



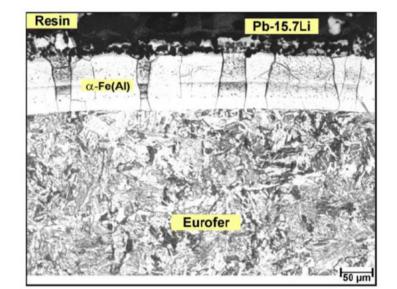
Institute for Applied Materials Material Process Technology (IAM-WPT)

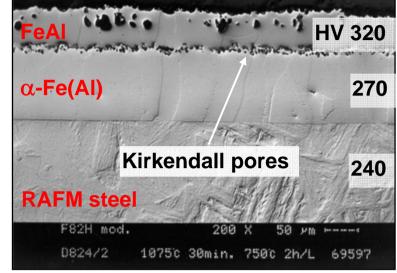
# Influence of deposition conditions on the microstructure of Al-based coatings for applications as corrosion and anti-permeation barrier

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## **Motivation**

Previous research revealed that the application of aluminum-based barriers is suitable to minimize corrosion rates of





**Reference scale made by** HDA after heat treatment (1040°C/0.5 h + 750°C /1 h)

Eurofer steel in Pb-15.7Li (picture to the right) and tritium-permeation from the liquid breeder into the cooling system (HCLL) in an envisaged future fusion reactor.

Developed deposition techniques to electroplate aluminum are based on water-free electrolytes. Either ECA or ECX process can provide several advantages compared to the established Hot-Dip Aluminization (HDA) process:

- Better thickness distribution
- **Controllable coating thicknesses**
- Reduced amounts of aluminum applicable (low activation)

However, the main challenge is to convert the electroplated Al-coatings during a subsequent heat treatment, to enable the formation of protective scales on Eurofer for applications as corrosion and anti-permeation barriers in Pb-15.7Li.

HDA coated sample after exposure in flowing Pb-15.7Li, taken from [1]

# **Electrodeposition of Aluminum**

## Aluminum is highly electronegative ( $E_0 = -1, 6$ V vs. NHE)

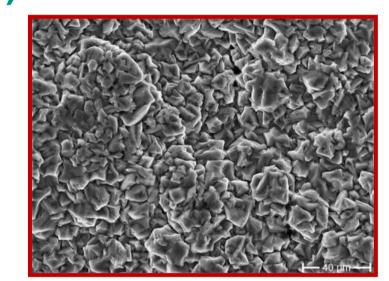
- Al cannot be deposited from aqueous electrolytes
- Water-free / aprotic electrolytes are needed

# **Principle types of electrolytes to electrodeposit**

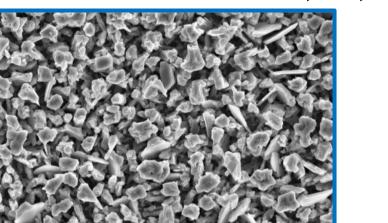
- Al at moderate temperatures (RT-100°C)
- Toluene-based electrolytes  $\rightarrow$  ECA process
- 2. Ionic liquids  $\rightarrow$  ECX process

### **Electrochemical processes have in common**

- No reaction with the substrate material
- Adjustable AI thickness (time, current density)
- Good adherence to the substrate
- Industrial relevance is given



AI-Surface ECA-plated 10 mA/cm<sup>2</sup>, 1 h, 100°C

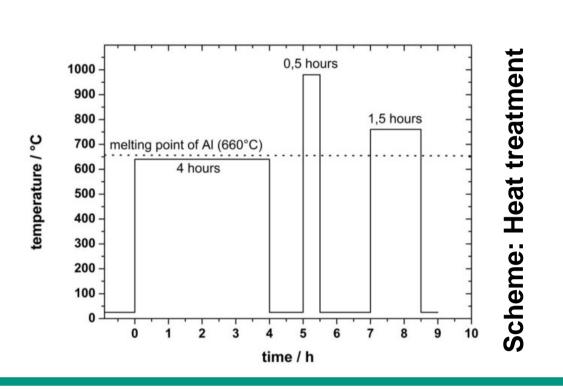


**AI-Surface ECX-plated** 

20 mA/cm<sup>2</sup>, 0.5 h, 100°C

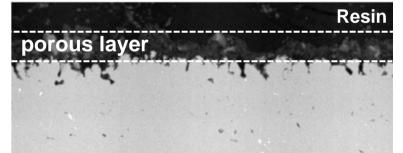
#### **Experimental**

- Heat treatment (HT) procedure according to [1]
  - 640°C, 4 h  $\rightarrow$  980°C, 0.5 h  $\rightarrow$  760°C, 1.5 h
- In flowing argon atmosphere
  - Reduction of oxidation effects
- Samples cooled down under flowing Ar

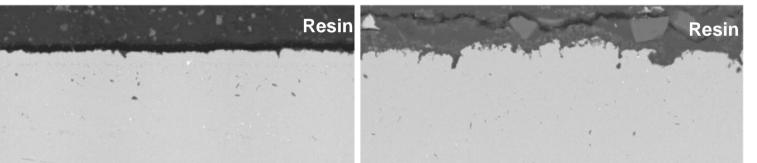


**P2** 

## **DC plated sample after HT**



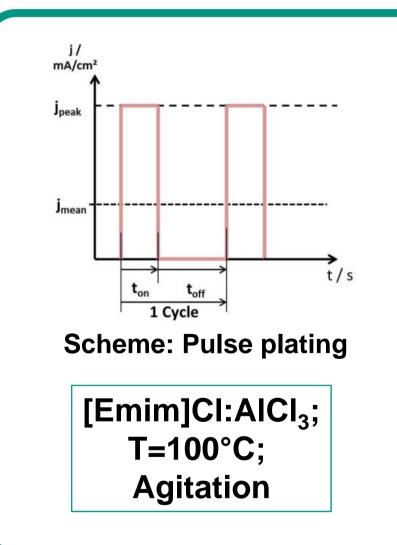
## **Pulse plated samples after HT**



#### Why ECX?

- Process is more flexible than ECA process
  - > Deposition parameters adjustable in a wider range (T, j, pulse plating,...)
- Electrolytes not oxygen sensitive  $\rightarrow$  non-inflammable

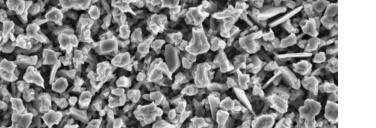
# **Electrodeposition of AI by ECX**

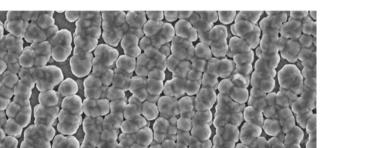


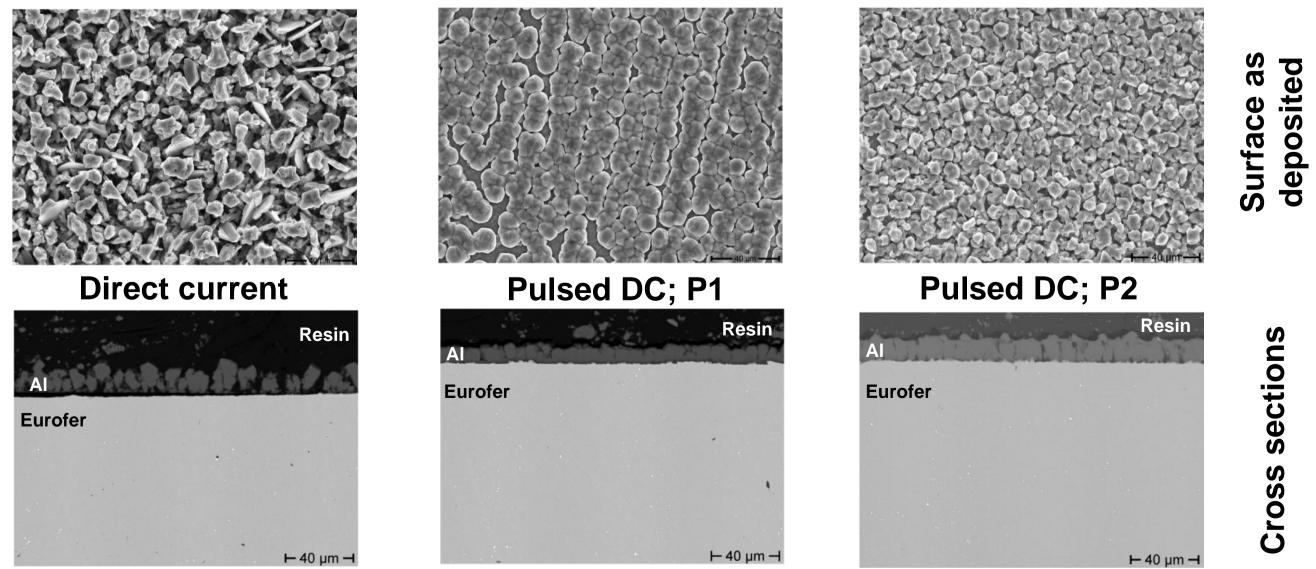
<ul> <li>From an imidazolium-based ionic liquid</li> <li>Direct current (DC) plated samples were compared to pulse plated (P) samples → j<sub>mean</sub> and t were held co</li> <li>Theoretical AI thickness should be about 12.5 µm</li> </ul>					
<ul> <li>Direct current (DC) plated samples were compared t</li> </ul>					
From an imidazoilum-based ionic ilquid					
Electrodeposition of AI on Eurofer steel samples by					

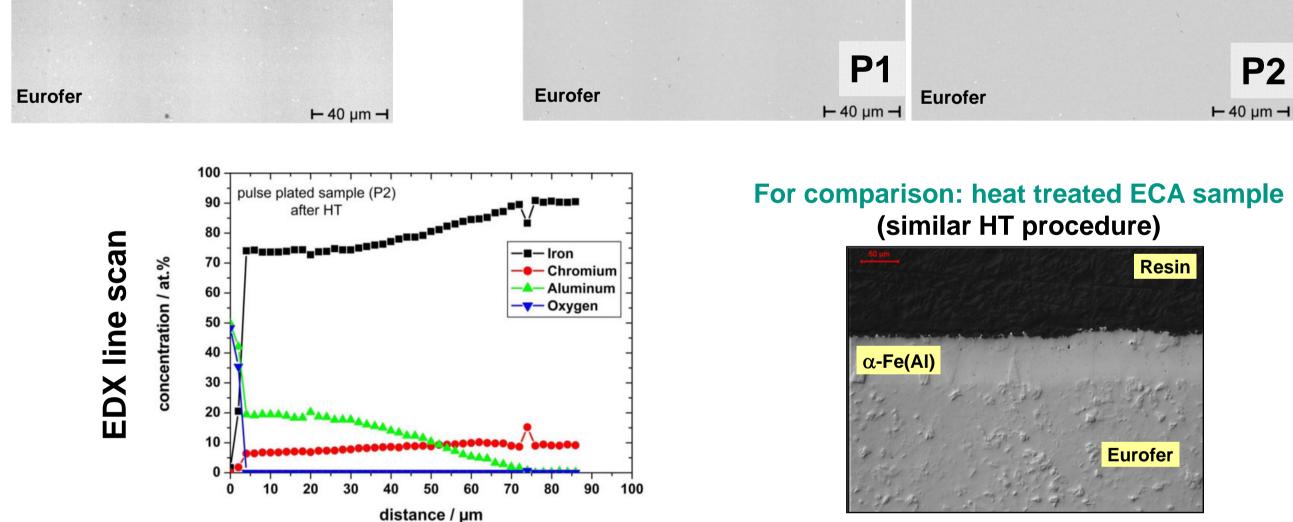
Parameter	DC	P1	P2
j <sub>mean</sub>		20 mA/cm <sup>2</sup>	
j <sub>peak</sub>	-	80 mA/cm <sup>2</sup>	25 mA/cm <sup>2</sup>
Duty Cycle	100%	25%	80%
t	30 minutes		

# Influence of deposition parameters









Heat treatment behavior

picture taken from [1]

#### Results

Quality depends on surface morphology of Al-layers on Eurofer after electroplating

Pulse plated (P2) sample after HT

- Pulse plated samples by ECX: uniform formation of Al-based scales, comparable to scales produced by ECA (see [1])
- Rough and jagged surface structure (in this case DC plated): formation of relatively thick porous, oxygen containing layer on top of the sample
- Al amount in the reacting zone of ca. 20 at.% (decreasing with distance from surface)

## Conclusions

ECX process show industrial relevance and deposit AI scales on Eurofer steel samples with controllable and reproducible thickness

Surface morphology is strongly influenced by deposition conditions

- DC plated sample : Jagged and rough surface; crystallites not grown together Pulse plated samples: Dense, even and homogeneous morphology
- Morphologies of AI coatings made by ECX process depend on deposition parameters Al layers have to be converted into protective scales by heat treatments (HT) HT behavior depends on the morphology of the coating produced by ECX Formation of Fe-Al scales during HT is influenced by morphology Rough and jagged surface structures cannot be recommended for further HT Even, dense and homogeneous morphologies are recommended for HT Heat treated samples with recommended morphology (pulse plated) were comparable to known ECA or HDA produced Al-based scales
  - The more flexible ECX process has proven to be a promising alternative to ECA

#### **References:**

[1] Konys, J. et al.: Impact of heat treatment on surface chemistry of Al-coated Eurofer for application as anti-corrosion and T permeation barriers in flowing Pb-15.7 environment, Fus. Eng. Des., 87 (2012), 1483.

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