

U redox state and speciation of U in contact with magnetite nanoparticles: high-resolution XANES, EXAFS, XPS and TEM study

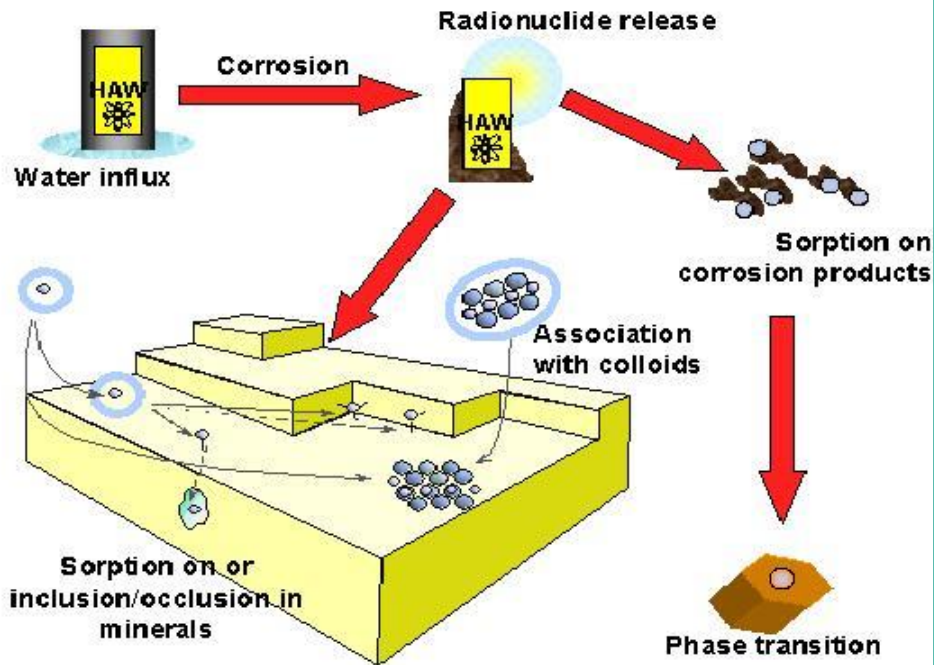
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Institute for Nuclear Disposal (INE), KARLSRUHE INSTITUTE FOR TECHNOLOGY (KIT)



Advanced synchrotron-based systematic investigations of actinide (An) and lanthanide (Ln) systems to understand and predict their reactivity

An geochemistry

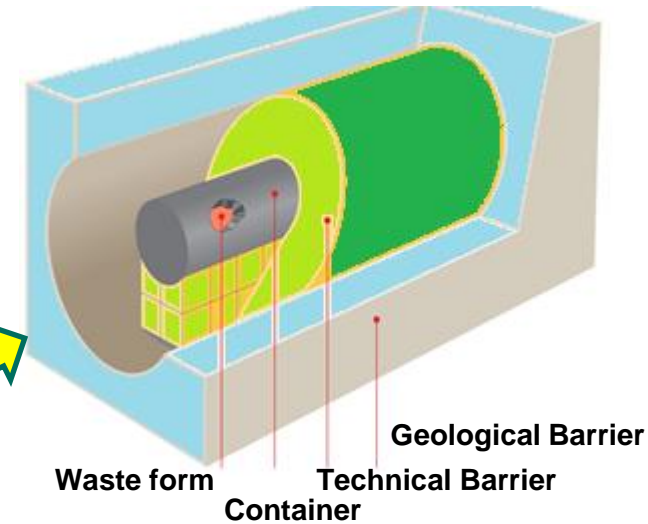
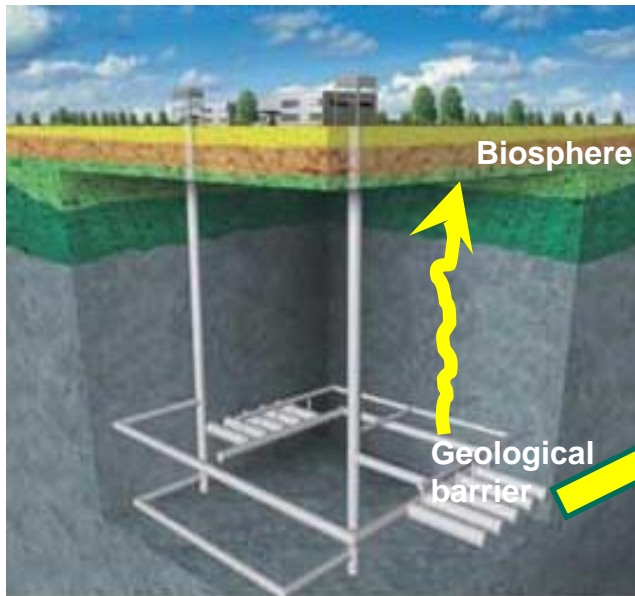


What is the **U/Pu** redox state and speciation after contact with magnetite?

Helmholtz-Young Investigators Group VH-NG-734 Institute for Nuclear Waste Disposal (KIT)



MOTIVATION



Anaerobic
corrosion
>1000 years



Fe canister
Corrosion
Studies
(INE)

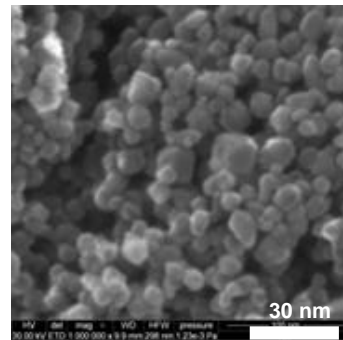
U⁶⁺ co-precipitation with magnetite

❑ U INCORPORATION INTO MAGNETITE ?

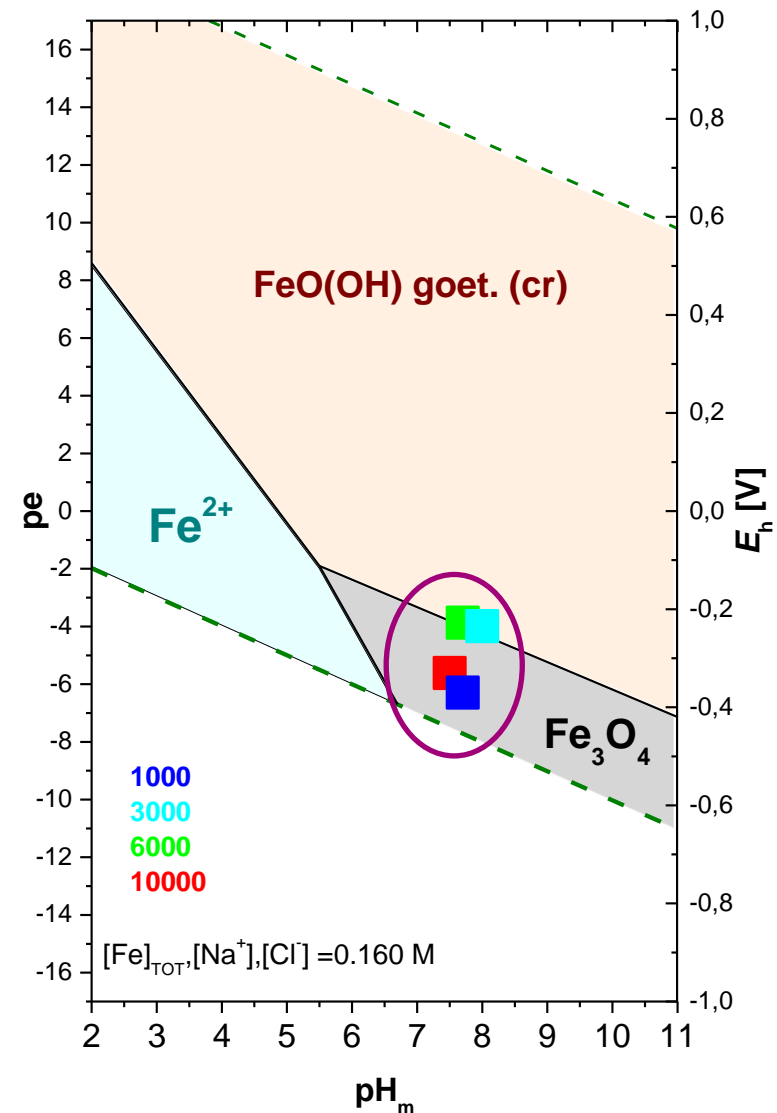
❑ IS U⁵⁺ POSSIBLE / STABLE ?

❑ WHAT IS U⁶⁺ REDOX KINETICS DURING MAGNETITE FORMATION ?

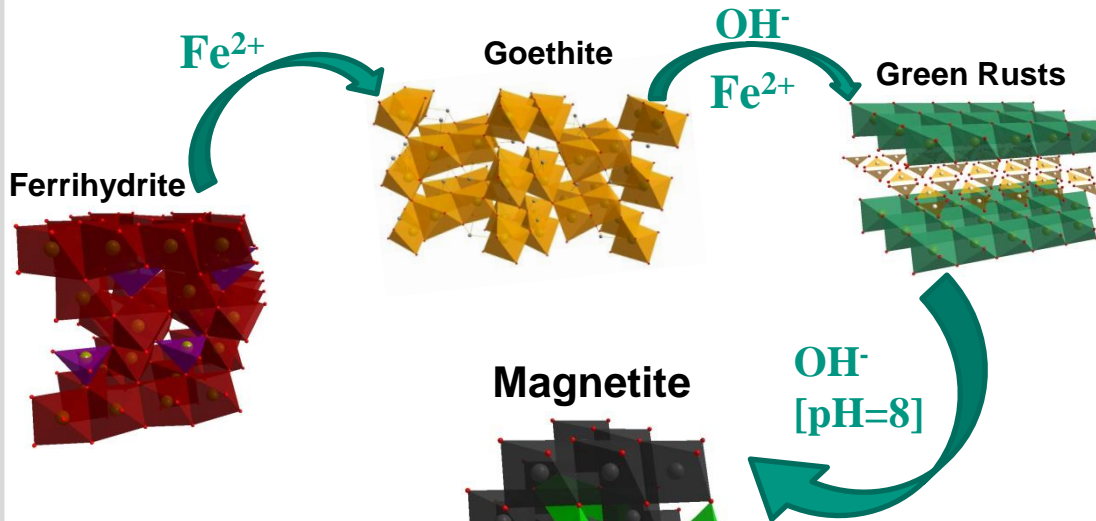
- FeCl₂ / FeCl₃ (1:2)
- NaOH (1 hour titration)
- 1 wt % NaCl (final)
- U 1000-10000 ppm
- 4 samples



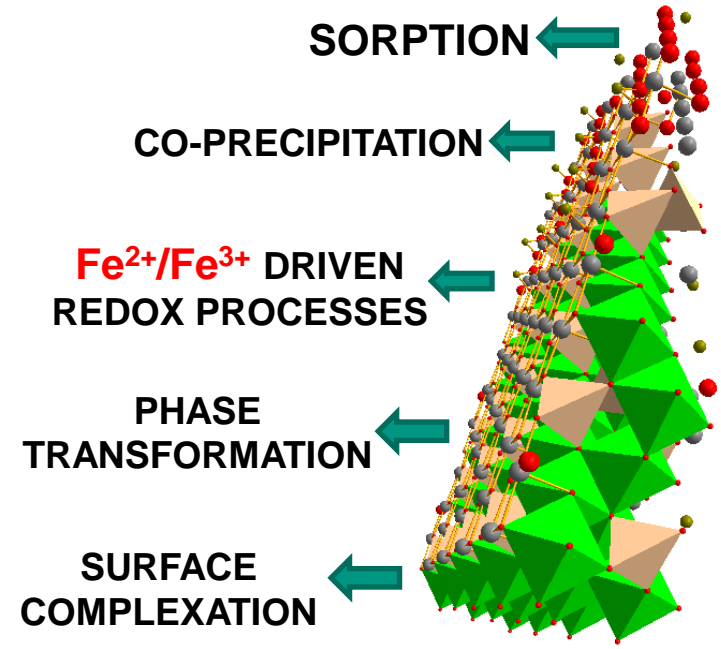
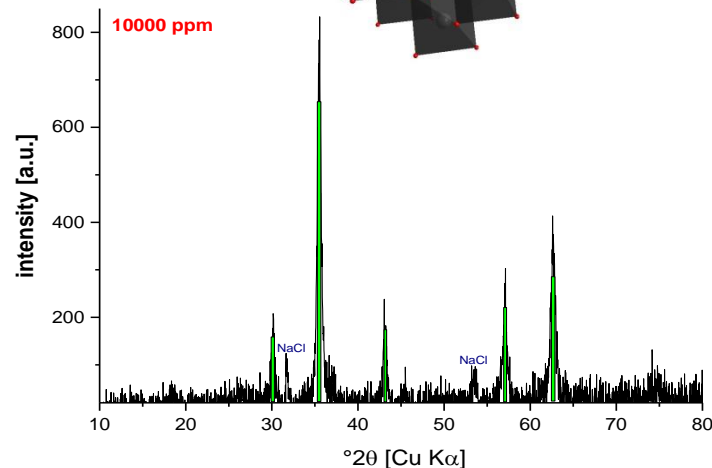
pH 2.5 → ~ 8.0
E_h 750 mV → - 370 mV



U⁶⁺ co-precipitation with magnetite



What happens to U?

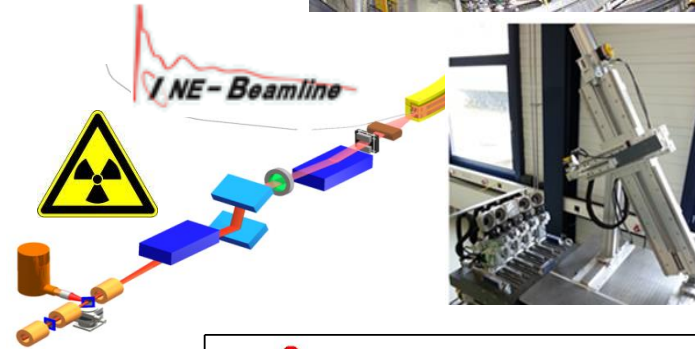
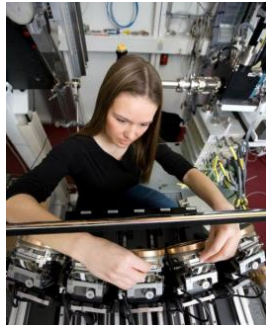


Jolivet et al, ChemCom, 2004
Massey et al, GCA, 2014

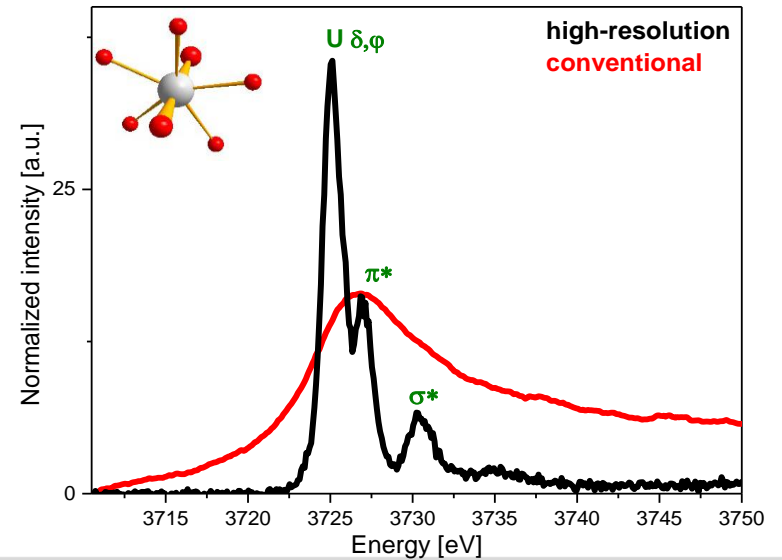
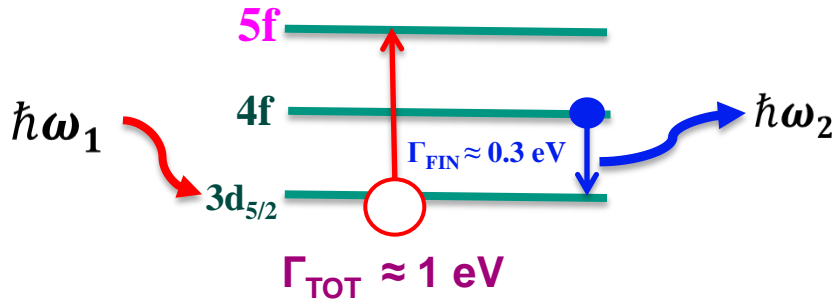
U M₄ HR-XANES / U L₃ EXAFS

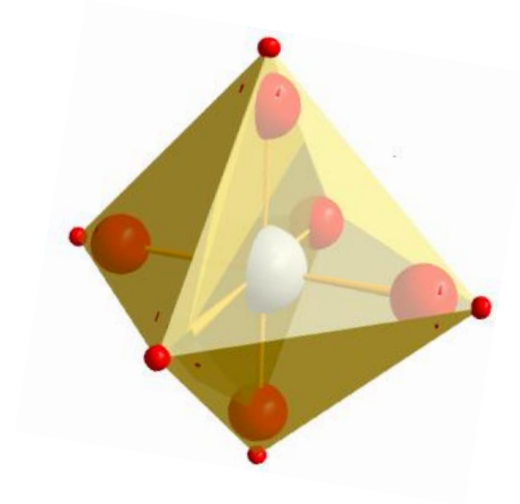
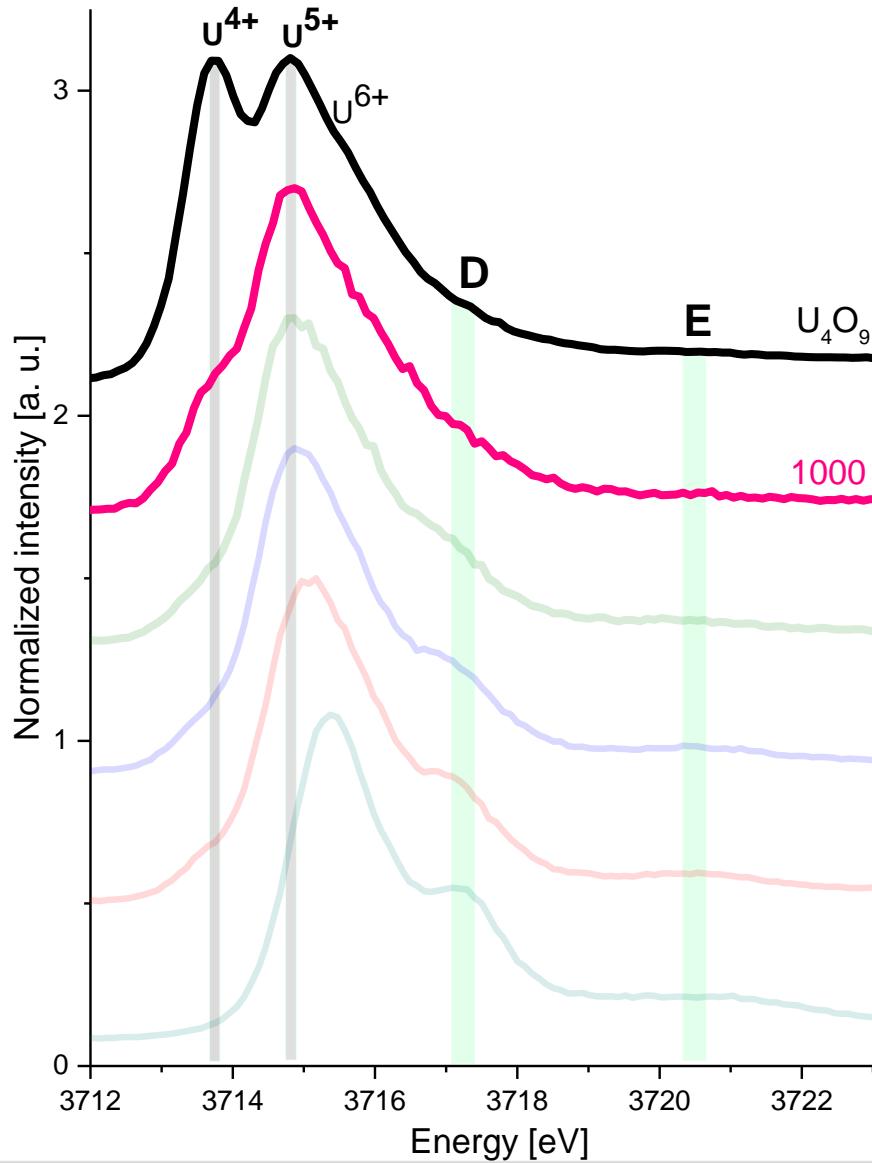


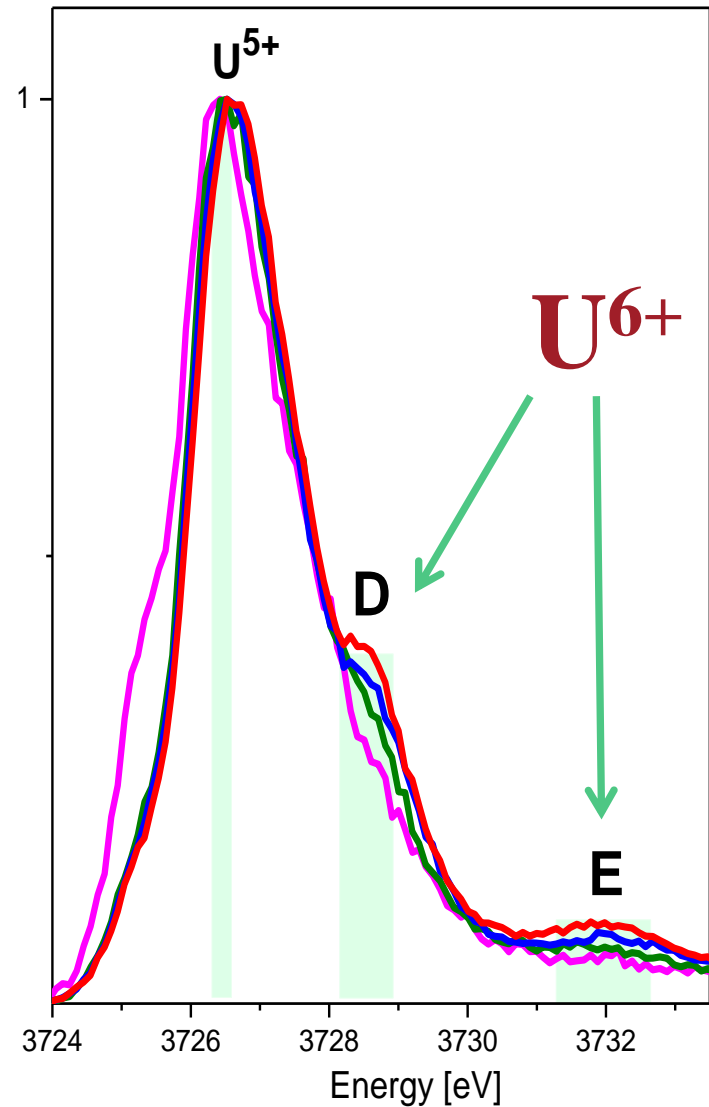
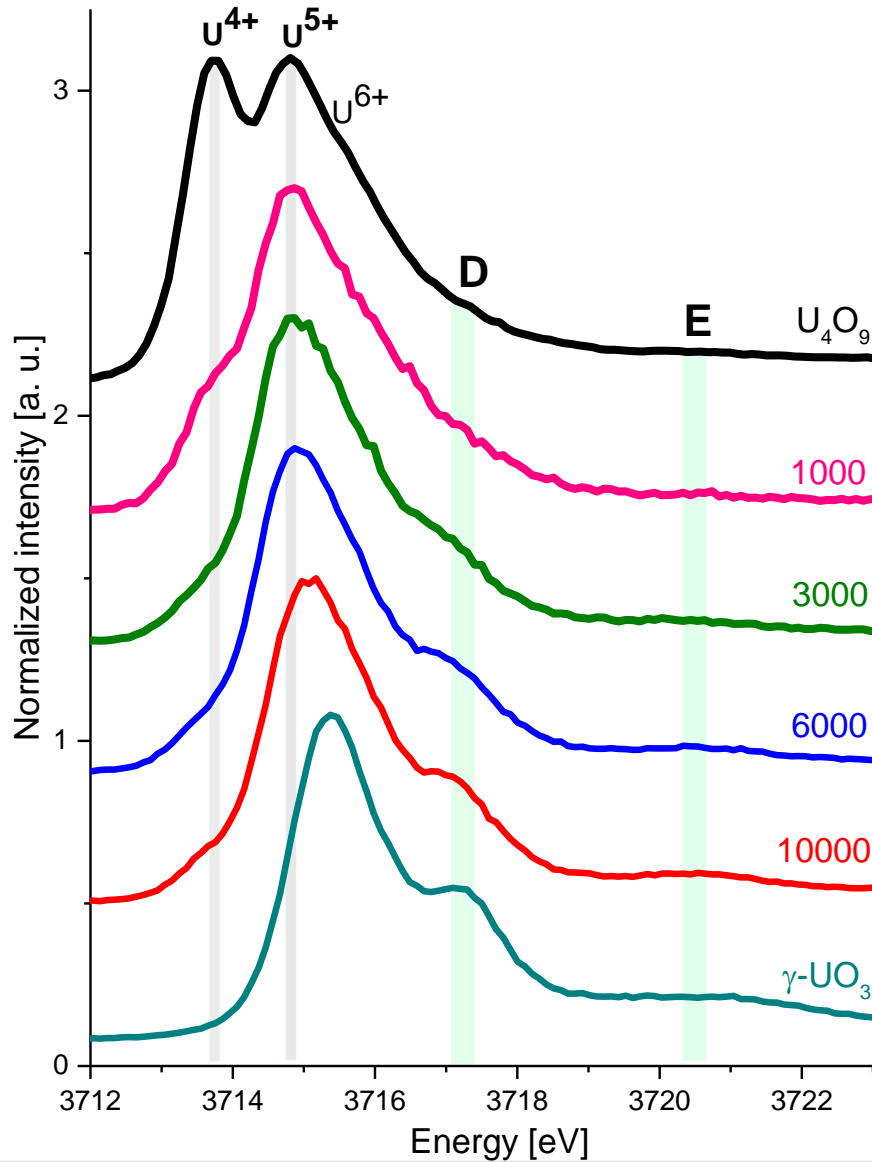
ID26 - High-Brilliance X-ray Spectroscopy

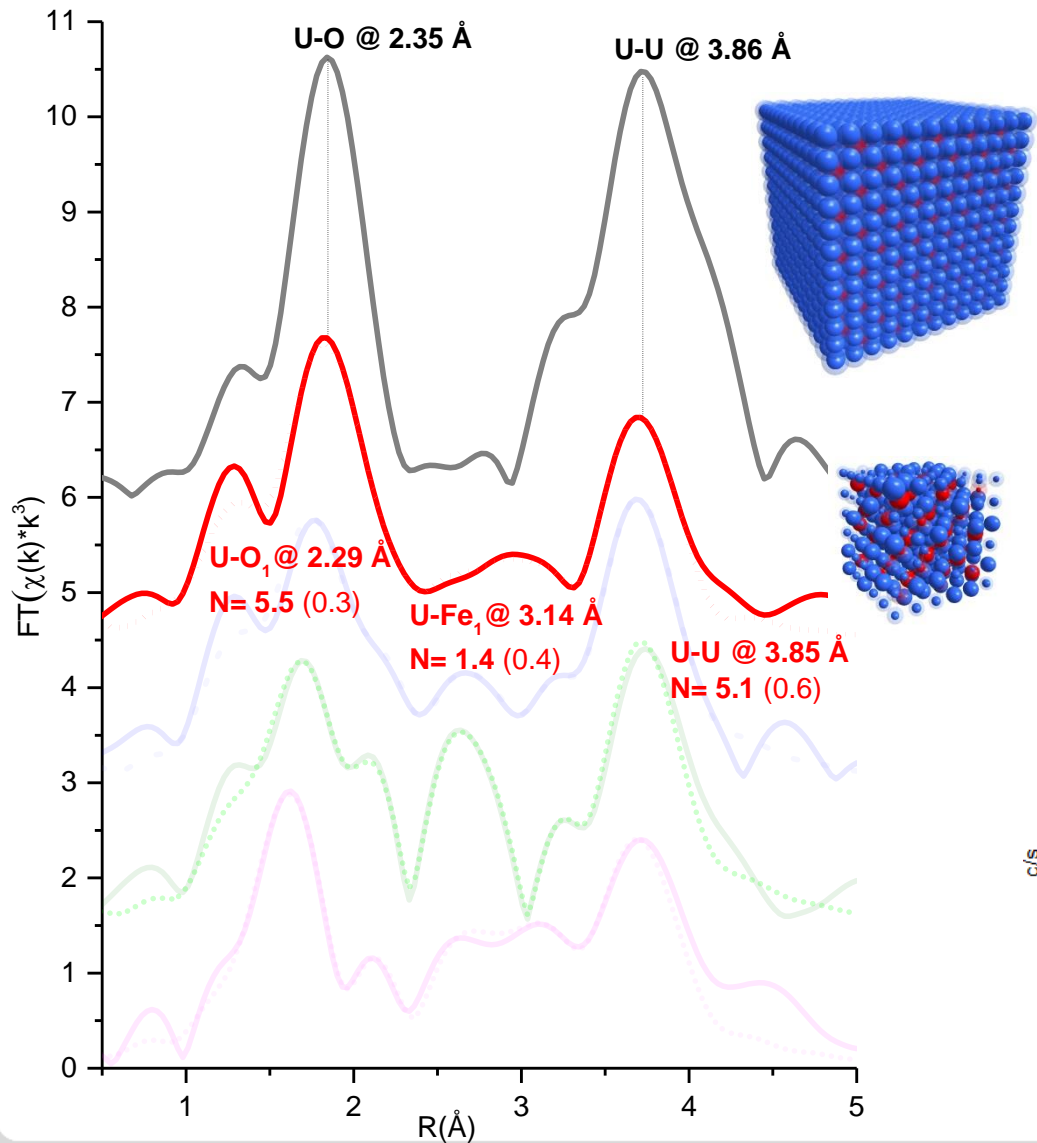


U M₄ XES/XAS principle

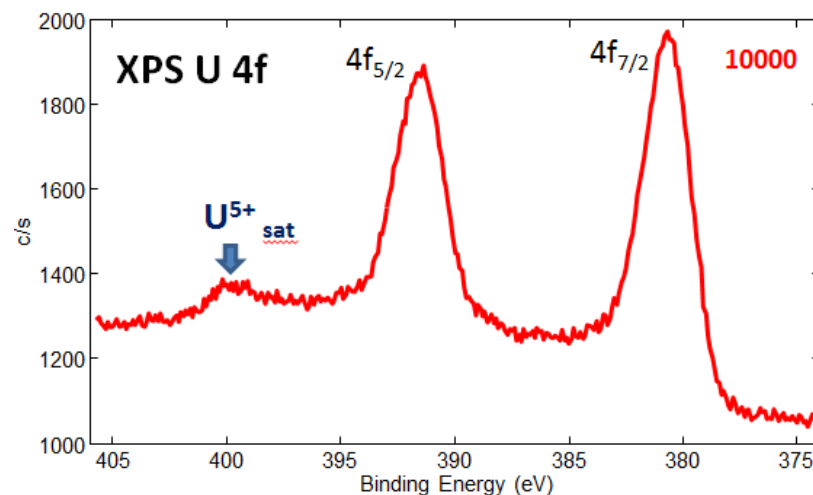
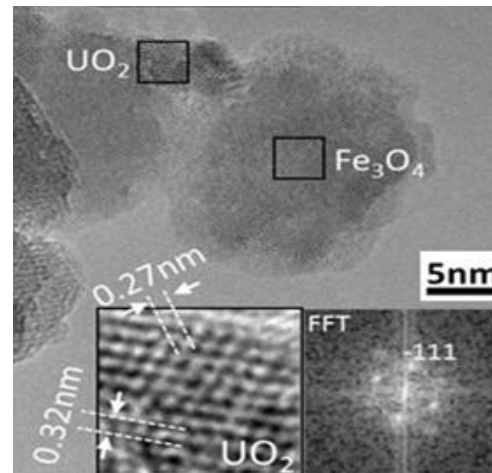


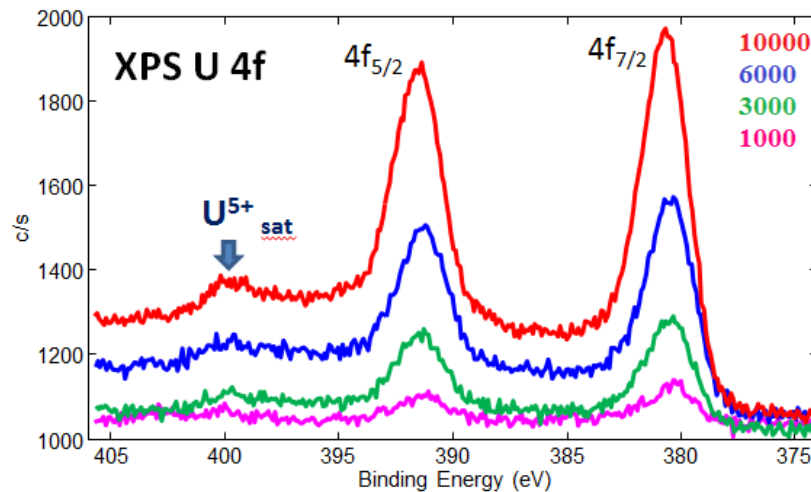
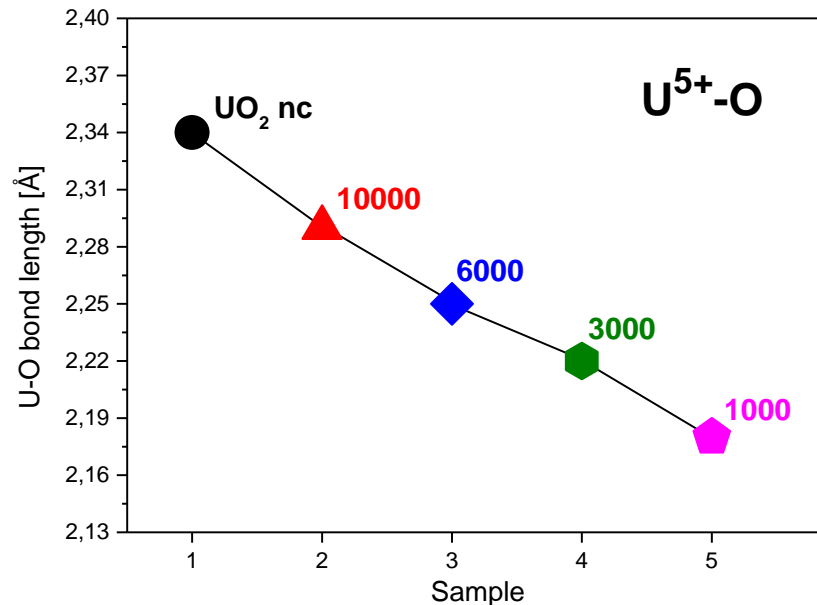
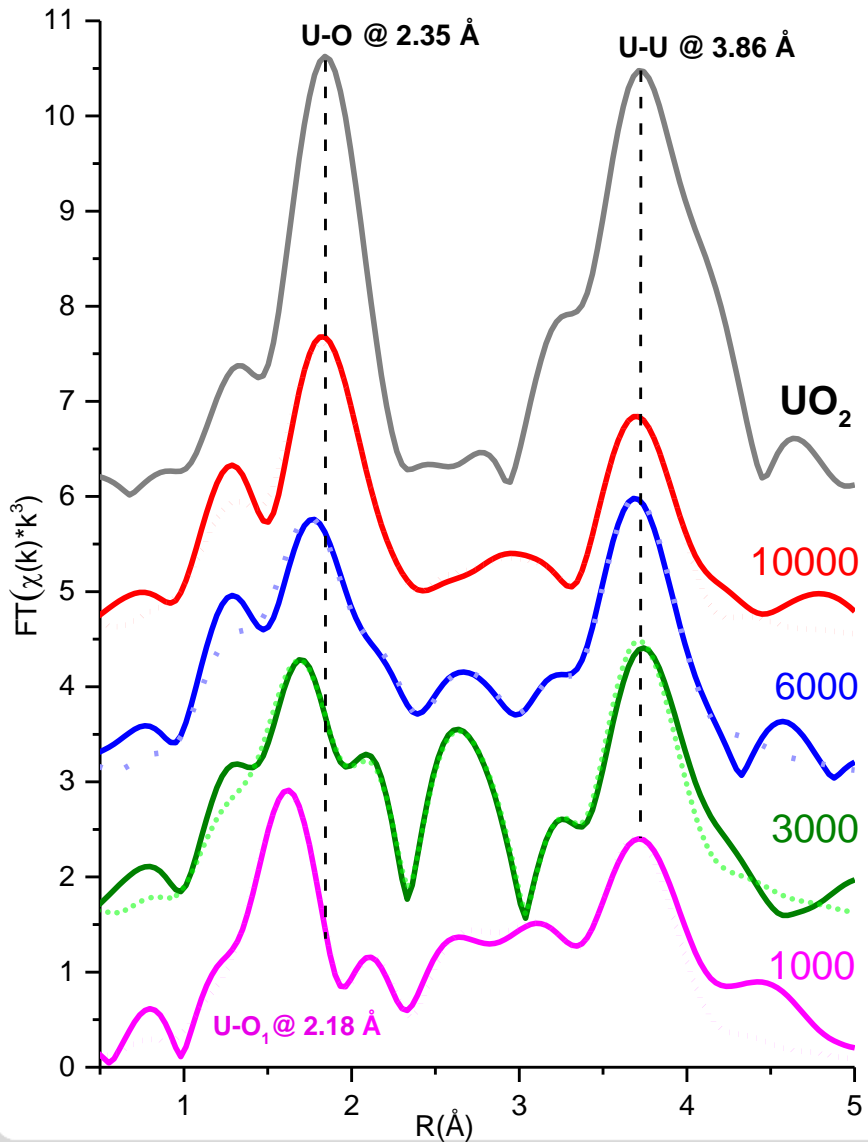




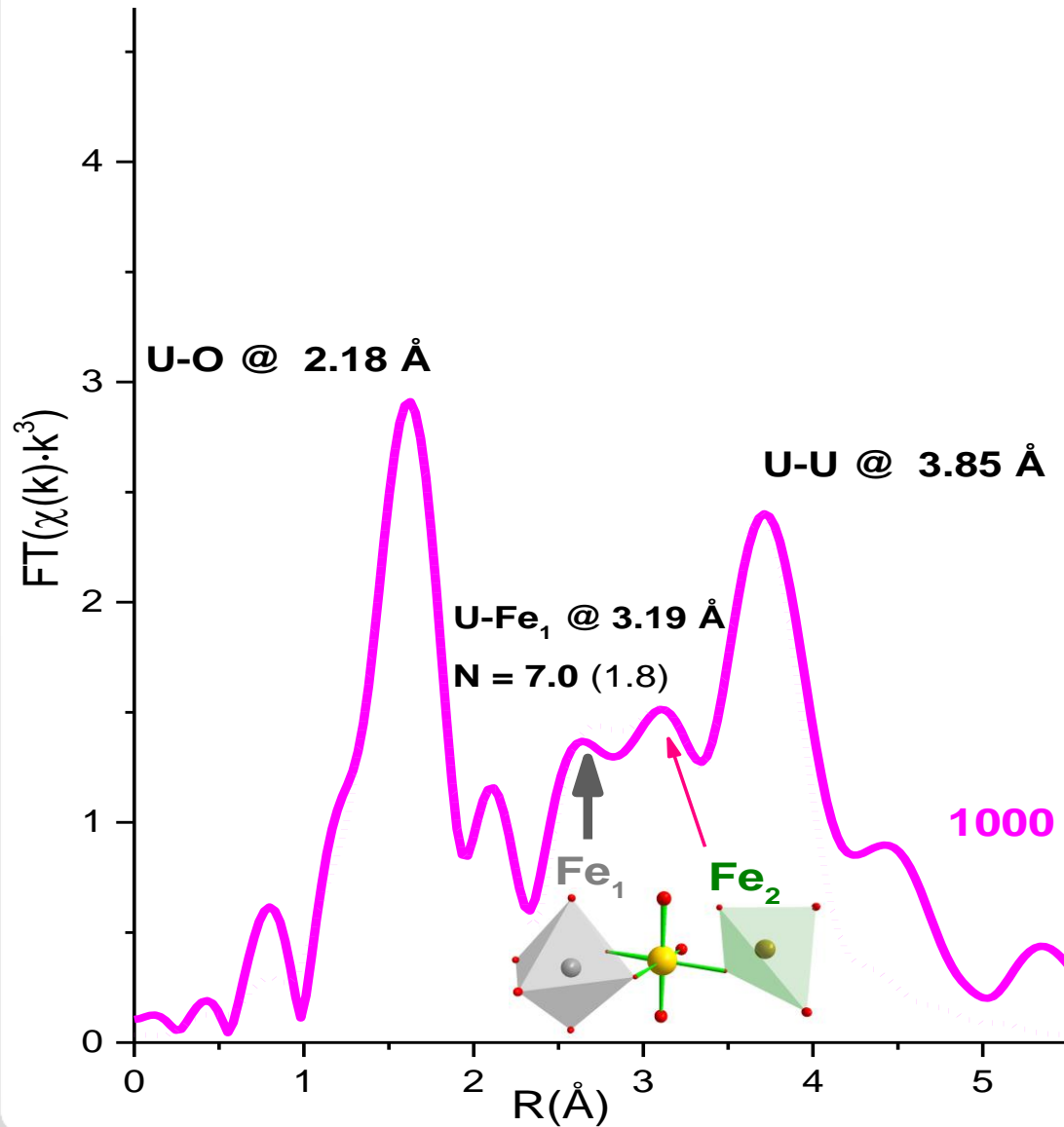


HR-TEM on **10000** (UO_{2+x})

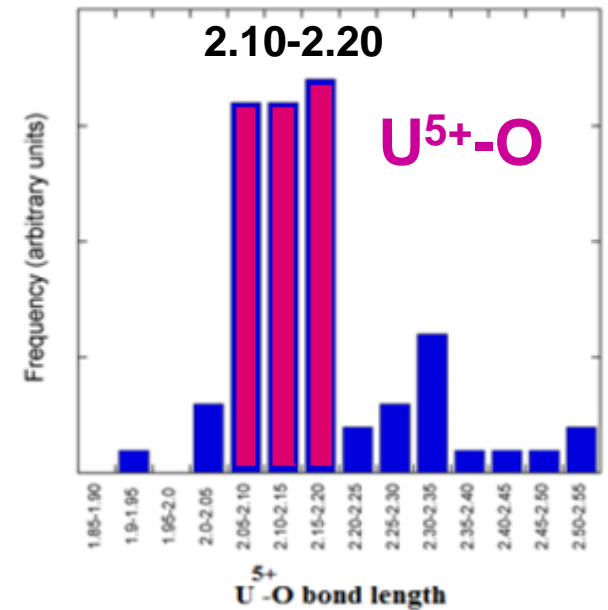
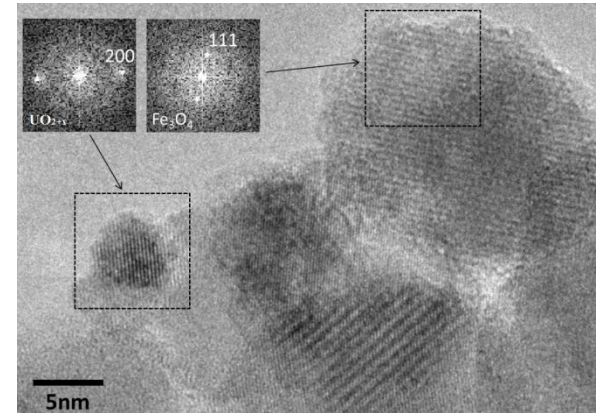




U L₃ XAFS/ ~200 days

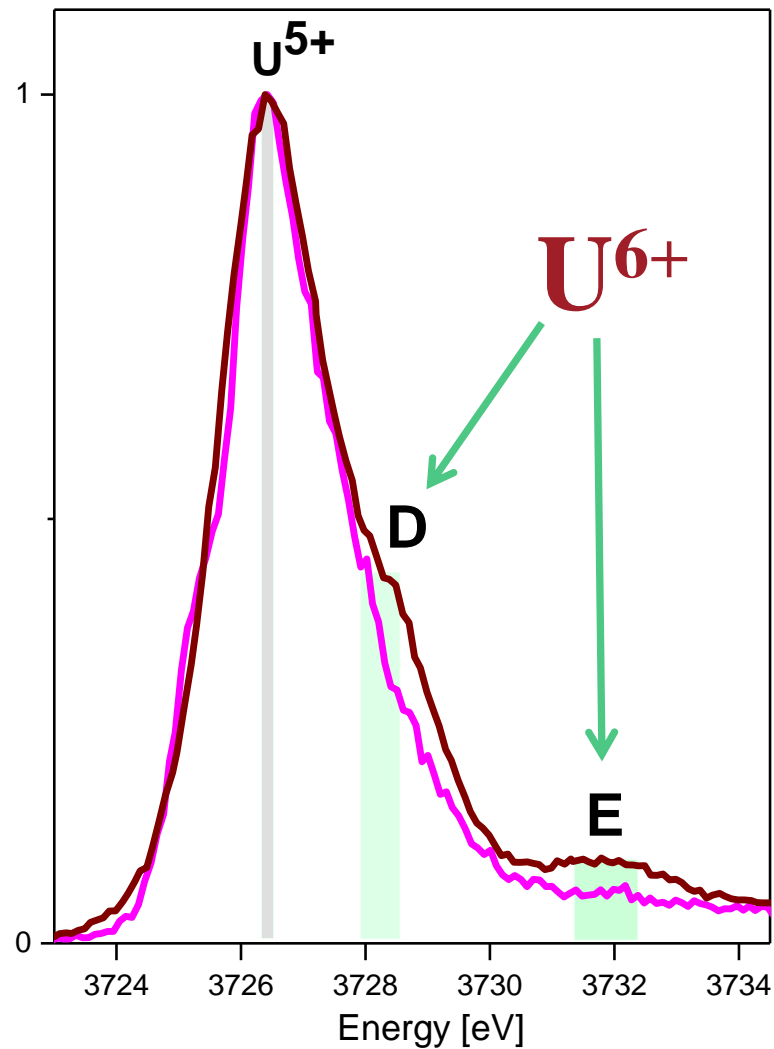
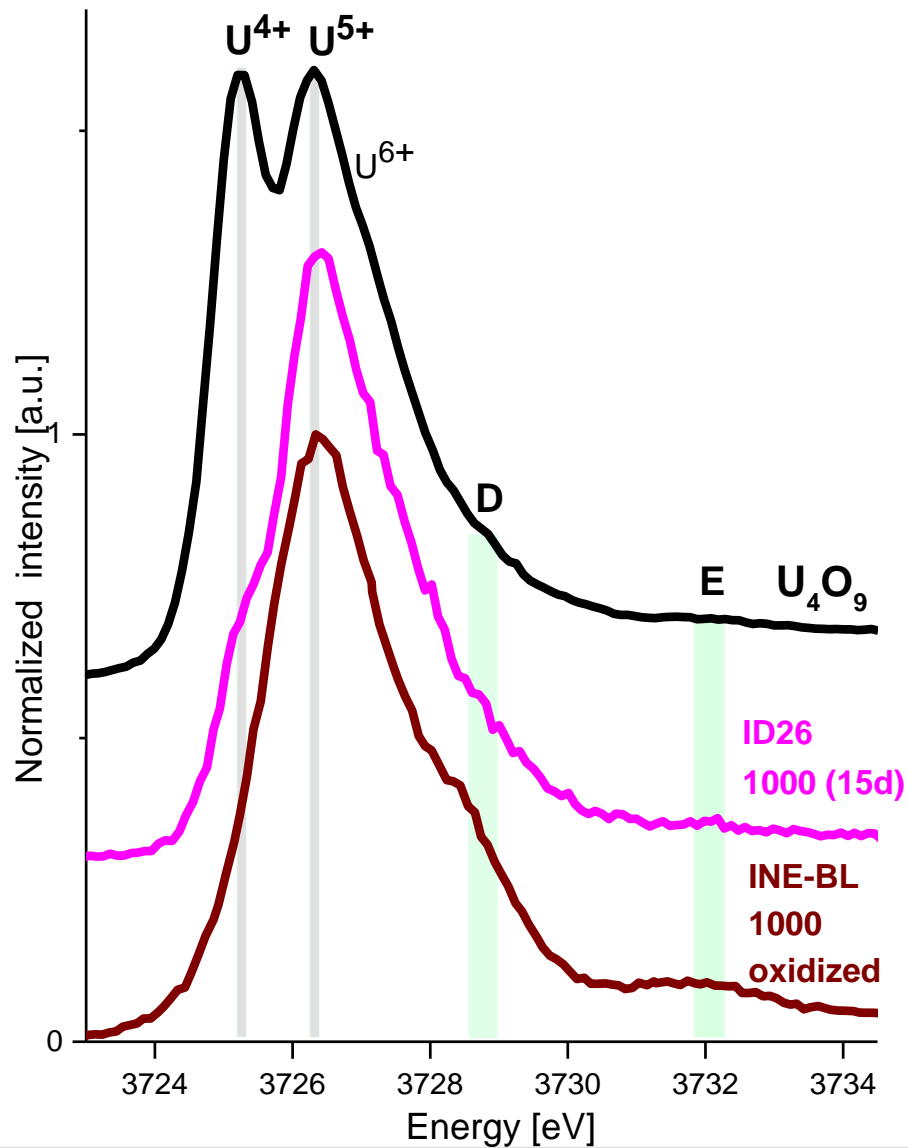


HR-TEM 1000 (UO_{2+x})

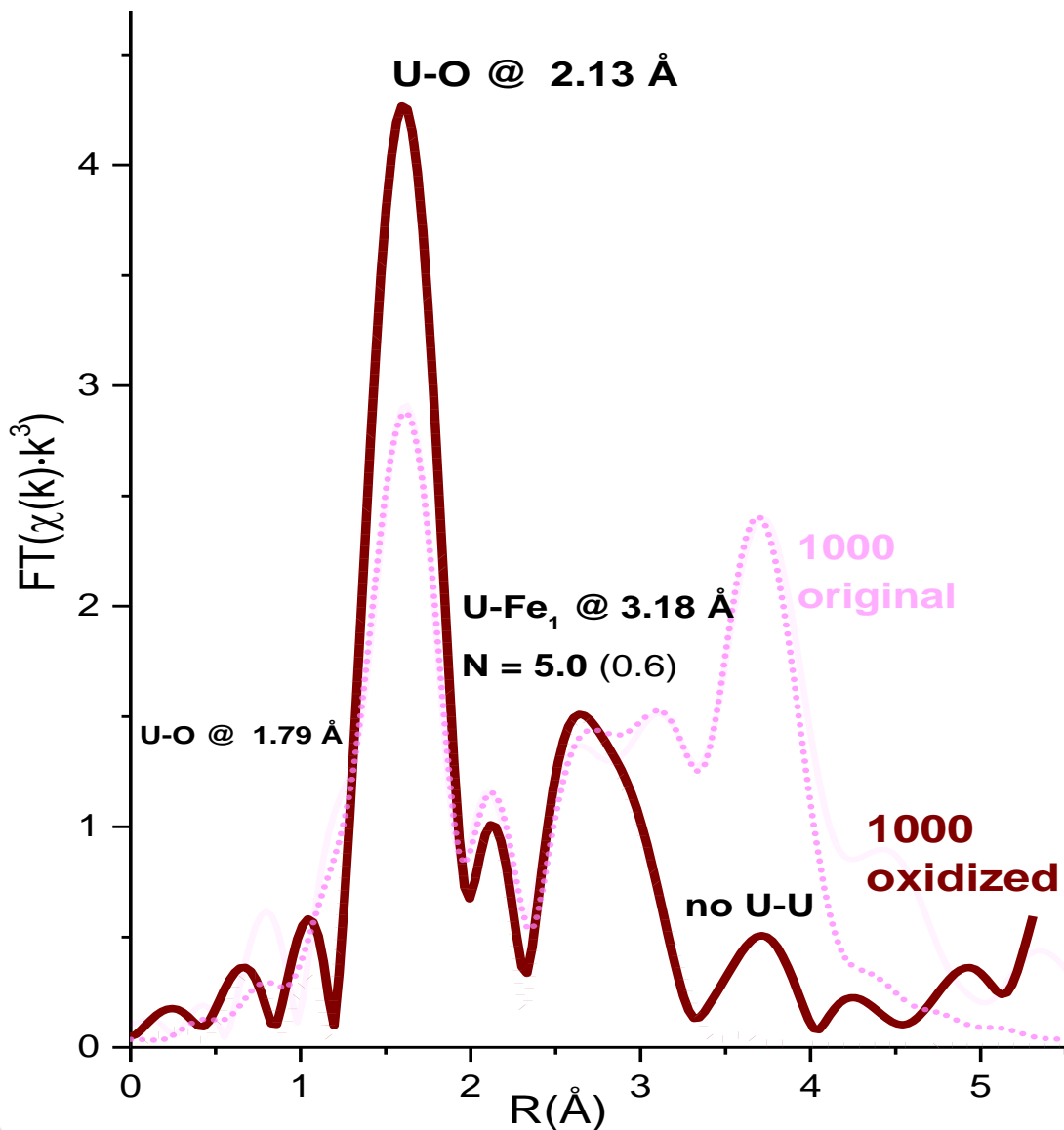


Ilton et al, EnvSci Tech 2010

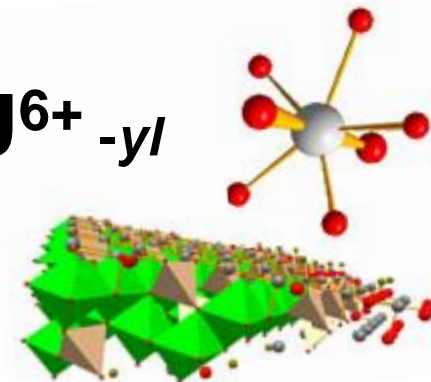
U M₄ HR-XANES / ~150 days on air



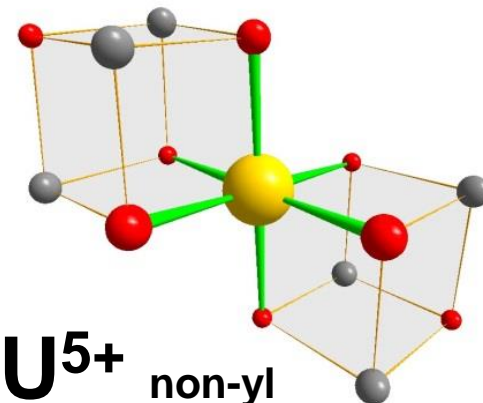
U L₃ XAFS/ ~250 days on air



~ 40% U⁶⁺ -yl

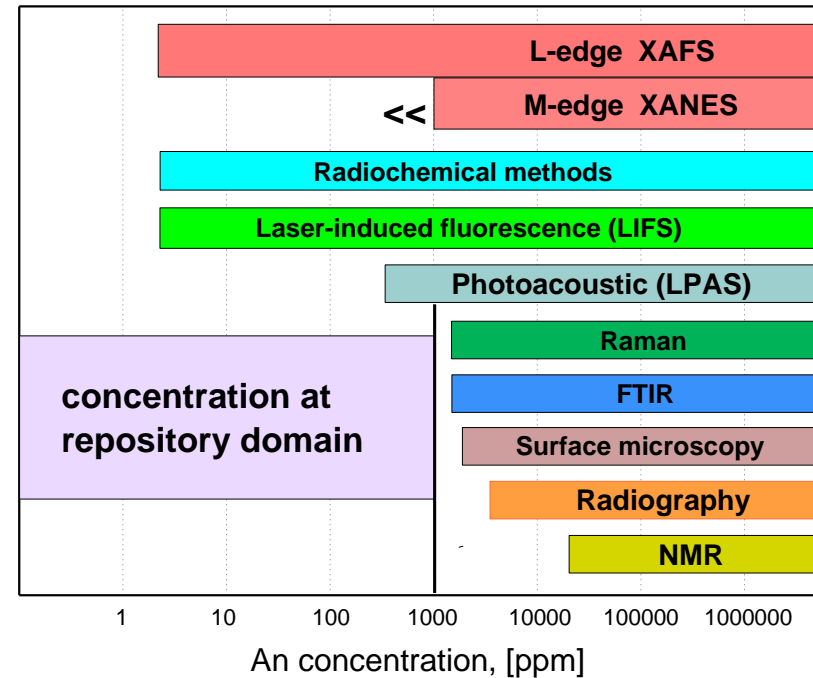


~ 60% U⁵⁺ non-yl
incorporated
octahedral site of Fe₃O₄



CONCLUSIONS

- ❑ **HR-XANES is an emerging tool for speciation studies of An with concentrations relevant to HLW repository**
- ❑ **The HR-XANES allows to resolve several U oxidation states in mixtures**
- ❑ **U⁶⁺ kinetics can be studied provided that the experimental conditions are the same**
- ❑ **Pentavalent U is stable on to air when incorporated into the structure of magnetite**



*Readopted from
W. Runde LA Science*

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Jörg Göttlicher
Ralph Steininger



Kristina Kvashnina
André Rossberg



Yulia Podkovyrina

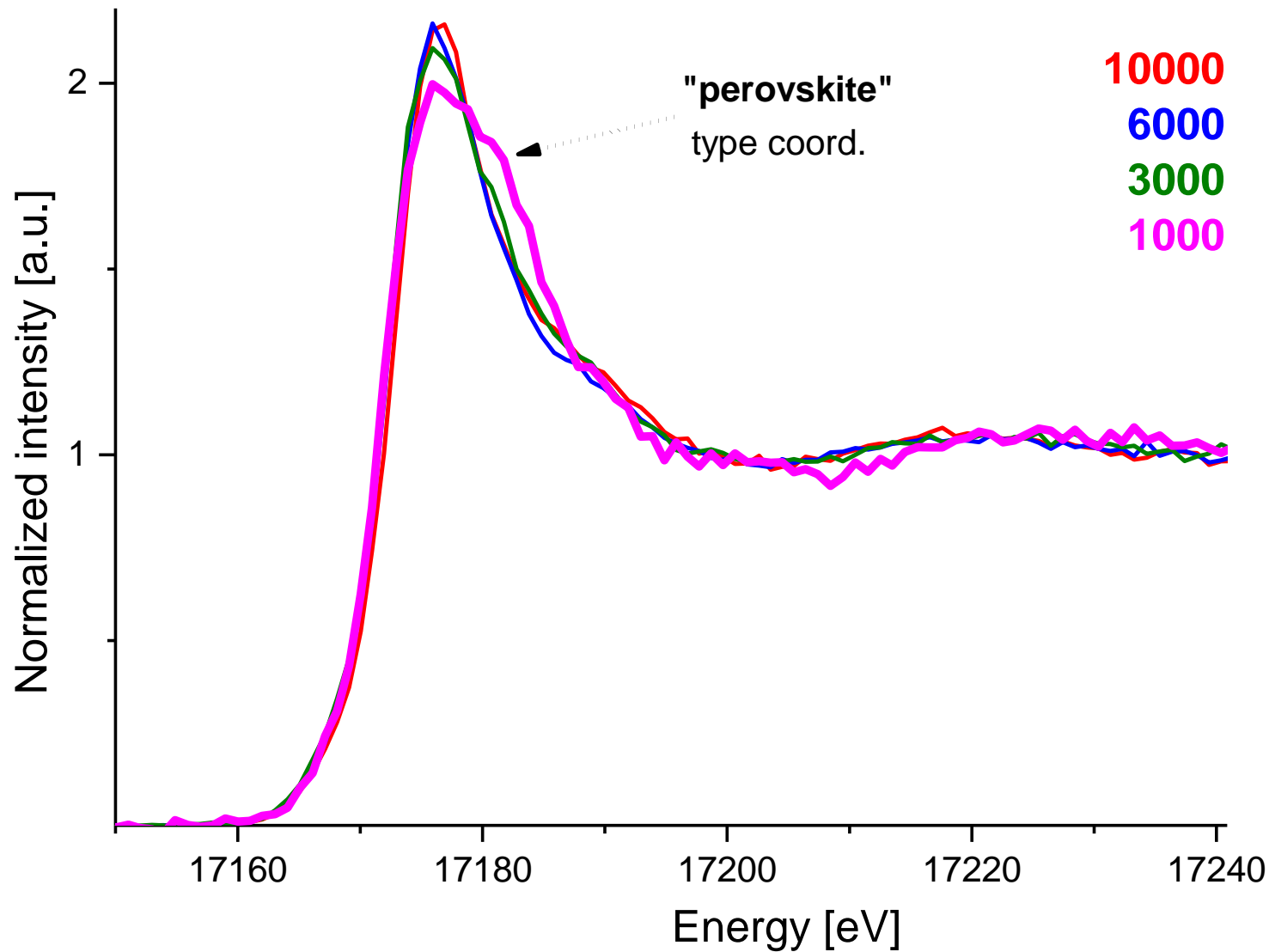
THANK YOU !



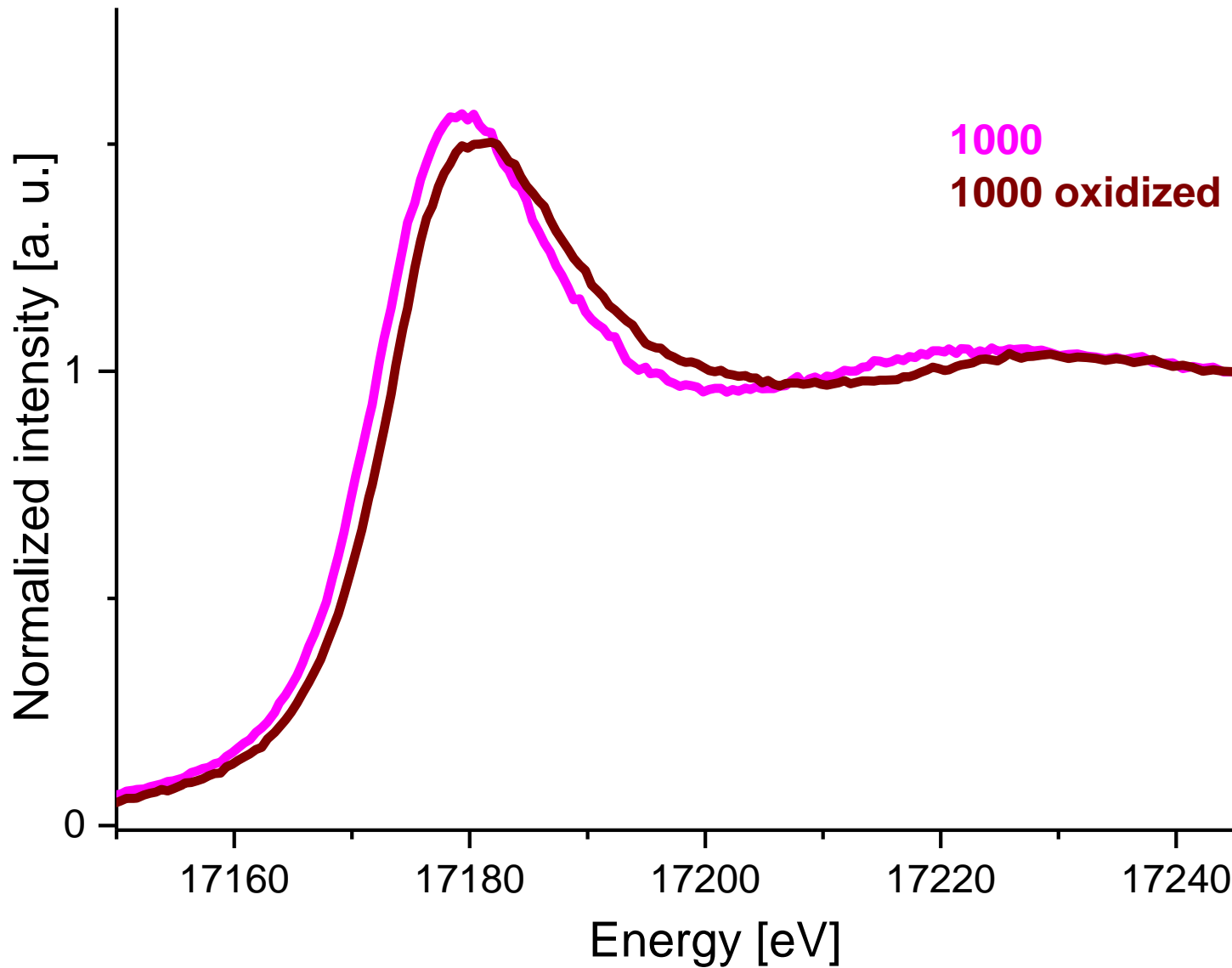
HYIG grant VH-NG-734

Supporting information

U L₃ HR-XANES / 15 days



U L₃ XANES / 200 days + oxidized



EXAFS fit results for 1000-10000 XAFS

sample	Scater Path	N	R (Å)	DW (Å ²)	ΔE_0 (eV)	S_0^2 (aver. from 4 UO2 fits)	r, (chi ²)
1000	U-O3	0.2 ± 0.1	1.73 (3)	0.006 (1)	6.6 (5)	0.77	0.008
	U-O1	3.0 ± 0.3	2.17 (1)	0.006 (1)			
	U-O2	1.7 ± 0.1	2.38 (2)	0.006 (1)			
	U-Fe _{oct}	7.0 ± 1.8	3.18 (2)	0.020 (2)			
	U-Fe _{tet}	0.6 ± 0.3	3.55 (3)	0.001 (1)			
	U-U1	2.9 ± 0.7	3.85 (2)	0.002 (1)			
1000 on air	U-O3	0.8 ± 0.2	1.79 (2)	0.004 (1)	5.0 (8)	0.77	0.008
	U-O1	3.5 ± 0.2	2.13 (1)	0.004 (1)			
	U-O2	0.8 ± 0.1	2.41 (1)	0.004 (1)			
	U-Fe _{oct}	5.0 ± 0.6	3.19 (1)	0.016 (2)			
3000	U-O3	0.3 ± 0.2	1.70 (1)	0.006 (1)	7.4 (3)	0.77	0.003
	U-O1	2.8 ± 0.3	2.22 (1)	0.006 (1)			
	U-O2	2.4 ± 0.2	2.43 (1)	0.006 (1)			
	U-Fe _{oct}	1.9 ± 0.4	3.11 (1)	0.006 (1)			
	U-Fe _{tet}	0.6 ± 0.3	3.34 (2)	0.006 (4)			
	U-U1	3.2 ± 0.7	3.84 (3)	0.002 (1)			
6000	U-O3	0.4 ± 0.2	1.69 (2)	0.006 (1)	7.0 (4)	0.77	0.003
	U-O1	2.7 ± 0.3	2.25 (1)	0.006 (1)			
	U-O2	2.1 ± 0.2	2.43 (1)	0.006 (1)			
	U-Fe _{oct}	4.8 ± 1.6	3.14 (2)	0.023 (2)			
	U-Fe _{tet}	0.6 ± 0.3	3.50 (2)	0.005 (3)			
	U-U1	3.4 ± 0.7	3.83 (3)	0.002 (1)			
10000	U-O3	0.4 ± 0.2	1.68 (2)	0.006 (1)	7.3 (3)	0.77	0.003
	U-O1	2.9 ± 0.2	2.29 (1)	0.006 (1)			
	U-O2	2.1 ± 0.2	2.44 (1)	0.006 (1)			
	U-Fe _{oct}	1.2 ± 0.4	3.14 (2)	0.013 (4)			
	U-Fe _{tet}	0.6 ± 0.2	3.50 (2)	0.004 (3)			
	U-U1	5.2 ± 0.8	3.83 (3)	0.006 (1)			

