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Microstructural evolution and trace element mobility in Witwatersrand pyrite

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1
2 **Abstract**
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4 Microstructural analysis of pyrite from a single sample of Witwatersrand conglomerate
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6 indicates a complex deformation history involving components of both plastic and brittle
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8 deformation. Internal deformation associated with dislocation creep is heterogeneously
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10 developed within grains, shows no systematic relationship to bulk rock strain or the location
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12 of grain boundaries and is interpreted to represent an episode of pyrite deformation that
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14 predates the incorporation of detrital pyrite grains into the Central Rand conglomerates. In
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16 contrast, brittle deformation, manifest by grain fragmentation that transects dislocation-related
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18 microstructures, is spatially related to grain contacts and is interpreted to represent post-
19
20 depositional deformation of the Central Rand conglomerates. Analysis of the low-angle
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22 boundaries associated with the early dislocation creep phase of deformation indicates the
23
24 operation of $\langle 010 \rangle \{ 100 \}$ slip systems. However, some orientation boundaries have
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26 geometrical characteristics that are not consistent with simple $\langle 010 \rangle \{ 100 \}$ deformation.
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28 These boundaries may represent the combination of multiple slip systems or the operation of
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30 the previously unrecognized $\langle 001 \rangle \{ 120 \}$ slip system. These boundaries are associated with
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32 order of magnitude enrichments in As, Ni and Co that indicate a deformation control on the
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34 remobilization of trace elements within pyrite and a potential slip system control on the
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36 effectiveness of fast-diffusion pathways. The results confirm the importance of grain-scale
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38 elemental remobilization within pyrite prior to their incorporation into the Witwatersrand
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40 gold-bearing conglomerates. Since the relationship between gold and pyrite is intimately
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42 related to the trace element geochemistry of pyrite, the results have implications for the
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44 application of minor element geochemistry to ore deposit formation, suggest a reason for
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46 heterogeneous conductivity and localized gold precipitation in natural pyrite and provide a
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48 framework for improving mineral processing.
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Introduction

Pyrite (FeS_2) is the most abundant metal sulfide in the Earth's crust and is found in a range of geological environments within sedimentary, igneous, and metamorphic rocks. In particular, pyrite is a major phase in sulfide-bearing ore deposits associated with base, ferroalloy and precious metals, most notably gold (Wells and Mullens 1973; Leistel et al. 1998; Simon et al. 1999). Pyrite displays trace element and isotopic compositional variability, which have led to the widespread use of pyrite geochemistry to provide constraints on a range of geological processes; for example, oxygenation of the Earth's atmosphere (Canfield et al. 2000; Bekker et al. 2004), microbial activity in ancient rocks (Schieber 2002) and the syn- to post-deposition fluid history of ore deposits (Craig et al. 1998; Clark et al. 2004). Inclusions trapped within pyrite have been utilized to date the timing of ore forming events (Smith et al. 2001; Kirk et al. 2002). In addition, trace element variations can modify the semi-conducting properties of pyrite, and hence its photovoltaic usage (Lehner et al. 2006), its flotation characteristics during mineral extraction (Abratis et al. 2004) and the chemical mechanisms of pyrite oxidation, the latter having critical implications for acid mine drainage and mineral processing (Evangelou and Zhang 1995; Lehner et al. 2007).

The widespread use of pyrite in geochemical studies requires a comprehensive understanding of the processes by which its composition may be modified under geological conditions. In this paper, microstructural data, obtained by orientation contrast imaging (OCI) and electron backscatter diffraction (EBSD), are presented from a sample from the Welkom goldfield to establish the microstructural evolution and deformation history of Witwatersrand pyrite grains. Microstructural data are then integrated with trace element maps to characterize the spatial and temporal relationships between different microstructures and trace element

1 compositional variations. These observations are then used to place constraints on the relative
2 timing of deformation and its effects on pyrite geochemistry. The trace elements arsenic (As),
3 nickel (Ni) and cobalt (Co) have been targeted because of their known occurrence in
4
5 Witwatersrand pyrites (Reimold et al. 2004), their close relationship to gold mineralization
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7 (Przybyłowicz et al. 1995; Foya et al. 1999) and their role in controlling the semiconductor
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9 and oxidation characteristics of pyrite (Lehner et al. 2006; Lehner et al. 2007). The results are
10
11 utilized to provide a framework for understanding trace element variations in pyrite and
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13 highlight potential complexities in the relationship between composition and deformation
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15 history.
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23 **Geological Background**

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26 The Witwatersrand Basin comprises a sequence of sedimentary rocks that represents the
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28 remnants of a larger sedimentary basin that formed over the Kaapvaal craton between ~2985-
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30 2849 Ma (Kositcin and Krapež 2004). At its base, the Witwatersrand Supergroup both
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32 conformably and unconformably overlies volcanic and clastic sedimentary rocks of the 3.07
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34 Ga Dominion Group (Armstrong et al. 1991) and are unconformable on Archaean basement
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36 rocks (>3.2Ga) (Poujol et al. 2003) of the Kaapvaal craton. The Witwatersrand Supergroup is
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38 overlain by the Ventersdorp Supergroup, which contains a basal conglomerate unit (the
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40 Ventersdorp Conglomerate Formation) that has been dated at 2729 Ma (Kositcin et al. 2003).
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47 The stratigraphy of the Witwatersrand Supergroup is well established and comprises the basal
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49 West Rand Group and the overlying Central Rand Group (McCarthy 2006). These groups
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51 record a transition from early deep marine to non-marine sequences and were deposited
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53 within an evolving retro-arc foreland basin system (Burke et al. 1986; Catuneanu 2001).
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55 Within this evolving, dynamic framework, the deposition of mudstones, sandstones and
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57 localized conglomerates represents a dominance of tectonic controls on sedimentation, upon
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1 which eustatic sea level changes were superimposed in the more distal, marine parts of the
2 basin (Karpeta and Els 1999). Unconformity surfaces, marked by the presence of fluvial
3 conglomerates in the more proximal parts of the Central Rand Group, contain gold. The
4 Central Rand Group rocks in the Welkom district have been metamorphosed at low
5 greenschist facies conditions ($T < 350^{\circ}\text{C}$, $P \sim 2\text{kbar}$) (Phillips and Law 1994).
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13 The occurrence of gold within the Central Rand Group is intimately linked to the presence
14 and origin of pyrite, but the nature of gold mineralization remains controversial. End member
15 models include a detrital or placer origin for gold and pyrite (Mellor 1916; Maclean and Fleet
16 1989; Fleet 1998; Minter 1999; Kirk et al. 2001; England et al. 2002; Meier et al. 2009) and a
17 hydrothermal or metamorphic origin for gold (Graton 1930; Phillips and Myers 1989; Phillips
18 and Law 1994; Barnicoat et al. 1997; Phillips and Powell 2011). However, a “modified
19 placer” model, involving remobilization of placer gold, is supported by some workers (Robb
20 and Meyer 1991; Robb and Meyer 1995; Frimmel 1997; Frimmel 2005; Schaefer et al. 2010).
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Despite increasingly compelling arguments for the latter, a consensus amongst the
Witwatersrand research community remains elusive (cf. Frimmel 1997; Phillips and Powell
2011).

The studied sample comes from the Welkom goldfield of the Witwatersrand Basin, South
Africa, a region where pyrite is intimately involved with extensive gold mineralization. The
structural evolution of the Welkom goldfield comprises east-verging thrust faults,
synchronous with deposition of the Central Rand Group at ~ 2.9 Ga, NNW-striking normal
faults at ~ 2.7 Ga, and the younger developments of folds and thrusts following Transvaal
Group deposition at ~ 2.2 Ga (Dankert and Hein 2010 and references therein).

Analytical Procedure

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3 A thin section was prepared normal to bedding, and polished with progressively finer grades
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5 of diamond paste (9-1 μ m), followed by a further 3 hours of polishing with 0.06 μ m colloidal
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7 silica in NaOH solution (pH10). OCI and EBSD data were collected using a combination of
8
9 W-sourced Philips XL30 and Zeiss EVO scanning electron microscopes (SEMs), housed at
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11 Curtin University, Perth, Australia. EBSD maps were collected from several pyrite grains and
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13 one of these was selected for further chemical analysis. Many other grains were observed with
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15 OCI in order to ensure the microstructural observations were representative. SEM and EBSP
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17 collection settings are similar to those reported elsewhere (see Table 1 of Reddy et al. 2009).
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19 Automatic EBSD mapping was undertaken by collecting and indexing an EBSP from each
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21 node of user-defined grids using Oxford Instruments' Channel 5 Flamenco software.
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23 Empirically obtained EBSPs from the pyrite were automatically indexed to theoretical (hkl)
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25 reflector intensities of the different Kikuchi bands of pyrite using crystallographic data
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27 (Brostigen and Kjekshus 1969) in the Channel 5 phase database. For all data, mean angular
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29 deviation (MAD), was generally low and MADs greater than 1.3° were rejected from the
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31 analysis.
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42 All EBSD data were noise reduced using a "wildspike" correction and a six-neighbour zero
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44 solution extrapolation. The EBSD data from each area was then processed in different ways,
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46 using the Channel 5 "Tango" module, to produce a series of maps that show different aspects
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48 of the microstructure. Band contrast is a measure of the average intensity of the Kikuchi
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50 bands in the electron backscatter pattern (EBSP) relative to the overall intensity and is a
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52 measurement of pattern quality that is a fundamental property of the EBSP (Maitland and
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54 Sitzman 2007). Band contrast is susceptible to variations in crystallographic orientation,
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56 structural integrity, crystal damage and surface topography and is therefore particularly useful
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1 for qualitatively delimiting the sample microstructure. During data processing, band contrast
2 maps were therefore used as a background over which orientation and boundary data were
3 superimposed. A variety of different maps can be created from orientation data (see Maitland
4 and Sitzman 2007 for recent summary). Orientation maps were produced using inverse pole
5 figures (IPF) by coloring the map with respect to the alignment of the map X direction with a
6 crystallographic IPF. The Channel 5 ‘texture’ component, in which each pixel is colored for
7 minimum misorientation relative to a user-defined reference orientation (white cross), was
8 also used. Local misorientation maps represent the average misorientation between a
9 diffraction pattern and its immediately adjacent eight neighbors in the mapping grid. This plot
10 emphasizes boundaries across which lattice orientations change and is therefore useful for
11 correlating with geochemical data.
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30 Boundaries showing variable misorientation angles were applied to maps using the Channel 5
31 ‘grain boundary’ component with different colors being assigned to different magnitudes of
32 misorientation angles. Coloring of the same boundaries to depict the 3D geometry of the
33 associated misorientation axis, relative to a colored stereonet, was done using the Channel 5
34 ‘misorientation axis’ component.
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45 Crystallographic orientation and misorientation axis data were plotted using Channel 5
46 “Mambo” software using lower hemisphere, equal area projections. All data are reported with
47 respect to an X-Y coordinate framework that is defined for each map that is also shown in the
48 orientation maps. The cubic crystal symmetry of pyrite means that 24 possible misorientation
49 angle/axes may be calculated between two different crystallographic orientations (Grimmer
50 1980). In this study, the angle/axis pair corresponding to the minimum misorientation angle
51 was used for analysis. This is justified because the changes in misorientation are progressive
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1 and can be mapped continuously across the grain. It should be noted that angular errors
2 associated with misorientation axis calculations can be up to $\pm 30^\circ$ for 2° misorientation
3 angles, and will vary with angular resolution of the data (Prior 1999). However, the similar
4 geometry of misorientation axes for a range of misorientation angles and the similarity to the
5 crystallographic dispersion axes indicate that misorientation axes data are geologically
6 meaningful. Graphs showing the relative frequency of misorientation angles were calculated
7 using 1000 random assigned lattice orientations (uncorrelated misorientation angles) using
8 Channel 5 software.
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10 Electron microprobe microanalyses were performed on a Cameca SX50, operating at 35kV
11 and 450nA. The sample was carbon coated. Maps were generated using a 2 μm beam size,
12 256 x 256 pixels, counting for 320 ms per pixel. Three maps were run over the same area
13 consecutively and the final maps were generated by digital averaging, which reduces the
14 signal to noise ratio. Arsenic was mapped using both the As $L\alpha$ peak and the As $L\alpha$
15 background whereas nickel was mapped using the Ni $K\alpha$ peak. A transect was collected using
16 at 2 μm step size with a 1 μm beam size to enable better spatial resolution across one of the
17 boundaries. Other operating conditions for these transects were identical to those for mapping.
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43 **Sample Characteristics**

44 The sample is a quartz-dominated, poorly-sorted, clast-supported conglomerate. Quartz clasts
45 range from 100 μm to several centimetres and comprise different textural types with
46 subrounded to subangular clasts of both high and low sphericity. Quartz pebbles show a range
47 of internal microstructures including single crystal clasts with undulose extinction,
48 deformation bands, subgrains and recrystallized grains. Such features provide evidence of
49 deformation by dislocation creep and dynamic recrystallization. Within individual quartz
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1 clasts the microstructural characteristics are consistent, but different grains show variable
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3 microstructure types (Fig 2c,d).
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6 Although the larger clasts are almost exclusively quartz, at finer grain size (<2mm), pyrite
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8 becomes a significant clast type. Pyrite shows a bimodal grain size distribution, with 1-2 mm
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10 coarser grains and a finer fraction of 100-200 μ m grains (Fig 2). The coarse grains comprise
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12 several different types including rounded, high sphericity, inclusion-free grains; rounded, low
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14 sphericity, inclusion-rich grains; and subangular, low sphericity grains showing bands of
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16 inclusions (Fig 2). These different pyrite types are similar to many other descriptions of
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18 Witwatersrand pyrite and correspond to “rounded compact pyrite”, “framboidal pyrite” and
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20 “banded porous pyrite”, respectively (e.g. Koglin et al. 2010).
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27 The matrix of the conglomerate comprises fine-grained quartz, pyrite, chlorite, sericite,
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29 bitumen, zircon and uraninite. Detailed analyses of matrix components in Witwatersrand
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31 conglomerates have been given elsewhere (England et al. 2001; Vollbrecht et al. 2002).
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35 Many of the pyrite grains are fractured. Fractures are commonly intragranular and, less
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37 commonly, transgranular (Fig. 2). The intensity of fracturing within individual grains is
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39 highly variable, and some grains are intensely fragmented. Most fractures are localized at the
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41 sites of impingement between coarse grains (Fig 2e,f) and are commonly filled by quartz,
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43 chlorite and sericite.
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48 Atomic number contrast (ANC) imaging of the coarse, rounded pyrite grains fails to show any
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50 internal microstructure within the grains (Fig. 3a). No resolvable internal chemical variations
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52 associated with either pyrite compositional zoning or mineral inclusions are apparent in ANC
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54 images. However, a very thin rim of new pyrite is visible on the edge of one pyrite grain (Fig
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56 3a). In contrast, OCI shows greyscale differences that indicate crystallographic orientation
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1 variations within the pyrite crystals (Fig. 3b-f) and a microstructural complexity not apparent
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3 by ANC imaging (cf. Fig. 3a & 3b). This complexity is manifest in a series of coarse
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5 (>100 μm) crystallographic orientation domains and a finer, polygonal microstructure
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7 (c.20 μm) that cross cut the boundaries of some of the larger orientation domains (Fig. 3b). In
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9 most cases, the orientation microstructure shows no spatial relationship to the grain
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11 boundaries, and orientation domains occur throughout the grains (Fig. 3). However, in a
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13 couple of examples, thin bands of orientation variations are localized at the grain boundaries
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15 (Fig 3b,c). In some cases, the orientation variations in the pyrite are clearly cut by fractures
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17 (Fig 3c, e), indicating that the fractures are younger.
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24 Orientation mapping via EBSD can be used to quantify the orientation variations within
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26 individual coarse pyrite grains and among fragments of the fractured grains. Orientation
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28 mapping confirms that the framboidal pyrites are multigrain aggregates of small, randomly
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30 oriented, pyrite grains (Fig. 4a,b). In contrast, individual round compact pyrites are single
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32 grains displaying discrete orientation domain boundaries (Fig. 4a). The orientation variations
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34 (misorientations) are systematic and related to crystallographic dispersion around a single
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36 $\langle 100 \rangle$ crystal axis that corresponds to the dominant orientation of low-angle ($<10^\circ$)
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38 misorientation axes calculated from adjacent diffraction patterns (Fig. 4c).
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47 In contrast to intragrain orientation boundaries, orientation variations associated with grain
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49 fragmentation are not systematic and show no consistent misorientation axis (Fig. 5, 6).
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52 Uncorrelated misorientation angles within fragmented and non-fragmented grains consistently
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54 show a larger range for fragmented grains than for those grains that only have low-angle
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56 boundaries (Fig 5, 6, 7). In almost all cases, low-angle boundaries are cut by fractures (Fig.
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58 6b) and the sites of impingement of pyrite grains show no evidence for the development of
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1 low-angle misorientation boundaries (Fig. 5 & 6). However in one grain (grain 1, Fig. 7), low-
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3 angle boundaries are developed at the contact of two grains (Fig 7). In this case, the
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5 misorientation angles associated with these boundaries are too small to accurately assess if the
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7 misorientation axes correspond to low index, rationale orientations associated with dislocation
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9 formation, or small fractures. High spatial resolution EBSD mapping close to fracture walls
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11 (data not shown) fails to show any convincing evidence of minute textural variations along
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13 the crack walls that could be attributed to the operation of dislocation glide in the early stages
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15 of fracture development.
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22 In some cases, pyrite grains show embayed grain boundaries, particularly when in contact
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24 with harder” accessory phases (e.g. zircon) (Fig. 8). In these cases there is no change of lattice
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26 orientation within the embayed region of the boundary indicating dissolution of the pyrite and
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28 deformation by a diffusive mass transfer process. In other pyrite grains within the sample,
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30 there is further evidence of minor pyrite dissolution and deformation accommodated by
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32 diffusive mass transfer at the grain boundary, specifically at sites of impingement with
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34 neighboring grains. However, none of this deformation appears to be associated with
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36 significant development of low-angle boundaries. Fine pyrite rims on the edge of some grains
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38 (Fig. 3a) may mark the sites of redeposition of removed pyrite.
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45 Compact round pyrite grains show a complex intragrain orientation microstructure. One grain,
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47 selected for more detailed microstructural and compositional analysis, and showing no signs
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49 of fracture (Fig. 3a,b), records a lattice distortion across the grain, with zones of localized
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51 recrystallization along discrete, grain-scale, low-angle boundaries (Fig. 9a,b). Orientation
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53 variations within the grain are large ($\sim 90^\circ$) and are dominated by dispersion of
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55 crystallographic axes around a single $\{100\}$ pole (Fig. 9c). Low-angle misorientation axes
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57 also coincide with the single $\{100\}$ pole (Fig. 9d). The orientation microstructure observed at
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1 the edge of the grain in orientation contrast images (Fig. 3b) are not apparent in either the IPF
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3 or local misorientation maps (Fig. 9a,b) indicating that they are caused by small lattice
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5 orientation variations that are insignificant in terms of the larger orientation variations
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7 identified in the grain (Fig. 9).
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11 A detailed analysis of a part of the grain shows a complex microstructure (Fig. 10a) with a
12
13 substructure of discrete low-angle boundaries that are overprinted by zones of more intense
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15 recrystallization that show orientation variations up to 45° (Fig. 10b, e). The recrystallized
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17 zones show irregular boundary orientations that are less discrete than the low-angle
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19 boundaries that they overprint. Orientation variations are again dominated by crystallographic
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21 dispersion and misorientations ($2\text{-}15^\circ$) around a $\{100\}$ pole (Fig. 10e,f).
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27 A chemical map for As (Fig. 10c) shows that the highest As concentrations correlate with
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29 several of the orientation boundaries recognised in the OCI images and EBSD data (Figs.
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31 10a,b). Similar patterns of enrichment of Ni coincide with the same boundaries (Fig 10d). A
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33 transect (A-A') shows an order of magnitude increase in As, Ni and Co associated with a
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35 specific orientation boundary (k) with a misorientation angle of $\sim 5^\circ$ (Fig. 10h). The boundary
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37 of enriched As, Ni and Co does not correlate with misorientation angle (Fig. 11a). However,
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39 mapping of misorientation axes (Fig. 11b) indicate that this boundary is associated with a
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41 different misorientation axis defined by an anomalous set of misorientation axes (shown by
42
43 the red ellipse in Fig 10f), which do not correspond to any low-index crystal axes but have a
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45 range of geometries that spread between two $\langle 210 \rangle$ axes (Fig. 10g).
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54 **Discussion**

55 *Deformation History of Welkom Pyrite*

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1 The origin of the Witwatersrand pyrites in rocks similar to those described here has received
2 considerable attention (see introduction). The sample studied here shows three different
3 types of pyrite grains that are similar to previously characterized pyrite types in the
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8 Witwatersrand Basin: fine-grained pyrite that is found in the matrix, framboidal pyrite, and
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10 rounded, high sphericity pyrites that are typically referred to as round, compact grains. The
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12 framboidal and compact pyrite have different microstructural features.
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17 Round, compact pyrite grains, such as the ones studied here, have previously been considered
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19 to be structurally simple, single detrital grains (Frimmel 1997; England et al. 2002). The OCI
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21 images and EBSD data in this study show considerable microstructural complexity and, in
22
23 contrast to previous studies, the grains show a range of intragrain crystallographic
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25 misorientation variations up to 90° around rational-low index directions. The magnitude of
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27 these misorientations indicates significant amounts of crystal-plastic deformation, yet the low-
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29 aspect ratio pyrite and quartz grain shapes are not consistent with the rock having undergone
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31 high strain deformation.
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40 The orientation boundaries identified in the pyrite grain are systematic in that they are
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42 associated with crystallographic dispersions and misorientation axes parallel to a dominant
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44 $\langle 100 \rangle$ direction. The observed orientation variations are accommodated by discrete low- to
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46 high-angle boundaries that indicate the operation of dislocation creep (Boyle et al. 1998;
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48 Barrie et al. 2007; Barrie et al. 2009). The patterns of varying crystallographic orientations
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50 (Fig. 10b,d,e) can be used to infer the slip systems that were active during the pyrite
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52 deformation (Reddy et al. 2007; Barrie et al. 2008). In this case, the microstructural data
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54 indicate the formation of tilt boundaries dominated by the operation of edge dislocations,
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56 associated with symmetrically equivalent $\langle 100 \rangle \{001\}$ slip systems, and the subsequent
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1 migration of these dislocations into tilt walls to reduce internal strain energy. These
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3 observations are consistent with the dominance of these slip systems in experimental studies
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5 on pyrite (Cox et al. 1981; Graf et al. 1981; Levade et al. 1982; Barrie et al. 2008). Fine-
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7 grained “new” grains within the core of some of these deformed pyrites (Fig.10b) are
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9 interpreted to represent the operation of subgrain rotation recrystallization to produce the
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11 high-angle boundaries.
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17 Significantly, the microstructures associated with dislocation creep are well developed
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19 throughout the pyrite grains and there is no clear relationship between these microstructures
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21 and grain boundaries or between different grains. Combined with the observation that grain
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23 shapes indicates low strain within the bulk sample, the dislocation creep microstructures are
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25 interpreted to predate the erosion and deposition of the detrital grains. The only exception to
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27 this interpretation is found, where small lattice misorientations ($<1^\circ$) are localised at grain
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29 boundaries (Fig 3b,c & Fig. 13), and these are interpreted to reflect the operation of
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31 dislocation glide driven by stresses associated with grain-grain impingement after deposition
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33 of the pyrite grains (e.g. Barrie et al. 2009).
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42 In contrast to the round compact grains, large framboidal pyrite grains have higher aspect
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44 ratios (ellipticity) and comprise aggregates of smaller ($\sim 20\mu\text{m}$ diameter) pyrite grains. These
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46 smaller grains have random crystallographic orientations and record no significant evidence
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48 of intragrain low-angle boundaries associated with the formation and migration of
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50 dislocations or fine-grained recrystallization. The framboidal pyrites are microstructurally
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52 similar to morphologically disordered frambooids that are interpreted to have developed by
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54 primary growth (Ohfuji et al. 2005).
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1 Many of the pyrite grains show intragranular and intergranular brittle fractures that emanate
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3 from grain boundaries at the sites of grain impingement. This spatial relationship indicates
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5 that fracture development is a response to the indentation of adjacent grains, and in three
6
7 dimensions fractures are likely to be cone cracks formed at the sites of localized high stress.
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9 These fractures therefore postdate the juxtaposition of coarse pyrite grains within the sample.
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11 Fractures also cut across and displace low-angle boundaries associated with dislocation creep
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13 (Figs. 3, 6). Grain fragmentation associated with localized, intense fracturing results in non-
14
15 systematic lattice orientation variations within individual grains.
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22 As well as brittle failure, the local presence of slightly embayed contacts between pyrite-
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24 pyrite grains without localized dislocation creep microstructures, is evidence for a limited
25
26 amount of pyrite dissolution. Diffusive mass transfer is more apparent in the embayment of
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28 pyrite in contact with accessory phases in the matrix (e.g. zircon) (Fig. 8). In most cases,
29
30 pyrite adjacent to sites of diffusive mass transfer preserves no significant deformation
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32 accommodated by the formation and migration of dislocations. Fracture edges also fail to
33
34 show any evidence of dislocation glide, though microstructures formed by dislocation glide
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36 may be present along some grain-grain contacts (Fig 7). The localization of diffusive mass
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38 transfer microstructures along impinging grain contacts indicates that deformation by this
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40 mechanism took place *in situ*, that is, after the deposition of the grains.
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51 The microstructural data presented here provide further constraints on the origin of
52
53 Witwatersrand pyrite and provide some new data regarding their evolution. Firstly, round
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55 compact grains comprise recrystallized single grains with a large range in misorientation
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57 angles that do not appear consistent with the bulk strain of the sample. Framboidal grains
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1 have elliptical shapes but show little evidence for internal deformation and have
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3 microstructural characteristics consistent with primary growth. In addition, low-angle
4
5 boundaries associated with dislocation creep (misorientation angles $>2^\circ$) show no relationship
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7 to grain edges and there is no systematic relationship between the internal microstructure of
8
9 the pyrites and their current positions within the sample (Fig. 12b). These observations are
10
11 consistent with a detrital origin for the pyrite grains, with crystal plasticity associated with the
12
13 development of low-angle boundaries and subgrain rotation recrystallization in rounded pyrite
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15 taking place prior to the erosion, transport and deposition of these pyrites in a placer
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17 environment. The different pyrite clast types are interpreted to represent multiple sources for
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19 pyrite detritus, a similar conclusion to that derived from the analysis of Witwatersrand quartz
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21 clasts (Vollbrecht et al. 2002).
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30 In contrast to the characteristics of dislocation creep microstructures, deformation in the form
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32 of fracture and grain fragmentation, is systematic in being associated with the impingement of
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34 neighbor grains (Fig 12b), and clearly post-dates deposition of the grains and formation of the
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36 conglomerate. This deformation is likely to have initiated during the early stages of
37
38 sedimentary compaction. However, evidence for diffusive mass transfer and dislocation glide
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40 processes, plus the presence of chlorite and sericite in some of these fractures, indicates that
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42 deformation probably continued to over a prolonged part of the burial history.
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49 The microstructural data therefore indicates a two-stage deformation history, manifest by the
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51 operation of different deformation mechanisms, that took place prior to, and after, deposition
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53 of the Central Rand group sediments.
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1 Experimental and empirical evidence indicates that pyrite deforms by a range of processes at
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3 natural strain rates within the Earth's crust (Atkinson 1975; Cox et al. 1981; Graf et al. 1981;
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5 McClay and Ellis 1983; Boyle et al. 1998; Freitag et al. 2004; Barrie et al. 2009).
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8 Experimental work indicates the operation of the dislocation creep field for pyrite
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10 deformation at higher temperatures than expected for fracture and diffusive mass transfer
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12 (McClay and Ellis 1983). However, recent detailed microstructural studies on experimental
13
14 and natural pyrites indicate the development of localized dislocation creep microstructures
15
16 may arise at lower temperature conditions (Fig 12a) (Barrie et al. 2009). Despite these
17
18 observations, the distinct characteristics of the two-stage deformation history in the Central
19
20 Rand sample point to a higher-temperature deformation event being responsible for the early
21
22 development of dislocation creep microstructures in the round compact pyrites.
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29 *Relationship between Deformation, Microstructure and Compositional Variation*

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31 Microstructures (dislocations and low-angle boundaries) formed by plastic deformation are
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33 known to have a dramatic effect on the diffusion properties of minerals by forming fast-
34
35 diffusion pathways that facilitate the chemical modification of minerals at much greater rates
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37 than volume diffusion (e.g. Lee 1995; Reddy et al. 2006; Timms et al. 2011). However, the
38
39 role that even small amounts of deformation play in facilitating chemical mobility in pyrite is
40
41 poorly known. The quantitative microstructural and geochemical data presented here
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43 demonstrate a systematic relationship between trace element enrichment and pyrite
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45 deformation.
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55 There are several possible explanations as to why specific orientation boundaries are enriched.
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57 The boundary enriched with As, Ni, and Co has a $<5^\circ$ misorientation angle, a value
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59 comparable to other orientation boundaries in the sample that do not show enrichment (Fig.
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1 11). Therefore, chemical enrichment does not correspond to the magnitude of boundary
2
3 misorientation angle. A further possibility is that the different enriched low-angle boundary
4
5 developed prior to low-angle boundaries associated with $\langle 001 \rangle \{ 100 \}$ slip and during
6
7 differing chemical conditions. Such an interpretation is allowed by the cross cutting nature of
8
9 the different microstructures (Fig. 10a,b). However, $\langle 001 \rangle \{ 100 \}$ slip dominates dislocation
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11 creep in pyrite and it is therefore expected that tilt boundaries associated with $\langle 001 \rangle \{ 100 \}$
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13 slip would also dominate any earlier deformation associated with the formation of the
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15 enriched boundary and would therefore also be chemically modified. There is no evidence of
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26 In detail, the enriched boundary (k) is segmented with each segment recording a different
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28 orientation of the boundary trace and associated misorientation axes that define a trend
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30 between two $\{ 210 \}$ poles (Fig. 10g). The geometry of the trace segments and associated
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32 misorientation axes (Fig. 10g) is not consistent with $\langle 001 \rangle \{ 100 \}$ slip or any simple tilt or
33
34 twist boundary model associated with any known pyrite slip systems. One possible
35
36 interpretation is that the boundary represents the activity of multiple slip systems at a scale
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38 that is unresolvable in the current data. Different boundary segments may then represent
39
40 differing degrees of slip system activity. These slip systems may represent various
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42 contributions from $\langle 001 \rangle \{ 100 \}$. However, the proximity of misorientation axes to $\langle 210 \rangle$
43
44 may point towards the operation of previously unrecognized $\langle 001 \rangle \{ 120 \}$. This slip system
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46 has not previously been recorded in pyrite. However, this may reflect the fact that most
47
48 studies of pyrite deformation have used either a multigrain fabric approach to infer slip
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50 systems (Cox et al. 1981; Siemes et al. 1993) or have analysed the geometry of individual
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52 dislocations using the transmission electron microscope (Cox et al. 1981; Graf et al. 1981;
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54 Levade et al. 1982). In both cases, the approach tends to recognize the dominant slip system
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1 and is unlikely to lead to the recognition of less common ones. Despite the inability of current
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3 data to fully resolve the microstructural nature of the enriched boundary, it appears that the
4
5 enrichment in As, Ni and Co takes place along orientation domain boundaries that are
6
7 complex and different from the ubiquitous boundaries associated with $\langle 001 \rangle \{ 100 \}$ slip
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9 systems. The variation in As, Ni and Co is therefore inferred to represent a slip system control
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11 on element mobility, such that the complex interaction of different slip planes, with
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13 different lattice spacings and Burgers vectors, facilitate the diffusion of different species at
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15 faster rates than volume diffusion alone.
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23 In Witwatersrand gold deposits, the occurrence of authigenic cobaltite (CoAsS) and
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25 gersdorffite (NiAsS) (Frimmel et al. 1993) indicate post-depositional As, Ni and Co migration
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27 within the rock matrix. However, the enrichment of As, Ni and Co within the studied pyrite is
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29 interpreted to be related to crystal plasticity that took place within the source region of
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31 compact pyrite prior to its erosion and redeposition in the Central Rand group placer deposits.
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33 Given the relationship between Au and As, the data presented here indicate the possibility of
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35 gold mobilization and redistribution prior to Central Rand Group deposition by enhanced
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37 diffusion of these elements from the matrix into the pyrite along specific low-angle
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39 boundaries that behaved as fast-diffusion pathways during an early, high-temperature
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41 ($>500^{\circ}\text{C}$) deformation event. The microstructural and trace element data are therefore
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43 consistent with mobilization of gold both prior to, and after, redeposition of eroded pyrite
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45 clasts into the Central Rand sediments. Such features may explain some of the controversies
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47 between placer and hydrothermal models for Witwatersrand gold mineralization and are
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49 consistent with recent models for remobilization of gold in modified placer models (Robb and
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51 Meyer 1991; Robb and Meyer 1995; Frimmel 1997; Frimmel 2005; Schaefer et al. 2010).
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1 In many deposits, it is common for gold to be intimately associated with pyrite as “invisible
2 gold” (Maddox et al. 1998). The nature of the relationship is complex and may be in the form
3 of a solid solution with the pyrite (Reich et al. 2005), or discrete, structurally-bound nano-
4 particles (Palenik et al. 2004). These end-member possibilities have been shown to reflect the
5 amount of As in the pyrite (Reich et al. 2005) and have been linked to the electrochemical
6 precipitation of gold at low / high As boundaries (Belcher et al. 2004). The positive
7 correlation between gold and As has been documented in the Witwatersrand gold deposits and
8 has been interpreted to be a result of mixed conductivity and electron exchange along
9 chemical (As) contrast boundaries (Meyer et al. 1990; Przybyłowicz et al. 1995).
10 Consequently, the development of zones of localized As, Ni and Co enrichment, and the
11 likely associated increase in localized conductivity, provides a mechanism of producing
12 chemical boundaries that facilitate heterogeneous gold precipitation at the intragrain scale.
13 The mobilization of As, combined with a knowledge that gold and As are linked, provides
14 further indirect support for syn-deformational remobilization of gold rather than a solely
15 placer origin for Witwatersrand gold mineralization. As well as controlling whether gold is
16 chemically alloyed with pyrite or exists as discrete nano-particles, trace element enriched
17 zones will also localize changes in pyrite reactivity and oxidation (Evangelou and Zhang
18 1995; Lehner et al. 2007). The microstructural development of pyrite will therefore also
19 dictate the nature of pyrite breakdown reactions, critical to the optimization of ore processing
20 techniques and gold recovery. However, further characterization of the range of conditions of
21 natural pyrite deformation mechanisms are needed if the enrichment of As, Ni and Co is to be
22 utilized in gold recovery processing.
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1 The intimate nature of pyrite and gold has allowed many researchers to obtain information on
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3 ore deposit fluid chemistry and the timing of gold mineralization through the analysis of
4
5 sulfur and argon isotopes (Smith et al. 2001; Clark et al. 2004). Such studies require that
6
7 pyrite remain a closed system for isotopic diffusion. However, the development of
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9 deformation-related orientation boundaries, which facilitate fast-path diffusion, indicates that
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11 the closed system assumption may not be valid. In many pyrite geochemical studies
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13 anomalous single point analysis are often attributed to the presence of discrete mineral
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15 inclusions. Our study however, shows that anomalous element concentrations can also be
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17 directly related to deformation microstructures.
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23 As the most common sulfide on Earth, pyrite is often considered an excellent record of host
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25 rock history, even where it has undergone high pressures and temperatures and from
26
27 geological environments as wide ranging as mantle xenolith sulfide inclusions, ore deposits,
28
29 early earth deposits and modern sediments. The combined microstructural and microchemical
30
31 approach outlined here reveals a new level of complexity that requires greater consideration
32
33 of microstructural characteristics in pyrite geochemical studies and precludes the assumption
34
35 of chemical mobility simply by volume diffusion. This finding has the potential to develop
36
37 new analytical approaches for constraining geological processes associated with ore body
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39 evolution, and their distal footprints, and the further development and optimization of the
40
41 industrial applications of pyrite.
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4
5 publication XXX and ARC CCFS COE publication YYY.
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Figure Captions

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3 Figure 1. Simplified geological map of the Witwatersrand region showing the distribution of
4 major sedimentary groups and goldfields. Figure modified from Rasmussen & Muhling
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6 (2009).
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12 Figure 2. Photomicrographs of the studied thin section. Yellow rectangles represent the
13 locality of reflected light photomicrographs (Fig 2d-f) and orientation maps (Figs 4-8).
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15 Labeled grains correspond to grains shown in Fig. 3. a) reflected light, image showing
16 distribution of pyrite (white) and quartz (grey). b & c) plane polarized and cross polarized
17 (with sensitive tint plate) images, respectively, showing grain size and microstructural
18 variations in quartz. Colour variations show the presence of subgrains and deformation bands.
19
20 d) reflected light image showing variations in round compact pyrite, framboidal pyrite and
21 finer grained pyrite. d & e) relationship between pyrite grains and fractures. Fracture density
22 is variable amongst grains and fractures tend to nucleate at the grain contacts.
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37 Figure 3. Scanning electron microscopy images of pyrite grains identified in Fig. 2. a) Atomic
38 number contrast image of round compact pyrite shown in 2d. The homogenous greyscale
39 contrasts the orientation contrast image (b). b-f) Orientation contrast images showing lattice
40 orientation variations in five round compact pyrite grains. b) Patchy greyscale variations
41 throughout the grain appear overprinted by zones of finer-grained recrystallized pyrite. On the
42 left hand margin of the grain, the striped appearance indicates a localized zone of lattice
43 distortion on the grain margin. A similar pattern is seen along the right hand margin of the
44 pyrite grain in (c). d-f show lattice orientation variations are present throughout the whole
45 grain. However, in all cases the greyscale variations are only qualitative and cannot be used
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3 Figure 4. Euler orientation map (a) showing microstructure in framboidal and large compact
4 pyrite grains. Colour differences represent different lattice orientations in pyrite and
5 correspond to the colours shown in the {100} pole figures (b & c). Red lines represent low-
6 angle boundaries with misorientation angles between 2-10°. Projections show the orientation
7 of {100} poles and 2-5° misorientation axes. Framboidal pyrite comprises a large number of
8 small grains (~20µm grains) that have random orientations (b). The compact grain comprises
9 a single crystal that has a systematically changing lattice orientation (c) associated with
10 misorientation axes that lie parallel to one of the {100} poles.
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25 Figure 5. a) Band contrast and b) IPF maps of compact rounded pyrites. c) pole figures for
26 {100}, {110} and {111} for grains 1, 2 & 3 (top, middle & bottom respectively). Colors in the
27 map correlate with those in the pole figures. d) Misorientation angle distribution, calculated
28 from 1000 uncorrelated orientations per grain, for grains 1 (blue), 2 (red) and 3 (green). The
29 theoretical random distribution is shown by the black line. n is the number of EBSPs (in
30 thousands) used to construct each grain.
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42 Figure 6. a) Band contrast and b) IPF maps of an intensely fractured rounded pyrite. Map was
43 created from a 560 x 350 grid at 2.5µm step size (~97,000 EBSPs). b) low-angle boundaries
44 that can be traced across the different fragments. c) misorientation angle distribution,
45 calculated from 1000 uncorrelated orientations per grain. The theoretical random distribution
46 is shown by the black line. d) pole figures for {100}, {110} and {111}. Colors in the IPF map
47 correlate with colors in the pole figures.
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1 Figure 7. Euler orientation maps to show microstructural relationships in round compact
2 pyrite grains. b) shows detail of area shown in the white square of (a). c) Pole figure
3 orientations for the green / red grain at the top (grain 1) and the blue grain in the middle of the
4 map (grain 2). The fractured grain shows a more dispersed and less systematic range of
5 orientations than grain 1.
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15 Figure 8. EBSD orientation map showing deformation around a small zircon grain. Colours
16 show Euler orientations of pyrite grains and different colours represent different orientations.
17 Greyscale variations show band contrast variations in non-pyrite phases. The fractured grain
18 in the centre of the image is zircon. The embayed boundaries at sites of impingement (marked
19 DMT) indicate deformation by a diffusive mass transfer process. Orientations variations in the
20 purple pyrite grain are $<1^\circ$.
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32 Figure 9. IPF (a) and local misorientation (b) maps of the rounded compact pyrite shown in
33 Fig 3a. Maps were created using a 400 x 450 grid and 6 μ m step size. Lattice orientations vary
34 significantly associated with dispersion (c) and misorientation axes (d) dominated by a single
35 $<100>$ direction. Note that the orientation variations localized at the grain margin in the
36 orientation contrast image (Fig. 3b) are not seen in the IPF map, indicating that the lattice
37 orientation variations associated with these microstructures are small.
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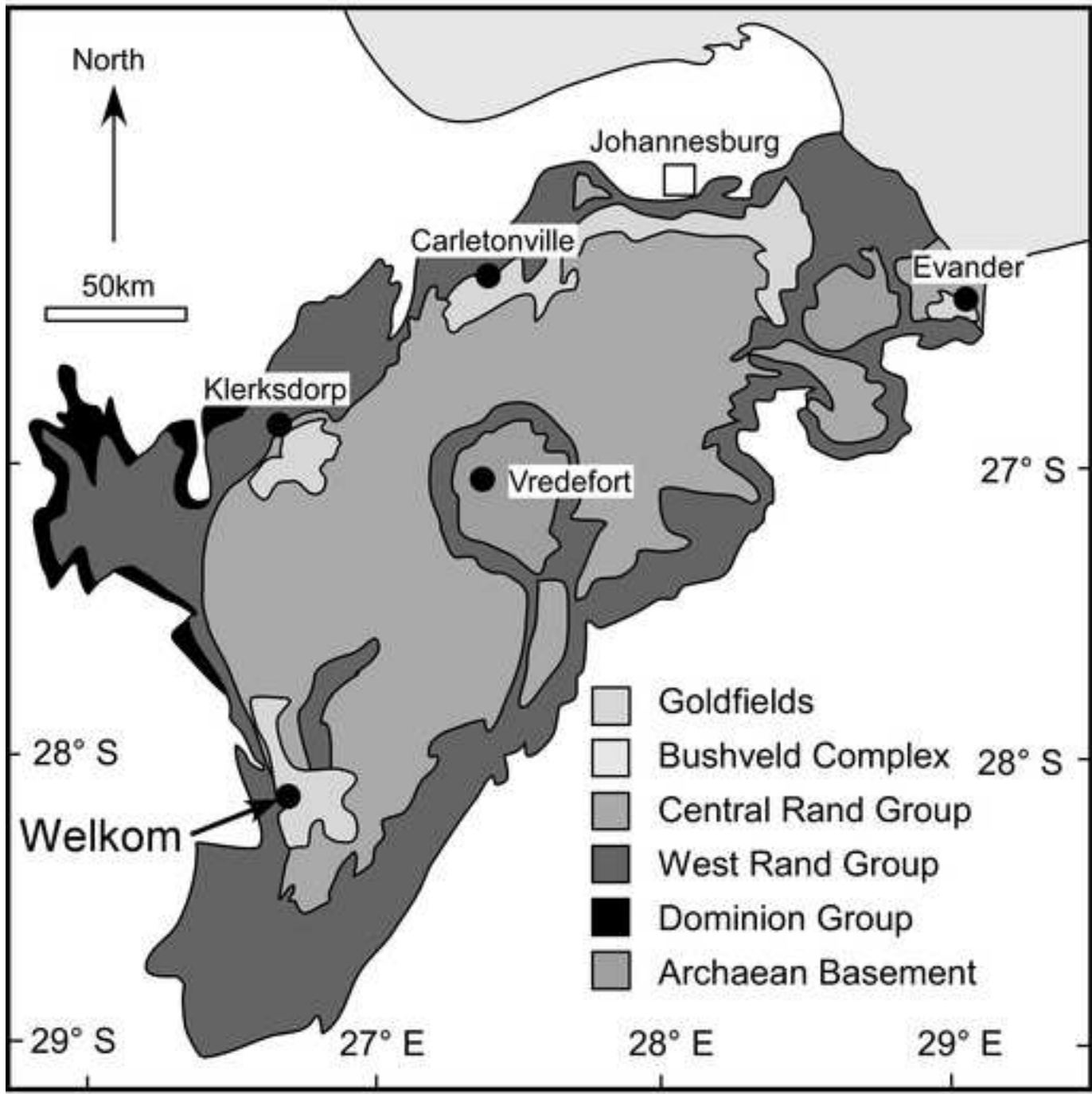
49 Figure 10. Pyrite microanalytical data from area shown in Fig. 9. a) Detailed orientation
50 contrast image of pyrite microstructure. White box shows location of b, c and d; A-A'
51 indicates orientation of profile shown in (h); b) Orientation map of pyrite comprising 33,000
52 individual EBSD determinations. Color corresponds to a misorientation from reference point
53 (white cross) from blue (0°) to red (45°). Sharp changes correspond to orientation boundaries
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1 associated with dislocation migration (e.g. boundary k). c) and d) Abundance of arsenic (c)
2 and Nickel (d) within the pyrite. Note the heterogeneous enrichment (lighter areas) in both As
3 and Ni that correlates exactly with some orientation boundaries (k). e) Pole figures of
4 and Ni that correlates exactly with some orientation boundaries (k). e) Pole figures of
5 crystallographic orientation variations shown by colors in b. f) Geometry of misorientation
6 axis / angle pairs shown for 2-4°, 4-6° and 6-15° misorientation angles. The majority of data
7 correspond to axis of rotation (r) shown in (e). Dispersion of crystallographic axes around the
8 single <100> direction can be related to the operation of two <010> (100) slip systems during
9 deformation. The red ring corresponds to data associated with boundary k shown in detail in
10 (g). g) Projection of misorientation axes and the boundary traces for different segments (k1-
11 k4) of boundary k. The three great circles correspond to {100} planes. h) Variation in trace
12 element abundance (As, Ni, Co), and corresponding variations in misorientation angle along
13 profile A-A' in (a). The increase in As, Ni, and Co correlates with boundary k. Orientation
14 variations associated with crystallographic dispersion around <001> and <001>
15 misorientation axes show no significant enrichment.

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37 Figure 11. EBSD orientation maps showing boundary characteristics for area shown in Fig 9b
38 draped over a greyscale band contrast map. The 'k' marks the boundary shown in Fig. 9. a)
39 Misorientation angle distributions in the boundary network. b) shows the orientation of
40 misorientation axes with respect to the inset projection. The boundary k differs from the
41 majority of other blue boundaries and this reflects the variation of misorientation axes shown
42 in Fig. 9e.

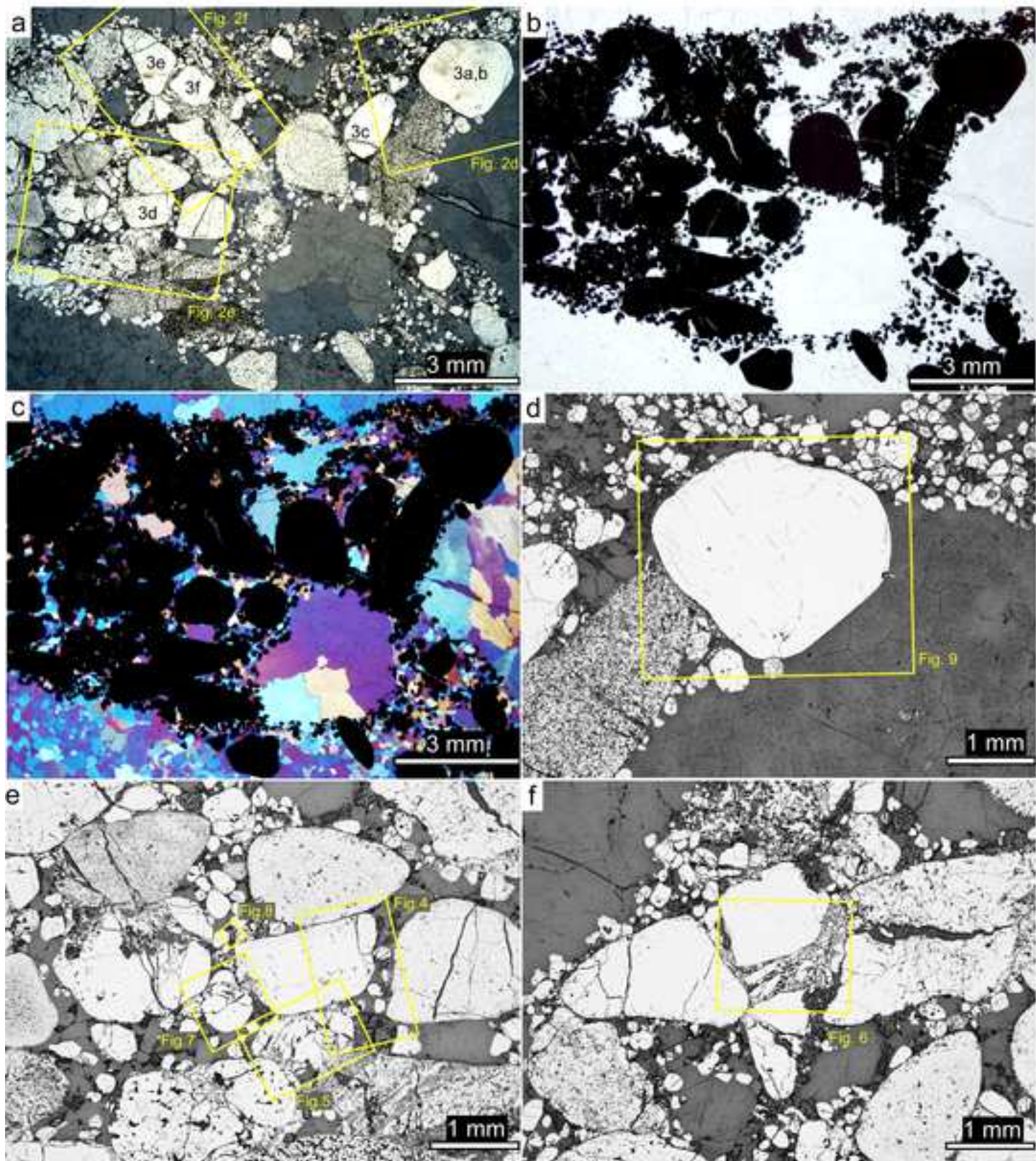
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54 Figure 12. Schematic summary of deformation mechanisms in deformed pyrite a) after Barrie
55 et al (2009) and b) for the grains shown in Fig. 2a. The red grains in b) are round compact
56 pyrites that are similar to those shown in yellow. However, the red depicts grains that have
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1 been studied by OCI (Fig. 3) or EBSD (Figs. 4-9) and show evidence for dislocation creep
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3 and subgrain rotation recrystallization. Black lines cutting across grains in b) represent
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5 fractures. Orange represents zones where diffusive mass transfer (DMT) processes are
6
7 inferred from the convex-concave shapes of grain-grain contacts. The matrix comprises fine-
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9 grained quartz and pyrite with minor bitumen, chlorite, sericite and zircon f. pyrite =
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11 framboidal pyrite; rc pyrite = round compact pyrite.
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Reddy & Hough 2012. Fig. 1
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Figure 2
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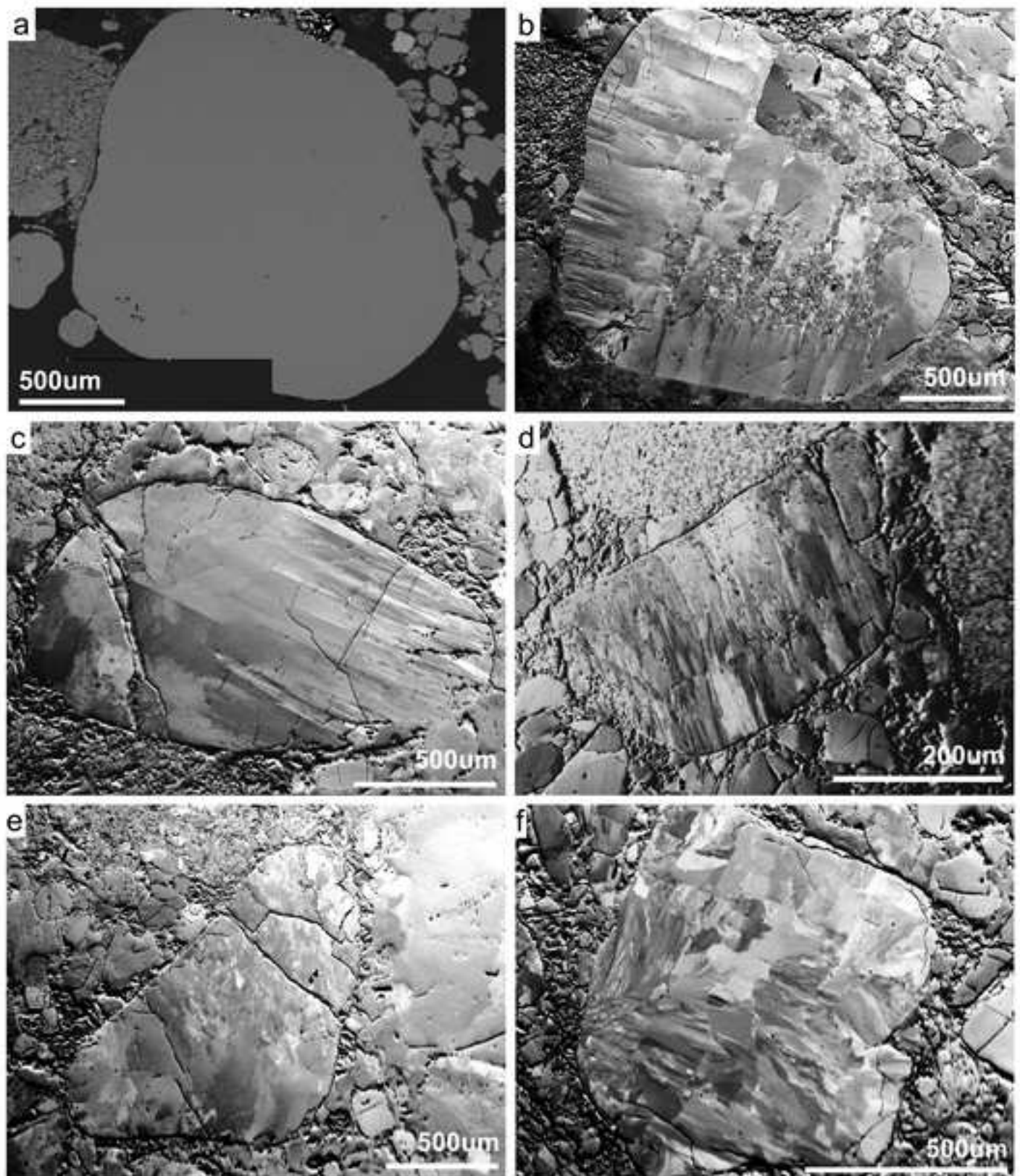


Reddy & Hough 2013. Fig. 2

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Figure 3

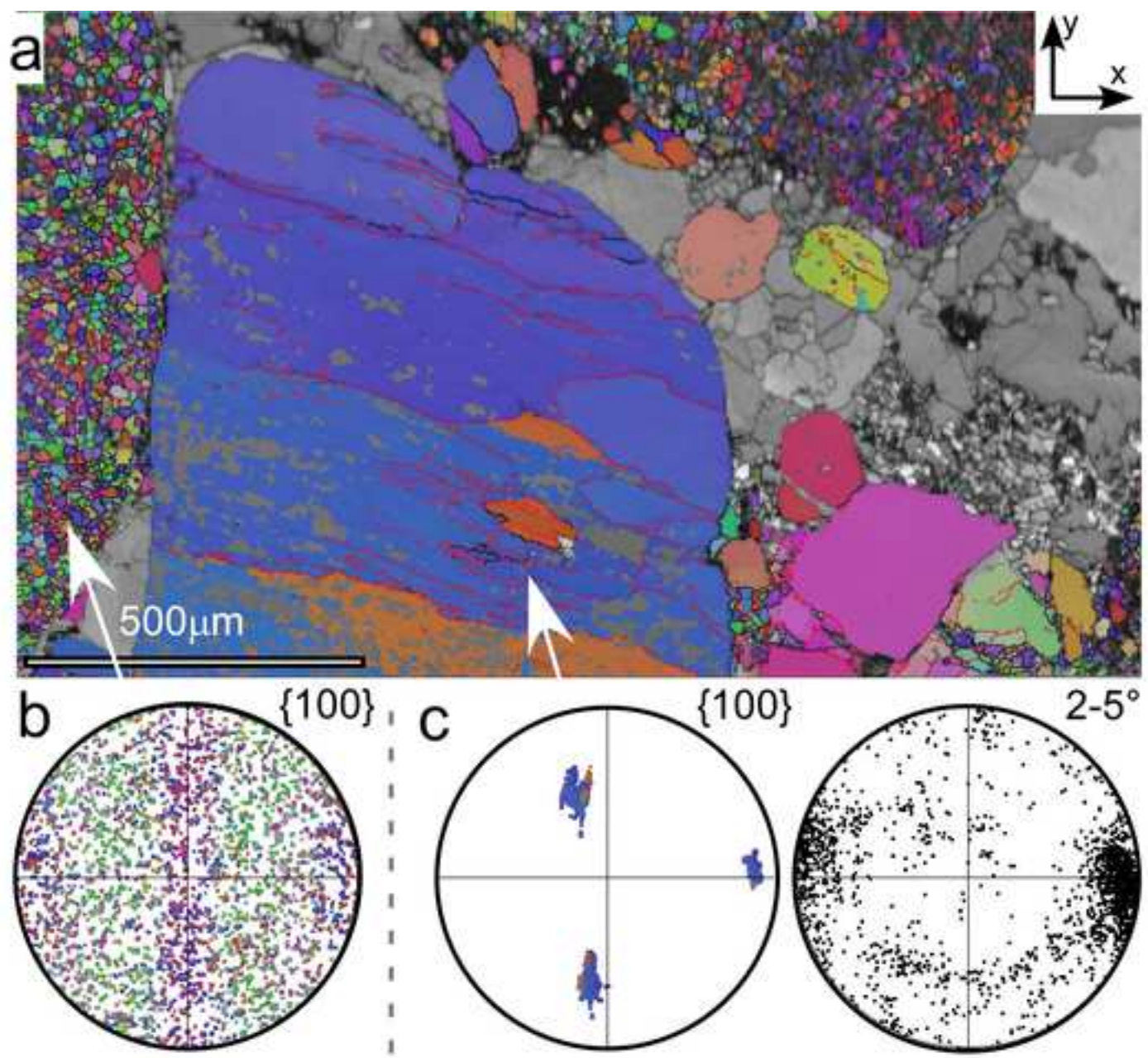
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Reddy & Hough 2013: Figure 3

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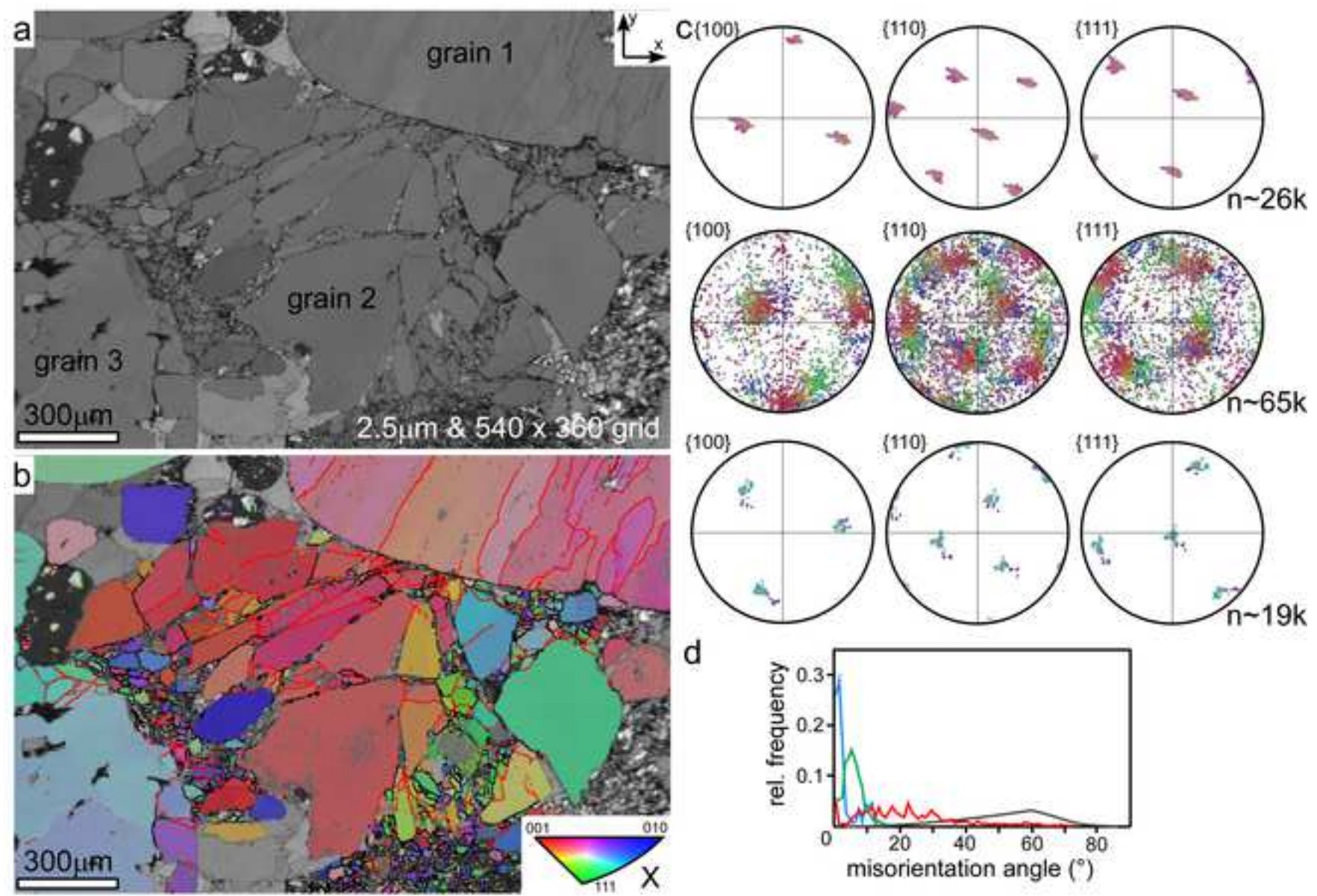
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Reddy & Hough 2013 Figure 4

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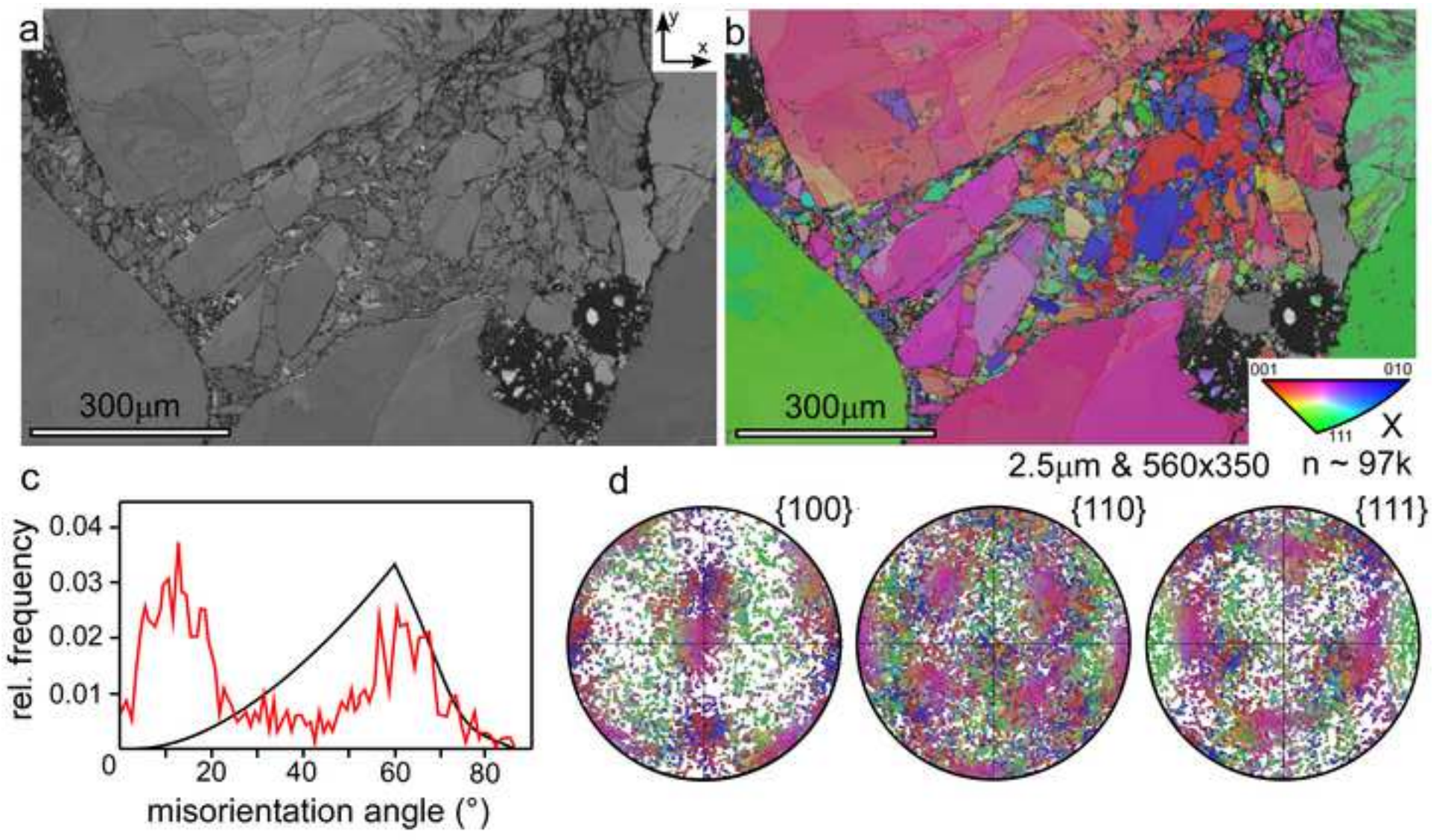
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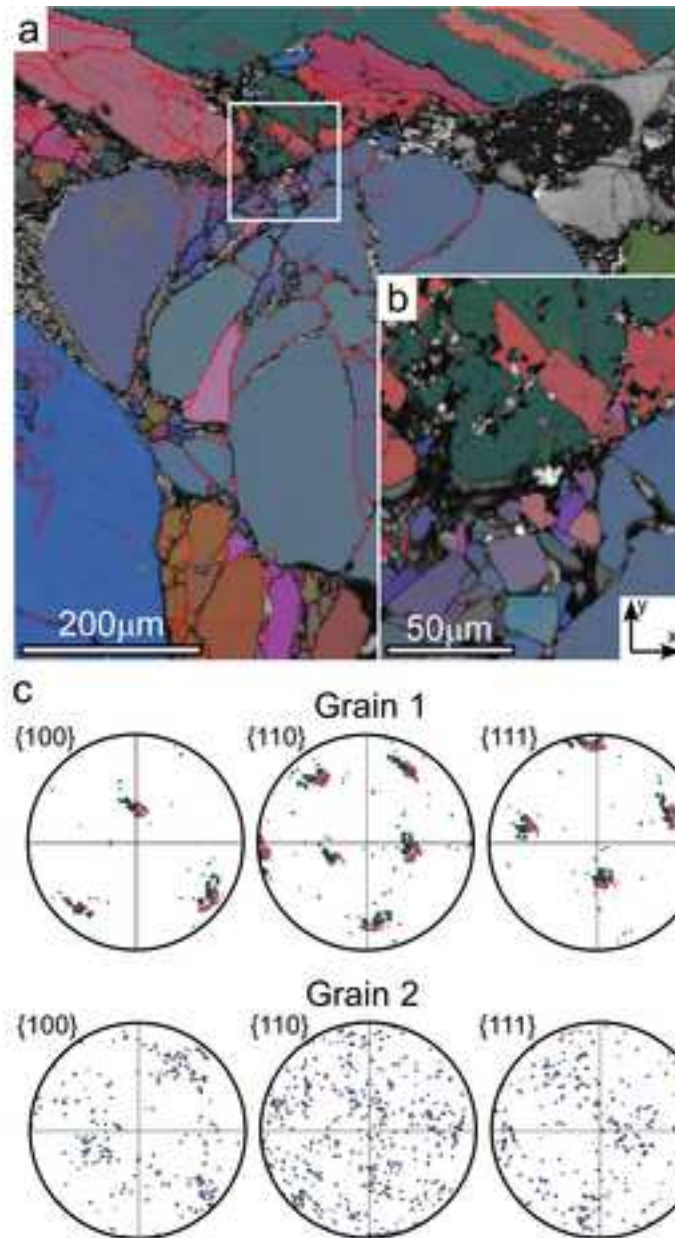
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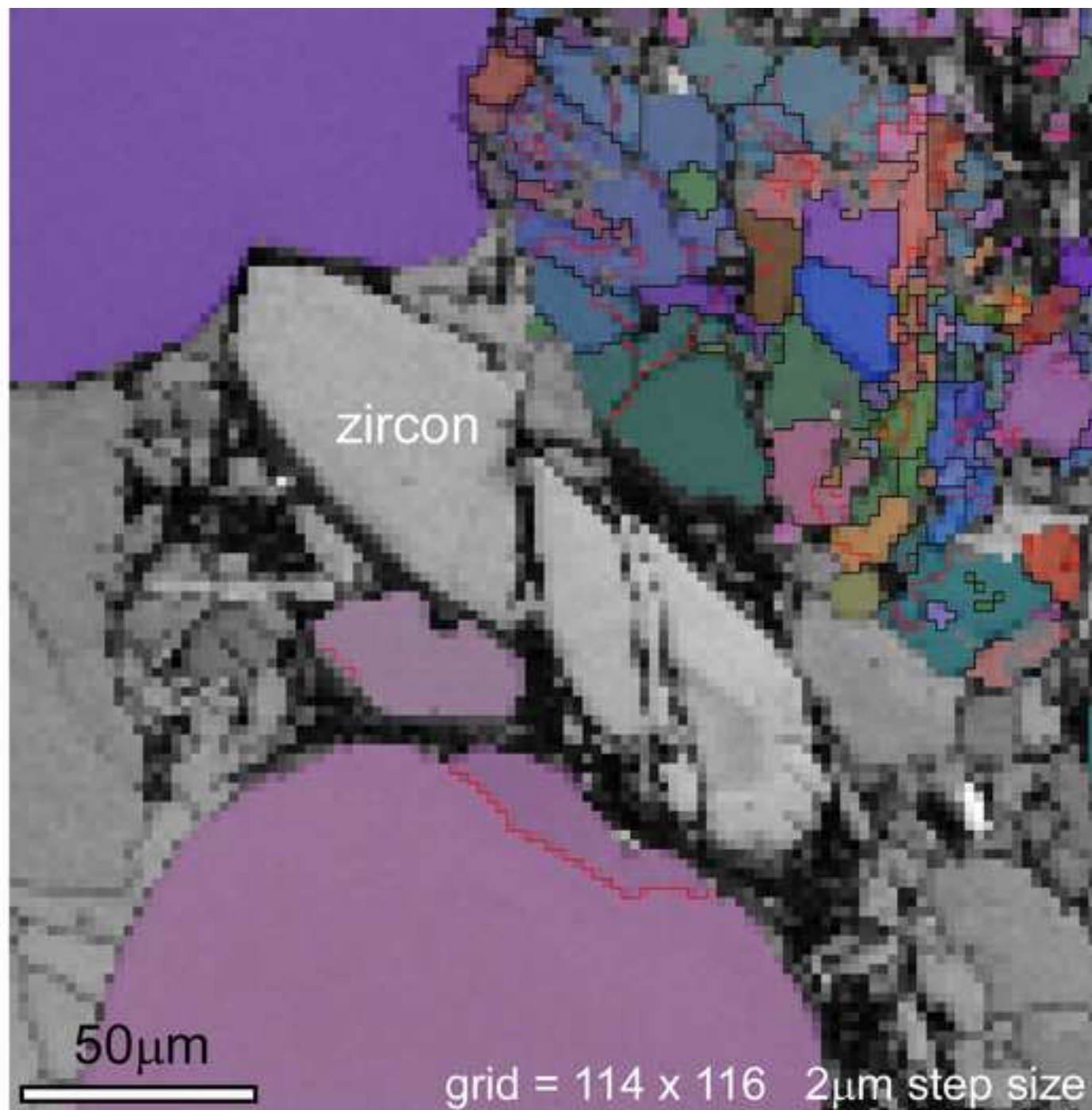


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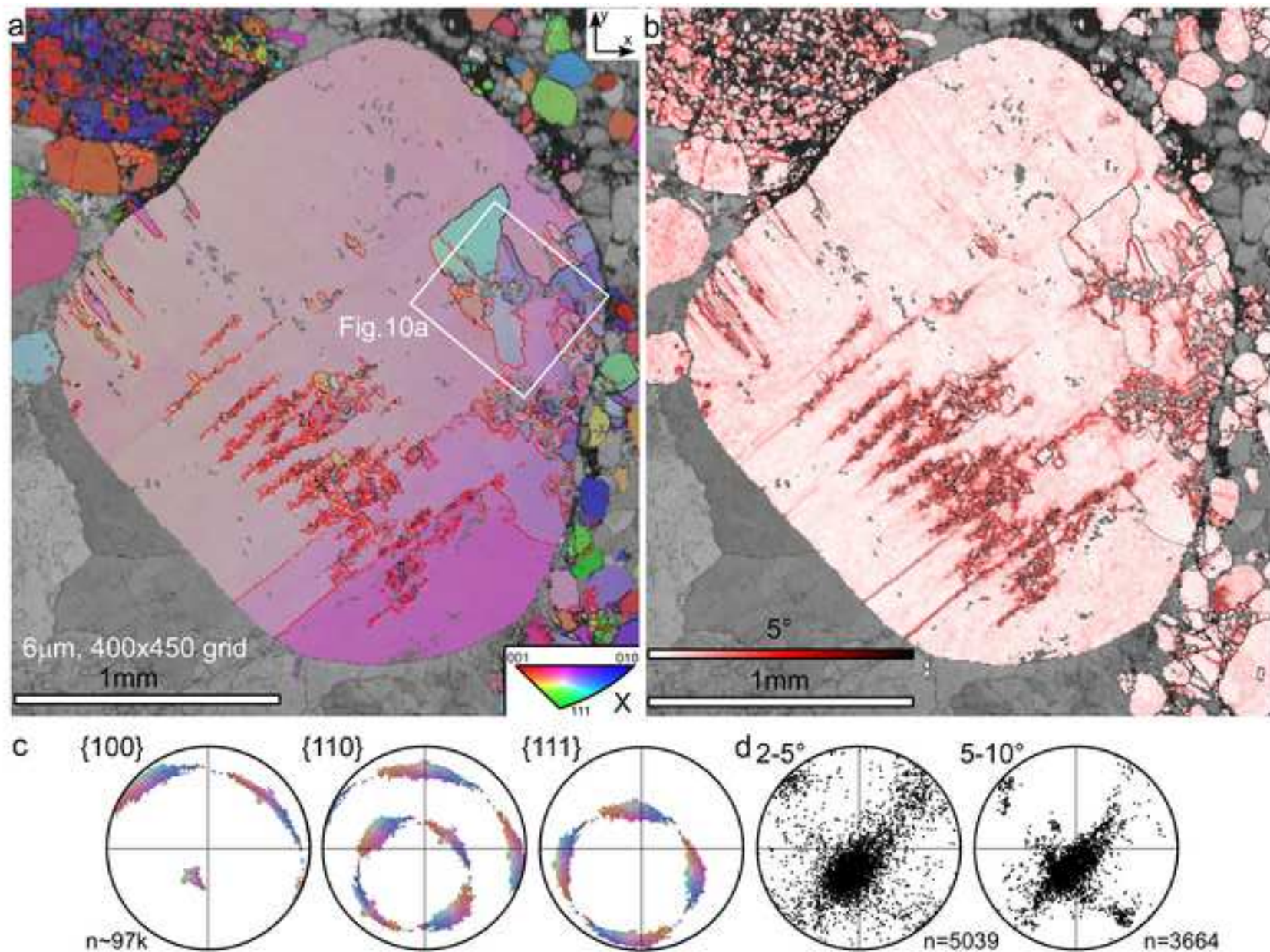
Figure 8

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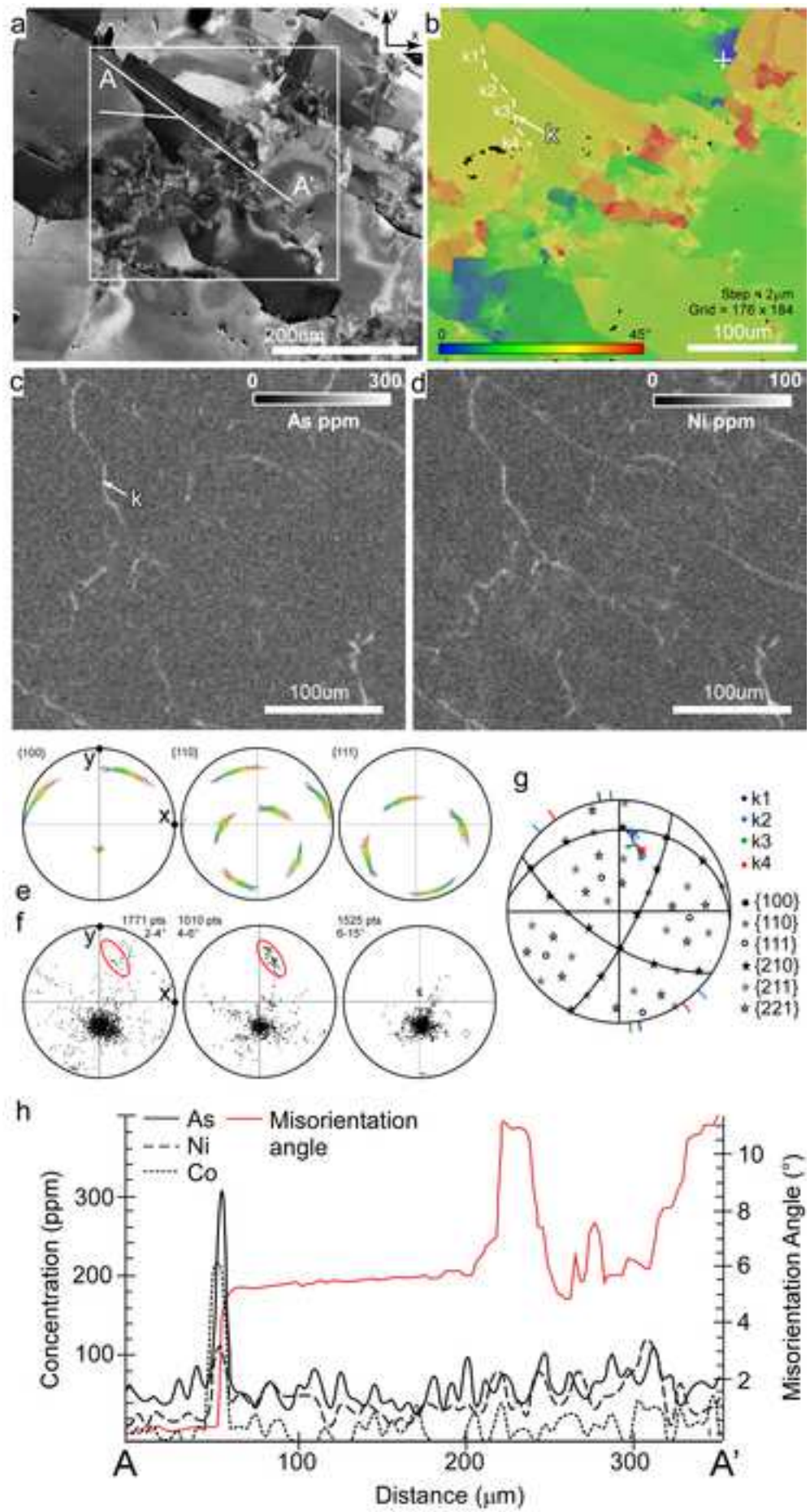
Reddy and Hough 2013. Figure 8
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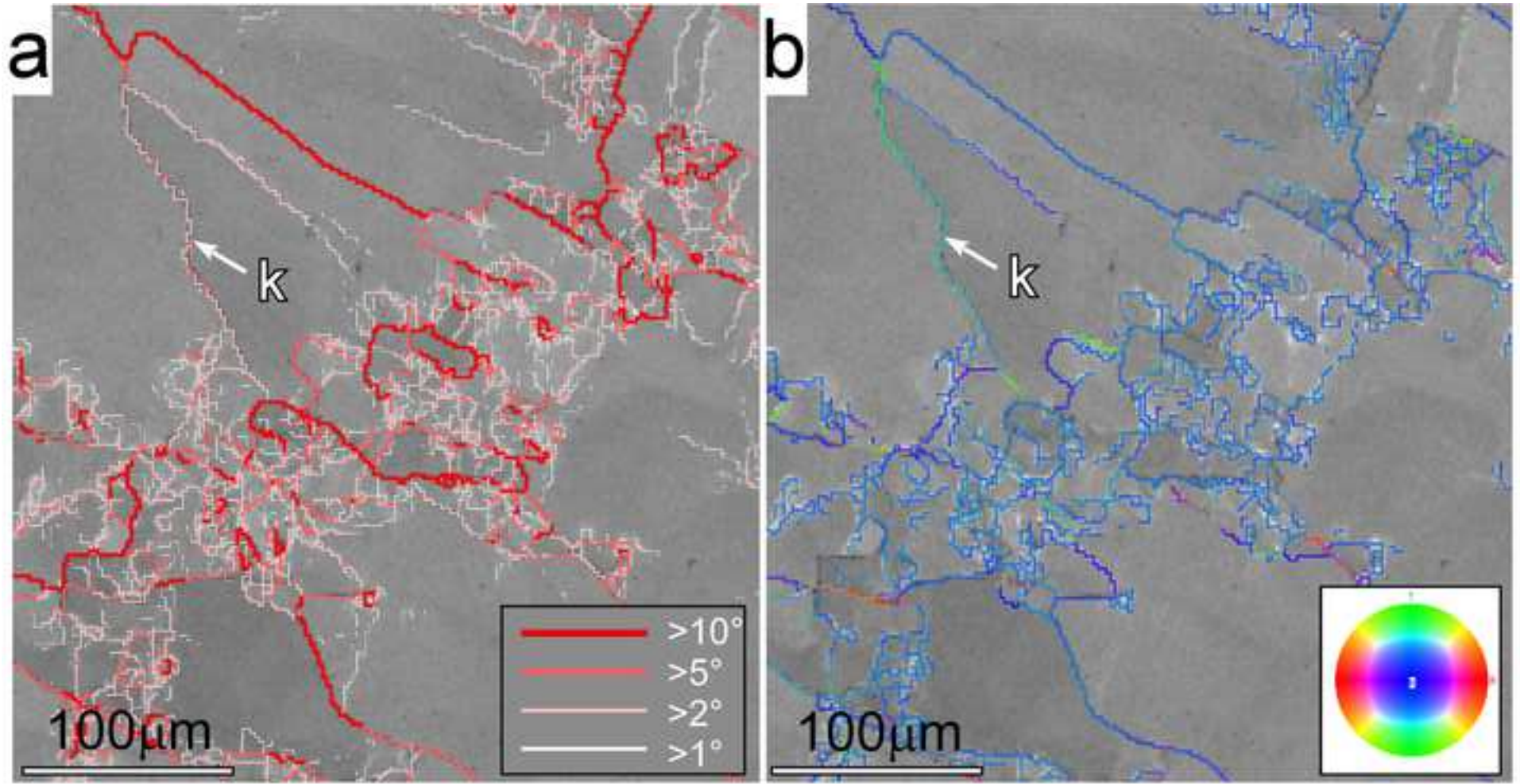
Figure 9
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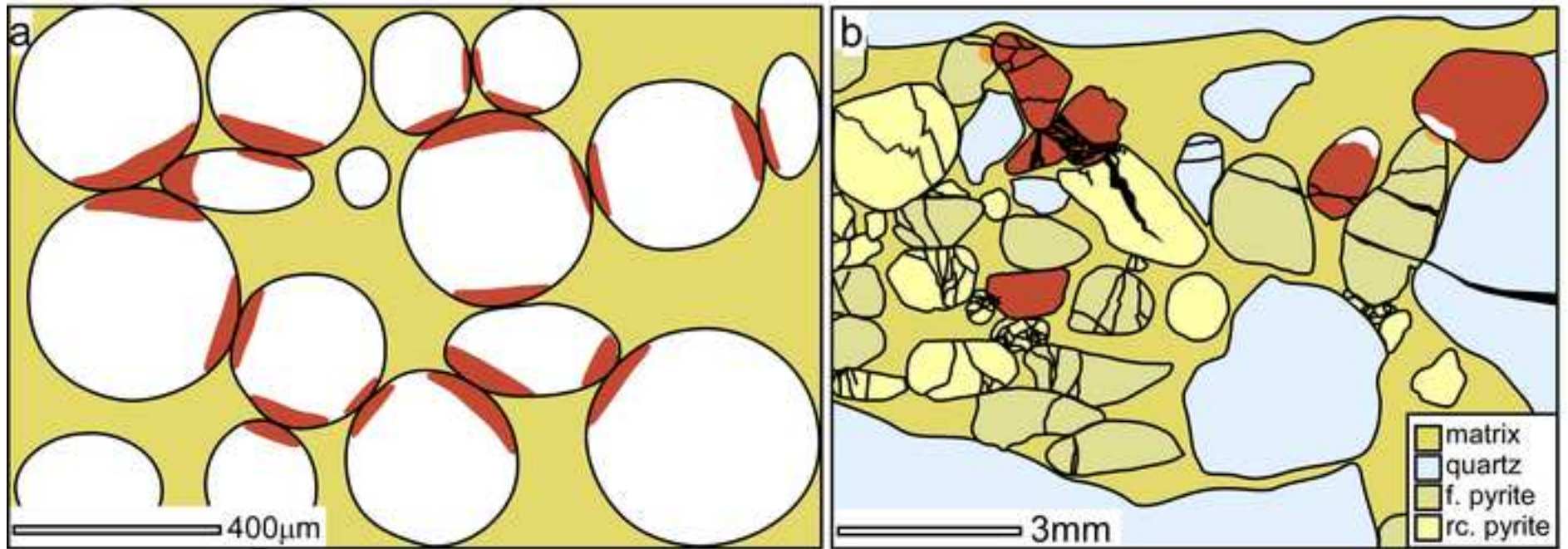
Reddy & Hough 2012: Figure 9
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Figure 10
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Reddy & Hough / Figure 11
Recommended width = current size



□ Dislocation glide dominated ■ Dislocation creep dominated (round compact pyrite) ■ DMT dominated

Reddy & Hough 2013/ Figure 12
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