Graphical Abstract

TETRAHEDRON

Preparation of penta-azole containing cyclopeptides: challenges in macrocyclization

Delia Hernández,^a Estela Riego,^aAndrés Francesch, ^b Carmen Cuevas,^b Fernando Albericio a,c, and Mercedes Álvarez a,d,*

> *ºBarcelona Science Park Josep Samitier 1-5, E-08028 Barcelona, Spain b Pharma Mar, Avda Reyes Católicos 1, E- 28770 Colmenar Viejo, Madrid, Spain ^cDepartment of Organic Chemistry, University of Barcelona, E-08028-Barcelona d Laboratory of Organic Chemistry, Faculty of Pharmacy, University of Barcelona, E-08028-Barcelona*

Abstract—Herein is described the synthesis of several analogs of the natural product IB-01211 from concatenated azoles, via a biomimetic pathway based on cyclization-oxidation of serine containing peptides combined with the Hantzsch synthesis. The macrocyclization of rigid peptide compounds **1** and **2** to give IB-01211 and its epimer **12b** was explored, and the results are compared here to those previously obtained for the macrocyclization of more flexible structures in the syntheses of YM-216391, telomestatin, and IB-01211. Lastly, the preliminary results of anti-tumor activity screening of the synthesized analogs are discussed. © 2013 Elsevier Science. All rights reserved

1. Introduction

Directly-linked 2,4-azoles are found in natural products that have interesting biological activities and fascinating structures. Numerous bis- and tris-oxazoles, as well as a few oxazole-thiazoles, have been isolated from marine organisms, whereas linked thiazole-containing natural products have generally been obtained from microorganisms.¹ Marine organism secondary metabolites such as ulapualides,² halichondramides, 3 $kabinamides³$ mycalolides,⁴ halishigamides,⁵ and jaspisamides,⁶ all contain a trisoxazole fragment. These compounds show a broad range of unusual biological activities.

Telomestatin, a potent telomerase inhibitor isolated from *Streptomyces anulatus* 3533-SV4⁸ that interacts specifically with the human telomeric intramolecular Gquadruplex without affecting DNA polymerases or reverse transcriptases, contains a novel macrocyclic structure comprised of seven linked oxazoles and one thiazoline unit. A related cyclopeptides, YM-216391,

———

containing only four oxazoles and one thiazole, has been isolated from *Streptomyces nobilis*. 9 **Figure 1.** Natural compounds with 2,4-concatenated azoles

* Corresponding author. Tel.: +34 93 403 7086; fax: +34 93 403 7126; e-mail: albericio@pcb.ub.es; malvarez@pcb.ub.es.

Scheme 1. Macrocyclization in the syntheses of YM-216391, tetomestatin, and IB-01112

A more recently discovered macrocyclic peptide, IB-01211 (Figure 1), has been isolated from the marinederived microorganism strain ES7-008, which is phylogenetically close to *Thermoactinomyces genus*. 10,11 It is strongly cytotoxic against several tumor cell lines, 12 and contains four oxazoles and one thiazole.

Concatenated azoles have been prepared following several strategies,^{1c} including cyclodehydration of peptides containing serine, threonine or cysteine followed by oxidation of the azoline to azole; the classical Hantzsch synthesis allowing a one-pot synthesis of the azole-ring; Pd(0) catalyzed cross-coupling reaction—despite difficulties in the preparation of several precursors; sequential $[3 + 2]$ cycloaddition of an appropriate rhodium carbene with nitriles, a new route to polyoxazoles from an ulapualide fragment; sequential Chan-type rearrangements of tertiary amides for the preparation of trisoxazoles; and finally, iterative oxazole assembly via base-promoted cyclization of alkynyl glycine derivatives prepared from the corresponding α chloroglycinates by reaction with alkynyl dimethylaluminum reagents. The total syntheses of YM-216391,¹³ telomestatin, 14° and IB-01211¹⁵ have recently been described. The three procedures possess a common feature, a macrocyclization of flexible precursors (Scheme 1).

The key step in the aforementioned synthesis of IB-01211 is macrocyclization by Hantzsch formation of the thiazole ring. Alternatively, the macrocyclization could be envisaged through formation of an amide bond between penta-azole peptides **1** and **2** as the last synthetic step (see Figure 2). Thus, with the aim of synthesizing IB-01211 and related derivatives, we studied the macrocyclization of penta-azole containing peptides. We describe here the preparation of several open chain IB-01211 derivatives,

including subsequent macrocyclization studies and antitumor activity screening.

Results and Discussion

Figure 2 Retrosynthetic analysis of IB-01211

The retro-synthetic strategy used for this work is shown in Figure 2. Disconnection of the amide bonds between the D-Val and the aminoethylidene (disconnection a), or between the D-Allo-Ile and the phenyloxazolcarbonyl (disconnection b), affords the penta-azolepeptides **1** or **2**, respectively. Both peptides possess a *tert*-butyl protected alcohol, which can be readily transformed into the exocyclic methylidene present in the natural compound. The strategy involving intermediate **1** was thought to be more favorable than that with **2**, which involves a coupling reaction of a hindered α -amine and a poorly reactive carboxylic group. A less convergent strategy through amide bond formation between the hindered Ile and Val residues was rejected. The common precursors

for both azole-peptides are the penta-azole **3** and the dipeptide **4.** Finally, assembly of the middle thiazole present in compound **3** was planned from appropriately functionalized bisoxazoles **5** and **6**.

Synthesis of penta-azole **3**

Preparation of **3** was attempted by transformation of peptide **7a** into thioamide **7b**, followed by cyclization and oxidation (Scheme 2A).¹⁶ Reaction of the acid **5a** and the amine **6a** using the general procedure described in the experimental section for peptide bond formation afforded **7a** in good yield.¹⁷ The bis-oxazole derivative **5a** was obtained by cyclization of the proper Ser peptide followed by oxidation of the resulting oxazolines, as previously described by our group.¹⁵ However, reaction of **7a** with the Lawesson reagent to produce the thioamide **7b** gave a complex mixture from which no product could be

Bisoxazole 10, containing the protected α -bromoketone residue for the Hantzsch synthesis, was obtained by two sequential oxazole-ring formations in order to minimize the amount of byproduct resulting from water elimination, and also because formation of a conjugated carbon-carbon double bond is favored by the presence of the phenyl ring. Cyclization and oxidation of dipeptide **8** 19 afforded an oxazole with a protected amino ethanol-substituent at position 2 of the ring, which was then deprotected with 95% trifluoroacetic acid (TFA) to give the amino alcohol **9**. Compound **9** was used in the following amide bond formation by reaction with bromopyruvic acid dimethyl acetal. Subsequent ring closure and oxidation then provided **10**. Elimination of the acetal protecting group of **10** by treatment with formic acid at reflux gave the bromoketone **6b** in quantitative yield. Penta-azole **3** was obtained in a 62% yield by Hantzsch thiazole synthesis using the bis-oxazolethioamide **5b** and the bis-oxazolyl α bromomethyl ketone **6b**.It was characterized by ¹H-

Scheme 2. Synthesis of penta-azole **3**

isolated.

After exploring a broad range of reagents and conditions to afford the thiazole moiety, including the use of a cysteine-containing building block, ¹⁸ we took a major shift in strategy: a classical Hantzsch synthesis using a thioamide and an α -bromoketone (Scheme 2B).

NMR, whereby the four singlets in the aromatic region, and the five aromatic protons, were taken as representative signals.

Macrocyclization reaction

Peptides **4** and **4a** were synthesized in solution from Boc-*D-allo*-Ile-OH and H-Val-OMe, and from Boc-*D-*Ile-OH and H-Val-OMe, respectively, using EDC·HCl/HOBt and DIEA as coupling reagents. The *N*-deprotected and *C*deprotected peptides were obtained by TFA treatment and

considered two options for macrocyclization, both of which implied preparation of peptides **1** and **2** (Scheme 3). The macrocyclization studies were performed with **4a**, as it is the cheaper of the two peptides.

Scheme 3. Macrocyclization of **1** and **2**

by methyl ester hydrolysis, respectively. With pentaazole **3** and epimeric peptides **4** and **4a** in hand, we

Compound **1a** was prepared in good yield by methyl ester hydrolysis of **3** with LiOH, followed by condensation with *N*-deprotected-**4a** using EDC·HCl and HOBt as activating agents, and *N,N-*diisopropylethylamine (DIEA). In parallel, **2a** was also obtained with good yields, by *N*and *O*-deprotection of **3** with 95% TFA, followed by condensation with the acid obtained in the saponification of **4a**, as described above. *C*- and *N*-deprotection of the linear precursors **1a** (R^1 = Me, R^2 = Boc) and **2a** (R^1 = Me, $R^2 = Boc$) were obtained *in situ* by methyl ester hydrolysis with LiOH followed by TFA treatment. Several macrocyclization trials were performed from *C*and *N*-deprotected **1a** ($R^1 = R^2 = H$) using the following activating agents: 1-[bis(dimethylamino)methylene]-1*H*-

1,2,3-triazolo[4,5-*b*] pyridinium hexafluorophosphate 3 oxide (HATU)/*N*-methylmorpholine (NMM), HATU/DIEA,¹² (benzotriazol-1yloxy)tris(pyrrolidino)phosphonium hexafluorophosphate (PyBOP)/HOBt/DIEA, and (7-azabenzotriazol-1 yloxy)tris(pyrrolidino)phosphonium hexafluorophosphate (PyAOP)/DIEA. No traces of cyclized compound were detected by either 1 H-NMR or MS. Likewise, no sign of **12b** was detected when the same cyclization conditions were applied to the methylidene derivative **11** $(R^1 = R^2 =$ H), obtained by dehydration of $2a (R^1 = Me, R^2 = Boc)$ with mesyl chloride (MsCl) in Et_3N followed by deprotection. Finally, macrocyclization of **1a** $(R^1 = R^2 =$ H) was achieved using pentafluorophenol (PfpOH) as activating agent. Thus, methyl ester hydrolysis of $1a(R¹)$ $=$ Me, $R^2 =$ Boc), conversion of the resulting acid into the pentafluorophenyl ester, *N*-deprotection²⁰ and, finally, macrocyclization by treatment with DIEA in a highly diluted THF solution gave the cyclic peptide **12a** (Scheme 3).

The exocyclic methylidene of **12b** was prepared by *O*deprotection and dehydration of **12a** using mesyl chloride as activating agent and *N,N,N-*triethylamine (TEA) in THF. It was characterized by 1 H-NMR, whereby the two methylidene singlets at 6.06 and 6.70 ppm, and the four singlets of the azole rings $(7.96, 8.20, 8.27,$ and 8.30 ppm), were taken as representative signals.²¹ As the yield of this macrocyclization was poor (less than 10%), we decided to test cyclization using a copper salt template.

The peptide-heterocycle **1** (R^1 = Me, R^2 = Boc) was obtained by condensation of the free carboxylic acid of **3** with the free amine of **4**, using the same reaction conditions described above for **1a**. Similar conditions were also used for the methyl ester hydrolysis of **1**, conversion of the resulting acid into the pentafluorophenyl ester followed by selective *N*deprotection. Macrocyclization was then attempted using DIEA and $CuSO₄$ in a highly dilute THF solution. Unfortunately, no trace of the cyclized product was found, and only the pentafluorophenyl ester **13** of the open chain peptide-heterocycle was obtained.

Biological activity

The cytotoxicity of the concatenated azoles was evaluated against a panel of three human tumor cell lines: A-549 lung carcinoma NSCL, HT-29 colon carcinoma, and MDA-MB-231 breast adenocarcinoma. A conventional colorimetric assay was run to estimate $GI₅₀$ values (i.e. the drug concentration at which 50% of cell growth is inhibited after 72 hours of continuous exposure to the test molecule), in which IB-01211 was for comparison.

The results obtained are shown in Table 1. A decrease of activity of compounds **3**, **1**, and **13** in relation with IB-01211 has been observed. Penta-azole **3** possess activity at micromolar (μM) concentration in the three cell lines, whereas **1** possess the same activity only in A-549 and **13** is inactive. The natural compound, IB-01211, shows activity in the three lines at nanomolar concentration. None of the peptides with *D*-Ile, **2a**, **12a**, and **12b**, has notable activity. These results demonstrate the importance of the configuration of the stereocenter to the activity of these compounds.

Table 1. Cell growth inhibition (GI_{50}) of synthetized azoles.

Cytotoxicity (GI⁵⁰ M)

a: A-549 lung carcinoma NSCL; b: HT-29 colon carcinoma cells; c: MDA-MB-231 breast adenocarcinoma; d: n.a. = not active at 10 μ g/mL; d: R¹ = Me, R² = Boc

Conclusions

Herein we have reported preparation of the functionalized polyazoles **6a**, **6b**, **7a**, the penta-azole **3**, and the peptideheterocycles **1**, **1a**, **2a**, **12a, 12b,** and **13**. Attempts at macrocyclization using the amino acids from the deprotection of **1** and **2** only led to a small amount of cyclized product, underscoring the need to change the synthetic strategy. Likewise, macrocyclization of *N*- and *C*-deprotected **1** using a copper chelating group also failed. While **1** and **2** possess an almost planar polyheterocyclic system of penta-azoles that separate the reactive groups the shown precursor of IB-01211 posses bigger conformational freedom which made the macrocyclization easier.

The open chain polyazoles with the same stereocenter configuration as the natural product inhibit growth of human carcinoma cells (GI_{50}) at micromolar concentrations, whereas the epi-analogs obtained with *D*-Ile are inactive.

Experimental section

General. Melting points (m. p.) were determined in a Büchi Melting Point B540 in open capillaries and are uncorrected. Reversed-phase analytical HPLC was performed on a Waters Alliance separation module 2695 using a Waters Xterra MS C_{18} column (150 x 4.6 mm, 5 m) and a Waters 996 PDA with a photodiode array detector with MeCN $(0.036\%$ TFA) and H₂O $(0.045\%$ TFA). Sonication was performed in a Branson ultrasound bath. Polarimetry studies were performed in a Perkin Elmer 241 polarimeter. ¹H NMR and ¹³C NMR spectra were recorded on a Varian Mercury 400 MHz and Varian 500 MHz spectrometers. Multiplicity of the carbons was assigned with DEPT and gHSQC experiments. Usual abbreviations for off-resonance decoupling have been used here: (s) singlet, (d) doublet, (t) triplet, and (q) quartet. The same abbreviations have also been used for the multiplicity of signals in 1 H-NMR, plus: (m) multiplet, (dd) double doublet, (bs) broad singlet, and (bd) broad doublet. Spectra were referenced to appropriate residual

solvent peaks $(CDCl₃$ and $d₆-DMSO)$. CI-MS were measured in a Hewlett-Packard model 5890A with ammonia (NH_3) . MALDI-TOF and ES-MS were performed in a PerSeptive Biosystems Voyager DE RP, using an ACH matrix, for the former, and a Waters alliance 2795 HPLC equipped with a 2487 UV-Vis detector and coupled to a ZQ electrospray mass detector, for the latter. The samples were run with MeCN (0.07% $HCO₂H$) and $H₂O$ (0.1% $HCO₂H$). HRMS were performed on a Bruker Autoflex high resolution mass spectrometer by the Mass Spectrometry Service of the University of Santiago de Compostela.¹

Syntheses of compounds **4** ($[\alpha]_D$ +22.3 (*c* 0.56, CHCl₃), **5a, 5b, 8, 9,** and **10** have previously been reported.¹⁵

Peptide bond formation. Sample procedure:

D-**Boc-Ile-***L***-Val-OMe (4a)** *D*-Boc-Ile-OH. ¹/₂ H₂O (795) mg, 3.31 mmol), EDC•HCl (698 mg, 3.64 mmol), HOBt (491 mg, 3.64 mmol), and DIEA (1.21 mL, 7.11 mmol) were added to a solution of *L*-H-Val-OMe•HCl (550 mg, 3.31 mmol) and dry CH₂Cl₂ (28 mL) at 0 °C. The mixture was stirred at room temperature for 20 h. The organic solution was washed with 5% $NaHCO₃$ and NH₄Cl, dried and concentrated to give the title compound (1.02 g, 90%) as a white solid, m. p. 105-107 °C. $[\alpha]_D$ +16.6 (*c* 0.7, CHCl₃). ¹H NMR (CDCl₃, 400 MHz) δ 0.88-0.93 (m, 12H); 1.09-1.21 (m, 1H); 1.42 (s, 9H); 1.49-1.55 (m, 1H); 1.86 (m, 1H); 2.11-2.20 (m, 1H); 3.72 (s, 3H); 3.92-3.95 (m, 1H); 4.51-4.55 (m, 1H); 5.04-5.06 (d, 6.8 Hz, 1H); 6.37-6.39 (d, 6.8 Hz, 1H). ¹³C NMR (CDCl₃, 100 MHz) δ 11.31 (q); 15.5 (q); 17.71 (q); 18.9 (q); 24.78 (t); 28.27 (q); 31.20 (d); 36.91 (d); 52.09 (q); 57.01 (d); 59.37 (d); 79.88 (s); 155.73 (s); 171.55 (s); 172.08 (s). MS (CI): m/z 345 (M + 1, 100), 289 (M, 29), 245 (M, 43). HRMS m/z calcd. for $C_{17}H_{32}N_2NaO_5$ (M + Na) 367.2203, found 367.2206.

Hydrolysis of methyl esters 2M LiOH (9 mmol) was added to a solution of methyl ester (3 mmol) in THF-H₂O-MeOH (50:6:0.2, 14 mL), and the reaction mixture was stirred at room temperature for 1 h. The pH of was brought to 3 by addition of 1M HCl, and then the solution was extracted with EtOAc. The organic layer was dried over $MgSO_4$ and concentrated to afford the acid as a white solid.

Deprotection with TFA

———

Method A. Elimination of N-Boc and O-tert-Bu protecting groups. 95% TFA (5 mL) was added to a solution of the di-protected (*N*-Boc and *O*-*tert-*Bu) compound (2.38 mmol), and the solution was stirred at room temperature for 5 h. The TFA was then removed under reduced pressure, and the crude material was used for subsequent chemistry without further purification.

Method B. Selective elimination of N-Boc protecting group. A solution of *N*-Boc protected compound (27 μ mol) in TFA-CH₂Cl₂ (1 mL, 3:7) was stirred at room temperature for 1 h. The TFA and the solvent were removed, and the crude material was used for subsequent chemistry without further purification.

Peptide 1 (\mathbb{R}^1 = **Me,** \mathbb{R}^2 = **Boc).** Coupling of the free carboxylic acid of **3** to the free amine of **4** using the general procedure for peptide formation provided **1** (70%) as a white solid, m. p. (MeCN) 186-188 °C. $[\alpha]_D +4.0$ (*c* 0.56, CHCl₃). ¹H NMR (CDCl₃, 400 MHz) δ 0.88-1.08 (m, 12H); 1.12 (s, 9H); 1.22-1.34 (m, 1H); 1.48 (s, 9H); 1.52-1.62 (m, 1H); 2.11-2.25 (m, 2H); 3.70 (s, 3H); 3.74 (dd, *J*= 4.4, 9.2 Hz, 1H); 4.57 (dd, *J*= 5.2 Hz and 8.6 Hz, 1H); 4.65 (dd, *J*= 5.6 Hz and 8.8 Hz, 1H); 5.06-5.14 (m, 1H); 5.62 (bs, 1H); 6.56 (bs, 1H); 7.42-7.53 (m, 3H), 7.81 (bs, 1H); 8.24 (s, 1H); 8.29 (s, 1H); 8.33-8.38 (m, 2H); 8.42 (s, 1H); 8.47 (s, 1H). ¹³C NMR (CDCl₃, 100 MHz) δ 11.6 (q); 14.7 (q); 17.8 (q); 18.9 (t); 27.3 (q); 28.3 (q); 29.7 (t); 31.2 (d); 37.1 (d); 50.1 (d); 52.1 (q); 57.1 (d); 57.2 (t); 62.9 (t); 73.7 (s); 80.2 (s); 122.0 (d); 126.6 (s); 128.3 (d); 128.5 (d); 129.6 (s); 129.9 (s); 130.2 (d); 131.0 (s); 136.2 (d); 136.3 (s); 139.3 (d); 143.4 (d); 151.9 (s); 153.0 (s); 155.3 (s); 155.6 (s); 158.2 (s); 161.2 (s); 161.3 (s); 164.8 (s); 171.0 (s); 172.0 (s). MS (MALDI): m/z 937.33 (M + 23, 100). HRMS m/z calcd. for $C_{45}H_{54}N_8NaO_{11}S$ (M + Na) 937.3525, found 937.3527.

Peptide 1a $(\mathbb{R}^1 = \mathbb{M}\mathbb{e}, \mathbb{R}^2 = \mathbb{B}\mathbb{O}\mathbb{e})$. Coupling of the free carboxylic acid of **3** to the free amine of **4a** using the general procedure for peptide formation provided **1a** (70%) as a white solid, m. p. (MeCN) 188-190 °C. $[\alpha]_D$ +3.4 (*c* 0.64, CHCl₃). ¹H NMR (CDCl₃, 400 MHz) δ 0.87-1.02 (m, 12H); 1.03-1.06 (m, 1H); 1.11 (s, 9H); 1.21-1.33 (m, 1H); 1.47 (s, 9H); 1.60-1.75 (m, 1H); 2.07- 2.24 (m, 1H); 3.69 (s, 3H); 3.71-3.76 (m, 1H); 3.84-3.91 (m, 1H); 4.47-4.61 (m, 2H); 5.04-5.14 (m, 1H); 5.60 (bs, 1H); 6.50 (bs, 1H); 7.42-7.53 (m, 3H), 7.82 (bs, 1H); 8.23 (s, 1H); 8.28 (s, 1H); 8.32-8.38 (m, 2H); 8.4 (s, 1H); 8.47 (s, 1H). ¹³C NMR (CDCl₃, 100 MHz) δ 11.4 (q); 15.7 (q); 17.6 (q); 19.0 (q); 25.1 (t); 27.3 (q); 28.3 (q); 31.2 (d); 36.9 (d); 50.1 (d); 52.1 (q); 57.2 (d); 58.1 (d); 62.9 (t); 73.7 (s); 80.2 (s); 122.0 (d); 126.7 (s); 128.4 (d); 128.5 (d); 129.7 (s); 130.0 (s); 130.2 (s); 131.1 (s); 136.2 (d); 136.4 (s); 139.3 (d); 143.4 (d); 151.8 (s); 153.1 (s); 155.3 (s); 155.6 (s); 158.2 (s); 161.1 (s); 161.2 (s); 164.8 (s); 170.7 (s); 171.9 (s). (MALDI): m/z 937.31 (M + Na, 100). HRMS m/z calcd. for $C_{45}H_{54}N_8NaO_{11}S$ (M+Na) 937.3525, found 937.3529.

Peptide-heterocycle 2a $(\mathbb{R}^1 = \mathbb{M}\mathbb{e}, \mathbb{R}^2 = \mathbb{B}\mathbb{0}\mathbb{c})$. The free amino-alcohol, obtained by *N*- and *O*-deprotection of **3**

using 95% TFA (2 mL), was coupled to the acid of **4a** following the general procedure for peptide formation to provide **2a** (50 %) as a white solid, m. p. (MeCN) 224- 226 °C. ¹H NMR (DMSO, 400 MHz) δ 0.76-0.88 (m, 12H); 0.95-1.10 (m, 1H); 1.37 (s, 9H); 1.45-1.46 (m, 1H); 1.63-1.72 (m, 1H); 1.94-2.06 (m, 1H); 3.57-3.63 (m, 1H); 3.76-3.84 (m, 1H); 3.86 (s, 3H); 3.97-4.07 (m, 1H); 4.11- 4.17 (m, 1H); 5.14-5.21 (m, 1H); 5.30-5.36 (bs, 1H); 6.77 (bs, 1H); 7.55-7.61 (m, 3H); 7.73-7.80 (bs, 1H); 8.06-8.11 (m, 2H); 8.67 (s, 1H); 9.03 (s, 1H); 9.08 (s, 1H); 9.2 (s, 1H). ¹H NMR (DMSO, 400 MHz) δ 0.76-0.88 (m, 12H); 0.95-1.10 (m, 1H); 1.37 (s, 9H); 1.45-1.46 (m, 1H); 1.63- 1.72 (m, 1H); 1.94-2.06 (m, 1H); 3.57-3.63 (m ,1H); 3.76- 3.84 (m, 1H); 3.86 (s, 3H); 3.97-4.07 (m, 1H); 4.11-4.17 (m, 1H); 5.14-5.21 (m, 1H); 5.30-5.36 (bs, 1H); 6.77 (bs, 1H); 7.55-7.61 (m, 3H); 7.73-7.80 (bs, 1H); 8.06-8.11 (m, 2H); 8.67 (s, 1H); 9.03 (s, 1H); 9.08 (s, 1H); 9.2 (s, 1H). 13° C NMR (CDCl₃, 100 MHz) δ 10.8 (q); 15.3 (q); 17.9 (q); 19.0 (q); 29.0 (t); 29.8 (q); 36.27 (d); 40.6 (d); 50.2 (d); 52.0 (q); 57.7 (d); 58.7 (d); 60.5 (t); 69.7 (s); 78.8 (s); 123.4 (d); 126.1 (s); 127.2 (s); 128.2 (2d); 128.5 (2d); 129.8 (s); 129.9 (s); 130.6 (d); 135.4 (d); 139.1 (s); 141.2 (d); 141.3 (s); 141.5 (s); 142.6 (d); 152.8 (s); 154.3 (s); 155.2 (s); 157.4 (s); 160.4 (s); 161.6 (s); 163.6 (s); 164.1 (s); 172.6 (s). (MALDI): m/z 881.07 (M + Na, 78); 783 (95); 759 (100). HRMS m/z calcd. for $C_{41}H_{46}N_8NaO_{11}S$ (M + Na) 881.2899, found 881.2886.

Methyl 2'-{2-[2'-(2-*tert***-butoxy-1-***tert***-butoxycarbonylaminoethyl)-[2, 4']bis-oxazol-4-yl]thiazol-4-yl}-5 phenyl[2,4']bis-oxazolyl-4-carboxylate (3)** Bromoketone **6b** (64 mg, 0.16 mmol) was added to a suspension of **5b** $(45 \text{ mg}, 0.11 \text{ mmol})$ and NaHCO₃ $(29 \text{ mg}, 0.35 \text{ mmol})$ in THF (2 mL). The mixture was stirred for 8 h at room temperature, at which point it was filtered over alumina and washed with CH_2Cl_2 -MeOH (4:1). The organic layer was concentrated to give a crude material which was dissolved in dry THF (2 mL) and cooled to -10 °C. Lutidine $(0.10 \text{ mL}, 0.87 \text{ mmol})$ and TFAA $(46 \mu L, 0.33 \text{ m})$ mmol) were then added to the solution, and the reaction mixture was stirred at room temperature overnight. Concentration *in vacuo* gave a brown residue, which was purified by silica gel chromatography. Elution with $CH_2Cl_2/EtOAc$ (9:1) gave 3 (47.8 mg, 62%) as a white solid, m. p. (MeCN) 216-218 °C. [α]_D -12.2 (*c* 0.51, CHCl₃). ¹H NMR (CDCl₃, 400 MHz) δ 1.11 (s, 9H); 1.47 (s, 9H); 3.73 (dd, *J* = 4.4 and 9.2 Hz, 1H); 3.86-389 (m, 1H); 3.98 (s, 3H); 5.06-5.12 (m, 1H); 5.58-5.60 (d, *J* = 8.4 Hz, 1H); 7.49-7.52 (m, 3H); 8.17-8.20 (m, 2H); 8.22 (s, 1H); 8.28 (s, 1H); 8.46 (s, 1H); 8.49 (s, 1H). ¹³C NMR $(CDCl₃, 100 MHz)$ δ 27.29 (q); 28.32 (q); 29.69 (q); 50.14 (s); 52.43 (d), 62.94 (t); 73.75 (s); 121.92 (d); 126.46 (s); 127.68 (s); 128.44 (d); 128.66 (d); 129.95 (s); 130.63 (s); 131.0 (s); 136.17 (d); 136.4 (s); 139.30 (s); 139.37 (d); 143.51 (d); 153.03 (s); 155.61 (s); 158.01 (s); 161.17 (s); 162.35 (s); 164.79 (s). MS (FAB): m/z 720.1 $(M + 18, 65)$, 589 $(M + 1, 100)$. HRMS m/z calcd. for $C_{34}H_{34}N_6NaO_9S$ (M + Na) 725.200, found 725.1993.

Methyl 2'-(2-*tert***-butoxy-1-***tert***-butoxycarbonylaminoethyl)-5-phenyl[2, 4']bis-oxazolyl-4-carboxylate (6)**

The free amine of **9** was coupled with Boc-L-Ser(tBu)- OH following the general procedure for peptide formation. Cyclization using $DAST/K_2CO_3$ and oxidation with DBU-CC l_4 in Pyr and ACN as solvents gave a crude which was purified by column chromatography on silica gel. Elution with hexane/ EtOAc (3:1) gave **6** (50%) as a solid, m. p. (MeCN) 66-68 °C. ¹H NMR (CDCl₃, 400 MHz) δ 1.1 (s, 9H); 1.47 (s, 9H); 3.72 (dd, *J* = 4.0 and 9.2 Hz, 1H); 3.83-3.89 (m, 1H); 3.96 (s, 3H); 5.02-5.12 (m, 1H); 5.6 (bs, 1H); 7.45-7.51 (m, 3H); 8.10-8.15 (m, 2H); 8.33 (s, 1H). ¹³C NMR (CDCl₃, 100 MHz) δ 27.24 (q); 28.30 (q); 50.10 (d); 52.32 (q); 62.89 (t), 73.70 (s); 80.13 (s); 126.50 (s); 127.60 (s); 128.40 (2d); 128.56 (2d) 129.80 (s); 130.50 (s); 139.41 (d); 153.31 (s); 155.34 (s); 162.35 (s); 164.50 (s). MS (ES) m/z 486.53 (M + 1, 100). HRMS m/z calcd. for $C_{25}H_{32}N_3O_7$ 486.2235, found 486.2222.

Methyl 2'-(2-bromoacetyl)-5-phenyl[2, 4']bis-oxazolyl-4-carboxylate (6b). A mixture of 10 (200 mg, 452 μ mol) and formic acid (3.5 mL) was refluxed for 2 h, and then cooled to room temperature. The organic solution was poured into an aqueous solution of $NAHCO₃$ and extracted with CH₂Cl₂. The organic layer was dried over $MgSO₄$ and concentrated to give **6b** (164 mg, 91%) as a yellow oil. ¹H NMR (CDCl₃, 400 MHz) δ 3.98 (s, 3H); 4.68 (s, 2H); 7.51-7.53 (m, 3H); 8.13-8.16 (m, 2H); 8.58 (s, 1H). 13 C NMR (CDCl₃, 100 MHz) δ 30.3 (t); 52.5 (q); 126.1 (s); 127.8 (s); 128.5 (2d); 128.6 (2d); 130.9 (d); 131.7 (s); 142.5 (d); 151.3 (s); 155.9 (s); 156.1 (s); 162.1 (s); 178.7 (s). MS (ES): m/z 408.1 (MBr⁷⁹ + 18, 61); 410.1 (MBr⁸¹ + 18, 62); 391 (MBr⁷⁹, 62); 393 (MBr⁸¹, 63); 313 (100). HRMS m/z calcd. for $C_{16}H_{12}BrN_2O_5$ 390.9924, found 390.9911.

Peptide 7a. The free carboxylic acid of **5** (40 mg, 0.10 mmol) was coupled to **6a** (33 mg, 0.10 mmol) using the general procedure for peptide formation to provide **7a** (31 mg, 43%) as a yellow solid, m. p. (MeCN) $98-100$ °C. α _D –13.4 (*c* 0.35, CHCl₃). ¹H NMR (CDCl₃, 400 MHz) 1.09 (s, 18H); 1.45 (s, 9H); 3.6-3.70 (m, 1H); 3.79-3.84 (m, 1H); 3.93 (s, 3H); 4.09-4.13 (m, 1H); 4.28-4.32 (m, 1H); 5.01-5.09 (m, 1H); 5.55-5.59 (m, 1H); 7.43-7.48 (m, 4H); 7.95 (d, *J* = 8.0 Hz, 1H) 8.07-8.10 (m, 2H); 8.20 (s, 1H); 8.26 (s, 1H); 8.33 (s, 1H).¹³C NMR (CDCl₃, 100 MHz) δ 27.2 (q); 28.2 (q); 48.5 (d); 50.1 (d); 52.4 (q); 62.9 (t); 63.2 (t); 73.7 (s); 80.3 (s); 126.6 (s); 127.8 (s); 128.7 (d); 128.8 (d); 129.9 (s); 130.9 (d); 136.5 (s); 139.5 (d); 140.2 (d); 141.8 (d); 153.0 (s); 153.6 (s); 155.6 (s); 155.8 (s); 155.9 (s); 160.5 (s); 160.8 (s); 163.2 (s); 164.2 (s); 171.3 (s). MS (MALDI-TOF) 745 (M + K); 729.24 (M + Na, 100); 707.64 (60); 674 (60). HRMS m/z calcd for $C_{34}H_{38}N_6NaO_{11}$ (M + Na) 729.2491, found 729.2480.

Peptide 12a. A solution of the carboxylic acid (125 mg, 190 mol) obtained from **1a** in dry THF-DMF (120 mL and 5 mL) was cooled to 0 $°C$. EDC·HCl (190 mg, 99 μ mol), DIEA (167 μ L, 99 μ mol), and Pfp-OH (194 mg, 1.05 mmol) were then added, and the mixture was stirred at room temperature for 20 h. The solvents were removed *in vacuo*, and the residue was diluted with CH_2Cl_2 and washed with 5% aqueous NaHCO₃ and aqueous NH₄Cl. The organic solution was dried and concentrated, and the residue was diluted with TFA-CH₂Cl₂ (6 mL and 14 mL). The solution was stirred for 1 h at room temperature, and then the TFA was removed, and the crude material was dissolved in THF (250 mL). DIEA (479 μ L, 2.82 mmol) was added, and the mixture was stirred for 96 h at room temperature. The solvents were removed, and the residue was then washed with MeOH. The residue obtained after removing the solvent was purified by preparative HPLC. A gradient of $H₂O$ (0.045% TFA) / MeCN (0.036% TFA) from 6:4 until 1:9 in 15 min gave a white solid (10 mg, 9%, rt = 7.47 min), m. p. (MeCN) 206-208 °C. $[\alpha]_D$ – 48.3 (*c* 0.35, DMSO). ¹H NMR (d₆-DMSO, 500 MHz) δ 0.82 -0.95 (m, 12H); 1.02-1.06 (m, 2H); 1.22 (s, 9H); 1.49-1.57 (m, 1H); 1.85-1.91 (m, 1H); 3.85-3.94 (m, 1H); 3.97-4.04 (m, 1H); 4.17-4.24 (m, 1H); 4.62-4.71 (m, 1H); 5.16-5.21 (m, 1H); 5.28-5.35 (m, 1H); 7.53-7.58 (m, 3H); 8.01 (bs, 1H); 8.03 (bs, 1H); 8.27-8.37 (m, 2H); 8.49 and 8.62 (2bs, 1H); 8.73 (s, 1H); 9.04 (s, 1H); 9.12 (s, 1H); 9.23 (s, 1H); 10.17 (bs, 1H). ¹³C NMR (d₆-DMSO, 125 MHz) 11.3 (q); 14.2 (q); 15.6 (q); 18.7 (q); 18.8 (q); 24.5 (t); 38.1 (d); 50.5 (d); 56.5 (d); 57.7 (d); 60.9 (t); 67.8 (d); 123.9 (d); 128.4 (2d); 128.9 (2d); 130.5 (d); 138.0 (d); 141.8 (d); 141.9 (d). MS (MALDI): m/z 782.19 (M, 40).

Peptide 12b. Peptide **12a** (9 mg, 0.0114 mmol) was deprotected with 95% TFA (1 ml). The crude was dissolved in dry THF (1 mL), and the solution was cooled to 0 °C, TEA (15.82 µL, 0.114 mmol) and MsCl (4.4 µL, 0.057 mmol) were added drop-wise. The resulting solution was stirred for 2 h at 0° C, then washed with NH4Cl and water, dried, and concentrated. The solvents were removed and the residue wa**s** washed with MeCN to give a white solid (7.4 mg, 90%), m. p. (MeCN) 147-149 ^oC. $[\alpha]_D$ +12.4 (*c* 0.15, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) δ 0.79-0.92 (m, 12H); 0.99-1.09 (m, 2H); 1.41-1.45 (m, 1H); 1.52-1.55 (m, 1H); 3.65-3.75 (m, 2H); 3.78-3.89 (m, 1H); 6.06 (s, 1H); 6.70 (s, 1H); 7.43-7.52 (m, 4H); 7.96 (s, 1H); 8.20 (s, 1H); 8.27 (s, 1H); 8.30 (s, 1H); 8.31 (bs, 1H); 8.36-8.37 (m, 2H); 8.47 (bs, 1H). ¹³C NMR $(CDCl₃, 125 MHz)$ δ 11.29 (q); 13.9 (q); 15.2 (q); 18.7 (q); 25.2 (t); 30.2 (d); 30.7 (d); 53.6 (d); 63.2 (d); 111.7 (t); 119.5 (d) ; (d); 127.9 (d) ; 128.4 (2d); 130.2 (2d); 137.7 (d); 139.0 (d); 139.4 (d). MS (MALDI): m/z 749.3 $(M + Na, 100)$, 765.3 $(M + K, 47)$.

Peptide 13 A solution of carboxylic acid (150 mg, 0.166 mmol) obtained from **1** in dry THF-DMF (10 mL and 2 mL) was cooled to 0 ºC. EDC·HCl (0.22 mg, 1.16 mmol), DIEA (0.19 ml, 1.16 mmol) and Pfp-OH (0.229 mg, 1.24 mmol) were then added, and the reaction mixture was stirred at room temperature for 20 h. The solvents were removed *in vacuo*, and the residue was diluted with CH_2Cl_2 , and then washed with 5% aqueous NaHCO₃ and aqueous NH4Cl. The organic solution was dried and concentrated, and the residue was diluted in $TFA-CH_2Cl_2$ (1 mL/3 mL). The solution was stirred for 1 h at room temperature, and then the TFA was removed. The crude material was dissolved in THF (300 mL). DIEA (0.28 ml, 1.66 mmol) and $CuSO₄$ (132 mg, 0.83 mmol) were added,

and the mixture was stirred for 72 h at room temperature. The solvents were removed, and the crude was purified by silica gel chromatography $(9:1 \text{ CH}_2\text{Cl}_2$ - MeOH) to give 13 (58.8 mg, 37%) as a yellow solid, m. p. (MeCN) 140-142 ^oC. $[\alpha]_D$ +7.7 (*c* 0.39, CHCl₃). ¹H NMR (CDCl₃, 400 MHz) δ 0.94-1.09 (m, 12H); 1.13 (s, 9H); 1.20-1.32 (m, 1H); 1.51-1.61 (m, 1H); 2.08-2.26 (m, 1H); 2.32-2.43 (m, 1H) 3.77-3.83 (m, 1H); 3.96-4.01 (m, 1H); 4.55-4.65 (m, 1H); 4.82-4.91 (m, 1H); 5.38-5.45 (m, 1H); 6.67 (d, *J= 8.4 Hz*, 1H); 6.81 (d, *J= 8.4 Hz*, 1H); 7.42-7.47 (m, 3H); 8.24 (s, 1H); 8.29-8.30 (m, 2H); 8.31 (s, 1H); 8.40 (s, 1H); 8.47 (s, 1H). ¹³C NMR (CDCl₃, 100 MHz) δ 11.4 (q); 14.7 (q); 17.6 (q); 18.9 (q); 26.2 (t); 27.3 (q); 30.9 (d); 36.4 (d); 48.9 (d); 57.2 (d); 57.5 (d); 61.8 (t); 74.3 (s); 122.3 (d); 125.1 (s); 126.4 (s); 126.5 (s); 128.3 (2d); 128.5 (2d); 128.7 (s); 129.4 (s); 130.1 (s); 130.3 (d); 130.9 (s); 131.5 (s); 136.3 (s); 136.4 (d); 139.3 (d); 139.8 (d); 143.3 (s); 151.8 (s); 153.2 (s); 155.1 (s); 156.8 (s); 157.2 (s); 158.2 (s); 161.0 (s); 161.5 (s); 162.0 (s); 167.7 (s); 171.5 (s). ¹⁹F NMR (CDCl₃, 400 MHz) δ 84.0 (s). (MALDI): m/z 1006.7 (M + K, 100).

Acknowledgments

This study was partially supported by CICYT (BQU 2003-00089 and BQU2006-03794), *Generalitat de Catalunya*, and the Barcelona Science Park. We gratefully acknowledge PharmaMar S.L. for performing the preliminary biological tests. D.H. thanks the *Ministerio de Educación y Ciencia* for a doctoral fellowship, and E.R. thanks the *Principado de Asturias* for a postdoctoral fellowship.

References

- 1. Recent revisions about the chemistry and properties can be found in: Roy, R. S.; Gehring, A. M.; Milne, J. C.; Belshaw P. J.; Walsh C. T. *Nat. Prod. Rep.* **1999**, *16*, 249; (b) Yeh, V. S. C. *Tetrahedron*, **2004**, *60*, 11995; (c) Riego, E.; Hernández, D.; Albericio, F.; Álvarez, M. *Synthesis*, **2005**, 1907.
- 2. Roesener, J. A.; Scheuer, P. J. J. Am. Chem. Soc. 1986, 108, 846. For absolute stereochemistry of ulapualide A, see: Allingham, J. S.; Tanaka, J.; Marriott, G.; Rayment, I. *Org. Lett.* **2004**, *6*, 597.
- 3. Matsunaga, S.; Fusetani, N.; Hashimoto, K.; Koseki, K.; Noma, M.; Noguchi, H.; Sankawa U. *J. Org. Chem.* **1989**, *54*, 1360.
- 4. Fusetani, N.; Yasumuro, K.; Matsunaga, S.; Hashimoto, K. *Tetrahedron Lett.* **1989**, *30*, 2809. (b) Rashid, M. A.; Gustafson, K. R.; Cardeilina II, J. H.; Boyd, M. R. *J. Nat. Prod.* **1995**, *58*, 1120. (c) Matsunaga, S.; Nogata, Y.; Fusetani, N*. J. Nat. Prod.* **1998**, *61*, 663. (d) Matsunaga, S.; Sugawara, T.; Fusetani, N. *J. Nat. Prod.* **1998**, *61*, 1164. (e) Phuwapraisirisan, P.; Matsunaga, S.; van Soest, R. W. M.; Fusetani, N. *J. Nat. Prod.* **2002**, *65*, 942.
- 5. Kobayashi, J.; Tsuda, M.; Fuse, H.; Sasaki, T.; Mikami, Y. *J. Nat. Prod.* **1997**, *60*, 150.
- 6. Kobayashi, J.; Murata, O.; Shigemori, H.; Sasaki, T. *J. Nat. Prod.* **1993**, *56*, 787.
- 7. Michael, J. P.; Pattenden, G. *Angew. Chem. Int. Ed. Engl.* **1993**, *32*, 1.
- 8. Shin-ya, K.; Wierzba, K.; Matsuo, K.; Ohtani, T.; Yamada, Y.; Furihata, K.; Hayakawa, Y.; Seto, H. *J. Am. Chem. Soc*. **2001**, *123*, 1262. (b) Kim, M.-Y.; Vankayalapati, H.; Shinya, K.; Wierza, K.; Hurley, L. H. *J. Am. Chem. Soc.* **2002**, *124*, 2098.
- 9. Hayata, A.; Takebashi, Y.; Nagai, K.; Hiramoto, M. Jpn. Kokai Tokkyo Koho JP11180997-A, 1999; *Chem. Abstr.* **1999**, *131*, 101.
- 10. Romero, F.; Malet, L.; Cañedo, M. L.; Cuevas, C.; Reyes, F. WO 2005/000880 A2, 2005.
- 11. Same structure was proposed for Merchercharmycin A isolated from a marine-derived Thermoactinomices sp. by Kanoh, K.; Matsuo, Y.; Adachi, K.; Imagawa, H.; Nishizawa, M.; Shizuri, Y. *J. Antibiot.* **2005**, *58*, 289.
- 12. Cañedo, M. L.; Martínez, M, Sánchez, J. M.; Fernández-Puentes J. L.; Malet, L.; Pérez J.; Romero, F.; García, L. F. 4th Eur. Conference on Marine Natural Products, Paris 2005, poster 54.
- 13. Deeley, J.; Pattenden, G. *Chem. Commun*. **2005**, 797.
- 14. T. Doi, M. Yoshida, K. Shin-ya, T. Takahashi, *Org. Lett.* **2006**, *8*, 4165. A penta-azole related to telomestatin has been recently described by Marson, M, C.; Saadi, M. *Org. Biomol. Chem*, **2006**, *4*, 3892.
- 15. Hernández, D.; Vilar, G.; Riego, E.; Cañedo, L. M.; Cuevas, C.; Albericio, F.; Álvarez, M. *Org. Lett.* **2007**, *9*, 809-811.
- 16. Kaleta, Z.; Tárkányi, G.; Gömöry, A.; Kálmán, F.; Nagy, T.; Soós, T. *Org. Lett.* **2006**, *8*, 1093. (b) Sowinski, J. A.; Toogood, P. L. *J. Org. Chem.* **1996**, *61*, 7671.
- 17. **7a** was obtained as unic stereoisomer as indicate its H and C-NMR maintaining the configuration of starting *L*-Ser used in the preparation of **5a** and **6a**.
- 18. *S*-Trityl cysteine containing peptide was treated with TiCl⁴ and Ph₃PO/Tf₂O for a concomitant removal of the trityl group and cyclization following the methods developed by Kelly and co-workers; (a) Raman, P.; Razavi, H.; Kelly, J. W. *Org. Lett.* **2000**, *2*, 3289; (b) You, S.-L.; Razavi, H.; Kelly, J. W. *Angew. Chem. Int. Ed*. **2003**, *42*, 83; (c) You, S.-L.; Kelly, J. W. *J. Org. Chem.* **2003**, *68*, 9506.
- 19. The peptide **8** was prepared as a stereoisomers mixture from *N*-Boc-*O*-*t*-Bu-*L*-SerOH and *D,L*-PhSerOMe as it is described in ref 15. The three stereocenters of **8** were lost in the bis-oxazole **10**.
- 20. Selective deprotection of the *N*-Boc in front of *O*-*t*Bu group was afforded by treatment with a solution of TFA in $CH₂Cl₂$ (30:70) during 1 h at room temperature.
- 21. Assays of coelution in the HPLC of the natural product and the obtained macrocyclic peptide demonstrated that both compounds were different. A sample of IB-01211 was kindly supplied by PharmaMar S.L.