Comment on "A spherical cavity model for quadrupolar dielectrics" [J. Chem. Phys. 144, 114502 (2016)]

Iglika M. Dimitrova, Radomir I. Slavchov, Tzanko Ivanov, and Sebastian Mosbach

Citation: The Journal of Chemical Physics **146**, 187101 (2017); doi: 10.1063/1.4979717 View online: http://dx.doi.org/10.1063/1.4979717 View Table of Contents: http://aip.scitation.org/toc/jcp/146/18 Published by the American Institute of Physics





Comment on "A spherical cavity model for quadrupolar dielectrics" [J. Chem. Phys. 144, 114502 (2016)]

Iglika M. Dimitrova,¹ Radomir I. Slavchov,^{1,2,a)} Tzanko Ivanov,³ and Sebastian Mosbach² ¹Department of Physical Chemistry, Faculty of Chemistry and Pharmacy, Sofia University, 1164 Sofia, Bulgaria ²Department of Chemical Engineering and Biotechnology, Cambridge University, CB2 3RA Cambridge, United Kingdom

³Department of Theoretical Physics, Faculty of Physics, Sofia University, 1164 Sofia, Bulgaria

(Received 6 February 2017; accepted 23 March 2017; published online 11 May 2017)

[http://dx.doi.org/10.1063/1.4979717]

Recently, a spherical cavity (Onsager) model for quadrupolar dielectrics has been proposed and used to calculate the macroscopic quadrupolarizabilities of several common solvents.¹ In this Comment, one of the boundary conditions used in Ref. 1 (for continuity of the displacement field) is revised to account for the surface displacement field that occurs at non-homogeneously polarized interfaces. We show that the revised condition leads to more reasonable results.

The displacement field D in a quadrupolar medium involves the divergence of the quadrupolarization tensor Q,

$$\boldsymbol{D} = \boldsymbol{\varepsilon} \boldsymbol{E} - \frac{1}{2} \nabla \cdot \mathbf{Q}, \tag{1}$$

where ε is the dielectric permittivity and E is the electric field intensity. In order to solve Coulomb's equation ($\nabla \cdot D = 0$) for problems involving an interface between two quadrupolar media, a boundary condition is required for D. In Ref. 1, we used the condition for continuity of the normal displacement field (D_r) at the surface of the spherical cavity, Eq.¹ (8) (we cite the equations from Ref. 1 with superscript 1),

$$D_r(r = R_{cav} - 0) = D_r(r = R_{cav} + 0),$$
(2)

where R_{cav} is the radius of the cavity. This condition is valid only in case that **D** has no δ -term in its singular expansion at the surface of discontinuity (i.e., if **D** can be expressed as $D^+\eta^+$ $-D^-\eta^-$, where $\eta^+ = \eta(r - R_{cav})$ and $\eta^- = \eta(R_{cav} - r)$, and D^+ and D^- are continuous functions of the coordinates in the vicinity of $r = R_{cav}$; η and δ are the Heaviside and Dirac functions). However, the quadrupolarization tensor is discontinuous at the interface between the cavity and the medium, i.e., $\mathbf{Q} = \mathbf{Q}^+\eta^+ + \mathbf{Q}^-\eta^-$; from Eq. (1), it then follows that **D** does have a δ -term,

$$\boldsymbol{D} = \boldsymbol{D}^{+}\boldsymbol{\eta}^{+} + \boldsymbol{D}^{-}\boldsymbol{\eta}^{-} + \boldsymbol{D}^{\mathrm{S}}\boldsymbol{\delta}, \qquad (3)$$

where the bulk displacements D^{\pm} stand for

$$\boldsymbol{D}^{\pm} = \boldsymbol{\varepsilon}^{\pm} \boldsymbol{E}^{\pm} - \frac{1}{2} \nabla \cdot \boldsymbol{Q}^{\pm}, \qquad (4)$$

and the surface displacement field D^{S} is given by

$$\boldsymbol{D}^{\mathrm{S}} = -\frac{1}{2} \mathbf{e}_{r} \cdot \left(\mathbf{Q}^{+} - \mathbf{Q}^{-} \right).$$
 (5)

0021-9606/2017/146(18)/187101/5/\$30.00

Here, $\delta \equiv \delta(r - R_{cav})$, \mathbf{e}_r is the unit vector normal to the surface, and we used the relations $\nabla \eta^{\pm} = \pm \mathbf{e}_r \delta$. The correct boundary condition in the case of non-zero \mathbf{D}^{S} reads^{2–4}

$$D_r(r = R_{cav} - 0) = D_r(r = R_{cav} + 0) + \nabla^{S} \cdot \boldsymbol{D}^{S} \Big|_{r = R_{cav}}, \qquad (6)$$

where ∇^{S} is the surface nabla operator ($\nabla^{S} = \nabla - \mathbf{e}_{r}\partial/\partial r$). The derivations of this equation in Refs. 3 and 4 were for a flat surface, and \mathbf{D}^{S} in Ref. 4 is produced by intrinsic surface polarization. Nevertheless, Eq. (6) is valid² also for the case we investigate—spherical surface and \mathbf{D}^{S} induced by the discontinuity of the bulk quadrupolarization, Eq. (5) (cf. the supplementary material). The term $\nabla^{S} \cdot \mathbf{D}^{S}$ is zero for the problems of quadrupolar electrostatics we solved previously,^{5–7} since the involved surfaces were homogeneous. However, this is not the case with the four basic problems required by the Onsager model of quadrupolar dielectrics solved in Ref. 1. The revised expressions follow.

Reaction fields of a dipole and a quadrupole. With regard to the problem for a dipole p in the centre of a cavity, the solution for the potential ϕ_{p0} inside the cavity ($r < R_{cav}$) remains formally the same (Eq.¹ (13)), but the one outside the cavity is, instead of Eq.¹ (14),

$$\phi_p = \frac{\boldsymbol{p}_{\text{ext}} \cdot \boldsymbol{r}}{4\pi\varepsilon r^3} \left(1 - \frac{9}{g_p} \frac{L_Q^2}{R_{\text{cav}}^2} \frac{r + L_Q}{R_{\text{cav}}} e^{-\frac{r - R_{\text{cav}}}{L_Q}} \right). \tag{7}$$

Here L_Q is the quadrupolar length. The *external dipole moment* p_{ext} in Eq. (7) is proportional to p,

$$\boldsymbol{p}_{\text{ext}} = Y_E \boldsymbol{p},\tag{8}$$

where Y_E is given by Eq.¹ (28), $Y_E = 3f_E \varepsilon / (2\varepsilon + f_p \varepsilon_0)$. The quadrupolar factors f_p , g_p , and f_E in the formulae above stand for the expressions

$$f_p = \frac{2+8x}{2g_p + 9x^2 + 9x^3}, \quad f_E = \frac{2g_p}{2g_p + 9x^2 + 9x^3},$$
$$g_p = 1 + 4x + 9x^2 + 9x^3, \tag{9}$$

where $x = L_Q/R_{cav}$; the formula Eq.¹ (17) for f_p is invalid. The expression Eq.¹ (15) for the reaction field E_{react} remains formally unchanged but with the new factor f_p . The calculated limit in Eq.¹ (18) holds true with the new formulae presented here; note that it is actually incorrect with the old ones due

^{a)}E-mail: ris26@cam.ac.uk

187101-2 Dimitrova et al.

to an arithmetic mistake. This means that the revised Onsager model following from the new boundary condition (6) leads to the correct continuum single particle limit of the perturbation theory of Milischuk and Matyushov,⁸ while the old one¹ stemming from Eq. (2) does not.

Similar corrections are required in the formulae¹ (20)– (24) for the reaction field of a quadrupole **q** in a cavity. Instead by Eq.¹ (21), the potential ϕ_q at $r > R_{cav}$ is given by

$$\phi_{q} = \frac{3 \mathbf{r} \cdot \mathbf{q}_{\text{ext}} \cdot \mathbf{r}}{8\pi\varepsilon r^{5}} \left(1 - \frac{18}{g_{q}} \frac{L_{Q}^{2}}{R_{\text{cav}}^{2}} \frac{r^{2} + 3L_{Q}r + 3L_{Q}^{2}}{R_{\text{cav}}^{2}} e^{-\frac{r - R_{\text{cav}}}{L_{Q}}} \right).$$
(10)

The *external quadrupole moment* \mathbf{q}_{ext} is still proportional to \mathbf{q} , but instead of Eq.¹ (23), the relation is

$$\mathbf{q}_{\text{ext}} = Y_{\nabla E} \mathbf{q},\tag{11}$$

where $Y_{\nabla E}$ is given by Eq.¹ (33), i.e., $Y_{\nabla E} = 5f_{\nabla E}\varepsilon/(3\varepsilon + 2f_q\varepsilon_0)$. The quadrupolar factors $f_q, f_{\nabla E}$, and g_q above stand for

$$f_q = \frac{1 + 6x + 6x^2}{g_q + 12x^2 + 18x^3 + 18x^4},$$

$$f_{\nabla E} = \frac{g_q}{g_q + 12x^2 + 18x^3 + 18x^4},$$

$$g_q = 1 + 6x + 24x^2 + 54x^3 + 54x^4.$$
(12)

Eq.¹ (23) for the reaction field gradient is valid but with the new f_q -factor above.

Cavity field and cavity field gradient. The solution to the problem for a cavity in a quadrupolar continuum polarized by an external field E_0 , with the new boundary condition (6), also changes compared to Eqs.¹ (26)–(30). The potential ϕ_E at $r > R_{cav}$ is

$$\phi_E = -\boldsymbol{E} \cdot \boldsymbol{r} + \frac{\boldsymbol{p}_{\text{ext},E} \cdot \boldsymbol{r}}{4\pi\varepsilon r^3} \left(1 - \frac{9}{g_p} \frac{L_Q^2}{R_{\text{cav}}^2} \frac{\boldsymbol{r} + L_Q}{R_{\text{cav}}} e^{-\frac{\boldsymbol{r} - R_{\text{cav}}}{L_Q}} \right).$$
(13)

The cavity field E_{cav} is proportional to the external field E_0 , formally with the same Y_E coefficient, Eq.¹ (28), but with the new quadrupolar factors (9). The vector $p_{ext,E}$ in Eq. (13) is the excess dipole moment of the cavity induced by the external field, and instead by Eq.¹ (29), it is given by

$$\boldsymbol{p}_{\text{ext},E} = -\frac{4\pi}{3} R_{\text{cav}}^3 Y_E \left(\varepsilon - \varepsilon_0\right) \boldsymbol{E}_0. \tag{14}$$

Finally, let us summarize the solution to the problem for a cavity in a quadrupolar medium placed in an external electric field gradient $(\nabla E)_0$. The cavity field gradient $(\nabla E)_{cav}$ is proportional to $(\nabla E)_0$ with the $Y_{\nabla E}$ coefficient from Eq.¹ (33). The potential outside the cavity is

$$\phi_{\nabla E} = -\frac{1}{2} \mathbf{r} \cdot (\nabla E)_0 \cdot \mathbf{r} + \frac{3\mathbf{r} \cdot \mathbf{q}_{\text{ext},\nabla E} \cdot \mathbf{r}}{8\pi\varepsilon r^5} \times \left(1 - \frac{18}{G_q} \frac{L_Q^2}{R_{\text{cav}}^2} \frac{r^2 + 3L_Q r + 3L_Q^2}{R_{\text{cav}}^2} e^{-\frac{r-R_{\text{cav}}}{L_Q}}\right).$$
(15)

The excess quadrupole moment $\mathbf{q}_{\text{ext},\nabla_E}$ of the sphere induced by the external gradient $(\nabla E)_0$ is given by the equation

$$\mathbf{q}_{\text{ext},\nabla E} = -\frac{8\pi}{15} R_{\text{cav}}^5 Y_{\nabla E} A_{\nabla E} \nabla E_0.$$
(16)

The quadrupolar factors in these formulae are given by Eq. (12) and the expressions

$$g_{\nabla E} = 1 + 9x + 39x^{2} + 90x^{3} + 90x^{4},$$

$$A_{\nabla E} = \left(1 + 9x^{2}\frac{g_{\nabla E}}{2g_{q}}\right)\varepsilon - \left(1 - 3x^{2}\frac{1 + 3x + 3x^{2}}{g_{q}}\right)\varepsilon_{0},$$

$$G_{q} = \frac{12g_{q}A_{\nabla E}}{15(1 + 6x^{2})\varepsilon - 10\varepsilon_{0}}.$$
(17)

Reciprocal relationships. Onsager⁹ observed that the following reciprocal relation holds for the cavity field and the external dipole in a dipolar medium: $E_{cav}/E_0 = p_{ext}/p = 3\varepsilon/(2\varepsilon + \varepsilon_0)$ (his Eq. (11)). An important outcome from the revised boundary condition (6) is that such a relation is valid also in a quadrupolar medium

$$\frac{p_{\text{ext}}}{p} = \frac{E_{\text{cav}}}{E_0} = Y_E.$$
(18)

Analogous reciprocal relationship occurs for the external quadrupole and the cavity field gradient

$$\frac{\boldsymbol{q}_{\text{ext}}}{\boldsymbol{q}} = \frac{(\nabla \boldsymbol{E})_{\text{cav}}}{(\nabla \boldsymbol{E})_0} = Y_{\nabla \boldsymbol{E}}.$$
(19)

The old boundary condition (2) leads¹ to results that disagree with Eqs. (18) and (19).

The macroscopic polarizability and quadrupolarizability. The macroscopic polarizability and quadrupolarizability of the medium are obtained in the manner described in Ref. 1. The result for the polarizability α_P reads

$$\alpha_P \equiv \varepsilon - \varepsilon_0 = \frac{Y_E}{1 - \alpha_p X_p} \left(\alpha_p + \frac{1}{1 - \alpha_p X_p} \frac{p_0^2}{3k_{\rm B}T} \right) C.$$
(20)

This is the sought generalization of Onsager's⁹ formula for ε to a quadrupolar medium. The respective equation for the macroscopic quadrupolarizability α_O of an Onsager fluid is

$$\alpha_Q \equiv 3\varepsilon L_Q^2 = \frac{Y_{\nabla E}}{1 - \alpha_q X_q} \left(\alpha_q + \frac{1}{1 - \alpha_q X_q} \frac{\mathbf{q}_0 : \mathbf{q}_0}{10k_{\rm B}T} \right) C.$$
(21)

Here, the factors X_p and X_q are given by Eqs.¹ (15) and (22), and Y_E and $Y_{\nabla E}$ are given by Eqs.¹ (28) and (33)—but with the new expressions for f_p , f_q , f_E , and $f_{\nabla E}$, Eqs. (9) and (12). Eqs. (20) and (21) are setting the correct Onsager model for a quadrupolar liquid, instead of Eqs.¹ (50) and (53). The change is significant enough to affect most of the results presented in Ref. 1. Qualitatively, the results are similar, with one exception: according to Eqs.¹ (51) and (52), the quadrupoles have a stronger effect on ε in the case where $p_0 = 0$, i.e., non-polar liquids; according to Eq. (20), the relative effect is not so different for polar and non-polar fluids. Quantitatively, in most cases, the new term $\nabla^{\rm S} \cdot D^{\rm S}$ in Eq. (6) leads to a significant change of the final values of α_Q (up to 20%) and cannot be neglected. Correspondingly, Table¹ I and most figures in Ref. 1 are inaccurate—the revised ones follow.

Results. As in the previous work,¹ we used the following empirical relationship between the cavity radius and the density of the fluid:

$$\frac{m}{\frac{4}{3}\pi R_{\rm cav}^3} = k_{\rho}\rho + k_0,$$
(22)

TABLE I. Values of the molecular multipole moments and polarizabilities, and the coefficients in dependence (22) of R_{cav} on ρ .

	Dipole moment $p_0 (C \cdot m) \times 10^{30}$	Polarizability $\alpha_p/4\pi\varepsilon_0$ (Å ³)	Quadrupole moment $(\mathbf{q}_0:\mathbf{q}_0)^{1/2} (\mathbf{C}\cdot\mathbf{m}^2) \times 10^{40}$	Quadrupolarizability $lpha_q/4\pi arepsilon_0 ({ m \AA}^5)$	k_0 (kg/m ³) in Eq. (22)	$k_{ ho}$ in Eq. (22)	$dev_{\varepsilon}/\varepsilon_0$ of Eq. (22) vs. the data for ε
Ar	0	1.639	0	0.454	655.5	0.3134	0.0009
Kr	0	2.488	0	0.913			
Xe	0	4.105	0	1.936			
CH_4	0	2.597	0	1.681	122.84	0.7019	0.0004
N ₂	0	1.737	4.08	1.12	342.2	0.5445	0.0005
CO_2	0	2.98	11.43	2.21	235.64	0.9102	0.0009
CS_2	0	8.215	8.88	11.40	721.2	0.5513	0.0017
C ₆ H ₆	0	10.25	24.87	18.42	695.2	0.3300	0.0019
H ₂ O	6.204	1.470	8.073	0.496	Eq. (23)		0.19
CH ₃ OH	5.638	3.32	16.436	3.121	1991.3	0.3534	0.7

where *m* is the molecular mass. The values of the coefficients k_{ρ} and k_0 we determined previously in Table¹ I are based on the old defective variant of the Onsager model. The correct k_{ρ} and k_0 are given in Table I here; they were determined by regression over the experimental data for ε vs. the theoretical permittivity following from our new Eqs. (20)–(22) (solved for the unknown R_{cav} , L_Q , and ε). In all cases but water, the analysis of the data for ε suggested that R_{cav} is a function of ρ only. For water, as in Ref. 1, statistically significant temperature dependence was evident, and therefore, instead of Eq. (22) we used

$$\frac{m}{\frac{4}{3}\pi R_{\rm cav}^3} = k_\rho \rho - k_T T + k_0.$$
(23)

For H₂O, we compared the theoretical permittivities that follow from Eqs. (20)–(23) with the data points for ε ; the dispersion was minimized with respect to the parameters of Eq. (23), leading to best values $k_0 = 2853.7 \text{ kg/m}^3$, $k_\rho = 0.1195$, and $k_T = 1.057 \text{ kg/m}^3$ K (slightly different from those determined in Ref. 1). The details for the experimental data and the values of the molecular parameters used in this Comment are described in Ref. 1.

Figs.¹ 5–9 in Ref. 1 are inaccurate, due to the defective Eqs.¹ (50) and (53). Figs. 1–5 here show the correct results. For non-polar and non-quadrupolar fluids (Ar, Kr, Xe, CH₄), the change is very small, so the respective Figs.¹ 3 and 4 do not require a correction. For quadrupolar but non-polar molecules (N₂, CO₂, CS₂, C₆H₆), the values of L_Q calculated here at

the highest experimental densities are about 10% smaller than those in Ref. 1 (corresponding to an error in α_Q of about 20%). In Fig. 3, the correct quadrupolar lengths of benzene calculated via Eqs. (20) and (21) are compared with those from Ref. 1, to highlight the importance of the term $\nabla^{\rm S} \cdot D^{\rm S}$ in Eq. (6). For polar species (H₂O, CH₃OH), the new values of L_Q are at most 5% smaller than those in Ref. 1 (corresponding to α_Q by 10% smaller). The change in the values of $R_{\rm cav}$ compared to those in Ref. 1 is relatively small (e.g., 1%-2%) yet significant in view of the strong dependence of all properties of the quadrupolar liquid on $R_{\rm cav}$.

The value of the dipole moment p of a water molecule in liquid environment is larger than the intrinsic dipole p_0 of a molecule in the gas phase by a factor of $1/(1 - \alpha_p X_p)$ = 2.17 (which differs from the value 2.07 calculated previously¹ due to the different f_p factor in the expression for X_p). The quadrupole moment of water increases by a factor of \mathbf{q}/\mathbf{q}_0 = $1/(1 - \alpha_q X_q) = 1.39$ (instead of ¹ 1.36). This suggests that one of the approximations of the Onsager model—for the negligible molecular hyperpolarizabilities—fails even worse for the revised model of Onsager.

An important feature of the quadrupolar Onsager model is that it leads to a much smaller field intensity E near the cavity compared to the original model: in the range $r = R_{cav}...3R_{cav}$, using the results for the parameters of water at room temperature, one can obtain that the maximal value of E_r varies in the range $3 \times 10^8...3 \times 10^7$ V/m (2 orders of magnitude smaller than the range obtained previously,¹ due to another arithmetic



FIG. 1. (a) Cavity radius of N₂ calculated from Eqs. (20) and (21) and experimental data for ε and ρ . The colour indicates the temperature. The gray symbols are Böttcher's R_{cav} following from the original theory of Onsager (Eq. (20) with $L_Q = 0$). The black line is Eq. (22) with $k_\rho = 0.545$ and $k_0 = 342$ kg/m³. (b) The quadrupolar length L_Q following from the same model and data. See Ref. 1 for details.



FIG. 2. (a) Cavity radius of CO₂ and CS₂ as a function of their concentration *C*, calculated with Eqs. (20) and (21) from experimental data for ε and ρ . The colour indicates the temperature. The gray symbols are Böttcher's *R*_{cav} following from Onsager's original equation (Eq. (20) with $L_Q = 0$). Gray line is Eq. (22) with $k_\rho = 1$ and $k_0 = 0$; black line is Eq. (22) with $k_\rho = 0.910$ and k_0 = 236 kg/m³ for CO₂. (b) The quadrupolar length L_Q following from the same model. See Ref. 1 for details.

FIG. 3. (a) Cavity radius of benzene as a function of ρ , calculated with Eqs. (20) and (21) and experimental data for ε and ρ . The red circles correspond to various temperatures (300-340 K) and pressure 1 atm, and the black ones-various pressures (1-1600 atm) and temperature 323 K. The gray symbols are Böttcher's Rcav following from Onsager's equation (Eq. (20) with $L_Q = 0$). The black line is Eq. (22) with $k_{\rho} = 0.330$ and k_0 $=695 \text{ kg/m}^3$. (b) The quadrupolar length Lo following from the same model (circles). Gray dots: α_0 of an ideal gas. Crosses: L_Q calculated¹ with the wrong boundary condition 1. See Ref. 1 for details.

error). This field is still high and dielectric saturation can still be expected, but not as massive as estimated in Ref. 1. The dipole moment per each water molecule in the first coordination shell that follows from the Onsager cavity model can be estimated to be of the order of $0.5 \times p_0$. For comparison, the original model⁹ of Onsager (with $L_Q = 0$) predicts significantly higher field (up to 8×10^8 V/m at $r = R_{cav}$).

The comparison of the values of the macroscopic quadrupolarizabilities determined by Jeon and Kim¹¹ from Stokes shift data of coumarin in CO₂ and C₆H₆ ($\alpha_Q = 0.42 \times 10^{-30}$ Fm at $\rho = 800$ kg/m³ for CO₂ and 2.6 × 10⁻³⁰ Fm for C₆H₆) with those following from the revised Onsager model

under the same conditions ($\alpha_Q = 0.46 \times 10^{-30}$ Fm for CO₂ and 1.9×10^{-30} Fm for C₆H₆) shows good agreement for CO₂ and acceptable for C₆H₆. The coincidence might be accidental since the Stokes shift formula of Jeon and Kim is based on a set of boundary conditions different from ours.¹ The calculated quadrupole length of methanol varies in the range L_Q = 0.79-0.82 Å, Fig. 5, in satisfactory agreement with the value obtained from the data for the activity coefficient of NaBr in methanol solution,⁶ $L_Q = 1.1 \pm 0.2$ Å. The difference is most probably due to the neglected hyperpolarizabilities: according to the revised Onsager model, the dipole moment **p** of methanol is higher than **p**₀ by a factor of 2.39 (compared to the



FIG. 4. (a) Cavity radius of H₂O calculated with Eqs. (20) and (21) and experimental data for ε and ρ . The colour indicates the temperature. The lines are Eq. (23) at the indicated temperatures. (b) The quadrupolar length L_Q following from the same model. See Ref. 1 for details.



FIG. 5. (a) Cavity radius of methanol calculated from Eqs. (20) and (21) and experimental data for static permittivity and density. The colour indicates the temperature. The black line is Eq. (22) with the coefficients from Table I. The gray symbols are Böttcher's radii (obtained from Eq. (20) with $L_Q = 0$). (b) The quadrupolar length L_Q following from the same model. See Ref. 1 for details.

previous¹ value 2.16), which is too high, and the reaction field is of magnitude 2×10^{10} V/m, large enough for the hyperpolarizabilities to play a role. Another problem with methanol is that its molecule is not spherical, cf. the discussion in Ref. 1.

Clausius-Mossotti-Debye fluid. We would like to use this document to answer a question that we have been asked on several occasions regarding the quadrupolarizability of a liquid, namely, what would a Clausius-Mossotti-Debye-like model give for the properties of a quadrupolar fluid? This model¹⁰ assumes that the local field $E_{\rm loc}$ acting on a molecule in the cavity in the liquid is the sum of the external field E_0 plus the field $E_P = P/3\varepsilon_0$ of a homogeneously polarized medium of polarization P outside the cavity (in contrast, in Onsager's model, the polarization is a function of the coordinates). A similar approach can be applied to a cavity in a homogeneously quadrupolarized medium of constant **Q**. However, it is easy to show that the potential inside such a cavity is constant, and therefore, the respective electric field gradient $(\nabla E)_O$ produced by the medium is zero. Consequently, the local electric field gradient is equal to the outer one, i.e., $(\nabla E)_{loc} = (\nabla E)_0$ (compared to the relation Eq.¹

(43) in an Onsager fluid). Therefore, the quadrupolarizability of a Clausius-Mossotti-Debye liquid is precisely equal to the quadrupolarizability of an ideal gas, Eq.¹ (1).

See supplementary material for (A) some of the derivations and (B) a sample Maple code for solving Eqs. (20) and (21).

- ¹I. M. Dimitrova, R. I. Slavchov, T. Ivanov, and S. Mosbach, J. Chem. Phys. **144**, 114502 (2016).
- ²I. M. Dimitrova, "Cavitation model of quadrupolar liquids and electric properties of surfaces with dipole moment," Ph.D. thesis, Sofia University, 2017
- ³E. B. Graham and R. E. Raab, Proc. R. Soc. A **456**, 1193 (2000).
- ⁴R. Slavchov, T. Ivanov, and B. Radoev, J. Phys.: Condens. Matter 19, 226005 (2007).
- ⁵R. I. Slavchov and T. I. Ivanov, J. Chem. Phys. 140, 074503 (2014).
- ⁶R. I. Slavchov, J. Chem. Phys. **140**, 164510 (2014).
- ⁷R. I. Slavchov, I. M. Dimitrova, and T. I. Ivanov, J. Chem. Phys. **143**, 154707 (2015).
- ⁸A. A. Milischuk and D. V. Matyushov, J. Chem. Phys. **123**, 044501 (2005).
- ⁹L. Onsager, J. Am. Chem. Soc. 58, 1486 (1936).
- ¹⁰P. Debye, Phys. Z. **13**, 97 (1912).
- ¹¹J. Jeon and H. J. Kim, J. Chem. Phys. **119**, 8626 (2003).