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ELECTRON MICROSCOPY OF ANIONIC SURFACTANT-DIRECTED SYNTHESIS OF MAGNETITE NANOPARTICLES

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Abstract. Here we report new work on the synthesis and an electron microscopy study of the earliest known magnetic material, magnetite (Fe_3O_4). We have synthesized a variety of magnetite nanoparticles which appear to have biogenic signatures and could give insights into how the nanomagnetite in biological systems forms, and how they may be associated with Alzheimer's disease. We have also synthesized mesoporous magnetite nanoparticles which have potential use in the targeted drug delivery.

Keywords: magnetite, magnetic nanoparticles, hydrothermal synthesis, drug delivery.

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Introduction

Magnetite (Fe_3O_4) in the form of lodestone is the original magnetic material used by mankind to navigate the Earth by following geomagnetic field lines. It is also used for navigation in Nature by honey bees, homing pigeons and bacteria. Currently, magnetic nanoparticles are used for data storage, magnetic resonance imaging (MRI) and are the potential colloidal mediators for cancer magnetic hyperthermia. This area of nanotechnology spans the boundaries of chemistry, biology, medicine and materials science.

The earliest reported medical use of magnetite powder for internal applications was in the 11th century A.D. by the Persian physician and philosopher Avicenna of Bokhara [1,2]. Today magnetic nanoparticles are used for magnetic resonance imaging (MRI) and as colloidal mediators for magnetic hyperthermia cancer treatment [3].

In 1992, it was reported that magnetite (Fe₃O₄) is present in the human brain [4] and more recently that nanoscale biogenic magnetite is associated with neurodegenerative diseases such as Alzheimer's [5]. Although this association was first discovered over fifty years ago by Goodman [6], to date the magnetite's origin and formation are not certain. Here we show that the synthesis of magnetite in the presence of the surfactant sodium dodecyl sulphate (SDS) gives rise to a variety of nanoscale morphologies, some of which look remarkably similar to magnetite found in organisms, suggesting that similar processes may be involved. Furthermore, these 0D materials with nanoscale dimensions and with the potential to show quantum confinement effects are of interest in the areas of biosensors [7], biomedical imaging [8] and as tunneling magnetoresistance (TRM) sensor elements in magnetic random access memory (MRAM) configurations, which can be used to form magnetic, programmable biochips [9].

In the more extensively studied magnetotactic bacteria [10], biogenically controlled morphologies of magnetite, which normally crystallises in a cubic space group, occur in the form of nanocrystals. These have been found to have elongated {111} axes often with rhombic or truncated dodecahedral morphology. They are part of intracellular structures called magnetosomes [11], which consist of a nanocrystal surrounded by a lipid bilayer membrane. They arrange in a 1D array to form a "compass needle", which is utilised by the bacteria to sense the Earth's magnetic field [12,13]. Figure 1 shows two typical examples of these magnetotactic bacteria [14].

Taking this lead from biology, where the biogenic templates produce magnetite of defined shapes and sizes, we have been interested in investigating how surfactant molecules could similarly influence magnetite formation.



Figure 1. Two examples of magnetite in magnetotactic bacteria. Left: *Magnetospirillum magnetotacticum* (whole organism) showing the location of magnetoite crystals and Right: an example of a string of bulletshaped crystals within a magnetotactic bacterium [14], Copyright (2008) Federation of European Microbiological Societies.

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Experimental *Materials*

All chemicals were reagent grade and used without further purification. Ferrous chloride tetrahydrate (FeCl₂·4H₂O > 99%), sodium hydroxide (NaOH > 97%) and sodium dodecyl sulphate (SDS > 99%) were purcheased from Sigma-Aldrich. Double distilled water, sodium hydroxide solutions and SDS solutions were deoxygenated by bubbling through N₂ gas for 1 h prior to use. The synthesis of magnetite was performed under nitrogen in an autoclave using a 20 mL Teflon insert.

Compounds 1-5 preparation

To an aqueous solution containing $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ and SDS, a deoxygenated aqueous NaOH solution was added (Table 1). The autoclave was heated to 200 °C for 24 h and after cooling the magnetite was isolated as detailed below.

Reagent quantities and concentrations used to prepare compounds 1-5.		
<i>FeCl</i> ₂ ·4H ₂ <i>O</i> (0.2 <i>M</i>)	NaOH (1 M)	SDS
5.0 mL	2.0 mL	0.5 mL (15 mM)
5.0 mL	1.0 mL	2.0 mL (15 mM)
5.0 mL	1.0 mL	0.5 mL (35 mM)
5.0 mL	1.0 mL	0.5 mL (35 mM)
5.0 mL	1.0 mL	0.5 mL (35 mM)
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The suspended magnetite/surfactant product was put in a sonication bath (Qualilab USR-30 H) for 1 h. The suspension was then diluted with a 20 fold excess of distilled water before being sonicated again for 10 min and centrifuged (Heraeus Instruments, Labofuge 400) at 3500 rcf for 30 min. The samples containing particles less than 100 nm in size were filtered (NC10; Schleicher & Schuell) prior to sonication and centrifugation. The magnetite sediment was suspended in ethanol and sonicated for a few seconds and then "spotted" onto carbon coated copper grids of 400 mesh size for TEM examination and onto 1 cm squares of polished silicon wafers for SEM examination and then dried before analysis.

Equipment

Scanning electron microscopy (SEM) was performed using a LEO 1530 scanning electron microscope operating at 3 KeV. Transmission electron microscopy (TEM) measurements were performed using a Tecnai T20 ST transmission electron microscope operating at 200 KeV.

Results and discussion

In the course of our experiments, we found that hydrothermal conditions can be used to produce octahedral magnetite (Figure 2) as typically observed under geological conditions (Figure 3).





Table 1

Figure 2. Octahedrally shaped magnetite. SEM of magnetite formed hydrothermally without SDS present.

Figure 3. Geological octahedrally shaped magnetite crystallised on feldspar and quartz found in Bolivia (photo by S. Malik).

However, when SDS (Figure 4) is present, we found instead a variety of magnetite morphologies, which appear to have biogenic trademarks. SDS (Figure 4) is commonly used to mimic hydrophobic binding environments such as cell membranes [15] and has been used to study the folding and thermal stability of cytochrome c (cyt c), which is a biologically important electron transfer system [16].



Figure 4. The anionic surfactant SDS.

The synthesis of magnetite in the presence of SDS under hydrothermal conditions yielded a variety of magnetite morphologies, as revealed by SEM and TEM. We discuss five of these morphologies here.

Firstly, bullet shaped needles with an aspect ratio of 1:6-8 and 1-2 μ m in length with square-pyramidal endings (system 1, Figure 5) can be observed as aligned bundles and rather unusually these crystals frequently grow with a 90° bend. Acicular magnetite and maghemite with similar aspect ratios and sizes are commonly used for magnetic media devices. However, the route of synthesis of these crystallites is not straightforward and typically involves the synthesis of goethite (α -FeOOH) or lepidocrocite (γ -FeOOH), followed by dehydration to form hematite (α -Fe₂O₃) and then reduction to magnetite. Then, this can be re-oxidised to give maghemite [17,18].

Arrays of magnetite octahedra (system 2, Figure 6) were observed having 80-100 nm dimensions and forming 1D stacks of up to 4 μ m in length. In 2008, Schumann *et al.* reported magnetofossils derived from magnetotactic bacteria which, as seen from their SEM images and drawings depicted in Figure 7, show similar morphological features with spearhead-like magnetite particles.



Figure 5. Bullet shaped needles which are frequently observed with a 90° bend.



Figure 6. SEM of ordered 1D array of nanocrystallites.



Figure 7. Magnetofossils derived from magnetotactic bacteria [19]. Copyright (2008) National Academy of Sciences, USA.

Whiskers with an aspect ratio of 1:10 (system 3, Figures 8 (*a*) to (*d*)), typically around 100-200 nm in length, are also formed and they self-assemble into 2D hexagonal plates and 3D "flower" motifs.



(d) TEM detail of whisker tip.

String-like crystallite assemblies are observed (system 4, Figure 9) with the strings composed of nanocrystals less than 10 nm in diameter arranged in helical bundles of 70-200 nm in diameter and up to 5 μ m in length. These are reminiscent of the magnetite spheres reported by Hou *et al.* [20], which were prepared hydrothermally by reducing [Fe(acac)₃] in the presence of various surfactants, and is the only other report of such an unusual form of magnetite produced hydrothermally. Finally, mesoporous nanocrystallites form (system 5, Figure 10 (*a*) and (*b*)) with diameters of 40-100 nm and morphology similar to the widely reported truncated dodecahedral shape seen in certain magnetotactic bacteria [12]. These nanocrystallites, as observed for system 3, also seem to self-align. The "question-mark" motif seen in Figure 10 (*a*) is typical.





Figure 9. SEM of string-like assemblies composed of nanocrystallites (the insert shows a TEM image).

Figure 10. (a) TEM overview and (b) TEM detail of mesoporous magnetite nanocrystallites.

It has been reported that magnetosome vesicles are present before magnetite formation in magnetotactic bacteria [21] and we can draw a parallel to this situation given the presence of SDS micelles providing a reaction space for formation of nanocrystalline magnetite arrays. In addition to its "biogenic signature", this mesoporous nanomagnetite may prove useful for targeted drug delivery [22].

Conclusions

In Nature, the synthesis of the bacterial magnetosome seems to be a complex process that involves several discrete steps, including magnetosome vesicle formation, iron uptake by the cell, iron transport into the magnetosome vesicle and controlled Fe_3O_4 biomineralization within the magnetosome vesicle. In general, there has been little research regarding the methods by which bacteria synthesize these mineral crystals in specific morphologies.

In our experiments, we have shown that the synthesis of magnetite in the presence of the surfactant sodium

dodecyl sulphate gives rise to a variety of nanoscale morphologies, some of which look remarkably similar to magnetite found in organisms, suggesting that similar morphological control processes may be involved.

Furthermore, these 0D materials with nanoscale dimensions are of interest in the areas of biosensors and biomedical imaging. The magnetite whiskers, in particular, are ideally suited for use in "lab-on-a-bead" devices, which have whisker morphology and are set to be the next-generation lab-on-a-chip devices. They are both cheaper and better suited to use in field diagnostics and theranostics. In addition, the whiskers are the ideal shape for use as tunneling magnetoresistance (TRM) sensor elements in magnetic random access memory (MRAM) configurations, which can be used to form magnetic, programmable biochips.

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