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1	Using $\delta^{13}$ C to reveal the importance of different water
2	transport pathways in two nested karst basins, Southwest
3	China
4	
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17	Abstract: This study used carbon (C) isotope sourcing to determine transport
18	processes of dissolved inorganic carbon (DIC) from the land surface to river
19	catchments in Southwest China. Both nested karst watersheds investigated (Chenqi
20	and Houzhai) are representative of typical karst landform environments (e.g., primary

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21	forest, secondary forest, and farmland). We measured DIC concentrations and the
22	$\delta^{13}$ C values of rainfall, river water, groundwater, soil, and plants. To do so, we used
23	IsoSource (a Visual Basic program) to determine source partitioning over time
24	(seasonal) and across the two nested watersheds. In 2017, the mean DIC concentration
25	was 0.06 $\pm$ 0.03 mmol $L^{\text{-1}}$ and the rainfall $\delta^{13}C_{\text{DIC}}$ value was -14.4‰ $\pm$ 1.9‰. We
26	found similar DIC concentrations in the surface and groundwater of both watersheds,
27	ranging from 0.20 to 0.71 mmol L <sup>-1</sup> (seasonal) and from -3.7‰ to -9.4‰ ( $\delta^{13}C_{DIC}$ ) in
28	the Chenqi catchment and from 0.33 to 0.60 mmol $L^{-1}$ (seasonal) and from -10.3‰ to
29	-6‰ ( $\delta^{13}C_{DIC}$ ) in the Houzhai watershed. The average $\delta^{13}C$ values of soil and local
30	plants were -24.6 $\pm$ 1.4‰ and -28.9 $\pm$ 1.2‰ in the Chenqi catchment and -25.8 $\pm$
31	0.9‰ and -27.2 $\pm$ 1.8‰ in Houzhai watershed, respectively. In addition, carbonate
32	bedrock and groundwater were the main sources of surface water in the Chenqi and
33	Houzhai nested watersheds, both being greater than 30%. Source percentages were
34	~20% from atmospheric deposition and ~10% from soil. Furthermore, $HCO_3^-$ was the
35	predominant form of DIC (pH values > 8), and the contribution rates of dissolved
36	carbonate minerals (HCO3 <sup>-</sup> ) were approximately 10.4% and 19.6% in the Chenqi
37	catchment and the Houzhai watershed, respectively.

### **1. Introduction**

41 Continual increases in atmospheric carbon dioxide (CO<sub>2</sub>) that result in global
42 warming and associated climate changes necessitate a deeper understanding of global

**Keywords:** carbon cycle; source partition; hydrological pathway; catchment

carbon (C) cycles (Huang et al., 2015; Zhao et al., 2015; Shin et al., 2011; Liu et al., 43 2010). Rivers play the leading role in transporting C from terrestrial to marine 44 ecosystems in the global carbon cycle, being the main conduit of dissolved inorganic 45 carbon (DIC) from the land surface to the oceans (McClanahan et al., 2016). 46 Compared to organic and particulate fractions, DIC concentrations can provide more 47 information on C sources and processes involved in riverine C cycles (Li et al., 2010; 48 Cao et al., 2016; Brunet et al., 2009; Wachniew, 2006). The approximate global C flux 49 from DIC through river transport is  $0.4 \times 10^{15}$  gC yr<sup>-1</sup>, representing ~50% of the total 50 C flux (Cole et al., 2007). 51

DIC concentrations are influenced by water-air CO<sub>2</sub> exchange processes, 52 hydrologic inputs, vegetation, and carbonate weathering (McClanahan et al., 2016; 53 54 Shin et al., 2011; Tobias and Böhlke, 2011; Brunet et al., 2009). Sources of DIC in river water include inputs from soil CO<sub>2</sub> (through groundwater), atmospheric CO<sub>2</sub> 55 exchange, planktonic respiration, chemical weathering, and the dissolution of 56 57 carbonate rocks. Recent studies have used stable isotopes to trace DIC sources which are usually characterized by  $\delta^{13}C_{DIC}$  values. Each DIC source has a different  $\delta^{13}C_{DIC}$ 58 isotopic signature (from -26‰ to -9‰ for soil organic matter (SOM) (Wang et al., 59 2017a), -8% to -6% for atmospheric  $CO_2$  and approximately 0% for carbonate rocks 60 (Brunet et al., 2005). Moreover, C isotopes are used to assess riverine C 61 transformation as well as follow the riverine DIC transport into oceans (Zeng et al, 62 2015 Brunet et al., 2005; Deirmendjian and Abril, 2018; Hu et al., 2017; Ye et al., 63 2017). Monitoring changes in spatial and temporal DIC concentrations and  $\delta^{13}C_{DIC}$ 64

values from inland water systems can provide a better understanding of C sources. At the same time, this approach is useful in revealing reaction pathways and transportation processes which would be difficult to discern with the normative carbonate systems involved in C cycling in freshwater systems (Cao et al., 2016; Tallini et al., 2014).

Approximately 67% of the global DIC in rivers is known to originate from soil 70  $CO_2$  through  $C_3$  plants, the typical photosynthetic pathway (van Geldern et al., 2015). 71  $C_3$  plants have  $\delta^{13}C$  values that range from -22% to -30%, with an average 72 approximate value of -27‰ (Marx et al., 2018; van Geldern et al., 2015). Along with 73 photosynthesis, the main soil CO<sub>2</sub> transport pathways include rivers and streams. In 74 stream zones, the contribution of groundwater to water flow is significant and the 75 76 input of DIC-containing groundwater is derived from soil CO2. Soil CO2 tends to contain lower  $\delta^{13}C_{DIC}$  values in river water because soil CO<sub>2</sub> mainly derives from the 77 microbial degradation of SOM (Shin et al., 2011). 78

Karst zones are dynamic (McGee et al., 2010), being characterized by surface 79 erosion and extensive subsurface drainage (Song et al., 2017; Chang et al., 2015). 80 81 Moreover, karst aquifers are apt to form in areas of limestone and dolostone (Williams and Fong, 2010), wherein water is transported under conditions of dissolution. Water 82 transport that takes place in bedrock, sinkholes, sinking streams, and runoff then 83 percolates into groundwater and surface water (Lawhon. 2014). Such areas are 84 environmentally sensitive and are known for their rapid hydrologic and chemical 85 response to changing surface conditions (Yan et al., 2014; Li et al., 2012; Hartmann et 86

al., 2009; Macpherson et al., 2008). Karst landscapes comprise approximately 11.2%
of the surface area of the planet, namely, approximately 15 million km<sup>2</sup> (Song et al.,
2017; Dürr et al., 2005). Additionally, karst landscapes cover approximately 0.45
million km<sup>2</sup> in Southwest China (Yan et al., 2012). However, karst desertification (i.e.,
total soil loss) has increasingly become a serious problem in this region and could
potentially become a global problem (e.g., changing climatic patterns) (Song et al.,
2017).

Guizhou Province is suitable to study karst environments being located in the 94 95 center of the South China Karst, the largest karst area in the world (Liao et al., 2015; Wang et al., 2017b; Zeng et al., 2015). Moreover, karst water resources account for 80% 96 of the total water resources in this province (Li et al., 2018; Lu, 2007). This 97 98 subtropical region has abundant rainfall (with an average of 1340 mm) as well as distinct rainy (May to August) and dry seasons. In addition, the surface-groundwater 99 hydraulic system is a typical karst system, which facilitates the study of 100 rainfall-driven hydrological C processes and will help in our understanding of C 101 transport pathways and their implications. Accordingly, the objective of this study was 102 to determine the importance of different transport pathways in surface and 103 groundwater in a karst river basin (comprised of nested karst watersheds) using 104  $\delta^{13}C_{DIC}$  to trace C sources. However, the present researches generally focused on 105 relationship between surface water and groundwater or C sources of groundwater in 106 karst watersheds (Zeng et al., 2015; Deirmendjian and Abril 2018; Marx et al. 2018). 107 Therefore, the novelty of our research lies in using  $\delta^{13}C_{DIC}$  from precipitation, plant, 108

soil to water throughout the whole watershed to trace the C transport and analyzing
seasonal differences in C sources, thereby contributing to the further understanding of
C transport and storage in large karst basins.

112 **2. Materials and Methods** 

113 Study site

This study was conducted in the Houzhai watershed ( $80.65 \text{ km}^2$ ) ( $26^{\circ}15' \text{ N}$ ; 114 105°41' E) located in Guizhou Province, Southwest China (Figure 1a). The bedrock 115 type in the drainage area is predominantly limestone and dolomite of the Middle 116 117 Triassic Guanling Formation (Liu et al., 2010a, 2010b; Zhao et al., 2010). The drainage basin ranges in altitude between 1218 and 1585 m above mean sea level 118 (AMSL). This region is under the influence of a subtropical monsoon climate with an 119 annual mean temperature of 15.2°C (Yan et al., 2012). Annual average precipitation is 120 1314.6 mm, and 85% of rainfall occurs during the rainy season (from May to 121 October) (Yan et al., 2012). Tianlong Mountain (26°14'48" N; 105°45'51" E) is 122 located within the Houzhai watershed. The area of this mountain is 0.5 km<sup>2</sup> and its 123 altitude is 1460 m AMSL (Figure 1a). The mountain is covered by nearly 100% 124 natural forests where the average tree height is 6.26 m, the average shrub layer is 1.3 125 m, and the average herbaceous layer is 0.4 m. In the study area, the black "residual" 126 soil from 50 to 80 cm deep follows the Chinese soil taxonomic system. The Chenqi 127 catchment (105°42' E; 26°14' N), which has an area of 1.31 km<sup>2</sup>, is nested within the 128 Houzhai watershed (see Figure 1b). Annual mean temperature is 14.2°C, mean 129 precipitation is 1336 mm, and the altitude ranges from 1338 to 1491 m AMSL. The 130

131	land-use type of the Chenqi catchment is predominantly agricultural, including dry
132	crops (i.e., maize; 56%), rice paddies (14%), abandoned areas dominated by shrubs
133	(23%), and fruit trees (7%).

134 Water sampling

Surface and groundwater samples were collected from the Houzhai watershed and 135 the Chenqi catchment twice a month from June 28<sup>th</sup>, 2016, to May 23<sup>rd</sup>, 2017. Figure 136 1b shows the 10 sampling locations in the Houzhai watershed: No. 1 is comprised of 137 three sampling point (No. 1 surface water, No. 1G groundwater, and No. 1M the 138 139 surface-groundwater mixed sampling points) and No. 2 to No. 10 are all comprised of surface water sampling points. Figure 1c shows that the Chengi catchment is 140 comprised of four groundwater sampling points (No. 1G, No. 3G, No. 6G, and No. 141 142 8G), six surface water sampling points (No. 1 through No. 6), and one surface-groundwater mixed sampling point (No. 7). 143

Samples were collected in 100 mL polyethylene plastic bottles and stored prior to 144 analysis (4°C). Samples were filtered through a 0.45 µm membrane filter (Millipore) 145 (Jiangsu Jiuding Group Co., Ltd., China). Water samples used to determine DIC, 146 taken from the Millipore membranes, were thereafter heated to 80°C and maintained 147 at this temperature for 8 h to remove impurities. We determined DIC using a Vario 148 TOC Analyzer (Elementar, Langenselbold, Hesse, Germany) and  $\delta^{13}C_{DIC}$  using a 149 Finnigan MAT-252 mass spectrometer (Thermo Fisher Scientific, Darmstadt, Hesse, 150 Germany) We measured  $K^+$ ,  $Ca^{2+}$ ,  $SiO_2$ ,  $Na^+$ , and  $Mg^{2+}$  using an inductively coupled 151 plasma optical emission spectrometer (ICP-OES) (Thermo Fisher Scientific, 152

Darmstadt, Hesse, Germany) and Cl<sup>-</sup> and SO<sub>4</sub><sup>2-</sup> using an ion chromatograph (Thermo Scientific Aquion IC, USA). We measured electrical conductivity (COND), redox potential (ORP), total dissolved solids (TDS), and water acidity (pH) using Ultrameter II pH meters (Myron L company, Carlsbad, CA, USA). To collect rainfall samples, we installed a rain gauge on the roof of the station near No.1 sampling point in the test area of Houzhai watershed.

#### 159 **Plant and soil sampling**

We collected plant and soil samples from the Chengi catchment and the Tianlong 160 161 Mountain in July 2016. For the Tianlong Mountain, we chose four height gradients (A, B, C, and D), and each height gradient was divided into four separate soil layers (0-10, 162 10-30, 30-50, and 50-100 cm) for soil sampling. Plant samples were collected from 163 164 leaves, roots and twigs of different tree species. In the Chenqi catchment, two transects were established (Figure 1d) downslope on the hillside. Four quadrats were 165 established on each transect from which plant and soil samples were collected. A total 166 of 18 soil samples were collected from farmland, orchard, and forest land use types. 167 Continuous soil cores were taken and subsampled at depth intervals of S1 (0–10 cm), 168 S2 (10-30 cm), S3 (30-50 cm), and S4 (50-100 cm). Plant samples were collected as 169 mixed samples according to type (i.e., tree, shrub, or herb). These plant samples were 170 oven-dried at 60°C and then ground to  $<150 \mu m$ . Two milligrams of the subsamples 171 were prepared for stable isotope analysis, and soil samples were then air-dried and 172 ground to 150 µm. Finally, subsamples (3 g) were acidified (5% HCl) and then 173 analyzed for  $\delta^{13}$ C using the Finnigan MAT-252 mass spectrometer. 174

#### 175 Statistical analysis

#### 176 Isotopic mixing model

177 Utilizing one isotope system and three sources, we used the following mass balance 178 equations to determine the proportions ( $f_A$ ,  $f_B$ , and  $f_c$ ) of isotopic signature sources ( $\delta_A$ , 179  $\delta_B$ , and  $\delta_c$ ), which coincide with the observed signatures of the mixture ( $\delta_M$ ):

180 
$$\delta_M = f_A \delta_A + f_B \delta_B + f_C \delta_C$$

$$181 \qquad 1 = f_A + f_B + f_C$$

however, with n (A, B, C, ...) isotope systems and >n+1 sources, we were still
able to apply the requirement for mass balance conservation to determine multiple
combinations of source proportions that offer feasible solutions (Phillips and Gregg,
2003; Hao et al., 2018).

186 Stable isotope compositions of DIC in water, soil, and plant samples were 187 expressed as conventional delta notations, the difference between measured ratios of 188 the samples and references over the measured ratios of the references:

189 
$$\delta^{13} C_{\text{DIC}} (\%) = \left( \frac{R_{Sample} - R_{S \tan dard}}{R_{S \tan dard}} \right) \times 1000$$

where  $R_{sample}$  and  $R_{standard}$  represent <sup>13</sup>C/<sup>12</sup>C in the samples and the standard references. Carbon isotope data were reported on the Vienna Pee Dee Belemnite (VPDB) (‰) scale, with a standard deviation (1 $\sigma$ ) of 0.15‰ (Zhao et al., 2015).

Principal component analysis (PCA) and factor analysis were used with SPSS 17.0 software (SPSS Inc., Chicago, IL, the USA). The model outputs presented feasible ranges of isotope contributions given that the mass balance with qualitative statistics were calculated by IsoSource.

#### 198 **3 Results**

#### **3.1 Seasonal hydrochemical variations**

Figure 2a shows the seasonal variation of pH ranges in the Houzhai watershed. The 200 pH ranges were highest in autumn, namely, from 8.6 to 9.1 (except for the outliers) and 201 lowest in summer, namely, from 8.3 to 8.5, and from 8.2 to 8.7 and from 8.2 to 8.5 in 202 winter and spring, respectively. Figure 2b shows the seasonal variation of pH in the Chenqi 203 catchment. Compared to the Houzhai watershed, pH ranges in the Chenqi catchment were 204 205 narrow with the exception of spring. The highest values occurred in summer, namely, from 8.4 to 9.1 (except for the outliers) and the lowest in spring (from 7.8 to 8.2). Autumn and 206 winter pH values were similar. 207

208 Figure 2c shows linear relationships between pH, COND, TDS, and ORP of 21 rainfall events, respectively. We found that COND and TDS values were positively 209 correlated to pH values, while ORP values were negatively correlated to pH values. Table 1 210 211 shows the seasonal variation in water quality parameters of surface water, groundwater, 212 and mixed sampling points in the Houzhai and Chenqi nested watersheds. As Table 1 shows, seasonal variation of ORP and pH in surface water, groundwater and mixed water 213 were only slightly different in Houzhai watershed, while ORP values were much higher in 214 summer and spring in Chenqi catchment. Furthermore, pH values in the Houzhai 215 watershed were lower than Chenqi catchment with the exception of spring. Compared to 216 217 other seasons, only COND and TDS values were higher in surface water than groundwater in the Houzhai watershed. Additionally, COND and TDS values in Chenqi were obviously 218

higher than the Houzhai watershed all year round. However, ORP values were particularly
lower in the Chenqi catchment compared to the Houzhai watershed with the exception of
spring in surface water.

222

#### 3.2 Variations in dissolved inorganic carbon and $\delta^{13}C$

We collected data on 21 rainfall events from June 28<sup>th</sup>, 2016, to May 23<sup>rd</sup>, 2017, 223 and found that DIC values for eight rain events were nearly 0 mmol  $L^{-1}$ . As Figure 3a 224 shows, seasonal changes were significant, where the maximum concentration was 225 0.09 mmol  $L^{-1}$  in the spring of 2017 and the minimum concentration was nearly 0 226 mmol L<sup>-1</sup> in the autumn of the same year. The average DIC concentration for all rain 227 events was  $0.02 \pm 0.03$  mmol L<sup>-1</sup>. Additionally, DIC concentrations for rainfall events 228 in 2017 were significantly higher compared to 2016, where the average DIC 229 concentration for rain events from January to May 2017 was  $0.06 \pm 0.03$  mmol L<sup>-1</sup>. 230 Figure 3b shows variation in  $\delta^{13}$ C values for 11 rain events throughout 2016 to 2017. 231 Seasonal variation was not obvious, wherein the maximum was -12.3% in the spring 232 of 2017 and the minimum was -18.3% in the autumn of 2016. The average for all 11 233 rain events was  $-14.4\% \pm 1.9\%$ . 234

As Figure 4a shows, DIC concentrations in the Houzhai watershed ranged from 0.33 to 0.48 mmol  $L^{-1}$  during the summer, from 0.33 to 0.49 mmol  $L^{-1}$  during the autumn, from 0.39 to 0.58 mmol  $L^{-1}$  during the winter, and from 0.37 to 0.57 mmol  $L^{-1}$  during the spring. We found greater variation in DIC during the winter and spring compared to the summer and autumn. In sampling point No.1, surface water and groundwater exhibited similar variation, and DIC concentrations in groundwater were 241 higher than surface water.

Figure 4b shows DIC variation in the Chenqi catchment. We were able to obtain a 242 complete dataset of groundwater sampling points for all four seasons (winter, spring, 243 summer, and autumn). With the exception of No. 8G, namely, the groundwater 244 sampling points from the hilly area, all groundwater sampling points exhibited similar 245 variation, wherein the highest DIC concentration was measured in spring and lowest 246 in autumn. The No. 7 sampling point was also a surface and groundwater mixed 247 sampling point (where No. 6 and No. 6G were combined), and we found little 248 difference in DIC concentrations between surface and groundwater for all four 249 seasons. The maximum DIC concentration in the Chengi catchment was 0.71 mmol 250  $L^{-1}$ in spring and the minimum was 0.20 mmol  $L^{-1}$  in summer. 251

In both watersheds, DIC concentrations between surface water and groundwater were only slightly different. Moreover, we found higher concentrations for both during spring, when the averages were 0.50 and 0.56 mmol  $L^{-1}$ , and lower concentrations were measured during autumn, when the averages were 0.39 and 0.41 mmol  $L^{-1}$  in the Houzhai and Chenqi nested catchments, respectively. Surface water was abundant in summer due to high rainfall frequency and abated in the autumn and winter, while abundant groundwater was available all year around.

Figure 4c shows the 2016–2017  $\delta^{13}$ C values in the Houzhai watershed during the autumn, winter, and spring. The range in variation of  $\delta^{13}$ C values was from -10.3‰ to -6‰, and the average  $\delta^{13}$ C values in the autumn, winter, and spring were -9.0‰ ± 0.9‰, -8.1‰ ± 1.0‰, and -7.4‰ ± 0.9‰, respectively.  $\delta^{13}$ C values were more depleted in autumn and were more enriched in spring, Figure 4d shows that  $\delta^{13}$ C values in surface water and groundwater of the Chenqi catchment exhibited little difference, wherein the average  $\delta^{13}$ C values were -7.4‰ ± 2.0‰ in surface water and -7.2‰ ± 2.2‰ in groundwater. Furthermore, different months yielded obviously different values in sampling point No. 6, where the maximum value was -3.7‰ in surface water in August and the minimum value was -9.4‰ in groundwater in August due to frequent rainfall.

#### 269 **3.3 Variation in \delta^{13}C values for soil and vegetation**

Figure 5a shows the  $\delta^{13}$ C soil values of the four sampling depths of the different 270 height gradients (from A to D) in the Tianlong Mountain, wherein the lowest values 271 occurred in topsoil at all sampling points. The  $\delta^{13}$ C values in sampling points B and C 272 exhibited the same trend, namely, the deeper the soil layers were, the more depleted the 273  $\delta^{13}$ C values were. The minimum value was -27.0% for the 0 to 10 cm soil layer in 274 sampling point A, and the maximum value was -24.0% for the 50 to 100 cm soil layer in 275 sampling point B. Figure 5b and 5c show the differences in two transects of a ditch 276 between two hills in the Chengi catchment. For example, the  $\delta^{13}$ C value range in CQ-1 277 was from -28.3% to -22.6% and the  $\delta^{13}$ C value range in CQ-2 was from -27.6 % to 278 -22.0‰. The averages of CQ-1 in S1 (0-10 cm), S2 (10-30 cm), S3 (30-50 cm), and S4 279 (50–100 cm) were -24.9‰, -24.8‰, -24.2‰, and -24.5‰, respectively, and the averages 280 of CQ-2 were -25.5‰, -24.3‰, -24.7‰, and -24.6‰, respectively. We found that <sup>13</sup>C 281 values were more enriched in the deeper soil layers (10-50 cm). 282

Figure 6a shows the  $\delta^{13}$ C values of the four plant types in the Tianlong Mountain, which is near sampling point No. 8 in the Houzhai watershed. The highest  $\delta^{13}$ C value was -24.6‰ (cedar wood) and the lowest  $\delta^{13}$ C value was -27.6‰ (*Broussonetia papyrifera*) among the four plant types. Figure 6b shows the mixed samples of trees divided into leaves, twigs, and roots, wherein the average  $\delta^{13}$ C values were -30.2‰ ± 0.3‰, -28.7‰ ± 0.6‰, and -27.7‰ ± 0.5‰, respectively. Therefore, <sup>13</sup>C concentrations increased in the roots of plants.

290

291 **4. Discussion** 

#### 292 4.1 Inorganic river chemistry

Dissolved and particulate C is likely associated with the chemical weathering of 293 basin soil and bedrock (McClanahan et al., 2016). COND is an indicator that reflects 294 ion changes under some conditions. Higher COND values indicate higher ionic 295 296 concentrations in water bodies, which may stem from the dissolution of carbonate minerals. Both surface water and groundwater in the Chenqi catchment (TDS values: 297 510~723 mg·L<sup>-1</sup>) were much more carbonate-rich compared to the Houzhai watershed 298 (TDS values:  $262 \sim 432 \text{ mg} \cdot \text{L}^{-1}$ ) all year around (Table 1). This indicated that there was 299 not only an increase in karst dissolution processes adding to the dissolved constituents 300 in the water, but other potential sources along the surface water path were 301 contributing COND and TDS values in karst water. These potential sources included 302 overland flow and anthropogenic impacts which also have influences on ionic 303 concentrations (COND) especially agricultural activities including potassium, nitrates, 304 and phosphates fertilizers (Lawhon, 2014; Moore et al., 2009). 305

306

The nature of redox reactions and redox-sensitive aqueous species is quite

different from most other reactions in aquifer systems (Liu et al., 2017). Therefore, it 307 is necessary to set a unique redox status indicator in aquifer systems (Kumar and 308 309 Rivazuddin, 2012). In water, ORP is strongly associated with temperature, pH, salinity, and dissolved oxygen concentrations (Liu et al., 2009; Li et al., 2014). The greater the 310 dissolved oxygen (DO) is, the higher the ORP will be. As Table 1 shows, the Houzhai 311 watershed yielded higher ORP compared to the Chenqi catchment; thus, its living 312 conditions were better suited for aquatic organisms, when considering variations in 313 pH and DO simultaneously. 314

In karst areas, such as the Houzhai and Chenqi nested watersheds, given the abundant carbonate bedrock, carbonates will help remove  $CO_2$  from the system. Under most conditions,  $CO_2$  is consumed because carbonate minerals dissolve in H<sub>2</sub>O-CO<sub>2</sub> solutions, and pH will buffer to higher values due to adjustments in relevant reactions (chemical equilibrium).

As Figure 2c and Table 1 show, pH values were generally high in the sampling 320 321 areas, and pH values in the Chenqi catchment were higher than in the Houzhai 322 watershed. The CO<sub>2</sub> that escaped from streams in the Chenqi catchment reacted much more significantly than river water in the Houzhai watershed. Guizhou Province has a 323 rich karst landscape, including carbonates and other minerals; therefore, water quality 324 325 parameters, such as COND and TDS, are much higher in water than other areas. Moreover, these parameters have a direct effect on pH values in that they are higher 326 than other areas (McClanahan et al., 2016; Lawhon, 2014; Mondal et al., 2010). 327 However, previous literature reported that karst water bodies had been found to have 328

higher pH ranges in general (Hatcher, 2013).

## 330 4.2 Using $\delta^{13}$ C values to distinguish between sources of dissolved inorganic 331 carbon

Many internal and external factors, such as soil respiration, carbonate mineral 332 dissolution, atmospheric precipitation, and the pH value of water bodies, etc., have an 333 effect on  $\delta^{13}C_{DIC}$  in water catchments (Shin et al., 2011; Doctor et al., 2008). In this 334 study, the average  $\delta^{13}$ C values of surface water were lower than groundwater in the 335 Houzhai and Chenqi nested watersheds. This could be explained in that groundwater 336 may be derived more from "old" water in matrix pores that is more enriched in 337  $\delta^{13}C_{DIC}$  due to the longer time required for CaCO<sub>3</sub> dissolution to take place, while 338 river water has many sources, including groundwater inputs, atmospheric deposition, 339 soil water, and carbonate reactions. In order to determine  $\delta^{13}C$  proportions from 340 various sources, the isotope model that we used to calculate results from IsoSource is 341 provided for the Houzhai and Chenqi nested watersheds (see Figure 7). As it pertains 342 343 to river water (surface water) and groundwater time spans, atmospheric deposition results are typically taken from long-term monitoring, where, for example, average 344 soil  $\delta^{13}$ C values are from -21% to -22% in karst areas of USA (McClanahan et al., 345 2016). In China, the average soil  $\delta^{13}$ C value is -24‰ (Zhao et al., 2015). 346 In this study, soil water was assumed to be -24‰ and carbonate mineral sources 347 were hypothesized to be 0‰ (Das et al., 2005; Hu et al., 2017; Li et al., 2010). The 348

main sources in the Chenqi catchment were groundwater and carbonate reactions, wherein the mean percentages were 31.7% and 38.1%, respectively; the minimum

source was soil water, with a mean percentage of only 10.1% during the summer 351 (Figure 7). In autumn, when stream flows were lower, percentages of groundwater 352 353 were higher compared to carbonate reactions (Yan et al., 2014; Zhao et al., 2010). The proportion in the Houzhai watershed was similar to that of the Chengi catchment, 354 wherein the main sources were groundwater and carbonate reactions, which showed 355 little differences between seasons. The mean proportions of the four sources were 32.4% 356 in the autumn and 35.5% in the spring from groundwater, 20.5% in the autumn and 357 18.8% in the spring from atmospheric deposition, 11.4% in the autumn and 10.5% in 358 359 the winter from soil water, and 35.7% in the autumn and 34.6% in the spring from carbonate reactions. In autumn, rainfall contributed more, which reduced the overall 360 groundwater impact. Furthermore, winter and spring showed little seasonal 361 362 differences in the difference in Houzhai watershed.

Figure7 shows the mean percentage of each source. All values are provided in 363 Table 2, which also summarizes seasonal trends. To present model results more 364 365 accurately, median values of model averages are provided along with the standard. The median value of carbonate reactions was highest  $(38\% \pm 13\%)$  in summer and 366 groundwater was highest  $(32\% \pm 24\%)$  in autumn. The lowest DIC source was found 367 in soil water, which likely derived from the interflow of water being infiltrated along 368 slopes within the basin (9%  $\pm$  8%) in summer and (11%  $\pm$  9%) in the Chenqi 369 catchment in autumn. In summer, rainfall was more frequent and subsequently surface 370 371 water was more abundant; thus groundwater and soil water were lower than in the autumn. We found little differences between seasons in the Houzhai watershed. The 372

highest mean for all *in situ* carbonate reactions was 36%, and the lowest mean for allsoil was 8% as shown in Table 2.

# 4.3 Influence and contribution rates of carbonate dissolution on dissolved inorganic carbon concentrations

River chemistry is strongly influenced by carbonate dissolution, which has an 377 obvious effect on pH value and supplies a large quantity of DIC in the form of 378 bicarbonates (Schulte et al., 2011). The study area lies within the carbonate area of the 379 region, and the distribution of pH in groundwater ranged from 7.9 to 9.0. Therefore, 380 the type of inorganic C in water is predominantly  $HCO_3^-$  (Doctor et al., 2007). COND 381 and pH values are shown in Table 1, which were completely consistent for both 382 sampling points. They showed that carbonate minerals control dissolution, including 383 384 the main DIC components in both the Chenqi catchment and the Houzhai watershed.

Cations ( $Ca^{2+}$ ,  $Mg^{2+}$ ,  $K^+$ , and  $Na^+$ ) and anions ( $HCO_3^-$ ,  $Cl^-$ , and  $SO_4^{2-}$ ) were the 385 main ions found in the Chenqi catchment and the Houzhai watershed (Table 3). The 386 Piper diagram reflects not only the chemical composition of river water but also 387 distinguishes between different weathered sources of species composition (Stallard et 388 al., 1983). In the cationic ternary diagram (Figure 8), evaporite mineral weathering 389 products fall on the high  $Na^++K^+$  line, while limestone weathering products fall on 390 Mg<sup>2+</sup>-Ca<sup>2+</sup> line. Due to dolomite characteristics, weathering products fall in the 391 middle of the  $Mg^{2+}$  and  $Ca^{2+}$  (Ca: Mg = 1:1) lines(Li et al., 2010). 392

Furthermore, the weathering products of silicate minerals fall on the  $Mg^{2+}-Ca^{2+}$ line, which is relatively biased towards one side of  $Na^++K^+$ . In the anion ternary graph,

the weathering material of pure carbonates is primarily  $HCO_3^{-1}$ ; thus, the data points 395 fall at the relatively high end of HCO3<sup>-</sup>. The weathering products of evaporated salt 396 minerals fall on the Cl<sup>+</sup>  $SO_4^{2-}$  line, and the proportion is relatively high, while the 397 weathering of silicate minerals causes both  $HCO_3^-$  and  $SiO_2$  to be present in river 398 systems. In this study, the proportion of  $Ca^{2+}$  and  $Mg^{2+}$  were higher than  $Na^++K^+$ . In 399 particular,  $Ca^{2+}$ ,  $HCO_3^{-}$ , and  $SO_4^{2-}$  were obviously higher than  $Cl^{-}$  and  $Mg^{2+}$  in the 400 Chenqi catchment and the Houzhai watershed, which suggested that river water in 401 these two nested watersheds are controlled by carbonates, silicate minerals, and small 402 amounts of evaporites. 403

In order to further quantify the influence of different rock types on the main ions 404 found in rivers, we used the main ion concentrations ( $Ca^{2+}$ ,  $Mg^{2+}$ ,  $K^+$ ,  $Na^{2+}$ ,  $Cl^-$ ,  $SiO_2$ , 405 406 and  $HCO_3$ ) of surface water in the Chenqi catchment and the Houzhai watershed for PCA and factor analysis of river water chemistry as shows in Table3 and Figure 9. In 407 the Chenqi catchment, as Figure 9 shows, the cumulative contribution rate of the first 408 three variables reached 79.6% (<80%), but eigenvalues from the extraction factor 409 were less than 1 of the fourth variable, and therefore deleted. Factor 1 was highly 410 correlated to Ca<sup>2+</sup>, Mg<sup>2+</sup>, Na<sup>2+</sup>, and SiO<sub>2</sub>, indicating that the dissolution of silicate 411 minerals contributes to chemical weathering in the sampling areas. Factor 2 was more 412 significantly correlated to  $K^+$  and  $HCO_3^-$ , and Factor 3 was more significantly 413 correlated to  $Cl^{-}$  and  $SO_{4}^{2-}$ , indicating that the contribution of dissolved carbonate 414 minerals was relatively high compared to silicate minerals (Figure 9). Bare carbonate 415 rocks accounted for almost 75% of the study area, which suggested that carbonate 416

dissolution should play a key role in rock weathering (Cao et al., 2005; Li et al., 2010). 417 In the Houzhai watershed, as Figure 9 shows, the cumulative contribution rate of the 418 first three variables reached 81.2% (>80%), and the relationship between each factor 419 was similar to the Chengi catchment. The squares of each variable factor load divided 420 by the common variance is the relative variance contribution of dissolution of each 421 type of rock to each variable. After calculations, the contribution rates of dissolved 422 carbonate minerals to  $HCO_3^-$  were approximately 10.4% and 19.6% in the Chenqi 423 catchment and the Houzhai watershed, respectively. 424

425 4.4

## 4.4 $\delta^{13}$ C characteristics of vegetation and soil

The  $\delta^{13}$ C values of the C<sub>3</sub> and C<sub>4</sub> photosynthetic pathway plants were distinctly 426 different, namely, the average value of C<sub>3</sub> plants is -27.1‰ (from -21‰ to -35‰) and 427 the average value of C<sub>4</sub> plants is -13.1‰ (from -10‰ to -14‰) (Rouw et al., 2015). In 428 our study, all plant  $\delta^{13}$ C values in the Houzhai watershed were from -24.6% to -27.6% 429 and all plant  $\delta^{13}$ C values in the Chengi catchment were from -27.7% to -30.2%. For 430 CQ-1, soil  $\delta^{13}$ C values were more enriched in the deeper soil layers, with an order 431 rank of S1<S2<S4<S3, and the order rank of soil  $\delta^{13}$ C values in CQ-2 was S1< 432 S3<S4<S2. The reason for this could potentially be that surface soil is easily 433 transported by rain, promoting nutrient accumulation in deeper soil layers (10–50 cm) 434 in karst areas. The 0–10 cm soil layer in these soil profiles had the lowest  $\delta^{13}$ C values 435 of the corresponding profile. This is due to the fact that the top of the soil profile had 436 a relatively large amount of undecomposed plant residue and better preserved plant 437  $\delta^{13}$ C signatures, while underneath the surface soil, due to the uneven distribution of 438

439 plant roots, different degrees of plant residue decomposition and differences in 440 organic or acid species in the soil itself, etc., could lead to differences in  $\delta^{13}$ C soil 441 sample values (Wynn et al., 2006; Wang et al., 2008).

The average  $\delta^{13}$ C value of SOC for the four soil profiles was -24.6‰±1.4‰ in the 442 Chengi catchment, with an  $\delta^{13}$ C enrichment value of 4‰ compared to the weighted 443 average  $\delta^{13}$ C value of local plants (-28.9‰±1.2‰). This trend was also seen in the 444 Houzhai watershed, with an average value of -25.8‰±0.9‰, an increase of nearly 2‰ 445 compared to local plants. During isotope fractionation, which occurs through 446 microbial decomposition of organic matter and leads to the uneven distribution of <sup>13</sup>C 447 and <sup>12</sup>C during the different C phases between reactants and products, there is 448 relatively more  ${}^{12}C$  in released CO<sub>2</sub>, whereas  ${}^{13}C$  is more prevalent in C derived from 449 microbial biomass, eventually being returned as SOM (Luo and Wang, 2009; Li et al., 450 2012). Furthermore, even for individual plants, <sup>13</sup>C tends to concentrate in root 451 systems followed by twigs and leaves, and this is because the lighter <sup>12</sup>C isotope exits 452 453 plants through transpiration. In order to fully understand C cycling, more research is required on interannual changes, photosynthesis and transpiration of  $C_3$  and  $C_4$  plants, 454 and soil microbial respiration to better understand internal C cycling dynamics that 455 could influence DIC sources, transportation, and storage, which could even lead to an 456 improvement in global C sink estimations (Figure 10). 457

458

#### 459 **5. Conclusions**

460 Carbonate karst environments may play an indispensable role in the global C

budget. Investigations into the impacts of karst watersheds such as C transportation 461 and storage within this system on a local level are vital to understanding the global C 462 cycle. International and domestic academics have provided credible scientific data 463 regarding many aspects of karst watersheds in China. However, additional research is 464 needed for a better comprehensive understanding of DIC and its sources in this type 465 of system. In this study, carbonate reactions and groundwater were the main sources 466 of river water in Houzhai and Chenqi watersheds, whose proportions were both 467 greater than 30% in river water. Sources proportions were influenced by seasonality 468 469 which related to the frequency of rainfall, especially regarding groundwater and soil water. Furthermore, DIC mainly exists in the form of  $HCO_3^-$  in the sampling areas of 470 this study, of which the contribution rate of dissolved carbonate minerals (HCO $_3^{-}$ ) was 471 472 approximately 10.4%-19.6%. Silicate minerals and evaporites had a negligible effect on DIC in the nested watersheds investigated in this study. The amount of carbonate 473 mineral dissolution had a significant influence on surface water and groundwater (in 474 475 the form of a C sink) under karst hydrological conditions. Additionally, water-soil and soil-plant processes together comprised the complete karst eco-hydrological C cycle 476 in this study which is not mentioned in other studies at present. More relevant studies, 477 however, are required in order to gain a more comprehensive understanding of 478 biogeochemical cycles in large karst-influenced watersheds. 479

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705	Table 1 Water quality parameters of surface water, groundwater and mixed water in
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Sample		Summer			Autum	n			
	COND	TDS	ORP	DU	COND	TDS	ORP	DU	
Point	$(uS cm^{-1})$	$(mg.L^{-1})$	(mv)	PH	$(uS cm^{-1})$	$(mg.L^{-1})$	(mv)	РН	
HZ-S	410	276	216	7.9	391	262	187	8.6	
HZ-G	397	266	219	7.9	423	285	186	8.8	
HZ-M	403	271	216	8.3	456	308	181	8.8	
CQ-S	452	510	213	9.0	836	581	124	8.9	
CQ-G	491	586	194	9.0	780	542	130	9.0	
CQ-M	747	514	203	9.0	1042	723	134	8.9	
	Winter				Spring				
Sample		Winter				Sprin	g		
Sample	COND	Winter TDS	ORP	DII	COND	Sprin <sub>.</sub> TDS	g ORP	DU	
Sample Point	COND (uS cm <sup>-1</sup> )	Winter TDS (mg.L <sup>-1</sup> )	ORP (mv)	РН	COND (uS cm <sup>-1</sup> )	Sprin TDS (mg.L <sup>-1</sup> )	g ORP (mv)	РН	
Sample Point HZ-S	COND (uS cm <sup>-1</sup> ) 396	Winter TDS (mg.L <sup>-1</sup> ) 266	ORP (mv) 211	PH 8.4	COND (uS cm <sup>-1</sup> ) 467	Sprin, TDS (mg.L <sup>-1</sup> ) 315	g ORP (mv) 208	РН 8.3	
Sample Point HZ-S HZ-G	COND (uS cm <sup>-1</sup> ) 396 451	Winter TDS (mg.L <sup>-1</sup> ) 266 304	ORP (mv) 211 212	PH 8.4 8.5	COND (uS cm <sup>-1</sup> ) 467 635	Sprin, TDS (mg.L <sup>-1</sup> ) 315 432	g ORP (mv) 208 219	РН 8.3 8.2	
Sample Point HZ-S HZ-G HZ-M	COND (uS cm <sup>-1</sup> ) 396 451 405	Winter TDS (mg.L <sup>-1</sup> ) 266 304 272	ORP (mv) 211 212 211	PH 8.4 8.5 8.5	COND (uS cm <sup>-1</sup> ) 467 635 626	Sprin, TDS (mg.L <sup>-1</sup> ) 315 432 426	g ORP (mv) 208 219 223	PH 8.3 8.2 8.3	
Sample Point HZ-S HZ-G HZ-M CQ-S	COND (uS cm <sup>-1</sup> ) 396 451 405 770	Winter TDS (mg.L <sup>-1</sup> ) 266 304 272 526	ORP (mv) 211 212 211 125	PH 8.4 8.5 8.5 8.9	COND (uS cm <sup>-1</sup> ) 467 635 626 838	Sprin, TDS (mg.L <sup>-1</sup> ) 315 432 426 573	g ORP (mv) 208 219 223 218	PH 8.3 8.2 8.3 7.9	
Sample Point HZ-S HZ-G HZ-M CQ-S CQ-G	COND (uS cm <sup>-1</sup> ) 396 451 405 770 740	Winter TDS (mg.L <sup>-1</sup> ) 266 304 272 526 514	ORP (mv) 211 212 211 125 127	PH 8.4 8.5 8.5 8.9 9.1	COND (uS cm <sup>-1</sup> ) 467 635 626 838 855	Sprin, TDS (mg.L <sup>-1</sup> ) 315 432 426 573 596	g ORP (mv) 208 219 223 218 217	PH 8.3 8.2 8.3 7.9 8.1	

707	(Note: pH, COND, TDS and ORP were measured by Ultrameter II (Myron L company USA.
708	HZ-S values represent the average for all the surface water points (No. 1 through No. 10) in the
709	Houzhai watershed; HZ-G values represent the average for No.1G in different months; HZ-M
710	values represent the average for No.1M in different months. CQ-G values represent the average of
711	the four groundwater sampling points in different months; CQ-S values represent the average of
712	the six surface water sampling points in different months and CQ-M values represent the average
713	for the No.7 in different months.)

Houzhai and Chenqi.

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716	
717	Table 2 Seasonal trends of model results for Chenqi and Houzhai

		G	Ground water Atmosphere deposition		Soil water		Carbonate reaction		
		value	median standard	value	median standard	value	median standard	value	median standard
(Chenqi)	Median	28%	23%	18%	15%	9%	8%	38%	13%
Summer	Minimum	0%		0%		0%		6%	
	Maximum	94%		60%		31%		69%	
	Median	32%	24%	20%	16%	11%	9%	29%	14%
Fall	Minimum	0%		0%		0%		0%	
	Maximum	93%		64%		37%		61%	
(Houzhai)	Median	29%	23%	18%	15%	10%	8%	36%	13%
Fall	Minimum	0%		0%		0%		4%	
	Maximum	95%		61%		34%		65%	
	Median	30%	25%	17%	14%	9%	8%	36%	15%
Winter	Minimum	0%		0%		0%		0%	
	Maximum	97%		57%		30%		67%	
	Median	32%	24%	17%	13%	10%	8%	36%	16%
Spring	Minimum	0%		0%		0%		0%	
	Maximum	92%		53%		31%		66%	

(N**3t9** IsoSource calculation automatic generated results. Groundwater, atmosphere deposition, soil water and carbonate **72t** actions as the four sources of river water in this study. The median, minimum and maximum of possibility within **72th** river routinely produced and median standard contributions of 20–25% and 8–16%, respectively).

#### 731 Table 3 Cations and anions concentrations of surface water, groundwater in Houzhai

and	Chen	qi.
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Points	Summer			Autumn				
Concentrations (mmol • $L^{-1}$ )	CQ-S	CQ-G	HZ-S	HZ-G	CQ-S	CQ-G	HZ-S	HZ-G
Ca <sup>2+</sup>	4.55	2.86	1.06	1.28	6.73	2.89	1.74	1.01
$Mg^{2+}$	2.17	1.13	0.83	0.91	3.38	1.33	0.96	0.87
$Na^+$	0.29	0.08	0.13	0.15	0.32	0.09	0.19	0.16
$\mathbf{K}^+$	0.38	0.06	0.07	0.07	0.13	0.05	0.11	0.08
$SO_4^{2-}$	1.36	2.97	0.60	0.60	2.68	1.31	0.90	0.60
Cl	0.15	0.19	0.19	0.30	0.18	0.22	0.25	0.21
HCO <sub>3</sub> <sup>-</sup>	0.45	0.45	0.51	0.57	0.39	0.41	0.54	0.50
SiO <sub>2</sub>	0.19	0.14	0.02	0.05	0.28	0.14	0.09	0.07
season	Winter			Spring				
Point	CQ-S	CQ-G	HZ-S	HZ-G	CQ-S	CQ-G	HZ-S	HZ-G
Ca <sup>2+</sup>	2.44	2.64	1.19	1.00	3.41	3.34	2.25	1.98
$Mg^{2+}$	1.16	1.38	0.78	0.86	1.30	1.74	1.23	1.19
$Na^+$	0.07	0.09	0.21	0.20	0.15	0.59	0.37	0.54
$\mathbf{K}^+$	0.03	0.03	0.10	0.05	0.16	0.06	0.15	0.07
$SO_4^{2-}$	1.97	2.61	1.19	0.77	3.10	2.44	1.61	0.99
Cl	0.17	0.11	0.31	0.31	0.23	0.13	0.39	0.39
HCO <sub>3</sub> <sup>-</sup>	0.39	0.44	0.52	0.57	0.55	0.55	0.56	0.58
SiO <sub>2</sub>	0.13	0.14	0.06	0.06	0.17	0.15	0.05	0.06

733 (Note: HZ-S values represent the average for all the surface water points (No. 1 through No. 10) in

the Houzhai watershed; HZ-G values represent the average for No.1G in different months. CQ-G

values represent the average of the four groundwater sampling points in different months; CQ-S

values represent the average of the six surface water sampling points in different months.)

743	
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746	Figure Caption
747	Eigure 1 (a) Study sites leastion in Puding County Cuickou (b) The water compling
/4/	Figure 1 (a) Study sites location in Fuding County, Guizhou. (b) The water sampling
748	points in Houzhai watershed. (c) The water sampling points in Chenqi catchment. (d)
749	The soil sampling points in Chenqi catchment.
750	Figure 2 (a) The seasonal variation of pH values in Houzhai. (b) The seasonal
751	variation of pH values in Chenqi.(c) Correlation between pH with water quality
752	parameters of atmospheric deposition from 2016 to 2017.
753	Figure 3(a) Variation of [DIC] of rainfall during 2016 to 2017. (b) The variation of
754	$\delta^{13}C_{DIC}$ of rainfall during 2016 to 2017.
755	Figure 4 (a) The seasonal variation of DIC in Houzhai watershed. (b) The seasonal
756	variation of DIC in Chenqi catchment. (c) Seasonal variation of $\delta^{13}$ C values in
757	Houzhai watershed. (d) The variation of $\delta^{13}C$ values in surface water and groundwater
758	in Chenqi catchment.
759	Figure 5(a) The $\delta^{13}$ C values of soil collected at four different locations (A to D) on
760	Tianlong Mountain. The $\delta^{13}$ C values of two transects of a ditch between two hills with
761	four depths in Chenqi (b) CQ-1 and (c) CQ-2.
762	Figure 6 (a) The $\delta^{13}$ C values of arbor plants in Tianlong Mountain. (b) The $\delta^{13}$ C values
763	of leaves, twig and roots of tree in Chenqi.
764	Figure 7 Season series analysis of mixing model results for Chenqi and Houzhai.
765	Figure 8 The Piper diagram of cationic and anion in surface water and groundwater in
766	Chenqi and Houzhai watershed.
767	Figure 9 Component loadings of principal component analysis of Chenqi and Houzhai
768	watershed

- Figure 10<sup>13</sup>C cycle in a nested karst watershed and proportion of carbon sources in the
- river.

Figure 1 





**Figure 2** 













- 812 Figure 4





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Figure 5 824



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Sampling points

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**Figure 6** 





## 857 Figure 7



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874	Figure 8			
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**Figure 9** 





