

# Environmental life cycle feasibility assessment of hydrogen as an automotive fuel in Western Australia

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## ABSTRACT

A life cycle assessment has been undertaken in order to determine the environmental feasibility of hydrogen as an automotive fuel in Western Australia. The criterion for environmental feasibility has been defined as having life cycle impacts equal to or lower than those of petrol. Two hydrogen production methods have been analyzed. The first is steam methane reforming (SMR), which uses natural gas (methane) as a feedstock. The second method analysed is alkaline electrolysis (AE), a mature technology that uses water as a feedstock. The life cycle emissions and impacts were assessed per kilometer of vehicle travel.

Initial results found that hydrogen production under the SMR scenario produced less greenhouse gas, photochemical oxidation and eutrophication emissions per kilometer than petrol. Petrol produced less greenhouse gas and eutrophication emissions than hydrogen produced under the AE scenario, but the only improvement was in the terms of photochemical oxidation emissions. “Hotspot” analysis showed that while the usage life cycle phase of hydrogen produced very few emissions, the reliance on electricity and fossil fuels

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24 during production was responsible for emission levels higher than those from petrol. After  
25 wind-generated electricity was incorporated, the emissions were significantly reduced below  
26 the levels of those from petrol under both SMR and AE scenarios. However, with the  
27 incorporation of wind-generated electricity, the production of hydrogen, particularly from  
28 electrolysis, is more environmentally friendly than the SMR process.

29

30 *Keywords:*

31 Hydrogen fuel

32 Life cycle assessment

33 Environmental feasibility

34

## 35 1. Introduction

36

37 There is a growing necessity for an alternate energy carrier to replace the ever decreasing,  
38 and high emissions generating, supplies of fossil fuels. This is particularly notable in the  
39 transport sector, where the overwhelming majority of vehicles operate on petroleum products  
40 [1]. Considering the enormous environmental, and economic impact of the transport industry,  
41 the introduction of alternative fuels will be key to a sustainable transport sector [2].

42 With petrol as the most common vehicle fuel, the Western Australian transport sector  
43 generates approximately 14% of the state's total greenhouse gas (GHG) emissions. This is  
44 attributable to the heavy reliance on passenger vehicles for most West Australians, coupled  
45 with the sparsely populated landscape and large distances between population centers [3]. In  
46 2007, approximately 78.9% of the total vehicle fleet was registered as using unleaded petrol  
47 and 85.9% of these vehicles were classed as passenger vehicles [4]. With ownership of  
48 private vehicles in Australia on the increase (up 13.1% from 2004 to 2009) [4], transportation  
49 is a major factor in the ever increasing demand for fossil fuels [5], in turn having a significant  
50 effect on the Western Australian environment [6].

51 With the overwhelming majority of Western Australia's vehicles operating on petrol,  
52 environmentally damaging emissions are constantly being introduced into the atmosphere,  
53 resulting in the per capita GHG emissions for Western Australia being significantly higher  
54 than for other Australia states [3]. These passenger vehicles are also the primary emitters of  
55 nitrogen oxides (NO<sub>x</sub>) and volatile organic compounds (VOCs) causing photochemical smog  
56 and negative health impacts [7].

57 Considering the growing atmospheric pollution and the current energy crisis, studies  
58 have been conducted in Australia that assess the environmental feasibility of alternative  
59 transport fuels such as liquefied petroleum gas (LPG), liquefied natural gas (LNG), bio-diesel

60 and ethanol. While the use of these fuels reduces GHG emissions, they can have other  
61 environmental impacts during the combustion stage. For example, ethanol was a potentially  
62 renewable fuel with reduced carbon monoxide (CO) emissions compared to petrol, but the  
63 NO<sub>x</sub> emissions resulting from combustion were significantly higher than those from  
64 petroleum products [8].

65 Alternative fuels may produce relatively less GHGs than conventional fuel during  
66 combustion, but more emissions are produced during the production process. For example, a  
67 study in 2011 by Biswas *et al.* [9] found that biodiesel production and combustion from  
68 canola is not “carbon neutral”, as GHGs are emitted from production of farm inputs and  
69 during crop growth. Similarly, LNG has been considered one of the safest and cleanest fossil  
70 fuels [10, 11, 12, 13, 14, 15] in comparison with other fossil fuels such as coal and oil in  
71 terms of NO<sub>x</sub>, sulphur dioxide (SO<sub>2</sub>) and carbon dioxide (CO<sub>2</sub>) emissions, but the production  
72 and liquefaction of LNG is energy intensive and not free of environmental impacts.  
73 Therefore, a life cycle assessment (LCA) that takes into account emissions from all stages  
74 needs to be conducted to assess the environmental impacts of alternative fuels and to identify  
75 the most polluting processes for applying mitigation strategies.

76 Many alternative fuels have been studied over the years; however, the fuel which appears  
77 to be a more promising alternative is hydrogen due to its clean burning characteristics and  
78 limitless supply. Although research into hydrogen fuel is limited in Australia, a 2003  
79 Australian study identified a number of hydrogen feed stocks suitable for mass production in  
80 Australia. These feed stocks included coal, fuel oils, industrial chemical by-products, coal,  
81 coal seam methane and natural gas [16].

82 One of numerous foreign studies into hydrogen as an automotive fuel, a life cycle  
83 emissions study for hydrogen fuel production found that hydrogen could potentially be  
84 produced with comparatively less emissions than petrol [17]. Similarly, a 2005 Canadian

85 study [18] found that the life cycle emissions from hydrogen could also be comparable to  
86 those of petrol when producing hydrogen from natural gas feed stocks. Other studies have  
87 assessed the viability of hydrogen from alternative production sources and processes [19, 20,  
88 21].

89 Western Australia possesses abundant fossil fuel resources, particularly coal and natural  
90 gas. Black coal accounts for around 49% of total fossil fuel resources within the state, with  
91 natural gas accounting for around 40% and growing as more sources are identified [22]. This  
92 makes reforming of natural gas, or steam methane reforming (SMR), an attractive option for  
93 Western Australia due to its availability in large reserves. While there are available resources  
94 to produce environmentally friendly hydrogen fuel in Western Australia, the upstream  
95 activities, such as feedstock production, processing and storage stages, can have adverse  
96 environmental impacts because of the state's fossil fuel dependent electricity mix and  
97 scattered settlements [8, 22].

98 This study aims to assess the life cycle environmental feasibility of using hydrogen as an  
99 automotive fuel in Western Australia through two commonly used hydrogen production  
100 process (SMR and electrolysis). This study utilizes the functional unit VKT (vehicle  
101 kilometer travelled) in order to assess the well-to-wheel emissions of vehicles per kilometer  
102 of travel, so that there is a common unit of measure between the petrol and hydrogen results.

103 Firstly, the paper discusses the methodology for carrying out the life cycle environmental  
104 feasibility study of hydrogen fuel in Western Australia. Secondly, the life cycle  
105 environmental impact of hydrogen fuel has been compared with that of petrol and the  
106 “hotspot” – the inputs causing the most pollution – is identified. Finally, appropriate  
107 mitigation strategies have been considered for reducing the life cycle environmental impacts  
108 of hydrogen fuel use in passenger transport.

109

## 110 **2. Methodology**

111

112 LCAs model the complex interactions between a product and the environment throughout all  
113 phases of the product's life. The methodology for this LCA of hydrogen as an automotive  
114 fuel has followed the guidelines set out by ISO14040–14043 [23]. The LCA methodology  
115 consists of four steps

- 116 i. goal and scope
- 117 ii. life cycle inventory (LCI) analysis, which provides information on the input data  
118 (chemicals and energy) used to determine the life cycle emissions during each life  
119 cycle phase
- 120 iii. impact assessment, which evaluates the environmental impacts of the emissions of  
121 each life cycle phase and classifies impacts into environmental impact categories  
122 (e.g. global warming)
- 123 iv. Interpretation, which evaluates the LCA model by identifying significant issues  
124 based on the results of LCI and LCA, considering completeness and consistency  
125 and making conclusions and recommendations (as presented in the results and  
126 discussions section of this paper).

127

### 128 **2.1. Goal and scope definition**

129

130 The goal of this life cycle study is to evaluate the environmental feasibility of hydrogen as an  
131 automotive fuel in Western Australia. The study also provides a reasonable comparison of the  
132 life cycle environmental impacts of hydrogen compared to petrol as a vehicle fuel. For the  
133 purposes of this comparative study, the functional unit used is VKT. This allows the  
134 identification and comparison of life cycle impacts between hydrogen and petrol vehicles.

135 Road tests using hydrogen fuel in a Volkswagen Polo (1.4 liter engine) in 2011 gave a  
136 maximum speed of 125 km/h and an estimated consumption of 1 kg of hydrogen per 100 km  
137 at an average speed of 90 km/h [24]. Therefore, the average consumption of hydrogen (0.01  
138 kg hydrogen/VKT) [24] was used as a functional unit. The same model vehicle with the same  
139 engine size consumes 0.059 liters of regular unleaded petrol per kilometer [25]. Using the  
140 density of BP unleaded petrol ( $730 \text{ kg/m}^3$  [26]), the fuel consumption by mass was found to  
141 be 0.043 kg/VKT, where VKT is the functional unit for petrol. It should be noted that  
142 Volkswagen Polo cars are sold in Australia [27], which justifies their use in this case study.

143 The life cycle environmental impacts of the use of 0.01 kg hydrogen have been  
144 compared with 0.043 kg of petrol for driving a passenger car for 1 km.

145 This LCA study considers the well-to-wheel approach, which means that it takes into  
146 account all stages from resource extraction to eventual fuel consumption.

147 Three system scenarios have been assessed within this LCA. The first is the LCA of  
148 hydrogen as an automotive fuel when the hydrogen is produced by SMR. The second  
149 scenario will assess the LCA of hydrogen when the hydrogen is sourced from alkaline  
150 electrolysis (AE). Finally, the third scenario is the LCA of petrol for comparison.

151 The determination of impacts associated with the modification of the existing  
152 Volkswagen engine into a petrol-H<sub>2</sub> engine was beyond the scope of this research.

153

## 154 **2.2. Life cycle inventory analysis**

155

156 LCI is the collection of data that describes the inputs required for each stage of the well-to-  
157 wheel life cycle. The purpose of these inventories is to provide the basis for an assessment of  
158 the environmental impacts of running a vehicle on hydrogen compared to running a vehicle

159 on conventional petrol. Figure 1 presents the life cycle pathways for SMR and AE to produce  
160 the same amount of hydrogen required to drive a passenger vehicle for 1 km.

### 161 *2.2.1. Steam methane reforming scenario*

162

163 The SMR scenario includes seven life cycle stages of well-to-wheel (or production to  
164 combustion), which are as follows:

- 165 1 Natural gas extraction and distribution: this phase takes into account the energy  
166 and resources required to extract and distribute the gas.
- 167 2 SMR: this phase takes into account the natural gas, steam and electricity required  
168 for the process. The SMR process is assumed to occur at 20 bar.
- 169 3 Compression of the hydrogen into large transport trailers: SMR produces  
170 hydrogen gas at pressures of around 20 bar; however, large-scale CP-12 hydrogen  
171 delivery trucks have 12 storage tubes which operate at 165 bar [28]. Therefore a  
172 compressor is used to increase the pressure of the hydrogen to 165 bar for travel  
173 and delivery.
- 174 4 The distribution of hydrogen gas by tanker truck: the CP-12 hydrogen delivery  
175 trailers weigh 42.5 tons and are typically pulled by large diesel trucks. The mean  
176 delivery distance was also calculated based on Western Australia. BP locations  
177 and the average distance were found to be 233 km. This phase takes into account  
178 delivery distance and diesel consumption by a tanker truck.
- 179 5 The compression of the hydrogen into medium-term storage tanks at the fuelling  
180 station: mid-term storage tanks at fuelling stations contain hydrogen at 300 bar to  
181 allow for faster refueling of vehicle tanks [29]. This means that the hydrogen  
182 must again be compressed from 165 bar in the delivery tanker tubes to 300 bar  
183 using an electrical compressor. The energy required to pump petrol into a fuel



184 tank was not considered as it is negligible when compared with the energy  
185 required to compress hydrogen into a vehicle tank.

186 6 The compression of the hydrogen into smaller vehicle fuel tanks: from the 300  
187 bar storage cylinders at the fuelling station, the hydrogen gas needs to be  
188 compressed to 350 bar inside the vehicle fuel tank [30, 31]. Again, this process is  
189 performed by an electrical compressor.

190 7 Hydrogen used by vehicle: the emissions associated with hydrogen combustion  
191 have been sourced from Wallnera *et al.* [32].

192 Table 1 details the inputs and quantities required for production, delivery and  
193 combustion of 0.01 kg of hydrogen gas produced through SMR.

194

#### 195 2.2.2: Alkaline electrolysis scenario

196

197 LCA for the AE scenario includes five life cycle stages of well-to-wheel analysis, which are  
198 as follows:

199 1 Electrolysis process: this phase takes into account the water, electricity and  
200 electrolytes used during the electrolytic process (Table 2). The process used as a basis  
201 for this research operated at 8.14 bar [31].

202 2 Compression of the hydrogen into large transport trailers: compression into the  
203 transport trailer requires more energy when the hydrogen is produced by AE as the  
204 hydrogen gas is produced at a lower pressure than during SMR. This phase takes into  
205 account the electricity required to compress the hydrogen from 8.14 bar to 165 bar for  
206 transport.

207 3 The distribution of hydrogen gas by tanker truck: the distribution method is identical  
208 to when hydrogen is produced by SMR.

209 4 The compression of the hydrogen into medium-term storage tanks at the fuelling  
210 station: as with the SMR scenario, the electricity required to compress the hydrogen  
211 from 300 bar to 350 bar is taken into account in this phase.

212 5 Hydrogen use by vehicle: this is same as for SMR.

213 A separate inventory for petrol has not been developed as the software used has the  
214 emission values of petrol production and use.

215

### 216 **2.3 Life cycle impact assessment**

217

218 The environmental impacts associated with the production and use (combustion) of hydrogen  
219 includes two steps. Firstly, the energy and material flow data provided in the LCI were input  
220 to *Simapro 7.24* software [33] to calculate the environmental impacts of the production and  
221 use of hydrogen fuel. Secondly, the program categorized the emissions for all impact  
222 categories and then converted them to equivalent environmental impacts, including global  
223 warming, photochemical oxidation, eutrophication, carcinogens, land use, water use, solid  
224 waste, embodied energy and mineral depletion impacts.

225 Step 1: The input and output data in the LCI were input to the *Simapro* software to  
226 calculate the emissions for different environmental impact categories due to the use of  
227 hydrogen and petrol per VKT. The input/output data of the LCI were linked to relevant  
228 libraries in *Simapro*. The LCA Library is a database of energy consumption, emissions and  
229 materials data for the production of one unit of an input (e.g. electricity, diesel).

230 This study utilized the Australian LCA libraries [34] developed by RMIT University for  
231 Australian conditions to calculate the emissions associated with the production and use of  
232 inputs. The library for the Western Australian electricity generation mix was used to calculate

233 the environmental impacts associated with the use of electricity for hydrogen production,  
234 storage and compression [34].

235 Step 2: *Simapro* software calculated the environmental impacts once the inputs and  
236 outputs were linked to the relevant libraries. The program sorted the relevant emissions for  
237 particular impacts, and then converted them to an equivalent amount of environmental  
238 impacts. The Australian Environmental Impact calculation method, developed locally [34],  
239 was used to assess the environmental impacts of the use of hydrogen and petrol for VKT.

240

### 241 **3. Results and discussions**

242

#### 243 **3.1. Comparison of environmental performance of hydrogen with petrol**

244

245 The comparative environmental performance of three scenarios has been carried out. The first  
246 scenario is the life cycle of hydrogen when the hydrogen is produced by SMR. The second  
247 scenario is for hydrogen produced by AE. The last scenario is the life cycle of petrol.

248 Contributions to global warming, photochemical smog and eutrophication have been  
249 found to be the predominant environmental impacts in these three scenarios (Figure 2). While  
250 hydrogen is a cleaner burning fuel than petrol, the AE scenario produces more life cycle  
251 global warming and eutrophication impacts than the latter in the petrol scenario. This is  
252 mainly due to the emissions of CO<sub>2</sub> (causing global warming) and NO<sub>x</sub> (nitrogen oxides  
253 causing eutrophication) from electricity and diesel consumption during upstream activities  
254 (alkaline electrolysis, compression for distribution and storage, and transportation) being  
255 higher than those for petrol.

256 The life cycle global warming impacts due to the use of hydrogen produced in the AE  
257 scenario are 2.3 times greater than those of petrol. Walwijk *et al.* [35] also found that CO<sub>2</sub>-e

258 emissions from electrolytic hydrogen production and use would be higher (approximately 1.6  
259 times) than those from petrol. There are similar results in terms of emissions for  
260 eutrophication. Figure 2 indicates that  $\text{PO}_4^-$ -e eutrophication emissions from the AE scenario  
261 are significantly greater than for petrol. However, in terms of photochemical oxidation  
262 emissions, the results are quite different. Both the hydrogen scenarios produce less  $\text{SO}_x$  and  
263  $\text{NO}_x$  ( $\text{C}_2\text{H}_2$ -e emissions) throughout the life cycle from a photochemical perspective.

264 The SMR scenario produces slightly lower environmental impacts than the petrol scenario.  
265 About 4%, 91% and 23% of the global warming, photochemical smog and eutrophication  
266 impacts, respectively, can be avoided due to the replacement of petrol with hydrogen fuel  
267 produced under the SMR scenario. In addition, hydrogen production from the SMR scenario  
268 is less harmful to the environment than the from AE scenario in its global warming,  
269 photochemical smog and eutrophication impacts, because electricity consumption in the AE  
270 scenario is about 6.7 times higher than that in the SMR process (Tables 1 and 2).

271 The life cycle emissions from the AE scenario were found to be significantly higher than  
272 for the SMR scenario across every environmental impact category. This is likely attributable  
273 to the large quantities of coal (37%) and natural gas (60%) in the Western Australian energy  
274 mix required to produce the electricity for electrolysis; however, this will be examined in  
275 more detail in the following section. Further investigation has been carried out to determine  
276 the inputs or processes causing the most environmental impacts (hotspots) so that the  
277 appropriate mitigation strategies can be considered for making hydrogen fuel  
278 environmentally competitive with petrol.

279

280

281 *3.2 Breakdown of environmental impacts of the use of hydrogen produced by steam*  
282 *methane reforming*

283

284 In order to find the hotspots, the percentage distribution of global warming, photochemical,  
285 and eutrophication impacts in terms of inputs for the SMR and AE scenarios have been  
286 determined (Table 3).

287 Greenhouse Gas Emissions: The majority (88.64%) of GHGs are generated by SMR, the  
288 generation of electricity and the production of steam. The SMR process itself produces the  
289 largest amount of CO<sub>2</sub>-e (44.9% of the total emissions).

290 The generation of electricity, in particular from coal and natural gas, produces the second  
291 largest amount of CO<sub>2</sub>-e. This life cycle phase accounts for 29.6% of the total emissions due  
292 to the heavy reliance on fossil fuels as the primary source of fuel for generating electricity.  
293 The production of steam is also a carbon intensive process, accounting for 15.5% of the total  
294 emissions.

295 The results of the SMR model in a 2007 Canadian study (0.3602 kg CO<sub>2</sub>-e per VKT) are  
296 similar to those in the current study (0.252 kg CO<sub>2</sub>-e per VKT) [36]. The difference in  
297 emission output is likely attributable to the technical efficiency improvement during this  
298 period. The average hydrogen fuel consumption during 2006–12 was 0.0227 kg/VKT, while  
299 the present study considered the latest consumption figure in 2011 (0.01 kg/VKT). The  
300 emissions breakdown clearly indicates that for GHG emissions to be reduced, improvements  
301 need to be made to the aforementioned CO<sub>2</sub>-e intensive life cycle phases.

302 Photochemical Smog Emissions: The major life cycle phases contributing to  
303 photochemical emissions are also the production of steam, the steam reforming operation and  
304 electricity generation. Together, these three life cycle phases represent 63% of the total C<sub>2</sub>H<sub>2</sub>-  
305 e emissions due to significant levels of NO<sub>x</sub> and VOCs released into the atmosphere. The  
306 second largest contribution is from tailpipe emissions (30%), mainly NO<sub>x</sub>.

307 Eutrophication Emissions: Eutrophication emissions are produced primarily from the  
308 production of steam, the production of electricity and from the steam reforming process. In  
309 total, these processes account for 86.83% of the total of eutrophication emissions. Producing  
310 the steam required for reforming emits 0.016 g of PO<sub>4</sub><sup>-e</sup> per VKT while the generation of  
311 electricity for the steam reforming and compression processes produces 0.0385 g of PO<sub>4</sub><sup>-e</sup>  
312 per VKT.

313

### 314 ***3.3 Breakdown of environmental impacts of the use of hydrogen produced by alkaline*** 315 ***electrolysis***

316

317 Table 3 also shows the breakdown of global warming, photochemical, and eutrophication  
318 impacts that would result from the production and use of hydrogen fuel generated by AE.  
319 Greenhouse Gas Emissions: The overwhelming majority of life cycle GHGs emitted during  
320 the alkaline electrolysis scenario are attributable to the generation of electricity. Table 1  
321 shows that 93.1% of the total GHG emissions are generated from the electricity supply, of  
322 which 78.3% of the CO<sub>2</sub>-e comes from electricity generation from coal and 14.8% comes  
323 from electricity generation from natural gas. AE is very energy intensive, requiring 62.7 kWh  
324 per kilogram of hydrogen production which equates to 0.63 kWh per VKT. Although AE  
325 itself is virtually emission free, generating the required electricity is currently very carbon  
326 intensive.

327 Photochemical Oxidation Emissions: Table 3 clearly shows that electricity generation  
328 from coal and gas accounts for 73.4% of total C<sub>2</sub>H<sub>2</sub>-e emissions; however, vehicle tailpipe  
329 emissions are also significant. Tailpipe emissions account for 16.5% of the total C<sub>2</sub>H<sub>2</sub>-e  
330 emissions and this is attributable to the combustion of hydrogen within the vehicle engine.  
331 NO<sub>x</sub>, as well as fugitive hydrocarbon emissions, are also emitted during electricity generation  
332 and contribute to the development of photochemical smog.

333 Eutrophication Emissions: The majority of the emissions (about 93.9%) causing  
334 eutrophication are generated during the production of electricity from coal and natural gas,  
335 with these processes contributing 84.6% and 9.3% respectively. The first compression stage  
336 of hydrogen gas is somewhat significant with a 2.5% contribution. Producing the electricity  
337 required for electrolysis emits 0.3 g of PO<sub>4</sub><sup>-e</sup> per VKT while the compression processes  
338 produces 0.008 g of PO<sub>4</sub><sup>-e</sup> per VKT.

339

### 340 ***3.4 Mitigation and reduction of emissions using wind***

341

342 The previous sections identified electricity generation as a major source of global warming,  
343 photochemical oxidation and eutrophication emissions for both the SMR and AE scenarios. It  
344 is clear from the breakdowns of the life cycle emissions that reducing the carbon intensity of  
345 electricity production would have the greatest environmental benefit and would significantly  
346 reduce total emissions in each impact category.

347 The implementation of wind-generated electricity for hydrogen production has the  
348 potential to substantially reduce the emissions across all impact categories in every life cycle  
349 phase, excluding for the vehicle use phase as the only input is hydrogen gas.

350 Wind power is a promising technology in Australia with a potential to generate  
351 renewable and virtually emissions-free electricity. As of 2009, Western Australia's wind

352 energy capacity was 202.7 MW which represents a significant investment [37] and currently  
353 Western Australia has 42 operating wind farms [38,39].

354 Wind technology is poised to be a potential solution to reducing emissions during  
355 hydrogen production by greatly reducing reliance on coal and gas. The potential benefits are  
356 greatest for the AE scenario as the only life cycle phase which relies directly on fossil fuels is  
357 the transportation of hydrogen by diesel truck.

358 The emissions from the SMR scenario will also benefit from lower emission levels;  
359 however, there is still a reliance on fossil fuels, particularly natural gas, during the extraction  
360 and steam reforming processes. This means that although emissions from electricity  
361 production will be reduced, there is still potential for significant environmental impacts  
362 resulting from the use of fossil fuels.

363 The efficacy of wind electricity needs to be assessed for both the SMR and AE scenarios  
364 before any conclusions can be made regarding the net environmental effects. Figure 3 shows  
365 that the environmental impacts can be significantly reduced due to the use of wind energy in  
366 the production, delivery and storage of hydrogen fuel. This is because the substitution of coal  
367 and natural gas powered electricity with wind generated electricity for production and storage  
368 purposes have significantly reduced the emissions of CO<sub>2</sub>, NO<sub>x</sub> and O<sub>3</sub>, which cause global  
369 warming, eutrophication and photochemical smog impacts, respectively. About 31%, 19%  
370 and 35% of the total global warming, photochemical smog and eutrophication impacts can be  
371 reduced by using wind electricity in the SMR scenario. In the AE scenario, global warming  
372 and eutrophication impacts have been almost completely eliminated (by 99%) with the use of  
373 wind energy in the life cycle of hydrogen fuel.

374 The replacement of grid electricity with wind electricity could make hydrogen fuel  
375 environmentally competitive with petrol from the global warming, photochemical smog and  
376 eutrophication impacts perspectives. Although the SMR scenario using grid electricity (coal



377 and natural gas mix) produced less environmental impacts than petrol, a further reduction in  
378 environmental impacts is possible when grid electricity is replaced with wind-generated  
379 electricity. About 37%, 91% and 64% of the total global warming, photochemical smog and  
380 eutrophication impacts can be reduced by replacing petrol with hydrogen fuel under the SMR  
381 scenario with wind-generated electricity. The AE scenario has significant potential to reduce  
382 global warming (97%), photochemical smog (96%) and eutrophication (98%) impacts due to  
383 replacement of petrol with hydrogen fuel.

384 Therefore, the use of wind-generated electricity in the hydrogen fuel cycle not only  
385 reduces overall environmental impacts in hydrogen fuel production but also makes the  
386 hydrogen fuel environmentally friendlier than petrol. When grid electricity was used for  
387 hydrogen production, the SMR scenario appeared to be more environmentally friendly than  
388 the AE scenario. Interestingly, if wind is only source of electricity used in hydrogen  
389 production, then the AE scenario becomes much more environmentally friendly than the  
390 SMR scenario.

391

#### 392 **4 Conclusions and recommendations**

393

394 LCA has been demonstrated as an effective tool for modeling and quantifying the  
395 environmental impacts from the use of hydrogen as an automotive fuel. Global warming,  
396 photochemical smog and eutrophication have been found to be the predominant  
397 environmental impacts associated with the use of hydrogen fuel produced from both SMR  
398 and AE. The initial results of the models found that the SMR scenario emitted 0.252 kg of  
399 CO<sub>2</sub>-e, 0.000079 kg of C<sub>2</sub>H<sub>2</sub>-e and 0.00012 kg of PO<sub>4</sub><sup>-</sup>-e per VKT. The AE scenario was  
400 found to emit 0.67 kg CO<sub>2</sub>-e, 0.000139 kg of C<sub>2</sub>H<sub>2</sub>-e and 0.000322 kg of PO<sub>4</sub><sup>-</sup>-e per VKT.

401 In order to determine the feasibility of hydrogen as an automotive fuel, the life cycle  
402 impacts were compared to those of petrol. When grid electricity is used in the hydrogen fuel  
403 life cycle, the use of hydrogen fuel was found to be environmentally friendlier than petrol  
404 from global warming, photochemical oxidation and eutrophication perspectives under the  
405 SMR scenario. Except for the photochemical smog impact, the AE scenario produces higher  
406 global warming and eutrophication impacts than petrol. The global warming and  
407 eutrophication impacts associated with the production and use of petrol have been found to be  
408 2.3 and 1.8 times lower than hydrogen fuel produced from the AE scenario, respectively. For  
409 both the SMR and AE scenarios, electricity was a major source of emissions; however, the  
410 AE model required nearly seven times the electricity of the SMR model, hence the greater  
411 environmental impacts. Natural gas was also a major source of emissions, particularly in the  
412 SMR model, as it was required in large quantities during the SMR process.

413 In order to mitigate the environmental impacts further, the LCAs were reworked so as to  
414 incorporate electricity from wind turbines to reduce the reliance on coal and gas. The results  
415 from the wind hydrogen models revealed significant improvements in all impact categories  
416 and emissions reduction below the levels of petrol.

417 However, the situation is different when electricity generated by wind is incorporated  
418 into the LCA analysis. The incorporation of wind-generated electricity into the SMR model  
419 reduced the global warming impact ( $\text{CO}_2\text{-e}$ ), photochemical smog ( $\text{C}_2\text{H}_2\text{-e}$ ) and  
420 eutrophication ( $\text{PO}_4\text{-e}$ ) emissions by 31%, 19% and 35.0% respectively. More impressively,  
421 the  $\text{CO}_2\text{-e}$ ,  $\text{C}_2\text{H}_2\text{-e}$  and  $\text{PO}_4\text{-e}$  emissions from the AE model were reduced by 99%, 84% and  
422 99% respectively. Also, hydrogen production can be environmentally feasible compared to  
423 petrol under the AE and SMR scenarios when the electricity is generated by wind.

424 The results of this study could be improved by widening the scope to include  
425 consideration of economic factors. The study has indicated that, from an environmental

426 perspective, both hydrogen models can be made feasible by incorporating wind-generated  
427 electricity. However, the capital costs of wind-generated electricity have not been considered,  
428 nor the prices of grid electricity. For instance, a preliminary review of capital costs found that  
429 South West Interconnected System (SWIS) connected wind farms commissioned in Western  
430 Australia after 2000 cost, on average, \$2.22 million/MW of output [40]. The cost of natural  
431 gas and water could also be incorporated into the models to provide an improved  
432 environmental-economic analysis, particularly for the SMR model.

433 This study also assumed that for the wind scenario, the electricity needed for  
434 compressing the hydrogen gas was sourced from wind generation. Given that the models  
435 employed centralized hydrogen production, where hydrogen gas was transported from a  
436 production facility to fuelling stations within the metropolitan area, it is inaccurate to assume  
437 that the electricity used at the fuelling station would be sourced from wind turbines. A more  
438 accurate emissions model could be developed if the electricity required for compressing the  
439 hydrogen was sourced from SWIS.

440 The study could also include alternative hydrogen storage systems, such as cryogenic  
441 liquid hydrogen tanks or hydride systems, as opposed to compressed hydrogen tanks, which  
442 may require less energy during refueling.

443

#### 444 **Acknowledgements**

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447

448 **References**

449

- 450 1. Taljan G, Fowler M, Canizares C, Verbic G. Hydrogen storage for mixed wind–nuclear  
451 power plants in the context of a Hydrogen Economy. *Int J. Hydrogen Energy* 2008; 33:  
452 4463-75.
- 453 2. Ramesohl S, Merten F. Energy system aspects of hydrogen as an alternative fuel in  
454 transport. *Energy Policy* 2006; 34 : 1251-9.
- 455 3. <http://www.soe.wa.gov.au/report/about.html>.
- 456 4. Australian Bureau of Statistics. *Motor Vehicle Census 2007*. Canberra: Australian  
457 Bureau of Statistics; 2007.
- 458 5. Li ZJ, Rose M, Hensher DA. Forecasting automobile petrol demand in Australia: an  
459 evaluation of empirical models. *Transportation Research Part A* 2009; 44 (1): 16-38.
- 460 6. Lenzen M. Total requirements of energy and greenhouse gases for Australian transport.  
461 *Transportations Res Part D4:Transportation and Environment* 1999; 4(4): 265-90.
- 462 7. McLellan B, Shoko E, Dicks AL, Costa JCD. Hydrogen production and utilization  
463 opportunities for Australia. *Int J Hydrogen Energy* 2004; 30: 669-79.
- 464 8. Western Australian Planning Commission. *State planning strategy. Discussion Paper*.  
465 Perth: Western Australian Planning Commission ; 1995.
- 466 9. Biswas WK, Barton L, Carter D. Biodiesel production in a semiarid environment: a life  
467 assessment approach, *Environ Sci Technol* 2011; 45(7): 3069–74.
- 468 10. Dinca C, Rousseaux P, Badea A. A life cycle impact of the natural gas used in the energy  
469 sector in Romania. *J Cleaner Prod* 2007; 15: 1451-1462.
- 470 11. <http://nglicensing.conocophillips.com/EN/about/gas/Pages/index.aspx>.
- 471 12. <http://www.naturalgas.org/environment/naturalgas.asp>.

- 472 13. Tamura I, Tanaka T, Kagajo T, Kuwabara S, Yoshioka T, Nagata T et al. Life cycle CO<sub>2</sub>  
473 analysis of LNG and city gas. *Applied Energy* 2001; 68: 301-19.
- 474 14. Okamura T, Furukawa M, Ishitani H. Future Forecast for Life-cycle Greenhouse Gas  
475 Emissions of LNG and city gas 13A. *Applied Energy* 2007; 84: 1136–49.
- 476 15. May JR, Brennan DJ. Life cycle assessment of Australian fossil energy options.  
477 *Sustainable Dev Technol* 2003; 81 (5): 317–30.
- 478 16. Dicks AL, Diniz da Costa JC, Simpson A, McLellan B. Fuel cells, hydrogen and energy  
479 supply in Australia. *J Power Sources* 2003;131 (12): 1-12.
- 480 17. Spath PL, Mann MK. Life cycle assessment of hydrogen production via natural gas  
481 steam reforming. Colorado: National Renewable Energy Laboratory; 2001.
- 482 18. Granovskii M, Dincer I, Rosen MA. Life cycle assessment of hydrogen fuel cell and  
483 gasoline vehicles. *Int J Hydrogen Energy* 2005; 31 (3): 337-352.
- 484 19. Koroneos C, Dompros A, Roumbas G, Moussiopoulos N. Life cycle assessment of  
485 hydrogen fuel production processes. *Int J Hydrogen Energy* 2004; 29: 1443–50.
- 486 20. Dufoura J, Serrano DP, Gálvez JL, Moreno J, Garcí'a C. 2008. Life cycle assessment of  
487 processes for hydrogen production: Environmental feasibility and reduction of  
488 greenhouse gases emissions. *Int J Hydrogen Energy* 2009; 34: 1370-6.
- 489 21. Cetinkaya E, Dincer I, Naterer GF. Life cycle assessment of various hydrogen  
490 production methods. *Int J Hydrogen Energy* 2011; 37: 2071-80.
- 491 22. Office of Energy. Energy WA Gas Reserves, edited by Office of Energy. Perth.  
492 Government of Western Australia; 2006.
- 493 23. International Organization for Standardization (ISO). ISO environmental management –  
494 life cycle assessment – principles and framework. In ISO14041. Geneva. International  
495 Organization for Standardization (ISO); 1997.

- 496 24. Sa´inz D, Die´guez P M, Sopena C, Urroz J C, Gandi´ L M. Conversion of a commercial  
497 gasoline vehicle to run bi-fuel (hydrogen-gasoline). *Int J Hydrogen Energy* 2011; 37:  
498 1781-9.
- 499 25. <http://www.volkswagen.com.au/en/models/polo.html>
- 500 26. BP Australia Pty Ltd. Material safety data sheet - BP Regular Unleaded Petrol.  
501 Melbourne: BP Australia Pty Ltd; 2010.
- 502 27. [http://www.carshowroom.com.au/newcars/reviews/1661/Volkswagen\\_Polo\\_1\\_4l\\_Trendl](http://www.carshowroom.com.au/newcars/reviews/1661/Volkswagen_Polo_1_4l_Trendl)  
503 [ine\\_Review\\_and\\_Road\\_Test](http://www.carshowroom.com.au/newcars/reviews/1661/Volkswagen_Polo_1_4l_Trendl)
- 504 28. Cooper R. personal communication. BOC Australia, Perth; 2010.
- 505 29. Ally J, Pryor T. Life-cycle assessment of diesel, natural gas and hydrogen fuel cell bus  
506 transportation systems. *J Power Sources* 2007; 170: 401-11.
- 507 30. <http://automobiles.honda.com/fcx-clarity/specifications.aspx>
- 508 31. Lee JY, An S, Cha K, Hur T. Life cycle environmental and economic analyses of a  
509 hydrogen station with wind energy. *Int J Hydrogen Energy* 2009; 35: 2213-2225.
- 510 32. Wallnera T, Lohse-Buscha H, Gurskia S, Duobaa M, Thielb W, Martinb D et al. Fuel  
511 economy and emissions evaluation of BMW Hydrogen 7 Mono-Fuel demonstration  
512 vehicles. *Int J Hydrogen Energy* 2008; 33: 7607-18.
- 513 33. PRé Consultants. *Simapro* Version 7.2. The Netherlands: PRé Consultants; 2010.
- 514 34. RMIT. Australian LCA database 2005, Centre for Design. Victoria: Royal Melbourne  
515 Institute of Technology; 2005.
- 516 35. Walwijk M, Buckman M, Troelstra W. Elam N. *Automotive Fuels for the Future.*  
517 *Technology & Engineering.* Paris: OECD/IEA; 1999.
- 518 36. Hussain MM, Dincer I, Li X. A preliminary life cycle assessment of PEM fuel cell  
519 powered automobiles. *Applied Thermal Engineering* 2007; 27 (2007): 2294-9.

- 520 37. Cuevas-Cubria C, Schultz A, Petchey R, Maliyasena A. Sandu S. Energy in Australia  
521 2010. Canberra: Department of Resources, Energy and Tourism, Australia; 2010.
- 522 38. Zahedi, A. Australian renewable energy progress. Renewable and Sustainable Energy  
523 Reviews 2010; 14(8): 2208-13.
- 524 39. Ehsani M Y, Gao SGay E, Emadi A. Fuel cell vehicles. in modern electric, hybrid  
525 electric, and fuel cell vehicles: fundamentals, theory, and design. Florida: CRC Press  
526 LLC; 2005.
- 527 40. Office of Energy. Electricity from Renewable Energy-Fact Sheet, edited by Office of  
528 Energy. Perth: Government of Western Australia; 2010.
- 529 41. Fanning I. personal communication, Mack Trucks Brisbane, 2010.
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532

533 **List of Tables**

534

535

536 Table 1- Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using SMR

537

538 Table 2- Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using AE

539 Table 3 – Breakdown of three major impacts in terms of inputs for two hydrogen production

540 processes

541



542

543 **Table 1 Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using**

544 **SMR**

<b>Inputs</b>	<b>Amount</b>	<b>Unit</b>	<b>Reference</b>
<b>Extraction of natural gas</b>			
Electricity	2.22E-02	kWh	[35]
<b>SMR of natural gas</b>			
Electricity	6.56E-02	kWh	[17]
Natural gas	3.92E-02	kg	
Steam	1.88E-01	kg	
<b>Compression for distribution</b>			
Electricity	2.27E-02	kWh	[39]
<b>Distribution to fuelling station</b>			
Diesel Fuel	0.20	L	[41]
<b>Compression for storage at fuelling station</b>			
Electricity	2.23E-03	kWh	[39]

### **Compression for storage on board vehicle**

Electricity	5.37E-04	kWh	[39]
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### **Vehicle usage**

NO <sub>x</sub> emissions	2.20E-05	kg	[30, 32]
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CO <sub>2</sub> emission	8.19E-04	kg	
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548

549 **Table 2 Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using**550 **AE**

<b>Input</b>	<b>Amount</b>	<b>Unit</b>	<b>Reference</b>
<b>AE process</b>			
Electricity	0.63	kWh	[31]
Potassium hydroxide (KOH)	7.05E-05	kg	
Water	0.11	kg	
<b>Compression for distribution</b>			
Electricity	4.00E-02	kWh	[39]
<b>Distribution to fuelling station</b>			
Diesel fuel	0.20	L	[41]
<b>Compression for storage at fuelling station</b>			
Electricity	2.23E-03	kWh	[39]

---

**Compression for storage on board vehicle**

Electricity

5.37E-04

kWh

[39]

**Vehicle usage**

Hydrogen

1.00E-2

kg

[30, 32]

551 Table 3 Breakdown of three major impacts in terms of inputs for two hydrogen production processes

552

	Global warming impact		Photochemical smog		Eutrophication	
	kg CO <sub>2</sub> -e/VKT	%	kg C <sub>2</sub> H <sub>2</sub> -e/VKT	%	kg PO <sub>4</sub> -e/VKT	%
<b>SMR process</b>						
Electricity supply	7.47E-02	29.6%	1.27E-05	17.3%	3.85E-05	32.0%
Steam reforming operation	1.13E-01	44.9%	1.17E-05	16.0%	1.68E-05	14.0%
Steam production from natural gas	3.92E-02	15.5%	2.2E-05	30.1%	4.91E-05	40.8%
Natural gas extraction for steam reforming	1.30E-02	5.2%	2.56E-06	3.5%	5.94E-06	4.9%
Compression of hydrogen for tanker delivery	9.52E-03	3.8%	1.72E-06	2.4%	5.10E-06	4.2%
Compression of hydrogen at the Fuelling station	2.12E-03	0.8%	3.8E-07	0.5%	1.13E-06	0.9%
Hydrogen distribution via tanker truck	2.27E-04	0.1%	6.58E-08	0.1%	1.20E-07	0.1%
Compression of hydrogen for vehicle tank	5.05E-04	0.2%	8.78E-08	0.1%	2.65E-07	0.2%
Vehicular emission	0.00E+00	0.0%	2.2E-05	30.1%	3.28E-06	2.7%

	Total	2.52E-01	100.0%	7.31E-05	100.0%	1.20E-04	100.0%
<b>AE process</b>							
Electricity supply		6.23E-01	93.09%	1.02E-04	73.37%	3.03E-04	93.86%
Compression of hydrogen for tanker delivery		3.24E-02	4.84%	8.32E-06	5.98%	7.96E-06	2.47%
Electrolysis of water		8.04E-03	1.20%	4.87E-06	3.50%	6.61E-06	2.05%
Compression of hydrogen for storage		4.35E-03	0.65%	7.52E-07	0.54%	1.74E-06	0.54%
Compression of hydrogen for vehicle tank		1.07E-03	0.16%	1.11E-07	0.08%	2.58E-07	0.08%
Hydrogen distribution via tanker truck		2.68E-04	0.04%	6.96E-08	0.05%	9.67E-08	0.03%
Production of KOH		1.34E-04	0.02%	2.78E-08	0.02%	3.22E-08	0.01%
Vehicular emissions		0.00E+00	0.00%	2.29E-05	16.46%	3.10E-06	0.96%
	Total	6.70E-01	100.00%	1.39E-04	100.00%	3.22E-04	100.00%

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554

555 **List of figures**

556

557 Figure 1 Simplified block diagram for hydrogen fuel life cycle models

558

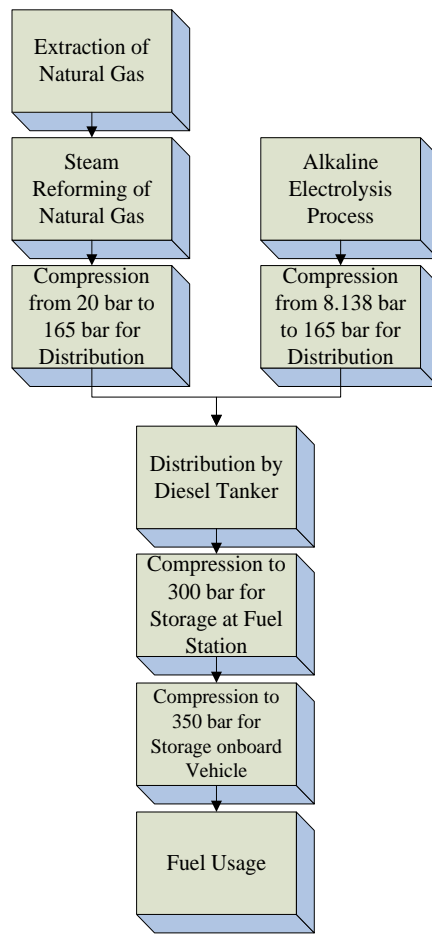
559 Figure 2 Hydrogen models compared to conventional petrol model on an environmental

560 impact basis

561 Figure 3 Implication of mitigation strategies

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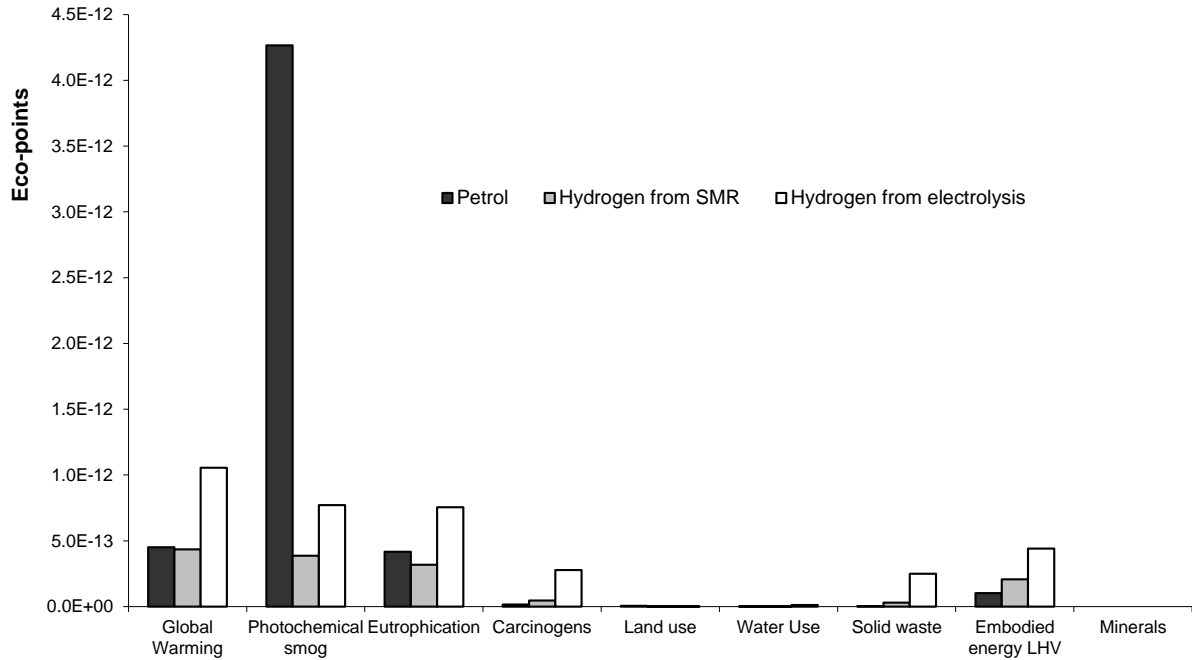


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Figure 1 Simplified block diagram for hydrogen fuel life cycle models



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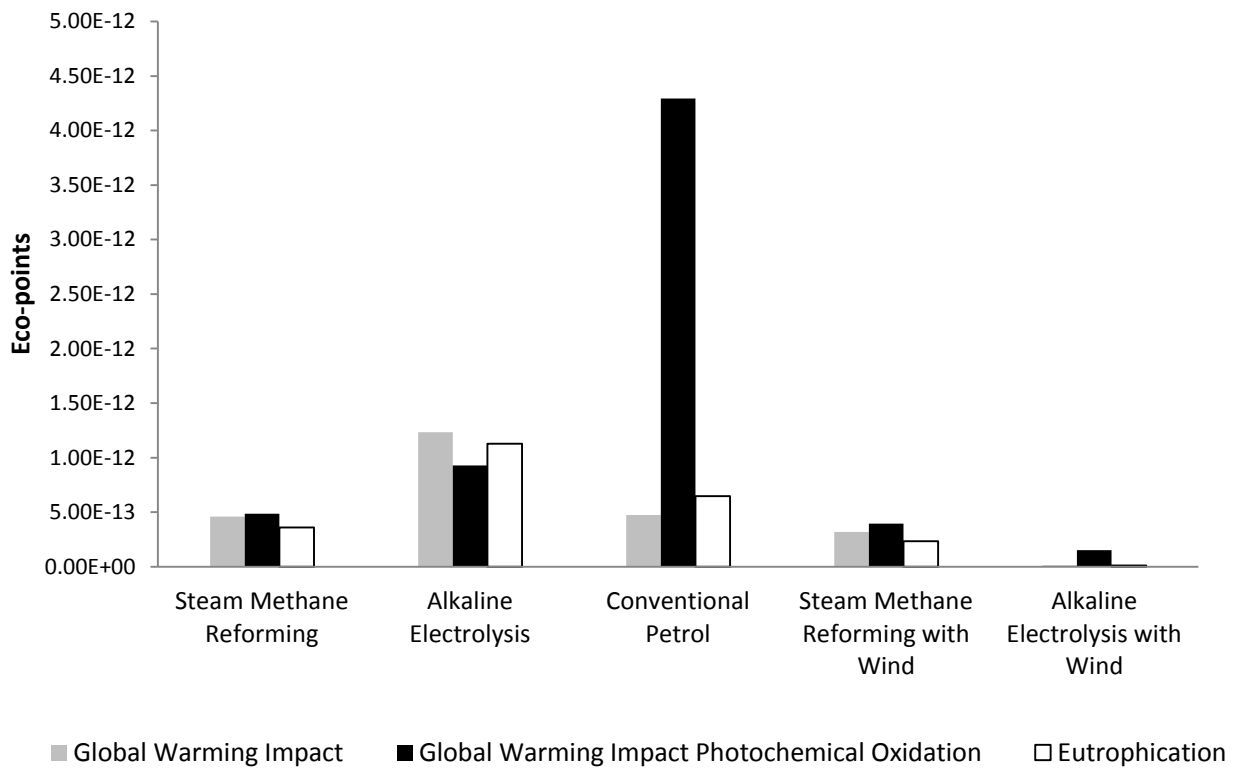
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572 Figure 2 Hydrogen models compared to conventional petrol model on an environmental  
573 impact basis

574 Note: eco-points represent the relative importance of environmental impacts assigned by  
575 industry and society.

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578

579 Figure 3 Implication of mitigation strategies

580