

**SYNTHESIS AND CHARACTERIZATION OF ALUMINUM
DOPED ZINC OXIDE NANOWIRES ON NON-CATALYTIC
SILICA SUBSTRATES**

TASHI DORJI

UNIVERSITI TEKNOLOGI MALAYSIA

2013

**SYNTHESIS AND CHARACTERIZATION OF ALUMINUM
DOPED ZINC OXIDE NANOWIRES ON NON-CATALYTIC
SILICA SUBSTRATES**

TASHI DORJI

A dissertation submitted in partial fulfillment of the
requirements for the award of the degree of
Master of Science (Physics)

**Faculty of Science
Universiti Teknologi Malaysia**

JUNE 2013

To my beloved son, RIGKUEN YESHEY CHABDAG

ACKNOWLEDGEMENT

My sincere and heartfelt gratitude goes to all those individuals who were involved directly or indirectly for the completion of this thesis. In particular; I owe my gratitude to my supervisor, Prof. Dr. Samsudi Sakrani for all the profound guidance, motivations and eminent helps, without which the project wouldn't be as presented. Indeed, it was a great privilege to work together and I am eagerly looking forward to working together again in the near future.

I will always remember the Vacuum laboratory, Physics Department for wonderful working environment, for which the credit goes to Mr. Mohd. Nazari Kamarudin. Also, I am equally grateful to the staff at Nanostructure and Nanophysics and Chemical Laboratory, Ibnu Sina Institute, UTM, Skudai for their support.

Without the scholarship from Royal Government of Bhutan, I wouldn't have made it here to pursue my studies. So, I shall remain grateful to RGoB eternally. Joining UTM was a right choice I made because it is one of the top universities in Malaysia with state of the art facilities which facilitates innovative ideas.

Finally, I would like to thank all the faculty professors, colleagues and other staff for the support, love and care. Last but not the least; I would like to appreciate my parents and wife for all the moral support and above all "KONCHU-SUM" for ensuring good health and success.

ABSTRACT

The undoped and Al-doped ZnO nanostructures were fabricated on the Si (100) substrates via catalyst free thermal evaporation method using a horizontal quartz tube under controlled supply of O₂ gas. The substrate was placed vertically above the source materials unlike the conventional methods. The undoped ZnO nanowires were randomly oriented. When both Al dopant and when doping concentrations were increased, ZnO showed various morphologies in which ZnO changed from randomly orientated nanowires to hexagonal shaped, 'pencil-like' nanorods. Further increase in dopant concentrations beyond 2.4 at% lead to spikey ZnO:Al morphology. The morphology and crystalline structure of ZnO nanostructures were characterized using X-ray diffraction, field emission scanning electron microscopy, scanning electron microscopy and photoluminescence (PL) spectroscopy. ZnO:Al nanorods were found to have diameter roughly between 260 to 350 nm and the length about 720 nm. The as prepared ZnO:Al nanorods also exhibited a strong UV emission. The Al doping concentrations played an important role on the morphology and optical properties of ZnO nanostructures. The significance of the experiment is the simplicity, low cost and fewer necessary apparatus of the process that would suit the high-throughput fabrication of ZnO:Al nanorods. They are expected to have potential applications in functional Si based nanodevices.

ABSTRAK

Nanostruktur ZnO yang didopkan dan tidak didopkan dengan Al di atas substrat Si (100) tanpa penggunaan sebarang pemankin telah dihasilkan melalui kaedah penyejatan terma dengan tiub kuarza digunakan bersama gas oksigen yang dilepaskan melaluinya secara terkawal. Substrat tersebut diletakkan betul-betul di atas bahan asas tidak seperti yang selalu dipraktikkan dalam kaedah konvensional. Tanpa sebarang dopan wayar nano yang dihasilkan adalah secara rawak. Namun, apabila kesemua bahan pendopan Al dilepaskan dan kepekatan dopan meningkat, struktur ZnO yang dihasilkan menunjukkan pelbagai imej dengan struktur tersebut berubah daripada menumbuh secara orientasi rawak kepada struktur yang berbentuk heksagon seperti pensil. Jika kepekatan dopan terus ditingkatkan melebihi 2.4 at%, ZnO:Al akan membentuk struktur seperti paku-paku tajam. Morfologi dan struktur kristal ZnO yang bersaiz nano ini dianalisis ciri-cirinya dengan menggunakan alat XRD, FESEM, SEM dan Photoluminescence (PL). Rod-rod ZnO yang bersaiz nano ini didapati mempunyai jejari sekitar 260 ke 350 nm dan panjangnya mencecah sekitar 720 nm. Rod-rod tersebut juga mempamerkan sinaran UV yang kuat. Kepekatan dopan Al memainkan peranan penting kepada morfologi dan ciri-ciri optikal struktur ZnO. Kepentingan ujikaji ini adalah supaya penghasilan ZnO:Al nanorod dapat dihasilkan dalam kuantiti yang banyak dengan penggunaan kaedah yang mudah, ringkas, berkos rendah dan tidak memerlukan alatan yang banyak. Saintis menjangkakan struktur ini mempunyai potensi yang tinggi dalam alatan-alatan bersaiz nano yang bersasaskan bahan Si.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENTS	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	x
	LIST OF FIGURES	xi
	LIST OF ABBREVIATIONS/SYMBOLS	xiii
1.	INTRODUCTION	1
	1.1 Background of the study	1
	1.2 Research problems	2
	1.3 Research objectives	3
	1.4 Scope of the study	4
	1.5 Significance of the study	4
2.	LITERATURE REVIEW	6
	2.1 Introduction	6
	2.2 Previous studies on aluminum doped ZnO nanowires (ZnO:Al)	6
	2.3 General properties of zinc oxide (ZnO)	11
	2.3.1 Crystal structure of zinc oxide (ZnO)	12
	2.4 General properties of aluminum (Al)	13
	2.5 Thermal evaporation method	15

2.5.1	Vapour Transport Mechanism	16
2.5.2	Catalyst free / self-catalytic growth mechanism of ZnO nanowires	16
2.5.3	Alloying evaporation deposition method (AED)	18
2.5.3.1	Source transformation during the growth of ZnO:Al nanostructures	19
2.5.4	Doping mechanism for ZnO:Al nanostructures	20
2.6	Effects of aluminum doped on ZnO nanowires	21
2.7	Structural characterization of ZnO:Al nanostructures	22
3.	RESEARCH METHODOLOGY	23
3.1	Introduction	23
3.2	Research frame work	23
3.3	Experimental setup	25
3.4	Substrate preparation	28
3.4.1	Cutting substrate	28
3.4.2	Cleaning substrate	29
3.5	Source preparation	30
3.6	Deposition process	30
3.7	Research procedure	30
3.8	Sample characterization	32
3.8.1	Scanning Electron Microscopy (SEM)	33
3.8.1.1	Sample Preparation for SEM	34
3.8.2	Transmission Electron Microscopy (TEM)	35
3.8.3	Field Emission Scanning Electron Microscopy	37
3.8.4	X-Ray Diffraction (XRD)	38
3.8.5	Photoluminescence (PL) Spectroscopy	40
3.8.5.1	Uses of photoluminescence spectroscopy	41
4.	RESULTS AND DISCUSSIONS	43
4.1	Introduction	43
4.2	Structural properties of zinc oxide (ZnO)	44
4.3	Effect of dopant concentration on aluminum doped ZnO (ZnO:Al) nanostructure	48

4.3.1	Structural properties of aluminum doped ZnO (ZnO:Al) nanostructures	49
4.3.2	Photoluminescence properties of aluminum doped ZnO (ZnO:Al) nanostructures	54
4.4	Doping mechanism of ZnO:Al nanostructures	57
5.	CONCLUSION	58
5.1	Introduction	58
5.2	Structural properties of aluminum doped ZnO (ZnO:Al) nanostructures	58
5.3	Photoluminescence properties of aluminum doped ZnO (ZnO:Al) nanostructures	59
5.4	Doping mechanism for ZnO:Al nanostructures	59
5.5	Future works on ZnO:Al nanorods	60
6.	REFERENCES	61

LIST OF TABLES

TABLE NO.	TITL	PAGE
2.1	Physical properties of wurtzite ZnO	12
2.2	Physical properties of aluminum	14
4.1	Varying dopant concentrations at constant temperature, growth time and flow rate of O ₂	49

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
2.1	Wurtzite structure of ZnO	13
2.2	The self-catalytic growth model of ZnO nanowire	17
2.3	Schematic illustration of different steps of source transformation during the growth of ZnO nanorods	19
3.1	Research frame work chart	24
3.2	Schematic experimental setup for synthesis of aluminum doped ZnO (ZnO:Al)	27
3.3	Experimental setup for synthesis of aluminum doped ZnO (ZnO:Al) Vacuum Lab, UTM	28
3.4	Ultrasonic bath, Vacuum Lab, UTM	29
3.5	Graphical representation of deposition process	31
3.6	Schematic of SEM showing the working parts	34
3.7	Scanning Electron Microscopy (SEM), Ibnu Sina Institute, UTM	35
3.8	Schematic of TEM showing the working parts	36
3.9	FESEM installed in Ibnu Sina Institute, UTM	38
3.10	Reflection of X-rays from two planes of atoms	39
3.11	Methods of photoluminescence	42
3.12	Photoluminescence spectroscopy, Faculty Lab, UTM	42
4.1	XRD pattern of ZnO nanowire synthesized on Si substrate	45
4.2	SEM images of randomly oriented mesh of ZnO nanowires synthesized on silicon substrates (a) low magnification and (b)high magnification	46

4.3	FESEM images of undoped ZnO nanowires synthesized on Si substrate, (a-b) Surface view, (a) Low magnification, (b) High magnification, (c-d) Cross-sectional view (c) Low magnification and (d) High magnification	46
4.4	(a) Detection position of EDAX spectra of the ZnO nanowires sample (b) EDAX spectra of ZnO nanowires on Si substrate with inset showing element composition in a table. (c) Image of element mapping of the sample and (d) EDAX spectra of the mapping with inset showing element composition in percentage	48
4.5	Comparative SEM images of undoped and Al doped ZnO nanowires. (a) 0 at% Al, (b) 0.6 at% Al, (c) 1.2 at% Al, (d) 2.4 at% Al, (e) 4.7 at% Al and (f) 11.3 at% Al	50
4.6	SEM images of Al doped ZnO nanowires. (a-b) 1.2 at% Al low and high magnification, (c-d) 2.4 at% Al, low and high magnification	53
4.7	FESEM images of Al doped ZnO nanowires. (a-b) 1.2 at%, (a) Surface view with inset showing high magnification, (b) Cross sectional view with inset showing high magnification, (c-d) 2.4 at% (c) Surface view with inset showing high magnification, (d) Cross sectional view with inset showing high magnification	53
4.8	(a) Detection position of EDAX spectra of 2.4 at% Al doped ZnO:Al nanorods sample (b) EDAX spectra of ZnO:Al nanorods, the inset showing element composition % in a tabular form, (c) image of element mapping of the sample and (d) EDAX spectra of the mapping with inset showing element composition in mass and atomic percentage	54
4.9	Photoluminescence spectra of the as-synthesized ZnO:Al nanowires on silicon substrate showing intensity versus wavelength	55
4.10	Photoluminescence spectra of the as-synthesized ZnO:Al nanowires on a silicon substrate showing intensity versus energy	56

LIST OF ABBREVIATIONS / SYOMBOLS

a, a_0	-	Lattice parameter
Å	-	Angstrom
Al ₂ O ₃	-	Aluminate
Ar	-	Argon
Au	-	Gold
c, c_0	-	Lattice parameter
C	-	Carbon/Graphite
cm ³	-	Cubic centimeter
cm ² /Vs	-	Centimeter square per volt second
CO	-	Carbon monoxide
CO ₂	-	Carbon dioxide
CROs	-	Cathode ray oscilloscopes
CVD	-	Chemical Vapour Deposition
°C	-	Degree Celsius
d	-	Interatomic spacing
D	-	Dimension
eV	-	Electron volt
EDS	-	Energy Dispersive Spectroscopy
EDAX	-	Energy Dispersive Analysis of X-rays
E _g	-	Energy band gap
FESEM	-	Field Emission Electron Microscope
FTO	-	Fluorine doped tin oxide
g	-	Gram
g/cm ³	-	Gram per cubic centimeter (density)

h	-	Hour
HCl	-	Hydrochloric acid
HF	-	Hydrofluoric acid
H ₂ O	-	Water
HRTEM	-	High-resolution transmission electron Microscope
ICDD	-	International Centre for Diffraction Data
IOSCs	-	Inverted organic solar cells
ITO	-	Indium tin oxide
K	-	Kelvin scale
kJ mol^{-1}	-	Kilo joule per mole
λ	-	Lambda (Wavelength)
μ	-	Micron (10^{-6} m)
m	-	Metre
MRs	-	Microrods
mM	-	Molar mass
MOCVD	-	Metalorganic chemical vapor deposition
MOVPE	-	Metal-Organic Vapor Phase Epitaxy
meV	-	Milli electron volt (10^{-3} eV)
n	-	Wave number, Number of moles
NC	-	Nano-crystals
Ne	-	Neon gas
nm	-	Nanometer (10^{-9} m)
N ₂ O	-	Nitrous oxide
NPs	-	Nanoplates
NSs	-	Nanostructures / nanosheets
NTs	-	Nanotubes
$\text{Na}_2[\text{Zn}(\text{OH})_4]$	-	Sodium zincate
NaOH	-	Sodium hydroxide (base)
NWs	-	Nanowires
O	-	Oxygen atom
Ωcm^{-1}	-	Omega per centimeter (resistivity)
PET	-	Polyethylene terephthalate
PL	-	Photoluminescence

Q1D	-	Quasi-one-dimensional
RF	-	Radio frequency
sccm	-	Standard cubic centimeter per minute
SEM	-	Scanning Electron Microscope
Si	-	Silicon
SnO ₂	-	Tin oxide
SPM	-	Scanning probe microscope
T	-	Temperature
TCO	-	Transparent conducting oxide
θ	-	Theta (Diffraction angle)
UV	-	Ultraviolet radiation
VLS	-	Vapor-liquid-solid
VS	-	Vapour Solid
XRD	-	X-ray diffraction
Zn(C ₅ H ₇ O ₂) ₂ .xH ₂ O	-	Zinc acetylacetonate hydrate
ZnCl ₂	-	Zinc chloride
ZnO	-	Zinc Oxide
ZnO:Al	-	Aluminum doped zinc oxide
ZnS	-	Zinc sulfide

CHAPTER 1

INTRODUCTION

1.1 Background of the Study

The ease of life we enjoy can be attributed to achievements in science and technology to a very large extent. For that matter, nanotechnology has become the seed of peace and tranquility of the modern world. It has stimulated great interest due to their importance in basic scientific research and potential technological applications [1]. Zinc oxide (ZnO) is an interesting and a well-known wide band gap II-VI semiconductor with a direct band gap of ~ 3.3 eV with large exciton binding energy (60 eV). The immense excitement in this area of research arises from understanding the fact that ZnO gives rise to new phenomena and multi-functionality which ultimately leads to unprecedented integration density with nanometer-scale structures [2].

One dimensional (1D) zinc oxide nanostructures (NSs) have been synthesized by various methods [3]. Nanotechnology is an intensively and extensively pursued topic of the 21st century. The influence on the electronic and optical properties due to doping of ZnO has been often reported [4]. However, there are very few studies done on the topic and only limited data is available. The influence of dopants on the

formation of doped 1D ZnO NSs would indeed contribute to a better understanding of their growing mechanisms.

Curiosity and need to understand the novel physical properties of one-dimensional nanoscale has been the driving force behind the synthesis and characterization of nanowires (NWs). Undeniably, there is a wide area of applications. As the ever growing research continues, new findings are leading to new applications. The unique structural, electrical, magnetic and optical properties of ZnO NWs is due to its extremely thin nanocrystal (NC) structure which contributes to play a predominant role in the fabrication of sophisticated electronic devices. Nanowires (NWs), nanobelts (NBs), nanorods (NRs), nanotubes (NTs), nanoplates (NPs) and nanohelices (NHs) are all sister nanostructures (NSs) which can be formed from ZnO synthesis under certain conditions. According to Kwon *et al.* [5], NWs with different compositions has been explored using various methods. Various growth mechanisms have also evolved over time. The details of growth methods will be discussed later. Since the structural, optical, magnetic and electrical properties of ZnO are dependent on growth parameters, hence their applications. So, the prime interest here is to synthesize catalyst free doped ZnO and learn the influence of dopant concentrations on the structural and optical properties. Over the time, researchers have used various dopants to dope ZnO NSs.

1.2 Statement of Problem

Intrinsic semiconductor is electrically neutral and not much of use. However the physical properties of semiconductors can be effectively manipulated by impurity doping. Recently, doping of nanostructures has become an important issue for the more diverse range of applications [6]. So far as we know, there are very few reports on Al-doped 1D ZnO nanowires synthesis particularly via the method we have adopted. Therefore, my focus here is to synthesize Al-doped ZnO nanowires via

thermal oxidation of Zn at different dopant concentrations and understand its influence on the structural and optical properties of ZnO:Al.

In particular, catalyst free synthesis of aluminum doped zinc oxide nanowires by placing the substrate vertically above the source is a rarely attempted method. So, it is indeed an opportunity to explore this method and compare its efficiency with other methods. In the conventional systems, source and substrate lie at the same horizontal level separated by certain distance in between. Also, reduction of ZnO with graphite has been a common method unlike the current attempt.

1.3 Research Objectives

In general, aluminum doped ZnO NWs will be synthesized on silicon wafer using high purity zinc powder, aluminum powder and oxygen along with argon as carrier gas. The structural properties and optical properties under various dopant concentrations will be investigated. The following objectives will be under focus in this research;

- a) To synthesize aluminum doped ZnO (ZnO:Al) nanowires by a thermal evaporation method.
- b) To determine the influence of dopant concentrations on structural and optical properties of ZnO nanowires.
- c) To establish the mechanism leading to the growth of ZnO:Al nanowires.

1.4 Scope of the Study

Most of the previous works being devoted on undoped ZnO, there has been very little effort devoted to investigate the doped counterparts [7]. So, it is an opportunity and little concerted effort from my part to explore little further. Doping is a powerful tool to tailor the electrical and optical properties, facilitating the construction of many electronic and optoelectronic devices. Additionally, when doped with Al, it behaves as an acceptor in ZnO with its energy level locating at 0.1 eV below the bottom of the conduction band, making itself a good candidate for creating a p-type ZnO [8]. Similarly, Al doped ZnO nanostructure is found to be one of the best n-type thermoelectric oxide. When doped with Al, the fraction of Al is expected to influence the morphologies of ZnO:Al NWs. In recent years, researchers have shifted their focus on the doped NSs because doping has brought about enhanced conductivity and optical behaviour of the NSs. Likewise, when we choose a dopant, it must possess close lattice matching with ZnO which is satisfied by group III elements which includes Al. As a consequence of limited time, we cannot experiment with dopant other than Al.

Emphasis here will be to synthesize and characterize the structural properties of aluminum (Al) doped ZnO with dopant concentrations ranging from 0.6 at% to 11.3 at% as stated in the research objectives. Also, the process will be completed without use of catalyst unlike many popular methods. Thus, there awaits a huge appetite for pure Al doped ZnO NWs.

1.5 Significance of the Study

Expansion of nano research over the past decades clearly indicates the vitality of nanotechnology for the benefit of mankind. The impacts can be felt in every

sphere of our life from economy, politics, entertainment, recreation, space exploration, communication, transportation, etc. Due to their unique density of electronic states, NWs in the limit of small diameters are expected to exhibit significantly different optical, electrical and magnetic properties from their bulk 3-D crystalline counterparts. It is mainly attributed to the extremely thin nanocrystals of ZnO. Nanowires, compared to other low dimensional systems, have two quantum confined directions, while still leaving one unconfined direction for electrical conduction.

The study on structural characterization of Al doped ZnO with respect to various dopant concentrations will hopefully contribute in its own manner towards nanotechnology and its pool of knowledge. Although it appears insignificant at the initial stage; however it has a large scope in fulfilling the recent surge for demand and knowledge of Al doped nanomaterials. We are optimistic that it will have a profound impact the world over and least to my needs.

REFERENCES

- [1] A. Janotti and C. G. V. Walle. Fundamentals of zinc oxide as a semiconductor. *Reports on Progress in Physics*. 2009, volume 72(1): 148-154.
- [2] Y. Cui, Z. Zhong, D. Wang, W. U. Wang and C. M. Lieber. High performance transistors. *Nano Letter*. 2003, volume 3(2): 149-152.
- [3] L. M. Charles and W. L. Zhong. Functional Nanowires Serving the International, *Material Research Community*. 2007, volume 32: 217-230.
- [4] P. O. Amin. *Structural and Optical Properties of Randomly Oriented ZnO Nanowire*: Master Thesis, Universiti Teknologi Malaysia; 2012.
- [5] B. J. Kwon, K. M. Lee, H. Y. Shin, J. Kim, J. Liu, S. Yoon, S. Lee, Y. H. Ahn and J. Y. Park. Synthesis of vertical arrays of ultra-long ZnO nanowires on noncrystalline substrates. *Materials Science and Engineering*. 2012, volume 177: 132-139.
- [6] X. S. Fang, C. H. Ye, L. D. Zhang, Y. Li and Z. D. Xiao. Formation and Optical Properties of Thin and Wide Tin-doped ZnO Nanobelts. *Chemistry Letters*. 2005, volume 34(3):436-437.
- [7] L. Wang. Zinc oxide nanostructures: growth, properties and Applications. *Journal of Physics: Condensed Matter*. 2004, volume 16: 830-845.
- [8] Y. Kanai. Admittance Spectroscopy of Cu-Doped ZnO Crystals. *Japan Journal of Applied Physics*. 1991, volume 30: 703-707.
- [9] L. J. Lauhon, M. S. Gudiksen, and C. M. Lieber. Semiconductor nanowire heterostructures. *Philosophical Transactions of the Royal Society of London*. 2004, volume 362: 1247–1260.

- [10] X. Fan, G. Fang, S. Guo, N. Liu, H. Gao, P. Qin, S. Li, H. Long, Q. Zheng and X. Zhao. Controllable synthesis of flake-like Al-doped ZnO nanostructures and its application in inverted organic solar cells. *Nanoscale Research Letters*. 2011, volume 6: 546.
- [11] H. B. Zhou, H. Y. Zhang, Z. G. Wang and M. L. Tan. Highly conductive and transparent Al-doped ZnO films on glass substrate via incorporating hydrogen at low substrate temperatures. *Materials Letters*. 2012, volume 74: 96–99.
- [12] D. G. Ping, L. Wang, F. M. Gong, C. Nan, F. Xue, W. Y. Qun and Y. M. Ming. Synthesis of tetrapod-shaped ZnO whiskers and microrods in one crucible by thermal evaporation of Zn/C mixtures. *Transitions of Nonferrous Metals Society of China*. 2008, volume 18: 158-161.
- [13] T. J. Kuo, C. N. Lin, C. L. Kuo, and M. H. Huang. Growth of Ultralong ZnO Nanowires on Silicon Substrates by Vapor Transport and Their Use as Recyclable Photocatalysts. *Chemistry Materials*. 2007, volume 19: 5143-5147.
- [14] J.T. Chen , J. Wang, R.F. Zhuo, D. Yan, J.J. Feng, F. Zhang and P.X. Yan. The effect of Al doping on the morphology and optical property of ZnO nanostructures prepared by hydrothermal process. *Applied Surface Science*. 2009, volume 255: 3959–3964.
- [15] C. M. García, E. D. Valdés, A. Ma. P. Mercado, A. F. M. Sánchez, J. A. A. Adame, V. Subramaniam, J. R. Ibarra. Synthesis of Aluminum-doped Zinc Oxide Nanowires hydrothermally grown on plastic substrate. *Advances in Materials Physics and Chemistry*. 2012, volume 2: 56-59.
- [16] S. N. Bai, H. H. Tsai and T. Y. Tseng. Structural and optical properties of Al-doped ZnO nanowires synthesized by hydrothermal method. *Thin Solid Films*. 2007, volume 516(2–4):155–158.
- [17] J. Yu, B. Huang, X. Qin, X. Zhang, Z. Wang and H. Liu. Hydrothermal synthesis and characterization of ZnO films with different nanostructures. *Applied Surface Science*. 2011, volume 257: 5563–5565.
- [18] J. J. Wu and S. C. Liu. Catalyst-Free Growth and Characterization of ZnO Nanorods. *Journal of Physics and Chemistry. B*. 2002, volume 106: 9546-9551.
- [19] C. L. Hsu, S. J. Chang, H. C. Hung, Y. R. Lin, C. J. Huang, Y. K. Tseng and I. C. Chen. Well-Aligned, Vertically Al-Doped ZnO Nanowires Synthesized on ZnO:Ga/Glass Templates. *Journal of Electrochemical Society*. 2005, volume 152(5):G378-G381.

- [20] R. Deng, Y. Zou, Z. Chen, G. Jiang and H. Tang. Growth of single-crystalline Al:ZnO nanorods by simple vapor deposition method. *Materials Letters*. 2007, volume 61(17): 3582-3584.
- [21] W.I. Park, D.H. Kim, S.W. Jung and G.C. Yi. Metalorganic vapor-phase epitaxial growth of vertically well-aligned ZnO nanorods. *Applied Physics Letters*. 2002, volume 80(22): 4232.
- [22] Y. Ortega, D. Haussler, J. Piqueras, P. Fernandez and W. Jager. Complex hierarchical arrangements of stacked nanoplates in Al-doped ZnO. *Physica Status Solidi A*. 2012, volume 209(8): 1487–1492.
- [23] S. C. Erwin, L. Zu, M. I. Haftel, A.L. Efros, T. A. Kennedy and D.J. Norris. Doping semiconductor nanocrystals. *Nature*. 2005, volume 436: 91-94.
- [24] S. Ilcan, M. Caglar and Y. Caglar. Sn doping effects on the electro-optical properties of sol gel derived transparent ZnO films. *Applied Surface Science*. 2005, volume 256: 7204-7210.
- [25] J. Gao, Q. Zhao, Y. Sun, G. Li, J. Zhang and D. Yu. A Novel Way for Synthesizing Phosphorus-Doped ZnO Nanowires. *Nanoscale Research Letter*. 2011, volume 6: 45-46.
- [26] H.Y. Dang, J. Wang and S.S. Fan. The synthesis of metal oxide nanowires by directly heating metal samples in appropriate oxygen atmospheres. *Nanotechnology*. 2003, volume 14:738–741.
- [27] H. Y. Dang, J. Wang and S. S. Fan. The synthesis of metal oxide nanowires by directly heating metal samples in appropriate oxygen atmospheres. *Nanotechnology*. 2003, volume 14:738–741.
- [28] Y. Ortega, P. Fernandez and P. Piqueras. Growth and luminescence of oriented nanoplate arrays in tin doped ZnO. *Nanotechnology*. 2007, volume 18: 2-3.
- [29] R. C. Wang, C. P. Liu, J. L. Huang and S. J. Chen. Single-crystalline Al:ZnO nanowires/nanotubes synthesized at low temperature. *Applied Physics Letters*. 2005, volume 86: 251104.
- [30] C. P. Liu. *Multi-channel ZnO nanoconductors with tunable opto-electrical properties*. National Cheng Kung University, Tainan, Taiwan.
- [31] D. J. Lee, H. M. Kim, J. Y. Kwon, H. Choi, S. H. Kim and K. B. Kim. Structural and Electrical Properties of Atomic Layer Deposited Al-Doped ZnO Films. *Advanced Functional Materials*. 2011, volume 21:448-455.

- [32] K. L. Chen, F. Y. Hung, Y. T. Chen, S. Y. Chang and Z. S. Hu. Surface Characteristics, Optical and Electrical Properties on Sol-Gel Synthesized Sn-Doped ZnO Thin Film. *Materials Transactions*. 2010, volume 51 (7) 1340-1345.
- [33] S. Guozhen, J. H. Cho, Jung, S., Lee, C. J. Synthesis and characterization of S-doped ZnO nanowires produced by a simple solution-conversion process. *Chemical Physics Letters*. 2005, volume 401: 529–533.
- [34] L. Yang. *Synthesis and characterization of ZnO nanostructures*. Department of Science and Technology, Linköping University, Sweden.
- [35] O. Lupan, G. A. Emelchenko, V. V. Ursaki, G. Chai, A. N. Redkin, A. N. Gruzintsev, I. M. Tiginyanu, L. Chow, L. K. Ono, B. Roldan Cuenya, H. Heinrich and E. E. Yakimov. Synthesis and characterization of ZnO nanowires for nanosensor applications. *Materials Research Bulletin*. 2010, volume 45:1026-1032.
- [36] S. Lin, H. Tang, Z. Ye, H. He, Y. Zeng, B. Zhao and L. Zhu. Synthesis of vertically aligned Al-doped ZnO nanorods array with controllable Al concentration. *Materials Letters*. 2008, volume 62: 603–606.
- [37] J. Yu, B. Huang, X. Qin, X. Zhang, Z. Wang and H. Liu. Hydrothermal synthesis and characterization of ZnO films with different Nanostructures. *Applied Surface Science*. 2011, volume 257: 5563–5565.
- [38] S. Yun, J. Lee, J. Yang and S. Lim. Hydrothermal synthesis of Al doped ZnO nanorods arrays on Si substrate. *Physical B*. 2010, volume 405:413–419.
- [39] T. J. Kuo, C. N. Lin, C. L. Kuo, and M. H. Huang. Growth of Ultralong ZnO Nanowires on Silicon Substrates by Vapor Transport and Their Use as Recyclable Photocatalysts. *Chemistry Materials*. 2007, volume 19: 5143-5147.
- [40] Transmission and scanning electron microscopy, accessed 25 May 2013, <<http://www.expertsmind.com/topic/looking-at-microbes/transmission-and-scanning-electron-microscopy-92282.aspx>>