

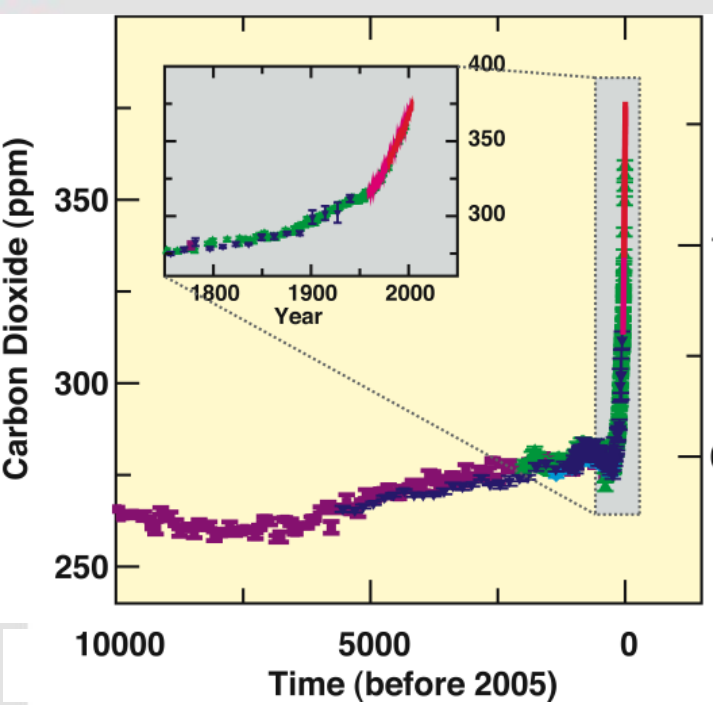
The role of the biosphere in air quality and climate

Klaus Butterbach-Bahl

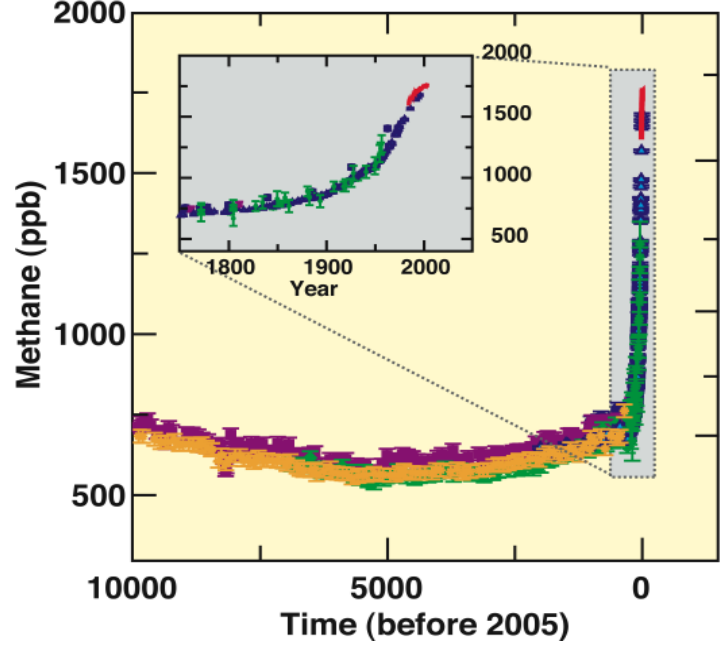
Institute for Meteorology and Climate Research

Garmisch-Partenkirchen, Germany

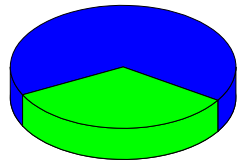
Why to study biosphere-atmosphere interactions when talking about climate change?



Radiative Forcing ($W m^{-2}$)



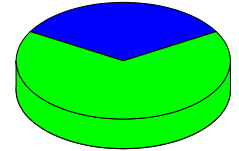
Fossil fuel burning



Land use change

CO₂

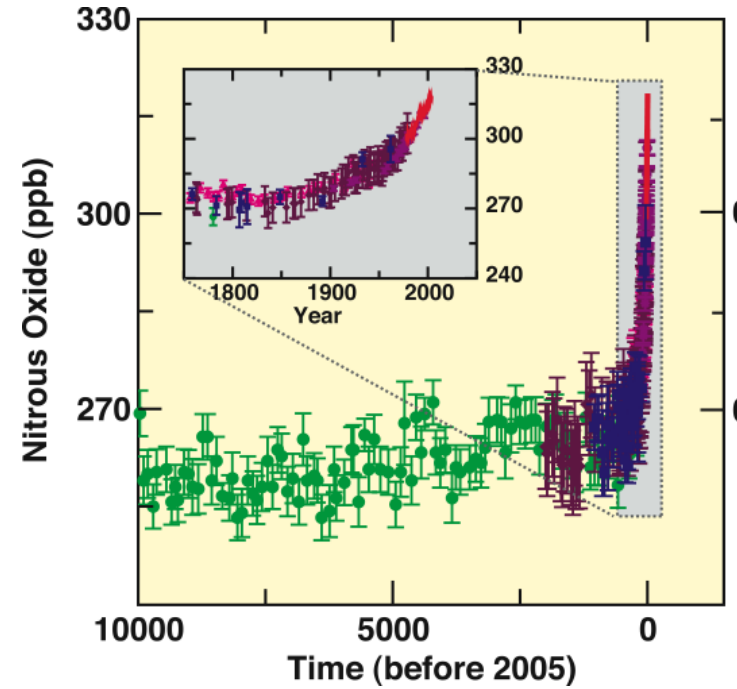
Industrial sources



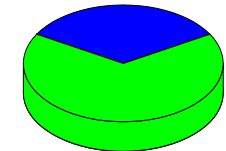
Rice paddies, wetlands, ruminants

CH₄

Radiative Forcing ($W m^{-2}$)



Industrial sources

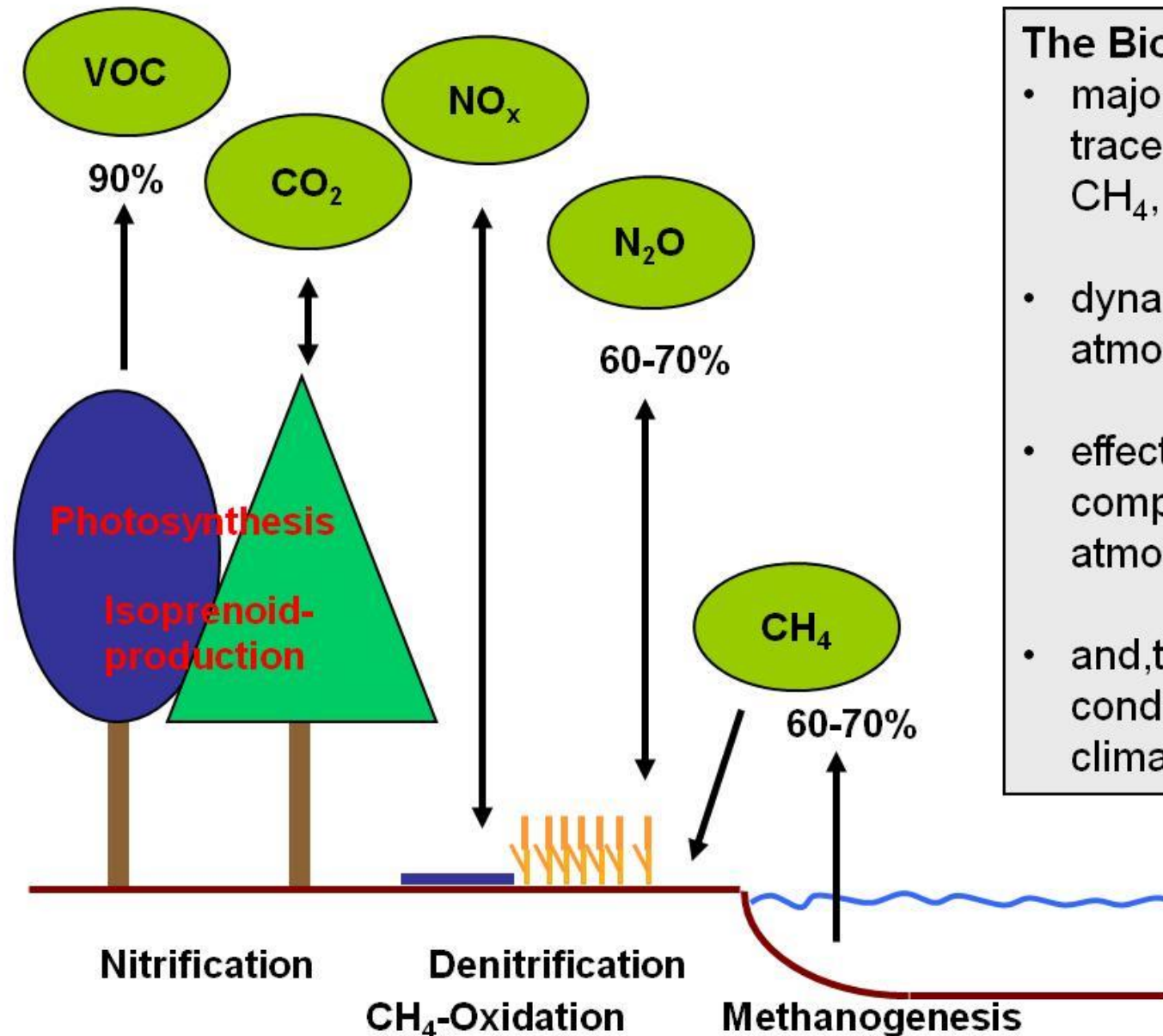


Agriculture, forests, oceans

N₂O

Soils: 60-70%

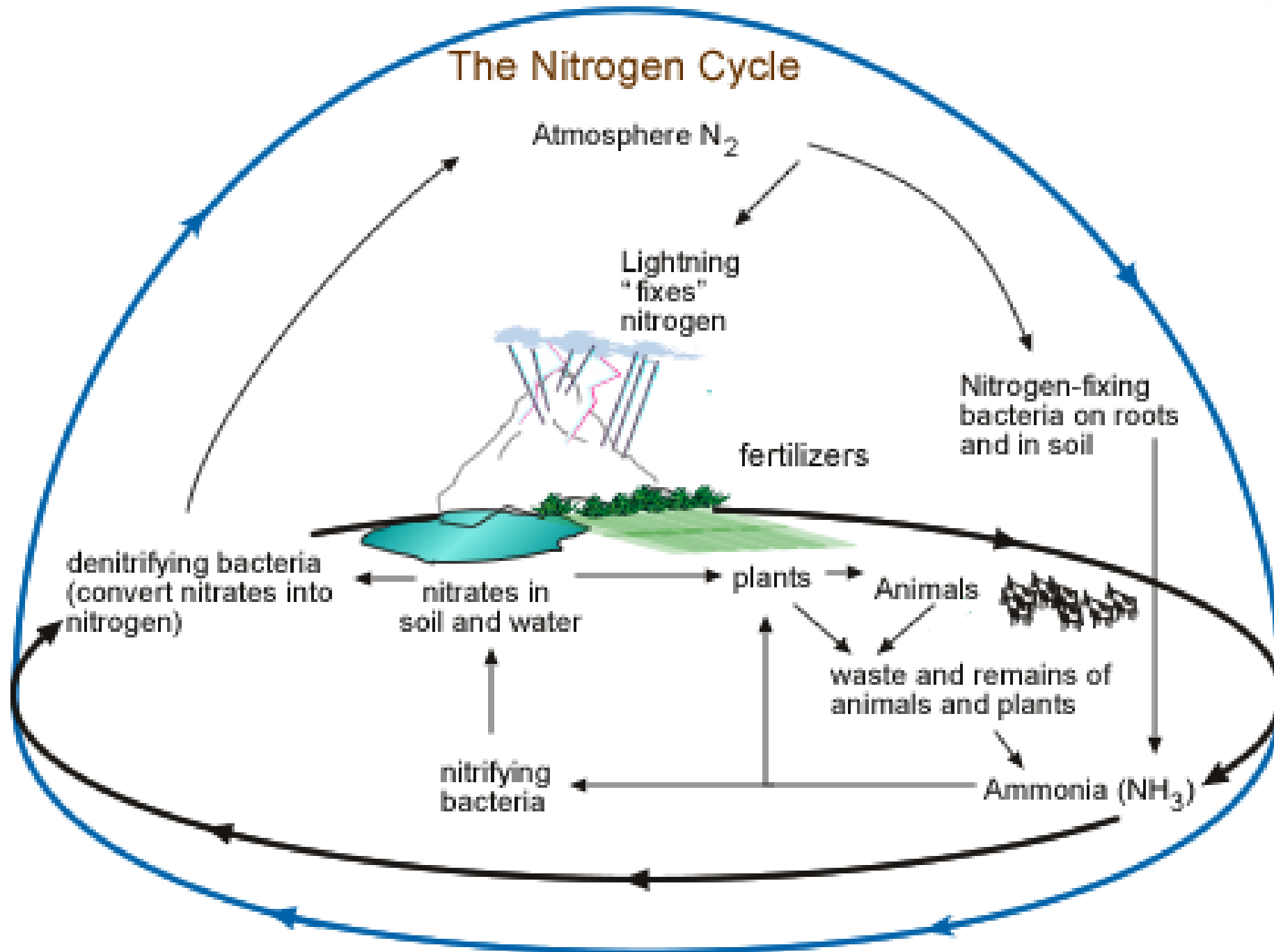
The biosphere as source and sink of trace gases



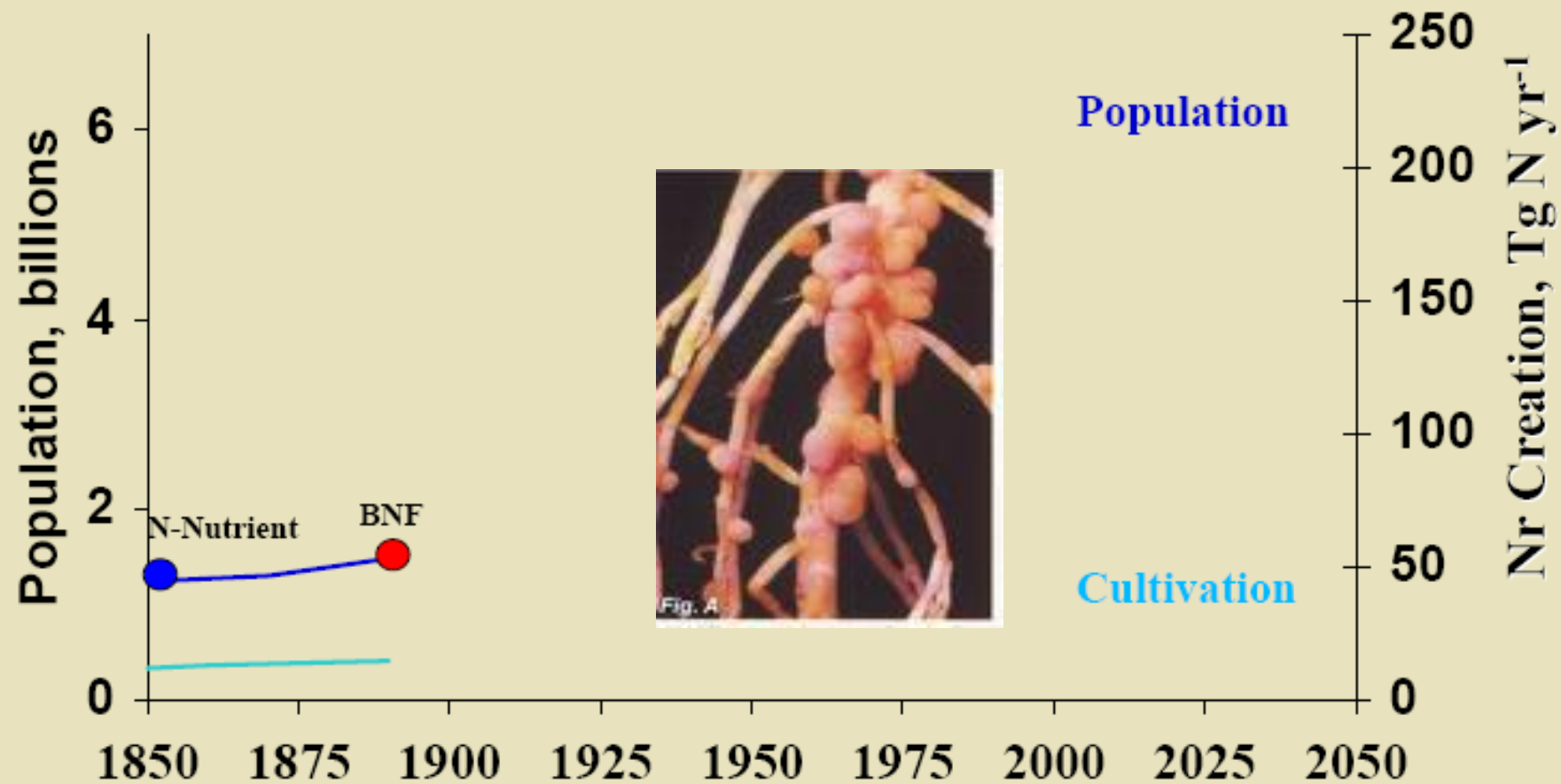
The Biosphere

- major source/ sink for trace substances (N₂O, CH₄, NO_x, CO₂, VOC)
- dynamic exchange with atmosphere
- effects chemical composition of the atmosphere
- and, thus, environmental conditions on earth (e.g. climate and air pollution)

The Nitrogen Cycle

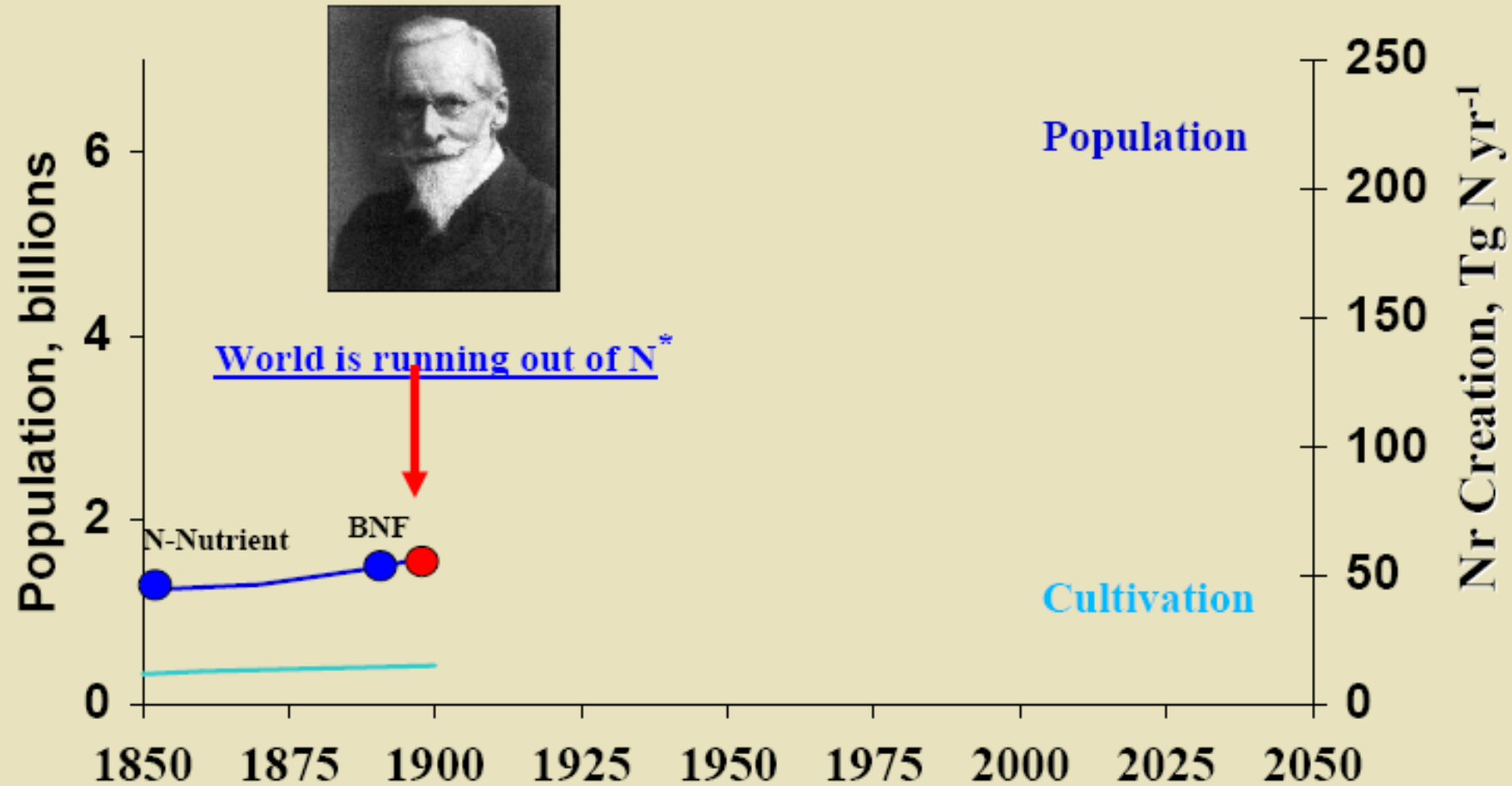


Timeline of Global Reactive N Creation by Human Activity 1850 to 2000



Galloway et al., 2003

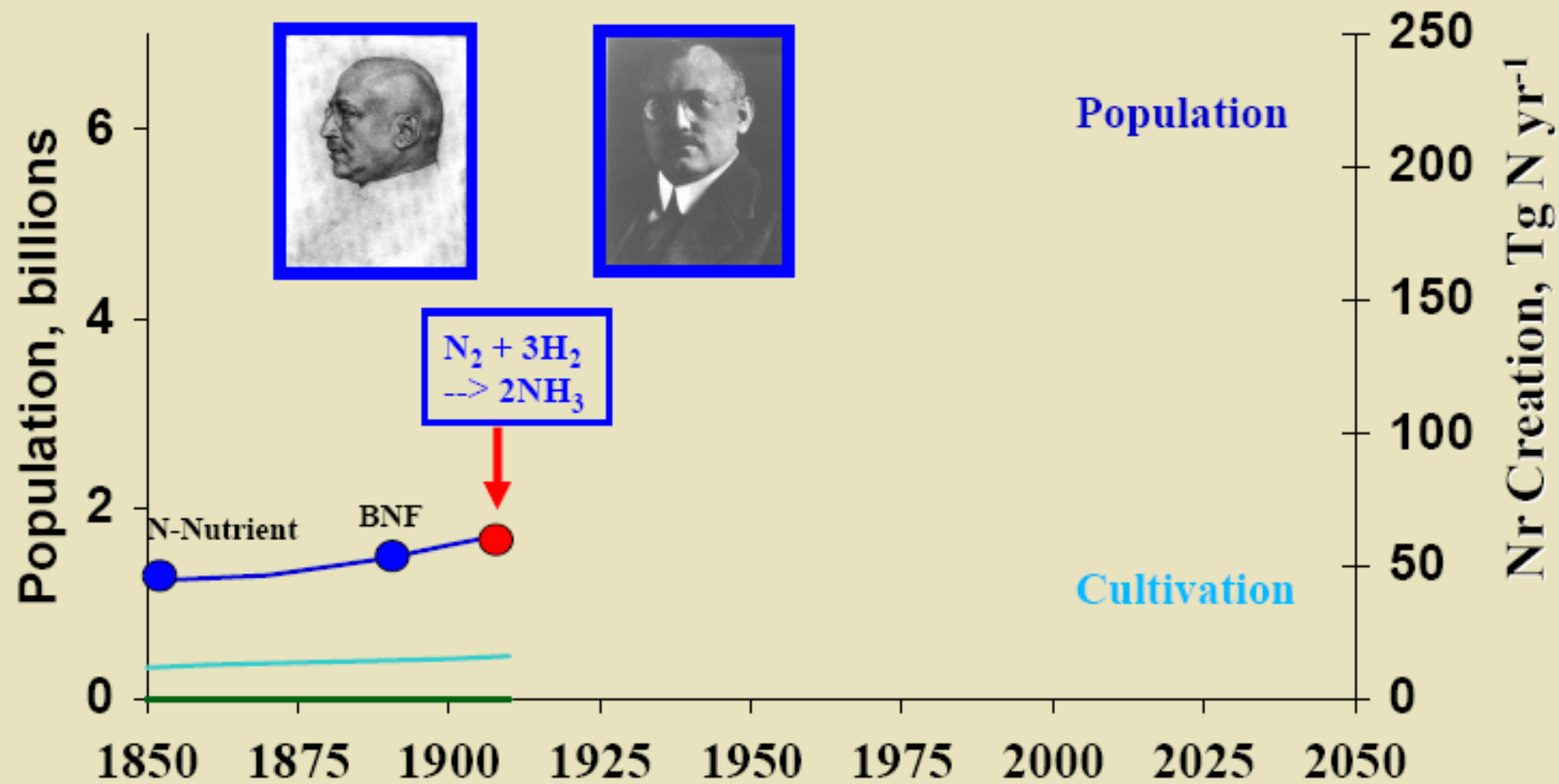
Timeline of Global Reactive N Creation by Human Activity 1850 to 2000



*1898, Sir William Crookes, president of the British Association for the Advancement of Science

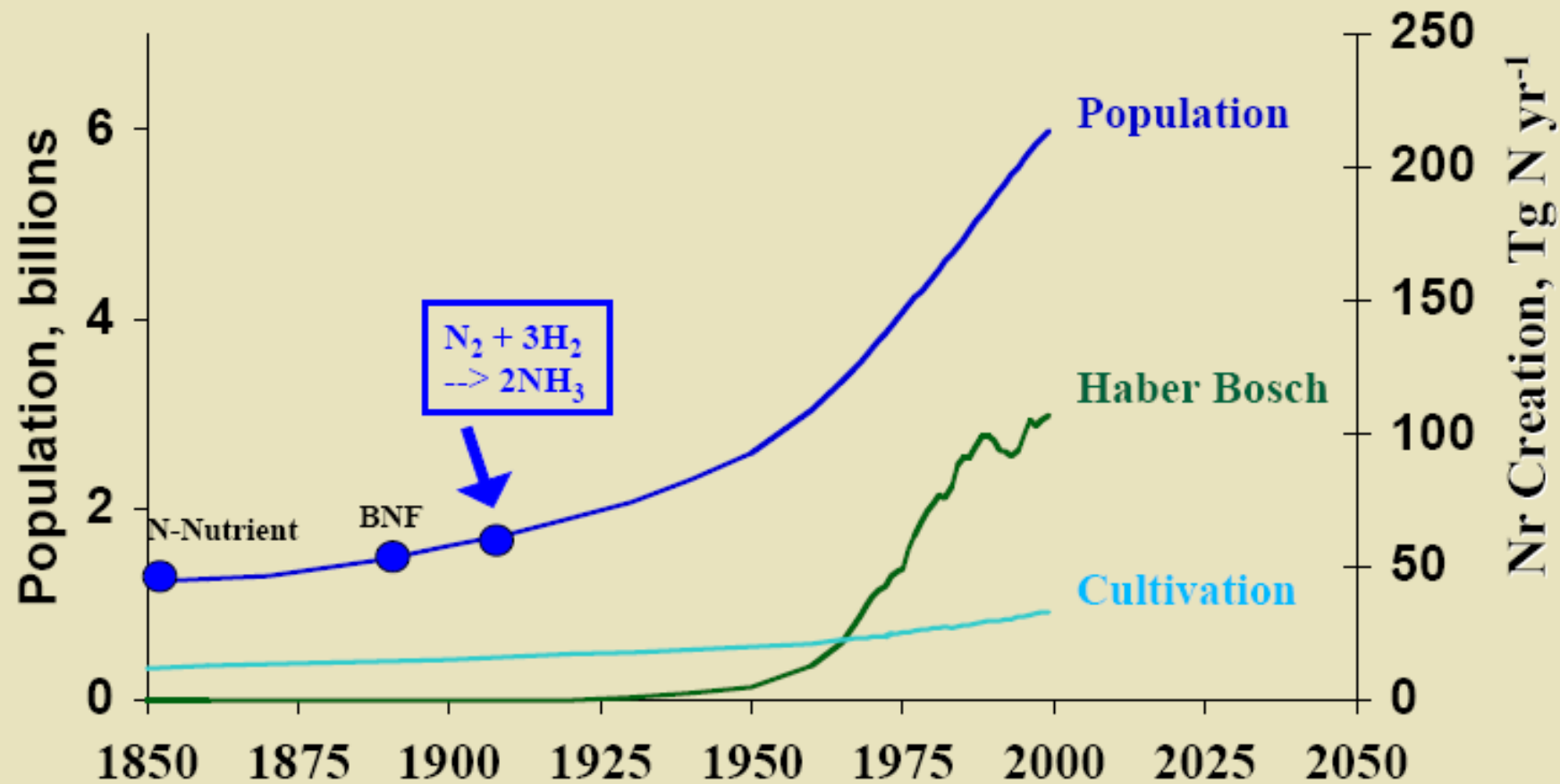
Galloway et al., 2003

Timeline of Global Reactive N Creation by Human Activity 1850 to 2000



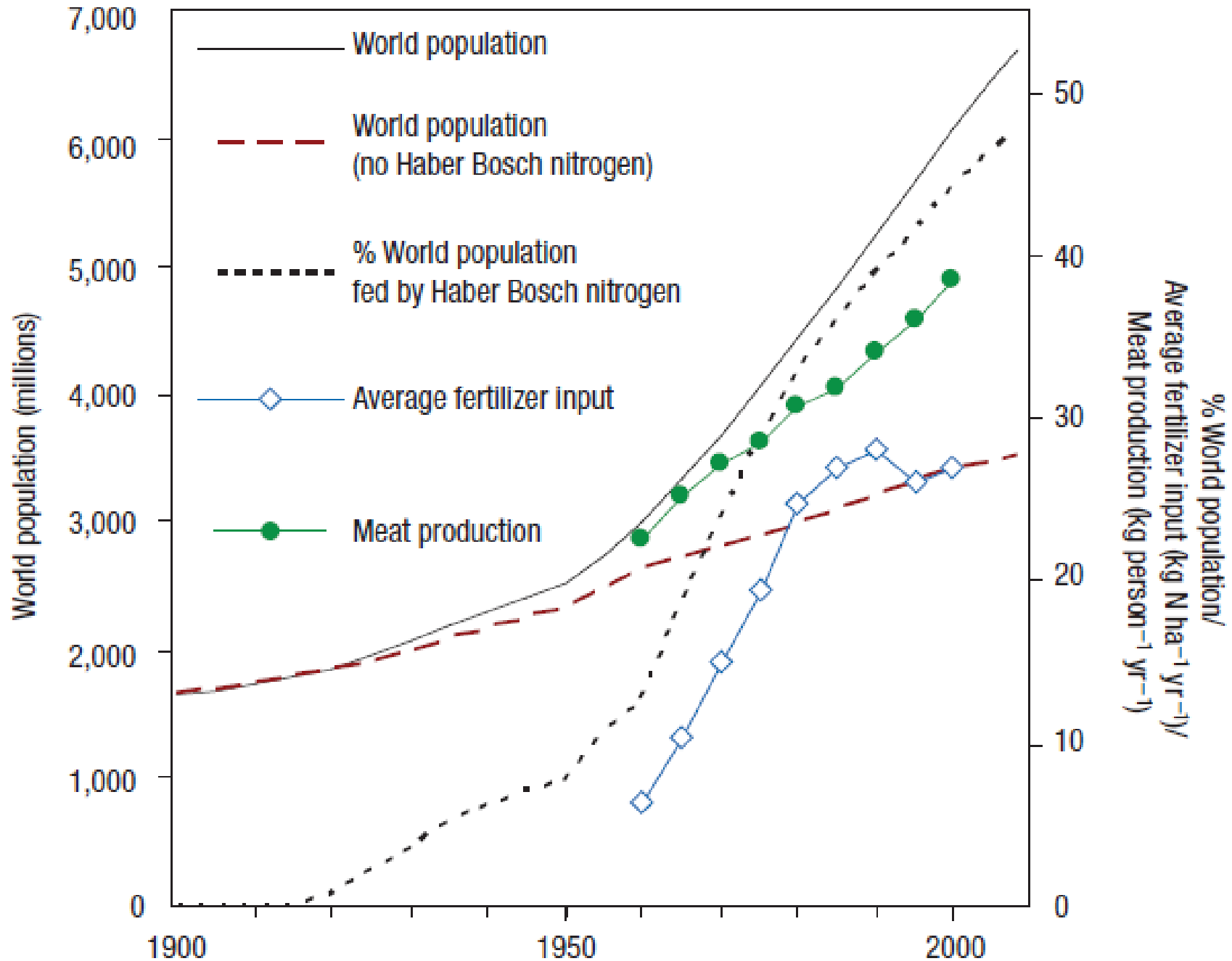
Galloway et al., 2003

Timeline of Global Reactive N Creation by Human Activity 1850 to 2000



Galloway et al., 2003

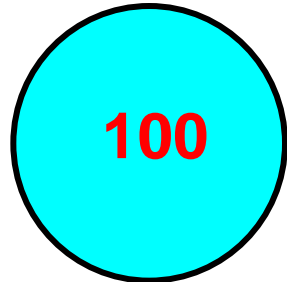
Nitrogen and global population increase



Erisman et al., 2008; Nature - Geosciences

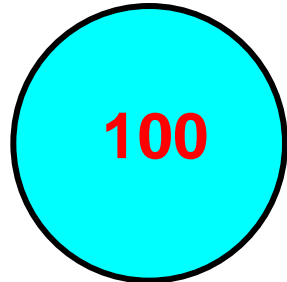
The fate of nitrogen

**N Fertilizer
Produced**



The fate of nitrogen

**N Fertilizer
Produced**



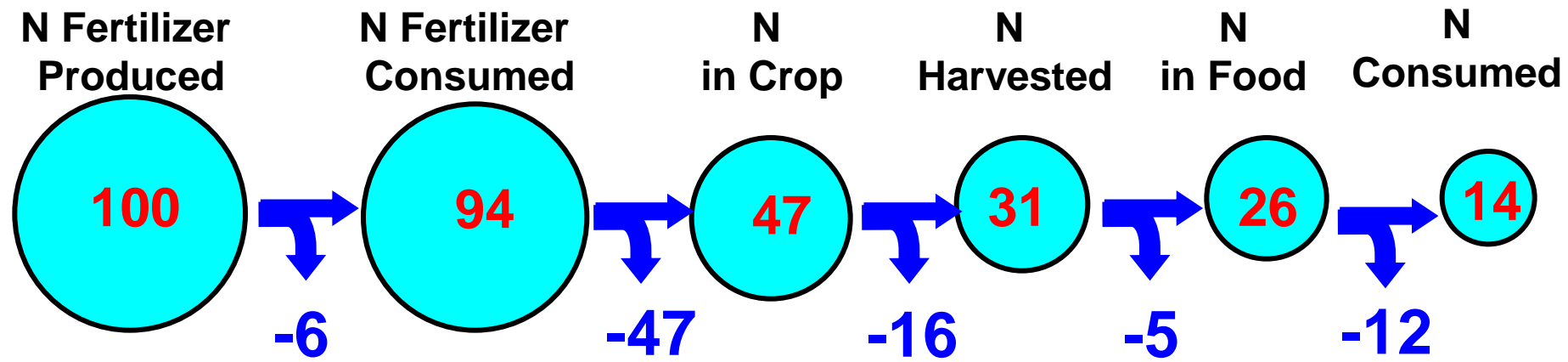
**N
Consumed**



14% of the N produced in the Haber-Bosch process enters the human mouth.....

Galloway JN and Cowling EB. 2002

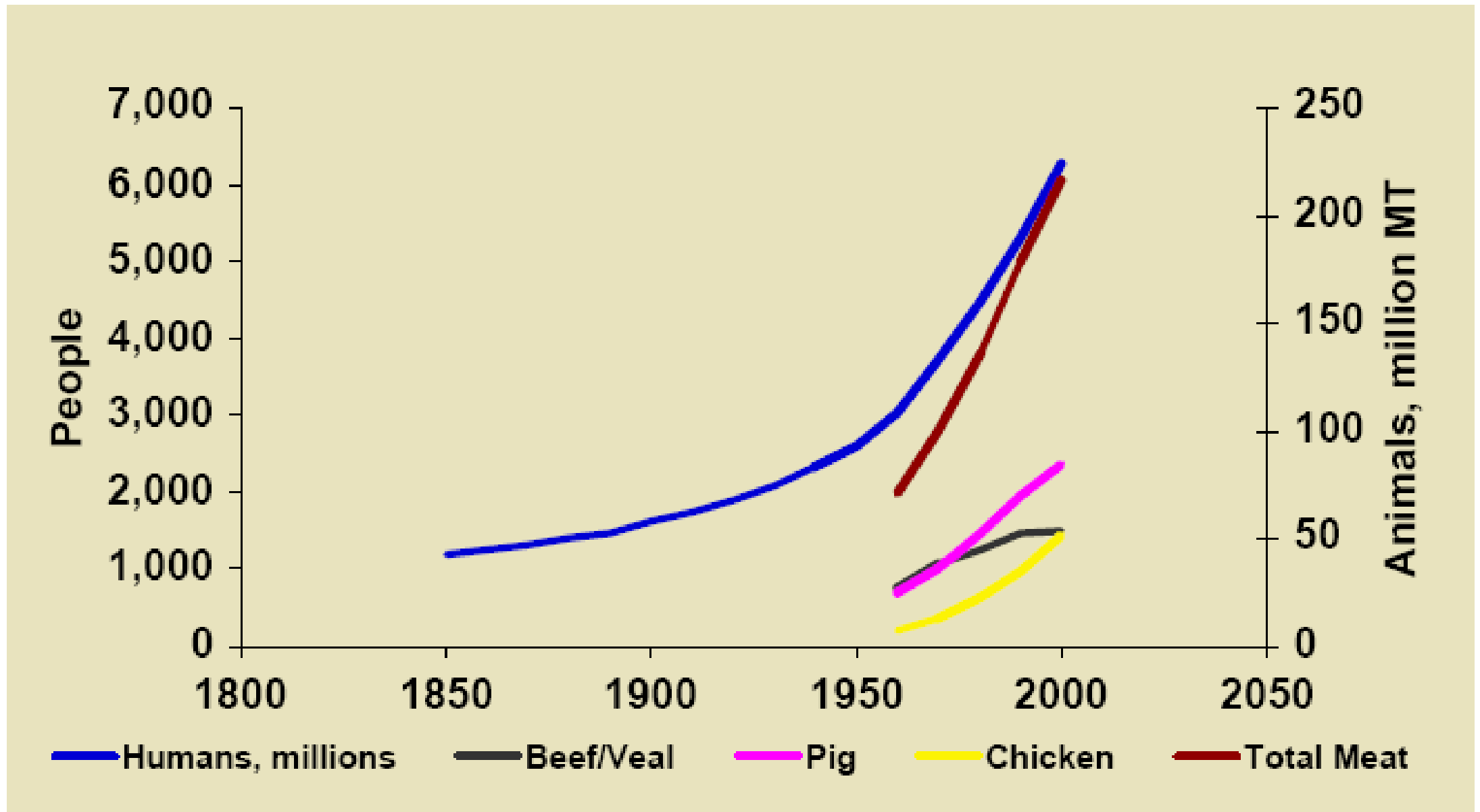
The fate of nitrogen



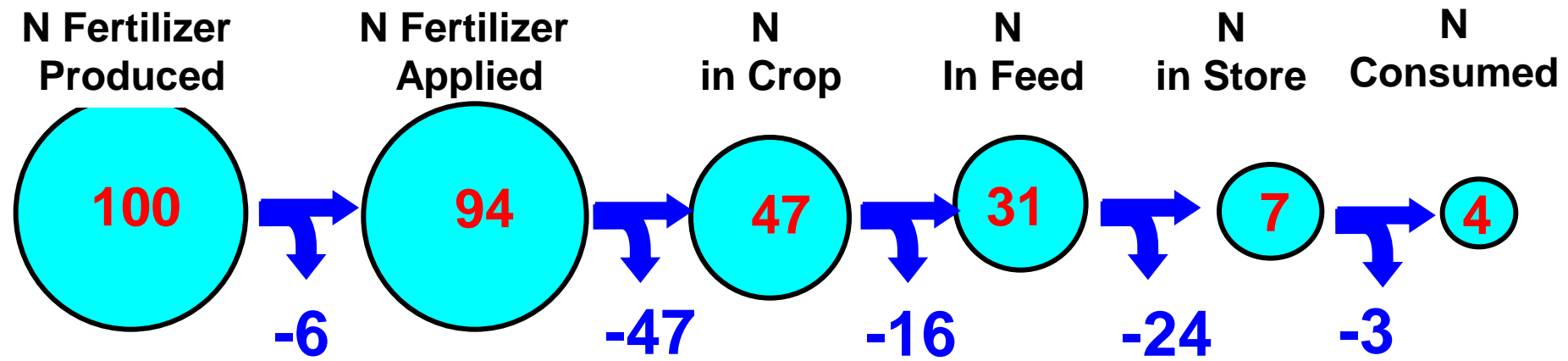
14% of the N produced in the Haber-Bosch process enters the human mouth.....if you are a vegetarian.

Galloway JN and Cowling EB. 2002

Global human population and meat production



The fate of nitrogen



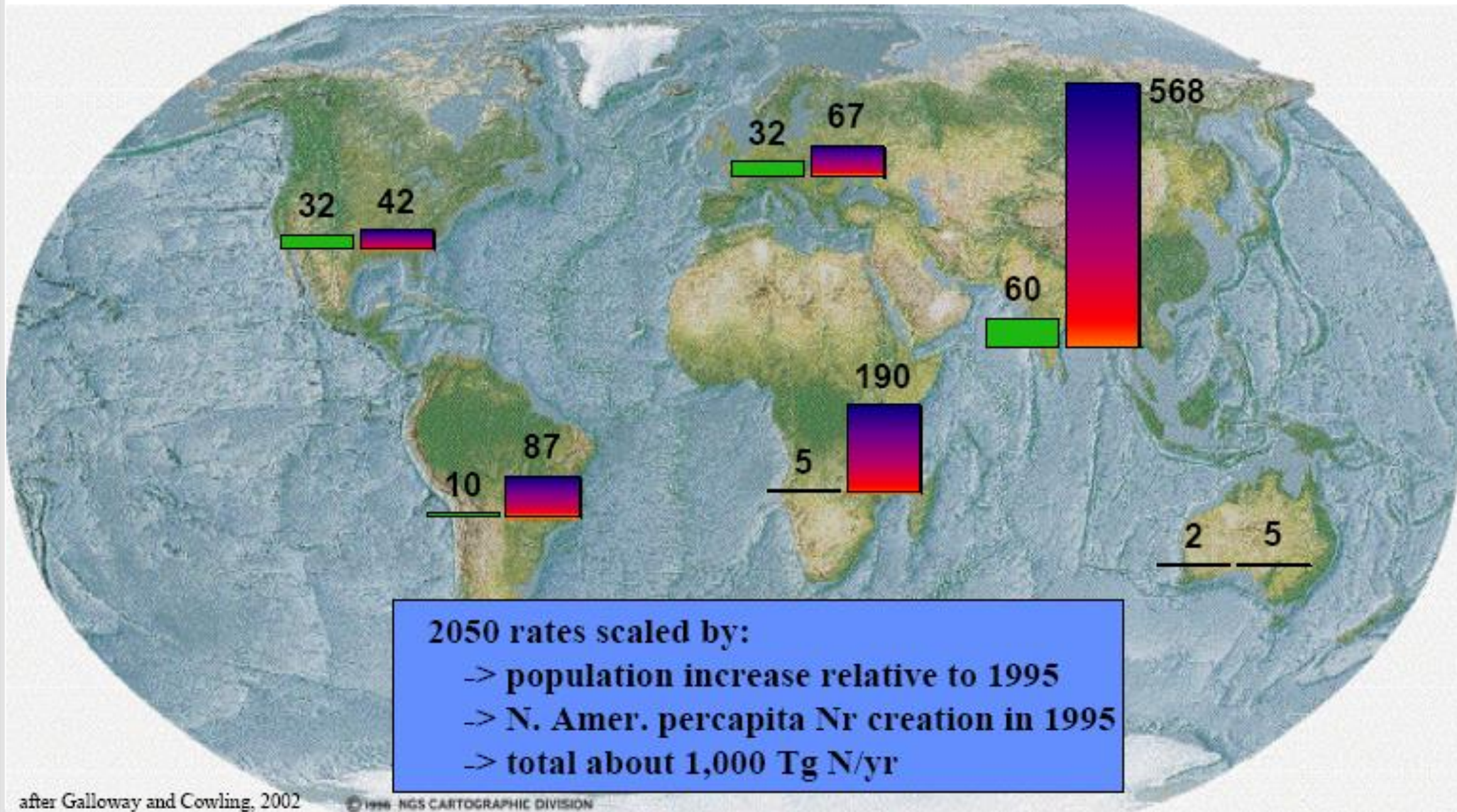
4% of the N produced in the Haber-Bosch process and used for animal production enters the human mouth.

Galloway JN and Cowling EB. 2002

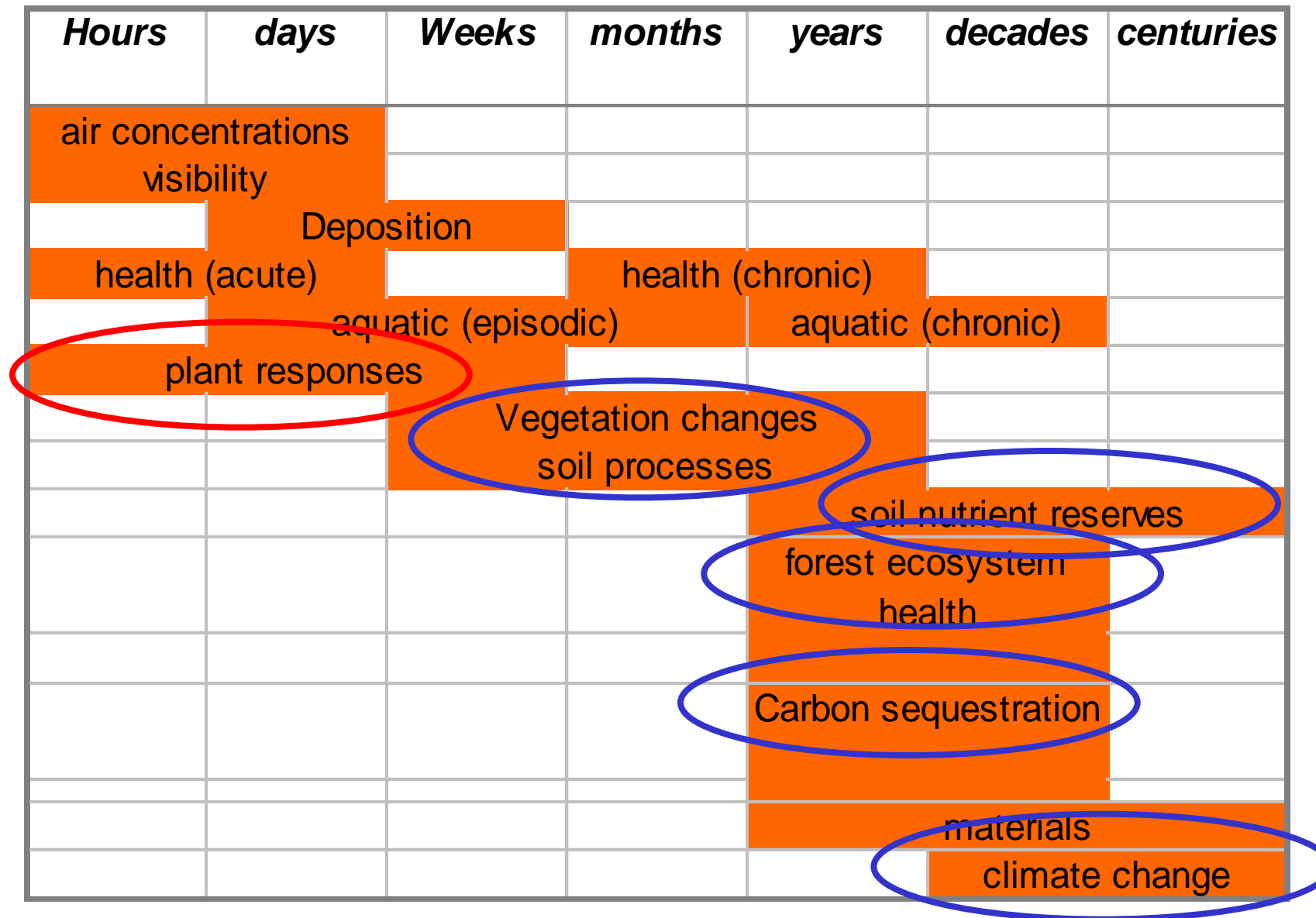
The global nitrogen cycle



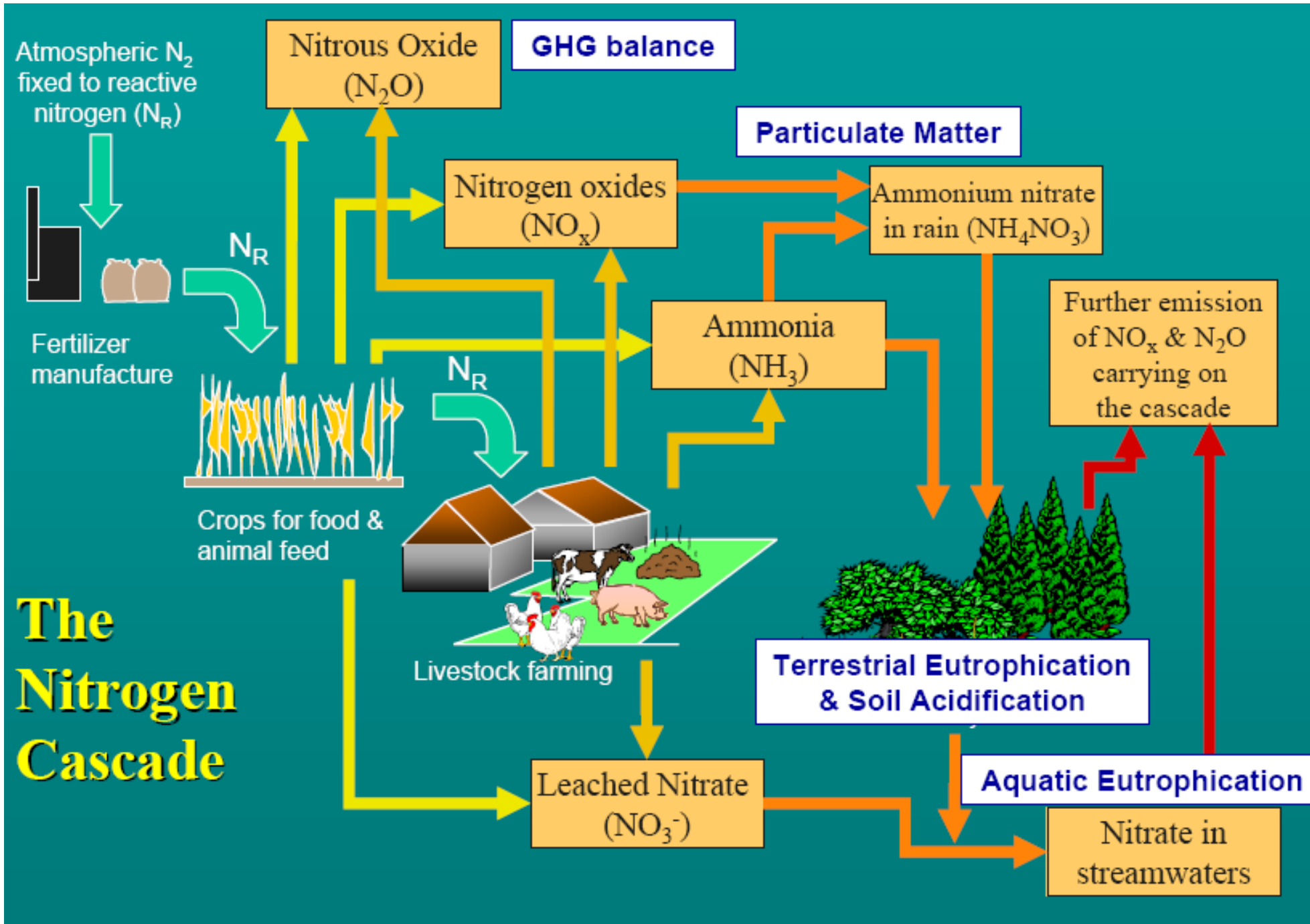
Nr creation 1995 (left) and 2050 (right) [Tg N yr⁻¹]



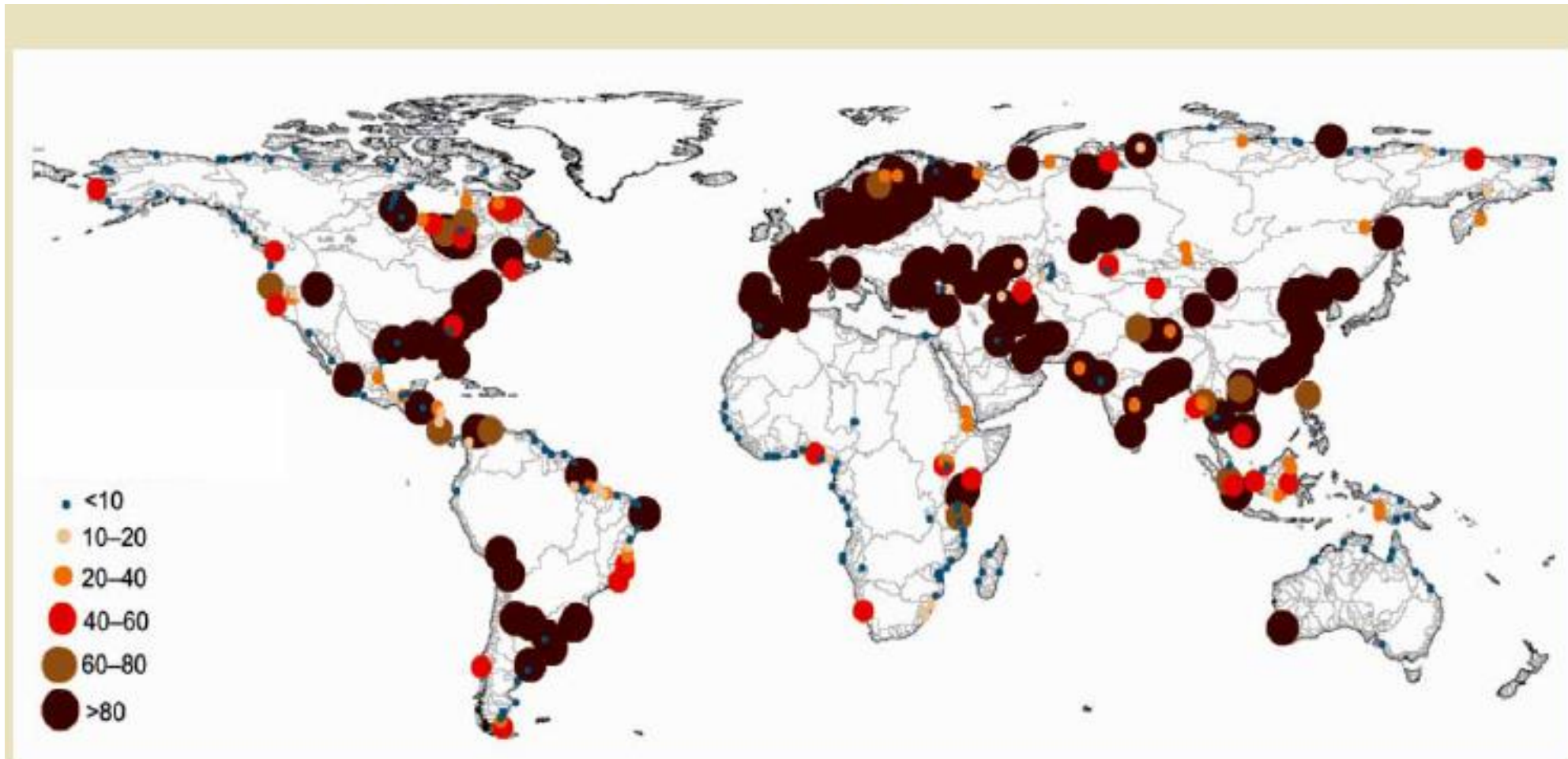
Ecological and social consequences of Nr



The N_r cascade



Increase in nitrogen flows in rivers

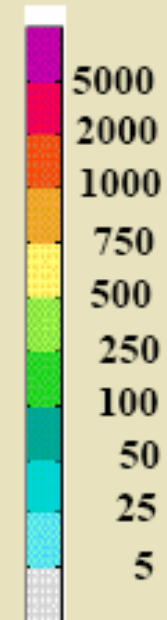
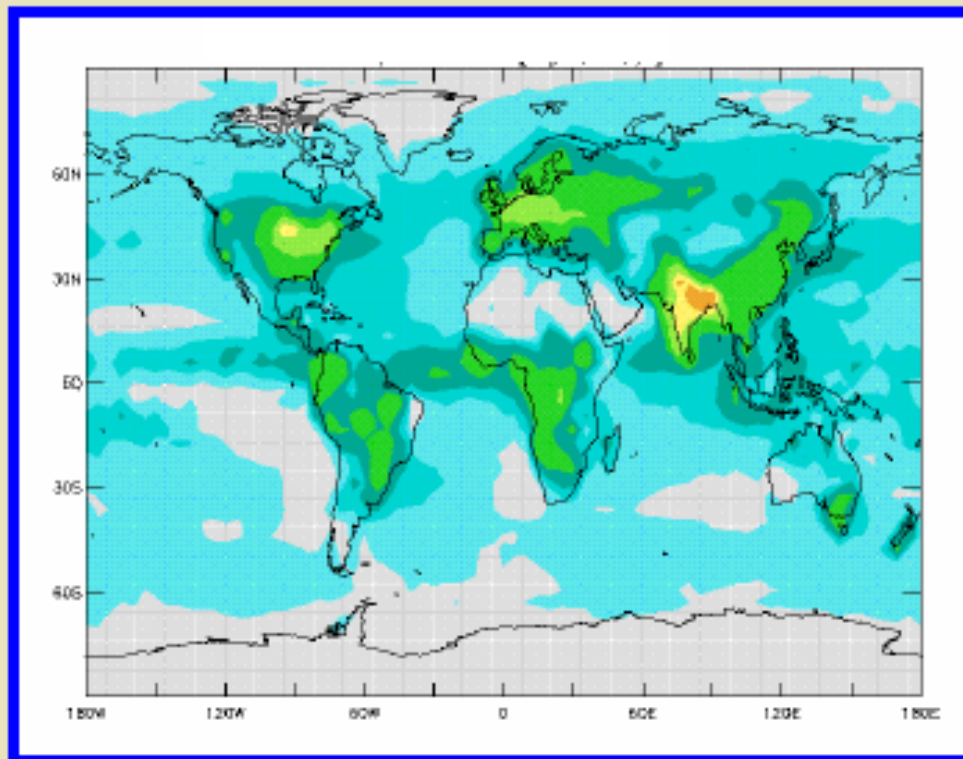


◆ Most of the Nr created for food production, is released to the environment

◆ About 25% is discharged to the coast via rivers

Source: Millennium Ecosystem Assessment

Nr deposition in 1860 and 1993 [$\text{mg m}^{-2} \text{yr}^{-1}$]

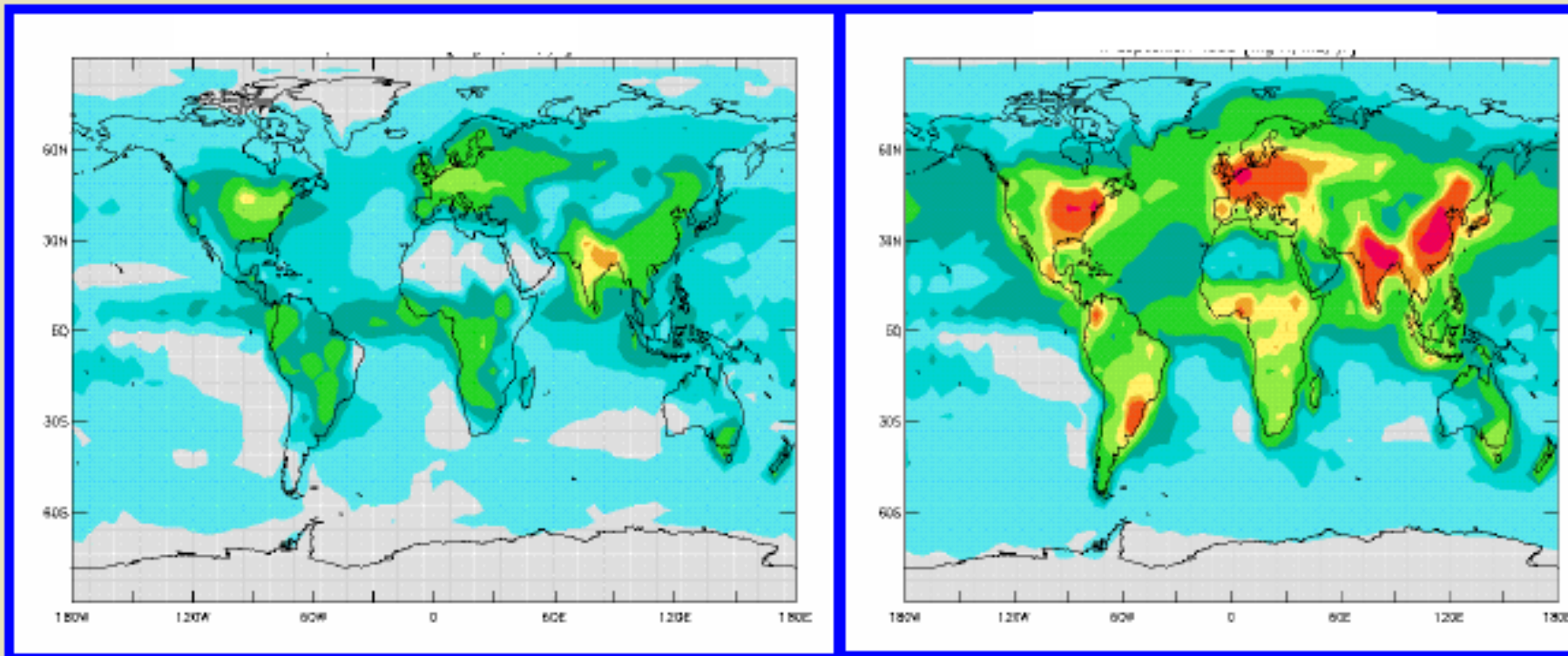


1860

- Nitrogen is emitted as NO_x to the atmosphere by fossil fuel combustion
- Nitrogen is emitted as NH_3 and NO_x from food production.
- Once emitted, it is transported and deposited to ecosystems.
- In 1860, human activities had limited influence on N deposition.

Galloway et al., 2003b

Nr deposition in 1860 and 1993 [$\text{mg m}^{-2} \text{yr}^{-1}$]



1860

1993

- Nitrogen is emitted as NO_x to the atmosphere by fossil fuel combustion
- Nitrogen is emitted as NH_3 and NO_x from food production.
- Once emitted, it is transported and deposited to ecosystems.
- In 1860, human activities had limited influence on N deposition.
- By 1993, the picture had changed.

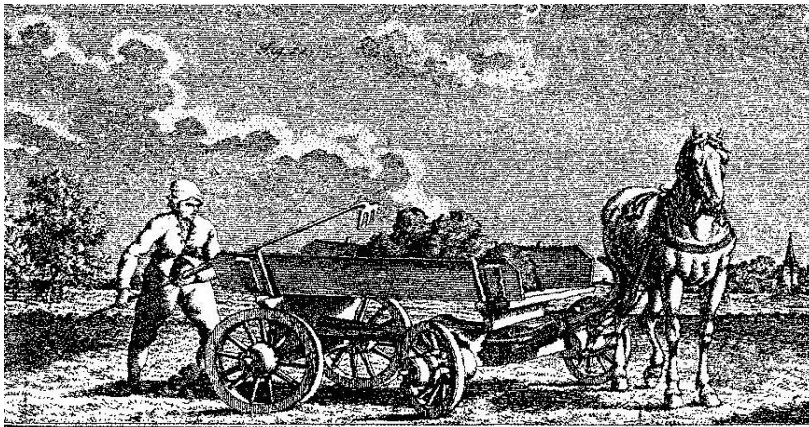
Galloway et al., 2003b

Historical development

Closed nutrient cycles



Intensive agriculture

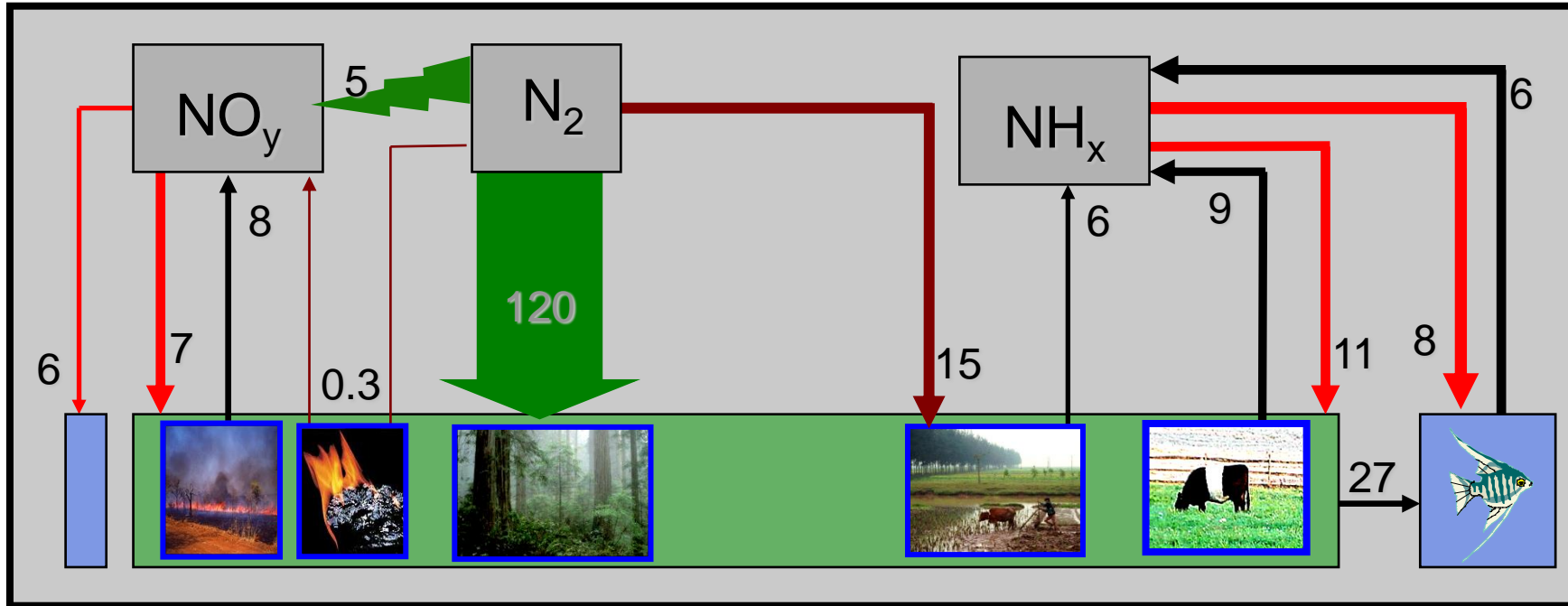


Industrialisation

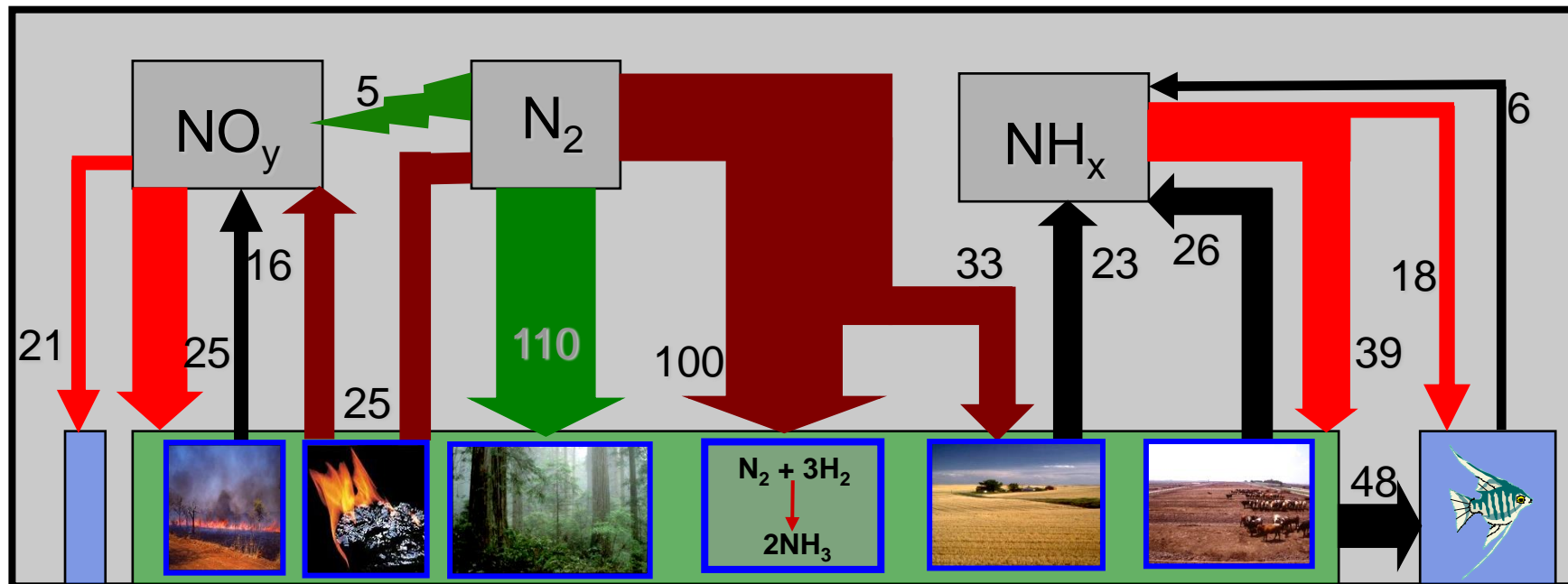
Man labor

The Global Nitrogen Budget in 1860 and mid-1990s, TgN/yr

1860

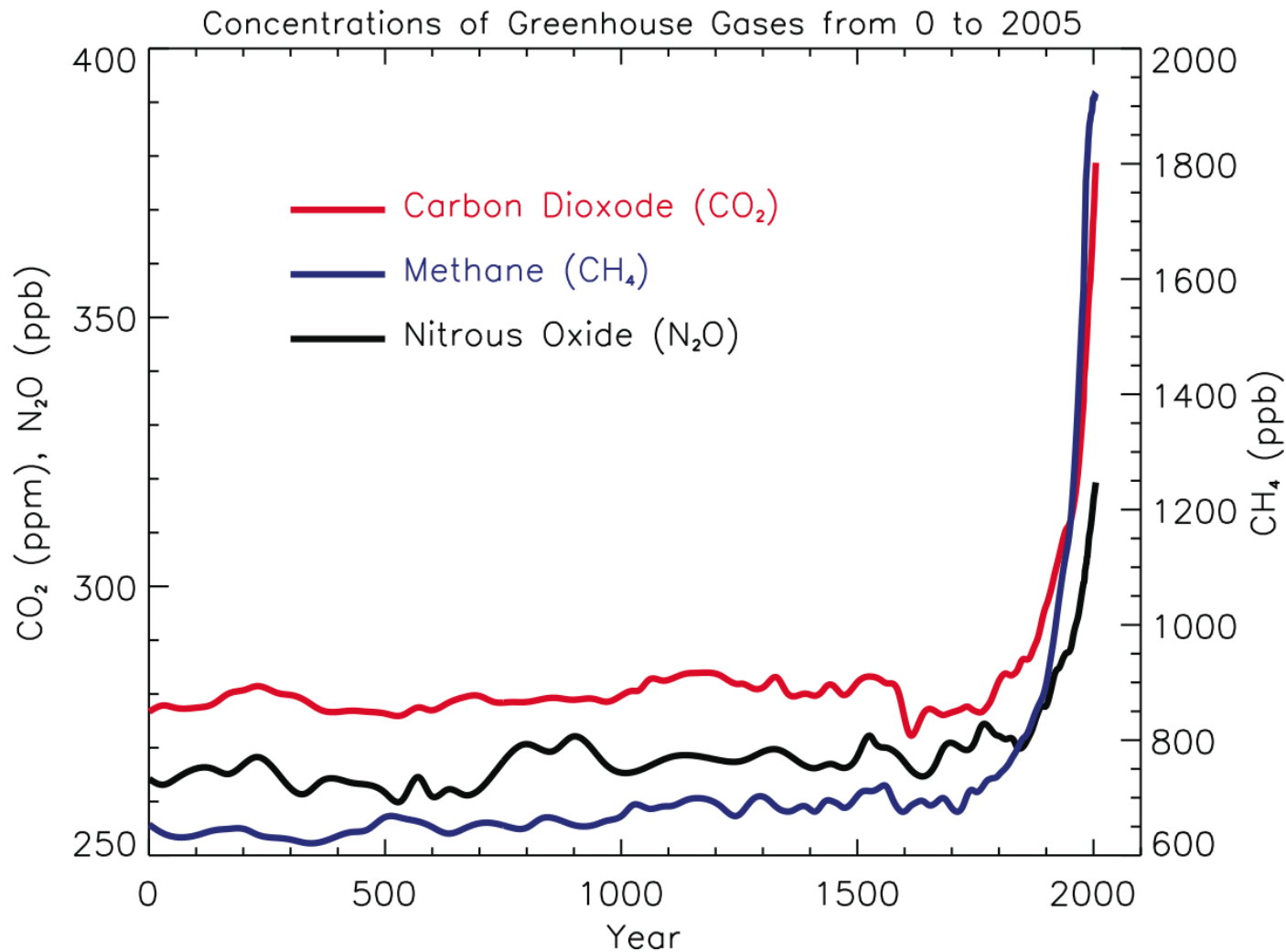


mid-1990s



Galloway et al., 2003b

N₂O

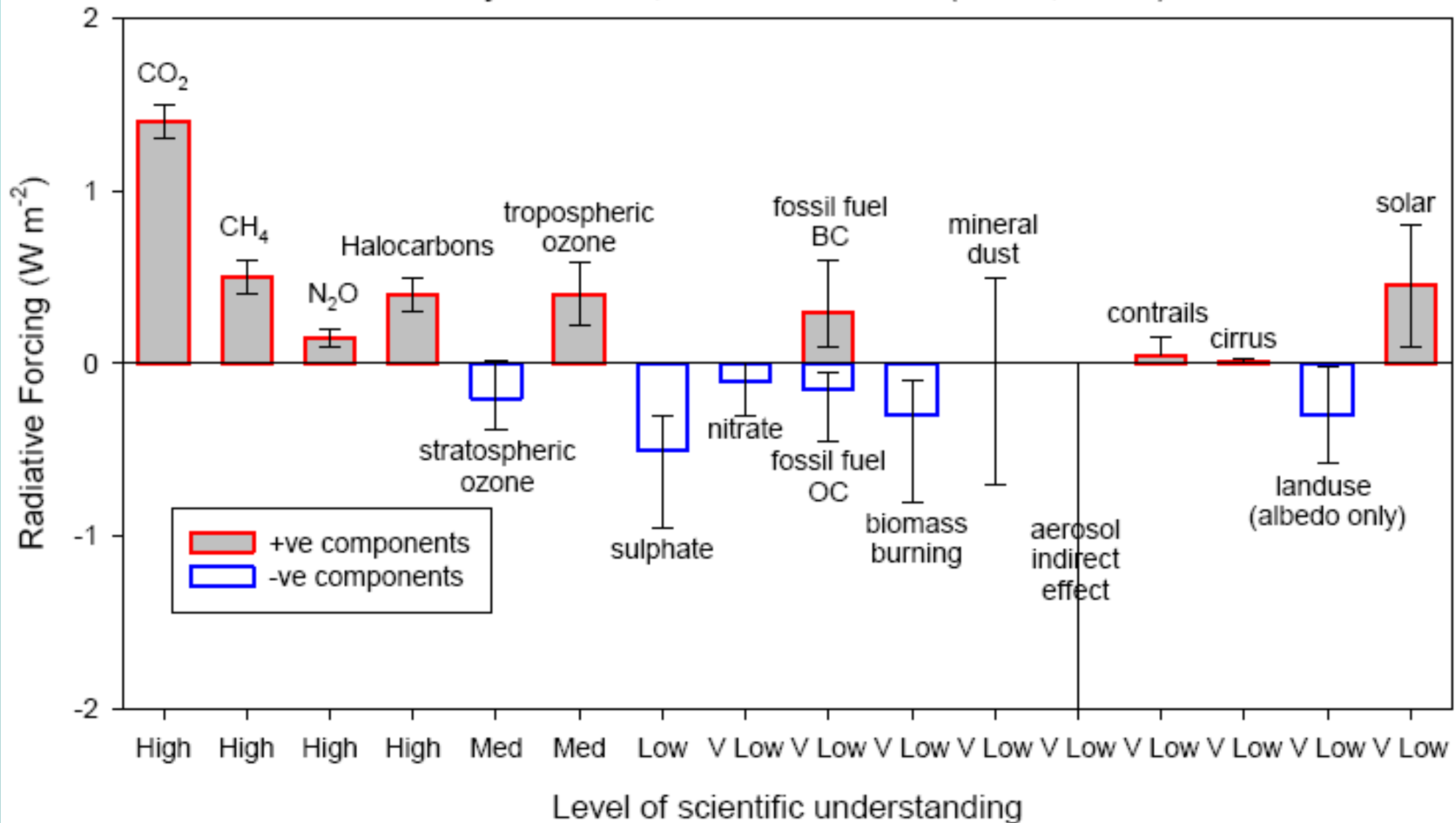


IPCC 2007

Industrial Designation or Common Name (years)	Chemical Formula	Lifetime (years)	Radiative Efficiency (W m ⁻² ppb ⁻¹)	Global Warming Potential for Given Time Horizon			
				SAR [†] (100-yr)	20-yr	100-yr	500-yr
Carbon dioxide	CO ₂	See below ^a	^b 1.4x10 ⁻⁵	1	1	1	1
Methane ^c	CH ₄	12 ^c	3.7x10 ⁻⁴	21	72	25	7.6
Nitrous oxide	N ₂ O	114	3.03x10 ⁻³	310	289	298	153

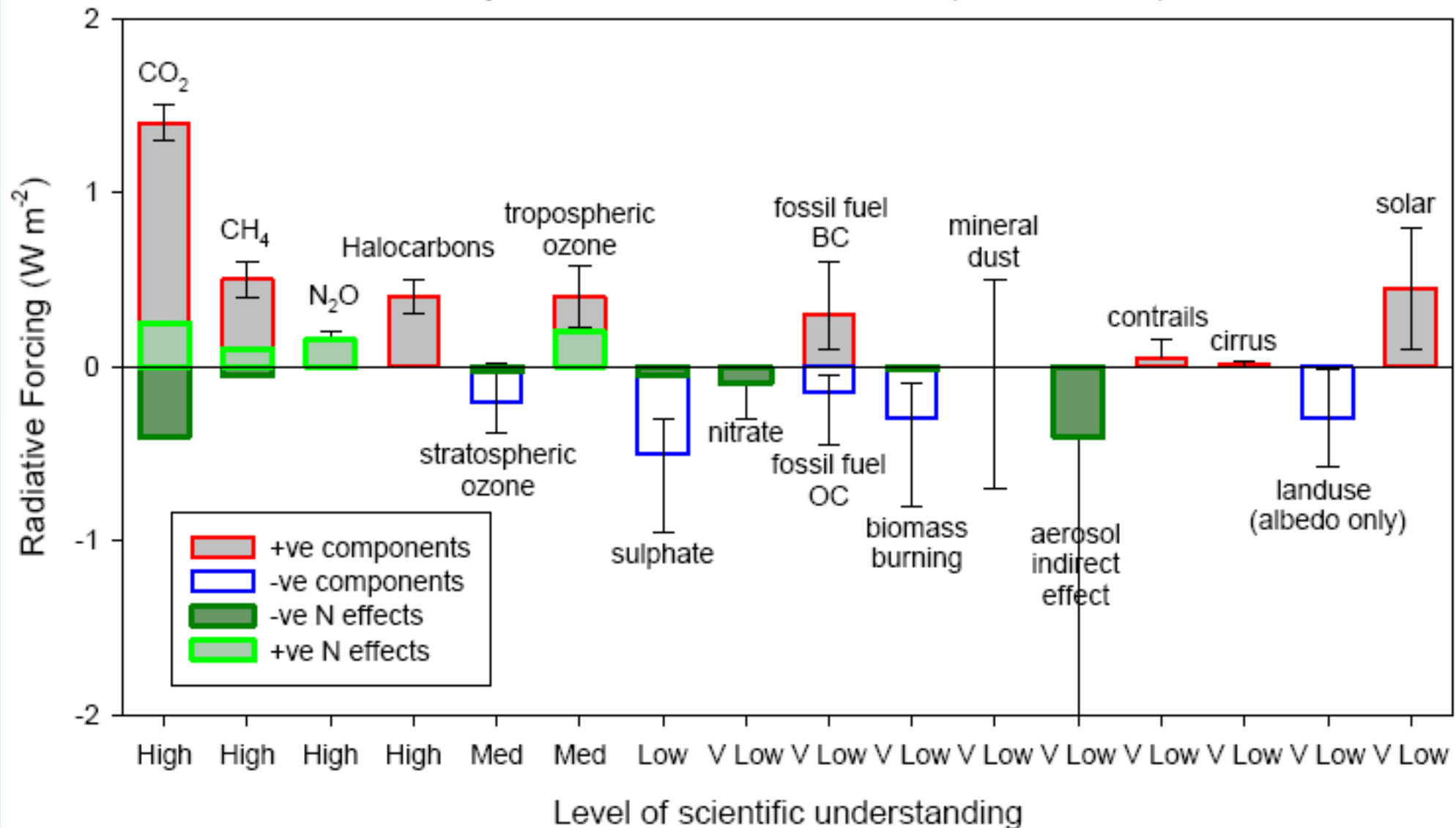
The effect of N on the GHG balance

The mean global radiative forcing of the climate system for the year 2000, relative to 1750 (IPCC, 2001)

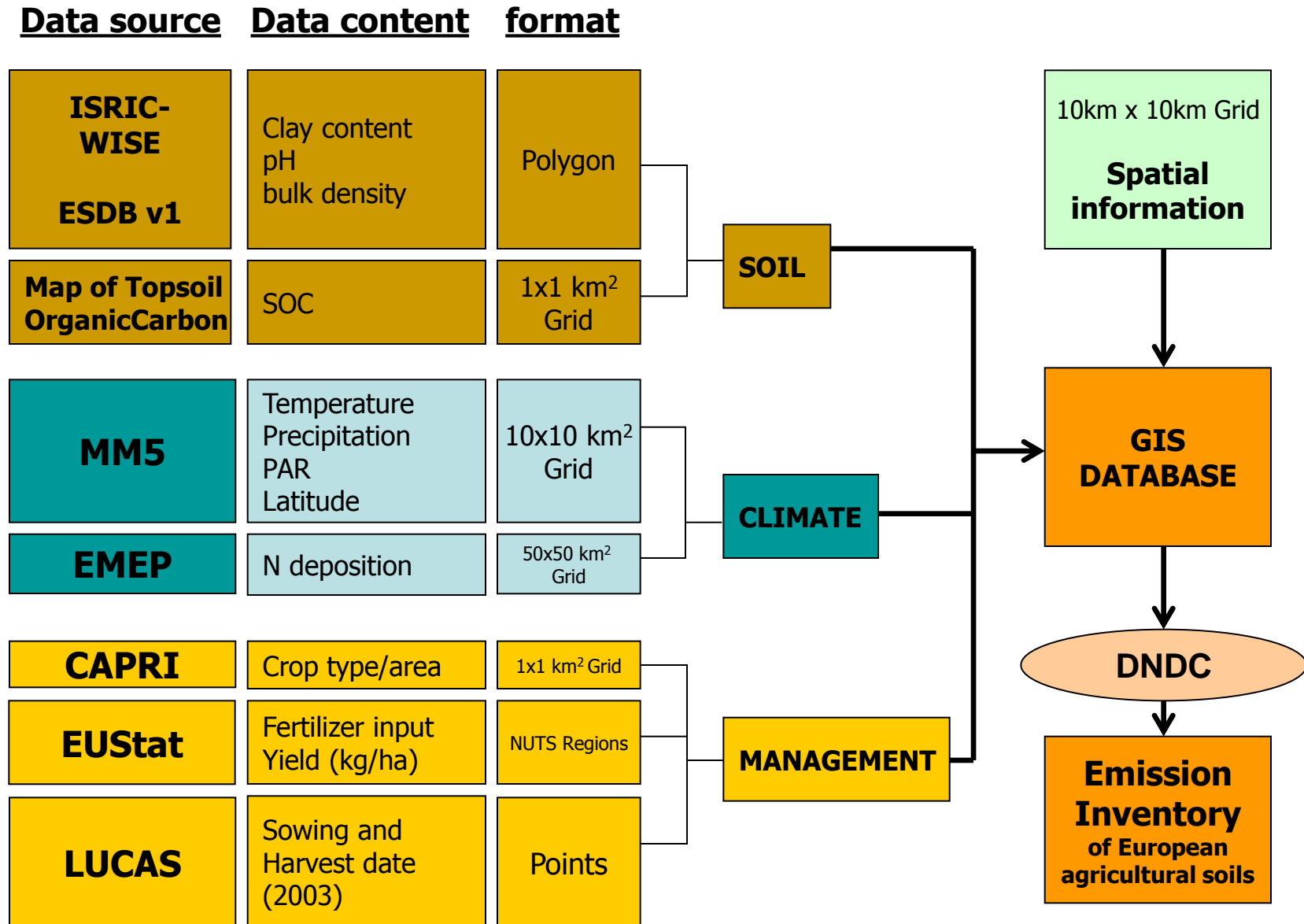


The effect of N on the GHG balance

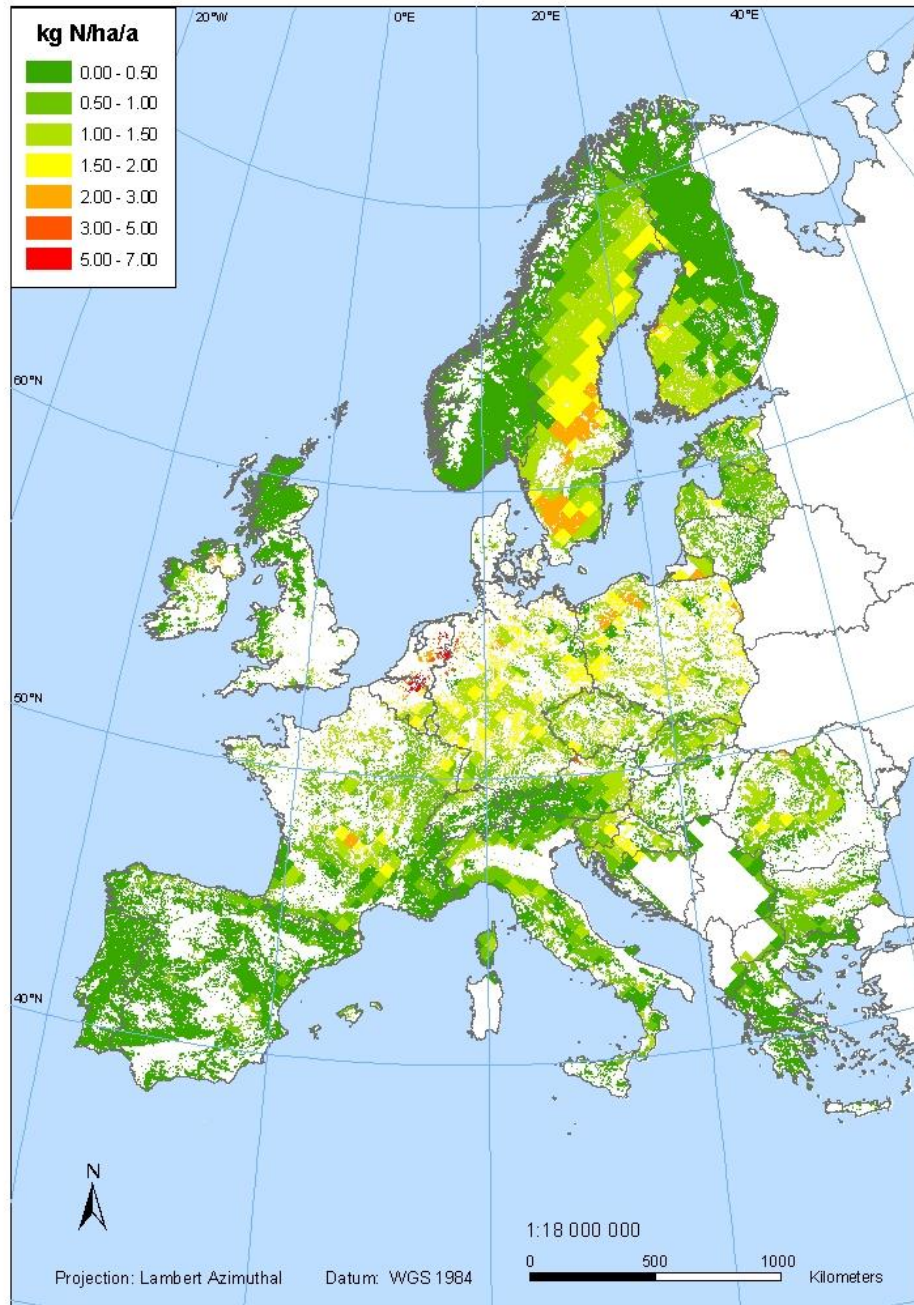
The mean global radiative forcing of the climate system for the year 2000, relative to 1750 (IPCC, 2001)



GIS database for DNDC



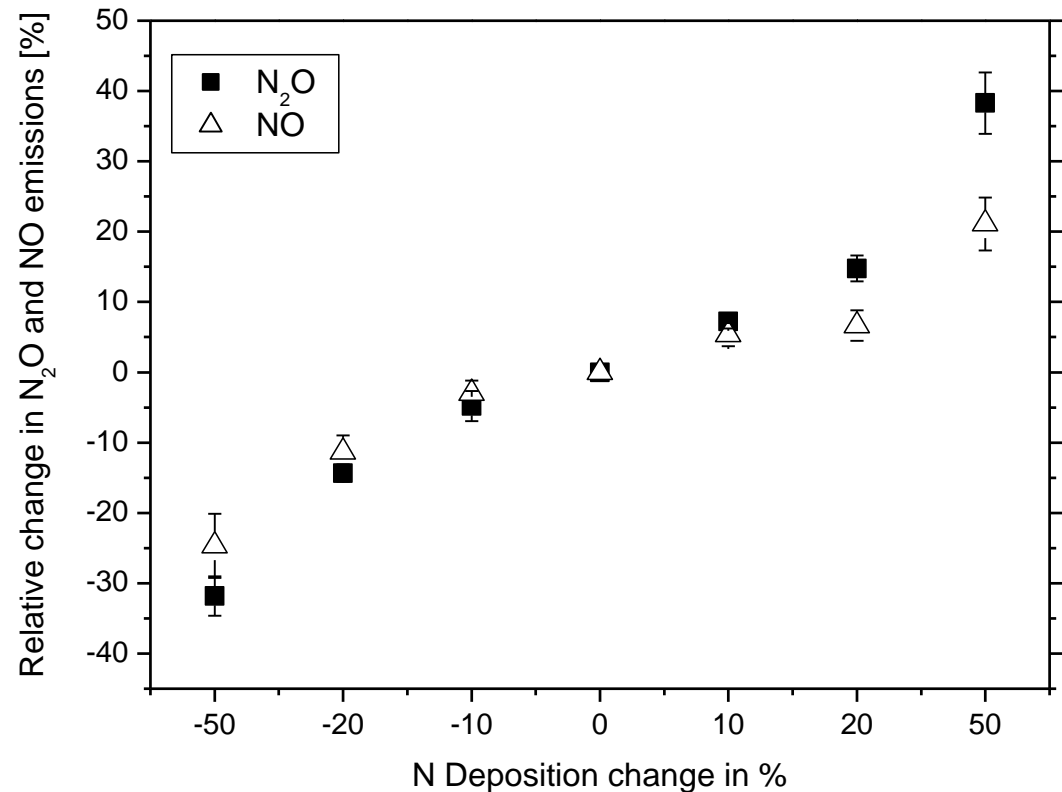
Inventorizing soil N trace gas fluxes and identifying feedbacks



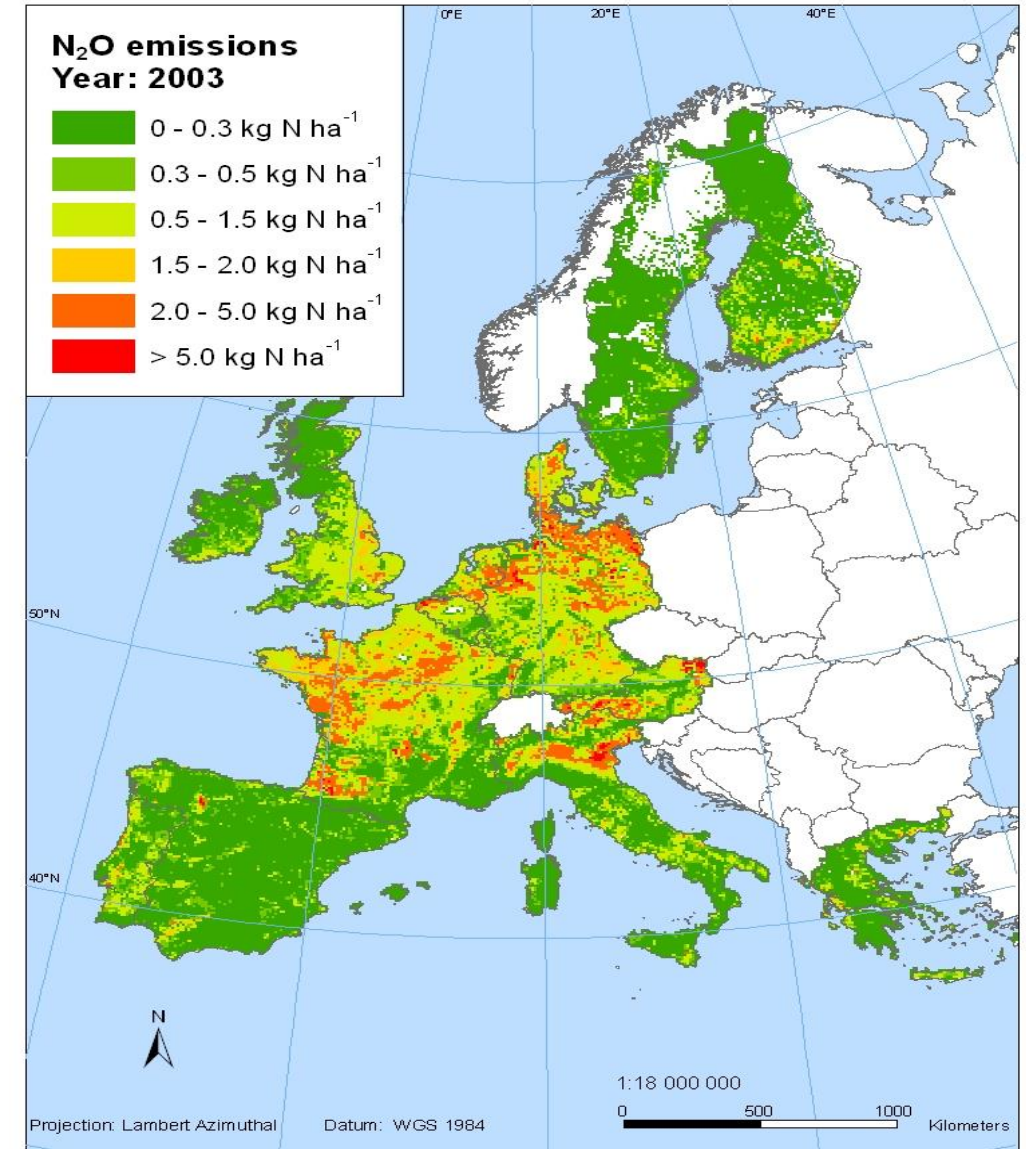
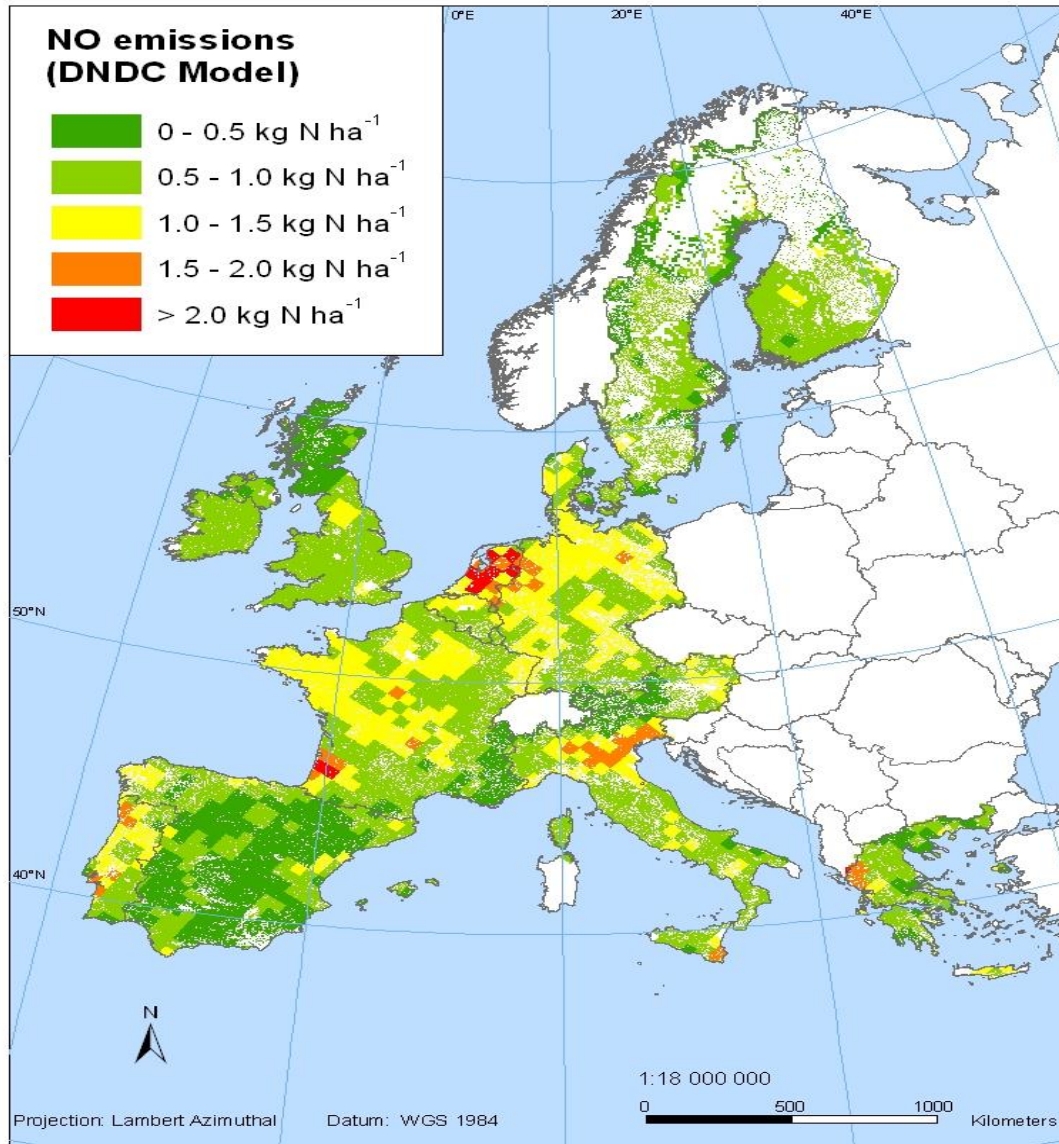
Kesik et al., 2006; Biogeosciences

NO Emissions	Minimum Scenario kt N a ⁻¹	Average Scenario kt N a ⁻¹	Maximum Scenario kt N a ⁻¹
1990	45	98	248
1995	38	85	220
2000	45	99	254

Simulated forest area of Europe: 1 410 477km²



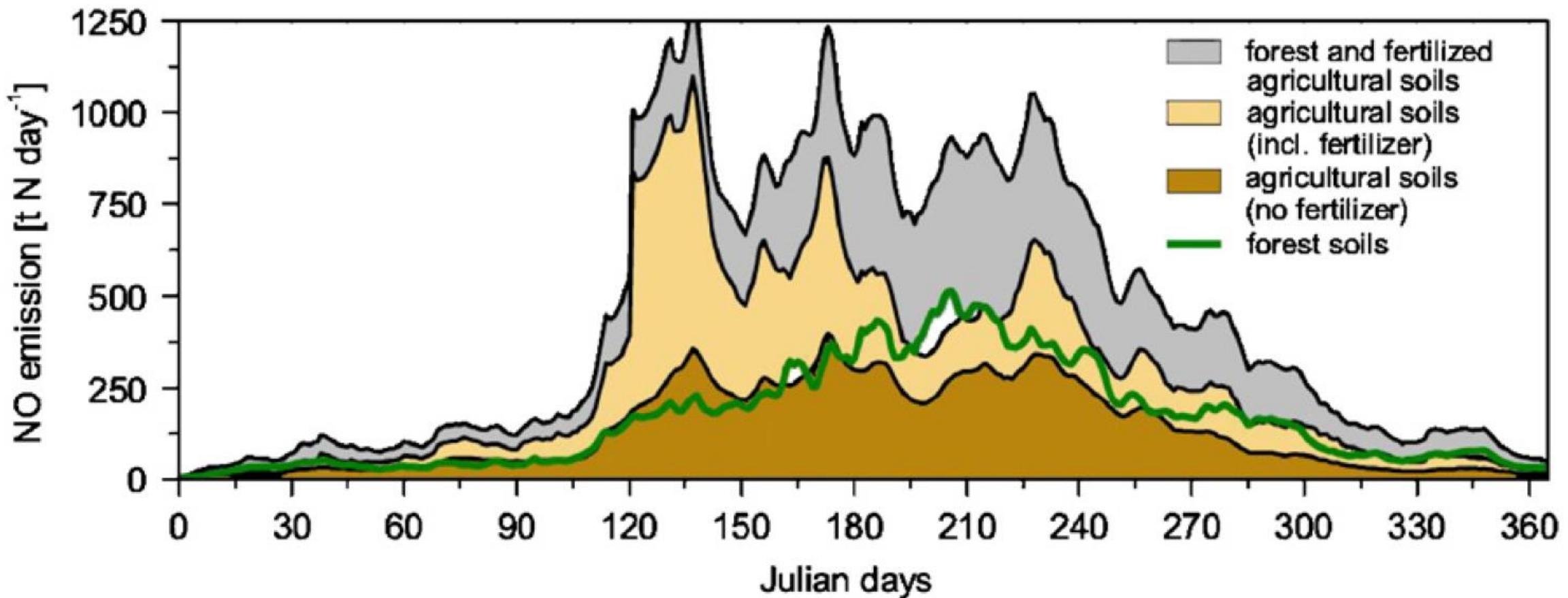
Agricultural soils – NO & N₂O emissions



Butterbach-Bahl et al., 2008; Atm. Environm.

Soil sources versus industrial sources

Soil NO_x emissions are contributing in average for entire Europe up to 10% to the tropospheric NO_x burden



Butterbach-Bahl et al., 2009; *Atm. Environm.*

Summary

- Human activities have perturbed the global nitrogen cycling
 - Acceleration by a factor of approx. 2 (C <10%)
- Large scale environmental impacts, e.g.
 - eutrophication
 - Biosphere-atmosphere exchange of GHG's
- Long-term effects and biosphere feedbacks to climate change unknown
- Increase in N use efficiency urgently needed