# Enhancement in Boiling Heat Transfer for Water Using a Polished Plate Surface

Masahiro KATOH<sup>1</sup>, Norimasa YOSHIDA<sup>2</sup>, Naoto YAMADA<sup>2</sup>, Daiki MINE<sup>2</sup>, Kenji OHNISHI<sup>3</sup>, Daisuke YONEKURA<sup>4</sup>, and Shigeru SUGIYAMA<sup>1</sup>

<sup>1</sup>Department of Applied Chemistry, Graduate School of Technology, Industrial and Social Sciences, Tokushima University, 2-1 Minamijosanjima-cho, Tokushima-shi, Tokushima 770-8506, Japan <sup>2</sup>Department of Chemical Science and Technology, Tokushima University, 2-1 Minamijosanjima-cho, Tokushima-shi, Tokushima 770-8506, Japan

<sup>3</sup>Daikatech Co. LTD., 9-1 Yonezu, Kawauchi-cho, Tokushima-shi, Tokushima 771-0139, Japan
<sup>4</sup>Department of Mechanical Engineering, Graduate School of Technology, Industrial and Social Sciences, Tokushima University, 2-1 Minamijosanjima-cho, Tokushima-shi, Tokushima 770-8506, Japan

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The thermal characteristic of a plate evaporator for boiling water have been experimentally investigated. Stainless-steel plates with five kinds of surface roughness (a mirror-polished surface and four kinds of polished plate surface, F2, F0, F-1, and F-2 polishing) were used for the investigation. The roughness order is F-2 > F-1 > F0 > F2 > mirror. The local boiling heat transfers were measured on the evaporator surface using ten thermocouples inside the evaporator. For a comparison of the heat transfer characteristic of the evaporator, the local boiling heat transfer coefficient was measured when the polishing direction was perpendicular or parallel to the water flow at a range of pressures (0.3–0.4 MPa) and flow rates (20–60 mL/min). The results indicate that the local heat transfer coefficient increased with increasing vapor quality at all surface conditions. For F0 polishing, the heat transfer coefficient on the surface with perpendicular polishing was much larger than that on the surfaces with parallel and mirror polishing. In particular, the heat transfer coefficient of the perpendicularly polished surface (F2 and F0 polishing) was increased six-fold in comparison with that of the mirror surface at a vapor quality of 0.35. However, for the F-1 and F-2 perpendicularly polishing plates, the heat transfer coefficients were smaller than those of the F0 and F2 perpendicularly polishing plates. For the F2 perpendicularly polished plate, heat transfer enhancement was confirmed, along with an increase in the number of bubbling points, on increasing the system pressure. For the polished surface, the heat transfer coefficient increased with increasing flow rate. This indicates that the bubble formation cycle was promoted by an increased flow rate.

## Introduction

In many industries, plate heat exchangers are used because of their high efficiency, easy maintenance, and small size (Li *et al.*, 2010). However, further reduction in the energy requirement is necessary. Use of an uneven plate surface is an easy way to enhance heat transfer by increasing the heat transfer area and generating a swirling current (Wang *et al.*, 2003; Turk *et al.*, 2016). Recently, many studies on millimeter-order uneven heat plates have been reported. Piper *et al.* (2016) reported the numerical investigation of turbulent forced convection heat transfer in pillow plates.

However, studies of micro- or nanoscale fine fabricated heat plates are scarce. Okamoto *et al.* (2010) reported the local heat transfer coefficient of a micro-grooved surface was remarkably larger (10–40%) than that of a flat surface. In their study, the depth of flutes for plates was 30  $\mu$ m, and the width of the protrusions was 100  $\mu$ m.

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Corresponding concerning this article should be addressed to M. Katoh (E-mail address: katoh@tokushima-u.ac.jp).

We have focused on our unique fabricated technique, socalled "F polishing" (Katoh *et al.*, 2008; Yonekura *et al.*, 2010; Yonekura *et al.*, 2011). As we have previously reported, an antiadhesion effect of particles is observed on these uniquely fabricated steel plates. In this study, we have aimed to enhance heat transfer by applying this technology to the heat transfer surface of a plate heat exchanger.

#### 1. Experimental

In this study, SUS 304 stainless steel was used. The surface of the plate was mirror polished using waterproof abrasive papers (#180–#2000) and alumina powders (particle diameter:  $0.3 \mu$ m). The F-polished surfaces were unidirectionally polished using four kinds of polishing material on the mirror surface (Katoh *et al.*, 2008). Hereafter, the four polishing conditions are denoted F2, F0, F-1, and F-2. The difference is only the grain size of the polishing materials used. The roughness was measured using a Surface Profiler (Dektak, Veeco Instruments Inc.). The root mean square roughness of the five types of polished SUS surfaces are shown in **Table 1**. Scanning electron microscopy (SEM) and atomic force microscopy (AFM) images of the polished SUS surface were obtained using a SEM (TM- 1000, Hitachi High-Technologies Corp.) and an AFM (Nano Wizard II, JPK Instruments AG), respectively. **Figures 1** and **2** show SEM and AFM images of the five kinds of polishing SUS surface. Heat transfer plates with four kinds of F polishing surfaces were used, in order of surface roughness, these are F-2, F-1, F0, and F2. A mirror-polished plate was also prepared for comparison.

Table 1The root mean square roughness (Rq) of the fivekinds of polished SUS surfaces.

Polishing	mirror	F2	F0	F-1	F-2
Surface roughness	6	71	206	552	655
$R_q$ [nm]					



Fig. 1 SEM images of the five kinds of polished SUS surfaces: (a) mirror, (b) F2, (c) F0, (d) F-1, and (e) F-2.



**Fig. 2** AFM images of the five kinds of polished SUS surfaces: (a) mirror, (b) F2, (c) F0, (d) F-1, and (e) F-2.

**Figure 3** shows the experimental setup, which consists of a heat transfer test plate, a pre-heater, an oil bath, a feed pump, and coolers (Koyama *et al.*, 2014). **Figure 4** shows the heat transfer test plate. The plate consists of microfabricated SUS304 heat transfer plate and aluminum heater block. Figure 4 also shows the location of thermocouples in the plate. Ten thermocouples were embedded in the plate, and five of them are located close to the passage. Pairs of thermocouples were set every 59 mm along the working fluid flow direction. **Figure 5** shows the detailed configuration of a pair of thermocouples; each pair is separated by 40-mm intervals to allow the measurement of the temperature distribution in the test plate. The experimental procedure was as follows. First, distilled water (the working

fluid) was circulated through the setup at an arbitrary flow rate. Then, the fluid was heated by the pre-heater and the heater attached to the heat transfer test plate. After the desired experimental conditions had been obtained and a steady state had been reached, the temperatures of the test plates were measured. In this study, the effects on the heat transfer performance of the test plates were investigated by changing the flow rate of the working fluid, the system pressure, and the heater output. The experimental conditions in this study are shown in **Table 2**.



Fig. 3 Schmatic diagram of the experimental setup.



Fig. 4 Configration of the heat transfer plate in the experimental setup.



Fig. 5 Configration of a pair of thermocouples.

 Table 2 Experimental conditions.

Parameter	Range	
Flow rate of the working	20, 40, 60	
fluid [mL/min]		
System pressure [MPa]	0.3, 0.4, 0.45, 0.5	
Heater output [W]	360, 640, 1000	
Outlet temperature of	60–138.7	
pre-heater [°C]		

## 2. Data Reduction

The local heat transfer coefficient and vapor quality were

calculated in this study (Koyama *et al.*, 2014). The local heat flux, q, was obtained using Eq. (1) and the data from the five pairs of thermocouples, as shown in Figure 4. Eq. (1) is derived from the one-dimensional heat conduction equation

$$q = k_{\rm sus} \frac{T_1 - T_2}{l_1} \tag{1}$$

The Plate surface temperature  $T_{\text{wall}}$ , is calculated by extrapolation:

$$T_{wall} = T_2 - \frac{ql_2}{k_{sus}} \tag{2}$$

Then, the local heat transfer coefficient h, is obtained:

$$h = \frac{q}{T_{wall} - T_{sat}} \tag{3}$$

The local vapor quality is obtained using Eq. 4:

$$\mathbf{x}_n = \frac{i_{\text{test,n}} - i_{sat,liq}}{i_{fg}} \quad (n = 1 - 5) \tag{4}$$

First, the specific enthalpy at the heat test plate entrance  $(i_{\text{test,in}})$  was calculated using Eq. 5.

$$i_{test,in} = i_{sat.liq} - c(T_{sat} - T_{in})$$
<sup>(5)</sup>

Next, the local specific enthalpy ( $i_{test,n}$ ) at temperature measuring point *n*, as shown in **Figure 6**, was calculated by adding the increase in the specific enthalpy at the plate entrance (Arima *et al.*, 2010).

Further,  $i_{\text{test,n}}$  is expressed as follows.

(a) In the case that n=1,

$$i_{test,1} = i_{test,in} + \frac{Q_1}{m}, \ Q_1 = q_1 A_1$$

(b) In the case that n=2-5

 $i_{test,n} = i_{test,n-1} + \frac{Q_n}{m}, Q_n = \frac{Q'_n + Q'_{n-1}}{2}, Q'_n = q_n A_n$  (6)

The local Nusselt numbers are defined by Eq. 7



Fig. 6 Schematic diagram of the effective heat transfer surface for calculating local entalpies.

# 3. Results and Discussion

The thermal characteristics of boiling water flow in a plate evaporator were experimentally investigated. The effects of the type of F-polishing, pressure, and flow rate on the local heat transfer coefficient are discussed.

#### 3.1 Effect of polishing direction

Figure 7 shows the local heat transfer coefficient (h) for

F0 and F-2-polished surfaces polished in two different directions (perpendicular or parallel) with respect to the direction of water flow, as well as that of a mirror surface. The figure shows the changes in the local heat transfer coefficient with vapor quality at the set operating conditions (flow rate: 40 mL/min, heater output: 1000 W, system pressure: 0.3 MPa). According to Figure 7, for a parallel polished plate, the local heat transfer coefficient increased slightly with increasing vapor quality. The coefficients for F0 and F-2 polished plates were almost the same as that of the mirror-polished plate, and the coefficient was independent of the surface roughness. On the other hand, for the perpendicularly polished plates, the coefficient increased sharply with increasing vapor quality. The coefficient of the F0-polished plate was higher than that of F-2polished plate. Furthermore, the perpendicularly polished plates showed better heat transfer characteristics than the parallel and mirror-polished plates. These results indicate that perpendicular F polishing was effective under boiling conditions with nuclear boiling. An enhancement in the heat transfer was obtained when using perpendicular F polishing. In the case of perpendicular polishing, the flow is continuously subjected to a rough surface, resulting in vortex flow. The vortex flow separates the superheated liquid layer from the heat transfer surface and accelerates the bubble formation cycle. Thus, the vortex flow contributes to the enhancement of heat transfer performance. On the other hand, in the case of parallel polishing, the roughness that causes vortex flow is not in the flow direction, and the vortex flow does not occur. The valued of h for F0 and F-2 parallelpolished surfaces are very similar that of mirror surface. Thus, all polishing was carried out perpendicular to the water flow direction.



Fig. 7 Comparison of local heat transfer coefficient versus vapor quality for F0 and F-2-polished plates and perpendicular or parallel polishing directions, as well as that for a mirror-polished plate

## 3.2 Effect of pressure drop

**Figure 8** shows the pressure drops for three kinds of plate surfaces (F0, F-2, and mirror polished). The results showed pressure drop increased with increasing heater output (a) and flow rate (b). Concerning heater output (a), the heat quantity provided to a heat transfer plate from the heater increased with increasing heater output. As a result, the temperature difference between the temperature of the working fluid and the saturated temperature increased with increasing plate surface temperature. By increasing the temperature difference, nucleate boiling was promoted, the vapor quality in the flow channel increased and

the pressure drop increased. On the other hand, concerning the flow rate (b), it is thought that the influence of forced convection was dominant. Because the area of flow channel exit was small, pressure drop increased with increasing flow rate. However, these tendencies were not dependent on the polishing surface.



**Fig. 8** Comparison of the pressure drop at F0, F-2, and mirror polished plates with (a) heater output and (b) flow rate.

## 3.3 Effect of polishing the surface

Figure 9 shows the local heat transfer coefficient (*h*) for the five types of plate surfaces. The figure shows the changes in h with vapor quality for the different operating conditions (flow rate: 40 mL/min, heater output: 1000 W, system pressure: 0.3 MPa). As shown in Figure 9, h decreased in the order F0 > F2 >F-2 > F-1 > mirror; in addition, h is dependent on the differences in the roughness of the F2 and F0-polished plates at a vapor quality of around 0.2. We believe that the difference depends on the flow-pattern (Arima et al., 2010). The difference is remarkable from nuclear boiling to slug flow at a vapor quality of ca. 0.2, but the difference is negligible in annular flow at a vapor quality of 0.35. In the vapor quality range of 0.3 to 0.4, values of h approximately six-times that of the mirror-polished plate were observed for the F0- and F2-polished plates. However, for the F-1 and F-2-polishing plate, the heat transfer coefficients were smaller than those of F0 and F2-polished plates, and the local heat transfer coefficient was approximately three times that of the mirror-polished plate. The local heat transfer coefficient at a vapor quality of 0.35 of the different types of polished surface are summarized in Figure 10.



Fig. 9 Comparison of the local heat transfer coefficient versus vapor quality for different polished plates.

Based on these results, a significant heat transfer enhancement was confirmed with F-polished plates for heat transfer in boiling water. Possibly, many cavities were formed, which could yield bubbling points in the case of boiling induced by F polishing. Because the F-1 and F-2-polished surfaces have large surface roughness values, the individual cavities were larger than those of the F0 and F2-polished surfaces. As a result, the number of possible bubbling points decreased. In the studied experimental range, we found that the number and frequency of cavities contributed significantly more to the improvement of thermal conductivity than the size of the cavities.



Fig. 10 Relationship of local heat transfer coefficient and surface rooughness.

## 3.4 Effect of system pressure

Figure 11 shows the heat transfer coefficients for two kinds of plate surfaces (F2 and F0). The figure shows the changes in local heat transfer coefficient with vapor quality at set operating conditions (flow rate: 40 mL/min, heater output: 1000 W, system pressure: 0.3 or 0.4 MPa). Although the coefficient is independent of the pressure on the F0-polished plate, the coefficient increased with increasing pressure above a vapor quality of 0.3 for the F2-polished surface. This result indicates that there was a heat transfer enhancement and an increase in the number of bubbling points with increasing pressure for the low-roughness F2 plate because the number of active cavities on the plate surface increased with increasing pressure (Jpn. Soc. Mech. Eng. 2005). That is, under low pressure, larger active cavities only are used as bubbling points. However, at high pressure, water can enter the smaller active cavities and boil at the boiling points.



**Fig. 11** The relationship between vapor quality and the local heat transfer coefficient on each plate at different pressures: (a) F2 and (b) F0.

## 3.5 Effect of flow rate

**Figure 12** shows the heat transfer coefficients for three kinds of plate surfaces (F2, F0, and F-2). The figure shows the changes in the local heat transfer coefficient with vapor quality at different operating conditions (flow rate: 20, 40 and 60 mL/min, heater output: 1000 W, system pressure: 0.3 MPa). For all polishing conditions, the coefficient increased with

increasing flow rate because the bubble formation cycle in the boiling water was promoted by the increasing flow rate.



Fig. 12 The relationship between vapor quality and local heat transfer coefficient on each plate at different flow rate: (a) F2, (b) F0, and (c) F-2.

**Figure 13** shows the Nusselt numbers (Nu) for three kinds of plate surfaces (F2, F0, and F-2) at a vapor quality of 0.2. Nuincreased with increasing flow rate for all polishing conditions. The difference in Nu with polishing condition became significant as the flow rate increased. The results indicate that the local heat transfer coefficient increases with increasing flow rate.



Fig. 13 Relationship between flow rate and Nusselt number on each plate (F2, F0, and F-2).

# Conclusions

In this study, the effects of polishing SUS surface on boiling water heat transfer were investigated, and the following conclusions were drawn.

- 1. An enhancement in heat transfer on the F-polished plates was confirmed under various operating conditions.
- For the F0 or F-2-polished surfaces, the heat transfer coefficient on the surfaces polished perpendicularly to the water flow direction was much larger than that on the surfaces with parallel and mirror polishing.
- 3. For the F0 and F2 perpendicularly polished plates, a heat transfer enhancement of ca. six-times that of the mirror-polished plate was obtained at vapor qualities of 0.3 to 0.4. However, for the F-1 and F-2 perpendicularly polished plates, the heat transfer coefficients were smaller than those of the F0 and F2-polished plates.
- 4. In the studied experimental range, the number and frequency of the cavities contributed more significantly to the improvement in thermal conductivity than the size of the cavities.

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#### Nomenclature

$A_{\rm n}$	= local heat transfer area	[m <sup>2</sup> ]
С	= specific heat	[J/kg·K]
$D_{ m h}$	= heated equivalent diameter	[m]
h	= heat transfer coefficient	$[W/m^2K]$
$i_{\rm test,in}$	= specific enthalpy at the heat transfer test plate	entrance
		[J/kg]
$i_{\text{test,n}}$	= local specific enthalpy in test plate	[J/kg]
$i_{ m fg}$	= latent heat of evaporation	[J/kg]
<i>i</i> sat,liq	= specific enthalpy of saturated liquid	[J/kg]
$k_{ m sus}$	= thermal conductivity of the heat transfer plate	
		$[W/m \cdot K]$
$l_1$	= distance between a pair of the heat transfer su	rface and
	the working fluid side of the thermocouple	[m]
$l_2$	= distance between a pair of thermocouples	[m]
т	= mass flow rate	[kg/s]
Ν	= segment number	[—]
Nu	= Nusselt number	[—]
$Q_{\rm n}$	= local heat flow	[W]
q	= heat flux	$[w/m^2]$
$q_{ m n}$	= local heat flux	$[W/m^2]$
$T_1$	= plate temperature of heater side	[°C]
$T_2$	= plate temperature of flow channel side	[°C]
$T_{\rm in}$	= temperature at the heat transfer test plate entra	nce [°C]
$T_{\rm sat}$	= saturation temperature	[°C]
$T_{\text{wall}}$	= surface temperature of the heat transfer plate	[°C]

- $\chi$  = vapor quality
- $\lambda$  = water thermal conductivity [W/m·K]

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