

Orthonasal and retronasal detection thresholds of 26 aroma compounds in a model alcohol-free beer: effect of threshold calculation method

Article

Accepted Version

Creative Commons: Attribution-Noncommercial-No Derivative Works 4.0

Piornos, J. A., Delgado, A., de La Burgade, R. C. J., Methven, L., Balagiannis, D. P., Koussissi, E., Brouwer, E. and Parker, J. K. (2019) Orthonasal and retronasal detection thresholds of 26 aroma compounds in a model alcohol-free beer: effect of threshold calculation method. Food Research International, 123. pp. 317-326. ISSN 0963-9969 doi: https://doi.org/10.1016/j.foodres.2019.04.034 Available at http://centaur.reading.ac.uk/83299/

It is advisable to refer to the publisher's version if you intend to cite from the work. See <u>Guidance on citing</u>.

To link to this article DOI: http://dx.doi.org/10.1016/j.foodres.2019.04.034

Publisher: Elsevier

All outputs in CentAUR are protected by Intellectual Property Rights law, including copyright law. Copyright and IPR is retained by the creators or other copyright holders. Terms and conditions for use of this material are defined in



the End User Agreement.

www.reading.ac.uk/centaur

CentAUR

Central Archive at the University of Reading

Reading's research outputs online

1	Orthonasal and retronasal detection thresholds of 26 aroma
2	compounds in a model alcohol-free beer: Effect of threshold
3	calculation method
4	
5	José A. PIORNOS ¹ , Alexia DELGADO ¹ , Rémi C.J. DE LA BURGADE ¹ , Lisa METHVEN ¹ , Dimitrios
6	P. BALAGIANNIS ¹ , Elisabeth KOUSSISSI ^{2, †} , Eric BROUWER ² , Jane K. PARKER ^{1,*}
7	
8	¹ Department of Food and Nutritional Sciences, University of Reading, RG6 6AP, UK.
9	² Heineken Supply Chain BV, Global Innovation & Research, Burgemeester Smeetsweg, 1, 2382 PH
10	Zoeterwoude, The Netherlands.
11	
12	
13	
14	
15	
16	
17	
18	
19	*Corresponding author: E-mail address: j.k.parker@reading.ac.uk
20	[†] Current address: Department of Wine, Vine and Beverage Sciences, University of West Attica, Ag.
21	Spyridona Str., 12210 Athens, Greece

22 Abstract

23 Detection thresholds are used routinely to determine the odour-active compounds in foods. The composition of a food matrix, such as hydrophobicity or solids content, has an impact on the release 24 25 of flavour compounds, and thus on thresholds. In the case of beer, thresholds determined in alcoholic 26 beer may not be the same for alcohol-free beer (AFB). Therefore, the aim of this study was to 27 determine detection thresholds for aroma compounds typically found in beer within a model AFB. The model was designed to match the sugar concentration and pH of an AFB brewed by a cold 28 29 contact process. Thresholds were measured using a 3-AFC procedure and calculated using either Best 30 Estimate Threshold (BET) method or by logistic regression. Moreover, an algorithm for the removal 31 of false positives was applied to adjust the assessors' raw responses. Retronasal thresholds were generally lower than orthonasal. Those calculated by BET were significantly higher (p < 0.05) than 32 those from logistic regression, and removal of false positives also produced significantly higher 33 34 thresholds than those from raw data. The use of logistic regression has the advantage of providing the 35 mathematical model describing the behaviour of the group. The results from this study can be used to better understand the role of flavour compounds in AFB and the effect of the calculation method to 36 37 prevent under- or overestimated results. 38

- 39
- 40
- 41
- 42 Keywords: alcohol-free beer, orthonasal threshold, retronasal threshold, best estimate threshold,
 43 logistic regression, method comparison

44 1. Introduction

Detection thresholds are commonly used in flavour science as a measure of the potency of flavour 45 compounds. They are defined as the minimum concentration of a flavour compound at which its 46 47 presence can be detected in a food or beverage, but this concept has also been applied to other research fields, such as air pollution (Leonardos, Kendall, & Barnard, 1969). Flavour compounds can 48 49 be ranked according to their odour activity by comparing their concentration in a food and their 50 detection threshold. Odour activity values are an important tool in flavour research and have been 51 used to identify key odorants in a wide variety of foods, including virgin olive oil (Guth & Grosch, 52 1993), rape honey (Ruisinger & Schieberle, 2012), and wheat beer (Langos, Granvogl, & Schieberle, 2013). It is also recognised that flavour compounds may contribute to the overall aroma of a food at 53 subthreshold concentrations due to synergistic effects with other odorants (Kishimoto, Noba, Yako, 54 55 Kobayashi, & Watanabe, 2018).

56 Aroma detection thresholds depend on many variables and are difficult to predict, if not impossible. 57 Apart from the natural differences in sensitivity of humans to different flavour compounds (Schranz, 58 Lorber, Klos, Kerschbaumer, & Buettner, 2017), other factors affect perception too. One source of 59 difference relates to the way that individuals are exposed to the odorant, either orthonasally or 60 retronasally. When sniffing a food, flavour molecules have to be released from the food matrix to the 61 air and then travel through the nasal cavity to reach the olfactory mucosa (Espinosa Díaz, 2004). This 62 corresponds to orthonasal perception of the odorant, whereas in the case of retronasal perception the flavours are released in the mouth and cross the nasopharynx via the posterior nares before reaching 63 64 the nasal cavity and olfactory mucosa.

The release of the flavour compounds from the food matrix is the starting point for both orthonasal and retronasal sensory experiences. Along with other factors, such as temperature, the composition of the food matrix plays a key role in the release of volatiles compounds (Hansson, Andersson, & Leufvén, 2001). For example, the orthonasal detection threshold for the sweaty, cheesy flavour compound 3-methylbutanoic acid in water has been reported to be 490 µg/L (Czerny et al., 2008), whereas in sunflower oil the reported threshold was only 22 µg/L (Reiners & Grosch, 1998). Other

food components, such as sugars or ethanol, also have a significant effect on the release of volatiles from the food to the air phase. Perry and Hayes (2016) concluded that thresholds determined in one food matrix should not be translated to a different food system. Such assumptions can lead to underor overestimation of the real potency of flavour chemicals in foods when comparing their concentration with inappropriate threshold values.

76 Alcoholic and alcohol-free beers are a good example of two similar food matrices where different 77 composition may affect volatile release. Lager beers usually contain 5 % alcohol by volume (ABV) 78 and low remaining fermentable sugars, i.e. glucose, fructose, sucrose, maltose and maltotriose. There 79 are studies in the literature reporting detection thresholds of flavour compounds in Lager beers (Meilgaard, 1975; Saison, De Schutter, Uyttenhove, Delvaux, & Delvaux, 2009). However, thresholds 80 determined in this alcohol-containing matrix may not be applicable to alcohol-free beers (AFB). In the 81 case of AFB, the absence of alcohol (below 0.05 % ABV), and the presence of non-fermented sugars 82 83 from wort in beers brewed by cold contact fermentation, are likely to make the release of flavour 84 compounds from this matrix different from alcoholic Lager beers.

85 The sensory method most commonly employed in determining thresholds is the three-alternative forced choice (3-AFC) discrimination method. However, even where this sensory method is applied 86 87 consistently across studies, another source of variation in published threshold values is due to the 88 calculation method used. The most commonly used calculation method is Best Estimate Threshold 89 (BET) (Czerny et al., 2008; Plotto, Margaría, Goodner, & Baldwin, 2008; Plotto, Margaría, Goodner, Goodrich, & Baldwin, 2004). According to ISO 13301:2002, this method consists of calculating the 90 91 geometrical mean of "the highest concentration missed and the next higher concentration". This is 92 done for every assessor's response and the average of the group is then calculated, this being the final 93 threshold value. This ISO standard discloses some of the disadvantages of this method, such as the calculation of thresholds out of the range of concentrations assessed when an assessor's threshold falls 94 95 above or below the range evaluated. Moreover, BET values do not give any further information about 96 the behaviour of the group for concentrations of the odorant other than the calculated threshold. In recent years, authors have started using an alternative calculation approach by means of psychometric 97

98 sigmoid functions. These functions consider the probability of perceiving the presence of the flavour compound (i.e. the probability of identifying the correct sample during the experiment) against 99 compound concentration. When using this approach, the threshold is often defined as the 100 101 concentration at which there is a 50 % probability of detecting the flavour compound (Lawless, 2010). 102 Several mathematical models have been used for this purpose, such as Weibull distribution, logistic function (Hough, Methven, & Lawless, 2013) or the Hill equation, often used in biochemistry (Perry 103 104 & Hayes, 2016). By using this modelling approach, concentrations other than 50 % probability can be 105 easily calculated, and these may be useful in certain cases, for instance, to avoid detection of off-notes 106 in foods by very sensitive consumers (Lawless, 2010). By comparing thresholds calculated using BET 107 and fitting the data to the Hill equation, Perry and Hayes (2016) observed differences between both 108 methods, BET values being lower than detection thresholds (DTs) calculated from the Hill equation in 109 most of the experiments reported. The authors did not discuss the differences between both algorithms 110 that led to the different threshold values. Furthermore, false positives, i.e. correct answers given by 111 chance, could have an effect in the final threshold values. Hough et al., (2013) proposed a threshold 112 calculation method by logistic regression using different functions, which included the application of 113 an algorithm for the adjustment of false positives. The weight of these false positive responses was 114 not evaluated nor their impact on the threshold value. Certainly, the false positives are expected to influence the final threshold values. 115

116 It is reasonable to consider that the release of flavour compounds from AFBs brewed by cold contact 117 fermentation is not comparable to water or Lager beer-like systems (usually 5 % ethanol in water). 118 Considering the impact of alcohol on flavour release, it was hypothesised that orthonasal and 119 retronasal DTs from the AFB would be different to those previously published in alcoholic beers. 120 Furthermore, the second hypothesis of this study was that the threshold calculation method had a 121 significant effect on the final value, as well as the presence of false positives. Hence, the aim of this 122 study was to determine orthonasal and retronasal detection thresholds in a model AFB of aroma 123 compounds typically found in beer. The effect of the calculation method (BET and logistic regression) 124 and the impact of false positives on the final threshold values were tested too.

125 **2.** Materials and methods

126 **2.1. Materials**

127 Carbonated water (Sparkling spring water, Aldi Stores Ltd., UK), sucrose (> 90 %, Silver Spoon,

- 128 UK), fructose (> 90 %, Tate & Lyle, UK), and glucose powder (> 90 %, Thornton & Ross Ltd., UK)
- 129 were purchased at a local store. C☆SweetTM glucose syrup (composition in dry base: 5 % w/w
- 130 glucose, 75 % w/w maltose, 10 % w/w maltotriose, 10 % w/w unspecified components) was donated

131 by Cargill (Manchester, UK).

132 **2.2.** Aroma compounds

- 133 The following aroma compounds were purchased from Sigma-Aldrich (purity in parenthesis):
- acetaldehyde (\geq 99 %), acetic acid (\geq 99.5 %), 2,3-butanedione (97 %), butanoic acid (\geq 99 %),
- dimethyl sulfide (≥99 %), 5(or 2)-ethyl-4-hydroxy-2(or 5)-methyl-3(2*H*)-furanone (homofuraneol,
- 136 96 %), Z-4-heptenal (≥98 %), 3-hydroxy-4,5-dimethyl-2(5*H*)-furanone (sotolone, 10 % in propylene
- 137 glycol), 4-hydroxy-2,5-dimethyl-3(2*H*)-furanone (furaneol, \geq 98 %), methional (\geq 97 %), 2'-
- 138 methoxyacetophenone (99 %), 2-methoxy-4-methylphenol (≥98 %), 2-methoxyphenol (≥99 %), 2-
- 139 methoxy-4-vinylphenol (\geq 98%), 2-methylbutanal (\geq 95%), 3-methylbutanal (\geq 97%), 3-
- 140 methylbutanoic acid (99 %), 3-methyl-1-butanol (≥98%), methylpropanal (≥98 %), 2-
- 141 methylthiophene (98 %), 2,3-pentanedione (≥96 %), phenylacetaldehyde (10 % in ethanol), 2-
- 142 phenylacetic acid (≥99 %), 2-phenylethanol (≥99 %), vanillin (≥97 %), 4-vinylphenol (10 % in
- 143 propylene glycol). All were food grade except 2'-methoxyacetophenone and 2-methylthiophene.
- 144

2.3. Preparation of the model alcohol-free beer

A model beer was prepared to match the sugar content of an alcohol-free beer brewed following a standard cold contact fermentation procedure, bottling and pasteurisation carried out at Heineken's pilot brewery (Zoeterwoude, The Netherlands). First, a five-fold concentrated solution of sugars was prepared in tap water. Then, one part of the sugar solution was diluted into four parts of carbonated water, reaching the final concentration of sugars: 7.2 g/L glucose, 2.1 g/L fructose, 0.6 g/L sucrose, 26.9 g/L maltose and 3.6 g/L maltotriose. In parallel, a stock solution of odorants was prepared in absolute ethanol (Sigma-Aldrich, UK). Then, 400 μ L stock solutions containing the odorant (absolute ethanol for blanks) were added to one litre of model beer. The final pH of the model was 4.50 and the final ethanol content was 0.04 %.

154 **2.4.** Sensory methodology

155 For each compound, the aim was to collect threshold data from 24 trained and experienced sensory 156 assessors. To achieve this, allowing for absences, there was a pool of 33 assessors (8 men, 25 women, ages 25 to 60). The assessors were recruited from the flavour and sensory groups of The University of 157 Reading, all of whom had experience in describing a wide range of aroma chemicals. Preliminary 158 159 sensory experiments were carried out in order to establish the range of concentrations for the 160 threshold experiments, as well as to familiarise the panellists with the aroma chemicals. For 7 out of 161 26 compounds for orthonasal assessment and 2 for retronasal assessment, only 12 assessors were available. The experiments were designed following a three-alternative forced choice (3-AFC) 162 163 methodology (ISO13301:2002). Each sample (10 mL) was presented in a screw-capped 27-mL clear 164 glass vial (height 72 mm, internal diameter 23 mm) at a temperature between 9 and 14 °C. Six 165 concentrations of each compound were presented in ascending order, each being 3 times more concentrated than the previous sample. Each concentration was presented along with two blank 166 167 samples per level. Within each set of three, the order of blanks and the sample was balanced and 168 randomised (AAB, ABA, or BAA) across the panellists, and all samples were coded with 3-digit 169 random numbers. During each one-hour sensory session, three compounds were presented to the panel. After sniffing all the samples to assess orthonasal perception, the samples were presented for a 170 171 second time, in a random but balanced order, and the panellists were asked to taste them for retronasal 172 perception. The vials were presented uncapped to avoid interference with aroma from the headspace when assessing the samples for retronasal perception. Compusense Cloud (Compusense Inc., Guelph, 173 ON, Canada) was used to guide panellists during the study as well as to collect responses. The 174 experiments were carried out in individual sensory booths (controlled temperature 18-20°C) at the 175 176 Sensory Science Centre of The University of Reading.

177 **2.5. Data analysis**

178 **2.5.1.**Adjustment of assessors' responses by chance

In order to remove false positives, i.e. positive responses given by chance, the methodology published
by Hough et al. (2013) was followed. Responses were classified into four different cases exemplified
in Table 1:

Case 1: Negative response. If the panellist could not identify the sample containing the aroma
compounds, this remained as "no" in all cases.

Case 2: "Yes before or next to no". This applies to all positive responses before a negative answer, and also those just after a negative response (i.e. those first in a row of correct answers). In these cases, first, the proportion of discriminators (P_d) was calculated (Lawless, 2010) (Eq. 1):

188
$$P_{d} = \frac{P_{corr} - P_{chance}}{1 - P_{chance}} \qquad Eq. 1$$

189where P_{corr} is the proportion of correct answers at a concentration level and P_{chance} is the190probability of getting a correct answer by chance (in 3-AFC tests, this is 1/3). Then, the ratio191 P_d/P_{corr} was calculated and compared with a random number X from 0.000 to 1.000 generated192using the function "RAND". If $P_d/P_{corr} < X$, the original positive response was corrected and193replaced by a negative answer.

- Case 3: "Second yes after last no". In this case, the same procedure as in case 2 was followed,
 although the P_{chance} used in this case was 1/9. This was because this positive response is the
 second in a row, so the chance of getting two correct answers is (1/3) × (1/3).
- Case 4: "Third and further yes after no". The probability of choosing a third correct answer by
 chance is (1/3) × (1/3) × (1/3). This is below 5 %, so it was assumed that these were real
 positives and consequently kept as positives.

200 The different steps and criteria were implemented into an Excel spreadsheet (Microsoft Office 365201 ProPlus).

202

2.5.2.Best estimated threshold (BET)

BETs were calculated from raw and adjusted data according to the procedure reported in ISO
13301:2002. BETs for each assessor and compound were calculated as the geometric mean of the
highest concentration for a negative response and the next concentration. In the case where an
assessor's response was either negative or positive for all the concentrations presented, the BET was
calculated as the geometrical mean using the next concentration in the series (up or down,
respectively) which had not been tested.

209 **2.5.3.Logistic regression**

210 The raw and adjusted data were fitted to the logistic function (Eq. 2) using XLSTAT 2012:

211
$$P_{c}(\ln C) = \frac{1}{1 + e^{-(\alpha + \beta \ln C)}}$$
 Eq. 2

212 Where P_c is the probability of a correct answer, α is the factor that sets the displacement of the curve 213 along the abscissa axis, and β is the steepness factor. The detection threshold was considered as the 214 concentration at which the probability of correct answer was 0.50.

215 **2.5.4.Statis**

2.5.4. Statistical analysis

Thresholds calculated by BET and logistic regression, from raw data and after removal of false positives (adjusted data), were compared aiming to determine significant differences between these four different methods. T-test for paired samples ($\alpha = 0.05$) was applied to the logarithms of the threshold values grouped into methods, i.e. not distinguishing between orthonasal and retronasal thresholds for this purpose.

221 **3. Results**

222 **3.1.** Orthonasal and retronasal thresholds in a model AFB

Table 2 shows the orthonasal detection thresholds for 26 aroma compounds in a model alcohol-free

beer, calculated by the four different methods. The overall range of values obtained for different

225 compounds was noticeably broad, from below 1 μ g/L to more than 100,000 μ g/L. The highest

- orthonasal DTs, (those over 1,000 µg/L, i.e. 1 ppm), were found for acetic acid (131,000-
- 227 391,000 μg/L), 2-methylthiophene (1,732-11,800 μg/L), and 2-phenylacetic acid (1,174-5,830 μg/L).

228 On the other hand, the lowest values (those below 1 μ g/L, i.e. 1 ppb) were found for Z-4-heptenal (0.0035-0.022 µg/L), methional (0.19-0.68 µg/L), and 3-methylbutanal (0.31-0.64 µg/L). A similar 229 scenario was observed for these compounds when assessed for retronasal perception. Table 3 shows 230 the results for retronasal detection thresholds for 20 aroma compounds. The compounds with the 231 232 highest retronasal detection thresholds were acetic acid (22,100-104,000 µg/L), 4-vinylphenol (90.0-4,210 μ g/L), and 2-phenylacetic acid (12.6-1,690 μ g/L). As for orthonasal perception, methional 233 234 (0.040-1.78 µg/L) and 3-methylbutanal (0.22-0.74 µg/L) exhibited the lowest retronasal threshold 235 values. Orthonasal threshold values were higher than retronasal for most of the compounds evaluated. 236 The only exceptions were dimethyl sulfide and 3-methyl-1-butanol, for which retronasal detection 237 thresholds were higher than orthonasal. For other compounds (methional, 3-methylbutanal, and 4-238 vinylphenol), the difference between orthonasal and retronasal thresholds was less apparent as it was 239 dependent on the method used to calculate the threshold.

240

3.2. Comparison of calculation methods

241 In this study, two different threshold calculation methods were used, as well as an algorithm for the 242 removal of false positives. As shown in Tables 2 and 3, both orthonasal and retronasal detection thresholds were affected by the calculation method (BET or logistic regression) and the removal of 243 false positives (raw and adjusted data). Figure 1 shows the comparison plots for the different 244 245 calculation approaches, where orthonasal and retronasal thresholds from each method are plotted against each other. Thresholds calculated from adjusted data were higher than those from raw data, 246 independently of the compound assessed, this increase being higher in the case of the logistic 247 regression than the BET. This can be observed when comparing the trendline equations (Figure 1a 248 249 and 1b), where, although the slopes were very close to one, the lines do not pass through zero and there is a significant intercept. The interpretation of these trendline equations and the meaning of this 250 251 intercept is complicated by the fact that the thresholds are plotted on a log plot. The trendline equations were expressed in the following terms: $\ln DT_1 = a \times \ln DT_2 + \ln (b)$ where $a \approx 1$ and the 252 253 intercept is ln (b). Using the standard rules of logarithms, $DT_1 = DT_2 \times b$, so b represents the constant

ratio between the methods. The intercept from the graph gives ln (b), so the constant ratio is theexponential of the intercept, or exp (b).

256 In the case of the adjustment of false positives, the intercept in Figure 1a (+1.4698) was higher than in 257 Figure 1b (+0.3792). This means that the values from logistic regression and adjusted data were, on 258 average, 4.3 times (i.e. exp (+1.4698)) higher than those from raw data, whereas this difference was only 1.5 times (exp (+0.3792)) in the case of BET. Differences were also found between BET and 259 logistic regression methodologies from the same sets of data (raw and adjusted data) (Figures 1c and 260 261 1d). In both cases, BET produced higher threshold values than logistic regression and the difference 262 was greater for raw data (intercept +1.5825, ratio 4.9) than adjusted data (intercept +0.4804, ratio 1.6). In order to identify significant differences between methods, t-tests for paired samples were applied. 263 P-values from these tests showed significant differences (p < 0.05) between the results from BET and 264 logistic regression (p = 1.4×10^{-14} for BET raw vs. logistic regression raw; p = 1.2×10^{-9} for BET 265 266 adjusted vs. logistic regression adjusted), as well as for those calculated from raw and adjusted data for both methods ($p = 7.2 \times 10^{-27}$ for BET raw vs. BET adjusted; $p = 7.1 \times 10^{-21}$ for logistic regression 267 raw vs. logistic regression adjusted). Surprisingly, thresholds from logistic regression from adjusted 268 data and standard BET from raw data were not significantly different (p = 0.31). 269

270

3.3. Logistic regression for the calculation of thresholds

271 Supplementary Tables A.1 and A.2 show the parameters that define the logistic models for the 272 probability of a correct answer (i.e. correct identification of the aroma compound) against the logarithm of the concentration of the compound. The logistic model used here is defined by two 273 274 parameters: α sets the displacement along the x-axis, and β is the steepness factor. According to Eq. 2, 275 a lower value of α is translated in a higher value for the inflexion point of the sigmoidal curve, 276 whereas higher values of β give steeper curves. For both orthonasal and retronasal studies, the 277 adjustment of the data for the removal of false positives produced a decrease in the α parameter, which resulted in a displacement of the curve towards the right and, thus, higher thresholds. The 278 steepness factor β was also affected by the adjustment of the data because the β -values from adjusted 279 data were higher than those from raw data. An exception to this trend was the orthonasal model for Z-280

4-heptenal, for which the α-factor was higher after the removal of false positives. Despite this, the orthonasal detection thresholds for these compounds were still higher because the effect of the αfactor was compensated for by a higher β-factor.

284 The removal of false positives also affected the goodness of fit of the logistic model. The adjustment of the data produced an increase of the pseudo- R^2 values in all cases, for both orthonasal and 285 286 retronasal models (Supplementary Tables A.1 and A.2). Furthermore, the confidence interval for the thresholds calculated using this method were considerably narrower after the removal of false 287 288 positives (Figure 2). For example, the error bar for the retronasal detection threshold of vanillin was 289 reduced from three orders of magnitude to only one (Figure 2b). For a few compounds (2methylbutanal and 3-methyl-1-butanol for orthonasal, and methional, 2-methoxy-4-methylphenol, 2-290 phenylacetic acid, and 4-vinylphenol for retronasal detection thresholds) confidence intervals could 291 not be calculated properly when using raw data because the calculation method could not converge to 292 293 a solution after 100 iterations. This issue was resolved after the removal of false positives, when 294 confidence intervals could be calculated in all cases.

295 **4. Discussion**

296

4.1 Threshold calculation method

Thresholds calculated by BET and logistic regression were found to be significantly different 297 298 (p < 0.05) for both orthonasal and retronasal data. Logistic regression generated lower threshold 299 values from both raw and adjusted data. Psychometric functions take into consideration all the 300 positive responses along the entire range of concentrations. On the other hand, BET only considers 301 positive answers that are not followed by negative answers. This makes logistic curves displaced towards the left to lower concentrations, resulting in lower threshold values. Previous studies have 302 303 compared the standardised BET method with logistic regression. Perry and Hayes (2016) found that 304 thresholds from BET were lower than those calculated by using logistic regression. These results, which may seem to be contradictory to those from the present study, might be explained by the fact 305 306 that these authors used an equation model that it is restricted from 33 % to 100 % probabilities on the

307 ordinate axis. In our study, we did not use a restricted model, as shown in Eq. 2, so the probability of 308 correct answer can vary from 0 % to 100 %. In our opinion, the use of an algorithm for the removal of 309 false positives already discards the correct answers given by chance, so the restriction at 33 % chance 310 should not be necessary anymore. When using restricted models, it is common to define the threshold 311 at 66.6 % chance as the middle point of the curve (between 33.3 % and 100 %). This might be another 312 reason why these authors obtained higher threshold values with logistic regression. Lawless (2010) 313 also used 66.6 % probability as the corrected 50 % detection level following a similar reasoning.

314 The effect of the removal or correction of false positives was also covered in the present study. As 315 shown above, threshold values increased significantly after the application of this algorithm. In previous studies, differences between BET raw and logistic regression adjusted thresholds were 316 observed. Hough et al. (2013) reported that the BET method using raw data produced lower 317 thresholds than logistic regression from adjusted data. This was associated with the fact that in logistic 318 319 regression the adjustment of the responses pushed the threshold upwards, whereas this data treatment was not applied when using BET. In our study, there was not a clear trend when comparing these two 320 321 sets of thresholds. Not all BET raw thresholds were lower than the corresponding logistic regression 322 adjusted threshold (Figure 1e) and on average the results from these methodologies were not 323 significantly different (p = 0.31). This demonstrated that logistic regression along with the removal of 324 false positives is a methodology comparable to the standardised BET, with the advantage of providing 325 further information such as the mathematical model describing the response of the group at different 326 concentrations of an aroma compound.

327

4.2 Orthonasal thresholds

In the literature, perception thresholds are available for different aroma compounds determined in a variety of matrices, e.g. water (Czerny et al., 2008), air (Schranz et al., 2017), and beer (Meilgaard, 1975, 1982; Saison et al., 2009). In Figure 3, those found in the literature (diamonds) for water, 9.4 % ethanol or beer are compared to those from the present study (horizontal bars) for both orthonasal (Fig. 3a) and retronasal (Fig. 3b) perception. Full details of these threshold values from the literature

can be found in Appendix B. Before plotting them, all the thresholds units were converted into µg/Lfor comparison.

The impact of ethanol on aroma release was demonstrated instrumentally by Perpète and Collin (2000), who observed higher retention of 2-methylbutanal and 3-methylbutanal when increasing the concentration of ethanol from 0 to 5 % in an aqueous solution. This was explained by the 'cosolvent' effect of ethanol in water, thus increasing the solubility of these aldehydes and reducing their partition coefficients between the water/ethanol solution and the air (Tsachaki et al., 2008). However, Figure 3a shows that the literature detection thresholds which had been determined in 9.4 % ethanol fell within the same range as those determined in water in 4 out of 5 cases.

342 The role of sugar on flavour release has been studied more extensively. Perpète and Collin (2000) 343 demonstrated that the presence of sugars in beer produced an increase in the release of 2- and 3-344 methylbutanal, up to a maximum sugar concentration of 40 g/L. Tsitlakidou, Van Loey, Methyen, & 345 Elmore (2019), and Hansson et al. (2001) also showed this salting-out effect of non-polar compounds 346 in soft drinks when sugar increased from ~40-150 g/L and 200-600 g/L, respectively. Bredie, 347 Mottram, & Birch (1994) also showed an increase in volatility with added glucose (200 g/L) for hydrophobic compounds such as menthol and limonene in a maltodextrose solution, but no effect with 348 349 the more polar compounds (3-methylbutyl acetate and 2,3-butanedione). Banavara, Rabe, Krings, & 350 Berger (2002) modelled flavour release and predicted a salting-out effect for most compounds. 351 However, experimentally they reported that the effect was much less than predicted, and not statistically significant for more polar compounds. These literature studies in accord with our data, 352 353 which cover a range of more polar compounds, rather than the terpenes and longer chain aldehydes 354 which showed the biggest salting-out effects in these literature studies. On average the more polar compounds in our study (6: homofuraneol, 8: sotolone, 9: furaneol, 12: 2-methoxy-4-methylphenol, 355 14: 2-methoxy-4-vinylphenol, 25: vanillin, and 26: 4-vinylphenol) showed no evidence of salting-out 356 357 and presented higher orthonasal thresholds than those from the literature (Fig. 3a). This may be due to 358 the interaction between the sugars and these more polar volatiles.

359 The effect of carbonation on flavour release has been studied, particularly in relation to champagne. Pozo-Bayón, Santos, Martín-Álvarez, & Reineccius (2009) showed an increase in aroma release with 360 carbonation but stressed the importance of the physicochemical character of the volatiles, showing the 361 most hydrophobic, most volatile compounds were affected the most. Saint-Eve et al. (2009) looked at 362 363 the effect of adding 10 g/L sucrose on aroma release of carbonated beverages. Carbonation had by far the bigger effect and increased volatile release, but added sucrose had no impact on aroma release in 364 365 the carbonated samples. Our results did not show a corresponding decrease in aroma threshold with 366 carbonation, but surface activity, bubble size and bubble frequency are important parameters which 367 we could not readily control.

368

4.3 Retronasal thresholds

369 Retronasal thresholds were much scarcer in the literature, most of them being comparable to our 370 results (Fig. 3b). In this study, retronasal thresholds for 2,3-butanedione (3), butanoic acid (4), 2-371 methoxy-2-vinylphenol (14), 3-methyl-1-butanol (18) and 2-phenylacetic acid (23) were lower than 372 those from the literature, whereas furaneol (9) showed a higher threshold in the AFB model. Apart 373 from the matrix effect, the differences between thresholds from the literature and our results could be due to the diversity of methodologies employed. This includes differences in calculation method 374 375 (BET, interpolation using probability vs. concentration graphs), number of panellists and sample 376 presentation (triangle, 3-AFC, duo-trio test, sets of samples presented in either ascending or 377 descending concentrations) (Guadagni, Buttery, & Okano, 1963; Langos et al., 2013; Rothe, Wölm, Tunger, & Siebert, 1972). All too often, authors of threshold studies do not fully specify the details of 378 379 their studies, this making comparisons less valid. This was demonstrated in a comprehensive literature 380 search, and summarised in Appendix B, which shows thresholds in the literature and the main 381 characteristics of the sensory study.

382 Comparing the results for orthonasal and retronasal perception, retronasal DTs tended to be lower

than orthonasal for most of the compounds assessed, independently of the data treatment (Tables 2

and 3). The reason behind this does not seem to be very clear. Retronasal perception is a more

385 complex process which also involves changes in temperature of the foodstuff, dilution with saliva,

386 binding to mucous membranes in mouth and tongue, increase of air/food surface area and the mixing effect of swallowing (Taylor & Roozen, 1996). Due to the higher complexity of the retronasal 387 pathway, Espinosa Díaz (2004) hypothesised a higher efficiency of the orthonasal pathway, thus 388 requiring lower concentrations of odorants for the same odour intensity as the retronasal pathway. On 389 390 the other hand, the opposite behaviour was observed by Voirol and Daget (1986) for vanillin and citral, which was related to a higher concentration of these odorants in the vapor phase when put in 391 392 the mouth, as well as the influence of other non-chemical interactions. From the results of the current 393 study it appears that most of the compounds studied corresponded with the latter theory as their 394 retronasal thresholds were lower. For the compounds that were the exceptions to this, there is no clear 395 reason why they were all detected at lower levels orthonasally. Dimethyl sulfide is a highly volatile 396 compound and hence it is perhaps unsurprising that its orthonasal DT would be lower. However, this 397 was not the case for the other three less volatile compounds (homofuraneol, furaneol, and 3-methyl-1-398 butanol). The relatively low log P values of these four compounds did not seem to be the reason 399 behind this behaviour either, since other compounds with similar log P values (methylpropanal), 2-400 methoxyphenol and 2-methylbutanal) did not show the same effect.

401 **5.** Conclusions

402 Orthonasal and retronasal detection thresholds of 26 and 20 aroma compounds, respectively, are 403 reported in a model AFB for the first time. Four different methodologies for threshold calculation 404 were applied and compared, elucidating the role of the calculation procedure in the final threshold value. Threshold values were found to be method-dependent (BET and logistic regression), as well as 405 406 affected by the presence of false positives or correct answers given by chance. Although BET is a 407 standard commonly used threshold calculation method, logistic regression is recommended for the additional information extracted from the data. Additionally, data treatment for the removal of false 408 409 positives is strongly recommended in order to obtain a more realistic mathematical model.

410 The determination of perception thresholds in the correct matrix is crucial for estimating the potency 411 of flavour compounds in conditions closer to the real beverage. After a comprehensive literature 412 research, we have shown that for many of the compounds studied, our results in a model AFB were

413 comparable to those reported in water. However, a group of polar compounds (mainly furanones and 414 phenols) consistently showed higher orthonasal detection thresholds in the model AFB compared to water (literature values). Comparison of threshold values from different studies may be very risky due 415 to the lack of consistency of the methods for threshold determination so it is strongly recommended 416 417 that the experimental setup, matrix in which the odorant was presented and threshold calculation method are all extracted from the primary source wherever possible to ensure they are appropriate. 418 The results reported in the present study can be of great importance for the brewing industry when 419 420 studying the aroma composition of alcohol-free beers brewed by cold contact fermentation. The 421 market for alcohol-free beers is currently undergoing huge growth worldwide, and the determination 422 of perception thresholds is essential to understand the role of flavours compounds and their contribution to the overall aroma. 423

424 Acknowledgments

425 This study has been funded by Heineken Supply Chain BV. We would also like to acknowledge

426 Compusense Inc. for provision of the sensory software under an academic consortium agreement as

427 well as all the panellists that participated in this study.

428 Conflicts of interest

429 The authors declare no conflicts of interest.

431 References

- 432 Banavara, D. S., Rabe, S., Krings, U., & Berger, R. G. (2002). Modeling dynamic flavor release from
- 433 water. *Journal of Agricultural and Food Chemistry*, *50*(22), 6448–6452.
- 434 <u>https://doi.org/10.1021/jf020232c</u>
- 435 Bredie, W. L. P., Mottram, D. S., & Birch, G. G. (1994). Aroma binding in maltodextrose solutions.
- 436 In H. Maarse & D. G. Van der Heij (Eds.), *Trends in Flavour Research* (pp. 139–143). New York:
- 437 Elsevier.
- 438 Czerny, M., Christlbauer, M., Christlbauer, M., Fischer, A., Granvogl, M., Hammer, M., ...
- 439 Schieberle, P. (2008). Re-investigation on odour thresholds of key food aroma compounds and
- 440 development of an aroma language based on odour qualities of defined aqueous odorant solutions.
- 441 European Food Research and Technology, 228(2), 265–273. https://doi.org/10.1007/s00217-008-
- 442 0931-x
- Espinosa Díaz, M. (2004). Comparison between orthonasal and retronasal flavour perception at
- different concentrations. *Flavour and Fragrance Journal*, *19*(6), 499–504.
- 445 https://doi.org/10.1002/ffj.1475
- 446 Guadagni, D. G., Buttery, R. G., & Okano, S. (1963). Odour thresholds of some organic compounds
- 447 associated with food flavours. *Journal of the Science of Food and Agriculture*, 14(10), 761–765.
- 448 https://doi.org/10.1002/jsfa.2740141014
- 449 Guth, H., & Grosch, W. (1993). Quantitation of Potent Odorants of Virgin Olive Oil by Stable-Isotope
- 450 Dilution Assays. Journal of the American Oil Chemists' Society, 70(5), 513–518. Retrieved from
- 451 https://link.springer.com/article/10.1007/BF02542586
- 452 Hansson, A., Andersson, J., & Leufvén, A. (2001). The effects of sugars and pectin on flavour release
- 453 from a soft drink-related model system. *Food Chemistry*, 72(3), 363–368.
- 454 https://doi.org/https://doi.org/10.1016/S0308-8146(00)00243-0
- 455 Hough, G., Methven, L., & Lawless, H. T. (2013). Survival Analysis Statistics Applied to Threshold
- 456 Data Obtained from the Ascending Forced-Choice Method of Limits. Journal of Sensory Studies,

- 457 28(5), 414–421. https://doi.org/10.1111/joss.12067
- 458 ISO 13301:2002. (n.d.). Sensory analysis Methodology General guidance for measuring odour,
- 459 flavour and taste detection thresholds by a three-alternative forced-choice (3-AFC) procedure.
- 460 Kishimoto, T., Noba, S., Yako, N., Kobayashi, M., & Watanabe, T. (2018). Simulation of Pilsner-type
- 461 beer aroma using 76 odor-active compounds. Journal of Bioscience and Bioengineering, 126(3), 330–
- 462 338. https://doi.org/10.1016/J.JBIOSC.2018.03.015
- 463 Langos, D., Granvogl, M., & Schieberle, P. (2013). Characterization of the Key Aroma Compounds in
- 464 Two Bavarian Wheat Beers by Means of the Sensomics Approach. Journal of Agricultural and Food
- 465 *Chemistry*, *61*(47), 11303–11311. https://doi.org/10.1021/jf403912j
- 466 Lawless, H. T. (2010). A simple alternative analysis for threshold data determined by ascending
- 467 forced-choice methods of limits. *Journal of Sensory Studies*, 25(3), 332–346.
- 468 https://doi.org/10.1111/j.1745-459X.2009.00262.x
- 469 Leonardos, G., Kendall, D., & Barnard, N. (1969). Odor Threshold Determinations of 53 Odorant
- 470 Chemicals. Journal of the Air Pollution Control Association, 19(2), 91–95.
- 471 https://doi.org/10.1080/00022470.1969.10466465
- 472 Meilgaard, M. C. (1975). Flavor Chemistry of Beer. II. Flavor and threshold of 239 aroma volatiles.
- 473 *Master Brewers Association of the Americas Technical Quarterly*, *12*(3), 151–168.
- 474 Meilgaard, M. C. (1982). Prediction of Flavor Differences between Beers from Their Chemical
- 475 Composition. *Journal of Agricultural and Food Chemistry*, *30*(6), 1009–1017.
- 476 https://doi.org/10.1021/jf00114a002
- 477 Perpète, P., & Collin, S. (2000). Influence of beer ethanol content on the wort flavour perception.
- 478 Food Chemistry, 71(3), 379–385. https://doi.org/10.1016/S0308-8146(00)00179-5
- 479 Perry, D., & Hayes, J. (2016). Effects of Matrix Composition on Detection Threshold Estimates for
- 480 Methyl Anthranilate and 2-Aminoacetophenone. *Foods*, *5*(2), 35.
- 481 https://doi.org/10.3390/foods5020035

- 482 Plotto, A., Margaría, C. A., Goodner, K. L., & Baldwin, E. A. (2008). Odour and flavour thresholds
- 483 for key aroma components in an orange juice matrix: esters and miscellaneous compounds. *Flavour*
- 484 *and Fragrance Journal*, 23(6), 398–406. https://doi.org/10.1002/ffj.1888
- 485 Plotto, A., Margaría, C. A., Goodner, K. L., Goodrich, R., & Baldwin, E. A. (2004). Odour and
- 486 flavour thresholds for key aroma components in an orange juice matrix: terpenes and aldehydes.
- 487 Flavour and Fragrance Journal, 19(6), 491–498. https://doi.org/10.1002/ffj.1470
- 488 Pozo-Bayón, M. Á., Santos, M., Martín-Álvarez, P. J., & Reineccius, G. (2009). Influence of
- 489 carbonation on aroma release from liquid systems using an artificial throat and a proton transfer
- 490 reaction-mass spectrometric technique (PTR-MS). *Flavour and Fragrance Journal*, 24(5), 226–233.
- 491 https://doi.org/10.1002/ffj.1934
- 492 Reiners, J., & Grosch, W. (1998). Odorants of Virgin Olive Oils with Different Flavor Profiles.
- 493 Journal of Agricultural and Food Chemistry, 46(7), 2754–2763. https://doi.org/10.1021/jf970940b
- 494 Rothe, M., Wölm, G., Tunger, L., & Siebert, H.-J. (1972). Schwellenkonzentrationen von
- 495 Aromastoffen und ihre Nutzung zur Auswertung von Aromaanalysen. *Die Nahrung*, *16*(5), 483–495.
- 496 https://doi.org/https://doi.org/10.1002/food.19720160509
- 497 Ruisinger, B., & Schieberle, P. (2012). Characterization of the key aroma compounds in rape honey
- 498 by means of the molecular sensory science concept. Journal of Agricultural and Food Chemistry, 60,
- 499 4186–4194. https://doi.org/10.1021/acs.jafc.6b04499
- 500 Saint-Eve, A., Déléris, I., Aubin, E., Semon, E., Feron, G., Rabillier, J.-M., ... Souchon, I. (2009).
- 501 Influence of composition (CO2 and sugar) on aroma release and perception of mint-flavored
- 502 carbonated beverages. *Journal of Agricultural and Food Chemistry*, 57(13), 5891–5898.
- 503 https://doi.org/10.1021/jf900542j
- Saison, D., De Schutter, D. P., Uyttenhove, B., Delvaux, F., & Delvaux, F. R. (2009). Contribution of
- staling compounds to the aged flavour of lager beer by studying their flavour thresholds. *Food*
- 506 Chemistry, 114(4), 1206–1215. https://doi.org/10.1016/J.FOODCHEM.2008.10.078

- 507 Schranz, M., Lorber, K., Klos, K., Kerschbaumer, J., & Buettner, A. (2017). Influence of the chemical
- 508 structure on the odor qualities and odor thresholds of guaiacol-derived odorants, Part 1: Alkylated,
- alkenylated and methoxylated derivatives. *Food Chemistry*, 232, 808–819.
- 510 https://doi.org/10.1016/j.foodchem.2017.04.070
- 511 Taylor, A. J., & Roozen, J. P. (1996). Volatile Flavor Release from Foods during Eating. *Critical*
- 512 *Reviews in Food Science and Nutrition*, *36*(8), 765–784. https://doi.org/10.1080/10408399609527749
- 513 Tsachaki, M., Gady, A.-L., Kalopesas, M., Linforth, R. S. T., Athès, V., Marin, M., & Taylor, A. J.
- 514 (2008). Effect of Ethanol, Temperature, and Gas Flow Rate on Volatile Release from Aqueous
- 515 Solutions under Dynamic Headspace Dilution Conditions. Journal of Agricultural and Food
- 516 *Chemistry*, 56(13), 5308–5315. <u>https://doi.org/10.1021/jf800225y</u>
- 517 Tsitlakidou, P., Van Loey, A., Methven, L., & Elmore, J. S. (2019). Effect of sugar reduction on
- flavour release and sensory perception in an orange juice soft drink model. *Food Chemistry*, 284,
- 519 125–132. https://doi.org/10.1016/j.foodchem.2019.01.070
- 520 Voirol, E., & Daget, N. (1986). Comparative study of nasal and retronasal olfactory perception. LWT -
- 521 *Food Science and Technology*, *19*(4), 316–319.

522 FIGURE CAPTIONS

523 Figure 1. Comparison of methods. Natural logarithms of orthonasal and retronasal thresholds (in

524 μ g/L) calculated by the different methodologies have been plotted, as well as the linear trend line

525 (red) and the line of equality (grey). BET raw: Best Estimate Threshold from raw data; BET adj: BET

526 from adjusted data (i.e. with false positives removed); LR raw: Logistic regression from raw data; LR

- 527 adj: Logistic regression with adjusted data.
- **Figure 2**. Detection thresholds calculated by logistic regression showing confidence intervals ($\alpha = 95\%$) for orthonasal (a) and retronasal (b) perceptions, *Confidence interval not available.
- Figure 3. Comparison of orthonasal (a) and retronasal (b) detection thresholds determined in this study
 and those found in the literature. Legend: Thresholds calculated by (-) BET from raw data, (-) BET
 from adjusted data, (-) logistic regression from raw data, (-) logistic regression from adjusted data;
 thresholds from the literature: (*) in water and (*) other matrices (9.4 % ethanol in Fig. 3a or beer in
 Fig. 3b).

- **Table 1.** Example of an assessor's response showing the different cases according to the algorithm for
- 537 the removal of false positives.

Concentration, µg/L	1	3	9	27	81	273
Assessor's response	no	yes	no	yes	yes	yes
Case	1	2	1	2	3	4

	Compound	Odour quality	Orthonasal detection threshold, µg/L						
No.			Logistic	regression	B	ЕТ	Threshold range in		
		-	Raw	Adjusted	Raw	Adjusted	literature, µg/L		
1	acetaldehyde*	fruity, solvent	14.5	45.8	37.5	49.3	$11.7^{a} - 900^{b}$		
2	acetic acid	vinegar	131,000	355,000	297,000	391,000	$100^{\circ} - 522,000^{\circ}$		
3	2,3-butanedione	caramel, raw meat, butter	1.25	5.19	4.28	6.18	$1^{e} - 15^{f, g}$		
4	butanoic acid	cheese, sour, vomit	907	2,080	1,390	2,190	$1^{c} - 4,752^{d}$		
5	dimethyl sulfide*	vegetables, garlic, savoury	13.4	48.4	47.2	89.5	$0.24^{h} - 5^{b}$		
6	5-ethyl-4-hydroxy-2- methyl-3(2 <i>H</i>)-furanone (homofuraneol)	candy floss, caramel	35.3	102	83.2	131	1.15 ⁱ		
7	Z-4-heptenal*	lamb fat, rancid oil, fish, rubber	0.0035	0.016	0.014	0.022	0.0087 ^e		
8	3-hydroxy-4,5-dimethyl- 2(5H)-furanone (sotolone)*	curry, cooked sugar	8.68	28.3	22.9	27.5	$0.3^{g,j} - 20^{i}$		
9	4-hydroxy-2,5-dimethyl- 3(2 <i>H</i>)-furanone (furaneol)	candy floss, strawberry	49.4	148	87.3	158	$1^{\circ} - 1,000^{\circ}$		
10	methional	boiled potato, metallic	0.19	0.47	0.47	0.68	$0.2^{g, k, 1} - 1.8^{e, j}$		
11	2'-methoxyacetophenone	plastic, chemical, petrol	688	2,260	2,880	3,300			
12	2-methoxy-4-methylphenol	smoky, bacon, vanilla	20.7	37.2	27.7	34.8	21 ^e		
13	2-methoxyphenol	smoky, chemical	0.67	2.10	1.59	2.51	$0.84^{e} - 3.39^{a}$		
14	2-methoxy-4-vinylphenol	cloves, medicinal, bacon	33.1	81.5	79.5	99.9	$3^{m} - 100^{j}$		
15	2-methylbutanal	fruity, sweet	1.88	23.4	37.0	50.9	$1.5^{e} - 5.6^{d}$		
16	3-methylbutanal	malty, cheese	0.31	0.61	0.47	0.64	$0.15^{n} - 8^{b}$		
17	3-methylbutanoic acid*	cheese, fruity, sour	89.4	376	360	624	$132^{\circ} - 2,754^{d}$		
18	3-methyl-1-butanol	banana, nail polish remover	23.3	89.0	96.5	127	$203^{h, p} - 4,750^{q}$		
19	methylpropanal	nutty, chemical	1.01	4.32	3.44	5.69	$0.49^{e} - 43.5^{o}$		
20	2-methylthiophene*	vegetable stock, onion, solvent	1,732	7,970	9,000	11,800			
21	2,3-pentanedione*	butter, caramel	3.06	12.9	13.7	18.0	$30^{\rm f} - 500,000^{\rm b}$		
22	phenylacetaldehyde	rose, floral	1.63	5.42	4.38	6.04	$4^{k,l} - 9^{b}$		
23	2-phenylacetic acid	floral	1,174	5,150	3,860	5,830	$68^{\rm r} - 6,100^{\rm e}$		
24	2-phenylethanol	floral, rose, bread dough	569	1,880	1,580	3,000	$140^{e} - 1,122^{a, h}$		

Table 2. Orthonasal detection thresholds for 26 aroma compounds in an alcohol-free beer model system, calculated by four different methods.

	25	vanillin	vanilla, caramel	396	1,490	1,040	1,880	$4.9^{j} - 53^{e, s}$
	26	4-vinylphenol	leather, chemical, plastic	665	2,980	2,540	4,020	$10.4^{a} - 78^{j}$
540	*Compo	ounds assessed by 12 panellists.	remaining compounds by 24 panellists	. ^a Buttery, 7	Furnbaugh, & I	Ling (1988), ^b I	Rothe et al. (19	72), ^c Larsen & Poll,
541	(1992),	^d Schnabel, Belitz, & von Ranse	on (1988), ^e Czerny et al. (2008), ^f Blank,	Sen, & Gro	osch (1991), ^g	Guth & Grosch	(1994), ^h Butte	ry, Teranishi, Flath, &
542	Ling (1	990), ⁱ Semmelroch, Laskawy, I	Blank, & Grosch (1995), ^j Langos et al. (2013), ^k But	tery, Seifert, C	uadagni, & Li	ng (1971), ¹ Gua	adagni, Buttery, &
543	Turnba	ugh (1972), ^m Buttery, Guadagn	i, Ling, Seifert, & Lipton (1976), "Guad	agni et al. (1963), °Amoo	re, Venstrom, a	& Davis (1968)), ^p Baldwin, Scott,
544	Shewm	aker, & Schuch (2000), ^q Karah	adian, Josephson, & Lindsay (1985), ^r W	agner, Gra	nvogl, & Schie	berle (2016), ^s	Sellami, Mall,	& Schieberle (2018).

545 Full references in Appendix B.

	Compound	Odour quality	Retronasal detection threshold, µg/L						
No.			Logistic	regression	BET		Threshold range in literature, μg/L		
			Raw	Adjusted	Raw	Adjusted	In water	In beer	
2	acetic acid	vinegar	22,100	60,000	68,600	104,000	54,000 ^a	175,000 ^h	
3	2,3-butanedione	butter, dairy	0.19	0.74	1.30	1.64	$0.2^{b} - 5^{c}$	$17^{i} - 150^{h}$	
4	butanoic acid	cheese	255	575	462	666	6,800 ^a	2,200 ^h	
5	dimethyl sulfide*	sweet, vegetable, savoury	39.3	74.8	56.7	81.7		50 ^h	
6	5-ethyl-4-hydroxy-2- methyl-3(2 <i>H</i>)-furanone (homofuraneol)	candy floss, caramel	27.9	134	131	238			
8	3-hydroxy-4,5-dimethyl- 2(5H)-furanone (sotolone)*	curry, molasses	1.24	3.59	4.41	5.80			
9	4-hydroxy-2,5-dimethyl- 3(2 <i>H</i>)-furanone (furaneol)	candy floss, strawberry	81.5	270	190	300	30 ^d		
10	methional	boiled potato, metallic	0.040	0.73	1.12	1.78	0.04 ^{c, e}	$4.2^{i} - 250^{i}$	
12	2-methoxy-4-methylphenol	smoky, bacon, vanilla	0.079	1.86	4.65	5.85			
13	2-methoxyphenol	vanilla, smoky	0.42	0.99	1.21	1.91	0.75 ^e		
14	2-methoxy-4-vinylphenol	cloves, medicinal, bacon	1.90	8.33	24.2	30.4		300 ^h	
15	2-methylbutanal	fruity, sweet, cheesy	1.57	8.99	15.5	22.3	$0.03^{\mathrm{b}}-40^{\mathrm{f}}$	$45^{i} - 1,250$	
16	3-methylbutanal	nutty, cheesy	0.22	0.44	0.56	0.74	$0.04^{b} - 60^{f}$	600^{h}	
18	3-methyl-1-butanol	banana, cheese, fermented	128	262	220	303	4,750 ^f	70,000 ^h	
19	methylpropanal	chocolate	0.16	0.86	1.65	2.17	$0.006^{b} - 180^{f}$	1,000 ^h	
22	phenylacetaldehyde	rose, floral, green	0.10	0.68	1.33	2.11	40^{f}	105 ⁱ – 1,600 ^h	
23	2-phenylacetic acid	floral, metallic, musty	12.6	218	1,290	1,690		2,500 ^h	
24	2-phenylethanol	floral, beer, rose	110	278	579	874	$240^{\rm f}-750^{\rm g}$	40,000 ^j – 125,000 ^h	
25	vanillin	vanilla	45.9	448	754	1,040			
26	4-vinylphenol	chemical, medicinal	90.0	2,340	2,540	4,210			

Table 3. Retronasal detection thresholds for 20 aroma compounds in an alcohol-free beer model system, calculated by four different methods.

- 548 Compounds 1, 7, 11, 17, 20, and 21 in Table 2 were not assessed for retronasal perception. *Compounds assessed by 12 panellists; remaining compounds by
- 549 24 panellists. ^aPatton (1964), ^bRothe & Thomas (1962), ^cMilo & Grosch (1993), ^dPittet, Rittersbacher, & Muralidhara (1970), ^eCerny & Grosch (1993),
- ⁵⁵⁰ ^fSheldon, Lindsay, Libbey, & Morgan (1971), ^gOhloff (1978), ^hMeilgaard (1975), ⁱSaison et al. (2009), ^jEngan (1972). Full references in Appendix B.











