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## **Nitrogen losses to the environment following food-based digestate and compost applications to agricultural land**

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1 Nitrogen losses to the environment following food-based digestate and  
2 compost applications to agricultural land

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13 **Abstract**

14 The anaerobic digestion of food waste for energy recovery produces a nutrient-rich digestate which  
15 is a valuable source of crop available nitrogen (N). As with any 'new' material being recycled to  
16 agricultural land it is important to develop best management practices that maximise crop available  
17 N supply, whilst minimising emissions to the environment. In this study, ammonia (NH<sub>3</sub>) and nitrous  
18 oxide (N<sub>2</sub>O) emissions to air and nitrate (NO<sub>3</sub><sup>-</sup>) leaching losses to water following digestate, compost  
19 and livestock manure applications to agricultural land were measured at 3 sites in England and  
20 Wales. Ammonia emissions were greater from applications of food-based digestate (c.40% of total N  
21 applied) than from livestock slurry (c.30% of total N applied) due to its higher ammonium-N content  
22 (mean 5.6 kg/t compared with 1-2 kg/t for slurry) and elevated pH (mean 8.3 compared with 7.7 for  
23 slurry). Whilst bandspreading was effective at reducing NH<sub>3</sub> emissions from slurry compared with  
24 surface broadcasting it was not found to be an effective mitigation option for food-based digestate  
25 in this study. The majority of the NH<sub>3</sub> losses occurred within 6 hours of spreading highlighting the  
26 importance of rapid soil incorporation as a method for reducing NH<sub>3</sub> emissions. Nitrous oxide losses  
27 from food-based digestates were low, with emission factors all less than the IPCC default value of 1%  
28 (mean 0.45 ± 0.15%). Overwinter NO<sub>3</sub><sup>-</sup> leaching losses from food-based digestate were similar to  
29 those from pig slurry, but much greater than from pig farmyard manure or compost. Both gaseous N  
30 losses and NO<sub>3</sub><sup>-</sup> leaching from green and green/food composts were low, indicating that in these  
31 terms compost can be considered as an 'environmentally benign' material. These findings have been

32 used in the development of best practice guidelines which provide a framework for the responsible  
33 use of digestates and composts in agriculture.

34

### 35 **Capsule**

36 Field measurements of nitrogen losses as ammonia, nitrous oxide and nitrate leaching have been  
37 used to develop best practice guidance for food-based digestate and compost use in agriculture.

38

39 **Keywords:** Emissions; nitrate leaching; ammonia; nitrous oxide; digestate

40

### 41 1. Introduction.

42 The United Kingdom generates around 14 million tonnes of food waste each year, the highest rate in  
43 the European Union, which in total generates nearly 90 million tonnes; the quantity produced by the  
44 different member states depends on numerous factors such as cultural practices, climate, diet and  
45 socio-economic conditions (EC, 2010). A large proportion of this waste is disposed of to landfill, with  
46 the UK sending around 8 million tonnes of biodegradable municipal waste (including food waste) to  
47 landfill every year (Defra, 2016). Redirecting this material away from landfill will significantly reduce  
48 greenhouse gas (GHG) emissions, in particular methane (CH<sub>4</sub>), which has a global warming potential  
49 around 25-fold greater than carbon dioxide (CO<sub>2</sub>), thereby contributing to GHG reduction targets. To  
50 this end, the EU Landfill Directive states that by 2020 the amount of biodegradable municipal waste  
51 disposed of in landfill sites must be reduced by 65%, compared with 1995 levels (EC, 1999).

52 As part of the UK's commitment to reduce GHG emissions and to meet EU renewable energy targets,  
53 policies and strategies have been implemented (DECC/Defra, 2011) to increase the treatment of  
54 food waste through anaerobic digestion (AD; Styles *et al.*, 2016) which, as well as providing  
55 renewable energy, generates a nutrient-rich digestate or 'biofertiliser'. The most recent estimates  
56 suggest that by 2020 around 5 Mt of the 7 Mt of food waste currently sent to landfill could be  
57 available for digestion annually in the UK (DECC/Defra, 2011). However, AD is not without its  
58 problems. In a recent study of the UK biogas sector, Styles *et al.* (2016) found that whilst biogas  
59 energy has a lower GHG intensity than fossil fuels, it can increase acidification and eutrophication  
60 burdens. Life-cycle assessment (LCA) studies have highlighted the fact that the environmental  
61 outcomes are very sensitive to factors such as feedstock type, fugitive emissions, biomethane use,  
62 energy conversion efficiency and digestate management. Nevertheless, Styles *et al.* (2016)

63 concluded that to maximise the potential for GHG abatement, climate change policies should  
64 encourage digestion of food waste whilst restricting digestion of crop inputs and wastes that could  
65 be used as animal feed.

66 The AD sector in the UK has been developing rapidly, with a 34% increase in the total number of  
67 operational AD plants between 2012 and 2013, and an increase of 51% in the tonnage of organic  
68 material processed (WRAP, 2014). Over 2 million tonnes of digestate were applied to agricultural  
69 land in the UK in 2013, supplying a total of 92 kt nitrogen (N). Although this is only half the total N  
70 supplied by composts, c.80% of the total N within food-based digestate is readily available (WRAP,  
71 2016) i.e. in the form of ammonium-N ( $\text{NH}_4\text{-N}$ ). A regular survey of the organics processing industry  
72 has been undertaken since the mid-1990s; in 2012 the survey indicated that there had been little  
73 year-on-year change in the UK composting sector, with a total of 3.5 million tonnes (fresh weight)  
74 produced in 2012, 68% of which was recycled to agricultural land, supplying 192 kt N.

75 When applying organic materials such as digestate and compost to land, it is essential that their  
76 application, agricultural or otherwise, is not harmful to the environment (i.e. to soil, water and air  
77 quality) or human health. The European Nitrogen Assessment (Sutton *et al.*, 2011) highlighted how  
78 the overall environmental costs of all N losses in Europe (estimated at €70–€320 billion per year at  
79 current rates) outweigh the direct economic benefits of N in agriculture, due largely to loss of air and  
80 water quality. The land application of organic materials therefore needs to be carefully managed to  
81 maximise their crop available nutrient value and minimise their impact on the wider environment.  
82 Food-based digestate is of particular interest being a new, less well understood material which will  
83 vary in its compositional characteristics depending on the type of food waste used to produce it (e.g.  
84 vegetable-based feed stocks produce digestate with lower N concentrations than those produced  
85 from mixed food wastes). It generally has a relatively high total N content of around  $5 \text{ kg/m}^3$  (WRAP,  
86 2016) compared with c. $3 \text{ kg/m}^3$  for cattle slurry and c. $4 \text{ kg/m}^3$  for pig slurry (Defra, 2010) and might  
87 therefore be expected to have the potential for greater N losses to the environment than livestock  
88 manures or manure-based digestates.

89 Ammonia ( $\text{NH}_3$ ) emissions to air contribute to acid deposition and can cause eutrophication of  
90 sensitive ecosystems; in addition,  $\text{NH}_3$  reacts with acids in the atmosphere to form particulate  
91 matter which may pose a threat to human health (Webb *et al.*, 2004). The amount and rate of  $\text{NH}_3$   
92 release following land spreading depends on a range of organic material (e.g. pH, readily available N,  
93 dry matter), spreading (e.g. application rate, method and timing), soil (e.g. moisture content) and  
94 environmental (e.g. temperature, wind speed, rainfall) factors (Nicholson *et al.*, 2013). There is an  
95 extensive body of research in the UK (and elsewhere) on  $\text{NH}_3$  emissions following land application of

96 livestock manures and slurries (see for example Pain *et al.*, 1989; Chambers *et al.*, 1997; Sommer *et al.*, 1997; Huijsmans *et al.*, 2001; Misselbrook *et al.*, 2002; Webb *et al.*, 2004), which has been used  
97 to populate the National Ammonia Emissions Inventory (Misselbrook *et al.*, 2015) and provide  
98 guidance for farmers to minimise NH<sub>3</sub> emissions from manures in the UK (Defra, 2009) and  
99 elsewhere. Research has also been undertaken using manure-based and crop-based digestates  
100 applied to land (e.g. Rubaek *et al.*, 1996; Wulf *et al.*, 2002); however, very little information is  
101 available for food-based digestates produced and applied under conditions pertinent to the UK. A  
102 recent study by Tiwary *et al.* (2015) showed that surface applied food-based digestate applications  
103 led to NH<sub>3</sub> losses of 35-65% of the total N applied in the week following application, with an  
104 abatement of 85% achieved if the material was incorporated into the soil immediately following  
105 application. However, this study was undertaken at a field site in India and used digestates that may  
106 not be comparable with those currently produced in the UK. Composts produced from green wastes  
107 such as grass clippings and hedge trimmings (green compost) or from a mixture of green and food  
108 wastes (green/food compost) tend to be applied to agricultural soils for soil conditioning purposes  
109 (WRAP, 2016), although they also contain valuable amounts of plant available nutrients. The low  
110 readily available N content of composts (generally <5% of total N; Defra, 2010) would suggest that  
111 NH<sub>3</sub> losses following land spreading are also likely to be low, although there is little evidence  
112 currently available to support this assertion.  
113

114 Nitrous oxide (N<sub>2</sub>O) is a greenhouse gas with a global warming potential c.300-fold greater than  
115 carbon dioxide (IPCC, 2006). The UK Greenhouse Gas Emissions Inventory (2014) estimated that  
116 c.70% of N<sub>2</sub>O produced in the UK comes from agriculture (Brown *et al.*, 2016), of which the majority  
117 (75%) is emitted from soils following N applications/returns (e.g. manufactured fertiliser N, crop  
118 residue incorporation organic materials and urine from grazing returns) to land. Around 17% of  
119 agricultural N<sub>2</sub>O is emitted indirectly from soils following re-deposition of emitted NH<sub>3</sub> and from  
120 leached nitrate (NO<sub>3</sub><sup>-</sup>) (Brown *et al.*, 2016). As with NH<sub>3</sub>, there is little information available on N<sub>2</sub>O  
121 losses following food-based digestate and compost applications to agricultural land. The current  
122 IPCC Tier 1 default emission factor (EF) for N<sub>2</sub>O losses from animal manure, compost, sewage sludge  
123 and other organic N additions (e.g. digestates) is 1% of the total N applied (IPCC, 2006). By way of  
124 comparison, Tiwary *et al.* (2015) found that N<sub>2</sub>O emissions from food-based digestate were 4-10% of  
125 the total N applied, which is much higher than the default IPCC EF of 1%, although these  
126 measurements were made in India under very different soil and climatic conditions from those in the  
127 UK. Measurements in Scotland showed that cumulative N<sub>2</sub>O emissions following green compost  
128 applied at 35, 100 and 200 t/ha ranged from 0.32 to 4.54 kg N<sub>2</sub>O-N/ha/yr, with the higher values  
129 measured following the 200 t/ha application in the wet spring of 2008 (Ball *et al.*, 2014). Although

130 compost application rates were very high in this experiment (35-200 t/ha compared with a more  
131 typical rate of 30 t/ha), the maximum N<sub>2</sub>O EFs were still only around 1% of the total N applied.

132 There is still much uncertainty over the factors which control N<sub>2</sub>O emissions from food-based  
133 digestates following application to agricultural land. For example, Pezzolla *et al.* (2012) found that  
134 applying food –based digestate to a UK grassland did not increase emissions compared to the  
135 untreated control, although measurements were made during an exceptionally dry growing season.  
136 In contrast, an incubation study showed much larger emissions from food-based digestate compared  
137 to ammonium sulphate applied under high soil moisture conditions (Koester *et al.*, 2011). A later  
138 study under similar conditions found emissions were twice as high from cattle slurry than from food-  
139 based digestate (Koester *et al.*, 2015). Following a laboratory incubation study using food-based and  
140 other digestates, Rigby & Smith (2013) concluded that “the significance and influence of the  
141 interaction between soil type and digestate stability and physical properties on denitrification  
142 processes in digestate-amended soils require urgent investigation.”

143 Additionally, poorly-managed applications of digestate and compost to land have the potential to  
144 impact on water quality as a result of NO<sub>3</sub><sup>-</sup> and other pollutant losses in drainage water. Indeed the  
145 1991 EU Nitrates Directive (91/676/EEC) aims to protect water quality across Europe by preventing  
146 NO<sub>3</sub><sup>-</sup> from agricultural sources from polluting ground and surface waters by promoting good farming  
147 practices. Again, there is a substantial body of evidence on NO<sub>3</sub><sup>-</sup> losses following manure application  
148 to agricultural land (e.g. Chambers *et al.*, 2000; Smith *et al.*, 2002), but very few field studies  
149 available for food-based digestates. Goberna *et al.* (2011) reported that digestate application to soil  
150 doubled NO<sub>3</sub><sup>-</sup> leaching compared with animal manure applied at an equivalent N rate (80 kg N/ha),  
151 however this was a pot-based study using manure-based digestate. Another pot experiment using  
152 manure-based digestate applied to grass showed that there was less potential for NO<sub>3</sub><sup>-</sup> leaching  
153 losses compared with undigested slurry or mineral fertiliser (Walsh *et al.*, 2012).

154 The study described in this paper was part of a wider programme of work to establish a robust  
155 scientific evidence base on the nutrient supply properties of digestate and compost applications to  
156 land in the UK (Bhogal *et al.*, 2016). The objective of the work described in this paper was to  
157 measure N losses following autumn and spring applications of food-based digestate, green/food  
158 compost and green compost to agricultural land via NH<sub>3</sub> volatilisation and N<sub>2</sub>O emissions to air, and  
159 NO<sub>3</sub><sup>-</sup> leaching to water, and to compare these losses with those from livestock manures. The effect  
160 of application techniques (i.e. surface broadcast compared with band spreading) were also assessed  
161 to establish the effectiveness of this potential mitigation technique. The findings will be used to  
162 develop best management practices that maximise crop nutrient utilisation, whilst minimising  
163 environmental emissions of NH<sub>3</sub>, N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>.

## 164 2. Materials and methods

### 165 2.1 Experimental sites

166 Experimental sites were established at 3 locations on a range of contrasting soil types and  
167 agroclimatic zones (Table 1). To characterise each site, representative topsoil samples (0-15cm at the  
168 arable site – Wensum - and 0-7.5cm at the grassland sites) were taken prior to the start of the  
169 experiment and analysed for pH, sand (%), clay (%), silt (%), total N and organic carbon (C) using  
170 standard methodologies (Anon., 1986), with the results shown in Table 1.

### 171 2.2 Treatments and design

172 The organic materials used were food-based digestate, compost, green or green/food compost,  
173 green compost, solid farmyard manure (FYM) and livestock slurry. All organic materials were surface  
174 broadcast, with additional treatments where food-based digestate and slurry were applied using a  
175 bandspreading technique.

176 At WE, the only site where leaching losses to water were quantified, treatments were applied in  
177 August 2011 (autumn), and then repeated in February 2012 (spring) on different experimental plots;  
178 at the other sites, only spring applications (March 2012 at NW and May 2012 at PW) were evaluated.  
179 For practical reasons, the spring and autumn treatments were grouped separately, with each group  
180 of plots having an untreated control. Each treatment was replicated three times and arranged in a  
181 randomised block design. Plot sizes were 3-7 m wide by 8-15m long.

182 Cattle FYM and slurries were used at NW and PW, and pig FYM and slurry at WE. The green compost  
183 (PW and NW) or green/food compost (WE) and food-based digestates were sourced from  
184 commercial enterprises local to the experimental sites, with the digestates from AD plants using  
185 mainly commercial and municipal food wastes as a feedstock. The liquid materials (livestock slurry  
186 and food-based digestate) were applied using a specially designed small plot applicator and the solid  
187 materials (FYM and compost) were applied by hand (at the NW site the liquid organic materials were  
188 also applied by hand). To be representative of commercial practice, at the grassland sites the  
189 bandspread liquid material applications were by trailing shoe with 20cm spacing between the bands  
190 and at WE, the bandspread applications were by trailing hose with 30cm spacing between the bands.  
191 At WE, the autumn applied broadcast pig slurry and food-based digestate were incorporated into  
192 the soil within 6 hours of application, using a rotavator to comply with Nitrate Vulnerable Zone (NVZ)  
193 rules (SI, 2008; WSI, 2008).

194 The amount of organic material applied to each plot was weighed and recorded, so that the  
195 application rate to each plot could be accurately determined, with the aim of applying the materials

196 at typical agronomic rates i.e. 20-30 m<sup>3</sup>/ha digestate, 30-40 m<sup>3</sup>/ha livestock slurry, 20 t/ha compost  
197 and 25-30 t/ha FYM. Organic material application rates and N loadings are shown in Table 2.

### 198 2.3 Organic material analysis

199 At each site and application timing, a representative sample of each organic material type from each  
200 experimental block was taken (c.2 litres or 2 kg per block), giving three replicate samples of each  
201 material per site. These samples were analysed for pH, dry matter (DM), total N and NH<sub>4</sub>-N using  
202 standard methodologies (Anon., 1986).

### 203 2.4 Crop management

204 The grass at PW and NW was permanent pasture; the wheat crop at WE was grown according to  
205 best farm practice using commercially recommended seed rates, with crop protection products  
206 applied as needed and according to good agricultural practice to control weeds, pests and diseases  
207 equally across all treatments. No manufactured N fertiliser was applied to the organic material  
208 treatments, but all treatments, including the untreated control, had manufactured fertilisers (P, K  
209 and S) applied based on the requirements of the untreated control (Defra, 2010).

### 210 2.5 Ammonia emissions

211 Wind tunnels were used to assess NH<sub>3</sub> emissions from the livestock manure treatments at each site,  
212 based on the design developed by Lockyer (1984) and as recently used by Tiwary *et al.* (2015) and  
213 Misselbrook *et al.* (2014). Each wind tunnel consisted of two parts; a transparent polycarbonate  
214 canopy (2.0 m x 0.5 m) which covered the plot area, and a stainless steel duct housing a fan which  
215 drew air through the canopy at a speed of 1 m/s; an anemometer measuring the wind speed, which  
216 was recorded using a pulse counter. A sub-sample of the air entering and leaving the tunnel was  
217 drawn through absorption flasks containing 80 ml of 0.02 M orthophosphoric acid. The absorption  
218 flasks were changed after 24 hours and then daily for 7 days. The loss of NH<sub>3</sub> from beneath each  
219 tunnel was calculated as the product of air flow through the tunnel and the difference between the  
220 concentrations of NH<sub>3</sub> in the air entering (i.e. the background NH<sub>3</sub> concentration) and leaving the tunnel  
221 as follows:

$$222 \quad \text{NH}_3 \text{ loss} = ((b/a)_{\text{outlet}} - (b/a)_{\text{inlet}}) * c \quad (1)$$

223 where, a is the volume of air sampled by each acid trap, b is the quantity of NH<sub>3</sub>-N in each trap over  
224 the sampling period and c is the volume of air drawn through each tunnel. The rate of loss was  
225 calculated over each time period so that the pattern of loss could be quantified and cumulative  
226 losses were then calculated by summing over all sampling periods.



227 2.6 Nitrous oxide emissions

228 Nitrous oxide emissions were measured using the static chamber technique (Chadwick *et al.*, 2014),  
229 from three replicate plots per treatment, using 5 chambers per plot (giving a total of 15 replicate  
230 chambers per treatment). Each chamber had dimensions of 40 cm x 40 cm square and was 25 cm  
231 tall, giving a soil surface area coverage of 0.16 m<sup>2</sup>. The chambers were installed immediately after  
232 organic material application and positioned in a 5 cm deep slot cut in the soil. The chambers were  
233 designed to completely enclose growing arable crops and grassland, without damage, with chamber  
234 extensions fitted to enable measurements to be taken from mature cereal and grass crops. On each  
235 sampling occasion, the chambers were covered for at least 40 minutes before the headspace was  
236 sampled. Sampling was normally conducted between 10 am and 2 pm, as previous studies have  
237 shown that emissions at this time of day approximate to average diurnal emission rate (Clayton *et al.*,  
238 1994). The samples were transferred to evacuated vials prior to Gas Chromatography analysis  
239 using an Electron Capture Detector. To verify the assumption of linear gas accumulation within a  
240 chamber's headspace, 3 chambers were selected on each sampling occasion from which a time  
241 series of headspace samples was taken every 15 minutes up to 60 minutes after closure. The  
242 following steps were taken to help ensure that linearity in gas accumulation was achieved, by (i)  
243 ensuring an air-tight seal between the chamber and soil; (ii) ensuring an air-tight seal between the  
244 chamber and lid; (iii) using 'large' chambers to provide as much headspace as practically possible,  
245 whilst retaining analytical sensitivity.

246 Data from previous UK studies have indicated that c.75% of total *direct* N<sub>2</sub>O emissions are likely to  
247 occur in the first 4-6 weeks following slurry application. Therefore the sampling strategy was  
248 weighted accordingly, with c.50% of sampling events carried out during the (likely) period of highest  
249 N<sub>2</sub>O fluxes (i.e. 4-6 weeks after application), giving a total of at least 30 measurements over a 12  
250 month period.

251 2.7 N leaching losses

252 At WE, N leaching losses to water were measured from the plots receiving autumn applications of  
253 organic materials, using Teflon cup water samplers. Five water samplers were installed on each plot  
254 to a depth of 90 cm. Samples of soil water were collected every 2 weeks or after 50 mm drainage,  
255 whichever occurred sooner, throughout the drainage period and analysed for NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub>-N using  
256 standard methodologies (Anon, 1986). Drainage estimates were obtained using IRRIGUIDE (Bailey &  
257 Spackman, 1996) and were combined with the pollutant concentration data to calculate N losses in  
258 drainage water.

259

## 260 2.8 Other measurements

261 Soil samples for mineral nitrogen (SMN) determination were taken periodically (from 0-10 cm depth)  
262 throughout the experimental period to quantify changes in soil mineral N following the organic  
263 material applications that could influence gaseous N losses. Soil moisture content measurements  
264 (per block) were also made on each N<sub>2</sub>O sampling occasion, and converted to water filled pore space  
265 (WFPS) using the following formula:

$$266 \quad \text{WFPS (\%)} = \text{Soil moisture (\% w/w)} \times \text{BD} / (1 - (\text{BD} / 2.65)) \quad (2)$$

267 Where BD is the bulk density of the studied soil (g/cm<sup>3</sup>) and 2.65 g/cm<sup>3</sup> is the bulk density of mineral  
268 soils.

269 Daily rainfall and mean air and soil temperature (at 5cm depth) data were measured at each site or  
270 obtained from a nearby meteorological station.

## 271 2.9 Statistical analysis

272 At each experimental site, conventional analysis of variance (ANOVA) comparisons were undertaken  
273 between the different treatments in terms of cumulative N<sub>2</sub>O and NH<sub>3</sub> emissions to air, and NO<sub>3</sub><sup>-</sup> and  
274 NH<sub>4</sub>-N losses to water, with comparison of P statistics (quoted in the text). A separate ANOVA was  
275 carried out at each site, after which *post-hoc* testing was undertaken to evaluate which treatment  
276 means were different from each other using a Duncan's multiple range test (using Genstat version  
277 12; VSN International Ltd, 2010).

278 Additionally, the pooled data were analysed to assess effects of N<sub>2</sub>O and NH<sub>3</sub> emissions to air (*i.e.* a  
279 cross-site analysis ANOVA was undertaken). This enabled us to assess whether a particular  
280 treatment had had a statistically significant effect at each site, and across *all* study sites. The ANOVA  
281 also indicated if there were significant interactions between sites and treatments.

## 282 3. Results

### 283 3.1 Organic material analysis

284 The composition of the organic materials applied at each site and application timing is shown in  
285 Table 2. As expected the food-based digestate had higher total N contents (mean 7.0 kg/t across all 4  
286 site/timings) than the pig slurry (mean 2.8 kg/t at WE) or cattle slurry (mean 2.4 kg/t at NW and PW).  
287 The NH<sub>4</sub>-N contents of the food-based digestate and pig slurry (*c.*80% of total N) were higher than

288 the cattle slurry (mean 54% of total N), with the solid materials containing <10% of the total N in the  
289  $\text{NH}_4\text{-N}$  form.

### 290 3.2 Ammonia emissions

291 Ammonia emissions from the spring 2012 applications at NW were similar from the food-based  
292 digestate and cattle slurry (23-31% total N applied), and both these treatments had higher emissions  
293 than the green compost and cattle FYM (<5% total N applied;  $P<0.05$ ). Perhaps surprisingly, there  
294 were no differences ( $P>0.05$ ) in  $\text{NH}_3$  emissions between the bandspread and broadcast food-based  
295 digestate and cattle slurry treatments.

296 At PW,  $\text{NH}_3$  emissions following the spring 2012 organic material applications were higher from  
297 food-based digestate (48-63% total N applied) than from cattle slurry (12-37% total N applied), with  
298 the green compost and cattle FYM treatments having much lower emissions (<3% total N applied;  
299  $P<0.05$ ). Bandspreading reduced  $\text{NH}_3$  emissions from the food-based digestate by c.25% (although  
300 this was not significant;  $P>0.05$ ) and cattle slurry by 70% ( $P<0.05$ ) compared with the respective  
301 broadcast applications.

302 At WE,  $\text{NH}_3$  emissions following the autumn 2011 organic material applications were again greater  
303 from food-based digestate (56-60% total N applied) than from pig slurry (c.40% total N applied), with  
304 the pig FYM and green/food compost having the lowest emissions (4-14% total N applied;  $P<0.05$ ).  
305 There were no significant differences in  $\text{NH}_3$  emissions between the bandspread and broadcast (soil  
306 incorporation within 24 hours) food-based digestate and pig slurry applications probably because  
307 the bandspread digestate and slurry stayed on the soil surface due to the dry (hydrophobic) soil  
308 conditions.  $\text{NH}_3$  emissions were lower following the spring 2012 food-based digestate and pig slurry  
309 applications than in autumn because the soil was moist leading to more rapid soil infiltration  
310 compared with the dry, hydrophobic soil surface of the arable stubble in autumn which restricted  
311 infiltration. Ammonia emissions were similar from the food-based digestate and pig slurry  
312 treatments (c.20% total N applied), with the pig FYM and green/food compost again having lower  
313 emissions (c.2% total N applied;  $P<0.05$ ). There were no significant differences between the  
314 bandspread and broadcast food-based digestate and pig slurry treatments, which was most probably  
315 due to the bandspread applications not staying in a band (plus some temporary ponding on the soil  
316 surface), followed by 'rapid' infiltration into the soil on both treatments.

317 The cross-site analysis of the total  $\text{NH}_3$  losses from the autumn 2011 applications at WE, and the  
318 spring 2012 organic material applications at NW, PW and WE is summarised in Table 3. Ammonia  
319 emissions were similar on the broadcast and bandspread food-based digestate (c.40% of total N

320 applied) and were greater than those from the broadcast and bandspread livestock slurry, with FYM  
321 and compost having the lowest emissions. There was no difference in emissions due to spreading  
322 method for the food-based digestate, but broadcast slurry gave rise to greater emissions than  
323 bandspread slurry. Total NH<sub>3</sub> losses from food-based digestate were of a similar magnitude to the  
324 35-65% of total N applied reported by Tiwary *et al.* (2015).

325 The cross-site NH<sub>3</sub> emission curves (Figure 1) indicate that the majority of the NH<sub>3</sub> losses from food-  
326 based digestate and livestock slurry occurred within 6 hours of spreading.

### 327 3.3 Nitrous oxide emissions

328 Nitrous oxide fluxes at NW in spring 2012 were generally low throughout the measurement period  
329 at <20 g N<sub>2</sub>O-N/ha/day on all treatments (Figure 2a). There was a small peak in emissions (*c.*30 g  
330 N<sub>2</sub>O-N/ha/day) at the end of May which corresponded with a drop of *c.*10% in the WFPS. Cumulative  
331 N<sub>2</sub>O losses (net of the control) were all low (<0.5 kg N<sub>2</sub>O/ha) and there were no significant  
332 differences in EFs between the different treatments (Figure 2b).

333 At PW in spring 2012, N<sub>2</sub>O fluxes were generally higher than at NW. Emissions peaked (particularly  
334 on the FYM treatment at *c.*110 g N<sub>2</sub>O-N/ha/day) around 2 weeks after the organic materials were  
335 applied in early May; after this, emissions on all the organic material treatments returned to  
336 background levels (*c.*10 g N<sub>2</sub>O-N/ha/day; Figure 3a). As at NW, cumulative N<sub>2</sub>O losses (net of the  
337 control) were all low (<0.5 kg N<sub>2</sub>O/ha). Emissions from the green compost treatment were  
338 significantly lower than from the cattle slurry and FYM treatments (*P*<0.05). Although bandspreading  
339 significantly increased the EF from cattle slurry compared with surface broadcasting (*P*<0.05), there  
340 was no difference in the EF between the broadcast and bandspread digestates. However, the  
341 bandspread digestate had a significantly (*P*<0.05) lower EF than the bandspread cattle slurry (Figure  
342 3b).

343 Nitrous oxide fluxes at WE in autumn peaked at *c.*100 g N<sub>2</sub>O-N/ha/day, shortly after the organic  
344 materials were applied in early August 2011; emissions on all the organic material treatments had  
345 returned to background levels (*c.*10 g N<sub>2</sub>O-N/ha/day) by the end of November 2011 (Figure 4a). Net  
346 cumulative N<sub>2</sub>O losses ranged from 0 kg N<sub>2</sub>O/ha on the compost treatment to 1.2 kg/ha on  
347 broadcast digestate treatment. Although the EF for green/food compost was very low, and  
348 bandspreading slurry and digestate resulted in numerical reductions in the EF, none of the treatment  
349 effects were significant (Figure 4b).

350 At WE in spring, N<sub>2</sub>O fluxes peaked at *c.*70 g N<sub>2</sub>O-N/ha/day in March about 1 month after the organic  
351 materials were applied in late February 2012; emissions on all the organic material treatments had

352 returned to background levels (c.10 g N<sub>2</sub>O-N/ha/day) by the end of April 2012 (Figure 5a). Net  
353 cumulative N<sub>2</sub>O losses ranged from <0 kg N<sub>2</sub>O/ha on the compost treatment to 1.8 kg/ha on the  
354 bandspread digestate treatment. Emissions from the green compost were significantly lower than  
355 from the digestate treatments and the bandspread pig slurry ( $P<0.05$ ). There was no effect of  
356 bandspreading on the EF for digestate or pig slurry compared with surface broadcasting (Figure 5b).  
357 Overall, there was no significant effect of food-based digestate/livestock slurry application method  
358 (i.e. surface broadcast compared with bandspreading) on N<sub>2</sub>O emissions (Table 3).

### 359 3.3.1 Leaching losses

360 Drainage volumes at WE over the winter of 2011-12 were low (92 mm) due to the lower than  
361 average over-winter rainfall of c.200mm compared to the 25 year average of just over 300 mm.  
362 Nitrate concentrations in the drainage water were c.50 mg/l on all treatments at the start of  
363 drainage in November/December 2011; concentrations peaked in January/February 2012 and were  
364 highest (c.150 mg/l) on the surface broadcast food-based digestate treatment (Supplementary  
365 Information Figure S1a). The peak in leachate NO<sub>3</sub><sup>-</sup> concentrations occurred somewhat later than the  
366 maximum soil NO<sub>3</sub><sup>-</sup> concentrations which were measured in September 2011, shortly after the  
367 organic materials were applied (Supplementary Information Figure S1b). This was probably because  
368 the NO<sub>3</sub><sup>-</sup> present in the top 15cm of the soil took this time to move through the soil profile to 30cm,  
369 which was the depth at which the leachate was sampled; additional NO<sub>3</sub><sup>-</sup> would also be mineralised  
370 over the autumn period from the organic N applied with the organic materials, contributing to NO<sub>3</sub><sup>-</sup>  
371 leaching losses.

372 Cumulative NO<sub>3</sub><sup>-</sup> leaching losses following the food-based digestate and pig slurry treatments were  
373 greater ( $P<0.05$ ) than from the pig FYM and compost treatments, with no significant differences  
374 between bandspread and broadcast food-based digestate treatments (Figure 6). Ammonium-N  
375 concentrations in the drainage waters were very low on all treatments (<0.05 mg/l) and cumulative  
376 leaching losses were <0.02 kg/ha (i.e. <0.01% of the total N applied).

377 The winter wheat crop (drilled in early October) was only expected to take up 5-10 kg N/ha during  
378 the autumn growth phase which effectively ceases by the end of October (Nicholson *et al.*, 2013).  
379 This is equivalent to <10% of the total N applied with the organic materials and hence would have  
380 had little effect on the quantities of N leached over the winter period from the different organic  
381 materials.

## 382 4. Discussion

### 383 4.1 Ammonia emissions

384 The higher NH<sub>3</sub> emissions from the food-based digestate than from livestock slurry ( $P<0.05$ ) were  
385 most probably due to the greater NH<sub>4</sub>-N content of the food-based digestate (mean 5.6 kg/t)  
386 compared with the livestock slurries (mean 2.2 kg/t for pig slurry and 1.3 kg/t for cattle slurry), Table  
387 2. Additionally, the mean pH of the food-based digestate was 8.5 compared with 7.8 for pig slurry  
388 and 7.6 for cattle slurry (Table 2). It is known that pH values greater than 8 are particularly conducive  
389 to elevated NH<sub>3</sub> emissions from digestates (e.g. Hoeksma *et al.*, 2012) and digestion of livestock  
390 slurry has been shown to increase pH with a concomitant increase in NH<sub>3</sub> volatilisation (Sommer *et*  
391 *al.*, 2006). Acidification (i.e. decreasing the pH) has been adopted as the Best Available Technology  
392 (BAT) for reducing NH<sub>3</sub> losses from livestock slurry in some European countries (Kai *et al.*, 2008),  
393 however further research into the costs, practicalities and effectiveness of acidification of food-  
394 based digestates as a method of controlling NH<sub>3</sub> emissions is still required.

395 Overall, bandspreading was effective at reducing NH<sub>3</sub> emissions from livestock slurry compared with  
396 surface broadcasting as reported in many previous studies (e.g. Sommer *et al.*, 1997; Webb *et al.*,  
397 2005), because the bandspreading technique reduces the surface area of slurry exposed to the  
398 atmosphere from which NH<sub>3</sub> can be volatilised. However in this study bandspreading was not found  
399 to be an effective mitigation option for food-based digestate. Bandspreading of liquid organic  
400 materials (such as food-based digestate) is now a common practice, with the majority of contractor-  
401 spread digestate applied using bandspreaders. In this study, the failure to observe a reduction in NH<sub>3</sub>  
402 emissions when bandspreading food-based digestate (in comparison with surface broadcast  
403 applications) was most probably due to soil and/or organic material properties that meant that the  
404 digestate did not rapidly infiltrate into the soil or did not stay in a narrow band on the soil surface.  
405 Dry matter content is known to affect NH<sub>3</sub> emissions from cattle slurry, with emissions increasing as  
406 slurry dry matter content increases (e.g. Sogaard *et al.*, 2002; Misselbrook *et al.*, 2004); it is likely  
407 that this relationship will also hold when comparing food-based digestates with different dry matter  
408 contents, although we do not know of any research data specific to digestates. The importance of  
409 crop height as a factor affecting NH<sub>3</sub> emissions is not clear; previous UK/Danish research has shown  
410 that the abatement efficiency of bandspread slurry applications increased with grass height and was  
411 typically 60% (Thorman *et al.*, 2008). In contrast, more recent Irish data (Lalor *et al.*, 2012) has not  
412 supported a relationship between NH<sub>3</sub> emission reductions from bandspread slurry applications and  
413 grass height. There is a requirement for further research on this topic to enable us to better assess  
414 the importance of crop height in controlling NH<sub>3</sub> losses from food-based digestates. It is important to

415 bear in mind that bandspreading technologies provide numerous other advantages over broadcast  
416 applications (e.g. more even digestate application and hence more accurate assessment of  
417 application rates, the ability to apply from tramlines, reduced odour and crop damage, and a cleaner  
418 sward) implying that it is still the best application method available.

419 In line with findings from previous studies using livestock slurry (Nicholson *et al.*, 2013), Figure 1  
420 indicates that the majority of the NH<sub>3</sub> losses from food-based digestate occurred within 6 hours of  
421 spreading. In contrast, Tiwary *et al.* (2015) found that NH<sub>3</sub> losses from food based digestates  
422 continued for about 2 days following spreading (although NH<sub>3</sub> emission patterns in tropical climates  
423 are likely to be different from those in temperate conditions), but that incorporation immediately  
424 following application reduced NH<sub>3</sub> losses by 85%. These findings highlight the importance of rapid  
425 soil incorporation as an alternative method for preventing N losses via this pathway on arable  
426 (cultivated) soils. Rapid soil incorporation has previously been shown to reduce NH<sub>3</sub> losses following  
427 slurry applications (e.g. Sommer & Hutchins, 2001; Webb *et al.*, 2005), although UK survey results  
428 have shown that a significant proportion of cattle and pig slurry (c.65%) applied to tillage fields is  
429 either not incorporated, or only incorporated more than 24 hours after spreading (Benford, 2016).  
430 This study has demonstrated that rapid incorporation could have the same mitigation potential  
431 when used with food-based digestate. However, the technique has also been shown to increase N<sub>2</sub>O  
432 emissions following slurry application as (Thorman *et al.*, 2011) suggesting that the potential for so  
433 called 'pollution swapping' could also exist when it is used with digestates.

434 In addition to representing the loss of a valuable resource, NH<sub>3</sub> emissions from digestate applications  
435 present a challenge to the UK meeting EU directives on NH<sub>3</sub> emissions. Under the EU National  
436 Emissions Ceiling Directive, the UK has a proposed target to reduce ammonia emissions by 8%  
437 (relative to a 2005 baseline) between 2020 and 2029, and by 21% from 2030. The UK Ammonia  
438 Emissions Inventory (UKAEI) includes emissions from food and crop-based digestates (but in the non-  
439 agricultural Inventory), although better data are needed to improve the current estimates. Based on  
440 the emission factors measured in this study (i.e. 38% for bandspread food-based digestate; Table 3)  
441 and the estimated 1.4 million m<sup>3</sup> of food-based digestate currently applied to agricultural land  
442 (WRAP, 2014) with an average total N content of 5 kg/m<sup>3</sup>, this implies that food-based digestate will  
443 emit an additional 3.3 kt of NH<sub>3</sub>, equivalent to 1% of the UK emission target (297 kt for 2010). If this  
444 is scaled up to the predicted 2020 production volumes (4.25 million m<sup>3</sup>), then food-based digestate  
445 would emit an additional 9.8 kt of NH<sub>3</sub> or 3.3% of the UK 2010 target. However, these estimates do  
446 not account for reductions in ammonia emissions from manufactured (mineral) N fertiliser use.

447 4.2 Nitrous oxide emissions

448 Nitrous oxide emissions from agricultural soil are predominately produced via the microbially  
449 mediated processes of nitrification and denitrification (Firestone & Davidson, 1989). The factors  
450 which control the magnitude of N<sub>2</sub>O emission include soil mineral nitrogen (SMN) content, soil  
451 temperature, soil moisture content (Dobbie & Smith, 2001; 2003) and available carbon (Weier *et al.*,  
452 1993). Davidson (1991) showed that there is an optimum production of N<sub>2</sub>O, which occurs at a  
453 water-filled pore space (WFPS) roughly equal to a transition point below which N<sub>2</sub>O is predominantly  
454 emitted from the aerobic process of nitrification and above which N<sub>2</sub>O is predominantly emitted  
455 from the anaerobic process of denitrification. Davidson (1991) suggested that this transition occurs  
456 at a WFPS of 60%; other studies, however, have shown that the position of the maximum emission  
457 can vary with soil type and conditions. Notably, UK studies have indicated that the highest N<sub>2</sub>O  
458 emissions frequently occur as a result of the anaerobic process of denitrification i.e. at a WFPS >60%  
459 (e.g. Dobbie & Smith, 2001; 2003).

460 At all the 3 experimental sites, most N<sub>2</sub>O emissions occurred in the few weeks following the organic  
461 materials being spread to land, and had generally returned to background levels within c.2 months  
462 (Figures 2a, 3a, 4a and 5a). Tiwary *et al.* (2015) also reported that N<sub>2</sub>O emission rates were “feeble”  
463 past 24 hours following spreading of food-based digestate at a site in India, and emissions from soil  
464 incorporated digestate were negligible; however they only measured emissions for 30 days.

465 At NW and WE, the highest N<sub>2</sub>O emissions corresponded to a peak in soil NH<sub>4</sub>-N concentrations (see  
466 Supplementary Information, Figure S2) suggesting that N<sub>2</sub>O was being produced as a result of the  
467 nitrification of the NH<sub>4</sub>-N in the applied organic materials to NO<sub>3</sub><sup>-</sup>-N by soil micro-organisms. In this  
468 study, there was no clear relationship observed between N<sub>2</sub>O emission rates and changes in the  
469 WFPS (Figures 2a, 3a, 4a and 5a). However, at NW there was a peak in emissions corresponding to a  
470 c.10% drop in WFPS which may have been caused by a change from anaerobic to aerobic conditions  
471 which moved the end product of denitrification from N<sub>2</sub> to N<sub>2</sub>O.

472 Importantly, N<sub>2</sub>O emission factors (EFs) from all the organic material treatments were below the  
473 IPCC Tier 1 default value of 1% of total N applied, and in the case of compost was not significantly  
474 different from background values, although the variability associated with emissions from some  
475 treatments suggests that the IPCC value Tier 1 default value could be exceeded on some occasions  
476 (see Figures 3b and 5b). The EF for food-based digestate (mean 0.45% ± 15%) was much lower than  
477 those obtained by Tiwary *et al.* (2015) of 4 – 10%, although the latter values were obtained under  
478 tropical climatic conditions which are very different to the UK and may have promoted N<sub>2</sub>O losses.  
479 Data derived from the present study could be used in conjunction with other recently published  
480 research results on N<sub>2</sub>O emissions from fertilisers and manures (Bell *et al.*, 2015a,b; Bell *et al.*, 2016)



481 to reduce some of the uncertainty in the UK national agricultural N<sub>2</sub>O inventory through the  
482 generation of robust and experimentally verified Tier 2 EFs in compliance with the requirements of  
483 the IPCC (IPCC, 1996; 2006).

484

#### 485 4.3 Nitrate leaching

486 Nitrate leaching losses from autumn food-based digestate applications were c.15% of the total N  
487 applied compared with the IPCC default value of 30% for all agricultural soil N sources (i.e. synthetic  
488 fertilisers, manures, compost, sewage sludge, crop residues and other organic N sources), which is  
489 used when calculating indirect N<sub>2</sub>O losses from N applications to land. Research in the 1990s using  
490 livestock manures showed that NO<sub>3</sub><sup>-</sup> leaching losses can be greatly reduced by applying the materials  
491 in spring compared with autumn applications (e.g. Chambers *et al.*, 2000), and this has led to the  
492 introduction of no-spreading periods for high available N materials being an integral part of NVZ  
493 legislation in England, Scotland and Wales (SI, 2015; SSI, 2013; WSI, 2013). Despite this, recent UK  
494 survey data have shown that significant percentages of biosolids (64%) and other non-farm organic  
495 materials (32%) are still applied to winter sown crops in August, September and October (Benford,  
496 2016) and hence may be prone to nitrate leaching losses. This is mainly for practical reasons due to  
497 storage capacity pressures, and so that farmers can spread materials to land while soils are still  
498 trafficable and before a crop is sown. The results from our study strongly suggest that, as for  
499 livestock slurry and other high readily available N organic materials, farmers should be advised to  
500 apply food-based digestate in the spring where practically possible, or in autumn to an actively  
501 growing crop such as grass or oilseed rape which will take up available N from the soil so it will not  
502 be lost via overwinter NO<sub>3</sub><sup>-</sup> leaching.

#### 503 5. Conclusion

504 The results of this study have shown that N losses via NH<sub>3</sub> volatilisation following land spreading of  
505 food-based digestate were high at both the arable and grassland sites (30-50% of total N applied).  
506 This was due to the high readily available N content and high pH of the food-based digestate (mean  
507 8.5), and the soil conditions at the time of spreading which affected the rate at which the materials  
508 infiltrated into the soil matrix. Precision application (i.e. bandspreading) can reduce NH<sub>3</sub> emissions,  
509 but the effectiveness of these techniques is dependent on the prevailing soil conditions. Because the  
510 majority of the NH<sub>3</sub> losses occurred within 6 hours of spreading, it is important that where possible  
511 farmers are encouraged to rapidly incorporate food-based digestate into the soil as a method for  
512 conserving N so it can be utilised by the crop (assuming application to spring crops). In contrast, N<sub>2</sub>O

513 losses from food-based digestate were low, with measured emission factors all less than the 1% IPCC  
514 default value (mean  $0.45\% \pm 0.15\%$ ). There was no significant difference between  $N_2O$  losses from  
515 food-based digestate applied by surface broadcast or bandspread techniques; however, we would  
516 recommend that further research is undertaken on different soil types and under different climatic  
517 conditions to confirm this finding, as these are important factors controlling  
518 denitrification/nitrification processes. Autumn applications could potentially lead to a total of 75% of  
519 the applied N being lost by ammonia volatilisation (c.60%) and leaching (c.15%), constituting the loss  
520 of a valuable resource with consequent economic and environmental implications. This strongly  
521 suggests that farmers should be advised to apply these materials in the spring where practically  
522 possible. Emissions ( $NH_3$ ,  $N_2O$ ,  $NO_3^-$ ) from green compost were all low, indicating that in these terms  
523 compost can be considered as a low risk material in terms of N losses, which can be used to build up  
524 soil long-term (organic) N reserves and to improve soil condition.

525 The information produced from this study has been used to develop best practice guidelines for  
526 digestate and compost use in the UK that seek to maximise crop nutrient utilisation and to minimise  
527 emissions to the environment (WRAP, 2016) Furthermore, the  $N_2O$  and  $NH_3$  EFs derived in this study  
528 for both food-based digestate and composts could be used to improve the estimates of emissions  
529 from these sources in the UK GHG and Ammonia Emission Inventories, although emissions under  
530 other agroclimatic conditions are likely to be different.

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724 **List of Tables**

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726 Table 2. Organic material analysis at each site

727 Table 3. Cross-site NH<sub>3</sub> and N<sub>2</sub>O emission factors. *Means labelled with different superscript letters*  
728 *are significantly (P<0.05) different from each other.*

729

730

**Table 1. Baseline soil characteristics, rainfall and cropping at the experimental sites.**

Site code	Site name and location	Soil texture	Clay (%)	Total N (%)	Organic C (%)	pH	Annual rainfall (mm)	Cropping
NW	North Wyke (Devon, England)	Clay loam	38	0.4	6.9	5.5	1,031	Grassland
PW	Pwllpeiran (Ceredigionshire, Wales)	Clay loam	28	0.5	4.7	5.1	975	Grassland
WE	Wensum (Norfolk, England)	Sandy loam	11	0.2	1.3	6.7	594	Winter wheat

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732

**Table 2. Organic material analysis, application rates and N loadings at each site**

Determinand	Food-based digestate	Compost*	Slurry**	FYM**
<b>Site: NW</b>				
Dry matter (%)	5.1	60	6.1	20
Total N (kg/t fw)	8.0	14	2.6	5.8
NH <sub>4</sub> -N (kg/t fw)	5.8	0.7	1.4	<0.1
NO <sub>3</sub> <sup>-</sup> -N (kg/t fw)	<0.1	<0.1	<0.1	<0.1
NH <sub>4</sub> -N (% total N)	73	5	56	<1
pH	8.1	8.5	8.2	9.0
Application rate (t/ha)	20	20	30	25
N loading (kg/ha)	160	271	77	144
<b>Site: PW</b>				
Dry matter (%)	6.1	51	4.9	24
Total N (kg/t fw)	5.4	7.0	2.2	4.9
NH <sub>4</sub> -N (kg/t fw)	3.9	<0.1	1.2	<0.1
NO <sub>3</sub> <sup>-</sup> -N (kg/t fw)	<0.1	<0.1	<0.1	0.4
NH <sub>4</sub> -N (% total N)	72	1	53	8
pH	8.4	7.9	7.0	7.3
Application rate (t/ha)	20	20	30	25
N loading (kg/ha)	107	140	67	122
<b>Site: WE autumn</b>				
Dry matter (%)	5.4	54	2.3	24
Total N (kg/t fw)	7.8	11	3.0	8.1
NH <sub>4</sub> -N (kg/t fw)	6.3	1.5	2.2	0.8
NO <sub>3</sub> <sup>-</sup> -N (kg/t fw)	<0.1	<0.1	<0.1	0.1
NH <sub>4</sub> -N (% total N)	82	14	75	9
pH	8.8	7.0	7.5	8.4
Application rate (t/ha)	32	20	41	30
N loading (kg/ha)	245	216	122	244
<b>Site: WE spring</b>				
Dry matter (%)	4.4	48	2.7	23
Total N (kg/t fw)	6.9	9.0	2.6	9.2
NH <sub>4</sub> -N (kg/t fw)	6.2	0.2	2.2	0.2
NO <sub>3</sub> <sup>-</sup> -N (kg/t fw)	<0.1	<0.1	<0.1	0.6
NH <sub>4</sub> -N (% total N)	89	2	86	3
pH	8.7	8.3	8.0	7.4
Application rate (t/ha)	30	20	38	30
N loading (kg/ha)	207	181	98	277

734 fw = fresh weight

735 \*Green compost at NW and PW; green/food compost at WE

736 \*\*Cattle slurry and FYM at NW and PW; pig slurry and FYM at WE

737 **Table 3. Cross-site NH<sub>3</sub> and N<sub>2</sub>O emission factors. Means labelled with different superscript letters**  
 738 **are significantly ( $P<0.05$ ) different from each other.**

Treatment	Mean NH <sub>3</sub> EF (% total N applied)	Mean N <sub>2</sub> O EF
Food based digestate – surface broadcast	42 <sup>d</sup>	0.47 <sup>b</sup>
Food based digestate – bandspread	38 <sup>d</sup>	0.43 <sup>b</sup>
Livestock slurry – surface broadcast	31 <sup>c</sup>	0.35 <sup>b</sup>
Livestock slurry – bandspread	24 <sup>b</sup>	0.55 <sup>b</sup>
Livestock FYM	4.5 <sup>a</sup>	0.28 <sup>b</sup>
Compost	3.3 <sup>a</sup>	<0.01 <sup>a</sup>
<i>SED</i>	3.3	0.12
<i>LSD</i>	6.7	0.24

739 SED: Standard error of difference of means

740 LSD: Least significant difference of means (5% level)

741

742 **List of figures**

743 Figure 1. Cross-site ammonia emissions curves.

744 Figure 2. NW spring: a) daily mean N<sub>2</sub>O fluxes and b) N<sub>2</sub>O emission factors. *Error bars show the*  
745 *standard error; SED = standard error of difference of means. No significant treatment differences*

746 Figure 3. PW spring: a) daily mean nitrous oxide fluxes and b) emission factors. *Error bars show the*  
747 *standard error; SED = standard error of difference of means. Columns labelled with different letters*  
748 *are significantly ( $P < 0.05$ ) different from each other.*

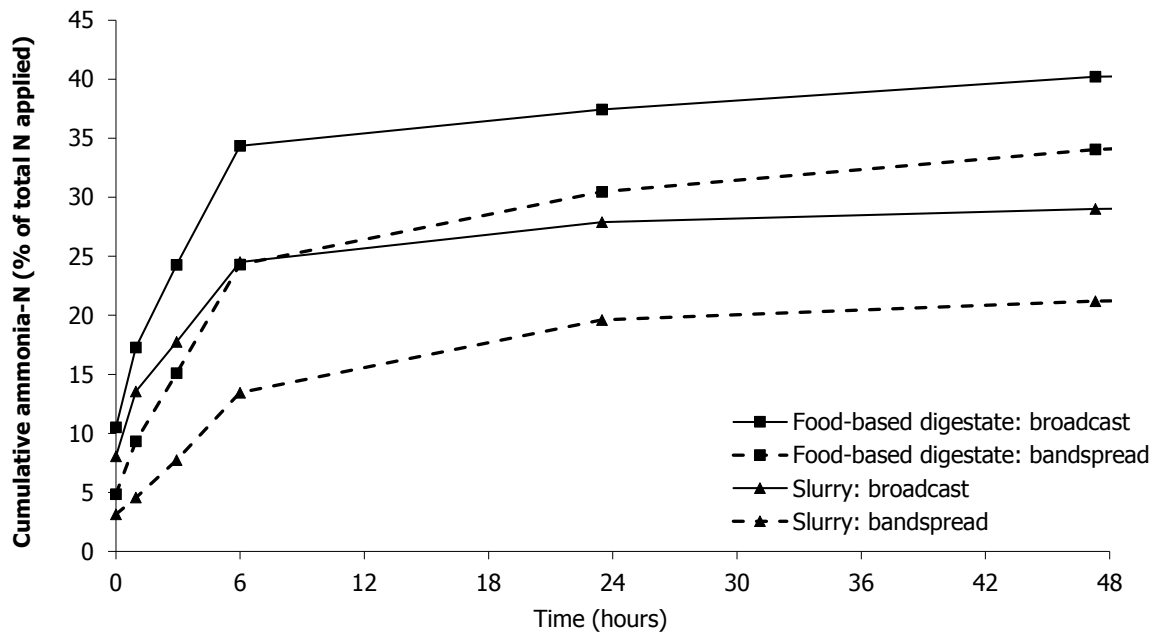
749 Figure 4. WE autumn: a) daily mean N<sub>2</sub>O fluxes and b) N<sub>2</sub>O emission factors. *Error bars show the*  
750 *standard error; SED = standard error of difference of means. No significant treatment differences.*

751 Figure 5. WE spring: a) daily mean N<sub>2</sub>O fluxes and b) N<sub>2</sub>O emission factors. *Error bars show the*  
752 *standard error; SED = standard error of difference of means. Columns labelled with different letters*  
753 *are significantly ( $P < 0.05$ ) different from each other.*

754 Figure 6. Leaching losses (% of total N applied) following the autumn 2011 organic material  
755 applications. *Error bars show the standard error; SED = standard error of difference of means.*  
756 *Columns labelled with different letters are significantly ( $P < 0.05$ ) different from each other.*

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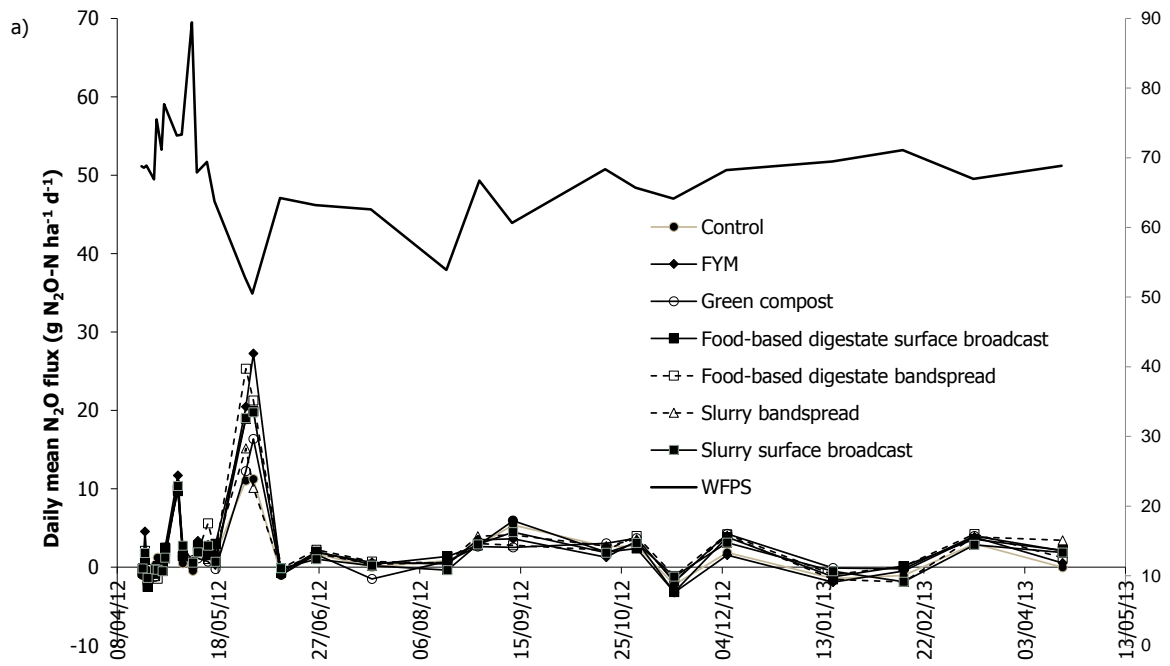


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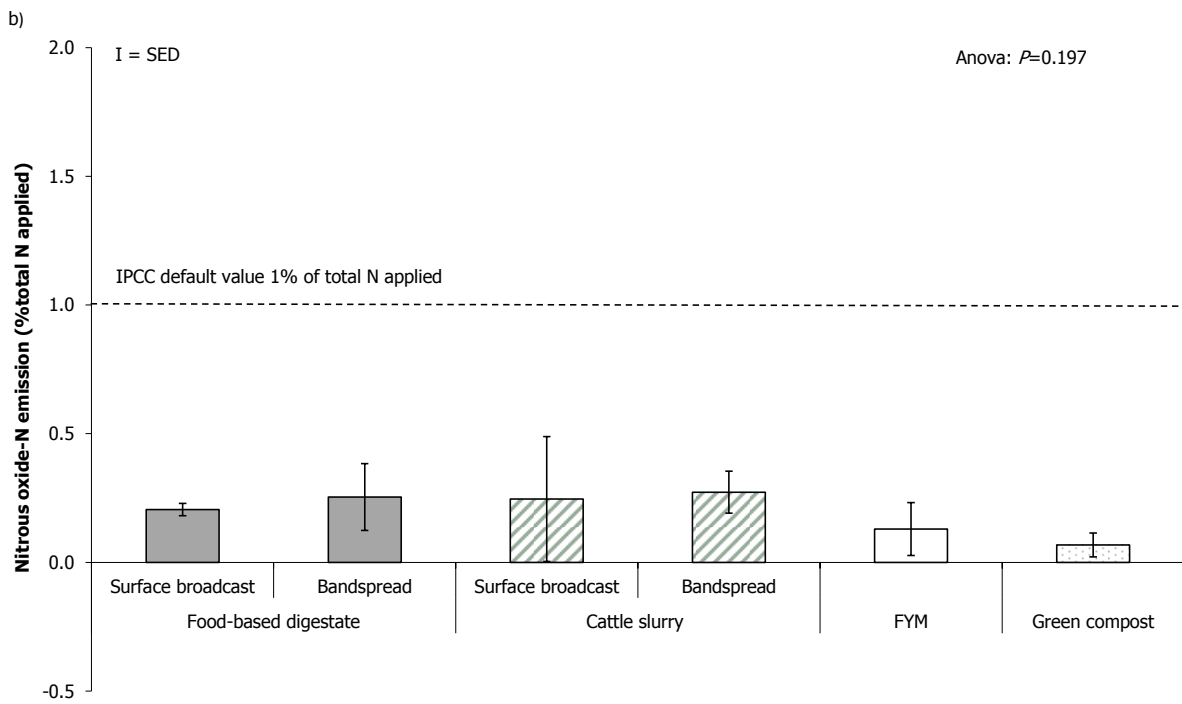
760 **Figure 1. Cross-site ammonia emissions curves.**

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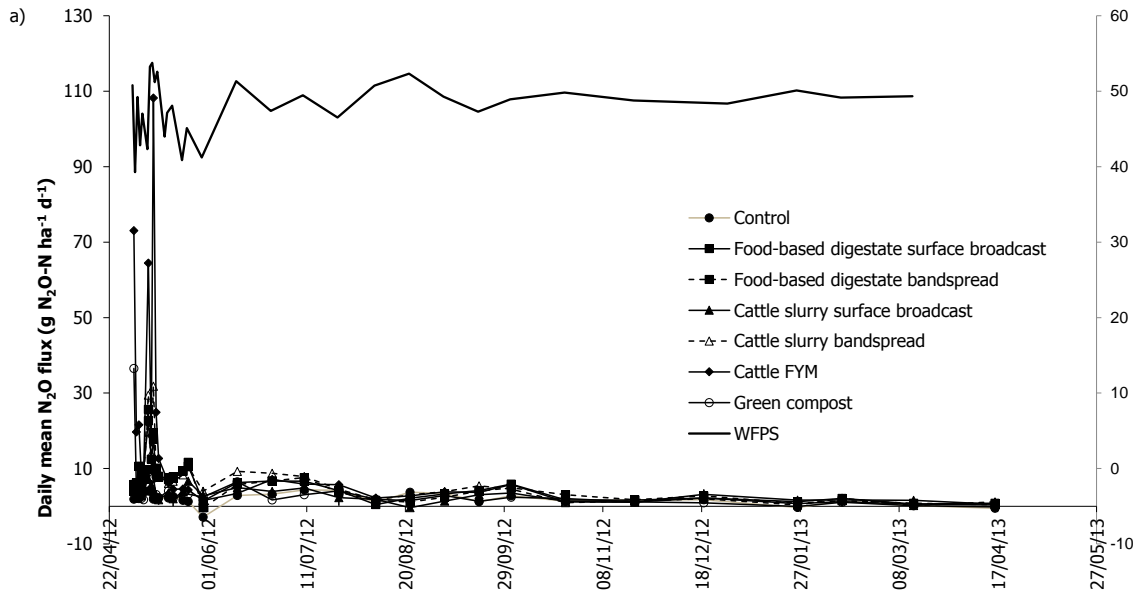
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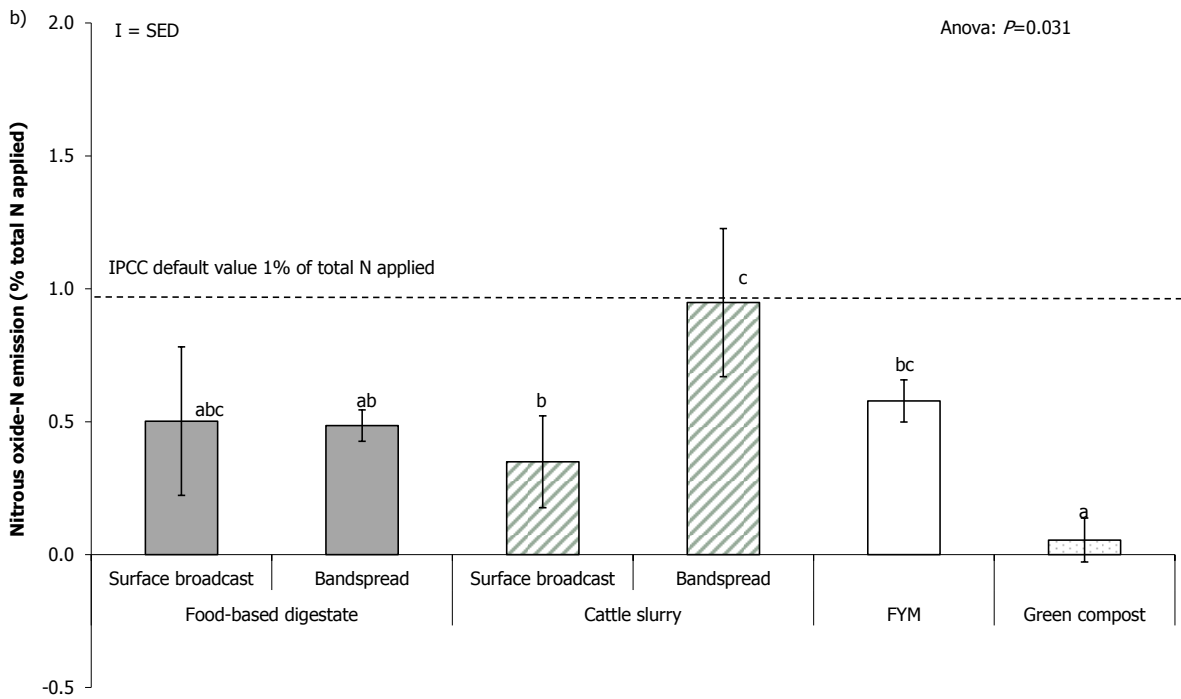
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**Figure 2. NW spring: a) daily mean N<sub>2</sub>O fluxes and b) N<sub>2</sub>O emission factors. Error bars show the standard error; SED = standard error of difference of means. No significant treatment differences.**



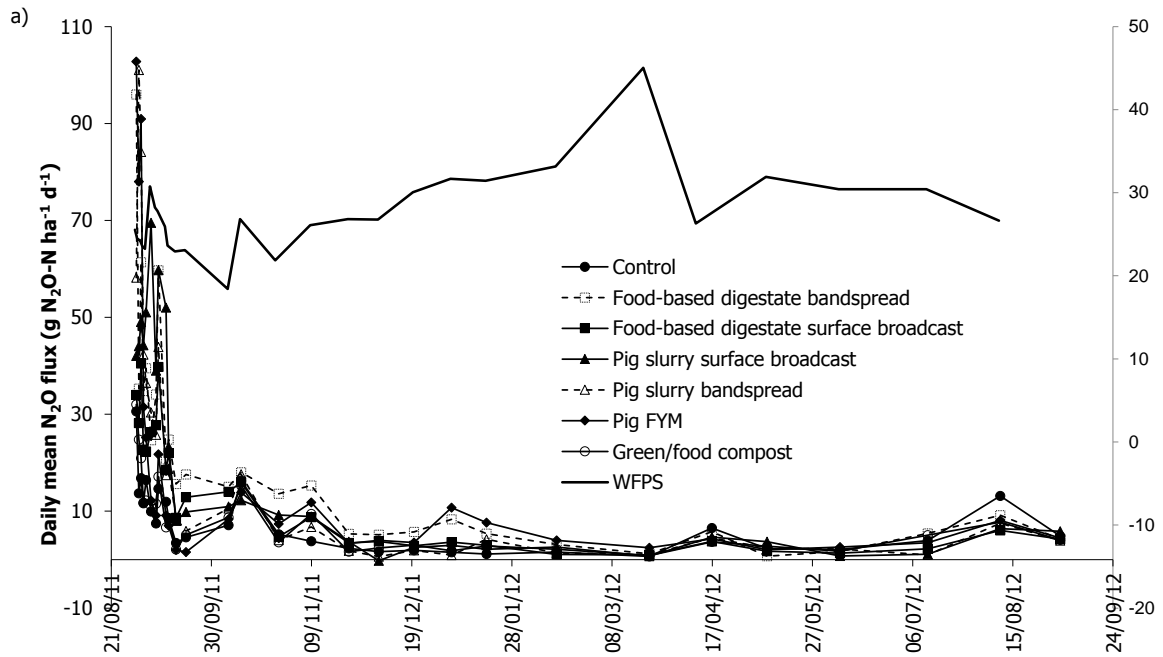
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**Figure 3. PW spring: a) daily mean nitrous oxide fluxes and b) emission factors. Error bars show the standard error; SED = standard error of difference of means. Columns labelled with different letters are significantly ( $P < 0.05$ ) different from each other.**

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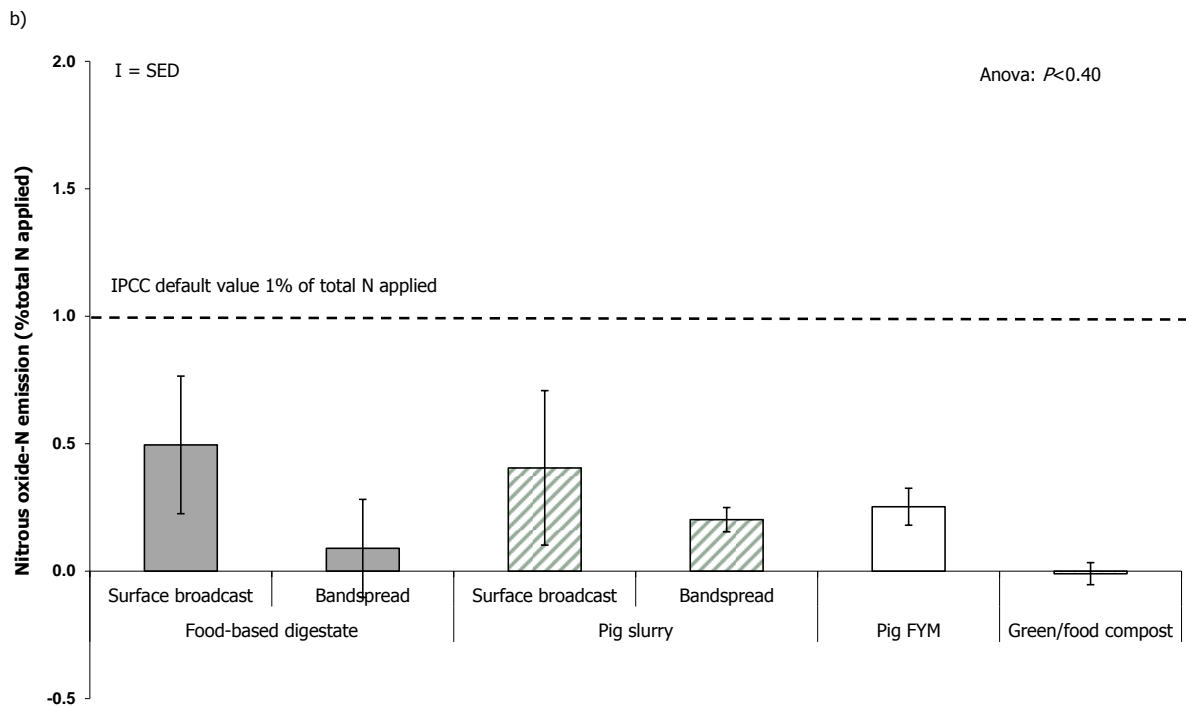
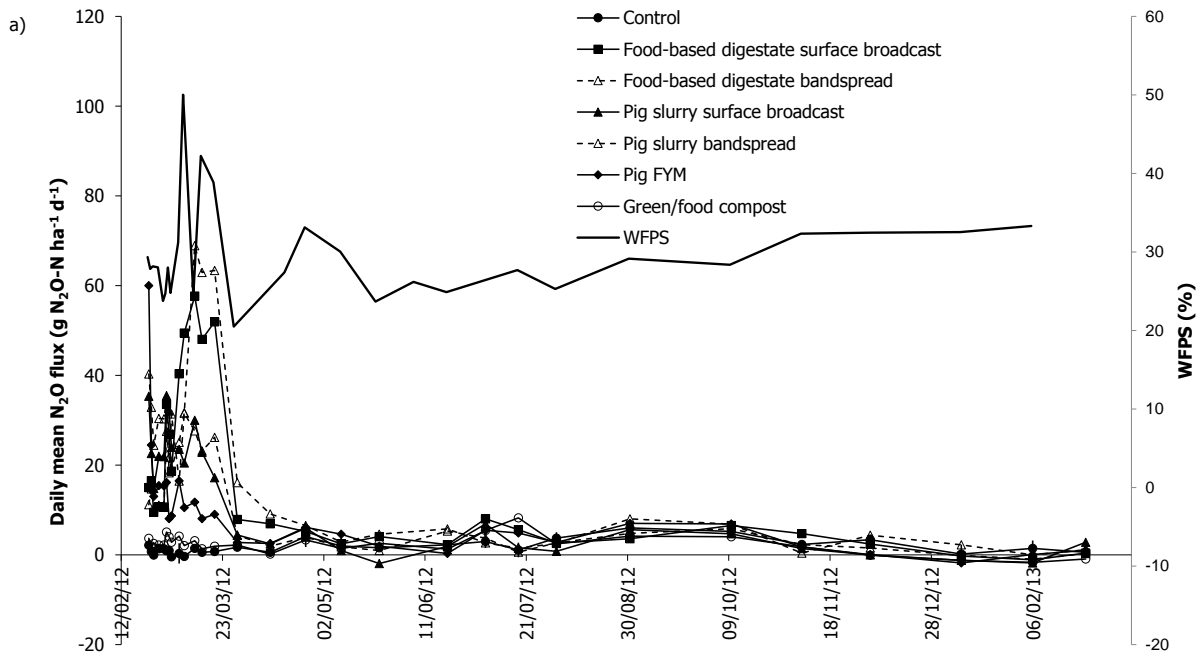


Figure 4. WE autumn: a) daily mean  $N_2O$  fluxes and b)  $N_2O$  emission factors. Error bars show the standard error; SED = standard error of difference of means. No significant treatment differences.

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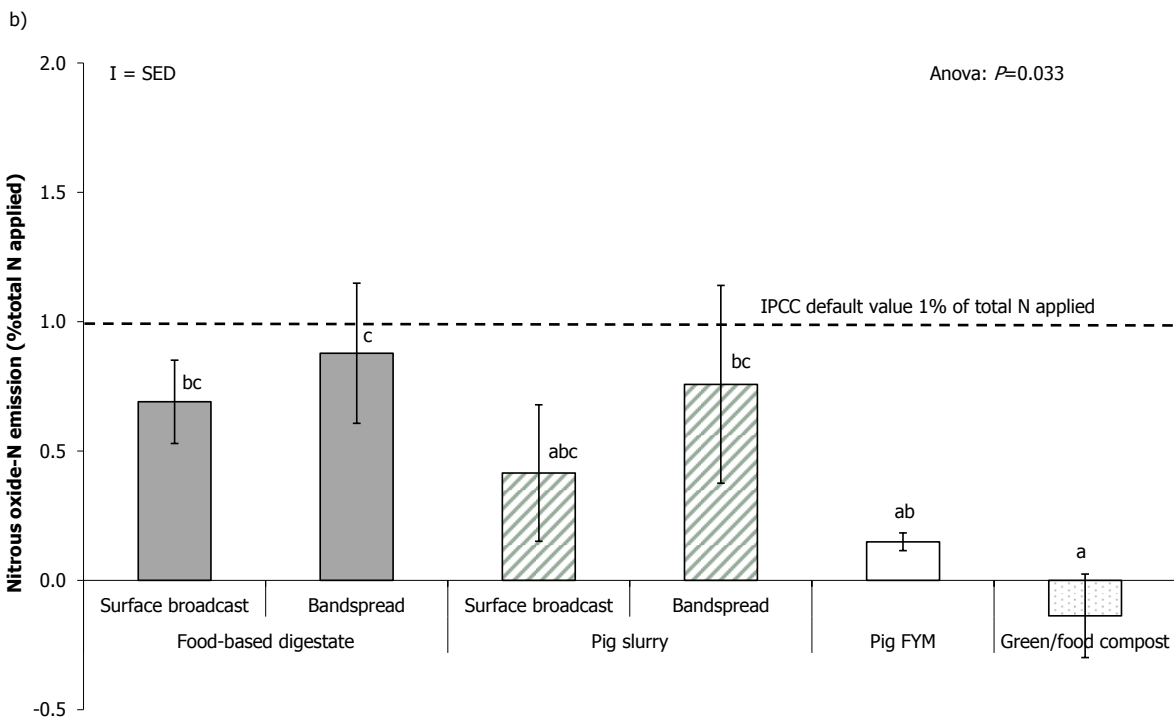
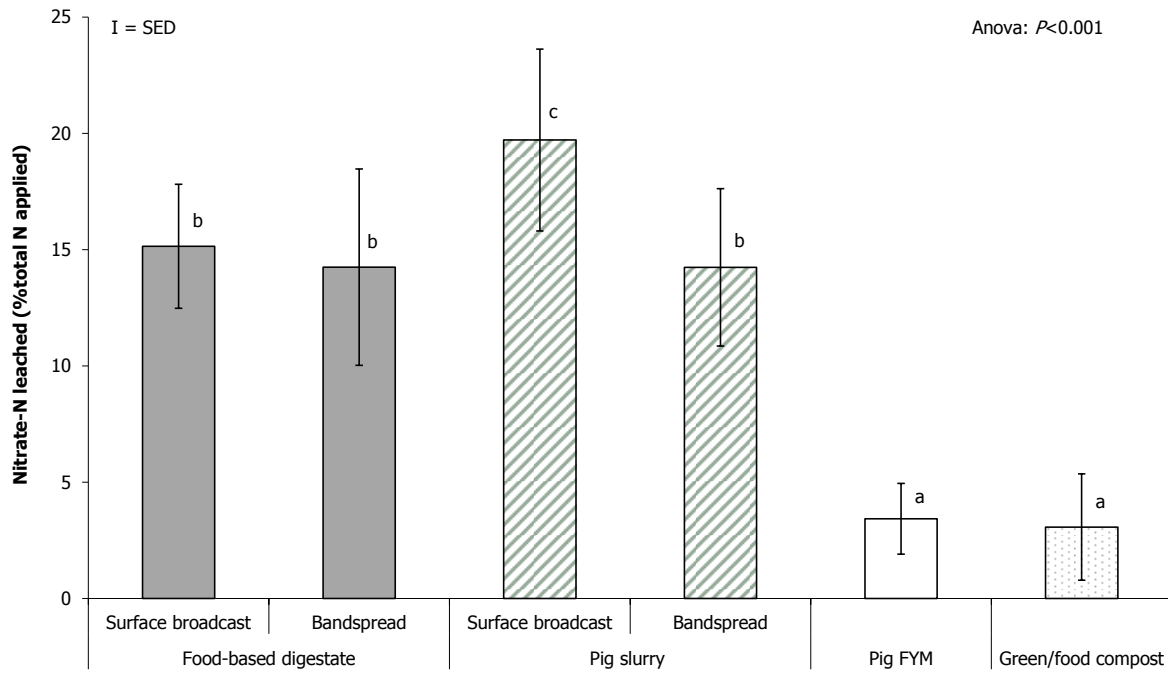


Figure 5. WE spring: a) daily mean N<sub>2</sub>O fluxes and b) N<sub>2</sub>O emission factors. Error bars show the standard error; SED = standard error of difference of means. Columns labelled with different letters are significantly (P<0.05) different from each other.

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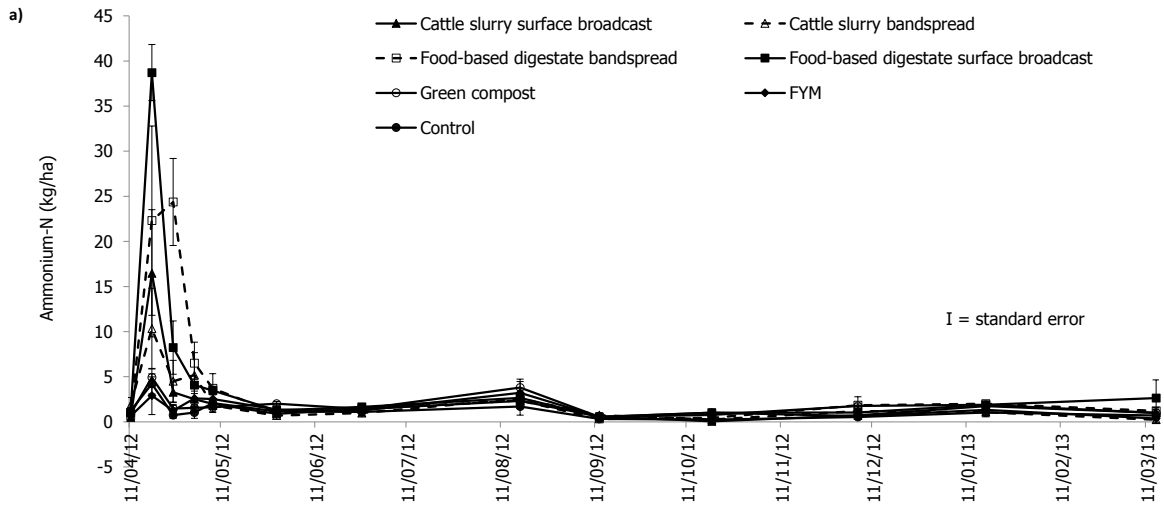
771 **Figure 6. Leaching losses (% of total N applied) following the autumn 2011 organic material**  
 772 **applications. Error bars show the standard error; SED = standard error of difference of means.**  
 773 **Columns labelled with different letters are significantly ( $P < 0.05$ ) different from each other.**

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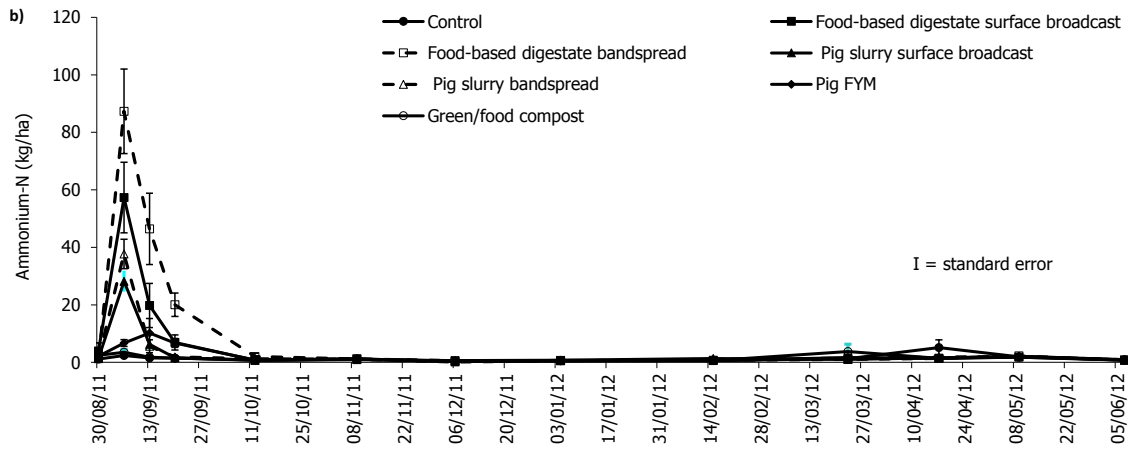
775 **Supplementary Information**

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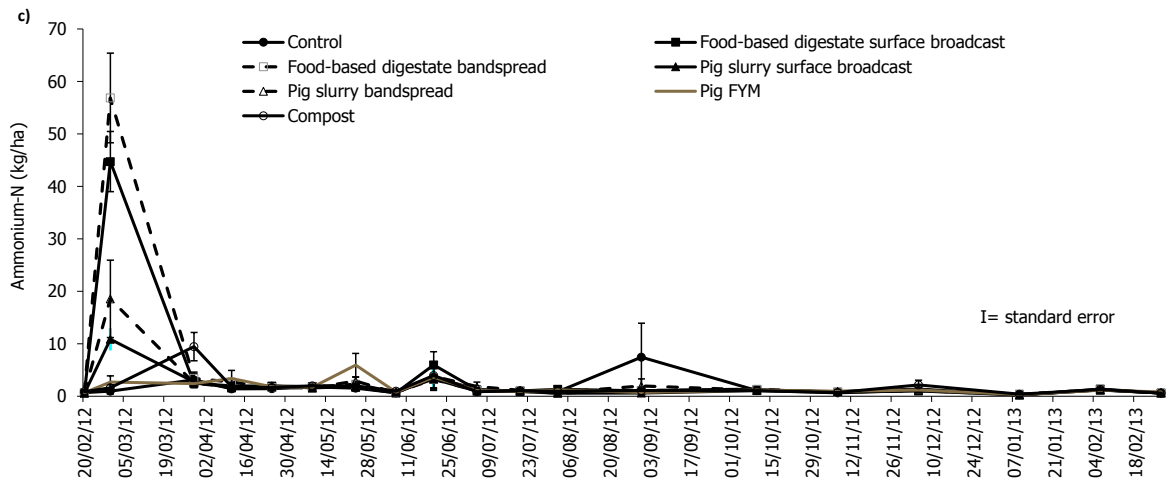
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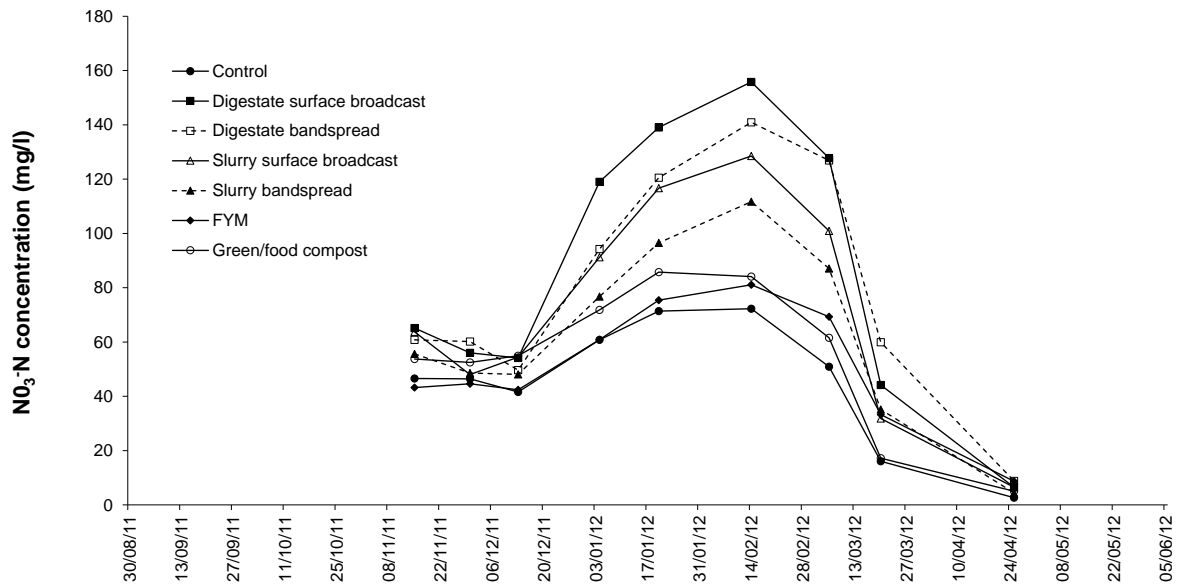


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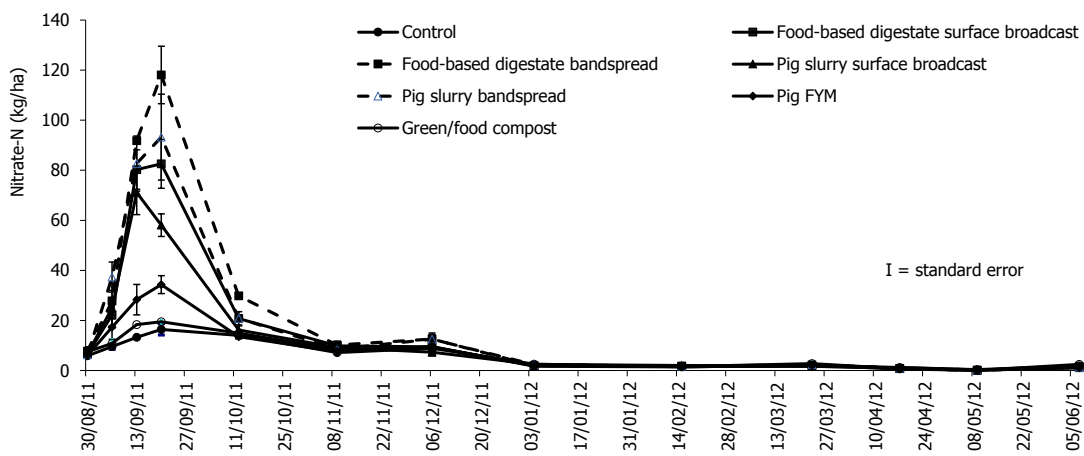
780 Figure S1. Soil  $\text{NH}_4\text{-N}$  concentrations at a) NW, b) WE autumn and c) WE spring

781 a)



782

783 b)



784

785 **Figure S2. WE autumn a)  $\text{NO}_3^-$  concentrations in leachate following the autumn 2011 organic**  
786 **material applications and b) soil  $\text{NO}_3^-$  concentrations**

787