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4	Urochloa ruziziensis cover crop increases the cycling of soil inositol
5	phosphates
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18	Acknowledgments: This research was supported by São Paulo Research Foundation
19	(FAPESP) grants #2014/23707-5 and #2015/04200-0. This work was performed as part of the
20	Organic Phosphorus Utilisation in Soils (OPUS) project, funded by Biotechnology and
21	Biological Sciences Research Council (BBSRC) responsive mode grant (BB/K018167/1) in
22	the UK to explore cropping strategies to target the use of recalcitrant soil organic phosphorus.
23	We would also like to thank the Coordination for the Improvement of Higher Education
24	Personnel (CAPES) for granting a scholarship to the first author.
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28 Abstract

Ruzigrass (Urochloa ruziziensis) is a cover crop that is commonly used in Brazil and exudes 29 high concentrations of organic acids from its roots, and is therefore expected to mobilize soil 30 31 organic P such as inositol phosphates. However, is it not known if this can occur only under P 32 deficient conditions. Specifically, we aimed to test the hypothesis that the degradation of inositol phosphates is increased by growing ruzigrass at two different P levels. To investigate 33 this, we studied soil organic P in a nine-year old field experiment, with treatments consisting 34 of ruzigrass or fallow during the soybean (Glycine max) off-season, with or without P addition. 35 36 Organic P was extracted in NaOH-EDTA, followed by colorimetric quantification of organic P hydrolysable by phytase, and *myo*-inositol hexakisphosphate by hypobromite oxidation and 37 HPLC separation. Ruzigrass dry matter yield increased by about 80% with P application. 38 39 Ruzigrass reduced the concentration of phytase-labile P and myo-inositol hexakisphosphate, 40 but only in soil receiving P. A corresponding increase in unidentified inositol phosphates, presumably representing lower-order esters, was also observed after ruzigrass in soil with P 41 42 application. We deduce that the degradation of inositol phosphates under ruzigrass with P application is due to greater ruzigrass productivity in the more fertile treatment, increasing the 43 release of root exudates that solubilize inositol phosphates and promote their decomposition by 44 phytase. We conclude that ruzigrass cover cropping can promote the cycling of recalcitrant soil 45 organic P, but only when fertility is raised to a sufficient level to ensure a productive crop. 46

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48 Keywords: *myo*-inositol hexakisphosphate, organic phosphorus, no-till, cover-crops,
49 hypobromite oxidation.

51 Introduction

The use of ruzigrass [Urochloa ruziziensis (R. Germ. and C.M. Evrard) Morrone and Zuloaga] 52 in the off-season of cash crops in systems under no-till has been largely adopted in Brazil as a 53 54 conservationist practice able to keep the soil covered over the off-season with minimal disturbance, to reduce soil erosion, promote weed control (Franzluebbers et al. 2014), and even 55 to promote soil phosphorus (P) cycling, due to its high production of biomass (Almeida and 56 Rosolem 2016). Recent studies have promoted the notion that growing certain species able to 57 scavenge recalcitrant P forms can affect P cycling (Calegari et al. 2013; Merlin et al. 2014). 58 59 Inorganic P fertilizers are the main source of P in agriculture, with a low use efficiency, and are obtained from phosphate rocks, which is a non-renewable source (MacDonald et al. 2011). 60 Thus, the adoption of management practices able to improve P cycling is considered a step into 61 62 sustainability of agricultural systems (Menezes-Blackburn et al. 2017). However, most species 63 and microorganisms express their traits to scavenge recalcitrant P in P deficient soils (Li et al. 1997; Olander and Vitousek 2000), which is an unusual condition in agricultural soils. The 64 high P application rates associated with the high P fixing capacity of tropical soils leads to most 65 of added fertilizer P being fixed and accumulated in recalcitrant forms, such as inositol 66 67 phosphates (MacDonald et al. 2011; Rodrigues et al. 2016). To seek and identify species able to promote soil P cycling, not only in P deficient soils is, therefore of potential benefit to 68 sustainable management of agricultural areas in the tropics (Lagos et al. 2016; Menezes-69 70 Blackburn et al. 2017).

Ruzigrass has a C4-type photosynthesis and is well-adapted to low P availability due to a high P uptake efficiency (Rao et al. 1996). According to Onthong and Osaki (2006), the increase in the root:shoot ratio is the main P uptake mechanism in ruzigrass. Furthermore, ruzigrass can mobilize and take up soil recalcitrant P bound to Fe and Al (Merlin et al. 2015). The P uptake from non-labile P forms by ruzigrass may be related to an extensive root system 76 and the ability of roots to exude low-molecular weight organic acid anions, such as citrate and 77 oxalate (Louw-Gaume et al. 2017; Wenzl et al. 2001). The high exudation of organic acids from ruzigrass roots may also mobilize organic P (P₀) forms (Martin et al. 2004). At least two 78 79 mechanisms have been related to P mobilization by organic acid anions: the competition of organic acid for P adsorption sites on the surface of Fe and Al oxides, leading to P desorption, 80 and the complexation of metal cations by organic acid anions inducing the solubilization of 81 82 precipitated P (Hinsinger 2001; Jones 1998). The exudation of phosphatases from ruzigrass roots is also a mechanism to hydrolyze P_o, accounting for the high adaptation of ruzigrass to 83 84 low P soils (Louw-Gaume et al. 2017).

Cycling of Po has been considered as key factor to meet plant P demand, mainly in 85 weathered soils under no-till, where the proportions of Po tends to increase in the long-term 86 87 (Rodrigues et al. 2016). In Brazilian Oxisols under no-till, it has been shown that Po 88 corresponds to 20 to 50% of total P (Costa et al. 2014; Olibone and Rosolem 2010; Rodrigues et al. 2016). This P_o portion consists in a range of compounds including orthophosphate esters, 89 90 phosphonates, and phosphoric acid anhydrides (Condron et al. 2005). The orthophosphate monoesters include the inositol phosphates, which have been shown to be the dominant forms 91 of Po in many soils (Shears and Turner 2007; Turner et al. 2002), including Oxisols (Chapuis-92 Lardy et al. 2001). In particular, myo-inositol hexakisphosphate (also known as IP₆, InsP₆ or 93 phytate) is one of the nine stereoisomers of inositol phosphates, with all the six hydroxyl groups 94 95 esterified as phosphates moieties (Shears and Turner 2007). Among all soil Po compounds, myo-inositol hexakisphosphate has the highest affinity to soil particles and the lowest 96 bioavailability (Martin et al. 2004; Shang et al. 1996), which results in its accumulation in soils 97 98 (Giaveno et al. 2010). The high soil-stability of myo-inositol hexakisphosphate is due to the capacity of numerous of its six phosphate groups to bind with cations and soil adsorption sites 99 simultaneously (Celi et al. 1999; Shang et al. 1996). Inositol phosphate sorbs strongly to Fe 100

and Al oxides, kaolinite, montmorillonite, and soil organic matter (SOM) (Celi and Barberis
2007; Karathanasis and Shumaker 2009).

The extent to which inositol phosphates accumulate in soil depends on the presence of 103 104 active phytate degrading enzymes (phytases), environmental conditions (pH, temperature, soil texture, and soil mineralogy), enzyme inactivation/inhibition, phytase mobility, and phytate 105 106 solubility (Menezes-Blackburn et al. 2013; Nannipieri et al. 2011; Turner et al. 2002). Phytases 107 are a special group of phosphatases that can initiate the dephosphorylation of *myo*-inositol hexakisphosphate (Shears and Turner 2007). In cases where the extracellular activity of phytase 108 is present in soils, the phytase labile P (P_{phy-lab}) may be depleted and made available to plants 109 (Yadav and Tarafdar 2004). A very small proportion of phytases (less than 1% of total 110 phosphatase activity) was observed in ruzigrass roots by Louw-Gaume et al. (2010), and in a 111 112 study conducted by Onthong and Osaki (2006) no phytase secretion by ruzigrass roots was observed. However, some phytase activity in soil under ruzigrass has been shown (Louw-113 Gaume et al. 2017), due to soil microbial activity, which provides an important mechanism by 114 which plants can acquire P from recalcitrant P_0 forms (Harrison 1987; Richardson et al. 2001). 115 A higher phyatase activity has been observed mainly in the rhizosphere due to the favorable 116 conditions for microbial proliferation (Yadav and Tarafdar 2004), which should also be favored 117 in no-till, with higher SOM content(Alvear et al. 2005). 118

Understanding the complex interrelations that affect soil P bioavailability is a challenge, but also a great opportunity for the development of technologies to reduce the accumulation of recalcitrant P_0 forms in soil (Haygarth et al. 2018; Menezes-Blackburn et al. 2017). Most information on the soil P_0 composition has come from ³¹P nuclear magnetic resonance (³¹P-NMR) spectroscopy of alkali-extractable P (Cade-Menun 2017). The use of high performance liquid chromatography (HPLC) to identify and quantify soil P_0 forms has been proposed as an alternative technique, potentially more sensitive and less expensive than 126 ³¹P-NMR spectroscopy. Espinosa et al. (1999) previously used HPLC to identify and quantify P_0 forms in soil leachates, but this method does not achieve high resolution and is unsuitable 127 for the NaOH-EDTA extracts commonly used for ³¹P-NMR spectroscopy. An alternative to 128 129 improve the resolution and detection of soil P_0 forms is the use of hypobromite oxidation of soil NaOH-EDTA extracts (Turner and Richardson 2004). The hypobromite oxidation digests 130 non-phytate organic matter (Irving and Cosgrove 1981) and therefore eliminates organic matter 131 interferences in HPLC separation of inositol phosphates, which may turn possible the use of a 132 technique much more accessible than ³¹P-NMR spectroscopy. 133

134 The objective of this study was to assess the effect of long-term soybean-ruzigrass rotation and P fertilizer application on the dynamics of soil P₀ forms, mainly the *myo*-inositol 135 hexakisphosphate, using an innovative approach by performing hypobromite oxidation of soil 136 137 extracts and separation of *myo*-inositol hexakisphosphate with HPLC. Specifically, we aimed to test the hypothesis that the degradation of inositol phosphates is increased by growing a 138 known soil P mobilizer cover crop (Ruzigrass) at two different P levels, in comparison to 139 140 fallow. This study aims to improve the understanding of the soil P cycling and dynamics using crop rotation in the tropics. 141

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143 Material and methods

144 *Experimental site and treatments*

A long-term field experiment established in 2006 was monitored for two years (2014 and 2015). The experiment is located in Botucatu, State of São Paulo, Brazil (22°50′00″ S; 48°25′31″ W; and altitude of 806 m) in an area that had been under no-till since 1998. The climate of the region, according to Köppen's classification, is Cwa (mesothermic with dry winter climate). The dry season is well defined from May to September. The average annual precipitation is 1,450 mm, with mean temperatures of 24 °C during the warmest month and 13 151 °C during the coldest month (Cepagri 2015). The soil is a Rhodic Hapludox (Soil Survey Staff, 2014) with 67% of sand, 12% of silt, and 21% of clay. From 2006, soybean [Glycine max (L.) 152 Merrill] has been grown in rotation with ruzigrass [Urochloa ruziziensis (R. Germ. and C.M. 153 154 Evrard) Morrone and Zuloaga] or fallowed during the off-season, with or without the application of 26 kg ha⁻¹ of P as triple superphosphate (TSP) in the soybean seed furrow. 155 Therefore, for ruzigrass, there was only the residual effect of P application to soybean. The 156 experiment design was a 2×2 factorial in randomized complete block design, with four 157 replications. Plots were 8 m long \times 6 m wide. 158

159 The off-season period starts after soybean harvest in April and goes to November in the soybean sowing. Ruzigrass was planted shortly after soybean harvest and all the plots (with 160 ruzigrass and fallow treatments) were desiccated using glyphosate (1.44 kg ha⁻¹ a.i.) in 161 November. Spontaneous vegetation grew in the fallow plots, with the predominance of 162 Rhynchelytrum repens (Willd.) C.E. Hubb., and Cenchrus echinatus L., Gnaphalium spicatum 163 Lam. However, weed growth was random and sparse, with insignificant dry matter production 164 by the end of the off-season period. Four subsamples of ruzigrass shoots were randomly taken 165 using a 0.5 m x 0.5 m frame and combined into one sample per plot, oven dried at 60 °C for 72 166 h, and then weighed to determine dry matter yield at November 2014 and 2015. 167

Soil samples were taken from depths of 0–5, 5–10, 10–20 and 20–40 cm in November 168 169 2014 and 2015, before chemical desiccation with herbicide. Six soil subsamples were randomly 170 taken from each plot using a 50 mm diameter core sampler, and were combined into one composite sample per depth per field replication. The soil samples were air-dried and passed 171 through a 2-mm sieve for chemical analysis. Briefly, soil pH was determined in a 0.01 M 172 173 calcium chloride (CaCl₂) suspension (1:2.5 soil/solution), potential acidity (H+Al) was estimated by the Shoemaker-McLean-Pratt (SMP) pH buffer method (Shoemaker et al. 1961), 174 Ca, Mg, K, and P were extracted using pearl resin according to Raij et al. (1986), cation 175

exchange capacity (CEC) was calculated as the sum of (H+Al)+Ca+Mg+K, and soil organic
matter (SOM) was determined by chromic acid wet oxidation method (Walkley and Black
1934). Selected chemical characteristics of the soil of the experimental area are presented in
Table 1.

180

181 Soil phosphorus extraction

Soil P was extracted with 0.25 M sodium hydroxide and 0.05 M disodium 182 ethylenediaminetetraacetic acid (NaOH-EDTA). The NaOH-EDTA solution is a widely used 183 184 extractant for soil P_o analysis and has been suggested as standard method for soil P_o extraction (Cade-Menun and Liu 2014). One gram of air-dried soil was extracted with 10 mL of NaOH-185 EDTA on a reciprocal shaker for 16 h. The extracts were then filtered using Whatman N1 filter 186 187 paper and analyzed for total P (P_t), inorganic P (P_i), and organic P (P_o). The P_i concentration was determined immediately after extraction. Total P was obtained by autoclave digestion for 188 1 h at 121 °C and 124 kPa, with potassium persulfate and sulfuric acid. Total P and P_i were 189 190 obtained with the molybdate reactive P, which was quantified using the malachite green colorimetry of the digested and undigested extracts respectively (Van Veldhoven and 191 Mannaerts 1987). The difference between Pt and Pi corresponds to the molybdate unreactive P, 192 here termed as Po. Figure 1 contains a scheme of the soil P extraction and the subsequent 193 determinations (Pi, Po, Pt, phytase labile P before hypobromination, steps of hypobromine 194 195 oxidation of NaOH-EDTA soil extract, phytase labile P after hypobromination, and P speciation). 196

197

198 Phytase labile phosphorus

199 Phytase labile P (P_{Phy-lab}) was determined using a commercially available Aspergillus niger

200 phytase (Natuphos, EC 3.1.3.8; BASF SE, Ludwigshafen, Germany) with high activity towards

sodium phytate (Wyss et al. 1999). The phytase was added to a final activity in excess of 50 nKat mL⁻¹. Briefly, 100 μ L of the same soil extracts obtained with NaOH-EDTA were combined with 100 μ L of phytase (100 nKat mL⁻¹) diluted in a buffer (50 mM acetate-acetic acid, pH 5.5) and incubated at 37 °C for 16 h. Phytase labile P was measured via malachite green colorimetry. Phytase labile P was inferred by the difference of P_i content measured after the incubation of samples with phytases, and samples incubated with denatured phytases (autoclaved for 1 h at 121 °C and 124 kPa) were used as blanks.

208

209 Hypobromite oxidation

The hypobromite oxidation was performed according to Irving and Cosgrove (1981), following 210 211 the modifications described by Turner et al. (2012), in the same soil extracts obtained after soil 212 P extraction with NaOH-EDTA. Prior to hypobromite oxidation, the pH of the soil extract was increased to 13 (to produce hypobromine after add the bromine), by adding 2 g of solid NaOH 213 to a digestion tube containing 10 mL of soil extract, and the digestion tubes were cooled using 214 an ice-bath. Then, 1 mL of pure bromine was added to the soil extract. After 1 hour, the extract 215 was boiled at 140 °C for 5 min to hydrolyse the non inositol P. To release all the bromine from 216 the extract, the pH was lowered by adding 10 mL of 10 M hydrochloric acid. After bromine 217 volatilization, the pH was adjusted to near 8.5 by adding 6 mL of 10 M NaOH. The final volume 218 was adjusted to 30 mL with ultra-pure Milli-Q (MQ) water (Millipore). The phosphates from 219 220 the extract were precipitated with barium, by adding 15 mL of barium acetate (10% w/v), and 15 mL of ethanol (50% v/v). The suspension was then centrifuged at $4000 \times g$ for 10 minutes. 221 The supernatant was discarded, and the precipitate was washed with 30 mL of 50% ethanol, 222 centrifuged, and the supernatant was discarded. The barium-phosphate precipitate were 223 resuspended in 20 mL of amberlite IR120 cation exchange resin (hydrogen form) and 20 mL 224 of MQ water, and shaken for 16 hours. Then, the extracts containing the remaining P forms 225

(inositol phosphates) and orthophosphate were separated from the resin by syringe-filtering ($0.2 \mu m$) and the final volume of each sample was adjusted to 35 mL. The extracts were neutralized and subjected to P_i, P_t determination using the malachite green colorimetry of the digested and undigested extracts respectively (Van Veldhoven and Mannaerts 1987), and phytase labile P determination as described in the previous section. Due to practical issues related to the large number of samples and the lengthiness of this protocol, only the soil depth of 0–10 cm was assayed with hypobromite oxidation.

233

234 Phosphorus speciation

The extracts obtained from the hypobromite oxidation were separated using High Performance 235 Liquid Chromatography (HPLC), to determine the component P species, as in Espinosa et al. 236 237 (1999). The separation was performed on an Agilent 1100 Series HPLC system (Agilent Technologies, California, U.S.), with an Agilent IonoSpher 5 A column (length 250 mm, 238 internal diameter 4.6 mm). The extracts were pH adjusted to 5.5 with 1 M HCl and diluted in 239 240 MQ water (1:10), and 30 µL of the diluted extract was injected into the HPLC column. The ionic linear gradient increased from 0.11 to 0.75 M NaCl over the 160-minute run, with a 241 constant concentration of 0.5 mM EDTA and 50 mM MES. The flow rate was maintained at 242 0.5 mL min⁻¹ and the buffers were pH adjusted to 5.5. A fraction collector was used to collect 243 1 mL of eluted sample every 2 min in glass vials. The eluted sample was then used for the 244 245 quantification of Pt, Pi and PPhy-lab as described in the soil P extraction and phytase labile P sections. To confirm elution time of the P component peaks, a mixture of 100 µM sodium 246 phosphate and 100 µM sodium myo-inositol hexakisphosphate, Sigma-68388 (Sigma-Aldrich, 247 248 Missouri, U.S.), were used as a standard.

250 *Statistical analysis*

Data were first examined for homogeneity of variance using Levene's tests. Then, the results were subjected to analysis of variance by soil depth, considering a 2×2 factorial in randomized complete block design, with four replications, using a general linear model (Proc GLM) in SAS software (version 9.4, SAS Inst., North Carolina, U.S.). When the F test was significant (p < 0.05), treatment means were compared by Student's t-test (p < 0.05).

256

257 **Results**

Ruzigrass yield was almost twice as high in soil with P application than without P, in both
years (Fig. 2). However, the yield in 2014 was lower than 2015, due to a low rainfall during
the 2014 off-season.

261 On average, 40% of the Pt extracted with NaOH-EDTA corresponded to Po in the 0–20 cm of soil depth, in both years. Despite the cumulative increase of SOM by growing ruzigrass 262 as cover crop compared to fallow (Table 2), Po concentration was not affected by ruzigrass. 263 The concentration of Pt extracted with NaOH-EDTA was also not affected by the long term 264 ruzigrass cover crop rotation (Table 3). The P fertilizer application affected Pt and Po up to the 265 10-20 cm soil depth (Table 3). At the 20-40 cm soil depth, there was no effect of the 266 treatments, and Pt and Po averaged 57 and 23 mg kg⁻¹, respectively, in 2014 as well as in 2015. 267 The P_{phy-lab} concentration before the hypobromite oxidation of the NaOH-EDTA was 268 269 lower in the uppermost soil layer with ruzigrass than after fallow in soil fertilized with P in 2014 (Table 3). In 2015, P_{phy-lab} was lower after growing ruzigrass than after fallow regardless 270 of P application, at the depth of 0-5 and 5-10 cm. The P_{phy-lab} concentration was lower with P 271 272 fertilizer application than without (Table 3). There was no effect at the 20-40 cm soil depth, where the average P_{phy-lab} was 5 and 6 mg kg⁻¹, in 2014 and 2015, respectively. 273

The P_{phy-lab} concentration after hypobromite oxidation of the NaOH-EDTA extracts was markedly higher than before hypobromite oxidation in soil receiving P fertilizers (Table 4). However, there was no effect of ruzigrass on P_{phy-lab} concentration after hypobromite oxidation (Table 4).

The HPLC retention time of P_i and the *myo*-inositol hexakisphosphate was on average 278 30 min apart, and phytate was the last observed peak. The confirmation of the *myo*-inositol 279 hexakisphosphate peak was obtained both by co-elution with a standard and by phytase 280 digestion of the collected HPLC fractions. A clear resolution of the soil extracts digested with 281 282 bromine were obtained using the chosen HPLC method, as shown in Figure 3, nevertheless, peaks were somewhat broad, and resolution could be improved in further studies by coupling 283 the HPLC to an inductively coupled plasma (ICP) for the online detection of the eluted 284 285 phosphorus forms. Other unidentified small peaks were observed and labeled here as unidentified inositol phosphates. 286

The P_i concentration in the brominated NaOH-EDTA extracts was not different between ruzigrass and fallow treatments (Table 5). The concentration of unidentified inositol phosphates increased after ruzigrass in soil with P application compared to fallow, in both years. Growing ruzigrass resulted in lower *myo*-inositol hexakisphosphate concentration in soil with P application than in soil without P application (Table 5). In contrast, the fallowed soil with P application showed a higher *myo*-inositol hexakisphosphate concentration than soil without P application.

295 Discussion

Previous studies have shown that ruzigrass grown as a cover crop promotes the accumulation 296 of P_o in the upper soil layers in the long term (Almeida and Rosolem 2016; Merlin et al. 2014), 297 298 despite an increase in the phosphatase activity in soil (Rosolem et al. 2014). Here we intended to have a better understanding of field dynamics of Po species under ruzigrass and specifically 299 explore the following questions: how the concentration of *myo*-inositol hexakisphosphate and 300 P_{phy-lab} is affected by the agricultural system and if it gives insights on the field inositol 301 phosphate cycling. How strong is the effect of P fertilizer application on these dynamics? Are 302 303 there possible management perspectives from these results?

304

305 Phytase labile phosphorus and hypobromite oxidation

306 Soil Po accounted for 40% of the total P, which is in agreement with that observed by Chapuis-Lardy et al. (2001) in weathered soils from Brazil. However, there was no effect of ruzigrass 307 on soil P_o compared with soils kept fallow during off-season, despite the higher SOM content 308 309 after ruzigrass. The soil P₀ concentration is not necessarily correlated to the SOM concentration; nonetheless, there is usually a good correlation of soil Po and total P 310 concentration (Appelhans et al. 2016), as observed in the present study, where soil Po 311 concentration was higher with P application. It is worth mentioning that the increase of soil P_o 312 by applying inorganic P fertilizers is a long-term effect, conversely, only an effective increase 313 314 in the P_i should be expected in the short-term by applying inorganic P fertilizers. According to George et al. (2007), the continuous application of inorganic phosphate increases both soil P_i 315 and Po concentrations. The Po increases with inorganic P fertilizer applications is a result of a 316 317 higher P uptake by plants and the consequential greater amount of plant residue deposition (shoot and roots) as well as through the synthesis of Po by soil microorganisms (Stewart and 318 319 Tiessen 1987).

320 Interestingly, the soil P_{phy-lab} and HPLC quantified *myo*-inositol hexakisphosphate were lower after ruzigrass in soil with P fertilizer application in comparison with both fallow and 321 unfertilized treatments, indicating a synergistic effect of P fertilizer application and ruzigrass 322 323 in increasing phytate bioavailability and cycling in these soils. However, despite a lower myoinositol hexakisphophate concentration after ruzigrass, the Pt has not changed, but it is 324 important to keep in mind that this is not the Pt from soil, it is the Pt extracted with NaOH-325 EDTA, corresponding to only a fraction of the soil Pt. It is also important to note that the Pphy-326 lab is not necessarily myo-inositol hexakisphosphate, since the P_{phy-lab} method use phytases with 327 328 broad substrate specificity, and therefore includes non-phytate orthophosphate monoesters (Menezes-Blackburn et al. 2013). The P_{phy-lab} method quantifies P_o lability, whilst the HPLC 329 quantifies myo-inositol hexakisphosphate concentration, and perhaps HPLC could be used to 330 331 quantify other hexakisphosphate isomers, such as *scyllo*-inositol hexakisphosphate.

The hypobromite oxidation of NaOH-EDTA soil extracts increased the measured P_{phy-} 332 lab concentration mainly in the treatment with P application, which is the opposite of the 333 observed before hypobromite oxidation. This indicates that a significant part of P_{phy-lab} in the 334 NaOH-EDTA was not available to the phytase used. Is important to note that although the 335 phytase used is only one of many phytases that might be found in soil, it has a wide substrate 336 specificity. According to Hayes et al. (2000), both substrate availability and enzyme 337 presence/activity are determinant of the hydrolysis of inositol phosphates in soils. Interactions 338 339 with surface-reactive particles and the entrapment of phytases within humic molecules extracted from soil may act inhibiting the phytase activity (Nannipieri et al. 2011). A strong 340 adsorption of inositol phosphates has been demonstrated in many soil compounds, such as 341 342 calcite, illite, montmorillonite, goethite, and Al hydroxides, which limits the action of the enzyme (Menezes-Blackburn et al. 2013). According to Bowman and Moir (1993), NaOH 343 promotes the solubilization of Po, whereas EDTA is able to complex cations that binds Po to 344

soil solid phase, overcoming a possible resistance of P_o to the extraction by NaOH. Possibly, part of the phytase added to the soil extract was inhibited by interactions with colloids remaining in the extracts after filtering and/or part of inositol phosphates precipitated or adsorbed to the soil colloids remaining in the extracts were not solubilized with NaOH-EDTA extraction, and were only accessed by the enzyme after hypobromite oxidation.

The concentration of P_i measured after hypobromite oxidation was about 16 mg kg⁻¹ 350 higher than before hypobromite oxidation. This increase in the Pi corresponds to 37% of the 351 total P₀ extracted with NaOH-EDTA, and this increase is due to the hydrolysis of P₀ forms that 352 353 are not resistant to the hypobromite oxidation (Irving and Cosgrove 1981). The hypobromite oxidation has been already successfully used to digest non-phytate organic matter in soil 354 355 extracts for ³¹P-NMR analysis (Turner et al. 2012; Turner and Richardson 2004). This 356 degradation of other organic compounds by the hypobromite oxidation was crucial to ensure a 357 pure extract, free of SOM interferents, and thereby allowing the use of HPLC for the determination of inositol phosphates in the soil NaOH-EDTA extracts. 358

359

360 New insights into the dynamics of myo-Inositol hexakisphosphate in soil with ruzigrass

Clearly, the HPLC analysis of NaOH-EDTA extracts after hypobromite oxidation allowed for 361 new insights into the dynamics of recalcitrant Po forms in soil. The higher concentration of 362 P_{phy-lab} observed after hypobromite oxidation was not necessarily exclusively due to myo-363 364 inositol hexakisphosphate, since the presence of other forms of P with retention times greater than orthophosphate and smaller than myo-inositol hexakisphosphate were found. These P 365 forms were termed unidentified inositol phosphates, and are possibly products of myo-inositol 366 367 hexakisphosphate degradation (e.g. *myo*-inositol 1,2,3,4,5-pentakisphosphate). The inositol phosphates (myo-inositol hexakisphosphate + unidentified inositol phosphates) accounted for 368

about 30% of the total P extracted with NaOH-EDTA, which is consistent with results from
Cerrado soils in Brazil (22-39%) (Chapuis-Lardy et al. 2001).

The use of ruzigrass as a cover crop combined with P fertilizer applications resulted in 371 lower soil concentrations of myo-inositol hexakisphosphate and higher concentrations of other 372 forms of inositol phosphates, when compared to soil without P application. Several studies 373 have shown that the application of soluble phosphates results in the suppression of phosphatase 374 activity, with consequent increase of P_{phy-lab} in the soil (George et al. 2007; Olander and 375 Vitousek 2000; Rosolem et al. 2014), since the higher availability of P reduces the demand for 376 377 P_o mineralization (Turner et al. 2002). However, in the present study, P was applied at soybean planting, about one year before soil sampling, thereby, this may be considered as a residual 378 379 effect of P application. In addition, the concentration of P in fertilized soil is still considered low, resin-P is below 40 mg dm⁻³. The low resin-P concentration is associated with the low 380 total soil P concentration, on average 400 mg kg⁻¹ (Almeida and Rosolem 2016), which is much 381 lower than that observed in some European soils that receive large amounts of phosphate 382 fertilizers (over 1000 mg kg⁻¹) over a long period (Menezes-Blackburn et al. 2017). 383

Ruzigrass may have favored the enzyme-producing microbial community capable of 384 degrading inositol phosphates, since the various organic compounds exuded by ruzigrass roots, 385 such as organic acid anions (Wenzl et al. 2001), are a source of energy for soil microorganisms 386 (Hinsinger 2001; Menezes-Blackburn et al. 2016). Furthermore, the exudation of organic acid 387 388 anions, such as citrate, by roots can complex cations and compete with soil sorption sites, preventing the adsorption of enzymes on soil particles (Hayes et al. 2000). Nevertheless, 389 several factors affects the activity of enzymes in soil, such as electrostatic interactions, enzyme 390 391 entrapment/adsorption, presence of inhibitors, and type of mineral precipitates formed with the substrate as revised by Nannipieri et al. (2011). A higher activity of phosphatases, including 392 393 phytases, has been shown from soil microbial activity in the rhizosphere of ruzigrass (Louw394 Gaume et al. 2017; Rosolem et al. 2014). Even with the addition of P fertilizer, the activity of acid phosphatase in soil with ruzigrass is higher than in soil under fallow, as observed 395 previously by Rosolem et al. (2014) using soil from the same experimental area of the present 396 397 study. It is well stablished that the phosphatase activity is high in soil with ruzigrass (Simon et al. 2017). Compared to fallow and other cover crops, such as sorghum (Sorghum bicolor), 398 millet [Pennisetum glaucum (L.) R. Br.], stylosanthes (Stylosanthes spp / CV. BRS), forage 399 turnip (Raphanus sativus L.), crambe (Crambe abyssinica Hochst), soil with ruzigrass showed 400 the highest activity of acid phosphatase (Simon et al. 2017). The increase of acid phosphatase 401 402 activity in soil with ruzigrass has been correlated with an increase of P_i concentration (Louw-Gaume et al. 2010), showing an effective P cycling. 403

404 The lower concentration of P_{phy-lab} after ruzigrass than after fallow, may also have been 405 a result of the phytase activity from the ruzigrass roots. According to Louw-Gaume et al. 406 (2010), ruzigrass roots show some phytase activity, which is not affected by P applications. Ruzigrass is a highly adapted species to tropical soils with low P availability (Begum et al. 407 408 2006), with roots able to increase P acquisition by physiological (Merlin et al. 2015) and morphological adjustments (Louw-Gaume et al. 2010; Wenzl et al. 2001). In addition, the 409 continuous input of a great amount of ruzigrass residues, as well as the higher SOM 410 concentration after ruzigrass than fallow may result in higher soil moisture, and fewer 411 412 oscillations in soil temperature (Awan 1964), favoring the soil microbial activity. According 413 to Nannipieri et al. (2011), it is well stablished that phosphatase activity is correlated with the 414 content of SOM, which is usually higher in no-till systems. However, the lower concentration of P_{phy-lab} after ruzigrass was only observed in soil with P application, which may indicate that 415 416 P application was able to promote a more pronounced priming effect of soil phytate under ruzigrass than in fallow. As observed by Lagos et al. (2016), Luo et al. (2017), and Margenot 417 418 et al. (2017) the P application does not necessarily suppress the activity of microorganismsharboring phosphatases. Some studies have attributed the increase in phosphatase activity to the increase in SOM (Alvear et al. 2005), P_0 (Redel et al. 2007), and microbial biomass (Costa et al. 2013), when P is applied.

422 Since the ruzigrass yield was almost twice higher in soil with P application, a greater amount of organic compounds exuded by ruzigrass roots should be expected due to the 423 increased root biomass, favoring the proliferation of microbial community, complexing cations 424 (Lienhard et al. 2012), and solubilizing inositol phosphates (Gerke 2015; Martin et al. 2004). 425 Despite a greater organic acid anion exudation per length of root be expected under low P soils 426 427 than under high P soils, a greater amount of organic acid anion is also expected with the increase of root length. Additionally, relieving nutrient limitation by applying P favors the 428 429 increase of crop biomass production and result in greater residue additions to soil (Margenot et 430 al. 2017), and soil C may stimulate microbial activity because C has been found to be more 431 limiting than P in some P-fertilized soils, such as in Kenya (Bünemann et al. 2004). According to Gerke (2015), future research considering P acquisition from myo-inositol hexakisphosphate 432 433 should emphasize the mobilization of *myo*-inositol hexakisphosphate from the soil solid phase by root exudates, mainly di- and tricarboxylic acids, which may increase the solubility of myo-434 inositol hexakisphosphate. 435

Although *myo*-inositol hexakisphosphate represented on average only 12% of the total 436 P extracted with NaOH-EDTA, the concentration decreased by 50% under the ruzigrass 437 438 treatment when compared with the fallow in presence of P fertilization. This comparison is important, since fallow during off-season is still widely used in large areas in Brazil (Simon et 439 al. 2017). Therefore, the P fertilized soybean-ruzigrass crop rotation, as recommended for 440 441 soybean cultivation, can be an important management practice to induce the cycling of myoinositol hexakisphosphate, considered the most recalcitrant Po form in the soil. This cycling of 442 myo-inositol hexakisphosphate may be even more effective in soils receiving organic 443

amendments, such manures, which has high *myo*-inositol hexakisphosphate concentration
(Gatiboni et al. 2005). More studies are needed to evaluate the community of microorganisms
and phytase activity in soil relating the expression of genes that codifies this enzyme (Lagos et
al. 2016; Luo et al. 2017; Margenot et al. 2017), as well as the factors that favor the
solubilization of *myo*-inositol hexakisphosphate and the development of microorganisms
harboring phytases, such as the exudation of organic acid anions by ruzigrass roots.

450

451 **Conclusion**

452 Long-term soybean-ruzigrass crop rotation increases SOM content. However, the increase of SOM has no correlation with Po and Pphy-lab, including myo-inositol hexakisphosphate 453 454 accumulation in the soil. The soil *myo*-inositol hexakisphosphate concentration is reduced by 455 growing ruzigrass as a cover crop in the soybean off-season, compared with fallow in the presence of P fertilizer applications, accepting the hypothesis that the degradation of inositol 456 phosphates is increased by growing ruzigrass in soil receiving P applications. This is the 457 458 opposite of what is usually observed: an increase of *myo*-inositol hexakisphosphate concentration when P fertilizers are applied. The P fertilizer application results in a great 459 increase of ruzigrass biomass, which likely may have caused a higher exudation of organic 460 acids and a consequent higher mobilization of recalcitrant P forms. The concentration of 461 unidentified inositol phosphates was higher after ruzigrass than fallow. These unidentified 462 463 inositol phosphates may be products of degradation of *myo*-inositol hexakisphosphates, which is also an evidence of the effect of ruzigrass in stimulating the degradation of soil *myo*-inositol 464 hexakisphosphates. 465

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