



# Effects of Copper Availability on the Physiology of Marine Heterotrophic Bacteria

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Marine heterotrophic bacteria play a crucial role in the cycling of energy and nutrients in the ocean. Copper (Cu) belongs to the repertoire of essential trace nutrients for bacterial growth, yet physiological responses of marine heterotrophic bacteria to Cu deficiency remain unexplored. Here, we examined these responses in oceanic and coastal isolates of heterotrophic bacteria from ecologically significant microbial clades (Flavobacterija class from Bacteroidetes phylum, and marine Roseobacter clade within Alphaproteobacteria class and Alteromonadales within Gammaproteobacteria class, both from Proteobacteria phylum). Bacterial growth, Cu quotas (Cu:P), macronutrient content and stoichiometry (cellular C, N, P, S, and C:N, S:P), as well as carbon metabolism (respiration, productivity, carbon demand, growth efficiency) were monitored across a gradient of Cu conditions, characteristic of coastal and open-ocean surface waters. Cu deficiency had most severe effects on a Flavobacteriia member Dokdonia sp. strain Dokd-P16 for which we observed significant reductions in growth, C metabolism and Cu guotas. Other strains did not significantly reduce their growth rate, but adjusted their Cu content and some C metabolic rates (Ruegeria pomeroyi DSS-3, Roseobacter clade) or were unaffected (Pseudoalteromonas sp. strains PAlt-P26 and PAlt-P2, Alteromonadales clade). These diverse bacterial responses were accompanied by constant cellular composition of major elements and stoichiometric ratios. Changes in bacterial Cu guotas occurred within a modest range (~5-fold range) relative to the 50-fold variation in total Cu in the media. We hypothesize that this may reflect a well-controlled Cu homeostasis in marine heterotrophic bacteria. In a preliminary assessment, we found that Cu quotas of bacteria and those of eukaryotic phytoplankton are not statistically different. However, compared to eukaryotic phytoplankton, the variability of Cu quotas in marine heterotrophic bacteria is smaller, which could reflect differences in their Cu homeostasis. Using Cu quotas obtained in our study, we assessed the contribution of bacterial Cu to the biogenic Cu pool in an oceanic euphotic zone in the NE Pacific. These preliminary estimates suggest that up to 50% of the biogenic Cu could be contained in the biomass of marine heterotrophic bacteria. Our study sheds light on the interactions between Cu and marine heterotrophic bacteria, demonstrating the potential for Cu to influence microbial ecology and for microbes to play role in Cu biolgeochemical cycle.

Keywords: copper, Cu quota, limitation, marine heterotrophic bacteria, *Flavobacteriia*, Roseobacteria, *Alteromonadales*, *Dokdonia* sp.

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# INTRODUCTION

Planktonic bacteria represent most of the living biomass in the world's oceans (Whitman et al., 1998). In surface waters, prokaryotic assemblages are dominated by autotrophic and heterotrophic bacteria, while archaea are more abundant in the ocean's interior (Karner et al., 2001). Heterotrophic members of the bacterioplankton community play a critical role as the recyclers of organic material, thus influence fluxes of energy and essential elements within oceanic ecosystems. Their activity is governed by the availability of organic substrates as well as inorganic nutrients. The latter group includes bioactive metals, such as iron (Fe), copper (Cu), manganese (Mn), zinc (Zn), and cobalt (Co), that facilitate a variety of biological pathways in a prokaryotic cell. In this group, iron has been studied most intensively and has a well-established role as a factor limiting growth and metabolism of marine heterotrophic bacteria (Pakulski et al., 1996; Tortell et al., 1996; Kirchman et al., 2003a; Bertrand et al., 2011; Fourquez et al., 2014). Furthermore, previous studies have linked heterotrophic bacterioplankton to the marine biogeochemical cycle of Fe (Tortell et al., 1996; Maldonado and Price, 1999; Maldonado et al., 2016). By contrast, there is a paucity of information on how the availability of other essential metals regulate fundamental processes of heterotrophic bacteria, nor is it clear how these metals shape bacterial ecophysiology in the sea.

Copper plays a unique biological role because even though it is required for growth, it can be extremely toxic at very low concentrations. Its toxicity in marine bacteria and phytoplankton was found to be dependent on the abundance of the free Cu (II) ion (Sunda and Lewis, 1978; Schreiber et al., 1985), which is present in pM levels in seawater due to complexation with natural organic chelators (Santos-Echeandía et al., 2008; Heller and Croot, 2015). Copper has the potential to induce oxidative stress as it easily enters Fenton-like reactions that generate harmful hydroxyl radicals in the cell (Rowley and Halliwell, 1983). Another mode of Cu-induced toxicity is the destruction of Fe-S clusters of enzymes by direct complexation of Cu(I) ion to the coordinating sulfur atom (Macomber and Imlay, 2009). It is therefore essential to strictly control cellular Cu levels so that bacterial nutritional requirements are met while Cu toxicity is minimized. Prokaryotic Cu homeostasis has been the subject of extensive research in model bacteria (see reviews of: Rensing and Grass, 2003; Solioz and Stoyanov, 2003; Solioz et al., 2010; Argüello et al., 2013; Bondarczuk and Piotrowska-Seget, 2013), with a focus on elucidating mechanisms for Cu detoxification. In this context, the interactions between Cu and marine bacteria have also been explored (Gordon et al., 1993, 1994; Moffett and Brand, 1996; Mann et al., 2002). However, far less is known of the impacts of low Cu availability on prokaryotic cells (bacteria and archaea) and what adaptations these microorganisms have for dealing with Cu shortage in their environment.

To our best knowledge, the effects of Cu deficiency have only been investigated in model strains from two groups of marine microbes: heterotrophic denitrifying bacteria and autotrophic ammonia-oxidizing archaea (Granger and Ward, 2003; Moffett et al., 2012; Amin et al., 2013); whereas aerobic heterotrophic bacteria are yet to be explored. Copper serves as a catalyst in thirteen known prokaryotic cuproenzymes (as reviewed by Argüello et al., 2013), with the respiratory enzyme cytochrome c oxidase (COX) having the most prominent use in aerobic bacteria (Ridge et al., 2008). Hence, Cu starvation is likely to affect C metabolism in these microorganisms, impacting a cell's energetic status and growth.

The aim of our study was to gain insight into the role of Cu as a nutrient in marine heterotrophic bacteria. We focused on examining key aspects of bacterial physiology: growth, carbon metabolism, cellular Cu quotas, as well as macronutrient content and stoichiometry across a gradient of Cu availability typical of coastal and open-ocean surface waters. We investigated four marine heterotrophic bacteria, which included three isolates (Dokdonia sp. strain Dokd-P16; Pseudoalteromonas sp. strains PAlt-P2 and PAlt-P26) from surface waters of the Northeast Pacific Ocean (including Fe-limited waters), and the model bacterium Ruegeria pomeroyi DSS-3. These strains belong to ecologically important microbial groups within the world's ocean, namely Flavobacteriia class within the phylum of Bacteroidetes, Roseobacter clade within the Alphaproteobacteria class (Proteobacteria phylum), and the order Alteromonadales within the Gammaproteobacteria class (Proteobacteria phylum). These bacterial groups are known for their importance in the transformation of phytoplankton-derived organic matter during bloom events (as reviewed in Buchan et al., 2014).

#### MATERIALS AND METHODS

#### Study Organisms

The heterotrophic bacteria used in this study were isolated from different locations along Line P, a transect extending from the coast of British Columbia, Canada, into the open-ocean waters of the Gulf of Alaska, 1,500 km offshore. Near-surface seawater ( $\sim$ 25 m) was collected at four different stations along the transect (P2, P4, P16, and P26) during the June cruise in 2012 (Line P program, cruise 2012-12)<sup>1</sup>. The seawater was then plated onto marine agar supplemented with either  $0.5 \text{ g L}^{-1}$  (for seawater from the coastal stations P2 and P4) or 0.05 g  $L^{-1}$  (for seawater from oceanic stations P16 and P26) of organic substrate (bactopeptone and casein hydrolysate in a 1:1 ratio). The agar plates were prepared using filtered seawater (0.22 µm) collected from  $\sim$ 40 m at station P26. The colonies which grew on the plates were purified by sequential plating on marine agar. Pure cultures were preserved and stored in 15% glycerol stock at -80°C and plated as needed for the experiments.

Line P bacteria isolates were identified using 16S ribosomal DNA PCR amplification. PCR was performed using liquid cultures from colonies originally grown on agar plates. Liquid cultures were grown by adding a single colony to 2 mL of sterile liquid media made with station P26 seawater (from 10 m depth) supplemented with 0.5 g DOC (0.25 g L<sup>-1</sup> bactopeptone and 0.25 g L<sup>-1</sup> casein hydrolysate). Test tubes were placed in a shaker for 24 h at room temperature. Following culture

<sup>&</sup>lt;sup>1</sup>http://linep.waterproperties.ca/2012-12/index.php

growth, the full-length 16S rRNA gene was PCR amplified with primers 27F (5'-AGAGTTTGATCMTGGCTCAG-3') and 1492R (5'-GGTTACCTTGTTACGACTT-3'). PCR amplification was performed in a 25 µL reaction volume with 16.7 µL of nuclease free water, 2.5 µL of buffer, 1.5 µL MgCl<sub>2</sub> (Sigma Aldrich), 2 µL dNTPs, 0.5 µL forward and reverse primers,  $0.3 \,\mu\text{L}$  of Taq polymerase and  $1 \,\mu\text{L}$  of liquid culture. The reaction mixture was placed in the Thermocycler and run under the following conditions 95°C for 10 min, followed by 25 cycles of 95°C for 30 s, 55°C for 30 s, 72°C for 90 s, followed by a final extension for 10 min at 72°C. The amplicon size was verified with an agarose gel, and the amplicons were purified with the NucleoSpin Gel and PCR Clean-up kit (Macherey-Nagel). Samples were diluted to  $\sim 4 \text{ ng } \mu l^{-1}$  and sent to the GeneWiz sequencing center (GeneWiz, Inc.) for sequencing using three primers, 27F, 1492R, and a center primer, 519R (5'-GNTTTACCGCGGCKGCTG-3'). The three sequences for each clone were aligned using Sequencher 5.1 and a consensus sequence for each clone was obtained. The sequences were compared to the SILVA database version 115<sup>2</sup> and isolates were taxonomically identified to the genus level (Table 1). The sequence data can be found in publicly available repository on Github<sup>3</sup>.

With the exception of one strain, bacteria isolated from Line P surface waters belonged to the genus Pseudoalteromonas within the Gammaproteobacteria class (Table 1). Members of this genus may play a role in processing of organic carbon during phytoplankton blooms (e.g., transient peaks in Pseudolateromonas sp. in North Sea bloom reported by Lucas et al., 2015) and influencing Fe speciation and bioavailability in the ocean given their ability to produce Fe-binding siderophores (Granger and Price, 1999; Armstrong et al., 2004; Sijerčić and Price, 2015). One isolate from Line P belonged to genus Dokdonia sp. within the Flavobacteriia class. Many of Dokdonia sp. isolates from previous studies carry out rhodopsin-based bacterial phototrophy (Gómez-Consarnau et al., 2007; Kimura et al., 2011; Riedel et al., 2013; Bogachev et al., 2016), suggesting that this taxon may play a role in the cycling of carbon and energy in the ocean. In addition to the Line P isolates, we studied the model marine bacterium Ruegeria pomeroyi DSS-3, DMS 15171 (isolated from coastal Georgia), which was obtained from the American Type Culture Collection (ATCC, Manassas, VA, United States). The Roseobacter member Ruegeria pomeroyi DSS-3 has a range of adaptations that have been implicated in various biogeochemical cycles (e.g., Green et al., 2012; Cunliffe, 2013; Durham et al., 2015), such as an ability to degrade the algal osmolite dimethylsulphoniopropionate (DMSP, Reisch et al., 2011; Todd et al., 2012). The bacterial strains examined here are diverse in terms of their provenance and phylogeny (Table 1), and thus are likely to have distinct Cu requirements and adaptations to changing Cu availability. In our preliminary studies, we found a similar growth response to low and high Cu availability in different Pseudoalteromonas sp. isolates (Supplementary Figure 1). Ultimately, we selected

<sup>2</sup>http://www.arb-silva.de/

one coastal (station P2, strain PAlt-P2) and one oceanic isolate (station P26, strain PAlt-P26) of *Pseudoalteromonas* sp. for more detailed investigations.

# Limitations of Using Isolates as Models for Bacteria in the Sea

Using readily cultivable bacteria (as those in our study) to describe the ecophysiology of marine microbes is challenging as it remains unclear if such bacteria are good models for natural microbial populations. This can be inferred from low abundance of the types of bacteria commonly isolated using traditional plate methods in natural communities (e.g., 16S RNA surveys by Eilers et al., 2000). Furthermore, these bacteria are commonly characterized by having a copiotrophic lifestyle, thus a preference for high nutrient availability, a condition that is typically episodic in surface open-ocean. These considerations have put the ecological relevance of bacterial isolates and the rare bacterial biosphere into question.

However, there is increasing evidence that rare bacteria can have a significant impact on the ecosystem dynamics, which may be in part related to their high activity when conditions favor growth. For example, a large, fast-growing single strain of Alteromonas sp. AltSIO (Alteromonadaceae) was shown to consume the entire pool of dissolved organic carbon in a coastal ecosystem (Pedler et al., 2014). Furthermore, peaks in abundance of bacteria classically considered rare, such as Ulvibacter and Formosa sp. of Flavobacteriia, were observed during phytoplankton blooms (Teeling et al., 2012). There are also some reports of culturable bacteria accounting for 12-14% of the indigenous bacterial populations (Lebaron et al., 2001). Collectively, these observations are beginning to challenge perceptions of the ecological irrelevance of rare and cultivable bacteria. Our views on the ecology of culturable bacteria may change as we begin to accumulate information on their identity, physiology and genetic characteristics and integrate this information with culture-independent studies (Hagström et al., 2017).

# Experimental Design and Culture Conditions

Bacterial physiology and metabolism were examined at five dissolved Cu levels (dCu) from limiting to replete, with a 50-fold variation in the total dCu (Table 2). At each Cu concentration, the following measurements were obtained: (1) growth rates; (2) cellular Cu quota; (3) major cellular elemental composition [carbon (C), nitrogen (N), phosphorus (P) and sulfur (S)]; and (4) bacterial O<sub>2</sub> consumption rates. Bacteria were grown at  $19 \pm 1^{\circ}$ C as semi-continuous batch cultures in the chemically well-defined artificial seawater medium Aquil (Price et al., 1989) modified for culturing marine heterotrophic bacteria (Granger and Price, 1999). Organic substrates were added to a final concentration of 0.5 g L<sup>-1</sup> and consisted of casein hydrolysate and bactopeptone (1:1 ratio, Sigma-Aldrich). Organic substrate stocks were purified separately with Chelex 100 resin (25 g per 100 mL of organic substrate, Bio-Rad) prepared following the protocol of Price et al. (1989). These purified stocks were

<sup>&</sup>lt;sup>3</sup>https://github.com/aposacka/Marine\_het\_bacteria-copper

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Isolated from	Strain	Latitude (N)	Longitude (W)	Habitat	[dCu] (nmol L <sup>-1</sup> )	pCu (-log[Cu <sup>2+</sup> ])	[Cu'] (fmol L <sup>-1</sup> )	Phylum	Class	Order	Family	Relative in Silva database h	Shared sequence omology (%)
Coastal Georgia	DSS-3	31.98° N	81.022° W	Coastal	pq	nd	nd	Proteobacteria	ø	Rhodobacterales	Roseobacteraceae	Ruegeria pomeroyi DSS-3 (ATCC collection)	n/a
NE Pacific Stn P2	PAIt-P2	48.60° N	126.01° W	Coastal	pu	nd	pu	Proteobacteria	Y	Alteromonadales	Pseudoalteromondaceae	Pseudoalteromonas sp. BSw20080	62.66
NE Pacific Stn P4	PAlt-P4	48.66° N	126.66° W	Coastal	1.8–2.4	14.7–18.8	44-66	Proteobacteria	~	Ateromonadales	Pseudoalteromondaceae	Pseudoalteromonas sp. (148Z-14, 146Z1-2, BSw10087)	99.65
NE Pacific Stn P16	PAlt-P16	49.28° N	134.66° W	Oceanic	1.7-1.9	14.6	59-60	Proteobacteria	Y	Alteromonadales	Pseudoalteromondaceae	Pseudoalteromonas sp. 148Z-14	99.38
NE Pacific Stn P16	Dokd-P16	49.28° N	134.66° W	Oceanic	1.7-1.9	14.6	59-60	Bacteroidetes f	Tavobacteriia	Flavobacteriales	Flavobacteriaceae	Dokdonia sp. 4H-3-7-5, Dokdonia diaphoros	99.29 99.28
NE Pacific Stn P26	PAlt-P26	49.99° N	144.99° W	Oceanic	2.1–2.2	14.9–15	19-60	Proteobacteria	~	Alteromonadales	Pseudoalteromondaceae	Pseudoalteromonas sp. (148Z-14, 146Z1-2, BSw10002)	99.72
Information ranges of v <sub>i</sub>	on the isc alues in sai	lation locat mples from	ion and the p. 10 to 40 m aı	hylogeny o t each stati	if the model o	rganism Rueger Semeniuk et al	ia pomeroyi (2016). nd,	DSS-3 (isolated ir. not determined; n	June 1996; . /a, not applic	ATCC culture colle able. Dissolved Cu	ction) is also included. Diss (dCu); Free Cu ion concenti	olved Cu (dCu), pCu an ation expressed as a ne	d Cu', represent gative log (pCu);

TABLE 2   Total dissolved copper (Cutot), inorganic Cu (Cu') and free Cu ion
concentrations (pCu, -log[Cu <sup>2+</sup> ]) in bacterial growth media.

Cu treatment	[Cu <sub>tot</sub> ] (nmol L <sup>-1</sup> )	[Cu'] (fmol L <sup>-1</sup> )	[pCu] (-log[Cu <sup>2+</sup> ])
Deplete (no addition)	0.6 <sup>1</sup>	4.6	15.49
Low	2	19.8	14.86
Replete	10	81.4	14.25
High	25	204.4	13.85
High	50	388.9	13.57

Calculations were performed with Visual MINTEQ version 3.1 (software is available from https://vminteq.lwr.kth.se/) excluding the organic substrate additions from calculations. The dissolved Cu reported in the lowest Cu treatment (without Cu addition) represents background Cu contamination in the chelexed SOW used for media preparation (measured by FIA-CL analysis, Semeniuk, 2014). <sup>1</sup>From background Cu contamination.

then used to amend 250 mL of artificial seawater (SOW) in trace metal-clean polycarbonate bottles and the media were microwave-sterilized for approximately 7 min. Once cooled, the SOW with organic substrates was amended with filter-sterilized additions of vitamins and trace metal stocks. Trace metals were buffered with 100  $\mu$ M ethylenediaminetetraacetic acid (EDTA), and except for Cu, their total concentrations in the media were identical to those in Maldonado et al. (2006) and as reported in **Supplementary Table 1**. Copper was either omitted or added separately as Cu-EDTA complex (1:1.1) to create 5 different Cu treatments (**Table 2**). All plastics used in the growth experiments were sterilized and rigorously cleaned prior to use by storage in Extran, followed by 10% HCl for at least 24 h. All bacterial culture manipulations were done under sterile and trace metal clean conditions using a laminar flow hood.

In Aquil medium, EDTA and trace metals are manipulated to buffer the concentrations of the free metal ions-typically the most bioavailable metal species—at environmentally relevant concentrations. Because of the unknown affinity of Cu for the organic substrate added to the bacterial growth media, Cu speciation in these media cannot be precisely estimated. For this reason, Cu treatments are primarily reported here in terms of the total dissolved Cu (Cutot). Nevertheless, we also provide the estimates for the concentrations of free Cu (pCu =  $-\log$  $[Cu^{2+}]$ ) and of inorganic Cu (Cu') in the growth media (ignoring the organic substrate additions) using the chemical equilibrium model Visual MINTEQ (version 3.1, Gustafsson, 2016, Table 2). These values are only approximate but are included here as a reference, as previously done for other studies of marine heterotrophic bacteria involving organically amended Aquil (Granger and Price, 1999; Fourquez et al., 2014).

Granger and Price (1999) found that the concentration of kinetically labile Fe was higher in Aquil modified for heterotrophic bacteria compared to the traditional Aquil (with the same amount of Fe and EDTA) and suggested that the additions of organic substrates for bacterial growth may have enhanced Fe lability. This may be related to the competition between the stronger (EDTA) and the weaker ligands for trace metals (i.e., organics such as bactopeptone and casein hydrolysate) in bacterial culture media. Indeed, there is some evidence that the presence of weaker ligands (e.g., cysteine and histidine) in seawater media containing a stronger ligand (e.g., EDTA) can increase Cu and Zn bioavailability (Aristilde et al., 2012; Kim et al., 2015; Semeniuk et al., 2015; Walsh et al., 2015). Unfortunately, the kinetically labile (bioavailable) concentrations of Cu, Co, Mn, and Zn in organics-amended Aquil have not been evaluated, but it is possible that, as with Fe, their levels are higher than in the traditional Aquil media. Therefore, the levels of Cu' and pCu we report for our organically enriched Aquil media are only approximate values that probably underestimate the true values of Cu bioavailability. Nevertheless, the approximate levels of inorganic Cu [Cu'] in our experiments encompass a range of concentrations that may be experienced by bacteria in both coastal and open-ocean environments (Moffett, 1995; Jacquot et al., 2013; Semeniuk et al., 2016).

#### **Growth Rate Measurements**

Bacterial growth rates were determined as previously described (Granger and Price, 1999). Briefly, cells acclimated to different Cu treatments (for eight generations) were inoculated into trace metal clean, sterile polystyrene cuvettes with caps and a stir bar (all sterilized by microwaving). Cultures were incubated in the spectrophotometer (Cary 1E UV-Vis, Varian) at 19°C and their optical density (A = 600 nm) was measured at 5 min interval. Absolute growth rates were determined during the exponential phase from log-linear regressions of absorbance versus time.

## **Enumeration of Bacteria**

Bacterial cell numbers were estimated using flow cytometry analysis as described in Brussaard (2004). Cultures were diluted (x5) with pre-filtered SOW (0.22  $\mu$ m polycarbonate filters) before the addition of glutaraldehyde (Electron Microscopy Grade, Sigma-Aldrich) to a final concentration of 0.5 % (i.e., 980 µL of culture sample containing 20  $\mu$ L of 25% glutaraldehyde). Samples were taken in duplicate, flash frozen with liquid nitrogen and stored at -80°C until analysis. For flow cytometry analysis with FACS Calibur (Becton-Dickinson, Franklin Lakes, NJ, United States) samples were diluted up to 1:1000 in sterile 0.1 µm filtered 10:1 TE buffer (10 mM Tris-HCL and 1 mM EDTA, pH 8), stained with SYBR Green I (x10000, Invitrogen, Waltham, MA, United States) at a final concentration of 0.5  $\times$   $10^{-4}$  of commercial stock. Samples were incubated for 15 min in the dark before the processing run for 1 min at a high rate. Event rates were kept between 100 and 1000 per second and green fluorescence and side scatter detectors were used. Data were processed and gated using Cell-Quest software (Becton-Dickson).

# Intracellular Carbon and Nitrogen Analysis

Samples in mid- to late-exponential growth phase (10–20 mL) were filtered by gravity onto 25 mm pre-combusted Whatman GF/F filters (>4 h, 450°C). Cells were rinsed with 2 mL of 0.6 M NaCl to remove extracellular organics and the filters were preserved at  $-20^{\circ}$ C until analysis. To normalize the C and N data to per cell basis, samples for flow cytometry analysis were collected from the bacterial cultures before filtration onto GF/F

filters, as well as from the filtrate. This allowed us to correct the cell numbers for any cell losses during filtration. However, on average, more than 99% of the bacteria were retained on the filters, similarly to a previous study using gravity filtration to collect cultured heterotrophic bacteria for C analysis (Pedler et al., 2014). Procedural blanks consisted of filters that were rinsed with media containing no bacterial cells. These blanks were typically < 5 % of the signal of the sample and the precision of the analysis ranged from 1 to 6% (range in RSD of 20 duplicate samples). Filters were dried overnight at 50°C, fumed with HCl, and dried again at 50°C before analysis on Perkin Elmer analyzer.

#### Intracellular P, S, and Trace Metal Analysis

Samples for trace metal analysis were collected in midexponential growth phase. From each biological replicate, 50 mL of culture was collected and centrifuged at 9,000  $\times$  g for 10 min. The supernatant was decanted, and cells were rinsed with cold (4°C) 20 mL of oxalate wash (pH 7.0, Tang and Morel, 2006) for 15 min at 4°C. Preliminary experiments suggested that washing bacterial cells with the oxalate solution does not lead to significant leakage of intracellular material, as described in Methods Section "Assessment of Cell Leakage Due to Washes." Cells were then centrifuged for 10 min (9,000  $\times$  g, 4°C) and rinsed with 20 mL of chelexed 0.6 M NaCl for 5 min at 4°C and centrifuged again. The pellet was then re-suspended in 1 mL of chelexed 0.6 M NaCl and transferred to a clean microcentrifuge tube. An aliquot  $(5 \,\mu L)$ was taken for cell enumeration by flow cytometry. Subsequently, the sample was centrifuged for 10 min (10,000  $\times$  g, 4°C), the supernatant was removed, and the pellets were immediately stored at  $-20^{\circ}$ C until analysis. The metal content of bacterial pellets was determined via inductively-coupled plasma-mass spectrometry (ICP-MS) using an instrument equipped with a hyperbolic quadrupole mass analyzer and octopole collision cell (Agilent, 7700). Bacterial pellets were digested in 5 mL Teflon digestion vials, which were cleaned following these steps: (1) detergent Extran for 24 h ( $50^{\circ}$ C); (2) 6N HCl for 1-week ( $50^{\circ}$ C); and (3) 1N HNO<sub>3</sub> for 1-week (50°C). Vials were rinsed in between the cleaning steps with MilliQ water (x 6 rinses) and were leached in MilliQ for a week at 50°C before use. Pellets were digested in 3 mL of ultrapure HNO<sub>3</sub> at 120°C for 7 h, cooled down, and evaporated to dryness overnight at 120°C. Subsequently, 1 mL of ultrapure H<sub>2</sub>O<sub>2</sub> was added to each vial and evaporated to dryness at 120°C for  $\sim$  4 h. Procedural blanks consisted of vials with no pellets treated in the same way as the samples. For ICP-MS analysis, vials were filled with a matrix solution, consisting of 2% HNO3 spiked with 10 ppb In as an internal standard, and allowed to stand for 10 min before being transferred to 15 mL trace metal clean polyethylene tube. The final volume of samples was subsequently adjusted to  $\sim 10$  mL.

The instrument was calibrated using multi-element standard (High-Purity Standards, Charleston, SC, United States) prepared in the same matrix solution as the samples. Two isotopes of copper (Cu 63/65) were measured to ensure that background and interference were acceptably low. Average procedural blank

was 0.05 nmol Cu  $L^{-1}$  and the limit of detection calculated as 3  $\times$  SD of the blank was 0.07 nmol L<sup>-1</sup>, with samples values on average > 40-100 times higher than the detection limit. Determination of phosphorus (<sup>31</sup>P) by quadrupole ICP-MS is challenging due to high background and interferences from polyatomic ions such as <sup>15</sup>N<sup>16</sup>O<sup>+</sup>; <sup>14</sup>N<sup>17</sup>O<sup>+</sup>; <sup>14</sup>N<sup>16</sup>O<sup>1</sup>H<sup>+</sup> and <sup>30</sup>Si H<sup>+</sup>. We assessed P data obtained using ICP-MS by comparison with measurements of P in the same samples by ICP-Optical Emission Spectroscopy (OES). There was an excellent agreement between the P measurements made by these two different instruments (Supplementary Figure 2), thus confirming the accuracy of the P concentrations obtained in our samples with ICP-MS. Sulfur data were obtained using ICP-OES. Trace metal data were normalized using P, cell numbers or cellular carbon data. We focus on P normalized quotas because this element was measured simultaneously with Cu during ICP-MS analysis, thus eliminating errors due to different analytical techniques. Phosphorous normalization allows for comparison of metal contents independently of bacterial cell volume, which was not measured during this study. While bacterial P content, can be plastic depending on environmental conditions (e.g., organic substrate quality, Godwin and Cotner, 2015), we found no statistically significant changes in cell normalized P content in response to Cu (see Results, Section "Macronutrient Cellular Composition in Response to Cu"). Indeed, P-normalized Cu quotas followed the same trends as those of the cell- and C-normalized Cu quotas (Supplementary Figure 3), suggesting that organic P was an appropriate biomass proxy.

# Assessment of Cell Leakage Due to Washes

Determination of intracellular elements requires a washing step to remove any elements adsorbed to the outside of the cells. There are various washing protocols, depending on the element of interest (e.g., Fe vs. Cu) and the method used for metal quota determination (e.g., radioisotope techniques vs. ICP-MS). However, washing solutions could affect cell membrane integrity and lead to a loss of intracellular elements, resulting in underestimated elemental quotas.

In this study, we examined whether two commonly used washes, oxalate-EDTA (Tang and Morel, 2006) and DTPA (Croot et al., 2003), induce cell leakage in bacteria using <sup>14</sup>C radiolabeled glucose as a tracer of intracellular material. For these experiments, the oceanic Pseudoalteromonas sp. PAlt-P26 strain was selected as it grew successfully in modified media containing glucose as a C source ( $\mu \sim 3 d^{-1}$ ), unlike *Dokdonia* sp. strain Dokd-P16, which we were unable to grow in this medium. The modified medium contained 35 mM glucose, 6 mM ammonium and 0.1 mM phosphate in filtered seawater (Weaver et al., 2003). We used water from Ocean Station Papa (OSP, station P26, ~10 m) amended with vitamins and trace metals, as for the Aquil media recipe. A small inoculum of exponentially growing PAlt-P26 acclimated to the modified media was transferred into 250 mL medium containing 20 µCi <sup>14</sup>C-glucose and cultured overnight at room temperature. Wash

test experiments involved filtering two aliquots (2.5 mL) of exponentially growing bacterial culture (0.22  $\mu$ m polycarbonate filters) and incubating one aliquot with a wash (2.5 mL, oxalate or DTPA solutions) and the other aliquot with sterile SOW (2.5 mL) for 15 min in the dark at room temperature. This procedure was performed in triplicates. The oxalate and the DTPA incubated aliquots of cell culture were rinsed with sterile SOW before final filtration. Filter blanks (n = 3) were collected by subjecting filters to the same washing procedure but without bacterial cells and were used to correct the <sup>14</sup>C activity of sample filters. Cell leakage was determined by comparing the <sup>14</sup>C activity of wash-treated (oxalate or DTPA) and SOW only treated cells.

A paired Student's *t*-test was used to determine whether the differences between the washes (oxalate or DTPA) and SOW were statistically significant. There was no statistically significant difference between wash-treated and SOW-treated cells (oxalate, T-value = 3.85; p = 0.061; DTPA, T-value = 1.6; p = 0.24), supporting our conclusion that the washing solutions do not induce a significant loss of cellular contents (Supplementary Figure 4). We thus assumed that just like with <sup>14</sup>C-glucose, the loss of cellular Cu would be minimal. We do acknowledge, however, that the lability of the intracellular C and Cu pools may not be the same. In gram-negative bacteria, Cucontaining proteins are believed to be mostly located in the inner membrane and/or the periplasmic space (Tottey et al., 2005). We suspect that damage to the cell's membranes would lead to a loss of cytoplasmic C constituents, as well as membranebound Cu and Cu ions in the periplasm. Ideally, to assess Cu leakage, cellular Cu pools in the cells would have been labeled with a radioactive Cu tracer (e.g., <sup>64</sup>Cu or <sup>67</sup>Cu). Unfortunately, these tracers are not commercially available. Ultimately, oxalate solution was selected for cell washing during the collection of samples for ICP-MS analysis (as described in Section "Intracellular P, S, and Trace Metal Analysis") due to its previous use in the removal of multiple biogenic elements (e.g., Wilhelm et al., 2013), as opposed to the DTPA wash, which was developed specifically for extracellular Cu removal.

#### **Oxygen Consumption Rates**

Bacterial respiration was measured using a S1 Clark type oxygen electrode (Hansatech Instruments, Norfolk, England). Cultures were sampled in mid-exponential phase as 1 mL aliquots that were transferred to a clean O<sub>2</sub> electrode chamber connected to a 19°C water bath. Oxygen consumption rates were derived as the slope of O<sub>2</sub> consumption over time (10 min) in the chamber containing the bacterial culture. Prior to these measurements, the electrode was calibrated by filling the electrode chamber with O<sub>2</sub>-saturated synthetic seawater (SOW, bubbled with O<sub>2</sub> for 1 h prior to experiment in a water bath at 19°C). Subsequently, the SOW in the electrode chamber was gently bubbled with N<sub>2</sub> to record the O<sub>2</sub>-minimum signal. The electrode was controlled by a CB1D O<sub>2</sub> electrode control box (Hansatech) and the data were collected and processed with the Labjackoxy software (Hansatech). For each biological replicate, two or more technical replicates were collected during the exponential phase

and averaged. The  $O_2$  consumption rates were normalized to bacterial cell abundance.

# Bacterial Carbon Metabolism Calculations

Growth and respiration rates, as well as C content data, were used to derive different estimates of heterotrophic C metabolism as previously described (del Giorgio and Cole, 1998):

$$BP = Growth rate (d^{-1}) \times Cellular C (fmol C cell^{-1})$$
(1)

$$BCD = BP (fmol C cell-1 d-1) + BR_{cell} (fmol C cell-1 d-1)$$
(2)

$$BGE = BP/BCD \tag{3}$$

where BP stands for bacterial productivity (fmol C cell<sup>-1</sup> d<sup>-1</sup>), BCD for bacterial carbon demand (fmol C cell<sup>-1</sup> d<sup>-1</sup>), BR for bacterial respiration (fmol C cell<sup>-1</sup> d<sup>-1</sup>), and BGE for bacterial growth efficiency (ratio, unitless).

These metabolic rates were derived for *Dokdonia* sp. strain Dokd-P16, oceanic *Pseudoalteromonas* sp. strain PAlt-P26 and *R. pomeroyi* DSS-3 using triplicates measurements. For these estimates, bacterial respiration rates were converted from the amount of oxygen consumed (fmol  $O_2$  cell<sup>-1</sup> d<sup>-1</sup>) to the amount of carbon respired (fmol C cell<sup>-1</sup> d<sup>-1</sup>) using a respiratory quotient (RQ – ratio of  $O_2$  consumed to CO<sub>2</sub> produced) of 1, as has been assumed in other studies (del Giorgio and Cole, 1998; Reinthaler and Herndl, 2005; Hörtnagl et al., 2011; Pradeep Ram et al., 2016). While RQ can vary depending on the organic

substrate quality (0.7–1.2, Berggren et al., 2012), it is less clear if changes in inorganic nutrients affect RQ. Fourquez et al. (2014) found that Fe-limited marine heterotrophic bacterium *Alteromonas macleodii* alters certain C metabolic pathways, which may have an effect on RQ. However, no studies have yet quantified the magnitude of this effect on RQ due to Fe limitation or any other metal.

#### **Statistical Analysis and Figures**

The statistical analyses were conducted using an open-source programming language R (R Core Team, 2016). Data processing was done using package "dplyr" (Wickham, 2016), while all figures were generated using "ggplot" (Wickham, 2009) and "cowplot" (Wilke and Wickham, 2016) packages in R. All the data presented were calculated as means  $\pm$  standard deviation. One-way Analysis of Variance (ANOVA) was used to determine the significance among different Cu treatments for individual bacterial strains with a significance level  $\alpha$  of 0.05. Post hoc analyses were performed using a pairwise t-test with Bonferroni correction of  $\alpha$  (at 0.05) when the variance was homogeneous using the R base package "stats". For data that displayed heterogeneity of variance, we applied Tukey's honest significance difference test, using "multcomp" package in R (version 1.4-6, Herberich et al., 2010). Data and code used to produce figures, summary tables and statistical tests in this manuscript can be found in publicly available repository on Github<sup>4</sup>.





results are presented in Table 4. Note x-axis is not linear.

## RESULTS

## Growth Rates and Cu Quotas in Response to Changing Cu Availability

We found diverse responses of bacterial growth rates and Cu quotas to changing Cu availability (Figure 1). Both coastal and oceanic strains of Pseudoalteromonas sp. (PAlt-P2 and PAlt-P26, respectively) had the fastest growth rates, and minor reductions in growth rates in the lowest Cu treatments ( $Cu_{tot} = 0.6$  nmol  $L^{-1}$ ,  $\mu/\mu_{max}$  of 0.72 and 0.84 for *Pseudoalteromonas* sp. PAlt-P2, and PAlt-P26, respectively, Table 3). These strains also had similar Cu quotas (Cu:P, mmol:mol) across all Cu treatments (Figure 1). Statistical analysis confirmed that neither their growth rates nor the Cu:P ratio were significantly affected by Cu concentrations (Table 4). Interestingly, lack of response in stoichiometry when growth rates are above 0.6  $\mu/\mu_{max}$ has been previously reported for freshwater bacterial isolates, although for macronutrients (C, N, P, Godwin and Cotner, 2018). In contrast, while the growth rate of the Roseobacter member, R. pomeroyi DSS-3, was not affected by changes in Cu availability (F = 2.1, p = 0.158, Table 4 and Figure 1), its Cu quota (Cu:P, mmol:mol) varied significantly (F = 20, p < 0.001, Table 4 and Figure 1) showing a 2-fold reduction from the highest to the lowest Cu treatment (Cu:P =  $0.15 \pm 0.01$ versus  $0.05 \pm 0.01$  mmol:mol, respectively; Table 3). The growth rate of the Flavobacteriia member, Dokdonia sp. strain Dokd-P16, was significantly affected by Cu availability (F = 79, p < 0.001, Table 4), and we found up to  $\sim 80\%$  reduction in growth when the cultures were most deprived of Cu (at 0.6 nmol Cu  $L^{-1}$ , Table 3) compared to the maximal growth rate under Cu replete conditions. For this strain, reductions in growth rates were accompanied by  $\sim$ 50% decrease in Cu quotas at the two lowest Cu levels (average 0.7 mmol Cu:mol P for the 0.6 and 2 nM Cu cultures) compared to the two highest Cu replete treatments (average 0.13 mmol Cu:mol P for 25 and 50 nM Cu treatments, Figure 1 and Table 3). Interestingly, similar decreases in Cu quota were observed at moderate Cu levels in the media (10 nmol Cu  $L^{-1}$ ) even though the growth rate was unaffected (Figure 1 and Table 3).

# Macronutrient Cellular Composition in Response to Cu

The major element composition of bacterial strains at varying Cu levels were assessed as cell-normalized carbon (C), nitrogen (N), phosphorus (P) and sulfur (S) (**Figure 2** and **Supplementary Table 2**). In general, cellular macronutrients did not vary significantly in response to the Cu treatments (**Figure 2**; ANOVA results, **Supplementary Table 3**). When averaged across all treatments, the oceanic *Pseudoalteromonas* sp. strain PAlt-P26 had the highest macronutrient quotas [cellular fmolar quota (fmol cell<sup>-1</sup>) of 27 C, 6 N, 0.5 P, and 0.2 S], followed by the coastal *Pseudoalteromonas* sp. strain PAlt-P2 (fmol cell<sup>-1</sup> of 22 C, 6 N, 0.5 P, and 0.2 S), *R. pomeroyi* DSS-3 (fmol cell<sup>-1</sup> of 16 C, 4 N, 0.2 P, and 0.15 S) and *Dokdonia* sp. strain Dokd-P16 (fmol cell<sup>-1</sup> of 10 C, 2 N, 0.2 P, and

0.12 S). The larger cellular macronutrient content of the *Pseudoalteromonas* sp. strains may reflect their larger cell size compared to *R. pomeroyi* DSS-3 and Dokd-P16. To further assess the effects of Cu on the elemental composition of bacteria, we determined the stoichiometric ratios of major elements (C:N and S:P, **Figure 2**). We consider ratios only for those elements that were analyzed simultaneously (C and N, P, and S) to minimize the error due to differences in analytical techniques. In all bacterial strains, the molar ratios of both C:N and S:P were not significantly affected by changing Cu availability (ANOVA results, **Supplementary Table 3**). The elemental ratios determined in this study are on par with those of native and cultured marine and freshwater bacteria reported in Fagerbakke et al. (1996) (**Figure 2**).

#### Regulation of Bacterial Carbon Metabolism by Cu Availability

Using growth rate, cellular C, and bacterial respiration data, we estimated various aspects of heterotrophic C metabolism and examined its sensitivity to changes in Cu availability for three strains: Dokd-P16, oceanic PAlt-P26, and R. pomeroyi DSS-3 (Figure 3). Carbon metabolism of bacteria can be distinguished into anabolic (bacterial productivity, BP) and catabolic reactions (bacterial respiration, BR), which can be used to derive bacterial carbon demand (BCD; the carbon biomass that is required to sustain net bacterial metabolic needs) and bacterial growth efficiency (BGE; carbon assimilation efficiency into biomass) (del Giorgio and Cole, 1998). Of all strains, only Dokd-P16 was found to significantly reduce all metabolic rates (BR, BP, and BCD) and BGE in response to low Cu availability (One-way ANOVA, Table 4). Copper limitation (0.6 and 2 nmol Cu L<sup>-1</sup> treatments) was associated with  $\sim$ 20% reduction in cellular respiration (BR<sub>cell</sub>),  $\sim$  40% reduction in productivity, and  $\sim$ 40-50% reduction in both BCD and BGE, compared to Cu-replete conditions (25 and 50 nmol Cu  $L^{-1}$ , Figure 3 and Supplementary Table 4). Furthermore, BR and BCD were significantly reduced at moderate Cu levels (10 nmol Cu  $L^{-1}$ ) compared to the two highest Cu treatments (Figure 3), yet, BGE estimates were similar across these treatments. In R. pomerovi DSS-3, BR and BCD were also significantly affected by Cu availability (Table 4), but in contrast to Dokd-P16, these rates were substantially elevated under Cu limitation (higher at 0.6 than at 50 nmol Cu  $L^{-1}$  treatment; p < 0.05, Figure 3 and Supplementary Table 4). Finally, none of the carbon metabolism parameters varied significantly for PAlt-P26 in response to changing Cu availability (Table 4 and Figure 3).

The BGE estimates in this study varied from 0.3 to 0.6 and are in the range of those reported for *Vibrio harveyi* (<0.1-0.5, Kirchman et al., 2003a), and are typical of bacterial communities in eutrophic systems (as reviewed in del Giorgio and Cole, 1998). Respiration rates obtained for PAlt-P26 (ranging from 344 to 449 fmol O<sub>2</sub> cell<sup>-1</sup> d<sup>-1</sup>, **Supplementary Table 4**) are higher than those of coastal and oceanic *Alteromonas macleodii* strains

**TABLE 3** | Specific growth rates (d<sup>-1</sup>), relative growth rates ( $\mu/\mu_{max}$ ), and phosphorus normalized Cu quotas (Cu:P, mmol:mol) of bacterial strains at various levels of Cu in the growth media (nmol Cu L<sup>-1</sup>).

Strain	$Cu_{tot}$ (nmol L <sup>-1</sup> )	n	Growth rate (d <sup>-1</sup> )	$\mu/\mu_{max}$	n	Cu:P (mmol:mol)
Dokdonia sp.	0.6	6	1.61 ± 0.7	0.18	4	0.06 ± 0.03
strain Dokd-P16	2	11	$2.83\pm0.5$	0.33	6	$0.08\pm0.02$
	10	7	$6.48\pm0.8$	0.74	7	$0.08\pm0.03$
	25	9	$6.94 \pm 0.7$	0.79	6	$0.14\pm0.03$
	50	10	$8.72 \pm 2.4$	1	3	$0.12\pm0.003$
Pseudoalteromonas sp.	0.6	4	$9.00\pm0.9$	0.72	4	$0.03\pm0.02$
strain PAlt-P2 (coastal)	2	nd	nd	nd		nd
	10	5	$12.4 \pm 2.06$	1	3	$0.04 \pm 0.01$
	25	7	$11.3 \pm 1.07$	0.92	3	$0.03\pm0.04$
	50	nd	nd	nd		nd
Pseudoalteromonas sp.	0.6	20	$15.2 \pm 1.7$	0.84	4	$0.06\pm0.03$
strain PAlt-P26 (oceanic)	2	6	$18.2 \pm 1.8$	1	4	$0.05\pm0.02$
	10	33	$18.0 \pm 2.2$	1	3	$0.05 \pm 0.04$
	25	3	$17.1 \pm 0.6$	0.95		nd
	50	6	$15.2 \pm 2.2$	0.84	3	$0.03\pm0.01$
R. pomeroyi DSS-3	0.6	11	$4.68 \pm 0.7$	0.85	3	$0.05\pm0.01$
	2	4	$5.47 \pm 0.7$	1	4	$0.12 \pm 0.03$
	10	9	$4.42 \pm 1.0$	0.80	3	$0.15 \pm 0.02$
	25	3	$3.94\pm0.5$	0.72	nd	nd
	50	9	$5.47\pm0.8$	1	3	$0.15\pm0.01$

Data are means  $\pm$  standard deviation. nd, not determined.

TABLE 4 | One-way ANOVA values for the effect of Cu concentration in the media on various metabolic variables of the bacterial strains.

	ANOVA results											
	Dokd	·P16	PAlt	·P2	PAIt-	PAIt-P26		ri DSS-3				
Variable	F statistic	p-value	F statistic	p-value	F statistic	p-value	F statistic	<i>p</i> -value				
Growth rate	79	<0.001	2.08	0.170	0.25	0.612	2.1	0.158				
Cu quota (Cu:P)	7.1	<0.001	0.84	0.467	0.74	0.698	20	< 0.001				
BR <sub>cell</sub>	14	<0.001	nd	nd	0.97	0.452	7.7	0.022				
BR <sub>carb</sub>	4.1	0.032	nd	nd	0.74	0.554	5.6	0.043				
BP	17	<0.001	nd	nd	0.32	0.805	0.3	0.756				
BCD	46	<0.001	nd	nd	0.25	0.857	11.9	0.008				
BGE	9.0	0.002	nd	nd	1.11	0.399	1.98	0.218				

Cell normalized bacterial respiration (BR<sub>cell</sub>), carbon normalized respiration (BR<sub>carb</sub>), bacterial productivity (BP), bacterial carbon demand (BCD), bacterial growth efficiency (BGE). Statistically significant effects are shown in bold. Cellular C, N, P, S as well as C:N and S:P were tested with no significance (results of ANOVA can be found in **Supplementary Table 3**). nd, not determined.

(order *Alteromonadales*), cultured in Aquil modified media with replete Fe concentrations (226 fmol  $O_2$  cell<sup>-1</sup> d<sup>-1</sup>, Fourquez et al., 2014).

## DISCUSSION

# Responses of Bacterial Isolates to Changing Cu Availability

#### Bacterial Growth and Cellular Cu Quotas

We investigated four strains of marine heterotrophic bacteria from three distinct taxa, and found that these bacteria showed

three contrasting responses to changes in Cu availability. First, with the exception of macronutrient content and stoichiometry, all physiological assessments of the *Flavobacteriia* member *Dokdonia* sp. Dokd-P16 (oceanic) were affected by low Cu, suggesting that this bacterium has a metabolism highly dependent on Cu availability. The growth rate of Cu-limited Dokd-P16 was severely reduced, as well as its cellular Cu. Second, the *Roseobacter* member *R. pomeroyi* DSS-3 (coastal) also reduced its Cu content under low Cu concentrations but its growth rate remained unchanged. Interestingly, both Dokd-P16 and *R. pomeroyi* DSS-3 had similar cellular Cu quotas at various Cu levels (mean 0.05–0.15 mmol Cu:mol P, **Figure 1** 



**FIGURE 2** Cellular carbon (fmol C cell<sup>-1</sup>), nitrogen (fmol N cell<sup>-1</sup>), phosphorus (fmol P cell<sup>-1</sup>), sulfur (fmol S cell<sup>-1</sup>), and molar macronutrient ratios (C:N and S:P) in *Dokdonia* sp. strain Dokd-P16, *R. pomeroyi* DSS-3, coastal *Pseudoalteromonas* sp. strain PAlt-P2, oceanic *Pseudoalteromonas* sp. strain PAlt-P26 at different levels of Cu (nmol L<sup>-1</sup>). The horizontal line represent the mean and the superimposed colored circles are data for each biological replicate that was used in the calculation of mean. The horizontal dashed lines indicate the average C:N and S:P ratios for native and cultured bacteria reported in a study by Fagerbakke et al. (1996), with the shaded regions spanning the range of each of the macro-element ratios reported in Fagerbakke et al. (1996). The data used to produce the figures can be found in **Supplementary Table 2**, and the ANOVA results for the effect of Cu on major elemental composition in the **Supplementary Table 3**. Note *x*-axis is not linear.

and **Table 3**). Thus, their contrasting growth responses to Cu deficiency cannot be readily explained by their cellular Cu. The third response characterizes both strains of *Pseudoalteromonas* sp. These bacteria had unaffected growth rates, as well as low and invariant average cellular Cu quotas despite a 50-fold variation in Cu availability (mean 0.03–0.06 mmol Cu:mol P, **Table 3**, One-way ANOVA, **Table 4**), indicating tightly controlled intracellular Cu levels. Our results suggest that both strains of

*Pseudoalteromonas* sp., as well as *R. pomeroyi* DSS-3 would be more successful under low Cu availability than Dokd-P16. While future genomic and proteomic studies are needed to identify the metabolic pathways affected by low Cu in Dokd-P16, our findings indicate that this strain may have unique Cu requirements.

One important observation in our study is that, although there was some strain-specific variability in bacterial Cu quotas,



bacterial productivity (BP, fmol C d<sup>-1</sup>), bacterial carbon demand (BCD, fmol C d<sup>-1</sup>) and bacterial growth efficiency (BGE, unitless) of *Dokdonia* sp. strain Dokd-P16, *Ruegeria pomeroyi* DSS-3, *Pseudoalteromonas* sp. strain PAlt-P26. The horizontal line represents the mean and the superimposed colored circles are data for each biological replicate that was used in the calculation of mean. Significant effects (p < 0.05) are shown with different letters for Dokd-P16 and *R. pomeroyi* DSS-3 (pairwise *t*-test with Bonferroni correction of  $\alpha$  at 0.05). The data used to produce the figure can be found in the **Supplementary Table 4**. Note *x*-axis is not linear.

the range of Cu:P ratios was narrow (~5-fold, 0.03–0.15 mmol Cu:mol P) relative to the 50-fold variation of Cu concentrations in the growth media. Much of our understanding of bacterial Cu homeostasis comes from studies with model bacteria such as *Escherichia coli*. These studies suggest that bacteria are highly sensitive to Cu toxicity and that their Cu trafficking and cellular inventories are under strict management via a variety of mechanisms (as reviewed by: Rensing and Grass, 2003; Solioz and Stoyanov, 2003; Solioz et al., 2010; Argüello et al., 2013; Bondarczuk and Piotrowska-Seget, 2013). In *E.coli*, Cu efflux (via a regulatory protein CueR) is activated at zeptomolar ( $10^{-21}$ ) Cu<sup>2+</sup> levels (<600 atoms per cell), suggesting that this bacterium

is extremely sensitive to free Cu in the cytosol, and operates Cudetoxification mechanisms even under Cu deprivation (Changela et al., 2003). Furthermore, with the exception of photosynthetic bacteria, no cytosolic Cu-containing proteins have been reported in bacteria, and in *E. coli* these proteins are confined to the periplasm or the cytoplasmic membrane (Tottey et al., 2005). Taken together, bacteria are required to maintain low cellular Cu in order to avoid Cu toxicity, which can cause oxidative stress and destruction of Fe–S clusters in proteins (reviewed in Dupont et al., 2011). Therefore, the narrow range of Cu quotas displayed by marine heterotrophic bacteria from this study relative to the broad Cu availability in the growth media appears to fit with the model of strict Cu regulation in gram-negative bacteria.

#### Copper Availability and Dokdonia sp. Dokd-P16

As shown in this study, Cu is a critical micronutrient for the Flavobacteriia member Dokdonia sp. Dokd-P16. Indeed, Dokd-P16 significantly reduces its growth rates (33% of  $\mu_{max}$  at 2 nmol Cu L<sup>-1</sup>) at free divalent Cu (Cu<sup>2+</sup>) concentrations  $(1.38 \times 10^{-15} \text{ mol } \text{Cu}^{2+} \text{ L}^{-1})$  that are similar to those  $(1 \times 10^{-15} \text{ mol } \text{Cu}^{2+} \text{ L}^{-1})$  limiting several coastal phytoplankton (e.g., Thalassiosira weissflogii and Chaetoceros decipiens, the coccolithophore Emiliania huxleyi, and the prymnesiophyte Phaeocystis cordata; Annett et al., 2008; Guo et al., 2012). In contrast to eukaryotic phytoplankton, little is known about Cu conditions that limit the growth of marine prokaryotes; so far responses to Cu deprivation have only been explored in some species of marine denitrifying bacteria and one ammonia oxidizing archaeon (Matsubara et al., 1982; Granger and Ward, 2003; Amin et al., 2013; Moffett et al., 2012). The levels of Cu<sup>2+</sup> that limit the growth of Dokd-P16 are one order of magnitude higher ( $[Cu^{2+}] = 10^{-15} \text{ mol } L^{-1}$ ) than those found to reduce the growth of some marine denitrifiers in very similar media ( $[Cu^{2+}] < 10^{-16} \text{ mol } L^{-1}$ , Moffett et al., 2012). In contrast, free divalent  $Cu^{2+}$  levels ( $10^{-15}$  mol  $L^{-1}$ ) are two orders of magnitute lower than those limiting ammonia oxidizing archaeon ( $10^{-13}$  mol L<sup>-1</sup>, Amin et al., 2013). These results suggest that the Flavobacteriia member Dokd-P16 may have Cu requirements higher than those of denitrifying bacteria but lower than those of AOA.

Furthermore, the levels of Cu<sup>2+</sup> reported for surface waters at station P16 in the NE Pacific [Cu<sup>2+</sup> =  $2.5 \times 10^{-15}$  mol L<sup>-1</sup> (10 and 35 m), Semeniuk et al., 2016], where Dokd-P16 was isolated from, fall within the range of Cu levels in our study that resulted in relative growth rates of 74 and 33%  $\mu_{max}$  in this strain (Cu<sub>tot</sub> treatments of 10 nmol Cu L<sup>-1</sup>, Cu<sup>2+</sup> =  $5.6 \times 10^{-15}$  nmol L<sup>-1</sup> and 2 nmol Cu L<sup>-1</sup>, Cu<sup>2+</sup> =  $1.38 \times 10^{-15}$  nmol L<sup>-1</sup>, respectively). Hence, this bacterium could be experiencing Cu limitation *in situ*.

At present, little information is available on the ecology of Dokd-P16 isolate and whether it serves as a good model for natural Flavobacteriia populations. However, it is promising that the members of Flavobacteriia responded preferentially to an increase in Cu availability in a recent microcosm study in the Southern Ocean (Flavobacteria-Cytophagia cluster; Ramaiah et al., 2015), supporting our hypothesis of an important role for Cu in Flavobacteriia. However, the causes for the positive response of Flavobacteriia to Cu were not explored in study of Ramaiah et al. (2015). It remains to be determined if Flavobacteriia in wild populations favor high Cu levels for their growth and metabolism as observed for Dokdonia sp. in our study. Given the significant representation of Flavobacteriia in marine surface waters (10-70%, Glöckner et al., 1999; Eilers et al., 2000; Kirchman et al., 2003b; Abell and Bowman, 2005; Williams et al., 2013), as well as their significance in organic matter transformation (particularly the complex, high molecular weight compounds), evaluating the role of Cu in this group is of interest.

#### Composition of Major Elements in Response to Cu

Microbial biosynthesis requires a balanced stoichiometry of carbon (C), nitrogen (N), and phosphorus (P) (Herbert, 1976). Since bacterial biomass represents a major sink for these elements in the aquatic environment, understanding factors that influence bacterial stoichiometry is critical to predicting the fate of C, N, and P in the ocean. Sulfur is not naturally limiting in the ocean (seawater sulfate concentrations are approximately 28 mmol  $SO_4 L^{-1}$ ; Emerson and Hedges, 2008), and rarely considered in stoichiometric studies. However, bacterial S content is of interest given the involvement of S-containing compounds (e.g., glutathione) in intracellular Cu binding and trafficking (Helbig et al., 2008; Solioz et al., 2010).

In the present study, bacterial C:N and S:P molar ratios did not vary significantly across various Cu growth conditions. In agreement with our findings, C:N ratios were also constant for marine heterotrophic bacteria grown under different Fe regimes (Tortell et al., 1996; Fourquez et al., 2014). These invariant bacterial C:N ratios indicate that assimilatory pathways of C and N in bacteria remain highly interconnected under changing availability of bioactive metals, such as Fe and Cu. The molar ratios of all elements normalized to P (C:N:S:P) and averaged across all Cu levels for different bacterial strains are summarized in Table 5 and compared to the data from literature. Surprisingly, there are only few reports on the major elemental stoichiometry of native marine heterotrophic bacteria (i.e., Fagerbakke et al., 1996; Gundersen et al., 2002). In general, our data are comparable with those studies. The elemental stoichiometry of bacteria investigated here, for both major and minor elements, will be discussed in an upcoming publication.

#### Cu Regulation of Microbial Carbon Metabolism

Microbial growth and metabolism is dependent on the consumption of organic carbon, which at a cellular level is partitioned between biosynthetic (anabolic - bacterial production) and energy-yielding (catabolic - bacterial respiration) processes (del Giorgio and Cole, 1998; Carlson et al., 2007). This partitioning is flexible (Teixeira de Mattos and Neijssel, 1997), a strategy that allows bacteria to maximize growth depending on environmental conditions (Tempest and Neijssel, 1992; Russell and Cook, 1995). One way to parameterize the relationship between cellular catabolism and anabolism is to estimate bacterial growth efficiency (BGE - sometimes termed carbon use efficiency, CUE), which represents the ratio of C allocated toward bacterial biomass production relative to the total amount of C consumed to support bacterial respiration and growth [BGE = BP/(BP + BR)] (del Giorgio and Cole, 1998; Rivkin and Legendre, 2001). The higher the BGE values the more efficient Cu use (i.e., allocation toward bacterial growth) and lower portion of the organic C consumed being respired (as  $CO_2$ ). In contrast, low BGE's point to a greater loss of C (as  $CO_2$ ) and inefficient energy metabolism, which has been linked to imbalanced substrate stoichiometry (Keiblinger et al., 2010), viral infection (Pradeep Ram et al., 2016), UV exposure (Hörtnagl et al., 2011), and iron deficiency (Tortell et al., 1996; Kirchman et al., 2003a). However, the role of bioactive trace metals, such as Cu, in controlling BGE remains unexplored.

Reference	Organisms	C:P	N:P	S:P	S:C	C:N
This study <sup>a</sup>	Dokd-P16	<b>59.6</b> (45.6 – 75.8)	<b>13.3</b> (9.6 – 16.9)	<b>0.84</b> (0.61 – 1.1)	<b>0.014</b> (0.008 - 0.014)	<b>4.52</b> (4.3 – 4.8)
	R. pomeroyi DSS-3	<b>84.9</b> (62.1 – 106.3)	<b>20.7</b> (15 – 25)	<b>0.79</b> (0.6 - 1.1)	<b>0.008</b> (0.007 - 0.01)	<b>4.14</b> (8.9 – 4.2)
	PAIt-P2	<b>40.8</b> (35.6 – 51.7)	<b>11.7</b> (7.3 – 14.1)	<b>0.45</b> (0.4 - 0.45)	<b>0.012</b> (0.008 - 0.012)	<b>4.08</b> (3.45 - 5.0)
	PAIt-P26	<b>57.6</b> (47.2 - 81)	<b>13.9</b> (11.3 – 19.1)	<b>0.57</b> (0.4 – 0.76)	<b>0.009</b> (0.007 – 0.011)	<b>4.17</b> (3.9 – 4.6)
Fagerbakke et al., 1996 <sup>b</sup>	Native bacteria (lakes and coastal seawater)	<b>54</b> (39 – 66)	<b>10</b> (7.7 – 13.8)	<b>0.7</b> (0.28 – 1.2)	<b>0.014</b> (0.005 – 0.03)	<b>5.4</b> (4.4 – 7.2)
	E. coli	29	7.1	0.28	0.014	4.5
	V. natriegens	53	11.7	0.74	0.010	4.1
Gundersen et al., 2002 <sup>c</sup>	Native bacteria (Sargasso Sea)	58 - 142.9	16	nd	nd	5.3 – 9.1
Chan et al., 2012 <sup>d</sup>	R. pomeroyi DSS-3					
	Carbon-limited	55.6	12.7	nd	nd	4.4
	Nitrogen-limited	166.7	14.8	nd	nd	11.2
	Phosphorous-limited	335.3	35.0	nd	nd	9.5
	Sulfur-limited	62.5	12.7	nd	nd	4.9

TABLE 5 | Comparison of the molar macronutrient ratios of bacteria used in this study and values reported in literature.

<sup>a</sup>Values in bold are averages of ratios from all Cu treatments. <sup>b</sup>For native bacteria values in bold are averages of elemental ratios from five different habitats, for E. coli and V. natriegens values in bold are averages for exponentially growing biological replicates as reported in **Table 1** of the study. <sup>c</sup>C:P and C:N ratios represent the ranges of geometric mean measured for bacteria from Sargasso Sea plotted in **Figure 3** of the study (n = 5) and the N:P is the slope. <sup>d</sup>Values in bold represent average of two technical replicates from triplicate chemostat cultures as reported in **Table 1** of the study.

Here, we found a strong connection between C metabolism and Cu availability in a Flavobacteriia member Dokdonia sp. strain Dokd-P16. In this bacterium, Cu starvation led to drastically reduced growth rates (Figure 1), as well as BCD, and slower BR and BP rates (Figure 3). We suspect that slower BR rates could be partly related to the impairment of the respiratory system, since in aerobic bacteria this is the major repository of cellular Cu (in the form of COX; Ridge et al., 2008). On average, the reduction in BP ( $\sim$ 40%) was greater than that in BR ( $\sim$ 20%), indicating a higher flux of C for energy production than for biomass in Cu-limited Dokd-P16, which was confirmed by a reduction in BGE (by 50%). We also found a poor correlation between its growth and respiration rates ( $R^2 = 0.12$ , **Supplementary Figure 5**), which has been similarly noted for bacteria growing under Fe limitation, although with a higher statistical strength ( $R^2 = 0.45$ , Fourquez et al., 2014). This may be explained by the fact that bacteria allocate energy to multiple processes that are independent of cell biomass formation, notably, overflow metabolism, futile cycles and maintenance metabolism (reviewed by Russell and Cook, 1995). The energetic requirements for processes such as cell maintenance (e.g., activation of nutrient uptake system) are likely to increase under constrained growth (Russell and Cook, 1995), which may explain the trends we observed in BR and BP, and ultimately reduced efficiency of C use in Cu-limited Dokd-P16. Iron limitation led to a similar response, lowering BGE in 3 out of 5 marine heterotrophic bacteria (up to 70%, Tortell et al., 1996). Our results suggest a link between Cu availability and the fate of organic C transformed by Dokdonia sp. strain Dokd-P16.

In contrast, the growth rate of *R. pomeroyi* DSS-3 was not affected by Cu limitation, but its BR and BCD increased substantially (**Figure 3** and **Supplementary Table 4**). Given that BCD provides an estimate of the C needed to support bacterial production (Carlson et al., 2007), its increase in *R. pomeroyi* DSS-3 under low Cu suggest that the energetic demand to support BP

under these conditions is higher than when Cu is replete. Perhaps this reflects the energetic investment in mechanisms allowing *R. pomeroyi* DSS-3 to maintain constant BP even when Cu is scarce (**Figure 3**). In this regard, our data suggested that this strain is lowering its Cu use under Cu-limitation (**Figure 1**). On the other hand, strain PAlt-P26 did not change either its growth nor its Cu quota under varying Cu availability, which may explain why we did not observe any significant modifications to its C metabolism.

#### Comparison of Bacterial Cu Quotas From This Study With Other Bacteria and Eukaryotic Phytoplankton

The trace metal content of microorganisms reflects the trace metal availability in their growth environment (varying in time and space) and their physiological requirements (as reviewed by Twining and Baines, 2013). There is also some evidence that trace metal availability during microbial evolution played a role in shaping the metal composition of phytoplankton superfamilies (Quigg et al., 2011). Phytoplankton and bacteria originally evolved under drastically different Cu availabilities, but presently little is known about whether their evolutionary histories may be reflected in their present-day metal quotas. To explore this, we compared the Cu quotas of marine heterotrophic bacteria with published values for other prokaryotes (as discussed below), as well as eukaryotic phytoplankton (compared in **Figure 4**).

Copper quotas of marine heterotrophic bacteria agree well with those measured for the model heterotroph *E. coli*, despite differences in culturing conditions (1.8–5.3  $\mu$ mol Cu:mol C, Outten and O'halloran, 2001; 0.5–3.3  $\mu$ mol Cu:mol C, Cameron et al., 2012). The range of Cu quotas of our strains and *E. coli* (0.6–5  $\mu$ mol Cu:mol C) may, therefore, provide a good approximation of Cu requirements in aerobic heterotrophic prokaryotes.



Copper quotas of heterotrophic bacteria (data pooled from all Cu treatments) are also similar to those of eukaryotic phytoplankton cultured in EDTA-buffered growth Aquil media under various Cu conditions (not statistically significant, Wilcoxon test, p = 0.095; **Figure 4**). However, relative to marine heterotrophic bacteria Cu quotas of eukaryotic phytoplankton (both cultured in EDTA-buffered growth medium Aquil) are significantly more variable (Levene test, p = 0.026, **Figure 4**). This comparison should be considered preliminary as the heterotrophic bacteria dataset consists of considerably fewer measurements than the well-studied phytoplankton.

The larger range of Cu quotas in eukaryotic phytoplankton may reflect greater tolerance to a wider range of intracellular Cu or differences in Cu availability in culture media. To explore these factors, we plotted Cu quota data in **Figure 4** versus bioavailable Cu concentrations in culture media (inorganic Cu [Cu']) of both marine heterotrophic bacteria and eukaryotic phytoplankton (**Figure 5**). In general, heterotrophic bacteria maintain a well-controlled inventory of Cu across varying levels of Cu', and their average Cu content agrees particularly well with that of eukaryotic phytoplankton and cyanobacteria under low Cu availability ( $-\log [Cu'] < 13$ , Figure 5). However, as Cu availability increases there is a larger variation in Cu quotas in eukaryotic phytoplankton compared to quotas of heterotrophs. The larger variation in phytoplankton may reflect greater taxonomic variability in their Cu requirements and/or cell size, greater ability to store and/or detoxify Cu intracellularly. In eukaryotic phytoplankton, excess intracellular metals are sequestered by phytochelatins and metallothioneins (cysteine-rich peptides binding metals) and delivered to vacuoles, where these metals can be safely stored (Cobbett and Goldsbrough, 2002). In contrast, bacteria detoxify metals mainly by efflux (see reviews of Nies, 1999; Rensing and McDevitt, 2013), suggesting a tight intracellular trace metal control over a range of concentrations outside the cell. Furthermore, as opposed to eukaryotic phytoplankton and cyanobacteria that have Cu-dependent proteins in the cytosol (e.g., cytochrome c oxidase in the thylakoid plasma and membrane of cyanobacteria, Paumann et al., 2003), there are no known cytosolic Cu proteins in heterotrophic bacteria (Tottey et al., 2005).

It has been hypothesized that trace metal availability during the evolution of various organisms influenced metal selection for biological usage (Williams and Fraústo da Silva, 2003). Early prokaryotic cells (i.e., cyanobacteria) evolved in an anoxic ocean (Archean, ~2750 Myr) where Cu chemistry, as indicated by speciation models, was dominated by complexation with sulfide, rendering Cu less bioavailable (Saito et al., 2003). This contrasts with the evolutionary environment of eukaryotic phytoplankton (Paleozoic  $\sim$ 1000 Myr to modern ocean  $\sim$ 540 to 120 Myr), which was characterized by higher Cu availability as a result of Earth's oxygenation (Saito et al., 2003). These considerations have been used to explain the greater sensitivity of modern cyanobacteria to Cu toxicity (Brand et al., 1986; Saito et al., 2003). The narrow range of Cu quotas of heterotrophic bacteria relative to eukaryotic phytoplankton in our preliminary comparison can be linked to greater sensitivity of bacteria to Cu and their strict Cu homeostasis. This may be a result of the contrasting early evolutionary metal conditions of those groups.

#### Preliminary Assessment of the Role of Marine Heterotrophic Bacteria in the Cu Cycle

Bacteria dominate the living carbon biomass in the ocean (Whitman et al., 1998) and may equate or even exceed phytoplankton biomass in euphotic zones, particularly in oligotrophic regions (Cho and Azam, 1990; Ducklow, 1999). Hence, Cu in these microbes might represent a significant portion of Cu associated with the living biomass (biogenic Cu) in the upper ocean. To explore this, we calculated biogenic Cu associated with heterotrophic bacteria and phytoplankton (algae and cyanobacteria) biomass, using mean Cu:C ratios from culture studies of both groups (**Table 6**, using 1.36 and 2.4  $\mu$ mol Cu: for heterotrophic bacteria and phytoplankton, respectively, values represented by dashed lines in **Figure 5**). For these estimates,



we employed euphotic zone integrated heterotrophic bacterial and phytoplankton biomass measurements at station P26 in the NE Pacific in different seasons (Sherry et al., 1999, **Table 6**). As exemplified using this model system, heterotrophic bacteria may account for ~50% of the total biogenic Cu (associated with planktonic organisms), and the total biogenic Cu (5.6– 34 pM Cu, **Table 6**) might be a significant fraction (5.9–36%) of the particulate Cu pool in surface waters of the North Pacific (~94 pM, Nakatsuka et al., 2009). Similar findings have been reported for the North Atlantic by Twining et al. (2015)—the biogenic Cu pool contributing up to ~10% to the total particulate Cu pool. Furthermore, the impact of heterotrophic bacteria on the dissolved Cu pool can be estimated by calculating their biological Cu demand relative to the dissolved Cu supply. Using conservative estimates of heterotrophic bacterial growth rates  $(1-5 d^{-1}, Table 3)$  and an average biogenic heterotrophic bacteria

TABLE 6 | Calculations of biogenic Cu concentration (pmol L<sup>-1</sup>) for marine bacteria and phytoplankton (algae and cyanobacteria) in the seasonal euphotic zone (winter, spring, summer, Sherry et al., 1999) at station P26 in the subarctic NE Pacific.

		Heterotrop	hic bacteria	Phytop	lankton			
		(1.36 µmol Cu:mol C)		(2.4 μmol	Cu:mol C)			
Season	BB:PB	Biomass (μmol C L <sup>-1</sup> )	Biogenic Cu (pmol L <sup>-1</sup> )	Biomass (μmol C L <sup>-1</sup> )	Biogenic Cu (pmol L <sup>-1</sup> )	Total (pmol L <sup>−1</sup> )	Bacteria % of total	Phytoplankton % of total
Winter	0.08	1.1	1.5	14	23	34	4	96
	0.64	0.8	1.0	1.2	2.0	3.8	27	73
Spring	0.87	2.1	2.8	1.1	2.0	5.6	51	49
	1.82	1.8	2.4	2.0	3.5	7.2	33	67
Summer	1.29	2.3	3.1	1.7	3.0	7.2	42	58
	1.49	2.0	2.7	1.3	2.3	5.9	46	54

Estimates were performed using average Cu:C quotas of cultured phytoplankton from literature and the heterotrophic bacteria in this study (the dashed lines in **Figure 5**). Bacterial biomass to phytoplankton biomass ratio (BB:PB) and bacterial biomass were obtained from Sherry et al. (1999), with values for two different years (summer: 1995 and 1996, winter and spring: 1996 and 1997). Cu pool of 1 pM (**Table 6**), the annual heterotrophic bacterial Cu demand would range between 0.36 and 1.8 nM per year. This biogenic Cu demand could be significant relative to the supply rate of Cu mediated by upwelling at P26 (0.58–1.95 nM, Posacka et al., 2017). However, bacterially assimilated Cu is unlikely to be exported into the deep ocean as bacteria are not dense enough to sink on their own. Therefore, accumulation of Cu by heterotrophic bacteria is probably more important in retaining Cu in surface waters, where it could be subject to internal cycles (e.g., recycling through viral lysis) or enter the food web via protozoans that graze upon them.

#### DATA AVAILABILITY STATEMENT

Data and code for reproducing analyses can be found in publicly available repository on Github: https://github.com/aposacka/ Marine\_het\_bacteria-copper.

## **AUTHOR CONTRIBUTIONS**

AP, MM, and DS carried out the experimental design. AP conducted the sample collection and analysis and mainly wrote the manuscript. All authors contributed to data interpretation.

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## SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fmars. 2018.00523/full#supplementary-material

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**Conflict of Interest Statement:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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