in order to avoid damage and disturbance within the tubes, especially post ultracentrifugation.
- 34. Balancing of two tubes against each other should be done very accurately using a microbalance (least count: 0.00001 g). A small imbalance will prevent the ultracentrifuge from starting and can cause it to break down (if started) owing to the high centrifugation speeds. Take utmost care to not mix the tubes, their adapters and centrifuge lids that have been balanced against each other.
- 35. Pre-heat (~20 seconds) the tube topper before use. Place the tube topper on the opening of the Quick Seal tube only briefly to seal the tube. Quickly place a small metallic lid to cool the sealed tube.
- 36. Ultracentrifugation is carried out at maximum acceleration and deceleration is set to coast so that the slowing of the ultracentrifuge does not disrupt the gradient. CsTFA protects RNA from degradation over the duration of ultracentrifugation run at 20 °C.
- 37. From this step onwards, avoid shaking and disturbance of the ultracentrifuge tubes. Always place the tubes in their stands and avoid free handling of the tubes.

- 38. Pierce the needle into the ultracentrifuge tube in one stroke. This is done to prevent the formation of a bigger hole on top which can exert higher pressure within the tube thereby leading to a faster outflow of the liquid from the other end of the tube. Placing the tip of the needle on the opposing wall of the tube stabilizes it and prevents it from rotating inside the tube.
- 39. Refer to [18] for a video on SIP of nucleic acids.
- 40. Lighter fractions may contain some water and their RI will be much lower than the expected density range. Water containing fractions can be disregarded for further processing.
- 41. Instead of weighing 100 μ L from each fraction in a separate new tube, remove 100 μ L, record the difference, and put the sample back into the tube. Set microbalance to zero ("tare") and repeat estimation twice. As the microbalance is very sensitive and densities of the neighboring fractions are similar, pipetting should be done slowly, carefully and in the absence of any sudden movements.
- 42. LPA is a non-interfering agent which acts as a carrier for DNA precipitation and increases the efficiency of the process [18]. Owing to its dense nature, it also helps in locating the DNA pellet after precipitation.

2.5. Figures

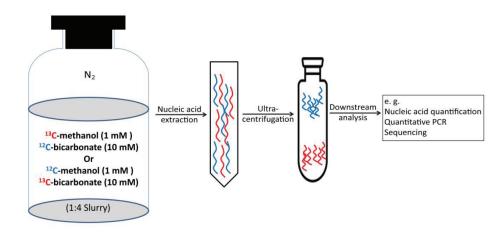


Figure 1. Workflow for DNA- and RNA-SIP for targeting methylotrophic methanogens in marine sediments.

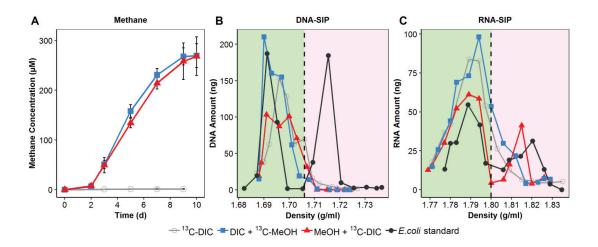


Figure 2. Isotope probing of methylotrophic methanogens in marine sediment incubations. (A) Kinetics of methane production over 10 days in sediment slurry incubations with different C1 substrates. (B) Isopycnic separation of unlabeled and labeled DNA and (C) RNA from sediment slurry incubations and controls (*E. coli*). Each data point represents the amount of DNA or RNA from a single fraction collected after density separation. Fractions containing unlabeled DNA or RNA are represented by the green area and fractions containing labeled DNA or RNA are represented by the pink area. The dotted line represents the density of the sample mixture before ultracentrifugation (**Note 31**).

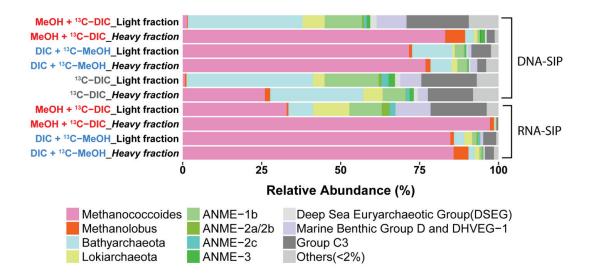


Figure 3. Total sum scaling of abundances of archaeal 16S rRNA gene sequences from selected light and heavy fractions of DNA- and RNA-SIP. Sequencing was done on an Illumina HiSeq 4000 platform as stated in [23]. Relative abundances of population < 2% were grouped together as "Others".

2.6. Table

Table 1. Sediment slurry incubation setup using different C-1 substrates, n=3 per treatment.

Incubation setup	Unlabeled methanol (1 mM)	¹³ C-methanol (1 mM)	Unlabeled DIC (10 mM)	¹³ C-DIC (10 mM)
¹³ C-DIC				Х
DIC + ¹³ C-MeOH		Х	Х	
MeOH + ¹³ C-DIC	Х			Х

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Chapter 3

CO₂ conversion to methane and biomass in obligate methylotrophic

methanogens in marine sediments

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Experimental concept and design	70%
Acquisition of experimental data	60%
Data analysis and interpretation	60%
Preparation of figures and tables	60%
Drafting of manuscript	50%

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Abstract

Methyl substrates are important compounds for methanogenesis in marine sediments but diversity and carbon utilization by methylotrophic methanogenic archaea have not been clarified. Here, we demonstrate that RNA-stable isotope probing (SIP) requires ¹³C-labeled bicarbonate as co-substrate for identification of methylotrophic methanogens in sediment samples of the Helgoland mud area, North Sea. Using lipid-SIP, we found that methylotrophic methanogens incorporate 60 to 86% of dissolved inorganic carbon (DIC) into lipids, and thus considerably more than what can be predicted from known metabolic pathways (~40% contribution). In slurry experiments amended with the marine methylotroph *Methanococcoides methylutens*, up to 12% of methane was produced from CO₂, indicating that CO₂-dependent methanogens is an alternative methanogenic pathway and suggesting that obligate methylotrophic methanogens grow in fact mixotrophically on methyl compounds and DIC. Although methane formation from methanol is the primary pathway of methanogenesis, the observed high DIC incorporation into lipids is likely linked to CO₂-dependent methanogenesis rates are much lower in marine sediments than under optimal conditions in pure culture, CO₂ conversion to methano is an important but previously overlooked methanogenic process in sediments for methylotrophic methanogens.

3.1. Introduction

Methanogenesis is the terminal step of organic matter mineralization in marine sediments [1]. There are three main pathways producing methane, i.e., hydrogenotrophic (H₂/CO₂), acetoclastic (acetate), and methylotrophic (e.g., methanol, methylamine, methoxylated benzoate) methanogenesis [2-4] with the former two pathways considered dominant. However, the importance of methylated compounds for methanogenesis in marine sediments has been acknowledged in recent years. Geochemical profiles and molecular analysis have shown that methylotrophic methanogenesis is the most significant pathway for methane formation in hypersaline sediments [5, 6] and in the sulfate reduction zone (SRZ) in marine environments [7, 8] where methanol concentration of up to 69 μ M had been measured [9, 10]. Especially in the SRZ, methylated compounds are regarded as non-competitive substrates for methanogenesis, since sulfate reducing microorganisms apparently do not compete with methanogens for these compounds [11, 12]; in addition, methylated compounds can be used by marine homoacetogens [13] however, in marine sediments evidence for this activity in competition with methanogens has not been obtained so far [14]. In sediments of the Helgoland mud area, specifically, high relative abundances of potential methylotrophic methanogens were observed [15] of which many are unknown. The potential for methylotrophic methanogenesis was recently even predicted from two metagenome-assembled genomes of uncultivated Bathyarchaeota, assembled from a shotgun metagenome [16].

The formation of methane via the three main pathways in methanogenic archaea has been studied intensively [17-19]. Much less is known regarding assimilation of carbon into biomass under *in situ* conditions and to which extent different carbon sources in the environment are utilized. Discrepancies between the predicted pathways known and the actual carbon metabolism measured appear to be based on 1) different cellular functions of carbon dissimilation and assimilation originated from reaction equilibria operative, 2) intermediate carbon cross utilization, and 3) interplay between different microbial communities [18, 20-23]. For example, mixotrophically growing cultures of *Methanosarcina barkeri* form their biomass equally from methanol and CO_2 , however, almost all the methane is formed from methanol rather than from CO_2 since methanol is disproportionated to methane and CO_2 according to the following reaction:

$$4 \operatorname{CH}_{3}\operatorname{OH} \xrightarrow{} 3 \operatorname{CH}_{4} + \operatorname{CO}_{2} + 2 \operatorname{H}_{2}\operatorname{O}$$
(1)

But apart from such culture studies using the nutritionally versatile *Methanosarcina barkeri*, the respective contribution of CO_2 and methylated carbon substrates to biomass formation during

methylotrophic methanogenesis, especially for "obligate" methylotrophic methanogens, in natural sediments has not been studied to date.

Nucleic acids (RNA ~20%, DNA ~3%, of dry biomass, respectively), lipids (7–9%) and proteins (50– 55%) are crucial cell components in living microorganisms [24], and thus, are suitable markers of carbon assimilation. In order to characterize carbon assimilation capabilities, stable isotope probing (SIP) techniques exist, among which RNA-SIP is very powerful for identifying active microorganisms based on separating ¹³C-labeled from unlabeled RNA using isopycnic centrifugation [25, 26]. In combination with downstream sequencing analysis, RNA-SIP provides high phylogenetic resolution in detecting transcriptionally active microbes [27, 28] but is limited in its sensitivity by requiring more than 10% of ¹³C incorporation into RNA molecules for separating ¹³C-labeled from unlabeled RNA [29]. To date, a number of SIP studies successfully detected methylotrophic bacteria [30-32] but the detection of methylotrophic methanogens by RNA-SIP with ¹³C labeled methyl compounds might be hampered by mixotrophic growth [33].

In contrast to RNA-SIP, lipid-SIP has a lower phylogenetic resolution, but can detect very sensitively δ^{13} C-values in lipid derivatives by gas chromatography combustion isotope ratio mass spectrometry (GC-c-IRMS), thereby facilitating quantitative determination of small amounts of assimilated carbon [34, 35].

In this study, we aimed to identify methylotrophic methanogens by RNA-SIP and elucidate carbon assimilation patterns in marine sediments. We hypothesized that the large pool of ambient dissolved inorganic carbon (DIC) in sediments [7] alters carbon utilization patterns in methylotrophic methanogens compared to pure cultures. To address this hypothesis, we tracked carbon dissimilation into methane and quantified assimilation into lipids by lipid-SIP in slurry incubations and pure cultures. In contrast to known pathways, we found a high degree of methane generation from DIC during methylotrophic methanogenesis by obligate methylotrophic methanogens, i.e., using only methyl groups for methane formation. This mixotrophic methanogenesis from both, methanol and DIC, might be the basis for our observation that more inorganic carbon was assimilated into biomass than could be expected from known pathways.

3.2. Materials and Methods

3.2.1. Sediment incubation setup for SIP

Sediment was collected from the Helgoland mud area (54°05.23'N, 007°58.04'E) by gravity coring in 2015 during the RV HEINCKE cruise HE443. The geochemical profiles were previously described [15].

Sediments of the SRZ (16–41 cm) and MZ (238–263 cm) from gravity core HE443/077-1 were selected for incubations; typically, sulfate concentration for SRZ sediment is in the range of ~3-25 mM and for MZ sediment is below the detection threshold (~50 μ M) as reported in Oni et al. 2015 [15]. Anoxic slurries (1:4; w/v) were prepared by mixing sediments with sterilized artificial sea water without sulfate [36]. Slurries of 50 mL were dispensed into sterile 120-mL serum bottles and sealed with butyl rubber stoppers. Residual oxygen was removed by exchanging bottle headspace 3 times with N₂ gas. A 10-day pre-incubation was performed, followed by applying vacuum (3 min at 100 mbar) to remove most of the headspace CO₂. Triplicate incubations were conducted by supplementing 1 mM ¹³C-labeled methanol (~33 mg L⁻¹ slurry) and unlabeled 10 mM sodium bicarbonate (~610 mg L⁻¹ slurry), or 1 mM unlabeled methanol and 10 mM ¹³C-labeled sodium bicarbonate (¹³C-labeled substrates provided by Cambridge Isotope Laboratories, Tewksbury, Massachusetts, USA) at 10 °C. The proportion of ¹³C dissolved inorganic carbon (DIC) was determined by GC-c-IRMS.

3.2.2. Pure culture setup

The carbon assimilation patterns were compared between SIP sediment incubations and the obligate methylotrophic methanogen, *Methanococcoides methylutens*. *M. methylutens* strain MM1 (DSM 16625) was obtained from the German Collection of Microorganisms and Cell Cultures (DSMZ, Braunschweig, Germany). Initial cultivation was performed using Medium 280 according to DSMZ protocols. After several transfers of the culture in anoxic marine Widdel medium [37], 5% of the culture were inoculated into fresh Widdel medium supplemented with 30 mM methanol, trace element solution SL 10 [38], and 50 mM sodium bicarbonate (i.e., DIC) with carbon sources containing 5% of ¹³C-label. Pure cultures were grown at 30 °C in triplicates.

3.2.3. Slurry incubations inoculated with M. methylutens

To test methanogenesis from CO₂, incubations were performed with *M. methylutens* in autoclaved (n=3) slurry from the SRZ with different amendments of electron donor (H₂), electron shuttles (humic acid; anthraquinone-2,6-disulfonic acid - AQDS), and electron acceptors/electron conductors (hematite, α -Fe₂O₃; magnetite, Fe₃O₄; Lanxess, Germany). Incubations were separately prepared with 50% H₂ in headspace, 100 μ M AQDS, 30 mM magnetite, 30 mM hematite and 500 mg L⁻¹ humic acid (Sigma-Aldrich, Steinheim, Germany). The pure culture (5%) was inoculated into these setups, and amended with 20 mM unlabeled methanol and ~10% of ¹³C-labeled DIC (1 mM) for measuring carbon partitioning into methane. The control incubation comprised autoclaved slurry, 50% H₂ and 20 mM methanol without

addition of *M. methylutens*. All experiments were setup with a total volume of 50 mL in 120-mL serum bottles sealed with butyl rubber stoppers, and incubated at 30 $^{\circ}$ C in triplicates.

3.2.4. Gas analysis

The concentration of methane in the headspace was measured by gas chromatography as previously described [39]. Headspace H_2 was determined with a reduction gas detector (Trace Analytical, Menlo Park, California, USA). Gas samples of 100 μ L and 1 mL from triplicate bottles were used for measuring methane and H_2 , respectively. The parameters were as follows: carrier gas (nitrogen) 50 mL min⁻¹, injector temperature 110 °C, detector 230 °C, column (Porapak Q 80/100) 40 °C.

The δ^{13} C values of methane and CO₂ in the headspace, DIC as well as total dissolved inorganic carbon (TIC) in slurries were determined using a Thermo Finnigan Trace GC connected to a DELTA Plus XP IRMS (Thermo Scientific, Bremen, Germany) as described previously [40]. Prior to analyses of δ^{13} C-DIC and -TIC, 1 mL of supernatant or slurry was converted to CO₂ by adding 1 mL phosphoric acid (85%, H₃PO₄) overnight at room temperature.

3.2.5. Nucleic acids extraction, quantification and DNase treatment

The nucleic acids were extracted according to Lueders et al. [41]. Briefly, 2 mL of wet sediment without supernatant from biological triplicates was used for cell lysis by bead beating, nucleic acid purification by phenol-chloroform-isoamyl alcohol extraction and precipitation with polyethylene glycol. For the RNA extract, DNA was removed by using the RQ1 DNase kit (Promega, Madison, Wisconsin, USA). DNA and RNA were quantified fluorimetrically using Quant-iT PicoGreen and Quant-iT RiboGreen (both Invitrogen, Eugene, Oregon, USA), respectively.

3.2.6. Isopycnic centrifugation, gradient fractionation and reverse transcription

Isopycnic centrifugation and gradient fractionation were performed according to the previously described method with modifications [41]. In brief, 600 to 800 ng RNA from biological replicates (n=3) was combined and loaded with 240 μ L formamide, 6 mL cesium trifluoroacetate solution (CsTFA, GE Healthcare, Buckinghamshire, UK) and gradient buffer solution. RNA was density separated by centrifugation at 124 000 ×*g* at 20 °C for 65 h using an Optima L-90 XP ultracentrifuge (Beckman Coulter, Brea, California, USA). As standard, a mixture of equivalent amounts of fully ¹³C-labeled and unlabeled *E. coli* RNA was used in density separation for defining heavy and light gradient fraction

density ranges. RNA was quantified and reverse transcription was conducted using the high capacity cDNA reverse transcription kit (Applied Biosystems, Foster City, California, USA).

3.2.7. Quantitative PCR (qPCR)

Archaeal 16S rRNA and *mcrA* genes were quantified from each biological replicate (n=3) using primer sets 806F/912R and ME2 mod/ME3'Fs 1011 (Table S1), respectively; *mcrA* encodes the alpha subunit of methyl coenzyme M reductase, a key enzyme of methanogenic and methanotrophic archaea [42]. Standard curves were based on the 16S rRNA gene of *M. barkeri* and the *mcrA* gene clone A4-67 for archaea and methanogens, respectively. The setup of PCR reaction was described previously [36]. The qPCR protocol comprised an initial denaturation for 5 min at 95 °C and 40 cycles amplification (95 °C for 30 sec, 58 °C for 30 sec and 72 °C for 40 sec). The detection thresholds were 100–1 000 gene copies with an efficiency of 90–110%.

3.2.8. Sequencing and bioinformatics analysis

Based on the RNA-SIP profiles of *E.coli* standard RNA (Figure S1) and a previously reported density shifts in SIP fractions [41], "heavy" (1.803–1.823 g mL⁻¹, combination of fraction 3, 4, and 5) and "light" (1.777–1.780 g mL⁻¹, fraction 11) fractions of RNA-SIP samples were selected. Library construction and sequence read processing were as described previously [39].

3.2.9. Lipid analysis

Total lipids were extracted from ~4 g of freeze-dried sediment samples from single labeling incubations (one substrate labeled, the other unlabeled) using a modified Bligh-Dyer protocol [43]. Intact polar archaeal ether lipids were purified by preparative high performance liquid chromatography with fraction collection according to the method by Zhu et al. [44]. Considering the origin and complexity of sediment samples and similar proportion of carbon atoms in lipid molecules (archaeol and hydroxyarchaeol), phytenes, biphytane, and biphytanes containing cycloalkyl rings (Figure 3C, Figure S2) were obtained from the intact archaeal lipid fraction [45]. The detailed chromatographic and mass spectrometric parameters were described by Kellermann et al. [46].

3.2.10. δ^{13} C calculation

The proportion of methane from DIC (f_{DIC/CH_4}) was calculated based on the fractional abundance of ¹³C (¹³F) of methane, methanol (MeOH) and DIC in the incubation with ¹³C-DIC and MeOH. According to a

two-end member model, DIC and MeOH are two main carbon sources for methane production expressed as follows:

$$f_{DIC/CH_4}{}^{13}F_{DIC} + (1 - f_{DIC/CH_4}){}^{13}F_{MeOH} = {}^{13}F_{CH_4}$$
(Eq.1)
$$f_{DIC/CH_4} = \frac{{}^{13}F_{CH_4} - {}^{13}F_{MeOH}}{{}^{13}F_{DIC} - {}^{13}F_{MeOH}} \times 100\%$$
(Eq.2)

where ¹³*F* is obtained from the δ notation according to F = R/(1+R) and R = (δ /1000+1) * 0.011180 [47]. ¹³*F*_{CH₄} and ¹³*F*_{DIC} were the fractional ¹³C abundance of methane and DIC at harvest time, and ¹³*F*_{MeOH} that of MeOH in the medium at the start.

¹³C label incorporation ratios from MeOH or DIC in single labeling experiments were calculated from the ¹³C abundance increase relative to the ¹³C label strength via Eq. 3 and Eq. 4. X_{MeOH} and X_{DIC} signify the ¹³C incorporation ratio from MeOH and DIC, respectively. ¹³ $F_{t_{end}}$ and ¹³ F_{t_0} are the ¹³C fractional abundance of lipids harvested at t_{end} and t₀.

$$X_{MeOH} = \frac{{}^{13}F_{t_{end}} - {}^{13}F_{t_0}}{{}^{13}F_{MeOH}}$$
(Eq.3)
$$X_{DIC} = \frac{{}^{13}F_{t_{end}} - {}^{13}F_{t_0}}{{}^{13}F_{DIC}}$$
(Eq.4)

Given that the single labeling incubations were conducted with the same treatment, i.e., 1 mM methanol and 10 mM DIC, the relative proportion of DIC for lipids biosynthesis ($f_{DIC/lipid}$) was estimated from the ¹³C incorporation ratios (X_{MeOH} and X_{DIC}) in these single labeling incubations as follow:

$$f_{DIC/lipid} = \frac{X_{DIC}}{X_{DIC} + X_{Me}}$$
(Eq.5)

3.3. Results

3.3.1. Methylotrophic methanogenesis and increase in methanogenic archaea

In order to examine carbon labeling into RNA and lipids of methylotrophic methanogens in anoxic marine environments, sediment slurries amended with or without ¹³C-methanol (1 mM) and ¹³C-DIC (10 mM) were incubated at 10 °C. Sediment incubations from the zones of sulfate reduction (SRZ) and methanogenesis (MZ) showed a divergent methane production rate, i.e., methanogenesis finished after 40 and 20 days, respectively (Figure 1A). In incubations amended with DIC and ¹³C-methanol, carbon recovery from methanol of ~80% was measured from both sediment incubations (Table S2). Amended

¹³C-DIC was diluted into the sediment endogenous DIC pool to about 70–84%, which was more obvious in samples from MZ than SRZ (Table 1). In SRZ sediment incubations with ¹³C-DIC and unlabeled methanol, up to 10.3% of methane originated from ¹³C-DIC (Table 1).

The dynamics of the archaeal communities in all incubations was tracked by qPCR of archaeal 16S rRNA genes and *mcrA* genes after methanogenesis ceased (Figure 1B). Archaeal and *mcrA* gene copy numbers increased strongly by 10–14 and 19–30 times for all treatment incubations respectively, while gene copies in control incubations were not elevated (Figure 1B).

3.3.2. Carbon assimilation into RNA and identification of metabolically active archaea

In preliminary sediment incubations, SIP experiments with ¹³C-methanol had shown that RNA could not be labeled to a sufficiently high extent to become detectable in heavy gradient fractions (e.g., >1.803 g mL⁻¹) after isopycnic separation of RNA. Contrastingly, methanol dissimilation was strong and archaeal and *mcrA* gene copies increased compared to that on day 0 and ¹³C-DIC control, likewise indicating that methylotrophic methanogens were active (Figures 1A and B, Table S2). Because of mixotrophic assimilation capabilities in methylotrophic methanogens, i.e., utilizing methylated compounds and DIC, a series of SIP slurry experiments were conducted with combinations of methanol and DIC in order to improve the sensitivity of RNA-SIP: double ¹³C-label (methanol + DIC), single ¹³C-label (one of the substrates labeled), both substrates unlabeled, and a ¹³C-DIC control (Figure 2, Figure S3). After density separation of RNA, different degrees of RNA labeling were detected in isotopically heavy gradient fractions, e.g., >1.803 g mL⁻¹ (Figure 2). Strongest ¹³C-labeling, as indicated by largest amounts of RNA found in gradient fractions > 1.803 g mL⁻¹, was detected in RNA from incubations with double ¹³Clabeling (Figure 2A), followed by single-label incubations with ¹³C-DIC. For single-label ¹³C-methanol incubations, however, RNA fraction shifts according to density were minor compared to unlabeled incubations.

In order to estimate ¹³C-labeling levels of methanogens in single SIP experiments (¹³C-methanol or ¹³C-DIC), a series of molecular techniques were applied including qPCR of cDNA in heavy fractions of RNA-SIP samples, archaeal 16S rRNA sequencing from RNA-SIP fractions and δ^{13} C value determination of methanogen lipids, e.g., phytanes derived from intact polar archaeol-based molecules. In incubations amended with ¹³C-DIC and unlabeled methanol, archaeal gene copies were substantially higher than that of ¹³C-DIC control and ¹³C-methanol incubations (Figure 2B). Up to 49 000-fold more RNA molecules were present in the heavy fraction (i.e., 1.803–1.823 g mL⁻¹) compared to the incubation amended with unlabeled DIC and ¹³C-methanol (Table S3). Correspondingly, Illumina sequencing of RNA revealed that sequences identified as related to the genera *Methanococcoides* were dominant in SRZ sediment

incubations, and the methylotrophic methanogens Methanococcoides and Methanolobus spp. were more dominant in MZ sediment incubations. In contrast to heavy fractions, the abundance of methanogens in the light fractions from double labeling incubations, i.e., 13 C-(DIC + methanol) were lowest (~30-60%) (Figure S3), followed by the incubations amended with methanol and 13 C-DIC (~50-70%) (Figure 2C). SIP incubations amended with DIC and ¹³C-methanol harbored the highest relative abundance of methanogens in the light fraction, which ranged in abundance from 80 to 90% of total archaea (Figure 2C). Light fractions were overall mainly composed of anaerobic methanotrophic (ANME) archaea, Bathyarchaeota and Lokiarchaeota except for methylotrophic methanogens (Figure 2C). For ¹³C-DIC control incubations, abundances of methanogens were low in SIP samples (Figure S3). For unlabeled methanol and DIC incubations, given the low amount of labeled RNA in heavy fractions, no amplicons were obtained, but light fractions showed a high abundance of methylotrophic methanogens (Figure S3). Classifications were confirmed by phylogenetic clustering of cloned 16S rRNA gene fragments (about 800 base pairs) with OTU sequences representing Methanococcoides and Methanolobus spp. (Figure S4). Sequences of these methanogens accounted for more than 97% of total archaea in heavy gradient fractions. However, known hydrogenotrophic methanogens were undetectable (Figure 2C) although 3 and 10% of methane was formed from DIC in incubations with MZ and SRZ sediments, respectively (Table 1).

In parallel to RNA-SIP, lipid-SIP incubations with SRZ sediment slurries demonstrated δ^{13} C values of phytane and phytenes being more positive in ¹³C-DIC and unlabeled methanol treatment than that in ¹³C-methanol amendments, while the opposite was found in MZ sediment incubations (Figure 3A). After elimination of ¹³C-DIC dilution effects by ambient inorganic carbon, DIC contributions to lipids ranged from 59.3% to 86.1% in SRZ sediment incubations, which was constantly higher than that of MZ sediment incubations (52.7% to 56.4%).

To understand how carbon is assimilated into lipids by methylotrophic methanogens, pure culture incubations of *M. methylutens* were performed with 5% of the ¹³C-labeled substrates (i.e., DIC or MeOH) and the dominating archaeal lipids archaeol (AR) and hydroxyarchaeol (OH-AR) were directly analyzed without cleavage. In contrast to sediment incubations, lipids showed lower $f_{\text{DIC/lipid}}$ (~49%) based on the carbon incorporation in single labeling incubations (Figure 3B).

3.3.3. Methane formation from DIC during methylotrophic methanogenesis

The high proportion of methane formed from DIC in methanol amended sediment slurry incubations (Table 1) prompted us to investigate the underlying mechanism in more detail. Thus, autoclaved sediment slurries were used as a surrogate of natural sediment, but with all microorganisms killed, and inoculated

with the obligate methylotroph *M. methylutens*. Hematite and magnetite known to serve as electron acceptors or conductors [48, 49] were added along with humic acid, and AQDS as electron shuttles, as well as an additional electron donor (H₂), which are all known to stimulate methanogenesis [48-51]; certain methylotrophic methanogens, e.g., *Methanomassiliicoccales* spp., require hydrogen for methanogenesis [52, 53].

Methane concentrations in incubations with hematite and humic acid were higher than that of the other incubations after 7 days (Figure 4A). Although methane production rates were low, methane proportions from DIC in treatments with *M. methylutens* alone, H₂, AQDS and magnetite were much higher ($f_{DIC/CH4}$, ~10%) than that in incubations with hematite and humic acid (~2%) (Figure 4B). Linear regression showed a strong correlation between methane production rate and CO₂-dependent methanogenesis by methylotrophic methanogenes on day 3 and 5 of the incubations, indicating that lower methanogenesis rates triggered higher levels of methane formation derived from ¹³C-DIC (Figure 4C).

3.4. Discussion

In this study, we utilized RNA-SIP employing ¹³C-DIC and methanol and successfully identified methylotrophic methanogens in both, SRZ and MZ sediments of the Helgoland mud area in the North Sea. We demonstrated that the addition of ¹³C-DIC is necessary to detect label in RNA of methylotrophic methanogens rather than using ¹³C-methanol as energy substrate alone. We further evaluated carbon utilization patterns of the methylotrophic methanogens by lipid-SIP and identified a high DIC assimilation into characteristic lipids within the SRZ sediment. Isotope probing experiments revealed that up to 12% of methane was formed from DIC by the "obligate" methylotrophic methanogen, *M. methylutens*, thereby suggesting an explanation for the elevated DIC incorporation into biomass.

3.4.1. Carbon assimilation by methylotrophic methanogens in sediment incubations

Nucleic acids-SIP techniques depend on ¹³C-labeling levels of DNA or RNA molecules, from which carbon assimilation can be reconstructed and compared to the known pathway of nucleic acid biosynthesis from methyl-groups in methanogens [54-59]. The current pathways show that only one carbon atom stems from methanol in ribose-5-phosphate while 25% to 40% of carbon in nucleobases originates from the methyl carbon of the substrate (Figure 5). This is corroborated by our RNA-SIP experiments using ¹³C-labeled methanol alone, but RNA was not found to be labeled effectively enough for density separation and further sequence analysis. However, by additionally using ¹³C-DIC, we found high 16S rRNA copy numbers (Figure 2B) and a high representation of known methylotrophic methanogens (Figure 2C) in the heavy RNA gradient fractions, successfully recovering ¹³C-labeled RNA

of methylotrophic methanogens in the SRZ and MZ sediments of the Helgoland mud area. Combined with downstream analysis including qPCR, 16S rRNA sequencing and cloning, we directly show that members of the genus *Methanococcoides* were the predominantly active methylotrophic methanogens in SRZ incubations, while *Methanococcoides* together with *Methanolobus* were dominant in MZ incubations. In addition, archaea with an abundance less than 0.01% showed a higher proportion in light fractions than in heavy fractions (Table S4), excluding the populations under high-sensitivity SIP conditions [60]. A small peak at 1.808 g mL⁻¹ was detected in RNA-SIP profiles from the SRZ incubations amended with ¹³C-methanol and unlabeled DIC, which originated most likely from methylotrophic methanogens as shown by relative abundances of methanogens in the heavy fractions (Fig. 2C). However, at this density, RNA was partially labeled only because of the lower contribution of methanol carbon to nucleic acid biosynthesis, which resulted in lower RNA amounts in heavy fractions than that of ¹³C-DIC and unlabeled methanol treatment. Consequently, RNA labeling will be more effective in methylotrophic methanogenic archaea by using DIC than by methanol.

The main reactions of inorganic carbon assimilation are the generation of acetyl-CoA and pyruvate, respectively (Figure 5). In principle, the generated CO_2 from methanol (Reaction 1) can be utilized for biomass synthesis but is in exchange with the large pool of ambient CO_2 (at least 10 mM in our experiments, up to 40 mM in marine sediment [7]. Thus, the methane formed by reduction of CO_2 will be largely recruited from ambient, unlabeled CO_2 molecules [61]. Hence, addition of ¹³C-labeled DIC or a combination of both substrates labeled enables tracking of methylotrophic methanogens via RNA-SIP techniques. For carbon assimilation into nucleic acids of these methanogens, both proposed biosynthesis pathway of nucleic acid and labeling strategy of RNA-SIP confirmed inorganic carbon as the main carbon source for nucleic acids.

Because of its proven accuracy, lipid-SIP was used for the relative quantification of carbon assimilation into biomass. In lipid-SIP analysis, we evaluated ¹³C-incorporation into intact polar archaeol- and hydroxyarchaeol diether molecules, which are the dominant lipids produced by moderately thermophilic methanogenic archaea [62-64], via phytane and phytene side-chain analysis (Figure 3). These moieties were the only ones being ¹³C-labeled while tetraether-derived biphytane and cycloalkylated biphytanes as indicators of archaea such as Thaumarchaeota [65], anaerobic methanotrophs [66, 67] or Bathyarchaeota [68] did not show a ¹³C incorporation (Figure S5). This was corroborated by our sequencing results demonstrating that methylotrophic methanogens were the dominant archaea in the heavy fractions and that the relative abundances of other archaea were very low or even below detection (Figure 2C).

By evaluating ¹³C incorporation into methanogen-derived phytane and phytenes, lipid-SIP provides insight into methanogen activities and carbon utilization. As the main precursors of ether lipids in archaea, biosynthesis of isopentenyl diphosphate (IPP) and dimethylallyl diphosphate (DMAPP) proceeds via the modified mevalonate pathway [69-71]. In this pathway, mevalonate-5-phosphate is decarboxylated to IPP, in which three out of five carbon atoms are derived from methanol (Figure 6). DMAPP is further converted to geranylgeranyl diphosphate (GGPP), which receives 60% of its carbon from methanol, suggestive of a lower DIC contribution to isoprenoid chains than methanol. This was supported by the fact that archaeol and hydroxyarchaeol contained more methanol-derived than DIC-derived carbon using a pure culture of *M. methylutens* (Figure 3B). However, unlike the proposed lipid biosynthesis pathway and the pure culture, clearly more DIC was assimilated into lipids than methanol in both sediment incubations, which was most prominent in the sediment from the SRZ (Figure 3A). We, moreover, detected that ~10% of methane produced was derived from DIC during the SIP experiments and using M. methylutens in autoclaved sediment slurry incubations (Table 1, Figure 4). Because of the reversibility of all reactions from CO_2 to methyl-tetrahydrosarcinapterin (CH₃-H₄SPT) [23], it is very likely that part of the DIC is converted to CH₃-H₄SPT. Thus, CH₃-H₄SPT generated from CO₂ will be available for lipid biosynthesis (Figure 6) leading to the ¹³C-enrichment of the lipid pool observed (Figure 3).

3.4.2. CO₂ reduction to methane by obligate methylotrophic methanogens

There are two types of CO₂-dependent methanogenesis: 1) Hydrogenotrophic methanogenesis [2, 4]. These methanogens contain F_{420} -reducing [NiFe]-hydrogenase to catalyze F_{420} reduction by H₂ [72]. 2) Mediation by interspecies electron transfer between bacteria and some members of the *Methanosarcinales*. CO₂ reduction to methane was observed in *Methanosaeta* and *Methanosarcina* during syntrophic growth with *Geobacter* species on alcohols (ethanol, propanol, and butanol), as electrons generated from *Geobacter* are directly transferred to methanogens to reduce CO₂ [73-76].

Based on our SIP incubations with SRZ sediment showing 10% of methane generation from DIC at low H_2 partial pressure (< 0.3 Pa) (Table 1) and the overall lack of hydrogenotrophic methanogens in RNA-SIP fractions (Figure 2C), we argue that H_2 -dependent methanogenesis does not play a role [2]. Similarly, in autoclaved sediment slurry (Figure 4B), the "obligate" methylotroph *M. methylutens* generated methane from CO₂ without a hydrogen (or electron) supplying partner microorganism.

Members of the genus *Methanococcoides* are considered as obligate methylotrophic methanogens since no F_{420} -reducing [NiFe]-hydrogenase was detected in their genomes [69, 77] ruling out hydrogenotrophic methanogenesis in sediment incubations. Nevertheless, part of the methane formed during methylotrophic methanogenesis by *M. methylutens* was from CO_2 , especially when methane production rates were low (Figure 4C). Apparently, at high rates of methanol dissimilation to CO_2 , the reverse pathway of CO_2 reduction to methane was outcompeted. Higher rates of methylotrophic methanogenesis can be achieved potentially by amendments in autoclaved slurries using hydrogen as electron donor, electron conductors (hematite, magnetite) and electron shuttles (humic acid, AQDS); in our incubations, we found humic acid and hematite most strongly stimulating methylotrophic methanogenesis. Although the underlying mechanism is beyond the scope of the current study, methylotrophic methanogens in our incubations could take advantage of hematite as potential electron conductor [48, 49] or humic acid as electron shuttle [51] as indicated by a higher rate of methanogenesis compared to the other treatments (Figure 4A).

It has been shown that 3% of methane was produced from CO_2 during methylotrophic methanogenesis of *Methanosarcina barkeri* (i.e., a facultative methylotroph) without the addition of H₂ [33], which is similar to about 2.5% of methane generated from CO_2 by *M. methylutens* (i.e., "obligate" methylotroph) in our study (Table S5). However, in SRZ sediment incubations the rate of methane production was lower than in MZ incubations, which resulted in a high proportion of methane generated from CO_2 (10%) (Table 1, Figure 1). Furthermore, CO_2 conversion to methane linked inorganic carbon assimilation into lipids, highlighting the importance of the activity of concomitant CO_2 reduction during methylotrophic methanogenesis rate are likely lower than those in pure cultures because of the limitation in methylated substrates [7, 18], strongly suggesting that methane generation from CO_2 by obligate methylotrophic methanogenesis is underestimated under *in situ* conditions. Thus, CO_2 conversion to methane sediments of *in situ* methylotrophic methanogenesis in marine sediment.

In summary, we have shown that ¹³C-DIC is required as co-substrate for successful identification of methylotrophic methanogens by RNA-SIP in marine sediments. DIC is the main carbon source for biosynthesis of nucleic acids in these methanogens and thus using ¹³C-methanol as energy and carbon substrate alone is insufficient in SIP experiments (Figure 5). Given the intricacies of known assimilatory pathways in methanogenic archaea as a functional group, it might be necessary to at least check for the possibility of DIC as a main assimilatory carbon component in all methanogens for successful SIP experiments. In general, it seems that archaea have a propensity for using DIC as a carbon source for assimilation [68, 78], possibly as an evolutionary adaptation to environments with limited availability of organic carbon [79]

But beyond known pathways, we detected an unexpectedly high amount of methane (> 10%) formed from DIC. Especially in SRZ incubations, the lower methane production rates resulted in increased CO₂ conversion to methane (~10%), which is linked to CO₂ conversion assimilation. This finding strongly suggests that the alleged obligate methylotroph studied here was rather mixotrophically converting both available substrates (DIC, methanol) to methane. Our detailed labeling studies showed that the kinetics of substrate utilization apparently is a decisive factor in channeling more or less CO₂ into the pathway of methanogenesis: more methane formed from CO₂ when the overall kinetics were slow, and vice versa. From an ecological perspective, DIC is a much more pertinent substrate than methanol (or other methyl compounds) in marine sediments [5, 7], and thus, we speculate that more DIC reduction by obligate methylotrophic methanogens occurs *in situ* than is currently known. A larger proportion of methane formed from DIC in methylotrophic methanogens should also impact interpretation of δ^{13} CH₄ values and associated carbon isotope fractionations, which might be overprinted by such mixotrophic methanogenesis. Thus, the CO₂ reduction to methane and assimilation into biomass by obligate methylotrophic methanogens plays a much more important role in the environment than was previously known.

3.5. Acknowledgments

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3.6. Conflict of interest

The authors declare that they have no conflict of interest.

3.7. Data availability

Sequencing data have been submitted to GenBank Short Reads Archive with accession numbers from SRR8207425 to SRR8207442.

3.8. Table

Table 1. ¹³C fractional abundance and H_2 partial pressures in SIP incubations. Data is presented as average values (n = 3).

Sediment	Substrates	$^{13}F_{DIC}$ (%)	f_{DIC/CH_4} (%) ^a	$H_2 (Pa)^b$	Incubation time (d)
SRZ	DIC + ¹³ C-MeOH	3.7 ± 0.4	89.3 ± 0.3	NA	43
MZ	$DIC + {}^{13}C-MeOH$	3.0 ± 0.0	96.4 ± 0.1	NA	19
SRZ	$MeOH + {}^{13}C-DIC$	83.6 ± 0.6	10.3 ± 0.2	0.1 ± 0.1	43
MZ	$MeOH + {}^{13}C-DIC$	69.8 ± 0.7	3.4 ± 0.1	0.3 ± 0.0	19

^aMethane proportion from DIC (f_{DIC/CH_4}) in "methanol + ¹³C-DIC" incubations was based on Eq.2. ^bH₂ partial pressure was measured on day 23 and 16 for incubation SRZ and MZ sediments, respectively. NA, not analyzed.

3.9. Figures

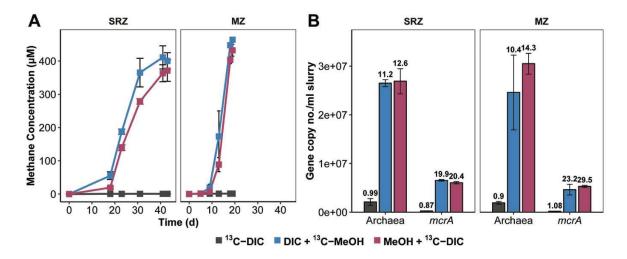


Figure 1. Dynamics of methane formation and archaeal populations in stable isotope probing (SIP) incubations with SRZ and MZ sediment samples. (A) Methane concentrations in SIP incubations. Methane data is presented as average values (n = 3, error bar = SD). (B) Gene copy numbers of archaea (16S rRNA genes) and methanogens (*mcrA* gene). Gene copies were quantified based on DNA extracts at harvest. Fold increase of gene copies were indicated above each histogram by comparing gene copies on day 0 after preincubation (n = 3, error bar = SD). DIC: dissolved inorganic carbon, i.e. bicarbonate; MeOH: methanol.

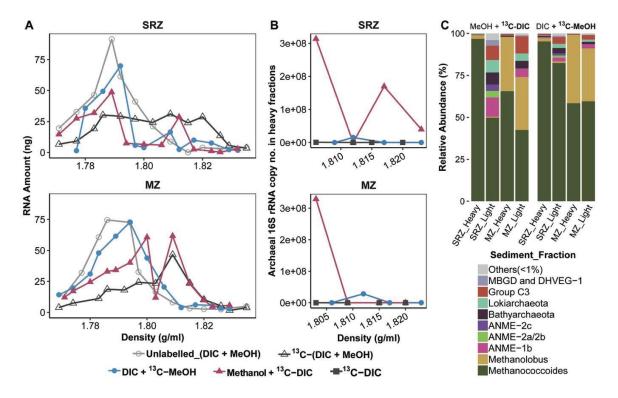


Figure 2. Density distribution of RNA, gene copy numbers, and community composition from SIP incubations with SRZ and MZ sediment after isopycnic separation. (A) RNA profiles from different RNA-SIP experiments. (B) Gene copy numbers of archaeal cDNA in heavy fractions (1.803 g mL⁻¹ to 1.823 g mL⁻¹) from RNA-SIP experiments. Archaeal gene copy numbers refer to the absolute abundance of 16S rRNA gene copies in cDNA from gradient fractions. (C) Relative abundances of density separated archaeal 16S rRNA from single-labeling incubations in light (1.771 g mL⁻¹ to 1.800 g mL⁻¹) and heavy (1.803 g mL⁻¹ to 1.835 g mL⁻¹) gradient fractions.

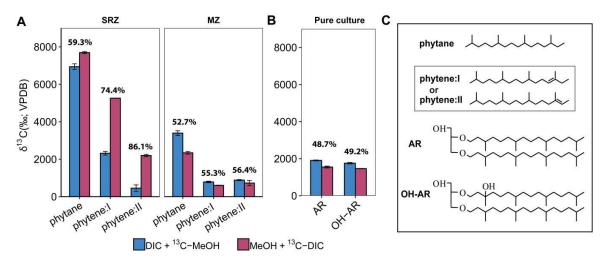


Figure 3. Lipid-SIP experiments from sediment incubations (natural community) and pure cultures in Widdel medium. Lipid δ^{13} C values were measured in homogenized samples after methanogenesis had ceased. (A) δ^{13} C values of phytanes in sediment incubations with 70% ¹³C-DIC. Phytane originates from intact polar archaeol lipids, phytenes (phytene I and phytene (II) derive from intact polar hydroxyarchaeol lipids. $f_{\text{DIC/lipid}}$ are indicated on the top of bars from single labeling incubations based on Eq.5. (B) δ^{13} C values of archaeol (AR) and hydroxyarchaeol (AR-OH) in pure culture of *M. methylutens* treated with 5% ¹³C-labeled substrates (methanol or DIC) (C) Structures of archaeal lipids. Enclosed structures of phytenes in Figure C were tentatively assigned according to GC-MS mass spectra (Figure S6) [80]. Data is expressed as average values (n = 3, error bar=SD).

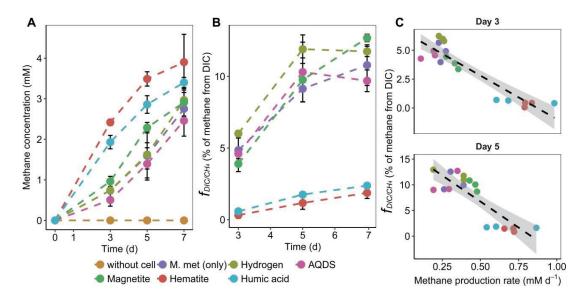


Figure 4. Methane production from DIC during methylotrophic methanogenesis in autoclave slurry supplemented with pure culture of *M. methylutens*. (A) Total methane concentrations in headspace. (B) Proportion of methane derived from DIC. Methane proportion from DIC (f_{DIC/CH_4}) was calculated according to Eq.2. Data is expressed as average values (n = 3, error bar = SD). (C) Linear correlation between methanogenesis rate and methane proportion from DIC after 3 and 5 days. Day 3: Pearson's r = -0.92, P < 0.001, CI (0.95) = -0.79 > r > -0.97; Day 5: Pearson's r = -0.85, P < 0.001, CI (0.95) = -0.62 > r > -0.94.

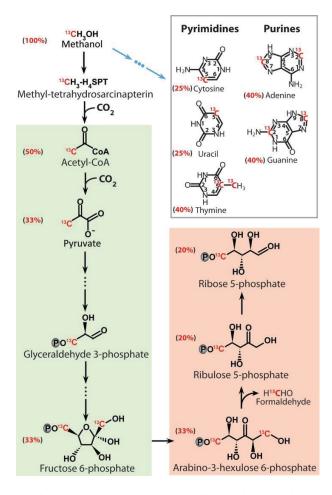


Figure 5. Biosynthesis of nucleotide moieties, the pyrimidine and purine bases, as well as the C5-carbon from ¹³C-labelled methanol in methylotrophic methanogens based on previous studies [54-59] with final carbon contribution from methanol added besides the compounds. Black arrows indicate ribose synthesis and the blue arrows represent synthesis of base moieties in nucleosides. The reverse gluconeogenesis pathway is displayed in green and the reverse ribulose monophosphate pathway in pink.

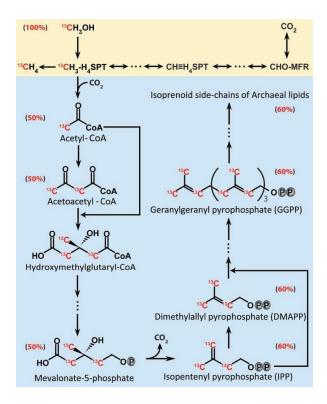


Figure 6. Methylotrophic methanogenesis pathway from methanol (yellow) and carbon assimilation pattern into isoprenoid chains of archaeal lipids (blue) with carbon contribution from ¹³C-methanol added besides the compounds. The pathway of archaeal lipid biosynthesis is based on previous studies [69, 81, 82].

3.10. Supplementary figures

Clone library construction

A clone library of archaeal 16S rRNA gene fragments (~800 bp) was constructed to confirm the accuracy of classification by using short Illumina sequences (143 base pairs). PCR was conducted with primer set of 109F/912R (Table S1) and ALLin RPH polymerase Kit (highQu, Kraichtal, Germany) according to the protocol of the manufacturer. The template cDNA was used from the heavy fractions of RNA-SIP sample of the MZ incubations amended with ¹³C-DIC and unlabeled methanol. Thermocycling was performed as follows: 95 °C for 3 min; 40 cycles at 95 °C for 30 sec, 58 °C for 45 sec and 72 °C for 45 sec; 72 °C for 10 min. Purified PCR products were ligated into the pGEM-T vector (Promega, Mannheim, Germany) and transformed into *Escherichia coli* JM109 competent cells (Promega, Mannheim, Germany) according to the manufacturer. White colonies were randomly picked and cell material directly subjected to colony PCR with the following cycling parameters: 95 °C for 5 min; 28 cycles at 95 °C for 30 sec, 55 °C for 45 sec and 72 °C for 1 min; 72 °C for 5 min. Amplicons of 8 clones were submitted to LGC Genomics (Berlin, Germany) for Sanger sequencing. Sequences have been deposited at GenBank with accession numbers from MK434328 to MK434335.

Table S1. I	Primers	used in	this	study
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Target gene	Primer	Reference
Archaeal 16S rRNA gene	806F (5'-ATTAGATACCCSBGTAGTCC-3')	[1]
Archaeal 16S rRNA gene	912R (5'-GTGCTCCCCGCCAATTCCTTTA-3')	[2]
mcrA	ME2 mod (5'-TCATBGCRTAGTTNGGRTAGT-3')	[3]
mcrA	ME3'Fs (5'-GTCNGGTGGHGTMGGSTTYAC -3')	[4]
Archaeal 16S rRNA gene	Arch519F (5'-CAGCMGCCGCGGTAA-3')	[5]
Archaeal 16S rRNA gene	Arch806R (5'-GGACTACVSGGGTATCTAAT-3')	[6]
Archaeal 16S rRNA gene	109F (5'-ACKGCTCAGTAACACGT-3')	[7]

Table.S2 Carbon recovery of methanol from incubations amended with DIC and ¹³C-methanol

	methane,	¹³ C-TIC,	Ratio of	Carbon
Sediment	μmol	μmol	methane to CO ₂	recovery, %
SRZ	27.0 ± 1.4	10.8 ± 0.7	2.5 ± 0.3	81.3 ± 1.7
MZ	28.6 ± 0.4	8.8 ± 0.3	3.2 ± 0.1	80.4 ± 0.4

¹³C-TIC (total inorganic carbon in bottle) and ¹³C-methane were quantified at harvest. Data is presented as average values (n = 3, error bar = SD). A total amount of 46.5 μ mol ¹³C-methanol was amended into SIP incubation.

Table S3. Comparison of fold increase of archaeal 16S rRNA gene copy number in heavy fractionsbetween "MeOH + 13 C-DIC" and "DIC + 13 C-MeOH" incubations

Sample	Density (g/mL)	Archaeal 16S rRNA copy number	Fold increase compared to ¹³ C- MeOH incubation	
	1.823	9.04E+00	-	
MZ_	1.817	3.05E+01	-	
DIC+ ¹³ C-MeOH	1.812	2.82E+07	-	
	1.806	1.59E+03	-	
	1.82	1.43E+01	1.59E+00	
MZ_	1.815	3.67E+02	1.21E+01	
MeOH+ ¹³ C-DIC	1.809	2.31E+02	8.20E-06	
	1.803	3.29E+08	2.07E+05	
	1.823	1.69E+02	-	
SMZ_	1.817	3.46E+02	-	
DIC+ ¹³ C-MeOH	1.812	1.50E+07	-	
	1.809	1.32E+04	-	
	1.823	3.96E+07	2.35E+05	
SMZ_	1.817	1.70E+08	4.90E+05	
MeOH+ ¹³ C-DIC	1.812	7.69E+03	5.11E-04	
	1.806	3.14E+08	2.37E+04	

Table S4. Relative abundance of archaeal reads above 0.01%.

	SRZ	SRZ	MZ	MZ	SRZ	SRZ	MZ	MZ
taxonomy	DIC+13C-MeOH	DIC+13C-MeOH	DIC+13C-MeOH	DIC+13C-MeOH	MeOH+13C-DIC	MeOH+13C-DIC	MeOH+13C-DIC	MeOH+13C-DIC
-	Heavy fraction	Light fraction						
Methanococcoides	95.2564	79.2561	58.4216	61.3458	96.7867	67.7758	65.5494	56.4008
Methanolobus	1.8769	1.1343	40.8147	36.9315	2.0298	1.6736	32.3346	36.3824
Group C3	0.8410	8.1593	0.2845	0.5795	0.2740	2.5437	0.6668	3.8382
Bathyarchaeota	0.6359	4.5811	0.1385	0.1736	0.2198	6.7875	0.7557	1.2871
ANME-2c	0.4051	0.2629	0.0075	0.0000	0.0181	3.7059	0.0178	0.0460
Lokiarchaeota	0.3385	4.6054	0.0936	0.2706	0.1897	3.0418	0.2400	0.7584
ANME-1b	0.2974	0.4576	0.0899	0.3721	0.0602	5.3995	0.0889	0.5516
ANME-2a/2b	0.1538	0.0633	0.0112	0.0135	0.2018	4.1841	0.0445	0.0230
Marine Benthic Group D and DHVEG-1	0.0154	0.1022	0.0262	0.0992	0.0843	2.0456	0.1245	0.1379
Thermoplasmatales_CCA47	0.0359	0.2629	0.0000	0.0000	0.0120	0.4251	0.0178	0.0000
Marine Hydrothermal Vent Group(MHVG)	0.0359	0.5696	0.0037	0.0113	0.0181	0.6907	0.0000	0.0689
Methanosarcina	0.0308	0.0097	0.0225	0.0316	0.0211	0.0199	0.0267	0.0230
ANME-3	0.0256	0.0195	0.0112	0.0225	0.0181	0.0731	0.0267	0.0230
Candidatus Nitrosopumilus	0.0205	0.0487	0.0112	0.0135	0.0060	0.0797	0.0178	0.0919
Unclassified Archaea	0.0103	0.0584	0.0037	0.0135	0.0030	0.0531	0.0089	0.0460
Thermoplasmatales_ANT06-05	0.0051	0.1753	0.0075	0.0068	0.0120	0.4649	0.0089	0.0000
Methanosaeta	0.0051	0.0146	0.0075	0.0248	0.0000	0.1195	0.0000	0.0919
Woesearchaeota (DHVEG-6)	0.0051	0.0682	0.0150	0.0338	0.0271	0.0930	0.0089	0.0460
Thermoplasmatales_20c-4	0.0000	0.0243	0.0000	0.0023	0.0000	0.1461	0.0000	0.0000
Archaea_AK8	0.0000	0.0049	0.0000	0.0000	0.0000	0.0066	0.0178	0.0000
Altiarchaeales	0.0000	0.0049	0.0037	0.0045	0.0000	0.0066	0.0178	0.0000
Thermoplasmatales_AMOS1A-4113-D04	0.0000	0.0097	0.0037	0.0000	0.0000	0.0332	0.0000	0.0000
Ancient Archaeal Group(AAG)	0.0000	0.0000	0.0000	0.0023	0.0030	0.0797	0.0000	0.0000
Candidatus Aenigmarchaeum	0.0000	0.0000	0.0000	0.0000	0.0000	0.0199	0.0000	0.0230
Deep Sea Euryarchaeotic Group(DSEG)	0.0000	0.0487	0.0000	0.0045	0.0120	0.1328	0.0089	0.0689
Marine Group I	0.0000	0.0000	0.0000	0.0023	0.0000	0.0000	0.0000	0.0230
Marine Group II	0.0000	0.0049	0.0037	0.0090	0.0000	0.0000	0.0089	0.0460
Thermoplasmatales_MKCST-A3	0.0000	0.0049	0.0000	0.0023	0.0000	0.0465	0.0000	0.0000
Soil Crenarchaeotic Group(SCG)	0.0000	0.0097	0.0000	0.0000	0.0030	0.0133	0.0089	0.0230
Terrestrial Miscellaneous Gp(TMEG)	0.0000	0.0292	0.0037	0.0090	0.0000	0.3121	0.0000	0.0000
Thermoplasmatales_VC2.1 Arc6	0.0000	0.0000	0.0000	0.0000	0.0000	0.0133	0.0000	0.0000
Others(<0.01%)	0.0051	0.0097	0.0150	0.0203	0.0000	0.0133	0.0000	0.0000

The gray color denotes archaea taxonomy showed in Figure 2. Pink and green colors indicate the scale of archaeal relative abundance.

Table S5. Methanogenesis from methanol and DIC in pure culture of *M. methylutens* grown in Widdel medium. Data is expressed as average values (n = 3).

Substrates	δ ¹³ C-methane, (‰; VPDB)	δ ¹³ C-DIC_day 0, (‰; VPDB)	δ ¹³ C-DIC_day 11, (‰; VPDB)	Methane from labeled substrate, %
DIC + 5% 13 C-MeOH	4620 ± 160	NA	NA	97.1 ± 3.2
MeOH + 5% 13 C-DIC	63.8 ± 5.5	4170 ± 80	3520 ± 82.0	$2.3 \pm 0.1 \sim 2.6 \pm 0.2$

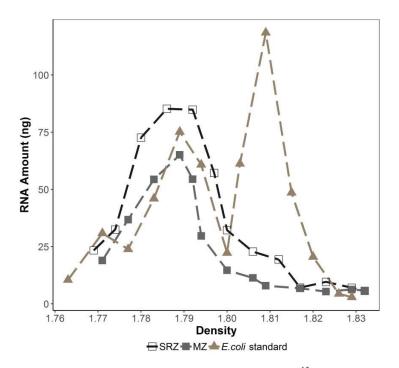


Figure S1. RNA-SIP profiles from slurry incubations amended with ¹³C-DIC only (no methanol added) and *E. coli* standard. Samples were harvested in parallel to methanol amended incubations.

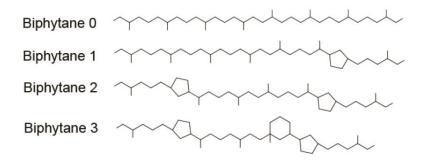


Figure S2. Structures of biphytane moieties released from the intact polar glycerol diphytanoyl glycerol tetraether fraction.

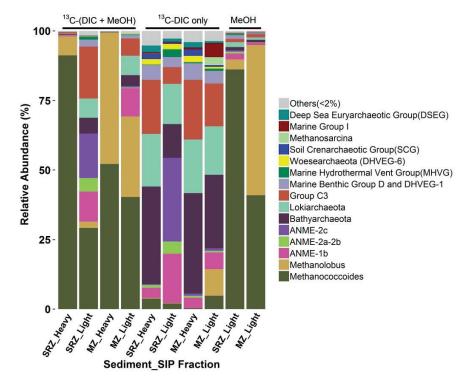


Figure S3. Relative abundance of archaeal 16S rRNA in the RNA-SIP samples from double-labeling incubations (13 C-DIC + 13 C-methanol) and control incubations (13 C-DIC or unlabeled-methanol). No data are shown for the unlabeled-methanol controls from the SRZ and MZ due to low amount of RNA in the heavy fraction.

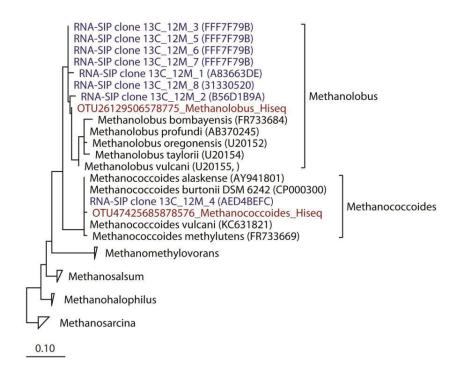


Figure S4. Phylogenetic tree of 16S rRNA genes from clone library (blue) and Illumina sequencing (red). Clone sequences were assembled by using SeqMan software (Version 8.0.2) and aligned online by Silva aligner (<u>https://www.arb-silva.de/aligner/</u>). The aligned sequences were input into ARB (Version 6.0.2). Aligned clone sequences and know sequences of *Methanosarcinaceae* in SILVA SSURef database (Release 132) were selected to build a phylogenetic tree using maximum likelihood algorithm and bootstrapping (n=1000). The two dominant OTUs of *Methanosarcinaceae* from Hiseq Illumina sequencing (red) were aligned and added to the tree using the ARB parismony tool.

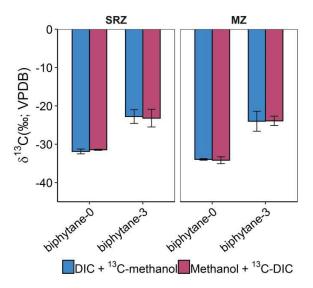


Figure S5. δ^{13} C values of biphytanes released from the intact ploar glycerol diphytanoyl glycerol tetraether fraction. Biphytane-1 and biphytane-2 concentrations were too low to be measured accurately. Determiantion of carbon isotope values was performed after methanogenesis had ceased. Data is expressed as average values (n = 3, error bar = SD).

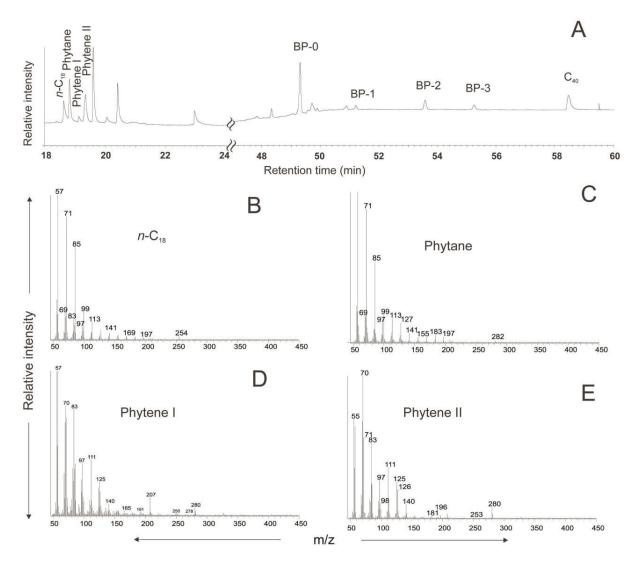


Figure S6. Chromatogram of phytane and phytenes released from intact polar lipid fraction in the incubations of the MZ sample amended with ¹³C-DIC and unlabeld methanol (A) and the mass spectra of compunds interests (B to E).

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Chapter 4

Asgard archaea are active in marine sediment carbon cycling

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Abstract

Asgard is an archaeal superphylum that might hold the key to understand the origin of eukaryotes, but its diversity and ecological role remains poorly understood. Here, we propose five new Asgard phyla, including Kariarchaeota, Balderarchaeota, Hodarchaeota, Lagarchaeota and Gerdarchaeota, based on the 16S rRNA genes phylogenetic analysis. Stable-isotope probing (SIP) revealed that Lokiarchaeota actively fix CO₂ and are associated with organic polymers degradation. Supporting the nucleic acid-SIP results, genomics and transcriptomics further evidence that Asgard archaea are active in the metabolism of fatty acids, alcohol, amino acids, proteins, peptides, benzoate and cellulose. Intriguingly, our analyses indicate that the new phylum Gerdarchaeota might utilize organic substrates by aerobic respiration, the Helarchaeota are capable of short-chain alkane oxidation, and Loki- and Thorarchaeota even have potential for medium-chain alkane degradation. Our findings substantially expand the known global diversity of Asgard archaea, and link their metabolic potentials and activities to their extended role in global carbon cycling.

4.1. Introduction

The domain of Archaea constitutes a considerable fraction of the biomass on Earth¹ and have been estimated to account for ~37% of the total microbial cells in the marine biosphere, especially in coastal sediments (up to 69%)². Meanwhile, archaea have been recognized as key players in the global carbon cycle³, participating in CO₂ fixation and multiple carbon dissimilation pathways such as methanogenesis⁴ and the degradation of proteins, carbohydrates, methylated compounds, fatty acids and lipids^{5, 6}. Nevertheless, the metabolic, physiological, and evolutionary functions of archaea remain largely unknown, especially for many of the newly discovered archaeal groups in recent years, such as DPANN⁷, Bathyarchaeota⁸, Verstraetearchaeota⁹, and Asgard archaea¹⁰.

Asgard archaea, proposed as the latest archaeal superphylum, are composed Lokiarchaeota¹¹, Thorarchaeota¹², Odinarchaeota¹⁰, Heimdallarchaeota¹⁰, and Helarchaeota¹³, mainly originated from lineages formerly named Marine Benthic Group B (MBG-B)¹⁴, Deep-Sea Archaeal Group (DSAG)¹⁵, Ancient Archaeal Group (AAG)¹⁶, and Marine Hydrothermal Vent Group (MHVG)^{16, 17}. Since Asgard archaea contain abundant eukaryotic signature proteins (ESPs) and form a monophyletic group with eukaryotes in the phylogenetic tree, they are regarded as the closest relatives of Eukarya and have attracted increasing research interest^{10, 11, 18}. Asgard archaea inhabit various environments (e.g., marine waters and sediment, lake sediment, mangrove sediment, estuarine sediment and mud volcano)^{10, 19, 20}. The Asgard-associated rRNA recently identified in natural environments²⁰ suggests that they might be more diverse than previously known. Based on metagenome analyses, Lokiarchaeota potentially perform hydrogen oxidation²¹, Thorarchaeota might be mixotrophic (i.e., using both, inorganic and organic carbon for growth)¹⁹ and acetogenic¹², Heimdallarchaeota harbour novel mechanisms of phototrophy^{22, 23}, and Helarchaeota might be capable of anaerobic hydrocarbon oxidation¹³. However, knowledge about their metabolic capabilities, activities and ecological roles^{10, 12} is still in its infancy due to the shortage of available Asgard archaea genomes^{10, 19, 24-26}, the lack of cultured representatives, and the limited molecular and physiological data (e.g., metatranscriptomic or stable isotope probing (SIP) analysis).

In the current study, we first investigated Asgard archaea based on all publicly available 16S rRNA gene sequences and expanded the known diversity and global distribution of Asgard archaea. Second, by using nucleic acid-SIP, we successfully revealed metabolic activities of Asgard archaea. Third, we explored the potential metabolic capabilities of Asgard archaea through metagenomics and metatranscriptomics, especially for the new phyla Gerdarchaeota and Helarchaeota. These findings substantially extend our knowledge of the metabolic capabilities and *in situ* activities of Asgard archaea.

4.2. Results and discussion

4.2.1. Asgard archaea are diverse and ubiquitous

To study the diversity of Asgard archaea, 16S rRNA gene sequences (> 600 bp) from all Asgard metagenome-assembled genomes (MAGs, December 2018 updated) and public databases (SILVA SSU 132 and NCBI databases, December 2018 updated), including a recent study of 16S rRNA gene sequences²⁰, were systematically analysed. After sequence filtering, we obtained >400 Asgard archaeal operational taxonomic units (OTUs, 95% cut-off, 10,448 sequences). The majority of OTUs (79%) and sequences (85%) fell into the Lokiarchaeota lineage (Fig. 1A and Table S2). The phylogenetically broad Lokiarchaeota group with a minimum intragroup similarity of 80% (Table S2) was divided into four subgroups (Lokiarchaeota-1 to Lokiarchaeota-4) based on tree nodes with high bootstrap support (Fig. 1A). Lokiarchaeota-2 and Lokiarchaeota-4 were the most abundant subgroups, accounting for 19% and 59% of the total Asgard 16S rRNA gene sequences, respectively. With a lower minimum intragroup similarity (71%, Table S2) than the recommended phylum-level threshold²⁷, we divided the lineage of Heimdallarchaeota into two subgroups: Heimdallarchaeota-AAG and Heimdallarchaeota-MHVG, following previous studies^{10, 28}. Based on the threshold of 75-83% intragroup similarity of 16S rRNA genes for the phylum level²⁷, we identified five new Asgard phyla and proposed their names as Kariarchaeota, Balderarchaeota, Hodarchaeota, Lagarchaeota and Gerdarchaeota, respectively (Fig. 1A and Table S3). Most sequences (99.7%) of these new phyla were from expressed 16S rRNA genes²⁰, indicating that the Asgard archaea are transcriptionally active and more diverse than previously proposed¹⁰.

A high proportion (~92%) of the OTUs of Asgard 16S rRNA gene sequences originated from sediment samples, especially from offshore and coastal sediments (Fig. 1 and Table S1). Besides, Asgard archaea inhabit other environments such as marine water columns, mud volcanos and soil. The global distribution of Asgard archaea appears to follow a biogeographical organization. For example, unique OTUs from hydrothermal environments placed within the Odinarchaeota and several distinct Lokiarchaeota subgroups, while sequences collected from hypersaline habitats (e.g., hypersaline microbial mat and saltworks belt) clustered within the Lokiarchaeota-4 subgroup (Fig. 1A).

4.2.2. Asgard archaea fix CO₂ and degrade organic polymers based on stable isotope probing

The carbon metabolism in Asgard archaea was investigated by RNA- and DNA-SIP in sediment samples from the Helgoland Mud Area (North Sea) by amending ¹³C-labelled bicarbonate in combination with different electron donors and/or electron acceptors. Due to the general slowness of metabolic processes at 10 °C under energy and carbon pool limitations, we opted for long-term incubations conducted for 255 and 386 days for RNA- and DNA-SIP, respectively.

By applying RNA-SIP, an ultra-high-sensitivity technique with a detection threshold of ≤0.001% of labelled RNA²⁹, we identified the active members of subgroups Lokiarchaeota-2 and Lokiarchaeota-4 in incubations fed with ¹³C-labelled bicarbonate and additional electron donors (Figs. 1A, 2A, and S1C). Specifically, the high relative abundance of the Lokiarchaeota-2 subgroup OTUs in ¹³C-labelled RNA-SIP fractions (>1.80 g/mL) indicated their involvement in the metabolism of elemental sulfur and iron oxide (lepidocrocite) (Figs. 2, S1C, and S1D), although a MAG could not be obtained (Fig. 4). OTUs within subgroup Lokiarchaeota-2 were also associated with the degradation of cellulose (or intermediates) (Figs. 2A and S1C), potentially involving sugar fermentation and acetogenesis as suggested by metagenomic and metatranscriptomic analysis (Fig. 4). OTUs within subgroup Lokiarchaeota-4 were stimulated in the presence of organic polymers (lignin and humic acid, Figs. 2, S1), which was corroborated by a significant increase (p < 0.05) in Lokiarchaeota 16S rRNA gene copy numbers (Fig. S1B). More specifically, Lokiarchaeota-4 OTU1 strongly incorporated ¹³CO₂ in the presence of lignin (25-50% of all archaeal sequences in fractions >1.80 g/ml, Figs. 2A and S1C), whereas Lokiarchaeota-4 OTU2 incorporated less of the label (in fractions >1.79 g/mL, Figs. 2A and S1C) but made up ~50% of all archaeal sequences. The SIP results revealed highly varied activities of Lokiarchaeota in sulfur metabolism, CO₂ assimilation and organic polymer degradation. The ability to utilize inorganic and recalcitrant carbon, such as lignin and humic acids, reflects a capability for mixotrophic growth in Lokiarchaeota subgroups.

4.2.3. Potential metabolic capabilities of Asgard archaea revealed by metagenomics and metatranscriptomics

To gain in-depth insight into the detailed metabolic capabilities and biological functions of Asgard archaea, especially for the new phyla, we collected isotopically heavy DNA fractions from SIP-incubations and coastal sediments for deep sequencing analysis (totally 2.2 Tbp, Table S4). We recovered

14 high-quality Asgard MAGs with completeness >75% (Table S5). As demonstrated by multi-locus phylogenetic analysis (Fig. 3), 16S rRNA phylogeny (Fig. 1A) and a pan-genome based gene clustering (Fig. S2), we classified six Asgard MAGs as the new Asgard phylum Gerdarchaeota (42-50% AAI to other Asgard MAGs, Fig. S3), one as Helarchaeota (43-47% AAI to other Asgard MAGs, Fig. S3), three as Lokiarchaeota-4, three as Thorarchaeota, and two as Heimdallarchaeota-MHVG, respectively.

All available Asgard MAGs (Table S5, except Odinarchaeota, December 2018 updated) were annotated against databases for function prediction, and the key metabolic pathways were reconstructed to investigate their metabolic capability in combination with the metatranscriptomic data (Fig. 4 and Table S6). Corresponding to the nucleic acid-SIP revealed the active CO_2 fixation in Lokiarchaeota, genes for tetrahydromethanopterin dependent Wood-Ljungdahl (THMPT-WL) and tetrahydrofolate dependent WL (THF-WL) pathways were identified in Lokiarchaeotal MAGs (Fig. 4). Similar with Lokiarchaeota, Thorarchaeotal MAGs also contain genes for both THMPT-WL and THF-WL pathways, however, transcripts for the THMPT-dependent WL pathway in Lokiarchaeota and Thorarchaeota were predominant, especially those for incorporating CO₂ into formyl-methanofuran (Fig. 4). Furthermore, in Lokiarchaeota and other Asgard phyla, the abundant transcripts for the generation of pyruvate from acetyl-CoA and CO₂ elevated the contribution of inorganic carbon to biomass³⁰. Additionally, genomic and transcriptomic evidence suggest that Lokiarchaeota and most Asgard phyla have carbon fixation potential via the incomplete rTCA cycle, and the nucleotide salvage pathway using the type III or IV ribulose 1,5-bisphosphate carboxylase (RuBisCO) (Fig. S4)^{19, 26, 31, 32}. The ability to perform inorganic carbon assimilation suggests that Asgard archaea are adaptive to various environments, especially in deep marine sediments where the energy supply is extremely low³³ but inorganic carbon is sufficient³⁴. We also identified genes involved in degradation of multi-sugar, cellulose, polymeric sugar, and the complete gene sets for benzoate degradation (Fig. 4 and Table S6), which strengthened the observation of active Lokiarchaeota in SIP incubations amended with various organic substrates (Fig. 2A). The presence and the high expression levels of genes involved in acetate, ethanol, 1-alcohol, amino acids, proteins and/or peptides degradation suggest that these compounds are also carbon or energy sources for Lokiarchaeota and other Asgard phyla.

For the new phylum Gerdarchaeota proposed in this study, the near-complete gene sets for only THMPT-WL pathway were identified with detectable transcripts in coastal sediments (Fig. 4), which reveals that members of this new Asgard archaeal phylum may use tetrahydromethanopterin as the single C_1 carrier. The results indicate that Gerdarchaeota are quite different to their neighbourhood Heimdallarchaeota-AAG and Heimdallarchaeota_MHVG (Figs. 3 and S5), which only encode the THF-WL pathway. In addition, Gerdarchaeota appear to use organic compounds as electron donors (Fig. 4 and Table S6), and the dominant carbohydrate-active enzymes and peptidases were glycosyltransferases and serine peptidases (Fig. S6 and Table S7). Intriguingly, Gerdarchaeota harbour genes encoding cytochrome c oxidase, nitrate reductase and respiratory chain complex I (Fig. 3), suggesting their potentials for organic substrates degradation by aerobic or anaerobic respiration, as reported for Heimdallarchaeota-AAG and Heimdallarchaeota-MHVG³². Meanwhile, differ to other Asgard phyla, Gerdarchaeota do not harbour the RuBisCO genes for nucleotide salvage (Fig. 3). Additionally, like Lokiarchaeota, Gerdarchaeota lack Glycerol-1-phosphate dehydrogenase (G1PDH) for the archaeal-type lipid biosynthesis, but contain the bacterial-type Glycerol-3-phosphate dehydrogenase (G3PDH). However, the Asgard archaeal G3PDH is encoded by the *glp* gene, thus it may participate in organic carbon degradation rather than archaeal lipid synthesis^{35, 36}.

4.2.4. Alkane metabolism

Methane formation is one of the earliest metabolisms on $earth^{37}$. This process was restricted to Euryarchaeota until the discovery of novel *mcr* genes across the domain of archaea, e.g., Bathy-, Verstraete-, Hades-, Nezha-, and Korarchaeaota^{8, 9, 38-40}. Notably, genes encoding *mcrABG* were identified in Helarchaeota genome SZ_4_bin10.384, as firstly reported by Seitz et. al.¹³, indicating that the MCR complex is widespread in the domain of Archaea. Phylogenetic analysis showed that Asgard archaea *mcrA* genes clustered together within the branch of the butane-oxidizing Syntrophoarchaeum⁴¹ and ethane-oxidizing Argoarchaeum⁴² (Fig. S7A), corroborated by the McrB and McrG protein trees (Fig. S7).

Molecular modelling and dynamics studies (see Supplementary Materials and Methods) showed that the MCR complex of Asgard archaea for butane binding was similar to that of *Ca.* Syntrophoarchaeum according to the root mean square deviation (RMSD) values (~5 Å, Fig. S8B). Thus, Asgard archaea might be able to oxidize short-chain alkanes, in which ethane and butane are preferentially used as revealed by the RMSD values (Fig. S8C). Based on the evolutionary analysis, a high percentage (> 10%) of *mcrA* genes in the Helarchaeota MAGs was most likely transferred horizontally from archaea, and most of these genes originated from methanogenic hosts/donors, e.g., Thermococci, Methanomicrobia, and Methanobacteria (Fig. S9). Although *mcrA* gene transcript has not been detected from the MAG SZ_4_bin10.384 (Fig. 3), the expressed Asgard-like *mcrA* genes in the unbinned scaffolds (e.g., SZ_4_scaffold_203331_2, Fig. S10) highlights the involvement in activities of alkane oxidation by Asgard archaea in marine sediments.

For Asgard archaeal C1 alkane degradation, the methyl-CoM formed by MCR might be oxidized to CO_2 via the reverse THMPT-dependent WL pathway⁴³, while for C2 to C4 alkane degradation, the mechanism of the conversion of $C_nH_{2n+1}COM$ to $C_nH_{2n-1}COA$ is enigmatic. Like *Ca.* Syntrophoarchaeum⁴¹, Asgard archaea might use the potential alkyl and methyl transferase, such as MtaA, MtbA or MttB, (Table S6) for this reaction. Additionally, other potential transferases that not identified in *Ca.* Syntrophoarchaeum, such as tetrahydromethanopterin S-methyltransferase (MtrA and MtrH)⁴¹, were also discovered in all Asgard phyla. In the subsequent step, $C_nH_{2n-1}COA$ is further oxidized to acetyl-CoA through the highly expressed β -oxidation pathway (Fig. 4 and Table S6). Intriguingly, as in ANME⁴⁴, Helarchaeota have the genes for assimilatory sulfate reduction, but involvement in anaerobic alkane oxidation linked to sulfate reduction has not been demonstrated yet³⁸. In contrast to methanogens⁴⁵ and methanotrophs⁴³, β -oxidation enzymes were also identified in Asgard archaea, Bathyarchaeota⁸ and *Ca.* Syntrophoarchaeum⁴¹, suggesting metabolic similarity among the three lineages. Although the MCR complex was not identified in all Asgard MAGs, the presence of β -oxidation enzymes, as well as the Asgard-like *mcrA* genes in the unbinned scaffolds (Fig. S10) revealed their metabolic potential for short-chain hydrocarbon oxidation.

Genes encoding aerobic alkane utilization (*alkB*) were found in Lokiarchaeota and Thorarchaeota MAGs (Fig. 4 and Table S6) retrieved from Shark Bay microbial mat²⁴. Hence, Asgard archaea might also participate in medium-chain alkane (C5–C16) oxidation, although the spectrum of alkanes utilized remains unclear. The construction of protein tree indicates that Asgard *alkB* genes are monophyletic with those from the sulfate-reducing bacterium *Desulfatibacillum aliphaticivorans*⁴⁶ (46% amino acid identity, Fig. S11). Similar to other sulfate-reducing bacteria^{46, 47}, Asgard archaea might oxidize medium-chain alkanes via a primary alcohol, which may be further transferred to acetyl-CoA through the incomplete β -oxidation process (Fig. 4). Meanwhile, we identified some auxiliary receiving and transcriptional gene sets, e.g., methyl-accepting chemotaxis proteins (MCPs) and the chemotaxis protein set CheARWY, which might be involved in alkane degradation⁴⁸.

Except for the core genes for alkane degradation, genes encoding enzymes for hydrogen metabolism and electron transport, including the [NiFe]-hydrogenase *mvhADG* and the hydrogenase heterodisulfide reductase $hdrABC^{49}$ were detected in all phyla and were highly expressed in Lokiarchaeota and Thorarchaeota (Fig. 4 and Table S6). Additionally, membrane-bound $F_{420}H_2$ dehydrogenase (Fpo) was identified in all Asgard phyla. The proton gradient generated by Fpo can be used for ATP synthesis through the highly expressed genes encoding V/A-type ATPase (Fig. 4). In contrast to most members of Asgard archaea, Heimdallarchaeota contain cytochrome c oxidase genes, suggesting the potential for electron transfer during respiration²³.

In conclusion, we expand the Asgard archaea diversity by reporting five new Asgard phyla. The wide distribution and large diversity of Asgard archaea suggest that they are globally important components of the microbial community. The experimentally verified ability to perform carbon fixation, organic polymer degradation and transcriptomic activity with a wide spectrum of organic carbon compounds, especially alkanes, indicates that Asgard archaea are important players in the marine sediment carbon cycle. However, their contribution to carbon metabolism remains to be quantified, especially for the newly discovered clades. Overall, our results prove that Asgard archaea are active participants of carbon cycling in the marine environment.

4.3. Methods

4.3.1. Asgard archaea 16S rRNA gene dataset construction

Archaeal 16S rRNA gene sequences were retrieved from the GenBank NCBI nucleotide database (September 2017) and SILVA SSU r132 database⁵¹. E-utilities⁵² was applied to search and retrieve the archaeal 16S rRNA gene sequences from the NCBI nucleotide database using the ESearch function with the following string: "16S AND 800:2000[Sequence Length] AND archaea[organism] AND rrna[Feature Key] AND isolation source[All fields] NOT genome NOT chromosome NOT plasmid". EFetch function was then used to retrieve the sequences and the corresponding GenBank-formatted flat file, which contained environmental information (e.g., location and isolation source). In the subsequent steps, custom scripts were designed to combine the two datasets and to remove low-quality (i.e., containing 'N' or shorter than 800 bp) and duplicate sequences, resulting in 100,786 archaeal sequences. To obtain potential Asgard 16S rRNA gene sequences, the above sequences were BLAST-searched against genome-based 16S rRNA gene sequences (\geq 800 bp, Table S2) using BLASTn with a cutoff E-value \leq 1e-5, sequence identity \geq 75%, and coverage \geq 50%. This resulted in a set of 9765 potential Asgard sequences from the public databases. OTUs were assigned using the QIIME UCLUST⁵³ wrapper, with a threshold of 95% nucleotide sequence identity, and the cluster centroid for each OTU was chosen as the representative OTU sequence. A set of OTU threshold (e.g., 90%, 95% and 97%) was verified to obtain the optimum value for phylogenetic tree building and data analysis.

In addition to public databases, archaeal 16S rRNA gene sequences were also retrieved from a recent study²⁰. All expressed 16S rRNA gene sequences in the reference paper were BLASTn-searched against a custom database containing 16S rRNA gene sequences retrieved from Asgard MAGs and potential Asgard OTUs obtained as described above, with a cutoff E-value $\leq 1e-5$, sequence identity $\geq 50\%$, and coverage $\geq 80\%$. Finally, 5588 potential Asgard sequences were obtained, including eight newly proposed

clusters DAS1–8 (79 sequences)²⁰. OTU re-formation of these sequences (15,353 sequences, from both databases and the reference paper) was performed using the QIIME UCLUST⁵³ wrapper, with a threshold of 95% pair-wise nucleotide sequence identity, resulting in 1836 OTUs.

4.3.2. Phylogenetic position and distribution of Asgard archaea 16S rRNA gene sequences

SINA-aligned⁵⁴ archaeal representative OTU sequences obtained in the previous step (1836 OTUs) were pre-filtered using a backbone tree, which was constructed based on updated 16S rRNA archaeal gene datasets^{55, 56}, using the ARB software (version 5.5) with the "Parsimony (Quick Add Marked)" tool⁵⁷. This resulted in 456 Asgard OTUs. Candidate representative OTU sequences (456 OTUs), genome-based 16S rRNA gene sequences \geq 600 bp (17 sequences), and reference sequences were used for phylogenetic tree construction. Maximum-likelihood tree was inferred with IQ-TREE (version 1.6.1)⁵⁸ using the GTR+I+G4 mixture model (recommended by the "TESTONLY" model) and ultrafast (-bb 1000). Asgard clade designations were made when either a group was defined in previous publications, or when groups with >20 sequences and intragroup similarity >75%⁵⁹ were monophyletic. Attributes [i.e., expressed rRNA gene, biotope, temperature, and salinity (Table S1)] for each representative OTU were extracted and visualized using iTOL software⁶⁰. Calculation of percent identity of new Asgard clades was based on 179 sequences with a long fragment of Asgardarchaeotal 16S rRNA gene. Fragments of 16S rRNA gene from position of *E. coli* 243 to 1414 (~1170 bp) were used for calculating the identity.

The corresponding environmental information (i.e., location and biotopes) for the 16S rRNA gene sequences of Asgard archaea was extracted from the GenBank-formatted flat file using custom scripts. This resulted in 172 libraries with latitude and longitude of sampling sites (Table S1). Information on the sample locations was plotted using the mapdata and ggplot2 packages in R software.

4.3.3. SIP incubations

Sediment for incubations was collected from Helgoland Mud Area methane zone (54°05.23'N, 007°58.04'E) by gravity coring during the RV HEINCKE cruise HE443 in 2015. Sediment from the methanic zone (239-263 cm depth) and artificial sea water (w:v=1:4, 50 mL) were homogenized and incubated anaerobically in sterile 120-mL serum bottles sealed with butyl rubber stoppers, and headspace refilled with N₂. A 10-day pre-incubation was performed by exchanging headspace with N₂ to remove CO_2 . Triplicate set ups were supplemented with electron donors (1 g L⁻¹ sulfur, 30 mg L⁻¹ lignin, 30 mg L⁻¹ humic acid, 30 mg L⁻¹ cellulose, respectively), electron acceptors (30 mM lepidocrocite) and 10 mM

sodium bicarbonate (¹³C-labelled bicarbonate provided by Cambridge Isotope Laboratories, Tewksbusry, Massachusetts, USA) and were incubated at 10 °C. An additional set up fed with ¹²C-bicarbonate was used as a control for comparison. Samples collected at day 255 were used for RNA-SIP analysis and those at day 386 were used for DNA-SIP analysis. DNA and RNA extraction were performed as previously described⁵⁰. Cross-feeding by indirectly utilization of ¹³C-DIC in these incubations can be neglected for the following reasons: i) organic substrates were unlabelled; ii) no obvious buoyant density shift in nucleic acid-SIP profiles of organic substrates incubations; iii) no considerable increase of bacterial gene copies from incubations amended with organic polymers (Fig. S1d); iv) no significant bacterial community shift between heavy and light fractions in incubations of "sulfur" and "sulfur + lepidocrocite" (Fig. S1c).

4.3.4. Isopycnic centrifugation, gradient fractionation and sequencing

Isopycnic centrifugation and gradient fractionation were performed to separate ¹³C- labelled from unlabelled nucleic acids as previously described⁵⁰. 0.3- 0.7 µg RNA and 4-6.5 µg DNA were used for DNA- and RNA-SIP, respectively. 13 fractions were obtained from each sample after ultracentrifugation. RNA was reverse transcribed using the high capacity cDNA reverse transcription kit (Applied Biosystems, Foster City, California, USA). cDNA from fractions 4 and 5 (ultra-heavy), 6 and 7 (heavy), 8 and 9 (light), as well as 10 and 11 (ultra-light) were combined for sequencing. DNA samples from fractions without combination were used for high-throughput sequencing. PCR targeting the V4 region of 16S rRNA gene sequences was performed with KAPA HiFi HotStart PCR kit (KAPA Biosystems, Cape Town, South Africa) and barcoded archaeal primer Arc519F (5'-CAGCMGCCGCGGTAA-3') and Arch806R (5'-GGACTACVSGGGTATCTAAT-3'). Thermocycling was performed as follows: 95 °C for 3 min; 35 cycles at 98 °C for 20 sec, 61 °C for 15 sec, and 72 °C for 15 sec; 72 °C for 1 min. PCR products were purified using the Monarch PCR Cleanup Kit (New England Biolabs, Ipswich, Massachusetts, USA) according to the manufacturer. Equimolar amounts of amplicons per samples were combined based on PicoGreen quantification. Amplicon sequencing was conducted using Illumina Hiseq 4000 platform with 150-bp paired-end reads at GATC Biotech (Konstanz, Germany). Raw reads were processed using the QIIME 1.9.0 software package⁶¹. OTUs were clustered at 97% identity using UPARSE-OTU⁶². Potential Asgard OTUs were taxonomically assigned by BLASTn search against OTUs in Fig. 1a with the following parameters "-perc identity 97, -qcov hsp perc 100, -evalue 1e-5".

4.3.5. Clone library construction

A clone library of archaeal 16S rRNA gene fragments (>800 bp) was constructed to confirm the accuracy of classification by using short Illumina sequences (143 bp). The DNA-SIP samples from heavy fractions of the incubations amended with sulfur and lepidocrocite (density = 1.715 g/ml), lignin and lepidocrocite (1.714 g/ml) as well as humic acid and lepidocrocite (1.714 g/ml) were used for cloning. PCR was performed with the primer set of Arc8F (5'-TCCGGTTGATCCTGCC-3')/Arc912R (5'-GTGCTCCCCGGCCAATTCCTTTA-3') using ALLin RPH polymerase Kit (highQu, Kraichtal, Germany). Thermocycling was performed as follows: 95 °C for 3 min; 40 cycles at 95 °C for 30 sec, 55 °C for 45 sec and 72 °C for 45 sec; 72 °C for 10 min. Purified PCR products were ligated into the pGEM-T vector (Promega, Mannheim, Germany) and transformed into Escherichia coli JM109 competent cells (Promega, Mannheim, Germany) according to the manufacturer. White colonies were randomly picked and cell material was directly subjected to colony PCR with the following cycling parameters: 95 °C for 5 min; 28 cycles at 95 °C for 30 sec, 55 °C for 45 sec, and 72 °C for 1 min; 72 °C for 5 min. Amplicons of 96 clones were submitted to LGC Genomics (Berlin, Germany) for Sanger sequencing. Similarity between OTUs of 16S rRNA gene sequences (Hiseq) and clones ranged from 97.9% to 100% (Fig. 1a), suggesting the accuracy of identification of short Illumina sequences.

4.3.6. Quantitative PCR (qPCR)

16S rRNA genes of Lokiarchaeota in cultures were quantified using the newly designed primer set Loki97F/Loki495R (5'-TTCCCATRGCAAACTGCTCA-3'/5'-CCTTGCCCTCTCCTTTCT-3'), which were designed in ARB using the PROBE DESIGN feature and showed specificity to Lokiarchaeota. qPCR was performed as described previously with modifications³⁰. Lokiarchaeota-4 16S rRNA gene clone was used as qPCR standard. Each reaction mixture contained Takyon Master Mix (Eurogentec, Seraing, Belgium), bovine serum albumin (Roche, Mannheim, Germany), 500 nM primers, 1 ng DNA templates or 2 μ L of standard. The qPCR protocol comprised an initial denaturation (5 min at 95 °C) and 40-cycle amplification (95 °C for 15 sec, 58 °C for 20 sec and 72 °C for 40 sec). Primer specificity for qPCR was tested by comparison with the known clones including Lokiarchaeota-4 (positive control), Lokiarchaeota-2 (positive control), and clones as negative control selected from Heimdallarchaeota, Odinarchaeota, Bathyarchaeota, Woesearchaeota, and ANME-3. The cycle threshold (Ct) values of Lokiarchaeota-2 was close to Ct values of the Lokiarchaeota-4 clone, while the Ct values of the negative control clones were higher than that of the standard with lowest quantity of DNA, indicating primers were specific for the Lokiarchaeota detected in our samples. Statistically significant differences of gene

abundances between treatments were assessed with the general linear hypothesis test and pairwise Tukey procedures⁶³ at a threshold of p <0.05. The tests were adjusted for heteroscedasticity, non-Gaussian gene abundance distribution, and for multiple comparisons to control the false discovery rate. Analyses were carried out in R $3.5.2^{64}$.

4.3.7. Sediment sample collection and processing

Samples for metagenome analysis were collected from coast sediment of China (Table S4). They were sampled using custom hollow column, sealed in plastic bags in duplicates, stored in sampling box with ice bags, and transported to the lab within 4 hours. The physiochemical parameters of the samples were determined as previously described⁶⁵. Samples for RNA extraction were preserved in RNALater (Ambion, Life Technologies). For each sample, 10 g sediment each was used for DNA and RNA isolation with the PowerSoil DNA Isolation Kit (MO BIO) and RNA PowersoilTM Total RNA Isolation Kit (QIAGEN), respectively. The rRNA genes were removed from the total RNA using the Ribo-Zero rRNA removal kit (Illumina, Inc., San Diego, CA, USA) and the remaining mRNA was reverse-transcribed. DNA and cDNA were sequenced using an Illumina HiSeq sequencer (Illumina) with 150-bp paired-end reads at BerryGenomics (Beijing, China). Metatranscriptomic reads were quality-trimmed using Sickle (version 1.33)⁶⁶ with the quality score ≥ 25 , and the potential rRNA reads were removed using SortMeRNA (version 2.0)⁶⁷ against both the SILVA 132 database and the default databases (E-value cutoff $\leq 1e-5$).

4.3.8. Metagenomic assembly, genome binning and gene annotation

Raw metagenomic DNA reads of the mangrove sediments and sediment enrichment were dereplicated (identical reads) and trimmed using sickle (version 1.33)⁶⁶ with the option "-q 25". Paired-end Illumina reads for each sample were *de novo* assembled using IDBA-UD (version 1.1.1)⁶⁸ with the parameters "- mink 65, -maxk 145, -steps 10". Scaffolds were binned into genomic bins using a combination of MetaBAT⁶⁹ and Das Tool⁷⁰. Briefly, twelve sets of parameters were set for MetaBAT binning, and Das Tool was further applied to obtain an optimized, non-redundant set of bins. To improve the quality of the bins (e.g., scaffold length and bin completeness), each Asgard-related bin was remapped with the short-read mapper BWA⁷⁵ and re-assembled using SPAdes (version 3.0.0)⁷¹ or IDBA-UD (version 1.1.1)⁶⁸, followed by MetaBAT and Das Tool binning. Asgard MAGs with high contamination were further refined with Anvi'o software (version 2.2.2)⁷². The completeness, contamination and strain heterogeneity of the genomic bins were estimated by CheckM (version 1.0.7) software⁷³. Anvi'o software (version 2.2.2)⁷² was applied for pan-genome analysis of Asgard archaea with the option "--min-occurrence 3".

Protein-coding regions were predicted using Prodigal (version 2.6.3) with the "-p meta" option⁷⁴. The KEGG server (BlastKOALA)⁷⁵, eggNOG-mapper⁷⁶, InterProScan tool (V60)⁷⁷, and BLASTp vs. NCBI-nr database searched on December 2017 (E-value cutoff $\leq 1e-5$) were used to annotate the protein-coding regions. ESP prediction was based on arCOG (eggNOG-mapper) and the eukaryote-specific IPR domains (InterProScan)¹⁰. Archaeal extracellular peptidases were identified using PRED-SIGNAL⁷⁸ and PSORTb⁷⁹ (Table S8).

4.3.9. Phylogenetic analyses of Asgard archaea and functional genes

Small subunit 16S rRNA gene sequences and a concatenated set of 122 archaeal-specific conserved marker genes⁹ were used for phylogenetic analyses of Asgard archaea. Ribosomal RNA genes in the Asgard bins were extracted by Barrnap (version 0.3. http://www.vicbioinformatics.com/software.barrnap.html). An updated 16S rRNA gene sequence dataset from reference papers^{55, 56} with genome-based 16S rRNA genes were aligned using SINA (version 1.2.11)⁵⁴. The 16S rRNA gene sequences and protein maximum-likelihood tree was built with IQ-TREE (version 1.6.1)⁵⁸ using the GTR+I+G4 and PMB+F+I+G mixture model (recommended by the "TESTONLY" model), respectively, with opition "-bb 1000". Marker genes for protein tree were identified using hidden Markov models (HMMs) and were aligned separately using hmmalign from HMMER3⁸⁰ with default parameters. The 122 archaeal marker genes were identified using hidden Markov models. Each protein was individually aligned using hmmalign⁸¹. The concatenated alignment was trimmed by BMGE with flags "-t AA -m BLOSUM30"82. Then, maximum-likelihood trees were built using IQ-TREE with the best-fit model of "LG+I+G4" followed by extended model selection with FreeRate heterogeneity and 1000 times ultrafast bootstrapping. The final tree was rooted with the DPANN superphylum and Euryarchaeota.

Key genes for enzymes involved in alkane degradation, i.e. *mcrABG* and *alkB*, were extracted from the Asgard MAGs. Amino acid sequences were aligned using MUSCLE (version 3.8.1551)⁸³ with default parameters. Un-rooted phylogenetic trees for each gene were built with corresponding gene sequences from NCBI database and reference paper⁸⁴ using IQ-TREE with model LG+F+I+G4 (*mcrABG*) or LG+F+G4 (*alkB*).

4.3.10. Metabolic pathway construction

Potential metabolic pathways were reconstructed based on the predicted annotations and the reference pathways depicted in KEGG and MetaCyc⁸⁵. Metatranscriptome data from mangrove and mudflat

sediments of Shenzhen Bay (Table S4) were analysed to clarify the transcriptomic activity of Asgard archaea. The abundance of transcripts for each gene was determined by mapping all non-rRNA transcripts to predicted genes using BWA with default setting^{6, 86}. Normalized expression was expressed in transcript per million units (TPM), followed by normalization by genome number of each phylum.

4.3.11. Molecular modelling and dynamics simulation

Helarchaeotal MCR amino acid sequences were blasted against the protein data bank (PDB)⁸⁷ to obtain high-similarity sequences. The geometry of these sequences were then predicted using MODELLER⁸⁸, and those with high DOPE score were kept for analysis. Protein–protein interaction of the MCR complex was predicted using ZDOCK (http://zdock.umassmed.edu/), and the protein–ligand docking was executed by the AutoDock4 (version 4.2.6)⁸⁹ with the Lamarckian genetic algorithm⁹⁰.

Molecular dynamics (MD) simulations of MCR complexes were performed using GROMACS (version 5.1.1)⁹¹ based on AMBER99SB-ILDN force field⁹² and TIP3P water box⁹³. The cutoff of non-bonded interactions involving van der Waals and electrostatics was set 10 Å. Long range electrostatic interactions were treated using the Partice-Mesh-Ewald (PME) algorithm⁹⁴. Energy minimization was carried out to obtain initial structure with force $\leq 1000 \text{ kJ} \cdot \text{mol}^{-1} \cdot \text{nm}^{-1}$. MD simulations were initially carried out using the NVT ensemble for 100 ps (300 K), followed by simulation with NPT ensemble for 100 ps (300 K and 1 bar). Then, 50 ns MD simulations were performed using NPT ensemble (300 K and 1 bar) with the time steps of 2 fs to get the equilibrium trajectories. Snapshot was saved at an interval of 10 ps for subsequent analysis. Snapshots were saved every 10 ps for subsequent analysis.

4.3.12. Evolutionary analysis

All genomes containing MCR complex were downloaded from NCBI (<u>https://www.ncbi.nlm.nih.gov/</u>) databases. CheckM was used to check the genome quality. Genomes with completeness < 70% were removed for the further analysis. We selected 117 genomes for final analysis. The phylogenetic tree of concatenated 122 archaeal marker genes were constructed using IQ-TREE (version 1.6.3) with "-m MFP - mset LG,WAG -mrate E,I,G,I+G -mfreq FU -bb 1000" flags, the genome of a DPANN archaea (Diapherotrites_AR10) was used as outgroup. Putative HGTs were inferred using HGTector⁹⁵. A standard database (version updated on 2017-6-30) were used for homologues searching by using MMseqs2⁹⁶. Quality cutoffs for valid hits were e-value $\leq 1e^{-20}$, sequence identity $\geq 30\%$, and coverage of query sequence $\geq 50\%$.

4.3.13. Data availability

Archaeal 16S rRNA gene sequences were retrieved from NCBI database, SILVA SSU r132 database, and a reference paper as described in Supplementary Methods. Public Asgard MAGs were from NCBI database and MG-RAST. The newly obtained Asgard MAGs and metatranscriptomic data are available in NCBI database under the project PRJNA495098 and PRJNA360036. Sequencing data of SIP samples have been submitted to GenBank Short Reads Archive with accession numbers from SRR8607872 to SRR8607991. Clone sequences have been deposited at GenBank with accession numbers of MK551261-MK551285.

4.4. Acknowledgements

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4.5. Author contributions

M.L. and M.C. conceived this study. M.C. analysed the 16S rRNA data, metagenomic data and metatranscriptomic data. X.Y., A.K., T.R.H and M.W.F., performed the SIP trials and analysed the 16S rRNA gene data. Y.L. collected samples and analysed the horizontal gene transfer. Z.Z. provided support for diversity analysis. M.C., Y.C.Y., J.P. and Z.Z. prepared the DNA and cDNA for sequencing. W.L. and X.W. analysed MCR complex protein structure and simulated the binding substrates. M.L., M.C., Y.L., and X.Y. wrote, and all authors edited and approved the manuscript.

4.6. Author Information

Reprints and permissions information is available at www.nature.com/reprints. The authors declare no competing financial interests. Readers are welcome to comment on the online version of the paper. Correspondence and requests for materials should be addressed to M.L. (limeng848@szu.edu.cn) or M.W.F. (michael.friedrich@uni-bremen.de).

4.7. Figures

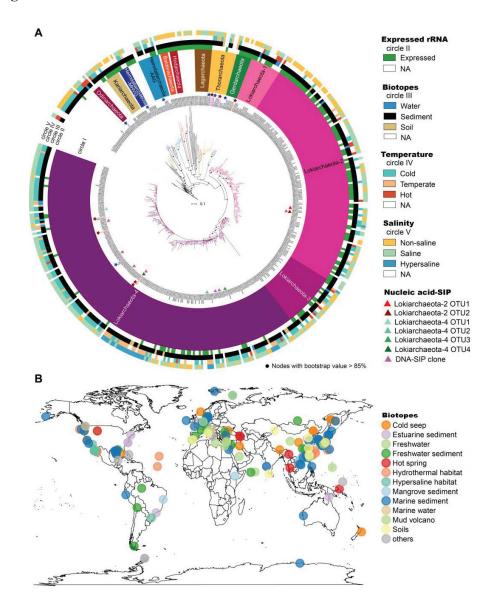


Fig. 1. Diversity and distribution of Asgard archaea. (**A**) Maximum likelihood tree of Asgard archaea constructed using genome-based and publicly available 16S rRNA gene sequences clustered at 95% sequence identity. Groups were designated based on sequences fulfilling the division criteria (see Supplementary Materials and Methods). The colour circles, from the inside to the outside, represent the expressed 16S rRNA gene sequences, habitat, temperature, and salinity, accordingly. Red stars represent 16S rRNA gene sequences from newly discovered Asgard MAGs, and blue stars represent sequences from reference Asgard MAGs. MAGs with the MCR complex and MAGs from ¹³C-bicarbonate labelled enrichment are highlighted with two red stars. The 16S rRNA gene sequences from DNA-SIP and RNA-

SIP trials are marked with triangles (refer to Fig. 2 for details). The tree was re-rooted with Crenarchaeota. Bootstrap support values >70% are shown. (**B**) Global distribution and biotopes of Asgard archaea from 172 public archaeal libraries. Detailed information can be found in Table S1 and Data S1.

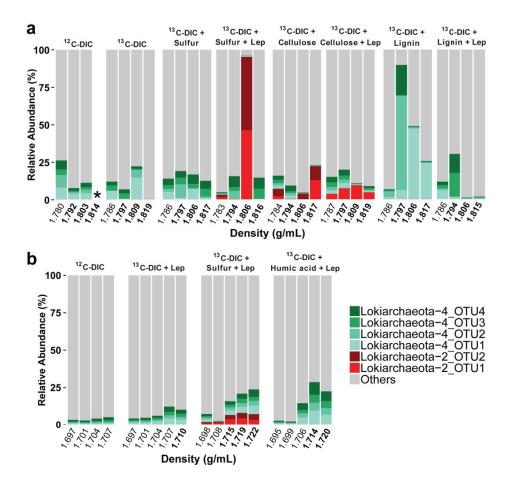


Fig. 2. Total sum scaling of Lokiarchaeota abundances of archaeal 16S rRNA gene sequences from selected "light" and "heavy" gradient fractions of (a) RNA- and (b) DNA-SIP samples. Considering the limited density shifts in incubations with organic substrates and similar bacterial communities in the RNA fractions of "sulfur" and "sulfur + lepidocrocite" incubations (see Supplementary Methods and Supplementary Fig. 1), cross-feeding was most likely negligible⁴⁷. For RNA-SIP, pairs of fractions (fractions 4 and 5, 6 and 7, 8 and 9, 10 and 11) were combined for Illumina sequencing, whereas individual samples were used for DNA-SIP. Density was indicated as the average density of combined fractions for RNA-SIP samples. The centrifugation gradients densities of RNA >1.79 g/mL and DNA >1.71 g/mL were regarded as containing ¹³C-labelled nucleic acids (bold)(Lueders et al., 2004a). *

indicates that cDNA synthesis failed because of the low amount of RNA in these fractions. DNA with densities >1.71 g/mL was not obtained from ¹²C-control incubations. Lep: lepidocrocite. DIC: dissolved inorganic carbon, i.e., bicarbonate.

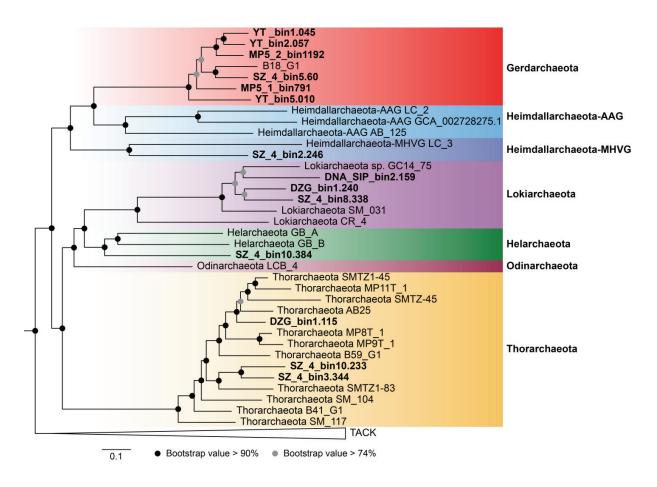


Fig. 3. Maximum likelihood tree of Asgard MAGs inferred from a concatenated alignment of 122 archaeal marker genes and re-rooted with DPANN and Euryarchaeota. The collapsed nodes are represented by triangles and sized in proportion to genome numbers. Asgard MAGs obtained in this study are marked bold.

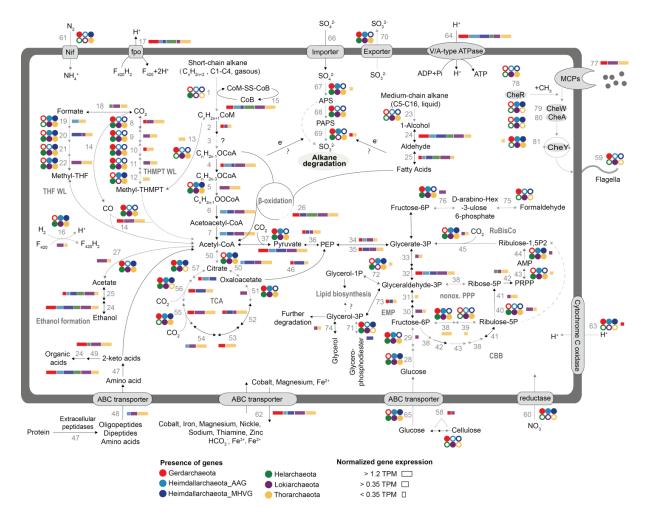


Fig. 4. Key potential metabolic pathways of Asgard archaea. The whole pathway was reconstructed based on all available Asgard MAGs (Table S5). Circles with different colours represent different phyla or MAGs. For comparison purposes, we used arrows of different types and colours. Black arrows indicate genes found in all Asgard MAGs, and grey arrows represent genes that are present in a subset of MAGs. Dashed grey arrows show pathways that are missing from all MAGs. Detailed metabolic information for the MAGs is available in Table S6. The relative abundance of the transcripts (Transcripts Per Million reads, TPM) for each gene was marked with proportionally sized rectangles and represented at the phylum level. Odinarchaeota were excluded because no Odinarchaeota MAG was identified in the present study.

4.8. Supplementary tables

Supplementary Table 1. Overview of 172 archaeal libraries that include Asgard archaeal 16S rRNA sequences.

sequences.							
Library/s tudy	Site location	Biotope	Salinity	Temperatur	Life style	pН	Depth (m
The context of this statement is not immediately apparent (specific instruction a	Mandovi and Zuari river, India	Estuarine sediment	2	2	2	2	5
Please list references, unless that is specified in the table.	Lake Ørn, Denmark	Freshwater sediment	1	1	2	NA	4,5
Microbial methane cycling in a terrestrial mud volcano in eastern Taiwan	Kuan-Shan area, eastern Taiwan	Mud volcano	2	2	2	2	NA
Comparison of archaeal and bacterial community structures in heavily oil-conta	Jidong Oilfield near Bohai Bay, China	Soils	1	1	3	NA	NA
An integrated study reveals diverse methanogens, Thaumarchaeota, and yet-un		Hot spring	2	3	2	2	NA
Novel uncultured Chloroflexi dechlorinate perchloroethene to trans-dichloroethe		Marine sediment	2	1	2	NA	NA
Archaeal diversity and the prevalence of Crenarchaeota in salt marsh sediment	<i>v</i> ,	Estuarine sediment	2	2	2	1	NA
Niche Separation of Methanotrophic Archaea (ANME-1 and -2) in Methane-S	· · · ·	Cold seep	2	1	2	NA	NA
Archaeal diversity in tidal flat sediment as revealed by 16S rDNA analysis	Dongmak, Korea	Marine sediment	1	2	3	1	0,05
An integrated study reveals diverse methanogens, Thaumarchaeota, and yet-un		Hot spring	2	3		2	NA 0
Life without light: microbial diversity and evidence of sulfur- and ammonium-ba Evidence for anaerobic oxidation of methane in sediments of a freshwater systemeter and the systemeter a		Freshwater sediment Freshwater sediment	1	2	2	2 NA	13
Comparison of microbial communities associated with phase-separation-induce		Hydrothermal habitat	•	3	2	NA	NA
Stratification of Archaeal communities in shallow sediments of the Pearl River		Estuarine sediment	2	2	2	2	0.5
Enrichment and cultivation of prokarvotes associated with the sulphate-methan	/	Marine sediment	2	1	2	NA	NA
Vertical Distribution and Diversity of Archaea in Nether Sediment from a Mud		Cold seep	2	1	2	NA	3288
Diversity, abundance and distribution of amoA-encoding archaea in deep-sea n	1 /	Cold seep	2	1	2	NA	783
Microbial Communities in Methane- and Short Chain Alkane-Rich Hydrotherm		Hydrothermal habitat	2	3	2	NA	NA
Microbial communities of deep-sea methane seeps at hikurangi continental mar		Cold seep	2	1	2	NA	1114
Composition of archaeal community in a paddy field as affected by rice cultivation	<u> </u>	Soils	1	2	3	NA	NA
Biogeochemistry and community composition of iron- and sulfur-precipitating m		Mud volcano	2	1	2	NA	3024
Biogeochemical processes and microbial diversity of the Gullfaks and Tommeli			2	1	2	NA	150
Physiological and molecular characterization of a microbial community establish		Soils	1	1	3	2	NA
Stratified communities of active archaea in shallow sediments of the pearl river		Estuarine sediment	2	2	2	2	0,6
Diversity of methanogenic archaea in a mangrove sediment and isolation of a n	Mzinga creek, Tanzania	Mangrove sediment	1	2	2	2	NA
Bacterial sulfur cycling shapes microbial communities in surface sediments of a	Logatchev hydrothermal field, Mid Atk	Hydrothermal habitat	2	3	2	NA	3000
Bacterial and archaeal diversity in surface sediments from the south slope of the	South China Sea, China	Marine sediment	2	1	2	NA	1285
Variability in microbial community and venting chemistry in a sediment-hosted	Okinawa Trough, Japan	Hydrothermal habitat	2	3	2	1	NA
Taxonomic and Functional Metagenomic Profiling of the Microbial Community		Freshwater sediment	2	1	2	2	2,4
Culture-Dependent and -Independent Characterization of Microbial Communitie		Hot spring	2	3	2	NA	23
Molecular biological and isotopic biogeochemical prognoses of the nitrification-			2	1	2	NA	9760
Methanogen diversity evidenced by molecular characterization of methyl coenz		Hydrothermal habitat	2	3	2	NA	NA
Stratified active archaeal communities in the sediments of Jiulong River estuary		Estuarine sediment	2	1	2	NA	3
Diverse and novel nifH and nifH-like gene sequences in the deep-sea methane		Cold seep	2	1	2	NA	840
Microbial diversity in sediments associated with surface-breaching gas hydrate	Gulf of Mexico	Marine sediment	2	1	2	NA	NA
Molecular characterization of potential nitrogen fixation by anaerobic methane-	· · ·	Mud volcano	2	1	2	NA	2050
Comparison of Archaeal and Bacterial Diversity in Methane Seep Carbonate N		Cold seep	2	1	2	NA	800
Isolation and Characterization of a Thermophilic, Obligately Anaerobic and Het		Hydrothermal habitat	2	3	2	NA	NA
Microbial diversity of cold-seep sediments in Sagami Bay, Japan, as determined Spatial distribution of viruses associated with planktonic and attached microbial		Cold seep Hydrothermal habitat	2	1 3	1	NA NA	NA NA
Use of 16S rRNA gene based clone libraries to assess microbial communities p		Mud volcano	2	1	2	1NA 2	1673
Anaerobic oxidation of methane at different temperature regimes in Guaymas I		Hydrothermal habitat	2	2	2	NA	NA NA
Influence of deglaciation on microbial communities in marine sediments off the		Marine sediment	2	1	2	NA	40
Prokaryotic diversity and metabolically active microbial populations in sediment		Marine sediment	3	1	2	NA	600
Comparison of the microbial diversity in cold-seep sediments from different dep		Cold seep	2	1	2	NA	615
Prokaryotic diversity, distribution, and insights into their role in biogeochemical		Marine sediment	2	1	2	NA	2135
Diversity and spatial distribution of prokaryotic communities along a sediment v		Mud volcano	2	1	2	NA	NA
Unique clusters of Archaea in Salar de Huasco, an athalassohaline evaporitic b	/	Freshwater sediment	2	1	2	NA	NA
Temporal evolution of methane cycling and phylogenetic diversity of archaea in		Marine sediment	2	1	2	NA	2893
A long-term cultivation of an anaerobic methane-oxidizing microbial community	Omine Ridge, Japan	Cold seep	2	1	2	NA	2533
Comparative analysis of archaeal 16S rRNA and amoA genes to estimate the a	East Sea, Korea	Marine sediment	2	1	2	NA	500
Diversity of Archaea in marine sediments from Skan Bay, Alaska, including cu	Skan Bay, USA	Marine sediment	2	1	2	NA	65
Diversity of prokaryotes and methanogenesis in deep subsurface sediments fro	Nankai Trough, Pacific ocean	Marine sediment	2	1	2	NA	4791
An anaerobic methane-oxidizing community of ANME-1b archaea in hypersali	Gulf of Mexico	Marine sediment	3	1	2	NA	876
Dominance of putative marine benthic Archaea in Qinghai Lake, north-western	Qinghai Lake, China	Freshwater	2	2	1	2	0,1
Diversity and distribution of methanotrophic archaea at cold seeps	Black Sea, Atlantic Ocean	Cold seep	2	1	2	NA	230
Subseafloor microbial communities associated with rapid turbidite deposition in	Gulf of Mexico	Marine sediment	2	1	2	NA	NA
Diversity and community structure of archaea in deep subsurface sediments fro		Marine sediment	2	1	2	NA	NA
Variations in archaeal and bacterial diversity associated with the sulfate-methan		Marine sediment	2	1	2	NA	587
Cultivation of methanogens from shallow marine sediments at Hydrate Ridge, O		Marine sediment	2	1	2	NA	1180
Archaeal diversity and distribution along thermal and geochemical gradients in		Hydrothermal habitat	2	3	2	NA	NA
Characteristics of Microbial Community Structures in the Ulleung Basin, East S	<i>U</i> /	Marine sediment	2	1	2	NA	NA
Microbial communities associated with geological horizons in coastal subseafloo	/	Marine sediment	2	1	2	NA	1225
Biogeographical distribution and diversity of microbes in methane hydrate-bear	6 /	Marine sediment	2	1	2	NA	900
Biogeographical distribution and diversity of microbes in methane hydrate-bear		Marine sediment	2	1	2	NA	5086
16S rDNA diversity analysis of mangrove soil	China	Mangrove sediment	NA	NA	NA	NA	NA
A Microbial Community within a Natural Asphalt Lake	Pitch Lake, Trinidad	Freshwater	NA	NA	NA	NA	NA
Aerobic and anaerobic methane oxidation in terrestrial mud volcanoes in the No	Napoli mud volcano, Eastern Mediterra	Mud volcano	NA	NA	NA	NA	NA

Acardine submanophic correntizes in solution promoted as used of The col Carel Line Science 1 AA NA	Library/study	Site location	Biotope	Salinity	Temperatur	Life style	рH	Depth (m
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Novel microbial diversity retrieved by an autonomous rover in the workd's deepe El Zacaton, Mexico Others NA NA <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>								
Photosynthesis versus exopolymer degradation in the formation of microbialites Kiritimati, Republic of Kiribati Others NA								
Prokaryotic Abundance and Community Composition in a Freshwater Iron-Ricl Hiroshima, Japan Freshwater sediment 1 2 2 1 NA Prokaryotic communities of the top 1 m below sea floor sediment in an eastern Amsterdam mud volcano, Mediterrane Mud volcano NA NA NA NA		· · · · ·			-			
Prokaryotic communities of the top 1 m below sea floor sediment in an eastern Amsterdam mud volcano, Mediterrane/Mud volcano NA NA NA NA NA		· · ·	1					
		· · ·		NA			NA	
			Mud volcano					

Library/study	Site location	Biotope	Salinity	Temperatur	Life style	pН	Depth (m
Recording of climate and diagenesis through sedimentary DNA and fossil pigme	Laguna Potrok Aike, Argentina	Freshwater sediment	NA	NA	NA	NA	NA
Seasonal variation of particle-associated and free-living prokaryotic communities	Gulf of Trieste, Northern Adriatic Sea	Marine water	NA	NA	NA	NA	NA
soil of JingHe saltworks belt in the Lake Ebinur	Lake Ebinur, China	Soils	NA	NA	NA	NA	NA
Spatial and temporal changes in microbial diversity of the Marmara Sea Sedime	Marmara Sea, Turkey	Marine sediment	2	2	2	NA	NA
Spatial and Temporal Changes in the Microbial Communities of Lithifying and N	Bahamas	Others	NA	NA	NA	NA	NA
Spatial distribution of viruses associated with planktonic and attached microbial	Okinawa, Japan	Hydrothermal habitat	NA	NA	NA	NA	NA
Spatial structure and activity of sedimentary microbial communities underlying a	Gulf of Mexico, Atlantic Ocean	Cold seep	NA	NA	NA	NA	NA
Studies on the archaeal diversity in the coconut husk retting zone, Kerala, India	Kerala, India	Soils	NA	NA	NA	NA	NA
The archaea community patterns from sediments of Poyang Lake, China	Poyang Lake, China	Freshwater sediment	NA	NA	NA	NA	NA
The Community Structure of Archaea in Deep Sea Sediment of the Ulleung Ba	Ulleung Basin, Korea	Marine sediment	NA	NA	NA	NA	NA
The Dark Side of the Mushroom Spring Microbial Mat: Life in the Shadow of C	NA	Hot spring	NA	NA	NA	NA	NA
The study of taxonomic composition of microorganisms in surface sediments an	Lake Baikal, Russia	Freshwater sediment	NA	NA	NA	NA	NA
Thermal and geochemical zonation of microbial biogeography in Guaymas Basir	Guaymas Basin, USA	Hydrothermal habitat	NA	NA	NA	NA	NA
Three spatial distribution of microbial communities in the Lei-Gong-Huo mud vo	Lei-Gong-Huo mud volcano, Taiwan	Mud volcano	NA	NA	NA	NA	NA
Tropical aquatic archaea show environment-specific community composition	Ilha Grande, Brazil	Freshwater	NA	NA	NA	NA	NA
Uncultivated archaea from sulfate-reducing cave biofilm	Frasassi cave, Italian	Freshwater	NA	NA	NA	NA	NA
Unexpected diversity and complexity of the Guerrero Negro hypersaline microb	Guerrero Negro	Hypersaline habitat	NA	NA	NA	NA	NA
Vertical distribution and diversity of bacteria and archaea in methane-rich cold s	Okhotsk Sea, Pacific Ocean	Marine sediment	NA	NA	NA	NA	NA
Vertical distribution of prokaryotes and responses to their environment in Hongh	Honghu Lake, China	Freshwater sediment	NA	NA	NA	NA	NA
Archaeal diversity in ODP legacy borehole 892b and associated seawater and se	Cascadia margin, Pacific Ocean	Cold seep	2	1	2	NA	NA
Archaeal diversity in ODP legacy borehole 892b and associated seawater and se	Cascadia margin, Pacific Ocean	Cold seep	2	1	2	NA	600
Mariana forearc serpentine mud volcanos harbor novel communities of extremop	NA	Mud volcano	NA	NA	NA	NA	NA
Novel bacterial and archaeal lineages from an in situ growth chamber deployed a	Snake Pit on, Mid-Atlantic Ridge	Hydrothermal habitat	2	3	2	NA	NA
Phylogenetic Diversity of the Archaeal Component in Microbial Mats on Coral-I	Black Sea, Atlantic Ocean	Cold seep	2	1	2	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Artificial pond at the Aalborg Universit	Freshwater sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Estuary of Mariagerfjord, Denmark.	Estuarine sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Madum lake, Denmark	Freshwater sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Nors lake, Denmark	Freshwater sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Poulstrup lake, Denmark	Freshwater sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Seashore close to Dokkedal, Denmark	Marine sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Seashore of Fanø, Denmark	Marine sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Ulvedybet in Limfjorden, Denmark	Marine sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Ulvedybet in Limfjorden, Denmark	Marine sediment	NA	NA	NA	NA	NA
Retrieval of a million high-quality, full-length microbial 16S and 18S rRNA gene	Voer Å, Denmark	Freshwater sediment	NA	NA	NA	NA	NA
Stratification of archaea in the deep sediments of a freshwater meromictic lake:	Lake Pavin, France	Freshwater sediment	1	1	2	NA	NA
The extent of protist diversity: insights from molecular ecology of freshwater eul	NA	Freshwater	NA	NA	NA	NA	NA

For samples collected from marine sediments without specific instructions, the salinity, temperature, and life style scores are presumed to be 2, 1, and 2, respectively. Relative abundance values are from published papers.

Groups and subgroups	Minimum intragroup similarity (%)	Number of OTUs	Number of DNA sequences	Number of RNA sequences	RNA proportion (%)	Bootstrap value
Kariarchaeota	81	7	1	157	99.4	100
Balderarchaeota	88	4	0	27	100	100
Hodarchaeota	78	4	0	24	100	100
Lagarchaeota	78	9	0	351	100	100
Gerdarchaeota	85	8	1	97	99.0	100
Lokiarchaeota-1	86	16	12	49	80.3	100
Lokiarchaeota-2	80	121	117	1771	93.8	100
Lokiarchaeota-3	88	28	47	873	94.9	96
Lokiarchaeota-4	83	200	4238	1784	29.7	88
Thorarchaeota	86	13	14	391	96.5	100
Odinarchaeota	92	4	4	0	0	100
Heimdallarchaeota-AAG	87	11	192	78	28.9	100
Heimdallarchaeota-MHVG	80	5	3	5	62.5	100
Heimdallarchaeota	71	16	195	83	29.8	NA
Lokiarchaeota (1-4)	80	365	4414	4477	50.4	98
ungrouped	NA	33	17	199	92.1	NA

Supplementary Table 2. Characteristics of groups within the Asgard archaea.

I I I I I I I I I I I I I I I I I I I	-								5-								-		_		_	_	_	_						
Groups/subgroups		В	aldera	irchae	ota		Hodar	chaeot	a			Karia	chaeo	ota					Lag	archa	eota					(Gerdar	chaeo	ta	
		P010	P010	P010	P010	P010	P010	P010	P010	P010	P011	P010	P010) P010	P010	P010	P011	P010	P010	P011	P011	P010	P010	P011	P010	P010	P010	P011	P010	P011
	OTUs	6434	7684	7772	7809	6967	7430	8456	8559	7942	3005	7042	8520	9197	6631	1903	4522	2360	9190	0414	0710	9294	2583	2970	9075	8535	8054	3413	8271	3292
	OBEP010643	06	81	15	72	48	07	64	42 69 3	68	13	39	77	44	14	42	07	85	98	46	51	60	42	44	79	00	04	76	44	19
	OBEP010043 OBEP010768	4 91.1	100.0	94.9	2 91.0	68.3	68.0	5 69.0	69.3	70.1	69.	2 70.6	6 68.	9 70.2	2 70.0	69.7	70.8	70.4	71.2	70.3	70.1	71.1	70.3	72.5	5 69.9	69.4	70.3	70.9	71.3	70.2
Balderarchaeota	OBEP010777	94.9			95.1	68.4			69.3	71.4	70.						70.6	71.0	71.2	70.8	69.9	71.1	70.8	71.9			71.2	71.5		71.1
	OBEP010780 OBEP010696	99.0 68.1	91.0 68	0 95.1 3 68.4		0 68.1	69. 80.8	68.5 96.6	69.4 93.1	71.5	70. 69.		69. 69		1 71.3 5 69.8	70.3	70.8	70.6	71.3	70.6	70.2	71.2	70.5	72.3		69.7	71.8 69.6	72.2		69.4
Hodarchaeota	OBEP010743	68.8	68.0	68.8	69.1	80.8	100.0	80.6	80.2	70.5	68.	3 69.9	69.	2 69.9	9 70.6	71.4	72.5	71.4	71.5	69.1	70.0	71.5	71.1	71.2	2 69.5	69.4	70.4	69.7	68.6	5 70.2
riouarchaeota	OBEP010845 OBEP010855	68.5	69.0	0 68.5	68.5	96.6	80.0		95.6	70.9	68.	7 70.9	0 69.		1 70.3	74.5	74.2	72.9	73.4	72.1	71.5	73.3	72.7	71.0		69.9	69.8 71.0	70.8	69.1	69.6
	OBEP010855 OBEP010794	9 69.3 2 71.7	70.			93.1			71.2	71.2	69.: 92.:	5 70.8 2 89.0	8 69. 94.		9 70.7		73.5	77.2	72.8	75.5	74.6	72.7	72.0	76.2			76.8	76.9		
	OBEP011300	5 70.3	69.2					68.7	69.5	92.2	100.	87.8	94.	5 91.0				76.3	76.4	74.9	74.3	76.2					75.4	74.9	73.8	8 75.4
Kariarchaeota	OBEP010704 OBEP010852	2 71.6 70.0	68.9	6 71.3 6 69.9		1 70.9 3 69.4		9 70.9 2 69.4	70.8	89.0 94.0	87. 94.	8 100.0 5 89.1	89.	1 88.0 0 94.2	0 89.1 2 94.0	78.5	79.0	78.7	78.0	76.6	76.1	77.9	78.3	77.2			77.2	78.0	76.9	
	OBEP010032	71.4	70.2		71.1	69.6		70.1	70.9	93.0	91.		94.		95.8	76.8	78.0	77.0	77.4	75.2	74.0	77.3	76.8	75.8			76.2	77.3	75.0	76.2
	OBEP010663	1 71.5	70.0	0 71.6		69.8	8 70.0		70.7	94.4	92.				8 100.0	77.0	77.9	77.1	77.5	75.6	74.4	77.3	76.9	75.8			76.6	77.3		
	OBEP010190 OBEP011452	70.4 70.8	69.	7 70.7		74.2 73.5			73.1 73.5	78.0	77.					91.7	91.7	92.8	91.5 92.7	86.9 87.5	86.2 86.5	91.7 92.8	92.3	85.1				77.5		
	OBEP010236	70.4	70.8	3 71.0		5 72.9	71.4	4 72.9	72.2	77.2	76.	3 78.7	77.			92.8	90.3	100.0	91.4	86.9	86.1	91.5	99.5	84.8	8 74.9	75.9	76.3	77.1		76.3
The second second	OBEP010919 OBEP011041	71.2				72.7			72.8	77.1	76.4			4 77.4	4 77.5	91.5	92.7	91.4	100.0	88.5	87.5	99.8	91.1	86.4				77.8		
Lagarchaeota	OBEP011041 OBEP011071	4 70.3 70.1	70.	3 70.8 1 69.9	3 70.6 70.2	5 71.1 2 70.3	69.1	72.1	71.4	75.5	74.	9 76.6 3 76.1	5 75.	1 /5.2 8 74.2	2 75.6 2 74.4	86.9	87.5	86.9	88.5 87.5	93.0	93.0 100.0	88.5 87.5	86.8	84.6		5 74.0 73.6	76.7	75.0	73.8	3 76.7 3 75.4
	OBEP010929	71.1			71.2	2 72.6			72.7	77.0	76.		77.	2 77.3	3 77.3	91.7	92.8	91.5	99.8	88.5	87.5	100.0	91.2	86.4	4 76.5	76.4	78.0	78.0	76.2	
	OBEP010258 OBEP011297	70.3 72.3				5 72.5 5 70.3			72.0	77.1	76.		8 76. 2 74.		8 76.9 8 75.8		89.8 85.6	99.5 84.8	91.1 86.4	86.8 84.6	85.8 85.7	91.2 86.4	84.5	84.5	5 74.8 75.0		76.0	76.8		
	OBEP010907	70.5	69.9			69.4		5 69.4	70.5	76.1	74.				7 76.1				76.4	74.6	74.0	76.5		75.0		93.6	94.8	94.6		94.6
	OBEP010853	71.1	69.4		71.2	69.7	69.4		71.3	75.8	74.				5 76.2		75.9	75.9	76.3	74.0	73.6	76.4		74.0		100.0	92.6	92.8	92.8	92.4
Gerdarchaeota	OBEP010805 OBEP011341		70.3		71.8	8 69.6 2 70.6	6 70.4	4 69.8 7 70.8	71.0	76.8	75.4						77.5	76.3	77.9	76.7	75.4	78.0	76.0	76.4		92.6 92.8	93.4	93.4	92.2	99.8
	OBEP010827	1 71.3	71.4	4 70.7	71.4	69.3	68.0	69.1	70.7	75.0	73.	8 76.9	74.	0 75.0	74.9	76.4	76.4	75.3	76.1	73.8	73.8	76.2	75.0	75.1	7 93.6	92.8	92.2	93.9	100.0	92.0
	OBEP011329 KP091041	2 71.7	70.2			8 69.4			70.8	76.6		4 77.2	2 76.	0 76.2	2 76.6		77.3	76.3	77.8	76.7	75.4	77.8	76.0	76.3		5 92.4 71.9	99.8	93.2 72.9	92.0	
	AB019716	71.6			71.6	71.5	71.3	3 72.7 5 72.9	72.2	71.2	71.	5 72.8 4 72.6	5 70. 5 70.	1 70.5	5 71.1	73.9	74.2	73.6	74.8	74.9	73.8	74.7	73.1	73.0			73.7	72.9		
	AB797480	71.3	70.		71.1	71.6	71.6	5 72.2	71.9	72.5	71.	7 73.0	71.		5 71.8	74.6	74.6	73.7	74.8	74.2	73.2	74.8	73.2	73.2	2 73.1	72.2	74.0	73.3	72.0	73.9
Haimdallamhasata AAC	OBEP010863 JX000838	70.8	70.2 70.8		70.5	5 71.1 7 70.6	70.0		71.3	71.7	71.	2 72.2 1 72.1		1 70.1 1 70.2	7 71.3	74.5		73.9	74.8	74.3	73.2	74.8	73.4	73.0			73.5 73.9	72.6		
Heimdallarchaeota-AAG	DQ640139	71.6				71.1			71.8	71.8	71.	4 72.8			9 71.1	74.3	74.7	73.6	75.0	74.0	72.9			73.0			73.7	72.9		
	GU553642	72.1			71.8	3 71.5			72.5	72.0	71.	1 72.5			5 71.2	74.0		73.7	75.0	74.4	73.1	74.9	73.2	73.6			74.4	73.3		
	JQ817966 AB797473	72.1			5 71.9 F 71.3	0 71.0 3 72.3			72.2	71.3	71.	4 72.4 0 72.6		2 70.6	5 71.1	73.3	73.5	73.0	74.3	74.5	73.4	74.2	72.7	73.4			74.0	72.8	71.9	73.9
	OBEP011316	69.7	68.0	68.8	3 70.0	68.0	67.1	69.2	69.2	77.0	76.	5 77.3	77.	0 76.4		73.2	73.3	73.0	73.5	73.6	73.3	73.3	72.9	72.2	2 73.4	72.8	72.8	73.7	71.6	72.8
Heimdallarchaeota-MHVG	OBEP010918	69.8	67.9			66.9			68.4	76.4	75.	7 76.5			5 76.3			72.5	73.9	73.3	73.0	73.8	72.5	72.3				73.8	71.6	72.5
	JX000774 JO989558	68.6 68.1	66.9	9 68.2 5 68.3	2 68.5 3 68.4	68.4 67.7		69.8 6 68.6	69.7 68.9	75.9	75.				5 76.3 7 75.9	72.6	72.4	72.3	73.5	73.1	72.2	73.4	72.1	73.0			71.2	72.6		71.2
	OBEP011394	2 73.5	72.	5 71.9	73.0	72.6	70.9	73.0	72.9	79.1	78.		78.		5 80.0	82.4			83.9	81.9	81.9	83.7	81.4	82.2			79.2	79.8	77.8	
	OBEP011353 KC682077	9 74.0 72.8	72.5		2 73.5	72.8	71.	1 73.6 1 73.3	73.7	79.0	77.	6 80.4 7 79.1	1 78. 78.		1 80.0 7 79.8	81.8	84.1	81.0	83.2 83.9	81.5	81.1	83.0 83.8	80.5	82.2	2 78.5 8 76.9	5 78.0 76.9	79.6 78.0	80.0	77.8	8 79.6 78.0
	AB019721	73.6				72.5		3 72.9	72.5	78.8	77.	5 79.7	78.		7 79.5		82.8	81.0	83.3	81.3	81.5	83.1	80.7	82.2			79.0	79.6		
Lokiarchaeota-1	OBEP011133	1 72.6				72.1			73.2	78.8	77.	8 79.0							83.8	81.4	81.0	83.7	81.2	81.5			77.8	78.3		
	OBEP011050 OBEP010829	8 72.9 8 73.8				72.6			73.3	78.1	77.	2 78.9			2 79.4 8 79.9	0.000	83.6	81.6	84.2 84.0	81.5	81.1	84.0 83.8	81.2	81.3			78.2	78.6		
	OBEP010681								72.4	78.0	76.	7 78.8			8 78.6			80.6	82.5		80.4	82.4		81.9			78.6	78.8		
	OBEP011307	73.3	72.3	3 72.2 2 72.8	2 72.8	8 71.6	5 70.3	3 72.6 5 72.9	72.7	79.3	77.	7 80.1	78.	7 79.2	2 80.1			80.8	83.5	81.2 82.0	81.0	83.3	80.4	82.0		2 77.7	79.4	79.7	77.5	5 79.4 0 79.0
	OBEP010822 OBEP010685	4 73.7 72.8	72.9			5 71.7	71.0		72.8	77.6	76.	2 78.3 6 79.0	77.	6 77.7	5 78.3 7 78.4			82.2	84.3 84.1	82.0	81.3	84.1 83.9	81.7	82.1		5 76.7 76.5	79.0 78.5	79.3	77.0	
	OBEP011043	72.8	72.3	3 71.8	3 72.9	72.8	3 71.8	3 73.6	72.4	78.3	76.	8 78.8	3 77.	8 77.9	9 78.8	81.3	84.3	81.6	84.1	81.7	80.9	83.9	81.2	81.4	4 77.6	5 76.6	78.3	79.1	76.6	5 78.3
	OBEP011240 KJ569649	9 72.8 73.1				71.4		2 72.7	72.1	78.1	76.		77.		2 78.6		83.8	80.8 81.9	83.1 84.2	80.9 82.1	80.2	82.9 84.1	80.4	81.6		76.6	78.8	79.5	77.0	
	OBEP011226	4 72.8	72.8	72.5		71.9	71.0	5 73.1	72.9	77.8	77.	0 78.9	77.		8 78.3	82.0	84.4	81.6	84.0	82.0	81.1	83.9	81.1	81.9		76.4	78.8	79.0	76.8	
	OBEP010059	72.9		7 72.5	72.6	5 71.9	71.2	2 73.1	72.7	77.9	76.	7 78.8	3 77.		7 78.5	81.7	84.4	82.0	83.8	81.6	81.1	83.6	81.5	82.2		76.7	78.9	79.0		
Lokiarchaeota-2	OBEP010887 OBEP011263	72.8				3 72.0) 72.7			73.0 73.6	78.1	77.	1 79.0 7 79.3			9 78.6 1 79.7			81.7	84.3 84.4	82.3 82.5	81.2	84.2 84.3	81.2	82.0		76.8	79.1 80.5	79.1		
	JQ989629	72.7	72.3	3 72.1	72.6				72.7	77.6	76.4		77.		1 78.0		83.3	81.4	83.8	81.4	80.7	83.6	80.9	82.0	77.3	76.4	78.6	79.0	76.7	78.6
	OBEP010865	73.0				72.6			73.3	77.8	76.4				3 78.2			81.9	84.1	81.8	81.6	83.9	81.4			3 77.0	79.0	79.4		
	OBEP010846 OBEP010679	72.5 73.1	72.0			71.3	71.2	2 72.5 4 73.0	72.0	78.0	76.	3 79.1 1 78.4		7 78.3	3 78.6 2 77.9			81.0 81.8	83.1 83.8	81.0 81.5	80.0 81.0	82.9 83.7	80.6	81.6		76.4 76.5	78.7 78.6	79.4	76.7	
	OBEP011303	9 73.2	72.0	5 72.3	73.2	2 72.3	71.9	73.5	73.1	78.2	77.	1 79.0	78.	0 78.1	1 78.8	81.8	84.5	81.7	84.0	82.2	81.4	83.9	81.2	81.8	8 78.4	77.7	79.8	79.5	77.2	2 79.8
	AB019720 OBEP010053	73.2				72.0			73.0	78.2	77.	1 79.2 5 78.2		8 78.1 0 77.4	1 78.8 4 78.3		84.0 81.9	81.8	84.2 82.0	82.1 80.6	81.0	84.1 81.8	81.3	82.4		76.4	78.9 78.2	78.8		
	OBEP010055 OBEP010166	8 72.7							72.5	78.1				-			83.4	81.0	82.9	81.0	80.4	82.8	80.5	80.0				79.4		
	OBEP010231	72.3	71.3	7 71.5	5 72.2	2 71.5	70.0	71.9	72.0	78.0	76.	7 78.2			3 78.5	81.0	82.9	80.9	82.8	80.7	79.9	82.7	80.4	79.8	8 76.8	3 76.7	77.8	78.8	76.7	77.8
	OBEP011071- OBEP011226	4 72.2 4 72.5	71.4	4 71.0 0 71.2	72.0	2 71.8	3 70.0	72.4	72.1	78.7	77.4	4 78.1 4 78.3	77.	4 77.0	5 78.8 4 78.2	8 79.8	82.0	79.1	81.7 82.5	80.6	78.9	81.5	78.8	8 78.8			79.0	79.0	76.3	
	OBEP 011220 OBEP 011057	71.9			71.8	3 71.1	70.4	5 71.7	72.3	77.2	76.		76.	8 77.3	3 78.0				82.2	80.4	80.0	82.0	79.9	80.1			78.7	78.8	77.1	
	OBEP010071	4 72.7	72.2	2 71.7	72.4			0 71.3	72.0	77.4	76.	3 77.9	77.	0 77.1	1 77.8	8 79.5	82.0	79.3	81.1	79.4	78.9				76.5	76.7	77.4	78.7		
	OBEP011392 OBEP011121					71.5			72.2	78.5	76.			5 77.9			83.0		82.8	80.9 81.1			80.4				78.6	78.9		
	OBEP011121 OBEP011062	72.2	71.0						71.5	78.7							001-	00	82.4					0.010			79.1			
	OBEP011059	72.0	71.3			71.4			72.5	77.6	77.	3 78.8			1 78.5	81.0			83.1	81.1	80.2	82.9	80.4				78.0	78.5	76.6	78.0
Lokiarchaeota-3	OBEP011151 OBEP010024								71.8	76.9									81.8 82.4	80.0 80.8		81.6 82.2								
Lonan chatota-o	OBEP011071	71.9	71.0	5 71.1	71.8	8 70.0	69.1	70.9	71.8	77.4	77.	2 77.3	77.	3 77.5	5 78.1	81.2	83.2	80.5	83.0	81.7	80.7	82.8	80.2	80.5	5 76.9	76.3	77.8	77.9	76.0	77.8
	OBEP011112								72.7	78.5									83.2			83.0							76.7	7 78.6
	OBEP011282 OBEP011031	9 73.3 4 72.3				71.6	5 70.8 7 70.4		72.3	78.5	77.						83.3	81.1	83.1 82.3	81.1 81.4	80.2	83.0 82.2	80.6	80.3			78.6	78.9		
	AF068822	72.4	71.3	7 71.3	72.2	2 71.6	70.4	4 72.0	72.5	77.7	76.	9 78.9	77.	5 77.9	9 78.7	80.7	83.2	80.7	82.7	81.0	79.8	82.6	80.3	80.1	1 77.4	77.0	78.7	79.3	77.0	78.7
	OBEP011131								72.6	78.1									82.3				80.1				78.1	78.4		
	AF419642 OBEP010875	73.0 8 72.1							72.1	77.9									82.9 81.3		79.6									
	OBEP011112	6 71.7	71.2	2 70.5	71.6	5 71.1	70.1	7 71.3	71.6	78.0	76.	8 78.3	77.	6 77.2	2 78.5	80.7	83.1	80.8	82.1	80.4	79.6	81.9	80.3	80.2	2 77.8	3 76.7	78.8	79.2	76.7	78.8
	OBEP011063 OBEP010255	71.8 73.1				3 71.0 71.1			71.8	78.7	77.								82.6 82.4		80.1	82.4 82.3						79.0		
	OBEP010233		71.4							78.2									82.8	81.1		82.7	80.4	80.4			78.2			
P		-	-	-	-	-	-	-	_	-	-		-	-	-	-	-	-								-		-	-	

Supplementary Table 3. The 16S rRNA gene identity (%) of Asgard groups.

Groups/subgroups		B	alde ra	rchae	ota	1	Hodar	chae of	ta		J	Karian	chaeo	ta					Lag	archae	eota					0	le rdar	chaeo	ta	
			1.010	P010	P010									P010				P010		P011				P011		P010				
	OTUs	6434 06	7684 81	7772 15	7809 72	6967 48	7430 07	8456 64	8559 42	7942 68	3005 13	7042 39	8520 77	9197 44	6631 14	1903 42	4522 07	2360 85	9190 98	0414 46	0710 51	9294 60	2583 42	2970 44	9075 79	8535 00	8054 04	3413 76	8271 44	3292 19
	OBEP010833 OBEP010148	72.9	71.8	71.1	72.7	71.2	70.3	72.0	71.9	77.4	76.2	78.0	76.7	76.6	77.6	80.0 80.9	82.0 83.1	80.2	81.4 82.9	80.1 81.1	79.6	81.2	79.7 80.4	79.4 80.5	76.5	76.7	78.0	78.2		1.010
	OBEP011065 OBEP011331	71.7	71.2				71.1	72.0	72.2	77.6	76.7	78.3	77.6	78.0	78.8	80.5	82.9	80.8	82.4	81.2	80.4		80.3 79.6	80.3	76.6	76.4	77.9	78.2		77.9
	OBEP010786	73.1	72.7	72.2	73.1	71.3	70.6	72.0	72.6	78.3	77.3	78.5	77.6	78.0	78.6	79.8	82.2	79.5	81.2	79.5	78.8	81.0	79.3	79.0	77.0	76.3	77.8	78.8	76.5	77.8
	OBEP010657 OBEP011047	72.8	71.7	71.5			70.7	72.0	72.2	78.4 78.1	77.1 77.3 77.2	79.4 78.4	78.0 77.9	78.4 78.0	79.3	80.7 80.5	83.1 82.8	80.6 79.9	82.9 81.9	81.0 81.1	80.1 79.3	82.8 81.8	80.3 79.6	80.9 79.4	77.1	76.9	78.5 79.5	79.1	76.7	78.5
	OBEP011022: OBEP011221	72.3	71.8	70.9		71.1	69.8 69.4	71.7	71.8	77.9	77.2	78.6	78.1	77.6	78.9 78.2	81.3 79.8	83.1 81.5	80.9 79.9	82.7 81.2	81.1 80.0	79.8 78.4		80.5 79.4	80.4 79.4	77.1	76.6	78.1 77.6	78.8 78.2		78.1
	AY835409 GU553554	73.3	72.1	71.6	73.3	71.5	70.6	72.0	72.4	78.2	76.6	78.5 78.8	76.9	77.0	78.1	80.0 80.9	82.3 82.8	79.7	81.6 82.5	80.8 81.0	80.1 79.6	81.4 82.3	79.2 80.0	79.5 80.3	77.2	76.9	78.0 78.5	79.1	77.1	78.0
	OBEP010650	73.1	72.0	71.6	72.9	70.7	69.8	71.3	71.8	77.9	76.3	78.3	76.9	77.0	78.0	80.5	82.7	80.7	81.7	80.6	80.4	81.5	80.3	80.1	78.0	77.6	78.5	79.5	77.4	78.5
	AY822005 OBEP011079	72.5	71.5	71.1	72.7	71.3	70.6	71.9	72.0 72.5	77.7 78.3	76.8	78.2 78.4	76.8 77.4	76.8	78.1 78.4	80.5 80.6	82.5 82.9	80.4 80.7	82.0 82.7	79.8 80.3	79.1 79.8	81.8 82.5	80.0 80.4	79.6 80.0	76.6	76.5	77.5 78.3	78.5 79.3		
	OBEP010625 OBEP010672	72.7	71.9	71.3		69.8	70.0	70.1	70.7	76.9 77.9	75.9 76.6	77.2	76.4	76.5	77.2	79.4 80.1	82.3 82.3	79.9 80.0	82.2 81.6	79.5 80.6	79.6	82.0 81.4	79.5 79.6	79.4 79.3	75.3	75.9	76.9 77.3	77.0	75.9	
	OBEP011017 OBEP011089	72.5	71.9	71.4	72.5	71.5	70.5	72.4	72.0	79.2 78.2	77.7	79.1 78.8	78.3 78.0	78.3	79.3 79.0	81.8 80.3	84.1 82.4	81.0	82.9 82.6	81.6	80.9 79.5	82.7 82.4	80.7 80.0	81.1	77.7	76.4	78.9	78.8	76.6	78.9
	EU910616	72.2	71.4	71.0	71.9	71.9	70.8	72.3	72.9	78.6	77.7	79.6	78.4	78.4	79.6	80.9	83.4	80.9	83.2	81.0	80.2	83.1	80.4	80.6	77.5	77.4	78.8	79.3	77.0	78.8
	OBEP011287. OBEP010921	73.0 72.9	72.3	71.9	72.8		70.7	72.9	72.8	78.4 78.0	77.3 76.5	78.8 78.5	77.6 77.2	77.8	78.5 78.3	80.7 80.7	82.7 82.9	80.4 81.1	82.4 82.8	80.8 80.8	79.3 79.7	82.2 82.6	80.0 80.6	80.1 80.1	77.0	76.6 76.6	78.2 78.1	78.7 78.3		78.1
	OBEP010136 OBEP011042	72.8	71.7	71.5 69.3	5 72.5 3 70.4	71.7	70.0	72.0	71.8	78.1	77.0	78.6	77.9	77.8	78.8 78.0	81.2 80.0	83.4 83.0	81.0	83.0 82.2	80.6 80.9	80.1 79.8	82.9 82.0	80.5 79.4	80.4 79.8	77.6	77.0	78.3	79.3	77.0	78.3
	OBEP011314 OBEP011045	72.7	72.1	71.6	72.5	71.0	69.5 70.5	71.4	71.5	78.0	76.7	78.3 78.4	77.3	77.5	78.3 78.7	80.6 81.0	82.6	80.5	81.7 82.5	80.5 81.4	79.2	81.5	80.0 80.2	80.0 80.5	76.2	76.5	77.8	78.5	76.6	77.8
	OBEP011038	72.4	72.0	71.3	72.2	71.8	71.1	72.3	72.7	78.1	77.0	78.4	77.8	77.8	79.0	80.7	83.4	80.4	82.4	81.3	80.4	82.2	80.0	80.5	77.1	76.2	78.4	78.5	76.2	78.4
Lokiarchaeota-4	OBEP010166 EU420711	72.7	72.0	71.6	5 72.6 3 72.7	71.1	70.0	71.7	71.9	78.2	77.4	78.6	77.5	77.7	78.8	80.6 80.8	82.7	80.5	82.7 82.0	80.3 80.6	79.3	82.5	80.2 80.0	80.1	77.0	76.5	78.2	78.6		
	OBEP011014 OBEP010899	71.9	71.2	70.8	3 71.6 3 72.5	70.2	69.7 69.8	71.0	71.3	77.7	77.3	78.2 79.0	78.1 78.3	78.6	78.9	80.6 81.3	83.3 83.7	80.2 81.2	82.3 83.2	80.5 81.0	79.9 79.8	82.1 83.1	79.7 80.8	80.2 80.1	77.3	76.1	77.9	78.8 79.9	76.5	77.9
	OBEP010004	71.6	71.5	70.7	71.2	71.3	70.6	72.0	71.0	76.7	75.3	76.8	76.8	76.4	77.5	80.0	81.5	79.4	80.1	79.9	78.5	79.9	79.2	78.6	76.8	76.0	76.9	78.3	76.3	76.9
	OBEP011017: OBEP010895	72.7	71.6	71.0	3 72.9	70.5	70.5	71.3	71.5	78.5 77.7	76.8	78.8	78.4 77.3	77.8	78.6	80.6 81.0		80.1 80.7	82.6 83.0	81.0 81.3	80.4 80.1		79.8 80.3	80.6 80.4	77.5	76.4	78.6 78.2	78.9	76.7	78.2
	OBEP011308: OBEP010080	71.4	71.8	70.9	71.6 73.0	71.1	70.9	71.6	71.7	77.7	76.6	76.3	77.0	77.9	77.8	78.0 81.2	80.0 83.1	77.6 81.0	78.9 83.2	79.1	78.4 79.9	78.7 83.1	77.3	78.1 80.8	77.8	76.6	78.4 78.4	78.1	78.0	
	OBEP011278 JF268336	72.4	71.6	71.2	72.2	72.0	71.0	72.6	72.5	78.8 78.2	77.6	79.1 78.8	78.4	78.0	79.2	80.8 80.9	83.5 83.2	80.9	82.3 82.3	81.3 80.8	80.1 79.9	82.1 82.1	80.6 80.1	80.1	77.7	76.8	79.3	79.1 79.4	76.6	79.3
	OBEP010719	72.6	72.1	71.4	72.4	71.2	70.4	71.3	71.4	78.2	77.1	78.9	78.0	77.6	78.9	80.4	82.9	80.4	82.7	80.5	79.5	82.5	80.0	80.6	778.4	77.1	79.3	79.7	77.4	79.3
	OBEP010098 OBEP010105	72.5	72.2				69.7 70.5	71.5	71.4	77.4	76.3	78.5 78.8	77.1	77.4	78.2	79.8 81.4	81.9	79.6	81.9 83.3	80.4 81.2	79.0		79.3 81.0	79.7	76.8	76.6	78.1 78.8	78.7		
	OBEP0113032 OBEP011039	73.0	72.2	71.9	72.8	71.3	70.1	71.5	71.6	78.8 77.1	77.9 76.9	79.1 78.4	78.5 78.1	78.4	79.5 78.4	81.2 80.4	83.7 82.4	81.3	83.2 82.3	81.0 80.7	79.9 79.6	83.1 82.2	80.8	80.1 80.1	78.5	77.2	78.9 77.6	79.7	77.2	78.9
	OBEP010025	73.1	72.4	71.9	72.9	71.6	70.4	72.0	72.4	78.7	77.7	79.2	78.4	78.7	79.6	81.3	83.7	81.3	83.3	81.3	80.4	83.2	80.9	80.5	77.7	77.3	78.3	79.4	77.3	78.3
	OBEP011240 OBEP010667	73.0	71.8	71.3	8 72.8 5 72.3	71.3	69.9 70.2	71.9	71.7	78.6 77.7	77.2 76.8	78.3 78.6	77.9 77.4	77.8	78.8 78.6	80.4 80.4	82.6	80.5	82.7 82.7	81.0 80.2	79.7 79.2	82.5 82.5	80.1 79.9	79.9 80.5	77.0	77.2	78.3 78.1	78.8 78.9		
	OBEP011480 KP814621	72.7	72.7	72.2		71.3	70.5	71.8	72.0	77.7	76.2	78.4	77.4	77.4	78.2	80.0 81.1	82.6 83.0	79.7 81.0	81.8 82.3	79.8	79.5 79.4	81.7	79.3	79.5	76.8	76.9	77.7	79.1	76.9	77.7
	OBEP011223- JX000783	73.2	72.3	71.6	73.0	71.6	70.5	72.4	72.1	77.8	77.0	78.7	77.5	77.7	78.6	80.6	82.8	80.4	82.6	80.7 80.4	79.5	82.4	80.1 79.6	80.1	77.1	76.8	78.3	78.8	76.9	78.3
	OBEP010745	72.8	71.8	71.2	72.6	71.0	70.1 69.8	71.0	71.5	77.6	77.1	77.6 78.1	77.0 77.6	77.0 77.4	78.6	80.3 80.2	82.5 82.2	80.2	82.2	80.6	79.2	82.0	79.7	79.7	76.8	76.9	77.9 78.1	78.5	76.4	78.1
	OBEP010143 OBEP011028	73.2	72.3	71.5			70.0	72.3	72.1	78.1 77.4	77.0	78.8 78.3	77.7	78.0	78.8 78.5	80.6 80.4	82.8	80.4	82.8 82.7	81.0 81.1	79.6	82.7 82.5	80.1 80.3	80.2 80.5	77.3	76.8	78.3 78.2	79.0 78.5	76.9	
	OBEP011478 OBEP011090	72.7	71.9	71.6	5 72.5 3 72.8	70.7	69.8 70.4	71.3	71.6	77.5 77.7	76.6 77.4	78.5 77.9	77.3 78.2	77.5	78.3 78.7	80.5 80.3	82.2 82.1	80.1	82.5 82.6	80.6 80.4	79.7 79.4	82.3 82.5	79.7 79.4	80.0 79.9	76.6	76.5 78.9	77.6 80.5	78.4 80.0	76.3	77.6
	FJ655658	73.3	73.2	72.0	72.9	71.8	70.6	71.9	72.1	77.7	77.2	77.9	77.6	78.0	78.4	80.3	82.2	79.5	82.8	80.7	79.7	82.7	79.0	80.3	79.1	79.3	80.1	79.9	78.9	80.1
	OBEP010911: GU553522	5 73.5 73.3	73.0 73.2	72.3	72.8	71.9	70.7	73.1	72.9	77.9 77.7	77.5 77.2	78.7 78.6	78.2 77.8	78.2	78.9 78.6	80.8 80.2	82.8 82.3	79.9 79.6	82.9 82.6	81.0 80.8	80.5 80.3	82.7 82.5	79.4 79.1	80.4 80.3	78.8 78.4	78.6	79.6 79.1	79.8 79.3	78.5	79.6 79.1
	JQ989539 AB800140	73.1	73.1	72.0			70.4	71.9	71.7	77.8	77.4	78.4	77.9	78.2	78.8 78.5	79.9 80.3	81.5 81.9	79.0	82.4 82.5	80.1 80.4	79.0 79.7	82.2 82.3	78.5 78.6	79.9 80.4	78.9	79.0	79.8 79.9	79.6 79.6		79.8
	AF412942 OBEP010634	72.6	72.4	71.6	72.1	70.8	70.1	71.5	71.7	76.7	76.5	77.6	76.9	77.1	77.8	80.3	81.9	79.1	82.5 82.6	80.4 80.4	79.7	82.3	78.6	79.7	78.5	78.5	79.6	79.4	78.3	79.6
	OBEP0113210	72.9	72.6	71.8	3 72.5	71.6	70.1	71.9	71.9	77.5	77.2	77.9	77.7	77.6	78.3	79.9	81.8	79.1	82.2	79.9	79.2	82.0	78.6	79.3	78.7	78.5	79.3	79.5	78.3	79.3
	JQ817971 JN123687	73.1			72.0		70.8	72.5	72.2	77.5	77.0	78.7	77.5	77.7	78.4	81.1 79.3	82.5 80.8	80.6	83.1 81.4	81.0 79.5	80.3 78.9	82.9 81.2	80.1	81.1 78.8	77.7	77.6	78.9 78.7	79.0 78.5		
	HM480259 AF412944	71.8	71.4	71.0	71.6	71.6	70.3	71.7	71.3	76.7 77.4	76.4	77.2 78.3	76.7	77.6	78.0 78.4	79.6 80.3	80.5 81.8	78.9	81.8 82.6	79.1 80.8	77.6	81.6 82.4	78.4	78.6	78.0 78.0	78.4 78.2	79.2 79.4	79.1 79.3	78.0	79.2
	JN123682	73.2	73.0	72.1	72.6	71.4	70.7	72.2	72.2	77.8	77.1	78.3	77.6	78.0	78.6	80.4	82.1	79.6	82.6	80.7	80.1	82.4	79.1	80.1	78.8	79.0	79.9	79.8	79.0	79.9
Lokiarchaeota-4	OBEP010618 OBEP011076	72.5	72.3	71.5	72.1	71.9	70.9	72.4	71.6	77.2	76.6	77.5	77.2	77.3	78.1 77.9	80.2	81.2 82.5	78.5	81.7 82.5	80.7	78.7 79.8		78.0 79.2	78.9 79.6	78.1	78.1	78.8 79.6	78.7 79.0		79.6
	JQ989549 JN123610	72.5	72.0	71.5			70.3	71.5	71.7	77.1 77.0	76.6	78.0 77.7	77.1	77.5	78.0 78.0	79.9 80.4	81.3 81.6	78.6	82.0 82.4	79.8 80.2	79.0 79.6	81.8 82.2	78.1 78.6	79.5 79.7	77.9	78.0	79.3 79.4	78.9 79.5		
	AB800489 OBEP010669	73.4	72.8	72.2	2 73.0	71.9	70.9	72.6	72.8	77.2	76.9	78.5 77.5	77.2	77.5	78.1	80.5 80.8	82.3	79.4	82.5 82.9	81.1 81.0	80.4 80.3	82.3	78.9	80.2 80.0	78.4	78.6	79.6 79.5	79.5	78.4	79.6
	OBEP011356	73.2	72.9	72.2	2 72.7	71.8	70.7	72.6	72.7	77.5	77.1	78.6	77.7	77.9	78.5	80.6	82.5	79.6	82.5	80.9	80.2	82.3	79.1	80.1	78.6	78.7	79.9	79.6	78.5	79.9
	AB800193 AM992695	73.2 69.0	72.0 69.1	68.2	68.5	71.9 67.6	70.8	68.5	72.2 69.2	77.5	77.0 72.3	78.7 73.6	77.5 72.5	77.7	78.4 73.7	81.2 75.9	82.6 77.0	80.7 75.1	83.2 78.2	81.1 75.1	80.4 75.0	83.0	80.2 74.7	81.2 75.9	77.8	77.7	79.0 75.5	79.1 75.8	78.0 74.8	75.5
	OBEP010636 AB799355	72.9	73.0	72.0		71.4	69.7	71.7	71.8	77.1 77.4	76.8 76.9	77.5 78.2	77.4	77.3	77.8 78.3	80.0 80.3	81.9 82.0	78.7	82.2 82.5	80.3 80.5	78.8 79.8	82.0 82.3	78.4 78.6	78.8 80.1	78.0 78.4	77.5	78.9 79.8	78.6	76.9	78.9
	GU553813	73.5	73.3	72.7	73.1	71.8	70.4	72.4	72.5	77.7	77.2	78.3	77.7	78.1	78.6	80.1	82.1	79.4	82.5	81.0	80.1	82.3	79.0	80.2	77.9	78.1	79.2	79.0	78.1	79.2
	DQ522906 OBEP0107794	73.6 73.1	73.4	72.4	73.2 72.6	71.9	70.9	72.8 72.3	73.0 72.4	77.5 77.7	77.0 76.8	78.7 78.3	77.5 77.5	77.7	78.4 78.4	80.5 80.1	82.3 82.1	79.5 79.6	82.5 82.7	81.1 80.9	80.4 80.2		79.0 79.1	80.2 80.5	78.5	78.7 78.5	79.7 79.3	79.6 79.0	78.3	79.7 79.3
	OBEP010777 OBEP011471	73.3	72.9	72.2		72.0	71.0	72.3	72.5	78.3 77.3	77.8	78.5 78.1	78.5 77.2	78.4	79.3	80.7 80.6	82.5 82.7	79.9	83.0 83.1	80.9 80.6	80.4 80.1	82.8	79.4 79.2	80.0 79.9	79.2	79.1	80.1 79.4	80.1 79.5	78.7	80.1
	OBEP010654	73.2	72.8	72.1	72.8	71.8	70.4	72.2	72.7	77.6	77.1	78.5	77.8	77.8	78.2	80.6	82.3	79.8	82.5	80.7	80.2	82.3	79.3	80.0	78.4	78.7	79.6	79.7	78.6	79.6
Odinarchaeota	KX764885 AB019719	72.5 71.6	71.8	72.3	8 72.4 8 71.4	71.8	73.0	71.8	72.2	80.0 80.0	79.0 78.8	80.7 80.6	78.6 78.6	78.8	79.3 79.5	81.3 80.8	81.4 80.9	81.4 80.9	81.1 80.5	80.4 79.2	79.8 78.6	80.9 80.3	81.1 80.4	81.0 80.9	79.9 80.1	78.2	80.6 80.7	80.0 80.4		
Thorarchaeota	OBEP011314	70.7	69.8	70.3	70.6	69.7	71.5	70.4	70.3	74.0	73.4	75.7	74.0	74.4	74.6	77.7	78.1	76.8	78.0	77.7	77.1	77.9	76.4	77.5	74.1	74.0	75.8	76.2		75.7

Calculation of percent identity of new Asgard groups was based on 179 sequences with a long fragment of Asgardarchaeotal 16S rRNA gene. Fragments of 16S rRNA gene from position of *E. coli* 243 to 1414 (~1170 bp) were used for calculating the identity. Novel clades are marked with colored background.

Sample	Combined samples for metagenomic assembly	Biotope	Sampling Location		cDNA Data size (Gbp)	Sampling time	latitude and longitude
				99.2	NA		
DZ	NA	Mangrove sediment	Xinying Bay, Danzhou, China (10-20 cm)	105.6	NA	Jan, 2018	19.7410 N, 109.2727 E
DZG	NA	Mangrove sediment	Dongzhai Bay, Haikou, China (10-20 cm)	107.6	NA	Jan, 2018	19.9516 N, 110.5764 E
LZ	NA	Mangrove sediment	Leizhou Bay, China (10-20cm)	112.1	NA	Jan, 2018	20.9262 N, 110.1628 E
Maipo_7		Mangrove sediment	Shenzhen Bay, Hong Kong, China (0-2 cm)	126.9	8.6	Sep, 2014	22.4979 N, 114.0295 E
Maipo_8		Mangrove sediment	Shenzhen Bay, Hong Kong, China (10-15 cm)	91.1	8.6	Sep, 2014	22.4979 N, 114.0295 E
Maipo_9	MP_5	Mangrove sediment	Shenzhen Bay, Hong Kong, China (20-25 cm)	88.6	8.9	Sep, 2014	22.4979 N, 114.0295 E
Maipo_10		Mudflat sediment	Shenzhen Bay, Hong Kong, China (0-5 cm)	98.1	9.5	Sep, 2014	22.4992 N, 114.0276 E
Maipo_11		Mudflat sediment	Shenzhen Bay, Hong Kong, China (13-16 cm)	86.0	12.2	Sep, 2014	22.4992 N, 114.0276 E
SZ_1	NA	Mangrove sediment	Shenzhen Bay, Shenzhen, China (0-2 cm)	105.6	12.4	Jan, 2018	22.5288 N, 114.0300 E
SZ_2	NA	Mangrove sediment	Shenzhen Bay, Shenzhen, China (6-8 cm)	121.1	12.4	Jan, 2018	22.5288 N, 114.0300 E
SZ_3	NA	Mangrove sediment	Shenzhen Bay, Shenzhen, China (12-14 cm)	90.6	11.7	Jan, 2018	22.5288 N, 114.0300 E
SZ_4	NA	Mangrove sediment	Shenzhen Bay, Shenzhen, China (28-30 cm)	96.6	NA	Jan, 2018	22.5288 N, 114.0300 E
XMD	NA	Mangrove sediment	Ximen Island, Leqing, China (10-20 cm)	114.3	NA	Jan, 2018	28.3484 N, 121.1782 E
YX	NA	Mangrove sediment	Zhangjiang Estuary, Yunxiao, China (10-20 cm)	96.9	NA	Jan, 2018	25.9179 N, 117.4215 E
C1-1		seagrass sediment	Rongcheng Swan Lake (0-4cm)	119.5	NA	Nov, 2018	37.3502 N, 122.5781 E
C1-7		seagrass sediment	Rongcheng Swan Lake (21-26cm)	105.4	NA	Nov, 2018	37.3502 N, 122.5781 E
C1-10	ΥT	seagrass sediment	Rongcheng Swan Lake (36-41cm)	101.1	NA	Nov, 2018	37.3502 N, 122.5781 E
FC1-1	11	mudflat sediment	Rongcheng Swan Lake (0-4cm)	150.9	NA	Nov, 2018	37.3464 N, 122.5700 E
FC1-7	FC1-7 mudflat sedimen		Rongcheng Swan Lake (21-26cm)	123.7	NA	Nov, 2018	37.3464 N, 122.5700 E
FC1-10		mudflat sediment	Rongcheng Swan Lake (36-41cm)	96.7	NA	Nov, 2018	37.3464 N, 122.5700 E

Supplementary Table 4. Information of sediment samples analyzed in this study.

Table S5. Overview of Asgard archaea genomic bins.

Metage	nomic bins	Bin size (Mbp)	Number of scaffolds	Largest scaffold (Mbp)	N50 (Kbp)	Number of ORFs	Completeness (%) ¹	Contamination (%) ¹	Strain heterogeneity (%) [*]	GC (%)	rRNA [#]
	MP5_1_bin791	4.8	441	0.08	20.8	4,598	90.65	18.69	0.00	31.7	NA
	MP5_2_bin1192	3.2	580	0.03	32.5	3,394	88.47	8.72	23.08	30.1	NA
	SZ_4_bin5.60	3.4	68	0.37	69.1	3,206	93.46	8.88	0.00	30.5	NA
Gerdarchaeota	YT_bin1.045	3.7	240	0.10	25.2	2,427	83.29	8.88	0.00	31.1	NA
	YT_bin2.057	4.3	223	0.15	31.6	4,042	94.86	9.81	0.00	32.2	SSU+LSU
	YT_bin5.010	2.6	599	0.02	4.9	2,841	75.24	9.63	23.53	34.1	NA
	B18_G1	2.9	261	0.06	18.4	2,921	86.45	7.94	0.00	38.9	LSU
	SZ_4_bin10.384	4.4	54	0.48	215.6	4,343	96.26	4.21	0.00	30.1	LSU
Helarchaeota	Hel_GB_A^	3.8	333	0.05	15.2	3,574	82.40	2.80	0.00	35.3	NA
	Hel_{GB_B}	3.5	180	0.07	28.9	3,164	86.92	1.40	0.00	28.0	NA
	DNA_SIP_bin2.159	3.4	362	0.04	12.5	3,716	75.23	6.22	41.18	29.3	SSU
	DZG_bin1.240	3.9	112	0.21	66.1	3,706	87.85	3.74	0.00	33.0	SSU
Lokiarchaeota	SZ_4_bin8.338	3.7	198	0.11	33.5	3,596	86.45	2.34	0.00	30.6	SSU+LSU
Lokiarchaeota	CR4	4.3	1,904	0.09	4.1	4,413	80.45	2.34	0.00	44.0	SSU+LSU
	GC14_75	5.1	504	0.07	15.4	5,384	90.29	45.15	78.21	31.1	SSU+LSU
	SM_031	4.8	728	0.05	8.7	5,404	85.21	5.14	0.00	30.1	LSU
	DZG_bin1.115	3.1	95	0.19	53.2	2,956	82.78	3.27	0.00	45.0	SSU
	SZ_4_bin3.344	1.7	22	0.47	222.1	1,674	83.57	2.34	0.00	45.5	SSU+LSU
	SZ_4_bin10.233	3.5	98	0.27	71.9	3,344	83.18	2.8	0.00	47.2	SSU+LSU
Thorarchaeota	AB_25	2.3	264	0.09	15.4	2,914	90.12	3.19	33.33	44.6	SSU+LSU
	B41_G1	3.0	559	0.09	7.0	3,154	94.86	8.88	0.00	51.9	SSU+LSU
	B59_G1	4.3	523	0.20	34.0	4,344	95.33	9.42	5.26	48.9	SSU+LSU
	MP8T	3.5	148	0.24	51.5	3,612	92.06	5.30	11.11	41.6	SSU+LSU

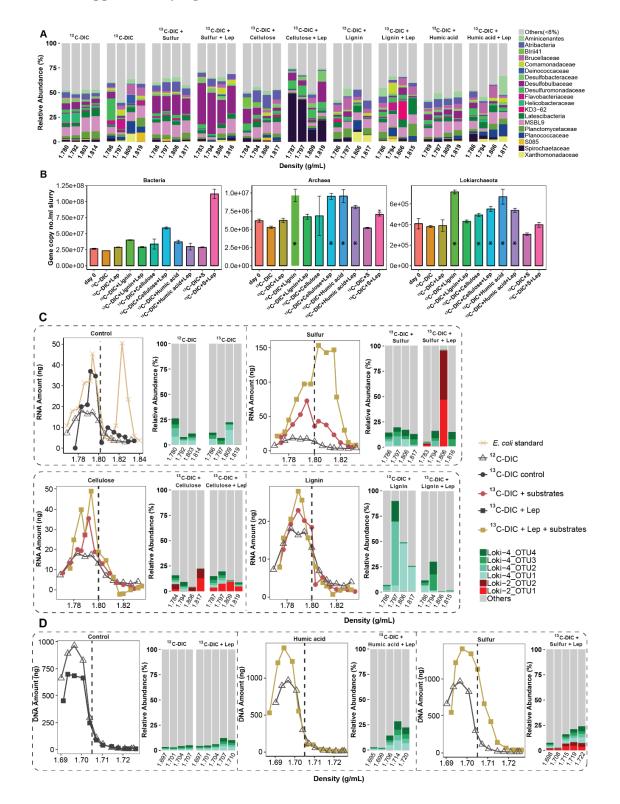
121

	MP9T	4.4	390	0.11	12.5	4,513	85.05	1.01	0.00	41.5	NA
	MP11T	3.5	235	0.12	18	3,693	90.19	2.34	0.00	42.6	SSU+LSU
	SMTZ1_45	3.5	119	0.18	51.2	3,208	87.69	5.14	0.00	42.2	SSU+LSU
	SMTZ1_83	3.3	266	0.07	15.3	3,029	90.19	6.54	14.29	49.3	SSU+LSU
	SM_104	4.3	243	0.10	35.6	4,337	91.30	5.14	14.29	39.5	LSU
	SM_117	4.0	320	0.14	28.2	4,363	86.45	6.83	14.29	45.3	NA
Odinarchaeota	LCB_4	1.5	9	1.18	1181.4	1,584	96.3	1.4		38.1	SSU+LSU
	AB_125	2.3	225	0.04	10.8	2,348	79.37	2.34	0.00	33.4	SSU+LSU
Heimdallarchaeota- AAG	GCA_002728275.1	1.4	76	0.07	26.4	1,386	83.18	1.87	50	30.1	SSU+LSU
	LC_2	4.8	177	0.15	38.7	4,586	72.43	4.21	40	32.6	LSU
Heimdallarchaeota-	SZ_4_bin2.246	3.0	16	0.65	525.5	2,686	87.38	1.94	0.00	34.9	NA
MHVG	LC_3	5.7	157	0.22	59.3	5,514	91.59	5.61	12.5	29.7	SSU+LSU

* Completeness and contamination were assessed using lineage-specific maker sets with checkM. # rRNA genes were predicted using Barrnap (version 0.9, <u>https://github.com/tseemann/barrnap</u>) with the parameter "1e-5".

[^] These MAGs were from Guaymas Basin hydrothermal vent sediments.

Asgard MAGs with completeness > 70% were chosen for analyses and those retrieved in this study are marked red.



4.9.Supplementary figures

Supplementary Fig. 1. (a) Total sum scaling of abundances of bacterial 16S rRNA gene sequences from selected "light" and "heavy" gradient fractions of RNA-SIP samples. Before Illumina sequencing, 2 fractions (fraction 4 and 5, 6 and 7, 8 and 9, 11 and 10) were combined together as one sample for library preparation. Density is indicated by the average density of combined fractions for RNA-SIP samples. Relative abundances are shown at the family level as taxonomic threshold. (b) 16S rRNA gene copy numbers of Lokiarchaeota of DNA extracts from SIP incubations on day 0 and 255. * indicates significant increase of gene copies compared to control incubations i.e., "day 0" and "¹³C-DIC" (c) RNA-SIP profiles and relative abundances of Lokiarchaeota 16S rRNA gene sequences. (d) DNA-SIP profiles and relative abundances of Lokiarchaeota 16S rRNA gene sequences. DIC: dissolved inorganic carbon (i.e. bicarbonate); S: sulfur; Lep: lepidocrocite. Dashed lines indicate starting density before ultracentrifugation.

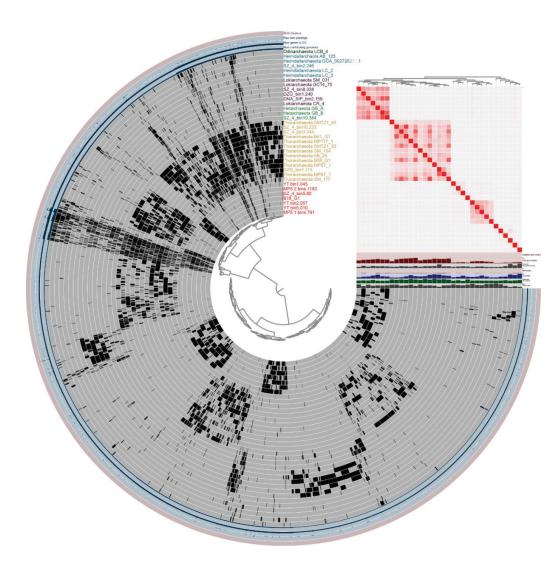


Fig. S2. Pan-genome analysis of protein clusters within all Asgard MAGs and their amino acid identity (ANI) using the Anvi'o software. The inner and upper trees were clustered using "Gene cluster presence

absence". Asgard MAGs with red stars are from this study. Those with MCR complex or from ¹³C-labelled enrichment were highlighted with two red stars. The ANI ranges from 60% to 100%, with deeper colour represent higher identity.

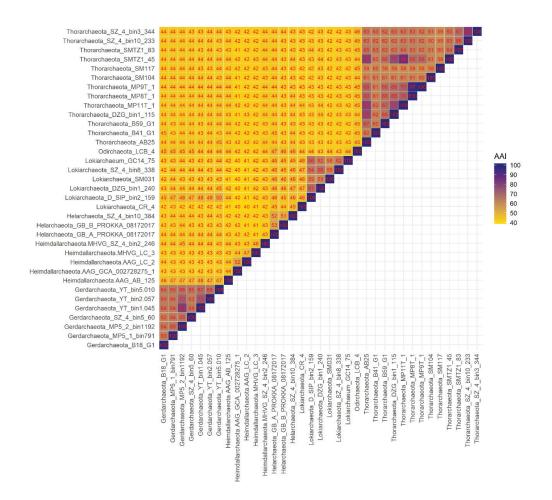


Fig. S3. Amino acid identity (AAI) of Asgard MAGs. AAI was calculated using CompareM (https://github.com/dparks1134/CompareM).

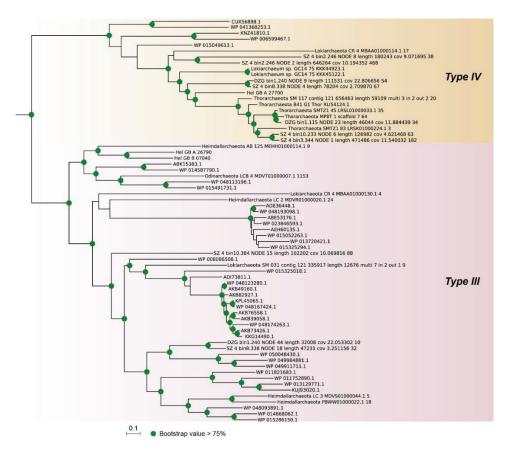


Fig. S4. Phylogenetic maximum likelihood tree of RuBisCO amino acid sequences (the large subunit). The tree was build using IQ-TREE with model LG+I+G4 and parameter "-bb 1000".

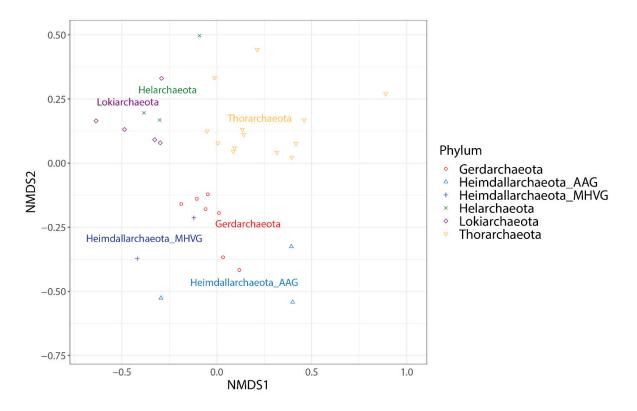


Fig. S5. Non-metric multidimensional scaling (NMDS) plot of KEGG in all available Asgard MAGs (December 2018 updated). Odinarchaeota was excluded because it has one MAG and it is not representative. KEGG data was analyzed by Bray–Curtis dissimilarity.

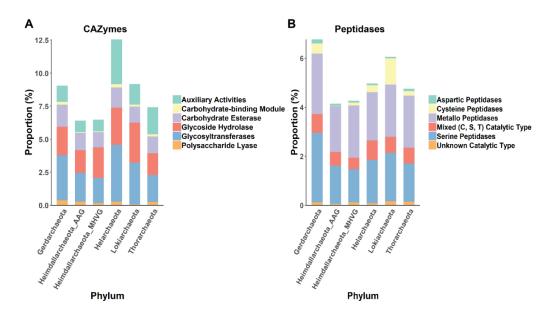


Fig. S6. Abundance of (A) carbohydrate-active enzymes and (B) peptidases in Asgard phyla. Carbohydrate-active enzymes (CAZymes) and peptidases were annotated using the dbCAN webserver and MEROPs database, respectively. The e-value cutoff is 1e-5 for both cases. Proportion was calculated by normalizing the MAGs number and the average protein numbers in each phylum. Detailed information is available in Table S7.

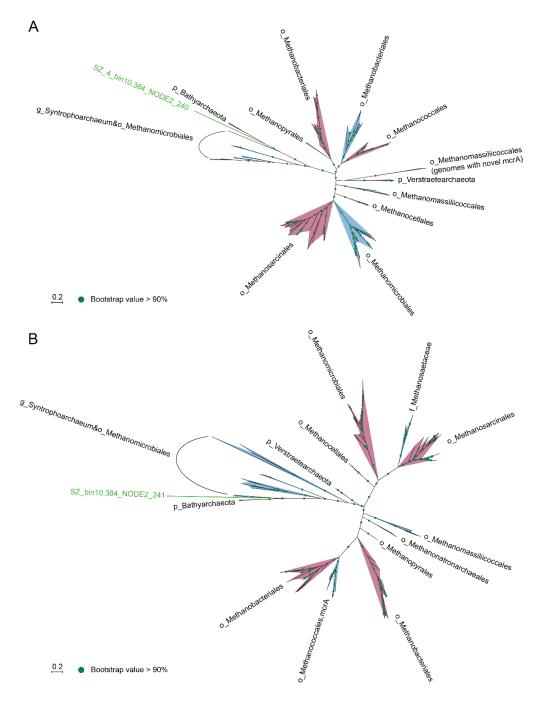


Fig. S7. Protein trees of (A) McrB and (B) McrG. Asgard archaea McrA gene obtained in this study is marked with green line. The maximum-likelihood trees were inferred using IQ-TREE tree with model LG+F+I+G4 and parameter "-bb 1000". The background-colour is added to differentiate adjacent groups.

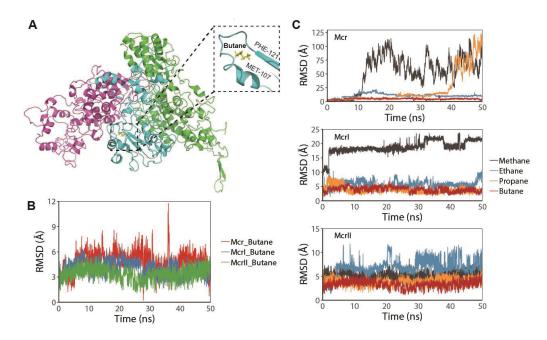


Fig. S8. (A) Equilibrium structure of the Lokiarchaeotal MCR complex and the docking model of butane (yellow). McrA, McrB and McrG are marked in green, magenta, and cyan, respectively. Magnified structure of butane binding with the pocket of McrG shown as inset. The main residues interacting with butane are from MET107 to PHE121 of McrG. (B) The RMSD of butane in Mcr (red), McrI (blue) and McrII (green) complexes as a function of time. The RMSD of butane in the three complexes are of the same magnitude (~5Å). (C) The RMSD of methane (black), ethane (blue), propane (yellow) and methane (red) in Mcr (top), McrI (middle) and McrII (bottom) complexes as a function of time. The RMSD of butane in all the three complexes are of smaller magnitude than the ones of methane, which implies that the binding affinities of butane to all the three Mcr complexes are higher than the ones of methane. Here, Mcr represents the Lokiarchaeotal MCR complex, while McrI and McrII represent the complexes from the *Ca*. Syntrophoarchaeum butanivorans. The RMSD is estimated as the distance deviation of methane/butane to the center of mass of each Mcr complex.

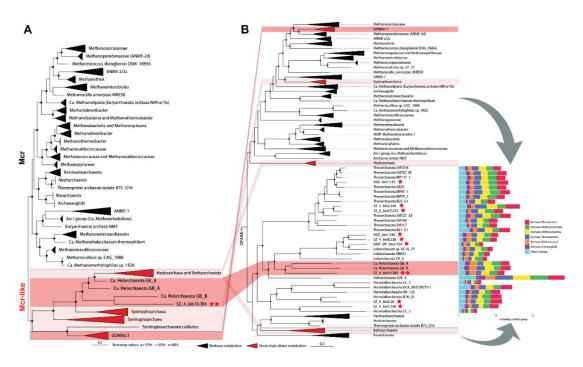


Fig. S9. Phylogenetic position and evolution of Asgard archaea *mcrA* genes. (A) Maximum likelihood phylogenetic tree of the *mcrA* genes. (B) Ancestral genome content reconstruction with the concatenated 122 archaeal marker gene tree inferred from IQ-TREE. Bar plot of the percentage of genes horizontally transferred from archaea to their genomes. The genes and MAGs obtained in this article are marked with red stars, and those with the MCR complex are highlighted with two red stars. The two grey arrows indicate genes that underwent extensive horizontal transfer from archaea.

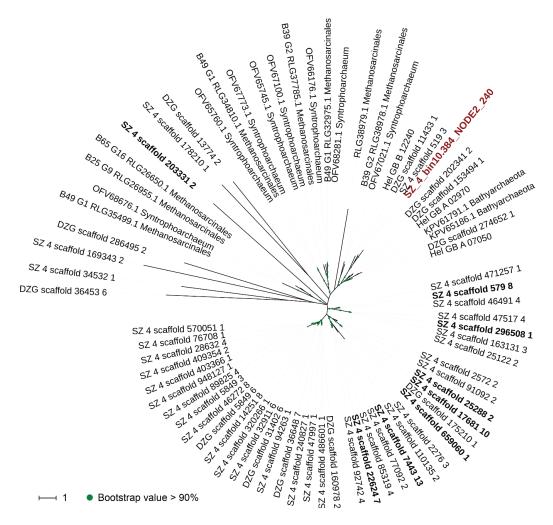


Fig. S10. Protein tree based on *mcrA* gene sequences as identified in the scaffolds. The protein sequences were obtained through BLASTP against novel McrA protein sequences of Asgard with E-value cutoff \leq 1e-5. Amino acids > 400 aa were kept for tree building. The tree was re-rooted with Euryarchaeotal and Verstraetearchaeotal McrA protein sequences. McrA sequences deduced from transcripts are marked in bold.

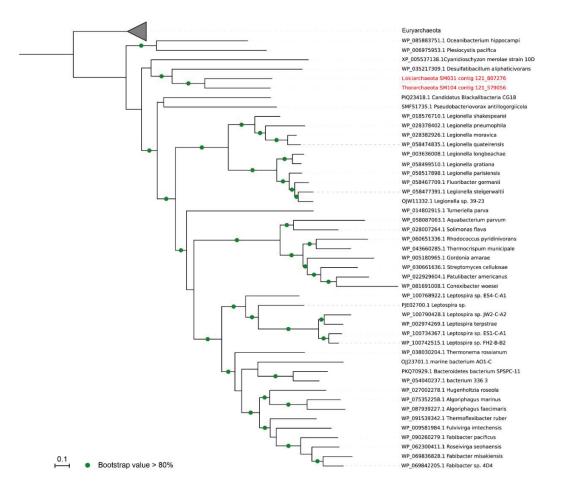


Fig. S11. Protein tree of *alkB* genes. The reference sequences were obtained through BLASTP search with NCBI nr protein database and clustered at 90% sequence identity with Usearch.

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Chapter 5

General Discussion

The main goal of the thesis was to examine the carbon metabolism in methylotrophic methanogens and Asgard archaea, especially for inorganic carbon assimilation. Based on nucleic acid-SIP, we have clearly shown that ¹³C-DIC is required as the co-substrate for identifying archaea (methylotrophic methanogens and Lokiarchaeota-4) and detecting their activities in marine sediments, which indicates that inorganic carbon is the main carbon source for nucleic acid synthesis in these archaea.

In methylotrophic methanogens (Chapter 2 and 3), the low methanogenesis rate is associated with methane production from CO_2 . This mixotrophic methanogenesis from both methanol and CO_2 in utilizing both substrates kinetically alters inorganic carbon assimilation into biomass, which is indicated by lipid-SIP. Thus, in marine sediments, the low methane production rate *in situ* most likely alters carbon utilization patterns in obligate methylotrophic methanogens. Presently, it is unknown whether methanogens can benefit from the process or whether it is only due to reaction kinetics. However, it is clear that carbon metabolism of methylotrophic methanogens is more divergent *in situ* than that in pure cultures under optimized conditions, and the large pool of inorganic carbon in sediments could alter carbon incorporation patterns of archaea.

Similarly, the ¹³C-DIC-dependent nucleic acid-SIP is also feasible for identifying uncultivated Asgard archaea (Lokiarchaeota) (Chapter 4). By amending organic polymers, i.e., lignin, humic acid and cellulose, activities of Lokiarchaeota were successfully detected in SIP incubations. In combination with metagenomic sequencing, we have shown that Lokiarchaeota might degrade the polymeric sugar part of these organic compounds via Embden–Meyerhof–Parnas pathway or potentially use the products of polymer breakdown (benzoate) as carbon or energy source. Asgard archaea harbor a more complex carbon metabolism than methanogens, which is indicated by the observation of oxidation pathways of both, short- and medium-chain alkanes. The finding of methyl-coenzyme M reductase genes in Asgard archaea widely extends the phylogenetic distribution of microorganisms involved in short-chain alkane oxidation way beyond methane as a substrate. Taken together, the new findings about activities and potential capabilities of Asgard archaea suggest multiple carbon utilization potentials and versatile activities in alkane oxidation, potentially linked to environmental adaption.

Both methylotrophic methanogens and Asgard archaea have genes encoding methyl-CoM reductase (MCR) for alkane metabolisms. Because of the reversible reactions catalyzed by relevant enzymes

including the Wood-Ljungdahl pathway and alkane metabolisms (Friedrich 2005; Hallam et al., 2004; Scheller et al., 2010), the presence of these relevant enzymes in archaea harbor versatile activities. In the case of methylotrophic methanogenesis, the reverse reactions of CO_2 to methyl-H₄SPT and further to methane suggest CO_2 conversion to methane in obligate methylotrophic methanogens, which will alter carbon incorporation patterns as methyl-H₄SPT can be used for biomass synthesis. Initially, MCR catalyzes the reaction of methane formation. In this process, methanogens can obtain decent amounts of energy for cell growth. However, the extension of *mcr* genes to anaerobic methanotrophs (ANMEs) indicates the reversibility of the methanogenesis (Hallam et al., 2004; Scheller et al., 2010). Furthermore, findings about alkane oxidation via the catalysis of MCR expanded the function of the MCR cluster in archaea (Evans et al., 2019). Therefore, alkane metabolism in methylotrophic methanogens and Asgard archaea offers a glimpse of the versatile abilities of archaea regarding to the survival in a variety of environments. It is likely that archaea can use many reversible activities to adapt to adverse environmental conditions.

Carbon fixation into organic carbon is the determining factor for existence of biosphere on earth. Although the origin of life is unclear, the lack of organic carbon and oxygen on the primitive earth (Lyons et al., 2014; Zahnle et al., 2010) indicates the importance of anaerobic inorganic carbon utilization. Phylogenetic analysis of prokaryotic genes encoding protein depicts the physiology of CO_2 fixation via the Wood–Ljungdahl pathway and methyl compound metabolisms in the last universal common ancestor (LUCA) (Weiss et al., 2016). Thus, the utilization of C1 substrates (methanol) and organic polymers as well as the activity of inorganic carbon assimilation by methylotrophic methanogens and Asgard archaea is potentially associated with the evolution of life.

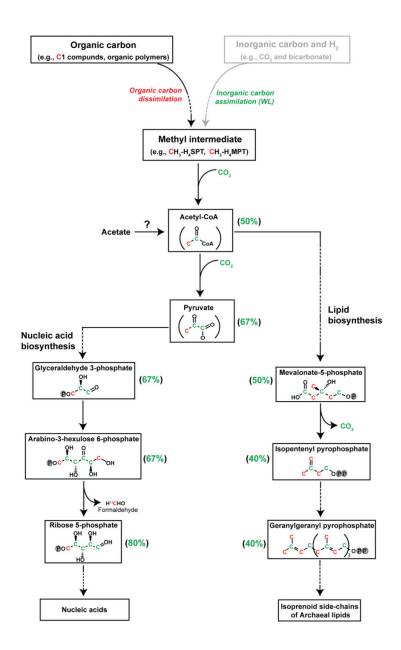


Fig. 1 Summary of the organic carbon dissimilation and carbon assimilation into nucleic acid and lipid biosynthesis in archaea from anoxic marine sediments

WL: Wood–Ljungdahl pathway; CH_3 - H_4SPT , methyl-tetrahydrosarcinapterin; CH_3 - H_4MPT , methyltetrahydromethanopterin. Square with black outline indicates utilization of organic substrates and carbon fixation in archaea. A combination of different reaction steps is indicated by dashed line. Grey color shows autotrophic CO_2 fixation via the reductive WL pathway, potentially by hydrogenotrophic methanogens and Asgard archaea. Red represents carbon assimilation of organic compounds into archaea. Green shows inorganic carbon assimilation and contribution of inorganic carbon to intermediates. Question marker denotes the potential acetate incorporation for biomass.

5.1. Carbon metabolism in methylotrophic methanogens and Asgard archaea

5.1.1. Methyl metabolism

Methylated compounds are the reduced carbon sources and are involved in carbon and energy metabolisms in archaea (Weiss et al., 2016). In the presence of C1 compounds or organic polymers, archaea preferentially use these organic substrates to generate methyl groups, which are the key intermediate of carbon source for biosynthesis (Fig. 1). For example, methanol conversion to CH₃-H₄SPT and methane in methylotrophic methanogens provides energy and methyl intermediate for cell growth. In contrast, Asgard archaea harbor more complex methyl sources. With organic polymers (lignin, humic acid and cellulose), the methyl intermediate originates from organic carbon dissimilation (Fig. 1), which might be involved in utilization of intermediates (fatty acids and benzoate) during the degradation of these organic polymers as shown in Chapter 4 (Fig. 2 and 3). Microorganisms would uptake organic carbon for producing methyl intermediate if methylated compounds or organic carbon are available, since CO₂ reduction to CH₃-H₄SPT is endergonic (Berg et al., 2010). On the contrary, dissimilation of methanol or organic compounds to methyl group and further to CO₂ provides energy for cell activities (Kleerebezem et al., 1999; Thauer et al., 2008; Watkins et al., 2014). Therefore, organic carbon is necessary for triggering the activity of archaea. For example, the standard free energy for methylotrophic methanogenesis from methanol is -105 kJ/mol CH₄ (Liu and Whitman 2008), which can provide sufficient energy for cell growth. For Asgard archaea, participating in degradation of organic polymers such as lignin, humic acids and cellulose or their products of polymer breakdown also provides energy for Asgard activities. As adaption strategy, Asgard archaea might harbor a low growth rate and participate in the degradation of complex compounds existing in sediments. The environmental filtering under the intricate conditions with low energy and carbon availability might lead to the characteristics of these archaea such as long doubling time and the difficulty of enrichment and isolation from sediments.

Marine sediment is a large pool of organic compounds originating from terrigenous sources and water column (Burdige 2005; Schlünz and Schneider 2000; Schubert and Calvert 2001). The continuous precipitation and sedimentation provide organic carbon for microbial growth. Archaea with a slow activity and long doubling time (e.g., Asgard archaea, Bathyarchaeota and ANMEs) might have a similar lifestyle considering the limitation of organic carbon utilization. With a long-time scale, these archaea might have a high abundance in marine sediments (Danovaro et al., 2016; Hoshino and Inagaki 2019), as shown that archaea and bacteria even have a similar abundance in marine sediments (Lloyd et al., 2013a; Schippers et al., 2012).

For carbon assimilation, the methyl intermediate generated from methyl substrates and organic polymers in methylotrophic methanogens and Asgard archaea, respectively, while CO_2 is utilized for the synthesis of carboxyl groups of acetyl-CoA in the presence of methyl intermediate (Fig.2; Fig. 5 and 6 in Chapter 3). The methyl group will be further delivered to acetyl-CoA for biosynthesis of lipid and nucleic acids. The scenario is like that C1 substrates such as methanol and methylamines are the small-molecule compounds without carbon–carbon bond, lacking the diversity of functional groups, which result in inorganic carbon assimilation to replenish carbon source for cell growth. In addition, because of carbon availability in organic polymers, inorganic carbon incorporation might be important for carbon utilization by archaea.

5.1.2. Inorganic carbon assimilation

Inorganic carbon assimilation by microorganisms is ubiquitous in marine sediments, including autotrophic, mixotrophic and heterotrophic carbon fixation (Jones et al., 2016; Sakai et al., 2011; Santruckova et al., 2005). As chemolithoautotrophs, microorganisms affiliated to sulfur-oxidizing bacteria (*Gammaproteobacteria*) fix inorganic carbon in dark sediments under energy supply from sulfur oxidation (Dyksma et al., 2016). For archaea, hydrogenotrophic methanogens in deeper marine sediments also perform dark carbon fixation with reduction of CO_2 to biomass by H_2 (Sakai et al., 2011). Mixotrophs are defined as having the similar contribution of carbon source from organic and inorganic carbon, such as methylotrophic bacteria with serine cycle, some acetogens (*Clostridium ljungdahlii*) and methylotrophic methanogens (Chistoserdova et al., 2009; De Marco 2004; Weimer and Zeikus 1978). Heterotrophs can fix CO_2 via anaplerotic reactions of the citric acid cycle and fermentation pathway (Krebs 1941; Santruckova et al., 2005).

Therefore, inorganic carbon utilization by microorganisms is intricate, especially for archaea in which the carbon source and activity are unclear. However, in consistence with the typical physiology of the LUCA, Euryarchaeota, Bathyarchaeota and Asgard archaea harbor the Wood-Ljungdahl pathway involving in CO₂ fixation (Weiss et al., 2016), which is followed by the reaction to pyruvate from acetyl-CoA and CO₂ (Santiago-Martinez et al., 2016). Methyl group of acetyl-CoA might be formed from organic substrate, i.e., methyl group of methanol by methylotrophic methanogens (Chapter 2) and fatty acids or alkane by Asgard archaea (Chapter 4). However, for the reactions such as acetyl-CoA and pyruvate synthesis, CO₂ incorporation is inevitable in archaea, indicating inorganic carbon is crucial for biomass synthesis. In marine sediments, the presence of organic substrates such as methyl compounds and recalcitrant

compounds might stimulate the activities of these archaea but the organic carbon availability is low. In this case, CO_2 fixation is the carbon replenishment of the limited organic carbon for archaeal growth.

5.2. Carbon incorporation in other archaea

Our study about carbon metabolism in methylotrophic methanogens and Asgard archaea provides insight into the potentials of carbon utilization pattern in archaea, which allows us to have a further discussion about carbon metabolism in other archaea. In Helgoland Mud area, archaea placed within the phyla of Thaumarchaeota, Euryarchaeota, Bathyarchaeota, Lokiarchaeota and Odinarchaeota were detected (Chapter 1). However, knowledge about carbon incorporation by the other archaea found in Helgoland Mud sediments is still limited to date, including Bathyarchaeota, anaerobic methanotrophs (ANMEs), marine benthic group D and DHVEG-1 (MBG-D and DHVEG-1), *Methanomassiliicoccales* and SG8-5.

5.2.1. Carbon metabolism in Bathyarchaeota

Bathyarchaeota is a ubiquitous phylum which is found in a variety of anaerobic marine sediments (He et al., 2016; Kubo et al., 2012; Zhou et al., 2018b). Except for ungrouped Bathyarchaeota, this phylum is classified into 17 subgroups (Kubo et al., 2012), indicating a high diversity in sediments. However, the activity of this archaea phylum has not been known before, only long-term incubations suggest that members of the subgroup Bathy-8 participate in lignin degradation and CO₂ fixation (Yu et al., 2018). Although two genes involved in the degradation of aromatic compounds are detected including catalase peroxidase and 4-oxalocrotonate tautomerase (Meng et al., 2014; Yu et al., 2018), carbon assimilation into biomass is unclear. Arguments such as incorporation of methoxy group and CO₂ into acetyl-CoA and further to biomass indicate that Bathy-8 is an autotroph (Yu et al., 2018) although the relevant gene was not shown for methoxy metabolism. However, it is not feasible to conclude that Bathy-8 are "organoautotrophs" since the methyl or methoxy group from lignin are apparently metabolized. Chapter 3 and Fig. 1 indicate that the methyl group will be assimilated into acetyl-CoA and further to biomass, suggesting a mixotrophic activity of these Bathyarchaeota in utilizing both organic carbon and inorganic carbon source for biomass synthesis. For archaea in anoxic sediments, the methyl group, as a reduced carbon source, and CO₂ as oxidized carbon source are necessary for synthesis of acetyl-CoA. Via the Wood-Ljungdahl pathway with lowest energetic cost (Berg et al., 2010), archaea most likely use both organic and inorganic carbon to produce acetyl-CoA for biomass, because only utilizing CO₂ would require more reducing equivalents (H₂ in the case of hydrogenotrophic methanogens) and energy. Therefore, similar to methylotrophic methanogens and Lokiarchaeota, Bathyarchaeota very likely harbor a mixotrophic carbon utilization pattern rather than solely autotrophic carbon fixation.

5.2.2. Carbon metabolism in anaerobic methanotrophs (ANMEs)

In ANMEs, methane generated from deeper marine sediments can be converted into CO_2 in presence of electron acceptors such as sulfate, iron oxides and nitrate (Fig. 2) (Yan et al., 2018). In the sulfate methane transition zone (SMTZ) of marine sediments, the syntrophic consortia between ANMEs and sulfate reducers is formed, in which ANMEs deliver electrons generated during anaerobic methane oxidation (AOM) to microorganisms for sulfate reduction (Hinrichs et al., 1999; Knittel and Boetius 2009). The reactions of anaerobic methane oxidation are as followed:

 $CH_4 + SO_4^{2+} \rightarrow HCO_3^{-} + HS^{-} + H_2O$ $\Delta G^{\circ} = -16.6 \text{ KJ/mol } CH_4 \text{ (Valentine and Reeburgh 2000)}$

Although energy formation from AOM coupled to sulfate reduction is feasible, carbon incorporation into biomass in ANMEs is not well estimated. Incubation-dependent AOM indicates that methane oxidation by ANME-1 strongly accelerates CO₂ incorporation into archaeal lipids (Kellermann et al., 2012). Lipid analysis with dual labeling strategy of ¹³C-DIC and D suggests that ANME-1 is "chemoorganoautotrophic" according to the ratio of inorganic carbon assimilation to lipid production (Kellermann et al., 2012). However, methane oxidation via the catalysis of methyl-CoM reductase (MCR) to methyl intermediate (CH₃-H₄MPT) will be used for acetyl-CoA synthesis and then for biomass (Fig. 2) (Meyerdierks et al., 2010), indicating a mixotrophic growth during AOM, which is supported by previous studies (Nauhaus et al., 2007; Weber et al., 2017; Wegener et al., 2008). In addition, it is unclear whether the dual labeling technique is feasible to assess carbon assimilation patterns in ANME-1, since the calculation of produced lipid is based on deuterium assimilation from D₂O, while hydrogen incorporation into biomass (lipid) from methane is complicated and unclear because four hydrogen atoms are available in methane. Therefore, it is very likely that ANMEs, as mixotrophs, harbor the similar carbon utilization patterns to methylotrophic methanogens as they share the similar methyl incorporation.

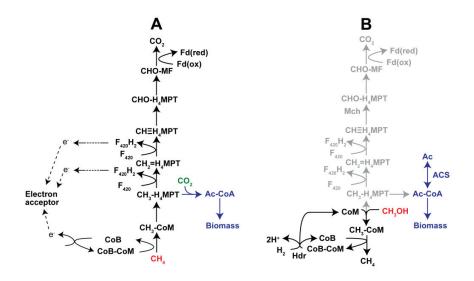


Fig. 2 Pathway of anaerobic methane oxidation (AOM) in ANMEs (A) and methanogenesis in *Methanomassiliicoccales* spp (B). Electron acceptor in Fig. A indicates that the reduction of potential electron acceptors such as sulfate, iron oxides and nitrate coupled to AOM. Ac-CoA: acetyl-CoA; Ac: acetate; ACS: acetyl-CoA synthase. Grey represents the methanogenic pathway absent in *Methanomassiliicoccales*. Pathway construction is based on the KEGG metabolic pathways and the previous studies (Borrel et al., 2013; Borrel et al., 2014; Cai et al., 2018; Chistoserdova et al., 2005; Havelsrud et al., 2011; Lang et al., 2015; Li et al., 2016; Meyerdierks et al., 2010).

5.2.3. Carbon metabolism in Thermoplasmata and potential acetate metabolism in

archaea

Thermoplasmata including *Thermoprofundales* (i.e., former marine benthic group D; MBG-D), SG8-5 (RC-III) and *Methanomassiliicoccales* are found in Helgoland Mud sediments. Metagenomic analysis suggests that *Thermoprofundales* and SG8-5 are anaerobic protein degraders (Lazar et al., 2017; Lloyd et al., 2013b; Zhou et al., 2018a). *Thermoprofundales* might fix CO₂ via the Wood–Ljungdahl pathway, which is identified in metagenome-assembled genomes (Zhou et al., 2018a).

As the seventh order of methanogens, *Methanomassiliicoccales* have been identified in human feces, rumen, termite gut and waste treatment sludge (Mihajlovski et al., 2008; Paul et al., 2012; Tajima 2001; Wright et al., 2004). *Methanomassiliicoccales* are able to produce methane by using H₂ and methyl substrates such as methanol, methylamins, methanethiol and dimethyl sulfide (Borrel et al., 2014). The reaction of methanol dependent methanogenesis is followed:

$CH_3OH + H_2 \rightarrow CH_4 + H_2O$ $\Delta G^{\circ'} = -113 \text{ kJ/mol } CH_4 \text{ (Liu and Whitman 2008)}$

Given that H₂ is directly used to reduce the methyl group and the pathway of oxidation of methyl group to CO₂ is absent, methyl intermediate (CH₃-H₄MPT) cannot be obtained during methane formation in *Methanomassiliicoccales* (Fig. 2), indicating that methane metabolism in these archaea is only involved in energy generation rather than carbon assimilation. Metagenomic analysis and the culture medium further suggest an additional carbon source, i.e., acetate, is necessary for cell carbon incorporation into biomass. This observation suggests *Methanomassiliicoccales* were not detected by RNA-SIP in methanol incubations without H₂ (Chapter 2 and 3), although evidences show that *Methanomassiliicoccales*-like group is detected in marine sediments (Zhou et al., 2015). Hence, unlike other obligate methylotrophic methanogens (*Methanococcoides* spp.), *Methanomassiliicoccales* likely to use acetate as carbon source, which is incorporated into acetyl-CoA via the catalysis of acetyl-CoA synthase (Fig. 2).

Apart from *Methanomassiliicoccales*, the functional genes involved in acetate assimilation or generation are detected in most archaea (Table 1 in Chapter 1), indicating acetate metabolism potentially plays an important role in carbon utilization in archaea. Previous studies from the perspective of energy metabolism suggest that archaea should produce acetate to obtain ATP instead of acetate assimilation (He et al., 2016; Lazar et al., 2017; Liu et al., 2018; Spang et al., 2019; Zhou et al., 2018a). For example, ANME-1 might perform methanogenesis with acetate or grow on acetate (Beulig et al., 2018; Jagersma et al., 2012). Based on the capability of acetate metabolism in Methanomassiliicoccales and ANME-1, our studies propose that except for CO₂ as the replenishment carbon source for archaeal growth when carbon availability is low, acetate might be the second crucial carbon source as the acetate metabolism genes are ubiquitous in archaea. Without participation of methanogens or additional electron acceptors, acetate degradation is intricate since acetate is the primary terminal product of organic decomposition in anaerobic conditions (Duddleston et al., 2002). In addition, acetate as the key intermediate for organic carbon degradation, is detectable (up to ~20 µM) (Beulig et al., 2018; King 1991; Zhuang et al., 2018) and does not directly link to methanogenesis in marine sediments (Beulig et al., 2018). Therefore, SIP technique should be applied based on incubations with ¹³C-labeled acetate as the co-substrate to target uncultured archaea in marine sediments.

5.3. Nucleic acid-SIP for detecting archaea-high sensitivity meets low activity

Nucleic acid-SIP based on ¹³C-labeled carbon incorporation is applied to detect the activity of microorganisms in a variety of environments. During nucleic acid-SIP, specific carbon source should be used to stimulate microorganisms in incubation setups, but for uncultured microorganisms with unknown

carbon source or complex carbon utilization, nucleic acid-SIP is not able to detect them. Cross-feeding of label might occur if a high molecular carbon source is supplied. Since next generation sequencing after nucleic acid-SIP provides a high phylogenetic resolution, the direct detection of microbial activity by SIP attracts the interests of scientists, especially combining with further analysis such as metagenomic or metatranscriptomic sequencing and Raman-FISH (Fortunato and Huber 2016; Grob et al., 2015; Huang et al., 2007).

In general, 12 to 14 fractions will be obtained after density gradient separation of nucleic acids. In order to have a successful separation of ¹³C-labeled nucleic acid, density shifts require a high labeling level in RNA or DNA (more than 10 atom% of C) (Manefield et al., 2002). However, nucleic acid-SIP harbors an ultra-high sensitivity as indicated by the successful separation of a mixture of 0.001% of ¹³C-labeled RNA (Aoyagi et al., 2015). This ultra-high sensitivity in identifying active microorganisms has been applied to detect dioxane degraders with a low activity (Aoyagi et al., 2018). Considering the high sensitivity of nucleic acid-SIP and low activity of Asgard archaea in marine sediments, the higher relative abundance of Lokiarchaeota in heavy fractions than light fractions suggests inorganic carbon assimilation into nucleic acids of these Lokiarchaeota (Fig. 2 in Chapter 4). Therefore, nucleic acid-SIP is a crucial technique to identify microorganisms with low activity.

Additionally, apart from methylotrophic methanogens and Lokiarchaeota, it is very likely that the nucleic acid-SIP method together with using ¹³C-labeled bicarbonate can detect other archaea such as Bathyarchaeota, some Thermoplasmata, ANMEs and so on. In marine sediments, these archaea are detectable and potentially active after feeding some organic substrates combining with ¹³C-DIC. Given that archaea activity in deeper marine sediment is very low, nucleic acid-SIP technique might be very useful to detect their activities.

5.4. Carbon fixation in bacteria in anoxic marine sediments

In the Helgoland mud sediments, divergent bacteria groups were detected. Specifically in the SIP incubations without amending electron donor, bacteria affiliated to Atribacteria, Planctomycetes (*Planctomycetaceae* and MSBL9), and Deltaproteobacteria (*Desulfobacteraceae*, *Desulfobulbaceae* and *Desulfuromonadaceae*) were identified, indicating the potential activity *in situ*. However, relative abundances of these bacteria did not show a substantial increase in the heavy fractions. Indeed, metagenomic analysis of these detected bacteria shows that Atribacteria are potentially heterotrophs in fermentation carbohydrates and organic acids (Nobu et al., 2016). Planctomycetes are also regarded as heterotrophs growing on carbohydrates or sugar polymers (Robbins et al., 2016; Woebken et al., 2007).

For the family of *Desulfobacteraceae*, *Desulfobulbaceae* and *Desulfuromonadaceae*, most members are heterotrophs, although some groups harbor the complete WL pathway and reverse citric acid cycle for carbon fixation with the reduction by H_2 (Brysch et al., 1987; Dorries et al., 2016; Finster et al., 2013), suggesting H_2 is required for autotrophic growth. Therefore, it seems that nucleic-SIP combining with incubations amended ¹³C-DIC is feasible for identifying active archaeal groups in marine sediments. The explanation is that some archaea such as methylotrophic methanogens, Lokiarchaeota, ANMEs and even Bathyarchaeota are mixotrophically assimilating both organic substrates and CO_2 as carbon source, coupling methyl intermediate production from organic carbon and CO_2 for biomass biosynthesis.

Based on these findings, further work should focus on the topics as follows:

1) Is it ubiquitous that CO_2 conversion to methane occurs in all methanogens? Except for hydrogenotrophic methanogens, methylotrophic methanogens can use CO_2 for methane production. But it is unknown whether acetoclastic methanogens can produce methane from CO_2 without syntrophic partners, although the complete acetyl-CoA pathway is detected in these methanogens.

2) RNA-SIP has shown that Lokiarchaeota is able to degrade lignin but the corresponding pathway was not detected. More evidence should be collected to prove how Asgard archaea participate in lignin degradation such as specific functional group of lignin molecular and secondary products of organic polymer degradation by obtaining highly complete metagenomes, enriching Asgard archaea and measuring intermediates.

3) Carbon fixation in other archaeal phyla. It is well known that some Euryarchaeota (hydrogenotrophic methanogens) and Thaumarchaeota (ammonia oxidizer) incorporate inorganic carbon for carbon source, but carbon assimilation such as in Bathyarchaeota, Woesearchaeota, ANMEs is not well studied. Nucleic acid- and lipid-SIP should be applied to study carbon assimilation patterns by these archaea from incubated samples. Further studies should explore factors influencing carbon fixation by archaea in marine sediments such as organic carbon content, geochemical profiles of sediments.

4) It is likely that archaea activities are associated with organic carbon degradation since organic carbon is necessary for archaea growth, i.e., methanol utilization in methylotrophic methanogens and organic polymer degradation by Lokiarchaeota. However, how organic compounds affect archaeal diversity and distribution in marine sediments is unclear. For example, is there any relationship between distributions of some specific archaea and organic carbon content, specific molecular structures of organic compounds or C:O:N:H ratios? Therefore, archaeal distributions, activities and organic carbon characteristics from different marine sediments should be investigated to establish their correlations.

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