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Abstract

Transparent barrier films based on vacuum deposited aluminium oxide (AIO_x) layers are continuing to create large interest in the market with regards to their use as food and healthcare packaging materials. Nevertheless, their postmetalliser conversion to the final packaging material still presents challenges to current AIO_x producers and the wider converting industry. In this work, AIO_x coated PET films have been converted in long duration industrial-scale trials via topcoating, printing, lamination and finally pouch making. Throughout this process, each conversion step has been investigated for its effects on the barrier performance. It was found that the printing processes, especially, induce significant damage to the ceramic barrier layer. However, by the use of a protective topcoat prior to any conversion step, the barrier properties of the AIO_x coated film were preserved, or could even be significantly enhanced, depending on the topcoat material. Furthermore, for a barrier topcoat, remarkable stretch- and flex-durability properties were achieved in the final laminate.

Aluminum oxide; transparent inorganic barrier layers; conversion; printing; lamination
Peter Kelly
Manchester Metropolitan University
Carolin Struller, Nicholas Copeland, Peter Kelly
Charles Bishop, Hazel Assender

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Research Data Related to this Submission

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Dear Editors,

We would like to submit our paper entitled "Conversion of aluminium oxide coated films for food packaging applications – From a single layer material to the finished pouch", which describes the impact of standard converting processes on the barrier properties of AlOx-based transparent packaging material. We can confirm that this paper isn't under consideration with any other journal.

'Declarations of interest: none'

Dear Authors,

The manuscript is well written and has useful information. It can be accepted for publication is the journal after revisions are made.

Please see below:

1) line 47/48/147: PVdC (polyvinylidene chloride), EVOH, HFFS etc. should be written as polyvinylidene chloride (PVdC), please be consistent is formatting with the rest of the manuscript.

Response - corrected

2) Please define the acronym PET before first use.

Response - corrected

3) line 142: Rohm and Hass Italia Sri (Italy)... no need to be bold.

Response - corrected

Thanks

-Reviewer 2

- A well-presented summary of the effects on barrier performance of AlOx/PET films of the various film conversion properties, and then an investigation of the benefits induced by two 'protection layers'. The absence of any information on the nature of these protective layer coatings and the paucity of information on their deposition process means that it is difficult to put any 'why' on the performance improvements observed, but this report does serve to illustrate what can be achieved.

I would recommend a rewording in lines 265-270 which seems unclear in it logic - the manuscript makes reference to a critical thickness of the adhesive that depends on the thickness of barrier layer and therefore concludes that the barrier performance of the adhesive dominates that of the laminated polymer. It is not clear why a critical thickness that depends on the barrier material should be critical in dictating the relative importance of the adhesive and laminated polymer.

Response – text amended:

The thickness of the adhesive layer was between 2.5 and 3 μ m (determined via light microscopy of microtome cross sections of the laminate). This means that the adhesive thickness is above the so-called critical thickness, which depends on the defect size distribution of the inorganic barrier layer and typically is less than 2.5 μ m for the type of industrially produced barrier layers discussed here (Langowski, 2002). According to Langowski (Langowski, 2008), if the thickness of the polymer layer directly adjacent to the inorganic barrier layer is thicker than the critical thickness, then this layer along with the inorganic barrier layer is dictating the barrier performance of the system. For the work

discussed here this means that in the final laminate, the barrier properties of the polyurethane adhesive (significantly better oxygen barrier properties compared to polypropylene) dominate over those of the material laminated on top (i.e. the CPP film), which is of importance for the above discussion on barrier enhancement due to adhesive lamination.

-Reviewer 3

- Well done investigation work. Very interesting from a practical point of view.

Conversion of aluminium oxide coated films for food packaging applications -

From a single layer material to the finished pouch

Carolin Struller^{a, b}, Peter Kelly^b, Nicholas Copeland^a

Highlights

- Full industrial conversion including topcoating, printing, lamination and pouching
- Most conversion damage induced to ceramic AlO_x layer during printing process
- Successful industrial conversion of AlO_x PET via use of protective offline topcoat
- Outstanding flex and stretch durability when protective barrier topcoat is used

'Declarations of interest: none'



- Conversion of aluminium oxide coated films for food packaging applications From a single
 layer material to a complete pouch
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10 Abstract

11 Transparent barrier films based on vacuum deposited aluminium oxide (AlO_x) layers are

- 12 continuing to create large interest in the market with regards to their use as food and
- 13 healthcare packaging materials. Nevertheless, their post-metalliser conversion to the final
- 14 packaging material still presents challenges to current AlO_x producers and the wider
- 15 converting industry. In this work, AlO_x coated polyethylene terephthalate (PET) films have
- 16 been converted in long duration industrial-scale trials via topcoating, printing, lamination and
- 17 finally pouch making. Throughout this process, each conversion step has been investigated
- 18 for its effects on the barrier performance. It was found that the printing processes, especially,
- 19 induce significant damage to the ceramic barrier layer. However, by the use of a protective
- topcoat prior to any conversion step, the barrier properties of the AlO_x coated film were
- 21 preserved, or could even be significantly enhanced, depending on the topcoat material.
- 22 Furthermore, for a barrier topcoat, remarkable stretch- and flex-durability properties were
- 23 achieved in the final laminate.
- 24 Keywords: aluminium oxide; transparent inorganic barrier layers; conversion; printing;
- 25 lamination
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- 29

30 1 Introduction

Inorganic transparent barrier layers such as aluminium oxide (AlO_x) or silicon oxide (SiO_x) 31 are still in demand for clear barrier packaging materials, with applications ranging from food 32 stuffs, which have rather moderate barrier requirements, to electronic products such as 33 displays, where ultra-high barrier levels are essential. A number of different techniques are 34 35 available to deposit such inorganic layers including atomic layer deposition (ALD), plasma 36 enhanced chemical vapour deposition (PECVD), sputtering and thermal evaporation 37 processes (electron beam and boat evaporation) (Groner, George, McLean, & Carcia, 2006; Kelly, 1994; Schiller, Neumann, Morgner, & Schiller, 1994). Whilst sputtering and ALD are 38 mainly of importance for high-end applications, PECVD and thermal evaporation techniques 39 are the processes of interest with regards to the cost-sensitive food packaging market. 40 41 Reactive evaporation using resistively heated boats represents an especially promising 42 candidate with great market potential due to the relatively low associated capital investment, 43 the use of inexpensive raw materials and the high process speeds that can be achieved, 44 without disadvantageous effects to the barrier performance, in comparison to other deposition 45 techniques.

46 The market for transparent oxide coated films has an estimated compound annual growth rate of around 7 to 8 %, compared to only 2 to 3 % for polyvinylidene chloride (PVdC) coated 47 films and 4 to 6 % for ethylene vinyl alcohol (EVOH) copolymer co-extruded films (Platt, 48 2016); both of which are polymer based transparent barrier films that can potentially be 49 50 replaced by oxide coated films. Nevertheless, the market volume for these oxide based barrier films is currently a lot smaller, compared to the polymer based options of PVdC and EVOH. 51 52 To date, the clear barrier flexible packaging market for AlO_x and SiO_x transparent barrier 53 films is still dominated by Japanese material producers, such as Toppan, Toray, DNP, Mitsubishi and others with well-known film grades such as GL Film, Barrialox, IB-Film or 54 TECHBARRIER (DNP America, 2017; Mitsubishi Chemical, 2018; Toppan USA, 2016; 55 Toray Advanced Film Co., 2018), although Camvac's Camclear (Camvac Limited, 2018) and 56 Amcor's AmLite (formerly Ceramis) (Amcor Limited, 2018) have also become well-57 58 established products. The Japanese producers predominantly apply electron beam evaporation or PECVD techniques for the deposition of the barrier layers (Naegli & Lohwasser, 2001), 59 which entail substantially higher production and investment costs and more technical 60 complexity, including potential reliability issues, compared to reactive boat evaporation. 61 62 Furthermore, their products are generally topcoated with a material that not only protects the

63 inorganic AlO_x or SiO_x barrier layer through handling and conversion, but also significantly enhances its barrier performance. This, in addition to the retortability of many of these 64 materials, makes them very high performance products, which comes at a certain cost level 65 and prevents real volume growth for cost-sensitive volume market applications. The reactive 66 67 boat evaporation process has been developed over the last few decades and is now wellestablished (Günther et al., 2009; Kelly, 1993, 1994; Schiller et al., 1994; Struller, Kelly, 68 Copeland, & Liauw, 2012). Its vast economic potential for low cost AlOx production leads to 69 new players continuously pushing to enter the attractive transparent barrier market and trying 70 71 to obtain market share from the more advanced Japanese players. However, unlike the 72 Japanese products, most of these 'new' products do not have a protective topcoat and their 73 conversion is challenging with inevitable barrier deterioration.

In order to obtain a fully viable commercial AlO_x product, the successful conversion of the 74 coated film also has to be taken into account. AlO_x coated polymer films will always need to 75 76 undergo further conversion steps in order to obtain the final packaging structure. These 77 conversion steps will include all or most of the following downstream processes; slitting, 78 coating, printing and lamination. Since each of these steps can potentially have a detrimental 79 impact on barrier performance, it is essential that barrier loss upon conversion is avoided or at 80 least minimised to achieve the barrier requirements of the target application. In this study, it 81 has been found that the printing step, especially rotogravure printing, is the most damaging conversion process with regards to the AlO_x barrier performance, which has also been 82 identified by other researchers (Jahromi, 2016). This paper is a unique study on optimising 83 and facilitating the conversion process of AlO_x coated films in order to meet market 84 85 requirements for food packaging applications (with the aim of dry food packaging applications such as snack foods, confectionary, biscuits or cereal bars). The focus, therefore, 86 87 is on developing a suitable offline topcoat solution, which is not only capable of protecting the AlO_x barrier layer through all conversion steps (including pouch making), but also offers 88 89 additional barrier functionality. Once this topcoat is successfully applied using an optimised/modified platform to prevent barrier loss upon web conveying and coating 90 91 application, the topcoated AlO_x polyethylene terephthalate (PET) films can be converted on standard industrial equipment, without the need for changes and without any subsequent 92 barrier deterioration. Trials have been conducted using AlO_x coated PET film and each 93 94 process step was characterised with regards to its post-processing barrier performance.

Additionally, flex-durability (Gelbo-Flex) testing as well as barrier retention upon elongation
investigations were carried out.

97 2 Material and methods

98 2.1 Coating/Conversion processes and platforms

99 Standard low-cost commodity grade PET films (12 µm film thickness; 1250 mm width; 100 corona treated by the film manufacturer) were chosen for this investigation (note: especially in the pre-assessment phase, PET films from a variety of film supplies were used). AIO_x 101 coating was conducted via reactive thermal evaporation using a Bobst Manchester Ltd 102 (Heywood, United Kingdom) K5000 vacuum metalliser with an optional AlO_x coating 103 104 system. The vacuum coater has a deposition source consisting of resistively heated evaporation boats (standard intermetallic composite) onto which aluminium is continuously 105 106 fed in the form of a wire. Oxygen is introduced into the aluminium vapour stream in order to produce a transparent aluminium oxide coating and an optical monitor beam and closed loop 107 108 control system is used to achieve consistent optical properties of the coated film across the web width and length. The pressure during aluminium oxide deposition is of the order of 0.05 109 Pa. AlO_x deposition was carried out at a web speed of 600 m/min, whilst also applying 110 optimised in-line plasma pre-treatment using a medium frequency plasma source with 111 magnetically enhanced water cooled electrodes. To achieve enhanced convertibility of the 112 AlO_x coated starting material, the original AlO_x process (Struller, Kelly, & Copeland, 2014) 113 was further optimised in order to facilitate the handling of the AlO_x coated film via increasing 114 115 its flexibility. Thereby, the thickness of the coating layer was reduced from an average value of 10 nm to a thickness in the range of 7 to 8 nm. For the final conversion work (including 116 topcoating, printing and lamination), 32 km reels were AlO_x coated. 117

118 Topcoating of the produced AIO_x PET reels was performed on an AIO_x optimised Bobst

119 CL 850D coater/laminator at Bobst Italia SpA (San Giorgio Monferrato, Italy) using a

120 forward gravure coating system at web speeds up to 250 m/min. The two coatings chosen are

121 water-based and were supplied by the project partner Michelman SARL (Windhof,

122 Luxembourg). These coatings are: a protective topcoat (offering protection through

123 conversion only) and a barrier topcoat (offering protection through conversion and additional

barrier functionality). Both coatings are compliant with relevant food contact legislations.

125 The rotogravure printing was also conducted at Bobst Italia using a standard Bobst

- 126 RS 4003MP 8 colour rotogravure printing press at web speeds of up to 400 m/min. For the
- 127 final conversion work (using a topcoat prior to printing), a solvent-based high performance
- 128 commercially available ink system was used, which was supplied by the project partner Flint
- 129 Group Italia SpA (Cinisello Balsamo, Italy).

Flexo printing was carried out on a standard Bobst 20SIX 8 colour CI (central impression) 130 flexo printing press at Bobst Bielefeld GmbH (Bielefeld, Germany). For the final conversion 131 work, printing was conducted at web speeds of 200 m/min using the flexo version of the 132 133 solvent-based high performance ink system used in the gravure printing trials, supplied by Flint Group Germany GmbH (Willstätt, Germany). Printing (flexo and gravure) was 134 135 conducted as 'reverse' printing with the white colour printed last, since the printed material itself will be on the outside of the packaging structure with the ink embedded within the 136 137 laminate (i.e. one looks through the PET film at the print or, in areas of no print, at the

138 packaged food stuffs).

Lamination was again carried out at Bobst Italia SpA using a solvent-based adhesive on a 139 Bobst CL 850D coater/laminator. A commercially available high performance, two 140 component polyurethane adhesive (supplied by the project partner Dow/Rohm and Haas 141 Italia Srl (Mozzate, Italy) was used and applied via a flexo trolley coating application 142 system at a speed of 150 m/min. For the lamination in the pre-assessment stage, a 40 µm 143 thick corona treated cast polypropylene (CPP) sealant film was used; whilst for the full 144 conversion work a 32 µm corona treated linear low-density polyethylene (LLDPE) sealant 145 film was applied as secondary material. 146

All further conversion steps (slitting, pouch making via HFFS (horizontal form fill & seal)
and pouch filling) were arranged and managed by collaborator Printpack Inc. (Atlanta, GA,
United States) using their facilities, contacts and customer base.

150 2.2 Analytical methods

151 Barrier properties in terms of oxygen and water vapour transmission rates (OTR/WVTR)

were analysed in accordance with ASTM F 1927 and ASTM F 1249/ISO 15106-3 using a

153 Mocon Oxtran 2/20 and Systech Illinois 8001 for oxygen permeation and a Mocon

- 154 Permatran-W 3/33 and Systech Illinois 7001 for water vapour permeation. Test conditions
- were 23 °C and 50 % relative humidity (RH) for OTR measurements and 37.8 °C and a

gradient of 90 % RH for WVTR measurements. Barrier properties were typically measured at
several points across the 1250 mm web width and average values are given.

158 A Zeiss Supra 40VP field emission gun scanning electron microscope (FEG SEM) was used

to acquire SEM images. All samples were examined using a low acceleration voltage to avoid

the need for coating with a conductive layer (which could potentially mask important surface

detail). Cross-sections were cut on a Leica RM2125 microtome prior to SEM analysis.

162 The flex durability or Gelbo-Flex (ASTM F392) of the laminated AlO_x coated films was

163 investigated using a Gelbo-Flex tester model 5000 manufactured by United States Testing

164 Co., Inc. For this investigation 1, 5 and 20 flexing cycles were performed and the barrier

165 performance was determined successively.

166 For the evaluation of the barrier retention on elongation behaviour, the films were stretched

to a pre-defined strain (between 0.5 and 5 %, stretching in the machine direction) by a tensile

testing unit (Hounsfield H10KS with QMat 5.52 software) and subsequently the barrier

properties were measured (test for stretch durability/resistance (Felts, 1993), however

samples are analysed in the relaxed state).

171

172 **3** Results and discussion

173 **3.1** Printing and lamination pre-assessment

174 **3.1.1** Flexographic and gravure printing

175 During the initial assessment, printing conversion processes were analysed for their impact 176 on the barrier performance of AlO_x coated PET film (i.e. when the non-protective topcoat is 177 present). Therefore, trials were conducted using standard 'non-optimised' conversion equipment at Bobst Italia and Bobst Bielefeld. During each interaction, several trials were 178 conducted on the printing presses, in order to investigate the variability that can be obtained 179 during direct printing onto the sensitive AlO_x PET material. Additionally, trials were 180 conducted on different machines of the same model, i.e. the RS 4003MP as well as the 20SIX 181 presses were investigated on two different occasions, whereby the machines were of the same 182 model, but not 'physically' the same machines. This was also part of the pre-assessment to 183 show the range of barrier deterioration that can occur on the same machine type. No special 184 care was taken in regards to the sensitiveness of the AlO_x PET film, but the web was printed 185

using parameters and run conditions typical for 12 µm uncoated PET film. Additionally, trials
were also carried out at a customer site in China. The latter was conducted in order to acquire
true field data of a trial that has not been executed under the supervision and control of the
authors/team at Bobst as well as using non-Bobst and older equipment (Chinese supplied 7
colour gravure printing press; make, model and age of machinery unknown to the authors;
solvent based inks used from Chinese supplier).

For each of the assessments, 4 km long rolls of AlO_x coated PET film were produced and 192 193 used to carry out the conversion trials and the results are summarised in Table 1. Barrier numbers before conversion are given as a range which is typical for the PET films coated in 194 this work. For the post printing barrier performance, a large spread of data was obtained in 195 the different assessments of the same machine types as well as the various trials run on the 196 same machine. Due to this variability of transmission rates, the post printing ranges obtained 197 198 are given for OTR and WVTR, rather than an average value. As can be seen, after printing (gravure and flexo) the barrier is deteriorated in all cases, although to a larger degree for the 199 gravure printing. This is due to the longer web path and larger amount of web handling 200 involved in the gravure process (printing units inline), versus CI flexo printing (printing units 201 arranged around central impression drum). The fact that web handling can have a detrimental 202 impact on the barrier performance of ceramic inorganic barrier layers, such as AlO_x coatings 203 deposited onto a polymer web, has also been shown by a group of researchers (Lahtinen, 204 Lahti, Johansson, Seppänen, & Cameron, 2014) who investigated the effect of post deposition 205 film handling on atomic layer deposited aluminium oxide coatings on biaxially oriented 206 207 polypropylene film. Also, the barrier data obtained from a gravure printing field trial at a customer site falls within the spread of post-printing data obtained on the Bobst gravure 208 209 platform. Furthermore, Jahromi (Jahromi, 2016), has reported comparable barrier values for printed unprotected AlO_x PET (OTR 6 to 12 cm³/(m² d) and WVTR 2 to 7 g/(m² d)), although 210 211 these samples were flexo printed, which in our case has been found to be significantly less destructive than gravure printing. 212

In order to investigate the nature and origin of the damage to the AlO_x layer, further analysis was conducted. In principle, the damage to the AlO_x layer (and hence barrier deterioration) could be caused by either chemical, thermal or mechanical stress to the film. However, AlO_x as a ceramic material should be chemically inert and also temperature resistant, which was

217 investigated and proven via laboratory assessment trials. Therefore, mechanical damage to

the AlO_x layer appeared to be the most likely cause of the barrier deterioration. In this case, 218 damage to the AlO_x coating could be due to stretching the film and hence cracking of the 219 AlO_x layer or mechanically scratching/scuffing the brittle AlO_x due to contact with rollers 220 221 (driven or idler) or the gravure printing head in the web handling mechanism. With regards to the stretch durability of AlO_x coated PET film, the authors have previously shown (Struller et 222 223 al., 2012) that OTR can withstand up to 3 % elongation, whilst water vapour barrier is retained up to 1 % elongation before deterioration. Based on the web tensions used during the 224 printing process, this is very unlikely to be the cause of the barrier decline seen after printing 225 and, hence, samples of the printing trials were analysed via scanning electron microscopy in 226 227 order to discover any visible kind of mechanical damage in the AlO_x layer. An example SEM 228 micrograph revealing the damaged AlO_x layer is show in Figure 1. The image on the left side is at lower magnification in order to show the location of the damage on a larger scale. 229 Obviously, damage to the AlO_x coating can only be detected in the unprinted area, since it 230 will be covered up by the ink in the printed areas. On the right hand side, a higher 231 magnification image of the scratch marked in the low magnification image can be seen. It is 232 clearly noticeable that the AlOx layer is damaged. Scratches like this were found to cover all 233 post-printing samples (exhibiting barrier deterioration) that were investigated and their 234 typical orientation was the machine direction (i.e. movement direction of the film during 235 processing). Consequently, it can be concluded that the barrier loss induced during printing is 236 caused by mechanical damage of the AlO_x coating during web handling on the printing press. 237

238 **3.1.2 Lamination**

In addition to assessment of the two printing processes, the lamination process was also 239 240 investigated for non-protected AlO_x coated PET film using a standard 'non-optimised' laminator. A range of lamination trials was conducted, whereby the adhesive was applied 241 directly onto the AlO_x layer and a CPP film was used as the secondary web. Unlike the 242 printing assessment, the post-lamination barrier performance was comparable to pre-243 lamination properties and, hence, average values and associated standard deviations are given 244 in Table 1 for the AlO_x PET as well as the laminated structure (note, the average given has 245 been obtained from several lamination trials). Overall, no significant barrier deterioration is 246 247 seen when non-protected AlO_x PET film is laminated via direct application of the adhesive onto the ceramic barrier layer using standard industrial equipment. Nevertheless, an oxygen 248 249 barrier improvement effect is frequently observed by lamination of vacuum coated (i.e.

metallised) films when using polyurethane based adhesives, based on the good oxygen barrier 250 properties of the adhesive itself (Mueller, Schoenweitz, & Langowski, 2012). It should be 251 noted here that the aluminium layer of a metallised film is less prone to damage during 252 conversion due to its ductility, compared to an AlO_x layer which is ceramic and hence brittle. 253 Langowski (Langowski, 2008) states that for a multilayer system with an inorganic barrier 254 layer embedded between two polymer layers 'when the two polymer layers have the same or 255 very similar permeation coefficients, the permeability (i.e. transmission rate) is halved 256 compared to a system comprising a single polymer layer and the inorganic layer'. As no 257 degree of barrier improvement was seen when laminating the AlO_x coated PET film, it may 258 well be that damage of the AlO_x layer took place during the conversion process of 259 lamination. However, as we obtain a post-lamination barrier level comparable to the pre-260 lamination level, it is assumed that the barrier performance loss is recovered due to the barrier 261 properties of the adhesive, as well as by a possible infiltration of the adhesive into the 262 defects/damage in the AlOx layer (Miesbauer, Schmidt, & Langowski, 2008). The thickness 263 of the adhesive layer was between 2.5 and 3 µm (determined via light microscopy of 264 microtome cross sections of the laminate). This means that the adhesive thickness is above 265 266 the so-called critical thickness, which depends on the defect size distribution of the inorganic barrier layer and typically is less than 2.5 µm for the type of industrially produced barrier 267 layers discussed here (Langowski, 2002). According to Langowski (Langowski, 2008), if the 268 thickness of the polymer layer directly adjacent to the inorganic barrier layer is thicker than 269 270 the critical thickness, then this layer along with the inorganic barrier layer dictates the barrier performance of the system. For the work discussed here this means that in the final laminate, 271 272 the barrier properties of the polyurethane adhesive (significantly better oxygen barrier properties compared to polypropylene) dominate over those of the material laminated on top 273 274 (i.e. the CPP film), which is of importance for the above discussion on barrier enhancement due to adhesive lamination. 275

276

Another important conversion step with the potential to damage is the slitting process. This has been found not to impair AIO_x barrier performance when conducted in a controlled manner on standard slitting equipment and has been discussed by the authors in more detail elsewhere (Struller et al., 2015).

281 **3.2** Topcoat development

After the initial assessment of conversion processes and their impact on the barrier 282 performance of AlO_x coated films, the approach taken in this research work was to apply a 283 topcoat prior to any damaging conversion (such as printing). This topcoat serves the purpose 284 of protecting the AlO_x layer during conversion, thus avoiding damage/barrier deterioration 285 and enabling the use of standard industrial equipment for downstream processing of AlO_x 286 287 coated films. In this manner, any conversion after topcoating can be decoupled. A topcoat can be applied in two ways, either inline in the vacuum chamber or offline via an atmospheric 288 289 pressure 'wet' coating process. The former can, for example, be achieved by the use of an acrylate-based topcoat, flash evaporated (Affinito, Eufinger, Gross, Graff, & Martin, 1997; 290 Affinito et al., 1996; Shipman, 2016) or flexo printed in vacuum (Ferrari, 2016), or also via a 291 melamine based topcoat (Jahromi, 2011). Acrylate coatings have been widely investigated, 292 including by the authors of this paper (Struller, Kelly, Copeland, et al., 2014). Nevertheless, 293 294 the guarantee of fully cured acrylate layers without any residual monomer (as would be required for food safety reasons) has been the main hurdle and, despite recent advances, 295 acrylate topcoated films have not yet found a place in the market for food packaging 296 materials, but are predominantly used for industrial applications. Melamine based topcoats on 297 298 the other hand, also suffer from low market acceptance; here, though, mainly based on the negative connotation associated with melamine. Hence, in this work, the focus lies on 299 300 investigating offline applied wet topcoats, which do not exhibit any of the above drawbacks.

301 The initial work started here with material screening trials, in order to find a suitable coating material (polymer) and assess the material properties required. These investigations were 302 303 conducted on a laboratory-scale as well as pilot-scale prior to moving onto industrial-scale, and won't be discussed in further detail here. An effective laboratory-scale conversion 304 simulation technique was developed in order to subject the topcoated AlO_x PET samples 305 (from laboratory, pilot and industrial coating work) to the levels of damage and stress that 306 they would generally endure during conversion processes such as gravure printing. Once this 307 308 proprietary technique was refined to match the effects of conversion, this method could then 309 be used to examine and test topcoated samples for the protective properties of the applied 310 topcoat without the need of having to conduct actual printing trials. When the step was taken to move onto industrial-scale, the right coating application technique had to be selected (a 311 forward gravure coating system) and adjusted and the web handling system also had to be 312 adapted in order to optimise the platform for the coating of the AlO_x PET material. This is in 313 order to avoid any damage of the sensitive AlO_x layer on the offline coater due to the web 314

- handling before the coating head as well as during the coating application process.
- 316 Nevertheless, these platform and coating technique optimisations are of confidential nature to
- Bobst and can't be discussed in further detail.
- 318 Results of some of the initial investigations conducted are summarised in Table 2. As can be
- seen, after the conversion simulation test on unprotected AlO_x PET, OTR and WVTR are
- drastically increased with the OTR rising from a value of around $1 \text{ cm}^3/(\text{m}^2 \text{ d})$ to more than
- $10 \text{ cm}^3/(\text{m}^2 \text{ d})$ and WVTR from values of less than $1 \text{ g}/(\text{m}^2 \text{ d})$ to an average value of 5.62
- $g/(m^2 d)$. These values fall within the range of results obtained during print tests (refer to
- Table 1) of non-topcoated AlO_x PET, which confirms the suitability of the conversion
- 324 simulation test developed.

325 Also displayed in Table 2 are the barrier performances obtained before and after conversion simulation tests for the two topcoats, which were selected for the conversion/pouch making 326 trials (see next section). As can be seen, the protective topcoat only offers a small OTR 327 improvement (from OTR values > 1 cm³/(m² d) to values consistently < 1 cm³/(m² d)), whilst 328 329 the barrier topcoat increases the oxygen barrier properties remarkably. Furthermore, the barrier topcoat also offers some water barrier improvement, from a typical value of around 330 331 0.7 g/(m^2 d) to values around 0.4 g/(m^2 d) . The barrier improvement obtained by applying the topcoat can be attributed to two aspects; a potential pore-filling effect (Affinito & Hilliard, 332 2004) (something the authors have previously reported for acrylate topcoats (Struller, Kelly, 333 Copeland, et al., 2014)) and the effect of the permeability/barrier properties of the topcoat 334 335 material itself (in comparison to the PET substrate), which is now adjacent to the inorganic barrier layer (Langowski, 2008). With regards to the conversion simulation test, it can be 336 seen that the barrier is unchanged or only marginally increased after conducting this test. The 337 latter is the case for the barrier topcoat (refer to oxygen barrier) and indicates that the coat 338 weight needs to be slightly increased. As will be shown later, the results obtained here are 339 very reproducible and are consistently achieved in the subsequent long duration industrial 340 341 trials.

342 **3.3 Conversion trials**

After the initial investigations presented in the previous section, as well as successful shortrun industrial topcoating trials, the next step was to conduct long-duration conversion trials, whereby the full downstream conversion chain was investigated, from topcoating, via printing, lamination and slitting to the final packaging structure (such as pouch/bag or lid)

including the filling of this packaging material with actual food stuffs. Being able to directly 347 print onto the topcoated AlO_x PET material and hence achieve a two-ply laminate structure 348 instead of a three-ply structure helps to fulfil the demand for cost reduction and down 349 gauging. Appropriate inks were selected based on printability and ink adhesion studies 350 carried out by the ink manufacturer. Furthermore, compatibility tests were conducted with the 351 inks selected in order to avoid any negative effects on the barrier performance due to a 352 potential incompatibility between topcoats and inks. Two topcoats were used in the 353 conversion work: 354

- Protective topcoat: offering protection through conversion only

- Barrier topcoat: offering protection and additional barrier functionality

357 The final laminate structure produced during the conversion trials is shown in an SEM image 358 (microtome cross section) in Figure 2. The picture clearly shows the individual layers added 359 in each conversion step: AlO_x, topcoat, inks & adhesive, apart from the LLDPE sealant web which would be on top of the adhesive. For the printing process, one can also distinguish 360 361 between the different coloured inks, based on the larger pigments (titanium dioxide) in the white ink. Finally, even the thin AlO_x layer can be made out as a lighter coloured line. This is 362 thought to be caused by electrons backscattered from the aluminium (higher atomic number 363 364 of aluminium compared to atoms in the surrounding PET film and polymer topcoat) in the AlO_x barrier layer. 365

The oxygen and water barrier was assessed after each individual conversion process step and 366 results are summarised in Table 3 for the gravure printing and in Table 4 for flexo printing 367 (note: not in all cases, the full conversion of the produced material was assessed). These 368 results are very much in agreement with the barrier performance obtained during the topcoat 369 development stages (see Table 2). If one compares the barrier properties measured after each 370 individual conversion step, it is clear that no oxygen or water vapour barrier deterioration 371 took place through the conversion exercise. This is the case for both the protective and the 372 373 barrier topcoat. Some fluctuations in the barrier data are noticeable, which are, however, put down to typical fluctuations within the large film reels converted and the samples being taken 374 at different positions (lengthwise). Although not all conversion steps have been conducted 375 for the flexo printing, it is clear that also in this case, no further barrier deterioration can be 376 expected, especially since flexo printing has been found to be less aggressive and damaging 377 compared to gravure printing. 378

379 **3.4** Gelbo-Flex testing and barrier retention on elongation

Laminated samples of the conversion trials with gravure printing have been further 380 investigated for their flex-durability (Gelbo-Flex). During this test, the laminate is repeatedly 381 382 twisted and crushed, which serves the purpose of simulating the strain that the laminated material may be subjected during further conversion (i.e. folding and forming into packaging 383 structures) and whilst handled in typical transport, storage and retail environments as finished 384 packages of food products. Results of this investigation are shown in Figure 3. As can be 385 seen, the laminate with the barrier topcoat reveals very good barrier performance. The OTR is 386 387 predominantly unaffected, even after 20 Gelbo-Flex cycles, whilst WVTR is slightly 388 increased after 20 cycles from values around 0.4 g/(m^2 d) for the unflexed laminate to values 389 around 0.7 g/(m^2 d). This flex-durability performance is comparable to or even outperforms data published by some of the large and more established transparent barrier film producers 390 391 (DNP America, 2017; Toppan USA, 2016). In the case of the protective topcoat, the effect of the Gelbo-Flex test is more pronounced. OTR increases from $0.77 \text{ cm}^3/(\text{m}^2 \text{ d})$ for the 392 393 unflexed sample to an average of $3.87 \text{ cm}^3/(\text{m}^2 \text{ d})$ after 20 cycles, whilst WVTR rises from 0.68 g/(m^2 d) to 1.44 g/(m² d). This performance is, however, still remarkable in regards to 394 395 the destructiveness of this test. Finally, it should also be noted here that, as stated previously (Struller, Kelly, Copeland, & Read, 2013), the specific effects of the Gelbo-Flex test are 396 influenced by many factors, such as the type/chemistry of the inorganic barrier layer (Chiba, 397 Mikami, Sakamoto, & Tsuchiya, 2006) and its thickness, the deposition process (Komada, 398 Oboshi, & Ichimura, 2000), the substrate used and the characteristics of the lamination 399 process (Abedin & Jopko, 2013) (type of adhesive, secondary material, duplex/triplex 400 laminate etc.). 401

The laminates were additionally tested for their barrier retention on elongation behaviour, 402 whereby the samples were subjected to uniaxial deformation in the machine direction. This 403 test has been previously used in our research in order to assess how the AlO_x coated film can 404 withstand downstream processing in terms of web tension (Struller et al., 2012). In the case 405 of conducting this test with laminates, the objective is similar, as also the laminate is further 406 converted to a packaging structure (pouch) via HFFS. On these types of machinery, the 407 laminate may be subjected to stretching due to high web tensions and it is important to 408 exclude any barrier deterioration based on this process step. 409

The barrier results from this test, presented in Figure 4, indicate that oxygen barrier remains 410 unchanged for both topcoats up to 2 % elongation. After this, the OTR rises for the barrier 411 topcoat to an average value of only $0.65 \text{ cm}^3/(\text{m}^2 \text{ d})$ at 5 % elongation. As observed for the 412 flex-tests, this again shows the durability of the barrier topcoat laminate when subjected to 413 different forms of tensile stress and strain. In the case of the protective topcoat. However, the 414 increase in OTR for 3 % elongation onwards is drastic. When investigating the water barrier 415 retention for both topcoats, it was found that water barrier properties are maintained up to 416 1.5 % elongation. For elongation values of 2 % and onwards, the WVTR increases; although, 417 418 the barrier topcoat performs significantly better than the protective topcoat. Looking back at the barrier performance obtained for the structure with the barrier topcoat post pouch-making 419 (see Table 3), one can see that neither OTR nor WVTR is deteriorated after this conversion 420 step, hence indicating that no stretching beyond 1.5 % elongation occurs due to the web 421 tensions on the pouch making equipment. 422

423 4 Summary and Conclusions

It has been shown with the work presented here that AlO_x coated PET films can be converted 424 without barrier deterioration using standard equipment once a protective offline topcoat has 425 426 been applied. Extensive research has been conducted in assessing the various conversion processes for their impact on the conversion of AlO_x coated PET film as well as developing 427 428 and characterising optimised topcoats. Two different topcoat variations have been established and investigated, with one topcoat giving protection through conversion only, and the other 429 430 topcoat offering additional and significant barrier enhancement. Long duration conversion trials, including printing (flexo and gravure), lamination, slitting and pouch making/filling 431 were successfully carried out with both topcoats. Furthermore, tests on flex-durability and 432 barrier retention on elongation also showed that the barrier topcoat laminate especially has 433 remarkably good properties when subjected to this kind of repetitive or uniaxial strain. 434

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4465References

- 447 Abedin, N., & Jopko, L. (2013, April). How good is your barrier packaging in the real
- world?. Conference presentation at the TAPPI PLACE Flexible packaging symposium: New technologies, Orlando, FL.
- 450 Affinito, J. D., Eufinger, S., Gross, M. E., Graff, G. L., & Martin, P. M. (1997).
- 451 PML/oxide/PML barrier layer performance differences arising from use of UV or electron
- beam polymerization of the PML layers. *Thin Solid Films*, *308–309*, 19-25.
- 453 Affinito, J. D., Gross, M. E., Coronado, C. A., Graff, G. L., Greenwell, I. N., & Martin, P. M.
- (1996). A new method for fabricating transparent barrier layers. *Thin Solid Films, 290–291*,
 63-67.
- Affinito, J. D., & Hilliard, D. (2004). A new class of ultra-barrier materials. 47th Annual
 Technical Conference Proceedings, Society of Vacuum Coaters, 563-593.
- 458 Amcor Limited. AmLite Standard Innovative metal-free film for high barrier packaging.
- 459 (2018). https://www.amcor.com/product-listing/amlite-standard Accessed 31 July 2018.
- 460 Camvac Limited. Camclear[®] Transparent barrier films. (2018).
- 461 http://www.camvaclimited.com/product/Camclear Accessed 31 July 2018.
- 462 Chiba, D., Mikami, K., Sakamoto, H., & Tsuchiya, H. (2006). Heat sterilizable gas barrier

film deposited on polyamide by plasma-enhanced CVD. 49th Annual Technical Conference

- 464 Proceedings, Society of Vacuum Coaters, 613-616.
- 465 DNP America, LLC. Transparent Barrier Film IB-PET-PIR. (2017).
- 466 http://www.dnpamerica.com/uploads/2017/01/ib_pet_pir.pdf Accessed 31 July 2018.
- Felts, J. T. (1993). Transparent barrier coatings update: Flexible substrates *Journal of Plastic Film and Sheeting*, 9(2), 139-158.
- 469 Ferrari, D. (2016, October). In vacuum flexo top-coating on metallized film: An update on
- the state of the art. Conference presentation at the AIMCAL Web Coating and HandlingConference USA, Memphis, TN.
- Groner, M. D., George, S. M., McLean, R. S., & Carcia, P. F. (2006). Gas diffusion barriers
 on polymers using Al₂O₃ atomic layer deposition. *Applied Physics Letters*, *88*(5), 051907.
- 474 Günther, S., Straach, S., Schiller, N., Quiceno, A. L., Contreras, A. G., Ludwig, R., et al.
- 475 (2009). Innovative transparent barrier for packaging. 52nd Annual Technical Conference
- 476 Proceedings, Society of Vacuum Coaters, 727-729.
- 477 Jahromi, S. (2011, October). Freshure[®] coatings: Environmentally friendly transparent high
- barrier coating. Conference presentation at the AIMCAL Fall Technical Conference, Reno,NV.

- 480 Jahromi, S. (2016, June). Advances in protection of aluminum oxide using inline vacuum
- 481 deposited organic top coats. Conference presentation at the AIMCAL Web Coating and Hendling Conference Dreaden Cormany
- 482 Handling Conference Europe, Dresden, Germany.
- Kelly, R. S. A. (1993). Development of clear barrier films in Europe. *36th Annual Technical Conference Proceedings, Society of Vacuum Coaters*, 312-316.
- Kelly, R. S. A. (1994). Development of aluminium oxide clear barrier films. *37th Annual Technical Conference Proceedings, Society of Vacuum Coaters*, 144-148.
- 487 Komada, M., Oboshi, T., & Ichimura, K. (2000). Novel transparent gas barrier film prepared
- by PECVD method. *43rd Annual Technical Conference Proceedings, Society of Vacuum Coaters*, 352-356.
- 490 Lahtinen, K., Lahti, J., Johansson, P., Seppänen, T., & Cameron, D. C. (2014). Influence of
- substrate contamination, web handling, and pretreatments on the barrier performance of
 aluminum oxide atomic layer-deposited BOPP film. *Journal of Coatings Technology and*
- 493 *Research*, *11*(5), 775-784.
- 494 Langowski, H.-C. (2002). Flexible Materialien mit ultrahohen Barriereeigenschaften
- 495 (Flexible ultra high barrier materials). *Vakuum in Forschung und Praxis*, 14(5), 297-302.
- 496 Langowski, H.-C. (2008). Permeation of gases and condensable substances through
- 497 monolayer and multilayer structures. In O. G. Piringer & A. L. Baner (Eds.), *Plastic*
- *packaging Interactions with food and pharmaceuticals* (pp. 297-347). Weinheim: WILEY VCH Verlag.
- 500 Miesbauer, O., Schmidt, M., & Langowski, H.-C. (2008). Stofftransport durch
- 501 Schichtsysteme aus Polymeren und dünnen anorganischen Schichten (Mass transport through
- 502 layer systems consisting of polymers and thin inorganic coatings). *Vakuum in Forschung und*
- 503 *Praxis*, 20(6), 32-40.
- 504 Mitsubishi Chemical. SiOx Vacuum Coated High Gas Barrier Film TECHBARRIERTM.
- 505 (2018). https://www.m-chemical.co.jp/en/products/departments/mcc/food-
- 506 packaging/product/1201194_8048.html Accessed 31 July 2018.
- 507 Mueller, K., Schoenweitz, C., & Langowski, H.-C. (2012). Thin laminate films for barrier
- 508 packaging application Influence of down gauging and substrate surface properties on the 509 permeation properties. *Packaging Technology and Science*, *25*(3), 137-148.
- 510 Naegeli, H. R., & Lohwasser, W. (2001, May). Processing and Converting of Vacuum
- 511 Coated Inorganic Barrier Films. Conference presentation at the European TAPPI PLC
- 512 Conference, Barcelona, Spain.
- 513 Platt, D. (2016). *The future of high-barrier packaging films to 2021*: Smithers Pira.
- 514 Schiller, S., Neumann, M., Morgner, H., & Schiller, N. (1994). Plasma-activated high-rate
- 515 deposition of oxides on plastic films. *37th Annual Technical Conference Proceedings*,
- 516 Society of Vacuum Coaters, 203-210.
- 517 Shipman, J. T. (2016, October). In vacuo coating (IVC) Improving resistive metallised films
- 518 through in line, in vacuum, polymeric coating A summary of development and performance.
- 519 Conference presentation at the AIMCAL Web Coating and Handling Conference USA,
- 520 Memphis, TN.
- 521 Struller, C. F., Kelly, P. J., Copeland, N., & Read, S. J. (2013, October). Characterisation and
- 522 conversion of aluminum oxide coated flexible barrier films. Conference presentation at the
- 523 AIMCAL Web Coating and Handling Conference USA, Charleston, SC.

- Struller, C. F., Kelly, P. J., & Copeland, N. J. (2014). Aluminum oxide barrier coatings on 524
- polymer films for food packaging applications. Surface and Coatings Technology, 241, 130-525 137. 526
- Struller, C. F., Kelly, P. J., Copeland, N. J., & Liauw, C. M. (2012). Characterization studies 527
- of aluminum oxide barrier coatings on polymeric substrates. Journal of Vacuum Science & 528 Technology A, 30(4), 041502. 529
- Struller, C. F., Kelly, P. J., Copeland, N. J., Tobin, V., Assender, H. E., Holliday, C. W., et al. 530
- (2014). Aluminium oxide barrier films on polymeric web and their conversion for packaging 531 532 applications. Thin Solid Films, 553, 153-156.
- 533 Struller, C. F., Kelly, P. J., Copeland, N. J., Tobin, V., Assender, H. E., Holliday, C. W., et al.
- (2015). Aluminium oxide barrier layers and their conversion for packaging applications. 58th 534 Annual Technical Conference Proceedings, Society of Vacuum Coaters, 518-523.
- 535
- Toppan USA, Inc.. Transparent Barrier Film GL FILM. (2016). http://www.toppan-536 usa.com/Toppan-GL-FILM.pdf Accessed 31 July 2018. 537
- Toray Advanced Film Co.,Ltd., Barrialox[™] (Coated Type), (2018), http://www.toray-538
- taf.co.jp/en/product/pro 040.html Accessed 31 July 2018. 539
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Table and figure headings 541

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Structure	Description	OTR	WVTR
		$cm^{3}/(m^{2} d)$	$\frac{g/(m^2 d)}{g/(m^2 d)}$
PET 12 μm		≈ 120	40 - 50
PET/AlO _x	Before printing	1.0 - 1.5	0.5 - 1.0
PET/AlO _x /ink	Bobst Italia, gravure press RS 4003MP 250 m/min	2.5 - 18	1 – 12
PET/AlO _x	Before printing	1.0 - 1.5	0.5 - 1.0
PET/AlO _x /ink	Chinese customer, gravure press 200 m/min	15 - 20	8-10
PET/AlO _x	Before printing	1.0 - 1.5	0.5 - 1.0
PET/AlO _x /ink	Bobst Bielefeld, CI flexo press 20SIX 250 and 400 m/min	1.5 – 8	1-3.5
PET/AlO _x	Before lamination	1.29 ± 0.21	0.82 ± 0.11
PET/AlO _x /ad*/CPP	Bobst Italia, laminator CL 850D 150 m/min	1.34 ± 0.23	0.85 ± 0.10

Table 1 – Barrier performance – Assessment of printing and lamination processes for'unprotected' $AlO_x PET$

*ad = adhesive layer

Structuro	Description	OTR	WVTR
Structure		cm ³ /(m ² d)	g/(m² d)
PET/AlO _x	-	1.07 ± 0.10	0.66 ± 0.05
PET/AlO _x	After conversion simulation test	13.47 ± 2.32	5.62 ± 2.12
PET/AlO_x/protective topcoat	-	0.73 ± 0.12	0.58 ± 0.02
PET/AlO _x /protective topcoat	After conversion simulation test	0.66 ± 0.11	0.74 ± 0.03
PET/AlO _x /barrier topcoat	-	0.15 ± 0.06	0.36 ± 0.01
PET/AlO _x /barrier topcoat	After conversion simulation test	0.27 ± 0.08	0.44 ± 0.04

Table 2 – Barrier performance – Effect of conversion simulation test

Conversion	Structure -	OTR	WVTR
process		cm ³ /(m ² d)	g/(m² d)
-	PET 12 μm	≈ 120	40 - 50
AlO _x coating	PET/AlO _x	1.0 - 1.5	< 1.0
	Protective topcoat		
Topcoating	PET/AlO _x /topcoat	0.65 ± 0.09	0.56 ± 0.10
Printing	PET/AlO _x /topcoat/ink	0.78 ± 0.10	0.70 ± 0.05
Lamination	PET/AlO _x /topcoat/ink/ad/P	0.77 ± 0.10	0.69 ± 0.06
Slitting	PET/AlO _x /topcoat/ink/ad/P	0.59 ± 0.02	0.69 ± 0.01
Barrier topcoat			
Topcoating	PET/AlO _x /topcoat	0.09 ± 0.04	0.42 ± 0.03
Printing	PET/AlO _x /topcoat/ink	0.12 ± 0.08	0.45 ± 0.06
Lamination	PET/AlO _x /topcoat/ink/ad/P	0.11 ± 0.07	0.44 ± 0.13
Slitting	PET/AlO _x /topcoat/ink/ad/P	0.11 ± 0.09	0.41 ± 0.01
Pouch making	PET/AlO _x /topcoat/ink/ad/P	0.11 ± 0.09	0.44 ± 0.01

 Table 3 – Barrier performance through conversion – Gravure printing

	Structure	OTR	WVTR
Conversion process		cm ³ /(m ² d)	g/(m² d)
-	PET 12 μm	≈ 120	40 - 50
AlO _x coating	PET/AlO _x	1.0 - 1.5	< 1.0
Protective topcoat			
Topcoating	PET/AlO _x /topcoat	0.67 ± 0.01	0.64 ± 0.01
Printing	PET/AlO _x /topcoat/ink	0.72 ± 0.06	0.67 ± 0.17
	Barrier topcoat		
AlO _x coating	PET/AlO _x	1.0 - 1.5	< 1.0
Topcoating	PET/AlO _x /topcoat	0.12 ± 0.03	0.39 ± 0.05
Printing	PET/AlO _x /topcoat/ink	0.12 ± 0.03	0.39 ± 0.03

Table 4 – Barrier performance through conversion – Flexo printing