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Processing and characterization of UHMWPE composite fibers with alumina particles in poly(ethylene-vinyl acetate) matrix

Jelena Zec¹, Nataša Tomić^{2*}, Milorad Zrilić¹, Smilja Marković³, Dušica Stojanović¹ and Radmila Jančić-Heinemann¹

¹University of Belgrade, Faculty of Technology and Metallurgy, Belgrade, Serbia

²University of Belgrade, Innovation Center of Faculty of Technology and Metallurgy, Belgrade, Serbia

³Institute of Technical Sciences of SASA, Belgrade, Serbia

***Corresponding author:** Nataša Tomić, University of Belgrade, Innovation Center of Faculty of Technology and Metallurgy, Belgrade, Serbia.
Email: ntomic@tmf.bg.ac.rs

Abstract

Processing of hybrid composites represents a challenge for engineers where the aim is to establish compatibility among several materials. The aim of this study is to evaluate the effects of different size and morphology of alumina fillers on the mechanical and thermal properties of the composite fibers based on ultra-high molecular weight polyethylene fibers (UHMWPE). These fibers have an outstanding elastic modulus and they are compatible with nonpolar sequences of the poly(ethylene-co-vinyl acetate) (EVA) matrix. Compared to the fibers, inferior mechanical properties of the matrix can be improved using alumina particles. Commercial Al₂O₃ nanoparticles, commercial whiskers and synthesized particles of Al₂O₃ doped with iron oxide, incorporated in different weight percent were used as fillers. The UHMWPE fibers were impregnated using the solution of EVA in toluene with dispersed particles. FT-IR and FE-SEM were

used for structural examination. Tensile testing revealed increasing of modulus of elasticity and strengths of obtained hybrid composite fibers. Thermal gravimetry showed improved thermal stability up to 350°C of the hybrid composite fibers with alumina particles doped with iron oxide. Results of tested samples showed that the best mechanical properties were for hybrid composite fibers with 1 wt.% of iron doped alumina filler.

Keywords

UHMWPE, Al₂O₃ reinforcement, EVA matrix, hybrid composite fibers, thermal stability

Introduction

Fiber-reinforced polymer composites are widely used due to their excellent properties, such as light weights, high tensile strength and moduli.¹ Poly(ethylene-co-vinyl acetate) (EVA) is a random copolymer synthesized by copolymerization of ethylene and vinyl acetate monomers with different acetate contents.² Due to their properties and processing characteristics, these copolymers have extensive applications in biomedical, food industry, construction, transport, wire and cable industry. They are commonly used as adhesives, electrical insulation, corrosion protection, waterproofing and packaging.²⁻

⁴As adhesives, EVA copolymers are often used in combination with different types of materials like wood, rubber, metals, glass and polymers.^{5,6} EVA properties depend on the content of each comonomer. Crystallization and melting characteristics of EVA are very sensitive to the VA content.⁷ The melting temperature and degree of crystallinity decreases with an increase of vinyl acetate groups, but the glass transition temperature (T_g) increases with the increase of vinyl acetate content.⁵⁻⁸ Presence of ethylene segments in EVA molecules provides characteristic to „plasticized” and increases the activity.⁹

The mechanical properties of the EVA as matrix are inferior compared to the fibers like ultra-high molecular weight polyethylene (UHMWPE) and they can be improved using inorganic fillers having a different morphology. The main reason to incorporate inorganic particles into polymer matrices is to obtain a composite with improved

mechanical properties like tensile strength, hardness, Young's modulus or stiffness applying different reinforcement mechanisms.¹⁰

A material that is widely used in composites is UHMWPE with long chains that provide sufficient load transfer between the macromolecules.¹¹ Due to its wear resistant characteristics, chemical inertness, strength, high impact resistance, toughness, tensile strength and excellent modulus, it has become one of the most modified polymers.¹² The most significant characteristics include high specific fiber strength that is about fifteen times stronger than steel. These advancements have led to the development of high performance polymer matrix composites.¹³ The combination of high tensile strength, stiffness, low density and damage tolerance made UHMWPE fibers an ideal material. Ropes and cables made of this material have various applications for kites, fishing lines, offshore mooring etc.¹¹ The extreme tensile strength and the possibility to apply thick cables allow overdesign of the cable up to load carrying capacity of a few tons.

UHMWPE fibers have lack polarity and, therefore, it is not easy to produce composites due to the poor interfacial bonding with the matrix. On the other side, lateral interactions between molecules are the result of Van der Waals forces. Their presence in composite materials helps transferring off-axis loads retaining good mechanical properties along the fiber axis.¹⁴ Although UHMWPE fibers have numerous useful characteristics and provide a load-bearing effect, their weakness is poor thermal resistance. In hybrid composite fibers the function of polymer is to transfer load to the

reinforcement and protect it from chemical and high temperature influences.¹⁵For this purpose, the most commonly used thermosetting resins are epoxy, but brittle, low impact resistance and low toughness, limits their usage. Weaknesses of epoxy resins can be overcome with the incorporation of the reinforcement phase such as UHMWPE fibers.¹⁶

In order to investigate the behavior of materials with different fillers, considerable studies have been conducted. That includes zirconium particles, TiO₂, kaolin, Al₂O₃, carbon fibers, carbon nanotubes (CNT), etc..¹²Combination of organic and inorganic components and obtaining new materials attract increasing attention, since they combine the properties of including constituents.¹⁷

Bo Jin *et. al* have studied the composites where EVA is used as the polymer matrix and alumina nanoparticles as fillers. There was no agglomeration for 1.5wt.% of Al₂O₃ nanoparticles. The influence of the Al₂O₃ content on the tensile strength of EVA/Al₂O₃ nanocomposites showed increasing values up to 1.5wt.% of Al₂O₃ and at a higher content, the tensile strength starts to decrease, probably due to the agglomeration of nanoparticles.¹⁸Baskaran et al.¹⁹ studied the effect of alumina nanoparticles in the mechanical properties of the nanocomposites. Chee et al.²⁰ noticed that the higher amount of alumina nanoparticles would reduce the reinforcing effect and mechanical properties of the composite due to poor dispersion and agglomeration.

In this work, different alumina fillers have been chosen due to their specific properties including hardness, wear resistance, high elastic modulus, thermal stability, high adsorption capacity, good thermal conductivity, high strength and stiffness, and good resistance to chemical and thermal environments.

Aluminum oxide has several crystal forms, and corundum is one of the most desired form for the reinforcement of composite materials when mechanical properties are important. Considering that α - Al_2O_3 is the hardest form of alumina, it was a good choice to be used as reinforcement in order to improve the mechanical and thermal properties of polymer matrix composites. Like the other metal oxides, when Al_2O_3 is exposed to the atmosphere, its surface becomes covered with water in the form of the terminal OH groups.²¹ Fine alumina particles exert influence on mechanical properties of the composite, even if the quantity is rather small.²² Opelt et al. reported that after the addition of 0.15wt.%, 0.50wt.% and 1.50wt.% of alumina, the modulus of elasticity had increased respectively. For tensile strength the results showed that there is no significant variation between the neat matrix and nanocomposite with small amount of Al_2O_3 fillers.²³ According to Wetzel et al. the similar behavior is observed for nanocomposites with up to 1% volume fraction (approximately 3.5 wt.%) of alumina fillers. On the other hand flexural modulus and strength depends on the filler volume fraction.²⁴

Particle sizes at micro scale have a low influence on the Young's modulus. However, at nanoscale, Young's modulus of the composite increases as the size of particles

decreases. It was also reported that tensile strength decreases in composites with 3 vol.% nanoparticles, due to poor dispersion. Conversely, at 1 vol.% loading of microparticles, the tensile strength increases.²⁵In the case of the Al₂O₃/epoxy nanocomposites, the formation of an interconnected network formed by the alumina nanoparticles was given as the main reason for the elastic modulus enhancements.²⁶However, the elastic modulus in nanocomposites does not increase continuously with increasing the weight percentage of fillers. After the optimum weight percentage is reached, due to stress concentration, the elastic modulus will decrease.²⁷

The aim of the present work is to research the properties of UHMWPE hybrid composite in poly (ethylene-co-vinyl acetate) matrix modified by commercial nanoparticles and whiskers of Al₂O₃ as fillers, and comparing the properties of composite in case of using new synthesized particles of alumina doped with iron oxide. Based on the results, the application of the hybrid composite material with the best characteristics will also be examined.

Experimental

Materials

The polyethylene used in this study was UHMWPE – DSM Dyneema, Netherlands (SK75, filament diameter 20 μm, density 0.970 g/cm³). Commercial EVA (Elvax[®] 410) produced by DuPont[™], containing 18 wt.% of vinyl acetate was chosen as a polymer

matrix. Commercial toluene (produced by Zorka Šabac) was used as a solvent. The aluminum oxide nanoparticles (< 50 nm particle size) and alumina whiskers (diameter 2-4 nm × 400 nm) were commercially supplied from Sigma Aldrich. Aluminum chloride hydroxide (Locron L) was purchased from the Clariant Company in the crystallized state of $\text{Al}_2\text{Cl}(\text{OH})_5 \cdot 2.5 \cdot \text{H}_2\text{O}$. This product contained 23.5% Al_2O_3 , 8.18% Cl, corresponding to an Al/Cl, molar ratio of 2/1. Iron chloride (FeCl_3) was also purchased from the same company.

Methods

Alumina particles doped with iron oxide were synthesized via sol-gel technique by the hydrolysis of aluminium ion. Solution of aluminum chloride hydroxide and iron chloride in water was mixed in a weight ratio of 19.5 g $\text{Al}_2(\text{OH})_5\text{Cl}$, 0.5 g FeCl_3 and 15 g H_2O . Obtained gel was calcined for 2 hours on 900°C .²⁸The mixture was grinded in a mortar and then sifted through a sieve. Nanoparticles, whiskers and synthesized particles of Al_2O_3 doped with iron oxide were annotated as n- Al_2O_3 , w- Al_2O_3 and Fe- Al_2O_3 . Obtained hybrid composite fibers with iron doped particles, alumina nanoparticles and alumina whiskers in UHMWPE fibers hybrid composite in EVA matrix were annotated as 1Fe- Al_2O_3 /EVA/UHMWPE, 3Fe- Al_2O_3 /EVA/UHMWPE, 5Fe- Al_2O_3 /EVA/UHMWPE, 1n- Al_2O_3 /EVA/UHMWPE, 3n- Al_2O_3 /EVA/UHMWPE, 5n- Al_2O_3 /EVA/UHMWPE, 1w- Al_2O_3 /EVA/UHMWPE, 3w- Al_2O_3 /EVA/UHMWPE,

5w-Al₂O₃/EVA/UHMWPE for the addition of 1 wt.%, 3 wt.% and 5 wt.% of the particles, respectively.

To obtain solutions with 1, 3 and 5 wt.% of fillers, n-Al₂O₃, w-Al₂O₃ and Fe-Al₂O₃ were added in toluene. Uniform dispersion of particles was achieved after one hour in ultrasound bath. Afterwards, 20 wt.% of EVA was added into prepared solution and mixed on 60°C on the magnetic stirrer with a rotation speed of 500 rpm for one hour. Prepared mixtures were used for UHMWPE fiber impregnation using apparatus showed in Figure 1.²⁹The nozzle was changed and its dimensions were 0.4 × 1.5 mm, adapted to the dimensions of the fibers that are impregnated. To avoid composite sticking on a drum, impregnated fibers were cured with hot air on 60°C before winding.

[insert Figure 1.]

Figure1. Schematic presentation of UHMWPE impregnation process.

Characterization

The distribution of the filler particles was determined using an optical microscope and the laser particle size analyzer (PSA) Mastersizer 2000.

Morphology and shape of the particles were observed using a Mira3 Tescan field emission scanning electron microscope (FE-SEM), operated at 20 kV. The samples were previously coated with a thin gold film. The particle size and the dispersion of synthesized fillers were obtained by image analysis using the Image ProPlus 6.0 software. Images from FE-SEM were further used for measuring diameters, particle size

distribution and particle shape characterization. The laser particle size analyzer (PSA) Mastersizer 2000 was used to measure the particle size distribution, which covers the particle size range of 0.02–2000 μm .

The chemical structure of the samples was analyzed using a Fourier transform infrared spectroscopy (FT-IR) at Nicolet 6700 spectrometer (Thermo Scientific) in the attenuated total reflectance (ATR) mode with a single bounce 45 °F Golden Gate ATR accessory with a diamond crystal, and an electronically cooled DTGS detector. The spectra were the co-addition of 64 scans at 4 cm^{-1} spectral resolution, and ATR corrected. The spectrometer was equipped with OMNIC software and recorded the spectra in the wavelength range from 2.5 μm to 20 μm . Infrared spectra were recorded in transmission mode between 400 and 4.000 cm^{-1} . Fourier transform, as a mathematical process was required to convert the raw data into the actual spectrum.

Thermal analyses of samples were conducted by using a differential scanning calorimetry (DSC) and thermogravimetric analysis (TGA). The DSC measurements were conducted using a TA Instrument, TA DSC Q10 calibrated with indium standards. The temperature range was from 30 °C to 180 °C at a heating rate of 10 °C/min with a dynamic nitrogen flow of 50 ml/min. The samples were heated up to 180 °C, kept at 180 °C for 5 min to erase thermal history, and cooled down to 30 °C at the same cooling rate. A second heating was done on the each sample. The thermogravimetric analysis

was performed on a TA Instruments TGA/STD Q600 under a nitrogen gas flow (100 ml/min) between 30 °C and 600°C with a heating rate of 10 °C/min.

The tensile strength of the samples was examined using a servo hydraulic, mechanical testing machine INSTRON 1332 with control electronics FASTtrack 8800. Strain rate was 5 mm/min. All samples were of the same length and three replicates of each sample were used in the testing procedure.

Results and Discussion

The displayed measurement and results were performed in order to determine which filler contributes the most to the mechanical properties of the hybrid composite EVA/UHWMPE.

The distribution of particles

Analyzing images from the FE-SEM and processing by Image ProPlus software, the distribution of Fe-Al₂O₃ particle dimension showed submicron range of diameters achieved by sol-gel technique. More than 850 objects were analyzed in every sample of particles using the automatic object recognition algorithm. The average diameter of 0.15 μm (Table 1) was determined as most of the objects had very small dimensions. A few agglomerates have been noticed (Figure2a) as the result of lower dispersion of the particles and their joining during the calcination process, but they were recognized as individual objects by the algorithm. The presence of particle aggregates was reflected by high values of maximal observed particle diameters in Table 1. The predominance of

a great number of particles with low diameters (obvious on Figure2b) led to small mean diameter of particles, as can be seen on Figure2c. The range of measured objects is even larger than presented in Figure2c where it was between 0 and 5 μm , but the very large objects were agglomerates that are dissipated on smaller particles in the processing of the composite so the most important information is the mean value of observed individual particles.

Table 1. Statistical data of particle size distribution obtained by image analysis of Fe-Al₂O₃.

| Sample | Area | Max diameter, μm | Min diameter, μm | Mean diameter, μm | Roundness |
|-----------------------------------|--------------------|-----------------------------|-----------------------------|------------------------------|-----------|
| Fe-Al ₂ O ₃ | Min | 0.0032 | 0.0565 | 0.0565 | 1.000 |
| | Max | 542.134 | 55.068 | 12.942 | 176.455 |
| | Mean | 0.966 | 0.239 | 0.150 | 1.542 |
| | Standard deviation | 201.418 | 23.139 | 0.530 | 1.093 |

[insert Figure 2.]

Figure2.(a) FE-SEM micrograph of Fe-Al₂O₃, (b) Magnified part of Fe-Al₂O₃, FE-SEM micrograph, (c) Statistical distribution of Fe-Al₂O₃, – diameter, and (d) Particle size distribution of Fe-Al₂O₃.

Particle size distribution indicated narrow distribution of Fe-Al₂O₃ in the interval between 0.3 and 1.3 μm , with a maximum peak at 0.48 μm (Figure2d). The obtained values for particle size were in accordance with the range of distribution of mean diameter (Figure2c) suggesting that the most of agglomerates dissipated on single submicron particles.

ATR-FT-IR analysis

Chemical structure and interactions were examined using FT-IR. Spectra of the n-Al₂O₃, w-Al₂O₃ and Fe-Al₂O₃ are given in Figure 3.

[insert Figure 3.]

Figure 3. FT-IR spectra for fillers and hybrid composite fibers: (a) EVA/UHMWPE, (b) n-Al₂O₃, w-Al₂O₃, Fe-Al₂O₃, and (c) 1n-Al₂O₃/EVA/UHMWPE, 1w-Al₂O₃/EVA/UHMWPE, 1Fe-Al₂O₃/EVA/UHMWPE hybrid composite fibers.

FT-IR identifies chemical bonds in a molecule of organic and inorganic samples by producing an infrared absorption spectrum. The resulting spectrum represents the molecular absorption, and creates a molecular fingerprint of the sample.

The scope of interest was acetate (1195 cm⁻¹), carbonyl (1731 cm⁻¹) and ethylene sequence C-H (2940 cm⁻¹) absorption region of the IR spectrum for EVA (Figure 3a). The UHMWPE fibers showed characteristic absorption of ethylene segments (1470 cm⁻¹, 2950/2920 cm⁻¹, and 720 cm⁻¹).

Spectra for n-Al₂O₃, w-Al₂O₃ and Fe-Al₂O₃ fillers (Figure 3b) showed changes in the intensities of absorption bands at 3300 cm⁻¹ that was attributed to the presence of the O-H groups. The peak intensity of O-H groups for n-Al₂O₃ and Fe-Al₂O₃ indicated that these groups were present in a greater content on the surface than w-Al₂O₃. The function of these groups was to establish a better bonding with the composite.

The FT-IR spectra for 1n-Al₂O₃/EVA/UHMWPE, 1w-Al₂O₃/EVA/UHMWPE and 1Fe-Al₂O₃/EVA/UHMWPE (Figure 3c) indicated the characteristic peaks at 1222 cm⁻¹, 1687

cm^{-1} , and 2740cm^{-1} that referred to presence of C–O–C, C=O and C–H groups, respectively. Also, no vibrations of the O–H groups were noted. This indicated that these groups were used for establishing of hydrogen bonds between particles and EVA/UHMWPE hybrid composite fibers.

FE-SEM analysis

In this study, EVA/UHMWPE was compounded with the same proportions (1 wt.%) of different particles. Based on the FE-SEM images the surface properties of the fillers, matrix and UHMWPE fiber, as reinforcement, were analyzed. Dimensions of the samples (Figure4) were: length (l) $30 (\pm 0.1)$ mm; width (d) $1.5 (\pm 0.05)$ mm; thickness (m) $0.43(\pm 0.03)$ mm. The mass proportion of UHMWPE fibers, EVA and particles in the matrix were 48.09 wt.%, 51.91 wt.% and 0.52 wt.%, respectively.

In order to obtain an insight into the impregnation quality, the cross-section of the sample was made by a sharp medical blade. To examine the morphology and the type of failure, the fractured surfaces of the samples after the tensile test were sputtered with gold and observed under SEM.

[insert Figure 4.]

Figure4. Dimensions of the samples.

The FE-SEM pattern of a fractured surface of EVA/UHMWPE composite is shown in Figure5.

[insert Figure 5.]

Figure5. The FE-SEM micrograph of unmodified EVA/UHMWPE composite:(a) Cross section before tensile test, (b) Surface of composite breakage.

Cross-sectional view in Figure5a showed that the fiber distribution in matrix was uniform, as well the wetting of UHMWPE fibers, which means that these fibers were a proper choice for EVA reinforcement due to established adhesion between these components. FE-SEM micrograph in Figure5b showed thinned UHMWPE fibers that acted as a main load bearing element. Good adhesion of EVA matrix to the fibers enabled simultaneous elongation of both components as a response to the applied load, which was followed by ductile fracture of the EVA/UHMWPE composite. Fractured surfaces of EVA/UHMWPE composite with n-Al₂O₃, w-Al₂O₃ and Fe-Al₂O₃ particles are shown in Figure6.

[insert Figure 6.]

Figure6. The FE-SEM micrograph of EVA/UHMWPE composite with: (a) 1n-Al₂O₃, (b) 1w-Al₂O₃, and (c) 1Fe-Al₂O₃.

SEM image after the tensile test of EVA/UHMWPE hybrid composite fibers with Al₂O₃ nanoparticles (Figure 6a) showed a brittle fracture. The transverse fracture surface of the matrix indicated the combined effect of mechanical interlocking between fibers and matrix with 1n-Al₂O₃. Better fiber-matrix interfacial bonding and propagation resistance across the cross sections of samples resulted in higher strengths in hybrid composite

fibers. This image was in accordance with the tensile stress results due to their higher resistance compared to the unmodified composite and w-Al₂O₃/EVA/UHMWPE.

Poor interfacial adhesion between the UHMWPE fibers and matrix is shown in Figure6. The whiskers didn't sufficiently strengthen the matrix and the fibers were pulled out. Agglomeration and segregation of whiskers were also detected. According to a separate pulled out fiber, it could be concluded that the connection between fiber and matrix was weak. The separation at the fiber-matrix interface indicated that the longitudinal cracks were nucleated at this interface.

The FE-SEM micrograph of fractured surfaces of Fe-Al₂O₃/EVA/UHMWPE showed longitudinal cracking of EVA with improved interfacial adhesion and mechanical properties of the hybrid composite fibers that caused cohesive failure of the matrix (Figure6c). It could be seen that the EVA covered the surface of UHMWPE fibers that led to higher interactions between the surfaces of the composite components.

DSC analysis

The difference in melting temperature is one of the essential indicators for the required demands of studied composites. DSC as a calorimetry method measures heat flow as a function of temperature. Heat flow can indicate structural changes driven by thermal processes. At the same time it is a good indicator of thermal active transitions such as melting temperature and glass transition temperature. Differential scanning calorimeter (DSC) thermograms of samples are shown in Figure7. From the represented

thermogram it was noticed only one melting peak for neat EVA and UHMWPE fibers at 78.4 °C and 137.2 °C, respectively.

For UHMWPE fibers, a slight difference in melting temperature (T_m) between obtained and the one reported by the producer (144-152 °C) was shown due to the environmental and storage conditions like humidity, increased storage temperature, DSC set parameters, etc³⁰. EVA copolymer is a partially crystalline polymer that consists of crystalline ethylene segments and also a non-crystalline (amorphous) vinyl acetate segments.⁷ An endothermic peak was typical for crystal melting of ethylene sequences in EVA copolymer³¹, which was low ($\Delta H_{m1} = 8.708$ J/g), Table 2. Unlike EVA and UHMWPE fibers, composites exhibited two clearly separated endothermic peaks corresponding to the melting phase transition of EVA and UHMWPE components. The endothermic effects were quantified by determining the total enthalpy change from the peak area for the different samples (Table 2). A significant change for neat EVA was observed at around 78 °C. According to A. Badie et al.³², it can be associated with internal structural changes or crystallization transitions of the matrix.

The addition of fillers reduced melting enthalpy of UHMWPE fibers. Comparing to melting temperature of neat EVA, fillers slightly increased the melting temperature of the matrix (2.2%). When the melting point of EVA/UHMWPE and hybrid composites were compared, it can be concluded that no change in melting point was observed by inducing particles. The main change in EVA phase melting transition was noticed by

melting enthalpy. Sharp decrease in melting enthalpy (ΔH_{m1}) for 52.9-65.5% suggested large-scale structural changes due to disturbance of polyethylene segment packing by the introduction of reinforcement as UHMWPE and particles. Maintained value for T_{m1} enabled preserved mechanical properties of EVA matrix concerning the elongation of the composites and the amount of absorbed load.

However, a change of the melting temperature of UHMWPE phase didn't follow specific order and the values for hybrid composite were similar. Melting enthalpy of UHMWPE phase for hybrid composites showed significant differences. The highest melting enthalpy of UHMWPE phase (ΔH_{m2}) for hybrid composite 1Fe- Al_2O_3 /EVA/UHMWPE showed the structural arrangement of reinforcement. Negligible UHMWPE melting point decrement (2.2%) in composites may be prescribed to structural changes due to disturbance of surface polyethylene sequences by the processing of hybrid composites and thus the use value of composites wasn't affected by the use of presented technology.

[insert Figure 7.]

Figure7. DSC thermograms of EVA, UHMWPE fibers and composite fibers.

Table 2. Temperatures and enthalpies obtained from DSC thermograms.

| Samples | T_{m1} (°C) (EVA phase) | ΔH_{m1} (J/g) (EVA phase) | T_{m2} (°C) (UHMWPE phase) | ΔH_{m2} (J/g) (UHMWPE phase) |
|--|------------------------------|--------------------------------------|---------------------------------|---|
| UHMWPE | | | 137.2 | 78.2 |
| EVA | 78.4 | 8.7 | | |
| EVA/UHMWPE | 79.7 | 3.6 | 133.8 | 24.7 |
| 1n-Al ₂ O ₃ /EVA/UHMWPE | 80.0 | 4.1 | 134.2 | 20.9 |
| 1w-Al ₂ O ₃ /EVA/UHMWPE | 79.5 | 3.0 | 134.2 | 24.4 |
| 1Fe-Al ₂ O ₃ /EVA/UHMWPE | 80.1 | 3.3 | 134.2 | 32.3 |

Overall endothermic enthalpy for six samples was classified in the following rank:

UHMWPE (78.2 J/g) > 1Fe-Al₂O₃/EVA/UHMWPE (35.6 J/g) > EVA/UHMWPE (28.3 J/g) > 1w-Al₂O₃/EVA/UHMWPE (27.4 J/g) > 1n-Al₂O₃/EVA/UHMWPE (24.9 J/g) > EVA (8.7 J/g).

Thermal degradation study of hybrid composite fibers

Thermogravimetric curves of EVA, UHMWPE fibers and composites are presented on Figure 8a. Thermal stability was evaluated at decomposition temperatures at 5% of weight loss ($T_{0.05}$) and at 50% of weight loss ($T_{0.5}$).

Degradation of EVA, as the matrix, was followed by two degradation steps. The first step was a deacetylation process that started at 286.5 °C and completed at 381.5 °C. In this temperature range, chain fragments as carbon monoxide, carbon dioxide ketene and water were probably formed during the thermal decomposition of the acetic acid.³³ Within this interval EVA yielded ethylene and acetic acid.³⁴

At the first point ($T_{0.05}$), the highest temperature of degradation occurred for UHMWPE fibers and 1Fe-Al₂O₃/EVA/UHMWPE, 408.1 °C and 383.3 °C, respectively. In interval from 156.6 °C to 351.4 °C, 1Fe-Al₂O₃/EVA/UHMWPE showed higher stability than other samples.

On the second point ($T_{0.5}$), all of the samples displayed similar degradation profiles. The second step referred to degradation of the EVA backbone (chain scission within the interval of 400-500 °C).^{35,36} It was noticeable that the presence of fillers in EVA improved the thermal stability of the matrix. According to the results presented in Table 3, decomposition temperatures $T_{0.05}$ and $T_{0.5}$ showed that the reinforcement of EVA matrix improved the thermal stability.³⁰ The experimental results (Figure 8a) showed that the addition of fillers improved thermal stability and prolonged degradation temperatures.

The higher mass residual for neat EVA (3.27%) was associated with networking of EVA polymer due to instability and formation of double bonds after the degradation. M. C. Costache et al. suggested that in the first step of degradation, acetic acid loss was related to the presence of hydroxyl groups on the edges of fillers.³⁷ That was in accordance with the results obtained for neat EVA. Results for hybrid composite fibers proved that the lowest mass weight loss was for 1n-Al₂O₃/EVA/UHMWPE and 1Fe-Al₂O₃/EVA/UHMWPE. Degradation processes of EVA matrix, UHMWPE fibers and UHMWPE/EVA hybrid composite fibers were represented by DTG curves in

Figure8(b). For all samples except UHMWPE fibers, degradation occurred at approximately 351 °C. The final point was similar for all samples and it is shown in Table 3.

Table 3. Temperatures obtained from TG/DTG diagrams.

| Samples | TG | | Mass res. (%) | DTG | |
|--|-----------------|----------------|---------------|------------|------------|
| | $T_{0.05}$ (°C) | $T_{0.5}$ (°C) | | T_1 (°C) | T_2 (°C) |
| UHMWPE | 408.1 | 477.5 | 0.421 | | 480.9 |
| EVA | 347.6 | 471.1 | 3.277 | 364.1 | 481.6 |
| EVA/UHMWPE | 345.7 | 471.7 | 3.243 | 352.3 | 477.2 |
| 1n-Al ₂ O ₃ /EVA/UHMWPE | 357.1 | 471.8 | 0.009 | 352.9 | 478.4 |
| 1w-Al ₂ O ₃ /EVA/UHMWPE | 349.5 | 472.7 | 1.154 | 351.6 | 480.8 |
| 1Fe-Al ₂ O ₃ /EVA/UHMWPE | 383.3 | 472.4 | 0.110 | 351.6 | 478.5 |

[insert Figure 8.]

Figure8. TG (a) and DTG (b) diagrams for EVA, UHMWPE fibers and hybrid composite fibers.

Tensile test

Tests were performed in order to determine tensile strength and modulus of elasticity of EVA/UHMWPE hybrid composite fibers with different fillers and their mass fractions.

The values of tensile strength of different concentration of fillers are presented in Figure9. The highest values for different fillers and concentrations were for 1Fe-Al₂O₃/EVA/UHMWPE (823.02 MPa).

[insert Figure 9.]

Figure9. Tensile test results of hybrid composites.

According to the results given in Figure9, dependence of tensile stress on relative displacement of four samples of 1n-Al₂O₃/EVA/UHMWPE, 1w-Al₂O₃/EVA/UHMWPE, 1Fe-Al₂O₃/EVA/UHMWPE and EVA were presented on Figure10. The addition of 1 wt.% of any of the used fillers showed that the highest tensile stress was for 1Fe-Al₂O₃/EVA/UHMWPE.

[insert Figure 10.]

Figure 10. Tensile test diagram for hybrid composite fibers with 1 wt.% of fillers.

The measured values of tensile strength for 1n-Al₂O₃/EVA/UHMWPE, 1w-Al₂O₃/EVA/UHMWPE and 1Fe-Al₂O₃/EVA/UHMWPE compared to EVA/UHMWPE showed an increase of 63.0, 31.9 and 100.7%, respectively. The significant improvement in the mechanical properties could be attributed to the strong interaction between the hydroxyl groups of Fe-Al₂O₃ and EVA because of the sintering process at lower temperatures, where OH groups partially remained on alumina surface, so the better binding with EVA was expected. Formation of agglomerates of w-Al₂O₃ resulted in a reduction of tensile strength value.

Resulting values of the modulus of elasticity (Figure11) had a different trend. The highest value of modulus of elasticity shown for 1n-Al₂O₃/EVA/UHMWPE indicated the increase of rigidity of the hybrid composites, resulting brittle failure of the matrix seen in the SEM image (Figure5a). A larger area under the stress-displacement curve for the 1Fe-Al₂O₃/EVA/UHMWPE sample was a consequence of more resilient composite

obtained that can absorb a higher amount of tensile energy. Such a behavior was reflected by lower modulus of elasticity. Alumina fillers in a form of whiskers in a sample 1w-Al₂O₃/EVA/UHMWPE caused the lowest mechanical properties in relation to other hybrid composites, while the modulus of elasticity was lower than the one obtained for EVA without particle reinforcement (EVA/UHMWPE). This behavior may be explained with higher amount of aggregates in the matrix that disabled interaction with the main reinforcement of the composites – UHMWPE.

[insert Figure 11.]

Figure11. Effect of the composition of the samples on the modulus of elasticity.

Conclusions

In this article, experimental results were presented for UHMWPE fibers as reinforcement with 1, 3 and 5 wt.% of alumina nanoparticles, whiskers and iron doped alumina particles as fillers in EVA matrix. The tensile results obtained for hybrid composites with 1 wt.% of particles indicated that the optimal weight percentage was used which was compliant with the literature.

In comparison with other used commercial fillers, Fe-Al₂O₃ showed a submicron particle size distribution. Average diameter of 0.15 μm had a significant effect on the mechanical properties due to better dispersion and distribution in the matrix.

FE-SEM analysis proved the best fiber-matrix interfacial adhesion for 1-Fe- Al_2O_3 /EVA/UHMWPE due to resilient longitudinal cracking of EVA that confirmed excellent bonding.

This was in agreement with tensile test results where 1-Fe- Al_2O_3 /EVA/UHMWPE showed improved mechanical properties in comparison with hybrid composite fibers with nanoparticles or whiskers.

FT-IR spectra indicated better interactions between Fe- Al_2O_3 and matrix due to established hydrogen bonds.

Presented results from the FT-IR, FE-SEM, and tensile stress analysis showed that in comparison with commercial available aluminum oxide nanoparticles with particle size less than 50 nm and alumina whiskers (diameter 2 – 4 nm \times 400 nm), synthesized iron oxide doped alumina particles were the best choice for hybrid reinforcement of EVA/UHMWPE composites.

DSC analysis presented the highest melting enthalpy of UHMWPE phase (32.3 J/g) for 1Fe- Al_2O_3 /EVA/UHMWPE, which indicated that the composite with the best structural arrangement of main reinforcement was obtained.

TG/DTG curves represented that degradation process temperatures of EVA matrix, UHMWPE fibers and UHMWPE/EVA composites were similar for all samples except UHMWPE fibers. Temperature of onset of degradation ($T_{0.05}$) showed that the particles improved thermal stability of EVA/UHMWPE composites, where the highest value was

observed for UHMWPE fibers. It was interesting to notice that iron doped particles improved thermal stability of hybrid composite until 350 °C even compared to UHMWPE fibers.

The incorporation of different combination of fillers had, in general, a significant improvement effect on modulus of elasticity causing different type of composite failure. The highest modulus of elasticity for composite with alumina nanoparticles (1n-Al₂O₃/EVA/UHMWPE) was followed by brittle matrix failure. The most resilient hybrid composite with iron doped alumina represented the composite that was able to absorb the highest amount of tensile energy.

It can be concluded that EVA/UHMWPE hybrid composite fibers with Fe-Al₂O₃ particles were superior to those of EVA/UHMWPE, or hybrid composites with Al₂O₃ nanoparticles or whiskers.

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FIGURE CAPTIONS

Figure1. Schematic presentation of UHMWPE impregnation process.

Figure2. (a) FE-SEM micrograph of Fe-Al₂O₃, (b) Magnified part of Fe-Al₂O₃, FE-SEM micrograph, (c) Statistical distribution of Fe-Al₂O₃, – diameter, and (d) Particle size distribution of Fe-Al₂O₃.

Figure3. FT-IR spectra for fillers and hybrid composite fibers: (a) EVA/UHMWPE, (b) n-Al₂O₃, w-Al₂O₃, Fe-Al₂O₃, and (c) 1n-Al₂O₃/EVA/UHMWPE, 1w-Al₂O₃/EVA/UHMWPE, 1Fe-Al₂O₃/EVA/UHMWPE hybrid composite fibers.

Figure4. Dimensions of the samples.

Figure5. The FE-SEM micrograph of unmodified EVA/UHMWPE composite: (a) Cross section before tensile test, (b) Surface of composite breakage.

Figure6. The FE-SEM micrograph of EVA/UHMWPE composite with: (a) 1n-Al₂O₃, (b) 1w-Al₂O₃, and (c) 1Fe-Al₂O₃.

Figure7. DSC thermograms of EVA, UHMWPE fibers and composite fibers.

Figure8. TG (a) and DTG (b) diagrams for EVA, UHMWPE fibers and hybrid composite fibers.

Figure9. Tensile test results of hybrid composites.

Figure 10. Tensile test diagram for hybrid composite fibers with 1 wt.% of fillers.

Figure11. Effect of the composition of the samples on the modulus of elasticity.

TABLE CAPTIONS

Table 1. Statistical data of particle size distribution obtained by image analysis of Fe-Al₂O₃.

Table 2. Temperatures and enthalpies obtained from DSC thermograms.

Table 3. Temperatures obtained from TG/DTG diagrams.

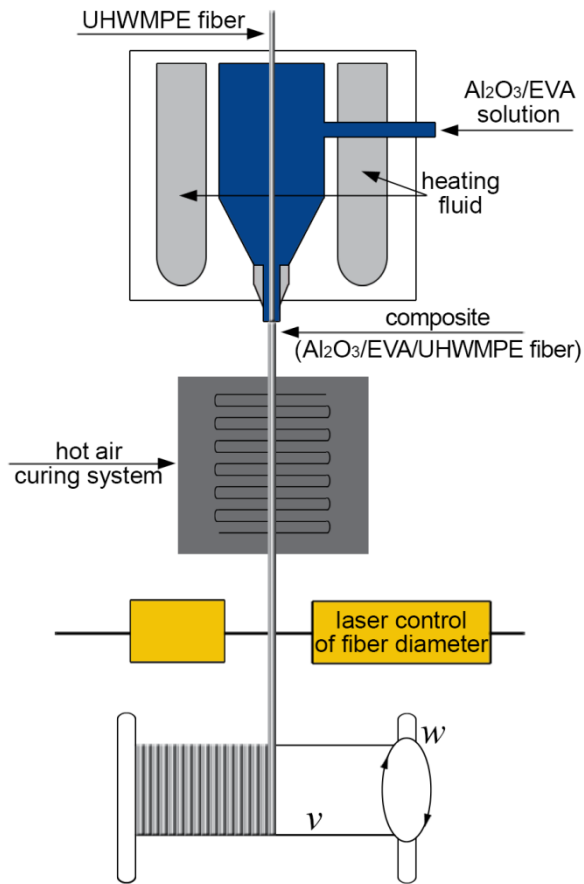


Figure 1

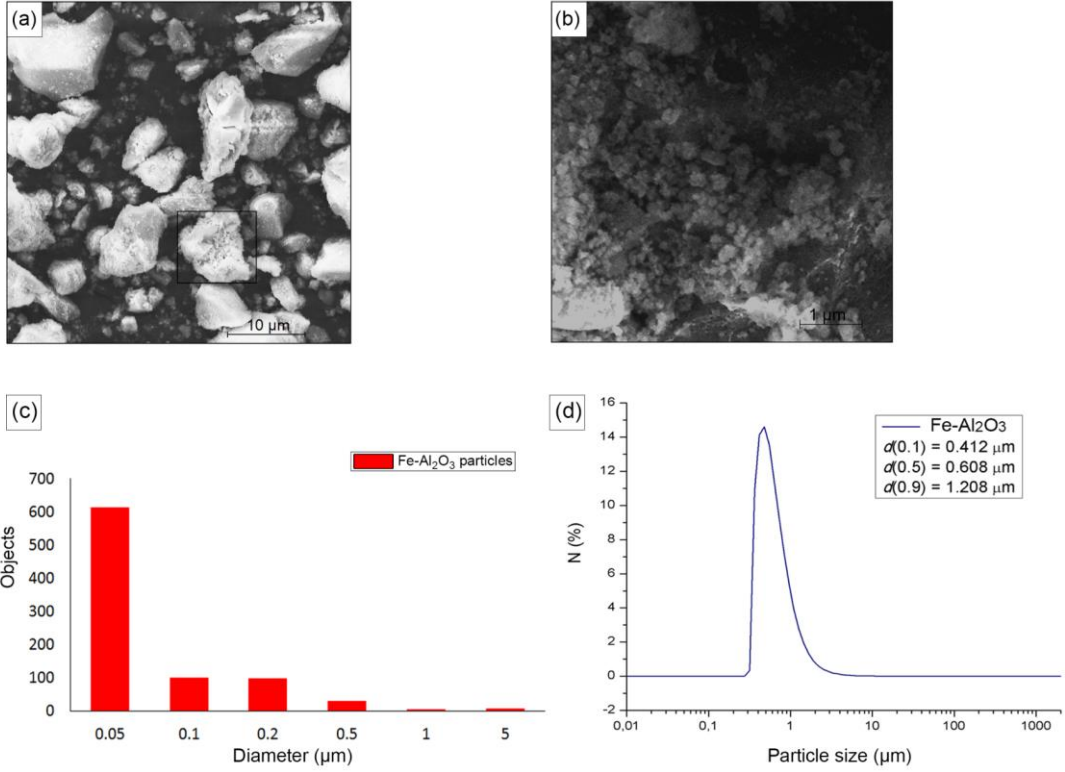


Figure 2

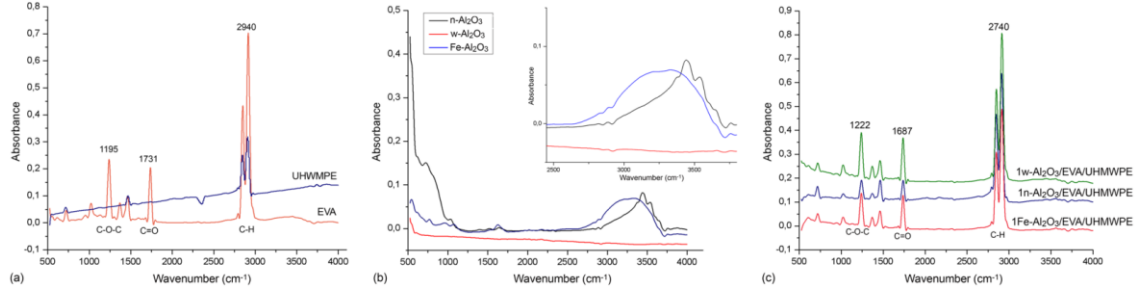


Figure 3

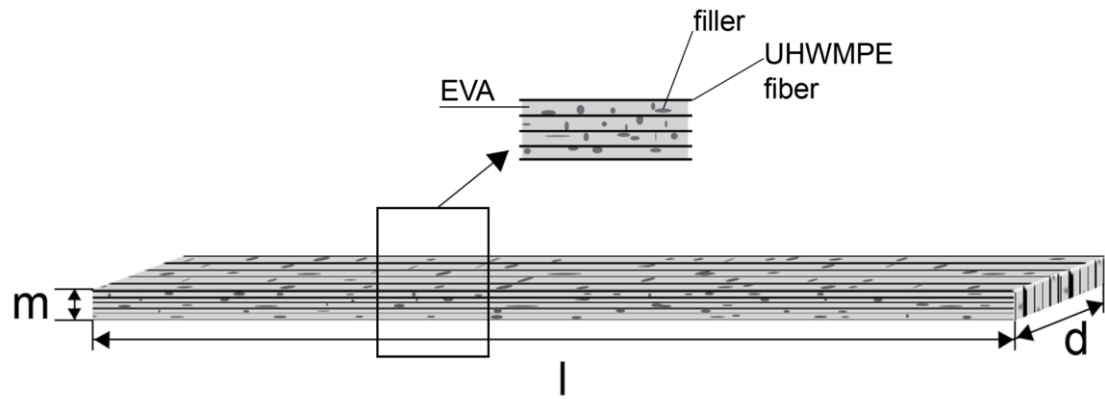


Figure 4

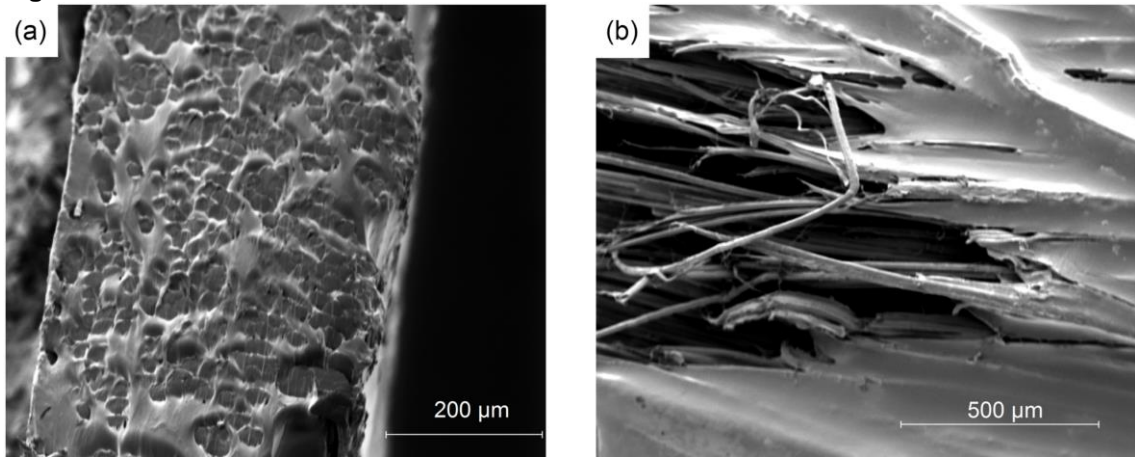


Figure 5

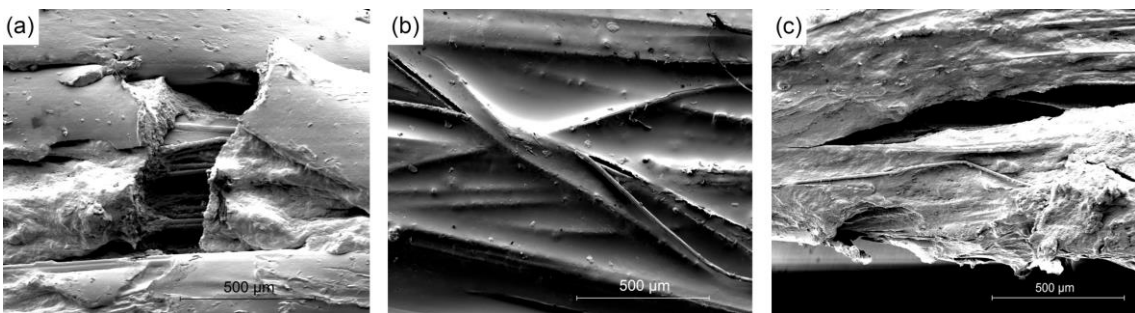


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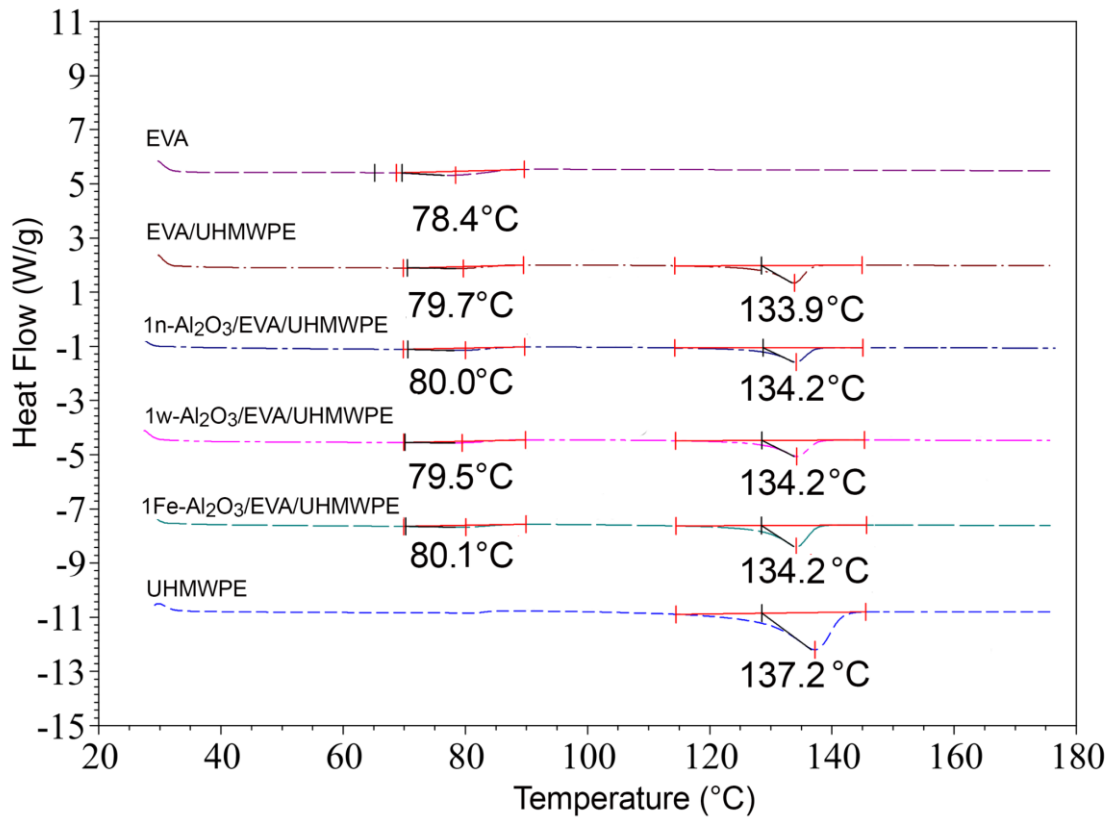


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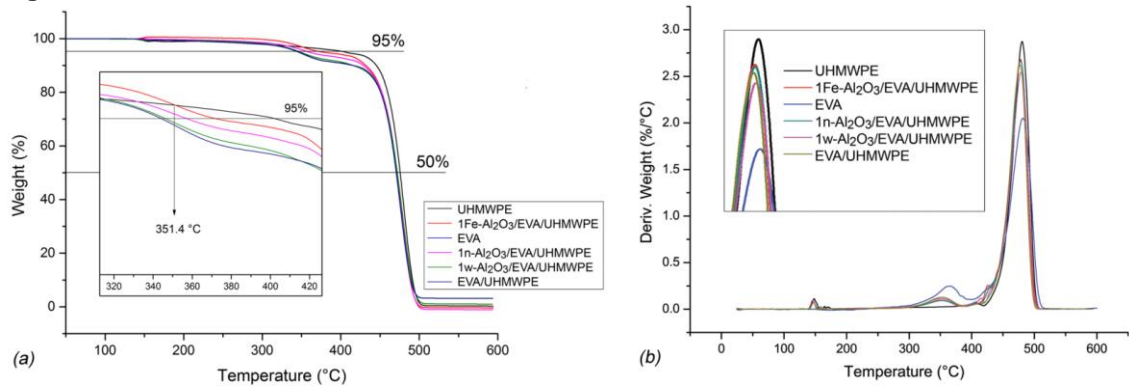


Figure 8

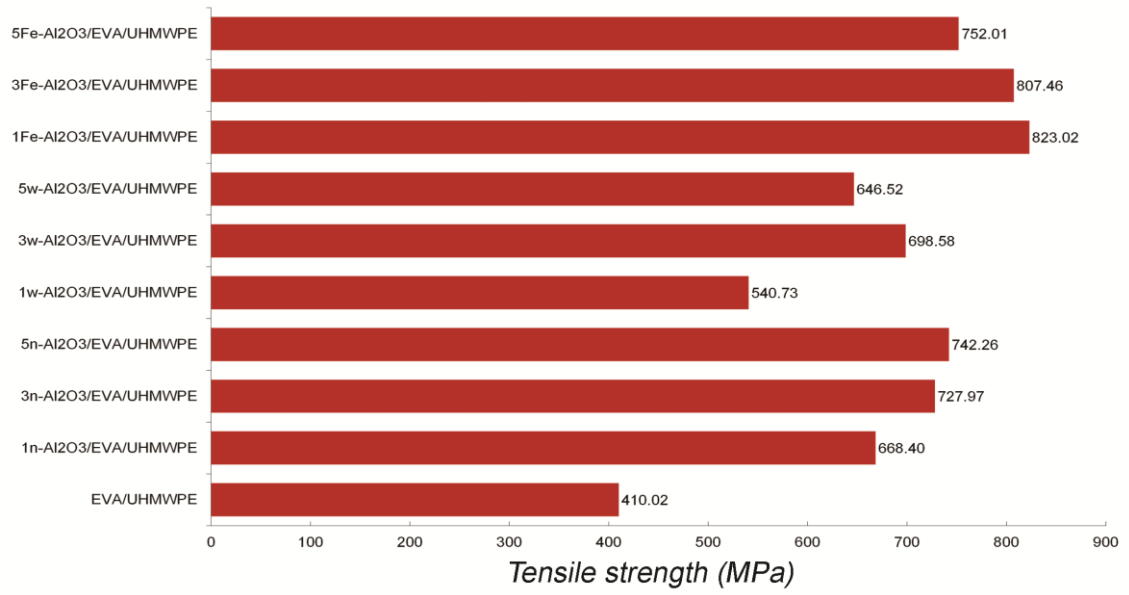


Figure 9

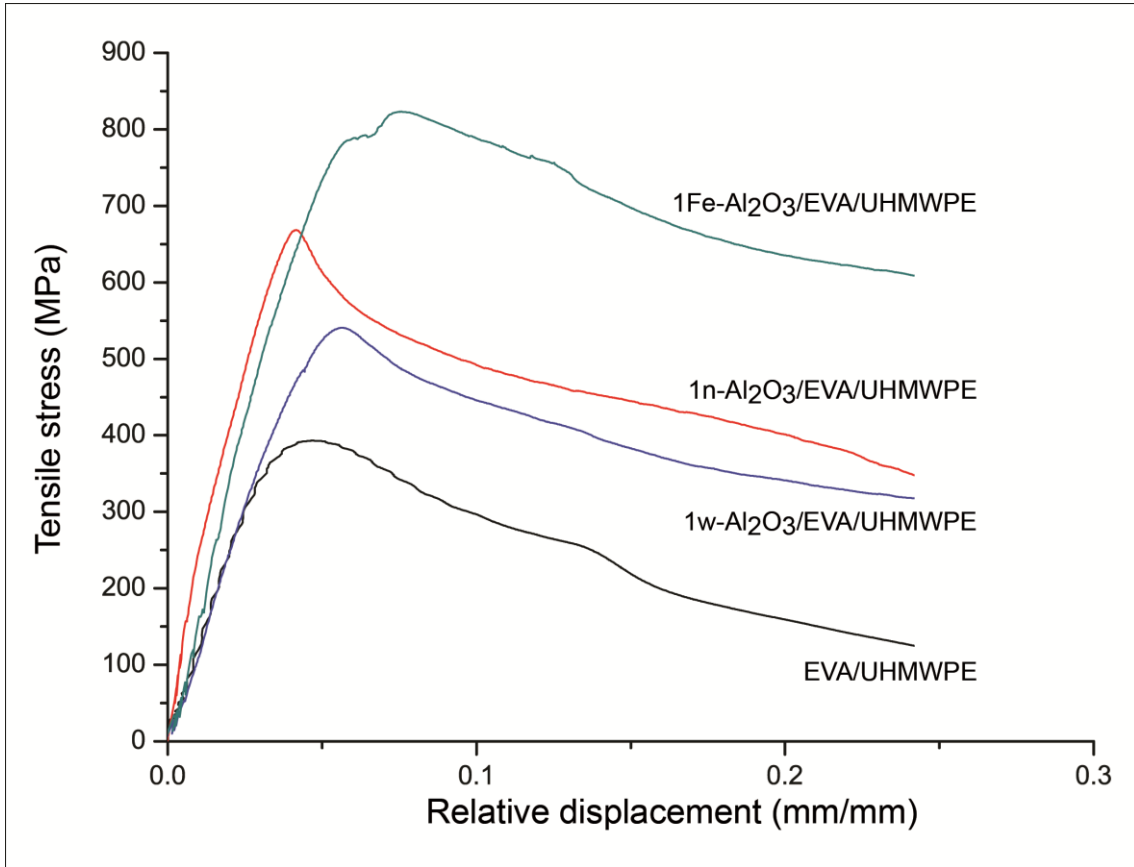


Figure 10

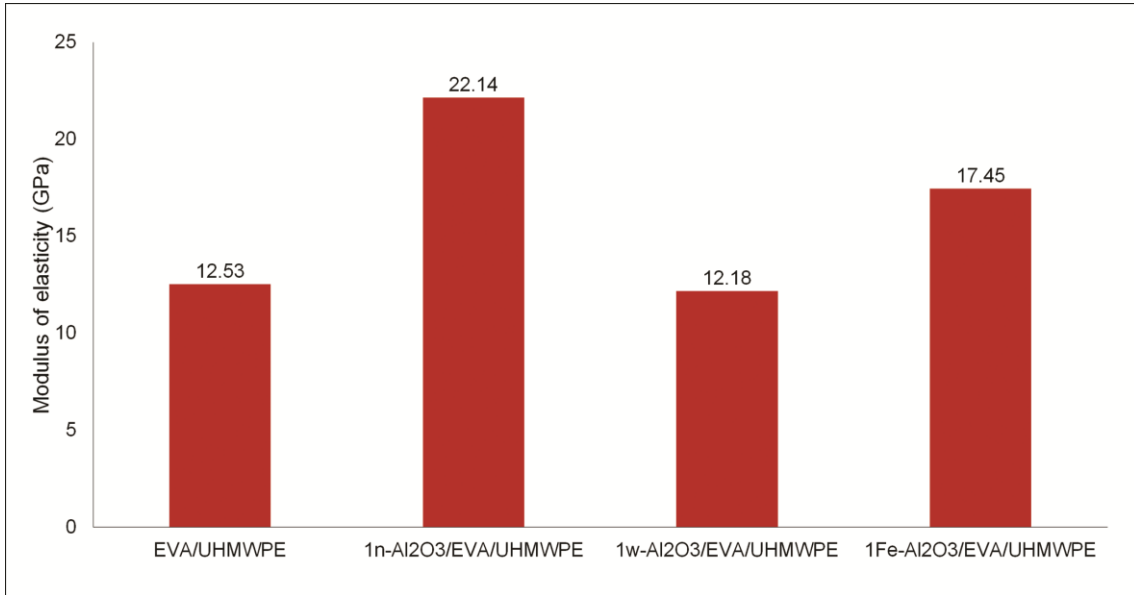


Figure 11