

# Azamacrocyclic Ca<sup>2+</sup> Sensitive Contrast Agents for MR Imaging

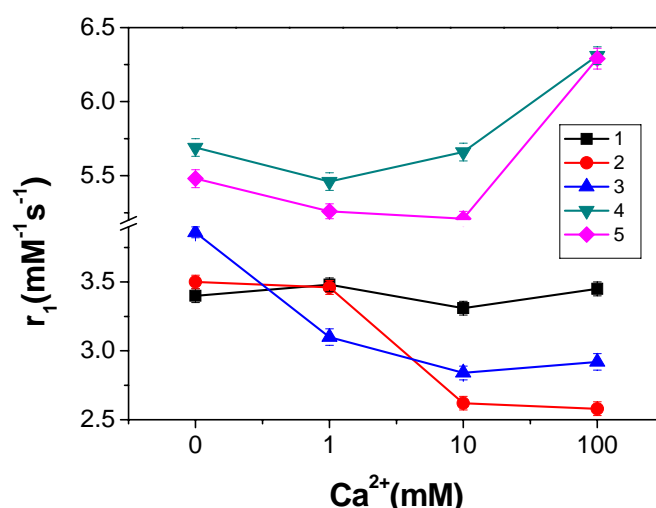
Ilgar Mamedov<sup>1</sup>, Goran Angelovski<sup>1</sup>, Jörg Henig<sup>2</sup>, Hermann A. Mayer<sup>2</sup> and Nikos K. Logothetis<sup>1</sup>,

<sup>1</sup>Department for Physiology of Cognitive Processes, Max-Planck Institute for Biological Cybernetics, Tübingen, Germany, <sup>2</sup>Institute for Inorganic Chemistry, University of Tübingen, Tübingen, Germany

**Introduction.** As calcium plays an important role in regulating a great variety of neuronal processes, many efforts are already made to generate gadolinium complexes that can act as a calcium-sensors in MRI.<sup>1</sup> We developed a series of the DO3A-based macrocyclic and bismacrocyclic gadolinium chelates, bearing phosphonate groups as an additional coordination sites. These complexes are hypothesized to change the MRI contrast dynamically with Ca<sup>2+</sup> concentration. Different lengths of the phosphonate side chains are exploited for fine-tuning the sensitivity of the agent to calcium ion concentration.

**Methods.** The macrocyclic ligands DO3A-**alkyl** aminobis(methylenphosphonates) (alkyl = propyl **1**; amyl **2**; hexyl **3**), and the bismacrocyclic bisDO3A-ethyleneaminobis (**alkyl**)phosphonates (alkyl = methyl **4**; ethyl **5**) were prepared in multistep reactions from cyclen. Gadolinium complexes were obtained by treating the ligands with GdCl<sub>3</sub> at pH 7.0 -7.5 and 70°C for 24h. Longitudinal and transverse relaxivities  $r_1$  and  $r_2$  were measured on a vertical 7 T/60 cm MRI Biospec system (Bruker Biospin, Germany), at pH 7.4 and different concentrations of Ca<sup>2+</sup>.

**Results.** The sensitivity of the complexes **1-3** for the changes in Ca<sup>2+</sup> concentrations increased with the chain length between the DO3A unit and the phosphonate functions (Fig. 1). Maximal observed change in relaxivity in presence of the physiological extracellular concentration of calcium (~1 mM) was up to 35% for compound **3**, and was selective to Ca<sup>2+</sup> compared to other physiologically important metals such as magnesium or zinc. The significant changes in the relaxivity of the bismacrocyclic compounds **4** and **5** were found in the presence of a higher concentration of Ca<sup>2+</sup> ions.



**Figure 1.** Dependence of the complex relaxivity to the Ca<sup>2+</sup> ions concentration

1. Li et al. J.Am.Chem.Soc., **1999**,121,1413-1414

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