

Hydrogen Uptake of Zry-4 during Reaction in N₂/H₂O Atmosphere in the Temperature Range of 600 - 1100°C

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Outline



- Introduction
- Annealing of zirconium alloys in steam/nitrogen atmosphere
- Neutron radiography
 - Measurements
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- Conclusions

Introduction



- High temperature oxidation of the zircaloy claddings in nitrogen/steam atmosphere can occur for instance:
 - During severe accidents in BWR where the containment is inerted by nitrogen
 - Spent fuel pool (SFP) accidents (The hydrogen released consumes the oxygen from the air and steam/nitrogen atmosphere remains at the fuel rods).

The reaction

 $2 Zr + N_2 = 2 ZrN$ (very simplified)

occurs only under oxygen and steam starvation conditions.

Introduction



Porous oxide scale is formed:



Do we get an enhanced hydrogen uptake like it happens during the so called breakaway oxidation at around 1000°C?

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Annealing of zirconium alloys in steam/nitrogen atmosphere



INRRO furnace

2 Zry-4 cladding tube segments (2 * 10 mm length) per annealing

Loading and withdrawal of the samples at test temperature

Cool down to nearly RT during about 5 min.

Atmospheres:

- pure steam
- 99 % steam + 1 % N₂
- 20 % steam + 80 % N₂.



Neutron radiography investigations



X-ray radiography



neutron radiography



GKSS Geesthacht 1991

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Neutron radiography investigations



facility	ICON
neutron source	SINQ
institution	PSI Villigen
resolution	~ 25 µm
time per image	300 s







Basis of neutron radiography



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Basis of neutron radiography



Correction of the effect of oxygen:

$$\Sigma_o = \frac{a}{\delta_m S} \cdot \Delta m = 0.32 \ g^{-1} \cdot \Delta m$$

(former work: M. Grosse et al.; Nucl. Instr.& Meth. In Phys. Res. A 651, (2011), 253)

Correction of the effect of nitrogen:

Using the samples annealed in synthetic air:

$$\Sigma_N = 0.60 \ mm^{-1} \cdot \overline{\delta_{ZrN}}$$



mean thickness of the nitride containing layer

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Results (600°C)





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Results (800°C)





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Results (1000°C)





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Results (1100°C)





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Discussion





"Hydrogen pump" effect known for breakaway oxides:

- Steam penetrates into the cracks and reaches the oxide/metal interface
- Steam reacts at the interface, hydrogen remains in the cracks resulting in a high hydrogen partial pressure



 Higher hydrogen partial pressure results in a stronger hydrogen uptake by the metallic Zry (Sieverts law). The total pressure in the cracks decreases. New steam is sucked into the cracks and the process continues at the beginning.

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Summary and Conclusion



- Nitrogen in steam strongly enhances oxidation and hydrogen uptake of Zircaloy-4.
- The effect is very strong at 1000 and 1100°C. Only some minutes annealing is needed to start the enhanced reaction.
- The formation of nitrides is accompanied by an strong increase in the hydrogen concentration. The effect is comparable with the effect occurring during breakaway oxidation
- Additionally to the faster degradation by the enhanced reaction, strong hydrogen embrittlement occurs.
- More details and discussions (e.g. hydrogen uptake fractions, influence of the total gas flow) are given in the proceedings.





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Thanks for your attention, questions?

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