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# Magneto-Optical Properties of the Magnetite-Graphene Oxide Composites in Organic Solvents

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## Abstract

Copyright © 2018 American Chemical Society. Graphene oxide (GO) aqueous solutions are known to form liquid crystals that can switch in electric fields. Magnetic fields as external stimuli are inefficient toward GO because of its diamagnetic properties, and GO is known to be insoluble in most of the organic solvents. In this study, composites of GO with oleate-protected magnetite nanoparticles were prepared as stable colloid solutions in the mixed isopropanol-chloroform solvents. The structure of the composite particles and the optical properties of their solutions can be controlled by the ratio of the mixing parent components. The as-prepared solutions are highly responsive to external magnetic field. As the consequence, the optical transmission and the direction of light scattering can be efficiently manipulated. These systems pave the way for fabricating functional materials, such as magneto-optical switches, density-gradient materials, and micromotors. Solubility in nonpolar organic solvents broadens the scope of their potential applications.

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## Keywords

graphene oxide, iron nanoparticles, light scattering, liquid crystals, magneto-optical properties

## References

- [1] Dreyer, D. R.; Todd, A. D.; Bielawski, C. W. Harnessing the Chemistry of Graphene Oxide. *Chem. Soc. Rev.* 2014, 43, 5288-5301, 10.1039/c4cs00060a
- [2] Eigler, S.; Hirsch, A. Chemistry with Graphene and Graphene Oxide-Challenges for Synthetic Chemists. *Angew. Chem., Int. Ed.* 2014, 53, 7720-7738, 10.1002/anie.201402780
- [3] Dimiev, A. M.; Eigler, S. *Graphene Oxide: Fundamentals and Applications*; Wiley: London, 2016.
- [4] Buelke, C.; Alshami, A.; Casler, J.; Lewis, J.; Al-Sayaghi, M.; Hickner, M. A. Graphene Oxide Membranes for Enhancing Water Purification in Terrestrial and Space-Born Applications: State of the Art. *Desalination* 2018, 448, 113-132, 10.1016/j.desal.2018.09.008
- [5] Parades, J. I.; Villar-Rodil, S.; Martínez-Alonso, A.; Tascón, J. M. D. Graphene Oxide Dispersions in Organic Solvents. *Langmuir* 2008, 24, 10560-10564, 10.1021/la801744a
- [6] Gudarzi, M. M. Colloidal Stability of Graphene Oxide: Aggregation in Two Dimensions. *Langmuir* 2016, 32, 5058-5068, 10.1021/acs.langmuir.6b01012
- [7] Vallés, C. Rheology of Graphene Oxide Dispersions. In *Graphene Oxide: Fundamentals and Applications*; Dimiev, A. M., Eigler, S., Eds.; Wiley: London, 2016; Chapter 4, pp 115-138.
- [8] Amirov, R. R.; Shayimova, J.; Nasirova, Z.; Dimiev, A. M. Chemistry of Graphene Oxide. Reactions with Transition Metal Cations. *Carbon* 2017, 116, 356-365, 10.1016/j.carbon.2017.01.095

- [9] Amirov, R. R.; Shayimova, J.; Nasirova, Z.; Solodov, A.; Dimiev, A. M. Analysis of Competitive Binding of Several Metal Cations by Graphene Oxide Reveals the Quantity and Spatial Distribution of Carboxyl Groups on its Surface. *Phys. Chem. Chem. Phys.* 2018, 20, 2320-2329, 10.1039/c7cp07055a
- [10] Singh, V. K.; Patra, M. K.; Manoth, M.; Gowd, G. S.; Vadera, S. R.; Kumar, N. In situ Synthesis of Graphene Oxide and its Composites with Iron Oxide. *N. Carbon Mater.* 2009, 24, 147-152, 10.1016/s1872-5805(08)60044-x
- [11] Shen, J.; Hu, Y.; Shi, M.; Li, N.; Ma, H.; Ye, M. One Step Synthesis of Graphene Oxide- Magnetic Nanoparticle Composite. *J. Phys. Chem. C* 2010, 114, 1498-1503, 10.1021/jp909756r
- [12] Yan, L.; Chang, Y.-N.; Zhao, L.; Gu, Z.; Liu, X.; Tian, G.; Zhou, L.; Ren, W.; Jin, S.; Yin, W.; Chang, H.; Xing, G.; Gao, X.; Zhao, Y. The Use of Polyethylenimine-Modified Graphene Oxide as a Nanocarrier for Transferring Hydrophobic Nanocrystals into Water to Produce Water-Dispersible Hybrids for Use in Drug Delivery. *Carbon* 2013, 57, 120-129, 10.1016/j.carbon.2013.01.042
- [13] Santhosh, C.; Daneshvar, E.; Kollu, P.; Peräniemi, S.; Grace, A. N.; Bhatnagar, A. Magnetic SiO<sub>2</sub>@ CoFe<sub>2</sub>O<sub>4</sub> Nanoparticles Decorated on Graphene Oxide as Efficient Adsorbents for the Removal of Anionic Pollutants from Water. *Chem. Eng. J.* 2017, 322, 472-487, 10.1016/j.cej.2017.03.144
- [14] Cao, X.; Jiang, Z.; Wang, S.; Hong, S.; Li, H.; Shao, Y.; She, Y.; Wang, J.; Jin, F.; Jin, M. One-Pot Synthesis of Magnetic Zeolitic Imidazolate Framework/Graphene Oxide Composites for the Extraction of Neonicotinoid Insecticides from Environmental Water Samples. *J. Sep. Sci.* 2017, 40, 4747-4756, 10.1002/jssc.201700674
- [15] Cen, Y.; Xiao, A.; Chen, X.; Liu, L. Screening and Separation of  $\alpha$ -amylase Inhibitors from *Solanum Nigrum* with Amylase-Functionalized Magnetic Graphene Oxide Combined with High-Speed Counter-Current Chromatography. *J. Sep. Sci.* 2017, 40, 4780-4787, 10.1002/jssc.201700333
- [16] Cong, H.-P.; He, J.-J.; Lu, Y.; Yu, S.-H. Water-Soluble Magnetic-Functionalized Reduced Graphene Oxide Sheets: In situ Synthesis and Magnetic Resonance Imaging Applications. *Small* 2010, 6, 169-173, 10.1002/sml.200901360
- [17] Zhao, H.; Zhu, Q.; Gao, Y.; Zhai, P.; Ma, D. Iron Oxide Nanoparticles Supported on Pyrolytic Graphene Oxide as Model Catalysts for Fischer Tropsch Synthesis. *Appl. Catal., A* 2013, 456, 233-239, 10.1016/j.apcata.2013.03.006
- [18] He, G.; Liu, W.; Sun, X.; Chen, Q.; Wang, X.; Chen, H. Fe<sub>3</sub>O<sub>4</sub>@ Graphene Oxide Composite: A Magnetically Separable and Efficient Catalyst for the Reduction of Nitroarenes. *Mater. Res. Bull.* 2013, 48, 1885-1890, 10.1016/j.materresbull.2013.01.038
- [19] Song, Y.; He, Z.; Hou, H.; Wang, X.; Wang, L. Architecture of Fe<sub>3</sub>O<sub>4</sub>-Graphene Oxide Nanocomposite and its Application as a Platform for Amino Acid Biosensing. *Electrochim. Acta* 2012, 71, 58-65, 10.1016/j.electacta.2012.03.077
- [20] Zubir, N. A.; Yacou, C.; Motuzas, J.; Zhang, X.; Da Costa, J. C. D. Structural and functional investigation of graphene oxide-Fe<sub>3</sub>O<sub>4</sub> nanocomposites for the heterogeneous Fenton-like reaction. *Sci. Rep.* 2014, 4, 4594, 10.1038/srep04594
- [21] Al-Zangana, S.; Iliut, M.; Turner, M.; Vijayaraghavan, A.; Dierking, I. Confinement Effects on Lyotropic Nematic Liquid Crystal Phases of Graphene Oxide Dispersions. *2D Mater.* 2017, 4, 041004, 10.1088/2053-1583/aa843a
- [22] Guo, F.; Kim, F.; Han, T. H.; Shenoy, V. B.; Huang, J.; Hurt, R. H. Hydration-Responsive Folding and Unfolding in Graphene Oxide Liquid Crystal Phases. *ACS Nano* 2011, 5, 8019-8025, 10.1021/nn2025644
- [23] Kim, J. E.; Han, T. H.; Lee, S. H.; Kim, J. Y.; Ahn, C. W.; Yun, J. M.; Kim, S. O. Graphene Oxide Liquid Crystals. *Angew. Chem., Int. Ed.* 2011, 50, 3043-3047, 10.1002/anie.201004692
- [24] Xu, Z.; Gao, C. Aqueous Liquid Crystals of Graphene Oxide. *ACS Nano* 2011, 5, 2908-2915, 10.1021/nn200069w
- [25] Sasikala, S. P.; Lim, J.; Kim, I. H.; Jung, H. J.; Yun, T.; Han, T. H.; Kim, S. O. Graphene Oxide Liquid Crystals: A Frontier 2D Soft Material for Graphene-Based Functional Materials. *Chem. Soc. Rev.* 2018, 47, 6013-6045, 10.1039/c8cs00299a
- [26] Shen, T.-Z.; Hong, S.-H.; Song, J.-K. Electro-Optical Switching of Graphene Oxide Liquid Crystals with an Extremely Large Kerr Coefficient. *Nat. Mater.* 2014, 13, 394-399, 10.1038/nmat3888
- [27] Laurent, S.; Forge, D.; Port, M.; Roch, A.; Robic, C.; Vander Elst, L.; Muller, R. N. Magnetic Iron Oxide Nanoparticles: Synthesis, Stabilization, Vectorization, Physicochemical Characterizations, and Biological Applications. *Chem. Rev.* 2008, 108, 2064-2110, 10.1021/cr068445e
- [28] Hu, D. L.; Goreau, T. J.; Bush, J. W. M. Flow Visualization Using Tobacco Mosaic Virus. *Exp. Fluid* 2009, 46, 477-484, 10.1007/s00348-008-0573-6
- [29] Jasper, C.; Sage, I. Liquid Crystals for Display Application. *Ullmann's Encyclopedia of Industrial Chemistry*; Wiley Online Library, 2013.
- [30] Freundlich, H.; Hatfield, H. S. *Colloid and Capillary Chemistry*; Methuen And Co. Ltd; London, 1926.
- [31] Shao, L.; Yang, Z.-J.; Andr n, D.; Johansson, P.; K ll, M. Gold Nanorod Rotary Motors Driven by Resonant Light Scattering. *ACS Nano* 2015, 9, 12542-12551, 10.1021/acsnano.5b06311
- [32] Fennimore, A. M.; Yuzvinsky, T. D.; Han, W.-Q.; Fuhrer, M. S.; Cumings, J.; Zettl, A. Rotational Actuators Based on Carbon Nanotubes. *Nature* 2003, 424, 408-410, 10.1038/nature01823

- [33] Dimiev, A. M.; Polson, T. A. Contesting the Two-Component Structural Model of Graphene Oxide and Reexamining the Chemistry of Graphene Oxide in Basic Media. *Carbon* 2015, 93, 544-554, 10.1016/j.carbon.2015.05.058
- [34] Jalili, R.; Aboutalebi, S. H.; Esrafilzadeh, D.; Konstantinov, K.; Moulton, S. E.; Razal, J. M.; Wallace, G. G. Organic Solvent-Based Graphene Oxide Liquid Crystals: A Facile Route Toward the Next Generation of Self-Assembled Layer-by-Layer Multifunctional 3D Architectures. *ACS Nano* 2013, 7, 3981-3990, 10.1021/nn305906z
- [35] Amirova, L.; Surnova, A.; Balkaev, D.; Musin, D.; Amirov, R.; Dimiev, A. M. Homogeneous Liquid Phase Transfer of Graphene Oxide into Epoxy Resins. *ACS Appl. Mater. Interfaces* 2017, 9, 11909-11917, 10.1021/acsami.7b02243
- [36] Wen, X.; Yang, J.; He, B.; Gu, Z. Preparation of Monodisperse Magnetite Nanoparticles Under Mild Conditions. *Curr. Appl. Phys.* 2008, 8, 535-541, 10.1016/j.cap.2007.09.003
- [37] Jiang, W.; Wu, Y.; He, B.; Zeng, X.; Lai, K.; Gu, Z. Effect of Sodium Oleate as a Buffer on the Synthesis of Superparamagnetic Magnetite Colloids. *J. Colloid Interface Sci.* 2010, 347, 1-7, 10.1016/j.jcis.2010.02.055