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Characterization of photovoltaics with In₂S₃ nanoflakes/*p*-Si heterojunction

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Abstract

We demonstrate that heterojunction photovoltaics based on hydrothermal-grown \ln_2S_3 on p-Si were fabricated and characterized in the paper. An n-type \ln_2S_3 nanoflake-based film with unique 'cross-linked network' structure was grown on the prepared p-type silicon substrate. It was found that the bandgap energy of such \ln_2S_3 film is 2.5 eV by optical absorption spectra. This unique nanostructure significantly enhances the surface area of the \ln_2S_3 films, leading to obtain lower reflectance spectra as the thickness of \ln_2S_3 film was increased. Additionally, such a nanostructure resulted in a closer spacing between the cross-linked \ln_2S_3 nanostructures and formed more direct conduction paths for electron transportation. Thus, the short-circuit current density (*Jsc*) was effectively improved by using a suitable thickness of \ln_2S_3 . The power conversion efficiency (PCE, η) of the AZO/In₂S₃/textured p-Si heterojunction solar cell with 100-nm-thick \ln_2S_3 film was 2.39%.

Keywords: Heterojunction; Nanoflake; In₂S₃

Background

Indium sulfide (In₂S₃) is one of the important semiconductor materials with direct bandgap and attracts intense interest due to its high photosensitivity, photoconductivity, and photocatalyst characteristics at ambient conditions [1-3]. In In_2S_3 , there are three polymorphic forms: defect cubic structure α -In₂S₃, defect spinel structure β -In₂S₃, and higher-temperature-layered structure γ -In₂S₃ [4]. Among them, β -In₂S₃ is an n-type semiconductor with superior photoelectric conversion function that can be employed in near-infrared to ultraviolet regions of solar energy absorption [5]. Hence, we may expect that β -In₂S₃ will act as a good absorber in heterojunction thin film solar cells [6]. On the other hand, In_2S_3 is a nontoxic semiconductor material which also offers potential advantage in process without Cd and Pb. A cell with ITO/ PEDOT:PSS/In₂S₃:P3HT/Al structure has been fabricated by Jia et al. [7], which showed the short-circuit current density (Jsc) of 0.68 mA cm⁻² and a power conversion efficiency of 0.04%.

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In recent years, In₂S₃ thin films have been grown by a variety of deposition techniques such as chemical bath deposition (CBD) [8], thermal evaporation [9], solvothermal synthesis [10], and atomic layer chemical vapor deposition (ALCVD) [11]. Among them, chemical bath deposition is a desirable method because of its low cost, arbitrary substrate shapes, simplicity, and can be easily prepared in large areas. There have been many reports for the heterojunction solar cell with CBD grown In₂S₃. For example, In₂S₃ was used for the n-type buffer layer of CIGS solar cells [12]. Crystalline silicon solar cells are presently the predominant photovoltaic devices among various solar cells due to their higher photovoltaic conversion efficiency, and long-term stability [13]. Recently, Abd-El-Rahman and Darwish et al. reported a p-Sb₂S₃/ n-Si heterojunction photovoltaic that was fabricated by using thermal evaporation technique [14], which showed $Jsc = 14.53 \text{ mA cm}^{-2}$, fill factor = 0.32, and $\eta = 4.65\%$.

In this study, the In_2S_3 thin films were deposited on a p-type silicon substrates via chemical bath deposition route. To our knowledge, works on In_2S_3 film deposited on textured Si-based solar cell by CBD are few. In addition, the advantages of chemical bath deposition process are low temperature and low-cost synthesis. This fact motivates this work which discusses the structure

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and electrical property of the $\rm AZO/In_2S_3/textured$ p-Si heterojunction devices.

Methods

The In_2S_3 nanoflakes were prepared according to the CBD procedure reported by Bai et al. [15]. Typically, aqueous solutions of 0.025 M InCl₃, 0.048 M thioacetamide (CH₃CSNH₂) (TAA), and 0.04 M acetic acid were mixed in a glass beaker under magnetic stirring. The beaker was maintained at a reaction temperature of 80°C using water bath. In addition, the samples of silicon wafer were cleaned using a standard wet cleaning process. Subsequently, KOH was diluted to isotropically etch the silicon wafer to form a surface with a pyramid texture [16].

The preparation process of In₂S₃/p-Si heterojunction solar cell was separated into three parts: First, the samples with 1.5×1.5 -cm² square were cut from a (100)-oriented p-type silicon wafer with $\rho = 10 \ \Omega$ cm and 200-µm thickness. For ohmic contact electrodes, we used the DC sputtering technique to deposit 2-µm-thick Al onto the back of the Si substrates, followed by furnace annealing at 450°C for 1 h in Ar ambient conditions to serve Al as the p-ohmic contact electrodes. Second, $50 \sim 400$ -nm-thick n-type In₂S₃ thin films were deposited on the prepared p-type Si substrates by chemical bath deposition route in order to form an In2S3/p-Si heterojunction structure. Finally, an AZO film and Al metal grid with thicknesses of 0.4 and 2 µm, respectively, were deposited by sputtering. The purpose of AZO deposition is to produce a transparent conductive film by RF magnetron sputtering using ZnO:Al (2 wt.% Al₂O₃) target with a purity of 99.99% with 300-W power. All devices with the same AZO thickness (approximately 400 nm) were deposited at the same conditions. The single-cell size of photovoltaic device is about 0.4 cm^2 .

The phase identification of materials was performed by X-ray powder diffraction (Rigaku Dmax-33, Tokyo, Japan). The morphology and microstructure were examined by high-resolution transmission electron microscopy (HR-TEM; Hitachi HF-2000, Tokyo, Japan). The absorption and reflectance spectra were measured at room temperature using a Hitachi U-4100 UV–Vis-NIR spectrophotometer. The current density-voltage measurements (Keithley 2410 SourceMeter, Cleveland, OH, USA) were obtained by using a solar simulator (Teltec, Mainhardt, Germany) with an AM 1.5 filter under an irradiation intensity of 100 mW cm⁻².

Results and discussion

XRD patterns of various In_2S_3 films with thicknesses of 50 to 300 nm are shown in Figure 1. The In_2S_3 films were formed directly from the amorphous precursors by using chemical bath deposition method. All of the peaks



for various thicknesses were identified to be the tetragonal β -In₂S₃ phase (JCPDS card no. 25-0390) [17]. It can be seen that the crystallinity of In₂S₃ increases as the thickness of In₂S₃ film increases. The peaks of (206), (0012), and (2212) was observably seen while the thickness of In₂S₃ film was increased up to 300 nm. In this experiment, In³⁺ ions could form a variety of complexes in a solution. As InCl₃ is dissolved in water, it is hydrolyzed and finally form In(OH)₃. The possible chemical reactions for the synthesis of In₂S₃ nanocrystals can be expressed as following [18]:

$$\begin{bmatrix} \text{In}(\text{H}_2\text{O})_6 \end{bmatrix}^{3+} \leftrightarrow \begin{bmatrix} \text{In}(\text{H}_2\text{O})_5(\text{OH}) \end{bmatrix}^{2+} \\ + \text{H}^+ \leftrightarrow \text{In}(\text{OH})_3 + 3\text{H}^+$$
(1)

$$\begin{array}{l} CH_{3}CSNH_{2}+H^{+}+2H_{2}O{\leftrightarrow}H_{2}S+CH_{3}COOH\\ +NH_{4}^{+} \end{array} \tag{2}$$

$$H_2S \leftrightarrow HS^- + H^+, HS^- \leftrightarrow S^{2-} + H^+$$
(3)

$$2\mathrm{In}^{3^+} + 3\mathrm{S}^{2^-} \rightarrow \mathrm{In}_2\mathrm{S}_3.$$
 (4)

During the reaction processes, sulfide ions were slowly released from CH_3CSNH_2 and reacted with indium ions. Consequently, the In_2S_3 nanoflakes were formed via an *in situ* chemical reaction manner. Equation (4) indicates that In_2S_3 is produced by the reaction of S^{2-} and In^{3+} .

TEM analysis provides further insight into the structural properties of as-synthesized nanoflakes In_2S_3 . Figure 2a shows the low-magnification TEM image, and the nanoflakes can be clearly observed. The crystalline In_2S_3 nanoflakes are identified by electron diffraction (ED) pattern in the inset of Figure 2a, which exhibits diffusing rings, indicating that the In_2S_3 hollow spheres are constructed of polycrystalline In_2S_3 nanoflakes. The





concentric rings can be assigned to diffractions from (101), (103), and (116) planes of tetragonal In_2S_3 , which coincides with the XRD pattern. It is possible that the assembled effect arising from the nanocrystals results in the decrease of surface energy. A representative HRTEM image for such a tetragonal In_2S_3 nanostructure is shown in Figure 2b. It was found the interplanar distance of the crystal fringe is 3.3 Å, corresponding to the spacing of the (109) plane of tetragonal In_2S_3 [19].

Figure 3a,b shows the side-view and top-view SEM images of the textured p-Si substrate by using wet etching process. The uniform pyramids had been made on the surface of the p-Si, which was defined as the anti-reflective structures for incident sunlight. The various thicknesses of In_2S_3 films were grown on the surface of the textured p-Si substrate; the thicknesses of the In_2S_3 films were about 50, 100, and 300 nm, respectively, as



shown in Figure 3c,d,e. The images of the In_2S_3 /textured p-Si substrate exhibit a rough surface. The EDS line profiles indicate that the film consists of indium and sulfur. The atomic concentrations of In = 56.6% and S = 43.4% are calculated from the EDS spectrum, as shown in Figure 3f. The In_2S_3 films were grown not only in the lateral direction, but also randomly in the vertical direction. In the inset of Figure 3f, we can see that the surface of the In_2S_3 film is with a cross-linked network structure.

We have measured the UV–Vis absorption spectra of the various thicknesses of the In_2S_3 film and estimated the bandgap energy from the absorption onset of data curves in Figure 4a. For a direct bandgap semiconductor, the absorbance in the vicinity of the onset due to the electron transition is given by

$$\alpha = \frac{C(h\nu - E_g)^{1/2}}{h\nu},\tag{5}$$

where α is the absorption coefficient, *C* is the constant, $h\nu$ is the photon energy, and $E_{\rm g}$ is the bandgap energy. The inset of Figure 4a reveals the relationship of $(\alpha h\nu)^2$ and $h\nu$ gives a bandgap energy of 2.5 eV by the extrapolation of the linear region. The result was similar to previous report that 120- and 68-nm thicknesses of thermal-evaporated tetragonal In₂S₃ are with the bandgap of 2.54 and 2.52 eV, respectively [20].

Figure 4b shows the transmittance spectra of the 400-nm-thick AZO films on In_2S_3 films with various thicknesses. While the pure 400-nm AZO film on the glass showed 90.2% of transmittance, the transmittance values of 400-nm-thick AZO on In_2S_3 with 50-, 100-, and 300-nm thickness were about 86.2%, 75.5%, and





68.6%, respectively. It can be seen that the transmittance is decreased as the thickness of In_2S_3 film increases.

Figure 5 shows the reflectance spectra of the planar p-Si, textured p-Si, and the In_2S_3 film with various thicknesses on textured p-Si substrate in the range of 200 ~ 1,100 nm. The average reflectance was about 11.3%, 10.9%, and 8.7% for the In_2S_3 film on the textured p-Si substrate with 50-, 100-, and 300-nm thicknesses, respectively. These values are lower than the average reflectance of planar p-Si and textured p-Si (32.0% and 16.2%, respectively). Therefore, the reflectance is obviously reduced by the nanoflake In_2S_3 and decreased as the thickness of In_2S_3 film increases. It could be attributed to the decreasing reflectance for In_2S_3 film at short wavelengths because the nanotexturization was on the surface [21].

Figure 6a displays the schematic structure of the heterojunction solar cell in which the nanotextured In_2S_3/p -Si was the photoactive layer of such a device. Photovoltaic performance of the AZO/In₂S₃/p-Si heterojunction solar cell with various In_2S_3 thicknesses is given in Table 1. All samples for the electrical measurement were performed with AZO film of about 400 nm. Characterization of the AZO/In₂S₃ film deposited on the textured p-Si substrate was studied for the first time. Figure 6b shows a SEM image of an inclined angle of the AZO/In₂S₃/p-Si heterojunction structure. The AZO deposited on the In_2S_3 (100 nm)/p-Si substrate exhibits

a well coverage and turns into a cylinder-like structure with a hemispherical top as shown in the inset of Figure 6b. The deposition thickness of the AZO was estimated to be 400 nm. Jiang et al. [22] revealed that they had fabricated the SnS/ α -Si heterojunction photovoltaic devices, which the junction exhibited a typical rectified diode behavior, and the short-circuit current density was 1.55 mA/cm². Hence, the AZO/In₂S₃/p-Si structure in the study was suitable for solar cell application.

The current–voltage (*J*-*V*) characteristics of the fabricated photovoltaic devices were measured under an illumination intensity of 100 mW/cm², as shown in Figure 6c. Such result shows that the short-circuit currents (*J*sc) were increased while the In_2S_3 films were deposited onto the p-Si. The power conversion efficiency (PCE) of the devices can be obviously improved from 0.47% to 2.39% by employing a 100-nm-thick In_2S_3 film.

Table 1 Photovoltaic performance of the AZO/ ln_2S_3 /p-Si heterojunction solar cell with various thicknesses of ln_2S_3

Device	$V_{\rm oc}$	J _{sc} (mA/cm ²)	F.F. (%)	Efficiency (%)
Non-In ₂ S ₃	0.20	10.68	21.95	0.47
ln ₂ S ₃ (50 nm)	0.28	21.18	30.55	1.81
ln ₂ S ₃ (100 nm)	0.32	23.43	31.82	2.39
ln ₂ S ₃ (200 nm)	0.24	16.37	32.14	1.26
ln ₂ S ₃ (300 nm)	0.24	16.08	28.10	1.08

The photovoltaic condition is AM 1.5 G at 100-mW/cm² illumination.

It was also found that the highest open-circuit voltage (Voc) and short-circuit current density are 0.32 V and 23.4 mA/cm², respectively. Therefore, the optimum thickness of the In₂S₃ film is 100 nm, with PCE of 2.39%. When the thickness of the In_2S_3 film increases, the efficiency decreased because of the decrease in Jsc and FF, as shown in Figure 6d. A similar phenomenon was also observed in the In₂S₃/CIGS heterojunction thin film solar cell [23]. It is possible that some defects on the interface of the AZO/In₂S₃/p-Si heterojunction with thicker In₂S₃ films will decrease the PCE. The cell performance improved markedly as the thickness of the In₂S₃ layer was increased to 100 nm. This improved cell performance is attributed to the reduction of possible shunt paths by the inclusion of a high-resistivity In_2S_3 buffer layer between the transparent conducting ZnO:Al and the p-Si layers. The cell performance, however, deteriorated in devices with 200- and 300-nm-thick In₂S₃ layers since the series resistance of the solar cell increased due to the high resistance of the In₂S₃ layer. Therefore, the 100-nm In₂S₃ sample shows the best performance.

Conclusions

In summary, we have successfully synthesized the nanoflake In_2S_3 by a chemical bath deposition route in the study. The well-crystallized single phase of tetragonal In_2S_3 that can be obtained at 80°C and deposited on *p*-Si substrate was investigated for the first time. The visible light absorption edge of the as-grown In_2S_3 film corresponded to the bandgap energy of 2.5 eV by UV–Vis absorption spectra. It can be seen that the lower reflectance spectra occurred while the thickness of In_2S_3 film on the textured p-Si was increased. The photovoltaic characteristics of the AZO/ In_2S_3 /textured p-Si heterojunction solar cells with various In_2S_3 thicknesses were also given in the investigation, and the PCE of such device with 100-nm-thick In_2S_3 film is 2.39% under 100-mW/cm² illumination.

Abbreviations

FE-SEM: Field emission scanning electron microscopy; HRTEM: High-resolution transmission electron microscopy; UV–Vis: Ultraviolet–visible absorption spectra; PCE: Power conversion efficiency.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

YJH and LWJ carried out the design of the study and drafted this manuscript. CHL and THM conceived of the study and participated in its design and coordination. YLC and HPC carried out the preparation of the samples and characteristic measurements. All authors read and approved the final manuscript.

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