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The application of compound-specific sulfur isotopes to the oil–source rock correlation of Kurdistan petroleum

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ABSTRACT

The concentrations and $\delta^{34}S$ values of thioaromatic compounds of a suite of oils from several major oil fields in Kurdistan and their corresponding regional Type II-S source rocks have been measured to investigate their source relationship. The oils of three fields (Khabbaz, Jambur, Ajeel) and the bitumen extracted from specific rock formations (Alan, Sargelu, Naokelekan, Chia Gara) showed particularly high abundances of thioaromatics consistent with a carbonate source deposited in a restricted sulfate-rich marine platform setting. The δ^{34} S [V-CDT] values of the major organosulfur compounds (OSCs) in these petroleum samples were measured with a gas chromatograph coupled to a multi-collector inductively coupled plasma mass spectrometer. δ^{34} S values of dibenzothiophenes and methyldibenzothiophenes were consistently in the range -4% to -12% and -9% to -18% for the oils and rocks, respectively. Separate groupings of oils and rocks were distinguishable by > 2%difference, given an analytical reproducibility of < 0.8%. OSCs from rocks were consistently $\sim 2-4\%$ depleted than in oils, reflecting a similar trend to previous bulk δ^{34} S studies from which an initial evolution of ³⁴S depleted H₂S during diagenesis and thermal maturation had been proposed. Distinctive $\delta^{34}S_{OSC}$ data of the oils and rocks with particularly high thioaromatic abundances did suggest several oils-source rock relationships: the Ajeel and Jambur oils and sediments from the Chia Gara formation yielded relatively enriched $\delta^{34}S_{OSC}$ values, whereas consistently depleted $\delta^{34}S_{OSC}$ values were observed for the Khabbaz oil and Naokelekan source rocks. Results

suggest that compound-specific S isotope analysis can help establish oil–source rock relationships of S-rich petroleum.

Keywords: Benzothiophenes; dibenzothiophenes; Type II-S kerogen; Chia Gara, Naokelekan and Sargelu formations; Kirkuk, Bai Hassan, Khabbaz, Ajeel oil fields.

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1. Introduction

The autonomous Kurdistan region of the Middle East is host to one of the world's largest active petroleum provinces. Exploration and field development in this region still affords tremendous oil exploration opportunities and continues to intensify.

Two major petroleum systems occur in Mesozoic strata: (1) the Middle–Upper Jurassic; and (2) the basal Cretaceous (Al-Ameri and Zumberge, 2012; Mohialdeen et al., 2013; Mohialdeen and Hakimi, 2016). Kurdistan oil fields in northern Iraq are located in the Zagros Fold Belt (of the Arabian platform), where the Jurassic and Cretaceous petroleum systems are generally well separated by the widespread carbonates/anhydrites of the Gotnia Formation. However, an absence of this unit or its replacement by the siliciclastic or brecciated Barsarin Formation can complicate the differentiation of the two petroleum systems and their respective contribution to many of the Kurdistan oils remains unclear.

Aliphatic terpenoid biomarkers indicative of deposition in Jurassic/Cretaceous shallow seas of the Arabian Carbonate Platform under restricted water circulation (e.g., high abundances of $C_{28} \alpha\beta$ -hopane, 29,30-dinorhopane and extended hopanes, enrichment of the C_{24} tetracyclic terpenoid relative to tricyclics and low abundances of extended tricyclics) were recently reported in a suite of Kurdistan oils (Mohammed and Schwark, 2013). Strong anoxic/euxinic conditions were developed in basins with sulfur-rich marl to carbonate-anhydrite lithologies. Many of the oils had a high S content, being derived from Type II-S kerogens deposited in an anoxic, sulfur-rich

and iron-deficient depositional environment. The S-rich oils typically contained relatively high abundances of OSCs, the distributions or δ^{34} S values of which may help in supporting oil–source rock correlations, identify oil families or improve oil–reservoir characterisation.

The δ^{34} S values of individual OSCs in oils or bitumen can now be measured with gas chromatography combined with a multi-collector inductively coupled plasma mass spectrometer (GC–ICPMS; Amrani et al., 2009). The δ^{34} S values of thioaromatic and thioadamantane compounds have been measured in a range of oils of different ages and petroleum provinces (Amrani et al., 2009, 2012; Greenwood et al., 2014; 2015; Cai et al., 2015, 2016; Gvirtzman et al., 2015; Li et al., 2015; Ellis et al., 2017). The δ^{34} S of the OSCs within and between oils can span a wide range, with values as low as -21‰ detected in Smackover oils (Gulf of Mexico; Gvirtzman et al., 2015) and > +40% in thermochemical sulfate reduction (TSR) impacted Tarim Basin oils (NE China; Cai et al., 2015, 2016; Greenwood et al., 2015; Li et al., 2015). This relatively large variance can be attributed to the variable sensitivity and isotopic fractionation of different OSCs to the different S-processes impacting petroleum systems, particularly TSR or microbial sulfate reduction (MSR), as well as potentially petroleum migration or alteration (e.g., thermal maturation, water washing, biodegradation). A slightly narrow range of δ^{34} S values measured from bulk oils (e.g., -8 to +32%; Faure and Mensing, 2005) have proved useful for oil-oil correlations (e.g., Gaffney et al., 1980; Thode, 1981; Orr, 1986; Cai et al., 2009a,b).

Here, we have measured the δ^{34} S values of organic sulfur compounds (δ^{34} S_{OSCs}) in a suite of S-rich oils from several commercial Kurdistan oil fields and a series of potential source rocks, and integrated this with other traditional organic geochemical data, to help: (i) identify the major source rocks responsible for the Kurdistan oils; and (ii) investigate the relationship between the δ^{34} S values of OSCs from oils and source rocks.

Prospective mid Jurassic to lower Cretaceous source rocks of the Kurdistan oils occur in rock intervals of the Chia Gara (Upper Jurassic-Lower Cretaceous), Naokelekan (Upper Jurassic) and Sargelu (Middle Jurassic) formations (Dunnington, 2005; Al-Ameri et al., 2014; Mohialdeen and Hakimi, 2016), but their relative source significance has not been fully resolved. Hydrocarbon biomarkers including hopanoids (including gammacerane), steroids, and tricyclic terpenoids detected in Chia Gara rocks were consistent with largely marine algal and microbial inputs deposited in anoxic and saline stratified bottom waters (Mohammed, 2016; Mohialdeen and Hakimi, 2016). Hydrocarbon biomarkers have also revealed several source rock-oil reservoir relationships: e.g., high C27 diasterane abundances and elevated Ts/(Ts+Tm) ratios indicated that the Miocene reservoir units in the area were charged by the Jurassic petroleum system (Mohammed and Schwark, 2013). However, most of the rocks analysed in this study showed very similar terpenoid biomarker compositions, due to comparable depositional settings on a shallow carbonate platform, that limited a more detailed oil-source rock correlation.

The source rock–oil correlation potential of compound-specific sulfur isotope analysis (CSSIA) has not yet been fully established although in some recent studies the $\delta^{34}S_{OSC}$ data of: (i) several Tarim Basin oils (NW China) were correlated with Cambrian source rocks (Cai et al., 2015) – although the much speculated Cambrian or Ordovician source of the vast majority of Tarim Basin oils that have been analysed by GC–ICPMS (Cai et al. 2015, 2016; Li et al 2015) could not be distinguished due to complications such as mixing or alteration by thermochemical sulfate reduction (TSR); and (ii) bitumen produced by the pyrolysis of Ghareb Formation rock was correlated with adjacent Late Cretaceous organic chalk sourced oils (e.g., Gurim, Emuna; Rosenberg et al., 2017).

2. Methods

2.1. Geological setting

2.1.1. Stratigraphy and lithology of potential source rock and oil accumulations

Sedimentary rocks deposited on the Arabian Platform during the Late Jurassic to Early Cretaceous comprise carbonate and anhydrite minerals and carbonaceous shales, which accumulated organic matter deposited under anoxic and sulfur-enriched conditions (Dunnington, 2005; Mohialdeen and Hakimi, 2016). Low iron concentrations during deposition led to the organic incorporation of reduced sulfur into Type II-S kerogens. The stratigraphy of the Jurassic/Cretaceous formations of the study area with source potential is indicated in Fig. 1.

Increases in the thermal maturity of sediments upon formation of the Zagros Fold Belt led to the generation and accumulation of large oil reservoirs, including the giant Kirkuk field. A map of the Kurdistan oil province and major oil fields is given in Fig. 2.

2.1.2. Source rocks stratigraphy and potential

Source rocks recovered from the Ajeel-12 well covering the time interval from the Lower Middle Jurassic (Aalenian) to the Early Cretaceous (Barremian), were made available from a previous study (Mohammed and Schwark, 2013; Mohammed, 2016). Prior high resolution (117 samples) Rock-Eval (RE) pyrolysis of Ajeel-12 rocks was acquired for a detailed appraisal of the petroleum generating potential of the Jurassic–Cretaceous strata and selected data is included in Fig. 1. High TOC and HI values, combined with T_{max} values largely within the oil window identified substantial source rock potential in Sargelu, Naokelekan and Chia Gara Formations, with possible further potential in transitional zones of the evaporitic Gotnia Formation (Fig. 1). A suite of 18 rock samples (Table 1) covering the six main formations of Ajeel-12 well were selected for the current CSSIA study.

2.1.3. Kurdistan production oils

A total of 29 production oils (Table 2) from the Kirkuk supergiant oil field and the adjacent south-west oil fields of Bai Hassan, Khabbaz, Jambur, Hamrin, Ajeel, Tikrit and Baji were analysed. These oils were all produced from Miocene (Euphrates,

Jeribe), Oligocene (Baba) and Eocene (Khurmala) reservoirs and largely charged from Jurassic petroleum systems.

2.1.4. Sample preparation, GC-MS and CSSIA

Mono- and polyaromatic fractions of all oils and aromatic fractions of the rock samples were made available from a previous study (Mohammed and Schwark, 2013; Mohammed, 2016). Briefly, bitumen had been recovered from potential source rocks by accelerated solvent extraction (DIONEX ASE-200) applying an azeotropic mixture (93:7) of dichloromethane (DCM) and methanol (MeOH) at 50 bars and 75 °C over a static phase of 25 min. Before further compound class separation, elemental sulfur from extracts and oils was removed by adding activated copper. Oils and bitumen extracts were separated into aliphatic, mono- and polyaromatic, or NSO fractions by automated SPE using an LC-TECH Freestyle robot. Activated silica (2.8 g, 25-40 µm) was dry filled into 8 ml SPE columns and overlain with 20 mg of oil/extract sorbed onto a minimum amount of silica gel. Aliphatic hydrocarbons were eluted with 7 ml of *n*-hexane, monoaromatics with a further 3 ml of hexane, polyaromatic hydrocarbons with 6 ml hexane/dichloromethane (3:2, v:v) and NSO compounds were eluted with 12 ml of DCM:MeOH (1:1, v:v). GC-MS measurements were performed on an Agilent 5975B MSD interfaced to an Agilent 7890A gas chromatograph (GC) equipped with a DB-5MS capillary column (30 m length, 0.25 mm i.d., 0.25 µm film thickness). The temperature program used was: 70 °C (5 min isothermal) to 140 °C at 10 °C/min, then to 325 °C at 2 °C/min (held for 10 min). The quadrupole mass

spectrometer was operated in scan mode in the m/z 50 to 550 range. Compounds presented and discussed here (Pr, Ph, n-C₁₇, n-C₁₈, dibenzothiophenes/phenanthrene and methylated analogues) were identified via characteristic mass spectra and relative retention times, and peak areas, used to calculate product ratios, were integrated using the GC–MSD ChemStation Software (Agilent Technologies).

It was anticipated the thioaromatics (TAs) would concentrate in the polyaromatic fraction, but this was found not to be the case for all samples. Consequently, the mono- or polyaromatic fraction of each sample which showed the highest GC–MS concentration of thioaromatics was used for GC–ICPMS analysis. Analysis of OSCs common to the mono- and polyaromatic fractions of selected samples showed negligible δ^{34} S variance (< 1‰), indicating no obvious ³⁴S fractionation between these two sub-fractions.

The δ^{34} S values of OSCs in the mono- or polyaromatic fractions of the oils and rocks were measured with a Thermo Neptune Plus multi-collector ICPMS coupled to an Agilent 6890 GC inlet system (GC–ICPMS; Greenwood et al., 2015). The GC was fitted with a DB-5 capillary column (30 m × 0.25 mm i.d. × 0.1 µm film thickness). The oven was temperature programmed to increase from 80 °C (held 0.5 min) to 280 °C at 8 °C/min and He carrier gas was used at a rate of 5 ml/min. Pulses of SF₆ reference gas (–8.6‰) were introduced at the beginning and end of each GC run to facilitate the anchor point δ^{34} S calibration of the acquired data.

The m/z 32 (³²S) and m/z 34 (³⁴S) ICPMS ion currents were acquired in 131 ms increments by the Neptune software and exported to Excel in ASCII format. Peak

areas from background subtracted signals were manually processed in Excel to obtain the ${}^{34}S/{}^{32}S$ analyte ratios (Ra) used to calculate $\delta^{34}S$ values:

 δ^{34} S = Ra/(Rs–1), where Rs = 34 S/ 32 S Standard (Vienna Canyon Diablo Troilite = 0.044151)

The ICPMS-measured δ^{34} S values were converted to true δ^{34} S with a normalisation curve established from the measurement of authentic standards of known δ^{34} S values (benzothiophene (BT) = +4.5‰; dibenzothiophene (DBT) = +13.7‰; 1-dodecanthiol (C₁₂SH) = -7.1‰; 1-octadecaenthiol (C₁₈SH) = +13.7‰) measured by Elemental Analysis isotope ratio mass spectrometry calibrated for δ^{34} S measurements with international standards provided by IAEA (IAEA-S1, IAEA-S2, IAEA-S3 and NBS127). External standards were also analysed daily to confirm the accuracy of the ICPMS measurements (precision typically < 0.5%). The δ^{34} S values reported for the OSCs of all samples are the mean of at least duplicate GC–ICPMS measurement typically with standard deviation values typically < 0.8‰, but larger for products of low concentration or those that were only partially GC resolved (i.e., some dimethylated and trimethylated DBT isomers).

Methyl DBT isomers were identified from previously reported GC elution data (Garcia et al., 2002). Isomeric identification of higher molecular weight alkyl DBTs (and alkyl BTs) was not attempted and these analytes are simply referred to as isomers i, ii or iii (i.e., δ^{34} S data was measured for up to 3 different isomers of some OSCs).

3. Results

3.1. Thioaromatic composition of oils and rocks

The m/z 32 chromatogram from GC–ICPMS analysis of the polyaromatic fractions of Ajeel oil sample 4407 and rock sample 4272 from the Chia Gara formation are shown in Fig. 3. The most abundant OSCs were alkyl-DBTs, whilst several alkyl-BTs were also evident. These thioaromatic profiles were generally representative of all the oils and rocks analysed although alkyl BTs were often absent or present in low concentrations presumably due to evaporative losses during sample preparation.

Several oils (Khabbaz, Jambur, Ajeel) and rock formations (Alan, Sargelu, Naokelekan, Chia Gara) showed particularly high abundances of thioaromatics consistent with a carbonate source deposited in sulfate-rich marine lacustrine setting. Other less S-rich oils (i.e., Bai Hassan, some Kirkuk) may have originated from mixed carbonate and shale sources deposited in a sulfate deficient environment.

3.2. $\delta^{34}S$ of thioaromatics in oils and rocks

 δ^{34} S values were measured for the most abundant OSCs in the oils and rock extracts. This consistently included DBT and the first (4-) and co-eluting (2/3-) methyl DBT peaks, three dimethyl DBT peaks (i, ii, iii) and an early eluting trimethyl DBT peak (i). The δ^{34} S value of other OSCs could be measured for some samples (e.g., 1-methyl DBT in several oils and rocks and alkyl BTs in most of the oils and a few rocks), but their low concentrations in many samples limited these data so they have been excluded from the detailed δ^{34} S oil source rock assessment.

The δ^{34} S values measured for the OSC in the production oils ranged from -12% to +3% (Table 2) and in the rocks from -18% to -3% (Table 1). The rocks showed consistently depleted δ^{34} S_{OSC} values than the oils.

The OSCs within a given sample had very similar δ^{34} S values apart from a common depletion of a few ‰ with increasing methylation and molecular weight. There was relatively little variance in the data from specific production fields or rock formations. These data trends were evident from the δ^{34} S_{OSC} profiles of oil samples shown in Fig. 4. Production oil data, representing the averaged value from all oils of a given field, are compared in Fig 4a and δ^{34} S_{OSC} data of the three Jambur oils and the five Khabbaz oils are separately shown in Fig. 4b. The δ^{34} S_{OSC} values of the Jambur oils were consistently heavier than the Khabbaz oils (Fig. 4b), and the average δ^{34} S_{OSC} values within each of these two field spanned a range of ~3‰ (Fig 4a).

4. Discussion

4.1. $\delta^{34}S$ character of oils and rocks

The δ^{34} S values of the BTs and DBTs of the production oils varied from -12%to +3%, with individual compounds showing very narrow value ranges. These values were similar to the -3.5% to -8.8% bulk δ^{34} S data of oils from several Kurdistan fields, including Jambur, Khabbaz and Bai Hussain, measured several decades ago (Thode and Munster, 1970) and considered remarkably close due to common Jurassic sources. The slightly larger range in the presently measured $\delta^{34}S_{OSC}$ may reflect the varied sensitivity of different OSCs to elements of the S-cycle of petroleum systems.

Alternatively, the oils may derive from two or more isotopically distinct sources to which different OSCs have a varied sensitivity. The unique source responses of compounds may be obscured by bulk δ^{34} S analyses. The bulk δ^{34} S analysis of kerogens or rocks might also be compromised by the persistence of inorganic sulfur species, such as pyrite which when finely dispersed in kerogen is difficult to fully remove by chemical treatment (Cai et al., 2009). This problem is avoided with the analyses of GC separated OSCs, although the solvent-extractable (i.e., bitumen) fractions of rocks analysed here might represent a quantitatively minor, and potentially isotopically unique, pool of organic S present in these samples.

The δ^{34} S values for the presently measured OSCs and previously of bulk oils previously are close to -5% which represents an approximate 20‰ depletion from Jurassic–Cretaceous seawater sulfate values predicted from the analysis of sulfate minerals (Thode and Munster, 1970; Wu et al., 2014). This difference suggests minimal impact from TSR which would be expected to produce OSCs approaching the heavier δ^{34} S of mineral SO₄^{2–} sources (Amrani et al., 2012). Kurdistan oils have generally shown little evidence of TSR and the modest thermal maturity of the present oil suite (C₂₉ $\beta\beta/(\beta\beta+\alpha\alpha)$ sterane ~ 0.54) suggests the high temperatures necessary to support TSR may not have been reached. Amrani et al. (2012) demonstrated the varied TSR stability and δ^{34} S sensitivity of BTs and DBTs and proposed that the δ^{34} S differential between these thioaromatic groups can often be a sensitive indicator of the early stages of TSR. There was a small but discernible distinction between δ^{34} S_{DBT} and more limited δ^{34} S_{BT} values of the rocks (Table 1) and oils (Table 2): The

 $\delta^{34}S_{BT-DBT}$ of the oils was ~4‰ and for the rocks the difference was slightly higher at ~8‰. The moderate magnitude of these $\delta^{34}S_{BT-DBT}$ values is not indicative of the occurrence of TSR. A recent CSSIA study of a suite of Bighorn Basin oils impacted to varying degrees by TSR was unable to attribute a TSR impact to oils with $\delta^{34}S_{BT-DBT} \le$ 6‰ (Ellis et al., 2017). It was intriguing to note the consistently greater $\delta^{34}S_{BT-DBT}$ values (by few ‰) of the Kurdistan rocks compared to the Kurdistan oils. Under favorable conditions TSR may occur in situ within rocks, but other factors (e.g., thermal maturity; Ellis et al., 2017) can also contribute to $\delta^{34}S$ differences between BTs and DBTs.

The average oil field δ^{34} S values of DBT and specific alky DBTs reflected a variance of ~4‰ between oil fields with enriched to depleted being: Jambur~Ajeel > Hamrin > Kirkuk ~ Bai Hassan > Khabbaz (Fig. 4a). δ^{34} S_{OSC} differences between samples is not likely due to thermal maturity since all of the oils studied were of similar maturity, based on consistent C₂₉ $\beta\beta/(\beta\beta+\alpha\alpha)$ sterane values (0.54 ± 0.03). Very narrow δ^{34} S_{OSC} ranges were measured for the production oils of a given field. The Khabbaz oil showed the largest variance of ~4‰ for most OSCs (Fig. 4b). Thode and Monster (1970) also reported negligible differences in δ^{34} S values of oils from the same field, even for a range of ages, but did note a small variation in δ^{34} S of similarly aged oils of different fields. Local conditions (e.g., SO₄^{2–} and organic reactant concentrations, temperature) can affect metabolic rates and the degree of isotopic fractionation associated with microbial sulfate reduction. The organic incorporation of reduced sulfur species is also typically associated with a small isotopic fractionation

(usually an enrichment; e.g., Amrani and Aizenshtat, 2004), which can vary for different diagenetic pathways. The δ^{34} S of the organic sulfur in petroleum can be subsequently influenced by thermal maturity (Ellis et al., 2017; Rosenberg et al., 2017) as well as potentially by other common geophysical or chemical impacts on petroleum.

The $\delta^{34}S_{OSC}$ profiles of all of the rock extracts were also generally similar, although an up to 8‰ range in values was evident within some formations (e.g., Gotnia). The span in $\delta^{34}S_{OSC}$ values of Chia Gara was 6‰ and for Naokelekan it was 4‰ (Table 1).

A plot of the δ^{34} S value of the first eluting methyl DBT and dimethyl DBT isomers for all oils and rocks is shown in Fig. 5. These were typically the most abundant isomers and gave the most precise δ^{34} S values (std. dev. < 0.6, i.e. smaller than the size of data points given in Fig. 5).

This consideration of just two of the available δ^{34} S values provided a good separation of most oil fields and rock formations demonstrating the sample correlation capacity of CSSIA. For example, the Khabbaz oils generally had the most depleted δ^{34} S values of the oils, and the Ajeel and Jambur oils generally the most enriched. Of the rocks, Chia Gara samples (apart from rock 4271) were heaviest and generally ~2–4‰ more enriched in ³⁴S than the Naokelekan extracts.

The enriched nature of the $\delta^{34}S_{OSC}$ values of the rock extracts (vs oils) is also clearly evident in Fig. 5. A similar trend has been reported from bulk $\delta^{34}S$ analysis of oils and source rocks. Monterey oils were reported to be 2‰ enriched than the

kerogen of their source rocks (Orr, 1986), and the oil expelled from laboratory conducted pyrolysis of immature kerogen showed a similar ³⁴S enrichment compared to the residual kerogen (Idiz et al., 1990; Amrani et al., 2005). These trends were attributed to the initial release of ³⁴S depleted H₂S on the thermal cracking of S–S and C-S kerogen bonds. This leads to residual kerogen being slightly enriched in ³⁴S from which oil was subsequently expelled at higher maturity. Alternatively, ³⁴S depleted H₂S might be formed via the secondary cracking of higher MW OSCs in the bitumen, as was suggested to account for ³⁴S enriched OSCs produced by the hydrogen pyrolysis (HyPy) treatment of several S-rich kerogens (Grotheer et al., 2017). S isotopic trends evident from a simulated thermal maturity study of Type II-S source rocks (Rosenberg et al., 2017) included the evolution of H₂S that was a few ‰ depleted in δ^{34} S than the immature kerogen, consistent with a kinetic control favouring release of ³²S, although the residual kerogen and oil released showed only a minor ³⁴S enrichment (δ^{34} S increase of +0.4‰). Furthermore, the kerogen S and oil phase OSCs showed a δ^{34} S homogenisation with increasing thermal maturity due to formation of new products from multiple S sources, thus becoming more reflective of the δ^{34} S of the bulk kerogen (Rosenberg et al., 2017).

4.2. Correlation of oils and rocks

The depositional environments and lithologies of petroleum source rocks can be inferred from a plot of DBT/phenanthrene (P) vs Pr/Ph (Hughes et al., 1995). These plots for the production oils and source rocks are shown in Fig. 6.

The production oils show two separate groupings in zones 1b and 2 as defined by Hughes et al. (1990), whereas most of the rocks plot in Zone 1b. Zone 1b is indicative of a carbonate or mixed source deposited in a sulfate-rich marine or lacustrine setting, whereas Zone 2 petroleum could derive from a number of possible sources deposited in a less sulfate-rich environment (Hughes et al., 1995). All of the Bai Hassan oils and several of the Kirkuk oils were categorised as moderate sulfate Zone 2 oils. The DBT/P and methyl DBT/methyl P plot (Fig. 6b; ratio values given in Table 2) provides a quite distinctive separation of these oils on the basis of their OSC content and reiterates the lower S content of the Bai Hassan oils. There was less resolution, however, of the other production oils. A subgroup of the Kirkuk oils and one Hamrin oil were also distinguished as being S-poor. All of the Ajeel, Khabbaz and Jambur oils, and all but a few rocks, gave relatively high abundance of thioaromatic compounds.

Triterpenoid biomarker profiles from GC–MS of the saturate fractions of three potential source units were previously shown to match well with the Ajeel and Hamrin oils (Mohammed and Schwark, 2013; Mohammed, 2016). This correlation included a consistently low abundance of tri- vs pentacyclic terpanes, a predominance of the C_{29} hopanes, the presence of 2-methylhopane but no gammacerane, and the peculiar occurrence of a 29,30-dinorhopane.

Sample resolution can be enhanced by considering a combination of $\delta^{34}S$ data from CSSIA and molecular data from GC–MS. This is demonstrated by plotting the $\delta^{34}S$ value of the first eluting dimethyl DBT ($\delta^{34}S_{dmDBTi}$) vs the DBT/P ratio of the

production oils (Fig. 7). The $\delta^{34}S_{dmDBTi}$ data help to resolve some of the S-rich oils, specifically distinguishing several of the Khabbaz oils from the Jambur, Ajeel and Kirkuk oils that had similar thioaromatic concentrations (e.g., Fig. 6b).

Several oils (Khabbaz, Ajeel and Jambur) and rocks (Chia Gara and Naokelekan) all plotted in Zone 1b (Fig. 6) consistent with a close molecular correlation, and potential oil–source rock relationship. δ^{34} S data may extend this correlation and help resolve further the contribution of the Chia Gara and Naokelekan rocks to these oil fields.

The δ^{34} S values of DBT and the first eluting of the methyl-, dimethyl- and trimethyl-DBTs for these three oils and two rocks are shown in Fig. 8. The Jambur and Ajeel oil showed almost identical δ^{34} S_{OSC} data and were consistently 2‰ enriched than the OSCs of the Khabbaz oil. The OSCs in Chia Gara rocks were similarly enriched by 3‰ compared to those in the Naokelekan rock. These trends suggest a greater OSC contribution of the S-isotopically enriched Chia Gara rocks to the enriched Jambur and Ajeel oils, and likewise the S-isotopically depleted Naokelekan rocks may have been more significant for the depleted Khabbaz oils.

4. Conclusions

The incorporation of $\delta^{34}S_{OSC}$ data measured by CSSIA with traditional molecular S indicators (e.g., DBT/P, methyl DBT/methyl P) enhanced the distinction of hydrocarbons from different Kurdistan oil fields and demonstrated how CSSIA can assist the organic geochemical characterization of S-rich oils. The OSCs extracted from rock formations thought responsible for large Kurdistan oil reservoirs were consistently ³⁴S depleted compared to the oils, making direct $\delta^{34}S$ correlations difficult. Nevertheless, similar trends such as relatively ³⁴S enriched oils and rocks allowed several potential relationships to be inferred. For instance, isotopically depleted OSCs from Ajeel and Jambur oils may have received a major contribution from Chia Gara rocks given the notable ³²S depletion of its products, and conversely relatively enriched $\delta^{34}S_{OSC}$ data suggest the Khabbaz oils may be predominantly sourced from Naokelekan rocks. The veracity of these inferences drawn here will need to be assessed by further CSSIA studies of oils and rocks.

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Table Captions

Table 1. List of source rock samples (from well Ajeel-12) and their: (A) molecular and (B) δ^{34} S data.

Table 2. List of production oils and their measured: (A) molecular and (B) δ^{34} S data.

Figure Captions

Fig. 1. Jurassic basal Cretaceous stratigraphy of the Arabian Platform; with TOC, HI and T_{max} data from Rock Eval pyrolysis of sediment samples (data from Mohammed and Schwark, 2013).

Fig. 2. Location of major oil fields in study with inset showing geographic map of Iraq.

Fig. 3. Thioaromatic product distributions evident in the m/z 32 chromatogram from GC–ICPMS analysis of Ajeel oil 4407 (top) and Chia Gara formation rock 4272 (bottom). Separated dimethyl DBT and trimethyl DBT isomers are identified as i, ii, or iii.

Fig. 4. $\delta^{34}S_{OSC}$ profiles of: (A) different production oil fields. i.e., averaged values from all oils (number of oils given in parenthesis) of specific field; (B) three Jambur oils and five Khabbaz oils.

Fig. 5. δ^{34} S plots of first eluting methyl DBT (4m-) and dimethyl DBT (-i) for the production oils (squares) and rock extracts (circles).

Fig. 6. Plots of: (A) DBT/P vs Pr/Ph for the production oils (left) and rock extracts (right), inferring different depositional environment and lithologies of source rocks (after Hughes et al., 1995); and (B) methyl DBT/methyl P vs DBT/P of production oils, distinguishing oils on basis of their OSC content.

Fig. 7. Plot of δ^{34} S dimethyl DBTi values vs DBT/P ratios for the production oils. Fig. 8. Average δ^{34} S of selected OSCs from three production oils and two candidate source rocks characterised as having very similar aliphatic biomarker and thioaromatic compositions. The number of oil or rock samples represented by average values are indicated in parenthesis.

R

Well										
AJ-12	Formation		Molecular Data							
Sample		Pr/Ph	$Pr/n-C_{17}$	<i>Ph/n-C</i> ₁₈	DBT/P	mDBT/mP				
4246	L. Sarmond	0.83	0.31	0.37	0.34	1.91				
4271	Chia Gara	0.65	0.22	0.44	1.59	2.02				
4272	Chia Gara	0.75	0.21	0.44	1.99	2.43				
4350	Chia Gara	0.42	0.24	0.55	1.95	2.34				
4360	Chia Gara	1.05	0.31	0.30	1.84	2.29				
4312	Gotnia	0.33	0.20	0.38	7.89	9.91				
4313	Gotnia	0.40	0.15	0.32	7.74	9.16				
6946	Gotnia	0.54	0.22	0.36	1.37	2.55				
6947	Gotnia	0.44	0.19	0.33	1.46	2.46				
4142	Naokelekan	0.62	0.09	0.14	1.93	3.10				
4318	Naokelekan	0.75	0.17	0.26	1.58	3.01				
4319	Naokelekan	0.83	0.19	0.28	1.61	3.03				
4370	Naokelekan	1.43	0.31	0.25	1.66	3.00				
4281	Sargelu	0.72	0.19	0.28	0.82	2.27				
4375	Sargelu	0.67	0.19	0.33	1.15	3.06				
4377	Sargelu	0.44	0.19	0.32	1.20	2.49				
4288	Alan	0.59	0.26	0.37	1.18	2.58				
4289	Alan	0.69	0.23	0.34	1.21	4.53				

Table 1. List of source rock samples (from well Ajeel–12) and their measured molecular (1A) and δ^{34} S (1B) data. **1A:**

Pr = pristane, Ph = phytane, n-C₁₇ = n-heptadecane, n-C₁₈ = n-octadecane, DBT = dibenzothiophene,

P = phenanthrene, mDBT = methyl dibenzothiophene, mP = methyl phenanthrene

ACC

ID;		1										
Wel							δ^{34} S d	ata (‰)				
1												
AJ–	Forma											
12	tion		1	1	1	1	1	1	1	1		
Sam		mВ	mВ	dm	dmB	DB	4mD	2/3m	dmD	dmD	dmD	tmD
ple		Ti	Tii	BTi	Tii	Т	BT	DBT	BTi	BTii	BTiii	BTi
	L.	-	_	-	-							
424	Sarmo					-9.	-12.		-14.			
6	nd					7	1	-10.3	6	-12.8	-13.1	
427	Chia	-2.	-2.	-3.		-1	-15.		-16.			
1	Gara	9	7	6	-4.8	2.8	1	-14.2	1	-16.6	-15.5	
427	Chia	-3.	-4.	-4.		-1	-11.		-12.	0		-11.
2	Gara	0	3	5	-6.8	1.9	6	-10.6	5	-11.4	-10.8	2
435	Chia	—	_	-	-	-8.			-12.			-10.
0	Gara					6	-9.9	-10.1	2	-11.7	-11.0	9
436	Chia	_	_	-	-	-1	-11.		-10.			-10.
0	Gara					1.9	6	-11.4	5	-10.6	-10.6	7
431		—	_	-	-	-1	-15.		-15.			-15.
2	Gotnia					4.7	5	-13.8	5	-14.5	-13.6	2
431		_	_	-3.		-1	-15.		-14.			-13.
3	Gotnia			2	-5.9	3.4	1	-13.8	5	-14.0	-14.0	6
694		_	_			-1	-14.		-14.			
6	Gotnia					1.3	0	-13.2	8	-14.8	-14.1	
694		_	_		-	-7.	-11.		-12.			-12.
7	Gotnia					4	1	-9.8	6	-11.7	-10.4	0
414	Naokel	_	-	_	_	-1	-14.		-15.			-15.
2	ekan					7.3	2	-15.6	2	-15.1	-13.5	0
431	Naokel	-5.	-6.	-7.		-1	-14.		-15.			-15.
8	ekan	6	2	6	-9.0	1.9	3	-13.3	5	-14.1	-13.6	2
431	Naokel	-4.	-6.	-7.		-1	-16.		-13.			-12.
9	ekan	5	4	9	-9.2	2.2	4	-15.1	7	-12.9	-11.9	6
437	Naokel	—	_	_	_	-1	-15.		-15.			-13.
0	ekan					5.6	0	-14.2	2	-13.9	-13.6	7
428	Sargel	-3.	-4.	-6.		-1	-12.		-13.			-11.
1	u	0	3	8	-3.4	0.0	3	-11.0	6	-12.6	-11.2	3
437	Sargel	-4.	-6.	-6.		-1	-14.		-14.			
5	u	4	0	5	-7.3	1.3	0	-13.2	8	-14.8	-14.1	
437	Sargel	_	_	_	_	-7.	-11.		-12.			-12.
7	u					4	1	-9.8	6	-11.7	-10.4	0
428		_	_	-4.		-1	-16.		-18.			
8	Alan			2	-7.5	3.7	7	-15.4	4	-17.2	-16.3	
428				_	_	_9	-12		-14		10.0	-12
9	Alan					4	9	_9 7	5	-12.9	-10.8	6

mBTi = methyl benzothiophene (isomer i), dmBT = dimethyl benzothiophene, DBT = dibenzothiophene, mBDT = methyl dibenzothiophene, dmDBT = dimethyl dibenzothiophene, tmDBT = trimethyl dibenzothiophene.

2A:						
Oil	Oil Field			Molecular	Data	
Sample		Pr/Ph	$Pr/n-C_{17}$	<i>Ph/n-C</i> ₁₈	DBT/P	mDBT/mP
5404	Jambur	0.92	0.31	0.37	2.66	3.33
5403	Jambur	0.91	0.28	0.34	2.65	3.30
5402	Jambur	0.96	0.29	0.35	2.52	3.30
5401	Hamrin	1.04	0.37	0.40	0.33	0.52
5400	Hamrin	0.98	0.35	0.42	1.75	2.23
4426	Kirkuk	1.01	0.34	0.38	0.36	0.71
4425	Kirkuk	0.87	0.32	0.41	0.41	0.95
4424	Kirkuk	0.86	0.30	0.38	2.14	2.82
4423	Kirkuk	0.90	0.33	0.38	0.96	1.44
4422	Kirkuk	0.89	0.31	0.38	0.67	1.09
4421	Kirkuk	0.89	0.32	0.40	2.22	2.81
4420	Kirkuk	0.94	0.34	0.39	2.17	2.76
4419	Kirkuk	0.92	0.33	0.41	2.21	2.73
4418	Bai Hassan	0.95	0.31	0.36	0.65	0.99
4417	Bai Hassan	0.93	0.31	0.38	0.74	0.89
4416	Bai Hassan	0.88	0.31	0.39	0.76	1.00
4415	Bai Hassan	0.91	0.32	0.39	1.00	1.06
5407	Bai Hassan	0.82	0.27	0.37	0.72	1.24
5405	Bai Hassan	0.83	0.28	0.38	0.44	0.91
4414	Bai Hassan	1.06	0.35	0.36	0.84	1.17
4413	Khabbaz	0.96	0.35	0.41	2.14	2.83
4412	Khabbaz	0.78	0.22	0.32	2.28	3.33
4411	Khabbaz	1.00	0.37	0.42	2.37	2.81
4410	Khabbaz	1.04	0.36	0.39	2.30	2.70
4409	Khabbaz	0.74	0.21	0.33	2.96	3.65
4408	Ajeel	1.03	0.35	0.38	1.67	2.33
4407	Ajeel	1.01	0.39	0.44	1.69	2.06
5406	Baiji	0.86	0.24	0.31	0.89	1.95
6993	Tikrit	0.91	0.28	0.35	0.79	1.56

Table 2: List of production oils and their measured molecular (2A) and δ^{34} S (2B) data.

Pr = pristane, Ph = phytane, $n-C_{17} = n$ -heptadecane, $n-C_{18} = n$ -octadecane, DBT = dibenzothiophene,

P = phenanthrene, mDBT = methyl dibenzothiophene, mP = methyl phenanthrene

2B:												
Oil	Oil Field						\Box^{34} S d	lata (‰)				2
Sam		mB	mB	dm	dmB	D	4mD	2/3m	dmD	dmD	dmDB	tmD
ple		Ti	Tii	BTi	Tii	BT	BT	DBT	BTi	BTii	Tiii	BTi
5404	Jamb ur	0.3	-1. 1	-1. 4	-1.6	-2 .5	-5.1	-4.4	-6.4	-6.2	-5.2	-7.4
5403	Jamb ur	1	-0. 3	-1	-0.9	-3 .3	-5.1	-3.3	-6.4	-5.5	-4.4	-8.1
5402	Jamb ur	3.4	-0. 3	1.8	-0.5	-3 .3	-5.1	-3.9	-7.2	-3.4	-4.7	-7.1
5401	Hamr in	-0. 5	-2. 5	-2. 4	-3.4	-5 .1	-4.7	-4.9	-6.7	-7.4	-7.2	-7.4
5400	Hamr in	-1. 8	-3. 2	-4. 8	-4.1	-4 .8	-5.1	-4.5	-6	-6.6	-6.6	-6.7
4426	Kirk uk	1.1	-0. 5	-0. 7	-1.4	-4 .4	-5.1	-4.7	-7	-6.7	-6.2	-8.7
4425	Kirk uk	0.2	-1	-1. 5	-2.6	-4 .5	-6.5	-7.1	-7.4	-8.1	-7.5	-8
4424	Kirk uk	-2. 4	-2. 5	-4. 9	-3.2	-5 .3	-7.4	-5.3	-7.9	-6.6	-6.8	-8
4423	Kirk uk	-	-	_	_	-6 .9	-7.3	-5.3	-7.9	-6.5	-5.5	-8.7
4422	Kirk uk	-	_	_	_	-7 .9	-7.2	-6.9	-8.7	-7.7	-7.1	-9.8
4421	Kirk uk	-1. 6	-2	-3. 4	-3	-5 .4	-6.3	-5.1	-7.3	-6.3	-5.3	-8.2
4420	Kirk uk	-0. 8	-1	-3	-1.7	-4 .8	-6.5	-3.8	-7.3	-5.8	-4.9	-8.4
4419	Kirk uk	-1. 4	-1. 5	-3. 1	-2.5	-5 .7	-6.4	-4.9	-7.7	-6.3	-5.2	-8
4418	Bai Hass an	-0. 7	-2. 2	-2. 5	-2.8	-5	-6.7	-6.1	-8.3	-7.4	-7	-8.4
4417	Bai Hass an	-2. 4	-3	-3. 3	-3.5	-5 .8	-7	-6.1	-8.2	-6.6	-7.8	-9.1
4416	Bai Hass	0.2	-1	-1. 9	-2.2	-6 .6	-6.5	-6.5	-7.5	-8.1	-7.7	-8.3

	an											
4415	Bai Hass an	-0. 6	-1. 6	-2. 4	-2.8	-5 .8	-7.1	-6.9	-7.8	-8.3	-7.7	-9.2
5407	Bai Hass an	-1. 6	-2. 1	-3. 3	-3	-5 .8	-7.9	-7.4	-9.3	-7.3	-7.4	-8.1
5405	Bai Hass an	-0. 8	-2. 2	-2. 6	-3.1	-7	-6.4	-5.7	-7.1	-8.1	-8.2	-9.2
4414	Bai Hass an	-0. 9	-1. 4	-2. 2	-2.6	-5 .6	-6.5	-5.7	-7	-7.7	-6.8	-9.3
4413	Khab baz	-0. 4	-2. 6	-3	-3.9	-6 .7	-6.9	-7	-8	-8	-8.5	-8.6
4412	Khab baz	0.4	-2. 6	-2. 5	-2	-3 .4	-4.6	-4.4	-7.3	-7.5	-7.1	-10
4411	Khab baz	1.7	-1. 6	-1. 1	-3	-5 .8	-9.4	-8.8	-11	-10.9	-10	-11. 6
4410	Khab baz	-5. 7	-5. 3	-5. 8	-5.5	-8 .4	-10. 5	-10.1	-11. 2	-10.5	-10	-10. 9
4409	Khab baz	-1	-2. 7	-2. 6	-3.9	-6 .6	-8.6	-6.6	-9.8	-8.4	-7.1	-10. 6
4408	Ajeel	-1. 4	-0. 5	-3. 5	-2.1	-4 .7	-5.6	-3.1	-6.1	-4.9	-3.1	-8.1
4407	Ajeel	1	0.7	0	-1.3	-2 .9	-4.8	-3	-6.3	-5	-4	-7.4
5406	Baiji	-0. 5	-1	-2. 4	-2.8	-6 .9	-8.6	-5.9	-10	-7.1	-6.4	-8.9
6993	Tikrit	2.7	2.1	0.8	0.7	-2 .3	-4	-2.5	-5.5	-4.7	-5	-6

mBTi = methyl benzothiophene (isomer i), dmBT = dimethyl benzothiophene, DBT = dibenzothiophene, mBDT = methyl dibenzothiophene, dmDBT = dimethyl dibenzothiophene, tmDBT = trimethyl dibenzothiophene.

Highlights

- Organic sulfur compounds (OSCs) were studied in Kurdistan oils and source • rocks
- δ^{34} S values indicate separate oil and rock groups (> 2‰ difference). •
- OSCs from rocks were 2–4‰ depleted than OSCs in oils •
- en or Related oils and rocks were inferred from relative ³⁴S enrichment or depletion. •

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