

1	COD/sulfate ratio does not affect the methane yield and microbial diversity
2	in anaerobic digesters
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20	ABSTRACT
21	Anaerobic digestion of organic matter is the major route of biomethane production.
22	However, in the presence of sulfate, sulfate-reducing bacteria (SRB) typically
23	outcompete methanogens, which may reduce or even preclude methane production
24	from sulfate-containing wastewaters. Although sulfate-reduction and methanogenesis
25	can occur simultaneously, our limited understanding of the microbiology of anaerobic
26	digesters treating sulfate-containing wastewaters constrains improvements in the
27	production of methane from these systems. This study tested the effects of carbon

sources and chemical oxygen demand-to-sulfate ratio (COD/SO<sub>4</sub><sup>2-</sup>) on the diversity and
interactions of SRB and methanogens in an anaerobic digester treating a high-sulfate
waste stream. Overall, the data showed that sulfate removal and methane generation
occured in varying efficiencies and the carbon source had limited effect on the
methane yield. Importantly, the results demonstrated that methanogenic and SRB
diversities were only affected by the carbon source and not by the COD/SO<sub>4</sub><sup>2-</sup> ratio.

Keywords: Anaerobic digestion, methanogens, sulfate reducers, COD/sulfate ratio

37 1. INTRODUCTION

38 Anaerobic digestion (AD) has been successfully deployed for decades to treat high-39 strength industrial wastewaters and sewage sludge. Since methane, a renewable 40 energy source, is generated as the major end product, AD is considered the most 41 sustainable treatment process with a global primary energy potential of 99 EJ/year 42 projected for 2050 (Koornneef et al., 2013). However, the most recent estimates 43 indicate that currently only around 2.1 EJ/year is produced from the anaerobic 44 digestion of waste (WBA, 2014). Efficient AD process (from complex organic matter 45 degradation to biomethane generation) requires the concerted action of a well-46 balanced microbial consortium composed of hydrolysers, fermenters, syntrophic 47 microorganisms and methanogens. Despite numerous studies characterising these key players, many unidentified microorganisms and unresolved metabolic pathways are 48 49 regularly observed in AD reactors, hence the AD process is still considered a 'black-50 box' (Schmidt et al., 2016).

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While many high-strength industrial wastewaters can be treated efficiently via
anaerobic digestion, anaerobic treatment of sulfate-containing wastewaters, such as
from the brewery, pulp and paper, food processing, and tannery industries, generates

55 very little methane. In sulfate-containing wastewaters terminal oxidation occurs via both 56 sulfate reduction and methanogenesis. Sulfate-reducing bacteria (SRB) use sulfate as 57 their terminal electron acceptor and can outcompete methanogenic archaea for carbon 58 and electrons (O'Flaherty et al., 1998). SRB may also compete with syntrophic bacteria 59 (e.g. acetogens) for short-chain volatile fatty acids such as propionate and butyrate 60 (Qatibi et al., 1990), while hydrogen sulfide production by SRB can inhibit both 61 methanogens and SRB (O'Reilly and Colleran, 2006). In addition to the competitive 62 interaction between methanogenic archaea and SRB, co-existence of methanogenesis 63 and sulfate reduction has been demonstrated in different ecosystems with high sulfate 64 concentrations such as estuarine sediments (Oremland and Polcin, 1982) and 65 anaerobic digesters (Isa et al., 1986). In environments with low sulfate concentrations, 66 H<sub>2</sub>-utilising methanogens scavenge hydrogen produced during acidogenesis and 67 provide energetically favourable conditions for syntrophic SRB or acetogens (Parkin et 68 al., 1990; Muyzer and Stams, 2008; Bae et al., 2015). Moreover, the flexible 69 metabolism of many SRB increases their chance of survival in the absence of sulfate 70 (Plugge et al., 2011).

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72 The interaction between methanogens and SRB is governed by several factors such as 73 the type and oxidation state of organic carbon as well as the carbon-to-sulfate ratio 74 (Bhattacharya et al., 1996; Raskin et al., 1996; Hu et al., 2015; Lu et al., 2016). For 75 instance, it has been shown that SRB in natural sediments prefer simple organic 76 compounds such as ethanol and acetate over more complex organic compounds and 77 usually outcompete methanogens if sulfate is available (Oremland and Polcin, 1982; 78 Pol et al., 1998). However, anaerobic metabolism in high-rate engineered systems 79 such as anaerobic digesters may differ significantly from natural sediments. In anaerobic reactors treating sulfate-containing wastewaters, the carbon (measured as 80 chemical oxygen demand, COD) to sulfate ratio (COD/SO<sub>4</sub><sup>2-</sup>) has been found to be 81

82 critical in determining the fate of the carbon; this ratio is usually kept above the 83 theoretical value of 0.67 to ensure complete sulfate removal. However, results from previous research on the effect of  $COD/SO_4^{2-}$  are contradictory. For instance, methane 84 85 production from an upflow anaerobic sludge bed (UASB) reactor greatly deteriorated when the COD/SO<sub>4</sub><sup>2-</sup> ratio fell below 2 (Choi and Rim, 1991; Lu et al., 2016), whilst 86 87 other studies did not observe a significant effect of sulfate on methanogenesis (Hoeks 88 et al., 1984; Hu et al., 2015). The inconsistency between these observations may be 89 due to the differences in operational conditions such as wastewater characteristics and 90 reactor type used. Our knowledge of the diversity and metabolism of microorganisms in 91 AD reactors receiving sulfate-containing wastewaters is still very limited, which restricts 92 our understanding of these systems and hinders the development of strategies to 93 improve the methane production from AD reactors. In particular, sulfate may affect the 94 degradation pathway of carbon compounds present in the influent and of the associated volatile fatty acids. Therefore, the effect of the  $COD/SO_4^{2-}$  ratio on the 95 96 interactions between SRB and methanogens as well as on the degradation pathway of 97 carbon compounds needs to be addressed.

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In this study, we systematically evaluated the impact of three different  $COD/SO_4^{2-}$  ratios and four different carbon sources on the methane yield and on the microbial population dynamics in anaerobic sludge samples collected from a full-scale anaerobic digester treating a sulfate-containing waste stream. Results revealed how the carbon source and  $COD/SO_4^{2-}$  ratio affected the methane yield, the interactions between SRB and methanogens and the metabolic pathways in anaerobic digester samples under sulfidogenic conditions previously considered as unfavourable for methane generation.

#### 107 2. MATERIALS AND METHODS

108 **2.1. Sample collection** 

Anaerobic sludge samples were collected in July 2015 from three different sampling ports of a UASB reactor of an industrial treatment plant that receives coffee production wastewater (Jacobs Douwe Egberts Ltd, Banbury, UK), which contains sulfate. So, the anaerobic sludge is acclimatised to sulfate. Samples were transferred to the laboratory immediately and kept at 4°C until the experiments were set up.

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#### 2.2. Potential methane production test

116 A potential methane production (PMP) test was conducted to determine the optimum 117 concentrations and incubation times for four carbon sources (acetate, propionate, 118 butyrate and trimethylamine) to maximise methane production. Acetate, propionate and 119 butyrate were chosen as competitive, whilst trimethylamine (TMA) was chosen as a 120 non-competitive substrate for methanogens. Sludge samples from the three sampling 121 ports were mixed and washed twice in anaerobic medium with vitamin solution (DSMZ 122 318 and DSMZ 141, respectively; Braunschweig, Germany) to remove sulfate and 123 organic compounds from the samples. The washed sludge was centrifuged at 4000 g 124 for five minutes, the supernatant was decanted and the resulting pellet was 125 resuspended in equal volume of anaerobic medium as the removed supernatant. 126 Triplicate incubations were set up in 60 ml crimp-top serum bottles with 30 ml liquid 127 volume. Seed sludge with 1000 mg/l volatile suspended solids (VSS) was added to the 128 bottles. Acetate was tested at final concentrations of 10 to 60 mM, the other three 129 carbon sources were tested at 10 to 25 mM. The bottles were closed with butyl-rubber 130 stoppers and crimp-sealed with aluminium caps, flushed with oxygen-free nitrogen gas 131 for 10 min and then incubated at 35°C with shaking (150 rpm, Innova 4300, New 132 Brunswick Scientific Ltd., UK). Headspace gas pressure was measured daily using a 133 handheld digital manometer (Dwyer Series 475, Dwyer Instruments Ltd, UK) and the 134 incubations were ceased once gas production stopped.

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136 PMP test results showed that the highest methane productions were obtained when 137 the samples were incubated with 45 mM acetate, 20 mM propionate, 15 mM butyrate 138 and 15 mM trimethylamine (Supplementary Figure 1). Incubations with acetate, 139 propionate and butyrate reached the highest PMP on day seven whilst TMA 140 incubations took 12 days. Therefore, experiments were set up using these 141 concentrations and incubated for seven (acetate, propionate and butyrate) or 12 days 142 (TMA) to provide conditions for maximum methane production and avoid substrate 143 inhibition.

144

145 2.3. Experimental design

146 Batch experiments were used to assess the impact on methane yield of acetate,

propionate, butyrate and TMA at three different COD/SO<sub>4</sub><sup>2-</sup> ratios (0.5, 1.5 and 5) and to 147 148 analyse interactions between anaerobic microbial populations. No-sulfate incubations 149 were set up as controls. Five replicated microcosms were prepared for each substrate and  $COD/SO_4^{2-}$  combination using inoculum adjusted to 1000 mg/l VSS in 60 ml serum 150 151 bottles with 30 ml liquid volume. Guided by the PMP test, different carbon (15, 20 or 45 152 mM) and sulfate (1.5 – 66.7 mM) concentrations were provided to establish the selected COD/SO<sub>4</sub><sup>2-</sup> ratios (Supplementary Table 1). The microcosms were run for 153 154 seven (acetate, butyrate and propionate) or 12 days (TMA).

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#### 156 **2.4. Methane and volatile fatty acids analysis**

At the end of the incubations, gas samples were collected using a gas-tight syringe (Hamilton Company, Reno, USA) and the methane production was monitored by gas chromatography (Agilent 6890N, Agilent Technologies, Cheshire, UK) fitted with a flame ionisation detector and Porapak Q column. Nitrogen with 20 ml/min was used as the carrier gas. Three measurements were taken for each microcosm and the mean was calculated.

164	Slurry samples were also collected at the end of the incubations and centrifuged at
165	4000 g for five minutes. Supernatant was collected, filtered through a 0.20 $\mu m$
166	polyethersulfone membrane and analysed for volatile fatty acids (VFA) and sulfate
167	using an ion exchange chromatography (Dionex ICS3000; Dionex Corp., Sunnyvale,
168	CA, USA). Anion analysis was done using an lonpac AS 18 column (2 mm x 50 mm)
169	equipped with an lonpac AS 18 guard column, while cation analysis was done using an
170	lonpac CS12A column (4 mm x 250 mm) equipped with an lonpac CG12A guard
171	column. A gradient of 0-30 mM KOH and 20 mM methylsulfonic acid was used as
172	eluent for anion and cation analyses, respectively.
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174	2.5. Molecular methods
175	2.5.1. DNA extraction and PCR
176	Three replicates (out of five) that had less than 5% difference in methane generation
177	from each treatment and controls were chosen for molecular analysis. Slurry samples
178	were collected as above and total genomic DNA was extracted from 500 mg of
179	centrifuged slurry from each selected incubation using the hydroxyapatite spin-column
180	method (Purdy, 2005). Bacterial and archaeal 16S rRNA genes and functional genes
181	specific to methanogens (methyl coenzyme M reductase, mcrA) and SRB (dissimilatory
182	sulfate reductase, dsrB) were amplified by PCR (Supplementary Table 2). All PCR
183	amplifications were carried out using a Mastercycler Pro thermal cycler (Eppendorf UK
184	Ltd., Stevenage, UK) with MyTaq Red DNA Polymerase (Bioline Reagents Ltd.,
185	London, UK). Amplification conditions for the 16S rRNA and the mcrA genes were as
186	follows: initial denaturation at 95°C for 5 min, 35 cycles of 95°C for 1min, 55°C for 1
187	min, 72°C for 1.5 min, a final elongation step at 72°C for 5 min. For the <i>dsrB</i> gene, the
188	PCR conditions were the same except the annealing temperature, which was 52°C.
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190	2.5.2. High-throughput sequencing and data analysis

16S rRNA and functional gene PCR products were sequenced on the Illumina MiSeq
platform (300 bp paired-end, Illumina, Inc, San Diego, CA, USA) at the University of
Warwick (UK). Before sequencing, the PCR products were cleaned using Charge
Switch PCR Clean-up kit (Invitrogen, CA, USA), quantified by Qubit dsDNA BR Assay
Kit with Qubit 2.0 Fluorometer (Invitrogen, CA, USA), and prepared for sequencing as
described by Caporaso et al. (2012).

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198 We obtained 4.7, 3.3, 8.6 and 7.5 Gb raw sequences for the mcrA, dsrB, bacterial and 199 archaeal 16S rRNA genes, respectively. Raw sequences were quality-trimmed using 200 Trimmomatic (Bolger et al., 2014). Merging and operational taxonomic unit (OTU) 201 picking were carried out by USEARCH v8 (Edgar, 2010) at 97% and 85% similarity cut-202 off for the 16S rRNA and the functional gene sequences, respectively. Chimeras were 203 checked using ChimeraSlayer (Haas et al., 2011) and removed from downstream 204 analysis. Taxonomy assignments were determined against the Greengenes database 205 (DeSantis et al., 2006) for bacteria and archaea, and custom dsrB and mcrA databases 206 (Müller et al., 2015; Wilkins et al., 2015) using RDP Classifier 2.2 (Wang et al., 2007) 207 via QIIME software, version 1.6.0 (Caporaso et al., 2010). Average relative abundance 208 for each OTU in the samples was calculated using the relative OTU read abundances 209 of three replicates. Sequence datasets have been submitted to the National Center for 210 Biotechnology Information (NCBI) Read Archive under the bioproject accession 211 number of PRJNA434657.

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2.5.3. Quantitative PCR (qPCR)

In order to relate the methane generation to the relative abundance of methanogens, total *mcrA* gene copies in the incubation bottles were quantified using a qPCR assay with the *mcrA*-specific PCR primers (Supplementary Table 2). A standard curve was produced using serial 10-fold dilutions of a plasmid containing the *mcrA* gene. PCR reaction volumes were 10 ul, comprising 2 ul of 1:10 diluted gDNA, 0.35 ul of each

219 primer, 2.3 ul H<sub>2</sub>0 and 5 ul SsoAdvanced Universal SYBR Green Supermix (Bio-Rad 220 Laboratories Ltd., Hertfordshire, UK). Samples were run on a Bio-Rad CFX Connect 221 Real-Time Detection System (Bio-Rad Laboratories Ltd., Hertfordshire, UK). The 222 cycling conditions were as follows: 98 °C for 3 min, followed by 40 cycles of 98 °C for 223 15 s, 55 °C for 15 s, 72 °C for 1 min. To check for non-specific DNA products, a melt 224 curve was performed by heating the reaction mixture from 65 to 95°C with 0.5°C increments. The efficiency of the reactions was between 103%-109%, while the R<sup>2</sup> 225 226 value for the standard curve was 96%.

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228 2.6. Statistical analysis

229 One-way ANOVA with Post-hoc Dunnett's test was conducted to determine the 230 statistical significance of difference in biomethane production in the microcosms. 231 Species richness (Chao1) and alpha diversity (Shannon's index) were calculated using 232 OTU numbers and relative abundances. Principal Components Analysis (PCA) was 233 also applied to the relative abundance of OTUs to discriminate the samples with 234 respect to treatments. Following this, Spearman's correlation analysis was carried out 235 to identify the factors that may have affected the OTU abundances by correlating the first two principal components to the experimental variables including the methane 236 generation, sulfate removal efficiency,  $COD/SO_4^2$  ratio as well as the concentrations of 237 238 sulfate and the carbon compound removed. Graphpad Prism 7 software (Graphpad 239 Software, CA, USA) was used for correlation analysis and one-way ANOVA test, while 240 PAST (version 3) was used for diversity indices and PCA (Hammer et al., 2001). 241

#### 242 3. RESULTS AND DISCUSSION

243 3.1. Methane production and sulfate reduction efficiencies under different  $COD/SO_4^{2-}$  ratios 244

Methane, VFA and sulfate concentrations in the microcosms were measured at the end of the incubation and mass balances were calculated (Table 1). Results showed that three carbon sources (acetate, propionate and butyrate), and any VFAs produced as by-products, were consumed during the incubation period (data not shown). However,  $\sim$ 30% of the added TMA (123 to 137 µmoles) was not consumed in the incubation time.

251 The methane production and sulfate reduction for each substrate were assessed by 252 comparison to the no-sulfate control microcosms. Methane production was also 253 compared to the theoretical methane yields based on the amount of substrate utilised 254 (Bushwell and Mueller, 1952). Both acetate- and propionate-amended microcosms 255 produced methane in amounts close to their theoretical maximum (1350 µmoles and 1050 µmoles, respectively, Figure 1a, Table 1), while butyrate- and TMA-amended 256 microcosms produced no more than 60% of their theoretical maximum (1500 µmoles 257 258 and 704-734 µmoles, respectively, Figure 1a, Table 1).

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260 In acetate-amended microcosms, there was no significant difference between the 261 methane generation in controls and sulfate-amended microcosms. Similarly, in a 262 previous study, an anaerobic sludge sample, acclimated to sulfate-rich pulp and paper 263 wastewater, utilized 2000 mg/L acetate and produced 700 mL methane, which was 264 approximately the theoretical maximum (Ince et al., 2007). On the other hand, propionate and TMA had lower methane yields when COD/SO<sub>4</sub><sup>2-</sup> ratio was 0.5 and 1.5 265 compared to the controls, while butyrate-amended samples with all COD/SO<sub>4</sub><sup>2-</sup> ratios 266 267 had lower methane yields compared to the controls. It should be noted that hydrogen 268 sulfide (H<sub>2</sub>S) produced by the reduction of sulfate might have an inhibitory effect on 269 some methanogenic species, which might lower the methane generation (Isa et al., 270 1986). However, we used sulfate-acclimated anaerobic sludge to set up the 271 experiments, so the inhibitory effect of H<sub>2</sub>S would likely be reduced in our microcosms.

This may be the reason why we did not observe any significant drop in methane generation from acetate-amended microcosms with or without sulfate. Furthermore, there is experimental evidence that the kinetic and thermodynamic advatages of sulfate reducers over methanogens are erased by their sensitivity to sulfide toxicity, which may explain the methanogenic activity observed in our microcosms amended with sulfate (Maillacheruvu and Parkin, 1996).

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The effect of COD/SO<sub>4</sub><sup>2-</sup> ratio on methane generation and sulfate removal in the 279 280 microcosms was limited, and depended on the carbon source utilised. There was no significant effect of changing the COD/SO $_{4}^{2-}$  ratio on the methane production in the 281 282 acetate-, butyrate- and TMA-amended samples (Figure 1a, p>0.05). There was a 283 small, but significant (p<0.01) decrease in methane production in propionate-amended samples at a  $COD/SO_4^{2-}$  ratio of 1.5. However, even within these microcosms there was 284 285 no pattern of decreasing methane production with increasing sulfate. The consistent methane production with an increasing COD/SO $_{4}^{2-}$  ratio suggests that sulfate reduction 286 287 does not affect methanogenesis in either the acetate- or propionate-amended 288 microcosms. This is despite the fact that, in other systems, both of these substrates are 289 preferentially utilised by SRB if sulfate is freely available (Purdy et al., 2003a, 2003b) 290 and acetate-based sulfate reduction is more thermodynamically favourable than 291 acetoclastic methanogenesis (Schönheit et al., 1982).

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293 Methane production in the butyrate- and TMA-amended microcosms was between 294 44% - 82% of their theoretical maximum (Figure 1a, Table 1) in all treatments and 295 significantly lower than the no-sulfate controls in all samples. However, there was no 296 significant difference in methane production across the three COD/SO<sub>4</sub><sup>2-</sup> ratios for both 297 substrates, which suggests that the presence but not the concentration of sulfate 298 affected the methane production. The limited methane production with butyrate and

TMA indicates that non-methanogenic pathways for both butyrate and TMAdegradation occurred in these incubations.

301

Sulfate removal efficiency increased with increasing COD/SO<sub>4</sub><sup>2-</sup> ratio in all four 302 treatments (Figure 1b, Table 1) with only the acetate-amended microcosms not 303 304 reaching ~100% removal of sulfate (maximum of 70% removal). The effect of sulfate 305 addition on methane production in TMA-amended microcosms is remarkable, as this 306 compound is not known to be a competitive substrate for SRB. Hence, our results 307 disagree with those of Vich et al (2011), who amended methylamine and sulfate to 308 sludge samples from a full-scale UASB reactor and observed no significant effect of 309 sulfate addition on methane generation. In the propionate-amended samples at 0.5 and 1.5  $COD/SO_4^{2-}$  ratios, available sulfate had a small but statistically significant effect 310 311 on methane production (p<0.01; Figures 1a and 1b). Our results contradict two recent studies, where the effect  $COD/SO_4^{2-}$  ratio on methane generation was investigated. In a 312 study by Lu et al. (2016) on the effect of influent  $COD/SO_4^{2-}$  ratio on the biodegradation 313 314 of starch wastewater in a lab-scale UASB reactor, sulfate addition enhanced sulfidogenesis and subsequently methanogenesis. However, when the COD/SO<sub>4</sub><sup>2-</sup> ratio 315 316 was lower than 2, methanogenesis was supressed, possibly due to the competition and H<sub>2</sub>S inhibition (Lu et al., 2016). Similarly, Kiyuna et al. (2017) found that high sulfate 317 318 concentrations significantly reduced methane production from sugarcane vinasse, however these authors used higher  $COD/SO_4^{2-}$  ratios (7.5, 10 and 12) than we used in 319 320 our study. 321 322 While our results showed that methane production and sulfate reduction are

323 independent pathways for readily biodegradable substrates, in full-scale applications,

324 both COD removal efficiency and methane production in anaerobic treatment of

325 complex, sulfate-rich wastewaters may be lower. This may be due to the low

326 biodegradability of wastewater and the inhibitory effect of high sulfate/sulfur

327 concentration on microbial activity (Lens et al., 1998).

328

#### 329 **3.2.** Taxonomic and functional diversities in the microcosms

330 Between 1.7 and 3.8 million quality-filtered, chimera-free sequences were obtained for 331 bacterial 16S rRNA, archaeal 16S rRNA, dsrB and mcrA genes. These sequences 332 were assigned to 1295 and 543 distinct OTUs at 97% identity for bacterial and 333 archaeal 16S rRNA genes, whilst 288 and 61 distinct OTUs were obtained at 85% 334 identity for *dsrB* and *mcrA* genes, respectively. There was no significant difference 335 between the observed and predicted numbers of OTUs for each marker gene within 336 each treatment as estimated by Chao1 (Supplementary Table 3). The Shannon 337 diversity index did not vary significantly across samples (Supplementary Table 3).

338

We observed distinct shifts in the specific microbial populations in the microcosms over the experimental period, which allowed us to draw conclusions about the impact of carbon sources and the COD/SO<sub>4</sub><sup>2-</sup> ratio on the diversity and metabolic interactions of SRB and methanogens.

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#### 3.2.1. Methanogenic diversity and abundance

345 *Methanobacterium* spp, which use  $H_2$  and  $CO_2$  to produce methane (Boone, 2001), 346 dominated the methanogenic communities in all incubations (67-82% of the mcrA 347 sequences, Figure 2a). This finding was confirmed by archaeal 16S rRNA sequencing 348 (Figure 3a). The strong dominance of hydrogenotrophic methanogens even in the 349 presence of sulfate demonstrates that H<sub>2</sub>-consuming methanogens were not 350 outcompeted by H<sub>2</sub>-consuming SRB, which has been suggested to be a characteristic 351 of nutritious, high-rate systems such as anaerobic digesters (Ueki et al., 1992). The 352 consistently low percentage of Methanosaeta sequences (0.1-0.7%) in all the

microcosms indicates that acetoclastic methanogenesis was not a significant process
in this bioreactor (Demirel and Scherer, 2008). This is clearly shown in the fact that
even the addition of acetate did not enhance *Methanosaeta* (Figure 2a), suggesting
acetoclastic methanogenesis was not active at all in these slurries, despite the fact that
100% of the predicted methane was produced in the acetate-amended samples
(Figure 1a).

359

360 In the TMA-amended microcosms, the methanogenic community structure shifted. In 361 these incubations, the relative abundance of the obligate methylotrophic genus 362 Methanomethylovorans (Lomans et al., 1999) increased significantly to 20.1% (±1.8%, p=0.003) from 1% in the other incubations, irrespective of the COD/SO $_4^{2-}$  ratio (Figure 363 364 2a). Methylotrophic methanogens dominate TMA degradation in marine sediments 365 (King, 1984; Purdy et al., 2003a), so it is not unexpected that sulfate reduction and 366 methanogenesis were independent in TMA-amended microcosms and the relative 367 abundance of Methanomethylovorans was not affected by the presence or the 368 concentration of sulfate (Figure 2a). PCA analysis of the mcrA sequences also 369 supported this finding, as it separated the TMA incubations from the rest of the 370 samples (Figure 2b). The first principal components explained 82% and 84% of the 371 total variability in the mcrA (Figure 2b) and archaeal diversities (Figure 3b) in the 372 samples, respectively.

373

In addition to the sequence analysis, we have also quantified the *mcrA* genes to reveal the abundance of the methanogens in the samples. The *mcrA* gene numbers increased about ten-fold, from about  $1.1 \times 10^5$  to about  $1.6 \times 10^6$  across all microcosms (Supplementary Figure 2). The pattern was different for different substrates, though. The average abundance of the methanogens increased from  $2.3 \times 10^5$  to  $1.6 \times 10^6$  in the acetate-amended microcosms as the COD/SO<sub>4</sub><sup>2-</sup> ratio increased, however this increase

380 was not statistically significant. There was also no statistically significant difference in 381 the methanogen abundance in propionate-amended microcosms, in spite of an increase in methane production at the highest  $COD/SO_4^{2-}$  ratio. This suggests an 382 383 increase in the specific methanogenic activity in these microcosms. The lowest 384 methanogen abundance was observed in the butyrate-amended microcosms, which 385 was consistent with the methane production in these incubations, where the methane 386 yield was lower than the other microcosms (Figure 1). The number of methanogens did 387 not change significantly in the TMA-amended microcosms and they had a similar 388 number of methanogens to acetate and propionate incubations although the methane 389 yield was lower. This might be due to the lower efficiency of Methanomethylovorans 390 spp in utilising TMA compared to hydrogenotrophic methanogens dominating other 391 incubations.

392

393 Correlation analyses revealed that the *mcrA* and archaeal diversities did not 394 significantly correlate with the  $COD/SO_4^{2-}$  ratio in the microcosms, while the first 395 principal component of the *mcrA* analysis significantly correlated with only the methane 396 yield (p <0.01; Figure 2b and 3b; Table 2). Methanogen abundance did not correlate 397 significantly with the methane yield in the microcosms, however sulfate removed was 398 significantly related to the archaeal diversity (Table 2).

399

400 *3.2.2. SRB diversity* 

The SRB diversity, as determined by sequencing the *dsrB* gene, did not change markedly with the  $COD/SO_4^{2-}$  ratio in the microcosms (Figure 4a). This counterintuitive result could be explained by the metabolic flexibility of SRB, which allows some of them act as fermenters when sulfate is not available (Plugge et al., 2011). Some SRB can form syntrophic associations with H<sub>2</sub> scavengers such as hydrogenotrophic methanogens, utilising the H<sub>2</sub> produced by SRB (Bryant et al., 1967; Stams and

Plugge, 2009). Indeed, hydrogenotrophic methanogenesis was the dominant
methanogenic pathway in the microcosms, which might enable SRB survival in the
sulfate-free control incubations.

410

411 The relative mean read abundance of the Desulfarculus baarsii lineage increased from 412 ~6% to 14%-23.5% in the butyrate-amended microcosms (Figure 4a). Desulfarculus 413 baarsii can oxidise acetate and fatty acids completely to CO<sub>2</sub> using sulfate as an 414 electron acceptor (Sun et al., 2010). Although they have not been shown to grow 415 without sulfate in syntrophy with methanogens to date (Muyzer and Stams, 2008; 416 Plugge et al., 2011), they were found in the control incubations without added sulfate. 417 However, presence does not mean activity: these *D. baarsii* species may have been 418 present but inactive in the control incubations without sulfate. PCA analysis of the dsrB 419 sequence data revealed that the first component accounted for 97.9% of the total 420 variability, separating the butyrate incubations from the rest of the samples (Figure 4b). 421 Interestingly, there was no significant correlation between the mcrA and dsrB diversities, and with the COD/SO $_4^{2-}$  ratio (Table 2), which further indicates that methane 422 423 production and sulfate reduction were independent processes in these samples. 424 However, the dsrB diversity was found to be correlated with the concentration of sulfate 425 removed, sulfate removal efficiency, the initial carbon concentration and the methane 426 yield (Table 2).

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428 3.2.3. Bacterial Diversity

The most striking result from the bacterial sequence analysis was the dramatic

430 increase in the relative abundance of the genus *Syntrophomonas* in the butyrate

431 incubations to  $8.9\% \pm 1.02\%$  from  $1.1\% \pm 0.3\%$  in the other microcosms (p=0.003,

432 Figure 5a). As in the *mcrA* and *dsrB* diversities, this change was not dependent on the

433 COD/SO<sub>4</sub><sup>2-</sup> ratio (Figure 5a and 5b). *Syntrophomonas* species can degrade butyrate to

acetate and H<sub>2</sub> (Schmidt et al., 2013), and have been shown to form syntrophic
interactions with hydrogenotrophic *Methanobacterium* spp (Sousa et al., 2007). We
suggest that the members of this genus worked in syntrophy with *Methanobacterium*spp., which utilised H<sub>2</sub> to produce methane, particularly in the butyrate-amended
microcosms. Similar cooperation was observed in co-cultures of *Syntrophomonas wolfei* and *Methanospirillum hungatei*, which coupled butyrate degradation to acetate
and H<sub>2</sub> formation during growth on butyrate (Schmidt et al., 2013).

441

442 All the microcosms, including the controls, consistently contained Syntrophobacter in 443 relatively high abundances (3.6-7%). This is in line with a previous research, showing 444 that Syntrophobacterales are a stable and resilient functional group of bacteria in 445 anaerobic digestion systems (Werner et al., 2011). Syntrophobacter species can grow 446 on acetate, propionate and butyrate, either by sulfate reduction or, in the absence of 447 sulfate, by fermentation in syntrophy with methanogens and other H<sub>2</sub>/formate oxidisers 448 (Sobieraj and Boone, 2006; Müller et al., 2010, 2013). Their metabolic flexibility may 449 explain their high relative abundance across the samples regardless of the carbon compound or the  $COD/SO_4^{2-}$  ratio used (Boone and Bryant, 1980; Muyzer and Stams, 450 451 2008; Plugge et al., 2011).

452

Bacterial diversity significantly correlated only with methane production (p <0.05, Table 2), which may be due to the effect of the carbon sources on the bacterial populations, as clearly observed in the butyrate set. There was no significant correlation between bacterial diversity and  $COD/SO_4^{2-}$  ratio across the samples (Table 2).

457

#### 458 **3.3. Metabolic interactions between the microbial communities**

459 Sulfate reduction and methane generation were observed in varying efficiencies in the460 microcosms, whilst the relative abundances of specific functional groups such as

461 syntrophic organisms and hydrogenotrophic or methylotrophic methanogens (Methanobacterium spp and Methanomethylovorans spp.) did not vary considerably 462 within each set despite the change in the COD/SO<sub>4</sub><sup>2-</sup> ratio (Figure 2-5). This may be 463 464 explained by the flexible metabolism of SRB, which allows these populations to survive 465 when there is no available sulfate to respire as discussed above for D.baarsii. 466 Furthermore, syntrophic associations between methanogens and SRB may have 467 facilitated their growth together, as was previously shown in sulfate-amended anaerobic reactors, which had high sulfate-reduction efficiency even when 468 469 hydrogenotrophic methanogens were dominant (Yang et al., 2015).

470

We derived metabolic pathways for the metabolism of the carbon compounds used in this study based on the dominant microbial populations as obtained by the sequence analysis. In acetate amended microcosms, efficient methane generation was observed with and without sulfate and there was no marked change in microbial diversity under different COD/SO<sub>4</sub><sup>2-</sup> ratios. According to the sequence analysis, different metabolic pathways for the mineralization of acetate could be active simultaneously in these microcosms, independent of the COD/SO<sub>4</sub><sup>2-</sup> ratio (Figure 6a).

478

479 Desulfarculus baarsii species can convert acetate to CO<sub>2</sub>, which can be further used to 480 produce methane. Similarly, syntrophic acetate oxidation coupled to hydrogenotrophic 481 methanogenesis, which is thermodynamically and physiologically feasible at mesophilic 482 temperatures, may have occurred efficiently in these microcosms (Schnürer and 483 Nordberg, 2008; Dolfing, 2014). We propose that methane generation from propionate 484 was via similar pathways (Figure 6b), with propionate being converted to acetate first 485 as it is not utilised by methanogens directly. The dominance of the members of the 486 Desulfarculus baarsii lineage and the genus Syntrophobacter suggests complete 487 oxidation of propionate to  $H_2+CO_2$  via acetate. Although propionate degradation to

acetate is thermodynamically unfavourable under standard conditions ( $\Delta G^{\circ}$  = +76 488 489 kJ/mol), hydrogenotrophic methanogenesis in the microcosms could have lowered the 490  $H_2$  partial pressure, providing suitable conditions for propionate conversion to acetate. 491 Similar interactions were observed in paddy soils, where Syntrophobacter spp were 492 found to be the dominant propionate degraders. These organisms were suggested to 493 degrade propionate in synthrophy with hydrogenotrophic methanogens in the absence 494 of sulfate, however they switch to sulfate reduction when sulfate became available (Liu 495 and Conrad, 2017).

496

497 Metabolic pathways were different in butyrate and TMA-amended incubations as 498 inferred from the bacterial and methanogenic community structures in these 499 microcosms. Results suggest that the genus Syntrophomonas degraded butyrate to 500 acetate. Meanwhile, members of the Desulfarculus baarsii lineage may have 501 completely oxidised butyrate and produced CO<sub>2</sub> while reducing sulfate (Figure 6c). In 502 the sulfate-free control incubations, they may have worked in syntrophy with  $H_2$ 503 oxidisers. Additionally, Syntrophobacter spp. likely degraded butyrate to CO<sub>2</sub> and H<sub>2</sub>. 504 Metagenomic analysis of samples from lab-scale anaerobic digesters demonstrated 505 that Syntrophobacterales have the metabolic potential to degrade reduced carbon 506 compounds such as butyrate and propionate to acetate, CO2 and H2 (Vanwonterghem 507 et al., 2016). The highest relative abundance of Syntrophobacter spp (12% of dsrB sequences) was in the 0.5 COD/SO $_4^{2-}$  ratio microcosms compared to 4.8-8% in control 508 and higher COD/SO<sub>4</sub><sup>2-</sup> ratios. The high abundance of these complete-oxidisers conflicts 509 510 with findings of Muyzer and Stams (Muyzer and Stams, 2008), who suggested that 511 incomplete oxidisers of SRB would dominate over complete oxidisers when degrading 512 butyrate.

513

514 The increased relative abundance of the genus *Methanomethylovorans* in the TMA 515 microcosms indicates that part of the TMA was converted to methane directly via 516 methylotrophic methanogenesis (Figure 6d). Interestingly, sulfate removal was also 517 observed in these incubations although TMA has not been shown to be a growth 518 substrate for SRB previously. Interspecies H<sub>2</sub> transfer between *Methanomethylovorans* 519 spp. and the SRB may well have been the mechanism behind the sulfate reduction 520 observed. As demonstrated previously, when methylotrophic methanogens and 521 hydrogenotrophic SRB are in the same environment, the methanogens produce H<sub>2</sub>, 522 which serves as the electron donor for hydrogenotrophic SRB via interspecies H<sub>2</sub> 523 transfer (Phelps et al., 1985; Finke et al., 2007). On the other hand, Methanobacterium 524 spp (hydrogenotrophic methanogens) used  $H_2+CO_2$  to generate methane. Hence, 525 together with the hydrogenotrophic SRB, they would have maintained low H<sub>2</sub> 526 concentrations, thus facilitating the H<sub>2</sub> production by methylotrophic methanogens 527 (Meuer et al., 2002). Finke et al. (2007) have suggested that this H<sub>2</sub> loss mechanism 528 allows the methanogens to be active even when sulfate is available. Indeed, in our 529 experiments, the availability of sulfate did not affect the methanogenic diversity. 530 However, further experiments are required to confirm the metabolic interaction 531 between SRB and methylotrophic methanogens when degrading TMA. 532 533 The results of this study should be useful to develop strategies to increase the 534 methane yield from full-scale anaerobic digesters receiving sulfate-containing

wastewaters. For instance, a two-stage anaerobic treatment may be operated toincrease the acetate and propionate concentrations during the acidification step. Since

- 537 we have demonstrated that the  $COD/SO_4^{2-}$  ratio does not affect the methane production
- 538 when acetate and propionate are the carbon surces, a higher methane yield may be
- 539 obtained in the second reactor than when a one-reactor strategy is followed. Moreover,

540 the acidification reactor can be operated under alkaline conditions to increase the 541 propionate production when the influent is a protein-rich wastewater.

542

#### 543 **4. CONCLUSION**

544 Our results demonstrate that in a microbial community sourced from a sulfate 545 acclimated reactor, methane production and sulfate reduction were independent processes and that the COD/SO<sub>4</sub><sup>2-</sup> ratio did not affect the microbial community 546 547 structure, although the presence of sulfate can result in a shift in the metabolic pathway 548 to simultaneous methanogenesis and sulfate reduction. The main factor influencing the 549 microbial community structure, and hence the metabolic pathways, was the carbon 550 source. This indicates a more important role for the substrate in anaerobic reactors than merely the COD/SO<sup>2-</sup><sub>4</sub> ratio, which was previously suggested to be the key 551 552 parameter.

553

554

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# COD/sulfate ratio does not affect the methane yield and microbial diversity in anaerobic digesters

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Supplementary Figures and Tables



**Supplementary Figure 1.** Biogas production during the specific methanogenic activity tests. Maximum gas production occurred at 45 mM acetate, 20 mM propionate, 15 mM butyrate or trimethylamine.

## *mcrA* gene abundance



Supplementary Figure 2. Abundance of the *mcrA* gene in the incubations.

Supplementary Table 1. Experimental conditions used to set up the incubations.

Carbon source	Incubation	Substrate added (mM)	Sulfate added (mM)		
	Control	45	0		
Acatata	COD/SO4 = 0.5	45	59.7		
Acetate	COD/SO4 = 1.5	45	19.9		
	COD/SO4 = 5	45	5.9		
	Control	20	0		
Dronionato	COD/SO4 = 0.5	20	46.5		
Propionate	COD/SO4 = 1.5	20	15.5		
	COD/SO4 = 5	20	4.6		
	Control	15	0		
Butwata	COD/SO4 = 0.5	15	66.7		
Dulyrale	COD/SO4 = 1.5	15	22.2		
	COD/SO4 = 5	15	6.67		
	Control	15	0		
Trimethylamine	COD/SO4 = 0.5	15	15		
	COD/SO4 = 1.5	15	5		
	COD/SO4 = 5	15	1.5		

Supplementary Table 2. Primers used in this study.

Target gene	Primer name	Sequence (5'-3')	Reference			
165 rDNA Bastoria	515F	GTGCCAGCMGCCGCGGTAA	Concress at al 2011			
103 I KINA Dacleria	806R	GGACTACHVGGGTWTCTAAT	Caporaso et al., 2011			
165 rDNA Archaga	Parch519F	CAGCCGCCGCGGTAA	Øvreås et al., 1997			
105 IKNA Archaea	ARC915R	GTGCTCCCCGCCAATTCCT	Stahl and Amann, 1991			
danD	DSR1762Fmix	and Def	Pelikan et al., 2016			
<b>USTB</b>	DSR2107Rmix	see kei.				
100 O 11 A	mcrIRD -F	TWYGACCARATMTGGYT	Lover and Teaks 2015			
mcrA	mcrIRD -R	ACRTTCATBGCRTARTT	Lever and Teske, 2015			

**Supplementary Table 3.** Alpha diversity indices calculated for each sample. C: No-sulfate control. 0.5, 1,5 and 5 represent  $COD/SO_4^{2-}$  ratios.

		Acetate			Propionate				Butyrate				ТМА				
		С	0.5	1.5	5	С	0.5	1.5	5	С	0.5	1.5	5	С	0.5	1.5	5
	OTU number	1070	1083	1016	1043	985	978	978	1029	926	880	953	884	1024	959	948	932
Bacteria	Shannon's index	4.27	4.29	4.35	4.36	4.18	4.27	4.10	4.35	4.21	4.13	4.18	4.16	4.31	4.25	4.26	4.22
	Chao1	1362	1259	1186	1217	1221	1221	1173	1234	1149	1111	1209	1079	1213	1147	1121	1109
	OTU number	528	482	502	505	500	484	510	516	474	448	458	452	476	478	492	455
Archaea	Shannon's index	3.07	3.13	2.81	2.95	2.83	3.02	3.06	2.92	2.77	2.91	2.79	2.89	2.93	2.93	3.13	2.96
	Chao1	581	529	522.0	525	519	529	546	538	505	534	503	515	525	515	534	493
	OTU number	247	253	239	256	238	245	258	254	259	259	259	251	258	255	232	230
dsrB	Shannon's index	2.44	2.63	2.42	2.41	2.35	2.26	2.50	2.53	2.69	3.27	3.03	3.11	2.60	2.69	2.47	2.24
	Chao1	267	263	255	269	271	272	300	259	269	274	275	277	289	281	262	245
mcrA	OTU number	45	61	58	56	42	43	54	45	61	59	53	44	43	43	43	43
	Shannon's index	1.72	1.51	1.45	1.48	1.94	1.62	1.73	1.35	1.73	1.75	1.77	1.63	1.97	1.85	1.60	1.78
	Chao1	46	61	62	63	42	43	57	48	61	60	55	44	43	43	43	43

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### Table 1.

ſ	1		Substrate Methane yield						Sulfate			
	Treatment	Added	Residual	Consumed	Actual	Theoretical	ļ	Added	Consumed			
		1	μmol	!	μmol	μmol	%		μmol	%		
	Control	1350	0	1350	1184±117	1350	88	0	0			
Acetate	$COD/SO_4^{2-} = 0.5$	1350	0	1350	1237±73	1350	92	1792	358	20±6		
45 mM	$COD/SO_4^{2-} = 1.5$	1350	0	1350	1276±169	1350	95	597	377	66±16		
	$COD/SO_4^{2-} = 5$	1350	0	1350	1307±78	1350	97	179	126	70±9		
	Control	600	0	600	1164±99	1050	111	0	0			
Propionate	$COD/SO_4^{2-} = 0.5$	600	0	600	961±118	1050	92	1397	662	47±12		
20 mM	$COD/SO_4^{2-} = 1.5$	600	0	600	869±74	1050	83	466	357	77±7		
]	$COD/SO_4^{2-} = 5$	600	0	600	1214±167	1050	116	140	138	99±1		
	Control	600	0	600	770±117	1500	51	0	0			
Butyrate	$COD/SO_4^{2-} = 0.5$	600	0	600	677±98	1500	45	2002	1694	84±2		
15 mM	$COD/SO_4^{2-} = 1.5$	600	0	600	683±76	1500	46	667	665	99±0.2		
]	$COD/SO_4^{2-} = 5$	600	0	600	656±90	1500	44	200	199	99±0.2		
	Control	450	123	327	602±83	734*	83	0	0			
TMA	$COD/SO_4^{2-} = 0.5$	450	137	313	466±43	704*	66	451	112	25±4		
15 mM	$COD/SO_4^{2-} = 1.5$	450	130	220	473±58	720*	66	150	113	74±4		
	$COD/SO_4^{2-} = 5$	450	127	323	527±79	727*	73	45	44	98±1		

\*TMA theoretical methane yield for 450  $\mu$ mols of substrate was 1012.5  $\mu$ mols, actual theoretical is based on total TMA consumed (70-73% of the calculated yield).



a





а



а





b



