# PRODUCTION AND CHARACTERIZATION OF PROTEASE FROM HALOPHILIC *VIRGIBACILLUS* SPECIES CD6

# LAM MING QUAN

A dissertation submitted in partial fulfilment of the requirements for the award of the degree of Master of Science (Biotechnology)

Faculty of Biosciences and Medical Engineering
Universiti Teknologi Malaysia

a						
Spec	cially dedicated	l to my beloved	family, future	e life partner,	soulmates an	d friends

#### **ACKNOWLEDGEMENT**

First and foremost, I would also like to express my deepest gratitude towards my supervisor, Dr. Chong Chun Shiong who had guided me in every step taken to accomplish my master dissertation throughout these 1.5 years. Even though he was at his post-doctoral position at remote country for 1 year, however, he still concerned about my progress and always kept in touch to update the latest situation. Besides that, I would like to thank my co-supervisor, Dr. Haryati Jamaluddin for being so supportive in term of giving advices and resources, especially during the period when my main supervisor was not around.

Not forgetting fellow seniors in the Enzyme Research Laboratory including Ms. Suganthi Thevarajoo, Ms. Chitra Selvaratnam and Mr. Lim Jia Chun, assistant science officers (Mrs. Fatimah Harun and Ms. Norsyuhada Jaafar), administrative assistant (Mrs. Zulbaidah Muhammad), coursemates (Ms. Tan Ee Yau, Ms. Sonia Nair P. Kreshnan and Mrs. Zetty Amirah Zulkifli), other seniors, juniors and staffs in Faculty of Biosciences and Medical Engineering. Thank you for all of your help during problem solving, listening ears and experiences that you have shared to me.

Apart from that, I would like to acknowledge MyBrain15 scholarship (MyMaster) from Ministry of Higher Education Malaysia, Ministry of Education Malaysia (Project number: 4F265) and Universiti Teknologi Malaysia RU grant (Project number: 07H43) as my financial support for master degree. Last but not least, I would like to take this opportunity to sincerely thank my parents (Lam Poy Leng and Lye Chou Ngo), siblings, relatives, soulmates and friends who have supported me always so that I am able to accomplish what I have in life so far either mentally or physically.

#### ABSTRACT

In enzyme production industries, the major challenges that hinder the efficient and economic commercial scale application of proteases are their stability in broad range of pH, temperature, salinity, as well as their optimal activity in the presence of metal ions, organic solvents and detergents. Moreover, the enzyme purification steps also contribute to the cost of production. To overcome this problem, characterization and production of crude protease with attractive properties from wild bacterial isolate could be an alternative as it is a more cost-effective way compared to production of protease that involves purification steps and protein engineering approach. Therefore, crude protease of Virgibacillus sp. CD6 isolated from salted-fish was characterized in this study using azocasein assay and bioinformatics tools. Protease production was found to be highest when using soybean meal and yeast extract as nitrogen source compared to other organic nitrogen sources. The protease exhibited vast range of stability with optimum activity at 10.0 % (w/v) NaCl, 60°C, pH 7 and 10, indicating its polyextremophilicity. The enzyme activity was enhanced by Mg<sup>2+</sup>, Mn<sup>2+</sup>, Cd<sup>2+</sup> and Al<sup>3+</sup>. Both PMSF and EDTA hindered protease activity, denoting the presence of serine protease and metalloprotease properties respectively. High protease stability (>80%) was demonstrated in presence of organic solvents and detergent constituents investigated, and surprisingly it is exceptionally compatible with commercial detergents. Phylogenetic analyses revealed that proteases of Virgibacillus sp. demonstrated far distance relationship with other species, which worth for further exploration. Attributes of this protease can actualize necessity of searching superlative enzymes from extremophiles for diverse applications, particularly in detergent industry.

#### **ABSTRAK**

Dalam industri penghasilan enzim, cabaran utama yang menghalang aplikasi komersial protease yang cekap dan ekonomi adalah ciri-ciri protease yang stabil dalam pelbagai pH, suhu, kadar garam serta aktiviti optimum dalam ion logam, pelarut organik, dan unsur detergen. Selain itu, proses penulenan enzim juga menyumbang kepada kos penghasilan. Bagi mengatasi masalah ini, pencirian dan penghasilan protease dari bakteria tanpa melibatkan proses penulenan boleh menjadi alternatif kerana ia adalah cara yang kos efektif berbanding dengan penghasilan protease yang melibatkan penulenan enzim dan kejuruteraan protein. Oleh itu, protease daripada Virgibacillus sp. CD6 yang dipencilkan daripada ikan masin telah dicirikan dalam kajian ini dengan penggunaan azocasein assay dan alat bioinformatik. Penghasilan protease didapati paling tinggi apabila menggunakan kacang soya dan ekstrak yis sebagai sumber nitrogen berbanding dengan sumber nitrogen organik yang lain. Protease tersebut mempamerkan luas kestabilan dengan aktiviti optimum pada 10.0% (w/v) NaCl, 60°C, pH 7 dan 10, menunjukkan ciri poli-ekstremofi. Aktiviti enzim telah dipertingkatkan oleh Mg<sup>2+</sup>, Mn<sup>2+</sup>, Cd<sup>2+</sup> dan Al<sup>3+</sup>. Kedua-dua PMSF dan EDTA didapati menghalang aktiviti protease, menandakan ciri protease serine dan metalloprotease masing-masing. Kestabilan protease yang tinggi (>80%) telah ditunjukkan dalam pelarut organik dan unsur detergen, serta amat serasi dengan bahan pencuci komersial. Analisis filogenetik menunjukkan bahawa protease daripada Virgibacillus sp. mempunyai hubungan yang jauh dengan spesies lain, bernilai untuk penerokaan selanjutnya. Sifat-sifat protease ini boleh merealisasi keperluan mencari enzim cemerlang dari esktremofi untuk pelbagai aplikasi, terutamanya dalam industri detergen.

# TABLE OF CONTENTS

CHAPTER			TITLE	PAGE	
	DECLARATION		N	ii	
	DED	<b>ICATION</b>		iii	
	ACK	NOWLED	GEMENT	iv	
	ABSTRACT				
	ABSTRAK TABLE OF CONTENTS			vi	
				vii	
	LIST	S OF TAB	LES	xii	
	LIST	S OF FIGU	JRES	xvi	
	LIST	OF SYMB	SOLS	xxii	
	LIST OF ABBREVIATION			xxiv	
	LIST	OF APPE	NDICES	xxviii	
1	INTE	RODUCTIO	ON		
	1.1	Backgro	und of study	1	
	1.2	Problem	statement / significance of study	3	
	1.3	Objective	es of study	3	
	1.4	Scope of	study	4	
2	LITE	RATURE	REVIEW		
	2.1	Halophil	ic bacteria	5	
		2.1.1	Ecology and phylogeny	6	
		2.1.2	Adaptations in saline environment	9	
	2.2	Protease-	-producing halophilic bacteria	12	
		2.2.1	Virgibacillus sp.	12	
	2.3	Protease		13	

		2.3.1	Protease source	14
		2.3.2	Protease classification	16
		2.3.3	Physiological function	19
		2.3.4	Protease engineering	20
	2.4	Applic	ations of protease	21
		2.4.1	Food industry	21
		2.4.2	Detergent industry	23
		2.4.3	Leather processing	25
		2.4.4	Pharmaceutical industry	26
		2.4.5	Waste management	27
3	MAT	ERIALS	AND METHODS	
	3.1	Experi	mental design	28
	3.2	Bacteri	al culture	30
	3.3	Gram s	staining	30
	3.4	Qualita	tive proteolytic screening	31
		3.4.1	Skim milk agar plate assay	31
		3.4.2	Gelatin liquefaction test	32
	3.5	Semiqu	nantitative analysis of proteolytic	32
		activity	1	
	3.6	Investi	gation of protease production medium	33
	3.7	Inoculu	am preparation and protease	34
		produc	tion	
	3.8	Quanti	tative protease activity investigation	35
		3.8.1	Azocasein assay	35
	3.9	Lowry	assay	36
		3.9.1	Construction of standard calibration	36
			curve	
		3.9.2	Protein content and specific activity	37
			determination in crude enzyme	
	3.10	Bacteri	al growth profiling in different	37
		nitroge	n sources	

	3.11	Effect of	f nitrogen sources on protease	38
		producti	on	
	3.12	Effect of	f temperature on protease activity	39
		and stab	ility	
	3.13	Effect of	f pH on protease activity and	39
		stability		
	3.14	Effect of	f salt concentration on protease	39
		activity	and stability	
	3.15	Stability	of protease in presence of organic	40
		solvent		
	3.16	Stability	of protease in presence of metal	40
		ions		
	3.17	Stability	of protease in presence of inhibitors	40
	3.18	Stability	of protease in presence of detergent	41
		constitue	ents	
	3.19	Compati	bility of protease with commercial	42
		detergen	ats	
	3.20	Substrat	e specificity of protease	42
	3.21	Bioinfor	matics analysis of protease of	43
		Virgibac	cillus sp.	
		3.21.1	Sequence retrieval and primary	43
			analysis	
		3.21.2	Multiple sequence alignment and	43
			phylogenetic analysis	
4	DECIII	TC AND	DISCUSSION	
4	4.1		logical characterization of	44
	4.1	1	cillus sp. strain CD6	44
		4.1.1	Gram staining	45
	4.2		ive screening of protease activity	43 47
	4.4	4.2.1	Skim milk agar plate assay	47 47
		4.2.1	Gelatin liquefaction test	47
		4.2.2	Gelatin inqueraction test	40

4.3	Semiquantivative analysis of proteolytic	49			
	activity				
4.4	Protease production medium investigation	50			
	4.4.1 Carbon source screening	50			
	4.4.2 Nitrogen source screening	51			
4.5	Growth profile of Virgibacillus sp. strain	52			
	CD6 in different nitrogen sources				
4.6	Effect of nitrogen source on protease	55			
	production				
4.7	Effect of temperature on protease activity	58			
	and stability				
4.8	Effect of pH on protease activity and	60			
	stability				
4.9	Effect of salt concentration on protease	63			
	activity and stability				
4.10	Effect of metal ions on protease activity and	l 66			
	stability				
4.11	Effect of inhibitors on protease activity and	69			
	stability				
4.12	Effect of organic solvents on protease	71			
	activity and stability				
4.13	Effect of surfactants and oxidizing agent on	73			
	protease activity and stability				
4.14	Compatibility of protease with commercial	75			
	detergents				
4.15	Substrate specificity	77			
4.16	Bioinformatics analysis of extracellular				
	protease of Virgibacillus sp.				
	4.16.1 Protein sequences retrieval and	78			
	prediction of extracellular protease	2			
	4.16.2 Primary sequence analysis	81			
	4.16.3 Conserved domain analysis and	85			
	multiple sequence alignment				

		4.16.4	Phylogenetic analysis	96
5	CON	CLUSIO	NS	
	5.1	Conclu	sion	104
	5.2	Recom	mendations	105
REFEREN	ICES			106
Appendices	s A - B			126 - 131

# LIST OF TABLES

TABLE NO.	TITLE	PAGE
2.1	Category of halophiles according to their salt requirement, in both percentage (%) and molarity (M)	5
2.2	Compatible solutes (zwitterionic, non-charged and charged solutes) and its properties, utilized by halophilic bacteria in osmoadaptation	11
2.3	Descriptions of proteases from animal and plant origins	14
2.4	Descriptions of proteases from microbial origins	15
2.5	Classification and descriptions of exopeptidases	16
2.6	Properties and mechanism of action of various endopeptidases classified based on chemical nature of catalytic site	17
2.7	Classification of proteases based on evolutionary relationships, showing diverse clans and families	18
2.8	Commercial proteases (with their microbial sources and suppliers) used in food industry until today	22
2.9	Commercial proteases (with their microbial sources and suppliers) used in detergent industry until today	24
2.10	Commercial proteases (with their microbial sources and suppliers) used in leather processing	25

2.11	Properties of purified proteases approved for use as therapeutics agents	26
3.1	Components and quantities of protease production medium for <i>Virgibacillus</i> sp. strain CD6. This medium was modified from chemically defined medium	34
3.2	Reagents preparation for Lowry assay	36
3.3	Different types of protease inhibitors used in protease stability assessment	41
3.4	Ionic surfactants, non-ionic surfactants and oxidizing agent used in protease stability assessment	41
4.1	Colony morphological characteristics of <i>Virgibacillus</i> sp. strain CD6 after 24 hours of incubation at 37°C	45
4.2	Gram stain results and bacterial shape observed under 1000X magnification of microscope	46
4.3	Semiquantitative analysis of proteolytic activity using crude protease of <i>Virgibacillus</i> sp. strain CD6. Hydrolysis index was expressed ( $\log_{10}$ mm <sup>2</sup> ) as indication of degree of proteolytic activity. Each value represents mean $\pm$ SD, n=3	49
4.4	Screening of carbon sources utilized by <i>Virgibacillus</i> sp. CD6 for growth after 24 hours of 37°C incubation under shaking condition at 200 rpm. +, growth was observed, solution changed turbid with $OD_{600} > 0.7$ ; -,	
	no bacterial growth, solution remained clear with $\ensuremath{\mathrm{OD}}_{600}$ $$<0.1$$	50

4.5	Screening of nitrogen sources utilized by Virgibacillus	
	sp. CD6 for growth after 24 hours of 37°C incubation	
	under shaking condition at 200 rpm. Trisodium citrate	
	was used as carbon source. +, growth was observed,	
	solution changed turbid with OD $_{600}$ > 0.7; -, no bacterial	
	growth, solution remained clear with $OD_{600} < 0.1$	
		51
4.6	Stability of protease from Virgibacillus sp. strain CD6	
	in presence of various metal ions after 1 hour of pre-	
	incubation at 50°C and pH 8. Enzyme activity without	
	any metal ions (control) was taken as 100%. Each	
	value represents mean $\pm$ SD, n=3	68
4.7	Stability of protease from <i>Virgibacillus</i> sp. strain CD6	
	in presence of different types of protease inhibitors	
	after 1 hour of pre-incubation at 50°C. Enzyme activity	
	without any inhibitors (control) was taken as 100%.	
	Each value represents mean $\pm$ SD, n=3	70
4.8	Stability of protease from <i>Virgibacillus</i> sp. strain CD6	
4.0	in presence of 25% (v/v) organic solvents after 4 hours	
	of pre-incubation at 50°C. Enzyme activity without	
	organic solvent (control) was taken as 100%. Each	
		72
	value represents mean $\pm$ SD, n=3	72
4.9	Stability of protease from Virgibacillus sp. strain CD6	
	in presence of various detergents after 1 hour of pre-	
	incubation at 50°C. Enzyme activity without any	
	detergent (control) was taken as 100%. Each value	
	represents mean ± SD, n=3	74

4.10	Compatibility of protease from <i>Virgibacillus</i> sp. strain	
	CD6 with various commercial detergents after 1 hour	
	of pre-incubation at 50°C. Enzyme activity without any	
	commercial detergents (control) was taken as 100%.	
	Each value represents mean $\pm$ SD, n=3	76
4.11	Substrate specificity of protease from <i>Virgibacillus</i> sp. strain CD6. Percentages of relative activity shown are relative to maximum activity of azocasein substrate. Each value represents mean $\pm$ SD, n=3	77
4.12	Physico-chemical characteristics of extracellular proteases predicted by ProtParam tool	82

# LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
3.1	Experimental design of this research according to objectives of study.	29
3.2	Hydrolysis index scale $(\log_{10} \text{ mm}^2)$ used for semiquantitative analysis of proteolytic activity. Values decrease from left to right as degree of proteolytic decrease	32
4.1	Virgibacillus sp. strain CD6 on Marine Agar (MA) after 24 hours of incubation at 37°C	44
4.2	Qualitative screening of caseinolytic activity using skim milk agar. Positive and negative control are plates inoculated with protease-producing bacterium <i>Shewanella</i> sp. strain CH1 (a) and without inoculation (b) respectively; Clear zones observed on skim milk agar inoculated with <i>Virgibacillus</i> sp. strain CD6 after 3 days of incubation at 37°C (c)	47
4.3	Qualitative screening of gelatinolytic activity using gelatin liquefaction test. A control was prepared without inoculation of bacterial culture (a); gelatin was liquefied by <i>Virgibacillus</i> sp. strain CD6 after 10 days of incubation at 37°C and then refrigerated at 4°C for 1 hour (b)	48
	at 4 C 101 1 110th (b)	40

4.4	Growth profile of <i>Virgibacillus</i> sp. strain CD6 in medium supplemented with different nitrogen sources, incubated at 37°C and under 200 rpm of shaking condition. Absorbance at 600 nm of cells were taken at every 2 hours' interval until 24 <sup>th</sup> hours. Mean values (n=3) were expressed and standard	
	deviations are indicated as error bars	54
4.5	Effect of various nitrogen sources on protease production (U/ml) of <i>Virgibacillus</i> sp. strain CD6. Samples were taken after 24 h of incubation at 37°C under shaking conditions (200 rpm). Mean values (n=3) are expressed and standard deviations are indicated as error bars	56
4.6	Effect of different nitrogen sources on protease production (specific activity, U/mg) of <i>Virgibacillus</i> sp. strain CD6, estimated using Lowry assay. Samples were taken after 24 h incubation at 37°C under shaking conditions (200 rpm). Mean values (n=3) are expressed and standard deviations are indicated as error bars	57
4.7	Effect of various temperature on activity and stability of protease from <i>Virgibacillus</i> sp. strain CD6. Relative activity (%) was calculated by relative to the case of reaction at which maximum activity was taken as 100%; Stability (%) was calculated by relative to enzyme activity before incubation was taken as 100%. Mean values (n=3) were reported and standard deviations are indicated	
	as error bars	59

4.8	Effect of pH on activity and stability of protease	
	from Virgibacillus sp. strain CD6. Relative activity	
	(%) and stability (%) were calculated by relative to	
	the case of reaction at which maximum activity was	
	taken as 100%. Mean values (n=3) were reported	
	and standard deviations are indicated as error bars	61
4.9	Effect of various salt concentration on activity and	
	stability of protease from Virgibacillus sp. strain	
	CD6. Relative activity (%) and stability (%) were	
	calculated by relative to the case of reaction at which	
	maximum activity was taken as 100%. Mean values	
	(n=3) were reported and standard deviations are	
	indicated as error bars	64
4.10	Signal peptide prediction by SignalP 4.1 server with	
	value above cutoff (>0.450), showing peptide	
	possessed in minor extracellular protease vpr,	
	indicating this protease can be secreted	
	extracellularly	79
4.11	Signal peptide prediction by SignalP 4.1 server with	
	value above cutoff (>0.450), showing peptide	
	possessed in thermostable alkaline protease,	
	indicating this protease can be secreted	
	extracellularly	79
4.12	Signal peptide prediction by SignalP 4.1 server with	
	value above cutoff (>0.450), showing peptide	
	possessed in zinc carboxypeptidase, indicating this	
	protease can be secreted extracellularly	80

4.13	Signal peptide prediction by SignalP 4.1 server with	
	value above cutoff (>0.450), showing peptide	
	possessed in neutral protease B, indicating this	
	protease can be secreted extracellularly	80
4.14	Composition of amino acids (with single alphabet	
	abbreviations) of minor extracellular protease vpr	
	(A), thermostable alkaline protease (B), zinc	
	carboxypeptidase (C) and neutral protease B (D),	
	analyzed by Statistical Analysis of Protein	
	Sequences (SAPS)	84
4.15	Family, domains and active sites of minor	
	extracellular protease vpr predicted by using	
	InterProscan (A) and ScanProsite (B)	88
4.16	Family, domains and active sites of thermostable	
	alkaline protease predicted by using InterProscan	
	(A) and ScanProsite (B)	89
4.17	Domain of zinc carboxypeptidase predicted by using	
	InterProscan	90
4.18	Family, domains and active sites of neutral protease	
	B predicted by using InterProscan (A) and	
	ScanProsite (B)	90
4.19	Multiple sequence alignment of minor extracellular	
	protease vpr of Virgibacillus massiliensis Vm-5 in	
	comparison with other bacterial minor extracellular	
	protease vpr by using Clustal Omega. The well-	
	conserved regions (I, II and III) are indicated by	
	black colour boxes; active sites are indicated by a red	
	asterisk (*). The dash (-) indicated the gap inserted	
	to optimize the sequence alignment	92

4.20	Multiple sequence alignment of thermostable	
	alkaline protease of Virgibacillus massiliensis Vm-5	
	in comparison with other thermostable alkaline	
	protease by using Clustal Omega. The well-	
	conserved regions (I, II and III) are indicated by	
	black colour boxes; active sites are indicated by a red	
	asterisk (*). The dash (-) indicated the gap inserted	
	to optimize the sequence alignment	93
4.21	Multiple sequence alignment of zinc	
	carboxypeptidase of Virgibacillus massiliensis Vm-	
	5 in comparison with other bacterial zinc	
	carboxypeptidase by using Clustal Omega. The	
	well-conserved region is indicated by black colour	
	box. The dash (-) indicated the gap inserted to	
	optimize the sequence alignment	94
4.22	Multiple sequence alignment of neutral protease B	
	of Virgibacillus massiliensis Vm-5 in comparison	
	with other bacterial neutral protease B by using	
	Clustal Omega. The well-conserved region is	
	indicated by black colour box; active site is indicated	
	by a red asterisk (*). The dash (-) indicated the gap	
	inserted to optimize the sequence alignment	95
4.23	The neighbor-joining tree based on analysis of minor	
	extracellular proteases vpr of 20 amino acid	
	sequences from different bacteria. Numbers on	
	nodes represent percentage bootstrap confidence	
	score for 1000 replicates. Red colour box indicates	
	minor extracellular protease vpr of this study	97

99
99
99
99
99
101
103

### LIST OF SYMBOLS

A<sub>420</sub> - Absorbance at 420 nm

A<sub>750</sub> - Absorbance at 750 nm

 $\alpha$  - Alpha

 $\approx$  - Approximately

 $\beta$  - Beta

°C - Degree celcius

D - Diameter

= **-** Equal

 $\gamma \qquad \quad \text{-} \qquad \text{Gamma}$ 

g - Gram

g/L - Gram per liter

> - Greater than

h - Hour

kPa - Kilo Pascal

< - Less than

L - Liter

log<sub>10</sub> - Logarithm to base 10

mg/ml - Milligram per milliliter

μl - Microliter

mg - Milligram

mg/L - Milligram per liter

ml - Milliliter

mm - Millimeter

mM - Millimolar

M - Molar mass

nm - Nanometer

- - Negative

n - Number

OD<sub>600</sub> - Optical density at 600 nm

/ - Or

% - Percent

cm<sup>-1</sup> - Per centimeter

M<sup>-1</sup> - Per molar

 $\pi$  - Pi

 $\pm$  - Plus-minus

+ - Positive

® - Registered trademark

 $^{2}$  - Square  $\times$  - Times

TM - Trademark

U/mg - Units per milligram

U/ml - Units per volume

v/v - Volume per volume

w/v - Weight per volume

#### LIST OF ABBREVIATIONS

A - Alanine

Al<sup>3+</sup> - Aluminum ion

 $Al_2(SO_4)_3$  - Aluminum sulfate

APC - Activated protein C

ATP - Adenosine triphosphate

BLASTp - Protein-protein Basic Local Alignment Search Tool

BSA - Bovine serum albumin

C - Cysteine

C<sub>6</sub>H<sub>5</sub>Na<sub>3</sub>O<sub>7</sub> - Trisodium citrate

C<sub>6</sub>H<sub>5</sub>Na<sub>3</sub>O<sub>7</sub>.2H<sub>2</sub>O - Trisodium citrate dihydrate

Ca<sup>2+</sup> - Calcium ion

CaCl<sub>2</sub> - Calcium chloride

Cd<sup>2+</sup> - Cadmium ion

Cd(NO<sub>3</sub>)<sub>2</sub> - Cadmium nitrate

Cl<sup>-</sup> - Chloride ion

Co<sup>2+</sup> - Cobalt ion

CoCl<sub>2</sub> - Cobalt chloride
Cu<sup>2+</sup> - Copper (II) ion

CuSO<sub>4</sub> - Copper (II) sulfate

CuSO<sub>4</sub>.5H<sub>2</sub>O - Copper (II) sulfate pentahydrate

D - Aspartic acid

DMSO - Dimethyl sulfoxide

DNA - Deoxyribonucleic Acid

DTT - Dithiothreitol

EC - Enzyme commission

EDTA - Ethylene Diamine Tetraacetic Acid

et al. - And friends

F - Phenylalanine

FDA - Food and Drug Administration

Fe<sup>3+</sup> - Ferum (III) ion

FeCl<sub>3</sub> - Ferum (III) chloride

G - Glycine

Glu, E
 H-bond
 Hydrogen bond
 Hydrogen ion

H<sub>2</sub>O<sub>2</sub>Hydrogen peroxideHClHydrochloric acid

His, H - Histidine
I - Isoleucine

IAA - Iodoacetic acid

ID - IdentifierK - Lysine

K<sup>+</sup> - Potassium ion

K<sub>2</sub>HPO<sub>4</sub> - Dipotassium hydrogen phosphate

KCl - Potassium chloride

KH<sub>2</sub>PO<sub>4</sub> - Potassium dihydrogen phosphate

KNO<sub>3</sub> - Potassium nitrate

L - Leucine

M - Methionine

MEGA 7.0 - Molecular Evolutionary Genetic Analysis version 7.0

Mg<sup>2+</sup> - Magnesium ion

MgCl<sub>2</sub> - Magnesium chloride

MgSO<sub>4</sub>.7H<sub>2</sub>O - Magnesium sulfate heptahydrate

Mn<sup>2+</sup> - Manganese ion

MnCl<sub>2</sub> - Manganese chloride

N - Asparagine  $Na^+$  - Sodium ion

Na<sub>2</sub>CO<sub>3</sub> - Sodium carbonate
NaCl - Sodium chloride

NaHCO<sub>3</sub> - Sodium bicarbonate

NaNO<sub>2</sub> - Sodium nitrite

NaOH - Sodium hydroxide

NH<sub>4</sub>Cl - Ammonium chloride

Ni<sup>2+</sup> - Nickel ion

NiSO<sub>4</sub> - Nickel sulfate
OH<sup>-</sup> - Hydroxide ion

P - Proline

PHB - Polyhydroxybutyrate

PMSF - Phenylmethylsulfonyl fluoride

pI - Isoelectric point

PSI-BLAST - Position-Specific Iterated Basic Local Alignment

Search Tool

Q - Glutamine R - Arginine

rcf - Relative centrifugal force

rpm - Rotary per minute

rRNA - Ribosomal ribonucleic acid

SAPS - Statistical Analysis of Protein Sequences

SD - Standard deviation

SDS - Sodium dodecyl sulfate

SDS-PAGE - Sodium dodecyl sulfate polyacrylamide gel

electrophoresis

Ser, S - Serine

sp. - Species (singular)spp. - Species (plural)

T - Threonine

t-PA - Tissue plasminogen activator

TCA - Trichloroacetic acid

Tris - 2-Amino-2-(hydroxymethyl)propane-1,3-diol

u-PA - Urokinase type plasminogen activator

USA - United States of America

USD - United States dollar

UV - Ultraviolet

V - Valine

W - Tryptophan

X, Xaa - Unknown amino acid

 $\begin{array}{cccc} Y & & - & Tyrosine \\ Zn^{2+} & & - & Zinc \ ion \end{array}$ 

 $ZnSO_4$  - Zinc sulfate

## LIST OF APPENDICES

APPENDIX	TITLE	PAGE
<b>A1</b>	Buffer solution preparations	126
B1	Standard calibration curve for Lowry assay	128
B2	Phylogenetic tree of genus <i>Virgibacillus</i> (16S rRNA gene)	129
В3	Annotated protease sequences of <i>Virgibacillus</i> massiliensis Vm-5	130

#### **CHAPTER 1**

#### INTRODUCTION

#### 1.1 Background of study

Halophilic bacteria has been recognized as one of the extremophiles that has valuable applications in industry and environment (Oren, 2010; Edbeib *et al.*, 2016; Yin *et al.*, 2015). They are found in natural saline and hypersaline habitats such as seawater, salt marshes and lagoon. Occurrence of halophiles can be from seawater to brines (Brock, 1979), some habitats include Dead Sea between Israel and Jordan and also Great Salt Lake in Utah (Oren, 2006). Besides that, salty environments inhabited by halophilic and halotolerant bacteria include food products such as salted fish and fermented food (Enache *et al.*, 2012), and these type of foods are commonly found in Malaysia.

Well-adapted strategies in saline environments utilized by halophilic bacteria made them useful in industrial applications. These halophilic bacteria has been used for production of valuable metabolites and solutes such as stress protectants (DasSarma and DasSarma, 2006), saline wastewater treatments (Shivanand and Mugeraya, 2011) and biodegradation of organic pollutants in environmental biotechnology (Le Borgne *et al.*, 2008). Halophilic bacteria can be classified under different phyla. Under different phylum, halophilic bacteria have different physiological requirements such as compatible solute used and salt concentration required. This diversity makes the halophilic bacteria as one of the source of opportunity and abundance, including industrial enzymes.

One of the enzymes produced by halophilic bacteria is protease, which is a type of hydrolase. Protease can be produced from animal, plant and microbial source. Protease from microbial source has been extensively used in various application especially in detergent industry since 1960 (Rao *et al.*, 1998) due their effectiveness in removing protein stains (Karn and Kumar, 2015). Until today, proteases contributed approximately 60% of the global industrial enzymes market (Anithajothi *et al.*, 2014). While from this amount, microbial proteases constitute 40% of total enzyme production (Raval *et al.*, 2014) which applied in various industries. The largest market undeniable is detergent industry, as this industry contributed to production of 13.5 billion tons per year (Adrio and Demain, 2014).

Apart from that, use of eco-friendly protease recovered from industrial sludge for bio-conversion of proteinaceous waste material into value-added products has become an increasingly concern due to it is a cost effective process (Karn and Kumar, 2015). And also, protease has been engineered using rational design and directed evolution approach to improve its properties and functions to be applied as therapeutic agents and in food processing (Li *et al.*, 2013). Based on huge demand of protease market and its application, new candidate of protease remained a worth for further discovery.

## 1.2 Problem statement / significance of study

Halophilic bacteria produce polyextremophilic enzymes that may have useful application in various biotechnological field. For instance, protease can act as fibrinolytic agent and also removing protein based stains such as blood and sweat effectively (Karn and Kumar, 2015). Most of the commercial bacterial proteases used in detergent industry are produced from Bacillus sp. (Gupta et al., 2002b), lesser investigation on protease from Virgibacillus sp., and until today, no commercial protease is originated from genus Virgibacillus as well. Furthermore, expenditure cost in detergent industry such as purification, production (Niyonzima and More, 2015b) and protein engineering to increase protease efficiency (Li et al., 2013) are expensive. To sort out these problems, a single step of production with the use of crude enzyme is required (Niyonzima and More, 2015a), a more cost effective way compared to purification. Moreover, exploration on novel enzymes with extraordinary properties from extremophiles is always in demand and continuously in research field. Therefore, this study was conducted to characterize extracellular protease produced from a halophilic bacterium, Virgibacillus sp. strain CD6 that is potentially to be applied in various industries, especially in detergent formulation.

#### 1.3 Objectives of study

The objectives of this research are:

- i. To select the best nitrogen source for protease production.
- ii. To assess the effect of physico-chemical factors on the activity and stability of protease from *Virgibacillus* sp. CD6.
- iii. To analyze extracellular protease sequences encoded for Virgibacillus sp.

## 1.4 Scope of study

The previously isolated halophilic bacteria, *Virgibacillus* sp. strain CD6 was initially screened for extracellular protease activity by using qualitative approaches, (skim milk agar and gelatin liquefaction). After that, medium for protease production was formulated and effect of nitrogen sources on protease production was investigated. The optimum conditions of protease activity and its stability in terms of pH, temperature and salt concentration were determined. Then, protease stability in presence of metal ions, inhibitors, detergent constituents and organic solvent was assessed. Compatibility of protease with commercial detergents and substrate specificity of protease were also investigated. Lastly, annotated protein sequences of extracellular proteases of *Virgibacillus* sp. were analyzed using bioinformatics approach and phylogenetic protein tree was constructed.

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