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Characterization of polyphenols in Australian sweet lupin (*Lupinus angustifolius*) seed coat by HPLC-DAD-ESI-MS/MS

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26 Abstract

27 Seeds of the legume lupin (Lupinus spp.) are becoming increasingly important as human food. 28 The seed coat, at ~25% of the whole seed of Lupinus angustifolius (Australian sweet lupin, 29 ASL), is the main by-product of lupin kernel flour production. The primary market for lupin 30 seed coat is low value feed with very limited use in foods. In this study, seed coats of six ASL 31 commercial varieties from two growing sites were sampled for identification and quantification 32 of polyphenols using a high-performance liquid chromatography (HPLC) with diode array 33 detector (DAD) and coupled with a triple quadrupole mass spectrometer which equipped with 34 electrospray ionization source (ESI-MS/MS). Three flavones (apigenin-7-O-β-apiofuranosyl-35 6,8-di-C-β-glucopyranoside, vicenin 2, and apigenin-7-O-β-glucopyranoside), one isoflavone 36 (genistein) and one dihydroflavonol derivative (aromadendrin-6-C-β-D-glucopyranosyl-7-O-37 $[\beta$ -D-apiofuranosyl- $(1\rightarrow 2)$]-O- β -D-glucopyranoside), and several hydroxybenzoic and 38 hydroxycinnamic acid derivatives were identified. Considerable variations in levels of 39 individual polyphenols found apigenin-7-*O*-β-apiofuranosyl-6,8-di-*C*-βwere but 40 glucopyranoside was the predominant polyphenol in all samples accounting for 73.08 - 82.89 % 41 of the total free polyphenols. These results suggest that ASL seed coat could be valuable dietary 42 source of polyphenols.

Keywords: Lupinus angustifolius; Australian sweet lupin; seed coat; polyphenols; genotype 43

44 by environment; by-product; HPLC-DAD-ESI-MS/MS **List of Abbreviations**

ASL	Australian sweet lupin
BP	Bound polyphenols
ER	Eradu
FP	Free polyphenols
Mw	Molecular weight
WA	Western Australia
WH	Wongan Hills

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45 **1. Introduction**

Seeds of the legume lupin are attracting worldwide attention as a potential future staple 46 47 food (Johnson, Clements, Villarino, & Coorey, 2017). This is due to their multiple roles in 48 farming systems (through nitrogen fixation ability), and their likely benefits for human 49 nutrition and health contributed by the high dietary fibre and protein content. Western Australia 50 (WA) is the world's largest lupin producer, with Lupinus angustifolius (Australian sweet lupin, ASL) being the major species under production. However, ASL has a relatively higher 51 52 percentage of seed coat, generally 25% of the whole seed, than most of other pulses like 53 soybean (5-8%) and pea (Pisum sativum L.) (9-14%). As a result, flour production from the 54 dehulled kernels for human consumption has a high proportion of commercial loss (~31% in 55 Australia). This represents a tough disposal problem for the industry, since the seed coat has 56 little market value or demand; it is primarily a low value animal feed (Sipsas, 2008).

57 The seed coat of pulses, including chickpea (Sreerama, Neelam, Sashikala, & Pratape, 58 2010), faba bean (Boudjou, Oomah, Zaidi, & Hosseinian, 2013), field pea (Marles, Warkentin, & Bett, 2013), lentil (Oomah, Caspar, Malcolmson, & Bellido, 2011), and mung bean (Luo, 59 60 Cai, Wu, & Xu, 2016), is the predominant contributor of polyphenols to the whole seeds 61 (Zhong et al., 2018). For example, 80.3-84.2% of the total polyphenol and over 83.9% of total 62 flavonoid content of whole mung bean seed is present in the seed coat (Luo et al., 2016). 63 Additionally, significantly higher polyphenol levels are found in dark coloured (black and red) 64 chickpea and lentil seeds than those in lighter coloured (white and beige) counterparts (Xu, 65 Yuan, & Chang, 2007). In contrast, the total polyphenol content in seed coats of L. mutabilis,

L. albus, and *L. angustifolius* grown in Brazil is reported to be 1.15 - 4.49 mg catechin
equivalents (CE)/g dry basis which is much lower than in cotyledons (7.38 - 12.42 mg CE/g
dry basis) (Ranilla, Genovese, & Lajolo, 2009). A majority of tannins of lupin, however, is
present in the seed coat (Petterson, 1998).

Despite these conflicting results, to the best of our knowledge, no work on identification and quantification of individual polyphenols in lupin seed coats have been reported. Moreover, effects of genotype and environment on the polyphenols in the ASL seed coat has not been investigated. To this end, in this paper, individual polyphenols in ASL seed coat were identified and quantified. Six commercial varieties of ASL grown in two locations in WA were used to evaluate the effects of genotype, environment, and their interaction ($G \times E$) on contents of the individual polyphenolics.

77 2. Materials and methods

78 2.1. Materials

79 Whole seeds (ca 4 kg) of six ASL genotypes (Coromup, Mandelup, Jenabillup, PBA Barlock, PBA Gunyidi and PBA Jurien) harvested from two locations within 2015 growing 80 season were obtained from the Department of Primary Industries and Regional Development -81 82 Agriculture and Food (Kensington, WA, Australia). The lupins were grown in Wongan Hills (WH; 30.54 °S, 116.43 °E) and Eradu (ER; 28.70 °S, 115.05 °E) WA. Wongan Hills has a 83 84 rainfall of 388.4 mm annually and 153.5 clear days with temperature ranging from 12 °C to 28 °C, whereas Eradu has a rainfall of 450.4 mm annually but a smaller temperature variation 85 (14.4 °C -24.7 °C) and less clear days (108.1 days) (BOM (Bureau of Meteorology), 2017). 86

The two locations belong to lupin Agzone 5 and Agzone 2 respectively based on their rainfalls
(White, French, McLarty, & Grains Research and Development Corporation, 2008). The seeds
were cleaned manually, vacuum-packed in polyethylene bags and stored in the dark at 4 °C
until use.

91 The seeds were dehulled by a laboratory-scale AMAR dehuller (NSIC.SSI, India). Coarse 92 seed coats were separated by a vacuum separator (Kimseed Pty Ltd, WA, Australia) with 93 manual removal of any remaining broken cotyledons. The seed coats were dried at 50 °C for 94 overnight and milled to flours using a ZM 200 Retch Mill (Retsch Gmbh & Co, Haan, 95 Germany), then passed through (> 97 %) a 500 µm screen. The resulting seed coat flours were 96 thereafter vacuum-packed in polyethylene bags and stored at 4 °C in the dark until analysis.

97 2.2. Reagents and polyphenols standards

Hydrochloric acid, sodium hydroxide, ethyl acetate, and LC-MS grade acetonitrile, formic
acid and methanol were purchased from Thermo Fisher Scientific (Scoresby, Vic, Australia).
Authentic standards, including caffeic acid, *trans*-cinnamic acid and ferulic acid were
purchased from Sigma-Aldrich (Castle Hill, NSW, Australia). Vitexin (apigenin-8-*C*glucoside), *p*-coumaric acid, protocatechuic acid and genistein were purchased from Cayman
Chemical (Redfern, NSW, Australia).

104 2.3. Extraction of free and bound phenolic compounds

105 *Free polyphenols extraction:* Polyphenols were extracted as described by our research 106 team (Wu et al., 2016). Duplicate 2 g samples of each ground lupin seed coat was mixed with 107 10 mL chilled 80% methanol (methanol/water, v/v) in a 15 mL tube with polyvinyl lined cap. 108 After vortexed for 10 s, the tube was covered by aluminium foil and mixed by rotational-109 shaking for 2 h in the dark followed by centrifugation at $3220 \times g$, 4 °C for 10 min. The residue 110 was extracted twice more with 7 mL of 80% methanol, the supernatants were then collected, 111 pooled and made up to 25 mL with 80% methanol. Extracts were stored at -80 °C before free 112 phenolics analysis (FP fraction).

113 **Bound polyphenols extraction:** The seed coat residue from free fraction extraction was 114 transferred to 50 mL tube and mixed with 15 mL 2 mol/L NaOH. The mixture was vortexed 115 for 10 s; then the tube was flushed with N₂, capped and sealed. After covered by aluminium 116 foil and rotational-shaken for 2 h at ambient temperature, the mixture was acidified to pH 2 117 with 12 mol/L HCl. Alkaline hydrolysed previously-matrix bound phenolics (BP fraction) was 118 extracted with 15 mL ethyl acetate, intermittently shaken for 10 min then centrifuged at 3220 119 ×g, 4 °C for 5 min. Organic layer was collected and the extraction was repeated for 5 times. 120 The combined extract was evaporated at 30 °C under N2 to dryness. Extract was re-suspended 121 with 10 mL of 100% methanol and stored at -80 °C before use.

122 2.4. HPLC-DAD-ESI-MS/MS analysis

Individual polyphenols were identified and quantified according to the procedure developed by our research group (Wu et al., 2016) with minor modifications. An Agilent 1200 auto-sampler HPLC system was coupled to diode array detector (DAD) and an MS/MS system (Agilent 6460 LC-QQQ, Agilent Technologies, Palo Alto, CA, USA). The DAD was set to monitor signals at 190-600 nm with resolution of 2 nm. After filtering through a millipore membrane (0.22 µm), samples (20 µL) were injected into the Kinetex XB-C 18 reversed phase129 HPLC column (5 µm, 250 × 4.6 mm, Phenomenex, Torrance, CA, USA) at 25 °C. Solvent A 130 consisted of 0.05% formic acid in water; solvent B was acetonitrile. Flow rate was set at 0.5 131 mL/min to facilitate the coupling to mass spectrometer. A linear gradient elution was conducted: 132 linear to 8% B from 0% B in 18 min, and changed to 18% B in 2 min, to 20% B in 15 min, 133 then linear to 80% B in 10 min, and finally to 100% B in 2 min. This was followed by extra 8 134 min for column washing (100% B) and re-equilibration (100% A) respectively. 135 Two stage MS/MS spectra were acquired in the electrospray ionisation (ESI) negative ion 136 mode with full scan ranging from m/z 50 to 1300. 5 L/min of N₂ was employed as the nebulizing 137 gas at 45 psi, 300°C. Capillary voltage and nozzle voltage were set at 3.5 kV and -500 V 138 respectively. Sheath gas (N₂) was maintained at 11 L/min and 250 °C. Collision energy was 139 selected as 25 eV based on the abundance of the daughter ions. Data acquisition and analyses 140 were performed on the Agilent MassHunter workstation. The identification confidence level 141 of each compound was presented in table 1 following La Barbera et al. (2017) described. 142 For quantification of individual phenolic compounds, only those peaks with signal/noise 143 (S/N) >10 were selected. Due to the lack of available standards, individual phenolic compound 144 (except protocatechuic acid, ferulic acid and genistein) were semi-quantified using compounds 145 with similar chemical structures and UV absorption. Apigenin-7-O-β-apiofuranosyl-6,8-di-C-146 β -D-glucopyranoside (F3), vicenin 2 (F4) and apigenin-7-O- β -glucopyranoside (F5/B5) were 147 quantified using vitexin as standard (Siger et al., 2012). Phenolic acid hexosides, including cinnamic acid glucoside (F1), p-coumaric acid glucoside (B3) and ferulic acid glucoside (B4) 148 were quantified using their corresponding phenolic acids. Dicaffeoylquinic acid (F6) content 149

150 was calculated using caffeic acid standard curve. Aromadendrin-6-C-β-D-glucopyranosyl-7-151 O-[β -D-apiofuranosyl-(1 \rightarrow 2)]-O- β -D-glucopyranoside (F2) quantified using was 152 dihydroquercetin (or taxifolin). B1 and F7/B7 were not quantified since the peaks were not 153 confidently identified. All results for the compounds above were expressed as µg standards 154 equivalent per g of dry sample (µg vitexin equivalent per g of dry sample for F3 and F4, for 155 example).

Linearity of the selected seven standards, and their spike recovery which was performed using lupin (Coromup, ER) seed coat and detected in free fraction are presented in table 2. The limit of detection (LOD) and quantification (LOQ) were calculated at S/N ratios of 3 and 10 respectively. Six replicates of a mixed standards solution at a same concentration were freshly prepared and analysed in a single day and six separate days to evaluate the intraday and interday precision respectively. The intra- and inter-day variations were calculated as relative standard deviation (RSD) of the peak area.

163 2.5. Statistical analysis

Fragmentations of flavonoid aglycones and glycosides were designated using nomenclature systems proposed by Ma, Li, Vanden Heuvel, and Claeys (1997) and Domon and Costello (1988) respectively. Proposed mass spectrum fragmentation pathways were drawn using ChemDraw Prime software (V 16.0, PerkinElmer, VIC, Australia).

168 All the results were reposted on dry basis (d.b.) and expressed as mean \pm standard deviation

169 $(n\geq 2)$. Two-way ANOVA with Tukey post hoc tests were conducted to investigate the main

170 effects of genotype (G), location (E) and their interaction (G×E). One-way ANOVA by Tukey

test was performed to find any significant differences (P < 0.05) between means of genotypes
within locations. Independent samples t-test was used to compare across different locations.

- 173 All analyses were performed on SPSS Statistics V22 (SPSS Inc., Chicago, Illinois, US).
- 174 **3. Results and discussion**
- 175 *3.1. Identifications of individual polyphenols*
- 176 The HPLC-DAD chromatograms of free polyphenol extract (FP) and bound polyphenol
- 177 extract (BP) of PBA Barlock are shown in Fig 1. Peaks were labelled as F1-8 (FP) and B1-8
- 178 (BP) respectively. HPLC retention time (Rt), UV-Vis absorption maximum wavelength (λ_{max}),
- deprotonated molecular ions and fragment ions of the 13 individual polyphenols are listed intable 1.
- 181 *3.1.1 Flavones in free faction*

182 F3 was the dominant individual polyphenol in FP fraction. The UV-Vis absorbance at 275 nm and 340 nm of F3 (Fig 2.) showed the characteristic UV absorption bands of flavones (Wu 183 184 et al., 2016). The absorption properties were largely the same as that of the authentic apigenin-8-C-β-D-glucopyranoside (vitexin) standard (270 nm, 332 nm). The mass spectra of the F3 185 revealed the $[M-H]^{-1}$ ion at m/z 725. The yield of two-fold neutral 120 amu fragments (m/z 725) 186 to m/z 605, ${}^{0,2}X_{1,1}$; and m/z 455 to m/z 335, ${}^{0,2}X_{1,2}$), as well as ions at m/z 383 (apigenin + 113) 187 and m/z 353 (apigenin + 83) are characteristics of apigenin-di-C-glycosyl flavone 188 fragmentations (Fig 2.&3.) (Ferreres, Silva, Andrade, Seabra, & Ferreira, 2003). In addition, 189 fragments at m/z 593 ([(M-H)-132]⁻, Y₀), m/z 575 ([(M-H)-150]⁻, Z_{0,1}) and m/z 455 ([(M-H)-190 191 120-150]⁻, ${}^{0,2}X_{1,1}Z_{0,2}$) suggested the presence of *O*-pentose which was supported by diagnostic

192	ion pairs at m/z 575 and at m/z 455 (Cuyckens & Claeys, 2004; Vukics & Guttman, 2010). The
193	pentose can be attached either to the flavone aglycone giving an -O-glycoside or one of the two
194	glucoses forming a C,O-glycoside, which is difficult to distinguish by MS/MS (Cavaliere,
195	Foglia, Pastorini, Samperi, & Lagana, 2005). However, this compound was tentatively
196	identified as apigenin-7-O-β-apiofuranosyl-6,8-di-C-β-D-glucopyranoside (Api-7-O-Apif-6,8-
197	di-C-Glcp), mainly because it was previously isolated and identified using MS and nuclear
198	magnetic resonance (NMR) spectra in the whole seeds of <i>L. hartwegii</i> by Kamel (2003) and <i>L</i> .
199	termis by Elbandy and Rho (2014). The identification is further supported by the results of
200	Siger et al. (2012) who identified the compound in <i>L. angustifolius</i> , <i>L. luteus</i> and <i>L. albus</i> seeds
201	grown in Poland, and reported a highly similar mass spectra.
202	The UV-Vis absorption properties and fragment performances of F4 were very similar
203	with those of F3, with typical UV absorption at 273 nm (Band II) indicating the introduction
204	of di-C-glycosides substituents at C6 or C8 positions (Santos-Buelga, García-Viguera, &
205	Tomás-Barberán, 2003). Likewise, the same two successive losses of 120 amu neutral portions
206	$(m/z 593 \text{ to } m/z 473, {}^{0,2}X_1; \text{ and } m/z 473 \text{ to } m/z 353, {}^{0,2}X_2)$, indicated that it was flavone di-C-
207	glycosides either. Moreover, the [M-H] ⁻ of F4 (m/z 593) was 132 amu smaller than that of F3
208	(<i>m</i> / <i>z</i> 725) but accompanied by the same fragment ions at <i>m</i> / <i>z</i> 383 ([(M-H)-120-90] ⁻) and <i>m</i> / <i>z</i>
209	353 ([(M-H)-120-120] ⁻), indicating it was apigenin di-C-glycosides but was not conjugated by
210	the pentose, i.e., apiose. Ions at m/z 297 [(M-H)-120-120-56] ⁻ corresponded to the further losses
211	of 2×CO. As such, F4 was tentatively identified as apigenin-6,8-di-C-β-D-glucopyranoside
212	(vicenin 2). This was consistent with the mass spectra of the standard (Cao, Yin, Qin, Cheng,

& Chen, 2014), and the compound was previously reported in seeds of lupins (Siger et al.,
2012).

215 Both F5 and B5 had λ_{max} at 278 nm and 320 nm. The deprotonated molecular ion [M-H]⁻ 216 of the compound was m/z 431 but this study failed to obtain its ion fragments (MS²). However, 217 based on the UV absorption and deprotonated molecular ion, as well as the slightly longer 218 elution time (33.94 min) then the authentic standard vitexin (33.75 min), it could be tentatively 219 identified as apigenin-7-O-β-glucopyranoside (Api-7-O-Glcp) (Santos-Buelga et al., 2003). 220 The longer retention time of F4 (di-C-glycosides) than that of F3 (O-pentosyl-di-C-glycosides) 221 in our study could be explained by the induction of a third sugar moiety in F3 which can 222 increase polarity and thus shorter retention time (Santos-Buelga et al., 2003). A longer retention 223 time of vitexin (mono-C-glycosides) than F4 supported the elution order. In contrast, Siger et 224 al. (2012) reported a reverse elution order of the two compounds (F3 and F4). This 225 inconsistency could be attributed to the different HPLC conditions and column used. In the 226 case of the longer retention time of F5/B5 (mono-O-glycosides) than vitexin (mono-C-227 glycosides), Santos-Buelga et al. (2003) revealed that flavone O-glycosides generally elute 228 after the corresponding C-glycosides.

Besides the compound **F3**, **F4** and **F5/B5** found in our ASL seed coat samples, much more complex but similar flavone-glycosides were also reported, e.g., apigenin-7-*O*-βapiofuranosyl-6-*C*-β-glucopyranosyl-8-*C*- (6''''-*O*-*E*-feruloyl)-β-glucopyranoside (molecular weight (Mw) = 902) in *L. hartwegii* (Kamel, 2003) ; apigenin-6-*C*-β-D-glucopyranosyl-8-*C*- $[\alpha$ -L-rhamnopyranosyl-(1→2)]-β-glucopyranoside (Mw = 740) and apigenin-6-*C*-β-D- glucopyranosyl-8-*C*-[β -D-apiofuranosyl-(1 \rightarrow 2)]- β -glucopyranoside (Mw = 726) in *L. termis* (Elbandy & Rho, 2014); apigenin-7-neohesperidoside (Mw = 578), apigenin-7-apioglucoside (Mw = 564) and several unidentified apigenin derivatives in raw and germinated *L. angustifolius* seeds (Dueñas, Hernandez, Estrella, & Fernandez, 2009). Apart from apigenin as the aglycone, luteolin derivatives and diosmetin derivatives are also found in *L. angustifolius* seeds. All of these studies suggest the high complexity of flavones in lupins.

240 3.1.2. Isoflavones in free (FP) and bound polyphenol (BP) faction

F8/B8 had a deprotonated ion at m/z 269, and showed a characteristic ion at m/z 133 which 241 242 was the same as that of the authentic genistein standard but different from those of the apigenin 243 standard (m/z 151 and m/z 117) (Vukics & Guttman, 2010). It was therefore identified as 244 genistein and this identification was also supported by its UV spectra ($\lambda_{max} = 267$ nm). Some 245 studies reported that no isoflavones were detected in L. angustifolius species, whereas high 246 level of isoflavones (e.g., mutabilein, mutabilin) were found in L. mutabilis (Dini, Schettino, 247 & Dini, 1998; Ranilla et al., 2009). Other researchers, however, found genistein and its derivatives in raw and/or germinated L. angustifolius seeds. Examples include genistein 248 apiofuranosyl diglycoside (Mw = 726), genistein diglucoside (Mw = 594), genistein-7-O- β -249 250 glucopyranoside (Mw = 432), genistein diglycoside (Mw = 564, attaching pentose-hexose 251 moiety) and several types of genistein acetyl glycosides (Dueñas et al., 2009). The first three 252 compounds which had the same deprotonated molecular ions as F3, F4 and F5/B5 in the 253 current study, however, were identified as flavones, with apigenin being the corresponding aglycone instead of genistein, due to their characteristic UV absorptions. 254

255 3.1.3. Dihydroflavonols in free faction

The strong UV-Vis absorption peak at 295 nm and the small peak of lower intensity 256 (shoulder) at 325 nm implied that the F2 could be flavanone or dihydroflavonol (Fig 2.) (Mabry, 257 258 Markham, & Thomas, 1970). However, flavanones almost only occur in Citrus genus, but 259 dihydroflavonols are ubiquitously distributed in legumes (Santos-Buelga et al., 2003). Dueñas 260 et al. (2009) have found 4 dihydroflavonols, namely 3 dihydroquercetin (taxifolin) derivatives 261 and dihydrokaempferol (aromadendrin) acetylglycoside in L. angustifolius seeds. 262 The 162 amu (m/z 743 to m/z 581, Y₁) and 180 amu (m/z 743 to m/z 563, Z₁) losses from 263 deprotonated molecular ion of F2 were typical pattern of O-glycoside fragmentations, 264 suggesting the presence of hexose then giving an O- or C, O- diglycosyl structure (Gattuso, 265 Barreca, Gargiulli, Leuzzi, & Caristi, 2007). Moreover, a series of ions, [M-H-18]⁻ (m/z 725, 266 E₁⁻), $[M-H-90]^{-}$ (*m*/*z* 653, ^{0,3}X_{2,1}), $[M-H-120]^{-}$ (*m*/*z* 623, ^{0,2}X_{2,1}), $[M-H-120-90]^{-}$ (*m*/*z* 533, $^{0.2}X_{2,1}^{0.3}X_{2,2}$, [M-H-120-120]⁻ (m/z 503, $^{0.2}X_{2,1}^{0.2}X_{2,2}$) were observed as the characteristic 267 268 fragments of two glucoside moieties. The further subsequent losses from m/z 461 ([M-H-162-120], $Y_1^{0,2}X_{1,1}$ to m/z 401 (60 amu, $Y_1^{0,2}X_{1,1}^{0,3}X_0$), m/z 371 (90 amu, $Y_1^{0,2}X_{2,1}^{0,2}X_0$) and m/z269 341 (120 amu, $Y_1^{0,2}X_{2,1}^{0,1}X_0$) were C-bound cleavages of pentose. Together, although further 270 271 aglycone analyses are needed, the compound was proposed to be aromadendrin-6-C-β-Dglucopyranosyl-7-O-[β -D-apiofuranosyl-(1 \rightarrow 2)]-O- β -D-glucopyranoside (Aro-6-C-Glcp-7-272 273 O-Apif Glcp).

274 3.1.4. Phenolic acid derivatives in free (FP) and bound polyphenol (BP) faction

275	F1 showed deprotonated molecular ion $[M-H]^-$ at m/z 309, with ion at m/z 291 being its
276	dehydrated fragment (Demarque, Crotti, Vessecchi, Lopes, & Lopes, 2016). Decarboxylated
277	ion (m/z 247), glycoside fragment (m/z 180) indicated that F1 could be cinnamic acid glucoside
278	(Mw = 310). F6 had a λ_{max} at 327 nm, and parent ion at m/z 515 accompanied by distinguished
279	fragment ion at m/z 179 and then lost a carboxyl group to m/z 135, indicating the caffeic acid
280	moiety. It was thus tentatively identified as dicaffeoylquinic acid (diCQA) (Clifford, Knight,
281	& Kuhnert, 2005). Czubinski, Siger, and Lampart-Szczapa (2016) also reported two nearby
282	peaks at the similar retention times of F5/B5 (Api-7-O-Glcp) and F6. Instead, the authors
283	postulated the compounds to be apigenin and cinnamic acid derivative respectively due to the
284	lack of MS data in their study.

285 In terms of phenolic acid hexosides in bound fraction, *p*-coumaric acid glucoside (B3) and 286 ferulic acid glucoside (B4) showed diagnostic 162 amu losses ([M-H-Glu]⁻) with resulting in the corresponding ions at deprotonated phenolic acids. Deprotonated protocatechuic acid (B2, 287 $[M-H]^-$ at m/z 153) and p-coumaric acid moiety ($[M-H-162]^-$ at m/z 163) of B3 showed 288 289 characteristic 44 amu carboxyl group losses and then giving signals at [M-H-COO]⁻ and [M-290 H-162-COO]⁻, namely ions at *m/z* 109 and *m/z* 119 respectively (Gruz, Novák, & Strnad, 2008; 291 Wu et al., 2016). Moreover, fragment at m/z 178 of **B4** was due to a methyl radical loss ([M-292 H-162-CH₃]⁻, 15 amu) of the ferulic acid moiety and then a further 44 amu loss to m/z 134 (Wu 293 et al., 2016). We failed to get MS data for B6, but it had same retention time and UV-Vis absorption properties with the authentic ferulic acid standard. B1 showed deprotonated ion at 294 m/z 405 but produced fragments at m/z 191 and m/z 111 which are characteristic for quinic acid 295

(Clifford et al., 2005). Therefore, the compound was denoted as quinic acid derivative. Due to
the lack of MS² data for either, F7 and B7 were designated as cinnamic acid derivative mainly
basing on their characteristic maximum UV absorption at 340 nm. Protocatechuic acid (B2),
ferulic acid glycoside (B4), ferulic acid (B6), *p*-coumaric acid and other hydroxybenzoic and
hydroxycinnamic compounds previously found in *L. angustifolius* seeds (Dueñas et al., 2009). *3.2. Quantification of individual polyphenols.*

Results for HPLC-DAD method validation are shown in Table 2. Briefly, the R² of the 7 analysed standards were all greater than 0.99, indicating good linearities within the ranges used. The intra- and inter-day variations of all the standards were lower than 0.48% and 2.74% respectively. Moreover, the percentage of recovery of these standards which spiked in lupin (Coromup, ER) seed coat ranged from 97.61% to 104.38% with acceptable precision. The results suggested that the HPLC-DAD method is adequate for quantifying the selected phenolics.

309 *3.2.1. Quantifications of individual polyphenols in free fraction.*

Of all individual polyphenols of the 12 lupin seed coat samples (6 genotypes by 2 locations), apigenin-7-*O*- β -apiofuranosyl-6,8-di-*C*- β -D-glucopyranoside (Api-7-*O*-Api*f*-6,8di-*C*-Glc*p*, **F3**) was the dominant compound in the free fraction, ranging from 697.85 µg/g d.b. to 1011.82 µg/g d.b. (as vitexin equivalent), which accounted for 73.08 - 82.89 % of the total polyphenols in free fraction (Table 3). High contents of this compound have also been found in Polish grown *L. angustifolius*, *L. luteus* and *L. albus* whole seeds, 409.6-428.8 µg/g d.b., 709.8-876.9 µg/g d.b. and 257.5-259.5µg/g d.b. respectively (Siger et al., 2012). Interestingly, 317 those authors also reported high levels of vicenin 2 (F4 in the present study) in the whole seeds 318 of the three-lupin species, being 277.8-302.5 µg/g d.b., 536.3-631.4 µg/g d.b. and 119.0-143.0 319 µg/g d.b. respectively. In contrast, much lower contents of vicenin 2 (F4) were found in our 320 seed coats, from 24.65 μ g/g d.b. up to 59.53 μ g/g d.b. The differences might partly be related 321 to the genetic (different genotypes) and environmental (different grown locations) differences, 322 because considerable variability was also apparent in our collection as will be discussed below. 323 In addition, the differences in distributions of the two compounds in the seed coat and cotyledon may have also contributed. Luo et al. (2016) reported provocatively high vitexin 324 325 (apigenin-8-C-glucoside) and isovitexin (apigenin-6-C-glucoside) contents (37,430 µg/g d.b. and 47,180 µg/g d.b., respectively) in mung bean seed coat, while the two compounds were 326 327 not detected in mung bean cotyledons. In the current study the high percentage of api-7-O-328 Apif-6,8-di-C-Glcp (F3) suggested that lupin seed coat could be a good plant source of the 329 compound. For example, raw celery which is regularly the main dietary source of apigenin 330 usually contains between 28.5 and 240.2 μ g/g, but up to 786.5 μ g/g in the seeds (Bhagwat & Haytowitz, 2014). Total flavone intakes have been associated with lower risk for all-cause 331 332 mortality (Mink, 2007). Followed by purification and investigations of the potential 333 bioactivities of F3, the compound also could be used to produce other apigenin derivatives 334 (e.g., vitexin and isovitexin) by removing particular glycosides (e.g., by using acid hydrolysis) 335 resulting in compounds that may have multiple pharmacological effects (He et al., 2016). Phenolic acids (in free or esterified form) of lupins seeds, including *p*-hydroxybenzoic acid, 336

337 procatechuic acid, chlorogenic acid, vanillin acid, p-coumaric acid and ferulic acid, have

338 previously been indicated to mainly occur in the seed coat of L. luteus, L. albus, and L. angustifolius rather than the cotyledon (Lampart-Szczapa et al., 2003). The HPLC conditions 339 340 in our study were carefully developed to maximise separation of gallic acid, protocacheuic acid, 341 catechin, caffeic acid, p-coumaric acid, ferulic acid and trans-cinnamic acid standards (not 342 shown). In this context, contrary to expectations, this study found only three hydroxycinnamic 343 derivatives (F1, F6 and F7) in free fraction. But total hydroxycinnamics level was up to 99.65 344 $\mu g/g$ d.b. (as standards equivalents), that was much higher than published data of 6.06 $\mu g/g$ d.b. 345 in L. albus seed coat, and around 2.50 µg/g d.b. in L. angustifolius seed coat (Lampart-Szczapa 346 et al., 2003). Oppositely, Sosulski and Dabrowski (1984) revealed that only trans-p-coumaric 347 acid, trans-ferulic acid and p-hydroxybenzoic acid occurred in L. albus seed coat but totally up to 44 µg/g d.b.. In fact, the existing data on content of phenolic acids in whole L. angustifolius 348 seeds show a large variability, from 4.9 μ g/g d.b. to 58.14 μ g/g d.b. as reported by Dueñas et 349 350 al. (2009) and Siger et al. (2012) respectively.

351 It was also somewhat surprising that high levels of isoflavones (genistein) were found in 352 the free fraction in our study at 22.30 - 62.60 μ g/g d.b., which differed from some previous 353 studies that reported very low levels of isoflavones (0.9 µg/g d.b.) in whole L. angustifolius 354 seeds and did not detect them in its seed coat (Dueñas et al., 2009; Ranilla et al., 2009). 355 However compared to the results of the present study, much higher levels of genistein and its 356 derivative were previously reported in L. mutabilis seed coats, 98.09-870 µg/g fresh weight 357 (Ranilla et al., 2009). Not unexpectedly, no anthocyanins were found in the lupin seed coats by 358 the HPLC-MS/MS, which could be supported by the results at 520 nm of the DAD. Oomah,

359	Tiger, Olson, and Balasubramanian (2006) reported that only negligible anthocyanins were
360	found in Australian sweet lupin seed using UV-Vis method. However, positive ion mode is
361	generally used to identify the anthocyanins using mass spectrometry (Motilva, Serra, & Macia,
362	2013). Therefore, further investigations should be performed. Nevertheless, the results of this
363	study indicated that the polyphenols in lupin seed coats were primary flavonoids (flavones plus
364	isoflavone), 90.64 % to 94.41 % of total free polyphenol content, a finding that in agreement
365	with that of the lentil seed coat polyphenol profile (Dueñas, Hernández, & Estrella, 2002), and
366	that of whole seeds of <i>L. angustifolius</i> , <i>L. luteus</i> and <i>L. albus</i> (Siger et al., 2012).
367	As mentioned, the quantity of individual phenolic compounds varied across the 6
368	genotypes and the 2 locations, whereas the profile was found to be constantly stable. Statistical
369	analysis revealed that the levels of free individuals and total free polyphenol content of ASL
370	seed coats were significantly affected by genotype, location and their interaction, with all of
371	the effects being significant at $P < 0.01$. In contrast, location exhibited no effects on total
372	flavones (P = 0.189). Among the 6 genotypes, the lowest F3 level, and accordingly, the lowest
373	total flavone content and total polyphenol content were measured in PBA Jurien ($P < 0.01$),
374	which followed by PBA Gunyidi (P < 0.01) and consistently so for both ER and WH.
375	Comparisons across the two locations, seeds of all genotypes harvested from WH accumulated
376	lower level of total hydroxycinnamics than that of the counterparts from ER ($P < 0.001$). The
377	higher temperature and UV radiation, but lower rainfall at WH tended to decrease free phenolic
378	acid derivatives levels in the lupin seed coats. A reverse trend, however, was observed in
379	genistein content. Collectively, location exerted effects on the levels of phenolic individuals

but were in genotype-dependent manner; genotype was the determining contributor of theobserved variations.

382 *3.2.2. Quantification of individual polyphenols in bound fraction.*

383 It was the first time to evaluate bound polyphenols in lupin seed coat. Bound polyphenols 384 could reach colon then largely be metabolized by gut bacteria and show physiological benefits 385 (Saura-Calixto, 2012). Although acid hydrolysis was used to hydrolyse the polyphenols from 386 seed coats of chickpea (Sreerama, Sashikala, & Pratape, 2010) and lentil (Dueñas, Sun, 387 Hernández, Estrella, & Spranger, 2003), alkaline hydrolysis was used in our study since 388 alkaline hydrolysis was reported to be a better procedure to release polyphenols from 389 polysaccharides than acid hydrolysis because it (1) can reduce polyphenols (especially 390 flavonoids and phenolic acids) losses; (2) is an effective method of cleavage of ester bonds 391 which bind polyphenols to the cell wall (Acosta-Estrada, Gutierrez-Uribe, & Serna-Saldivar, 392 2014). This approach has been employed to release polyphenols of 10 legumes seed coats 393 (Sosulski & Dabrowski, 1984), and more recently lentil seed coat (Dueñas et al., 2002). As 394 shown in table 4, the majority of phenolic individual types in bound fraction were phenolic 395 acid derivatives. No api-7-O-Apif-6,8-di-C-Glcp (F3) and vicenin 2 (F4) were detected in the 396 bound fraction. Ferulic acid derivatives widely occur in plant cell walls, and contribute to cell 397 wall rigidity by crosslinking polysaccharides and lignin (Rosazza, Huang, Dostal, Volm, & 398 Rousseau, 1995). In addition, they also esterify with various compounds (e.g., flavonoids, 399 sterols and hydroxycarboxylic acids), that can be cleaved by alkaline hydrolysis.

400	Following the trends in the results of polyphenol contents in the free fraction, the total
401	polyphenol contents varied among the genotypes and locations, but the results in bound fraction
402	showed much larger standard deviations. Generally, the effects of genotype, location and their
403	interaction on bound polyphenol quantity were found in great similarities with free fraction,
404	but with location and genotype × location showing no influences on levels of both api-7-O-
405	Glcp (B5, P = 0.117 and 0.269 respectively) and genistein (B8, P = 0.613 and 0.717
406	respectively). The hydroxycinnamics and api-7-O-Glcp were the dominant bound phenolic
407	individuals, totally accounted up to 96.58 % of total bound phenolic compounds. In addition,
408	individual and total bound polyphenol content of PBA Jurien of the two locations were the
409	lowest among the genotypes ($P < 0.001$). Conversely, PBA Barlock of the two locations had
410	the highest total bound polyphenol content which was mainly contributed by the highest levels
411	of total hydroxycinnamics. Particularly, the protocatechuic acid levels in PBA Jurien, $5.07 \pm$
412	2.99 μ g/g d.b. (ER) and 5.68 \pm 0.42 μ g/g d.b. (WH) respectively, were much lower than PBA
413	Barlock (51.45 \pm 4.96 $\mu g/g$ d.b. and 63.38 \pm 2.03 $\mu g/g$ d.b. respectively). In contrast to free
414	fraction with respect to total bound polyphenol content, only PBA Gunyidi and Mandelup seeds
415	from WH showed statistically significantly higher than those of ER, but the remaining did not.
416	4. Conclusions
417	Up to the present time, lupin seed coat is a low value animal feed and a waste disposal
418	issue for lupin flour millers. In this study, three flavones (apigenin-7-O-β-apiofuranosyl-6,8-

419 di-C-β-glucopyranoside, vicenin 2 and apigenin-7-O-β-glucopyranoside), one isoflavone

420 (genistein) and one dihydroflavonol derivative (aromadendrin-6-C-β-D-glucopyranosyl-7-O-

421 $[\beta$ -D-apiofuranosyl- $(1\rightarrow 2)$]-O- β -D-glucopyranoside), together with several hydroxybenzoic and hydroxycinnamic acid derivatives were, for the first time, identified in ASL seed coats 422 using HPLC-ESI-MS/MS. Mass spectrum fragmentation pathways for apigenin-7-O-β-423 424 apiofuranosyl-6,8-di-C- β -glucopyranoside (F3) and vicenin 2 were also proposed. The 425 remarkable concentration of F3 in the free polyphenol extracts highlighted the potential that 426 lupin seed coat could be good source for the compound. Taken together, these results of this study support and promote the idea that ASL seed coat could be further value added by 427 428 exploring the potential for it as a fibre-polyphenol bioactive ingredient, manufacturing 429 flavonoid-fortified high fibre foods and flavonoid-based nutraceuticals, for example. However, 430 future studies are required to further optimize the phenolic extraction method and MS/MS 431 conditions (like using positive ion mode and higher collision energy); and isolate and 432 standardize F3 to confirm structures (especially its glycosylation patterns) using higher 433 resolution MS spectrometry and NMR; investigate the influences of processing, including 434 harvest, storage, dehulling, milling and food development technologies (e.g., baking, boiling, 435 extrusion) on polyphenol composition and bioavailability in ASL seed coat.

This study has also examined the impacts of genotype, environment and their interaction on ASL seed coat polyphenols. The results demonstrated that both free and bound polyphenols in ASL seed coat were significantly affected by all of genotypic and environmental factors. However, the observed variation was largely attributed to genotype. Notwithstanding, only two growing sites were selected in this study. Their environmental conditions were similar and potential seasonal effects were not included. Further screenings, using a broader range of
environmental conditions, are important to obtain more comprehensive insights on this matter.

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458 **Conflict of interest statement**

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612	Figure Caption
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615	Figure 1. UV chromatogram at 280 nm (DAD) and base peak chromatogram (BPC, all) of the
616	free (A) and bound fraction (B) of PBA Barlock (ER) seed coat.
617	
618	Figure 2. ESI-MS/MS spectra (a) and UV-Vis absorption profile (b) of aromadendrin-6- C - β -
619	D-glucopyranosyl-7- O -[β -D-apiofuranosyl-(1 \rightarrow 2)]- O - β -D-glucopyranoside (F2) and
620	apigenin-7- <i>O</i> -β-apiofuranosyl-6,8-di- <i>C</i> -β-D-glucopyranoside (F3).
621	
622	Figure 3. Proposed mass spectrum fragmentation pathway for apigenin-7- <i>O</i> -β-apiofuranosyl-
623	6,8-di- C - β -D-glucopyranoside (F3) and apigenin-6,8-di- C - β -D-glucopyranoside (F4, dashed

624 box).





627 Figure 1. UV chromatogram at 280 nm (DAD) and base peak chromatogram (BPC, all) of the

- 628 free (A) and bound fraction (B) of PBA Barlock (ER) seed coat.
- 629
- 630



631 Figure 2. ESI-MS/MS spectra (a) and UV-Vis absorption profile (b) of aromadendrin-6-*C*-β-D-glucopyranosyl-7-*O*-[β-D-apiofuranosyl-(1 \rightarrow 2)]-

O-β-D-glucopyranoside (**F2**) and apigenin-7-*O*-β-apiofuranosyl-6,8-di-*C*-β-D-glucopyranoside (**F3**).



633 Figure 3. Proposed mass spectrum fragmentation pathway for apigenin-7-*O*-β-apiofuranosyl-6,8-di-*C*-β-D-glucopyranoside (**F3**) and apigenin-

634 6,8-di-C- β -D-glucopyranoside (F4, dashed box).

636 TABLES

637

638 Table 1. Polyphenols identified by HPLC-ESI-MS/MS in samples

Peak No.	t _R (min)	λ_{max}	[M-H] ⁻ <i>m/z</i>	<i>m/z</i> MS/MS (Abundance %)	Assigned identity	Identification confidence level	Reference
F1	20.63	275	309a	291(64); 247 (3); 180(30); 128(99)	Cinnamic acid glucoside	2a	Gruz, Novák, & Strnad, 2008
B1	9.89	202	405a	191(44);111(100)	Quinic acid derivative		Clifford, Knight, & Kuhnert, 2005
B2	23.08	275	153b	135(40); 109(80)	Protocatechuic acid	1	Gruz, Novák, & Strnad, 2008
B3	25.25	226; 310	325b	163 (87); 119(64)	<i>p</i> -coumaric acid glucoside	2a	Gruz, Novák, & Strnad, 2008
F2	25.57	290; 325sh	743b	653(17); 623(100); 581(32); 563 (13); 533(31); 503 (22); 461(54); 371(15)	Aromadendrin-6- <i>C</i> - β -D-glucopyranosyl- 7- <i>O</i> -[β -D-apiofuranosyl-(1 \rightarrow 2)]- <i>O</i> - β -D-glucopyranoside	2b	Dueñas et al., 2009
B4	25.73	290	355b	193(34);178(30);134(100)	Ferulic acid glucoside	2a	Gruz, Novák, & Strnad, 2008
F3	26.43	275; 340	725a	635(2); 605(10); 593(5); 575(7); 503(1); 455(7); 383(1); 353(1); 335 (3)	Apigenin-7- <i>O</i> -β-apiofuranosyl-6,8-di- <i>C</i> - β-glucopyranoside	2a	Siger et al., 2012

F4	26.92	273; 340	593b	503 (9); 473 (16); 383(11); 353(22); 297(3)	Vicenin 2	2a	Cao, Yin, Qin, Cheng, & Chen, 2014
F5/B5	33.94	278; 320	431c	n.d.	Apigenin-7- <i>O</i> -β-glucopyranoside	2a	Santos-Buelga et al., 2003
F6	34.58	327	515c	249(5); 179(2); 135(6)	Dicaffeoylquinic acid	2a	Clifford, Knight, & Kuhnert, 2005
B6	36.94	300sh; 330	n.d.	n.d.	Ferulic acid	1	Gruz, Novák, & Strnad, 2008
F7/B7	42.93	300sh; 335	361c	n.d.	Hydroxycinnamic acid derivative		Gruz, Novák, & Strnad, 2008
F8&B8	44.97	267; 330sh	269a	269(100); 195(25); 133 (52)	Genistein	1	Vukics & Guttman, 2010

639 t_R: Retention time; sh:Shoulder; n.d.: no data.

640 Abundances of ions in the MS spectra: (a) abundance over 1×10^5 ; (b) abundance in the range 1×10^4 - 1×10^5 ; (c) abundance below 1×10^4 ; n.d.: not

641 detected.

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642 Identification confidence level: Level 1, confirmed structures where a reference standard is available; level 2a, evidence by spectrum matching

with a spectrum from the literature; level 2b, diagnostic evidence where no other structure fits the experimental MS² information (La Barbera et
 al., 2017).

645 Table 2. Method validation parameters of the seven selected standards and their recovery in lupin (Coromup, ER) seed coat using the HPLC-

646 DAD.

Compound	Linear range (mg/L)	Regression equation	R ²	LOQ (µg/L)	LOD (µg/L)	Recovery (%)	RSD (%) of recovery	RSD (%) of intra- day (n=6)	RSD (%) of inter- day (n=6)
Protocatechuic acid	0.32-8.00	y=58.989x-3.4128	0.9994	35	119	97.94	1.80	0.37	0.52
Caffeic acid	0.29-7.28	y=122.62x-57.419	0.9989	9	30	100.76	4.43	0.10	1.06
Vitexin	0.80-20.00	y=47.05x-0.4737	1	30	100	97.72	4.25	0.37	0.46
Ferulic acid	0.26-6.40	y=146.81x+2.1822	1	11	36	97.61	0.75	0.43	0.56
Taxifolin	0.17-4.20	y=78.267x-23.037	0.9983	19	65	104.38	0.84	0.48	2.74
trans-Cinnamic acid	0.19-4.80	y=358.73+2.3522	1	1	4	99.32	2.30	0.11	0.34
Genistein	0.12-3.00	y=105.96x+1.5452	0.9997	4	13	97.89	4.42	0.23	0.37

647

			PBA Jurien	Coromup	PBA Gunyidi	Mandelup	PBA Barlock	Jenabillup
Hydroxycinnamics								
F1	CIA^2	ER	$19.51{\pm}1.69^{aA}$	$31.14{\pm}0.54^{dA}$	28.15 ± 0.52^{cdA}	27.12 ± 0.38^{bcA}	$24.32{\pm}0.82^{bA}$	$31.40{\pm}0.09^{dA}$
		WH	17.10±0.11 ^{aA}	$27.83{\pm}0.77^{bcB}$	$26.88{\pm}0.01^{bcA}$	$26.49{\pm}1.44^{bA}$	$25.63{\pm}0.51^{bA}$	29.68 ± 0.35^{cB}
F6	diCQA ³	ER	$67.88{\pm}2.60^{abA}$	$89.03{\pm}15.01^{bA}$	$71.50{\pm}2.69^{abA}$	$57.35{\pm}0.48^{aA}$	$63.18{\pm}0.51^{aA}$	$65.25{\pm}0.08^{abA}$
		WH	49.08 ± 0.20^{bB}	$55.52{\pm}0.35^{cA}$	$42.31{\pm}0.02^{aB}$	$43.62{\pm}1.63^{aB}$	$46.56{\pm}2.66^{abB}$	$46.90{\pm}0.02^{abB}$
Σ Su	bgroup	ER	87.40±0.91 ^{aA}	120.18 ± 15.55^{bA}	99.65±3.21 ^{abA}	$84.47{\pm}0.85^{aA}$	$87.50{\pm}1.32^{aA}$	$96.65{\pm}0.01^{abA}$
		WH	66.18±0.09 ^{aB}	83.35±1.13 ^{cA}	$69.19{\pm}0.03^{abB}$	$70.11 {\pm} 3.07^{abB}$	72.19 ± 3.17^{abB}	76.58 ± 0.32^{bcB}
Flav	ones							
F3	Api-Apif-di-Glcp ⁴	ER	717.28±2.22 ^{aA}	960.32±6.33 ^{cA}	$764.60{\pm}3.47^{aA}$	958.7±14.76 ^{cA}	883.18 ± 19.52^{bA}	900.52 ± 16.15^{bcA}
		WH	$697.85{\pm}10.96^{aA}$	$954.85{\pm}7.87^{cdA}$	804.18 ± 6.21^{bB}	977.00 ± 9.73^{deA}	1011.82 ± 8.25^{eB}	928.95±18.85 ^{cA}
F4	Vicenin 2	ER	$36.02{\pm}1.81^{bA}$	$25.70{\pm}0.06^{aA}$	$26.38{\pm}0.14^{aA}$	$33.68{\pm}0.92^{bA}$	$44.02{\pm}1.17^{dA}$	$28.48{\pm}0.05^{aA}$
		WH	34.51 ± 0.53^{bA}	$24.65{\pm}0.20^{aB}$	$25.78{\pm}0.28^{aA}$	$33.84{\pm}0.06^{bA}$	59.53 ± 1.05^{cB}	32.62 ± 2.35^{bA}
F5	Api-O-Glcp ⁵	ER	80.21 ± 8.62^{abcA}	$74.80{\pm}7.51^{abcA}$	81.96 ± 17.59^{bcA}	$46.50{\pm}1.20^{aA}$	91.67±0.61 ^{cA}	57.21 ± 0.59^{abA}
		WH	$40.76{\pm}2.27^{aB}$	$32.54{\pm}11.15^{aB}$	$22.42{\pm}0.03^{aB}$	$23.65{\pm}2.31^{aB}$	$48.02{\pm}11.77^{aB}$	$36.36{\pm}0.21^{aB}$
Σ Su	Σ Subgroup		$833.51 {\pm} 9.04^{aA}$	1060.82±13.9cA	872.94±21.21 ^{aA}	$1038.88 {\pm} 16.88^{bcA}$	1018.86 ± 21.3^{bcA}	986.21 ± 15.51^{bA}
		WH	$773.12{\pm}13.76^{aB}$	1012.04 ± 3.08^{cB}	$852.38{\pm}6.46^{bA}$	1034.49±12.11 ^{cA}	1119.37 ± 2.47^{dB}	$997.93{\pm}20.99^{cA}$
Isofl	avones							
F8	Genistein	ER	$24.89{\pm}0.39^{abA}$	$25.01{\pm}0.72^{bA}$	30.79 ± 0.25^{cA}	$22.30{\pm}1.35^{aA}$	$34.01{\pm}0.17^{dA}$	$25.82{\pm}0.36^{bA}$
		WH	41.79 ± 4.22^{abB}	$47.92{\pm}6.03^{abcB}$	$32.53{\pm}0.45^{\mathrm{aB}}$	$33.13{\pm}2.19^{aB}$	51.00 ± 5.59^{bcA}	62.60 ± 5.19^{cB}
Dihy	Dihydroflavonols							
F2	Aro-Glcp-ApifGlcp ⁶	ER	$35.83{\pm}1.08^{aA}$	$56.57{\pm}0.16^{dA}$	$42.88{\pm}0.01^{bA}$	$40.92{\pm}0.54^{bA}$	50.87 ± 1.05^{cA}	61.03 ± 0.74^{eA}
		WH	$37.52{\pm}0.75^{aA}$	$38.83{\pm}2.91^{aB}$	$34.20{\pm}0.13^{aB}$	$40.93{\pm}3.65^{aA}$	$48.29{\pm}0.14^{aA}$	$39.01{\pm}13.8^{aA}$
Tota	1	ER	$986.63{\pm}15.55^{aA}$	1262.58±28.89cA	1046.26±24.66ªA	1186.58±16.93 ^{bcA}	1191.25±23.85 ^{bcA}	1169.71 ± 15.89^{bA}
		WH	$918.6{\pm}10.38^{aB}$	1182.14 ± 5.08^{cA}	$988.3{\pm}6.76^{bA}$	1178.66±3.19cA	$1290.84{\pm}6.44^{dB}$	1176.12±1.68 ^{cA}

648 Table 3. Individual polyphenol profile ($\mu g/g dry basis$) in free fraction of Australian sweet lupin seed coats¹

- 649 ¹ Means \pm standard deviation (n=2).
- 650 ² cinnamic acid glucoside (as *trans*-cinnamic acid equivalent); ³ dicaffeoylquinic acid (as caffeic acid equivalent); ⁴ apigenin-7-*O*-β-apiofuranosyl-
- 651 6,8-di-C-glucopyranoside (as vitexin equivalent); ⁵ apigenin-7-O-β-glucopyranoside (as vitexin equivalent); ⁶ aromadendrin-6-C-β-D-
- 652 glucopyranosyl-7-*O*-[β-D-apiofuranosyl- $(1\rightarrow 2)$]-*O*-β-D-glucopyranoside glucopyranoside (as taxifolin equivalent).
- 653 ER, Eradu; WH, Wongan Hills.
- 654 Means assigned with different small letters in the same row, and capital letters in the same column within each dependent variable indicate
- 655 significant differences (P < 0.05).

			PBA Jurien	Coromup	PBA Gunyidi	Mandelup	PBA Barlock	Jenabillup
B2	Protocatechuic acid	ER	$6.46{\pm}1.02^{aA}$	8.12±0.27 ^{aA}	29.13 ± 3.28^{bA}	$34.72{\pm}0.22^{bA}$	51.45±4.96 ^{cA}	40.03±4.03 ^{bcA}
		WH	$5.68{\pm}0.42^{\mathrm{aA}}$	$13.08{\pm}2.37^{aA}$	$58.21{\pm}1.87^{bcB}$	$60.03{\pm}1.42^{bcB}$	63.38±2.03 ^{cA}	$44.96{\pm}0.50^{bA}$
B3	<i>p</i> -coumaric acid glucoside ²	ER	$5.31{\pm}0.84^{abA}$	$4.88{\pm}0.97^{abA}$	$6.19{\pm}0.33^{abA}$	$5.39{\pm}0.11^{abA}$	$7.09{\pm}0.47^{\mathrm{bA}}$	$4.00{\pm}0.48^{aA}$
		WH	$3.80{\pm}0.59^{\mathrm{aA}}$	$3.13{\pm}0.23^{aA}$	$6.40{\pm}0.18^{bA}$	$5.67{\pm}0.02^{bA}$	$6.02{\pm}0.27^{bA}$	$3.91{\pm}0.18^{aA}$
B4	Ferulic acid glucoside ³	ER	$7.36{\pm}1.10^{aA}$	$9.41{\pm}0.55^{aA}$	$8.32{\pm}0.11^{aA}$	$8.26{\pm}0.14^{aA}$	$8.24{\pm}0.31^{aA}$	$7.30{\pm}0.54^{aA}$
		WH	$6.30{\pm}0.63^{aA}$	$7.53{\pm}0.25^{abA}$	$8.75{\pm}0.51^{bA}$	$8.45{\pm}0.03^{bA}$	$7.91{\pm}0.26^{bA}$	$7.42{\pm}0.22^{abA}$
B6	Ferulic acid	ER	$5.83{\pm}0.22^{aA}$	$6.53{\pm}0.15^{abA}$	$7.51{\pm}0.26^{bA}$	$5.57{\pm}0.01^{aA}$	$6.20{\pm}0.11^{aA}$	$5.38{\pm}0.60^{aA}$
		WH	$5.63{\pm}0.21^{aA}$	$6.40{\pm}0.20^{aA}$	$9.15{\pm}0.48^{bA}$	$5.93{\pm}0.04^{aB}$	$6.32{\pm}0.03^{aA}$	$6.01{\pm}0.07^{aA}$
Б Нус	<i>L</i> Hydroxycinnamics		$24.96{\pm}3.18^{aA}$	$28.94{\pm}1.94^{abA}$	51.15 ± 3.98^{cA}	49.21 ± 7.14^{bcA}	$72.97{\pm}5.85^{dA}$	$56.70{\pm}5.65^{cdA}$
		WH	$21.41{\pm}1.84^{aA}$	$30.14{\pm}1.68^{aA}$	82.51 ± 2.08^{cB}	80.07 ± 1.47^{cB}	83.64 ± 1.54^{cA}	$62.30{\pm}0.18^{bA}$
B5	Apigenin-7-O-glucoside4	ER	$27.12{\pm}5.94^{aA}$	41.41 ± 9.42^{aA}	$68.33{\pm}5.45^{bA}$	$40.03{\pm}2.87^{aA}$	$51.6{\pm}1.27^{abA}$	$49.32{\pm}5.51^{abA}$
		WH	24.17 ± 2.76^{aA}	47.42 ± 7.23^{bA}	83.14±4.11 ^{cA}	$40.11{\pm}0.25^{bA}$	$50.57{\pm}2.14^{bA}$	$54.51{\pm}2.79^{bA}$
B8	Genistein	ER	$4.16{\pm}0.06^{aA}$	$4.67{\pm}0.31^{aA}$	$5.85{\pm}0.11^{aA}$	$4.42{\pm}0.25^{aA}$	$5.19{\pm}0.17^{aA}$	$4.78{\pm}0.94^{\mathrm{aA}}$
		WH	$4.02{\pm}0.19^{aA}$	$4.88{\pm}0.21^{abA}$	$6.00{\pm}0.57^{bA}$	$4.07{\pm}0.28^{aA}$	$4.75{\pm}0.06^{aA}$	$4.89{\pm}0.01^{abA}$
Total		ER	$56.24{\pm}9.07^{aA}$	$75.03{\pm}11.05^{abA}$	125.33±9.32 ^{cA}	$98.40{\pm}3.08^{abcA}$	129.77±7.28 ^{cA}	116.04 ± 8.69^{bcA}
		WH	$49.60{\pm}4.79^{aA}$	82.43 ± 5.34^{bA}	171.65 ± 2.59^{dB}	124.26 ± 0.95^{cB}	138.96±3.74 ^{cA}	121.48 ± 2.96^{cA}

656 Table 4. Individual polyphenol profile ($\mu g/g$ dry basis) in bound fraction of Australian sweet lupin seed coats¹

657 ¹ Means ± SD (n=2); ² *p*-coumaric acid equivalent; ³ as ferulic acid equivalent; ⁴ apigenin-7-*O*-β-glucopyranoside (as vitexin equivalent).

658 ER, Eradu; WH, Wongan Hills.

659 Means assigned with different small letters in the same row, and capital letters in the same column within each dependent variable indicate

660 significant differences (P < 0.05).