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Abstract: Diazoacetates 1a,b undergo BF<sub>3</sub> OEt<sub>2</sub> catalyzed carbenoid attack on the oxygen of the phosphoryl double bond of phosphate triesters 2a-c or on the sulfur of thiophosphoryl double bond of thiophosphate 9 to form corresponding O-alkoxycarbonylmethylphosphates 3a-c or S-alkoxycarbonylmethylphosphate 13.

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The reactions of carbenes or carbenoids with the carbon-oxygen double bond give synthetically valuable intermediate carbonyl ylides. <sup>1,2</sup> Surprisingly, that little is known about analogous reaction of diazo precursors with the phosphoryl-oxygen or phosphoryl-sulfur double bond. Recently, the anomaly low H/D isotope effect was found in the reaction of thermally generated fluorenyl carbene with dimethyl hydrogen phosphite indicating the carbene attack on the oxygen of P=O bond (Scheme 1).<sup>3</sup>

$$C: + O = P \xrightarrow{H(D)} - \overline{C} - O - P \xrightarrow{H(D)} H(D)$$

Scheme 1

It is interesting that fragmentation of the similar zwitterionic intermediate to carbene is often involved into a possible mechanism of the reactions of ketones or thioketones with tervalent phosphorus compounds (Scheme 2).<sup>4-6</sup>

$$C=X + :P \xrightarrow{\overline{C}} -X - P \xrightarrow{\overline{C}} C: + X=P \xrightarrow{\overline{C}}$$

Diazo esters 1a,b and trialkylphosphates 2a-c were first to be chosen for intermolecular generation of zwitterionic intermediate. On slow (2 h) dropwise addition of a solution of diazo ester 1a,b (5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) to a stirred boiling solution of catalyst (0.5 mmol) and trialkylphosphate 2a-c (12.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) (Scheme 3) followed by water-Na<sub>2</sub>CO<sub>3</sub> work-up and MgSO<sub>4</sub> drying O-alkoxycarbonylphosphates 3a-c<sup>7</sup> have been isolated by distillation.

BF<sub>3</sub> OEt<sub>2</sub> is the catalyst of choice what indicates a Lewis acid promoted process. It is often that BF<sub>3</sub> OEt<sub>2</sub> is a superior catalyst in the typical carbenoid transformations.<sup>8</sup>