

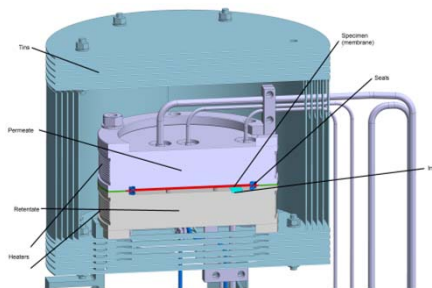
Permeation data analysis including a nonzero hydrogen concentration on the low pressure detector side

F. Arbeiter, D. Klimenko (Experiment QMS), K. Nagatou (analytische Lösungen der DGL.) V. Pasler (Open Foam, Simulation), G. Schlindwein (Experiment), A. von der Weth (Literatur, Theorie)

Karlsruhe , Computational Science and Mathematical Methods, 13th March, 2019

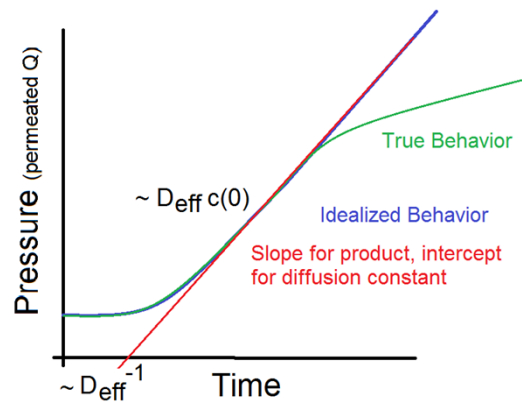
IKIT/INR/MET (Maschinenbau)

Determination of concentration dependent interstitial diffusion parameter regarding re-diffusion and small loading pressure: Situation of future fusion power plant 2 Pa tritium partial pressure (breeder unit) enriched to 1 Pa in purge gas system.



- 1.: Motivation
- 2.: Description of setup and simplification
- 3.: Status quo: FDM solver
- 4.: Status quo: Branch & Bond algorithm
- 5.: Desired Object: Gas release
- 6.: Other Projects

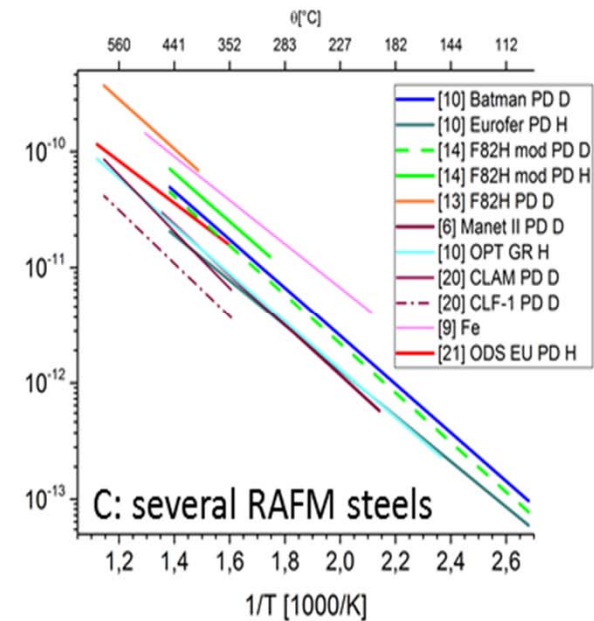
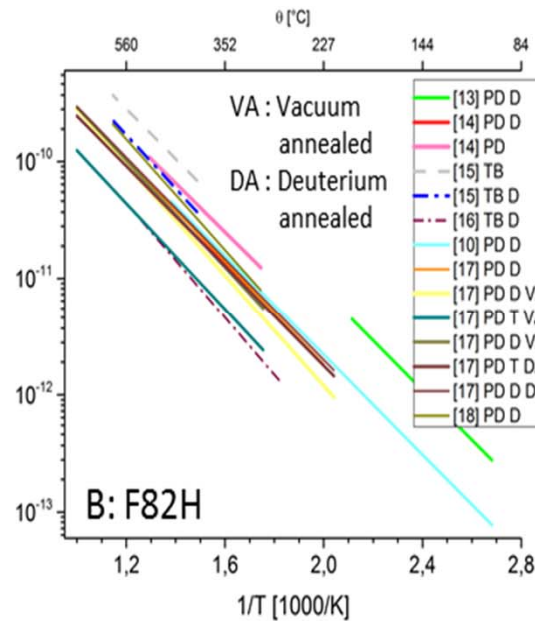
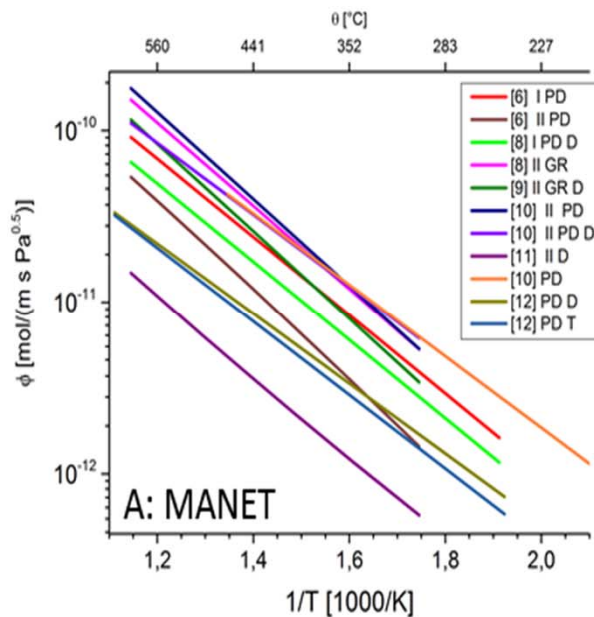
1.: Motivation



Time dependent non zero Q-concentration near measuring system (gauge or QMS) generates deviation from linear behavior.

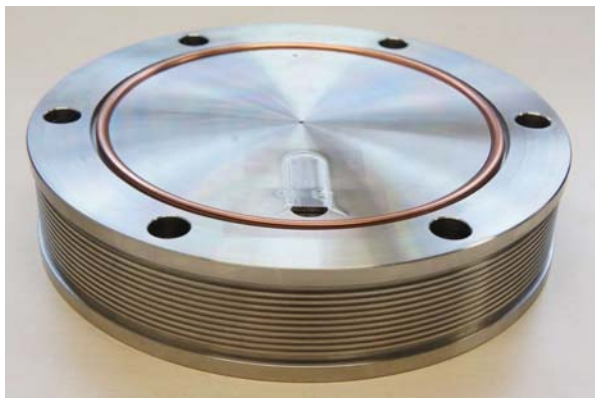
$$j(t)_{measure} = \frac{\Phi}{d_m 4 \pi} \frac{D_{eff} k_s \sqrt{p_{load}} w_m^2}{j_{steady\ state}} \left(1 + 2 \sum_{k=1}^{\infty} (-1)^k e^{-\frac{k^2 \pi^2 D_{eff} (t-t_{off})}{d_m^2}} \right)$$

Daynes, Forcey transport equation solution

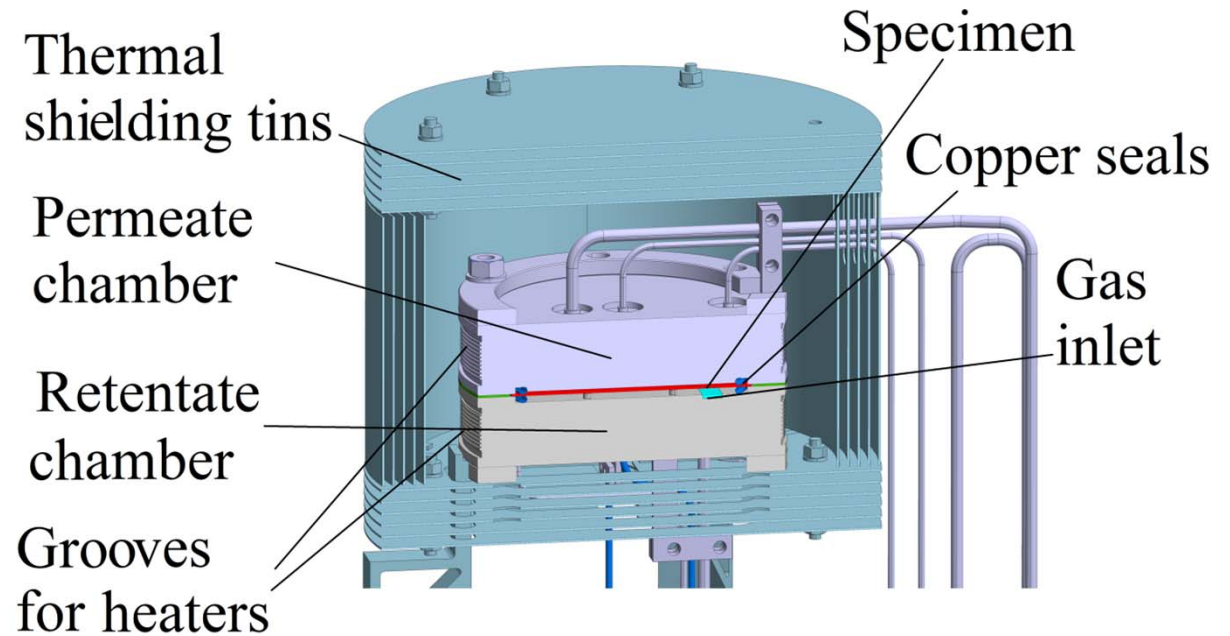


2.: Description of setup and simplification

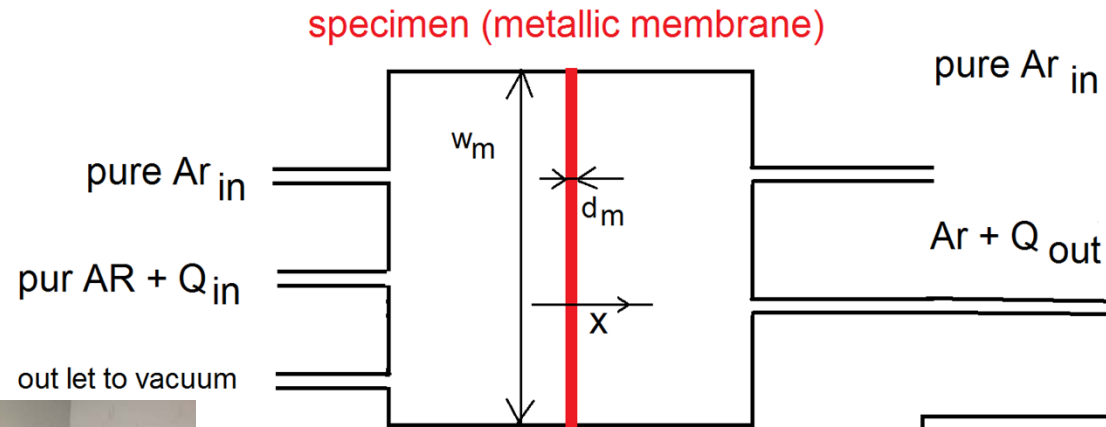
Therefore removing Q in permeate chamber.



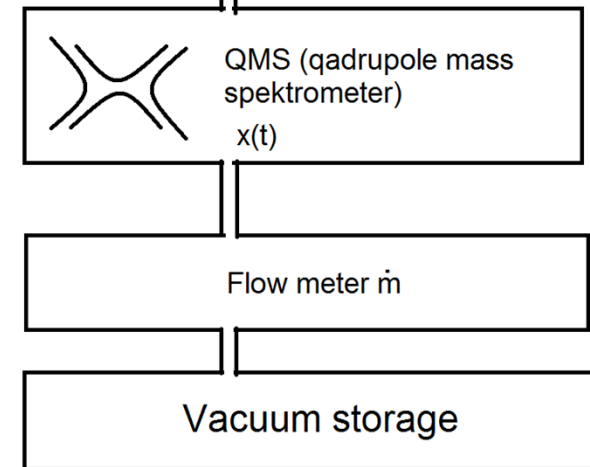
Permeate (secondary) chamber



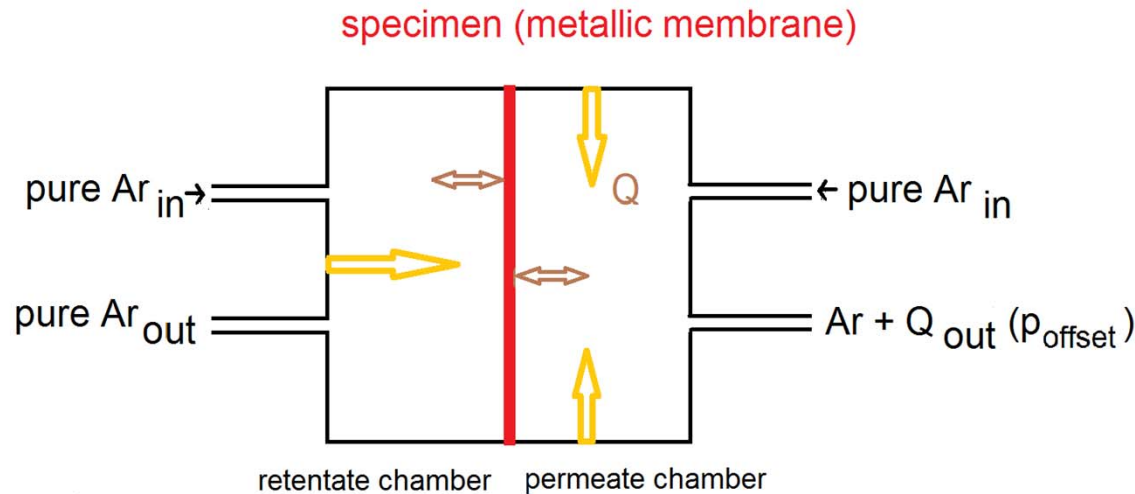
Simplified Q-PETE experiment



retentate chamber permeate chamber
 heated up to 600°C, in vacuum tank
 pressure gauges for both chambers



3.: Status Quo FDM-solver: Before beginning of experiment: Purging with pure Ar:



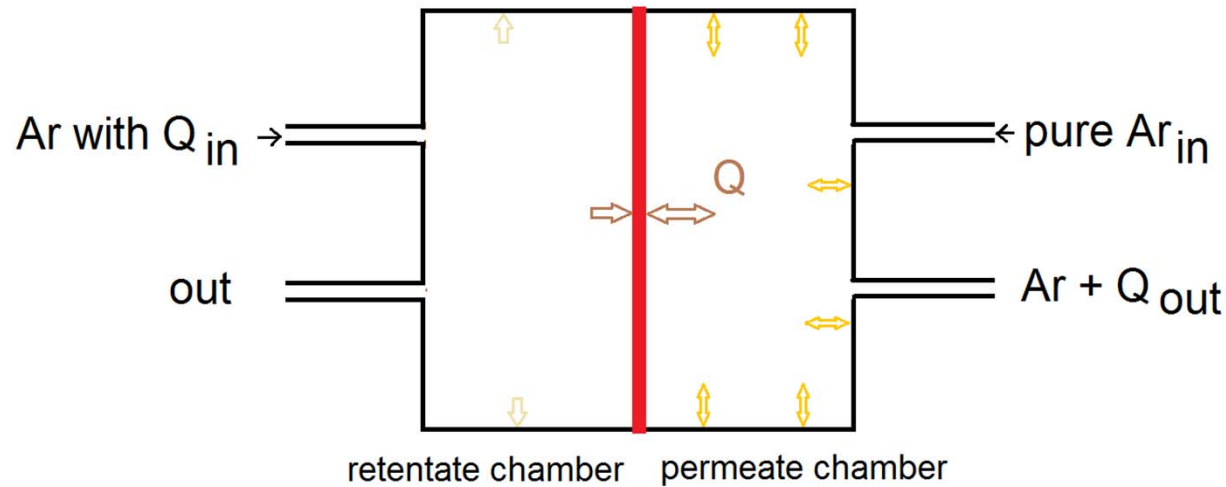
The stored Q is generating a constant assumed offset permeation. The membrane is in diffusive contact with the two volumes, equilibrium state given by k_S .

$$c(t, x) = k_S \sqrt{p_{offset}} = k_S \sqrt{\frac{j_{offset} p_{tot}}{\dot{m}_{Ar}}}, t < t_{start}, 0 \leq x \leq d_m$$

j_{offset} determined by QMS, specimen saturated with Q, no Q is additionally stored or emitted, p_{tot} absolute pressure in both volumes by pressure gauge, \dot{m}_{Ar} by mass flow controller @RT

FDM solver: Boundary conditions after start of experiment:

specimen (metallic membrane)



$$c(x = 0, t > t_{off}) \stackrel{FDM}{=} c(1, t > t_{off}) = c(o)$$

$$= k_s \sqrt{p_{load}}$$

$$p_{offset} < p_{load}$$

$$\dot{j}_{measure} = \dot{j}_{offset} + \underbrace{\dot{j}_{perm}}_{\text{from membrane}}$$

Partial Q-Pressure in retentate chamber, surface concentration linear increased in 1 s after t_{off}

\dot{j}_{offset} assumed constant, generated by thick structures thickness more than 20 mm (1.4404), emitting into vacuum also, membrane around 1.2 mm thickness.

FDM solver: Boundary condition of permeate (secondary)
 membrane side after start of experiment

$$j_{measure} = j_{offset} + -D_{eff} \frac{d_m^2 \pi}{4} \frac{\partial c(x = w_m, t > t_{off})}{\partial x} \stackrel{\text{FDM}}{\approx} j_{offset} + -D_{eff} \frac{d_m^2 \pi}{4} \frac{c(x = w_m - 4\Delta x, t > t_{off}) - c(x = w_m - 2\Delta x, t > t_{off})}{2 \Delta x}$$

$$(*) \quad c(n, t > t_{off}) = \underbrace{k_s \sqrt{j_{measure} \frac{p_{tot}}{\dot{m}_{Ar} \alpha}}}_{lit.=0, c(n,t>t_{off})=k_s \sqrt{\frac{RT_{abs}}{V_{gas}} \int_0^t j_{measure} dt}} \sim k_s \sqrt{\frac{D_{eff} (c(n-4,t) - c(n-2,t))}{2 \Delta x}}$$

$\Delta x = 12 \mu m (=d_m / n)$ from discretization of membrane in thickness direction normally $n=100$ elements, first element on retentate side, n^{th} element on permeate side. $\alpha=1$ for homogeneous purge gas inlet, $\alpha=2$ for point shaped inlet

Transient FDM-solver (any textbook):

$$\frac{\partial c}{\partial t} = D_{eff} \frac{\partial^2}{\partial x^2}$$

$$(**) \quad c(i, t + \Delta t) = c(i, t) + \frac{D_{eff} \Delta t}{\Delta x^2} (c(i + 1, t) - 2 c(i, t) + c(i - 1, t))$$

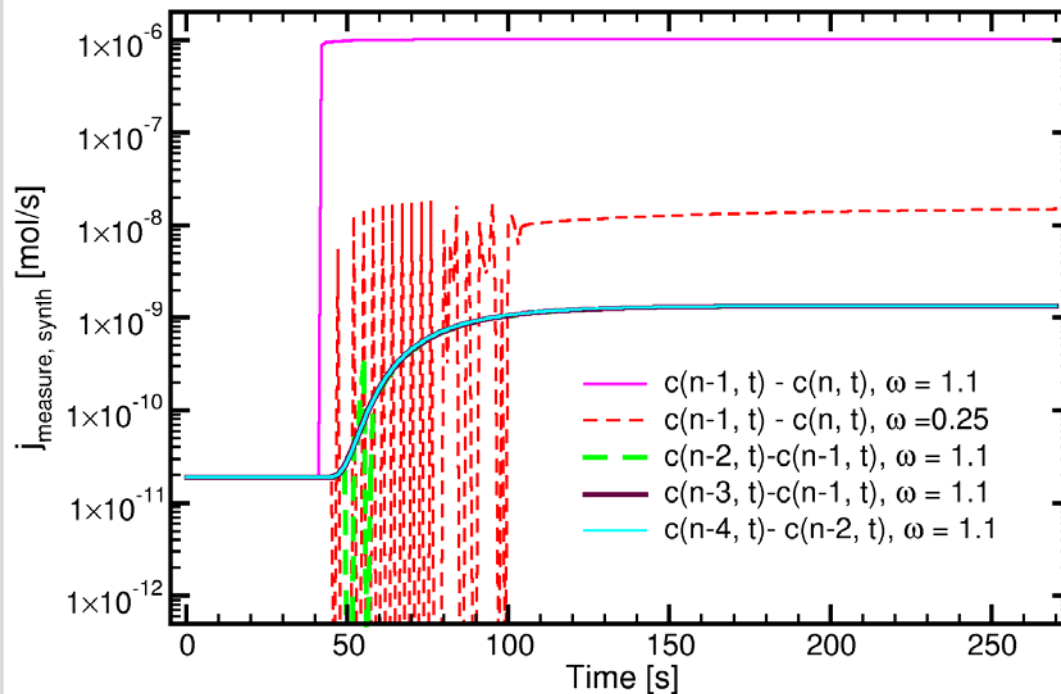
$$I = 2 \dots n - 1, t > t_{off}$$

Used FDM-SOR-step (successive over relaxation)

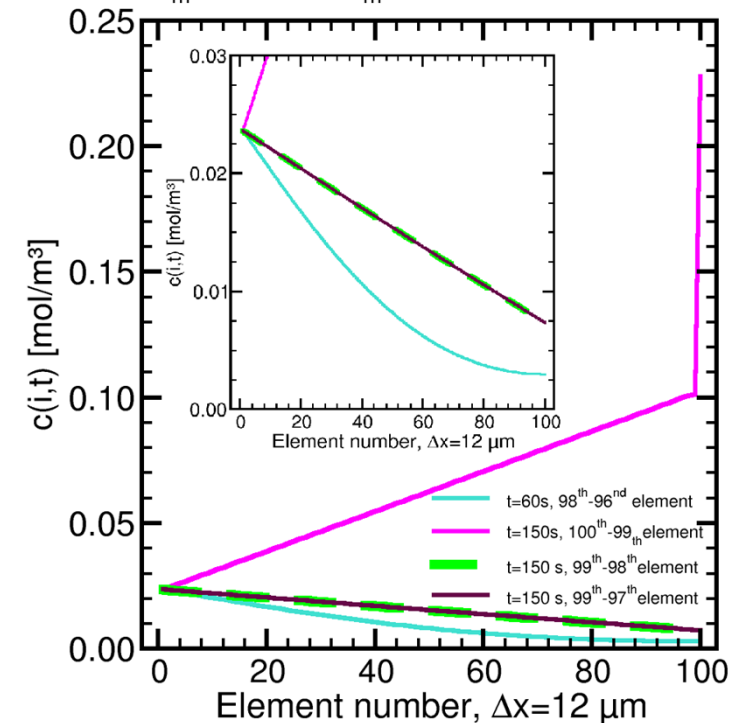
$$c(i, t + \Delta t)_{\omega} = \underbrace{\omega}_{\text{SOR}} (c(i, t + \Delta t) - c(i, t)) + c(i, t), \quad 0 < \omega < 2$$

Optimized ω with Eigenwertanalysis of (*, **), only proposal: Change of (*) to Henrys law, translation of (**) into matrix, calculation with QR method for λ_{\max} , now only $\omega=1.1$ carefully is used.

T=673 K, Optifer, 150 Pa, $w_m = 125$ mm, $d_m = 1.2$ mm, 30 ml/min
 ω SOR parameter, n=100

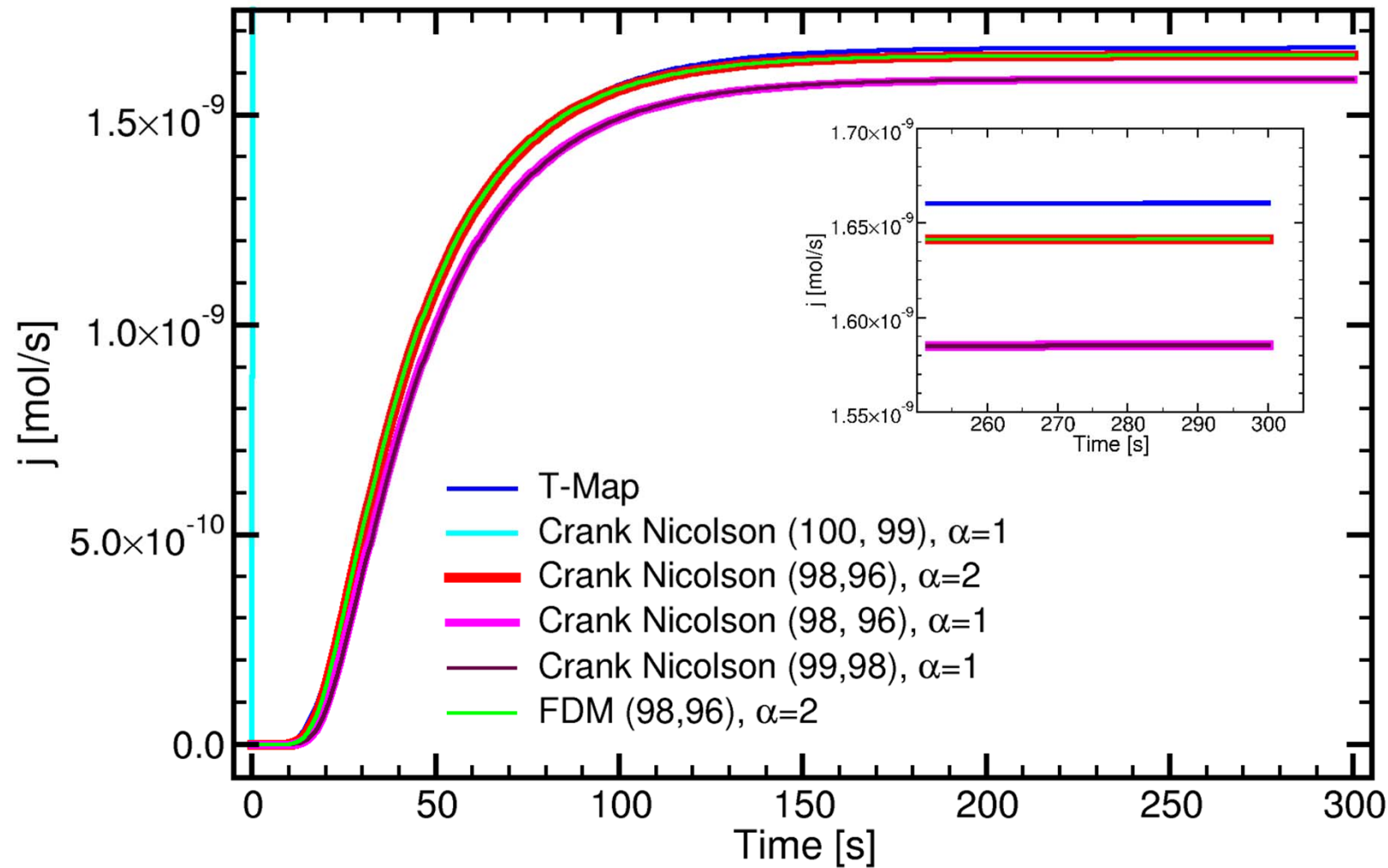


T=673 K, 150 Pa, 30 ml/min, $t_{\text{off}}=40$ s
 $w_m = 125$ mm, $d_m = 1.2$ mm, n=100, $\tau=19$ s



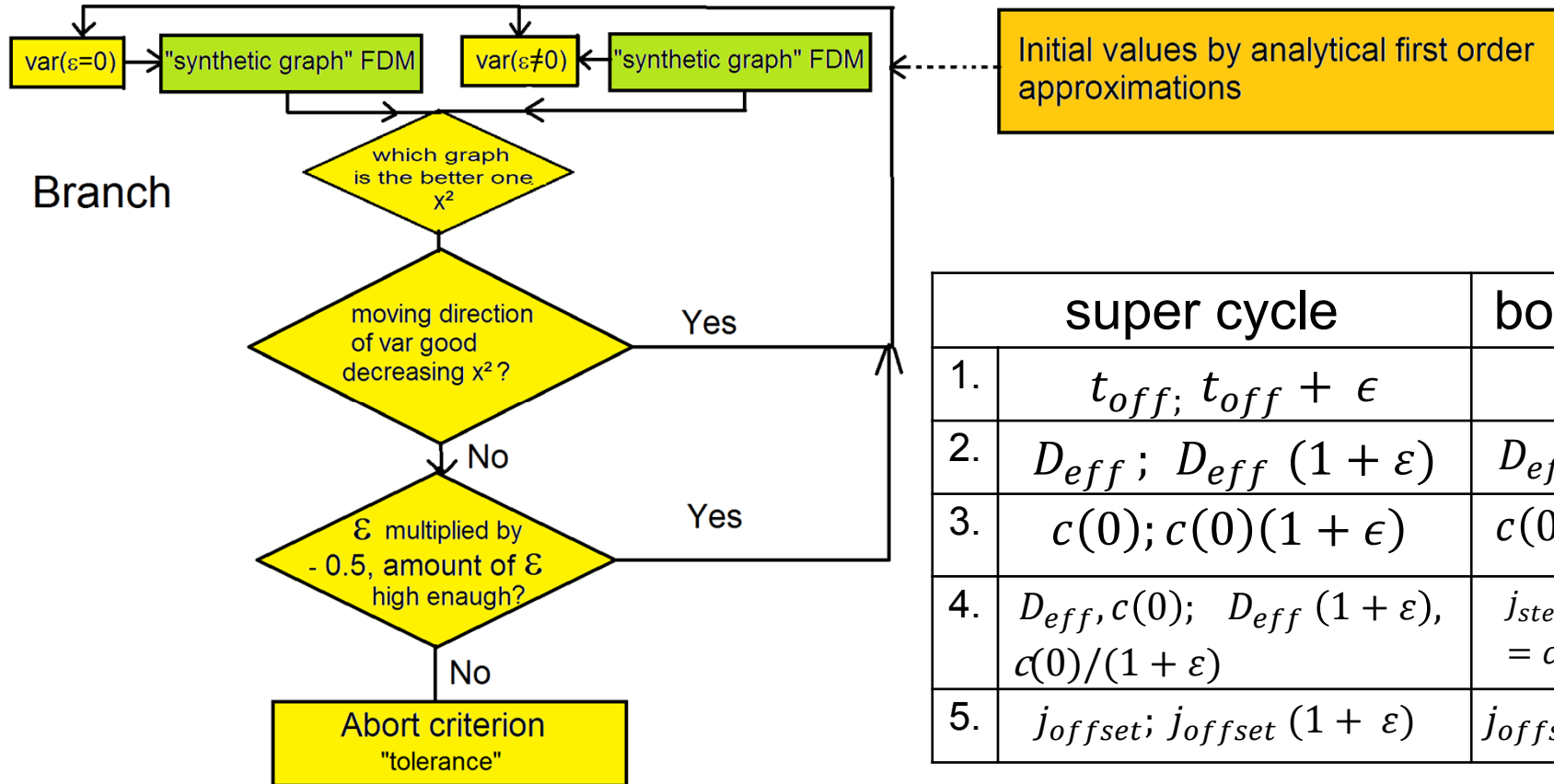
Comparison T-Map, Crank Nicolson and FDM

673 K, $2.5 \cdot 10^{-6} \text{ m}^3/\text{s}$, 280 Pa, $1.4 \cdot 10^5 \text{ Pa}$, D_2 , 100 elements



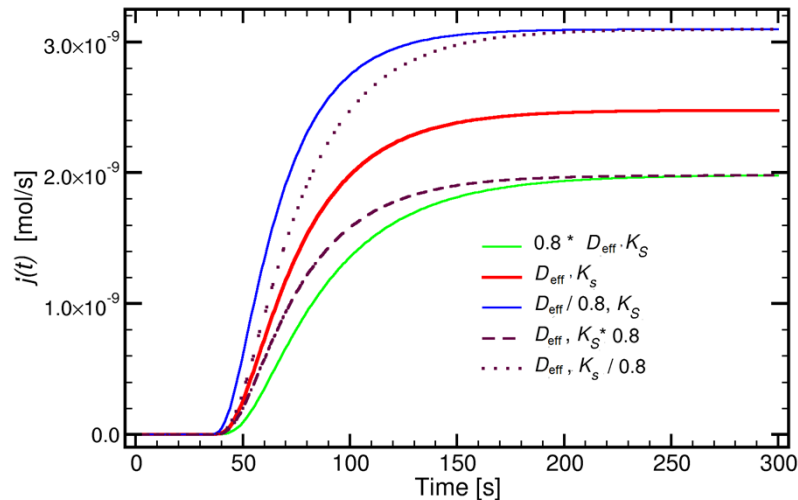
4.: Status quo: (Branch and Bound) B&B algorithm

Four desired variables: D_{eff} , $c(0)$ ($=c(1, t)$ res. $k_s \sqrt{p_{load}}$), t_{off} and j_{offset} serial treated within a super cycle. No explicit formulation possible, especially Daynes solution, comparison between measured permeation curve and "synthetic" graph from FDM module:

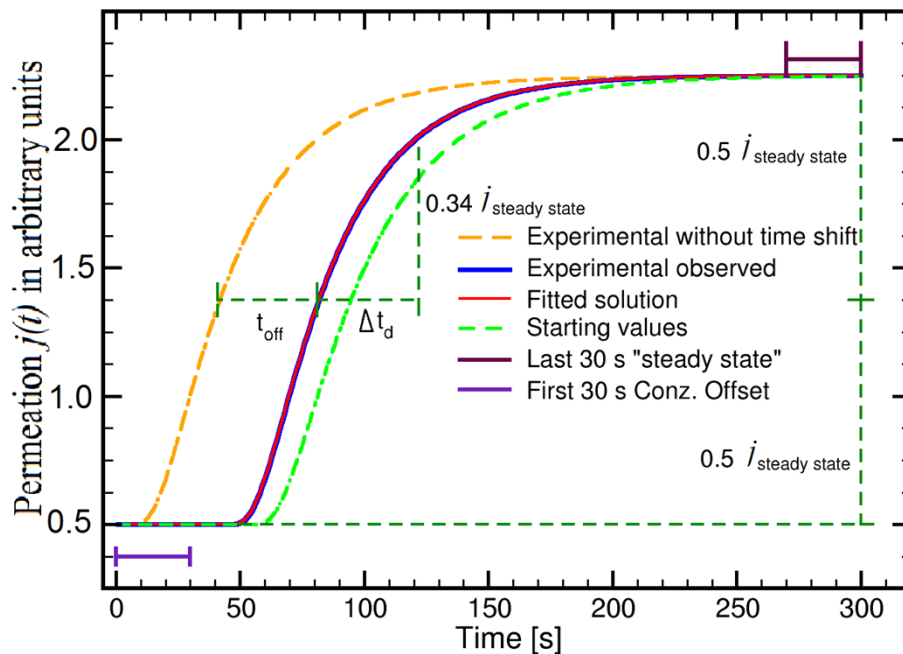


	super cycle	bounds
1.	$t_{off}; t_{off} + \epsilon$	
2.	$D_{eff}; D_{eff} (1 + \epsilon)$	$D_{eff} > 0$
3.	$c(0); c(0)(1 + \epsilon)$	$c(0) > 0$
4.	$D_{eff}, c(0); D_{eff} (1 + \epsilon), c(0)/(1 + \epsilon)$	$j_{steady state} = const.$
5.	$j_{offset}; j_{offset} (1 + \epsilon)$	$j_{offset} > 0$

$T = 573 \text{ K}$, $D_{st} = 4.78 \cdot 10^{-9} \text{ m}^2/\text{s}$, $K_{s,st} = 5.06 \cdot 10^{-2} \text{ mol/m}^3$
 $p_L = 3 \cdot 10^3 \text{ Pa}$, $\dot{m} = 180 \text{ ml/min}$, $d_m = 1.2 \text{ mm}$, $w_m = 125 \text{ mm}$



B&B: Determination of initial values:



$$j_{offset,initial} = \frac{1}{n_j} \sum_{i=1}^{n_j} j(i)_{measure}$$

$$D_{eff,initial} = \frac{d_m^2}{\pi^2 \Delta t_d}$$

$$t_{off,initial} = t_{1/2} - 1.25 \Delta t_d$$

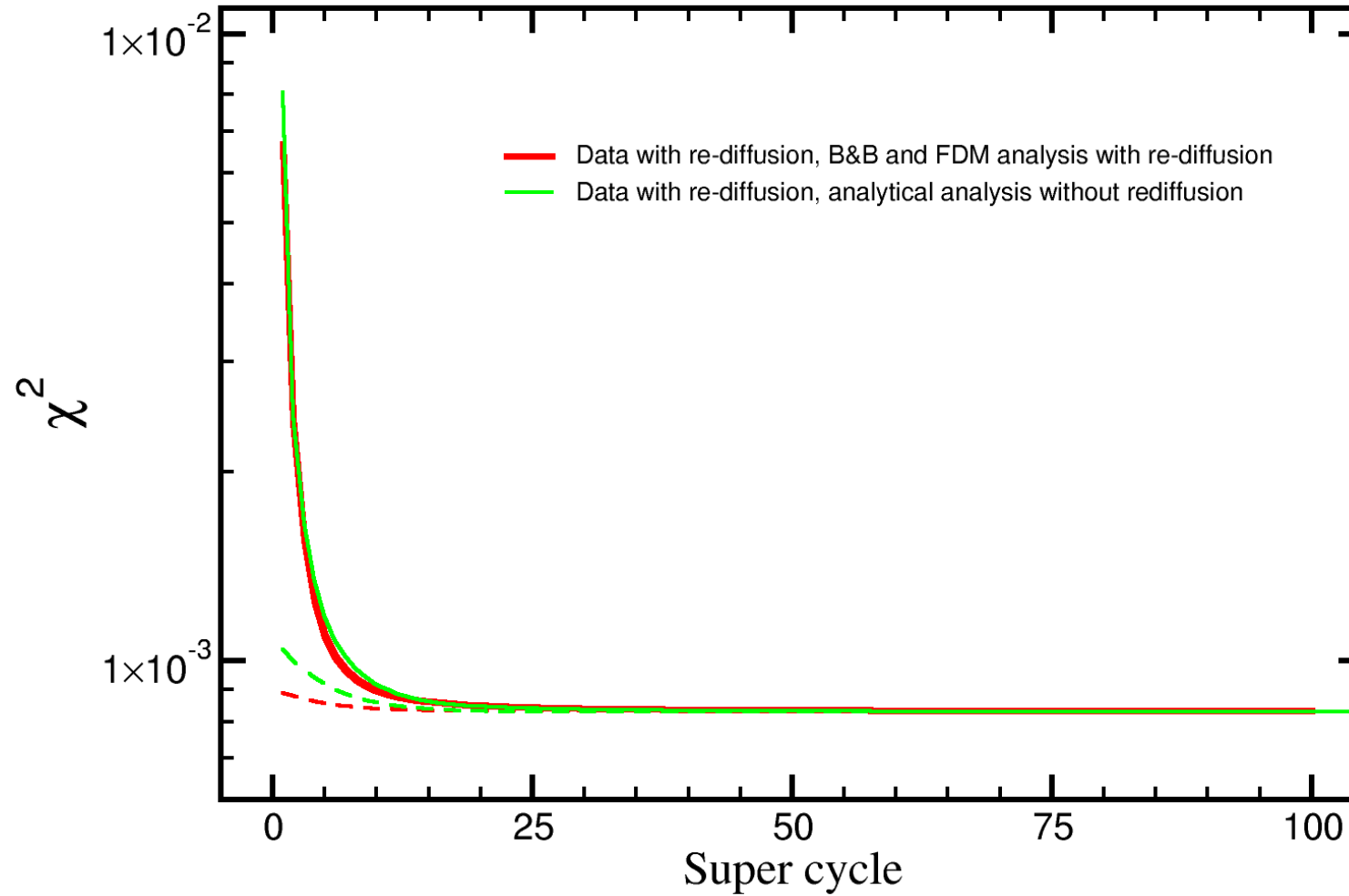
$$\Delta j = j_{steady\ state} - j_{offset,initial}$$

$$\begin{aligned}
 & c(1, t \geq t_{off})_{initial} \\
 &= \frac{\Delta j \cdot 4 d_m}{w_m^2 \pi D_{eff,initial} \sqrt{\dot{m}_{therm} p_{load}}} \sqrt{\frac{\Delta j p_{tot}}{\dot{m}_{therm} p_{load}}}
 \end{aligned}$$

Example for χ^2

Optifer 573 K, 0.15 kPa partial pressure, 11th run

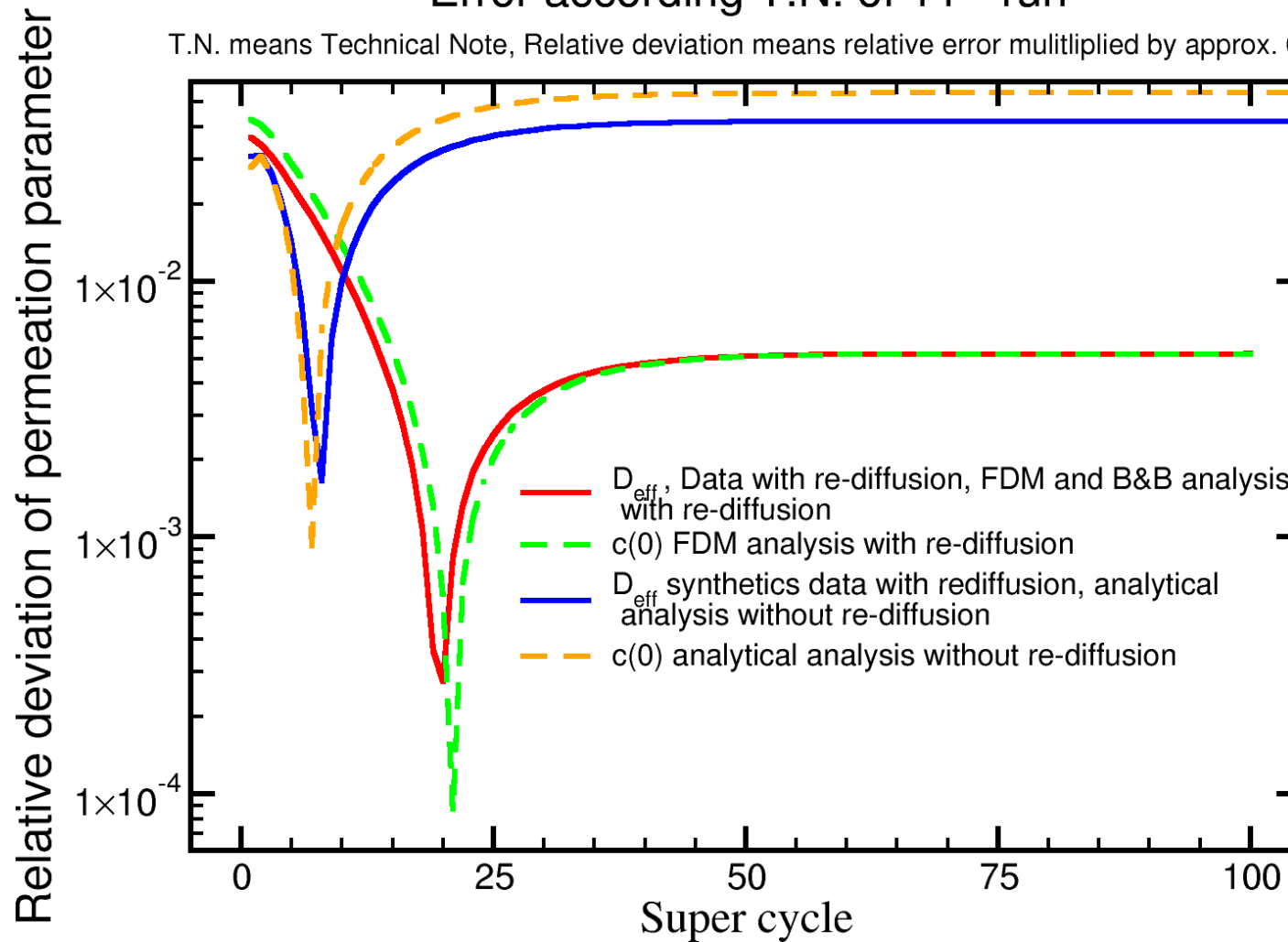
$\omega=1.1$, tolerance=0.001, $\Delta t=10^{-4}$ s, 100 FDMes, $w_m=0.125$ m, $d_m=1.2 \cdot 10^{-3}$ m



Example to former slide, comparison with analytical Daynes solution and FDM results

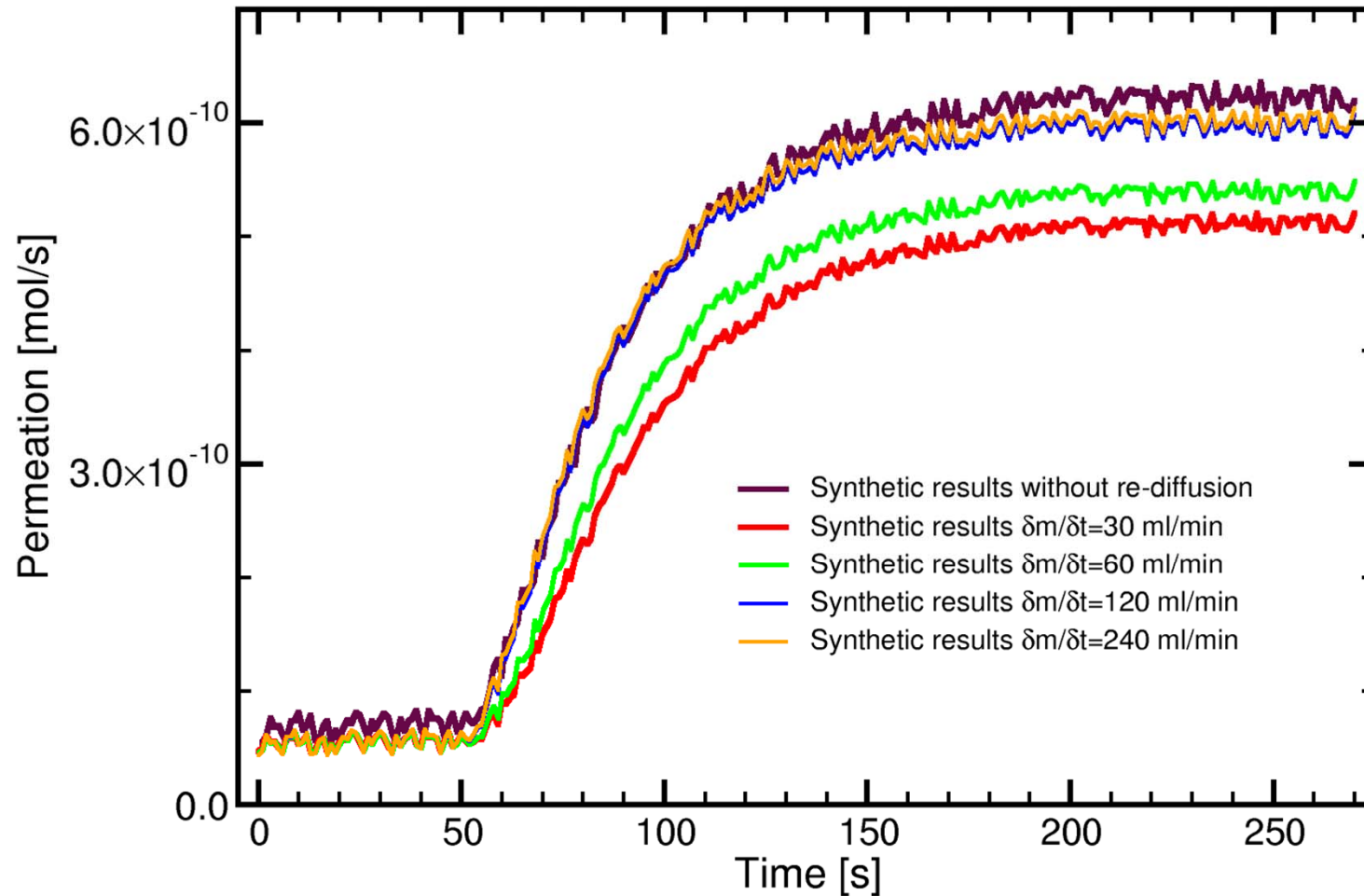
Error according T.N. of 11th run

T.N. means Technical Note, Relative deviation means relative error multiplied by approx. 0.5



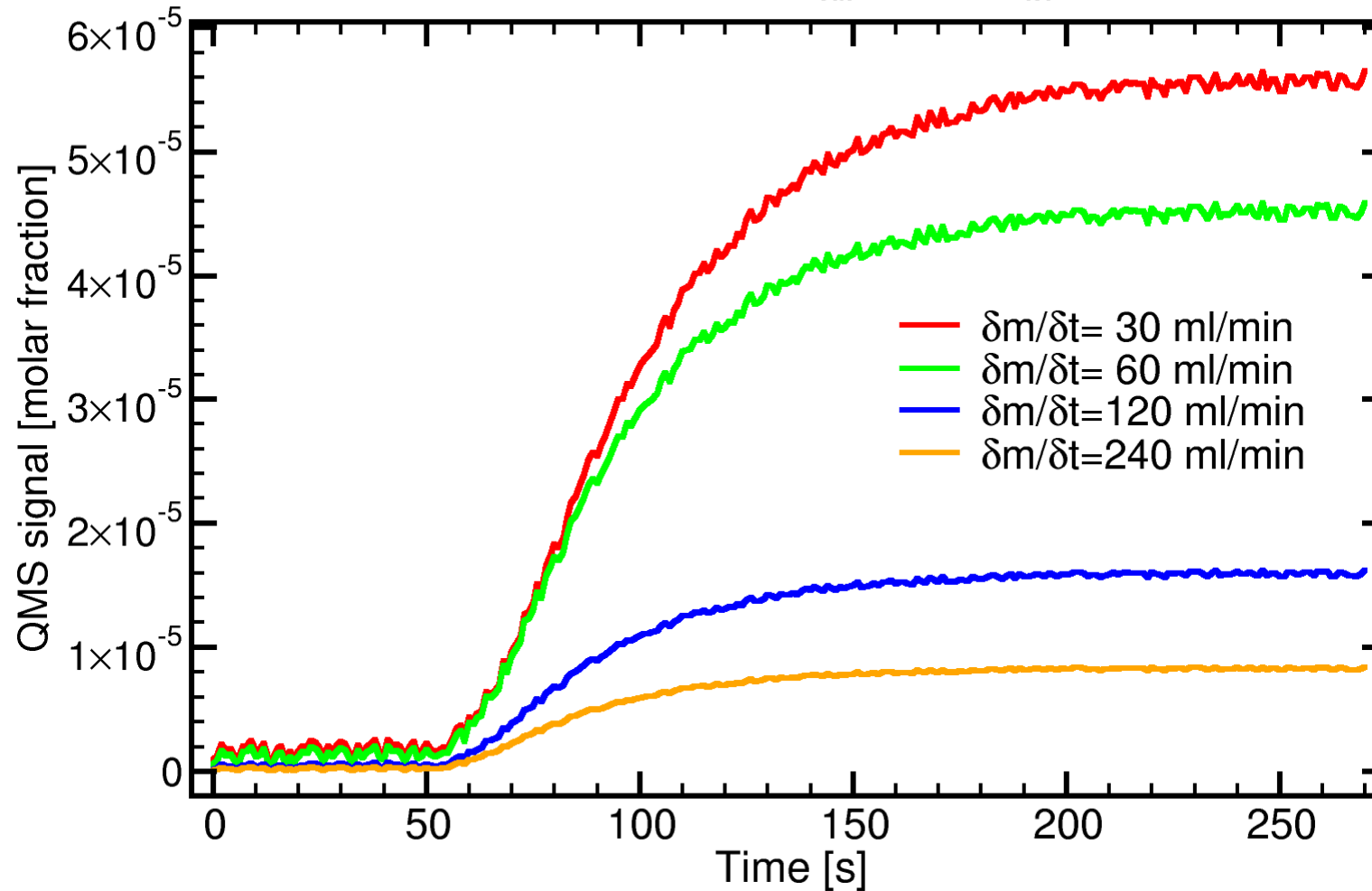
4. Expected advantage of Q-PETE setup: Influence on results by differing purge gas flux in permeate chamber:

573 K, 30-240 ml/min, $d_m=0.125$ m, $w_m = 1.2 \cdot 10^{-3}$ m
 $\Delta t=10^{-4}$ s, 100 FDMs, $\omega=1.1$, tolerance=0.001



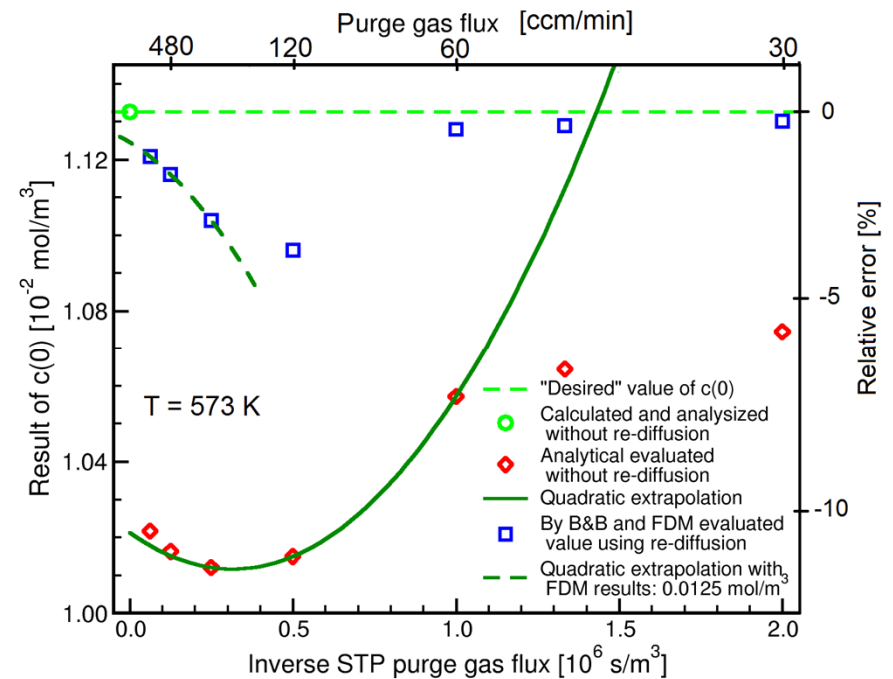
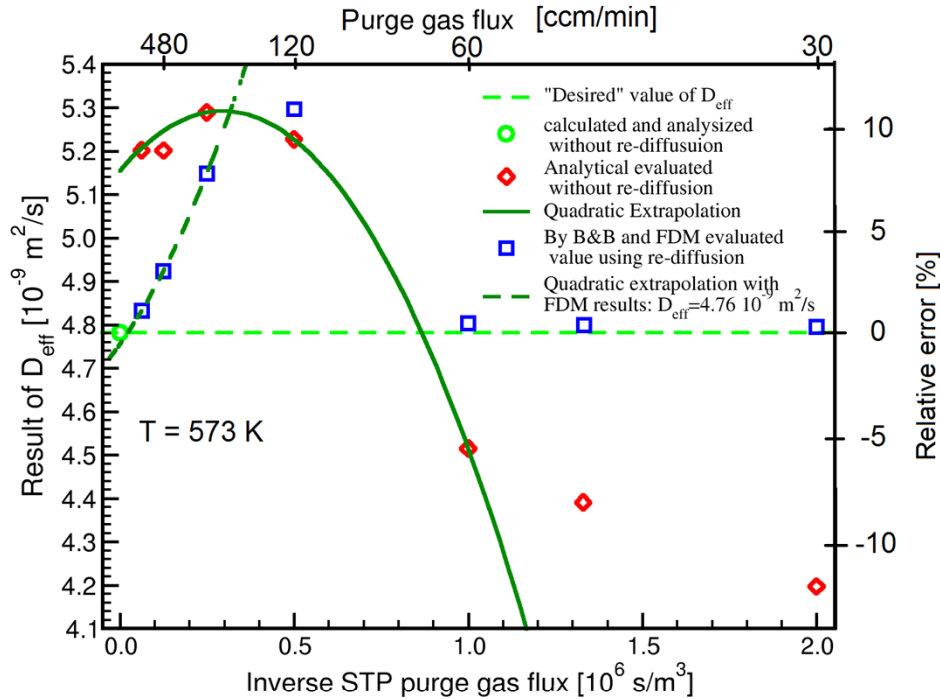
Expected advantage of Q-PETE setup: Influence on results by differing purge gas flux in permeate chamber aiming a differing Q-concentration there.

573 K, Optifer, $d_m = 0.125$ m, $w_m = 1.2 \cdot 10^{-3}$ m
 $\Delta t = 10^{-4}$ s, 100 FDMs, $\omega = 1.1$, $p_{load} = 150$ Pa, $p_{tot} = 1.5 \cdot 10^5$ Pa

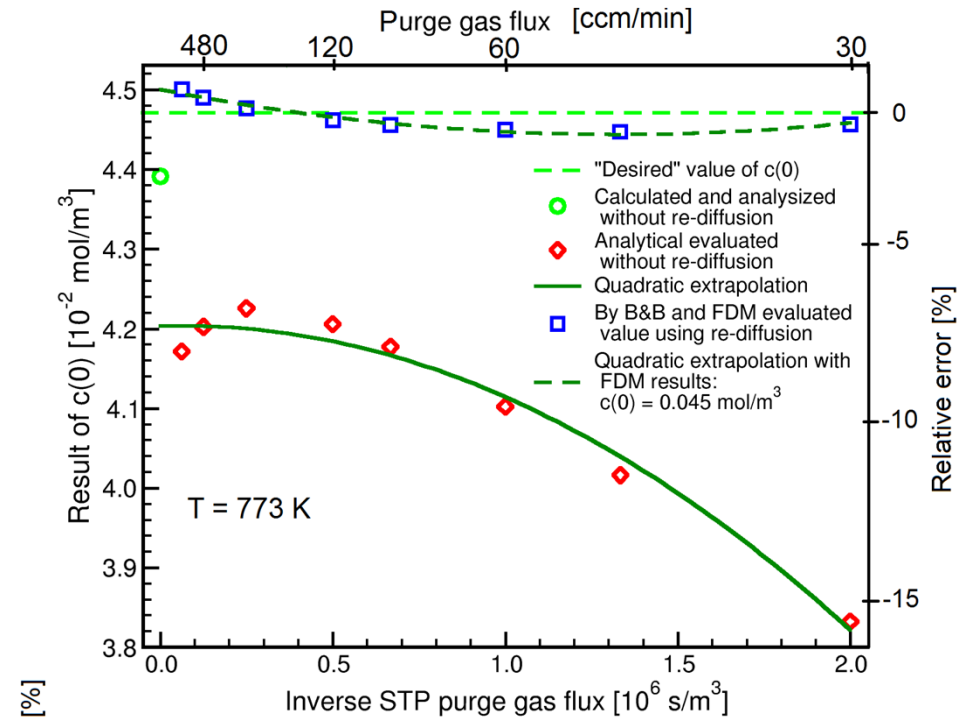
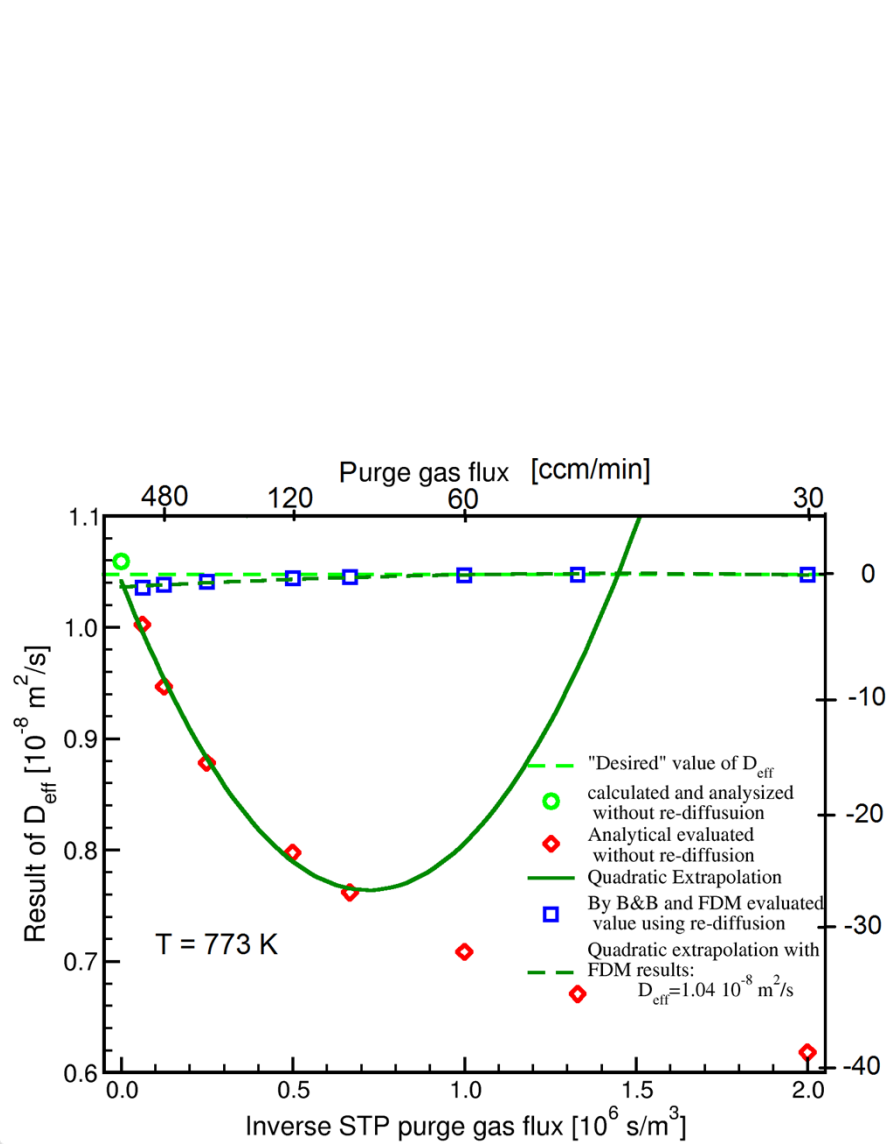


Re-recognition of diffusion constant and Sieverts' constant,

permeation constants of technical note as true regarded, time dependent synthetic permeation graph, analysis by FDM and B&B algorithm versus Daynes with B&B

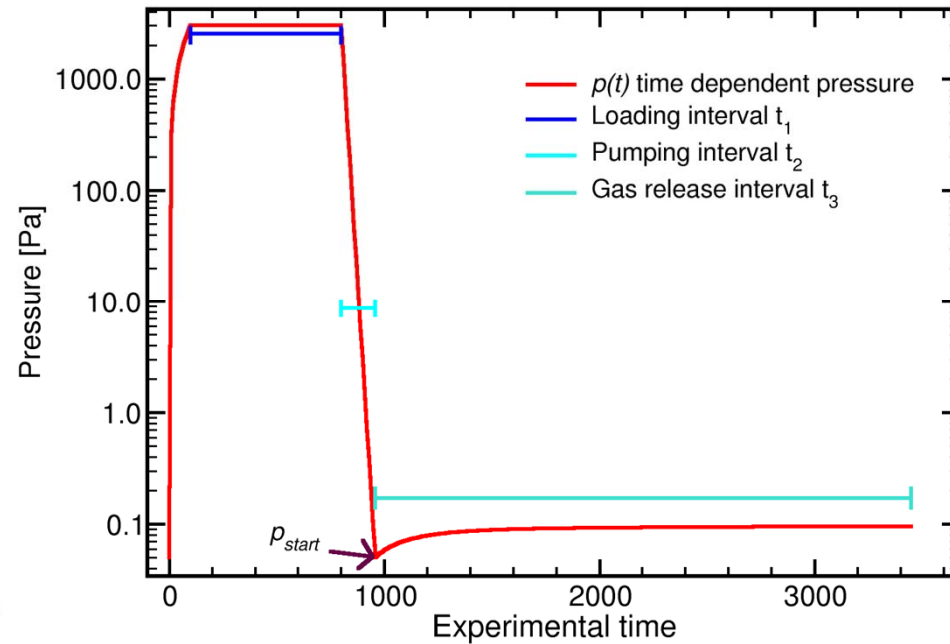
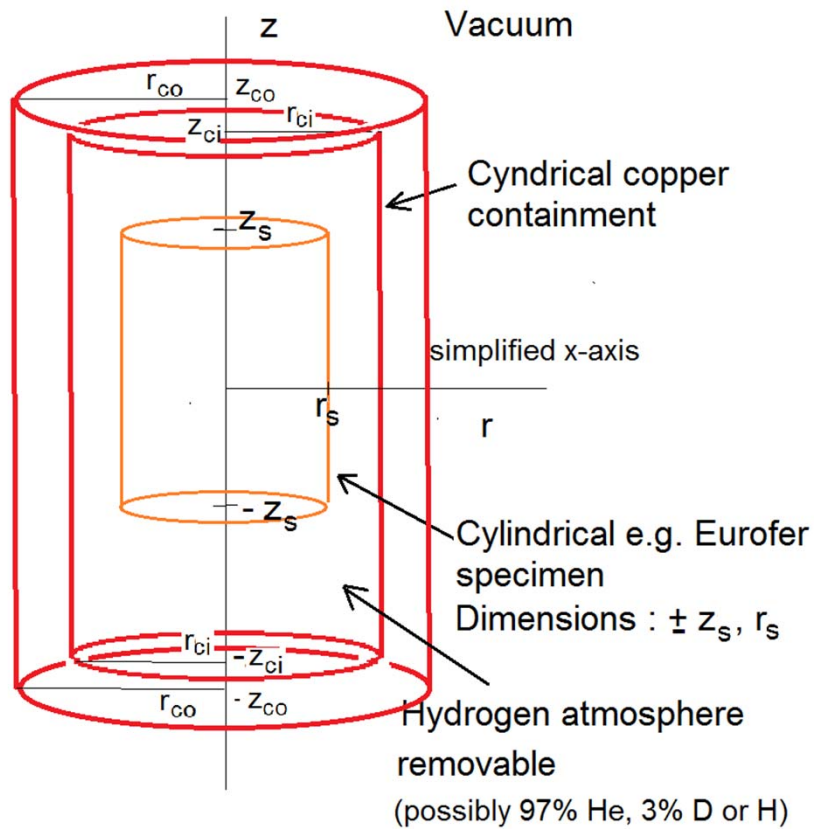


Re-recognition of diffusion and Sieverts' constant



5.: Desired object, gas release experiment

$T = 573 \text{ K}$



Equations simplified:

$$\frac{\partial c}{\partial t} = D_{sa} \Delta c \quad \frac{\partial d}{\partial t} = D_{cu} \Delta d \quad \Delta \underset{\text{only } r\text{-dependence}}{=} \frac{\partial^2}{\partial r^2} + \frac{1}{r} \frac{\partial}{\partial r}$$

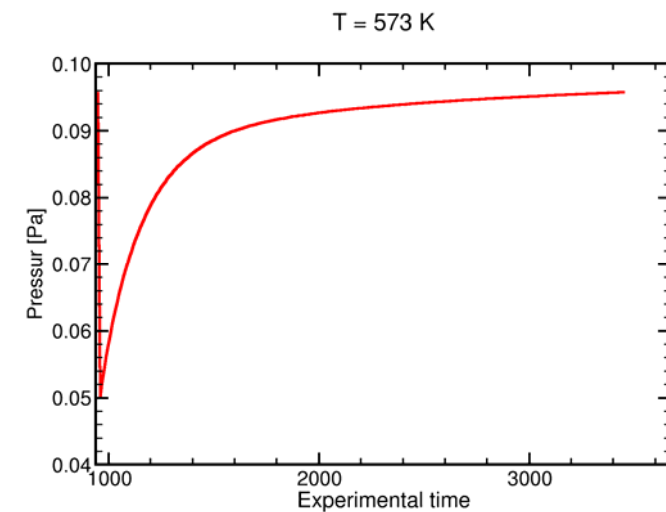
$$\frac{\partial m}{\partial t} = -A_{sa} D_{sa} \frac{\partial}{\partial r} c(r = r_{sa}, t) - A_{cu} D_{cu} \frac{\partial}{\partial r} d(r = r_{ci}, t)$$

$$p(t) = p_{start} + \underbrace{k_v}_{RT_{abs}/V_{gas}} \int_{t_1+t_2}^t \frac{\partial m}{\partial t} dt$$

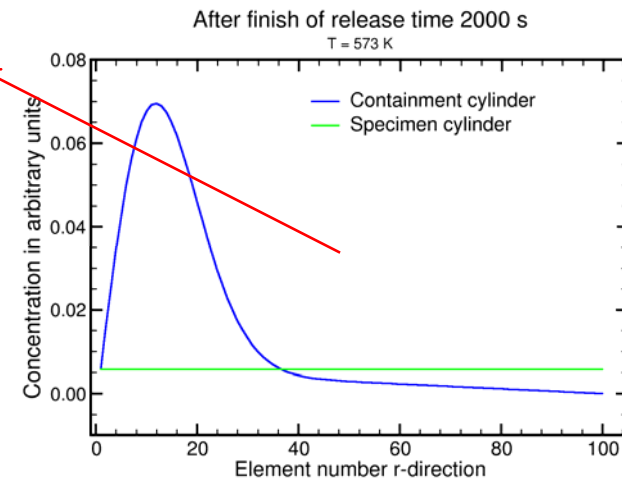
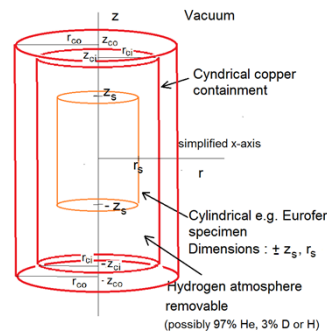
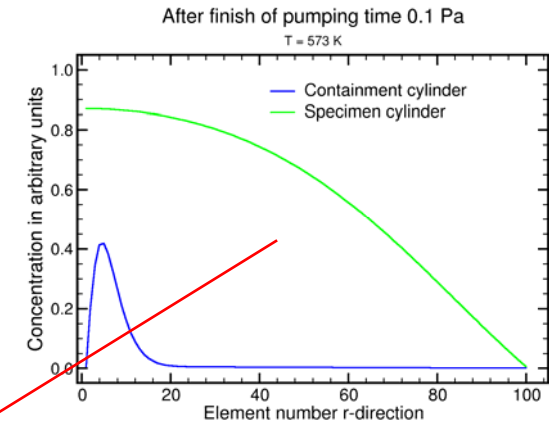
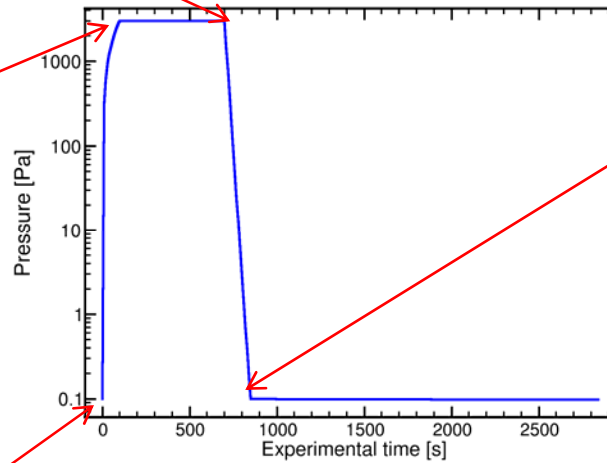
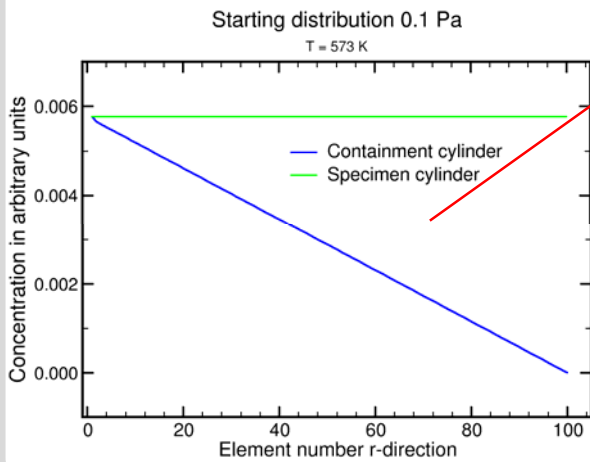
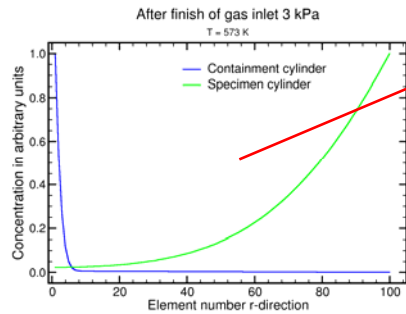
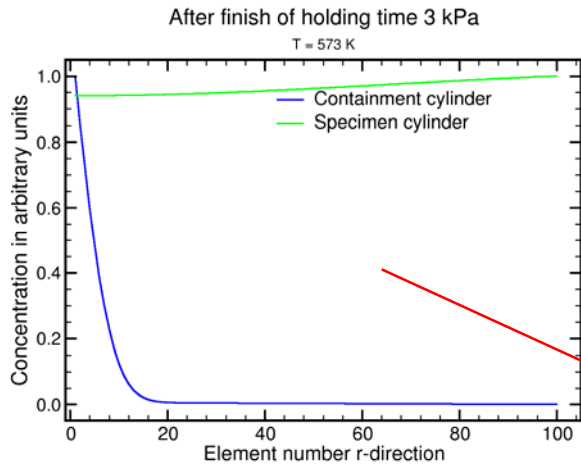
$$c(r = r_s, \forall t) = k_{s,sa} \sqrt{p(t)} \quad \frac{\partial}{\partial r} c(r = 0, \forall t) = 0$$

$$d(r = r_{ci}, \forall t) = k_{s,cu} \sqrt{p(t)}$$

$$d(r = r_{co}, \forall t) = 0$$

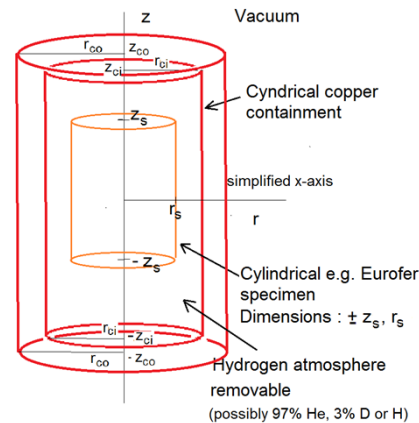
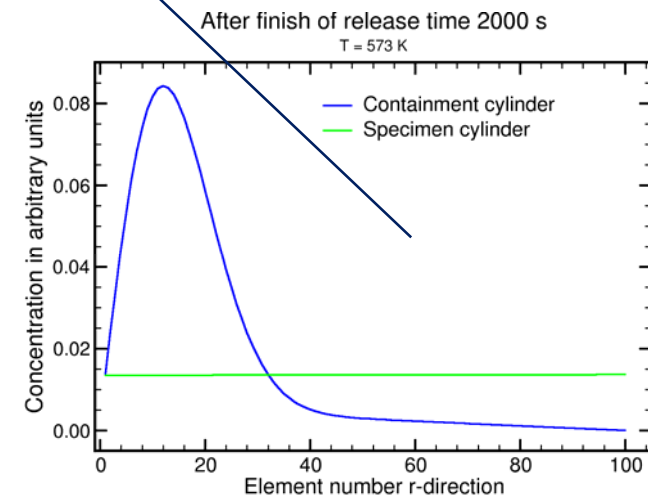
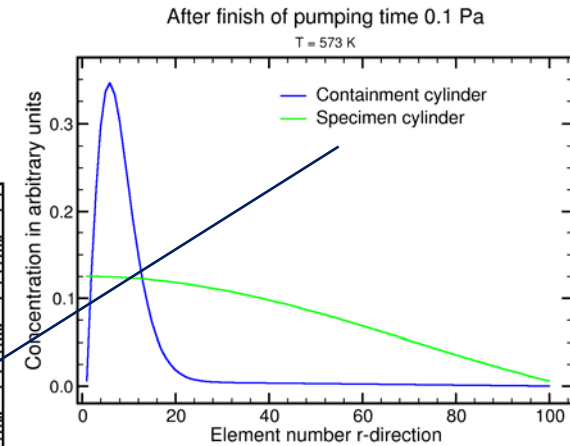
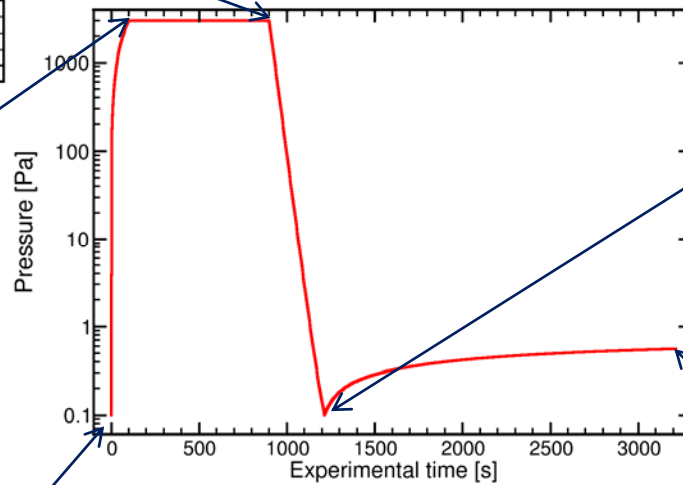
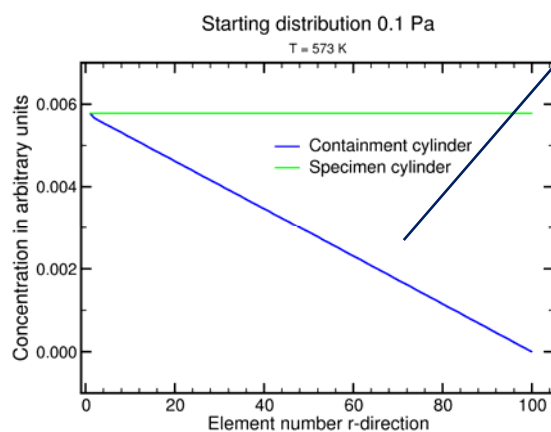
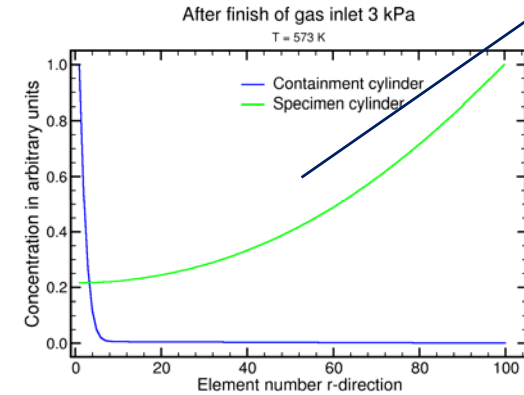
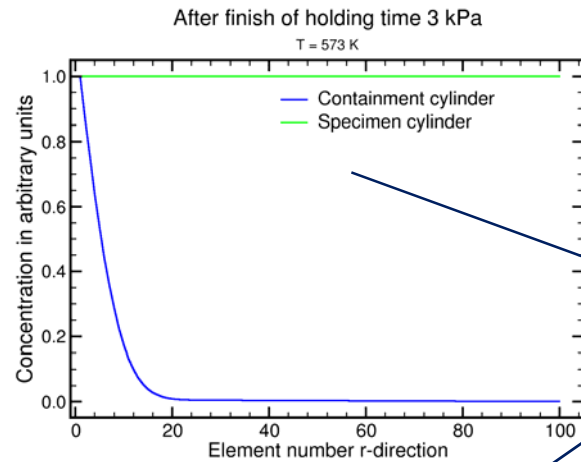


Gas release Experiment, current status



Gas release Experiment, current status

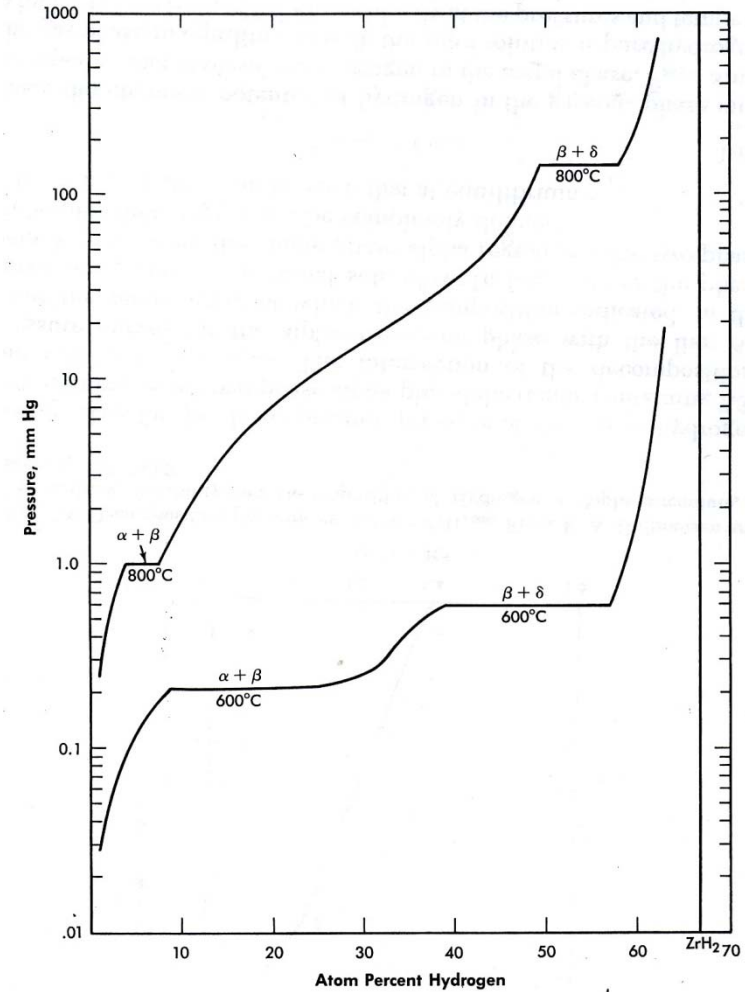
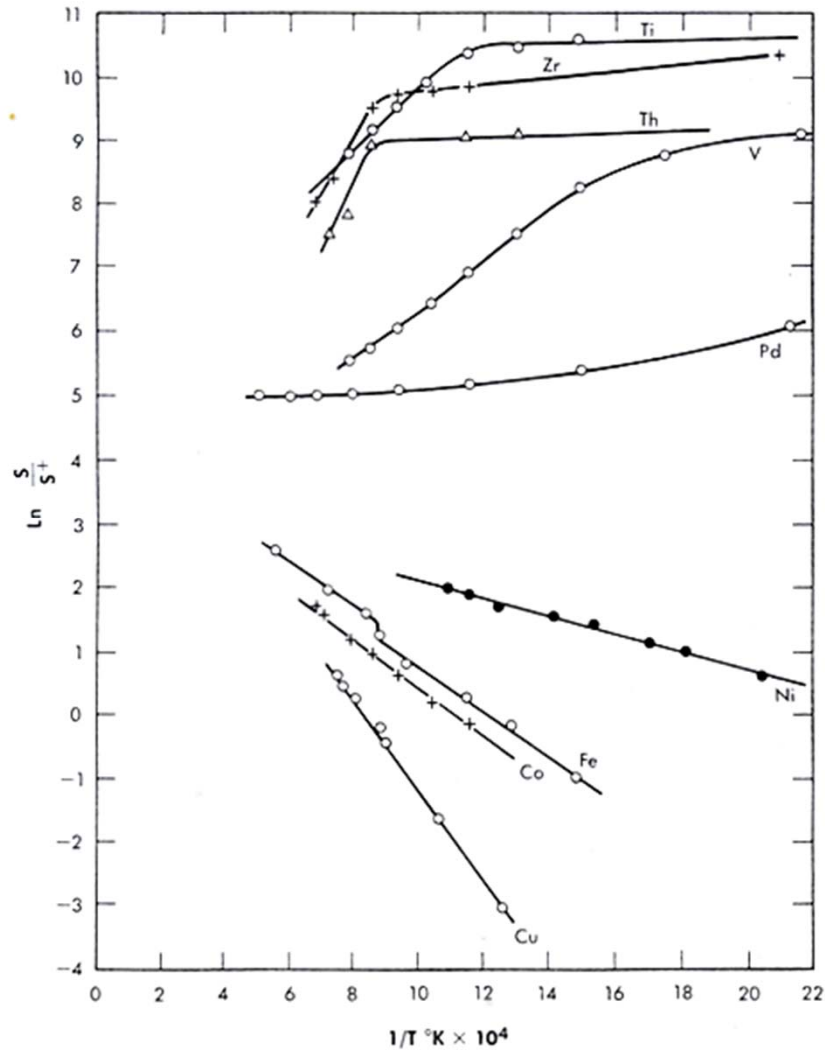
Extrapolation of inverse problem using 2D-Solver and additional cap surfaces 2400 h on bwUniCluster.



6.: Mögliche Anwendungsgebiete

Projekt	Problem-Typ	Analytische Lösung ohne Phasengleichgewicht	Numerische Lösung mit Phasengleichgewicht	Analytische Lösung mit Phasengleichgewicht
Q-PETE, (INR) Transportparameter	kartesisch	Vorhanden, Inverses Problem (B&B)	Vorhanden, Inverses Problem (B&B)	Fehlt
Gas release (INR) Transportparameter	zylinder	Teilweise vorhanden, Inverses Problem fehlt	In Arbeit, Inverses Problem fehlt (Open Foam)	In Vorbereitung
VSL (ITEP) Tritium Extraction	sphere	Vorhanden	Fehlt (Open Foam)	Fehlt
Tritium Extraction mit Superpermeation (ITEP) exothermes Material	kartesisch	Vorhanden für endothermes Material	Fehlt (Open Foam)	Fehlt
Wasserstoffspeicher exothermes Material	zylinder	Fehlt	Fehlt (Open Foam)	Fehlt

Exothermic/endothemic metallic hydrides



Aus Blackledge