

*Original Article*

## Preparation of aluminum doped zinc oxide targets and RF magnetron sputter thin films with various aluminum doping concentrations

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Received: 9 December 2016; Revised: 15 March 2017; Accepted: 29 April 2017

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**Abstract**

In this work, aluminum doped zinc oxide (AZO) ceramic targets were prepared from ZnO powder and Al<sub>2</sub>O<sub>3</sub> powder with varying amounts of Al<sub>2</sub>O<sub>3</sub> doped in a range of 1-5 wt%. The mixed ZnO and Al<sub>2</sub>O<sub>3</sub> powders were pressed at a pressure of 80 MPa into disks and sintered at 1,300 °C for 5 h in air. The crystal structures of the sintered targets were characterized by X-ray diffraction (XRD) technique. It was found that the XRD spectra showed a hexagonal (wurtzite) structure of ZnO for all Al<sub>2</sub>O<sub>3</sub> doped films. However, as the amount of Al<sub>2</sub>O<sub>3</sub> increased over 2 wt%, the gahnite (ZnAl<sub>2</sub>O<sub>4</sub>) phase could be observed in the XRD spectra. The AZO films were deposited on glass slides at room temperature and post-annealed at 500 °C in a vacuum for 1 h. The film structures, Al/Zn ratio between the Al and Zn atoms, and the electrical properties were characterized. It was found that an increase of Al<sub>2</sub>O<sub>3</sub> content in the target gave a higher Al doping concentration in the film which resulted in more Al substitutions in the Zn sites which in turn resulted in an increase of carrier concentration. The crystal structure of the AZO films deposited from undoped and 1 wt% Al<sub>2</sub>O<sub>3</sub>-doped targets showed the (002) preferred orientation and drastically decreased at higher Al doping concentrations. The mobility of the charge carrier was affected by lower crystallinity due to grain boundary scattering. In addition, the excess Al in the film may play a role as impurity scattering centers. A decrease of Hall mobility resulted in increased resistivity. The minimum resistivity of 2.01x10<sup>-3</sup> Ω.cm could be achieved for the AZO films deposited from 1 wt% Al<sub>2</sub>O<sub>3</sub>-doped ZnO targets.

**Keywords:** AZO films, Al-doping concentration, RF magnetron sputtering

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## 1. Introduction

Transparent conducting oxide (TCO) films, such as tin-doped indium oxide (ITO), fluorine-doped tin oxide (FTO) and aluminum-doped zinc oxide (AZO), have been widely used in many applications such as solar cells (Zhao *et al.*, 2016), gas sensors (Hosseinnejad *et al.*, 2016), and thin film transistors (Navamathavan *et al.*, 2008). Among these TCOs, the ITO is the mostly used because of high visible transmission and low electrical resistivity. However, indium is an expensive material, hence there are efforts to prepare other low-cost, good quality, and indium-free TCO films (Zhang *et al.*, 2016). The AZO film is especially attractive due to its good electrical properties with high transmission in the visible light region comparable to the ITO. Unique combinations of electrical and optical properties of the TCO films result from good crystalline structure and high carrier concentration with high mobility in the film. The carrier concentration in the AZO film originates from oxygen vacancy and the substitution of  $\text{Al}^{3+}$  into  $\text{Zn}^{2+}$  sites of the zinc oxide structure.

There are various deposition techniques used to prepare the AZO films such as spray pyrolysis (Muiva *et al.*, 2011), pulse laser deposition (Chen *et al.*, 2005), sol-gel (Mamat *et al.*, 2010), and radio frequency (RF) magnetron sputtering. In the magnetron sputtering technique, the oxygen vacancy and  $\text{Al}^{3+}$  substitution are dependent on deposition parameters such as working pressure (Houng *et al.*, 2008), sputtering power (Besleaga *et al.*, 2014), substrate temperature (Li *et al.*, 2009), distance from target to substrate (Jeong *et al.*, 2004), and oxygen partial pressure (Yim *et al.*, 2006) which have been widely investigated.

In addition, Al doping concentration is an important parameter that can affect the properties of the AZO films prepared by the sputtering technique (Besleaga *et al.*, 2014). In this work, the effects of Al doping concentrations were studied by varying the percentage of Al in AZO sputtering targets. Low-cost zinc oxide (ZnO, 99.7%) and aluminum oxide ( $\text{Al}_2\text{O}_3$ , 99.99%) powders were used as raw materials to prepare the AZO sputtering targets. The properties of the prepared AZO films were then investigated. The AZO films were prepared by RF magnetron sputtering at room temperature and post-annealed at 500 °C in a vacuum.

## 2. Experimental Details

### 2.1 Sputtering target preparations and characterizations

The experimental procedure began with preparations of the  $\text{Al}_2\text{O}_3$ -doped ZnO ceramic targets by mixing ZnO powder (99.7%) and  $\text{Al}_2\text{O}_3$  powder (99.99%) in the weight percent (wt%) ratios of ( $\text{Al}_2\text{O}_3/\text{ZnO}$ ) 0, 1, 2, 3, 4, and 5 wt%. The mixed powders were milled by a ring-mill machine and pressed into disks with a diameter of 82.2 mm and then sintered at 1,300 °C in air for 3 h. The diameters of the ceramic targets were reduced to about 74.0 mm after sintering. The crystal structures of the AZO targets were characterized using the X-ray diffraction technique (D8 diffractometer, Bruker). Surface morphologies and composition were analyzed using a scanning electron microscope (SEM) and energy dispersive X-ray (EDX) spectroscopy (SU3500, Hitachi) respectively. Peak intensity in the EDX measurement was

determined with ZAF correction using an energy dispersive X-ray microanalyzer (EMAX Energy, Horiba) attached to the SEM (SU3500, Hitachi). In addition, the accuracy of elemental assignment and peak position was confirmed using a commercial 2 wt%  $\text{Al}_2\text{O}_3$ -doped ZnO ceramic target (Kurt J. Lesker) as an internal reference to confirm peak position with which the atomic percentage (at.%) of the target compositions were investigated.

### 2.2 AZO film preparations and characterizations

For preparations of AZO films, glass slides (size 2.5x2.5 cm) were used as the substrates. They were ultrasonically cleaned by steeping the substrates in acetone and ethyl-alcohol respectively for 10 min each and blowing dry with nitrogen gas before placing them into a vacuum chamber. The AZO films were prepared in a home-built sputtering system at room temperature. Before the sputtering process, the vacuum chamber was evacuated to a background pressure of about  $1.0 \times 10^{-5}$  mbar by rotary and diffusion pumps. High-purity argon gas (99.995%) was then filled into the chamber and the working pressure was maintained at  $5.0 \times 10^{-3}$  mbar. RF power was maintained at 100 W for all depositions at a frequency of 13.56 MHz (Cito 1310, Comet).

The target was pre-sputtered for 5 min before film deposition. The deposition times were calibrated for a film thickness of 300 nm for 30-45 min. The film thicknesses were also determined by cross-section SEM images of the deposited film/glass substrate. The deposited films were then annealed in vacuum at 500 °C for 1 h. The crystal structure and the electrical properties of the annealed films were characterized using the X-ray diffraction technique (D8 diffractometer, Bruker) and Hall measurement (HMS-3000, Ecopia), respectively. In addition, atomic percentage (at.%) of Al composition in the film structure was investigated by EDX spectroscopy (SU3500, Hitachi).

## 3. Results and Discussion

### 3.1 Crystal structure of prepared AZO targets

The XRD patterns of the ZnO and 1-5 wt% of  $\text{Al}_2\text{O}_3$ -doped ZnO ceramic targets sintered at 1,300 °C for 3 h were collected (Figure 1). The XRD spectra of ZnO and 1-2 wt%  $\text{Al}_2\text{O}_3$ -doped ZnO ceramic targets show the hexagonal (wurtzite) structure of the ZnO (ICDD: 01-089-0510). The peak positions remained unchanged with the increase in the wt% of  $\text{Al}_2\text{O}_3$  doped in the ZnO structure. However, it can be noticed that peaks of gahnite ( $\text{ZnAl}_2\text{O}_4$ ) phase (ICDD: 01-070-8181) could be observed with the increase in the wt% of  $\text{Al}_2\text{O}_3$  doped in the target higher than 2 wt%. This implied residual  $\text{Al}_2\text{O}_3$  which could form the gahnite phase. Therefore, in high wt%  $\text{Al}_2\text{O}_3$  doped ZnO, not all Al atoms substituted into the Zn sites of the AZO crystal structure. This result was consistent with that reported by Besleaga *et al.* (2014) in which the gahnite phase could be detected for the AZO sputtering targets doped with 2.2 wt%  $\text{Al}_2\text{O}_3$ .

In order to analyze the gahnite phase that appeared in the XRD spectra, backscattered electron (BSE) images and the characteristic X-ray mapping of the elements were performed. Figure 2(a) shows the BSE image of the surface morphology for the 5 wt%  $\text{Al}_2\text{O}_3$ -doped ZnO ceramic target. It can be

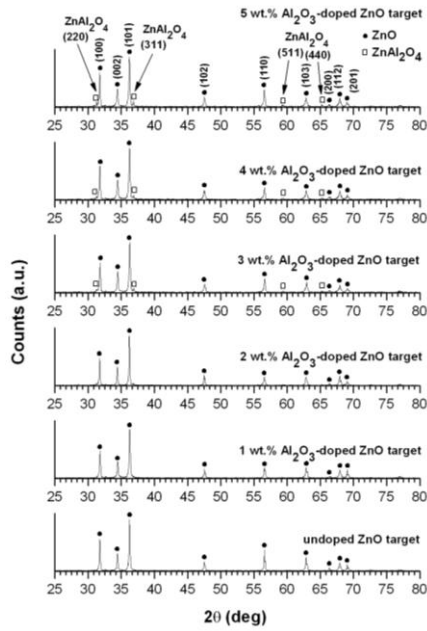


Figure 1. XRD pattern of AZO ceramic target doped with Al<sub>2</sub>O<sub>3</sub> from 0-5 wt%.

obviously seen that there were two distinguishable areas such as white and black areas. The BSE image was performed by secondary electrons scattered from the incident object. If the incident object is composed of various elements or morphology, then the scattered electrons will be different. Consequently, the image will show different colors corresponding to the scattered electrons as gray to white for high scattering and dark-gray to black for the scattering of lighter electrons. In order to specify the elements in these two areas, characteristic X-ray mapping was performed (Figure 2(b)-(e)). The images of element distribution were performed by the characteristic X-ray emitted from each element from the primary incident electrons. Therefore, from Figure 2, the amounts of Al and O in the black area can be seen. In Figures 2(b) and (d), is higher than that in the white area. On the other hand, the amount of Zn is higher in the white area than that in the black area as shown in the Figure 2(c). The atomic ratio of Zn:Al:O for ZnAl<sub>2</sub>O<sub>4</sub> is 1:2:4 while the Al substitution in the Zn position in the film structure was one by one. Thus, the ZnAl<sub>2</sub>O<sub>4</sub> phase needs more Al and O atoms than in the case of Al substitutions. Therefore, the black area represented the gahnite (ZnAl<sub>2</sub>O<sub>4</sub>) phase in the target which results in the peaks of gahnite phase in the XRD spectra (Figure 1). The compositions in the black and white areas were confirmed by the EDX

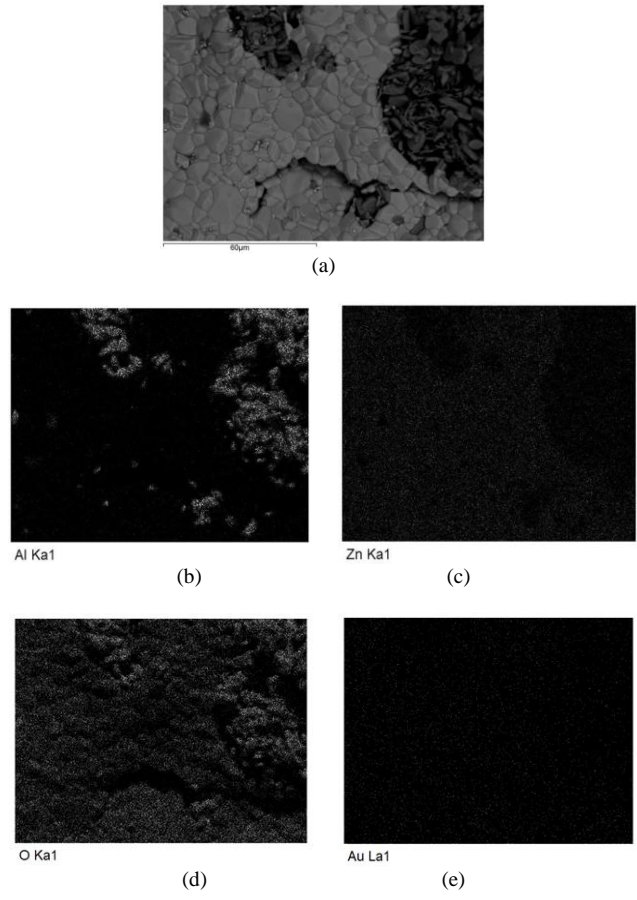


Figure 2. Surface morphology of 5 wt% Al<sub>2</sub>O<sub>3</sub>-doped ZnO ceramic target (a) backscattered electron (BSE) image and characteristic X-ray mappings for element distributions of (b) Al, (c) Zn, (d) O and (e) Au.

spectroscopic images which represents the wt% of each composition in both areas (Figure 3). In addition, the wt% and at.% of Zn, Al, and O are shown in the inset tables. It can be seen that the atomic ratio of Zn:Al:O is 1:1.67:4.37 which is very close to that of the atomic ratio of the  $ZnAl_2O_4$  molecule as discussed above. Moreover, Au film was coated on the specimen surface before the SEM. Therefore, the X-ray mapping of the Au element, as shown in Figure 2(e), showed a uniform distribution on the specimen surface which implied uniform coating of the Au film.

**3.2 Crystal structure of deposited AZO films**

The XRD spectra of the AZO films deposited from the undoped and 1-5 wt%  $Al_2O_3$ -doped ZnO targets were studied (Figure 4). The films deposited from undoped and 1 wt% of  $Al_2O_3$ -doped ZnO targets indicated the preferred orientation in (002) plane at  $2\theta \approx 34.54^\circ$  and  $34.74^\circ$ , respectively. The  $2\theta$  value of the film deposited from the undoped ZnO target was very close to the standard value of  $34.45^\circ$  of the ZnO crystal structure. For the AZO films deposited from 1 wt%  $Al_2O_3$ -doped ZnO target, the  $2\theta$  value shifted to a higher angle which may be attributed to different ionic radii of  $Zn^{2+}$  and  $Al^{3+}$  being 72 and 53 pm, respectively. Therefore, the Al substitution in the Zn sites of the film structure can bring about a shorter length of the *c*-axis. However, for film with a higher wt% of  $Al_2O_3$  doping concentration, the (002) peak drastically decreased and disappeared for the AZO film prepared from 5 %wt. For film with  $Al_2O_3$  doped, the low intensities (100) and (101) peaks were observed. However, there are no  $ZnAl_2O_4$  peaks detected in any of deposited films which may imply that the Al atoms were substituted in the ZnO sites or segregated to the grain boundary in the film structure.

The decrease in the (002) peak as the wt% of Al-doped in ZnO structure increased may be due to the difference in ion size between  $Zn^{2+}$  and  $Al^{3+}$  which created stress in the film structures. Appearance of the stress resulted in the loss of periodic arrangement, random orientations and, consequently, poor crystallinity in the film structures as discussed by Muiva *et al.* (2011), which reported AZO films prepared by spray pyrolysis technique at substrate temperature of 420 °C with different atomic percent (at.%) from 0-10 at.%.

**3.3 Al composition and electrical properties of deposited AZO films**

The percentage amounts of Al doping concentration in the AZO films structure were confirmed by EDX results for the films deposited from 0 to 5 wt% of  $Al_2O_3$  doped in the AZO targets (Figure 5). From linear regression, the atomic percentage of Al increased linearly with weight percentage of  $Al_2O_3$ -doped in the target. The atomic percentage of Al doping concentration increased from 3.45 to 12.77 at.% in the AZO film deposited by the targets as the wt% of  $Al_2O_3$  doping increased from 1 to 5 wt%. The higher at.% of Al doping concentrations was consistent with an increase of the carrier concentration in the AZO films which increased from  $3.40 \times 10^{19} \text{ cm}^{-3}$  for undoped ZnO film to  $9.67 \times 10^{20} \text{ cm}^{-3}$  for 4 wt%  $Al_2O_3$  doped film (Figure 6). The higher carrier concentration in Al-doped ZnO film resulted from Al substitutions in Zn sites in the film structure. Therefore, as the Al doping amount increased, there was an increase in Al substitution in the film structure and higher carrier concentration. However, as the  $Al_2O_3$  doping concentration increased to 5 wt%, the carrier concentration reduced to  $6.86 \times 10^{20} \text{ cm}^{-3}$ . The excess Al atoms in the film structure had no longer contributed to the Al substitutions but played a role as electron traps resulting in reducing the carrier concentration. This result was consistent with that reported by El Manounia *et al.* (2006).

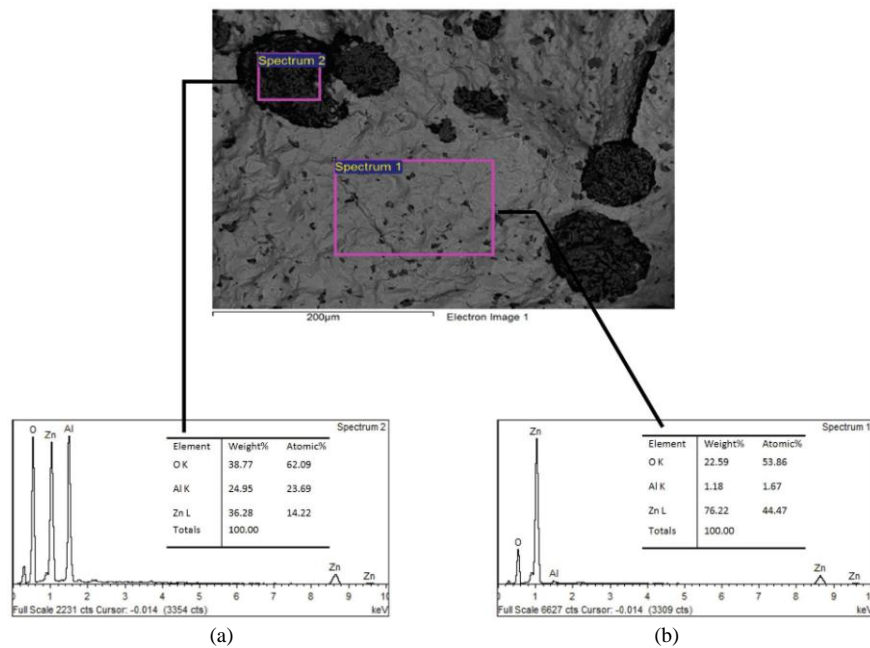


Figure 3. EDX measurements of sanded 5 wt%  $Al_2O_3$ -doped ZnO target in the white and black areas.

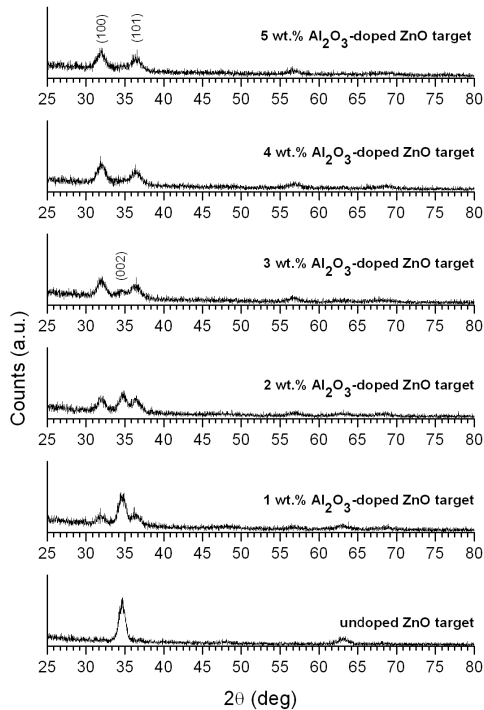


Figure 4. XRD spectra of AZO films deposited from prepared ceramic target doped with  $\text{Al}_2\text{O}_3$  from 0-5 wt%.

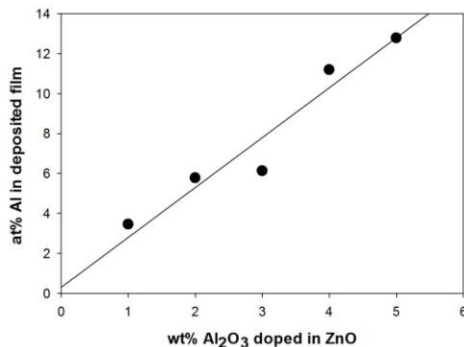


Figure 5. Al atomic percentage of AZO films deposited from prepared ceramic target doped with  $\text{Al}_2\text{O}_3$  from 0-5 wt% extracted from EDX results.

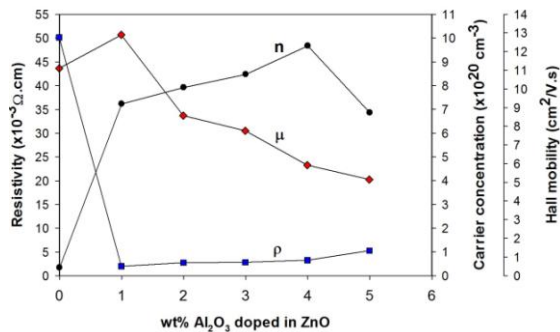


Figure 6. Electrical properties of AZO films deposited from prepared ceramic target doped with  $\text{Al}_2\text{O}_3$  from 0-5 wt%.

In addition, the excess Al atoms doped in the crystalline structure may segregate in the grain boundary or interstitial sites and act as impurity scattering centers which can reduce mobility of the carriers. Moreover, according to the XRD spectra (Figure 2), the decrease in the Hall mobility may be due to grain boundary scattering because of lattice distortion in the film structure with high  $\text{Al}_2\text{O}_3$ -doping concentration as reported by Zhao *et al.* (2016).

The electrical resistivity is inversely proportional to the carrier concentration and the mobility according to  $1/\rho = ne\mu$ , where  $e$  is the electron charge. For undoped ZnO film, the resistivity was as high as  $5.02 \times 10^{-2} \Omega \cdot \text{cm}$  due to the very low carrier concentration of  $3.40 \times 10^{19} \text{ cm}^{-3}$ . Then the resistivity drastically dropped to  $2.01 \times 10^{-3} \Omega \cdot \text{cm}$  due to the increased carrier concentration. For the AZO films deposited from 1-4 wt%  $\text{Al}_2\text{O}_3$  doped ZnO targets, the carrier concentration increased with higher Al doping concentration while the mobility became smaller. This resulted in increased resistivity from  $2.01 \times 10^{-3} \Omega \cdot \text{cm}$  to  $9.67 \times 10^{-3} \Omega \cdot \text{cm}$ . It can be noticed that the mobility had a much greater effect than the carrier concentration. Therefore, in order to achieve lower electrical resistivity of the AZO film, it is necessary to improve the mobility by increasing the crystallinity of the film structure.

#### 4. Conclusions

In this research, AZO targets were prepared from ZnO powder and  $\text{Al}_2\text{O}_3$  powder as 0-5 wt%  $\text{Al}_2\text{O}_3$ -doped. The sintering temperature was set at  $1300^\circ\text{C}$  for 3 h in air. It was found that the gahnite ( $\text{Zn Al}_2\text{O}_4$ ) phase appeared in XRD spectra of the sintered targets. After that AZO films were deposited from the sintered AZO targets by the RF magnetron sputtering technique. The deposited films were then annealed at  $500^\circ\text{C}$  in vacuum for 1 h. It was found that the amount of Al in the film increased as the wt% of  $\text{Al}_2\text{O}_3$  doping increased in the target.

The crystal structure and electrical properties of the films were affected by the concentration of Al doping in the film structures. Higher Al doping concentrations resulted in poor crystallinities and an increase in the carrier concentration. However, poor crystallinities can cause grain boundary scattering to be dominant which then contributed to low Hall mobility. In addition, for films from the 5 wt%  $\text{Al}_2\text{O}_3$  doped target, the excess Al atoms in the film played the role as electron traps leading to a decrease in the carrier concentration.

It can be concluded that high quality AZO films deposited by the RF magnetron sputtering technique can be achieved from prepared AZO targets with the appropriate wt% of  $\text{Al}_2\text{O}_3$ -doped target.

#### Acknowledgements

The authors would like to thank Asst. Prof. Rattapol Rangkupan, Metallurgy and Materials Research Institute Chulalongkorn University, for his kindness for the SEM and EDX measurements. The authors also would like to thank the Research and Development Institute, Nakhonpathom Rajabhat University for the research funding.

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