

Crystal structure of (Z)-2-hydroxy-N'-(4-oxo-1,3-thiazolidin-2-ylidene)-benzohydrazide

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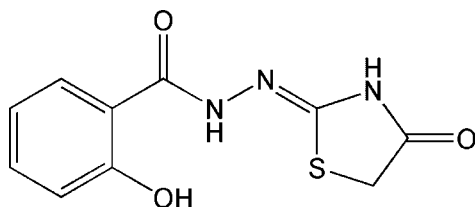
In the title compound, $C_{10}H_9N_3O_3S$, the five-membered ring adopts a slightly twisted conformation about the C_m-S ($m =$ methylene) bond. The dihedral angle between this ring and the benzene ring is $7.99(9)^\circ$. A bifurcated intramolecular $N-H \cdots (O,S)$ hydrogen bond helps to establish the near planar conformation of the molecule. In the crystal, molecules are linked by $N-H \cdots O$ and $O-H \cdots O$ hydrogen bonds to generate (001) sheets.

Keywords: crystal structure; benzohydrazide; 4-thiazolidinone derivatives; biological activity; hydrogen bonding.

CCDC reference: 1028612

1. Related literature

For background to the biological activities of 4-thiazolidinone derivatives, see: Singh *et al.* (1981); Verma & Shailendra (2008); Jain *et al.*, (2012).



2. Experimental

2.1. Crystal data

$C_{10}H_9N_3O_3S$
 $M_r = 251.26$

Monoclinic, $C2/c$
 $a = 18.788(2) \text{ \AA}$

$b = 8.9334(10) \text{ \AA}$
 $c = 12.7969(14) \text{ \AA}$
 $\beta = 92.667(2)^\circ$
 $V = 2145.6(4) \text{ \AA}^3$
 $Z = 8$

Mo $K\alpha$ radiation
 $\mu = 0.30 \text{ mm}^{-1}$
 $T = 298 \text{ K}$
 $0.22 \times 0.21 \times 0.20 \text{ mm}$

2.2. Data collection

Bruker SMART CCD diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2001)
 $T_{\min} = 0.937$, $T_{\max} = 0.942$

10873 measured reflections
2102 independent reflections
1784 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.032$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.103$
 $S = 0.99$
2102 reflections
181 parameters
9 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.36 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.30 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1A \cdots O1$	0.91 (1)	1.86 (1)	2.6166 (17)	139 (2)
$N1-H1A \cdots S1$	0.91 (1)	2.42 (2)	2.8879 (15)	112 (1)
$N3-H3A \cdots O3^i$	0.91 (1)	1.88 (1)	2.7767 (18)	168 (2)
$O1-H1 \cdots O2^ii$	0.82 (1)	1.83 (1)	2.6538 (16)	177 (2)

Symmetry codes: (i) $-x + \frac{1}{2}, -y - \frac{1}{2}, -z + 1$; (ii) $x, -y + 1, z + \frac{1}{2}$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: HB7294).

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supporting information

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S1. Comment

Derivatives of 4-thiazolidinone exhibit prominent biological activities such as antibacterial activity, antifungal activity, antitubercular activity, anticancer activity, anti-inflammatory activity, analgesic activity, anticonvulsant activity, antidepressant activity, antiviral/anti-HIV activity, antidiabetic activity, muscarinic receptor 1 agonist, FSH receptor agonist, trypanocidal (anti-epimastigote) activity and antiarrhythmic activity (Jain, *et al.*, 2012; Verma & Shailendra, 2008; Singh *et al.*, 1981). Besides, enough coordinational sites exist in these compounds, lead to a potential to form a supermolecular structure. As part of our ongoing studies, the preparation and X-ray structure determination of the title compound, (I), was undertaken.

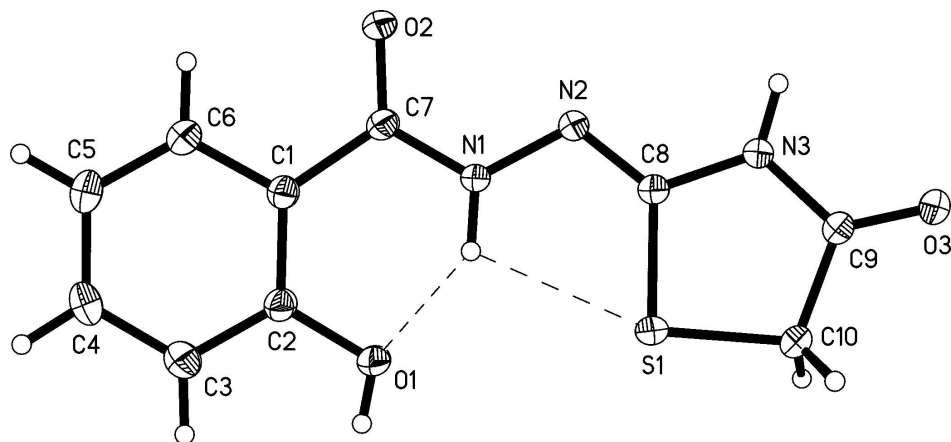
In the title molecule (Fig. 1), bond lengths and angles in (I) show normal values. The non-hydrogen atoms of the molecule lie in a plane with an r.m.s deviation of 0.002 Å. An intramolecular tricentered hydrogen bond is observed between the N—H of imino group, O atom of phenolic hydroxyl group and S atom of 4-thiazolidinone group (Fig. 1 and Table 1). The molecules translated one unit cell along the *b* direction are stacked with N3 \cdots O3ⁱ, O1 \cdots O2ⁱⁱ and O3 \cdots S1ⁱⁱⁱ [symmetry code: (i) 1/2 - *x*, 3/2 - *y*, 1 - *z*; (ii) *x*, -*y*, -1/2 + *z*; (iii) 1/2 - *x*, 1/2 + *y*, 1/2 - *z*] distances of 2.7766 (18) Å, 2.6538 (17) Å and 3.1028 (13) Å, respectively, indicating weak C—H \cdots π interactions.

S2. Experimental

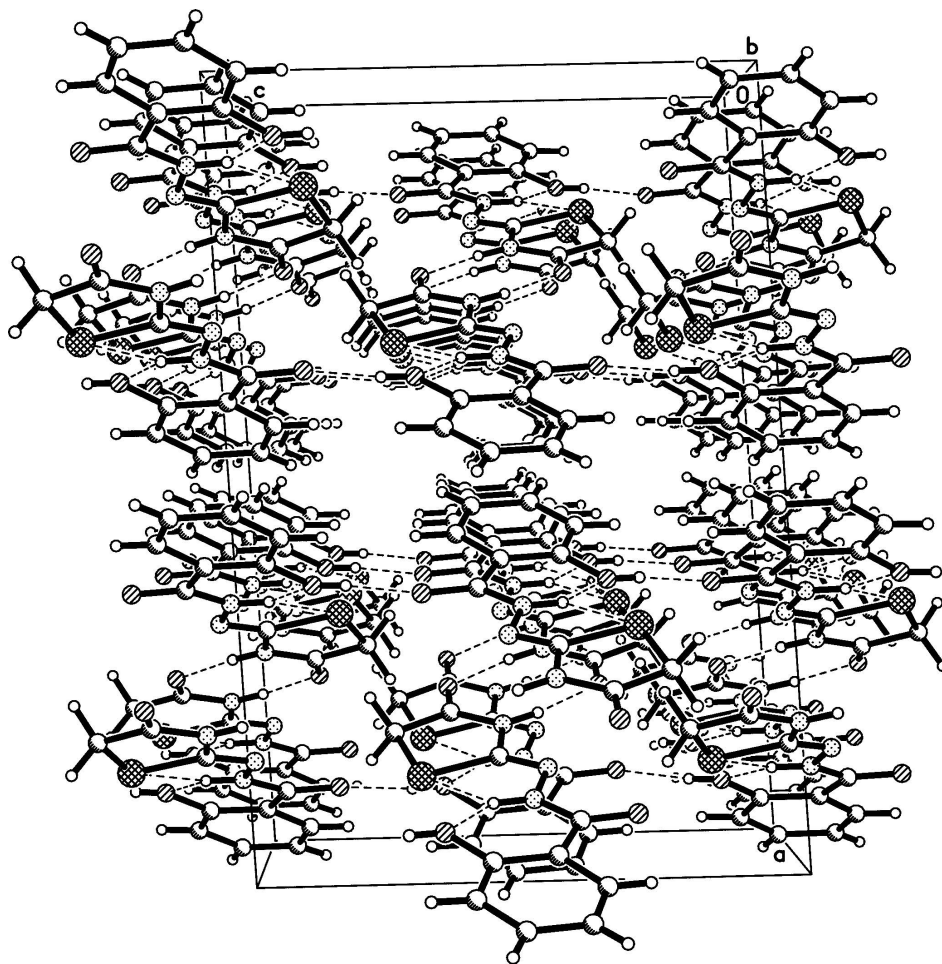
4-Salicyloyl thiosemicarbazide (2.11 g, 0.01 mol), ethyl bromoacetate (1.67 g, 0.01 mol), sodium acetate (3.28 g, 0.04 mol) and 40 ml of ethyl alcohol were added to a round-bottom flask. Stirred for 10 minutes, then the reaction mixture was slowly warmed to boiling and stirred for 10 h. After cooling to room temperature, 40 ml of water were added and staying for 12 h. The resulting precipitate was filtered and recrystallized with ethyl alcohol to give 1.30 g of the title compound. Colourless blocks were grown by slow evaporation from a mixed solution of methanol+*N,N*-dimethyl formamide (6:1) at room temperature.

S3. Refinement

All H atoms were placed in calculated positions, with N—H distances of 0.91 Å, O—H distances of 0.82 Å, and C—H distances of 0.97 Å (CH₂) and 0.93 Å (benzyl CH). They were included in the refinement in the riding-model approximation, with isotropic displacement parameters set to 1.2 U_{eq} of the carrier atom (1.5 U_{eq} for hydroxyl H atoms).

**Figure 1**

The molecular structure of (I), showing 30% probability displacement ellipsoids. Dashed lines indicate hydrogen bonds.

**Figure 2**

Packing diagram for (I). The hydrogen bonds are indicated by dashed lines.

(Z)-2-Hydroxy-N'-(4-oxo-1,3-thiazolidin-2-ylidene)benzohydrazide*Crystal data*C₁₀H₉N₃O₃S $M_r = 251.26$ Monoclinic, *C2/c*

Hall symbol: -C 2yc

 $a = 18.788$ (2) Å $b = 8.9334$ (10) Å $c = 12.7969$ (14) Å $\beta = 92.667$ (2)° $V = 2145.6$ (4) Å³ $Z = 8$ $F(000) = 1040$ $D_x = 1.556$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 4520 reflections

 $\theta = 2.5$ – 28.0 ° $\mu = 0.30$ mm⁻¹ $T = 298$ K

Block, colorless

 $0.22 \times 0.21 \times 0.20$ mm*Data collection*

Bruker SMART CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 ϕ and ω scans

Absorption correction: multi-scan

(SADABS; Bruker, 2001)

 $T_{\min} = 0.937$, $T_{\max} = 0.942$

10873 measured reflections

2102 independent reflections

1784 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.032$ $\theta_{\max} = 26.0$ °, $\theta_{\min} = 2.2$ ° $h = -22 \rightarrow 23$ $k = -11 \rightarrow 11$ $l = -15 \rightarrow 15$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.103$ $S = 0.99$

2102 reflections

181 parameters

9 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

 $w = 1/[\sigma^2(F_o^2) + (0.0666P)^2 + 0.9019P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.36$ e Å⁻³ $\Delta\rho_{\min} = -0.30$ e Å⁻³*Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.40566 (8)	0.54405 (17)	0.47258 (12)	0.0304 (4)
C2	0.40845 (9)	0.58782 (17)	0.57775 (12)	0.0318 (4)
C3	0.43987 (10)	0.7222 (2)	0.60795 (14)	0.0422 (4)

C4	0.46818 (12)	0.8145 (2)	0.53512 (17)	0.0523 (5)
C5	0.46518 (12)	0.7748 (2)	0.43080 (16)	0.0538 (5)
C6	0.43446 (11)	0.6417 (2)	0.40054 (14)	0.0421 (4)
C7	0.37628 (8)	0.39894 (17)	0.43052 (12)	0.0307 (4)
C8	0.31830 (9)	0.07860 (17)	0.55293 (12)	0.0302 (4)
C9	0.27480 (9)	-0.13806 (17)	0.63012 (12)	0.0329 (4)
C10	0.29374 (11)	-0.0495 (2)	0.72704 (13)	0.0398 (4)
H1	0.3791 (12)	0.539 (2)	0.7078 (9)	0.060*
H3	0.4396 (10)	0.749 (2)	0.6782 (5)	0.048*
H4	0.4891 (9)	0.9032 (12)	0.5589 (15)	0.048*
H5	0.4837 (10)	0.8367 (17)	0.3803 (11)	0.048*
H6	0.4346 (10)	0.612 (2)	0.3310 (5)	0.048*
H1A	0.3575 (10)	0.329 (2)	0.5703 (5)	0.048*
H3A	0.2773 (10)	-0.1079 (19)	0.4796 (7)	0.048*
H10A	0.3270 (8)	-0.1054 (19)	0.7719 (12)	0.048*
H10B	0.2520 (6)	-0.022 (2)	0.7644 (14)	0.048*
N1	0.35652 (8)	0.30049 (15)	0.50199 (10)	0.0338 (3)
N2	0.32887 (8)	0.16107 (15)	0.47420 (10)	0.0345 (3)
N3	0.28759 (8)	-0.06155 (15)	0.54179 (10)	0.0346 (3)
O1	0.38054 (7)	0.49708 (13)	0.65105 (9)	0.0441 (3)
O2	0.37177 (7)	0.37081 (13)	0.33575 (9)	0.0438 (3)
O3	0.25005 (7)	-0.26414 (14)	0.63173 (9)	0.0451 (3)
S1	0.33730 (3)	0.11903 (5)	0.68522 (3)	0.0514 (2)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0361 (8)	0.0267 (8)	0.0285 (8)	0.0000 (6)	0.0016 (6)	0.0019 (6)
C2	0.0403 (9)	0.0265 (8)	0.0287 (8)	0.0018 (6)	0.0034 (6)	0.0015 (6)
C3	0.0562 (11)	0.0344 (9)	0.0359 (9)	-0.0042 (8)	0.0016 (8)	-0.0053 (7)
C4	0.0696 (13)	0.0344 (10)	0.0531 (12)	-0.0202 (9)	0.0038 (10)	-0.0048 (9)
C5	0.0753 (14)	0.0414 (11)	0.0455 (12)	-0.0211 (10)	0.0100 (10)	0.0071 (9)
C6	0.0580 (11)	0.0381 (10)	0.0305 (9)	-0.0094 (8)	0.0044 (8)	0.0037 (7)
C7	0.0389 (9)	0.0290 (8)	0.0243 (8)	0.0002 (6)	0.0026 (6)	0.0024 (6)
C8	0.0409 (9)	0.0263 (8)	0.0237 (8)	-0.0013 (6)	0.0029 (6)	-0.0009 (6)
C9	0.0439 (9)	0.0282 (8)	0.0269 (8)	-0.0005 (7)	0.0039 (7)	0.0023 (6)
C10	0.0618 (11)	0.0324 (9)	0.0255 (8)	-0.0068 (8)	0.0053 (7)	0.0032 (7)
N1	0.0555 (9)	0.0249 (7)	0.0211 (7)	-0.0082 (6)	0.0038 (6)	-0.0015 (5)
N2	0.0534 (8)	0.0261 (7)	0.0239 (7)	-0.0070 (6)	0.0012 (6)	0.0002 (5)
N3	0.0540 (9)	0.0267 (7)	0.0230 (7)	-0.0076 (6)	0.0014 (6)	-0.0008 (5)
O1	0.0770 (9)	0.0316 (6)	0.0247 (6)	-0.0106 (6)	0.0122 (6)	-0.0041 (5)
O2	0.0752 (9)	0.0354 (7)	0.0207 (6)	-0.0124 (6)	0.0025 (5)	0.0010 (5)
O3	0.0724 (9)	0.0313 (7)	0.0320 (7)	-0.0149 (6)	0.0060 (6)	0.0027 (5)
S1	0.0938 (4)	0.0391 (3)	0.0215 (2)	-0.0258 (2)	0.0027 (2)	-0.00162 (17)

Geometric parameters (Å, °)

C1—C6	1.396 (2)	C8—N2	1.271 (2)
C1—C2	1.400 (2)	C8—N3	1.383 (2)
C1—C7	1.499 (2)	C8—S1	1.7514 (16)
C2—O1	1.3630 (19)	C9—O3	1.219 (2)
C2—C3	1.385 (2)	C9—N3	1.352 (2)
C3—C4	1.371 (3)	C9—C10	1.500 (2)
C3—H3	0.930 (2)	C10—S1	1.8064 (18)
C4—C5	1.380 (3)	C10—H10A	0.968 (4)
C4—H4	0.930 (2)	C10—H10B	0.969 (4)
C5—C6	1.370 (3)	N1—N2	1.3893 (18)
C5—H5	0.930 (2)	N1—H1A	0.909 (2)
C6—H6	0.930 (2)	N3—H3A	0.909 (2)
C7—O2	1.2376 (19)	O1—H1	0.820 (2)
C7—N1	1.334 (2)	O1—H1	0.820 (2)
C6—C1—C2	117.51 (15)	N2—C8—S1	127.90 (13)
C6—C1—C7	116.79 (14)	N3—C8—S1	110.58 (11)
C2—C1—C7	125.68 (14)	O3—C9—N3	124.33 (15)
O1—C2—C3	119.75 (15)	O3—C9—C10	123.33 (15)
O1—C2—C1	119.81 (14)	N3—C9—C10	112.33 (14)
C3—C2—C1	120.45 (15)	C9—C10—S1	106.74 (11)
C4—C3—C2	120.40 (17)	C9—C10—H10A	109.9 (12)
C4—C3—H3	121.5 (13)	S1—C10—H10A	108.5 (12)
C2—C3—H3	118.1 (13)	C9—C10—H10B	112.0 (12)
C3—C4—C5	120.22 (17)	S1—C10—H10B	109.0 (12)
C3—C4—H4	117.6 (13)	H10A—C10—H10B	110.6 (17)
C5—C4—H4	122.2 (13)	C7—N1—N2	121.89 (13)
C6—C5—C4	119.64 (17)	C7—N1—H1A	118.8 (13)
C6—C5—H5	119.1 (13)	N2—N1—H1A	119.1 (13)
C4—C5—H5	121.3 (13)	C8—N2—N1	112.77 (13)
C5—C6—C1	121.79 (17)	C9—N3—C8	117.45 (13)
C5—C6—H6	120.1 (12)	C9—N3—H3A	117.5 (12)
C1—C6—H6	118.0 (12)	C8—N3—H3A	125.0 (12)
O2—C7—N1	121.96 (15)	H1—O1—C2	111.5 (16)
O2—C7—C1	122.33 (14)	C2—O1—H1	111.5 (16)
N1—C7—C1	115.70 (13)	C8—S1—C10	92.28 (8)
N2—C8—N3	121.51 (14)		
C6—C1—C2—O1	179.46 (16)	N3—C9—C10—S1	-7.08 (19)
C7—C1—C2—O1	-2.4 (3)	O2—C7—N1—N2	-0.8 (3)
C6—C1—C2—C3	-1.1 (3)	C1—C7—N1—N2	-179.53 (14)
C7—C1—C2—C3	177.05 (16)	N3—C8—N2—N1	176.65 (14)
O1—C2—C3—C4	179.98 (18)	S1—C8—N2—N1	-2.5 (2)
C1—C2—C3—C4	0.5 (3)	C7—N1—N2—C8	174.89 (15)
C2—C3—C4—C5	0.4 (3)	O3—C9—N3—C8	-177.67 (16)
C3—C4—C5—C6	-0.7 (4)	C10—C9—N3—C8	3.0 (2)

C4—C5—C6—C1	0.1 (3)	N2—C8—N3—C9	-176.46 (16)
C2—C1—C6—C5	0.8 (3)	S1—C8—N3—C9	2.81 (19)
C7—C1—C6—C5	-177.51 (18)	C3—C2—O1—H1	9.9 (17)
C6—C1—C7—O2	-6.1 (2)	C1—C2—O1—H1	-170.6 (17)
C2—C1—C7—O2	175.83 (16)	N2—C8—S1—C10	173.24 (17)
C6—C1—C7—N1	172.68 (15)	N3—C8—S1—C10	-5.97 (13)
C2—C1—C7—N1	-5.4 (2)	C9—C10—S1—C8	7.23 (14)
O3—C9—C10—S1	173.59 (15)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1A \cdots O1	0.91 (1)	1.86 (1)	2.6166 (17)	139 (2)
N1—H1A \cdots S1	0.91 (1)	2.42 (2)	2.8879 (15)	112 (1)
N3—H3A \cdots O3 ⁱ	0.91 (1)	1.88 (1)	2.7767 (18)	168 (2)
O1—H1 \cdots O2 ⁱⁱ	0.82 (1)	1.83 (1)	2.6538 (16)	177 (2)

Symmetry codes: (i) $-x+1/2, -y-1/2, -z+1$; (ii) $x, -y+1, z+1/2$.